

START Experiments recorded for
 from SC-Database on Saturday, 01 January, 2000 at 00:45:09
 Software version = 5.81 Data version = 4.62
 Experiment list contains 153 experiments for
 (no ligands specified)
 Metal : Ge(IV)
 (no references specified)
 (no experimental details specified)

e- HL Electron (442)
 Electron;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) kin oth/un 25°C 1.00M U 1965REa (513) 1

K(Ge(IV) + 2e = Ge++)=0
 K(Ge(IV)+4e=Ge(s))=8.38, 124mV

Medium: H2SO4

 Ge(IV) EMF none 25°C 0.0 U 1959LBa (514) 2

K=-4.0(brown GeO, -118 mV)
 K=-9.2(yellow GeO, -273 mV)
 K(Ge(II)+2e=Ge(s))=7.81(231mV)

K: GeO2(s,hex)+2H+2e=GeO(s)+H2O. K(H2GeO3+4H+4e=Ge(s)+3H2O)=0.7(11 mV)

Br- HL Bromide CAS 10035-10-6 (19)
 Bromide;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) dis oth/un 25°C ? C 1991S0a (1943) 3

K(GeMe(OH)+H+L=GeMeL)=-2.57
 K(GeMe(OH)2+2H+2L=GeMeL2)=-4.3
 K(GeMe(OH)3+3H+3L=GeMeL3)=-4.7
 K(GeMe2(OH)+H+L=GeMe2L)=-1.59

K(GeMe2(OH)2+2H+2L=GeMe2L2)=-3.49

Cl- HL Chloride CAS 7647-01-0 (50)
 Chloride;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) dis oth/un 25°C ? C 1991S0a (4931) 4

K(Ge(OH)+H+L=GeL(H2O))=-3.02
 K(Ge(OH)2+2H+2L=GeL2)=-3.84
 K(Ge(OH)3+3H+3L=GeL3)=-4.82
 K(Ge(OH)4+4H+4L=GeL4)=-5.09

K(MeGe(OH)+H+L=MeGeL)=-2.31; K(MeGe(OH)2+2H+2L=MeGe(OH)2)=-2.95; K(MeGe(OH)3+3H+3L=MeGeL3)=-3.81; K(Me2Ge(OH)+H+L=Me2GeL)=-0.71; K(Me2Ge(OH)2+2H+2L)=-2.2

 Ge(IV) sp oth/un ? var U 1961ADb (4932) 5
 K5K6=-5.06

 F- HL Fluoride CAS 7644-39-3 (201)
 Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) ISE NaClO4 25°C 3.00M U I 1990CIc (6930) 6
 *B(2,-1)=2.9
 *B(4,0)=7.18
 *B(4,1)=6.65
 *B(6,1)=9.94

*B(6,2)=9.59. *B(p,q): $\text{Ge(OH)}_4 + \text{pHF} = \text{Ge(OH)}_x\text{F}_p + \text{qH} + 4-x \text{H}_2\text{O}$. Data also
 in 3.0 M LiClO4

Ge(IV) dis oth/un 20°C ? U K1=1.68 B2=3.03 1979NVa (6931) 7
 B3=4.18
 B4=5.17
 B5=6.07
 B6=7.24

Ge(IV) ix oth/un ? ? U K6=3.21 1972PAb (6932) 8

Ge(IV) ix KCl ? 0.50M U K6=3.86 1968PMF (6933) 9

Ge(IV) ISE oth/un 25°C var U T K=-30.9 1965RKa (6934) 10
 K(GeF6+2H2O=4H+6F+GeO2)=-25.8

K: $\text{K}_2\text{GeF}_6(\text{s}) + 2\text{H}_2\text{O} = 2\text{K} + 4\text{H} + 6\text{F} + \text{GeO}_2(\text{s, hex})$

Ge(IV) oth NaCl 50°C 0.40M U T H K(GeF5H2O+HF=GeF6+H3O)=0.34 1964RKb (6935) 11

Method: chemical analysis. K=0.66(0 C), 0.62(10 C), 0.58(20 C), 0.50(30 C),
 0.42(40 C). At 25 C: DH(K)=10.8 kJ mol⁻¹, DS=-26.3 J K⁻¹ mol⁻¹

Ge(IV) oth oth/un 25°C dil U T K(GeF4H2O+HF+F=GeF6)=5.3 1964RKc (6936) 12

Method: chemical analysis, quinhydrone electrode. At 0 C: K=5.9

Ge(IV) EMF NaClO4 25°C 0.50M U K(Ge(OH)4+4HF)=7.30 1963BPb (6937) 13
 K(Ge(OH)4+5HF)=8.94

 MoO4-- H2L Molybdate (443)
 Molybdate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	oth/un	?	?	U			1960KRb (8735)	14
K(H ₄ GeO ₄ +4H ₂ Mo ₃ O ₁₀ =H ₄ GeMo ₁₂ O ₄₀ +4H ₂ O)? =12.86 (pH 2.40)									

OH-		HL						(57)	
Hydroxide;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	gl	NaClO ₄	25°C	0.10M	C			2000KAa (11543)	15
							K(Ge(OH) ₄ =GeO(OH) ₃ +H)=-9.16		
Ge(IV)	gl	oth/un	25°C	0.0	C	T		1998PSb (11544)	16
							K(GeO ₂ (s)+2H ₂ O=Ge(OH) ₄)=-5.02		
							K(GeO(OH) ₃ +H=Ge(OH) ₄)=9.32		
Method: solubility of GeO ₂ (tetr) in dil KOH, 21-90 C. Also solubility data for GeO ₂ at pH 1.5-10 at 25-350 C.									

