

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 143 experiments for

(no ligands specified)

Metal : Hf++++

(no references specified)

(no experimental details specified)

e- HL Electron (442)

Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ oth none 25°C 0.0 U 1952LAb (515) 1

K=-106(-1570 mV)

K: HfO2(s)+4H+4e=Hf(s)+2H2O. From thermodynamic data

Br- HL Bromide CAS 10035-10-6 (19)

Bromide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis NaClO4 20°C 3.0M U K1=-0.1 1967HPc (1944) 2

CO3-- H2L Carbonate CAS 465-79-6 (268)

Carbonate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis oth/un 20°C 1.00M U I 1987JBb (3235) 3

B4=39.83

When I=2.5 M: B5=40.21

Hf++++ gl KCl 25°C 1.00M U 1982KCc (3236) 4

K(Hf(OH)2L+L)=11.0

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis NaClO4 25°C 4.00M U M 1976TSa (4933) 5

K(HfOH+Cl)=-0.54

Hf++++ ix NaClO4 20°C 2.0M U K1=0.07 B2=-0.48 1967EMc (4934) 6

B3=-0.40

Hf++++ dis NaClO4 20°C 3.0M U K1=0.34 B2=-0.02 1967HPc (4935) 7

Medium: HClO4. By cation exchange:K1=0.18

Hf++++	dis	NaClO4	27°C	2.0M	U		K1=-0.15	B2=-0.32	1965DKa	(4936)	8
Hf++++	dis	NaClO4	25°C	2.0M	U		K1=0.38 K3=-0.68 K4=-0.7 B4=1.3	B2=0.07	1963PAd	(4937)	9
Hf++++	ix	NaClO4	?	2.0M	U	I	K1=-0.02 B3=-1.15 B4=-1.10	B2=-0.92	1962MRc	(4938)	10
Medium: HClO4. In 4 M HClO4 K1=-0.03, B2=-0.74, B3=-1.10											

F-		HL	Fluoride						CAS 7644-39-3	(201)	
Fluoride;											
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values		Reference	ExptNo
Hf++++	cal	NaClO4	25°C	4.0M	U	H				1990AHa	(6938) 11
DH(Hf+HF=HfF+H)=15.2 kJ mol ⁻¹ ; DH(HfF+HF=HfF2+H)=15.0; DH(HfF2+HF=HfF3+H)=8.8; DH(HfF3+HF=HfF4+H)=27.4											
Hf++++	gl	NaCl	37°C	0.15M	C			B(HfL4)=30.16 B(HfL6)=40.48		1985IWb	(6939) 12
Hf++++	ISE	NaClO4	25°C	4.0M	U			*K1=5.5 *K2=4.01		1973N0a	(6940) 13
Medium: HClO4. *Kn=HfF(n-1)+HF=HfFn+H											
Hf++++	ix	oth/un	?	?	U			K6=3.83		1972PAb	(6941) 14
Hf++++	gl	NaClO4	25°C	3.0M	U			*K1=4.42 *K2=3.12 *K3=2.64 *K4=1.9		1969VAa	(6942) 15
Medium: (H,Na)ClO4. *K5=1.59, *K6=1.70. *Kn: HfF(n-1)+HF=HfFn+H											
Hf++++	dis	NaClO4	20°C	4.0M	U			K(Hg+HF=HfF+H)=5.52 K(HfF+HF=HfF2+H)=4.04 K(HfF2+F=HfF3+H)=3.04 K(HfF3+HL=HfF4+H)=2.20		1967N0a	(6943) 16
Medium: HClO4. *K5=1.7, *K6=0.5											
Hf++++	ix	NaClO4	20°C	4.0M	U			K(Hf+HL=HfF+H)=5.51		1967N0a	(6944) 17

K(HfF+HL=HfF2+H)=3.7?

