

Experiment No. :- 07

1) Aim:- To determine the structure of an unknown compound & using spectroscopic data

Given Molecular formula of the given compound is $C_9H_{10}O$

2) Theory:-

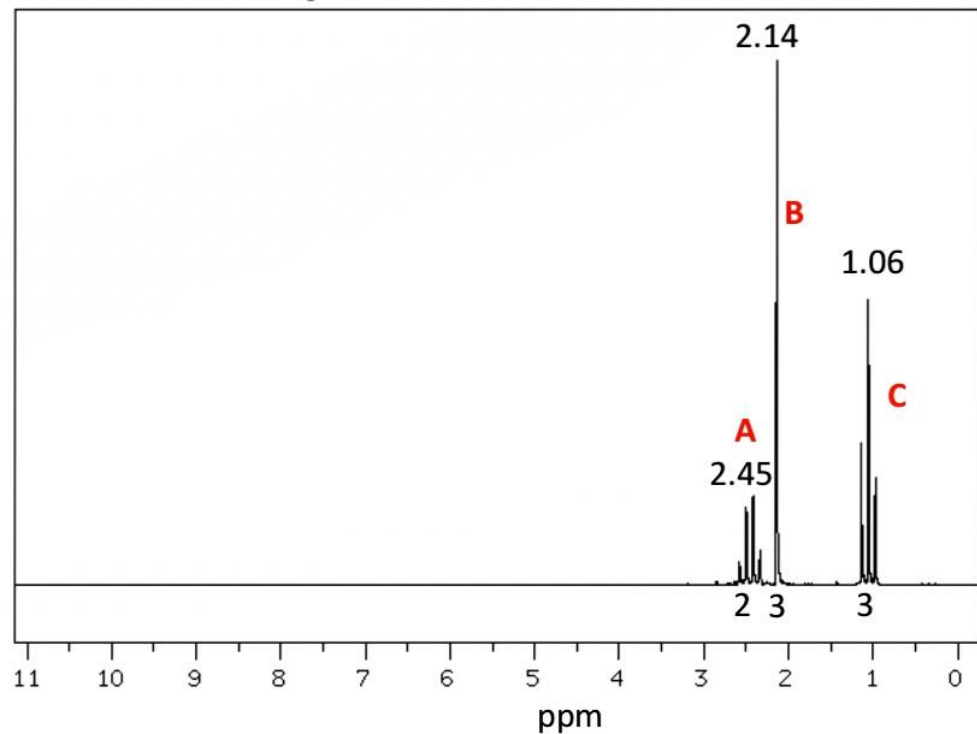
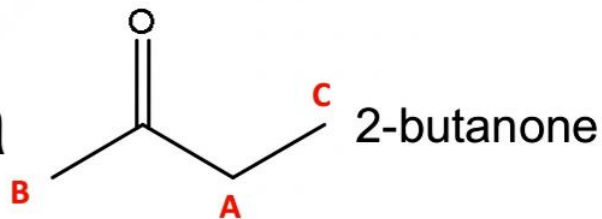
- 1) Spectroscopy is the study of the interaction of the E.M. radiation with the matter
- 2) Spectroscopy is an important tool for the qualitative and quantitative analysis of an unknown compound.
- 3) In the modern structure analysis the powerful spectroscopy methods of I.R. and N.M.R. play a major role
- 4) In order to determine the structure of an unknown compound it may be required to combine a variety of spectroscopic data

IR frequencies for important Functional Groups

Functional Group	Frequency (cm^{-1})
OH stretch (H_2O)	3700–3100
OH stretch (alcohol)	3600–3200
carboxylic acid OH stretch	3600–2500
N–H stretch	3500–3350
$\equiv\text{C}$ –H stretch	~ 3300
$=\text{C}$ –H stretch	3100–3000
–C–H stretch	2950–2840
–C–H aldehydic	2900–2800
$\text{C}\equiv\text{N}$ stretch	~ 2250
$\text{C}\equiv\text{C}$ stretch	2260–2100
$\text{C}=\text{O}$ aldehyde	1740–1720
$\text{C}=\text{O}$ anhydride	1840–1800, 1780–1740
$\text{C}=\text{O}$ ester	1750–1720

