

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 686 experiments for

(no ligands specified)

2 metals : Pd(IV), Pd++

(no references specified)

(no experimental details specified)

e- HL Electron (442)

Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	EMF	none	18°C	0.0	U				1924JIa	(793) 1

K=42.3(1220mV)

K'=32.9(950mV)

K: PdO₃(s)+2H+2e=PdO₂(s)+H₂O. K': PdO₂(s)+2H+2e=PdO(s)+H₂O

Br- HL Bromide CAS 10035-10-6 (19)

Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	EMF	NaClO ₄	25°C	0.40M	U				1971DUa	(2221) 2

K₅=3.48K₆=2.64Medium: HClO₄

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	EMF	NaClO ₄	25°C	0.40M	U				1971DUa	(5441) 3

K₅K₆=4.22Medium: HClO₄

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd(IV)	sol	NaCl	25°C	1.0M	U				1930WEa	(5442) 4

K(K₂PdL₆(s)=2K+PdL₆)=-5.22

e- HL Electron (442)

Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	none	25°C	0.00	U				1971JPa	(794) 5

K(Pd + 2e=Pd(s))=30.8(0.91V)

Pd++	EMF	oth/un	25°C	4.00M	U T				1970IEa	(795) 6
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$K(\text{Pd} + 2\text{e} = \text{Pd}(\text{s})) = 33.1(979\text{mV})$

Medium: HClO_4 . $K = 34.8(978\text{mV}, 10^\circ\text{C})$, $34.2(979\text{mV}, 15^\circ\text{C})$, $33.5(975\text{mV}, 20^\circ\text{C})$, $32.5(978\text{mV}, 30^\circ\text{C})$, $31.8(972\text{mV}, 35^\circ\text{C})$, $30.9(960\text{mV}, 40^\circ\text{C})$ (m units)

Pd++	oth none	25°C	0.0	U	K'=16.2 (480mV)	1968GHa	(796)	7
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Method:Literature evaluated data. K' : $\text{Pd(IV)}\text{I}_6 + 2\text{e}^- = \text{PdI}_4 + 2\text{I}^-$

Method:Literature evaluated data

Method: Literature evaluated data. $K(\text{PdBr}_4 + 2\text{e} = \text{Pd}(\text{s}) + 4\text{Br}) = 16.6 (0.49\text{V})$,
 $K(\text{PdI}_4 + 2\text{e} = \text{Pd}(\text{s}) + 4\text{I}) = 6.1 (0.18\text{V})$

Medium: HClO₄. I=3.46: K=32.19, 952 mV; I=2.22: K=31.25, 924.4 mV;
I=1.06: K=31.11, 920 mV

By calorimetry, 0.1 M NaI: $\Delta H(\text{Pd} + 3\text{I} = \text{Pd(s)} + \text{I}_3^-) = -104.1 \text{ kJ mol}^{-1}$

$$K=23.40, 692 \text{ mV (X=Br)}$$

$K=21.13, 625 \text{ mV (X=I)}$

$$\text{K: Pd(en)}_2\text{X}_2 + 2\text{e}^- = \text{Pd(en)}_2 + 2\text{X}^-$$

K: $\text{Pd(II)Br}_4 + 2\text{e} = \text{Pd(s)} + 4\text{Br}$. From thermodynamic data
 $\text{K(PdO}_2\text{(s)} + \text{H}_2\text{O} + 2\text{e} = \text{PdO(s)} + 2\text{OH}) = 25(730 \text{ mV})$ estimated

Medium: HCl. K: $\text{PdCl}_4 + 2\text{e} = \text{Pd(s)} + 4\text{Cl}$. At 15 C: $K = 21.8(623 \text{ mV})$, 35 C: $20.3(619 \text{ mV})$. In 4 M HClO_4 : $K(\text{Pd(II)} + 2\text{e} = \text{Pd(s)}) = 33.4(987 \text{ mV})$

$\text{K}(\text{Pd}(\text{IV})\text{Cl}_6 + 2\text{e}^- = \text{Pd}(\text{II})\text{Cl}_4 + 2\text{Cl}_2)$. In 1 M HCl: $K = 43.5$ (1286 mV). In 1 M KBr: $\text{K}(\text{PdBr}_6 + 2\text{e}^- = \text{PdBr}_4 + 2\text{Br}_2) = 33.6$ (994 mV). 1 M KI: $\text{K}(\text{PdI}_6 + 2\text{e}^- = \text{PdI}_4 + 2\text{I}_2) = 16.3$ (482 mV)

Medium: HCl. K: $\text{Pd(IV)Cl}_6 = \text{Pd(II)Cl}_4 + \text{Cl}_2(\text{aq})$. Method: chemical analysis

$K(\text{Pd(IV)Cl}_6 + 2e = \text{Pd(II)Cl}_4 + 2\text{Cl}) = 43.56 (1288 \text{ mV})$ from thermodynamic data

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	1.0M	U	M	1973ELa (2222) K(cis-trans-PdL2(H2O)2)=0.78 K2(cis)=4.19 K2(trans)=3.41 K3(cis)=3.37	(2222)	17
Medium: HClO4. K3(trans)=4.15. Kn: PdL2(H2O)2+nL									
Pd++	sp	NaClO4	?	1.0M	U		K1=2.23	1973GSc (2223)	18
Pd++	sp	NaClO4	25°C	1.0M	U		K1=5.17 B2=9.42 B3=12.7 B4=14.9	1972ELa (2224)	19
Medium: HClO4									
Pd++	sp	NaClO4	25°C	4.50M	U	M	1972FKa (2225) K(PdCl4+L=PdCl3L+Cl)=1.40 K(PdCl3L+L=PdCl2L2+Cl)=1.06 K(PdCl2L2+L=PdClL3+Cl)=0.72 K(PdClL3+L=PdL4+Cl)=0.27	(2225)	20
Medium: LiClO4									
Pd++	cal	NaClO4	25°C	1.0M	U	H	1972RHa (2226)	(2226)	21
Medium: HClO4. DH(K1)=-21.3 kJ mol-1, DS(K1)=27.2 J K-1 mol-1									
Pd++	ISE	diox/w	25°C	71%	U	TI	1968GFc (2227)	(2227)	22
B4=19.0 B4=16.2(0%), also B4 for several other % dioxan. At 40 C: B4=18.1(71%) 15.3(0%)									
Pd++	sol	NaClO4	20°C	0.10M	U		K1=6.8	1967GGa (2228)	23
Pd++	cal	oth/un	25°C	0.10M	U	H	1967IWa (2229)	(2229)	24
Medium:NaBr. DH(B4)=-54.8 kJ mol-1									
Pd++	gl	NaClO4	var	var	U		1967KPc (2230) K(PdBr3OH+Br=PdBr4+OH)=-4.23	(2230)	25
19-50 C, I=0.1-1.0									
Pd++	sp	NaCl	25°C	1.0M	U		1966BSa (2231) B4=13.05	(2231)	26
Pd++	sp	NaClO4	45°C	1.80M	U	T H	1966SBb (2232) K4=2.16	(2232)	27

K4=2.50(10 C),2.30(25 C). DH(K4)=-18.0 kJ mol⁻¹, DS=-14.6 J K⁻¹ mol⁻¹

Pd++ ISE oth/un 25°C var U 1965FKa (2233) 28

B4=14

Medium:KBr var. Also values for B4 at 10-60C assuming same K and 1/RTF as for 25C!

Pd++ sol oth/un 20°C 0.60M U I 1964PBa (2234) 29

B3=11.28

B4=13.42 ?

Kso=-12.54

K(PdL2(s)=PdL2)=-4.4

At I=0.4: Kso=12.96,K=-4.5, K(PdL2(s)+L)=-1.36, B3=11.60, B4=13.40?, K3=3.1, K4=1.8

Pd++ sp NaClO4 25°C 0.50M U 1964SBe (2235) 30

K4=2.20

Pd++ sp NaClO4 20°C 0.80M U K1=4.37 1964SLb (2236) 31

K4=3.50

Medium:0.8(ClO4),0.6 H+. By hypothesis method:K2=4.08, K3=3.79

Pd++ ISE oth/un 19°C var U 1963GKa (2237) 32

B4=16.1

Pd++ oth none 25°C 0.0 U 1952LAb (2238) 33

B4=13.10

Method: from thermodynamic data; I=0 corr.

CN- HL Cyanide CAS 74-90-8 (230)

Cyanide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ ISE oth/un 25°C ? C 1976HEb (2752) 34

B4=63 (60<B4<65)

K(Pd(CN)2(s)+2CN)=20.8

Kso(Pd(CN)2)=-42

Pd++ sol NaClO4 20°C 0.10M U K1=10.5 1967GGa (2753) 35

Pd++ ISE oth/un 25°C 0.0 U H 1967IWa (2754) 36

B4=42.4

K5=2.9

Medium:0 corr. By calorimetry:DH(B4)=-385.8 kJ mol⁻¹, DS=-485 J K⁻¹ mol⁻¹; DH(K5)=-0.8, DS=33. DH(PdBr4+4L=PdL4+4Br)=-329

Pd++ ISE oth/un 25°C var U T 1965FKa (2755) 37

B4=51.6

K(Pd+2e=Pd(s))=33.4

Medium: KCN var

C03-- H2L Carbonate CAS 465-79-6 (268)

Carbonate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaCl 25°C 0.11M C I M 2003CBa (3345) 38

Data for 0.105-1.0 M NaCl + H3BO3. $K(\text{PdCl}_4 + \text{HCO}_3 = \text{Pd}(\text{CO}_3)\text{Cl}_3 + \text{H} + \text{Cl}) = -6.68$

$K = -6.50$ (I=0.305), -6.62 (I=0.505), -6.71 (I=0.705), -6.95 (I=1.005)

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaCl 25°C 0.11M C I M 2003CBa (5443) 39

$K(\text{PdCl}_3 + \text{Cl}) = 1.08$

Data for 0.105-1.0 M NaCl, pH 3.0-8.5. $K(\text{PdCl}_4 + \text{H}_2\text{O} = \text{Pd}(\text{OH})\text{Cl}_3 + \text{H} + \text{Cl}) = -8.72$.

$K = -8.89$ (I=0.305), -8.97 (I=0.505), -8.98 (I=0.705), -8.96 (I=1.005)

Pd++ sp NaCl 25°C 0.50M C 2000BYa (5444) 40

$*K(\text{PdCl}_4) = -8.98$

*K: $\text{PdCl}_4 + \text{H}_2\text{O} = \text{PdCl}_3(\text{OH}) + \text{H}$.

Pd++ sol KCl 25°C 0.10M C TI M 1999VWa (5445) 41

$B_4 = 11.81$

$K(\text{Pd} + 3\text{Cl} + \text{OH}) = 20.21$

Data for 20 C and I=0.8 m and for 25 C and I=0.1-1.0 m. At I=0, $B_4 = 11.29$,

$K(\text{Pd} + 3\text{Cl} + \text{OH}) = 20.29$.

Pd++ sp NaNO3 37°C 0.16M C M 1998ESa (5446) 42

$K(\text{PdA}(\text{H}_2\text{O})_2 + \text{Cl}) = 3.563$

$K(\text{PdA}(\text{H}_2\text{O})\text{Cl} + \text{Cl}) = 2.28$

A is 1,3-diaminopropane.

Pd++ gl NaCl04 37°C 0.15M C M 1996GTa (5447) 43

$K(\text{PdA}(\text{H}_2\text{O})_2 + \text{L}) = 3.65$

$K(\text{PdA}(\text{H}_2\text{O})_2 + 2\text{L}) = 5.86$

$*K(\text{PdA}(\text{H}_2\text{O})_2 + \text{L}) = -2.68$

A=diaminosuccinate diethylester, $\text{EtO}_2\text{CCH}(\text{NH}_2).\text{CH}(\text{NH}_2)\text{CO}_2\text{Et}$

*K: $\text{PdA}(\text{H}_2\text{O})_2 + \text{L} = \text{PdA}(\text{H}_2\text{O})(\text{OH})\text{L} + \text{H}$

Pd++ sol NaCl 100°C 1.0M U T 1995GAa (5448) 44

$K_3 = 0.30$

Method: solubility of AgCl in Pt solution, 0.03-3.0 m HCl.

At 200 C, $K_3 = 1.20$, at 300 C, $K_3 = 1.36$

Pd++ kin NaCl04 25°C 0.10M U M 1993SHa (5449) 45

$K_{\text{out}}(\text{PdABH}_2\text{O} + \text{L}) = 2.28$

A=N,N,N',N'-Tetraethyldiaminoethane, B=Inosine, C=Inosine-5'-monophosphate

Pd++

sp

oth/un

19°C

var

U

TI

1991TJa

(5450)

46

K3=2.60

K4=1.25

19-90 C. Constants at I=0

Pd++

nmr

non-aq

24°C

100%

U

IHM

1982HBa

(5451)

47

K(PdI2+PdL2=2PdIL)=0.79

K(PdBr2+PdL2=2PdBrL)=0.63

K(PdCl2+PdL2=2PdClL)=0.61

Medium: CH2Cl2; Pd as Pd2(bis(diphenylphosphino)methane)2

For iodide complex, DH=-5.0 kJ mol-1, DS=12.6 J K-1 mol-1

Pd++

oth

NaClO4

25°C

0.0

M

I

K1=5.08

B2= 8.88

1980KRa

(5452)

48

K3=2.42

K4=0.88

Analysis of literature data using Pitzer coefficients. Data for 0.05 to 2.0 M NaClO4. Equation given for ionic strength dependence.

Pd++

sp

NaClO4

25°C

0.86M

U

K1=4.0

B2=7.2

1976YBa

(5453)

49

K3=2.3

When I=0.1 M NaClO4: K1=6.0, K2=4.6, K3=2.5

Pd++

sp

non-aq

20°C

100%

U

I

1974V0a

(5454)

50

K(Pd2L4+2L=Pd2L6)=6.4

K(Li+Pd2L6)=1.7

Medium: MeCN, LiCl at different concentrations. With Me4NCl, values are: 7.8, 1.5

Pd++

kin

NaClO4

25°C

1.0M

U

M

1973ELa

(5455)

51

K2(cis)=3.11

K2(trans)=2.79

K3(cis)=2.59

K3(trans)=2.90

Medium: HClO4. K(cis-PdL2(H2O)2=trans-PdL2(H2O)2)=0.32

Pd++

sp

NaClO4

?

1.0M

U

1973GSc

(5456)

52

K4=1.27

Pd++

sp

non-aq

?

100%

U

M

1973KFa

(5457)

53

K(PdBr4+L=PdBr3L+Br)=1.24

K(PdBr3L+L=PdBr2L2+Br)=1.84

K(PdBr2L2+L=PdBrL3+Br)=2.50

K(PdBrL3+L=PdL4+Br)=2.39

Medium: MeCN, 1.5 M Bu4N(Cl,Br)

Pd++

sp

NaClO4

25°C

1.0M

U

K1=4.47

B2=7.76

1972ELa

(5458)

54

B3=10.2

B4=11.5

Medium: HClO4

Pd++ cal NaClO4 25°C 1.0M U H 1972RHa (5459) 55
 Medium: HClO4. DH(K1)=-12.7 kJ mol⁻¹, DS=43.1 J K⁻¹ mol⁻¹; DH(K2)=-10.8,
 DS=26.8; DH(K3)=-10.7, DS=10.0; DH(K4)=-14.2, DS(K4)=-21.8

Pd++ oth non-aq 37°C 100% U M 1971HMB (5460) 56
 K(Li2Pd2L6+2LiL=2Li2PdL4)=-1.0

Medium: CH3COOH. Method: vapor phase osmometry

Pd++ vlt NaClO4 25°C 0.20M U 1971JPa (5461) 57
 B3=7.94
 K4=1.44

Medium: HClO4

Pd++ EMF oth/un 25°C 3.0M U 1971KMh (5462) 58
 K3=1.76
 K4=2.35

Medium: H2SO4

Pd++ sp NaClO4 ? 1.0M U K1=3.48 B2=6.27 1970RGa (5463) 59
 K3=2.35
 K4=1.1

Pd++ EMF oth/un ? var U K1=4.7 B2=7.70 1969GKd (5464) 60
 K3=2.6
 K4=1.6

Pd++ EMF NaClO4 25°C 1.0M U 1969KSc (5465) 61
 B4=12.15

Medium: H(ClO4,S04)

Pd++ ISE diox/w 25°C 72% U TI K1=17.7 1968GFC (5466) 62
 Also B4 for several dioxan percentages. At 40 C: B4=16.6(72% dioxan)

Pd++ sp NaClO4 25°C 4.0M U TI 1968LEc (5467) 63
 K4=2.00

Medium: LiClO4. K4=1.77(I=3),1.59(I=2),1.43(I=1)
 At I=2: K4=1.68(15 C),1.59(25 C),1.51(40 C)

Pd++ ISE NaClO4 25°C 3.40M U I 1968LMb (5468) 64
 B4=11.4

Medium: HClO4. By spectrophotometry:K4=1.77(I=3.4), 1.44(I=1.07)

Pd++ con oth/un 25°C dil U 1967CMb (5469) 65
 K(Pd(NH3)2L+L)=2.55 ?

Pd++ sol NaClO4 25°C 0.10M U K1=5.1 1967GGa (5470) 66

Pd++	cal	NaCl	25°C	0.10M	U	H		1967IWa	(5471)	67
DH(B4)=-23.0 kJ mol ⁻¹										

Pd++	gl	NaCl04	var	var	U			1967KPc	(5472)	68
							K(PdCl3OH+Cl=PdCl4+OH)=-5.7			

Pd++	gl	R4N.X	25°C	var	U	T		1967RBc	(5473)	69
							K(Pd(NH3)2L+L)=2.33			
							K(Pd(NH3)3+L)=3.0			
Medium:NH4(NO3). Also other constants and values at 30 C by spec.										

Pd++	sp	oth/un	25°C	1.0M	U	T	H	K1=4.00	B2=7.49	1966SBb (5474) 70
							B3=9.73			
							B4=11.11			
Med:1.0(NaCl04),0.8 H+. K4=1.50(10 C), 1.42(25 C), 1.28(45 C). DH(B4)=-11.7 kJ mol ⁻¹ , DS=-12.1 J K ⁻¹ mol ⁻¹										

Pd++	ISE	KCl	25°C	1.0M	U			1965FKa	(5475)	71
							B4=11.8			
							K(Pd+2e=Pd(s))=33.4			
also B4 values for 10-60C, assuming same K and 1/RTF as for 25C!										

Pd++	sp	oth/un	25°C	0.0	U	I		K1=6.0	B2=10.60	1964BSg (5476) 72
							K3=2.5			
							K4=2.0			
							B4=15.1			
also B4 for I=0.25 to 1.01 M NaCl04										

Pd++	oth	oth/un	25°C	1.0M	U			K1=3.88	B2=6.94	1964BUa (5477) 73
							K3=2.14			
							K4=1.34			
							B4=10.42			
K1 by solubility, otheres by EMF, spec,										

Pd++	sp	NaCl04	25°C	0.50M	U			1964SBe	(5478)	74
							K4=1.35			

Pd++	ISE	oth/un	19°C	var	U			1963GKa	(5479)	75
							B4=12.2			
							K(Pd+2e=Pd(s))=33.4			

Pd++	sol	none	25°C	0.0	U	M		1962REa	(5480)	76
							Ks=-3.02			
							K(trans-Pd(NH3)2L+L)=2.41			
I=0 corr. Ks: Pd(NH3)2L2(s)=Pd(NH3)2L2										

Pd++	sp	NaCl04	20°C	0.80M	U			K1=4.34	B2=7.88	1961SLc (5481) 77
							K3=2.68			
							K4=1.68			
							B4=12.24			

Pd++ sp none 21°C 0.0 U T H K1=6.2 B2=10.9 1957DBa (5482) 78
K3=2.5
K4=2.6

DH(K1)=-33 kJ mol⁻¹, DS=4.2 J K⁻¹ mol⁻¹; DH(K2)=-38, DS=-42; DH(K3)=-33, DS=-59; DH(K4)=-33, DS=-59. 38 C: K1=5.9, K2=4.1, K3=2.2, K4=2.5

Pd++ sp none 25°C 0.0 U T H K1=6.1 B2=10.7 1956DRa (5483) 79
K3=2.4
K4=2.6
K5=-2.1
K6=-2.1

DH(K5)=0, DS=-38 J K⁻¹ mol⁻¹; DH(K6)=0, DS=-38

Pd++ ISE NaCl04 25°C 4.0M U B4=13.22 1943TWa (5484) 80

FC1BrI HL (541)
Halides, comparative (for book data under ligand 80)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sol NaCl04 20°C 0.10M U M 1967GGd (7412) 81
K(Pd+Cl)=5.1
K(Pd+Br)=6.8
K(Pd+I)=10.0
K(Pd+CN)=10.5

Pd++ sp oth/un 27°C 0.50M U HM 1967HPb (7413) 82
K(PdACl+I=PdAI+Cl)=1.95
K(PdABr+I=PdAI+Br)=1.48
K(PdABr+SCN)=2.23

A=dien. DH(Cl,I)=-15.5 kJ mol⁻¹, DS=-14.2 j k⁻¹ MOL⁻¹. DH(Br,I)=-10.5,DS=-7;
DH(Br,SCN)=-19.6, DS=-22.6. Also other related data

Pd++ sp NaCl04 25?°C 4.50M U 1967SNa (7414) 83
K(PdBr4+I=PdBr3I+Br)=2.75
K(PdBr3I+I)=3.00
K(PdBr2I2+I)=1.70
K(PdBrI3+I=PdI4+Br)=0.80

Pd++ sp oth/un 25°C 1.10M U 1966BSd (7415) 84
K(PdCl4+2Br=PdCl2Br2+2Cl)=1.99
K(PdCl2Br2+2Br=PdBr4+2Cl)=-.06
B(PdCl2Br2)=13.11
B(PdCl4)=11.12

Pd++ sp NaCl04 25°C 4.50M U M 1966SNc (7416) 85
K(PdCl4+Br=PdCl3Br+Cl)=1.55
K(PdCl3Br+Br=PdCl2Br2+Cl)=1.09

$K(\text{PdCl}_2\text{Br}_2 + \text{Br} = \text{PdClBr}_3 + \text{Cl}) = 0.95$

$K(\text{PdClBr}_3 + \text{Br} = \text{PdBr}_4 + \text{Cl}) = 0.55$

Medium: LiClO₄

I- HL Iodide CAS 10034-85-2 (20)
Iodide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	kin	NaClO ₄	25°C	1.00M	U		K1=6.08	1986E0a (8319)	86
Pd++	sp	NaClO ₄	25°C	1.0M	U		K4=2.56 K(2PdL ₄ =Pd ₂ L ₆ +2L)=1.32	1977E0b (8320)	87

Pd++	sp	NaClO ₄	25°C	4.50M	U	M		1972SNc (8321)	88
							K(PdCl ₄ +L=PdCl ₃ L+Cl)=3.95 K(PdCl ₃ L=PdCl ₂ L ₂ +Cl)=4.1 K(PdCl ₂ L ₂ +L=PdClL ₃ +Cl)=2.8 K(PdClL ₃ +L=PdL ₄ +Cl)=1.30		

Medium: LiClO₄. Data also for complexes with Br in place of Cl: 2.75, 5.75, 7.45, 8.25. Data for L=Br : 1.55, 2.64, 4.14

Pd++	sol	NaClO ₄	20°C	0.10M	U		K1=10.0	1967GGa (8322)	89
Pd++	ISE	oth/un	25°C	1.0M	U		B4=24	1965FKa (8323)	90

Medium:KI. Also B4 for 10-60C, assuming same K and 1/RTF as for 25C!

