# Chapter 4 – Elements of the Wave Function Formalism. Schrödinger Equation. Infinite Potential Well

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Wave Function Physical Quantities Schrödinger Equation Wave Function and its Probabilistic Interpretation Trajectory, Average Position, and Uncertainty of Position Stationary States Mathematical Appendix (I). Inner Product

#### Wave Function

#### Wave Function

In quantum mechanics the state of a particle is represented by a **wave-function**. It is usually denoted by the Greek letter  $\Psi$ . The wave function is in general complex–valued. For the wave function in the position representation in 1D and in 3D, respectively,

$$\Psi = \Psi(x, t), \qquad \qquad \Psi = \Psi(\overline{r}, t),$$

#### Interpretation (Born)

$$|\Psi(x,t)|^2 dx = \Psi^*(x,t)\Psi(x,t)dx$$

$$x \quad x+dx$$

probability of finding the particle in (x, x + dx) at instant t

$$|\Psi(\overline{r},t)|^2 d^3r = \Psi^*(\overline{r},t)\Psi(\overline{r},t)d^3r$$



probability of finding the particle in the element of volume  $d^3r$  around  $\bar{r}$  at instant t

# Probabilistic Interpretation of Wave Function

**Note.** In the language of probability: x and  $\bar{r}$  are the values of a random variable, whereas  $|\Psi(x,t)|^2$  and  $|\Psi(\bar{r},t)|^2$  are the corresponding probability density functions.

Consequently

$$\int_{a}^{b} |\Psi(x,t)|^2 dx$$

is the probability of finding the particle in an interval [a, b], and

$$\iiint\limits_A |\Psi(\bar{r},t)|^2 \mathrm{d}^3 r$$

is the probability of finding the particle in a solid region A.

#### Normalization

Functions satisfying the condition

$$\int\limits_{-\infty}^{\infty} |\Psi(x,t)|^2 \mathsf{d}x = 1 \qquad ext{or} \qquad \iiint\limits_{\mathbb{R}^3} |\Psi(ar{r},t)|^2 \mathsf{d}^3 r = 1$$

are said to be normalized.

- If  $\int_{-\infty}^{\infty} |\Psi(x,t)|^2 dx < \infty$  or  $\iiint_{\mathbb{R}^3} |\Psi(\bar{r},t)|^2 d^3r < \infty$ , then  $\Psi$  is said to be **square-integrable**.
- Square-integrable functions (with the usual operations of function addition and multiplication by a scalar) form a linear space denoted as  $\mathcal{L}^2(\mathbb{R})$  or  $\mathcal{L}^2(\mathbb{R}^3)$ .

# Trajectory, Average Position, and Uncertainty of Position

*Implication.* Quantum mechanics gives up on the idea of trajectory in the classical sense.



Hence, we will rather use the **average position** of the particle in the state represented by the function  $\Psi$  (furthermore we will give the formulas in the 1D-version)

$$\langle x \rangle_{\Psi} = \int_{-\infty}^{\infty} x |\Psi(x,t)|^2 dx = \int_{-\infty}^{\infty} \Psi^*(x,t) x \Psi(x,t) dx.$$

Consequently, the variance of the position is given by

$$(\Delta_x^{(\Psi)})^2 = \langle (x - \langle x \rangle_{\Psi})^2 \rangle_{\Psi} = \langle x^2 - 2x \langle x \rangle_{\Psi} + \langle x \rangle_{\Psi}^2 \rangle_{\Psi}$$
$$= \langle x^2 \rangle_{\Psi} - 2\langle x \rangle_{\Psi} \langle x \rangle_{\Psi} + \langle x \rangle_{\Psi}^2 = \langle x^2 \rangle_{\Psi} - \langle x \rangle_{\Psi}^2$$

The standard deviation (a.k.a. dispersion or uncertainty) of the position is  $\Delta_x^{(\Psi)} = \sqrt{\langle x^2 \rangle_{\Psi} - \langle x \rangle_{\Psi}^2}$ .

# Stationary States

If  $|\Psi(x)|^2 = \Phi(x)$ , i.e. the modulus square does not depend on time, then the state represented by  $\Psi$  is called a stationary state.

