



Phytochemistry 53 (2000) 785-793

www.elsevier.com/locate/phytochem

Structure and synthesis of ether-linked proteracacinidin and promelacacinidin proanthocyanidins from *Acacia caffra**

Linette Bennie^a, Elfranco Malan^{a, 1}, Johan Coetzee^{a, 1}, Daneel Ferreira^{b,*}

^aDepartment of Chemistry, University of the Orange Free State, P.O. Box 339, Bloemfontein, 9300 South Africa
^bNational Center for Natural Products Research, Research Institute of Pharmaceutical Sciences, School of Pharmacy, The University of Mississippi,
University, MS 38677, USA

Received 26 August 1999; received in revised form 29 November 1999

Abstract

Two new ether-linked proanthocyanidins, epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4β -ol and epimesquitol- $(4\beta \rightarrow 4)$ -epioritin- 4β -ol, were isolated from the heartwood of *Acacia caffra*. Their structures and absolute configurations were established by spectroscopic methods and syntheses. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: Acacia caffra; Leguminosae; Proteracacinidins; Promelacacinidins; Proanthocyanidins; Synthesis

1. Introduction

The natural occurrence of carbon–carbon interflavanoid linked dimeric proanthocyanidins is well documented (Hemingway, 1989; Porter, 1994). The dioxane-type biflavanoids from *Acacia mearnsii* (Drewes & Ilsley, 1969) were unique in possessing a double ether linkage. Their identification was followed by the discovery of A-type biflavanoids with one of the two interflavanyl bonds and an ether linkage (Mayer, Goll, von Arndt & Mannschreck, 1966; Jacques, Haslam, Bedford & Greatbanks, 1974). Recently, a proteracacinidin analogue with a $(4\beta \rightarrow 7)$ ether linkage as well as a $C_5 \rightarrow C_6$ bond between the A- and Drings was reported from *A. caffra* (Malan, Sireeparsad, Burger & Ferreira, 1994).

The first single ether-linked (C_4 –O– C_4) dimeric promelacacinidins were identified in *A. melanoxylon* (Foo, 1989) and were followed by similar bis-teracacinidins (Coetzee, Malan & Ferreira, 1998b), and the first (C_4 –O– C_3) ether-linked bis-teracacinidins (Coetzee, Malan & Ferreira, 1998a) from *A. galpinii*. We now report the structure and synthesis of the new epioritin-($4\beta \rightarrow 3$)-epioritin- 4β -ol 2 and epimesquitol-($4\beta \rightarrow 4$)-epioritin- 4β -ol 10, the first biflavanoid comprising both leucomelacacinidin and leucoteracacinidin constituent units, from *A. caffra*.

2. Results and discussion

The novel proanthocyanidins **2** and **10** were isolated from the MeOH extract of the heartwood of *A. caffra* and were accompanied by *ent*-oritin- $(4\alpha \rightarrow 4)$ -epioritin- 4α -ol, epioritin- $(4\beta \rightarrow 4)$ -epioritin- $(4\beta \rightarrow 3)$ -epioritin- $(4\beta \rightarrow 3)$ -oritin- $(4\beta \rightarrow 3)$ -oritin- $(4\alpha$ -ol and epioritin- $(4\beta \rightarrow 3)$ -oritin- $(4\alpha$ -ol previously isolated from *A. galpinii* (Coetzee et al., 1998a, 1998b). Owing to the complexity of the phenolic mixture, the proanthocyanidins **2** and **10** were purified and identified as the hexa- and hepta-methyl ether diacetate derivatives **3** and **11**, respectively.

^{*} Part 32 in the series 'Oligomeric Flavanoids'. Part 31. Mciteka, L., Coetzee, J., Malan, E., & Ferreira, D. (2000). Structure and synthesis of novel cassiaflavan-epicatechin dimers. *Phytochemistry*, submitted.

^{*} Corresponding author. Tel.: +1-91-601-232-1572; fax: +1-91-601-232-7062.

E-mail address: dferreir@olemiss.edu (D. Ferreira).

¹ Correspondence may also be addressed to Malan or Coetzee.

The structures and relative configurations were established by ¹H-NMR data (Table 1) of derivatives 3 and 11. The presence of two AB- and two AA'BB' spin systems in the spectrum of 3, and two AB-, an ABX- and an AA'BB' system in the spectrum of 11 for the aromatic protons, as well as two AMX-systems for the heterocyclic protons suggested the dimeric nature of the two compounds.

2D COSY experiments differentiated the spin systems and showed the connectivities between aromatic and heterocyclic protons. The presence of six O-methyl- and two O-acetyl resonances for 3, and seven O-methyl- and two O-acetyl signals for 11 in conjunction with FABMS data indicating molecular ions of m/z 730 and m/z 760, were reminiscent of molecular formulae $C_{40}H_{42}O_{13}$ and $C_{41}H_{44}O_{14}$, respectively. The above information suggested ether-type interflavanyl linkages for both compounds.

