

## Experiment No. - 2.

1. Aim: To determine the planck's constant by photo-electric effect

2. Apparatus Required: Light source, Photoelectric tube  
Ammeter, Voltmeter, Ammeter (mA),  
Connecting wires

### 3. Theory:

When light source of suitable frequency falls on the metal surface, electrons are ejected from it and this phenomenon is called photoelectric effect. These electrons are known as photoelectrons and the resulting current is photoelectric current.

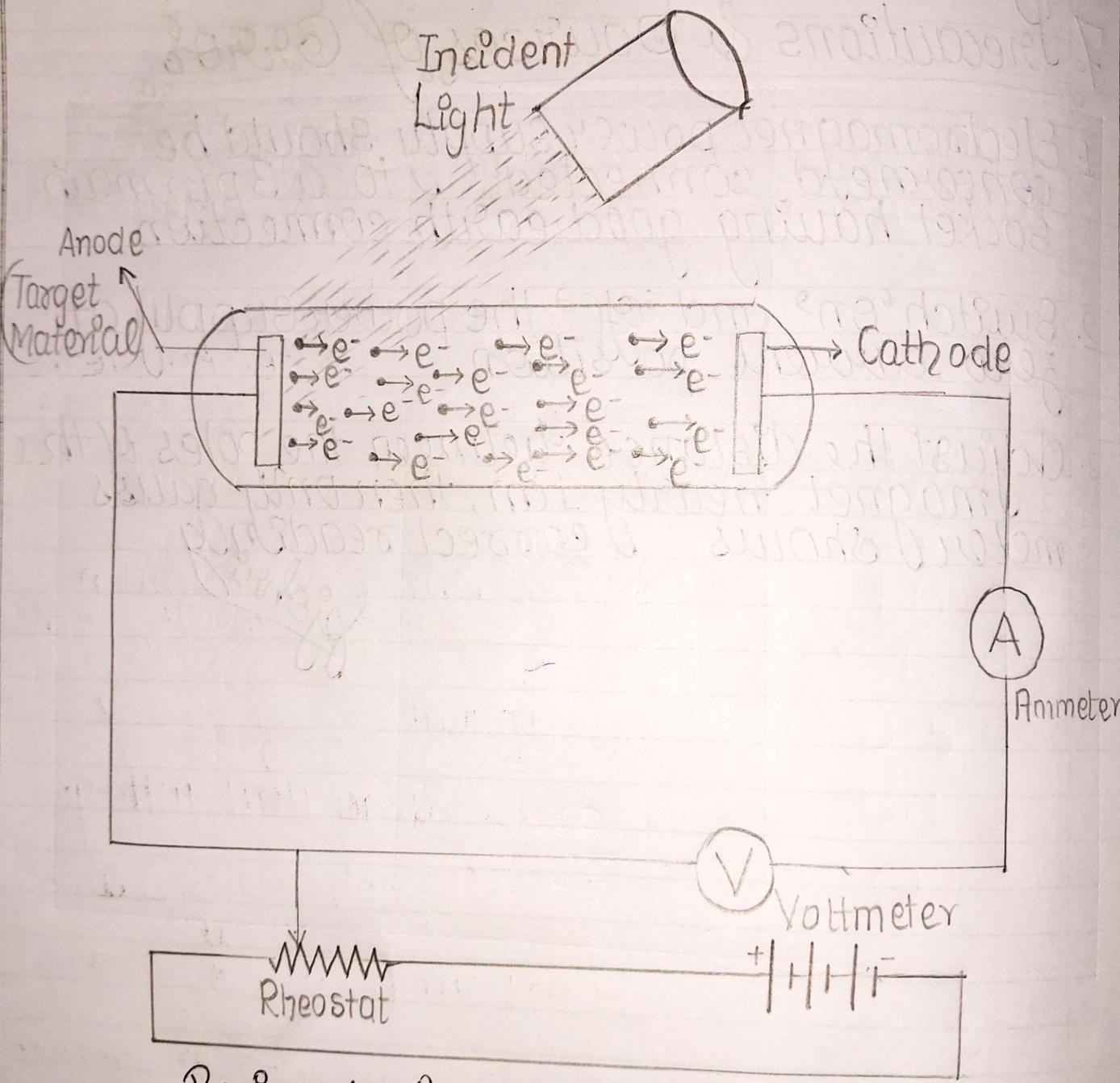
The photoelectric effect is an instantaneous process.

The number of photoelectrons emitted is proportional to the intensity of incident light. Also, the energy of emitted photoelectrons is directly independent of the intensity of incident light. The energy of emitted electrons is directly proportional to the frequency of incident light.

Equation for photoelectric effect

$$h\nu = KE_{\max} + W \quad (i)$$

*Jogesh*



Basic experimental setup for  
photoelectric effect.

Fig.Ci)

~~photo~~

where,

$h\nu$  = Energy of incident photon

$K_{E\text{max}}$  = Maximum KE of ejected electron

$$K_{E\text{max}} = eV_0 - (\text{PQ})$$

where,

$e = 1.6 \times 10^{-19} \text{ C}$  (electron charge)

$V_0$  = Stopping potential (Minimum Voltage required to stop ejection of photo-electrons)

$$W = h\nu_0 - (\text{PQ})$$

$W$  = Work function of metal

$\nu_0$  = Threshold frequency (Minimum frequency for photoelectric effect)

#### 4. Procedure

1. Select the material studying photoelectric effect
2. Select area of material, wavelength, intensity of incident light
3. Switch on the light source
4. Measure the reverse current
5. Determine the threshold Voltage
6. Repeat the experiment by varying the intensity for a particular wavelength of incident light
7. Repeat the experiment for appnbx 1D readings

*Rishabh*

# Observation Table

for copper.

