

START Experiments recorded for  
 from SC-Database on Saturday, 01 January, 2000 at 00:48:50  
 Software version = 5.81 Data version = 4.62  
 Experiment list contains 686 experiments for  
 (no ligands specified)  
 2 metals : Pd(IV), Pd++  
 (no references specified)  
 (no experimental details specified)

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e- HL Electron (442)  
 Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	EMF	none	18°C	0.0	U				1924JIa	(793) 1

K=42.3(1220mV)

K'=32.9(950mV)

K: PdO<sub>3</sub>(s)+2H+2e=PdO<sub>2</sub>(s)+H<sub>2</sub>O. K': PdO<sub>2</sub>(s)+2H+2e=PdO(s)+H<sub>2</sub>O

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Br- HL Bromide CAS 10035-10-6 (19)  
 Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	EMF	NaClO <sub>4</sub>	25°C	0.40M	U				1971DUa	(2221) 2

K5=3.48

K6=2.64

Medium: HClO<sub>4</sub>

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Cl- HL Chloride CAS 7647-01-0 (50)  
 Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	EMF	NaClO <sub>4</sub>	25°C	0.40M	U				1971DUa	(5441) 3

K5K6=4.22

Medium: HClO<sub>4</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	sol	NaCl	25°C	1.0M	U				1930WEa	(5442) 4

K(K<sub>2</sub>PdL<sub>6</sub>(s)=2K+PdL<sub>6</sub>)=-5.22

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e- HL Electron (442)  
 Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	none	25°C	0.00	U				1971JPa	(794) 5

K(Pd + 2e=Pd(s))=30.8(0.91V)

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Pd++ EMF oth/un 25°C 4.00M U T 1970IEa (795) 6  
 $K(\text{Pd} + 2\text{e} = \text{Pd(s)}) = 33.1(979\text{mV})$   
 Medium: HClO<sub>4</sub>. K=34.8(978mV, 10 C), 34.2(979mV, 15 C), 33.5(975mV, 20 C), 32.5(978mV, 30 C), 31.8(972mV, 35 C), 30.9(960mV, 40 C)(m units)

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Pd++ oth none 25°C 0.0 U 1968GHa (796) 7  
 $K' = 16.2(480\text{mV})$   
 Method: Literature evaluated data.  $K'$ :  $\text{Pd(IV)I}_6 + 2\text{e} = \text{PdI}_4 + 2\text{I}$

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Pd++ oth none 25°C 0.0 U 1968GHa (797) 8  
 $K(\text{Pd} + 2\text{e} = \text{Pd(s)}) = 31.1(920\text{mV})$   
 Method: Literature evaluated data

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Pd++ oth none 25°C 0.0 U M 1968GHa (798) 9  
 $K(\text{PdCl}_4 + 2\text{e} = \text{Pd(s)} + 4\text{Cl}) = 19.9$   
 Method: Literature evaluated data.  $K(\text{PdBr}_4 + 2\text{e} = \text{Pd(s)} + 4\text{Br}) = 16.6(0.49\text{V})$ ,  
 $K(\text{PdI}_4 + 2\text{e} = \text{Pd(s)} + 4\text{I}) = 6.1(0.18\text{V})$

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Pd++ EMF NaClO<sub>4</sub> 25°C 4.87M U I 1968LMb (799) 10  
 $K(\text{Pd} + 2\text{e} = \text{Pd(s)}) = 33.67, 996\text{ mV}$   
 Medium: HClO<sub>4</sub>. I=3.46: K=32.19, 952 mV; I=2.22: K=31.25, 924.4 mV;  
 I=1.06: K=31.11, 920 mV

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Pd++ EMF none 25°C 0.0 M H 1967IEa (800) 11  
 $K(\text{Pd} + 2\text{e} = \text{Pd(s)}) = 30.9, 915\text{ mV}$   
 By calorimetry, 0.1 M NaI:  $\text{DH}(\text{Pd} + 3\text{I} = \text{Pd(s)} + \text{I}_3^-) = -104.1\text{ kJ mol}^{-1}$

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Pd++ EMF oth/un 25°C ? U M 1965BKc (801) 12  
 $K = 38.21, 1130\text{ mV (X=Cl)}$   
 $K = 23.40, 692\text{ mV (X=Br)}$   
 $K = 21.13, 625\text{ mV (X=I)}$   
 $K: \text{Pd(en)}_2\text{X}_2 + 2\text{e} = \text{Pd(en)}_2 + 2\text{X}$

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Pd++ oth none 25°C 0.0 U 1952LAB (802) 13  
 $K = 20(600\text{ mV})$   
 $K: \text{Pd(II)Br}_4 + 2\text{e} = \text{Pd(s)} + 4\text{Br}$ . From thermodynamic data  
 $K(\text{PdO}_2(\text{s}) + \text{H}_2\text{O} + 2\text{e} = \text{PdO(s)} + 2\text{OH}) = 25(730\text{ mV})$  estimated

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Pd++ EMF KCl 25°C 1.0M U T 1943TWa (803) 14  
 $K = 21.0(621\text{ mV})$   
 Medium: HCl. K:  $\text{PdCl}_4 + 2\text{e} = \text{Pd(s)} + 4\text{Cl}$ . At 15 C: K=21.8(623 mV), 35 C: 20.3(619 mV). In 4 M HClO<sub>4</sub>:  $K(\text{Pd(II)} + 2\text{e} = \text{Pd(s)}) = 33.4(987\text{ mV})$

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Pd++ EMF NaCl 25°C 1.0M U I 1942GSa (804) 15  
 $K = 44.0(1301\text{ mV})$   
 $K(\text{Pd(IV)Cl}_6 + 2\text{e} = \text{Pd(II)Cl}_4 + 2\text{Cl}$ . In 1 M HCl: K=43.5(1286 mV). In 1 M KBr:  
 $K(\text{PdBr}_6 + 2\text{e} = \text{PdBr}_4 + 2\text{Br}) = 33.6(994\text{ mV})$ . 1 M KI:  $K(\text{PdI}_6 + 2\text{e} = \text{PdI}_4 + 2\text{I}) = 16.3(482\text{ mV})$

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Pd++ oth KCl 25°C 1.0M U 1930WEa (805) 16  
 $K = -3.62$

Medium: HCl. K: Pd(IV)Cl<sub>6</sub>=Pd(II)Cl<sub>4</sub>+Cl<sub>2</sub>(aq). Method:chemical analysis  
 K(Pd(IV)Cl<sub>6</sub>+2e=Pd(II)Cl<sub>4</sub>+2Cl)=43.56(1288 mV) from thermodynamic data  
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Br- HL Bromide CAS 10035-10-6 (19)  
 Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	kin	NaClO <sub>4</sub>	25°C	1.0M	U	M		1973ELa (2222)	17	
								K(cis-trans-PdL <sub>2</sub> (H <sub>2</sub> O) <sub>2</sub> )=0.78 K <sub>2</sub> (cis)=4.19 K <sub>2</sub> (trans)=3.41 K <sub>3</sub> (cis)=3.37		

Medium: HClO<sub>4</sub>. K<sub>3</sub>(trans)=4.15. Kn: PdL<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub>+nL

Pd++	sp	NaClO <sub>4</sub>	?	1.0M	U		K <sub>1</sub> =2.23	1973GSc (2223)	18	
Pd++	sp	NaClO <sub>4</sub>	25°C	1.0M	U		K <sub>1</sub> =5.17 B <sub>2</sub> =9.42 B <sub>3</sub> =12.7 B <sub>4</sub> =14.9	1972ELa (2224)	19	

Medium: HClO<sub>4</sub>

Pd++	sp	NaClO <sub>4</sub>	25°C	4.50M	U	M		1972FKa (2225)	20	
								K(PdCl <sub>4</sub> +L=PdCl <sub>3</sub> L+Cl)=1.40 K(PdCl <sub>3</sub> L+L=PdCl <sub>2</sub> L <sub>2</sub> +Cl)=1.06 K(PdCl <sub>2</sub> L <sub>2</sub> +L=PdClL <sub>3</sub> +Cl)=0.72 K(PdClL <sub>3</sub> +L=PdL <sub>4</sub> +Cl)=0.27		

Medium: LiClO<sub>4</sub>

Pd++	cal	NaClO <sub>4</sub>	25°C	1.0M	U	H		1972RHa (2226)	21	
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Medium: HClO<sub>4</sub>. DH(K<sub>1</sub>)=-21.3 kJ mol<sup>-1</sup>, DS(K<sub>1</sub>)=27.2 J K<sup>-1</sup> mol<sup>-1</sup>

Pd++	ISE	diox/w	25°C	71%	U	TI		1968GFc (2227)	22	
							B <sub>4</sub> =19.0			

B<sub>4</sub>=16.2(0%), also B<sub>4</sub> for several other % dioxan. At 40 C: B<sub>4</sub>=18.1(71%)  
 15.3(0%)

Pd++	sol	NaClO <sub>4</sub>	20°C	0.10M	U		K <sub>1</sub> =6.8	1967GGa (2228)	23	
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Pd++	cal	oth/un	25°C	0.10M	U	H		1967IWa (2229)	24	
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Medium:NaBr. DH(B<sub>4</sub>)=-54.8 kJ mol<sup>-1</sup>

Pd++	gl	NaClO <sub>4</sub>	var	var	U			1967KPc (2230)	25	
							K(PdBr <sub>3</sub> OH+Br=PdBr <sub>4</sub> +OH)=-4.23			

19-50 C, I=0.1-1.0

Pd++	sp	NaCl	25°C	1.0M	U			1966BSa (2231)	26	
							B <sub>4</sub> =13.05			

Pd++	sp	NaClO <sub>4</sub>	45°C	1.80M	U	T H		1966SBb (2232)	27	
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K4=2.50(10 C), 2.30(25 C). DH(K4)=-18.0 kJ mol<sup>-1</sup>, DS=-14.6 J K<sup>-1</sup> mol<sup>-1</sup>

B4=14

B3=11.28

B4=13.42 ?

$$K_{SO} = -12.54$$
$$K(\text{PdL2}(s) = \text{PdL2}) = -4.4$$
$$K_4 = 2.20$$
$$K1=4.37$$

K4=3.50

Medium:0.8(C104),0.6 H+. By hypothesis method:K2=4.08, K3=3.79

B4=16.1

B4=13.10

Method: from thermodynamic data; I=0 corr.

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CN-	HL	Cyanide	CAS 74-90-8 (230)
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Cyanide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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B4=63 (60<B4<65)

$$K(\text{Pd}(\text{CN})_2(\text{s}) + 2\text{CN}) = 20.8$$
$$K_{so}(\text{Pd}(\text{CN})_2) = -42$$

K1=10.5

B4=42.4

 $K_5 = 2.9$ 

Medium: 0 corr. By calorimetry:  $\Delta H(B4) = -385.8 \text{ kJ mol}^{-1}$ ,  $\Delta S = -485 \text{ J K}^{-1} \text{ mol}^{-1}$ ;

$$\text{DH}(\text{K5}) = -0.8, \text{DS} = 33. \text{DH}(\text{PdBr}_4 + 4\text{L} = \text{PdL}_4 + 4\text{Br}) = -329$$

B4=51.6

$$K(\text{Pd}+2\text{e}=\text{Pd}(\text{s}))=33.4$$

Medium: KCN var

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C03-- H2L Carbonate CAS 465-79-6 (268)

Carbonate;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	NaCl	25	0.11M	C	I M			2003CBa (3345)	38
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Data for 0.105-1.0 M NaCl + H3BO3.  $K(\text{PdCl}_4+\text{HCO}_3=\text{Pd}(\text{CO}_3)\text{Cl}_3+\text{H}+\text{Cl})=-6.68$

$K=-6.50$  (I=0.305),  $-6.62$  (I=0.505),  $-6.71$  (I=0.705),  $-6.95$  (I=1.005)

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Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	NaCl	25	0.11M	C	I M			2003CBa (5443)	39
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$$K(\text{PdCl}_3+\text{Cl})=1.08$$

Data for 0.105-1.0 M NaCl, pH 3.0-8.5.  $K(\text{PdCl}_4+\text{H}_2\text{O}=\text{Pd}(\text{OH})\text{Cl}_3+\text{H}+\text{Cl})=-8.72$ .

$K=-8.89$  (I=0.305),  $-8.97$  (I=0.505),  $-8.98$  (I=0.705),  $-8.96$  (I=1.005)

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Pd++	sp	NaCl	25	0.50M	C				2000BYa (5444)	40
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$$*K(\text{PdCl}_4)=-8.98$$

\*K:  $\text{PdCl}_4+\text{H}_2\text{O}=\text{PdCl}_3(\text{OH})+\text{H}$ .

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Pd++	sol	KCl	25	0.10M	C	TI M			1999VWa (5445)	41
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$$\text{B}_4=11.81$$

$$K(\text{Pd}+3\text{Cl}+\text{OH})=20.21$$

Data for 20 C and I=0.8 m and for 25 C and I=0.1-1.0 m. At I=0,  $\text{B}_4=11.29$ ,

$K(\text{Pd}+3\text{Cl}+\text{OH})=20.29$ .

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Pd++	sp	NaNO3	37	0.16M	C	M			1998ESa (5446)	42
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$$K(\text{PdA}(\text{H}_2\text{O})_2+\text{Cl})=3.563$$

$$K(\text{PdA}(\text{H}_2\text{O})\text{Cl}+\text{Cl})=2.28$$

A is 1,3-diaminopropane.

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Pd++	gl	NaClO4	37	0.15M	C	M			1996GTa (5447)	43
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$$K(\text{PdA}(\text{H}_2\text{O})_2+\text{L})=3.65$$

$$K(\text{PdA}(\text{H}_2\text{O})_2+2\text{L})=5.86$$

$$*K(\text{PdA}(\text{H}_2\text{O})_2+\text{L})=-2.68$$

A=diaminosuccinate diethylester,  $\text{EtO}_2\text{CCH}(\text{NH}_2).\text{CH}(\text{NH}_2)\text{CO}_2\text{Et}$

\*K:  $\text{PdA}(\text{H}_2\text{O})_2+\text{L}=\text{PdA}(\text{H}_2\text{O})(\text{OH})\text{L}+\text{H}$

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Pd++	sol	NaCl	100	1.0M	U	T			1995GAa (5448)	44
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$$\text{K}_3=0.30$$

Method: solubility of AgCl in Pt solution, 0.03-3.0 m HCl.

At 200 C,  $\text{K}_3=1.20$ , at 300 C,  $\text{K}_3=1.36$

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Pd++	kin	NaClO4	25	0.10M	U	M			1993SHa (5449)	45
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$$K_{out}(PdACH20+L)=1.86$$

Pd++ sp oth/un 197C var U TI 1991TJa (5450) 46

 $K_4 = 1.25$ 

Pd++	nmr non-aq 24°C 100% U IHM	1982HBa	(5451)	47
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$$K(\text{PdBr}_2 + \text{PdL}_2 = 2\text{PdBrL}) = 0.63$$
$$K(\text{PdCl}_2 + \text{PdL}_2 = 2\text{PdClL}) = 0.61$$

For iodide complex,  $\Delta H = -5.0 \text{ kJ mol}^{-1}$ ,  $\Delta S = 12.6 \text{ J K}^{-1} \text{ mol}^{-1}$

Pd++ oth NaCl04 25°C 0.0 M I K1=5.08 B2= 8.88 1980KRa (5452) 48

 $K_4 = 0.88$ 

Pd++ sp NaCl04 25C 0.86M U K1=4.0 B2=7.2 1976YBa (5453) 49

 $K_3 = 2.3$ 

When  $I=0.1 \text{ M NaClO}_4$ :  $K_1=6.0$ ,  $K_2=4.6$ ,  $K_3=2.5$

Pd++	sp	non-aq	20°C	100%	U	I	1974V0a	(5454)	50
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$$K(Li+Pd_2L_6)=1.7$$

Medium: MeCN, LiCl at different concentrations. With Me<sub>4</sub>NCl, values are: 7.8, 1.5

Pd++	kin NaClO4	25°C	1.0M U	M	1973ELa	(5455)	51
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$$K_2(\text{trans}) = 2.79$$
$$K3(\text{trans}) = 2.90$$

Medium: HClO<sub>4</sub>.  $K(\text{cis-PdL}_2(\text{H}_2\text{O})_2 = \text{trans-PdL}_2(\text{H}_2\text{O})_2) = 0.32$

Pd++	sp	NaClO4	?	1.0M U	1973GSc	(5456)	52
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$$K_4 = 1.27$$

Pd++	sp	non-aq	?	100%	U	M	1973KF	a	(5457)	53
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$$K(\text{PdBr}_3\text{L} + \text{L} = \text{PdBr}_2\text{L}_2 + \text{Br}) = 1.84$$
$$K(\text{PdBrL}_3 + \text{L} = \text{PdL}_4 + \text{Br}) = 2.39$$

Medium: MeCN, 1.5 M Bu<sub>4</sub>N(Cl,Br)

Pd++ sp NaCl04 25°C 1.0M U K1=4.47 B2=7.76 1972ELa (5458) 54

B3=10.2

B4=11.5

Medium: HClO4

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Pd++ cal NaClO4 25°C 1.0M U H 1972RHa (5459) 55

Medium: HClO4. DH(K1)=-12.7 kJ mol<sup>-1</sup>, DS=43.1 J K<sup>-1</sup> mol<sup>-1</sup>; DH(K2)=-10.8, DS=26.8; DH(K3)=-10.7, DS=10.0; DH(K4)=-14.2, DS(K4)=-21.8

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Pd++ oth non-aq 37°C 100% U M 1971HMb (5460) 56

K(Li2Pd2L6+2LiL=2Li2PdL4)=-1.0

Medium: CH3COOH. Method: vapor phase osmometry

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Pd++ vlt NaClO4 25°C 0.20M U 1971JPa (5461) 57

B3=7.94

K4=1.44

Medium: HClO4

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Pd++ EMF oth/un 25°C 3.0M U 1971KMh (5462) 58

K3=1.76

K4=2.35

Medium: H2SO4

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Pd++ sp NaClO4 ? 1.0M U K1=3.48 B2=6.27 1970RGa (5463) 59

K3=2.35

K4=1.1

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Pd++ EMF oth/un ? var U K1=4.7 B2=7.70 1969GKd (5464) 60

K3=2.6

K4=1.6

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Pd++ EMF NaClO4 25°C 1.0M U 1969KSc (5465) 61

B4=12.15

Medium: H(ClO4,S04)

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Pd++ ISE diox/w 25°C 72% U TI K1=17.7 1968GFc (5466) 62

Also B4 for several dioxan percentages. At 40 C: B4=16.6(72% dioxan)

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Pd++ sp NaClO4 25°C 4.0M U TI 1968LEc (5467) 63

K4=2.00

Medium: LiClO4. K4=1.77(I=3),1.59(I=2),1.43(I=1)

At I=2: K4=1.68(15 C),1.59(25 C),1.51(40 C)

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Pd++ ISE NaClO4 25°C 3.40M U I 1968LMb (5468) 64

B4=11.4

Medium: HClO4. By spectrophotometry: K4=1.77(I=3.4), 1.44(I=1.07)

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Pd++ con oth/un 25°C dil U 1967CMb (5469) 65

K(Pd(NH3)2L+L)=2.55 ?

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Pd++ sol NaClO4 25°C 0.10M U K1=5.1 1967GGa (5470) 66

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Pd++      cal NaCl  25°C 0.10M U  H      1967IWa (5471) 67
DH(B4)=-23.0 kJ mol-1
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Pd++      gl  NaClO4 var   var  U      1967KPc (5472) 68
                                           K(PdCl3OH+Cl=PdCl4+OH)=-5.7
-----
Pd++      gl  R4N.X  25°C var   U T      1967RBc (5473) 69
                                           K(Pd(NH3)2L+L)=2.33
                                           K(Pd(NH3)3+L)=3.0
Medium:NH4(NO3). Also other constants and values at 30 C by spec.
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Pd++      sp  oth/un 25°C  1.0M U T H      K1=4.00  B2=7.49  1966SBb (5474) 70
                                           B3=9.73
                                           B4=11.11
Med:1.0(NaClO4),0.8 H+. K4=1.50(10 C), 1.42(25 C), 1.28(45 C). DH(B4)=-11.7
kJ mol-1, DS=-12.1 J K-1 mol-1
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Pd++      ISE KCl   25°C  1.0M U      1965FKa (5475) 71
                                           B4=11.8
                                           K(Pd+2e=Pd(s))=33.4
also B4 values for 10-60C, assuming same K and 1/RTF as for 25C!
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Pd++      sp  oth/un 25°C  0.0  U  I      K1=6.0  B2=10.60  1964BSg (5476) 72
                                           K3=2.5
                                           K4=2.0
                                           B4=15.1
also B4 for I=0.25 to 1.01 M NaClO4
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Pd++      oth oth/un 25°C  1.0M U      K1=3.88  B2=6.94  1964BUa (5477) 73
                                           K3=2.14
                                           K4=1.34
                                           B4=10.42
K1 by solubility, others by EMF, spec,
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Pd++      sp  NaClO4 25°C  0.50M U      1964SBe (5478) 74
                                           K4=1.35
-----
Pd++      ISE oth/un 19°C  var   U      1963GKa (5479) 75
                                           B4=12.2
                                           K(Pd+2e=Pd(s))=33.4
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Pd++      sol none  25°C  0.0  U  M      1962REa (5480) 76
                                           Ks=-3.02
                                           K(trans-Pd(NH3)2L+L)=2.41
I=0 corr. Ks: Pd(NH3)2L2(s)=Pd(NH3)2L2
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Pd++      sp  NaClO4 20°C  0.80M U      K1=4.34  B2=7.88  1961SLc (5481) 77
                                           K3=2.68
                                           K4=1.68

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B4=12.24

-----  
 Pd++ sp none 21°C 0.0 U T H K1=6.2 B2=10.9 1957DBa (5482) 78  
 K3=2.5  
 K4=2.6  
 DH(K1)=-33 kJ mol<sup>-1</sup>, DS=4.2 J K<sup>-1</sup> mol<sup>-1</sup>; DH(K2)=-38, DS=-42; DH(K3)=-33, DS=-59; DH(K4)=-33, DS=-59. 38 C: K1=5.9, K2=4.1, K3=2.2, K4=2.5  
 -----

Pd++ sp none 25°C 0.0 U T H K1=6.1 B2=10.7 1956DRa (5483) 79  
 K3=2.4  
 K4=2.6  
 K5=-2.1  
 K6=-2.1  
 DH(K5)=0, DS=-38 J K<sup>-1</sup> mol<sup>-1</sup>; DH(K6)=0, DS=-38  
 -----

Pd++ ISE NaClO4 25°C 4.0M U 1943TWa (5484) 80

B4=13.22

\*\*\*\*\*

FCIBrI HL (541)  
 Halides, comparative (for book data under ligand 80)  
 -----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ sol NaClO4 20°C 0.10M U M 1967GGd (7412) 81  
 K(Pd+Cl)=5.1  
 K(Pd+Br)=6.8  
 K(Pd+I)=10.0  
 K(Pd+CN)=10.5  
 -----

Pd++ sp oth/un 27°C 0.50M U HM 1967HPb (7413) 82  
 K(PdACl+I=PdAI+Cl)=1.95  
 K(PdABr+I=PdAI+Br)=1.48  
 K(PdABr+SCN)=2.23  
 A=dien. DH(Cl,I)=-15.5 kJ mol<sup>-1</sup>, DS=-14.2 J K<sup>-1</sup> mol<sup>-1</sup>. DH(Br,I)=-10.5, DS=-7;  
 DH(Br,SCN)=-19.6, DS=-22.6. Also other related data  
 -----

Pd++ sp NaClO4 25°C 4.50M U 1967SNa (7414) 83  
 K(PdBr4+I=PdBr3I+Br)=2.75  
 K(PdBr3I+I)=3.00  
 K(PdBr2I2+I)=1.70  
 K(PdBrI3+I=PdI4+Br)=0.80  
 -----

Pd++ sp oth/un 25°C 1.10M U 1966BSd (7415) 84  
 K(PdCl4+2Br=PdCl2Br2+2Cl)=1.99  
 K(PdCl2Br2+2Br=PdBr4+2Cl)=-.06  
 B(PdCl2Br2)=13.11  
 B(PdCl4)=11.12  
 -----

Pd++ sp NaClO4 25°C 4.50M U M 1966SNc (7416) 85  
 K(PdCl4+Br=PdCl3Br+Cl)=1.55

$K(\text{PdCl}_3\text{Br} + \text{Br} = \text{PdCl}_2\text{Br}_2 + \text{Cl}) = 1.09$   
 $K(\text{PdCl}_2\text{Br}_2 + \text{Br} = \text{PdClBr}_3 + \text{Cl}) = 0.95$   
 $K(\text{PdClBr}_3 + \text{Br} = \text{PdBBr}_4 + \text{Cl}) = 0.55$

Medium: LiClO4

\*\*\*\*\*

I- HL Iodide CAS 10034-85-2 (20)  
 Iodide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	1.00M	U		K1=6.08	1986E0a (8319)	86
Pd++	sp	NaClO4	25°C	1.0M	U		K4=2.56 K(2PdL4=Pd2L6+2L)=1.32	1977E0b (8320)	87
Pd++	sp	NaClO4	25°C	4.50M	U	M	K(PdCl4+L=PdCl3L+Cl)=3.95 K(PdCl3L=PdCl2L2+Cl)=4.1 K(PdCl2L2+L=PdClL3+Cl)=2.8 K(PdClL3+L=PdL4+Cl)=1.30	1972SNc (8321)	88

Medium: LiClO4. Data also for complexes with Br in place of Cl: 2.75, 5.75, 7.45, 8.25. Data for L=Br : 1.55, 2.64, 4.14

Pd++	sol	NaClO4	20°C	0.10M	U		K1=10.0	1967GGa (8322)	89
Pd++	ISE	oth/un	25°C	1.0M	U		B4=24	1965FKa (8323)	90
Medium:KI. Also B4 for 10-60C, assuming same K and 1/RTF as for 25C!									
Pd++	sp	NaClO4	20°C	0.80M	U		K1=4.95 K4=2.92 B4=15.74	1965SLd (8324)	91
Pd++	ISE	oth/un	19°C	var	U		B4=24.9	1963GKa (8325)	92
Pd++	sol	oth/un	18°C	var	U		K(PdL2(s)+2I=PdL4)=-2.8	1948Tab (8326)	93

\*\*\*\*\*

NH3 L Ammonia CAS 7664-41-7 (414)  
 Ammonia

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	oth/un	25°C	1.0M	U	H	K1=9.56 B2=18.43 K3=7.52	1991NSb (9192)	94

Medium: H/NH3/NaClO4;

Pd++	gl	KNO3	25°C	?	M	M	K1=6.06	1988SKa (9193)	95
------	----	------	------	---	---	---	---------	----------------	----

K(PdA+L)=5.36

A=diethylenetriamine

Pd++ gl NaClO4 25°C 0.10M C M 1984KMe (9194) 96

K(PdGlyGly+L)=6.50

K(PdPheGly+L)=6.53

Data also for many other amines

Pd++ sp none 25°C 0.0 C 1975PJb (9195) 97

K(Pd(phen)+L)=7.45

K(Pd(phen)L+L)=6.3

Pd++ gl NaClO4 25°C 1.0M U K1=9.6 B2=18.50 1968RJa (9196) 98

K3=7.5

K4=6.8

Pd++ ISE oth/un 25°C 0.50M U 1965FKa (9197) 99

B4=29.6

K(Pd+2e=Pd(s))=33.4

Medium: L. Also B4 for 10-60 C but assuming same RT/F as at 25 C

\*\*\*\*\*

NO2- HL Nitrite CAS 7782-77-6 (635)

Nitrite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ ISE oth/un 25°C var U 1965FKa (9401) 100

B4=21

K(Pd+2e=Pd(s))=33.4

Medium: KL var. B4 values 10-60 C, but RT/F at value for 25 C

\*\*\*\*\*

OH- HL Hydroxide (57)

Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M C M 2002MSb (11918) 101

\*K(PdA(H2O)2)=-5.54

\*K(PdA(OH)H2O)=-15.01

K(2PdA(H2O)2=Pd2A2(OH)2+2H)=-7.90. A is N,N'-dimethylethylenediamine.

