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Software version = 5.81 Data version = 4.62
Experiment list contains 1840 experiments for
(no ligands specified)
3 metals : Hg++, Hg2++, HgR+
(no references specified)
(no experimental details specified)
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e- Electron;	HL	Electron	(442)
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo	
Hg++	EMF	non-aq	25°C	100%	C	IH	$E^0(\text{Hg(l)}/\text{Hg}^{2+}) = -301 \text{ mV}$	1980APa	(516)	1
Medium: DMSO, 1 M NH4ClO4. E^0 referred to $E^0(\text{aq})=0$ for the Ag(s)/Ag+ elect.										
Hg++	sol	oth/un	25°C	dil	U	T	$K(\text{Hg(l)}=\text{Hg(aq)}) = -6.52$	19740Na	(517)	2
K=-5.79(80 C)										
Hg++	EMF	non-aq	25°C	100%	U		$K(\text{Hg} + \text{Hg(l)}=\text{Hg}^{2+}) = 1.6$	1974SLa	(518)	3
Medium: DMSO, 1 M MClO4(M=Li or Na)										
Hg++	EMF	non-aq	25°C	100%	U		$K(\text{Hg} + \text{Hg(l)}=\text{Hg}^{2+}) = 2.70$	1973CCa	(519)	4
Medium: acetonitrile										
Hg++	EMF	non-aq	25°C	100%	U		$K(\text{Hg} + \text{Hg(l)}=\text{Hg}^{2+}) = 3.4$	1972FDa	(520)	5
Medium: DMSO, 0.1 M LiClO4										
Hg++	sol	NaCl	25°C	6.10M	U	T	$K(\text{Hg(l)}=\text{Hg(aq)}) = -6.78$	1972GHa	(521)	6
Conc. of NaCl:in M units; K=-7.24(5 C), -6.38(45 C), -5.95(70 C); log K=-6.043 -1639.8/T+1.92301log T										
Hg++	EMF	oth/un	25°C	1.00M	U		$K = -28.1 (-0.83\text{V})$ $K' = -34.5 (-1.02\text{V})$	1972GKa	(522)	7
Medium: Na2SO4; K: HgSe(s)+2e=Hg(l)+Se--. K': HgTe(s)+2e=Hg(l)+Te--										
Hg++	EMF	oth/un	250°C	100%	U		$K(\text{Hg} + \text{Hg(l)}=\text{Hg}^{2+}) = 2.18$	1972SDa	(523)	8
Medium: (Na,K)NO3										
Hg++	EMF	non-aq	25°C	100%	U		$K(\text{Hg} + \text{Hg(l)}=\text{Hg}^{2+}) = 0.6$	1971BGa	(524)	9

Medium: N,N-dimethylacetamide, 0.1 M LiClO₄

Hg++ sol none 25°C 0.00 U T 1971GHd (525) 10

$K(\text{Hg(l)}=\text{Hg(aq)})=-6.54$

$K=-6.69(0\text{ C}), -6.58(20\text{ C}), -6.40(40\text{ C}), -6.17(60\text{ C}), -5.91(80\text{ C}),$
 $-5.62(100\text{ C}), -5.31(120\text{ C})$

Hg++ sp oth/un 125°C 100% U T H 1971TFa (526) 11

$K(\text{Hg} + \text{Hg(l)}=\text{Hg}_{2++})=5.32$

Medium: (Na,Al)Cl(n(Na):n(Al)=35:65); DH=-68.2 kJ mol⁻¹; K=4.34(175 C),
3.18(250 C) M units; Method: also current-voltage studies and Raman

Hg++ sp oth/un 125°C 100% U T H 1971TFa (527) 12

$K=4.48$

Medium: (Na,Al)Cl(n(Na):n(Al)=35:65); DH=-56.9 kJ mol⁻¹;
 $K(\text{Hg}_{++} + \text{Hg}_{3++}=2\text{Hg}_{2++})=3.65(175\text{ C}), 2.70(250\text{ C})$. Raman also used

Hg++ EMF oth/un 135°C 100% U 1969APa (528) 13

$K(\text{Hg} + \text{Hg(l)}=\text{Hg}_{2++})=4.6$

Medium: (Na,K,Al)Cl

Hg++ EMF oth/un 150°C 100% U 1968HPa (529) 14

$K(\text{Hg} + \text{Hg(l)}=\text{Hg}_{2++})=4.56$

Medium: (Na,K,Al)Cl

Hg++ EMF non-aq 20°C 100% U 1967KBa (530) 15

$K(\text{Hg}_{++} + \text{Hg(l)} = \text{Hg}_{2++})=0.7$

Medium: Me₂NCHO, 0.1 M HClO₄

Hg++ EMF NaClO₄ 25°C 2.0M U I 1962ZSa (531) 16

$K=30.95(915.44\text{ mV})$

K: $2\text{Hg}+2\text{e}=\text{Hg}_2$. In 2 M HClO₄ $K=31.12(920.36\text{ mV})$

Hg++ EMF none 25°C 0.0 U 1956HSa (532) 17

$K=30.68(907.5\text{ mV})$

K: $2\text{Hg}+2\text{e}=\text{Hg}_2$. $K(\text{Hg}+\text{Hg(l)}=\text{Hg}_2)=1.94$

Hg++ EMF none 20°C 0.0 U H 1954SAa (533) 18

$K(\text{Hg}+\text{Hg(l)}=\text{Hg}_2)=1.93$

DH(K)=3.4 kJ mol⁻¹, DS=25 J K⁻¹ mol⁻¹, 0 to 40 C

Br- HL Bromide CAS 10035-10-6 (19)

Bromide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO₄ 25°C 0.15M C H K1=6.49 2003AGb (1945) 19

Metal is CH₃Hg+. By calorimetry, DH(K1)=-37.6 kJ mol⁻¹, DS(K1)=
-1.8 J K⁻¹ mol⁻¹.

Hg++	sp	alc/w	20°C	100%	U				1991GAa	(1946)	20
							K3=3.08				
Medium: MeOH											
Hg++	sp	oth/un	20°C	var	U	M			1991GAb	(1947)	21
							K(HgI2+L)=1.92				
							K(HgI2L+L)=0.84				
Hg++	sp	non-aq	20°C	100%	C				1991GAc	(1948)	22
							K4=0.42				
Medium: methanol.											
Hg++	ISE	non-aq	25°C	100%	C	H	K1=16.2	B2=31.44	1990PDa	(1949)	23
							K3=5.74				
							K4=3.26				
Medium: Acetonitrile, 0.1 M (PyH)CF3SO3											
DH(K1)=-58.9, DH(K2)=-49.1, DH(K3)=-31.1, DH(K4)=-31.9 kJ mol ⁻¹											
Hg++	ISE	non-aq	25°C	100%	C	H	K1=10.01	B2=18.08	1985AIa	(1950)	24
							K3=5.13				
							K4=2.35				
Medium: DMSO, 1 M NH4ClO4. DH(K1)=-24.0 kJ mol ⁻¹ , DH(K2)=-32.9, DH(K3)=-27.7, DH(K4)=-21.5. In 0.1 M: K1=10.81, B2=19.75, B3=25.34; DH(K1)=-20.0											
Hg++	ISE	non-aq	25°C	100%	C	H	K1=10.73	B2=19.46	1985AIa	(1951)	25
							K3=4.71				
							K4=1.9				
Medium: Pyridine, 0.1 M (PyH)CF3SO3;											
DH(K1)=-13.6, DH(K2)=-8.7, DH(K3)=-14.4, DH(K4)=-7.5 kJ mol ⁻¹											
Hg++	sp	none	20°C	0.0	U	M			1984BSe	(1952)	26
							B(HgBrI)=3.64				
							B(HgBrCl)=2.46				
Hg++	sp	NaClO4	20°C	1.0M	U	I			1984GAc	(1953)	27
							K3=2.34				
							K4=1.91				
At I=0 corr. K3=2.23, K4=1.40											
Hg++	cal	oth/un	25°C	0.50M	U	TIH			1984VKc	(1954)	28
							K(HgBr2+Br)=2.10				
							K(HgBr2+2Br)=3.89				
DH(K3)=-10.96 kJ mol ⁻¹ ; DS(K3)=3.5 J mol ⁻¹ K ⁻¹ ;											
DH(HgBr2+2Br)=-26.23 kJ mol ⁻¹ ; DS=-13.5											
Hg++	oth	non-aq	25°C	100%	C	H	K1=12.92	B2=21.96	1981ABc	(1955)	29
							B3=27.62				
							B4=30.22				
Medium: DMSO, 0.1 M NH4ClO4. Mean values from potentiometry (amalgam) and calorimetry. DH(B1)=-20.0; DH(B2)=-44.1; DH(B3)=-68.5; DH(B4)=-85.8 kJ mol ⁻¹											

Hg++	ISE	non-aq	25°C	100%	C	H	K1=12.14 B3=25.33 B4=27.87	B2=20.20	1981APb	(1956)	30
Medium: DMSO, 1.0 M NH4ClO4. DH(K1)=-24.7; DH(B2)=-56.6; DH(B3)=-84.4; DH(B4)=-103.5 kJ mol-1											
Hg++	sp	NaClO4	20°C	var	U	M	K(HgI2+HgBr2=2HgIBr)=1.26 K(HgCl2+HgBr2=2HgClBr)=0.70 K(HgBr2+Cl)=0.23		1980GAc	(1957)	31
Hg++	cal	NaClO4	25°C	0.5M	U	TI	K1=9.14 DH(K1)=-41.8 kJ/mol DH(B2)=-85.8	B2=17.33	1980VKb	(1958)	32
For T=60 C: K1=8.43; B2=15.87; DH1=-34.9 kJ/mol; DH(Hg+2L)=-72.0 kJ/mol Also data for 0 M and 2 M NaClO4; T=40 C											
Hg++	sp	oth/un	20°C	0.01M	U	M	K(Hg(SCN)2+HgBr2)=0.81		1978BSa	(1959)	33
Hg++	cal	NaClO4	25°C	0.50M	C	H	DH(K1)=-42.7 kJ mol-1, DH(K2)=-46.4, DH(K3)=-12.6, DH(K4)=-14.4.		1977EKc	(1960)	34
Hg++	vlt	non-aq	25°C	100%	C		K3=7.0		1977GPb	(1961)	35
Medium: hexamethylphosphotriamide											
Hg++	sol	NaClO4	20°C	1.00M	U	M	B(Hg(Oxalate)Br)=9.09 B(Hg(Oxalate)Br2)=8.91		1975ZCa	(1962)	36
Hg++	oth	non-aq	42°C	100%	U	TIH	K(2HgI2=Hg2I4)=1.2		1974EAa	(1963)	37
Medium: C6H6. K=3.0(55.6 C). In C6H5CH3: K=2.3(42.1 C), K=1.7(55.6 C). Also p-xylene, mesitylene. Method: Vapor phase osmometry											
Hg++	EMF	non-aq	30°C	100%	U		K1=12.0 B3=17.7	B2=15.1	1974JAc	(1964)	38
Medium: pyridine, containing 0.2 M LiClO4. m units											
Hg++	vlt	NaClO4	?	var	U		B4=20.2		1974KIb	(1965)	39
Hg++	dis	R4N.X	55°C	?	U	T H	K3=2.15 K4=1.79		1974NGc	(1966)	40
Medium: NH4NO3(H2O)2. DH(K3)=-15.9 kJ mol-1, DH(K4)=-18.0; K3=2.04, K4=1.65 (70 C); K3=1.93, K4=1.54(85 C) m units											

Hg++	EMF non-aq	25°C	100%	U			1974SLa	(1967)	41
						B3=28.32 B4=29.20			
Medium: DMSO, 1 M MClO4(M=Li,Na)									
Hg++	EMF non-aq	25°C	100%	U		B2=35.4 B3=42.4 B4=46.7	1973CCa	(1968)	42
Medium: MeCN									
Hg++	gl	NaClO4	25°C	0.50M	U I	K1=9.04 B2=17.3 B(Hg(OH)L)=5.68	1972ALa	(1969)	43
K1=9.3, B2=17.5, B(Hg(OH)L)=5.90(I=3). redox electrode also used									
Hg++	EMF non-aq	?	100%	U		B2=20.2	1972BGa	(1970)	44
Medium: ethylene glycol, 0.1 M LiClO4									
Hg++	EMF non-aq	25°C	100%	U		B2=22.2 B3=28.0 B4=30.4	1972FDa	(1971)	45
Medium: DMSO, 0.1 M LiClO4									
Hg++	EMF non-aq	90°C	100%	U T H		B2=14.00 B4=18.18	1972KRa	(1972)	46
Medium: (Me2NH2)2SO4. DH(B4)=-7.9 kJ mol ⁻¹ ; B2=14.83, B4=16.82(101 C); B2=12.69, B4=15.75(115 C) x units									
Hg++	EMF non-aq	?	100%	U			1971BGa	(1973)	47
Medium: N,N-dimethylacetamide, 0.1 M LiClO4									
Hg++	sp	NaClO4	20°C	0.10M	U M		1971BPd	(1974)	48
K(HgCl2+L=HgL+2Cl)=2.37 K(HgCl2+2L=HgL2+2Cl)=4.08 K(HgCl3+L=HgL+3Cl)=2.40 K(HgCl3+2L=HgL2+3Cl)=4.0									
Medium: HClO4. K(HgCl3+3L=HgL3+3Cl)=5.4. K(HgCl4+L=HgL+4Cl)=2.5, K(HgCl4+2L=HfL2+4Cl)=4.3, K(HgCl4+3L=HgL3+4Cl)=5.4, K(HgCl4+4L=HgL4+4Cl)=6.2									
Hg++	EMF	NaClO4	25°C	0.40M	U	K1=9.07 B2=17.21 B3=19.8 B4=21.8	1970DSe	(1975)	49
Medium: HClO4									
Hg++	vlt non-aq	25°C	100%	U		B2=24.17 B3=31.62 B4=34.73	1970MDa	(1976)	50
Medium: DMF, containing 0.1 M Et4NClO4									
Hg++	dis	NaClO4	25°C	0.50M	U		1970SIa	(1977)	51

$$K_4 = 1.60$$
$$K(\text{HgA}+\text{L})=5.8$$
$$K_s(\text{HgBr}_2(s) + \text{Hg} = \text{Hg}_2\text{Br}_2) = -0.7$$
$$K(\text{Hg}(\text{EDTA}) + \text{L}) = 5.5$$

B3=9.5

 $K_4 = 1.2$
$$K_3 = 2.76$$
$$K_4 = 1.49$$

Medium: 0.5 M NaClO₄, 0.1 HClO₄, DH(K1)=-46.4 kJ mol⁻¹(7 C), -42.6(25 C), -41.8 (40 C); DH(B2)=-94.5(7 C), -88.7(25 C), -89.5(40 C). Also DS values

$$K(\text{Hg}(\text{CF}_3)_2 + \text{L}) = 0.2$$
$$K_d(\text{HgBr}_2(\text{salt melt})) = 0.96$$
$$K3=0.90$$
$$K_4 = 0.95$$
$$K_3 = 2.26$$
$$K_4 = 1.38$$
$$K1/K2 = \text{ca. } 0.9$$

K3=2.1

 $K_4 = 1.7$

DH(K1)=-42.3, DH(K2)=-44.8, DH(K3)=-13, DH(K4)=-17, DH(B2)=87.0, DH(B4)=-117 J K⁻¹ mol⁻¹.

Hg++ ix NaClO4 23°C 3.0M U I 1963EMb (1989) 63

K3=1.6

K4=1.0

Method:anion exchange. At I=0.3: K3=2.5, K4=2.1

Hg++ EMF NaClO4 40°C 0.60M U T K1=8.70 B2=16.46 1963HIa (1990) 64

Medium:0.5 M NaClO4,0.1 HClO4. At 7 C:K1=9.53, K2=8.85

Hg++ cal none 25°C 0.0 U H 1961MPa (1991) 65

I=0 corr. DH(K1)=-44.4 kJ mol⁻¹. DS=40 J K⁻¹ mol⁻¹

Hg++ cal NaClO4 25°C 0.50M U H 1960GKa (1992) 66

Medium: HClO4. DH(B4)=-116 kJ mol⁻¹. DS=13 J K⁻¹ mol⁻¹

Hg++ sp NaClO4 25°C 0.50M U T H 1958PAa (1993) 67

K3=2.10

K4=1.5

K(HgL2(s)=HgL2)=-1.18

K3=2.23(1.2 C), 2.20(13.8 C); K4=1.1(1.2 C), 1.91(13.8 C); K=-2.03(5.1 C), -1.66(40 C). DH(K3)=-8.8 kJ mol⁻¹, DH(K)=18 J K⁻¹ mol⁻¹, DS=26 J K⁻¹ mol⁻¹

Hg++ dis NaClO4 25°C 0.50M U T K1=2.13 1958STa (1994) 68

K4=1.36

K(HgL4+2L)=0.5

K3=2.36(5 C), 2.04(35 C). DH(K3)=-18 kJ mol⁻¹, DS=-20; DH(K4)=-7.9, DS=0
K4=1.41(5 C), 1.26(35 C). m units

Hg++ sol oth/un ? var U 1957KPa (1995) 69

K(HgO(s)+H2O+3L=HgL3+2OH)=0.3

Hg++ dis NaClO4 25°C 0.0 U I 1957MAa (1996) 70

Kd(HgL2 into C6H6)=0.05

At I=0.5 M, Kd=0.15

Hg++ EMF NaClO4 25°C 0.50M U K1=8.94 B2=16.88 1957MAa (1997) 71

K3=2.27

K4=1.75

Hg++ con non-aq 25°C 100% U 1952ECa (1998) 72

K3=6.00

K4=2.04

K(2HgBr2+Br=Hg2Br5)=7.00

Medium:MeCN.

Hg++ ISE NaClO4 25°C 0.50M U K1=9.05 B2=17.33 1948BJb (1999) 73

K3=2.41

K4=1.26

B4=21.00									
Hg++	EMF	NaCl04	25°C	0.50M	C			1946SIa	(2000) 74
						K(HgBr2+Hg=2HgBr)=0.76			
Method: Pt/Hg(II),Hg(I) electrode.									
Hg++	sol	none	25°C	0.0	U			1939GAa	(2001) 75
						K3=2.04			
						K(HgBr2(s)=HgBr2)=-1.77			
						K(HgBr2(s)+Br=HgBr3)=0.27			
Hg++	sp	oth/un	22°C	var	U			1933FLa	(2002) 76
						K3.K4=4.15			
Medium: KBr									
Hg++	sp	oth/un	16°C	var	U			1928JOa	(2003) 77
						K3.K4=5.80			
Medium: KBr.									
Hg++	sol	oth/un	34°C	dil	U T			1924TPa	(2004) 78
						K(HgBr2(s)=HgBr2)=-1.72			
K=-1.39(55.5 C), -1.15(78 C), -0.92(96.5 C)									
Hg++	dis	oth/un	25°C	dil	U			1906PAa	(2005) 79
						B4=21.89			
Hg++	ISE	oth/un	25°C	var	U			1903SHa	(2006) 80
						B2=17.18			
						B4=21.63			
						K3.K4=4.45			
Method: Hg electrode. Medium: KBr. Kd(HgL2 into C6H6)=0.05									
Hg++	sol	oth/un	25°C	var	U			1902MOa	(2007) 81
						K1=9.40 B2=17.7?			
Medium: Hg(NO3)2									

CN-		HL	Cyanide		CAS 74-90-8 (230)				
Cyanide;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaCl04	25°C	2.00M	U		B2=34.01	1981BPb	(2690) 82
Hg++	vlt	oth/un	22°C	0.01M	U		B2=35.3	1981KTc	(2691) 83
						B3=38.9			
						B4=42.0			
Hg++	sp	oth/un	25°C	2.00M	U	M		1980BPa	(2692) 84
						B(HgBr3(CN))=6.60			
						B(HgBr2(CN)2)=14.40			
						B(HgBr(CN)3)=17.18			
						B(Hg(CN)4)=19.51			

Hg++	sp	NaCl	25°C	2.00M	U			1977BPb	(2693)	85
							K(HgCl ₄ +4CN=Hg(CN) ₄)=26.32			
Hg++	EMF	NaClO ₄	25°C	1.00M	U			1972FAa	(2694)	86
							K(Hg+Cr(III)CN=CrNCHg)=7.5 K(CrCN+Cr(CN)Hg)=7.6			
Hg++	sp	none	25°C	0.0	U	M		1971BPf	(2695)	87
							K(Hg(CN) ₂ +Fe(II)(CN) ₆)=2.4 K(Hg(CN) ₂ +Mo(IV)(CN) ₈)=1.3 K(Hg(CN) ₂ +Ru(II)(CN) ₆)=1.89. By potentiometry: K(Hg(CN) ₂ +Fe(III)(CN) ₈)=2.2, K(Hg(CN) ₂ +Mo(V)(CN) ₈)=1.3			
Hg++	sp	NaClO ₄	25°C	2.00M	U			1971EBa	(2696)	88
							K(Hg+Cr(III)CN)=1.45			
Medium: LiClO ₄ . By kinetics, K=1.38										
Hg++	con	oth/un	0°C	var	U			1971GHf	(2697)	89
							K(2HgI ₂ +2HF=Hg ₂ I ₃ +HL+HF ₂)=-0.5			
Medium: HF										
Hg++	sp	none	25°C	0.0	U			1971GMa	(2698)	90
							B(Hg(SCN)(CN))=-0.28			
Hg++	ISE	KNO ₃	30°C	3.0M	U			1968ADc	(2699)	91
							B ₅ =40.5 B ₆ =42.6 B ₇ =38.2(?)			
Hg++	vlt	NaNO ₃	25°C	2.0M	U	M		1968AMa	(2700)	92
							B(Hg(CN) ₂ SCN)=36.3 B(Hg(CN) ₃ SCN)=38.7 B(Hg(CN) ₂ Cl)=36.2 B(Hg(CN) ₃ Cl)=39.4			
B(Hg(CN) ₂ Br)=36.0, B(Hg(CN) ₃ Br)=39.1										
Hg++	oth	oth/un	25°C	5.0M	U		K ₁ =2.27 B ₂ =5.99	1968CPa	(2701)	93
Method: Raman spectra. Medium: NaI										
Hg++	sp	oth/un	25°C	var	U	I		1968CPa	(2702)	94
							K(HgI ₂ +HgI ₂ =2HgLI)=-0.89			
Medium: HgL ₂ . In 0.4 M HgI ₂ , by solubility: K=-0.85										
Hg++	ISE	mixed	25?°C	20%	U	I	K ₁ =22.9 B ₂ =26.5	1968PMe	(2703)	95
							B ₃ =28.4			
Also by: polarography. Medium: pyridine/H ₂ O. B ₂ =28.6(80%) and other values										
Hg++	vlt	non-aq	195°C	100%	U		B ₂ =6.50	1967ETa	(2704)	96
Medium: molten KSCN										

Hg++ ISE oth/un 40°C 0.0 U T H T K1=16.26 B2=31.28 1965CIa (2705) 97
K3=3.37
K4=2.42

Medium:0 corr. 10 C: K1=17.97, K2=16.74, K3=3.81, K4=2.81; 25 C: K1=17.00,
K2=15.75, K3=3.56, K4=2.66

Hg++ cal oth/un 25°C 0.0 U H 1965CIa (2706) 98
DH(K1)=-96.1 kJ mol⁻¹, DH(K2)=-106.6, DH(K3)=-31.8, DH(K4)=-30.1;
DS(K1)=2.9 J K⁻¹ mol⁻¹, DS(K2)=-56.0, DS(K3)=-37.6, DS(K4)=-50.6

Hg++ ISE oth/un 25°C var U B2=36.57 1964ASa (2707) 99
B3=40.64
B4=43.54
B5=44.40
B6=45.19

Hg++ sp oth/un 25°C dil U I M 1964BGB (2708) 100
K(HgL2+HgCl2=2HgLC1)=0.93
K(HgL2+HgBr2=2HgLBr)=0.29
K(HgL2+HgI2=2HgLI)=-0.96
In dioxan: K=1.08(Cl), 0.24(Br), -0.7(I)

Hg++ vlt none 25°C 0.0 U B2=35.21 1961MUa (2709) 101
B3=38.85
B4=41.47

Hg++ vlt NaNO3 30°C 2.0M U I M 1961NHa (2710) 102
K(HgL3+Cl)= -0.4
K(HgL3+Br)=0.36
K(HgL3+SCN)=0.46
In 4M NaCl04 (spectrophotometry) K(HgL2+Cl)= -0.25

Hg++ oth oth/un ? var U 1961PJa (2711) 103
K4/K3=-1
Method: ir.

Hg++ sol NaNO3 23°C 2.0M U 1959NHa (2712) 104
K(Hg(OH)2+HgL2=2Hg(OH)L)=2.42
K(HgO(s)+HgL2=2Hg(OH)L)= -0.99
By polarography: K(Hg(OH)2+HgL2=2Hg(OH)L)=2.45. At 30 C: B(Hg(OH)L)=28.86

Hg++ vlt NaNO3 30°C 2.0M U B2=33.9 1958NCa (2713) 105
K3=4.2
K4=2.5

Hg++ gl NaNO3 20°C 0.10M U T K1=18.00 B2=34.70 1957ANc (2714) 106
K3=3.83
K4=2.98
B4=41.52

Hg++	EMF	none	25°C	0.0	U		1957TMb	(2715)	107
						B4=41.51			
Method: emf with Hg electrode									
Hg++	vlt	none	25°C	0.0	U		1957TMb	(2716)	108
						B2=35.25 K3=3.71			
Hg++	oth	NaCl04	25°C	var	U		1954WDa	(2717)	109
						K2/K1=-1.3			
Medium: 0.01 to 1 M; method:chemical analysis; 0-25 C									
Hg++	sp	NaCl04	?	1.0M	U		1953PEc	(2718)	110
						K1/K2=-0.95			
Hg++	cal	oth/un	?	?	U	HM	1951YAa	(2719)	111
DH(B3)=-224.7, DH(B4)=-246.9 kJ mol ⁻¹ . DH(HgL2Cl)=-195.8, DH(HgL2Br)=-197.5, DH(HgL2I)=-205.9, DH(HgL2I2)=-210.9									
Hg++	gl	oth/un	rt	var	U		1949GAa	(2720)	112
						B2=21.7 B4=27.7(?)			
Hg++	gl	oth/un	2°C	dil	U		1946BJa	(2721)	113
						K(HgL2+HL=Hg _L 3+H)=-5.52 K(HgL2+2HL=Hg _L 4+2H)=-12			
Hg++	ISE	oth/un	12°C	var	U		1941BJa	(2722)	114
						K4=ca.3			
Hg++	ISE	oth/un	12°C	var	U		1932BDa	(2723)	115
						B4=41.7 to 43.2			
Hg++	oth	oth/un	100°C	var	U		1928BRa	(2724)	116
						K(HgL2+Cl)=-0.11 K(HgL2+Br)=0.06			
Method: boiling point									
Hg++	sol	oth/un	20°C	var	U		1909Hwa	(2725)	117
						K(HgL2+NO3)=-0.36 Ks(HgL2(s)=Hg _L 2)=-0.40 K(HgL2+OH=Hg _L 2(OH))=0.2			
Hg++	EMF	oth/un	25°C	var	U		1906PAa	(2726)	118
						B4=41.65			
Medium: not specified; method: emf with Hg electrode									
Hg++	EMF	oth/un	25°C	var	U		1905SSa	(2727)	119
						B4=41.66			
Medium: not specified; method: emf with Hg electrode									

Hg++ EMF oth/un ? var U 1903B0a (2728) 120

B4=41.4

Medium: not specified; method: emf with Hg electrode.

Hg++ EMF oth/un 25°C var U 1903SHa (2729) 121

B4=41.4

Medium: not specified; method: emf with Hg electrode

Hg++ dis oth/un 25°C var U 1903SHa (2730) 122

K(HgL2(aq)=HgL2(Et20))=-1.64

CO3-- H2L Carbonate CAS 465-79-6 (268)

Carbonate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl none 25°C 0 C R 2005PBa (3237) 123

K(Hg(OH)2+CO2(g)=HgCO3)=-0.70

K(Hg(OH)2+HCO3=Hg(OH)CO3)=0.98

IUPAC evaluated. K(Hg(OH)2+CO2(g)+H=HgHCO3)=3.63

Ks(HgCO3.2H2O(s)+3H2O=3Hg(OH)2+CO2(g))=-11.27 (3M NaClO4)

Hg++ gl NaClO4 25°C 0.50M U K1=11.01 B2=14.50 1980BMb (3238) 124

B(HgHL)=15.08

K(Hg+H2O+L=Hg(OH)L+H)=4.40

Hg++ ISE NaClO4 25°C 3.00M C 1976HHa (3239) 125

*Ks=7.20

*Ks: HgL.2HgO(s)+6H=3Hg+3H2O+CO2(g)

Hg++ sol NaClO4 25°C 3.00M C K1=10.65 1976HHb (3240) 126

B(1,1,1)=15.05

B(-1,1,1)=4.40

B(p,q,r): pH+qHg+rL=Hp(Hg)qLr

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl none 25°C 0 C H R K1=7.31 B2=14.00 2005PBa (4939) 127

K3=0.93

K4=0.61

K(Hg+HgCl2=2HgCl)=0.61

B(HgH-1Cl)=4.27

IUPAC evaluated (R). DH(K1)=-21.3 kJ mol⁻¹, DH(B2)=-49.1, DH(K3)=0.5

DH(K4)=-10.5. D(SIT) values:K1=-0.22(04),B2=-0.39(03),K3=0.01(05)

Hg++ gl NaClO4 25°C 0.15M C H K1=5.21 2003AGb (4940) 128

Metal is CH3Hg+. By calorimetry, DH(K1)=-23.7 kJ mol⁻¹, DS(K1)=

20.4 J K⁻¹ mol⁻¹.

Hg ⁺⁺	sp	alc/w	20°C	100%	U			1991GAa	(4941)	129
							K3=1.28			

Medium: MeOH

Hg ⁺⁺	sp	oth/un	20°C	var	U	M		1991GAb	(4942)	130
							K(HgBr ₂ +L)=3.17			
							K(HgBr ₂ L+L)=0.25			
							K(HgI ₂ +L)=0.99			
							K(HgI ₂ L+L)=-0.64			

Hg ⁺⁺	sp	non-aq	20°C	100%	C			1991GAc	(4943)	131
							K4=0.75			

Medium: methanol.

Hg ⁺⁺	ISE	non-aq	25°C	100%	C	H	K1=15.1	B2=29.61	1990PDa	(4944)	132
							K3=2.81				
							K4=2.90				

Medium: Acetonitrile, 0.1 M (PyH)CF₃SO₃

DH(K1)=-50.5, DH(K2)=-41.5, DH(K3)=-31.1, DH(K4)=-19.7 kJ mol⁻¹

Hg ⁺⁺	ISE	non-aq	25°C	100%	C	H	K1=8.89	B2=15.94	1985AIa	(4945)	133
							K3=3.97				
							K4=1.97				

Medium: Dimethylsulfoxide 1M NH₄ClO₄;

DH(K1)=-19.2, DH(K2)=-29.7, DH(K3)=-20.1, DH(K4)=-16.5 kJ mol⁻¹

Hg ⁺⁺	ISE	non-aq	25°C	100%	C	H	K1=10.59	B2=18.99	1985AIa	(4946)	134
							K3=4.60				
							K4=2.64				

Medium: Pyridine, 0.1 M (PyH)CF₃SO₃;

DH(K1)=-14.9, DH(K2)=-6.0, DH(K3)=-10.4, DH(K4)=-6.8 kJ mol⁻¹

Hg ⁺⁺	sp	NaClO ₄	20°C	0.50M	U	I			1984GAc	(4947)	135
							K3=0.81				

At I=0 corr. K3=0.70, K4=0.50

Hg ⁺⁺	gl	NaClO ₄	25°C	0.50M	C		K1=6.76	B2=13.24	1982NBb	(4948)	136
							B3=14.19				
							B4=15.24				

Hg ⁺⁺	cal	NaClO ₄	25°C	0.50M	U	TIH			1982VKa	(4949)	137
							K3=1.03				
							K(HgL ₂ +2L)=1.98				

DH(K3) = -0.25 kJ mol⁻¹; DH(HgL₂+2L) = -9.38 kJ mol⁻¹.

Hg ⁺⁺	ISE	non-aq	25°C	100%	C	H	K1=10.87	B2=17.97	1981APb	(4950)	138
							B3=21.96				
							B4=24.04				

Medium: DMSO, 1.0 M NH₄ClO₄. DH(K₁)=-20.3 kJ mol⁻¹, DH(B₂)=-48.8,
DH(B₃)=-68.9, DH(B₄)=-82.9

Hg++ cal NaClO₄ 25°C 0.5M U TI K₁=6.74 B₂=13.22 1980VKb (4951) 139
DH(K₁)=-23.6 kJ mol⁻¹
DH(B₂)=-51.7

For T=60 C: K₁=6.37; B₂=12.37; DH₁=-16.8 kJ/mol; DH(Hg+2L)=-40.0 kJ/mol
Also data for 0 M and 2 M NaClO₄; T=40 C

Hg++ sp alc/w 20°C 100% U M 1979GAa (4952) 140
K(HgI₂+HgCl₂=2HgICl)=1.09
K(HgBr₂+HgCl₂=2HgBrCl)=0.70

Medium: MeOH

Hg++ sp oth/un 20°C 0.01M U M 1978BSa (4953) 141
K(Hg(SCN)₂+HgCl₂)=1.0

Hg++ vlt non-aq 25°C 100% C 1977GPb (4954) 142
K₃=5.8

Medium: hexamethylphosphotriamide

Hg++ gl NaClO₄ 25°C 3.00M C I M K₁=7.22 B₂=14.00 1977SJb (4955) 143
K₃=1.07
K₄=1.07
B(-1,1,-1)=-9.87
B(-1,2,-2)=-15.25

B(-2,3,-5)=-37.79

pH+qHgCl₂+rCl=Hp(HgCl₂)qCl_r; B(p,q,r)

Hg++ oth NaNO₃ 25°C 1.00M U 1976BAb (4956) 144
K₃=0.894
K(HgCl₂+2Cl)=2.025

Method: Refractometry.

Hg++ cal NaClO₄ 25°C 1.0M C K₁=6.72 B₂=13.23 1975CGe (4957) 145
K₃=1.00
K₄=0.97

DH(K₁)=23.2 kJ mol⁻¹, DS=51 J K⁻¹ mol⁻¹; DH(K₂)=27.7, DS=32;

DH(K₃)=1.0, DS=16; DH(K₄)=7.6, DS=-7

Hg++ oth non-aq 42°C 100% U TIH 1974EAa (4958) 146
K(2HgI₂=Hg₂I₄)=1.8

Medium: C₆H₆. Method: vapour phase osmometry. K=3.6(55.6 C).

In C₆H₅CH₃: K=2.0(42 C), 1.8(55.6 C)

Hg++ EMF non-aq 30°C 100% U K₁=8.8 B₂=11.1 1974JAc (4959) 147
B₃=13.1

Medium: pyridine, 0.2 M LiClO₄; m units

Hg++ EMF oth/un ? var U 1974KIb (4960) 148

B4=15.1

Hg++ EMF R4N.X 55°C ? U T H K1=5.30 B2=10.30 1974NGb (4961) 149
Medium: $\text{NH}_4\text{NO}_3(\text{H}_2\text{O})_x (x=2)$. $\text{DH}(\text{K}_1) = -30.1 \text{ kJ mol}^{-1}$, $\text{DS} = 10.5 \text{ J K}^{-1} \text{ mol}^{-1} (x=2)$.
 $\text{K}_1 = 5.12$, $\text{K}_2 = 4.85 (x=2, 70^\circ\text{C})$, $\text{K}_1 = 4.92$, $\text{K}_2 = 4.58 (x=2, 85^\circ\text{C})$ x units

Hg++ dis R4N.X 55°C ? U T H 1974NGc (4962) 150
K3=1.18
K4=0.68
Medium: $\text{NH}_4\text{NO}_3(\text{H}_2\text{O})_2$. $\text{DH}(\text{K}_3) = -26.4 \text{ kJ mol}^{-1}$; $\text{DH}(\text{K}_4) = 27.2$, $\text{K}_3 = 0.96$, $\text{K}_4 = 0.88 (70^\circ\text{C})$
 $\text{K}_3 = 0.81$, $\text{K}_4 = 1.04 (85^\circ\text{C})$ m units

Hg++ EMF non-aq 25°C 100% U B2=21.11 1974SLa (4963) 151
B3=26.93
B4=28.32
Medium: DMSO, 1 M (Li,Na)ClO₄

Hg++ sp alc/w 25°C 60% U T K1=3.0 B2=9 1973ABd (4964) 152
Medium: 60% v/v EtOH/H₂O, 0.5 M LiClO₄. K or log K ?

Hg++ ISE none 25°C 0.0 C IH K2=6.53 1973ACa (4965) 153
K3=1.22
 $\text{K}(\text{HgCl}_2(\text{s}) = \text{HgCl}_2(\text{aq})) = -0.59$

Data also for various MeOH/water mixtures

Hg++ ISE alc/w 25°C 8.1% U I K2=6.93 1973ACa (4966) 154
K3=1.06
Medium: 8.1% w/w MeOH/H₂O. $\text{K}_2 = 6.53$, $\text{K}_3 = 1.22 (0\%)$; $\text{K}_2 = 6.90$, $\text{K}_3 = 1.30 (16.5\%)$

Hg++ EMF non-aq 25°C 100% U B2=35.1 1973CCa (4967) 155
K3=6.0
K4=4.2

Medium: MeCN

Hg++ EMF non-aq ? 100% U B2=16.3 1972BGa (4968) 156
Medium: ethylene glycol, 0.1 M LiClO₄

Hg++ EMF NaClO₄ 25°C 1.0M U 1972CIa (4969) 157
 $\text{K}(\text{Hg} + \text{OH} + \text{L} = \text{HgOHL} + \text{H}) = 3.67$

Hg++ EMF non-aq 25°C 100% U B2=21.2 1972FDa (4970) 158
B3=26.9
B4=23.5

Medium: DMSO, 0.1 M LiClO₄, Et₄NClO₄

Hg++ EMF non-aq 90°C 100% U T H B2=12.20 1972KRa (4971) 159
B4=14.70
Medium: dimethylammoniumsulphate. $\text{DH}(\text{B}_4) = -5.1 \text{ kJ mol}^{-1}$; B2=11.38, B4=13.51 (101°C); B2=9.85, B4=12.75(115°C) x units

Hg++	EMF non-aq 25°C 100% U		1971BGa (4972) 160
		B3=37.7	
Medium: N,N-dimethylacetamide			

Hg++	EMF NaClO4 25°C 0.40M U		1970DSe (4973) 161
		K3=0.96	
		K4=0.38	
Medium: HClO4			

Hg++	vlt non-aq 25°C 100% U	B2=24.16 B3=31.23 B4=34.57	1970MDa (4974) 162
Medium: DMF, 0.1 M Et4NClO4			

Hg++	dis NaClO4 25°C 0.50M U		1970SIa (4975) 163
		K3=0.85	
		K4=1.00	

Hg++	oth none 50°C 0.0 U T	K1=5.99 B2=12.51 B3=14.50	1969HEa (4976) 164
Evaluated from literature data. At 100 C: K1=5.65, B2=11.41, B3=13.25; 150 C: K1=5.50, B2=10.71, B3=12.46			

Hg++	sp NaCl ? 0.10M U M		1969NOc (4977) 165
		K(HgA+L)=3.9	
H6A=methylthymolblue			

Hg++	dis oth/un 200°C 100% U T M		1969ZAa (4978) 166
		Kd(HgLA(melt))=HgLA(org)=0.49	
Medium: (Li,K)NO3, LiNO3 content 43%. org=polyphenol. A=Br. When X=I, K=0.78. Many other data, 165 to 200 C			

Hg++	EMF NaClO4 25°C 1.0M U		1968CGa (4979) 167
		K(Hg(OH)2+2H+2L)=19.6	
		K(HgOHL+H+L)=9.56	
		K(HgClOH+H)=3.1	

Hg++	EMF NaClO4 25°C 1.0M U	K1=6.72 B2=13.23 K3=1.00 K4=0.97 B4=15.20	1968CGB (4980) 168

Hg++	ISE mixed 25°C 20% U I	K1=2.2 B2=4.3	1968PMe (4981) 169
Also polarography. Medium:20% pyridine. B2=7.9,B3=10(80%). Also in 10,40,60%			

Hg++	oth oth/un 23°C var U	K2=>2 K3 > 1	1968SCc (4982) 170
Method:electrical migration or transference number. Medium:LiCl var			

Hg++	vlt non-aq 25°C 100% U T	B2=31.1	1966FSa (4983) 171

B3=37.1

B4=39.3

Medium:MeCN

Hg++ EMF NaClO4 35°C 0.50M U T K2=6.35 1966VSb (4984) 172
K1/K2=0.24

K2=6.87(0 C),6.67(12.5 C),6.48(25 C); K1/K2=0.31(0 C),0.28(12.5 C),0.26(25C)

Hg++ ISE NaClO4 25°C 3.0M U H K1=7.07 B2=13.98 1965ARa (4985) 173
K3=0.75
K3K4=2.13

By calorimetry: DH(K1)=-24.2 kJ mol⁻¹, DH(K2)=-27.2, DH(K3)=-4.18,
DH3+DH4=-10.5. DS(K1)=53.9 J K⁻¹ mol⁻¹,DS(K2)=41.4,DS(K3)=0, DS3+DS4=5.4

Hg++ EMF NaClO4 25°C 0.10M U B2=15.5 1965BPg (4986) 174
Medium:HClO4

Hg++ gl oth/un 25°C 0.0 U H 1965PIa (4987) 175
K(Hg(OH)2+Cl=HgClOH+OH)=-4.09
K(HgOHCl+Cl=HgCl2+OH)=-3.77

By calorimetry:DH(Hg(OH)2+Cl)=5.0 kJ mol⁻¹, DS=-61.0 J K⁻¹ mol⁻¹;
DH(HgOH+Cl)=5.1, DS=-55.2

Hg++ cal NaClO4 24°C 0.60M U T H 1964CIa (4988) 176
DH(K1)=-28.2kJ mol⁻¹, DS=37.6 J K⁻¹ mol⁻¹(8 C); -23.0,49.7(25 C); -23.4,
60.0(40 C). DH(B2)=-58.7,DS=57.3(8 C); -53.3,69.0(25 C); -52.3,76.9(40 C)

Hg++ dis non-aq 150°C 100% U H 1964ZMb (4989) 177
Kd(HgL2(melt)=HgL2(org))=0.38
K3=0.8
K4=0.8

Medium: (Li/K)NO3 eutectic; org=polyphenyl eutectic; m unit.
DH(Kd)=-15.5 kJ mol⁻¹

Hg++ ix NaClO4 23°C 3.0M U I 1963EMb (4990) 178
K3=0.70
K4=0.60

Method:anion exchange. At I=0.3:K3=1.1, K4=1.2

Hg++ EMF NaClO4 40°C 0.60M U T K1=6.58 B2=12.76 1963HIa (4991) 179
K1=7.23(7 C),6.62(25 C); K2=6.72(7 C),6.36(25 C)

Hg++ ix oth/un 25°C 10.0M U I 1962MIa (4992) 180
K(H+HgL4)=0.6
K(H+HHgL4)=-0.05

Medium: LiCl.

Hg++ cal none 25°C 0.0 U H 1961MPa (4993) 181
I=0 corr. DH(K1)=-20 kJ mol⁻¹, DS=71 J K⁻¹ mol⁻¹

Hg++ cal NaClO4 25°C 0.50M U H 1960GKa (4994) 182
Medium: HClO4. DH(K1)=-25 kJ mol⁻¹, DS=46.4 J K⁻¹ mol⁻¹; DH(K2)=-29, DS=26;
DH(K3)=-9.2, DS=-13; DH(K4)=0.4; DS=21; DH(B4)=-62.3, DS=81.6

Hg++ vlt non-aq 20°C 100% U 1959GPa (4995) 183
B4=ca.30
Medium: HCONMe2, 1 M LiClO4

Hg++ sp NaClO4 25°C 0.50M U T 1958PAa (4996) 184
K3=0.66
K3K4=1.76
K3=1.12(1.5C), 0.83(39.7C), K3K4=2.04(1.5C), 1.72(39.7C). DH(K3)=-10 kJ m⁻¹,
DS=-2.5 J K⁻¹ mol⁻¹; DH(B2)=-56

Hg++ dis NaClO4 25°C 0.50M U K1=6.74 B2=13.22 1957MAa (4997) 185
K3=0.95
K4=1.05
Kd(HgCl2=HgCl2(in C6H6))=-1.06 (I=0), -0.96(I=0.5)

Hg++ gl oth/un RT var U M K1=6.8 B2=13.4 1955DWb (4998) 186
K3=0.57
K4=1.46
K(HgOH+H+2Cl=HgCl2+H2O)=12.40

Hg++ sp NaClO4 18°C 1.0M U 1953PEc (4999) 187
K3=0.50
K1/K2=0.40

Hg++ con non-aq 25°C 100% U 1952ECa (5000) 188
K3=6.00
K4=2.23
K(2HgL2+L=Hg2L5)=6.70
Medium: MeCN

Hg++ EMF NaClO4 25°C 0.50M U K1=6.74 B2=13.22 1947LJa (5001) 189
K3=0.85
K4=1.00
B4=15.07

Hg++ EMF NaClO4 25°C 0.50M C 1946SIa (5002) 190
K(HgCl2+Hg=2HgCl)=0.255
Method: Pt/Hg(II),Hg(I) electrode.

Hg++ sol none 25°C 0.0 U 1939GAa (5003) 191
Ks(HgCl2(s)=HgCl2)=-0.57
Ks(HgCl2(s)+Cl=HgCl3)=1.30
K3=1.87

Hg++ sol KCl 25°C var U T 1934TOa (5004) 192
K(2HgL2+L=Hg2L5)=1.26

K=1.51(0 C), 1.16(34 C), 0.72(100 C); K3K4=1.66(34 C), 1.89(100 C)

Hg++	sp	KCl	22°C	var	U		1933FLa	(5005)	193
						K3K4=2.06			

Hg++	oth	oth/un	100°C	var	U		1928BRa	(5006)	194
						K3K4=1			
						K(3HgL2=Hg3L6)=-0.41			

Hg++	sp	KCl	16°C	var	U		1928JOa	(5007)	195
						K3=0.70			

Hg++	dis	NaCl	25°C	var	U		1916LIa	(5008)	196
						K3=0.99			
						K4=0.69			
						K(Hg2L4+L=Hg2L5)=1.57			
						K(Hg2L5+L=Hg2L6)=1.28			

Hg++	dis	oth/un	25°C	var	U T		1915LIa	(5009)	197
						K(2HgL2+2L=Hg2L6)=-0.55			

Medium: HgCl2. K=-0.44(40 C). By freezing point, 0 C, K=-0.79. 100 C(B. Pt)
K=-0.17. Kd(HgL2 into C6H6)=-1.08(25 C), -1.03(40 C)

Hg++	dis	oth/un	25°C	var	U		1912DRa	(5010)	198
						K3=1.13			

By EMF B2=14.03

Hg++	sol	oth/un	25°C	var	U	K1=7.28	1908BHa	(5011)	199
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Hg++	dis	KCl	25°C	var	U		1907SBa	(5012)	200
						K3=1.06			

Hg++	ISE	KCl	25°C	var	U		1906PAa	(5013)	201
						B4=16.20			

Hg++	con	oth/un	25°C	var	U		1904LHa	(5014)	202
						K(2HgL2+H2O=Hg2L2O+2H+2L)=-16			

Hg++	EMF	KCl	25°C	var	U	B2=14.0	1903SHa	(5015)	203
						B4=15.95			
						K3K4=1.95			

Hg++	sol	oth/un	25°C	var	U	K1=7.46	B2=14.01	1902MOa	(5016)	204
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Hg++	EMF	oth/un	16°C	var	U	B2=13.82	1901LUa	(5017)	205
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CrO4-- H2L Chromate CAS 7738-94-5 (2382)

Chromate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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 Hg++ sol non-aq 147°C 100% U 1969MBd (6492) 206
 log Kso=-1.37-3537/T

Medium: fused (Li,K)NO3, 43% Li. Temp: 147-192 C

F- HL Fluoride CAS 7644-39-3 (201)
 Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	sp	oth/un	20°C	var	U	M		1979GAb (6949)	207
							K(HgCl2+L)=0.12		
							K(HgBrL+L)=-0.85		
							K(HgI2+L)=-0.96		

 Hg++ EMF NaClO4 25°C 0.50M U T H K1=1.03 1955PAa (6950) 208
 K1=1.01(15 C), 1.05(35 C). DH(K1)=3.6 kJ mol⁻¹, DS=33 J K⁻¹ mol⁻¹. At I=0
 K1=1.56, DS=55

FClBrI HL (541)
 Halides, comparative (for book data under ligand 80)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	vlt	non-aq	20°C	100%	U			1966BKa (7394)	209
							B4(Cl)=32.9		
							Medium: DMF, 0.1 M LiClO4. B4=35.2(Br), 36.6(I), 22.5(SCN), 26.5(SC(NH2)2)		

Hg++	sol	oth/un	25°C	1.0M	U	I M		1966CZb (7395)	210
							B(HgBrI2)=25.7		
							B(HgBr2I2)=27.35		
							Ks(HgI2)=-28.19		

Medium: LiNO3. Data also in other nitrates

Hg++	con	alc/w		100%	U	M		1965CEa (7396)	211
							K((F3CSe)2Hg+HgCl2)=1.3		

Also Raman spectra, solubility, infrared spectra. Medium: MeOH. K=1.18(Br), 0.31(I), 0.39(SCN), low for CN

Hg++	dis	non-aq	25°C	100%	U	IHM		1964ELa (7397)	212
							K'(HgClBr)=1.16		
							K'(HgClI)=1.50		
							K'(HgBrI)=0.76		

Medium: C6H6. In 0.5 M NaClO4: K'(HgClBr)=1.20

Hg++	dis	non-aq	150°C	100%	U	M		1964ZMa (7398)	213
							K(HgCl2+Cl+Br)=2.79		
							K(HgCl2+2Br)=2.61		
							K(HgCl2+2I)=1.77		
							K(HgBr2+2I)=1.52		

Medium:(Li/K)NO3 eutectic. Data also for many other mixed complexes

Hg++ dis non-aq 150°C 100% U M 1964ZMb (7399) 214

Kd(HgClBr)=0.59

K(HgCl2+HgBr2=2HgClBr)=1.85

K(HgCl2+HgI2=2HgClI)=-0.35

Med:(Li/K)NO3 eutectic. Kd: HgXY(melt)=HgXY(polyphenyl eutectic)

Kd=1.95(HgClI), 2.57(HgBrI)

Hg++ sp oth/un 23°C 1.0M U M 1963SHa (7400) 215

K(HgBr4+I=HgBr3I+Br)=3.20

K(HgBr4+2I=HgBr2I2+2Br)=5.49

K(HgBr4+3I=HgBrI3+3Br)=7.40

K(HgBr4+4I=HgI4+4Br)=8.76

Hg++ sp NaClO4 25°C .001M U M 1961SHc (7401) 216

K(HgCl2+HgBr2=2HgClBr)=1.14

K(HgCl2+HgI2=2HgClI)=1.35

K(HgBr2+HgI2=2HgBrI)=1.07

Hg++ sol NaNO3 20°C 0.50M U M 1960CZa (7402) 217

K(Hg(SCN)2(s)+Br)=0.43 ?

Hg++ sp alc/w ? 100% U M 1957DEa (7403) 218

K(HgCl2+HgBr2=2HgClBr)=0.3

Medium: MeOH

Hg++ dis NaClO4 25°C 0.50M U M 1957MAa (7404) 219

K(HgCl2+HgBr2=2HgClBr)=2.0

K(HgCl2+HgI2=2HgClI)=1.75

K(HgBr2+HgI2=2HgBrI)=1.10

K(HgBrI+Br)=1.85

K(HgBr2I+Br)=1.49, K(HgBrI2+Br)=1.88, K(HgI3+Br)=1.10. Also Kd(C6H6) values

I- HL Iodide CAS 10034-85-2 (20)

Iodide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.15M C H K1=8.61 2003AGb (8030) 220

Metal is CH3Hg+. By calorimetry, DH(K1)=-54.8 kJ mol-1, DS(K1)=

-19 J K-1 mol-1.

Hg++ sp alc/w 20°C 100% U 1991GAa (8031) 221

K3=5.26

Medium: MeOH

Hg++ sp non-aq 20°C 100% C 1991GAc (8032) 222

K4=-0.40

Medium: methanol.