In oligomeric proanthocyanidin derivatives with C–C interflavanyl linkages, 4-H(C) of the ABC chain extender unit is shielded (1.32–1.82 ppm) relative to the chemical shift of the same proton in the permethylaryl ether 3,4-di-O-acetyl derivative of the flavan-3,4-diol precursor (Steynberg, Steynberg, Brandt, Ferreira & Hemingway, 1997; Young, Brandt, Young, Ferreira & Roux, 1986; Malan & Sireeparsad, 1995). Application of this observation indicated a C₄–O–C₃ ether bond for 3 (CDCl₃) with a shielding of 1.52 ppm for 4-H(C) (δ 4.38) and 0.99 ppm for 3-H(F) (δ 4.25). The same principle was applied to compound 11 (C₆D₆), where 4-H(C) (δ 5.40) and 4-H(F) (δ 5.42) displayed shielding of 1.09 and 0.64 ppm, respectively. These

shielding phenomena were in accordance with the results reported for the promelacacinidins from A. melanoxylon (Foo, 1989), hence, indicating a C_4 –O– C_4 interflavanyl bond for derivative 11.

The coupling constants of the two pairs of heterocyclic systems $[J_{2,3}(C) = J_{2,3}(F) = 1.5 \text{ Hz}; J_{3,4}(C) = J_{3,4}(F) = 3.0 \text{ Hz}$ for both 3 and 11] indicated 2,3-cis-3,4-trans relative configurations for the C- and F-rings of compounds 3 and 11, respectively (Coetzee et al., 1998a, 1998b; Malan & Sireeparsad, 1995; Malan, 1995). The conspicuous absence of NOE associations between 2- and 4-H of the C- and F-rings in both derivatives could be interpreted as confirmation of the 2,4-trans relative configuration for the heterocyclic systems.

Phase sensitive NOESY experiments of 3 showed important associations of 3-H(F) with 3- and 4-H(C) and of 4-H(F) with 5-H(A) and 5-H(D). These associations were verified from Dreiding models and the 3-H(C) association with 2',6'-H(E) is only possible in the event of a C_4 -O- C_3 coupling. The same experiment on derivative 11 showed associations of 4-H(C) with 5-H(D), 4-H(F) with 4-H(C), 5-H(A) and 5-H(D) with 5-H(A), and the 3-H(C) association with 5-H(D) which is only possible for a C_4 -O- C_4 interflavanyl coupling.

In designing a synthetic sequence towards etherlinked proanthocyanidins of types 2 and 10, cognizance had to be taken of the acid-lability of the C-4 benzylic ether functionality (Foo, 1989) and of the previous focus on the synthesis of this class of compounds as permethylaryl ether derivatives (Coetzee et al., 1998a, 1998b). The synthesis of the proteracacinidin 2

Table 1 ¹H-NMR (300 MHz, 296 K) data of compounds **2**, **3**, **4**, **11**^a

Ring	Н	2 (Acetone- d_6)	3 (CDCl ₃)	4 (Acetone- d_6)	11 (C_6D_6)
A	5	6.93 (d, 8.5)	6.76 (d, 9.0)	7.12 (d, 8.5)	7.56 (d, 9.0)
	6	6.50 (d, 8.5)	6.53 (d, 9.0)	6.78 (d, 8.5)	6.57 (d, 9.0)
В	2',6'	7.41 (d, 8.5)	6.87 (d, 9.0)	7.11 (<i>d</i> , 8.5)	
	3',5'	6.82 (d, 8.5)	6.82 (d, 9.0)	6.99 (d, 8.5)	
	2′				7.34 (<i>d</i> , 2.0)
	5′				6.74 (d, 8.5)
	6′				7.30 (<i>dd</i> , 2.0, 8.5)
С	2	5.25 (s, 1.0)	4.64 (br.s, 1.5)	4.68 (br.s, 1.5)	5.82 (br.s, 1.5)
	3	3.89 (dd, 1.0, 3.5)	4.42 (dd, 1.5, 3.0)	4.57 (dd, 1.5, 3.0)	6.04 (dd,1.5,3.0)
	4	4.65 (d, 3.5)	4.38 (d, 3.0)	4.66 (d, 3.0)	5.40 (d, 3.0)
D	5	6.50 (d, 8.5)	7.22 (d, 9.0)	7.52 (d, 8.5)	7.60 (d, 9.0)
	6	6.40 (d, 8.5)	6.63 (d, 9.0)	6.92 (d, 8.5)	6.58 (d, 9.0)
E	2',6'	7.40 (d, 8.5)	7.56 (d, 9.0)	7.71 (d, 8.5)	7.67 (d, 9.0)
	3',5'	6.68 (d, 8.5)	6.97 (d, 9.0)	7.29 (d, 8.5)	6.95 (d, 9.0)
F	2	5.44 (s, 1.0)	5.32 (br.s, 1.5)	5.48 (s, 1.5)	5.82 (br.s, 1.5)
	3	5.84 (dd, 1.0, 4.0)	4.25 (dd, 1.5, 3.0)	4.51 (dd, 1.5, 3.0)	6.00 (dd, 1.5, 3.0)
	4	5.46 (d, 4.0)	6.17 (d, 3.0)	6.27 (d, 3.0)	5.42 (d, 3.0)
OMe		, , , ,	3.99, 3.98, 3.95, 3.94, 3.93, 3.80		4.01, 4.00, 3.59, 3.49, 3.48, 3.47, 3.39
OAc			2.20, 1.83	2.32, 2.28 (×2), 2.26, 2.21, 2.19, 2.12, 1.76	1.55, 1.54

