

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 348 experiments for

(no ligands specified)

5 metals : Mo(0), Mo(III), Mo(IV), Mo(V), Mo(VI)

(no references specified)

(no experimental details specified)

C3H9O3P L CAS 121-45-9 (1786)

Trimethylphosphite; (CH3O)3.P

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(0) cal non-aq 25°C 100% U HM 1991ZGa (28002) 1

Medium: THF. DH(Mo(CO)3A2+L)=-100.4 kJ mol⁻¹, A=P(C6H11)3

C18H33P L CAS 2622-14-2 (169)

Tri-(cyclohexyl)phosphine; (C6H11)3P

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(0) cal non-aq 25°C 100% U HM 1991ZGa (98314) 2

K(Mo(CO)3py2+L)=-4.73

Medium: THF. DH=-70.7 kJ mol⁻¹

e- HL Electron (442)

Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) EMF oth/un 25°C 4.0M U I 1963MAb (680) 3

K=14.64, 866 mV

K: Mo(Cn)6--- + e = Mo(CN)6---- . K=13.79(I=0.5 M; 816 mV). Medium: H2SO4

CO L Carbon monoxide CAS 630-08-0 (551)

Carbon monoxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) EMF non-aq 22°C 100% U 1992PMa (2815) 4

K([MoL3A]2=2MoL3A)=-16.15

Metal:Mo+. Medium: MeCN, 0.1 M Bu4NPF6. A=C5H5. Dimer-monomer equilibrium

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) EMF oth/un 25°C 3.00M U 1975ZSa (5236) 5

$$K(\text{Mo}(\text{OH})+4\text{Cl}=\text{Mo}(\text{OH})\text{Cl}_4)=2.60$$

Medium: ethanoic acid

Mo(III) kin oth/un 25°C 1.0M U K1=1.03 1974SSd (5237) 6

Medium: lithium p-toluenesulfonate

FCIBrI HL (541)

Halides, comparative (for book data under ligand 80)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) kin alc/w 22°C 100% U M 1960SHa (7409) 7

Metal:Mo++. Medium: EtOH. $K(\text{Mo}_6\text{Cl}_{18}\text{Cl}_6+6\text{Br}=\text{Mo}_6\text{Cl}_{18}\text{Br}_6+6\text{Cl})=-0.1$

OH- HL Hydroxide (57)

Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) dis NaCl 20°C 1.00M U K1=12.0 B2=23.4 1978MMb (11754) 8

B3=34.7

S-- H2L Sulfide CAS 7783-06-4 (705)

Sulfide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) oth none 25°C 0 U 1988LIa (14420) 9

$K_{\text{so}}(\text{Mo}_2\text{S}_3)=-107.8$

$*K_{\text{so}}(\text{Mo}_2\text{S}_3)=-55.8$

Derived from thermodynamic data and $K(\text{H}+\text{S}=\text{HS})=17.3$.

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) kin oth/un 25°C 2.00M U 1997Nca (15153) 10

$K(\text{Mo}_4\text{S}_4(\text{H}_2\text{O})_{12}+\text{L})=3.11$

$K(\text{Mo}_7\text{S}_8(\text{H}_2\text{O})_{18}+\text{L})=2.94$

Medium: Li-p-toluenesulfonate.

Mo(III) kin oth/un 25°C 2.00M U 1993HLA (15154) 11

$K(\text{Mo}_4\text{S}_4+\text{L})=3.11$

Medium: Li toluene-p-sulfonic acid. For Mo(IV), $K=3.72$; for mixed Mo(III)/Mo(IV) ($\text{Mo}_4\text{S}_4++++$), $K=3.48$.

Mo(III) kin oth/un 25°C 1.0M U K1=5.0 1974SSd (15155) 12

medium:lithium p-toluenesulfonate

Mo(III) sp oth/un ? 1.0M U K1=0.6 1972KTa (15156) 13
Medium: p-toluenesulfonic acid

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) kin oth/un ? ? U K1=3.38 1956YAc (18966) 14

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) vlt oth/un 25°C 0.20M U K(?)=3.17 1962ZRa (31304) 15

Medium: 0.2 Na2SO4, 0.1 H2SO4, 0.04 KNO3

C5H9N L t-Butylnitrile CAS 7188-38-7 (913)
t-Butylcyanide; (CH3)3C.CN

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) con non-aq 40°C 100% U M 1992LIa (38452) 16
K(MoL7+Cl)=3.4
K(MoL7+Br)=3.18
K(MoL7+I)=2.6
K(MoL6Cl+Cl)=3.5

Medium: MeCN, 0.0063 M Bu4NClO4. K(MoL6Br+Br)=3.18, K(MoL6I+I)=3. Mo++

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) vlt oth/un 25°C 0.20M U K(?)=3.47 1962ZRa (46178) 17

Medium: 0.2 Na2SO4, 0.1 H2SO4, 0.04 KNO3

C8H19P L (6822)
Di(t-Butyl)phosphine; ((CH3)3C)2PH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(III) nmr none ? 0.0 U T HM 1992BCc (63201) 18
Method:NMR. Medium:toluene. DH(1,2-Mo2L2(NMe2)4 (anti-gauche isomerization))
=-1.3 kJ mol-1, DS=-6.3 J K-1 mol-1. Data also for other phosphides

C10H10O2 HL Benzoylacetone CAS 93-91-4 (197)

1-Phenylbutane-1,3-dione; C₆H₅.CO.CH₂.CO.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(III)	dis	NaCl	20°C	1.0M	U		K1=5.71 B2=11.68 B3=18.64	1978MMi (70752)	19

C₁₂H₂₄O₆ L 18-Crown-6 CAS 17455-13-9 (577)
1,4,7,10,13,16-Hexaoxacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(III)	cal	alc/w	25°C	100%	U	H	K1=2.44	1977ILb (83472)	20

Medium: Methanol. DH=20.0 kJ mol⁻¹.

C₁₇H₁₄N₂O₂ L CAS 4551-69-3 (698)
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(III)	dis	NaCl	20°C	1.0M	U		K1=6.55 B2=12.99 B3=20.15	1978MMi (95893)	21

C₁₈H₁₅P L CAS 603-35-0 (621)
Triphenylphosphine; (C₆H₅)₃P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(III)	kin	non-aq	35°C	100%	U	M		1975EDa (97143)	22

Kout(Mo(CO)₅(NHC₅H₁₀)+L)=2.78

Medium: hexane

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	EMF	oth/un	?	4.50M	U			1958CHb (681)	23

K(Mo+e=Mo(III))=1.7(100 mV)

Medium: H₂SO₄

CN- HL Cyanide CAS 74-90-8 (230)
Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	nmr	KNO ₃	25°C	0.10M	C			1994RLa (2741)	24

*K(MoO(CN)₄(H₂O))=-9.88

Method: N.M.R.

Mo(IV) con oth/un 25°C dil U M 1974FIb (2742) 25
 $K(K+Mo(CN)_8)=1.8$
 $K(Me_4N+Mo(CN)_8)=2.5$
 $K(Et_4N+Mo(CN)_8)=2.3$

Mo(IV) gl none 25°C 0.0 U T H 1973BKa (2743) 26
 $K(MoOOH(CN)_4+H)=8.81$
 $K=8.86(30\text{ }^\circ\text{C})$. $K=8.90(35\text{ }^\circ\text{C})$. $K=8.97(40\text{ }^\circ\text{C})$. $K=9.04(45\text{ }^\circ\text{C})$. $K=9.13(50\text{ }^\circ\text{C})$.
 $DH=23.4\text{ kJ mol}^{-1}$

Mo(IV) sp NaClO4 25°C var U 1973MHa (2744) 27
 $K(Fe+Mo(CN)_8)=2.6$

Mo(IV) sp NaClO4 25°C var U M 1971JSb (2745) 28
 $K(Fe+Mo(CN)_8)=2.6$

Mo(IV) sp oth/un 25°C var U M 1969KBc (2746) 29
 $K(UO_2+Mo(CN)_4(OH)_3(H_2O))=3.71$

Mo(IV) sp oth/un 25°C var U M 1968DBb (2747) 30
 $K(VO+MoL_4(OH)_3H_2O)=4.86$

Mo(IV) gl oth/un 25°C 0.0 U 1968PNb (2748) 31
 $K(H+MoO_2L_4)=12.62$
 $K(H+MoOOHL_4)=9.98$

Mo(IV) con oth/un 25?°C dil U M 1958SEa (2749) 32
 $K_s(KAg_2Y(s))=-13.96$
 $K_s(Ag_3Y(s))=-13.83$
 $K_s(Mn_3Y_2(s))=-12.35$
 $K_s(Fe_3Y_2(s))=-16.28$
 $Y=MoSOHL_4(H_2O)_2---$. $K_s(Co_3Y_2)=-13.92$; $K_s(Ni_3Y_2)=-18.23$; $K_s(Cu_3Y_2)=-18.46$;
 $K_s(Zn_3Y_2)=-13.62$; $K_s(Cd_3L_2)=-18.32$; $K_s(Hg_3Y_2)=-18.73$; $K_s(Pb_3Y_2)=-18.52$

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) kin oth/un 25°C 2.00M U 1993HLa (5238) 33
 $K(Mo_4S_4+L)=0.30$

Medium: Li toluene-p-sulfonic acid. For mixed Mo(III)/Mo(IV) (Mo_4S_4+++++)

ClO4- HL Perchlorate CAS 7001-90-3 (287)
Perchlorate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) kin oth/un ? 2.50M U $K_1=0.91$ 1952HSc (6326) 34
Medium: H2SO4

NO3- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	NaClO4	20°C	2.0M	U	I	K1=0.14 B2=-0.17	1967VDA (9773)	35
Metal:MoO2++. Medium: HClO4. K1=0.10(I=1), 0.14(0.5); B2=-0.26(I=1), -0.08(I=0.5), corrected for assumed mononuclear hydrolysis									

Mo(IV)	ix	NaClO4	20°C	1.0M	U		K1=0.15 B2=-0.15	1967VDA (9774)	36
Metal:MoO2++. Medium: HClO4									

Mo(IV)	dis	oth/un	20°C	var	U			1967VDb (9775)	37
Kd(MoO2+2L+2TBP(org))=-0.95									

Metal:MoO2++. Medium: HL var. Org=kerosene

OH- HL Hydroxide (57)
Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	gl	KCl	25°C	1.2M	C			1998ARa (11755)	38
*K(MoO(H2O)(CN)4)=-9.88									
*K(MoO(OH)(CN)4)=<-14									

Medium: KCl/KN03

Mo(IV)	kin	NaClO4	25°C	2.00M	U			1993LMb (11756)	39
*K(Mo3Se4(H2O)9)=-0.49									
*K(Mo3O3Se(H2O)9)=-0.36									

Medium: LiClO4.

Mo(IV)	sp	NaClO4	25°C	2.00M	U			1992RSb (11757)	40
*K(Mo3S4(H2O)9)=-0.74									

Medium: 2.0 M LiClO4.

Mo(IV)	EMF	oth/un	16°C	var	U			1959LMA (11758)	41
K(Mo(CN)4OH+OH)=8.10									
K(Mo(CN)4(OH)2+OH)=5.47									
K(Mo(CN)4(OH)3+OH)=1.55									

O2 L Oxygen CAS 7782-44-7 (83)
Dioxygen, also oxide; O-- , and superoxide, O2-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	kin	oth/un	25°C	2.0M	U		K1=2.56	1986HNa (12630)	42
K(MoO2+Mo=Mo2O2)=2.73									

Medium: Li-p-toluenesulphonate

O2-- H2L Peroxide CAS 7772-84-1 (2813)
 Peroxide; -0.0-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	oth/un	RT	0.04M	M			1989CHb (12671)	43

Medium: 0.04 M phosphate buffer. $K(\text{MoO}_4 + \text{H}_2\text{O}_2 = \text{MoO}_2(\text{O}_2)\text{OH} + \text{OH}) = -6.04$
 $K(\text{MoO}_2(\text{O}_2)\text{OH} + \text{H}_2\text{O}_2 = \text{MoO}(\text{O}_2)_2\text{OH} + \text{H}_2\text{O}) = 5.43$. Also tris and tetra peroxo cpds.

P04--- H3L Phosphate CAS 7664-38-2 (176)
 Phosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	vlt	NaClO4	25°C	1.00M	U			1962ZRa (13249)	44

$K(\text{Mo(IV)} + \text{H}_3\text{L}) = 1.16$

ReO4- HL Perrhenate (2581)
 Rhenate(VII), Perrhenate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	dis	none	20°C	0.0	U			1977PRa (14105)	45

$K(\text{MoO}_2 + \text{L}) = 1.43$

S-- H2L Sulfide CAS 7783-06-4 (705)
 Sulfide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	oth	none	25°C	0	U			1988LIa (14421)	46

$K_{\text{so}}(\text{MoS}_2) = -75.6$
 $*K_{\text{so}}(\text{MoS}_2) = -40.9$
 Derived from thermodynamic data and $K(\text{H} + \text{S} = \text{HS}) = 17.3$.

SCN- HL Thiocyanate CAS 463-56-9 (106)
 Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	kin	NaClO4	25°C	2.00M	U			1993LMb (15157)	47

$K(\text{Mo}_3\text{Se}_4 + \text{NCS}) = 3.38$
 $K(\text{Mo}_3\text{OSe}_3 + \text{NCS}) = 3.23$
 $K(\text{Mo}_3\text{O}_2\text{Se}_2 + \text{NCS}) = 3.66$
 $K(\text{Mo}_3\text{O}_3\text{Se} + \text{NCS}) = 3.18$
 $K(\text{Mo}_3\text{O}_4 + \text{NCS}) = 2.99$. Medium: 2.0 M HClO4.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	kin	NaClO4	25°C	2.00M	U			1993VSA (15158)	48

$K(\text{Mo}_3\text{S}_4(\text{H}_2\text{O})_9 + \text{L}) = 3.36$
 $K(\text{Mo}_2\text{WS}_4(\text{H}_2\text{O})_9 + \text{L}) = 3.48$

$$K(\text{MoW}2\text{S}4(\text{H}_2\text{O})_9+\text{L})=3.68$$

Medium: 2.0 M HClO₄. For mixed Mo/W species data refer to L binding to Mo.
Metals are Mo(IV) and W(IV).