Functional Group	Frequency (cm^{-1})
$\text{C}=\text{O}$ ketone	1745–1715
$\text{C}=\text{O}$ amide	1700–1500
$\text{C}=\text{C}$ alkene	1680–1600
$\text{C}=\text{C}$ aromatic	1600–1400
CH_2 bend	1480–1440
CH_3 bend	1465–1440, 1390–1365
$\text{C}-\text{O}-\text{C}$ stretch	1250–1050
$\text{C}-\text{OH}$ stretch	1200–1020
NO_2 stretch	1600–1500 and 1400–1300
$\text{C}-\text{F}$	1400–1000
$\text{C}-\text{Cl}$	800–600
$\text{C}-\text{Br}$	750–500
$\text{C}-\text{I}$	~ 500

Example: NMR spectra



Peak	Integration	Chemical shift (δ)	Multiplicity	Adjacent Hs
A	2	2.45	q	3
B	3	2.14	s	0
C	3	1.06	t	2

Example: IR spectra

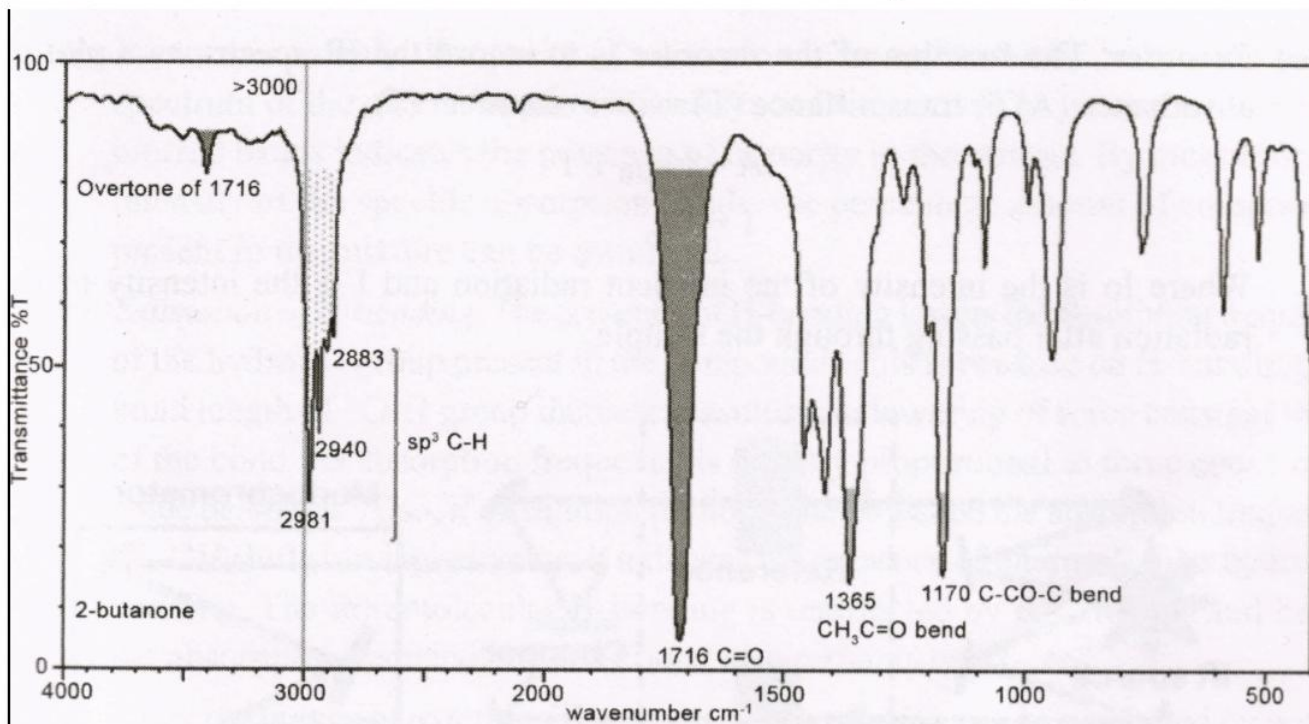
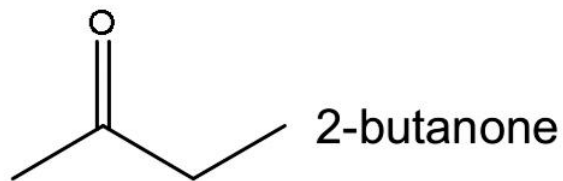