^a Splitting patterns and *J* values (Hz) are given in parentheses.

was, thus, attempted by using silver tetrafluoroborate (AgBF₄) to activate the free phenolic epioritin-4β-ol 1 towards self-condensation. Treatment of epioritin-4βol 1 with AgBF₄ in THF at 0°C (Scheme 1) gave a complex mixture which had to be acetylated to enable separation. Compounds 2 and 5 were, therefore, purified and identified as their per-O-acetyl derivatives 4 and 7. The C-C-linked analogue 7 had ¹H-NMR and CD data identical to those previously reported (Malan & Sireeparsad, 1995). The ether-linked derivative 4 was subjected to mild alkaline hydrolysis to obtain the free phenolic proteracacinidin 2 (¹H-NMR, Table 1). The latter was methylated and acetylated to obtain the hexa-O-methyl-di-O-acetyl derivative 3 with ¹H-NMR and CD spectra identical to those of the same derivative of the natural product, hence, confirming the 2R,3R,4S(C):2R,3R,4S(F) absolute configuration of the novel C₄–O–C₃ linked proteracacinidin 2.

The stereochemical course of the coupling step is

explicable in terms of a neighboring group mechanism triggered by interaction of the Lewis acid and the near-axial C-4 hydroxyl group of the flavan-3,4-diol 1 (Coetzee et al., 1998a, 1998b). We could not detect any of the anticipated C_4 –O– C_4 isomer, probably as a result of the low overall yields. The proteracacinidin 2 is apparently stable towards racemization at either C-2 or C-4 of both the C- and F-rings under the mild hydrolytic conditions (1% KOH in MeOH at ca. 20°C) used for its generation from the peracetate 4. In the absence of AgBF₄, epioritin-4 β -ol 1 did not undergo self-condensation, hence, indicating that the Lewis-acid is a prerequisite for product formation.

In a second approach (Scheme 2), epimesquitol- 4α ol **8** was converted into the 4β -benzylsulfanyl derivative **9** via an adapted thiolytic procedure (Hemingway,
Karchesy, McGraw & Wielesek, 1983). Compound **9**was then activated by the thiophilic Lewis-acid, AgBF₄
(Steynberg, Nel, van Rensburg, Bezuidenhoudt & Fer-

Scheme 1. Synthesis of proteracacinidins 2 and 5.

reira, 1998). When the thioether 9/AgBF₄ complex was treated with epioritin- 4β -ol 1, the latter compound acted as ambient nucleophile to give both the C₄-O-C₄ and C-C coupled analogues 10 (4.1%) and 5 (2.1%), respectively. Activation of the near-axial C₄-S bond of 9 by AgBF₄ once again triggers a neighboring group mechanism (Coetzee et al., 1998a, 1998b), hence, explaining the stereoselective formation of compounds 5 and 10. Owing to the complexity of the mixture, dimers 5 and 10 were identified as their methyl ether acetate derivatives 6 and 11, the additional chromatographic steps offered by derivatization being a pre-requisite for sample purity. The permethylaryl tri-O-acetyl derivative 6 showed ¹H-NMR and CD data identical to those previously reported (Malan & Sireeparsad, 1995). The hepta-O-methyldi-O-acetyl deriva-

tive 11 of the ether-linked analogue 10 was identical to the same derivative of the natural product 11 by comparison of 1 H-NMR (Table 1) and CD data. It thus possesses 2R,3R,4S(C):2R,3R,4S(F) absolute configuration.

Pursuing our studies on the synthesis of free phenolic proanthocyanidins, the $AgBF_4$ catalyzed process was applied to a mixture of the epioritin-4 β - and 4 α -ols 1 and 12 (Scheme 3) with a view to form epioritin-(4 $\beta \rightarrow 4$)-epioritin-4 α -ol 13 and epioritin-(4 $\beta \rightarrow 3$)-epioritin-4 α -ol 16, previously synthesized as permethylaryl ether diacetates 14 and 17 (Coetzee et al., 1998a, 1998b). Both the ether-linked compounds 13 and 16 indeed formed and were purified and identified as the octa-O-acetyl derivatives 15 (9.1%) and 18 (7.8%) (Table 2). They were accompanied by the C–C-linked

Scheme 2. Synthesis of the 'mixed' promelacacinidin/proteracacinidin 10.

