```
Software version = 5.81 Data version = 4.62
Experiment list contains 118 experiments for
(no ligands specified)
Metal : Cm+++
(no references specified)
(no experimental details specified)
***********************************
            HL
               Electron
                          (442)
Electron:
       Mtd Medium Temp Conc Cal Flags Lg K values
                                Reference ExptNo
-----
Cm+++ oth none 25°C 0.0 U
                               1969NBa (409) 1
                      K(Cm+e=Cm(II))=-84.5(-5.0V)
Method: Estimated data
-----
Cm+++ sp oth/un 25°C ? U
                               1965MIb (410) 2
                      K(Cm++++++e)=55 (3250 mV)
*********************************
Br-
                        CAS 10035-10-6 (19)
            HL
               Bromide
Bromide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     dis NaClO4 20°C 3.00M U K1=0.39 B2=0.22 1982FKb (1820)
                                            3
********************************
CO3--
           H2L Carbonate CAS 465-79-6 (268)
Carbonate;
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Cm+++ sp NaCl 25°C 0.0 C I K1=8.1 B2=13.00 1999FKa (3167)
                      B3=15.2
                      B4=13.0
Method: Laser fluorescence spectroscopy. Media: 0-6 m NaCl.
****************************
C1-
            HL
               Chloride
                        CAS 7647-01-0 (50)
Chloride;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     dis NaClO4 20°C 3.00M U
                      K1=0.56 B2=0.20 1982FKb (4609)
                                            5
______
    dis NaCl 30°C 1.00M U K1=0.21 B2=-0.03 1980KMa (4610)
                                            6
-----
Cm+++ ix none ? 0.0 U K1=1.17
                               1956WWa (4611) 7
**********************************
            HL Perchlorate CAS 7001-90-3 (287)
C104-
```

SC-Database

Perchlor	ate;	
Metal	Mtd Medium Temp Conc Cal Fla	ags Lg K values Reference ExptNo
Cm+++ ******	**********	B2=0.38 1981LMa (6176) 8  ***********************************
Fluoride		
Metal	Mtd Medium Temp Conc Cal Fla	ags Lg K values Reference ExptNo
Cm+++	dis NaClO4 25°C 0.50M C	K1=3.34 B2= 6.18 1970ALc (6809) B3=9.08
	extraction of 244Cm from 0.50 M hylhexyl)phosphoric acid. Mediu	
		K1=3.34 B2=6.18 1969ALd (6810) B3=9.1
CIII+++	sol NaClO4 23°C 0.10M U TIH	1954FEa (6811) 11 K3=3.90
Medium: I At 0 C: I	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr,	1954FEa (6811) 11 K3=3.90 Ks(CmF3(s)=CmF3)=-4.91 =139 J K-1 mol-1; DH(Ks)=16.7, DS=-38 47 C K3=4.37, Ks=-4.75
Medium: I At 0 C: I ****** NO3- Nitrate;	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr, ***********************************	1954FEa (6811) 11  K3=3.90  Ks(CmF3(s)=CmF3)=-4.91  F139 J K-1 mol-1; DH(Ks)=16.7, DS=-38
Medium: I At 0 C: I ****** NO3- Nitrate;	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr, ***********************************	1954FEa (6811) 11 K3=3.90 Ks(CmF3(s)=CmF3)=-4.91 =139 J K-1 mol-1; DH(Ks)=16.7, DS=-38 47 C K3=4.37, Ks=-4.75  ***********************************
Medium: I At 0 C: I ******* NO3- Nitrate; Metal Cm+++	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr, **************  HL Nitrate  Mtd Medium Temp Conc Cal Fladis NaCl 30°C 1.00M U	1954FEa (6811) 11  K3=3.90  Ks(CmF3(s)=CmF3)=-4.91  139 J K-1 mol-1; DH(Ks)=16.7, DS=-38  47 C K3=4.37, Ks=-4.75  ***********************************
Medium: I At 0 C: I ******* NO3- Nitrate; Metal Cm+++	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr, ***************  HL Nitrate  Mtd Medium Temp Conc Cal Fla dis NaCl 30°C 1.00M U  dis R4N.X 25°C 2.0M U	1954FEa (6811) 11  K3=3.90  Ks(CmF3(s)=CmF3)=-4.91  139 J K-1 mol-1; DH(Ks)=16.7, DS=-38  47 C K3=4.37, Ks=-4.75  ***********************************
Medium: I At 0 C: I ******* NO3- Nitrate; Metal Cm+++ Cm+++ Medium:NI Cm+++	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr, *****************************  HL Nitrate  Mtd Medium Temp Conc Cal Fla dis NaCl 30°C 1.00M U  dis R4N.X 25°C 2.0M U  H4SCN  ix R4N.X 20°C 1.0M U  cation exchange. Medium: NH4Cl,	1954FEa (6811) 11  K3=3.90  Ks(CmF3(s)=CmF3)=-4.91  E139 J K-1 mol-1; DH(Ks)=16.7, DS=-38  47 C K3=4.37, Ks=-4.75  Ex***********************************
Medium: I At 0 C: I ******* NO3- Nitrate; Metal Cm+++ Cm+++ Medium:NI Cm+++ Medium:NI	HClO4. DH(K3)=17.9kJ mol-1, DS= K3=3.87, Ks=-5.22. At I=0 corr, *****************************  HL Nitrate  Mtd Medium Temp Conc Cal Fla dis NaCl 30°C 1.00M U  dis R4N.X 25°C 2.0M U  H4SCN  ix R4N.X 20°C 1.0M U  cation exchange. Medium: NH4Cl,	1954FEa (6811) 11  K3=3.90  Ks(CmF3(s)=CmF3)=-4.91  139 J K-1 mol-1; DH(Ks)=16.7, DS=-38  47 C K3=4.37, Ks=-4.75  ***********************************