Ge(IV)	sol	NaCl	25°C	0.10M	C			1998PSc (11545)	17
							Ks(GeO ₂ +2H ₂ O=Ge(OH) ₄)=-1.38		
Method: solubility of GeO ₂ (hex) in NaCl.									

Ge(IV)	sol	none	RT	0.0	C			1990DEa (11546)	18
							Ks(Ge(OH) ₄ +2H)= -19.26		
							K(4Ge(OH) ₄ (s)+GeO ₂ (OH) ₂)=13.15		
K: 4Ge(OH) ₄ (s)+GeO ₂ (OH) ₂ =Ge ₅ O ₁₁ +9H ₂ O									

Ge(IV)	sp	KN ₃	25°C	0.10M	U	I	K ₁ =14.18 B ₂ =27.98	1968NFa (11547)	19
							B ₃ =41.52		
							B ₄ =54.81		
K ₁ =13.73, B ₂ =29.28, B ₃ =43.47, B ₄ =56.98(I=1). Also when I=0.3, 0.5									

Ge(IV)	dis	oth/un	25°C		U		K ₁ =14.78 B ₂ =29.18	1966ANa (11548)	20
							B ₃ =43.32		
							B ₄ =56.85		
Medium: LiCl									

Ge(IV)	sol	oth/un	25°C	var	U			1964GZa (11549)	21
							*Ks(Ge(OH) ₂ (s)+H=GeOH)=-1.26		
							K(GeOH+H=Ge(II))=-1.7 ?		

Ge as Ge++ ?

O ₂ --		H ₂ L					CAS 7772-84-1	(2813)	
Peroxide; -0.0-									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	U			1960ARb (12665)	22
							K(GeO(OH) ₃ +2H ₂ L)=1.68		

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Medium: H_2SO_4

CH4O	L	Methyl alcohol	CAS 67-56-1	(597)
Methanol; CH3.OH				

Method: H electrode. Medium: MeOH, 1.0 M Me₄NC₁

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH2.COOH

C2H6OS HL CAS 60-24-2 (841)
2-Mercaptoethanol; HS.CH2.CH2.OH

C2H6O2 L Ethyleneglycol CAS 107-21-1 (924)
1,2-Dihydroxyethane (Ethane-1,2-diol); HO.CH2.CH2.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	EMF	KCl	25°C	0.10M	U				1959ANa (22147)	28

$$K(\text{HGeO}_3 + 2\text{L}) = -0.37$$

$$K(\text{Ge}(\text{OH})_2 + \text{H}_3\text{L}) = 2.52$$

$$K(\text{Ge}(\text{OH})_2 + 2\text{L}) = 8.81$$

$$K(\text{Ge}(\text{OH})_2 + 2\text{L}) = 0.46$$
$$K(\text{H}_2\text{GeO}_3 + 2\text{HL}) = 0.6(?)$$

$$K(\text{H}_2\text{GeO}_3 + 2\text{HL}) = 1.9(?)$$

$$K(\text{HGeO}_3 + \text{L}) = 0.28$$
$$K(\text{HGeO}_3 + 2\text{L}) = 0.06$$

$K(\text{HGeO}_3 + \text{L}) = 1.21$

$K(\text{HGeO}_3 + 2\text{L}) = 1.94$

$K(\text{HGeO}_3 + 2\text{L}) = 1.105 - 1.700\sqrt{I}$

C3H12N09P3 H6L NTPA CAS 6419-19-8 (2920)

Nitrilotris(methylenephosphonic acid); $\text{N}(\text{CH}_2\text{PO}_3\text{H}_2)_3$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV)	sp	KN03	20°C	0.10M	U				1984SBa (28570)	36
--------	----	------	------	-------	---	--	--	--	-----------------	----

$K(\text{Ge} + \text{H}_2\text{L}) = 13.64$

C4H6O5 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; $\text{HOOC}.\text{CH}_2.\text{CH}(\text{OH}).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV)	con	NaCl	18°C	1.0M	U				1957VAa (30639)	37
--------	-----	------	------	------	---	--	--	--	-----------------	----

$K(\text{H}_2\text{GeO}_3 + 2\text{H}_2\text{L}) = 0.68$

Ge(IV)	gl	oth/un	18°C	0.0	U				1957VAa (30640)	38
--------	----	--------	------	-----	---	--	--	--	-----------------	----

$K(\text{H}_2\text{GeO}_3 + 2\text{H}_2\text{L}) = 2.92$

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)

L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; $\text{HOOC}.\text{CH}(\text{OH}).\text{CH}(\text{OH}).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV)	sp	NaNO3	25°C	0.10M	U				1973BPa (31266)	39
--------	----	-------	------	-------	---	--	--	--	-----------------	----

$K(\text{Ge}(\text{OH})_2 + \text{H}_2\text{L}) = 4.13$

Ge(IV)	gl	NaCl	18°C	1.0M	U				1957VAa (31267)	40
--------	----	------	------	------	---	--	--	--	-----------------	----

$K(\text{H}_2\text{GeO}_3 + \text{HL}) = 5.2$

C4H10O2 L Butanediol CAS 26171-83-5 (3574)

Butanediol (1,2-/1,3-/1,4- etc not stated)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV)	EMF	KCl	25°C	0.10M	U				1959ANa (34667)	41
--------	-----	-----	------	-------	---	--	--	--	-----------------	----

$K(\text{HGeO}_3 + \text{L}) = 0.64$

$K(\text{HGeO}_3 + 2\text{L}) = 0.04$

Method: quinhydrone electrode.