S.No	Wavelength (nm) ( $\text{\AA}$ )	Frequency ( $\nu$ ) ( $10^{15}$ ) Hz	Stopping Potential ( $V_0$ ) in V
1	100	3	7.8
2	120	2.5	5.7
3	140	2.14	4.2
4	160	1.875	3.1
5	180	1.67	2.2
6	200	1.5	1.6
7	220	1.36	1
8	240	1.25	0.5
9	260	1.15	0.1
10	280	1.07	0

~~graph~~

# Calculations

$$h = \text{slope} \times e$$

$$\text{slope} = \frac{(6.7 - 3.6)}{(2.75 - 2) \times 10^{15}}$$

$$= 4.13 \times 10^{-15}$$

$$h = (4.13 \times 10^{-15}) \times (1.6 \times 10^{-19})$$

$$h = \underline{6.608 \times 10^{-34} \text{ Js}} \text{ Ans}$$

# Percentage Error calculations

$$\text{actual } h = 6.626 \times 10^{-34}$$

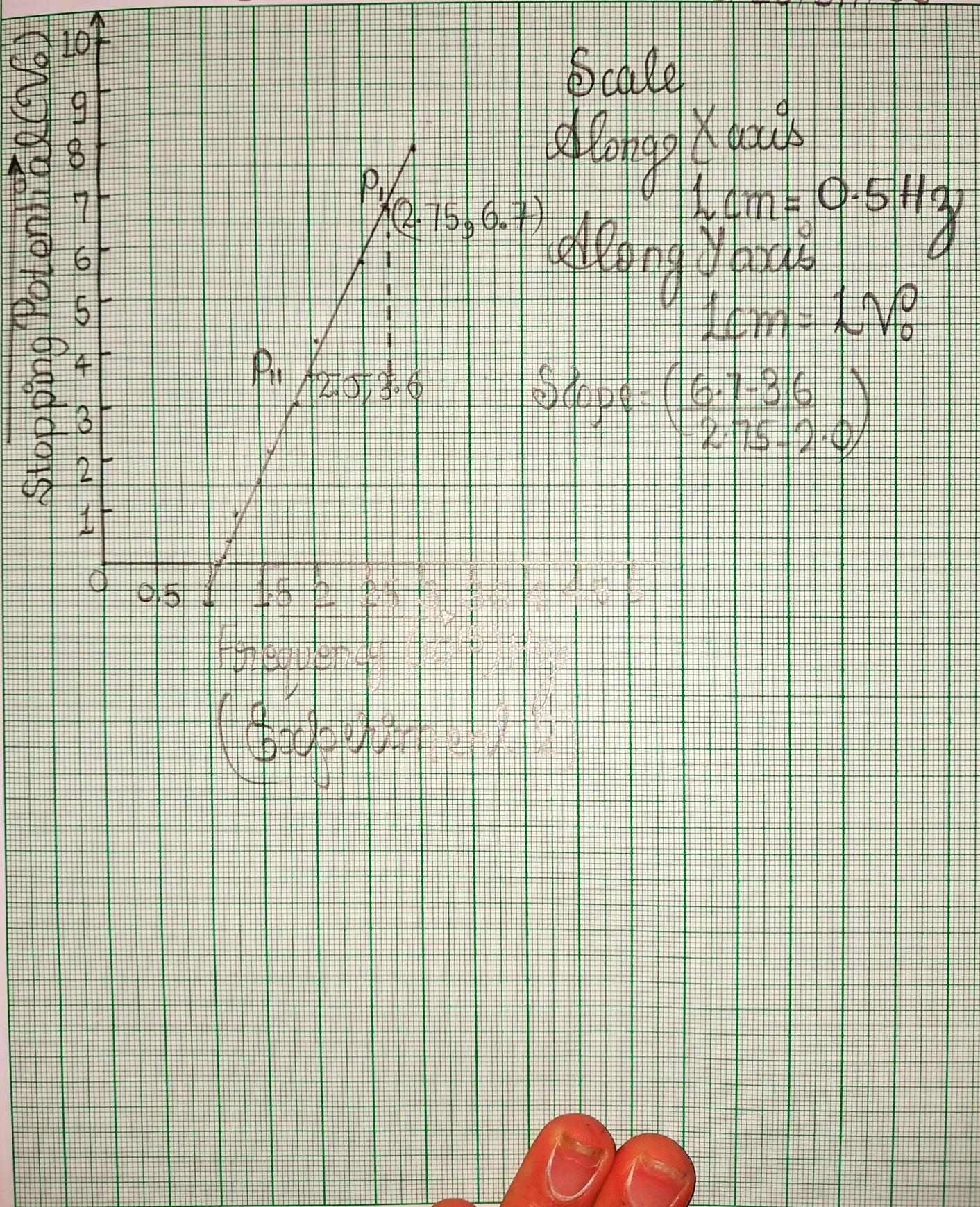
$$\% \text{ error} = \frac{(6.626 - 6.608) \times 10^{-34}}{6.696} \times 100\%$$

$$= \underline{0.27\% \text{ Ans}}$$

~~higher~~

# Experiment no:- 2

Parth Jadhav  
2K20/817/33



## 5. Result:

Planck's constant comes out to be

$$6.608 \times 10^{-34} \text{ Js} \text{ Ams} \text{ and \% error} = 0.27\%$$

## 6. Precautions and Sources of Error

1. Switch on the light before applying voltage
2. Voltage should be lowered slowly until reached stopping potential
3. Make sure to set area of plots to maximum

*for now*