```
Software version = 5.81 Data version = 4.62
Experiment list contains 777 experiments for
(no ligands specified)
3 metals : Tl+, Tl++, Tl+++
(no references specified)
(no experimental details specified)
************************************
                HL
                    Electron
                                 (442)
Electron:
        .....
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ EMF oth/un 25°C 0.0 C T 1993MCb (969) 1
In KI. T1(Hg)/T1(I) electrode. K(T1I(s)+e=T1(Hg)+I-)=-12.297(-0.72872 mV).
At 40 C, K=12.364 (-0.73266). Tl(Hg) has 0.001-0.002 mol fraction Tl.
-----
Tl+ EMF KNO3 25°C 0.0 M TIH
                                       1981GLc (970) 2
                            K=-10.988 (-650.10 \text{mV})
                            K' = -10.432 (-617.18 \text{ mV})
Method: T1(Hg) and T1/T1Br electrodes in 0.005-0.10 M KNO3. Data for
10-70 C. K: TlBr+e=Tl(s)+Br-; K': TlBr+Hg+e=Tl(Hg)+Br-.
______
                           1974BNb (971) 3
Tl+ EMF non-aq 30°C 100% U
                            K=-9.97(-599.4mV) M units
Medium: N.N-dimethylformamide; K: TlCl(s)+e=Tl(s)+Cl-
    EMF non-aq 23°C 100% U
                                       1974CRa (972) 4
                            K=-16.56(-973mV)
Medium: n-hexanol; K: TlCl(s)+e=Tl(s)+Cl-
Tl+ EMF non-ag 23°C 100% U
                                       1974SRg (973) 5
                           K=-15.06(-886.8mV)
Medium: N.N-dimethylformamide; K: TlCl(s)+e=Tl(s)+Cl-
_____
T1+
                                       1967KRb (974)
       EMF NaClO4 25°C 3.0M U I
                            K(T1+e=T1(s))=-6.649, -393.3mV
                            K'=-9.72, -575 \text{ mV}
K': TlCl(s)+e=Tl/Hg+Cl. I=2.0: K=-6.336, -374.8 mV, K'=-9.43, -558 mV;
I=1.0: K=-6.038, -357.2 mV, K'=-9.23, -546 mV
______
T1+
      EMF NaClO4 25°C 3.00M U
                                       1966GKb (975) 7
                           K(T1+ +e=T1/Hg)=-6.606, -390mV
______
Tl+ EMF none 0°C 0.0 U T
                                       1965MLa (976)
                                                    8
                           K=-9.352, -506.8 \text{ mV}
K: TlCl(s) + e = Tl/Hg + Cl-. K=-8.796(25 C), -8.372(50 C), -7.992(80 C)
______
    EMF NaClO4 25°C 3.0M U
T1+
                                       1959SCb (977) 9
```

SC-Database

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K: Tl+e=Tl(in Hg,saturated)
______
     EMF non-aq 25°C 100% U T
                               1954PSa (978) 10
                        K=-5.81(-344mV) M units
Medium: formamide; K: Tl+ + e=Tl(s). K=-5.85(-338mV,18 C) M units
______
Tl+ EMF none 25°C 0.0 U T
                                  1934CMa (979) 11
                        K(T1+e=T1(s))=-5.68(-336.0 \text{ mV})
                        K(T1C1(s)+e=T1(s))=-9.42(-557)
                        K(TlBr(s)+e=Tl(s))=-11.11
                        K(TII(s)+e=TI(s))=-12.95(-766)
K(T1+e=T1(s))=-5.595(0 C, -303.2 mV), -5.639(12.5 C, -319.6 mV), -5.718(37.5 C,
-352.4 mV),-5.819(50 C,-373.1 mV)
*******************
             HL
                            (2497)
Tetrafluoroborate:
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
    vlt non-aq 22°C 100% U K1=2.9
                                 1988BEb (1204) 12
Medium: CH2Cl2
______
    con non-ag 25°C 100% U
                        K1=1.15
                                  1970YKb (1205) 13
Medium: MeCN
********************************
             HL Bromide
                           CAS 10035-10-6 (19)
Bromide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      nmr NaClO4 25°C 4.0M U I
                        K1=9.6 B2=16.9 1981GHa (2333) 14
                        B3=22.2
                        B4=26.5
Medium: 1M NaClO4/3M HClO4, [Tl]=1.0 M
______
     EMF KNO3 25°C 0.0 M H
                                  1981GLc (2334) 15
                        Kso = -5.459
Method: measurements with Tl(Hg) and Tl/TlBr elecrodes in 0.005-0.10 M
KNO3. Data for 10-70 C.
______
     EMF non-ag 25°C 100% C TIH
                                  1981STb (2335) 16
Method: Tl/Hg electrode. Medium: DMF. DH(K1)=-29.38 kJ mol-1, DS=15.95
J K-1 mol-1, Kso(TlCl)=-7.94. In PC: DH(K1)=-70.40, DS=-8.55, Kso=-10.84.
