



Veronika Zinovyeva 16 August 2022

Gadolinium

Reaction	Baes and Mesmer, 1976	Brown and Ekberg, 2016
$Gd^{3+} + H_2O \rightleftharpoons GdOH^{2+} + H^+$	-8.0	-7.87 ± 0.05
$Gd^{3+} + 2 H_2O \rightleftharpoons Gd(OH)_2^+ + 2 H^+$	(-16.4)	
$Gd^{3+} + 3 H_2O \rightleftharpoons Gd(OH)_3 + 3 H^+$	(-25.2)	
$Gd^{3+} + 4 H_2O \rightleftharpoons Gd(OH)_4^- + 4 H^+$	-34.4	
$2 \text{ Gd}^{3+} + 2 \text{ H}_2\text{O} \rightleftharpoons \text{Gd}_2(\text{OH})_2^{4+} + 2 \text{ H}^+$		-14.16 ± 0.20
$3 \text{ Gd}^{3+} + 5 \text{ H}_2\text{O} \rightleftharpoons \text{Gd}_3(\text{OH})_5^{4+} + 5 \text{ H}^+$		-33.0 ± 0.3

$Gd(OH)_3(s) + 3 H^+ \rightleftharpoons Gd^{3+} + 3 H_2O$	15.6	17.20 ± 0.48
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C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976, p. 137.

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 284–287.