

* Vision & Mission of D.T.U. *

I*) Vision :-

To be a world class university through education, innovation and research for service of humanity

II*) Mission :-

- 1.) To establish centres of excellence in emerging areas of science, engineering, management in allied areas
- 2.) To power foster an ecosystem for innovation, development of technology and entrepreneurship
- 3.) To create environment of collaboration, imagination and creativity.
- 4.) To develop human potential, abilities, ethics & integrity
- 5.) To provide environment friendly and sustainable solutions for global needs.

Vision & Mission of Department of Applied Chemistry.

I) Vision *

To be a centre of excellence in areas of applied chemistry, polymer science & technology.

II) Mission *

- 1) To be a world class centre for education, research & innovation in field of applied chemistry, focus on cutting-edge technologies and foster environment of seamless mess b/w academics & industries.
- 2) To meet need for engineers & scientists at all levels of tertiary education ; UG, PG & doctoral

* Course Objectives *

- 1) To familiarize students with concepts of engineering chemistry, material characteristics & green chemistry
- 2) To give hands on training on various subjects taught in theory classes, to help them do work systematically

* Course Outcomes *

- 1) Understand theoretical principles behind conventional and modern methods of analysis and apply them to perform chemical analysis experiments.
- 2) Understand synthesis, properties & applications of various materials
- 3) Apply knowledge of thermodynamics in studying chemical synthesis
- 4) Understand concepts of chemistry for achieving sustainable development

- General Safety.
- Practices.

- 1) Never work alone in labs
- 2) Wear lab coat, safety goggles and proper shoes while working on the lab.
- 3) Locate fire extinguishers, first-aid kits etc in lab
- 4) No food/drinks are allowed in labs
- 5) Never taste any chemicals or inhale any vapours.
- 6) Spilled chemicals (acid/base) should be rinsed with water & dry sponge ASAP (as soon as possible)
- 7) Seek Medical attention, post-rinsing of eyes if chemicals get into them.
- 8) Broken glass items should be placed in appropriate containers.
- 9) Keep your working area clean
- 10) All Lab packages must be free from blockage

11.) Wash hands after experiment is done.

12.) Report injuries to concerned staff.

* Material Safety (MSDS) Data sheet *

Section 1: Chemical product and company identification

Section 2: Composition & Information on ingredients

Section 3: Hazards Identification

Section 4: First Aid Measures

Section 5: Fire and Explosion Data

Section 6: Accidental release measures

Section 7: Handling & Storage

Section 8: Exposure controls/personal protection

Section 9: Physical & Chemical properties

Section 10: Stability & Reactivity Data

Section 11: Toxicological Information

Section 12: Ecological Information

Section 13: Disposal Considerations.

Section 14: Transport Information

Section 15: Other regulatory

Section 16: Other Information

GHS - Hazard Pictograms and Related Hazard Classes

		
Exploding Bomb <ul style="list-style-type: none">• Explosives• Self-reactives• Organic Peroxides	Corrosion <ul style="list-style-type: none">• Skin corrosion/burns• Eye damage• Corrosive to metals	Flame Over Circle <ul style="list-style-type: none">• Oxidizing gases• Oxidizing liquids• Oxidizing solids
		
Gas Cylinder <ul style="list-style-type: none">• Gases under pressure	Environment <ul style="list-style-type: none">• Aquatic toxicity	Skull & Crossbones <ul style="list-style-type: none">• Acute toxicity (fatal or toxic)
		
Exclamation Mark <ul style="list-style-type: none">• Irritant (eye & skin)• Skin sensitizer• Acute toxicity• Narcotic effects	Health Hazard <ul style="list-style-type: none">• Carcinogen• Mutagenicity• Reproductive toxicity• Respiratory sensitizer	Flame <ul style="list-style-type: none">• Flammables• Pyrophorics• Self-heating• Emits flammable gas

Globally Harmonized System (GHS) of Classification & Labelling of Chemicals

The GHS pictogram consists of a symbol on a white background framed within a red border and represents distinct hazards.

National Fire Protection Association Diamond,

The NFPA has developed a hazard warning symbol called 'NFPA' Diamond. It conveys information regarding hazards quickly. The type of hazard is indicated by colour & location.

Hazards are rated from 0 (very low) to 4 (extreme) in each category, with a provision for special hazards.

#Special Hazards

OX

Oxidizer

ACID

Acid

ALK

Alkali

W

Use of no water

COR

Corrosive

BIO

Biohazard

POI

Poisonous.

Rating Summary.

#Health (BLUE)

4: Danger: Maybe fatal on short exposure
Specialized protective equipment required

3: Warning: Corrosive; avoid skin contact

2: Warning :- May be harmful on Inhalation

1: Caution :- May be irritating

0: No Unusual Hazard.

#flammability (RED)

4:- Danger :- Flammable gas / liquid.

3:- Warning :- Flammable; flash point below
 100°F (37.8°C)

2:- Caution :- Combustible, flash point below
 100°F to 200°F (93.8°C)

1:- Combustible if heated

0:- Combustible stability

#Reactivity (Yellow)

4:- Danger :- Explosive material at room temp.

3:- Danger :- Explosive if heated under

confinement / mixed with water.

2:- Caution :- Unstable / Reacts violently with water

1:- Caution :- May react with water.

0:- Stable :- Not reactive when mixed with water.

Program Outcomes.

Engineering graduates will be able to :-

- 1.) Apply knowledge of maths, science and engineering, specialization to solution of complex engineering problems.
- 2.) Identify, formulate, review and analyze complex engineering problems reaching substantial conclusions using maths, science.
- 3.) Design solutions for complex engineering problems & design systems / components that meet specified needs with apt. conditions for health, culture & environmental consideration
- 4.) Conduct investigations of complex problems. Use research-based knowledge for design of experiments and synthesis of information to provide valid conclusions

- 5.) Enhance modern tool usage, create, select & apply techniques, resources & tools with an understanding of limitations.
- 6.) Apply reasoning paired with continual knowledge to access societal, health issues and responsibilities relevant to professional engineering practice.
- 7.) Understand impact of engineering control solutions in environmental contexts & demonstrate need for sustainable development.
- 8.) Apply ethical principles & commit to professional ethics & norms of engineering practice.
- 9.) Function effectively as an individual and as a member/leader in diverse teams & in multi disciplinary settings.
- 10.) Communicate effectively on complex activities with engineering community & society at large, design & present instructions.
- 11.) Demonstrate knowledge & understanding of engineering & management principles.

apply these to one's own work to manage projects

- 12) Recognise need for preparation & ability to engage in independent learning in broader context of technological change

