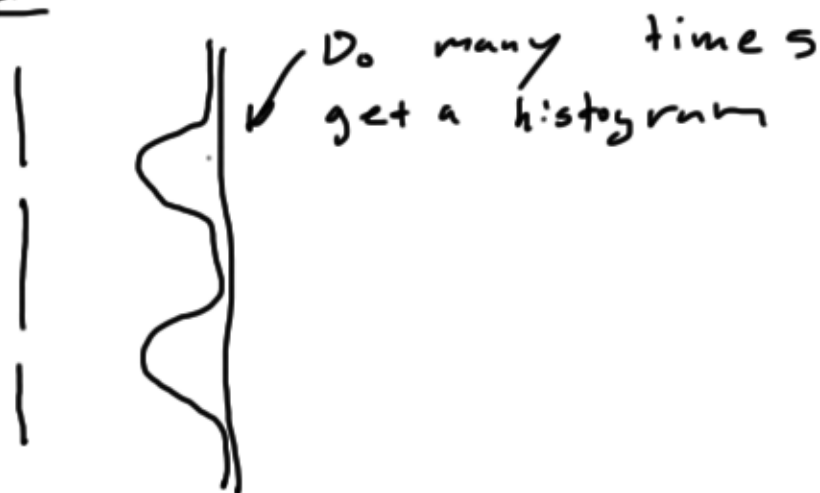


Wave-Particle Duality

What happens if you send an e^- through the double slit experiment.

→ for large slits e^- behaves like a ball)

Large slit



But keep making the experiment smaller and you will get

Small slits



Sometimes e^- behaves as ball

ex → we only find a amount of charge and only measure one electron at the end of each run.

Other times e^- behaves as wave
ex. interference pattern implies e^- is
traveling as a wave.

This is wave particle duality.

1923 Louis de Broglie proposed
matter exhibits wave behavior

Proposed: $\lambda = \frac{h}{p}$ $p = \frac{mv}{\sqrt{1 - \frac{v^2}{c^2}}}$

↑
as p gets large
 λ gets small.

Could we do double slit experiment with
a whole atom? Whole molecule? bacteria? cat!!?
Person!!?

the larger (more massive) the object
the larger p . large $p \rightarrow$ small λ .

ex

$\lambda_{\text{baseball?}}$ for $v_{\text{baseball}} = 13 \text{ m/s}$

$$\lambda \approx 3 \times 10^{-34} \text{ m}$$

↑
 $10^{-35} \rightarrow$ plank length

would not see wave nature of
baseball.

What kind of waves are matter waves?

See Richard Feynman vid

truth is its not clear

→ Probably best to think of them as probability waves.

If probability waves does that mean we just have an incomplete theory that is governed by something deeper?

debatable but one thing is clear

→ Quantum effects are quite different than classical physics.

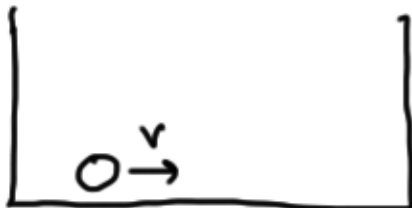
We often describe a quantum state in terms of this wave → called a wave function

ψ Ψ

Schrödinger & Heisenberg

developed theory to determine wave function
→ Quantum mechanics.

Let's put a ball in a box



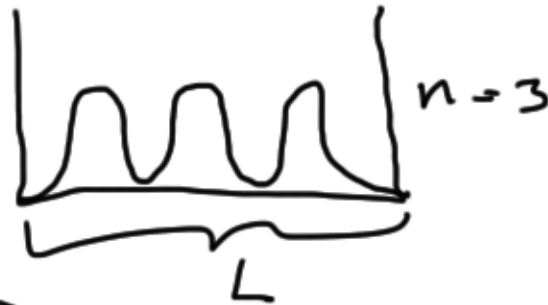
say the ball is moving fast
so its position is random
if the time scale of the measurements is long compared to ball's vel.

