```
START Experiments recorded for
   from SC-Database on Saturday, 01 January, 2000 at 00:05:55
Software version = 5.81 Data version = 4.62
Experiment list contains 52 experiments for
(no ligands specified)
Metal : Ta
(no references specified)
(no experimental details specified)
************************************
                HL Electron
e -
                                  (442)
Electron;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       sp NaClO4 25�C 0.10M U
                                        1971CKa
                                              (948)
                             K(Ta6Cl12 + e)=14.0 (830mV)
Medium: HC104. K=8.3(0.49V,(+++)). For Br complex, values are 15.0(0.89V) and
10.0(0.59V) (Esce=0.242V). Method:spectroscopy and current-voltage studies
______
       kin NaClO4 15♦C 0.10M U
                                        1966EMb
                                               (949) 2
Ta
                             K = -1.74
Medium 0.1M HClO4. By spectrophotometry, 22 C: K'=-1.80
                        Ta
      oth none 25�C 0.0 U
                                        1952LAb (950) 3
                             K=-68.6(-810 \text{ mV})
K: 0.5Ta2O5(s)+5H+5e=Ta(s)+2.5H2O. From thermodynamic data
*******************************
             HL Chloride CAS 7647-01-0 (50)
C1-
Chloride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
       ISE oth/un 175�C ? C
                                        1992BMa (5760) 4
                            K6=3.89
Medium: NaCl-AlCl3 melt.
                           М
Ta nmr oth/un -90♦C var U
                                        1971BIb (5761) 5
                             K'=0.52(cis)
                             K'=1.15(trans)
                             K''=2.39(cis)
                             K''=-1.35(trans)
K': 5TaF2L4=TaF6+4TaFL5. K": 5TaF3L3=2TaF6+3TaFL5. K(5TaF4L2=3TaF6+2TaFL5)=
-1.29(cis); -0.96(trans). Data also for other complexes and Br analogues
_____
       gl alc/w 25♦C 100% U
                                        1965GSd (5762) 6
                             K' = -3.55
                             K'' = -7.83
Medium: MeOH. K':TaCl5+MeOH=TaOMeCl4+H+Cl, K":TaCl5+2MeOH=Ta(OMe)2Cl3+2H+2Cl
______
```

```
Ta
      oth non-aq 300♦C 100% U T
                                   1959COa (5763) 7
                         K6=3.2 , x units
                         K(TaCl5(g)+Cl(melt))=1.8
Method: by partial pressure of TaCl5. Medium: liquid NaFeCl4.
K(TaCl5(g)+Cl(in melt))=TaCl6(in melt))=0.5(400 C) atm and x units
*********************************
            HL Fluoride CAS 7644-39-3 (201)
F-
Fluoride:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Та
     ISE non-aq ? 100% C
                                   1978GRa (7209) 8
                         K6=8.88
                         K(TaF5+TaF6=Ta2F11)=1.50
Medium: liquid anhydrous HF
_____
Ta sp oth/un ? ? U
                              1973LCa (7210) 9
                         K(TaH202+F)=2.39
                         B(TaH202+2F)=2.35
                         B(TaH202+3F)=1.93
                         B(TaH202+4F)=1.83
______
                         K1=6.37 B2=11.85 1972BAb (7211) 10
Ta
    ISE NaClO4 25�C 2.0M U
                         B3=16.03
                         B4=19.63
                         B5=23.29
                         B7 = 30.21
Metal: Ta(V)
            .....
Ta ix NaClO4 25♦C 1.0M U
                                  1969VAa (7212) 11
                         K4=5.90
                         K4K5=10.80
Metal: Ta(V). Method: quinhydrone electrode also
     .-----
                         1969VAa (7213) 12
Ta ix NaClO4 25�C 3.0M U
                         K4.K5.K6.K7=18.9
                         K4.K5.K6.K7.K8.K9=25.2
Medium: HClO4. Metal: Ta(V). Medium: quinhydrone electrode also used
______
     EMF NaClO4 25�C 1.0M U
                                   1966BFb (7214) 13
                         K6=3.75
                         K7 = 3.10
                         K8=0.66
Method: quinhydrone electrode
______
Ta
     dis NaClO4 25�C 3.0M U
                                   1965VWa (7215) 14
                         K4=5.86
                         K4K5=10.77
                         B6/B3=15.67
                         B7/B3=20.15
```

```
ix oth/un 25♦C 1.00M U
                                1962VFa (7216) 15
Ta
                       K5=4.8
                       K6=3.6
                       K7 = 3.3
                       K8 = 3.0
K9=3.6. Method: anion exchage and quinhydrone electrode.