Pd++	sp	NaClO ₄	20°C	0.80M	U		K1=4.95 K4=2.92 B4=15.74	1965SLd (8324)	91
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Pd++	ISE	oth/un	19°C	var	U		B4=24.9	1963GKa (8325)	92
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Pd++	sol	oth/un	18°C	var	U		K(PdL ₂ (s)+2I=PdL ₄)=-2.8	1948TAb (8326)	93
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NH₃ L Ammonia CAS 7664-41-7 (414)
Ammonia

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	oth/un	25°C	1.0M	U	H	K1=9.56 K3=7.52	1991NSb (9192)	94

Medium: H/NH₃/NaClO₄;

Pd++	gl	KNO ₃	25°C	?	M	M	K1=6.06 K(PdA+L)=5.36	1988SKa (9193)	95
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A=diethylenetriamine

Pd++ gl NaClO4 21°C 0.10M C M 1984KMe (9194) 96
K(PdGlyGly+L)=6.50
K(PdPheGly+L)=6.53

Data also for many other amines

Pd++ sp none 25°C 0.0 C 1975PJb (9195) 97
K(Pd(phen)+L)=7.45
K(Pd(phen)L+L)=6.3

Pd++ gl NaClO4 25°C 1.0M U K1=9.6 B2=18.50 1968RJa (9196) 98
K3=7.5
K4=6.8

Pd++ ISE oth/un 25°C 0.50M U 1965FKa (9197) 99
B4=29.6
K(Pd+2e=Pd(s))=33.4

Medium: L. Also B4 for 10-60 C but assuming same RT/F as at 25 C

NO2- HL Nitrite CAS 7782-77-6 (635)
Nitrite;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ ISE oth/un 25°C var U 1965FKa (9401) 100
B4=21
K(Pd+2e=Pd(s))=33.4

Medium: KL var. B4 values 10-60 C, but RT/F at value for 25 C

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M C M 2002MSb (11918) 101
*K(PdA(H2O)2)=-5.54
*K(PdA(OH)H2O)=-15.01

K(2PdA(H2O)2=Pd2A2(OH)2+2H)=-7.90. A is N,N'-dimethylethylenediamine.

Pd++ gl NaClO4 25°C 0.10M C 2001BPd (11919) 102
*K(Pd(dien)(H2O))=-7.16
K(2Pd(dien)(H2O)=Pd2(dien)2(OH)2)=-10.56.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (11920) 103
*K(Pd(bpy)(H2O)2)=-3.91
*K(Pd(bpy)(OH)H2O)=-8.09

K(2Pd(bpy)=Pd2H-2(bpy)2)=-4.70

Pd++ sol NaClO4 25°C 0.50M C TI K1=11.95 B2=23.20 1999VWa (11921) 104

At I=1.0, B2=23.4, B3=26.2. At I=0.1, B2=23.8. Data for 25-85 C.

Pd++ sp NaCl04 25°C 1.0M C 1998SEb (11922) 105
*K(Pt(H2O)4)=-3.0

Pd++ gl NaCl04 37°C 0.15M C M 1996GTa (11923) 106
*K(PdA(H2O)2)=-5.25
*K(dimer)=-6.55

A=diaminosuccinate diethylester, EtO2CCH(NH2).CH(NH2)CO2Et

*K: PdA(H2O)2=PdA(H2O)(OH)L+H, *K(dimer): 2PdA(H2O)2=(PdA(H2O)2(OH)2)2+2H

Pd++ sol oth/un 25°C var M B2=18.9 1991W0a (11924) 107
B3=20.9

Pd++ gl NaCl 25°C 0.50M C I 1984MBa (11925) 108
*K1=-9.23
*B(4,4)=-28.81

Data for 0.5-3.0 M NaCl. At I=1.0 M, *K1=-9.30, *B(4,4)=-29.10

Pd++ sol NaCl04 17°C 0.10M U K1=11.72 B2=23.57 1970NKb (11926) 109
K3=1.85
K4=1.0
Kso(Pd(OH)2(s))=-28.96

Pd++ sp none 25°C 0.0 M K1=12.4 B2=26.5 1967IEa (11927) 110
By glass electrode: K1=13.0, B2=25.8. By solubility: Ks(PdL2(s)=PdL2)=-2.65

Pd++ sp oth/un 25°C var U 1966WYa (11928) 111
*K1(PdCl2(H2O)2)=-2

Pd++ oth none 25°C 0.0 U 1957ZMa (11929) 112
*Kso(Pd(OH)2)=-2.35
*Kso(PdO)=-3.02

*Kso: K(Pd(OH)2(s)+2H=Pd2+2H2O); *Kso(PdO(s)+2H=Pd2+H2O); method:
combination of thermodynamic data

Pd++ oth none 25°C 0.0 U 1952LAb (11930) 113
Kso(Pd(OH)2)=-31

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Pd++ gl KNO3 25°C ? M M K1=3.10 1988SKa (13301) 114
K(PdA+L)=2.63

A=diethylenetriamine

S-- H2L Sulfide CAS 7783-06-4 (705)
Sulfide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sol	oth/un	200°C	var	U	T	Ks(PdS+H2S)=-7.0 Ks(PdS+2H2S)=-11.2	1993GBa (14450)	115
Constants at I=0. 30-300 C									
Pd++	oth	none	25°C	0.0	C		KPd+HS=PdS+H)=43.4	1989DYa (14451)	116
Calculated from literature data, based on K(H+S)=17.0.									
Pd++	oth	none	25°C	0	U		Kso(PdS)=-62.1 *Kso(PdS)=-44.8	1988LIa (14452)	117
Derived from thermodynamic data and K(H+S=HS)=17.3.									

SCN-		HL		Thiocyanate			CAS 463-56-9 (106)		
Thiocyanate;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaCl04	30°C	0.10M	U	I	B2=16.2 B4=25.2	1973JPa (15222)	118
Medium: HCl04. At I=0, B2=16.9, B4=25.6									
Pd++	sp	oth/un	25°C	1.0M	U	M	B4=28.67 K(PdCl4+L=PdCl3L+Cl)=6.03 K(PdCl3L+L=PdCl2L2+Cl)=4.09 K(PdCl2L2+L=PdClL3+Cl)=3.59	1967BSc (15223)	119
Medium: 1 NaCl, 0.1 H+. K(PdClL3+L=PdL4+Cl)=3.03									
Pd++	sp	oth/un	25°C	1.0M	U		B(PdBrL3)=25.85 B(PdBr2L2)=22.25 B(PdBr3L)=18.15 B(PdBr4)=13.05	1966BSd (15224)	120
B(PdClL3)=25.19; B4=28.22. Medium: 1 M Na+, 0.1 M H+									
Pd++	ISE	oth/un	25°C	var	U		B4=26 K(Pd+2e=Pd(s))=33.4	1965FKa (15225)	121
Medium: KI. Also B4 values 10 to 60 C - doubtful since RT constant									
Pd++	ISE	oth/un	25°C	dil	U	T	B4=19.46	1964GPa (15226)	122
Kso=-17.8. By spectrophotometry: B2=8.4. By solubility, 20 C: K(PdL2(s)+2L)=1.63									

Pd++ sol oth/un 20°C var U 1964GPa (15227) 123
 $K_s(\text{PdI}_2(\text{s})+\text{L}=\text{PdL}_2\text{L})=-0.47$

Pd++ ISE oth/un 19°C var U 1963GKa (15228) 124
 $B_4=27.6$
 $K(\text{Pd}+2\text{e}=\text{Pd}(\text{s}))=33.4$

 SO3-- H2L Sulfite CAS 7782-99-2 (801)
 Sulfite;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ con oth/un ? var U 1960EAa (15474) 125
 $K(\text{PdL}_2(\text{OH})(\text{H}_2\text{O})+\text{H})=9$

 SO4-- H2L Sulfate CAS 7664-93-9 (15)
 Sulfate;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 1.0M C 1998SEb (16469) 126
 $K(\text{Pd}+\text{SO}_4)=1.28$
 $K(\text{Pd}+\text{HSO}_4)=-0.15$
 $K(\text{PdSO}_4+\text{H})=-0.40$

Pd++ vlt NaClO4 25°C 0.20M U B2=3.16 1971JPa (16470) 127
 Medium:HClO4

 Se-- H2L Selenide (6335)
 Selenide;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ oth none 25°C 0.0 U 1964BUE (16947) 128
 $K_{\text{so}}=-73.4$

 CH2O2 HL Formic acid CAS 64-18-6 (37)
 Methanoic acid; H.CO₂H

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=3.67 1997SEa (17639) 129
 $K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=0.15$
 $\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-9.7 \text{ kJ mol}^{-1}$, $\text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-29 \text{ J K}^{-1} \text{ mol}^{-1}$

Pd++ gl KNO3 25°C ? M M K1=2.22 1988SKa (17640) 130
 $K(\text{PdA}+\text{L})=2.14$

A=diethylenetriamine

 CH4N2O L Urea CAS 57-13-6 (2018)

Carbamide, Urea; (H₂N)₂CO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	nmr	oth/un	40°C	0.90M	U				1998KKf (17723)	131

K(Pt(H₂O)₂en+L)=0.11

Method: ¹³C nmr. K is for N-bound ligand. For O-bound urea, K=1.36.

Also data for many other alcohol/H₂O mixtures.

CH ₄ N ₂ S	L	Thiourea	CAS 62-56-6	(51)
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Thiocarbamide, Thiourea; (H₂N)₂CS

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	?	1.0M	U	M			1966SBb (17848)	132

K(PdCl₂L₂+L=PdClL₃+Cl)=4.86

K(PdBr₂L₂+L=PdBrL₃+Br)=4.65

K(PdClL₃+L=PdL₄+Cl)=4.24

K(PdBrL₃+L=PdL₄+Br)=4.18

K(Pd(SCN)₂L₂+L=Pd(SCN)L₃+SCN)=2.95; K(Pd(SCN)L₃+L=PdL₄+SCN)=2.52. I=1 or 0.2

CH ₅ N	L	Methylamine	CAS 74-89-5	(155)
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Methylamine; CH₃.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaNO ₃	25°C	0.10M	C	M			2002MSb (18023)	133

K(PdA+L)=7.64

K(PdA+2L)=13.46

K(PdA+B+L)=16.57

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutane dicarboxylic acid.

Pd++	gl	NaNO ₃	25°C	0.10M	U	M			1999SSd (18024)	134
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K(Pd(pn)+L)=6.96

K(Pd(pn)+2L)=13.57

pn is 1,2-diaminopropane. For amine protonation, K₁=10.43.

Pd++	gl	KNO ₃	25°C	0.10M	M	M	K ₁ =7.56		1991SKe (18025)	135
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K(Pd(dien)+L)=4.86

Also data for complexes with homologous alkylamines.

Pd++	gl	NaClO ₄	21°C	0.10M	C	M			1984KMe (18026)	136
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K(PdGlyGly+L)=7.18

K(PdPheGly+L)=7.31

Data also for many other amines

CH ₆ NO ₃ P	H ₂ L	AMPA	CAS 1066-51-3	(1981)
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Aminomethylphosphonic acid; H₂N.CH₂.PO₃H₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp KCl ? 1.00M U 1973RRc (19244) 144
B4=32.4

Medium: HCl

C2H4 L Ethylene CAS 74-85-1 (478)
Ethene; H2C:CH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sol oth/un 25°C 2.00M U M 19720La (19428) 145
K(PdCl3+L)=4.79
K(PdBr3+L)=3.64
K(PdI3+L)=1.68
K(Pd(SCN)3+L)=2.00

Medium : MgSO4 K(Pd(NO2)3+L)=1.34

Pd++ sol NaClO4 13°C 2.0M U I 1966PMb (19429) 146
K(PdCl4+L=PdCl3L+Cl)=1.19
K(PdCl3L=PdCl2(H2O)L+Cl)=-1.5

Medium:HClO4. K(PdCl4+L=PdCl2(H2O)L+2Cl)=-0.7. I=3.0: K values: 1.2, -0.7, 0.4. I=4.5(LiClO4+HClO4): K values: 1.21, -0.4, 0.81

C2H4O2 HL Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH3.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 1.00M U K1=4.34 1996SEa (20127) 147

Pd++ sp alc/w 25°C 100% U M 1994PAa (20128) 148
K(Pd3A3CO+L)=2.86

Medium: MeOH. A=Bis(diphenylphosphino)methane

Pd++ gl KNO3 25°C ? M M K1=2.73 1988SKa (20129) 149
K(PdA+L)=2.52

A=diethylenetriamine

Pd++ sp NaClO4 25°C 0.92M U K1=4.9 B2=8.0 1976YBa (20130) 150
K3=2.6

Pd++ sp non-aq 25°C 100% U M 19720Ma (20131) 151
K(PdL2+CeL3=CePdL5)=4.8

Medium: CH3COOH

C2H4O2S H2L Thioglycolic CAS 68-11-1 (596)
Mercaptoethanoic acid; HS.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 25°C 1.00M C B2=47.5 2000SAb (20356) 152

K(Pd+HL)=22.2
K(Pd+2HL)=37.1

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 1.00M U K1=3.81 1996SEa (20608) 153

C2H5NO L Acetamide CAS 60-35-5 (2886)
Ethanoic acid amide; CH3.CO.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un 25°C .001M U K1=4.46 1958MCa (20673) 154

C2H5NO2 HL Glycine CAS 56-40-6 (85)
2-Aminoethanoic acid; H2N.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (21677) 155

K(Pd(pn)+L)=11.01

pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.60, B2=11.93.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (21678) 156
K(PdA+L)=10.76

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.50M U 1978LIa (21679) 157
K(Pd(en)+L)=11.21

Pd++ gl NaClO4 20°C 1.00M C K1=15.25 B2=27.50 1976AMa (21680) 158
K(PdL+2Br)=6.47

Pd++ gl oth/un 25°C 0.01M U K1=9.12 B2=17.55 1949MMa (21681) 159

C2H6N2O L Glycinamide CAS 598-41-4 (60)
2-Aminoethanoic acid amide; H2N.CH2.CO.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (21951) 160

K(Pd(pn)+L)=8.58

K(Pd(pn)+L=PdH-1(pn)L+H)=5.35

pn is 1,2-diaminopropane. For amine protonation, K1=7.88.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (21952) 161
K(PdA+L)=7.41

$$K(\text{PdA}+\text{L}=\text{PdAH}-1\text{L}+\text{H})=4.20$$

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.10M U M 1977LIb (21953) 162

$$K(\text{Pd}(\text{en})+\text{L})=8.64$$

$$K(\text{Pd}(\text{en})\text{L}=\text{PdH}-1(\text{en})\text{L}+\text{H})=-2.47$$

C2H6OS L DMSO CAS 67-68-5 (329)

Dimethylsulfoxide; (CH3)2.SO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un 25°C ? U T H K1=0.954 B2=0.56 1987DMA (22117) 163
DH(K1)=-7.3 kJ mol-1; DS(K1)=-8.4 J K-1 mol-1

Pd++ sp alc/w 25°C 95% U I 1982CCa (22118) 164

$$K(\text{PdCl}_4+\text{L}=\text{PdLCl}_3+\text{Cl})=1.8$$

$$K(\text{PdLCl}_3+\text{L}=\text{PdL}_2\text{Cl}_2+\text{Cl})=-1.6$$

Medium: 95% MeOH/H2O

C2H7NO L Ethanolamine CAS 141-43-5 (1057)

2-Aminoethanol; H2N.CH2.CH2.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (22415) 165

$$K(\text{Pd}(\text{pn})+\text{L})=7.30$$

$$K(\text{Pd}(\text{pn})+\text{L}=\text{PdH}-1(\text{pn})\text{L}+\text{H})=1.94$$

pn is 1,2-diaminopropane. For amine protonation, K1=9.31.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (22416) 166

$$K(\text{PdA}+\text{L})=6.81$$

$$K(\text{PdA}+\text{L}=\text{PdAH}-1\text{L}+\text{H})=1.85$$

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.10M M M 1991SKe (22417) 167

$$K(\text{Pd}(\text{dien})+\text{L})=5.29$$

Pd++ gl KNO3 25°C 0.10M U M 1981LIb (22418) 168

$$K(\text{Pd}(\text{H}_2\text{O})_2\text{A}+\text{L}=\text{PdLA}+2\text{H}_2\text{O})=7.88$$

$$K(\text{Pd}(\text{H}-1\text{L})\text{A}+\text{H})=5.16$$

A=1,2-diaminoethane

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)

1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C M 2001NSa (23218) 169

$$*B_2(\text{PdL}(\text{H}_2\text{O})_2)=-15.21$$

$K(2PdL(H_2O)_2 = Pd_2(OH)L_2 + H) = -3.04$, $K(2PdL(H_2O)_2 = Pd_2(OH)L_2 + 2H) = -8.41$,
 $K(3PdL(H_2O)_2 = Pd_3(OH)_3L_3 + 3H) = -11.80$

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (23219) 170
 $K(Pd(bpy)(H_2O)_2 + L) = 17.08$
 $K(Pd(bpy)(H_2O)_2 + H + L) = 20.87$

Pd++ gl KNO3 25°C 0.10M M M 1991SKe (23220) 171
 $K(Pd(dien) + L) = 6.70$
 $K(Pd(dien) + H + L) = 14.63$

Pd++ gl NaClO4 25°C 1.00M C M K1=23.6 B2=42.20 1986ANa (23221) 172
 Ternary complex with Br-. Combined pot. and spectrophotometric study

Pd++ gl KNO3 23°C 0.20M U 1976LMa (23222) 173
 $K(2PdL(OH)_2 = LPd(OH)_2PdL) = 8.3$

 C2H8O7P2 H4L HEDPA CAS 2809-21-4 (436)
 1-Hydroxyethane-1,1-diphosphonic acid; $CH_3.C(OH)(PO_3H_2)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	U		K1=5.74 $K(Pd+HL) = 4.44$ $K(Pd+H_2L) = 2.41$	1980ZRC (23394)	174

 C3H4N2 L Imidazole CAS 288-32-4 (90)
 1,3-Diazole, imidazole; $C_3H_4N_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	C	M	$K(PdA+L) = 7.92$ $K(PdA+2L) = 14.64$ $K(PdA+B+L) = 14.20$	2002MSb (23917)	175

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutane dicarboxylic acid.

Pd++	gl	NaNO3	37°C	0.16M	M	M	$K(PdA+L) = 7.29$ $K(PdA+2L) = 13.87$	1998ESa (23918)	176
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A is 1,3-diaminopropane.

Pd++	gl	KNO3	25°C	?	M	M	K1=6.40 $K(PdA+L) = 5.62$	1988SKa (23919)	177
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A=diethylenetriamine

 C3H4O4 H2L Malonic acid CAS 141-82-2 (79)
 Propanedioic acid; $CH_2(COOH)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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 Pd++ gl NaClO4 37°C 0.15M C M 2003Tmb (24533) 178
 K(Pd(en)+L)=5.40

Pd++ kin NaClO4 25°C 1.00M U H K1=3.40 1997SEa (24534) 179
 K(Pd+HL=PdL+H)=0.8

DH(Pd+HL=PdL+H)=-7.5 kJ mol⁻¹, DS(Pd+HL=PdL+H)=-10 J K⁻¹ mol⁻¹

C3H6 L Propylene CAS 115-07-1 (702)

Propene; CH₃.CH:CH₂

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sol oth/un 25°C 2.00M U M 19720La (24755) 180

K(PdCl₃+L)=4.97

K(PdBr₃+L)=3.72

K(Pd(NO₂)₃+L)=1.39

K(PdI₃+L)=1.70

Medium: MgSO₄

C3H6O2 HL Propionic acid CAS 79-09-4 (35)

Propanoic acid; CH₃.CH₂.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 1.00M U K1=4.32 1996SEa (25040) 181

Pd++ sp alc/w 25°C 100% U M 1994PAa (25041) 182

K(Pd3A3CO+L)=3.41

Medium: MeOH. A=Bis(diphenylphosphino)methane

 Pd++ gl KNO₃ 25°C ? M M K1=2.94 1988SKa (25042) 183

K(PdA+L)=2.60

A=diethylenetriamine

C3H6O2S HL CAS 2444-37-3 (1074)

(Methylthio)ethanoic acid; CH₃.S.CH₂.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 25°C 1.00M C K1=13.2 B2=22.30 2000SAb (25092) 184

Pd++ kin oth/un 25°C 1.00M U 1996SEa (25093) 185

K1eff=4.08

Medium: 1.00 M HClO₄.