Scheme 3. Synthesis of ether-linked proteracacinidins 13 and 16 and the C-C coupled analogue 19.

compound, epioritin- $(4\beta \rightarrow 6)$ -epioritin- 4α -ol **19** (Malan & Sireeparsad, 1995), identified as the per-O-acetyl derivative **20** (7.3%). Derivatives **15** and **18** were subjected to mild alkaline hydrolysis (1% KOH in

MeOH) yielding for the first time the free phenolic proanthocyanidins **13** and **16**, respectively (¹H-NMR, Table 2). Subsequent methylation and acetylation gave the hexa-*O*-methylether diacetates **14** and **17** with ¹H-

Table 2 1 H-NMR (300 MHz, 296 K) data of compounds **13**, **15**, **16**, **18** a

Ring	Н	13 (Acetone- d_6)	15 (CDCl ₃)	16 (Acetone- d_6)	18 (CDCl ₃)
A	5	6.66 (d, 8.5)	7.28 (d, 8.5)	6.96 (d, 8.5)	6.92 (d, 8.5)
	6	6.45 (d, 8.5)	6.91 (<i>d</i> , 8.5)	6.57 (d, 8.5)	6.72 (d, 8.5)
В	2',6'	7.39 (d, 8.5)	7.51 (d, 8.5)	7.39 (d, 8.5)	7.08 (d, 8.5)
	3′,5′	6.83 (d, 8.5)	7.11 (d, 8.5)	6.84 (d, 8.5)	7.01 (d, 8.5)
	2'				
	5′				
	6′				
С	2	5.49 (s, 1.0)	5.43 (<i>br.s</i> , 1.5)	5.37 (s, 1.0)	4.89 (br.s, 1.5)
	3	5.85 (<i>dd</i> , 1.0, 4.0)	5.28 (dd, 1.5, 3.0)	5.23 (<i>dd</i> , 1.0, 3.5)	4.72 (dd, 1.5, 3.0)
	4	5.47 (d, 4.0)	4.75 (d, 3.0)	4.70 (d, 3.5)	4.08 (d, 3.0)
D	5	6.96 (d, 8.5)	7.16 (<i>d</i> , 8.5)	6.66 (d, 8.5)	7.11 (<i>d</i> , 8.5)
	6	6.57 (d, 8.57)	6.81 (d, 8.5)	6.45 (d, 8.5)	6.85 (d, 8.5)
E	2',6'	7.37 (d, 8.5)	7.47 (d, 8.5)	7.37 (d, 8.5)	7.61 (d, 8.5)
	3',5'	6.86 (d, 8.5)	7.14 (d, 8.5)	6.86 (d, 8.5)	7.18 (d, 8.5)
F	2	5.37 (s, 1.0)	5.47 (<i>br.s</i> , 1.0)	5.49 (s, 1.0)	5.45 (br.s, 1.0)
	3	5.23 (dd, 1.0, 3.5)	5.88 (dd, 1.0, 4.0)	5.86 (dd, 1.0, 4.0)	4.69 (dd, 1.0, 4.0)
	4	5.70 (d, 3.5)	5.39 (d, 4.0)	5.47 (d, 4.0)	6.32 (d, 4.0)
OMe					
OAc			2.35, 2.33, 2.32, 2.32, 2.30, 2.28, 1.95, 1.86		2.32, 2.31, 2.31, 2.30, 2.28, 2.24, 1.99, 1.74

 $^{^{\}rm a}$ Splitting patterns and J values (Hz) are given in parentheses.

NMR and CD data identical to those previously reported (Coetzee et al., 1998a, 1998b). Formation of the ether-linked analogues **13** and **16** is explicable in terms of activation of the more reactive axial C-4 hydroxyl group of epioritin-4 β -ol **1** (Clark-Lewis & Williams, 1967), epioritin-4 α -ol **12** with its remarkable stable equatorial C-4 hydroxyl group (Clark-Lewis & Williams, 1967; Coetzee, Malan & Ferreira, 1999) then serving as the ambient nucleophile.

Although the yields of the ether-linked proanthocyanidins 2, 10, 13 and 16 were low, the AgBF₄ catalyzed process of activation of either the flavan-3,4-diol or its C-4 benzylsulfanyl derivative has merit, since it produced these unique ethers in free phenolic form for the first time.

