

Method of Quantum Mechanics.

classical.

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* Introduction of Quantum Mechanics.

→ The classical mechanics which deals with the matter of macroscopic phenomenon. These phenomenon are described by Newton's laws of motion. The second law of motion is widely used in macroscopic. Hence phenomenon. Hence this law is known as fundamental law of Newtonian mechanics. But classical mechanics could not explain the phenomenon like

- 1) The stability of atoms
- 2) Observed spectrum of black body radiation
- 3) Specific heat capacity at low temperature
- 4) The several phenomena like photo electric effect, Compton effect, Raman effect, plasma state, emission of α, β, γ etc.

particle ray

These phenomenon are explained by hypothesis of Planck's, Bohr's postulates etc under the limitation. These theory proposed by them led to development of quantum mechanics such quantum mechanics is called ~~new~~ quantum mechanics.

Deficiency of old Quantum Mechanics.

1) The Bohr's Theory explains well spectrum of monoatomic Hydrogen, singly ionized He-atom, but it fails to explain the spectrum of the simplest multi-electron atom. For example: He

2) The most serious criticism of the old quantum theory is that it is intellectually unsatisfied that is it is not a unified or general theory.

3) It deals with microscopic particles of a well defined path by uncertainty principle is not possible.

4) This theory can be applied only to periodic system for example: harmonic oscillator, circular motion etc. but there are many ^{imp} physical system that are not periodic.

These deficiencies help to develop the new quantum theory whose fundamental principle is wave function. (eqn)

$$\sin \theta - i \cos \theta = e^{-i\theta}$$

* wave equation (wave function) :-

→ The wave equation of the progressive wave at any instant of time 't' is given by :-

$$y = a \sin(\omega t - kx)$$

where,

a = amplitude, k = wave number $= \frac{2\pi}{\lambda}$

It can be written in the exponentially form as

$$y = a e^{-i(\omega t - kx)} \quad \text{--- (1)}$$

$$E = h\nu$$

$$= \frac{h}{2\pi} \times 2\pi\nu$$

$$= h\omega \Rightarrow \omega = \frac{E}{h}$$

$$\text{and } \lambda = \frac{h}{p}$$

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For x-axis

$$\frac{E}{P_m} = \frac{h}{P_m}$$

$$P_m = \frac{h}{\lambda}$$

$$= \frac{h}{2\pi} \times \frac{2\pi}{\lambda}$$

$$= \hbar k$$

$$\text{or } k = \frac{P_m}{\hbar}$$

using value of k and ω in eqn i)

$$y = a e^{-i \left(\frac{E}{\hbar} t - \frac{P_m x}{\hbar} \right)}$$

Then this eqn can be represented by a function.

$$\psi(x, t) \quad \therefore \psi(x, t) = a e^{-i \left[\frac{E}{\hbar} t - \frac{P_m x}{\hbar} \right]}$$

$$= a e^{-\frac{i}{\hbar} [Et - P_m x]}$$

$$= a e^{\frac{i}{\hbar} [P_m x - Et]}$$

$$r = \sqrt{x^2 + y^2 + z^2}$$

This is called wavefunction of particle in ~~3D~~-coordinate. In 3-dimension it can be written as

$$\Psi(r, t) = A e^{\frac{i}{\hbar} [p \cdot r - Et]}$$

This is the fundamental wave eqn in terms of wave function. for new Quantum Mechanics.

* Quantum Mechanical Operator

* Introduction:

→ In classical mechanics, energy and momentum can be expressed in terms of co-ordinate of a particle whereas in quantum mechanics, energy and momentum cannot be expressed in terms of co-ordinate of a particle to avoid a violation of uncertainty principle. Hence to obtain the information of a quantum mechanical ~~part~~ particle, an operator is introduced. The quantum mechanical operator is defined as a function which indicates an operation to be performed over a space of physical state to that of another.