**Observation.** For a stationary state,  $\langle x \rangle_{\Psi}$  and  $\Delta_x^{(\Psi)}$  are time-independent.

**Example.** The wave function  $\Psi(x,t)=e^{-\frac{i}{\hbar}Et}f(x)$ , where E is a constant and f is a square-integrable function, is representing a quantum system (particle) in a stationary state.

Wave Function Physical Quantities Schrödinger Equation Wave Function and its Probabilistic Interpretation Trajectory, Average Position, and Uncertainty of Position Stationary States Mathematical Appendix (I). Inner Product

Mathematical Appendix (I). Inner Product

#### Inner Product

In the space  $\mathcal{L}^2(\mathbb{R})$  the **inner product**  $\langle \cdot, \cdot \rangle$  (an equivalent of the scalar product in the space of usual vectors) can be introduced as

$$\langle \Psi, \Phi \rangle \stackrel{\text{def}}{=\!=\!=} \int_{-\infty}^{\infty} \Psi^*(x, t) \Phi(x, t) dx,$$

where  $\Psi^*(x,t)\Phi(x,t)$  is in general a complex number.

**Observation** (exercise). For any  $\Psi$ ,  $\Phi_1$  and  $\Phi_2 \in \mathcal{L}^2(\mathbb{R})$  and  $\alpha \in \mathbb{C}$ :

$$\langle \Psi, \Phi_1 + \Phi_2 \rangle \ = \ \langle \Psi + \Phi_1 \rangle + \langle \Psi, \Phi_2 \rangle \tag{1}$$

$$\langle \Psi, \alpha \Phi \rangle = \alpha \langle \Psi, \Phi \rangle \tag{2}$$

$$\langle \Psi, \Phi \rangle = \langle \Phi, \Psi \rangle^*$$
 (3)

$$\langle \Psi, \Psi \rangle \ge 0; \qquad \langle \Psi, \Psi \rangle = 0 \Leftrightarrow \Psi \equiv 0$$
 (4)

Comment\*. Any functional satisfying the four axioms (1)–(4) is an inner product.

The formula  $\langle \Psi, \Phi \rangle$  quoted above is just a particular realization of the inner product in  $\mathcal{L}^2(\mathbb{R})$ .

#### Note

- (3) and (2) imply  $\langle \alpha \Psi, \Phi \rangle = \alpha^* \langle \Psi, \Phi \rangle$ .
- \* (4) induces the norm  $\|\Psi\| = \sqrt{\langle \Psi, \Psi \rangle}$  (and hence the metric  $d(\Psi, \Phi) = \|\Psi \Phi\|$ ).
- \*\* A linear space with an inner product is called a unitary spaceand a unitary space that is complete in the norm-introduced metric is called a Hilbert space.
- \*\*\* Any Hilbert space is a Banach space, but a Banach space is not necessarily a Hilbert space.

#### **Examples**. Rewrite using the inner product notation.

(a) Average position  $\langle x \rangle_{\Psi}$ , where x is a real number.

$$\int_{-\infty}^{\infty} \Psi^*(x,t) x \Psi(x,t) dx = \langle \Psi, x \Psi \rangle$$

(b) The normalization condition.

$$\langle \Psi, \Psi \rangle = 1$$

Hermitian Operators in Quantum Mechanics. Examples Average Values of Physical Quantities Hamiltonian

Physical Quantities

# Physical Quantities and Hermitian Operators

In quantum mechanics, physical (i.e. measurable) quantities are represented by **Hermitian operators**.

What is an operator?

An operator transforms one function into another:  $\hat{A}\Psi=\Phi.$  We will use hats  $\hat{\ }$  to denote operators.

