

## SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 119 experiments for  
(no ligands specified)

5 metals : W(0), W(III), W(IV), W(V), W(VI)

(no references specified)

(no experimental details specified)

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CO L Carbon monoxide CAS 630-08-0 (551)

Carbon monoxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(0)	gl	oth/un	0°C	var	U	M			1959HEb (2824)	1
								K(H3W3L9(OH)2(H2O)+H)=1.5		
								K(H2W3L9(OH)2(H2O)+H)=5.4		

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C3H9O3P L CAS 121-45-9 (1786)

Trimethylphosphite; (CH3O)3P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(0)	cal	non-aq	25°C	100%	U	HM			1991ZGa (28004)	2
								Medium: THF. DH(Mo(CO)3A2+L)=-110.9 kJ mol <sup>-1</sup> , A=P(C6H11)3		

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C18H33P L CAS 2622-14-2 (169)

Tri-(cyclohexyl)phosphine; (C6H11)3P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(0)	cal	non-aq	25°C	100%	U	HM			1991ZGa (98316)	3
								K(W(CO)3py2+L)=-6.84		

Medium: THF. DH=-79.1 kJ mol<sup>-1</sup>

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C3H9P L CAS 594-09-2 (1732)

Trimethyl phosphine; (CH3)3P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(III)	nmr	non-aq	30°C	100%	U	T HM			1992RZa (28058)	4
								K(WL6=WL4A(H)+L)=1.25		

Metal::W(0). Method:NMR. Medium:C6D6. T=30-70C. K=1.48(40C); 1.78(50C); 1.92(60C); 2.00(70C). DH=38.9 kJ mol<sup>-1</sup>; DS=155. A:P(CH3)2(CH2).

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C8H19P L (6822)

Di(t-Butyl)phosphine; ((CH3)3C)2PH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(III) nmr non-aq ? 100% U T HM 1992BCc (63202) 5  
 Metal:W+++ . Method:NMR. Medium:toluene. DH(1,2-W2L2(NMe2)4(anti-gauche  
 isomerization)=-2.1 kJ mol-1, DS=-1.3. Data also for other phosphides

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CN- HL Cyanide CAS 74-90-8 (230)  
 Cyanide;

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(IV) kin KNO3 25°C 1.00M U 1995SPb (2775) 6  
 $K(WO(H2O)L4+L)=3.0$   
 $K(WO(H2O)L4+HL=WO(HL)L4)=0.0$   
 $*K(WO(HL)L4)=-5.8$

-----  
 W(IV) EMF KCl 20°C 0.10M U I 1973HKa (2776) 7  
 $K(WO2(CN)4+H)=11.7$   
 $K'(WO2(CN)4H+H)=8.25$   
 $K=12.0(I=0.014); 12.1(I=0.04); 11.8(I=0.06); 11.6(I=0.2); 11.6(I=0.5);$   
 $-12.1(I=0). I=0: K=12.1. K'=8.57(I=0.014); -7.76(I=0.5). I=0: K'=8.84$

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 W(IV) gl oth/un 20°C 0.10M U 1971SKc (2777) 8  
 $K(H+W(CN)8)=1.6$

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F- HL Fluoride CAS 7644-39-3 (201)  
 Fluoride;

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(IV) sp KNO3 20°C 1.00M U M 1986LBa (7331) 9  
 $K(WO(H2O)(CN)4+F)=2.15$

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OH- HL Hydroxide (57)  
 Hydroxide;

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

W(IV) gl KCl 25°C 1.2M C 1998ARa (12484) 10  
 $*K(WO(H2O)(CN)4)=-7.89$   
 $*K(WO(OH)(CN)4)=-14.5$

Medium: KCl/KNO3

-----  
 W(IV) EMF oth/un 16°C var U 1959LMa (12485) 11  
 $K(W(CN)4(OH)+OH)=9.10$   
 $K(W(CN)4(OH)2+OH)=6.67$   
 $K(W(CN)4(OH)3+OH)=3.28$

Metal is W(IV). Method: Bi electrode

-----  
 W(IV) gl oth/un 25°C var U 1956BAa (12486) 12  
 $K1(W(CN)8+H) < 2$   
 $K1(W(CN)8+H) < 2$

C2H4	L	Ethylene	CAS 74-85-1	(478)
Ethene; H2C:CH2				

Method:NMR. Medium:C6D6. A=PMePh2. When A=PMe3, K=-3.02

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e-           HL      Electron          (442)
Electron;

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W(V) vlt oth/un 25°C 12.0M U 1952LAb (1030) 15  
K(W+e=W(IV))=-5(-300 mV)  
K(W+2e=W(III))=-7(red WC15,-200 mV), -3.4(W(III),-100 mV), 3.4(green W2C19)

W(V) EMF KCl  $\theta^{\circ}\text{C}$  1.0M U I 1924COa (1031) 16  
K(W(CN)<sub>8</sub>+e)=11.0(597 mV)  
At I=0.5 M K=K=10.7(580 mV), I=0.25 M: K=10.5(568 mV)

CN- Cyanide;	HL	Cyanide	CAS 74-90-8 (230)
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W=W(IV). Method: N.M.R.