C4H10O3 L CAS 623-39-2 (3577)

3-Methoxypropan-1,2-diol; $\text{CH}_2(\text{OH}).\text{CH}(\text{OH}).\text{CH}_2.\text{OCH}_3$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV)	oth	KCl	25°C	0.10M	U				1959ANa (34707)	42
--------	-----	-----	------	-------	---	--	--	--	-----------------	----

$K(\text{HGeO}_3 + \text{L} = \text{GeO}_2\text{H} - 2\text{L}) = 0.84$
 $K(\text{HGeO}_3 + 2\text{L} = \text{HGeO}(\text{H} - 2\text{L})_2) = 0.58$

Method: quinhydrone electrode

C4H11NO8P2 H5L CAS 2439-99-8 (2129)
 N-Carboxymethyl-N,N-bis(methylenephosphonic acid); $\text{HOOC} \cdot \text{CH}_2 \cdot \text{N}(\text{CH}_2 \cdot \text{PO}_3\text{H}_2)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U				1988SBb (35109)	43
									$K(\text{Ge} + \text{HL}) = 17.1$	

Phosphate buffer pH=6

Ge(IV)	sp	KNO3	20°C	0.10M	U				1986SBb (35110)	44
									$K(\text{Ge}(\text{OH})_2 + \text{H}_2\text{L}) = 4.18$	

C5H4O3 HL Pyromeconic aci CAS 496-63-9 (3600)
 3-Hydroxy-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U				1967CBb (36272)	45
									$K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 2.86$	

C5H5N L Pyridine CAS 110-86-1 (31)
 Pyridine, Azine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	cal	non-aq	25°C	100%	U	H			1967M0b (36638)	46
Medium: n-hexane. $\text{DH}(\text{GeF}_4(\text{l}) + 2\text{L}(\text{l}) = \text{GeF}_4\text{L}_2(\text{c})) = -202.3 \text{ kJ mol}^{-1}$, $\text{DH}(\text{GeF}_4(\text{g}) + 2\text{L}(\text{l}) = \text{GeF}_4\text{L}_2(\text{c})) = -224$; $\text{DH}(\text{GeCl}_4(\text{g}) + 2\text{L}(\text{l}) = \text{GeCl}_4\text{L}_2(\text{c})) = -207$. Plus others										

C5H10N07P H4L PMIDA CAS 5994-61-6 (2433)
 N-(Phosphonomethyl)iminodiethanoic acid; $\text{H}_2\text{O}_3\text{P} \cdot \text{CH}_2 \cdot \text{N}(\text{CH}_2 \cdot \text{COOH})_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U				1988SBb (39676)	47
									$K(\text{Ge}(\text{OH}) + \text{HL}) = 10.4$	

Phosphate buffer pH 6

Ge(IV)	sp	KNO3	20°C	0.10M	U				1986SBb (39677)	48
									$K(\text{Ge}(\text{OH})_2 + \text{HL}) = 6.48$	

C5H10O5S2 HL CAS 110-50-9 (591)
 (Butoxy)dithiomethanoic acid; $\text{CH}_3 \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{OSSH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV) sp KNO3 20°C 0.10M U I 1982SGc (40158) 49
K(Ge(OH)2+2L)=8.82

C5H10O4 L Deoxy-Ribose CAS 533-67-5 (7470)
2-Deoxy-D-ribose, 2-Deoxy-D-erythro-pentose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1979HUa (40327) 50
K(H2GeO3+L)=3.44

C5H10O5 L D-Arabinose CAS 10323-20-3 (3606)
D-Arabinose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1959ATa (40335) 51
K(HGeO3+2L=HGeO(H-2L)2)=3.52

C5H10O5 L D-Xylose CAS 58-86-6 (3607)
D-Xylose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1959ATa (40362) 52
K(HGeO3+2L=HGeO(H-2L)2)=3.38

C5H10O5 L L-Arabinose CAS 5328-37-0 (1616)
L-Arabinose

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1959ATa (40370) 53
K(HGeO3+2L=HGeO(H-2L)2)=3.63

C6H03Cl3 HL CAS 69173-78-0 (3668)
Trichlorohydroxy-1,4-benzoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1966BBb (42031) 54
K(Ge(OH)4+2HL) < 1.4

C6H2N2O8 H2L Nitroanilic aci CAS 479-22-1 (3669)
3,6-Dinitro-2,5-dihydroxy-1,4-benzoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaClO4 25°C 3.0M U 1967BBa (42034) 55
K(Ge(OH)4+2HL=Ge(OH)2L2)=4.9

Medium: HCl

$$K(Ge(OH)_3I + HI) = 6.30$$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			1967BBa (42333)	63
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.35									

C6H5NO4		H2L		3-Nitrocatechol		CAS 6665-98-1		(2685)	
1,2-Dihydroxy-3-nitrobenzene; O2N.C6H3(OH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	NaClO4	25°C	0.10M	U			1970NLc (42861)	64
B3=59.59									

C6H5NO4		H2L		4-Nitrocatechol		CAS 3316-09-4		(890)	
1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	KCl	25°C	0.10M	U			1967PBd (42928)	65
K(Ge(OH)4+3H2L=GeL3+2H)=3.90									

C6H5O2Cl		H2L		4-Cl-Catechol		CAS 2138-22-9		(1656)	
1,2-Dihydroxy-4-chlorobenzene; Cl.C6H3(OH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.10M	U			1967PBd (43082)	66
K(Ge(OH)4+3H2L=GeL3+2H)=0.65									

C6H5O4Cl		HL		Chlorokojic aci		(3086)			
3-Chloro-5-hydroxy-2-hydroxymethyl-4-pyrone;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			1967CBb (43133)	67
K(Ge(OH)4+2HL=Ge(OH)2L2)=2.33									

C6H5O4I		HL		Iodokojic acid		CAS 40838-33-3		(3681)	
3-Iodo-5-hydroxy-2-hydroxymethyl-4-pyrone;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			1967CBb (43143)	68
K(Ge(OH)4+2HL=Ge(OH)2L2)=2.49									

C6H6O2		H2L		Catechol		CAS 120-80-9		(534)	
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

Ge(IV) gl NaCl 25°C 0.10M M 1998PSc (43768) 69
 $K(\text{Ge}(\text{OH})_4 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H} + 4\text{H}_2\text{O}) = -1.39$
 Method: solubility of $\text{GeO}_2(\text{hex})$ in 0.1 m NaCl/0.01-0.05 m H_2L .

