KSSOLV User's Guide

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Chapter 1

Introduction

KSSOLV is a MATLAB toolbox for solving a class of nonlinear eigenvalue problems also known as the Kohn-Sham equations. This type of problem arises from electronic structure calculation which is nowadays an essential tool for studying the microscopic quantum mechanical properties of molecules, solids and other nanoscale materials. Through the density functional theory (DFT) formalism, one can reduce the many-body Schrödinger equation used to describe the electron-electron and electron-nuclei interactions within an atomistic system to a set of single-electron equations that have far fewer degrees of freedom. These equations are first derived by Kohn and Sham in [14], and they have the form

$$H(\rho)\psi_i = \lambda_i \psi_i, \quad i = 1, 2, ..., n_e,$$
 (1.1)

where n_e is the number of electrons in the system, $\{\psi_i(r)\}\ (i=1,2,...,n_e)$ is a set of single electron wavefunctions that are mutually orthonormal, $\rho(r)$ is the total electron charge defined by

$$\rho(r) = \sum_{i=1}^{n_e} |\psi_i(r)|^2, \tag{1.2}$$

and

$$\lambda_1 \le \lambda_2 \le \ldots \le \lambda_{n_e}$$

are the n_e smallest eigenvalues of $H(\rho)$, a linear operator that depends on ρ . The operator H is known as the Kohn-Sham Hamiltonian and defined by

$$H(\rho) = -\frac{1}{2}\Delta + V_{ion}(r) + \rho \star \frac{1}{|r|} + V_{xc}(\rho),$$
(1.3)

where Δ is the Laplacian operator, V_{ion} and V_{xc} are known as the ionic and exchange-correlation potentials respectively [14], and \star denotes the convolution operator.

The Kohn-Sham equations are the Euler-Lagrange equations for the total energy minimization problem

$$\min_{s.t} E_{total}^{KS}(\psi_i)
s.t \psi_i^* \psi_j = \delta_{i,j},$$
(1.4)

where

$$E_{total}^{KS} = \frac{1}{2} \sum_{i=1}^{n_e} \int_{\Omega} |\nabla \psi_i|^2 dr + \int_{\Omega} \rho V_{ion} dr + \frac{1}{2} \int_{\Omega} \int_{\Omega} \frac{\rho(r_1)\rho(r_2)}{|r_1 - r_2|} dr_1 dr_2 + E_{xc}(\rho), \tag{1.5}$$

 $E_{xc}(\rho)$ is the exchange-correlation energy defined in [14].

After (1.1) is discretized, it becomes a set of nonlinear equations that bear similarity to algebraic eigenvalue problems found in standard linear algebra textbook. The main feature that distinguishes the Kohn-Sham equations from the standard linear eigenvalue problem is that the matrix operators in these equations are functions of the eigenvectors to be computed. Due to this nonlinear coupling between the matrix operator and its eigenvector, the Kohn-Sham equations are much more difficult to solve than a standard linear eigenvalue problem. Currently, the most widely used numerical method for solving the this type of problem is the Self Consistent Field (SCF) iteration, which we will examine in detail in Chapter 4. An alternative approach which is designed to minimize the total energy (1.5) directly will also be reviewed there. Both methods have been implemented in KSSOLV to allow users to compute the solution to the Kohn-Sham equations associated with various molecules and solids.

KSSOLV is designed using the object oriented programming features available in MATLAB. Several classes are created in KSSOLV to define both physical objects such as atoms and molecules and mathematical entities such as wavefunctions and Hamiltonians. These classes are defined in a hierarchical fashion so that a molecule object can be easily constructed from atom objects. The use of objects not only allows users with a minimal physics or chemistry background to quickly assemble a realistic atomistic system, but also allows them to develop and compare new numerical methods for solving the Kohn-Sham equations in a user friendly environment.

This user's guide will help users get familiar with various features of the toolbox. It is organized as follows. In next chapter, we will demonstrate how the toolbox can be used to set up simple molecular systems and how the Kohn-Sham equations associated with these systems can be solved by calling the numerical methods currently implemented in KSSOLV. In Chapter 3, we will take a closer look at various classes defined in KSSOLV, and show how KSSOLV objects can be created and modified. Chapter 4 focuses on numerical methods for solving the Kohn-Sham equations that have been implemented in KSSOLV. In particular, we will examine the SCF iteration and the DCM algorithm. We will discuss the importance of using the trust region technique to maintain monotonic reduction of the total energy function and how charge mixing can be used to accelerate and stabilize the convergence of the SCF iteration.

Chapter 2

Getting Started

2.1 Download KSSOLV

KSSOLV can be downloaded from

http://www.crd.lbl.gov/~chao/KSSOLV

2.2 Setting the environmental variable KSSOLVPATH

Once you retrieve the package, take a look at the README file in the directory kssolv. The file contains a minimal set of instructions for using the package.

Before running KSSOLV, one should define the environmental variable KSSOLVPATH. If you are running MATLAB under Unix/Linux operating system, this can be done by adding, for example,

setenv KSSOLVPATH /home/cyang/kssolv

to your .cshrc if you use csh. If you use bash, use

export KSSOLVPATH=/home/cyang/kssolv

If you are running MATLAB under Windows, click on Start->My Computer, and then click View system information on the left (System Task) panel. Continue with a click on Advanced, followed by a click on Environmental variables. If KSSOLVPATH is not among the defined environmental variables, click new, type in the variable name KSSOLVPATH, and the directory in which KSSOLV is installed.

2.3 Test Problems

The package contains a number of test problems that one can experiment with. Each test problem represents a particular molecule or periodic system. The system is created in a setup file. Table 2.1 contains the names of all setup files and a brief description for each of them. It also shows the number of electron pairs (occupied states (nocc) for most systems (with the exception of the electron quantum dot example in which nocc is the actual number of electrons in the dot).

setup file name noce		Description
c2h6_setup.m	7	an ethane molecule
co2_setup.m	8	a carbon dioxide molecule
h2o_setup.m	4	a water molecule
hnco_setup.m	8	an isocyanic acid molecule
qdot_setup.m	8	a 8-electron quantum dot confined by an external potential
si2h4_setup.m	6	a planar singlet silylene molecule
sibulk_setup.m	4	a silicon bulk system
sih4_setup.m 4		a silane molecule
ptnio_setup.m	43	a Pt_2Ni_6O molecule

Table 2.1: Setup files for examples included in KSSOLV

2.4 Setting up an Molecular System

To create a new system, a user can simply take one of the existing setup files and modify the construction of the Molecule object. A user can also change of the size of the primary cell or the kinetic energy cutoff of the existing setup file to examine the change in the convergence behavior of the numerical method or the quality of the computed solution.

In the following, we will use sih4_setup.m as an example to demonstrate how a molecular system can be easily set up in KSSOLV. The source code contained in sih4_setup.m is shown in Figure 2.2.

2.4.1 Initialization

In KSSOLV, an molecular system is defined by first creating an Molecule object, an particular instance of the Molecule class to be defined in Chapter 3, using

```
mol = Molecule();
```

Here mol is a user defined variable name and Molecule is a KSSOLV keyword. This single line of code creates an empty Molecule object mol with no particular attribute.

2.4.2 Specifying the Atom List

Properties of an Molecular object are specified by using the set method associated with the object. For example,

```
mol = set(mol, 'atomlist', alist);
```

is used in Figure 2.2 to specify the list of atoms contained in mol. Here the string 'atomlist' is the name of an attribute associated with an Molecule object predefined by KSSOLV, and alist is a user defined variable which supplies the actual atom list. In Figure 2.2, the variable alist is defined earlier as an array of Atom objects. Each Atom object can be defined by its chemical symbol or atomic number. For example, the silicon atom can be defined by

```
a = Atom('Si')
```

```
a = Atom(14);
```

Here, Atom is a KSSOLV keyword associated with the atom class (see Chapter 3 for details.)

The atom list array alist can also be defined by

```
alist(1) = Atom('Si');
for j = 2:5
    alist(j) = Atom('H');
end
```

2.4.3 Specifying the Positions of the Atoms

The position of each atom in the atom list is specified by an $n_a \times 3$ array (xyzmat in Figure 2.2). The *i*th row of this array gives the x, y, z coordinates of the *i*th atom in the atom list. This array can be passed to the mol object through the set method as one can see in Figure 2.2. The unit of the coordinates should be in Bohr (1 Bohr = 0.5291772083 Angstrom).

Although it is perfectly OK to define this array directly, it is sometimes more convenient to first define the relative position of each atom in a unit (super)cell (of a periodic lattice) specified by three lattice vectors \mathbf{c}_1 , \mathbf{c}_2 and \mathbf{c}_3 . If we denote the relative displacement of the atom along \mathbf{c}_i by ξ_i , then the absolute x, y, z coordinates of the atom correspond to the three components of the row vector

$$r = \xi_1 \mathbf{c}_1^T + \xi_2 \mathbf{c}_2^T + \xi_3 \mathbf{c}_3^T = (\xi_1 \, \xi_2 \, \xi_3) C,$$

where

$$C = \begin{pmatrix} \mathbf{c}_1^T \\ \mathbf{c}_2^T \\ \mathbf{c}_3^T \end{pmatrix}. \tag{2.1}$$

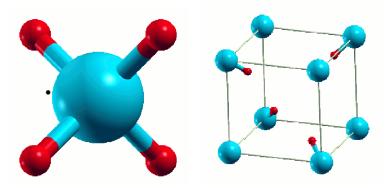
The 3×3 matrix C defines a unit supercell in KSSOLV. It can be used to set the 'supercell' attribute of a Molecule object. For example,

```
mol = set(mol, 'supercell', 10*eye(3));
```

sets the unit supercell of mol to a $10 \times 10 \times 10$ cube.