______
            25°C 0.0 C T H
T1+
      EMF none
                                  1981STb (2336) 17
Method: Tl/Hg electrode. DH(K1)=-34.57 kJ mol-1, DS=-3.13 J K-1 mol-1.
______
Tl+ sol NaClO4 10°C 0.50M U TIH
                         K1=0.92
                                 1974FRd (2337) 18
                        Kso = -5.60
```

```
Medium: LiClO4. K1=0.89(I=1); K1=0.86, B2=0.38(I=2); K1=0.88, B2=0.30, B3=-0.10
(I=3). Kso=-5.49(I=1), -5.34(I=3). Also at 10-60 C and I to 4 M LiClO4
______
      sol none 25°C 0.0 U T H K1=1.08 B2=0.60 1974FRd (2338)
T1+
                        Kso = -5.45
K1=1.2,Kso=-5.95(10 C); K1=0.98, B2=0.52, Kso=-5.03(40 C); K1=0.80, B2=0.40,
Kso=-4.62(60 C)
______
Tl+ sol non-aq 25°C 100% U I
                                  1974MUa (2339) 20
                        Kso = -8.08
Medium: DMF. In DMSO: Kso=-5.32. In propene carbonate: Kso=-11.11
______
     sol non-aq 25°C 100% U
                        B2=6.4
                              1973BNa (2340) 21
T1+
                        Kso=-8.1
Medium: N,N-dimethylacetamide
-----
     vlt NaClO4 25°C 1.0M U K1=0.93 B2=1.1 1972BHb (2341)
                                               22
_____
    sp none 25°C 0.0 U K1=0.79
                                  1972CPa (2342) 23
_____
Tl+ sol none 25°C 0.0 U T
                                  1972KEa (2343) 24
                        Kso = -5.356
Kso=-5.882(10.1 C), -5.705(15 C), -5.528(20 C), -5.189(30 C), -5.001(35 C),
-4.852(40 C), -4.730(45 C)
                -----
     sol none 25°C 0.0 U T
                                  1972KEa (2344) 25
                        Kso = -5.596
In D20; Kso=-6.179(10 C), -5.965(15 C), -5.802(20 C), -5.447(30 C), -5.292(35 C),
-5.167(40 C),-4.999(45 C)
______
     EMF non-aq 25°C 100% U
                                  1970SAc (2345) 26
T1+
                        Kso = -12.66
Medium: propene carbonate
______
      sol none 25°C 0.0 U K1=0.93 1969CPa (2346) 27
T1+
-----
                         K1=2.5 B2=3.0 1969LUb (2347)
Tl+
     sol non-aq 24°C 100% U
                        B3=2.9
                        B12=2.6
                        Kso=-5.3
Medium: DMSO, 1 M LiClO4. TlHg electrode also used
______
      sol non-aq 275°C 100% U T K1=0.70 1965SPa (2348) 29
T1+
Medium: (Na,K)NO3. K1=0.48(300 C) m units
______
     sol none 20°C 0.0 U T H K1=0.98 B2=1.10 1964PCa (2349)
T1+
                        Kso(TlBr)=-5.60
At 30 C: K1=0.87, K2=-0.01; 40 C: K1=0.73, K2=-0.15. At I=0 corr.,25 C: K1=
0.93, K2=0.06. DH(K1)=-17.7 kJ mol-1, DS=-41 J K-1 mol-1; DH(K2)=22.9, DS=76
-----
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```
sol none 25°C 0.0 U
T1+
                           K1=0.62 B2=1.14 1962SDc (2350) 31
                          Kso(TlBr)=-5.42
                          K(TlBr(s)=TlBr)=-4.80
I=0 corr. By solubility in KBr B4/B2=-1.20
   sol NaClO4 25°C 4.0M U K1=0.34 B2=0.18 1960KMa (2351)
T1+
                                                  32
                          K3 = -0.23
-----
T1+
     sol oth/un 25°C var U I K1=1.05 B2=0.77 1958KMa (2352)
                                                  33
                          B3=0.24
                          K(T1Br(s)=T1Br)=-4.34
                          Kso(TlL)=-5.38
Medium: LiBr. In NaBr K1=0.92,B2=0.80,B3=0.31 and K(TlBr(s)=TlBr)=-4.45. In
KBr K1=0.92,B2=0.92,B3=0.40,K=-4.45; in CsBr K1=1.05,B2=1.00,B3=0.64,K=-4.31
-----
    sol NaClO4 25°C 4.0M U
T1+
                                    1958MIb (2353) 34
                          Kso = -4.82
In dilute solution: Kso=-5.38
Tl+ sol NaClO4 25°C 4.0M U
