
Atomistic Modelling of Aggregate Chlorophyll Systems

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ABSTRACT

Here goes the abstract

DEDICATION AND ACKNOWLEDGEMENTS

Here goes the dedication.

AUTHOR'S DECLARATION

I declare that the work in this dissertation was carried out in accordance with the requirements of the University's Regulations and Code of Practice for Research Degree Programmes and that it has not been submitted for any other academic award. Except where indicated by specific reference in the text, the work is the candidate's own work. Work done in collaboration with, or with the assistance of, others, is indicated as such. Any views expressed in the dissertation are those of the author.

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INTRODUCTION

Naturally occurring light harvesting systems present an interesting scientific challenge. With near perfect efficiency, the energy from a photon will be taken and transferred to a reaction centre, leading to charge transfer processes that culminate in powering biological systems. Making models that can predict and explain these effects are key to making similarly efficient photovoltaic systems.

1.1 Quantum Exploits in Light Harvesting Systems

Begins a section.

1.1.1 Electronic Energy Transfer

Begins a subsection.

1.1.2 Coherence

1.2 Light-Matter Response

1.3 Electronic Structure for Large Systems

1.4 The Aim

MEAN-FIELD EXCITED STATES

Previous Published Work

The work presented in this chapter is based on joint work with Dr Susannah Bourne-Worster, and was published in March 2021[26]. The account given in this chapter includes research that was also reported in this publication, namely the parts of section 2.2 from 2.2.1 up to but excluding 2.2.5.

This chapter collates all of the work done on investigating and benchmarking Δ -SCF methods, at both an ab initio DFT level of theory as well as with semi-empirical, tight-binding approximations. The reduced cost and moderate accuracy of these methods make them an ideal substitute for full TD-DFT or high-level methods when investigating large systems like chlorophyll. Transition properties were calculated for a range of molecules, as well as for a small set of chlorophyll geometries, using variety of different basis sets, density functionals, response methods and electronic structure methods. Most of the work was compared to either high-standard EOM-CCSD or SCS-CC2 reference, from which conclusions were made on the accuracy of each method. The non-orthogonality issue of the ground and excited states were also investigated for the mean-field Δ -SCF method, as well as assignment of transitions based on symmetry. It was found that while the DFT methods give reasonable results, the semi-empirical Δ -SCF methods are not as accurate. This result is unexpected in the context of the benchmarking, however has reasonable explanations, and will guide the work of the later chapter on developing a new semi-empirical method.

2.1 Theory

2.1.1 Δ -SCF and eigenvalue difference

Δ -SCF predicts the transition energy ΔE of a system as the difference of the single point energy E_n of two states:

$$(2.1) \quad \Delta E = E_2 - E_1$$

Finding the excited state correctly can be an issue, and for ease is usually assumed to be similar to the ground state. In its simplest form, the Δ -SCF method calculates the ground state with normal DFT or other mean-field methods, and then calculates the excited state by rerunning the same method with the excited state occupation numbers. A full description of both the ground as excited state is given as the orbital coefficients output from the two calculations.

Initially, the excited state was calculated by relaxing the orbitals which contain the excited electron and hole in the ground state space, so that the excited state and ground state are orthogonal[15]. However, it was argued that this procedure would exacerbate errors from finding the ground state, and that the excited state was not a proper SCF solution[8]. Alternatively, an SCF like method was proposed, where instead of populating orbitals according to the aufbau principle, orbitals which most resemble the previous iteration's orbitals should be occupied. Each iteration in an SCF procedure produces new molecular orbital coefficients by solving the Roothaan-Hall equations[23], generally given as an eigenvalue problem:

$$(2.2) \quad \mathbf{FC}^{\text{new}} = \mathbf{SC}^{\text{new}}\epsilon$$

where \mathbf{C}^{new} are the next orbital coefficient solutions, \mathbf{S} is the overlap, and ϵ are the orbital energies. The Fock matrix \mathbf{F} is calculated from the previous set of orbital coefficients:

$$(2.3) \quad \mathbf{F} = f(\mathbf{C}^{\text{old}})$$

The amount of similarity of orbitals can be estimated from their overlap:

$$(2.4) \quad \mathbf{O} = (\mathbf{C}^{\text{old}})^{\dagger} \mathbf{SC}^{\text{new}}$$

and for a single orbital can be evaluated as a projection:

$$(2.5) \quad p_j = \sum_i^n O_{ij} = \sum_v^N \left[\sum_\mu^N \left(\sum_i^n C_{i\mu}^{\text{old}} \right) S_{\mu v} \right] C_{vj}^{\text{new}}$$

where μ, ν are orbital indices. The population can then be given by the set of orbitals with the highest projection p_j . This method can be used for any excited state, with the caveat that the orbital solution will most likely be in the same region as the ground state solution. For a small number of low lying states, this is generally true, and so Δ -SCF can be used to calculate a small spectrum of excited states[8]. The method of using this orbital overlap is called the maximum overlap method (MOM)[8].

Δ -SCF has been shown to be cheap alternative to TD-DFT and other higher level methods[4, 7, 18], without considerable losses of accuracy in certain cases, especially for HOMO-LUMO transitions[17]. Additionally, as the excited state is given as solutions to SCF equations, the gradient of this solution can be given by normal mean-field theory. These gradients would be much cheaper than TD-DFT or coupled cluster methods, and so would be better for any dynamic simulation[7].

2.1.2 Eigenvalue Difference

Another approximation to full response theory would be to eigenvalue difference method. Here there is assumed to be no response of the orbital energies and shapes when interacting with light. This would be recovered from the complete Cassida equation if the coupling elements in the **A** and **B** matrices are set to zero. This gives the difference between the excited state energy and the ground state energy, the transition energy, as just the difference of the orbital energies between the orbital an electron has been excited (ϵ_e) to and the orbital has been excited from (ϵ_g):

$$(2.6) \quad \Delta E = \epsilon_e - \epsilon_g$$

Additionally, transition properties can be calculated by construction transition density matrices from only the ground state solution, needing only a single SCF optimization. Generally, eigenvalue difference methods are not seen as accurate response methods, but can offer a quick and easy initial value[9].

2.1.3 Transition Density and Dipole Moments

Δ -SCF transition properties, such as the transition dipole moment, can be calculated from the SCF solutions for the ground and excited states. The reduced one-particle transition density matrix **D**²¹ can be written as:

$$(2.7) \quad \mathbf{D}^{21} = |\Psi_1\rangle\langle\Psi_2|$$

where $|\Psi_n\rangle$ is the Slater determinant of state n , constructed from the set of spin orbitals $\{\phi_j^{(n)}\}$. Expressed in terms of the molecular orbitals coefficients $\mathbf{C}^{(n)}$, the transition density matrix is given by

$$(2.8) \quad \mathbf{D}^{21} = \mathbf{C}^{(2)} \text{adj}(\mathbf{S}^{21}) \mathbf{C}^{(1)\dagger}$$

where \mathbf{S}^{mn} is the overlap of the two states. The adjunct of the overlap correspond to Löwdin's normal rules for non-orthogonal determinants [19]. For transition dipole elements, this is:

$$(2.9) \quad \langle \Psi_2 | \hat{\mu} | \Psi_1 \rangle = \sum_{jk} \mu_{jk}^{21} \text{adj}(\mathbf{S}^{21})_{jk}$$

where $\hat{\mu}$ is the one-electron transition dipole operator, and μ_{jk} is the element of this operator corresponding to orbital indices j, k . The determinant of this overlap, the orbital inner products, can be defined as the inner product of the two states:

$$(2.10) \quad |\mathbf{S}^{21}| = \langle \Psi_2 | \Psi_1 \rangle$$

The general definition of the transition dipole

$$(2.11) \quad \mu^{1 \rightarrow 2} = \langle \Psi_2 | \hat{\mu} | \Psi_1 \rangle$$

can be expressed with this transition density matrix as:

$$(2.12) \quad \begin{aligned} \langle \Psi_2 | \hat{\mu} | \Psi_1 \rangle &= \text{tr}(\hat{\mu} | \Psi_1 \rangle \langle \Psi_2 |) \\ &= \text{tr}(\hat{\mu} \mathbf{D}^{21}) \end{aligned}$$

2.1.4 Semi-empirical extensions

A main investigation of this chapter is whether the range of DFT methods that could be used for Δ -SCF and eigenvalue difference methods could be extended by using tight-binding methods to predict transition properties. This mainly focused on the recently published GFN-xTB method, parameterized by the Grimme group [12]. This method has been parameterized for geometries, frequencies and non-covalent interactions, and uses an extended version of Hückel theory. The name GFN-xTB is acronym for "Geometries, Frequencies, Non-Covalent - eXtended Tight Binding". This method was chosen for two reasons. The first being that the GFN-xTB method was already implemented in the QCORE package. This significantly reduced the amount of effort required for this project. Additionally, there would be other users and developers who would help with implementation of this new method. Second is that a similar method has already been published that calculates transition properties, the precursor to the GFN-xTB methods. This is the sTDA-xTB method [11].

2.1.4.1 sTDA-xTB

sTDA-xTB ("simplified Tann-Danoff Approximation - eXtended Tight Binding") is another method in the family of xTB methods developed by the Grimme group, and is parameterised for transition properties. The accuracy in calculating transition energies with this method is very good, with the error to high-level results being around 0.3 - 0.5 eV.

Similar to other xTB methods, the sTDA-xTB method is a tight-binding method that uses empirically fit parameters and a minimal basis set. It was trained on a test set of highly accurate coupled cluster and density functional theory excitation energies, as well as atomic partial charges for inter-electronic interactions.

Unlike other xTB methods, coefficients in the basis set for sTDA-xTB are dependent on the D3 coordination number. This makes basis functions far more flexible, which would usually be achieved in with fixed basis functions by using diffuse or other additional orbitals in the basis set. Additionally, it uses two sets of parameterized basis sets - a smaller valence basis set (VBS) and an extended basis set (XBS). Whilst this reduces the cost of having larger basis sets, it makes calculating the gradient of transition properties much more difficult.

The two basis sets are used to construct formally similar Fock matrix elements, however in practice they use different global parameters. The core Hamiltonian is similar to other DFTB methods that use a self-consistent charge (SCC) method, as opposed to an SCF method, to obtain molecular orbital coefficients. It is given by:

$$(2.13) \quad \langle \psi_\mu | H^{\text{EHT, sTDA-xTB}} | \psi_\mu \rangle = \frac{1}{2} \left(k_\mu^l k_\nu^{l'} \right) \frac{1}{2} \left(h_\mu^l h_\nu^{l'} \right) S_{\mu\nu} - k_T \langle \psi_\mu | \hat{T} | \psi_\nu \rangle$$

where, μ, ν, l, l' are orbital and shell indices k_μ^l are shell-wise Hückel parameters, h are effective atomic-orbital energy levels, S is the overlap, k_T is a global constant and \hat{T} is the kinetic energy operator. The charges used in the inter-electronic repulsion function are given by CM5 [20] charges for the XBS Fock matrix. These are calculated using Mulliken charges obtained from diagonalising the Fock matrix with the VBS. The charges for the initial VBS Fock matrix are based on Gasteiger charges [6], modified by the parameterised electronegativities of atoms in the system.

The whole process for determining molecular orbitals can be summarized as:

1. Calculate modified Gasteiger charges for the first initial guess
 2. Diagonalise Fock matrix in the VBS to get the first set of Mulliken charges
 3. Compute CM5 charges
 4. Diagonalise Fock matrix in the VBS again for final set of Mulliken charges.
 5. Recalculate CM5 charges with this final set, and diagonalize the Fock matrix in the XBS.
- The molecular orbital coefficients from this are then fed to the response theory.

The response theory for this method is based on the previous work in the Grimme group on the simplified Tamm-Danoff Approximation [10]. There are several approximations made between full linear response theory and the sTDA method. First is the Tamm-Dannoff approximation, where the **B** matrix is ignored. The second approximation is to use monopole approximations with Mataga-Nishimoto-Ohno-Klopman operators instead of explicit 2 electron integrals to calculate matrix elements, as well as neglecting the density functional term.

Transition charges are used to calculate these MNOK integrals. The charge q_{nm}^A , the charge centred on atom A given by transition density for the transition of $n \rightarrow m$, are computed using a Löwdin population analysis:

$$(2.14) \quad q_{nm}^A = \sum_{\mu \in A} C'_{\mu n} C'_{\mu m}$$

where the transformed coefficients $C'_{\mu n}$ are given by the orthogonalising the original MO coefficients **C**:

$$(2.15) \quad \mathbf{C}' = \mathbf{S}^{\frac{1}{2}} \mathbf{C}$$

and μ is an index that runs over the atomic orbitals (AO).

The operator is the MNOK[21][22][16] damped coloumb function, with different exponents y_K and y_J for exchange and coulomb integral respectively. The α_x parameter is included to recover the amount of Fock exchange mixing in the original matrix element equation, and is a free parameter. These will be discussed in more detail in the next chapter, as they are a crucial part of designing a new response method for chlorophyll systems.

Third is the truncation of single particle excited space that is used to construct the **A** matrix. This reduces the number of elements that need to be calculated, and so reduces the time taken for diagonalisation, whilst also capturing a broad enough spectrum of excitation energies. The sTDA-xTB has many of the same goals as this project, except in one respect, which is the gradient theory. As the sTDA-xTB method still requires constructing and diagonalizing the **A** matrix, albeit with a tight-binding method for molecular orbital coefficients, the gradient of the transition properties would still be difficult to calculate.

2.2 Benchmarking

The first step in testing whether a GFN-xTB based Δ -SCF method (which we name Δ -xTB) could predict TD-DFT transition properties with decent accuracy was to test a set of small molecules, as well as bacterial chlorophylls from the LHII protein against high-level results.

2.2.1 Reference Data and test set

The test set comprised of 109 small molecules. Each system was closed-shell, contained 12 atoms or less, and contained on H, C, N, O and F atoms. The size and relative simplicity of this test set was chosen to minimise any other factors that could cause complication in analysing the results. For example choosing a consistently correct basis set and functional for the Δ -SCF results would have been challenging if it was necessary to include transition metal complexes, as well considering CPU walltime constraints. Additionally, the size of these molecules made symmetries much easier to inspect by hand.

Reference data was calculated as the three lowest energy singlet excited states, using EOM-CCSD with an aug-cc-pVTZ basis set. These results were generated using the Gaussian 16 program [5].

2.2.2 Small Systems

Transition properties for this test set were calculated using TD-DFT and Δ -SCF, both using a CAM-B3LYP functional and aug-cc-pVTZ basis set. The transitions were assigned to the EOM-CCSD results by comparing transition dipoles, energies and the character of the MOs involved in the transitions. Where the symmetries could be assigned, these were also used, however this was not the case for all systems as many were unsuccessful in labelling symmetry or defaulted to a non-Abelain group. Symmetry labelling was also only available for TD-DFT calculations, as these were performed with Gaussian 16. Δ -SCF calculations were done with the QCORE program.

The Δ -SCF singlet transition is not a correct representation of a true singlet excitation, as this is a superposition of both spin-same $\alpha \rightarrow a, \alpha$ and spin-flipping $\alpha \rightarrow a, \beta$ excitations. The spin-purification formula:

$$(2.16) \quad \Delta E_S = 2\Delta E^{i,\alpha \rightarrow a,\alpha} - \Delta E^{i,\alpha \rightarrow a,\beta}$$

was used to correct for the true singlet excitation energy ΔE_S .

The results of comparing transition energies and transition dipole magnitudes are shown in figures 2.1 and 2.2.

Overall, the excitation energies calculated with Δ -SCF are as accurate at predicting EOM-CCSD energies as TD-DFT. The mean error is 0.35 eV, with a standard deviation of 0.25 eV. This is a marginal improvement on the TD-DFT results, which has a mean and standard deviation of 0.41 eV and 0.27 eV respectively. Transition dipoles were similarly accurate to the reference data, although Δ -SCF performs slightly worse in this respect. The mean and standard deviation in the absolute value of transition dipole moment, $|\mu|$, was 0.07 a.u. and 0.08 a.u. respectively. For TD-DFT, the mean and standard deviation were 0.03 a.u. and 0.06 a.u. (the atomic unit here being equal to ea_0). The outlier circled in figure 2.1 is an ethene dimer system, and shows the inability of Δ -SCF to capture a mixed excited state. The two LUMO orbitals in this dimer system

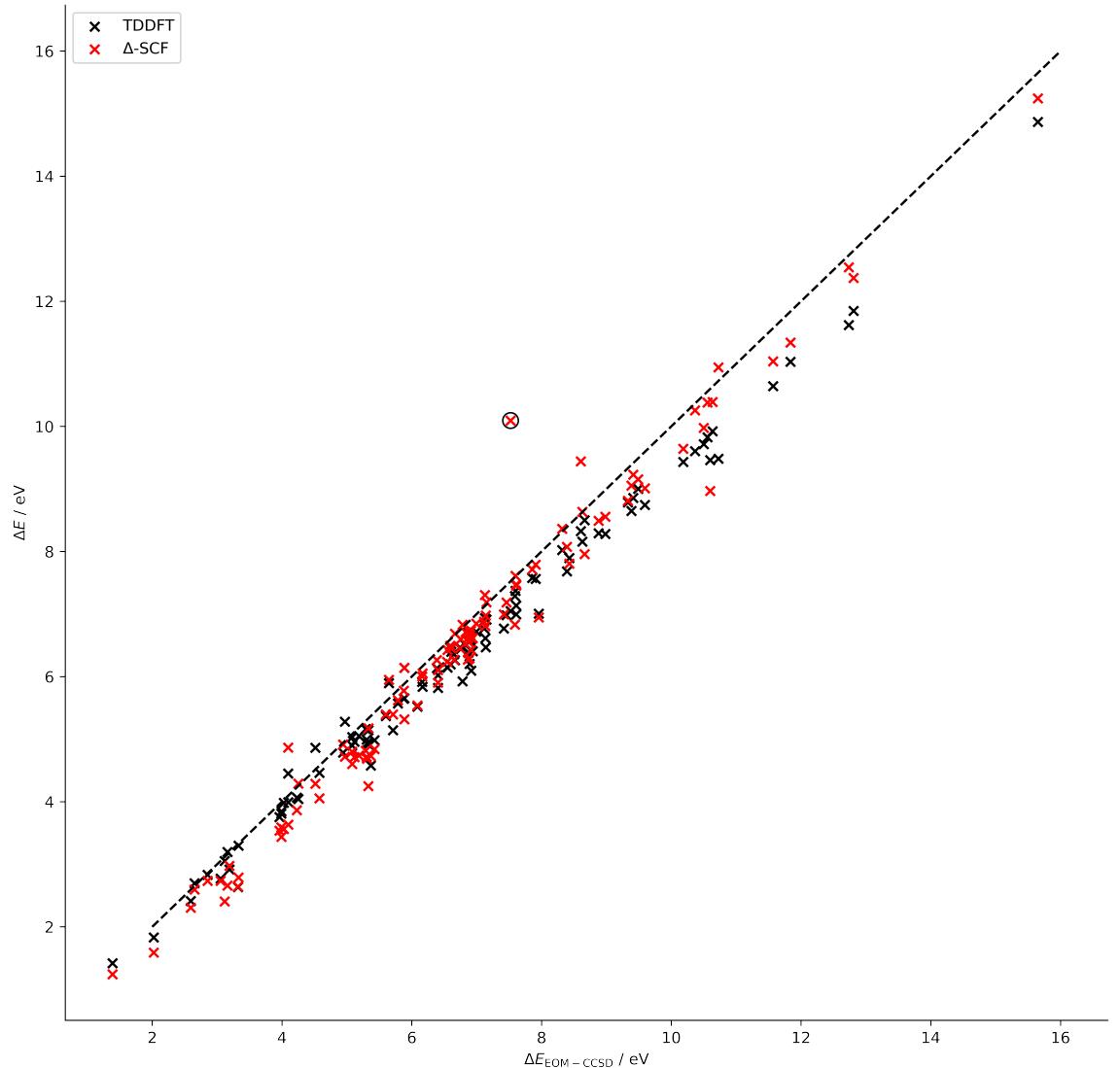


Figure 2.1: Transition energies ΔE from TD-DFT (black) and Δ -SCF (red) plotted against EOM-CCSD energies, with the line $y = x$ (dashed) for reference. The ethene dimer outlier has been circled.

