


Periodic Table of the Elements

Group 1

18

1 H hydrogen 1.008		2																	
3 Li lithium 6.9675		4 Be beryllium 9.0122		<div></div>															
11 Na sodium 22.99		12 Mg magnesium 24.3055		3		4		5		Z Sy element saw									
19 K potassium 39.098		20 Ca calcium 40.078		21 Sc scandium 44.956		22 Ti titanium 47.867		23 V vanadium 50.942		41 Nb niobium 92.906		73 Ta tantalum 180.95		105 Db dubnium (268)					
37 Rb rubidium 85.468		38 Sr strontium 87.62		39 Y yttrium 88.906		40 Zr zirconium 91.224		41 Nb niobium 92.906		73 Ta tantalum 180.95		105 Db dubnium (268)							
55 Cs caesium 132.91		56 Ba barium 137.33		*		72 Hf hafnium 178.49		73 Ta tantalum 180.95		105 Db dubnium (268)									
87 Fr francium (223)		88 Ra radium (226)		** actinides		104 Rf rutherfordium (267)		105 Db dubnium (268)											

Z	Sy
element	symbol
saw	saw



Z: atomic number

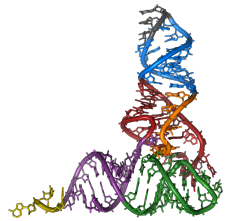


Sy: symbol
element: element name
saw: standard atomic weight†

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57 La lanthanum 138.91	58 Ce cerium 140.12	59 Pr praseodymium 140.91	60 Nd neodymium 144.24	61 Pm promethium (145)	62 Sm samarium 150.36	63 Eu europium 151.96	64 Gd gadolinium 157.25	65 Tb terbium 158.93	66 Dy dysprosium 162.5	67 Ho holmium 164.93	68 Er erbium 167.26	69 Tm thulium 168.93	70 Yb ytterbium 173.05	71 Lu lutetium 174.97
89 Ac actinium (227)	90 Th thorium 232.04	91 Pa protactinium 231.04	92 U uranium 238.03	93 Np neptunium (237)	94 Pu plutonium (244)	95 Am americium (243)	96 Cm curium (247)	97 Bk berkelium (247)	98 Cf californium (251)	99 Es einsteinium (252)	100 Fm fermium (257)	101 Md mendelevium (258)	102 No nobelium (259)	103 Lr lawrencium (266)



†Standard atomic weights (average terrestrial atomic weight) taken from the Commission on Isotopic Abundances and Atomic Weights (<http://www.ciaaw.org/abridged-atomic-weights.htm>). If CIAAW indicates a range for the standard atomic weight of an element, I used the arithmetic mean of the boundaries of the range. Elements with atomic weight in parentheses (e.g., Francium (223)) have no known stable isotopes and it is therefore impossible to provide a standard atomic weight. For these elements, the mass of a representative isotope is provided.
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