Title Here

Group 50

Section ????? April 22, 2022

Abstract—This experiment uses knowledge of the two-dimensional principal stress state to determine the internal pressure in an unopened soda can. Thin-walled pressure vessel theory is used in conjunction with Mohr's theories on principal stresses to report the internal can pressure. This result is compared to another method based upon the ideal gas law. The hoop stress in the can is $XX \pm XX$ Pa and the longitudinal stress is $XX \pm XX$ Pa. The internal pressure reported by the mechanics of materials approach is $XX \pm XX$ kPa. The internal pressure reported by the ideal gas law method is $XX \pm XX$ kPa.

Index Terms—plane stress, principal stresses, strain gauge rosette, thin-walled pressure vessel

I. INTRODUCTION

YLINDRICAL pressure vessels are an effective way to store gasses or liquids which must be kept at certain high pressures. As such, knowing the operating and current pressures in these vessels is imperative to ensuring user safety; catastrophic failures in these vessels can sometimes be fatal. The stresses in the walls of these vessels can be used to determine the overall pressure within the vessel. This study seeks the pressure in a sealed soda can.

The can is said to be a thin-walled pressure vessel. The diameter of the can is much greater than its wall thickness, leading to the assumption that there is no stress in the direction of the thickness of the can, so the stresses in the vessel wall can be analyzed as a two-dimensional stress state, the plane stress condition.

To analyze this stress state, strain gauges can be affixed in a 0-45-90° rosette on the side of the can to record the strains on the walls. Using multiple strain gauges allows an experimenter to determine the complete strain state, including the shear strain. As the strain gauge is a transducer, the strains are recorded as a voltage difference, V_{amp} , and are calculated according to (1), where V_s is the excitation voltage for the circuit, and G_f is the gauge factor of the strain gauge, given by the manufacturer.

$$\varepsilon = \frac{4\Delta V_{amp}}{V_s G_f} \tag{1}$$

These strain gauges work in conjunction with a quarter Wheatstone bridge circuit to record meaningful voltage differences (Fig. I). The strain gauge acts as a variable resistor. The change in voltage across the bridge as the strain gauge moves and varies its resistance is the voltage used to compute strain in (1). Each member of the strain gauge rosette has its own individual Wheatstone bridge circuit fed by an excitation voltage.

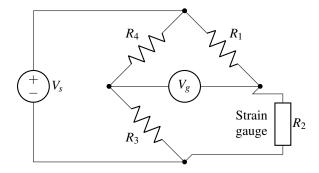


Fig. 1. A simple quarter Wheatstone bridge circuit.

In plane strain, the shear strain γ_{xy} can be computed using the 45° strain gauge value according to (2), where ε_x and ε_y are the strains from the 0° and 90° strain gauges, and θ is 45° for this rosette [CITATION NEEDED].

$$\varepsilon_{\theta} = \frac{\varepsilon_{x} + \varepsilon_{y}}{2} - \frac{\varepsilon_{x} - \varepsilon_{y}}{2} \cos(2\theta) + \gamma_{xy} \sin(2\theta)$$
 (2)

While the experimenter should attempt to align the strain gauge rosette along the principal axes, along the can's central axis and along its circumference, the true principal strain should be found according to (3), where $\varepsilon_{1,2}$ are the principal strains for the plane strain condition. This is based upon the concepts of Mohr's circle, a geometric way to identify principal stresses and strains in a material.

$$\varepsilon_{1,2} = \frac{\varepsilon_x + \varepsilon_y}{2} \pm \sqrt{\left(\frac{\varepsilon_x - \varepsilon_y}{2}\right)^2 + \left(\frac{\gamma_{xy}}{2}\right)^2} \tag{3}$$

With principal strains in hand, the principal stresses in the vessel can thus be computed. These principal stresses in the wall arise from the internal pressure pushing on the can wall, so knowing the principal stresses in the wall may lead to the pressure within the can. In 2D strain, the well-known Hooke's law relating stress to strain is given directionally as (4a) and (4b), where $\sigma_{1,2}$ are the principal stress on the wall, E is the modulus of elasticity for the can material, aluminum alloy, ν is Poisson's ratio, and $\varepsilon_{1,2}$ are the principal strains, found by using (3) with the recorded strains [1].

$$\sigma_{1} = \frac{E}{1 - v^{2}} (\varepsilon_{1} + v \varepsilon_{2}) \tag{4a}$$

$$\sigma_2 = \frac{E}{1 - v^2} \left(\varepsilon_2 + v \varepsilon_1 \right) \tag{4b}$$

For thin-walled pressure vessels, the 2D stress state is described by the longitudinal stress, along the vessel's central axis, and hoop stress, along its circumference, which can be computed with the principal strains above. The free-body diagram (Fig.

