XII PHYSICS FORMULAE SHEET

HEAT

Section	Topic	Formula
	Conversion of Fahrenheit into Celsius	$T_c = \frac{5}{9} (T_F - 32)$
	Conversion of Celsius into Fahrenheit	$T_F = \frac{9}{5}T_c + 32$
Section – I	Conversion of Kelvin into Celsius	$T_{\rm c} = T_{\rm K} - 273$
Heat and Temperature	Conversion of Celsius into Kelvin	T _K = T _C + 273
	Conversion of Kelvin into Fahrenheit	$T_F = \frac{9}{5} (T_k - 273) + 32$
	Conversion of Fahrenheit into Kelvin	$T_K = \frac{s}{9} (T_F - 32) + 273$
	Linear Thermal Expansion	$\alpha = \frac{\Delta L}{L_0 \Delta T} \& L' = L_0 (1 + \alpha \Delta T)$
Section - II Thermal Expansion	Volumetric Thermal Expansion	$\beta = \frac{\Delta V}{V_o \Delta T} \& V' = V_o (1 + \beta \Delta T)$
	Relation between linear and volumetric expansion	β = 3 α
	•	PV = K
	Boyle's Law	$P_1V_1 = P_2V_2$
	boyle 3 Law	$P_1V_1 = P_2V_2$ $\frac{P_1V_1}{P_2V_2} = \frac{P_2V_2}{P_2V_2}$
Section – III		m ₁ m ₂ V/T = K
Gas Laws	Charles Law	The state of the s
	Charles Law	$\frac{V_1}{T_2} = \frac{V_2}{T_2}$
	General Gas Equation	P V = nRT
	Kinetic Gas Equation	$P = 1/3 \rho V^2$
Section – IV Kinetic Molecular Theory of Gases	Relation between Kinetic Energy and Temperature	$\frac{3}{2} kT = \frac{1}{2} mV^2 \text{ (For one molecule)}$ $K.E. \text{ per mole} = \left(\frac{3}{2} kT\right) \times N_0$
		$V_{rms} = \sqrt{\frac{3kT}{m}}$
	Root mean square velocity	$V_{rms} = \sqrt{\frac{3RT}{M}}$ $V_{rms} = \sqrt{\frac{3P}{\rho}}$
	Isoshavia Drossas	102.52
	Isochoric Process Isobaric Process	$\Delta Q = \Delta U$ $\Delta Q = \Delta U + P\Delta V$
Section – V	Isothermal Process	$\Delta Q = \Delta U + P \Delta V$ $\Delta Q = \Delta W$
1 st Law of Thermodynamics	Adiabatic Process	$\Delta U = \Delta W$ $\Delta U = -\Delta W$
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	Heat Capacity	$Q = c\Delta T$
	Specific Heat Capacity	$C = \frac{Q}{m\Delta T}$
	Molar Specific heat Capacity	$C = \frac{Q}{n\Delta T}$
Section VI	MSHC at constant pressure	$Q_p = nCp\Delta T$
Heat Capacity	MSHC at constant volume	$Q_v = nCv\Delta T$
	Relation between 2 capacities	Cp - Cv = R
	Specific heat ratio	$\frac{C_p}{C_n} = \gamma$
	Latent Heat	Q = mH _f Q = mHv
Section VII 2 nd Law of Thermodynamics	Work done	$\Delta W = Q_1 - Q_2$
	Efficiency	$ \eta = \frac{\text{Work done}}{\text{Heat supplied}} \text{ or } \eta = \frac{Q_1 - Q_2}{Q_1} $
	% Efficiency	$\% \eta = \frac{Q_1 - Q_2}{Q_1} \times 100$
	% Efficiency	$\% \eta = \left[1 - \frac{Q_2}{Q_1}\right] \times 100$
	% Efficiency	$\% \eta = \left[1 - \frac{T_2}{T_1}\right] \times 100$
	Entropy	$\Delta S = \frac{\Delta Q}{T}$