                          K1=0.32 B2=0.15 1957NIa (2354) 35
                          K3 = -0.45
                          K4 = -0.75
                          K(TlBr(s)+2Br=TlBr3)=-5.10
                          K(T1Br(s)+3Br=T1Br4)=-5.80
Kso(TlL)=-4.81, K(TlL(s)=TlL)=-4.48, K(TlL(s)+L=TlL2)=-4.62
By Tl/Hg electrode Kso=-4.81
T1+
      sol none 25°C 0.0 U T H
                          K1=0.88
                                    1957NNa (2355) 36
                          K(TlL(s)+TlL)=-1.55
I=0 corr. DH(K1)=-10.3 kJ mol-1, DS=-18 J K-1 mol-1. At 40 C: K1=0.80,
K(TlL(s)=TlL)=-1.19
_______
   sp NaClO4 ? 2.20M U K1=1.60 1956PVa (2356) 37
______
Tl+ sol none 25°C 0.0 U T H K1=1.05 1955ANd (2357) 38
                          Kso(TlBr)=-5.47
I=0 corr. K1=1.26(5 C), 1.00(45 C). Kso=-6.23(5 C), -4.89(45 C). DH(K1)=-10
kJ mol-1, DS=-16; DH(so)=56.4, DS=84.5
______
      sol none 25°C 0.0 U H
                                    1953ADa (2358) 39
I=0 corr. DH(K1)=-5.82 kJ mol-1; DS=0.4 J K-1 mol-1
______
Tl+ ISE none 25°C 0.0 U
                                     1934ITa (2359) 40
                         Kso(TlBr)=-5.41
______
     con none 26°C 0.0 U T
T1+
                                     1923BOa (2360) 41
                         Kso(TlBr)=-5.41
I=0 corr. Kso=-6.02(9.4 C), -5.68(18 C), -4.20(68.5 C)
______
      con oth/un 20°C dil U
T1+
                                 1903B0b (2361) 42
```

```
Kso(TlBr)=-5.60
```

```
**********************************
Br03-
                 Bromate
                             (6017)
Bromate:
         Metal Mtd Medium Temp Conc Cal Flags Lg K values
______
     sol oth/un 45°C 0.0 U T
                                  1968KHa (2435) 43
                        K1=0.3
                         Kso = -3.34
Kso=-3.78(30 C),-3.62(35 C),-3.47(40 C)
     sol none 40°C 0.0 U
                                  1923B0a (2436) 44
T1+
                         Kso(TlL)=-3.41
______
T1+
     con oth/un 20°C dil U
                                   1903B0b (2437) 45
                         Kso(TlL)=-4.07
**********************************
             H2L
CO3--
                 Carbonate
                            CAS 465-79-6 (268)
Carbonate;
          ______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
-----
     sol NaClO4 20°C 3.40M U K1=0.51 B2=0.11 1980FPa (3400)
                                                46
*********************************
C6N6Fe----
             H4L
                             (2191)
Hexacyanoferrate (II); Fe(II)(CN)6----
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
T1+
      ISE oth/un 25°C 1.00M U TIH K1=0.78
                                  1984FIa (3608) 47
Medium: KF
      ISE NaClO4 25°C 3.0M U H K1=0.82
                                   1967MKc (3609) 48
Method: amalgam electrode. Medium: LiClO4. By solubility: K1=0.5 ?
By calorimetry: DH(K1)=-7.4 kJ mol-1, DS=9.2 J K-1 mol-1
______
T1+
     sol oth/un 18°C dil U M
                                   1958DTb (3610) 49
                         Kso = -10.17
                         Ks(Ag3T1L) = -23.55
                         Ks(Ag2T12L) = -17.95
      sp none 25°C 0.0 U T H K1=3.00
                                  1958PWa (3611) 50
DH(K1)=4.6 \text{ kJ mol-1,DS}=72.8 \text{ J K-1 mol-1}(25 \text{ C}). K1=3.05(35 \text{ C}), 3.06(50 \text{ C})
-----
      sol none 0°C 0.0 U T H K1=3.19
                                   1953BGb (3612) 51
DH(K1)=3.4 kJ mol-1, DS=72.8 J K-1 mol-1(25 C). K1=3.22(25 C), 3.27(40 C)
********************************
C6N6Fe---
             H3L