Method: cation exchange. Medium: HClO4

Hf++++ ix KCl ? 0.50M U 1967PMd (6945) 18
K6=3.83

Hf++++ dis NaClO4 27°C 2.0M U 1965DKa (6946) 19
K(Hf+HL=HfF+H)=4.62

Medium: HClO4

Hf++++ dis oth/un 25°C 3.0M U 1964VHa (6947) 20
*K1=4.16
*B2=7.79
*B3=10.10
*B4=12.03

Medium:HClO4. *B6=12.62; HfF5 negligible. *Bn: K(Hf+nHF=HfFn+nH)

Hf++++ dis NaClO4 25°C 3.00M U 1963VHa (6948) 21
K(Hf+HF=HfF+H)=4.89
K(HfF+HF=HfF2+H)=3.67
K(HfF2+HF=HfF3+H)=2.97
K(HfF3+HF=HfF4+H)=2.77

Methdo: quinhydrone elec. *K5(HfF4+HF=HfF5+H)=1.55, *K6=2.54

I- HL Iodide CAS 10034-85-2 (20)
Iodide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hf++++ dis NaClO4 20°C 3.0M U K1=-0.46 1967HPc (8029) 22

NO3- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hf++++ dis NaClO4 4.0M U K1=-0.4 B2=-1.3 1969HSb (9696) 23
Medium: HClO4
Hf++++ ix oth/un 20°C 2.0M U K1=-0.15 B2=-0.96 1967EMc (9697) 24
Hf++++ dis NaClO4 27°C 2.0M U K1=-0.1 1965DKa (9698) 25
Hf++++ dis NaClO4 25°C 2.0M U K1=0.34 B2=0.00 1963PAd (9699) 26
K3=-0.72
K4=-0.80
B4=-1.52

Hf++++ ix NaClO4 ? 4.0M U I K1=-0.22 B2=-0.92 1962MRc (9700) 27
Method: Cation exchange. Medium: HClO4. I=2 M: K1=-0.05, B2=-0.26

Hf++++	dis	NaClO4	20°C	4.0M	U	M	K1=0.92 B3=1.89 B4=2.08 B5=2.08 B6=1.81	B2=1.51	1962PBc (9701)	28
Medium: HClO4. Kd(HfL4+2TBP(org)=HfL4(TBP)2(org))=-0.12, org=hydrog.kerosene *****										
OH-		HL		Hydroxide			(57)			
Hydroxide;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hf++++	dis	oth/un	25°C	0.00	U			*B2=-24.4	1975CCa (11550)	29
Hf++++	dis	NaClO4	20°C	4.00M	U	T		*K1=-0.2	1973NOa (11551)	30
Medium: HClO4; *K1=-1.1(25 C) determined with fluoride-ion selective electrode										
Hf++++	sp	KNO3	25°C	0.10M	U	I	K1=14.05 B3=40.86 B4=53.37	B2=27.66	1971NAd (11552)	31
K1=14.07, B2=27.65, B3=40.74, B4=53.54(I=0.3); K1=14.10, B2=27.68, B3=40.62, B4=53.18(I=0.5); K1=14.15, B2=27.83, B3=41.07, B4=54.11(I=1)										
Hf++++	sp	none	20°C	0.0	U			Kso(Hf(OH)4)=-53.43	1962KBc (11553)	32
Hf++++	dis	NaClO4	25°C	1.0M	U		*K1=0.12 *K2=-0.23 *K3=-0.42 *K4=-0.52		1962PAC (11554)	33
Hf++++	dis	NaClO4	25°C	1.0M	U		*B(3,4)=4.37 *B(4,8)=8.00		1962PAC (11555)	34
Hf++++	sol	oth/un	25°C	1.0M	U			K(Hf(OH)4(s)+OH=Hf(OH)5)=-3.2	1960SPa (11556)	35
Hf++++	gl	oth/un	25°C	var	U			Ks(Hf(OH)4=Hf(OH)2+2OH)=-25.4	1950LGa (11557)	36

O2--		H2L		Peroxide			CAS 7772-84-1 (2813)			
Peroxide; -0.0-										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Hf++++ sp oth/un 25°C var U 1973KPf (12666) 37
K(HfO+H2L)=1.9 to 4.5

Hf++++ kin oth/un 0°C var U 1970RAB (12667) 38
K(Hf'+H2L)=5.1

[Hf']=[Hf(OH)]+[Hf(OH)2]+[Hf(OH)3]