Ge(IV) gl oth/un 25°C 0.0 U 1963ANc (43769) 70
 $K(\text{HGeO}_3 + 3\text{H}_2\text{L} = \text{HGeL}_3) = 8.67$

Ge(IV) gl KCl 25°C 0.10M U 1959AMa (43770) 71
 $K(\text{H}_2\text{GeO}_3 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H}) = -0.77$

 C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)
 1,2,3-Trihydroxybenzene; $\text{C}_6\text{H}_3(\text{OH})_3$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl oth/un 25°C 0.0 U 1963ANc (43961) 72
 $K(\text{HGeO}_3 + 3\text{H}_3\text{L} = \text{HGe}(\text{HL})_2) = 9.05$

Ge(IV) gl KCl 25°C 0.10M U 1959AMa (43962) 73
 $K(\text{H}_2\text{GeO}_3 + 3\text{H}_3\text{L} = \text{Ge}(\text{HL})_3 + 2\text{H}) = -0.22$

 C6H6O3 HL Maltol CAS 118-71-8 (2442)
 3-Hydroxy-2-methyl-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KNO_3 20°C 0.10M C 1979MBf (44088) 74
 $K(\text{GeO}_2 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 4.2$
 $K(\text{GeO}_2 + 3\text{HL} + \text{H} = \text{GeL}_3 + 2\text{H}_2\text{O}) = 8.3$

Ge(IV) sp NaCl 25°C 0.50M U 1966BBb (44089) 75
 $K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 3.90$
 $K(\text{Ge}(\text{OH})_4 + 3\text{HL} + \text{H} = \text{GeL}_3) = 8.05$

 C6H6O3 HL Allomaltol CAS 644-46-2 (2688)
 5-Hydroxy-2-methyl-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967CBb (44127) 76
 $K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 3.43$

 C6H6O4 HL Kojic acid CAS 501-30-4 (1800)
 5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KNO_3 20°C 0.10M C 1979MBf (44219) 77
 $K(\text{GeO}_2 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 3.2$
 $K(\text{GeO}_2 + 3\text{HL} + \text{H} = \text{GeL}_3 + 2\text{H}_2\text{O}) = 6.0$

```

-----
Ge(IV)      sp  NaCl   25°C 0.50M U                      1967CBb (44220) 78
                                     K(Ge(OH)4+2HL=Ge(OH)2L2)=2.81
*****
C6H6O8S2          H4L    Tiron          CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ge(IV)      sp  NaCl   25°C 1.00M U  I                      1967PBd (44453) 79
                                     K(Ge(OH)4+2H2L=GeL2)=3.89
                                     K' (Ge(OH)4+3H2L=GeL3+2H)=3.70
K=2.30(I=0.11), 3.10(I=0.26), 3.50(I=0.50)
-----
Ge(IV)      gl  KCl    25°C var U                      1966ATc (44454) 80
K(Ge(OH)4+3H2L=GeL3+2H)=-2.307+27.49SQRTI/(1+2.851SQRTI)-0.370I
-----
Ge(IV)      gl  KCl    25°C 0.10M U                      1959Ama (44455) 81
                                     K(Ge(OH)4+3H2L=GeL3+2H)=-2.74
*****
C6H8O7          H3L    Citric acid      CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ge(IV)      gl  NaCl   25°C 0.10M M                      1998PSc (46126) 82
K(Ge(OH)4+2H3L=Ge(OH)2(H2L)2+2H2O)=6.2, K(Ge(OH)4+H2L=Ge(OH)3HL+H2O)=
2.4. Method: solubility of GeO2(hex) in 0.1 m NaCl/0.02 m H3L.
-----
Ge(IV)      sp  NaNO3  25°C 0.10M U                      1973BP a (46127) 83
                                     K(Ge(OH)2+H3L)=2.01 pH 1-2
*****
C6H9NO6          H3L    NTA          CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ge(IV)      sp  KNO3   20°C 0.10M U                      1988SBb (46843) 84
                                     K(Ge(OH)2+HL)=4.06
Phosphate buffer pH=6
-----
Ge(IV)      sp  KNO3   20°C 0.10M U                      1986SBb (46844) 85
                                     K(Ge(OH)2+HL)=3.82
-----
Ge(IV)      gl  KNO3   20°C 0.10M U                      1981MMe (46845) 86
                                     K(GeO2+H2L=Ge(OH)2L)=4.42
*****
C6H10O7          HL     Glucuronic acid CAS 6556-12-3 (599)
D-Glucuronic acid;
-----

```

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=1.70	1986HPb (48420)	87

C6H11NO5		H2L		HIMDA				CAS 93-62-9	(192)	
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2										
Ge(IV)	sp	KN03	20°C	0.10M	U			K(Ge(OH)2+L)=8.42	1988SBb (48739)	88
Phosphate buffer pH=6										

C6H12O5		HL						CAS 123-97-7	(6144)	
Pentoxydithiomethanoic acid; C5H11.0.C(S)SH										
Ge(IV)	sp	KN03	20°C	0.10M	U	I		K(Ge(OH)2+2L)=8.72	1982SGc (49411)	89