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)

L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (26240) 192
K(PdA+L)=10.90

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.50M U 1978LIa (26241) 193
K(Pd(en)+L)=11.22

Pd++ gl KNO3 20°C 0.5M U K1=9.98 B2=18.33 1974KHb (26242) 194

C3H7NO2 HL B-Alanine CAS 107-95-9 (575)
3-Aminopropanoic acid; H2N.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Pd++ gl KNO3 20°C 0.5M U T K1=8.73 B2=15.79 1974KHb (26473) 195

C3H7NO2 HL Sarcosine CAS 107-97-1 (87)
N-Methyl-2-aminoethanoic acid; CH3.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Pd++ gl KNO3 25°C 0.50M U 1978LIa (26606) 196
K(Pd(en)+L)=11.28

C3H7NO2S H2L Cysteine CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Pd++ gl NaClO4 25°C 1.00M C B2=51.6 2000SAb (26822) 197
K(Pd+HL)=27.3
K(Pd+2HL)=45.0

C3H7NO3 HL Serine CAS 56-45-1 (49)
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (27167) 198
K(Pd(pn)+L)=12.00
K(Pd(pn)+L=PdH-1(pn)L+H)=3.74
pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.14, B2=11.40.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (27168) 199
K(PdA+L)=10.19
K(PdA+L=PdAH-1L+H)=1.90

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.10M U M T 1981LIb (27169) 200
K(PdA(H2O)2+L=PdAL+2H2O)=11.01

$$K(\text{PdA}(\text{H}-1\text{L})+\text{H})=8.51$$

A=1,2-diaminoethane

C3H8O3S3 H3L Unithiol CAS 74-61-3 (1271)

2,3-Dimercaptopropanesulfonic acid; HS.CH2.CH(SH).CH2.SO3H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ EMF KNO3 ? 1.00M U B2=21.1 1969SOa (27797) 201

Medium:HNO3

C3H9N2O4P H2L CAS 30211-73-5 (7117)

Glycylaminomethylphosphonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.10M C B2=26.27 1997BLc (27968) 202

$$B(\text{PdH}-2\text{L}2)=10.99$$

$$B(\text{PdLCl})=20.54$$

$$B(\text{PdH}-1\text{LCl})=16.74$$

$$B(\text{PdH}-2\text{L})=8.67$$

$$B(\text{PdH}-3\text{L})=-1.51$$

Pd++ gl KCl 25°C 0.10M U 1996BRa (27969) 203

$$K(\text{Pd}+\text{L}+2\text{Cl}+\text{H})=24.48$$

$$K(\text{Pd}+2\text{L})=27.50$$

$$K(\text{Pd}+\text{L}+\text{Cl})=21.35$$

C3H10N2 L CAS 78-90-0 (2905)

1,2-Diaminopropane; CH3.CH(NH2)CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U 1999SSd (28170) 204

$$*K(\text{PdL})=-5.62$$

$$*K(\text{Pd}(\text{OH})\text{L})=-9.35$$

C3H10N2 L Propanediamine CAS 109-76-2 (123)

1,3-Diaminopropane; H2N.CH2.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 37°C 0.16M M 1998ESa (28319) 205

$$*K(\text{PdL}(\text{H}2\text{O})2)=-5.45$$

$$*B2(\text{PdL}(\text{H}2\text{O})2)=-14.58$$

C3H11N3 L CAS 21292-99-6 (2975)

Propane-1,2,3-triamine; H2N.CH2.CH(NH2).CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

$$K(\text{PdA}+2\text{L})=13.92$$

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.10M U M 1981LIa (28867) 213

$$K(\text{Pd}(\text{en})(\text{H}_2\text{O})_2+\text{L})=8.59$$

$$K(\text{Pd}(\text{en})(\text{H}_2\text{O})\text{L}+\text{L})=6.79$$

$$K(\text{Pd}(\text{dien})(\text{H}_2\text{O})+\text{L})=8.01$$

C4H6N2S HL Methimazole CAS 60-56-0 (1824)

N-Methyl-2-mercaptoimidazole; C3H2N2(CH3).SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp KNO3 25°C 0.50M C K1=7.43 B2=11.30 1977LWa (29666) 214

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)

1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=4.03 1997SEa (30025) 215

$$K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=0.08$$

$$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=10 \text{ kJ mol}^{-1}, \text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=36 \text{ J K}^{-1} \text{ mol}^{-1}$$

C4H6O4 H2L Me-Malonic Acid CAS 516-15-2 (816)

Methylpropanedioic acid; HOOC.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 37°C 0.15M C M 2003TMb (30134) 216

$$K(\text{Pd}(\text{en})+\text{L})=5.68$$

C4H6O5 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=3.65 1997SEa (30699) 217

$$K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=0.54$$

$$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-3.5 \text{ kJ mol}^{-1}, \text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-1 \text{ J K}^{-1} \text{ mol}^{-1}$$

C4H6O5 H2L Diglycolic acid CAS 110-99-6 (243)

Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOC.CH2.O.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=3.46 1997SEa (30914) 218

$$K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=0.64$$

$$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-12 \text{ kJ mol}^{-1}, \text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-27 \text{ J K}^{-1} \text{ mol}^{-1}$$

C4H7NO4 H2L Aspartic acid CAS 56-84-8 (21)
 Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl	37°C	0.10M	C			2003GZa (31919)	219
							B(Pd(bpy)L)=13.91		
Pd++	gl	none	25°C	0.0	U			1979FWa (31920)	220
							K(PdL2+H)=4.52		
							K(PdHL2+H)=3.68		
							K(PdCl4+2HL=PdH2L2+4Cl)=11.3		
Pd++	gl	NaCl04	25°C	0.10M	U		K1=10.44 B2=18.14	1972SSe (31921)	221
Pd++	gl	KNO3	30°C	0.10M	U		K1=10.55 B2=18.25	1971STc (31922)	222
Pd++	oth	KNO3	30°C	0.13M	U			1971TKe (31923)	223
							K(Pd+H2L=PdHL+H)=10.45		
							K(PdHL+H2L=Pd(HL)2+H)=7.76		

C4H7NO4 H2L IDA CAS 142-73-4 (118)
 Iminodiethanoic acid; HN(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl04	20°C	1.00M	C	M	K1=17.5 B2=26.80	1976AMa (32336)	224
							K(Pd+HL)=9.0		
							K(PdL+2Br)=3.83		
Pd++	gl	KCl	25°C	0.1M	U		K1=9.62 B2=14.87	1975CGc (32337)	225
Pd++	EMF	KCl	25°C	0.10M	U		K1=9.62 B2=14.87	1975VCa (32338)	226

C4H8 L But-1-ene CAS 106-98-9 (2985)
 But-1-ene; CH2:CH.CH2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sol	NaCl04	15°C	5.0M	U	I M		1966PMb (32458)	227
							K(PdCl4+L=PdCl3L+Cl)=1.05		

Medium: LiCl04+HCl04. K=1.14(I=2), 1.13(I=3), 1.13(I=4)
 K(PdCl4+L=PdCl2(H2O)L+2Cl)=-0.5(I=2), 0.1(I=3), 0.65(I=4), 0.95(I=5)

C4H8N2O2 H2L Dimethylglyoxim CAS 95-45-4 (2032)
 2,3-Butanedione dioxime, Dimethylglyoxime; CH3.(C:NOH).(C:NOH).CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	dis	NaCl04	25°C	1.0M	U		B2=34.1	1963BDa (32547)	228

$$K(\text{PdL}_2+\text{OH})=5.50$$

Pd++ sol oth/un 25°C ? U 1958BBb (32548) 229

$$K_{s2}=-3.30$$

C4H8N2O3 HL Asparagine CAS 70-47-3 (17)

2-Aminobutanedioic acid 4-amide; $\text{H}_2\text{N}.\text{CH}(\text{CH}_2.\text{CO}.\text{NH}_2).\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (32720) 230

$$K(\text{Pd}(\text{pn})+\text{L})=12.79$$

$$K(\text{Pd}(\text{pn})+\text{L}=\text{PdH}-1(\text{pn})\text{L}+\text{H})=6.38$$

pn is 1,2-diaminopropane. For amide protonation, $K_1=8.55$.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (32721) 231

$$K(\text{PdA}+\text{L})=10.19$$

$$K(\text{PdAH}-1\text{L}+\text{H})=3.33$$

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.50M U M 1977LIa (32722) 232

$$K(\text{Pd}(\text{en})+\text{L})=10.46$$

$$K(\text{Pd}(\text{en})\text{H}-1\text{L}+\text{H})=6.46$$

Pd++ gl NaClO4 25°C 3.00M C 1974Gwa (32723) 233

$$B(\text{PdHL})=12.11$$

$$B(\text{PdH}-1\text{L})=9.1$$

$$B(\text{PdHLC1})=18.29$$

$$B(\text{PdH}-1\text{LC1})=17.0$$

C4H8N2O3 HL Gly-Gly CAS 556-50-3 (54)

Glycyl-glycine; $\text{H}_2\text{N}.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}_2.\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C 1999AJa (33047) 234

$$B(\text{PdLC1})=18.08$$

$$B(\text{PdH}-1\text{LC1})=15.56$$

$$B(\text{PdH}-2\text{L})=4.89$$

$$B(\text{PdH}-1\text{L}_2)=19.30$$

Medium: 0.1 M KNO3, 0.1 M KCl. $B(\text{PdH}-2\text{L}_2)=13.90$; $B(\text{PdH}-1\text{L})=13.57$.

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (33048) 235

$$K(\text{Pd}(\text{pn})+\text{L})=9.41$$

$$K(\text{Pd}(\text{pn})+\text{L}=\text{PdH}-1(\text{pn})\text{L}+\text{H})=6.02$$

pn is 1,2-diaminopropane. For aminoacid protonation, $K_1=7.97$.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (33049) 236

$$K(\text{PdA}+\text{L})=7.53$$

$$K(\text{PdA}+\text{L}=\text{PdAH}-1\text{L}+\text{H})=2.88$$

A is 1,3-diaminopropane.

Pd++ gl KNO3 25°C 0.10M U M 1977LIb (33050) 237
K(Pd(en)+L)=9.60
K(Pd(en)L=PdH-1(en)L+H)=-3.76

C4H8O5 L (1882)
Tetramethylenesulfoxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 95% U 1982CCa (33192) 238
K(PdCl4+L=PdLCl3+Cl)=1.7

C4H8O2 HL CAS 107-92-6 (1118)
n-Butanoic acid; CH3.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=4.38 1997SEa (33346) 239
K(Pd+HL=PdL+H)=-0.29
DH(Pd+HL=PdL+H)=-14.6 kJ mol⁻¹, DS(Pd+HL=PdL+H)=-55 J K⁻¹ mol⁻¹

C4H8O3 HL CAS 594-61-6 (81)
2-Hydroxy-2-methylpropanoic acid; (CH3)2C(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.00M U H K1=4.05 1997SEa (33504) 240
K(Pd+HL=PdL+H)=1.88
DH(Pd+HL=PdL+H)=-8.1 kJ mol⁻¹, DS(Pd+HL=PdL+H)=-21 J K⁻¹ mol⁻¹

C4H9NO L CAS 127-19-5 (477)
N,N-Dimethylacetamide; CH3.CO.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 100% U M 1994PAa (33767) 241
K(Pd3A3CO+L)=-1.15

Medium: MeOH. A=Bis(diphenylphosphino)methane

C4H9NO2 L CAS 623-33-6 (3011)
Glycine ethyl ester; H2N.CH2.CO.OCH2CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M M M K1=6.01 1987SKb (34002) 242
K(Pd(dien)+L)=2.81

Pd++ gl KNO3 25°C 0.50M U 1983LIb (34003) 243

K(Pd(en)+L)=7.12

C4H9NO2 HL Dimethylglycine CAS 1118-68-9 (88)
N,N-Dimethyl-2-aminoethanoic acid; (CH3)2N.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.50M U 1978LIa (34031) 244

K(Pd(en)+L)=11.02

C4H9NO2S HL Methylcysteine CAS 1187-84-4 (84)
2-Amino-3-methylmercaptopropanoic acid; H2N.CH(CH2.S.CH3)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 25°C 0.10M M 2002BSa (34101) 245

*K(PdL)=-4.13

K(2PdL=Pd2H-1L2)=-0.01

*B2(PdL)=-15.77

Pd++ gl NaClO4 25°C 1.00M C K1=19.9 B2=36.30 2000SAb (34102) 246

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (34103) 247

K(Pd(pn)+L)=10.83

pn is 1,2-diaminopropane. For aminoacid protonation, K1=8.65.

Pd++ gl KNO3 25°C 0.50M U 1978LIa (34104) 248

K(Pd(en)+L)=9.38

K(Pd(en)+HL)=1.18

C4H9NO3 HL Threonine CAS 72-19-5 (48)
2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH3)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (34319) 249

K(Pd(pn)+L)=11.76

K(Pd(pn)+L=PdH-1(pn)L+H)=3.83

pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.06, B2=11.03.

Pd++ gl KNO3 25°C 0.10M U M T 1981LIb (34320) 250

K(PdA(H2O)2+L=PdAL+2H2O)=10.96

K(PdA(H-1L)+H)=8.05

A=1,2-diaminoethane

C4H9NO3 HL Homoserine CAS 1927-25-9 (578)
2-Amino-4-hydroxybutanoic acid; HO.CH2.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

C4H12N2	L	Dimeen	CAS 110-70-3 (125)
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N,N'-Dimethyl-1,2-diaminoethane; CH₃.NH.CH₂.CH₂.NH.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	gl	NaNO ₃	25°C	0.10M	C			2001MSb (35422)	257
							*K(PdL(H ₂ O) ₂)=-5.54 *K(PdL(OH)H ₂ O)=-9.47 K(2PdL=Pd ₂ H-2L ₂ +2H)=-7.90		

Pd++	gl	NaNO ₃	25°C	0.10M	C	M		2001MSb (35423)	258
							K(PdL+gly)=11.79 K(PdL+ala)=10.89 K(PdL+pro)=11.14 K(PdL+val)=11.59		

Also data for phe, met, imidazole, ser, his, histamine, orn, lys, asp and glu. Amino acid protonation constants also reported.

Pd++	gl	NaNO ₃	25°C	0.10M	C	M		2001MSb (35424)	259
							K(PdL+A)=16.31 K(PdL+B)=15.12 K(PdL+C)=16.31 K(PdL+D)=7.64		

K(PdL+E)=6.46. HA=mercaptoethylamine, H₃B=glutathione, H₂C=cysteine, D=methylamine, E=ethanolamine. Protonation constants also reported.

Pd++	gl	NaNO ₃	25°C	0.10M	C	M		2001MSb (35425)	260
							K(PdL+A)=6.38 K(PdL+B)=6.28 K(PdL+C)=4.35 K(PdL+D)=4.09		

Acids: H₂A=oxalic, H₂B=malonic, H₂C=succinic, H₂D=adipic. Also data for 1,1-cyclobutane dicarboxylic & fumaric. Protonation constants reported.

Pd++	gl	NaNO ₃	25°C	0.10M	C	M		2001MSb (35426)	261
							K(PdL+A)=8.70 K(PdL+B)=8.35 K(PdL+C)=8.56 K(PdL+D)=8.75		

HA=uridine, HB=uracil, HC=thymine, HD=thymidine. Also data for inosine, IMP and adenine. Protonation constants are reported.

Pd++	gl	NaNO ₃	25°C	0.10M	C	M		2001MSb (35427)	262
							K(PdL+A)=7.40 K(PdL+B)=10.73 K(PdL+C)=12.31		

A=glycinamide, HB=glutamine, HC=asparagine. Protonation constants are reported.

Pd++	gl	NaNO ₃	25°C	0.10M	C	M		2001MSb (35428)	263
							K(PdL+A)=7.75		

K(PdL+B)=7.63

K(PdL+C)=8.36

HA=glycylglycine, HB=glycylalanine, HC=glycylleucine.

Protonation constants are reported.

C4H13N3 L Dien CAS 111-40-0 (584)
1,4,7-Triazaheptane, 2,2'-Iminobis(ethylamine), diethylenetriamine;
NH2.(CH2)2.NH.(CH2)2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	?	M	M	B(PdH-1L)=-8.68 B(PdL(Butyrate))=2.58 B(PdL(pyridine))=4.04	1988SKa (35805)	264

Pd++	gl	NaClO4	25°C	1.00M	C	M	K1=32.6 B2=40.40	1986ANa (35806)	265
Ternary complexes with Cl- and Br-. A combined pH-metric and spec. study.									

Pd++	gl	NaClO4	25°C	0.50M	C	I	*K(PdL)=-7.589 K(PdL+PdLOH=Pd2L2OH)=2.19	1981GMf (35807)	266
In 0.5 NaNO3: *K(PdL)=-7.543, K(PdL+PdLOH)=2.10									

Pd++	gl	NaNO3	25°C	1.00M	U	M	K1=34 K(PdL+NH3)=6.9	1969RJa (35808)	267

C5H5N		L		Pyridine			CAS 110-86-1 (31)		
Pyridine, Azine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	non-aq	25°C	100%	U		K(PdAB+2L=PdAL2+B)=1.09 K(PdAC+2L=PdAL2+C)=-1.39	1994CVa (36663)	268
Medium: CHCl3. A:n3-allyl; B:4-MeOC6H4.N=CH.CH=N.C6H4OMe; C:4-MeOC6H4.N=C(Me).C(Me)=N.C6H4OMe. Also data for 4-substituted pyridines.									

Pd++	sp	NaClO4	25°C	1.0M	U	I M	K1=8.4 B2=16.10 K3=6.6 K4=5.9	1986AHb (36664)	269
Ternary complexes with 2,2'-bipyridine and 1,10-phenanthroline. In 0.1M NaCH3SO3, K1=8.5, K2=7.5, K3=6.3, K4=5.7.									

Pd++	sp	NaClO4	25°C	1.0M	U	M	K(Pd(en)Cl2+L=PdenLC1+Cl)=4.31 K(Pd(en)LC1+L=PdenL2+Cl)=3.15	1984ETa (36665)	270

Pd++	sp	none	25°C	0.0	C		K(Pd(phen)+L)=7.02	1975PJb (36666)	271

K(Pd(phen)L+L)=6.18

C5H5N5 L Adenine CAS 73-24-5 (237)
6-Aminopurine; H2N.C5H3N4

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	C	M		2001SHc (36975) 272 K(Pd(bpy)(H2O)2+L)=11.95 K(Pd(bpy)(H2O)2+H+L)=15.97 K(Pd(bpy)(H2O)2+2L)=16.59 K(Pd(bpy)(H2O)2+2L+H)=25.76		

K(Pd(bpy)(H2O)2+2L+2H)=30.25.

Pd++	gl	NaNO3	25°C	0.10M	U	M		1999SSd (36976) 273 K(Pd(pn)+L)=11.14		
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pn is 1,2-diaminopropane. For nucleotide protonation, K1=9.59, B2=13.77.

Pd++	gl	NaNO3	37°C	0.16M	M	M		1998ESa (36977) 274 K(PdA+L)=10.83 K(PdA+2L)=14.62		
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A is 1,3-diaminopropane.

C5H6N2O2 HL 1-Methyluracil CAS 615-77-0 (7923)
1-Methyl-2,4(1H,3H)-pyrimidinedione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	C	M		2003NFa (37222) 275 K(PdA+L)=9.26		

A is bis-((2-pyridyl)methyl)amine

Pd++	gl	KNO3	25°C	0.20M	C	M		2001NSa (37223) 276 K(Pd(en)+L)=9.07 K(Pd(en)+2L)=14.88 *K(Pd(en)(H2O)L)=-8.53		
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K(Pd(en)(H2O)+L=Pd(en)(OH)L+H)=0.54,
K(2Pd(en)(H2O)2+2L=Pd2(en)2(OH)L2+H)=12.58

Pd++	gl	KNO3	25°C	0.20M	C	M		2001NSa (37224) 277 K(Pd(pic)+L)=9.57 K(Pd(pic)+2L)=15.73 *K(Pd(pic)(H2O)L)=-7.73		
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K(Pd(pic)(H2O)+L=Pd(pic)(OH)L+H)=1.84,
K(2Pd(pic)(H2O)2+2L=Pd2(pic)2(OH)L2+H)=14.58. Hpic=picric acid.

Pd++	gl	KNO3	25°C	0.20M	C		K1=7.51	2000NFa (37225) 278		
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C5H6N2O2 HL Thymine CAS 65-71-4 (413)
2,4-Dihydroxy-5-methylpyrimidine; C4HN2(CH3)(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	C	M	K(PdA+L)=8.56 K(PdA+2L)=15.14 K(PdA+B+L)=15.71	2002MSb (37282)	279
A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutanedicarboxylic acid.									
Pd++	gl	NaNO3	25°C	0.10M	U	M	K(Pd(pn)+L)=8.90 K(Pd(pn)+2L)=15.80	1999SSd (37283)	280
pn is 1,2-diaminopropane. For nucleotide protonation, K1=9.59.									
Pd++	gl	NaNO3	37°C	0.16M	M	M	K(PdA+L)=8.37 K(PdA+2L)=14.60	1998ESa (37284)	281
A is 1,3-diaminopropane.									
Pd++	gl	KNO3	25°C	0.20M	C		K(PdACl+L=PdAL+Cl)=6.97	1997WKA (37285)	282
PdA is [PdH-1(gly-met)].									

C5H7N3O		HL		1-MeCytosine		CAS	1122-47-0 (2268)		
1-Methyl-4-aminopyrimidin-2-one;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	C	M	K(PdA+L)=5.84 K(2PdA+L=Pd2A2H-1L+H)=1.76	2003NFA (37588)	283
A is bis-((2-pyridyl)methyl)amine									
Pd++	gl	KNO3	25°C	0.20M	C	M	K(Pd(en)+L)=6.13 K(Pd(en)+2L)=11.44 *K(Pd(en)(H2O)L)=-5.69	2001NSa (37589)	284
K(Pd(en)(H2O)+L=Pd(en)(OH)L+H)=0.44, K(2Pd(en)(H2O)2+2L=Pd2(en)2(OH)L2+H)=10.41									
Pd++	gl	KNO3	25°C	0.20M	C	M	K(Pd(pic)+L)=8.07 K(Pd(pic)+2L)=13.35 *K(Pd(pic)(H2O)L)=-5.22	2001NSa (37590)	285
K(Pd(pic)(H2O)+L=Pd(pic)(OH)L+H)=2.85, K(2Pd(pic)(H2O)2+2L=Pd2(pic)2(OH)L2+H)=14.06. Hpic=picric acid.									
Pd++	cal	KNO3	25°C	0.20M	C	HM		2000NFA (37591)	286
DH(Pd(dien)+L)=-38.5 kJ mol-1; DH(Pd(gly-ala)+L)=-33.1, DH(Pd(gly-met)+L)=-32.8.									

$$K(\text{Pd(en)}+\text{L})=12.16$$

 Pd++ gl KNO3 20°C 0.5M U K1=10.26 B2=19.10 1974KHb (38637) 295

C5H9NO3 HL Hydroxyproline CAS 51-35-4 (416)
 4-Hydroxy-2-pyrrolidinecarboxylic acid; C4H7N(OH)(COOH)

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.10M U M 1981LIb (38746) 296
 K(PdA(H2O)2+L=PdAL+2H2O)=11.47
 K(PdA(H-1L)+H)=10.82

A=1,2-diaminoethane

 Pd++ gl KNO3 20°C 0.5M U K1=9.88 B2=19.45 1974KHb (38747) 297

C5H9NO4 H2L Glutamic acid CAS 56-86-0 (22)
 2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KCl 25°C 0.1M U B2=38.0 2004AEa (39116) 298
 K(Pd+HL)=46.7
 K(Pd+2H2L+2Cl)=54.2
 K(Pd+2L+OH)=30.1
 K(P2d+L+2Cl)=41.9

 Pd++ gl none 25°C 0.0 U 1979FWa (39117) 299
 K(PdL2+H)=4.76
 K(PdHL2+H)=4.06
 K(PdCl4+2HL=PdH2L2+4Cl)=10.0

 Pd++ gl NaClO4 25°C 0.10M U K1=10.38 B2=17.84 1972SSe (39118) 300

C5H9NO4 H2L MIDA CAS 4408-64-4 (190)
 N-Methyliminodiethanoic acid; CH3.N(CH2.COOH)2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp KCl 20°C 0.10M U B2=24.88 1987KUa (39277) 301

C5H9NO4S H2L (1736)
 3-(Carboxymethyl)thio-L-alanine; HOOC.CH2.S.CH2.CH(NH2)COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin NaClO4 25°C 1.0M U 1998VTa (39312) 302
 K(Pd+HL=PdL+H)=1.82
 K(Pd+H2L=PdHL+H)=2.43

 C5H9N3 L Histamine CAS 51-45-6 (103)

4(5)-(2'-Aminoethyl)imidazole; C3H3N2.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	U	M		1999SSd (39543) 303		
								K(Pd(pn)+L)=13.22		

pn is 1,2-diaminopropane. For amine protonation, K1=9.59, B2=15.65.