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ sp NaCl04 20°C 1.00M U 1972DSg (13207) 39
K(Hf+3HL)=1.57

SCN- HL Thiocyanate CAS 463-56-9 (106)
Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis NaCl04 25°C 3.50M U M 1976TSa (15019) 40
K(HfOH+SCN)=1.04

Hf++++ dis NaCl04 ? 3.0M U K1=1.13 B2=2.33 1971LFb (15020) 41
B4=2.22

Hf++++ sp non-aq 20°C 100% U I K1=1.77 B2=3.49 1970GLa (15021) 42
B3=4.12
B4=6.68
B5=8.13
B6=9.49
B7=10.79, B8=12.05; Medium: N,N-dimethylformamide. In acetonitrile, B6=16.5

Hf++++ sp NaCl04 ? ? U I K1=2.6 B2=4.9 1966GSI (15022) 43
B3=7.1
B4=9.2
B5=11.1
B6=12.9

B7=14.7, B8=16.5; constants for 0.8 M H+. In 0.1 M H+: K1=2.0, B2=4.0,
B3=5.7, B4=7.2, B5=8.7, B6=10.0, B7=11.1, B8=12.2

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ kin NaCl04 25°C 0.02M U 1979ABb (16227) 44
K(Hf(OH)3+HSO4=Hf(OH)2SO4+H2O)=5.42; K(Hf(OH)2+2HSO4=Hf(OH)(SO4)2+2H2O)=9.11

Hf++++ dis NaClO4 25°C 4.00M U K1=1.00 B2=2.26 1976TSa (16228) 45

Hf++++ dis NaClO4 25°C 1.0M U 1971LSa (16229) 46

*K1=1.9

*B2=2.8

Hf++++ ix NaClO4 20°C 2.0M U K1=3.10 B2=5.42 1967EMc (16230) 47

B(HfLC1)=3.04

B(HfL(NO3))=3.23

B(HfL2(NO3))=5.7

Hf++++ dis NaClO4 27°C 2.0M U 1965DKa (16231) 48

*K1=2.04

*B2=3.7

Hf++++ ix NaClO4 ? 2.30M U 1964RMd (16232) 49

*K1=2.11

*B2=3.32

*B3=6.48

Hf++++ dis NaClO4 25°C 2.0M U K1=3.11 B2=5.48 1963PAd (16233) 50

Hf++++ ix NaClO4 ? 2.30M U 1962REb (16234) 51

*K1=2.11

*B2=3.32

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)

Ethanedioic acid; (COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis NaClO4 25°C 4.00M U M 1976TSa (18915) 52

K(HfOH+H2L=HfL+H)=4.40

Hf++++ dis NaClO4 20°C 2.00M U K1=10.22 B2=18.88 1971NSa (18916) 53

Hf++++ ix NaClO4 ? 2.0M U I 1962MRb (18917) 54

K(Hf+H2L=HfL+2H)=5.2

K(Hf+2H2L=HfL2+4H)=9.7

Medium: HClO4. I=4: K(Hf+H2L=HfL+2H)=5.2

C2H4O2 HL Acetic acid CAS 64-19-7 (36)

Ethanoic acid; CH3.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ kin none 25°C 0.00 U M 1973VPa (19981) 55

K(Hf(OH)2+L)=6.71

Hf++++ EMF oth/un 25°C 0.01M U M 1969KPb (19982) 56

$K(\text{Hf}(\text{OH})_3 + \text{L}) = 3.03$

$K(\text{Hf}(\text{OH})_3 + \text{L}) = 1.90$

Medium: 0.01 HfOCl₂

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO ₄	25°C	2.00M	U		K ₁ =4.94 B ₂ =9.85	1981HLA (20553)	57

Medium: 2.0 M HClO₄

Hf++++ kin oth/un 25°C 0.10M U I 1973KPg (20554) 58
 $K(\text{Hf}(\text{OH})_2 + \text{L}) = 7.3$

I=0: K=7.3; I=0.01: K=7.1

C2H5NO₂ HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H₂N.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	kin	oth/un	25°C	0.10M	U			1971KPc (21570)	59

