

## SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 52 experiments for  
(no ligands specified)

Metal : Ta

(no references specified)

(no experimental details specified)

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e- HL Electron (442)  
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	sp	NaClO4	25°C	0.10M	U			1971CKa (948) 1		
K(Ta6Cl12 + e)=14.0 (830mV)										
Medium:HClO4. K=8.3(0.49V,(+++)). For Br complex, values are 15.0(0.89V) and 10.0(0.59V) (Esce=0.242V). Method:spectroscopy and current-voltage studies										

Ta	kin	NaClO4	15°C	0.10M	U			1966EMb (949) 2		
K=-1.74										
Medium 0.1M HClO4. By spectrophotometry, 22 C: K'=-1.80										

Ta	oth	none	25°C	0.0	U			1952LAB (950) 3		
K=-68.6(-810 mV)										
K: 0.5Ta2O5(s)+5H+5e=Ta(s)+2.5H2O. From thermodynamic data										

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Cl- HL Chloride CAS 7647-01-0 (50)  
Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	ISE	oth/un	175°C	?	C			1992BMA (5760) 4		
K6=3.89										
Medium: NaCl-AlCl3 melt.										

Ta	nmr	oth/un	-90°C	var	U	M		1971BIb (5761) 5		
K'=0.52(cis)										
K'=1.15(trans)										
K''=2.39(cis)										
K''=-1.35(trans)										
K': 5TaF2L4=TaF6+4TaFL5. K'': 5TaF3L3=2TaF6+3TaFL5. K(5TaF4L2=3TaF6+2TaFL5)=-1.29(cis); -0.96(trans). Data also for other complexes and Br analogues										

Ta	gl	alc/w	25°C	100%	U	M		1965GSd (5762) 6		
K'=-3.55										
K''=-7.83										
Medium: MeOH. K':TaCl5+MeOH=TaOMeCl4+H+Cl, K'':TaCl5+2MeOH=Ta(OMe)2Cl3+2H+2Cl										

Ta	oth	non-aq	300°C	100%	U	T		1959COa (5763) 7		
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K6=3.2 , x units  
K(TaCl5(g)+Cl(melt))=1.8  
Method: by partial pressure of TaCl5. Medium: liquid NaFeCl4.  
K(TaCl5(g)+Cl(in melt)=TaCl6(in melt))=0.5(400 C) atm and x units  
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F- HL Fluoride CAS 7644-39-3 (201)  
Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	ISE	non-aq	?	100%	C			K6=8.88 K(TaF5+TaF6=Ta2F11)=1.50	1978GRa (7209)	8

Medium: liquid anhydrous HF

Ta	sp	oth/un	?	?	U			K(TaH2O2+F)=2.39 B(TaH2O2+2F)=2.35 B(TaH2O2+3F)=1.93 B(TaH2O2+4F)=1.83	1973LCa (7210)	9
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Ta	ISE	NaClO4	25°C	2.0M	U			K1=6.37 B2=11.85 B3=16.03 B4=19.63 B5=23.29 B7=30.21	1972BAb (7211)	10
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Metal: Ta(V)

Ta	ix	NaClO4	25°C	1.0M	U			K4=5.90 K4K5=10.80	1969VAa (7212)	11
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Metal: Ta(V). Method: quinhydrone electrode also

Ta	ix	NaClO4	25°C	3.0M	U			K4.K5.K6.K7=18.9 K4.K5.K6.K7.K8.K9=25.2	1969VAa (7213)	12
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Medium: HClO4. Metal: Ta(V). Medium: quinhydrone electrode also used

Ta	EMF	NaClO4	25°C	1.0M	U			K6=3.75 K7=3.10 K8=0.66	1966BFb (7214)	13
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Method: quinhydrone electrode

Ta	dis	NaClO4	25°C	3.0M	U			K4=5.86 K4K5=10.77 B6/B3=15.67 B7/B3=20.15	1965VWa (7215)	14
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Ta ix oth/un 25°C 1.00M U 1962VF a (7216) 15

K5=4.8

K6=3.6

K7=3.3

K8=3.0

K9=3.6. Method: anion exchange and quinhydrone electrode.