2) can be a helpful tool in relating these stresses to the can's internal pressure [1].

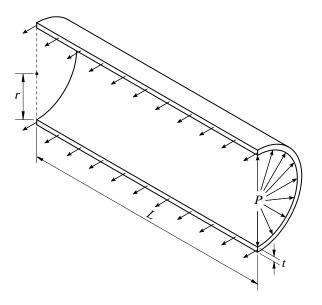


Fig. 2. Free-body diagram of the pressures acting on the pressure vessel internally.

The force balance for this diagram is (5),

$$(2rL)P = (2Lt)\sigma_H \tag{5}$$

as pressure P acts on the can inner surface area, 2rL, and is resisted by the circumferential stress σ_H acting on wall area 2Lt. The expression (5) can be solved for the hoop stress σ_H as (6).

$$\sigma_H = \frac{Pr}{t} \tag{6}$$

This hoop stress is related to its orthogonal longitudinal stress σ_L with relationship (7), known from thin-walled pressure vessel theory [1].

$$\sigma_L = \frac{1}{2}\sigma_H \tag{7}$$

These hoop and longitudinal stresses are assumed to be the principal stresses σ_1 and σ_2 . Thus the relationships from (6) and (4) can be rearranged to yield a functional relationship between principal strain and internal pressure (8) [1].

$$\varepsilon_1 = \frac{Pr}{2Et}(2 - v) \tag{8}$$

II. PROCEDURE

A. Instrumenting the Soda Can

The can must first be outfitted with the strain gauge rosette. First, the decorative wrapping on the beverage can is scrubbed off with methanol so that this wrapping does not affect the strain of the aluminum. Once the surface is properly cleaned, painter's tape is used to create an approximately 90° L-shaped guideline towards the outer left and bottom edges of the cleaned surface for strain gauge placement.

To place the strain gauge rosette, clear packing tape is placed over the top of the rosette, opposite the side that will contact the surface of the can. The tacky side of the taped strain rosette is then gently laid onto the can surface, such that the 0° and 90° strain gauges are aligned with the taped guidelines. When the placement of the strain rosette is properly aligned, part of the taped side is lifted, and glue is applied to the can. The strain gauge rosette is replaced over top of the glue and allowed to cure.

B. Wheatstone Bridge

Each strain gauge works as a variable resistor in a Wheatstone bridge, whose voltage difference is recorded for the strain. Three Wheatstone bridge configurations are wired onto a simple breadboard. Each bridge also makes use of a pre-amplifier, as the voltages passing through the circuit are expected to be very small. An excitation voltage of 3.3 V is wired so as to pass across each bridge from an external power supply. This excitation voltage V_s and the voltage difference V_{amp} for each of the three bridges is recorded with a SADI DAQ device.

C. Data Acquisition

The SADI DAQ which records the voltage difference is fed into a LabVIEW virtual instrument. The LabVIEW VI reads the V_{amp} for each bridge as V_{amp_x} for the 0° gauge, V_{amp_t} for the 45° gauge, and V_{amp_y} for the 90° gauge. These are observed with a ± 5.12 V gain window.

D. Ideal Gas Law Pressure Determination Method

A secondary method is also used based upon the ideal gas law. For this method, the temperature of the room is measured and the can is allowed to reach thermodynamic equilibrium with the room. The unopened, instrumented can is weighed on the lab scale, and its initial weight is recorded. After the can is opened, the instrumented can is weighed again to record its final weight. Next, a bottle filled with an arbitrary amount of water is weighed on the lab scale. Water is poured from the bottle into the open can until the headspace between the soda and the top of the can is filled. The bottle is weighed again.