Pd++ gl NaClO4 25°C 0.10M C 2001BPd (11919) 102

\*K(Pd(dien)(H2O))=-7.16

K(2Pd(dien)(H2O)=Pd2(dien)2(OH)2)=-10.56.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (11920) 103

\*K(Pd(bpy)(H2O)2)=-3.91

\*K(Pd(bpy)(OH)H2O)=-8.09

K(2Pd(bpy)=Pd2H-2(bpy)2)=-4.70

Pd++ sol NaClO4 25°C 0.50M C TI K1=11.95 B2=23.20 1999VWa (11921) 104  
 At I=1.0, B2=23.4, B3=26.2. At I=0.1, B2=23.8. Data for 25-85 C.

Pd++ sp NaClO4 25°C 1.0M C 1998SEb (11922) 105  
 \*K(Pt(H2O)4)=-3.0

Pd++ gl NaClO4 37°C 0.15M C M 1996GTa (11923) 106  
 \*K(PdA(H2O)2)=-5.25  
 \*K(dimer)=-6.55

A=diaminosuccinate diethylester, EtO2CCH(NH2).CH(NH2)CO2Et

\*K: PdA(H2O)2=PdA(H2O)(OH)L+H, \*K(dimer): 2PdA(H2O)2=(PdA(H2O)2(OH)2)2+2H

Pd++ sol oth/un 25°C var M B2=18.9 1991W0a (11924) 107  
 B3=20.9

Pd++ gl NaCl 25°C 0.50M C I 1984MBa (11925) 108  
 \*K1=-9.23  
 \*B(4,4)=-28.81

Data for 0.5-3.0 M NaCl. At I=1.0 M, \*K1=-9.30, \*B(4,4)=-29.10

Pd++ sol NaClO4 17°C 0.10M U K1=11.72 B2=23.57 1970NKb (11926) 109  
 K3=1.85  
 K4=1.0  
 Kso(Pd(OH)2(s))=-28.96

Pd++ sp none 25°C 0.0 M K1=12.4 B2=26.5 1967IEa (11927) 110  
 By glass electrode: K1=13.0, B2=25.8. By solubility: Ks(PdL2(s)=PdL2)=-2.65

Pd++ sp oth/un 25°C var U 1966WYa (11928) 111  
 \*K1(PdCl2(H2O)2)=-2

Pd++ oth none 25°C 0.0 U 1957ZMa (11929) 112  
 \*Kso(Pd(OH)2)=-2.35  
 \*Kso(PdO)=-3.02

\*Kso: K(Pd(OH)2(s)+2H=Pd2+2H2O); \*Kso(PdO(s)+2H=Pd2+H2O); method:  
 combination of thermodynamic data

Pd++ oth none 25°C 0.0 U 1952LAb (11930) 113  
 Kso(Pd(OH)2)=-31

\*\*\*\*\*

P04--- H3L Phosphate CAS 7664-38-2 (176)  
 Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C ? M M K1=3.10 1988SKa (13301) 114  
 K(PdA+L)=2.63

A=diethylenetriamine

\*\*\*\*\*

S-- H2L Sulfide CAS 7783-06-4 (705)

Sulfide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	sol	oth/un	200°C	var	U	T		1993GBa (14450)	115
------	-----	--------	-------	-----	---	---	--	-----------------	-----

Ks(PdS+H2S)=-7.0  
Ks(PdS+2H2S)=-11.2

Constants at I=0. 30-300 C

Pd++	oth	none	25°C	0.0	C			1989DYa (14451)	116
------	-----	------	------	-----	---	--	--	-----------------	-----

KPd+HS=PdS+H)=43.4

Calculated from literature data, based on K(H+S)=17.0.

Pd++	oth	none	25°C	0	U			1988LIa (14452)	117
------	-----	------	------	---	---	--	--	-----------------	-----

Kso(PdS)=-62.1  
\*Kso(PdS)=-44.8

Derived from thermodynamic data and K(H+S=HS)=17.3.

\*\*\*\*\*

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	sp	NaClO4	30°C	0.10M	U	I	B2=16.2	1973JPa (15222)	118
------	----	--------	------	-------	---	---	---------	-----------------	-----

B4=25.2

Medium: HClO4. At I=0, B2=16.9, B4=25.6

Pd++	sp	oth/un	25°C	1.0M	U	M		1967BSc (15223)	119
------	----	--------	------	------	---	---	--	-----------------	-----

B4=28.67  
K(PdCl4+L=PdCl3L+Cl)=6.03  
K(PdCl3L+L=PdCl2L2+Cl)=4.09  
K(PdCl2L2+L=PdClL3+Cl)=3.59

Medium: 1 NaCl, 0.1 H+. K(PdClL3+L=PdL4+Cl)=3.03

Pd++	sp	oth/un	25°C	1.0M	U			1966BSd (15224)	120
------	----	--------	------	------	---	--	--	-----------------	-----

B(PdBrL3)=25.85  
B(PdBr2L2)=22.25  
B(PdBr3L)=18.15  
B(PdBr4)=13.05

B(PdClL3)=25.19; B4=28.22. Medium: 1 M Na+, 0.1 M H+

Pd++	ISE	oth/un	25°C	var	U			1965FKa (15225)	121
------	-----	--------	------	-----	---	--	--	-----------------	-----

B4=26  
K(Pd+2e=Pd(s))=33.4

Medium: KI. Also B4 values 10 to 60 C - doubtful since RT constant

Pd++	ISE	oth/un	25°C	dil	U	T		1964GPa (15226)	122
------	-----	--------	------	-----	---	---	--	-----------------	-----

B4=19.46

Kso=-17.8. By spectrophotometry: B2=8.4. By solubility, 20 C: K(PdL2(s)+2L)=1.63

-----  
Pd++ sol oth/un 20°C var U 1964GPa (15227) 123  
Ks(PdI2(s)+L=PdL2L)=-0.47  
-----

Pd++ ISE oth/un 19°C var U 1963GKa (15228) 124  
B4=27.6  
K(Pd+2e=Pd(s))=33.4  
\*\*\*\*\*

S03-- H2L Sulfite CAS 7782-99-2 (801)  
Sulfite;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ con oth/un ? var U 1960EAa (15474) 125  
K(PdL2(OH)(H2O)+H)=9  
\*\*\*\*\*

S04-- H2L Sulfate CAS 7664-93-9 (15)  
Sulfate;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp NaClO4 25°C 1.0M C 1998SEb (16469) 126  
K(Pd+S04)=1.28  
K(Pd+HS04)=-0.15  
K(PdS04+H)=-0.40  
-----

Pd++ vlt NaClO4 25°C 0.20M U B2=3.16 1971JPa (16470) 127  
Medium:HClO4  
\*\*\*\*\*

Se-- H2L Selenide (6335)  
Selenide;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth none 25°C 0.0 U 1964BUE (16947) 128  
Kso=-73.4  
\*\*\*\*\*

CH2O2 HL Formic acid CAS 64-18-6 (37)  
Methanoic acid; H.CO0H  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ kin NaClO4 25°C 1.00M U H K1=3.67 1997SEa (17639) 129  
K(Pd+HL=PdL+H)=0.15  
DH(Pd+HL=PdL+H)=-9.7 kJ mol-1, DS(Pd+HL=PdL+H)=-29 J K-1 mol-1  
-----

Pd++ gl KNO3 25°C ? M M K1=2.22 1988SKa (17640) 130  
K(PdA+L)=2.14  
-----

A=diethylenetriamine  
\*\*\*\*\*

CH4N2O                      L      Urea                      CAS 57-13-6    (2018)  
 Carbamide, Urea; (H2N)2CO

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++	nmr	oth/un	40°C	0.90M	U				1998KKf (17723)	131
------	-----	--------	------	-------	---	--	--	--	-----------------	-----

K(Pt(H2O)2en+L)=0.11

Method: 13C nmr. K is for N-bound ligand. For O-bound urea, K=1.36.

Also data for many other alcohol/H2O mixtures.

\*\*\*\*\*

CH4N2S                      L      Thiourea                      CAS 62-56-6    (51)  
 Thiocarbamide, Thiourea; (H2N)2CS

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	oth/un	?	1.0M	U	M			1966SBb (17848)	132
------	----	--------	---	------	---	---	--	--	-----------------	-----

K(PdCl2L2+L=PdClL3+Cl)=4.86

K(PdBr2L2+L=PdBrL3+Br)=4.65

K(PdClL3+L=PdL4+Cl)=4.24

K(PdBrL3+L=PdL4+Br)=4.18

K(Pd(SCN)2L2+L=Pd(SCN)L3+SCN)=2.95; K(Pd(SCN)L3+L=PdL4+SCN)=2.52. I=1 or 0.2

\*\*\*\*\*

CH5N                      L      Methylamine                      CAS 74-89-5    (155)  
 Methylamine; CH3.NH2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	C	M			2002MSb (18023)	133
------	----	-------	------	-------	---	---	--	--	-----------------	-----

K(PdA+L)=7.64

K(PdA+2L)=13.46

K(PdA+B+L)=16.57

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutane dicarboxylic acid.

---

Pd++	gl	NaNO3	25°C	0.10M	U	M			1999SSd (18024)	134
------	----	-------	------	-------	---	---	--	--	-----------------	-----

K(Pd(pn)+L)=6.96

K(Pd(pn)+2L)=13.57

pn is 1,2-diaminopropane. For amine protonation, K1=10.43.

---

Pd++	gl	KNO3	25°C	0.10M	M	M	K1=7.56		1991SKe (18025)	135
------	----	------	------	-------	---	---	---------	--	-----------------	-----

K(Pd(dien)+L)=4.86

Also data for complexes with homologous alkylamines.

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Pd++	gl	NaClO4	21°C	0.10M	C	M			1984KMe (18026)	136
------	----	--------	------	-------	---	---	--	--	-----------------	-----

K(PdGlyGly+L)=7.18

K(PdPheGly+L)=7.31

Data also for many other amines

\*\*\*\*\*

CH6NO3P                      H2L      AMPA                      CAS 1066-51-3    (1981)  
 Aminomethylphosphonic acid; H2N.CH2.PO3H2

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	C			B2=27.51 B(PdH2L2Cl2)=38.76 B(PdHL2Cl2)=35.68 B(PdHLC12)=24.65 B(PdLC12)=21.08	1997BLc (18228)	137
B(PdH-2L)=4.73										

Pd++	gl	KCl	25°C	0.10M	U			K(Pd+L+Cl)=21.52 K(Pd+2L)=27.70 K(Pd+L+H+2Cl)=24.66	1996BRa (18229)	138
------	----	-----	------	-------	---	--	--	---	-----------------	-----

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C2H2O4		H2L		Oxalic acid				CAS 144-62-7 (24)		
Ethanedioic acid; (COOH)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	ix	NaClO4	18°C	0.20M	U			K1=8.72	1972NKb (19031)	139
Medium : HClO4										

Pd++	oth	oth/un	18°C	?	U			K2=3.55	1972NKb (19032)	140
Method : ion-migration										

\*\*\*\*\*

C2H2S4		H2L						CAS 82766-65-2 (2965)		
Tetrathio-oxalic acid; HSSC.CSSH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	?	0.05M	U			B(Pd2L)=8.11	1957JBa (19170)	141

\*\*\*\*\*

C2H3N		L		Cyanomethane				CAS 75-05-8 (1399)		
Acetonitrile; CH3.CN										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		K(Pd3A3CO+L)=-1.0	1994PAa (19192)	142

Medium: MeOH. A=Bis(diphenylphosphino)methane

Pd++	sp	NaClO4	25°C	1.00M	C	T	H	K1=15.5 B2=17.70	1988HEa (19193)	143
Medium: HClO4. DH(K1)=-8.6 kJ mol-1. DS(K1)=-6 J K-1 mol-1. At 5										
C, K1=19.6, K2=1.94; at 15 C, K1=16.7, K2=1.67										

\*\*\*\*\*

C2H3N3S		L						CAS 3179-31-5 (4221)		
1,2,4-Triazoline-3-thione;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ sp KCl ? 1.00M U 1973RRc (19244) 144  
 B4=32.4

Medium: HCl

\*\*\*\*\*

C2H4 L Ethylene CAS 74-85-1 (478)

Ethene; H2C:CH2

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ sol oth/un 25°C 2.00M U M 19720La (19428) 145

K(PdCl3+L)=4.79

K(PdBr3+L)=3.64

K(PdI3+L)=1.68

K(Pd(SCN)3+L)=2.00

Medium : MgSO4 K(Pd(NO2)3+L)=1.34

-----  
 Pd++ sol NaClO4 13°C 2.0M U I 1966PMb (19429) 146

K(PdCl4+L=PdCl3L+Cl)=1.19

K(PdCl3L=PdCl2(H2O)L+Cl)=-1.5

Medium: HClO4. K(PdCl4+L=PdCl2(H2O)L+2Cl)=-0.7. I=3.0: K values: 1.2, -0.7, 0.4. I=4.5(LiClO4+HClO4): K values: 1.21, -0.4, 0.81

\*\*\*\*\*

C2H4O2 HL Acetic acid CAS 64-19-7 (36)

Ethanoic acid; CH3.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ sp NaClO4 25°C 1.00M U K1=4.34 1996SEa (20127) 147

Pd++ sp alc/w 25°C 100% U M 1994PAa (20128) 148

K(Pd3A3CO+L)=2.86

Medium: MeOH. A=Bis(diphenylphosphino)methane

-----  
 Pd++ gl KNO3 25°C ? M M K1=2.73 1988SKa (20129) 149

K(PdA+L)=2.52

A=diethylenetriamine

-----  
 Pd++ sp NaClO4 25°C 0.92M U K1=4.9 B2=8.0 1976YBa (20130) 150

K3=2.6

-----  
 Pd++ sp non-aq 25°C 100% U M 19720Ma (20131) 151

K(PdL2+CeL3=CePdL5)=4.8

Medium: CH3COOH

\*\*\*\*\*

C2H4O2S H2L Thioglycolic CAS 68-11-1 (596)

Mercaptoethanoic acid; HS.CH2.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ gl NaClO4 25°C 1.00M C B2=47.5 2000SAb (20356) 152  
 K(Pd+HL)=22.2  
 K(Pd+2HL)=37.1

\*\*\*\*\*  
 C2H4O3 HL Glycolic acid CAS 79-14-1 (33)  
 2-Hydroxyethanoic acid; HO.CH2.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ sp NaClO4 25°C 1.00M U K1=3.81 1996SEa (20608) 153  
 \*\*\*\*\*  
 C2H5NO L Acetamide CAS 60-35-5 (2886)  
 Ethanoic acid amide; CH3.CO.NH2

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ sp oth/un 25°C .001M U K1=4.46 1958MCa (20673) 154  
 \*\*\*\*\*  
 C2H5NO2 HL Glycine CAS 56-40-6 (85)  
 2-Aminoethanoic acid; H2N.CH2.COOH

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (21677) 155  
 K(Pd(pn)+L)=11.01  
 pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.60, B2=11.93.

-----  
 Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (21678) 156  
 K(PdA+L)=10.76  
 A is 1,3-diaminopropane.

-----  
 Pd++ gl KNO3 25°C 0.50M U 1978LIa (21679) 157  
 K(Pd(en)+L)=11.21

-----  
 Pd++ gl NaClO4 20°C 1.00M C K1=15.25 B2=27.50 1976AMa (21680) 158  
 K(PdL+2Br)=6.47

-----  
 Pd++ gl oth/un 25°C 0.01M U K1=9.12 B2=17.55 1949MMA (21681) 159  
 \*\*\*\*\*  
 C2H6N2O L Glycinamide CAS 598-41-4 (60)  
 2-Aminoethanoic acid amide; H2N.CH2.CO.NH2

-----  
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
 -----

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (21951) 160  
 K(Pd(pn)+L)=8.58  
 K(Pd(pn)+L=PdH-1(pn)L+H)=5.35  
 pn is 1,2-diaminopropane. For amine protonation, K1=7.88.

-----  
 Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (21952) 161

K(PdA+L)=7.41  
K(PdA+L=PdAH-1L+H)=4.20

A is 1,3-diaminopropane.

-----  
Pd++ gl KNO3 25°C 0.10M U M 1977LIb (21953) 162

K(Pd(en)+L)=8.64  
K(Pd(en)L=PdH-1(en)L+H)=-2.47

\*\*\*\*\*

C2H6OS L DMSO CAS 67-68-5 (329)  
Dimethylsulfoxide; (CH3)2.SO

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ sp oth/un 25°C ? U T H K1=0.954 B2=0.56 1987DMa (22117) 163  
DH(K1)=-7.3 kJ mol-1; DS(K1)=-8.4 J K-1 mol-1

-----  
Pd++ sp alc/w 25°C 95% U I 1982CCa (22118) 164  
K(PdCl4+L=PdLC13+Cl)=1.8  
K(PdLC13+L=PdL2Cl2+Cl)=-1.6

Medium: 95% MeOH/H2O

\*\*\*\*\*

C2H7NO L Ethanolamine CAS 141-43-5 (1057)  
2-Aminoethanol; H2N.CH2.CH2.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (22415) 165  
K(Pd(pn)+L)=7.30  
K(Pd(pn)+L=PdH-1(pn)L+H)=1.94  
pn is 1,2-diaminopropane. For amine protonation, K1=9.31.

-----  
Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (22416) 166  
K(PdA+L)=6.81  
K(PdA+L=PdAH-1L+H)=1.85

A is 1,3-diaminopropane.

-----  
Pd++ gl KNO3 25°C 0.10M M M 1991SKe (22417) 167  
K(Pd(dien)+L)=5.29

-----  
Pd++ gl KNO3 25°C 0.10M U M 1981LIb (22418) 168  
K(Pd(H2O)2A+L=PdLA+2H2O)=7.88  
K(Pd(H-1L)A+H)=5.16

A=1,2-diaminoethane

\*\*\*\*\*

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)  
1,2-Diaminoethane; H2N.CH2.CH2.NH2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ gl KNO3 25°C 0.20M C M 2001NSa (23218) 169

\*B2(PdL(H2O)2)=-15.21  
 $K(2PdL(H2O)2=Pd2(OH)L2+H)=-3.04$ ,  $K(2PdL(H2O)2=Pd2(OH)L2+2H)=-8.41$ ,  
 $K(3PdL(H2O)2=Pd3(OH)3L3+3H)=-11.80$

---

Pd++      gl   NaNO3   25°C   0.10M   C      M      2001SHc (23219) 170  
 $K(Pd(bpy)(H2O)2+L)=17.08$   
 $K(Pd(bpy)(H2O)2+H+L)=20.87$

---

Pd++      gl   KNO3   25°C   0.10M   M      M      1991SKe (23220) 171  
 $K(Pd(dien)+L)=6.70$   
 $K(Pd(dien)+H+L)=14.63$

---

Pd++      gl   NaClO4   25°C   1.00M   C      M      K1=23.6   B2=42.20   1986ANa (23221) 172  
Ternary complex with Br-. Combined pot. and spectrophotometric study

---

Pd++      gl   KNO3   23°C   0.20M   U      1976Lma (23222) 173  
 $K(2PdL(OH2)2=LPd(OH)2PdL)=8.3$

\*\*\*\*\*  
C2H8O7P2                      H4L      HEDPA                      CAS 2809-21-4    (436)  
1-Hydroxyethane-1,1-diphosphonic acid; CH3.C(OH)(PO3H2)2

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	U			K1=5.74 K(Pd+HL)=4.44 K(Pd+H2L)=2.41	1980ZRc (23394)	174

\*\*\*\*\*  
C3H4N2                      L      Imidazole                      CAS 288-32-4    (90)  
1,3-Diazole, imidazole; C3H4N2

---

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	C	M		K(PdA+L)=7.92 K(PdA+2L)=14.64 K(PdA+B+L)=14.20	2002MSb (23917)	175

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutane dicarboxylic acid.

---

Pd++      gl   NaNO3   37°C   0.16M   M      M      1998ESa (23918) 176  
 $K(PdA+L)=7.29$   
 $K(PdA+2L)=13.87$

A is 1,3-diaminopropane.

---

Pd++      gl   KNO3   25°C   ?   M      M      K1=6.40      1988SKa (23919) 177  
 $K(PdA+L)=5.62$

A=diethylenetriamine

\*\*\*\*\*  
C3H4O4                      H2L      Malonic acid                      CAS 141-82-2    (79)  
Propanedioic acid; CH2(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	37°C	0.15M	C	M		2003TMb (24533)	178
							K(Pd(en)+L)=5.40		

Pd++	kin	NaClO4	25°C	1.00M	U	H	K1=3.40	1997SEa (24534)	179
							K(Pd+HL=PdL+H)=0.8		
							DH(Pd+HL=PdL+H)=-7.5 kJ mol <sup>-1</sup> , DS(Pd+HL=PdL+H)=-10 J K <sup>-1</sup> mol <sup>-1</sup>		

\*\*\*\*\*

C3H6	L	Propylene	CAS 115-07-1	(702)
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Propene; CH<sub>3</sub>.CH:CH<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sol	oth/un	25°C	2.00M	U	M		19720La (24755)	180
							K(PdCl <sub>3</sub> +L)=4.97		
							K(PdBr <sub>3</sub> +L)=3.72		
							K(Pd(NO <sub>2</sub> ) <sub>3</sub> +L)=1.39		
							K(PdI <sub>3</sub> +L)=1.70		

Medium: MgSO<sub>4</sub>

\*\*\*\*\*

C3H6O2	HL	Propionic acid	CAS 79-09-4	(35)
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Propanoic acid; CH<sub>3</sub>.CH<sub>2</sub>.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	1.00M	U		K1=4.32	1996SEa (25040)	181
Pd++	sp	alc/w	25°C	100%	U	M		1994PAa (25041)	182
							K(Pd3A3CO+L)=3.41		

Medium: MeOH. A=Bis(diphenylphosphino)methane

Pd++	gl	KN03	25°C	?	M	M	K1=2.94	1988SKa (25042)	183
							K(PdA+L)=2.60		

A=diethylenetriamine

\*\*\*\*\*

C3H6O2S	HL		CAS 2444-37-3	(1074)
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(Methylthio)ethanoic acid; CH<sub>3</sub>.S.CH<sub>2</sub>.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	1.00M	C		K1=13.2 B2=22.30	2000SAb (25092)	184
Pd++	kin	oth/un	25°C	1.00M	U			1996SEa (25093)	185
							K1eff=4.08		

Medium: 1.00 M HClO<sub>4</sub>.

\*\*\*\*\*

C3H6O3	HL	L-Lactic acid	CAS 79-33-4	(82)
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L-2-Hydroxypropanoic acid; CH<sub>3</sub>.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	kin	NaClO4	25°C	1.00M	U	H	K1=3.79 K(Pd+HL=PdL+H)=1.42	1997SEa (25515)	186
------	-----	--------	------	-------	---	---	--------------------------------	-----------------	-----

DH(Pd+HL=PdL+H)=-6.6 kJ mol<sup>-1</sup>, DS(Pd+HL=PdL+H)=-18 J K<sup>-1</sup> mol<sup>-1</sup>

Pd++	gl	KNO3	25°C	?	M	M	K1=2.02 K(PdA+L)=1.89	1988SKa (25516)	187
------	----	------	------	---	---	---	--------------------------	-----------------	-----

A=diethylenetriamine

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C3H6O3	HL	Methoxyacetic	CAS 625-45-6	(29)
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Methoxyethanoic acid; CH3.O.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	kin	NaClO4	25°C	1.00M	U	H	K1=3.60 K(Pd+HL=PdL+H)=1.60	1997SEa (25605)	188
------	-----	--------	------	-------	---	---	--------------------------------	-----------------	-----

DH(Pd+HL=PdL+H)=-6.8 kJ mol<sup>-1</sup>, DS(Pd+HL=PdL+H)=-18 J K<sup>-1</sup> mol<sup>-1</sup>

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C3H7NO	HL		CAS 127-06-0	(7906)
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Acetoxime;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	sp	non-aq	40°C	100%	C	I M		2001KKa (25641)	189
------	----	--------	------	------	---	-----	--	-----------------	-----

K(cis-Pd(en)(S)2+L)=2.51  
K(cis-Pd(en)L(S)+L)=1.52  
K(cis-Pd(A)(S)2+L)=1.59  
K(cis-Pd(A)L(S)+L)=0.48

Medium: acetone (S). Also data for D2O/acetone mixtures.

Additional methods: 1H and 13C nmr. A is 3,6-dithia-1,8-octanediol.

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C3H7NO	L	DMF	CAS 68-12-2	(598)
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N,N-Dimethylformamide; HCO.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	sp	alc/w	25°C	100%	U	I M		1994PAa (25666)	190
------	----	-------	------	------	---	-----	--	-----------------	-----

K(Pd3A3CO+L)=-0.27

Medium: MeOH. A=Bis(diphenylphosphino)methane. In toluene, K=-0.15;

in CH3CN, K=-0.35; in acetone, K=-0.62; in CH2Cl2, K=-0.59

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C3H7NO2	HL	Alanine	CAS 56-41-7	(86)
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2-Aminopropanoic acid; H2N.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	gl	NaNO3	25°C	0.10M	U	M		1999SSd (26239)	191
------	----	-------	------	-------	---	---	--	-----------------	-----

K(Pd(pn)+L)=11.42

pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.69, B2=11.88.

-----  
Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (26240) 192  
K(PdA+L)=10.90

A is 1,3-diaminopropane.