W(V)	gl	oth/un	20°C	0.10M	U	1971SKc	(2779)	18
						K(H+W(CN)8)=2.35		
						K(H+HW(CN)8)=1.7		

Cl- Chloride;	HL	Chloride	CAS 7647-01-0 (50)
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K:  $2\text{WOC15} + 6\text{H}_2\text{O} = \text{W2L204}(\text{OH})_2(\text{H}_2\text{O})_2 + 6\text{H} + 8\text{Cl}$ .  $K = 6.61(1 \text{ C})$ ,  $6.30(40 \text{ C})$ .

DH(K)=-14.6 kJ mol<sup>-1</sup>, DS=74 J K<sup>-1</sup> mol<sup>-1</sup>. Method: magnetic susceptibility

W(V)	oth KCl	40°C	var	U T	1967JRa	(5949)	20
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$$K(2WOCl_5 + 6H_2O = X + 8Cl + 6H) = 12.58$$

Method:magnetic susceptibility. Medium:HCl var. K=13.23(1 C),12.94(20 C)  
X=W2O2(OH)6Cl2

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OH- HL Hydroxide (57)  
Hydroxide;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(V) kin oth/un 25°C 2.00M U 1993PSa (12487) 21  
\*K(Mo2W04(H2O))=-0.013  
\*K(Mo2W04(NCS))=-1.7

Metals are W(IV) and Mo(IV). Medium: 2.0 M Li(tetrafluoromethane sulfonate).

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SCN- HL Thiocyanate CAS 463-56-9 (106)  
Thiocyanate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(V) kin NaCl04 25°C 2.00M U 1993VSa (15338) 22  
K(W3S4(H2O)9+L)=3.18  
K(MoW2S4(H2O)9+L)=3.29  
K(Mo2WS4(H2O)9+L)=3.02

Medium: 2.0 M HCl04. Metals are W(IV) and Mo(IV). For mixed Mo/W species the data refer to binding of L to W.

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C5H9N L t-Butylnitrile CAS 7188-38-7 (913)  
t-Butylcyanide;(CH3)3C.CN

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(V) con non-aq 40°C 100% U M 1992LIa (38458) 23  
K(WL6I+I)=3.11

Medium: MeCN, 0.0063 M Bu4NCl04, W++. Contradictory data in Tables and text

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e- HL Electron (442)  
Electron;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

W(VI) vlt oth/un 25°C 0.0 U I 1952LAb (1032) 24  
K(WO2Cl3+e)=4.4(260 mV)  
K'=-9(-90 mV)  
K"=-107(1050 mV)

K': WO3(s)+6H+6e=W(s)+3H2O. K'': WO4+4H2O+6e=W(s)+8)H. K(WO2Cl3+e) in 12M HCl

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CN- HL Cyanide CAS 74-90-8 (230)  
Cyanide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(VI) sp KCl 20°C 0.20M U 1978SSc (2780) 25  
K(WL7+H)=4.82  
K(WL70H+H)=9.17

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F- HL Fluoride CAS 7644-39-3 (201)  
Fluoride;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
W(VI) nmr non-aq -75°C 100% U K1=1.7 B2=2.40 1974SBc (7332) 26  
Medium: vinyl chloride

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W(VI) kin KCl 25°C 0.20M U 1964YPa (7333) 27  
K1eff=3.0

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W(VI) con non-aq -5°C 100% U 1960NVa (7334) 28  
K(WO2F2+4HF=WF6+2H2O)=-2.8

Medium: liquid HF, m units

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MoO4-- H2L Molybdate (443)  
Molybdate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
W(VI) nmr oth/un 25°C >6 U 1994AHa (8761) 29  
Beff(1,6)=1.00  
Beff(2,5)=1.57  
Beff(3,4)=1.83  
Beff(4,3)=1.83

Beff(5,2)=1.55, Beff(6,1)=0.98. Beff(q,r): pH+qMo7024+rW7024 at pH 6.0

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NH2SO3- H2L Sulfamate CAS 5329-14-6 (452)  
Sulfamate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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W(VI) sp NaCl 25°C 1.0M U 1958SAc (8804) 30  
K(2H+2L+WO4=WO3L2+H2O)=-8 ?