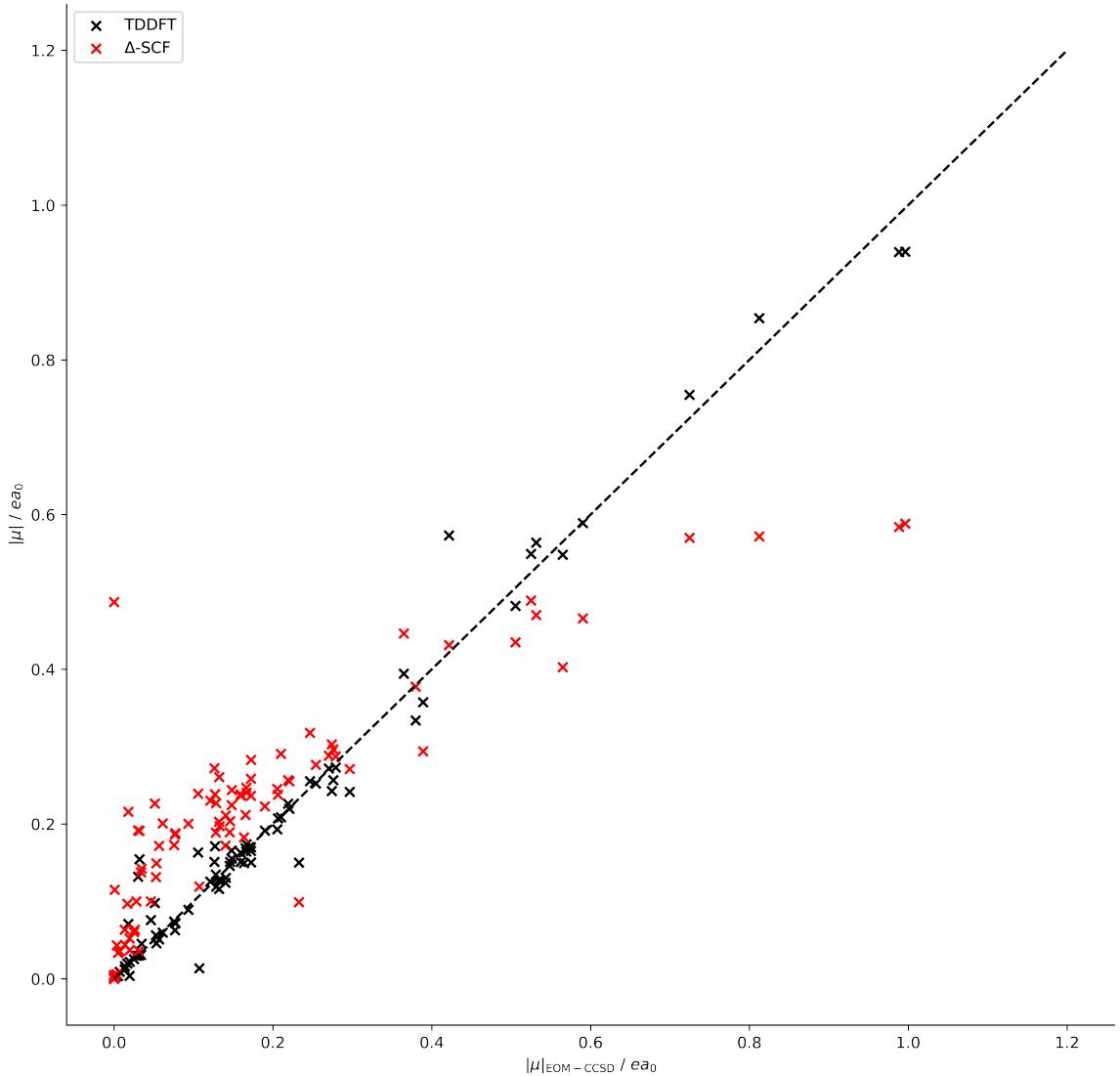


Figure 2.2: Transition dipole magnitudes from TD-DFT (black) and Δ -SCF (red) plotted against EOM-CCSD transition dipole magnitudes, with the line $y = x$ (dashed) for reference.

are in-phase and out-of-phase combinations of the π -antibonding orbitals, which are very close in energy. The HOMO orbitals are the same on both ethene molecules, being the π -bonding orbitals, which are degenerate in energy. The first excited state is predicted by TD-DFT and EOM-CCSD to be a mix of these two close HOMO-LUMO transitions, however Δ -SCF cannot include this mixed behaviour. Δ -SCF predicts two transitions of 7 eV and 10 eV, whilst TD-DFT and EOM-CCSD predict 7.5 eV.

In summary, Δ -SCF can be seen to accurately predict transition properties to a EOM-CCSD level of accuracy with as much success as TD-DFT, except in cases of mixed transitions. It might then be expected a tight-binding method, with good electronic structure treatment, could also be accurate whilst drastically reducing the cost of calculation.

2.2.3 Non-orthogonality

There is, however, another caveat with Δ -SCF. The ground and excited states, solutions to two separate SCF cycles, will not be truly orthogonal. The Slater determinants $|\Psi_n\rangle$, are constructed from the set of orbitals $\{|\phi_j^{(n)}\rangle\}$ that will be orthogonal within the same state, however the overall states will not be orthogonal to each other, such that the inner product:

$$(2.17) \quad S_{jk}^{21} = \langle \phi_j^{(2)} | \phi_k^{(1)} \rangle$$

will be non-zero. Similarly, there will be a non-zero transition charge:

$$(2.18) \quad q^{21} = \langle \Psi^2 | \Psi^2 \rangle$$

which breaks the origin-independence property of the transition dipole moment. In this way, any transition dipoles that do not have their centre at the origin will have a systematic error based on this overlap and the distance from the origin. For vertical transitions, this transition charge should be zero, and so all transition dipole moments calculated with non-orthogonal Δ -SCF would always have this error.

In order to fix this issue, a transformation to symmetrically orthogonalise the two states was applied, which also would preserve as much character of the original states as possible. The transformation is given by:

$$(2.19) \quad |\Psi_{\tilde{\nu}}\rangle = \sum_{\nu} |\Psi_{\nu}\rangle \left[\mathbf{S}^{-\frac{1}{2}} \right]_{\nu\tilde{\nu}}$$

where \mathbf{S} here is a block matrix:

$$(2.20) \quad \mathbf{S} = \begin{pmatrix} 1 & S \\ S & 1 \end{pmatrix}$$

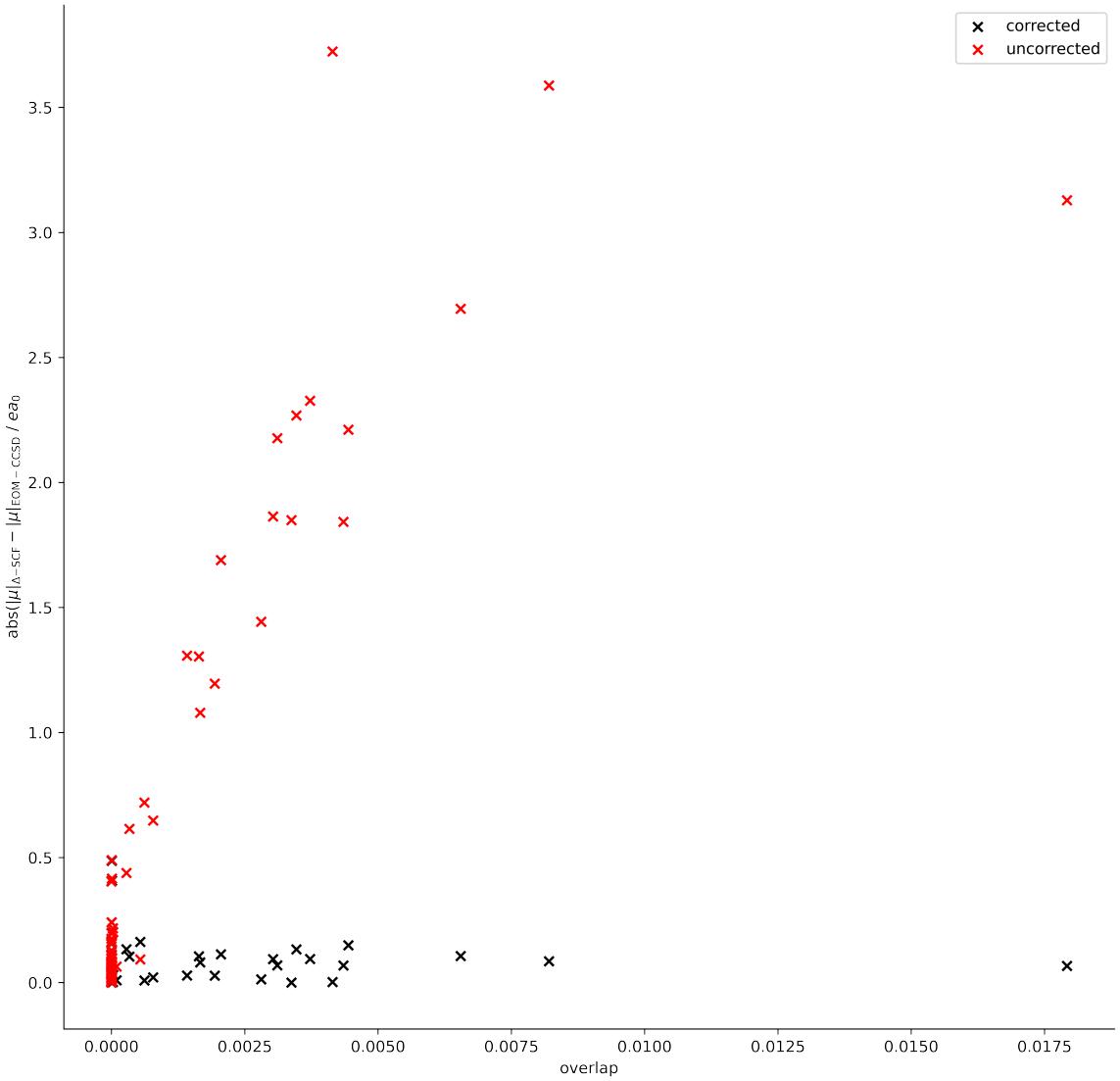


Figure 2.3: The absolute value of the error in transition dipole magnitude between Δ -SCF and EOM-CCSD, plotted against the Δ -SCF overlap of the ground and excited state. All systems were translated by 100 Å in all cartesian axes. Transition dipole magnitudes calculated without any correction are shown in red, whilst those with the symmetric orthogonalisation correction are shown in black.

with S being the overlap value of the two states $\langle \Psi_2 | \Psi_1 \rangle$.

It was found that using this method for correcting the non-zero overlap of states, the origin-independence of the transition dipole moment was recovered (see figure 2.2.3). The transition dipole for each molecule in the test set systems was calculated for molecules translated by 100 Å in each of the x , y and z axes. This would induce an error for the non-orthogonalised states, which has been corrected for in the calculations with the symmetric orthonormalisation. It should be noted that whilst this effect is dependent on how large the overlap may be, and it could be argued that with a small overlap this effect may not be large, having any large translation of the molecule (on the order of hundreds of angstroms) can lead to nonphysical transition dipole magnitudes. In a large protein system, where these distances can be on this scale, this would obviously present a much larger problem than for a vacuum phase small molecule.

2.2.4 LHII Chlorophyll

The accuracy of Δ -SCF was also tested on a set of bacterial chlorophyll A (BChla) molecules. These are a much larger and more complex system, and so present a tougher challenge for the Δ -SCF method. Additionally, it is a more relevant test for the goal of simulating a whole LHII complex.

As each BChla is 140 atoms, EOM-CCSD could not be used as the reference method due to resource constraints, so TD-DFT at a PBE0/Def2-SVP level of theory was used instead [2, 24]. This has been used previously for BChla structures, and has been shown to be a good balance between accuracy and cost [25]. The Δ -SCF was run with the same density functional and basis set.

Δ -SCF can be seen to give transition energies to within the accuracy of TD-DFT, such that the variations in transition energy can be said to be due to physical intra-molecular reasons rather than random error from the Δ -SCF method. Δ -SCF could therefore be used to accurately predict geometry dependent properties.

However, the error in transition dipole magnitudes is larger than that of the small test set. This error is about 0.42 a.u. larger, but without EOM-CCSD or another high-level method, it's unclear whether this error might be from TD-DFT or Δ -SCF. Additionally, there is a clear correlation between the transition dipole magnitudes from TD-DFT and Δ -SCF, and so whilst quantitative statements cannot be made, qualitative assessments can be confidently made from Δ -SCF data.

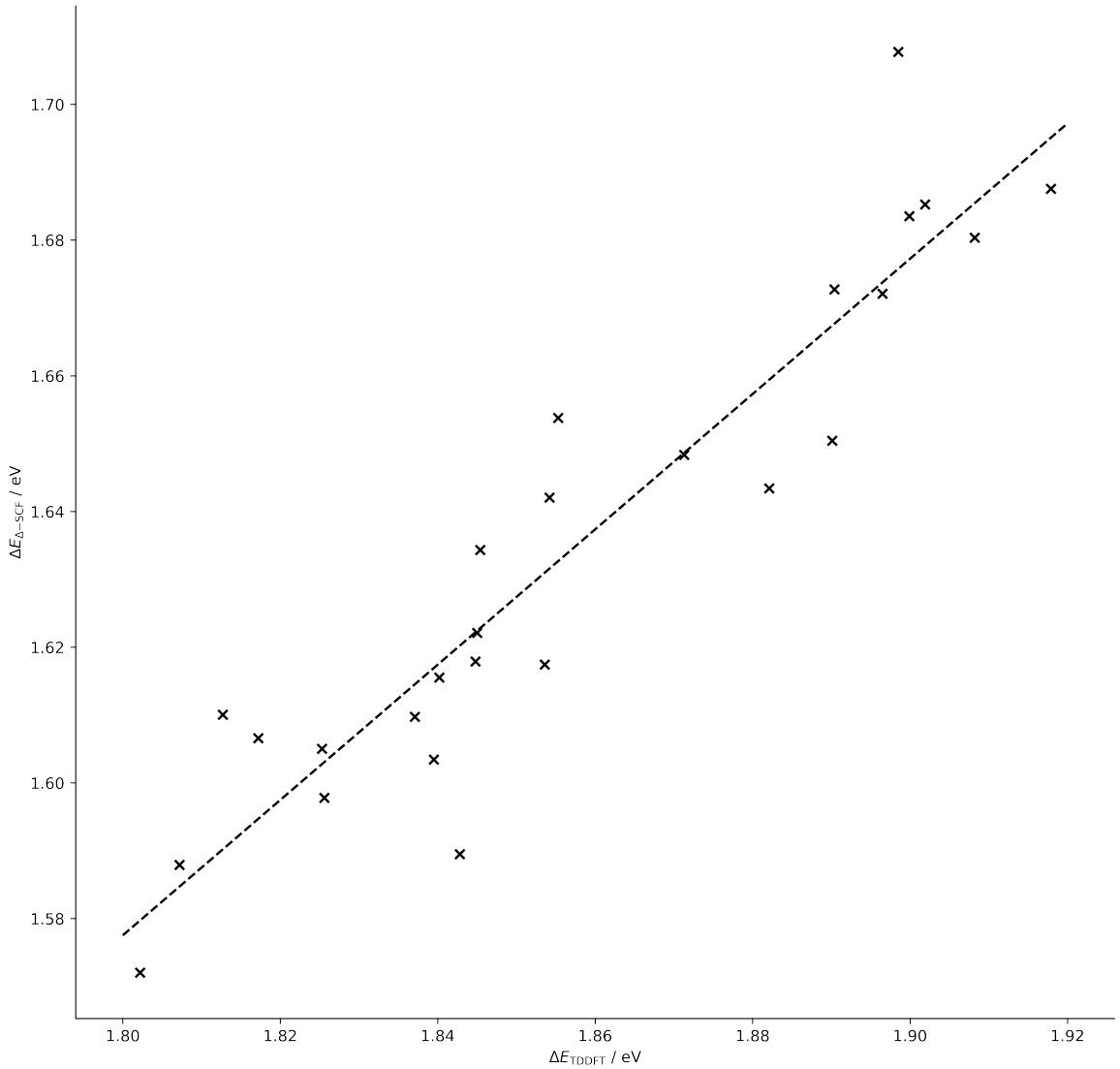


Figure 2.4: The transition energies from Δ -SCF for the 26 chlorophyll geometries, plotted against energies from TD-DFT. The line of best fit ($R^2 = 0.87$) is shown as the dashed line.