III. RESULTS

Morbi luctus, wisi viverra faucibus pretium, nibh est placerat odio, nec commodo wisi enim eget quam. Quisque libero justo, consectetuer a, feugiat vitae, porttitor eu, libero. Suspendisse sed mauris vitae elit sollicitudin malesuada. Maecenas ultricies eros sit amet ante. Ut venenatis velit. Maecenas sed mi eget dui varius euismod. Phasellus aliquet volutpat odio. Vestibulum ante ipsum primis in faucibus orci luctus et ultrices posuere cubilia Curae; Pellentesque sit amet pede ac sem eleifend consectetuer. Nullam elementum, urna vel imperdiet sodales, elit ipsum pharetra ligula, ac pretium ante justo a nulla. Curabitur tristique arcu eu metus. Vestibulum lectus. Proin mauris. Proin eu nunc eu urna hendrerit faucibus. Aliquam auctor, pede consequat laoreet varius, eros tellus scelerisque quam, pellentesque hendrerit ipsum dolor sed augue. Nulla nec lacus.

Suspendisse vitae elit. Aliquam arcu neque, ornare in, ullamcorper quis, commodo eu, libero. Fusce sagittis erat at erat tristique mollis. Maecenas sapien libero, molestie et,

lobortis in, sodales eget, dui. Morbi ultrices rutrum lorem. Nam elementum ullamcorper leo. Morbi dui. Aliquam sagittis. Nunc placerat. Pellentesque tristique sodales est. Maecenas imperdiet lacinia velit. Cras non urna. Morbi eros pede, suscipit ac, varius vel, egestas non, eros. Praesent malesuada, diam id pretium elementum, eros sem dictum tortor, vel consectetuer odio sem sed wisi.

IV. DISCUSSION

Sed feugiat. Cum sociis natoque penatibus et magnis dis parturient montes, nascetur ridiculus mus. Ut pellentesque augue sed urna. Vestibulum diam eros, fringilla et, consectetuer eu, nonummy id, sapien. Nullam at lectus. In sagittis ultrices mauris. Curabitur malesuada erat sit amet massa. Fusce blandit. Aliquam erat volutpat. Aliquam euismod. Aenean vel lectus. Nunc imperdiet justo nec dolor.

Etiam euismod. Fusce facilisis lacinia dui. Suspendisse potenti. In mi erat, cursus id, nonummy sed, ullamcorper eget, sapien. Praesent pretium, magna in eleifend egestas, pede pede pretium lorem, quis consectetuer tortor sapien facilisis magna. Mauris quis magna varius nulla scelerisque imperdiet. Aliquam non quam. Aliquam porttitor quam a lacus. Praesent vel arcu ut tortor cursus volutpat. In vitae pede quis diam bibendum placerat. Fusce elementum convallis neque. Sed dolor orci, scelerisque ac, dapibus nec, ultricies ut, mi. Duis nec dui quis leo sagittis commodo.

Aliquam lectus. Vivamus leo. Quisque ornare tellus ullamcorper nulla. Mauris porttitor pharetra tortor. Sed fringilla justo sed mauris. Mauris tellus. Sed non leo. Nullam elementum, magna in cursus sodales, augue est scelerisque sapien, venenatis congue nulla arcu et pede. Ut suscipit enim vel sapien. Donec congue. Maecenas urna mi, suscipit in, placerat ut, vestibulum ut, massa. Fusce ultrices nulla et nisl.

Etiam ac leo a risus tristique nonummy. Donec dignissim tincidunt nulla. Vestibulum rhoncus molestie odio. Sed lobortis, justo et pretium lobortis, mauris turpis condimentum augue, nec ultricies nibh arcu pretium enim. Nunc purus neque, placerat id, imperdiet sed, pellentesque nec, nisl. Vestibulum imperdiet neque non sem accumsan laoreet. In hac habitasse platea dictumst. Etiam condimentum facilisis libero. Suspendisse in elit quis nisl aliquam dapibus. Pellentesque auctor sapien. Sed egestas sapien nec lectus. Pellentesque vel dui vel neque bibendum viverra. Aliquam porttitor nisl nec pede. Proin mattis libero vel turpis. Donec rutrum mauris et libero. Proin euismod porta felis. Nam lobortis, metus quis elementum commodo, nunc lectus elementum mauris, eget vulputate ligula tellus eu neque. Vivamus eu dolor.

Nulla in ipsum. Praesent eros nulla, congue vitae, euismod ut, commodo a, wisi. Pellentesque habitant morbi tristique senectus et netus et malesuada fames ac turpis egestas. Aenean nonummy magna non leo. Sed felis erat, ullamcorper in, dictum non, ultricies ut, lectus. Proin vel arcu a odio lobortis euismod. Vestibulum ante ipsum primis in faucibus orci luctus et ultrices posuere cubilia Curae; Proin ut est. Aliquam odio. Pellentesque massa turpis, cursus eu, euismod nec, tempor congue, nulla. Duis viverra gravida mauris. Cras tincidunt. Curabitur eros ligula, varius ut, pulvinar in, cursus faucibus, augue.