H2L Peroxide CAS 7772-84-1 (2813)
02--
Peroxide; -0.0-
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp oth/un 20�C 78% U TIH
                                1973VZa (12699) 16
                       K(TaOSO4+H2L)=2.73
Medium: 78.4% H2SO4. K=2.80(15 C), 2.62(35 C), 2.55(55 C)
DH=-17 kJ mol-1 (TaO(SO4) assumed) also 63.5, 88.7, 94.5%
______
      sp oth/un 0♦C 90% U
                                1969CKa (12700) 17
                       K(Ta0S04+H2L)=3
Medium:H2SO4
______
      sp oth/un 0�C 10% U I
                                1969VZa (12701) 18
                       K(TaOSO4+H2L)=1.48
Medium: 10\% H2SO4 K=1.48(20%), 2.20(30%), 2.59(50%), 2.80(70%), 3.38(80%),
3.59(100%)
_____
     sp non-aq ? 100% U
                                1968VZa (12702) 19
                       K(Ta(V)+H2L)=3.43
Medium: H2SO4
______
     vlt oth/un 25�C 0.34M U
                                1964BRb (12703) 20
                       B(HTaO3+H2L)=2.0
Medium: H2SO4
***********************************
                Thiocyanate CAS 463-56-9 (106)
SCN-
             HL
Thiocyanate;
          Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp alc/w ? 100% U I K1=3.12 B2=5.48 1964GSa (15264) 21
Ta
                       B3=7.77
Medium: MeOH. In BuOH: K1=3.68, B2=7.05, B3=11.42. In Me2NCHO: K1=3.15,
B2=5.92, B3=8.55, B4=11.06, B5=13.52, B6=15.96
****************************
                Methyl alcohol CAS 67-56-1 (597)
CH40
Methanol; CH3.OH
 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
EMF alc/w 20♦C 100% U
                                      1965GBa (17902)
Ta
                           K(TaA(L')4+2L'=Ta(L')6+A)=7.1
                           K(Ta(H-1L)4+A)=11.04
                           K(TaA(H-1L)3+TaH-1L)=12.95
                           K(Ta(L')5+HA=TaA(L')4+L)=4.36
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. HA=acetylacetone, L'=H-1L
______
      EMF alc/w 20�C 100% U
                         Μ
                                      1965GBa (17903) 23
Ta
                           K' = 14.2
                           K''=7.85
                           K(TaA(L')3+L'=TaA(L')4)=9.04
                           K'''=2.5
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl; H2A=catechol; L'=H-1L. K':
TaA(L')3+H2A+L'=TaA2(L')2+2L. K":TaA2(L')2+H2A+L'=TaA3L'. K'":TaAL'4+TaAL'3
______
    EMF alc/w 20�C 100% U
Ta
                                      1964GUa (17904) 24
                           K(Ta(H-1L)4+H-1L)=11.47
                           K(Ta(H-1L)5+H-1L)=6.67
                           K(Ta(H-1L)6+H=Ta(H-1L)5+L)=5.1
                           K(Ta(H-1L)7+H=Ta(H-1L)6+L)=9.9
Method: H electrode; medium: MeOH, 1.0 M Me4NCl
********************************
C2H2O4
              H2L
                   Oxalic acid
                              CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sol oth/un ? 0.10M U
                                      1970ZPa (19077) 25
                           K3=5.91
Medium: HC104
Metal ion is TaO+++
______
      sol oth/un 19�C ? U
                                      1965BLd (19078) 26
                           K(Ta(OH)2+L)=11.10
                           K(Ta(OH)2+2L)=18.52
                           K(Ta(OH)2L+OH)=13.33
****************************
                   Cyanomethane CAS 75-05-8 (1399)
C2H3N
Acetonitrile; CH3.CN
------
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      nmr non-aq -60♦C 100% U M
Ta
                                      1974GMa (19196) 27
                          K(TaBr5A+L=TaBr5L+A)=-0.89
Medium: CH2Cl2. A=t-butylcyanide
______
       nmr non-aq -40♦C 100% U
                                      1972MBb (19197) 28
                           K(TaC15A+L=TaC15L+A)=0.32
Medium: CHCl3. A=dimethylether. K=0.36, A= 1,4-dioxan;
K=1.57, A=diethylether; K=0.70, A=1,4-dithiane.