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In order to introduce an operator a equation of ~~form~~ is given by

$$\hat{A} \psi = a \psi$$

is known as eigen value equation
where \hat{A} is quantum mechanical operator

ψ is wave function (eigen function)

$a \rightarrow$ eigen value

[In German, eigen means proper]
two basic

\rightarrow There are \uparrow types of quantum mechanical operators

- ① Momentum operator
- ② Energy operator

① Momentum Operator:-

\rightarrow we have,

wave function of free quantum particle,

$$\psi = A e^{-\frac{i}{\hbar} [Et - P_x x]}$$

for +ve x-axis \rightarrow

Taking derivative of eqn i) with respect to x ,

$$\frac{\partial \psi}{\partial x} = \frac{\partial A e^{-\frac{i}{\hbar} [Et - P_x x]}}{\partial x}$$

$$\frac{\partial \psi}{\partial x} = A \left(\frac{-i}{\hbar} \right) (-P_x) e^{-\frac{i}{\hbar} [Et - P_x x]}$$

$$= \frac{i P_x \psi}{\hbar}$$

$$\text{or } P \psi = \frac{\hbar}{i} \frac{\partial \psi}{\partial x} = \frac{i \hbar}{i^2} \frac{\partial \psi}{\partial x} = -i \hbar \frac{\partial \psi}{\partial x}$$

$$\therefore P \psi = -i \hbar \frac{\partial \psi}{\partial x}$$

$$\therefore \hat{p} \rightarrow -i \hbar \frac{\partial}{\partial x}$$

\rightarrow for x -axis

This is momentum operator

② Energy operator

we have,

$$\psi = a e^{\frac{i}{\hbar} [Et - P_n]} \quad \text{--- (9)}$$

Taking derivative wrt ~~to~~ time,

$$\frac{\partial \psi}{\partial t} = \frac{\partial}{\partial t} \left(a e^{\frac{i}{\hbar} [Et - P_n]} \right)$$

$$\frac{\partial \psi}{\partial t} = a \left(\frac{i}{\hbar} \right) E \left[e^{\frac{i}{\hbar} [Et - P_n]} \right]$$

$$\text{or, } \frac{\partial \psi}{\partial t} = \psi \left(\frac{i}{\hbar} \right) E$$

$$\text{or, } E \psi = \frac{-\hbar}{i} \frac{\partial \psi}{\partial t}$$

$$= \frac{-i\hbar}{1^2} \frac{\partial \psi}{\partial t}$$

$$= i\hbar \frac{\partial \psi}{\partial t}$$

$$\text{Hence } E \longrightarrow i\hbar \frac{\partial}{\partial t}$$

This is called energy operator.

This shows that there is association of energy with operator $i\hbar \frac{\partial}{\partial t}$.

* Schrodinger wave equation's

Introduction:-

→ Schrodinger wave eqn is a fundamental wave eqn in terms of variable wave function ψ which describes the ~~process~~ motion of quantum mechanical particle same as \downarrow second law of motion does in Newton's classical mechanics. There are two types of Schrodinger wave eqns:-

- (i) Time dependent Schrodinger wave eqn
- (ii) Time independent Schrodinger wave eqn.

1) Time dependent :-

In order to derive the time dependent Schrodinger wave eqⁿ. Consider a particle has well defined momentum and energy.

The wave function associated with motion of free particle in terms of momentum and energy is given by

$$\psi = A e^{\frac{i}{\hbar} [Et - P_x x]} \quad \text{--- i)}$$

for +ve x-axis,

Taking derivative of ψ wr to x ,

$$\begin{aligned} \frac{\partial \psi}{\partial x} &= A \left(\frac{-i}{\hbar} \right) (P_x) e^{\frac{i}{\hbar} (Et - P_x x)} \\ &= \frac{i P_x}{\hbar} \psi \end{aligned}$$

Again taking derivative of $\frac{\partial \psi}{\partial x}$ wr to x

$$\frac{\partial^2 \psi}{\partial x^2} = \left[\frac{i P_x}{\hbar} \right]^2 \psi = -\frac{P_x^2}{\hbar^2} \psi$$

$$\text{or } P_x^2 \psi = -\hbar^2 \frac{\partial^2 \psi}{\partial x^2} \quad \text{--- ii)}$$