#### Examples.

$$\frac{\partial}{\partial x}(x^4) = 3x^3$$
  $\hat{\alpha}f(x) = \alpha \cdot f(x)$   $\hat{Q}f(x) = (f(x))^*$ 

differentiation multiplication by a scalar complex conjugation

#### Hermitian Operators

If  $\langle \hat{B}\Psi, \Phi \rangle = \langle \Psi, \hat{A}\Phi \rangle$  for all  $\Psi, \Phi \in \mathcal{L}^2(\mathbb{R})$ , then  $\hat{B}$  is called the **Hermitian conjugate** of  $\hat{A}$  and it is denoted as  $\hat{A}^{\dagger}$ . (that is  $\langle \hat{A}^{\dagger}\Psi, \Phi \rangle = \langle \Psi, \hat{A}\Phi \rangle$ )

Operator  $\hat{A}$  is said to be **Hermitian**, if  $\hat{A}^{\dagger} = \hat{A}$  (that is  $\langle \hat{A}\Psi, \Phi \rangle = \langle \Psi, \hat{A}\Phi \rangle$  for all  $\Psi$  and  $\Phi$ .)

**Examples**. Which of the given operators are Hermitian?

(a) the position operator (in the position representation)  $\hat{x} = x$ , which is just the multiplication by a (real) number x.

Check whether  $\hat{x} = \hat{x}^{\dagger}$ .

$$\begin{split} \langle \Psi, \hat{x} \Phi \rangle &= \int_{-\infty}^{\infty} \Psi^* \hat{x} \Phi \, \mathrm{d}x = \int_{-\infty}^{\infty} \Psi^* (x \cdot \Phi) \, \mathrm{d}x \stackrel{x = x^*}{=} \int_{-\infty}^{\infty} (x \cdot \Psi)^* \Phi \, \mathrm{d}x \\ &= \int_{-\infty}^{\infty} (\hat{x} \Psi)^* \Phi \, dx = \langle \hat{x} \Psi, \Phi \rangle \end{split}$$

Therefore  $\hat{x}$  is Hermitian.

(b) the momentum operator (in the position representation)

$$\hat{p} = -i\hbar \frac{\partial}{\partial x}$$

where *i* is the imaginary unit  $(i^2 = -1)$ .

\* The form of the momentum operator can be derived by studying infinitesimal translations (c.f. Noether's theorem).

Check whether  $\hat{p}^{\dagger} = \hat{p}$  (i.e. whether  $\hat{p}$  is Hermitian).

$$\begin{array}{lll} \langle \Psi, \hat{p} \Phi \rangle & = & \int_{-\infty}^{\infty} \Psi^* \hat{p} \Phi \, \mathrm{d}x = \int_{-\infty}^{\infty} \Psi^* (-i\hbar \frac{\partial}{\partial x} \Phi) \mathrm{d}x \\ & \stackrel{by \ parts}{=} & [-i\hbar \Psi^* \Phi]_{-\infty}^{+\infty} + \int_{-\infty}^{\infty} (i\hbar \frac{\partial}{\partial x} \Psi^*) \Phi \mathrm{d}x \\ & = & \int_{-\infty}^{\infty} (-i\hbar \frac{\partial}{\partial x} \Psi)^* \Phi \mathrm{d}x = \int_{-\infty}^{\infty} (\hat{p} \Psi)^* \Phi \, \mathrm{d}x = \langle \hat{p} \Psi, \Phi \rangle \end{array}$$

Therefore  $\hat{p}$  is Hermitian.

# Average Value

The average value of a physical quantity A in state  $\Psi$  is

$$\langle A \rangle_{\Psi} = \langle \Psi, \hat{A} \Psi \rangle$$

where  $\hat{A}$  is the corresponding (Hermitian) operator.

#### Note

- $\langle A \rangle_{\Psi}$  is always real (simple exercise: use Hermiticity & properties of the inner product).
- Consequently, the standard deviation  $\frac{\Delta_A^{(\Psi)}}{\Delta_A^{(\Psi)}} = \sqrt{\langle A^2 \rangle_{\Psi} (\langle A \rangle_{\Psi})^2}.$
- $\Delta_x^{(\Psi)} \Delta_p^{(\Psi)} \geqslant \hbar/2$  is known as **Heisenberg Uncertainty Principle** and will be derived later.

# Other Important Operators. The Hamiltonian

(a) kinetic energy operator

$$\hat{K} = \frac{\hat{p}^2}{2m} = \frac{\left(-i\hbar\frac{\partial}{\partial x}\right)^2}{2m} = -\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2}$$

(b) potential energy operator (multiplication by scalar V(x))

$$\hat{V} = V(\hat{x}) = V(x)$$

(c) total energy operator — the **Hamiltonian** 

$$\hat{H} = \hat{K} + \hat{V} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x)$$

Exercise. Argue that they are Hermitian.