Fig. 2.7: IR spectra of 2-butanone

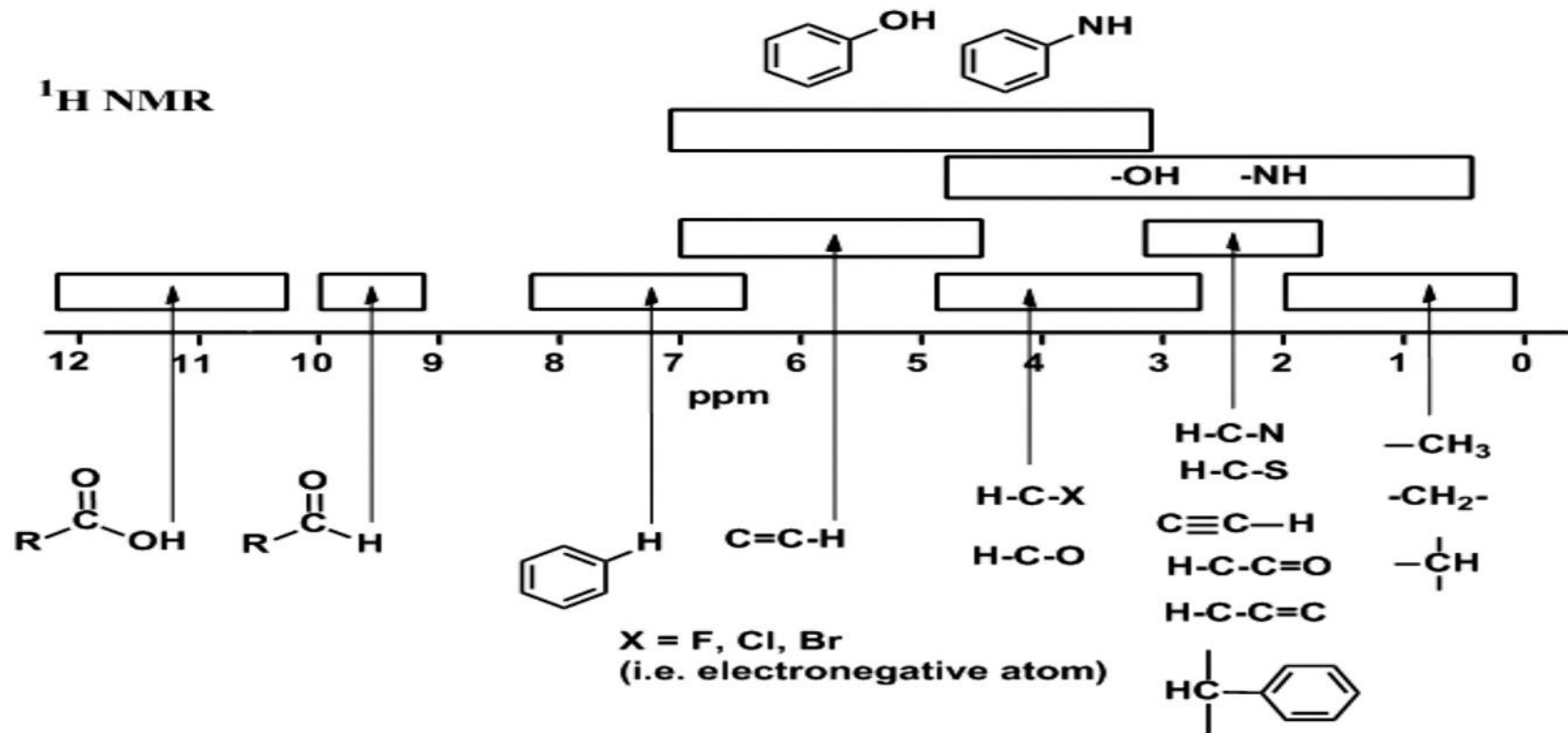
* Infrared Spectroscopy

- 1) Infrared spectroscopy is especially useful in identifying the functional groups in an unknown compound. This is a type of functional group spectroscopy.
- 2) Conventionally the IR spectrum is a plot of inherent transmittance ($\%T$) vs wave number (cm^{-1}).
- 3) The peaks/signals observed in the spectrum can be used for the detection of various important functional groups such as alcohol ($-\text{OH}$), ($-\text{NH}_2$), ($-\text{CO}-$) etc in unknown samples.
- 4) The IR spectrum is interpreted by referring to the reference tables containing the info regarding the functional groups & their corresponding wave numbers and the presence of functional groups is analysed accordingly.

* ^1H -NMR Spectroscopy

- 1) Proton nuclear magnetic resonance is very important & widely used analytical technique.

Characteristic ^1H -NMR Chemical shifts (δ)