ELECTROSTATICS

Section	Topic	Formula
Section I – Concept of Charge	Quantization of Charge	q = ne
and Coulomb's Law	Coulomb's Law	$F = k \frac{q_1 q_2}{r^2}$
Section II – Electric Field	In terms of test charge	$E = \frac{F}{q_o}$
	In terms of source charge	$E = \frac{k q}{r^2}$
	D 6 141	$\Delta \phi = \vec{E} \cdot \vec{A}$
	Definition	$\Delta \phi = EAcos\theta$
	Flux Density	$E = \frac{\Delta \phi}{A}$
	Gauss Law	$\Delta \phi = \frac{q}{\epsilon_0}$
Section III – Electric Flux and Gauss Law	Due to positively charged sheet	$E = \frac{\delta}{2\varepsilon_0}$
	Between two oppositely charged sheets	$E = \frac{\delta}{\epsilon_0}$
	Outside a charged sphere	$E = \frac{\delta}{\varepsilon_0} \left[\frac{a}{r} \right]^2 \text{ or } E = \frac{kq}{r^2}$
	On the surface of charged sphere	$E = \frac{\delta}{\epsilon_0}$
	Inside a charged sphere	E = 0

Section IV – Electric Potential	Definition	$\Delta V = \frac{\Delta W}{q}$
	Potential gradient	$E = \frac{\Delta V}{\Delta r}$
	Absolute potential	$V = \frac{kq}{r}$
	Relation b/w K.E. and Electric potential	K.E. = $q_o \Delta V$
Section V - Capacitors	Capacitance	$C = \frac{q}{V}$
	Capacitance when air is between plates	$C = \frac{A \epsilon_0}{d}$
	Capacitance in the presence of dielectric	$C' = \frac{A \epsilon_0 \epsilon_r}{d}$ or $C' = C \epsilon_r$
	Parallel combination	$C = C_1 + C_2 + C_3 + \dots$
	Series combination	$\frac{1}{C} = \frac{1}{C_1} + \frac{1}{C_2} + \frac{1}{C_3} + \dots$
	Compound capacitor	$C_{e} = \frac{\varepsilon_{0} A}{(d-t) + \frac{t}{\varepsilon_{r}}}$

CURRENT ELECTRICITY

Section	Topic	Formula
Section I – Electric Current	Definition	$I = \frac{q}{t}$
section 1 – Electric Current	Deninition	$I = \frac{ne}{t}$
Section II – Ohm's Law	In terms of Resistance	V = IR
Section II – Onm s Law	In terms of conductance	I = kV (k=conductance)
	In terms of Resistivity	$R = \frac{\rho L}{A}$
	Effect of Temperature	$\alpha = \frac{R_t - R_o}{R_o \Delta T}$
Section III – Concept of	THE RESIDENCE OF THE PROPERTY	$R_t = R_o (1 + \alpha \Delta T)$
Resistance and	Series Combination	R _e = R1 + R2 + R3 +
Conductance	Series Combination	$R_s = nR$
terrations to restrict an advantage of the control	Parallel Combination	$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3} + \dots$
		$R_p = \frac{R}{n}$
Section IV – Power Dissipation	Power Dissipation	$P = VI = I^2R = \frac{V^2}{R}$
	Heat Energy Dissipation	$H = VIt = I^2Rt = \frac{V^2}{R}t$
Section V – Concept of		EMF = T.P.D. + Loss Voltage
EMF, Terminal Potential	Definition	$\varepsilon = IR + Ir$
Difference and Loss		$\varepsilon = 1 (R + r)$
Voltage	Discharging Battery	V = E - Ir
CONTRACTOR AND CONTRACTOR	Charging Battery	V = E + Ir