Pd++	gl	NaNO3	37°C	0.16M	M	M		1998ESa (39544) 304		
								K(PdA+L)=12.56		

A is 1,3-diaminopropane.

C5H10N2O3	HL	Glutamine	CAS 56-85-9	(18)
2-Aminopentanedioic acid 5-amide; H2N.CH(CH2.CH2.CO.NH2)COOH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	U	M		1999SSd (39831) 305		
								K(Pd(pn)+L)=11.02		

pn is 1,2-diaminopropane. For amide protonation, K1=8.98.

Pd++	gl	NaNO3	37°C	0.16M	M	M		1998ESa (39832) 306		
								K(PdA+L)=9.29		
								K(PdA+L=PdAH-1L+H)=-0.43		

A is 1,3-diaminopropane.

Pd++	gl	KCl	25°C	0.50M	U	M		1977LIa (39833) 307		
								K(Pd(en)+L)=10.8		
								*K(Pd(en)L)=-9.03		

C5H10N2O3	HL	Ala-Gly	CAS 687-69-4	(55)
Alanyl-glycine; H2N.CH(CH3).CO.NH.CH2.COOH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.20M	C			1999AJa (39893) 308		
								B(PdLC1)=17.96		
								B(PdH-1LC1)=15.09		
								B(PdH-2L)=4.38		
								B(PdH-1L2)=18.70		

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-2L2)=13.37; B(PdH-1L)=13.10.

C5H10N2O3	HL	Gly-Ala	CAS 3695-73-6	(56)
Glycyl-alanine; H2N.CH2.CO.NH.CH(CH3).COOH				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.20M	C			1999AJa (40005) 309		
								B(PdLC1)=18.00		

B(PdH-1LC1)=16.01

B(PdH-2L)=4.80

B(PdH-1L2)=19.80

Medium: 0.1 M KNO₃, 0.1 M KCl. B(PdH-1L)=14.02.

Pd++ gl NaNO₃ 25°C 0.10M U M 1999SSd (40006) 310

K(Pd(pn)+L)=8.17

K(Pd(pn)+L=PdH-1(pn)L+H)=3.69

pn is 1,2-diaminopropane. For aminoacid protonation, K₁=8.04.

C5H10N2O3 HL Gly-b-Ala CAS 7536-21-2 (9057)

Glycyl-beta-alanine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl oth/un 25°C 0.20M C K₁=17.11 2003AMb (40010) 311

K(PdH-1L)=14.93

K(PdH-2L)=6.00

K(PdH-1L2)=20.60

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO₃/0.10 M KCl.

C5H10N2O3 HL B-Ala-Gly CAS 2672-88-0 (4323)

beta-Alanylglycine; H₂N.CH₂.CH₂.CO.NH.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl oth/un 25°C 0.20M C K₁=14.12 2003AMb (40051) 312

K(PdH-1L)=11.09

K(PdH-2L)=2.38

K(PdH-1L2)=17.43

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO₃/0.10 M KCl.

C5H10N4O3 L CAS 54376-69-1 (8335)

N,N'-Carbonylbis(2-aminoacetamide);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO₄ 25°C 0.10M U TIH K₁=10.75 B₂=17.25 1980SAc (40137) 313

Data for 0.075-0.15 M. At I=0, K₁=11.15, K₂=6.70. Also data for 30 C.

DH and DS values.

C5H10O5S2 HL CAS 110-50-9 (591)

(Butoxy)dithiomethanoic acid; CH₃.CH₂.CH₂.CH₂O.CSSH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ dis oth/un 25°C 0.25M U B₂=>24 1982SAa (40163) 314

C5H11N L Piperidine CAS 110-89-4 (105)

Perhydropyridine; cyclo(-CH₂.CH₂.CH₂.NH.CH₂.CH₂-) C5H11N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	non-aq	25°C	100%	U			1994CVa (40452)	315
							K(PdAB+2L=PdAL2+B)=3.14		
							K(PdAC+2L=PdAL2+C)=0.28		
Medium: CHCl3. A:n3-allyl; B:4-MeOC6H4.N=CH.CH=N.C6H4OMe; C:4-MeOC6H4.N=C(Me).C(Me)=N.C6H4OMe. Also data for L=morpholine, NEt2, N-methylaniline.									

C5H11NO2		HL	Valine			CAS	72-18-4	(43)	
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	U	M	K(Pd(pn)+L)=11.36 pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.57, B2=11.70.	1999SSd (40742)	316
Pd++	gl	NaNO3	37°C	0.16M	M	M	K(PdA+L)=9.55 A is 1,3-diaminopropane.	1998ESa (40743)	317

Pd++ g1 KN03 20°C 0.5M U K1=9.62 B2=17.76 1974KHB (40744) 318

C5H11NO2 L (8054)
Alanine ethyl ester;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.20M	M	M	K1=5.15 K(Pd(dien)+L)=3.92	1987SKb (40866)	319

C5H11NO2S			HL	Methionine			CAS 63-68-3 (42)		
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3)COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	1.00M	C			K1=16.8 B2=34.30	2000SAb (41115)	320
Pd++	gl	NaNO3	25°C	0.10M	U	M		K(Pd(pn)+L)=10.37	1999SSd (41116)	321
pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.10, B2=11.08.										
Pd++	gl	NaNO3	37°C	0.16M	M	M		K(PdA+L)=8.83	1998ESa (41117)	322
A is 1,3-diaminopropane.										
Pd++	gl	KNO3	25°C	0.50M	U			K(Pd(en)+L)=9.14 K(Pd(en)+HL)=0.74	1978LIa (41118)	323

C5H11NO2S H2L Penicillamine CAS 52-66-4 (350)
DL-2-Amino-3-mercapto-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	1.00M	C		B2=48.2 K(Pd+HL)=27.3 K(Pd+2HL)=44.8	2000SAb (41278)	324

C5H11NO3 L (8128)
Serine ethyl ester;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	M	M	K(Pd(dien)+L)=4.43	1987SKb (41312)	325

C5H11NS2 HL CAS 147-84-2 (2126)
Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	dis	oth/un	?	?	U	M	B2=64.9 B(PdLCl)=44.6	1969BHd (41367)	326

Pd++	sp	non-aq	?	100%	U	M	K(Pd(HA)2+2HL=PdL2+2H2A)=1.6	1968SRg (41368)	327
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Medium: CCl4. H2A=dithizone

C5H12N2O2 HL Ornithine CAS 1069-31-4 (46)
2,5-Diaminopentanoic acid; H2N.CH2.CH2.CH2.CH(NH2)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaNO3	25°C	0.10M	U	M	K(Pd(pn)+L)=13.65 K(Pd(pn)+H+L)=19.86	1999SSd (41582)	328

pn is 1,2-diaminopropane. For aminoacid protonation, K1=10.58, B2=19.43.

Pd++	gl	NaNO3	37°C	0.16M	M	M	K(PdA+L)=11.58 K(PdA+H+L)=18.56	1998ESa (41583)	329
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A is 1,3-diaminopropane.

C5H12N2O2S HL Met-hydroxamic CAS 19253-87-3 (5992)
2-Amino-4-(methylthio)butanehydroxamic acid, Methionine hydrox.a.;
CH3.S.CH2.CH2.CH(NH2).CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Medium: 40% v/v EtOH/H₂O, 0.10 M NaClO₄.

C6H6	L	Benzene	CAS 71-43-2	(2143)
Benzene, cyclohexatriene;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		K(Pd3A3CO+L)<-1.15	1994PAa (43169)	336

Medium: MeOH. A=Bis(diphenylphosphino)methane

C6H6NBr L 3-Bromoaniline CAS 591-19-5 (758)
3-Bromoaniline; H2N.C6H4.Br

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.00M	U	M		1972VGa (43177) 337 K(M(H2O)4+L=M(H2O)3L+H2O)=6.30		

Medium: HCl04

C6H6N2O2 L m-Nitroaniline CAS 99-09-2 (464)
3-Nitroaminobenzene; H2N.C6H4.NO2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.00M	U	M		1972VGa (43389) K(M(H2O)4+L=M(H2O)3L+H2O)=5.68		338

Medium: HC104

C6H6N2O2 L p-Nitroaniline CAS 100-01-6 (465)
4-Nitroaminobenzene; H2N.C6H4.NO2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	kin	NaClO4	25°C	2.0M	U				1972VGa (43405)	339
								K(Pd(H2O)4+L=Pd(H2O)3L)=4.53		

C6H6O2 H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.20M	U				1981Cmb (43807)	340
								K(Pd+H2L=PdL+2H)=-2.2		

C6H6O4 HL Kojic acid CAS 501-30-4 (1800)
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Pd++       gl  alc/w   RT   20% C I      K1=7.44      1989MEb (44236) 341
Medium: 20% v/v MeOH/H2O. Data for 20-50% v/v MeOH/H2O, EtOH/H2O,
acetone/H2O, DMF/H2O and dioxane/H2O.
*****
C6H7N      L      Aniline      CAS 62-53-3 (583)
Aminobenzene, aniline; C6H5.NH2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       kin NaClO4 25°C 2.00M U      M      1972VGa (44876) 342
K(M(H2O)4+L=M(H2O)3L+H2O)=7.20
Medium: HClO4
*****
C6H7N5     HL      9-Methyladenine CAS 700-00-5 (4347)
9-Methyl-6-aminopurine;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  KNO3    25°C 0.20M C      1997WKa (45172) 343
K(PdACl+L=PdAL+Cl)=4.33
PdA is [PdH-1(gly-Met)].
*****
C6H8N2     L      2-Picolylamine CAS 29722-36-9 (502)
2-(Aminomethyl)pyridine; C5H4N.CH2NH2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  NaClO4 25°C 0.10M C      1997RSa (45359) 344
*K(PdL(H2O)2)=-4.43
*B2(PdL(H2O)2)=-13.07
*****
C6H8N2O2   HL      1-Methylthymine CAS 4160-72-9 (7411)
2,4-Dihydroxy-1,5-dimethylpyrimidine;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  KNO3    25°C 0.20M C      M      2001NSa (45383) 345
K(Pd(en)+L)=9.05
K(Pd(en)+2L)=14.76
*K(Pd(en)(H2O)L)=-8.44
K(Pd(en)(H2O)+L=Pd(en)(OH)L+H)=0.61,
K(2Pd(en)(H2O)2+2L=Pd2(en)2(OH)L2+H)=12.70
-----
Pd++       gl  KNO3    25°C 0.20M C      M      2001NSa (45384) 346
K(Pd(pic)+L)=9.56
K(Pd(pic)+2L)=15.40
*K(Pd(pic)(H2O)L)=-8.00

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$K(\text{Pd}(\text{pic})(\text{H}_2\text{O})+\text{L}=\text{Pd}(\text{pic})(\text{OH})\text{L}+\text{H})=1.56$

$K(2\text{Pd}(\text{pic})(\text{H}_2\text{O})_2+2\text{L}=\text{Pd}_2(\text{pic})_2(\text{OH})_2\text{L}+\text{H})=14.30$. Hp_{pic}=picric acid.

Pd++	gl	KN03	25°C	0.20M	C		K ₁ =7.71	2000NFa (45385)	347
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Pd++	gl	KN03	25°C	0.20M	C			1997WKa (45386)	348
------	----	------	------	-------	---	--	--	-----------------	-----

$K(\text{PdACl}+\text{L}=\text{PdAL}+\text{Cl})=7.26$

PdA is [PdH-1(gly-Met)].

C₆H₈O₄ H₂L CAS 5445-51-2 (69)

Cyclobutane-1,1-dicarboxylic acid; C₄H₆(C₂O₄)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaCl04	37°C	0.15M	C	M		2003TMb (45509)	349
------	----	--------	------	-------	---	---	--	-----------------	-----

$K(\text{Pd}(\text{en})+\text{L})=6.16$

Pd++	gl	NaCl04	25°C	0.10M	M	M		2002BSa (45510)	350
------	----	--------	------	-------	---	---	--	-----------------	-----

$K(\text{PdA}+\text{L})=6.61$

$K(\text{PdA}+\text{H}+\text{L})=9.69$

HA is S-methyl cysteine.

Pd++	gl	NaNO3	25°C	0.10M	C	M		2002MSb (45511)	351
------	----	-------	------	-------	---	---	--	-----------------	-----

$K(\text{PdA}+\text{L})=7.17$

$K(\text{PdA}+\text{H}+\text{L})=9.44$

A is N,N'-dimethylethylenediamine.

Pd++	gl	NaNO3	25°C	0.10M	C	M		2001SHc (45512)	352
------	----	-------	------	-------	---	---	--	-----------------	-----

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L})=8.45$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{H}+\text{L})=11.37$

Pd++	gl	NaNO3	25°C	0.10M	C	M		2001SHc (45513)	353
------	----	-------	------	-------	---	---	--	-----------------	-----

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A})=18.31$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+\text{H})=24.76$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+2\text{H})=27.05$

HA is uracil.

Pd++	gl	NaNO3	25°C	0.10M	C	M		2001SHc (45514)	354
------	----	-------	------	-------	---	---	--	-----------------	-----

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A})=20.14$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+\text{H})=26.74$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+2\text{H})=28.62$

HA is uridine.

Pd++	gl	NaNO3	25°C	0.10M	C	M		2001SHc (45515)	355
------	----	-------	------	-------	---	---	--	-----------------	-----

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A})=16.64$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+\text{H})=22.77$

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+\text{A}+2\text{H})=25.58$

HA is inosine.

Pd++	gl	NaNO3	25°C	0.10M	C	M		2001SHc (45516)	356
------	----	-------	------	-------	---	---	--	-----------------	-----

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A})=17.06$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A}+\text{H})=23.24$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A}+2\text{H})=27.08$

A is adenine.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (45517) 357
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A})=16.00$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A}+\text{H})=22.42$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A}+2\text{H})=27.92$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2\text{L}+\text{A}+3\text{H})=31.49$

H3A is inosine-5'-monophosphate.

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (45518) 358
 $K(\text{Pd}(\text{pn})+\text{L})=6.05$

pn is 1,2-diaminopropane. For acid protonation, $K_1=5.42$ $B_2=8.06$.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (45519) 359
 $K(\text{PdA}+\text{L})=6.39$

A is 1,3-diaminopropane.

Pd++ gl NaClO4 25°C 0.10M C M 1997RSa (45520) 360
 $K(\text{PdA}+\text{L})=7.34$

A=2-(Aminomethyl)pyridine

C6H8O6S H3L CAS 99-68-3 (3692)
 (Carboxymethylthio)butanedioic acid; $\text{HOOC}.\text{CH}(\text{S}.\text{CH}_2.\text{COOH}).\text{CH}_2.\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++ gl KNO3 20°C 0.10M U $K_1=5.20$ 1977CAd (45707) 361

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
 2-Hydroxypropane-1,2,3-tricarboxylic acid; $\text{HOOCCH}_2.\text{CH}(\text{OH})(\text{COOH}).\text{CH}_2\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Pd++ kin NaClO4 25°C 1.00M U H $K_1=3.46$ 1997SEa (46230) 362
 $K(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=0.66$

$\text{DH}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=-3.4 \text{ kJ mol}^{-1}$, $\text{DS}(\text{Pd}+\text{HL}=\text{PdL}+\text{H})=2 \text{ J K}^{-1} \text{ mol}^{-1}$

C6H9NO6 H3L NTA CAS 139-13-9 (191)
 Nitrilotriethanoic acid; $\text{N}(\text{CH}_2.\text{COOH})_3$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++ gl NaClO4 20°C 1.00M C M T $K_1=17.1$ $B_2=23.70$ 1976AMa (46982) 363
 $K(\text{PdL}+\text{H})=2.48$
 $K(\text{PdHL}+\text{H})=0.5$
 $K(\text{PdL}(\text{OH})+\text{H})=7.82$
 $K(\text{PdL}+\text{PdL}(\text{OH})=\text{Pd}_2\text{L}_2(\text{OH}))=3.1$

By exchange with PdBr4. $K(\text{PdL}+\text{Br})=2.7$

C6H9N3O2 HL Histidine CAS 71-00-1 (1)
2-Amino-3-(4'-imidazolyl)propanoic acid; $\text{H}_2\text{N}.\text{CH}(\text{CH}_2.\text{C}_3\text{H}_3\text{N}_2)\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (47602) 364

$K(\text{Pd}(\text{pn})+\text{L})=14.75$

pn is 1,2-diaminopropane. For aminoacid protonation, $K_1=9.53$, $B_2=15.81$,
 $B_3=17.81$.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (47603) 365

$K(\text{PdA}+\text{L})=12.48$

A is 1,3-diaminopropane.

C6H10O4S H2L CAS 111-17-1 (139)
3,3'-Thiodipropanoic acid; $\text{HOOC}.\text{CH}_2.\text{CH}_2.\text{S}.\text{CH}_2.\text{CH}_2.\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaCl 25°C 0.50M U M 1987CMc (48193) 366

$K(\text{PdCl}_4+\text{L}=\text{PdCl}_3\text{L}+\text{Cl})=5.42$

$K(\text{PdCl}_3\text{L}+\text{L}=\text{PdCl}_2\text{L}_2+\text{Cl})=2.87$

$K(\text{PdL}_2+\text{Cl})=4.30$

$K(\text{PdL}_2\text{Cl}+\text{Cl})=2.51$

Pd++ sp NaClO4 25°C 0.50M U 1986CCe (48194) 367

$B(\text{PdH}_2\text{L})=16.71$

$B(\text{PdH}_4\text{L}_2)=31.60$

$K(\text{Pd}+\text{H}_2\text{L})=7.40$

$K(\text{PdH}_2\text{L}+\text{H}_2\text{L})=5.58$

C6H10O4S2 H2L CAS 7244-02-2 (438)
1,2-Bis(carboxymethylthio)ethane; $\text{HOOC}.\text{CH}_2.\text{S}.\text{CH}_2.\text{CH}_2.\text{S}.\text{CH}_2.\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un 25°C 0.10M U $K_1=4.48$ $B_2=6.91$ 1978POa (48249) 368

C6H10O4S2 H2L CAS 1119-62-6 (3697)
3,3'-Di(thiopropanoic acid); $\text{HOOC}.\text{CH}_2.\text{CH}_2.\text{S}.\text{S}.\text{CH}_2.\text{CH}_2.\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 0.50M U 1986CCe (48270) 369

$B(\text{PdH}_2\text{L})=15.25$

$B(\text{Pd}_2\text{H}_2\text{L})=19.67$

$K(\text{Pd}+\text{H}_2\text{L})=5.92$

$K(\text{PdH}_2\text{L}+\text{H}_2\text{L})=10.34$

C6H10O4Se2 H2L CAS 86515-79-7 (6099)
Ethylene-bis-selenoglycolic acid; HOOC.CH2.Se.CH2.CH2.Se.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ con NaCl 25°C 3.00M C K1=6.32 B2=11.97 1988PFb (48297) 370

C6H11NO2 HL CAS 89203-64-5 (3435)
1-Pyrrolidine-1-ethanoic acid, 1-Azacyclopentane-1-ethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp none 25°C 0.0 U K1=11.20 B2=21.23 1974HFa (48504) 371

C6H11NO4 H2L Amino adipic CAS 542-32-5 (1259)
2-Aminohexanedioic acid; HOOC.CH2.CH2.CH2.CH(NH2).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl none 25°C 0.0 U 1979FWa (48583) 372

K(PdL2+H)=5.04
K(PdHL2+H)=4.45
K(PdCl4+2HL=PdH2L2+4Cl)=10.5

C6H11N3O4 HL Gly-Gly-Gly CAS 556-33-2 (415)
Glycyl-glycyl-glycine; H2N.CH2.CO.NH.CH2.CO.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C B2=23.0 1999AJa (48987) 373

B(PdLC1)=17.91
B(PdH-1LC1)=14.64
B(PdH-2L)=9.07
B(PdH-3L)=-1.15

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-1L2)=19.81; B(PdH-2L2)=13.40.

Pd++ sp oth/un 25°C ? U 1978Cwa (48988) 374

K(PdH-2L+H)=2.2
K(PdH-1L+H)=1.5

C6H12 L CAS 592-41-6 (2771)
1-Hexene; CH2:CH(CH2)3.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ oth non-aq 30°C 100% U M 1974KKb (49012) 375

K(PdCl2+L)=0.28

Medium: N-methylacetamide

C6H12 L CAS 760-21-4 (2772)
2-Ethyl-1-butene; CH2:C(C2H5).CH2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	oth	non-aq	30°C	100%	U	M			1974KKb (49015)	376
									K(PdCl2+L)=-0.89	

Medium: N-methylacetamide

C6H12 L CAS 763-29-1 (2770)
2-Methyl-1-pentene; CH2:C(CH3).CH2.CH2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	oth	non-aq	30°C	100%	U	M			1974KKb (49017)	377
									K(PdCl2+L)=-0.85	

Medium: N-methylacetamide

C6H12 L CAS 691-37-2 (2767)
4-Methyl-1-pentene; CH2:CH.CH2.CH(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	oth	non-aq	30°C	100%	U	M			1974KKb (49019)	378
									K(PdCl2+L)=0.18	

Medium: N-methylacetamide

C6H12 L CAS 7668-21-3 (2774)
cis-2-Hexene; CH3.CH:CH.CH2.CH2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	oth	non-aq	30°C	100%	U	M			1974KKb (49021)	379
									K(PdCl2+L)=0.11	

Medium: N-methylacetamide

C6H12 L (2768)
cis-4-Methyl-2-pentene; CH3.CH:CH.CH(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	oth	non-aq	30°C	100%	U	M			1974KKb (49023)	380
									K(PdCl2+L)=0.26	

Medium: N-methylacetamide

C6H12 L CAS 4050-45-7 (2773)
trans-2-Hexene; CH3.CH:CH.CH2.CH2.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ oth non-aq 30°C 100% U M 1974KKb (49025) 381
K(PdCl₂+L)=-0.31

Medium: N-methylacetamide

C6H12 L CAS 4461-48-7 (2769)
trans-4-Methyl-2-pentene; CH₃.CH:CH.CH(CH₃)₂

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ oth non-aq 30°C 100% U M 1974KKb (49027) 382
K(PdCl₂+L)=-0.42

Medium: N-methylacetamide

C6H12N2O3 HL B-Ala-B-Ala CAS 34322-87-7 (2118)
3-Alanyl-3-alanine; H₂N.CH₂.CH₂.CO.NH.CH₂.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO₃ 25°C 0.20M C 2003AMb (49061) 383
K(PdH-1L)=11.19
K(PdH-2L)=2.52
K(PdH-1L2)=17.76

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO₃/0.10 M KCl.