$K(\text{Hf}(\text{OH})_3 + \text{L}) = 1.46$

C2H8O₇P₂ H5L CAS 76267-75-9 (4226)
2-Hydroxyethylidenediphosphonic acid; HO.CH₂.CH(PO₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	oth/un	25°C	2.0M	U			1999VKa (23409)	60

$K(\text{Hf} + \text{H}_3\text{L} = \text{HfH}_3\text{L}) = 7.97$

In 2.0 M HClO₄, T=room

C3H2O₅ H2L Mesoxalic acid (3544)
Oxopropanedioic acid; HOOC.CO.COOH (Ketomalonic acid)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	ix	NaClO ₄	?	2.0M	U		K ₁ =4	1960REa (23489)	61

C3H4N₂ L Imidazole CAS 288-32-4 (90)
1,3-Diazole, imidazole; C₃H₄N₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	gl	KN ₃	25°C	0.10M	C	H	K ₁ =7.02	1976EWa (23897)	62

By calorimetry: DH=-33.27 kJ mol⁻¹, DS=22.74

C3H6O₃ HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	25°C	2.00M	U		K1=5.61 B2=11.04 B3=15.30	1981HLA (25457)	63

Medium: 2.0 M HClO4

Hf++++	ix	NaClO4	25°C	2.0M	U		K(Hf+HL=HfL+H)=1.73 K(Hf+2HL=HfL2+2H)=2.0	1964RMD (25458)	64
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Hf++++	ix	oth/un	?	2.0M	U		K(Hf+HL=HfL+H)=1.73	1960REa (25459)	65
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 C3H7NO3 HL Serine CAS 56-45-1 (49)
 2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	kin	oth/un	25°C	0.10M	U		K(Hf(OH)3+HL)=1.28 pH 2	1973WIA (27138)	66

 C3H12NO9P3 H6L NTPA CAS 6419-19-8 (2920)
 Nitrilotris(methylenephosphonic acid); N(CH2PO3H2)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	oth/un	25°C	2.0M	U		K(Hf+H3L=HfH3L)=12.51	1999VKA (28571)	67

In 2.0 M HClO4, T=room

 C4H6O5 H2L Malic acid CAS 617-48-1 (393)
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	25°C	2.00M	U		K1=4.81 B2=8.98 B3=15.82	1981HLA (30641)	68

Medium: 2.0 M HClO4

Hf++++	ix	NaClO4	?	2.0M	U		K(Hf+H2L=HfHL+H)=1.83	1964RMD (30642)	69
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Medium: HClO4

Hf++++	ix	NaClO4	?	2.0M	U	I	K(Hf+H2L=HfHL+H)=1.53	1960REa (30643)	70
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Medium: HClO4. K=2.16(I=1)

 C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	25°C	2.00M	U		K1=4.68 B2=8.68	1981HLa (31268)	71
Medium: 2.0 M HClO4									
Hf++++	sp	KCl	25°C	1.00M	U		K(Hf(OH)+HL)=8.55	1978KKf (31269)	72
Hf++++	EMF	oth/un	25°C	?	U		K(Hf(OH)3+L)=6.03	1970KKb (31270)	73
Hf++++	ix	NaClO4	?	2.0M	U		K(Hf+H2L=HfHL+H)=2.99	1964RMd (31271)	74
Hf++++	ix	NaClO4	?	1.0M	U	I	K(Hf+H2L=HfHL+H)=2.45	1960REa (31272)	75
Medium: HClO4									
Hf++++	dis	NaClO4	?	2.0M	U		K(Hf+H2L=HfHL+H)=1.69	1959REa (31273)	76
Medium: HClO4									

C4H6O6		H2L		meso-Tartaric			CAS 147-73-9 (91)		
meso-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH									
Hf++++	sp	KCl	25°C	1.00M	U		K(Hf(OH)+HL)=8.70	1978KKf (31429)	77

C4H7NO4		H2L		IDA			CAS 142-73-4 (118)		
Iminodiethanoic acid; HN(CH2.COOH)2									
Hf++++	EMF	KNO3	35°C	0.10M	U		K1=10.9	1978RSa (32270)	78

