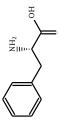


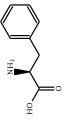
Group 1

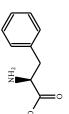
hydrogen 1.008

0.98

Li lithium 6.9675







He

28

helium 4.0026

argon 39.8775

Ar

3p

neon 20.18

krypton 83.798

 $\vec{\Gamma}$

46

Xe

2.60

24

5p

xenon 131.29

Rn

At

Po 2.0

98

*6*9

2.2

*6*9

6р

2.02 Bi 83 6р

82 6р

1.62

54

6:1

. ps

2.54

29

. ps

2.28

54

54

2.2

54

59

2.36

54

29

7.3 Ξ

米

0.89

26

6s

0.79

22

 C_{S}

 $\mathop{Rb}_{\text{rubidium}}$

potassium 39.098

0.82

 $\mathbf{K}^{0.82}$

 $\underset{\scriptscriptstyle{22.99}}{Na}$

0.93

 $\underset{\text{Darium}}{Ba}$

caesium 132.91

(97) 6.1 (222)

astatine (210)

polonium (209)

bis muth 208.98

 $\underset{\text{lead}}{Pb}$ <u>~</u>.

thallium

Hg mercury 200.59

204.385

nts Ho		71 91 51 14 17	S 2.04 2p 6 2.55 2p 7 3.04 2p 8 3.44 2p 9 3.98 2p S 2.04 2p P 2.04 2p	3p 14 1.90 3p 15 2.19 3p 16 2.58 3p 17	Al Si F S CI	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
Periodic Table of the Elements					12	$\sum_{\substack{\text{zinc}\\65.38}}^{30}$	48 1.69 Cd cadmium 112.41
re El					E	34 29 1.90 3d Copper 63.546	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
of th					01	3d 28 1.91 3a Ni nickel 58.693	$\mathop{Pd}_{\text{palladiur}}^{2.20}$
able					6	27 1.88 CO cobalt 58.933	45 2.28 4d° 46 Rh rhodium 102.91
ic T				٠.	∞	3d 26 1.83 3d Fe iron 55.845	$\underset{\text{ruthenium}}{Ru}$
riod			Z: atomic number X: Pauling electronegativity ss: last occupied subshell Sy: symbol element: element name	d atomic weight†	7	$\stackrel{\text{25}}{\text{Mn}} \stackrel{\text{1.55}}{\text{manganes}}$	$\begin{array}{cccc} 43 & 1.9 & 4d \\ & TC \\ & \text{technetium} \\ (97) \end{array}$
-			_	saw: standard atomic	9	$\sum_{\text{chromium}\atop{51.996}}^{24}$	42 2.16 4 <i>d</i> MO molybdenum 95.95
OHOO			$\sum_{\text{element}}^{\chi}$ ss	saw	5	23 1.63 3 <i>d</i> V vanadium 50.942	4d 41 1.6 4d ND niobium 92.906
					4	3d 22 1.54 3d Ti titanium 47.867	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
					æ	Scandium 44.956	Ss 39 1.22 4d Y Y Y Y X X X X X X X X X X X X X X X
		2	4 1.57 2s Be beryllium 9.0122	12	IVI B magnesium 24.3055	$\overset{\text{20}}{\text{Ca}}\overset{\text{1.00}}{\text{calcium}}$	5s 38 0.95 5s Ss Standard Strontium 87.62
	- s	<u>_</u>	25	3s		84 E 2	5s m

4f			<i>p</i> 9		_
4f 71 1.27	Lu	lutetium 174.97	5f 103		lawrencium (266)
	Yb	ytterbium 173.05	102 1.3 5 <i>f</i>	No	ium nobelium (259)
4f 69 1.25 4f 70	Tm	thulium 168.93	5f 101 1.3 5f	рW	mendelevium (258)
1.24	Er	erbium 167.26	5f 100 1.3 5f	Fm	fermium (257)
	Но	holmium 164.93	5f 99 1.3 5f	Es	einsteinium (252)
4f 66 1.22 4f 67	Dy	dysprosium 162.5	5f 98 1.3 5f	CŁ	californium (251)
	Тþ	terbium 158.93	5f 97 1.3 5f	Bk	berkelium (247)
4f 64 1.2 4f° 65	Вd	gadolinium 157.25	96	Cm	curium (247)
4f 63 4f	Eu	europium 151.96	sf 95 sf	Am	americium (243)
62 1.17 4	Sm	samarium 150.36	94 1.28	Pu	plutonium (244)
4f 61 4f	Pm	promethium (145)	93 1.36 5 <i>f</i> *	Np	neptunium (237)
	pΝ	neodymium 144.24	1.5 5f 92 1.38 5f 93	n	uranium 238.03
59 1.13 4 <i>f</i>	Pr	praseodymium 140.91	ı	Pa	protactinium 231.04
5d 58 1.12 4f 59 1.13 4f 60 1.14	G	cerium 140.12	1.1 6d* 90 1.3 5f* 91	Ц	thorium 232.04
57 1.1 5 <i>d</i> °	La	lanthanum 138.91	89 1.1 6 <i>d</i> *	Ac	actinium (227)
	*			* *	

 $\log_{\text{oganesson}}$

tennessine

(294)

livermorium (293)

moscovium (290)

flerovium (289)

nihonium

copernicium

roentgenium Rg

darmstadtium Ds

meitnerium

hassium

(269)

bohrium (270) Bh

Sg seaborgium (269)

dubnium (268)

rutherfordium

actinides 水水

 $\underset{\scriptscriptstyle{(226)}}{Ra}$

francium (223)

 $\mathbf{F}_{\mathbf{I}}$

6.0

88

Dp

Rf (267)

M(278)

Hs

(281)

(282)

(285)

Cu

р9

р9

*p*9

108

*b*9

107

р9

osmium 190.23

rhenium 186.21

tungsten 183.84

tantalum 180.95

hafnium 178.49

<u>L</u>a 7.5

(586)

 $\frac{1}{2}$

Mc

됴

†Standard atomic weights (average terrestrial atomic weight) taken from the Commission on Isotopic Abundances and Atomic Weights (http://www.ciaaw.org/abridged-atomic-weights.htm). If CIAAW indicates a range for the standard atomic weight of an element, I used the arithmetic mean of the boundaries of the range. Elements with atomic weight in parentheses (e.g., Francium (223)) have no known stable isotopes and it is therefore impossible to provide a standard atomic weight. For these elements, the mass of a representative isotope is provided.

'Indicates an anomalous (Aufbau rule-breaking) ground state electron configuration. Inspired by Ivan Griffin's Efi≵ Periodic Table. Efi≵code is released under the MIT open source license. Final product (this Table) is released under creative commons attribution/share-alike copyright terms. ⊛⊕© 2021. Paul N. Danese