3. Experimental

¹H-NMR spectra were recorded on a Bruker AM-300 spectrometer for solutions as indicated, with Me₄Si as internal standard. FAB mass spectra were recorded on a VG-70E instrument with a VG 11-250J data system and an iontech saddlefield FAB gun. TLC was performed on precoated Merck plastic sheets (silica gel 60 PF₂₅₄, 0.25 mm) and the plates were sprayed with H₂SO₄-HCHO (40:1, v/v) after development. Preparative plates (PLC) $[20 \times 22 \text{ cm}, \text{Kieselgel}]$ PF₂₅₄ (1.0 mm)] were air dried and used without prior activation. Column chromatography was done on Sephadex LH-20 in various columns, solvent systems and flow rates (to be specified in each instance). Methylations were performed with an excess of diazomethane in MeOH-Et₂O over a period of 48 h at -15° C, while acetylations were conducted in Ac₂Opyridine at ambient temperature. Evaporations were done under reduced pressure at ambient temperature in a rotary evaporator, and freeze drying of aqueous solutions on a Virtis 12 SL freezemobile.

3.1. General procedure for the alkaline hydrolysis

The compound was dissolved in a 1% KOH in MeOH solution (2 ml) under an Ar-atmosphere at room temperature and stirred for 10 min. The reaction was quenched by the addition of H_2O (10 ml) and the pH was adjusted to 6.5 (10% HCl) before it was extracted with EtOAc (3 × 10 ml). The combined layers were dried (Na₂SO₄) and the EtOAc was evaporated under reduced pressure at ambient temperature.

3.2. Isolation of phenolic compounds

Drillings (4.3 kg) from the heartwood of *A. caffra* were repeatedly extracted with MeOH (3×2.5 l) for 24 h periods at room temperature (23° C). The extract

was concentrated by evaporating the MeOH under vacuum at 35° C. The concentrate was dissolved in H_2O and freeze dried to give a pale brown powder (447 g).

A portion (20.5 g) of the extract was separated on a Sephadex LH-20/EtOH column (6 × 180 cm) with a flow rate of 1 ml/min, 32 min fractions. The first 1.5 l of EtOH was discarded and the fractions of two columns (41 g) were combined as follows: A (tubes 33– 41, 175 mg), B (42–48, 131 mg), C (65–69, 299 mg), D (71–77, 389 mg), E (81–87, 218 mg), F (89–101, 652 mg), G (102-109, 810 mg), H (113-125, 366 mg), I (129–134, 1.332 g), J (153–161, 1.057 g), K (163–175, 789 mg), L (177-191, 544 mg), M (192-208, 1.161 g), N (209–221, 1.834 g), O (222–235, 926 mg), P (239– 276, 2.291 g), Q (277-307, 769 mg), R (309-315, 1.138 g), S (317–331, 728 mg), T (333–354, 756 mg), U (355-372, 573 mg), V (373-389, 319 mg), W (390-465, 840 mg), X (467–503, 839 mg), Y (504–514, 308 mg), Z (515–535, 219 mg), AA (537–555, 188 mg), BB (557–576, 407 mg), CC (577–633, 952 mg), DD (637– 659, 290 mg), EE (660-670, 482 mg), FF (671-764, 909 mg), GG (765-813, 896 mg), HH (814-904, 830 mg), II (905-948, 476 mg), JJ (949-998, 679 mg), KK (999-1043, 674 mg), LL (1044-1115, 754 mg), MM (1116–1139, 207 mg), NN (1140–1230, 683 mg), OO (1231-1259, 505 mg), PP (1260-1300, 531 mg), QQ (1309–1130, 261 mg), RR (1131–1389, 345 mg), SS (1390-1429, 189 mg), TT (1430-1440, 117 mg), UU (1441-1480, 158 mg), column residue (812 mg).

3.3. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4β -ol hexa-O-methylether diacetate 3

Methylation of a portion (200 mg) of fraction S followed by PLC in benzene–Me₂CO–EtOAc (7:2:1, v/v) afforded five bands at $R_{\rm f}$ 0.43 (17.4 mg), 0.34 (13.3 mg), 0.31 (22.5 mg), 0.21 (21.1 mg) and 0.13 (14.1 mg). Acetylation of the $R_{\rm f}$ 0.31 band (22.5 mg) followed by PLC in CHCl₃–Et₂O (49:1, ×3, v/v) gave compound 3 ($R_{\rm f}$ 0.33, 9.7 mg) as a white amorphous solid. (Found: M⁺, 730.2702. C₄₀H₄₂O₁₃ requires M, 730.2704); $\delta_{\rm H}$ (Table 1); CD [θ]_{274.8} –1492, [θ]_{273.5} –5886 and [θ]_{221.8} 2182. The remaining bands contain related proanthocyanidin-type compounds which will be dealt with elsewhere.