C6H12O5		L		L-Rhamnose				CAS 634-74-2	(3659)	
6-Deoxy-L-mannose;										
Ge(IV)	gl	KCl	25°C	0.10M	U			K(HGeO3+2L=HGeO(H-2L)2)=3.24	1959ATa (49507)	90

C6H12O6		L		D-Fructose				CAS 57-48-7	(1561)	
D-Fructose										
Ge(IV)	gl	NaCl	25°C	0.10M	M			K(GeO(OH)3+2L=Ge(OH)(H-2L)2+3H2O)=5.48.	1998PSc (49548)	91
2.4. Method: solubility of GeO2(hex) in 0.1 m NaCl/0.02 m L.										
Ge(IV)	gl	KCl	25°C	var	U	I		K(HGeO3+2L)=4.273+1.155SQRTI	1963NFa (49549)	92

Ge(IV)	gl	KCl	25°C	0.10M	U			K(HGeO3+2L=HGeO(H-2L)2)=5.48	1958ANa (49550)	93

C6H12O6		L		D-Galactose				CAS 59-23-4	(1559)	
D-Galactose										
Ge(IV)	gl	KCl	25°C	var	U	I			1963NFa (49565)	94

$$K(\text{HGeO}_3+2\text{L})=2.117+1.297\text{SQRTI}$$

 Ge(IV) gl KCl 25°C 0.10M U 1958ANa (49566) 95
 $K(\text{HGeO}_3+2\text{L}=\text{HGeO}(\text{H}-2\text{L})_2)=3.29$

C6H12O6 L D-Glucose CAS 492-62-6 (1560)

D-Glucose

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ge(IV) gl KNO3 20°C 0.10M M 1980MBc (49589) 96

$$K(\text{GeO}_2+2\text{H}_2\text{L}=\text{Ge}(\text{OH})\text{L}_2+\text{H})=-6.33$$

$$K'(\text{Ge}(\text{OH})\text{L}_2+\text{H}_2\text{L}=\text{GeL}_3+\text{H})=-10.6$$

For L=D-dulcitol, K=-3.88, K'=-10.0; L=D-adonitol, K=-5.43, K'=-10.6.

 Ge(IV) gl KCl 25°C var U I 1963NFa (49590) 97

$$K(\text{HGeO}_3+2\text{L})=1.451+1.178\text{SQRTI}$$

 Ge(IV) gl KCl 25°C 0.10M U 1958ANa (49591) 98

$$K(\text{HGeO}_3+2\text{L}=\text{HGeO}(\text{H}-2\text{L})_2)=3.46$$

C6H12O6 L D-Mannose CAS 3458-28-4 (1562)

D-Mannose

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ge(IV) gl KCl 25°C 0.10M U 1958ANa (49606) 99

$$K(\text{HGeO}_3+2\text{L}=\text{HGeO}(\text{H}-2\text{L})_2)=4.13$$

C6H12O6 L Sorbose CAS 87-79-6 (930)

L(-)-Sorbose;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ge(IV) gl KCl 25°C 0.10M U 1959ATa (49614) 100

$$K(\text{HGeO}_3+2\text{L}=\text{HGeO}(\text{H}-2\text{L})_2)=5.35$$

C6H12O6 L Inositol CAS 87-89-8 (2285)

myo-Inositol, meso-Inositol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ge(IV) gl KCl 25°C 0.10M U 1967FAa (49638) 101

$$K(\text{HGeO}_3+2\text{L}=\text{HGeO}(\text{H}-2\text{L})_2)=2.140$$

C6H12O7 HL Gluconic acid CAS 526-95-4 (904)

D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=2.06 1986HPb (49721) 102

C6H13NO6 HL CAS 84518-56-9 (4387)
 2-Amino-2-deoxy-D-gluconic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl NaClO4 25°C 0.10M U M 2000KAa (50532) 103
 B(GeO(OH)H-1L)=3.01
 B(Ge(OH)2(H-1L)2)=6.63
 B(GeO(OH)H-2L)=-6.23
 B(Ge(OH)(H-1L)2)=14.35

Metal is Ge(OH)4. Also data for ternary species Ge(OH)4ML, M = Zn, Cd, Pb.

C6H14O6 L D-Dulcitol CAS 608-66-2 (3663)
 D-Galactitol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1959ARa (51061) 104
 K(HGeO3+2L=HGeO(H-2L)2)=4.71

 C6H14O6 L D-Mannitol CAS 69-65-8 (3664)
 D-Mannitol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KNO3 20°C 0.10M C 1979MBf (51081) 105
 K(GeO2+3H2L=GeL3+2H+2H2O)=-13.7; K(GeO2+2H2L=Ge(OH)L2+H+H2O)=-4.0;
 K(Ge(OH)L2+H2O=GeL3+H+H2O)=-9.7

Ge(IV) gl NaCl 25°C 0.50M U 1973PAb (51082) 106
 K(Ge(OH)4+L+H2O=GeH-1(OH)4L+H)=-6.43, K(Ge(OH)4+2L+H2O=GeH-1(OH)4L2+H)=-3.95
 K(2Ge(OH)4+2L+2H2O=(Ge(OH)4)2H-2L2+2H)=-10.68

