concept analysis for periodic table

- **≻**Concept label : Groups
- **Concept definition:** The vertical columns in the periodic table
- > Type of concept : concrete

Attribute to :-

- critical: groups, vertical columns, periodic table
- variable: Number of sub shells, Atomic number, metallic properties, non metallic properties

- Concept position :
- > supper ordinate : periodic table
- ➤ Coordinate :periods
- > Subordinate: families

Examples: group I.II,III,...

Non examples: periods 1,2,3,...

Concept analysis for Blocks

- Concept label: blocks
- Concept definition: a block of the periodic table of elements is a set of adjacent groups
- Concept type : concrete
- Attribute to:
- a. Critical: adjacent groups, periodic table, elements, blocks
- b. Variable: electron configuration, atomic orbitals, valence electrons

- Concept position :
- a. Supper ordinate: periodic table
- b. Coordinate: representative elements, transition elements, inner transition elements
- c. Subordinate: metals, non metals, metalloids noble gases,

Example: s-block, p-block, d-blocks, f-blocks

None example: period 1, 2,3,...

Concept analysis for Metals

- Concept label: metals
- Concept definition: elements those are on the left and bottom side of the periodic table
- Type of concept : concrete
- Attribute to
- a. Critical: left and bottom side, periodic table
- Variable: conductivity, magnetic character, hardness, density, melting point, valence electrons, ionization energy, atomic size,

- Concept position :
- a. Supper ordinate: elements
- b. Coordinate: non metals
- c. Subordinate: alkali metals, alkaline earth metals, transition metals, inner transition metals
- Examples: s-block elements with out H, d-block elements, lanthanides, actinides
- Non examples: noble gases, halogens, oxygen family except Po, Te,

Concept analysis for Non metals

- Concept label: non metals
- Concept definition: non metals are the upper right of the periodic table
- Concept type: concrete
- Attribute to:
- a. Critical: upper right, periodic table, non metals
- Variable: state, boiling point, melting point, reactivity, atomic size, electro negativity, electro affinity, non metallic character

- Concept position :
- a. Supper ordinate: elements
- b. Coordinate: metals
- c. Subordinate: none

Examples: noble gases, halogens, oxygen family except Po,Te

Non examples: transition elements, Group I and group 2 elements,

Concept analysis for Metalloids

- Concept label: metalloids
- Concept definition: metaloids have some properties of non metals and metals which are arranged in zigzag steps between the metals and non metals in the periodic table
- Concept type: concrete
- Attribute to:
- a. Critical: properties of metals and non metals, zigzag steps, periodic table, metalloids
- b. Variable: metalic character, groups, periods, atomic number, density, conductivity, Mp, Bp,

- Concept position :
- a. Supper ordinate: elements
- b. Coordinate: non metals and metals
- c. Sub ordinate: none

Examples: silcon, germanium, boron, galium

Non examples: group 1 and 2 elements, halogens, d – block elements, oxygen family except Po, Te

- Concept label: atomic number
- Concept definition: atomic number is the number of protons in the nucleus of an atom that determine the identity of an element
- Type of concept : abstract

Attribute to:-

- Concrete: atomic number, identity of element, number of protons in the nucleus, atom
- Variables: protons,

Concept position:

Supper ordinate: elements

Coordinate:mass number

Subordinate: none

Examples: atomic number of elements

Non examples: number of neutrons of elements , mass number of elements