(Hint. Use the Hermiticity of  $\hat{p}$  and  $\hat{x}$ .)

Introduction and Motivation Time-Dependent Schrödinger Equation Stationary (Time-Independent) Schrödinger Equation

#### Schrödinger Equation

#### Introduction

	classical mechanics	quantum mechanics
particle's state	position	wave function
	$ar{r}=ar{r}(t)$	$\Psi = \Psi(x,t)$
time-evolution of the state	Newton's equation	
	of motion	?
	$ar{{\sf F}}={\sf d}ar{{\sf p}}/{\sf d} t$	

What should stand at the position of the question mark is the **Schrödinger equation**. The Schrödinger equation is the quantum–mechanical equation of motion determining the time-evolution of particle's state.

# Schrödinger Equation. Motivation (not 'Derivation')

Recall the wave equation of photons (e-m wave)

$$\frac{\partial^2 \widetilde{\xi}}{\partial x^2} = \frac{1}{c^2} \frac{\partial^2 \widetilde{\xi}}{\partial t^2}.$$

For a plane wave  $\widetilde{\xi} = \widetilde{\xi}_0 e^{i(kx - \omega t)}$ , we have  $\begin{cases} \partial^2 \widetilde{\xi} / \partial x^2 = -k^2 \widetilde{\xi} \\ \partial^2 \widetilde{\xi} / \partial t^2 = -\omega^2 \widetilde{\xi} \end{cases}$ . Hence

$$k^2 = \frac{\omega^2}{c^2} \longrightarrow \omega = kc \xrightarrow{p=\hbar k, E=\hbar \omega} E = pc.$$

For non-relativistic particles (e.g. electrons) in a potential force field, the corresponding relationship reads  $E=\frac{p^2}{2m}+V$ , where V is the potential energy. Using de Broglie relations  $E=\hbar\omega$  and  $p=\hbar k$ , it can be rewritten as

$$\hbar\omega = \frac{\hbar^2 k^2}{2m} + V.$$

For a plane wave, the operators that yield the first term on the lhs and the first term on the rhs are  $\hbar\omega \sim i\hbar\frac{\partial}{\partial t}$  and  $\frac{\hbar^2k^2}{2m}\sim -\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2}$ .

### 1D Schrödinger Equation

The state of a particle with mass m, moving in a force field with the potential energy V(x), represented by the wave function  $\Psi(x,t)$  evolves with time according to the **Schrödinger equation**.

$$\boxed{-\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2}\Psi(x,t)+V(x)\Psi(x,t)=i\hbar\frac{\partial}{\partial t}\Psi(x,t)}.$$

Terminology. In QM the term "potential" is commonly used instead of "potential energy".

#### Comments

- Schrödinger equation is a 2<sup>nd</sup> order partial differential equation.
- ② Schrödinger equation is also a linear equation, i.e. if  $\Psi_1$ ,  $\Psi_2$  are solutions to the same Schrödinger equation and  $\alpha$ ,  $\beta$  are any complex numbers, then  $\alpha\Psi_1+\beta\Psi_2$  is also a solution to that equation.

# Digression. Superposition Principle

If  $\Psi_1$  and  $\Psi_2$  are wave functions describing two states of a particle, and  $\alpha$ ,  $\beta$  are any complex numbers, then  $\alpha\Psi_1 + \beta\Psi_2$  is the wave function representing another possible state of that particle.

**Note.**\* Postulating that states of a particle are elements (vectors) from a Hilbert space already implies the superposition principle (Hilbert space is a linear space).

#### Example. Quantum Bit

Recall that a classical bit can be either "not set"  $\boxed{0}$  or "set"  $\boxed{1}$ . On the contrary, for any two-level system with states  $\boxed{0}$  or  $\boxed{1}$ , the quantum bit (**qubit**) that can be defined on this system, may be in any of the superposition states  $\alpha \boxed{0} + \beta \boxed{1}$ .