Magnetism and Electromagnetism

Section	Topic	Formula
Section I – Introduction to Electromagnetism and Magnetic force	Force on a charge in magnetic field	$F = qVB \sin\theta$
	Force on a current carrying conductor in magnetic field	$F = BIL \sin\theta$
	e/m ratio in terms of Velocity	$\frac{e}{m} = \frac{V}{Br}$
	Radius of circular path	$r = \frac{b^2}{2a}$
Section II – Charge to Mass	Velocity by potential difference method	$V = \sqrt{\frac{2\Delta Ve}{m}}$
Ratio of Electron	Velocity selector method	$V = \frac{E}{B} = \frac{\Delta V}{dB}$
	e/m in terms of potential	$\frac{e}{m} = \frac{2\Delta V}{B^2 r^2}$
	e/m in terms of electric field intensity	$\frac{\mathbf{e}}{\mathbf{m}} = \frac{\mathbf{E}}{\mathbf{B}^2 \mathbf{r}}$
Section III - Torque	Torque on a current carrying coil in magnetic field	au = BINA cos $lpha$
	Biot-sawart Law	$B = \frac{\mu_{01}}{2\pi r}$
	Ampere's Law	$\sum \vec{B} \cdot \vec{\Delta L} = \mu_0 x$ net curren
Section IV – Ampere's Law	Magnetic field due to Solenoid	$B = \mu_0 n I (n = \frac{N}{L})$
and its applications	Magnetic Field due to Toroid	$B = \mu_0 n I \left(n = \frac{N}{2\pi r} \right)$
	Force between two wires	$F = \frac{\mu_0 I_1 I_2 I_2}{2\pi r}$
	Magnetic Flux	$\Delta \phi = \vec{B} \cdot \vec{A}$ $\Delta \phi = BA\cos\theta$
	Flux density	$B = \frac{\Delta \phi}{\Delta}$
	Encodov's Low	$emf = -N \frac{\Delta \phi}{\Delta t}$
	Faraday's Law	$emf = -\frac{Flux linkage}{\Delta t}$
Section V – Magnetic Flux and Electromagnetic	Mutual Induction and Inductance	$emf_s = -M \frac{\Delta I_p}{\Delta t}$
Induction		$M = \frac{\mathrm{emf}}{\frac{\Delta I_{\mathbf{p}}}{\Delta t}}$ or $M = \frac{\mathrm{N}\Delta \phi}{\Delta I_{\mathbf{p}}}$
	Self Induction and	$emf = -L \frac{\Delta I}{\Delta t}$
	Inductance	$L = \frac{\mathrm{emf}}{\frac{\Delta I}{\Delta t}} \text{ or } L = \frac{N\Delta \phi}{\Delta I}$
	Transformer Equation	$\frac{N_s}{N_p} = \frac{E_s}{E_p} = \frac{I_p}{I_s}$
	Motional EMF	$Emf = VBL sin\theta$
Section VI - Motional Emf	EMF in A.C. Generator	Emf = N ω AB sin θ Emf = N(2 π f)ABsin(2 π ft)

Electrical Measuring Instruments

Instrument	Description	Formula
Galvanometer	Current through the coil	$I = \frac{c}{BAN} \theta$
Galvarionietei	Sensitivity of Galvanometer	Sensitivity $\left(\frac{\theta}{l}\right) = \frac{\text{BAN}}{c}$
Ammeter	Resistance of Shunt	$Rs = \left(\frac{t_g}{t - t_g}\right) R_g$
Voltmeter	Resistance of Multiplier	$Rx = \frac{v}{l_g} - R_g$
Wheatstone Bridge	Balanced Wheatstone bridge	$\frac{R_1}{R_2} = \frac{R_3}{R_4}$
Meter Bridge	Unknown Resistance	$X = R \frac{l_x}{l_r}$
P.O. Box	Unknown resistance	$X = R \frac{Q}{P}$
Potentiometer	Compare two EMF's	$\frac{E_X}{E_S} = \frac{l_X}{l_S}$