```

```
************************************
C2H6NOC12P
                             CAS 667-43-0 (910)
Dichloro(dimethylamine)phosphine oxide; (CH3)2N.P(0)Cl2
  .-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
------
      nmr non-aq -60♦C 100% U M
                                     1974GMa (21901) 29
                         K(TaBr5A+L=TaBr5L+A)=1.48
Medium: CH2Cl2, A=acetonitrile
*****************************
                            CAS 115-10-6 (4214)
Dimethyl ether; CH3.0.CH3
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
   nmr non-aq -40♦C 100% U M
                                     1972MBb (22021) 30
                          K(TaC15A+L=TaC15L+A)=1.25
Medium: CHCl31. A=diethyl ether. K=0.04, A=dioxan. Metal ion: Ta(V)
*******************************
                           CAS 75-18-3 (151)
C2H6S
Dimethyl sulfide; CH3.S.CH3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                        М
Ta nmr non-ag -60♦C 100% U
                                     1974GMa (22195) 31
                        K(TaC15A+L=TaC15L+A)=0.72
Medium: CH2Cl2, A=pivalocyanide
     nmr non-ag -60♦C 100% U
                                    1974GMa (22196) 32
                          K(TaBr5A+L=TaBr5L+A)=0.24
Medium: CH2Cl2, A=pivalocyanide
                         M 1972MBb (22197) 33
      nmr non-aq -40♦C 100% U
                          K(TaC15A+L=TaC15L+A)=0.26
Medium: CHCl3. A=t-butylnitrile. K=0.40, A=acetonitrile, K=1.98, A=diethyl
ether. K=0.72, A=diethyl sulfide. K=0.72, A=dimethyl sulfide.
                        M 1972MBb (22198) 34
   nmr non-aq -60≎C 100% U
                          K(TaBr5A+L=TaBr5L+A)=0.24
Medium: CH2Cl2. A=pivalonitrile.
**********************************
              L DiMeSelenide CAS 81369-92-3 (911)
Dimethylselenide; CH3.Se.CH3
-----
      Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
______
      nmr non-ag -60♦C 100% U
                                     1974GMa (22207) 35
                          K(TaC15A+L=TaC15L+A)=0.06
                          K(TaBr5A+L=TaBr5L+A)=0.43
                          K(TaBr5B+L=TaBr5L+B)=0.67
```

```
Medium: CH2Cl2, A=dimethylthioether, B=t-butylcyanide
*******************************
                 DiMeTelluride
                           CAS 593-80-6 (912)
Dimethyltelluride; CH3.Te.CH3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      nmr non-ag -60♦C 100% U
                       М
                                  1974GMa (22209) 36
                        K(TaC15A+L=TaC15L+A)=0.25
                        K(TaCL5B+L=TaCl5L+B)=0.31
                        K(TaBr5A+L=TaBr5L+A)=0.95
Medium: CH2Cl2, A=selenobismethane, B=thiobismethane
********************************
                L-Tartaric acid CAS 87-69-4 (92)
             H2L
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values
                                   Reference ExptNo
______
      dis oth/un 22�C 3.00M U
                                  1972SSj (31366) 37
                        K(Ta(OH)4+L)=0.72
                        K(Ta(OH)4+H-1L)=8.83
                        K(Ta(OH)4+H-2L)=15.3
********************************
                 1,4-Thioxane
                          CAS 15980-15-1 (4266)
1,4-Oxathiane; cyclo(-0.CH2.CH2.S.CH2.CH2-)
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                       М
     nmr non-aq -60�C 100% U
                                  1972MBb (33191) 38
                        K(TaC15L+A=TaC15A+L)=0.70
Medium: CHCl3. A=t-butyl nitrile
***********************
C4H8S
                           CAS 110-01-0 (150)