Hg++ ISE non-aq 25°C 100% C H K1=17.4 B2=34.27 1990PDa (8033) 223
K3=7.41
K4=3.02

Medium: Acetonitrile, 0.1 M (PyH)CF₃SO₃
DH(K1)=-67.6, DH(K2)=-59.6, DH(K3)=-41.1, DH(K4)=-42.4 kJ mol⁻¹

Hg++ ISE non-aq 25°C 100% C H K1=11.48 B2=21.23 1985AIa (8034) 224
K3=5.96
K4=2.47

Medium: Dimethylsulfoxide 1M NH₄ClO₄;
DH(K1)=-32.4, DH(K2)=-40.4, DH(K3)=-28.1, DH(K4)=-23.0 kJ mol⁻¹

Hg++ ISE non-aq 25°C 100% C H K1=11.72 B2=21.55 1985AIa (8035) 225
K3=4.83
K4=2.0

Medium: Pyridine, 0.1 M (PyH)CF₃SO₃;
DH(K1)=-18.3, DH(K2)=-16.8, DH(K3)=-16.2, DH(K4)=-4.3 kJ mol⁻¹

Hg++ sp none 20°C 0.0 U M 1984BSe (8036) 226
B(HgClI)=5.88
B(HgBrI)=3.64

Hg++ sp NaClO₄ 20°C 0.50M U I 1984GAc (8037) 227
K3=3.69
K4=2.38

At I=0 corr. K3=3.79, K4=2.03

Hg++ gl NaClO₄ 25°C 0.50M C K1=12.87 B2=23.82 1982NBb (8038) 228
B3=27.60
B4=29.83

Hg++ ISE non-aq 25°C 100% C H K1=13.52 B2=23.27 1981APb (8039) 229
B3=29.28
B4=31.91

Medium: DMSO, 0.1 M NH₄ClO₄. DH(K1)=-33.2; DH(B2)=-72.6; DH(B3)=-100.7;
DH(B4)=-121.8 kJ mol⁻¹

Hg++ sp NaClO₄ 20°C var U M 1980GAc (8040) 230
K(HgCl₂+HgI₂=2HgClI)=1.40
K(HgBr₂+HgI₂=2HgBrI)=1.26
K(HgCl₂+I)=0.98
K(HgBr₂+I)=0.52

Hg++ ISE mixed 25°C 82% U K1=26.22 B2=29.23 1979TBc (8041) 231
B3=32.00
B4=35.79

Medium: 82%(vol) DMF/H₂O; In 82% Formamide, K1=22.38, B2=25.20, B3=28.00,
B4=31.74

Hg++	vlt non-aq 25°C 100% C		1977GPb (8042) 232
		K3=7.2	
	Medium: hexamethylphosphotriamide		
Hg++	oth non-aq 42°C 100% U T H		1974EAa (8043) 233
		K(2HgI2=(HgI2)2)=1.2	
		K(3HgI2=(HgI2)3)=3.1	
	Medium: benzene, 42.1 C. At 55.6 C K(2HgI2=(HgI2)2)=0, K(3HgI2=(HgI2)3)=3.5.		
	Also toluene (0.9,2.4),p-xylene and mesitylene. Method:vapor phase osmometry		
Hg++	EMF non-aq 30°C 100% U	K1=18.8 B2=23.6 B3=23.4	1974JAc (8044) 234
	Medium:pyridine, 0.2 M LiClO4		
Hg++	EMF non-aq 25°C 100% U		1974SLa (8045) 235
		B3=30.36	
		B4=32.34	
	Medium: DMSO, 1 M (M)ClO4(M=Li,Na)		
Hg++	oth alc/w ? 100% U I M		1974YMc (8046) 236
		K(HgBr4+L=HgBr3L+Br)=0.66	
	Medium: MeOH. K=1.15 in formamide. K=0.45 in DMF. Method: Raman		
Hg++	gl NaClO4 25°C 0.50M U		1973AHa (8047) 237
		K(Hg+I+H2O=Hg(OH)I+H)=8.9	
Hg++	sol NaClO4 25°C 0.50M U M		1973BPc (8048) 238
		K(HgL2+Cl)=0.75	
		K(HgL2+Br)=1.64	
		K(HgL3+L)=5.7	
		K(HgL2Cl+Cl)=1.18	
		K(HgI2Br+Br)=3.10	
Hg++	vlt non-aq 25°C 100% U	B2=38.2 B3=45.9 B4=48.7	1973CCa (8049) 239
	Medium: MeCN		
Hg++	EMF oth/un 20°C dil U T HM		1972BPe (8050) 240
		K(HgCl2+L=HgClL+Cl)=5.77	
		K(HgClL+L=HgL2+Cl)=4.87	
	DH(HgCl2+L=HgClL+Cl)=-46.9 kJ mol ⁻¹ ; K=6.05(10 C), 5.48(30 C)		
	DH(HgClL+L=HgL2+Cl)=-51.4; K=5.14(10 C), 4.52(30 C)		
Hg++	EMF non-aq 25°C 100% U	B2=24.2 B3=30.4 B4=32.6	1972FDa (8051) 241
	Medium: DMSO, 0.1 M LiClO4		
Hg++	EMF non-aq 90°C 100% U T	B2=19.46	1972KRa (8052) 242

B4=22.46
 Medium: dimethylammoniumsulphate; B2=17.49, B4=20.94(101 C); B2=16.00,
 B4=19.51(115 C) x units

Hg++ EMF non-aq ? 100% U 1971BGa (8053) 243
 B4=38.3

Medium: N,N-dimethylacetamide

Hg++ sol alc/w 25°C 100% U I 1971PKd (8054) 244
 Ks(HgI2(s)+I=HgI3)=0.43

Medium: MeOH; Ks=0.73 in acetone

Hg++ vlt non-aq 25°C 100% U B2=25.93 1970MDa (8055) 245
 B3=33.34
 B4=36.02

Medium: DMF, 0.1 M Et4NClO4

Hg++ sp non-aq 25°C 100% U 1970PLb (8056) 246
 K3=5.74
 K4=1.88

Medium: DMSO, 0 corr

Hg++ dis NaClO4 25°C 0.50M U 1970SIa (8057) 247
 K3=3.78
 K4=2.23

Hg++ vlt R4N.X 127°C 100% U K1=2.00 B2=3.95 1969PVa (8058) 248
 B3=4.95
 B4=5.09

Medium: Et4NCl

Hg++ sp alc/w 25°C 96% U 1968AJa (8059) 249
 K3=4.87

Medium: 96% MeOH

Hg++ ISE mixed 25°C 20% U I B2=14 1968PMe (8060) 250
 B3=17.5
 B4=19.3

Also polarography. Medium: 20% C5H5N. In 90%: B2=16.1, B3=20.1, B4=20.9
 Also values for w=10, 40, 60% pyridine

Hg++ sol oth/un 20°C 5.20M U M 1968YGa (8061) 251
 Ks(AgI(s)+Hg=AgIHg)=-1.14

Medium: Ca(NO3)2

Hg++ sol non-aq 25°C 100% U 1967KWb (8062) 252
 Ks2=-3.99
 Ks(Hg2I2(s)=Hg(l)+HgI2)=-5.85

Medium: iso-octane. In MeOH: Ks=-2.84

Hg++	dis	NaClO4	25°C	0.50M	U		1967Mwa	(8063)	253
							Kd(HgI2=HgI2(in CCl4))=1.1		
Hg++	cal	NaClO4	24°C	0.60M	U	T H	1964CIa	(8064)	254
Medium:0.5 NaClO4,0.1 HClO4. DH(K1)=-79.0(8 C) kJ mol-1, -75.2(25 C), -72.3 (40 C); DH(B2)=-142.1(8 C),-142.7(25 C),-144.2(40 C), with other values									
Hg++	sol	NaClO4	25°C	1.0M	U		B2=26.0 B3=27.65 B4=29.3 Kso=-28.19	1964CZc	(8065) 255
Hg++	dis	non-aq	150°C	100%	U	H	1964ZMb	(8066)	256
							Kd(HgI2(melt)=HrI2(org))=1.58 K3=1.18 K4=1.11		
Medium:(Li/K)NO3 eutectic. DH(D)=-11.3 kJ mol-1, org=polyphenyl eutectic									
Hg++	ix	NaClO4	23°C	3.0M	U	I		1963EMb	(8067) 257
							K3=3.0 K4=1.4		
Method:anion exchange. At I=0.3:K3=3.8, K4=2.6									
Hg++	EMF	NaClO4	40°C	0.60M	U		K1=12.40 B2=22.93	1963HIa	(8068) 258
							K(HgL2(s)=HgI2)=-3.87 K(HgL2(s)=HgI+L)=-14.42		
Also solubility. Medium:0.5 NaClO4,0.1 HClO4. At 7 C: K1=13.58, K2=11.63, K(HgL2(s)=HgI+L)=-16.20; K(HgL2(s)=HgI2)=-4.43(8C),-4.09(25C)									
Hg++	sol	NaClO4	24°C	0.60M	U	T H		1963HIa	(8069) 259
Medium:0.5 NaClO4,0.1 HClO4.DH(HgL2(S)=HgI2)=31.8(16 C) kJ mol-1,29.3(25 C), 29.3(25 C), 26.8(32 C)									
Hg++	sp	alc/w	25°C	100%	U			1961Dwa	(8070) 260
							K3=4.9		
Medium: MeOH									
Hg++	cal	none	25°C	0.0	U	H		1961MPa	(8071) 261
I=0 corr. DH(K1)=-73.6 kJ mol-1, DS=10 J K-1 mol-1									
Hg++	cal	NaClO4	25°C	0.50M	U	H		1960GKa	(8072) 262
Medium: HClO4, DH(B4)=-186 kJ mol-1, DS=-56.1 J K-1 mol-1									
Hg++	vlt	non-aq	20°C	100%	U			1959GPa	(8073) 263
							K3= ca.4 to 5 K4=2.79 B4=38.79		
Medium: HCONMe2, 0.1 LiClO4									
Hg++	sp	NaClO4	25°C	0.50M	U	T H		1958PAa	(8074) 264

$$K_4 = 2.39$$
[illegible]

Hg++ dis NaCl04 25°C 0.50M U K1=12.87 B2=23.82 1957MAa (8077) 267
K3=3.67
K4=2.37

At $I=0$ corr. $K3=3.71$, $K4=2.02$

Hg++ cal oth/un 25°C var U H 1954Ysa (8080) 270
DH(K1)=-69.5 kJ mol⁻¹, DS=25 J K⁻¹ mol⁻¹

Hg++ ISE oth/un ? var U 1953G0a (8081) 271

B4=30.18

Hg++ con non-aq 25°C 100% U T 1952ECa (8082) 272
 K3=5.95
 K4=1.61
 B(HgL2)2L)=6.95
Medium: MeCN. At 50 C K3=5.85, K4=1.48, B=6.85

Hg++ sol oth/un 25°C var U K1=13.15 1952YSa (8083) 273
B(Hg2L)=13.75

Hg++ sol oth/un 25°C var U K1=12.36 1951YAc (8084) 274
B(Hg2L)=13.62

Hg++ sol NaClO4 25°C 0.50M U I 1949BSb (8085) 275
 K(HgL2(s)=HgL2)=-4.13
 At I=0 corr. K=-3.98

Hg++ ISE NaCl04 25°C 0.50M U K1=12.87 B2=23.82 1949QSa (8086) 276
K3=3.78
K4=2.23

Hg++	vlt oth/un	17°C	0.0	U		1946MSa	(8087)	277
					B4=27			
Hg++	EMF NaCl04	25°C	0.50M	C		1946SIa	(8088)	278
					K(HgI2+Hg=2HgI)=2.06			
Method: Pt/Hg(II),Hg(I) electrode.								
Hg++	sol none	25°C	0.0	U		1941BRa	(8089)	279
					K(HgL2(s)=HgL2)=-4.01			
Hg++	sol none	25°C	0.0	U		1939GAa	(8090)	280
					K3=3.56			
					K4=1.86			
					K(HgL2(s)=HgL2)=-3.88			
					K(HgL2(s)+L=HgL3)=-0.32			
Hg++	sol none	18°C	0.0	U		1937TPa	(8091)	281
					K(HgL2(s)=HgL2)=-4.12			
Hg++	sp oth/un	22°C	var	U		1933FLa	(8092)	282
					K3.K4=5.04			
Hg++	sp oth/un	16°C	var	U		1928JOa	(8093)	283
					K3.K4=7.92			
Hg++	sol alc/w	25°C	100%	U I		1908HKa	(8094)	284
					K(HgL2(S)=HgL2)=-1.16			
Medium: MeOH. In EtOH: K=-1.40								
Hg++	ISE oth/un	25°C	var	U		1906PAa	(8095)	285
					B4=30.53			
Hg++	sol non-aq	25°C	100%	U		1903SHa	(8096)	286
					K(HgL2(s)=HgL2)=-2.31			
					Kd(HgL2 into C6H6)=1.58			
Medium: C6H6. In KI var, by Hg electrode, B2=24.42, B4=30.28, K3.K4=5.86								
Hg++	sol oth/un	25°C	var	U	K1=13.40 B2=25?	1902MOa	(8097)	287
					K(HgL2(s)=HgL2)=-3.88			

NH3	L	Ammonia			CAS 7664-41-7	(414)		
Ammonia								
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Hg++	vlt	non-aq	25°C	100%	U		B2=17.0	1973MTa (9160) 288
Medium: Me2NCHO, 0.5 M Et4NCl04								
Hg++	ISE	NaCl04	25°C	1.0M	U		B2=17.8	1964WDa (9161) 289

Hg++ vlt R4N.X 25°C 0.10M U 1962TRa (9162) 290
B4=19.26

Medium: NH4NO3.

Hg++ cal R4N.X 27°C 2.0M U H 1957YMa (9163) 291
Medium: NH4NO3. DH(B2)=-103.3 kJ mol⁻¹; DH(K3)=-13.8; DH(K4)=-15.1;
DS(B2)=-9.6; DS(K3)=-26.8; DS(K4)=-34.7.

Hg++ cal R4N.X rt 3.0M U H 1952FYa (9164) 292
Medium: NH4NO3. DH(B4)=-119.2 kJ mol⁻¹; DS=-32.2 J K⁻¹ mol⁻¹

Hg++ gl R4N.X 22°C 2.0M U K1=8.8 B2=17.5 1941BJa (9165) 293
K3=1.00
K4=0.78
B4=19.3

Medium: NH4NO3

NH3O L Hydroxylamine; CAS 5470-11-1 (1808)
Hydroxylamine; NH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE mixed 20°C 50% U K1=10.94 B2=12.88 1980IPa (9265) 294
B3=14.87
B4=15.78

Medium: 50% EtOH/acetone

NO2- HL Nitrite CAS 7782-77-6 (635)
Nitrite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE NaCl04 25°C 1.00M C K1=5.94 B2=9.91 1988EAa (9373) 295
B3=11.45
B4=11.86

Hg++ sol NaCl04 20°C 1.00M U M 1975ZCa (9374) 296
B(Hg(Oxalate)(NO2))=9.34
B(Hg(Oxalate)(NO2)2)=9.32

Hg++ sp oth/un 20°C 0.0 U K1=2.41 B2=3.85 1973CZa (9375) 297

Hg++ sp oth/un 20°C 0.0 U M 1973CZa (9376) 298
B(HgClL)=5.35
K(HgClL3)=23.15
K(HgCl2L2)=20.59

Hg++ sp NaCl04 25°C 0.01M U I 1971TLa (9377) 299
B4=11.6

I=0.012. When I=0.005, B4=11.2

Hg++ gl oth/un ? var U 1932Bwa (9378) 300
B4=14.9

Hg++ ISE oth/un 25°C var U 1906PAa (9379) 301
B4=13.54

NO3- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hg++ oth oth/un 25°C var U K1=0.11 B2=0.11 1968DIa (9702) 302
Methods: Raman spectra, infrared spectra.

Hg++ oth mixed 23°C 90% U K1=0.83 B2=1.56 1966Wfa (9703) 303
Medium: 90% i-PrOH, 0.5 M HNO3

Hg++ EMF NaNO3 25°C var U K1=0.35 1956HSd (9704) 304

Hg++ EMF NaClO4 25°C 3.0M U K1=0.11 K2=-0.1 1946ISa (9705) 305

N3- HL Azide CAS 7782-79-8 (441)
Azide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hg++ sp NaClO4 25°C 1.00M U H K1=6.98 B2=14.39 1976AAa (10229) 306
DH(K1)=-30.4 kJ mol⁻¹; DH(K2)=-37.1

Hg++ EMF non-aq 25°C 100% U 1971BGa (10230) 307
B4=28.5
Medium: MeCONMe2, 0.1 M LiClO4

Hg++ sp NaClO4 28°C 0.25M U I K1=7.15 B2=14.12 1965MKa (10231) 308
K1=7.30(I=0.15),7.48(I=0.05),7.75(I=0); K2=7.08(I=0.15),7.26(0.05),7.49(I=0)

Hg++ cal oth/un 25°C 0.0 U H 1956GWc (10232) 309
DH(Kso(Hg2L2(s)))=125 kJ mol⁻¹

Hg++ ISE oth/un 25°C 0.0 U 1952SUa (10233) 310
Kso(Hg2L2(s))=-9.15

OCN- HL Cyanate CAS 661-20-1 (6165)
Cyanate, Fulminate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hg++ vlt alc/w 25°C 25% U B2=29.8 1970KSb (10298) 311

B4=36.4

Medium: 25% MeOH

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl none 25°C 0 C H R 2005Pba (11558) 312
*K1=-3.40
*B2=-5.98
*B3=-21.1

IUPAC evaluated (R). DH(*K1)=26.2 kJ mol⁻¹, DH(*B2)=51.5

*Kso(HgO(s)+2H=Hg+H2O)=2.37, DH=-25.3. D(SIT): *K1=-0.14(3), *K2=-0.14(2)

Hg++ gl NaClO4 25°C 0.15M C H K1=9.18 2003AGb (11559) 313
K(CH3Hg+CH3HgOH)=2.32

Metal is CH3Hg+. By calorimetry, DH(K1)=-35.8 kJ mol⁻¹, DS(K1)=55.7
J K⁻¹ mol⁻¹; DH(K)=-16.1, DS(K)=-9.

Hg++ gl oth/un 25°C 0.10M C K1=10.20 B2=21.20 1987GGc (11560) 314
Medium: KSO3CH3

Hg++ gl NaClO4 25°C 0.10M U K1=-3.10 1984PTa (11561) 315

Additional method: Hg electrode.

Hg++ oth none 25°C 0.0 U K1=10.6 B2=21.83 1983RCa (11562) 316
Recalculation of literature data

Hg++ gl NaClO4 25°C 3.0M C K1=-3.48 1979CFb (11563) 317
*B2=-6.18

Hg++ gl NaClO4 25°C 3.00M C K1=-3.48 1979CIa (11564) 318
*B2=-6.18

Hg++ gl NaNO3 30°C 1.00M U K1=-3.2 1978THa (11565) 319
*K2=-3.0

Hg++ gl NaClO4 25°C 3.00M C K1=-3.58 1977SJB (11566) 320
*B(1,2)=-6.228
*B(2,2)=-4.84

Hg++ cal NaClO4 25°C 1.00M U HM K(HgClOH+H)=3.05 1976CGB (11567) 321
K(HgSCNOH+H)=3.4

$K(\text{Hg}(\text{OH})_2 + \text{H}) = 2.6$
 $K(.5\text{HgCl}_2 + .5\text{Hg}(\text{OH})_2) = -0.5, \text{DH} = 0$
 $K(\text{HgCl} + \text{HgOH} = \text{HgClOH} + \text{Hg}) = 0.6, \text{DH} = -0.8 \text{ kJ mol}^{-1}. \text{DH}(\text{HgClOH} + \text{H}) = -20,$
 $K(\text{HgSCN} + \text{HgOH} = \text{Hg}(\text{SCN})(\text{OH}) + \text{Hg}) = 0.3, \text{DH} = 1. \text{DH}(\text{HgSCNOH} + \text{H}) = -22, \text{DH}(\text{Hg}(\text{OH})_2 + \text{H}) = -20.7$

Hg++ cal NaClO4 25°C 1.0M C 1975CGe (11568) 322
 $*K_1 = -3.65$
 $*K_2 = -2.64$
 $\text{DH}(*K_1) = -20.7 \text{ kJ mol}^{-1}, \text{DS} = 0 \text{ J K}^{-1} \text{ mol}^{-1}; \text{DH}(*K_2) = -20.7, \text{DS} = 20$

Hg++ EMF NaClO4 25°C 1.00M U 1970CGc (11569) 323
 $*K_1 = -3.65$
 $*B_2 = -6.29$

Hg++ EMF NaClO4 25°C 1.00M U 1968CGa (11570) 324
 $*K_1 = -3.84$
 $*B_2 = -6.38$

Hg++ cal NaClO4 25°C 3.00M U H 1967AKb (11571) 325
 $\text{DH}(*K_1) = 30.2 \text{ kJ mol}^{-1}, \text{DS} = 33 \text{ J K}^{-1} \text{ mol}^{-1}; \text{DH}(*K_2) = 10.9, \text{DS} = -14;$
 $\text{DH}(*B(2,1)) = 12.8, \text{DS} = -9; \text{DH}(*B(2,2)) = 53.0, \text{DS} = 79$

Hg++ gl NaClO4 25°C 0.50M U 1963KOb (11572) 326
 $*K_1 = -3.68$
 $*K_2 = -2.57$

Hg++ gl NaClO4 25°C 3.0M U 1962AHa (11573) 327
 $*K_1 = -3.55$
 $*B_2 = -6.21$
 $*B(2,1) = -2.67$
 $*B(2,2) = -5.16$
 $*B(4,3) = -6.40. *B(m,n): m\text{Hg} + n\text{H}_2\text{O} = \text{Hg}_m(\text{OH})_n + n\text{H}$
 In 3M Ca(ClO4)2 $*K_1 = -3.49, *B_2 = -5.96, *B(2,1) = -2.67, *B(2,2) = -4.95$

Hg++ cal NaClO4 25°C 0.88M U H 1962LGa (11574) 328
 Medium: HClO4. $\text{DH}(*K_{\text{so}}) = -23.6 \text{ kJ mol}^{-1}. \text{Data also for } I = 8.76: \text{DH}(*K_{\text{so}}) = -28.1$
 Also DH for mixtures HClO4-HCl, HBr, HI; HgO(s)

Hg++ sol none 25°C 0.0 U T 1961AHb (11575) 329
 $K_{\text{s}}(\text{HgO}(\text{s}) + \text{H}_2\text{O} = \text{Hg}(\text{OH})_2) = -3.65$
 $\text{HgO}(\text{s}) = \text{rh, red}; K_{\text{s}} = -3.50(35 \text{ C}), -3.29(50 \text{ C}), -3.13(60 \text{ C})$
 $\text{HgO}(\text{s}) \text{rh, yellow}; K_{\text{s}} = -3.63(35 \text{ C}), -3.12(60 \text{ C}) \text{ plus others}$

Hg++ sol NaClO4 25°C 3.0M U 1961DTa (11576) 330
 $*K_{\text{so}} = 2.41$
 $K_{\text{s}}(\text{HgO}(\text{s}) + \text{H}_2\text{O} = \text{Hg}(\text{OH})_2) = -3.75$
 $*K_1 = -3.23$
 $*B_2 = -6.16$
 $K_{\text{so}} = -26.0$

Hg++	sol	NaNO3	23°C	2.0M	U		1959NHa (11577)	331
						$K_s(\text{HgO}(s)+\text{H}_2\text{O}=\text{Hg}(\text{OH})_2)=-3.41$		
Hg++	vlt	NaNO3	30°C	2.0M	U	B2=21.40	1959NHa (11578)	332
Hg++	sol	oth/un	35°C	var	U		1959SAa (11579)	333
						$K_s(\text{HgO}(s)+\text{H}_2\text{O}=\text{Hg}(\text{OH})_2)=-3.5$		
Hg++	cal	NaClO4	25°C	5.50M	U	H	1959SLd (11580)	334
						Medium:HClO4. $\text{DH}(*K_{so}=-18.7 \text{ kJ mol}^{-1})$. $\text{DH}(*K_{so})=-19.2(I=4.2)$, $-20.1(I=3.3)$, $-21.4(I=2.21)$, $-22.80(I=1.12)$		
Hg++	gl	NaNO3	25°C	0.10M	U	TIH	1958ASa (11581)	335
						$*B2=-6.52$		
						$\text{DH}(*B2)=45.2 \text{ kJ mol}^{-1}$. $*B2=-6.72(13 \text{ C})$, $-6.26(30 \text{ C})$, $-6.00(40 \text{ C})$		
Hg++	vlt	KNO3	25°C	1.0M	U		1954GOa (11582)	336
						$*K1=-3.5$		
						$*K2=-4.05$		
Hg++	gl	NaClO4	25°C	0.50M	U		1952HSa (11583)	337
						$*K1=-3.70$		
						$*K2=-2.60$		
Hg++	oth	none	25°C	0.0	U		1952LAb (11584)	338
						$K_{so}=-25.52$		
Hg++	con	oth/un	25°C	var	U		1941BJa (11585)	339
						$*K1=-2.4$		
Hg++	gl	oth/un	21°C	dil	U		1941BJa (11586)	340
						$*K1=-2.8$		
						$*K2=-3.5$		
Hg++	sol	none	25°C	0.0	U		1939GHa (11587)	341
						$K1=11.51$ $B2=22.66$		
						$*K1=-2.49$		
						$*K2=-2.85$		
						$*K_{so}=1.72$		
						$*K_s(\text{HgO}+\text{H}=\text{HgOH})=-0.77$		
Hg++	sol	none	25°C	0.0	U		1938GHa (11588)	342
						$*K3=-14.85$		
						$K_s(\text{HgO}+\text{H}_2\text{O}=\text{Hg}(\text{OH})_2)=-3.65$		
						$K_s(\text{HgO}+\text{H}_2\text{O}+\text{OH}=\text{Hg}(\text{OH})_3)=-4.50$		
Hg++	sol	none	25°C	0.0	U		1920FUa (11589)	343
						$*K3=-14.77$		
						$K_s(\text{HgO}+\text{H}_2\text{O}=\text{Hg}(\text{OH})_3)=-3.61$		
						$K_s(\text{Hg}+\text{H}_2\text{O}+\text{OH}=\text{Hg}(\text{OH})_3)=-4.30$		

Hg++ sol oth/un 18°C var U K1=11.86 B2=22.13 1917K0a (11590) 344
 Kso=-25.86
 Ks(HgO+H₂O=Hg(OH)₂)=-3.73

Hg++	oth oth/un	?	?	U		1904FAa (11592)	346
					K _{so} =-25.82		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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$$*K_s((\text{HgOH})_3\text{P}_4\text{O}_{10}(\text{s}) + 4\text{H} = 3\text{Hg} + \text{H}_3\text{P}_4\text{O}_{10} + 3\text{H}_2\text{O}) = -9.4, \quad K_s(\text{HgH}_2\text{P}_4\text{O}_{10}(\text{s}) = \text{Hg} + \text{H}_3\text{P}_4\text{O}_{10}) = -13.1$$

Hg++ sol oth/un 20°C var U 1972G0d (13210) 351
K(Hg3L2(s)+4H=3Hg+2H2L)=-6.0

P207---- H4L Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from (HO)2PO.O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF NaNO3 27°C 0.75M U 1960YDa (13595) 355
B(Hg(OH)L)=17.45

P3010----- H5L CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE oth/un 25°C 1.00M U I K1=7.16 B2=8.90 1962SIb (13864) 356
B(Hg2(OH)L)=13.65
B(Hg2(OH)2L)=20.05
K(Hg2+HL)=4.34

Medium: Na+. In K+: K1=7.84, B2=9.47, B(Hg2(OH)L)=14.22, B(Hg2(OH)2L)=20.35

S-- H2L Sulfide CAS 7783-06-4 (705)
Sulfide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sol oth/un 25°C 0.3M C 2000JMa (14377) 357
Ks(HgS+(x-1)S=HgSxOH+H)=-15.7

Dissolution of cinnabar in H2S solutions containing S(orth), pH 6-10, to form polysulfides. Medium not stated. Ks(HgS+HS+2(x-1)S=Hg(Sx)2+H)=-11.7.

Hg++ sol KCl 25°C 0.70M M 1997PHa (14378) 358
K(HgS(s)+H2S=Hg(HS)2)=-5.36
K(HgS(s)+HS=Hg(HS)S)=-5.34
K(HgS(s)+2HS=HgS2+H2S)=-7.14
K(HgS+HS+nS(0)=HgHS(Sn))=-3.97

For K(HgS(s)+HS+nS(rhom)): n=4-6.

Hg++ sol KCl 25°C 0.7M C 1995PHa (14379) 359
Ks(HgS+H2S(aq)=Hg(HS)2)=-5.76
Ks(HgS(s)+HS=HgS2H)=-4.82
Ks(HgS(s)+HS=HgS2+H)=-13.41
Ks(2HgS+H+H2O=Hg2SOH+H2S)=-8.4

Method: dissolution of cinnabar, red HgS, in 0.7 m KCl, 0.5-30 mM NaHS, buffered to pH 1-13. Ks(Hg3S2Cl2+H2S=3HgS(s)+2H+2Cl)=-1.26.

Hg++ sol KCl 25°C 0.7M C 1995PHa (14380) 360
Ks(2Hg(s)+2HS+nS=Hg2S4SnH2)=1.

Method: dissolution of cinnabar, red HgS, in 0.7 m KCl, 0.5-30 mM NaHS, buffered to pH 1-13, in presence of S(s).The product is Hg2(S(II)4(S(0))nH2

Hg++ vlt NaCl 25°C ? U 1994ZMa (14381) 361
 $K_{1eff}=7.8$
 $K_{2eff}=5.0$
 Medium: sea water, pH=8. Method: cathodic stripping square wave voltammetry

Hg++ oth none ? 0 U 1990DKa (14382) 362
 $*K_s(\text{HgS}(\text{black})+\text{H}=\text{Hg}+\text{HS})=-38.9$
 $*K_s(\text{HgS}(\text{red})+\text{H}=\text{Hg}+\text{HS})=-39.5$
 From recalculation of literature data.

Hg++ oth none 25°C 0.0 C 1989DYa (14383) 363
 $K_{\text{Hg}+\text{HS}=\text{HgS}+\text{H}}=28.5$
 $*K_{\text{so}}(\text{HgS})=-38.5$
 $K_{\text{so}}(\text{ZnS})=-10$
 Calculated from literature data, based on $K(\text{H}+\text{S})=17.0$. HgS is metacinnabar
 For cinnabar, $*K_{\text{so}}(\text{HgS})=-39.8$, $K_{\text{so}}(\text{HgS})=-11.3$.

Hg++ oth none 25°C 0 U 1988LIa (14384) 364
 $K_{\text{so}}(\text{HgS}, \text{red})=-56.4$
 $*K_{\text{so}}(\text{HgS}, \text{red})=-39.1$
 $K_{\text{so}}(\text{HgS}, \text{black})=-56.1$
 $*K_{\text{so}}(\text{HgS}, \text{black})=-38.7$
 Derived from thermodynamic data and $K(\text{H}+\text{S}=\text{HS})=17.3$.

Hg++ oth none 25°C 0 U 1988SBc (14385) 365
 $K_{\text{so}}(\text{HgS}, \text{cinnabar})=-58.36$
 $K_{\text{so}}(\text{HgS}, \text{metacinnabar})=-57.07$
 Method: recal. from literature data using $K(\text{H}+\text{S}=\text{HS})=18.57$ and $K(\text{H}+\text{HS})=6.99$

Hg++ dis oth/un 25°C 0.69M U 1985DYa (14386) 366
 $K(\text{Hg}+2\text{H}_2\text{S}=\text{HgHS}_2+3\text{H})=17.77$
 $K(\text{Hg}+2\text{H}_2\text{S}=\text{Hg}(\text{HS})_2+2\text{H})=23.96$

Hg++ sol oth/un 150°C 0.00 U 1981SZb (14387) 367
 $K(\text{HgS}(\text{s})+\text{H}_2\text{S}=\text{Hg}(\text{HS})_2)=-4.63$

Hg++ oth none 50°C 0.0 M T 1969HEa (14388) 368
 Estimated from literature data. Cinnabar: $K_{\text{so}}=-50.02(50\text{ C})$; $-44.25(100\text{ C})$;
 $-39.91(150\text{ C})$; $-36.51(200\text{ C})$; $-33.83(250\text{ C})$; $-31.72(300\text{ C})$

Hg++ sol none 25°C 0.0 U B2=50.23 1964PCa (14389) 369
 $K(\text{HgL}(\text{s})+\text{L}=\text{HgL}_2)=-1.5$
 $K(\text{HgL}(\text{s})+2\text{H}=\text{HgL}+\text{H}_2\text{S}(\text{g}))=-30.76$
 $K(\text{HgL}(\text{s})+\text{H}_2\text{L}(\text{g})=\text{Hg}(\text{HL})_2)=-6.2$
 Also by Hg electrode. I=0 corr.

Hg++ ISE KCl 20°C 1.0M U 1963SWa (14390) 370
 $K_{\text{so}}(\text{black HgS})=-50.96$
 $K(\text{HgS}(\text{s})+\text{H}_2\text{S}=\text{Hg}(\text{SH})_2)=-5.97$
 $K(\text{HgHS}_2+\text{H})=6.19$

$$K(\text{HgS}_2 + \text{H}) = -8.30$$

Hg++ oth none 25°C 0.0 U T 1959CZa (14391) 371

$$K_{\text{so}}(\text{HgL}) = -52.73$$

From thermodynamic data. $K_{\text{so}} = -43.40(100\text{ C}), -35.57(200\text{ C}), -26.89(400\text{ C}), -22.16(600\text{ C})$

Hg++ sol oth/un rt var U 1959DGc (14392) 372

$$K(\text{HgL}(\text{s}) + \text{L} = \text{HgL}_2) = 1.08$$

$$K(\text{HgL}_2 + 2\text{H}) = 20.88?$$

Hg++ ISE none 25°C 0.0 U 1952GGc (14393) 373

$$K_{\text{so}}(\text{HgL}) = -51.52$$

Hg++ oth none 25°C 0.0 U 1952LAb (14394) 374

$$K_{\text{so}}(\text{HgL}) = -53.8$$

$$K(\text{HgL}(\text{s}) + \text{L} = \text{HgL}_2) = 0.58$$

From thermodynamic data. $\text{HgL}(\text{s}) = \text{metacinnabarite}$

Hg++ sol oth/un 25°C var U 1950MAa (14395) 375

$$K(\text{HgL}(\text{s}) + \text{L} = \text{HgL}_2) = \text{ca. } 0$$

Hg++ oth none 25°C 0.0 U I 1946TSa (14396) 376

$$K(\text{HgL}(\text{s}) + 2\text{H} = \text{Hg} + \text{H}_2\text{S}(\text{g})) = -31.1$$

$$K(\text{HgS}_2 + 2\text{H} = \text{Hg}(\text{SH})_2) = 7.0$$

From thermodynamic data. $\text{HgL}(\text{s}) = \text{metacinnabarite}$

Hg++ sol oth/un 25°C var U 1931K0a (14397) 377

$$K_{\text{so}}(\text{HgL}) = -53.5$$

$$K(\text{HgS}(\text{s}) + 2\text{H} = \text{Hg} + \text{H}_2\text{S}(\text{g})) = -30.6$$

Also by Hg electrode

Hg++ ISE oth/un 18°C var U 1909BZa (14398) 378

$$K_{\text{so}}(\text{HgL}) = -47.17$$

Hg++ sol oth/un 25°C var U B2=54.7 1908KNa (14399) 379

$$K_{\text{so}}(\text{HgL}) = -53.5$$

Also by Hg electrode

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 25°C 0.10M U B2=16.53 1986ECa (15023) 380

Hg++ ISE non-aq 25°C 100% C H T K1=6.77 B2=11.8 1985AIa (15024) 381

$$K_3 = 3.1$$

Medium: pyridine, 0.1 M Et4NClO4. $\text{DH}(\text{K}_1) = -2.5\text{ kJ mol}^{-1}$, $\text{DH}(\text{K}_2) = 3.1$, $\text{DH}(\text{K}_3) = 0.1$; $\text{DS}(\text{K}_1) = 121\text{ J K}^{-1}\text{ mol}^{-1}$, $\text{DS}(\text{K}_2) = 102$, $\text{DS}(\text{K}_3) = 60$

 Hg++ ISE non-aq 25°C 100% C H T K1=7.49 B2=13.08 1985AIa (15025) 382
 K3=2.95
 K4=2.24

Medium: DMSO, 1 M NH₄ClO₄. DH(K1)=-29.7 kJ mol⁻¹, DH(K2)=-31.4, DH(K3)=-9.8,
 DH(K4)=-17.5; DS(K1)=50 J K⁻¹ mol⁻¹, DS(K2)=2, DS(K3)=24, DS(K4)=-16

Hg++ cal NaClO₄ 25°C 0.50M U TI 1984VKa (15026) 383
 DH(K1)=-50.08 kJ mol⁻¹; DS(K1)=5.9 J K⁻¹ mol⁻¹; DH(B2)=-102.4; DS(B2)=-22.8

Hg++ ISE non-aq 25°C 100% C H K1=9.33 B2=14.99 1981APb (15027) 384
 B3=17.95
 B4=20.26

Medium: DMSO, 0.1 M NH₄ClO₄. DH(K1)=-28.1; DH(B2)=-59.1; DH(B3)=-68.9;
 DH(B4)=-85.6 kJ mol⁻¹

Hg++ cal NaClO₄ 25°C 1.0M C K1=9.08 B2=16.86 1975CGe (15028) 385
 K3=2.80
 K4=0

DH(K1)=51.2 kJ mol⁻¹, DS=2 J K⁻¹ mol⁻¹; DH(K2)=51.2, DS=23;
 DH(K3)=18, DS=-7; DH(K4)=18, DS=-22

Hg++ sol NaClO₄ 20°C 1.00M U M 1975ZCa (15029) 386
 B(Hg(Oxalate)(SCN))=8.37

Hg++ EMF oth/un ? var U 1974KIb (15030) 387
 B4=20.3

Hg++ cal NaClO₄ 25°C 1.0M U H 1974KUe (15031) 388
 DH(B4)=-145.31 kJ mol⁻¹; DS=-72.0 J K⁻¹ mol⁻¹

Hg++ EMF non-aq 25°C 100% U B2=16.08 1974SLa (15032) 389
 B3=19.00
 B4=20.41

Medium: DMSO, 1 M (Li,Na)ClO₄

Hg++ EMF non-aq ? 100% U B2=16.85 1972BGa (15033) 390
 Medium: ethylene glycol, 0.1 M LiClO₄

Hg++ EMF non-aq 25°C 100% U B2=16.1 1972FDa (15034) 391
 B3=19.1
 B4=21.2

Medium: DMSO, 0.1 M LiClO₄

Hg++ sp NaClO₄ 25°C 1.0M U T H 1971AKb (15035) 392
 K(Hg+CrL)=4.22

Medium: HClO₄; B=4.42(15 C)

Hg++ cal NaClO₄ 25°C 1.0M U H 1971AKb (15036) 393
 DH(K1)=-49.8 kJ mol⁻¹, DS=7.1 J K⁻¹ mol⁻¹; DH(K2)=-50.6, DS=-20.1;

DH(K3)=-20.5, DS=-14.2; DH(K4)=-20.9, DS=-33.1

Hg++ EMF non-aq ? 100% U 1971BGa (15037) 394

B4=26.8

Medium: N,N-dimethylacetamide, 0.1 M LiClO4

Hg++ sp none 25°C 0.0 U M 1971GMa (15038) 395

K(HgL2+HgCl2=2HgClL)=0.91

K(HgL2+HgBr2=2HgBrL)=0.10

K(HgL2+HgI2=2HgIL)=-0.96

Hg++ sp oth/un 25°C dil U 1971GMa (15039) 396

B(HgLC1)=0.91

B(HgLBr)=0.10

B(HgLI)=-0.96

Hg++ EMF NaClO4 25°C 1.0M U T K1=9.08 B2=16.86 1970CGb (15040) 397

B3=19.70

B4=21.67

By solubility: Kso=-19.00, Ks(HgL(s)+L)=-2.70, K3=2.80, K3.K4=4.8

Hg++ gl NaClO4 25°C 1.0M U 1970CGc (15041) 398

K(Hg2+L+H2O=Hg(OH)L+H)=5.65

Hg++ dis NaClO4 25°C 0.50M U 1970SIa (15042) 399

K3=2.73

K4=2.02

Hg++ sp oth/un ? 0.10M U M 1969NOc (15043) 400

K(HgA+L)=5.7

Medium: KSCN. A=methyl thymol blue

Hg++ vlt non-aq 40°C 100% U I T K1=16.2 1969PVb (15044) 401

B4=22.7

Medium: 20% w/w N-methylacetamide-DMF. K1=17.7, B4=25.2(0%); K1=15.0, B4=21.1 (40%); K1=14.4, B3=17.3, B4=19.52(w=80); K1=14.6, B4=20.5(100%)

Hg++ sol oth/un 20°C 4.60M U 1968GYa (15045) 402

Ks(HgL2(s)+Hg=Hg2L2)=-0.81

Medium: Ca(NO3)2

Hg++ sp NaClO4 ? 0.10M U 1968NOa (15046) 403

K(Hg(EDTA)+L)=5.9

Hg++ ISE mixed 25?°C 20% U I B2=7.6 1968PMe (15047) 404

B3=9.2

B4=10

Also polarography. Medium: 20% C5H5N. In 80%: B3=9.7. Also 10%,40%,60%

Hg++ sol NaNO3 20°C 2.0M U 1968YGa (15048) 405

						$K_s(\text{HgL2(s)} + \text{Br} = \text{HgL2Br}) = 0.56$ $K_s(\text{HgL2(s)} + 2\text{NO2}) = -1.03$	
Hg++	sol	KN03	20°C	4.60M	U		1966GBb (15049) 406 $K_s(\text{HgL2(s)} = \text{HgL2}) = -1.83$ $K_s(\text{HgL2(s)} + \text{Hg} = \text{Hg2L2}) = -0.70$
Hg++	sp	NaCl04	25°C	1.0M	U		1964CZb (15050) 407 $K(\text{HgI2(s)} + 2\text{L} = \text{HgI2L2}) = -0.27$ $B(\text{HgI2L2}) = 27.92$ $K_s(\text{HgI2}) = -28.19$
Hg++	vlt	oth/un	30°C	var	U	B2=17.60 B3=20.40 B4=21.23	1964SKa (15051) 408
Hg++	sol	NaNO3	20°C	0.50M	U		1962CSa (15052) 409 $B(\text{HgL2(NO2)}) = 5.98$ $B(\text{HgL2Cl}) = 6.82$ $B(\text{HgL2Br}) = 7.57$ $B3 = 7.24$ $K(\text{HgL2(s)} = \text{Hg} + 2\text{L}) = -7.15$ assumed
Hg++	vlt	KN03	25°C	0.20M	U TI	T B2=16.43 B3=19.14 B4=21.12 B2=17.26, B3=19.85, B4=22.05(15 C); B2=15.74, B3=18.39, B4=20.23(35 C); At I=0 corr B2=18.08, B3=20.66, B4=22.61(15 C)); 16.58, 19.23, 20.83(35 C)	1962TEa (15053) 410
Hg++	sol	NaNO3	20°C	0.50M	U	T $K(\text{HgL2(s)} + \text{Br} = \text{HgL2Br}) = 0.43$ $B(\text{HgL2Br}) = 7.59(?)$	1960CZa (15054) 411
Hg++	ISE	KN03	20°C	3.5?M	U	B5=22.05 B6=21.67	1960GRd (15055) 412
Hg++	ISE	alc/w	20°C	100%	U I	B4=24.36 Medium: MeOH, 1.5?M KN03. B5=21.41(2.5 M MeOH), 22.29(10 M MeOH). Data also in acetone/H2O mixtures	1960GRd (15056) 413
Hg++	vlt	NaCl04	25°C	1.0M	U	T B2=16.07 K3=2.88 K4=1.99 B4=20.94	1960NAb (15057) 414
Hg++	sol	NaNO3	20°C	0.50M	U		1959CFd (15058) 415 $K(\text{HgL2(s)} + \text{Cl} = \text{HgL2Cl}) = -0.39$ $B(\text{HgL2Cl}) = 7.00(?)$

Hg++ ISE oth/un 25°C var U 1905SSa (15070) 427
B4=21.99

S03-- H2L Sulfite CAS 7782-99-2 (801)
Sulfite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF NaCl04 25°C 0.50M C H B2=23.33 1976MHc (15457) 428
B3=24.1

Method: Hg electrode. By calorimetry: DH(B2)=91 kJ mol⁻¹.

Hg++ sol NaNO3 var U 1972PBg (15458) 429
K(HgO(s)+2L+H2O=HgL2+2OH)=0.43

Hg++ sol oth/un 25°C var U I M 1966CZb (15459) 430
Ks(HgI2(s)+L)=0.12
B(HgI2L)=28.31

In 3 M NaCl04: B(HgI3L)=29.77, K(HgI4+L=HgI3L+I)=0.47

Hg++ sol NaCl04 21°C 0.50M U M 1964CZa (15460) 431
K(Hg(SCN)2(s)+L)=3.95

Hg++ ISE oth/un 18°C 3.0M U I B2=22.85 1955TBa (15461) 432
Method: Hg electrode. Medium: Na2S04. At I=0 corr.: B2=24.07, K3=0.89

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaCl04 25°C 0.5M C P K1=1.4 B2= 2.4 2005PBa (16235) 433
IUPAC evaluated(P). B2 uncertain

Hg++ con oth/un 25°C 0.0 U M 1968YMa (16236) 434
K(Hg(en)+L)=2.47

Hg++ EMF oth/un 25°C var U B2=1.5 1957KSb (16237) 435

Hg++ sp NaCl04 25°C 0.33M U K1=1.42 1957PTa (16238) 436

Hg++ EMF NaCl04 25°C 0.50M U K1=1.34 B2=2.44 1946ISa (16239) 437

S203-- H2L Thiosulfate CAS 73686-28-7 (177)
Thiosulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal NaCl04 25°C 1.0M U H 1974KUe (16852) 438

DH(B3)=-160.9 kJ mol⁻¹, DS=97.9 J K⁻¹ mol⁻¹

Hg++ sol NaNO3 var U 1972PBg (16853) 439
K(HgO(s)+2L+H2O=HgL2+2OH)=0.48

Hg++ vlt NaNO3 25°C 0.20M U 1970MSc (16854) 440
K3=2.48

Hg++ sp NaClO4 25°C 3.0M U M 1966CZa (16855) 441
K(HgI4+L=HgI3L+I)=0.36
B(HgI3L)=29.66

Hg++ sol NaClO4 25°C 1.0M U 1964CZc (16856) 442
K(HgI2(s)+L)=0.66
B(HgI2L)=28.86
Kso(HgL2)=-28.19

Hg++ ISE none 25°C 0.0 U B2=29.18 1961NSa (16857) 443
B3=30.3

Method: Hg electrode. By polarography B2=29.27, B3=30.8

Hg++ sol oth/un 25°C var U B2=29.4 1957T0a (16858) 444

Hg++ ISE none 25°C 0.0 U B2=29.86 1954T0b (16859) 445
K3=2.40
K4=1.35

Se-- H2L Selenide (6335)
Selenide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF oth/un 25°C 1.0M U 1972GKa (16941) 446
Kso=-58

Medium: Na2SO4

Hg++ sol oth/un 25°C dil C 1971MGd (16942) 447
Kso(HgSe)=-51.9
K(Hg+HSe+OH)=51.2
K(Hg+2HSe+2OH)=61.0
K(Hg+2HSe+OH)=52.8

Method: 203Hg radiometric measurement of HgSe dissolution in selenide media, 0.015-0.081 M.

Hg++ oth none 25°C 0.0 U 1964BUE (16943) 448
Kso=-64.5

Hg++ vlt oth/un 25°C var U 1948LNa (16944) 449
Kso(HgL)=-59

SeCN- HL Selenocyanate CAS 73102-11-2 (440)
Selenocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	NaClO4	25°C	1.0M	U	H			1974KUE (16984)	450
DH(B4)=-192.5 kJ mol-1, DS=-96.2 J K-1 mol-1										
Hg++	vlt	KNO3	25°C	0.30M	U	T		B2=21.4 B3=25.7	1972MTa (16985)	451
35 C: B2=20.4, B3=24.5										
Hg++	sol	NaClO4	25°C	1.0M	U	M		B(HgI2L)=28.0 B(HgI2L2)=29.42 Ks(HgI2)=-27.19 B(HgI3L)=29.74	1965CZa (16986)	452
K(HgI4+L=HgI3L+I)=0.44										
Hg++	sol	oth/un	18°C	dil	U				1960LCa (16987)	453
K(CuHgL4(s)=Cu+HgL4)=-9.80										
Hg++	ISE	NaNO3	25°C	0.30M	U	TIH		B3=26.40 K4=2.47 B4=28.87	1956TOa (16988)	454
B4=30.07(15 C), 29.46(20 C), 28.33(30 C). In 0.8 M NaNO3 B4=28.73 At I=0 corr: B4=29.95, DH(B4)=-194.6 kJ mol-1, DS=-87.0 J K-1 mol-1										

SeO3-- H2L Selenite CAS 7783-00-8 (2391)
Selenite;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sol	NaNO3	25°C	1.0M	U			B2=12.48 Kso(HgL)=-13.82	1957TOa (17061)	455
Hg++	sol	oth/un	25°C	var	U			K(HgL(s)+L=HgL2)=-1.38	1909RPa (17062)	456

SeO4-- H2L Selenate CAS 7783-08-6 (459)
Selenate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	NaNO3	25°C	1.0M	U			B2=36.8	1957TOa (17102)	457
Ligand: selenosulfate, SeSO3--										

CH2O2 HL Formic acid CAS 64-18-6 (37)
Methanoic acid; H.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	NaClO4	25°C	3.00M	U	I		K1=3.62 B3=6.72	1982KCd	(17611) 458
Hg++	gl	NaClO4	25°C	3.0M	C			K1=3.61 B2=7.10	1977RWa	(17612) 459
Hg++	EMF	oth/un	25°C	0.10M	U			K1=5.43	1973PZa	(17613) 460

CH3NO			L	Formamide				CAS 75-12-7	(3536)	
Methanoic acid amide; HCO.NH2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	1.0M	U			K1=2.06	1976BMg	(17677) 461

CH4N2O			L	Urea				CAS 57-13-6	(2018)	
Carbamide, Urea; (H2N)2CO										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	1.0M	U			K1=2.06	1976BMg	(17719) 462
Hg++	ISE	oth/un	30°C	0.10M	U			B2=2.1	1969GLa	(17720) 463

CH4N2S			L	Thiourea				CAS 62-56-6	(51)	
Thiocarbamide, Thiourea; (H2N)2CS										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	KNO3	25°C	0.09M	U	T H		K1=12.07 B3=20.70	1988MMe	(17827) 464
DH(K1)=-218 kJ mol-1; DH(B2)=-170.9; DH(B3)=-198										
Hg++	vlt	KNO3	25°C	0.10M	U			B2=22.43	1986ECa	(17828) 465
Hg++	ISE	mixed	25°C	82%	U			B4=28.42	1979MTd	(17829) 466
Medium: 82% DMF/H2O										
Hg++	ISE	KNO3	25°C	0.10M	U			K1=18.60 K(Hg+HL)=13.27 B(Hg(OH)L)=25.37	1978SGd	(17830) 467
Hg++	EMF	NaClO4	25°C	0.50M	C	H		K1=11.4 B3=25.15 B4=27.10 B(Hg2L3)=36.0	1976MHc	(17831) 468
Method: Hg electrode. By calorimetry: DH(K1)=-74 kJ mol-1; DH(B2)=-144,										

DS(B2)=-60 J K-1 mol-1; DH(B3)=-188, DS(B3)=-148, DH(B4)=-203, DS(B4)=-161

Hg++ sp oth/un 20°C 4.10M U 1973SUa (17832) 469
B(HgBr3L)=22.87

Hg++ cal none 25°C 0.0 C H 1972EIa (17833) 470
K(Hg(CN)2+L)=2.074
K(Hg(CN)2+2L)=2.644
DH(Hg(CN)2+L)=-5.35 kJ mol-1, DH(Hg(CN)2+2L)=-41.7.

Hg++ ISE oth/un 30°C 0.10M U B2=21.3 1969GLa (17834) 471
B3=24.2
B4=25.8

Hg++ cal alc/w 25°C 20% U IH 1968IEb (17835) 472
K'(Hg(CN)2+L)=2.03
K''(Hg(CN)2L+L)=0.77

Medium: 20% EtOH. DH'=-10.0 kJ mol-1, DS=5.4 J K-1 mol-1; DH''=-35.9, DS=-107
40%: K'=1.94, DH'=-18.9, DS'=-26.3; K''=1.04, DH''=-34.3, DS''=-95. Data to 92%

CH4N2Se L Selenourea CAS 630-10-4 (4207)
Selenocarbamide; (H2N)2CSe

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE oth/un 30°C 0.10M U B2=24.0 1969GLa (17865) 473
B3=30.2
B4=32.9

CH5N L Methylamine CAS 74-89-5 (155)
Methylamine; CH3.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaCl04 25°C 0.15M C H K1=7.78 2003AGb (18016) 474
Metal is CH3Hg+. By calorimetry, DH(K1)=-39.6 kJ mol-1, DS(K1)=
16 J K-1 mol-1.