Mo(IV) kin oth/un 25°C 2.0M U T K1=2.54 19760Sa (15159) 49
Medium: LiClO₄/HClO₄, metal: MoO⁺⁺. K1=2.89 (10 C); 2.73 (15 C); 2.61 (20 C)

C₂H₂O₄ H₂L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Mo(IV)	kin	oth/un	25°C	1.00M	U			1984KR _a (18967)	50
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K(Mo+HL=MoL+H)=3.07

C₂H₄ L Ethylene CAS 74-85-1 (478)
Ethene; H₂C:CH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Mo(IV)	nmr	non-aq	24°C	100%	U	M		1992HM _a (19427)	51
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K(MoO₃Cl₂+L=MoOLB₂Cl₂+B)=-1.0
Method:NMR. Medium:C₆D₆. A:PMePh₂. When A=PMe₃, K=-3.00

C₂H₄O₂S H₂L Thioglycolic CAS 68-11-1 (596)
Mercaptoethanoic acid; HS.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Mo(IV)	sp	oth/un	25°C	?	U			1976LAg (20347)	52
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K(MoO(OH)+H₂L=MoO₂H₂L+H)=0.20
K(MoO(OH)+HL=MoO₂H₂L)=3.80

C₂H₆O₅ L DMSO CAS 67-68-5 (329)
Dimethylsulfoxide; (CH₃)₂.SO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Mo(IV)	kin	non-aq	20°C	100%	U	I		1993BB _c (22112)	53
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K(MoO₃A+L)=1.22
Medium: CH₂Cl₂. In DMF: K=1.90. A: S-methyl-3-(2-hydroxyphenyl)methylene-
dithiocarbazate.

C₃H₆O₂S H₂L Thiolactic acid CAS 79-42-5 (366)
2-Mercaptopropanoic acid; CH₃.CH(SH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Mo(IV)	sp	oth/un	25°C	?	U			1976LAg (25159)	54
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K(MoO(OH)+H₂L=MoO₂H₂L+H)=0.08

C3H7NO2S H2L Cysteine CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) sp oth/un RT 0.5M U 1977LAB (26807) 55
K(MO(OH)+HL=M(OH)2L+H)=2.04

Medium: Na-toluenesulfonic acid

C4H6O4S H3L Thiomalic acid CAS 70-49-5 (109)
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) sp oth/un 25°C ? U 1976LAg (30347) 56
K(MoO(OH)+H2L=MoO2H2L+H)=0.04

C4H6O4S2 H4L CAS 2418-14-6 (4264)
2,3-Dimercaptobutanedioic acid; HOOC.CH(SH).CH(SH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) gl KNO3 25°C 0.10M C 1991HKb (30394) 57
K(Mo3S7L3+H)=11.7
K(Mo3S7HL3+H)=7.64
K(Mo3S7H2L3+H)=6.77
K(Mo3S7H3L3+H)=4.89

K(Mo3S7H4L3+H)=3.98, K(Mo3S7H5L3+H)=3.3

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) vlt NaClO4 25°C 1.0M U 1962ZRa (31305) 58
K(?)=2.06

C5H6 HL Cyclopentadiene CAS 542-92-7 (4288)
Cyclopentadiene; cyclo(-CH:CH.CH2.CH:CH-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(IV) nmr none 37°C 0.0 U 1991KKc (37080) 59
*K(MoL2)=-5.5
*K(MoH-1L2)=-8.5

C5H9NO4 H2L MIDA CAS 4408-64-4 (190)
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	nmr	oth/un	35°C	?	U			1966KUa (39266)	60
							K(MoO4+HL+H)=8.5 K(MoO3+L)=10.4 K(Mo7O24+7HL=7MoO3L+H)=8.9 K(MoO3L+H)=2.8		

K(2MoO3L+2H)=7

C6H8O7	H3L	Citric acid	CAS 77-92-9	(95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; H00CCH2.CH(OH)(C00H).CH2C00H				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	vlt	NaClO4	25°C	1.0M	U			1962ZRa (46179)	61
							K(?)=2.42		

Medium: HClO4

C6H9N3O2	HL	Histidine	CAS 71-00-1	(1)
2-Amino-3-(4'-imidazolyl)propanoic acid; H2N.CH(CH2.C3H3N2)C00H				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	gl	KNO3	25°C	0.15M	C			1981JJa (47588)	62
							B(MoO4+2H+A=MoO3A+H2O)=16.76		

C7H14N4S	L		(6856)
2,8-Dimethylnona-2,7-diene-3,4,6,7-tetraza-5-thione; CH3.C(CH3):N.NH.CS.NH.N:C(CH3)			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	alc/w	25°C	70%	U		K1=3.49 B2=6.72	1991LGa (57845)	63
Complexes probably MoO2L and MoO2L2									

C9H18N4S	L		(6855)
3,9-Dimethylundeca-3,8-diene-4,5,7,8-tetraza-6-thione; CH3CH2C(CH3):N.NH.CS.NH.N:C(CH3)CH2CH3			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	alc/w	25°C	70%	U		K1=4.14 B2=8.02	1991LGa (67960)	64
Complexes probably MoO2L and MoO2L2									

C15H10O8	H6L	Myricetin	CAS 529-44-2	(4055)
3,3',4',5,5',7-Hexahydroxyflavone;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(IV)	sp	oth/un	20°C	?	U			1965GKa (91026)	65
							K(MoO4+H6L=MoO3H4L)=4.62(?)		

e- HL Electron (442)
Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) EMF none 30°C 0.00 U 1971ESb (682) 66
K=2.96(89mV)

K: MoO₂+ + 4H+ + 2e=Mo+++ + 2H₂O

Mo(V) EMF oth/un 18°C 2.0M U 1952LAb (683) 67
K=-8.7(green Mo(III), -250 mV)
K=3.8(red Mo(III), 110 mV)

Medium:HCl. K: Mo(V)+2e=Mo(III)

Mo(V) EMF oth/un 20°C 8.75M U I 1941HGa (684) 68
K(Mo+2e=Mo(III))=7.9(230 mV)

Medium:H₂SO₄. At I=4.7 M: K=3.3(95 mV), 2.25 M:1.0(30 mV); I=0.45:-0.3(10mV)

Mo(V) EMF oth/un 25°C 0.25M U I 1936KTa (685) 69
K(Mo(CN)₈+e)=13.46(796 mV)

Medium: KCl, KBr, KNO₃. At I=0: K=12.28(726.0 mV)

Mo(V) EMF oth/un 0°C var U 1924COa (686) 70
K(Mo(CN)₈+e)=15.5(839 mV)

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) oth oth/un 20°C var U T H 1972JRa (2134) 71
K=13.4

K: Mo₄(OH)₄O₄L₁₂+4H₂O=Mo₄(OH)₄O₈L₄+8H+8L)=13.7(1 C), 13.1(40 C).

DH(K)=-24.3 kJ mol⁻¹, DS=173 J K⁻¹ mol⁻¹. Method: magnetic susceptibility

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) sp NaClO₄ 25°C 5.0M C 1984HSa (5239) 72
K(Mo₂O₄Cl₄+2H+4Cl=Mo₂O₃Cl₈+H₂O)=0.30

K(Mo₂O₃Cl₈+6H+2Cl=2H₂MoOCl₅)=-7.82

Mo(V) oth KCl 20°C var U T H 1972JRa (5240) 73
K=4.01

Medium: HCl. K: 2MoOCl₅+4H₂O=(MoOCl₂(H₂O))₂+4H+6Cl. K=3.82(30 C), 3.65(40 C)
DH=-31.4 kJ mol⁻¹, DS=39.7 J K⁻¹ mol⁻¹. Method: magnetic susceptibility

Mo(V) sp KCl 26°C 4.0M U TIH 1971YTa (5241) 74
K=3.07

Medium: HCl.K: $2\text{MoOCl}_5 + \text{H}_2\text{O} = (\text{MoOCl}_4)_2\text{O} + 2\text{H} + 2\text{Cl}$. $\text{DH}(\text{K}) = -59.8 \text{ kJ mol}^{-1}$. $\text{K} = 3.35$
(59 C), 2.24(78 C), 1.86(93 C). I=6 M: $\text{DH}(\text{K}) = -37.6$. $\text{K} = 2.50(24 \text{ C})$, 1.27(92 C)

Mo(V) oth KCl 40°C var U T 1967JRa (5242) 75
 $\text{K}(2\text{MoOCl}_5 + 4\text{H}_2\text{O} = \text{X} + 4\text{H} + 6\text{Cl}) = 7.31$

Method:magnetic susceptibility. Medium:HCl var. $\text{K} = 8.03(20 \text{ C})$, 7.64(30 C)
 $\text{X} = \text{Mo}_2\text{O}_2(\text{OH})_4\text{Cl}_4$

Mo(V) sp KCl ? var U 1959BGi (5243) 76
 $\text{K}(\text{MoO} + 3\text{Cl}) = -2.3$
 $\text{K}(\text{MoO}_2 + 2\text{H} + 3\text{Cl} = \text{MoOCl}_3 + \text{H}_2\text{O}) = -3.2$

FClBrI HL (541)
Halides, comparative (for book data under ligand 80)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) nmr oth/un 20°C 8.0M U 1966MDb (7410) 77
 $\text{K}: \text{MoOCl}_5 + n\text{L} = \text{MoOCl}_{5-n}\text{Ln} + n\text{Cl}$. $\text{L} = \text{Br}: \text{K} = -0.5(n=1)$, $-0.7(n=2)$, $-1.7(n=3)$,
 $-3.2(n=4)$ or $-4.3(n=5)$

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) gl NaCl 30°C 0.50M C 1991HYa (11759) 78
 $\text{K}(2\text{MoO} + 30\text{H}) = -0.89$
 $\text{K}(3\text{MoO} + 40\text{H}) = 1.00$
 $\text{K}(4\text{MoO} + 50\text{H}) = 2.98$

Mo(V) sp oth/un 18°C 0.10M U 1984NGa (11760) 79
 $\text{K}(\text{MoO}_2 + \text{OH} = \text{MoO}_2\text{OH}) = 10.60$
In 0.1 M HClO₄/NaClO₄; For 1.0 M HClO₄/NaClO₄ $\text{K} = 10.98$

Mo(V) sol oth/un 450°C 0.00 U 1980KKc (11761) 80
 $\text{K}(4\text{MoO}_2 + 2\text{H}_2\text{O} + \text{O}_2 = 4\text{MoO}_2(\text{OH})) = 2.2$
 $p(\text{O}_2) = 500 \text{ atm}$

Mo(V) sp KCl ? var U 1959BGi (11762) 81
 $\text{K}(\text{MoO}_2 + 2\text{H} = \text{MoO} + \text{H}_2\text{O}) = \text{ca. } -1$

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) sp mixed 20°C ? C 1986CZa (15160) 82

B(CuH-2L)=-7.88
B(CuH-3L)=-15.12

Medium: DMSO/acetone

Mo(V) kin NaClO4 25°C 1.00M U M 1976CSa (15161) 83
K(Mo2O4(C2O4)2+L)=0.74

By spectrophotometry: K=0.63

Mo(V) kin NaClO4 25°C 2.00M U T 1975STa (15162) 84
K(Mo2O4+L=Mo2O4L)=2.38

Medium: LiClO4

Mo(V) sp non-aq ? 100% U K1=2.88 1970BRb (15163) 85
Medium: (EtO)2PSSEt + EtOH(4:1)

Mo(V) nmr NaClO4 23°C 2.0M U M 1968MDf (15164) 86
K(MoOL4+A=MoOL3A+L)=-1.64
K(MoOL4+2A=MoOL2A2+2L)=-3.24
K(MoOL4+3A=MoOLA3+3L)=-6.19

Medium: HClO4. A=(NH2)2CS

Mo(V) sp non-aq ? 100% U I K1=5.0 B2=9.40 1965ULa (15165) 87
K3=4.0
K4=3.4

Medium: Me2CO, Mo as MoCl5. In MeOH: K1=3.85

Mo(V) sp oth/un ? 3.25M U I 1959NAb (15166) 88
K6?=1.35

Medium: H2SO4. In 3.1 M (NH4)2SO4 K3*K4*K5?=2.25

Mo(V) sp mixed ? 60% U K1=3.2 B2=6.2 1958PEb (15167) 89
K3=ca.2
K4=-1.6

Medium: 60% w/w acetone/H2O

Mo(V) sp mixed 20°C 60% U K1=3.2 B2=6.2 1958PEb (15168) 90
K3=1.85

Medium: 60% w/w acetone/H2O, 1 M HCl. Also by electrical migration

S04-- H2L Sulfate CAS 7664-93-9 (15)

Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) sp oth/un ? var U 1974RWa (16355) 91
K(2MoO(HL)5+4H2O=MoO2(OH)2(H2O)2(HL)6+4HL+2H)=-9.0

CH4O L Methyl alcohol CAS 67-56-1 (597)

Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(V)	EMF	alc/w	20°C	100%	U			1971GSa (17886)	92
							$K(\text{Mo}+2\text{L}=\text{Mo}(\text{L}')_2+2\text{H}) > 1$ $K(\text{Mo}(\text{L}')_2+2\text{L}'=\text{Mo}(\text{L}')_4)=24.35$		
Medium: MeOH, 1 M Me4NCl. L'=H-1L									

C3H7NO2S		H2L		Cysteine			CAS 52-90-4	(96)	
2-Amino-3-mercaptopropionic acid; H2N.CH(CH2.SH)COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(V)	gl	NaCl04	25°C	0.50M	C			1998MSa (26808)	93
							$K(\text{Mo}2\text{O}2\text{S}2+2\text{L})=36.24$ $K(\text{Mo}2\text{O}2\text{S}2+\text{H}+2\text{L})=38.63$ $K(\text{Mo}2\text{O}2\text{S}2+2\text{H}+2\text{L})=40.63$ $*K(\text{Mo}2\text{O}2\text{S}2\text{H}2\text{L}2)=-2.00$		
*K(Mo2O2S2HL2)=-2.39									