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OH- HL Hydroxide (57)  
Hydroxide;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
W(VI) sp NaClO4 25°C 2.00M U 1992RSb (12488) 31  
\*K(W3S4(H2O)9)=-0.59  
Medium: 2.0 M LiClO4. Metal is W(IV).

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W(VI) sol oth/un 300°C var M T H 1992W0a (12489) 32

300-600 C and P=1 kbar.  $\Delta H(K)=41 \text{ kJ mol}^{-1}$ . Constant at  $I=0$

In NaCl:  $K=0.20$ ; in LiCl:  $K=0.78$

W(VI)	gl	NaClO <sub>4</sub>	25°C	3.00M	C	B(2,1)=11.30 B(6,6)=52.46 B(7,6)=60.76 B(14,12)=123.24	1974ASa (12492)	35
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W(VI)=W02++. K1=13.28, B2=26.2, B3=38.7(I=2.5)

02--	H2L	Peroxide	CAS 7772-84-1	(2813)
Peroxide; -0.0-				

W(VI) sp oth/un 24°C var U 1963DLa (12758) 37  
K(WL4+H)=4.9 to 6.9  
K(WL4+H2L=HWL4+HL)=-6.9 to -8

SCN- Thiocyanate;	HL	Thiocyanate	CAS 463-56-9 (106)
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W(VI) sp non-aq ? 100% U I K1=3.5 B2=7.2 1967USa (15339) 38  
Medium: Me2CO. W added as WC15. Conductivity also used. In MeCOEt: K1=3.1,  
B2=6.3, B4=13.25. In cyclohexanone: B6=20.6

W04-- H2L Tungstate CAS 13783-36-3 (445)  
Tungstate;

W(VI)      gl   NaCl   25°C 1.00M U      1979IRa (17453)   39  
K(7H+6W04=HW6021+3H2O)=56.42  
K(9H+6W04=H3W6021+3H2O)=70.45

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CH2O2                      HL      Formic acid              CAS 64-18-6    (37)  
Methanoic acid; H.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	ix	oth/un	?	0.05M	U		K1=0.30    B2=2.28 B3=3.18 B4=4.90	1970SHa (17661)	40

Metal ion: WO2++. Medium: 0.01-0.05 HL, pH 2.5

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CH4O                      L      Methyl alcohol      CAS 67-56-1    (597)  
Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	EMF	alc/w	20°C	100%	U			1964GUa (17911)	41
							K(WO(L')3+L'=WO(L')4)=12.51 K(WO(L')5+H=WO(L')4+L)=4.09		

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. L'=H-1L (i.e. CH3O)

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C2H2O4                      H2L      Oxalic acid              CAS 144-62-7    (24)  
Ethanedioic acid; (COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	gl	KN03	21°C	0.22M	C			1978MBc (19152)	42
							K(WO4+2H+L=WO3L+H2O)=13.97		

Medium pH 5-7.

W(VI)	oth	oth/un	?	?	U		K1=1.48	1969SHd (19153)	43
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Metal ion: WO2++

W(VI)	vlt	oth/un	25°C	0.16M	U			1962YBa (19154)	44
							K(H2WO4+H2L)=4.85 K(H2WO3L+H2L)=7.5		

Medium: 0.08-0.24 H2SO4

W(VI)	kin	oth/un	25°C	?	U			1962YPb (19155)	45
							K(H2WO4+H2L)=5.13		

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C2H4O3                      HL      Glycolic acid              CAS 79-14-1    (33)  
2-Hydroxyethanoic acid; HO.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C			1995HCa (20656)	46
							Keff(WO4+2L+2H=WO2L2)=16.85		

Medium: 0.1 M acetate buffer, pH 4.7

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C3H4O4                      H2L      Malonic acid              CAS 141-82-2    (79)

Propanedioic acid; CH<sub>2</sub>(COOH)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	kin	oth/un	25°C	0.05M	U			1962YPa (24594)	47
							K(H <sub>2</sub> W <sub>04</sub> +H <sub>2</sub> L)=3.09		

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C<sub>3</sub>H<sub>6</sub>O<sub>3</sub> HL L-Lactic acid CAS 79-33-4 (82)  
L-2-Hydroxypropanoic acid; CH<sub>3</sub>.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C			1995HCa (25570)	48
							K <sub>eff</sub> (W <sub>04</sub> +2L+2H=W <sub>02</sub> L <sub>2</sub> )=18.15		

Medium:0.1 M acetate buffer, pH 4.7

W(VI)	gl	NaCl	25°C	1.00M	C	H		1993CKb (25571)	49
							B(1,2,2)=17.47		
							B(1,2,3)=18.38		
							B(1,1,2)=13.03		
							B(1,1,3)=14.56		

B(p,q,r): pW<sub>04</sub> +qHL +rH =(W<sub>04</sub>)pLqHq+r. B(2,2,3)=25.47. DH by calorimetry:  
DH(1,2,2)=-80 kJ mol<sup>-1</sup>.