If the relative positions of n_a atoms within a unit supercell defined by C are placed in an $n_a \times 3$ array coefs, then the absolute coordinates of these atoms can be obtained from coefs*C.

We should point out that the (x, y, z) coordinates associated with the atoms contained in a Molecule object can be negative even though the lattice vectors used to specify the unit supercell are usually in $\mathbb{R}^+ \times \mathbb{R}^+ \times \mathbb{R}^+$. Because the unit supercell is assumed to periodically extended throughout the entire domain, (x, y, z) is equivalent to $(x + m\mathbf{c}_1, y + n\mathbf{c}_2, z + \ell\mathbf{c}_3)$ for any integer m, n and ℓ , i.e., we can always bring an atom into the specified unit supercell if its coordinates are outside of the unit supercell. For example, if the coordinate of the silicon atom is chosen to be (0,0,0), then the atomic positions shown on the left side of Figure 2.1 are equivalent to those shown on the right side of the figure.



(a) Atomic positions outside of the (b) Atomic positions inside of the unit supercell unit supercell

Figure 2.1: Equivalent rendering of the atomic positions of Si and H in a silane molecule

2.4.4 Kinetic Energy Cutoff

In order to discretize the Kohn-Sham equations with an appropriate number of planewave basis functions, one must specify a kinetic energy cutoff E_{cut} through

where cutoff_value is a user supplied number in the unit of Rydberg (Ryd).

This energy cutoff defines a maximum magnitude limit for the wave numbers g_j allowed in the planewave expansion of a single-particle wavefunction $\psi(r)$. That is, when the cutoff is set to E_{cut} , $\psi(r)$ is expressed as

$$\psi(r) = \sum_{j=1}^{n_g} c_j e^{ig_j^T r},$$
(2.2)

where n_g is the number of all wave numbers g_j that satisfy

$$||g_i||^2 < 2E_{cut}, (2.3)$$

The frequency cutoff E_{cut} also allows us to determine the minimum number of sampling grid points (n_1, n_2, n_3) we need to represent $\psi(r)$ in real space without causing aliasing effects [31].

Clearly, the larger the E_{cut} value we use, the larger the number of planewave basis functions and real space sampling grid points are required in the representations of $\psi(r)$. The use of a large number of planewave basis leads to a more accurate finite dimensional approximation to the continuous Kohn-Sham equations. Unfortunately, this also leads to a high computational cost (in term of both time and storage) required to solve the finite dimensional Kohn-Sham equations.

The optimal choice of E_{cut} is system dependent. The choice also depends on the pseudopotential function used. However, for the purpose of understanding the numerical method for solving a finite dimensional KS equation, it is sufficient to set E_{cut} to a relatively small number, e.g., $E_{cut} = 25Ryd$. One can experimenting with different choices of E_{cut} to see how the minimum total energy changes as one increase the number of plane wave basis in the discretization.

```
% 1. construct atoms
a1 = Atom('Si');
a2 = Atom('H');
alist = [a1; a2; a2; a2; a2];
\% 2. set up primitive cell (Bravais lattice)
C = 10*eye(3);
% 3. define the coordinates the atoms
coefs = [
        0.0
0.0
                  0.0
0.161
        0.161
                 0.161
-0.161 -0.161
                0.161
0.161 -0.161
               -0.161
-0.161 0.161
               -0.161
];
xyzmat = coefs*C';
\% 4. Configure the molecule
mol = Molecule();
mol = set(mol, 'Blattice',C);
mol = set(mol, 'atomlist', alist);
mol = set(mol,'xyzlist',xyzmat);
mol = set(mol, 'ecut', 25);  % kinetic energy cut off
mol = set(mol, 'name', 'SiH4');
```

Figure 2.2: Setting up the Silane molecule

2.4.5 Checking the Attributes of an Molecule Object

After all attributes of an Molecule object mol have been set, one can view all attributes by simply typing mol (without semicolon) at the MATLAB prompt without a semicolon at the end. For example, typing mol after running sih4_setup gives the output shown in Figure 2.3.

```
>> mol
Molecule: SiH4
   primitive cell:
                                    0.000e+00
      1.000e+01
                     0.000e+00
      0.000e+00
                     1.000e+01
                                    0.000e+00
      0.000e+00
                     0.000e+00
                                    1.000e+01
   sampling size: n1 = 32, n2 = 32, n3 = 32
   atoms and coordinates:
                    0.000e+00
                                    0.000e+00
                                                    0.000e+00
       1
           Si
       2
            Η
                    1.610e+00
                                    1.610e+00
                                                    1.610e+00
       3
            Η
                   -1.610e+00
                                   -1.610e+00
                                                    1.610e+00
       4
            Н
                    1.610e+00
                                   -1.610e+00
                                                   -1.610e+00
       5
            Η
                                    1.610e+00
                                                   -1.610e+00
                   -1.610e+00
   number of electrons
   spin type
   kinetic energy cutoff: 2.500e+01
```

Figure 2.3: Displaying the attributes of the Silane molecule

Note that once the list of atoms is specified, the Molecule object automatically calculates the number of valence electrons contained in the molecular system. In the case of the SiH_4 molecule, there are 8 valence electrons. Four of them are contributed by the Si atom in the 2s and 2p atomic orbitals. The four hydrogen atoms each contributes an electron. Currently, KSSOLV does not distinguish the spin orientations of valence electrons in molecules or solids. This is indicated by the spin type attribute in the Molecule object, which is set to 1 in Figure 2.3. What that means is that for this particular system, electrons are considered in pairs, and the number of occupied states (Kohn-Sham wavefunctions) is half of the number of valence electrons.

For systems that do not contain atoms such as electron quantum dots that contain electrons confined by an external potential. The number of electrons **nel** must be specified explicitly through a statement like

```
mol = set(mol,'nel',4);
```

In this case, one may also set the spin type to 2 by

```
mol = set(mol, 'nspin', 2)
```

so that electrons will not be considered in pairs.

2.5 Solving the Kohn-Sham Equations

Once an Molecule object is set up properly, computing the electron wavefunctions associated with the minimum total energy of the molecular system can be performed in KSSOLV by calling one of the functions listed in Table 2.2. We will describe these methods in detail in chapter 4.

function	Description
scf.m	Self consistent field iteration
chebyscf.m	Chebyshev filtered self consistent field iteration
dcm.m	direct constrained minimization (DCM)
trdcm.m	direct constrained minimization
	with flexible trust region parameters
trdcm1.m	direct constrained minimization
	with a fixed trust region parameter

Table 2.2: Methods developed for Solving the KS equation in KSSOLV

The simplest way to use the **scf** function which contains an implementation of the self-consistent field (SCF) iteration with charge mixing [22, 23, 11] acceleration is:

```
>> [Etotvec, X, vtot, rho] = scf(mol);
```

The function call takes mol as its only input and produces the following outputs:

- 1. Etotvec is an array of total energies computed at each SCF iteration;
- 2. X is a Wavefun object that contains the approximate single-particle wavefunctions returned from the SCF calculation. The details on the structure of the Wavefun class are explained in Chapter 3;
- 3. vtot is a 3D array that contains the total potential associated with the returned single-particle wavefunctions of the mol object.
- 4. rho is a 3D array that contains the electron charge associate with the returned single-particle wavefunctions of the mol object.

All functions listed in 2.2 allow additional options to be passed in through an option structure that can be defined by calling the function setksopt.

Calling setksopt without any argument simply creates an option structure with default parameter settings. For example,

```
opts = setksopt;
```

creates an option structure opt that contains the following fields

```
opts =
```

verbose: 'on'
maxscfiter: 10
maxdcmiter: 10

```
maxinerscf: 3
maxcgiter: 10
    scftol: 1.0000e-08
    dcmtol: 1.0000e-08
    cgtol: 1.0000e-06
    mixtype: 'anderson'
    mixdim: 9
    betamix: 0.8000
        brank: 1
            X0: []
            rho0: []
```

Each field can be modified by simply resetting the value of the field through an assignment statement. For example, to change the maximum number of SCF iterations allowed, we can simply use

```
opts.maxscfiter = 20.
```

This parameter can also be modified by calling **setksopt** again with appropriate argument pairs. For example,

```
opts = setksopt(opts, 'maxscfiter', 20);
```

achieves the same effect. The advantage of calling setksopt is that the function checks the validity of the structure field and its value.

More information about the options and default values for all methods listed in Figure 2.2 can be found in Chapter 4.

2.6 Understanding the Output from KSSOLV

By default, both the SCF and the DCM methods implemented in KSSOLV print out a bunch of information on the screen. A typical SCF run produces diagnostic output like the one shown in Figure 2.4. After some initialization, the code uses the either the LOBPCG algorithm [13] to be described later in section 4 (which is the default) or some other subspace iteration based algorithms to solve a sequence of linear eigenvalue problem. In the case of the silane molecule (SiH_4) , there are 4 electron pairs. Thus 4 smallest eigenvalues of a fixed Hamiltonian are approximated at each SCF iteration. The diagnostic output shows both the eigenvalue approximation and the estimated error associated with these eigenvalues and their corresponding eigenvectors. At the end of each SCF iteration, the code also prints out the total energy evaluated at the current wavefunction approximation, the residual error associated with eigenvalue and wavefunction approximations (to the solution of the Kohn-Sham equations (1.1)), and the lack of self-consistency in the total potential.