                 Ferricyanide
                          (2491)
Hexacyanoferrate (III); Fe(III)(CN)6---
______
```

Metal	Mtd Medium Temp Conc Cal Flags Lg K values Reference Ex	xptNo
	nmr oth/un 25°C 0.30M U K1=1.83 1959GRa (3694) ************************************	
Metal	Mtd Medium Temp Conc Cal Flags Lg K values Reference Ex	xptNo
T1+	sp NaCl 25°C 0.0 C T H K1=0.49 B2= 0.00 1998BSa B4=-1.42	(58 01) 53
DH(K1)=-0.	ed from data for 0.0012-3.0 M NaCl solution. Data for 25-200 C. 0.90 kJ mol-1, DS(K1)=6.37 J K-1 mol-1; DH(K2)=-6.08, DS(K2)=-29	9.8
Tl+	sol KCl 25°C 0.0 C I 1993KFb (5802 3. Solubility measured in 0.005-4.0 m LiCl and KCl.) 54
	sol NaCl 25°C 0.0 C I 1992KFa (5803 3. Solubility measured in 0.005-4.0 m NaCl.) 55
T1+	sol NaCl 25°C 0.0 U I K1=0.60 B2=0.28 1992RAb B4=-2.72	(5804) 56
I=0 to 6.0		
	sol NaCl 25°C 0.0 C I 1991KFb (5805 Kso(TlCl)=-3.73 ed from data for 0.1-3.2 m NaCl/HCl, using Pitzer equation.) 57
Tl+ In HF solu) 58
	ISE NaClO4 25°C 1.50M M I K1=1.11 B2=0.35 1989CIa NaClO4, K1=0.95, B2=0.32	 (5807) 59
Medium: 4		` ,
T1+	ISE oth/un 25°C 0.50M U I K1=1.56 B2=1.94 1983EIa Na-acetate	
	ISE oth/un 25°C 0.50M U I K1=0.02 B2=-0.56 1983FIa	 (5810) 62
	nmr NaClO4 25°C 4.0M U I K1=7.18 B2=12.78 1981GHa B3=16.70 B4=19.68 B5=19.15 B6=19.04	 (5811) 63
Medium: 1M	1M NaClO4/3M HClO4. [Tl]=1.0 M. Also data for [Tl]=0.05 M	

```
EMF non-aq 25°C 100% C TIH
                                   1981STb (5812) 64
Method: Tl/Hg electrode. Medium: DMF. DH(K1)=-30.51 kJ mol-1, DS=18.46
J K-1 mol-1, Kso(TlCl)=-8.58. In PC: DH(K1)=-77.33, DS=-12.82, Kso=-11.31.
______
  EMF none 25°C 0.0 C T H 1981STb (5813) 65
Method: Tl/Hg electrode. DH(K1)=-26.30 kJ mol-1, DS=-4.97 J K-1 mol-1.
______
      vlt NaCl 25°C 1.0M C B2=1.15
                                  1975APd (5814) 66
Method: polarography.
______
      sol non-aq 25°C 100% U I
                                   1974MUa (5815) 67
                         Kso = -8.53
Medium: DMF. Kso=-5.58 (in DMSO), -11.45 (in propene carbonate)
                       B2=7.2 1973BNa (5816) 68
Tl+ sol non-aq 25°C 100% U
                        Kso=-8.9
Medium: N,N-dimethylacetamide
______
Tl+ sol none ? 0.0 U I K1=0.59 B2=0.32 1973P0b (5817)
0 corr from NaCl. K1=0.57,B2=0.36(0 corr from NH4Cl). K1=0.56,B2=0.38(0 corr
from HCl)
______
   sol oth/un 25°C var U K1=-0.6
T1+
                                1972AAb (5818) 70
______
    sol none 25°C 0.0 U K1=0.65
                                1972CPa (5819) 71
_____
      ISE NaCl04 25°C 1.0M U I K1=-0.02 B2=-0.39 1972FIb (5820)
Medium: LiClO4. K1=0.4, B2=-0.2(I=0); K1=-0.08, B2=-1.0(I=2). TlHg electrode
______
Tl+ sol none 25°C 0.0 U T
                                   1972KEa (5821) 73
                         Kso = -3.656
Kso=-4.046(10.1 C), -3.923(15 C), -3.811(20 C), -3.542(30 C), -3.436(35 C),
-3.312(40 C), -3.238(45 C)
______
Tl+
      sol none 25°C 0.0 U T
                                   1972KEa (5822) 74
                         Kso = -3.862
In D20. Kso=-4.305(10 \text{ C}), -4.139(15 \text{ C}), -4.006(20 \text{ C}), -3.764(30 \text{ C}),
-3.637(35 C), -3.571(40 C), -3.403(45 C)
______
      ISE NaClO4 25°C 3.0M U K1=0
                                 1971BSd (5823) 75
T1+
Method: Tl amalgam electrode
______
      ISE NaCl04 25°C 1.0M U I K1=0.08 B2=0.04 1971FRb (5824)
Medium: LiClO4. K1=0.40, B2=-0.60(I=0); K1=0.11(I=0.5); K1=-0.12, B2=0.05
(I=2); K1=-0.10, B2=-1.1(I=3); K1=-0.08, B2=-1.2(I=4). TlHg electrode
_____
Tl+ sp NaClO4 25°C 0.15M U K1=0.31 1971MMg (5825) 77
-----
      vlt NaClO4 30°C 1.0M U I K1=0.32