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C<sub>3</sub>H<sub>7</sub>N<sub>02</sub>S H<sub>2</sub>L Cysteine CAS 52-90-4 (96)  
2-Amino-3-mercaptopropanoic acid; H<sub>2</sub>N.CH(CH<sub>2</sub>.SH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	sp	NaCl	18°C	1.00M	U			1990CJa (26853)	50
							K(W <sub>04</sub> +L+2H=W <sub>03</sub> L+H <sub>2</sub> O)=18.8		

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C<sub>3</sub>H<sub>9</sub>P L CAS 594-09-2 (1732)  
Trimethyl phosphine; (CH<sub>3</sub>)<sub>3</sub>P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	nmr	non-aq	25°C	100%	U	T HM		1992Wwa (28059)	51
							K(WABC <sub>2</sub> +L)=2.73		

Method:NMR. Medium:Toluene. T=-10-40. K=4.23(-10C);3.73(0);3.28(10);2.96(18)  
;2.10(40). A:CHC(CH<sub>3</sub>)<sub>3</sub>. B:NC<sub>6</sub>H<sub>4</sub>(i-C<sub>3</sub>H<sub>7</sub>)<sub>2</sub> C:OC(CH<sub>3</sub>)<sub>3</sub>. DH=-65.7kJmol<sup>-1</sup>;DS=-170

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C<sub>4</sub>H<sub>6</sub>O<sub>4</sub> H<sub>2</sub>L Succinic acid CAS 110-15-6 (112)  
1,4-Butanedioic acid; HOOC.CH<sub>2</sub>.CH<sub>2</sub>.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	ix	oth/un	22°C	0.10M	U		K <sub>1</sub> =1.06	1973SDa (30076)	52
Metal ion: W <sub>02</sub> ++, pH 2.5									

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C4H6O5                      H2L      Malic acid                      CAS 617-48-1    (393)  
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	gl	NaCl	25°C	1.0M	C	H			1997CKa (30756)	53
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B(1,1,1)=8.85  
 B(1,1,2)=14.78  
 B(1,1,3)=17.26  
 B(1,2,2)=17.20

B(p,q,r):pW04+qHL+rH=(W04)p(HL)qHr. B(1,2,3)=21.70, B(1,2,4)=25.61, B(2,1,3)=23.07, B(2,2,2)=18.87, B(2,2,3)=25.72, B(2,2,4)=31.88. DH by calorimetry.

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W(VI)	oth	NaClO4	30°C	1.00M	U	M			1979CBa (30757)	54
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B((W04)H2L2)=23.1

Method: polarimetry

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C4H6O6                      H2L      DL-Tartaric acid                      CAS 133-37-9    (94)  
 DL-Tartaric acid,DL-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	sp	oth/un	25°C	0.10M	C				1995HCa (31037)	55
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Keff(2W04+2L+4H)=(W0)2L2)=33.55

Medium: 0.1 M acetate buffer, pH 4.7

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C4H6O6                      H2L      L-Tartaric acid                      CAS 87-69-4    (92)  
 L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	kin	oth/un	25°C	0.10M	U				1962YPb (31396)	56
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K(H2W04+H2L)=3.93

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C4H7NO4                      H2L      Aspartic acid                      CAS 56-84-8    (21)  
 Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	gl	NaClO4	25°C	0.10M	U			K1=8.20    B2=14.07	1972SSe (31978)	57
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K3=3.81

Metal ion is W02++

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C4H7NO4                      H2L      IDA                      CAS 142-73-4    (118)  
 Iminodiethanoic acid; HN(CH2.COOH)2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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W(VI)	gl	NaClO4	25°C	3.0M	U				1979ZLa (32398)	58
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B(W04+L+2H=W03L)=18.14

