

MAGMA FORMATION AND PROPERTIES

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What is a volcano?

- A vent in the surface of the Earth through which magma and associated gases and ash erupt.

(glossary of geology, fifth edition, 2005)



Crater Lake, Oregon



Augustine, Alaska



Mid Atlantic ocean

What is magma?

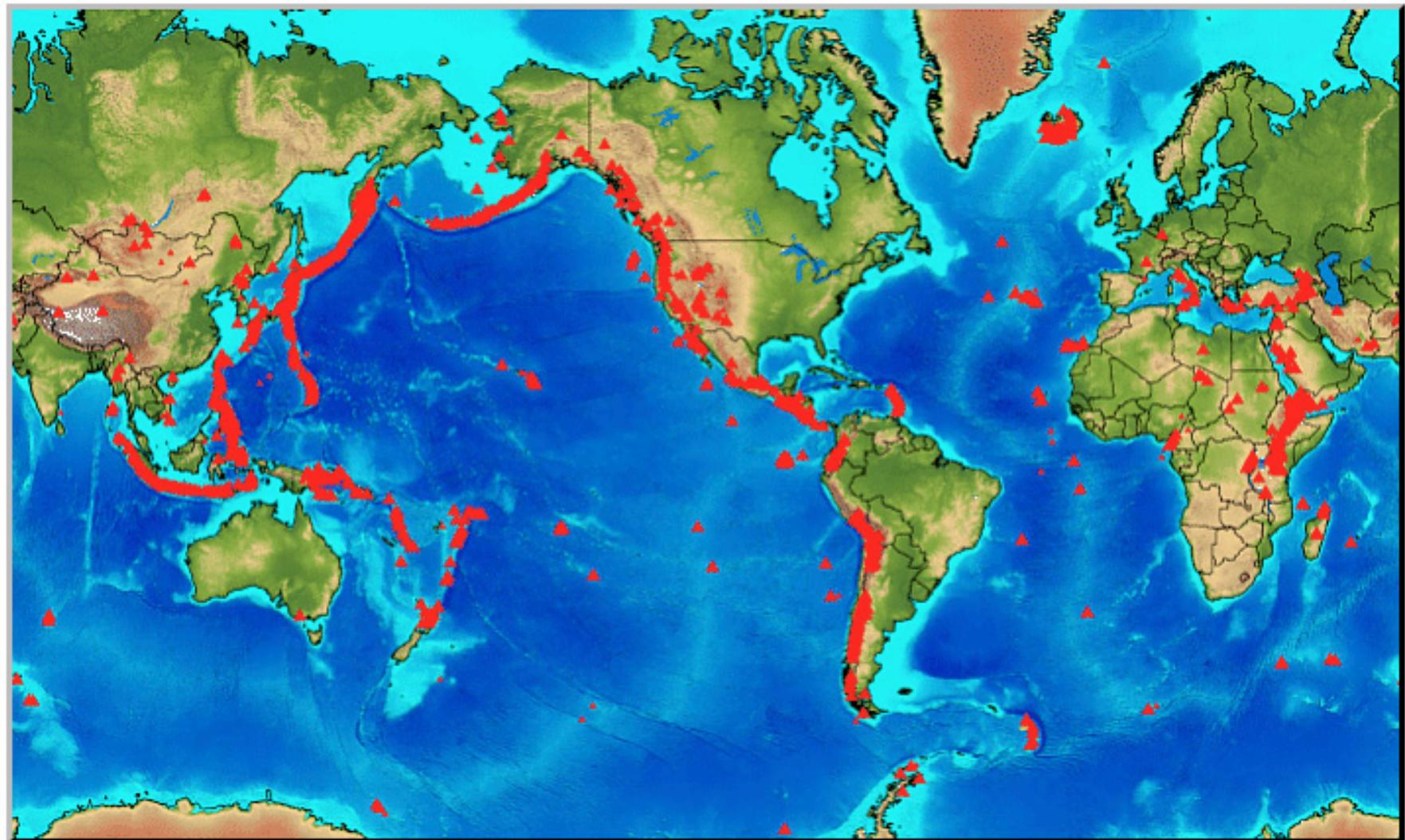
- Naturally occurring molten or partially molten rock material.. Capable of intrusion and extrusion. It may or may not contain suspended solids and/or gas phases

(glossary of geology, fifth edition, 2005)

Volcanoes are located

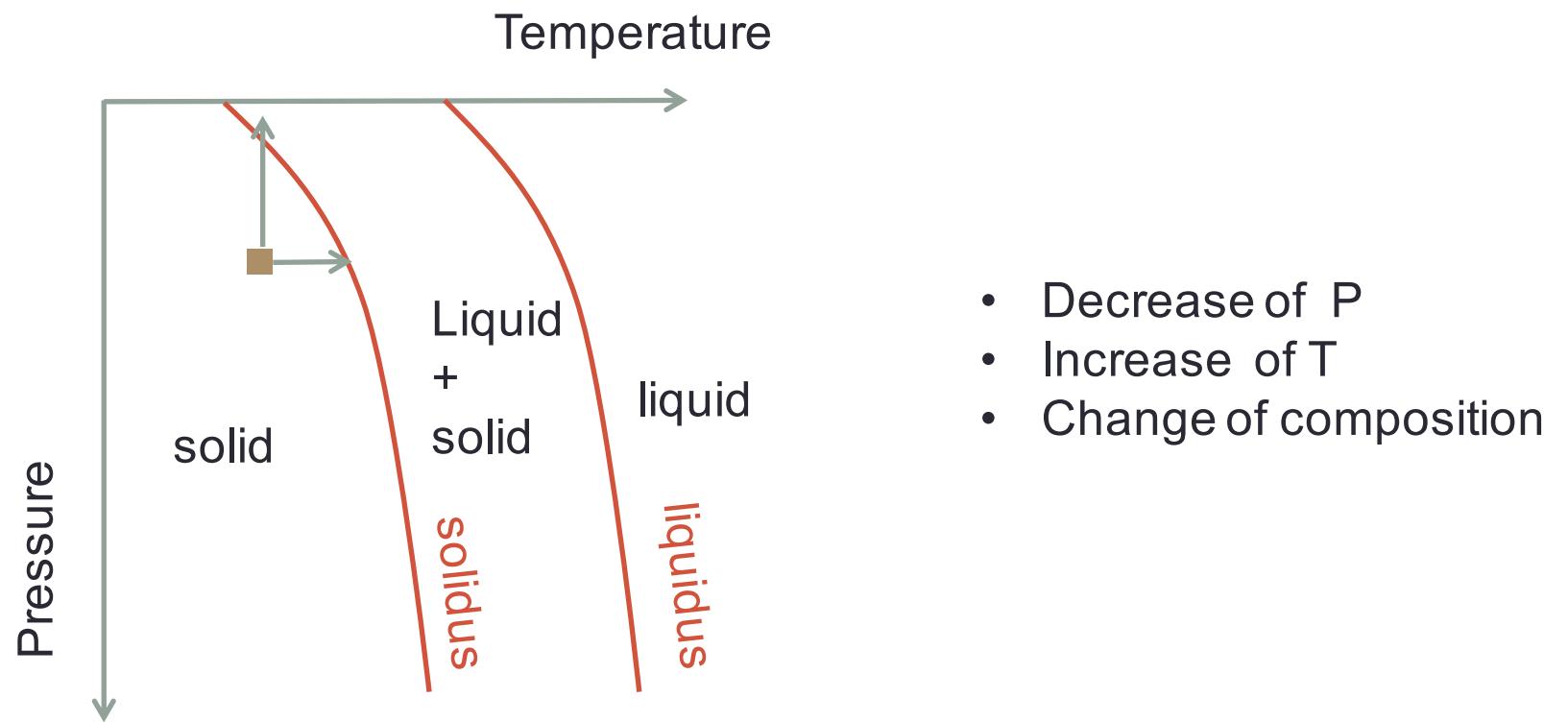
- At the boundaries of:
 - Divergent plates (Mid ocean ridges and continental rifts)
 - Convergent plates (volcanic arcs)
 - Transform boundaries
- Intraplate volcanism (hot spot)

Where are the volcanoes?



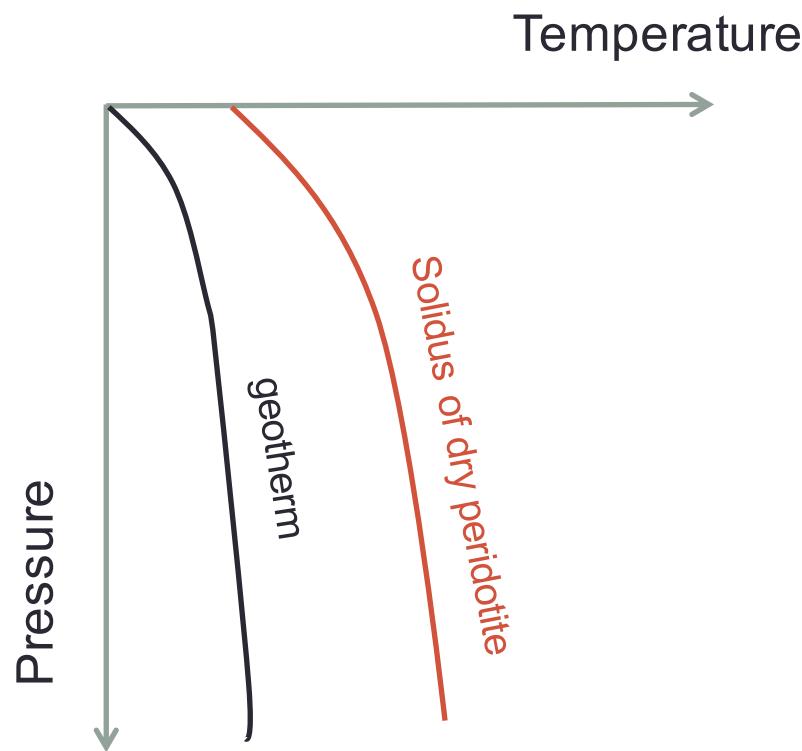
How does magma form?