                                 1970B0d (5826) 78
```

```
Medium: Na(F,ClO4). K1=0.0, B2=-0.44(I=4)
______
   EMF non-aq 25°C 100% U
                                    1970SAc (5827) 79
                          Kso = -12.39
Medium: propene carbonate
                      -----
_____
Tl+ cal none 25°C 0.0 U H
                                    1969BPa (5828) 80
DH(K1) = -6.3 \text{ kJ mol} -1
______
Tl+ con diox/w 25°C 20% U I K1=1.01 1969DFa (5829) 81
Medium: 19.8% w/w dioxan/H20. K1=0.72(0%), 1.40(35.8%), 1.61(43.6%),
2.06(53.6%), 2.31(58.2%), 2.94(66.2%)
_____
Tl+ sol non-aq 24°C 100% U
                          K1=2.3 B2=3.4 1969LUb (5830) 82
                          B3=3.2
                          B(T12L)=3.0
                          Kso=-5.5
Medium: DMSO, 1 M LiClO4. TlHg electrode
Tl+ sol none 25°C 0.0 U K1=0.62
                                   1969MPa (5831) 83
Using spect., K1 <0.3
______
Tl+ ISE non-aq 25°C 100% U
                                     1969SBa (5832) 84
                          K(TlL(s)=TiL)=-4.6
                          K(TiL(s)+L=TiL2)=-2.1
                          Kso(TlL(s)=Tl+L)=-9.0
Medium: DMF. In DMSO: Ks1=-2.95, Ks2=-1.8, Kso=-6.4. In propene carbonate:
Ks1=-6.4, Ks2=-4.1, Kso=-12.4
Tl+ sol none 30°C 0.0 U K1=0.60
Kso=-3.62
                          K1=0.60 1967KHa (5833) 85
-----
Tl+ sol none 25°C 0.0 U I K1=0.52
                                  1967KPa (5834) 86
                          Kso = -3.74
In 16.7% MeOH: K1=0.70. K1=0.90(30%), 1.34(60%); Kso=-4.15(16.7%),
-4.49(30%), -5.36(60%)
       ISE non-aq 25°C 100% U I K1=2.70 B2=3.95 1966CBa (5835) 87
T1+
                          Kso = -6.26
Medium: DMSO, 0.5 M LiClO4. Kso(TlCl(s)=Tl+Cl)=-7.21(I=0). TlHg electrode
-----
    sol none 25°C 0.0 U K1=0.62 1966MPa (5836) 88
                          Kso = -3.74
------
Tl+ sol oth/un 25°C 0.0 M K1=0.6 1964MPa (5837) 89
Tl+ vlt oth/un 25°C var U
                          B2=-0.80 1963KMd (5838) 90
                          B3 = -1.68
                          B4 = -2.64
Medium:LiCl var
```

```
Tl+ sol NaCl04 25°C 4.0M U T H K1=0.00 B2=-0.58 1963KMe (5839)
                                                91
Medium:LiClO4. K1=0.04,B2=-0.74(15 C); -0.05,-0.55(40 C); -0.10,-0.40(60 C);
-0.12, -0.5(80 C). DH(K1)=-4.6 kJ mol-1, DS=-17 J K-1 mol-1; DH(B2)=9.6, DS=-17
______
Tl+
      sol NaClO4 25°C 4.0M U T H
                                   1963KMe (5840) 92
                         K(K+T1C12)=-0.58
Medium: LiCl04. K=-0.37(15 C),-0.72(40 C),-0.69(60 C),-0.82(80 C)
DH(K)=-11 kJ mol-1. DS=-54 J K-1 mol-1(25 C)
______
Tl+ ISE none 25°C 0.0 U K1=0.80 1962APa (5841) 93
_____
   vlt NaClO4 25°C 2.0M U K1=0.19 1962BSc (5842) 94
______
Tl+ sol non-aq 275°C 100% U T K1=0.3 1962SIc (5843) 95
Medium: liquid (Na/K)NO3, m units
______
      sol none 25°C 0.0 U K1=0.74 1962SMc (5844) 96
Kso(TlL(s))=-3.75
______
      dis non-aq 480°C 100% U I K1=0.85 1961KEb (5845) 97
Medium: liquid KNO3. Kd(TlL(in AgCl(1)) = TlL(in KNO3(1)) = -1.7. In liquid
K2S2O7 K1=0.3, Kd=-1.4. In m units
______
Tl+ sol NaClO4 25°C 4.0M U
                        K1=0.09 B2=0.74 1961KMb (5846)
                                                98
                        Kso(Tll(s))=-2.8
Method: Tl/Hg electrode. Medium: LiClO4. Also data on addn. of Na, K, Rb, Cs
See: V Mironov, Zh.Neorg.Khim., 1963, 8,764
______
Tl+ vlt none 25°C 0.0 U K1=0.46 1961NRa (5847) 99
Tl+ vlt NaClO4 25°C 1.0M U K1=0.64 1961NRa (5848) 100
Tl+ ISE none 25°C 0.0 U K1=0.60 B2=0.40 1958B0b (5849) 101
______
Tl+ ix none 25°C 0.0 U K1=0.46 B2=-0.02 1958H0a (5850) 102 K3=-0.9?