\*

Hydroxide

HL

Cm+++

OH-

sp none 25°C 0.0 U K1=0.90 B2=1.38 1983MCb (10182) 15

(57)

Cm+++	dis oth/u	n 30°C 0.	01M C	*K1=-3.30	1989MKb (11127) 16	
Medium: Cl	LCH2COOH					
Cm+++	dis NaClO	4 ? 0.		*K1=-5.40	1973HHd (11128) 17	
Medium: Li						
Cm+++	oth R4N.X	25°C 0.	01M U	K1=10.6 B2=18	8.9 1972SSf (11129) transference number	18
Cm+++	dis NaClO	4 23°C 0.	10M U	*1/4 5 02	1969DHa (11130) 19	
Medium: Li	iC104			*K1=-5.92		
Cm+++	dis NaClO	4 23°C 0.	10M U	*K1=-6.05	1969GMa (11131) 20	
Medium: Li	iC104					
Cm+++	dis NaClO	4 25°C 0.		*K2=-7.85	1969MGf (11132) 21	
	iClO4 + tri		*****		******	
PO4 Phosphate;				CAS 7664-3		
Metal	Mtd Mediu	m Temp Co	nc Cal Flag	s Lg K values	Reference ExptNo	
	Mtd Mediu ix R4N.X			K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84	Reference ExptNo 4.1 1971MOd (13131)	22
Cm+++ Medium:NH4	ix R4N.X	20°C 1.	00M U	K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84 K(Cm+4H2L)=3.1	4.1 1971MOd (13131)	22
Cm+++ Medium:NH4	ix R4N.X	20°C 1.	00M U	K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84 K(Cm+4H2L)=3.1		22
Medium:NH4 Cm+++  Methods: s	ix R4N.X	20°C 1.	.0 U	K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84 K(Cm+4H2L)=3.1 K1=20.2 B2=36 K(Cm+H2L)=2.40 K(Cm+H2L)=3.60 K(Cm+2H2L)=3.60 K(Cm+3H2L)=5.61 K(Cm+4H2L)=6.2 ibution, EMF	4.1 1971MOd (13131)	
Medium:NH4 Cm+++  Methods: s ******** P309 Cyclotrime	ix R4N.X  C1  oth none  solubility,  ********	20°C 1.  20°C 1.  ion exch  ******  H3L e;	00M U	K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84 K(Cm+4H2L)=3.1 K1=20.2 B2=36 K(Cm+H2L)=2.40 K(Cm+H2L)=3.60 K(Cm+2H2L)=5.61 K(Cm+4H2L)=6.2 ibution, EMF ************************************	4.1 1971MOd (13131)  6.8 1969MOc (13132)	
Medium:NH4 Cm+++  Methods: s ******** P309 Cyclotrime	ix R4N.X  Cl oth none  colubility, ********	ion exch ****** H3L e;	00M U 0 U  ange, distr:	K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84 K(Cm+4H2L)=3.1  K1=20.2 B2=36 K(Cm+H2L)=2.40 K(Cm+2H2L)=3.60 K(Cm+3H2L)=5.61 K(Cm+4H2L)=6.2 ibution, EMF ************************************	4.1 1971MOd (13131)  6.8 1969MOc (13132)	
Medium:NH4 Cm+++  Methods: s ******** P309 Cyclotrime Metal	ix R4N.X  Cl oth none  caphosphat Mtd Mediu	ion exch ******  H3L e; m Temp Co	00M U  ange, distr: ************************************	K1=17.5 B2=34 K(Cm+H2L)=1.48 K(Cm+2H2L)=2.08 K(Cm+3H2L)=2.84 K(Cm+4H2L)=3.1  K1=20.2 B2=36 K(Cm+H2L)=2.40 K(Cm+2H2L)=3.60 K(Cm+3H2L)=5.61 K(Cm+4H2L)=6.2 ibution, EMF ************************************	4.1 1971MOd (13131)  6.8 1969MOc (13132)  ***********************************	