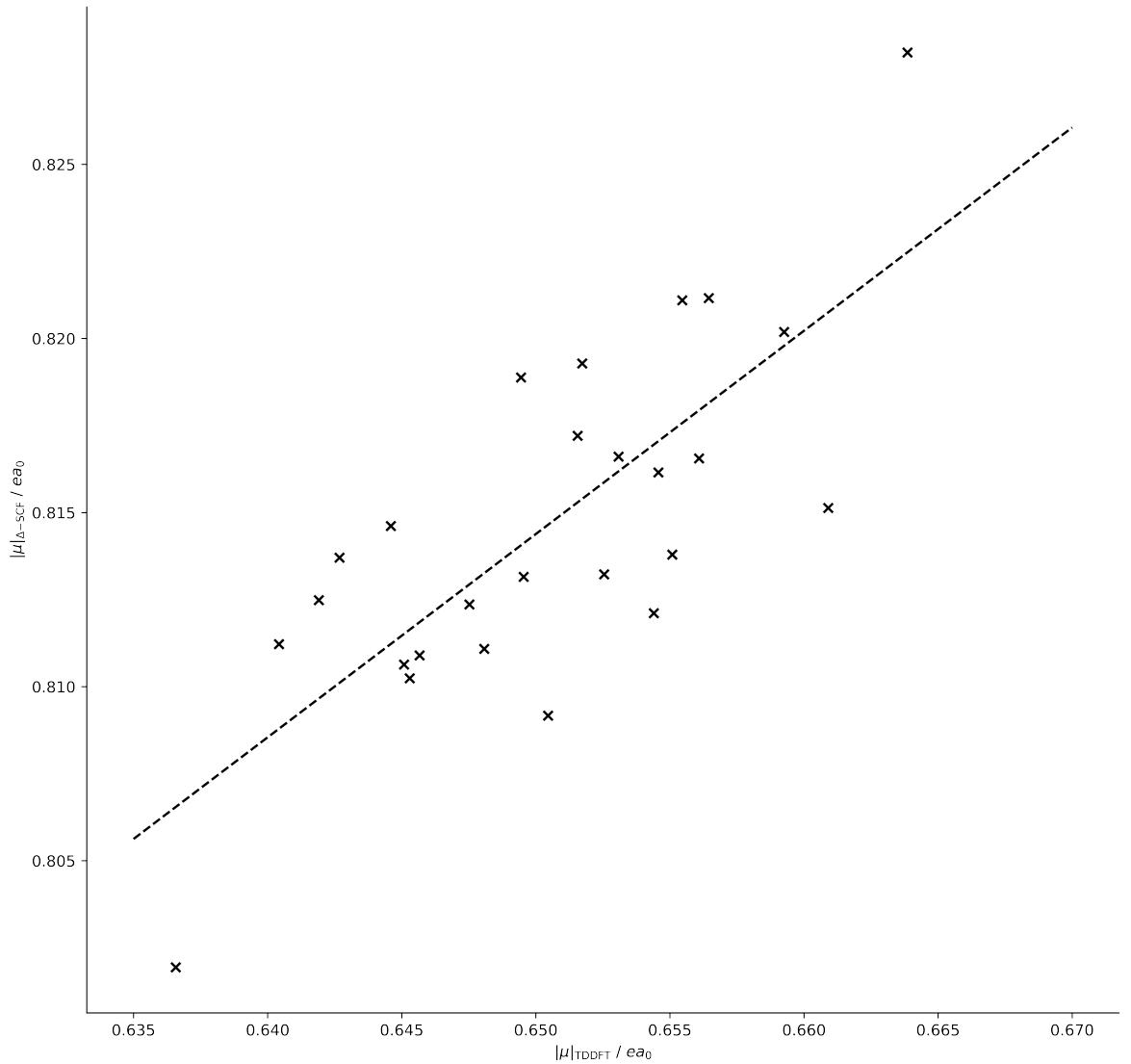


Figure 2.5: The transition dipole magnitudes from Δ -SCF for the 26 chlorophyll geometries, plotted against dipole magnitudes from TD-DFT. The line of best fit ($R^2 = 0.57$) is shown as the dashed line.

2.2.5 xTB methods

The same benchmarking set was used with the Δ -xTB method. The number of methods in the comparison was also expanded to cover a range of approximations in calculating transition properties, including the aforementioned sTDA-xTB method. To include this in the comparison, it was necessary to use the spin-component-scaled second order coupled-cluster (SCS-CC2) [13, 14] reference data produced by the Grimme group. Concurrent to this work, the GFN0-xTB method was also implemented in QCORE. This method is similar to the GFN1-xTB method, but excludes any charge dependent terms in its Fock matrix so is not self-consistent. This method therefore only uses a single diagonalisation, with orbital solutions being the same for both ground and excited state. A transition energy from Δ -SCF with this method would functionally be the same as an eigenvalue difference, and this was included in the benchmarking to test whether it would be a possibility for predicting transition properties. Also included in this benchmarking is using the eigenvalue differences from sTDA-xTB, given by the `xtb4stda` program, to observe how much improvement to transition properties is caused by the sTDA procedure. `xtb4stda` is the first of two programs that runs a version of xTB, providing molecular orbital coefficients and energies for the sTDA program to then use to calculate the transition properties.

All methods included were:

- High level TD-DFT, with a range separated functional CAM-B3LYP and a aug-cc-pVTZ basis set.
- Lower level TD-DFT, with a PBE0 functional and smaller Def2-SVP basis set.
- Δ -SCF with CAM-B3LYP/aug-cc-pVTZ.
- Δ -SCF with PBE0/Def2-SVP.
- linear response with GFN1-xTB and GFN0-xTB.
- Δ -SCF with GFN1-xTB and GFN0-xTB, named Δ -xTB .
- Full sTDA-xTB, for additional reference.
- sTDA-xTB eigenvalue difference.

A source of error that hasn't been discussed in much detail so far is the assignment of transitions between different methods. A known problem of Δ -SCF methods is that the excited state SCF cycle may not converge to the correct state, or it might collapse back to the ground state. This could be seen in the symmetry of the excitation, as if Δ -SCF excited state has converged to different transition than TD-DFT and CC2 it would have different symmetries of transition. The benchmarking discussed above used symmetry labels to assign TD-DFT transitions to EOM-CCSD, but as noted earlier this was not always possible, and assigning symmetry labels to

Δ -SCF was not possible. Instead, transition dipole orientations and plots of MOs were used as a proxy. This proved harder in the Δ -xTB case for two reasons. First is the valence basis set for the Δ -xTB calculations are very different to those used in DFT. Second is that this information was not available for the CC2 data, however the symmetry labels were. Additionally, inspecting the symmetry is very time-consuming and can not be automated. Every new method that was added to the benchmarking had to have every molecule individually inspected to make sure the symmetry of each transition was correct. Assigning symmetry to Δ -SCF results was investigated, but was not a straightforward implementation.

2.2.5.1 Post-SCF Assignment of Symmetry

Considering symmetry is a common thread in many parts of electronic structure theory. It appears in normal mode analysis, wave-function analysis and assignment of electronic transitions. For this project, assigning symmetry to the transitions for Δ -SCF would require assigning symmetry to the MOs and overall wave-function of a molecule. Broadly speaking, most electronic structure codes have two choices in assigning symmetry to orbitals - either all of the SCF code will treat symmetry from the outset, or nothing is assigned in the SCF code and assignment will happen post-SCF. Both these approaches have benefits and drawbacks. The first method allows the symmetry to be given at any point in the SCF procedure, and allows the Hamiltonian to be organized into a block diagonal matrix. This can be useful when solving for a large basis set or large system as the matrix diagonalisation can be partitioned and parallelised over several cores or nodes on a cluster computer. However, this works best if the system is highly symmetric, which is often not the case when treating unoptimized systems, such as those from a molecular dynamics simulation, and is definitely not the case when looking at biological systems. The second approach, assigning symmetry after the SCF cycles, doesn't fix these drawbacks, but it does allow for codes which originally didn't have symmetry assignment to be extended without rewriting SCF code. The obvious drawback of doing assignment post-SCF is that symmetry can't be utilized during the SCF procedure. The second approach was opted for, as this was the easiest to implement in QCORE. The open source library libmsym [1] was used for point group assignment routines and finding the symmetry adapted linear combination of atomic orbitals. The steps for assigning orbital symmetry is as follows:

1. Determine the point group of the molecule, from the atomic positions
2. Setup the atomic orbitals in the libmsym representation.
3. Get the symmetry adapted linear combinations (SALCs) of atomic orbitals for each subspace.
These subspaces are the groups of symmetries that can be found in the point group of the molecule.
4. The SALCs can then be used to construct the transformation matrix T .

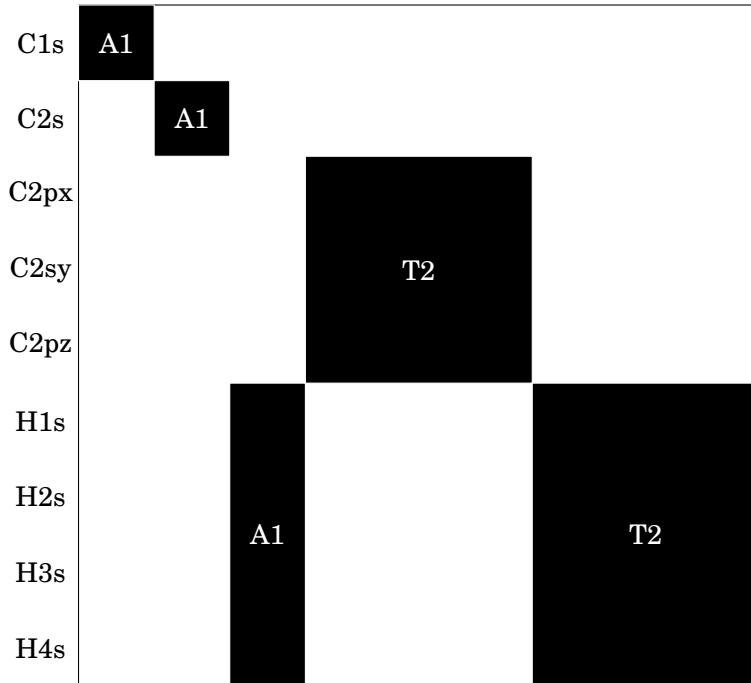


Figure 2.6: A breakdown of the symmetry orbitals in STO-3G methane into the subspaces present in the T_d point group.

5. Assign the one electron molecular orbital (MO) for these subspace characters with the symmetry adapted linear combinations.
6. Multiply the one electron MO symmetries together to find the symmetry of the overall wavefunction.

This procedure was implemented and tested on methane with a minimal STO-3G basis set. It was found that this could accurately assign the MOs and overall wavefunction of methane the ground state.

To assign a label to each molecular orbital, the character of each orbital in all subspaces was looked at. This required transforming the molecular orbital coefficients \mathbf{C} into the each subspace A by using the transformation matrix \mathbf{T} :

$$(2.21) \quad \tilde{\mathbf{C}}_A = \mathbf{T}_A^T \mathbf{S} \mathbf{C}$$

and then summing the coefficients to obtain the character P_A :

$$(2.22) \quad P_A = \sum_v |\tilde{\mathbf{C}}_{A,\mu v}|$$

where μ, ν are indices for the atomic and molecular orbitals respectively. The molecular orbital with character equal to 1 in a subspace would then have that symmetry label, and would be a well defined assignment. However in practice this was not so clear cut and so the highest subspace character was taken as the assignment.

The MOs for an optimised methane geometry with an STO-3G basis set were correctly assigned - two occupied orbitals and one unoccupied orbital of A1 symmetry, and three occupied and unoccupied orbitals of T2 symmetry. The overall wavefunction symmetry can then be expressed as the product of all MO symmetries, reduced with the reduction formula:

$$(2.23) \quad n = \frac{1}{h} \sum_R \xi_r(R) \xi_i(R)$$

where ξ_r, ξ_i are the reducible and irreducible representations respectively, h is the order of the group and R is the subspace. This correctly produced the overall symmetries of ground state systems for methane, as well as water.

However the assignment of MOs did not work well for excited states, due to the character from the subspace projection being unclear for many MOs. Additionally was that for non-abelian groups, where there are some degenerate E and T subspaces, this assignment did not work. This is a similar problem to the assignment of symmetry for the TD-DFT and EOM-CCSD transitions in the earlier benchmarking. Often the reduction of ground state wavefunctions gave non-physical answers.

Using symmetry decomposition for the non-abelian point groups and analysis based on the antisymmetric component of the direct products was discussed, however this would create a more open-ended project. Additionally, while this may be a useful feature for testing the benchmarking sets, chlorophyll molecules would be far from symmetric and so this type of assignment could not have been expected to have worked. In total transitions could not be confidently assigned for Δ -SCF with this method.

2.2.5.2 Δ -xTB Benchmarking Results

After considering this leading error, the assignment of symmetry was based on the previously used inspection of transition dipole orientations and transition density plots, however the difficulty of doing this for the entire range of methods should be noted.

The distributions of errors for each of the benchmarking methods, as well as a generated distribution of sTDA-xTB results, are shown in fig-2.7. The means and standard deviations are reported in table 2.1.

Overall, both Δ -xTB methods are inaccurate - far too inaccurate to be used as a viable method for transition properties of chlorophyll, or any other system. The mean error GFN1- Δ -xTB was -0.12 eV, and has a significant standard deviation of 2.11 eV. GFN0- Δ -xTB had a larger mean error

2.2. BENCHMARKING

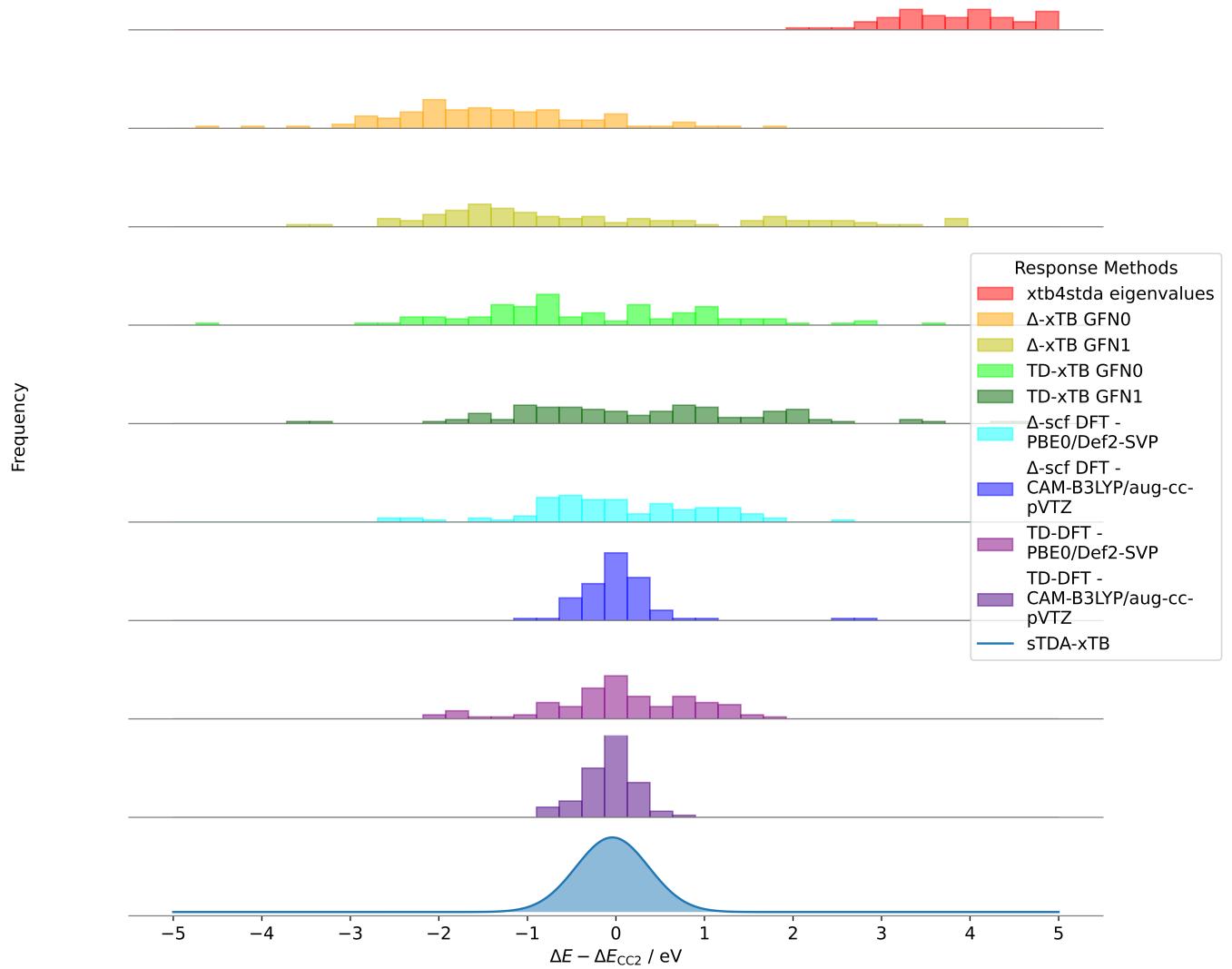


Figure 2.7: The distributions of errors to CC2 transition energies for the methods included in the Δ -xTB benchmarking.

Method	Mean / eV	Standard Deviation / eV
TD-DFT CAM-B3LYP/aug-cc-pVTZ	-0.18	0.34
TD-DFT PBE0/Def2-SVP	-0.06	0.79
Δ -SCF CAM-B3LYP/aug-cc-pVTZ	-0.14	0.28
Δ -SCF PBE0/Def2-SVP	-0.62	0.50
TD-GFN1-xTB	0.27	1.47
TD-GFN0-xTB	-0.41	1.32
Δ -SCF GFN1-xTB	-0.12	2.11
Δ -SCF GFN0-xTB	-1.50	1.08
xtb4stda	4.39	1.26

Table 2.1: Mean and standard deviations of the errors, in eV, against SCS-CC2 reference data. The `xtb4stda` entry represents the eigenvalue difference method that uses the eigenvalues output from this program.

of -1.50 eV, and whilst a slightly smaller standard deviation of 1.08 eV, this is still well beyond a usable accuracy.

The DFT methods, both the linear response and the Δ -SCF methods, are still shown to be accurate at predicting excitation energies, with means and standard deviations within ranges previously reported.

The sTDA-xTB line is generated from the mean and standard deviation of the errors as reported (-0.04 eV and 0.41 eV respectively)[11]. This was included as a reference line to show how the full sTDA-xTB method is arguably as accurate as TD-DFT with a range-based density functional and triple zeta basis sets. Again CAM-B3LYP/aug-cc-pVTZ methods, both the linear-response and Δ -SCF, are accurate to within this range as well. The outliers in the Δ -SCF results are known mixed transitions, as discussed earlier with the ethene dimer system. Overall, these results are inline with the previous set of benchmarking. The PBE0/Def2-SVP methods have a marked decrease in accuracy. On going from higher-level DFT to lower level, the standard deviation approximately doubles for both TD-DFT (0.34 eV to 0.79 eV) and Δ -SCF (0.28 eV to 0.50 eV). Additionally, Both linear-response and Δ -SCF methods perform similarly when compared to each other, and so the leading cause of error is attributed to be the electronic structure method, and not the response method.

It can be seen then that the GFN-xTB based methods performed much worse than the DFT methods. There is a relatively small difference between linear-response GFN-xTB and Δ -xTB results, again implying that poor electronic structure theory gives poor transition properties. Both the GFN0- and GFN1-xTB based methods proved inaccurate, and there is a marked drop when using GFN0. The systematic shift, shown in the mean error of -1.50 eV for response method based on out-of-the-box GFN0-xTB would not be viable. the GFN0- Δ -xTB method is especially bad. It's concluded from this that a

Overall, the most inaccurate method is the eigenvalue difference methods based orbital

energies (eigenvalues of the Hamiltonian diagonalisation) from the `xtb4stda` method. Arguably the sTDA method then makes up a large part of predicting transition properties accurately.

The result that GFN-xTB based methods are not accurate is not unexpected. As opposed to using *ab initio* or first principle parameters, the xTB methods fit to target properties, and so could not be expected to be suitable for every case [3]. Whilst the GFN-xTB methods are better than many other methods in this parameterisation, using a few pair-wise parameters as possible to improve extensibility to systems outside the training set, these parameters are trained on ground state properties and not excited state properties. These parameters are found to be unsuitable for predicting transition properties.