Tetrahydrothiophene; cyclo(-CH2.CH2.S.CH2.CH2-)
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     nmr non-aq -60♦C 100% U
                                  1972MBb (33742) 39
                        K(TaC15A+L=TaC15L+A)=0.63
A=t-butyl mercaptan. Medium: CHCl3
****************************
                          CAS 505-29-3 (4255)
                1,4-Dithiane
1,4-Dithiane; cyclo-(S.CH2.CH2.S.CH2.CH2-)
______
     Mtd Medium Temp Conc Cal Flags Lg K values
                                   Reference ExptNo
______
      nmr non-aq -60♦C 100% U
                                  1972MBb (33744) 40
                        K(TaC15A+L=TaC15L+A)=0.85
A=t-butyl nitrile. Medium: CHCl3
********************************
```

```
C4H10S
                          CAS 352-93-2 (4259)
Diethyl sulfide; C2H5.S.C2H5
_____
      Mtd Medium Temp Conc Cal Flags Lg K values
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                         М
     nmr non-aq -60�C 100% U
                                     1974GMa (34722) 41
                          K(TaBr5A+L=TaBr5L+A)=-0.61
A=t-butyl nitrile. Medium: CH2Cl2
______
      nmr non-aq -40♦C 100% U
                                     1972MBb (34723) 42
                         K(TaC15A+L=TaC15L+A)=0.00
A=dimethyl ether. Medium: CHCl3.
***********************************
                  Acetylacetone CAS 123-54-6 (164)
              HL
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3
_____
      Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
______
      EMF non-aq 20�C 100% U
                                     1971GSa (38090) 43
                          K(TaA5+HL=TaA4L+HA)=4.36
                          K(TaA3L+A)=12.95
                          K(TaA4L+2A=TaA6+L)=7.10
Medium: MeOH. HA=MeOH
************************************
                  t-Butylnitrile CAS 7188-38-7 (913)
               L
t-Butylcyanide; (CH3)3C.CN
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                        М
      nmr non-aq -60�C 100% U
Ta
                                     1974GMa (38456) 44
                          K(TaBr5A+L=TaBr5L+A)=0.11
                          K(TaBr5B+L=TaBr5L+B)=0.39
                          K(TaBr5C+L=TaBr5L+C)=1.72
Medium: CH2Cl2, A=acetonitrile, B=thiobismethane, C=dimethylether
______
Ta
      nmr non-aq -40♦C 100% U
                         М
                                     1972MBb (38457) 45
                          K(TaCl5A+L=TaCl5L+A)=0.47
Medium: CHCl3. A=dimethyl ether. When A=cyanomethane, K=0.15,
A=1,4-dioxan (-60 C), K=0.51
********************************
                             CAS 120-80-9 (534)
              H2L
                  Catechol
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      EMF alc/w 20�C 100% U
                        М
                                     1971GSa (43837) 46
                          K(TaA3L+A)=9.04
                          K(TaA3L+TaA4L=Ta2A7L2)=2.50
                          K(TaA3L+H2L+A=TaA2L2+2HA)=14.2
                          K(TaA2L2+H2L+A=TaAL3+2HA)=7.85
```

```
Medium: MeOH, 1.0 M Me4NCl. HA=CH3OH
*******************************
               Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
   sp oth/un ? ? U K1=9.5 1966SAb (45660) 47
****************************
               EDTA
                         CAS 60-00-4 (120)
C10H16N2O8
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
vlt oth/un 20�C 1.08M U
                                1969VSb (74191) 48
                       K(Ta(OH)2+L)=33.6
Medium: K2SO4
**********************************
       H2L
                         CAS 1141-59-9 (636)