Introduction and Motivation
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Stationary (Time-Independent) Schrödinger Equation

# 1D Time-Dependent Schrödinger Equation

Recalling the form of the Hamiltonian, the time-dependent Schrödinger equation can also be rewritten in a more compact form

$$-\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2}\Psi(x,t) + V(x)\Psi(x,t) = i\hbar\frac{\partial}{\partial t}\Psi(x,t)$$
$$\hat{H} = \frac{\hat{p}^2}{2m} + V(x) = -\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2} + V(x)$$
$$\left[\hat{H}\Psi(x,t) = i\hbar\frac{\partial}{\partial t}\Psi(x,t)\right]$$

where  $\hat{H}$  is the Hamiltonian, which represents the total energy of the particle.

### Separation of Variables

Suppose that  $\hat{H}$  does not depend on time explicitly. We will be looking for wavefunctions that are solutions to the Schrödinger equation in the form  $\Psi(x,t)=f(t)\psi(x)$ .

$$\hat{H}(f(t)\psi(x)) = i\hbar \frac{\partial}{\partial t}(f(t)\psi(x))$$

$$\downarrow$$

$$f(t)[\hat{H}\psi(x)] = \psi(x)[i\hbar \frac{\mathrm{d}f(t)}{\mathrm{d}t}]$$

$$\downarrow$$

$$\frac{\hat{H}\psi(x)}{\psi(x)} = \frac{i\hbar \frac{\mathrm{d}f(t)}{\mathrm{d}t}}{f(t)} = E$$

where E is the separation constant and each term has to be equal to this constant (proved by contraposition).

Hence, a single PDE turns into two ODEs:

$$\frac{\hat{H}\psi(x)}{\psi(x)} = E \implies \left[\hat{H}\psi(x) = E\psi(x)\right]$$

Note that here  $\hat{H}$  is an operator and E is a number, so it is an eigenproblem for the Hamiltonian. Its solution is a set of E — eigenvalues (eigenenergies or energy levels) and  $\psi(x)$  — eigenfunctions corresponding to eigenvalues E.

$$\frac{i\hbar\frac{\mathrm{d}f(t)}{\mathrm{d}t}}{f(t)} = E \quad \Rightarrow \quad i\hbar\frac{\mathrm{d}f(t)}{\mathrm{d}t} = Ef(t) \quad \Rightarrow \quad \boxed{f(t) = e^{-\frac{i}{\hbar}Et}}$$

where E is the eigenenergy.

2

Comment. The multiplicative constant resulting from solving the differential equation is included in  $\psi$  (eventually, we need to normalize the wavefunction anyway).

# Stationary Schrödinger Equation

With the explicit form of  $\hat{H}$ , the eigenproblem for the Hamiltionan can be written as

$$-\frac{\hbar^2}{2m}\frac{\mathrm{d}^2\psi(x)}{\mathrm{d}x^2}+V(x)\psi(x)=E\psi(x).$$

This equation is knwon as the **time-independent Schrödinger equation** or the **stationary Schrödinger equation**.

The wave function  $\psi$  should satisfy the following requirements

- $\mathbf{0} \ \psi$  is continuous;
- $\bigcirc$   $\psi'$  is continuous (may be relaxed for singular potentials);
- $\ensuremath{\mathfrak{g}}$   $\psi$  is square-integrable if it represents a bound state.

Full solution (with time dependence) corresponding to an eigenvalue *E* (remember to normalize!)

$$\Psi_E(x,t) = e^{-\frac{i}{\hbar}Et}\psi_E(x)$$

Example. Particle in Infinite Potential Well

# Infinite Potential Well

$$V(x) = \begin{cases} 0, & \text{for } 0 < x < L \\ \infty, & \text{otherwise} \end{cases}$$

No particle can be found outside of [0, L], i.e.  $\psi(x) \equiv 0$  for x < 0 or x > L.