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O2-- H2L Peroxide CAS 7772-84-1 (2813)

Peroxide; -0.0-

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Ta sp oth/un 20°C 78% U TIH 1973VZ a (12699) 16

K(TaOSO4+H2L)=2.73

Medium:78.4% H2SO4. K=2.80(15 C), 2.62(35 C), 2.55(55 C)

DH=-17 kJ mol-1 (TaO(SO4) assumed) also 63.5, 88.7, 94.5%

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Ta sp oth/un 0°C 90% U 1969CK a (12700) 17

K(TaOSO4+H2L)=3

Medium:H2SO4

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Ta sp oth/un 0°C 10% U I 1969VZ a (12701) 18

K(TaOSO4+H2L)=1.48

Medium: 10% H2SO4 K=1.48(20%), 2.20(30%), 2.59(50%), 2.80(70%), 3.38(80%), 3.59(100%)

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Ta sp non-aq ? 100% U 1968VZ a (12702) 19

K(Ta(V)+H2L)=3.43

Medium: H2SO4

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Ta vlt oth/un 25°C 0.34M U 1964BR b (12703) 20

B(HTaO3+H2L)=2.0

Medium: H2SO4

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SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Ta sp alc/w ? 100% U I K1=3.12 B2=5.48 1964GS a (15264) 21

B3=7.77

Medium:MeOH. In BuOH: K1=3.68, B2=7.05, B3=11.42. In Me2NCHO: K1=3.15,

B2=5.92, B3=8.55, B4=11.06, B5=13.52, B6=15.96

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CH4O L Methyl alcohol CAS 67-56-1 (597)

Methanol; CH3.OH

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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo  
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Ta EMF alc/w 20°C 100% U M 1965GB a (17902) 22

$K(\text{Ta}(\text{L}')_4 + 2\text{L}' = \text{Ta}(\text{L}')_6 + \text{A}) = 7.1$   
 $K(\text{Ta}(\text{H}-1\text{L})_4 + \text{A}) = 11.04$   
 $K(\text{Ta}(\text{H}-1\text{L})_3 + \text{TaH}-1\text{L}) = 12.95$   
 $K(\text{Ta}(\text{L}')_5 + \text{HA} = \text{TaA}(\text{L}')_4 + \text{L}) = 4.36$   
 Method: H electrode. Medium: MeOH, 1.0 M Me<sub>4</sub>NCl. HA=acetylacetone, L'=H-1L

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Ta	EMF	alc/w	20°C	100%	U	M	1965GBa (17903)	23
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$K' = 14.2$   
 $K'' = 7.85$   
 $K(\text{TaA}(\text{L}')_3 + \text{L}' = \text{TaA}(\text{L}')_4) = 9.04$   
 $K''' = 2.5$   
 Method: H electrode. Medium: MeOH, 1.0 M Me<sub>4</sub>NCl; H<sub>2</sub>A=catechol; L'=H-1L. K':  
 $\text{TaA}(\text{L}')_3 + \text{H}_2\text{A} + \text{L}' = \text{TaA}_2(\text{L}')_2 + 2\text{L}$ . K'':  $\text{TaA}_2(\text{L}')_2 + \text{H}_2\text{A} + \text{L}' = \text{TaA}_3\text{L}'$ . K''':  $\text{TaAL}'_4 + \text{TaAL}'_3$

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Ta	EMF	alc/w	20°C	100%	U		1964GUa (17904)	24
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$K(\text{Ta}(\text{H}-1\text{L})_4 + \text{H}-1\text{L}) = 11.47$   
 $K(\text{Ta}(\text{H}-1\text{L})_5 + \text{H}-1\text{L}) = 6.67$   
 $K(\text{Ta}(\text{H}-1\text{L})_6 + \text{H} = \text{Ta}(\text{H}-1\text{L})_5 + \text{L}) = 5.1$   
 $K(\text{Ta}(\text{H}-1\text{L})_7 + \text{H} = \text{Ta}(\text{H}-1\text{L})_6 + \text{L}) = 9.9$   
 Method: H electrode; medium: MeOH, 1.0 M Me<sub>4</sub>NCl

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C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	H <sub>2</sub> L	Oxalic acid	CAS 144-62-7	(24)
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Ethanedioic acid; (COOH)<sub>2</sub>