- Liquidus/solidus temperatures and curves



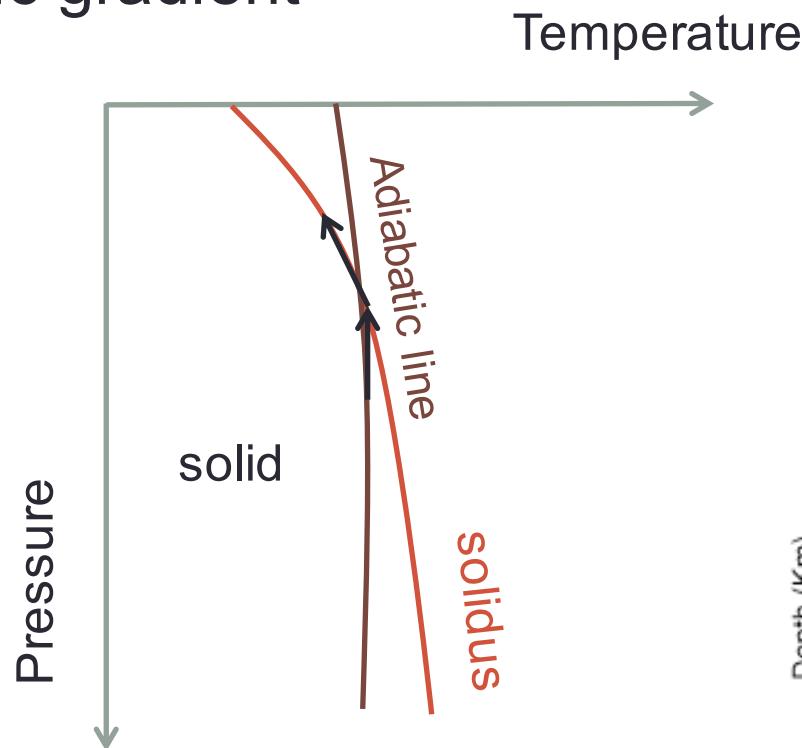
How is melting possible?

- Solidus and geotherm curves

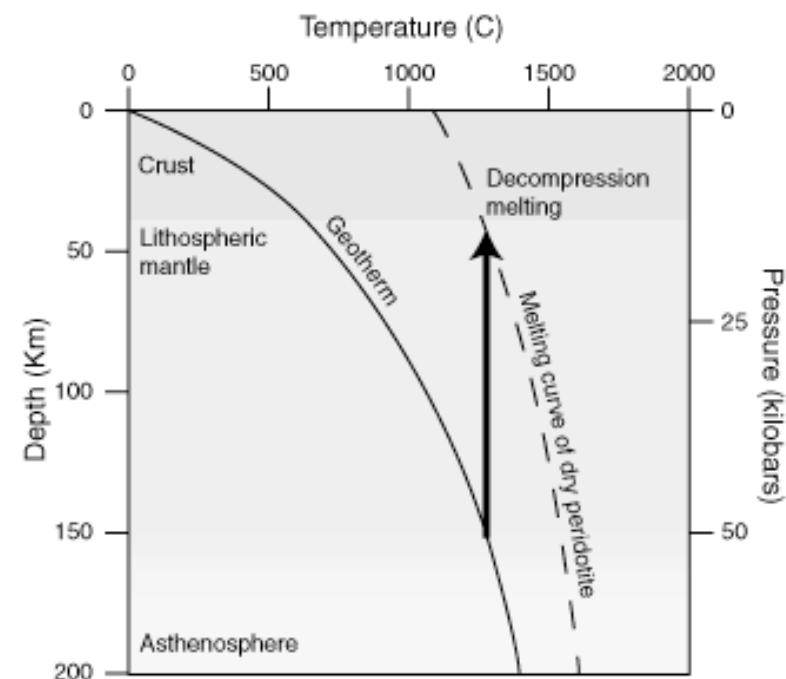


Decompression melting

- The role of convection in the mantle
- Adiabatic gradient

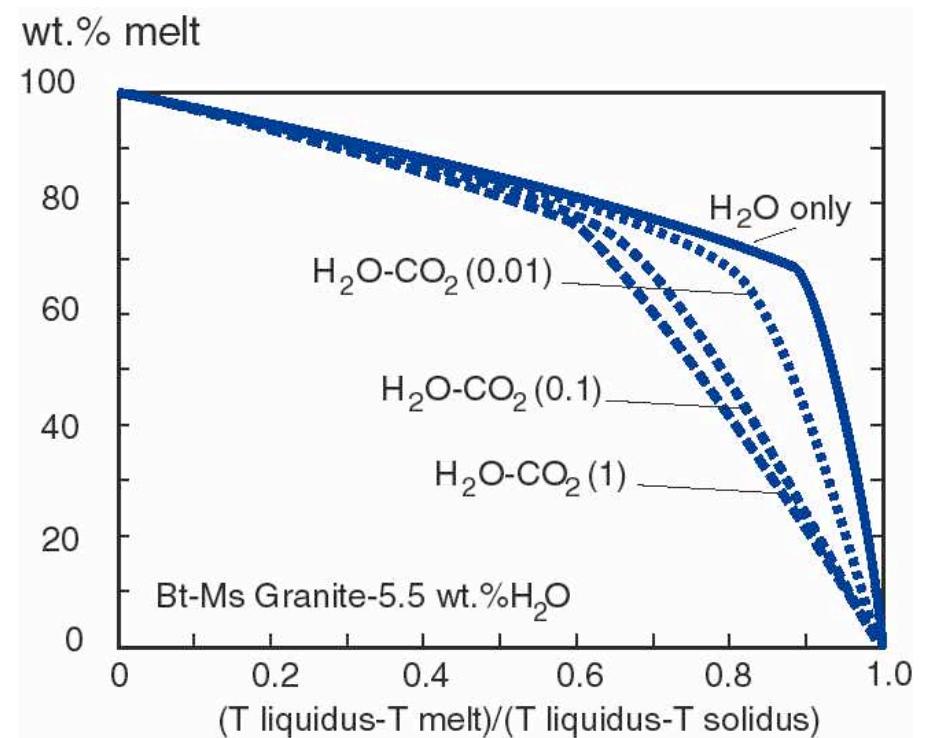
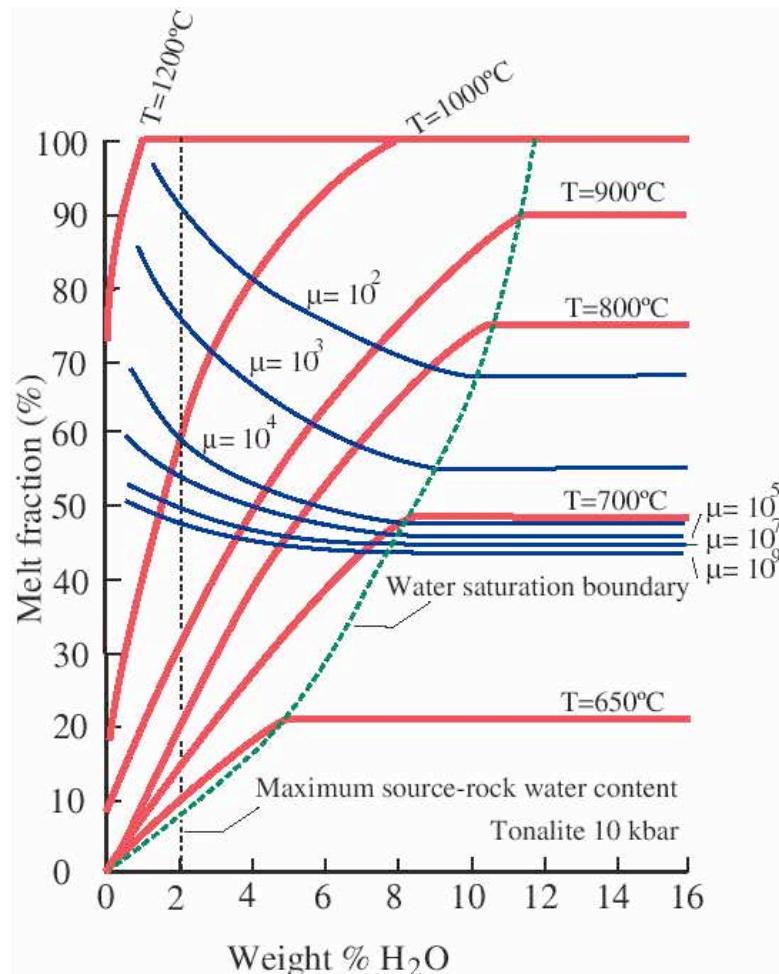


