

Student Number: _____

Name: _____

Quiz 6

FYI: $h = 6.626 \cdot 10^{-34} \text{ J} \cdot \text{s}$; $\hbar = 1.055 \cdot 10^{-34} \text{ J} \cdot \text{s}$; $1 \text{ u} = 1.66 \cdot 10^{-27} \text{ kg}$; $c = 2.998 \cdot 10^8 \frac{\text{m}}{\text{s}} = 2.998 \cdot 10^{10} \frac{\text{cm}}{\text{s}}$

1. The equilibrium bond length for the hydrogen molecule is $.7416 \cdot 10^{-10} \text{ m}$. **What is the frequency, expressed in wavenumbers ($\bar{\nu} = 1/\lambda$; give your answer in units of cm^{-1}) associated with the lowest-energy rotational transition, $J = 0 \rightarrow J = 1$?** Assume the rigid rotor approximation is acceptable.

2,3. A photon has a wavenumber of $k = \frac{2\pi}{\lambda} = 10^7 \text{ m}^{-1}$.

(a) What is the momentum of the photon?

(b) What is the energy of the photon?

4. Show that the eigenvalues of a Hermitian operator are always real.

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1. The equilibrium bond length for the hydrogen molecule is $.7416 \cdot 10^{-10} \text{ m}$. What is the frequency, expressed in wavenumbers ($\bar{\nu} = 1/\lambda$; give your answer in units of cm^{-1}) associated with the lowest-energy rotational transition, $J = 0 \rightarrow J = 1$? Assume the rigid rotor approximation is acceptable.

The rotational levels are given by $E_{\text{rot}}(J) = \frac{\hbar^2 J(J+1)}{2\mu r_e^2}$. So the change in rotational energy is

$$\Delta E = E_{\text{rot}}(J=1) - E_{\text{rot}}(J=0) = E_{\text{rot}}(J=1) = \frac{\hbar^2 \cdot 2}{2\mu r_e^2} = \frac{\hbar^2}{\mu r_e^2}$$

Using the equation for the change in energy in terms of wavenumber,

$$\Delta E = hc\bar{\nu}$$

$$\begin{aligned} \bar{\nu} &= \frac{\hbar^2}{hc\mu r_e^2} = \frac{h^2}{(2\pi)^2 hc\mu r_e^2} = \frac{h}{(2\pi)^2 c\mu r_e^2} \\ &= \frac{6.626 \cdot 10^{-34} \text{ kg m}^2 \text{ s}^{-1}}{(2\pi)^2 \left(\frac{1 \text{ u} \cdot 1 \text{ u}}{1 \text{ u} + 1 \text{ u}} \right) (1.66 \cdot 10^{-27} \text{ kg}) (2.998 \cdot 10^{10} \frac{\text{cm}}{\text{s}}) (.7416 \cdot 10^{-10} \text{ m})^2} \\ &= 122.6 \text{ cm}^{-1} \end{aligned}$$

2. A photon has a wavenumber of 10^7 m^{-1} .

- (a) What is the momentum of the photon?

$$p = \hbar k = 1.055 \cdot 10^{-27} \frac{\text{kg} \cdot \text{m}}{\text{s}}$$

- (b) What is the energy of the photon?

$$\begin{aligned} E = h\nu &= \frac{hc}{\lambda} = pc = 1.055 \cdot 10^{-27} \frac{\text{kg} \cdot \text{m}}{\text{s}} \cdot 2.998 \cdot 10^8 \frac{\text{m}}{\text{s}} \\ &= 3.163 \cdot 10^{-19} \frac{\text{kg} \cdot \text{m}^2}{\text{s}^2} \end{aligned}$$

4. Show that the eigenvalues of a Hermitian operator are always real.

Let \hat{C} be a Hermitian operator. Then, by definition,

$$\int (\hat{C}\Psi_1(x))^* \Psi_2(x) dx = \int \Psi_1(x) \hat{C}\Psi_2(x) dx$$

Furthermore, let γ_k and $\psi_k(x)$ be eigenfunctions of \hat{C} . Then, by definition,

$$\hat{C}\psi_k(x) = \gamma_k \psi_k(x)$$

Taking the complex conjugate of both sides of this equation, we have

$$(\hat{C}\psi_k(x))^* = \gamma_k^* \psi_k^*(x)$$

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Multiply the first equation on both sides by $\psi_k^*(x)$ and integrate; multiply the second equation on both sides by $\psi_k(x)$ and integrate. (The multiplication sign is used to indicate which side of the expression one multiplies on.) So we have:

$$\int \psi_k^*(x) \hat{C} \psi_k(x) dx = \gamma_k \int \psi_k^*(x) \psi_k(x) dx$$

$$\int (\hat{C} \psi_k(x))^* \psi_k(x) dx = \gamma_k^* \int \psi_k^*(x) \psi_k(x) dx$$

Rearranging, we can rewrite this as:

$$\frac{\int \psi_k^*(x) \hat{C} \psi_k(x) dx}{\int \psi_k^*(x) \psi_k(x) dx} = \gamma_k$$

$$\frac{\int (\hat{C} \psi_k(x))^* \psi_k(x) dx}{\int \psi_k^*(x) \psi_k(x) dx} = \gamma_k^*$$

However, because \hat{C} is Hermitian, the left-hand-sides of these equations are the same, and therefore their right-hand sides must also be equal. So $\gamma = \gamma^*$. But any number that is its own complex conjugate must be real (i.e. $(a + bi)^* = a - bi$ only if $b = \text{Im}[a + bi] = 0$). So the eigenvalues of a Hermitian operator are always real.