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Ta	sol	oth/un	?	0.10M	U			1970ZPa (19077)	25
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$K_3 = 5.91$

Medium: HClO<sub>4</sub>  
 Metal ion is TaO<sup>+++</sup>

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Ta	sol	oth/un	19°C	?	U			1965BLd (19078)	26
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$K(\text{Ta}(\text{OH})_2 + \text{L}) = 11.10$   
 $K(\text{Ta}(\text{OH})_2 + 2\text{L}) = 18.52$   
 $K(\text{Ta}(\text{OH})_2\text{L} + \text{OH}) = 13.33$

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C <sub>2</sub> H <sub>3</sub> N	L	Cyanomethane	CAS 75-05-8	(1399)
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Acetonitrile; CH<sub>3</sub>CN

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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (19196)	27
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$K(\text{TaBr}_5\text{A} + \text{L} = \text{TaBr}_5\text{L} + \text{A}) = -0.89$

Medium: CH<sub>2</sub>Cl<sub>2</sub>. A=t-butylcyanide

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Ta	nmr	non-aq	-40°C	100%	U	M		1972MBb (19197)	28
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$K(\text{TaCl}_5\text{A} + \text{L} = \text{TaCl}_5\text{L} + \text{A}) = 0.32$

Medium: CHCl<sub>3</sub>. A=dimethylether. K=0.36, A= 1,4-dioxan;  
 K=1.57, A=diethylether; K=0.70, A=1,4-dithiane.

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C2H6NOC12P L CAS 667-43-0 (910)  
Dichloro(dimethylamine)phosphine oxide; (CH3)2N.P(O)Cl2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (21901) K(TaBr5A+L=TaBr5L+A)=1.48		29

Medium: CH2Cl2, A=acetonitrile

C2H6O L CAS 115-10-6 (4214)  
Dimethyl ether; CH3.O.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	nmr	non-aq	-40°C	100%	U	M		1972MBb (22021) K(TaCl5A+L=TaCl5L+A)=1.25		30

Medium: CHCl3l. A=diethyl ether. K=0.04, A=dioxan. Metal ion: Ta(V)

C2H6S L CAS 75-18-3 (151)  
Dimethyl sulfide; CH3.S.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (22195) K(TaCl5A+L=TaCl5L+A)=0.72		31

Medium: CH2Cl2, A=pivalocyanide

Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (22196) K(TaBr5A+L=TaBr5L+A)=0.24		32
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Medium: CH2Cl2, A=pivalocyanide

Ta	nmr	non-aq	-40°C	100%	U	M		1972MBb (22197) K(TaCl5A+L=TaCl5L+A)=0.26		33
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Medium: CHCl3. A=t-butyl nitrile. K=0.40, A=acetonitrile, K=1.98, A=diethyl ether. K=0.72, A=diethyl sulfide. K=0.72, A=dimethyl sulfide.

Ta	nmr	non-aq	-60°C	100%	U	M		1972MBb (22198) K(TaBr5A+L=TaBr5L+A)=0.24		34
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Medium: CH2Cl2. A=pivalonitrile.

C2H6Se L DiMeSelenide CAS 81369-92-3 (911)  
Dimethylselenide; CH3.Se.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (22207) K(TaCl5A+L=TaCl5L+A)=0.06 K(TaBr5A+L=TaBr5L+A)=0.43 K(TaBr5B+L=TaBr5L+B)=0.67		35

Medium: CH2Cl2, A=dimethylthioether, B=t-butylcyanide

C2H6Te L DiMeTelluride CAS 593-80-6 (912)  
Dimethyltelluride; CH3.Te.CH3

Medium: CH<sub>2</sub>Cl<sub>2</sub>, A=selenobismethane, B=thiobismethane

C4H6O6                      H2L      L-Tartaric acid    CAS 87-69-4    (92)  
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid;  $\text{HOOC} \cdot \text{CH}(\text{OH}) \cdot \text{CH}(\text{OH}) \cdot \text{COOH}$

C4H8OS L 1,4-Thioxane CAS 15980-15-1 (4266)  
1,4-Oxathiane; cyclo(-O.CH2.CH2.S.CH2.CH2-)

