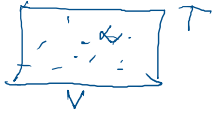


$$P = \frac{2}{3} \frac{N}{V} \left\langle \frac{1}{2} m v^2 \right\rangle$$

$$PV = N k_B T$$



$$\left\langle \frac{1}{2} m v^2 \right\rangle = \frac{3}{2} k_B T$$

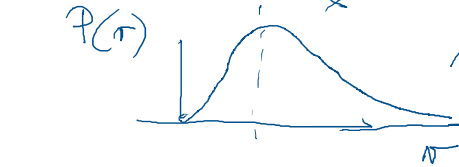
kinetička
energija
molekuly

$$\langle E_{\text{molekuly}} \rangle_{\text{IP}} = \frac{df}{2} k_B T$$

$$P(r) \sim e^{-\frac{m v^2}{2 k_B T}}$$

Maxwellovo
rozdelenie

$$v^2 = v_x^2 + v_y^2 + v_z^2$$



$$\int_{-\infty}^{\infty} e^{-x^2} dx$$



$$u = x^2$$

$$x = \sqrt{u}$$

$$dx = \frac{1}{2} \frac{1}{\sqrt{u}} du$$

$$\frac{1}{2} \int \frac{1}{\sqrt{u}} e^{-u} du$$

$$I = \int_{-\infty}^{\infty} e^{-x^2} dx$$

$$I^2 = \int_{-\infty}^{\infty} e^{-x^2} dx \cdot \int_{-\infty}^{\infty} e^{-y^2} dy$$

$$= \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} dx dy e^{-(x^2 + y^2)}$$

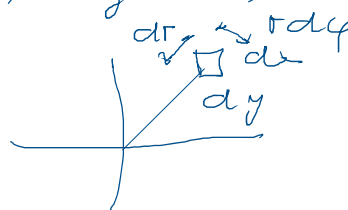
$$r^2 = x^2 + y^2$$

$$r^2 = x^2 + y^2$$

$$\varphi = \arctan \frac{y}{x}$$

$$dx dy \Rightarrow r dr d\varphi$$

$$\Rightarrow \int_0^{\infty} r dr \int_0^{2\pi} d\varphi e^{-r^2} = 2\pi \int_0^{\infty} r e^{-r^2} dr$$



$$u = r^2$$

$$du = 2r dr$$

$$= \pi \int_0^{\infty} e^{-u} du$$

$$= \pi \left[-e^{-u} \right]_0^{\infty} =$$

$$= \pi (0 - (-1)) = \pi$$

$$I^2 = \pi$$

$$I = \sqrt{\pi}$$



"Profil"

$$\int_{-\infty}^{\infty} e^{-x^2} dx = \int_{-\infty}^0 e^{-x^2} dx + \int_0^{\infty} e^{-x^2} dx$$

$$u = x^2$$

$$x = \sqrt{u}$$

$$dx = \frac{1}{2} \frac{1}{\sqrt{u}} du$$

$$= 2 \int_0^{\infty} e^{-x^2} dx$$

$$I = 2 \frac{1}{2} \int_0^{\infty} u^{-\frac{1}{2}} e^{-u} du = \Gamma\left(\frac{1}{2}\right) = \sqrt{\pi}$$

$$\Gamma(z) = \int_0^{\infty} x^{z-1} e^{-x} dx$$

$$\Gamma(z) = (z-1)! \quad \text{pre } z \in \mathbb{N} \quad \Gamma(z+1) = z \Gamma(z)$$

$$\Gamma\left(\frac{1}{2}\right) = \sqrt{\pi}$$

$$\Gamma\left(\frac{3}{2}\right) = \frac{1}{2} \Gamma\left(\frac{1}{2}\right) = \frac{\sqrt{\pi}}{2}$$

$$\left\langle \frac{1}{2} m r^2 \right\rangle = \frac{3}{2} kT$$

$$\langle E_{\text{mol}} \rangle = \frac{d \cdot f}{2} kT$$

↓

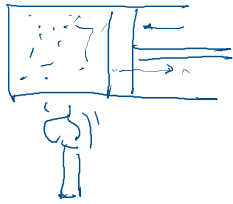
$$\frac{E}{N}$$

→ celková energia
molekul plynu
vnitřní energie

$$dQ = dU + dW$$

(m. teplo,) (zmena vn.) (praca,)

$$dQ = \left(\begin{array}{l} \text{m.k. tepla} \\ \text{přít. do} \\ \text{systemu} \end{array} \right) + \left(\begin{array}{l} \text{změna vn.} \\ \text{energie} \end{array} \right) + \left(\begin{array}{l} \text{práce,} \\ \text{vykon.} \\ \text{teleso} \end{array} \right)$$