Inside of [0,L], we have  $V(x)\equiv 0$  and the stationary Schrödinger equation reads

$$\begin{cases} -\frac{\hbar^2}{2m} \frac{\mathrm{d}^2 \psi(x)}{\mathrm{d}x^2} = E \psi(x) \\ \psi(0) = \psi(L) = 0 \quad \text{(continuity of } \psi\text{)} \end{cases}$$

Hence, formally, we need to solve a boundary value problem.  $E = ? \psi_E(x) = ?$ 

# Solution: Eigenvalues (Energy Levels)

$$\frac{\mathrm{d}^2\psi(x)}{\mathrm{d}x^2} + \frac{2mE}{\hbar}\psi(x) = 0 \quad \Longleftrightarrow \quad \frac{\mathrm{d}^2\psi(x)}{\mathrm{d}x^2} + k^2\psi(x) = 0,$$

where  $k^2 \stackrel{\text{def}}{=} 2mE/\hbar$ . Since  $k^2 > 0$ , the general solution to this ODE is

$$\psi(x) = A\cos kx + B\sin kx.$$

#### **Eigenvalues**

(2)

Apply the continuity conditions (boundary conditions):

(1) 
$$\psi(0) = A = 0 \implies \psi(x) = B \sin kx$$

(1) 
$$\psi(0) = A = 0 \implies \psi(x) = B \sin kx$$
(2) 
$$\psi(L) = B \sin kL = 0 \implies kL = n\pi \implies k = \frac{n\pi}{L} = k_n$$

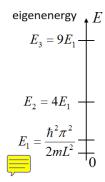
where 
$$n = 1, 2, ...$$
 is the quantum number. (Why  $n \neq 0$ ?)

But  $k^2 = \frac{2mE}{\hbar^2}$ , hence

$$E = E_n = \frac{\hbar^2 \pi^2}{2mL^2} n^2, \quad n = 1, 2, ...$$

# Energy Levels in Infinite Potential Well. Discussion

discrete spectrum



# Solution: Eigenfunctions

$$\psi(x) = \psi_n(x) = \left\{ egin{array}{ll} B \sin k_n x, & ext{for } 0 \leqslant x \leqslant L \\ 0, & ext{otherwise} \end{array} \right.$$

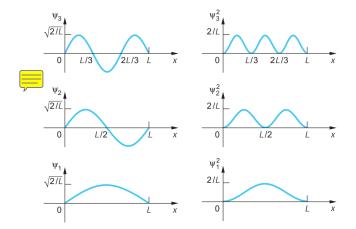
But  $k_n = n\pi/L$ , hence

$$\psi_n(x) = \begin{cases} B \sin \frac{n\pi}{L} x, & \text{for } 0 \leqslant x \leqslant L \\ 0, & \text{otherwise} \end{cases}$$

The normalization condition  $\langle \psi_n, \psi_n \rangle = 1$  (that is, explicitly,  $\int_{-\infty}^{\infty} \psi_n^*(x) \psi_n(x) \mathrm{d}x = 1$ ) yields  $B = \sqrt{2/L}$ . Hence

$$\psi_n(x) = \begin{cases} \sqrt{\frac{2}{L}} \sin \frac{n\pi}{L} x, & \text{for } 0 \leq x \leq L \\ 0, & \text{otherwise} \end{cases}$$

# Wave Functions in Infinite Potential Well. Discussion



#### Solutions to time-dependent equation (stationary states)

$$\Psi_n(x,t) = e^{-\frac{i}{\hbar}E_n t} \psi_n(x) = \sqrt{\frac{2}{L}} e^{-\frac{\hbar^2 \pi^2}{2mL^2}n^2 t} \sin \frac{n\pi}{L} x$$

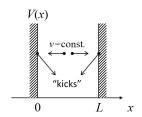
for  $0 \le x \le L$  and zero otherwise.

#### The Correspondence Principle

**Question:** How does the solution to the Schrödinger equation compare with classical mechanics?

- Discrete spectrum vs continuous spectrum of energy levels.
- But in the limit of large *n* classical results expected (the correspondence principle).

#### Classical analysis



Probability of finding the particle in (x, x + dx)

$$\frac{\mathrm{d}t}{T} = \frac{2\frac{\mathrm{d}x}{v}}{2\frac{L}{L}} = \frac{1}{L}\,\mathrm{d}x$$

### The Correspondence Principle

 $0 \le x \le L$ , and zero otherwise.