C5H8O2		HL		Acetylacetone			CAS 123-54-6 (164)		
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3									
Hf++++	dis	oth/un	25°C	1.0M	U		K1=7.40 B2=14.68	1961PAa (37984)	79
K3=6.74									
K4=6.68									

C5H8O7		H2L					CAS 40120-71-6 (3022)		
2,3,4-Trihydroxypentanedioic acid, Trihydroxyglutaric acid; HOOC.(CH(OH))3.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	KCl	25°C	1.00M	U		K(Hf(OH)+HL)=9.37	1978KKf (38424)	80
Hf++++	EMF	oth/un	25°C	0.10M	U		K(Hf(OH)3+L)=6.50	1970KKb (38425)	81
Hf++++	ix	NaClO4	?	2.0M	U		K(Hf+H2L=HfHL+H)=2.72 K(Hf+2H2L=Hf(HL)2+2H)=5.2	1964RMd (38426)	82
Hf++++	ix	NaClO4	?	2.0M	U		K(Hf+H2L=HfHL+H)=2.42 K(Hf+2H2L=Hf(HL)2+2H)=4.9	1960REa (38427)	83

Medium: HClO4

C6H2O4Cl2 H2L Chloranilic acid CAS 87-88-7 (1281)
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	NaClO4	25°C	3.0M	U		K(Hf+H2L=HfL+2H)=3.73 K(Hf+3H2L=HfL3+6H)=11.63	1967VVb (42054)	84

Medium: HClO4

C6H6O2 H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	ix	NaClO4	?	1.0M	U		K1=22.58	1967EKb (43771)	85

Medium: HClO4

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)
1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	KCl	19°C	0.10M	U	I		1966PRc (43963)	86

K(Hf(OH)3+H3L=Hf(OH)2HL+H)=4.44(I=0), 4.33(I=0.1)

C6H6O3 HL Maltol CAS 118-71-8 (2442)
3-Hydroxy-2-methyl-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	20°C	1.00M	U	I	K1=13.16 B2=24.48	1972HSc (44090)	87

Medium: HClO4. I=2.0 M, K1=13.24, B2=24.18

C6H6O4 HL Kojic acid CAS 501-30-4 (1800)
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	20°C	1.00M	U	I	K1=12.04 B2=22.59	1972HSc (44221)	88

Medium: HClO4. I=2.0 M, K1=12.20

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	ix	NaClO4	?	1.0M	U	I	K1=23.00 K(Hf+H2L=HfL+2H)=2.61 K'(Hf+2H2L=HfL2+4H)=4.05	1967EKb (44456)	89

Medium: HClO4. K=3.15(I=0.5), 2.28(I=2.0); K'=4.4(I=0.5)

Hf++++	dis	NaClO4	?	0.20M	U		K1=24.66 B3=66.92	1966KEa (44457)	90
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Medium: HClO4

C6H8O6 H2L Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	oth/un	?	?	U		K1=8.0 K(Hf+HL)=4.2	1966SAb (45642)	91

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	25°C	2.00M	U		K1=5.33 B2=9.11	1981HLA (46128)	92

Medium: 2.0 M HClO4

Hf++++	ix	NaClO4	?	2.0M	U		K(Hf+H3L=HfH2L+H)=2.54	1964RMD (46129)	93
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Hf++++	ix	oth/un	?	2.0M	U		K(Hf+H3L=HfH2L+H)=2.24 ?	1960REa (46130)	94
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Medium: 2 M HClO4. K=3.05(?) (I=1)

C6H9NO6 H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hf++++ cal oth/un 25°C 0.0 U TIH 1981VBa (46846) 95
DH(K1)=20.7 kJ mol-1

Hf++++ ix NaClO4 ? 0.23M U K1=20.34 1966EMd (46847) 96

Hf++++ ix NaClO4 ? 2.0M U I 1964EMc (46848) 97
K(Hf+H3L=HfL+3H)=3.83

Medium: HClO4. K=5.05(I=1)