C5H5N		L		Pyridine			CAS 110-86-1	(31)	
Pyridine, Azine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(V)	sp	NaCl04	25°C	1.00M	U	M		1976CSa (36659)	94
							$K(\text{Mo}2\text{O}4(\text{C}2\text{O}4)_2+\text{L})=1.60$		
By kinetics: K=1.83									

C5H8O2		HL		Acetylacetone			CAS 123-54-6	(164)	
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(V)	dis	oth/un	?	?	U			1968ABb (38030)	95
							$K(\text{MoO}(\text{OH})+2\text{L})=20.36$		

C6H2O4Br2		H2L		Bromanilic acid			CAS 4379-59-6	(1279)	
3,6-Dibromo-2,5-dihydroxy-1,4-benzoquinone;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(V)	sp	NaCl04	25°C	2.00M	U			1986VPa (42037)	96
							$K(\text{MoO}2+\text{H}2\text{L}=\text{MoO}2\text{L}+2\text{H})=3.58$		

C10H8O8S2		H4L		Chromotropic ac			CAS 148-25-4	(1875)	
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(V)	sp	oth/un	?	0.10M	U			1968TKd (69960)	97

K(MoO+2H₂L)=8.46

Metal: MoO+++

C₁₀H₁₆N₂O₈ H₄L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) EMF NaClO₄ ? 0.10M U 1970HPa (73970) 98

K(Mo₂O₄+H₂L)=11.24

K(Mo₂O₄+L) > 27.4

Mo(V) sp none ? 0.0 U K₁=6.36 1958SAa (73971) 99

C₁₃H₁₁N₂O₂ HL CAS 304-88-1 (181)

N-Phenylbenzohydroxamic acid; C₆H₅.CO.N(C₆H₅).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) dis NaClO₄ 20°C ? U K₁=14.0 B₂=28.12 1967DBa (85166) 100

K(MoO(OH)+L)=11.83

K(MoO(OH)+2L)=23.31

C₁₃H₁₅N₂O₇ H₃L CAS 98531-21-6 (8057)

2-Hydroxybenzylamine-N,N,O-triethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(V) gl NaCl 30°C 0.50M C 1991HYa (85760) 101

K(MoO+H+L)=14.85

K(MoO+2H+2L)=28.51

e- HL Electron (442)

Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) oth none 25°C 0.00 U 19740Ha (687) 102

K=-92.30(-0.910V)

K: MoO₄-- + 4H₂O + 6e=Mo(s) + 8OH-; method:combination of thermodynamic data

Mo(VI) sp oth/un 25°C 8.6MM U 1964ANb (688) 103

K=0.4

Medium 8.6M HBr. K:Mo(VI)=(MoOBr₄-)₂ + Br₃-

Mo(VI) oth none 25°C 0.00 U 1956GHa (689) 104

K=-93.3(-0.92V)

K: MoO₄-- + 4H₂O + 6e=Mo(s) + 8OH-; method:combination of thermodynamic data

Mo(VI) EMF oth/un 30°C 0.0 U 1953EEa (690) 105

K(MoO ₂ +2H+e)=8.0(482.6 mV)									
Mo(VI)	EMF	oth/un	18°C	2.0M	U			1952LAb	(691) 106
K(Mo+e=Mo(V))=9.2(530 mV)									
Mo(VI)	oth	none	25°C	0.0	U			1952LAb	(692) 107
K=-106.1(-1050 mV)									
K: MoO ₄ +4H ₂ O+6e=Mo(s)+8OH from thermodynamic data									
Mo(VI)	EMF	oth/un	20°C	5.0M	U	I		1941HGa	(693) 108
K(Mo+e=Mo(V))=9.1(530 mV)									
Medium:HCl. In H ₂ SO ₄ : 9.25 M: K=9.1(530 mV), 4.85 M: K=81.(470 mV), 2.35 M: K=7.4(430 mV), 0.5 M: K=7.0(405 mV)									

Cl-			HL	Chloride				CAS 7647-01-0	(50)
Chloride;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
Mo(VI)	sp	NaCl	25°C	4.0M	C			1983HHa	(5244) 109
Medium: HCl. K(Mo(OH) ₅ (H ₂ O)+H+2Cl=Mo(OH) ₄ Cl ₂ +2H ₂ O)=-0.54									
K(Mo(OH) ₄ Cl ₂ +H+Cl=Mo(OH) ₃ Cl ₃ +H ₂ O)=-1.44									
Mo(VI)	sp	KCl	25°C	var	U			1966RCa	(5245) 110
K=-0.89									
K'=-1.42									
K(Mo ₂ Cl ₂ +Cl)=-2.64									
Medium: HCl. K: Mo(OH) ₅ H ₂ O+Cl=MoCl(OH) ₅ +H ₂ O). K': Mo(OH) ₅ Cl+H+L=Mo ₂ Cl ₂ +3H ₂ O (Mo(OH) ₅ H ₂ O=H ₃ MoO ₄ (H ₂ O) ₂). HMoO ₄ : K(H) ₁ =4.21, K ₂ =4.00, K ₃ =0.93									
Mo(VI)	ix	none	25°C	0.0	U			K ₁ =-0.3 B ₂ =-0.8	1964PCa (5246) 111
K ₃ =-1.89									
Mo(VI)	sp	NaClO ₄	?	5.30M	U			1959CSa	(5247) 112
K(HMo ₂ O ₆ +3H+4Cl=2MoO ₂ Cl ₂ +2H ₂ O)=-4.1									

F-			HL	Fluoride				CAS 7644-39-3	(201)
Fluoride;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference ExptNo
Mo(VI)	ix	oth/un	?	?	U			1973PMa	(7028) 113
K(MoO ₂ F ₂ +F)=2.91									
K(MoO ₂ F ₃ +F)=3.83									
Mo(VI)	sp	oth/un	?	var	U			1967KKb	(7029) 114
K(H ₂ MoO ₄ +F=MoO ₃ F+H ₂ O)=4.48									
K(H ₂ MoO ₄ +4F=MoO ₂ F ₄ (+2H))=10.58									
Mo(VI)	con	non-aq	-5°C	100%	U			1960NVa	(7030) 115

$K(\text{MoO}_2\text{F}_2 + 4\text{HF} = \text{MoF}_6 + 2\text{H}_2\text{O}) = -3.5$

Medium: liquid HF, m units

NH₂SO₃- H₂L Sulfamate CAS 5329-14-6 (452)
Sulfamate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 25°C ? U 1958SAc (8800) 116

$K(2\text{H} + 2\text{L} + \text{MoO}_4 = \text{MoO}_3\text{L}_2 + \text{H}_2\text{O}) = -7 ?$

NH₃O L Hydroxylamine; CAS 5470-11-1 (1808)
Hydroxylamine; NH₂.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl oth/un 20°C dil U 1968JDa (9268) 117

$K(\text{HMo}_{11036}\text{L} + \text{H}) = 3.84$

NO₃- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) vlt NaClO₄ 25°C 0.50M U 1983ZZa (9776) 118

$K(\text{MoO}_2 + \text{L}) = 0.32$

Mo(VI) dis non-aq 25°C 100% U I 1970CMb (9777) 119

$K(\text{Mo}(\text{OH})_6 + \text{HL}) = 2.26$

Medium: TBP, 0.5 M KNO₃. In 1 M KNO₃, K=2.22

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO₄ 25°C 0.0 U H 2000CHa (11763) 120

$K(\text{MoO}_3 + \text{H} = \text{MoO}_2(\text{OH})) = 0.95$

$K(\text{MoO}_2(\text{OH}) + \text{H} = \text{MoO}_2) = -1.18$

$K(2\text{MoO}_2(\text{OH}) = \text{Mo}_2\text{O}_5 + \text{H}_2\text{O}) = 1.99$

Medium: 0.4-8.1 M HClO₄. DH($K(\text{MoO}_2(\text{OH}) + \text{H}) = -21.4$ kJ mol⁻¹).

DS=-90 J K⁻¹ mol⁻¹ in 6.3 M HClO₄. DH(Mo₂O₅)=-30.5, DS=-63.

Mo(VI) sp NaClO₄ 25°C 1.0M C 1988CDc (11764) 121

B(2,1)=7.11

B(11,7)=62.9

B(12,8)=72.0

Method: distribution between HClO₄/NaClO₄ solution and tri-n-butyl phosphate. B(p,q): pH + qMoO₄=Hp(MoO₄)q

Mo(VI) sp NaClO4 25°C 4.0M C 1983HUa (11765) 122
 $K(\text{Mo}(\text{OH})_6 + \text{H} = \text{Mo}(\text{OH})_5(\text{H}_2\text{O})) = 1.35$

Mo(VI) gl NaCl 25°C 0.10M C TIH K1=3.47 B2=7.21 1976CRa (11766) 123
 At I=0 by extrapolation: K1=3.55, B2=7.20

Mo(VI) ix NaCl 25°C 0.70M U 1976SKb (11767) 124
 $K(\text{MoO}_2 + \text{OH}) = 12.04$
 $B(\text{MoO}_2 + 2\text{OH}) = 23.60$
 $B(\text{MoO}_2 + 3\text{OH}) = 31.78$
 $B(2\text{MoO}_2 + 3\text{OH}) = 36.78$

Mo(VI) sp KNO3 25°C 0.10M U I K1=13.81 B2=27.06 1971NSd (11768) 125
 $K_3 = 12.48$
 Mo(VI)=MoO2++. K1=14.06, K2=13.52, K3=12.75(I=0.3). K1=14.17, K2=13.62,
 K3=12.79(I=0.5). K1=14.68, K2=14.15, K3=13.37(I=1)

 O2-- H2L Peroxide CAS 7772-84-1 (2813)
 Peroxide; -O.O-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	nmr	oth/un	25°C	0.30M	C T	M		2002TAa (12672)	126
							$B(1,1,2,0) = 11.61$		
							$B(2,1,2,0) = 13.77$		
							$B(2,1,2,1) = 14.50$		
							$B(2,2,4,0) = 23.77$		
Method: 017 nmr. Medium: 0.30 M Na2SO4. Also data for 5 C.									
$B(p,q,r,s): \text{pH} + q\text{MoO}_4 + r\text{H}_2\text{O}_2 + s\text{SO}_4 = \text{Hp}(\text{MoO}_4)q(\text{H}_2\text{O}_2)r(\text{SO}_4)s.$									
Mo(VI)	nmr	NaCl	5°C	0.60M	C T			2002TAa (12673)	127
							$B(1,1,2,0) = 11.61$		
							$B(2,1,2,0) = 13.86$		
							$B(2,1,2,1) = 13.87$		
							$B(2,2,4,0) = 24.08$		
Method: 017 nmr. $B(3,2,4,0) = 26.23$, $B(2,2,6,0) = 23.9$.									
$B(p,q,r,s): \text{pH} + q\text{MoO}_4 + r\text{H}_2\text{O}_2 + s\text{Cl} = \text{Hp}(\text{MoO}_4)q(\text{H}_2\text{O}_2)r(\text{Cl})s.$									
Mo(VI)	gl	oth/un	25°C	0.30M	C	M		2002THa (12674)	128
							$B(1,1,1,0) = 8.53$		
							$B(2,1,1,0) = 11.22$		
							$B(1,1,2,0) = 11.61$		
							$B(2,1,2,0) = 13.77$		
Medium: 0.30 M Na2SO4. $B(2,1,2,1) = 14.50$, $B(2,2,4,0) = 23.77$, $B(8,7,1,0) =$ 56.71. $B(p,q,r,s): \text{pH} + q\text{MoO}_4 + r\text{H}_2\text{O}_2 + s\text{SO}_4 = \text{Hp}(\text{MoO}_4)q(\text{H}_2\text{O}_2)r(\text{SO}_4)s.$									
Mo(VI)	gl	oth/un	25°C	0.30M	C	M		2002THa (12675)	129
							$B(9,7,1,0) = 62.00$		
							$B(10,7,1,0) = 65.74$		
							$B(11,7,1,0) = 68.23$		

Medium: 0.30 M Na2SO4. B(p,q,r,s): pH+qMoO4+rH2O2+sSO4=Hp(MoO4)q(H2O2)r(SO4)s.