T1+
      ISE NaClO4 25°C 4.0M U K1=0.00 B2=-0.80 1957NIa (5851) 103
Tl+
                         Kso(TlL(s))=-3.04
Method: Tl/Hg electrode. By solubility K1=-0.1, K2=-0.7, K3=-0.9, Kso=-3.04
Ks(TlL(s)=TlL)=-3.15, K(TlL(s)+L=TlL2)=-3.74, K(TlL(s)+2L=TlL3)=-4.70
______
      oth none 25°C 0.0 U H K1=0.49 1957NNa (5852) 104
extrapolated to zero ionic strength,
DS(K1) = -7.5 \ J \ K-1 \ mol-1
______
Tl+ oth none 40°C 0.0 U K1=0.44 1957NNa (5853) 105
_____
   sp NaCl04 ? 2.20M U K1=-0.13 1956PVa (5854) 106
```

Tl+ sol none 25°C 0.0 U T H K1=0.68 1955ANd (5855) 107	
Tl+ sol none 25°C 0.0 U K1=0.60 B2=0.17 1955HSa (5856) 1	108
Tl+ sol none 25°C 0.0 U H 1953ADa (5857) 109 I=0 corr. DH(K1)=1.1 kJ mol-1, DS=17 J K-1 mol-1	
Tl+ sol none 25°C 0.0 U T K1=0.68 1953BGb (5858) 110 I=0 corr. K1=0.78(0 C), 0.64(40 C). DH(K1)=-6.0 kJ mol-1, DS=-7.1 J K-1 m-1	
Tl+ con none 25°C 0.0 U I K1=0.64 1945GVa (5859) 111 I=0 corr. Also in (CH2OH)2/H2O mixtures	
Tl+ sol none 25°C 0.0 U K1=0.66 1943BGa (5860) 112	
Tl+ sol none 25°C 0.0 U K1=0.66 1941HGb (5861) 113	
Tl+ con none 18°C 0.0 U K1=0.60 1937RDa (5862) 114	
Tl+ con none 25°C 0.0 U K1=0.47 1937RDa (5863) 115	
Tl+ ISE none 25°C 0.0 U K1=0.51 1934CMa (5864) 116	
Tl+ sol none 18°C 0.0 U K1=0.52 1930BDa (5865) 117	
Tl+ ISE alc/w 25°C 100% U 1929BHa (5866) 118 Kso(TlL(s))=-4.54	
Medium: MeOH	
Tl+ sol none 25°C 0.0 U 1928RVa (5867) 119 Kso(TlL(s))=-3.726	
Tl+ con none 18°C 0.0 U K1=0.51 19270Na (5868) 120	
Tl+ sol none 25°C 0.0 U T 1926BHa (5869) 121 Kso(TlL(s))=-3.72	
I=0 corr. Kso=-4.43(0 C), -3.16(50 C)	
Tl+ sol none 40°C 0.0 U 1923B0a (5870) 122 Kso(TlL(s))=-3.32	
Tl+ con none 26°C 0.0 U T 1923B0a (5871) 123 Kso(TlL(s))=-3.67 I=0 corr. Kso=-4.12(9.5 C)	
Tl+ sol none 25°C 0.0 U 1923B0a (5872) 124	

Kso(TlL(s))=-3.66

Tl+	con none	20°C	0.0 U	Kso(TlL(s))=-3.	1903B0b (5873) 125 82
******	******	*****	*********		**********
ClO3- Chlorate;				CAS 7790-9	
Metal	Mtd Medium	Temp C	Conc Cal Flag	s Lg K values	Reference ExptNo
T1+ ******	sol none ******	25°C *****	0.0 U	K1=0.47 *******	1892NOa (6064) 126 ********
ClO4- Perchlorat		HL	Perchlorate	CAS 7001-9	0-3 (287)
Metal	Mtd Medium	Temp C	Conc Cal Flag	s Lg K values	Reference ExptNo
Tl+ Medium: CH		22°C 1	.00% U	K1=3.3	1988BEb (6383) 127
T1+	sol NaClO4	25°C	? U	K1=-0.5	1973J0a (6384) 128
Tl+ Medium: KF		30°C	1.0M U	K1=-0.49	1970BOd (6385) 129
	con non-aq CN, 0 corr	25°C 1	.00% U	K1=1.51	1970YKb (6386) 130
	•		0.0 U T -0.34(60 C)		1967ZBa (6387) 131
Tl+	oth oth/un	25?°C	0.0 M		1966MBb (6388) 132
Tl+ ******	con none ******	25°C *****	0.0 U	K1=0.0 ******	1937RDa (6389) 133 ***********************************
CrO4 Chromate;		H2L	Chromate	CAS 7738-9	4-5 (2382)
Metal	Mtd Medium	Temp C	Conc Cal Flag	s Lg K values	Reference ExptNo
	sol oth/un		2.40M U	Kso=-10.0	1974FEa (6512) 134
Tl+	sol NaClO4	20°C 3	3.00M U	Kso=-9.85	1974FGe (6513) 135
	sol oth/un			Kso=-11.70	1958KGb (6514) 136
T1+	ISE none	25°C	0.0 U		1953SUa (6515) 137

Kso=-12.01

F- Fluoride;			HL	Fluc	oride	CAS 764	4-39-3	(201)	
Metal	Mtd	Medium	Temp	Conc (Cal Flags	Lg K values		Refer	ence	ExptNo
	con	oth/un	25°C	1.00M	U I	K1=-0.07 Bind 3.0 M				