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W(VI)      gl  oth/un 25°C 0.15M U      1966KR a (32399)  59
                                         K(W04+L+2H=W03L)=18.5
*****
C4H8N2O3      HL      Asparagine      CAS 70-47-3  (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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W(VI)      gl  NaCl04 25°C 0.10M U      K1=5.84   B2=10.95  1973TSe (32746)  60
                                         K3=3.30
*****
C4H8O3      HL      CAS 594-61-6  (81)
2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
W(VI)      sp  oth/un 25°C 0.10M C      1995HCa (33537)  61
                                         Keff(W04+2L+2H=W02L2)=18.75
Medium: 0.1 M acetate buffer, pH 4.7
*****
C4H8O3      HL      CAS 965-70-8  (423)
2-Hydroxybutanoic acid; CH3.CH2.CH(OH).COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
W(VI)      sp  oth/un 25°C 0.10M C      1995HCa (33588)  62
                                         Keff(W04+2L+2H=W02L2)=18.25
Medium: 0.1 M acetate buffer, pH 4.7
*****
C5H9N04      H2L      Glutamic acid  CAS 56-86-0  (22)
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
W(VI)      gl  NaCl04 25°C 0.10M U      K1=7.95   B2=13.40  1972SSe (39147)  63
                                         K3=3.60
*****
C5H9N04      H2L      MIDA      CAS 4408-64-4  (190)
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
W(VI)      gl  oth/un 25°C 0.15M U      1966KR a (39293)  64
                                         K(W04+L+2H=W03L)=18.70
-----
W(VI)      nmr oth/un 35°C 1.00M U      1966KR a (39294)  65
                                         K(W04+L+2H=W03L)=18.6
*****
C5H10N2O3     HL      Glutamine      CAS 56-85-9  (18)

```

2-Aminopentanedioic acid 5-amide; H<sub>2</sub>N.CH(CH<sub>2</sub>.CH<sub>2</sub>.CO.NH<sub>2</sub>)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	gl	NaClO <sub>4</sub>	25°C	0.10M	U		K <sub>1</sub> =5.76 B <sub>2</sub> =10.85 K <sub>3</sub> =3.20	1973TSe (39847)	66

\*\*\*\*\*  
C<sub>5</sub>H<sub>10</sub>O<sub>5</sub> L CAS 1114-34-7 (6113)  
D-Lyxose

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	gl	KCl	25°C	0.10M	C		B(2W <sub>04</sub> +L+2H)=18.08	1989VCa (40340)	67

\*\*\*\*\*  
C<sub>5</sub>H<sub>12</sub>O<sub>5</sub> L Xylitol CAS 87-99-0 (2139)  
Xylitol; HO.CH<sub>2</sub>.HCOH.HOCH.HCOH.CH<sub>2</sub>.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	gl	KCl	RT	0.10M	M I		K(2W <sub>04</sub> +2H+L=W <sub>207</sub> L+H <sub>2</sub> O)=18.50	1990VSc (41694)	68

Data for 0.01-1.0 M KCl and NaCl. In 0.01 M KCl, K=19.65.  
\*\*\*\*\*  
C<sub>6</sub>H<sub>20</sub>Cl<sub>2</sub> H<sub>2</sub>L Chloranilic acid CAS 87-88-7 (1281)  
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	sp	oth/un	30°C	?	U		K <sub>1</sub> =5.23	1981BMD (42061)	69

\*\*\*\*\*  
C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>O<sub>6</sub> H<sub>2</sub>L CAS 7659-29-2 (2694)  
1,2-Dihydroxy-3,5-dinitrobenzene; (HO)<sub>2</sub>.C<sub>6</sub>H<sub>2</sub>(NO<sub>2</sub>)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	sp	NaNO <sub>3</sub>	25°C	0.10M	U		K(W <sub>02</sub> +2L)=20.74	1972PSb (42268)	70

\*\*\*\*\*  
C<sub>6</sub>H<sub>5</sub>N<sub>04</sub> H<sub>2</sub>L 4-Nitrocatechol CAS 3316-09-4 (890)  
1,2-Dihydroxy-4-nitrobenzene; O<sub>2</sub>N.C<sub>6</sub>H<sub>3</sub>(OH)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U		K(H <sub>2</sub> W <sub>04</sub> L+H <sub>2</sub> L=W <sub>02</sub> L <sub>2</sub> +2H <sub>2</sub> O)=3.15 K(W <sub>04</sub> +H <sub>2</sub> L)=3.31	1980NKA (42946)	71

Medium: 0.1 M NH<sub>4</sub>OH, 0.08 M Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>. pH 8  