-----  
Pd++ gl KNO3 25°C 0.50M U 1978LIa (26241) 193  
K(Pd(en)+L)=11.22

-----  
Pd++ gl KNO3 20°C 0.5M U K1=9.98 B2=18.33 1974KHb (26242) 194  
\*\*\*\*\*  
C3H7NO2 HL B-Alanine CAS 107-95-9 (575)  
3-Aminopropanoic acid; H2N.CH2.CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ gl KNO3 20°C 0.5M U T K1=8.73 B2=15.79 1974KHb (26473) 195  
\*\*\*\*\*  
C3H7NO2 HL Sarcosine CAS 107-97-1 (87)  
N-Methyl-2-aminoethanoic acid; CH3.NH.CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ gl KNO3 25°C 0.50M U 1978LIa (26606) 196  
K(Pd(en)+L)=11.28  
\*\*\*\*\*  
C3H7NO2S H2L Cysteine CAS 52-90-4 (96)  
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ gl NaClO4 25°C 1.00M C B2=51.6 2000SAb (26822) 197  
K(Pd+HL)=27.3  
K(Pd+2HL)=45.0  
\*\*\*\*\*  
C3H7NO3 HL Serine CAS 56-45-1 (49)  
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----  
Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (27167) 198  
K(Pd(pn)+L)=12.00  
K(Pd(pn)+L=PdH-1(pn)L+H)=3.74  
pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.14, B2=11.40.

-----  
Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (27168) 199  
K(PdA+L)=10.19  
K(PdA+L=PdAH-1L+H)=1.90

A is 1,3-diaminopropane.

-----  
Pd++ gl KNO3 25°C 0.10M U M T 1981LIb (27169) 200

$K(\text{PdA}(\text{H}_2\text{O})_2 + \text{L} = \text{PdAL} + 2\text{H}_2\text{O}) = 11.01$

$K(\text{PdA}(\text{H}-1\text{L}) + \text{H}) = 8.51$

A=1,2-diaminoethane

\*\*\*\*\*

C3H8O3S3                      H3L      Unithiol                      CAS 74-61-3    (1271)

2,3-Dimercaptopropanesulfonic acid; HS.CH2.CH(SH).CH2.SO3H

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	EMF	KNO3	?	1.00M	U			B2=21.1	1969SOa (27797)	201
------	-----	------	---	-------	---	--	--	---------	-----------------	-----

Medium: HNO3

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C3H9N2O4P                      H2L                      CAS 30211-73-5    (7117)

Glycylaminomethylphosphonic acid;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.10M	C			B2=26.27	1997BLc (27968)	202
------	----	------	------	-------	---	--	--	----------	-----------------	-----

B(PdH-2L2)=10.99

B(PdLC1)=20.54

B(PdH-1LC1)=16.74

B(PdH-2L)=8.67

B(PdH-3L)=-1.51

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Pd++	gl	KCl	25°C	0.10M	U				1996BRa (27969)	203
------	----	-----	------	-------	---	--	--	--	-----------------	-----

$K(\text{Pd} + \text{L} + 2\text{Cl} + \text{H}) = 24.48$

$K(\text{Pd} + 2\text{L}) = 27.50$

$K(\text{Pd} + \text{L} + \text{Cl}) = 21.35$

\*\*\*\*\*

C3H10N2                      L                      CAS 78-90-0    (2905)

1,2-Diaminopropane; CH3.CH(NH2).CH2.NH2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++	gl	NaNO3	25°C	0.10M	U				1999SSd (28170)	204
------	----	-------	------	-------	---	--	--	--	-----------------	-----

\*K(PdL)=-5.62

\*K(Pd(OH)L)=-9.35

\*\*\*\*\*

C3H10N2                      L      Propanediamine      CAS 109-76-2    (123)

1,3-Diaminopropane; H2N.CH2.CH2.CH2.NH2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	37°C	0.16M	M				1998ESa (28319)	205
------	----	-------	------	-------	---	--	--	--	-----------------	-----

\*K(PdL(H2O)2)=-5.45

\*B2(PdL(H2O)2)=-14.58

\*\*\*\*\*

C3H11N3                      L                      CAS 21292-99-6    (2975)

Propane-1,2,3-triamine; H2N.CH2.CH(NH2).CH2.NH2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	C	M		*K(PdL)=-5.30 *B2(PdL)=-16.67 K(PdL+Gly)=6.16 K(PdL+Ala)=6.38 K(Gly+H)=9.45; K(Ala+H)=9.59. Also K(PdL+Pro)=6.58, K(PdL+methionine)=6.09, K(PdL+imidazole)=4.22, K(PdL+inosine)=3.96, K(PdL+guanosine)=4.43.	1996SEb (28489)	206
Pd++	gl	NaNO3	25°C	0.10M	C	M		K(PdL+cysteine)=5.79 K(PdL+H+cysteine)=14.45 K(PdL+penicillamine)=6.05 K(PdL+H+penicillamine)=14.64 k(PdL+His)=8.52, K(PdL+H+His)=15.63, K(PdL+histamine)=8.85, K(PdL+H+hist- amine)=15.34. Also K(PdL+A=PdL(H-1A)) for A=Gly-Val (-4.43), Leu-Ala (-3.7) ***** C4H3N2O2F HL 5-Fluorouracil CAS 51-21-8 (4277) 5-Fluoro-2,4(1H,3H)-pyrimidinedione;	1996SEb (28490)	207
Pd++	ISE	KNO3	20°C	0.10M	U	M		B(PdCl2L2)=21.7 K(PdCl2L+L)=7.82 ***** C4H4N2O2 HL Uracil CAS 66-22-8 (412) 2,4-Dihydroxypyrimidone, 2,4-Pyrimidinedione;	1969GKc (28694)	208
Pd++	gl	NaNO3	25°C	0.10M	C	M		K(PdA+L)=8.35 K(PdA+2L)=14.88 K(PdA+B+L)=16.18 A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutanedicarboxylic acid.	2002MSb (28863)	209
Pd++	gl	NaNO3	25°C	0.10M	C	M		K(Pd(bpy)(H2O)2+L)=10.96 K(Pd(bpy)(H2O)2+H+L)=13.50 K(Pd(bpy)(H2O)2+2L)=17.17 K(Pd(bpy)(H2O)2+2L+H)=22.15	2001SHc (28864)	210
Pd++	gl	NaNO3	25°C	0.10M	U	M		K(Pd(pn)+L)=8.74 K(Pd(pn)+2L)=15.43 pn is 1,2-diaminopropane. For nucleotide protonation, K1=9.13.	1999SSd (28865)	211
Pd++	gl	NaNO3	37°C	0.16M	M	M			1998ESa (28866)	212

A is 1,3-diaminopropane.

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*****
C4H6N2S          HL   Methimazole      CAS 60-56-0  (1824)
N-Methyl-2-mercaptoimidazole; C3H2N2(CH3).SH
*****
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C4H6O4	H2L	Succinic acid	CAS 110-15-6	(112)
1,4-Butanedioic acid; HOOC.CH2.CH2.COOH				

$$DH(Pd+HL=PdL+H)=10 \text{ kJ mol}^{-1}, DS(Pd+HL=PdL+H)=36 \text{ J K}^{-1} \text{ mol}^{-1}$$

\*\*\*\*\*  
C4H6O4                    H2L      Me-Malonic Acid    CAS 516-15-2    (816)  
Methylpropanedioic acid; HOOC.CH(CH3).COOH

\*\*\*\*\*  
C4H6O5                    H2L        Malic acid                    CAS 617-48-1    (393)  
2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; H00C.CH2.CH(OH).COOH

$$\Delta H(\text{Pd} + \text{HL} = \text{PdL} + \text{H}) = -3.5 \text{ kJ mol}^{-1}, \Delta S(\text{Pd} + \text{HL} = \text{PdL} + \text{H}) = -1 \text{ J K}^{-1} \text{ mol}^{-1}$$

\*\*\*\*\*  
C4H6O5                    H2L      Diglycolic acid    CAS 110-99-6    (243)  
Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOC.CH2.O.CH2.COOH

$$\Delta H(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-12 \text{ kJ mol}^{-1}, \Delta S(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-27 \text{ J K}^{-1} \text{ mol}^{-1}$$

\*\*\*\*\*

C4H7NO4                      H2L      Aspartic acid      CAS 56-84-8    (21)  
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaCl	37°C	0.10M	C				2003GZa (31919)	219
								B(Pd(bpy)L)=13.91		
Pd++	gl	none	25°C	0.0	U				1979FWa (31920)	220
								K(PdL2+H)=4.52 K(PdHL2+H)=3.68 K(PdCl4+2HL=PdH2L2+4Cl)=11.3		
Pd++	gl	NaClO4	25°C	0.10M	U			K1=10.44    B2=18.14	1972SSe (31921)	221
Pd++	gl	KNO3	30°C	0.10M	U			K1=10.55    B2=18.25	1971STc (31922)	222
Pd++	oth	KNO3	30°C	0.13M	U				1971TKe (31923)	223
								K(Pd+H2L=PdHL+H)=10.45 K(PdHL+H2L=Pd(HL)2+H)=7.76		

\*\*\*\*\*

C4H7NO4                      H2L      IDA                      CAS 142-73-4    (118)  
Iminodiethanoic acid; HN(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	20°C	1.00M	C	M		K1=17.5    B2=26.80	1976AMa (32336)	224
								K(Pd+HL)=9.0 K(PdL+2Br)=3.83		
Pd++	gl	KCl	25°C	0.1M	U			K1=9.62    B2=14.87	1975CGc (32337)	225
Pd++	EMF	KCl	25°C	0.10M	U			K1=9.62    B2=14.87	1975VCa (32338)	226

\*\*\*\*\*

C4H8                              L      But-1-ene                      CAS 106-98-9    (2985)  
But-1-ene; CH2:CH.CH2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sol	NaClO4	15°C	5.0M	U	I M			1966PMb (32458)	227
								K(PdCl4+L=PdCl3L+Cl)=1.05		
Medium: LiClO4+HClO4. K=1.14(I=2), 1.13(I=3), 1.13(I=4) K(PdCl4+L=PdCl2(H2O)L+2Cl)=-0.5(I=2), 0.1(I=3), 0.65(I=4), 0.95(I=5)										

\*\*\*\*\*

C4H8N2O2                      H2L      Dimethylglyoxim      CAS 95-45-4    (2032)  
2,3-Butanedione dioxime, Dimethylglyoxime; CH3.(C:NOH).(C:NOH).CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++ dis NaClO4 25°C 1.0M U B2=34.1 1963BDa (32547) 228  
K(PdL2+OH)=5.50

Pd++ sol oth/un 25°C ? U 1958BBb (32548) 229  
Ks2=-3.30

\*\*\*\*\*

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)  
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (32720) 230  
K(Pd(pn)+L)=12.79  
K(Pd(pn)+L=PdH-1(pn)L+H)=6.38

pn is 1,2-diaminopropane. For amide protonation, K1=8.55.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (32721) 231  
K(PdA+L)=10.19  
K(PdAH-1L+H)=3.33

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.50M U M 1977LIa (32722) 232  
K(Pd(en)+L)=10.46  
K(Pd(en)H-1L+H)=6.46

Pd++ gl NaClO4 25°C 3.00M C 1974Gwa (32723) 233  
B(PdHL)=12.11  
B(PdH-1L)=9.1  
B(PdHLC1)=18.29  
B(PdH-1LC1)=17.0

\*\*\*\*\*

C4H8N2O3 HL Gly-Gly CAS 556-50-3 (54)  
Glycyl-glycine; H2N.CH2.CO.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C 1999AJa (33047) 234  
B(PdLC1)=18.08  
B(PdH-1LC1)=15.56  
B(PdH-2L)=4.89  
B(PdH-1L2)=19.30

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-2L2)=13.90; B(PdH-1L)=13.57.

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (33048) 235  
K(Pd(pn)+L)=9.41  
K(Pd(pn)+L=PdH-1(pn)L+H)=6.02

pn is 1,2-diaminopropane. For aminoacid protonation, K1=7.97.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (33049) 236  
K(PdA+L)=7.53

$$K(\text{PdA}+\text{L}=\text{PdAH}-1\text{L}+\text{H})=2.88$$

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.10M U M 1977LIb (33050) 237

$$K(\text{Pd}(\text{en})+\text{L})=9.60$$

$$K(\text{Pd}(\text{en})\text{L}=\text{PdH}-1(\text{en})\text{L}+\text{H})=-3.76$$

\*\*\*\*\*

C4H8OS L (1882)

Tetramethylenesulfoxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 95% U 1982CCa (33192) 238

$$K(\text{PdCl}_4+\text{L}=\text{PdLCl}_3+\text{Cl})=1.7$$

\*\*\*\*\*

C4H8O2 HL CAS 107-92-6 (1118)

n-Butanoic acid; CH3.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=4.38 1997SEa (33346) 239

$$K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-0.29$$

$$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-14.6 \text{ kJ mol}^{-1}, \text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-55 \text{ J K}^{-1} \text{ mol}^{-1}$$

\*\*\*\*\*

C4H8O3 HL CAS 594-61-6 (81)

2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=4.05 1997SEa (33504) 240

$$K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=1.88$$

$$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-8.1 \text{ kJ mol}^{-1}, \text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-21 \text{ J K}^{-1} \text{ mol}^{-1}$$

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C4H9NO L CAS 127-19-5 (477)

N,N-Dimethylacetamide; CH3.CO.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 100% U M 1994PAa (33767) 241

$$K(\text{Pd}_3\text{A}_3\text{CO}+\text{L})=-1.15$$

Medium: MeOH. A=Bis(diphenylphosphino)methane

\*\*\*\*\*

C4H9NO2 L CAS 623-33-6 (3011)

Glycine ethyl ester; H2N.CH2.CO.OCH2CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M M M K1=6.01 1987SKb (34002) 242

$$K(\text{Pd}(\text{dien})+\text{L})=2.81$$

-----

Pd++ gl KNO3 25°C 0.50M U 1983LIb (34003) 243  
 $K(\text{Pd}(\text{en})+\text{L})=7.12$   
 \*\*\*\*\*  
 C4H9NO2 HL Dimethylglycine CAS 1118-68-9 (88)  
 N,N-Dimethyl-2-aminoethanoic acid; (CH3)2N.CH2.COOH  
 -----  

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo																																																																																																																																																																																			
Pd++	gl	KNO3	25°C	0.50M	U				1978LIa (34031)	244																																																																																																																																																																																			
$K(\text{Pd}(\text{en})+\text{L})=11.02$ ***** C4H9NO2S HL Methylcysteine CAS 1187-84-4 (84) 2-Amino-3-methylmercaptopropanoic acid; H2N.CH(CH2.S.CH3)COOH ----- <table border="1"> <thead> <tr> <th>Metal</th> <th>Mtd</th> <th>Medium</th> <th>Temp</th> <th>Conc</th> <th>Cal</th> <th>Flags</th> <th>Lg</th> <th>K values</th> <th>Reference</th> <th>ExptNo</th> </tr> </thead> <tbody> <tr> <td>Pd++</td> <td>gl</td> <td>NaClO4</td> <td>25°C</td> <td>0.10M</td> <td>M</td> <td></td> <td></td> <td></td> <td>2002BSa (34101)</td> <td>245</td> </tr> <tr> <td colspan="9"> <math>*K(\text{PdL})=-4.13</math>  <math>K(2\text{PdL}=\text{Pd}2\text{H}-1\text{L}2)=-0.01</math>  <math>*B2(\text{PdL})=-15.77</math>            -----  <table border="1"> <thead> <tr> <th>Metal</th> <th>Mtd</th> <th>Medium</th> <th>Temp</th> <th>Conc</th> <th>Cal</th> <th>Flags</th> <th>Lg</th> <th>K values</th> <th>Reference</th> <th>ExptNo</th> </tr> </thead> <tbody> <tr> <td>Pd++</td> <td>gl</td> <td>NaClO4</td> <td>25°C</td> <td>1.00M</td> <td>C</td> <td></td> <td></td> <td>K1=19.9 B2=36.30</td> <td>2000SAb (34102)</td> <td>246</td> </tr> <tr> <td>Pd++</td> <td>gl</td> <td>NaNO3</td> <td>25°C</td> <td>0.10M</td> <td>U</td> <td>M</td> <td></td> <td></td> <td>1999SSd (34103)</td> <td>247</td> </tr> <tr> <td colspan="11"> <math>K(\text{Pd}(\text{pn})+\text{L})=10.83</math>            pn is 1,2-diaminopropane. 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Pd++ gl KNO3 25°C 0.10M U M 1981LIb (34357) 251  
K(PdA(H2O)2+L=PdAL+2H2O)=11.09  
K(PdA(H-1L)+H)=9.60

A=1,2-diaminoethane

\*\*\*\*\*

C4H9NS L CAS 123-90-0 (3777)  
Thiomorpholine, tetrahydro-4H-1,4-thiazine, thiomorpholine;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ kin oth/un 25°C 1.00M U 1996SEa (34405) 252  
K1eff=4.30

Medium: 1.00 M HClO4.

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C4H10N2 L CAS 56123-06-9 (8023)  
1,3-Diamino-2-methylenepropene;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.50M U K1=13.64 B2=25.27 1975HSb (34490) 253

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C4H10OS L CAS 110-77-0 (3516)  
Ethyl-2-hydroxyethyl sulfide, 2-(ethylthio)ethanol; CH3CH2SCH2CH2OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ kin oth/un 25°C 1.00M U 1996SEa (34661) 254  
K1eff=4.45

Medium: 1.00 M HClO4.

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C4H10O2S L CAS 111-48-8 (4275)  
3-Thiapentane-1,5-diol; HO.CH2.CH2.S.CH2.CH2.OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ kin oth/un 25°C 1.00M U 1996SEa (34687) 255  
K1eff=4.34

Medium: 1.00 M HClO4.

\*\*\*\*\*

C4H11N L Diethylamine CAS 109-89-7 (1331)  
Diethylamine, 3-azapentane; (C2H5)2NH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp non-aq 25°C 100% U 1994CAa (34820) 256  
K(PdAB+2L=PdAL2+B)=3.14

A:C3H5 (n(3)-allyl); B:N,N'-di(4-methoxyphenyl)-1,2-diaminoethane.

Additional data for other allyl and amino derivatives.

\*\*\*\*\*

C<sub>4</sub>H<sub>12</sub>N<sub>2</sub> L Dimeen CAS 110-70-3 (125)  
 N,N'-Dimethyl-1,2-diaminoethane; CH<sub>3</sub>.NH.CH<sub>2</sub>.CH<sub>2</sub>.NH.CH<sub>3</sub>

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C 2001MSb (35422) 257  
 \*K(PdL(H<sub>2</sub>O)<sub>2</sub>)=-5.54  
 \*K(PdL(OH)H<sub>2</sub>O)=-9.47  
 K(2PdL=Pd<sub>2</sub>H-2L<sub>2</sub>+2H)=-7.90

---

Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C M 2001MSb (35423) 258  
 K(PdL+gly)=11.79  
 K(PdL+ala)=10.89  
 K(PdL+pro)=11.14  
 K(PdL+val)=11.59

Also data for phe, met, imidazole, ser, his, histamine, orn, lys, asp and glu. Amino acid protonation constants also reported.

---

Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C M 2001MSb (35424) 259  
 K(PdL+A)=16.31  
 K(PdL+B)=15.12  
 K(PdL+C)=16.31  
 K(PdL+D)=7.64

K(PdL+E)=6.46. HA=mercaptoethylamine, H<sub>3</sub>B=glutathione, H<sub>2</sub>C=cysteine, D=methylamine, E=ethanolamine. Protonation constants also reported.

---

Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C M 2001MSb (35425) 260  
 K(PdL+A)=6.38  
 K(PdL+B)=6.28  
 K(PdL+C)=4.35  
 K(PdL+D)=4.09

Acids: H<sub>2</sub>A=oxalic, H<sub>2</sub>B=malonic, H<sub>2</sub>C=succinic, H<sub>2</sub>D=adipic. Also data for 1,1-cyclobutane dicarboxylic & fumaric. Protonation constants reported.

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Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C M 2001MSb (35426) 261  
 K(PdL+A)=8.70  
 K(PdL+B)=8.35  
 K(PdL+C)=8.56  
 K(PdL+D)=8.75

HA=uridine, HB=uracil, HC=thymine, HD=thymidine. Also data for inosine, IMP and adenine. Protonation constants are reported.

---

Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C M 2001MSb (35427) 262  
 K(PdL+A)=7.40  
 K(PdL+B)=10.73  
 K(PdL+C)=12.31

A=glycinamide, HB=glutamine, HC=asparagine. Protonation constants are reported.

---

Pd++ gl NaNO<sub>3</sub> 25°C 0.10M C M 2001MSb (35428) 263



K(PdL+A)=7.75

K(PdL+B)=7.63

K(PdL+C)=8.36

HA=glycylglycine, HB=glycylalanine, HC=glycylleucine.

Protonation constants are reported.

\*\*\*\*\*

C4H13N3                      L      Dien                      CAS 111-40-0 (584)  
1,4,7-Triazaheptane, 2,2'Iminobis(ethylamine), diethylenetriamine;  
NH2.(CH2)2.NH.(CH2)2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	?	M	M		1988SKa (35805)	264
							B(PdH-1L)=-8.68 B(PdL(Butyrate))=2.58 B(PdL(pyridine))=4.04		

Pd++	gl	NaClO4	25°C	1.00M	C	M	K1=32.6    B2=40.40	1986ANa (35806)	265
Ternary complexes with Cl- and Br-. A combined pH-metric and spec. study.									

Pd++	gl	NaClO4	25°C	0.50M	C	I		1981GMf (35807)	266
							*K(PdL)=-7.589 K(PdL+PdLOH=Pd2L2OH)=2.19		

In 0.5 NaNO3: \*K(PdL)=-7.543, K(PdL+PdLOH)=2.10

Pd++	gl	NaNO3	25°C	1.00M	U	M	K1=34	1969RJa (35808)	267
							K(PdL+NH3)=6.9		

\*\*\*\*\*

C5H5N                      L      Pyridine                      CAS 110-86-1 (31)  
Pyridine, Azine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	non-aq	25°C	100%	U			1994CVa (36663)	268
							K(PdAB+2L=PdAL2+B)=1.09 K(PdAC+2L=PdAL2+C)=-1.39		
Medium: CHCl3. A:n3-allyl; B:4-MeOC6H4.N=CH.CH=N.C6H4OMe; C:4-MeOC6H4.N=C(Me).C(Me)=N.C6H4OMe. Also data for 4-substituted pyridines.									

Pd++	sp	NaClO4	25°C	1.0M	U	I M	K1=8.4    B2=16.10	1986AHb (36664)	269
							K3=6.6 K4=5.9		

Ternary complexes with 2,2'-bipyridine and 1,10-phenanthroline.

In 0.1M NaCH3SO3, K1=8.5, K2=7.5, K3=6.3, K4=5.7.

Pd++	sp	NaClO4	25°C	1.0M	U	M		1984ETa (36665)	270
							K(Pd(en)Cl2+L=PdenLCl+Cl)=4.31 K(Pd(en)LCl+L=PdenL2+Cl)=3.15		

Pd++	sp	none	25°C	0.0	C			1975PJb (36666)	271
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K(Pd(phen)+L)=7.02

K(Pd(phen)L+L)=6.18

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C5H5N5 L Adenine CAS 73-24-5 (237)

6-Aminopurine; H2N.C5H3N4

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (36975) 272

K(Pd(bpy)(H2O)2+L)=11.95

K(Pd(bpy)(H2O)2+H+L)=15.97

K(Pd(bpy)(H2O)2+2L)=16.59

K(Pd(bpy)(H2O)2+2L+H)=25.76

K(Pd(bpy)(H2O)2+2L+2H)=30.25.

-----  
Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (36976) 273

K(Pd(pn)+L)=11.14

pn is 1,2-diaminopropane. For nucleotide protonation, K1=9.59, B2=13.77.

-----  
Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (36977) 274

K(PdA+L)=10.83

K(PdA+2L)=14.62

A is 1,3-diaminopropane.

\*\*\*\*\*

C5H6N2O2 HL 1-Methyluracil CAS 615-77-0 (7923)

1-Methyl-2,4(1H,3H)-pyrimidinedione;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C M 2003NFa (37222) 275

K(PdA+L)=9.26

A is bis-((2-pyridyl)methyl)amine

-----  
Pd++ gl KNO3 25°C 0.20M C M 2001NSa (37223) 276

K(Pd(en)+L)=9.07

K(Pd(en)+2L)=14.88

\*K(Pd(en)(H2O)L)=-8.53

K(Pd(en)(H2O)+L=Pd(en)(OH)L+H)=0.54,

K(2Pd(en)(H2O)2+2L=Pd2(en)2(OH)L2+H)=12.58

-----  
Pd++ gl KNO3 25°C 0.20M C M 2001NSa (37224) 277

K(Pd(pic)+L)=9.57

K(Pd(pic)+2L)=15.73

\*K(Pd(pic)(H2O)L)=-7.73

K(Pd(pic)(H2O)+L=Pd(pic)(OH)L+H)=1.84,

K(2Pd(pic)(H2O)2+2L=Pd2(pic)2(OH)L2+H)=14.58. Hpic=picric acid.

-----  
Pd++ gl KNO3 25°C 0.20M C K1=7.51 2000NFa (37225) 278

\*\*\*\*\*

C5H6N2O2 HL Thymine CAS 65-71-4 (413)

2,4-Dihydroxy-5-methylpyrimidine; C<sub>4</sub>H<sub>5</sub>N<sub>2</sub>(CH<sub>3</sub>)(OH)<sub>2</sub>

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO <sub>3</sub>	25°C	0.10M	C	M			2002MSb (37282)	279
------	----	-------------------	------	-------	---	---	--	--	-----------------	-----

K(PdA+L)=8.56  
K(PdA+2L)=15.14  
K(PdA+B+L)=15.71

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutanedicarboxylic acid.

---

Pd++	gl	NaNO <sub>3</sub>	25°C	0.10M	U	M			1999SSd (37283)	280
------	----	-------------------	------	-------	---	---	--	--	-----------------	-----

K(Pd(pn)+L)=8.90  
K(Pd(pn)+2L)=15.80

pn is 1,2-diaminopropane. For nucleotide protonation, K<sub>1</sub>=9.59.

---

Pd++	gl	NaNO <sub>3</sub>	37°C	0.16M	M	M			1998ESa (37284)	281
------	----	-------------------	------	-------	---	---	--	--	-----------------	-----

K(PdA+L)=8.37  
K(PdA+2L)=14.60

A is 1,3-diaminopropane.

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Pd++	gl	KNO <sub>3</sub>	25°C	0.20M	C				1997WKa (37285)	282
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K(PdACl+L=PdAL+Cl)=6.97

PdA is [PdH-1(gly-met)].

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C5H7N3O	HL	1-MeCytosine	CAS	1122-47-0	(2268)
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1-Methyl-4-aminopyrimidin-2-one;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO <sub>3</sub>	25°C	0.20M	C	M			2003NFa (37588)	283
------	----	------------------	------	-------	---	---	--	--	-----------------	-----

K(PdA+L)=5.84  
K(2PdA+L=Pd2A2H-1L+H)=1.76

A is bis-((2-pyridyl)methyl)amine

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Pd++	gl	KNO <sub>3</sub>	25°C	0.20M	C	M			2001NSa (37589)	284
------	----	------------------	------	-------	---	---	--	--	-----------------	-----

K(Pd(en)+L)=6.13  
K(Pd(en)+2L)=11.44  
\*K(Pd(en)(H<sub>2</sub>O)L)=-5.69

K(Pd(en)(H<sub>2</sub>O)+L=Pd(en)(OH)L+H)=0.44,  
K(2Pd(en)(H<sub>2</sub>O)+2L=Pd<sub>2</sub>(en)<sub>2</sub>(OH)<sub>2</sub>L+H)=10.41

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Pd++	gl	KNO <sub>3</sub>	25°C	0.20M	C	M			2001NSa (37590)	285
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K(Pd(pic)+L)=8.07  
K(Pd(pic)+2L)=13.35  
\*K(Pd(pic)(H<sub>2</sub>O)L)=-5.22

K(Pd(pic)(H<sub>2</sub>O)+L=Pd(pic)(OH)L+H)=2.85,  
K(2Pd(pic)(H<sub>2</sub>O)+2L=Pd<sub>2</sub>(pic)<sub>2</sub>(OH)<sub>2</sub>L+H)=14.06. Hpic=picric acid.