2.3 Conclusions

The transition properties of a test set of small molecules has been benchmarked with multiple Δ -SCF, TD-DFT and high-level methods. It has been shown that DFT based Δ -SCF and TD-DFT methods can reproduce the same transition energies and transition dipole magnitudes as EOM-CCSD to within reasonable levels of accuracy. For the set of small molecules, transition energies were predicted with a mean of less than 0.5 eV, and 0.07 a.u. for transition dipole magnitudes. Additionally, the issue of breaking the origin independence property of transition dipoles has been shown to be fixed by using a symmetric orthogonalisation of the two originally non-orthogonal states.

For a small set of BChla geometries, it was found that the same level of accuracy for transition energy could be found between Δ -SCF and TD-DFT, where EOM-CCSD was too expensive to calculate. The error was well within the range of TD-DFT energy variation, shown in the high correlation coefficient, and so Δ -SCF could be reasonably expected to give correct geometry-dependent transition energies. Whilst the accuracy is slightly reduced for transition dipole moments, the appreciable degree of correlation implies that qualitative statements would be valid.

With all of the above benchmarking, reliably obtaining and assigning transitions predicted from Δ -SCF has proved to be an unsolved issue. Either Δ -SCF is formally unable to predict the correct character of transitions, as showcased in the ethene dimer mixed transition outlier, or it is unreliable in finding excited state solutions. This is best shown in the exclusion of a geometry of chlorophyll, that would not be made to converge to the correct excited state, as well as the necessary use of Fock damping, altered DIIS procedures and intermediate initial guesses.

To solve the inability of currently implemented Δ -SCF to assign symmetry labels, a post-SCF method of assigning MO and full wavefunction symmetry was investigated, but ultimately proved beyond the scope of this project. Whilst able to assign labels for small, trivial systems of STO-3G water and methane, non-trivial excited states and more complex systems did not work. Whilst there is more work that could be done in this area, it was decided that this should be moved to

potential further work on Δ -SCF methods.

Additionally, whilst DFT based Δ -SCF methods are shown to be accurate, GFN-xTB based methods were found to be inaccurate, to the point where it cannot be claimed to be a useful proxy to higher level methods. Due to the similarity in results for linear response and Δ -SCF methods over a range of electronic structure methods, from high level DFT to lower level GFN-xTB methods, this drop in accuracy is attributed to the different electronic structure theory used, rather than the response methods. This implies that altering the electronic structure method could lead to great improvements in the accuracy of a new response method.

CHAPTER



CHLOROPHYLL SPECIFIC METHODS

Preamble this is a preamble.

3.1 The Cassida equation

3.1.1 Approximations to Solutions

3.1.2 MNOK Integrals

3.2 Parameterization

3.2.1 Reference Data

3.2.2 Objective Function

3.2.3 Minimization Algorithms

3.3 Benchmarking

3.3.1 Transition properties

3.3.2 Potential Energy Surfaces

3.3.3 Absorption Spectra

Excitation Energies correlation

	PBE0	dscf	eigdiff	camb3
PBE0	1	0.915415	0.930766	0.952334
dscf	0.915415	1	0.846607	0.863979
eigdiff	0.930766	0.846607	1	0.925034
camb3lyp	0.952334	0.863979	0.925034	
BLYP	0.730346	0.637576	0.606743	0.6376

Figure 3.1: correlations of energies

Transition Dipole Magnitudes correlation				
	PBE0	dscf	eigdiff	camb3lyp
PBE0	1	0.5891	0.566434	0.762861
dscf	0.5891	1	0.977046	0.800859
eigdiff	0.566434	0.977046	1	0.773252
camb3lyp	0.762861	0.800859	0.773252	1
BLYP	0.0832043	-0.0344407	-0.0098512	0.0500000

Figure 3.2: correlations of transition dipole moments

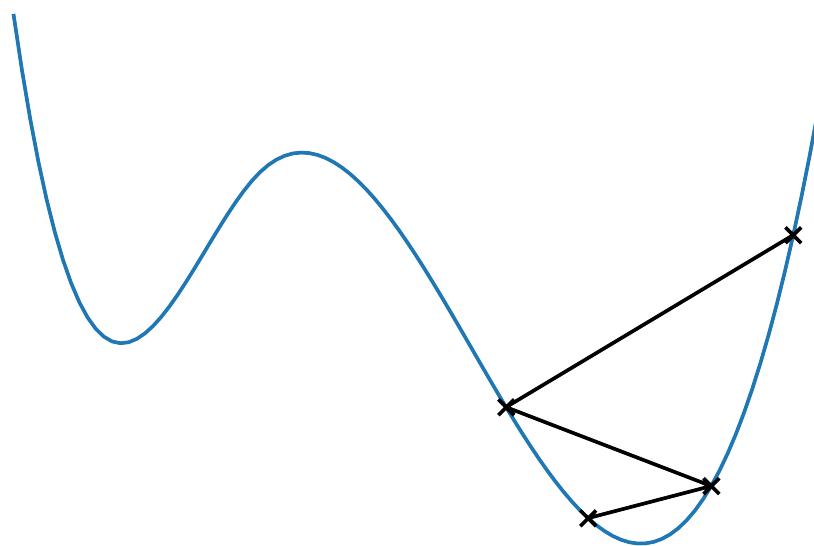


Figure 3.3: Nelder-mead

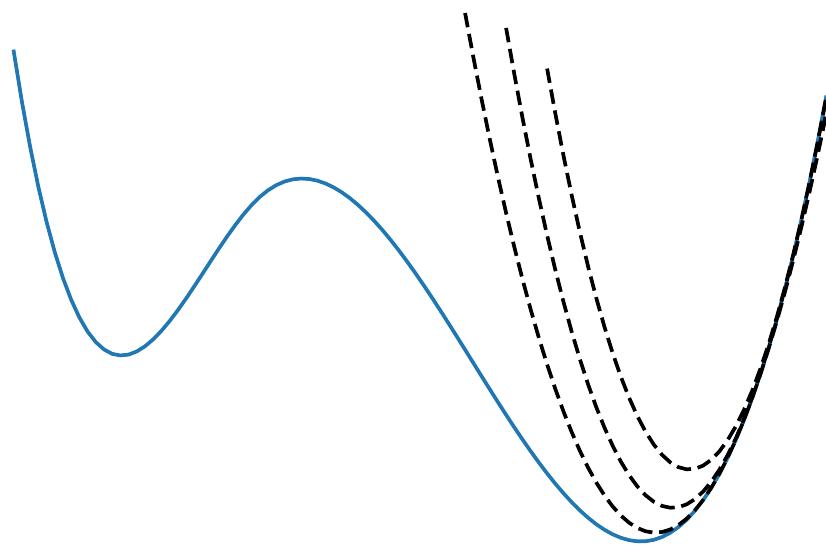
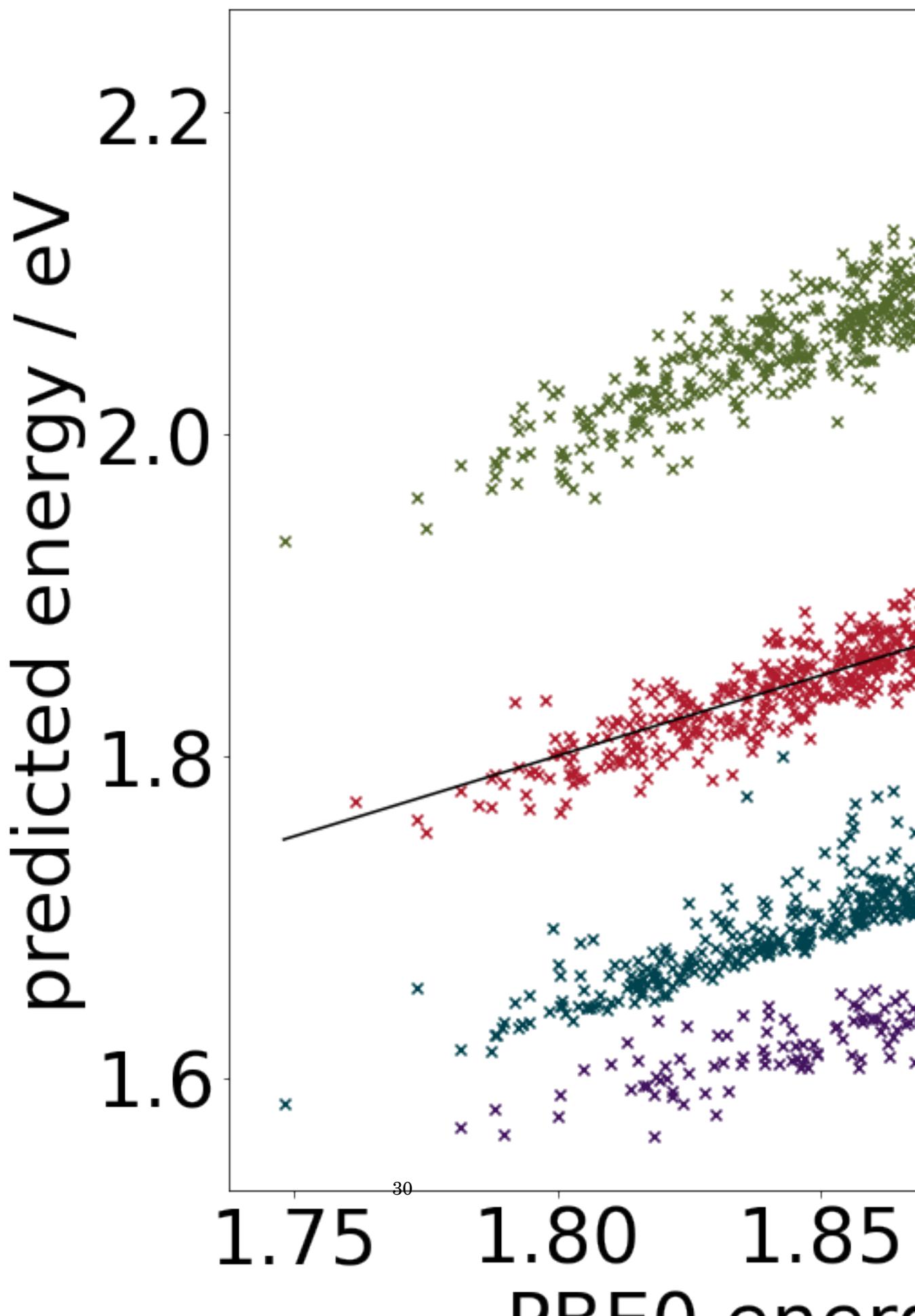
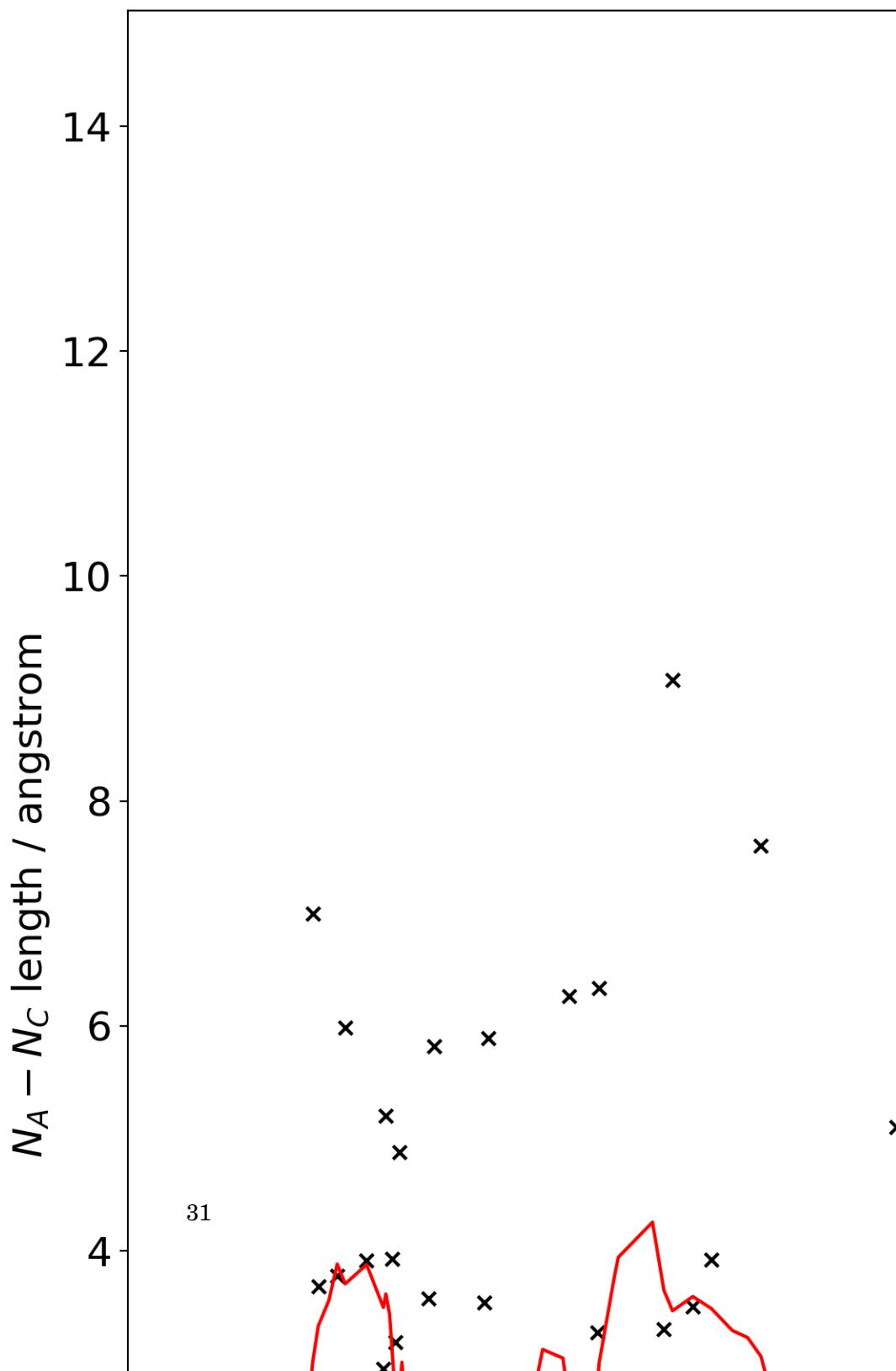