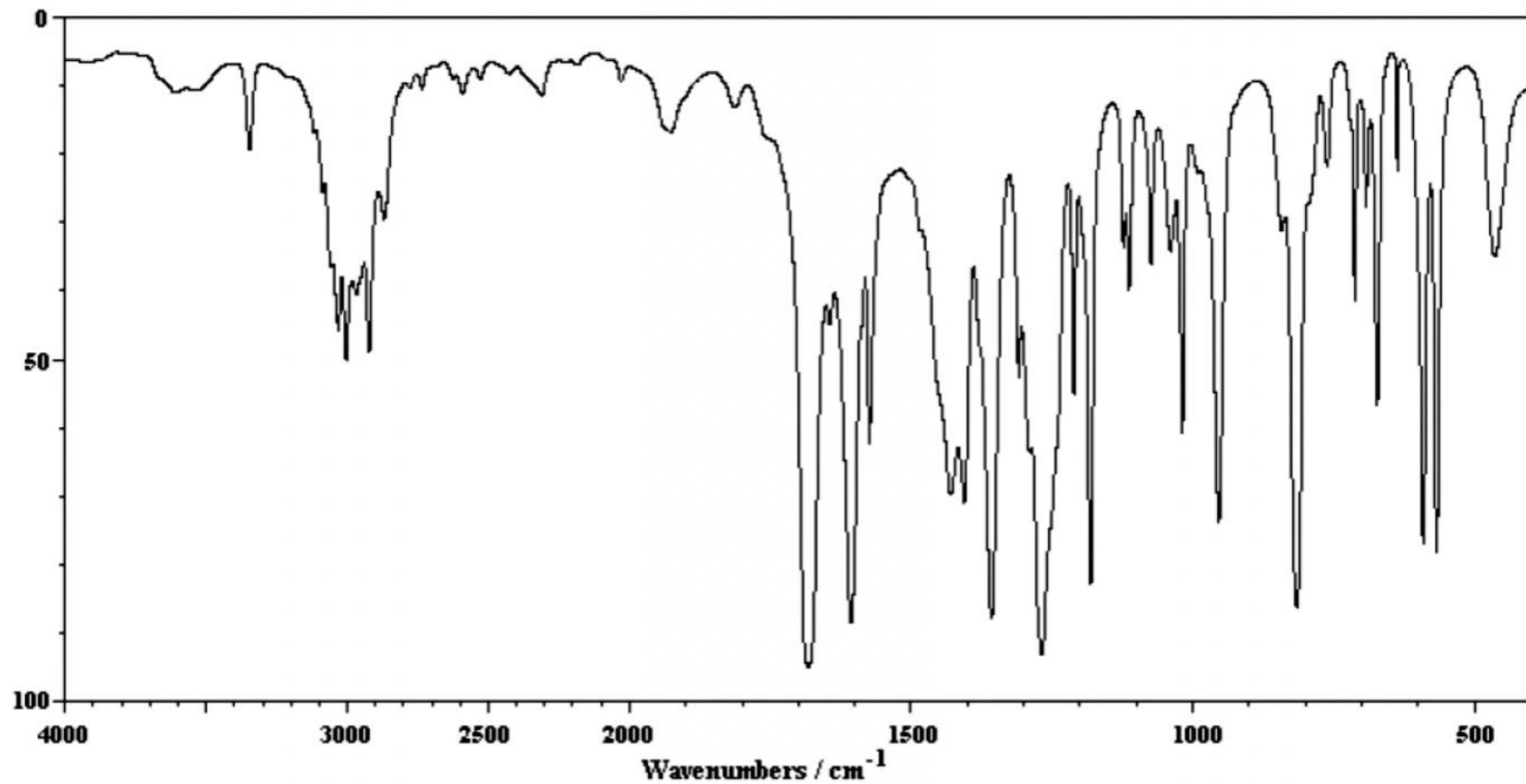


- 2) It is primarily used in the characterization of organic molecules
- 3) ^1H -NMR spectrum is a plot of signal intensity vs chemical shift (δ)
- 4) The info from the ^1H -NMR spectrum is extracted & is often combined with the info from other spectroscopic techniques such as IR, for the characterisation of unknown compounds

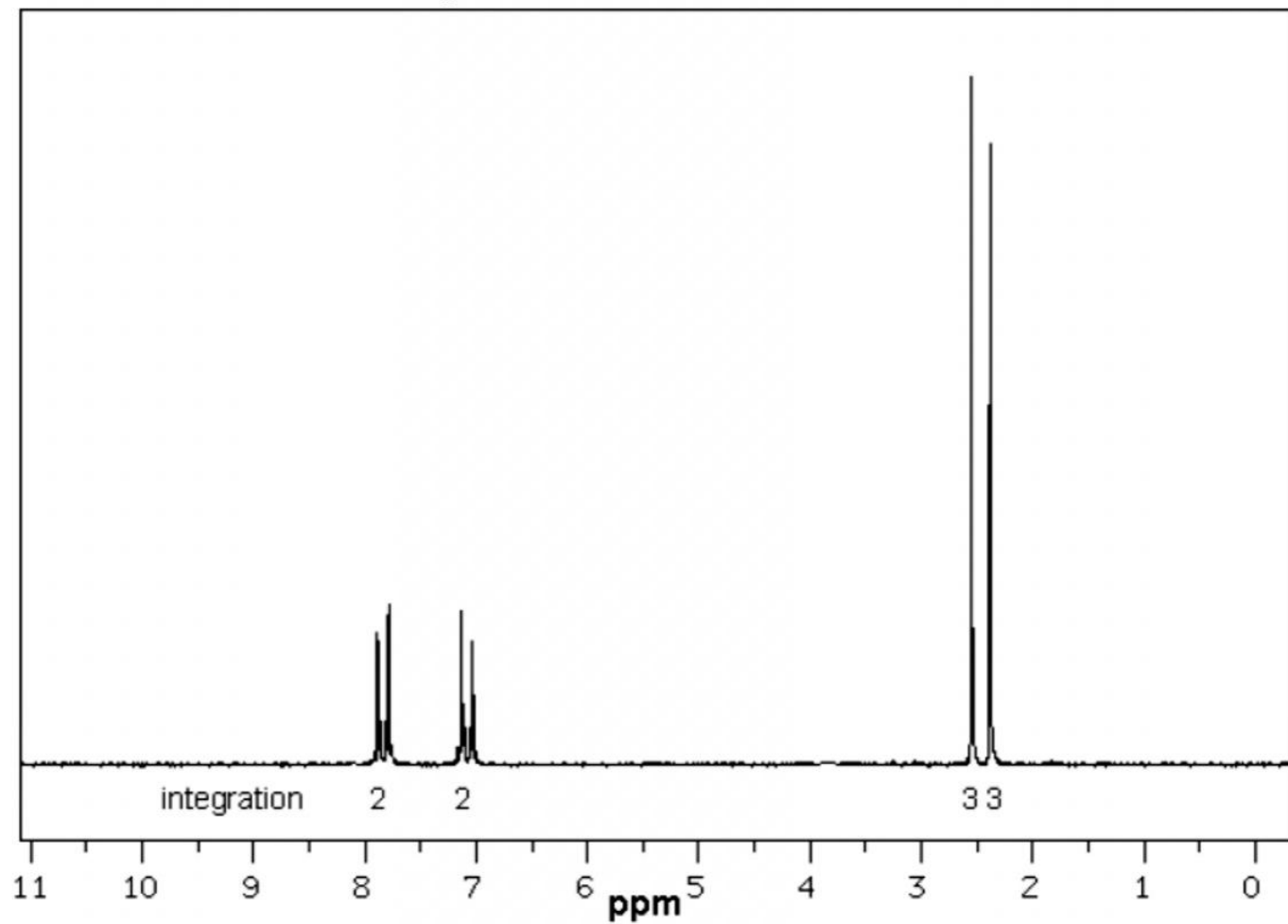
★ Interpretation of ^1H -NMR spectroscopy spectrum
Important info includes

