

**Problem 1:**

(a)  $\frac{dy}{dx} = e^x e^{-2y} \implies \int e^{2y} dy = \int e^x dx \implies \frac{1}{2} e^{2y} = e^x + c' \implies e^{2y} = 2e^x + c \implies y = \frac{1}{2} \ln(2e^x + c).$

(b)  $\frac{dy}{dx} - \frac{1}{x}y = x \sin(x).$  Integration factor:  $e^{-\int \frac{1}{x} dx} = e^{-\ln x} = \frac{1}{x}$   
 $\frac{1}{x} \frac{dy}{dx} - \frac{1}{x^2}y = \sin(x) \implies \frac{d}{dx} \left[ \frac{y}{x} \right] = \sin(x) \implies \frac{y}{x} = -\cos(x) + c \implies y = x(c - \cos(x)).$

(c)  $\frac{dy}{dx} + \frac{1}{x}y = \frac{1}{x^2}.$  Integration factor:  $e^{\int \frac{1}{x} dx} = e^{\ln x} = x$   
 $x \frac{dy}{dx} + y = \frac{1}{x} \implies \frac{d}{dx} [xy] = \frac{1}{x} \implies xy = \ln|x| + c \implies y = \frac{1}{x}(\ln|x| + c).$

**Problem 2:**

(a)  $\frac{dN}{dt} = -\lambda N \implies N(t) = N_0 e^{-\lambda t}.$  We also know that  $N_0 = N(0) = 100$  and  $N(6) = 0.97 \times N_0 = 97$ ,  
 i.e.  $97 = 100e^{-\lambda(6)},$  so that  $\lambda = 0.0050765.$  Therefore,  $N(24) = 100e^{-\lambda(24)} = 88.53$  mg.

(b) The half-life is the time  $t$  for which  $100e^{-\lambda t} = 50.$  Hence  $t = \frac{1}{\lambda} \ln(100/50) = 136.54$  hours.

**Problem 3:**

$\frac{dT}{dt} + kT = kT_m.$  Integration factor:  $e^{\int k dt} = e^{kt} \implies \frac{d}{dt} [e^{kt}T] = ke^{kt}T_m \implies e^{kt}T = e^{kt}T_m + c$

$T(0) = T_0 : T_0 = T_m + c \implies c = T_0 - T_m. \implies e^{kt}T = e^{kt}T_m + T_0 - T_m \implies T(t) = T_m + (T_0 - T_m)e^{-kt}.$

**Problem 4:**

Let  $T = T(t)$  represent the core temperature, where  $t$  is measured in hours from the time of death. According to Newton's law, with  $T_0 = 37$  and  $T_m = 21$ , we have  $T(t) = 21 + 16e^{-kt}.$  Let  $t = t^*$  be the time 16:00.

We are given that  $T(t^*) = 29$  and  $T(t^* + 1) = 26.5$ , hence  $29 - 21 = 16e^{-kt^*}$  and  $26.5 - 21 = 16e^{-k(t^*+1)}.$

Divide the second equation by the first, so that  $\frac{26.5-21}{29-21} = e^{-k},$  that is  $k = \ln(8/5.5) = 0.374693.$

Now we can determine  $t^*$  from  $29 - 21 = 16e^{-kt^*} \implies t^* = \frac{1}{k} \ln(16/8) = 1.849905.$

The time of death ( $t = 0$ ) is therefore 1.849905 hours before 16:00, which is approximately 14:09.

**Problem 5:**

Let  $T = T(t)$  represent the temperature of the thermometer.

We first consider  $t$  as being the time elapsed since the first minute (when the temperature was 55). Then  $T_0 = 55$ ,  $T_m = 5$ , and we have  $T(5 - 1 = 4) = 5 + (55 - 5)e^{-4k} = 30 \implies k = 0.25 \ln(2) \approx 0.173.$

We next consider  $t$  as being the time elapsed since we left the room, and we seek  $T_0.$  Now we have  $T(1) = 5 + (T_0 - 5)e^{-k} = 55.$  Substituting the value of  $k$  we found above and solving, we have  $T_0 = 50e^{\frac{1}{4} \ln 2} + 5 \approx 64.46.$

**Problem 6:**

While the bar is in  $A$  we have:  $dT_1/dt = k_1(0 - T_1) = -k_1T_1,$  with  $T_1(0) = 100 \implies T_1(t) = 100e^{-k_1t}.$

After 1 min the bar's temperature is  $90^\circ\text{C},$  hence  $90 = 100e^{-k_1}$  so that  $k_1 = \ln(100/90) = 0.105361.$

After 2 min the temperature is  $T(2) = 100e^{-2k_1} = 81.$  We now instantly transfer it to  $B.$

The temperature of the bar is now described by the DE:  $dT_2/dt = k_2(100 - T_2)$  with  $T_2(0) = 81.$

Note: here we measure time from the moment of transfer. We therefore get  $T_2(t) = 100 + (81 - 100)e^{-k_2 t}$ . After 1 min the temperature increases by  $10^\circ$ , hence  $91 = 100 - 19e^{-k_2}$  so that  $k_2 = \ln(19/9) = 0.747214$ .

How long does it take to reach  $99.9^\circ\text{C}$ ? We seek the time  $t$  for which  $99.9 = 100 - 19e^{-k_2 t}$ , therefore  $t = 7.02$  min after transfer, i.e. 9.02 min from the start of the entire process.

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**Problem 7:**

We now have  $\frac{dm}{dt} = 0.1F - \frac{m}{1500}F$ ,  $m(0) = 450$ . Solving again by integrating factor (omitted) we have  $m(t) = 150 + 300e^{-tF/1500}$ . We require that  $m(60) = 0.2 \times 1500 = 300 = 150 + 300e^{-60F/1500}$  and solving for  $F$  we find  $F = \frac{1500}{60} \ln(2) \approx 17.3 \text{ l/min}$ .