Probability dist. (Classical)



no more likely to be
found in one place than another

Quantum



$$P_{\text{quant}}(x) = |\psi(x)|^2 = \frac{2}{L} \sin^2\left(\frac{n\pi x}{L}\right)$$

↑
You can not ever know for sure
where you will measure the particle to
be you can only know the probability
of finding it at a given location,

Place a measuring device in double slit

experiment



looks
to see
if electron
passes



Changes
interference
pattern

In pop culture this is an argument for the importance of the conscious observer but lets think about whats really happening.

How do we see $e^- \rightarrow$ light

→ large λ does not disturb e^- as much
but less likely to find e^-

→ only slightly changes interference pattern.

→ small $\lambda \rightarrow$ greatly disturbs e^-

→ more likely finds e^-

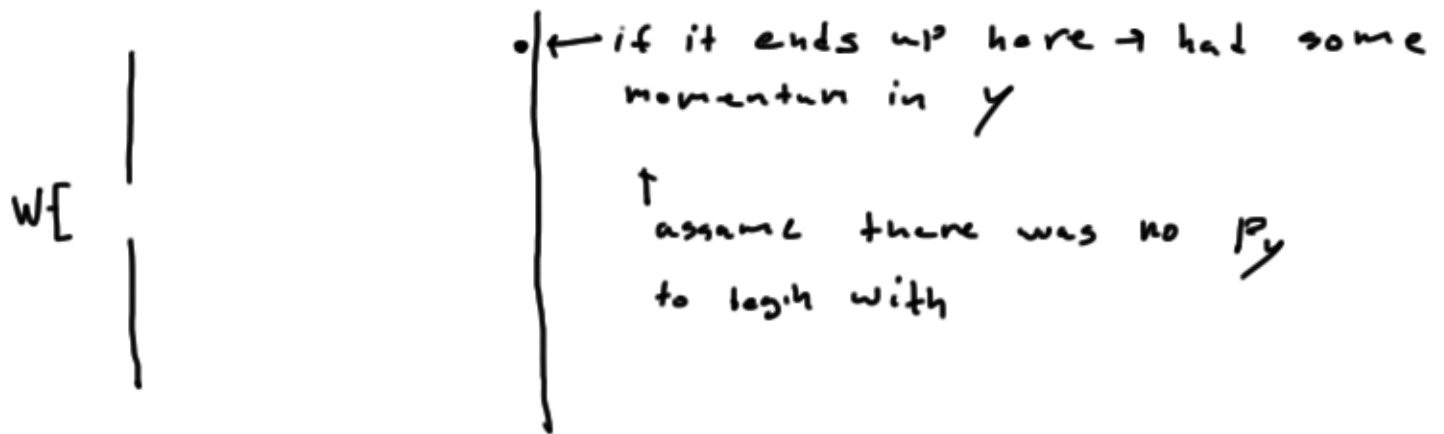
→ drastically changes interference pattern.

Conscious observer not important

→ definition of observation is.

The Heisenberg uncertainty Principle

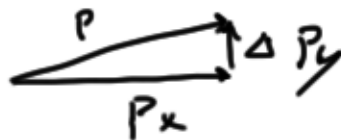
Take single slit



$$\sin \theta = \frac{\lambda}{W} \leftarrow \text{first dark fringe}$$

↑
for θ small $\sin \theta \approx \tan \theta$

$$\tan \theta = \frac{\Delta p_y}{p_x}$$



so $\frac{\Delta p_y}{p_x} \approx \frac{\lambda}{W}$

also know $\lambda = \frac{h}{p}$ so $p_x = \frac{h}{\lambda}$

$$\frac{\Delta p_y}{h/\lambda} = \frac{\lambda}{W} \rightarrow$$

$$\boxed{\Delta p_y = \frac{h}{W}}$$

Also uncertainty in position

$$\Delta y = \frac{W}{2} \quad [W = 2 \Delta y] \leftarrow \text{plug in}$$

$$\Delta p_y = \frac{h}{2 \Delta y} \rightarrow \boxed{\Delta p_y \Delta y = \frac{h}{2}} \leftarrow \text{roughly}$$

Gives us the idea. Actual expression:

$$\boxed{\Delta x \Delta p \geq \frac{\hbar}{2} \quad \hbar = \frac{h}{2\pi}}$$

↗

Heisenberg Uncertainty Principle

So what if we take our e^- in a box but make the box small?



probability dist for $n=0$

↖ we know the position pretty well but

$$\Delta x \Delta p_x \geq \frac{\hbar}{2}$$

↗
if this
is small

↖ then this must be
pretty large.