Hence, the classical probability density:

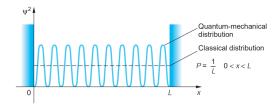
$$g_{cl}(x) = \frac{1}{L}$$
 for

#### Quantum-mechanical probability density

$$g_q(x) = |\Psi_n(x,t)|^2 = |\psi_n(x)|^2 = \frac{2}{l}\sin^2\left(\frac{n\pi}{l}x\right); \text{ for } 0 \leqslant x \leqslant L$$

For  $n \to \infty$ ,  $g_q(x)$  can be replaced by the average value

$$g_q(x) \xrightarrow{n \to \infty} \frac{1}{L} \int_0^L |\psi_n(x)|^2 dx = \frac{1}{L}$$



Schrödinger Equation with an Initial Condition

#### **Initial Condition**

**Question:** How to incorporate an initial condition?

$$\begin{cases} \hat{H}\Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t} \\ \Psi(x,0) = \Phi(x) \end{cases}$$

#### Idea

- Solve the stationary Schrödinger equation  $\hat{H}\psi_n(x) = E_n\psi_n(x)$ , i.e. find eigenvalues and eigenfunctions of  $\hat{H}$ .
- **2** Express the solution to the time-dependent Schrödinger equation as  $\Psi(x,t) = \sum_n c_n e^{\frac{i}{\hbar} E_n t} \psi_n(x)$ , where  $c_n$ 's are to be found.
- **3** Find  $c_n$ 's by using the initial condition  $\Psi(x,0) = \sum_n c_n \psi_n(x)$ .

For Steps 2 and 3, can we construct such expansions? In other words, do  $\psi_n$ 's form a basis of the corresponding space?

# Mathematical Appendix (II). Properties of Hermitian Operators

Let  $\hat{\boldsymbol{A}}$  be the Hermitian operator corresponding to a physical quantity  $\boldsymbol{A}.$  Then

- (1) Eigenvalues of  $\hat{A}$  are real.
- (2) Eigenfunctions corresponding to different eigenvalues are orthogonal (i.e.  $\langle \psi_n, \psi_m \rangle = 0$  if  $n \neq m$ ).
- (3) any square-integrable function can be represented as  $\Psi = \sum_{n} c_n \psi_n$  where  $\psi_n$ 's are the eigenfunctions of  $\hat{A}$ .

**Note.** Statements (2) and (3) will get slightly modified if there are uncountably many eigenvalues of  $\hat{A}$ .

\*\*\* Statement (3) follows from the spectral theorem.

Justification of (1) and (2)

$$\begin{cases}
\hat{A}\psi_{n} = \alpha_{n}\psi_{n} & \stackrel{\langle \psi_{m}, \dots \rangle}{\Rightarrow} \\
\hat{A}\psi_{m} = \alpha_{m}\psi_{m} & \stackrel{\langle \dots, \psi_{n} \rangle}{\Rightarrow} \end{cases}
\begin{cases}
\langle \psi_{m}, \hat{A}\psi_{n} \rangle = \alpha_{n}\langle \psi_{m}, \psi_{n} \rangle \\
\langle \hat{A}\psi_{m}, \psi_{n} \rangle = \alpha_{m}^{*}\langle \psi_{m}, \psi_{n} \rangle
\end{cases}$$

Since  $\hat{A}$  is Hermitian

$$\begin{cases} \langle \psi_{m}, \hat{A}\psi_{n} \rangle = \alpha_{n} \langle \psi_{m}, \psi_{n} \rangle \\ \langle \psi_{m}, \hat{A}\psi_{n} \rangle = \alpha_{m}^{*} \langle \psi_{m}, \psi_{n} \rangle \end{cases}$$

Subtracting both equations side by side

$$0 = (\alpha_n - \alpha_m^*) \langle \psi_m, \psi_n \rangle$$

- If n = m, then since  $\langle \psi_n, \psi_n \rangle \neq 0$ , we have  $\alpha_n = \alpha_n^*$ , i.e.  $\alpha_n$  is real [property (1)].
- If  $n \neq m$ , and  $\alpha_n \neq \alpha_m$ , then  $\langle \psi_n, \psi_m \rangle = 0$ , i.e.  $\psi_n$  and  $\psi_m$  are orthogonal to each other [property (2)].