C6H10O8 H2L Mucic acid CAS 526-99-8 (3650)
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; HOOC.(CHOH)4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ sp KCl 25°C 1.00M U 1978KKf (48436) 98
K(Hf(OH)+HL)=9.68

C6H10O8 H2L Saccharic acid CAS 87-73-0 (1191)
D-2,3,4,5-Tetrahydroxy-1,6-hexanedioic acid, Glucaric acid; HOOC.(CHOH)4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ ix NaClO4 20°C 2.0M U 1963RCa (48476) 99
K(Hf+H2L=HfL+2H)=2.29

Medium: HClO4

C6H11NO5 H2L HIMDA CAS 93-62-9 (192)
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ EMF KNO3 35°C 0.10M U K1=9.7 1978RSa (48740) 100
K(HfL(OH)+H)=8.65

Hf++++ ix NaClO4 ? 0.23M U K1=14.63 1966EMd (48741) 101

Medium: HClO4

C6H12O7 HL Gluconic acid CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ ix NaClO4 20°C 2.0M U 1963RCa (49722) 102
K(Hf+HL=HfL+H)=1.49

Medium: HClO4

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	kin	oth/un	25°C	?	U			1969KMf (54231)	103
							K(Hf(OH)3+HL)=5.5		

C7H6O6S	H3L	CAS 5965-83-3	(399)
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; HO3S.C6H3(OH).COOH			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	kin	oth/un	25°C	?	U			1969KMf (55012)	104
							K(Hf(OH)3+HL)=6.5		

C7H7NO2	H2L	Salicylaldoxime	CAS 94-67-7	(1486)
2-Hydroxybenzaldehyde oxime; HO.C6H4.CH:N.OH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	gl	KCl	25°C	0.10M	U	I	K1=11.05	1968MDe (55310)	105
							K1=16.7(I=0),15.82(I=0.01),14.15(I=0.025),13.00(I=0.05),12.30(I=0.075)		

C8H5O2F3S	HL	TTA	CAS 326-91-0	(165)
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	oth/un	25°C	1.0M	U		K1=10.60 B2=21.44	1962PAa (58628)	106
							B3=31.50		
							B4=41.52		

C8H5O2F3Se	HL	CAS 713-15-5	(3842)
4,4,4-Trifluoro-1-(2'-selenoyl)-butane-1,3-dione; F3C.CO.CH2.CO.C4H3Se			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	oth/un	20°C	1.0M	U		K1=10.46 B2=20.74	1962PAa (58703)	107
							B3=30.22		
							B4=39.70		

C8H7O3Br	HL	CAS 1878-91-7	(3819)
2-(4'-Bromophenyl)-2-hydroxyethanoic acid, p-bromomandelic acid;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	25°C	1.0M	U		K1=7.00 B2=13.15	1961AHa (59244)	108
							K3=6.61		
							K4=6.26		

Medium: HClO4

C8H8O2	HL	Phenylacetic	CAS 103-82-2	(1361)
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Phenylethanoic acid; C6H5.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	kin	none	25°C	0.0	M			1973KPg (59548)	109
							$K(\text{Hf}(\text{OH})_2+2\text{L})=7.0$		

C8H8O3 HL Mandelic Acid CAS 611-72-3 (80)
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaNO3	25°C	1.0M	U			1971PKb (59839)	110
							$K(\text{HfO}+2\text{H}-1\text{L})=17.74$ (?)		

C10H6O8Cl2S2 H4L CAS 6155-33-5 (4761)
2,7-Dichlorochromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	oth/un	?	?	U			1973DMb (68536)	111
							$K(\text{Hf}(\text{OH})_2+2\text{HL})=12.64$		

C10H6O9S2 H3L CAS 58425-39-1 (2004)
8-Hydroxy-1,2-naphthoquinone-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	NaClO4	20°C	0.10M	U			1975MDa (68540)	112
							$B(\text{Hf}(\text{OH})_2\text{L}_2)=17.66$		