Electronics

Topic	Description	Formula
Electromagnetic Waves	Speed of EM Wave	$C = \frac{1}{\mu_0 \varepsilon_0}$
	Current	$I_E = I_B + I_C$
Transistor	Current Gain	$\alpha = \frac{I_C}{I_E}$

Advent of Modern Physics

Topic	Description	Formula
Special Theory of Relativity	Mass increment	$m = \frac{m_0}{\sqrt{1 - \frac{V^2}{c^2}}}$
	Length Contraction	$l = l_o \sqrt{1 - \frac{V^2}{c^2}}$
	Time dilation	$t = \frac{t_0}{\sqrt{1 - \frac{v^2}{c^2}}}$
	Mass Energy Relation	$E = mc^2$
	Rest Mass Energy	$E_o = m_o c^2$
	T-1-1 F-1-1-	$E = E_o + K.E.$
	Total Energy	$mc^2 = m_0c^2 + K.E$
	Wien's Displacement Law	$\lambda_{max} \times T = constant$
-272	Stefan-Boltzmann Law	$E = \sigma T^4$
Black Body Radiation	Raleigh-Jeans Law	Eλ ⁴ = constant
	Planck's Quantum Theory	$E = \frac{hc}{\lambda}$
		Total Energy = Work
		Function + Kinetic Energy
Photoelectric Effect	DL	$E = \phi_o + K.E$
	Photoelectric Equations	$h\nu = h\nu_o + \frac{1}{2} \text{ mV}^2$
		$h\nu = h\nu_o + V_o e$
		$V_o e = hc \left[\frac{1}{\lambda} - \frac{1}{\lambda_o} \right]$

Compton's Effect	Compton's shift in wavelength	$\Delta \lambda = \frac{h}{m_o c} \left(1 - \cos \theta \right)$
Inter-conversion of matter	Pair Production	$h\nu = 2m_oc^2 + K.E_{e^-} + K.E_{e^+}$
and Energy	Annihilation of Matter	$2m_{o}c^{2}+K.E_{e}-+K.E_{e}+=2h\nu$
Wave nature of matter	De-Broglie's Wavelength	$\lambda = \frac{h}{P} = \frac{h}{mV}$
Heisenberg's Uncertainty Principle	Uncertainty of Position and momentum	ΔP. Δx = ħ
	Uncertainty of Energy and time	ΔΕ. Δt = ħ
	Kinetic Energy	$K.E. = \frac{\Delta P^2}{2m}$

Atomic Spectra

Topic	Description	Formula
Bohr's Model	Angular Momentum	$L = \frac{nh}{2\pi}$ or $L = mVr$
	Radius of nth orbital	$r_n = n^2 \times 0.53 ^{\circ} A$
	Energy of nth orbit	$E_n = \frac{-13.6 \text{ eV}}{n^2}$
Hydrogen spectrum	Wave number	$\frac{1}{\lambda} = R_{H} \left[\frac{1}{p^2} - \frac{1}{n^2} \right]$

Nuclear Physics

Topic	Description	Formula
	Radioactive decay Law	$\frac{\Delta N}{\Delta t} = -\lambda N$
Radioactivity	Activity	$A = \frac{\Delta N}{\Delta t} \text{ or } A = \lambda N$
	Relation b/w radioactivity and time	$N = N_o e^{-\lambda t}$
Half - Life	Relation between Half life and decay constant	$T_{1/2} = \frac{0.693}{\lambda}$
	Calculate the number of nuclei decayed	$\frac{N}{N_0} = \left(\frac{1}{2}\right)^{21}$
Mass defect and binding Energy	Mass Defect	Δm = Mass of nucleons – Mass of Nucleus
	Binding Energy	Binding Energy = Δmc ²
	Binding Energy per nucleon (Packing Fraction)	Packing Fraction = $\frac{B.E}{A}$