$$\left[\frac{dT}{dP} \right]_{s=0} = \frac{\alpha T}{\rho C_p}$$



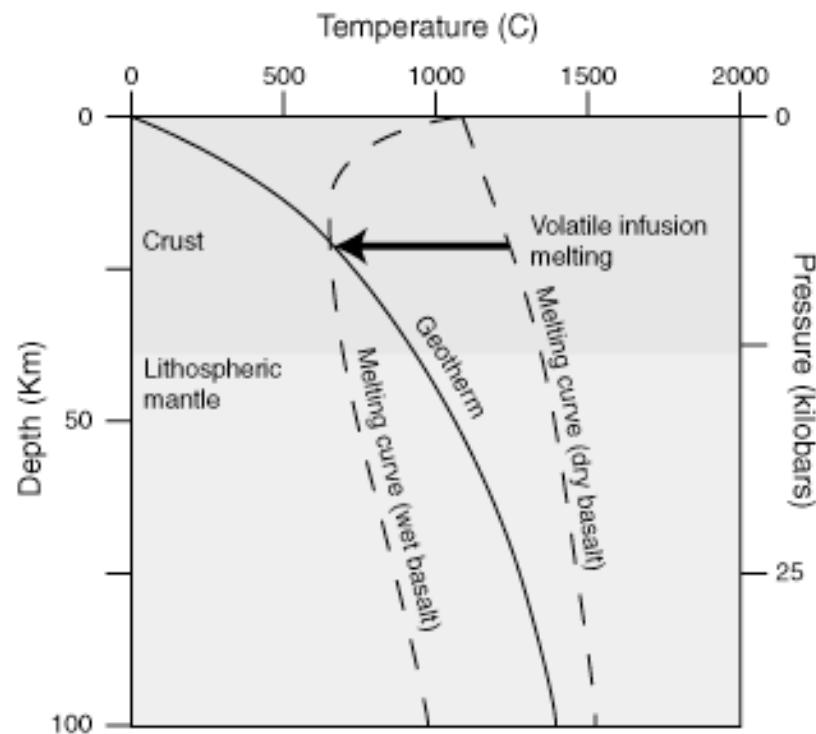
What does adiabatic mean?

Effect of volatiles on melting

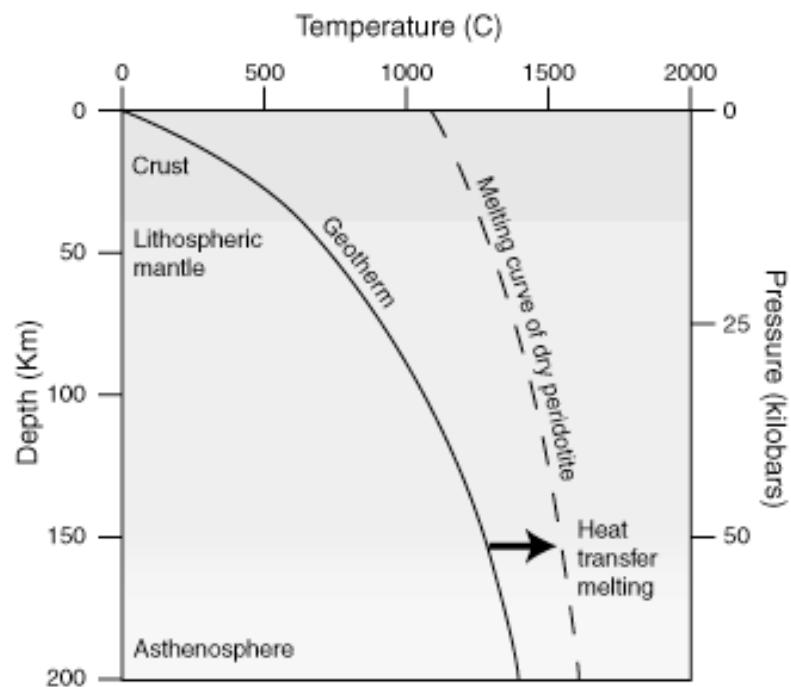


Muscovite bearing granite

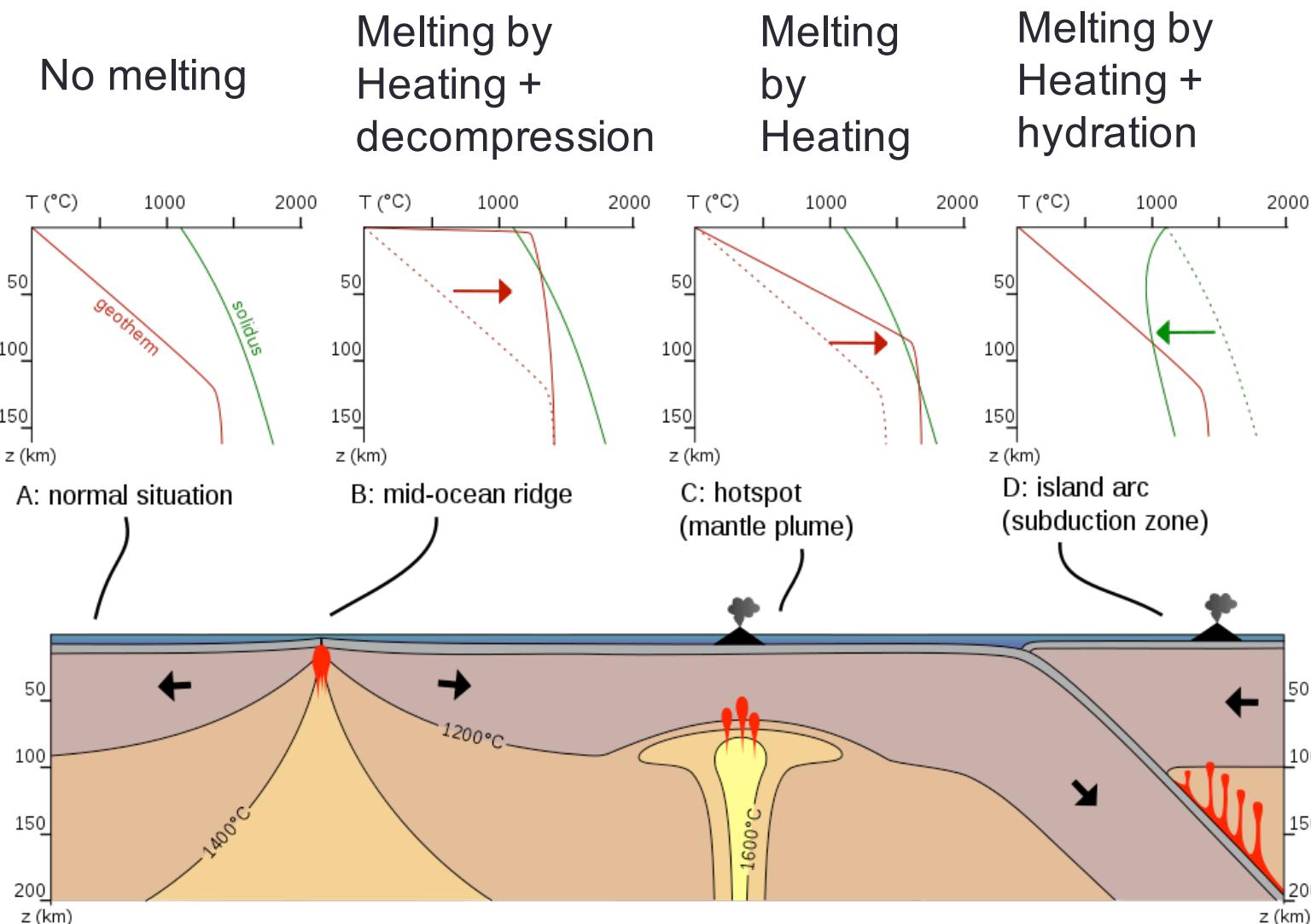
Effect of volatiles on melting



Melting by heat transfer



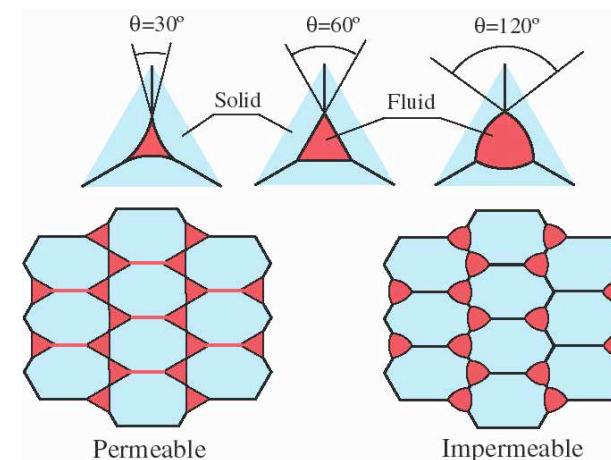
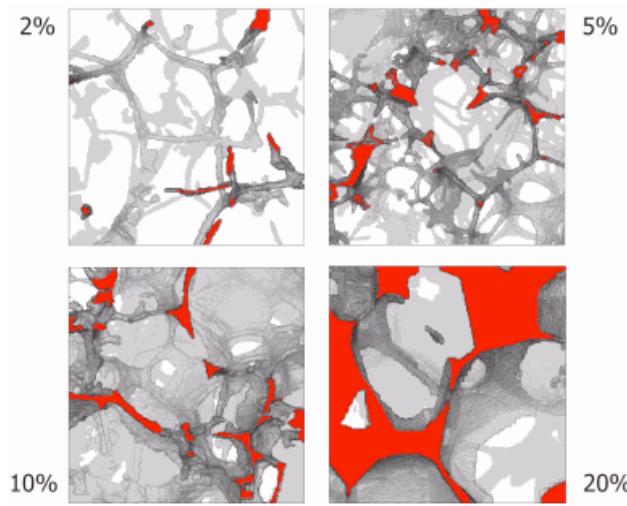
Melting and tectonic settings