Where do we find this in nature?

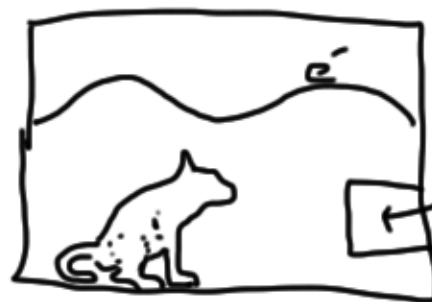
white dwarf stars

→ not enough matter to become black hole
since all matter that's crunched up
has large momentum.

White dwarf

- mass of sun size of earth,
- burned all fusible material
- will gradually cool until $\Delta x \Delta p \geq \frac{\hbar}{2}$
- is supplying pressure that keeps it from collapsing.

Schrödinger's Cat



Completely closed system

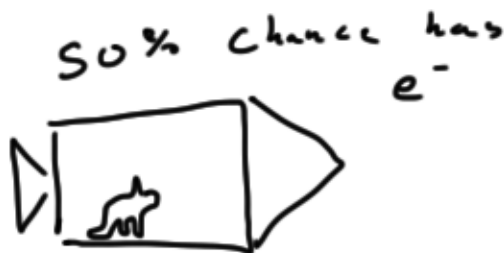
deadly device

↑
if it detects electron
is the cat alive or dead?

Superposition of states

(In reality → no) ← possible in quantum world.

2 cats + device the box
send them far away



← each in superposition

Entanglement

→ look in box 1 → find alive or dead
→ immediately determines fate of other cat. → faster than light? yes but cannot share info faster than light

The Hydrogen atom

$$E_H = \frac{-E_0}{n^2} \quad n = 1, 2, 3, \dots$$

$n \equiv$ Principal quantum #

$l \equiv$ orbital quantum #

$m \equiv$ magnetic quantum #

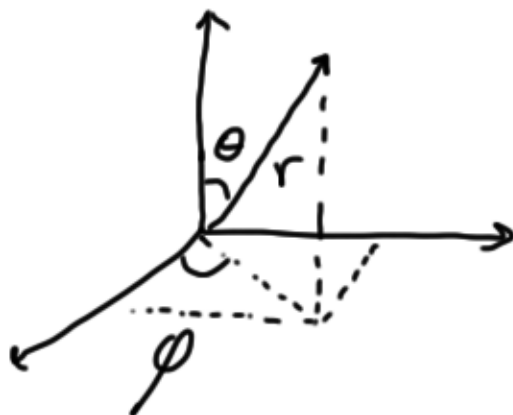
$s \equiv$ spin quantum #.