SCN- Thiocyanat	:e;	HL	Thiocyana	te CAS	463-56-9	(106)	
 Metal	Mtd Medium	Temp	Conc Cal Fl	ags Lg K val			
 Cm+++ IUPAC eval				R K1=0.44	199	•	19) 26
In 1.0 M N	NH4C104/NH4S	30°C SCN. Ex	1.0M C T H traction of	244Cm into C. DH(K1)=-2	198 hexane/amm	30KMe (1485 nonium dinc	50) 27 onyl-
				T K1=0.62 By calorime		•	•
Medium: NH	14C104/NH4SC	N, pH	2.8	T K1=0.18			` ,
Cm+++	dis R4N.X	25°C	2.0M U	K1=0.6 B3=1.15	B2=0.70	1973CDd	(14853)
				T K1=0.45 B3=-0.08	B2=-0.07		
 Cm+++	dis NaClO4	25°C	1.0M U	T K1=0.43	B2=0.85	1965CKb	(14855)
	NH4ClO4 K1=0	.67. I	n I=0 corr	K1=0.27 K1=1.62. Met	hod: catio	n exchange	·
SO4 Sulfate;		H2L	Sulfate	CAS	7664-93-9	(15)	
Metal	Mtd Medium	n Temp	Conc Cal Fl	ags Lg K val	ues	Reference	ExptNo
Cm+++	dis NaCl	30°C		K1=1.51		1980KMa	(16063)
 Cm+++	dis NaClO4	25°C	1.0M U I	K(Cm+HL)= K(Cm+2HL)	197 0.52 =0.83	'8RBa (1606	54) 35
Method: el	lectrical me	gratio	0.10M U T	K1=2.41 erence numbe C)	197	'3STe (1606	55) 36
Cm+++	dis none	25°C	0.0 U	K1=3.88 B3=5.15		1972MCc	(16066)
 Cm+++	dis NaClO4	25°C		K1=1.85			

```
By cation exchange: K1=1.86, B2=2.37
-----
      dis NaCl04 55°C 2.0M U T H K1=1.61 B2=2.30 1967CCd (16068)
K1=1.08(0 C), 1.34(25 C), 1.49(40 C); B2=1.66(0 C), 1.86(25 C), 2.05(40 C)
DH(K1)=17.1 kJ mol-1, DS=83.6 J K-1 mol-1
      ix R4N.X 25°C 0.75M U I K1=1.75 B2=1.93 1960LPb (16069)
                                               40
Medium: NH4Cl,ClO4. I=0 corr.: K1=3.66, K2=0.81
********************************
                         CAS 13590-71-1 (1752)
Methylphosphonic acid; CH3.PO3H2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Cm+++ ix none 25°C 0.00 U I
                                 1967BEa (18127) 41
                        K(Cm+HL)=2.82
At I=0.5 M NH4ClO4: K(Cm+HL)=1.86
*******************************
                          CAS 2617-47-2 (1977)
Hydroxymethylphosphonic acid; HO.CH2.PO3H2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Cm+++ ix R4N.X 25°C 0.20M U
                                  1972EZd (18147) 42
                        K(Cm+HL)=1.65
                        K(Cm+2HL)=3.28
Medium: NH4ClO4
**********************************
                 Oxalic acid CAS 144-62-7 (24)
             H2L
Ethanedioic acid; (COOH)2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     oth oth/un 25°C 0.10M U K1=5.30 B2=8.90 1971STe (18832)
Method : electrical migration or transference number
______
Cm+++ ix NaClO4 25°C 0.50M U K1=4.80 B2=8.61 1968ALd (18833)
                        B2=8.8 1966STa (18834) 45
Cm+++ dis R4N.X 20°C 0.10M U
                        B3=12.1
Medium : NH4Cl
______
Cm+++ ix oth/un 23°C 0.20M U K1=5.96 B2=10.15 1960LPa (18835)
*********************************
             HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH3.COOH
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     cal NaCl 25°C 2.0M U H K1=1.92
                                1985CLb (19922) 47
```

dis NaClO4 25°C 2.00M U T K1=2.03

1969MOc (19923) 48

49

51

0-55 C. 0 C: K1=1.73, 40 C: K1=2.11, 55 C: K1=2.27

From literature, I=0: K1=3.31, B2=4.72, B3=6.30, B4=6.56

\_\_\_\_\_

dis NaCl04 20°C 0.50M U K1=2.06 B2=3.09 1963GRa (19924) \*

Glycolic acid CAS 79-14-1 (33) HL

2-Hydroxyethanoic acid; HO.CH2.COOH

\_\_\_\_\_\_

Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo \_\_\_\_\_

50 0.5 C: K1=2.63, K2=1.83; 52 C: K1=2.52, K2=1.82

----dis NaClO4 20°C 0.50M U K1=2.85 B2=4.75 1963GRa (20516)

\*

Glycine CAS 56-40-6 (85)

2-Aminoethanoic acid; H2N.CH2.COOH

\_\_\_\_\_\_

Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo \_\_\_\_\_\_

Cm+++ dis NaClO4 25°C 2.0M U T H T 1968TCa (21515) 52

K(Cm+HL)=0.80

K=0.62(0 C), 0.66(11 C), 0.95(40 C). DH=13.8 kJ mol-1, DS=62.7 J K-1 mol-1 \*

CAS 4408-78-0 (4225) C2H505P H3L

Phosphonoethanoic acid; HOOC.CH2.PO3H2

------

Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo \_\_\_\_\_\_ Cm+++ ix none 25°C 0.00 U 1972EZc (21891) 53

> K(Cm+HL)=5.17B(Cm+2HL)=8.5K(Cm+H2L)=2.72

\*

C3H6O3 L-Lactic acid CAS 79-33-4 (82) HL

L-2-Hydroxypropanoic acid; CH3.CH(OH).COOH

\_\_\_\_\_\_

Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo -----10°C 1.50M U K1=2.59 B2=4.29 1972SNa (25424) 54 oth KCl

Method: (gelatinized cellulose acetate), electrophoresis

-----

dis R4N.X 20°C 0.50M U Cm+++ B3=6.46

Medium: NH4ClO4. By ix B3=5.78

\*

C3H9O3P HL CAS 38585-11-9 (4238)

