



Material Streams							
		Stream_1_inlet	Stream_1_outlet	Stream_2_inlet	Stream_2_outlet	Stream_3_inlet	Stream_3_outlet
Vapour Fraction		0.6000	0.8000	1.0000	0.0000	1.0000	0.0000
Temperature	C	150.0	150.0	250.0	170.4	250.0	250.0
Pressure	kPa	476.0	476.0	800.0	800.0	800.0	3978
Molar Flow	kgmole/h	5.551e-002	5.551e-002	5.551e-002	5.551e-002	5.551e-002	5.551e-002
Mass Flow	kg/h	1.000	1.000	1.000	1.000	1.000	1.000
Liquid Volume Flow	m3/h	1.002e-003	1.002e-003	1.002e-003	1.002e-003	1.002e-003	1.002e-003
Heat Flow	kJ/h	-1.403e+004	-1.360e+004	-1.297e+004	-1.520e+004	-1.297e+004	-1.484e+004

Energy Streams				
		Heater1_duty	Heater_2_duty	Heater_3_duty
Heat Flow	kJ/h	422.8	-2230	-1865

2.6d: $1 \text{ kg/h} = 1/60 \text{ kg/s}$, $422.8 \text{ kJ/h} = 7.046 \text{ kW}$; $\Delta H_{\text{specific}} = (7.046 \text{ kJ/s}) / (1/60 \text{ kg/s}) = 422.8 \text{ kJ/kg}$

$\Delta H = m \cdot \Delta H_{\text{specific}} = (3 \text{ kg})(422.8 \text{ kJ/kg}) = 1268 \text{ kJ}$

2.6e: $1 \text{ kg/h} = 1/60 \text{ kg/s}$, $-2230 \text{ kJ/h} = -37.16 \text{ kW}$; $\Delta H_{\text{specific}} = (-37.16 \text{ kJ/s}) / (1/60 \text{ kg/s}) = -2230 \text{ kJ/kg}$

$\Delta H = m \cdot \Delta H_{\text{specific}} = (3 \text{ kg})(-2230 \text{ kJ/kg}) = -6690 \text{ kJ}$

2.6f: $1 \text{ kg/h} = 1/60 \text{ kg/s}$, $-1865 \text{ kJ/h} = -31.08 \text{ kW}$; $\Delta H_{\text{specific}} = (-31.08 \text{ kJ/s}) / (1/60 \text{ kg/s}) = -1865 \text{ kJ/kg}$

$\Delta H = m \cdot \Delta H_{\text{specific}} = (3 \text{ kg})(-1865 \text{ kJ/kg}) = -5595 \text{ kJ}$