$$\Psi_H = R(r) \Theta(\theta) \Phi(\phi)$$

↑
radial
 n

↑
Polar
 l

↑
azimuthal
 m



$$l = 0, 1, \dots, n-1$$

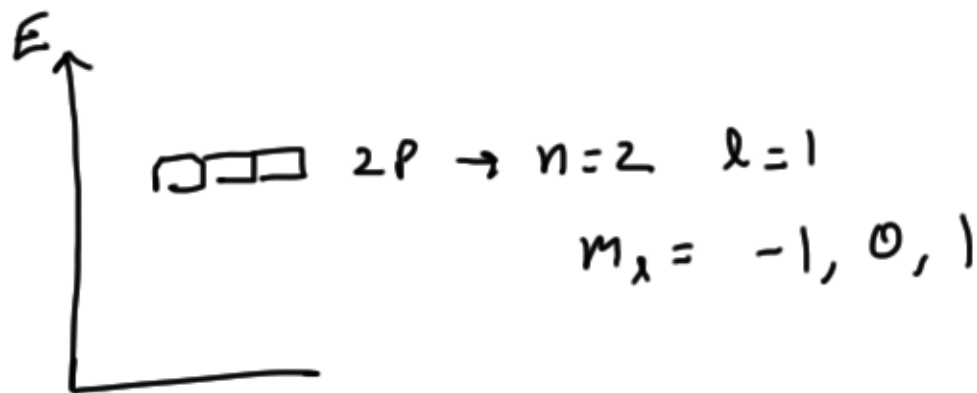
$$m = -l, -l+1, \dots, l-1, l$$

$$s = \pm \frac{1}{2}$$

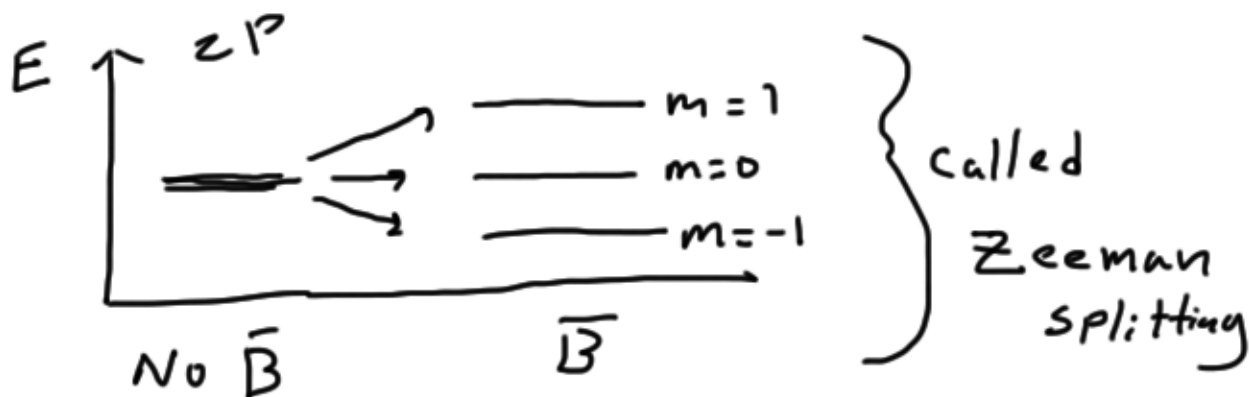
$m \rightarrow$ only have dif.
energy in B field

