





Introduction to modeling Magma generation and properties

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Modelling Volcanic Processes

Course structure:

- 12/11/2020 Eruption characterisation
- 13/11/2020 Magma properties and transport processes
- 26/11/2020 Plume and ballistic modelling
- 27/11/2020 Gravity currents in volcanology

Magma properties and transport processes:

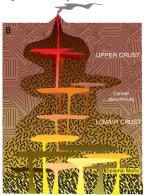
- Introduction to modelling, magma generation and properties
- Magma density and viscosity
- Magma transport processes
- Exercises



Model - Informative representation of an object or system

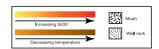
Types of model:

 Conceptual model - Composition of ideas to aid knowledge and understanding



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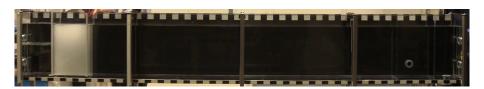


Cashman et al. (2017)



Model - Informative representation of an object or system

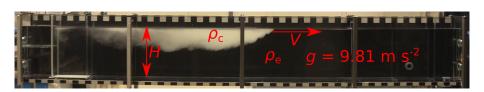
- Conceptual model
- Experimental model



Model - Informative representation of an object or system

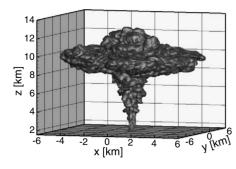
- Conceptual model
- Experimental model
- Mathematical model
 - Analytical model

$$V = \left(\frac{(\rho_{\rm c} - \rho_{\rm e})gH}{2\rho_{\rm c}}\right)^{1/2}$$



Model - Informative representation of an object or system

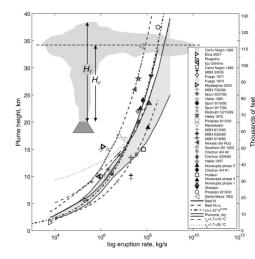
- Conceptual model
- Experimental model
- Mathematical model
 - Analytical model
 - Numerical model



Suzuki et al. (2016)

Model - Informative representation of an object or system

- Conceptual model
- Experimental model
- Mathematical model
 - Analytical model
 - Numerical model
 - Empirical model



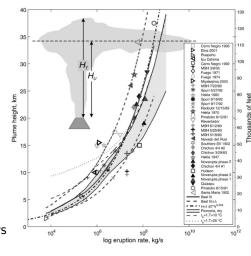
Model - Informative representation of an object or system

Types of model:

- Conceptual model
- Experimental model
- Mathematical model
 - Analytical model
 - Numerical model
 - Empirical model

 $\mathsf{Model} \neq \mathsf{Reality}$ All models contain simplifying assumptions

→ Validity conditions must always be considered



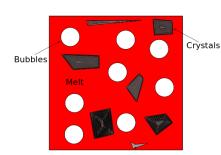
What is magma?

 Naturally occurring molten or partially molten rock material, generated within the Earth and capable of intrusion and extrusion, from which igneous rocks are derived through solidification and related processes. It may or may not contain suspended solids (such as crystals and rock fragments) and/or gas phases.

Magma is

(Glossary of Geology, 2011)

- three-phase:
 - solid unmelted crystals
 - liquid molten rock
 - gas exsolved volatiles
 - multi-component:
 - Many chemical species
 - SiO₂, K₂O, Na₂O, H₂O, etc.



Thermodynamic controls on magma properties

Physical and chemical properites of magma are controlled by three parameters:

- Temperature T
- Pressure P
- Bulk composition $\mathbf{X} = (X_{SiO_2}, X_{K_2O}, X_{Na_2O}, X_{H_2O}, ...)$

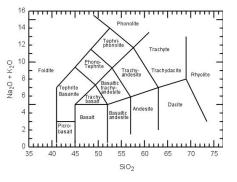
$$\sum_{i} X_{i} = X_{SiO_{2}} + X_{K_{2}O} + X_{Na_{2}O} + X_{H_{2}O} + \dots = 1$$

 X_i = Fraction of the ith component ($i = SiO_2$, K_2O , Na_2O , H_2O , ...) X_i can be expressed in mol% or wt%

Magma composition: Dry classification

Classification normally performed according to a subset of components

Volcanic rocks often classified on a Total Alkali Silica (TAS) diagram



 SiO_2 normally largest component (\sim 37-77) wt%

 $X_{\rm SiO_2}, X_{\rm K_2O}, X_{\rm Na_2O}$ determine composition of many crystals

