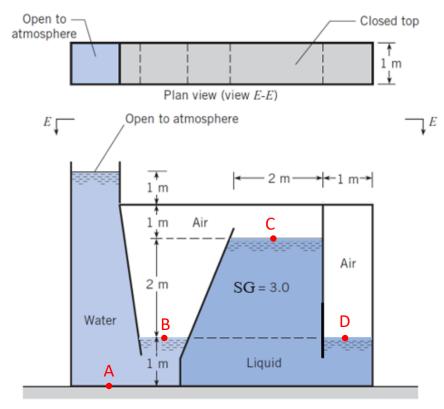
Mechanical, Automotive and Materials Engineering Fluid Mechanics I MECH3233-F23

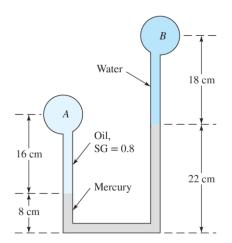
Assignment Problems Set #2

Due: Thursday, September 28, 2023, at 11:59 p.m.

Problem 1 (2 points): The tank with open and closed compartments contains water and immiscible liquid with SG = 3 as shown in the figure below. Calculate the **gage** pressures at points A, B, C and D. Assume density of water $\rho = 998 \text{ kg/m}^3$.



Problem 2 (3 points): For the system shown below, compute the pressure difference P_A - P_B . Assume density of water $\rho_w = 998 \text{ kg/m}^3$ and density of mercury $\rho_{Hg} = 13550 \text{ kg/m}^3$.



September 21, 2023

Problem 3: The Martian atmosphere where $g = 3.71 \text{ m/s}^2$ is almost entirely made up of carbon dioxide $(R_{CO_2} = 189 \frac{J}{kg.K})$. The average pressure on the surface of Mars is 700 Pa and the temperature drops off exponentially with altitude, z, as given below

$$T = T_0 e^{-Cz}$$

where $C = 1.3 \times 10^{-5} \text{ m}^{-1}$ and $T_0 = 250 \text{ K}$.

- (a) (3 points) Determine the formula for the variation of pressure with altitude.
- (b) (3 points) Find the altitude where pressure on Mars has dropped to 1 Pa.

Problem 4 (3 points): Determine the gage pressure at point A in Pa. Is it higher or lower than p_{atm} ? Assume density of water $\rho_w = 998 \text{ kg/m}^3$, density of mercury $\rho_{Hg} = 13550 \text{ kg/m}^3$ and density of air $\rho_{air} = 1.2 \text{ kg/m}^3$.