Tl+ Method: po						K1=2.88				4) 139
T1+							197	'3J0a	(726	·
Tl+ Medium: HF	EMF				U		196			
Tl+ Medium: li	EMF iquid	non-aq HF, I=0	0°C	100%	U	K1=3.33	196	1CZa	(726	7) 142
Tl+	sol	none	25°C	0.0	U	K1=0.10	195	3BGb	(726	8) 143
FClBrI			HL			************ (541) `ligand 80)				*****
FClBrI Halides, o	compar	rative (HL (for b	ook da	ata under Cal Flags	(541) `ligand 80) S Lg K values		 Refer	 ence	 ExptNo
FClBrI Halides, d Metal 	compar Mtd	rative (Medium	HL (for b Temp	oook da Conc (ata under Cal Flags U M	(541) r ligand 80) s Lg K values B(TlClBr)=0.8 B(TlClBr1)=2.2	 196 80 .93	 Refer	ence	 ExptNo
FClBrI Halides, c Metal Tl+	compar Mtd sol	rative (Medium oth/un	HL (for t	oook da Conc (var	ata under Cal Flags U M	(541) r ligand 80) r Lg K values B(TlClBr)=0.8 B(TlClBr2)=0 B(TlBrI)=2.24 B(TlBrI2)=2.4	 196 80 .93 4 42 *****	 Refer 2FSa	 ence (743	 ExptNo 6) 144
FC1BrI Halides, c Metal T1+ *********	compar Mtd sol	rative (Medium oth/un	HL (for t	oook da Conc (var	ata under Cal Flags U M	(541) r ligand 80) s Lg K values B(TlClBr)=0.8 B(TlClBr2)=0 B(TlBrI)=2.24 B(TlBrI2)=2.4	 196 80 .93 4 42 *****	 Refer 2FSa	 ence (743	 ExptNo 6) 144
FC1BrI Halides, 0 Metal T1+ ********* H20 Water	compar Mtd Sol	rative (Medium oth/un	HL (for b Temp 25°C	oook da Conc (var ******	ata under Cal Flags U M	(541) r ligand 80) r Lg K values B(TlClBr)=0.8 B(TlClBr2)=0 B(TlBrI)=2.24 B(TlBrI2)=2.4	 196 80 .93 4 42 ******	 Refer 2FSa ***** (611	 ence (743 ***** 5)	 ExptNo 6) 144 ******
FClBrI Halides, 0 Metal Tl+ ******** H20 Water Metal	compar Mtd Sol	rative (Medium oth/un *******	HL (for b Temp 25°C	cook da	ata under Cal Flags U M ******* er Cal Flags	(541) r ligand 80) s Lg K values B(TlClBr)=0.8 B(TlClBr2)=0 B(TlBrI)=2.24 B(TlBrI2)=2.4 ************************************	 196 80 .93 4 42 ****** 2-18-5	 Refer 2FSa ***** (611 Refer	 ence (743 ***** 5) ence	 ExptNo 6) 144 ******
FClBrI Halides, C Metal Tl+ ******** #20 Water Metal Tl+ Medium: Et	compar Mtd sol ****** Mtd vlt	Medium oth/un ******* Medium alc/w	HL (for b Temp 25°C ****** L Temp 25°C	var ***** Wate	ata under Cal Flags U M ******* er Cal Flags	(541) r ligand 80) s Lg K values B(TlClBr)=0.8 B(TlClBr2)=0 B(TlBrI)=2.2 B(TlBrI2)=2.4 ************************************	 196 80 .93 4 42 ****** 2-18-5	 Refer 2FSa ***** (611 Refer 8VAa	 ence (743 ***** 5) ence (761	 ExptNo 6) 144 ****** ExptNo 3) 145
FClBrI Halides, C Metal Tl+ ******** #20 Water Metal Tl+ Medium: Et	compar Mtd sol ****** Mtd vlt	Medium oth/un ******* Medium alc/w	HL (for b) Temp 25°C ****** L Temp 25°C	oook da	ata under Cal Flags U M ********* Cal Flags	(541) f ligand 80) G Lg K values G Lg K values G B(TlClBr)=0.8 G B(TlClBr2)=0 G B(TlBrI)=2.2 G B(TlBrI)=2.3 G CAS 773: G Lg K values K2=-1.48 K3=-1.66	 196 80 .93 4 42 ****** 2-18-5 195	 Refer 2FSa ***** (611 Refer 8VAa	 ence (743 ***** 5) ence (761	 ExptNo 6) 144 ****** ExptNo 3) 145

```
T1+
      EMF oth/un 25°C 0.0 C T
                                   1993MCb (8386) 146
                         Kso(T11) = -7.354
Method: T1(Hg)/T1(I) electrode. At 40 C, Kso=-6.729; at 55 C, Kso=-6.198.