Rock melting



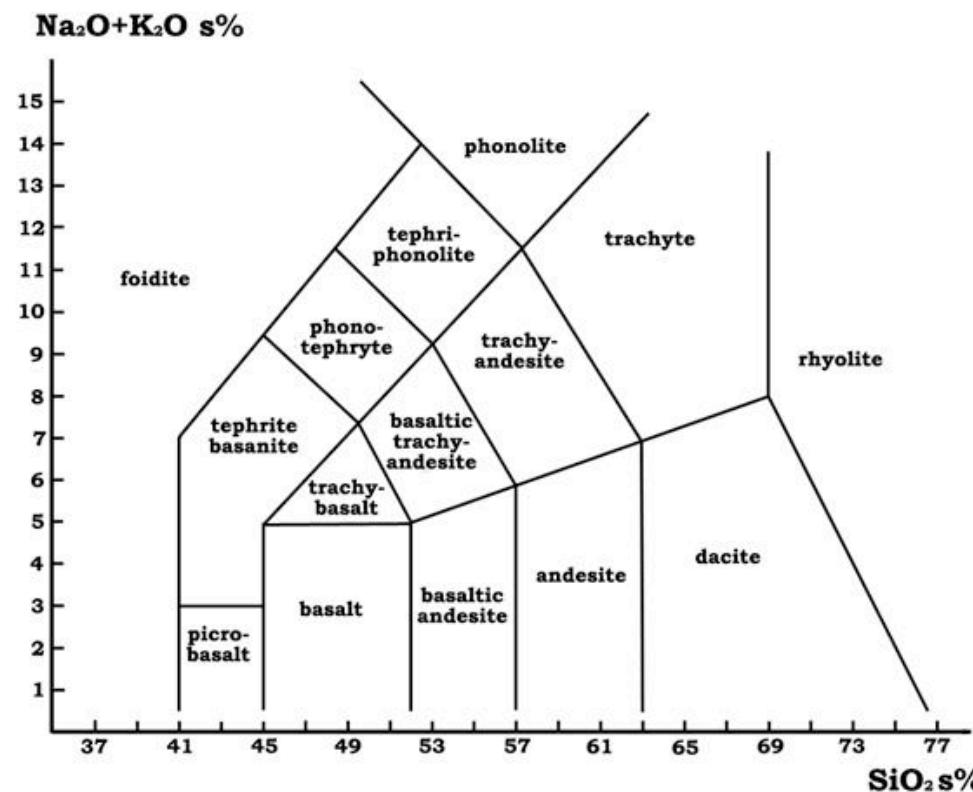
Progressive melting of gneiss with formation of a granitic melt



Melting of mantle rocks (experimental)

Magma properties-composition

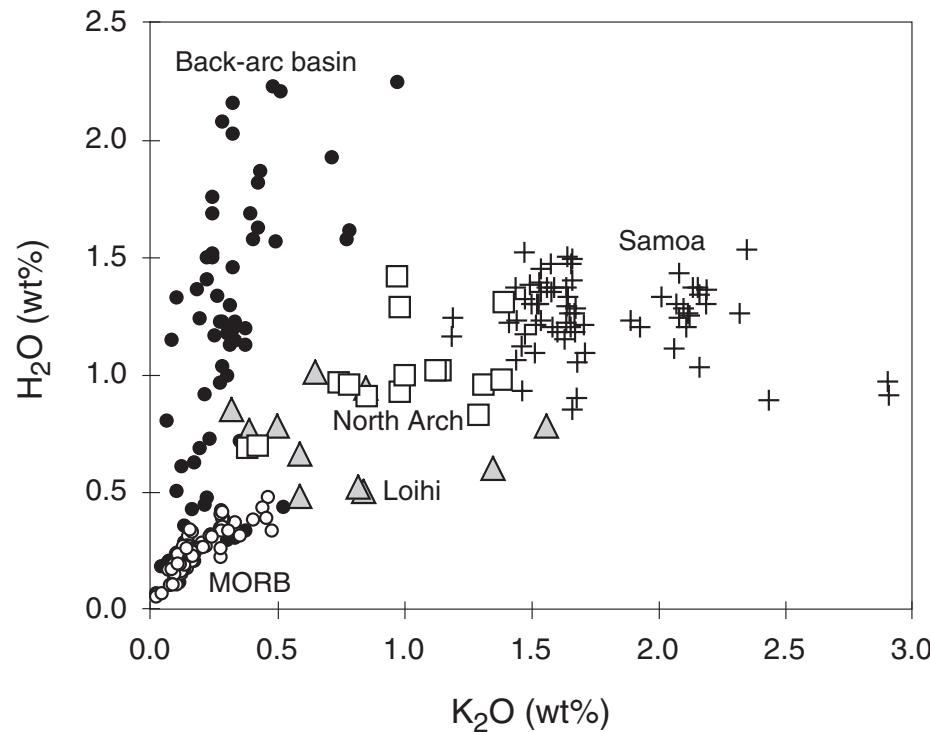
- Composition: Silica ranging from 37 to >70%, alkali content also very variable.