# Incorporating an Initial Condition. Expansion in Eigenstates

$$\Psi(x,t) = \sum_{n} c_n e^{-\frac{i}{\hbar} E_n t} \psi_n(x)$$

Assume  $\langle \psi_n, \psi_n \rangle = 1$  (normalized)

$$\Psi(x,0) = \Phi(x) = \sum_{n} c_n \psi_n(x)$$

Find

$$\langle \psi_m, \Phi \rangle = \langle \psi_m, \sum_n c_n \psi_n \rangle = \sum_n c_n \langle \psi_m, \psi_n \rangle = \sum_n c_n \delta_{mn} = c_m$$

$$c_m = \langle \psi_m, \Phi \rangle$$

Hence, each of the coefficients  $c_n$  is the orthogonal projection of  $\Phi$  onto  $\psi_n$ .

[Recall cartesian components of vectors in a 3D Euclidean space.]

# **Incorporating an Initial Condition**

Hence, the time dependent solution satisfying the given initial condition

$$\Psi(x,t) = \sum_{n} \langle \psi_{n}, \Phi \rangle e^{-\frac{i}{\hbar} E_{n} t} \psi_{n}(x)$$

where  $\langle \psi_n, \Phi \rangle = \int_{-\infty}^{\infty} \psi_n^*(x) \Phi(x) dx$ .

#### Exercises (Also, see problem set)

- (a) For a particle in an infinite potential well, find  $\Psi(x,t)$  for t>0, if  $\Psi(x,0)=\frac{1}{\sqrt{2}}(\psi_1(x)+\psi_2(x))$  (linear combination of only two eigenfunctions  $\Rightarrow$  only  $c_1,\ c_2\neq 0$ ).
- (b) For a particle in an infinite potential well, find  $\Psi(x,t)$  for t>0, if  $\Psi(x,0)=\sqrt{\frac{30}{L^5}}x(x-L) \text{ (infinitely many eigenfunctions contribute} \\ \Rightarrow \text{ infinitely many } c_n\text{'s}\neq 0\text{)}.$
- (c) Expanding well (see the applet on the St. Andrews University's website).

# Comment. Interpretation of the Expansion Coefficients

**Example.** Suppose that at some instant, a quantum system is described by the wavefunction

$$\psi = \sqrt{\frac{2}{5}}\psi_1 + \sqrt{\frac{3}{5}}\psi_3$$

where  $\psi_1$  and  $\psi_3$  are normalized eigenfunctions of the Hamiltonian, corresponding to eigenvalues  $E_1$  and  $E_3$ .

Let us measure the total energy of the system

possible outcome probability
$$E_1 \qquad \left| \sqrt{\frac{2}{5}} \right|^2 = \frac{2}{5}$$

$$E_3 \qquad \left| \sqrt{\frac{3}{5}} \right|^2 = \frac{3}{5}$$

**Interpreetation.** If we had 1000 copies of this particles in state  $\psi$ , then statistically  $\frac{2}{5} \times 1000 = 400$  and  $\frac{3}{5} \times 1000 = 600$  measurements of the total energy would yield  $E_1$  and  $E_3$  respectively. No other values ( $E_2$ ,  $E_4$ ,...) will be measured.

#### Generalization

If  $\psi = \sum_n c_n \phi_n$ , where  $\hat{A}\phi_n = a_n \phi_n$ , i.e.  $\phi_n$  is the eigenfunction of an operator  $\hat{A}$  corresponding to the eigenvalue  $a_n$ , then

$$|c_n|^2 = \Pr(A = a_n)$$

where  $Pr(A = a_n)$  is the probability that a measurement of the corresponding physical quantity A will yield  $a_n$ .

#### Note.

In the example above  $\hat{A}=\hat{H}$  (physical quantity is the total energy),  $c_1=\sqrt{\frac{2}{5}}$ ,  $c_2=\sqrt{\frac{3}{5}}$  and other  $c_n$ 's are zero.