all play
a role
in wavefunction

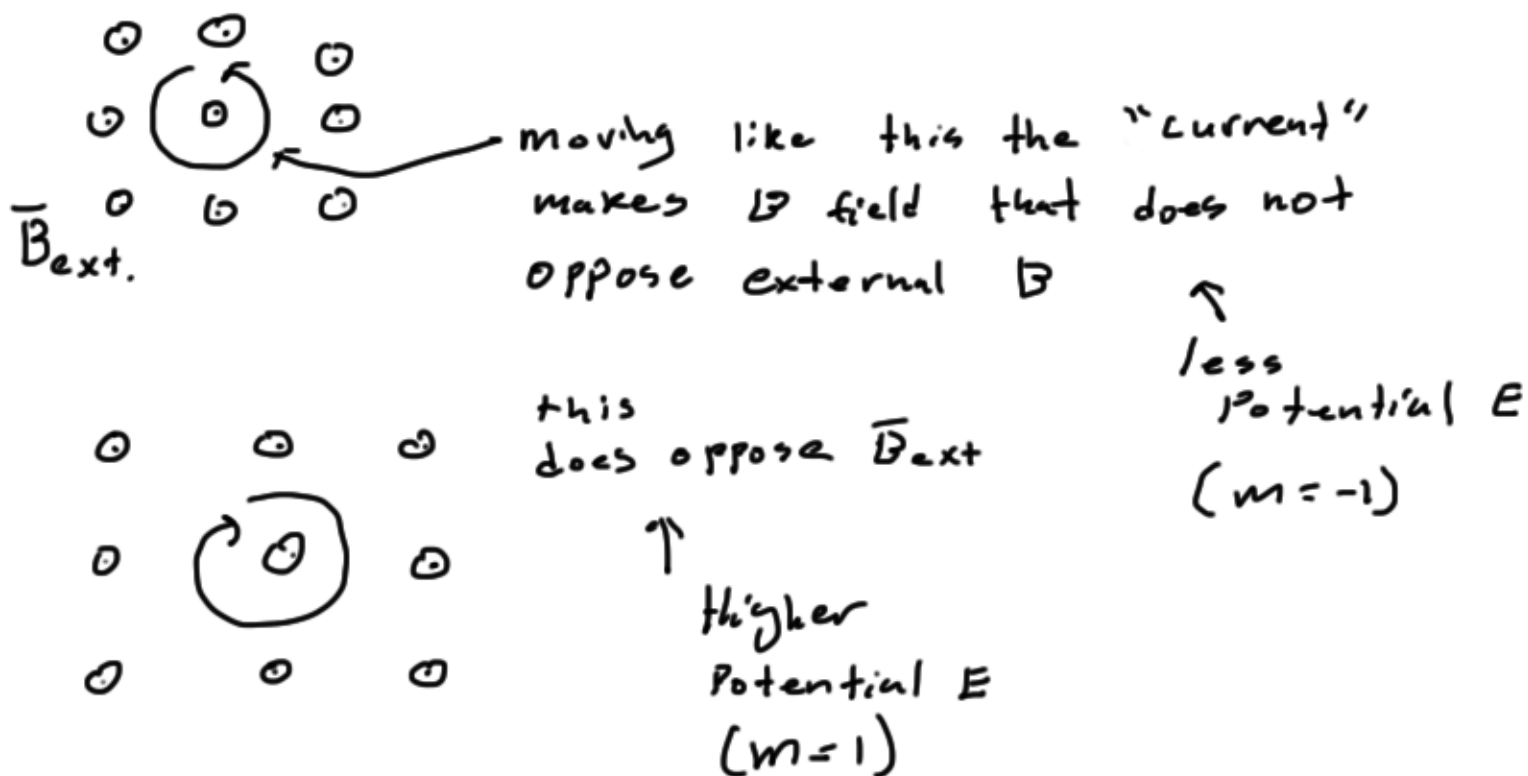




Add B field



Dif. levels correspond to (classical P/L)



Some Review

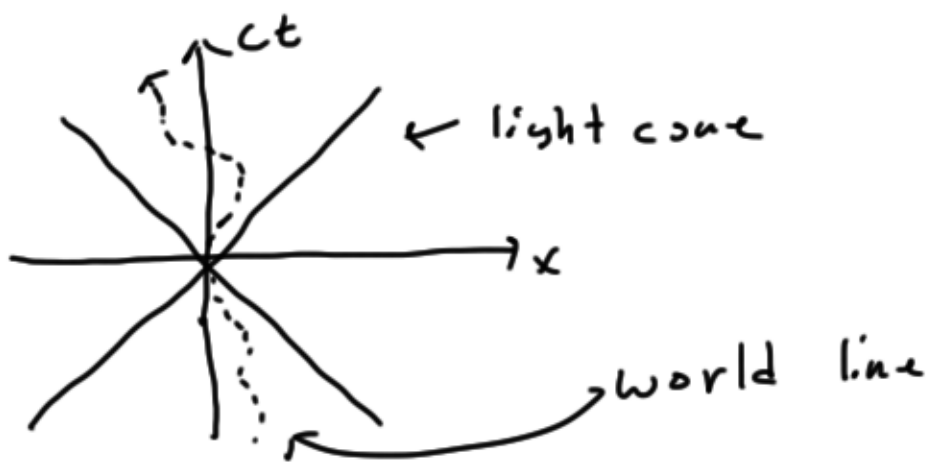
Postulates of relativity

- 1) c always const
- 2) laws of phys. same in all inertial ref frames.