Total Alkali Silica
(TAS) diagram

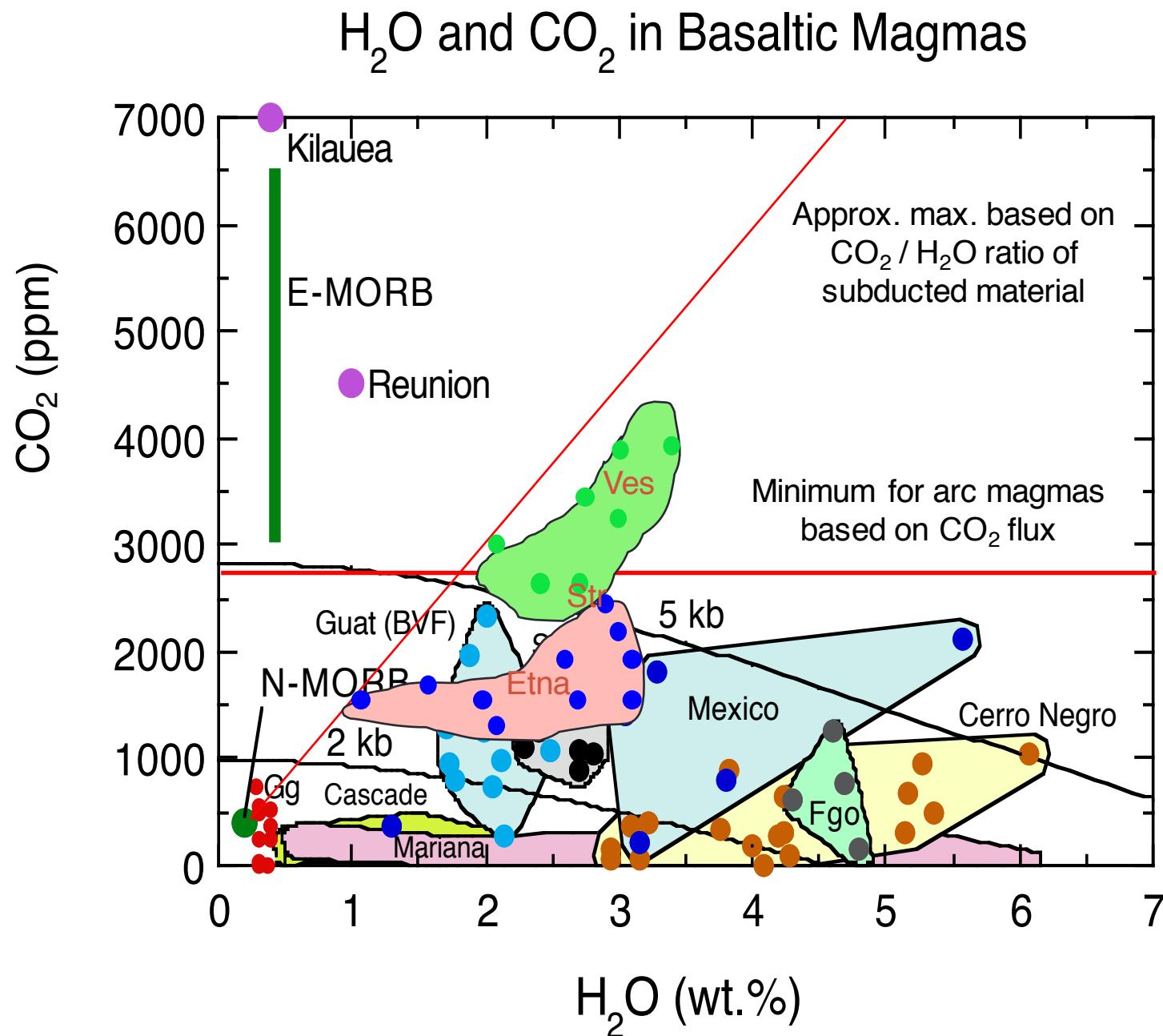
Le Maitre et al. 2002

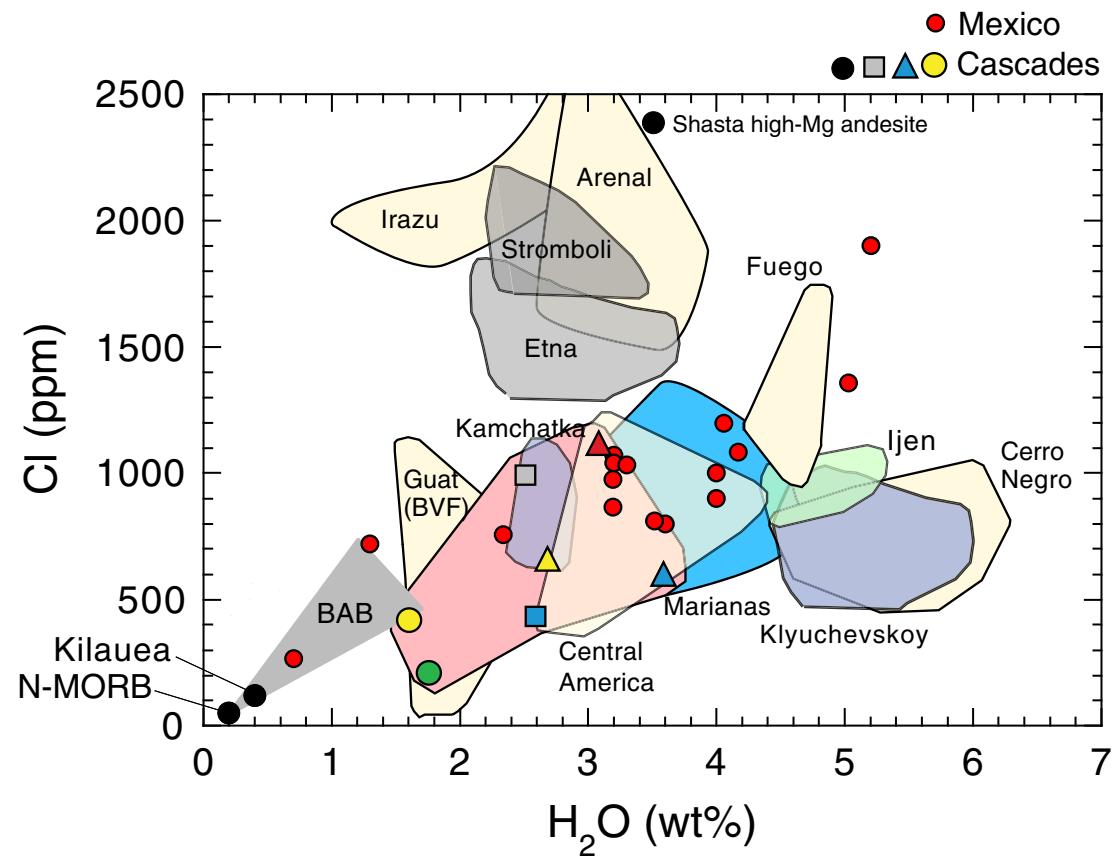
Volatiles in mafic magmas



Wallace et al., 2016

Magma properties-composition



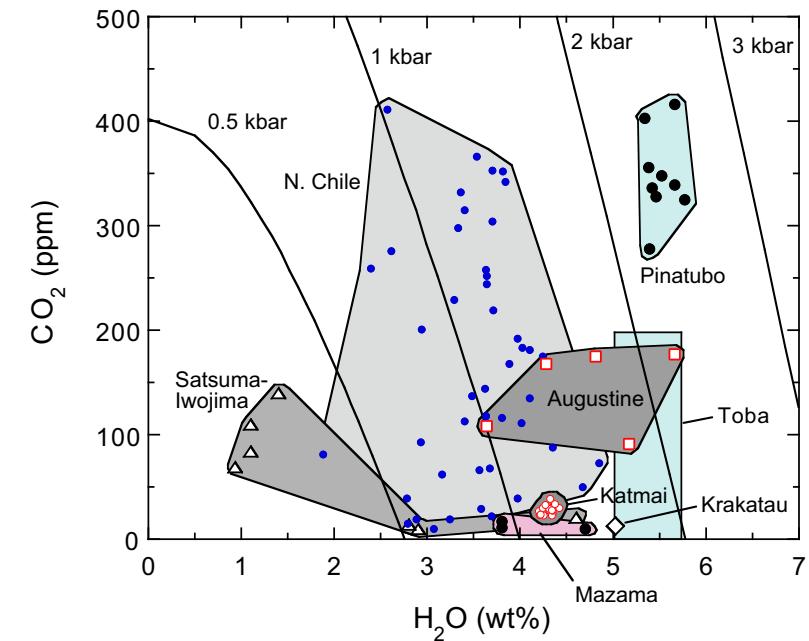
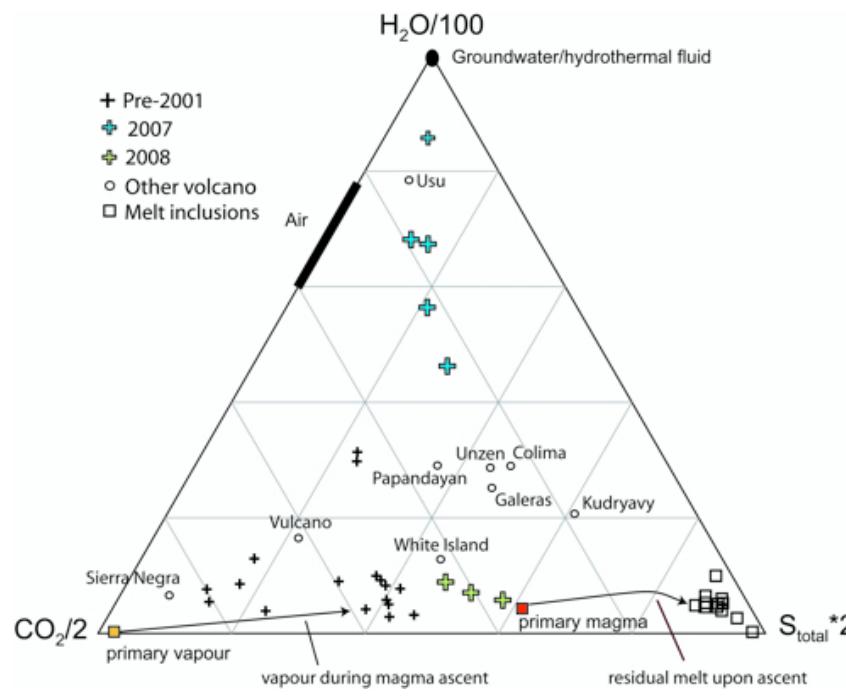


Metrich and Wallace, 2008

Magma properties-composition

- Volatile content

Volatile content in evolved magmas from arc settings



From http://www.sfu.ca/volcanology/research/volatile_history.html

Wallace, 2005

Magma properties- temperature

- Eruptive Temperature:

Basalts: 1200-1050° C

liquidus T: 1200-1250° C

- Solidus T 950-1000° C

- Rhyolites: 800-1000° C

liquidus: 1050° C

Solidus T 650-750° C

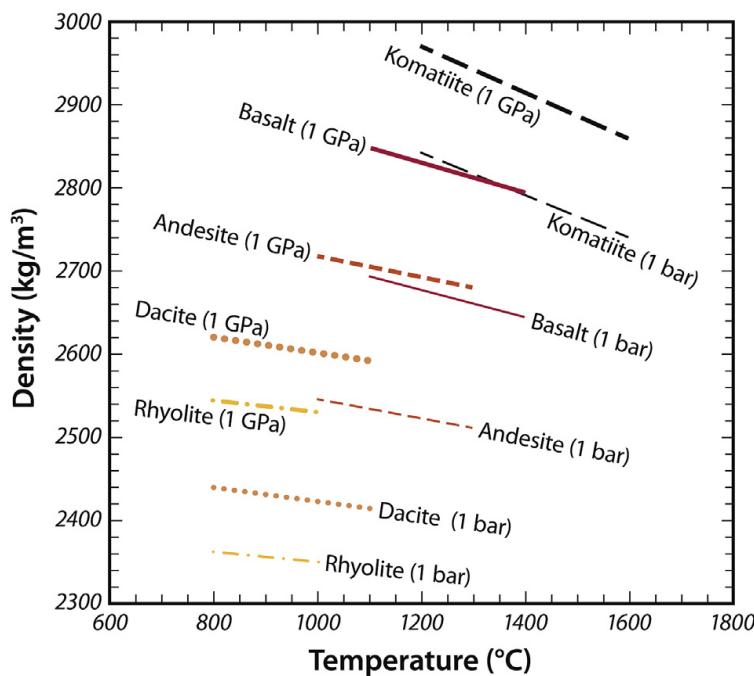
Magma density

It depends on composition

$$\rho = \sum X_i M_i / \sum X_i V_i$$

and on P and T

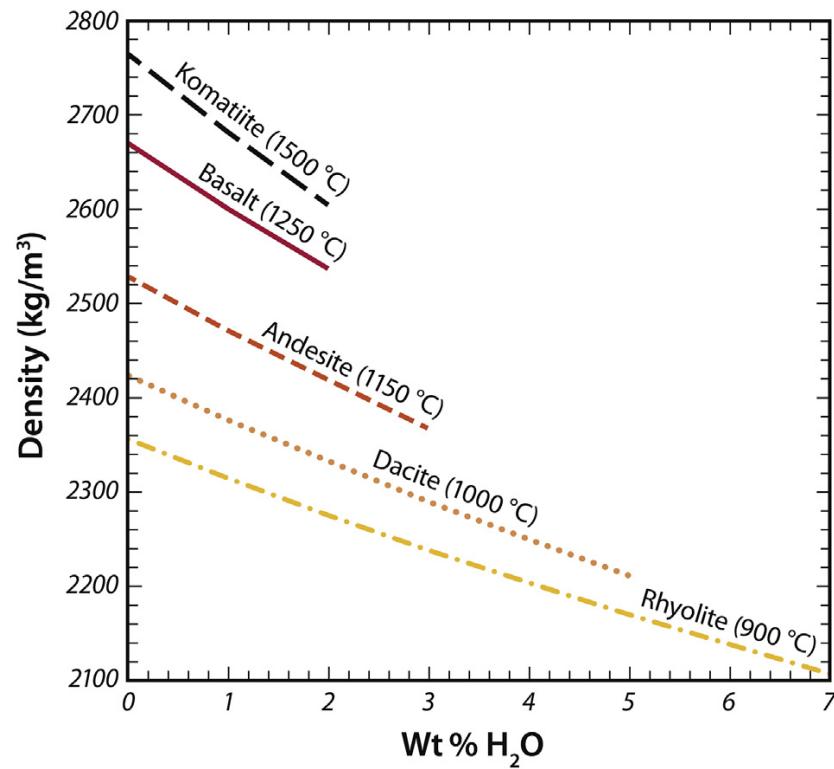
$$V(T, P, X) = \sum X_i \left[V_{i,T_r} + \left(\frac{\partial \bar{V}_i}{\partial T} \right)_P (T - T_r) + \left(\frac{\partial \bar{V}_i}{\partial P} \right)_T (P - P_r) \right]$$



Where T_r and P_r are reference conditions

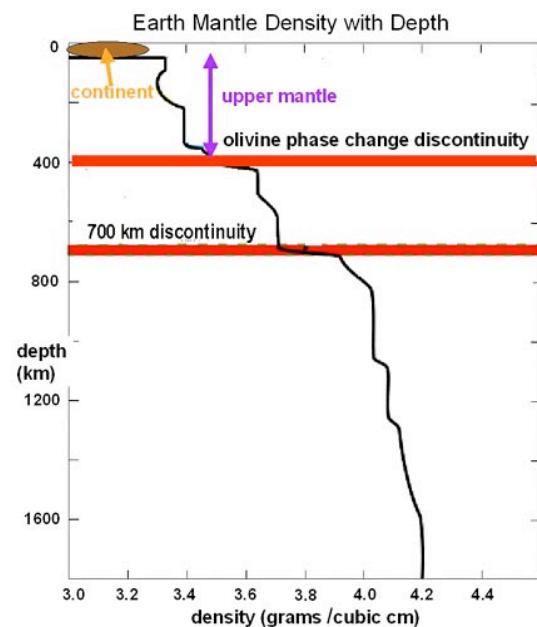
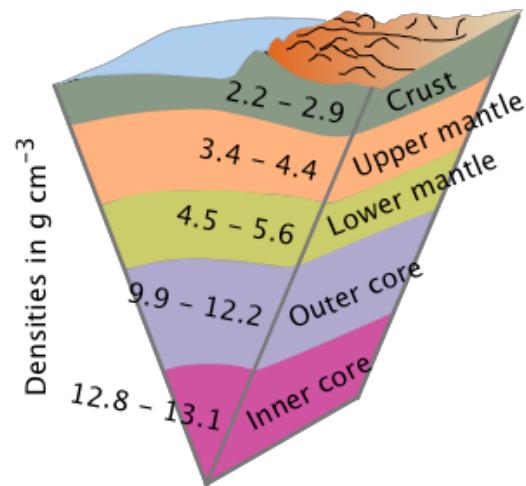
Density at
1 GPa of
pure liquid
magmas

Magma density



Effect of
water
content at
1 bar of
pressure

Magma density



- Why magmas rise up to crustal levels?
- Why magmas are erupted?

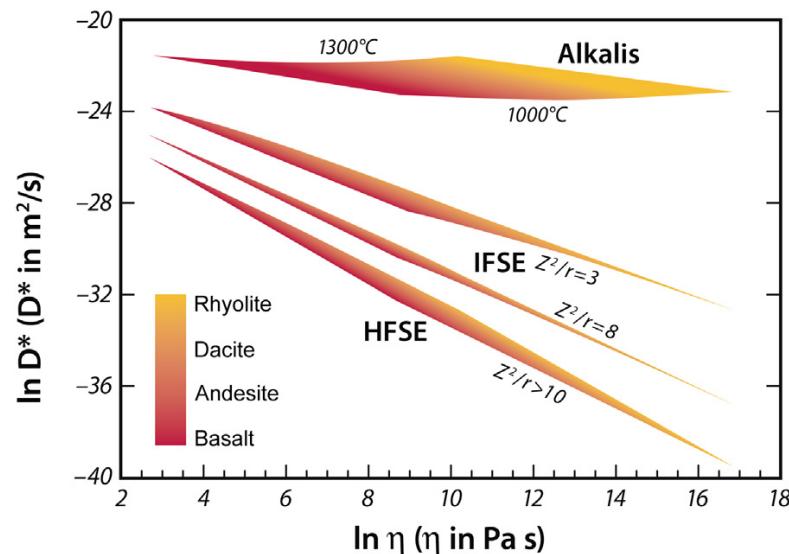
Magma diffusivity

$$\ln D_i = -16.16 + \frac{42.6(r_i - 1.03)^2 + 6.63Y_1 - [1239Y_2 + 27424(r_i - 1.03)^2 + 6975]}{T}$$

where r_i is in Å, $Y_1 = \text{M}/\text{O} = [\text{FeO} + \text{MnO} + \text{MgO} + \text{CaO} + 2(\text{Na}_2\text{O} + \text{K}_2\text{O} + \text{H}_2\text{O})]/[2(\text{SiO}_2 + \text{TiO}_2) + 3\text{Al}_2\text{O}_3 + \text{FeO} + \text{MnO} + \text{MgO} + \text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O} + \text{H}_2\text{O} + 5\text{P}_2\text{O}_5]$,