 Mo(VI) kin none 25°C 0.0 C 1990CSb (12676) 130
 K(MoO4+H2L=MoO2L(OH)+OH)=-6.04
 K(MoO2L(OH)+H2L=MoOL2(OH)+H2O)=5.43

Mo(VI) sp NaClO4 25°C 1.00M U T K1=6.90 1987LSa (12677) 131

Mo(VI) kin oth/un 25°C 0.05M U 1969AYa (12678) 132
 K(H2MoO4+H2L=H2MoO5+H2O)=4.64
 K'(Mo2O7+H2L=H2Mo2O9)=3.4
 By spectrophotometry, K'=3.5, K(Mo2O7+2H2L=H2Mo2O1+2H)=3.3

Mo(VI) oth oth/un 25°C var U 1965MOb (12679) 133
 K(MoL4+H2L=HMoL4+HL)=-4.7
 K(MoL4+H2O=HMoL4+OH)=ca. -9

Mo(VI) gl oth/un ? var U 1958CSb (12680) 134
 K(MoO4+2H2L=HMoO2L2+OH+H2O)=-3.6

Mo(VI) gl oth/un ? var U 1958CSb (12681) 135
 K(H+HMoO2L2)=2.5
 K(H+MoO2L2)=9.15

Mo(VI) sp oth/un ? 10.0M U 1955CSa (12682) 136
 Medium: HClO4,H2SO4. K(HMo2O2L2HSO4+2H2L+H2O=H2Mo2O3L4+HSO4+5H)=8.50

P04--- H3L Phosphate CAS 7664-38-2 (176)
 Phosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	0.60M	C	M		2000SAa (13250)	137
							B(22,11,1,1)=141.5 B(21,10,2,1)=146.9 B(22,10,2,1)=147.2 B(20,9,3,1)=147.1		
							B(p,q,r,s): pH+qMoO4+rHV04+sHPO4. Additional methods: 31P and 51V nmr. B(21,9,3,1)=150.2, B(22,9,3,1)=151.7, B(23,9,3,1)=152.2.		
Mo(VI)	sp	NaClO4	25°C	3.00M	C			1980LPe (13251)	138
							K(19H+11MoO4+HPO4)=125.96		
Mo(VI)	gl	NaClO4	25°C	3.00M	C			1975PEb (13252)	139
							B(8,5,2)=61.97 B(9,5,2)=67.12 B(10,5,2)=70.69 B(14,9,1)=98.41 B(15,9,1)=102.83; B(16,9,1)=105.85; B(17,9,1)=106.85;		

B(p,q,r): $pH + qMoO_4 + rHPO_4 = Hp(MoO_4)q(HPO_4)r$

Mo(VI) sp oth/un 25°C var U M 1969SAb (13253) 140
K((MoO2)12(H3L)H-27)=-14.9

Mo(VI) vlt oth/un 25°C var U 1961YBb (13254) 141
K(H2MoO4+H3L=MoO2HL+2H2O)=3.16
K(MoO2HL+H3L=MoO2(H2L)2)=0.19

Mo(VI) kin oth/un 22°C 0.48M U 1956YAc (13255) 142
K(H2MoO4+H3L=MoO2L+H+2H2O)=1.02; K(H2MoO4+2H3L=MoO2L2+4H+2H2O)=2.64

S-- H2L Sulfide CAS 7783-06-4 (705)
Sulfide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un 25°C ? U 1988BHa (14422) 143
K(MoO4+HS+H=MoO3S+H2O)=-10.80
K(MoO3S+HS+H=MoO2S2+aq)=-10.41
K(MoO2S2+HS+H=MoOS3+aq)=-10.05
K(MoO3S+HS+H=MoS4+H2O)=-9.49

Mo(VI) sol oth/un 60°C dil U T 1968SJB (14423) 144
Kso(Tl2MoO4S)=-11.36
K=-12.42(25 C), -11.87(40 C), -11.69(50 C)

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un ? var U M 1969MDb (15169) 145
K(MoOL4+A=MoOL3A+L)=-1.5
K(MoOL4+2A=MoOL2A2+2L)=-3.1
K(MoOL4+3A=MoOLA3+3L)=-5.1
K(MoOL4+4A=MoOA4+4L)=-7.6

A=Br-. Other ternary complexes also reported. Method: esr

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 25°C 4.0M C 1983HUa (16356) 146
K(Mo(OH)5+2HL=Mo(OH)4L2)=0.52

K: Mo(OH)5(H2O)+2HSO4=Mo(OH)4(SO4)2+H

Mo(VI) sp NaClO4 ? 8.0M U 1959CSa (16357) 147
K(HMo2O6+3H+2HL=2MoO3LH3)=-4.74

SiO3--	H2L	Silicate	CAS 7699-41-4	(747)
Silicate; SiO2(OH)2--				

Mo(VI) sp NaClO4 25°C 1.0M U Keff=32.2 1982KCb (17215) 148

V04---	H3L	CAS 15457-75-7	(1586)
Vanadate; V02(OH)3-- or polymers			

Mo(VI)	gl	NaCl	25°C	0.60M	C	1991HPa	(17383)	149
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B(15,1,9)=134.56
B(16,1,9)=137.33
B(16,2,8)=133.0
B(13,4,5)=105.22

$B(p, q, r): pH + q(MoO_4^{--}) + r(HVO_4^{--})$. $B(14, 4, 5) = 107.58$, $B(8, 4, 2) = 60.37$, $B(9, 4, 2) = 64.11$, $B(9, 5, 1) = 59.14$, $B(15, 9, 1) = 95.0$, $B(11, 7, 1) = 74.63$.

Mo(VI) g1 NaCl 25°C 0.60M C 1989HPa (17384) 150

$$\begin{aligned} B(15, 1, 9) &= 134.56 \\ B(16, 1, 9) &= 137.33 \\ B(16, 2, 8) &= 132.97 \end{aligned}$$

From combined emf/nmr study. Also $pK(16,1,9)=2.77$.

$$B(p,q,r): pH+q[MoO_4]+r[HV_4]=Hp[MoO_4]q[HV_4]r.$$

W04-- H2L Tungstate CAS 13783-36-3 (445)
Tungstate;

Mo(VI)	dis oth/un 25°C	U T H	1974PRa (17441) 151
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 $K' = 5.77$

K: $\text{MoO}_4 + \text{HCrO}_4$. 60 C; $K' = 4.65$. $\Delta H = -32 \text{ kJ mol}^{-1}$

CH2O2	HL	Formic acid	CAS 64-18-6	(37)
Methanoic acid; H.COOH				

Mo(VI)	ix	oth/un	?	0.05M U	K1=1.02	B2=2.20	1970SHa (17625)	152
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B3=2.78
B4=4.93

Medium: 0.01-0.05 HL. Metal ion: MoO₂++. pH 2.5

CH4O L Methyl alcohol CAS 67-56-1 (597)
Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	nmr	oth/un	-70°C	?	U	M			1971BPg (17887)	153
								K(MoO2F2(H2O)2+L)=0.08		
								K(MoO2F2(H2O)L+L)=-0.60		

CH5AsO3 H2L Me-Arsonic acid CAS 124-58-3 (585)
Methylarsonic acid; CH3.AsO3H2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	nmr	NaCl	22°C	1.00M	U				1997KYa (17966)	154
								B(11,6,2)=83.4		
								B(12,6,2)=88.3		
								B(10,5,2)=75.2		

B(p,q,r): pH + qMoO4 + rCH3AsO3 = Hp(MoO4)q(CH3AsO3)r

CH5O3P H2L CAS 13590-71-1 (1752)
Methylphosphonic acid; CH3.PO3H2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	1.0M	C				1998KYa (18130)	155
								B(10,5,2)=69.51		
								B(11,5,2)=71.07		
								B(11,7,1)=72.69		
								B(12,7,1)=76.23		

Additional method: nmr. B(p,q,r): pH+qMoO4+rL=Hp(MoO4)qLr.
B(12,6,1)=70.31.

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	1.0M	C	H			1986CHa (18968)	156
								B(MoH2L)=13.619		
								B(Mo2H5L2)=31.20		
								B(Mo2H6L2)=34.08		

Mo=MoO4--. DH(MoH2L)=-59.5, DH(Mo2H5L2)=-123.0, DH(Mo2H6L2)=-117.0 kJ mol-1

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KNO3	25°C	0.15M	C				1984JJa (18969)	157
								K(MoO4+2H+L=MoO3L+H2O)=13.816		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	NaClO4	30°C	1.00M	U				1981BCb (18970)	158
								K(MoO4+2L+2H)=15.52		
								K(2MoO2(OH)2L2+2H)=16.5		

$$K(\text{MoO}_5(\text{OH})_2\text{L}_2+\text{H})=14.6$$

Mo(VI) gl KNO₃ 21°C 0.22M C 1978MBc (18971) 159
 $K(\text{MoO}_4+2\text{H}+\text{L}=\text{MoO}_3\text{L}+\text{H}_2\text{O})=13.98$

Medium pH 5-7

Mo(VI) oth oth/un ? ? U K1=1.57 1969SHd (18972) 160
 Metal ion is MoO₂++

Mo(VI) dis NaClO₄ 20°C 0.10M U 1963STc (18973) 161
 $K(\text{H}_2\text{MoO}_4+2\text{HL})=7.37$

Mo(VI) vlt oth/un 25°C 0.11M U I 1961YBa (18974) 162
 $K(\text{H}_2\text{MoO}_4+\text{H}_2\text{L})=3.91$

$K=3.80(I=0.179)$, $3.49(I=0.345)$

 C₂H₄N₂S₂ L Rubeanic acid CAS 79-40-3 (2782)
 Dithiooxamide; H₂N.CS.CS.NH₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp mixed 90°C 80% U K1=23.91 1975WHb (19453) 163
 Medium: 80% 2-propanol/H₂O

 C₂H₄O₃ HL Glycolic acid CAS 79-14-1 (33)
 2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un -70°C 0.10M U M 1971BPg (20584) 164
 $K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})_2+\text{L}=\text{MoO}_2\text{F}_2(\text{H}_2\text{O})\text{L}+\text{H}_2\text{O})=-0.07$,
 $K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})\text{L}+\text{L}=\text{MoO}_2\text{F}_2\text{L}_2+\text{H}_2\text{O})=-0.70$.

 C₂H₅N₂O₂ HL Acetohydroxamic CAS 546-88-3 (2766)
 Acetohydroxamic acid, N-Hydroxyacetamide; CH₃.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.20M C 1998FMa (21814) 165
 $K(\text{MoO}_4+\text{L}+2\text{H}=\text{MoO}_3\text{L}+\text{H}_2\text{O})=17.16$

$K(\text{MoO}_4+2\text{L}+4\text{H}=\text{MoO}_2\text{L}_2+2\text{H}_2\text{O})=32.46$

 C₂H₆N₂O₂ HL CAS 5549-80-4 (833)
 2-Amino-N-hydroxyacetamide, Glycine hydroxamic acid; H₂N.CH₂.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.20M C 1999FCa (21994) 166
 $K(\text{MoO}_4+6\text{H}+2\text{L}=\text{MoO}_2\text{H}_2\text{L}_2)=47.06$
 $K(\text{MoO}_4+3\text{H}+\text{L}=\text{MoO}_3\text{HL})=24.43$

$$K(\text{MoO}_4 + 2\text{H} + \text{L} = \text{MoO}_3\text{L}) = 19.15$$

C2H7O2As HL Cacodylic acid CAS 75-60-5 (586)
Dimethylarsinic acid; (CH₃)₂AsO₂H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaCl 25°C 0.60M C 1997KSc (22537) 167
K(4MoO₄+L+7H)=46.20

Additional method: nmr

C3H4O4 H2L Malonic acid CAS 141-82-2 (79)
Propanedioic acid; CH₂(COOH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO₄ 23°C 1.0M C 1983BCd (24501) 168
K(2MoO₄+2H+2L=Mo₂O₅(OH)₂L₂+H₂O)=6.96
K(Mo₂O₅(OH)₂L₂+H=Mo₂O₅(OH)L₂+H₂O)=8.64

C3H6O2S H2L CAS 107-96-0 (437)
3-Mercaptopropanoic acid; HS.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 25°C ? U 1963SCe (25219) 169
K(MoO₄+3HL=MoO₃L₃+3OH)=23(?)

Medium: acetate buffer

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaCl 25°C 1.00M C H 1993CKb (25484) 170

$$B(1,2,1)=7.46$$

$$B(1,2,2)=15.71$$

$$B(1,2,3)=16.78$$

$$B(1,1,2)=11.76$$

B(p,q,r): pMoO₄ +qHL +rH =(MoO₄)pLqHq+r. B(1,1,3)=12.66, B(2,2,2)=16.07,
B(2,2,3)=21.70, B(2,2,4)=24.97, B(2,1,3)=18.44; other B(p,q,r); also DH

Mo(VI) sp NaClO₄ 25°C 1.0M C 1983BCc (25485) 171

$$K(\text{MoO}_4 + 2\text{HL} + 2\text{H} = \text{MoO}_2\text{L}_2) = -3.5$$

$$K(2\text{MoO}_2\text{L}_2 + 2\text{H} + 3\text{H}_2\text{O} = \text{Mo}_2\text{O}_5\text{L}_2(\text{H}_2\text{O})_2 + 2\text{H}_2\text{L}) = -8.6$$

$$K(\text{Mo}_2\text{O}_5\text{L}_2(\text{H}_2\text{O})_2 + \text{H} = \text{Mo}_2\text{O}_3(\text{OH})_3\text{L}_2 + \text{H}_2\text{O}) = -7.6$$

C3H7NO2 HL B-Alanine CAS 107-95-9 (575)
3-Aminopropanoic acid; H₂N.CH₂.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C				1998FMa (26467)	172
K(MoO4+2L+4H=MoO2L2+2H2O)=33.26										

C3H7NO2S		H2L		Cysteine				CAS 52-90-4 (96)		
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH										
Mo(VI)	sp	NaClO4	25°C	0.10M	C	I			2003GDa (26809)	173
K(MoO4+L+2H=MoO3L+H2O)=21.4										
Data for 0.1-1.0 M NaClO4. K=21.1 (I=0.30), 20.9 (I=0.50), 20.7 (I=0.7), 21.2 (I=1.0).										
Mo(VI)	sp	NaCl	18°C	1.00M	U				1990CJa (26810)	174
K(MoO4+L+2H=MoO3L+H2O)=18.8										
Mo(VI)	sp	oth/un	25°C		?	U			1963SCe (26811)	175
K(MoO4+3HL=MoOL3+3OH)=18(?)										
Medium: acetate buffer. K(?) Mo(V)=6.0										

C3H8N2O2		HL		Ala-hydroxamic				CAS 16707-85-0 (1582)		
2-Amino-N-hydroxypropanamide, Alanine hydroxamic acid; CH3.CH(NH2).CO.NH.OH										
Mo(VI)	gl	KCl	25°C	0.20M	C				1999FCa (27580)	176
K(MoO4+6H+2L=MoO2H2L2)=45.90										
K(MoO4+3H+L=MoO3HL)=24.00										
K(MoO4+2H+L=MoO3L)=18.65										

C3H8N2O2		HL						(6039)		
Sarcosinehydroxamic acid; CH3.NH.CH2.CO.NH.OH										
Mo(VI)	gl	KCl	25°C	0.20M	C				1999FCa (27586)	177
K(MoO4+6H+2L=MoO2H2L2)=47.04										
K(MoO4+3H+L=MoO3HL)=24.52										
K(MoO4+2H+L=MoO3L)=19.48										

