## Quiz 6

### Chemistry 3BB3; Winter 2005

1. The operator for x-component of the dipole moment for a P-atom N-electron molecule is defined as

$$\hat{\mu}_r =$$

- 2. Write down Fermi's golden rule for dipole transitions. You can leave out constants of proportionality (just in case you do not know them yet).
- 3. The sinc function is defined as
- (a) sinc  $x \equiv x \cdot \sin x$
- (e)  $\sin x \equiv \sin^2 x \cos x$
- (i) sinc  $x \equiv x \arcsin x$

- (b) sinc  $x \equiv \sin x e^x$
- (f)  $\operatorname{sinc} x \equiv x^2 \sin^2 x$
- (j) sinc  $x \equiv x^2 \arcsin x$ (k) sinc  $x = x \sin x^{-1}$

- (c) sinc  $x \equiv x^{-1} \cdot \sin x$ (d) sinc  $x \equiv \sin x \cos x$
- (g) sinc  $x \equiv \sin^2 x \ e^x$ (h) sinc  $x \equiv x^{-1} \cdot \sin^2 x$
- (1) sinc  $x = x \sin x^{-1}$
- 4. What is the relationship between the frequency of absorbed radiation and the difference in energy between the initial and final states?
- 5. Write down the time-dependent Schrödinger Equation.
- 6. The long-time approximation led to the conclusion that:
  - (a) In the absence of broadening, the absorption spectrum of a molecule is a collection of deltafunction peaks.
  - (b) Electric quadrupole-allowed and magnetic dipole allowed transitions have low intensity.
  - (c) Using time-dependent perturbation theory to model the interaction between light and the molecule is justified.
  - (d) Transitions which change the multiplicity of the electronic state (that is, transitions which change the spin of one or more electrons) are forbidden.
  - (e) Transitions that correspond to double-excitations in the orbital picture are forbidden in the orbital picture and (usually) low in intensity.
- 7. You put a molecule in the presence of light. The electric field oscillates in the z-direction and the magnetic field oscillates in the x-direction. Which axis is associated with the direction of propagation of the light?
  - (a) x -axis
- (b) y -axis

(c) z -axis

Name:

8.	The momentum of a photon can be expressed in terms of the wave number, k.	What is this
	relationship?	

- 9. The energy of a photon can be expressed in terms of the wave number, k. What is this relationship?
- 10. The period of light can be expressed in terms of its wave number, k. What is this relationship?

# Quiz 6 (Key)

### Chemistry 3BB3; Winter 2005

1. The operator for x-component of the dipole moment for a P-atom N-electron molecule is defined as

$$\hat{\mu}_{\boldsymbol{x}} = \sum_{i=1}^{N} -e x_{i} + \sum_{\alpha=1}^{P} Z_{\alpha} e X_{\alpha}$$

2. Write down Fermi's golden rule for dipole transitions. You can leave out constants of proportionality (just in case you do not know them yet).

$$W_{\!\scriptscriptstyle fi} = \frac{2\pi V^2 \ g \ \hbar \omega_{\!\scriptscriptstyle fi} \ + g \ - \! \hbar \omega_{\!\scriptscriptstyle fi}}{\hbar} \Big| \! \left\langle \Phi_{\!\scriptscriptstyle f} \left| \hat{\mu} \right| \Phi_{\!\scriptscriptstyle i} \right\rangle \! \Big|^2$$

- 3. The sinc function is defined as
- (a)  $\sin x \equiv x \cdot \sin x$
- (e)  $\sin c \ x \equiv \sin^2 x \cos x$  (i)  $\sin c \ x \equiv x \arcsin x$  (f)  $\sin c \ x \equiv x^2 \sin^2 x$  (j)  $\sin c \ x \equiv x^2 \arcsin x$  (g)  $\sin c \ x \equiv \sin^2 x \ e^x$  (k)  $\sin c \ x = x \sin x$  (l)  $\sin c \ x = x \sin x$

- (b) sinc  $x \equiv \sin x e^x$

- (c) sinc  $x \equiv x^{-1} \cdot \sin x$

- (d)  $\sin x \equiv \sin x \cos x$

- 4. What is the relationship between the frequency of absorbed radiation and the difference in energy between the initial and final states?

$$\nu_{\scriptscriptstyle fi} = \frac{E_{\scriptscriptstyle f} - E_{\scriptscriptstyle i}}{h}$$

5. Write down the time-dependent Schrödinger Equation.

$$\hat{H}\Psi = i\hbar \frac{\partial \Psi}{\partial t}$$

- 6. The long-time approximation led to the conclusion that:
  - In the absence of broadening, the absorption spectrum of a molecule is a collection of delta-function peaks.
  - Electric quadrupole-allowed and magnetic dipole allowed transitions have low intensity. (b)
  - Using time-dependent perturbation theory to model the interaction between light and the molecule is justified. (c)
  - Transitions which change the multiplicity of the electronic state (that is, transitions which change the spin of one or more electrons) are forbidden.
  - Transitions that correspond to double-excitations in the orbital picture are forbidden in the orbital picture and (e) (usually) low in intensity.
- 7. You put a molecule in the presence of light. The electric field oscillates in the z-direction. and the magnetic field oscillates in the x-direction. Which axis is associated with the direction of propagation of the light?
  - (a) x -axis

(b) y -axis

(c) z -axis

### Name:

8. The momentum of a photon can be expressed in terms of the wave number, k. What is this relationship?

$$p = \hbar k$$

9. The energy of a photon can be expressed in terms of the wave number, k. What is this relationship?

$$E=\hbar kc$$

10. The period of light can be expressed in terms of its wave number, k. What is this relationship?

$$T = \frac{2\pi}{kc}$$