C10H10O2 HL Benzoylacetone CAS 93-91-4 (197)
1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	dis	NaClO4	25°C	1.00M	C			1975LUa (70733)	113
							$K_d(\text{Hf})=0.6$		
							$K(\text{HfL}_4=\text{HfL}_4(\text{org}))=3.6$		

Organic phase=benzene; $B(\text{HfO}+4\text{L}+2\text{H}=\text{HfL}_4+\text{H}_2\text{O})=41.8$
 $K_d(\text{Hf}): \text{HfO}+4\text{HL}(\text{org})=\text{HfL}_4(\text{org})+2\text{H}+\text{H}_2\text{O}$

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	ISE	KN03	25°C	0.10M	C	M		1996YHa (73830)	114
							$K(\text{HfL}+\text{F})=4.56$		
							$K(\text{HfLF}+\text{F})=2.9$		

Method: Fluoride ISE.

Hf++++ EMF KNO3 35°C 0.10M U 1978RSa (73831) 115
K(HfL(OH)+H)=5.24
K(2HfL(OH)2+2H=2HfL)=9.00

Hf++++ sp KCl 25°C 0.50M U K1=15.1 1978TSa (73832) 116

Hf++++ ix NaClO4 ? 0.23M U K1=29.5 1966EMd (73833) 117
Medium: HClO4

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hf++++	ISE	KNO3	25°C	0.10M	C	M		K(HfL+F)=5.37 K(HfLF+F)=3.82 K(HfH-1L+H)=1.55 K(HfH-1L+F)=2.9	1996YHa (75410)	118

Method: Fluoride ISE. K(HfH-1LF+F)=1.9.

Hf++++ EMF KNO3 35°C 0.10M U 1978RSa (75411) 119
K(HfL(OH)+H)=9.24
K(2HfL(OH)2+2H=2HfL)=16.15

C11H9N3O2 H2L PAR CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hf++++	vlt	alc/w	25°C	50%	U			K(Hf(OH)3+HL)=16.9	1975TBa (77548)	120

Medium: 50% EtOH/H2O

C11H12O9 H3L CAS 69065-58-3 (2714)
1,2,4-Trihydroxy-3,4,5-trimethoxycarbonylcyclopentadiene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hf++++	sp	NaCl	19°C	0.1M	U			K(Hf(OH)2+HL)=13.28 K(Hf(OH)3+HL)=12.15	1977Lba (78428)	121

C11H18N2O8 H4L CAS 4408-81-5 (923)
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOCH2)2N.CH2.)2.CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hf++++ dis NaClO4 ? 1.0M U K1=27.65 1968EMa (79447) 122
K(Hf+2HL)=54.43

C12H12Si L (6825)
Diphenylsilane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ nmr non-aq 25°C 100% U M 1992WHa (81170) 123
K(HfABCD+L=HfABC(H-1L)+E)=-0.3

Method:NMR. Medium:benzene. A:cyclopentadienide. B:pentamethylcyclopentadienide. C:Cl. D:Si(C6H11)H2. E:Si(C6H11)H3.

C12H19O3P HL CAS 66170-45-4 (8310)
Phenylphosphonic acid monoethyl ester;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis NaCl RT 2.0M C 1977NAc (81992) 124
K(Hf+5HL(org)+Cl=HfL3Cl(HL)2(org)+3H)=23.3

Method: extraction from 2.0 M NaCl solution into benzene.

C12H27O4P L CAS 126-73-8 (2432)
Tri-n-butyl phosphate; (C4H9O)3PO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis oth/un 20°C ? U K1=-0.12 1962PBa (84120) 125
Metal: Hf+++

C12H27O6P HL CAS 14260-97-0 (8268)
Di-(n-butoxyethyl)phosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis non-aq RT 100% C 1977NAb (84126) 126
Medium: benzene. By distribution from 2 M NaCl/HCl or 2 M NaClO4/HClO4.