Ge(IV) gl KCl 25°C var U 1963NFa (51083) 107
 K(HGeO3+2L)=3.394+1.055SQRTI

Ge(IV) gl KCl 25°C 0.10M U 1959ARa (51084) 108
 K(HGeO3+2L=HGeO(H-2L)2)=4.53

 C6H14O6 L Glucitol CAS 50-70-4 (2878)
 D-Sorbitol;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KNO3 20°C 0.10M C 1979MBf (51103) 109
 K(GeO2+3H2L=GeL3+2H+2H2O)=-12.3; K(GeO2+2H2L=Ge(OH)L2+H+H2O)=-3.7;
 K(Ge(OH)L2+H2O=GeL3+H+H2O)=-8.6

Ge(IV) gl KCl 25°C 0.10M U 1959ARa (51104) 110
 $K(\text{HGeO}_3 + 2\text{L} = \text{HGeO}(\text{H}-2\text{L})_2) = 5.09$

C6H15NO3 Triethanolamine CAS 102-71-6 (447)
 Tris-(2-hydroxyethyl)amine; L

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KNO3 20°C 0.10M U 1981MMe (51292) 111
 $K(\text{GeO}_2 + \text{H}_3\text{L} = \text{Ge}(\text{OH})\text{L} + \text{H}_2\text{O}) = 5.26$

C7H4O7 H3L Meconic acid CAS 497-59-6 (3723)
 3-Hydroxy-4-pyrone-2,6-dicarboxylic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967CBb (52565) 112
 $K(\text{Ge}(\text{OH})_4 + 2\text{HL}) < 1$

C7H6O2 HL Tropolone CAS 533-75-5 (3129)
 2-Hydroxycyclohepta-2,4,6-trien-1-one;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1966BBb (53675) 113
 $K(\text{Ge}(\text{OH})_4 + 2\text{HL}) = 8.03$
 $K(\text{Ge}(\text{OH})_4 + 3\text{HL} + \text{H}) = 13.3$

C7H6O3 H2L CAS 139-85-5 (881)
 3,4-Dihydroxybenzaldehyde, protocatechuic aldehyde; $\text{C}_6\text{H}_3(\text{OH})_2.\text{CHO}$

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1968A0a (54355) 114
 $K(\text{HGeO}_3 + 3\text{H}_2\text{L}) = 2.78$

C8H6O4 H2L CAS 6272-27-1 (4474)
 2,3-Dihydro-6,7-dihydroxy-3-oxobenzofuran;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp oth/un 25°C ? U M 1967NP a (58815) 115
 $K(\text{Ge}(\text{OH})_2 + 2\text{H}_2\text{L} = \text{Ge}(\text{OH})_2\text{L}_2 + 4\text{H}) = 11.4$

C8H8O3 HL Mandelic Acid CAS 611-72-3 (80)
 2-Phenyl-2-hydroxyethanoic acid; $\text{C}_6\text{H}_5.\text{CH}(\text{OH}).\text{COOH}$

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) EMF KCl 25°C 0.10M U 1962CLb (59836) 116

K(GeL3+H)=3.83
K(GeHL3+H)=2.53

Ge(IV) con NaCl 18°C 1.0M U 1957VAa (59837) 117
K(H2GeO3+2HL)=2.0

Ge(IV) gl oth/un 18°C 0.0 U 1957VAa (59838) 118
K(H2GeO3+2HL)=2.92

C9H5NOC12 HL CAS 773-76-2 (3278)
5,7-Dichloro-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967Tmd (63543) 119
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.7

C9H6NO4IS H2L Ferron CAS 547-91-1 (275)
7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967Tmd (63805) 120
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.78

C9H7N L CAS 119-65-3 (487)
Isoquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) cal non-aq 25°C 100% U H 1967M0b (64026) 121
Medium: n-hexane. Many data; DH(GeF4(l)+2L(l)=GeF4L2(c))=-149.2 kJ mol-1
DH(GeF4(g)+2L(l)=GeF4L2(c))=-170.9, DH(GeCl4(l)+2L(l)=GeCl4L2(c))=-93.2

C9H7NO HL Oxine CAS 148-24-3 (504)
8-Hydroxyquinoline (8-quinolinol);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967Tmd (64279) 122
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.61

C9H7NO4S H2L Sulfoxine CAS 84-88-8 (448)
8-Hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967Tmd (64549) 123
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.55

C9H28N3O15P5 10L DTPPH CAS 15827-60-8 (2921)
 Diethylenetriamine-N,N,N',N'',N''-penta(methylphosphonic acid);
 H2O3PCH2.N(CH2CH2.N(CH2PO3H2)2)2 H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U			K(Ge+H6L)=9.45	1984SBa (68410)	124

 C10H6O3 HL CAS 83-72-7 (3294)
 2-Hydroxy-1,4-naphthoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			K(Ge(OH)4+2HL=Ge(OH)2L2) < 3.0	1966BBb (68460)	125

 C10H8O5S H3L DHNSA (877)
 2,3-Dihydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KCl	25°C	0.10M	U			K(H2GeO3+3H2L=GeL3+2H)=2.0	1967PBd (69849)	126

 C10H8O8S2 H4L Chromotropic ac CAS 148-25-4 (1875)
 1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KNO3	20°C	0.10M	C			K(GeO2+3H2L=GeL3+2H+2H2O)=-4.8	1979MBf (69953)	127

Ge(IV)	sp	KCl	25°C	0.10M	U			K(H3GeO4+3H2L=HGeL3+2H)=2.30	1967PBd (69954)	128
--------	----	-----	------	-------	---	--	--	------------------------------	-----------------	-----

 C10H9NO HL 8-OH-Quinaldine CAS 826-81-3 (998)
 2-Methyl-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			K(Ge(OH)4+2HL=Ge(OH)2L2)=3.4	1967Tmd (70047)	129

 C10H9NO4S H2L CAS 29021-67-8 (3926)
 2-Methyl-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U				1967Tmd (70198)	130