Medium: 0.1 M NH<sub>4</sub>OH, 0.08 M Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>, pH 8  
\*\*\*\*\*

C6H5O2Cl                      H2L      4-Cl-Catechol      CAS 2138-22-9    (1656)  
 1,2-Dihydroxy-4-chlorobenzene; Cl.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U			1980Nka (43086)	72	
								K(H2WO4L+H2L=WO2L2+2H2O)=3.41		
								K(WO4+H2L)=3.25		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										

W(VI)	sp	KCl	25°C	0.10M	U			1963HAc (43087)	73	
								K(WO4+2H2L)=7.1		

\*\*\*\*\*

C6H6O2                      H2L      Catechol                      CAS 120-80-9    (534)  
 1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U			1980Nka (43864)	74	
								K(H2WO4L+H2L=WO2L2+2H2O)=3.98		
								K(WO4+H2L)=2.59		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										

W(VI)	sp	oth/un	20°C	0.10M	U			1964PCa (43865)	75	
								K(WO4+2H2L=WO2L2)=6.53		

Medium: 0.1 M NaHSO3

W(VI)	sp	oth/un	20°C	?	U			1959HAa (43866)	76	
								K(WO4+2H2L=WO2L2)=6.53		

\*\*\*\*\*

C6H6O3                      H3L      Pyrogallol                      CAS 87-66-1    (696)  
 1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U			1980Nka (43998)	77	
								K(H2WO4L+H2L=WO2L2+2H2O)=3.21		
								K(WO4+H2L)=2.98		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										

W(VI)	sp	oth/un	20°C	?	U			1959HAa (43999)	78	
								K(WO4+2H3L=WO2(HL)2)=6.98		

W(VI)	sp	oth/un	20°C	?	U			1958PIa (44000)	79	
								K(WO4+2H3L=WO2(HL)2)=7.34		

\*\*\*\*\*

C6H8O7                      H3L      Isocitric acid      CAS 1637-73-6    (2527)  
 2-Hydroxy-3-carboxypentanedioic acid; HOOC.CH(OH).CH(COOH).CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----  
W(VI) sp oth/un 25°C 0.10M C 1995HCa (45736) 80  
Keff(2W04+2HL+2H=(W02)2O2L2)=25.7. Medium: 0.1 M acetate buffer, pH 4.7  
\*\*\*\*\*

C6H8O7 H3L Citric acid CAS 77-92-9 (95)  
2-Hydroxypropane-1,2,3-tricarboxylic acid; H00CCH2.CH(OH)(C00H).CH2C00H  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----

W(VI)	gl	NaCl	25°C	1.00M	U	H			1995CRa (46309)	81
								B(111)=10.21		
								B(121)=17.03		
								B(131)=21.67		
								B(141)=22.82		

B(pqr): pW04 + qH + rL = (W04)pHqLr  
-----

W(VI)	gl	NaCl	25°C	1.00M	C	H			1991CKa (46310)	82
								B(1,1,1)=10.21		
								B(1,2,1)=17.03		
								B(1,3,1)=21.67		
								B(1,4,1)=22.82		

B(2,4,2)=34.89, B(2,5,2)=39.33, B(1,6,2)=34.51, B(2,4,1)=31.68.

B(p,q,r): pW04+qH+rL=W04pHqLr. Also DH by calorimetry. Ligand defined as H4L

\*\*\*\*\*

C6H9NO6 H3L NTA CAS 139-13-9 (191)  
Nitrilotriethanoic acid; N(CH2.C00H)3  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----

W(VI)	sp	NaCl04	25°C	0.10M	C	I			2004MZA (47096)	83
								Keff(W04+2H+L)=19.00		

Data for 0.3-1.0M NaCl04. At I=1.0 M, Keff=19.40.  
-----

W(VI)	sp	NaCl04	25°C	0.5M	C				1976CLa (47097)	84
								K(W04+2H+L=W03L+H2O)=17.75		

Method: stopped flow spectrophotometry  
-----

W(VI)	gl	oth/un	25°C	0.15M	U				1966KRa (47098)	85
								K(W04+L+2H=W03L)=18.86		

-----

W(VI)	nmr	oth/un	35°C	2.00M	U				1966KRa (47099)	86
								K(W04+L+2H=W03L)=19.1		

\*\*\*\*\*

C6H12N2O4 H2L EDDA CAS 5657-17-0 (119)  
1,2-Diaminoethane-N,N'-diethanoic acid; H00C.CH2.NH.CH2.CH2.NH.CH2.C00H  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----

W(VI)	gl	NaCl04	25°C	3.0M	U				1979ZLa (49284)	87
								B(W04+L+2H=W03L)=19.62		

\*\*\*\*\*

C6H12O5                      L      L-Rhamnose                      CAS 634-74-2    (3659)  
6-Deoxy-L-mannose;

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

W(VI)            gl    KCl        25°C 0.10M C    1989VCa (49509)    88