1967ESa (25425) 55

```
Ethyl(hydroxymethyl)phosphinic acid; C2H5(HO.CH2).PO2H
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      ix R4N.X 25°C 0.20M U K1=1.78 1972EZd (27998) 56
Medium: NH4ClO4
**********************************
                 Squaric acid
                          CAS 2892-51-5 (439)
             H2L
3,4-Dihydroxy-3-cyclobutene-1,2-dione;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      ix R4N.X 25°C 1.00M U K1=2.34 B2=3.46 1972CSb (28641)
Medium: NH4ClO4
***********************************
                 L-Tartaric acid CAS 87-69-4 (92)
             H2L
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      dis oth/un 20°C 0.10M U K1=6.84
                                 1966STa (31219) 58
Medium: NH4Cl
______
    dis NaCl ? 0.10M U
                        B2=7.40
                                  1965MSd (31220) 59
Method: paper electrophoresis
**********************************
                           CAS 39156-77-9 (3008)
Hydrazine-N,N-diethanoic acid; H2N.N(CH2.COOH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     oth KNO3 25°C 0.10M U
                         K1=10.98 B2=19.97 1971LSc (33102)
                        K(Cm+HL)=4.13
Method: electrical migration or transference number
*************************
                           CAS 594-61-6 (81)
C4H803
2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Cm+++
     oth oth/un 25°C 0.10M U
                         K1=2.96 B2=5.15 1971SHb (33457) 61
                        B3=6.36
Method: electrical migration or transference number
______
     ix R4N.X 25°C 0.50M U
                         K1=2.46 B2=4.48 19560Ca (33458)
                                               62
                        K3=1.04
Medium: NH4ClO4
**********************************
                           CAS 37107-07-6 (4287)
Ethylenebis(iminomethylenephosphonous acid)
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     ix oth/un 25°C 0.50M U
                                  1971EZd (35829) 63
                        K(Cm+H2L)=6.40
****************
            H2L EDDPO CAS 1733-49-9 (2435)
C4H14N2O6P2
1,2-Diaminoethane-N,N'-bis(methylenephosphonic) acid; (H2O3P.CH2.NH.CH2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
      ix R4N.X 25°C 0.50M U
                                  1973EZa (35873) 64
                        K(Cm+H2L)=6.40
Medium: NH4ClO4
Cm+++ oth oth/un 25°C 0.10M U
                         K1=16.57 1971SHb (35874) 65
                        K(Cm+HL)=12.24
                         K(Cm+H2L)=7.80
                         K(Cm+H3L)=6.13
Method: electrical migration or transference number
*************************
             H3L
                Citric acid
                          CAS 77-92-9 (95)
C6H807
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Cm+++ dis NaClO4 25°C 0.10M U
                                  1974HHa (46062) 66
                        K(CmL2+6H)=14.23
                        K(Cm(HL)+5H)=9.63
 -----
                        K1=7.68 1971GBa (46063) 67
     dis oth/un 25°C 0.10M U
                        K(Cm+2H3L=CmHL2+5H)=-9.7
______
                        K1=7.93 B2=11.23 1971STe (46064)
Cm+++ oth oth/un 25°C 0.10M U
                        K(CmL+HL)=2.50
Constants obtained by survey of literature data
Cm+++ oth oth/un ? ? U
                                  1969MGf (46065) 69
                         K(CmOH+L)=5.3
                         K(CmOH+2L)=9.32
                        K(Cm(OH)2+L)=5.38
********************************
C6H9N06
             H3L
                 NTA
                          CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
------
Cm+++ cal NaClO4 25°C 0.50M C H K1=11.30 1987CRa (46744) 70
DH(K1)=-11.5 \text{ kJ mol}-1; DS(K1)=178 \text{ J K}-1 \text{ mol}-1
______
```

```
Cm+++ dis oth/un rt 6.00M U K1=11.18 B2=20.61 1975KPb (46745) 71
Method: distribution of Am betw. 1M trioctylamine in Toluole/EDTA in 6M
LiNO3 aq.; pH=3-4
_____
Cm+++ ix R4N.X 20°C 1.00M U K1=10.93 1971MOc (46746) 72
                        K(Cm+L+HL)=13.70
Medium: NH4Cl
-----
Cm+++ oth oth/un 20°C 0.10M U
                        K1=11.52 B2=19.57 1971SHb (46747) 73
                        K(Cm+L+HL)=13.72
Method: electrical migration or transference number
______
      oth none 25°C 0.00 M K1=13.53 1969MOc (46748) 74
Constants from survey of literature data
-----
Cm+++ ix R4N.X 25°C 0.10M U T K1=11.80 B2=20.58 1968EAa (46749)
                                               75
Medium: NH4ClO4
______
    dis R4N.X 20°C 0.10M U
                       B2=20.13
                                1966STa (46750) 76
Medium: NH4Cl
**********************************
            H2L HIMDA
                          CAS 93-62-9 (192)
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Cm+++
     oth KNO3 25°C 0.10M U
                                  1972SHb (48704) 77
                        K(Cm+HL)=9.20
                        K(Cm+2HL)=17.60
Method: electrical migration or transference number
-----
Cm+++ dis oth/un 25°C 0.10M U K1=9.27 1971EVb (48705) 78
                         K1=9.2 B2=16.7 1971SHb (48706) 79
      oth oth/un 25°C 0.10M U
Method: electrical migration or transference number
______
    ix R4N.X 25°C 0.10M U
                       K1=9.21 B2=17.13 1969EBa (48707) 80
Medium: NH4ClO4
**********************************
         H8L EDTPA
                          CAS 1429-50-1 (434)
Ethane-1,2-bis(iminobis(methylenephosphonic acid)); ((H2O3PCH2)2NCH2.)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                         K1=21.89
      oth KNO3 25°C 0.10M U
                                 1971SHb (52326) 81
                        K(Cm+HL)=17.74
                        K(Cm+H2L)=14.47
                        K(Cm+H3L)=10.83
                        K(Cm+H4L)=6.43
K(Cm+H5L)=4.73. Method: electrical migration or transference number
```