C3H8N2O2		HL						(6666)		
beta-Alaninehydroxamic acid; NH2.CH2.CH2.CO.NHOH										
Mo(VI)	gl	KCl	25°C	0.20M	C				1998FMa (27608)	178
K(MoO4+L+3H=MoO3HL+H2O)=25.81										
K(MoO4+2L+6H=MoO2H2L2+2H2O)=49.76										

 C3H8O L n-Propanol CAS 71-23-8 (1914)
 1-Propanol; CH3.CH2.CH2.OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Mo(VI) nmr mixed -80°C ? U M 1971BPg (27644) 179
 Medium: aq.propanol. -80 - +30 C
 $K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})_2 + \text{L} = \text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O}) + \text{H}_2\text{O}) = -0.09$; $K(\text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O}) + \text{L} = \text{MoO}_2\text{F}_2\text{L}_2 + \text{H}_2\text{O}) = -0.62$

C3H8O L isoPropanol CAS 67-63-0 (2024)
 2-Propanol; CH3.CH(OH).CH3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Mo(VI) nmr mixed -80°C ? U M 1971BPg (27646) 180
 Medium: aq.propan-2-ol. -80 - +30 C
 $K(\text{MoO}_2\text{F}_2(\text{H}_2\text{O})_2 + \text{L} = \text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O}) + \text{H}_2\text{O}) = -0.60$; $K(\text{MoO}_2\text{F}_2\text{L}(\text{H}_2\text{O}) + \text{L} = \text{MoO}_2\text{F}_2\text{L}_2 + \text{H}_2\text{O}) = -1.15$

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)
 1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Mo(VI) ix oth/un 22°C 0.10M U K1=1.20 1973SDa (29999) 181
 Metal ion: MoO2++. pH 2.5

C4H6O5 H2L Malic acid CAS 617-48-1 (393)
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Mo(VI) gl NaCl 25°C 1.0M C H 1997CRb (30675) 182
 $B(1,1,1)=7.47$
 $B(1,1,2)=13.23$
 $B(1,1,3)=15.87$
 $B(1,2,2)=15.48$
 $B(p,q,r): p\text{MoO}_4 + q\text{L} + r\text{H} = (\text{MoO}_4)p(\text{L})q(\text{H})r. B(1,2,3)=20.13, B(1,2,4)=24.14,$
 $B(4,2,8)=52.92, B(4,2,9)=54.35, B(4,2,10)=55.27, \text{etc. DH by calorimetry.}$

 Mo(VI) gl KNO3 20°C 0.20M U 1986BHd (30676) 183
 $K(\text{MoO}_2(\text{OH})_4 + \text{L} + 2\text{H} = \text{MoO}_2(\text{OH})\text{H} - \text{L} + 3\text{H}_2\text{O}) = 13.7$

 Mo(VI) oth oth/un RT ? U 1981BCd (30677) 184
 $K(\text{MoO}_2\text{L}_2 + 2\text{H} + 3\text{H}_2\text{O} = \text{Mo}_2\text{O}_5\text{L}_2\text{OH}) = 22.2$
 $K(\text{MoO}_4 + 2\text{L} + 2\text{H} = \text{MoO}_2(\text{OH})_2\text{L}_2) = 13.9, K(\text{MoO}_2\text{L}_2(\text{OH})_2 + \text{H}) = 8.20$

 C4H6O6 H2L D-Tartaric acid CAS 147-71-7 (93)
 D-Tartaric acid, D-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KNO3	20°C	0.20M	M				1982HHb (30977)	185
K(MoO2(OH)2+2L+2H=MoO2(H-1L)2+4H2O)=16.64										
K(2MoO2(OH)2+2L+4H=(MoO2)2(H-2L)2+8H2O)=30.90										

C4H6O6		H2L		L-Tartaric acid	CAS	87-69-4	(92)			
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOCH(OH).CH(OH).COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	1.00M	C	H			1990CHc (31306)	186
B(1,2,2)=16.33										
B(1,3,2)=19.99										
B(1,4,2)=22.92										
B(2,4,1)=24.81										
B(2,5,1)=26.16, B(4,7,4)=56.22, B(4,8,4)=61.53, B(4,9,4)=63.98, B(4,6,2)=43.4, B(4,7,2)=48.2. B(p,q,r): pMoO4 + qH + rL										

Mo(VI)	sp	NaCl	18°C	1.00M	U				1989CPa (31307)	187
K(MoO4+2L+2H=MoO2H-2L2)=16.2										
Data obtained from circular dichroism measurements										

Mo(VI)	oth	oth/un	?	?	M				1969PFa (31308)	188
K(HMO4+HL)=2.36										
Method: polarimetry										

Mo(VI)	dis	NaClO4	20°C	0.10M	U				1963STc (31309)	189
K(H2MoO4+2L)=7.66 ?										

C4H7NO4		H2L		Aspartic acid	CAS	56-84-8	(21)			
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	NaClO4	25°C	0.10M	C	I			2001GZa (31891)	190
K(MoO4+2H+L=MoO3L)=18.7										
Data for 0.1-0.8 M NaClO4.										

Mo(VI)	sp	NaClO4	25°C	0.15M	C				1995GZb (31892)	191
K(MoO4+2H+L=MoO3L)=1.26										

Mo(VI)	gl	NaCl	25°C	1.00M	C	H			1993CHa (31893)	192
B(1,1,1)=6.54										
B(1,1,2)=9.84										
B(1,2,1)=6.57										
B(1,2,2)=11.47										
B(p,q,r): pMoO4+qHL+rH=(MoO4)p(HL)qHr. B(2,1,4)=21.20, B(2,1,5)=23.50, B(4,4,9)=50.86, B(4,4,10)=53.48, B(2,4,8)=37.74. Also DH by calorimetry.										

Mo(VI) sp oth/un 25°C 1.0M C 1982CPa (31894) 193
K(MoO4+2H+L=MoO3L+H2O)=15.74

Medium not defined. pH 6.0.

Mo(VI) gl oth/un 25°C 0.16M M 1977RGa (31895) 194
B(MoO4+2H+L=MoO3L)=16.79

Mo(VI) gl NaClO4 25°C 0.10M U K1=9.29 B2=17.00 1972SSe (31896) 195
K3=3.88

Metal ion: MoO2++

C4H7NO4 H2L IDA CAS 142-73-4 (118)
Iminodiethanoic acid; HN(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaClO4 25°C 3.0M U 1979ZLa (32304) 196
B(MoO4+L+2H=MoO3L)=18.48

Mo(VI) gl oth/un 25°C 0.15M U 1966KR a (32305) 197
K(MoO4+L+2H=MoO3L)=18.3

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaClO4 25°C 0.10M U K1=8.06 B2=15.29 1973TSe (32712) 198
K3=3.45

C4H9NO2 HL 2-Aminobutyric CAS 2835-81-6 (571)
2-Aminobutanoic acid; CH3.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 25°C 0.10M U TIH K1=8.16 B2=15.61 1980SSf (33919) 199
K3=3.62

C4H10O L Isobutanol CAS 78-83-1 (4256)
2-Methylpropan-1-ol; CH3.CH(CH3).CH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un -70°C ? U M 1971BPg (34650) 200
K(MoO2F2(H2O)2+L=MoO2F2(H2O)L+H2O)=-0.22
K(MoO2F2(H2O)L+L=MoO2F2L2+H2O)=-0.72

C4H10O L Butan-2-ol CAS 15892-23-6 (3572)
sec-Butyl alcohol; C2H5.CH(OH)CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	nmr	oth/un	-70°C	?	U	M		1971BPg (34655)	201
K(MoO2F2(H2O)2+L=MoO2F2(H2O)L+H2O)=-0.24									
K(MoO2F2(H2O)L+L=MoO2F2L2+H2O)=-0.70									

C4H10O4	L	Erythritol	CAS 149-32-6	(2706)
1,2,3,4-Tetrahydroxybutane; HO.CH2.CH(OH).CH(OH).CH2.OH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.10M	U			1990CVb (34712)	202
							B(2,2,1)=15.20		
							B(2,3,1)=19.50		
B(p,q,r): pMoO4+qH+rL=MoO4pHqLr. With (2R,2R)-butantetrol, D-threitol,									
B(2,2,1)=14.60, B(2,3,1)=18.20									

C4H11N3O2	HL		CAS 471915-94-3	(8550)
2,4-Diamino-N-hydroxybutanamide;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			2002ECa (35179)	203
							K(MoO4+L+4H=MoO3H2L+H2O)=32.7		
							K(MoO4+L+3H=MoO3HL+H2O)=27.8		
K(MoO4+2L+8H=MoO2H4L2+2H2O)=63.65.									

C4H14N2O6P2	H2L	EDDP0	CAS 1733-49-9	(2435)
1,2-Diaminoethane-N,N'-bis(methylenephosphonic) acid; (H2O3P.CH2.NH.CH2)2				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	0.60M	C	M		1987YAA (35889)	204
							B(10,5,2)=68.07		
							B(11,5,2)=69.40		
							B(11,7,1)=71.96		
							B(12,7,1)=75.70		

B(12,6,1)=69.04. B(p,q,r)=pH+q(MoO4)+r(C6H5PO3)=Hp(MoO4)q(C6H5PO3)r

C5H8O2	HL	Acetylacetone	CAS 123-54-6	(164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	dis	oth/un	25°C	?	U		K1=10.57 B2=20.49	1968ABb (38031)	205
Metal ion: MoO2++									

C5H9NO4	H2L	Glutamic acid	CAS 56-86-0	(22)
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	NaClO4	25°C	0.10M	C	I			2000GZa (39100)	206
K(MoO4+2H+L=MoO3L+H2O)=17.54 Data for 0.1-1.0 M NaClO4. K=16.94 (I=0.4 M), 16.93 (I=0.5 M), 16.84 (I=0.7 M), 16.76 (I=1.0 M).										
Mo(VI)	gl	KNO3	25°C	0.16M	M				1977RGa (39101)	207
B(MoO4+2H+L=MoO3L)=16.79										
Mo(VI)	gl	NaClO4	25°C	0.10M	U			K1=9.24 B2=16.84 K3=3.50	1972SSe (39102)	208
***** C5H9NO4 H2L MIDA CAS 4408-64-4 (190) N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	nmr	oth/un	35°C	1.50M	U	I			1966KR a (39267)	209
K(MoO4+L+HL=MoO3L)=18.2 At 25 C, using glass electrode, I=0.15 M, K=18.73 *****										
C5H10N2O3 HL Glutamine CAS 56-85-9 (18) 2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaClO4	25°C	0.10M	U			K1=7.90 B2=14.83 K3=3.35	1973TSe (39826)	210
***** C5H10O5 L CAS 1114-34-7 (6113) D-Lyxose										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.10M	C				1989VCa (40339)	211
B((MoO4)2H2L)=14.98 B((MoO4)2H3L)=18.68 K((MoO4)2H2L+H)=3.70 M=MoO4 *****										
C5H11NO2S HL CAS 93964-73-9 (3633) Cysteine ethyl ester; H2N.CH(CH2.SH).CO.OCH2.CH3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	?	U				1963SCe (41146)	212
K(MoO4+3HL=MoOL3+3OH)=23(?) Medium: acetate buffer *****										

C5H12O L n-Pentanol CAS 71-41-0 (4298)
1-Pentanol; CH₃(CH₂)₄.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un -70°C ? U M 1971BPg (41640) 213
K(MoO₂F₂(H₂O)₂+L=MoO₂F₂(H₂O)L+H₂O)=-0.30
K(MoO₂F₂(H₂O)L+L=MoO₂F₂L₂+H₂O)=-0.72

C5H12O L Isopentanol CAS 34713-94-5 (4299)
Isopentanol; CH₃.CH₂.CH(CH₃).CH₂.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) nmr oth/un -70°C 0.10M U M 1971BPg (41641) 214
K(MoO₂F₂(H₂O)₂+L=MoO₂F₂(H₂O)L+H₂O)=-0.24
K(MoO₂F₂(H₂O)L+L=MoO₂F₂L₂+H₂O)=-0.70

C5H12O5 L Arabitol CAS 488-82-4 (5403)
Arabitol; HO.CH₂.HOCH.HCOH.HCOH.CH₂.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (41676) 215
B(2,2,1)=16.35
B(2,3,1)=20.45

B(p,q,r): pMoO₄+qH+rL=MoO₄pHqLr

C5H12O5 L Ribitol CAS 488-81-3 (3009)
Ribitol, Adonitol; HO.CH₂.HCOH.HCOH.HCOH.CH₂.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (41680) 216
B(2,2,1)=15.55
B(2,3,1)=19.45

B(p,q,r): pMoO₄+qH+rL=MoO₄pHqLr

C5H12O5 L Xylitol CAS 87-99-0 (2139)
Xylitol; HO.CH₂.HCOH.HOCH.HCOH.CH₂.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (41688) 217
B(2,2,1)=16.25
B(2,3,1)=19.65

B(p,q,r): pMoO₄+qH+rL=MoO₄pHqLr

C6H2O4Br2 H2L Bromanilic acid CAS 4379-59-6 (1279)
3,6-Dibromo-2,5-dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	sp	NaClO4	25°C	2.00M	U		K(HMoO3+H2L=MoO3HL+2H)=3.85	1986VPa (42038)	218
Mo(VI)	sp	oth/un	25°C	1.00M	U		K(MoO4+L+2H=MoO3L)=13.35	1980VPa (42039)	219
Mo(VI)	sp	oth/un	25°C	0.20M	U		K(MoO4+L+2H)=13.60	1979PVa (42040)	220

