



Veronika Zinovyeva 31 May 2021

Ytterbium

Reaction	Baes and Mesmer, 1976	Brown and Ekberg, 2016
$Yb^{3+} + H_2O \rightleftharpoons YbOH^{2+} + H^+$	-7.7	-7.31 ± 0.18
$Yb^{3+} + 2 H_2O \rightleftharpoons Yb(OH)_2^+ + 2 H^+$	(-15.8)	
$Yb^{3+} + 3 H_2O \rightleftharpoons Yb(OH)_3 + 3 H^+$	(-24.1)	
$Yb^{3+} + 4 H_2O \rightleftharpoons Yb(OH)_4^- + 4 H^+$	-32.7	
$2 \text{ Yb}^{3+} + 2 \text{ H}_2\text{O} \rightleftharpoons \text{Yb}_2(\text{OH})_2^{4+} + 2 \text{ H}^+$		-13.76 ± 0.20
$3 \text{ Yb}^{3+} + 5 \text{ H}_2\text{O} \rightleftharpoons \text{Yb}_3(\text{OH})_5^{4+} + 5 \text{ H}^+$		-30.6 ± 0.3

$Yb(OH)_3(s) + 3 H^+ \rightleftharpoons Yb^{3+} + 3 H_2O$	14.7	15.35 ± 0.20
$Yb(OH)_3(c) + OH^- \rightleftharpoons Yb(OH)_4^-$	-4.0 ± 0.2	
$Yb(OH)_3(c) \rightleftharpoons Yb(OH)_3$	~-9.4	

C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976.

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 247, 250–251 and 300–303.