Nemáme práci :

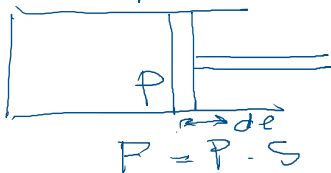
$$dQ = dU$$

$$dQ = \frac{3}{2} N k_B dT$$

$$\frac{dQ}{dT} = \frac{3}{2} N k_B = C_V \quad \text{tepelná kapacita}$$

$$C_V = \left(\frac{\partial U}{\partial T} \right)_V$$

Máme práci :



$$dW = P \cdot \underbrace{S \cdot dl}_{dV} = P dV$$

$$dQ = dU + P dV$$

$$= \frac{3}{2} N k_B dT + N k_B dT = \frac{5}{2} N k_B dT$$

$$C_P = \left(\frac{\partial U}{\partial T} \right)_P = \frac{5}{2} N k_B = C_V + N k_B$$

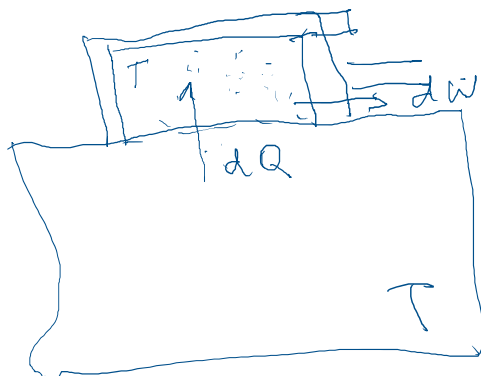
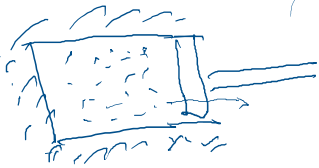
1 mol :

$$C_P = C_V + N_A k_B = C_V + R$$

$$C_X = \left(\frac{\partial U}{\partial T} \right)_X$$

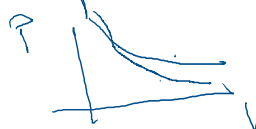
A diabatický systém
= izolovaný

A diabaticky system
= izoborany



$$P = \frac{N}{V} k_B T$$

$P \cdot V = \text{konst}$
izotermický
proces



$$dQ = 0$$

$$0 = dU + P dV$$

$$0 = C_V dT + \frac{N}{V} k_B T dV$$

$$0 = C_V \frac{dT}{T} + N k_B \frac{dV}{V}$$

$$0 = \frac{dT}{T} + \frac{N k_B}{C_V} \frac{dV}{V}$$

$$0 = d \left(\ln T + \frac{N k_B}{C_V} \ln V \right)$$

$$\ln T + \frac{N k_B}{C_V} \ln V = K$$

$$T V^{\frac{N k_B}{C_V}} = K'$$

$$T V^{\gamma-1} = K$$

$$N k_B = C_P - C_V$$

$$\frac{N k_B}{C_V} = \frac{C_P}{C_V} - 1$$

$$P V = N k_B T$$

$$T = P V \cdot \frac{1}{N k_B}$$

$$P V V^{\gamma-1} = K''$$

$$P V^{\gamma} = K'''$$

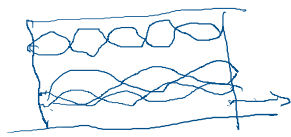
$$P_1 V_1^{\gamma} = P_2 V_2^{\gamma}$$

$$\Rightarrow \frac{P_1}{P_2} = \left(\frac{V_2}{V_1} \right)^{\gamma}$$

$$PV^\gamma = \text{const}$$

$$\gamma = \frac{C_p}{C_v}$$

$$\frac{\frac{5}{2}Nk}{\frac{3}{2}Nk} = \frac{5}{3}$$



$$\psi(x)$$

$$P(x) \sim |\psi|^2$$

Nabudece Uč 1830:

- nerovinné procesy
- entropia