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Now,

Taking derivative ψ wrt to t , from eqn (i)

$$\frac{\partial \psi}{\partial t} = A \left(\frac{-i}{\hbar} \right) \times E e^{\frac{-i}{\hbar} [Et - p_x x]}$$

$$= \frac{-i E \psi}{\hbar}$$

$$E \psi = \frac{-\hbar}{i} \frac{\partial \psi}{\partial t}$$

$$= \frac{-i \hbar \partial \psi}{i^2 \partial t} = i \hbar \frac{\partial \psi}{\partial t} \quad \text{--- (ii)}$$

The total Energy of free particle

$$E = \text{total KE} + PE$$

$$E = \frac{p^2}{2m} + V \quad \text{--- (iii)}$$

Multiplying by ψ on both sides,

$$E \psi = \frac{p^2 \psi}{2m} + V \psi \quad \text{--- (iv)}$$

Now using eqn (i) and (ii) in eqn (iv)

$$\therefore i \hbar \frac{\partial \psi}{\partial t} = \frac{-\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V \psi$$

$$\text{or, } i\hbar \frac{\partial \psi}{\partial t} = \left(-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V \right) \psi$$

1-D

This is a time dependent schrodinger wave eqn.

Let $H = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V$ is called Hamiltonian operator

Hence,

$$i\hbar \frac{\partial \psi}{\partial t} = H\psi$$

1-D

This is another form of time dependent schrodinger wave eqn.

2) Time independent:-

It is found that in many situations potential V acting on the particle does not depend on time and varies only with its position. For this condition time independent form of Schrodinger wave eqn is applicable.

In order to derive ^{time independent} the wave a particle has well defined momentum and energy can be given by wave function ψ as:-

Total Energy of free particle is given by

$$E = \frac{p^2}{2m} + V$$

using eqⁿ i) in eqⁿ ii)

$$E\psi = \frac{p^2\psi}{2m} + V\psi = -\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi$$

$$\text{or } -\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi - E\psi = 0$$

$$\text{or, } \frac{\partial^2 \psi}{\partial x^2} + \frac{2m}{\hbar^2} [E - V] \psi = 0 \quad \text{--- i)}$$

This is time independent wave

In three dimension. It can be written as

$$\frac{\partial^2 \psi}{\partial x^2} + \frac{\partial^2 \psi}{\partial y^2} + \frac{\partial^2 \psi}{\partial z^2} + \frac{2m}{\hbar^2} [E - V] \psi = 0$$

$$\textcircled{\ast} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right) \psi + \frac{2m}{\hbar^2} [E - V] \psi = 0$$

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Since $\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} = \nabla^2$ Laplacean
 operator as Helmholtz

$$\nabla^2 \psi + \frac{2m}{\hbar^2} [E - V] \psi = 0 \quad \text{Hame}$$

$$\left[H = \frac{-\hbar^2 \nabla^2}{2m} + V \right]$$

$$\nabla^2 \psi + \frac{2m}{\hbar^2} E \psi - \frac{2m}{\hbar^2} V \psi = 0$$

$$\frac{2m}{\hbar^2} E \psi = -\nabla^2 \psi + \frac{2m}{\hbar^2} V \psi$$

$$E \psi = \frac{-\hbar^2}{2m} \nabla^2 \psi + V \psi$$

$$= \left(\frac{-\hbar^2}{2m} \nabla^2 + V \right) \psi$$

$$\boxed{E \psi = H \psi}$$

$\therefore H = \frac{-\hbar^2}{2m} \nabla^2 + V$ is a Hamiltonian operator.

This is time independent SWE in 3D. in terms of Hamiltonian operator

* Schrodinger wave eqn for free particle.

we have,

$$\nabla^2 \psi + \frac{2m}{\hbar^2} [E - V] \psi = 0 \quad \text{--- (1)}$$

For free particle $V = 0$

Hence eqn (1) reduces to

$$\nabla^2 \psi + \frac{2m}{\hbar^2} [E] \psi = 0$$

This is SWE for free particle

* Physical significance of wave function (ψ)