Hg++ vlt non-aq 25°C 100% U B2=17.0 1973MTa (18017) 475
Medium: 0.5 Et4N.Cl04 in HCON(CH3)2

Hg++ gl KNO3 25°C 0.50M U K1=8.66 B2=17.86 1972BJa (18018) 476

Hg++ gl oth/un 25°C 0.0 U H 1966PCa (18019) 477
K(HgCl2+L=HgClL+Cl)=2.40
K(HgClL+L=HgL2+Cl)=2.21

Medium: 0 corr, By calorimetry: DH(HgCl2+L)=-28.4 kJ mol-1, DS=-50.2;
DH(HgClL)=-4.2, DS=29.3

Hg++ gl oth/un 25°C 0.50M U K1=8.6 B2=17.9 1950BJa (18020) 478

$K3=0.4$

CH5N3 L Guanidine (3535)
Diaminomethylimine; H2N.C(:NH)NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl NaCl04 27°C 1.0M U B2=24.96 1964WDa (18045) 479

CH5N3O	L	Semicarbazide	CAS 563-41-7	(373)
Semicarbazide, N-Aminourea; H2N.CO.NH.NH2				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ ISE oth/un 30°C 0.10M U B2=11.6 1969GLa (18053) 480
B3=15.2

CH5N3S	L	CAS 79-19-6 (372)
Thiosemicarbazide; H2N.CS.NH.NH2		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ ISE oth/un 30°C 0.10M U B2=22.4 1969GLa (18078) 481
B3=24.8

Hg++ ISE a1c/w 25°C 40% U B4=27.3 1961TKb (18079) 482

Medium: 40% EtOH, 0.05 NaC₂H₃O₂

Hg++ ISE NaNO3 20°C 0.80M U T 1960TKa (18080) 483
B4=26.77

B4=26.25(25 C), 25.75(30 C), 24.80(40 C), 23.93(50 C)

CH5N3Se L CAS 21198-79-8 (371)
Selenosemicarbazide; H2N.CSe.NH.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	ISE oth/un 30°C 0.10M U	B2=26.9 B3=30.4 B4=32.4	1969GLa (18088) 484
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C2H02Cl3 HL Trichloroacetic CAS 76-03-9 (1205)
Trichloroethanoic acid; Cl3C.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ ISE oth/un 25°C 0.10M U K1=3.08 1973LUa (18334) 485

C2H2	L	Acetylene	CAS 74-85-1 (703)
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Ethyne; HCCH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sol	oth/un	25°C	0.0	U			1965TUa (18356)	486
							Kso(HgL2(OH)2)=-37.10		

C2H2N2S3		H2L	Bismuthiol I	CAS	1072-71-5	(6261)
2,5-Dimercapto-1,3,4-thiadiazole;						

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U	M		1970GSb (18369)	487
							K(HgCl2+L=HgClL+Cl)=7.15		
							K(HgClL+L=HgL2+Cl)=6.54		
							K(HgClL+OH=HgLOH+Cl)=4.86		

C2H2O4		H2L	Oxalic acid	CAS	144-62-7	(24)
Ethanedioic acid; (COOH)2						

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sol	NaCl04	20°C	1.00M	U	I M	K1=7.25	1975ZCa (18918)	488
							B(HgLN02)=9.34		
							B(HgL(NO2)2)=9.32		
							B(HgLBr)=9.09		
							B(HgLBr2)=8.91		
							B(HgL(SCN))=8.37, B(HgL(acetate))=8.31, B(HgL(acetate)2)=8.57		
							B(HgL(Tartrate))=7.66, B(HgL(Tartrate)2)=9.16		

Hg++	ISE	oth/un	25°C	0.05M	U		K1=9.87	1973CSd (18919)	489
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Hg++	ISE	oth/un	25°C	0.10M	U		K1=9.66	1973LUa (18920)	490
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Hg++	dis	NaCl04	20°C	0.10M	U		K1=<4	1963STc (18921)	491
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C2H3O2Cl		HL	Chloroacetic	CAS	79-11-8	(34)
Chloroethanoic acid; ClCH2.COOH						

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaCl04	25°C	3.0M	C		K1=2.95 B2=5.61	1977RWa (19365)	492
Hg++	ISE	oth/un	25°C	0.10M	U		K1=4.64	1973LUa (19366)	493

C2H4		L	Ethylene	CAS	74-85-1	(478)
Ethene; H2C:CH2						

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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(Cyanomethyl)amine; CN.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	lg K values	Reference	ExptNo
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Medium: 0.5 M LHN03

C2H4N2S3 HL CAS 97049-30-4 (4220)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	lg K values	Reference	ExptNo
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Hg++	sp	oth/un	20°C	0.10M	U	K1=4.15	1970GKc	(19456)	496
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C₂H₄N₄HLCAS 61-82-5 (1265)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	lg K values	Reference	ExptNo
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Hg++ gl KN03 25°C 0.10M U I 1997DBa (19478) 497

Data also for $I=0.5$ and 1.0 M

C2H4OS2 HL CAS 2042-42-4 (592)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	lg K values	Reference	ExptNo
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Hg++ sp oth/un 25°C var U M 1970AFa (19513) 498

C2H4O2	HL	Acetic acid	CAS 64-19-7 (36)
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	lg K values	Reference	ExptNo
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Hg++ gl NaClO4 25°C 0.15M C H K1=3.36 2003AGb (19983) 499

Hg++ gl alc/w 25°C 100% M K1=8.5 B2=16.1 1994MPc (19984) 500

Hg++ ISE NaCl04 25°C 3.00M U I K1=4.00 B2=7.39 1982KCd (19985) 501

Hg++ gl NaNO3 30°C 1.00M U K1=3.32 B2=7.01 1978THa (19986) 502

Hg++ gl NaClO4 25°C 3.0M C K1=4.22 B2=8.45 1977RWa (19987) 503
Also measured by redox. K1=4.15, B2=8.44

Hg++ con NaNO3 25°C 0.10M U K1=6.1 B2=8.60 1975LBa (19988) 504

Hg++ sol NaClO4 20°C 2.00M U M 1975ZCa (19989) 505
B(Hg(Oxalate)L)=8.31
B(Hg(Oxalate)L2)=8.57

Hg++ gl NaNO3 34°C 1.00M U TI K1=3.48 B2=6.09 1974KBa (19990) 506
B3=9.24

In EtOH, K1=5.60, B2=10.62

Hg++ ISE oth/un 25°C 0.10M U K1=5.89 1973LUa (19991) 507

Hg++ EMF oth/un 20°C 0.02M U I K1=4.65 B2=5.65 1972KBa (19992) 508
B3=6.00

Medium: 0.02-0.5 acetate. In MeOH: K1=7.00, B2=9.60, B3=11.00

Hg++ gl NaClO4 30°C 1.0M U K1=5.55 B2=9.30 1964BSe (19993) 509
B3=13.28
B4=17.06

Hg++ sol oth/un ? ? U B2=8.43 1953MAa (19994) 510

C2H4O2S H2L Thioglycolic CAS 68-11-1 (596)
Mercaptoethanoic acid; HS.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.15M C H K1=16.72 2003AGb (20326) 511
K(CH3Hg+CH3HgL)=6.25
K'(CH3Hg+HL)=10.26
K''(CH3Hg+(CH3Hg)2L)=5.2

Metal is CH3Hg+. Calorimetry: DH(K1)=-71 kJ mol⁻¹, DS(K1)=82 J K⁻¹ mol⁻¹;
DH(K)=-38, DS(K)=-8; DH(K')=-43.1, DS(K')=52; DH(K'')=-49, DS(K'')=-64.

Hg++ EMF NaClO4 25°C 0.10M U K1=34.5 B2=40.50 1981BCc (20327) 512
K(Hg+HL)=36.1

Hg++ nmr oth/un 28°C ? U M 1975HMa (20328) 513
K(HgLC1+OH=HgLOH+Cl)=3.16
K(HgLC1+His=HgL(His)+Cl)=2.61

Hg++ EMF KNO3 12°C 1.0M U T B2=45.52 1954SKa (20329) 514
25 C: B2=43.82

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)

2-Hydroxyethanoic acid; HO.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	NaClO4	25°C	3.0M	C	T	K1=3.06 B2=7.05	1977RWa (20555)	515
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C2H5NO2		HL		Glycine			CAS 56-40-6	(85)	
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2-Aminoethanoic acid; H2N.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	NaClO4	25°C	0.15M	C	H	K1=7.85	2003AGb (21571)	516
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K(CH3Hg+CH3HgL)=2.33

K'(CH3Hg+HL)=2.00

Metal is CH3Hg+. Calorimetry: DH(K1)=-43.7 kJ mol⁻¹, DS(K1)=4 J K⁻¹ mol⁻¹

DH(K)=-6.2, DS(K)=24; DH(K')=-2.9, DS(K')=28.

Hg++	gl	KNO3	35°C	0.10M	C	M	K1=9.71 B2=18.25	1998ZWa (21572)	517
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Data for ternary complexes with 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime

Hg++	gl	NaCl	37°C	0.15M	U	TI	K1=3.962 B2=7.33	1992MTa (21573)	518
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at 0.5 M KNO3, 20 C: B1=10.3, B2=19.2; at 0.6 M NaNO3, 25 C: B2=18.36

Hg++	nmr	KNO3	25°C	0.10M	U	M	K1=10.5 B2=19.10	1988RPa (21574)	519
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Medium: D2O

Hg++	gl	NaNO3	25°C	0.10M	U		K1=12.2 B2=19.20	1974VBa (21575)	520
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Hg++	gl	oth/un	25°C	0.0	U	HM		1966PCa (21576)	521
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K'(HgCl2+L=HgClL+Cl)=3.42

K''(HgClL+L=HgL2+Cl)=2.61

Medium:0 corr. By calorimetry:DH(K')=-25.5 kJ mol⁻¹, DS=-20.9 J K⁻¹ mol⁻¹;

DH(K'')=-12.3, DS=8.4

Hg++	vlt	KNO3	25°C	0.60M	U		B2=18.36	1966TAb (21577)	522
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Hg++	gl	oth/un	22°C	0.01M	U		B2=18.2	1952PEa (21578)	523
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Medium: Hg(NO3)2

Hg++	gl	KNO3	20°C	0.50M	U		K1=10.3 B2=19.2	1945FLa (21579)	524
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C2H5N3O2		L		Biuret			CAS 108-19-0	(1126)	
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Carbomoylurea (Allophanic acid); H2N.CO.NH.CO.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	NaClO4	25°C	0.01M	U	T H	K1=10.90	1979SBa (21851)	525
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DH(K1)=-136 kJ mol⁻¹

Hg++ gl NaClO4 25°C 0.01M U K1=10.90 1975SSb (21852) 526

C2H6N2 L CAS 124-42-5 (4215)
Acetamidine; CH3.C(:NH)NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.10M U M 1970GSb (21903) 527
K(HgCl2+L=HgClL+Cl)=5.06
K(HgClL+L=HgL2+Cl)=4.49
K(HgClL+OH=HgLOH+Cl)=4.98

C2H6N2S L Methyl-Thiourea CAS 598-52-7 (1077)
N-Methylthiourea; CH3.NH.CS.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF KNO3 25°C 0.09M U T H K1=14.19 B2=18.64 1988MMe (22010) 528
B3=21.48
DH(K1)=-19.7 kJ mol-1; DH(B2)=-63; DH(B3)=-175

C2H6OS HL CAS 60-24-2 (841)
2-Mercaptoethanol; HS.CH2.CH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ nmr oth/un 28°C ? U M 1975HMa (22066) 529
K(HgLC1+OH=HgLOH+Cl)=3.69

C2H6O2S2 CAS 51554-68-8 (2123)
Ethylthiosulfonic acid; C2H5.S2O2H HL

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 20°C 1.00M U B2=17.98 1974GSb (22163) 530
B3=20.51
B4=22.89

C2H7N L Dimethylamine CAS 124-40-3 (802)
Dimethylamine; CH3.NH.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.15M C H K1=7.82 2003AGb (22225) 531
Metal is CH3Hg+. By calorimetry, DH(K1)=-31.8 kJ mol-1, DS(K1)=
43 J K-1 mol-1.

C2H7N L Ethylamine CAS 75-04-7 (156)
Ethylamine; CH3.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	oth/un	25°C	0.50M	U		K1=8.90 B2=18.44	1983HNa (22272)	532
Medium: 0.5 M LHN03									

Hg++	gl	oth/un	25°C	0.10M	U	M		1976PBa (22273)	533
							K(Hg(CN)2+L)=0.75		

Hg++	vlt	non-aq	25°C	100%	U		B2=16.5	1973MDb (22274)	534
Medium: HCON(CH3)2, 0.4 M Et4NC104									

C2H7NO L Ethanolamine CAS 141-43-5 (1057)

2-Aminoethanol; H2N.CH2.CH2.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	none	25°C	0.0	U		K1=8.86 B2=17.63	1970EAa (22403)	535

Hg++	ISE	mixed	25°C	20%	U	I	K1=16.32 B2=17.78	1969MPe (22404)	536
							B3=17.90		
							B4=18.43		

Medium:EtOH, 0.4 LiNO3. K1=16.29, B2=17.72, B3=18.16 (0% EtOH). K1=16.61, B2=17.81, B3=18.08, B4=18.93 (40%). K1=16.83, B2=18.17, B3=18.4, B4=19.09(60%)

Hg++	ISE	mixed	25°C	80%	U	I	K1=16.95 B2=18.25	1969MPe (22405)	537
							B3=18.8		
							B4=19.48		

Medium:EtOH, 0.4M LiNO3. K1=17.00, B2=18.33, B3=19.15, B4=19.94 (90%), B2=18.34, B3=19.65, B4=20.0 (95%). B3=19.71, B4=20.49 (100%)

Hg++	gl	NaCl04	30°C	1.0M	U	T	K1=8.92 B2=17.56	1964PCa (22406)	538
Medium: KCl04. 50 C: K1=8.64, K2=8.26									

Hg++	oth	KN03	25°C	0.40M	U		K1=8.51 B2=17.32	1956BJb (22407)	539
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C2H7NS HL CAS 60-23-1 (588)

2-Aminoethanethiol; H2N.CH2.CH2.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	sp	NaCl04	20°C	1.00M	U	M		1972GSh (22493)	540
							K(Hg+NiL2)=7.11		

C2H7O2PS2 HL CAS 5930-72-3 (4229)

O,O-Dimethyldithiophosphoric acid; (CH3O)2.PS.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	EMF	alc/w	25°C	40%	U		B2=25.5	1970TCa (22544)	541
							B3=28.0		

B4=30.5

Medium: 40% EtOH, 0.3 N KNO₃

C₂H₈N₂ L Ethylenediamine CAS 107-15-7 (23)

1,2-Diaminoethane; H₂N.CH₂.CH₂.NH₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO₄ 25°C 0.15M C H K₁=8.17 2003AGb (23164) 542
K(CH₃Hg+CH₃Hg_L)=4.85
K'(CH₃Hg+HL)=5.41

Metal is CH₃Hg+. By calorimetry, DH(K₁)=-48.6 kJ mol⁻¹, DS(K₁)=-7 J K⁻¹ mol⁻¹. DH(K)=-35.2, DS=-25; DH(K')=-42.5, DS=-39.

Hg++ gl KNO₃ 20°C 0.10M C M 2003LBa (23165) 543
B(HgAL)=18.43
K(HgA+L)=11.72
K(HgA+L=HgAL(OH)+H)=11.07

A is cytidine.

Hg++ gl KNO₃ 20°C 0.10M U K₁=13.85 B₂=22.22 1999LBa (23166) 544
B(HgHL)=16.95

Hg++ gl NaClO₄ 25°C 3.00M U K₁=16.749 B₂=25.970 1988WIa (23167) 545
B(HgHL)=8.82
B(HgHL₂)=20.75

Hg++ oth NaClO₄ 25°C 0.10M U K₁=14.179 B₂=23.34 1988WIa (23168) 546
K(Hg+OH+L)=23.865
K(Hg+HL+L)=18.459
K(Hg+2HL+L)=21.994
K(Hg+2HL)=13.1012

Recalculation of literature data.

Hg++ vlt KNO₃ 25°C 0.10M U B₂=22.92 1986ECa (23169) 547

Hg++ gl KNO₃ 25°C 0.10M C TIH R K₁=14.3 B₂=23.3 1984PAa (23170) 548
IUPAC evaluation

Hg++ vlt NaNO₃ 20°C 1.0M U K₁=10.79 1980BMc (23171) 549

Hg++ vlt non-aq 25°C 100% U B₂=14.00 1979SZa (23172) 550
Medium: DMSO, 0.1 M NaClO₄

Hg++ gl NaNO₃ 20°C 1.0M U K₁=12.13 1976BMg (23173) 551

Hg++ gl oth/un 25°C 0.10M U M 1976PBa (23174) 552
K(Hg(CN)₂+L)=2.12
K(Hg(CN)₂+HL)=0.60

Hg++ vlt non-aq 25°C 100% U B2=23.5 1973MKa (23175) 553
Medium: HCON(CH3)2, 0.5 M Et4NClO4

Hg++ gl oth/un 25°C 0.0 U HM 1966PCa (23176) 554
K'(HgCl2+L=HgClL+Cl)=5.54
K''(HgClL+L=HgL2+Cl)=4.19
Medium: 0 corr.By calorimetry: DH(K')=-36.4 kJ mol-1, DS=-16.7 J K-1 mol-1;
DH(K'')=-37.6, DS=-46.0

Hg++ vlt KNO3 40°C 0.10M U T H B2=21.94 1961RMa (23177) 555
B3=21.74
B2=24.36(10 C), 23.18(25 C); B3=24.1(10 C), 23.09(25 C).
DH(B2)=-137.5 kJ mol-1, DS=-20.9 J K-1 mol-1

Hg++ vlt KNO3 25°C 0.10M U K1=14.3 B2=23.3 1956WMc (23178) 556
B(HgL(OH))=23.8
K(Hg+L+HL)=18.6
K(Hg+L+2HL)=22.2
K(Hg+2HL)=12.9

Hg++ vlt KNO3 25°C 0.10M U B2=23.18 1955NMa (23179) 557
B3=23.06

Hg++ ISE KNO3 25°C 1.0M U B2=23.42 1950BJa (23180) 558

C3H4N2 L Imidazole CAS 288-32-4 (90)
1,3-Diazole, imidazole; C3H4N2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hg++ gl NaClO4 25°C 0.30M C TIH T K1=9.18 B2=18.18 1997SJa (23898) 559
IUPAC evaluation. I=0.5 M: DH(K1)=-75.0 kJ mol-1, DH(K2)=-32.1; B2=18.19

Hg++ gl NaClO4 25°C 0.50M U H K1=9.18 B2=18.18 1978MHa (23899) 560
B(Hg(OH)L)=21.86
By calorimetry, DH1=-75 kJ mol-1, DS1=77, DH(B2)=-107, DH(HgOHL)=-114

Hg++ gl NaClO4 25°C 3.00M C I M K1=9.18 B2=18.19 1977SJc (23900) 561
K(HgL+Cl)=7.23
K(HgL+2Cl)=7.95
K(HgL+3Cl)=8.97
K(HgL+OH)=12.68
K(HgL2+Cl=HgL2Cl)=0.70; K(HgL2+2Cl=HgL2Cl2)=1.0

Hg++ gl KCl 25°C .058M U T K1=3.57 B2=6.95 1961SMa (23901) 562
K1=3.84(0 C), 3.26(45 C)

Hg++ ISE oth/un 27°C 0.15M U B2=16.74 1960BDa (23902) 563

C3H4N2S HL Imidazolethiol CAS 872-35-5 (1823)

2-Mercaptoimidazole; C₃H₃N₂.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	NaClO ₄	25°C	0.10M	U		K ₁ =8.34	1977STc (23971)	564
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C ₃ H ₄ O ₃		HL		Pyruvic acid			CAS 127-17-3	(1152)	
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2-Oxopropanoic acid; CH₃.CO.CO₂H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	vlt	NaClO ₄	30°C	0.20M	C		K ₁ =1.28	1989GMc (24053)	565
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Method: polarography. Medium pH 2.5.

Hg++	gl	NaClO ₄	25°C	0.11M	U	TIH	K ₁ =1.63	1984GMc (24054)	566
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Data for 30-50 C. Also data for 0.03-0.11 M NaClO₄. At I=0.0 M, K₁=2.32

DH(K₁)=31.9 kJ mol⁻¹, DS(K₁)=138 J K⁻¹ mol⁻¹.

C ₃ H ₄ O ₄		H ₂ L		Malonic acid			CAS 141-82-2	(79)	
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Propanedioic acid; CH₂(CO₂H)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	con	oth/un	25°C	var	U			1968SAa (24466)	567
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B(Hg+2HL)=4.58

C ₃ H ₅ N ₂ S ₂		H ₂ L					CAS 29596-83-6	(3558)	
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N-(Dithiocarboxy)aminoethanoic acid; HS.CS.NH.CH₂.CO₂H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	ISE	KN ₃	20°C	1.00M	U		B ₂ =32.0	1968TTa (24658)	568
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B₃=33.72

C ₃ H ₆ N ₂ O ₅		L					CAS 591-08-2	(1423)	
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N-Acetylthiourea; CH₃.CO.NH.CS.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	ISE	mixed	25°C	82%	U			1979MTd (24772)	569
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B₄=19.05

Medium: 82% DMF/H₂O

C ₃ H ₆ N ₂ O ₂		L		D-Cycloserine			CAS 68-41-7	(907)	
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D-4-Amino-1,2-oxazolidine-3-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	ISE	KN ₃	25°C	0.40M	U		B ₂ =14.7	1963L0a (24795)	570
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K(Hg(H-1L)₂+H)=5.93

$$K(\text{Hg}(\text{H}-1\text{L})\text{L}+\text{H})=5.44$$

C3H6OS2 HL Xanthic acid CAS 151-01-9 (590)
(Ethoxy)dithiomethanoic acid; CH3.CH2O.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp oth/un 25°C var U M 1970AFa (24874) 571
B(HgL(CN)2)=35.26

C3H6O2 HL Propionic acid CAS 79-09-4 (35)
Propanoic acid; CH3.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 3.0M C K1=4.33 B2=8.80 1977RWa (25008) 572

C3H6O2S HL CAS 2444-37-3 (1074)
(Methylthio)ethanoic acid; CH3.S.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 25°C 0.10M U B2=11.93 1986ECa (25091) 573

C3H6O3 HL Methoxyacetic CAS 625-45-6 (29)
Methoxyethanoic acid; CH3.O.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 3.00M C K1=3.54 B2=6.91 1977RWa (25599) 574

C3H7NOS2 HL CAS 59333-68-5 (4240)
N-(2-Hydroxyethyl)dithiocarbamic acid; HO.CH2.CH2.NH.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 20°C 1.00M U B2=31.7 1968TTa (25670) 575
B3=34.55

C3H7NO2 HL Alanine CAS 56-41-7 (86)
2-Aminopropanoic acid; H2N.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=12.4 B2=19.60 1974VBa (26185) 576

Hg++ vlt KNO3 25°C 0.60M U B2=18.5 1966TAb (26186) 577

Hg++ gl oth/un 21°C .005M U B2=18.8 1953PEa (26187) 578
Medium: 0.005 M Hg(NO3)2

 Hg++ gl oth/un 21°C 0.01M U B2=18.4 1952PEa (26188) 579
 Medium: Hg(NO3)2

 C3H7NO2 HL B-Alanine CAS 107-95-9 (575)
 3-Aminopropanoic acid; H2N.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaCl	37°C	0.15M	U		K1=3.49 B2=6.77	1992MTa (26455)	580

Hg++	gl	NaCl04	25°C	0.50M	C	T	K1=11.306 B2=19.481	1986GGa (26456)	581
							B(HgH-1L)=6.51		
							B(HgH-1L2)=10.02		

 C3H7NO2 HL Sarcosine CAS 107-97-1 (87)
 N-Methyl-2-aminoethanoic acid; CH3.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	oth/un	20°C	0.01M	U		B2=18.7	1952PEa (26603)	582

 Medium: Hg(NO3)2

 C3H7NO2S H2L Cysteine CAS 52-90-4 (96)
 2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaCl04	25°C	0.15M	C	H	K1=16.60	2003AGb (26783)	583
							K(CH3Hg+CH3HgL)=8.70		
							K'(CH3Hg+HL)=15.18		

Metal is CH3Hg+. Calorimetry: DH(K1)=-76.2 kJ mol⁻¹, DS(K1)=62 J K⁻¹ mol⁻¹; DH(K)=-57, DS(K)=-25; DH(K')=-86.2, DS(K')=1.

Hg++	gl	NaCl04	25°C	0.15M	C	H		2003AGb (26784)	584
							K(CH3Hg+(CH3Hg)2L)=3.47		
							K'(CH3Hg+CH3HgHL)=4.15		
							K"(CH3Hg+H2L)=8.96		

Metal is CH3Hg+. Calorimetry: DH(K)=-22 kJ mol⁻¹, DS(K)=-8 J K⁻¹ mol⁻¹; DH(K')=-36.2, DS(K')=-42; DH(K")=-56, DS(K")=-16.

Hg++	nmr	KNO3	25°C	0.30M	U			1988CAb (26785)	585
							K3=0.87		
							K(HgL2+HL)=1.01		
							K(HgHL2+HL)=1.49		

Hg++	dis	NaCl04	25°C	1.0M	C		B2=42.7	1988SKf (26786)	586
							K(Hg+2HL)=40.0		

Method: extraction of 203Hg-labelled complex into dithizone/CCl4 solution.

Hg++ EMF NaClO4 25°C 0.10M U K1=37.8 B2=44.00 1981BCc (26787) 587
K(Hg+2HL)=38.3

Hg++ gl oth/un 25°C 1.00M U K1=43.68 1977NZa (26788) 588
Medium: KI

Hg++ gl NaNO3 25°C 0.10M U B2=39.4 1974VBa (26789) 589

Hg++ gl KNO3 25°C 0.10M U K1=14.21 1964LMa (26790) 590

Hg++ gl oth/un 20°C 0.00 U B2=20.5 1953PEa (26791) 591
Medium: 0.0025 M Hg(NO3)2

Hg++ ISE KNO3 25°C 1.0M U T K1=43.57 1953SKb (26792) 592
K1=45.40(12 C)

C3H7NO3 HL Serine CAS 56-45-1 (49)
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Hg++ gl KNO3 35°C 0.10M C M K1=8.96 B2=17.60 1998ZWa (27139) 593
Data for ternary complexes with 3,3,9,9-tetramethyl-4,8-diazaundecane-
2,10-dione dioxime

Hg++ gl KNO3 25°C 0.10M M M K1=4.42 1996AEa (27140) 594
Data for ternary complexes with dipicolinic acid.

Hg++ gl NaNO3 25°C 0.10M U K1=11.7 B2=19.10 1974VBa (27141) 595

Hg++ vlt KNO3 25°C 0.60M U B2=17.34 1966TAb (27142) 596

Hg++ gl oth/un 20°C .005M U B2=17.5 1953PEa (27143) 597
Medium: 0.005 Hg(NO3)2

C3H7NS2 HL CAS 128-04-1 (2125)
Dimethyldithiocarbamic acid; (CH3)2N.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 25°C 0.10M U B2=36.0 1991BSe (27275) 598

C3H8N2O2 HL CAS 71292-18-7 (356)
2,3-Diaminopropanoic acid; H2N.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 25°C 0.10M U K1=8.48 B2=15.90 1971HMd (27551) 599
K(Hg+HL)=4.56
K(Hg+L+HL)=12.32

C3H8N2S L DiMe-Thiourea CAS 61805-96-7 (1078)
1,3-Dimethylthiourea; CH3.NH.CS.NH.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	KNO3	25°C	0.09M	U	T H		K1=13.45 B2=17.62 B3=21.02	1988MMe (27627)	600

DH(K1)=-194 kJ mol⁻¹; DH(B2)=-216; DH(B3)=-195

C3H8N2S L Ethyl-thiourea CAS 625-53-6 (1079)
N-Ethylthiourea; C2H5.NH.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	KNO3	25°C	0.09M	U	T H		K1=14.42 B2=18.66 B3=22.46	1988MMe (27633)	601

DH(K1)=-110 kJ mol⁻¹; DH(B2)=-28; DH(B3)=-93

C3H8OS2 H2L BAL CAS 59-52-9 (379)
2,3-Dimercaptopropan-1-ol; HS.CH2.CH(SH).CH2(OH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	NaClO4	25°C	0.10M	U			K1=44.1 B2=51.90 K(Hg+2HL)=45.1	1981BCc (27659)	602

Hg++	ISE	NaClO4	25°C	0.10M	U			K1=44.8 B2=51.91	1980CJa (27660)	603
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Hg++	gl	NaClO4	25°C	0.10M	U			K1=25.74 B2=34.35	1977CJb (27661)	604
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C3H8O3S3 H3L (1324)
1,3-Dimercaptopropanesulfonic acid; HS.CH2.CH2.CH(SH).SO3H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	NaClO4	25°C	0.10M	U			K1=41.8 B2=53.10 K(Hg+2HL)=41.0	1981BCc (27763)	605

Hg++	EMF	KNO3	?	0.10M	U			K1=41.29	1966PRa (27764)	606
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C3H8O3S3 H3L Unithiol CAS 74-61-3 (1271)
2,3-Dimercaptopropanesulfonic acid; HS.CH2.CH(SH).CH2.SO3H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.50M	U	TI		K1=26.47 B2=35.45	1995NUa (27787)	607

Data also for I=0, I=1.0, I=1.5

Hg++	ISE	NaClO4	25°C	0.10M	U			K1=42.2 B2=53.10	1980CJa (27788)	608
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Hg++	dis	oth/un	?	?	U	B2=53.1	1973RPa (27789)	609
Hg++	EMF	KN03	?	0.10M	U	K1=39.71	1966PRa (27790)	610

C3H8S		HL		Propylmercaptan	CAS 75-33-2	(2515)		
2-Propanethiol, Isopropylmercaptan; CH3.CH(SH).CH3								
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Hg++	gl	alc/w	20°C	25%	U T H	K1=7.84	1978SKf (27807)	611
DH=-44.02 kJ mol-1. Data also available when T=10 and 30. Alternative methods: Conductivity and amperometric techniques.								
Hg++	con	alc/w	20°C	25%	C TIH		1978SKj (27808)	612
Kso(HgL2)=-7.84								
Medium: 25% v/v EtOH/H2O. Additional methods: potentiometry (25% EtOH/H2O) polarography (25% EtOH/H2O, 0.2 M NaClO4). Data for 10 and 30 C. DH values								

C3H9As		L				CAS 593-88-4	(2296)	
Trimethylarsine; (CH3)3.As								
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Hg++	cal	non-aq	30°C	100%	U H		1976GGb (27809)	613
In benzene. DH(HgCl2L)=-74.8 kJ mol-1; DH(HgBr2L)=-71.9.								
Hg++	cal	non-aq	30°C	100%	U H		1976GGb (27810)	614
K(HgI2+L=0.5(HgI2L)2)=5.78								
In benzene. DH=-70.0 kJ mol-1(Hg); DS=-120.								

C3H9N		L		n-Propylamine	CAS 107-10-8	(2356)		
1-Aminopropane; H2N.CH2.CH2.CH3								
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Hg++	vlt	non-aq	25°C	100%	U	B2=16.7	1973MTa (27832)	615
Medium: DMF, 0.5M (C2H5)4NClO4								

C3H9N		L		iso-Propylamine	CAS 75-31-0	(157)		
2-Propylamine; CH3.CH(CH3).NH2								
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference ExptNo
Hg++	gl	oth/un	25°C	0.50M	U	K1=8.75 B2=18.27	1983HNa (27845)	616
Medium: 0.5 M LHN03								
Hg++	vlt	non-aq	25°C	100%	U	B2=16.3	1973MTa (27846)	617
Medium: DMF, 0.5M (C2H5)4NClO4								

C3H9NO L CAS 109-85-3 (1575)
 2-Methoxyethylamine; CH3O.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	30°C	1.0M	U	T	K1=9.19 B2=17.84	1964PCa (27903)	618

50 C: K1=8.27, K2=8.36

C3H9OPS2 HL CAS 999-83-7 (4241)
 Methyl(ethyl)dithiophosphonic acid; (CH3S)(C2H5S)PO.H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U		B2=28.3 B3=30.3	1970TCa (27993)	619

Medium: 40% EtOH, 0.3 M KNO3

C3H10N2 L CAS 78-90-0 (2905)
 1,2-Diaminopropane; CH3.CH(NH2)CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	non-aq	25°C	100%	U		B2=24.0	1973MKa (28165)	620

Medium: DMF, (C2H5)4NClO4

Hg++	vlt	KNO3	40°C	0.10M	U	T H	B2=22.23 B3=21.85	1961RMa (28166)	621
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B2=24.75(10 C), 23.51(25 C); B3=24.57(10 C), 23.30(25 C).
 DH(B2)=-141.3 kJ mol⁻¹, DS=-25.1 J K⁻¹ mol⁻¹

Hg++	vlt	KNO3	25°C	0.10M	U		B2=23.53 B3=23.25	1955NRa (28167)	622
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C3H10N2 L Propanediamine CAS 109-76-2 (123)
 1,3-Diaminopropane; H2N.CH2.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.15M	C	H	K1=7.81 K(CH3Hg+CH3HgL)=3.92 K'(CH3Hg+HL)=6.77	2003AGb (28304)	623

Metal is CH3Hg+. By calorimetry, DH(K1)=-51 kJ mol⁻¹, DS(K1)=-23 J K⁻¹ mol⁻¹. DH(K)=-38.6, DS=-56; DH(K')=-45.9, DS=-24.

Hg++	gl	KNO3	20°C	0.10M	C	M	B(HgAL)=19.55 K(HgA+L)=12.84 K(HgA+L=HgAL(OH)+H)=11.07	2003Lba (28305)	624
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A is cytidine.

Hg++ gl KNO3 20°C 0.10M U K1=15.42 B2=19.90 1999LBa (28306) 625
B(HgH2L2)=37.50

Hg++ vlt non-aq 25°C 100% U B2=11.85 1979SZa (28307) 626
Medium: DMSO, 0.1 M NaClO4

Hg++ gl oth/un 25°C 0.10M U M 1976PBa (28308) 627
K(Hg(CN)2+L)=1.08
K(Hg(CN)2+HL)=0.25

C3H10N2 L CAS 109-81-9 (1308)
N-Methyl-1,2-diaminoethane; CH3.NH.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 1.00M C H B2=22.45 1982ABc (28362) 628
By calorimetry: DH(B2)=-110.5 kJ mol⁻¹, DS(B2)=58.6

C3H11N3 L CAS 21292-99-6 (2975)
Propane-1,2,3-triamine; H2N.CH2.CH(NH2).CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.15M C H K1=8.54 2003AGb (28487) 629
K(CH3Hg+CH3HgL)=6.11
K'(CH3Hg+HL)=7.30
K''(CH3Hg+H2L)=2.99
K'''(CH3Hg+(CH3Hg)2L)=2.65

Metal is CH3Hg+. Calorimetry: DH(K1)=-46.8 kJ mol⁻¹, DS(K1)=7 J K⁻¹ mol⁻¹;
DH(K)=-36, DS=-4; DH(K')=-44.4, DS=-9; DH(K'')=-32, DS=-32, DH(K''')=-12.

Hg++ gl KCl 20°C 0.50M U I K1=19.6 1950PSa (28488) 630
K(Hg+HL)=17.9

In 0.5 M KBr: K1=21.8, K(Hg+HL)=21.3

C4H2N2O4 L CAS 50-71-5 (8155)
1,3-Diazacyclohexane-2,4,5,6-tetraone, Pyrimidine-2,4,5,6-tetrone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 1.0M U K1=7.09 1976BMg (28617) 631

C4H2N2S2 H2L CAS 104409-71-4 (569)
1,2-Dicyano-1,2-dimercaptoethylene, Dimercaptomaleonitrile; (NC.C(SH):)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp NaClO4 25°C 0.30M U B2=43.31 1981ISa (28619) 632

C4H4N2 L Pyrimidine CAS 289-95-2 (4247)

1,3-Diazine, pyrimidine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	1.0M	U			K1=1.99	1976BMg (28778)	633

C4H4N2O2		HL		Uracil				CAS 66-22-8	(412)	
2,4-Dihydroxypyrimidone, 2,4-Pyrimidinedione;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=8.10 B2=14.96	1989MPa (28858)	634
Hg++	gl	NaClO4	30°C	0.10M	U			K1=7.30 B2=13.59	1978SSa (28859)	635

C4H4N2S		HL						CAS 1450-85-7	(1521)	
2-Mercapto-1,3-diazine, 2-Mercaptopyrimidine; C4H3N2.SH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	30°C	0.50M	U			K1=6.88 B2=11.88 B3=15.0	1989WIa (28937)	636

C4H5N3O		HL		Cytosine				CAS 71-30-7	(1096)	
2-Oxy-6-aminopyrimidine;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=7.85 B2=14.12	1989MPa (29411)	637
Hg++	ISE	NaClO4	27°C	0.10M	U			B2=10.90	1961FBa (29412)	638

C4H6N2		L		Methylpyrazole				CAS 453-58-3	(368)	
3-Methyl-1,2-diazole; C3H3N2.CH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	1.0M	U			K1=3.31	1976BMg (29503)	639

C4H6N2S		HL		Methimazole				CAS 60-56-0	(1824)	
N-Methyl-2-mercaptoimidazole; C3H2N2(CH3).SH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.10M	U			K1=8.50	1977STc (29663)	640

C4H6O4		H2L		Succinic acid				CAS 110-15-6	(112)	
1,4-Butanedioic acid; HOOC.CH2.CH2.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++      ISE oth/un 27°C 0.75M U      B2=7.28      1960YDa (29980) 641
              B(HgL(OH))=13.45
*****
C4H6O4S      H2L      Thiodiacetic      CAS 123-93-3 (140)
2,2'-Thiodiglycolic acid, Thiodiethanoic acid; H00C.CH2.S.CH2.C00H
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++      vlt KNO3      25°C 0.10M U      B2=13.51      1986ECa (30218) 642
-----
Hg++      sp NaCl04 25°C 0.50M C      K1=8.83      B2=15.95      1976NAb (30219) 643
              B(HgHL)=10.68
*****
C4H6O4S      H3L      Thiomalic acid      CAS 70-49-5 (109)
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; H00C.CH(SH).CH2.C00H
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++      gl KNO3      25°C 0.10M U      K1=9.94      B2=18.07      1965LMa (30335) 644
*****
C4H6O5      H2L      Diglycolic acid      CAS 110-99-6 (243)
Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; H00C.CH2.O.CH2.C00H
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++      ISE NaCl04 25°C 1.00M U      K1=6.7      1985MMa (30881) 645
              B(HgHL)=8.50
*****
C4H6O6      H2L      DL-Tartaric acid      CAS 133-37-9 (94)
DL-Tartaric acid,DL-2,3-Dihydroxybutanedioic acid; H00C.CH(OH).CH(OH).C00H
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++      sol NaCl04 20°C 3.00M U      M      1975ZCa (31024) 646
              B(Hg(Oxalate)L)=7.66
              B(HgOxL2)=9.16
*****
C4H6O6      H2L      L-Tartaric acid      CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; H00C.CH(OH).CH(OH).C00H
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++      con NaNO3      25°C 0.10M U      K1=7.0      1975LBa (31274) 647
-----
Hg++      dis NaCl04 20°C 0.10M U      K1=<4      1963STc (31275) 648
*****
C4H7NO2      HL      (8137)
(S)-Azetidine-2-carboxylic acid;
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	25°C	0.10M	C		K1=10.55 B2=20.97	1989ARa (31442)	649

C4H7NO4		H2L		Aspartic acid			CAS 56-84-8	(21)	
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	25°C	0.10M	M	M	K1=6.72	1996AEa (31865)	650
Data for ternary complexes with dipicolinic acid.									

Hg++	gl	KN03	25°C	0.10M	M		K1=6.10 B2=10.18	1981GVa (31866)	651

Hg++	gl	NaNO3	25°C	0.10M	U		K1=13.2 B2=20.00	1974VBa (31867)	652

C4H7NO4		H2L		IDA			CAS 142-73-4	(118)	
Iminodiethanoic acid; HN(CH2.COOH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.15M	C	H	K1=7.86 K(CH3Hg+CH3HgL)=2.50 K'(CH3Hg+HL)=2.33	2003AGb (32271)	653
Metal is CH3Hg+. Calorimetry: DH(K1)=-41.5 kJ mol-1, DS(K1)=11 J K-1 mol-1 DH(K)=-5.8, DS=28; DH(K')=-2.8, DS=35.									

Hg++	con	NaNO3	25°C	0.10M	U		K1=13.1 B2=20.20	1975LBa (32272)	654

Hg++	sol	NaClO4	25°C	0.10M	U		K1=11.76	1967SKg (32273)	655

C4H8N2O3		HL		Asparagine			CAS 70-47-3	(17)	
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	25°C	0.10M	M	M	K1=4.16	1996AEa (32701)	656
Data for ternary complexes with dipicolinic acid.									

Hg++	gl	NaNO3	25°C	0.10M	U		K1=11.4 B2=18.60	1974VBa (32702)	657

C4H8N2O3		HL		Gly-Gly			CAS 556-50-3	(54)	
Glycyl-glycine; H2N.CH2.CO.NH.CH2.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.20M	U		K1=2.58	1982RRd (33029)	658

Hg++	gl	oth/un	21°C	0.01M	U		B2=12.4	1952PEa (33030)	659
Medium: Hg(NO3)2									

C4H8N2S L Thiosinamine CAS 109-57-9 (2377)
1-Allylthiourea; CH2:CH.CH2.NH.CS.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF KNO3 25°C 0.09M U T H K1=15.03 B2=18.53 1988MMe (33156) 660
DH(K1)=9 kJ mol-1; DS(K1)=320 J K-1 mol-1; DH(B2)=-73.6; DS(B2)=107

C4H8N2S HL CAS 2055-46-1 (1522)
3,4,5,6-Tetrahydro-pyrimidine-2-thiol; C4H7N2.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 30°C 0.50M U K1=4.2 B2=11.50 1989WIa (33163) 661

C4H8O2 L Dioxan CAS 123-91-1 (2281)
1,4-Diethylene dioxide; cyclo(-CH2.CH2.O.CH2.CH2.O-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1974FGa (33201) 662
K(HgCl2+L)=0.58
K(HgBr2+L)=2.36

In benzene. For HgCl2, DH=-26.9 kJ mol-1; DS=-77.
For HgBr2, DH=-22.7; DS=-68.

C4H8O3 HL CAS 300-85-6 (30)
3-Hydroxybutanoic acid; CH3.CH(OH).CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 3.00M C K1=4.25 B2=8.36 1977RWa (33622) 663

C4H8O3 HL CAS 2544-06-1 (28)
3-Methoxypropanoic acid; CH3.O.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 3.00M C K1=4.24 B2=8.45 1977RWa (33634) 664

C4H8O3 HL CAS 591-81-1 (39)
4-Hydroxybutanoic acid; HO.CH2.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 3.0M C K1=4.35 B2=8.45 1977RWa (33655) 665

C4H8S L CAS 110-01-0 (150)
Tetrahydrothiophene; cyclo(-CH2.CH2.S.CH2.CH2-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	alc/w	25°C	50%	C		K1=>3.6	1979SRa (33735)	666
Hg++	cal	non-aq	30°C	100%	U	H	K(Hg(SCN)2+L)=1.36	1976FGa (33736)	667

Medium: MeCN. DH=-38.6 kJ mol⁻¹; DS=-102

C4H9NO2 HL Aminoisobutyric CAS 144-90-1 (188)
2-Amino-2-methylpropanoic acid; H2N.C(CH3)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	35°C	0.10M	C	M	K1=9.04 B2=18.09	1998ZWa (33839)	668

Data for ternary complexes with 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime

Hg++	gl	oth/un	19°C	0.01M	U		B2=18.3	1952PEa (33840)	669
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Medium: Hg(NO3)2

C4H9NO2 HL 2-Aminobutyric CAS 2835-81-6 (571)
2-Aminobutanoic acid; CH3.CH2.CH(NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaCl04	25°C	0.10M	U		K1=6.55 B2=12.02	1976SSf (33915)	670

Hg++	gl	oth/un	17°C	0.01M	U		B2=18.5	1952PEa (33916)	671
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Medium: 0.005-0.01 M Hg(NO3)2, 15-20 C

C4H9NO2 HL CAS 927-60-6 (4268)
Hydroxyisobutyramide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U	M		1970GSb (34005)	672

K(HgCl2+L=HgClL+Cl)=5.48

K(HgClL+L=HgL2+Cl)=5.09

K(HgClL+OH=HgLOH+Cl)=4.69

C4H9NO2S HL CAS 88806-98-8 (3019)
2-Amino-3-mercaptopropanoic acid methyl ester, cysteine methyl ester;
HSCH2CH(NH2)COOCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	25°C	0.10M	U		K2=6.33	1969PPd (34057)	673

K(2HgL+L=Hg2L3)=9.89

K(3HgL+L=Hg3L4)=12.86

C4H9NO2S HL Methylcysteine CAS 1187-84-4 (84)
2-Amino-3-methylmercaptopropanoic acid; H2N.CH(CH2.S.CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	dis	NaCl04	35°C	0.10M	U	M	K1=7.43 B2=13.38	1990TSb (34096)	674

Method: electrophoresis. Ternary complexes with NTA

Hg++	gl	KNO3	25°C	0.10M	U		K1=7.20 B2=13.01	1964Lma (34097)	675
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C4H9NO3 HL Threonine CAS 72-19-5 (48)
2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH3)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U		K1=11.7 B2=18.70	1974VBa (34307)	676
Hg++	gl	oth/un	20°C	.005M	U		B2=17.5	1953PEa (34308)	677

Medium: 0.005 Hg(NO3)2

C4H9N3S L (3587)
5-Methyl-2-thioxo-1,3,4-triazahex-4-ene (acetone thiosemicarbazone)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	alc/w	25°C	40%	U		B4=27.3	1961TKb (34433)	678

Medium: 40% EtOH, 0.052 M NaC2H3O2

C4H10N2O HL CAS 27620-10-6 (4273)
alpha-Hydroxyisobutyramidine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U	M		1970GSb (34516)	679

K(HgCl2+HL=HgClHL+Cl)=4.02
K(HgClHL+HL=Hg(HL)2+Cl)=3.57
K(HgClHL+OH=Hg(OH)HL+Cl)=4.86

C4H10N2O2 L (3588)
2,3-Diaminopropanoic acid methyl ester; CH2(NH2).CH(NH2).CO.OCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	oth/un	25°C	0.10M	U		K1=6.38 B2=11.48	1971HMD (34523)	680

K(HgLOH+H)=7.81

Hg++	gl	oth/un	25°C	0.10M	U		K1=6.38 B2=11.48	1968HMB (34524)	681
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K(HgLOH+H)=7.81

C4H10O2S L CAS 111-48-8 (4275)
3-Thiapentane-1,5-diol; HO.CH2.CH2.S.CH2.CH2.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	1.0M	C		K1=>3.85	1979SRa (34684)	682
Hg++	EMF	NaClO4	25°C	0.50M	C	H	K1=6.37 B3=11.5 B4=14.1	1976MHc (34685)	683

Method: Hg electrode. By calorimetry: DH(K1)=-33.3 kJ mol⁻¹; DH(B2)=-69.6, DS(B2)=-33 J K⁻¹ mol⁻¹; DH(B3)=-62, DS(B3)=13, DH(B4)=-70, DS(B4)=34

C4H11N L Butylamine CAS 109-73-9 (159)
1-Aminobutane; CH3.CH2.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	non-aq	25°C	100%	U		B2=16.8	1973MTa (34766)	684
Medium: DMF, 0.5 M Et4NClO4									
Hg++	gl	KNO3	25°C	0.50M	U		K1=8.74 K3=0.9 K4=1.0	1972BJa (34767)	685
Hg++	gl	oth/un	25°C	0.50M	U		K1=8.7 K3=0.9 K4=1	1950BJa (34768)	686

Medium: C4H11N.HNO3

C4H11N L t-Butylamine CAS 75-64-9 (158)
2-Amino-2-methylpropane; H2N.C(CH3)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	oth/un	25°C	0.50M	U		K1=8.77 B2=26.77	1983HNa (34789)	687
Medium: 0.5 M LHN03									

C4H11NOS L (1220)
1-Hydroxy-3-thia-5-aminopentane; HO.CH2.CH2.S.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	30°C	1.0M	U T		K1=10.51 B2=18.26	1953MCA (34889)	688
50 C: K1=8.99, K2=7.32									

C4H11NO2 L Diethanolamine CAS 111-42-2 (89)
2,2'-Iminodiethanol; HN(CH2.CH2.OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.50M	U	H		K1=7.84 B2=15.66	1978MHa (34958)	689

By calorimetry, DH1=-80 kJ mol⁻¹, DS1=117 J K⁻¹ mol⁻¹, DH(B2)=-115, DS(B2)=87

Hg++	ISE	alc/w	25°C	60%	U	I		B2=18.79 B3=19.6	1969MPe (34959)	690
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Medium: 0-100% EtOH, 0.4 M LiNO₃
 B2(0%)=17.70, B3(0%)=19.14, B3(100%)=20.26, B4(100%)=20.42

Hg++	gl	KN03	25°C	0.50M	U			K1=7.84 B2=15.66	1956BJb (34960)	691
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 C4H11NO3 L Tris buffer CAS 77-86-1 (550)
 2-Amino-2-(hydroxymethyl)-propan-1,3-diol; (HO.CH2)3C.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.15M	C	H		K1=6.71	2003AGb (35056)	692

Metal is CH3Hg+. By calorimetry, DH(K1)=-44.6 kJ mol⁻¹, DS(K1)=-21 J K⁻¹ mol⁻¹.

 C4H11NS HL CAS 108-02-1 (1792)
 1-Mercapto-2-(N,N-dimethyl)aminoethane; HS.CH2.CH2.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KN03	20°C	0.25M	U	I		K1=10.36 B2=20.28	1973MSd (35136)	693

0.25 KN03, 25% MeOH: K1=10.52, K2=9.93; 25% EtOH: K1=10.55, K2=10.32

 C4H11OPS HL CAS 866-53-5 (3595)
 Diethylphosphinothioic acid; (C2H5)2.POSH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	alc/w	25°C	40%	U			B2=26.2 B3=29.8 B4=33.5	1967TSa (35201)	694

Medium: 40% EtOH, 0.8 M KN03

 C4H11OPS2 HL CAS 999-87-1 (4284)
 O-(1-Methylethyl) hydrogen P-methylphosphonodithioate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U			B2=28.5 B3=31.0	1970TCa (35202)	695

Medium: 40% EtOH, 0.3 M KN03

 C4H11OPS2 HL CAS 995-79-9 (4283)

O-Ethyl hydrogen P-ethylphosphonodithioate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U		B2=29.0 B3=31.6	1970TCa (35206)	696

Medium: 40% EtOH, 0.3 M KNO₃

C4H110PS2 HL CAS 1000-53-9 (4285)

O-Propyl hydrogen P-methylphosphonodithioate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U		B2=28.6 B3=31.2	1970TCa (35208)	697

Medium: 40% EtOH, 0.3 M KNO₃

C4H1102PS2 H3L CAS 298-06-6 (210)

O,O'-Diethyldithiophosphoric acid; (C₂H₅O)₂P(S)SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U		B2=27.7 B3=28.2 B4=29.9	1970TCa (35232)	698

Medium: 40% EtOH, 0.3 M KNO₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	alc/w	25°C	40%	U		B2=29.1 B3=30.9 B4=33.7	1967TSa (35233)	699

Medium: 40% EtOH, 0.8 M KNO₃

C4H1103P HL CAS 762-04-9 (1329)

Diethylphosphonic acid; (C₂H₅O)₂P(O)H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	nmr	non-aq	25°C	100%	U	I M		1981PEa (35245)	700
							K(HgCl ₂ +HL)=0.11 K(HgBr ₂ +HL)=-0.60 K(HgI ₂ +HL)=-1.04 K(Hg(SCN) ₂ +HL)=0.76		

Medium: pyridine: K(Hg(CN)₂+HL)=-2.0; K(HgCl₂+HL)=-0.17, K(HgBr₂+HL)=-0.32,
In DMSO: K(HgI₂+HL)=-0.54, K(Hg(SCN)₂+HL)=0.11

C4H1103PS HL CAS 2465-65-8 (3596)

Phosphorothioic acid OO-diethyl ester; (C₂H₅O)₂.POSH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ ISE alc/w 25°C 40% U B2=24.0 1967TSa (35250) 701
 B3=26.4
 B4=27.7

Medium: 40% EtOH, 0.8 M KNO3

Hg++ vlt oth/un 25°C 1.0M U B2=23.5 1961TSa (35251) 702
 B3=22.6
 B4=20.1

 C4H11O3PSe HL CAS 41118-97-2 (3597)
 Phosphoroselenoic acid 00-diethyl ester; (C2H5O)2.PSeH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	KNO3	25°C	1.0M	U		B2=28.6 B3=30.3 B4=32.7	1966TSa (35253)	703

 C4H11PS2 HL CAS 886-54-6 (3591)
 Diethylphosphinodithioic acid; (CH3.CH2)2PSSH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U		B2=31.1	1970TCa (35294)	704
Medium: 40% EtOH, 0.3 M KNO3									
Hg++	ISE	alc/w	25°C	40%	U		B2=33.6 B3=36.0 B4=38.2	1967TSa (35295)	705

Medium: 40% EtOH, 0.8 M KNO3

 C4H12N2 L Putrescine CAS 110-60-1 (360)
 1,4-Diaminobutane; H2N.(CH2)4.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	C	M	B(HgAL)=20.50 K(HgA+L)=13.79 K(HgA+L=HgAL(OH)+H)=11.36	2003Lba (35363)	706

A is cytidine.

Hg++	gl	KNO3	20°C	0.10M	U		K1=16.94 B(HgH2L2)=39.44	1999Lba (35364)	707
Hg++	gl	oth/un	20°C	1.0M	U		K1=17.96 K(Hg+HL)=10.99 K(HgHL+HL)=6.83	1962SSc (35365)	708

Medium: Ba(ClO4)2

C4H12N2 L CAS 563-86-0 (59)
DL-2,3-Diaminobutane; H2N.CH(CH3).CH(CH3).NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U		K1=21.51	1962SKa (35379)	709

C4H12N2 L Dimeen CAS 110-70-3 (125)
N,N'-Dimethyl-1,2-diaminoethane; CH3.NH.CH2.CH2.NH.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	non-aq	25°C	100%	U		B2=11.78	1979SZa (35421)	710

Medium: DMSO, 0.1 M NaClO4

C4H12N2 L Butanediamine CAS 20759-15-3 (58)
meso-2,3-Diaminobutane; H2N.CH(CH3).CH(CH3).NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U		K1=20.33	1962SKa (35489)	711

C4H13N3 L Dien CAS 111-40-0 (584)
1,4,7-Triazaheptane, 2,2'-Iminobis(ethylamine), diethylenetriamine;
NH2.(CH2)2.NH.(CH2)2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	C	M		2004LBa (35784)	712

B(HgAL)=21.60
K(HgA+L)=9.80
B(HgAHL)=28.15
K(HgA+HL)=6.39

H2A is cytidine-5'-monophosphoric acid.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	C	M		2003LBa (35785)	713

B(HgAL)=22.08
K(HgA+L)=15.37
K(HgA+L=HgAL(OH)+H)=14.21
B(HgAHL)=27.47

A is cytidine. B(HgA+HL)=10.80.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U		K1=17.62 B2=24.41	1999LBa (35786)	714

B(HgHL)=20.72
B(HgHL2)=32.86

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	oth/un	25°C	0.10M	U	M		1976PBa (35787)	715

K(Hg(CN)2+L)=2.52
K(Hg(CN)2L+Hg(CN)2)=0.52
K(Hg(CN)2+HL)=1.68

Hg++ vlt KNO3 40°C 0.10M U T H B2=23.76 1961RMa (35788) 716
B3=23.43

At 25 C: B2=25.02, B3=24.5; DH(B2)=-150.5kJ mol⁻¹, DS=-29.3 J K⁻¹ mol⁻¹

Hg++ vlt KNO3 25°C 0.10M U B2=25.06 1955NRa (35789) 717
B3=24.00

Hg++ gl KCl 20°C 0.50M U K1=21.8 1950PSa (35790) 718
K(HgOHL+H)=7.7

C5H4N2O4 H2L Orotic acid CAS 65-86-1 (624)

1,2,3,6-Tetrahydro-2,6-dioxo-4-pyrimidinecarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=8.6 1987MPa (36113) 719
B(Hg2L3)=20.03
K(Hg+L2)=13.2

L2=orotic acid dimer

C5H4N4O HL Hypoxanthine CAS 68-94-0 (1174)

6-Hydroxypurine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 35°C 0.10M U TIH K1=7.37 1979RPa (36191) 720
Medium: KClO4. DH(K1)=-9.37 kJ mol⁻¹, DS(K1)=109 J K⁻¹ mol⁻¹.