			010 10100	-0 4 (04.5-)
C7H11NO6 N-(2'-Carl	H3I boxyethyl)iminod	=	CAS 40199- ncid; HOOC.CH2.CH2.N(	-58-4 (3165) (CH2.COOH)2
Metal 	Mtd Medium Ter	mp Conc Cal	Flags Lg K values	Reference ExptNo
Cm+++	ix R4N.X 25	°C 0.10M U	K1=10.65 B2=1 K(Cm+HL)=4.12	17.95 1968EAa (56881)
Medium: NH *******		*****	******	*******
C7H15O3P	н	L		99-6 (4458)
Metal	Mtd Medium Ter	mp Conc Cal	Flags Lg K values	Reference ExptNo
 Cm+++ Medium: NH	H4C104			1972EZa (58025) 83
C8H502F3S	н	L TTA	CAS 326-91 1,3-dione; F3C.CO.CH	L-0 (165)
Metal	Mtd Medium Ter	mp Conc Cal	Flags Lg K values	Reference ExptNo
Cm+++	dis oth/un 25	°C 0.10M U	B3=13.40	1969KSa (58609) 84
******** C8H22N2O6F	************** P2 H4I	************* L EDDIPH	*******	**************************************
******** C8H22N2O6F Diaminoeth	**************************************	********* L EDDIPH opropylphosp	**************************************	**************************************
******** C8H22N2O6F Diaminoeth  Metal  Cm+++	******************** P2 H4I hane-N,N'-di(iso Mtd Medium Ter oth oth/un 25	************* L EDDIPH opropylphosp mp Conc Cal	CAS 13516- Chonic)acid; (CH2.NH.C) CH2.NH.CO CH2.NH.CO CH3.NH.CO C	**************************************
********* C8H22N2OGF Diaminoeth Metal Cm+++	**************  P2 H4I hane-N,N'-di(iso  Mtd Medium Ter  oth oth/un 25	********** L EDDIPH opropylphosp mp Conc Cal °C 0.10M U	CAS 13516-  Chonic)acid; (CH2.NH.C)  Flags Lg K values  K1=17.70  K(Cm+HL)=13.85  K(Cm+H2L)=9.04  K(Cm+H3L)=6.26	**************************************
**************************************	**************  P2	********** L EDDIPH opropylphosp mp Conc Cal °C 0.10M U  ation or tra ************************************	CAS 13516-  Chonic)acid; (CH2.NH.C)  Flags Lg K values  K1=17.70  K(Cm+HL)=13.85  K(Cm+H2L)=9.04  K(Cm+H3L)=6.26	**************************************
**************************************	**************************************	********** L EDDIPH opropylphosp mp Conc Cal °C 0.10M U  ation or tra ********* L oacetone; CF	CAS 13516- Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Flags Lg K values  K1=17.70  K(Cm+HL)=13.85  K(Cm+H2L)=9.04  K(Cm+H3L)=6.26  Consference number  CAS 326-06  CAS 326-06  CAS 326-06	**************************************
********* C8H22N2O6F Diaminoeth Metal Cm+++  Method : 6 ******* C10H7O2F3 3-Benzoyl Metal	**************************************	*********  L EDDIPH opropylphosp mp Conc Cal °C 0.10M U  ation or tra *********  L oacetone; CF mp Conc Cal	CAS 13516- Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Flags Lg K values  K1=17.70  K(Cm+HL)=13.85  K(Cm+H2L)=9.04  K(Cm+H3L)=6.26  Consference number  CAS 326-06  CAS 326-06  CAS 326-06	**************************************
********* C8H22N2O6F Diaminoeth Metal Cm+++  Method : 6 ******** C10H7O2F3 B-Benzoyl Metal Metal	**************  P2 H4I hane-N,N'-di(iso iso iso iso iso iso iso iso iso iso	********** L EDDIPH opropylphosp mp Conc Cal °C 0.10M U  ation or tra ******** L oacetone; CF mp Conc Cal C 0.10M U	CAS 13516- Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Flags Lg K values  K1=17.70  K(Cm+HL)=13.85  K(Cm+H2L)=9.04  K(Cm+H3L)=6.26  CAS 326-06	**************************************
**************************************	**************************************	*************** L EDDIPH opropylphosp mp Conc Cal °C 0.10M U  ation or tra ********* L oacetone; CF mp Conc Cal °C 0.10M U  ***********************************	CAS 13516- Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Chonic)acid; (CH2.NH.C) Flags Lg K values  K1=17.70  K(Cm+HL)=13.85  K(Cm+H2L)=9.04  K(Cm+H3L)=6.26  CAS 326-06  CA	**************************************