2.7 Visualizing the Results

The electron charge density, total potential and single particle wavefunctions can be easily visualized using either the MATLAB's isosurface rendering function isosurface or the volume rendering function written by Joe Conti, which we also include in KSSOLV. To see the isosurface of the charge density rho, one can simply type

```
[Etotvec, X, vtot, rho] = scf(mol);
initialization...
Beging SCF calculation...
LOBPCG iter =
              1
eigval(1) = 7.195e+00, resnrm =
                                   3.277e+00
eigval(2) = 7.368e+00, resnrm =
                                   3.347e+00
eigval(3) = 7.551e+00, resnrm =
                                   3.233e+00
eigval(4) =
             7.703e+00, resnrm =
                                   3.239e+00
LOBPCG iter = 2
eigval(1) = 5.097e-01, resnrm =
                                   8.297e-01
eigval(2) = 6.200e-01, resnrm =
                                   8.719e-01
eigval(3) = 6.912e-01, resnrm =
                                   9.045e-01
eigval(4) = 9.038e-01, resnrm =
                                   9.130e-01
. . .
SCF iter 1:
norm(vout-vin) = 5.807e+00
Total energy = -5.8719608972592e+00
resnrm = 1.857e-02
resnrm =
           2.161e-02
           2.161e-02
resnrm =
           2.162e-02
resnrm =
```

Figure 2.4: SCF output

>> isosurface(rho).

The scf, dcm and trdcm functions in KSSOLV return a charge density defined within the specified unit supercell even though the atomic coordinates defined during the initial setup may not lie within that cell. To see the a charge density in the original atomic coordinates, one may use the MATLAB fftshift function first to shift density before rendering. That is, one may use

>> isosurface(fftshift(rho))

to view the isosurface of the charge density.

Figure 2.5 shows the isosurface rendering of the charge density of the SiH_4 molecule associated with the atomic positions defined in Figure 2.2 and displayed in Figure 2.1(a). Similarly,

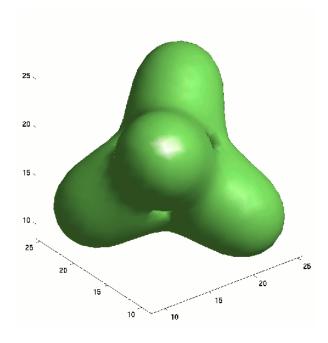


Figure 2.5: The computed charge density of the SiH_4 molecule.

```
one can use
>> view(3);
>> vol3d('cdata',fftshift(rho));
to view a volume rendering of rho.
```

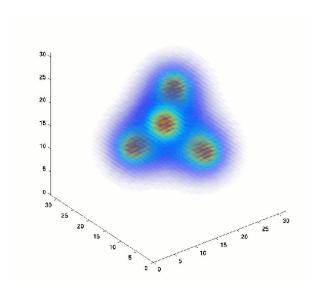


Figure 2.6: A volume rendering of the computed charge density of the SiH_4 molecule.

Chapter 3

KSSOLV Classes

In this chapter, we will describe all classes defined in KSSOLV in detail. In particular, we will examine the structure of these classes and methods that can be invoked. We will also discuss overloaded operators that can be used directly on objects that belong to certain classes.

In MATLAB, a user-defined class is identified by a directory whose name begins with the '@' symbol. Therefore, once KSSOLV is installed, one should find the following directories in KSSOLV's root directory kssolv/:

```
@Atom/
@Molecule/
@Ham/
@Wavefun/
@FreqMask/
```

Each @-prefixed directory contains an M-file with the same name (excluding the @ symbol). This M-file is a *constructor* that can be used to instantiate a specific KSSOLV object that belongs to a certain class. The function defined in this M-file can be called without any input argument to create a null object that needs to be refined later by calling the set method. For example, the following call creates a null Atomobject.

```
a = Atom();
```

One may also call the constructor with one or a few arguments to initialize the object when it is created. For example, the following call creates an Hamiltonian associated with a predefined Molecule object mol using a predefined charge density rho.

```
H = Ham(mol, rho);
```

Each class contains a number of attributes specific to that class. These attributes must be defined properly before an object belonging to that class is used. When these attributes are not initialized upon the initial construction, they can be set by using the set method defined for that class. For example, if an Atom object a is created by calling the constructor with no input argument (as illustrated above), we can use

```
a = set(a,'symbol','H');
```

to specify that a is an hydrogen atom. Here the string 'symbol' is a keyword that represents the name of a specific attribute, which in this case refers to the chemical symbol of an atom. The user provided string 'H' is a value assigned to that attribute.

Note that not all attributes of an object need to be set before the object can be used in subsequent calculations. Setting some key attributes will ensure that other attributes of the same object are properly set automatically. In fact, for an Atom object, the only attribute that needs to be set is either the atomic symbol 'symbol', or the atomic number 'anum'. All other attributes of the atom, which we will examine below, will be defined automatically once 'anum' or 'symbol' is set.

All attributes of an object can be retrieved by using the get method defined for the class to which the object belongs. For example,

```
nvel = get(a,'vnum');
```

retrieves the number of valence electrons in atom a, and assigns that number to the variable nvel.

The essential attributes of an object can also be displayed by typing the name of object at the MATLAB prompt without a semicolon. For example, typing

```
>> a = Atom('Si')
shows
atomic symbol: Si
atomic number: 14
number of valence electrons: 4
```

3.1 Atom

Atoms are the basic constituents of molecules and solids. Therefore, it is natural to define an Atom class in KSSOLV so that we can easily create atom objects and use them to construct molecules and bulk systems. An atom can be declared in KSSOLV as an Atom object using an Atom constructor. For example

```
a = Atom('Si');
or
a = Atom(14);
```

defines a silicon atom object named a. Note that an atom can be defined by either its chemical symbol or by its atomic number. An atom object contains a number of attributes listed and described in Table 3.1. Most of these attributes are properly set once the chemical symbol or the atomic number of the atom is initialized. They are mainly used by KSSOLV internally to construct ionic pseudopotentials.

All these attributes are accessible through the get method defined for the Atom class. However, a user would rarely need to retrieve attributes other than the atomic number and the number of valence electrons. 3.2. MOLECULE 21

attribute name	description	data type
'symbol'	chemical symbol	string
'anum'	atomic number	integer
'venum'	number of valence electrons	integer

Table 3.1: Attributes of an Atom object

3.2 Molecule

The keyword Molecule is used in KSSOLV broadly to refer to both a single molecule and a periodic (bulk) system. The Molecule class is created to encapsulate both the physical attributes of a molecule and the planewave discretization parameters that must be defined before the Kohn-Sham equations associated with a particular Molecule object can be solved.

To create an Molecule object, one typically uses

mol = Molecule();

to first construct an empty object called mol (a user defined variable name). This call simply sets up the required data structure that would be used to describe various attributes of mol. The actual attributes of mol such as the number and type of atoms in this molecule, the size and shape of the supercell that contains the molecule etc. can be defined through the set method.

mol = set(mol,attrname,attrvalue);

where the input argument attrname is a predefined string associated with the Molecule class that specifies the name of a particular attribute, and attrvalue is a user supplied quantity that will be stored in mol. Table 3.2 lists the all attributes of an Molecule object. Some of these attributes such as the name of the molecule and the external potential are optional. Some attributes such as the number of sampling grid points n1, n2, n3 will be properly initialized after other attributes such as ecut are defined.

The datatypes associated with most of the attributes are straightforward. A few require further explanation. The 'supercell' attribute defines the shape and size of the unit supercell that contains the basic atomic constituents of a single molecule or solid. In KSSOLV, the super cell is described by a 3×3 matrix. Each row of this matrix defines a lattice vector emanating from the origin. For example,

```
mol = set(mol, 'supercell', 10*eye(3));
```

sets the unit supercell of mol to a $10 \times 10 \times 10$ cube whose three basic lattice vectors are parallel to the x, y and z axes respectively. Atomic units (Bohr) are used in the definition of the unit cell.

The 'atomlist' attribute is defined to be an array of Atom objects. An example of how to construct such an array has been shown in Chapter 2. The coordinates associated with n_a atoms listed in 'atomlist' must be defined in an $n_a \times 3$ array that contains the x, y, z coordinates (in atomic units) for each atom.

Currently, KSSOLV makes no distinction between the wavefunctions associated with electrons occupying the same orbitals. In this case, the spin type is normally set to nspin=1.

However, KSSOLV can also be used to solve the Kohn-Sham equations for electron-only systems. In this case, electron wavefunctions are not associated with any atoms, and they must be treated differently, which indicated in KSSOLV by setting nspin to 2.

attribute name	purpose	value type
'name'	The name of the Molecule	string
'supercell'	Bravais lattice box	3×3 matrix
'atomlist'	list of atoms	array of Atom objects
'xyzlist'	list of atomic coordinates	an $n_a \times 3$ matrix
'vol'	The volume of the primary (super)cell	
	that contains the molecule	float
'natoms'	number of atoms in the primary cell	integer
'ecut'	planewave energy cutoff	floating point scalar
'n1'	the number of sampling points along the x -axis	integer
'n2'	the number of sampling points along the y -axis	integer
'n3'	the number of sampling points along the y -axis	integer
'nspin'	the spin type	1 or 2
'nel'	the number of valence electrons	integer
'vext'	external potential	3D array of size $n1 \times n2 \times n3$

Table 3.2: Attributes to be set in an Molecule object

3.3 Ham

A properly defined Molecule object mol can be used to initialize a KS Hamiltonian. The initialization can be done by simply calling the constructor

H = Ham(mol);

The constructed Ham object should contain all information one would need to solve the Kohn-Sham equations. Although the KS Hamiltonian H(X) is treated as a matrix in equation (4.12), it is not stored as a matrix in KSSOLV. Instead, the Ham class keeps L, \hat{V}_{ion} and total potential $V_{tot} = \hat{V}_{ion} + \text{Diag}(L^{\dagger}\rho) + \text{Diag}(\mu_{xc}(\rho))$ as separate attributes. This separation makes it easy to update the Hamiltonian in both the SCF and DCM calculations. Table 3.3 shows all other important attributes of an Ham object.

attribute name	purpose	value type
'gkk'	The magnitude square of wave numbers associated with the	1D array
'idxnz'	kinetic energy that satisfy the cutoff constraint (2.3) positions of the reciprocal lattice points associated with all	1D array
'vtot'	entries of gkk within a 3D array total local potential	3D array
'vion'	local ionic pseudopotential in real space	3D array
'wqmat'	Fourier representation of the non-local pseudopotential	2D array
'vext'	external potential	(matrix) 3D array
'vnp'	charge dependent portion of the potential	3D array

Table 3.3: Attributes to be set in an Molecule object

The kinetic component of a Ham object is represented and stored as a 1D array (gkk) of integers that represent the magnitude squares of the wave numbers that satisfy the energy cutoff constraint (2.3). These values correspond to the eigenvalues of a Laplacian operator

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(with periodic boundary conditions) sampled in the reciprocal (frequency) space. Only wave numbers (or frequencies) that satisfy the kinetic energy cutoff set in mol are stored. There numbers correspond to reciprocal space lattice points that lie within a sphere with a radius of $\sqrt{2E_{cut}}$. The locations of these lattice points in a 3D array of all lattice points (including those outside of the cutoff sphere) are kept in a separate array idxnz so that wave numbers associated with reciprocal space lattice points outside of the kinetic energy cutoff sphere are not used or stored.

