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START Experiments recorded for
   from SC-Database on Saturday, 01 January, 2000 at 00:56:14
Software version = 5.81 Data version = 4.62
Experiment list contains 35 experiments for
(no ligands specified)
Metal : Os
(no references specified)
(no experimental details specified)
*************************************
e -
               HL Electron
                                 (442)
Electron;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Os kin oth/un ? 1.00M U H
                                       1968BHc (762) 1
                           K=0.89.
Medium: 1 M KOH. K: OsO6---- + OsO4-- = 2Os(VII)
                           1966BDb (763) 2
Os EMF none 25¢C 0.0 M
                            K=14.96, 884.7 mV
                            K'=8.18, 483.6 \text{ mV}
K: Os(bpy)3+++ + e = Os(bpy)3++. K': Os(bpY)2(py)C1++ + e = Os(bpy)2(py)C1+
17 similar reactions
______
0s
      EMF none 25�C 0.0 U
                                       1956CAa (764) 3
                            K=65.3(964 \text{ mV})
K: 0s04(aq)+4H+4e=0s02(H20)x(s)+2H20
Os oth none 25¢C 0.0 U
                                      1952LAb (765) 4
                            K=10(0.3 V)
K(HOs(VIII)O5+2e=Os(VI)O4+OH); from thermodynamic data. Estimated values:
K(0s04+8H+8e=0s(s)+4H20)=114(850 \text{ mV}), K(0s(IV)Cl6+e)=14(850 \text{ mV})
______
                             1950MCa (766) 5
Os EMF oth/un 250C 2.11M U I
                            K=5.90(349 \text{ mV})
Medium: M HBr; K: Os(IV)Br6+e=Os(V)Br6. In I=4 M: K=5.24(310 mV), I=3.25 M:
K=5.43(321 mV), I=2.12: K=5.90(349 mV)
______
Os EMF oth/un 2000 0.10M U I
                                       1946DHa (767) 6
                            K=7.67(446 \text{ mV})
Medium: HCl; K: Os(IV)+e=Os(III). For I=5 M HCl: K=5.28(307 mV); 3.5 M: K=
5.80(337 mV); 2 M: 6.66(387 mV). Also in HBr: 1 M: K=6.63(392 mV)
********************************
           L Carbon monoxide CAS 630-08-0 (551)
Carbon monoxide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
   kin non-aq 25�C 100% U T HM
                                       1993PSb (2816) 7
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K(0s3L10H2+L)=2.77
Medium: Decalin. T. 25-90 C. K=2.63(30C); 2.51(35); 2.41(40); 1.96(60); 1.87
(70); 1.64(80); 1.49(90). At 25-40C, DH=-42.3 kJ mol-1; at 50-90 C, DH=-39.8
-----
       gl non-ag 25%C 100% U HM
                                      1993PSb (2817)
                           B(0s3L10H2+2L)=ca. 7.70
Medium:Decalin. DH=-79.9 kJ mol-1; DS=-113.0.
0s
       kin alc/w 25♦C 100% U
                                      1983WPa (2818) 9
                           K(H30s4(C0)12+H)=12.0
                           K(HOs3(CO)12+H)=14.7
                           K(HOs(CO)4+H)=15.2
**********************************
                             CAS 7727-37-9 (5686)
                   Nitrogen
Dinitrogen, also Nitrous oxide; N2O
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       sp oth/un 25�C 0.30M U M
                                      1971EGa (10024) 10
                           K' = 3.62
Medium:(K,H)SO4. K': cis-Ru(NH3)4(H2O)2+Os(NH3)5N2. K'=3.61 by kinetics
******************************
N2H4
                   Hydrazine
                             CAS 302-01-2 (2117)
Hydrazine; H2N.NH2
______
      Mtd Medium Temp Conc Cal Flags Lg K values
______
       kin oth/un 25�C var U K1=6.76
                                      1972RKc (10085) 11
Metal: OsO4 (?) Medium: HCl
*****************************
                   Hydroxide
                                (57)
OH-
               HL
Hvdroxide;
         ------
Metal Mtd Medium Temp Conc Cal Flags Lg K values
-----
0s
       gl KCl
              25♦C 1.2M C
                                      1998ARa (11828) 12
                           *K(0s0(H20)(CN)4)=3
                           *K(0s0(OH)(CN)4) => -1
Medium: KCl/KNO3
       sp oth/un 25♦C var C
                                      1983GZa (11829) 13
0s
                           *K(H20s02(OH)4)=-8.5
                           *K(HOsO2(OH)2=-10.4
Metal is Os(VI).
______
       sp oth/un 25♦C var C
0s
                                      1983GZa (11830) 14
                           *K(0s04)=-12.2
Metal is Os(VIII).
      sp NaClO4 10�C 4.00M U M
                                      1982BMa (11831) 15
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K(Os(VI)C14(OH)2+H)=0.8
                           K(Os(VI)C14(H2O)OH+H)=0.3
_____
      kin oth/un 20�C 0.10M C
                                      1978LDa (11832) 16
                           K(0s04+0H=H0s05)=2.48
Metal is Os(VIII). Medium: 0.10 M NaHCO3/Na2CO3.
______
      kin oth/un 35�C 1.0M U
                                      1977MGa (11833) 17
0s
                           K(0s03L3+L)=1.50
     dis oth/un 24�C U
                                      1972LEb (11834) 18
                           *K1(0s04+H20=0s040H+H)=-12.5
medium:KOH at various concentrations
______
Os gl oth/un 20♦C 0.25M U
                                      1967BNa (11835) 19
                           *K1=-7.24
Os as OsO2(OH)4. Medium: 0.25 M Na2SO4. In 'dilute' soln: *K1=-7.2,
*K2=12.2, *K3=-13.95, *K4=K(0s050H=0s056+H)=-14.17
Os gl oth/un 25?�C dil U
                                      1966WSa (11836) 20
                           *K1(0s(en)3)=-5.10
Metal: Os+++
______
Os dis none 25°C 0.0 U I
                                      1963G0b (11837) 21
                           Kd(M(aq)=M(CC14))=1.09
M is OsO4(H2O)n; data also for 1 M-NaClO4 (Kd=1.16); no ev polynuclear
complex for <10**-3 M-Os in CCl4
______
Os gl oth/un ? var U
                                  1955DHa (11838) 22
                           *K1(0s(en)3) > 0
                           *K2 = -5.8
                            1953SSb (11839) 23
Os dis NaClO4 25♦C 1.0M U
                           *K1(0s04(H20)n)=-12.0
                           *K2=-14.85?
By spectrophotometry *K1=-12.0, *K2=-14.52
______
Os dis non-aq 25♦C 100% U
                                      1938AYa (11840) 24
                           Kd=1.89(x units)
                           K=0.80 in CCl4
In CCl4; metal is 0s04(H20)n; Kd: K(0s04(aq)=0s04(CCl4)); ev (0s04)n in CCl4,
K(40s04=(0s04)4?); method:also partial pressure of CCl4
______
Os dis oth/un 25♦C dil U
                                      1928YWa (11841) 25
                           *K1(0s04(H20)n)=-12.10
                           Kd=1.09
metal is OsO4(H2O)n; Kd(OsO4(aq)=OsO4(CCl4))
******************************
             H3L Phosphate CAS 7664-38-2 (176)
P04---
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Phosphate;