Nulla mattis luctus nulla. Duis commodo velit at leo. Aliquam vulputate magna et leo. Nam vestibulum ullamcorper leo. Vestibulum condimentum rutrum mauris. Donec id mauris. Morbi molestie justo et pede. Vivamus eget turpis sed nisl cursus tempor. Curabitur mollis sapien condimentum nunc. In wisi nisl, malesuada at, dignissim sit amet, lobortis in, odio. Aenean consequat arcu a ante. Pellentesque porta elit sit amet orci. Etiam at turpis nec elit ultricies imperdiet. Nulla facilisi. In hac habitasse platea dictumst. Suspendisse viverra aliquam risus. Nullam pede justo, molestie nonummy, scelerisque eu, facilisis vel, arcu.

Curabitur tellus magna, porttitor a, commodo a, commodo in, tortor. Donec interdum. Praesent scelerisque. Maecenas posuere sodales odio. Vivamus metus lacus, varius quis, imperdiet quis, rhoncus a, turpis. Etiam ligula arcu, elementum a, venenatis quis, sollicitudin sed, metus. Donec nunc pede, tincidunt in, venenatis vitae, faucibus vel, nibh. Pellentesque wisi. Nullam malesuada. Morbi ut tellus ut pede tincidunt porta. Lorem ipsum dolor sit amet, consectetuer adipiscing elit. Etiam congue neque id dolor.

Donec et nisl at wisi luctus bibendum. Nam interdum tellus ac libero. Sed sem justo, laoreet vitae, fringilla at, adipiscing ut, nibh. Maecenas non sem quis tortor eleifend fermentum. Etiam id tortor ac mauris porta vulputate. Integer porta neque vitae massa. Maecenas tempus libero a libero posuere dictum. Vestibulum ante ipsum primis in faucibus orci luctus et ultrices posuere cubilia Curae; Aenean quis mauris sed elit commodo placerat. Class aptent taciti sociosqu ad litora torquent per conubia nostra, per inceptos hymenaeos. Vivamus rhoncus tincidunt libero. Etiam elementum pretium justo. Vivamus est. Morbi a tellus eget pede tristique commodo. Nulla nisl. Vestibulum sed nisl eu sapien cursus rutrum.

Nulla non mauris vitae wisi posuere convallis. Sed eu nulla nec eros scelerisque pharetra. Nullam varius. Etiam dignissim elementum metus. Vestibulum faucibus, metus sit amet mattis rhoncus, sapien dui laoreet odio, nec ultricies nibh augue a enim. Fusce in ligula. Quisque at magna et nulla commodo consequat. Proin accumsan imperdiet sem. Nunc porta. Donec feugiat mi at justo. Phasellus facilisis ipsum quis ante. In ac elit eget ipsum pharetra faucibus. Maecenas viverra nulla in massa

Nulla ac nisl. Nullam urna nulla, ullamcorper in, interdum sit amet, gravida ut, risus. Aenean ac enim. In luctus. Phasellus eu quam vitae turpis viverra pellentesque. Duis feugiat felis ut enim. Phasellus pharetra, sem id porttitor sodales, magna nunc aliquet nibh, nec blandit nisl mauris at pede. Suspendisse risus risus, lobortis eget, semper at, imperdiet sit amet, quam. Quisque scelerisque dapibus nibh. Nam enim. Lorem ipsum dolor sit amet, consectetuer adipiscing elit. Nunc ut metus. Ut metus justo, auctor at, ultrices eu, sagittis ut, purus. Aliquam aliquam.

V. CONCLUSION

Etiam pede massa, dapibus vitae, rhoncus in, placerat posuere, odio. Vestibulum luctus commodo lacus. Morbi lacus dui, tempor sed, euismod eget, condimentum at, tortor. Phasellus aliquet odio ac lacus tempor faucibus. Praesent sed sem.

Praesent iaculis. Cras rhoncus tellus sed justo ullamcorper sagittis. Donec quis orci. Sed ut tortor quis tellus euismod tincidunt. Suspendisse congue nisl eu elit. Aliquam tortor diam, tempus id, tristique eget, sodales vel, nulla. Praesent tellus mi, condimentum sed, viverra at, consectetuer quis, lectus. In auctor vehicula orci. Sed pede sapien, euismod in, suscipit in, pharetra placerat, metus. Vivamus commodo dui non odio. Donec et felis.