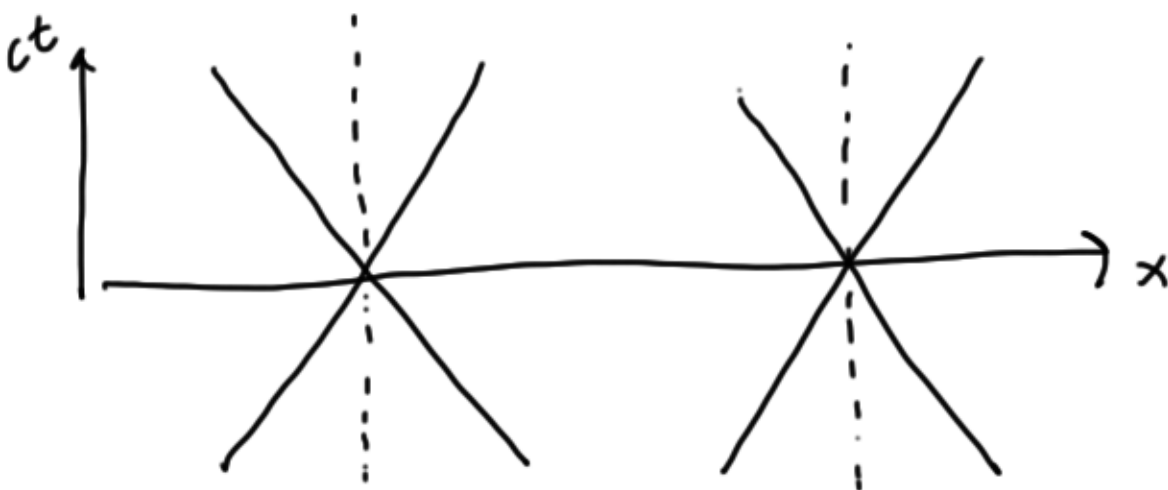
inertial ref. \rightarrow not accel

this means no faster than light communication

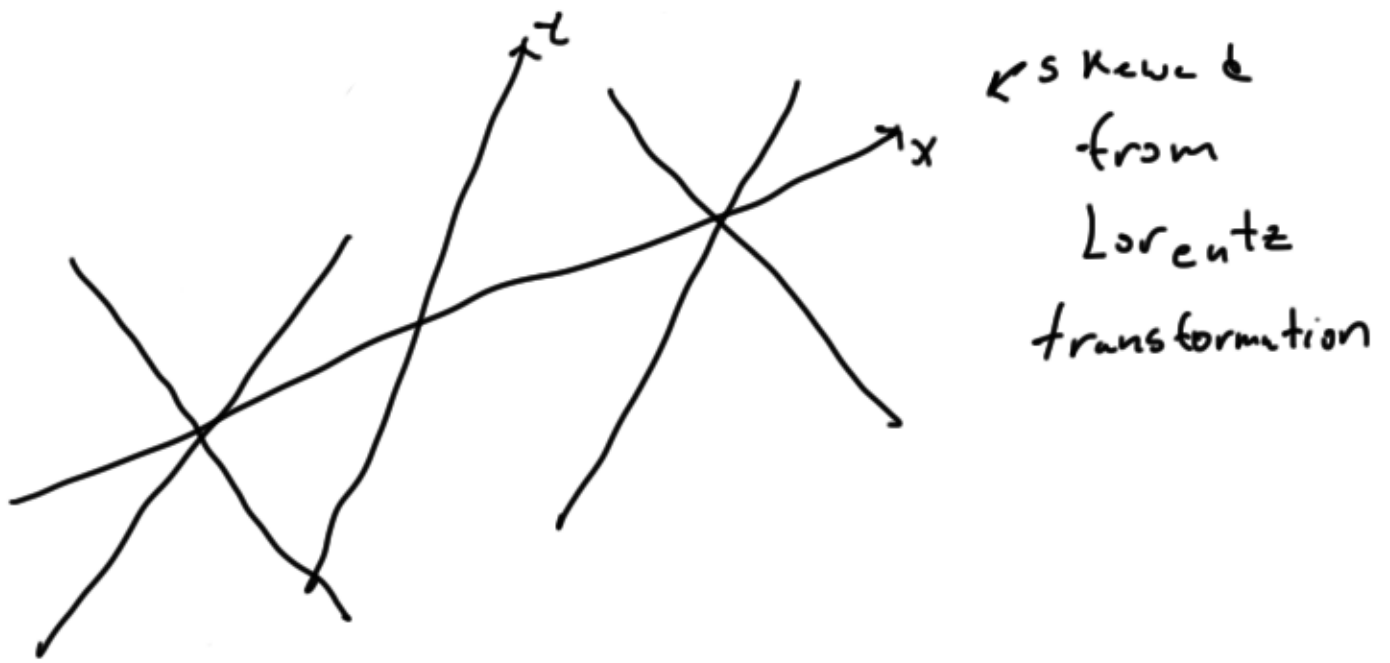
\uparrow why is this good? \rightarrow breaking speed of light breaks causality... why?



