

## Uranium(IV)

Reaction	Baes and Mesmer, 1976	Thoenen et al., 2014	Brown and Ekberg, 2016	Grenthe et al., 2020
$\text{U}^{4+} + \text{H}_2\text{O} \rightleftharpoons \text{UOH}^{3+} + \text{H}^+$	$-0.65 \pm 0.04$	$-0.54 \pm 0.06$	$-0.58 \pm 0.08$	$-0.54 \pm 0.06$
$\text{U}^{4+} + 2 \text{H}_2\text{O} \rightleftharpoons \text{U}(\text{OH})_2^{2+} + 2 \text{H}^+$	(-2.6)	$-1.1 \pm 1.0$	$-1.4 \pm 0.2$	$-1.9 \pm 0.2$
$\text{U}^{4+} + 3 \text{H}_2\text{O} \rightleftharpoons \text{U}(\text{OH})_3^+ + 3 \text{H}^+$	(-5.8)	$-4.7 \pm 1.0$	$-5.1 \pm 0.3$	$-5.2 \pm 0.4$
$\text{U}^{4+} + 4 \text{H}_2\text{O} \rightleftharpoons \text{U}(\text{OH})_4 + 4 \text{H}^+$	(-10.3)	$-10.0 \pm 1.4$	$-10.4 \pm 0.5$	$-10.0 \pm 1.4$
$\text{U}^{4+} + 5 \text{H}_2\text{O} \rightleftharpoons \text{U}(\text{OH})_5^- + 5 \text{H}^+$	-16.0			
$\text{UO}_2(\text{am, hyd}) + 4 \text{H}^+ \rightleftharpoons \text{U}^{4+} + 2 \text{H}_2\text{O}$		$1.5 \pm 1.0$		

$\text{UO}_2(\text{am,hyd}) + 2 \text{H}_2\text{O} \rightleftharpoons \text{U}^{4+} + 4 \text{OH}^-$			$-54.500 \pm 1.000$	
$\text{UO}_2(\text{c}) + 4 \text{H}^+ \rightleftharpoons \text{U}^{4+} + 2 \text{H}_2\text{O}$	-1.8			
$\text{UO}_2(\text{c}) + 2 \text{H}_2\text{O} \rightleftharpoons \text{U}^{4+} + 4 \text{OH}^-$				$-60.860 \pm 1.000$

C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976.

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 336–349.

I. Grenthe, X. Gaona, A.V. Plyasunov, L. Rao, W.H. Runde, B. Grambow, R.J.M. Konings, A.L. Smith and E.E. Moore, Second Update on the Chemical Thermodynamics of Uranium, Neptunium, Plutonium, Americium and Technetium, OECD Pub., 2020.

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