APPENDIX A: UNCERTAINTY CALCULATION

The uncertainty of each variable is given by Table I. The uncertainties in V_{amp_x} , V_{amp_t} , V_{amp_y} , and V_s were calculated by finding the standard deviation of a large sample set of these measurements and approximating a 95% confidence interval. The uncertainties in d and t were determined from the measuring device used, which was a micrometer. The uncertainties in the weights of the can and the water were determined from the scale that was used and analyzing the sensitivity of the last digit. The uncertainties in E, V, G, R, M, and ρ were estimated to be $\pm 1\%$ of the given value due to variability associated with these material properties [????].

TABLE I UNCERTAINTY IN EACH VARIABLE

Variable (Symbol)	Value	Uncertainty
Initial Amplified Voltage $x(V_{amp_{xi}})$	-0.1389 V	±0.00056 V
Initial Amplified Voltage t ($V_{amp_{ti}}$)	0.0673 V	$\pm 0.00039 \text{ V}$
Initial Amplified Voltage $y(V_{amp_{yi}})$	-0.1064 V	$\pm 0.00052 \text{ V}$
Final Amplified Voltage $x(V_{amp_{xf}})$	−0.237 V	$\pm 0.0011 \text{ V}$
Final Amplified Voltage t $(V_{amp_{tf}})$	−0.0253 V	$\pm 0.00027 \text{ V}$
Final Amplified Voltage $y(V_{amp_{yf}})$	-0.1398 V	$\pm 0.00055 \text{ V}$
Excitation Voltage (V_s)	3.348 V	$\pm 0.0015 \text{ V}$
Can Diameter (d)	65.9 mm	$\pm 0.896~\text{mm}$
Can Thickness (t)	0.09 mm	$\pm 0.016~\text{mm}$
Modulus of Elasticity (E) [3]	69 GPa	$\pm 0.69~\mathrm{GPa}$
Poisson's Ratio (v) [3]	0.35	± 0.0035
Shear Modulus (G) [3]	25 GPa	$\pm 0.25~\mathrm{GPa}$
Gauge Factor (G_f)	2.09	± 0.01045
Gain (A)	64	± 0.64
Initial Can Weight (W_{ci})	406.8 g	± 0.1 g
Final Can Weight (W_{cf})	406.2 g	± 0.1 g
Initial Water Weight (W_{wi})	152.01 g	$\pm 0.01~\mathrm{g}$
Final Water Weight (W_{wf})	141.35 g	$\pm 0.01~\mathrm{g}$
Temperature (T)	22.8°C	$\pm 0.28^{\circ}\mathrm{C}$
Gas Constant (R) [???]	$8.31 \text{ J/(K} \cdot \text{mol)}$	$\pm 0.083~J/(K\cdot mol)$
CO_2 Molar Mass (M) [4]	44 g/mol	± 0.44 g/mol
Density (ρ) [???]	0.998 g/cm^3	$\pm 0.00998~{\rm g/cm^3}$

Uncertainty in ΔV_{amp}

The change in V_{amp} for each strain gauge can be calculated using (9), where $V_{amp_{xi}}$, $V_{amp_{ti}}$, and $V_{amp_{yi}}$ are the initial voltages and $V_{amp_{xf}}$, $V_{amp_{tf}}$, and $V_{amp_{yf}}$ are the final voltages.

$$\Delta V_{amp_x} = V_{amp_{xf}} - V_{amp_{xi}} \tag{9a}$$

$$\Delta V_{amp_t} = V_{amp_{tf}} - V_{amp_{ti}} \tag{9b}$$

$$\Delta V_{amp_{y}} = V_{amp_{yf}} - V_{amp_{yi}} \tag{9c}$$

The uncertainty in each ΔV_{amp} can be calculated using the root sum square (RSS) method and (9), where U is the uncertainty.

$$U_{\Delta V_{amp}} = \left[U_{V_{amp_i}}^2 \left(\frac{\partial (\Delta V_{amp})}{\partial V_{amp_i}} \right)^2 + U_{V_{amp_f}}^2 \left(\frac{\partial (\Delta V_{amp_i})}{\partial V_{amp_f}} \right)^2 \right]^{\frac{1}{2}}$$
(10)

For V_{amp_x} , using the values given in Table I give an uncertainty in ΔV_{amp_x} of ± 0.27 V.