2 people (not moving fast)

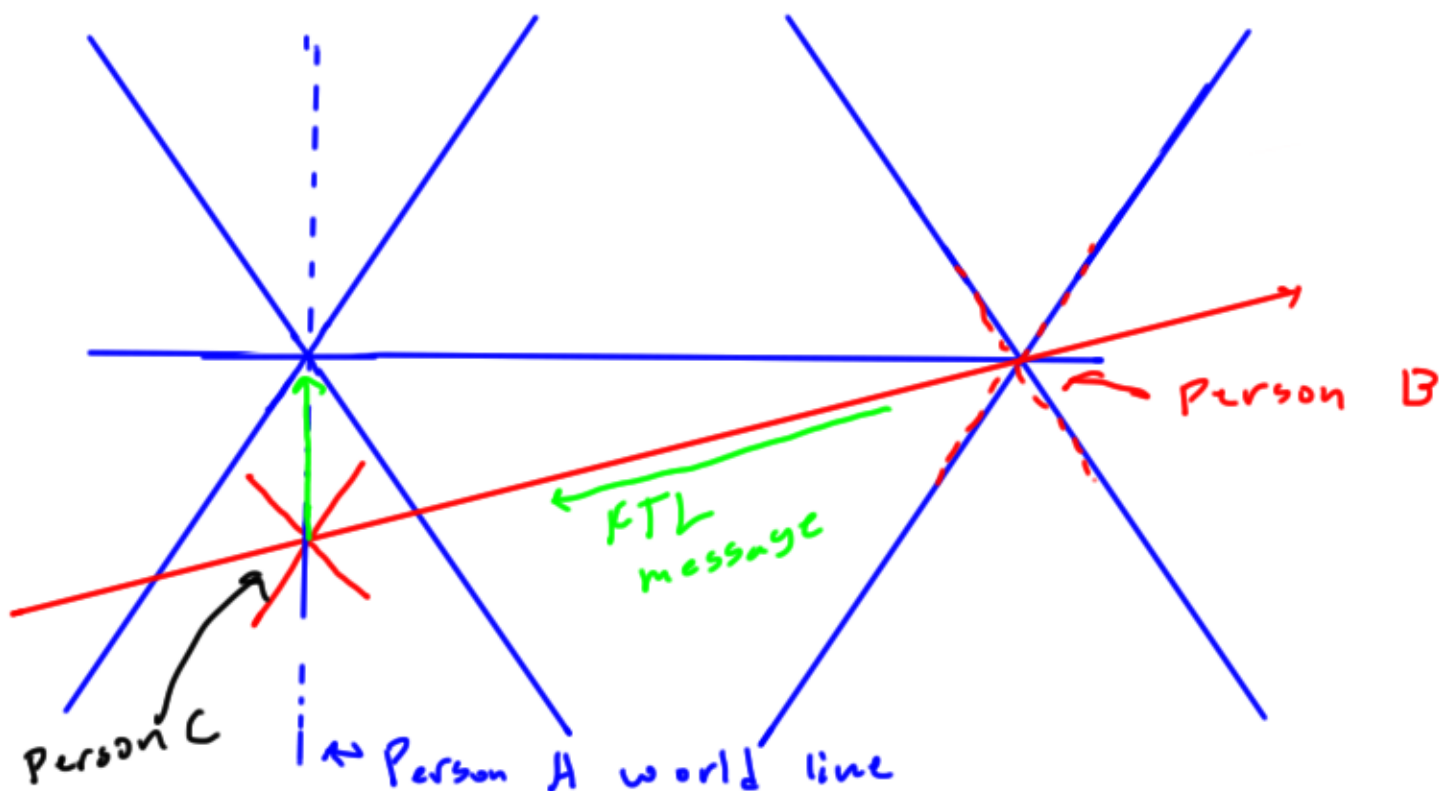


2 People moving fast



↑
if they can exchange info faster than light what happens?

↑
Let's say they do this right as they pass the first 2 ppl.



Person A can receive message from Person C because they are (momentarily) in the same place.

If A and B have FTL communication
A can send a message to B who sends it to C who sends it to A.

↑

Thus A has sent a message back in time to her/himself!