At 45 C, K1=7.46. At 35 C, I=0.0 M: K1=9.55.

C5H5N L Pyridine CAS 110-86-1 (31)

Pyridine, Azine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.50M U H K1=5.1 B2=10.0 1978MHa (36639) 721
By calorimetry, DH1=-36 kJ mol⁻¹, DS1=23, DH(B2)=-69, DS(B2)=40

Hg++ cal non-aq 30°C 100% U M 1976AGa (36640) 722
K(HgI2+Py)=1.74

Medium: MeCN

Hg++ cal non-aq 30°C 100% U H 1976FGa (36641) 723
K(Hg(SCN)2+L)=1.84

Medium: MeCN. DH=-41.5 kJ mol⁻¹; DS=-102.

Hg++ cal non-aq 30°C 100% U H 1974DGa (36642) 724
K(HgA2+2L)=0.52

In benzene. HA=thiobenzoyl-1,1,1-trifluoroacetone; DH=-31 kJ mol⁻¹; DS=-93

Hg++ gl KNO3 25°C 0.50M U K1=5.1 B2=10.00 1972BJa (36643) 725
K3=0.3
K4=0.3

Hg++ gl oth/un 25°C 0.50M U K1=5.1 B2=10.0 1950BJa (36644) 726
K3=0.4

Medium: 0.5 M C5H5N.HNO3

C5H5N5 L Adenine CAS 73-24-5 (237)
6-Aminopurine; H2N.C5H3N4

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ dis NaClO4 35°C 0.10M C M K1=7.81 B2=13.92 1989MMf (36971) 727
K(Hg(nta)+L)=5.95
B(Hg(nta)L)=19.25

Method: paper electrophoresis. Medium pH=8.5.

C5H6N2O2 HL Thymine CAS 65-71-4 (413)
2,4-Dihydroxy-5-methylpyrimidine; C4HN2(CH3)(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.10M U K1=10.65 B2=20.70 1984PTa (37276) 728
Additional method: Hg electrode.

Hg++ gl NaClO4 30°C 0.10M U K1=7.93 B2=14.70 1978SSa (37277) 729

Hg++ ISE NaClO4 27°C 0.10M U B2=21.2 1961FBa (37278) 730

C5H7NO2 HL Glutarimide CAS 1121-89-7 (4312)
Piperidine-2,6-dione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 45°C 50% C K1=8.69 B2=15.77 1996MMc (37509) 731
Medium: 50% v/v MeOH/H2O, 0.10 M KNO3.

C5H7N04S2 H3L CAS 36061-59-3 (1953)
Bis(carboxymethyl)dithiocarbamic acid; (HOOC.CH2)2.N.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF KNO3 20°C 1.00M U B2=30.92 1968TTa (37557) 732
B3=32.33

C5H8N2S L (6089)
N,N'-Divinylthiourea; CH2:CH.NH.CS.NH.CH:CH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Hg++ EMF KNO3 25°C 0.09M U T H K1=13.38 B2=17.04 1988MMe (37724) 733
 DH(K1)=-141 kJ mol⁻¹; DS(K1)=-218 J K⁻¹ mol⁻¹; BH(B2)=-196; DS(B2)=-332

 C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
 Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ con NaNO3 25°C 0.10M U K1=12.9 B2=20.10 1975LBa (37985) 734

 C5H8O4S2 H2L CAS 2068-24-8 (908)
 2,2'-(Methylenebis(thio))bis-ethanoic acid; HOOC.CH2.S.CH2.S.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ vlt KNO3 25°C 0.10M U B2=10.25 1986ECa (38395) 735

 C5H9NO2 HL Proline CAS 147-85-3 (44)
 Pyrrolidine-2-carboxylic acid; C4H8N.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ gl KNO3 35°C 0.10M C K1=9.44 B2=18.74 1998ZWa (38618) 736
 Data for ternary complexes with 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime.

 Hg++ gl NaNO3 25°C 0.10M U K1=12.2 B2=20.10 1974VBa (38619) 737

Hg++ vlt KNO3 25°C 0.60M U B2=19.25 1966TAb (38620) 738

Hg++ gl oth/un 17°C 0.01M U B2=20.5 1952PEa (38621) 739
 Medium: Hg(NO3)2

 C5H9NO3 HL Hydroxyproline CAS 51-35-4 (416)
 4-Hydroxy-2-pyrrolidinecarboxylic acid; C4H7N(OH)(COOH)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ gl oth/un 17°C 0.01M U B2=17.7 1952PEa (38735) 740
 Medium: Hg(NO3)2

 C5H9NO3S H2L N-Acetyl-Cys CAS 616-91-1 (1187)
 N-Acetylcysteine; CH3.CO.NH.CH(CH2.SH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ EMF NaClO4 25°C 0.10M U K1=38.4 B2=45.60 1981BCc (38816) 741
 K(Hg+2HL)=35.0

C5H9NO4 H2L Glutamic acid CAS 56-86-0 (22)
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.10M	M		K1=5.60 B2= 9.66	1981GVa (39087)	742
Hg++	gl	NaNO3	25°C	0.10M	U		K1=12.8 B2=19.20	1974VBa (39088)	743
Hg++	EMF	oth/un	25°C	?	U		K1=6.30 B2=11.36	1972SSe (39089)	744

C5H9NO4 H2L MIDA CAS 4408-64-4 (190)
N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.10M	U		K1=5.47 B2=9.15	1955SAa (39256)	745

K(HgL(OH)2+H=HgLOH)=9.18

C5H9NS2 HL CAS 25769-03-3 (3623)
Pyrrolidine-N-carboxydithioic acid; C4H8N-CSSH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	25°C	0.10M	U		B2=37.5	1991BSe (39333)	746

C5H9N3 L Histamine CAS 51-45-6 (103)
4(5)-(2'-Aminoethyl)imidazole; C3H3N2.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	.058M	U T		K1=6.01 B2=10.63	1961SMa (39539)	747

0 C: K1=6.82, K2=4.52; 45 C: K1=5.26, K2=4.36

C5H9N3S HL (1822)
2-Mercaptohistamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.10M	U		K1=12.45	1977STc (39608)	748

C5H10NO7P H4L PMIDA CAS 5994-61-6 (2433)
N-(Phosphonomethyl)iminodiethanoic acid; H2O3P.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	oth	KNO3	RT	0.10M	C		K1=>16	1980MVa (39678)	749

Method: paper electrophesis.

C5H10N2O2 HL CAS 2762-32-5 (3041)

Piperazine-2-carboxylic acid; C4H9N2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	22°C	0.10M	U			K1=4.1	1960REb (39723)	750

C5H10N2O2S		L						CAS 29061-28-7	(2621)	
4,5-Dimethoxyimidazolidine-2-thione;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE mixed		25°C	82%	U			K1=15.43 B2=18.02 B3=20.87	1979MTd (39728)	751

Medium: 82% DMF/H2O

C5H10N2O3		HL	Glutamine					CAS 56-85-9	(18)	
2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=11.5 B2=18.70	1974VBa (39818)	752
Hg++	gl	NaClO4	25°C	0.10M	U			K1=5.70 B2=10.35	1973TSb (39819)	753

C5H10N2O3		HL	Ala-Gly					CAS 687-69-4	(55)	
Alanyl-glycine; H2N.CH(CH3).CO.NH.CH2.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.20M	U			K1=2.42	1982RRd (39889)	754

C5H10N2O3		HL	Gly-DL-Ala					CAS 926-77-2	(66)	
Glycyl-DL-alanine; H2N.CH2.CO.NH.CH(CH3).COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.20M	U			K1=2.80 B2=5.00	1982RRd (39938)	755

C5H10O5S2		HL						CAS 110-50-9	(591)	
(Butoxy)dithiomethanoic acid; CH3.CH2.CH2.CH2O.CSSH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	dis	oth/un	25°C	0.25M	U			B2=>27	1982SAa (40159)	756
Hg++	sp	oth/un	25°C	var	U	M		B(HgL(CN)2)=35.43	1970AFa (40160)	757

C5H10O2		HL	Pivalic acid					CAS 75-98-9	(3026)	
Trimethylethanoic acid, 2,2-Dimethylpropanoic acid; (CH3)3C.COOH										

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       ISE oth/un 25°C 0.10M U          K1=5.92      1973LUa (40216) 758
*****
C5H11N          L                      CAS 1003-03-8 (304)
Cyclopentylamine;
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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl  oth/un 25°C 0.50M U          K1=8.89   B2=18.51  1983HNa (40394) 759
Medium: 0.1 M LHN03
*****
C5H11N          L    Piperidine      CAS 110-89-4 (105)
Perhydropyridine; cyclo(-CH2.CH2.CH2.NH.CH2.CH2-) C5H11N
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       ISE non-aq 25°C 100% U   H    K1=5.48   B2=11.92  1989PSa (40447) 760
B3=13.5
In DMSO; ionic medium: 0.1M Et4NC104
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Hg++       gl  KCl    25°C 0.50M U   H    K1=8.70   B2=17.44  1978MHa (40448) 761
By calorimetry, DH(K1)=-45 kJ mol-1, DS=23 J K-1 mol-1; DH(B2)=-74, DS=87
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Hg++       gl  KNO3   25°C 0.50M U          K1=8.74   B2=17.44  1972BJa (40449) 762
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Hg++       gl  KNO3   25°C 0.50M U          K1=8.70   B2=17.44  1950BJa (40450) 763
*****
C5H11NO          L                      CAS 109-02-4 (2279)
N-Methyl-tetrahydro-1,4-oxazine, N-methyl-morpholine;
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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       cal non-aq 30°C 100% U   H                      1974FGa (40464) 764
K(HgCl2+L)=2.44
K(HgBr2+L)=2.61
K(HgI2+L)=2.45
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In benzene. For HgCl2, DH=-37.9 kJ mol-1; DS=-79 J K-1 mol-1. For HgBr2,
DH=-37.1; DS=-73. For HgI2, DH=-33.6; DS=-64.
*****
C5H11NO2          HL    Valine      CAS 72-18-4 (43)
2-Amino-3-methylbutanoic acid; H2N.CH(CH3)2COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl  KNO3   35°C 0.10M C    M    K1=8.90   B2=17.42  1998ZWa (40715) 765
Data for ternary complexes with 3,3,9,9-tetramethyl-4,8-diazaundecane-
2,10-dione dioxime
-----

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Hg++ gl NaNO3 25°C 0.10M U K1=3.46 B2=6.83 1989MPa (40716) 766

Hg++ gl NaNO3 25°C 0.10M U K1=11.7 B2=18.90 1974VBa (40717) 767

C5H11NO2 HL Nor-Valine CAS 760-78-1 (689)
2-Aminopentanoic acid; CH3.CH2.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 19°C .005M U B2=17.7 1953PEa (40837) 768
Medium: 0.005 Hg(NO3)2.

Hg++ gl oth/un 20°C 0.00 U B2=17.6 1952PEa (40838) 769
Medium: 0.0005 Hg(NO3)2.

C5H11NO2S HL Methionine CAS 63-68-3 (42)
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=12.8 B2=19.50 1974VBa (41098) 770

Hg++ vlt KNO3 25°C 0.60M U B2=17.62? 1966TAb (41099) 771

Hg++ gl KNO3 25°C 0.10M U K1=6.52 B2=11.45 1964LMa (41100) 772

C5H11NO2S H2L D-Penicillamine CAS 52-67-5 (1323)
D-2-Amino-3-mercapto-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 0.10M C K1=18.86 B2=24.95 1983SLc (41187) 773
K(Hg+HL+L)=22.83

Hg++ EMF NaClO4 25°C 0.10M U K1=37.8 B2=44.50 1981BCc (41188) 774
K(Hg+2HL)=37.8

C5H11NO2S H2L Penicillamine CAS 52-66-4 (350)
DL-2-Amino-3-mercapto-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.50M U 1999KPa (41261) 775

B(HgHL2)=52.03

B(HgH3L3)=72.43

By solvent extraction: B(HgH2L2)=59.03, B(HgH3L2)=61.02, B(HgH4L2)=62.47.

Hg++ gl NaNO3 25°C 0.10M M M 1990SHa (41262) 776

K(HgL2+H)=8.88

K(HgL2+2H)=16.35

K(HgL2+3H)=19.06

K(HgL2+Ni)=8.22

K(HgL2+H+Ni)=13.53; K(HgL2+Co)=5.95

Hg++ nmr KNO3 25°C 0.30M U 1988CAb (41263) 777

K3=3.59

K(HgL2+HL)=3.35

K(HgHL2+HL)=3.11

Hg++ ISE NaClO4 25°C 0.10M U K1=38.3 B2=44.40 1980CJa (41264) 778

Hg++ gl KNO3 25°C 0.10M U K1=16.15 1964LMa (41265) 779

Hg++ gl KNO3 25°C 0.15M U K1=17.5 B2=23.50 1962KRa (41266) 780

C5H11NO2S2 HL CAS 1528-32-9 (2127)

Di(2-hydroxyethyl)dithiocarbamic acid; (HO.CH2.CH2)2N.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 20°C 1.00M U K1=31.96 B2=34.92 1968TTa (41298) 781

C5H11NS2 HL CAS 147-84-2 (2126)

Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 25°C 0.10M U B2=38.2 1991BSe (41352) 782

Also data for n-Pr(K1=38.0), i-Pr(42.9), n-Bu(39.4), i-Bu(42.0) and many other N-disubstituted dithiocarbamic acids.

Hg++ vlt non-aq 25°C 100% U K1=42.3 1977ZCa (41353) 783

Medium: DMSO, 0.2 M Bu4NClO4

Hg++ sp non-aq ? 100% U M 1968SRg (41354) 784

K(Hg(HA)2+2HL=HgL2+2H2A)=2.31

Medium: CCl4. H2A=dithizone

C5H12N2O HL CAS 62626-14-6 (4329)

2-Hydroxy-2-methylbutyramidine; H2N.C(:NH).C(OH)(CH3).CH2.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.10M U M 1970GSb (41462) 785

K(HgCl2+HL=HgClHL+Cl)=4.27

K(HgClHL+HL=Hg(HL)2+Cl)=3.95

K(HgClHL+OH=Hg(OH)HL+Cl)=4.76

C5H12N2O L TMU CAS 632-22-4 (146)

Tetramethylurea; (CH3)2N.CO.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	cal	oth/un	25°C	?	U	H		1980ACa (41478)	786
------	-----	--------	------	---	---	---	--	-----------------	-----

HgX2(s)+2L=HgL2X2(s) DH = -36.5 X = Cl, DH = -29.4 X = Br

C5H12N2O2	HL	Ornithine	CAS 1069-31-4	(46)
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2,5-Diaminopentanoic acid; H2N.CH2.CH2.CH2.CH(NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	KNO3	25°C	0.10M	U	I		1970CMc (41576)	787
------	----	------	------	-------	---	---	--	-----------------	-----

K(Hg+HL)=4.83
K(HgHL+HL)=4.32

I=1.0 M, K(Hg+HL)=4.52, K(HgHL+HL)=3.29

C5H12N2S	L	CAS 105-55-5	(2379)
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1,3-Diethylthiourea; C2H5.NH.CS.NH.C2H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	EMF	KNO3	25°C	0.09M	U	T H	K1=14.52 B2=18.67 B3=21.88	1988MMe (41623)	788
------	-----	------	------	-------	---	-----	----------------------------	-----------------	-----

DH(K1)=-155 kJ mol⁻¹; DH(B2)=-170; DH(B3)=-133

C5H12N2S	L	CAS 2782-91-4	(6088)
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N,N,N',N'-Tetramethylthiourea; (CH3)2N.CS.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	EMF	KNO3	25°C	0.09M	U	T H	K1=14.04 B2=18.36 B3=21.90	1988MMe (41628)	789
------	-----	------	------	-------	---	-----	----------------------------	-----------------	-----

DH(K1)=-190 kJ mol⁻¹; DH(B2)=-133; DH(B3)=-134

C5H12O3S4	H3L	CAS 19872-38-9	(4331)
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2,3-Dimercaptopropylthioethanesulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	dis	oth/un	?	?	U		B2=52.4	1973RPa (41654)	790
------	-----	--------	---	---	---	--	---------	-----------------	-----

Hg++	EMF	KNO3	?	0.10M	U		K1=41.33	1968PRc (41655)	791
------	-----	------	---	-------	---	--	----------	-----------------	-----

C5H12O4S3	H3L	CAS 19872-36-7	(4332)
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2,3-Dimercaptopropanoxyethanesulfonic acid; HS.CH2.CH(SH).CH2.O.CH2.CH2.HSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	dis	oth/un	?	?	U		B2=54.5	1973RPa (41668)	792
------	-----	--------	---	---	---	--	---------	-----------------	-----

Hg++ EMF KNO3 ? 0.10M U K1=41.45 1968PRc (41669) 793

 C5H12O5S4 H3L CAS 35617-14-2 (4333)
 2,3-Dimercaptopropanesulfonethanesulfonic acid; HS.CH2.CH(SH).CH2.SO2.CH2CH2.HSO3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ dis oth/un ? ? U B2=52.5 1973RPa (41699) 794

Hg++ EMF KNO3 ? 0.10M U K1=42.16 1968PRc (41700) 795

C5H13N L CAS 616-24-0 (5502)
 3-Aminopentane CH3.CH2.CH(NH2).CH2.CH3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 25°C 0.50M U K1=8.67 B2=18.08 1983HNa (41716) 796
 Medium: 0.1 M LHN03

C5H13OPS L CAS 14806-54-3 (3640)
 Diethylphosphinothioic acid O-methyl ester

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE alc/w 25°C 40% U B2=13.7 1967TSa (41805) 797
 B3=16.0
 B4=17.4

Medium: 40% EtOH; 0.8 M KNO3

C5H13OPS2 HL CAS 24392-61-8 (4340)
 O-(1-Methylethyl-hydrogen-P-ethylphosphonodithioate;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF alc/w 25°C 40% U B2=29.7 1970TCa (41806) 798
 B3=32.3

Medium: 40% EtOH, 0.3 M KNO3

C5H15N3 L CAS 13531-52-7 (738)
 1,4,8-triazaoctane, N-(2-Aminoethyl)propane-1,3-diamine; H2NCH2CH2NHCH2CH2CH2NH2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M C M 2004LBa (42004) 799
 B(HgAL)=24.27
 K(HgA+L)=12.47
 B(HgAHL)=30.43
 K(HgA+HL)=8.23

H2A is cytidine-5'-monophosphoric acid.

B(HgAL)=28.59
K(HgA+L)=21.88
B(HgAHL)=33.95
K(HgA+HL)=16.84

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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$$K(\text{HgL2+py})=1.30$$
$$K(\text{HgL2+bpv})=1.74$$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Hg++ ISE NaNO3 20°C 0.10M U K1=7.7 B2=15.4 1960ANb (42549) 804

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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$$K(\text{Hg}+\text{HL})=7.02$$
$$B(\text{Hg}(\text{OH})_2\text{L})=29.16$$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	---	--------	-----------	--------

Hg++ sp non-aq 25°C 100% U M 1974VPb (43167) 807
 K(L+HgCl2)=-0.69
 K(L+HgBr2)=-0.521

Medium: dichloromethane

Hg++ sp NaClO4 22°C 1.10M U 1964WDa (43168) 808
 K(Hg+L=HgH-1L+H)=2.5

 C6H6N2O HL CAS 873-69-8 (1258)
 Pyridine-2-aldoxime; C5H4N.CH:NOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	24°C	0.10M	U		K1=6.5 B2=12.20	1962BEa (43300)	809

 C6H6N2O L Nicotinamide CAS 98-92-0 (1473)
 Pyridine-3-carboxylic acid amide, Vitamin PP, C5H4N.CO.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	1.0M	U		K1=3.16	1976BMg (43342)	810

 C6H6N4O4 HL Furacilin CAS 59-87-0 (4360)
 5-Nitro-2-furfurylidene semicarbazone; NO2.C4H2O.CH:N.NH.CO.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	oth/un	?	?	U		B2=9.56 B(Hg2L3)=9.43	1968BAa (43503)	811

 C6H6O HL Phenol CAS 108-95-2 (457)
 Hydroxybenzene, phenol; C6H5.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	1.00M	C		K1=8.24 B2=15.75	1990ERa (43539)	812

 C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
 4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	NaClO4	25°C	0.10M	U		K1=19.86	1972GKc (44458)	813

 C6H7N L gamma-Picoline CAS 108-89-4 (325)
 4-Methylpyridine; C5H4N.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	M		1976AGa (44824)	814

$$K(\text{HgI}_2+\text{L})=1.80$$

Medium: MeCN

Hg++ cal non-aq 30°C 100% U H 1976FGa (44825) 815

$$K(\text{Hg}(\text{SCN})_2+\text{L})=2.07$$

Medium: MeCN. DH=-45.4 kJ mol⁻¹; DS=-110

C6H7N L Aniline CAS 62-53-3 (583)

Aminobenzene, aniline; C6H5.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 27°C 1.0M U K1=4.61 B2=9.21 1964WDa (44872) 816

C6H8N2 L 2-Picolylamine CAS 29722-36-9 (502)

2-(Aminomethyl)pyridine; C5H4N.CH2NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF NaNO3 20°C 0.10M U B2=20.08 1971ANa (45356) 817

C6H8N2O2S HL (8159)

4-Aminobenzenesulfamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 20°C 1.0M U K1=14.45 1976BMg (45394) 818

Medium: 1.0 M NaNO3 in 60% v/v EtOH/H2O

C6H8N2O3 HL (8157)

6-Methyl-5-methoxyuracil, 6-Methyl-5-methoxy-2,4-pyrimidinedione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 1.0M U K1=8.14 1976BMg (45399) 819

C6H8N2O4 H2L (3100)

Cyanomethyliminodiethanoic acid; NC.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M U K1=2.77 B2=4.97 1955SAa (45417) 820

C6H8N4O2S L CAS 42026-60-8 (8288)

6-Amino-3-methyl-2-(methylthio)-5-nitroso-4(3H)-pyrimidinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.1M U IH K1=5.76 B2=11.07 1984MMh (45442) 821

Data for I=0.01-0.20 M and 25-40 C. At I=0.0 M, K1=7.01, K2=6.73.

DH(K1)=-29.7 kJ mol⁻¹, DS(K1)=11.3 J K⁻¹ mol⁻¹; DH(K2)=-20.7, DS(K2)=32.7.

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; H₂OCCH₂.CH(OH)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	con	NaNO ₃	25°C	0.10M	U		K1=13.3 B2=18.80 K(Hg+HL)=6.1 K(Hg+H ₂ L)=4.1	1975LBa (46131)	822

Hg++ ISE NaClO₄ 25°C 0.10M U K1=10.9 1967SKe (46132) 823

C6H9NO₆ H3L NTA CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH₂.COOH)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	dis	NaClO ₄	25°C	0.10M	C		K1=13.30	1989MMf (46849)	824

Method: paper electrophoresis. Medium pH=8.5.

Hg++	sp	NaCl	20°C	0.10M	U		K1=13.51	1980KVa (46850)	825
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Hg++	ISE	NaClO ₄	25°C	0.50M	U		K1=13.48	1977GGb (46851)	826
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Hg++	ISE	KNO ₃	25°C	0.10M	U		K1=14.31	1977GNb (46852)	827
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Method: Hg-electrode

Hg++	sp	oth/un	20°C	?	U		K1=16.39 K(Hg+HL)=6.60	1969CAAd (46853)	828
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Hg++ ISE NaClO₄ 25°C 0.10M U T K1=14.6 1967SKe (46854) 829

C6H9N3OS L (7299)
4-Amino-6-methoxy-2-methylthio-pyrimidine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	C		K1=13.76 B(HgH ₂ L ₂)=38.4	1996LGc (47145)	830

C6H9N3O₂ HL Histidine CAS 71-00-1 (1)
2-Amino-3-(4'-imidazolyl)propanoic acid; H₂N.CH(CH₂.C₃H₃N₂)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO ₃	25°C	0.10M	U		B2=21.6	1974VBa (47565)	831
Hg++	vlt	KNO ₃	25°C	0.60M	U		B2=20.62	1966TAb (47566)	832

Hg++ gl KCl 25°C .058M U T K1=7.38 B2=12.38 1961Sma (47567) 833
0 C, K1=7.90, K2=5.50; 45 C, K1=6.85, K2=4.43

Hg++ ISE NaCl04 27°C 0.15M U B2=21.22 1960BDa (47568) 834
K(Hg+L+HL)=18.4
K(Hg+2HL)=15.0

C6H9N3O2S H2L Thiolhistidine CAS 13552-61-9 (5659)
1-Amino-2-(2-Mercaptoimidazole)-propionic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaCl04 25°C 0.10M U K1=12.43 1982TSb (47640) 835

C6H9O6P H3L CAS 4408-72-4 (7015)
Phosphinotriethanoic acid; P(CH2.COOH)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE NaCl04 25°C 0.10M U I K1=19.93 B2=32.27 1979PPc (47658) 836
B3=34.16

Method: Hg electrode. In 50% v/v dioxan/H2O: K1=22.8; B2=33.76; B3=34.93

Hg++ gl NaCl04 25°C 0.10M C I K1=19.93 B2=32.27 1979PPd (47659) 837
B3=34.16

Additional methods: polarography and Hg electrode.
In 50% dioxane/H2O: K1=22.8, B2=33.76, B3=34.93.

C6H10 L Cyclohexene CAS 110-83-8 (3054)
Cyclohexene; C6H10

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ dis KNO3 25°C 1.0M U K1=4.34 1939LHa (47669) 838

C6H10N2O4 H2L CAS 96705-91-8 (3103)
Piperazine-2,5-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 22°C 0.10M U K1=3.8 B2=7.1 1964PCa (47727) 839

C6H10N2O4 H2L (3104)
Piperazine-2,6-dicarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 22°C 0.10M U K1=5.1 B2=9.2 1964PCa (47737) 840

C6H10N2O4 H2L CAS 89601-09-2 (3102)

trans-Piperazine-2,3-dicarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	22°C	0.10M	U		K1=5.6 B2=9.6	1964PCa (47748)	841

C6H10N2O5		H2L	ADA				CAS 26239-55-4	(2747)	
N-(2-Acetamido)iminodiethanoic acid; H2N.CO.CH2.N(CH2.COOH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.10M	M	M	K1=6.44	1996AEa (47843)	842
Data for ternary complexes with dipicolinic acid									

Hg++	gl	KCl	20°C	0.10M	U		K1=3.82 B2=6.47	1955SAa (47844)	843

C6H10N4O5		L					(2622)		
4,5-Dimethyl-2,4,6,8-tetraazabicyclo[3,3,0]-octane-3-one-7-thione;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE mixed		25°C	82%	U		K1=15.77 B2=18.59 B3=21.22	1979MTd (47891)	844
Medium: 82% DMF/H2O									

C6H10O4S		H2L					CAS 111-17-1	(139)	
3,3'-Thiodipropionic acid; HOOC.CH2.CH2.S.CH2.CH2.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	25°C	0.10M	U		B2=12.78	1986ECa (48182)	845

C6H10O4S2		H2L					CAS 7244-02-2	(438)	
1,2-Bis(carboxymethylthio)ethane; HOOC.CH2.S.CH2.CH2.S.CH2.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	25°C	0.10M	U		B2=18.05	1986ECa (48243)	846

Hg++	gl	NaClO4	25°C	0.50M	U		B2=18.99	1980NAC (48244)	847

Hg++	gl	NaClO4	25°C	0.50M	C			1978NAb (48245)	848
K(HgL2+H)=2.10									
K(HgHL2+H)=0.41									

C6H10O8		H2L	Mucic acid				CAS 526-99-8	(3650)	
2,3,4,5-Tetrahydroxyhexanedioic acid, Galactaric acid; HOOC.(CHOH)4.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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C6H11NO2 HL CAS 37910-65-9 (6018)
2-Aminocyclopentane-1-carboxylic acid;

cis isomer

C6H11NO4S H3L CAS 58033-48-5 (3124)
N-2-Mercaptoethyliminodiethanoic acid; HS.CH2.CH2.N(CH2.COOH)2

 C6H11NO5 H2L HIMDA CAS 93-62-9 (192)
 N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2

 C6H11NO5 H2L (7174)
 N-Carboxymethylthreonine; HOOCCH2NHCH(CH(OH)CH3)COOH

C6H11O4P H2L CAS 85931-58-4 (5652)
Ethylphosphinediethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	g1	NaClO4	25°C	0.10M	U				1983NPa (49010)	854
								B4=40.9		
								B(HgHL4)=44.8		
								B(HgH2L4)=47.9		

B(HgH3L4)=50.7

Additional method: spectrophotometry. B(HgH4L4)=53.2, B(HgH5L4)=55.3,
B(HgH6L4)=57.2.

C6H12N2O3 HL DL-Ala-DL-Ala CAS 2867-20-1 (67)
DL-Alanyl-DL-alanine; H2N.CH(CH3).CO.NH.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 20°C 0.20M U K1=2.83 1982RRd (49129) 855

C6H12N2O4 H2L EDDA CAS 5657-17-0 (119)
1,2-Diaminoethane-N,N'-diethanoic acid; HOOC.CH2.NH.CH2.CH2.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp oth/un 25°C 0.10M U M K1=15.4 B2=24.2 1975Nka (49244) 856
Beff(MLCl)=9.9; Beff(MLBr)=12.0; Beff(MLSCN)=10.8. Conditions not stated.

C6H12N2O4 H2L N,N-EDDA CAS 5835-29-0 (2333)
1,2-Diaminoethane-N,N'-diethanoic acid; H2N.CH2.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 20°C 0.10M U K1=9.75 B2=15.80 1955SAa (49303) 857
K(HgLOH+H)=10.15

C6H12O7 HL Gluconic acid CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M C B(HgH-1L)=-0.9
B(HgH-2L)=-4.03

C6H13NO2 HL Isoleucine CAS 73-32-5 (424)
2-Amino-3-methylpentanoic acid; CH3.CH2.CH(CH3).CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 35°C 0.10M C M K1=9.01 B2=17.30 1998Zwa (49904) 859
Data for ternary complexes with 3,3,9,9-tetramethyl-4,8-diazaundecane-
2,10-dione dioxime

Hg++ gl NaNO3 25°C 0.10M U K1=12.4 B2=19.80 1974VBa (49905) 860

Hg++ gl oth/un 20°C 0.01M U B2=17.6 1952PEa (49906) 861
Medium: Hg(NO3)2.

C6H13NO2 HL Leucine CAS 61-90-5 (47)
2-Amino-4-methylpentanoic acid; H2N.CH(CH2.CH(CH3)2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=11.9 B2=19.50	1974VBa (50077)	862

Hg++	gl	oth/un	20°C	0.01M	U			B2=17.5	1952PEa (50078)	863
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Medium: Hg(NO3)2

C6H13NO2 HL Norleucine CAS 616-06-8 (602)
2-Aminohexanoic acid (2-Aminocaproic acid) CH3.(CH2)3.CH(NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaClO4	20°C	0.10M	U T H			K1=6.81 B2=12.40	1981SDb (50181)	864

Data for 20-40 C. DH(B2)=-59.5 kJ mol⁻¹, DS(B2)=33.8 J K⁻¹ mol⁻¹.

Hg++	gl	oth/un	19°C	0.00	U			B2=17.8	1952PEa (50182)	865
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Medium: 0.0005-0.005 Hg(NO3)2

C6H13NO2S HL Ethionine CAS 67-21-0 (1909)
2-Amino-4-(ethylthio)butanoic acid; CH3.CH2.S.CH2.CH2.CH(NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.10M	U			K1=7.25 B2=13.17	1964LMa (50265)	866

C6H13NO4 HL Bicine CAS 150-25-4 (2124)
N,N-Bis(2-hydroxyethyl)glycine; (HO.CH2.CH2)2N.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U			K1=14.17	1962SKa (50369)	867

C6H13N3O3 HL Citrulline (579)
2-Amino-5-ureidovaleric acid; H2N.CO.NH.CH2.CH2.CH2.CH(NH2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.10M	U			K1=6.02 B2=10.95	1970CMc (50578)	868

Hg++	gl	oth/un	20°C	.005M	U			B2=18.8	1953PEa (50579)	869
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Medium: 0.005 Hg(NO3)2

C6H14N2 L CAS 25155-35-5 (2282)
N,N-Dimethylpiperazine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ cal non-aq 30°C 100% U H 1974FGa (50656) 870

$K(\text{HgI}_2 + \text{L}) = 3.35$

In benzene. $\text{DH} = -41.1 \text{ kJ mol}^{-1}$; $\text{DS} = -72 \text{ J K}^{-1} \text{ mol}^{-1}$.

C6H14N202 HL Lysine CAS 56-87-1 (41)
2,6-Diaminohexanoic acid; $\text{H}_2\text{N} \cdot (\text{CH}_2)_4 \cdot \text{CH}(\text{NH}_2)\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=11.3 B2=18.70 1974VBa (50825) 871

C6H14N402 HL Arginine CAS 74-79-3 (40)
2-Amino-5-guanidopentanoic acid; $\text{H}_2\text{N} \cdot \text{CH}((\text{CH}_2)_3 \cdot \text{NH} \cdot \text{C}(:\text{NH})(\text{NH}_2)\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=11.5 B2=18.80 1974VBa (51010) 872

Hg++ gl KNO3 25°C 0.10M U K1=5.34 B2=10.21 1970CMc (51011) 873

Hg++ gl oth/un 19°C 0.00 U B2=17.4 1953PEa (51012) 874
Medium: 0.005 Hg(NO3)2

C6H14S L Isopropyl sulfi CAS 625-80-9 (5674)
2,2'-Thiodipropene, diisopropyl sulfide; $(\text{CH}_3)_2\text{CH} \cdot \text{S} \cdot \text{CH}(\text{CH}_3)_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE non-aq 25°C 100% U K1=7.84 B2=13.29 1986MMb (51138) 875
B3=15.92
B4=16.95

Medium: acetone, Bu4NClO4

C6H15N L Hexylamine CAS 111-26-2 (4352)
Hexylamine; $\text{CH}_3 \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{NH}_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE non-aq 25°C 100% U B2=13.43 1989PSa (51159) 876
In DMSO; ionic medium: 0.1M Et4NClO4

C6H15N L Triethylamine CAS 121-44-8 (1340)
N,N,N-Triethylamine; $(\text{C}_2\text{H}_5)_3\text{N}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 25°C 0.40M U K1=7.8 B2=14.8 1950BJa (51178) 877
Medium: 0.4 C6H15N, HNO3

C6H15NO3 Triethanolamine CAS 102-71-6 (447)

Tris-(2-hydroxyethyl)amine; L

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	ISE	alc/w	25°C	20%	U	I	B2=16.88 B3=17.31 B4=17.52	1969MPe (51293)	878
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Medium: 0-100% EtOH, 0.4 M LiNO₃. 0%: B2=16.9, B3=16.63, B4=17.27
B3(60%)=18.14, B4(100%)=18.96, B5(100%)=19.57

Hg++	gl	KN03	25°C	0.50M	U		K1=6.90 B2=13.08	1950BJa (51294)	879
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C6H150PS2 HL CAS 996-17-8 (4400)

O-Butyl hydrogen P-ethylphosphonodithioate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	EMF	alc/w	25°C	40%	U		B2=29.8 B3=32.4	1970TCa (51478)	880
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C6H1502PS2 HL (2059)

O,O'-Dipropyl dithiophosphoric acid; (C₃H₇O)₂P(S)SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	cal	non-aq	30°C	100%	U		K(2Hg _{L2} =Hg _{2L4})=0.00	1971DGB (51488)	881
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Medium: benzene

Hg++	EMF	alc/w	25°C	40%	U		B2=28.0 B3=29.4 B4=31.3	1970TCa (51489)	882
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Medium: 40% EtOH, 0.3 M KN03

C6H1502PS2 HL CAS 25134-38-7 (4401)

Phosphorodithioic acid O,O-diisopropyl ester; (CH₃.CH(CH₃)O)₂PS.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	ISE	alc/w	25°C	40%	U		B2=28.4 B3=29.7 B4=31.8	1970TCa (51501)	883
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Medium: 40% EtOH, 0.3 M KN03

C6H15015P3 H6L Ins(1,2,6)P3 CAS 28841-62-5 (6479)

D-myo-Inositol 1,2,6-trisphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	R4N.X	25°C	0.10M	U			1991LSa (51535)	884
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B(HgHL)=18.75

B(Hg2L)=20.92

B(Hg3L)=25.54

C6H15PS2 HL CAS 32338-34-4 (4394)

P,P-Di-(1-methylethyl)phosphinodithioic acid; ((CH3)2CH)2.PS.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF alc/w 25°C 40% U B2=33.2 1970TCa (51550) 885

Medium: 40% EtOH, 0.3 M KNO3

C6H15PS2 HL CAS 22689-71-0 (4395)

P,P-Dipropylphosphinodithioic acid; (CH3.CH2.CH2)2.PS.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF alc/w 25°C 40% U B2=32.8 1970TCa (51555) 886

Medium: 40% EtOH, 0.3 M KNO3

C6H16N2 L CAS 124-09-4 (358)

1,6-Diaminohexane; H2N.(CH2)6.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt NaNO3 20°C 1.0M U K1=10.89 1980BMc (51584) 887

Hg++ gl NaNO3 20°C 1.0M U K1=21.9 1976BMg (51585) 888

C6H16N2 L Tetrameen CAS 110-18-9 (124)

N,N,N',N'-Tetramethyl-1,2-diaminoethane; (CH3)2N.CH2.CH2.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt non-aq 25°C 100% U K1=2.60 B2=5.95 1979SZa (51645) 889

Medium: DMSO, 0.1 M NaClO4

Hg++ cal non-aq 30°C 100% U H 1976FGa (51646) 890

K(Hg(SCN)2+L)=4.48

Medium: MeCN. DH=-64.0 kJ mol-1; DS=-126

Hg++ cal non-aq 30°C 100% U H 1974FGa (51647) 891

K(HgCl2+L)=6.0

K(HgBr2+L)=6.60

K(HgI2+L)=6.30

In benzene. For HgCl2, DH=-90.2 kJ mol-1; DS=-182 J K-1 mol-1. For HgBr2, DH=-94.7; DS=-187. For HgI2, DH=-96.3; DS=-196.

C6H16N2O2 L CAS 93798-65-3 (3119)

3,6-Diaza-1,8-dihydroxyoctane; HO.CH2.CH2.NH.CH2.CH2.NH.CH2.CH2.OH

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl  NaNO3   25°C 0.10M U          K1=12.10          1986TSa (51688) 892
*****
C6H16N2O2          L                      CAS 929-59-4 (915)
3,6-Dioxaoctane-1,8-diamine; H2N.CH2.CH2.O.CH2.CH2.O.CH2.CH2.NH2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl  R4N.X   25°C 0.10M C  H    K1=18.55          1975ANa (51701) 893
Medium: Me4NNO3. DH(K1)=-102.5 kJ mol-1, DS=11.3
*****
C6H17N3          L                      CAS 35513-87-2 (292)
1,4,9-Triazanonane, 3-Azaheptane-1,7-diamine; H2NCH2CH2NHCH2CH2CH2CH2NH2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl  KNO3    20°C 0.10M U          K1=19.17          1999LBa (51849) 894
*****
C6H17N3          L                      CAS 56-18-8 (968)
1,5,9-Triazanonane, 4-azaheptane-1,7-diamine; H2N.CH2.CH2.CH2.NH.CH2.CH2.CH2.NH2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Hg++       gl  KNO3    20°C 0.10M C  M          2004LBa (51899) 895
                                B(HgAHL)=30.63
                                K(HgA+HL)=8.08
                                B(HgAH2L2)=46.69
H2A is cytidine-5'-monophosphoric acid.
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Hg++       gl  KNO3    20°C 0.10M C  M          2003LBa (51900) 896
                                B(HgAL)=23.28
                                K(HgA+L)=16.57
                                K(HgA+L=HgAL(OH)+L)=12.52
A is cytidine.
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Hg++       gl  KNO3    20°C 0.10M U          K1=19.75          1999LBa (51901) 897
                                B(HgHL)=25.10
*****
C6H18N4          L    Trien-tetramine CAS 112-24-3 (11)
1,4,7,10-Tetraazadecane; H2N.CH2.CH2.NH.CH2.CH2.NH.CH2.CH2.NH2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Hg++       EMF KNO3   25°C 1.0M C          K1=24.53          1983DWa (52106) 898
Method: Hg electrode.
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Hg++       ISE KNO3   25°C 1.00M C  H    K1=24.15          1982ABc (52107) 899
                                B(HgHL)=26.35
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By calorimetry: DH1=-123.8 kJ mol⁻¹, DS1=47.2

Hg++ EMF non-aq 25°C 100% U B2=18.00 1979SZa (52108) 900
Medium: DMSO

Hg++ vlt NaClO4 25°C 0.20M U H K1=24.5 1976KKb (52109) 901
DH=-126.3 kJ mol⁻¹, DS=45.2

Hg++ gl oth/un 25°C 0.10M U M 1976PBa (52110) 902
K(Hg(CN)2+L)=2.73
K(Hg(CN)2L+Hg(CN)2)=1.81
K(Hg(CN)2+HL)=2.19
K(Hg(CN)2HL+Hg(CN)2)=0.82

K(Hg(CN)2+H2L)=0.90

Hg++ gl KCl 25°C 0.10M U K1=25.0 1957RSb (52111) 903

Hg++ gl KCl 20°C 0.50M U K1=26.26 1950SCa (52112) 904
K(Hg+HL)=20.9

In 0.5 M KBr K1=26.35, K(Hg+HL)=23.6. Corrected for Hg-halide complexes

C6H18N4 L Tren CAS 4097-89-6 (817)

2,2',2''-Triaminotriethylamine; (H2N.CH2.CH2)3N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 20°C 0.50M U I K1=25.8 1950PSa (52199) 905

In 0.5 M KBr K1=27.3. Values corrected for Hg-halide complexes

In 0.1 M KCl K(Hg+HL)=4.5

C7H4N2O7 H2L CAS 609-99-4 (400)

3,5-Dinitrosalicylic acid; (O2N)2.C6H2(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 25°C 70% U K1=7.24 B2=12.15 1985ARa (52483) 906

Hg++ gl KNO3 35°C 0.10M U K1=3.20 1970DDa (52484) 907

C7H4O3Br2 H2L CAS 3147-55-5 (1116)

3,5-Dibromosalicylic acid; C6H2(OH)(Br)2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 25°C 70% U K1=9.64 B2=5.65 1985ARa (52542) 908

C7H4O3I2 H2L CAS 133-91-5 (4431)

3,5-Iodosalicylic acid; I2.C6H2.(OH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 25°C 70% U K1=9.52 B2=16.47 1985ARa (52559) 909

C7H5NO4 H2L Dipicolinic aci CAS 449-83-2 (418)
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 0.10M M M K1=5.45 1996AEa (52778) 910
Data for ternary complexes with aspartic acid, serine, asparagine and
N-(2-acetamido)iminodiacetic acid

Hg++ ISE NaNO3 20°C 0.10M U B2=20.28 1960ANb (52779) 911

C7H5O3Cl H2L CAS 321-14-2 (1113)
5-Chlorosalicylic acid; Cl.C6H3(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 25°C 70% U K1=11.52 B2=6.66 1985ARa (53345) 912

C7H6N2O4 H2L CAS 2683-49-0 (3753)
4-Aminopyridine-2,6-dicarboxylic acid (4-aminodipicolinic acid)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 20°C 0.10M U K1=15.80 B2=24.49 1965ABa (53509) 913

C7H6O2 HL Benzoic Acid CAS 65-85-0 (462)
Benzenecarboxylic acid; C6H5.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ nmr oth/un 28°C ? U M 1975HMa (53836) 914
K(LHgCl+OH=LHgOH+Cl)=4.04
K(LHgCl+His=LHgHis+Cl)=1.74

Hg++ nmr oth/un 28°C ? U M 1975HMa (53837) 915
K(LHgBr+OH=LHgOH+Br)=2.47

C7H6O2S H2L Thiosalicylic CAS 147-93-3 (236)
2-Mercaptobenzoic acid; HS.C6H4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 20°C 50% U K1=12.35 B2=21.1 1987SIa (53907) 916
Medium: 50% EtOH/H2O, 0.1 M NaClO4

Hg++ gl alc/w 25°C 70% U K1=15.98 B2=25.74 1985ARa (53908) 917

Hg++ sol oth/un 25°C 0.10M U K1=24.84 B2=33.47 1973KDb (53909) 918

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 25°C 100% M K1=7.5 B2=13.8 1994MPc (54232) 919

Hg++ gl alc/w 25°C 70% U K1=11.62 B2=19.53 1985ARa (54233) 920

C7H7NO2 HL Salicylamide CAS 65-45-2 (3155)
2-Hydroxybenzamide; HO.C6H4.CO.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl diox/w 30°C 50% U T H K1=4.61 B2= 8.36 1973PSc (55330) 921
Medium: 50% dioxane/H2O, 0.3 M KNO3. DH and DS values reported.
Data for 40 C.

C7H7NO2 HL 2-Pyridylacetic CAS 16179-97-8 (2211)
2-Pyridylethanoic acid; C5H4N.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 1.00M M K1=6.68 B2=10.60 1984CPa (55349) 922
B(HgL(OH))=16.76
B(HgL2(OH))=19.12
B(HgL(OH)2)=22.91

C7H7NO2 HL CAS 3222-47-7 (3154)
6-Methylpyridine-2-carboxylic acid; CH3.C5H3N.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE NaNO3 20°C 0.10M U B2=16.2 1960ANb (55429) 923

C7H7NO2 HL CAS 495-18-1 (184)
Benzohydroxamic acid; C6H5.CO.NH.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M M K1=4.88 B2= 9.66 1996KSc (55503) 924

C7H7NO3 H2L CAS 89-73-6 (204)
2-Hydroxybenzohydroxamic acid (salicylhydroxamic acid); HO.C6H4.CO.NHOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M M K1=6.01 B2= 9.51 1996KSc (55598) 925

 C7H7NS L Thiobenzamide CAS 2227-79-4 (1660)
 Thiobenzamide; C6H5.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	U			1977SWa (55705)	926
							K(HgCl2+L)=3.63 K(HgBr2+L)=3.88		

Medium: Et2O

 C7H7N2O2F3S HL CAS 73255-69-3 (559)
 2-(Trifluoromethanesulfonamidomethyl)pyridine; C5H4NCH2S(:O)2NHCF3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	diox/w	30°C	45%	U		K1=7.03 B2=12.56	1982MYb (55714)	927

Medium: 45% v/v dioxan/H2O, 0.01 M KN03

 C7H8 L CAS 108-88-3 (2144)
 Toluene; C6H5.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	U	M		1974VPb (55784)	928
							K(L+HgCl2)=-0.555 K(L+HgBr2)=-0.435		

Medium: dichloromethane

 C7H8N2OS L CAS 3394-05-6 (3182)
 N-3-Hydroxyphenylthiourea; HO.C6H4.NH.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ix	KN03	20°C	0.50M	U		K1=5.8	1958HOb (55851)	929

 C7H8N2O2 HL Salicylic hydra CAS 936-02-7 (2646)
 2-Hydroxybenzoic acid hydrazide; HO.C6H4.CO.NH.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	diox/w	25°C	25%	U		K1=5.87 B2=9.20	1975GSb (55875)	930

 C7H8N2S HL Phenylthiourea CAS 103-85-5 (625)
 1-Phenyl-2-thiourea; C6H5.NH.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KN03	25°C	0.10M	U		B2=20.43	1986ECa (55945)	931

B4=26.48

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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$B(\text{HgHL}) = 23.6$
 $B(\text{HgH}_2\text{L}_2) = 42.1$
 $B(\text{HgH}_2\text{L}(\text{EDTA})) = 44.0$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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$$\begin{aligned} K(\text{Hg} + \text{HL}) &= 16.71 \\ K(\text{Hg} + 2\text{HL}) &= 11.69 \end{aligned}$$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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$$B(Hg(OH)L) = 22.35$$

C7H10N2O2S HL (560)
2-(Methanesulfonamidomethyl)pyridine; C5H4N.CH2S(=O)2NHCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	diox/w	30°C	45%	U			K1=9.51 B2=17.61	1982MYb (56685)	938
In 45% v/v dioxan/H2O, 0.01 M KNO3 K1=9.67, B2=17.90										

C7H11NO2		HL						CAS 54162-90-2	(6019)	
2-Aminocyclohexene(4)-1-carboxylic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.50M	C			K1=15.901 B2=22.342	1986GGa (56768)	939
B(HgH-1L)=9.67										
B(HgH-1L2)=13.39										
cis isomer. For trans isomer K1=25.184, B2=31.625, B(HgH-1L)=18.93,										
B(HgH-1L2)=22.65										

C7H11NO6		H3L						(2926)		
2-Aminobutanoic-N-propane-1,3-dioic acid; HOOCH(C2H5)NH.CH(COOH)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	KNO3	25°C	0.10M	U			K1=13.03	1985KKb (56845)	940

C7H11N3O		HL						CAS 18259-63-7	(2265)	
N,N-Dimethyl-1-methyl-4-aminopyrimidin-2-one;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	nmr	non-aq	25°C	100%	U	M			1980MCb (56963)	941
K(HgCl2+L)=0.11										
Medium: DMSO-d6										

C7H11N3O2		L						CAS 7389-87-9	(3162)	
Histidine methyl ester										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	oth/un	25°C	0.10M	U			K1=5.33 B2=9.47	1971HMc (57003)	942

C7H12N2O3		HL		Gly-Pro				CAS 704-15-4	(257)	
Glycyl-proline; H2N.CH2.CO.NC4H7.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.20M	U			K1=3.13 B2=5.28	1982RRd (57123)	943

C7H12N2O3		HL		Pro-Gly				CAS 2578-97-6	(262)	
Prolyl-glycine; C4H8N.CO.NH.CH2.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

 Hg++ gl KCl 20°C 0.20M U K1=3.28 1982RRd (57149) 944

C7H13NO2 HL CAS 5691-19-0 (4449)
 2-Aminocyclohexanecarboxylic acid; H2N.C6H10.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	NaCl04	25°C	0.50M	C			K1=11.457 B2=18.93 B(HgH-1L)=6.89 B(HgH-1L2)=13.45	1986GGa (57445)	945
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cis isomer. For trans isomer K1=11.291, B2=18.656, B(HgH-1L)=6.61,
 B(HgH-1L2)=9.30

C7H13NO3 HL (7175)
 3,3'-Dimethylglutaramide; HOOCH2C(CH3)2CH2CONH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KN03	25°C	0.10M	U			K1=1.61 B2=8.61	1995MWb (57473)	946
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C7H13NO3S H2L CAS 59-53-0 (1269)
 N-Acetyl-penicillamine; CH3.CO.NH.CH(COOH)C(CH3)2SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	EMF	NaCl04	25°C	0.10M	U			K1=35.5 B2=41.90 K(Hg+2HL)=34.6	1981BCc (57490)	947
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Hg++	ISE	NaCl04	25°C	0.10M	U			K1=35.41 B2=41.61	1980CJa (57491)	948
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C7H13NO4S H2L (3184)
 N-(2-Methylthioethyl)iminodiethanoic acid; CH3.S.CH2.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KCl	20°C	0.10M	U			K1=8.01 B2=13.83 K(Hg(OH)L+HL)=9.82	1955SAa (57547)	949
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C7H13NO5 H2L CAS 62117-07-1 (3171)
 N-(2-Methoxyethyl)iminodiethanoic acid; CH3.O.CH2.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KCl	20°C	0.10M	U			K1=5.94 B2=10.03 K(HgLOH+H)=9.62 K(HgL(OH)2+H)=-10.75 K(HgL2OH+H)=10.35	1955SAa (57575)	950
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C7H13NO6 H2L CAS 32013-58-4 (6079)

N-(2,3-Dihydroxypropyl)iminodiethanoic acid; HO.CH2.CH(OH).CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U			B2=12.57	1980MRc (57613)	951

C7H14O5S2		HL							CAS 6791-11-3 (4456)	
Hexylxanthogenic acid; CH3(CH2)5.O.CSSH (hexoxydithioformic acid);										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	oth/un	25°C	var	U	M		B(HgL(CN)2)=35.54	1970AFa (57850)	952

C7H16S		L							CAS 26158-99-6 (5696)	
Pentyl-ethylsulfide; C2H5.S.C5H11										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	non-aq	25°C	100%	U			K1=7.96 B2=13.40 B3=16.04 B4=17.11	1986MMb (58095)	953

Medium: acetone, Bu4NC104

C7H18N2		L							CAS 110-95-2 (2277)	
N,N,N',N'-Tetramethyl-1,3-diaminopropane; (CH3)3N.CH2.CH2.CH2.N(CH3)3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H		K(HgCl2+L)=4.48 K(HgBr2+L)=4.85 K(HgI2+L)=5.00	1974FGa (58243)	954

In benzene. For HgCl2, DH=-80.0 kJ mol-1; DS=-179 J K-1 mol-1. For HgBr2, DH=-84.8; DS=-184. For HgI2, DH=-83.6; DS=-180.

C7H19N3		L							CAS 124-20-9 (13)	
1,5,10-Triazadecane, 4-Azaoctane-1,8-diamine; H2N.(CH2)3.NH.(CH2)4.NH2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	C	M		B(HgAL)=23.32 K(HgA+L)=11.52 B(HgAHL)=31.47 K(HgA+HL)=8.81	2004LBa (58308)	955

B(HgAH2L)=38.38, K(HgA+H2L)=5.80, B(HgH-1AL)=12.65.

H2A is cytidine-5'-monophosphoric acid.

Hg++	gl	KNO3	20°C	0.10M	C	M			2003LBa (58309)	956
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B(HgAL)=21.62
 K(HgA+L)=14.91
 K(HgA+L=HgAL(OH)+L)=11.40
 B(HgAHL)=29.12

A is cytidine. K(HgA+HL)=11.55.