- 1) No. of signals/peaks (No. of ^1H environments)
 - 2) Type of H s (Aliphatic/Aromatic/Functional) based on chemical shifts
 - 3) No. of H s each type (Integration)
 - 4) No. of adjacent H s (splitting/multiplicity of each signal)
- 3) Analysis & Interpretation
- a) Probable structure for unknown compound A is

IR Spectrum of the Unknown Compound A



NMR Spectrum of the Unknown Compound A



Observations

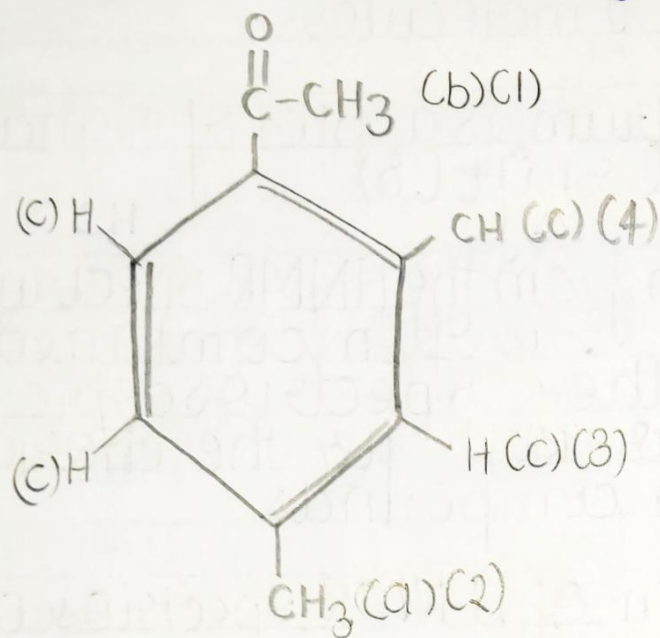
1) IR Spectrum

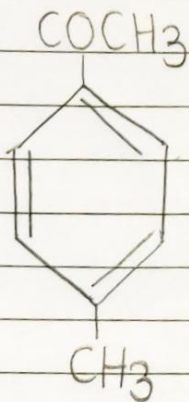
Sr No	Wave (cm ⁻¹) Number	Possible Group	Explanation	Group Present
1	3100-3210	-CH (Alkyl)	Medium & Sharp	-CH (a)
2	2360-2930	=CH (Alkyl/arom)	Medium & sharp	-CH (c)
3	1690-1750	-C=O	High & Broad	-CH=O (d)
4	1560-1620	-CH (attached to C=O)	High & Broad	-CH (b)

2) NMR-Spectrum

Sr No	Chemical Shift	Possible Group	Explanation	Group Present
1	2.4-2.6	-CH (attached to C=O)	High & sharp	-CH (1)
2	2.3-2.5	-CH (Alkyl)	High & sharp	-CH (2)
3	6.9-7.1	-CH (attached to Benzene)	High & sharp	-CH (3)
4	7.3-8.1	-CH (attached to Benzene near C=O)	Low & sharp	-CH (4)

Therefore, the structure of the compound is





b) Does it satisfy the observation from IR and HNMR spectra \rightarrow Yes

4) Result \rightarrow Based on the available spectroscopic data (IR & HNMR) the proposed structure of the unknown compound is // Ans