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Pd++	cal	KNO <sub>3</sub>	25°C	0.20M	C	HM			2000NFa (37591)	286
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DH(Pd(dien)+L)=-38.5 kJ mol<sup>-1</sup>; DH(Pd(gly-ala)+L)=-33.1,

DH(Pd(gly-met)+L)=-32.8.

Pd++ gl KNO3 25°C 0.20M C 1997WKa (37592) 287  
K(PdACl+L=PdAL+Cl)=5.04

PdA is [PdH-1(gly-Met)].

\*\*\*\*\*

C5H8N2O5 H2L (6682)  
5,5-Dimethyl-2-thioxoimidazolidin-4-one;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaCl 25°C 0.10M C B2=22.96 1993CCa (37688) 288  
B(PdH2L2)=44.10  
B(PdHL2)=34.58  
B(Pd2HL2)=45.46  
B(Pd2L2)=38.31

B(Pd2H-1L2)=28.34

\*\*\*\*\*

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)  
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl oth/un 20°C 0.0 U T H K1=16.7 1957DBa (38056) 289  
DH(K1)=-75 kJ mol-1, DS=63. 30 C: K1=16.2, K2=10.9; 40 C: K1=15.4, K2=10.5

Pd++ gl diox/w 25°C 50% U K1=8.71 B2=16.84 1949MMa (38057) 290

\*\*\*\*\*

C5H8O4 H2L CAS 595-46-0 (1144)  
Dimethylmalonic acid; HOOC.C(CH3)2.COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Pd++ gl NaClO4 37°C 0.15M C M 2003TMb (38216) 291  
K(Pd(en)+L)=5.22

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C5H9NO2 HL Proline CAS 147-85-3 (44)  
Pyrrolidine-2-carboxylic acid; C4H8N.COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (38634) 292  
K(Pd(pn)+L)=11.55

pn is 1,2-diaminopropane. For aminoacid protonation, K1=10.52, B2=12.03.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (38635) 293  
K(PdA+L)=10.48

A is 1,3-diaminopropane.

-----  
Pd++ gl KNO3 25°C 0.50M U 1978LIa (38636) 294

$$K(\text{Pd}(\text{en})+\text{L})=12.16$$

-----  
Pd++ gl KNO3 20°C 0.5M U K1=10.26 B2=19.10 1974KHb (38637) 295  
\*\*\*\*\*

C5H9NO3 HL Hydroxyproline CAS 51-35-4 (416)  
4-Hydroxy-2-pyrrolidinecarboxylic acid; C4H7N(OH)(COOH)

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.10M U M 1981LIb (38746) 296  
K(PdA(H2O)2+L=PdAL+2H2O)=11.47  
K(PdA(H-1L)+H)=10.82

A=1,2-diaminoethane

-----  
Pd++ gl KNO3 20°C 0.5M U K1=9.88 B2=19.45 1974KHb (38747) 297  
\*\*\*\*\*

C5H9NO4 H2L Glutamic acid CAS 56-86-0 (22)  
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Pd++ gl KCl 25°C 0.1M U B2=38.0 2004AEa (39116) 298  
K(Pd+HL)=46.7  
K(Pd+2H2L+2Cl)=54.2  
K(Pd+2L+OH)=30.1  
K(P2d+L+2Cl)=41.9

-----  
Pd++ gl none 25°C 0.0 U 1979FWa (39117) 299  
K(PdL2+H)=4.76  
K(PdHL2+H)=4.06  
K(PdCl4+2HL=PdH2L2+4Cl)=10.0

-----  
Pd++ gl NaClO4 25°C 0.10M U K1=10.38 B2=17.84 1972SSe (39118) 300  
\*\*\*\*\*

C5H9NO4 H2L MIDA CAS 4408-64-4 (190)  
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp KCl 20°C 0.10M U B2=24.88 1987KUa (39277) 301  
\*\*\*\*\*

C5H9NO4S H2L (1736)  
3-(Carboxymethyl)thio-L-alanine; HOOC.CH2.S.CH2.CH(NH2)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ kin NaClO4 25°C 1.0M U 1998VTa (39312) 302  
K(Pd+HL=PdL+H)=1.82  
K(Pd+H2L=PdHL+H)=2.43

\*\*\*\*\*

C5H9N3 L Histamine CAS 51-45-6 (103)  
 4(5)-(2'-Aminoethyl)imidazole; C3H3N2.CH2.CH2.NH2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	U	M			1999SSd (39543)	303
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K(Pd(pn)+L)=13.22

pn is 1,2-diaminopropane. For amine protonation, K1=9.59, B2=15.65.

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Pd++	gl	NaNO3	37°C	0.16M	M	M			1998ESa (39544)	304
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K(PdA+L)=12.56

A is 1,3-diaminopropane.

\*\*\*\*\*

C5H10N2O3 HL Glutamine CAS 56-85-9 (18)  
 2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	U	M			1999SSd (39831)	305
------	----	-------	------	-------	---	---	--	--	-----------------	-----

K(Pd(pn)+L)=11.02

K(Pd(pn)+L=PdH-1(pn)L+H)=2.12

pn is 1,2-diaminopropane. For amide protonation, K1=8.98.

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Pd++	gl	NaNO3	37°C	0.16M	M	M			1998ESa (39832)	306
------	----	-------	------	-------	---	---	--	--	-----------------	-----

K(PdA+L)=9.29

K(PdA+L=PdAH-1L+H)=-0.43

A is 1,3-diaminopropane.

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Pd++	gl	KCl	25°C	0.50M	U	M			1977LIa (39833)	307
------	----	-----	------	-------	---	---	--	--	-----------------	-----

K(Pd(en)+L)=10.8

\*K(Pd(en)L)=-9.03

\*\*\*\*\*

C5H10N2O3 HL Ala-Gly CAS 687-69-4 (55)  
 Alanyl-glycine; H2N.CH(CH3).CO.NH.CH2.COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.20M	C				1999AJa (39893)	308
------	----	------	------	-------	---	--	--	--	-----------------	-----

B(PdLC1)=17.96

B(PdH-1LC1)=15.09

B(PdH-2L)=4.38

B(PdH-1L2)=18.70

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-2L2)=13.37; B(PdH-1L)=13.10.

\*\*\*\*\*

C5H10N2O3 HL Gly-Ala CAS 3695-73-6 (56)  
 Glycyl-alanine; H2N.CH2.CO.NH.CH(CH3).COOH

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.20M	C				1999AJa (40005)	309
------	----	------	------	-------	---	--	--	--	-----------------	-----

B(PdLCl)=18.00  
B(PdH-1LCl)=16.01  
B(PdH-2L)=4.80  
B(PdH-1L2)=19.80

Medium: 0.1 M KNO<sub>3</sub>, 0.1 M KCl. B(PdH-1L)=14.02.

-----  
Pd++ gl NaNO<sub>3</sub> 25°C 0.10M U M 1999SSd (40006) 310

K(Pd(pn)+L)=8.17  
K(Pd(pn)+L=PdH-1(pn)L+H)=3.69

pn is 1,2-diaminopropane. For aminoacid protonation, K<sub>1</sub>=8.04.

\*\*\*\*\*

C5H10N2O3 HL Gly-b-Ala CAS 7536-21-2 (9057)

Glycyl-beta-alanine;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl oth/un 25°C 0.20M C K<sub>1</sub>=17.11 2003Amb (40010) 311

K(PdH-1L)=14.93  
K(PdH-2L)=6.00  
K(PdH-1L2)=20.60

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO<sub>3</sub>/0.10 M KCl.

\*\*\*\*\*

C5H10N2O3 HL B-Ala-Gly CAS 2672-88-0 (4323)

beta-Alanylglycine; H<sub>2</sub>N.CH<sub>2</sub>.CH<sub>2</sub>.CO.NH.CH<sub>2</sub>.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl oth/un 25°C 0.20M C K<sub>1</sub>=14.12 2003Amb (40051) 312

K(PdH-1L)=11.09  
K(PdH-2L)=2.38  
K(PdH-1L2)=17.43

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO<sub>3</sub>/0.10 M KCl.

\*\*\*\*\*

C5H10N4O3 L CAS 54376-69-1 (8335)

N,N'-Carbonylbis(2-aminoacetamide);

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl NaClO<sub>4</sub> 25°C 0.10M U TIH K<sub>1</sub>=10.75 B<sub>2</sub>=17.25 1980SAc (40137) 313

Data for 0.075-0.15 M. At I=0, K<sub>1</sub>=11.15, K<sub>2</sub>=6.70. Also data for 30 C.

DH and DS values.

\*\*\*\*\*

C5H10O5S2 HL CAS 110-50-9 (591)

(Butoxy)dithiomethanoic acid; CH<sub>3</sub>.CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>O.CSSH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ dis oth/un 25°C 0.25M U B<sub>2</sub>=>24 1982SAa (40163) 314

\*\*\*\*\*

C5H11N L Piperidine CAS 110-89-4 (105)

Perhydropyridine; cyclo(-CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>.NH.CH<sub>2</sub>.CH<sub>2</sub>-) C<sub>5</sub>H<sub>11</sub>N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	non-aq	25	100%	U				1994CVa (40452)	315
------	----	--------	----	------	---	--	--	--	-----------------	-----

K(PdAB+2L=PdAL<sub>2</sub>+B)=3.14

K(PdAC+2L=PdAL<sub>2</sub>+C)=0.28

Medium: CHCl<sub>3</sub>. A:n<sup>3</sup>-allyl; B:4-MeOC<sub>6</sub>H<sub>4</sub>.N=CH.CH=N.C<sub>6</sub>H<sub>4</sub>OMe; C:4-MeOC<sub>6</sub>H<sub>4</sub>.N=C(Me).C(Me)=N.C<sub>6</sub>H<sub>4</sub>OMe. Also data for L=morpholine, NHEt<sub>2</sub>, N-methylaniline.

\*\*\*\*\*

C<sub>5</sub>H<sub>11</sub>N<sub>02</sub> HL Valine CAS 72-18-4 (43)

2-Amino-3-methylbutanoic acid; H<sub>2</sub>N.CH(CH(CH<sub>3</sub>)<sub>2</sub>)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO <sub>3</sub>	25	0.10M	U	M			1999SSd (40742)	316
------	----	-------------------	----	-------	---	---	--	--	-----------------	-----

K(Pd(pn)+L)=11.36

pn is 1,2-diaminopropane. For aminoacid protonation, K<sub>1</sub>=9.57, B<sub>2</sub>=11.70.

Pd++	gl	NaNO <sub>3</sub>	37	0.16M	M	M			1998ESa (40743)	317
------	----	-------------------	----	-------	---	---	--	--	-----------------	-----

K(PdA+L)=9.55

A is 1,3-diaminopropane.

Pd++	gl	KNO <sub>3</sub>	20	0.5M	U			K <sub>1</sub> =9.62 B <sub>2</sub> =17.76	1974KHb (40744)	318
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C<sub>5</sub>H<sub>11</sub>N<sub>02</sub> L (8054)

Alanine ethyl ester;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO <sub>3</sub>	25	0.20M	M	M		K <sub>1</sub> =5.15	1987SKb (40866)	319
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K(Pd(dien)+L)=3.92

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C<sub>5</sub>H<sub>11</sub>N<sub>02</sub>S HL Methionine CAS 63-68-3 (42)

2-Amino-4-(methylthio)butanoic acid; H<sub>2</sub>N.CH(CH<sub>2</sub>.CH<sub>2</sub>.S.CH<sub>3</sub>)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaClO <sub>4</sub>	25	1.00M	C			K <sub>1</sub> =16.8 B <sub>2</sub> =34.30	2000SAb (41115)	320
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Pd++	gl	NaNO <sub>3</sub>	25	0.10M	U	M			1999SSd (41116)	321
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K(Pd(pn)+L)=10.37

pn is 1,2-diaminopropane. For aminoacid protonation, K<sub>1</sub>=9.10, B<sub>2</sub>=11.08.

Pd++	gl	NaNO <sub>3</sub>	37	0.16M	M	M			1998ESa (41117)	322
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K(PdA+L)=8.83

A is 1,3-diaminopropane.

Pd++	gl	KNO <sub>3</sub>	25	0.50M	U				1978LIa (41118)	323
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K(Pd(en)+L)=9.14



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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	1.00M	C			B2=48.2 K(Pd+HL)=27.3 K(Pd+2HL)=44.8	2000SAb (41278)	324

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.20M	M	M		K(Pd(dien)+L)=4.43	1987SKb (41312)	325

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	dis	oth/un	?	?	U	M	B2=64.9 B(PdLC1)=44.6	1969BHd (41367)	326
Pd++	sp	non-aq	?	100%	U	M	K(Pd(HA)2+2HL=PdL2+2H2A)=1.6	1968SRg (41368)	327

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	U	M	K(Pd(pn)+L)=13.65 K(Pd(pn)+H+L)=19.86	1999SSd (41582)	328
pn is 1,2-diaminopropane. For aminoacid protonation, K1=10.58, B2=19.43.									

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C5H12N2O2S                      HL            Met-hydroxamic            CAS 19253-87-3    (5992)  
2-Amino-4-(methylthio)butanehydroxamic acid, Methionine hydrox.a.;  
CH3.S.CH2.CH2.CH(NH2).CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KCl	25°C	0.15M	U	M		K1=13.230 B2=23.333 B(PdL(GlyGly))=21.058 B(PdHL(GlyGly))=24.370 B(Pd(GlyGly))=9.155	1990MSa (41607)	330

\*\*\*\*\*

C5H13N3 L (1866)  
cis-3,5-Diaminopiperidine; C5H9N(NH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	20°C	0.10M	U			K(PdL2+H)=6.35 K(PdHL2+H)=4.16	1979MSa (41795)	331

pK's for the other isomer of PdL2  
together with X-ray structure

\*\*\*\*\*

C6H3N3O7 HL Picric acid CAS 88-89-1 (593)  
2,4,6-Trinitrophenol; HO.C6H2(NO2)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	C	M		*K(PdL(H2O)2)=-5.00 *B2(PdL(H2O)2)=-13.79 K(2PdL(H2O)2=Pd2(OH)L2+H)=-2.28, K(2PdL(H2O)2=Pd2(OH)L2+2H)=-6.59	2001NSa (42142)	332

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C6H4N2O7S H3L (2023)  
2,4-Dinitroso-6-sulfonoresorcinol; (HO)2.C6H(N:O)2(SO3H)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	none	25°C	0.0	U			K1=4.2 B2=8.3	1980MGa (42269)	333

\*\*\*\*\*

C6H5NO2 L Nitrobenzene CAS 98-95-3 (3085)  
Nitrobenzene; C6H5.NO2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		K(Pd3A3CO+L)=0.24	1994PAa (42701)	334

Medium: MeOH. A=Bis(diphenylphosphino)methane

\*\*\*\*\*

C6H5NO2S H2L (6876)  
2-Mercaptopyridine-3-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	40%	C				1996ABc (42708)	335

K(Pd+HL)=16.10  
 K(Pd+H2L)=8.75  
 K(Pd+H3L=PdH2L+H)=3.05  
 \*K(PdH2L)=-3.15

Medium: 40% v/v EtOH/H2O, 0.10 M NaClO4.

\*\*\*\*\*

C6H6 L Benzene CAS 71-43-2 (2143)  
 Benzene, cyclohexatriene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		1994PAa (43169)	336
							K(Pd3A3CO+L)<-1.15		

Medium: MeOH. A=Bis(diphenylphosphino)methane

\*\*\*\*\*

C6H6NBr L 3-Bromoaniline CAS 591-19-5 (758)  
 3-Bromoaniline; H2N.C6H4.Br

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.00M	U	M		1972VGa (43177)	337
							K(M(H2O)4+L=M(H2O)3L+H2O)=6.30		

Medium: HClO4

\*\*\*\*\*

C6H6N2O2 L m-Nitroaniline CAS 99-09-2 (464)  
 3-Nitroaminobenzene; H2N.C6H4.NO2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.00M	U	M		1972VGa (43389)	338
							K(M(H2O)4+L=M(H2O)3L+H2O)=5.68		

Medium: HClO4

\*\*\*\*\*

C6H6N2O2 L p-Nitroaniline CAS 100-01-6 (465)  
 4-Nitroaminobenzene; H2N.C6H4.NO2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.0M	U			1972VGa (43405)	339
							K(Pd(H2O)4+L=Pd(H2O)3L)=4.53		

\*\*\*\*\*

C6H6O2 H2L Catechol CAS 120-80-9 (534)  
 1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.20M	U			1981Cmb (43807)	340
							K(Pd+H2L=PdL+2H)=-2.2		

\*\*\*\*\*

C6H6O4 HL Kojic acid CAS 501-30-4 (1800)

5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	alc/w	RT	20%	C	I	K1=7.44	1989MEb (44236)	341
Medium: 20% v/v MeOH/H2O. Data for 20-50% v/v MeOH/H2O, EtOH/H2O, acetone/H2O, DMF/H2O and dioxane/H2O.									
*****									
C6H7N		L	Aniline				CAS 62-53-3	(583)	
Aminobenzene, aniline; C6H5.NH2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.00M	U	M		1972VGa (44876)	342
							K(M(H2O)4+L=M(H2O)3L+H2O)=7.20		
Medium: HClO4									
*****									
C6H7N5		HL	9-Methyladenine				CAS 700-00-5	(4347)	
9-Methyl-6-aminopurine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.20M	C			1997WKa (45172)	343
							K(PdACl+L=PdAL+Cl)=4.33		
PdA is [PdH-1(gly-Met)].									
*****									
C6H8N2		L	2-Picolylamine				CAS 29722-36-9	(502)	
2-(Aminomethyl)pyridine; C5H4N.CH2NH2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.10M	C			1997RSa (45359)	344
							*K(PdL(H2O)2)=-4.43		
							*B2(PdL(H2O)2)=-13.07		
*****									
C6H8N2O2		HL	1-Methylthymine				CAS 4160-72-9	(7411)	
2,4-Dihydroxy-1,5-dimethylpyrimidine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.20M	C	M		2001NSa (45383)	345
							K(Pd(en)+L)=9.05		
							K(Pd(en)+2L)=14.76		
							*K(Pd(en)(H2O)L)=-8.44		
K(Pd(en)(H2O)+L=Pd(en)(OH)L+H)=0.61,									
K(2Pd(en)(H2O)2+2L=Pd2(en)2(OH)L2+H)=12.70									
*****									
Pd++	gl	KN03	25°C	0.20M	C	M		2001NSa (45384)	346
							K(Pd(pic)+L)=9.56		
							K(Pd(pic)+2L)=15.40		

\*K(Pd(pic)(H2O)L)=-8.00

K(Pd(pic)(H2O)+L=Pd(pic)(OH)L+H)=1.56

K(2Pd(pic)(H2O)2+2L=Pd2(pic)2(OH)L2+H)=14.30. Hpic=picric acid.

-----  
Pd++ gl KNO3 25°C 0.20M C K1=7.71 2000NFa (45385) 347  
-----

Pd++ gl KNO3 25°C 0.20M C 1997WKa (45386) 348  
K(PdACl+L=PdAL+Cl)=7.26

PdA is [PdH-1(gly-Met)].

\*\*\*\*\*

C6H8O4 H2L CAS 5445-51-2 (69)

Cyclobutane-1,1-dicarboxylic acid; C4H6(COOH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaClO4 37°C 0.15M C M 2003TMb (45509) 349  
K(Pd(en)+L)=6.16

-----  
Pd++ gl NaClO4 25°C 0.10M M M 2002BSa (45510) 350  
K(PdA+L)=6.61  
K(PdA+H+L)=9.69

HA is S-methyl cysteine.

-----  
Pd++ gl NaNO3 25°C 0.10M C M 2002MSb (45511) 351  
K(PdA+L)=7.17  
K(PdA+H+L)=9.44

A is N,N'-dimethylethylenediamine.

-----  
Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45512) 352  
K(Pd(bpy)(H2O)2+L)=8.45  
K(Pd(bpy)(H2O)2+H+L)=11.37

-----  
Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45513) 353  
K(Pd(bpy)(H2O)2+L+A)=18.31  
K(Pd(bpy)(H2O)2+L+A+H)=24.76  
K(Pd(bpy)(H2O)2+L+A+2H)=27.05

HA is uracil.

-----  
Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45514) 354  
K(Pd(bpy)(H2O)2+L+A)=20.14  
K(Pd(bpy)(H2O)2+L+A+H)=26.74  
K(Pd(bpy)(H2O)2+L+A+2H)=28.62

HA is uridine.

-----  
Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45515) 355  
K(Pd(bpy)(H2O)2+L+A)=16.64  
K(Pd(bpy)(H2O)2+L+A+H)=22.77  
K(Pd(bpy)(H2O)2+L+A+2H)=25.58

HA is inosine.

-----

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45516) 356  
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A})=17.06$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+\text{H})=23.24$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+2\text{H})=27.08$

A is adenine.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45517) 357  
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A})=16.00$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+\text{H})=22.42$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+2\text{H})=27.92$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+3\text{H})=31.49$

H3A is inosine-5'-monophosphate.

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (45518) 358  
 $K(\text{Pd}(\text{pn})+\text{L})=6.05$   
 pn is 1,2-diaminopropane. For acid protonation,  $K_1=5.42$   $B_2=8.06$ .

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (45519) 359  
 $K(\text{PdA}+\text{L})=6.39$   
 A is 1,3-diaminopropane.

Pd++ gl NaClO4 25°C 0.10M C M 1997RSa (45520) 360  
 $K(\text{PdA}+\text{L})=7.34$

A=2-(Aminomethyl)pyridine

\*\*\*\*\*

C6H8O6S H3L CAS 99-68-3 (3692)  
 (Carboxymethylthio)butanedioic acid;  $\text{HOOC}.\text{CH}(\text{S}.\text{CH}_2.\text{COOH}).\text{CH}_2.\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 20°C 0.10M U K1=5.20 1977CAd (45707) 361

\*\*\*\*\*

C6H8O7 H3L Citric acid CAS 77-92-9 (95)  
 2-Hydroxypropane-1,2,3-tricarboxylic acid;  $\text{HOOCCH}_2.\text{CH}(\text{OH})(\text{COOH}).\text{CH}_2\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=3.46 1997SEa (46230) 362

$K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=0.66$

$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-3.4 \text{ kJ mol}^{-1}$ ,  $\text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=2 \text{ J K}^{-1} \text{ mol}^{-1}$

\*\*\*\*\*

C6H9NO6 H3L NTA CAS 139-13-9 (191)  
 Nitrilotriethanoic acid;  $\text{N}(\text{CH}_2.\text{COOH})_3$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 20°C 1.00M C M T K1=17.1 B2=23.70 1976AMa (46982) 363

$K(\text{PdL}+\text{H})=2.48$

$K(\text{PdHL}+\text{H})=0.5$

$K(\text{PdL}(\text{OH})+\text{H})=7.82$

$$K(\text{PdL}+\text{PdL}(\text{OH})=\text{Pd}_2\text{L}_2(\text{OH}))=3.1$$

By exchange with  $\text{PdBr}_4$ .  $K(\text{PdL}+\text{Br})=2.7$

\*\*\*\*\*

C6H9N3O2                      HL            Histidine                      CAS 71-00-1    (1)  
2-Amino-3-(4'-imidazolyl)propanoic acid;  $\text{H}_2\text{N}.\text{CH}(\text{CH}_2.\text{C}_3\text{H}_3\text{N}_2).\text{COOH}$

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Pd++            gl    NaNO3    25°C 0.10M U            M                      1999SSd (47602) 364

$$K(\text{Pd}(\text{pn})+\text{L})=14.75$$

pn is 1,2-diaminopropane. For aminoacid protonation,  $K_1=9.53$ ,  $B_2=15.81$ ,  $B_3=17.81$ .

-----  
Pd++            gl    NaNO3    37°C 0.16M M            M                      1998ESa (47603) 365

$$K(\text{PdA}+\text{L})=12.48$$

A is 1,3-diaminopropane.

\*\*\*\*\*

C6H10O4S                      H2L                      CAS 111-17-1    (139)  
3,3'-Thiodipropanoic acid;  $\text{HOOC}.\text{CH}_2.\text{CH}_2.\text{S}.\text{CH}_2.\text{CH}_2.\text{COOH}$

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Pd++            sp    NaCl    25°C 0.50M U            M                      1987CMc (48193) 366

$$K(\text{PdCl}_4+\text{L}=\text{PdCl}_3\text{L}+\text{Cl})=5.42$$

$$K(\text{PdCl}_3\text{L}+\text{L}=\text{PdCl}_2\text{L}_2+\text{Cl})=2.87$$

$$K(\text{PdL}_2+\text{Cl})=4.30$$

$$K(\text{PdL}_2\text{Cl}+\text{Cl})=2.51$$

-----  
Pd++            sp    NaClO4    25°C 0.50M U                      1986CCe (48194) 367

$$B(\text{PdH}_2\text{L})=16.71$$

$$B(\text{PdH}_4\text{L}_2)=31.60$$

$$K(\text{Pd}+\text{H}_2\text{L})=7.40$$

$$K(\text{PdH}_2\text{L}+\text{H}_2\text{L})=5.58$$

\*\*\*\*\*

C6H10O4S2                      H2L                      CAS 7244-02-2    (438)  
1,2-Bis(carboxymethylthio)ethane;  $\text{HOOC}.\text{CH}_2.\text{S}.\text{CH}_2.\text{CH}_2.\text{S}.\text{CH}_2.\text{COOH}$

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Pd++            sp    oth/un    25°C 0.10M U                       $K_1=4.48$      $B_2=6.91$     1978POa (48249) 368

\*\*\*\*\*

C6H10O4S2                      H2L                      CAS 1119-62-6    (3697)  
3,3'-Di(thiopropoic acid);  $\text{HOOC}.\text{CH}_2.\text{CH}_2.\text{S}.\text{S}.\text{CH}_2.\text{CH}_2.\text{COOH}$

-----  
Metal            Mtd Medium Temp Conc Cal Flags Lg K values                      Reference ExptNo  
-----

Pd++            sp    NaClO4    25°C 0.50M U                      1986CCe (48270) 369

$$B(\text{PdH}_2\text{L})=15.25$$

$$B(\text{Pd}_2\text{H}_2\text{L})=19.67$$

$$K(\text{Pd}+\text{H}_2\text{L})=5.92$$

$$K(\text{PdH2L}+\text{H2L})=10.34$$

\*\*\*\*\*

C6H10O4Se2                      H2L                      CAS 86515-79-7 (6099)  
Ethylene-bis-selenoglycolic acid; HOOC.CH2.Se.CH2.CH2.Se.CH2.COOH

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              con NaCl      25°C 3.00M C              K1=6.32      B2=11.97      1988PFb (48297) 370

\*\*\*\*\*

C6H11NO2                      HL                      CAS 89203-64-5 (3435)  
1-Pyrrolidine-1-ethanoic acid, 1-Azacyclopentane-1-ethanoic acid;

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              sp none      25°C 0.0 U              K1=11.20      B2=21.23      1974HFa (48504) 371

\*\*\*\*\*

C6H11NO4                      H2L      Amino adipic              CAS 542-32-5 (1259)  
2-Aminohexanedioic acid; HOOC.CH2.CH2.CH2.CH(NH2).COOH

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              gl none      25°C 0.0 U                                      1979FWa (48583) 372

$$K(\text{PdL2}+\text{H})=5.04$$

$$K(\text{PdHL2}+\text{H})=4.45$$

$$K(\text{PdCl4}+2\text{HL}=\text{PdH2L2}+4\text{Cl})=10.5$$

\*\*\*\*\*

C6H11N3O4                      HL      Gly-Gly-Gly              CAS 556-33-2 (415)  
Glycyl-glycyl-glycine; H2N.CH2.CO.NH.CH2.CO.NH.CH2.COOH

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              gl KNO3      25°C 0.20M C              B2=23.0              1999AJa (48987) 373

$$B(\text{PdLCl})=17.91$$

$$B(\text{PdH-1LCl})=14.64$$

$$B(\text{PdH-2L})=9.07$$

$$B(\text{PdH-3L})=-1.15$$

Medium: 0.1 M KNO3, 0.1 M KCl.  $B(\text{PdH-1L2})=19.81$ ;  $B(\text{PdH-2L2})=13.40$ .