Hg++	gl	KNO3	20°C	0.10M	U	K1=18.20	1999LBa (58310)	957
						B(HgHL)=25.50		

 C7H20N4 L CAS 4741-99-5 (12)
 1,4,8,11-Tetraazaundecane; H2N.CH2.CH2.NH.CH2.CH2.CH2.NH.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	vlt	NaClO4	25°C	0.20M	U	H	K1=22.1	1976KKb (58357)	958
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DH=-114.6 kJ mol⁻¹, DS=38.1

 C8H6N2S L CAS 53911-41-4 (3815)
 4-(2'-Pyridyl)-1,3-thiazole;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	ISE	oth/un	25°C	0.10M	U		K1=8.73	B2=14.97	1968EHa (58803)	959
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 C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)
 Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	NaClO4	25°C	0.10M	U	TI	K1=2.41	1985GMc (58973)	960
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Data for 0.02-0.10 M NaClO4 and for 25-45 C.
 At I=0.0 M, K1=3.40

 C8H7N3 L CAS 18653-75-3 (3792)
 2-(2'-Pyridyl)imidazole;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	EMF	KNO3	25°C	0.10M	U		K1=10.07	B2=18.28	1967EHc (59184)	961
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 C8H8NO2F3S HL CAS 50790-31-3 (211)
 Trifluoromethanesulfonamidomethylbenzene; C6H5.CH2.S(:O)2.NH.CF3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	diox/w	30°C	45%	M		K1=4.4(5)	B2=7.9(2)	1984MYa (59290)	962
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 C8H8O4 L (601)
 4,5-Dimethoxy-1,2-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++		nmr non-aq	34°C	100%	U	M		1981KKc (60112)	963
							K(HgCl ₂ +L)=0.63		

Medium: nitromethane

C8H9NO2	HL	CAS 4410-31-5	(4513)
2-Hydroxy-2-phenylacetamide; C6H5.CH(OH).CO.NH2			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U	M		1970GSb (60182)	964
							K(HgCl ₂ +L=HgClL+Cl)=5.41		
							K(HgClL+L=HgL2+Cl)=5.05		
							K(HgClL+OH=HgLOH+Cl)=4.93		

C8H9NO3	HL	CAS 2292-53-7	(8860)
Mandelohydroxamic acid;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U		K1=5.46 B2=10.82	1989SMc (60445)	965

C8H9N2O2F3S	HL	CAS 58157-03-2	(212)
2-(Trifluoromethanesulfonamidoethyl)pyridine; C5H4NCH2CH2S(:O)2NHCF3			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	diox/w	30°C	45%	M		K1=6.7(1) B2=8.5(0)	1984MYa (60531)	966

C8H10	L	o-Xylene	CAS 95-47-6	(3072)
1,2-Dimethylbenzene, 2-Xylene; CH3.C6H4.CH3				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	U	M		1974VPb (60676)	967
							K(L+HgCl ₂)=-0.485		
							K(L+HgBr ₂)=-0.376		

Medium: dichloromethane

C8H10N2O	HL	Mandelamidine	CAS 700-63-0	(3825)
2-Hydroxy-2-phenylacetamide; C6H5.CH(OH).C(:NH)NH2				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U	M		1970GSb (60716)	968
							K(HgCl ₂ +HL=HgClHL+Cl)=3.86		
							K(HgClHL+HL=Hg(HL)2+Cl)=3.96		
							K(HgClHL+OH=Hg(OH)HL+Cl)=4.42		

C8H10N2OS HL (4577)

N-Methylaminothioformyl-N-phenylhydroxylamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	oth	NaCl04	30°C	0.10M	U			K1=11.30 B2=22.20	1972MBe (60727)	969

C8H10N2O3S HL CAS 144-80-9 (8160)

4-Aminobenzenesulfonylacetamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	1.0M	U			K1=8.22	1976BMg (60749)	970

C8H10N2S L CAS 538-28-3 (2599)

2-Benzyl-2-thiopseudourea; C6H5.CH2.S.C(:NH)(NH2)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE mixed		25°C	82%	U			K1=18.89 B2=21.82 B3=24.53	1979MTd (60768)	971

Medium: 82% DMF/H2O

C8H10N2S L (2598)

2-Tolylthiocarbamide; CH3.C6H4.NH.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE mixed		25°C	82%	U			B4=27.30	1979MTd (60774)	972

Medium: 82% DMF/H2O

C8H10O2 L CAS 589-29-7 (3801)

1,4-Dimethoxybenzene (4-methoxyanisole); C6H4(OCH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	NaCl04	26°C	1.0M	U			K(Hg+L=HgH-1L+H)=2.7	1964WDa (60831)	973

C8H10O9 H4L CAS 137172-86-2 (6612)

SS-Oxydisuccinic acid; O(CH(COOH)CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	C			K1=13.82 K(HgL+H)=5.89 K(HgHL+H)=4.62 K(HgH2L+H)=3.16 K(Hg+HL)=13.75	1992MMa (60904)	974

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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$$K(\text{Hg}+\text{H}_2\text{L})=13.97, \quad K(\text{Hg}+\text{H}_3\text{L})=13.72$$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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$$K(HgH-3L+H)=9.88$$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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In benzene. $\Delta H(\text{HgCl}_2\text{L}) = -75.1 \text{ kJ mol}^{-1}$; $\Delta H(\text{HgBr}_2\text{L}) = -66.7$; $\Delta H(\text{HgI}_2\text{L}) = -55.0$, $\Delta S = -113$.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Hg++ cal non-aq 30°C 100% U H 1976GGa (61324) 980
HgI2+L in benzene. DH(K1)=-92; DH(K2)=-36 kJ mol-1

C8H12N2O2 HL Pyridoxamine CAS 85-87-0 (1175)

4-(Aminomethyl)-5-hydroxy-6-methyl-3-pyridinemethanol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.50M U K1=7.07 B2=12.25 1976EEa (61421) 981

C8H12N2O8 H4L CAS 35039-85-1 (4537)

1,2-Diaminoethane-N,N'-dimalononic acid; (HOOC)2.CH.NH.CH2.CH2.NH.CH(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=18.64 1972GBd (61507) 982

K(Hg+HL)=14.08

B(HgL(OH))=25.41

C8H13NO6 H3L (5681)

2-Aminobutanoic-N,N-diethanoic acid; CH3CH2CH(COOH)N(CH2COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=14.38 1985KKb (61788) 983

K(Hg+HL)=7.04

C8H14N2O3 HL Pro-Ala CAS 6422-36-2 (263)

Prolyl-alanine; C4H8N.CO.NH.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 20°C 0.20M U K1=3.19 B2=6.11 1982RRd (61928) 984

C8H14O4S3 H2L (2526)

3,6,9-Trithiaundecanedioic acid; HOOC.CH2.S.C2H4.S.C2H4.S.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.50M U B2=19.05 1980NAc (62124) 985

Hg++ gl NaClO4 25°C 0.50M C 1979NPb (62125) 986

K(HgL2+H)=3.76

K(HgHL2+H)=1.58

C8H14O6S2 H2L (1227)

3,6-Dithia-1,8-octanedioic acid; (HO.C2H4.S.CH(COOH))2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.10M U K1=4.67 1978MJa (62138) 987

C8H16N2O3 HL Gly-Leu CAS 869-19-2 (255)
Glycyl-leucine; H2N.CH2.CO.NH.CH(CH2.CH(CH3)2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 20°C 0.20M U K1=2.77 B2=4.61 1982RRd (62388) 988

C8H16N2O3 HL Leu-Gly CAS 686-50-0 (1248)
Leucyl-glycine; H2N.CH(CH2.CH(CH3)2).CO.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 20°C 0.20M U K1=2.31 1982RRd (62433) 989

C8H16N2O4S2 H2L (1226)
3,6-Dithiaoctanediamine-4,5-dicarboxylic acid; (H2N.C2H4.S.CH(COOH))2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.10M U K1=17.53 B2=28.66 1978MJa (62558) 990

C8H16N10 L (7005)
N,N'-Di(2-(5-tetraazolyl)ethyl)-1,2-diaminoethane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 0.10M U K1=12.50 1981ESa (62614) 991

C8H18N2OS L [12]aneN2OS CAS 124775-44-6 (7839)
1-Oxa-7-thia-4,10-diazacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=10.5 1999AMa (62822) 992
Medium: 0.10 M Et4NClO4.

C8H19N L CAS 111-92-2 (849)
Dibutylamine, 5-azanonane; (C4H9)2NH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE non-aq 25°C 100% U H K1=4.46 B2=10.08 1989PSa (63023) 993
B3=11.6

In DMSO; ionic medium: 0.1M Et4NClO4

C8H19N3S L CAS 87071-53-2 (719)
1-Thia-4,7,10-triazacyclododecane; cyclo(-S.(C2H4.NH)3.C2H4-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.20M	C		K1=24.32	1984KKa (63146)	994

C8H19O2PS2 HL CAS 2253-44-3 (2060)
 O,O'-Dibutyl dithiophosphoric acid; (C4H9O)2P(S)SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	M		1971DGb (63156)	995

K(HgL2+py) = 0.79

Medium: benzene

Hg++	EMF	alc/w	25°C	40%	U		B2=29.40 B3=30.20 B4=32.0	1970TCa (63157)	996
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Medium: 90% EtOH, 0.3 M KNO3

C8H19O2PS2 HL CAS 2253-52-3 (4584)
 O,O-Di-isobutyl phosphorodithioic acid; ((CH3)2.CH.CH2O)2P(S)SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	M		1971DGb (63167)	997

K(HgL+py)=0.84

Medium : benzene

Hg++	EMF	alc/w	25°C	90%	U		B2=29.70 B3=30.20 B4=32.0	1970TCa (63168)	998
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Medium: 90% EtOH, 0.3 M KNO3

C8H19PS2 HL CAS 32435-51-5 (4552)
 Di-n-butyl phosphinedithioic acid; (C4H9)2PSSH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	40%	U		B2=33.6	1970TCa (63207)	999

Medium: 40% EtOH, 0.3 M KNO3

C8H20N2 L CAS 111-51-3 (2278)
 N,N,N',N'-Tetramethyl-1,4-diaminobutane; (CH3)2N.CH2.CH2.CH2.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H		1974FGa (63218)	1000

K(HgCl2+L)=5.30
 K(HgBr2+L)=5.30
 K(HgI2+L)=4.30

In benzene. For HgCl2, DH=-85.3 kJ mol-1; DS=-179 J K-1 mol-1. For HgBr2,

DH=-74.3; DS=-143. For HgI2, DH=-78.7; DS=-177.

C8H20N4 L Cyclen CAS 294-90-6 (10)
1,4,7,10-Tetraazacyclododecane; cyclo(-(NH.CH2.CH2.)4-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt NaClO4 25°C 0.20M U H K1=25.5 1976KKb (63292)1001
DH=-98.7 kJ mol⁻¹, DS=157.7

C8H22N4 L CAS 35513-90-7 (1545)
1,4,9,12-Tetraazadodecane; NH2.(CH2)2.NH.(CH2)4.NH.(CH2)2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 1.00M C H K1=23.28 1982ABc (63382)1002
By calorimetry: DH1=-109.6 kJ mol⁻¹, DS1=77.8

C8H23N5 L Tetren CAS 112-57-2 (715)
1,4,7,10,13-Pentaazatridecane (Tetraethylenepentamine);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt NaClO4 25°C 0.20M M H K1=24.8 1978KKb (63472)1003
DH1=-139.7 kJ mol⁻¹

Hg++ gl KNO3 25°C 0.10M U K1=27.7 1958RHa (63473)1004

C9H6N3OClS HL CAS 27004-41-7 (216)
2-(2'-Thiazolylazo)-4-chlorophenol; C3H2NS.N:N.C6H3(Cl).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp diox/w ? 2% U K1=6.21 B2=12.31 1969KAb (63925)1005
Medium: 2% dioxan, 0.1 M KNO3

C9H9NO3 HL Hippuric acid CAS 495-69-2 (1184)
Benzoylaminoethanoic acid, N-benzoylglycine; C6H5.CO.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U M K1=3.01 B2= 5.73 2001Sma (65056)1006
B(Hg(bpy)L)=17.59
B(Hg(bpy)H-1L)=11.66

C9H11NO2 HL Phenylalanine CAS 63-91-2 (2)
2-Amino-3-phenylpropanoic acid; H2N.CH(CH2.C6H5)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=12.4 B2=19.60 1974VBa (65943)1007

Hg++ vlt KNO3 25°C 0.60M U B2=18.06 1966TAb (65944)1008

Hg++ gl oth/un 20°C .005M U B2=18.7 1953PEa (65945)1009
Medium: 0.005 Hg(NO3)2

C9H11NO2 HL CAS 6052-10-4 (4647)
2-Hydroxy-2-phenylpropionamide; CH3.C(C6H5)(OH)CO.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.10M U 1970GSb (65992)1010
K(HgCl2+L=HgClL+Cl)=5.60
K(HgClL+L=HgL2+Cl)=5.45
K(HgClL+OH=HgLOH+Cl)=4.71

C9H11NO3 H2L Tyrosine CAS 60-18-4 (4)
2-Amino-3-(4-hydroxyphenyl)propanoic acid; HO.C6H4.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt oth/un 25°C 0.10M C 2002MRa (66225)1011
Beff(Hg2L)=2.80

Method: oxidation of tyrosine at a Pt electrode. Medium: phosphate buffer,
pH 7.0. Complex is adsorbed on electrode.

Hg++ gl NaNO3 25°C 0.10M U K1=12.3 B2=19.50 1974VBa (66226)1012

Hg++ gl oth/un 20°C .002M U B2=17.1 1953PEa (66227)1013
Medium: 0.002 Hg(NO3)2

C9H11NO3 HL Phenylserine CAS 2180-37-2 (2546)
2-Amino-3-hydroxy-3-phenylpropanoic acid; C6H5.CH(OH).CH(NH2)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 17°C .005M U B2=17.3 1953PEa (66259)1014
Medium: 0.005 Hg(NO3)2

C9H11NO4S H2L CAS 1080-44-0 (4682)
N-(4-Toluenesulfonyl)glycine, N-tosylglycine; CH3.C6H4.SO2.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U M K1=2.97 B2= 5.28 2001SMa (66421)1015
K(Hg+2H-1L)=16.91
B(Hg+bpy+H-1L)=19.08

Also data for 4-nitrophenylsulfonylglycine complexes.

Hg++ gl diox/w 30°C 45% U K1=13.15 1984MYa (66422)1016

Hg++ vlt oth/un 25°C 0.10M U 1968RFa (66423)1017

B4=15.46

C9H11N04S2 H3L CAS 97512-83-9 (1330)

N-Benzenesulfonyl-L-cysteine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl diox/w 30°C 50% M 1980MDc (66443)1018

*K(HgH2L2)=-10.49

*K(HgHL2)=-11.55

Medium: 50% v/v dioxane/H2O, 0.50 M NaClO4.

Hg++ gl alc/w 30°C 50% M 1978GMf (66444)1019

*K(Hg(H2L)2)=-3.13

*K(Hg(HL)H2L)=-4.19

*K(Hg(HL)2)=-10.49

*K(HgHL2)=-11.54

Medium: 50% EtOH/H2O, 0.50 M NaClO4.

C9H12 L Mesitylene CAS 108-67-8 (3242)

1,3,5-Trimethylbenzene; C6H3(CH3)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp non-aq 25°C 100% U M 1974VPb (66541)1020

K(L+HgCl2)=-0.368

K(L+HgBr2)=-0.269

Medium: dichloromethane

C9H12N2O HL Atrolactamidine CAS 27906-16-1 (3878)

2-Hydroxy-2-phenylpropanoylamidine; C6H5.C(OH)(CH3)C(:NH)NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.10M U 1970GSb (66559)1021

K(HgCl2+HL=HgClHL+HL)=4.0

K(HgClHL+HL=Hg(HL)2+Cl)=3.93

K(HgClHL+OH=Hg(OH)HL+Cl)=4.65

C9H12N2O10 H5L CAS 80921-06-8 (2924)

2,3-Diaminopropanoic-N,N'-di-1,3-propanedioic acid;

(HOOC)2CH.NH.CH(COOH).CH2.NH.CH(COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=17.64 1983KBd (66736)1022

B(Hg(OH)L)=24.27

Method: Hg-electrode

Hg++ ISE KNO3 25°C 0.10M U K1=17.64 1982KBb (66737)1023

C9H13N08 H4L (7012)
1,3-Dicarboxypropane-1-iminodiethanoic acid; HOOC.CH(N(CH2COOH)2)CH2CH2COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=14.33 1977GNb (66908)1024
Method: Hg-electrode

C9H13N304 HL CAS 3992-42-5 (2266)
Deoxycytidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ nmr non-aq 25°C 100% U M 1980MCb (66991)1025
K(HgCl2+L)=1.30

Medium: DMSO-d6

C9H13N305 L Cytidine CAS 65-46-3 (2152)
Cytidine, Cytosine-1-beta-D-ribofuranoside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M U K1=6.71 B2=11.89 1999LBa (67058)1026

Hg++ nmr non-aq 32°C 100% U 1980MCa (67059)1027
K(HgCl2+L)=1.3

Medium: DMSO-d6

Hg++ nmr non-aq 36°C 100% U K1=1.53 1970KLc (67060)1028

Medium: (CH3)2SO. method: nmr

C9H14N203 L (8153)
5-(2-Propyl)-5-ethyl-2,4,6-trihydroxypyrimidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 1.0M U K1=9.20 1976BMg (67124)1029

C9H14N209 H4L CAS 56360-11-3 (2576)
2-Hydroxy-1,3-diaminopropane-N,N'-di(1,3-propanedioic acid)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=19.32 1976GKb (67136)1030
B(Hg(OH)L)=26.45

C9H14N3O8P H2L CMP-5 CAS 63-37-6 (1243)
Cytidine-5'-monophosphoric acid, Cytidilic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U			K1=11.80	1999LBa (67253)	1031

C9H14N4O3 HL Carnosine CAS 305-84-0 (272)
3-Alanyl-histidine; H2N.CH2.CH2.CO.NH.CH(CH2.C3H3N2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.10M	U			K1=8.08 K(Hg+HL)=5.27	1964LMa (67318)	1032

C9H16N2O6 H2L CAS 24709-35-8 (3274)
N-(2-(2-Ethoxycarbonylamino)ethyl)iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.10M	U			K1=4.5 B2=8.0	1955SAa (67628)	1033

C9H18N2O3 HL Ala-Leu CAS 1999-42-4 (264)
Alanyl-leucine; H2N.CH(CH3).CO.NH.CH(CH2.CH(CH3)2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.20M	U			K1=2.79	1982RRd (67907)	1034

C9H19N2O4+ H2L (3277)
2-Di(carboxymethyl)aminoethyltrimethylammonium cation
+

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.10M	U			K1=2.77 B2=5.0	1955SAa (68002)	1035

C9H20N2O2 L 13-AneN202 CAS 60350-15-4 (5662)
1,4-Dioxa-7,11-diazacyclotridecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=13.30	1986TSa (68037)	1036

C9H21N3O L (2479)
1-Oxa-4,7,11-triazacyclotridecane; cyclo(-O.(CH2.CH2.NH)2.CH2.CH2.CH2.NH.CH2.CH2-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=18.05	1986TSa (68203)	1037

 C9H22N4 L CAS 295-14-7 (9)
 1,4,7,10-Tetraazacyclotridecane; cyclo(-(NH.CH2.CH2.)4.CH2-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	vlt	NaCl04	25°C	0.20M	U	H		K1=25.3	1976KKb (68248)	1038
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DH=-103.3 kJ mol⁻¹, DS=139.3

 C9H24N3O9P3 H6L NOTPH CAS 83843-39-3 (224)
 1,4,7-Triazacyclononane-N,N',N"-tris(methylenephosphonic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KCl	25°C	1.0M	U			K1=23.0 K(Hg+HL)=17.2	1984KMa (68321)	1039
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 C9H24N4 L CAS 129880-56-4 (1533)
 1,4,10,13-Tetraazatridecane; H2N.(CH2)2.NH.(CH2)5.NH.(CH2)2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	ISE	KNO3	25°C	1.00M	C	H		K1=22.2 B(HgH2L)=31.5	1982ABc (68335)	1040
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By calorimetry: DH1=-98.7 kJ mol⁻¹, DS1=93.7

 C9H24N4 L CAS 4963-47-7 (546)
 Tris-(3-aminopropyl)amine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	NaNO3	20°C	0.10M	U			K1=20.49	1962TAb (68389)	1041
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 C10H6O3 HL CAS 83-72-7 (3294)
 2-Hydroxy-1,4-naphthoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	sp	alc/w	20°C	40%	U			K1=8.77 B2=16.75 K(Hg+HL)=4.805 B(Hg(OH)2L)=32.37	1990SIb (68461)	1042
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Medium: 40% v/v EtOH/H2O, 0.1 M NaCl04

 C10H7NO5S H2L CAS 3682-32-4 (1812)
 2-Nitroso-1-hydroxynaphthalene-4-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KNO3	35°C	0.10M	U			K1=4.32	1974LSa (68885)	1043
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C10H7N08S2 H3L Nitroso-R acid CAS 525-05-3 (1811)
 1-Nitroso-2-hydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaCl04	10°C	0.10M	U	H		K1=6.15 B2=8.87	1979GBf (69014)	1044

C10H7O2F3 HL CAS 326-06-7 (196)
 3-Benzoyl-1,1,1-trifluoroacetone; CF3.CO.CH2.CO.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	dis	NaCl04	25°C	0.1M	U			K1=8.65 B2=15.10	1981MIc (69150)	1045

Distribution into CCl4 with trioctylphosphate

C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)
 2,2'-Bipyridine; (C5H4N)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	oth/un	25°C	0.10M	C			K1=8.30 B2=16.33 B3=19.04 B(HgLOH)=18.15 B(HgL2OH)=20.65	1987GGc (69581)	1046

Medium: KSO3CH3

Hg++	cal	non-aq	30°C	100%	U	M			1976AGa (69582)	1047
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K(HgI2+L)=ca.4

Medium: MeCN

Hg++	cal	non-aq	30°C	100%	U	H			1976FGa (69583)	1048
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K(Hg(SCN)2+L)=4.30

Medium: MeCN. DH=-40.7 kJ mol-1; DS=-52

Hg++	cal	non-aq	30°C	100%	U	H			1974DGA (69584)	1049
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K(HgA2+L) < 0.0

In benzene. HA=thiobenzoyl-1,1,1-trifluoroacetone; DH=<-5 kJ mol-1

Hg++	cal	non-aq	30°C	100%	U	H			1974FGa (69585)	1050
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K(HgCl2+L) > 4
 K(HgBr2+L) > 4
 K(HgI2+L)=3.63

In benzene. For HgCl2, DH=-32.8 kJ mol-1, DS=<-32. For HgBr2, DH=-40.1, DS=<-56. For HgI2, DH=-48.8, DS=-91.

Hg++	ISE	NaNO3	20°C	0.10M	U			K1=9.64 B2=16.74 K3=2.8	1963ANG (69586)	1051
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C10H9N08 H2L CAS 83785-11-9 (685)
 2-Nitro-1,4-di(carboxymethoxy)benzene; O2N.C6H3.(OCH2COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	none	30°C	0.0	U		K1=3.93	1985TZa (70235)	1052
Hg++	gl	oth/un	30°C	?	U		K1=4.30 B2=8.24	1985TZa (70236)	1053

C10H9NS		L					CAS 46000-25-3	(4721)	
1,4-Dihydro-1-methyl-4-thioquinoline;									
Hg++	kin	oth/un	20°C	0.10M	U		K(CH3Hg+L)=9.70	1973GEb (70240)	1054

C10H9N3OS		HL					CAS 1823-44-5	(4780)	
2-(2'-Thiazolylazo)-4-methylphenol; CH3.C6H3(OH).N:N.C3H3NS									
Hg++	gl	diox/w	25°C	10%	U		K1=6.11 B2=12.22	1969KAa (70348)	1055

C10H9N3OS		HL					CAS 60321-26-8	(4671)	
2-(2-Thiazolylazo)methylphenol; C3H2NS.N:N.C6H3(CH3)OH									
Hg++	sp	diox/w	25°C	10%	U T		K1=16.07	1973KSd (70361)	1056
Medium: 10% dioxan, 0.1 M KNO3. 15 C: K1=16.10; 35 C: K1=16.03									

C10H9N3O2S		HL					CAS 3012-52-0	(217)	
2-(2'-Thiazolylazo)-4-methoxyphenol; CH3O.C6H3(OH).N:N.C3H2N2									
Hg++	sp	NaClO4	?	0.01M	U I		K(Hg+2HL)=8.56	1972LHa (70400)	1057
Medium: 0.01-2 M HClO4. In 50% EtOH: K1=11.05, B2=20.4									

C10H10O2		HL					Benzoylacetone CAS 93-91-4	(197)	
1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3									
Hg++	dis	NaClO4	25°C	0.1M	U		K1=12.08 B2=23.69	1981MIc (70734)	1058
Distribution into CCl4 with trioctylphosphate									

C10H11NO3S		H2L					Benzoylcysteine CAS 60199-84-0	(2580)	
N-Benzoyl-2-amino-3-mercaptopropionic acid; C6H5.CO.NHCH(COOH)CH2SH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.15M	U			1979ZNa (70955)	1059
							K(Hg+2HL)=37.45 K(HgHL2+H)=7.39 K(HgL2+H)=8.17		

Hg++	gl	oth/un	25°C	1.00M	U		B2=41.33	1977NZa (70956)	1060
Medium: KI									

C10H11NO4		H2L					CAS 1137-73-1	(2567)	
N-Phenyliminodiethanoic acid; C6H5.N(CH2.COOH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	KNO3	20°C	0.10M	U		B2=12.9	1964PIa (71003)	1061

C10H1104As		H2L					CAS 51525-18-9	(3907)	
As-Phenylarsinodiethanoic acid; C6H5.As(CH2.COOH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	KNO3	20°C	0.10M	U		K1=14.7 B2=19.92	1964PIa (71130)	1062

C10H1104P		H2L					CAS 58942-13-5	(7014)	
Phenylphosphino-P,P-diethanoic acid, Diphenylphosphinediethanoic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	NaClO4	25°C	0.10M	U	I	K1=19.7 B2=31.92 B3=34.21	1979PPc (71138)	1063
Method: Hg electrode. In 50% v/v dioxan/H2O: K1=22.2; B2=33.27; B3=35.14									

Hg++	gl	NaClO4	25°C	0.10M	C	I	K1=19.7 B2=31.92 B3=34.21	1979PPd (71139)	1064
Additional methods: polarography and Hg electrode. In 50% dioxane/H2O: K1=22.2, B2=33.27, B3=35.14.									

C10H12N2O4		H2L					CAS 16598-05-3	(967)	
2-Pyridylmethyliminodiethanoic acid; C5H4N.CH2.N(CH2.COOH)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	NaNO3	20°C	0.10M	C	H	K1=14.90 B2=24.10	1981ANb (71261)	1065
DH(K1)=-50.0 kJ mol ⁻¹ , DH(K2)=-48.5 kJ mol ⁻¹									

C10H12N4O4		H2L					Thymine-thymine CAS 3660-32-0	(8529)	
Hexahydrodimethyl-Cyclobuta[1,2:4,3]dipyrimidine-2,4,5,7-tetrone, cis-syn Thymine dimer;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.10M	U			K1=13.55 B(HgH-1L)=6.73 B(HgH-2L)=-1.69	1984PTa (71325)	1066

Additional method: Hg electrode.

 C10H13N04S H2L N-Tosylalanine (1584)
 N-(4-Toluenesulfonyl)-3-aminopropanoic acid; CH3.C6H4.SO2.NH.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U	M		K1=3.20 B2= 5.99 K(Hg+2H-1L)=16.42 B(Hg+bpy+H-1L)=18.88	2001SMa (71771)	1067

 C10H13N5O4 L Adenosine CAS 58-61-7 (2154)
 Adenosine, Adenine-9-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	nmr	non-aq	25°C	100%	U	M		K(HgCl2+L)=0.079	1980MCb (71942)	1068

Medium: DMSO-d6

Hg++	nmr	non-aq	36°C	100%	U			K1=0.86	1970KLc (71943)	1069
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Medium: (CH3)2SO

Hg++	ISE	NaClO4	27°C	0.10M	U			B2=8.50	1961FBa (71944)	1070
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 C10H13N5O5 HL Guanosine CAS 118-00-3 (1402)
 2-Aminopurin-6-one-9-riboside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	nmr	non-aq	25°C	100%	U	M		K(HgCl2+L)=0.23	1980MCb (72010)	1071

Medium: DMSO-d6

Hg++	nmr	non-aq	21°C	100%	U			K(Hg+HL)=0.44	1973SFa (72011)	1072
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Medium: (CH3)2SO

Hg++	nmr	non-aq	36°C	100%	U			K(Hg+HL)=0.77	1970KLc (72012)	1073
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Medium: (CH3)2SO

 C10H14N2O6 L alpha-Thymidine CAS 4449-43-8 (695)
 Thymine-2-desoxyribofuranosyl-5-methyluracil;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U		K1=7.88 B2=14.94	1987MPa (72103)	1074
Hg++	ISE	NaClO4	27°C	0.10M	U		B2=21.2	1961FBa (72104)	1075

C10H14N5O7P H2L AMP-5 CAS 18422-05-4 (842)									
Adenosine-5'-monophosphoric acid, 5-Adenylic acid;									
Hg++	gl	KNO3	20°C	0.10M	U		K1=12.57 B(HgHL)=17.68	1999Lba (72454)	1076

C10H15NO L Ephedrine CAS 299-42-3 (1836)									
(1-Methylaminoethyl)benzyl alcohol; C6H5.CH(OH)CH(CH3)NHCH3									
Hg++	gl	KCl	25°C	.058M	U T		K1=4.24 B2=8.03	1961SMa (72645)	1077
At 0 C: K1=4.50, B2=8.54; 45 C: K1=4.04, B2=7.68									

C10H15N3O4 HL CAS 838-07-3 (2267)									
5-Methyldeoxycytidine;									
Hg++	nmr	non-aq	25°C	100%	U M		K(HgCl2+L)=1.3	1980MCb (72711)	1078
Medium: DMSO-d6									

C10H16N2O8 H4L EDDS CAS 52759-67-8 (1100)									
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH2.NH.CH(COOH)CH2.COOH)2									
Hg++	gl	KNO3	25°C	0.10M	U		K1=14.40 K(Hg+HL)=8.44 B(Hg(OH)L)=23.58	1993VZa (73140)	1079
Hg++	ISE	KNO3	25°C	0.10M	U		K1=17.50 K(Hg+HL)=12.49 K(Hg+OH+L)=24.69	1972Gsi (73141)	1080

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)									
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestic acid;									
Hg++	vlt	KNO3	25°C	0.10M	U		K1=19.98	1986ECa (73834)	1081

Hg++	EMF	NaNO ₃	29°C	1.0M	C	H		1986IKb (73835)1082
							K(Hg+H ₄ L=HgL+4H)=6.8	
Method: Hg/Hg++ electrode. Medium: pH 2.2. DS(K ₁)=-25 J K ⁻¹ mol ⁻¹ .								
Hg++	gl	KNO ₃	25°C	0.10M	C	T	H	1985HWc (73836)1083
Data for 5-35 C. Method: Hg and glass electrodes.								
DH(K ₁)=-54.8 kJ mol ⁻¹ , DS(K ₁)=229 J K ⁻¹ mol ⁻¹ .								
Hg++	sol	KNO ₃	25°C	1.00M	U			1979JPb (73837)1084
							K(HgL+H)=3.27	
							K(HgHL+H)=2.21	
Hg++	vlt	KNO ₃	20°C	0.10M	U		K ₁ =21.80	1978NLb (73838)1085
Hg++	ISE	NaClO ₄	25°C	0.50M	U		K ₁ =20.80	1977GGb (73839)1086
Hg++	sp	NaClO ₄	25°C	0.10M	U	M	K ₁ =21.75	1975TTa (73840)1087
							K(HgL+p-anisidine)=3.40	
							K(HgL+p-toluidine)=3.11	
							K(HgL+p-chloroaniline)=2.34	
							K(HgL+p-nitroaniline)=1.18	
K(HgL+aniline)=2.98, K(HgL+m-toluidine)=2.73, K(HgL+m-chloroaniline)=2.44								
K(HgL+m-nitroaniline)=2.29, K(HgL+p-iodoaniline)=2.74)								
Hg++	vlt	KNO ₃	20°C	0.10M	U		K ₁ =21.78	1970SNa (73841)1088
Hg++	gl	KNO ₃	25°C	0.10M	U		K ₁ =22.02	1969BNa (73842)1089
							K(HgL+H)=3.07	
							K(Hg+HL)=14.56	
Hg++	ISE	KNO ₃	20°C	0.10M	U	T	H	1966Mca (73843)1090
							K(HgL+H)=3.19	
K ₁ =21.44(30 C), 21.23(40 C). At 25 C: DH(K ₁)=-37.6 kJ mol ⁻¹ , DS=280 J K ⁻¹ m ⁻¹								
Hg++	cal	KNO ₃	25°C	0.10M	U	H		1965WHa (73844)1091
DH(K ₁)=-80.2 kJ mol ⁻¹ , DS=155 J K ⁻¹ mol ⁻¹								
Hg++	gl	KNO ₃	20°C	0.10M	U		K ₁ =21.8	1964ANa (73845)1092
							K(Hg+HL)=14.6	
Hg++	cal	KNO ₃	20°C	0.10M	U	H		1963ANf (73846)1093
DH(K ₁)=-79.1 kJ mol ⁻¹ , DS=151 J K ⁻¹ mol ⁻¹								
Hg++	oth	none	?	0.0	U			1959SRb (73847)1094
							K(HgL+NH ₃)=6.4	
							K(H+HgLOH)=8.9	
Method: vol								
Hg++	ISE	NaNO ₃	22°C	0.10M	U	T	K ₁ =21.78	1957SAb (73848)1095

Glutamyl-cysteinyl-glycine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	NaNO3	25°C	0.10M	M	M			1990SHa (75121)	1106
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K(HgL2+H)=9.89
K(HgL2+2H)=19.13
K(HgL2+3H)=23.43
K(HgL2+4H)=26.78

K(HgL2+5H)=29.26, K(HgL2+6H)=31.47; K(HgL2+Ni)=9.10, K(HgL2+H+Ni)=15.82;
K(HgL2+Co)=7.21, K(HgL2+H+Co)=14.48

Hg++	nmr	non-aq	25°C	100%	U				1988SCa (75122)	1107
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K(HgL2+L)=3.18

Hg++	vlt	KNO3	25°C	1.0M	U	T		B2=41.58	1953SKb (75123)	1108
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B2=43.47(12 C)

C10H18N2O7 H3L HEDTA CAS 150-39-0 (392)

N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	ISE	NaClO4	25°C	0.50M	U			K1=19.35	1977GGb (75412)	1109
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Hg++	gl	KNO3	25°C	0.10M	U			K1=19.97	1969BNa (75413)	1110
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2nd method: calorimetry

Hg++	ISE	KNO3	20°C	0.10M	U	T	H	K1=19.47	1966Mca (75414)	1111
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K(HgL+H)=2.57

K1=19.3(30 C),19.1(40 C). At 25 C: DH(K1)=-29.3 kJ mol⁻¹, DS=272 J K⁻¹ mol⁻¹

Hg++	cal	KNO3	25°C	0.10M	U	H			1965WHa (75415)	1112
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DH(K1)=-83.6 kJ mol⁻¹, DS=105 J K⁻¹ mol⁻¹

Hg++	ISE	KNO3	25°C	0.10M	U			K1=20.1	1960HRa (75416)	1113
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Hg++	oth	R4N.X	25°C	0.10M	U				1959SRb (75417)	1114
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K(HgL+NH3)=6.1

K(HgLOH+H)=8.4

C10H19N04 H2L (3328)

N-(3,3-Dimethylbutyl)iminodiethanoic acid; (CH3)3C.CH2.CH2.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KCl	20°C	0.10M	U			K1=5.91 B2=10.71	1955SAa (75639)	1115
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C10H20O5 L 15-Crown-5 CAS 33100-27-5 (576)

1,4,7,10,13-Pentaoxacyclopentadecane; cyclo(-(O.CH2.CH2)5-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	oth/un	25°C	0.10M	U	H T	K1=1.68	1976ITb (76005)	1116
DH=-15.0 kJ mol-1.									

C10H20S4		L		14-Ane-S4			CAS 24194-61-4	(175)	
1,4,8,11-Tetrathiacyclotetradecane; cyclo(-(S.CH2.CH2)2.CH2.(S.CH2.CH2)2.CH2-)									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	dis	non-aq	24°C	100%	U	M		1977SMa (76158)	1117
							K(HgLC104+C104)=3.18		
							K(HgL+C104)=3.87		

C10H21N04		L					CAS 66943-05-3	(5818)	
1-Aza-4,7,10,13-tetraoxacyclopentadecane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	alc/w	25°C	95%	U		K1=10.3	1993BDd (76184)	1118
Medium: 95% v/v MeOH/H2O, 0.1 M Et4NC104									

C10H21N11		L						(7006)	
1,7-Di(2-(5-tetraazolyl)ethyl)-1,4,7-triazaheptane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	0.10M	U		K1=20.25	1981ESa (76212)	1119

C10H22N203		L		Cryptand 2,1			CAS 31249-95-3	(835)	
4,7,13-Trioxa-1,10-diazacyclopentadecane (Trioxa(2,1)cryptand);									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	R4N.X	25°C	0.10M	U		K1=16.65	1982NSb (76318)	1120

C10H24N4		L		Cyclam			CAS 295-37-4	(8)	
1,4,8,11-Tetraazacyclotetradecane; cyclo(-(HN.CH2.CH2.NH.(CH2)3)2-)									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.50M	U		K1=23.0	1997BLd (76666)	1121

Hg++	sp	NaClO4	25°C	0.50M	U	M		1997GHb (76667)	1122
							Keff(Hg+L+I)=30.8 (0.1 M OH-)		
Data also for 3,6,10,13,16,19-Hexaazabicyclo[6.6.6]icosane and its 1,8-diami no derivative									

Hg++	vlt	NaClO4	25°C	0.20M	U	H	K1=23.0	1976KKb (76668)	1123

DH=-137.6 kJ mol⁻¹, DS=-20.5

C10H25N5 L 15-Ane-N5 CAS 295-64-7 (99)
1,4,7,10,13-Pentaazacyclopentadecane; cyclo(-(HN.CH2.CH2)5-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.20M M H K1=28.5 1978KKb (76734)1124
DH1=-136.8 kJ mol⁻¹

C10H26N2O12P4 H8L CAS 28698-30-8 (3342)
N,N,N',N'-Tetra(phosphomethyl)cyclohexane-1,2-diamine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 25°C 0.10M U K1=9.51 1959BYa (76759)1125

C10H26N4 L Spermine CAS 71-44-3 (291)
4,9-Diazadodecane-1,12-diamine; (H2N.CH2.CH2.CH2.NH.CH2.CH2.)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M U K1=18.40 1999Lba (76795)1126
B(HgHL)=26.56

C10H26N4S4 L CAS 55677-43-5 (1178)
1,1,2,2-Tetramercaptoethylamine-ethane; (CH(S.CH2.CH2.NH2)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.10M U K(Hg+H2L)=18.57 1976CJa (76818)1127

C10H28N2O12P4 H8L CAS 23605-74-5 (435)
(Hexamethylenedinitrilo)tetra(methylenephosphonic acid);
(CH2.CH2.CH2.N(CH2.PO3H2)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 0.10M U K1=9.51 1980ZRb (76839)1128
K(HgL+H)=7.30
K(HgHL+H)=6.51
K(HgH2L+H)=5.97
K(HgH3L+H)=5.46

C10H28N6 L PENTEN CAS 4097-90-9 (3315)
N,N,N',N'-Tetra-(2-aminoethyl)diaminoethane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaCl04 25°C 1.0M C K1=29.3 2001GLb (76874)1129
B(HgHL)=37.76
B(HgH2LC1)=43.7, B(HgH-1L)=18.5

Hg++ gl KCl 20°C 0.50M U K1=29.59 1953SMa (76875)1130
K(Hg+HL)=27.93
K(Hg+H2L)=23.7
K(HgL+H)=8.59

C11H8N6O7S2 H4L CAS 35322-95-7 (909)
3-Hydroxy-4-(1H-tetrazol-5-ylazo)-2,7-naphthalenedisulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp NaCl04 25°C 0.10M U 1983PEa (76938)1131
K(Hg+H2L=HgHL+H)=2.06
K(Hg+H2L=HgH-2L+4H)=-10.77

C11H8N6O8S2 H5L CAS 74385-48-1 (897)
2-(1H-Tetrazol-5-ylazo)chromotropic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp NaCl04 25°C 0.10M U 1983PEa (76951)1132
K(Hg+H3L=HgH2L+H)=0.22

C11H8O3 H2L CAS 86-48-6 (1129)
1-Hydroxy-2-naphthoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 30°C 0.10M U T H K2=0.89 1976SSb (77011)1133

C11H8O3 L CAS 18916-57-9 (581)
4-Methoxy-1,2-naphthoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp non-aq 34°C 100% U HM 1981KKb (77139)1134
K(HgCl2+L)=0.49

Medium: nitromethane

C11H12N2O2 HL Tryptophan CAS 73-22-3 (3)
2-Amino-3-(3-indolyl)propanoic acid; H2N.CH(CH2.C8H6N)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.10M U K1=11.7 B2=18.70 1974VBa (78209)1135

C11H13NO6S H3L CAS 20531-36-6 (4872)

N-Benzenesulfonyl-1-glutamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	EMF	none	30°C	0.0	U				1970GDb (78698)	1136
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K(HgL2+H)=5.68

K(HgHL2+H)=4.80

C11H14N2O

L

CAS 51036-80-7 (444)

1-(1-Ethoxyethyl)benzimidazole;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	kin	oth/un	80°C	0.20M	C				1980LKa (78771)	1137
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K(HgCL2+HL=HgCL2L+H)=4.05

Medium: 0.2 M HgCL2

C11H14N2O

L

(4854)

Methylglyoxal 4-dimethylaminoanil

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	sp	oth/un	?	?	U			K1=6.08	1969SMa (78776)	1138
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C11H14N2O4

H2L

(1880)

N-(6-Methyl-2-pyridylmethyl)iminodiethanoic acid; CH3C5H3NCH2N(CH2COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	ISE	NaNO3	20°C	0.10M	C	H		K1=14.90 B2=24.00	1981ANb (78884)	1139
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DH(K1)=-48.7 kJ mol⁻¹, DS=119 J K⁻¹ mol⁻¹; DH(K2)=-49.6, DS=5

C11H15NO3

L

(6281)

Benzaldehyde:tris-buffer Schiff's base; C6H5.CH:N.C(CH2.OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	alc/w	26°C	60%	U			K1=5.17 B2=8.78	1978TPb (79032)	1140
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C11H15NO4

HL

CAS 18212-81-2 (6280)

Salicylaldehyde:tris-buffer Schiffs base;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	alc/w	26°C	60%	U			K1=6.23	1978TPb (79044)	1141
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C11H15N5O5

HL

CAS 2140-65-0 (2184)

1-Methylguanosine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ nmr non-aq 25°C 100% U M 1980MCb (79074)1142
K(HgCl2+L)=0.48

Medium: DMSO-d6

C11H16N2O10 H5L CEDTA CAS 62394-58-5 (1080)

1-Carboxy-1,2-diaminoethane-N,N,N',N'-tetraethanoic acid;
(HOOCCH2)2NCH(COOH)CH2N(CH2COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=20.33 1980KBe (79109)1143
K(Hg+HL)=15.15
B(Hg(OH)L)=27.30

Method: Hg-electrode

C11H16N2S2 L CAS 771500-52-8 (9193)

2,8-Dithia-5-aza-2,6-pyridinophane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=10.68 2004BBe (79118)1144
K(HgL+OH)=3.24

Medium: 0.1 M Me4NO3

C11H16O8S4 H4L CAS 51865-18-0 (1138)

(Propanediyldenetetrathio)tetra-ethanoic acid; ((HOOC.CH.S)2.CH)2.CH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.10M U K1=16.61 1975JBa (79146)1145
The species has not been identified but it is thought to be Hg+H4L=HgH4L.

C11H18N2O3 L CAS 76-74-4 (8154)

5-(1-Methylbutyl)-5-ethyl-2,4,6-trihydroxypyrimidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 1.0M U K1=9.23 1976BMg (79203)1146

C11H18N2O8 H4L PDTA CAS 4408-81-5 (1655)

1,2-Diaminopropane-N,N,N',N'-tetraethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 20°C 0.10M U K1=20.20 1981NSc (79290)1147

Hg++ ISE KNO3 25°C 0.10M U K1=20.33 1980KBc (79291)1148

Hg++ vlt KNO3 20°C 0.10M U K1=22.52 1978NLb (79292)1149

Hg++ vlt KNO3 25°C 1.00M U 1977HDa (79293)1150
K1eff=18.82
Keff at pH 7

Hg++ ISE KNO3 20°C 0.10M U K1=22.81 1964ICb (79294)1151
K(Hg+HL)=3.12

C11H18N2O8 H4L CAS 38539-29-0 (2573)
1,3-Diaminopropane-N,N'-di(1,4-butanedioic acid)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	KNO3	25°C	0.10M	U			K1=17.09 K(Hg+HL)=12.78 K(Hg+H2L)=9.59 B(Hg(OH)L)=24.92	1976GKb (79365)	1152

C11H18N2O8 H4L CAS 4408-81-5 (923)
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((H00C.CH2)2N.CH2.)2.CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	1.0M	C	I		B(Hg2L2)=40.43	1999ANa (79448)	1153

Method: Glass electrode plus Hg electrode. At I = 0.1 M, B(Hg2L2)=42.1.

Hg++ gl KNO3 20°C 0.10M U H 1964ANa (79449)1154
K(Hg+HL)=13.46
By calorimetry: DH(K1)=-79.0 kJ mol-1, DS=111 J K-1 mol-1

Hg++ gl KNO3 20°C 0.10M U 1964LAa (79450)1155
K(HgL+H)=4.00
By ion-selective electrode: K1=19.92

Hg++ ISE NaNO3 20°C 0.10M C K1=19.70 1957SSa (79451)1156
K(Hg+HL)=12.53

C11H18N2O9 H4L HDPTA CAS 3148-72-9 (431)
1,3-Diamino-2-hydroxypropane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U			K1=18.35 B(HgHL)=13.34 K(Hg+HgL=Hg2L)=8.0	1975HAa (79558)	1157

Hg++ vlt KNO3 20°C 0.10M U K1=20.59 1962SKa (79559)1158

C11H18N2O9 H4L CAS 668-21-1 (2562)

2-Hydroxy-1,3-diaminopropane-N,N'-di(1,4-butanedioic) acid

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	KN03	25°C	0.10M	U			K1=17.73 K(Hg+HL)=12.90 B(Hg(OH)L)=25.10	1976GKb (79596)	1159

C11H20N2O3 HL Pro-Leu CAS 52899-07-7 (258)
 Prolyl-leucine; C4H8N.CO.NH.CH(CH2.CH(CH3)2).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	20°C	0.20M	U			K1=3.13 B2=6.02	1982RRd (79706)	1160

C11H22N4O2S L CAS 91328-04-4 (1610)
 1,5,8,14-Tetraaza-11-thiacyclohexadecane-2,4-dione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	NaNO3	25°C	0.20M	C	M		B(HgH-1L)=8.40	1989KKa (79838)	1161

C11H23N5O2 L CAS 76201-28-0 (1606)
 1,4,8,11,14-Pentaazacyclohexadecane-5,7-dione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	NaNO3	25°C	0.20M	C	M		B(HgH-1L)=10.14	1989KKa (79896)	1162

C11H25N5O L CAS 91328-02-8 (1605)
 1,5,8,11,14-Pentaazacyclohexadecane-2-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	NaNO3	25°C	0.20M	C	M		K1=22.82 B(HgH-1L)=14.10	1989KKa (79949)	1163

C11H26N4 L CAS 15439-16-4 (7)
 1,4,8,12-Tetraazacyclopentadecane; cyclo(-(NH.CH2.CH2.(N.(CH2)3.)3-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	NaClO4	25°C	0.20M	U	H		K1=23.7	1976KKb (79992)	1164

C11H26N4S L CAS 80846-36-2 (720)
 1-Thia-4,7,11,14-tetraazacyclohexadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.20M	C		K1=25.15	1984KKa (80023)	1165

C11H27N5		L					CAS 29783-72-0	(98)	
1,4,7,10,13-Pentaazacyclohexadecane; cyclo(-(NH.CH2.CH2)5.CH2-)									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.20M	M	H	K1=27.4	1978KKb (80033)	1166
DH1=-143.9 kJ mol-1									

C11H28N4S4		L					CAS 55677-44-6	(1179)	
1,1,3,3,-Tetramercaptoethylamine-propane; CH2(CH(S.CH2.CH2.NH2)2)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.10M	U		K(Hg+H2L)=18.79	1976CJa (80041)	1167

C12H7N2Cl		L					CAS 4199-89-7	(2751)	
5-Chloro-1,10-phenanthroline;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	35°C	0.10M	C	M	K1=12.64 B2=16.48	1998LYa (80146)	1168
							B(HgLA)=21.71		
							B(HgHLA)=29.78		
A is 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime.									

C12H7N3O2		L					CAS 4199-88-6	(449)	
5-Nitro-1,10-phenanthroline;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	35°C	0.10M	C	M	K1=11.40 B2=14.96	1998LYa (80177)	1169
							B(HgLA)=20.46		
							B(HgHLA)=28.32		
A is 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime.									

C12H8N2		L					Phenanthroline CAS 66-71-7	(144)	
1,10-Phenanthroline;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	35°C	0.10M	C	M	K1=13.23 B2=18.12	1998LYa (80459)	1170
							B(HgLA)=22.86		
							B(HgHLA)=30.60		
A is 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime.									

Hg++ gl oth/un 25°C 0.10M C K1=9.85 B2=19.04 1987GGc (80460)1171
B3=23.13
B(HgLOH)=19.85
B(HgL2OH)=23.35
B(HgL(OH)2)=24.2

Medium: KSO3CH3

Hg++ cal non-aq 30°C 100% U H 1976FGa (80461)1172
K(Hg(SCN)2+L)=4.30

Medium: MeCN. DH=-50.5 kJ mol⁻¹; DS=-84

Hg++ cal non-aq 30°C 100% U H 1974FGa (80462)1173
K(HgBr2+L) > 4
K(HgI2+L) > 4

In benzene. For HgBr2, DH=-70.8 kJ mol⁻¹, DS=<-158. For HgI2, DH=-68.5,
DS=<-150

Hg++ ISE NaNO3 20°C 0.10M U B2=19.65 1963ANg (80463)1174
K3=3.7

Hg++ ISE NaNO3 20°C 0.10M U B2=19 1959ANc (80464)1175
Kso=-24.70

C12H10N2O HL CAS 10354-53-7 (3970)
2-Benzoylpyridine oxime; C5H4N.C(:N.OH).C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 40°C 40% U TIH K1=10.16 B2=17.98 1965SSa (80659)1176
Medium: 40% acetone, 0.05 NaClO4. K1=9.44(20 C), 9.79(30 C); K2=6.94(20 C),
7.20(30 C). Also at I=0 to 0.1.

C12H12N2 L CAS 4916-40-9 (4895)
1,2-Bis(2-pyridyl)-ethane; C5H4N.CH2.CH2.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M U K1=7.0 B2=11.40 1970BAa (80992)1177

C12H12N2O3 H2L Phenobarbital CAS 50-06-6 (5924)
5-Ethyl-5-phenyl-barbituric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 20°C 1.0M U K1=10.13 1976BMg (81088)1178

Medium: 1.0 M NaNO3 in 60% v/v EtOH in H2O

C12H13N3 L CAS 1539-42-0 (932)
bis-((2-Pyridyl)methyl)-amine (Di-2-picolylamine); C5H4N.CH2NHCH2.C5H4N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	C	H	B2=22.25	1977AHc (81285)	1179
Calorimetry: DH(B2)=-101.9 kJ mol ⁻¹ , DS(B2)=52.5									

C12H14O14		H6L					CAS 111451-17-3 (5895)		
3,6-Dioxaoctane-1,2,4,5,7,8-hexacarboxylic acid; (CH2(COOH).CH(COOH).O.CH(COOH)-)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	C		K1=15.49 K(HgL+H)=5.48 K(HgHL+H)=5.20 K(HgH2L+H)=4.04 K(HgL+Hg)=8.9	1989MMd (81417)	1180
K(HgH-1L+H)=8.26, B(Hg2H-1L)=22.33, B(Hg2H-3L)=5.16, B(Hg2H-5L)=-13.18									

C12H15NO6S		H2L					CAS 34605-45-3 (4959)		
4-Toluenesulfonyl glutamic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KCl	25°C	0.10M	U		B4=16.10	1968RFa (81523)	1181

C12H16N2O8S4		H6L					(7852)		
N,N'-Bis(dithiocarboxy)-N,N'-bis-1,1'-(1,2-dicarboxyethyl)ethylenediamine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.1M	U		K1=27.0	1999SAa (81616)	1182

C12H16O4S6		L					CAS 66785-63-5 (7805)		
1,4,7,10,13,16-Hexathiacyclooctadecane-2,3,11,12-tetraone;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	con	none	25°C	0.0	C	T H	K1=7.5	1998GRa (81690)	1183
DH(K1)=-108 kJ mol ⁻¹ , DS(K1)=-240 J K ⁻¹ mol ⁻¹ .									
Also data for 15-45 C.									

