

Fundamentals of Quantum Computing

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1 The Probabilistic Nature of Qubits

A qubit is comparable to a bit in the “classical” world, but it exists on a sub-atomic level.

When a qubit is measured, its state will either be a “1” or a “0”.

The crucial aspect is that before the measurement, a qubit is in a *superposition* of both states.

For example, a given qubit $|\Psi\rangle$ can be represented as

$$|\Psi\rangle = \alpha|0\rangle + \beta|1\rangle$$

which means a linear combination of the two states $|0\rangle$ and $|1\rangle$.

The coefficients α and β represent the probability of the qubit collapsing into one of the two states when measured. The probability of the qubit collapsing into $|0\rangle$ is $|\alpha|^2$, while the probability of collapsing into $|1\rangle$ is $|\beta|^2$.

Since there is 100% chance of the qubits collapsing into one of the two states, α and β must satisfy the following requirement:

$$|\alpha|^2 + |\beta|^2 = 1$$

A uniform superposition of the two states looks like this:

$$|\Psi\rangle = \frac{|0\rangle + |1\rangle}{\sqrt{2}}$$

which means that we have 50% probability of the state collapsing into a $|0\rangle$ or $|1\rangle$ since $\left(\frac{1}{\sqrt{2}}\right)^2 = \frac{1}{2}$.

When the qubit is measured, the superposition is destroyed, leaving it in a “classical” binary state.

You might have noticed that I used the absolute value in $|\alpha|^2$ or $|\beta|^2$, which doesn’t usually make sense since squaring the value is already going to give us a positive result.

This is because the coefficients α and β can also be complex numbers. The absolute value of a complex number is defined as its distance from the origin.

The states $|1\rangle$ and $|0\rangle$ can also be written as $|+\rangle$ and $|-\rangle$.

2 Spin of a particle

2.1 Definition

Spin is intrinsic angular momentum associated with elementary particles. The particle isn't actually rotating, but spin is a property just like momentum, position, charge and mass.

We can represent a spin S as we would in the classical world, with a vector representing its components

$$S = \begin{pmatrix} S_x \\ S_y \\ S_z \end{pmatrix}$$

This vector has a fixed length depending on the type of the particle.

The spin of an electron, which is known as spin $1/2$, has a magnitude of $\frac{\sqrt{3}\hbar}{2}$.

Since the magnitude of the spin is always the same, a common way to express its direction is with the polar and azimuthal angles, θ and ϕ .

2.2 Measurement

A spin S can be measured. For example, we could measure the projection of the spin on its z -axis. This is written as measuring S_z . We could measure S_x , S_y or we could measure its projection from an arbitrary direction, $S_{\vec{n}}$.

We are given a set of electrons with randomly-oriented spins. For each of them, we measure S_z . We would expect a bunch of values between $\frac{\sqrt{3}\hbar}{2}$ and $-\frac{\sqrt{3}\hbar}{2}$, instead, we find that every measurement has an output of either $\frac{\hbar}{2}$ or $-\frac{\hbar}{2}$, equally distributed.

There are a couple of things to notice:

1. When we measure the spin, its state collapses in either one of the two *eigenvalues*, $|+\rangle$ or $|-\rangle$.
2. The quantity measured is not the entire length of the vector even if it should be either completely up or down. This is due to the Heisenberg uncertainty principle, if the spin was any closer to the vertical $\pm z$ -axis, we would have too much simultaneous knowledge about S_x and S_y .

The state of S_z is now 'locked', if we measure S_z of the same particle again, the result will stay the same.

We then measure S_x of each particle. Since we've already measured S_z we would expect S_x or S_y to satisfy $\sqrt{|S_x|^2 + |S_y|^2 + \frac{\hbar^2}{4}} = \frac{\sqrt{3}\hbar}{2}$, instead, we get random eigenvalues as before.

Now, if we come back to measuring S_z , the previously locked states are lost, we get new random eigenvalues. The act of measuring S_x has destroyed the information contained in S_z .

We notice that the eigenvalues $|+\rangle$ and $|-\rangle$ have the same probability of being measured.

We can make a rotation of θ of the direction of measurement from the z -axis, the state Ψ is now

$$|\Psi\rangle = \cos\left(\frac{\theta}{2}\right)|+\rangle + \sin\left(\frac{\theta}{2}\right)|-\rangle$$

Before, the rotation was $\frac{\pi}{2}$ from the z -axis, so the state was $|\Psi\rangle = \frac{|+\rangle + |-\rangle}{\sqrt{2}}$, which means that it is equally likely that the spin will collapse into a $|+\rangle$ or a $|-\rangle$.

3 Time Independent Quantum Mechanics

In this section the time t is not considered, everything pertains a single instant.

3.1 Particle in One-Dimension

A particle of mass m can move freely in a one-dimensional bounded space. The energy of this particle is quantized, meaning it can possess only certain discrete energy values. Given the energy we can form a probability curve that predicts the likelihood of finding the particle at various locations within our bounded space.

3.2 The First Postulate of Quantum Mechanics

3.2.1 The State Space

For each physical system \mathcal{S} , we associate a Hilbert space \mathcal{H} . Each physical state in \mathcal{S} is represented by a point on the projective sphere (all unit vectors) of \mathcal{H} .

$$\text{physical state} \in \mathcal{S} \quad \longleftrightarrow \quad \hat{v} \in \mathcal{H}$$

The Hilbert space \mathcal{H} is often referred to as *the state space*.

4 Types of Algorithms

4.1 Deterministic

Example: the quantum circuit is run and we measure the output.

4.2 Probabilistic

Example: The more we run the circuit, the higher the chance of getting the right or a more precise result.