C6H12O2S2 HL CAS 35088-67-6 (2829)
1-Ethylthio-2-thiocarboxymethylethane; C₂H₅.S.CH₂.CH₂.S.CH₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un 25°C 0.10M U K1=6.34 B2=11.03 1978POa (49451) 384

C6H13NO2 HL Isoleucine CAS 73-32-5 (424)
2-Amino-3-methylpentanoic acid; CH₃.CH₂.CH(CH₃).CH(NH₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO₃ 20°C 0.5M U K1=9.71 B2=18.15 1974KHb (49911) 385

C6H13NO2 HL Leucine CAS 61-90-5 (47)
2-Amino-4-methylpentanoic acid; H₂N.CH(CH₂.CH(CH₃)₂).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO₃ 25°C 0.50M U T 1978LIa (50096) 386
K(Pd(en)+L)=11.41

Pd++ gl KNO₃ 20°C 0.5M U K1=9.94 B2=18.17 1974KHb (50097) 387

C6H13NO2S HL Ethionine CAS 67-21-0 (1909)
2-Amino-4-(ethylthio)butanoic acid; CH₃.CH₂.S.CH₂.CH₂.CH(NH₂).COOH

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  NaCl04 25°C 1.00M C          K1=16.8   B2=34.00  2000SAb (50266) 388
-----
Pd++       gl  NaCl   25°C 0.16M U          K1=9.112  B2=14.361 1986AEa (50267) 389
                                           B(Pd2L)=18.487
                                           B(Pd2H2L)=23.979
                                           B(PdH-1L)=5.059

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*****
C6H13N3O3          HL      Citrulline          (579)
2-Amino-5-ureidovaleic acid; H2N.CO.NH.CH2.CH2.CH2.CH(NH2).COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  KNO3   25°C 0.10M C          B2=16.23          1991GLb (50585) 390
-----
C6H14N2O2          HL      Lysine          CAS 56-87-1 (41)
2,6-Diaminohexanoic acid; H2N.(CH2)4.CH(NH2)COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  NaNO3  25°C 0.10M U      M          1999SSd (50830) 391
                                           K(Pd(pn)+L)=11.49
                                           K(Pd(pn)+H+L)=20.44
pn is 1,2-diaminopropane. For aminoacid protonation, K1=10.44, B2=19.66.
-----
Pd++       gl  NaNO3  37°C 0.16M M      M          1998ESa (50831) 392
                                           K(PdA+L)=9.28
                                           K(PdA+H+L)=19.03

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A is 1,3-diaminopropane.
*****
C6H14O2S          L          CAS 10595-09-2 (3698)
3,3'-Thiodipropanol; S(CH2CH2CH2OH)2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       kin oth/un 25°C 1.00M U          K1eff=4.51          1996SEa (51036) 393
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Medium: 1.00 M HClO4.
*****
C6H14O2S2          L          CAS 5244-34-8 (4390)
3,6-Dithiaoctan-1,8-diol; HO.CH2.CH2.S.CH2.CH2.S.CH2.CH2.OH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp  KCl    25°C 1.00M U          B2=21.63          1991ZPa (51038) 394
-----
C6H18N4          L      Trien-tetramine CAS 112-24-3 (11)
1,4,7,10-Tetraazadecane; H2N.CH2.CH2.NH.CH2.CH2.NH.CH2.CH2.NH2

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  NaClO4 25°C 1.00M C I M    K1=40.1        1985YAa (52131) 395
*****
C6H18N4           L    Tren                      CAS 4097-89-6 (817)
2,2',2''-Triaminotriethylamine; (H2N.CH2.CH2)3N
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  KNO3   25°C 1.00M C    M                      1986ANa (52207) 396
                      B(Pd2L2)=77.4
Ternary complexes with Cl-, Br-, I- and SCN-. pH-metric and spec. study.
*****
C7H5N           L    Cyanobenzene      CAS 100-47-0 (4406)
Cyanobenzene, benzonitrile; C6H5.CN
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp  alc/w  25°C 100% U    M                      1994PAa (52570) 397
                      K(Pd3A3CO+L)=0.13
Medium: MeOH. A=Bis(diphenylphosphino)methane
*****
C7H5NO4           H2L    Dipicolinic aci  CAS 449-83-2 (418)
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Pd++       gl  KCl     25°C 0.20M U          K1=16.0        1980KDb (52799) 398
*****
C7H6O2           HL    Salicylaldehyde  CAS 90-02-8 (193)
2-Hydroxybenzaldehyde, Salicylaldehyde; HO.C6H4.CHO
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       gl  diox/w 25°C 50% U          K1=7.74  B2=14.77  1949MMA (53631) 399
*****
C7H6O2           HL    Benzoic Acid     CAS 65-85-0 (462)
Benzenecarboxylic acid; C6H5.COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Pd++       sp  alc/w  25°C 100% U    M                      1994PAa (53853) 400
                      K(Pd3A3CO+L)=4.0
Medium: MeOH. A=Bis(diphenylphosphino)methane
*****
C7H7NO2           HL                      CAS 150-13-0 (1376)
4-Aminobenzoic acid; H2N.C6H4.COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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 Pd++ sp alc/w 25°C 100% U M 1994PAa (55390) 401
 $K(\text{Pd3A3CO}+\text{L})=3.52$

Medium: MeOH. A=Bis(diphenylphosphino)methane

C7H7NO2 HL CAS 495-18-1 (184)

Benzohydroxamic acid; C6H5.CO.NH.OH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=9.52 B2=17.55 1969JSa (55512) 402

C7H8 L CAS 108-88-3 (2144)

Toluene; C6H5.CH3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 100% U M 1994PAa (55785) 403

$K(\text{Pd3A3CO}+\text{L})=-1.10$

Medium: MeOH. A=Bis(diphenylphosphino)methane

C7H8OS L CAS 1193-82-4 (1881)

Phenylmethysulfoxide; C6H5.SO.CH3

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 95% U 1982CCa (56055) 404

$K(\text{PdCl4}+\text{L}=\text{PdLC13}+\text{Cl})=0.94$

C7H8O2 H2L Methylcatechol CAS 452-86-8 (525)

1,2-Dihydroxy-4-methylbenzene; CH3.C6H3(OH)2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un 25°C 0.20M U 1981Cma (56075) 405

$K(\text{Pd}+\text{H2L}=\text{PdL}+2\text{H})=2.40$

C7H9N L 3-Methylaniline CAS 108-44-1 (755)

3-Methylaniline (3-Toluidine); CH3.C6H4.NH2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin oth/un 25°C ? U M 1972VGa (56309) 406

$K(\text{M}(\text{H2O})4+\text{L}=\text{M}(\text{H2O})3\text{L}+\text{H2O})=7.57$

C7H9N L 4-Methylaniline CAS 106-49-0 (754)

4-Methylaniline (4-Toluidine); CH3.C6H4.NH2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin oth/un 25°C ? U M 1972VGa (56343) 407
K(M(H2O)4+L=M(H2O)3L+H2O)=8.04

C7H9NO L p-Anisidine CAS 104-94-7 (3764)
4-Methoxyaniline; CH3O.C6H4.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ kin oth/un 25°C 2.0M U M 1972VGa (56397) 408
K(M(H2O)4+L=M(H2O)3L+H2O)=7.81

C7H9N5O HL 9-Ethylguanine CAS 879-08-3 (6679)
9-Ethyl-2-amino-6-hydroxypurine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C M 2003Nfa (56518) 409
K(PdA+L)=8.11
K(PdA+H+L)=15.06
K(2PdA+L)=14.95

A is bis-((2-pyridyl)methyl)amine

C7H10N4O2S L Sulfaguanidine CAS 57-67-0 (4469)
4-Aminobenzenesulfonyl guanidine; H2N.C(:NH).NH.SO2.C6H4.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 ? 1.0M U M 1970RGa (56704) 410
K(PdCl2+L)=5.42
K(PdCl2L+L)=4.38

C7H11N3O L Acetylhistamine CAS 673-49-4 (7412)
4-(2'-Acetylaminoethyl)imidazole; C3H3N2.CH2CH2.NH.COCH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C M 2003Nfa (56961) 411
K(PdA+L)=7.72
K(2PdA+L=(PdA)2H-1L+H)=6.13

A is bis-((2-pyridyl)methyl)amine

Pd++ gl KNO3 25°C 0.20M C 1997Wka (56962) 412
K(PdACl+L=PdAL+Cl)=5.48
K(2PdACl+L=Pd2A2H-1L+2Cl)=-0.9

PdA is [PdH-1(gly-Met)].

C7H11N3O2 L CAS 7389-87-9 (3162)
Histidine methyl ester

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Pd++ gl KNO3 25°C 0.20M M M 1987SKb (57005) 413
 K(Pd(dien)+L)=4.61
 K(Pd(dien)+H+L)=10.58

C7H13NO2 HL CAS 3235-67-4 (3772)
 Piperidine-N-ethanoic acid; C5H10N-CH2.COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp none 25°C 0.0 U K1=10.32 B2=19.72 1974HFa (57456) 414

C7H13NO3S HL CAS 65-82-7 (8508)
 N-Acetylmethionine;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C M 2003NFa (57495) 415

K(PdA+L)=3.41

A is bis-((2-pyridyl)methyl)amine. Competitive method using uridine.

 Pd++ gl KNO3 25°C 0.20M C HM 2000NFa (57496) 416

K(Pd(dien)+H+L)=8.49

K(Pd(dien)+L)=5.61

K(PdA+H+L)=7.07

K(PdA+L)=3.66

Method: uridine as a competitive ligand. A is terpyridine. K(Pd(dien)+OH)=
 6.25, K(PdA+OH)=6.91. By calorimetry: DH(Pd(dien)+L)=-38.8 kJ mol⁻¹.

 Pd++ gl KNO3 25°C 0.20M C HM 2000NFa (57497) 417

K(Pd(gly-gly)+H+L)=8.74

K(Pd(gly-gly)+L)=4.89

K(Pd(gly-ala)+H+L)=8.76

K(Pd(gly-ala)+L)=4.91

Method: uridine as a competitive ligand. K(Pd(gly-gly)+OH)=4.64,
 K(Pd(gly-ala)+OH)=4.72. By calorimetry: DH(Pd(gly-ala)+L)=-38.0 kJ mol⁻¹.

 Pd++ gl KNO3 25°C 0.20M C HM 2000NFa (57498) 418

K(Pd(gly-met)+H+L)=7.29

K(Pd(gly-met)+L)=3.24

K(Pd(gly-met)+OH)=4.82

Method: uridine as a competitive ligand.
 By calorimetry: DH(Pd(gly-met)+L)=-21.2 kJ mol⁻¹.

C7H13NO4 H2L Aminopimelic CAS 627-76-9 (1260)
 2-Amino-heptanedioic acid; HOOC.(CH2)4.CH(NH2).COOH

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KCl 25°C 0.10M U 1979FWa (57501) 419

$K(\text{Pd}(\text{HL})_2=\text{Pd}(\text{HL})\text{L}+\text{H})=4.58$
 $K(\text{Pd}(\text{HL})\text{L}=\text{PdL}_2+\text{H})=5.33$
 $K(\text{PdCl}_4+2\text{HL}=\text{Pd}(\text{HL})_2+4\text{Cl})=11.5$

C7H13N3O4 HL Ala-Asn CAS 1999-41-3 (5934)
 Alanyl-asparagine; $\text{NH}_2.\text{CH}(\text{CH}_3.\text{CO}.\text{NH}.\text{CH}(\text{CH}_2.\text{CO}.\text{NH}_2).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaCl	20°C	0.15M	U			1990YKa (57648)	420

$\text{Keff}(\text{Pd}+\text{L}+2\text{Cl})=21.5$
 Eff constant : stability of PdCl_4 is not accounted

C7H13N3O4 HL Gly-b-Ala-Gly CAS 42538-54-5 (9058)
 Glycyl-beta-alanylglycine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	oth/un	25°C	0.20M	C		$\text{K}_1=16.26$ $\text{K}(\text{PdH}-1\text{L})=12.06$ $\text{K}(\text{PdH}-2\text{L})=11.79$	2003AMb (57660)	421

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO_3 /0.10 M KCl .

C7H13N3O4 HL Gly-Gly-b-Ala CAS 42538-53-4 (4453)
 Glycylglycyl-beta-alanine; $\text{H}_2\text{N}.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}_2.\text{CH}_2.\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	oth/un	25°C	0.20M	C		$\text{K}_1=16.79$ $\text{K}(\text{PdH}-1\text{L})=10.97$ $\text{K}(\text{PdH}-2\text{L})=10.98$	2003AMb (57679)	422

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO_3 /0.10 M KCl .

C7H13N3O4 HL Gly-Gly-Ala CAS 19729-30-7 (3775)
 Glycylglycylalanine; $\text{H}_2\text{N}.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}(\text{CH}_3).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO_3	25°C	0.20M	C		$\text{B}_2=23.7$ $\text{B}(\text{PdLCl})=17.91$ $\text{B}(\text{PdH}-1\text{LCl})=14.45$ $\text{B}(\text{PdH}-2\text{L})=8.99$ $\text{B}(\text{PdH}-3\text{L})=-2.40$	1999AJa (57687)	423

Medium: 0.1 M KNO_3 , 0.1 M KCl . $\text{B}(\text{PdH}-1\text{L}_2)=19.60$; $\text{B}(\text{PdH}-2\text{L}_2)=15.74$.

C7H13N3O4 HL b-Ala-Gly-Gly CAS 42538-55-6 (4452)
 beta-Alanylglycylglycine; $\text{H}_2\text{N}.\text{CH}_2.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}_2.\text{CO}.\text{NH}.\text{CH}_2.\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++ gl oth/un 25°C 0.20M C K1=14.40 2003AMB (57694) 424

K(PdH-1L)=8.76

K(PdH-2L)=9.03

Method: competition with chloride (0.1 M). Medium: 0.10 M KNO3/0.10 M KCl.

C7H14N2O3S HL Gly-Met CAS 554-94-9 (726)

Glycyl-methionine; H2N.CH2.CO.NH.CH(CH2.CH2.S.CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KCl 25°C 0.20M C M 2001BNa (57800) 425

*K(PdL)=-3.61

K(PdH-1L+Cl)=2.03

*K(PdH-1L)=-5.34

K(PdH-1L+H+L)=11.47

K(PdH-1L+L)=4.56, K(PdH-1L+glygly)=4.72.

C7H15NO5S HL MOPSO CAS 68399-77-9 (1967)

3-(N-Morpholino)-2-hydroxypropane sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.10M C M K1=3.50 2001AAa (57996) 426

Also data for ternary complexes with 5'-GMP, 5'-IMP and 5'-CMP.

C7H17N2O4PS H2L CAS 82611-22-1 (7392)

Methionyl-1-aminoethylphosphonic acid; H2L

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.10M C B2=21.60 1997LBa (58201) 427

B(PdHLC12)=24.01

B(PdLCl)=17.87

B(PdH-1L)=10.76

B(PdH-2L)=-0.01

Data are for (S,S)-isomer. B(PdH2L2)=34.96, B(PdHL2)=28.74, B(PdH-1L2)=12.51

B(PdH-2L2)=2.48. Data also for (R,S)-isomer.

Pd++ gl KCl 25°C 0.10M U 1996BRa (58202) 428

K(Pd+2L+2H)=35.35

K(Pd+2L)=21.99

K(Pd+2L+H)=29.14

H2L: S,S-diastereoisomer

Pd++ gl KCl 25°C 0.10M U 1996BRa (58203) 429

K(Pd+2L+2H)=35.01

K(Pd+2L)=21.54

K(Pd+2L+H)=28.71

H2L: S,R-diastereoisomer

C7H20N4 L CAS 4741-99-5 (12)
1,4,8,11-Tetraazaundecane; H2N.CH2.CH2.NH.CH2.CH2.CH2.NH.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	1.00M	C			K1=46.3	1985YAa (58359)	430

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)
Benzene-1,2-dicarboxylic acid; C6H4(COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	?	M	M		K1=3.31 K(PdA+L)=3.04	1988SKa (59003)	431

A=diethylenetriamine

C8H8NO2Cl HL CAS 61756-69-2 (4569)
N-Acetyl-N-(4-chlorophenyl)hydroxamine; Cl.C6H4.N(CO.CH3).OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U			K1=9.63 B2=17.37	1968JSb (59281)	432

Medium: 70% dioxan, 0.1 M KCl

C8H8O2 HL p-Toluic acid CAS 99-94-5 (1372)
4-Methylbenzoic acid; CH3.C6H4.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		K(Pd3A3CO+L)=3.99	1994PAa (59501)	433

Medium: MeOH. A=Bis(diphenylphosphino)methane

C8H9NO5 HL CAS 4822-44-0 (3240)
N-(Mercaptoacetyl)aniline (thioglycolanilide); C6H5.NH.CO.CH2.SH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	oth	diox/w	30°C	70%	U			B2=24.34	1973BSa (60163)	434

Medium: 0.1 M KCl

C8H9NO3 HL CAS 5663-54-7 (1095)
2,4-Dihydroxy-acetophenone oxime; (HO)2.C6H3.C(CH3):NOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	27°C	60%	U	I		K1=8.99 B2=17.80	1974SRa (60400)	435

C8H9NO4S H2L CAS 7717-21-7 (3846)
N-(Phenylsulfonyl)aminoethanoic acid; C6H5SO2NHCH2COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	vlt	NaClO4	25°C	0.10M	U			K1=18.9 B2=24.4	1990GBb (60517)	436

C8H9N3O7		H2L						Uramildiacetic CAS 13055-06-5 (185)		
5-Amino-2,4,6-trioxo-1,3-perhydrodiazimino-N,N-diethanoic acid;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	KCl	20°C	0.10M	U	M			1987KUa (60652)	437
B(PdCl(OH)L)=22.12										

C8H9O3P		H2L						CAS 1707-08-0 (1969)		
2-Styrylphosphonic acid; C6H5.CH:CH.PO3H2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.12M	U			K1=3.33 B2=6.55	1979RZb (60673)	438

C8H10		L						p-Xylene CAS 106-42-3 (2145)		
1,4-Dimethylbenzene, 4-Xylene; CH3.C6H4.CH3										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M			1994PAa (60682)	439
K(Pd3A3CO+L)=-1.15										
Medium: MeOH. A=Bis(diphenylphosphino)methane										

C8H10N3OCl		HL						CAS 5756-79-6 (4578)		
3-Ethyl-3-hydroxy-1-(2-chlorophenyl)triazene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U			K1=10.49 B2=20.46	1968DSa (60784)	440
Medium: 70% dioxan, 0.1 M KCl										

C8H10N3OCl		HL						CAS 5756-78-5 (4579)		
3-Ethyl-3-hydroxy-1-(4-chlorophenyl)triazene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U			K1=10.68 B2=20.66	1968DSa (60789)	441
Medium: 70% dioxan, 0.1 M KCl										

C8H11N		L						DiMethylaniline CAS 121-69-7 (1343)		
N-Phenyl-N,N-dimethylamine; C6H5.N(CH3)2										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo

Pd++ sp non-aq 25°C 100% U M 1979SSa (60989) 442

K(PdA+L)=1.0

A=Tetraphenylporphyrin (in its excited triplet state)

C8H11N30 HL CAS 5956-70-7 (4529)

3-Hydroxy-3-methyl-1-(4-tolyl)triazene; CH3.C6H4.N:N.N(OH).CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=11.77 B2=23.10 1970DSb (61244) 443

Medium: 70% dioxan, 0.1 M KCl

C8H11N302 HL CAS 5756-72-9 (4533)

3-Hydroxy-3-methyl-1-(4'-methoxyphenyl)triazene; CH3O.C6H4.N:N.N(OH).CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=12.25 B2=23.70 1970DSb (61257) 444

Medium: 70% dioxan, 0.1 M KCl

C8H11N303 HL CAS 2497-02-1 (3230)

Acetyl-L-histidine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C M 2003NFa (61275) 445

K(PdA+L)=8.47

K(PdA+H+L)=11.58

K(2PdA+L=(PdA)2H-1L+H)=6.99

A is bis-((2-pyridyl)methyl)amine

Pd++ gl KNO3 25°C 0.20M C 1997Wka (61276) 446

K(PdACl+L=PdAL+Cl)=5.33

K(2PdACl+L=Pd2A2H-1L+2Cl)=0.1

PdA is [PdH-1(gly-Met)].

C8H14N405 HL Tetraglycine CAS 637-84-3 (1849)

Glycyl-Glycyl-Glycyl-Glycine; H2N.CH2.CO.NH.CH2.CO.NH.CH2.CO.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C 1999AJa (62024) 447

B(PdLC1)=18.25

B(PdH-1LC1)=14.81

B(PdH-2L)=10.13

B(PdH-3L)=2.45

Medium: 0.1 M KNO3, 0.1 M KCl.

C8H14O5S2 H2L CAS 4408-66-6 (8332)

Oxybis(ethylenethio)diethanoic acid;

$$K(\text{Pd}(\text{pn})+\text{L}=\text{PdH}-1(\text{pn})\text{L}+\text{H})=3.30$$

pn is 1,2-diaminopropane. For aminoacid protonation, $K_1=8.13$.

C8H22N4 L CAS 35513-90-7 (1545)

1,4,9,12-Tetraazadodecane; $\text{NH}_2.(\text{CH}_2)_2.\text{NH}.(\text{CH}_2)_4.\text{NH}.(\text{CH}_2)_2.\text{NH}_2$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaClO4 25°C 1.00M C K1=42.0 1985YAa (63383) 454

C9H6N04IS H2L Ferron CAS 547-91-1 (275)

7-Iodo-8-hydroxyquinoline-5-sulfonic acid; $(\text{HO})(\text{HO}_3\text{S})\text{C}_9\text{H}_4\text{NI}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un 25°C 0.0 U 1967MBe (63822) 455

$$K(?)=9.05$$

C9H7N04S H2L Sulfoxine CAS 84-88-8 (448)

8-Hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ oth oth/un ? ? U K1=11.6 B2=20.14 1973BIb (64573) 456

Method: fluorescence

C9H11N0S HL CAS 36076-50-3 (4680)

N-Phenyl-N-methyl-2-mercaptoacetamide; $\text{HS}.\text{CH}_2.\text{CO}.\text{N}(\text{CH}_3).\text{C}_6\text{H}_5$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ oth diox/w 30°C 70% U K1=9.87 B2=18.84 1973BSc (65682) 457

C9H11N02 HL Phenylalanine CAS 63-91-2 (2)

2-Amino-3-phenylpropanoic acid; $\text{H}_2\text{N}.\text{CH}(\text{CH}_2.\text{C}_6\text{H}_5).\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.50M U 1978LIa (65966) 458

$$K(\text{Pd}(\text{en})+\text{L})=10.86$$

Pd++ gl KNO3 20°C 0.5M U K1=9.32 B2=18.27 1974KHb (65967) 459

C9H11N02 HL B-Phenylalanine CAS 614-19-7 (187)

3-Amino-3-phenyl-propanoic acid; $\text{H}_2\text{N}.\text{CH}(\text{C}_6\text{H}_5).\text{CH}_2.\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (66011) 460

$$K(\text{Pd}(\text{pn})+\text{L})=11.06$$

pn is 1,2-diaminopropane. For aminoacid protonation, K1=9.12, B2=11.01.