K(Hf+6HL(org)=HfL4(HL)2(org)+4H)=27.1

C13H11N02 HL CAS 304-88-1 (181)
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ sp NaClO4 25°C 1.0M U K1=13.66 B2=26.90 1968FOa (85155) 127
K3=12.25
K4=12.15

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	oth/un	25°C	?	U		K(?)=10.4	1962BDa (86733)	128

Hf++++	sp	oth/un	25°C	?	U		B2=10.4	1959DBb (86734)	129
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C14H22N2O8 H4L CDTA CAS 482-54-2 (200)

trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	ISE	KNO3	25°C	0.10M	C	M		1996YHa (88673)	130

K(HfL+F)=4.50

K(HfLF+F)=3.1

Method: Fluoride ISE.

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)

Diethylenetriamine-pentaethanoic acid; H00C.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	ix	NaClO4	?	0.23M	U		K1=35.40	1966EMd (89271)	131
Hf++++	ix	NaClO4	?	2.0M	U	I		1964EMd (89272)	132

K(Hf+H5L=HfL+5H)=3.13

Medium: HClO4. K=4.86(I=1)

C15H11N3O HL PAN CAS 85-85-8 (572)

1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	vlt	alc/w	25°C	50%	U			1975TBa (91220)	133

K(Hf(OH)3+HL)=14.4

Medium: 50% EtOH/H2O

C16H11N3O10S2 H4L (5174)

2-Hydroxy-1-(2'-hydroxy-4'-nitro)phenylazo-3,6-disulfonaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hf++++	sp	oth/un	25°C	?	U		K(?)=5.33	1971RCd (92881)	134

C16H12O6 H2L CAS 475-25-2 (5141)

Hematein, haematin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hf++++ sp oth/un 20°C 0.10M U 1973DVa (93154) 135
K(HfO+2L)=6.20 pH=2

C16H13N2O10AsS2 H5L Thorin I CAS 3688-92-4 (2609)
1-((2-Arsonophenyl)azo)-2-hydroxy-3,6-naphthalylidysulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ sp oth/un 25°C ? U 1966SAd (93194) 136
K(?)=8.2

C16H27O3P HL CAS 52299-33-9 (8311)
Phenylphosphonic acid monodecyl ester;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ dis NaCl RT 2.0M C 1977NAC (94696) 137
K(Hf+3HL(org)+Cl=HfL3Cl(org)+3H)=12.4
Method: extraction from 2.0 M NaCl solution into benzene.

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)
Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ ISE KNO3 25°C 0.10M C M 1996YHa (98044) 138
K(HfHL+F)=5.58
K(HfL+H)=3.86
K(HfL+F)=2.8

Method: Fluoride ISE.

Hf++++ ix NaClO4 ? 0.50M U I K1=19.08 1966EMc (98045) 139
K(Hf(OH)+H6L=HfH2L+3H)=3.90
Medium: HClO4. K(Hf+H6L=HfH2L+4H)=3.5(I=2), 2.84(I=2)

C22H37N5O14 H7L CAS 3234-59-1 (2425)
Tetraethylenepentamineheptaethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hf++++ ix NaClO4 ? 0.20M U I 1966EMc (102328) 140
K(Hf(OH)+H7L=HfH4L+2H)=6.11

Medium: HClO4. K(Hf(OH)+H7L=HfH4L+2H)=5.72(I=0.5)
K(Hf+H7L=HfH4L+3H)=5.18(I=1), 4.23(I=2)

C23H24N4O2 L Trichachnine CAS 1251-85-0 (2606)
4,4'-Diantipyrylmethane,
4,4'-phenylmethylen-bis-(1,2-dihydro-1,5-dimethyl-2-phenylpyrazol-3-one

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hf++++	sp	KCl	25°C	0.10M	U			K1=7.82 B2=11.92	1982SKb (102673)	141
Hf++++	dis	oth/un	RT	2.0M	M			B2=7.40 B3=11.20	1975HSa (102674)	142
Medium: 2 M HCl. Extraction into benzene from 2 M HCl, using 181Hf. *****										
C31H32N2O13S		H6L		Xylenol orange		CAS 63721-85-5		(432)		
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hf++++	sp	NaClO4	?	0.30M	U			K(?)=6.51	1960CHa (105472)	143

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
 I Data with various BACKGROUNDS
 H Data for THERMOCHEMICAL quantities
 M Data for TERNARY Complexes

END