Not suitable for all volcanic rocks, e.g., High MgO

Le Maitre et al. (2002) Igneous Rock: A Classification and Glossary of Terms

Magma composition: volatile content

Magmas can contain dissolved gas species

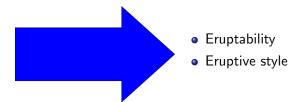
H2O and CO2 are most abundant, then S, Cl and F

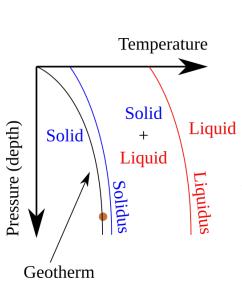
Solubility - Maximum amount of a species that can be dissolved - depends on P, T, \mathbf{X}

Once solubility exceeded, bubbles of exsolved phases form (vesiculation)

Consequnces for:

- Crystallisation
- Density
- Viscosity



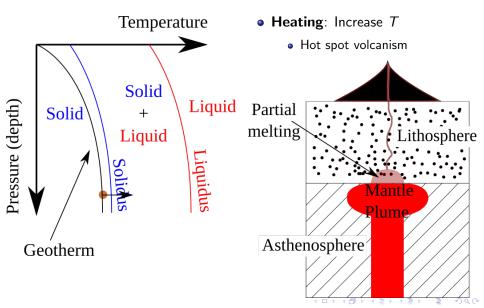


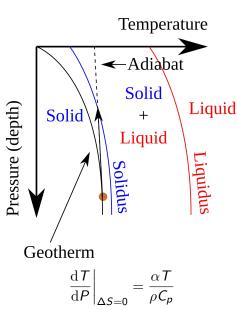
Thermodynamics \implies 3 ways to melt rocks:

- Increase temperature
- Decrease pressure
- Change composition

Can **model** these assuming **Chemical equilibrium** - System has sufficient time to respond to changes in:

- P
- T
- X



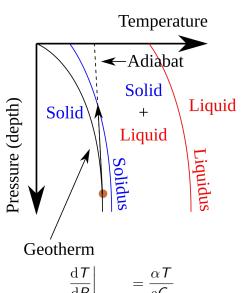


- **Depressurisation**: Reduce *P*
 - Adiabatic process No transfer of heat Q or mass between system and surroundings
 - Adiabatic decompression -Magma rises faster than it can transfer heat to its surroundings

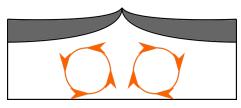
For reversible processes, can define entropy S as:

$$\mathrm{d}S = \frac{\mathrm{d}Q}{T}$$

Thus, adiabatic process is also isentropic (dS = 0)



- **Depressurisation**: Reduce *P*
 - Mid-ocean ridge volcanism

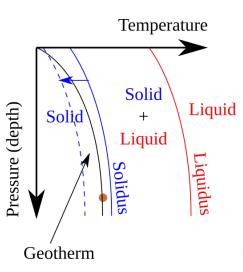


 $\alpha = \mathsf{Thermal} \ \mathsf{expansion} \ \mathsf{coefficient}$

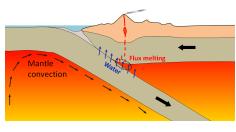
 $\rho = \mathsf{Density}$

 $C_p = \text{Heat capacity}$

 $\Delta S = \mathsf{Change} \; \mathsf{in} \; \mathsf{entropy}$

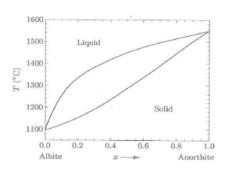


- Compositional change: Change Х
 - Arc volcanism



Decreasing X_{H_2O} shifts solidus to smaller T

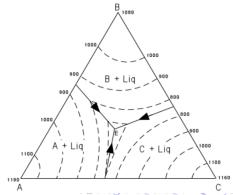
Phase diagrams



Three parameters can be considered on a ternary phase diagram

All diagrams are just *slices* of the full picture

State of magma depends on many parameters (T, P, X_i) Can consider two parameters on a simple **binary** phase diagram Determine compositions and relative proportions of different phases



Summary

- Magma properties controlled by P, T and X
- Models Representations of reality
- Models can describe:
 - Processes, e.g., magma generation by:
 - Heating
 - Decompression
 - Mixing
 - Characteristics
 - State of matter
 - Mineralogy