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Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      kin oth/un 25♦C var U
                                 1973KRb (13282) 26
                        K(0s04+L)=2.4 (?)
*************
C2H5N02
                 Glycine CAS 56-40-6 (85)
             HL
2-Aminoethanoic acid; H2N.CH2.COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
             25¢C 1.00M U T H K1=1.30
0s
      kin NaCl
                                 1984MKa (21655) 27
Data at 25-40C. DH = 8 kJ mol-1. Os = OsO4(OH)2
***********************************
             HL
                          CAS 56-41-7 (86)
C3H7N02
                Alanine
2-Aminopropanoic acid; H2N.CH(CH3).COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
0s
      kin NaCl 25♦C 1.00M U T H K1=0.93
                                 1984MKa (26229) 28
Data at 25-40C. DH = 17 kJ mol-1. Os = OsO4(OH)2
**********************************
                           CAS 147-84-2 (2126)
C5H11NS2
             HL
Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp non-aq 30♦C 100% U
                                 1993PBa (41358) 29
K(trans-OsL2A2=cis-OsL2A2)=0.78 (for Os(II)) and -3.87 (for Os(III)).
************************
                           CAS 2246-46-0 (707)
C9H7N3O2S
             H2L
                 TAR
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
0s
     sp oth/un ? 0.10M U
                                 1969IBc (64718) 30
                        K(?)=8.59
Metal:Os(IV). For Os(VIII), K(?)=6.49
*******************************
                        CAS 60435-22-5 (2819)
C10H11N7S2
Phthalimide-dithiosemicarbazone;
______
   Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
0s
      sp none 25♦C 0.0 U
                                 1976GPc (71099) 31
                        Keff=4.18
Os(VIII) at pH 4
***********************************
C12H8N2O4
                          CAS 6813-38-3 (5904)
4,4'-Dicarboxy-2,2'-bipyridine;
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       sp none 25≎C 0.0 U
                                           1990KNb (80549) 32
                           K(0sH2L3=0sL3+2H)=-1.70
CAS 25005-96-3 (5906)
C14H10N4
2,3-Bis(2-pyridyl)pyrazine;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
sp none 25�C 0.0 U
                                           1990KNb (86913) 33
0s
                             K(OsHL3=OsL3+H)=4.60
******************************
                 HL PAN
C15H11N3O
                                  CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
0s
        sp mixed ? 50% U K1=9.34 1969BIc (91235) 34
Os(IV). Medium: 50% DMF, 0.1 M NaClO4). With Os(VIII), K1(?)=8.62
********************************
C15H12N2O3S
                                    (4070)
2-(3'-Benzoylthioureido)benzoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
                                            Reference ExptNo
______
Os sp alc/w 30♦C 96% U
                                           1966MBa (91440) 35
                             K(?)=4.38
Metal: Os(VIII). Medium: 95% EtOH
REFERENCES
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1966WSa G Watt,J Summers,E Potrafke,E Birnbaum; Inorg.Chem.,5,857 (1966)
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## **EXPLANATORY NOTES**

## DATA Flags are :-

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS
- H Data for THERMOCHEMICAL quantities
- M Data for TERNARY Complexes

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END Experiments recorded for

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