$$Y_2 = \text{Al}/(\text{Na} + \text{K} + \text{H})$$

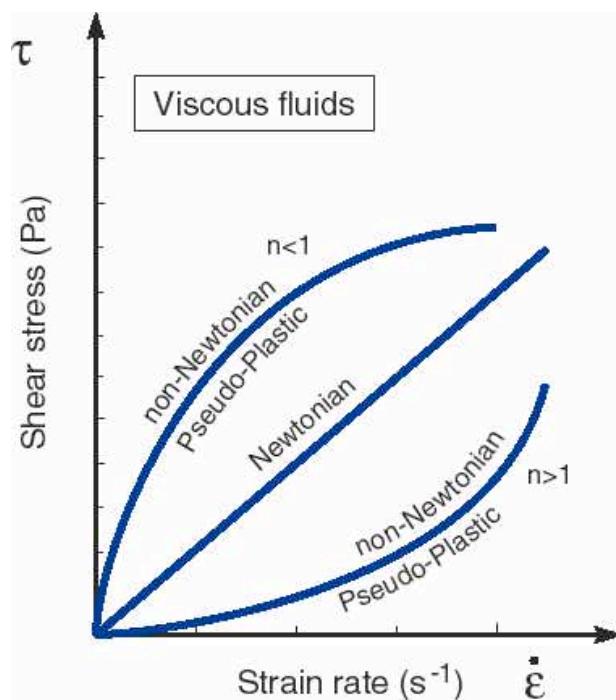
Mungall, 2002 and Zhang, 2010



IFSE= Intermediate field strength elements
HFSE= High field strength elements

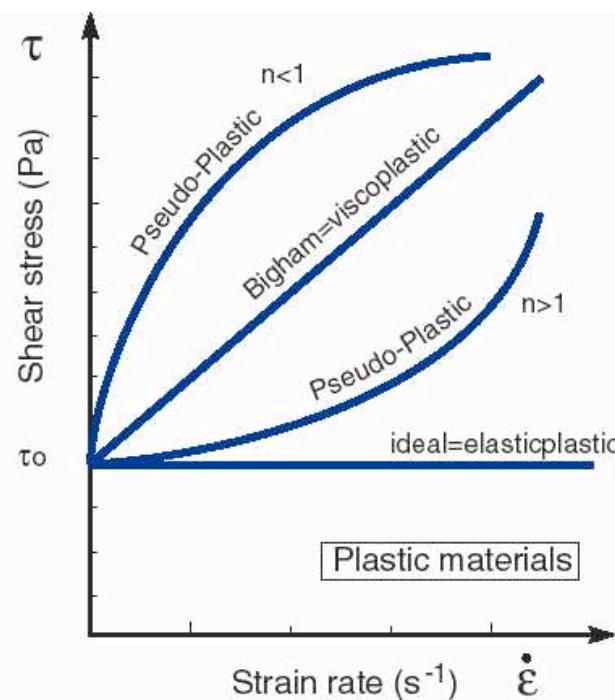
From Lehser and Spera, 2016

Magma rheology



$$\tau = \eta \dot{\varepsilon}$$

$$\tau = A \dot{\varepsilon}^n$$



$$\tau = \tau_0 + \eta \dot{\varepsilon}$$

$$\tau = \tau_0 + A \dot{\varepsilon}^n$$

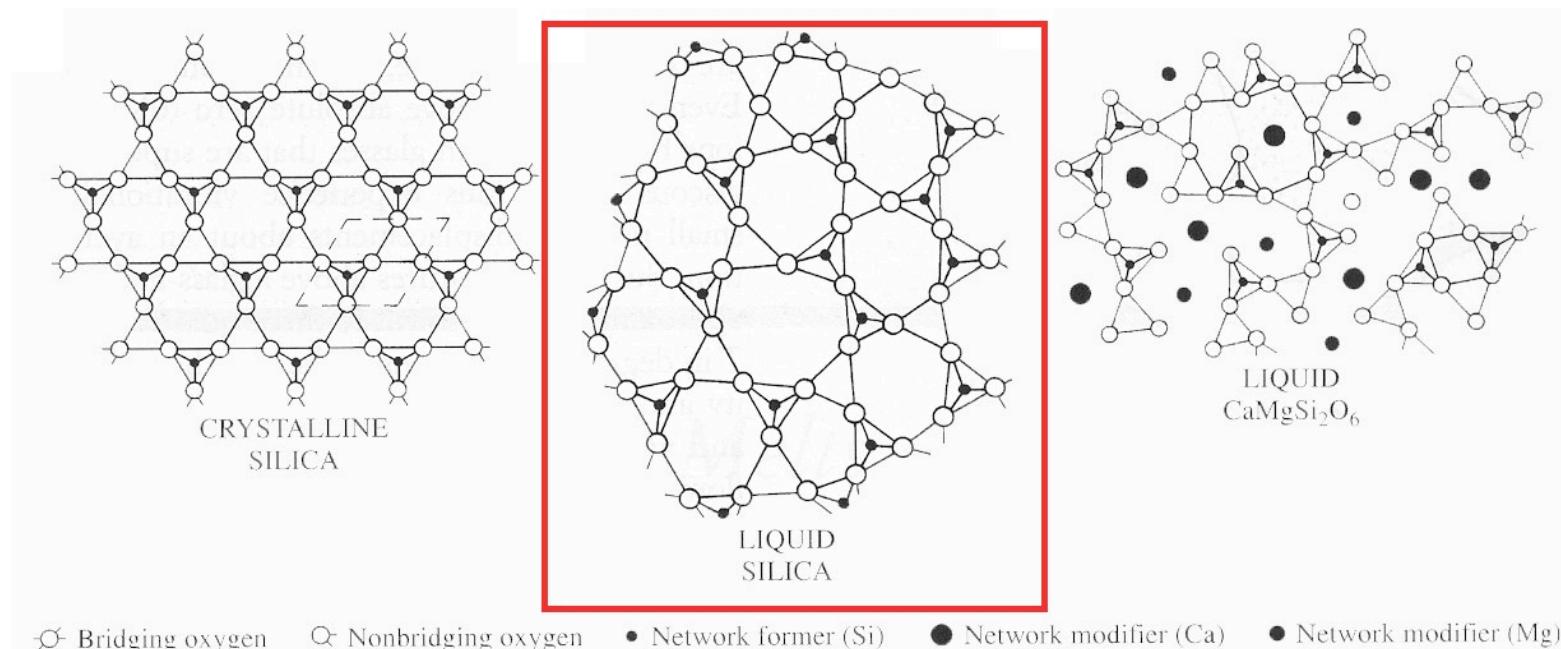
Viscosity is a measure of the resistance of a fluid to flow

Check the online course of P. Rey!

<http://www.geosci.usyd.edu.au/users/prey/Granite/Granite.html#Ch3>

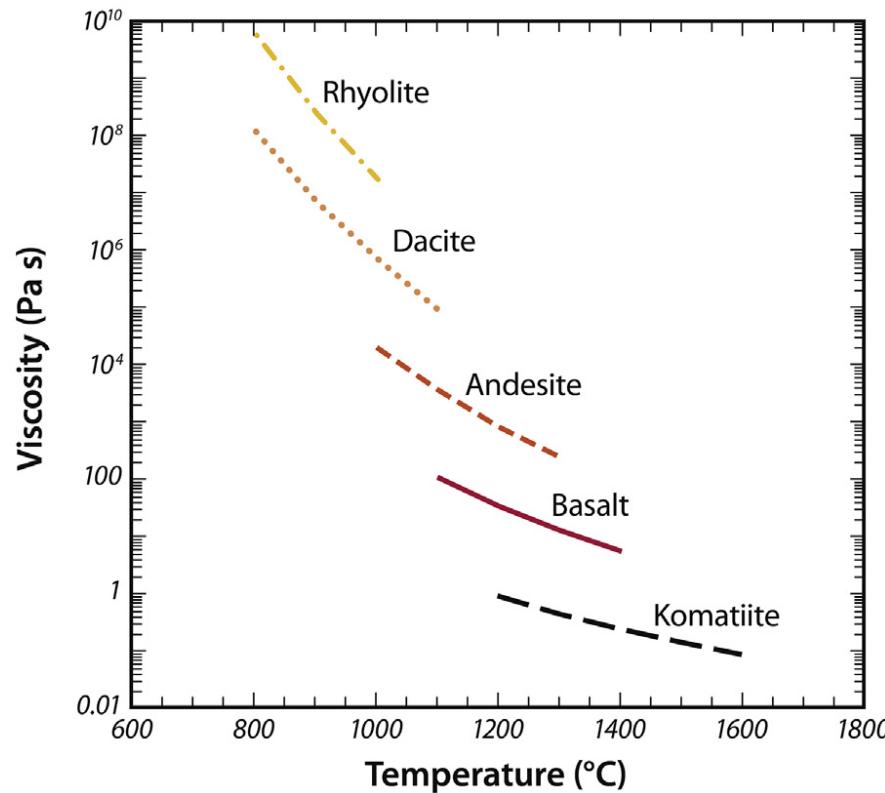
Magma viscosity

- Main controls: temperature (-) silica content (+), water content (-), alkali (-),



From Carmichael et al.

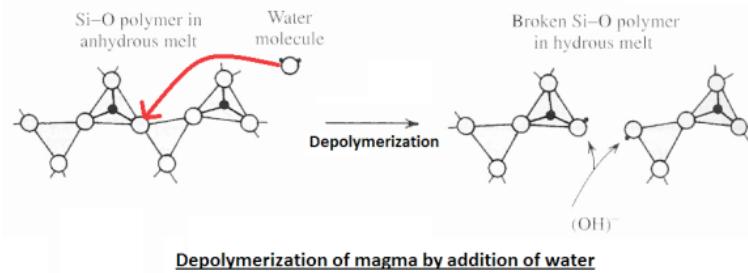
Magma viscosity



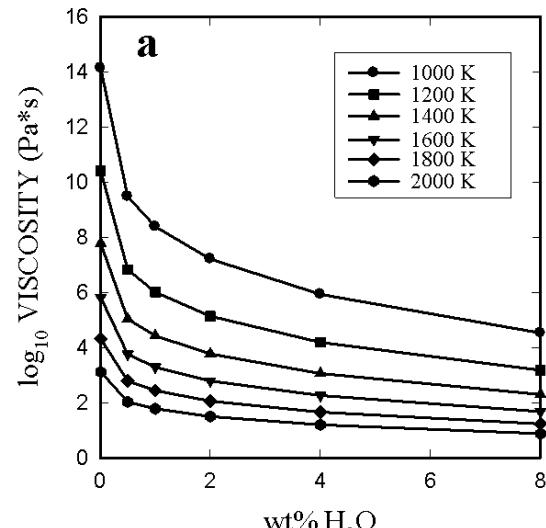
Compositional and
temperature effects

Magma viscosity

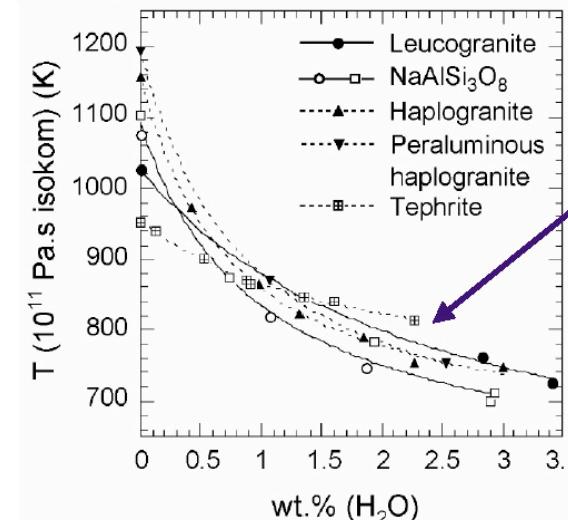
What is the role of water in the chain structure?



$$\eta = (-3.545 + 0.833 \ln(\text{wt \%})) + (9601 - 2368 \ln(\text{wt \%})) / [T - (195.7 + 32.25 \ln(\text{wt \%}))]$$