B((W04)2H2L)=17.04

B((W04)2H3L)=20.54

K((W04)2H2L+H)=3.50

\*\*\*\*\*

C6H12O6                      L      D-Mannose                      CAS 3458-28-4    (1562)  
D-Mannose

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

W(VI)            gl    KCl        25°C 0.10M C    1989VCa (49608)    89

B((W04)2H2L)=17.50

M=W04

\*\*\*\*\*

C6H14O6                      L      D-Mannitol                      CAS 69-65-8    (3664)  
D-Mannitol;

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

W(VI)            gl    KNO3      21°C 0.10M C    1978MBc (51090)    90

Medium pH 3-5. K(2W04+2H+L=W205(H-4L)+3H2O)=18.78

\*\*\*\*\*

C6H14O6                      L      Glucitol                      CAS 50-70-4    (2878)  
D-Sorbitol;

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

W(VI)            gl    KCl        RT   0.10M M    I    1990VSc (51110)    91

K(2W04+2H+L=W207L+H2O)=19.15

Data for 0.01-1.0 M KCl and NaCl. In 0.01 M KCl, K=19.30.

-----  
W(VI)            gl    KNO3      21°C 0.10M C    1978MBc (51111)    92

Medium pH 3-5. K(2W04+2H+L=W205(H-4L)+3H2O)=19.26

\*\*\*\*\*

C7H6O3                      H2L                                      CAS 139-85-5    (881)  
3,4-Dihydroxybenzaldehyde, protocatechuic aldehyde; C6H3(OH)2.CHO

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

W(VI)            sp    oth/un    25°C .575M U    1980NKa (54359)    93

K(H2W04L+H2L=W02L2+2H2O)=3.39

K(W04+H2L)=3.10

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----

W(VI) sp oth/un 20°C ? U 1959HAa (54360) 94  
K(WO4+H2L=WO2L2)=7.76

\*\*\*\*\*

C7H6O4 H3L Protocatechuic CAS 99-50-3 (875)  
3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

W(VI) sp oth/un 25°C .575M U 1980NKa (54709) 95  
K(H2WO4L+H2L=WO2L2+2H2O)=3.18  
K(WO4+H2L)=3.23

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
W(VI) sp oth/un 20°C ? U 1959HAa (54710) 96  
K(WO4+2H3L=WO2H2L2)=7.30

\*\*\*\*\*

C7H6O5 H4L Gallic acid CAS 149-91-7 (446)  
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

W(VI) sp oth/un 25°C .575M U 1980NKa (54772) 97  
K(H2WO4L+H2L=WO2L2+2H2O)=3.39  
K(WO4+H2L)=3.34

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

-----  
W(VI) sp oth/un 20°C ? U 1959HAa (54773) 98  
K(WO4+H3L=WO3HL)=3.37

\*\*\*\*\*

C7H7NO2 HL CAS 495-18-1 (184)  
Benzohydroxamic acid; C6H5.CO.NH.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

W(VI) dis KCl ? 4.0M U 1967PNa (55526) 99  
K(WO2+2HL=WO2L2+2H)=7.11

\*\*\*\*\*

C7H8O2 H2L CAS 488-17-5 (1657)  
1,2-Dihydroxy-3-methylbenzene; CH3.C6H3(OH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

W(VI) sp oth/un 25°C .575M U 1980NKa (56057) 100  
K(H2WO4L+H2L=WO2L2+2H2O)=3.60  
K(WO4+H2L)=3.22

Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8

\*\*\*\*\*

C7H8O2 H2L Methylcatechol CAS 452-86-8 (525)  
1,2-Dihydroxy-4-methylbenzene; CH3.C6H3(OH)2

-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKa (56082)	101
								K(H2WO4L+H2L=WO2L2+2H2O)=3.76 K(WO4+H2L)=2.42		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										
W(VI)	sp	KCl	25°C	0.10M	U				1963HAc (56083)	102
								K(WO4+2H2L)=6.31		
*****										
C8H8O3		HL		Mandelic Acid				CAS 611-72-3 (80)		
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	0.10M	C				1995HCa (59888)	103
								Keff(WO4+2L+2H=WO2L2)=18.15		
Medium: 0.1 M acetate buffer, pH 4.7										
*****										
C9H7NO3S2		H2L						CAS 58447-10-2 (4675)		
8-Mercaptoquinoline-5-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	?	?	U				1968ABa (64433)	104
								K(W2O5+L)=10.6 K(W2O5+2L)=19.7		
*****										
C9H7NO4S		H2L		Sulfoxine				CAS 84-88-8 (448)		