Cm+++ dis R4N.X 25°C 0.10M U 1971EVa (71040) 87 K(Cm+HL)=6.80K(Cm+2HL)=11.94Medium: 0.1 M NH4ClO4 \* H2L CAS 16598-05-3 (967) 2-Pyridylmethyliminodiethanoic acid; C5H4N.CH2.N(CH2.COOH)2 \_\_\_\_\_\_ Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo -----Cm+++ ix R4N.X 25°C 0.10M U K1=9.21 B2=17.69 1969EBa (71254) 88 Medium: 0.1 M NH4ClO4 \* H4L EDTA CAS 60-00-4 (120) C10H16N2O8 1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid; \_\_\_\_\_\_ Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo -----Cm+++ cal NaCl 25°C 2.0M U H K1=16.86 1985CLb (73662) 89 DH(K1) = -29.3 kJ mol -1\_\_\_\_\_\_ Cm+++ dis oth/un rt 6.00M U K1=17.33 1975KPb (73663) 90 Method: distribution of Am betw. 1M Trioctylamine in Toluol/EDTA in 6M LiNO3 aq.; pH=3-4 \_\_\_\_\_\_ Cm+++ oth KNO3 25°C 0.10M U T K1=17.10 1972SHc (73664) 91 K(Cm+HL)=9.32Method: electrical migration or transference number ix R4N.X ? 0.10M U I K1=17.29 1971EZb (73665) 92 Medium: (NH4ClO4), I= near zero, K1=19.95 Cm+++ oth oth/un 20°C 0.10M U K1=17.10 1971SHb (73666) 93 K(Cm+HL)=9.32Method: electrical migration or transference number \_\_\_\_\_\_ Cm+++ ix R4N.X 25°C 0.10M U K1=18.45 1957FSa (73667) 94 Medium: 0.1 M NH4ClO4 \* H3L HEDTA CAS 150-39-0 (392) N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid; \_\_\_\_\_\_ Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo -----B2=4.1 dis KCl 25°C 0.10M U 1971ZMa (75347) 95 K(Cm+L+HL)=2.82K(Cm+2HL)=1.0

C11H11N06 H3L CAS 1147-65-5 (425) N-(2'-Carboxyphenyl)iminodiethanoic acid; H00C.C6H4.N(CH2.C00H)2

\*

```
Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
 -----
      ix R4N.X 25°C 0.10M U K1=9.27
                                1969EBa (77825) 96
Medium: NH4ClO4
**********************************
                          CAS 4408-81-5 (923)
C11H18N2O8
            H4L
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.)2.CH2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      cal NaClO4 25°C 0.50M C H K1=13.05
                                1987CRa (79430) 97
DH(K1)=12.6 kJ mol-1; DS(K1)=292 J K-1 mol-1
Cm+++
                        K1=13.79 1985CMc (79431) 98
      dis NaCl 25°C 0.10M C
Method: extraction of 244Cm from 0.1 M NaCl (pH 5.5) into toluene/HDEHP.
****************************
C12H12N2O2
                          CAS 4173-74-4 (4915)
1-Phenyl-3-methyl-4-acetylpyrazol-5-one;
  -----
    Mtd Medium Temp Conc Cal Flags Lg K values
_____
     dis oth/un 25°C 0.10M U
                                1973BKc (81042) 99
                     B3=12.82
************************************
                           (3429)
1,1,1-Trifluoro-1'-naphthoylacetone;
 .....
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     dis oth/un 25°C 0.10M U
                                1969KSa (86872) 100
                      B3=18.17
************************************
                          CAS 1638-77-3 (5072)
(Methylenephosphinylmethyl)phenylphosphinic acid;
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      ix R4N.X 25°C 0.20M U I K1=3.35
                                1972EZb (88026) 101
Medium: NH4ClO4. I=0: K1=4.18
***********************************
            H4L
                CDTA
                          CAS 482-54-2 (200)
C14H22N208
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     dis R4N.X 20°C 0.10M U K1=18.79
                               1990GBc (88611) 102
Medium: NH4ClO4
______
     cal NaClO4 25°C 0.50M C H K1=18.10
                                1987CRa (88612) 103
```

