

Quantum Mechanics and Spectroscopy

CHEM 3PA3

Assignment 2

Name: _____

1. For the photoelectric effect, the work function is the minimum energy required to eject an electron. The work function for potassium is 3.69×10^{-19} J. If potassium is irradiated with radiation of 400 nm, determine the maximum kinetic energy that the ejected electrons can reach.
2. In the particle in a one dimensional box model, what is the probability of finding the particle in $a/4 \leq x \leq 3a/4$?
3. Show that the eigenvalues of a Hermitian operator are real.
4. Some textbooks define that a Hermitian operator is one that satisfies $\int \Psi_1^*(x) \hat{A} \Psi_2(x) dx = \int \Psi_2(x) (\hat{A} \Psi_1(x))^* dx$. Is this equivalent to $\int \Psi^*(x) \hat{A} \Psi(x) dx = \int (\hat{A} \Psi(x))^* \Psi(x) dx$?
5. For $\hat{A} = x^2$ and $\hat{B} = \frac{d}{dx}$, show that $\hat{A}(\hat{B}f(x)) \neq \hat{B}(\hat{A}f(x))$.
6. If a wavefunction $\Psi(x)$ is solution of the operator \hat{A} , show that a constant times the wavefunction is also a solution of \hat{A} .