C6H2O4Cl2		H2L		Chloranilic acid		CAS 87-88-7	(1281)		
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	sp	oth/un	30°C	?	U		K1=6.05	1981BMd (42055)	221
Mo(VI)	sp	oth/un	25°C	1.00M	U		K(MoO4+L+2H=MoO3L)=13.28	1980VPa (42056)	222
Mo(VI)	sp	oth/un	25°C	0.20M	U		K(MoO4+L+2H)=13.90	1979PVa (42057)	223
Mo(VI)	sp	NaClO4	25°C	0.37M	U		K(Mo3O11+3HL)=2.30	1964LSd (42058)	224

C6H4O4		H2L				CAS 615-94-1	(1280)		
2,5-Dihydroxy-1,4-benzoquinone;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	1.00M	U		K(MoO4+L+2H=MoO3L)=15.30	1980VPa (42308)	225

C6H5Li		L				CAS 591-51-5	(2352)		
Phenyl lithium;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	sp	non-aq	25°C	100%	U	I M	K(MoA+L)=3.4	1981PKa (42341)	226
Medium: THF. A=Mo(CO)4(Ph2P(OCH2CH2)3.OPPh2. In benzene: K > 5									

C6H5NO4		H2L		4-Nitrocatechol		CAS 3316-09-4	(890)		
1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

Mo(VI) sp oth/un 25°C .575M U 1980Nka (42936) 227
 $K(H_2MoO_4L+H_2L=MoO_2L_2+2H_2O)=3.3$
 $K(MoO_4+H_2L)=2.21$
Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8
Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅, pH 8

C₆H₅O₂Cl H₂L 4-Cl-Catechol CAS 2138-22-9 (1656)
1,2-Dihydroxy-4-chlorobenzene; Cl.C₆H₃(OH)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 25°C .575M U 1980Nka (43084) 228
 $K(H_2MoO_4L+H_2L=MoO_2L_2+2H_2O)=3.4$
 $K(MoO_4+H_2L)=2.07$
Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

Mo(VI) sp KCl 25°C 0.10M U 1962HAb (43085) 229
 $K(MoO_4+2H_2L)=5.85$

C₆H₆N₂O₂ HL CAS 5657-61-4 (1430)
Nicotinyhydroxamic acid; C₅H₄N.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO₄ 25°C 0.10M U 1964Rma (43437) 230
 $K(?)=6.3$
 $K(?)=6.7$

C₆H₆O₂ H₂L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C₆H₄.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 25°C .575M U 1980Nka (43790) 231
 $K(H_2MoO_4L+H_2L=MoO_2L_2+2H_2O)=3.3$
 $K(MoO_4+H_2L)=1.49$
Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (43791) 232
 $K(MoO_4+2H_2L=MoO_2L_2+2H_2O)=5.09$
By electrophoresis, phosphate buffer, K=5.21

Mo(VI) sp KNO₃ 20°C 0.10M U 1969HBa (43792) 233
 $K(MoO_4+2H_2L=MoO_2L_2)=5.00$

Mo(VI) sp oth/un 20°C 0.10M U 1964PCa (43793) 234
 $K(MoO_4+2H_2L=MoO_2L_2)=5.27$
Medium: 0.1 M NaHSO₃

Mo(VI) sp oth/un 26°C 0.10M U 1960HAa (43794) 235

$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 4.61$

Medium: 0.1 M NaHSO₃

Mo(VI) sp oth/un 20°C ? U 1959HAa (43795) 236

$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 5.27$

C₆H₆O₃ H₃L Pyrogallol CAS 87-66-1 (696)
1,2,3-Trihydroxybenzene; C₆H₃(OH)₃

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 25°C .575M U 1980NKA (43967) 237

$K(\text{H}_2\text{MoO}_4\text{L} + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 3.2$

$K(\text{MoO}_4 + \text{H}_2\text{L}) = 1.97$

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (43968) 238

$K(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_2(\text{HL})_2 + 2\text{H}_2\text{O}) = 5.43$

By electrophoresis, phosphate buffer, K=5.57

Mo(VI) sp oth/un 20°C ? U 1959HAa (43969) 239

$K(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_2(\text{HL})_2) = 5.48$

Mo(VI) sp oth/un 20°C ? U 1958PIa (43970) 240

$K(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_2(\text{HL})_2) = 5.68 ?$

C₆H₆O₅S H₃L CAS 7134-09-0 (3687)
3,4-Dihydroxybenzenesulfonic acid; (HO)₂C₆H₃.SO₃H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (44284) 241

$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 5.28$

C₆H₆O₈S₂ H₄L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)₂C₆H₂(SO₃H)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.20M C M 2002FCa (44474) 242

$K(\text{MoO}_4 + 2\text{H} + \text{L} = \text{MoO}_3\text{L} + \text{H}_2\text{O}) = 23.8$

$K(\text{MoO}_4 + 4\text{H} + 2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 46.96$, $K(2\text{MoO}_4 + 6\text{H} + 2\text{L} = \text{Mo}_2\text{O}_6\text{H}_2\text{L}_2 + 2\text{H}_2\text{O}) = 61.6$,

$K(\text{MoO}_4 + 4\text{H} + \text{A} + \text{L} = \text{MoO}_2\text{AL} + 2\text{H}_2\text{O}) = 41.5$. A is acetohydroxamic acid.

Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (44475) 243

$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 6.59$

C₆H₇O₃As H₂L Phenylarsonic CAS 98-05-5 (3690)
Benzenearsonic acid, phenylarsonic acid; C₆H₅AsO₃H₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	1MM	C			1987YTa (45177)	244

Values given for B(pH+qMoO2+C) where C=C6H5AsO3H- and H2AsO4-.

C6H8O6		H2L		Ascorbic acid			CAS 50-81-7 (285)		
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Ascorbic acid (Vitamin C);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	sp	oth/un	?	?	U			1966SAb (45648)	245

K(?)=4.6

C6H8O7		H3L		Citric acid			CAS 77-92-9 (95)		
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2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCCH2.CH(OH)(COOH).CH2COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	NaCl	25°C	1.00M	U T H		K1=5.12 B2=9.17 B3=11.94	1995CRa (46180)	246

DH(B1)=-1.3; DH(B2)=-5.8; DH(B3)=-10.1 kJ mol-1. TDS(B1)=28; TDS(B2)=47; TDS(B3)=58 kJ mol-1. Data also at 275 K

Mo(VI)	gl	NaCl	25°C	1.00M	U T H		B(111)=8.35 B(121)=15.00 B(131)=19.62 B(141)=21.12	1995CRa (46181)	247
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B(242)= 31.02; B(252)= 35.86; B(262)= 40.08; B(142)= 25.34; B(152)= 29.54; B(162)= 33.34; etc. B(pqr): pMoO4 + qH + rL = (MoO4)pHqLr

Mo(VI)	gl	NaCl	25°C	1.00M	C H		B(1,1,1)=8.25 B(1,2,1)=15.08 B(1,3,1)=19.66	1991CKa (46182)	248
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B(p,q,r): p(WO4--)+qH+rL=(WO4--)pHqLr. By calorimetry, DH(1,1,1)=-49, DH(1,2,1)=-60, DH(1,3,1)=-67 kJ mol-1.

Mo(VI)	gl	KNO3	20°C	0.20M	U		K(MoO2(OH)4+L+2H=MoO2H-1L)=16	1986BHd (46183)	249
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K(MoO2(OH)4+L+3H=MoO2H-1L+4H2O)=20.4
K(MoO2(OH)H-1L+H=MoO2H-1L+H2O)=4.8

Mo(VI)	gl	NaCl	25°C	1.00M	C H		B(1,1,1)=8.25 B(1,1,2)=15.08 B(1,1,3)=19.66 B(2,1,4)=27.27	1986CVa (46184)	250
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B(2,1,5)=31.86. B(p,q,r): pMoO4+qL+rH=(MoO4)pLqHr. DH(1,1,1)=-49.4 kJ mol-1; DH(1,1,2)=-60.2; DH(1,1,3)=-67.4; DH(2,1,4)=-124

C6H9NO6 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH₂.COOH)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	NaClO ₄	25°C	0.10M	C	I		2003MZa (46922)	251
							K(MoO ₄ +2H+L=MoO ₃ L+H ₂ O)=18.72		
Also data for I=0.5, 0.7 and 1.0 M NaClO ₄ . For I=1.0 M, K=17.97.									

Mo(VI)	gl	NaCl	25°C	1.00M	C	H		1994CHb (46923)	252
							B(1,1,2)=17.78		
							B(1,1,3)=21.02		
							B(1,1,4)=22.57		
							B(2,2,7)=45.16		
B(p,q,r): pMoO ₄ +qL+rH=(MoO ₄)pLqHr. Also B(2,2,8)=47.95, B(2,1,5)=30.74, B(2,1,6)=33.09. DH(1,1,2)=-69, DH(1,1,3)=-71.2, DH(2,2,7)=-123, DH(2,2,8)=-132.									

Mo(VI)	sp	NaClO ₄	25°C	0.5M	C			1976CLa (46924)	253
							K(MoO ₄ +2H+L=MoO ₃ L+H ₂ O)=17.90		
Method: stopped flow spectrophotometry									

Mo(VI)	nmr	oth/un	28°C	1.30M	U			1967MEa (46925)	254
							K(MoO ₄ +W ₃ L=MoO ₃ L+W ₄ O)=0.15		

Mo(VI)	gl	oth/un	25°C	0.15M	U			1966KR a (46926)	255
							K(MoO ₄ +L+2H=MoO ₃ L)=18.94		

Mo(VI)	nmr	oth/un	35°C	2.00M	U			1966KR a (46927)	256
							K(MoO ₄ +L+2H=MoO ₃ L)=18.90		

C6H10N4O2 HL CAS 25486-00-4 (2554)
2-Amino-3-(4'-imidazolyl)propanehydroxamic acid, Histidine-hydroxamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			1999FCa (47907)	257
							K(MoO ₄ +6H+2L=MoO ₂ H ₂ L ₂)=48.5		
							K(MoO ₄ +3H+L=MoO ₃ HL)=24.77		
							K(MoO ₄ +2H+L=MoO ₃ L)=18.44		
							K(MoO ₄ +8H+2L=MoO ₂ H ₄ L ₂)=55.1		
K(MoO ₄ +7H+2L=MoO ₂ H ₃ L ₂)=52.88									

C6H10O8 H2L Mucic acid CAS 526-99-8 (3650)
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; HOOC.(CHOH)₄.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	kin	oth/un	20°C		?	U		1971FPb (48438)	258
							K(?)=7.57		

 C6H10O8 H2L Saccharic acid CAS 87-73-0 (1191)
 D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid, Glucaric acid; $\text{HOOC}(\text{CHOH})_4\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	kin	oth/un	25°C	0.10M	U			$K(?)=7.64$	1971FPb (48484)	259

 C6H12N2O4 H2L EDDA CAS 5657-17-0 (119)
 1,2-Diaminoethane-N,N'-diethanoic acid; $\text{HOOC}.\text{CH}_2.\text{NH}.\text{CH}_2.\text{CH}_2.\text{NH}.\text{CH}_2.\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaClO4	25°C	3.0M	U			$B(\text{MoO}_4+\text{L}+2\text{H}=\text{MoO}_3\text{L})=19.69$	1979ZLa (49254)	260

 C6H12O5 L L-Rhamnose CAS 634-74-2 (3659)
 6-Deoxy-L-mannose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.10M	C			$B((\text{MoO}_4)2\text{H}_2\text{L})=13.89$ $B((\text{MoO}_4)2\text{H}_3\text{L})=17.59$ $K((\text{MoO}_4)2\text{H}_2\text{L}+\text{H})=3.70$	1989VCa (49508)	261

 C6H12O6 L D-Mannose CAS 3458-28-4 (1562)
 D-Mannose

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.10M	C			$B((\text{MoO}_4)2\text{H}_2\text{L})=14.50$ $B((\text{MoO}_4)2\text{H}_3\text{L})=18.10$ $K((\text{MoO}_4)2\text{H}_2\text{L}+\text{H})=3.60$	1989VCa (49607)	262

M=MoO4

 C6H12O7 HL Gluconic acid CAS 526-95-4 (904)
 D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; $\text{HO}.\text{CH}_2(\text{CHOH})_4.\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	kin	oth/un	?	?	U			$K(2\text{MoO}_3+\text{L}=(\text{HMoO}_3)_2(\text{H}-2\text{L}))=7.12$	1972FPb (49737)	263

 C6H13NO2 HL CAS 4312-93-0 (4386)
 Hexanohydroxamic acid; $\text{CH}_3.\text{CH}_2.\text{CH}_2.\text{CH}_2.\text{CH}_2.\text{CO}.\text{NH}.\text{OH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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 Mo(VI) sp oth/un ? ? U 1971PMd (50228) 264
 $K(\text{MoO}_2+\text{L})=15.02$
 $K(\text{MoO}_2+2\text{L})=18.04$

C6H14O6 L D-Dulcitol CAS 608-66-2 (3663)
 D-Galactitol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (51062) 265
 $B(2,2,1)=17.30$
 $B(2,3,1)=20.90$

$B(p,q,r): p\text{MoO}_4+q\text{H}+r\text{L}=\text{MoO}_4p\text{H}q\text{L}r$

C6H14O6 L D-Mannitol CAS 69-65-8 (3664)
 D-Mannitol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (51085) 266
 $B(2,2,1)=16.70$
 $B(2,3,1)=20.80$

$B(p,q,r): p\text{MoO}_4+q\text{H}+r\text{L}=\text{MoO}_4p\text{H}q\text{L}r$

 Mo(VI) gl KNO3 21°C 0.10M C 1978MBc (51086) 267
 Medium pH 3-5. $K(2\text{MoO}_4+2\text{H}+\text{L}=\text{Mo}_2\text{O}_5(\text{H}-4\text{L})+3\text{H}_2\text{O})=16.89$
 $K(\text{Mo}_2\text{O}_5(\text{H}-4\text{L})+\text{H}=\text{HM}_2\text{O}_5(\text{H}-4\text{L}))=3.82$