DH(K1)=-9.	.7 kJ n	nol-1;	DS(K1)=	=314 J H				
In I=0, K1	L=21.62	<u> </u>			I K1=18.96 =19.49, K1(0.07)	1971EZc	(88613)	
					K1=18.34 K(Cm+HL)=9 nsference number	.30	(88614)	105
From surve	ey of ]	literat	ure dat	ta	K1=21.6			
Cm+++ Method: ic	onic mi	(Cl Igratio	20°C 0. on	.10M U	K1=18.7	1967SMa	(88616)	107
	ix F			.10M U	K1=18.81	1966BAc	(88617)	
Medium: NF ******** C14H23N3O1	14C1 ****** L0	*****	******* H5L	.10M U ****** DTPA	K1=18.40  ********  CAS 6  id; HOOC.CH2.N(C	1966STa ******** 7-43-6 (238)	(88618) ******	
Metal	Mtd M	nedium	Temp Co	onc Cal	Flags Lg K valu	es Refer	ence Exp	tNo
Cm+++		 +b /p	2000 0	 50M II	K1=22.85	1072DDc	(00100)	
Cm+++ Medium: NH	ix F				K1=23.81 K(Cm+HL)=1	 1971BRa		
Medium: NH	ix F 44ClO4 ix F	 R4N.X  R4N.X	25°C 0	.10M U	K1=23.81 K(Cm+HL)=1 K1=21.1	1971BRa 5.48 1971MOc	(89190)  (89191)	111
Medium: NH Cm+++ Medium: NH Cm+++	ix F H4ClO4  ix F H4Cl 	R4N.X R4N.X R4N.X Dth/un	25°C 0.	.10M U	K1=23.81 K(Cm+HL)=1	1971BRa 5.48 1971MOc 1971SHb	(89190)  (89191)	111
Medium: NH Cm+++ Medium: NH Cm+++ Method: el Cm+++ From surve	ix F H4ClO4 ix F H4Cl oth c	A4N.X A4N.X Ahrinan	25°C 0.  20°C 1  25°C 0.  gration  ? @cure dat	.10M U	K1=23.81 K(Cm+HL)=1 K1=21.1 K1=22.83 K(Cm+HL)=1 nsference number	1971BRa 5.48  1971MOc  1971SHb 4.40 . 1969MOc	(89190) (89191) (89192) (89192)	111
Medium: NH Cm+++ Medium: NH Cm+++  Method: el Cm+++ From surve	ix F H4ClO4 ix F H4Cl oth c lectric	R4N.X R4N.X oth/un cal migoth/un literat	25°C 0.  25°C 0.  gration  ? (cure dat	.10M U	K1=23.81 K(Cm+HL)=1 K1=21.1 K1=22.83 K(Cm+HL)=1 nsference number K1=25.7	1971BRa 5.48	(89190) (89191) (89192) (89192) (89193) (89194)	111 112 113 114
Medium: NH Cm+++ Medium: NH Cm+++  Method: el Cm+++ From surve Cm+++ Method: el Cm+++ Literature	ix F  H4ClO4  ix F  H4Cl  oth c  lectric  oth c  oth k  lectron  oth c  deta	R4N.X R4N.X R4N.X Ral migoth/un Riterat RNO3 Rigrati ROM (NO)	25°C 0.  25°C 0.  gration  25°C 0.  25°C 0.  25°C 0.	.10M U	K1=23.81 K(Cm+HL)=1 K1=21.1 K1=22.83 K(Cm+HL)=1 nsference number K1=25.7	1971BRa 5.48  1971MOc  1971SHb 4.40  1969MOc  1968LFb	(89190) (89191) (89192) (89192) (89193) (89194) (89194)	111 112 113 114 115

```
Medium: NH4ClO4
**********************************
                                             CAS 4551-69-3 (698)
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Cm+++
         dis oth/un 25°C 0.10M U
                                                        1973BKc (95876) 118
                                        B3=16.81
REFERENCES
 1999FKa T Fanghanel, T Konnecke, H Weger; J.Solution Chem., 28,447 (1999)
 1997BPa A Bahta, G Parker, D Tuck; Pure & Appl. Chem., 69,1489 (1997)
 1990GBc B Gorski, G Buklanov et al; Radiochim. Acta, 51,59 (1990)
 1989MKb P Mohapatra, P Khopkar; Polyhedron, 8, 2071 (1989)
 1987CRa G Choppin, E Rizkalla, J Sullivan; Inorg. Chem., 26, 2318 (1987)
 1985CLb G Choppin, Q Liu, J Sullivan; Inorg. Chem., 24, 3968 (1985)
 1985CMc G Choppin, A Muscatello; Inorg. Chim. Acta, 109, 67 (1985)
 1983MCb C Musikas, C Cuillerdier, J Livet et al; Inorg. Chem., 22, 2513 (1983)
 1982FKb T Fukusawa, I Kawasuji et al; Bull.Chem.Soc.Jpn.,55,726 (1982)
 1981LMa I Lebedev, Y Mazur; Radiokhim., 23, 359 (1981)
 1980KMa P Khopkar, J Mathur; J.Inorg.Nucl.Chem., 42,109 (1980)
 1980KMe P Khopkar, J Mathur; Thermochim. Acta, 37,71 (1980)
 1978RBa P Rao, S Bagawde et al; J.Inorg.Nucl.Chem., 40,123 (1978)
 1975KPb G Korpusov, E Patrusheva et al; Radiokhim., 17,512 (1975)
 1974HHa S Hubert, M Hussonois, L Brillard et al; J.Inorg.Nucl.Chem., 36,2361 (1974)
 1974KCa W Kinard, G Choppin; J.Inorg.Nucl.Chem., 36,1131 (1974)
 1974KMa P Khopkar, J Mathur; J.Inorg.Nucl.Chem., 36,3819 (1974)
 1973BKc W Bacher, C Keller; J.Inorg.Nucl.Chem., 35, 2945 (1973)
 1973CDd R Chiarizia, P Danesi, G Scibona et al; J.Inorg.Nucl.Chem., 35, 3595 (1973)
 1973EZa A Elesin, A Zaitsev, G Sergeev et al; Radiokhim., 15,64 (1973)
 1973HHd M Hussonois, S Hubert, L Brillard et al; Radiochem. Radioanal. Lett., 15,47
(1973)
 1973STe A Stepanov; Zh.Neorg.Khim., 18,371(E:194) (1973)