C9H11NO4S H2L CAS 1080-44-0 (4682)
N-(4-Toluenesulfonyl)glycine, N-tosylglycine; CH3.C6H4.SO2.NH.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ vlt NaClO4 25°C 0.10M U 1994BGa (66427) 461

Keff(Pd+H2L=PdL)=19.9

Beff(Pd+2H2L=PdL2)=23.3

Complex formation involves loss of the amide proton.

Pd++ vlt NaClO4 25°C 0.10M U K1=17.8 B2=23.4 1990GBb (66428) 462

C9H11NO4S H2L (6960)
N-(Phenylsulfonyl)-2-aminopropanoic acid; C6H5.SO2.NH.CH(CH3)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ vlt NaClO4 25°C 0.10M U 1994BGa (66429) 463

Keff(Pd+H2L=PdL)=20.6

Beff(Pd+2H2L=PdL2)=23.0

Complex formation involves loss of the amide proton.

C9H11NO4S H2L (6961)
N-(Phenylsulfonyl)-3-aminopropanoic acid; C6H5.SO2.NH.CH2.CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ vlt NaClO4 25°C 0.10M U 1994BGa (66430) 464

Keff(Pd+H2L=PdL)=17.1

Beff(Pd+2H2L=PdL2)=20.8

Complex formation involves loss of the amide proton.

C9H12N2O6 HL Uridine CAS 58-96-8 (828)
Uracil-1-beta-D-ribofuranoside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C M 2003NFa (66703) 465

K(PdA+L)=8.90

A is bis-((2-pyridyl)methyl)amine

Pd++ gl NaNO3 25°C 0.10M C M 2002MSb (66704) 466

K(PdA+L)=8.70

K(PdA+2L)=14.37

K(PdA+H+L)=15.17

A is N,N'-dimethylethylenediamine.

Pd++ gl KNO3 25°C 0.20M C M 2001NSa (66705) 467

$K(\text{Pd}(\text{en})+\text{L})=8.98$
 $K(\text{Pd}(\text{en})+2\text{L})=14.80$
 $*K(\text{Pd}(\text{en})(\text{H}_2\text{O})\text{L})=-7.67$
 $K(\text{Pd}(\text{en})(\text{H}_2\text{O})+\text{L}=\text{Pd}(\text{en})(\text{OH})\text{L}+\text{H})=1.31,$
 $K(2\text{Pd}(\text{en})(\text{H}_2\text{O})_2+2\text{L}=\text{Pd}_2(\text{en})_2(\text{OH})\text{L}_2+\text{H})=12.14$

Pd++ gl KNO₃ 25°C 0.20M C M 2001NSa (66706) 468
 $K(\text{Pd}(\text{pic})+\text{L})=9.20$
 $K(\text{Pd}(\text{pic})+2\text{L})=15.09$
 $*K(\text{Pd}(\text{pic})(\text{H}_2\text{O})\text{L})=-7.94$
 $K(\text{Pd}(\text{pic})(\text{H}_2\text{O})+\text{L}=\text{Pd}(\text{pic})(\text{OH})\text{L}+\text{H})=1.26,$
 $K(2\text{Pd}(\text{pic})(\text{H}_2\text{O})_2+2\text{L}=\text{Pd}_2(\text{pic})_2(\text{OH})\text{L}_2+\text{H})=13.82.$ Hpic=picric acid.

Pd++ gl NaNO₃ 25°C 0.10M C M 2001SHc (66707) 469
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L})=9.71$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{H}+\text{L})=13.29$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L})=16.88$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+\text{H})=22.65$

Pd++ gl KNO₃ 25°C 0.20M C K₁=7.42 2000NFa (66708) 470

Pd++ gl KCl 25°C 0.20M U M 1997KFa (66709) 471
 $K(\text{Pd}(\text{dien})\text{Cl}+\text{L})=7.42$
 $K(\text{Pd}(\text{terpy})\text{Cl}+\text{L})=7.56$
 dien=diethylenetriamine, terpy=2,2'-6',2''-terpyridine. Data also for many related nucleobases

Pd++ gl KNO₃ 25°C 0.20M C 1997WKa (66710) 472
 $K(\text{PdACl}+\text{L}=\text{PdAL}+\text{Cl})=7.00$
 PdA is [PdH-1(gly-met)].

Pd++ gl KNO₃ 25°C 0.50M U M 1981LIa (66711) 473
 $K(\text{Pd}(\text{en})(\text{H}_2\text{O})_2+\text{L})=8.65$
 $K(\text{Pd}(\text{en})(\text{H}_2\text{O})\text{L}+\text{L})=5.92$
 $K(\text{Pd}(\text{dien})(\text{H}_2\text{O})+\text{L})=8.08$

 C9H13N2O9P H3L UMP-5 CAS 58-97-9 (2948)
 Uridine-5'-monophosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	nmr	oth/un	23°C	0.30M	U	M		1985PGa (66980) 474		
$\text{Keff}(\text{PdA}+\text{HL})=2.99$ A=Tetrakis(4-N-methylpyridyl)porphyrin. pD=7.0										

 C9H13N3O5 L Cytidine CAS 65-46-3 (2152)
 Cytidine, Cytosine-1-beta-D-ribofuranoside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl KNO3 25°C 0.20M C M 2003NFa (67073) 475
K(PdA+L)=5.83

A is bis-((2-pyridyl)methyl)amine

Pd++ sp NaClO4 25°C 1.0M U M 1984ETa (67074) 476
K(PdCl4+L=PdLC13+Cl)=4.49
K(PdLC13+L=PdL2C12+Cl)=3.45
K(Pd(en)Cl2+L=PdenLC1+Cl)=3.32
K(Pd(en)LC1+L=PdenL2+Cl)=2.56

C9H14N3O8P H2L CMP-5 CAS 63-37-6 (1243)
Cytidine-5'-monophosphoric acid, Cytidilic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.10M C M K1=3.35 2001AAa (67262) 477
Also data for ternary complexes with MOPSO, TAPSO and ACES.

C9H17N3O4S HL Gly-Met-Gly CAS 51529-34-1 (7566)
Glycylmethionylglycine; NH2CH2CONHCH(CH2CH2SCH3)CONHCH2COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KCl 25°C 0.20M C M 2001BNa (67872) 478
*K(PdL)=-3.65
K(PdH-1L+Cl)=2.21
*K(PdH-1L)=-5.57
K(PdH-1L+H+L)=11.49

K(PdH-1L+L)=4.71, K(PdH-1L+glygly)=4.50, K(PdH-1L+L=PdH-2L2+H)=-4.99.

C9H18N2O3 HL Leu-Ala CAS 7298-84-2 (4659)
Leucylalanine- H2N.CH(CH2.CH(CH3)2).CO.NH.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (67912) 479
K(Pd(pn)+L)=8.19
K(Pd(pn)+L=PdH-1(pn)L+H)=3.74

pn is 1,2-diaminopropane. For aminoacid protonation, K1=8.13.

C9H23N3 L CAS 3030-47-5 (4605)
N,N,N',N'',N'''-Pentamethyl-diethylenetriamine; (CH3)2NCH2CH2N(CH3)CH2CH2N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl R4N.X 25°C 0.10M C M 1998BBa (68282) 480
B(PdLC1)=24.9
B(PdH-1L)=14.1

Medium: 0.1 M NMe4Cl

Pd++ gl NaClO4 25°C 0.50M C I 1981GMf (68283) 481

K(PdL=PdLOH+H)=-7.293

K(PdL+PdLOH)=1.08

In 0.5 NaNO3, K(PdL=PdLOH+H)=-7.241, K(PdL+PdLOH)=0.70

C9H24N4 L CAS 129880-56-4 (1533)

1,4,10,13-Tetraazatridecane; H2N.(CH2)2.NH.(CH2)5.NH.(CH2)2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.10M C K1=37.9 1985YAA (68336) 482

C10H7NO2 HL CAS 131-91-9 (2668)

1-Nitroso-2-naphthol, alpha-Nitroso-beta-naphthol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl alc/w RT 40% M K1=5.53 B2= 8.86 1993RAB (68585) 483

Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

C10H7NO2 HL CAS 132-53-6 (2524)

2-Nitroso-1-naphthol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl alc/w RT 40% M K1=4.46 B2= 8.92 1993RAB (68653) 484

Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

C10H7NO5S H2L CAS 3682-32-4 (1812)

2-Nitroso-1-hydroxynaphthalene-4-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl oth/un RT 0.10M M K1=4.46 B2= 8.68 1993RAB (68892) 485

Medium not stated.

C10H7NO8S2 H3L Nitroso-R acid CAS 525-05-3 (1811)

1-Nitroso-2-hydroxynaphthalene-3,6-disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 0.10M U 1964MSa (69024) 486

K(?)=8.9

Pd++ sp oth/un 25°C ? U 1963BGB (69025) 487

K(?)=8.8

C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)

2,2'-Bipyridine; (C5H4N)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	vlt	NaClO4	25°C	0.10M	U	M		1994BGa (69635)	488
							Beff(Pd(bpy)A)=30.3 Beff(Pd(bpy)HB)=30.7 Beff(Pd(bpy)HC)=30.8 Beff(Pd(bpy)D)=23.2		
H2A=N-tosylglycine, H2B=N-phenylsulfonylglycine, H2C=tosyl-alpha-alanine, H2D=benzoylglycine. Data for other L.									

C10H8N2O4		HL		2-Furil dioxime	CAS 522-27-0		(3319)		
1,2-Di(2'-furyl)ethane-1,2-dione dioxime; (C4H3O.C(:N.OH))2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	dis	NaClO4	20°C	0.10M	U		B2=43.7	1967STa (69702)	489

C10H9NO4S		H2L			CAS 116-63-2		(4781)		
1-Amino-2-naphthol-4-sulfonic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	?	U		B2=7.90	1968MBa (70187)	490

C10H9N3O3		HL			(1933)				
4-(5'-Methyl-3'-isoxazolylazo)-1,3-dihydroxybenzene; (H0)2C6H3.N:N.C3H2N0									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	25°C	0.10M	U			1989TSb (70412)	491
							K(PdHL+H)=2.65 K(PdL+H)=7.09 K(PdH-1L+H)=9.57		

C10H11N3S		L			CAS 5351-70-2		(4734)		
Cinnamaldehyde thiosemicarbazone; C6H5.CH:CH.CH:N.NH.CS.NH2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	20°C	50%	U		B2=11.74	1972KLa (71085)	492
Medium: 50% EtOH, 0.1 M, pH=5									

C10H12N4O5		HL		Inosine	CAS 58-63-9		(2344)		
Hypoxanthine-9-beta-D-ribofuranoside;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.10M	M	M		2002BSa (71392)	493
							K(PdA+L)=6.94 K(PdA+H+L)=11.00		

$$K(\text{PdA}+2\text{L})=10.27$$

HA is S-methyl cysteine.

Pd++ gl NaNO3 25°C 0.10M C M 2002MSb (71393) 494

$$K(\text{PdA}+\text{L})=8.03$$

$$K(\text{PdA}+2\text{L})=12.74$$

$$K(\text{PdA}+\text{B}+\text{L})=12.29$$

$$K(\text{PdA}+\text{H}+\text{B}+\text{L})=17.72$$

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutanedicarboxylic acid.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (71394) 495

$$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L})=9.73$$

$$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{H}+\text{L})=12.55$$

$$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L})=14.89$$

$$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+\text{H})=20.11$$

$$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+2\text{H})=25.37.$$

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (71395) 496

$$K(\text{Pd}(\text{pn})+\text{L})=6.83$$

$$K(\text{Pd}(\text{pn})+2\text{L})=11.26$$

pn is 1,2-diaminopropane. For nucleotide protonation, K1=8.55.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (71396) 497

$$K(\text{PdA}+\text{L})=6.92$$

$$K(\text{PdA}+2\text{L})=11.58$$

A is 1,3-diaminopropane.

Pd++ gl KCl 25°C 0.20M U M 1997KFa (71397) 498

$$K(\text{Pd}(\text{dien})\text{Cl}+\text{L})=6.82$$

$$K(\text{Pd}(\text{dien})\text{Cl}+\text{H}+\text{L})=12.79$$

$$K(2\text{Pd}(\text{dien})\text{Cl}+\text{L})=11.56$$

$$K(\text{Pd}(\text{terpy})\text{Cl}+\text{L})=6.92$$

dien=diethylenetriamine, terpy=2,2'-6',2''-terpyridine. $K(\text{Pd}(\text{terpy})+\text{H}+\text{L})=12.10$

Data also for many related nucleobases

Pd++ gl NaClO4 25°C 0.10M C M 1997RSa (71398) 499

$$K(\text{PdA}+\text{L})=7.43$$

$$K(\text{PdA}+2\text{L})=11.77$$

A=2-(Aminomethyl)pyridine

Pd++ gl KNO3 25°C 0.20M C 1997WKa (71399) 500

$$K(\text{PdACl}+\text{L}=\text{PdAL}+\text{Cl})=6.38$$

$$K(\text{PdACl}+\text{H}+\text{L}=\text{PdAHL}+\text{Cl})=12.73$$

$$K(2\text{PdACl}+\text{L}=\text{Pd}_2\text{A}_2\text{L}+2\text{Cl})=10.93$$

PdA is [PdH-1(gly-Met)].

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (71400) 501

$$K(\text{PdACl}_2+\text{L})=6.04$$

$$K(\text{PdACl}_2+2\text{L})=9.56$$

A is N,N,N',N'-tetramethyl-1,2-diaminoethane. Also data at 15.5, 20, 30

Pd++ gl NaClO₄ 25°C 0.10M M T H 1996SEc (71401) 502
K(PdACl₂+L)=5.78
K(PdACl₂+2L)=10.48
A is N,N,N',N'-tetraethyl-1,2-diaminoethane. Also data at 15.5, 20, 30
and 35.2 C. DH(PdACl₂+L)=30.6 kJ mol⁻¹, DH(PdACl₂+2L)=35.6.

C10H12N6S L CAS 91262-80-9 (6101)
3-(4',5'-Dimethyl-2'-thiazolylazo)-2,6-diaminopyridine;

C10H13NO4S H2L N-Tosylalanine (1584)
N-(4-Toluenesulfonyl)-3-aminopropanoic acid; CH3.C6H4.SO2.NH.CH2.CH2.COOH

Complex formation involves loss of the amide proton.

C10H13N3OS L (4791)
alpha-Ethylfurylacrolein thiosemicarbazone;

C10H13N4O8P H3L IMP CAS 131-99-7 (843)
Inosine-5'-monophosphoric acid;

HA is S-methyl cysteine.

Pd++ gl NaNO3 25°C 0.10M C M 2001SHc (71864) 508

$K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L})=10.17$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{H}+\text{L})=16.65$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+\text{L}+2\text{H})=20.98$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L})=14.80$
 $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+\text{H})=21.49$, $K(\text{Pd}(\text{bpy})(\text{H}_2\text{O})_2+2\text{L}+2\text{H})=28.50$.

Pd++ gl NaNO3 25°C 0.10M U M 1999SSd (71865) 509
 $K(\text{Pd}(\text{pn})+\text{L})=8.13$
 $K(\text{Pd}(\text{pn})+2\text{L})=11.92$
 $K(\text{Pd}(\text{pn})+\text{H}+\text{L})=14.03$
 pn is 1,2-diaminopropane. For nucleotide protonation, $K_1=8.67$, $B_2=14.63$.

Pd++ gl NaNO3 37°C 0.16M M M 1998ESa (71866) 510
 $K(\text{PdA}+\text{L})=9.82$
 $K(\text{PdA}+2\text{L})=14.82$
 $K(\text{PdA}+\text{H}+\text{L})=15.14$
 A is 1,3-diaminopropane.

Pd++ gl NaClO4 25°C 0.10M C M 1997RSa (71867) 511
 $K(\text{PdA}+\text{L})=10.79$
 $K(\text{PdA}+\text{H}+\text{L})=17.02$
 $K(\text{PdA}+2\text{L})=14.65$
 A=2-(Aminomethyl)pyridine

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (71868) 512
 $K(\text{PdACl}_2+\text{L})=4.43$
 $K(\text{PdACl}_2+2\text{L})=9.20$
 A is N,N,N',N'-tetramethyl-1,2-diaminoethane. Also data at 15, 20, 30 and 35 C. $\text{DH}(\text{PdACl}_2+\text{L})=-73.1 \text{ kJ mol}^{-1}$, $\text{DH}(\text{PdACl}_2+2\text{L})=-62.4$.

Pd++ gl NaClO4 25°C 0.10M M T H 1996SEc (71869) 513
 $K(\text{PdACl}_2+\text{L})=4.39$
 $K(\text{PdACl}_2+2\text{L})=9.73$
 A is N,N,N',N'-tetraethyl-1,2-diaminoethane. Also data at 15, 20, 30 and 35 C. $\text{DH}(\text{PdACl}_2+\text{L})=-88.6 \text{ kJ mol}^{-1}$, $\text{DH}(\text{PdACl}_2+2\text{L})=-2.09$.

Pd++ sp NaClO4 25°C 0.10M U M 1994SEa (71870) 514
 $K(\text{PdACl}+\text{L}=\text{PdALCl})=-1.46$

A=N,N,N',N'-Tetramethylethylenediamine

 C10H13N5O5 HL Guanosine CAS 118-00-3 (1402)
 2-Aminopurin-6-one-9-riboside;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaNO3	37°C	0.16M	M	M		$K(\text{PdA}+\text{L})=7.85$	1998ESa (72015)	515

A is 1,3-diaminopropane.

 C10H14N2O5 H2L Thymidine CAS 50-89-5 (8256)

Thymine deoxyriboside, 1-(2-Deoxy-beta-ribofuranosyl)-5-methyluracil;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	25°C	0.10M	C	M			2002MSb (72085)	516
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K(PdA+L)=8.75
K(PdA+2L)=14.53
K(PdA+B+L)=16.26

A is N,N'-dimethylethylenediamine, B is 1,1-cyclobutanedicarboxylic acid.

Pd++	gl	NaNO3	25°C	0.10M	U	M			1999SSd (72086)	517
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K(Pd(pn)+L)=8.92
K(Pd(pn)+2L)=14.84

pn is 1,2-diaminopropane. For nucleotide protonation, K1=9.54.

Pd++	gl	NaNO3	37°C	0.16M	M	M			1998ESa (72087)	518
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K(PdA+L)=8.27
K(PdA+2L)=13.57

A is 1,3-diaminopropane.

C10H14N2O6 L alpha-Thymidine CAS 4449-43-8 (695)

Thymine-2-desoxyribofuranosyl-5-methyluracil;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	NaNO3	20°C	1.0M	M			K1=8.15 B2=15.82	1997WYa (72106)	519
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K3=6.37
K4=3.56

Pd++	gl	KNO3	25°C	0.50M	U	M			1981LIa (72107)	520
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K(Pd(en)(H2O)2+L)=8.84
K(Pd(en)(H2O)L+L)=5.85
K(Pd(dien)(H2O)+L)=8.31

C10H14N3 L CAS 29198-32-1 (6921)

4-Diazo-N,N-diethylaniline; N:N.C6H4.N(C2H5)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	alc/w	25°C	100%	U	M			1994PAa (72122)	521
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K(Pd3A3CO+L)=2.30

Medium: MeOH. A=Bis(diphenylphosphino)methane

C10H14N5O7P H2L AMP-2 CAS 81012-86-4 (2437)

Adenosine-2'-monophosphoric acid, 2-Adenylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KNO3	25°C	0.10M	C	M			2002WBa (72190)	522
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K(Pd(dien)+H+L)=11.1

K(Pd(dien)+L)=5.30
K(2Pd(dien)+H+L)=13.5

C10H14N5O8P H3L GMP-2 CAS 130-50-7 (8778)
Guanosine-2'-monophosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	C	M		2002WBa (72525)	523
							K(Pd(dien)+2H+L)=20.82 K(Pd(dien)+H+L)=15.40 K(Pd(dien)+L)=7.50 K(2Pd(dien)+H+L)=19.45		
K(2Pd(dien)+L)=13.70, K(3Pd(dien)+L)=16.59.									

Pd++	gl	KNO3	25°C	0.10M	C	M		2002WBa (72526)	524
							K(Pd(en)+4H+2L)=39.57 K(Pd(en)+2H+2L)=28.4 K(Pd(en)+2L)=13.52 K(Pd(en)+H+L)=15.98		

K(Pd(en)+L)=9.54

C10H14N5O8P H3L GMP-5 CAS 85-32-5 (2947)
Guanosine-5'-monophosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.10M	M	M		2002BSa (72593)	525
							K(PdA+L)=11.96 K(PdA+H+L)=18.75 K(PdA+2H+L)=22.00		

HA is S-methyl cysteine.

Pd++	gl	KNO3	25°C	0.10M	C	M		2002WBa (72594)	526
							K(Pd(en)+4H+2L)=40.96 K(Pd(en)+2H+2L)=28.8 K(Pd(en)+2L)=11.7 K(Pd(en)+H+L)=16.37		

K(Pd(en)+L)=9.83.

Pd++	gl	KNO3	25°C	0.10M	C	M	K1=3.60	2001AAa (72595)	527
Also data for ternary complexes with MOPSO, TAPSO and ACES.									

Pd++	gl	NaClO4	25°C	0.10M	C	M		1997RSa (72596)	528
							K(PdA+L)=10.82 K(PdA+H+L)=17.35 K(PdA+2L)=14.46		

A=2-(Aminomethyl)pyridine

Pd++	gl	NaClO4	25°C	0.10M	M	T H		1996SEc (72597)	529
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$$K(\text{PdACl}_2 + 2\text{HL}) = 8.03$$

Pd++ g1 NaCl04 25°C 0.10M M T H 1996SEc (72598) 530

$$K(\text{PdACl}_2 + \text{HL}) = 4.00$$
$$K(\text{PdACl}_2 + 2\text{HL}) = 7.14$$

C10H15N5O4	HL	Gly-Gly-His	CAS 93404-95-6	(74)
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Glycyl-glycyl-histidine; $\text{H}_2\text{N} \cdot \text{CH}_2 \cdot \text{CO} \cdot \text{NH} \cdot \text{CH}_2 \cdot \text{CO} \cdot \text{NH} \cdot \text{CH}(\text{CH}_2 \cdot \text{C}_3\text{H}_3\text{N}_2) \cdot \text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl KCl 25°C 0.20M C 1997BCb (72800) 531

$$*K(\text{PdH-4L}) = -11.30$$

*K corresponds to deprotonation of coordinated -NH₂.