C12H22O11 L Turanose CAS 547-25-1 (2701)
3-O-D-Glucopyranosyl-D-fructose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=2.32 B2=3.25	1986HPb (82866)	138

C12H22O11		L						alpha-Lactose CAS 5989-81-1 (2486)		
4-D-Beta-D-Galactopyranosyl-alpha-D-glucose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=1.97	1986HPb (82875)	139

C12H22O11		L						Maltose CAS 6363-53-7 (2705)		
4-O-alpha-D-Glucopyranosyl-D-glucose, Maltobiose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=1.48	1986HPb (82880)	140

C12H22O11		L						Cellobiose CAS 528-50-7 (2697)		
4-O-beta-D-Glucopyranosyl-D-glucose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=1.58	1986HPb (82886)	141

C12H22O11		L						Melibiose CAS 66009-10-7 (2699)		
6-O-D-Galactopyranose-D-glucose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=2.32 B2=3.24	1986HPb (82890)	142

C12H22O11		L						Gentiobiose CAS 554-91-6 (2698)		
6-O-D-Glucopyranosyl-D-glucose, Amygdalose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=1.27	1986HPb (82893)	143

C12H22O11		L						Trehalose CAS 6138-23-4 (2700)		
D-Glucopyranosyl-D-glucopyranoside;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	M			K1=1.12	1986HPb (82900)	144

C12H22O11		L						Sucrose CAS 57-50-1 (2523)		
beta-D-Fructofuranosyl-alpha-D-glucopyranoside; Saccharose;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Ge(IV) gl KCl 25°C 0.10M M K1=1.00 1986HPb (82910) 145

C12H24O11 L Maltitol CAS 585-88-6 (2709)
4-O-alpha-D-Glucopyranosyl-D-glucitol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=3.22 1988HLA (83683) 146

C12H24O11 L Lactitol CAS 535-94-4 (2710)
4-O-beta-D-Galactopyranosyl-D-glucitol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=3.08 1988HLA (83686) 147

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp oth/un 25°C 0.10M U 1972NFb (86732) 148
B3=52.80

Medium: acetate

C15H11N3O HL PAN CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp oth/un 27°C ? U M 1974ZSa (91219) 149
Keff(GeCl4+L)=3.3

C18H12O6 H2L (4124)
2,5-Dihydroxy-3,6-diphenoxy-1,4-benzoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967BBa (96886) 150
K(Ge(OH)4+2HL=Ge(OH)2L2)=8.8

C19H12O9Br2S H6L Bromo Pyrog.Red CAS 16574-43-9 (706)
5',5''-Dibromopyrogallolsulfonephthalein;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaNO3 ? 0.10M U 1969NMa (99011) 151
K(Ge(OH)3+3H2L)=12.9

C22H20O13 H5L Carminic acid CAS 1260-17-9 (714)

Carminic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	oth/un	25°C	?	U				1970BRa (101702)	152
									K(Ge(OH)4+H5L)=4.58	

Medium: conc H2SO4

C28H15NO4 L CAS 82-22-4 (3522)

1,1'-Iminodianthraquinone; (1,1'-dianthrimide)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	mixed	?	93%	U				1968LNa (104653)	153
									K(HGeO2+HL)=2.35(?)	

Medium: 93.2% H2SO4

REFERENCES

- 2000KAa Y Kanekiyo, S Aizawa, S Funahashi; Inorg.Chim.Acta, 298, 154 (2000)
1998PSb G Pokrovski, J Schott; Geochim.Cosmo.Acta, 62, 1631 (1998)
1998PSc G Povrovski, J Schott; Geochim.Cosmo.Acta, 63, 3413 (1998)
1991SOa Y Sohrin; Bull.Chem.Soc.Jpn., 64, 3363 (1991)
1990Cic L Ciavatta, M Iuliano, R Porto, E Vasca; Polyhedron, 9, 1263 (1990)
1990DEa A De La Cuadra; An.Quim., 86, 221 (1990)
1988HLa P Hakkinen et al; Finn.Chem.Lett., 15, 7 (1988)
1988SBb I Seifullina, T Batalova et al; Koord.Khim., 14(6)767 (1988)
1986HPb P Hakkinen et al; Finn.Chem.Lett., 13, 93 (1986)
1986SBb I Seifullina, T Batalova et al; Koord.Khim., 12(6)747 (1986)
1984SBa I Seifullina, T Batalova, A Kireeva; Koord.Khim., 10(3)336 (1984)
1983SBb I Seyfullina, T Batalova, A Kyreeva; Koord.Khim., 9, 67 (1983)
1982SGc I Seifullina, E Guzhavina, I Nazarova; Koord.Khim., 8, 1095 (1982)
1981MMe E Mikhailova, M Mikesova, M Bartusek; Coll.Czech.Chem.Comm., 46, 701 (1981)
1980Mbc A Mikan, M Bartusek; Coll.Czech.Chem.Comm., 45, 2645 (1980)
1979HUa E Huttunen; Finn.Chem.Lett., 236 (1979)
1979MBf M Mikesova, M Bartusek; Coll.Czech.Chem.Comm., 44, 3256 (1979)
1979NVa V Nazarenko, N Varlamova; Ukr.Khim.Zh., 45, 596 (1979)
1975BPf E Belousova, N Pozaritski et al; Zh.Neorg.Khim., 20, 2787(1542) (1975)
1974NOa G Nordheim; Acta Chem.Scand., A28, 115 (1974)
1974ZSa S Zaidi, K Siddiqi; Indian J.Chem., 12, 540 (1974)
1973BPa E Belousov, A Pozharitskii et al; Zh.Neorg.Khim., 18, 10, 2766 (1973)
1973PAb L Pettersson, I Andersson; Acta Chem.Scand., 27, 977; 1019 (1973)
1972NFb V Nazarenko, G Flyantikova, T Selyatina; Zh.Anal.Khim., 27, 12, 2369 (1972)
1972PAb N Parpiev; Uzbeksk.Khim.Zh., 6, 17 (1972)
1970BRa R Brown; Anal.Chim.Acta, 50, 157 (1970)
1970NLc V Nazarenko, N Lebedeva, L Vinogradova; Zh.Neorg.Khim., 15, 11, 2990 (1970)
1970NLd V Nazarenko, N Lebedeva, L Vinogradova; Zh.Neorg.Khim., 15, 3, 643 (1970)
1969KMe N Konopik, P Meszaros; Monatsh.Chem., 100, 649 (1969)
1969NMa V Nazarenko, N Makrinich; Zh.Anal.Khim., 24, 11, 1694 (1969)

