

# Quiz 1

## Chemistry 3BB3; Winter 2005

Name/PID:

1. Write the Schrödinger Equation for the Hydrogen atom in atomic units. You may use the Born-Oppenheimer approximation.
2. In order to solve the Hydrogen atom, we used the solution of the following exactly solvable system
  - (a) particle in a box
  - (b) harmonic oscillator
  - (c) rigid rotor
  - (d) Hückel Hamiltonian
3. Which of the following phrases correctly describe the Born-Oppenheimer approximation (there may be more than one correct answer)
  - (a) The electronic state is perfectly coupled to the nuclear positions and is independent of the motion of the atomic nuclei
  - (b) The nuclear state is perfectly coupled to the electronic positions and is independent of the motion of the electrons.
  - (c) The nuclei move independently of the electronic motions.
  - (d) The electronic wave function does not depend on the nuclear positions.
  - (e) There is no transfer of momentum between nuclei and electrons.
4. The mass of the antimuon,  $\mu^+$ , is about  $\frac{1}{9}$ <sup>th</sup> the mass of the proton. The Born-Oppenheimer approximation will be more accurate for:
  - (a) A system that consists of antimuons and electrons.
  - (b) A system that consists of protons and electrons.
5. Which of the following is *not* an approximations used in the electronic Hamiltonian we wrote in class?
  - (a) The effects of relativity are ignored.
  - (b) Nuclear forces are ignored.
  - (c) Interactions between electrons are ignored altogether.
  - (d) Atomic nuclei are assumed to be a point charges.
  - (e) The effects of gravity are ignored.

# Quiz 1 Key

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Name/PID:

1. Write the Schrödinger Equation for the Hydrogen atom in atomic units. You may use the Born-Oppenheimer approximation.

$$\left( \frac{-\nabla_r^2}{2} - \frac{Z}{r} \right) \Psi(r) = E \Psi(r) \quad \text{for Hydrogen, } Z = 1.$$

2. In order to solve the Hydrogen atom, we used the solution of the following exactly solvable system

- |                         |                        |
|-------------------------|------------------------|
| (a) particle in a box   | (c) rigid rotor        |
| (b) harmonic oscillator | (d) Hückel Hamiltonian |

3. Which of the following phrases correctly describe the Born-Oppenheimer approximation (there may be more than one correct answer)

- (a) The electronic state is perfectly coupled to the nuclear positions and is independent of the motion of the atomic nuclei
- (b) The nuclear state is perfectly coupled to the electronic positions and is independent of the motion of the electrons.
- (c) The nuclei move independently of the electronic motions.
- (d) The electronic wave function does not depend on the nuclear positions.
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4. The mass of the antimuon,  $\mu^+$ , is about  $\frac{1}{9}$ <sup>th</sup> the mass of the proton. The Born-Oppenheimer approximation will be more accurate for:

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5. Which of the following is *not* an approximations used in the electronic Hamiltonian we wrote in class?

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- (b) Nuclear forces are ignored.
- (c) Interactions between electrons are ignored altogether.
- (d) Atomic nuclei are assumed to be a point charges.
- (e) The effects of gravity are ignored.