 Mo(VI) kin oth/un ? ? U 1972FPa (51087) 268
 $K(2\text{H}_2\text{MoO}_4+\text{L}=(\text{H}_2\text{MoO}_4)_2\text{L})=7.12$

C6H14O6 L Glucitol CAS 50-70-4 (2878)
 D-Sorbitol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KCl 25°C 0.10M U 1990CVb (51105) 269
 $B(2,2,1)=16.60$
 $B(2,3,1)=20.50$

$B(p,q,r): p\text{MoO}_4+q\text{H}+r\text{L}=\text{MoO}_4p\text{H}q\text{L}r$

 Mo(VI) gl KNO3 21°C 0.10M C 1978MBc (51106) 270
 Medium pH 3-5. $K(2\text{MoO}_4+2\text{H}+\text{L}=\text{Mo}_2\text{O}_5(\text{H}-4\text{L})+3\text{H}_2\text{O})=16.90$
 $K(\text{Mo}_2\text{O}_5(\text{H}-4\text{L})+\text{H}=\text{HM}_2\text{O}_5(\text{H}-4\text{L}))=3.88$

 Mo(VI) kin oth/un ? ? U 1972FPa (51107) 271
 $K(2\text{H}_2\text{MoO}_4+\text{L}=(\text{H}_2\text{MoO}_4)_2\text{L})=6.64$

C6H15N3O2 HL CAS 52760-35-7 (6670)

Lysine hydroxamic acid; $\text{H}_2\text{N}(\text{CH}_2)_4\text{CH}(\text{NH}_2)\text{CO.NHOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C				2002ECa (51429)	272
								$\text{K}(\text{MoO}_4 + \text{L} + 4\text{H} = \text{MoO}_3\text{H}_2\text{L} + \text{H}_2\text{O}) = 34.16$		
								$\text{K}(\text{MoO}_4 + \text{L} + 3\text{H} = \text{MoO}_3\text{HL} + \text{H}_2\text{O}) = 29.15$		
								$\text{K}(\text{MoO}_4 + 2\text{L} + 8\text{H} = \text{MoO}_2\text{H}_4\text{L}_2 + 2\text{H}_2\text{O}) = 66.65$		

C7H6O3		H2L						CAS 139-85-5 (881)		
3,4-Dihydroxybenzaldehyde, protocatechuic aldehyde; $\text{C}_6\text{H}_3(\text{OH})_2\text{CHO}$										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	.575M	U				1980NKa (54356)	273
								$\text{K}(\text{H}_2\text{MoO}_4\text{L} + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 3.4$		
								$\text{K}(\text{MoO}_4 + \text{H}_2\text{L}) = 2.15$		

Medium: 0.1 M NH_4OH , 0.08 M $\text{Na}_2\text{S}_2\text{O}_5$. pH 8

Mo(VI)	sp	oth/un	20°C	?	U				1959HAa (54357)	274
								$\text{K}(\text{MoO}_4 + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 7.75$		

C7H6O4		H3L						CAS 409-79-9 (1115)		
2,5-Dihydroxybenzoic acid; $\text{C}_6\text{H}_3(\text{OH})_2\text{COOH}$										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	0.10M	U			$\text{K}_1 = 2.58$	1976DVa (54588)	275

C7H6O4		H3L						Protocatechuic CAS 99-50-3 (875)		
3,4-Dihydroxybenzoic acid; $\text{C}_6\text{H}_3(\text{OH})_2\text{COOH}$										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	.575M	U				1980NKa (54683)	276
								$\text{K}(\text{H}_2\text{MoO}_4\text{L} + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 3.1$		
								$\text{K}(\text{MoO}_4 + \text{H}_2\text{L}) = 1.98$		

Medium: 0.1 M NH_4OH , 0.08 M $\text{Na}_2\text{S}_2\text{O}_5$. pH 8

Mo(VI)	sp	oth/un	20°C	?	U				1959HAa (54684)	277
								$\text{K}(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_2\text{H}_2\text{L}_2) = 6.68$		

C7H6O5		H4L						CAS 610-02-6 (3725)		
2,3,4-Trihydroxybenzoic acid; $(\text{HO})_3\text{C}_6\text{H}_2\text{COOH}$										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	20°C	0.10M	U				1971SBd (54721)	278
								$\text{K}(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_2(\text{HL})_2 + 2\text{H}_2\text{O}) = 5.24$		

C7H6O5 H4L Gallic acid CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI) sp oth/un 25°C .575M U 1980Nka (54754) 279
 $K(H_2MoO_4L + H_2L = MoO_2L_2 + 2H_2O) = 3.2$
 $K(MoO_4 + H_2L) = 2.32$

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

Mo(VI) sp oth/un 20°C 0.10M U 1971SBd (54755) 280
 $K(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_2(\text{HL})_2 + 2\text{H}_2\text{O}) = 5.38$

Mo(VI) sp oth/un 20°C ? U 1959HAa (54756) 281
K(MoO4+2H3L=MoO2H2L2)=6.83

C7H7NO2 HL CAS 495-18-1 (184)
Benzohydroxamic acid; C6H5.CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI) g1 none 25°C 0.0 U T H K1=7.49 B2=14.40 1980ABa (55509) 282
At 35 C: K1=7.38, K2=6.79; DH(K1)=-4.6, DH(K2)=-5.0

C7H8O2 H2L CAS 488-17-5 (1657)
1,2-Dihydroxy-3-methylbenzene; CH₃.C₆H₃(OH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI) sp oth/un 25°C .575M U 1980Nka (56056) 283
K(H2MoO4L+H2L=MoO2L2+2H2O)=3.3
K(MoO4+H2L)=1.52

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

C7H8O2 H2L Methylcatechol CAS 452-86-8 (525)
1,2-Dihydroxy-4-methylbenzene; CH3.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI) sp oth/un 25°C .575M U 1980Nka (56071) 284
 $K(H_2MoO_4L + H_2L = MoO_2L_2 + 2H_2O) = 3.1$
 $K(MoO_4 + H_2L) = 1.28$

Medium: 0.1 M NH₄OH, 0.08 M Na₂S₂O₅. pH 8

Mo(VI) sp oth/un 20°C .014M U 1962HAb (56072) 285
K(MoO4+2H2L)=4.74

C8H7O3Cl H2L CAS 99-40-1 (3818)
4-Chloro-2,3-dihydroxyacetophenone, 3-acetyl-6-chlorocatechol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1963HAb (59247)	286
									K(MoO4+2H2L)=7.03	

C8H8O3	HL	Mandelic Acid	CAS 611-72-3	(80)
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	vlt	NaClO4	25°C	0.50M	U				1984ZZa (59853)	287
									K(MoO2+HL)=2.42	

C8H8O3	H2L		CAS 2848-25-1	(3799)
3,4-Dihydroxyacetophenone, (4-acetylcatechol)				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	20°C	?	U				1961HAa (59893)	288
									K(MoO4+2H2L=MoO2L2)=6.74	

C8H11NO2	H2L	Dopamine	CAS 579-59-9	(251)
2-(3',4'-Dihydroxyphenyl)ethylamine; (HO)2.C6H3.CH2.CH2.NH2				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1963HAc (61082)	289
									K(MoO4+2H2L=MoO2L2)=5.57	

C8H11NO3	H2L	Noradrenaline	CAS 138-65-8	(253)
Norepinephrine, 3,4-Dihydroxyphenylethanolamine; (HO)2C6H3.CH(CH2.NH2).OH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U				1962HAb (61166)	290
									K(MoO4+2H2L)=5.82(?)	

C9H5NOBr2	HL		CAS 521-74-4	(3279)
5,7-Dibromo-8-hydroxyquinoline;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	dis	oth/un	20°C	var	U		K1=14.22	B2=28.32	1967DBa (63522)	291
Metal: MoO2++ , Medium: var (HCl,HClO4)										

C9H6O4	H2L	Esculetin	CAS 305-01-1	(3853)
6,7-Dihydroxycoumarin;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI) sp alc/w ? 50% U 1963JSa (63953) 292
K(MoO4+2H2L)=3.65(?)

Medium: 50% EtOH

C9H7NO HL Oxine CAS 148-24-3 (504)
8-Hydroxyquinoline (8-quinolinol);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaNO3 25°C 0.20M U 1968KDa (64319) 293
K(H+HL+MoO4)=10.36

By kinetics, K=10.29

Mo(VI) dis oth/un ? ? U K1=17.04 B2=33.02 1967BDa (64320) 294
K(MoO(OH)+L)=16.44
K(MoO(OH)+2L)=30.02

Metal: MoO2++

Mo(VI) dis oth/un ? ? U K1=12.7 1967BDa (64321) 295

Metal: MoO++++

C9H7NO3S2 H2L CAS 58447-10-2 (4675)
8-Mercaptoquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un ? ? U B2=22.8 1968ABa (64427) 296

Metal: MoO2++

C9H7NO4S H2L Sulfoxine CAS 84-88-8 (448)
8-Hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 16°C 0.10M U 1969GTa (64565) 297
K(MoO4+L+2H=MoO3L)=19.53

C9H11NO2 HL (4650)
5-Methyl-2-hydroxyacetophenone oxime; (CH3)(HO).C6H3.C(:N.OH).CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp alc/w 30°C ? U 1970GMF (66027) 298
K(MoO4+2H2L=MoO2L2)=8.69

Medium: EtOH

C9H12N2O4 H3L (6664)
3,4-Dihydroxyphenylalanine hydroxamic acid, DOPA hydroxamic acid;
H2N.CH(CH2.C6H3(OH)2CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C	M	K(MoO4+8H+2L)=65.29 K(MoO4+6H+2L)=56.81 K(2MoO4+8H+2L)=70.65 K(MoO4+3H+L)=28.81 K(MoO4+2H+L)=20.91.	2002FCa (66621)	299

C9H13NO3		H2L					Corbadrine	CAS 50731-42-5	(3880)
1-(3',4'-Dihydroxyphenyl)-2-aminopropanol;									

Mo(VI)	sp	KCl	25°C	0.10M	U		K(MoO4+2H2L=MoO2L2)=5.92	1962HAb (66818)	300

C9H13NO3		H2L					(-)Adrenaline	CAS 51-43-4	(252)
4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene, Epinephrine;CH3NHCH(OH)C6H3(OH)2									

Mo(VI)	sp	KCl	25°C	0.10M	U		K(MoO4+2H3L=MoO2(HL)2)=5.76(?)	1962HAb (66865)	301

C10H8O2		H2L						CAS 92-44-4	(1658)
2,3-Dihydroxynaphthalene;									

Mo(VI)	sp	oth/un	25°C	.575M	U		K(H2MoO4L+H2L=MoO2L2+2H2O)=3.5 K(MoO4+H2L)=2.43	1980NKA (69774)	302
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8									

Mo(VI)	sp	oth/un	20°C	0.10M	U		K(MoO4+2H2L=MoO2L2+2H2O)=6.31	1973PAC (69775)	303
Medium: phosphate buffer. By electrophoresis: K(MoO4+2H2L=MoO2L2+2H2O)=6.15									

C10H8O4		H2L					4-Me-Esculetin	CAS 529-84-0	(3890)
4-Methyl-6,7-dihydroxycoumarin									

Mo(VI)	sp	alc/w	?	50%	U		K(MoO4+2H2L=MoO2L2)=7.55	1963JSA (69790)	304
Medium: 50% EtOH									

C10H8O5S		H3L					DHNSA	(877)	

2,3-Dihydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI)	sp	oth/un	20°C	0.10M	U				1971SBd (69854)	305
								$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + 2\text{H}_2\text{O}) = 6.25$		

By electrophoresis and phosphate buffer: $K = 6.32$

C10H8O8S2 H4L Chromotropic ac CAS 148-25-4 (1875)

1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI)	sp	NaClO4	20°C	0.10M	U				1970BGB (69961)	306
								$K(\text{H}_2\text{MoO}_4 + \text{H}_2\text{L} = \text{HMoO}_3\text{L} + \text{H}) = -0.8$		
								$K(\text{HMoO}_3\text{L} + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2 + \text{H}) = -2.8$		

Metal: MoO4--

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Mo(VI)	gl	NaClO4	25°C	3.0M	U				1979ZLa (73972)	307
								$B(\text{MoO}_4 + \text{L} + 2\text{H} = \text{MoO}_3\text{L}) = 18.76$		
								$B(2\text{MoO}_4 + \text{L} + 2\text{H} = \text{Mo}_2\text{O}_6\text{L}) = 36.0$		

Mo(VI)	gl	NaClO4	25°C	0.10M	U	T			1968NPa (73973)	308
								$K(\text{MoO}_3 + \text{HL}) = 8.22$		
								$K(\text{MoO}_3 + \text{L}) = 10.0$		
								$K(2\text{MoO}_3 + \text{L}) = 19.16$		
								$K(\text{H} + \text{Hn}-1(\text{MoO}_3)_2\text{L}) = 3.16, n=1$		

$K(n=2) = 2.87, K(n=3) = 2.21, K(n=4) = 1.93$

Mo(VI)	gl	oth/un	25°C	0.15M	U				1966KR a (73974)	309
								$K(\text{MoO}_4 + \text{L} + 2\text{H}) = 18.6$		
								$K(\text{MoO}_4 + \text{MoO}_3\text{L} + 2\text{H}) = 17.5$		
								$K(\text{MoO}_3\text{L} + \text{H}) = 8.1$		

Mo(VI)	nmr	oth/un	35°C	1.0M	U				1966KR a (73975)	310
								$K(\text{MoO}_4 + \text{L} + 2\text{H}) = 18.5$		
								$K(\text{MoO}_4 + \text{MoO}_3\text{L} + 2\text{H}) = 17.2$		
								$K(\text{MoO}_3\text{L} + \text{H}) = 7.5$		

$I = 1.0-2.5$

Mo(VI)	nmr	oth/un	35°C	?	U				1966KU b (73976)	311
								$K(\text{MoO}_4 + \text{HL} + \text{H} = \text{MoO}_3\text{L}) = 8.8$		
								$K(\text{MoO}_3\text{L} + \text{H}) = 7.5$		
								$K(2\text{MoO}_4 + \text{L} + 4\text{H} = (\text{MoO}_3)_2\text{L}) = 35.1$		
								$K(\text{MoO}_3 + \text{L}) = 10.7$		