8-Hydroxyquinoline-5-sulfonic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KN03	16°C	0.10M	U				1969GTa (64589)	105
								K(WO4+L+2H=WO3L)=19.87		
*****										
C10H8O2		H2L						CAS 92-44-4 (1658)		
2,3-Dihydroxynaphthalene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	.575M	U				1980NKa (69784)	106
								K(H2WO4L+H2L=WO2L2+2H2O)=3.47 K(WO4+H2L)=3.57		
Medium: 0.1 M NH4OH, 0.08 M Na2S2O5. pH 8										
*****										
C10H16N2O8		H4L		EDTA				CAS 60-00-4 (120)		
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo



-----  
W(VI) gl NaClO4 25°C 3.0M U 1979ZLa (74306) 107  
B(WO4+L+2H=WO3L)=19.67  
B(2WO4+L+2H=W2O6L)=36.22  
-----

W(VI) gl NaClO4 25°C 0.10M U K1=9.92 1975PPb (74307) 108  
K(WO3L+H)=7.31  
B((WO3)2L)=18.41  
-----

K1: WO3+L=WO3L  
-----

W(VI) gl oth/un 25°C 0.15M U 1966KRa (74308) 109  
K(WO4+L+2H)=18.9  
K(WO4+WO3L+2H)=16.9  
-----

W(VI) nmr oth/un 35°C 1.0M U 1966KRa (74309) 110  
K(WO4+L+2H)=18.7  
K(WO4+WO3L+2H)=16.7  
K(WO3L+H)=7.5  
-----

\*\*\*\*\*

C10H25N5 L 15-Ane-N5 CAS 295-64-7 (99)  
1,4,7,10,13-Pentaazacyclopentadecane; cyclo(-(HN.CH2.CH2)5-)  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	vlt	NaClO4	25°C	0.20M	C			K(WO4+H3L)=2.11	1999SSe (76739)	111

Method: differentail pulse polarography.  
-----

\*\*\*\*\*

C12H30N6 L CAS 296-35-5 (143)  
1,4,7,10,13,16-Hexaazacyclooctadecane; cyclo(-(NH.CH2.CH2)6-)  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	vlt	NaClO4	25°C	0.20M	C			K(WO4+H3L)=2.21	1999SSe (84361)	112

Method: differentail pulse polarography.  
-----

\*\*\*\*\*

C14H8O7S H3L DASA CAS 83-61-4 (950)  
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	25°C	?	U			B2=7.8	1959DBb (86771)	113

Method: differentail pulse polarography.  
-----

\*\*\*\*\*

C15H11N3O4S H2L (5130)  
7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO3	16°C	0.10M	U				1969GTa (91339)	114

Method: differentail pulse polarography.  
-----

$$B((W04)H2L)=18.34$$

\*\*\*\*\*

C15H11N3O7S2                      H3L                      CAS 17852-90-3 (5131)  
7-(4-Sulfophenylazo)-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO3	16°C	0.10M	U				1969GTa (91352)	115

$$B((W04)H2L)=18.05$$

\*\*\*\*\*

C19H13N3O7S2                      H3L                      SNAZOXS                      CAS 117-87-3 (995)  
8-Hydroxy-7-(4'-sulfo-1'-naphthylazo)-quinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	KNO3	16°C	0.10M	U				1969GTa (99051)	116

$$K(W04+L+2H)=18.00$$

\*\*\*\*\*

C22H20O13                      H5L                      Carminic acid                      CAS 1260-17-9 (714)  
Carminic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	sp	oth/un	22°C	?	U				1966KWb (101707)	117

$$K(W04+H5L=W03H3L)=5.5(?)$$

\*\*\*\*\*

C22H24N2O8                      L                      Deoxycycline                      CAS 564-25-0 (2204)  
Deoxycycline, 6-Deoxy-5-hydroxytetracycline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	none	20°C	0.0	C				1991JMa (101767)	118

$$K(W04+H3L=W03HL)=8.39$$

$$K(W04+2H3L=W03(H2L)2)=8.26$$

\*\*\*\*\*

C22H24N2O8                      H2L                      Tetracycline                      CAS 60-54-8 (2201)  
Tetracycline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
W(VI)	gl	none	20°C	dil	C				1989VJa (101831)	119

$$K(W04+HL)=7.86$$

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#### EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
 I Data with various BACKGROUNDS  
 H Data for THERMOCHEMICAL quantities  
 M Data for TERNARY Complexes

EVALUATION Flags are :-

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END