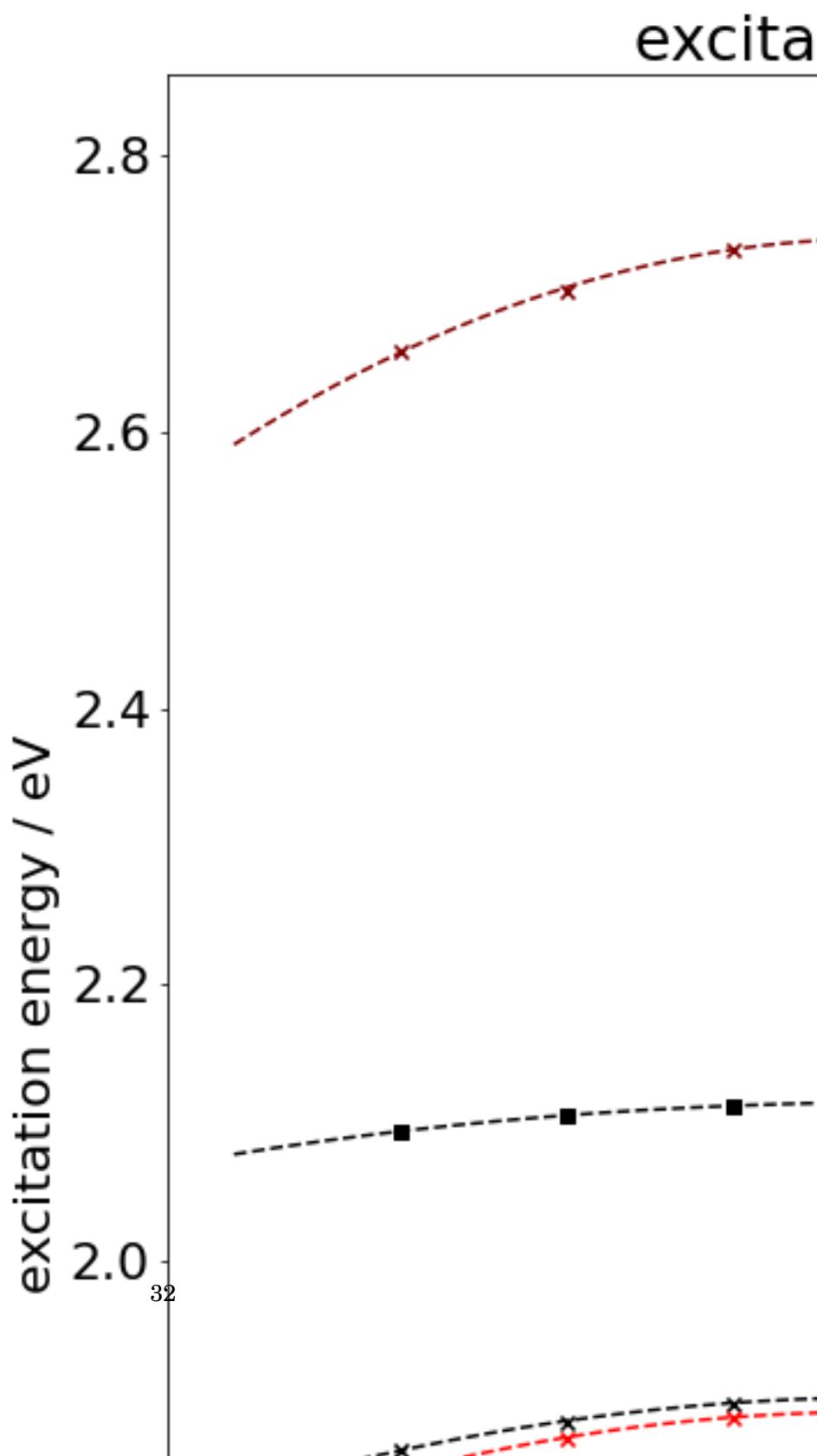
Figure 3.4: SLSQP

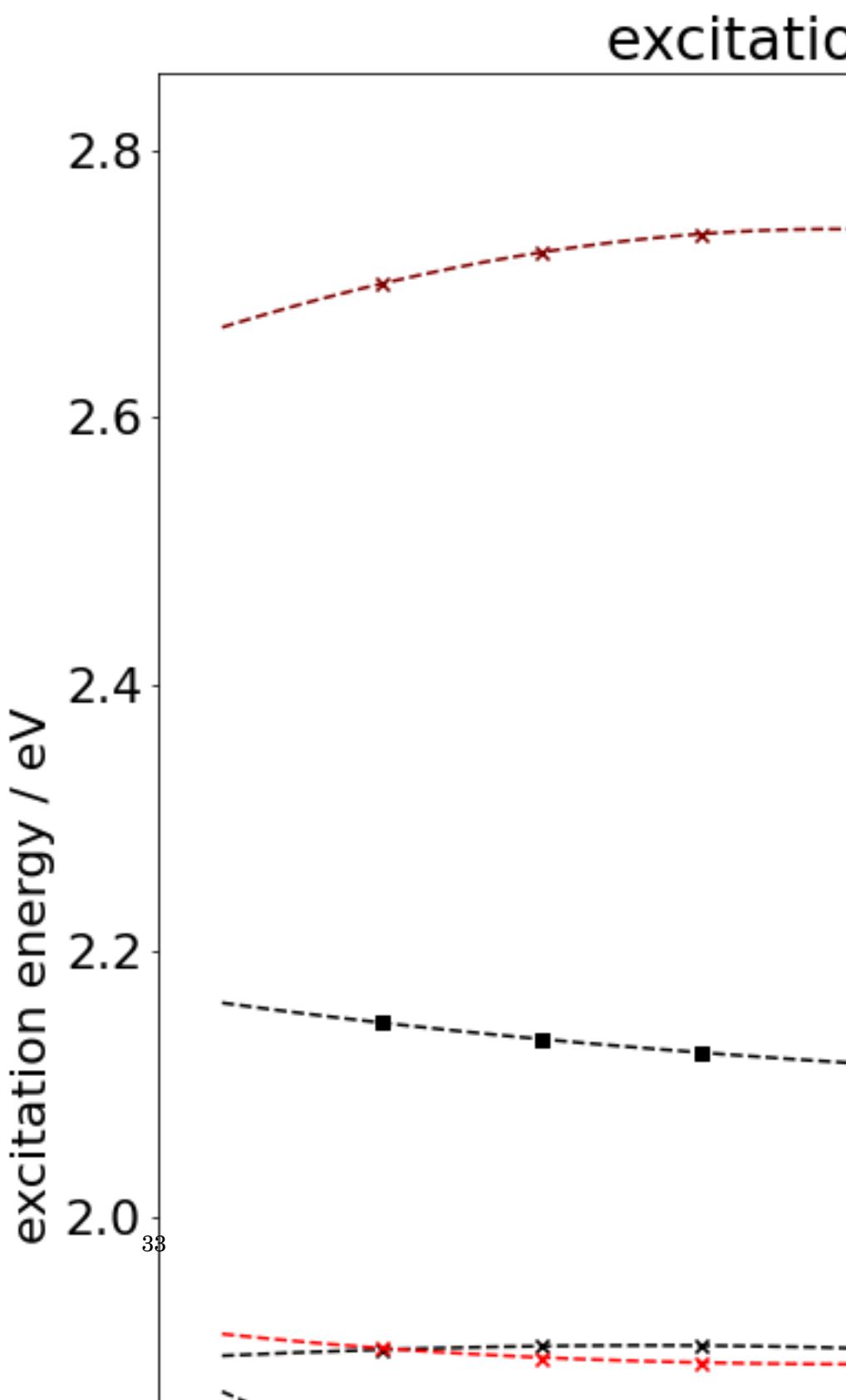
y_K	1.0
y_J	1.0
a_x	1.0

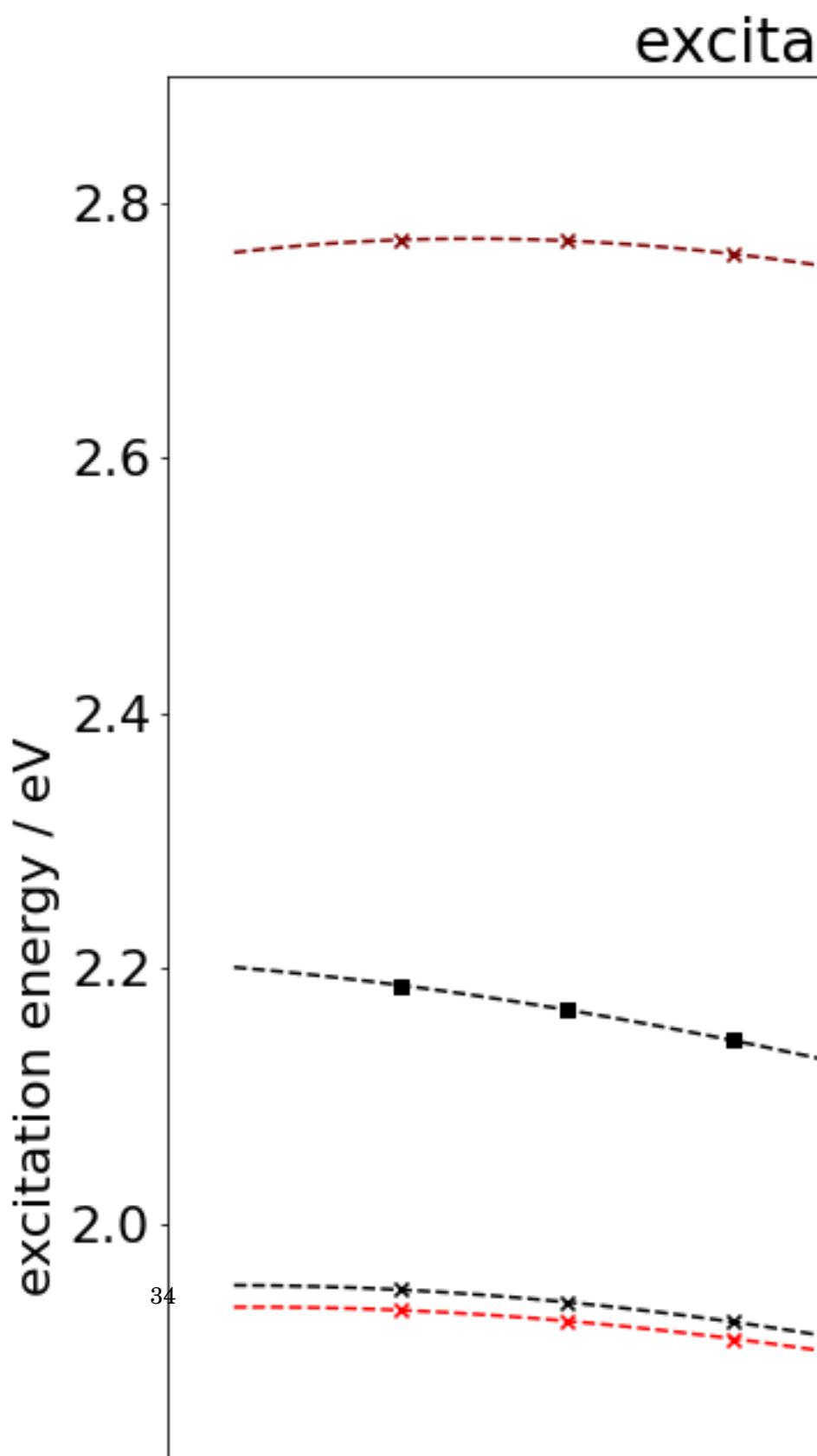
Figure 3.5: optimized parameters from SLSQP procedure.