C10H15N5O10P2 H3L ADP CAS 20398-34-9 (2181)

Adenosine-5'-diphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ nmr oth/un 23°C 0.30M U M 1985PGa (73011) 532

$$K_{eff}(PdA+L)=4.08$$

A=Tetrakis(4-N-methylpyridyl)porphyrin. pD=7.0

C10H16N2O8 H4L EDDS CAS 52759-67-8 (1100)

1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; $(\text{CH}_2.\text{NH}.\text{CH}(\text{COOH})\text{CH}_2.\text{COOH})_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp NaClO4 20°C 0.10M U M 1986PKa (73171) 533

$$K(\text{PdCl}+\text{H}_2\text{L})=10.93$$
$$K(\text{PdCl}+\text{L})=23.67$$

Pd++ g1 KNO3 30°C 0.10M U K1=13.6 1971STc (73172) 534

C10H16N2O8 H4L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp NaCl04 21°C 0.20M U M K1=25.6 1983KDa (74071) 535

$$K(PdL+Cl)=5.4$$

Pd++ sp NaCl04 25°C 1.00M U M 1981SDa (74072) 536

$$K(\text{PdL}+\text{Cl})=2.26$$

$$K(\text{PdL}+\text{OH})=4.41 \quad K(\text{PdL}+\text{NH}_3)=4.84 \quad K(\text{PdL}+\text{S}_2\text{O}_3)=4.66 \quad K(\text{PdL}+\text{thiocarbamate})=4.00$$

Pd++	sp	none	25°C	0.0	U	K1=26.4	1978KRa (74073)	537
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Pd++ gl oth/un 20°C 1.00M U I M K1=24.5 1976AMa (74074) 538
K(PdL+H)=3.01
K(PdHL+H)=3.21
K(PdH2L+H)=0.09

Medium: NaBr/NaClO₄. By exchange with PdBr₄

Pd++ oth NaClO4 25°C 0.20M U K1=18.5 1955MKa (74075) 539

C10H16N6S	L	Cimetidine	CAS 51481-61-9	(5716)
Cimetidine; CH3.C3H2N2.CH2.S.CH2.CH2.NH.C(:NCN)NH.CH3				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Pd++ g1 NaCl 25°C 0.10M U K1=7.63 B2=15.13 1995CCa (74912) 540
B(PdH-1L)=0.52
B(PdH-2L)=-10.95
B(PdH-1L2)=7.87
B(PdH-2L2)=-1.18

C10H26N4 L CAS 66475-54-5 (5756)
3,10-Diazadodecane-1,12-diamine; NH2.CH2.CH2.NH.(CH2)6.NH.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
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Pd++ gl KN03 25°C 0.10M C I M K1=38.14 1985YAa (76766) 541

C10H28N2O12P4 H8L CAS 23605-74-5 (435)
(Hexamethylenedinitrilo)tetra(methylenephosphonic acid);
(CH2.CH2.CH2.N(CH2.PO3H2)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	gl	KN03	25°C 0.10M U	K1=10.83	1980ZrB (76841) 542
				K(PdL+H)=9.56	
				K(PdHL+H)=6.71	
				K(PdH2L+H)=5.73	
				K(PdH3L+H)=4.65	

C11H7NO4 H2L CAS 122844-38-6 (8293)
1-Hydroxy-4-nitroso-2-naphthalenecarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ gl alc/w RT 40% M K1=4.95 B2= 8.72 1993Rab (76893) 543
Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

C11H7NO4 H2L CAS 32446-26-7 (8294)

3-Hydroxy-4-nitroso-2-naphthalenecarboxylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl alc/w RT 40% M K1=3.89 B2= 7.86 1993Rab (76901) 544

Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

C11H8O3 H2L CAS 86-48-6 (1129)

1-Hydroxy-2-naphthoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl alc/w RT 40% M K1=6.88 B2=13.31 1993Rab (77015) 545

Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

C11H8O3 H2L CAS 92-70-6 (1130)

2-Hydroxy-3-naphthoic acid (3-Hydroxy-2-naphthoic acid);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl alc/w RT 40% M K1=11.84 B2=17.04 1993Rab (77128) 546

Medium: 40% v/v EtOH/H2O, 0.1 M NaClO4.

C11H9NO2S HL CAS 29556-13-6 (1450)

N-Phenyl-2-thenoylhydroxamic acid; C4H3SCON(C6H5)OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 50% M T H K1=9.22 B2=16.97 1977ABb (77351) 547

50% v/v dioxan - water; Data also for complexes with Cu(II), Zn, Ni, Co, Mn

C11H9NO3 H2L CAS 80690-05-7 (872)

3-Hydroxy-2-methyl-1,4-naphthoquinone monoxime;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 30°C 0.10M U K1=5.63 B2=10.22 1981KSa (77365) 548

K3=4.61

C11H9N3O HL CAS 10335-29-2 (3937)

2-(2'-Pyridylazo)phenol; C5H4N.N:N.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 20°C 50% U K1=17.1 1967ANa (77461) 549

Medium: 50% MeOH, 0.1 M NaClO4

 C11H10N4 L PAPHY CAS 2215-33-0 (1305)
 Pyridine-2-aldehyde-2'-pyridyl-hydrazone; C5H4N.CH:N.NH.C5H4N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaCl	25°C	3.00M	U	M			1981MIb (77710)	550
								K(PdCl2+HL=PdHLC1+Cl)=4.00		
								K(PdLC1+H)=5.30		
								K(PdCl2+2PdLC1=Pd3L2C14)=8.30		

 C11H12N2O L Antipyrine CAS 60-80-0 (2026)
 2,3-Dimethyl-1-phenyl-3-pyrazolin-5-one, Phenazone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	?	0.60M	U			K1=5.58	1971KBe (78005)	551
									Medium: K2S04	

 C11H12N2O2 HL Tryptophan CAS 73-22-3 (3)
 2-Amino-3-(3-indolyl)propanoic acid; H2N.CH(CH2.C8H6N)COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.50M	U				1978LIa (78230)	552
								K(Pd(en)+L)=10.83		

 C11H14N2O3 HL Gly-Phe CAS 3321-03-7 (829)
 Glycyl-phenylalanine; H2N.CH2.CO.NH.CH(CH2.C6H5).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	C				1999AJa (78815)	553
								B(PdLC1)=17.94		
								B(PdH-1LC1)=16.09		
								B(PdH-2L)=5.30		
								B(PdH-1L2)=20.10		

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-1L)=14.10.

 C11H14N2O3 HL Phe-Gly CAS 721-90-4 (830)
 Phenylalanyl-glycine; H2N.CH(CH2.C6H5).CO.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.20M	C				1999AJa (78828)	554
								B(PdLC1)=17.58		
								B(PdH-1LC1)=15.20		
								B(PdH-2L)=4.50		
								B(PdH-1L2)=19.20		

Medium: 0.1 M KNO3, 0.1 M KCl. B(PdH-1L)=13.2; B(PdH-2L2)=13.70.

 C11H18N2O8 H4L CAS 4408-81-5 (923)
 1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOCH₂)₂NCH₂)₂CH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	gl	oth/un	20°C	1.00M	C		K1=28.8	1976AMa (79465)	555
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Medium: NaBr/NaClO₄. By exchange with PdBr₄

 C11H20N2 L (6343)
 3,5-Dipropyl-4-ethylpyrazole

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	nmr	non-aq	32°C	100%	U			1987FKa (79696)	556
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K(PdCl₂+2L)=6.7

Medium: Deuterated DMSO (D₆). With N-methyl analogue: K(PdCl₂+2L)=9.6;
 N-benzyl: K=7.9; N-allyl: K=10.8

 C12H8N2 L Phenanthroline CAS 66-71-7 (144)
 1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	sp	none	25°C	0.0	C		K2=9.4	1975PJb (80508)	557
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*K(PdL)=-3.3
 *K(Pd(OH)L)=-5.9
 *K(Pd(OH)2L)=-9.6

 C12H10N2O5S H3L Tropeolin 0 CAS 547-57-9 (1090)
 Chrysoin; HS03.C6H4.N:N.C6H3(OH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	sp	oth/un	25°C	?	U			1963SDd (80738)	558
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K(Pd+2HL)=9.4(?)

 C12H10N3OBr HL CAS 5756-88-7 (4001)
 1-(4'-Bromophenyl)-3-hydroxy-3-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++	gl	diox/w	25°C	70%	U		K1=10.86 B2=21.30	1965PSd (80754)	559
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Medium: 70% dioxan, 0.1 M KCl

 C12H10N3OCl HL CAS 52756-05-6 (3998)
 1-(2'-Chlorophenyl)-3-hydroxy-3-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++ gl diox/w 25°C 70% U K1=10.43 B2=20.43 1964PSg (80761) 560
Medium: 70% dioxan, 0.1 M KCl

C12H10N3OCl HL CAS 5756-86-5 (3999)

1-(4'-Chlorophenyl)-3-hydroxy-3-phenyltriazene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=10.70 B2=20.95 1964PSb (80767) 561

Medium: 70% dioxan, 0.1 M KCl

C12H11N3O4S H2L (4003)

3-Hydroxy-3-phenyl-1-(4'-sulfonyl)triazene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=9.71 B2=19.03 1964PSf (80942) 562

Medium: 70% dioxan, 0.1 M KCl

Pd++ sp oth/un 25°C ? U 1958DSa (80943) 563

K(?)=11.52

Acetate buffer

C12H12N2S2 HL CAS 1141-88-4 (7739)

2,2'-Dithiodianiline, 2,2'-Diaminodiphenyl disulfide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp none C K1=6.17 2000GNa (81110) 564

C12H13N3 L CAS 1539-42-0 (932)

bis-((2-Pyridyl)methyl)-amine (Di-2-picolyamine); C5H4N.CH2NHCH2.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 25°C 0.20M C 2003Nfa (81289) 565

*K(PdL(H2O))=-7.08

C12H20N2O8 H4L CAS 40623-42-5 (1101)

1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH2NHCH(COOH)CH2CH2COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl KNO3 30°C 0.10M U K1=13.4 1971STc (82091) 566

C12H20N2O8 H4L CAS 2458-58-4 (922)

1,4-Diaminobutane-N,N,N',N'-tetraethanoic acid; (HOOC.CH2)2N.(CH2)4.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl oth/un 20°C 1.00M C K1=25.8 1976AMa (82233) 567
 Medium: NaBr/NaClO4. By exchange with PdBr4

 C12H24O2S4 L (6657)
 1,4,7,10-Tetrathia-13,16-dioxacyclooctadecane, 1,4,7,10-Tetrathia-18-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	ix	none	25°C	0.0	U		K1=32.3	1991BTa (83119)	568

 C12H24O4S2 L CAS 296-39-9 (4938)
 1,4,10,13-Tetraoxa-7,16-dithiacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	cal	oth/un	25°C	0.50M	M	H	K1=21.1	1990IWa (83141)	569

 Medium: 0.5M HNO3. DH(K1)=-82.4 kJ mol-1, DS(K1)= 127.6 J K-1 mol-1.

 C12H24O4S2 L (6528)
 7,10,13,16-Tetraoxa-1,4-dithiacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	cal	oth/un	25°C	0.50M	M	H	K1=25.1	1990IWa (83151)	570

 Medium: 0.5M HNO3. DH(K1)=-184.1 kJ mol-1, DS(K1)= -137 J K-1 mol-1.

 C12H29N3 L CAS 123-12-6 (4904)
 (N,N,N",N"-Tetraethyl-diethylenetriamine; (C2H5)2N.CH2.CH2.NH.CH2.CH2.N(C2H5)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.50M	C	I		1981GMf (84249)	571

 K(PdL=PdLOH+H)=-7.688
 K(PdL+PdLOH)=0.90
 In 0.5 NaNO3, K(PdL=PdLOH+H)=-7.677, K(PdL+PdLOH)=0.48

 C12H30N4 L (7251)
 2,5,8,11-Tetramethyl-2,5,8,11-tetraazadodecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C	M	K1=23.38	1998BBa (84292)	572

 K(PdL+H+Cl)=6.85
 B(PdH-1L)=13.9
 K(PdL+OH)=4.3
 Medium: 0.1 M NMe4Cl

 C12H30N4 L (6740)
 Tris(2-(dimethylamino)ethyl)amine; N(CH2CH2.N(CH3)2)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl	25°C	1.00M	U	I	K1=30.5	1993AMa (84303)	573
Pd++	gl	oth/un	25°C	1.00M	U	M	K(Pd(H2O)L+Cl=PdClL)=2.6 K(Pd(H2O)L+Br=PdBrL)=2.8 K(Pd(H2O)L+SCN=Pd(SCN)L)=5.57	1993AMa (84304)	574

C12H30N6 L CAS 296-35-5 (143)
 1,4,7,10,13,16-Hexaazacyclooctadecane; cyclo(-(NH.CH2.CH2)6-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	cal	NaCl	25°C	0.50M	U	HM	DH(2PdCl4+L=Pd2LC12+6Cl)=-110.8 kJ mol-1	1993BBa (84349)	575
Pd++	gl	NaCl	25°C	0.50M	C	H	K1=29.2 B(PdHL)=37.47 B(PdH2L)=42.40 B(Pd2LC12)=51.8	1992BBf (84350)	576

By calorimetry: DH(PdCl4+H6L)=-6.3 kJ mol-1.

C13H8N2O6Cl2S H3L CAS 60743-06-8 (8478)
 2-[(3,5-Dichloro-2-hydroxyphenyl)azo]-5-sulfobenzoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaClO4	RT	0.10M	C		K1=15.51	1978GSc (84477)	577

								C13H9NO2BrCl HL CAS 104614-71-3 (9109)	
								4-Bromo-N-(3-chlorophenyl)-N-hydroxybenzamide;	

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	C	M	K1=11.13 B(Pd(gly)L)=20.61	2001AMc (84578)	578

Medium: 50% v/v dioxane/H2O

C13H9NO2ClF HL CAS 104614-72-4 (9107)
 N-(3-Chlorophenyl)-4-fluoro-N-hydroxybenzamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	C	M	K1=11.34 B(Pd(gly)L)=20.98	2001AMc (84586)	579

Medium: 50% v/v dioxane/H2O

C13H9NO2Cl2 HL CAS 67201-86-9 (9108)
 4-Chloro-N-(3-chlorophenyl)-N-hydroxybenzamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	C	M	K1=11.15 B(Pd(gly)L)=20.58	2001AMc (84594)	580

Medium: 50% v/v dioxane/H2O

 C13H9N3O4S2 H2L CAS 2536-61-0 (4031)
 1-(1',3'-Thiazol-2'-ylazo)-2-hydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	alc/w	25°C	50%	U	I	K1=13 B2=19.4	1967NPb (84644)	581

Medium: 50% MeOH, 0.1 M NaClO4. In 0% MeOH: K1=13, K2=5.7

 C13H10N2O2Cl HL CAS 36016-24-7 (1818)
 N-(4-Chlorophenyl)benzohydroxamic acid; C6H5.CO.N(C6H4Cl)OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U		K1=9.80 B2=18.21	1967JSa (84719)	582

Medium: 70% dioxan, 0.1 M KCl

 C13H10N2O2Cl HL CAS 78154-49-1 (5649)
 N-3-Chlorophenylbenzohydroxamic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	C	M	K1=11.47 B(Pd(gly)L)=21.29	2001AMc (84742)	583

Medium: 50% v/v dioxane/H2O

 C13H10N2O2 HL CAS 56288-80-1 (4980)
 2-Hydroxy-4-(phenylazo)benzaldehyde; C6H5.N:N.C6H3(OH).CHO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	30°C	50%	U		B2=7.64	1972DTb (84840)	584

 C13H10N2O5S H2L CAS 98789-35-6 (5012)
 4-Hydroxy-3-formylazobenzene-4'-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	30°C	aq	U		B2=7.22	1972DTb (84923)	585

 C13H10N4O4S H2L (6644)
 4-Hydroxy-3-(1H-imidazol-2-ylazo)-2-naphtalenesuphonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
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Pd++ sp NaClO4 25°C 0.50M U K1=15.53 1992VMa (84960) 586
For -3-ylazo analogue: K1=10.22; for 3,3-bis(1H-pyrazol-3-ylazo) analogue:
K1=10.99

C13H10N4S HL CAS 3788-81-6 (4014)
2-Picolinylaldehyde 2-benzothiazolylhydrazone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 50% U K1=10.33 1965HRa (84967) 587

C13H10O2S H2L CAS 88220-26-2 (6572)
3-(1-Naphthyl)-2-mercaptopropenoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 0.10M C K1=15.56 B2=26.40 1989IBb (84976) 588
Medium: Aqueous 0.1 M NaClO4 containing 1-2% EtOH.

C13H11NO2 HL CAS 304-88-1 (181)
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ EMF diox/w 25°C 70% U K1=10.11 B2=18.85 1967JSb (85171) 589
Medium: 70% dioxan, 0.1 M KCl

C13H11N3O6S H2L (2811)
1-(2-Carboxy-5-sulfonatophenyl)-3-hydroxy-phenyltriazene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp none 25°C 0.0 U K1=10.398 1974CHa (85304) 590

C13H12N2S HL CAS 156873-11-9 (8362)
2-[[1-(2-Pyridinyl)ethylidene]amino]benzene thiol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ dis NaCl 25°C 2.0M C 1998BMd (85391) 591
K(Pd+2HL(org)=PdL2(org)+2H)=2.9. Method: extraction into CHCl3.

C13H12N4S L Dithizone CAS 60-10-6 (1801)
Diphenylthiocarbazone; C6H5.NH.NH.CS.N:N.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 25°C 0.10M U K1=11.39 B2=21.78 1973BSe (85471) 592

C13H13N3O HL (4018)

3-Hydroxy-1-(2'-methylphenyl)-3-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KCl	25°C	0.10M	U		K1=11.70 B2=22.97	1964PSa (85508)	593

C13H13N3O HL CAS 5756-83-2 (4019)

3-Hydroxy-1-(4'-methylphenyl)-3-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KCl	25°C	0.10M	U		K1=11.89 B2=23.35	1964PSa (85514)	594

C13H13N3O2 HL CAS 5756-89-8 (4021)

3-Hydroxy-1-(4'-methoxyphenyl)-3-phenyltriazene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	70%	U		K1=12.06 B2=23.74	1965PSb (85522)	595

Medium: 70% dioxan, 0.1 M KCl

C13H20N2O2 L Procaine CAS 59-46-1 (4029)

2-(Diethylamino)ethyl 4-aminobenzoate; H2N.C6H4.CO2.CH2.CH2.N(C2H5)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	?	U		B2=7.88	1968SPd (86097)	596

C13H22N2O8 H4L CAS 1798-14-7 (921)

(Pentamethylenedinitrilo)tetraethanoic acid; ((HOOC.CH2)2N.CH2.CH2)2CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	oth/un	20°C	1.00M	C		K1=26.4	1976AMa (86203)	597

Medium: NaBr/NaClO4. Corrected for PdBrx complexes

C13H22N4O3S L Ranitidine CAS 66357-35-5 (7144)

N(2-(5-Dimethylaminomethyl)-2-furanylmethyl)thioethyl-N-methyl-2-nitro-1-ethenediamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl	25°C	0.10M	U		K1=9.97	1995CCa (86332)	598

B(PdH-1L)=2.41

B(PdH-2L)=-6.88

C13H26O4S2 L (6656)

1,5-Dithia-8,11,14,17-tetraoxacyclononadecane, 1,5-Dithia-19-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	ix	none	25°C	0.0	U		K1=29.8	1991BTa (86462)	599

C13H32N4			L				(7403)		
2,5,9,12-Tetramethyl-2,5,9,12-tetraazatridecane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C		K1=28.3 K(PdL+H+Cl)=3.6 B(PdH-1L)=16.9 K(PdL+OH)=2.4	1998BBa (86579)	600

Medium: 0.1 M Me4NCl

C14H9NO3		HL					CAS 116-85-8	(1020)	
1-Amino-4-hydroxyanthraquinone;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	20°C	50%	U		K1=14.21 B2=23.66 K(Pd+HL)=5.71 K(Pd(OH)2L)=31.81	1990ISa (86796)	601

Medium: 50% EtOH/H2O, 0.1 M NaClO4

C14H10O4		H2L					CAS 482-05-3	(8247)	
Diphenyl-2,2'-dicarboxylic acid; diphenic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	30°C	50%	U T H		K1=7.00 B2=13.24	1978SJC (86933)	602
Medium: 50% dioxane/H2O, 0.10 M NaClO4. At 40 C, K1=6.52, K2=5.52.									
DH and DS values reported.									

C14H12Cl2S2		L					CAS 33451-44-4	(5055)	
1,2-Bis(4-chlorophenylthio)ethane; Cl.C6H4.S.CH2.CH2.S.C6H4.Cl									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M	K(PdI4+L=PdLI2+2I)=-0.68	1969CCb (87034)	603

C14H12NO2Cl		HL					CAS 67055-92-9	(6301)	
N-(3-Chlorophenyl)-4-methylbenzohydroxamic acid; CH3.C6H4.CO.N(C6H4Cl)OH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	diox/w	25°C	50%	C	M	K1=11.65 B(Pd(gly)L)=21.66	2001AMc (87066)	604

Medium: 50% v/v dioxane/H2O

Pd++ gl diox/w 25°C 50% U K1=9.94 B2=18.79 1989PMb (87067) 605

Pd++ gl diox/w 25°C 50% U K1=10.05 B2=19.15 1989PMb (87068) 606

Data also for 4-fluoro, 4-chloro, 4-bromo, 4-nitro and 4-methoxy analogues

C14H12N03Cl HL CAS 67135-47-1 (9106)

N-(3-Chlorophenyl)-N-hydroxy-4-methoxybenzamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 50% C M K1=11.82 2001AMc (87097) 607

B(Pd(gly)L)=22.06

Medium: 50% v/v dioxane/H2O

C14H12N4O2Br2 HL CAS 72833-87-5 (2533)

2-(2-(3,5-Dibromopyridyl)azo)-5-dimethylaminobenzoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp diox/w 25°C 40% C K1=10.36 1986KHa (87319) 608

C14H13N02 HL CAS 1503-92-0 (1817)

N-(4-Tolyl)benzohydroxamic acid; C6H5.CO.N(C6H4.CH3).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=10.34 B2=19.19 1969JSa (87450) 609

C14H13N02 HL CAS 1143-74-2 (4044)

N-2-Tolylbenzohydroxamic acid; C6H5.CO.N(C6H4.CH3).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ oth diox/w 25°C 70% U K1=19.02 1968JSc (87482) 610

C14H13N3O2 HL (4045)

1-(4'-Acetylphenyl)-3-hydroxy-3-phenyltriazene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 70% U K1=10.97 B2=21.51 1964PSe (87594) 611

Medium: 70% dioxan, 0.1 M KCl

C14H13N5OS HL (5394)

1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)thiourea;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp mixed 25°C 40% U 1985RGa (87617) 612

K1eff=5.57

Medium: 40% DMF, pH 4.5

C14H13N5O2 HL (5393)
1-(2-Pyridylmethylideneamino)-3-(salicylideneamino)urea;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	mixed	25°C	32%	U			1985RGa (87624)	613

K1eff=5.38

Medium: 32% DMF, pH 4.5

C14H14N4OBr2 HL CAS 35601-32-2 (5092)
5-(3,5-Dibromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	?	?	U		K1=6.94	1967GUa (87688)	614

C14H14S2 L CAS 42311-15-9 (5031)
1,2-Bis(phenylthio)ethane; C6H5.S.CH2.CH2.S.C6H5

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (87708)	615

K(PdI4+L=PdLI2+2I)=0.72

C14H15N4OBr HL CAS 14337-50-9 (5095)
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	?	?	U			1967GUa (87768)	616

K(?)=7.35

C14H16N2O2S2 L CAS 729600-10-6 (9255)
2,3,5,6,8,9-Hexahydro[1,4,7,10]dioxadithiacyclododecino[2,3-b]quinoxaline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	nmr	mixed	25°C	60%	C		K1=4.20	2004HHa (87879)	617

Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.