Uncertainty in Strain

Strain is given by (1). The uncertainty in strain can be calculated by using (11).

$$U_{\varepsilon} = \left[U_{\Delta V_{amp}}^2 \left(\frac{\partial \varepsilon}{\partial (\Delta V_{amp})} \right)^2 + U_A^2 \left(\frac{\partial \varepsilon}{\partial A} \right)^2 + U_{V_s}^2 \left(\frac{\partial \varepsilon}{\partial V_s} \right)^2 + U_{G_f}^2 \left(\frac{\partial \varepsilon}{\partial G_f} \right)^2 \right]^{\frac{1}{2}}$$
(11)

For V_{amp_x} , using Table ???? and the values given in Table I give an uncertainty in strain ε_x of ± 0.0025 mm/mm.

Uncertainty in Shear Strain

Shear strain is given by (???). The uncertainty in shear strain can be calculated by using (12).

$$U_{\chi_{y}} = \left[U_{\varepsilon_{x}}^{2} \left(\frac{\partial \chi_{y}}{\partial \varepsilon_{x}} \right)^{2} + U_{\varepsilon_{t}}^{2} \left(\frac{\partial \chi_{y}}{\partial \varepsilon_{t}} \right)^{2} + U_{\varepsilon_{y}}^{2} \left(\frac{\partial \chi_{y}}{\partial \varepsilon_{y}} \right)^{2} \right]^{\frac{1}{2}}$$
(12)

Using Table ???, the uncertainty in shear strain is ± 0.0015 mm/mm.

Uncertainty in Principle Strain

Principle strain is given by (3). The uncertainty in principle strain can be calculated by using (13).

$$U_{\varepsilon_{1,2}} = \left[U_{\varepsilon_x}^2 \left(\frac{\partial \varepsilon_{1,2}}{\partial \varepsilon_x} \right)^2 + U_{\varepsilon_y}^2 \left(\frac{\partial \varepsilon_{1,2}}{\partial \varepsilon_y} \right)^2 + U_{\chi_y}^2 \left(\frac{\partial \varepsilon_{1,2}}{\partial \chi_y} \right)^2 \right]^{\frac{1}{2}}$$
(13)

For ε_1 , using Table ??? gives an uncertainty in principle strain ε_1 of ± 0.0024 mm/mm.

Uncertainty in Principle Stress

Principle stress is given by (4). The uncertainty in principle stress can be calculated by using (14).

$$U_{\sigma_{1,2}} = \left[U_E^2 \left(\frac{\partial \sigma_{1,2}}{\partial E} \right)^2 + U_V^2 \left(\frac{\partial \sigma_{1,2}}{\partial V} \right)^2 + U_{\varepsilon_1}^2 \left(\frac{\partial \sigma_{1,2}}{\partial \varepsilon_1} \right)^2 + U_{\varepsilon_2}^2 \left(\frac{\partial \sigma_{1,2}}{\partial \varepsilon_2} \right)^2 \right]^{\frac{1}{2}}$$
 (14)

For σ_1 , using Table ???? and the values given in Table I gives an uncertainty in principle stress σ_1 of ± 0.19 GPa.

Uncertainty in Pressure from Strain

Pressure from strain is given by (???). The uncertainty in pressure can be calculated by using (15).

$$U??? \tag{15}$$

Using Table ??? and the values given in Table I gives an uncertainty in pressure from strain of \pm ???kPa.

REFERENCES

- [1] G. Subhash and S. Ridgeway, "Thin-walled Pressure Vessels," in *Mechanics of Materials Laboratory Course*. San Rafael, CA: Morgan & Claypool, 2018, ch. 3, pp. 113–132.
- [2] D. R. Askeland, F. Haddleton, P. Green, and H. Robertson, "Nonferrous Alloys," in *The Science and Engineering of Materials*, 3rd ed. Berlin, Germany: Springer, 1996, ch. 13, pp. 401–436.
- [3] J. W. Bray, "Aluminum Mill and Engineered Wrought Products," in *Properties and Selection: Nonferrous Alloys and Special-Purpose Materials* (ASM Handbook, Vol. 2). Russell Township, OH: ASM International, 1990.
- [4] Engineering ToolBox (2018). Carbon Dioxide Thermophysical Properties. [Online] Available: https://www.engineeringtoolbox.com/ CO2-carbon-dioxide-properties-d_2017.html