-----  
Pd++              sp oth/un 25°C ? U                                      1978Cwa (48988) 374

$$K(\text{PdH-2L}+\text{H})=2.2$$

$$K(\text{PdH-1L}+\text{H})=1.5$$

\*\*\*\*\*

C6H12                      L                      CAS 592-41-6 (2771)  
1-Hexene; CH2:CH(CH2)3.CH3

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              oth non-aq 30°C 100% U      M                                      1974KKb (49012) 375

$$K(\text{PdCl2}+\text{L})=0.28$$

Medium: N-methylacetamide



\*\*\*\*\*

C6H12 L CAS 760-21-4 (2772)  
2-Ethyl-1-butene; CH<sub>2</sub>:C(C<sub>2</sub>H<sub>5</sub>).CH<sub>2</sub>.CH<sub>3</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth non-aq 30°C 100% U M 1974KKb (49015) 376  
K(PdCl<sub>2</sub>+L)=-0.89

Medium: N-methylacetamide

\*\*\*\*\*

C6H12 L CAS 763-29-1 (2770)  
2-Methyl-1-pentene; CH<sub>2</sub>:C(CH<sub>3</sub>).CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>3</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth non-aq 30°C 100% U M 1974KKb (49017) 377  
K(PdCl<sub>2</sub>+L)=-0.85

Medium: N-methylacetamide

\*\*\*\*\*

C6H12 L CAS 691-37-2 (2767)  
4-Methyl-1-pentene; CH<sub>2</sub>:CH.CH<sub>2</sub>.CH(CH<sub>3</sub>)<sub>2</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth non-aq 30°C 100% U M 1974KKb (49019) 378  
K(PdCl<sub>2</sub>+L)=0.18

Medium: N-methylacetamide

\*\*\*\*\*

C6H12 L CAS 7668-21-3 (2774)  
cis-2-Hexene; CH<sub>3</sub>.CH:CH.CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>3</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth non-aq 30°C 100% U M 1974KKb (49021) 379  
K(PdCl<sub>2</sub>+L)=0.11

Medium: N-methylacetamide

\*\*\*\*\*

C6H12 L (2768)  
cis-4-Methyl-2-pentene; CH<sub>3</sub>.CH:CH.CH(CH<sub>3</sub>)<sub>2</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth non-aq 30°C 100% U M 1974KKb (49023) 380  
K(PdCl<sub>2</sub>+L)=0.26

Medium: N-methylacetamide

\*\*\*\*\*

C6H12 L CAS 4050-45-7 (2773)  
trans-2-Hexene; CH<sub>3</sub>.CH:CH.CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>3</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

-----  
Pd++ oth non-aq 30°C 100% U M 1974KKb (49025) 381  
K(PdCl2+L)=-0.31

Medium: N-methylacetamide

\*\*\*\*\*

C6H12 L CAS 4461-48-7 (2769)

trans-4-Methyl-2-pentene; CH3.CH:CH.CH(CH3)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth non-aq 30°C 100% U M 1974KKb (49027) 382  
K(PdCl2+L)=-0.42

Medium: N-methylacetamide

\*\*\*\*\*

C6H12N2O3 HL B-Ala-B-Ala CAS 34322-87-7 (2118)

3-Alanyl-3-alanine; H2N.CH2.CH2.CO.NH.CH2.CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C 2003AMB (49061) 383

K(PdH-1L)=11.19

K(PdH-2L)=2.52

K(PdH-1L2)=17.76

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO3/0.10 M KCl.

\*\*\*\*\*

C6H12O2S2 HL CAS 35088-67-6 (2829)

1-Ethylthio-2-thiocarboxymethylethane; C2H5.S.CH2.CH2.S.CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un 25°C 0.10M U K1=6.34 B2=11.03 1978POa (49451) 384

\*\*\*\*\*

C6H13NO2 HL Isoleucine CAS 73-32-5 (424)

2-Amino-3-methylpentanoic acid; CH3.CH2.CH(CH3).CH(NH2).COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 20°C 0.5M U K1=9.71 B2=18.15 1974KHb (49911) 385

\*\*\*\*\*

C6H13NO2 HL Leucine CAS 61-90-5 (47)

2-Amino-4-methylpentanoic acid; H2N.CH(CH2.CH(CH3)2)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.50M U T 1978LIa (50096) 386

K(Pd(en)+L)=11.41

-----  
Pd++ gl KNO3 20°C 0.5M U K1=9.94 B2=18.17 1974KHb (50097) 387

\*\*\*\*\*

C6H13NO2S HL Ethionine CAS 67-21-0 (1909)

2-Amino-4-(ethylthio)butanoic acid; CH<sub>3</sub>.CH<sub>2</sub>.S.CH<sub>2</sub>.CH<sub>2</sub>.CH(NH<sub>2</sub>).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO <sub>4</sub>	25°C	1.00M	C			K <sub>1</sub> =16.8 B <sub>2</sub> =34.00	2000SAb (50266)	388
Pd++	gl	NaCl	25°C	0.16M	U			K <sub>1</sub> =9.112 B <sub>2</sub> =14.361 B(Pd <sub>2</sub> L)=18.487 B(Pd <sub>2</sub> H <sub>2</sub> L)=23.979 B(PdH-1L)=5.059	1986AEa (50267)	389

\*\*\*\*\*

C6H<sub>13</sub>N<sub>3</sub>O<sub>3</sub> HL Citrulline (579)

2-Amino-5-ureidovaleric acid; H<sub>2</sub>N.CO.NH.CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>.CH(NH<sub>2</sub>).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO <sub>3</sub>	25°C	0.10M	C			B <sub>2</sub> =16.23	1991GLb (50585)	390

\*\*\*\*\*

C6H<sub>14</sub>N<sub>2</sub>O<sub>2</sub> HL Lysine CAS 56-87-1 (41)

2,6-Diaminohexanoic acid; H<sub>2</sub>N.(CH<sub>2</sub>)<sub>4</sub>.CH(NH<sub>2</sub>).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaNO <sub>3</sub>	25°C	0.10M	U	M		K(Pd(pn)+L)=11.49 K(Pd(pn)+H+L)=20.44	1999SSd (50830)	391
pn is 1,2-diaminopropane. For aminoacid protonation, K <sub>1</sub> =10.44, B <sub>2</sub> =19.66.										
Pd++	gl	NaNO <sub>3</sub>	37°C	0.16M	M	M		K(PdA+L)=9.28 K(PdA+H+L)=19.03	1998ESa (50831)	392

A is 1,3-diaminopropane.

\*\*\*\*\*

C6H<sub>14</sub>O<sub>2</sub>S L CAS 10595-09-2 (3698)

3,3'-Thiodipropanol; S(CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	kin	oth/un	25°C	1.00M	U			K <sub>1</sub> eff=4.51	1996SEa (51036)	393

Medium: 1.00 M HClO<sub>4</sub>.

\*\*\*\*\*

C6H<sub>14</sub>O<sub>2</sub>S<sub>2</sub> L CAS 5244-34-8 (4390)

3,6-Dithiaoctan-1,8-diol; HO.CH<sub>2</sub>.CH<sub>2</sub>.S.CH<sub>2</sub>.CH<sub>2</sub>.S.CH<sub>2</sub>.CH<sub>2</sub>.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	KCl	25°C	1.00M	U			B <sub>2</sub> =21.63	1991ZPa (51038)	394

\*\*\*\*\*

C6H<sub>18</sub>N<sub>4</sub> L Trien-tetramine CAS 112-24-3 (11)

1,4,7,10-Tetraazadecane; H2N.CH2.CH2.NH.CH2.CH2.NH.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	1.00M	C	I M		K1=40.1	1985YAa (52131)	395
*****										
C6H18N4		L	Tren					CAS 4097-89-6	(817)	
2,2',2''-Triaminotriethylamine; (H2N.CH2.CH2)3N										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	1.00M	C	M			1986ANa (52207)	396
B(Pd2L2)=77.4										
Ternary complexes with Cl-, Br-, I- and SCN-. pH-metric and spec. study.										
*****										
C7H5N		L	Cyanobenzene					CAS 100-47-0	(4406)	
Cyanobenzene, benzonitrile; C6H5.CN										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M			1994PAa (52570)	397
K(Pd3A3CO+L)=0.13										
Medium: MeOH. A=Bis(diphenylphosphino)methane										
*****										
C7H5NO4		H2L	Dipicolinic aci					CAS 449-83-2	(418)	
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KCl	25°C	0.20M	U			K1=16.0	1980KDb (52799)	398
*****										
C7H6O2		HL	Salicylaldehyde					CAS 90-02-8	(193)	
2-Hydroxybenzaldehyde, Salicylaldehyde; H0.C6H4.CHO										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	U			K1=7.74 B2=14.77	1949MMa (53631)	399
*****										
C7H6O2		HL	Benzoic Acid					CAS 65-85-0	(462)	
Benzenecarboxylic acid; C6H5.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M			1994PAa (53853)	400
K(Pd3A3CO+L)=4.0										
Medium: MeOH. A=Bis(diphenylphosphino)methane										
*****										
C7H7NO2		HL						CAS 150-13-0	(1376)	
4-Aminobenzoic acid; H2N.C6H4.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		K(Pd3A3CO+L)=3.52	1994PAa (55390)	401
Medium: MeOH. A=Bis(diphenylphosphino)methane ***** C7H7NO2 HL CAS 495-18-1 (184) Benzohydroxamic acid; C6H5.CO.NH.OH										
Pd++	gl	diox/w	25°C	70%	U		K1=9.52 B2=17.55		1969JSa (55512)	402
***** C7H8 L CAS 108-88-3 (2144) Toluene; C6H5.CH3										
Pd++	sp	alc/w	25°C	100%	U	M		K(Pd3A3CO+L)=-1.10	1994PAa (55785)	403
Medium: MeOH. A=Bis(diphenylphosphino)methane ***** C7H8OS L CAS 1193-82-4 (1881) Phenylmethanesulfoxide; C6H5.SO.CH3										
Pd++	sp	alc/w	25°C	95%	U			K(PdCl4+L=PdLC13+Cl)=0.94	1982CCa (56055)	404
***** C7H8O2 H2L Methylcatechol CAS 452-86-8 (525) 1,2-Dihydroxy-4-methylbenzene; CH3.C6H3(OH)2										
Pd++	sp	oth/un	25°C	0.20M	U			K(Pd+H2L=PdL+2H)=2.40	1981CMA (56075)	405
***** C7H9N L 3-Methylaniline CAS 108-44-1 (755) 3-Methylaniline (3-Toluidine); CH3.C6H4.NH2										
Pd++	kin	oth/un	25°C	?	U	M		K(M(H2O)4+L=M(H2O)3L+H2O)=7.57	1972VGa (56309)	406
***** C7H9N L 4-Methylaniline CAS 106-49-0 (754) 4-Methylaniline (4-Toluidine); CH3.C6H4.NH2										

-----  
Pd++ kin oth/un 25°C ? U M 1972VGa (56343) 407  
K(M(H2O)4+L=M(H2O)3L+H2O)=8.04

\*\*\*\*\*  
C7H9NO L p-Anisidine CAS 104-94-7 (3764)  
4-Methoxyaniline; CH3O.C6H4.NH2  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ kin oth/un 25°C 2.0M U M 1972VGa (56397) 408  
K(M(H2O)4+L=M(H2O)3L+H2O)=7.81

\*\*\*\*\*  
C7H9N5O HL 9-Ethylguanine CAS 879-08-3 (6679)  
9-Ethyl-2-amino-6-hydroxypurine;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C M 2003NFA (56518) 409  
K(PdA+L)=8.11  
K(PdA+H+L)=15.06  
K(2PdA+L)=14.95

A is bis-((2-pyridyl)methyl)amine

\*\*\*\*\*  
C7H10N4O2S L Sulfaguanidine CAS 57-67-0 (4469)  
4-Aminobenzenesulfonyl guanidine; H2N.C(:NH).NH.SO2.C6H4.NH2  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp NaClO4 ? 1.0M U M 1970RGa (56704) 410  
K(PdCl2+L)=5.42  
K(PdCl2L+L)=4.38

\*\*\*\*\*  
C7H11N3O L Acetylhistamine CAS 673-49-4 (7412)  
4-(2'-Acetylaminioethyl)imidazole; C3H3N2.CH2CH2.NH.COCH3  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C M 2003NFA (56961) 411  
K(PdA+L)=7.72  
K(2PdA+L=(PdA)2H-1L+H)=6.13

A is bis-((2-pyridyl)methyl)amine

-----  
Pd++ gl KNO3 25°C 0.20M C 1997WKA (56962) 412  
K(PdACl+L=PdAL+Cl)=5.48  
K(2PdACl+L=Pd2A2H-1L+2Cl)=-0.9

PdA is [PdH-1(gly-Met)].

\*\*\*\*\*  
C7H11N3O2 L CAS 7389-87-9 (3162)  
Histidine methyl ester  
-----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	M				1987SKb (57005)	413
K(Pd(dien)+L)=4.61 K(Pd(dien)+H+L)=10.58 ***** C7H13NO2 HL CAS 3235-67-4 (3772) Piperidine-N-ethanoic acid; C5H10N-CH2.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	none	25°C	0.0	U			K1=10.32 B2=19.72	1974HFa (57456)	414
***** C7H13NO3S HL CAS 65-82-7 (8508) N-Acetylmethionine;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	C	M			2003NFa (57495)	415
K(PdA+L)=3.41 A is bis-((2-pyridyl)methyl)amine. Competitive method using uridine.										

Pd++	gl	KNO3	25°C	0.20M	C	HM			2000NFa (57496)	416
K(Pd(dien)+H+L)=8.49 K(Pd(dien)+L)=5.61 K(PdA+H+L)=7.07 K(PdA+L)=3.66 Method: uridine as a competitive ligand. A is terpyridine. K(Pd(dien)+OH)=6.25, K(PdA+OH)=6.91. By calorimetry: DH(Pd(dien)+L)=-38.8 kJ mol <sup>-1</sup> .										

Pd++	gl	KNO3	25°C	0.20M	C	HM			2000NFa (57497)	417
K(Pd(gly-gly)+H+L)=8.74 K(Pd(gly-gly)+L)=4.89 K(Pd(gly-ala)+H+L)=8.76 K(Pd(gly-ala)+L)=4.91 Method: uridine as a competitive ligand. K(Pd(gly-gly)+OH)=4.64, K(Pd(gly-ala)+OH)=4.72. By calorimetry: DH(Pd(gly-ala)+L)=-38.0 kJ mol <sup>-1</sup> .										

Pd++	gl	KNO3	25°C	0.20M	C	HM			2000NFa (57498)	418
K(Pd(gly-met)+H+L)=7.29 K(Pd(gly-met)+L)=3.24 K(Pd(gly-met)+OH)=4.82 Method: uridine as a competitive ligand. By calorimetry: DH(Pd(gly-met)+L)=-21.2 kJ mol <sup>-1</sup> .										

***** C7H13NO4 H2L Aminopimelic CAS 627-76-9 (1260) 2-Amino-heptanedioic acid; HOOC.(CH2)4.CH(NH2).COOH										
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KCl	25°C 0.10M U	1979FWa (57501) 419
				K(Pd(HL)2= Pd(HL)L+H)=4.58
				K(Pd(HL)L= PdL2+H)=5.33
				K(PdCl4+2HL= Pd(HL)2+4Cl)=11.5

\*\*\*\*\*

C7H13N3O4                      HL            Ala-Asn                      CAS 1999-41-3    (5934)  
Alanyl-asparagine; NH2.CH(CH3.CO.NH.CH(CH2.CO.NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp NaCl 2000C 0.15M U 1990YKa (57648) 420  
Keff(Pd+L+2Cl)=21.5

Eff constant : stability of  $\text{PdCl}_4$  is not accounted

\*\*\*\*\*

C7H13N3O4	HL	Gly-b-Ala-Gly	CAS 42538-54-5 (9058)
Glycyl-beta-alanylglycine;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Pd++      gl oth/un 25°C 0.20M C      K<sub>1</sub>=16.26      2003AMB (57660) 421  
K(PdH-1L)=12.06  
K(PdH-2L)=11.79

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO<sub>3</sub>/0.10 M KCl.

\*\*\*\*\*

C7H13N3O4                      HL           Gly-Gly-b-Ala           CAS 42538-53-4    (4453)  
Glycylglycyl-beta-alanine; H2N.CH2.CO.NH.CH2.CO.NH.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++ gl oth/un 25°C 0.20M C K1=16.79 2003AMB (57679) 422  
K(PdH-1L)=10.97  
K(PdH-2L)=10.98

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO<sub>3</sub>/0.10 M KCl.

\*\*\*\*\*

C7H13N3O4                      HL           Gly-Gly-Ala                      CAS 19729-30-7    (3775)  
Glycylglycylalanine; H2N.CH2.CO.NH.CH2.CO.NH.CH(CH3).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KN03	25♦C	0.20M C	B2=23.7	1999AJa (57687) 423
					B(PdLCl)=17.91	
					B(PdH-1LCl)=14.45	
					B(PdH-2L)=8.99	
					B(PdH-3L)=-2.40	

Medium: 0.1 M KNO<sub>3</sub>, 0.1 M KCl. B(PdH-1L2)=19.60; B(PdH-2L2)=15.74.

\*\*\*\*\*

C7H13N3O4 HL b-Ala-Gly-Gly CAS 42538-55-6 (4452)  
beta-Alanylglycylglycine; H2N.CH2.CH2.CO.NH.CH2.CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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 Pd++ gl oth/un 25°C 0.20M C K1=14.40 2003AMB (57694) 424  
 K(PdH-1L)=8.76  
 K(PdH-2L)=9.03

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO3/0.10 M KCl.  
 \*\*\*\*\*

C7H14N2O3S HL Gly-Met CAS 554-94-9 (726)  
 Glycyl-methionine; H2N.CH2.CO.NH.CH(CH2.CH2.S.CH3).COOH

-----  

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	gl	KCl	25°C	0.20M	C	M		2001BNa (57800)	425
							*K(PdL)=-3.61		
							K(PdH-1L+Cl)=2.03		
							*K(PdH-1L)=-5.34		
							K(PdH-1L+H+L)=11.47		

K(PdH-1L+L)=4.56, K(PdH-1L+glygly)=4.72.  
 \*\*\*\*\*

C7H15NO5S HL MOPSO CAS 68399-77-9 (1967)  
 3-(N-Morpholino)-2-hydroxypropane sulfonic acid;

-----  

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.10M	C	M	K1=3.50	2001AAa (57996)	426
------	----	------	------	-------	---	---	---------	-----------------	-----

Also data for ternary complexes with 5'-GMP, 5'-IMP and 5'-CMP.

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C7H17N2O4PS H2L CAS 82611-22-1 (7392)  
 Methionyl-1-aminoethylphosphonic acid; H2L

-----  

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.10M	C		B2=21.60	1997LBa (58201)	427
							B(PdHLC12)=24.01		
							B(PdLC1)=17.87		
							B(PdH-1L)=10.76		
							B(PdH-2L)=-0.01		

Data are for (S,S)-isomer. B(PdH2L2)=34.96, B(PdHL2)=28.74, B(PdH-1L2)=12.51  
 B(PdH-2L2)=2.48. Data also for (R,S)-isomer.

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Pd++	gl	KCl	25°C	0.10M	U			1996BRa (58202)	428
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K(Pd+2L+2H)=35.35  
 K(Pd+2L)=21.99  
 K(Pd+2L+H)=29.14

H2L: S,S-diastereoisomer

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Pd++	gl	KCl	25°C	0.10M	U			1996BRa (58203)	429
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K(Pd+2L+2H)=35.01  
 K(Pd+2L)=21.54  
 K(Pd+2L+H)=28.71

H2L: S,R-diastereoisomer

\*\*\*\*\*

C7H20N4 L CAS 4741-99-5 (12)  
1,4,8,11-Tetraazaundecane; H2N.CH2.CH2.NH.CH2.CH2.CH2.NH.CH2.CH2.NH2

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaClO4 25°C 1.00M C K1=46.3 1985YAa (58359) 430

\*\*\*\*\*

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)  
Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C ? M M K1=3.31 1988SKa (59003) 431

K(PdA+L)=3.04

A=diethylenetriamine

\*\*\*\*\*

C8H8NO2Cl HL CAS 61756-69-2 (4569)  
N-Acetyl-N-(4-chlorophenyl)hydroxamine; Cl.C6H4.N(CO.CH3).OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=9.63 B2=17.37 1968JSb (59281) 432

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C8H8O2 HL p-Toluic acid CAS 99-94-5 (1372)  
4-Methylbenzoic acid; CH3.C6H4.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp alc/w 25°C 100% U M 1994PAa (59501) 433

K(Pd3A3CO+L)=3.99

Medium: MeOH. A=Bis(diphenylphosphino)methane

\*\*\*\*\*

C8H9NOS HL CAS 4822-44-0 (3240)  
N-(Mercaptoacetyl)aniline (thioglycolanilide); C6H5.NH.CO.CH2.SH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth diox/w 30°C 70% U B2=24.34 1973BSa (60163) 434

Medium: 0.1 M KCl

\*\*\*\*\*

C8H9NO3 HL CAS 5663-54-7 (1095)  
2,4-Dihydroxy-acetophenone oxime; (HO)2.C6H3.C(CH3):NOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 27°C 60% U I K1=8.99 B2=17.80 1974SRa (60400) 435

\*\*\*\*\*

C8H9NO4S H2L CAS 7717-21-7 (3846)

N-(Phenylsulfonyl)aminoethanoic acid; C6H5SO2NHCH2COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	NaClO4	25°C	0.10M	U			K1=18.9 B2=24.4	1990GBb (60517)	436
*****										
C8H9N3O7		H2L							Uramildiacetic CAS 13055-06-5 (185)	
5-Amino-2,4,6-trioxo-1,3-perhydrodiazimino-N,N-diethanoic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	KCl	20°C	0.10M	U	M			1987KUa (60652)	437
B(PdCl(OH)L)=22.12										
*****										
C8H9O3P		H2L							CAS 1707-08-0 (1969)	
2-Styrylphosphonic acid; C6H5.CH:CH.PO3H2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.12M	U			K1=3.33 B2=6.55	1979RZb (60673)	438
*****										
C8H10		L							p-Xylene CAS 106-42-3 (2145)	
1,4-Dimethylbenzene, 4-Xylene; CH3.C6H4.CH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M			1994PAa (60682)	439
K(Pd3A3CO+L)=-1.15										

Medium: MeOH. A=Bis(diphenylphosphino)methane

*****										
C8H10N3OCl		HL							CAS 5756-79-6 (4578)	
3-Ethyl-3-hydroxy-1-(2-chlorophenyl)triazene;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U			K1=10.49 B2=20.46	1968DSa (60784)	440
Medium: 70% dioxan, 0.1 M KCl										
*****										

C8H10N3OCl		HL							CAS 5756-78-5 (4579)	
3-Ethyl-3-hydroxy-1-(4-chlorophenyl)triazene;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U			K1=10.68 B2=20.66	1968DSa (60789)	441
Medium: 70% dioxan, 0.1 M KCl										
*****										

C8H11N		L							DiMethylaniline CAS 121-69-7 (1343)	
N-Phenyl-N,N-dimethylamine; C6H5.N(CH3)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp non-aq 25°C 100% U M 1979SSa (60989) 442

K(PdA+L)=1.0

A=Tetraphenylporphyrin (in its excited triplet state)

\*\*\*\*\*

C8H11N3O HL CAS 5956-70-7 (4529)

3-Hydroxy-3-methyl-1-(4-tolyl)triazene; CH3.C6H4.N:N.N(OH).CH3  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=11.77 B2=23.10 1970DSb (61244) 443

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C8H11N3O2 HL CAS 5756-72-9 (4533)

3-Hydroxy-3-methyl-1-(4'-methoxyphenyl)triazene; CH3O.C6H4.N:N.N(OH).CH3  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=12.25 B2=23.70 1970DSb (61257) 444

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C8H11N3O3 HL CAS 2497-02-1 (3230)

Acetyl-L-histidine;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C M 2003NFa (61275) 445

K(PdA+L)=8.47

K(PdA+H+L)=11.58

K(2PdA+L=(PdA)2H-1L+H)=6.99

A is bis-((2-pyridyl)methyl)amine  
-----

Pd++ gl KNO3 25°C 0.20M C 1997WKa (61276) 446

K(PdACl+L=PdAL+Cl)=5.33

K(2PdACl+L=Pd2A2H-1L+2Cl)=0.1

PdA is [PdH-1(gly-Met)].

\*\*\*\*\*

C8H14N4O5 HL Tetraglycine CAS 637-84-3 (1849)

Glycyl-Glycyl-Glycyl-Glycine; H2N.CH2.CO.NH.CH2.CO.NH.CH2.CO.NH.CH2.CO.OH  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C 1999AJa (62024) 447

B(PdLC1)=18.25

B(PdH-1LC1)=14.81

B(PdH-2L)=10.13

B(PdH-3L)=2.45

Medium: 0.1 M KNO3, 0.1 M KCl.