C14H16N2O2S2 L CAS 729600-11-7 (9256)
2,3,5,6,8,9-Hexahydro[1,4,7,10]dioxadithiacyclododecino[8,9-b]quinoxaline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	nmr	mixed	25°C	60%	C		K1=3.61	2004HHa (87880)	618

Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.

C14H16N2O4S H2L Dansyl-Gly CAS 1091-85-6 (5845)
N-Dansylglycine, (5-Dimethylamino)naphthalene-1-sulfonoglycine;
(CH3)2N.C10H6.SO2.NH.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	vlt	NaCl04	25°C	0.10M	U		K1=17.8 B2=21.8 B(PdL(OH))=21.6 Beff(PdH-2L2)=21.8 Beff(PdH-2L2(OH))=21.6	1990GBb (87901)	619

C14H18N4 L DPEN CAS 4608-34-3 (1850)
N,N'-Bis-(2-pyridylmethyl)-1,2-diaminoethane; (C5H4N.CH2.NH.CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	1.00M	C		K1=35.6	1985YAA (88117)	620

Medium: NaBr

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl04	20°C	1.00M	U	M	K1=29.7 K(PdL+H)=3.49 K(PdHL+H)=2.93 K(PdH2L+H)=2.56 K(PdH3L+H)=1.93	1976AMa (89356)	621

K(PdL+SCN=PdL(SCN))=1.45; K(PdL+Br=PdBr)=-1.41 in NaBr by exchange with PdBr4

Pd++	EMF	oth/un	25°C	0.20M	U		K1=24.60	1972KIA (89357)	622
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C14H24N2O8 H4L HMDTA CAS 1633-00-7 (920)
1,6-Diaminohexane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.CH2.CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaCl04	20°C	0.10M	U	I	K(PdCl+HL)=17.3 B(PdClL(OH))=39.72 K(PdClL+OH)=15.60 K(PdClL+2OH)=28.35	1983KVa (89596)	623

B(Pd(OH)ClL)=43.72; B(Pd2Cl2(OH)2L)=57.43. Data also at 1.0 M

Pd++	gl	oth/un	20°C	1.00M	C		K1=26.3	1976AMa (89597)	624
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Medium: NaBr/NaCl04. By exchange with PdBr4

C14H26N2O8 H2L (6658)

1,4,10,13-Tetraoxa-7,16-diaza-2,3-dicarboxycyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	U		K1=8.5 B(PdHL)=15.6	1990AFa (90224)	625

C14H28N6O4 H2L (832)
N,N,N',N'-Tetrakis(2-carbamoylethyl)diaminoethane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.10M	C		K(Pd+H2L)=11.24 K(Pd+H2L=PdHL+H)=8.35 K(Pd+H2L=PdL+2H)=4.37 K(PdH2L=PdHL+H)=-2.89	1986HPa (90505)	626

K(PdHL=PdL+H)=-3.98

C14H34N4 L (7402)
2,6,9,13-Tetramethyl-2,6,9,13-tetraazatetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C		K1=22.95 K(PdL+H+Cl)=9.30 B(PdH-1L)=13.96 K(PdL+OH)=4.82 K(PdClHL+H+Cl=PdCl2H2L)=4.52	1998BBa (90834)	627

Medium: 0.1 M Me4NCl

C14H35N7 L CAS 296-85-5 (9052)
1,4,7,10,13,16,19-Heptaazacycloheicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl	25°C	0.50M	C	H	K1=24.55 B(PdHL)=34.92 B(PdH2L)=42.63 B(PdH3L)=47.13 B(Pd2LC1)=>52	1992BBf (90857)	628

By calorimetry: DH(PdCl4+H7L)=-6.3 kJ mol-1.

C14H37N7 L CAS 298-85-5 (5606)
1,4,7,10,13,16,19-Heptaazacycloheicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	cal	NaCl	25°C	0.50M	U	HM		1993BBa (90917)	629

DH(2PdCl4+L=Pd2LC12+6Cl)=-119.2 kJ mol-1

C15H10N3OBr HL (5128)

4-(5-Bromo-2-pyridylazo)-1-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ dis NaCl ? ? U 1967GVc (90944) 630
K(Pd+HL=PdL+H)=7.05

C15H12N4 L (4056)

2-Picolinaldehyde 2'-quinolylhydrazone; C5H4N.CH:N.NH.C9H6N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 50% U K1=10.57 1965HRa (91454) 631

C15H14N3O3Cl HL CAS 113581-14-9 (9105)

N-(3-Chlorophenyl)-4-ethoxy-N-hydroxybenzamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 25°C 50% C M K1=11.91 2001AMc (91706) 632
B(Pd(gly)L)=22.34

Medium: 50% v/v dioxane/H2O

C15H16N2O2 HL CAS 7397-15-1 (6853)

Peonolphenylhydrazone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ gl diox/w 20°C 75% U T K1=13.27 B2=26.05 1991NNa (91927) 633
30 C: K1=13.08, K2=12.42; 40 C: K1=12.92, K2=12.38

C15H16N4O2Br2 HL CAS 14337-54-3 (993)

2-(3,5-Dibromo-2-pyridylazo)-5-diethylaminophenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un ? ? U 1967GVb (91942) 634
K(Pd+HL=PdL+H)=6.3

C15H16S2 L CAS 42837-97-3 (5105)

1,3-Bis(phenylthio)propane; C6H5.S.CH2.CH2.CH2.S.C6H5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp alc/w 25°C 100% U M 1969CCa (91967) 635
K(PdI4+L=PdLI2+2I)=-1.18

Medium: CH3OH.

C15H17N4OBr HL CAS 14357-53-2 (712)
2-(5-Bromo-2-pyridylazo)-5-diethylaminophenol; BrC5H3N.N:N.C6H3(OH)N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	?	?	U				1967GVb (91982)	636
									K(Pd+HL=PdL+H)=7.0	

C15H18N2O2S2 L CAS 729600-13-9 (9258)
2,3,6,7,9,10-Hexahydro-5H-[1,4,7,11]dioxadithiocyclotridecino[2,3-b]quinoxalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	nmr	mixed	25°C	60%	C			K1=3.53	2004HHa (92008)	637
Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.										

C15H18N4O HL CAS 14337-52-1 (5124)
5-Diethylamino-2-(2-pyridylazo)phenol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	?	?	U				1967GVa (92098)	638
									K(?)=6.0	

C15H20N4 L DPTN CAS 63671-70-5 (1851)
N,N'-Bis-(2-pyridylmethyl)-1,3-diaminopropane; (C5H4N.CH2.NH.CH2)2CH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	1.00M	C			K1=39.1	1985YAa (92185)	639
Medium: NaBr										

C15H25N3O10 H5L (5127)
Diethylenetriamine-N,N,N'',N''-tetraethanoic acid-N'-propanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	dis	NaCl	?	?	U				1967GVc (92380)	640
									K(Pd+HL=PdL+H)=6.57	

C15H37N5 L CAS 3803-11-2 (1798)
2,5,8,11,14-Pentamethyl-2,5,8,11,14-pentaazapentadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C	M		K1=21.41	1998BBa (92627)	641
									K(PdL+H)=8.68	
									K(PdHL+H+Cl)=5.76	
									B(PdH-1L)=10.95	
									K(PdL+OH)=3.4	

Medium: 0.1 M NMe4Cl

C16H11N2O7ClS2 H3L CAS 4768-88-1 (7743)
4-Chloro-phenylazo-R-acid, 1-(4-Chlorophenylazo)-2-naphthol-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		C			1999HAa (92772)	642
							K1eff=3.75 B2eff=8.55		

Medium: Universal Buffer, pH 6.0

C16H12N2O4S H2L CAS 13964-82-4 (3475)
1-(4-Sulfophenylazo)-2-hydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		? U			1968SDa (93003)	643
							B2eff=9.8 (pH 4)		

C16H12N2O11S3 H5L CAS 548-81-2 (5180)
2-(4'-Sulfophenylazo)chromotropic acid,
2-(4-sulfophenylazo)-1,8-dihydroxyaphthalene-3,6-diHSO3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	dil	C			1985SSg (93099)	644
							B2eff=10.0 (pH 3.5) B3eff=15.0 (pH 10.5)		

Medium: dilute buffer solution (not stated).

C16H13NO4S HL (5182)
N-4-Toluenesulfonyl-benzofur-2-yl-carboxamide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	EMF	alc/w		? 70%	U		B2=6.10	1971MSc (93160)	645

Medium: 70% MeOH

C16H13N2O10AsS2 H5L Thorin I CAS 3688-92-4 (2609)
1-((2-Arsonophenyl)azo)-2-hydroxy-3,6-naphthalylldisulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		? U			1964SDd (93205)	646
							K1eff=4.4 (pH 3)		

C16H14N4O2S HL CAS 83688-78-2 (2534)
2-(2-Benzothiazolylazo)-5-dimethylaminobenzoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	diox/w	25°C	40%	C		K1=9.37	1986KHa (93483)	647

C16H18S2		L					(5144)		
1,2-Bis(3-tolylthio)ethane; CH3.C6H4.S.CH2.CH2.S.C6H4.CH3									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (93901)	648
							K(PdI4+L=PdLI2+2I)=1.23		

Medium: MeOH

C16H18S2		L					(5145)		
1,2-Bis(4-tolylthio)ethane; CH3.C6H4.S.CH2.CH2.S.C6H4.CH3									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (93902)	649
							K(PdI4+L=PdLI2+2I)=1.96		

C16H22N4		L		DPTE			CAS 81747-99-1 (1852)		
N,N-Bis-(2-pyridyl-methyl)-1,4-diaminobutane; (C5H4N.CH2.NH.CH2.CH2)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	1.00M	C		K1=37.0	1985YAA (94183)	650
Medium: NaBr									

C16H24N6O5		L		Pro-Gly-Ala-His			(7404)		
Prolyl-glycyl-alanyl-histidine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	KN03	25°C	0.20M	U		K1=15.72	1997THa (94338)	651
							B(PdHL)=17.58		
							B(PdH-1L)=11.95		

Results confirmed by H nmr measurements.

C16H26N2O12		H4L					(6659)		
1,4,10,13-Tetraoxa-7,16-diaza-2,3,11,12-tetracarboxycyclooctadecane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	U		K1=12.1	1990AFa (94591)	652
							B(PdHL)=18.4		

C16H26N2O12		H4L					CAS 130190-52-2 (6660)		
1,4,10,13-Tetraoxa-7,16-diaza-2,3,7,16-tetracarboxycyclooctadecane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	U		K1=14.1 B(PdHL)=20.0	1990AFa (94605)	653

C16H29N3O8 H3L (6699) 1,7-Dioxa-4,10,13-triazacyclopentadecane-N,N',N''-triethanoic acid;									

Pd++	gl	KCl	25°C	0.10M	C		K1=16.58 K(PdL+H)=4.88 K(PdHL+H)=2.18 B(Pd2L)=19.82 K(Pd(OH)L+H)=10.77	1993DSa (94977)	654

C16H40N8 L CAS 297-11-0 (5588) 1,4,7,10,13,16,19,22-Octaazacyclotetracosane;									

Pd++	cal	NaCl	25°C	0.50M	U	HM		1993BBa (95661)	655
DH(2PdCl4+L=Pd2LC12+6Cl)=-118.8 kJ mol-1									

C17H16N4O2S HL CAS 202867-34-3 (7313) 2-[2-(5-Methylbenzothiazolyl)azo]-5-dimethylaminobenzoic acid;									

Pd++	sp	alc/w	RT	16%	C		K1eff=6.88	1998FZa (96110)	656
Medium: 16% EtOH/H2O.									

C17H20S2 L (5209) 1,3-Bis(3-tolylthio)propane; CH3.C6H4.S.CH2.CH2.CH2.S.C6H4.CH3									

Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (96359)	657
K(PdI4+L=PdLI2+2I)=-1.04									
Medium: CH3OH.									

C17H20S2 L (5210) 1,3-Bis(4-tolylthio)propane; CH3.C6H4.S.CH2.CH2.CH2.S.C6H4.CH3									

Pd++	sp	alc/w	25°C	100%	U	M		1969CCb (96360)	658
K(PdI4+L=PdLI2+2I)=-0.77									

C17H24N4 L CAS 49764-71-3 (5757)
N,N'-Bis((2-pyridyl)methyl)-1,5-pentanediamine; C5H4N.CH2.NH.(CH2)5.NH.CH2.C5H4N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	oth/un	25°C	1.00M	C	I M		K1=34.7	1985YAa (96436)	659
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Medium: NaBr. Ternary complex with Br-

C18H15O3PS HL CAS 16704-71-5 (3365)
3-Diphenylphosphino-benzene sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	ISE	NaClO4	25°C	1.0M	U			K1=10.2 K3=6.3 K4=4.9	1972CBa (97110)	660
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C18H15P L CAS 603-35-0 (621)
Triphenylphosphine; (C6H5)3P

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	vlt	non-aq	20°C	100%	C			K(Pd(Ph)L2+Cl)=4.28 K(Pd(Ph)L2+Br)=3.56 K(Pd(Ph)L2+I)=3.23 K(Pd(Ph)L2+acetate)=2.88	1998ACd (97144)	661
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Medium: DMF. Method: chronoamperometry.

Pd++	kin	non-aq	25°C	100%	U T HM			K(PdABL+L=PdAL2+B)=2.39	1988JHc (97145)	662
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Medium: acetonitrile. A=2,5-dioxo-3,6-dichloro-1,4-benzoquinone, B=CH3CN
Data also at 30, 35, 40, 45 C

C18H18N2O2S2 L CAS 729600-12-8 (9257)
2,3,5,6,8,9-Hexahydrobenzo[g][1,4,7,10]dioxadithiacyclododecino[2,3-b]quinoxaline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	nmr	mixed	25°C	60%	C			K1=3.82	2004HHa (97229)	663
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Method: 1H nmr. Medium: 60% CD2Cl2/CD3CN.

C18H26N4 L CAS 80284-81-7 (5758)
N,N'-Bis((2-pyridyl)methyl)-1,6-hexanediamine; C5H4N.CH2.NH.(CH2)6.NH.CH2.C5H4N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++	sp	oth/un	25°C	1.00M	C			K1=34.7	1985YAa (97679)	664
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Medium: KBr. K is only a limiting value

C18H30N4O12 H6L TTHA CAS 869-52-3 (694)
 Triethylenetetraaminehexaethanoic acid;((HOOCH2)2NCH2CH2N(CH2COOH)CH2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.5M	C			K1=18.73 K(PdL+H)=6.92 K(PdH2L+H)=2.50 K(PdHL+H)=2.90 K(PdH3L+H)=2.45	1984NAb (98082)	665

K(2Pd+L)=27.50; K(Pd2L+H)=3.20
 K(Pd2HL+H)=2.0

C18H40N4O4 L CAS 89066-60-2 (867)
 N,N',N'',N'''-Tetrakis(2-hydroxyethyl)-1,4,8,11-tetraazacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	NaClO4	25°C	0.10M	C			K1=18.32	1987HPa (98924)	666

C18H44N6 L (7252)
 2,5,8,11,14,17-Hexamethyl-2,5,8,11,14,17-hexaazaoctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C	M		B(PdHL)=30.83 K(PdHL+H)=7.16 B(PdH-1L)=10.75 B(Pd2LC12)=44.0	1998BBa (98955)	667

Medium: 0.1 M NMe4Cl. B(Pd2H-1LC1)=33.9, K(Pd2LC12+H+Cl)=5.8

C19H14O7S H4L Pyrocatechol Vi CAS 369596-29-2 (709)
 Pyrocatechol Violet,
 3-[3,4-Dihydroxyphenyl-3-hydroxy-4-oxo-2,5-cyclohexadien-1-ylidenemethyl-b.;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	U			K(Pd+H2L)=8.29 K(Pd+HL)=13.67 K(PdL+OH)=3.60 K(2Pd+HL=Pd2L+H)=15.91	1997USa (99112)	668

K(Pd2L+OH)=8.25.

C19H24N2O5 L (2547)
 10-(3-Dimethylamine-2-methyl-propyl)-2-methoxyphenothiazine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Pd++ sp KCl 25°C 1.00M U K1=4.32 1978JOa (99349) 669

C21H17N2O3P L CAS 215457-01-5 (8001)
Diphenyl-3-(4-methoxyphenylsydnonyl)phosphine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp non-aq 25°C 100% C 2001LPb (101075) 670
K(PdLC12+benzylamine)=-0.49
K(PdLC12+dibenzylamine)=-0.38
K(PdLC12+diethylamine)=-0.44
K(PdLC12+triethylamine)=-0.69

Medium: CH2Cl2. Also data for dimethylamine, 2-aminopyridine, 4-anisidine, pyridine, 4-toluidine and aniline.

C21H18N4O6S H2L CAS 86170-15-2 (8412)
2-[5-(2-Methoxy-5-sulfophenyl)-3-phenyl-1-formazano]-benzoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp NaClO4 26°C 0.10M C K1=14.90 1983UCa (101119) 671
For the ligand, K1=14.4, K2=3.6.

C21H21P L CAS 6163-58-2 (600)
Tri(2-methylphenyl)phosphine (or 4-methyl where indicated); (CH3.C6H4)3P

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp non-aq 25°C 100% U TIHM 1981MKa (101193) 672
K(PdA2+L)=3.14

Medium: benzene. HA = trifluoroacetylacetone

C21H22N4O HL CAS 56932-30-0 (5308)
1-Hydroxy-2-(2-N-methylanabasiny1-alpha-azo)naphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp oth/un ? ? U B2=10.53 1966APa (101203) 673

C22H26N3OF3S L Fluphenazine CAS 146-56-5 (2548)
10-[3]-4-(2-Hydroxyethyl)piperazine-1-yl-propyl-2-trifluoromethylphenothiazine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++ sp KCl 25°C 1.00M U K1=5.13 1978JOa (101926) 674

C23H16O9Cl2S H4L Chrome azuro1 S CAS 1667-99-8 (711)
Chromazuro1 S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Pd++	sp	KCl	25°C	0.10M	C		K1=6.36	1975ISa (102565)	675
Pd++	sp	NaNO3	25°C	0.10M	U		B(Pd2L)=9.80 B(Pd2L2)=15.27 K(Pd+HL)=4.90	1972MSd (102566)	676
Pd++	sp	oth/un	25°C		? U		K(?)=4.8	1963SDc (102567)	677

C23H18O9S		H4L				Eriochrome cyan	CAS 3564-18-9	(433)	
4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchsone-5,5'-dicarboxylic acid;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	oth/un	25°C		? U		K2eff=5.0 (pH=4.5)	1970SMd (102634)	678

C23H31N3O4		H2L					(7088)		
1,4,7-Trimethyl-1,7-bis(4-carboxybenzyl)-1,4,7-triazaheptane;									
CH3N(CH2CH2N(CH3)CH2C6H4COOH)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	gl	NaCl	25°C	0.15M	C	M	B(PdLCl)=19.10 B(PdHLC1)=23.15 K(PdLCl+H)=4.05	1995BBc (102773)	679

C24H23N9O2		HL					(5330)		
1,5-Bis(4-antipyrinyl)-3-cyanoformazan;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	NaCl04	25°C	0.10M	U		K1=28.1	1971BSf (102934)	680

C26H25NO9S		H4L				Semi-Xylenol O	(426)		
3-(N,N-Di(carboxymethyl)aminomethyl)-2-cresolsulfonephthalein;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Pd++	sp	KNO3	25°C	0.10M	C	I	B(Pd2L)=26.62	1991HKg (103948)	681

C27H29NO11		L				Adriamycin	CAS 25316-40-9	(2407)	
Doxorubicin;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

Pd++ gl oth/un 25°C 0.10M U 1986FGa (104461) 682
K(Pd+HL=0.5(PdL)2)=22.1

Medium not stated.

C30H50N6 L (7089)
1,4,7,16,19,22-Hexamethyl-1,4,7,16,19,22-hexaaza[9.9]paracyclophane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C				1999BBd (105354)	683
								B(PdH2LC1)=37.44		
								B(Pd2LC12)=42.9		
								B(Pd2HLC13)=47.3		

Medium: NMe4Cl. Additional method: 1H and 13C nmr.

C31H32N2O13S H6L Xylenol orange CAS 63721-85-5 (432)
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	sp	oth/un	25°C	?	U				19630Ta (105488)	684
								K(?)=10.3		

C32H44N10O4 L CAS 702699-42-1 (9126)
2,9-Di[4-(1,4,7,10-tetraazacyclotridecane-11,13,-dione)methyl]-1,10-phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	KNO3	25°C	0.10M	U				2004GLa (105774)	685
								B(PdH2L)=19.82		
								B(Pd2L)=15.83		
								B(Pd3H-2L)=9.93		
								B(Pd3H-3L)=-3.52		

B(Pd3H-4L)=-13.72.

C35H57N5O4 L CAS 160320-59-2 (7393)
1,4,7-Trimethyl-19,22,28,31-tetraoxa-1,4,7,12,23-pentaaza[9.25]-4-cyclophane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Pd++	gl	R4N.X	25°C	0.10M	C				1999BBd (106212)	686
								B(PdLC1)=23.7		
								B(PdHLC1)=31.1		
								B(PdH2LC1)=36.7		
								K(PdLC1+H)=7.4		

Medium: NMe4Cl. Additional method: 1H and 13C nmr. K(PdHLC1+H)=5.6.

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END