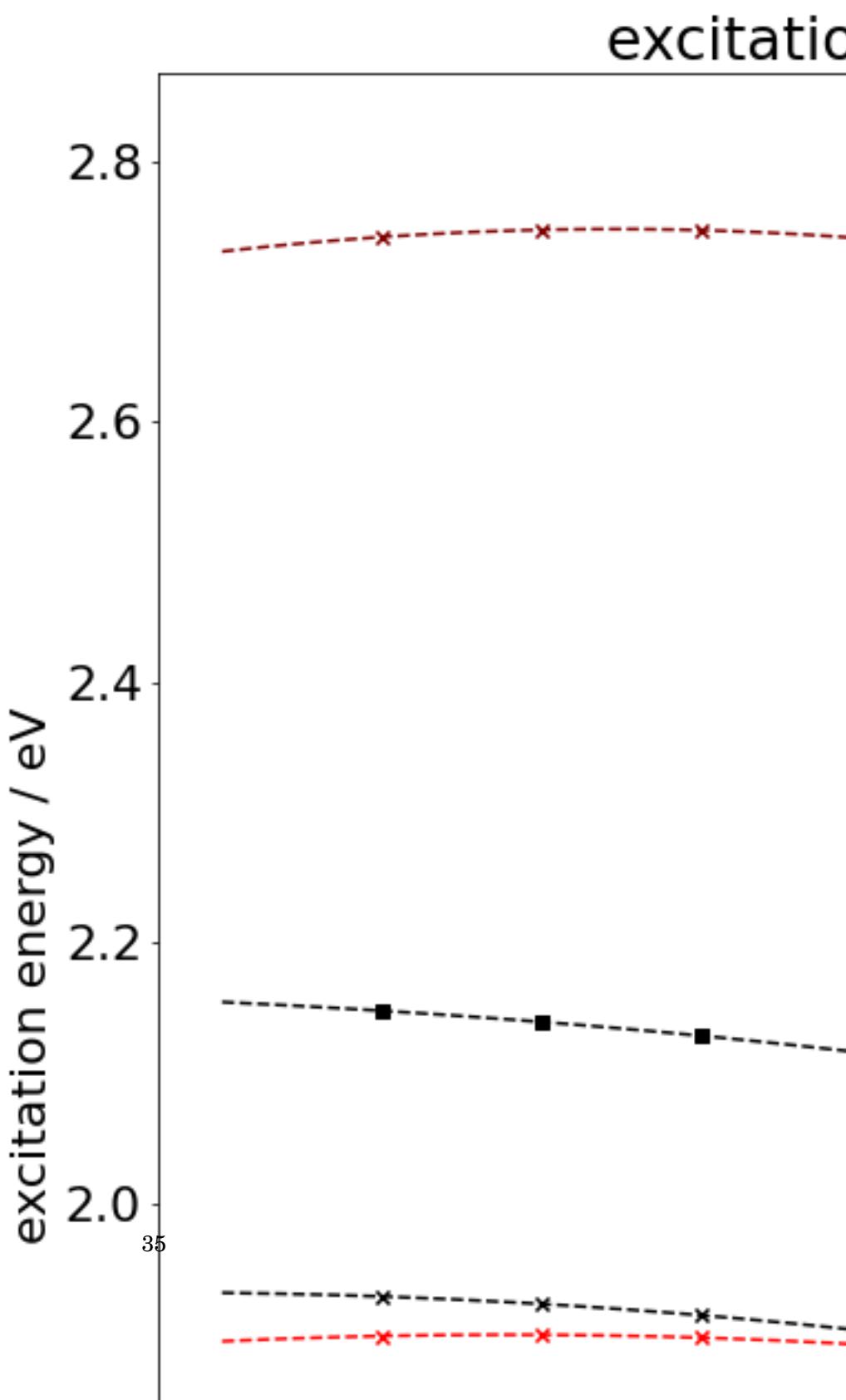


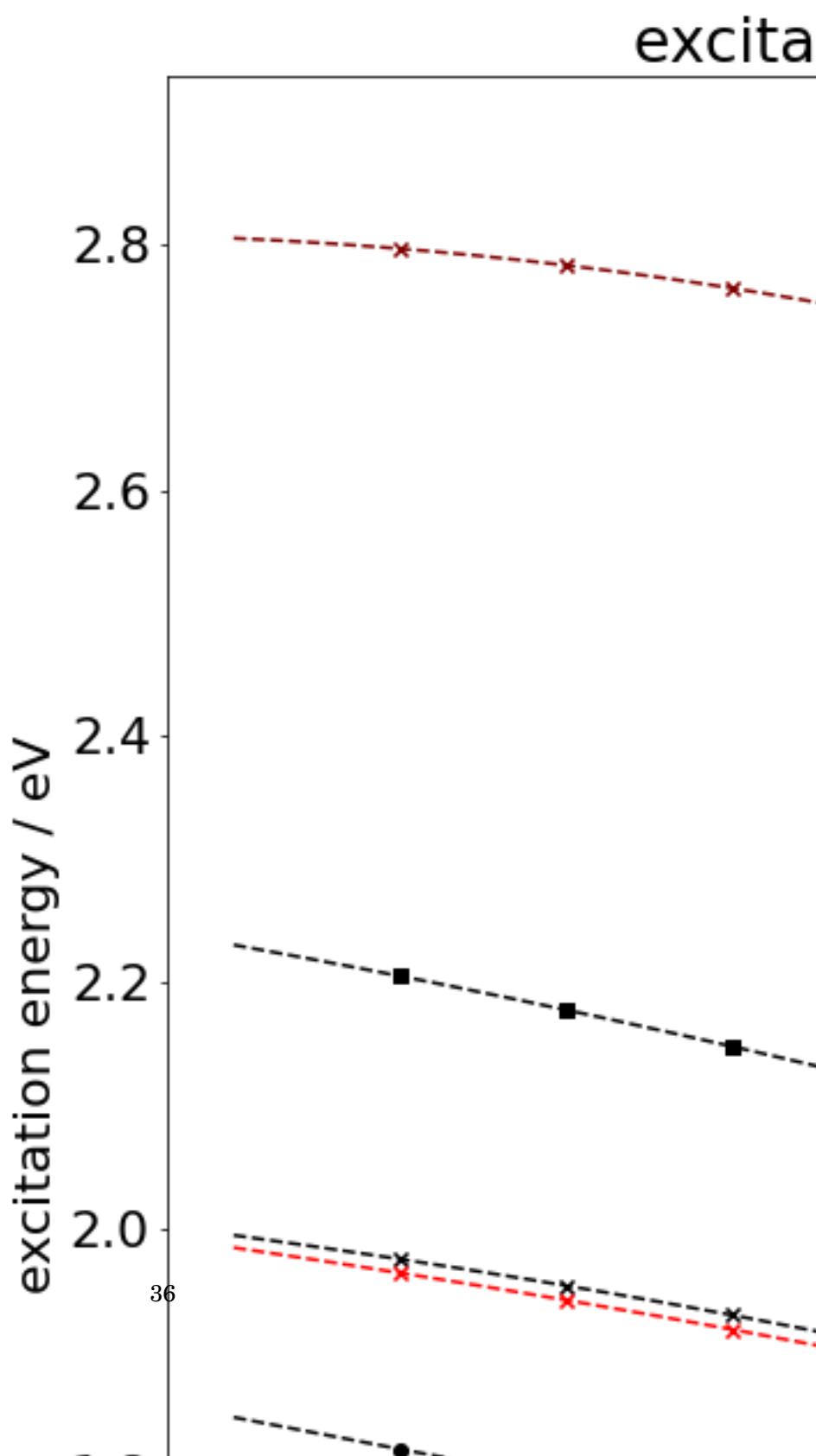


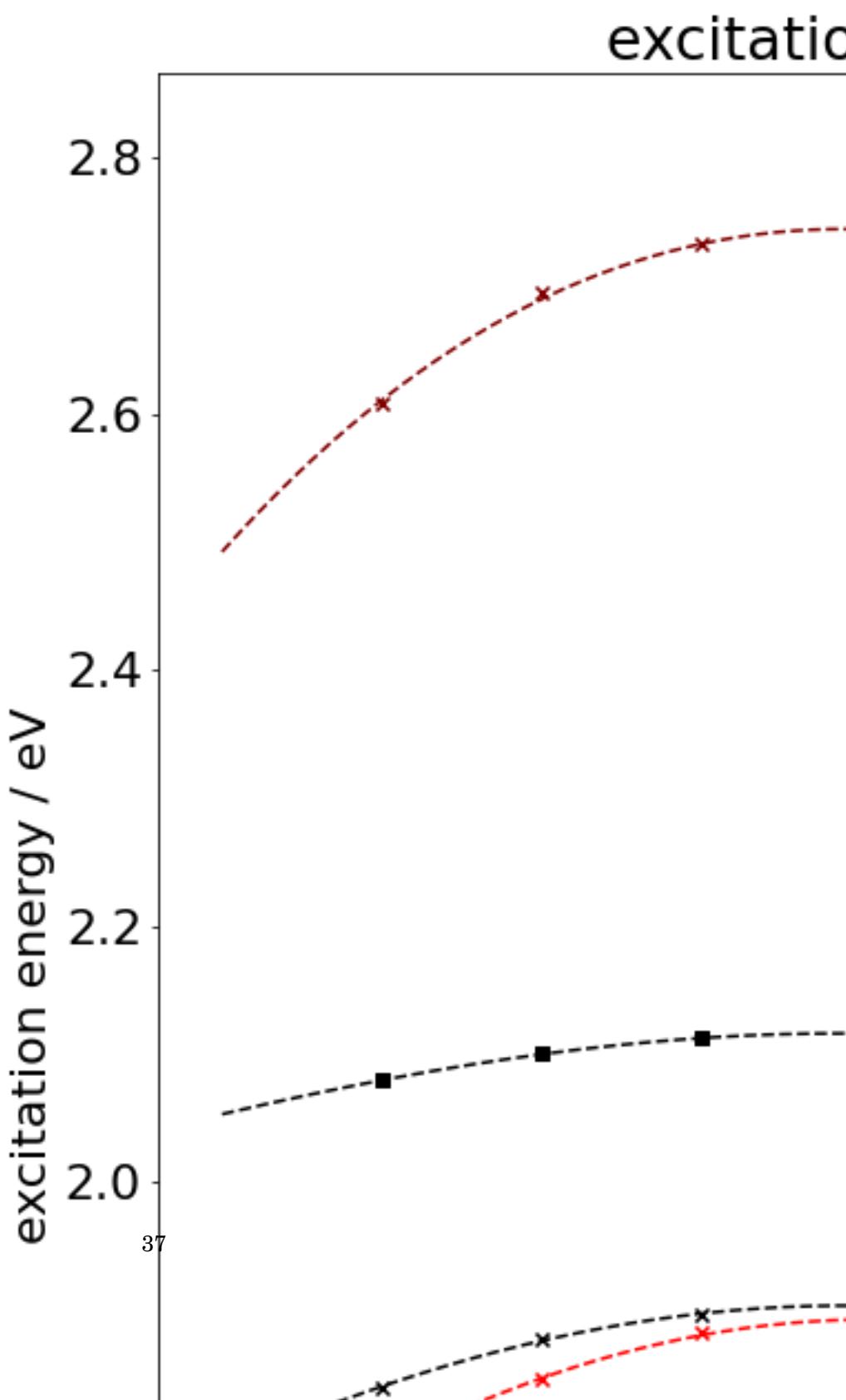


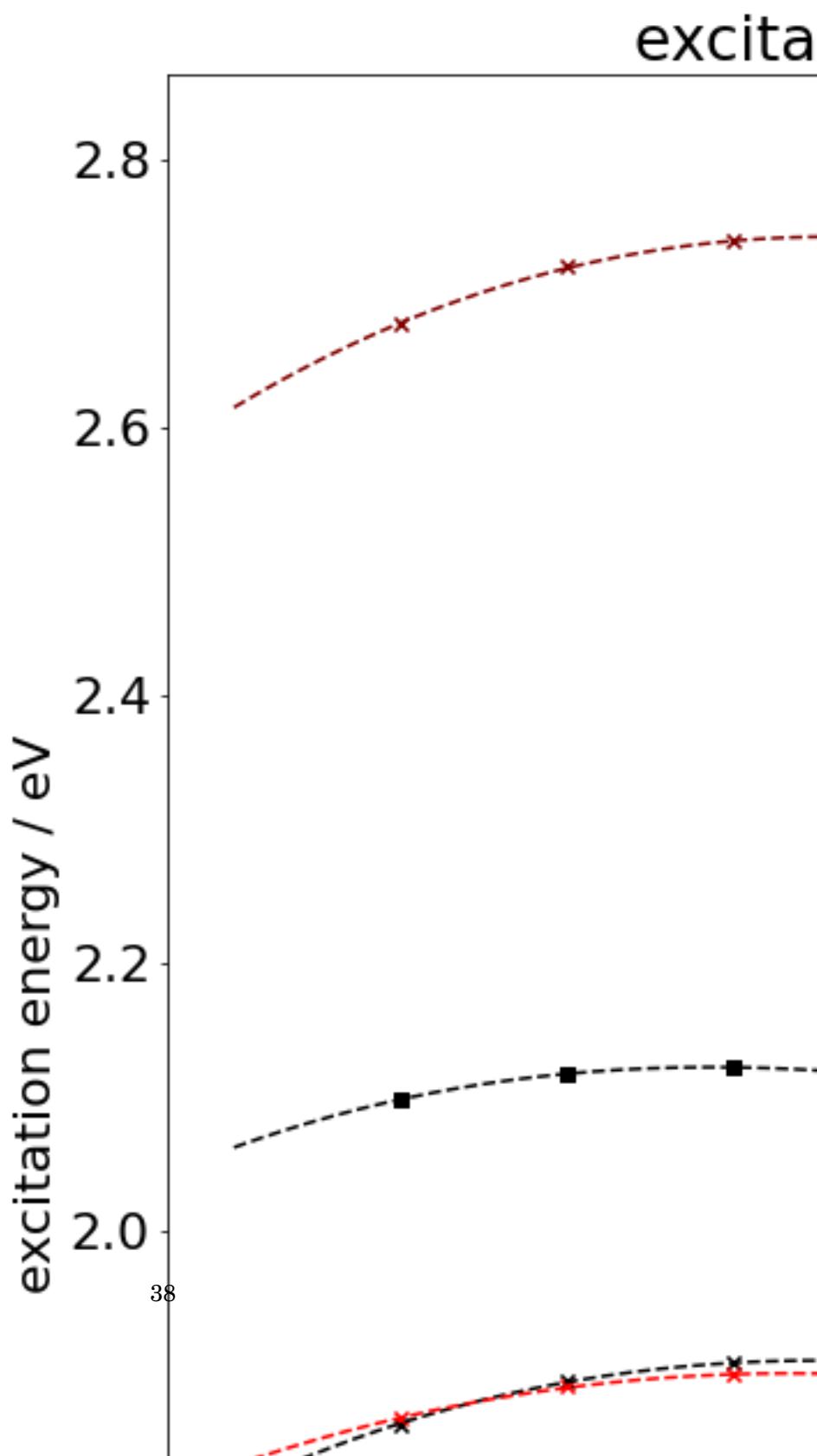


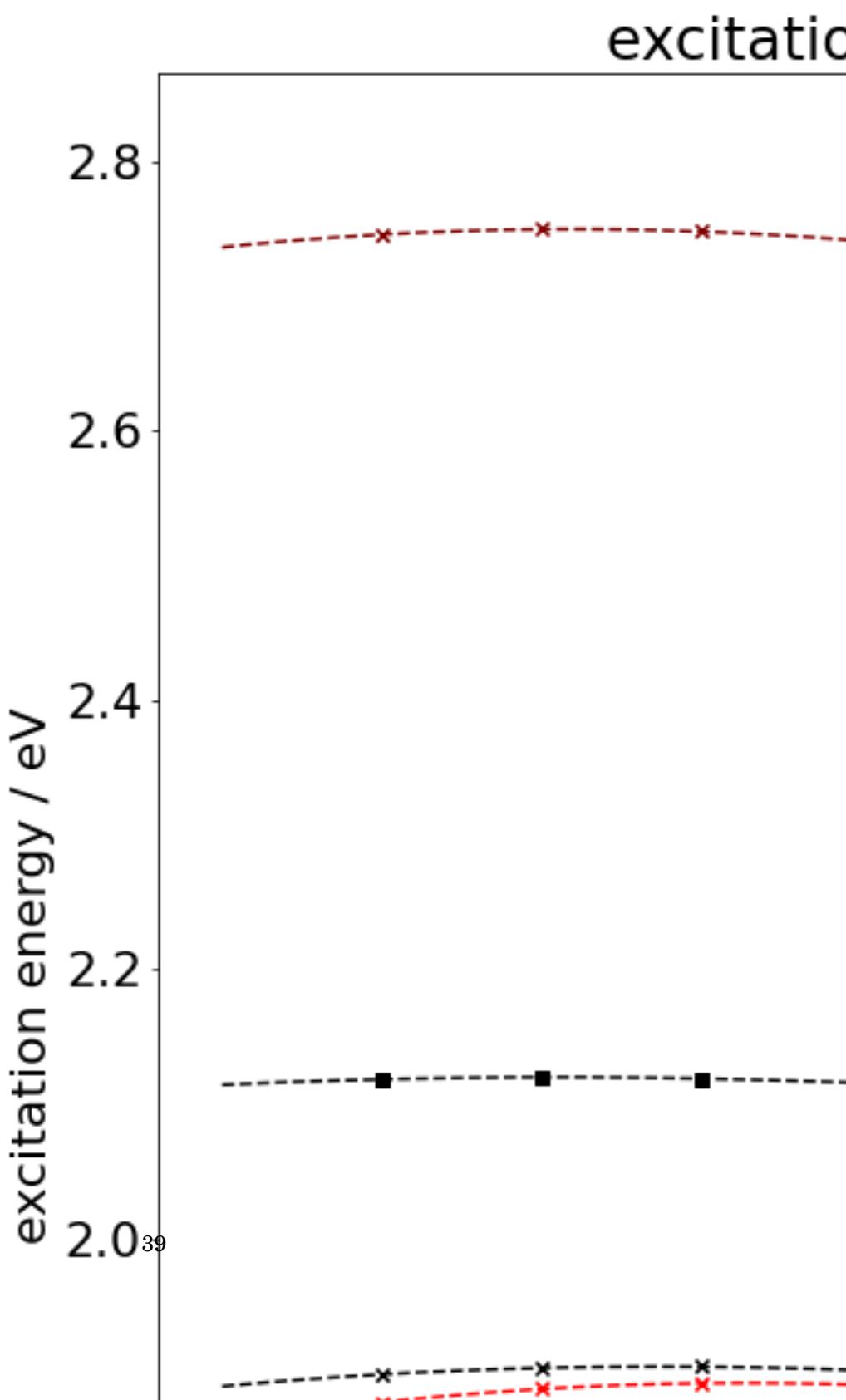


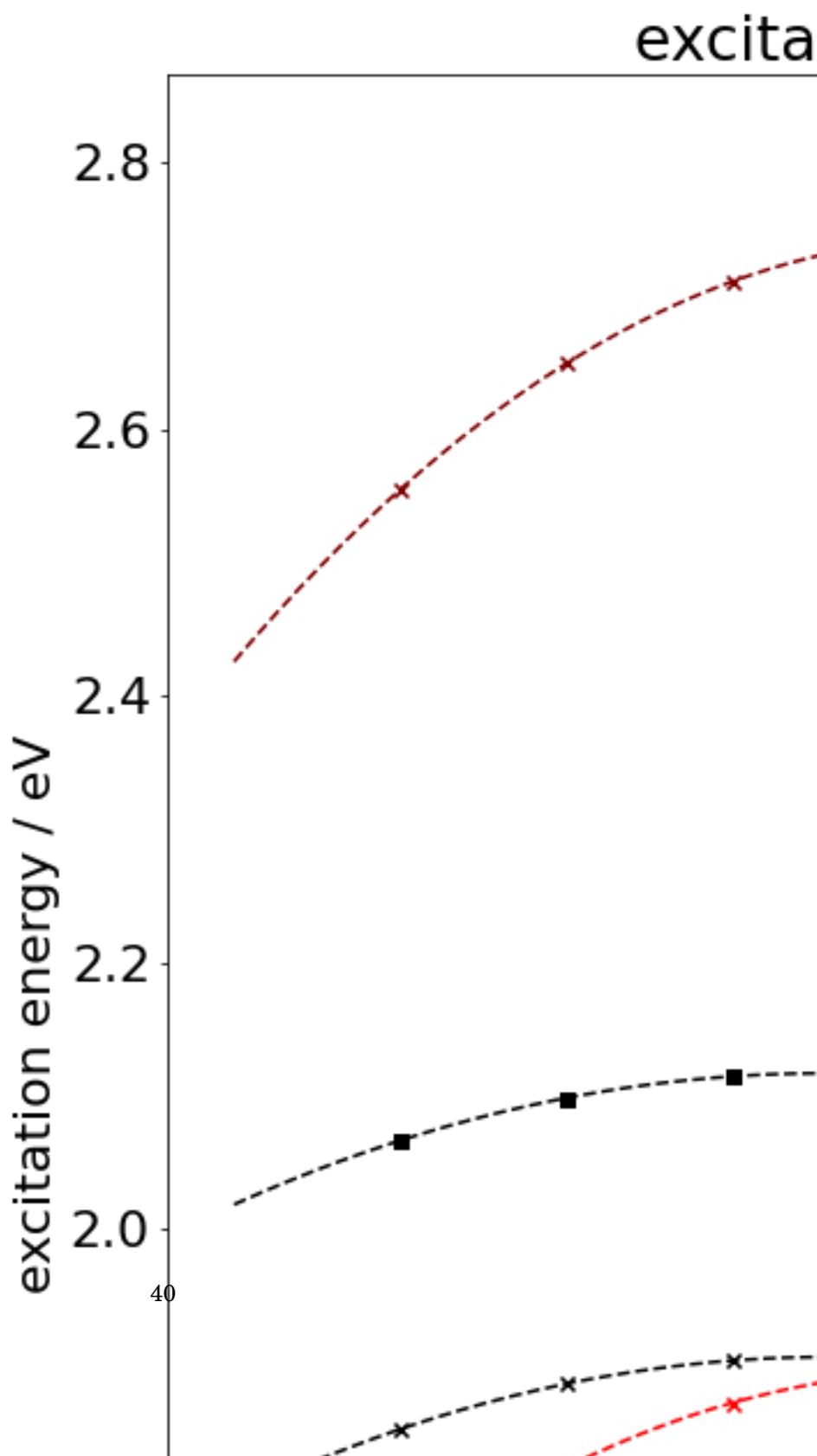


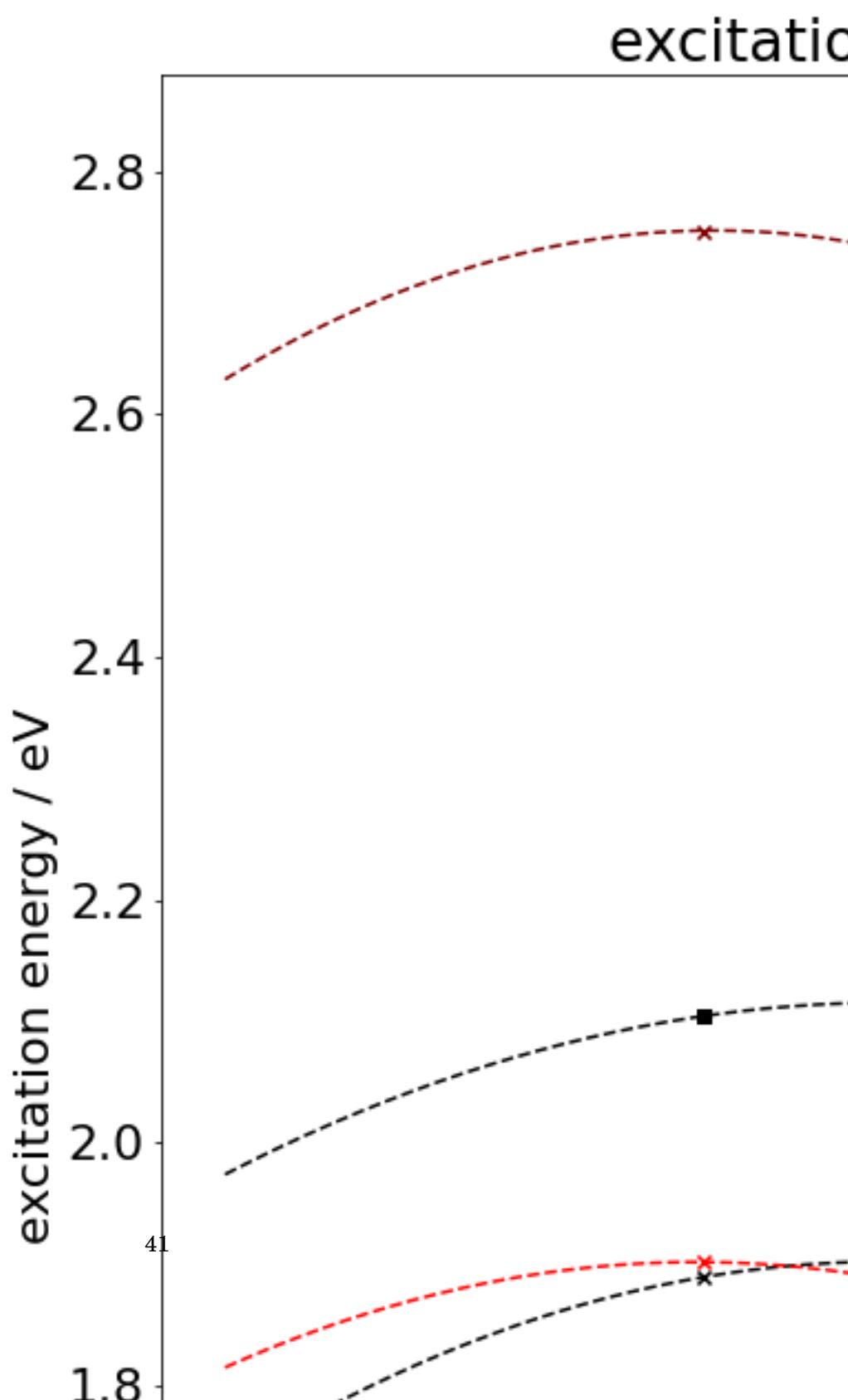


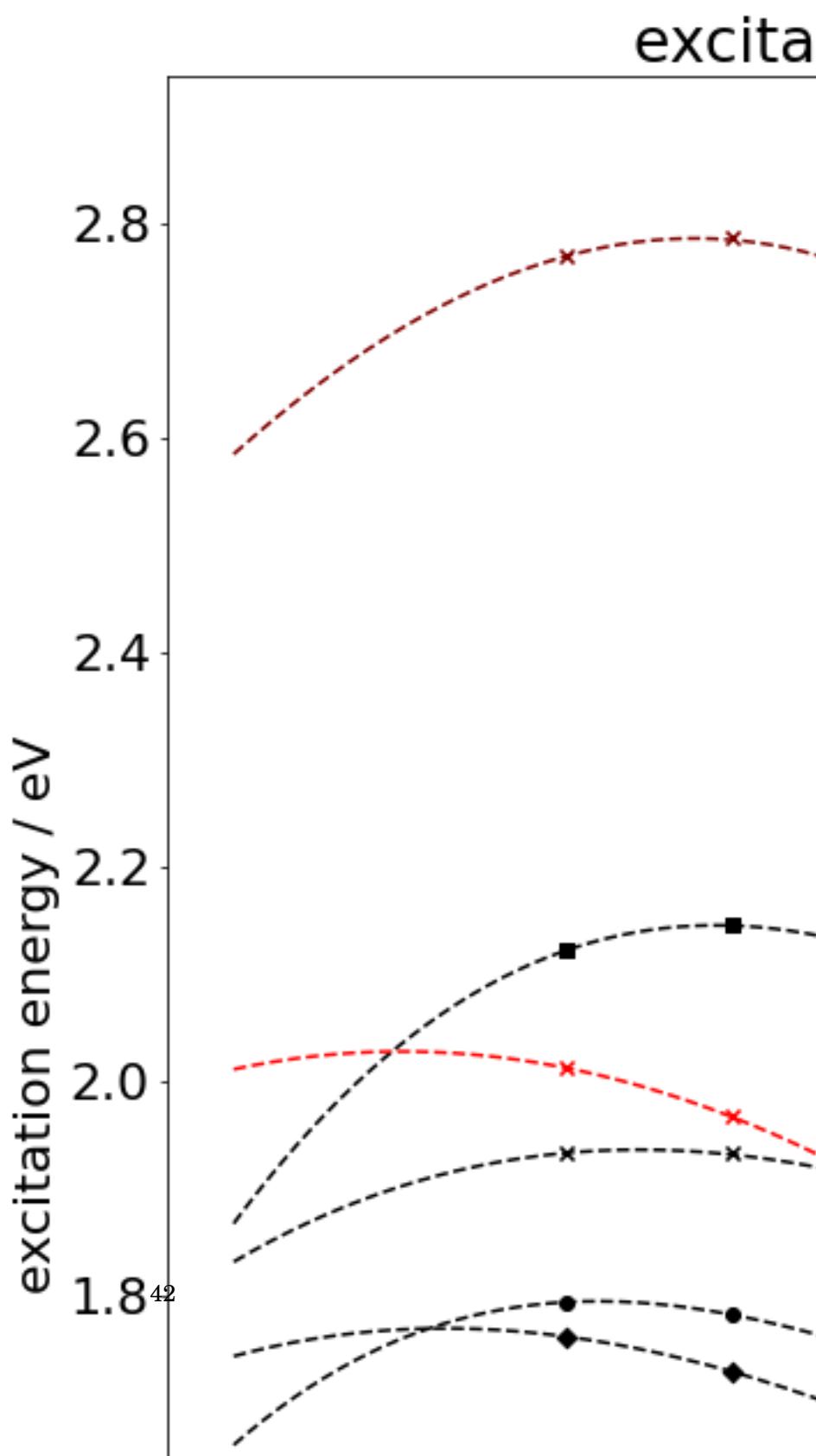












C H A P T E R



EXCITON METHOD

P^{reamble}

4.1 Theory

4.1.1 Exciton States

4.1.2 Embedding

4.2 Truncated Chlorophylls

4.2.1 Rotation

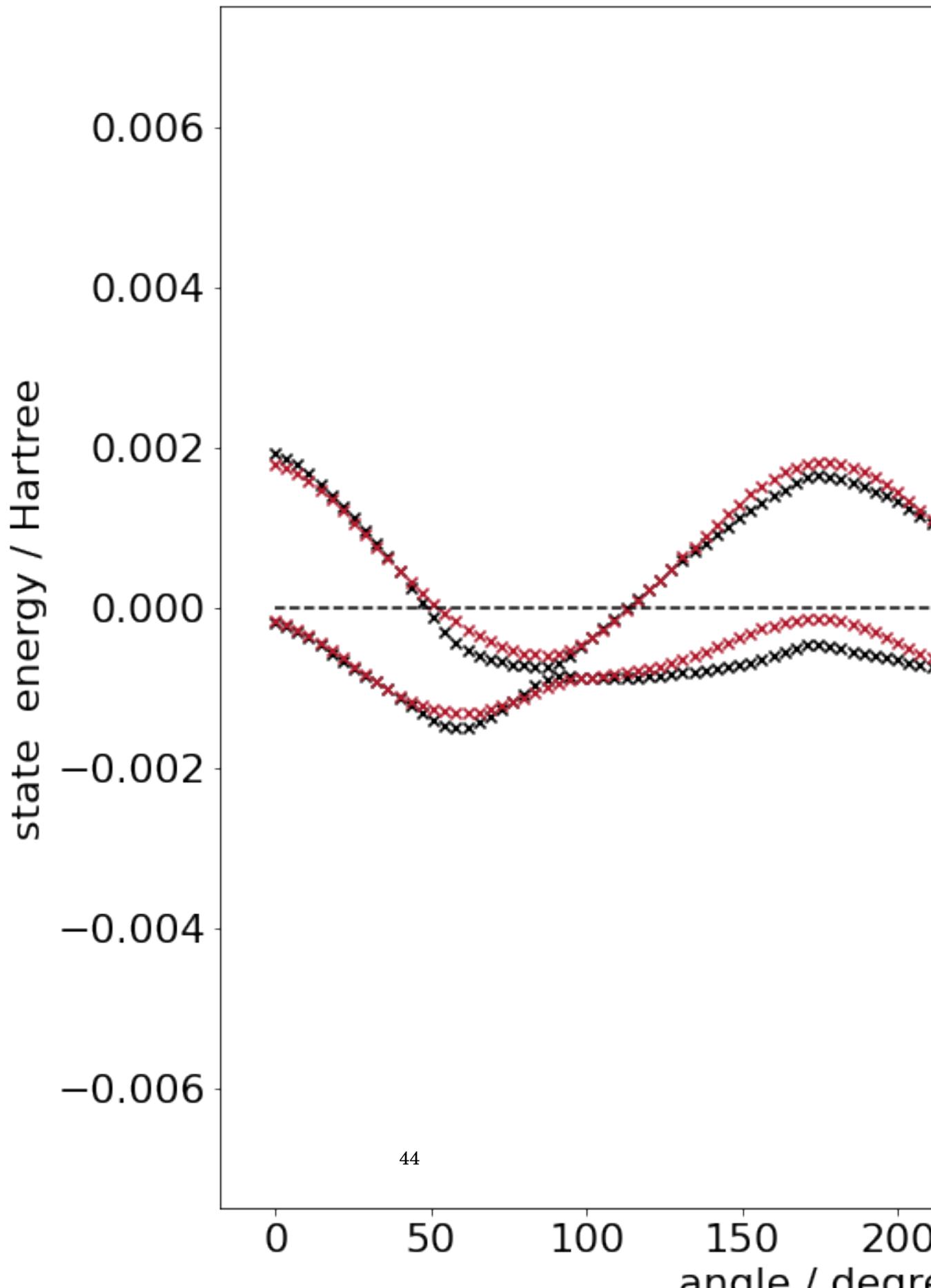
4.2.2 Distance

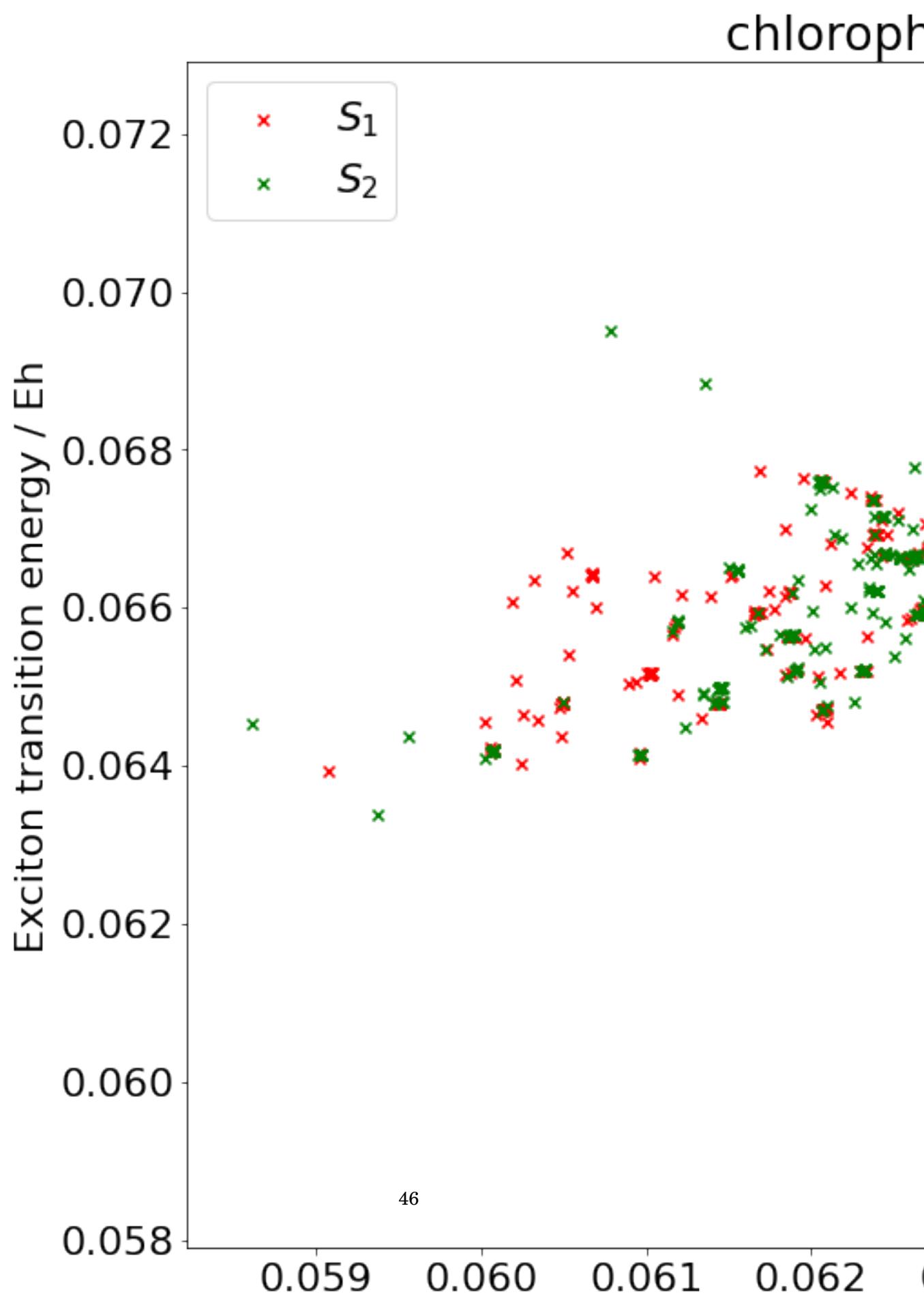
4.3 LHII Pairs

4.3.1 Assignment of States

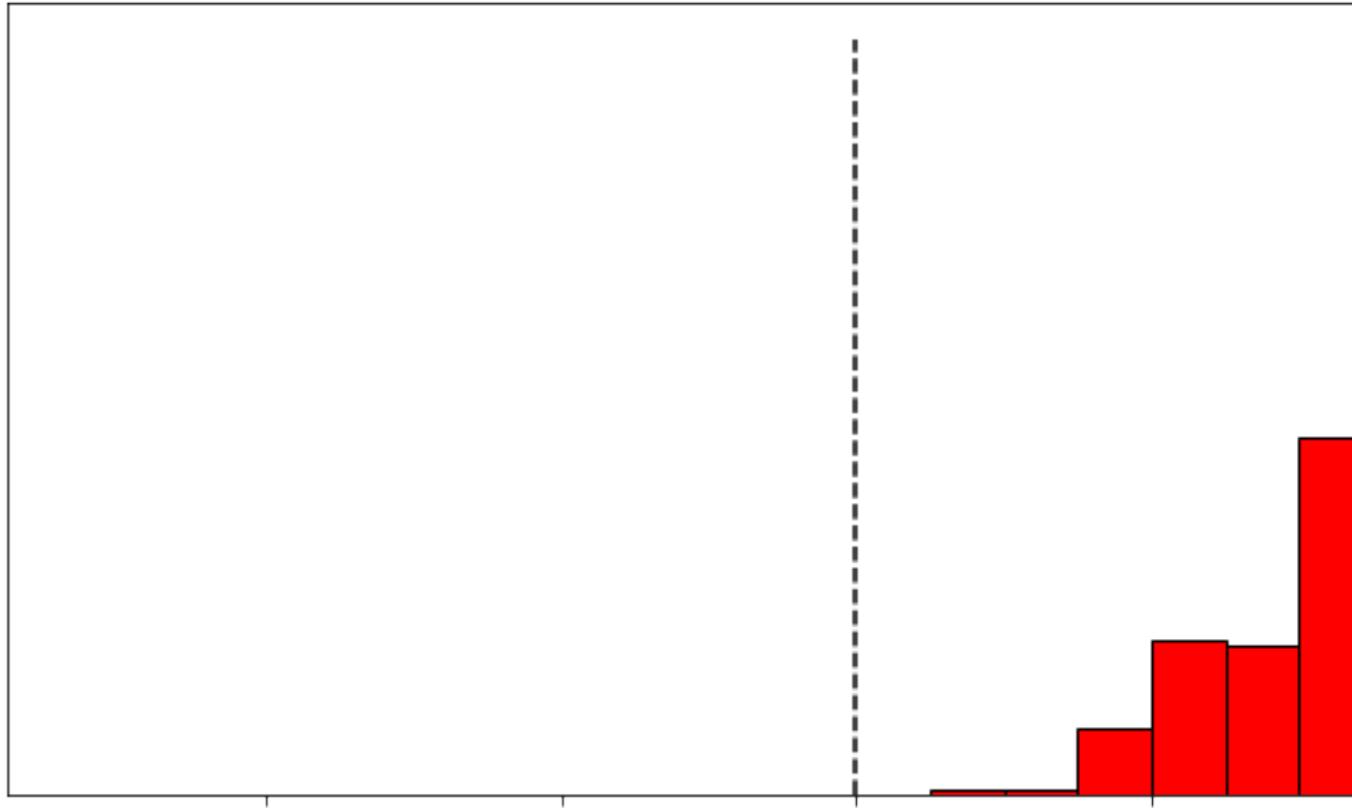
4.3.2 Comparison

Excited states of a ro

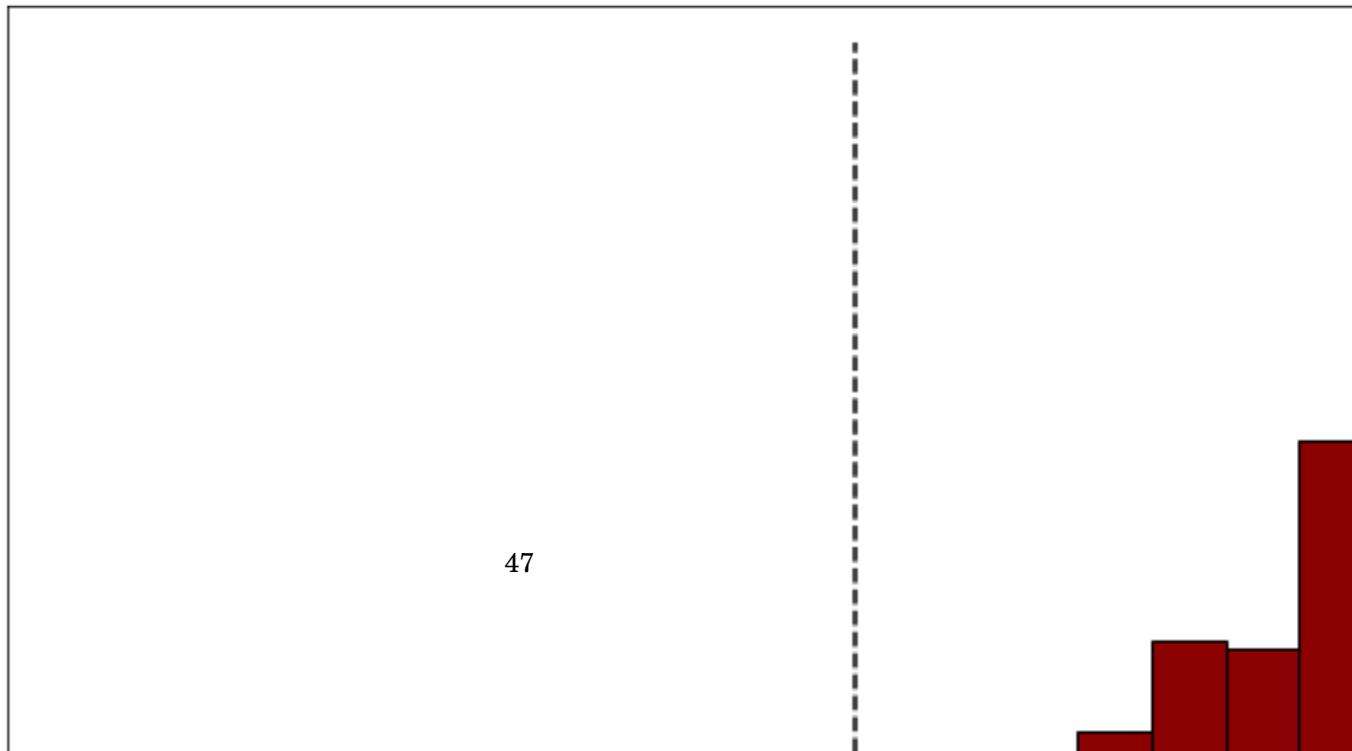




energy ordered



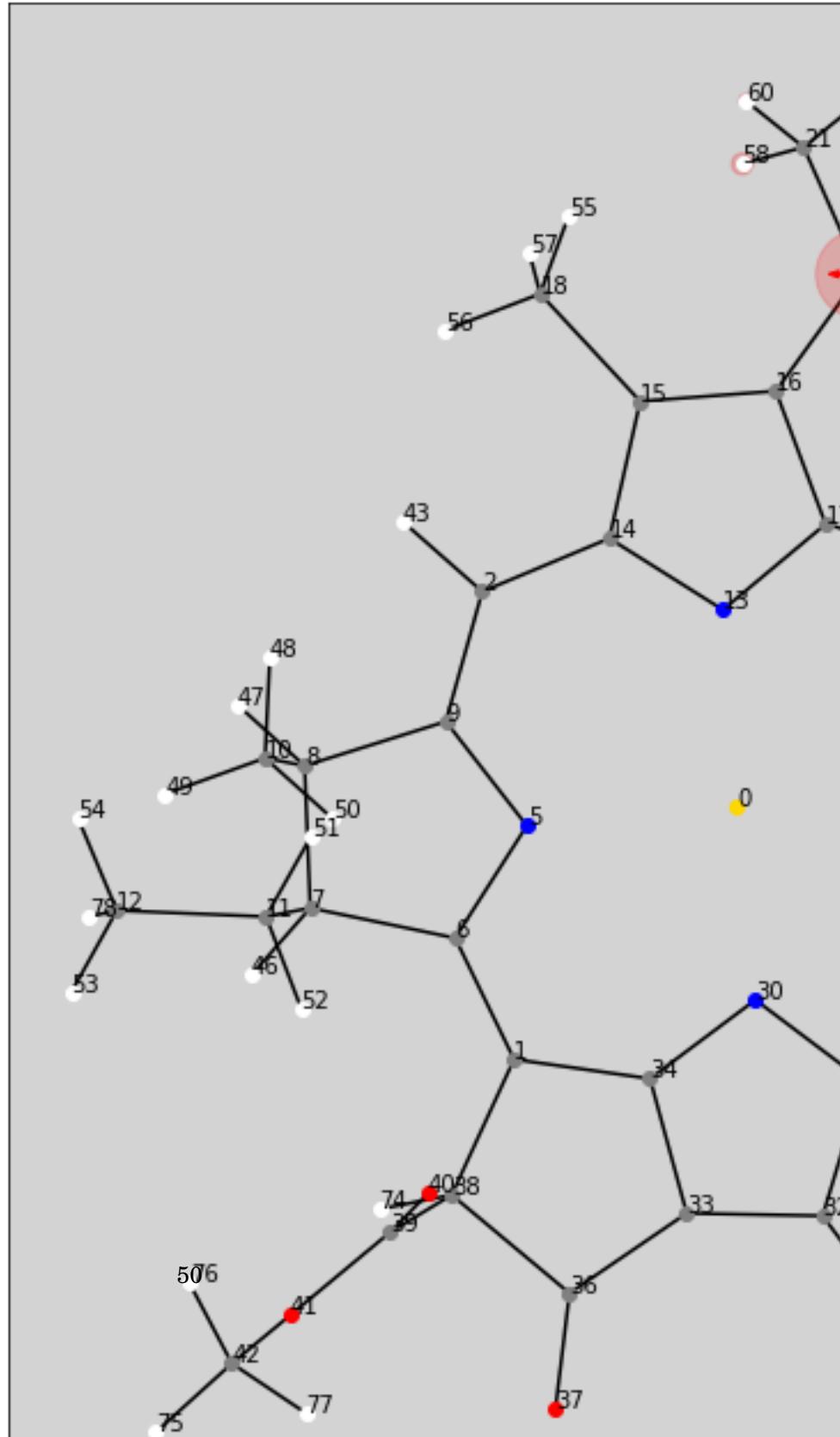
centre ordered

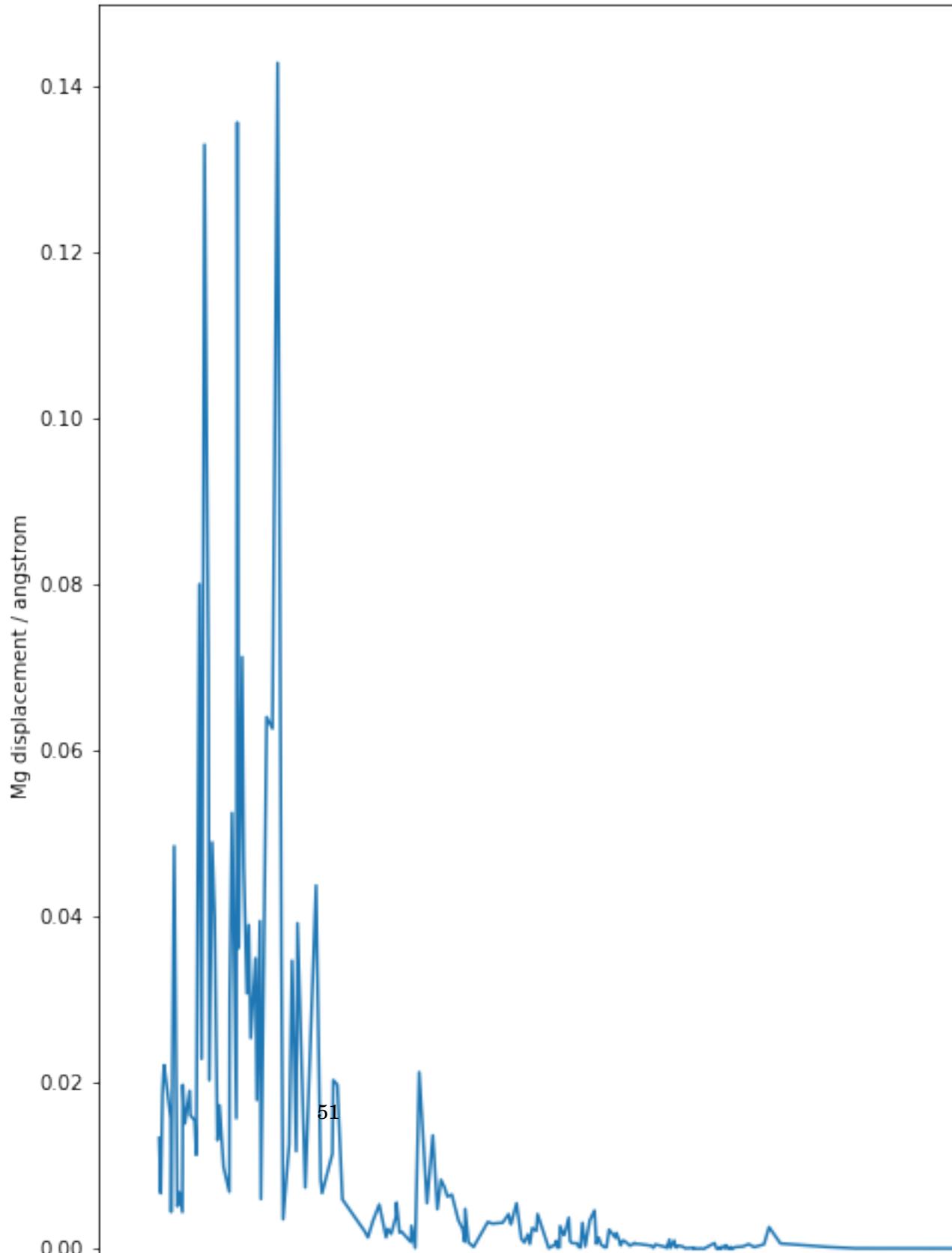


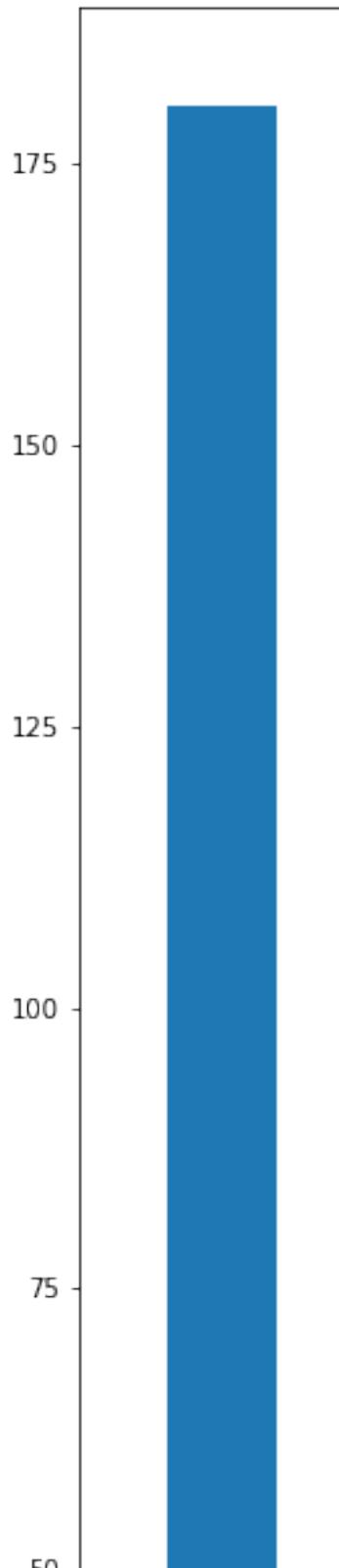