C12H18N2O8		H4L					CAS 76079-31-7 (2587)		
trans-1,2-Diaminocyclohexane-N,N'-di(propanedioic acid)									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	KNO3	25°C	0.10M	U		K1=19.62 K(Hg+HL)=12.31 K(Hg+H2L)=9.7	1978SGb (81866)	1184

C12H18O8S4 H4L CAS 51865-19-1 (1140)
 (Butanediyliidenetetraethio)tetraethanoic acid; ((HOOC.CH2.S)2.CH.CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	EMF	NaClO4	25°C	0.10M	U				1975JBa (81966)	1185
									K(Hg+H4L)=16.88	

C12H19P L CAS 7650-83-1 (2291)
 Dipropyl-phenyl-phosphine; (CH3.CH2.CH2)2P.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (81995)	1186
HgI2+L in benzene. DH(K1)=-87; DH(K2)=-51 kJ mol-1										

Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (81996)	1187
HgBr2+L in benzene. DH(K1)=-105; DH(K2)=-48 kJ mol-1										

Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (81997)	1188
HgCl2+L in benzene. DH(K1)=-100; DH(K2)=-37 kJ mol-1										

C12H20N2O8 H4L CAS 40623-42-5 (1101)
 1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH2NHCH(COOH)CH2CH2COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	ISE	KNO3	25°C	0.10M	U			K1=16.66	1972GBe (82074)	1189
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Hg++	ISE	KNO3	25°C	0.10M	U			K1=16.64	1972GSi (82075)	1190
									K(Hg+HL)=12.0	
									K(HgHL+HL)=8.15	
									B(Hg(OH)L)=23.20	

C12H20N2O8 H4L CAS 61368-60-3 (3389)
 1,2-Diaminoethane-N,N'-diethanoic-N,N'-di-2-propanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	vlt	KNO3	20°C	0.10M	U			K1=21.33	1970SNa (82134)	1191
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C12H20N2O8 H4L CAS 2458-58-4 (922)
 1,4-Diaminobutane-N,N,N',N'-tetraethanoic acid; (HOOC.CH2)2N.(CH2)4.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	KNO3	25°C	1.0M	C	I			1999ANa (82220)	1192
									B(Hg2L2)=42.50	

Method: Glass electrode plus Hg electrode. At I = 0.1 M, B(Hg2L2)=44.3.

Hg++ cal KNO3 20°C 0.10M U H 1964ANa (82221)1193
DH(K1)=-79.8 kJ mol-1, DS=129 J K-1 mol-1

Hg++ ISE KNO3 20°C 0.10M U K1=20.99 1964LAa (82222)1194

Hg++ EMF NaNO3 20°C 0.10M C K1=20.80 1957SSa (82223)1195
Method: H electrode

Hg++ EMF NaNO3 20°C 0.10M C K1=20.81 1955SAC (82224)1196
Method: H electrode

C12H20N2O8 H4L BDTA CAS 868-43-9 (1742)
DL-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOC.CH2)2N.CH(CH3).CH(CH3).N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 20°C 0.10M U K1=24.13 1971ISa (82305)1197
K(Hg+HL)=3.58

Hg++ vlt KNO3 20°C 0.10M U K1=21.51 1962SKa (82306)1198

C12H20N2O8 H4L CAS 63818-08-6 (2584)
meso-2,3-Diaminobutane-N,N'-di(1,4-butanedioic acid);
(CH(CH3).NH.CH(COOH)(CH2.COOH))2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U K1=17.10 1978SGd (82352)1199
K(Hg+HL)=11.39
K(Hg+H2L)=7.74
B(Hg(OH)L)=23.62

C12H20N2O8 H4L CAS 22968-57-6 (3992)
meso-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOC.CH2)2N.CH(CH3).CH(CH3).N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 20°C 0.10M U K1=22.06 1971ISa (82399)1200
K(Hg+HL)=3.31

Hg++ vlt KNO3 20°C 0.10M U K1=20.33 1962SKa (82400)1201

C12H20N2O8S H4L TEDTA CAS 923-74-0 (3394)
2,2'-Thiobis(ethyliminodiethanoic acid); S(CH2.CH2.N(CH2.COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M U H K1=23.9 1964ANa (82459)1202

K(Hg+HL)=17.5
By calorimetry: DH(K1)=-95.3 kJ mol⁻¹, DS=132 J K⁻¹ mol⁻¹

Hg++ ISE NaNO3 20°C 0.10M C K1=23.81 1957SSa (82460)1203
K(Hg+HL)=17.57

C12H20N2O9 H4L EEDTA CAS 923-73-9 (2112)
Oxa-bis(ethyleneimino)diethanoic acid; ((HOOC.CH2)2N.CH2.CH2)2O

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal KNO3 25°C 0.10M U H 1965WHa (82538)1204
DH(K1)=-81.5 kJ mol⁻¹, DS=167 J K⁻¹ mol⁻¹

Hg++ gl KNO3 20°C 0.10M U H K1=23.09 1964ANa (82539)1205
K(Hg+HL)=16.1

By calorimetry: DH(K1)=-85.7 kJ mol⁻¹, DS=149 J K⁻¹ mol⁻¹

Hg++ ISE KNO3 20°C 0.10M U K1=22.88 1962MMc (82540)1206

Hg++ ISE NaNO3 20°C 0.10M U K1=23.1 1960HRa (82541)1207

Hg++ ISE NaNO3 20°C 0.10M U K1=23.09 1957SSa (82542)1208
K(Hg+HL)=16.14

C12H20O8N2 H4L (6908)
2-Methyl-1,2-diaminopropane-N,N,N'-tetraethanoic acid;
(HOOC.CH2)2N.CH2.C(CH3)2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 20°C 0.10M C K1=22.51 1978NLa (82675)1209

C12H22O12 HL Lactobionic acid CAS 96-82-2 (2487)
4-O-Beta-D-Galactopyranosyl-D-gluconic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 20°C 0.10M C 1997FEb (82931)1210
B(HgH-2L)=-6.64
B(HgH-1L)=-0.29

C12H22S L CAS 7133-46-2 (5698)
S,S-Dicyclohexylsulfide; C6H11.S.C6H11

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE non-aq 25°C 100% U K1=8.04 B2=13.56 1986MMb (82937)1211
B3=15.94
B4=16.87

Medium: acetone, Bu4NC104

C12H24N2O12P4 H8L (1351)
1,3-Diaminomethylbenzene-N,N,N'-tetra(methylenephosphonic) acid;
C6H4(CH2.N(CH2.PO3H2)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.10M	U		K1=9.78 H(Hg+HL)=8.98 K(Hg+H2L)=5.14 K(Hg+H3L)=4.01 K(Hg+H4L)=4.61 (?)	1982PBa (83059)	1212

C12H24N4O4 H2L (7522)
1,4,8,11-Tetraazacyclotetradecane-6,13-dicarboxylic acid

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.50M	U		K1=21.8 K(HgL+H)=8.2 K(HgHL+H)=5.9 *K(HgL)=-13.4	1997BLd (83102)	1213

C12H24O2S4 L (6657)
1,4,7,10-Tetrathia-13,16-dioxacyclooctadecane, 1,4,7,10-Tetrathia-18-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ix	none	25°C	0.0	U		K1=17.4	1991BTa (83118)	1214

C12H24O4S2 L CAS 296-39-9 (4938)
1,4,10,13-Tetraoxa-7,16-dithiacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	oth/un	25°C	0.50M	M	H	K1=19.5	1990IWa (83137)	1215

Medium: 0.5M HNO3. DH(K1)=-74.0 kJ mol⁻¹, DS(K1)= 125 J K⁻¹ mol⁻¹.

C12H24O4S2 L (6528)
7,10,13,16-Tetraoxa-1,4-dithiacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	oth/un	25°C	0.50M	M	H	K1=22.2	1990IWa (83150)	1216

Medium: 0.5M HNO3. DH(K1)=-116.5 kJ mol⁻¹, DS(K1)= 34.9 J K⁻¹ mol⁻¹.

C12H24O6 L 18-Crown-6 CAS 17455-13-9 (577)
1,4,7,10,13,16-Hexaoxacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	nmr	non-aq	27°C	100%	C	M		2003JLa (83367)	1217
							K(Hg(CN) ₂ +L=HgL(CN) ₂)=-0.8		
Medium: DMSO. Method:199Hg nmr.									
Hg++	nmr	non-aq	RT	0.0	C			1991LBa (83368)	1218
							K(Hg(CN) ₂ +L)=>4		
Method: 1H nmr. Medium: acetone/CDC13 1:0.8.									
Hg++	vlt	R4N.X	22°C	0.03M	C	I	K1=2.91	1991PSa (83369)	1219
Medium: 0.025 M Et4NClO4. Method: differential pulse polarography. Data for 15-75% w/w CH3CN/H2O, 0.025 M Et4NClO4.									
Hg++	cal	oth/un	25°C	0.10M	U	H T	K1=2.42	1976ITb (83370)	1220
DH=-19.6 kJ mol ⁻¹ .									

C12H26N2O4 L Cryptand 2,2 CAS 23978-55-4 (925)									
4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	R4N.X	25°C	0.10M	C		K1=15.38	1985CSb (83843)	1221
Medium: 0.10 M Et4NClO4.									
Hg++	gl	R4N.X	25°C	0.10M	C	H	K1=17.85	1975ANa (83844)	1222
Calorimetry: DH1=-71.8 kJ mol ⁻¹ , DS1=101.7									

C12H26N6O2 L CAS 72975-82-7 (1612)									
1,4,7,10,13,16-Hexaazacyclooctadecane-2,6-dione;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	NaNO3	25°C	0.20M	C	M		1989KKa (83963)	1223
							B(HgH-1L)=9.64		

C12H26N12 L (7007)									
1,10-Di(2-(5-tetraazolyl)ethyl)-1,4,7,10-tetraazadecane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaNO3	20°C	0.10M	U		K1=16.94	1981ESa (83971)	1224

C12H26OS L CAS 2180-20-3 (5699)									
S,S-Dihexylsulfoxide; C6H13.SO.C6H13									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	non-aq	25°C	100%	U		K1=2.37 B2=4.11 B3=5.06	1986MMb (83975)	1225

B4=5.40

Medium: acetone, Bu4NClO4

C12H26S L CAS 6294-31-3 (5697)

S,S-Dihexylsulfide; C6H13.S.C6H13

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	non-aq	25°C	100%	U			K1=7.80 B2=13.31 B3=15.90 B4=16.98	1986MMb (84032)	1226

Medium: acetone, Bu4NClO4

C12H27N5O2 HL (7521)

6-Methyl-1,4,8,11-tetraazacyclotetradecane-6-amino-3-carboxylic acid

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.50M	U			K1=17.3 K(HgL+H)=6.8 K(HgHL+H)=6.0 *K(HgL)=-7.8	1997BLd (84111)	1227

C12H27P L CAS 998-40-3 (170)

Tri-n-butylphosphine; (CH3.(CH2)3)3P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	non-aq	25°C	100%	U	H		K1=11.5 B2=19.5 B3=21.7	1987HPc (84136)	1228

Medium: pyridine, 0.1M Et4NClO4

Hg++	cal	non-aq	30°C	100%	U	H		1976FGa (84137)	1229
K(Hg(SCN)2+L) > 5 K(Hg(SCN)2L+L)=1.81									

Medium: MeCN. DH(Hg(SCN)2L)=-83 kJ mol⁻¹; DS=<-178. DH(Hg(SCN)2L2)=-117;

DS=-348

C12H28N4O L (7305)

1-(2-Hydroxyethyl)-1,4,8,11-tetraazacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	R4N.X	25°C	0.10M	C			K1=17.4	1997RWa (84208)	1230

Medium: Et4NClO4

C12H29N5 L CAS 82583-20-6 (97)

1,4,7,11,14-Pentaazacycloheptadecane; cyclo(-(NH.C2H4)3.CH2(NH.C2H4)2.CH2-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl NaClO4 25°C 0.20M M H K1=26.5 1978KKb (84259)1231
DH1=-139.7 kJ mol-1

C12H30N4S4 L CAS 55677-45-7 (1180)
1,1,4,4-Tetramercaptoethylamine-butane; (CH2.CH(S.CH2.CH2.NH2)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaClO4 25°C 0.10M U 1976CJa (84306)1232
K(Hg+H2L)=19.36

C12H30N6 L CAS 296-35-5 (143)
1,4,7,10,13,16-Hexaazacyclooctadecane; cyclo(-(NH.CH2.CH2)6-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaNO3 25°C 0.20M C K1=29.7 1991KKa (84331)1233

Hg++ vlt NaClO4 25°C 0.20M U H K1=29.1 1980KKb (84332)1234
DH=-176 kJ mol-1, DS=-17 J K mol-1

C12H30N6 L (6409)
6,13-Dimethyl-1,4,8,11-tetraazacyclotetradecane-6,13-diamine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KCl 25°C 0.50M U K1=10.5 1997BLd (84376)1235
K(HgL+H)=7.5
K(HgHL+H)=5.7

Hg++ gl KCl 25°C 0.50M U I K1=12.2 1994LLb (84377)1236
K(HgL+H)=6.8
K(HgHL+H)=6.8
K(HgH-1L+H)=8.0

Data are for the syn isomer. For the anti isomer, K1=10.5, K(HgL+H)=7.5,
K(HgHL+H)=5.7, K(Hg2+L)=11.2, K(Hg2L+H)=9.1, K(Hg2H-1L+H)=7.7.

C12H32N4O12P4 H8L DOTPH CAS 91987-74-5 (229)
1,4,7,10-Tetraazacyclododecane-N,N',N'',N'''-tetramethylenephosphonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 1.0M U K1=25.1 1984KMb (84411)1237
K(Hg+HL)=21.1

C13H9N3O7S3 H3L CAS 2172-27-2 (5007)
1-(2-Thiazolylazo)-2-naphthol-3,6-disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Hg++ sp NaCl04 25°C 0.10M U 1983PEa (84652)1238
 $K(Hg+HL=HgL+H)=3.34$
 $K(2Hg+HL=Hg2H-1L+2H)=6.38$

C13H9N3O8S3 H3L CAS 28467-51-8 (898)
 2-(2-Thiazolylazo)chromotropic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp NaCl04 25°C 0.10M U 1983PEa (84664)1239
 $K(2Hg+H2L=Hg2HL+H)=5.28$

C13H10N02Cl HL (8130)
 N-(2-Chlorophenyl)benzohydroxamic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl diox/w 25°C 50% U $K1=7.00$ $B2=12.40$ 1986ARb (84709)1240
 Also data for the N-(2-chlorophenyl)-3-methoxy, 3-methyl, 3-fluoro,
 3-chloro, 3-bromo-, 3-iodo and 3-nitro-benzohydroxamic acids.

C13H10N2 L CAS 3003-78-6 (2752)
 5-Methyl-1,10-phenanthroline;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 35°C 0.10M C M $K1=14.03$ $B2=19.03$ 1998LYa (84815)1241
 $B(HgLA)=23.86$
 $B(HgHLA)=31.61$

A is 3,3,9,9-tetramethyl-4,8-diazaundecane-2,10-dione dioxime.

C13H10N2O5S H2L CAS 98789-35-6 (5012)
 4-Hydroxy-3-formylazobenzene-4'-sulfonic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF alc/w 25°C 42% U 1972DSc (84921)1242
 $K(Hg+HL=HgL+H)=3.73$
 $K(HgL+HL=HgL2+H)=3.22$

Medium: 42% EtOH, 0.2 M NaCl04

C13H11N3O5S H3L (5019)
 4-Hydroxy-3-oximinomethylazobenzene-4'-sulfonic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 25°C 50% U $K1=3.65$ $B2=6.88$ 1973DSa (85298)1243
 Medium: 42% EtOH, 0.2 M NaCl04

C13H12N2S L diPh-thiourea CAS 102-08-9 (1075)
1,3-Diphenyl-2-thiourea; C6H5.NH.CS.NH.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	25°C	0.10M	U		B2=21.34	1986ECa (85388)	1244

C13H12N2S L (2601)
N,N-Diphenylthiocarbamide; (C6H5)2N.CS.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE mixed		25°C	82%	U		B4=23.17	1979MTd (85393)	1245

Medium: 82% DMF/H2O

C13H12N4S L Dithizone CAS 60-10-6 (1801)
Diphenylthiocarbazone; C6H5.NH.NH.CS.N:N.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	NaCl04	25°C	0.10M	U		K1=20.64 B2=40.30	1973BSe (85459)	1246
Hg++	sp	non-aq	?	100%	U		B2=20.7	1971BBa (85460)	1247

Medium: N-methylpyrrolidone, 0.1 M

Hg++	dis	oth/un	25°C	var	U			1960DTa (85461)	1248
							K(Hg+2HL)=40.34		
							K(Hg(HL)2)=(Hg(HL)2)org)=3.54		

At 35 C: K(Hg(HL)2)=(Hg(HL)2)org)=3.57

Hg++	sp	NaCl04	25°C	0.50M	U			1956BRa (85462)	1249
							K(Hg+2H2L=Hg(HL)2+2H)=26.85		

C13H13As L CAS 954-48-2 (2298)
Methyldiphenylarsine; CH3.As.(C6H5)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H		1976GGb (85487)	1250
							K(HgCl2+L)=3.47		
							K(HgBr2+L)=3.05		
							K(HgI2+L)=2.05		

In benzene. DH(HgCl2L)=-72.4 kJ mol⁻¹; DS=-173. DH(HgBr2L)=-77.3; DS=-198.
DH(HgI2L)=-71.7; DS=-198.

C13H13P L CAS 1486-28-8 (1731)
Diphenyl-methyl-phosphine; CH3(C6H5)2P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ cal non-aq 30°C 100% U H 1976GGa (85550)1251
HgCl2+L in benzene: DH(K1)=-92; DH(K2)=-37 kJ mol-1

Hg++ cal non-aq 30°C 100% U H 1976GGa (85551)1252
HgBr2+L in benzene: DH(K1)=-89; DH(K2)=-39 kJ mol-1.

C13H14N2 L CAS 104986-55-2 (4972)
1,3-Bis(2'-pyridyl)-propane; C5H4N.CH2.CH2.CH2.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 20°C 0.10M U K1=7.8 B2=11.80 1970BAa (85574)1253

C13H14N2O3 HL Antineoplaston CAS 91531-30-5 (8098)
3-(N-Phenylacetyl amino)-2,6-piperidinedione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl alc/w 45°C 50% C K1=8.23 B2=13.82 1996MMc (85627)1254
Medium: 50% v/v MeOH/H2O, 0.10 M KNO3.

C13H14N4 L CAS 13103-75-8 (473)
4-(2-Pyridylazo)-N,N-dimethylaniline; C5H4N.N:N.C6H4.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp NaNO3 25°C 0.15M U K1=5.06 1953KMa (85683)1255

C13H17N3O4S HL Analgin CAS 57904-20-8 (6340)
1-Phenyl-2,3-dimethylaminopyrazolon-5-N-methansulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE NaNO3 25°C 0.10M U I K1=3.11 B2=5.40 1988GAa (86004)1256
B3=8.80
B4=12.40

In 5.7 Mol EtOH/H2O: B1=3.40; B2=6.60; B3=9.10; B4=12.40

C13H20N2O8 H4L CAS 22991-70-4 (3413)
trans-1,2-Cyclopentane-iminodiethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF NaNO3 20°C 0.10M U K1=23.24 1971PSc (86113)1257
B(HgHL)=26.01 (DL isomer)

C13H20O8S4 H4L CAS 51865-20-4 (1139)
(Pentanediy lidenetetrathio)tetra-ethanoic acid; ((HOOCCCH2S)2CHCH2)2.CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.10M	U		K1=18.97	1975JBa (86156)	1258

C13H22N2O8 H4L CAS 1798-14-7 (921)
(Pentamethylenedinitrilo)tetraethanoic acid; ((HOOCH2)2N.CH2.CH2)2CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	1.0M	C I		B(Hg2L2)=43.5	1999ANa (86196)	1259

Method: Glass electrode plus Hg electrode.

C13H22N2O8 H4L (7164)
2,4-Diaminopentane-N,N,N',N'-tetraethanoic acid;
(HOOCH2)2NCH(CH3)CH2CH(CH3)N(CH2COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U		K1=20.68	1981NSc (86255)	1260

C13H23N3O8 H4L (3414)
N-Methyl-2,2'-iminobis(ethyliminodiethanoic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	NaNO3	20°C	0.10M	C		K1=7.21 K(Hg+HL)=2.61	1957SSa (86396)	1261

C13H26O4S2 L (6656)
1,5-Dithia-8,11,14,17-tetraoxacyclononadecane, 1,5-Dithia-19-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ix	none	25°C	0.0	U		K1=13.9	1991BTa (86461)	1262

C13H28N4O2 L CAS 17023-02-8 (7247)
3,3,9,9-Tetramethyl-4,8-diazaundecane-2,10-dione dioxime;
(HON:C(CH3)C(CH3)2NHCH2)2CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	35°C	0.10M	C M		K1=11.79 B(HgHL)=18.60	1998LYa (86534)	1263

Ternary complexes with 5-substituted-1,10-phenanthrolines.

C13H32N4S4 L (1181)
1,1,5,5-Tetramercaptoethylamine-pentane; CH2(CH2.CH(S.CH2.CH2.NH2)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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$$K(Hg+H_2L)=19.55$$

C14H807S H3L DASA CAS 83-61-4 (950)

1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl NaCl04 30°C 0.0 U I K1=8.38 B2=15.28 1972GDa (86735)1265

I=0.02: K1=8.43, K2=6.68; 0.05: K1=8.64, K2=6.33;

0.15: $K_1=8.56$, $K_2=6.25$; 0.2: $K_1=8.04$, $K_2=6.10$

C14H11N5O8S2 H5L CAS 1105-53-9 (5084)

1,5-Bis(2-hydroxy-5-sulfohenyl)-3-cyanoformazan;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ g1 NaNO3 20°C 0.10M U K1=12.11 1971SEa (87019)1266

C14H12O2 HL CAS 119-53-9 (2739)

2-Hydroxydeoxybenzoin, 2-hydroxyphenylacetophenone; $\text{HO.C}_6\text{H}_5.\text{CH}_2.\text{CO.C}_6\text{H}_5$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl alc/w 30°C 50% U K1=7.10 1986SBa (87330)1267

C14H12O3 H2L CAS 3669-41-8 (2740)

2,4-Dihydroxydeoxybenzoin, 2,4-dihydroxyphenylacetophenone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl alc/w 30°C 50% U K1=3.65 1986SBa (87341)1268

C14H12O4 H3L (2741)

2,4,6-Trihydroxydeoxybenzoin, 2,4,6-trihydroxyphenylacetophenone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl alc/w 30°C 50% U K1=4.65 1986SBa (87357)1269

C14H13N5OS HL (5394)

1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)thiourea;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ sp mixed 25°C 40% U 1985RGa (87615)1270

$$K_{1eff}=6.43$$
$$B2_{eff}=12.31$$

Medium: 40% DMF, pH 4.5

C14H13O2P HL CAS 3064-56-0 (7013)
2-(Diphenylphosphino)-ethanoic acid; (C6H5)2P.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	NaCl04	25°C	0.10M	U	I	K1=18.7 B2=28.91 B3=35.38 B4=37.2	1979PPc (87634)	1271

Method: Hg elec. In 50% v/v dioxan/H2O: K1=19.4; B2=30.3 ; B3=35.39; B4=36.5

Hg++	gl	NaCl04	25°C	0.10M	C	I	K1=18.7 B2=28.90 B3=35.35 B4=37.2	1979PPd (87635)	1272
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Additional methods: polarography and Hg electrode.

In 50% dioxane/H2O: K1=19.4, B2=30.3, B3=35.39, B4=36.5

C14H13P L CAS 2155-96-6 (2289)
Diphenyl-ethenyl-phosphine; (C6H5)2.P.CH:CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H		1976GGa (87638)	1273

HgI2+L in benzene. DH(K1)=-60; DH(K2)=-49 kJ mol-1

Hg++	cal	non-aq	30°C	100%	U	H		1976GGa (87639)	1274
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HgBr2+L in benzene. DH(K1)=-78; DH(K2)=-42 kJ mol-1

Hg++	cal	non-aq	30°C	100%	U	H		1976GGa (87640)	1275
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HgCl2+L in benzene. DH(K1)=-85; DH(K2)=-38 kJ mol-1

C14H16N2 L CAS 1620-43-7 (5033)
1,4-Bis(2'-pyridyl)butane; C5H4N.CH2.CH2.CH2.CH2.C5H4N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U		K1=8.3 B2=11.50	1970BAa (87837)	1276

C14H16N2O8 H4L CAS 40774-59-2 (1901)
1,2-Diaminobenzene-N,N,N',N'-tetraethanoic acid; C6H4(N(CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaCl04	25°C	1.00M	C	H	K1=17.38	1992ANb (87953)	1277

By calorimetry: DH(K1)=-50.6 kJ mol-1, DS=165 J K-1 mol-1

C14H16N4O HL PAAC CAS 13059-69-3 (5067)
5-Ethylamino-4-methyl-2-(2'-pyridylazo)phenol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++ sp oth/un ? ? U 1968GKb (88017)1278

K(Hg+HL)=5.77

C14H18N4 L DPEN CAS 4608-34-3 (1850)

N,N'-Bis-(2-pyridylmethyl)-1,2-diaminoethane; (C5H4N.CH2.NH.CH2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 0.10M U H K1=19.1 B2=24.68 1975APc (88113)1279

DH(K1)=-92.0 kJ mol-1, DS(K1)=57.3 J K-1 mol-1

DH(K2)=-23.0 kJ mol-1 DS(K2)=30.1 J K-1 mol-1

C14H21NO7 HL CAS 85906-10-1 (6635)

2-(Benzylamino)-2-deoxy-D-glycero-D-gulo-heptonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaCl04 25°C 0.10M U K1=5.41 B2=9.78 1992Vda (88409)1280

B(HgH-1L)=-1.80

B(HgHL2)=16.51

B(HgH2L2)=21.6

C14H22N2O8 H4L CDTA CAS 482-54-2 (200)

trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 0.10M C T H K1=22.56 1985HWc (88674)1281

Data for 5-35 C. Method: Hg and glass electrodes.

DH(K1)=-71.6 kJ mol-1, DS(K1)=192 J K-1 mol-1.

Hg++ ISE NaCl04 25°C 0.50M U K1=23.28 1977GGb (88675)1282

Hg++ sp NaCl04 25°C 0.10M U M 1966JMa (88676)1283

K(HgL+OH)=3.20

K(HgL+Cl)=2.16

K(HgL+Br)=3.20

K(HgL+I)=5.3

K(HgL+SCN)=4.29

Hg++ ISE KNO3 30°C 0.10M U T H K1=23.47 1965HWa (88677)1284

K1=24.16(10 C), 23.77(20 C). At 25 C: DH(K1)=-57.3 J K-1 mol-1, DS=259 J K-1m-1

Hg++ cal KNO3 25°C 0.10M U H 1965WHa (88678)1285

DH(K1)=-79.0 kJ mol-1, DS=201 J K-1 mol-1

Hg++ cal KNO3 20°C 0.10M U T H 1963ANb (88679)1286

DH(K1)=-69.4 kJ mol-1, DS=246.6 J K-1 mol-1

Hg++ ISE KNO3 20°C 0.10M U H K1=24.95 1963ANf (88680)1287
By calorimetry: DH(K1)=-69.5 kJ mol⁻¹, DS=247 J K⁻¹ mol⁻¹

Hg++ ISE KNO3 25°C 0.10M U T H K1=23.43 1962MHa (88681)1288
DH(K1)=-58.9 kJ mol⁻¹, DS=247 J K⁻¹ mol⁻¹. At 20 C: K(HgL+H)=3.51

Hg++ ISE KNO3 25°C 0.10M U K1=24.4 1960HRa (88682)1289

Hg++ oth oth/un ? ? U 1959SRb (88683)1290
K(HgL+NH3)=5.5

Hg++ vlt oth/un ? ? U K1=24.3 1955MDa (88684)1291

Hg++ ISE NaNO3 20°C 0.10M U K1=24.30 1954SGa (88685)1292
K(HgLOH+H)=10.46
K(HgL+H)=9.50

C14H22O8S4 H4L (1160)
Ethane-tetramercaptopropanoic acid; (CH.(S.CH2.CH2.COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl NaCl04 25°C 0.10M U K1=17.3 1975PJa (89000)1293

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 0.10M C T H K1=27.23 1985HWc (89273)1294
Data for 5-35 C. Method: Hg and glass electrodes.
DH(K1)=-109.0 kJ mol⁻¹, DS(K1)=155 J K⁻¹ mol⁻¹.

Hg++ ISE NaCl04 25°C 0.50M U K1=26.08 1977GGb (89274)1295

Hg++ sp oth/un 20°C <0.1 U K1=28.4 1967KAb (89275)1296

Hg++ cal KNO3 20°C 0.10M U T H 1965ANa (89276)1297
DH(K1)=-99.1 kJ mol⁻¹, DS=172.6 J K⁻¹ mol⁻¹

Hg++ cal KNO3 25°C 0.10M U H 1965WHa (89277)1298
DH(K1)=-98.6 kJ mol⁻¹, DS=184 J K⁻¹ mol⁻¹

Hg++ ISE KNO3 25°C 0.10M U H K1=26.27 1962MTc (89278)1299
DH(K1)=-92.5 kJ mol⁻¹, DS=193 J K⁻¹ mol⁻¹

Hg++ vlt KNO3 20°C 0.10M U K1=25.4 1962SKa (89279)1300

Hg++ ISE KNO3 25°C 0.10M U K1=27.0 1960HRa (89280)1301

Hg++ gl KNO3 25°C 0.10M C K1=27.0 1960WAa (89281)1302
K(HgL+H)=3.6

Hg++ EMF oth/un 20°C 0.10M U K1=26.70 1959AND (89282)1303
K(Hg+HL)=20.36

C14H23N3S2 L CAS 771500-58-4 (9194)
5-(3-Aminopropyl)-2,8-dithia-5-aza-2,6-pyridinophane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=13.1 2004BBe (89459)1304
K(HgL+H)=4.9
K(HgL+OH)=2.9

Medium: 0.1 M Me4NO3

C14H23P L CAS 6372-44-7 (2292)
Dibutyl-phenyl-phosphine; (C4H9)2(C6H5)P

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1976GGa (89479)1305
HgCl2+L in benzene. DH(K1)=-108; DH(K2)=-49 kJ mol-1

Hg++ cal non-aq 30°C 100% U H 1976GGa (89480)1306
HgI2+L in benzene. DH(K1)=-86; DH(K2)=-46 kJ mol-1.

C14H24N2O7 H3L (3440)
N-(2-Hydroxycyclohexyl)ethylenediamine-N,N',N'-triethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE oth/un 25°C 0.10M U K1=20.14 1960SAc (89495)1307
K(Hg+HL)=12.56

C14H24N2O8 H4L (5075)
1,2-Diaminoethane-N,N'-diethanoic-N,N'-di-2-butyric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO3 20°C 0.10M U K1=20.53 1970SNa (89510)1308

C14H24N2O8 H4L HMDTA CAS 1633-00-7 (920)
1,6-Diaminohexane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.CH2.CH2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 1.0M C I 1999ANa (89579)1309
B(Hg2L2)=43.4

Method: Glass electrode plus Hg electrode. At I = 0.1 M, B(Hg2L2)=45.5.

Hg++ gl KNO3 20°C 0.10M U H K1=21.58 1964ANa (89580)1310
By calorimetry: DH(K1)=-87.7 kJ mol⁻¹, DS=114.1 J K⁻¹ mol⁻¹

Hg++ ISE NaNO3 20°C 0.10M U K1=21.38 1957SSa (89581)1311

C14H24N2O8S2 H4L (3441)
2,2'-Ethylenebisthio(ethyliminodiethanoic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ ISE NaNO3 20°C 0.10M U K1=23.93 1957SSa (89697)1312
K(Hg+HL)=18.50

C14H24N2O9 H4L CAS 87720-52-3 (1593)
2,2'-Oxybis(propyliminodiethanoic acid)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl KCl 20°C 0.10M U K1=21.67 1961ISa (89711)1313
K(Hg+HL)=15.0

C14H24N2O9 H4L BPETA CAS 87720-52-3 (5077)
Bis-(3-di(carboxymethyl)aminopropyl)ether;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl KCl 20°C 0.10M U K1=21.67 1961ISa (89730)1314
K(Hg+HL)=15.0

C14H24N2O10 EGTA CAS 67-42-5 (349)
Ethyleneglycol-0,0'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl KNO3 25°C 0.10M C T H K1=22.62 1985HWc (89870)1315
Data for 5-35 C. Method: Hg and glass electrodes.
DH(K1)=-70.0 kJ mol⁻¹, DS(K1)=212 J K⁻¹ mol⁻¹.

Hg++ gl KNO3 25°C 0.10M U K1=23.6 1982JGa (89871)1316
K(HgL+H)=3.8
K(HgL+2H)=3.1

Hg++ gl NaClO4 25°C 0.10M U K1=23.86 1970FTa (89872)1317
K(HgL+H)=3.06

Hg++ cal KNO3 25°C 0.10M U H 1965WHa (89873)1318
DH(K1)=-97.4 kJ mol⁻¹, DS=129.6 J K⁻¹ mol⁻¹

Hg++ gl KNO3 20°C 0.10M U H K1=23.2 1964ANa (89874)1319

K(Hg+HL)=16.8

By calorimetry: DH(K1)=-99.1 kJ mol⁻¹, DS=105.3 J K⁻¹ mol⁻¹

Hg++ ISE KNO3 20°C 0.10M U K1=23.12 1962MMc (89875)1320

Hg++ ISE KNO3 25°C 0.10M U K1=23.8 1960HRa (89876)1321

Hg++ ISE NaNO3 20°C 0.10M U K1=23.20 1957SRa (89877)1322
K(Hg+HL)=16.76

By glass electrode K1=23.8

C14H28N2O4 L Cryptand 2,1,1 CAS 31250-06-3 (836)

1,10-Diaza-4,7,13,18-tetraoxabicyclo[8,5,5]eicosane (2,1,1);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M U K1=15.97 1982NSb (90369)1323
B(HgHL)=18.71

Hg++ EMF non-aq 25°C 100% C K1=15.9 1979BLb (90370)1324
Method: Ag electrode; competition with Ag+. Medium: MeOH, 0.05 M
Me4NC104.

C14H28O7 L 21-Crown-7 CAS 33089-36-0 (2264)

1,4,7,10,13,16,19-Heptaoxacycloheptadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ nmr non-aq RT 0.0 C K(Hg(CN)2+L)=3.18 1991LBa (90521)1325

Method: 1H nmr. Medium: acetone/CDCl3 1:0.8.

C14H30N2O5 L (6722)

7,13-Bis(2-hydroxyethyl)-1,4,10-trioxa-7,13-diazacyclopentadecane

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=15.07 1995LLa (90628)1326
Medium: Et4NC104

C14H32N2O4 L CAS 102-60-3 (2678)

Tetra(2-hydroxypropyl)-N,N,N',N'-diaminoethane; (-CH2.N(CH2.CH(OH).CH3)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl oth/un 27°C 0.05M U K1=8.08 1959KEc (90744)1327

C14H32N4 L 4-Mecyclam-14 CAS 41203-22-9 (935)

1,4,8,11-Tetramethyl-1,4,8,11-tetraazacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO3	25°C	0.10M	U			K1=20.3	1983Nwa	(90803)1328

C15H11N3O4S		H2L		1-PAN-4S				(7010)		
2-(2-Pyridylazo)-1-naphthol-4-sulfonic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	alc/w	25°C	30%	U			K1=12.4 K(Hg+HL)=4.7	1979KKd	(91325)1329
Medium: 30% EtOH/H2O, 1.0 M HClO4										

C15H14N4O6S		H3L						(5132)		
3-Methyl-(2'-hydroxy-5'-sulfophenyl)-(2''-carboxyphenyl)formazan;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	NaNO3	20°C	0.10M	U			B2=14.57 K(Hg+HL)=7.62 K(HgHL+HL)=6.95	1970VEb	(91749)1330

C15H17P		L						CAS 7650-84-2	(2287)	
Diphenyl-propyl-phosphine; (C6H5)2PCH2.CH2.CH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H			1976GGa	(91989)1331
HgCl2+L in benzene: DH(K1)=-90; DH(K2)=-41 kJ mol-1										

C15H18N2		L						CAS 25382-73-6	(5106)	
1,5-Bis(2-pyridyl)-pentane; C5H4N.(CH2)5.C5H4N										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U			K1=9.6	1970BAa	(92003)1332

C15H20N4		L		DPTN				CAS 63671-70-5	(1851)	
N,N'-Bis-(2-pyridylmethyl)-1,3-diaminopropane; (C5H4N.CH2.NH.CH2)2CH2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	KNO3	25°C	0.10M	U	H		K1=18.8	1975APc	(92182)1333
DH(K1)=-94.5 kJ mol-1, DS=42.3 J K-1 mol-1										

C15H24O8S4		H4L						CAS 53480-91-4	(1161)	
Propane-1,1,3,3-tetramercaptopropanoic acid; CH2(CH(S.CH2.CH2.COOH)2)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Hg++ gl NaClO4 25°C 0.10M U K1=17.5 1975PJa (92353)1334

C15H32N4S3 L CAS 150148-73-5 (7926)
1-Methyl-3,13,16-trithia-6,10,19-triazabicyclo[6.6.6]eicosan-8-amine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO3 25°C 0.10M C K1=17.7 2001SGa (92559)1335
K(HgL+H)=5.3

C16H9N2OBr3 HL CAS 84317-74-8 (5169)
1-(2,4,6-Tribromophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=5.90 B2=10.43 1972MCb (92654)1336
Medium: 75% acetone, 0.1 M KNO3

C16H11N2OBr HL CAS 7150-24-5 (5172)
1-(4-Bromophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=6.74 B2=12.70 1972MCb (92699)1337
Medium: 75% acetone, 0.1 M KNO3

C16H11N2OCl HL CAS 24390-65-6 (5170)
1-(2-Chlorophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=6.24 B2=11.44 1972MCb (92714)1338
Medium: 75% acetone, 0.1 M KNO3

C16H11N2OCl HL CAS 10149-93-6 (5171)
1-(4-Chlorophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=6.68 B2=12.60 1972MCb (92729)1339
Medium: 75% acetone, 0.1 M KNO3

C16H11N2OI HL CAS 25023-35-2 (5173)
1-(4-Iodophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=6.88 B2=12.89 1972MCb (92744)1340
Medium: 75% acetone, 0.1 M KNO3

C16H11N3O3 HL CAS 6410-09-9 (5151)
1-(2-Nitrophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=3.17 1972MCb (92798)1341
Medium: 75% acetone, 0.1 M KNO3

C16H11N3O3 HL CAS 6410-46-1 (5152)
1-(4-Nitrophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=3.98 B2=7.45 1972MCb (92813)1342
Medium: 75% acetone, 0.1 M KNO3

C16H12N2O HL CAS 842-07-9 (5156)
1-Phenylazo-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=7.46 B2=13.88 1972MCb (92919)1343
Medium: 75% acetone, 0.1 M KNO3

C16H12N2O2 H2L CAS 9486-98-2 (3462)
1-(2-Hydroxyphenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U 1972MCb (92953)1344

K(Hg+HL)=7.65

K(HgHL+HL)=6.87

Medium: 75% acetone, 0.1 M KNO3

C16H12N2O2 H2L CAS 14934-27-1 (5157)
1-(4-Hydroxyphenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U 1972MCb (92971)1345

K(Hg+HL)=7.26

K(HgHL+HL)=6.04

Medium: 75% acetone, 0.1 M KNO3

C16H12N2O4S H2L CAS 13964-82-4 (3475)
1-(4-Sulfophenylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl mixed 25°C 75% U K1=3.70 B2=6.66 1972MCb (92999)1346

Medium: 75% acetone, 0.1 M KNO₃

C16H18O8S4 H4L CAS 51865-21-5 (239)

1,2-Dimethylbenzene-tetrathioethanoic acid; C₆H₄(CH(S.CH₂.COOH)₂)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF NaCl04 25°C 0.10M U K1=45.0 1975JBa (93887)1347
K1eff=18.8 (pH < 3.8)

Hg++ ISE KNO₃ 25°C 0.10M U 1974JBa (93888)1348

K(Hg+H₂L)=18.8

B(HgL(OH)₂)=45

C16H19P L CAS 6372-41-4 (2288)

Diphenyl-butyl-phosphine; (C₆H₅)₂P(C₄H₉)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1976GGa (93944)1349
HgCl₂+L in benzene. DH(K₁)=-90; DH(K₂)=-37 kJ mol⁻¹

Hg++ cal non-aq 30°C 100% U H 1976GGa (93945)1350
HgBr₂+L in benzene. DH(K₁)=-92; DH(K₂)=-43 kJ mol⁻¹

Hg++ cal non-aq 30°C 100% U H 1976GGa (93946)1351
HgI₂+L in benzene. DH(K₁)=-80; DH(K₂)=-48 kJ mol⁻¹

C16H20N₂ L (5146)

1,6-Bis(2-pyridyl)-hexane; C₅H₄N.(CH₂)₆.C₅H₄N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO₃ 20°C 0.10M U K1=9.2 1970BAa (93957)1352

C16H20N₂O8 H4L CAS 6411-02-5 (1919)

1-Phenyl-ethylenediamine-N,N,N',N'-tetraethanoic acid (DL)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ vlt KNO₃ 20°C 0.10M U K1=22.06 1989SLa (94039)1353

C16H22N₄ L DPTE CAS 81747-99-1 (1852)

N,N-Bis-(2-pyridyl-methyl)-1,4-diaminobutane; (C₅H₄N.CH₂.NH.CH₂.CH₂)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO₃ 25°C 0.10M U H K1=17.52 1975APc (94181)1354
DH(K₁)=-87.0 kJ mol⁻¹ DS=43.1 J K⁻¹ mol⁻¹

C16H22N4O L (3471)
2-(N-(2-Dimethylaminoethyl)-N-(4-methoxybenzyl)amino)pyrimidine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KCl	25°C	0.06M	U T		K1=3.61 B2=7.05	1961SMa (94199)	1355

K1=4.02(0 C), 3.54(45 C)

C16H25NO4 L (7444)
1-Aza-4,7,10,13-tetraoxa-1-phenyl-cyclopentadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	con	mixed	25°C	80%	C IH			1999MFa (94517)	1356

K(Hg(NO3)2+L)=4.66

Medium: 80% acetonitrile/H2O. Also data for 95% acetonitrile/H2O, and for 20-35 C. DH(K)=-51.85 kJ mol⁻¹, DS(K)=-78.4 J K⁻¹ mol⁻¹.

C16H26O8S4 H4L CAS 53480-92-5 (1162)
Butane-1,1,4,4-tetramercaptopropanoic acid; (CH2.CH(S.CH2.CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaClO4	25°C	0.10M	U		K1=17.7	1975PJa (94639)	1357

C16H28N2O8 H4L (5167)
1,2-Diaminoethane-N,N'-diethanoic-N,N'-di-2-(3-methyl)butanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U		K1=18.03	1970SNa (94712)	1358

C16H28N2O8 H4L (5168)
1,2-Diaminoethane-N,N'-diethanoic-N,N'-di-2-pentanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U		K1=20.54	1970SNa (94738)	1359

C16H28N2O8 H4L (5138)
1,2-Diaminooctane-N,N,N',N'-tetraethanoic acid;
(HOOCCCH2)2N.CH2.CH(C6H13)N(CH2COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U		K1=22.77	1979MBd (94764)	1360

C16H28N2O8 H4L (2850)
1,8-Diaminooctane-N,N,N',N'-tetraethanoic acid; ((HOOCCCH2)2N(CH2)4)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KNO3	20°C	0.10M	U	H	K1=21.83	1964ANa (94792)	1361
By calorimetry: DH(K1)=-82.5 kJ mol ⁻¹ , DS=136 J K ⁻¹ mol ⁻¹									
Hg++	oth	NaNO3	20°C	0.10M	U		K1=21.83	1957SSa (94793)	1362

C16H28N4O8		H4L		DOTA			CAS 60239-18-1	(1017)	
1,4,7,10-Tetraazacyclododecane-1,4,7,10-tetraethanoic acid;									
Hg++	vlt	KNO3	25°C	0.20M	C	H	K1=26.4	1994KOb (94904)	1363
DH(K1)=-86 kJ mol ⁻¹ , DS=79 J K ⁻¹ mol ⁻¹									

C16H30N4O8		H4L					(3473)		
N,N'-Dimethyl-2,2'-ethylenedi-iminobis(ethylenediethanoic acid);									
Hg++	ISE	NaNO3	20°C	0.10M	U		K1=27.68 K(Hg+HL)=21.24	1957SSa (95083)	1364

C16H30N4S4		H4L					CAS 55677-47-9	(1203)	
Tetramercaptoethylamine-1,2-dimethylbenzene; C6H4(CH(NH.CH2.CH2.SH)4)2									
Hg++	gl	NaClO4	25°C	0.10M	U		K1=19.65	1976CJa (95090)	1365

C16H32N2O5		L		Cryptand		2,2,1	CAS 31364-42-8	(837)	
1,10-Diaza-4,7,13,16,21-pentaoxabicyclo[8,8,5]tricosane (2,2,1);									
Hg++	gl	R4N.X	25°C	0.10M	U		K1=19.97	1982NSb (95209)	1366

C16H32N8O4		L					CAS 157599-02-5	(8676)	
1,4,7,10-Tetraazacyclododecane-1,4,7,10-tetraacetamide;									
Hg++	gl	NaNO3	25°C	0.10M	C		K1=15.53	1995MHa (95376)	1367

C16H34N2O5		L					(6953)		
7,13-Bis(2-methoxyethyl)-1,4,10-trioxa-7,13-diazacyclopentadecane;									
Hg++	gl	R4N.X	25°C	0.10M	C		K1=15.2	1995LLa (95414)	1368

Medium: Et4NC104

C16H34N2O6 L CAS 69930-74-1 (1321)
N,N'-Bis(2-hydroxyethyl)-1,7,10,16-tetraoxa-4,13-diazacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=14.58 1995LLa (95448)1369

Medium: Et4NC104

C16H34N4O2 L CAS 60598-04-1 (1530)
4,7-Dimethyl-1,4,7,10-tetraaza-13,18-dioxabicyclo[8,5,5]eicosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M U K1=26.6 1978LMa (95470)1370
K(Hg+HL)=19.3

C16H34OS L CAS 1986-89-6 (5700)
S,S-Dioctylsulfoxide; C8H17.SO.C8H17

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE non-aq 25°C 100% U K1=2.44 B2=4.08 1986MMb (95483)1371
B3=5.20
B4=5.61

Medium: acetone, Bu4NC104

C16H36N4O2 L (7297)
1,11-Bis(2-hydroxyethyl)-4,8-dimethyl-1,4,8,11-tetraazacyclotetradecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=18.1 1996BCc (95549)1372
B(HgH-1L)=10.8

Medium: Et4NC104

C16H36N4O2 L (7296)
1,4-Bis(2-hydroxyethyl)-8,11-dimethyl-1,4,8,11-tetraazacyclotetradecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl R4N.X 25°C 0.10M C K1=17.3 1996BCc (95557)1373
B(HgH-1L)=10.0

Medium: Et4Cl04

C16H40N4O12P4 H8L CAS 41007-47-0 (2070)
1,4,7,10-Tetraethylphosphonic acid-1,4,7,10-tetraazacyclododecane;
C8H16N4(CH2CH2.PO(OH)2)4

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	1.00M	U			K1=29.6 K(Hg+HL)=25.1 K(Hg+H2L)=21.4 K(Hg+H3L)=19.6 K(Hg+H4L)=17.2	1989PBb (95638)	1374

C17H13NO3S H2L CAS 119516-70-0 (6185)
7-Hydroxy-8((2-mercaptophenyl)iminomethyl)-4-methyl-2H-1-benzopyran-2-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	diox/w	20°C	70%	U	T H		K1=22.40 25 C:K=21.27; 32 C: K=19.76; 45 C:K=17.25. DH=-382 kJ mol-1, DS=-876	1988K0b (95749)	1375

C17H14N2O HL CAS 2046-17-5 (5214)
1-(2-Methylphenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	mixed	25°C	75%	U			K1=7.76 B2=14.12	1972MCb (95796)	1376

Medium: 75% acetone, 0.1 M KNO3

C17H14N2O HL CAS 6756-41-8 (5215)
1-(4-Methylphenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	mixed	25°C	75%	U			K1=7.94 B2=14.88	1972MCb (95811)	1377

Medium: 75% acetone, 0.1 M KNO3

C17H14N2O2 HL CAS 1229-55-6 (5216)
1-(2-Methoxyphenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	mixed	25°C	75%	U			K1=8.19 B2=15.64	1972MCb (95830)	1378

Medium: 75% acetone, 0.1 M KNO3

C17H14N2O2 HL CAS 13441-91-1 (5217)
1-(4-Methoxyphenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	mixed	25°C	75%	U			K1=7.74 B2=14.90	1972MCb (95845)	1379

Medium: 75% acetone, 0.1 M KNO3

C17H14N4 L CAS 24929-06-4 (2810)
2-(6-Benzoylpyridine)-2'-pyridylhydrazone C6H5.CO.C5H3N.N(NH2)C5H4N

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       sp  none   25°C  0.0  U          K1=8.1   B2=15.0   1974GSd (95949)1380
*****
C17H19N3           L    Antazoline      CAS 91-75-8 (3486)
2-(N-(Benzyl)-N-phenylaminomethyl)-1,4,5H-1,3-diazole, antistine;
C3H5N2.CH2.N(C6H5)CH2.C6H5
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl   KCl    25°C  0.06M U T      K1=6.38   B2=11.44  1961SMa (96266)1381
K1=7.16(0 C), 5.80(45 C); K2=5.44(0 C), 4.88(45 C)
*****
C17H21NO           L    Benadryl      CAS 58-73-1 (3492)
N,N-Dimethyl-2-(diphenylmethoxy)ethylamine;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl   KCl    25°C  0.06M U T      K1=4.02   B2=7.56   1961SMa (96373)1382
At 0 C: K1=4.26, K2=3.60; 45 C: K1=3.82, K2=3.56
*****
C17H28O8S4          H4L                      (1163)
Pentane-1,1,5,5-tetramercaptopropionic acid; CH2(CH2.CH(S.CH2.CH2.COOH)2)2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl   NaClO4 25°C  0.10M U          K1=17.9           1975PJa (96564)1383
*****
C17H38N4O3           L                      (7318)
1,4,8-Tris(2-hydroxyethyl)-11-methyl-1,4,8,11-tetraazacyclotetradecane;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       gl   R4N.X  25°C  0.10M C          K1=18.4           1997RWa (96797)1384
B(HgH-1L)=10.8
Medium: Et4NC1O4
*****
C18H15As           L                      CAS 603-32-7 (2653)
Triphenylarsine; (C6H5)3As
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg++       ISE non-aq 25°C 100% C H      K1=6.77   B2=8.97   1978ABb (96971)1385
Medium: DMSO, 0.1 M NH4ClO4; DH(K1)=-34, DH(K2)=-27 kJ mol-1
-----
Hg++       sp  alc/w  25°C  75% U    M          19660Ba (96972)1386
K(HgCl2+L=HgClL+Cl)=1.11
K(HgClL+L=HgL2+Cl)=0.83
Medium: 74.5% MeOH, 0.10 M NaClO4
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C18H15AsS L CAS 3937-40-4 (2303)
Triphenylarsine sulfide; (C6H5)3AsS

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1976GGb (96973)1387

K(HgCl2+L)=2.94

K(HgBr2+L)=3.70

K(HgI2+L)=2.90

In benzene. DH(HgCl2L)=-35.3 kJ mol⁻¹; DS=-61. DH(HgBr2L)=-25.0; DS=-11.

DH(HgI2L)=-29.5; DS=-40.

C18H15O3PS HL CAS 16704-71-5 (3365)
3-Diphenylphosphino-benzene sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE KNO3 25°C 1.0M U K1=14.3 B2=24.6 1962SBa (97108)1388

B3=29.7

B4=33.0

C18H15O3PS HL CAS 54262-24-7 (327)
4-(Diphenylphosphino)benzenesulfonic acid; (C6H5)2P.C6H4.SO3H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal NaClO4 25°C 0.10M U H K1=14.5 B2=24.8 1976HMa (97113)1389

B3=29.9

B4=33.2

DH(K1)=-86 kJ mol⁻¹, DS=-10; DH(B2)=-174, DS=-107; DH(B3)=-217, DS=-154;

DH(B4)=-236

C18H15O9AsS3 H3L CAS 103953-83-9 (326)
Tris-(3-sulfophenyl)arsine; (HO3S.C6H4)3As

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal NaClO4 25°C 0.50M U H K1=9.3 B2=14.36 1976HMa (97118)1390

B3=17.0

Calorimetry: DH(K1)=-43 kJ mol⁻¹, DS=34 J K⁻¹ mol⁻¹, DH(B2)=-73, DS=30

C18H15P L CAS 603-35-0 (621)
Triphenylphosphine; (C6H5)3P

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ EMF non-aq 25°C 100% U H K1=4.62 B2=6.21 1987HPc (97135)1391

Medium: pyridine, 0.1M Et4NClO4

Hg++ nmr non-aq -90°C 100% U M 1979AGa (97136)1392

K(HgL+A)=1.21

K(HgL+B)=1.36

K(HgL+C)=0.72

K(HgL+D)=0.22

Medium: CH₂Cl₂. A=(Ph₃P)₂(O₂CCF₃)₂, B=(Ph₃P)(O₂CCHF₂)₂, C=(Ph₃P)(O₂CCH₂F)₂, D=(Ph₃P)₂(O₂CCH₃)₂

Hg++ ISE non-aq 25°C 100% C H K1=11.06 B2=17.61 1978ABb (97137)1393

Medium: DMSO, 0.1 M NH₄ClO₄; DH(K1)=-57, DH(K2)=-51 kJ mol⁻¹

Hg++ cal non-aq 30°C 100% U M 1976AGa (97138)1394

K(HgI₂+L)=ca. 4

K(HgI₂L+L)=ca. 7

Medium: MeCN

Hg++ cal non-aq 30°C 100% U H 1976FGa (97139)1395

K(Hg(SCN)₂+L) > 5

K(Hg(SCN)₂L+L)=4.0

Medium: MeCN. DH(Hg(SCN)₂L)=-57.0 kJ mol⁻¹; DS=<-93. DH(Hg(SCN)₂L₂)=-35.3; DS=-41

C18H15PS L CAS 3878-45-3 (2301)

Triphenylphosphinesulfide; (C₆H₅)₃PS

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1976GGb (97150)1396

K(HgCl₂+L)=2.08

K(HgBr₂+L)=2.26

K(HgI₂+L)=1.93

In benzene. DH(HgCl₂L)=-12.9 kJ mol⁻¹; DS=-4. DH(HgBr₂L)=-13.8; DS=-3.

DH(HgI₂L)=-13.3; DS=-6.