The local ionic potential \hat{V}_{ion} is also constructed and stored as a 3D array at initialization. The construction process involves loading the atomic pseudopotential stored in the directory Pseudopot and summing them up using the scheme suggested by Kleinman and Bylander [12]. KSSOLV provides Troullier-Martins [27] atomic pseudopotentials associated with all atoms listed in the first four rows of the periodic table. Some elements in higher row of the table are also provided.

The determination of both the Hartree and exchange correlation potential requires the availability of the charge density ρ which is in turn a function of the wavefunctions to be computed. Since a good approximation to the desired wavefunctions is not available at initiation, the initial ρ is computed in KSSOLV by combining atomic charge densities associated with each atom in the mol object [12].

Once a Ham object has been defined, one can retrieve various attributes of the object through the get function, e.g.,

```
vt = get(H,'vtot');
```

returns the total potential from a Ham object named H and assigns it to a user defined variable vt. This potential can be used and updated in a subsequent SCF or DCM calculation. An updated vt can be passed into H by using the set function

```
H = set(H,'vtot',vt);
```

3.4 Wavefun

We created a special class Wavefun in KSSOLV to specify one or a set of wavefunctions rather than representing them by a MATLAB vector or matrix. The rationale for creating such a class is mainly to reduce the amount of codes a user would have to write to reshape a 3D array (which is the data structure required for a 3D FFT call) into a 1D vector (which is the most convenient data structure for linear algebra operations) and vice versa. Representing a wavefunction as a Wavefun object also makes it easy to perform the multiplication of a KS Hamiltonian with a wavefunction and basic linear algebra operations between wavefunctions through operator overloading.

In KSSOLV, a Wavefun object stores the Fourier coefficients of one or several wavefunctions that are expanded in a planewave basis defined by (2.2). The object can be constructed using either a noncompact scheme or a compact scheme. In a noncompact representation, a Wavefun object X can be constructed through the command

```
X = Wavefun(psi),
```

where psi is a MATLAB 3D array if X represents a single wavefunction, or a cell array that contains a list of 3D arrays if X represents a set of wavefunctions.

Under the compact scheme, a Wavefun object stores only the Fourier coefficients $(c_{i,j})$ of a truncated expansion of a single wavefunction $\psi(r)$ or a set of wavefunctions $\{\psi_i(r)\}_{i=1}^{n_e}$. These coefficients are associated with wave numbers (g_j) that satisfy the cutoff constraint (2.3), and stored contiguously as a MATLAB cell array of size n_q by nocc, where n_q is the number of these Fourier expansion coefficients, and nocc is the number of occupied states (electron pairs). All other Fourier coefficients that may appear in a full planewave expansion are considered zeros, thus never used or stored. Storing the nonzero Fourier expansion coefficients contiguously in a 1D array makes linear algebra operations on X more efficient. However, to perform 3D FFT operations on X, we must place these coefficients in a 3D array. Hence, in the compact scheme, we must also record the locations of these Fourier coefficients in a 3D array, which are the locations of the reciprocal lattice points associated with the corresponding wave numbers in the planewave basis. These locations are stored in a separate array which is labeled as the 'idxnz' attribute of the Wavefun object. Recall that the same strategy is used to store the magnitude squares of the wave numbers and their locations in a 3D array for the Ham object. In signal processing, functions that are expanded by planewaves up to certain frequency or wave number are called *bandlimited* functions.

The size of the 3D array that is used to hold the Fourier coefficients when an FFT is performed is specified by the 'n1', 'n2' and 'n3' attributes of the Wavefun object. These numbers also give the dimension of the corresponding wavefunction in real space. They should be the same as those specified in a mol object.

The constructor for a Wavefun object can be called in a number of ways. In addition to creating a null object by using,

X = Wavefun();

we may also create a null object of certain size by using

```
X = Wavefun(n1,n2,n3);
```

where n1, n2 and n3 are predefined integers. The following use of the constructor creates a Wavefun object using a noncompact scheme.

```
X = Wavefun(Q);
```

where Q is either a 3D array or a cell array that contains a number of 3D arrays. These 3D arrays should contain the Fourier expansion coefficients of one or several wavefunctions.

Finally, a fixed size Wavefun object in compact scheme can be constructed by using

```
X = Wavefun(Q,n1,n2,n3,idxnz);
```

where Q is a 1D array that stores the Fourier expansion coefficients continuously and idxnz is a 1D array that describes the locations of these coefficients in a 3D array of size $n1 \times n2 \times n3$.

In the following section, we will see that all major matrix operations have been overloaded for Wavefun objects for both the compact and noncompact schemes. KSSOLV also provides a utility function genX0 that allows one to easily construct initial Wavefun objects for the SCF or DCM calculations. To generate a set of random bandlimited wavefunctions using the kinetic energy cutoff specified in a Molecule object mol, one can simply use the command

```
X = genX0(mol).
```

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Converting a Wavefun object X to a 3D array (or a list of 3D arrays) is straightforward when X is constructed using a noncompact scheme. The following command

```
X3D = get(X, 'psi')
```

returns the wavefunctions as a cell array X3D of 3D arrays. Although rarely needed when using KSSOLV, the following lines of codes show how the same conversion can be accomplished for an X constructed using a compact representation scheme

```
n1 = get(X,'n1');
n2 = get(X,'n2');
n3 = get(X,'n3');
psi = get(X,'psi');
idx = get(X,'idxnz');
X3D = zeros(n1,n2,n3);
X3D(idx) = psi{1}.
```

3.5 FreqMask

The FreqMask class is created to define an object that stores the wave numbers (frequencies) that appear in (2.2) and their magnitudes. The three components of each wave number g_j are stored in three separate arrays gkx, gky and gkz. These arrays are attributes of a FreqMask. They are used in the pseudopotential calculation (getvion.m, getwq.m) and a number of other KSSOLV functions. Only the wave numbers that satisfy the energy cutoff criterion (2.3) are defined and stored when a FreqMask object is constructed. The number of wave numbers that satisfy (2.3) is stored as the attribute ng.

A FreqMask object can be constructed by calling

```
gm = FreqMask(mol).
```

In this case, the energy cutoff defined in the Molecule object mol is used to determine which wave numbers should be set in gkx, gky and gkz. The constructor of the object also calculates and stores the magnitude square of each Fourier expansion coefficient that satisfies (2.3) in an separate array gkk. Because such a quantity is used frequently in the numerical procedure for solving the Kohn-Sham equations we store it in a FreqMask object instead of computing it repeatedly from gkx, gky and gkz each time it is used.

Alternatively, one can pass in a different energy cutoff and create a FreqMask object by using

```
gm = FreqMask(mol,ec),
```

where ec is a user provide energy cutoff value.

3.6 Operator Overloading

Because the Kohn-Sham Hamiltonian H(X) and wavefunction X are viewed as matrices in the algorithms to be described in the next chapter, it is desirable to allow Hamilt and Wavefun objects to be manipulated in KSSOLV as if they are matrices. This feature is made possible in KSSOLV by overloading some basic algebraic operations for a Wavefun object. These

overloaded operations are listed in Table 3.4. One should be careful about the use of some of these operators. For example, since the wavefunctions used in the SCF and DCM calculation all have the same dimension, the multiplication operator * is never used between two Wavefun objects except when the first Wavefun object is transposed or conjugate transposed, i.e., it is valid to perform x'*y or x.'*y, and the multiplication returns a standard MATLAB matrix object. The overloaded multiplication operator * for Wavefun objects allows the second operand to be a standard matrix object with proper dimension. The result of the multiplication is a Wavefun object.

Operations	Description
x + y	Add two wavefunctions
х - у	Subtract one wavefunction from another
x * y	Multiply two wavefunctions and return a matrix
x * a	Multiply several wavefunctions with a matrix
х .* у	Element-wise multiplication of two wavefunctions
х. у	Element-wise division of two wavefunctions
x,	Complex conjugate transpose of a wavefunction
x.,	Transpose of a wavefunction
[x y]	Horizontal concatenation of several wavefunctions
x(:,i:j)	Subscripted reference of wavefunctions

Table 3.4: Overload operations for Wavefun objects in KSSOLV

The multiplication operator is also overloaded for the Hamilt class so that the multiplication of a Hamilt object H and a Wavefun object X can be accomplished in KSSOLV by a simple expression

Y = H*X,

which hides all the complexity of the operation from the user.

