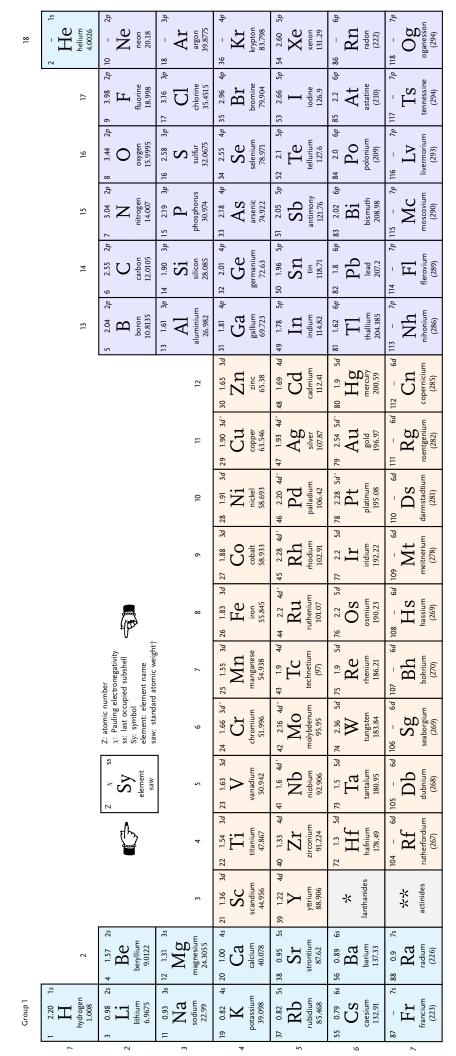




Periodic Table of the Elements







| 4 | | | | <i>p</i> 9 | | _ | |
|---|------------------|------------|----------|---------------------------|-------------------------|---------|----------|
| .27 | Lu | etinm | 74.97 | 1 | Ľ | enciun | (997 |
| 1 1 | _ | 크 | 11 | 103 | | lawr | ت |
| 4f | | _ | | . 2 <i>f</i> | | _ | |
| 1 | Хþ | rbiun | 73.05 | 1.3 | $\frac{N}{N}$ | beliun | 259) |
| | | ytte | | 102 1.3 | _ | no | _ |
| 4f 70 | | | | 2 <i>f</i> | | E | |
| 1.25 | Tm | ulium | 58.93 | 1.3 | Md | Jelevi | 258) |
| . 69 | H | ÷ | Ξ | 5 <i>f</i> 101 1.3 | | mend | _ |
| 4f 69 | | | | . £ | | | |
| 1.24 | $\overline{\Pi}$ | bium | 67.26 | 1.3 | Fm | miun | 257) |
| . 89 | | ē | <u>~</u> | 5f 100 1.3 | щ | fer | _ |
| 4f 68 | | _ | | . £ | | E | |
| .23 | Ho | miun | 4.93 | 1.3 | ES | einiu | 252) |
| 27 | Щ, | hol | 16 | | _ | einst | ٽ |
| 4f 67 | | ٦ | | 2f 99 | | L | |
| 1.22 | Dy | rosiun | 52.5 | 1.3 | Ċţ | rniun | (12) |
| .e | Ц | dyspi | . 16 | . 8 | | califo | (2 |
| 4f 66 | | | | 2 <i>f</i> 98 | | | |
| | Γ | inm | 3.93 | 1.3 5 | Bķ | elium | (7 |
| | H | terb | 158 | 7 1 | М | berk | (2) |
| 4f 65 | | | | £ 64 | | | |
| 2 4 | В | inium | .25 | - 5f* | Cm | ш | 5 |
| - | U | gadoli | 157. | | $\overline{\mathbf{C}}$ | curi | (24 |
| 4f 64 | | | | 96 £s | | | |
| 4. | ゴ | inm | 96 | 2. | П | cium | 3) |
| ' | Eu | enrop | 151 | | Am | ameri | (24 |
| f 63 | | | | 2 <i>f</i> 95 | | | |
| 7 | П | rium | .36 | | Pu | nium | 4 |
| Ξ | Sm | samai | 150. | 1.28 | Ъ | olutor | (24 |
| 4f 62 | | | | . 94 | | _ | |
| 4 | IJ | hium | 2) | 6 5 <i>f</i> ° | Д | inm | (|
| 1 | Pm | rometh | (145) | 1.3 | Z | neptur | (237) |
| <u>f</u> | | ď | | . 93 | | _ | |
| 4 | J | nium | 24 | 3 5f | _ | E | 93 |
| 1.14 | pN | eodymium | 144. | 1.38 | \supset | uraniu | 238.03 |
| 4f 60 1.14 4f 61 | | _ | | 1.5 5f 92 1.38 5f 93 1.36 | | | |
| 1 4 | | minm | 11 | 5 <i>f</i> . | ج. | nium | 4 |
| 1.13 | \mathbf{P}_{1} | aseodymium | 140.91 | 1.5 | Pa | otactin | 231.04 |
| 29 | | pra | | 16 | | pre | |
| 4 <i>f</i> | (L) | E | 7 | 5 <i>f</i> . | J | Ē | 4 |
| 1.12 | Ge | ceriui | 140.12 | 1.3 | Th | thoriu | 232.04 |
| 28 | | | | 96 | | | |
| 1,1 5 <i>d</i> 58 1,12 4 <i>f</i> 59 1,13 | | шп | Į. | 1.1 6d* 90 1.3 5f* 91 | | ш | |
| Ξ | Γ_{a} | ınthanum | 138.91 | Ξ | Ac | ctinium | (227) |
| 22 | | ā | | 68 | | ď | |
| | | | | | 1. | | |
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+Standard atomic weights (average terrestrial atomic weight) taken from the Commission on Isotopic Abundances and Atomic Weights (http://www.ciaaw.org/abridged-atomic-weights.html). If CIAAW indicates a range for the standard atomic weight in parentheses (e.g., Francium (223)) have no known stable isotopes. Therefore, the mass of a representa-'Indicates an anomalous (Aufbau rule-breaking) ground state electron configuration. Inspired by Ivan Griffin's Effzk Periodic Table. Distributed under the MIT open source license. 2019. Paul N. Danese tive isotope is provided.



