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Dysprosium

Reaction	Baes and Mesmer, 1976	Brown and Ekberg, 2016
$Dy^{3+} + H_2O \rightleftharpoons DyOH^{2+} + H^+$	−8.0	−7.53 ± 0.14
$Dy^{3+} + 2\ H_2O \rightleftharpoons Dy(OH)_2^{+} + 2\ H^+$	(–16.2)	
$Dy^{3+} + 3\ H_2O \rightleftharpoons Dy(OH)_3 + 3\ H^+$	(–24.7)	
$Dy^{3+} + 4\ H_2O \rightleftharpoons Dy(OH)_4^{-} + 4\ H^+$	–33.5	
$2\ Dy^{3+} + 2\ H_2O \rightleftharpoons Dy_2(OH)_2^{4+} + 2\ H^+$		−13.76 ± 0.20
$3\ Dy^{3+} + 5\ H_2O \rightleftharpoons Dy_3(OH)_5^{4+} + 5\ H^+$		−30.6 ± 0.3

$Dy(OH)_3(s) + 3\ H^+ \rightleftharpoons Dy^{3+} + 3\ H_2O$	15.9	16.26 ± 0.30
$Dy(OH)_3(c) + OH^- \rightleftharpoons Dy(OH)_4^-$	−3.6 ± 0.3	
$Dy(OH)_3(c) \rightleftharpoons Dy(OH)_3$	−8.8	

C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976.

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 247, 250−251 and 290−292.