Chapter 4

KSSOLV Methods

In this chapter, we will briefly describe the numerical methods employed in KSSOLV to obtain an approximate solution to the Kohn-Sham equations (1.1). We begin by discussing the planewave discretization scheme that turns the continuous nonlinear problem into a finite dimensional problem. The finite dimensional problem is expressed as a matrix nonlinear eigenvalue problem. We present two different approaches to solving the matrix nonlinear eigenvalue problem in sections 4.4 and 4.5.

4.1 Planewave Discretization

To solve the minimization problem (1.4) or the Kohn-Sham equation (1.1) numerically, we must first discretize the continuous problem. Standard discretization schemes such as finite difference, finite elements and other basis expansion (Ritz-Galerkin) methods [25] all have been used in practice. The discretization scheme we have implemented in the current version of KSSOLV is a Ritz type of method that expresses a single electron wavefunction $\psi(r)$ as a linear combination of planewaves $\{e^{-ig_j^Tr}\}$, where $g_j \in \mathbb{R}^3$ (j=1,2,...,K) are wave numbers (frequency vectors) arranged in a lexicographical order. The planewave basis is a natural choice for studying periodic systems such as solids. It can also be applied to non-periodic structures (e.g., molecules) by embedding these structures in a ficticious supercell [18] that is periodically extended throughout an open domain. The use of the planewave basis has the additional advantage of making various energy calculations in density functional theory easy to implement. It is the most convenient choice for developing and testing numerical algorithms for solving the Kohn-Sham equations within the MATLAB environment, partly due to the availability of efficient fast Fourier transform (FFT) functions.

It is natural to assume that the potential for R-periodic atomistic systems is a periodic function with a period $R \equiv (R_1, R_2, R_3)$. Consequently, we can restrict ourselves to one canonical period often referred to as the primitive cell and impose periodic boundary condition on the restricted problem. It follows from the Bloch's theorem [3, 4] that eigenfunctions of the restricted problem $\psi(r)$ can be periodically extended to the entire domain (to form the eigenfunction of the original Hamiltonian) by using the following formula:

$$\psi(r+R) = e^{ik^T R} \psi(r), \tag{4.1}$$

where $k = (k_1, k_2, k_3)$ is a frequency or wave vector that belongs to a primitive cell in the reciprocal space (e.g., the first *Brillouin* zone [3]). If the *R*-periodic system spans the entire

infinite open domain, the set of k's allowed in (4.1) forms a continuum in the first Brillouin zone. That is, each $\psi(r)$ generates an infinite number of eigenfunctions for the periodic structure. It can be shown that the corresponding eigenvalues form a continuous cluster in the spectrum of the original Hamiltonian [3]. Such a cluster is often referred to as an energy band in physics. Consequently, the complete set of eigenvectors of H can be indexed by the band number i and the Brillouin frequency vector k (often referred to as a k-point), i.e., $\psi_{i,k}$. In this case, the evaluation of the charge density must first be performed at each k-point by replacing $\psi_i(r)$ in (1.2) with $\psi_{i,k}$ to yield

$$\rho_k = \sum_{i=1}^{n_e} |\psi_{i,k}|^2.$$

The total charge density $\rho(r)$ can then be obtained by integrating over k, i.e.,

$$\rho(r) = \frac{|\Omega|}{(2\pi)^3} \int_{BZ} \rho_k(r) dk, \tag{4.2}$$

where $|\Omega|$ denotes the volume of the primitive cell in the first Brillouin zone. Furthermore, an integration with respect to k must also be performed for the kinetic energy term in (1.5).

When the primitive cell (or supercell) in real space is sufficiently large, the first Brillouin zone becomes so small that the integration with respect to k can be approximated by a single k-point calculation in (4.2).

To simplify our exposition, we will, from this point on, assume that a large primitive cell is chosen in the real space so that no integration with respect to k is necessary. Hence we will drop the index k in the following discussion and use $\psi(r)$ to represent an R-periodic single particle wavefunction. The periodic nature of $\psi(r)$ implies that it can be represented (under some mild assumptions) by a Fourier series, i.e.,

$$\psi(r) = \sum_{j=-\infty}^{\infty} c_j e^{ig_j^T r},\tag{4.3}$$

where c_i is a Fourier coefficient that can be computed from

$$c_j = \int_{-R/2}^{R/2} \psi(r) e^{-ig_j^T r} dr.$$

In practice, the exact number of terms used in (4.3) is determined by a kinetic energy cutoff E_{cut} . Such a cutoff yields an approximation

$$\psi(r) = \sum_{j=1}^{n_g} c_j e^{ig_j^T r}, \tag{4.4}$$

where n_q is chosen such that

$$||g_j||^2 < 2E_{cut}, (4.5)$$

for all $j = 1, 2, ..., n_g$. Although, the value of n_g will depend on many parameters such as the size and type of the system being studied, it is typically an order of magnitude smaller than $n = n_1 \times n_2 \times n_3$.

Once E_{cut} is chosen, the minimal number of samples of r along each Cartesian coordinate direction (n_1, n_2, n_3) required to represent $\psi(r)$ (without the aliasing effect) can be determined

from the sampling theorem [17]. That is, we must choose n_k (k = 1, 2, 3) sufficiently large so that

$$\frac{1}{2} \left(\frac{2\pi n_k}{R_k} \right) > 2\sqrt{2E_{cut}},\tag{4.6}$$

is satisfied, i.e., n_k must satisfy $n_k > 2R_k \sqrt{2E_{cut}}/\pi$.

We will denote the uniformly sampled $\psi(r)$ by a vector $x \in \mathbb{R}^n$, where $n = n_1 n_2 n_3$ and the Fourier coefficients c_j in (4.4) by a vector $c \in \mathbb{C}^n$ with zero paddings used to ensure the length of c matches that of x. If the elements of x and c are ordered properly, these two vectors satisfy

$$c = Fx. (4.7)$$

where $F \in \mathbb{C}^{n \times n}$ is a discrete Fourier transform matrix [28].

After a sampling grid has been properly defined, the approximation to the total energy can be evaluated by replacing the integrals in (1.5) with simple summations over the sampling grid.

4.2 Pseudopotentials

To solve the Kohn-Sham equations numerically, the Fourier series expansion (4.3) must be truncated to allow a finite number of terms only. If all electrons are treated equally, the number of terms required in (4.3) will be extremely large. This is due to the observation that the strong interaction between a nucleus and the inner electrons of an atom, which can be attributed to the presence of singularity in $V_{ion}(r)$ at the nuclei position \hat{r}_i , must be accounted for by high frequency planewaves. However, because the inner electrons are held tightly to the nuclei, they are not active in terms of chemical reactions, and they usually do not contribute to chemical bonding or other types of interaction among different atoms. On the other hand, the valence electrons (electrons in atomic orbits that are not completely filled) can be represented by a relatively small number of low frequency planewaves. These electrons are the most interesting ones to study because they are responsible for a majority of the physical properties of the atomistic system. Hence, it is natural to focus only on these valence electrons and treat the inner electrons as part of an ionic core. An approximation scheme that formalizes this approach is called the pseudo-potential approximation [19, 20, 34]. The details of pseudopotential construction and their theoretical properties are beyond the scope of this user's guide. The user should just keep in mind that the use of pseudo-potentials allows us to

- 1. remove the singularity in V_{ion} ;
- 2. reduce the number of electrons n_e in (1.5) and (1.2) to the number of valence electrons;
- 3. represent the wavefunction associated with a valence electron by a small number of low frequency planewaves.

In KSSOLV, the ionic pseudopotential experienced by each single particle wavefunction is constructed from atomic pseudopotentials that have been previously constructed and stored in the Pseudopot directory. After a Molecule object mol has been properly created, the user can simply call

pseudovar = pseudoinit(mol)

which returns a MATLAB structure pseudovar (user defined variable name). The pseudovar contains various pieces of information required by KSSOLV to construct the ionic pseudopotential. In most cases, a user does not need know what is stored in pseudovar.

The ionic pseudopotential consists of two components. One of them is a local component $V_{local}(r)$ that operates pointwise on a single particle wavefunction $\psi(r)$. The other component is nonlocal. It is a projection operator that can be represented as

$$V_{nonlocal}(r) = \frac{1}{\Delta} \sum_{i=1}^{\ell} w_i(r) w_i(r)^*,$$

where Δ is a normalization factor, and the rank of the projection operator ℓ is the total number of quantum number triplets (n, l, m) associated with all atoms in the system, where n is the major electron shell number of and (l, m) represent a angular momentum.

In KSSOLV, the local ionic potential can be obtained by calling function getvion, i.e.

vlocal = getvion(pseudovar),

where vlocal is a 3D array that contains the potential define at each real space sampling grid points.

The nonlocal potential can be computed by calling the function getwq, i.e.

wqmat = getwq(pseudovar),

where wq is a n_g by ℓ matrix and n_g is the number of planewave basis used to represent each single particle wavefunction.

Both vlocal and wqmat are computed when a Ham object is constructed initially using

H = Ham(mol);

for some properly defined Molecule object mol. These arrays are saved as different attributes of H.