1969NVa V Nazarenko, L Vinarova, N Lebedeva; Zh.Neorg.Khim., 14, 3, 700 (1969)
 1968AOa P Antikainen, H Oksanen; Acta Chem.Scand., 22, 2867 (1968)
 1968K Mb N Konopik, P Meszaros; Monatsh.Chem., 99, 902 (1968)
 1968LNa F Langmyhr, G Norheim; Anal.Chim.Acta, 41, 341 (1968)
 1968NFa V Nazarenko, G Flyantikova; Zh.Neorg.Khim., 13, 1855(E:966) (1968)
 1968PMf N Parpiev, I Maslennikov; Uzbeksk.Khim.Zh., 2, 6 (1968)
 1967BBa A Beauchamp, R Benoit; Bull.Soc.Chim.Fr., 672 (1967)
 1967CBb G Choux, R Benoit; Bull.Soc.Chim.Fr., 2920 (1967)
 1967FAa R Frostell, P Antikainen; Suomen Kem., B40, 86 (1967)
 1967KOc N Konopik; Z.Anal.Chem., 224, 107 (1967)
 1967MOb J Miller, M Onyszchuk; J.Chem.Soc.(A), 1132 (1967)
 1967NP a V Nazarenko, E Poluektova; Zh.Anal.Khim., 22, 6, 895 (1967)
 1967PBd P Pichet, R Benoit; Inorg.Chem., 6, 1505 (1967)
 1967TMD J Tsau, S Matsuo, P Clerc, R Benoit; Bull.Soc.Chim.Fr., 1039 (1967)
 1966ANa A Andrianov, V Nazarenko; Zh.Neorg.Khim., 11, 1527(E:816) (1966)
 1966ATc P Antikainen, K Tevanen; Suomen Kem., B39, 2 (1966)
 1966BBb A Beauchamp, R Benoit; Can.J.Chem., 44, 1607; 1615 (1966)
 1965REa W Reid; J.Phys.Chem., 69, 2269 (1965)
 1965R Ka I Ryss, N Kulish; Zh.Neorg.Khim., 10, 1827 (1965)
 1964BBb A Beauchamp, R Benoit; Can.J.Chem., 42, 2161 (1964)
 1964GUa R Gut; Helv.Chim.Acta, 47, 2262 (1964)
 1964GZa K Gayer, O Zajicek; J.Inorg.Nucl.Chem., 26, 2123 (1964)
 1964KSd G Kurnevich, G Shagisultanova; Zh.Neorg.Khim., 9, 1383 (2559) (1964)
 1964RKb I Ryss, N Kulish; Zh.Neorg.Khim., 9, 1103 (1964)
 1964RKc I Ryss, N Kulish; Zh.Neorg.Khim., 9, 1382 (1964)
 1963ANC A Andrianov, V Nazarenko; Zh.Neorg.Khim., 8, 1194 (2281) (1963)
 1963ATa P Antikainen, K Tevanen; Suomen Kem., B36, 199; B35, 224 (1963)
 1963BPb R Benoit, J Place; Can.J.Chem., 41, 1170 (1963)
 1963NFa V Nazarenko, G Flyantikova; Zh.Neorg.Khim., 8, 1189 (2271); 712 (137)
 (1963)
 1962CLb E Clark; J.Inorg.Nucl.Chem., 24, 81 (1962)
 1962NFa V Nazarenko, G Flyantikova; Zh.Neorg.Khim., 1, 1210 (1962)
 1961ADb A Angerstein, W Davidson; Z.Anorg.Chem., 310, 26 (1961)
 1960ARb P Antikainen, V Rossi; Suomen Kem., B33, 38; 210 (1960)
 1960KRb W Kemula, S Rosolowski; Roczn.Chem., 34, 835 (1960)
 1959AMa P Antikainen, P Malkonen; Suomen Kem., B32, 179 (1959)
 1959ANa P Antikainen; Acta Chem.Scand., 13, 312 (1959)
 1959ARa P Antikainen, V Rossi; Suomen Kem., B32, 182; 185 (1959)
 1959ATA P Antikainen, K Tevanen; Suomen Kem., B32, 214 (1959)
 1959L Ba B Lovrecek, J Bockris; J.Phys.Chem., 63, 1368 (1959)
 1958ANa P Antikainen; Suomen Kem., B31, 255 (1958)
 1958ISb F Iimura, M Shima, H Sano; Nippon Kagaku Kaishi, 79, 1032/7; 1041/5/8 (1958)
 1957ANa P Antikainen; Suomen Kem., B30, 45; 147 (1957)
 1957VAa O Vartapetian; Ann.Chim.(France), 2, 916 (1957)

EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES

I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

END Experiments recorded for
from SC-Database on Saturday, 01 January, 2000 at 00:45:09