3.4. Epimesquitol- $(4\beta \rightarrow 4)$ -epioritin- 4β -ol hepta-O-methylether diacetate 11

A portion (200 mg) of fraction BB was methylated and the mixture separated by PLC in benzene–Me₂CO–EtOAc (7:2:1, \times 2, v/v) to give six bands at $R_{\rm f}$ 0.72 (11.8 mg), 0.67 (5.1 mg), 0.58 (10.3 mg), 0.49 (13.2 mg), 0.38 (5.9 mg) and 0.28 (2.3 mg). Acetylation of the $R_{\rm f}$ 0.38 band followed by PLC in CHCl₃–Et₂O

(49:1, ×2, v/v) afforded compound **11** ($R_{\rm f}$ 0.48, 2.1 mg) as a *white amorphous solid*. (Found: M⁺, 760.2731. C₄₁H₄₄O₁₄ requires M, 760.2731); $\delta_{\rm H}$ (Table 1); CD [θ]₂₈₂₀ 790.2, [θ]_{243.9} – 72,410, [θ]_{233.1} 4251 and [θ]_{219.4} 1788. The quantities and complexity of the remaining bands did not merit further investigation.

3.5. Synthesis of proteracacinidin derivatives 2-4

Epioritin-4β-ol 1 (75 mg) was dissolved in dry THF (10 ml). AgBF₄ (0.8 eq.) was added under N₂ and the mixture was stirred at 0°C. After 90 min, the reaction was quenched by the addition of H₂O (50 ml) and the mixture was extracted with EtOAc (6 × 30 ml). The solvent was removed under vacuum and separation of the mixture by PLC in benzene–Me₂CO–MeOH (6:3:1, ×2, v/v) gave three main bands at R_f 0.72 (3.8 mg), 0.64 (6.5 mg) and 0.54 (13.1 mg). Acetylation of the R_f 0.54 band followed by an aqueous workup and subsequent PLC in benzene–Me₂CO (9:2, ×2, v/v) gave two bands at R_f 0.40 (4.2 mg) and 0.28 (3.8 mg).

3.6. Epioritin- $(4\beta \rightarrow 6)$ -epioritin- 4β -ol nona-acetate 7

The $R_{\rm f}$ 0.40 band was identified as the nona-acetate derivative 7 with physical data identical to those in the literature (Malan & Sireeparsad, 1995).

3.7. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4β -ol octa-acetate **4**

The $R_{\rm f}$ 0.28 band yielded the octa-acetate derivative of the C_4 –O– C_3 -linked compound 4 as a *white amorphous solid*. $\delta_{\rm H}$ (Table 1). FABMS, ${\rm M}^+$, 898.2322. $C_{46}{\rm H}_{42}{\rm O}_{19}$ requires ${\rm M}^+$ 898.2321.

3.8. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4β -ol 2

Alkaline hydrolysis of derivative **4** yielded the free phenolic compound **2** as a *light brown amorphous solid*. (Found: M^+ , 562.1474. $C_{30}H_{26}O_{11}$ requires M, 562.1475); δ_H (Table 1).

3.9. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4β -ol hexa-O-methylether diacetate 3

Subsequent methylation and acetylation of the free phenolic compound 2 yielded a compound with ¹H-NMR, CD and MS data identical to those of the natural product derivative 3.

3.10. Synthesis of (2R,3S,4S)-2,3-cis-3,4-trans-3,3',4',7,8-pentahydroxy-4-benzylsulfanylflavan **9**

Epimesquitol- 4α -ol **8** (1.0 g) was dissolved in EtOH (100 ml) and an excess of toluene- α -thiol (3 ml) and

HOAc (4 ml) were added, while the solution was purged with Ar gas. The solution was transferred to a Ar-purged vial and sealed. After 24 h at 100°C in a steam bath, the solvent was removed under a stream of N_2 . The residue was transferred into a container with H₂O (100 ml) and the excess toluene-α-thiol was removed by washing with hexane $(2 \times 50 \text{ ml})$. The H_2O -layer was extracted with Et_2O (5 × 20 ml). The Et₂O was removed under a stream of N₂ and the oil was applied to a Sephadex LH-20 column (4 × 150 cm) with the use of a small amount of EtOH. The column was eluted with CHCl3-EtOH (4:1), at a flow rate of 1 ml/min and 16 ml fractions were collected. The fractions were combined as follows: fraction A (tubes 58–72, 100 mg), fraction B (tubes 73–170, 3.01 g), fraction C (tubes 181–229, 208 mg).

Fraction C comprised the remaining epimesquitol- 4α -ol **8**. Fraction B yielded the title compound **9** as a white amorphous powder. $\delta_{\rm H}$ (Acetone- d_6) 7.48–7.25 (5H, m, D-ring), 7.35 (2H, d, 9.0, H-2',6'), 6.83 (2H, d, 9.0, H-3',5'), 6.53 (1H, d, 9.0, 9.0, H-5), 6.433 (1H, d, 9.0, H-6), 5.33 (1H, br.s, 1.0, H-2), 4.07 (1H, dd, 1.0, 3.0, H-3), 4.04 (1H, d, 14.0, -CH₂-), 3.95 (1H, d, 3.0, H-4), 3.93 (1H, d, 14.0, -CH₂-).