 1972CDb G Choppin, G Degischer; J.Inorg.Nucl.Chem., 34, 3473 (1972)
 1972CSb L Cilindro, E Stadlbauer, C Keller; J.Inorg. Nucl. Chem., 34, 2577 (1972)
 1972EZa A Elesin, A Zaitsev; Radiokhim., 14,370 (1972)
 1972EZb A Elesin, A Zaitsev, V Karaseva et al; Radiokhim., 14, 374(E:385) (1972)
 1972EZc A Elesin, A Zaitsev, S Kazakova, G Yakovlev; Radiokhim., 14,541 (1972)
 1972EZd A Elesin, A Zaitsev, N Ivanovich et al; Radiokhim., 14,546 (1972)
 1972HPb H Harmon, J Peterson et al; J.Inorg.Nucl.Chem., 34, 1381 (1972)
 1972MCc W McDowell, C Coleman; J.Inorg.Nucl.Chem., 34, 2837 (1972)
 1972PRc E Piskunov, A Rykov; Radiokhim., 14,2,260;265;330;332;641 (1972)
 1972SHb A Shalinets; Radiokhim., 14,1,33;2,269 (1972)
 1972SHc A Shalinets; Radiokhim., 14,2,275 (1972)
 1972SNa M Sakanoue, M Nakatani; Bull.Chem.Soc.Jpn.,45,3429 (1972)
 1972SSf A Shalinets, A Stepanov; Radiokhim., 14,280(E:290) (1972)
 1971BRa E Brandau; Inorg. Nucl. Chem. Lett., 7,1177 (1971)
 1971EVa V Ermakov, V Vorobeva, A Zaitsev et al; Radiokhim., 13,5,692 (1971)
```

```
1971EVb V Ermakov, V Vorobeva, A Zaitsev et al; Radiokhim., 13,6,840 (1971)
 1971EZb A Elesin, A Zaitsev; Radiokhim., 13, 5, 775 (1971)
 1971EZc A Elesin, A Zaitsev; Radiokhim., 13, 6, 902 (1971)
 1971EZd A Elesin, A Zaitsev, G Sergeev et al; Sci.Res.At.React.Inst.Rep., SRARI, 109
(1971)
 1971GBa R Guillaumont, L Bourderie; Bull.Soc.Chim.Fr., 2806 (1971)
 1971LSc B Levakov, A Shalinets; Radiokhim., 13,2,295 (1971)
 1971MOc A Moskvin; Radiokhim., 13,4,575;582;641 (1971)
 1971MOd A Moskvin; Radiokhim.,13,668;674;682(E:688;694;700) (1971)
 1971SHb A Shalinets; Radiokhim., 13, 4, 566 (1971)
 1971STe A Stepanov; Zh.Neorg.Khim., 16, 11, 2981 (1971)
 1971ZMa N Zaman, E Merciny, G Duyckaerts; Anal. Chim. Acta, 56, 271 (1971)
 1970ALc A Aziz, S Lyle; J.Inorg.Nucl.Chem., 32, 2383 (1970)
 1969ALd A Aziz, S Lyle; J.Inorg.Nucl.Chem., 31, 3471 (1969)
 1969DHa B Desire, M Hussonois et al; Compt.Rend., 269C, 448 (1969)
 1969EBa S Eberle, I Bayat; Inorg. Nucl. Chem. Lett., 5,229 (1969)
 1969GMa R Guillaumont, C de Miranda et al; Compt.Rend., 268C, 140 (1969)
 1969KSa C Keller, H Schreck; J.Inorg. Nucl. Chem., 31, 1121 (1969)
 1969MGf A Moutte, R Guillaumont; Rev.Chim.Minerale, 6, 603 (1969)
 1969MOc A Moskvin; Radiokhim., 11,458(E:447) (1969)
 1969NBa L Nugent, R Baybarz, J Burnett; J.Phys.Chem., 73,1177 (1969)
 1968ALd A Aziz, S Lyle, S Naqvi; J.Inorg. Nucl. Chem., 30, 1013 (1968)
 1968EAa S Eberle, S Ali; Z. Anorg. Allg. Chem., 361,1 (1968)
 1968LFb I Lebedev, V Filimonov et al.; Radiokhim., 10,93 (1968)
 1968TCa S Tanner, G Choppin; Inorg. Chem., 7, 2046 (1968)
 1967BEa M Borisov, A Elesin, I Lebedev; Radiokhim., 8, 2, 166 (1967)
 1967CCd R Carvalho, G Choppin; J.Inorg. Nucl. Chem., 29,725;737 (1967)
 1967ELa A Elesin, I Lebedev, E Piskunov, G Yakovlev; Radiokhim., 9, 161 (1967)
 1967ESa V Ermakov, I Stary; Radiokhim., 9, 197 (1967)
 1967SMa A Stepanov, T Makarova; Radiokhim., 9, 6, 710 (1967)
 1966BAc R Baybarz; J.Inorg.Nucl.Chem., 28, 1055 (1966)
 1966STa I Stary; Radiokhim., 8, 5, 504; 509 (1966)
 1966STb J Stary; Talanta, 13,421 (1966)
 1965BAc R Baybarz; J.Inorg.Nucl.Chem., 27, 1831 (1965)
 1965CKb G Choppin, J Ketels; J.Inorg.Nucl.Chem., 27, 1335 (1965)
 1965MIb J Miles; J.Inorg.Nucl.Chem., 27, 1595 (1965)
 1965MSd G Marcu, K Samochocka; Stud. Univ. Babes - Bolyai, 10,71 (1965)
 1963GRa I Grenthe; Acta Chem. Scand., 17, 1814 (1963)
 1962LYb I Lebedev, G Yakovlev; Radiokhim., 4,304 (1962)
 1960LPa I Lebedev, S Pirozhkov, G Yakovlev; Radiokhim., 2,559 (1960)
 1960LPb I Lebedev, S Pirozhkov, G Yakovlev; Radiokhim., 2,549 (1960)
 1957FSa J Foreman, T Smith; J.Chem.Soc., 1752 (1957)
 19560Ca B Odenheimer, G Choppin; J.Chem.Educ., 36,462 (1956)
 1956WWa M Ward, G Welch; J.Inorg.Nucl.Chem., 2, 395 (1956)
 1954FEa D Feay; Thesis, Berkeley, UCRL-2547 (1954)
```

## **EXPLANATORY NOTES**

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS
- H Data for THERMOCHEMICAL quantities

## EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC R or IUP=R signifies EVALUATION RATING = Recommended by IUPAC

-----

END