C18H15PSe L CAS 3878-44-2 (2302)

Triphenylphosphine selenide; (C₆H₅)₃PSe

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1976GGb (97151)1397

K(HgI₂+L)=2.30

In benzene. DH(HgI₂L)=-45 kJ mol⁻¹; DS=-105.

C18H18N2O5S₂ L CAS 350014-32-3 (8596)

3,5,6,8,9,11-Hexahydro-2,17:12,14-dietheno-7,4,10,1,13-benzoxadithiadiazacyclopentadecine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ sp non-aq 25°C 100% C K1=5.34 B2= 8.03 2002AAa (97221)1398

Medium: CH3CN. Method: fluorescence.

C18H18N2S3 L CAS 183310-21-6 (8595)

2,5,8-Trithia[9],(2,9)-1,10-phenanthrolinophane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	C		K1=5.95 B2= 8.88	2002AAa (97236)	1399

Medium: CH3CN. Method: fluorescence.

C18H18N4 L CAS 16858-01-8 (1528)

Tris(2-pyridylmethyl)amine; (C5H4NCH2)3N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	ISE	KNO3	20°C	0.10M	C	H	K1=17.15 B2=24.10	1977AHc (97261)	1400

K(HgL(OH)+H)=7.21

DH1=-91.4 kJ mol⁻¹, DS1=16.7DH(K2)=-53.6, DS(K2)=27.6

C18H20N2OS2 L CAS 244271-40-7 (8949)

2,2'-Oxybis[N-(phenylmethyl)]-ethanethioamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	C		K1=5.8 B2=10.70	1999RPa (97322)	1401

Medium: acetonitrile.

C18H20N2O6 H4L EHPG CAS 10328-28-6 (429)

N,N'-Ethylene-bis-(2-(2'-hydroxyphenyl))glycine; (HOOCCCH(C6H4OH)NHCH2.)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	EMF	KNO3	25°C	0.10M	C	T		1985HWb (97430)	1402

K(HgHL+H)=6.68

K(Hg+HL)=21.53

Method: Hg (and glass) electrode, using Hg(II) as competitive indicator ion. Data for 10-35 C.

C18H21N5 L (7482)

2,5,8-Triaza[9]-[9](2,9)[1,10]-phenanthrolinophane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	R4N.X	25°C	0.10M	U			2004BBa (97500)	1403

B(HgLC1)=28.7

Medium: Me4NCl. Method: spectrofluorimetry.

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)

Triethylenetetraaminehexaethanoic acid;((HOOCC.H2)2N.CH2.CH2.N(CH2.COOH).CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	KNO3	25°C	0.10M	C				1987HCa (98046)	1404

K(HgL+H)=6.10
K(HgHL+H)=3.54

Method: Hg electrode.

Hg++	gl	KNO3	25°C	0.10M	U			K1=19.3 K(HgL+H)=6.03 K(HgHL+H)=3.52 K(HgH2L+H)=2.96 B(Hg3L)=26.7	1971YMb (98047)	1405
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Hg++	ISE	KNO3	25°C	0.10M	U			K1=26.8 B(Hg2L)=39.1	1970HAa (98048)	1406
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By glass electrode: K(HgL+H)=6.3, K(HgHL+H)=3.6, K(Hg2L+H)=3.6
K(Hg2HL+H)=2.7, K(Hg2L+20H)=12.8

Hg++	EMF	KNO3	25°C	0.10M	U	M			1970LAd (98049)	1407
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K(Hg2L+2NH3)=10.9
K(Hg2L+NH3)=5.6
K(Hg2L+2A)=10.1
K(Hg2L+A)=5.4

K(Hg2L+2B)=3.8, K(Hg2L+B)=2.2. A=imidazole, B=hexamethylenetetramine

Hg++	gl	NaClO4	25°C	0.10M	U			K1=25.27 B2=33.67 K(HgL+H)=6.55 K(HgHL+H)=3.30 K(Hg2L+OH)=6.1 K(Hg2LOH+OH)=5.9	1966SCb (98050)	1408
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C18H30N6O2 L (1607)
6-(2-(2Pyridyl)ethyl)-1,4,8,11,14-Pentaazacyclohexane-5,7-dione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	vlt	NaNO3	25°C	0.20M	C	M			1989KKa (98112)	1409
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B(HgH-1L)=11.46

C18H32N4O8 H4L TETA CAS 60239-22-7 (1019)
1,4,8,11-Tetraazacyclotetradecane-1,4,8,11-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	vlt	KNO3	25°C	0.20M	C	H		K1=25.7	1994KOb (98207)	1410
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DH(K1)=-123 kJ mol⁻¹, DS=101 J K⁻¹ mol⁻¹

Hg++	gl	NaNO3	25°C	0.20M	C			K1=25.71	1991KKa (98208)	1411
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C18H33P L CAS 2622-14-2 (169)

Tri-(cyclohexyl)phosphine; (C₆H₁₁)₃P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	non-aq	25°C	100%	U	H		K1=11.8 B2=18.0 B3=20.5	1987HPc (98309)	1412

Medium: pyridine, 0.1M Et₄NClO₄

Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (98310)	1413
HgI ₂ +L in benzene. DH(K ₁)=-102; DH(K ₂)=-58 kJ mol ⁻¹										

Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (98311)	1414
HgBr ₂ +L in benzene. DH(K ₁)=-111; DH(K ₂)=-60 kJ mol ⁻¹										

Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (98312)	1415
HgCl ₂ +L in benzene. DH(K ₁)=-119; DH(K ₂)=-59 kJ mol ⁻¹										

C₁₈H₃₆N₂O₆ L Cryptand 2,2,2 CAS 23978-09-8 (514)
1,10-Diaza-4,7,13,16,21,24-hexaoxabicyclo[8.8.8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	R4N.X	22°C	0.03M	C	I		K1=17.07	1991PSa (98580)	1416
Medium: 0.025 M Et ₄ NClO ₄ . Method: differential pulse polarography. Data for 15-75% w/w CH ₃ CN/H ₂ O, 0.025 M Et ₄ NClO ₄ .										

Hg++	EMF	R4N.X	25°C	0.05M	C			K1=18.2	1979BLb (98581)	1417
Method: Ag electrode; competition with Ag+. Medium: 0.05 M Me ₄ NClO ₄ .										

Hg++	gl	R4N.X	25°C	0.10M	C	H		K1=18.2	1975ANa (98582)	1418
Medium: Me ₄ NNO ₃ . DH(K ₁)=-66.7 kJ mol ⁻¹ , DS=125										

Hg++	gl	R4N.X	25°C	0.05M	C			K1=18.2	1975LSc (98583)	1419

C ₁₈ H ₄₀ N ₄ O ₄ L CAS 89066-60-2 (867)										
N,N',N'',N'''-Tetrakis(2-hydroxyethyl)-1,4,8,11-tetraazacyclotetradecane;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO ₃	25°C	0.10M	U			K1=17.94	1984MMc (98922)	1420

C ₁₉ H ₂₀ N ₂ S ₂ L CAS 403819-60-3 (8597)										
3,6,7,8,9,11-Hexahydro-2,17:12,14-Dietheno-5H-4,10,1,13-benzodithiadiazacyclopentadecine;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	C			K1=5.56	2002AAa (99302)	1421
Medium: CH ₃ CN. Method: fluorescence.										

C19H31N5O2 L (1614)
3-(2-Phenylethyl)-1,5,8,11,14-pentaazacyclohexadecane-2,4-dione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	vlt	NaNO3	25°C	0.20M	C	M			1989KKa (99451)	1422
B(HgH-1L)=10.03										

C19H39N3O5 L CAS 60598-00-7 (1537)
4-Methyl-1,4,10-triaza-7,13,16,21,24-pentaoxa-bicyclo[8,8,8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	R4N.X	25°C	0.10M	U			K1=21.7 K(Hg+HL)=16.7	1978LMa (99491)	1423
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C20H13N7 HL CAS 30842-84-3 (5288)
1,5-Bis(8-quinolyl)-3-cyanoformazan;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	sp	NaClO4	25°C	0.10M	U				1971BSf (99589)	1424
B(HgHL2)=32.0										

C20H14N2O HL (5291)
1-(1-Naphthylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	mixed	25°C	75%	U			K1=6.70 B2=12.72	1972MCb (99600)	1425
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Medium: 75% acetone, 0.1 M KNO3

C20H14N2O HL CAS 2653-64-7 (5292)
1-(2-Naphthylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	gl	mixed	25°C	75%	U			K1=7.32 B2=13.67	1972MCb (99615)	1426
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Medium: 75% acetone, 0.1 M KNO3

C20H16N4O6S H3L Zincon CAS 135-52-4 (990)
2-Carboxy-2'-hydroxyl-5'-sulfoformazylbenzene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++	EMF	NaNO3	20°C	0.10M	U			B2=14.59 K(Hg+HL)=8.03 K(HgHL+HL)=6.56	1970VEb (99803)	1427
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C20H24O6 L DiBz-18-Crown-6 CAS 14187-32-7 (604)

2,3:11,12-Dibenzo-1,4,7,10,13,16-hexaoxacyclooctadeca-2,11-diene

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	nmr	non-aq	RT	0.0	C			1991LBa (100123)	1428
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K(Hg(CN)₂+L)=>4

Method: 1H nmr. Medium: acetone/CDC13 1:0.8.

C₂₀H₂₆N₆ L CAS 303955-27-3 (9162)

5-Aminoethyl-2,5,8-triaza-[9]-10,23-phenanthrolineophane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	R4N.X	25°C	0.10M	U		K ₁ =30.28	2004BBa (100342)	1429
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K(HgL+H+Cl)=7.79

K(HgHLC1+H+Cl)=4.49

Medium: Me₄NCl. Combined with spectrofluorimetric measurements.

C₂₀H₃₅N₅O₁₀ H5L (6545)

1,4,7,10,13-Pentaazacyclopentadecane-N,N',N'',N''',N''''-pentaethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	vlt	KNO ₃	25°C	0.20M	C	H	K ₁ =27.7	1994KOb (100537)	1430
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DH(K₁)=-128 kJ mol⁻¹, DS=101 J K⁻¹ mol⁻¹

Hg++	gl	NaNO ₃	25°C	0.20M	C		K ₁ =27.76	1991KKa (100538)	1431
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C₂₀H₃₆O₆ L DiCy-18-crown-6 CAS 16069-36-6 (1653)

2,3:11,12-Dicyclohexyl-1,4,7,10,13,16-hexaoxacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	con	non-aq	25°C	100%	C	TIH	K ₁ =1.8	2001RKa (100648)	1432
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Medium: DMF. Data for 15-55 C. Also data for 25-75% mol% DMF/AN.

DH(K₁)=-50 kJ mol⁻¹, DS(K₁)=-205 J K⁻¹ mol⁻¹.

Hg++	cal	oth/un	25°C	0.10M	U	H		1976ITb (100649)	1433
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K₁=2.75 (cis-syn-cis isomer)

K₁=2.60 (cis-anti-cis isomer)

DH(K₁)=-3.0 kJ mol⁻¹ for the cis-syn-cis isomer, -10.7 for cis-anti-cis.

C₂₀H₃₉N₅O₂ HL CAS 333309-52-7 (8662)

16-Aminodocosahydro-16-methyl-dibenzo[b,i][1,4,8,11]tetraazacyclotetradecine-7-carboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg++	gl	KCl	25°C	0.5M	U		K ₁ =11.55	2002WHa (100769)	1434
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K(HgL+H)=5.8

$$K(\text{HgL}=\text{HgH}-1\text{L}+\text{H})=-10.7$$

Data for the trans isomer. For the cis-isomer $K_1=11.2$, $K(\text{HgL}+\text{H})=5.2$

$$K(\text{HgL}=\text{HgH}-1\text{L}+\text{H})=-7.5$$

C20H42N4O4 L CAS 39678-14-3 (1543)

4,7-Dimethyl-1,4,7,10-tetraaza-13,16,21,24-tetraoxa-bicyclohexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	R4N.X	25°C	0.10M	U		$K_1=24.9$ $K(\text{Hg}+\text{HL})=18.6$	1978LMa (100887)	1435

K_1 is corrected for HgClx complexes

C21H14N4O2 HL CAS 194480-84-7 (8524)

2-Hydroxy-1-naphthalenecarboxaldehyde benzofuro[2,3-d]pyrimidin-4-ylhydrazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	diox/w	30°C	10%	U		$K_1=6.673$	1997HVa (101035)	1436

Medium: 10% v/v dioxane/H₂O, 0.10 M NaClO₄.

C21H24N4 L (931)

Tris((6-methyl-2-pyridyl)methyl)-amine; (CH₃.C₅H₃N.CH₂)₃N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	20°C	0.10M	C	H	$K_1=15.55$ $K(\text{HgL}(\text{OH})+\text{H})=5.76$	1977AHc (101246)	1437

Calorimetry: $\text{DH}_1=-64.9$ kJ mol⁻¹, $\text{DS}_1=72.8$

C22H17P L (2293)

1,2,5-Triphenylphosphole;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H	$K(\text{HgCl}_2+\text{L})=2.18$	1976GGa (101589)	1438

In benzene. $\text{DH} = -67$ kJ mol⁻¹.

Hg++	cal	non-aq	30°C	100%	U	H	$K(\text{HgBr}_2+\text{L})=2.20$	1976GGa (101590)	1439
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In benzene. $\text{DH} = -61$ kJ mol⁻¹.

C22H24N2O8 H2L Tetracycline CAS 60-54-8 (2201)

Tetracycline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	NaN03	25°C	0.10M	C		$K_1=6.5$	1992GAa (101817)	1440

C22H24N2O8 H4L CAS 91044-24-5 (1920)
meso-1,2-Diphenyl-1,2-diaminoethane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U			K1=17.50	1989SLa (101840)	1441

C22H24N2O8 H4L CAS 91044-25-6 (1921)
rac-1,2-Diphenyl-1,2-diaminoethane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	KNO3	20°C	0.10M	U			K1=23.04	1989SLa (101857)	1442

C22H24N2O10 H4L CAS 132796-79-3 (8113)
1,2-Bis(2-aminophenoxy)ethane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	EMF	KNO3	25°C	0.10M	C T H			K1=18.75 K(HgL+H)=2.94	1990HLA (101897)	1443

Method: Hg indicator electrode. Data for 15-35 C.
DH(K1)=-102.1 kJ mol⁻¹, DS(K1)=16.4 J K⁻¹ mol⁻¹.

C22H37N5O14 H7L CAS 3234-59-1 (2425)
Tetraethylenepentamineheptaethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO3	25°C	0.10M	C			K1=27.0 K(HgL+H)=12.0 K(HgH2L+H)=3.3 K(HgHL+H)=5.0 K(HgH3L+H)=2.3	1999LLa (102329)	1444

K(HgL+Hg)=20.6; K(Hg2L+H)=4.1; K(Hg2HL+H)=2.0

C22H48N4O4 L (7292)
N,N',N'',N'''-Tetrakis(3-hydroxypropyl)-1,4,8,11-tetraazacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	R4N.X	25°C	0.10M	C			K1=17.3	1996DTa (102469)	1445

Medium: Et4ClO4

C22H48N6O2 L CAS 39678-22-3 (1542)
4,7,13,16-Tetramethyl-1,4,7,10,13,16-hexaaza-21,24-dioxabicyclohexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	R4N.X	25°C	0.10M	U			K1=26.1	1978LMA (102486)	1446

$$K(\text{Hg}+\text{HL})=20.6$$

K1 is corrected for HgClx complexes

C23H25N05S L CAS 464185-98-6 (9292)

4'-[(2-Benzothiazole)ethenyl]-2:3-benzo-15-crown-5;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	non-aq	20°C	100%	C			K1=5.6 B(Hg2L)=9.4	2003FFa (102691)	1447

Medium: CH3CN.

C24H32O8 L DiBz-24-Crown-8 CAS 14174-09-5 (580)

2,3:14,15-Dibenzo-1,4,7,10,13,16,19,22-octaaxacyclotetracos-2,14-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	nmr	non-aq	27°C	100%	C	I M		K(Hg(CN)2+L=HgL(CN)2)=-0.4	2003JLa (103129)	1448

Medium: DMSO. Method:199Hg nmr. In DMF, K=0.6.

C24H42N6O12 H6L (6546)

1,4,7,10,13,16-Hexaazacyclooctadecane-N,N',N'',N''',N''',N''''-hexaethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	vlt	KN03	25°C	0.20M	C	H		K1=25.2	1994K0b (103376)	1449
DH(K1)=-121 kJ mol-1, DS=77 J K-1 mol-1										
Hg++	gl	NaNO3	25°C	0.20M	C			K1=25.27 K(Hg+H2L)=21.26	1991KKa (103377)	1450

C25H19N5O10 L CAS 611183-31-4 (9129)

8,9,18,19-Tetrahydro-3,23-dinitro-15,11-nitrilodibenzo[1,15,4,12]dioxadiazacyclohen eicosinetetron

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	nmr	non-aq	25°C	100%	C	M		K(Hg(CN)2+L=HgL(CN)2)=1.6	2003CAa (103599)	1451

Medium: acetonitrile. Method: 1H nmr.

C25H22As2 L CAS 21892-63-7 (2299)

Methylenebis(diphenylarsine); (C6H5)2As.CH2.As(C6H5)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H		K(HgCl2+L)=3.50 K(HgBr2+L)=2.99	1976GGb (103613)	1452

$$K(\text{HgI}_2+\text{L})=2.03$$

In benzene. $\text{DH}(\text{HgCl}_2\text{L})=-50.0 \text{ kJ mol}^{-1}$; $\text{DS}=-98$. $\text{DH}(\text{HgBr}_2\text{L})=-53.0$; $\text{DS}=-128$.
 $\text{DH}(\text{HgI}_2\text{L})=-52.0$; $\text{DS}=-133$.

C25H22As2S2 L CAS 69289-43-6 (2305)
 Methylenebis(diphenylarsine sulfide); $(\text{C}_6\text{H}_5)_2\text{AsS.CH}_2.\text{As}(:\text{S})(\text{C}_6\text{H}_5)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H			1976GGb (103614)	1453

$$K(\text{HgCl}_2+\text{L})=3.60$$

$$K(\text{HgBr}_2+\text{L})=3.07$$

In benzene. $\text{DH}(\text{HgCl}_2\text{L})=-44.9 \text{ kJ mol}^{-1}$; $\text{DS}=-79$. $\text{DH}(\text{HgBr}_2\text{L})=-52.5$; $\text{DS}=-114$.

C25H22O2P2 L CAS 207-21-8 (2099)
 Methylenebis(diphenylphosphine oxide); $\text{Ph}_2\text{P}(\text{O})\text{CH}_2\text{P}(\text{O})\text{Ph}_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	ISE	mixed	25°C	82%	U			K1=18.89 B2=21.82 B3=24.53	1979MTd (103628)	1454

Medium: 82% DMF/H2O

C25H22P2 L CAS 2071-20-7 (2294)
 Methylenebis(diphenylphosphine); $(\text{C}_6\text{H}_5)_2\text{P.CH}_2.\text{P}(\text{C}_6\text{H}_5)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (103646)	1455

In benzene. $\text{DH}(\text{HgCl}_2+\text{L})=-113 \text{ kJ mol}^{-1}$; $\text{DH}(2\text{HgCl}_2+\text{L})=-80 \text{ kJ mol}^{-1}$.

Hg++	cal	non-aq	30°C	100%	U	H			1976GGa (103647)	1456
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In benzene. $\text{DH}(\text{HgBr}_2+\text{L})=-113 \text{ kJ mol}^{-1}$; $\text{DH}(2\text{HgBr}_2+\text{L})=-69 \text{ kJ mol}^{-1}$.

C25H22P2S2 L CAS 14633-92-2 (2304)
 Methylenebis(diphenylphosphine sulfide); $(\text{C}_6\text{H}_5)_2\text{PS.CH}_2.\text{P}(:\text{S})(\text{C}_6\text{H}_5)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H			1976GGb (103648)	1457

$$K(\text{HgCl}_2+\text{L})=3.50$$

$$K(\text{HgBr}_2+\text{L})=3.50$$

$$K(\text{HgI}_2+\text{L})=3.28$$

In benzene. $\text{DH}(\text{HgCl}_2\text{L})=-39.7 \text{ kJ mol}^{-1}$; $\text{DS}=-64$. $\text{DH}(\text{HgBr}_2\text{L})=-39.0$; $\text{DS}=-62$.
 $\text{DH}(\text{HgI}_2\text{L})=-20.5$; $\text{DS}=-2$.

C26H24As2 L (2300)
 Ethane-1,2-diylbis(diphenylarsine); $(\text{C}_6\text{H}_5)_2\text{As.CH}_2.\text{CH}_2.\text{As}(\text{C}_6\text{H}_5)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ cal non-aq 30°C 100% U H 1976GGB (103892)1458

K(HgCl₂+L)=2.82

K(HgBr₂+L)=3.26

K(HgI₂+L)=2.54

In benzene. DH(HgCl₂L)=-73.0 kJ mol⁻¹; DS=-188. DH(HgBr₂L)=-72.6; DS=-177.

DH(HgI₂L)=-57.0; DS=-140.

C26H24P2 L CAS 28240-60-0 (2280)

Ethylenebis(diphenylphosphine); (C₆H₅)₂P.CH₂.CH₂.P(C₆H₅)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ cal non-aq 30°C 100% U H 1974FGa (103930)1459

K(HgCl₂+L) > 7

K(HgBr₂+L)=7

K(HgI₂+L)=6.30

In benzene. For HgCl₂, DH=-130 kJ mol⁻¹.

For HgBr₂, DH=-130.1; DS=-295. For HgI₂, DH=-118.8; DS=-270.

C26H25N09S H4L Semi-Xylenol O (426)

3-(N,N-Di(carboxymethyl)aminomethyl)-2-cresolsulfonephthalein;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ gl KNO₃ 25°C 0.10M U 1981MUa (103945)1460

K(Hg+HL)=6.8

C26H28N6 L CAS 16858-02-9 (933)

N,N,N',N'-Tetrakis-(2-pyridylmethyl)-diaminoethane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ISE NaNO₃ 25°C 0.10M C K1=25.00 1995CCb (104004)1461

Method: mercury pool electrode

Hg++ gl KNO₃ 20°C 0.10M C H K1=25.05 1977AHc (104005)1462

Calorimetry: DH1=-124.5 kJ mol⁻¹, DS1=55.6

C26H38N204 L CAS 80757-23-9 (2450)

N,N'-Bis(benzyl)-1,10-diaza-4,7,13,16-tetraoxacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg++ ix none 25°C 0.0 U K1=17.85 1988IBa (104185)1463

Ligand covalently attached to silica gel

C27H26P2 L CAS 6372-42-4 (2295)

Propane-1,3-diylbis(diphenylphosphine); (C₆H₅)₂P.CH₂.CH₂.CH₂.P(C₆H₅)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	cal	non-aq	30°C	100%	U	H		1976GGa (104412)	1464
In benzene. DH(2HgX2+L) = -93 (X=Cl); -78 (X=Br) kJ mol-1.									

C27H41N3O4		L					CAS 262610-61-7 (7222)		
3,4:5,6-Dibenzo-14-methyl-4',4''-bis(dimethylamino)1,8,11,17-tetraoxa-14-azacyclononadecan3,5diene									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	mixed	25°C	70%	C			2000CMa (104592)	1465
							B(HgHL)=12.90		
							B(HgH-1L)=5.55		
							B(HgH-2L)=-3.50		

Medium: 70% v/v dioxane/H2O, 0.10 M KNO3.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
C28H35N3O6		L					CAS 114880-42-1 (7377)		
3-(p-13-Aza-1,4,7,10-tetroxacyclopentadecan-13ylstyryl)-7-dimethylamino-1,4-benzoxazin-2-one;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	RT	100%	C		K1=6.15	1998ABc (104762)	1466
Medium: acetonitrile. Method: fluorescence spectroscopy.									

C28H36Fe2N4		L					CAS 174322-18-0 (7771)		
1,1':1',1'''-Bis[1,2-ethanediylbis(iminomethylene)]bis[ferrocene];									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	mixed	25°C	70%	C		K1=6.47	1998LBa (104772)	1467
							B(HgHL)=13.31		
							B(HgH2L)=19.50		
							K(HgL+H)=6.84		
							K(HgHL+H)=6.19		

Medium: 70% (v/v) THF/H2O, 0.1 M KCl. K(Pb+L+OH)=13.38, B(HgH-1L)=-1.60.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
C29H45N3O5		L					CAS 262610-63-9 (7249)		
3,4:5,6-Dibenzo-14-methyl-4',4''-bis(dimethylamino)-1,8,11,17,20-pentaoxa14azacyclodocosan3,5diene									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	mixed	25°C	70%	C		K1=8.15	2000CMa (105154)	1468
							B(HgH-1L)=4.17		
							B(HgH-2L)=-4.59		

Medium: 70% v/v dioxane/H2O, 0.10 M KNO3.

C32H44N10O4 L CAS 702699-42-1 (9126)
2,9-Di[4-(1,4,7,10-tetraazacyclotridecane-11,13,-dione)methyl]-1,10-phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	gl	KN03	25°C	0.10M	U		B(HgH3L)=28.38 B(Hg2L)=16.20 B(Hg3H-2L)=12.30 B(Hg3H-3L)=1.18	2004GLa	(105773)1469

$$B(\text{Hg}3\text{H}-4\text{L}) = -10.08.$$

C34H54O8	H2L	Lasalocid	CAS 25999-20-6 (2335)
Lasalocid acid;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	alc/w	25°C	100%	M		K1=8.5	B2=15.7	1994MPc	(106136)1470
Medium: MeOH										

C35H36N6 L CAS 750635-82-6 (9186)
2,9-[2,5,8-Triaza-5-(N-anthracene-9-methylamino)ethyl]-[9]-1,10-phenantrolinophane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	mixed	25°C	50%	C			K1=14.4 K(HgL+H)=4.2	2004BBd (106187)	1471

Medium: 50% v/v CH₃CN/H₂O, 0.1 M Me₄NCI.

C36H38N4O8 H6L CAS 531-14-6 (7709)
3,8,13,18-Tetramethylporphine-2,7,12,17-tetrapropanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++ Medium not specified.	sp	oth/un	20°C	0.5M	U		K1=3.42	1998GBb (106288)	1472

C36H60N8O8 L CAS 121925-84-6 (7152)
Cyclo(Gly-eLL-Gly)2 (eLL=N,N'-ethylene-bridged (S)-leucyl-(S)-leucine

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++ Medium: MeCN	sp	non-aq	25°C	100%	U		K1=3.71	1994MKa (106455)	1473

C36H62O11 HL Monensin CAS 17090-79-8 (737)
Monensin, 1,6-dioxaspiro[4,5]decane derivative;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg++ gl alc/w 25°C 100% M K1=9.9 B2=18.9 1994MPc (106500)1474
Medium: MeOH

C37H44N2O13S H6L MeThymol Blue (428)
3,3'-Bis(N,N-di(carboxymethyl)aminomethyl)thymolsulfonephthalein;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	oth/un	25°C	0.10M	C	M		K(HgL+SCN)=5.76 K(HgL+Br)=5.46 K(HgL+I)=8.32	1983ZJa (106602)	1475

Medium: 0.46M NaH2PO4, 0.26M NaOH, PH=7.0

Hg++	vlt	KN03	25°C	0.10M	C			K1eff=10.3	1977RSb (106603)	1476
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Method: a.c. polarography. Medium, pH 3.97.

Measured in competition with edta.

Hg++	sp	oth/un	?	?	U			K(Hg+H3L)(?)=6.05	1971ANb (106604)	1477
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C40H44N4O2S4 CAS 244271-42-9 (8951)
4,7,13,16-Tetrakis(phenylmethyl)-1,10-dioxa-4,7,13,16-tetraazacyclooctadecane-3,8,12,17-tetrathi

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	C			K1=7.3 B2=12.50 B(Hg2L)=14.0	1999RPa (106759)	1478

Medium: acetonitrile.

C40H46N4O2S4 L CAS 244271-41-8 (8950)
Dimethyl-N,N',6,9-tetrakis(phenylmethyl)-5,10-dithione-3,12-dioxa-6,9-diazatetradecanedithioamide

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	C			K1=9.4 B2=15.10 B(Hg2L)=16.7	1999RPa (106769)	1479

Medium: acetonitrile.

C43H58N4O12 H3L Rifampicin CAS 13292-46-1 (8977)
3-[[[4-Methyl-1-piperazinyl]imino]methyl]rifamycin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	alc/w	30°C	50%	C	T H		K(Hg+H2L)=6.42 K(HgH2L+H2L)=4.73	2001SKd (107019)	1480

Medium: 50% v/v MeOH/H₂O, 0.05 M KCl. DH(Hg+H₂L)=-46.39 kJ mol⁻¹, DS=-30.0 J K⁻¹ mol⁻¹; DH(HgH₂L+H₂L)=-36.50, DS=-26.7. Also data for 35 and 40 C.

C44H₂₂N₄O₁₂Br₈S₄ H₆L CAS 176173-80-1 (6959)
2,3,7,8,12,13,17,18-Octabromo-5,10,15,20-tetrakis(4-sulfonatophenyl)porphyrin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	NaNO ₃	25°C	0.1M	C				1996TNa (107037)	1481
									K(Hg+H ₂ L=HgL+2H)=0.12	

C44H₃₀N₄ H₂L Tetraphenylpor. CAS 917-23-7 (1781)
5,10,15,20-Tetraphenyl-21H,23H-porphine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	U	M			1970KHa (107071)	1482
									K(HgL+A)=2.92	
									K(HgL+B)=1.62	
									K(HgL+C)=1.41	
									K(HgL+py)=1.21	

Medium: benzene. K(HgL+D)=0.22, A=4-aminopyridine, B=4-methylpyridine, C=3-methylpyridine, D=4-cyanopyridine

C44H₃₈N₈ H₂L CAS 48242-70-2 (6629)
5,10,15,20-Tetrakis(1-methylpyridinium-4-yl)porphine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	sp	oth/un	25°C	1.90M	U	I M			1992RHa (107105)	1483
									K(2Hg+H ₂ L=Hg ₂ L+2H)=4.58	
									K(2Hg+H ₃ L=Hg ₂ L+3H)=1.58	
									K(2Hg+H ₄ L=Hg ₂ L+4H)=0.62	
									Keff(HgA ₂ +H ₂ L=HgA ₂ L+2H)=-5.64	

Medium: LiNO₃. A=OH. Keff in 0.1 M LiNO₃/2x10⁻³ M 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid, pH 7.5. Data also for 2-pyridyl- and 4-quinolyl-

C47H₅₇N₉S₃ L CAS 529487-44-3 (8769)
Ethane-1,1,1-tris(2-thia-5-aza-hexane-6-(3-(3'-methyl)dipyridine));

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg++	gl	KNO ₃	25°C	0.10M	C	M			2002KGa (107324)	1484
									K(FeL+Hg)=4.60	
									K(FeL+Hg+H)=9.30	
									K(FeL+Hg+OH)=8.84	

C48H₅₈N₂O₄S₂ L CAS 403518-26-3 (8260)
11,23-Diprop-2-enyl-25,27-bis(dimethylaminothiocarbonylmethoxy)-26,28-dipropoxycalix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	non-aq	25°C	100%	C		K1=7.8	2001ACa (107394)	1485
Medium: acetonitrile.									

C69H102N4O9		L					CAS 116352-85-3	(9286)	
para-t-Butyldihomooxacalix[4]arene tetra(diethyl)amide;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	sp	alc/w	25°C	100%	C		K1=3.4	2004MFa (107833)	1486
Medium: MeOH, 0.01 M Et4NCl.									

Polymer					CPA		CAS 11075-17-5	(1758)	
Carboxypeptidase A									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	oth	NaCl	4°C	1.0M	U			1961VWa (108113)	1487
							K(Hg+HxL=HgHyL+(x-y)H)=21.0		
Medium: 0.05 M tris buffer pH 8									

Polymer							(4203)		
Procarboxypeptidase;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg++	oth	NaCl	4°C	1.0M	U		K1=18.3	1967PVa (108397)	1488
Method: dialysis									

e-		HL			Electron		(442)		
Electron;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	EMF	non-aq	25°C	100%	C	I		1980APa (534)	1489
							E0(Hg(l)/Hg2++)=-342 mV		
Medium: DMSO, 1 M NH4ClO4. E0 referred to E0(aq)=0 for the Ag(s)/Ag+ elect.									

Hg2++	EMF	non-aq	25°C	100%	U			1972FDa (535)	1490
							K=23.12(684mV)		
Medium: DMSO, 0.1 M LiClO4; K: Hg2++ + 2e=2Hg(l)									

Hg2++	EMF	none	25°C	0.00	U	T		1971DFa (536)	1491
							K=9.062(268.05mV) or 9.056		
K: Hg2Cl2(s)+2e=2Hg(l)+2Cl-. 70 C: K=7.330(249.55mV) or 7.304. 100 C: K=6.293 or 6.260. 200 C: K=3.075 or 3.042 depending on choice for Eo(AgCl,Ag)									

Hg2++	EMF	non-aq	25°C	100%	U			1971NSc (537)	1492

Medium: formamide; K: $\text{Hg}_2\text{Cl}_2(\text{s}) + 2\text{e} = 2\text{Hg}(\text{l}) + 2\text{Cl}^-$

$$K = 26.67 (788.9 \text{ mV})$$
 $K=0.85$
$$K=9.067(268.18\text{mV})$$
$$K = 8.864 (266.59 \text{ mV})$$
$$K=4.706(139.21\text{mV})$$
$$K=4.713(139.41\text{mV})$$
$$K=4.436(131.20\text{mV})$$
$$K=4.113(121.65\text{mV})$$
$$K=3.061(90.55\text{mV})$$
$$K=0.222(6.57\text{mV})$$

Medium: 90% w/w i-propanol/H₂O; K: Hg₂Br₂+2e=2Hg(l)+2Br⁻.
K=0.746(21.32mV, 15 °C), -0.308(-9.42mV, 35 °C)

Hg2++	EMF mixed	25°C	99%	U T	1970SMc	(548)1503
=-3.058(-90.45mV)						
Medium: 99% w/w i-propanol/H2O; K: Hg2Br2+2e=2Hg(l)+2Br-.						
K=-2.610(-74.60mV,15 C), -3.569(-109.10mV,35 C)						

Hg2++	EMF none	25°C	0.00	U T	1970SPc	(549)1504
K=20.74(613.5mV)						
K: Hg2SO4(s)+2e=2Hg(l)+SO4--. K=22.81(629.4mV,5 C), 21.74(621.4mV,15 C),						
19.78(604.6mV,35 C)						

Hg2++	EMF mixed	25°C	20%	U I	1969SMb	(550)1505
K=8.547(252.80mV)						
Medium: 20% w/w propanol/H2O; K: Hg2Cl2(s)+2e=2Hg(l)+2Cl-. K=9.082(268.65mV,						
w=0), 7.929(234.54mV,w=40), 6.942(205.34mV,w=60), 2.543(75.22mV,w=95)						

Hg2++	sol none	25°C	0.0	U IH	1968SVa	(551)1506
K(Hg(l) = Hg(aq))=-6.55						
DH=64.5 kJ mol-1. Also K values in 16 solvents						

Hg2++	EMF none	25°C	0.0	U	1967CDa	(552)1507
K=9.059 to 9.062						
K: Hg2Cl2(s)+2e=2Hg(l)+2Cl						

Hg2++	EMF non-aq	25°C	100%	U	1967RPe	(553)1508
K=8.287, 245.1 mV						
Medium: H2NCHO. K: Hg2Cl2(s) + 2e = 2Hg(l) + 2Cl-						

Hg2++	EMF none	15°C	0.0	U TI	1966SFa	(554)1509
K=4.921, 140.67 mV						
K: Hg2Br2(s) + 2e = 2Hg(l) + 2Br-. K=4.713(25 C), 4.501(35 C). In MeOH:						
K=-0.375(15 C), -0.701(25 C), -1.024(35 C). Also MeOH/H2O mixtures						

Hg2++	EMF none	25°C	0.0	U	1965CDc	(555)1510
K=20.71; 6123mV.						
K: Hg2SO4(s) + 2e = 2Hg(l) + SO4--						

Hg2++	EMF none	25°C	0.0	U I	1965SFb	(556)1511
K=20.81, 615.5 mV						
Medium: M2SO4, corr 0. K: Hg2SO4 + 2e = 2Hg(l) + SO4--						

Hg2++	EMF none	15°C	0.0	U T	1965SFb	(557)1512
K=9.476, 270.88 mV						
K: Hg2Cl2(s) + 2e = 2Hg(l) + 2Cl-. K=9.068(25 C), 8.663(35 C)						

Hg2++	EMF mixed	25°C	95%	U	1965SWa	(558)1513
K=-0.42, -12.5 mV						
Medium: 95% Me2CO. K: Hg2Cl2(s) + 2e = 2Hg((l) + 2Cl-. Also Me2CO/H20 mix.						

Hg2++	oth none	0°C	0.0	M T	1964ACb	(559)1514

K=10.092, 273.5 mV

K=9.066(25 C;268.16 mV), K=6.951(80 C;243.52 mV), K=6.211(100 C;229.92 mV) K: Hg ₂ Cl ₂ (s)+2e=2Hg(l)+2Cl									
Hg ₂ ++	EMF none	25°C	0.0	U	I		1964SUa	(560)	1515
K=4.713; 139.40mV. K: Hg ₂ Br ₂ (s) + 2e = 2Hg(l) + 2Br-. In 95% EtOH: K=0.784. Also EtOH/H ₂ O mix.									
Hg ₂ ++	EMF none	5°C	0.0	U	T H		1963GHb	(561)	1516
K=9.890, 272.9 mV K=9.270(20 C;269.57 mV), K=8.865(30 C;266.60 mV), K=8.467(40 C;263.0 mV), K=8.271(45 C;261.04 mV). K:Hg ₂ Cl ₂ (s)+2e=2Hg(l)+2Cl. DH=-34.4 kJ mol ⁻¹ , DS=-29									
Hg ₂ ++	EMF none	5°C	0.0	U	T H		1963GHb	(562)	1517
K=5.110, 141.0 mV K=4.812(20 C;139.93 mV), K=4.707(25 C;139.23 mV), K=4.602(30 C;138.39 mV), K=4.285(45 C;135.24 mV). DH=-17.8 kJ mol ⁻¹ , DS=-15. K:Hg ₂ Br ₂ (s)+2e=2Hg(l)+2Br									
Hg ₂ ++	sol oth/un	25°C	dil	U	T		1962CTc	(563)	1518
K(Hg(l)=Hg(aq))=-6.50 Metal: Hg(0) K=-6.26(35 C), -6.05(50 C), -5.97(65 C), -5.89(80 C), -5.78(90 C)									
Hg ₂ ++	EMF NaClO ₄	25°C	2.0M	U	I		1962ZSa	(564)	1519
K=26.07(770.94 mV) K: Hg ₂ +2e=2Hg(liq)). In 2 M HClO ₄ K=26.04(770.32 mV)									
Hg ₂ ++	EMF none	35°C	0.0	U			1961SJa	(565)	1520
K=19.71(602.5 mV) K: Hg ₂ SO ₄ (s)+2e=2Hg(liq)+SO ₄									
Hg ₂ ++	EMF none	25°C	0.0	U	H		1960BDb	(566)	1521
K=20.82(615.81 mV) K: Hg ₂ SO ₄ (s)+2e=2Hg(liq)+SO ₄ . DH(K)=-168.4 kJ mol ⁻¹									
Hg ₂ ++	EMF none	25°C	0.00	U			1959BLa	(567)	1522
K=20.69(611.9mV) K: Hg ₂ SO ₄ (s)+2e=2Hg(l)+SO ₄ --									
Hg ₂ ++	dis none	25°C	0.0	U			1957MVa	(568)	1523
K(Hg ₂ =Hg(II)+Hg(aq))=-8.26 By solubility K(Hg(liq)=Hg(aq))=-6.52									
Hg ₂ ++	EMF none	25°C	0.0	U			1956HSa	(569)	1524
K=26.79(792.5 mV) K:Hg ₂ +2e=2Hg(liq)									
Hg ₂ ++	EMF none	25°C	0.0	U			1937BVa	(570)	1525
K=-1.37(-40.5 mV) K: Hg ₂ I ₂ (s)+2e=2Hg(liq)+2I									

Hg2++ EMF none 25°C 0.0 U T 1935HHa (571)1526

K=20.81(615.15 mV)

K: Hg2SO4(s)+2e=2Hg(liq)+SO4. K=23.44(0 C;634.95 mV),21.80(15 C;623.07 mV),
18.56(50 C;594.87 mV),17.75(60 C;586.59 mV)

Br- HL Bromide CAS 10035-10-6 (19)

Bromide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE non-aq 25°C 100% U 1973CCa (2008)1527

Kso(Hg2L2(s)=Hg2+2L)=-37.4

Medium: MeCN. Method: Ag electrode

Hg2++ EMF KNO3 177°C 100% U T 1969BMF (2009)1528

Kso(Hg2L2(s)=Hg2+2L)=-16.69

Medium: (Li,K)NO3. log Kso=3.666-9158.7/T m units at 137-207 C

Hg2++ ISE non-aq 180°C 100% U T 1966FBa (2010)1529

Kso=5.28-9785/T(T K)

Medium: (Li/K)NO3 eutectic. Kso=-16.46(450 K) m units

Hg2++ ISE NaCl04 40°C 0.60M U T 1963HIa (2011)1530

Kso=-20.17

Medium:0.5 M NaCl04,0.1 HCl04. Kso=-22.85(7 C)

Hg2++ ISE NaCl04 25°C 0.50M U I 1948BJb (2012)1531

Kso(Hg2L2)=-21.285

At I=0 corr. Kso=-22.24

Hg2++ ISE none 27°C 0.0 U TIH 1929BRa (2013)1532

Kso(Hg2L2)=-22.16

Method: Hg electrode. Kso=-23.26(10.8 C), -23.00(14.9 C), -22.41(19.2 C)

At I=0 corr. Kso=-22.26; DH(so)=39.0 kJ mol-1

Hg2++ ISE none 25°C 0.0 U 1903SHa (2014)1533

Kso(Hg2L2)=-20.90

CN- HL Cyanide CAS 74-90-8 (230)

Cyanide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE none 25°C 0.0 U 1929BRa (2731)1534

Kso(Hg2L2(s)=Hg2+2L)=-39.3

CO3-- H2L Carbonate CAS 465-79-6 (268)

Carbonate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

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-----
Hg2++      ISE NaCl04 25°C 3.00M C      1976HHa (3241)1535
                                         *Ks=4.19
*Ks: Hg2L(s)+2H=Hg2+H2O+CO2(g)
-----
Hg2++      EMF none 25°C 0.0 U      1929BRa (3242)1536
                                         Kso(Hg2CO3(s)=Hg2+CO3)=-16.05
*****
C6N6Co---      H3L Cyanocobaltate (5470)
Hexacyanocobaltate; [Co(CN)6]---
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg2++      ISE oth/un 25°C 0.0 U      1965R0a (3492)1537
                                         Kso((Hg2)3L2)=-36.72
*****
C6N6Fe----      H4L (2191)
Hexacyanoferrate (II); Fe(II)(CN)6----
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg2++      sol oth/un 25°C dil U      1962BBb (3569)1538
                                         Kso((Hg2)2L)=-11.95
*****
C6N6Fe---      H3L Ferricyanide (2491)
Hexacyanoferrate (III); Fe(III)(CN)6---
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg2++      sol oth/un 25°C dil U      1962BBb (3652)1539
                                         Kso((Hg2)3L2)=-20.07
*****
Cl-      HL Chloride CAS 7647-01-0 (50)
Chloride;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg2++      EMF non-aq 25°C 100% U      1973CCa (5018)1540
                                         Kso(Hg2Cl2)=-37.3
Medium: MeCN
-----
Hg2++      EMF non-aq 177°C 100% U T H      1969BMF (5019)1541
                                         Kso(Hg2Cl2)=-12.48
Medium: (Li,K)NO3; DH(Kso)=138 kJ mol-1(m units); Kso=3.608-7239.6/T
137-207 C
-----
Hg2++      ISE non-aq 450°C 100% U T      1966FBa (5020)1542
                                         Kso=4.22-7495/T
                                         Kso=-12.45
Medium: (Li/K)NO3 eutectic. m units

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-----
Hg2++      ISE NaCl04 40°C 0.50M U T      1963HIa (5021)1543
Kso=-16.16
Kso=-18.19(7 C)
-----
Hg2++      ISE NaCl04 25°C 0.50M U      K1=<1.3      1947JQa (5022)1544
Kso(Hg2L2(s))=-16.88
-----
Hg2++      ISE NaCl04 25°C 0.50M U      1947LJa (5023)1545
K(Hg2L2(s)=Hg(l)+HgL2)=-5.77
K(Hg2L2(s)+L=Hg(l)+HgL3)=-4.92
-----
Hg2++      ISE none 25°C 0.0 U T      1946LAa (5024)1546
Kso(Hg2L2(s))=-17.273
I=0 corr. Kso=-17.775(15 C), -16.791(35 C), -16.323(45 C)
-----
Hg2++      sol none 25°C 0.0 U      1942GNa (5025)1547
Ks(Hg2L2(s)=Hg2L2)=-5.23 ?
Ks(Hg2L2(s)+L=Hg2L3)=-4.06 ?
-----
Hg2++      ISE none 26°C 0.0 U T H      1929BRa (5026)1548
Kso(Hg2L2(s))=-17.82
I=0 corr. Kso=-18.65(10.8 C), -18.48(14.9 C), -18.27(19.2 C)
DH(Kso)=30.6 kJ mol-1
-----
Hg2++      ISE oth/un 25°C var U      1904LHa (5027)1549
Kso(Hg2L2(s))=-17.7
-----
Hg2++      sol none 25°C 0.0 U      1903SHa (5028)1550
Kso(Hg2L2(s))=-17.46
*****
Cl04-      HL Perchlorate CAS 7001-90-3 (287)
Perchlorate;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg2++      kin oth/un 25°C var U      K1=0      1957AHa (6249)1551
Medium: 3 M H+,Na+
-----
Hg2++      ISE NaCl04 25°C var U      K1=-0.05      1956HSd (6250)1552
*****
Cr04--      H2L Chromate CAS 7738-94-5 (2382)
Chromate;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Hg2++      ISE none 25°C 0.0 U      1929BRa (6493)1553
Kso=-8.70
*****
F-          HL Fluoride CAS 7644-39-3 (201)

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Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg2++	ISE	NaCl04	25°C	0.50M	U		K1=<0	1955PAa	(6951)1554
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Method: Hg electrode. At I=0 corr K1 < 0.5

I-		HL		Iodide			CAS 10034-85-2	(20)	
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Iodide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Hg2++	ISE	none	25°C	0.0	U			1973CCa	(8098)1555
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Kso(Hg2I2(s)=Hg2+2I)=-39.3

Hg2++	sol	non-aq	142°C	100%	U			1969MBd	(8099)1556
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Kso=0.13-9986/T

Medium: (Li,K)NO3; Kso=K(Hg2I2(s)=Hg2+2I); 142-192 C

Hg2++	sol	none	25°C	0.0	U			1967KWb	(8100)1557
-------	-----	------	------	-----	---	--	--	---------	------------

K(Hg2I2(s)=Hg(l)+HgI2(s))=-1.8

Hg2++	ISE	non-aq	177°C	100%	U			1966FBa	(8101)1558
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Kso=-22?

Medium: (Li/K)NO3 eutectic, m units

Hg2++	ISE	NaCl04	40°C	0.60M	U T			1963HIa	(8102)1559
-------	-----	--------	------	-------	-----	--	--	---------	------------

Kso=-26.46

Medium:0.5 M NaCl04,0.1 HCl04. Kso=-29.34(7 C)

Hg2++	ISE	NaCl04	25°C	0.50M	U			1949QSa	(8103)1560
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Kso(Hg2L2)=-27.465

Hg2++	ISE	none	25°C	0.0	U			1938LAa	(8104)1561
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Kso(Hg2L2)=-28.35

Hg2++	ISE	none	25°C	0.0	U T			1929BRa	(8105)1562
-------	-----	------	------	-----	-----	--	--	---------	------------

Kso(Hg2L2)=-28.31

Method: Hg electrode, I=0 corr. Kso=-29.70(10.8 C), -29.29(14.9 C), -28.98(19.2 C), -28.13(26.5 C). DH(so)=52.7 kJ mol-1(25 C)

Hg2++	ISE	oth/un	25°C	0.04M	U			1903SHa	(8106)1563
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Kso(Hg2L2)=-27.92

I03-		HL		Iodate			CAS 7782-68-5	(1257)	
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Iodate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Hg2++	ISE	none	25°C	0.0	U			1943TAa	(8518)1564
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Kso(Hg2L2)=-13.71

Hg2++ ISE none 25°C 0.0 U 1929BRa (8519)1565

Kso(Hg2L2)=-17.89

N02- HL Nitrite CAS 7782-77-6 (635)
Nitrite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ sp oth/un 25°C var U B2=6.1 1973TUa (9380)1566

N03- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE NaCl04 25°C 3.00M U K1=-0.08 B2=-1.14 1983BMb (9706)1567

Hg2++ ISE NaNO3 25°C var U K1=0.40 1956HSd (9707)1568

Hg2++ ISE NaCl04 25°C 3.0M U I K1=0.02 B2=-0.30 1946ISa (9708)1569
K1=0.08(I=0.5 M)

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ kin NaCl04 25°C 0.10M U K2=1.20 1977SAa (11594)1570

Hg2++ gl NaCl04 25°C 3.00M C 1976HHc (11595)1571

*K1=-4.88

*B(2,1)=-2.68

*B(5,4)=-8.48(?)

*Kso=2.35

*Kso:Hg2(OH)1.3(Cl04)0.7(s)+1.3H=Hg2++ + 0.7Cl04 +1.3H2O

Hg2++ kin oth/un 25°C 0.05M U 1952FHa (11596)1572

*K1=-3.6

Hg2++ gl oth/un ? var U 1936NEa (11597)1573

*K1=ca.-4.3

Hg2++ gl oth/un 18°C var U 1934BEa (11598)1574

Kso=-23.11(?)

Hg2++ ISE none 25°C 0.0 U 1929BRa (11599)1575

Kso=-23.74(?)