3.11. Synthesis of epimesquitol- $(4\beta \rightarrow 4)$ -epioritin- 4β -ol hepta-O-methylether diacetate 11

Epimesquitol-4β-thiobenzylether 9 (100 mg) and epioritin-4 β -ol 1 (141 mg, 2.0 eq.) were dissolved in dry THF (20 ml) at 0°C and AgBF₄ (118 mg, 2.5 eq.) was added in one portion. After the reaction was stirred under N₂ for 90 min at 0°C, it was quenched with H_2O (80 ml) and extracted with EtOAc (6 × 50 ml). The solution was dried (Na₂SO₄), the solvent was evaporated and the resulting mixture methylated and then acetylated. PLC in benzene-Me₂CO (9:1, v/v) afforded three bands at $R_{\rm f}$ 0.62 (46.9 mg), 0.55 (23.5 mg) and 0.39 (22.6 mg). The $R_{\rm f}$ 0.62 band gave epioritin-4β-ol tri-O-methylether diacetate. After further PLC in CHCl₃-hexane-Me₂CO (45:3:2, v/v), the R_f 0.55 band afforded epioritin- $(4\beta \rightarrow 6)$ -epioritin- 4β -ol hexa-O-methylether triacetate 6 (R_f 0.60, 8.7 mg) with physical data identical to those in the literature (Malan & Sireeparsad, 1995).

3.12. Epimesquitol- $(4\beta \rightarrow 4)$ -epioritin- 4β -ol hepta-O-methylether diacetate 11

Further PLC separation in CHCl₃-hexane–Me₂CO (45:3:2, v/v) of the initial $R_{\rm f}$ 0.39 band gave compound 11 ($R_{\rm f}$ 0.58, 7.2 mg) as a *light brown amorphous solid* with ¹H-NMR, CD and MS data identical to those of the natural product derivative 11.

3.13. Synthesis of dimeric ethers 13 and 16

Epioritin-4α-ol **12** (160 mg) and epioritin-4β-ol **1** (80 mg) were dissolved in dry THF (20 ml) at 0°C and AgBF₄ (134 mg, 2.5 eq.) was added. The mixture was stirred for 90 min at 0°C under N₂ before H₂O (100 ml) was added. The mixture was extracted with Et₂O (6 × 50 ml) and the combined ether layers dried (Na₂SO₄) before evaporation under reduced pressure. PLC separation in benzene–Me₂CO–MeOH (6:3:1, ×2, v/v) gave four bands at R_f 0.75 (88.6 mg), 0.73 (36.9 mg), 0.66 (17.0 mg) and 0.55 (20.9 mg). The R_f 0.75 and 0.73 bands yielded remaining starting materials epioritin-4α-ol **12** and epioritin-4β-ol **1**, respectively.

3.14. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4α -ol octa-acetate 18

Acetylation of the $R_{\rm f}$ 0.66 band followed by PLC in benzene–Me₂CO (9:1, ×2, v/v) afforded derivative **18** ($R_{\rm f}$ 0.30, 10.5 mg). $\delta_{\rm H}$ (Table 2).

3.15. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4α -ol **16**

Alkaline hydrolysis of derivative **18** yielded the free phenolic compound **16** as a *light brown amorphous solid*. (Found: M^+ , 562.1477. $C_{30}H_{26}O_{11}$ requires M, 562.1475); δ_H (Table 2).

3.16. Epioritin- $(4\beta \rightarrow 3)$ -epioritin- 4α -ol hexa-O-methylether diacetate 17

Subsequent methylation and acetylation of free phenolic compound **16** yielded derivative **17** with physical data identical to those in the literature (Coetzee et al., 1998a).

3.17. Epioritin- $(4\beta \rightarrow 6)$ -epioritin- 4α -ol nona-acetate **20**

Acetylation of the initial $R_{\rm f}$ 0.55 band followed by PLC in benzene–Me₂CO (99:1, ×2, v/v) gave two bands at $R_{\rm f}$ 0.47 (10.7 mg) and 0.35 (8.4 mg). The 0.47 band yielded derivative **20** with ¹H-NMR, CD and MS data identical to those in the literature (Malan & Sireeparsad, 1995).

3.18. Epioritin- $(4\beta \rightarrow 4)$ -epioritin- 4α -ol octa-acetate 15

The $R_{\rm f}$ 0.35 band was subjected to further PLC in hexane–benzene–Me₂CO–MeOH (43:42:10:5, ×3, v/v) and gave one prominent band at $R_{\rm f}$ 0.38 (6.1 mg) which was identified as the per-*O*-acetyl derivative **15**. $\delta_{\rm H}$ (Table 2).

3.19. Epioritin- $(4\beta \rightarrow 4)$ -epioritin- 4α -ol 13

Alkaline hydrolysis of derivative 15 afforded the free

phenolic compound **13** as a *light brown amorphous* solid. (Found: M^+ , 562.1478. $C_{30}H_{26}O_{11}$ requires M, 562.1475); δ_H (Table 2).