4.3 The finite dimensional Kohn-Sham problem

If we let $X \equiv (x_1, x_2, ..., x_{n_e}) \in \mathbb{C}^{n \times n_e}$ be a matrix that contains n_e discretized wavefunctions, the approximation to the kinetic energy (1.5) can also be expressed by

$$\hat{E}_{kin} = \frac{1}{2} \operatorname{trace}(X^* L X), \tag{4.8}$$

where L is a finite-dimensional representation of the Laplacian operator in the planewave basis. Due to the periodic boundary condition imposed in our problem, L is a block circulant matrix with circulant blocks that can be decomposed as

$$L = F^* D_a F, (4.9)$$

where F is the discrete Fourier transform matrix used in (4.7), and D_g is a diagonal matrix with $||g_j||^2$ on the diagonal [5]. It follows from (4.7) and (4.9) that the kinetic energy term (4.8 can be evaluated alternatively as

$$\frac{1}{2} \sum_{\ell=1}^{n_e} \sum_{j=1}^{n_g} ||g_j c_j^{(\ell)}||^2.$$

In the planewave basis, the convolution that appears in the third term of (1.5) may be viewed as the $L^{-1}\rho(X)$, where $\rho(X) = \operatorname{diag}(XX^*)$. (To simplify notation, we will drop X in $\rho(X)$ in the following.) However, since L is singular (due to the periodic boundary condition), its inverse does not exist. Similar singularity issues appear in the planewave representation of the pseudopotential and the calculation of the ion-ion interaction energy. However, it can be shown that the net effects of these singularities cancel out for a system that is electrically neutral [10, 21]. Thus, one can simply remove these singularities by replacing $L^{-1}\rho$ with $L^{\dagger}\rho$, where L^{\dagger} is the pseudo-inverse of L defined as

$$L^{\dagger} = F^* D_a^{\dagger} F,$$

where D_g^{\dagger} is a diagonal matrix whose diagonal entries (d_i) are

$$d_j = \begin{cases} ||g_j||^{-2} & \text{if } g_j \neq 0; \\ 0 & \text{otherwise.} \end{cases}$$

Consequently, the third term in (1.5), which corresponds to an approximation to the Coulomb potential, can be evaluated as

$$\hat{E}_{coul} = \rho^T L^{\dagger} \rho = [F \rho]^* D_q^{\dagger} [F \rho].$$

However, removing these singularities results in a constant shift of the total energy, for which a compensation must be made. It has been shown in [10] that this compensation can be calculated by adding a term $E_{\rm rep}$ that measures the degree of repulsiveness of the local pseudopotential with a term that corresponds to the nonsingular part of ion-ion potential energy. Because the second term can be evaluated efficiently by using a technique originally developed by Ewald [6], it is denoted by $E_{\rm Ewald}$. Both $E_{\rm rep}$ and $E_{\rm Ewald}$ can be computed once and for all in a DFT calculation. We will not go into further details of how they are computed since they do not play any role in the algorithms we will examine in this section. In KSSOLV, the $E_{\rm Ewald}$ calculation is performed in the function getewald.m and $E_{\rm rep}$ is calculated in the function getealpht.m Both functions are call once at the beginning of scf.m, dcm.m, trdcm.m etc.

To summarize, the use of a planewave basis allows us to define a finite-dimensional approximation to the total energy functional (1.5) as

$$\hat{E}_{total}(X) = \text{trace}[X^*(\frac{1}{2}L + \hat{V}_{ion})X] + \frac{1}{2}\rho^T L^{\dagger}\rho + \rho^T \epsilon_{xc}(\rho) + E_{\text{Ewald}} + E_{\text{rep}}, \tag{4.10}$$

where \hat{V}_{ion} denotes the ionic pseudopotentials sampled on the suitably chosen Cartesian grid of size $n_1 \times n_2 \times n_3$.

It is easy to verify that the KKT condition associated with the constrained minimization problem

$$\min_{X^*X=I} \hat{E}_{total}(X) \tag{4.11}$$

is

$$H(X)X - X\Lambda_{n_e} = 0,$$

$$X^*X = I,$$
(4.12)

where

$$H(X) = L + \hat{V}_{ion} + \text{Diag}(L^{\dagger}\rho) + \text{Diag}(\mu_{xc}(\rho)), \tag{4.13}$$

 $\mu_{xc}(\rho) = d\epsilon_{xc}(\rho)/d\rho$, and Λ_{n_e} is a $n_e \times n_e$ symmetric matrix of Lagrangian multipliers. Because $\hat{E}_{total}(X) = \hat{E}_{total}(XQ)$ for any orthogonal matrix $Q \in \mathbb{C}^{n_e \times n_e}$, we can always choose a particular Q such that Λ_{n_e} is diagonal. In this case, Λ_{n_e} contains n_e eigenvalues of H(X). We are interested the n_e smallest eigenvalues and the invariant subspace X associated with these eigenvalues. Because the Hamiltonian matrix H depends on the invariant subspace to be computed, (4.12) is often called a nonlinear eigenvalue problem.

4.4 SCF

Currently, the most widely used algorithm for solving (4.12) is the self-consistent field (SCF) iteration which we outline in Figure 4.1 for completeness.

SCF Iteration

Input: An initial guess of the wavefunction $X^{(0)} \in \mathbb{C}^{n \times n_e}$, pseudopotential:

Output: $X \in \mathbb{C}^{n \times n_e}$ such that $X^*X = I_{n_e}$ and columns of X spans the invariant subspace associated with the smallest n_e eigenvalues of H(X) defined in (4.13).

- 1. for $i = 1, 2, \dots$ until convergence
- 2. Form $H^{(i)} = H(X^{(i-1)})$;
- 3. Compute $X^{(i)}$ such that $H^{(i)}X^{(i)} = X^{(i)}\Lambda^{(i)}$, and $\Lambda^{(i)}$ contains the n_e smallest eigenvalues of $H^{(i)}$;
- 4. end for

Figure 4.1: The SCF iteration

In [33], we viewed the SCF iteration as an indirect way to minimize \hat{E}_{total} through the minimization of a sequence of quadratic surrogate functions of the form

$$q(X) = \frac{1}{2} \operatorname{trace}(X^* H^{(i)} X),$$
 (4.14)

on the manifold $X^*X = I_{n_e}$. This constrained minimization problem is solved in KSSOLV by running a small number of locally optimal preconditioned conjugate gradient (LOBPCG) iterations [13].

Since the surrogate function share the same gradient with \hat{E}_{total} at $X^{(i)}$, i.e.,

$$\nabla \hat{E}_{total}(X)_{|X=X^{(i)}} = H^{(i)}X^{(i)} = \nabla q(X)_{|X=X^{(i)}},$$

moving along a descent direction associated with q(X) is likely to produce a reduction in \hat{E}_{total} . However, because gradient information is local, there is no guarantee that the minimizer of q(X), which may be far from $X^{(i)}$, will yield a lower \hat{E}_{total} value. This observation partially explains why SCF often fails to converge. It also suggests at least two ways to improve the convergence of SCF.

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4.4.1 Charge mixing

One possible improvement is to replace the simple gradient-matching surrogate q(X) with another quadratic function whose minimizer is more likely to yield a reduction in \hat{E}_{total} . In practice, this alternative quadratic function is often constructed by replacing the charge density $\rho^{(i)}$ in (4.13) with a linear combination of m previously computed charge densities, i.e.,

$$\rho_{mix} = \sum_{j=0}^{m-1} \alpha_j \rho_{\text{in}}^{(i-j)} + \beta_j \rho_{\text{out}}^{(i-j)}, \tag{4.15}$$

where $\rho_{\text{in}}^{(i-j)}$ is the *input* charge density used to construct H at the i-jth iteration, $\rho_{\text{out}}^{(i-j)}$ is the *output* charge density obtained from

$$\rho_{\text{out}}^{(i-j)} = f(H(\rho_{\text{in}}^{(i-j)})) \tag{4.16}$$

and $f(H(\rho_{\text{in}}))$ is a function that we will discuss in the next section. Here, we can simply think of $f(H(\rho_{\text{in}}^{(i-j)}))$ as the diagonal part of the projection matrix XX^* , where columns of the $n \times n_e$ matrix X are approximations to the eigenvectors of $H(\rho_{\text{in}}^{(i-j)})$ corresponding to the n_e smallest eigenvalues.

The coefficients $\alpha_0, \alpha_2, ..., \alpha_{i-m+1}$ and $\beta_0, \beta_2, ..., \beta_{i-m+1}$ in (4.15) are chosen to reduce the lack of self consistency between the input charge density used to define the surrogate function and the output charge density defined by (4.16). Because ρ_{mix} is a mixture of previous input and output change densities, this type of surrogate construction technique is often called *charge mixing*.

Pulay mixing

A particular charge mixing scheme first proposed by Pulay for Hartree-Fock calculations [22, 23] chooses the coefficients in (4.15) by solving the following constrained minimization problem

$$\min_{\rho} \|f(\rho) - \rho\|_2^2. \tag{4.17}$$

within a subspace that consists of input charge densities computed in previous SCF iterations. We will derive Pulay's scheme in this section. Note that in computational chemistry, Pulay mixing is referred to as the method of *direct inversion of iterative subspace* or simply DIIS.

To derive the Pulay mixing scheme, let us assume that ρ_* is the exact solution to the equation

$$\rho = f(\rho). \tag{4.18}$$

Due to the way f is defined, it is easy to show that $\rho_* \in \mathbb{R}^+$ and $\|\rho_*\|_1 = n_e$.

Suppose we already have a few approximations to ρ , say, $\rho_1, \rho_2, ..., \rho_k$. Pulay's mixing scheme seeks an optimal approximation ρ_{opt} to (4.17) from

$$S \equiv \{\rho | \rho = \sum_{i=1}^{k} \alpha_i \rho_i, \sum_{i=1}^{k} \alpha_i = 1\}.$$

$$(4.19)$$

It follows form the Taylor's expansion that

$$f(\rho_{\text{opt}}) = f(\rho_*) + J_*(\rho_{\text{opt}} - \rho_*) + \text{higer order terms in } (\rho_{\text{opt}} - \rho_*)$$
 (4.20)

$$= \rho_* + J_*(\rho_{\text{opt}} - \rho_*) + \text{higer order terms in } (\rho_{\text{opt}} - \rho_*), \tag{4.21}$$

where J_* is the Jacobian matrix of f evaluated at ρ_* .