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE NaCl04 25°C 3.00M C 1975QDa (13214)1576

Kso=-10.70

*Ks=-21.40

Kso:Hg2HPO4(s)=(Hg2) + HPO4; *Ks=(Hg2)3(PO4)2(s) + 2H = 3(Hg2) + 2HPO4

P207---- H4L Pyrophosphate CAS 2466-09-3 (198)

Diphosphate; from (HO)2PO.O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE oth/un 25°C 1.00M U I K1=8.83 B2=11.71 1962SIb (13596)1577

B(Hg2(OH)L)=15.08

B(Hg2(OH)2L)=20.42

Medium: Na+, with Hg electrode. In K+: K1=9.25, B2=12.27, B(Hg2(OH)L)=15.85,
B(Hg2(OH)2L)=20.05, K(Hg2+HL)=5.93

Hg2++ sp NaNO3 28°C 0.75M U 1960YDa (13597)1578

B(Hg2(OH)L)=15.64

Also Hg electrode

Hg2++ EMF NaNO3 28°C 0.75M U B2=12.38 1959YDa (13598)1579

B(Hg2(OH)L)=16.11

P3010----- H5L CAS 10380-08-2 (1001)

Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE NaNO3 27°C 0.75M U B2=11.23 1960YDa (13865)1580

B(Hg2L(OH))=15.0

S-- H2L Sulfide CAS 7783-06-4 (705)

Sulfide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE oth/un 18°C var U 1931K0a (14400)1581

Kso(Hg2L)=-47.0

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE none 25°C 0.0 U 1929BRa (15071)1582

K(Hg2L2(s)=Hg2+2L)=-19.52

Hg2++ sol oth/un 25°C var U 1905GRa (15072)1583
K(Hg2L2(s)=Hg2+2L)=-19.84

Also Hg electrode

Hg2++ sol oth/un 25°C var U 1905SSa (15073)1584
K(Hg2L2(s)=Hg2+2L)=-19.74

Also Hg electrode

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ EMF none 25°C 0.0 U T H 1970SPb (16240)1585
Kso=-6.09

Kso=-6.13(15 C); -6.09(25 C); -6.06(35 C); DHso=5.4 kJ mol-1

Hg2++ ISE oth/un 25°C 0.0 U 1964PCa (16241)1586
Kso(Hg2L)=-6.13

Hg2++ sol oth/un 25°C 0.0 U 1957BLa (16242)1587
Kso(Hg2L)=-6.17

Hg2++ ISE NaCl04 25°C 0.50M U I K1=1.30 B2=2.40 1946ISa (16243)1588
In 3 M NaCl04 Kso=-4.46

Hg2++ ISE oth/un 25°C 0.0 U 1932HJa (16244)1589
Kso(Hg2L)=-6.32

Hg2++ ISE oth/un 25°C 0.0 U 1929BRa (16245)1590
Kso(Hg2L)=-6.33

W04-- H2L Tungstate CAS 13783-36-3 (445)
Tungstate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE NaNO3 18°C 0.10M U 1933BHa (17440)1591
Kso(Hg2L)=-16.96

CH2O2 HL Formic acid CAS 64-18-6 (37)
Methanoic acid; H.CO0H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ gl NaCl04 25°C 3.0M C K1=2.94 B2=5.45 1977RWb (17614)1592

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)

Ethanedioic acid; (COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Hg2++	ISE	NaCl04	25°C	1.0M	C				2000VCa (18922)	1593
-------	-----	--------	------	------	---	--	--	--	-----------------	------

Kso=-11.19

Method: Hg,Hg2 oxalate electrode

Hg2++	ISE	oth/un	27°C	2.50M	U			B2=6.98 B(Hg2(OH)L)=13.04	1960YDa (18923)	1594
-------	-----	--------	------	-------	---	--	--	------------------------------	-----------------	------

C2H3O2Cl HL Chloroacetic CAS 79-11-8 (34)
Chloroethanoic acid; ClCH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Hg2++	gl	NaCl04	25°C	3.0M	C			K1=2.40 B2=4.4	1977RWb (19367)	1595
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C2H4O2 HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH₃.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Hg2++	gl	alc/w	25°C	100%	M			K1=7.1 B2=12.8	1994MPc (19995)	1596
-------	----	-------	------	------	---	--	--	----------------	-----------------	------

Medium: MeOH

Hg2++	gl	NaCl04	25°C	3.0M	C			K1=3.57 B2=6.63	1977RWb (19996)	1597
-------	----	--------	------	------	---	--	--	-----------------	-----------------	------

Hg2++	vlt	non-aq	?	100%	U	M			1963DTa (19997)	1598
-------	-----	--------	---	------	---	---	--	--	-----------------	------

K(Hg2A2+2HL=Hg2L2+2HA)=7.03

Medium: ethanoic acid. HA=HClO₄

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Hg2++	gl	NaCl04	25°C	3.0M	C	T		K1=3.01 B2=5.71	1977RWb (20556)	1599
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C3H6O2 HL Propionic acid CAS 79-09-4 (35)
Propanoic acid; CH₃.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Hg2++	gl	NaCl04	25°C	3.0M	C			K1=3.72 B2=6.99	1977RWb (25009)	1600
-------	----	--------	------	------	---	--	--	-----------------	-----------------	------

C3H6O3 HL Methoxyacetic CAS 625-45-6 (29)
Methoxyethanoic acid; CH₃.O.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Hg2++ gl NaClO4 25°C 3.00M C K1=2.98 B2=8.39 1977RWb (25600)1601

C4H7NO4 H2L IDA CAS 142-73-4 (118)
Iminodiethanoic acid; HN(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ sol NaClO4 25°C 0.10M U K1=10.81 1967SKg (32274)1602

C4H8O3 HL CAS 300-85-6 (30)
3-Hydroxybutanoic acid; CH3.CH(OH).CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ gl NaClO4 25°C 3.00M C K1=3.3 1977RWb (33623)1603

C4H8O3 HL CAS 2544-06-1 (28)
3-Methoxypropanoic acid; CH3.O.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ gl NaClO4 25°C 3.00M C K1=3.50 B2=6.63 1977RWb (33635)1604

C5H8O4 H2L CAS 595-46-0 (1144)
Dimethylmalonic acid; HOO.C(CH3)2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ ISE NaNO3 27°C 0.75M U B2=7.52 1960YDa (38212)1605
B(Hg(OH)L)=13.58

C6H4Cl2 L CAS 106-46-7 (2405)
1,4-Dichlorobenzene; Cl.C6H4.Cl

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ nmr non-aq 32°C 100% U K1=-0.7 1976DIa (42169)1606
Medium: liquid SO2

C6H7N L Aniline CAS 62-53-3 (583)
Aminobenzene, aniline; C6H5.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Hg2++ gl NaClO4 27°C 1.0M U K1=3.71 1964WDa (44873)1607

C7H6NO3Cl H2L (205)
3-Chlorosalicylhydroxamic acid; Cl.C6H3(OH).CO.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	gl	diox/w	30°C	50%	U		K1=11.38 B2=19.16	1977DJb (53416)	1608

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)
 2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	gl	alc/w	25°C	100%	M		K1=6.1 B2=11.2	1994MPc (54234)	1609

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)
 Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	sp	oth/un	18°C	0.10M	U		K1=4.90	1966Gcb (58974)	1610

C8H10 L p-Xylene CAS 106-42-3 (2145)
 1,4-Dimethylbenzene, 4-Xylene; CH3.C6H4.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	nmr	non-aq	32°C	100%	U		K1=3.5 B2=5.70	1976DIa (60681)	1611

Medium: liquid SO2

C11H10N2 L CAS 1132-37-2 (2427)
 (2,2'-Dipyridyl)methane; C5H4N.CH2.C5H4N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	gl	KN03	20°C	0.10M	U		K1=7.8 B2=14.30	1970BAa (77659)	1612

C12H8N2 L Phenanthroline CAS 66-71-7 (144)
 1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	ISE	NaNO3	25°C	0.50M	U			1967SPa (80465)	1613

Ks(Hg2L2(NO3)2)=-23.20

C12H18 L CAS 87-85-4 (2406)
 Hexamethylbenzene; C6(CH3)6

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Hg2++	nmr	non-aq	32°C	100%	U		K1=5.0 B2=8.50	1976DIa (81791)	1614

Medium: liquid SO2

C15H33N5S2 L CAS 143634-65-5 (7925)

1-Methyl-8-amino-3,13-dithia-6,10,16,19-tetraazabicyclo[6.6.6]icosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg2++	gl	KN03	25°C	0.10M	C			K1=19.5 K(HgL+H)=5.9	2001SGa (92584)	1615

C20H36O6 L DiCy-18-crown-6 CAS 16069-36-6 (1653)
2,3:11,12-Dicyclohexyl-1,4,7,10,13,16-hexaoxacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg2++	cal	oth/un	25°C	0.10M	U	H		K1=1.93 (syn isomer) K1=1.57 (anti isomer) K2=0.60 (syn isomer) K2=1.1 (anti isomer)	1976ITb (100650)	1616

DH(K1,syn)=-9.04 and DH(K1,anti)=-18.2 kJ mol⁻¹. DH(K2,cis-syn-cis)=26.0,
DH(K2,cis-anti-cis)=-24.0

C34H54O8 H2L Lasalocid CAS 25999-20-6 (2335)
Lasalocid acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg2++	gl	alc/w	25°C	100%	M			K1=6.8 B2=12.6	1994MPc (106137)	1617

Medium: MeOH

C44H22N4O12Br8S4 H6L CAS 176173-80-1 (6959)
2,3,7,8,12,13,17,18-Octabromo-5,10,15,20-tetrakis(4-sulfonatophenyl)porphyrin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Hg2++	gl	NaNO3	25°C	0.1M	C			K(Hg2+H2L=Hg2L+2H)=3.57	1996TNa (107038)	1618

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	vlt	NaNO3	20°C	0.10M	C			K(CH3Hg+Br)=6.61	1986GLa (2015)	1619

Methods: normal and DP polarography.

HgR+	nmr	alc/w	25°C	100%	U	M		K(MeHg+Br)=0.021	1981BRa (2016)	1620
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Medium: EtOH

HgR+	dis	non-aq	25°C	100%	U	I			1978JIb (2017)	1621
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$K(\text{MeHg}+\text{Br}=\text{MeHgBr}(\text{aq}))=6.60$
 $K(\text{MeHg}+\text{Br}=\text{MeHgBr}(\text{org}))=8.30$
 $K(\text{CH}_3\text{Hg}+2\text{Br}=\text{CH}_3\text{HgBr}_2(\text{aq}))=5.98$

In o-xylene using the two-phase system: o-xylene/H₂O, 2.5 M NaNO₃.

Data also for o-xylene, 1.0 M (Na,H)(Br,Cl,NO₃)-medium

HgR+	nmr non-aq 25°C 100% U	1975BWc (2018)1622
	$K(\text{MeHg}+\text{L})=6.62$	

Medium: DMF

HgR+	kin oth/un 20°C 0.10M U	1973GEb (2019)1623
	$K(\text{MeHg}+\text{L})=6.62$	

HgR+	oth oth/un 25°C 0.0 U	1968SRf (2020)1624
	$K(\text{EtHg}+\text{L})=0.3$	
	$K(\text{EtHg}+\text{L}_2)=1.1$	

HgR+	vlt mixed 25°C 60% U M	1967BBb (2021)1625
	$K(\text{HgC}_6\text{H}_5+\text{L})=5.28$	
	$K(\text{HgC}_6\text{H}_5\text{L}+\text{L})=0.74$	
	$K(\text{HgCH}_2\text{C}_6\text{H}_5+\text{L})=5.42$	
	$K(\text{HgCH}_2\text{C}_6\text{H}_5\text{L}+\text{L})=1.10$	

Medium: 60% DMF, 0.1 Na,LiClO₄. Data also with HgCH₂CH₂C₆H₅(7.95,1.23), HgCH₂COCMe₃(5.18,2.75) and HgC₆H₅CHC₀₀Et(4.43,2.95)

HgR+	sol NaClO ₄ 25°C 1.0M U	1965BBa (2022)1626
	$K_s(\text{EtHg}+\text{L})=-3.1$	

HgR+	gl KNO ₃ 20°C 0.10M U H	1965SSd (2023)1627
	$K(\text{MeHg}+\text{L})=6.62$	

By calorimetry: DH=-41.4 kJ mol⁻¹, DS=-15.1 J K⁻¹ mol⁻¹

HgR+	gl oth/un 25°C dil U	1962POd (2024)1628
	$K(\text{CF}_3\text{Hg}+\text{L})=7.24$	
	$K(\text{C}_2\text{F}_5\text{Hg}+\text{L})=7.16$	
	$K(\text{C}_3\text{F}_7\text{Hg}+\text{L})=7.16$	

HgR+	dis oth/un 25°C 0.0 U	1961SIa (2025)1629
	$K_d(\text{CH}_3\text{Hg}+\text{L} \text{ into } \text{C}_6\text{H}_5\text{CH}_3)=1.65$	

HgR+	gl oth/un 25°C dil U	1955WWa (2026)1630
	$K(\text{CH}_3\text{Hg}+\text{L})=6.70$	
	$K(\text{CH}_3\text{Hg}+\text{L}(\text{s})=\text{CH}_3\text{Hg}+\text{L})=-2.19$	
	$K(\text{CH}_3\text{Hg}+\text{L}(\text{s})=\text{CH}_3\text{Hg}+\text{L})=-8.90$	

HgR+	gl oth/un 25°C dil U	1955WWa (2027)1631
	$K_{so}(\text{C}_6\text{H}_5\text{Hg}+\text{Br}=\text{C}_6\text{H}_5\text{Hg}+\text{Br})=-11.75$	

CN-	HL	Cyanide	CAS 74-90-8 (230)
Cyanide;			

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	kin	oth/un	20°C	0.10M	U		T	K1=14.1	1973GEb	(2732)1632
Metal ion: MeHg+										
HgR+	gl	KNO3	20°C	0.10M	U	H	T	K(MeHg+L)=14.0 K(MeHg+L+L) < 0.5	1963SCc	(2733)1633
By calorimetry: DH(MeHg+L)=-92.4 kJ mol ⁻¹ , DS=-47.7 J K ⁻¹ mol ⁻¹										
HgR+	gl	oth/un	25°C	0.1?M	U		T	K(MeHg+L)=14.2	1961SIa	(2734)1634

CO3--			H2L		Carbonate			CAS 465-79-6	(268)	
Carbonate;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	?	C	M		K1=6.10	1976RTa	(3243)1635

C6N6Co---			H3L		Cyanocobaltate			(5470)		
Hexacyanocobaltate; [Co(CN)6]---										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	20°C	0.10M	U			K(MeHg+L)=4.15 K(MeHg+2L)=7.65	1963SCc	(3493)1636

Cl-			HL		Chloride			CAS 7647-01-0	(50)	
Chloride;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	NaCl	25°C	0.0	C	I		K1=5.45	1998DFb	(5029)1637
Metal: (CH3)Hg+. Extrapolated from data for 0.1-3.0 M NaCl Pitzer interaction parameters derived.										
HgR+	nmr	KNO3	25°C	0.30M	C			K(HgR+Cl)=4.50	1987RBa	(5030)1638
Method: 1H nmr. R is p-benzenesulfonate.										
HgR+	vlt	NaNO3	20°C	0.10M	C			K(CH3Hg+Cl)=5.26	1986GLa	(5031)1639
Methods: normal and DP polarography.										
HgR+	dis	none	25°C	0.0	C			K(EtHg+Cl)=5.43	1984SNa	(5032)1640
Method: extraction of RHg+ from dilute acidified chloride solution into										

benzene.

HgR+ nmr non-aq 25°C 100% U 1981BRa (5033)1641
K(MeHg+Cl)=0.06

Medium: CH₂Cl₂

HgR+ dis non-aq 25°C 100% U I 1978JIb (5034)1642
K(MeHg+Cl=MeHgCl(aq))=5.50
K(MeHg+Cl=MeHgCl(org))=6.59

In o-xylene using the two-phase system: o-xylene/H₂O, 2.5 M NaNO₃.

Data also for o-xylene, 1.0 M (Na,H)(Br,Cl,NO₃)-medium

HgR+ dis NaClO₄ 25°C 1.0M C 1978SKi (5035)1643
K(CH₃Hg+Cl)=5.51
K(C₂H₅Hg+Cl)=5.32
K(C₆H₅Hg+Cl)=5.77

Medium: 1.0 M HClO₄. Method: extraction into dithizone/CCl₄.

HgR+ nmr non-aq 25°C 100% U 1975BWc (5036)1644
K(MeHg+L)=5.25

Medium: DMF

HgR+ dis NaClO₄ 25°C 1.0M U 1973BIa (5037)1645
K(MeHg+L)=5.32

HgR+ kin oth/un 20°C 0.10M U 1973GEb (5038)1646
K(MeHg+L)=5.25

HgR+ sol NaClO₄ 25°C 1.0M U 1965BBa (5039)1647
Ks(EtHgL)=-2.4

HgR+ gl KNO₃ 20°C 0.10M U H 1965SSd (5040)1648
K(MeHg+L)=5.25

By calorimetry: DH=-25.1 kJ mol⁻¹, DS=15.1 J K⁻¹ mol⁻¹

HgR+ oth KCl 0°C var U 1964D0a (5041)1649
K(Hg(CF₃)₂+L)=-0.5

Method:freezing point.

HgR+ gl none 25°C 0.0 U 1962P0d (5042)1650
K(CF₃Hg+L)=5.78
K(C₂F₅Hg+L)=5.64
K(C₃F₇Hg+L)=5.56

HgR+ dis none 25°C 0.0 U 1961SIa (5043)1651
Kd(MeHgL into toluene)=1.0

HgR+ gl none 25°C 0.0 U 1955WWa (5044)1652
Kso(MeHgL(s))=-7.16
Ks(MeHgL(s)+L)=-1.71

K(MeHg+L)=5.45

Kso(PhHgL(s))=-9.30

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	20°C	0.10M	U				1965SSd	(6952)1653

K(MeHg+L)=1.50

FCIBrI HL (541)
Halides, comparative (for book data under ligand 80)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	25°C	0.10M	U T M				1968ZPa	(7405)1654

K(MeHg+Cl)=4.90

K(MeHg+Br)=5.98

K(MeHg+I)=7.70

K(Cl)=4.78(Et),4.65(Pr),4.55(Bu). K(Br)=5.90(Et),5.80(Pr),5.74(Bu).

K(I)=7.85(Et),8.20(Pr). At 20 C:K(MeHg+I)=8.70

HPO3-- H2L Phosphite CAS 13598-36-2 (6305)
Phosphite;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	20°C	0.10M	U				1963SCc	(7511)1655

K(MeHg+L)=4.67

I- HL Iodide CAS 10034-85-2 (20)
Iodide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	kin	oth/un	20°C	0.10M	U				1973GEb	(8107)1656

K(MeHg+L)=8.60

HgR+	sol	NaClO4	25°C	1.0M	U				1965BBa	(8108)1657
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K(EtHgI+I)=-0.67

K(EtHgI2+I)=0.75

K(EtHgL(s)=EtHgL)=-4.11

HgR+	gl	KNO3	20°C	0.10M	U				1965SSd	(8109)1658
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Kso(MeHgI)=-11.46

K(MeHg+I)=8.60

K(MeHgI+I)=0.26 ?

HgR+	oth	oth/un	0°C	var	U				1964DOa	(8110)1659
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K(Hg(CF₃)₂+L)=0.98

Method:freezing point. Medium:KI var

HgR+ gl oth/un 25°C dil U 1962POd (8111)1660

K(CF₃Hg+L)=9.63
K(C₂F₅Hg+L)=9.66
K(C₃F₇Hg+L)=9.96

HgR+ gl oth/un 25°C 0.10M U 1961SIa (8112)1661

Kd(MeHgL into C₆H₅CH₃)=2.6

HgR+ gl oth/un 25°C dil U 1955WWa (8113)1662

K(MeHgL(s)=MeHg+L)=-11.72
K(MeHg+L)=8.7
K(C₆H₅HgL(s)=C₅H₅Hg+L)=-15.01

NH₃ L Ammonia CAS 7664-41-7 (414)
Ammonia

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ sp oth/un 24°C var U 1978SMA (9166)1663

K(MeHg(OH)+HL=MeHgL+H₂O)=2.42

Medium: phosphate buffer, pH 6-8, ionic strength 0.4 to 0.6M.

HgR+ nmr none 25°C 0.0 U 1974ROa (9167)1664

K(MeHg+L)=7.25

HgR+ gl KNO₃ 20°C 0.10M U 1965SSd (9168)1665

K(MeHg+L)=7.60

HgR+ gl oth/un 25°C 0.1?M U 1961SIa (9169)1666

K(MeHg+L)=8.4

NO₃- HL Nitrate CAS 7697-37-2 (288)
Nitrate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ dis non-aq 25°C 100% U I 1978JIb (9709)1667

K(MeHg+NO₃=MeHgNO₃(org))=-1.2

In o-xylene using the two-phase system: o-xylene/H₂O, 2.5 M NaNO₃.

Data also for o-xylene, 1.0 M (Na,H)(Br,Cl,NO₃)-medium

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl NaNO₃ 25°C 0.0 C I 1998DFb (11600)1668

*K1=-4.528

*B(2,1)=-2.15

Metal: (CH₃)Hg⁺. Extrapolated from data for 0.01-3.25 M NaNO₃.

Pitzer interaction parameters derived.

HgR⁺ nmr KNO₃ 25°C 0.30M C 1987Rba (11601)1669

K(HgR+OH)=9.12

Method: ¹H nmr. R is p-benzenesulfonate.

HgR⁺ vlt NaNO₃ 20°C 0.10M C 1986GLa (11602)1670

K(CH₃Hg+OH)=9.36

Methods: normal and DP polarography.

HgR⁺ dis none 25°C 0.0 C 1984SNa (11603)1671

K(EtHg+OH)=9.23

K(n-PrHg+OH)=9.20

K(n-BuHg+OH)=9.04

K(PhHg+OH)=9.98

Method: extraction of RHg⁺ from dilute phosphate buffer into benzene/
8-hydroxyquinoline.

HgR⁺ nmr oth/un 25°C ? U 1982MCb (11604)1672

K(MeHg+L)=9.41

HgR⁺ gl NaNO₃ 25°C 1.00M C 1978JIc (11605)1673

*K1=-4.686

*B(2,1)=-1.725

Kso(MeHgOH)=-13.66

HgR⁺ gl KNO₃ 25°C 0.10M C I 1974ANa (11606)1674

K(MeHg+OH)=4.56

K(PhHg+OH)=4.03

In 75% dioxan, 0.1 M NaClO₄: K(MeHg+OH)=5.2, K(PhHg+OH)=4.5, K(MeHg+MeHgOH=
(MeHg)₂OH)=3.7, K(PhHg+PhHgOH=(PhHg)₂OH)=3.85

HgR⁺ dis NaNO₃ 25°C 1.0M C 1974ILb (11607)1675

*K(CH₃Hg)=-4.40

HgR⁺ kin oth/un 20°C 0.10M U 1973GEb (11608)1676

K(MeHg+OH)=9.37

HgR⁺ nmr oth/un 25°C U 1973LRa (11609)1677

K(MeHg+OH)=9.30

K(MeHgOH+MeHg)=2.37

HgR⁺ sp oth/un ? U 1971SBa (11610)1678

*K1(M+H₂O=MOH+H)=-2.74

M=HgC(NO₂)₃⁺. Dissociation of cation neglected

HgR⁺ gl KNO₃ 25°C 0.10M U T 1968ZPa (11611)1679

K(MeHg+L)=9.00
K(EtHg+L)=8.80
K(PrHg+L)=8.66
K(BuHg+L)=8.61

K(MeHg+L)=9.32(20 C)

HgR+ gl KNO3 20°C 0.10M U H 1963SCc (11612)1680

*K1(MeHg)=-4.59

K(MeHgOH+MeHg)=2.37

DH(*K1)=-35.5 kJ mol⁻¹, DS=-57.3 J K⁻¹ mol⁻¹

HgR+ gl oth/un 25°C dil U 1962POd (11613)1681

K(CF3Hg+L)=10.76

K(C2F5Hg+L)=10.58

K(C3F7Hg+L)=10.50

HgR+ gl oth/un 25°C dil U 1961PSb (11614)1682

K(MeHg+L)=9.89

HgR+ gl oth/un 24°C dil U 1955WWa (11615)1683

K(MeHg+L)=9.50

K(EtHg+L)=9.1

K(PhHg+L)=10.0

P04--- H3L Phosphate CAS 7664-38-2 (176)

Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ dis none 25°C 0.0 C 1984SNa (13215)1684

K(EtHg+P04)=8.45

K(EtHg+HP04)=4.08

K(n-PrHg+P04)=8.04

K(n-PrHg+HP04)=4.15

Method: extraction of RHg+ from dilute phosphate buffer into benzene/
8-hydroxyquinoline.

HgR+ dis none 25°C 0.0 C 1984SNa (13216)1685

K(PhHg+P04)=9.49

K(PhHg+HP04)=5.26

K(n-BuHg+P04)=8.43

K(n-BuHg+HP04)=4.26

Method: extraction of RHg+ from dilute phosphate buffer into benzene/
8-hydroxyquinoline.

HgR+ dis NaNO3 25°C 1.0M C 1974ILb (13217)1686

K(MeHg+H2P04=MeHgHP04+H)=-1.74

HgR+ gl KNO3 20°C 0.10M U 1965SSd (13218)1687

K(MeHg+HL)=5.03

S-- H2L Sulfide CAS 7783-06-4 (705)
Sulfide;

HgR+ gl oth/un 25°C 0.1?M U 1961SIa (15081)1696

K(MeHg+L)=6.1

K(CH₃Hg+OH)=9.5 assumed

S03-- H2L Sulfite CAS 7782-99-2 (801)

Sulfite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr oth/un 25°C ? C M K1=7.96 1976RTa (15462)1697

HgR+ gl KNO3 20°C 0.10M U 1965SSd (15463)1698

K(MeHg+L)=8.11

S04-- H2L Sulfate CAS 7664-93-9 (15)

Sulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl oth/un 25°C 0.0 C I K1=2.64 1998DFb (16246)1699

Metal: (CH₃)Hg+. Extrapolated from data for 0.1-1.0 M Na₂SO₄

Pitzer interaction parameters derived.

HgR+ gl oth/un 25°C ? U M K1=0.94 1976RTa (16247)1700

S203-- H2L Thiosulfate CAS 73686-28-7 (177)

Thiosulfate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ cal KNO3 20°C 0.10M U H 1963SCc (16860)1701

K(MeHg+L)=10.90

DH=-48.9 kJ mol⁻¹, DS=41.4 J K⁻¹ mol⁻¹

SeCN- HL Selenocyanate CAS 73102-11-2 (440)

Selenocyanate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr oth/un 25°C ? C M K1=6.79 1976RTa (16989)1702

Se03-- H2L Selenite CAS 7783-00-8 (2391)

Selenite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr oth/un 25°C ? C M K1=6.46 1976RTa (17063)1703

K(MeHg+HL)=2.70

Se04-- H2L Selenate CAS 7783-08-6 (459)

Selenate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	?	C	M		K(MeHg+L)=1.12	1976RTa (17103)	1704

CH2O2		HL		Formic acid			CAS	64-18-6	(37)	
Methanoic acid; H.CO ₂ H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	NaNO ₃	25°C	1.00M	C			K(MeHg+L)=2.681	1978JIc (17615)	1705

HgR+	nmr	oth/un	25°C	0.40M	U			K(MeHg+L)=2.67	1973LRa (17616)	1706

CH ₅ N		L		Methylamine			CAS	74-89-5	(155)	
Methylamine; CH ₃ .NH ₂										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	sp	oth/un	24°C	var	U			K(MeHg(OH)+HL=MeHg ₂ L+H ₂ O)=1.23	1978SMa (18021)	1707
Medium: phosphate buffer, pH 6-8, ionic strength 0.4 to 0.6M.										
HgR+	nmr	none	25°C	0.0	U			K(MeHg+L)=7.57	1974ROa (18022)	1708

C ₂ H ₂ O ₂ Cl ₂		HL					CAS	79-43-6	(1282)	
Dichloroethanoic acid; Cl ₂ CH.CO ₂ H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	0.40M	U			K(MeHg+L)=1.14	1973LRa (18397)	1709

C ₂ H ₃ O ₂ Cl		HL		Chloroacetic			CAS	79-11-8	(34)	
Chloroethanoic acid; ClCH ₂ .CO ₂ H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	0.40M	U			K(MeHg+L)=2.19	1973LRa (19368)	1710

C ₂ H ₄ O ₂		HL		Acetic acid			CAS	64-19-7	(36)	
Ethanoic acid; CH ₃ .CO ₂ H										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

HgR+ gl NaNO3 25°C 1.00M C 1978JIc (19998)1711
 K(MeHg+L)=3.204
 B(2MeHg+L)=5.279

HgR+ nmr non-aq 25°C 100% U 1975BWc (19999)1712
 K(MeHg+L)=3.6

Medium: DMF

HgR+ nmr oth/un 25°C 0.40M U 1973LRa (20000)1713
 K(MeHg+L)=3.18

C2H5NO2 HL Glycine CAS 56-40-6 (85)
 2-Aminoethanoic acid; H2N.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	dis	NaClO4	25°C	1.0M	C			K1=8.78	1988KSa (21580)	1714

Metal is C6H5Hg+. Method: extraction of 203Hg(C6H5)Br into benzene from a ligand solution in 1.0 M NaClO4.

HgR+ gl NaNO3 25°C 1.00M C 1978JIc (21581)1715
 K(MeHg+L)=7.518
 K(MeHg+2L)=9.468

HgR+ sp oth/un 24°C var U 1978SMa (21582)1716
 K(MeHg(OH)+HL=MeHgL+H2O)=2.67
 Medium: phosphate buffer, pH 6-8, ionic strength 0.2 to 0.3M.

HgR+ gl oth/un 22°C ? U 1976HSa (21583)1717
 K(C6H5Hg+L)=7.13

HgR+ gl oth/un 22°C ? U 1976HSa (21584)1718
 K(MeHg+L)=6.07

HgR+ nmr none 25°C 0.0 U 1974R0a (21585)1719
 K(MeHg+L)=7.88

C2H6OS HL CAS 60-24-2 (841)
 2-Mercaptoethanol; HS.CH2.CH2.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	KNO3	25°C	0.30M	C			K(HgR+L)=16.35	1987RBa (22067)	1720

Method: 1H nmr. R is p-benzenesulfonate.

HgR+ gl oth/un 22°C ? U 1976HSa (22068)1721
 K(MeHg+L)=8.76
 K(C6H5Hg+L)=6.67

HgR+ nmr non-aq 25°C 100% U 1975BWc (22069)1722
K(MeHg+L)=16.1

Medium: DMF

C2H6S2 L CAS 624-92-0 (152)
Dimethyl disulfide; CH3.S.S.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr non-aq 25°C 100% U I M 1981BRa (22202)1723
K(MeHgA+L)=-1.30

Medium: CH2Cl2. HA=EtOH. In EtOH: K(MeHgA+L)=-0.16

C2H7N L Dimethylamine CAS 124-40-3 (802)
Dimethylamine; CH3.NH.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr none 25°C 0.0 U 1974R0a (22226)1724
K(MeHg+L)=6.76

C2H7N L Ethylamine CAS 75-04-7 (156)
Ethylamine; CH3.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ sp oth/un 24°C var U 1978SMa (22275)1725
K(MeHg(OH)+HL=MeHgL+H2O)=1.31

Medium: phosphate buffer, pH 6-8, ionic strength 0.4 to 0.6M

HgR+ nmr none 25°C 0.0 U 1974R0a (22276)1726
K(MeHg+L)=7.64

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)
1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl KNO3 22°C 0.10M U M 1977HSb (23181)1727
K(MeHg+L)=5.52, B2=7.90
K(EtHg+L)=5.52, B2=7.28
K(PrHg+L)=5.56, B2=7.35

C3H4N2 L Imidazole CAS 288-32-4 (90)
1,3-Diazole, imidazole; C3H4N2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr NaCl 25°C 0.50M U M K1=9.0 1981BEa (23903)1728
R=Hg(p-chlorobenzoate)

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HgR+      nmr oth/un 25°C  var  U  TI      K1=11.76      1977ERb (23904)1729
              K(MeHg+HL)=6.93
              K(MeHg+MeHgL)=8.26
Also K(MeHg+L)=11.79 at 20 C, by pH-metric titn.
*****
C3H4O3      HL      Pyruvic acid      CAS 127-17-3 (1152)
2-Oxopropanoic acid; CH3.CO.COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
HgR+      gl  NaNO3  25°C 1.00M U      1981JIa (24055)1730
              K(MeHg+L)=2.31
*****
C3H6O2      HL      Propionic acid  CAS 79-09-4 (35)
Propanoic acid; CH3.CH2.COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
HgR+      nmr oth/un 25°C 0.40M U      1973LRa (25010)1731
              K(MeHg+L)=3.39
*****
C3H6O2S      H2L      Thiolactic acid  CAS 79-42-5 (366)
2-Mercaptopropanoic acid; CH3.CH(SH).COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
HgR+      gl  oth/un 22°C  ?  U      1976HSa (25148)1732
              K(MeHg+L)=9.03
              K(C6H5Hg+L)=7.37
*****
C3H7NO2      HL      Alanine      CAS 56-41-7 (86)
2-Aminopropanoic acid; H2N.CH(CH3).COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
HgR+      gl  NaNO3  25°C 1.00M C      1978JIc (26189)1733
              K(MeHg+L)=7.516
              K(MeHg+2L)=9.450
*****
C3H7NO2      HL      B-Alanine      CAS 107-95-9 (575)
3-Aminopropanoic acid; H2N.CH2.CH2.COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
HgR+      sp  oth/un 24°C  var  U      1978SMa (26457)1734
              K(MeHg(OH)+HL=MeHgL+H2O)=1.80
Medium: phosphate buffer, pH 6-8, ionic strength 0.2 to 0.3M.
-----
HgR+      nmr none  25°C  0.0  U      1974R0a (26458)1735

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K(MeHg+HL)=2.52 (O coord)

K(MeHg+L)=7.56 (N coord)

C3H7NO2S H2L Cysteine CAS 52-90-4 (96)

2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ dis NaClO4 25°C 1.0M C K1=16.5 1988KSa (26793)1736

Metal is C6H5Hg+. Method: extraction of 203Hg(C6H5)Br into benzene from a
ligand solution in 1.0 M NaClO4.

HgR+ dis NaClO4 25°C 1.0M C K1=16.58 1988SKf (26794)1737

K(CH3Hg+HL)=15.56

Method: extraction of 203Hg-labelled complex into dithizone/CCl4 solution
Metal is CH3Hg+.

HgR+ dis NaClO4 25°C 1.0M C K1=17.7 1988SKf (26795)1738

K(C6H5Hg+HL)=16.5

Method: extraction of 203Hg-labelled complex into dithizone/CCl4 solution.
Metal is C6H5Hg+.

HgR+ nmr KNO3 25°C 0.30M C 1987RBa (26796)1739

K(HgR+L)=16.97

K(HgR+HL)=15.50

Method: 1H nmr. R is p-benzenesulfonate.

HgR+ gl NaNO3 25°C 1.00M U K1=15.70 1981JIa (26797)1740

K(MeHg+H2L)=6.69

B((MeHg)HL)=24.96

K(MeHgL+H)=9.25

HgR+ nmr oth/un 25°C 0.30M M K1=16.67 1981RRd (26798)1741

K(MeHgL+H)=8.80

K(MeHg+HL)=15.38

K(MeHgHL+H)=2.44

HgR+ gl oth/un 22°C ? U 1976HSa (26799)1742

K(C6H5Hg+L)=4.77

K(C6H5HgL+H)=8.64

HgR+ gl oth/un 22°C ? U 1976HSa (26800)1743

K(MeHg+L)=7.19

K(MeHgL+L)=5.96

K(MeHgL+H)=8.92

C3H7NO3 HL Serine CAS 56-45-1 (49)

2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

C3H10N2 L Propanediamine CAS 109-76-2 (123)
1,3-Diaminopropane; H2N.CH2.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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HgR+	gl	KNO3	22°C	0.10M	U	M			1977HSb (28309)	1751
								K(MeHg+L)=5.24, B2=9.01		
								K(EtHg+L)=5.08, B2=8.60		
								K(PrHg+L)=5.09, B2=8.39		

C4H6N2 L N-Me-Imidazole CAS 616-47-7 (354)
N-Methyl-1,3-diazole; C3H3N2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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HgR+	nmr	oth/un	25°C	var	U				1977ERb (29600)	1752
								K(MeHg+HL)=6.96		

C4H6O4S H3L Thiomalic acid CAS 70-49-5 (109)
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; H00C.CH(SH).CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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HgR+	nmr	KNO3	25°C	0.30M	C				1987RBa (30336)	1753
								K(HgR+L)=17.28		

Method: 1H nmr. R is p-benzenesulfonate.

HgR+	nmr	oth/un	25°C	0.30M	M			K1=17.31	1981RRd (30337)	1754
								K(MeHgL+H)=4.85		
								K(MeHgHL+H)=3.53		

HgR+	gl	oth/un	22°C	?	U				1976HSa (30338)	1755
								K(C6H5Hg+L)=7.78		

HgR+	gl	oth/un	22°C	?	U				1976HSa (30339)	1756
								K(MeHg+L)=8.16		

C4H6O4S2 H4L CAS 2418-14-6 (4264)
2,3-Dimercaptobutanedioic acid; H00C.CH(SH).CH(SH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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HgR+	nmr	KNO3	25°C	0.30M	C			K1=18.4	1985ACa (30393)	1757
								K(CH3Hg+CH3HgL)=16.9		

Metal is CH3Hg+. Method: 1H nmr by competition with mercaptoethanoic acid.

C4H7NO3 HL CAS 543-24-8 (3586)
N-Acetylglycine; CH3.CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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HgR+ nmr oth/un 25°C 0.40M U 1973LRa (31502)1758
K(MeHg+L)=2.68

C4H7N04 H2L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ dis NaCl04 25°C 1.0M C K1=8.71 1988K5a (31868)1759
Metal is C6H5Hg+. Method: extraction of 203Hg(C6H5)Br into benzene from a
ligand solution in 1.0 M NaCl04.

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl NaNO3 25°C 1.00M U 1981JIa (32703)1760
K(MeHg+L)=6.32

C4H9NO2 HL 4-Aminobutyric CAS 56-12-2 (574)
4-Aminobutanoic acid; H2N.CH2.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr none 25°C 0.0 U 1974R0a (33983)1761
K(MeHg+HL)=2.74 0 coord
K(MeHg+L)=7.54 N coord

C4H9NO2S HL Methylcysteine CAS 1187-84-4 (84)
2-Amino-3-methylmercaptopropanoic acid; H2N.CH(CH2.S.CH3)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl oth/un 22°C ? U 1976HSa (34098)1762
K(C6H5Hg+L)=5.84
K(MeHg+L)=5.15

C4H9NO2S HL CAS 29768-80-7 (2597)
2-Amino-4-mercaptoputanoic acid; H0OC.CH(NH2).CH2.CH2.SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ nmr oth/un 25°C 0.30M M K1=16.45 1981RRd (34113)1763
K(MeHg+L+H)=9.12
K(MeHgHL+H)=2.26

C4H10O2S2 H2L Dithiothreitol CAS 3483-12-3 (8164)
Threo-2,3-Dihydroxy-1,4-dithiobutane

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
HgR+	nmr	KNO3	25°C	0.30M	C		K1=17.0	1985ACa (34696)	1764
Metal is CH3Hg+. Method: 1H nmr by competition with mercaptoethanoic acid.									

C4H11N		L	t-Butylamine				CAS 75-64-9	(158)	
2-Amino-2-methylpropane; H2N.C(CH3)3									
HgR+	sp	oth/un	24°C	var	U			1978SMa (34790)	1765
K(MeHg(OH)+HL=MeHgL+H2O)=1.20									
Medium; phosphate buffer, pH 6-8, ionic strength 0.3 to 0.4M.									
HgR+	nmr	none	25°C	0.0	U			1974R0a (34791)	1766
K(MeHg+L)=7.52									

C4H11N		L	Diethylamine				CAS 109-89-7	(1331)	
Diethylamine, 3-azapentane; (C2H5)2NH									
HgR+	nmr	NaCl	25°C	0.50M	U	M	K1=10.0	1981BEa (34819)	1767
R=Hg(p-chlorobenzoate)									

C4H11NO3		L	Tris buffer				CAS 77-86-1	(550)	
2-Amino-2-(hydroxymethyl)-propan-1,3-diol; (HO.CH2)3C.NH2									
HgR+	nmr	NaCl	25°C	0.50M	U	M	K1=9.9	1981BEa (35057)	1768
R=Hg(p-chlorobenzoate)									

C4H12N2		L	Putrescine				CAS 110-60-1	(360)	
1,4-Diaminobutane; H2N.(CH2)4.NH2									
HgR+	gl	KNO3	22°C	0.10M	U	M		1977HSb (35366)	1769
K(MeHg+HL)=4.67, B2=10.23									
K(EtHg+HL)=4.44, B2=9.95									
K(PrHg+HL)=4.33, B2=9.91									

C5H5N		L	Pyridine				CAS 110-86-1	(31)	
Pyridine, Azine;									
HgR+	nmr	NaCl	25°C	0.50M	U	M	K1=8.3	1981BEa (36645)	1770

HgR+	kin	oth/un	20°C	0.10M	U				1973GEb (36646)	1771
K(CH3Hg+L)=4.72										

C5H8O3	HL	Laevulinic acid			CAS 123-76-2	(941)				
4-Ketopentanoic acid; CH3.CO.CH2.CH2.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

HgR+	gl	NaNO3	25°C	1.00M	U				1981JIa (38170)	1772
K(MeHg+L)=2.51										

C5H9NO3S	H2L	N-Acetyl-Cys			CAS 616-91-1	(1187)				
N-Acetylcysteine; CH3.CO.NH.CH(CH2.SH)COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

HgR+	gl	oth/un	22°C	?	U				1976HSa (38817)	1773
K(C6H5Hg+L)=7.29										
K(MeHg+L)=8.87										

C5H9NO4	H2L	Glutamic acid			CAS 56-86-0	(22)				
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

HgR+	dis	NaClO4	25°C	1.0M	C			K1=8.50	1988KSa (39090)	1774
Metal is C6H5Hg+. Method: extraction of 203Hg(C6H5)Br into benzene from a ligand solution in 1.0 M NaClO4.										

C5H10O2	HL	Pivalic acid			CAS 75-98-9	(3026)				
Trimethylethanoic acid, 2,2-Dimethylpropanoic acid; (CH3)3C.COOH										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

HgR+	nmr	oth/un	25°C	0.40M	U				1973LRa (40217)	1775
K(MeHg+L)=3.40										

C5H11N	L	Piperidine			CAS 110-89-4	(105)				
Perhydropyridine; cyclo(-CH2.CH2.CH2.NH.CH2.CH2-) C5H11N										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

HgR+	nmr	NaCl	25°C	0.50M	U	M		K1=10.2	1981BEa (40451)	1776
R=Hg(p-chlorobenzoate)										

C5H11NO2	L	Betaine			CAS 107-43-7	(4326)				
(Carboxymethyl)trimethylammonium hydroxide inner salt; (CH3)3.N.CH2.CO2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	0.40M	U			K(MeHg+L)=1.71	1973LRa (40467)	1777

C5H11NO2		HL		Valine				CAS 72-18-4 (43)		
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH										

HgR+	sp	oth/un	24°C	var	U			K(MeHg(OH)+HL=MeHgL+H2O)=2.09	1978SMa (40718)	1778
Medium; phosphate bufer, pH 6-8, ionic strength 0.2 to 0.3M.										

HgR+	nmr	none	25°C	0.0	U			K(MeHg+HL)=2.7 0 coord K(MeHg+L)=7.41 N coord	1974R0a (40719)	1779

C5H11NO2		HL						CAS 660-88-8 (1845)		
5-Aminopentanoic acid; H2N.CH2.CH2.CH2.CH2.COOH										

HgR+	nmr	none	25°C	0.0	U			K(MeHg+HL)=2.98 0 coord K(MeHg+L)=7.75 N coord	1974R0a (40859)	1780

C5H11NO2		HL		DL-Valine				CAS 516-06-3 (186)		
DL-2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2).COOH										

HgR+	gl	NaNO3	25°C	1.00M	C			K(MeHg+L)=7.268 K(MeHg+2L)=9.157	1978JIc (40894)	1781

C5H11NO2S		HL		Methionine				CAS 63-68-3 (42)		
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3)COOH										

HgR+	dis	NaClO4	25°C	1.0M	C			K1=8.42	1988KSa (41101)	1782
Metal is C6H5Hg+. Method: extraction of 203Hg(C6H5)Br into benzene from a ligand solution in 1.0 M NaClO4.										

HgR+	gl	NaNO3	25°C	1.00M	U			K(MeHg+L)=7.17	1981JIa (41102)	1783

HgR+	nmr	oth/un	25°C	0.20M	C			K1=7.40	1975FRb (41103)	1784
Method: 1H nmr. HgR is CH3Hg. Self-medium: 0.2 M CH3Hg, 0.03-0.29 M H2L.										

K1 from chemical shift values of CH₃Hg protons; from CH₂ protons, K₁=7.5.

C₅H₁₁N₂O₂S H₂L Penicillamine CAS 52-66-4 (350)
DL-2-Amino-3-mercapto-3-methylbutanoic acid; (CH₃)₂C(SH)CH(NH₂)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	KNO ₃	25°C	0.30M	C				1987Rba (41267)	1785
								K(HgR+L)=17.26 K(HgR+HL)=15.07		

Method: ¹H nmr. R is p-benzenesulfonate.

HgR+	nmr	oth/un	25°C	0.30M	M			K ₁ =16.94 K(MeHg _L +H)=8.14 K(MeHg _{HL} +H)=2.33	1981RRd (41268)	1786
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HgR+	gl	oth/un	22°C			? U			1976HSa (41269)	1787
								K(C ₆ H ₅ Hg+L)=11.86		

HgR+	gl	oth/un	22°C			? U			1976HSa (41270)	1788
								K(MeHg+L)=6.72 K(MeHg _L +L)=5.73		

HgR+	nmr	oth/un	25°C	0.20M	C				1975RFb (41271)	1789
								K(CH ₃ Hg _{HL} +H)=2.0 K(CH ₃ Hg _L +H)=9.0		

Self medium, ca. 0.2 M CH₃Hg⁺, 0.2 M H₂L. Method: ¹³C nmr.

C₅H₁₁N₂O₂Se HL CAS 1464-42-2 (1900)
2-Amino-4-(methylseleno)butanoic acid; CH₃.Se.CH₂.CH₂.CH(NH₂).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	KNO ₃	25°C	0.30M	U				1985IAa (41303)	1790
								K(MeHg+HL)=3.73 K(MeHg+L)=7.63		

C₅H₁₄N₂ L CAS 462-94-2 (359)
1,5-Diaminopentane; H₂N.(CH₂)₅.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO ₃	22°C	0.10M	U	M			1977HSb (41864)	1791
								K(MeHg+HL)=5.17, B ₂ =10.83 K(EtHg+HL)=4.93, B ₂ =10.59 K(PrHg+HL)=5.02, B ₂ =10.46		

C₆H₅N₂O₂S HL CAS 1849-36-1 (4397)
4-Nitrothiophenol; NO₂.C₆H₄.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	kin	oth/un	20°C	0.10M	U			K(CH ₃ Hg+L)=12.3	1973GEb (42710)	1792

C6H6S		HL		Thiophenol				CAS 108-98-5 (883)		
Phenyl mercaptan, thiophenol; C6H5.SH										

HgR+	nmr	non-aq	25°C	100%	U			K(MeHg+L)=14.1	1975BWc (44547)	1793
Medium: DMF										

C6H12N2O4S2		H2L		Cystine				CAS 923-32-0 (1404)		
DL-Dithio-bis(2-amino-3-propanoic acid); (HOOC.CH(NH ₂).CH ₂ .S) ₂										

HgR+	dis	NaCl04	25°C	1.0M	C			K1=8.77	1988KSa (49364)	1794
Metal is C6H5Hg+. Method: extraction of 203Hg(C6H5)Br into benzene from a ligand solution in 1.0 M NaCl04.										

C6H13NO2		HL						CAS 60-32-2 (1846)		
6-Aminohexanoic acid; H2N.CH2.CH2.CH2.CH2.CH2.COOH										

HgR+	nmr	none	25°C	0.0	U			K(MeHg+HL)=3.10 0 coord K(MeHg+L)=7.83 N coord	1974R0a (50217)	1795

C6H15NO3				Triethanolamine				CAS 102-71-6 (447)		
Tris-(2-hydroxyethyl)amine; L										

HgR+	nmr	NaCl	25°C	0.50M	U	M		K1=6.9	1981BEa (51295)	1796
R=Hg(p-chlorobenzoate)										

C6H15O3P		L						CAS 122-52-1 (1723)		
Triethylphosphite; (C2H5O)3P										

HgR+	nmr	non-aq	25°C	100%	U	M		K(MeHgA)+L)=-0.26	1981BRa (51512)	1797
Medium: CH2Cl2. HA=EtOH										

C6H16N2		L						CAS 124-09-4 (358)		

1,6-Diaminohexane; H2N.(CH2)6.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	22°C	0.10M	U				1977HSb (51586)	1798
								B(MeHgL2)=10.95, K(M+HL)=5.34		
								B(EtHgL2)=11.03, K(M+HL)=5.40		
								B(PrHgL2)=10.80		
								K(PrHg+HL)=5.21		

C7H8S2		L						CAS 14173-25-2	(2345)	
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Phenyl-methyl-disulfide; C6H5.S.S.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	non-aq	25°C	100%	U	M			1981BRa (56179)	1799
								K(MeHgA+L)=0.055		

Medium: CH2Cl2. HA=EtOH

C7H13NO3S		H2L						CAS 59-53-0	(1269)	
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N-Acetyl-penicillamine; CH3.CO.NH.CH(COOH)C(CH3)2SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	0.30M	M			K1=16.76	1981RRd (57492)	1800
								K(MeHgL+H)=3.10		

C8H17NO2		HL						CAS 2187-07-7	(1847)	
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8-Aminooctanoic acid; H2N.CH2.CH2.CH2.CH2.CH2.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	none	25°C	0.0	U				1974R0a (62748)	1801
								K(MeHg+HL)=3.15	O coord	
								K(MeHg+L)=7.60	N coord	

C8H18S		L						CAS 544-40-1	(2346)	
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Bis(n-butyl)sulfide; C4H9.S.C4H9

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	non-aq	25°C	100%	U	M			1981BRa (63007)	1802
								K(MeHgA+L)=-0.82		

Medium: CH2Cl2. HA=EtOH

C9H5NOClI		HL						CAS 130-26-7	(1541)	
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5-Chloro-7-iodo-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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HgR+ gl diox/w 25°C 75% C 1974ANa (63528)1803

K(MeHg+L)=9.2

K(PhHg+L)=9.8

Medium: 75% dioxan, 0.1 M NaClO4

C9H6N04IS H2L Ferron CAS 547-91-1 (275)

7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl KNO3 25°C 0.10M C 1974ANa (63806)1804

K(MeHg+L)=8.1

K(PhHg+L)=8.7

HgR+ sp oth/un 26°C 0.01M U K1=10.40 1973DNa (63807)1805

Metal: C6H5Hg+

C9H7N L CAS 91-22-5 (1538)

Quinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ gl KNO3 25°C 0.10M C 1974ANa (64062)1806

K(MeHg+L)=4.05

C9H7NO HL Oxine CAS 148-24-3 (504)

8-Hydroxyquinoline (8-quinolinol);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

HgR+ dis none 25°C 0.0 C 1984SNa (64280)1807

K(EtHg+L)=10.46

K(n-PrHg+L)=11.15

K(n-BuHg+L)=11.88

K(PhHg+L)=13.30

Method: extraction of RHg+ from dilute phosphate buffer into benzene/
8-hydroxyquinoline. Reaction: RHg+L=RHgL(org).

HgR+ gl KNO3 25°C 0.10M C I 1974ANa (64281)1808

K(MeHg+L)=8.8

K(MeHg+HL)<4

In 75% dioxan, 0.1 M NaClO4: K(MeHg+L)=11.5, K(MeHg+HL)=4.7

HgR+ sp oth/un 26°C 0.01M U K1=10.82 1973DNa (64282)1809

Metal : C6H5Hg, 25-27 C

C9H7N04S H2L Sulfoxine CAS 84-88-8 (448)

8-Hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	oth/un	25°C	?	U			K(MeHg+L)=13.7	1982RRb (67558)	1816
Medium: D20										

C10H7NO2		HL		Quinaldic acid	CAS	93-10-7		(2209)		
Quinoline-2-carboxylic acid;										

HgR+	gl	KN03	25°C	0.10M	U			K1=5.75	1974ANa (68711)	1817
Metal: MeHg+. For C6H5Hg+, K1=6.35										

C10H7NO2		HL			CAS	86-59-9		(873)		
Quinoline-8-carboxylic acid;										

HgR+	gl	KN03	25°C	0.10M	C			K(MeHg+L)=5.75 K(PhHg+L)=6.35	1974ANa (68762)	1818

C10H7NO3		H2L		Kynurenic acid	CAS	492-77-3		(1540)		
4-Hydroxy-2-quinolinecarboxylic acid;										

HgR+	gl	diox/w	25°C	75%	C	I		K(MeHg+L)=10.3 K(MeHg+HL)=4.6	1974ANa (68787)	1819
In 0.1(KN03): K(MeHg+L)=8.4										

C10H7NO4		H3L		Xanthurenic aci	CAS	59-00-7		(1539)		
4,8-Dihydroxy-2-quinolinecarboxylic acid;										

HgR+	gl	diox/w	25°C	75%	C			K(MeHg+HL)=6.6 K(MeHg+H2L)=4.1	1974ANa (68794)	1820

C10H8N2		L		2,2'-Bipyridyl	CAS	366-18-7		(25)		
2,2'-Bipyridine; (C5H4N)2										

HgR+	gl	KN03	25°C	0.10M	C			K(MeHg+L)=5.86	1974ANa (69587)	1821

C10H10N2	L	Benzylimidazole	CAS 4238-71-5	(2189)
Benzylimidazole;				

Medium: 0.02M Phosphate buffer.

C10H12N2	L	Tryptamine	CAS 61-54-1	(2190)
3-(2-Aminoethyl)indole;				

Medium: 0.04 M Phosphate buffer, pH 8. At 0.02 M, $K=2.69$; 0.005 M, $K=2.87$

C10H16N2O8	H4L	EDTA	CAS 60-00-4	(120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;				

HgR+ gl KNO₃ 22°C 0.10M U 1977HSb (73859)1828
K(PhHg+L)=9.15
K(PhHgL+L)=3.66
*K((PhHg)₂L)=-5.33

C10H17N3O6S	H3L	Glutathione	CAS 70-18-8 (333)
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Glutamyl-cysteinyl-glycine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	nmr	KNO3	25°C	0.30M	C			K(HgR+L)=15.91	1987RBa (75124)	1829

Method: 1H nmr. R is p-benzenesulfonate.

HgR+	nmr	oth/un	25°C	?	U			K1=16.00 K(MeHgL+H)=9.25 K(MeHgHL+H)=3.63 K(MeHg+HL)=15.85	1982RRb (75125)	1830
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Medium: D2O

HgR+	gl	oth/un	22°C	?	U			K(C6H5Hg+L)=7.09 K(C6H5HgL+H)=9.37	1976HSa (75126)	1831
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HgR+	gl	oth/un	22°C	?	U			K(MeHg+L)=8.11 K(MeHgL+H)=9.37	1976HSa (75127)	1832
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C11H12N2O2 HL Tryptophan CAS 73-22-3 (3)
2-Amino-3-(3-indolyl)propanoic acid; H2N.CH(CH2.C8H6N)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	sp	oth/un	24°C	0.02M	U			K(MeHg(OH)+HL=MeHgL+H2O)=3.92	1978SMa (78210)	1833

Medium: 0.02 M Phosphate buffer, pH 8. At 0.005 M, K=4.18

C12H8N2 L Phenanthroline CAS 66-71-7 (144)
1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	25°C	0.10M	U			K1=7.15	1974ANa (80466)	1834

Metal: CH3Hg+

HgR+	sp	oth/un	26°C	0.01M	U			K(C6H5Hg(H2O)+L)=8.24	1973DNa (80467)	1835
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Metal: C6H5Hg+

C12H28N2 L CAS 2783-17-7 (357)
1,12-Diaminododecane; H2N.(CH2)12.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	gl	KNO3	22°C	0.10M	U	M		K(MeHg+HL)=6.42	1977HSb (84143)	1836

K(EtHg+HL)=6.29

K(PrHg+HL)=6.28

K(PhHg+HL)=7.45

C13H12N4O L Diphenylcarbraz. CAS 538-62-5 (1195)

Diphenylcarbrazone; C6H5.NH.NH.CO.N:N.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	sp	non-aq	?	100%	U			K1=6.5 B2=11.90	1971TBa (85412)	1837
Metal: Hg(4-dimethylaminophenyl)+										
With Hg(4-diethylaminophenyl)+ : K1=6.2, K2=5.5										

C13H14N4 L CAS 13103-75-8 (473)

4-(2-Pyridylazo)-N,N-dimethylaniline; C5H4N.N:N.C6H4.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	kin	oth/un	20°C	0.10M	U				1973GEb (85684)	1838
K(CH3Hg+L)=5.75										

C13H16N2O2 H2L CAS 9479-05-2 (1104)

Tryptophan ethyl ester; C8H6N.CH2.CH(NH2).COOC2H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	sp	oth/un	25°C	0.01M	U				1980HMB (85936)	1839
K(MeHgOH+H2L=MeHgHL+H2O)=3.36										

Medium: 0.01 M phosphate, pH 8

C18H15O3PS HL CAS 16704-71-5 (3365)

3-Diphenylphosphino-benzene sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
HgR+	kin	oth/un	20°C	0.10M	U				1973GEb (97109)	1840
K(CH3Hg+L)=9.15										

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
 I Data with various BACKGROUNDS
 H Data for THERMOCHEMICAL quantities
 M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC
R or IUP=R signifies EVALUATION RATING = Recommended by IUPAC

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