Medium: CHCl<sub>3</sub>. A=t-butyl nitrile

C4H8S	L	CAS 110-01-0	(150)
Tetrahydrothiophene; cyclo(-CH2.CH2.S.CH2.CH2-)			

A=t-butyl mercaptan. Medium: CHCl<sub>3</sub>

C4H8S2                      L        1,4-Dithiane                      CAS 505-29-3    (4255)  
1,4-Dithiane; cyclo-(S.CH2.CH2.S.CH2.CH2-)

A=t-butyl nitrile. Medium: CHCl<sub>3</sub>

C4H10S L CAS 352-93-2 (4259)

Diethyl sulfide; C<sub>2</sub>H<sub>5</sub>.S.C<sub>2</sub>H<sub>5</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (34722)	41	
									K(TaBr5A+L=TaBr5L+A)=-0.61	

A=t-butyl nitrile. Medium: CH<sub>2</sub>Cl<sub>2</sub>

Ta	nmr	non-aq	-40°C	100%	U	M		1972MBb (34723)	42	
									K(TaCl5A+L=TaCl5L+A)=0.00	

A=dimethyl ether. Medium: CHCl<sub>3</sub>.

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C5H8O2                      HL      Acetylacetone      CAS 123-54-6      (164)  
Pentane-2,4-dione; CH<sub>3</sub>.CO.CH<sub>2</sub>.CO.CH<sub>3</sub>

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	EMF	non-aq	20°C	100%	U	M		1971GSa (38090)	43	
									K(TaA5+HL=TaA4L+HA)=4.36	
									K(TaA3L+A)=12.95	
									K(TaA4L+2A=TaA6+L)=7.10	

Medium: MeOH. HA=MeOH

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C5H9N                      L      t-Butylnitrile      CAS 7188-38-7      (913)  
t-Butylcyanide;(CH<sub>3</sub>)<sub>3</sub>C.CN

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	nmr	non-aq	-60°C	100%	U	M		1974GMa (38456)	44	
									K(TaBr5A+L=TaBr5L+A)=0.11	
									K(TaBr5B+L=TaBr5L+B)=0.39	
									K(TaBr5C+L=TaBr5L+C)=1.72	

Medium: CH<sub>2</sub>Cl<sub>2</sub>, A=acetonitrile, B=thiobismethane, C=dimethylether

Ta	nmr	non-aq	-40°C	100%	U	M		1972MBb (38457)	45	
									K(TaCl5A+L=TaCl5L+A)=0.47	

Medium: CHCl<sub>3</sub>. A=dimethyl ether. When A=cyanomethane, K=0.15,

A=1,4-dioxan (-60 C), K=0.51

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C6H6O2                      H2L      Catechol      CAS 120-80-9      (534)  
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	EMF	alc/w	20°C	100%	U	M		1971GSa (43837)	46	
									K(TaA3L+A)=9.04	
									K(TaA3L+TaA4L=Ta2A7L2)=2.50	
									K(TaA3L+H2L+A=TaA2L2+2HA)=14.2	
									K(TaA2L2+H2L+A=TaAL3+2HA)=7.85	

Medium: MeOH, 1.0 M Me<sub>4</sub>NCI. HA=CH<sub>3</sub>OH

C6H8O6                      H2L            Ascorbic acid            CAS 50-81-7    (285)  
Ascorbic acid (Vitamin C);

C10H16N2O8	H4L	EDTA	CAS 60-00-4	(120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;				

Medium: K<sub>2</sub>S<sub>04</sub>

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C11H9N3O2                      H2L        PAR                      CAS 1141-59-9    (636)  
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

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C14H8O7S                    H3L      DASA                    CAS 83-61-4 (950)  
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;  
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C17H17NO3                      HL                      CAS 58434-59-6    (1213)  
2'-Hydroxy-4-methoxy-5'-methylbenzylidene acetophenone oxime

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ta	sp	oth/un	30°C	8.00M	U	M		K(TaO(SCN)+L)=3.56 K(TaO(SCN)L+L)=2.07	1980GKa (96191)	52

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#### EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES  
 I Data with various BACKGROUNDS  
 H Data for THERMOCHEMICAL quantities  
 M Data for TERNARY Complexes

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 END