If we ignore the higher order terms in (4.21), we obtain

$$f(\rho_{\text{opt}}) - \rho_{\text{opt}} \approx \rho_* - \rho_{\text{opt}} + J_*(\rho_{\text{opt}} - \rho_*). \tag{4.22}$$

Substituting $\rho_{\text{opt}} = \sum_{i} \alpha_{i} \rho_{i}$ into the right hand side of (4.22), we obtain

$$f(\rho_{\text{opt}}) - \rho_{\text{opt}} \approx \rho_* - \sum_{i=1}^k \alpha_i \rho_i - J_* \left[\rho_* - \sum_{i=1}^k \alpha_i \rho_i \right]$$

$$= \sum_{i=1}^k \alpha_i (\rho_* - \rho_i) - J_* \left[\sum_{i=1}^k \alpha_i (\rho_* - \rho_i) \right]$$

$$= \sum_{i=1}^k \alpha_i \left[(\rho_* - \rho_i) - J_* (\rho_* - \rho_i) \right]. \tag{4.23}$$

It follow from the Taylor expansion of $f(\rho_i)$ at ρ_* that

$$\Delta \rho_i \equiv f(\rho_i) - \rho_i \approx \rho_* - \rho_i - J_*(\rho_* - \rho_i). \tag{4.24}$$

It now follows from (4.23) and (4.24) that

$$f(\rho_{\text{opt}}) - \rho_{\text{opt}} \approx \sum_{i=1}^{k} \alpha_i \Delta \rho_i.$$
 (4.25)

Therefore, with linear approximation, the best solution to (4.17) from the subspace S can be obtained by solving the following equality constrained quadratic program

$$\min_{\alpha_i} \quad \left\| \sum_{i=1}^k \alpha_i \Delta \rho_i \right\|_2^2
\text{s.t.} \quad \sum_{i=1}^k \alpha_i = 1$$
(4.26)

The solution to (4.26), which can be obtained by eliminating the constraint and solving a standard linear least squares problem, gives the coefficients used in Pulay mixing.

Broyden mixing

Another class of charge mixing algorithms can be derived by formulating the Kohn-Sham DFT problem as a system of nonlinear equations that involves only the unknown electron charge density ρ and the chemical potential μ . These equations can be written as

$$F(\rho,\mu) = \begin{pmatrix} r(\rho,\mu) \\ \nu(\rho,\mu) \end{pmatrix} = \begin{pmatrix} \rho - \operatorname{diag}[f_{\beta,\mu}(H(\rho))] \\ \operatorname{trace}[f_{\beta,\mu}(H(\rho))] - n_e \end{pmatrix} = 0, \tag{4.27}$$

where $f_{\beta,\mu}(t)$ is the Fermi-Dirac distribution function

$$f_{\beta,\mu,}(t) = \frac{1}{1 + e^{\beta(t-\mu)}},$$

 β is a fixed parameter proportional to the reciprocal of the temperature, H is the Hamiltonian of the system, and n_e is the number of electrons.

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In theory, we can solve (4.27) by using a standard Newton method, which constructs approximations to ρ and μ iteratively by

$$\begin{pmatrix} \rho^{(k+1)} \\ \mu^{(k+1)} \end{pmatrix} = \begin{pmatrix} \rho^{(k)} \\ \mu^{(k)} \end{pmatrix} - \hat{J}_k^{-1} \begin{pmatrix} r_k \\ \nu_k \end{pmatrix}, \tag{4.28}$$

where \hat{J}_k is the Jacobian matrix of $F(\rho,\mu)$ evaluated at $(\rho^{(k)},\mu^{(k)}),\ r_k\equiv r(\rho^{(k)},\mu^{(k)})$ and $\nu_k \equiv \nu(\rho^{(k)}, \mu^{(k)}).$

Because J_k and its inverse are difficult to obtain in practice, quasi-Newton methods that replace J_k with an easy-to-evaluate approximation are more appropriate. Note that once $\rho^{(k)}$ is available, it is easy to find $\mu^{(k)}$ such that the second equation in

(4.27) is satisfied, due to the monotonicity of $\nu(\rho,\mu)$ with respect to μ . Therefore, we can effectively focus on the first equation in (4.28), i.e.,

$$\rho^{(k+1)} = \rho^{(k)} - \tau J_k^{-1} r_k, \tag{4.29}$$

where J_k^{-1} is the (1,1) block of \hat{J}_k^{-1} and τ is an appropriate step length. Different approximations to J_k or J_k^{-1} yield different quasi-Newton algorithms. These approximations can be viewed as different charge mixing schemes for accelerating the selfconsistent field (SCF) iteration, and are frequently discussed in the physics literature [7, 15]. For example, when $J_k^{-1} = I$, (4.29) reduces to the simple mixing scheme,

$$\rho^{(k+1)} = (1 - \tau)\rho^{(k)} + \tau \operatorname{diag}[f_{\beta,\mu}(H(\rho^{(k)}))],$$

used in the SCF iteration.

More sophisticated quasi-Newton updating schemes can be devised by using Broyden techniques [?] to construct better approximations to J_k or J_k^{-1} . Such techniques yield iterative procedures in which an approximation to J_k or J_k^{-1} is obtained by performing a sequence of low-rank modifications to some initial approximation using a recursive formula [8, 16].

1. For example, we can construct an approximation to J_k^{-1} by solving the following constrained minimization problem

$$\min_{C} \frac{1}{2} ||C - C_{k-1}||_{F}^{2}$$
s.t $S_{k} = CY_{k}$, (4.30)

where C_{k-1} is the approximation to J_{k-1}^{-1} , S_k and Y_k are defined as

$$S^{(k)} = (s^{(k)} s^{(k-1)} \cdots s^{(k-\ell)}), \tag{4.31}$$

$$Y^{(k)} = (y^{(k)} y^{(k-1)} \cdots y^{(k-\ell)}), \tag{4.32}$$

and

$$s^{(k)} = \rho^{(k)} - \rho^{(k-1)}, \quad y^{(k)} = r(\rho^{(k)}, \mu^{(k)}) - r(\rho^{(k-1)}, \mu^{(k)}). \tag{4.33}$$

The solution to (4.30) is

$$C_k = C_{k-1} + (S_k - C_{k-1}Y_k)Y_k^{\dagger}, \tag{4.34}$$

where $Y_k^{\dagger} = (Y_k^T Y_k)^{-1} Y_k^T$ is the pseudoinverse of Y_k .

Substituting (4.34) for J_k^{-1} in (4.29) yields the following recursive formula for updating $\rho^{(k)}$:

$$\rho^{(k+1)} = \rho^{(k)} - \tau C_{k-1} r_k - \tau (S_k - C_{k-1} Y_k) Y_k^{\dagger} r_k, \tag{4.35}$$

It is easy to show that when $\tau = 1$, (4.35) is equivalent to

$$\rho^{(k+1)} = \rho^{(k)} - C_0 r_k - \sum_{j=1}^k Y_j Y_j^{\dagger} r_k + \sum_{j=1}^k S_j Y_j^{\dagger} r_k.$$

When $\ell = 0$, (4.35) reduces to the standard Broyden's method of the second type.

2. We can also construct an approximation to J_{k-1} directly by solving

$$\min_{B} \frac{1}{2} ||B - B_{k-1}||_{F}^{2}$$
s.t
$$BS_{k} = Y_{k},$$
(4.36)

where B_{k-1} is the approximation to J_{k-1} , S_k and Y_k are the same as those defined in (4.31) and (4.32) respectively. The solution to (4.36) yields the following updating formula for B_k , i.e.,

$$B_k = B_{k-1} - B_{k-1} S_k S_k^{\dagger} + Y_k S^{\dagger}. \tag{4.37}$$

In order to obtain an approximation to J_k^{-1} , which is required in (4.29), we invoke the Sherman-Morrison-Woodbury formula that yields

$$C_k = C_{k-1} + (S_k - C_{k-1}Y_k)(S_k^T C_{k-1}Y_k)^{-1} S_k^T C_{k-1}.$$

A number of practical issues must be addressed in the implementation of Broyden schemes. These issues include

- The choice of the initial guess to J or J^{-1} . A commonly used initial guess is σI for some $0 < \sigma < 1$. However, it is not clear how one should select an optimal σ .
- The determination of an optimal rank for the Jacobian update. The standard Broyden method performs a rank-1 update of J_k or J_k^{-1} at each iteration. Our preliminary experiments indicate that a multiple rank update [8] is often more effective. However, it is not yet clear what the optimal rank should be.
- The choice of line search parameter τ in the Broyden update $\rho^{(k+1)} = \rho^{(k)} \gamma J_k^{-1} r_k$.
- The choice of when to restart Broyden schemes to minimize storage requirements.

Anderson mixing

A special type of mixing scheme called Anderson mixing [1] can be derived from (4.35) by simply replacing C_{k-1} with σI and setting $\tau = 1$. This yields the much simpler updating formula

$$\rho^{(k+1)} = \rho^{(k)} - \sigma r_k - (S_k - \sigma Y_k) Y_k^{\dagger} r_k, = (\rho^{(k)} - S_k g) - \sigma (r_k - Y_k g),$$

where $g = Y_k^{\dagger} r_k$.

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Other mixing schemes

Other mixing scheme include *Kerker mixing* [11], *Thomas-Fermi mixing* [24]. Charge mixing is often quite effective in practice for improving the convergence SCF even though its convergence property is still not well understood. In some cases, charge mixing may fail also.

Choosing a charge mixing scheme in KSSOLV

KSSOLV supports a number of charge mixing schemes. The default choice is the Anderson mixing scheme originally developed in [1] and further analyzed in [7, 8]. To change the default choice of charge mixing, one must first construct an option structure by using

```
option = setksopt;
```

This option structure can be modified by using, for example,

```
option = setksopt('mixtype', 'broyden');
```

which replace the default charge mixing scheme by a Broyden mixing scheme [16].