\*\*\*\*\*

C8H14O5S2 H2L CAS 4408-66-6 (8332)

Oxybis(ethylenethio)diethanoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 20°C 0.10M U K1=6.22 1977CAc (62135) 448  
\*\*\*\*\*

C8H15NO2 HL (4572)

1-Azacycloheptane-1-ethanoic acid, hexamethyleneimine-ethanoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp none 25°C 0.0 U K1=10.48 B2=20.45 1974HFa (62159) 449  
\*\*\*\*\*

C8H15N3O4 HL (1008)

Glycyl-b-alanyl-b-alanine; H2NCH2CONH(CH2)2CONH(CH2)2COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl oth/un 25°C 0.20M C K1=16.66 2003AMb (62255) 450  
K(PdH-1L)=13.24  
K(PdH-2L)=10.12

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO3/0.10 M KCl.

\*\*\*\*\*

C8H15N3O4 HL (1009)

b-Alanyl-glycyl-b-alanine; H2N(CH2)2CONHCH2CONH(CH2)2COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl oth/un 25°C 0.20M C 2003AMb (62263) 451  
K(PdH-1L)=12.64  
K(PdH-2L)=9.58

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO3/0.10 M KCl.

\*\*\*\*\*

C8H15N7O2S3 L Famotidine CAS 76824-35-6 (6502)

N'-(Aminosulfonyl)-3-((2-(diaminomethyleneamino)-4-thiazolyl)methylthio)propanamid  
ine

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaCl 25°C 0.10M U K1=6.20 B2=12.69 1995CCa (62274) 452  
B(PdH-1L)=1.20  
B(PdHL2)=18.40  
B(PdH-1L2)=6.23

\*\*\*\*\*

C8H16N2O3 HL Gly-Leu CAS 869-19-2 (255)

Glycyl-leucine; H2N.CH2.CO.NH.CH(CH2.CH(CH3)2).COOH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (62393) 453

$$K(\text{Pd}(\text{pn}) + \text{L} = \text{PdH-1}(\text{pn})\text{L} + \text{H}) = 3.30$$

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Pd++      gl    NaNO3    25°C   0.10M   U      M      1999SSd (66011) 460

$K(\text{Pd}(\text{pn})+\text{L})=11.06$

pn is 1,2-diaminopropane. For aminoacid protonation,  $K_1=9.12$ ,  $B_2=11.01$ .  
 \*\*\*\*\*

C9H11NO4S                      H2L                      CAS 1080-44-0 (4682)  
 N-(4-Toluenesulfonyl)glycine, N-tosylglycine; CH3.C6H4.SO2.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	NaClO4	25°C	0.10M	U				1994BGa (66427)	461
									$K_{\text{eff}}(\text{Pd}+\text{H}_2\text{L}=\text{PdL})=19.9$	
									$B_{\text{eff}}(\text{Pd}+2\text{H}_2\text{L}=\text{PdL}_2)=23.3$	

Complex formation involves loss of the amide proton.

Pd++	vlt	NaClO4	25°C	0.10M	U			$K_1=17.8$ $B_2=23.4$	1990GBb (66428)	462
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C9H11NO4S                      H2L                      (6960)  
 N-(Phenylsulfonyl)-2-aminopropanoic acid; C6H5.SO2.NH.CH(CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	NaClO4	25°C	0.10M	U				1994BGa (66429)	463
									$K_{\text{eff}}(\text{Pd}+\text{H}_2\text{L}=\text{PdL})=20.6$	
									$B_{\text{eff}}(\text{Pd}+2\text{H}_2\text{L}=\text{PdL}_2)=23.0$	

Complex formation involves loss of the amide proton.

\*\*\*\*\*

C9H11NO4S                      H2L                      (6961)  
 N-(Phenylsulfonyl)-3-aminopropanoic acid; C6H5.SO2.NH.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	NaClO4	25°C	0.10M	U				1994BGa (66430)	464
									$K_{\text{eff}}(\text{Pd}+\text{H}_2\text{L}=\text{PdL})=17.1$	
									$B_{\text{eff}}(\text{Pd}+2\text{H}_2\text{L}=\text{PdL}_2)=20.8$	

Complex formation involves loss of the amide proton.

\*\*\*\*\*

C9H12N2O6                      HL      Uridine                      CAS 58-96-8 (828)  
 Uracil-1-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.20M	C	M			2003NFa (66703)	465
									$K(\text{PdA}+\text{L})=8.90$	

A is bis-((2-pyridyl)methyl)amine

Pd++	gl	NaNO3	25°C	0.10M	C	M			2002MSb (66704)	466
									$K(\text{PdA}+\text{L})=8.70$	
									$K(\text{PdA}+2\text{L})=14.37$	
									$K(\text{PdA}+\text{H}+\text{L})=15.17$	

A is N,N'-dimethylethylenediamine.

Pd++ gl KNO3 25°C 0.20M C M 2001NSa (66705) 467

$K(\text{Pd}(\text{en})+\text{L})=8.98$   
 $K(\text{Pd}(\text{en})+2\text{L})=14.80$   
 $*K(\text{Pd}(\text{en})(\text{H}_2\text{O})\text{L})=-7.67$

$K(\text{Pd}(\text{en})(\text{H}_2\text{O})+\text{L}=\text{Pd}(\text{en})(\text{OH})\text{L}+\text{H})=1.31$ ,  
 $K(2\text{Pd}(\text{en})(\text{H}_2\text{O})_2+2\text{L}=\text{Pd}_2(\text{en})_2(\text{OH})\text{L}_2+\text{H})=12.14$

Pd++ gl KNO3 25°C 0.20M C M 2001NSa (66706) 468

$K(\text{Pd}(\text{pic})+\text{L})=9.20$   
 $K(\text{Pd}(\text{pic})+2\text{L})=15.09$   
 $*K(\text{Pd}(\text{pic})(\text{H}_2\text{O})\text{L})=-7.94$

$K(\text{Pd}(\text{pic})(\text{H}_2\text{O})+\text{L}=\text{Pd}(\text{pic})(\text{OH})\text{L}+\text{H})=1.26$ ,  
 $K(2\text{Pd}(\text{pic})(\text{H}_2\text{O})_2+2\text{L}=\text{Pd}_2(\text{pic})_2(\text{OH})\text{L}_2+\text{H})=13.82$ . Hpic=picric acid.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (66707) 469

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L})=9.71$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{H}+\text{L})=13.29$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L})=16.88$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+\text{H})=22.65$

Pd++ gl KNO3 25°C 0.20M C K1=7.42 2000NFa (66708) 470

Pd++ gl KCl 25°C 0.20M U M 1997KFa (66709) 471

$K(\text{Pd}(\text{dien})\text{Cl}+\text{L})=7.42$   
 $K(\text{Pd}(\text{terpy})\text{Cl}+\text{L})=7.56$

dien=diethylenetriamine, terpy=2,2'-6',2"-terpyridine. Data also for many related nucleobases

Pd++ gl KNO3 25°C 0.20M C 1997WKa (66710) 472

$K(\text{PdACl}+\text{L}=\text{PdAL}+\text{Cl})=7.00$

PdA is [PdH-1(gly-met)].

Pd++ gl KNO3 25°C 0.50M U M 1981LIa (66711) 473

$K(\text{Pd}(\text{en})(\text{H}_2\text{O})_2+\text{L})=8.65$   
 $K(\text{Pd}(\text{en})(\text{H}_2\text{O})\text{L}+\text{L})=5.92$   
 $K(\text{Pd}(\text{dien})(\text{H}_2\text{O})+\text{L})=8.08$

C9H13N2O9P H3L UMP-5 CAS 58-97-9 (2948)

Uridine-5'-monophosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ nmr oth/un 23°C 0.30M U M 1985PGa (66980) 474

$\text{Keff}(\text{PdA}+\text{HL})=2.99$

A=Tetrakis(4-N-methylpyridyl)porphyrin. pD=7.0

C9H13N3O5 L Cytidine CAS 65-46-3 (2152)

Cytidine, Cytosine-1-beta-D-ribofuranoside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo



-----  
Pd++ gl KNO3 25°C 0.20M C M 2003NFa (67073) 475  
K(PdA+L)=5.83

A is bis-((2-pyridyl)methyl)amine

-----  
Pd++ sp NaClO4 25°C 1.0M U M 1984ETa (67074) 476  
K(PdCl4+L=PdLC13+Cl)=4.49  
K(PdLC13+L=PdL2Cl2+Cl)=3.45  
K(Pd(en)Cl2+L=PdenLC1+Cl)=3.32  
K(Pd(en)LC1+L=PdenL2+Cl)=2.56

\*\*\*\*\*  
C9H14N3O8P H2L CMP-5 CAS 63-37-6 (1243)

Cytidine-5'-monophosphoric acid, Cytidilic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl KNO3 25°C 0.10M C M K1=3.35 2001AAa (67262) 477

Also data for ternary complexes with MOPSO, TAPSO and ACES.

\*\*\*\*\*  
C9H17N3O4S HL Gly-Met-Gly CAS 51529-34-1 (7566)

Glycylmethionylglycine; NH2CH2CONHCH(CH2CH2SCH3)CONHCH2COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl KCl 25°C 0.20M C M 2001BNa (67872) 478

\*K(PdL)=-3.65

K(PdH-1L+Cl)=2.21

\*K(PdH-1L)=-5.57

K(PdH-1L+H+L)=11.49

K(PdH-1L+L)=4.71, K(PdH-1L+glygly)=4.50, K(PdH-1L+L=PdH-2L2+H)=-4.99.

\*\*\*\*\*  
C9H18N2O3 HL Leu-Ala CAS 7298-84-2 (4659)

Leucylalanine- H2N.CH(CH2.CH(CH3)2).CO.NH.CH(CH3).COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (67912) 479

K(Pd(pn)+L)=8.19

K(Pd(pn)+L=PdH-1(pn)L+H)=3.74

pn is 1,2-diaminopropane. For aminoacid protonation, K1=8.13.

\*\*\*\*\*  
C9H23N3 L CAS 3030-47-5 (4605)

N,N,N',N'',N'''-Pentamethyl-diethylenetriamine; (CH3)2NCH2CH2N(CH3)CH2CH2N(CH3)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
Pd++ gl R4N.X 25°C 0.10M C M 1998BBa (68282) 480

B(PdLC1)=24.9

B(PdH-1L)=14.1

Medium: 0.1 M NMe4Cl

-----  
 Pd++ gl NaClO4 25°C 0.50M C I 1981GMf (68283) 481  
 K(PdL=PdLOH+H)=-7.293  
 K(PdL+PdLOH)=1.08

In 0.5 NaNO3, K(PdL=PdLOH+H)=-7.241, K(PdL+PdLOH)=0.70  
 \*\*\*\*\*

C9H24N4 L CAS 129880-56-4 (1533)  
 1,4,10,13-Tetraazatridecane; H2N.(CH2)2.NH.(CH2)5.NH.(CH2)2.NH2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl KNO3 25°C 0.10M C K1=37.9 1985YAa (68336) 482  
 \*\*\*\*\*

C10H7NO2 HL CAS 131-91-9 (2668)  
 1-Nitroso-2-naphthol, alpha-Nitroso-beta-naphthol;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl alc/w RT 40% M K1=5.53 B2= 8.86 1993RAB (68585) 483  
 Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

\*\*\*\*\*

C10H7NO2 HL CAS 132-53-6 (2524)  
 2-Nitroso-1-naphthol;

-----  

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl alc/w RT 40% M K1=4.46 B2= 8.92 1993RAB (68653) 484  
 Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

\*\*\*\*\*

C10H7NO5S H2L CAS 3682-32-4 (1812)  
 2-Nitroso-1-hydroxynaphthalene-4-sulfonic acid;

-----  

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl oth/un RT 0.10M M K1=4.46 B2= 8.68 1993RAB (68892) 485  
 Medium not stated.

\*\*\*\*\*

C10H7NO8S2 H3L Nitroso-R acid CAS 525-05-3 (1811)  
 1-Nitroso-2-hydroxynaphthalene-3,6-disulfonic acid;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp NaClO4 25°C 0.10M U 1964MSa (69024) 486  
 K(?)=8.9

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 Pd++ sp oth/un 25°C ? U 1963BGb (69025) 487  
 K(?)=8.8

\*\*\*\*\*

C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)  
 2,2'-Bipyridine; (C5H4N)2

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       vlt NaClO4 25♦C 0.10M U      M      1994BGa (69635) 488
                                     Beff(Pd(bpy)A)=30.3
                                     Beff(Pd(bpy)HB)=30.7
                                     Beff(Pd(bpy)HC)=30.8
                                     Beff(Pd(bpy)D)=23.2
H2A=N-tosylglycine, H2B=N-phenylsulfonylglycine, H2C=tosyl-alpha-alanine,
H2D=benzoylglycine. Data for other L.
*****
C10H8N2O4      HL      2-Furil dioxime CAS 522-27-0 (3319)
1,2-Di(2'-furyl)ethane-1,2-dione dioxime; (C4H3O.C(:N.OH))2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       dis NaClO4 20♦C 0.10M U      B2=43.7      1967STa (69702) 489
*****
C10H9NO4S      H2L      CAS 116-63-2 (4781)
1-Amino-2-naphthol-4-sulfonic acid;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp oth/un 25♦C ? U      B2=7.90      1968MBa (70187) 490
*****
C10H9N3O3      HL      (1933)
4-(5'-Methyl-3'-isoxazolylazo)-1,3-dihydroxybenzene; (HO)2C6H3.N:N.C3H2NO
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp NaClO4 25♦C 0.10M U      1989TSb (70412) 491
                                     K(PdHL+H)=2.65
                                     K(PdL+H)=7.09
                                     K(PdH-1L+H)=9.57
*****
C10H11N3S      L      CAS 5351-70-2 (4734)
Cinnamaldehyde thiosemicarbazone; C6H5.CH:CH.CH:N.NH.CS.NH2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp alc/w 20♦C 50% U      B2=11.74      1972KLa (71085) 492
Medium: 50% EtOH, 0.1 M, pH=5
*****
C10H12N4O5      HL      Inosine      CAS 58-63-9 (2344)
Hypoxanthine-9-beta-D-ribofuranoside;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl NaClO4 25♦C 0.10M M      M      2002BSa (71392) 493
                                     K(PdA+L)=6.94

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$$K(\text{PdA}+2\text{L})=10.27$$

Pd++ g1 NaNO3 25°C 0.10M C M 2002MSb (71393) 494

$$\begin{aligned} K(\text{PdA}+\text{L}) &= 8.03 \\ K(\text{PdA}+2\text{L}) &= 12.74 \\ K(\text{PdA}+\text{B}+\text{L}) &= 12.29 \\ K(\text{PdA}+\text{H}+\text{B}+\text{L}) &= 17.72 \end{aligned}$$

Pd++      g1    NaNO3    25°C   0.10M   C      M      2001SHc (71394) 495

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2 + \text{L}) = 9.73$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2 + \text{H} + \text{L}) = 12.55$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2 + 2\text{L}) = 14.89$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2 + 2\text{L} + \text{H}) = 20.11$

Pd++ g1 NaNO3 25°C 0.10M U M 1999SSd (71395) 496

$$K(\text{Pd}(\text{pn})+\text{L})=6.83$$
$$K(\text{Pd}(\text{pn})+2\text{L})=11.26$$

Pd++ g1 NaNO3 37°C 0.16M M M 1998ESa (71396) 497

$$K(\text{PdA}+\text{L})=6.92$$
$$K(\text{PdA}+2\text{L})=11.58$$

Pd++ g1 KCl 25°C 0.20M U M 1997KFa (71397) 498

$K(\text{Pd}(\text{dien})\text{Cl}+\text{L})=6.82$   
 $K(\text{Pd}(\text{dien})\text{Cl}+\text{H}+\text{L})=12.79$   
 $K(2\text{Pd}(\text{dien})\text{Cl}+\text{L})=11.56$   
 $K(\text{Pd}(\text{terpy})\text{Cl}+\text{L})=6.92$

Pd++ gl NaClO4 25°C 0.10M C M 1997RSa (71398) 499

$$K(\text{PdA}+\text{L})=7.43$$
$$K(\text{PdA}+2\text{L})=11.77$$

Pd++ g1 KN03 25C 0.20M C 1997Wka (71399) 500

$$\begin{aligned} K(\text{PdACl} + \text{L} &= \text{PdAL} + \text{Cl}) = 6.38 \\ K(\text{PdACl} + \text{H} + \text{L} &= \text{PdAHL} + \text{Cl}) = 12.73 \\ K(2\text{PdACl} + \text{L} &= \text{Pd}_2\text{A}_2\text{L} + 2\text{Cl}) = 10.93 \end{aligned}$$

Pd++ g1 NaCl04 25°C 0.10M M T H 1996SEc (71400) 501

$$\begin{aligned} K(\text{PdACl}_2 + \text{L}) &= 6.04 \\ K(\text{PdACl}_2 + 2\text{L}) &= 9.56 \end{aligned}$$

A is N,N,N',N'-tetramethyl-1,2-diaminoethane. Also data at 15.5, 20, 30 and 35.2 C. DH(PdACl2+L)=17.0 kJ mol<sup>-1</sup>, DH(PdACl2+2L)=-10.7.

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (71401) 502

K(PdACl2+L)=5.78

K(PdACl2+2L)=10.48

A is N,N,N',N'-tetraethyl-1,2-diaminoethane. Also data at 15.5, 20, 30 and 35.2 C. DH(PdACl2+L)=30.6 kJ mol<sup>-1</sup>, DH(PdACl2+2L)=35.6.

\*\*\*\*\*

C10H12N6S L CAS 91262-80-9 (6101)

3-(4',5'-Dimethyl-2'-thiazolylazo)-2,6-diaminopyridine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 0.25M U 1988SSe (71519) 503

B(PdHL)=16.36

B(PdH3L2)=31.47

B(PdH4L2)=27.36

\*\*\*\*\*

C10H13N04S H2L N-Tosylalanine (1584)

N-(4-Toluenesulfonyl)-3-aminopropanoic acid; CH3.C6H4.SO2.NH.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ vlt NaClO4 25°C 0.10M U 1994BGa (71773) 504

Keff(Pd+H2L=PdL)=16.8

Beff(Pd+2H2L=PdL2)=20.5

Complex formation involves loss of the amide proton.

\*\*\*\*\*

C10H13N3OS L (4791)

alpha-Ethylfurylacrolein thiosemicarbazone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 20°C 50% U B2=12.68 1972KLa (71797) 505

Medium: 50% EtOH, 0.1 M, pH=5

\*\*\*\*\*

C10H13N4O8P H3L IMP CAS 131-99-7 (843)

Inosine-5'-monophosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 25°C 0.10M M M 2002BSa (71862) 506

K(PdA+L)=7.45

K(PdA+H+L)=14.10

HA is S-methyl cysteine.

Pd++ gl KNO3 25°C 0.10M C M K1=3.27 2001AAa (71863) 507

Also data for ternary complexes with MOPSO, TAPSO and ACES.

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Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (71864) 508  
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L})=10.17$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{H}+\text{L})=16.65$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+2\text{H})=20.98$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L})=14.80$   
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+\text{H})=21.49$ ,  $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+2\text{H})=28.50$ .

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (71865) 509  
 $K(\text{Pd}(\text{pn})+\text{L})=8.13$   
 $K(\text{Pd}(\text{pn})+2\text{L})=11.92$   
 $K(\text{Pd}(\text{pn})+\text{H}+\text{L})=14.03$   
 pn is 1,2-diaminopropane. For nucleotide protonation,  $K_1=8.67$ ,  $B_2=14.63$ .

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (71866) 510  
 $K(\text{PdA}+\text{L})=9.82$   
 $K(\text{PdA}+2\text{L})=14.82$   
 $K(\text{PdA}+\text{H}+\text{L})=15.14$   
 A is 1,3-diaminopropane.

Pd++ gl NaClO4 25°C 0.10M C M 1997RSa (71867) 511  
 $K(\text{PdA}+\text{L})=10.79$   
 $K(\text{PdA}+\text{H}+\text{L})=17.02$   
 $K(\text{PdA}+2\text{L})=14.65$   
 A=2-(Aminomethyl)pyridine

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (71868) 512  
 $K(\text{PdACl}_2+\text{L})=4.43$   
 $K(\text{PdACl}_2+2\text{L})=9.20$   
 A is N,N,N',N'-tetramethyl-1,2-diaminoethane. Also data at 15, 20, 30 and 35 C.  $\text{DH}(\text{PdACl}_2+\text{L})=-73.1$  kJ mol<sup>-1</sup>,  $\text{DH}(\text{PdACl}_2+2\text{L})=-62.4$ .

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (71869) 513  
 $K(\text{PdACl}_2+\text{L})=4.39$   
 $K(\text{PdACl}_2+2\text{L})=9.73$   
 A is N,N,N',N'-tetraethyl-1,2-diaminoethane. Also data at 15, 20, 30 and 35 C.  $\text{DH}(\text{PdACl}_2+\text{L})=-88.6$  kJ mol<sup>-1</sup>,  $\text{DH}(\text{PdACl}_2+2\text{L})=-2.09$ .

Pd++ sp NaClO4 25°C 0.10M U M 1994SEa (71870) 514  
 $K(\text{PdACl}+\text{L}=\text{PdALCl})=-1.46$

A=N,N,N',N'-Tetramethylethylenediamine  
 \*\*\*\*\*  
 C10H13N5O5 HL Guanosine CAS 118-00-3 (1402)  
 2-Aminopurin-6-one-9-riboside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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 Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (72015) 515  
 $K(\text{PdA}+\text{L})=7.85$

A is 1,3-diaminopropane.  
 \*\*\*\*\*

C10H14N2O5                      H2L      Thymidine                      CAS 50-89-5 (8256)  
 Thymine deoxyriboside, 1-(2-Deoxy-beta-ribofuranosyl)-5-methyluracil;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	C	M			2002MSb (72085)	516
								K(PdA+L)=8.75		
								K(PdA+2L)=14.53		
								K(PdA+B+L)=16.26		

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutanedicarboxylic acid.

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Pd++	gl	NaNO3	25°C	0.10M	U	M			1999SSd (72086)	517
								K(Pd(pn)+L)=8.92		
								K(Pd(pn)+2L)=14.84		

pn is 1,2-diaminopropane. For nucleotide protonation, K1=9.54.

---

Pd++	gl	NaNO3	37°C	0.16M	M	M			1998ESa (72087)	518
								K(PdA+L)=8.27		
								K(PdA+2L)=13.57		

A is 1,3-diaminopropane.

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C10H14N2O6                      L      alpha-Thymidine                      CAS 4449-43-8 (695)  
 Thymine-2-desoxyribofuranosyl-5-methyluracil;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	20°C	1.0M	M			K1=8.15    B2=15.82	1997WYa (72106)	519
								K3=6.37		
								K4=3.56		

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Pd++	gl	KNO3	25°C	0.50M	U	M			1981LIa (72107)	520
								K(Pd(en)(H2O)2+L)=8.84		
								K(Pd(en)(H2O)L+L)=5.85		
								K(Pd(dien)(H2O)+L)=8.31		

\*\*\*\*\*

C10H14N3                      L                      CAS 29198-32-1 (6921)  
 4-Diazo-N,N-diethylaniline; N:N.C6H4.N(C2H5)2

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	alc/w	25°C	100%	U	M			1994PAa (72122)	521
								K(Pd3A3CO+L)=2.30		

Medium: MeOH. A=Bis(diphenylphosphino)methane

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C10H14N5O7P                      H2L      AMP-2                      CAS 81012-86-4 (2437)  
 Adenosine-2'-monophosphoric acid, 2-Adenylic acid;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.10M	C	M			2002WBa (72190)	522
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.10M	C	M		$K(\text{Pd}(\text{dien})+2\text{H}+\text{L})=20.82$ $K(\text{Pd}(\text{dien})+\text{H}+\text{L})=15.40$ $K(\text{Pd}(\text{dien})+\text{L})=7.50$ $K(2\text{Pd}(\text{dien})+\text{H}+\text{L})=19.45$ $K(2\text{Pd}(\text{dien})+\text{L})=13.70$ , $K(3\text{Pd}(\text{dien})+\text{L})=16.59$ .	2002WBa (72525)	523

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaCl04	25°C	0.10M	M			K(PdA+L)=11.96 K(PdA+H+L)=18.75 K(PdA+2H+L)=22.00	2002BSa (72593)	525

Pd++ g1 KN03 25°C 0.10M C M 2002W Ba (72594) 526

K(Pd(en)+4H+2L)=40.96  
K(Pd(en)+2H+2L)=28.8  
K(Pd(en)+2L)=11.7  
K(Pd(en)+H+L)=16.37  
K(Pd(en)+L)=9.83.

Pd++	g1	NaCl04	25♦C	0.10M	C	M		1997RSa (72596)	528
							K(PdA+L)=10.82		
							K(PdA+H+L)=17.35		
							K(PdA+2L)=14.46		

A=2-(Aminomethyl)pyridine



Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (72597) 529  
 K(PdACl2+HL)=4.14  
 K(PdACl2+2HL)=8.03

A is N,N,N',N'-tetramethyl-1,2-diaminoethane. Also data at 15, 20, 30 and 35 C. DH(PdACl2+HL)=-11.8 kJ mol<sup>-1</sup>, DH(PdACl2+2HL)=-14.8.

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (72598) 530  
 K(PdACl2+HL)=4.00  
 K(PdACl2+2HL)=7.14

A is N,N,N',N'-tetraethyl-1,2-diaminoethane. Also data at 15, 20, 30 and 35 C. DH(PdACl2+HL)=-76.2 kJ mol<sup>-1</sup>, DH(PdACl2+2HL)=-111.

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C10H15N5O4 HL Gly-Gly-His CAS 93404-95-6 (74)  
 Glycyl-glycyl-histidine; H2N.CH2.CO.NH.CH2.CO.NH.CH(CH2.C3H3N2).COOH

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Pd++ gl KCl 25°C 0.20M C 1997BCb (72800) 531  
 \*K(PdH-4L)=-11.30

\*K corresponds to deprotonation of coordinated -NH2.