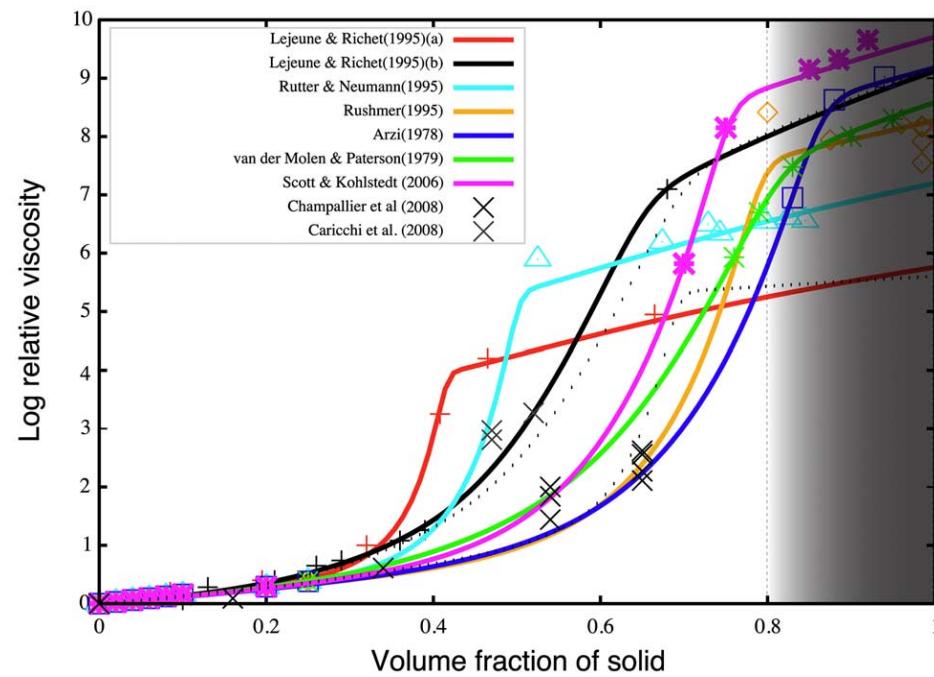
H₂O and different melt compositions



Sources:
Himalayan Leucogranite: Whittington et al. 2004
NaAlSi₃O₈: Whittington et al. 2004, Romano et al. 2001
Haplogranite: Dingwell et al. 1996
Peraluminous haplogranite: Dingwell et al. 1998
Tephrite: Whittington et al. 2000

Magma viscosity

Effect of crystals



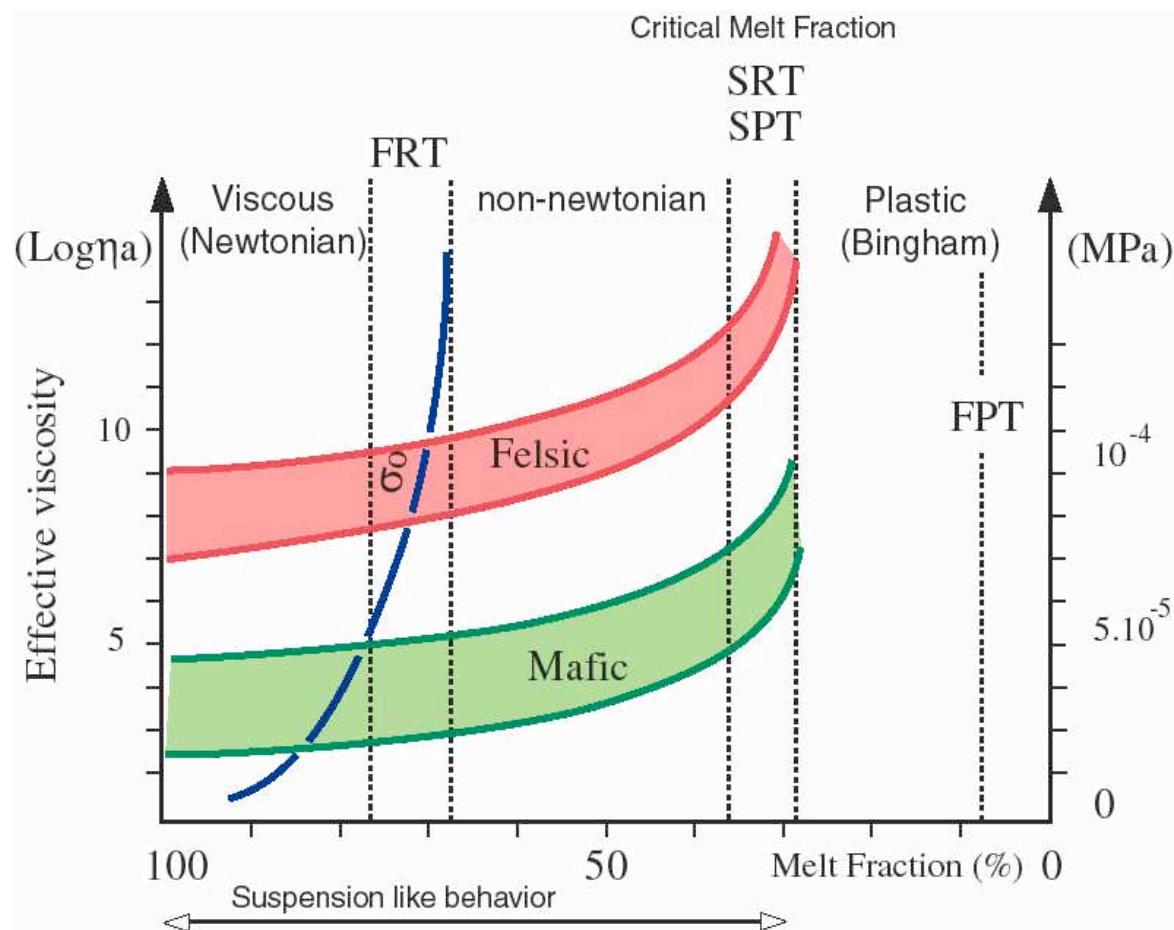
Complex relationships:

Viscosity is affected by
crystal content, shape, size
distribution and strain rate

Costa et al. 2009

Magma viscosity

Effect of crystals



FRT= first rheological threshold
SRT= second rheological threshold

- What does *suspension* mean?

Magma viscosity

Effect of bubbles

First experiments showed that magma viscosity either increased or decreased with increasing bubble volume fraction



Magma viscosity

Effect of bubbles

Capillary number

$$N_{Ca} = \frac{R\eta_c \dot{\gamma}}{\sigma}$$

σ = surface tension

η = viscosity

$\dot{\gamma}$ = strain

R= bubble radius



Bubbles in stretched liquid
sugar



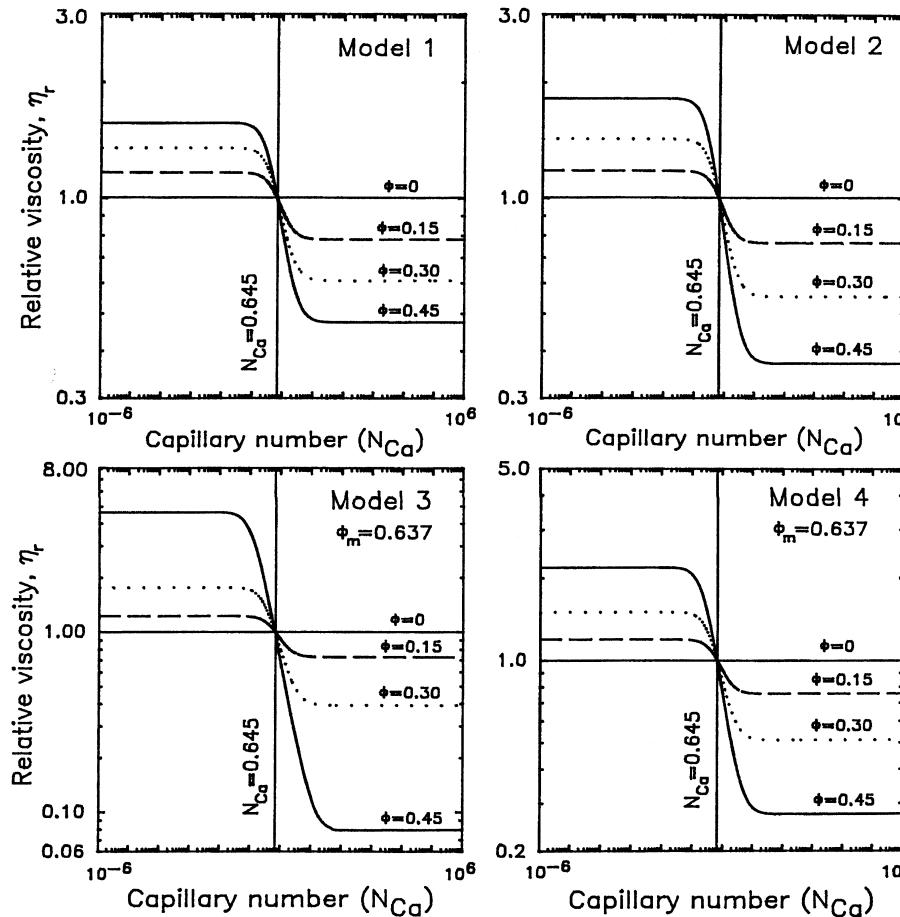
Bubbles in obsidian



Bubbles in water

Magma viscosity

Effect of bubbles



Φ = vesicularity

Pal (2004)

What is the relative viscosity?