4.4.2 Trust Region

Another way to improve the convergence of the SCF iteration is to impose an additional constraint to the surrogate minimization problem (4.14) so that the wavefunction update can be restricted within a small neighborhood of the gradient matching point $X^{(i)}$, thereby ensuring a reduction of the total energy function as we minimize the surrogate function. In [33], we showed that the following constraint

$$||XX^* - X^{(i)}X^{(i)^*}||_F^2 \le \Delta$$

is preferred because it is rotationally invariant (i.e., post-multiplying X by an unitary matrix does not change the constraint), and because adding such a constraint does not increase the complexity of solving the surrogate minimization problem. It is not difficult to show [33] that that solving the following constrained minimization

$$\min_{XX^* = I} q(X)$$

$$\|XX^* - X^{(i)}X^{(i)^*}\|_F^2 \le \Delta$$
(4.38)

is equivalent to solving a low rank perturbed linear eigenvalue problem

$$\[H(X^{(i)}) - \sigma X^{(i)} X^{(i)*} \] X = X\Lambda, \tag{4.39}$$

where σ is essentially the Lagrange multiplier for the inequality constraint in (4.38) and Λ is a diagonal matrix that contains the n_e smallest eigenvalues of the low rank perturbed $H^{(i)}$. When σ is sufficiently large (which corresponds to a trust region radius Δ that is sufficiently small), the solution to (4.39) is guaranteed to produce a reduction in $\hat{E}_{total}(X)$.

4.4.3 LOBPCG for Solving Linear Eigenvalue Problems

As we mentioned earlier, Step 3 of the SCF iteration in which one must compute approximations to desired eigenvectors of a fixed Hamiltonian must is solved by the LOBPCG algorithm [13]. The algorithm is an iterative procedure that seeks the minimum of the quadratic surrogate function (4.14) by projecting the function into a sequence of low dimensional subspaces. The minimizer within each subspace, which can be obtained by solving a small generalized eigenvalue problem produces the optimal search direction and "step length" simultaneously. We will refer users to [13] for details of the algorithm. We would like to point out that a user can easily experiment with this algorithm in KSSOLV after a Ham object H has been properly constructed. To invoke LOBPCG, one can use

where H is a predefined Ham object, XO is a Wavefun object that contains the initial guess to the minimizer of (4.14), prec is a preconditioner (which should be constructed as a Wavefun object), tol is the convergence tolerance a user must set, and maxit is the maximum number of LOBPCG iterations allowed. Upon completion, lobpcg returns the approximation to the desired wavefunction X, the corresponding eigenvalue approximation stored in the array lambda, and residual norms (RVEC) and eigenvalues approximation (LVEC) computed at each iteration.

4.5 DCM

Instead of focusing on Kohn-Sham equations (4.12) and minimizing the total energy indirectly in the SCF iteration, we can minimize the total energy directly in an iterative procedure that involves finding a sequence of search directions along which $\hat{E}_{total}(X)$ decreases and computing an appropriate step length. In most of the earlier direct minimization methods developed in [2, 9, 15, 18, 26, 29, 30], the search direction and step length computations are carried out separately. This separation sometimes results in slow convergence. We recently developed a new direct constrained minimization (DCM) algorithm [32, 33] in which the search direction and step length are obtained simultaneously in each iteration by minimizing the total energy within a subspace spanned by columns of

$$Y = \left(X^{(i)}, M^{-1}R^{(i)}, P^{(i-1)}\right),\,$$

where $X^{(i)}$ is the approximation to X obtained at the ith iteration, $R^{(i)} = H^{(i)}X^{(i)} - X^{(i)}\Lambda^{(i)}$, M is a hermitian positive definite preconditioner, and $P^{(i-1)}$ is the search direction obtained in the previous iteration. It was shown in [32] that solving the subspace minimization problem is equivalent to computing the eigenvectors G associated with the n_e smallest eigenvalues of the following nonlinear eigenvalue problem

$$\hat{H}(G)G = BG\Omega, \quad G^*BG = I, \tag{4.40}$$

where

$$\hat{H}(G) = Y^* \left[\frac{1}{2} L + V_{ion} + \text{Diag} \left(L^{\dagger} \rho(YG) \right) + \text{Diag} \left(\mu_{xc}(\rho(YG)) \right) \right] Y, \tag{4.41}$$

and $B = Y^*Y$.

Because the dimension of $\hat{H}(G)$ is at most $3n_e \times 3n_e$, which is normally much smaller than that of H(X), it is relatively easy to solve (4.40) by, for example, a trust region enabled SCF (TRSCF) iteration. We should note that it is not necessary to solve (4.40) to full accuracy in the early stage of the DCM algorithm because all we need is a G that yields sufficient reduction in the objective function.

Once G is obtained, we can update the wave function by

$$X^{(i+1)} \leftarrow YG$$
.

The search direction associated with this update is defined, using the MATLAB submatrix notation, to be

$$P^{(i)} \equiv Y(:, n_e + 1: 3n_e)G(n_e + 1: 3n_e, :).$$

A complete description of the constrained minimization algorithm is shown in Figure 4.2. We should point out that solving the projected optimization problem in Step 7 of the algorithm requires us to evaluate the projected Hamiltonian $\hat{H}(G)$ repeatedly as we search for the best G. However, since the first two terms of \hat{H} do not depend on G, they can be computed and stored in advance. Only the last two terms of (4.41) need to be updated. These updates require the charge density, the Coulomb and the exchange-correlation potentials to be recomputed.

4.6 Options for SCF and DCM functions

KSSOLV allows one to change a number of parameters used in either SCF or DCM functions. These parameters can be changed by defining and modifying an option structure that can be created by simply calling

```
options = setksopt;
```

This call creates a structure named options that contains a number of default parameters used in SCF and DCM functions. To see these parameters, one can simply type options at the MATLAB prompt without a semicolon at the end.

options =

```
verbose: 'on'
maxscfiter: 10
maxdcmiter: 10
maxinerscf: 3
maxcgiter: 10
scftol: 1.0000e-08
dcmtol: 1.0000e-08
cgtol: 1.0000e-06
mixtype: 'anderson'
mixdim: 9
betamix: 0.8000
brank: 1
X0: []
rho0: []
degree: 10
```

```
Algorithm: A Constrained Minimization Algorithm for Total Energy
Minimization
 Input:initial set of wave functions X^{(0)} \in \mathbb{C}^{n \times n_e}; ionic pseudopo-
          tential; a preconditioner M;
 Output \in \mathbb{C}^{n \times k} such that the Kohn-Sham total energy functional
          E_{total}(X) is minimized and X^*X = I_k.
          Orthonormalize X^{(0)} such that X^{(0)*}X^{(0)} = I_k;
 1.
          for i = 0, 1, 2, \dots until convergence
 2.
             Compute \Theta = X^{(i)*}H^{(i)}X^{(i)};
 3.
             Compute R = H^{(i)}X^{(i)} - X^{(i)}\Theta,
 4.
             if (i > 1) then
 5.
               Y \leftarrow (X^{(i)}, M^{-1}R, P^{(i-1)})
               Y \leftarrow (X^{(i)}, M^{-1}R);
             endif
             B \leftarrow Y^*Y:
 6.
             Find G \in \mathbb{C}^{2n_e \times n_e} or \mathbb{C}^{3n_e \times n_e} that minimizes E_{total}(YG)
 7.
             subject to the constraint G^*BG = I_{n_e};
             Set X^{(i+1)} = YG;
 8.
             if (i > 1) then
 9.
               P^{(i)} \leftarrow Y(:, n_e + 1: 3n_e)G(n_e + 1: 3n_e, :);
               P^{(i)} \leftarrow Y(:, n_e + 1 : 2n_e)G(n_e + 1 : 2n_e, :);
             endif
 10.
          end for
```

Figure 4.2: A Direct Constrained Minimization Algorithm for Total Energy Minimization

Each field can be modified by simply resetting the value of the field through an assignment statement. For example, to change the maximum number of SCF iterations allowed, we can simply use

```
opts.maxscfiter = 20.
```

This parameter can also be modified by calling **setksopt** again with appropriate argument pairs. For example,

```
opts = setksopt(opts, 'maxscfiter', 20);
```

achieves the same effect. The advantage of calling setksopt is that the function checks the validity of the structure field and its value.

Table 4.1 lists all options that can be defined by **setksopt** and explains what they are used for.

option name	allowed value	purpose
verbose	'on' or 'off'	display diagonostic info
maxscfiter	positive integer	maximum SCF iterations allowed
maxdcmiter	positive integer	maximum DCM iterations allowed
maxinerscf	positive integer	maximum number of inner SCF iterations al-
		lowed in the DCM algorithm
maxcgiter	positive integer	maximum LOBPCG iterations allowed
scftol	positive float	convergence tolerance for SCF
dcmtol	positive float	convergence tolerance for DCM
cgtol	positive float	convergence tolerance for LOBPCG
mixtype	'anderson'	charge or potential mixing types
	'broyden'	
	'broyden1'	
	'pulay' 'kerker'	
	'pulay+kerker'	
mixdim	positive integer	maximum number of previous potentials to be
1	0	mixed
betamix	positive float	The damping parameter used in Anderson
brank	nagitive integer	and Broyden mixing The rank of the Broyden update
	positive integer	v -
ХО	Wavefun object	An initial guess of the wavefunctions
rho0	3-D array	An initial guess of the charge. To restart
		an SCF or DCM calculation using previously
		computed results One should pass in both the
	•,•	previously obtained wavefunction and charge.
degree	positive integer	The degree of the Chebyshev polynomial used
		in chebyscf.

Table 4.1: Options that can be defined by calling setksopt

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