CHAPTER

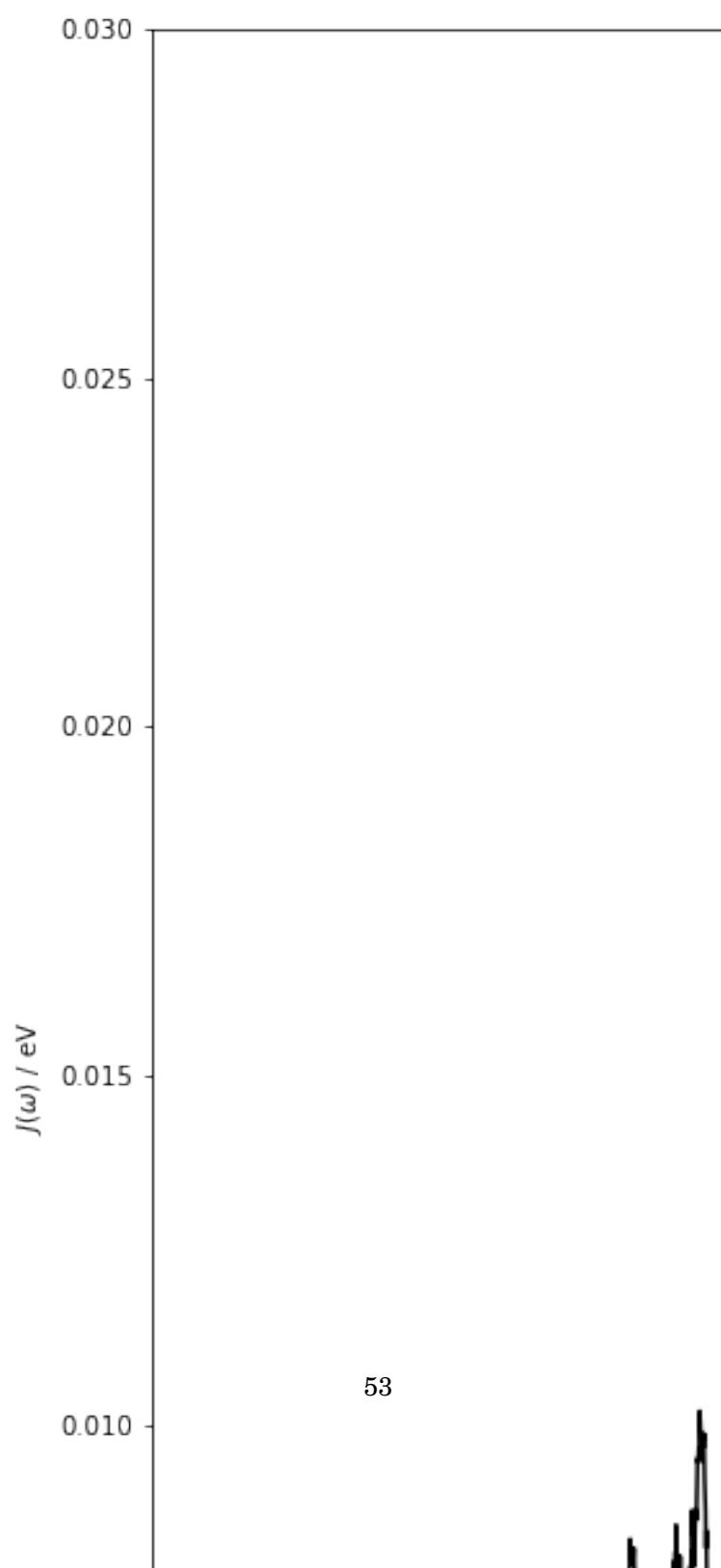


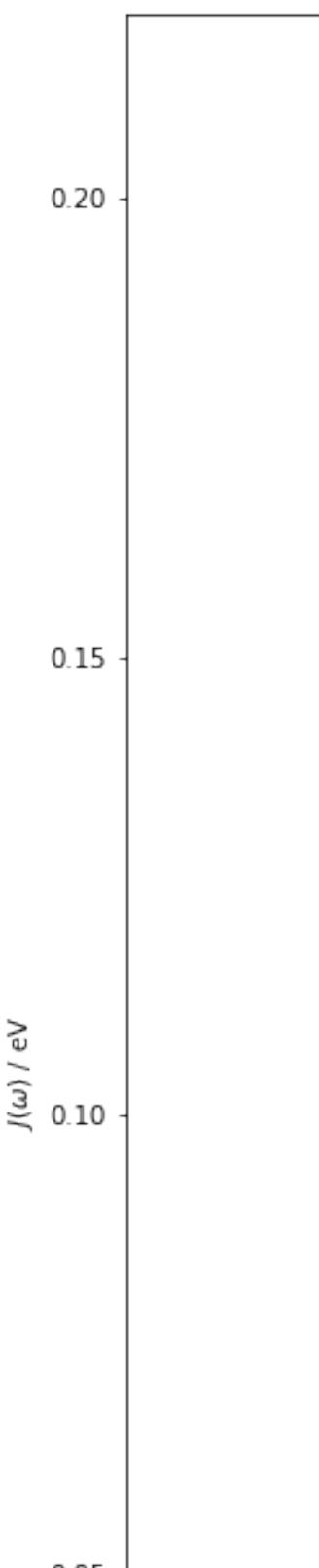
ATOMISTIC MODELLING OF LIGHT HARVESTING COMPLEXES

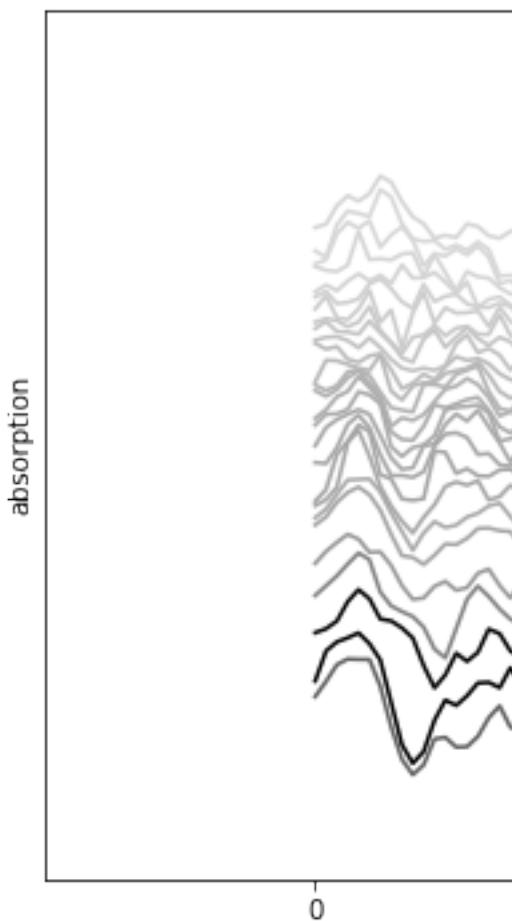


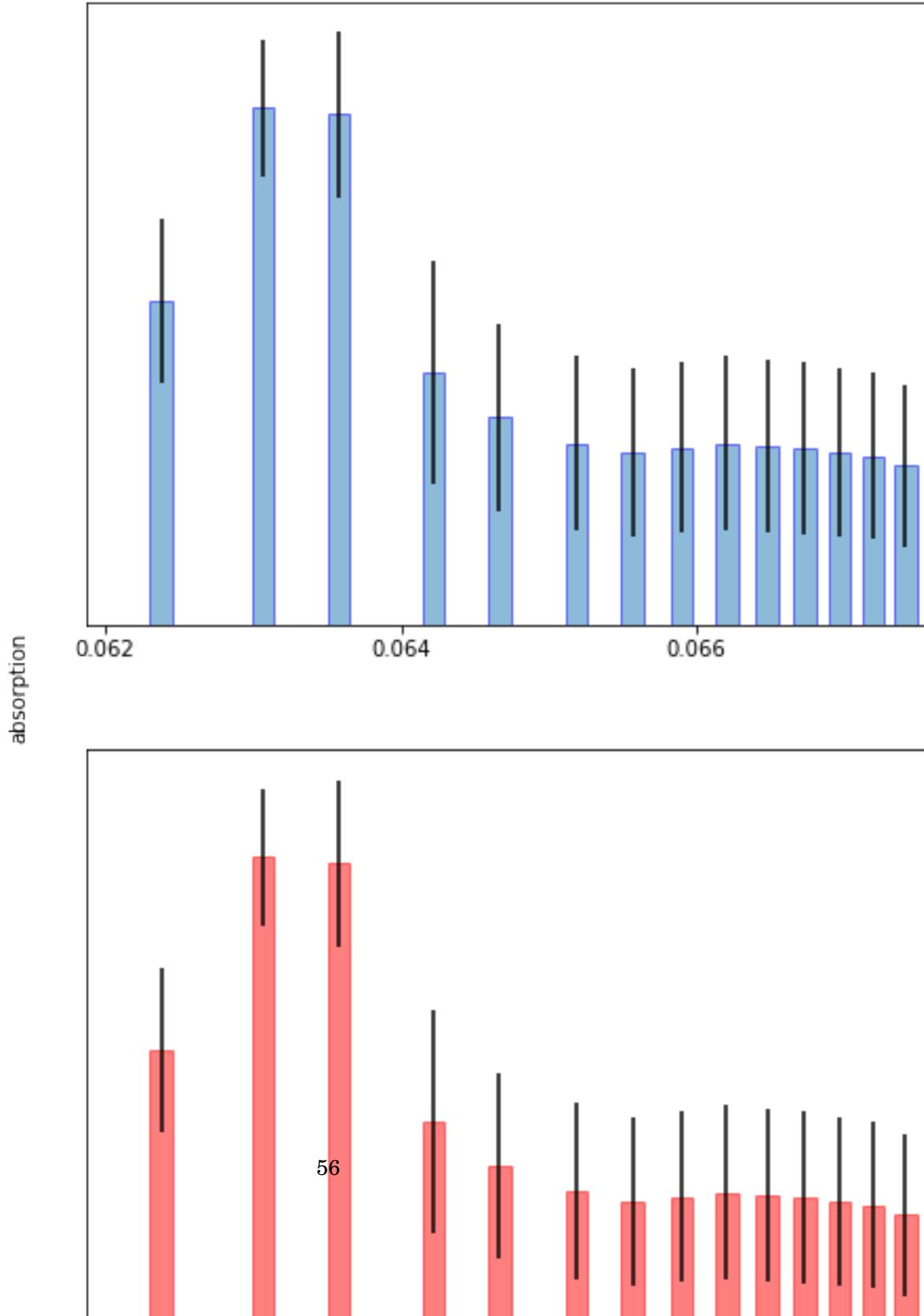












5.1 LHII

5.1.1 Spectral Density Method

5.1.2 Molecular Dynamics Method

5.2 Approximating Spectral Densities

5.2.1 Hessians

5.2.2 Huang Rhys Factors

5.2.3 Chlorophyll distances

5.3 Environmental Effects

5.3.1 Screening

5.3.2 Embedding

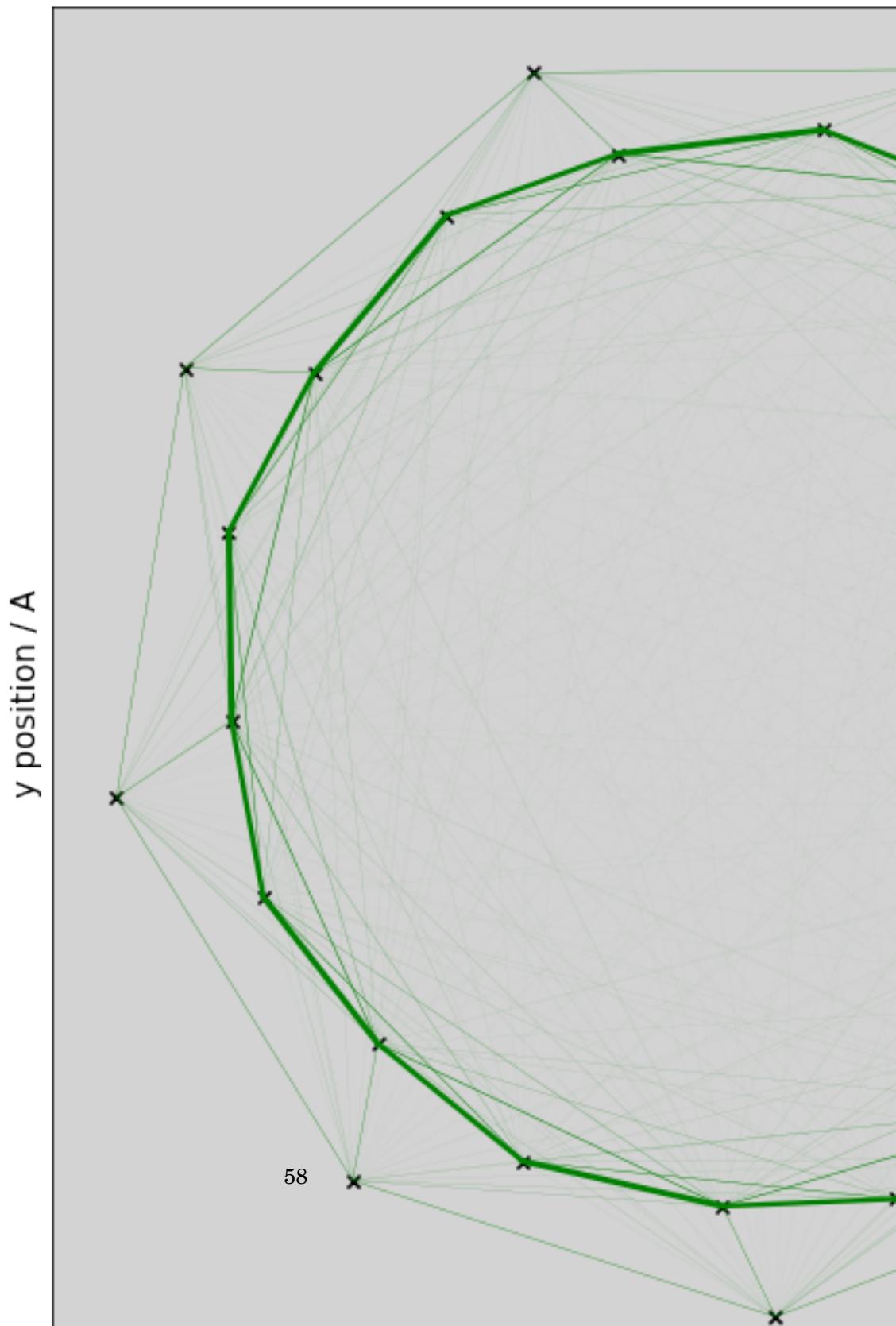
5.4 Sites, states and couplings

5.4.1 Sites

5.4.2 Exciton states

5.4.3 Coupling

5.5 Excitation Energies



DISCUSSION

Preamble

6.1 Transition Property Approximations

6.2 Further Investigations into LHII

6.3 Coherence



APPENDIX A

This appendix covers the common computational details of this work. Included are the software packages, hardware used. These are not exhaustive list, and additional details are provided in the main chapters. However, wherever implementations or methodology details are missing, the information will be found here.

A.1 Electronic Structure Codes

This project has primarily used the QCORE software that is found as part of the ENTOS project. This is a software package for DFT and DFTB electronic structure calculations that has been written as a joint venture between the Miller group in California Institute of Technology and the Manby group in the University of Bristol. It is now being hosted by Entos Inc. It is a novel C++ implementation, with a focus on modularity, functional code and modern development practices to enable easier, cleaner and more reuseable code. All novel methods discussed in the chapters have been implemented in the QCORE package.

A.2 Computational Hardware

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