(a) special cases OH: hydroxide ion SCN: thiocyanate ion CN : cyanide ion (CH₃COO : acetate)

(b) oxoanion

(b) oxomion		
Symbol of oxoanion	Name of oxoanion	
CO ₃ ²	Carbonate ion	
HCO ₃	Hydrogen carbonate ion	
SO ₃ ²⁻	Sulfite ion	
SO ₄ ²⁻	Sulfate ion	
HSO ₄	Hydrogen sulfate ion	
NO_2	Nitrite ion	
NO ₃	Nitrate ion	
ClO-	Hypochlorite ion	
ClO ₂ -	Chlorite ion	
ClO ₃ -	Chlorate ion	
ClO ₄	Perchlorate ion	
BrO	Hypobromite ion	
BrO ₂	Bromite ion	
BrO ₃	Bromate ion	
BrO ₄	Perbromate ion	
IO ₃	Iodate ion	
PO ₄ ³⁻	Phosphate ion	
HPO ₄ ²⁻	Hydrogen phosphate ion	
H ₂ PO ₄	Dihydrogen phosphate ion	
MnO ₄	Permanganate ion	
CrO ₄ ²⁻	Chromate ion	
$\operatorname{Cr_2O_7}^{2^2}$	Dichromate ion	

Chemical formula	Name of cation	Name of anion	Name of salt
KIO ₃	Potassium	Iodate	Potassium iodate
KMnO ₄	Potassium	Permanganate	potassium permanganate
FeCl ₂	Iron(II)	Chloride	iron (II) chloride
CuS	Copper(II)	Sulfide	copper (II) sulfide
Ca(HCO ₃) ₂	Calcium	Hydrogen	calcium hydrogen carbonate
		carbonate	
NaClO	Sodium	Hypochlorite	sodium hypochlorite
(NH ₄) ₂ HPO ₄	Ammonium	Hydrogen	ammonium hydrogen phosphate
		phosphate	
Fe ₂ (SO ₄) ₃	Iron(III)	Sulfate	iron (III) sulfate
K ₂ Cr ₂ O ₇	Potassium	Dichromate	potassium dichromate