$K(2\text{MoO}_3 + \text{L}) = 19.5$, $K((\text{MoO}_3)_2\text{L} + \text{H}_2\text{L} = 2\text{MoO}_3\text{HL}) = 0.26$

C10H18N4O6 H2L (4504)

Hexanoic acid bis(3-hydroxycarbamoyl-methyl)amide; $\text{HONHCOCH}_2\text{NHC}(\text{CH}_2)_4\text{CONHCH}_2\text{CONHOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			1998FMa (75569)	312

$K(\text{MoO}_4 + \text{L} + 4\text{H} = \text{MoO}_2\text{L} + 2\text{H}_2\text{O}) = 30.45$

$K(\text{MoO}_4 + \text{L} + 3\text{H} = \text{MoO}_3\text{HL} + \text{H}_2\text{O}) = 25.52$

For the propylamide analogue Kvalues are 30.88; 26.76

C10H20N4O4 H2L CAS 475984-27-1 (6717)

Piperazine-1,4-bis(N-methylacetohydroxamic acid); $\text{C}_4\text{H}_8\text{N}_2(\text{CH}_2.\text{CO}.\text{N}(\text{OH})\text{CH}_3)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			2002FCb (75896)	313

$K(\text{MoO}_4 + 3\text{H} + \text{L} = \text{MoO}_3\text{HL} + \text{H}_2\text{O}) = 27.4$

$K(2\text{MoO}_4 + 9\text{H} + 2\text{L} = \text{Mo}_2\text{O}_4\text{HL}_2 + 4\text{H}_2\text{O}) = 74.7$, $K(2\text{MoO}_4 + 10\text{H} + 2\text{L} = \text{Mo}_2\text{O}_4\text{H}_2\text{L}_2 + 4\text{H}_2\text{O}) = 77.5$,

$K(2\text{MoO}_4 + 11\text{H} + 2\text{L} = \text{Mo}_2\text{O}_4\text{H}_3\text{L}_2 + 4\text{H}_2\text{O}) = 80.5$. An alternative model given also.

C10H25N5 L 15-Ane-N5 CAS 295-64-7 (99)

1,4,7,10,13-Pentaazacyclopentadecane; $\text{cyclo}(-(\text{HN}.\text{CH}_2.\text{CH}_2)_5-)$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	vlt	NaClO4	25°C	0.20M	C			1999SSe (76737)	314

$K(\text{MoO}_4 + \text{H}_3\text{L}) = 2.11$

Method: differentail pulse polarography.

Also data for selenate, selenite and pyrophosphate as guest ions.

C11H17N3O3 H2L Isoprenaline CAS 586-06-1 (3950)

3,4-Dihydroxy-1-(1'-hydroxy-2'-(propylamino)ethyl)benzene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	sp	KCl	25°C	0.10M	U			1963HAc (79158)	315

$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 5.87$

C11H21N3O5 H2L CAS 499238-77-6 (8837)

N-Hydroxy-N'-[4-(hydroxymethylamino)-4-oxobutyl]-N-methylpentanediamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			2002FBb (79795)	316

$K(\text{MoO}_4 + \text{L} + 4\text{H} = \text{MoO}_2\text{L} + 2\text{H}_2\text{O}) = 31.27$

$K(\text{MoO}_4 + \text{L} + 3\text{H} = \text{MoO}_3\text{HL} + \text{H}_2\text{O}) = 26.62$

C12H11N3O4S H2L (4003)

3-Hydroxy-3-phenyl-1-(4'-sulfonyl)triazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	?	U			K(?)=12.87	1958DSa (80941)	317

C12H23N3O5 H2L CAS 499238-78-7 (8836)

N-Hydroxy-N'-[5-(hydroxymethylamino)-5-oxopentyl]-N-methylpentanediamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			K(MoO4+L+4H=MoO2L+2H2O)=31.41 K(MoO4+L+3H=MoO3HL+H2O)=26.68	2002FBb (82985)	318

C12H23N3O5 H2L CAS 499238-79-8 (8835)

N-Hydroxy-N'-[6-(hydroxymethylamino)-6-oxohexyl]-N-methylbutanediamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	KCl	25°C	0.20M	C			K(MoO4+L+4H=MoO2L+2H2O)=33.07 K(MoO4+L+3H=MoO3HL+H2O)=26.79	2002FBb (82995)	319

C12H30N6 L CAS 296-35-5 (143)

1,4,7,10,13,16-Hexaazacyclooctadecane; cyclo(-(NH.CH2.CH2)6-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	vlt	NaClO4	25°C	0.20M	C			K(MoO4+H3L)=2.14	1999SSe (84343)	320

Method: differentail pulse polarography.

Also data for selenate, selenite and pyrophosphate as guest ions.

C13H10O3 H2L CAS 5876-92-6 (4009)

3,4-Dihydroxybenzophenone; C6H5.CO.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	sp	oth/un	25°C	0.01M	U			K(MoO4+2H2L=MoO2L2)=6.75	1962HAb (84991)	321

C13H15NO7 H3L CAS 98531-21-6 (8057)

2-Hydroxybenzylamine-N,N,O-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Mo(VI)	gl	NaCl	30°C	0.50M	C			K(MoO4+2H+L)=17.36	1991HYa (85761)	322

$$K(\text{MoO}_4 + 3\text{H} + \text{L}) = 20.60$$

C14H7O5Cl3 H3L (5107)

2,3,7-Trihydroxy-9-trichloromethylfluorone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp KNO3 25°C 0.50M U 1972ASb (86590) 323

$$K(\text{MoO}_2\text{OH} + \text{H}_2\text{L}) = 17.10$$

C14H8O6 H4L Quinalizarin CAS 81-61-8 (1056)

1,2,5,8-Tetrahydroxyanthraquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp alc/w 25°C 50% U 1970RBb (86682) 324

$$K(?) = 4.48 \quad (\text{pH} = 5.0)$$

C14H8O7S H3L DASA CAS 83-61-4 (950)

1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl NaNO3 25°C 0.10M U 1983KMa (86742) 325

$$B(\text{MoO}_4\text{H}_2\text{L}) = 20.18$$

$$B(\text{MoO}_4\text{H}_3\text{L}) = 24.93$$

Mo(VI) sp NaClO4 25°C 0.10M U 1963SDF (86743) 326

$$K(\text{MoO}_4 + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 9.2(?)$$

Mo(VI) sp oth/un 25°C ? U B2=9.6 1959DBb (86744) 327

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)

Diethylenetriamine-pentaethanoic acid; $\text{HOOC} \cdot \text{CH}_2 \cdot \text{N}(\text{CH}_2 \cdot \text{CH}_2 \cdot \text{N}(\text{CH}_2 \cdot \text{COOH})_2)_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 25°C 0.10M U 1971LUa (89325) 328

$$K(2\text{MoO}_4 + \text{H}_5\text{L} = (\text{MoO}_3)_2\text{HL}) = 15.01$$

$$K(\text{MoO}_4 + \text{H}_5\text{L} = \text{MoO}_3\text{H}_3\text{L} + \text{H}_2\text{O}) = 8.53$$

$$K(\text{MoO}_4 + \text{H}_4\text{L} = \text{MoO}_3\text{H}_2\text{L} + \text{H}_2\text{O}) = 7.31$$

$$K(\text{MoO}_4 + \text{H}_3\text{L} = \text{MoO}_3\text{HL} + \text{H}_2\text{O}) = 4.87$$

$$K(\text{MoO}_4 + \text{H}_2\text{L} = \text{MoO}_3\text{L} + \text{H}_2\text{O}) = 2.55$$

C15H10O4 H2L CAS 38183-04-9 (4051)

6,7-Dihydroxy-4-phenylcoumarin (4-phenylesculetin)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp alc/w ? 50% U 1963JSa (90987) 329

$$K(\text{MoO}_4 + 2\text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 8.20(?)$$

Medium: 50% EtOH

C15H10O4 H2L (4052)

7,8-Dihydroxy-3-phenylcoumarin (3-phenyldaphnetin)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp alc/w 32°C 50% U 1966JKb (90988) 330

$$K(\text{MoO}_4 + \text{H}_2\text{L} = \text{MoO}_2\text{L}_2) = 4.5(?)$$

Medium: 50% EtOH, 0.2 M KCl

C15H10O5 H3L Galangin CAS 548-83-4 (4053)

3,5,7-Trihydroxyflavone (3,5,7-Trihydroxy-2-phenylchromone)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 28°C ? U 1964KSc (90993) 331

$$K(\text{MoO}_4 + \text{H}_3\text{L} = \text{MoO}_3\text{HL}) = 4.58(?)$$

C15H11N3O4S H2L (5130)

7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 16°C 0.10M U 1969GTa (91337) 332

$$B((\text{MoO}_4)\text{H}_2\text{L}) = 18.12$$

C15H11N3O7S2 H3L CAS 17852-90-3 (5131)

7-(4-Sulfophenylazo)-8-hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 16°C 0.10M U 1969GTa (91350) 333

$$B((\text{MoO}_4)\text{H}_2\text{L}) = 17.98$$

C16H14O5 H3L CAS 966-64-3 (5143)

2,3,7-Trihydroxy-9-propylfluorone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp KNO3 25°C 0.50M U 1972ASb (93590) 334

$$K(\text{MoO}_2\text{OH} + \text{H}_2\text{L}) = 17.27$$

$$K(\text{MoO}_2 + 2\text{H}_2\text{L}) = 26.50$$

C16H35O4P HL CAS 298-07-7 (1625)

Di-(2-ethylhexyl)-phosphoric acid; (C2H5C6H12O)2P(O)OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) dis non-aq RT 100% C I 1992SGa (95510) 335

K'=4.28

Method: solvent extraction into CCl4. K':

Mo(VI) dis non-aq RT 100% C I 1992SGa (95511) 336

K'=4.28

By solvent extraction into CCl4. K': H2MoO4+H2L2=MoO2L2(HL)2(org)+H2O

Also data for C6H6 (K'=4.02), C2H4Cl2 (3.90), CHCl3 (3.55), MIBK (3.48).

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)

Triethylenetetraaminehexaethanoic acid;((HOOCH2)2N.CH2.CH2.N(CH2.COOH).CH2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl KNO3 25°C 0.10M U 1971LUa (98070) 337

K(2MoO4+H6L=(MoO3)2H2L)=13.81

K(2MoO4+H5L=(MoO3)2HL)=11.78

K(2MoO4+H4L=(MoO3)2L)=8.42

K(MoO4+H6L=MoO3H4L)=7.45

K(MoO4+H5L=MoO3H3L)=6.88, K(MoO4+H4L=MoO3H2L)=5.64, K(MoO4+H3L=MoO3HL)=3.16,

K(MoO4+H2L=MoO3L)=3.14

C19H13N3O7S2 H3L SNAZOXS CAS 117-87-3 (995)

8-Hydroxy-7-(4'-sulfo-1'-naphthylazo)-quinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 RT 0.10M C K1=9.20 1987APb (99047) 338

Mo(VI) gl KNO3 16°C 0.10M U 1969GTa (99048) 339

K(MoO4+L+2H)=17.82

C20H11N06S2 H2L CAS 66451-75-0 (8985)

6-Hydroxy-5-oxo-5H-dibenzo[a,j]phenoxazine-11-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99529) 340

B((MoO2)2L)=10.8

B((MoO2)2L) is the effective constant at pH 2.3.

C20H11N06S2 H2L CAS 55968-31-5 (8984)

6-Hydroxy-5-oxo-5H-dibenzo[a,j]phenoxazine-9-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99531) 341

K1eff=3.62

Medium pH 2.2.

C20H11N09S2 H3L CAS 65501-73-7 (8982)
6-Hydroxy-5-dibenzo[a,j]phenoxazone-8,11-disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99535) 342

B((MoO2)2L)=10.0

B((MoO2)2L) is the effective constant at pH 2.05.

C20H11N09S2 H3L CAS 66451-74-9 (8983)
6-Hydroxy-5-oxo-5H-dibenzo[a,j]phenoxazine-9,11-disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaClO4 25°C 0.10M C 1977SLb (99539) 343

B((MoO2)2L)=9.9

B((MoO2)2L) is the effective constant at pH 2.05.

C20H13N3O7S H3L EriochromeBla A CAS 16279-54-2 (5299)
3-Hydroxy-4-(2-hydroxy-1-alpha-naphthylazo)-7-nitronaphthalene-1-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp NaNO3 25°C 2.0M U 1971AAc (99584) 344

K(MoO2+HL)=9.80

C20H22N2O8 H5L Azotochelin CAS 23369-85-9 (6112)
N,N'-Bis(2,3-dihydroxybenzoyl)lysine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl oth/un 25°C 0.10M C K1=ca. 35 1998DHa (99918) 345

K1(eff)=4

Medium: 0.10 M HEPES, pH 6.6.

C22H20O13 H5L Carminic acid CAS 1260-17-9 (714)
Carminic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) sp oth/un 22°C ? U 1966KWb (101705) 346

K(MoO4+H5L=MoO3H3L)=3.8(?)

C22H24N2O8 L Deoxycycline CAS 564-25-0 (2204)
Deoxycycline, 6-Deoxy-5-hydroxytetracycline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Mo(VI) gl none 20°C 0.0 C 1991JMa (101766) 347

K(MoO4+H3L=MoO3HL)=7.99

$$K(\text{MoO}_4 + 2\text{H}_3\text{L} = \text{MoO}_3(\text{H}_2\text{L})_2) = 9.21$$

C22H24N2O8 H2L Tetracycline CAS 60-54-8 (2201)
Tetracycline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Mo(VI)	gl	none	20°C	dil	C			1989VJa (101823)	348

$K(\text{MoO}_3 + \text{HL}) = 7.80$

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

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