               PAR
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
___________
     sp oth/un 25♦C ? C K1=5.77
                                1980LZb (77582) 49
Medium: hexamethylenetetramine ((CH2)6N4) buffer solution, pH 5.6
_____
                ; П
     sp oth/un 25♦C
                               1967ADa (77583) 50
                      K(?)=4.5
******************************
                    CAS 83-61-4 (950)
            H3L DASA
C14H807S
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
------
     sp oth/un ? ? U
                               1968ADa (86757) 51
                      B((TaO)L2)=8.27
**********************
C17H17N03
                         CAS 58434-59-6 (1213)
2'-Hydroxy-4-methoxy-5'-methylbenzylidene acetophenone oxime
__________
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
sp oth/un 30≎C 8.00M U
Ta
                                1980GKa (96191) 52
                       K(TaO(SCN)+L)=3.56
                       K(TaO(SCN)L+L)=2.07
REFERENCES
 1992BMa J von Barner, L McCurry et al; Inorg. Chem., 31, 1034 (1992)
```

1980GKa S Gholse, R Kharat; Indian J.Chem., 19A, 823 (1980)

```
1980LZb Luo Zongming; Acta Chimica Sinica, 38, 433 (1980)
 1978GRa R Gut, J Rueede; J.Coord.Chem., 8,47 (1978)
 1974GMa R Good, A Merbach; Helv.Chim.Acta, 57, 1192 (1974)
 1973LCa G Latysh, A Chernyak, T Serebrennikova; Zh. Neorg. Khim., 18, 1014(E:533)
(1973)
 1973VZa V Vasilev, G Zaitseva; Zh. Neorg. Khim., 18, 139(E:70) (1973)
 1972BAb E Baumann; J.Inorg.Nucl.Chem., 34,687 (1972)
 1972MBb A Merbach, J Bunzli; Helv.Chim.Acta, 55,580 (1972)
 1972SSj G Shabanova, N Skorik; Zh.Obshch.Khim., 42, 204 (1972)
 1971BIb Y Buslaev, E Ilin, M Krutkina; Dokl. Akad. Nauk SSSR, 200, 1345(E:850) (1971)
 1971CKa N Cook, T Kuwana, J Espenson; Inorg. Chem., 10, 1081 (1971)
 1971GSa R Gut, E Schmid, J Serrallach; Helv. Chim. Acta, 54, 593; 609 (1971)
 1970ZPa E Zhurennikov, D Pobezhimoskaya; Radiokhim., 12,1,105 (1970)
 1969CKa A Chernyak, V Khomutnikov, A Batsuev et al; Zh. Neorg. Khim., 14,1251(E:655)
(1969)
 1969VAa L Varga; Anal.Chem., 41,323 (1969)
 1969VSb G Volkova, V Sochevanov; Zh. Neorg. Khim., 14,5,1245 (1969)
  1969VZa V Vasilev, G Zaitseva; Zh. Neorg. Khim., 14, 198(E:102) (1969)
  1968ADa B Agarwala, A Dey; Chim. Anal. (Paris), 50, 233 (1968)
  1968VZa V Vasilev, G Zaitseva; Zh. Neorg. Khim., 13,84 (1968)
 1967ADa B Agarwala, A Dey; Curr.Sci., 36,544 (1967)
 1966BFb M Bukhsh, J Flegenheimer, F Hall et al; J. Inorg. Nucl. Chem., 28,421 (1966)
 1966EMb J Espenson, R McCarley; J.Am. Chem. Soc., 88, 1063 (1966)
  1966SAb K Stolyarov, I Amantova; Vestnik Leningr. Univ., 4, 141; 155; 10, 133 (1966)
  1965BLd A Babko, V Lukachina, B Nabivanets; Zh. Neorg. Khim., 10,467 (865) (1965)
 1965GBa R Gut, H Buser, E Schmid; Helv. Chim. Acta, 48,878 (1965)
  1965GSd A Golub, A Sych; Zh. Neorg. Khim., 10,889 (1965)
 1965VWa L Varga, W Wakley, L Nicolson et al; Anal. Chem., 37, 1003 (1965)
 1964BRb L Budarin, T Rumyantseva, T Sherina; Izv. VUZ. Khim., 7,715 (1964)
 1964GSa A Golub, A Sych; Izv. Akad. Nauk Latv. SSR, 387 (1964)
 1964GUa R Gut; Helv.Chim.Acta,47,2262 (1964)
  1962VFa L Varga, H Freund; J. Phys. Chem., 66, 21 (1962)
  1959COa C Cook; J.Am.Chem.Soc., 81,535 (1959)
 1952LAb W Latimer; "Oxidation Potentials", Prentice Hall, NY (1952)
EXPLANATORY NOTES
  DATA Flags are :-
        T Data at other TEMPERATURES
        I Data with various BACKGROUNDS
        H Data for THERMOCHEMICAL quantities
        M Data for TERNARY Complexes
```

END Experiments recorded for

from SC-Database on Saturday, 01 January, 2000 at 00:05:55