

# Method description for the PE curves

28 August, 2017

This document presents a general overview of the different steps that were used to calculate the photosynthetic parameters. These parameters have been determined from the P vs E curves obtained according to a method derived from Lewis and Smith (1983) both on water samples taken from the rosette and ice core bottom (last cm, see an example in Fig. 1). Some samples were also collected directly under the ice (UISW).

**Date: 2016–160 Depth: 5**

**alpha = 0.00113; beta = 8.71e-05; p0 = -0.0113; ps = 0.0866; pm = 0.0655; ek = 58.2**

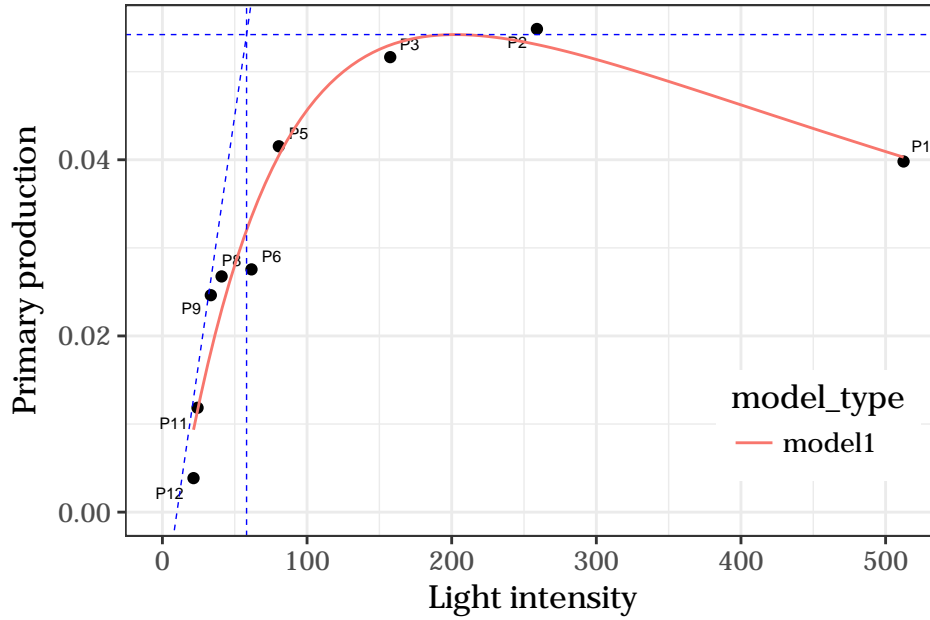


Figure 1: Example of a fitted PvsE curve.

Two different models based on the original definition proposed by (Platt et al., 1980) were used depending on the situation.

## Model with photoinhibition

When apparent photo-inhibition was present, a model including two exponential was fitted (equation 1).

$$p = ps \times (1 - e^{-\alpha \times \frac{light}{ps}}) \times e^{-\beta \times \frac{light}{ps}} + p0 \quad (1)$$

## Model without photoinhibition

When no apparent photo-inhibition was present, a model including only one exponential was fitted (equation 2).

$$p = ps \times (1 - e^{-\alpha \times \frac{light}{ps}}) + p0 \quad (2)$$

The non-linear fitting was done using the Levenberg-Marquardt algorithm implemented in the minpack.lm R package (Timur et al. 2016).

## Dissolved inorganic carbon (DIC) calculation

The calculation of photosynthetic production requires DIC values (see W in the following equation).

Photosynthetic production = (Rs-Rb) x W / (R x N)

- R: total activity (dpm) of bicarbonate added
- N: number of hours of incubation
- Rs: sample count (dpm) corrected for background quenching
- Rb: dark bottle count corrected for quenching
- W: weight of total carbon dioxide (mg C m-3) -> DIC

However, there was not always DIC measurements for each Production versus Irradiance (PvsE) curve. Therefore we tested three different methods to estimate DIC from salinity. After visual inspection and discussions with Marcel Babin and Patrick Raimbault it was decided to use fixed DIC concentrations (ug x L-1) for two ranges of salinity in the conversion of “Disintegrated count Per Minute” (dpm) to production values (mg C m-3 h-1, Parsons et al. 1984). DIC for salinity >= 30 was fixed to 25900 ug L-1, otherwise to 24000 ug L-1.

## Exported variables

The following variables are available:

Variable	Unit
ps	mg C m-3 h-1
$\alpha$	mg C m-3 h-1 [ $\mu$ mol photon m-2 s-1 ]-1
$\beta$	$\mu$ mol photon m-2 s-1
p0	mg C m-3 h-1
pm	mg C m-3 h-1
ek	$\mu$ mol photon m-2 s-1
im	$\mu$ mol photon m-2 s-1
ib	$\mu$ mol photon m-2 s-1
alpha_b	mg C m-3 chl -1 h-1 [ $\mu$ mol photonm-2 s-1 ]-1
beta_b	[ $\mu$ mol photonm-2 s-1] mg chl-1
pb_max	mg C mg chl-1 h-1

Variable  $ps$ ,  $\alpha$ ,  $\beta$  and  $p0$  are from equations 1 and 2. Note that  $\alpha\_b$  and  $pb\_max$  are normalized by chl-a measured from HPLC.

Other variables are defined as:

$$pm = ps \times \frac{\alpha}{\alpha + \beta} \times \frac{\beta}{\alpha + \beta}^{\frac{\beta}{\alpha}} \quad (3)$$

$$ek = \frac{pm}{\alpha} \quad (4)$$

$$im = \frac{ps}{\alpha} \times \ln\left(\frac{\alpha + \beta}{\beta}\right) \quad (5)$$

$$ib = \frac{ps}{\beta} \quad (6)$$

## References

- Lewis, M. R. and J. C. Smith (1983). “A small volume, short-incubation-time method for measurements of photosynthesis as a function of incident irradiance.” *Marine Ecology Progress Series* 13: 99-102.
- Plat, T., C.L. Gallegos and W.G. Harrison (1980). Photo-inhibition o photosynthesis in natural assemblages of marine phytoplankton. *J. Mar. Res.* 38:687-701.
- Timur V. Elzhov, Katharine M. Mullen, Andrej-Nikolai Spiess and Ben Bolker (2016). *minpack.lm*: R Interface to the Levenberg-Marquardt Nonlinear Least-Squares Algorithm Found in MINPACK, Plus Support for Bounds. R package version 1.2-1. <https://CRAN.R-project.org/package=minpack.lm>
- Parsons, T.R., Maita, Y., Lalli, C.M., 1984. *A Manual of Chemical and Biological Methodsfor Seawater Analysis*. Pergamon Press, Toronto