3.20. Epioritin- $(4\beta \rightarrow 4)$ -epioritin- 4α -ol hexa-O-methylether diacetate **14**

Subsequent methylation and acetylation of free phenolic compound 13 gave a derivative with ¹H-NMR, CD and MS data identical to those in the literature (Coetzee et al., 1998b).

Acknowledgements

Financial support by the Foundation for Research Development, Pretoria and by the 'Sentrale Navorsingsfonds' of UOFS is acknowledged. This work was supported in part by the United States Department of Agriculture, Agricultural Research Service Specific Cooperative Agreement No. 58-6408-7-012.

References

- Clark-Lewis, J. W., & Williams, L. R. (1967). Flavan derivatives. Part XVII: Epimerization of the benzylic 4-hydroxyl group in flavan-3,4-diols and the formation of 4-alkyl ethers by solvolysis. Aust. J. Chem, 20, 2152–2167.
- Coetzee, J., Malan, E., & Ferreira, D. (1998a). Oligomeric flavanoids. Part XXVIII: Structure and synthesis of ether-linked [4-O-3]-bis-teracacinidins, a novel class of naturally occurring proanthocyanidins. *Journal of Chemical Research* (S), 526–527; (M), 2287–2296.
- Coetzee, J., Malan, E., & Ferreira, D. (1998b). Oligomeric flavanoids. Part XXIX: Structure and synthesis of novel ether-linked [4-O-4] bis-teracacinidins. *Tetrahedron*, 54(31), 9153–9160.
- Coetzee, J., Malan, E., Ferreira, D. (1999). The formation and stability of flavans with 2,3-cis-3,4-cis configuration. *Tetrahedron*, 55, 999–1004.
- Drewes, S. E., & Ilsley, A. H. (1969). Dioxan-linked biflavanoid from the heartwood of *Acacia mearnsii*. *Journal of the Chemical Society* (C), 897–900.
- Foo, L. Y. (1989). Isolation of [4-O-4]-linked biflavanoids from Acacia melanoxylon: first examples of a new class of single etherlinked proanthocyanidin dimers. *Journal of the Chemical Society*, Chemical Communications, 20, 1505–1506.
- Hemingway, R. W. (1989). In J. W. Rowe, *Natural products of woody plants 1* (p. 571). New York: Springer–Verlag.
- Hemingway, R. W., Karchesy, J. J., McGraw, G. W., & Wielesek, R. S. (1983). Heterogeneity of interflavanoid bond location in loblolly pine bark procyanidins. *Phytochemistry*, 22(1), 275–281.
- Jacques, D., Haslam, E., Bedford, G.R., & Greatbanks, D. (1974).
 Plant proanthocyanidins. Part II: Proanthocyanidin A2 and its derivatives. *Journal of the Chemical Society, Perkin Transactions* 1, 2663–2671.
- Malan, E. (1995). A ($4\beta \rightarrow 5$)-linked proteracacinidin dimer from the heartwood of *Acacia caffra*. *Phytochemistry*, 40(5), 1519–1521.
- Malan, E., & Sireeparsad, A. (1995). The structure and synthesis of the first dimeric proteracacinidins from *Acacia galpinii*. *Phytochemistry*, 38(1), 237–239.
- Malan, E., Sireeparsad, A., Burger, J. F. W., & Ferreira, D. (1994).

- A novel doubly-linked proteracacinidin analogue from *Acacia caffra*. *Tetrahedron Letters*, 35(40), 7415–7416.
- Mayer, W., Goll, L., von Arndt, E.M., & Mannschreck, A. (1966). Procyanidino-(-)-epicatechin, ein zweiarmig verknüpftes, kondensiertes proanthocyanidin aus Aesculus hippocastanum. Tetrahedron Letters, 429–435.
- Porter, L. J. (1994). In J. B. Harborne, The flavonoids. Advances in research since 1986 (p. 23). London: Chapman & Hall.
- Steynberg, P. J., Nel, R. J. J., van Rensburg, H., Bezuidenhoudt, B. C. B., & Ferreira, D. (1998). Oligomeric flavanoids. Part XXVII: Interflavanyl bond formation in procyanidins under neutral conditions. *Tetrahedron*, 54(28), 8153–8158.
- Steynberg, P.J., Steynberg, J.P., Brandt, E.V., Ferreira, D., Hemingway, R.W. (1997). Oligomeric flavanoids. Part XXVI: Structure and synthesis of the first profisetinidins with epifisetinidol constituent units. *Journal of the Chemical Society, Perkin Transactions* 1, 1943–1950.
- Young, E., Brandt, E.V. Young, D.A., Ferreira, D., Roux, D.G. (1986). Synthesis of condensed tannins. Part XVII: Oligomeric (2R,3S)-3,3'4',7,8-pentahydroxyflavans: Atropisomerism and conformation of biphenyl and *m*-terphenyl analogues from *Prosopis glandulosa* ('mesquite'). *Journal of the Chemical Society, Perkin Transactions 1*, 1737–1749.