Name:		

Complete the equation, balance and give the type.

$$\frac{\sqrt[3]{2}-1}{9}$$
. zinc oxide + carbon \Rightarrow zinc + carbon dioxide $\frac{\sqrt[3]{2}-1}{2}$ $\frac{\sqrt[3]{2}-1}{2}$

$$-10.2c + 2H_2O \rightarrow CH_4 + CO_2$$

2 KC 103 -> 2 KC 1 +3 C2

$$-122N_2H_4 + N_2O_4 + N_2 H_2O$$