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C10H15N5O10P2 H3L ADP CAS 20398-34-9 (2181)

Adenosine-5'-diphosphoric acid;

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Pd++ nmr oth/un 23°C 0.30M U M 1985PGa (73011) 532  
 Keff(PdA+L)=4.08

A=Tetrakis(4-N-methylpyridyl)porphyrin. pD=7.0

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C10H16N2O8 H4L EDDS CAS 52759-67-8 (1100)

1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH2.NH.CH(COOH)CH2.COOH)2

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Pd++ sp NaClO4 20°C 0.10M U M 1986PKa (73171) 533  
 K(PdCl+H2L)=10.93  
 K(PdCl+L)=23.67

-----  
 Pd++ gl KNO3 30°C 0.10M U K1=13.6 1971STc (73172) 534

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C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestic acid;

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 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Pd++ sp NaClO4 21°C 0.20M U M K1=25.6 1983KDa (74071) 535  
 K(PdL+Cl)=5.4

-----  
 Pd++ sp NaClO4 25°C 1.00M U M 1981SDa (74072) 536

$K(\text{PdL}+\text{Cl})=2.26$   
 $K(\text{PdL}+\text{Br})=2.40$   
 $K(\text{PdL}+\text{I})=2.60$   
 $K(\text{PdL}+\text{SCN})=3.30$   
 $K(\text{PdL}+\text{OH})=4.41$     $K(\text{PdL}+\text{NH}_3)=4.84$     $K(\text{PdL}+\text{S}_2\text{O}_3)=4.66$     $K(\text{PdL}+\text{thiocarbamate})=4.00$

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Pd++	sp	none	25°C	0.0	U		K1=26.4	1978KR	(74073)	537
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Pd++	gl	oth/un	20°C	1.00M	U	I M	K1=24.5	1976AM	a (74074)	538
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$K(\text{PdL}+\text{H})=3.01$   
 $K(\text{PdHL}+\text{H})=3.21$   
 $K(\text{PdH}_2\text{L}+\text{H})=0.09$

Medium: NaBr/NaClO<sub>4</sub>. By exchange with PdBr<sub>4</sub>

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Pd++	oth	NaClO <sub>4</sub>	25°C	0.20M	U		K1=18.5	1955MK	a (74075)	539
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C10H16N6S                      L                      Cimetidine                      CAS 51481-61-9                      (5716)  
 Cimetidine; CH<sub>3</sub>.C<sub>3</sub>H<sub>2</sub>N<sub>2</sub>.CH<sub>2</sub>.S.CH<sub>2</sub>.CH<sub>2</sub>.NH.C(:NCN)NH.CH<sub>3</sub>

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaCl	25°C	0.10M	U			K1=7.63    B2=15.13	1995CC	a (74912) 540
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$B(\text{PdH}-1\text{L})=0.52$   
 $B(\text{PdH}-2\text{L})=-10.95$   
 $B(\text{PdH}-1\text{L}_2)=7.87$   
 $B(\text{PdH}-2\text{L}_2)=-1.18$

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C10H26N<sub>4</sub>                      L                      CAS 66475-54-5                      (5756)  
 3,10-Diazadodecane-1,12-diamine; NH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>.NH.(CH<sub>2</sub>)<sub>6</sub>.NH.CH<sub>2</sub>.CH<sub>2</sub>.NH<sub>2</sub>

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KN <sub>3</sub>	25°C	0.10M	C	I M		K1=38.14	1985YA	a (76766) 541
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C10H28N<sub>2</sub>O<sub>12</sub>P<sub>4</sub>                      H<sub>8</sub>L                      CAS 23605-74-5                      (435)  
 (Hexamethylenedinitrilo)tetra(methylenephosphonic acid);  
 (CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>.N(CH<sub>2</sub>.PO<sub>3</sub>H<sub>2</sub>)<sub>2</sub>)<sub>2</sub>

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KN <sub>3</sub>	25°C	0.10M	U			K1=10.83	1980ZR	b (76841) 542
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$K(\text{PdL}+\text{H})=9.56$   
 $K(\text{PdHL}+\text{H})=6.71$   
 $K(\text{PdH}_2\text{L}+\text{H})=5.73$   
 $K(\text{PdH}_3\text{L}+\text{H})=4.65$

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C11H7NO<sub>4</sub>                      H<sub>2</sub>L                      CAS 122844-38-6                      (8293)  
 1-Hydroxy-4-nitroso-2-naphthalenecarboxylic acid;

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ gl alc/w RT 40% M K1=4.95 B2= 8.72 1993Rab (76893) 543  
 Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

\*\*\*\*\*  
 C11H7NO4 H2L CAS 32446-26-7 (8294)  
 3-Hydroxy-4-nitroso-2-naphthalenecarboxylic acid;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ gl alc/w RT 40% M K1=3.89 B2= 7.86 1993Rab (76901) 544  
 Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

\*\*\*\*\*  
 C11H8O3 H2L CAS 86-48-6 (1129)  
 1-Hydroxy-2-naphthoic acid;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ gl alc/w RT 40% M K1=6.88 B2=13.31 1993Rab (77015) 545  
 Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

\*\*\*\*\*  
 C11H8O3 H2L CAS 92-70-6 (1130)  
 2-Hydroxy-3-naphthoic acid (3-Hydroxy-2-naphthoic acid);  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ gl alc/w RT 40% M K1=11.84 B2=17.04 1993Rab (77128) 546  
 Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

\*\*\*\*\*  
 C11H9NO2S HL CAS 29556-13-6 (1450)  
 N-Phenyl-2-thenoylhydroxamic acid; C4H3SCON(C6H5)OH  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ gl diox/w 25°C 50% M T H K1=9.22 B2=16.97 1977ABb (77351) 547  
 50% v/v dioxan - water; Data also for complexes with Cu(II), Zn, Ni, Co, Mn

\*\*\*\*\*  
 C11H9NO3 H2L CAS 80690-05-7 (872)  
 3-Hydroxy-2-methyl-1,4-naphthoquinone monoxime;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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-----  
 Pd++ gl diox/w 30°C 0.10M U K1=5.63 B2=10.22 1981KSa (77365) 548  
 K3=4.61

\*\*\*\*\*  
 C11H9N3O HL CAS 10335-29-2 (3937)  
 2-(2'-Pyridylazo)phenol; C5H4N.N:N.C6H4.OH  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----  
 Pd++ sp alc/w 20°C 50% U K1=17.1 1967ANa (77461) 549

Medium: 50% MeOH, 0.1 M NaClO4

\*\*\*\*\*

C11H10N4 L PAPHY CAS 2215-33-0 (1305)

Pyridine-2-aldehyde-2'-pyridyl-hydrazone; C5H4N.CH:N.NH.C5H4N

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaCl 25°C 3.00M U M 1981MIb (77710) 550

K(PdCl2+HL=PdHLC1+Cl)=4.00

K(PdLC1+H)=5.30

K(PdCl2+2PdLC1=Pd3L2Cl4)=8.30

\*\*\*\*\*

C11H12N2O L Antipyrine CAS 60-80-0 (2026)

2,3-Dimethyl-1-phenyl-3-pyrazolin-5-one, Phenazone;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un ? 0.60M U K1=5.58 1971KBe (78005) 551

Medium: K2SO4

\*\*\*\*\*

C11H12N2O2 HL Tryptophan CAS 73-22-3 (3)

2-Amino-3-(3-indolyl)propanoic acid; H2N.CH(CH2.C8H6N)COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.50M U 1978LIa (78230) 552

K(Pd(en)+L)=10.83

\*\*\*\*\*

C11H14N2O3 HL Gly-Phe CAS 3321-03-7 (829)

Glycyl-phenylalanine; H2N.CH2.CO.NH.CH(CH2.C6H5).COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C 1999AJa (78815) 553

B(PdLC1)=17.94

B(PdH-1LC1)=16.09

B(PdH-2L)=5.30

B(PdH-1L2)=20.10

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-1L)=14.10.

\*\*\*\*\*

C11H14N2O3 HL Phe-Gly CAS 721-90-4 (830)

Phenylalanyl-glycine; H2N.CH(CH2.C6H5).CO.NH.CH2.COOH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C 1999AJa (78828) 554

B(PdLC1)=17.58

B(PdH-1LC1)=15.20

B(PdH-2L)=4.50

B(PdH-1L2)=19.20

Medium: 0.1 M KNO<sub>3</sub>, 0.1 M KCl. B(PdH-1L)=13.2; B(PdH-2L2)=13.70.

\*\*\*\*\*

C11H18N2O8                      H4L                      CAS 4408-81-5 (923)  
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((H<sub>2</sub>OC.CH<sub>2</sub>)<sub>2</sub>N.CH<sub>2</sub>)<sub>2</sub>.CH<sub>2</sub>

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              gl   oth/un 20°C 1.00M C              K1=28.8              1976AMa (79465) 555

Medium: NaBr/NaClO<sub>4</sub>. By exchange with PdBr<sub>4</sub>

\*\*\*\*\*

C11H20N2                      L                      (6343)

3,5-Dipropyl-4-ethylpyrazole

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              nmr non-aq 32°C 100% U              1987FKa (79696) 556

K(PdCl<sub>2</sub>+2L)=6.7

Medium: Deuterated DMSO (D<sub>6</sub>). With N-methyl analogue: K(PdCl<sub>2</sub>+2L)=9.6;

N-benzyl: K=7.9; N-allyl: K=10.8

\*\*\*\*\*

C12H8N2                      L              Phenanthroline              CAS 66-71-7 (144)

1,10-Phenanthroline;

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              sp   none   25°C   0.0   C              K2=9.4              1975PJb (80508) 557

\*K(PdL)=-3.3

\*K(Pd(OH)L)=-5.9

\*K(Pd(OH)2L)=-9.6

\*\*\*\*\*

C12H10N2O5S                      H3L              Tropeolin 0              CAS 547-57-9 (1090)

Chrysoin; HS<sub>3</sub>.C<sub>6</sub>H<sub>4</sub>.N:N.C<sub>6</sub>H<sub>3</sub>(OH)<sub>2</sub>

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              sp   oth/un 25°C   ?   U              1963SDd (80738) 558

K(Pd+2HL)=9.4(?)

\*\*\*\*\*

C12H10N3OBr                      HL                      CAS 5756-88-7 (4001)

1-(4'-Bromophenyl)-3-hydroxy-3-phenyltriazene;

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

Pd++              gl   diox/w 25°C   70%   U              K1=10.86   B2=21.30   1965PSd (80754) 559

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C12H10N3OCl                      HL                      CAS 52756-05-6 (3998)

1-(2'-Chlorophenyl)-3-hydroxy-3-phenyltriazene;

-----  
Metal              Mtd Medium Temp Conc Cal Flags Lg K values              Reference ExptNo  
-----

-----  
Pd++ gl diox/w 25°C 70% U K1=10.43 B2=20.43 1964PSg (80761) 560  
Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*  
C12H10N3OCl HL CAS 5756-86-5 (3999)  
1-(4'-Chlorophenyl)-3-hydroxy-3-phenyltriazene;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=10.70 B2=20.95 1964PSb (80767) 561  
Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*  
C12H11N3O4S H2L (4003)  
3-Hydroxy-3-phenyl-1-(4'-sulfonyl)triazene;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=9.71 B2=19.03 1964PSf (80942) 562  
Medium: 70% dioxan, 0.1 M KCl

-----  
Pd++ sp oth/un 25°C ? U 1958DSa (80943) 563  
K(?)=11.52

Acetate buffer

\*\*\*\*\*  
C12H12N2S2 HL CAS 1141-88-4 (7739)  
2,2'-Dithiodianiline, 2,2'-Diaminodiphenyl disulfide;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp none C K1=6.17 2000GNa (81110) 564

\*\*\*\*\*  
C12H13N3 L CAS 1539-42-0 (932)  
bis-((2-Pyridyl)methyl)-amine (Di-2-picolylamine); C5H4N.CH2NHCH2.C5H4N  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.20M C 2003NFa (81289) 565  
\*K(PdL(H2O))=-7.08

\*\*\*\*\*  
C12H20N2O8 H4L CAS 40623-42-5 (1101)  
1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH2NHCH(COOH)CH2CH2COOH)2  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 30°C 0.10M U K1=13.4 1971STc (82091) 566

\*\*\*\*\*  
C12H20N2O8 H4L CAS 2458-58-4 (922)  
1,4-Diaminobutane-N,N,N',N'-tetraethanoic acid; (HOOC.CH2)2N.(CH2)4.N(CH2.COOH)2  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

-----  
 Pd++ gl oth/un 20°C 1.00M C K1=25.8 1976AMa (82233) 567  
 Medium: NaBr/NaClO4. By exchange with PdBr4

\*\*\*\*\*  
 C12H24O2S4 L (6657)  
 1,4,7,10-Tetrathia-13,16-dioxacyclooctadecane, 1,4,7,10-Tetrathia-18-crown-6;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ ix none 25°C 0.0 U K1=32.3 1991BTa (83119) 568

\*\*\*\*\*  
 C12H24O4S2 L CAS 296-39-9 (4938)  
 1,4,10,13-Tetraoxa-7,16-dithiacyclooctadecane;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++ cal oth/un 25°C 0.50M M H K1=21.1 1990IWa (83141) 569  
 Medium: 0.5M HNO3. DH(K1)=-82.4 kJ mol-1, DS(K1)= 127.6 J K-1 mol-1.

\*\*\*\*\*  
 C12H24O4S2 L (6528)  
 7,10,13,16-Tetraoxa-1,4-dithiacyclooctadecane;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----

Pd++ cal oth/un 25°C 0.50M M H K1=25.1 1990IWa (83151) 570  
 Medium: 0.5M HNO3. DH(K1)=-184.1 kJ mol-1, DS(K1)= -137 J K-1 mol-1.

\*\*\*\*\*  
 C12H29N3 L CAS 123-12-6 (4904)  
 (N,N,N",N"-Tetraethyl-diethylenetriamine; (C2H5)2N.CH2.CH2.NH.CH2.CH2.N(C2H5)2  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++ gl NaClO4 25°C 0.50M C I 1981GMf (84249) 571  
 K(PdL=PdLOH+H)=-7.688  
 K(PdL+PdLOH)=0.90

In 0.5 NaNO3, K(PdL=PdLOH+H)=-7.677, K(PdL+PdLOH)=0.48

\*\*\*\*\*  
 C12H30N4 L (7251)  
 2,5,8,11-Tetramethyl-2,5,8,11-tetraazadodecane;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++ gl R4N.X 25°C 0.10M C M K1=23.38 1998BBa (84292) 572  
 K(PdL+H+Cl)=6.85  
 B(PdH-1L)=13.9  
 K(PdL+OH)=4.3

Medium: 0.1 M NMe4Cl

\*\*\*\*\*  
 C12H30N4 L (6740)  
 Tris(2-(dimethylamino)ethyl)amine; N(CH2CH2.N(CH3)2)3

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  NaCl   25°C 1.00M U  I      K1=30.5          1993AMa (84303) 573
-----
Pd++       gl  oth/un 25°C 1.00M U   M          1993AMa (84304) 574
                                     K(Pd(H2O)L+Cl=PdClL)=2.6
                                     K(Pd(H2O)L+Br=PdBrL)=2.8
                                     K(Pd(H2O)L+SCN=Pd(SCN)L)=5.57

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*****
C12H30N6          L          CAS 296-35-5 (143)
1,4,7,10,13,16-Hexaazacyclooctadecane; cyclo(-(NH.CH2.CH2)6-)
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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       cal NaCl   25°C 0.50M U  HM          1993BBa (84349) 575
DH(2PdCl4+L=Pd2LC12+6Cl)=-110.8 kJ mol-1
-----
Pd++       gl  NaCl   25°C 0.50M C  H      K1=29.2          1992BBf (84350) 576
                                     B(PdHL)=37.47
                                     B(PdH2L)=42.40
                                     B(Pd2LC12)=51.8

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By calorimetry: DH(PdCl4+H6L)=-6.3 kJ mol-1.
*****
C13H8N2O6Cl2S      H3L          CAS 60743-06-8 (8478)
2-[(3,5-Dichloro-2-hydroxyphenyl)azo]-5-sulfobenzoic acid;
-----

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp  NaClO4  RT  0.10M C          K1=15.51          1978GSc (84477) 577
*****
C13H9NO2BrCl       HL          CAS 104614-71-3 (9109)
4-Bromo-N-(3-chlorophenyl)-N-hydroxybenzamide;
-----

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  diox/w 25°C 50% C   M      K1=11.13          2001AMc (84578) 578
                                     B(Pd(gly)L)=20.61

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```

Medium: 50% v/v dioxane/H2O
*****
C13H9NO2ClF        HL          CAS 104614-72-4 (9107)
N-(3-Chlorophenyl)-4-fluoro-N-hydroxybenzamide;
-----

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  diox/w 25°C 50% C   M      K1=11.34          2001AMc (84586) 579
                                     B(Pd(gly)L)=20.98

```

```

Medium: 50% v/v dioxane/H2O
*****
C13H9NO2Cl2        HL          CAS 67201-86-9 (9108)

```



4-Chloro-N-(3-chlorophenyl)-N-hydroxybenzamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	diox/w	25°C	50%	C	M		K1=11.15 B(Pd(gly)L)=20.58	2001AMc (84594)	580
------	----	--------	------	-----	---	---	--	-------------------------------	-----------------	-----

Medium: 50% v/v dioxane/H2O

\*\*\*\*\*  
C13H9N3O4S2                      H2L                      CAS 2536-61-0 (4031)  
1-(1',3'-Thiazol-2'-ylazo)-2-hydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	gl	alc/w	25°C	50%	U	I		K1=13      B2=19.4	1967NPb (84644)	581
------	----	-------	------	-----	---	---	--	--------------------	-----------------	-----

Medium: 50% MeOH, 0.1 M NaClO4. In 0% MeOH: K1=13, K2=5.7

\*\*\*\*\*  
C13H10N02Cl                      HL                      CAS 36016-24-7 (1818)  
N-(4-Chlorophenyl)benzohydroxamic acid; C6H5.CO.N(C6H4Cl)OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	diox/w	25°C	70%	U			K1=9.80      B2=18.21	1967JSa (84719)	582
------	----	--------	------	-----	---	--	--	-----------------------	-----------------	-----

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*  
C13H10N02Cl                      HL                      CAS 78154-49-1 (5649)  
N-3-Chlorophenylbenzohydroxamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	gl	diox/w	25°C	50%	C	M		K1=11.47 B(Pd(gly)L)=21.29	2001AMc (84742)	583
------	----	--------	------	-----	---	---	--	-------------------------------	-----------------	-----

Medium: 50% v/v dioxane/H2O

\*\*\*\*\*  
C13H10N2O2                      HL                      CAS 56288-80-1 (4980)  
2-Hydroxy-4-(phenylazo)benzaldehyde; C6H5.N:N.C6H3(OH).CHO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	alc/w	30°C	50%	U			B2=7.64	1972DTb (84840)	584
------	----	-------	------	-----	---	--	--	---------	-----------------	-----

\*\*\*\*\*  
C13H10N2O5S                      H2L                      CAS 98789-35-6 (5012)  
4-Hydroxy-3-formylazobenzene-4'-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	sp	oth/un	30°C	aq	U			B2=7.22	1972DTb (84923)	585
------	----	--------	------	----	---	--	--	---------	-----------------	-----

\*\*\*\*\*  
C13H10N4O4S                      H2L                      (6644)  
4-Hydroxy-3-(1H-imidazol-2-ylazo)-2-naphtalenesuphonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.50M	U		K1=15.53	1992VMa (84960)	586

For -3-ylazo analogue: K1=10.22; for 3,3-bis(1H-pyrazol-3-ylazo) analogue: K1=10.99

\*\*\*\*\*

C13H10N4S	HL						CAS 3788-81-6	(4014)	
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2-Picolinyaldehyde 2-benzothiazolyldiazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	U		K1=10.33	1965HRa (84967)	587

\*\*\*\*\*

C13H10O2S	H2L						CAS 88220-26-2	(6572)	
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3-(1-Naphthyl)-2-mercaptopropenoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.10M	C		K1=15.56 B2=26.40	1989IBb (84976)	588

Medium: Aqueous 0.1 M NaClO4 containing 1-2% EtOH.

\*\*\*\*\*

C13H11NO2	HL						CAS 304-88-1	(181)	
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N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	EMF	diox/w	25°C	70%	U		K1=10.11 B2=18.85	1967JSb (85171)	589

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C13H11N3O6S	H2L						(2811)		
-------------	-----	--	--	--	--	--	--------	--	--

1-(2-Carboxy-5-sulfonatophenyl)-3-hydroxy-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	none	25°C	0.0	U		K1=10.398	1974CHa (85304)	590

\*\*\*\*\*

C13H12N2S	HL						CAS 156873-11-9	(8362)	
-----------	----	--	--	--	--	--	-----------------	--------	--

2-[[1-(2-Pyridinyl)ethylidene]amino]benzene thiol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	dis	NaCl	25°C	2.0M	C			1998BMD (85391)	591

K(Pd+2HL(org)=PdL2(org)+2H)=2.9. Method: extraction into CHCl3.

\*\*\*\*\*

C13H12N4S	L	Dithizone					CAS 60-10-6	(1801)	
-----------	---	-----------	--	--	--	--	-------------	--------	--

Diphenylthiocarbazone; C6H5.NH.NH.CS.N:N.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.10M	U		K1=11.39 B2=21.78	1973BSe (85471)	592

\*\*\*\*\*

C13H13N3O HL (4018)

3-Hydroxy-1-(2'-methylphenyl)-3-phenyltriazene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KCl 25°C 0.10M U K1=11.70 B2=22.97 1964PSa (85508) 593

\*\*\*\*\*

C13H13N3O HL CAS 5756-83-2 (4019)

3-Hydroxy-1-(4'-methylphenyl)-3-phenyltriazene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KCl 25°C 0.10M U K1=11.89 B2=23.35 1964PSa (85514) 594

\*\*\*\*\*

C13H13N3O2 HL CAS 5756-89-8 (4021)

3-Hydroxy-1-(4'-methoxyphenyl)-3-phenyltriazene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=12.06 B2=23.74 1965PSb (85522) 595

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C13H20N2O2 L Procaine CAS 59-46-1 (4029)

2-(Diethylamino)ethyl 4-aminobenzoate; H2N.C6H4.CO2.CH2.CH2.N(C2H5)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un 25°C ? U B2=7.88 1968SPd (86097) 596

\*\*\*\*\*

C13H22N2O8 H4L CAS 1798-14-7 (921)

(Pentamethylenedinitrilo)tetraethanoic acid; ((HOOCH2)2N.CH2.CH2)2CH2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl oth/un 20°C 1.00M C K1=26.4 1976AMa (86203) 597

Medium: NaBr/NaClO4. Corrected for PdBrx complexes

\*\*\*\*\*

C13H22N4O3S L Ranitidine CAS 66357-35-5 (7144)

N(2-(5-Dimethylaminomethyl)-2-furanylmethyl)thioethyl-N-methyl-2-nitro-1-ethenediamine;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl NaCl 25°C 0.10M U K1=9.97 1995CCa (86332) 598

B(PdH-1L)=2.41

B(PdH-2L)=-6.88

\*\*\*\*\*

C13H26O4S2 L (6656)

1,5-Dithia-8,11,14,17-tetraoxacyclononadecane, 1,5-Dithia-19-crown-6;

```

-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       ix  none   25°C  0.0  U          K1=29.8          1991BTa (86462) 599
*****
C13H32N4          L                      (7403)
2,5,9,12-Tetramethyl-2,5,9,12-tetraazatridecane;
-----

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  R4N.X  25°C  0.10M C          K1=28.3          1998BBa (86579) 600
                      K(PdL+H+Cl)=3.6
                      B(PdH-1L)=16.9
                      K(PdL+OH)=2.4

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Medium: 0.1 M Me4NCl

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*****
C14H9NO3          HL                      CAS 116-85-8 (1020)
1-Amino-4-hydroxyanthraquinone;
-----

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp  alc/w  20°C  50%  U          K1=14.21  B2=23.66  1990ISa (86796) 601
                      K(Pd+HL)=5.71
                      K(Pd(OH)2L)=31.81

```

Medium: 50% EtOH/H2O, 0.1 M NaClO4

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*****
C14H10O4          H2L                      CAS 482-05-3 (8247)
Diphenyl-2,2'-dicarboxylic acid; diphenic acid;
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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  diox/w 30°C  50%  U T H    K1=7.00   B2=13.24  1978SJc (86933) 602
Medium: 50% dioxane/H2O, 0.10 M NaClO4. At 40 C, K1=6.52, K2=5.52.
DH and DS values reported.

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*****
C14H12Cl2S2          L                      CAS 33451-44-4 (5055)
1,2-Bis(4-chlorophenylthio)ethane; Cl.C6H4.S.CH2.CH2.S.C6H4.Cl
-----

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp  alc/w  25°C  100% U    M          1969CCb (87034) 603
                      K(PdI4+L=PdLI2+2I)=-0.68

```

```

*****
C14H12NO2Cl          HL                      CAS 67055-92-9 (6301)
N-(3-Chlorophenyl)-4-methylbenzohydroxamic acid; CH3.C6H4.CO.N(C6H4Cl)OH
-----

```

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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  diox/w 25°C  50%  C    M    K1=11.65          2001AMc (87066) 604
                      B(Pd(gly)L)=21.66

```

Medium: 50% v/v dioxane/H2O

-----  
Pd++ gl diox/w 25°C 50% U K1=9.94 B2=18.79 1989PMb (87067) 605  
-----

Pd++ gl diox/w 25°C 50% U K1=10.05 B2=19.15 1989PMb (87068) 606

Data also for 4-fluoro, 4-chloro, 4-bromo, 4-nitro and 4-methoxy analogues

\*\*\*\*\*

C14H12N03Cl HL CAS 67135-47-1 (9106)

N-(3-Chlorophenyl)-N-hydroxy-4-methoxybenzamide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 50% C M K1=11.82 2001AMc (87097) 607  
B(Pd(gly)L)=22.06

Medium: 50% v/v dioxane/H2O

\*\*\*\*\*

C14H12N4O2Br2 HL CAS 72833-87-5 (2533)

2-(2-(3,5-Dibromopyridyl)azo)-5-dimethylaminobenzoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp diox/w 25°C 40% C K1=10.36 1986KHa (87319) 608

\*\*\*\*\*

C14H13N02 HL CAS 1503-92-0 (1817)

N-(4-Tolyl)benzohydroxamic acid; C6H5.CO.N(C6H4.CH3).OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=10.34 B2=19.19 1969JSa (87450) 609

\*\*\*\*\*

C14H13N02 HL CAS 1143-74-2 (4044)

N-2-Tolylbenzohydroxamic acid; C6H5.CO.N(C6H4.CH3).OH

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ oth diox/w 25°C 70% U K1=19.02 1968JSc (87482) 610

\*\*\*\*\*

C14H13N3O2 HL (4045)

1-(4'-Acetylphenyl)-3-hydroxy-3-phenyltriazene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 70% U K1=10.97 B2=21.51 1964PSe (87594) 611

Medium: 70% dioxan, 0.1 M KCl

\*\*\*\*\*

C14H13N5O5 HL (5394)

1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)thiourea;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp mixed 25°C 40% U 1985RGa (87617) 612  
K1eff=5.57

Medium: 40% DMF, pH 4.5

\*\*\*\*\*

C14H13N5O2 HL (5393)  
1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)urea;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp mixed 25°C 32% U 1985RGa (87624) 613  
K1eff=5.38

Medium: 32% DMF, pH 4.5

\*\*\*\*\*

C14H14N4OBr2 HL CAS 35601-32-2 (5092)  
5-(3,5-Dibromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un ? ? U K1=6.94 1967GUa (87688) 614

\*\*\*\*\*

C14H14S2 L CAS 42311-15-9 (5031)  
1,2-Bis(phenylthio)ethane; C6H5.S.CH2.CH2.S.C6H5

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp alc/w 25°C 100% U M 1969CCb (87708) 615  
K(PdI4+L=PdLI2+2I)=0.72

\*\*\*\*\*

C14H15N4OBr HL CAS 14337-50-9 (5095)  
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un ? ? U 1967GUa (87768) 616  
K(?)=7.35

\*\*\*\*\*

C14H16N2O2S2 L CAS 729600-10-6 (9255)  
2,3,5,6,8,9-Hexahydro[1,4,7,10]dioxadithiacyclododecino[2,3-b]quinoxaline;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ nmr mixed 25°C 60% C K1=4.20 2004HHa (87879) 617  
Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.

\*\*\*\*\*

C14H16N2O2S2 L CAS 729600-11-7 (9256)  
2,3,5,6,8,9-Hexahydro[1,4,7,10]dioxadithiacyclododecino[8,9-b]quinoxaline;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ nmr mixed 25°C 60% C K1=3.61 2004HHa (87880) 618

Method: <sup>1</sup>H nmr. Medium: 60% CD<sub>2</sub>Cl<sub>2</sub>/CD<sub>3</sub>CN.

\*\*\*\*\*

C<sub>14</sub>H<sub>16</sub>N<sub>2</sub>O<sub>4</sub>S                      H<sub>2</sub>L              Dansyl-Gly                      CAS 1091-85-6 (5845)

N-Dansylglycine, (5-Dimethylamino)naphthalene-1-sulfonoglycine;

(CH<sub>3</sub>)<sub>2</sub>N.C<sub>10</sub>H<sub>6</sub>.SO<sub>2</sub>.NH.CH<sub>2</sub>.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	NaClO <sub>4</sub>	25°C	0.10M	U			K <sub>1</sub> =17.8    B <sub>2</sub> =21.8 B(PdL(OH))=21.6 B <sub>eff</sub> (PdH-2L <sub>2</sub> )=21.8 B <sub>eff</sub> (PdH-2L <sub>2</sub> (OH))=21.6	1990GBb (87901)	619

\*\*\*\*\*

C<sub>14</sub>H<sub>18</sub>N<sub>4</sub>                                      L              DPEN                                      CAS 4608-34-3 (1850)

N,N'-Bis-(2-pyridylmethyl)-1,2-diaminoethane; (C<sub>5</sub>H<sub>4</sub>N.CH<sub>2</sub>.NH.CH<sub>2</sub>)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	1.00M	C			K <sub>1</sub> =35.6	1985YAa (88117)	620

Medium: NaBr

\*\*\*\*\*

C<sub>14</sub>H<sub>23</sub>N<sub>3</sub>O<sub>10</sub>                                      H<sub>5</sub>L              DTPA                                      CAS 67-43-6 (238)

Diethylenetriamine-pentaethanoic acid; HOOC.CH<sub>2</sub>.N(CH<sub>2</sub>.CH<sub>2</sub>.N(CH<sub>2</sub>.COOH)<sub>2</sub>)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO <sub>4</sub>	20°C	1.00M	U	M		K <sub>1</sub> =29.7 K(PdL+H)=3.49 K(PdHL+H)=2.93 K(PdH <sub>2</sub> L+H)=2.56 K(PdH <sub>3</sub> L+H)=1.93	1976AMa (89356)	621

K(PdL+SCN=PdL(SCN))=1.45; K(PdL+Br=PdBr)=-1.K<sub>1</sub> in NaBr by exchange with PdBr<sub>4</sub>

Pd++	EMF	oth/un	25°C	0.20M	U			K <sub>1</sub> =24.60	1972KIa (89357)	622
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C<sub>14</sub>H<sub>24</sub>N<sub>2</sub>O<sub>8</sub>                                      H<sub>4</sub>L              HMDTA                                      CAS 1633-00-7 (920)

1,6-Diaminohexane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH<sub>2</sub>)<sub>2</sub>N.CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>)<sub>2</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	NaClO <sub>4</sub>	20°C	0.10M	U	I		K(PdCl+HL)=17.3 B(PdClL(OH))=39.72 K(PdClL+OH)=15.60 K(PdClL+2OH)=28.35	1983KV a (89596)	623

B(Pd(OH)ClL)=43.72; B(Pd<sub>2</sub>Cl<sub>2</sub>(OH)<sub>2</sub>L)=57.43. Data also at 1.0 M

Pd++	gl	oth/un	20°C	1.00M	C			K <sub>1</sub> =26.3	1976AMa (89597)	624
------	----	--------	------	-------	---	--	--	----------------------	-----------------	-----

Medium: NaBr/NaClO<sub>4</sub>. By exchange with PdBr<sub>4</sub>

\*\*\*\*\*

C14H26N2O8                      H2L                      (6658)  
1,4,10,13-Tetraoxa-7,16-diaza-2,3-dicarboxycyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	gl	R4N.X	25°C	0.10M	U			K1=8.5 B(PdHL)=15.6	1990AFa (90224)	625
------	----	-------	------	-------	---	--	--	------------------------	-----------------	-----

\*\*\*\*\*  
C14H28N6O4                      H2L                      (832)  
N,N,N',N'-Tetrakis(2-carbamoylethyl)diaminoethane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	gl	NaClO4	25°C	0.10M	C			K(Pd+H2L)=11.24 K(Pd+H2L=PdHL+H)=8.35 K(Pd+H2L=PdL+2H)=4.37 K(PdH2L=PdHL+H)=-2.89	1986HPa (90505)	626
------	----	--------	------	-------	---	--	--	--	-----------------	-----

K(PdHL=PdL+H)=-3.98

\*\*\*\*\*  
C14H34N4                      L                      (7402)  
2,6,9,13-Tetramethyl-2,6,9,13-tetraazatetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	gl	R4N.X	25°C	0.10M	C			K1=22.95 K(PdL+H+Cl)=9.30 B(PdH-1L)=13.96 K(PdL+OH)=4.82 K(PdClHL+H+Cl=PdCl2H2L)=4.52	1998BBa (90834)	627
------	----	-------	------	-------	---	--	--	---	-----------------	-----

Medium: 0.1 M Me4NCl

\*\*\*\*\*  
C14H35N7                      L                      CAS 296-85-5 (9052)  
1,4,7,10,13,16,19-Heptaazacycloheneicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	gl	NaCl	25°C	0.50M	C	H		K1=24.55 B(PdHL)=34.92 B(PdH2L)=42.63 B(PdH3L)=47.13 B(Pd2LC1)=>52	1992BBf (90857)	628
------	----	------	------	-------	---	---	--	--	-----------------	-----

By calorimetry: DH(PdCl4+H7L)=-6.3 kJ mol<sup>-1</sup>.

\*\*\*\*\*  
C14H37N7                      L                      CAS 298-85-5 (5606)  
1,4,7,10,13,16,19-Heptaazacycloheneicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Pd++	cal	NaCl	25°C	0.50M	U	HM			1993BBa (90917)	629
------	-----	------	------	-------	---	----	--	--	-----------------	-----



DH(2PdCl<sub>4</sub>+L=Pd<sub>2</sub>LCl<sub>2</sub>+6Cl)=-119.2 kJ mol<sup>-1</sup>

\*\*\*\*\*

C<sub>15</sub>H<sub>10</sub>N<sub>3</sub>OBr HL (5128)

4-(5-Bromo-2-pyridylazo)-1-hydroxynaphthalene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ dis NaCl ? ? U 1967GVc (90944) 630

K(Pd+HL=PdL+H)=7.05

\*\*\*\*\*

C<sub>15</sub>H<sub>12</sub>N<sub>4</sub> L (4056)

2-Picolinaldehyde 2'-quinolyldiazone; C<sub>5</sub>H<sub>4</sub>N.CH:N.NH.C<sub>9</sub>H<sub>6</sub>N

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 50% U K<sub>1</sub>=10.57 1965HRa (91454) 631

\*\*\*\*\*

C<sub>15</sub>H<sub>14</sub>N<sub>3</sub>Cl HL CAS 113581-14-9 (9105)

N-(3-Chlorophenyl)-4-ethoxy-N-hydroxybenzamide;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 25°C 50% C M K<sub>1</sub>=11.91 2001AMc (91706) 632

B(Pd(gly)L)=22.34

Medium: 50% v/v dioxane/H<sub>2</sub>O

\*\*\*\*\*

C<sub>15</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub> HL CAS 7397-15-1 (6853)

Peonolphenylhydrazone;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl diox/w 20°C 75% U T K<sub>1</sub>=13.27 B<sub>2</sub>=26.05 1991NNA (91927) 633

30 C: K<sub>1</sub>=13.08, K<sub>2</sub>=12.42; 40 C: K<sub>1</sub>=12.92, K<sub>2</sub>=12.38

\*\*\*\*\*

C<sub>15</sub>H<sub>16</sub>N<sub>4</sub>OBr<sub>2</sub> HL CAS 14337-54-3 (993)

2-(3,5-Dibromo-2-pyridylazo)-5-diethylaminophenol;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un ? ? U 1967GVb (91942) 634

K(Pd+HL=PdL+H)=6.3

\*\*\*\*\*

C<sub>15</sub>H<sub>16</sub>S<sub>2</sub> L CAS 42837-97-3 (5105)

1,3-Bis(phenylthio)propane; C<sub>6</sub>H<sub>5</sub>.S.CH<sub>2</sub>.CH<sub>2</sub>.CH<sub>2</sub>.S.C<sub>6</sub>H<sub>5</sub>

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp alc/w 25°C 100% U M 1969CCa (91967) 635

K(PdI<sub>4</sub>+L=PdLI<sub>2</sub>+2I)=-1.18

Medium: CH<sub>3</sub>OH.

\*\*\*\*\*

C15H17N4OBr HL CAS 14357-53-2 (712)  
2-(5-Bromo-2-pyridylazo)-5-diethylaminophenol; BrC5H3N.N:N.C6H3(OH)N(CH3)2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un ? ? U 1967GVb (91982) 636  
K(Pd+HL=PdL+H)=7.0

\*\*\*\*\*

C15H18N2O2S2 L CAS 729600-13-9 (9258)  
2,3,6,7,9,10-Hexahydro-5H-[1,4,7,11]dioxadithiocyclotridecino[2,3-b]quinoxalene;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ nmr mixed 25°C 60% C K1=3.53 2004HHa (92008) 637  
Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.

\*\*\*\*\*

C15H18N4O HL CAS 14337-52-1 (5124)  
5-Diethylamino-2-(2-pyridylazo)phenol;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un ? ? U 1967GVa (92098) 638  
K(?)=6.0

\*\*\*\*\*

C15H20N4 L DPTN CAS 63671-70-5 (1851)  
N,N'-Bis-(2-pyridylmethyl)-1,3-diaminopropane; (C5H4N.CH2.NH.CH2)2CH2

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un 25°C 1.00M C K1=39.1 1985YAA (92185) 639  
Medium: NaBr

\*\*\*\*\*

C15H25N3O10 H5L (5127)  
Diethylenetriamine-N,N,N'',N''-tetraethanoic acid-N'-propanoic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ dis NaCl ? ? U 1967GVc (92380) 640  
K(Pd+HL=PdL+H)=6.57

\*\*\*\*\*

C15H37N5 L CAS 3803-11-2 (1798)  
2,5,8,11,14-Pentamethyl-2,5,8,11,14-pentaazapentadecane;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl R4N.X 25°C 0.10M C M K1=21.41 1998BBa (92627) 641  
K(PdL+H)=8.68  
K(PdHL+H+Cl)=5.76  
B(PdH-1L)=10.95

$$K(\text{PdL}+\text{OH})=3.4$$

Medium: 0.1 M NMe<sub>4</sub>Cl

\*\*\*\*\*

C16H11N2O7ClS2                      H3L                      CAS 4768-88-1 (7743)  
4-Chloro-phenylazo-R-acid, 1-(4-Chlorophenylazo)-2-naphthol-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		C			K1eff=3.75 B2eff=8.55	1999HAa (92772)	642

Medium: Universal Buffer, pH 6.0

\*\*\*\*\*

C16H12N2O4S                      H2L                      CAS 13964-82-4 (3475)  
1-(4-Sulfophenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		? U			B2eff=9.8 (pH 4)	1968SDa (93003)	643

\*\*\*\*\*

C16H12N2O11S3                      H5L                      CAS 548-81-2 (5180)  
2-(4'-Sulfophenylazo)chromotropic acid,  
2-(4-sulfophenylazo)-1,8-dihydroxyaphthalene-3,6-diHSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	dil	C			B2eff=10.0 (pH 3.5) B3eff=15.0 (pH 10.5)	1985SSg (93099)	644

Medium: dilute buffer solution (not stated).

\*\*\*\*\*

C16H13N04S                      HL                      (5182)  
N-4-Toluenesulfonyl-benzofur-2-yl-carboxamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	EMF	alc/w		? 70%	U			B2=6.10	1971MSc (93160)	645

Medium: 70% MeOH

\*\*\*\*\*

C16H13N2O10AsS2                      H5L                      Thorin I                      CAS 3688-92-4 (2609)  
1-((2-Arsonophenyl)azo)-2-hydroxy-3,6-naphthalylldisulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		? U			K1eff=4.4 (pH 3)	1964SDd (93205)	646

\*\*\*\*\*

C16H14N4O2S                      HL                      CAS 83688-78-2 (2534)  
2-(2-Benzothiazolylazo)-5-dimethylaminobenzoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	sp	diox/w	25°C	40%	C		K1=9.37	1986KHa (93483)	647
------	----	--------	------	-----	---	--	---------	-----------------	-----

\*\*\*\*\*  
 C16H18S2 L (5144)  
 1,2-Bis(3-tolylthio)ethane; CH3.C6H4.S.CH2.CH2.S.C6H4.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (93901)	648
------	----	-------	------	------	---	---	--	-----------------	-----

K(PdI4+L=PdLI2+2I)=1.23

Medium: MeOH

\*\*\*\*\*  
 C16H18S2 L (5145)  
 1,2-Bis(4-tolylthio)ethane; CH3.C6H4.S.CH2.CH2.S.C6H4.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (93902)	649
------	----	-------	------	------	---	---	--	-----------------	-----

K(PdI4+L=PdLI2+2I)=1.96

\*\*\*\*\*  
 C16H22N4 L DPTE CAS 81747-99-1 (1852)  
 N,N-Bis-(2-pyridyl-methyl)-1,4-diaminobutane; (C5H4N.CH2.NH.CH2.CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	sp	oth/un	25°C	1.00M	C		K1=37.0	1985YAA (94183)	650
------	----	--------	------	-------	---	--	---------	-----------------	-----

Medium: NaBr

\*\*\*\*\*  
 C16H24N6O5 L Pro-Gly-Ala-His (7404)  
 Prolyl-glycyl-alanyl-histidine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	gl	KNO3	25°C	0.20M	U		K1=15.72 B(PdHL)=17.58 B(PdH-1L)=11.95	1997THa (94338)	651
------	----	------	------	-------	---	--	--	-----------------	-----

Results confirmed by H nmr measurements.

\*\*\*\*\*  
 C16H26N2O12 H4L (6659)  
 1,4,10,13-Tetraoxa-7,16-diaza-2,3,11,12-tetracarboxycyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++	gl	R4N.X	25°C	0.10M	U		K1=12.1 B(PdHL)=18.4	1990AFa (94591)	652
------	----	-------	------	-------	---	--	-------------------------	-----------------	-----

\*\*\*\*\*  
 C16H26N2O12 H4L CAS 130190-52-2 (6660)  
 1,4,10,13-Tetraoxa-7,16-diaza-2,3,7,16-tetracarboxycyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	U		K1=14.1 B(PdHL)=20.0	1990AFa (94605)	653
*****									
C16H29N3O8		H3L		(6699)					
1,7-Dioxa-4,10,13-triazacyclopentadecane-N,N',N''-triethanoic acid;									
-----									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KCl	25°C	0.10M	C		K1=16.58 K(PdL+H)=4.88 K(PdHL+H)=2.18 B(Pd2L)=19.82 K(Pd(OH)L+H)=10.77	1993DSa (94977)	654
*****									
C16H40N8		L		CAS 297-11-0 (5588)					
1,4,7,10,13,16,19,22-Octaazacyclotetracosane;									
-----									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	cal	NaCl	25°C	0.50M	U	HM		1993BBa (95661)	655
DH(2PdCl4+L=Pd2LC12+6Cl)=-118.8 kJ mol-1									
*****									
C17H16N4O2S		HL		CAS 202867-34-3 (7313)					
2-[2-(5-Methylbenzothiazolyl)azo]-5-dimethylaminobenzoic acid;									
-----									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	RT	16%	C			1998FZa (96110)	656
							K1eff=6.88		
Medium: 16% EtOH/H2O.									
*****									
C17H20S2		L		(5209)					
1,3-Bis(3-tolylthio)propane; CH3.C6H4.S.CH2.CH2.CH2.S.C6H4.CH3									
-----									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (96359)	657
							K(PdI4+L=PdLI2+2I)=-1.04		
Medium: CH3OH.									
*****									
C17H20S2		L		(5210)					
1,3-Bis(4-tolylthio)propane; CH3.C6H4.S.CH2.CH2.CH2.S.C6H4.CH3									
-----									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (96360)	658
							K(PdI4+L=PdLI2+2I)=-0.77		

\*\*\*\*\*

C17H24N4 L CAS 49764-71-3 (5757)  
N,N'-Bis((2-pyridyl)methyl)-1,5-pentanediamine; C5H4N.CH2.NH.(CH2)5.NH.CH2.C5H4N

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un 25°C 1.00M C I M K1=34.7 1985YAA (96436) 659  
Medium: NaBr. Ternary complex with Br-

\*\*\*\*\*

C18H15O3PS HL CAS 16704-71-5 (3365)  
3-Diphenylphosphino-benzene sulfonic acid;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ ISE NaClO4 25°C 1.0M U K1=10.2 B2=20.00 1972CBa (97110) 660  
K3=6.3  
K4=4.9

\*\*\*\*\*

C18H15P L CAS 603-35-0 (621)  
Triphenylphosphine; (C6H5)3P

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ vlt non-aq 20°C 100% C 1998ACd (97144) 661  
K(Pd(Ph)L2+Cl)=4.28  
K(Pd(Ph)L2+Br)=3.56  
K(Pd(Ph)L2+I)=3.23  
K(Pd(Ph)L2+acetate)=2.88

Medium: DMF. Method: chronoamperometry.

-----  
Pd++ kin non-aq 25°C 100% U T HM 1988JHc (97145) 662  
K(PdABL+L=PdAL2+B)=2.39

Medium: acetonitrile. A=2,5-dioxo-3,6-dichloro-1,4-benzoquinone, B=CH3CN  
Data also at 30, 35, 40, 45 C

\*\*\*\*\*

C18H18N2O2S2 L CAS 729600-12-8 (9257)  
2,3,5,6,8,9-Hexahydrobenzo[g][1,4,7,10]dioxadithiacyclododecino[2,3-b]quinoxaline;

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ nmr mixed 25°C 60% C K1=3.82 2004HHa (97229) 663  
Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.

\*\*\*\*\*

C18H26N4 L CAS 80284-81-7 (5758)  
N,N'-Bis((2-pyridyl)methyl)-1,6-hexanediamine; C5H4N.CH2.NH.(CH2)6.NH.CH2.C5H4N

-----  
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un 25°C 1.00M C K1=34.7 1985YAA (97679) 664  
Medium: KBr. K is only a limiting value

\*\*\*\*\*  
 C18H30N4O12                      H6L        TTHA                      CAS 869-52-3    (694)  
 Triethylenetetraaminehexaethanoic acid;((HOOCH2)2NCH2CH2N(CH2COOH)CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.5M	C			K1=18.73 K(PdL+H)=6.92 K(PdH2L+H)=2.50 K(PdHL+H)=2.90 K(PdH3L+H)=2.45	1984NAb (98082)	665

K(2Pd+L)=27.50;    K(Pd2L+H)=3.20  
 K(Pd2HL+H)=2.0

\*\*\*\*\*  
 C18H40N4O4                      L                      CAS 89066-60-2    (867)  
 N,N',N'',N'''-Tetrakis(2-hydroxyethyl)-1,4,8,11-tetraazacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.10M	C			K1=18.32	1987HPa (98924)	666

\*\*\*\*\*

C18H44N6                      L                      (7252)  
 2,5,8,11,14,17-Hexamethyl-2,5,8,11,14,17-hexaazaoctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C	M			1998BBa (98955)	667

B(PdHL)=30.83  
 K(PdHL+H)=7.16  
 B(PdH-1L)=10.75  
 B(Pd2LC12)=44.0

Medium: 0.1 M NMe4Cl.    B(Pd2H-1LC1)=33.9,    K(Pd2LC12+H+Cl)=5.8

\*\*\*\*\*

C19H14O7S                      H4L        Pyrocatechol Vi    CAS 369596-29-2    (709)  
 Pyrocatechol Violet,  
 3-[3,4-Dihydroxyphenyl-3-hydroxy-4-oxo-2,5-cyclohexadien-1-ylidenemethyl-b.;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	U				1997USa (99112)	668

K(Pd+H2L)=8.29  
 K(Pd+HL)=13.67  
 K(PdL+OH)=3.60  
 K(2Pd+HL=Pd2L+H)=15.91

K(Pd2L+OH)=8.25.

\*\*\*\*\*

C19H24N2OS                      L                      (2547)  
 10-(3-Dimethylamine-2-methyl-propyl)-2-methoxyphenothiazine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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 Pd++ sp KCl 25°C 1.00M U K1=4.32 1978J0a (99349) 669  
 \*\*\*\*\*

C21H17N2O3P L CAS 215457-01-5 (8001)

Diphenyl-3-(4-methoxyphenylsydnonyl)phosphine;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----

Pd++ sp non-aq 25°C 100% C 2001LPb (101075) 670

K(PdLCl2+benzylamine)=-0.49  
 K(PdLCl2+dibenzylamine)=-0.38  
 K(PdLCl2+diethylamine)=-0.44  
 K(PdLCl2+triethylamine)=-0.69

Medium: CH2Cl2. Also data for dimethylamine, 2-aminopyridine, 4-anisidine, pyridine, 4-toluidine and aniline.

\*\*\*\*\*

C21H18N4O6S H2L CAS 86170-15-2 (8412)

2-[5-(2-Methoxy-5-sulfophenyl)-3-phenyl-1-formazano]-benzoic acid;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++ sp NaClO4 26°C 0.10M C K1=14.90 1983UCa (101119) 671

For the ligand, K1=14.4, K2=3.6.

\*\*\*\*\*

C21H21P L CAS 6163-58-2 (600)

Tri(2-methylphenyl)phosphine (or 4-methyl where indicated); (CH3.C6H4)3P  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++ sp non-aq 25°C 100% U TIHM 1981MKa (101193) 672

K(PdA2+L)=3.14

Medium: benzene. HA = trifluoroacetylacetone

\*\*\*\*\*

C21H22N4O HL CAS 56932-30-0 (5308)

1-Hydroxy-2-(2-N-methylanabasinyl-alpha-azo)naphthalene;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

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Pd++ sp oth/un ? ? U B2=10.53 1966APa (101203) 673

\*\*\*\*\*

C22H26N3OF3S L Fluphenazine CAS 146-56-5 (2548)

10-[3]-4-(2-Hydroxyethyl)piperazine-1-yl-propyl-2-trifluoromethylphenothiazine;  
 -----

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

-----

Pd++ sp KCl 25°C 1.00M U K1=5.13 1978J0a (101926) 674

\*\*\*\*\*

C23H16O9Cl2S H4L Chrome azuro1 S CAS 1667-99-8 (711)

Chromazuro1 S;  
 -----



Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	KCl	25°C	0.10M	C		K1=6.36	1975ISa (102565)	675
Pd++	sp	NaNO3	25°C	0.10M	U		B(Pd2L)=9.80 B(Pd2L2)=15.27 K(Pd+HL)=4.90	1972MSd (102566)	676
Pd++	sp	oth/un	25°C	?	U		K(?)=4.8	1963SDc (102567)	677
*****									
C23H18O9S	H4L	Eriochrome cyan	CAS 3564-18-9	(433)					
4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchson-5,5'-dicarboxylic acid;									
*****									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	?	U		K2eff=5.0 (pH=4.5)	1970SMD (102634)	678
*****									
C23H31N3O4	H2L		(7088)						
1,4,7-Trimethyl-1,7-bis(4-carboxybenzyl)-1,4,7-triazaheptane;									
CH3N(CH2CH2N(CH3)CH2C6H4COOH)2									
*****									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl	25°C	0.15M	C	M	B(PdLC1)=19.10 B(PdHLC1)=23.15 K(PdLC1+H)=4.05	1995BBc (102773)	679
*****									
C24H23N9O2	HL		(5330)						
1,5-Bis(4-antipyrinyl)-3-cyanoformazan;									
*****									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.10M	U		K1=28.1	1971BSf (102934)	680
*****									
C26H25N9O9S	H4L	Semi-Xylenol O	(426)						
3-(N,N-Di(carboxymethyl)aminomethyl)-2-cresolsulfonephthalein;									
*****									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	KNO3	25°C	0.10M	C	I	B(Pd2L)=26.62	1991HKg (103948)	681
*****									
C27H29N011	L	Adriamycin	CAS 25316-40-9	(2407)					
Doxorubicin;									
*****									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

-----  
Pd++ gl oth/un 25°C 0.10M U 1986FGa (104461) 682  
K(Pd+HL=0.5(PdL)2)=22.1

Medium not stated.

\*\*\*\*\*

C30H50N6 L (7089)  
1,4,7,16,19,22-Hexamethyl-1,4,7,16,19,22-hexaaza[9.9]paracyclophane;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl R4N.X 25°C 0.10M C 1999BBd (105354) 683  
B(PdH2LC1)=37.44  
B(Pd2LC12)=42.9  
B(Pd2HLC13)=47.3

Medium: NMe4Cl. Additional method: 1H and 13C nmr.

\*\*\*\*\*

C31H32N2O13S H6L Xylenol orange CAS 63721-85-5 (432)  
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sul  
fonic acid;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ sp oth/un 25°C ? U 19630Ta (105488) 684  
K(?)=10.3

\*\*\*\*\*

C32H44N10O4 L CAS 702699-42-1 (9126)  
2,9-Di[4-(1,4,7,10-tetraazacyclotridecane-11,13,-dione)methyl]-1,10-phenanthroline  
;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl KNO3 25°C 0.10M U 2004GLa (105774) 685  
B(PdH2L)=19.82  
B(Pd2L)=15.83  
B(Pd3H-2L)=9.93  
B(Pd3H-3L)=-3.52

B(Pd3H-4L)=-13.72.

\*\*\*\*\*

C35H57N5O4 L CAS 160320-59-2 (7393)  
1,4,7-Trimethyl-19,22,28,31-tetraoxa-1,4,7,12,23-pentaaza[9.25]-4-cyclophane;  
-----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
-----

Pd++ gl R4N.X 25°C 0.10M C 1999BBd (106212) 686  
B(PdLC1)=23.7  
B(PdHLC1)=31.1  
B(PdH2LC1)=36.7  
K(PdLC1+H)=7.4

Medium: NMe4Cl. Additional method: 1H and 13C nmr. K(PdHLC1+H)=5.6.

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#### EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
 I Data with various BACKGROUNDS  
 H Data for THERMOCHEMICAL quantities  
 M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

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 END Experiments recorded for

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