Medium 0.10 M KI. Cell emf independent of [I].
-----
Tl+ EMF non-aq 25°C 100% C TIH
                               1981STb (8387) 147
Method: Tl/Hg electrode. Medium: DMF. DH(K1)=-26.86 kJ mol-1, DS=12.28
J K-1 mol-1, Kso(TlCl)=-6.86. In PC: DH(K1)=-46.62, DS=10.36, Kso=-9.99.
______
Tl+ EMF none 25°C 0.0 C T H
                               1981STb (8388) 148
Method: Tl/Hg electrode. DH(K1)=-70.05 kJ mol-1, DS=-29.69 J K-1 mol-1.
______
      sp NaClO4 25°C 0.02M U T H K1=2.86 1975PFa (8389) 149
L=the triiodide ion. K1=2.70 (5 C); 2.80 (15 C); 2.87 (35 C); 2.83 (45 C)
______
    sol non-aq 25°C 100% U I
T1+
                                   1974MUa (8390) 150
                         Kso = -7.01
Medium: DMF. In DMSO: Kso=-4.78. In propene carbonate: Kso=-9.99
______
Tl+ sol non-aq 25°C 100% U B2=6.2 1973BNa (8391) 151
                        Kso=-6.8
Medium: N,N-dimethylacetamide
-----
Tl+ EMF non-aq 25°C 100% U
                                   1970SAc (8392) 152
                        Kso = -12.22
Medium: propene carbonate
______
Tl+ ISE non-aq 24°C 100% U
                         K1=1.9 B2=2.3 1969LUb (8393) 153
                         B3=2.4
                         B(T12I)=1.0
                         B(T13I)=1.9
                         Kso(TlI(s)=Tl+I)=-4.9
Medium: DMSO, 1 M LiClO4. TlHg electrode
------
Tl+ con non-ag 140°C 100% U K1=2.91 1967BNb (8394) 154
                         K(T1I+T1)=3.10
Medium: liquid I2
______
  EMF NaClO4 25°C 7.0M U
T1+
                                   1966J0a (8395) 155
                         Kso(T1(I3)) = -7.74
Medium: 3 M HClO4,4 M NaClO4. Kso(TlI)=-6.77, Ks(TlI0.83(I3)0.17)=-6.16
______
      sol oth/un 25°C var U H 1963KMd (8396) 156
T1+
Medium:KI. DH(K1)=-17 kJ mol-1,DS=-29 J K-1 mol-1; DH(B2)=-30, DS=-67
______
Tl+
      sol NaClO4 25°C 4.0M U
                         K1=0.76 B2=0.90 1960KMa (8397) 157
                         K3=0.14
                         K4 = -0.19
                         B4=0.85
```

```
sol oth/un 20°C var U T H B2=2.20
T1+
                                      1958KMb (8398) 158
                           B3=1.95
                           B4=1.54
                           Kso(AgL)=-7.49
Medium:KI. DH(B2)=-35 kJ m-1,DH(B3)=-39,DH(B4)=45.2. 30 C:Kso=-7.07,K(TlL(s)
=T1L)=-5.40,B2=2.03,B3=1.80,B4=1.29. 40 C:Kso=-6.69,K=-5.05,B2=1.72,B3=1.51
______
    sol oth/un 50°C var U T
                           K1=1.38 B2=1.58 1958KMb (8399) 159
T1+
                           B3=1.31
                           B4=0.80
                           Kso(TlL)=-6.31
                           K(T1L(s)=T1L)=-4.92
Medium KI. At 60 C: Kso=-5.96, K=-4.74, K1=1.21, B2=1.40, B3=1.12, B4=0.58.
At 70 C: Kso=-5.63, K=-4.43, K1=1.17, B2=1.28, B3=0.94, B4=0.37
______
    EMF NaClO4 25°C 4.0M U
T1+
                                      1958MIa (8400) 160
                           Kso = -6.73
-----
      sol oth/un 70°C dil U T
                                      1958MIa (8401) 161
                           Kso = -5.63
Kso=-7.49(20 C), -7.24(25 C), -6.69(40 C), -6.31(50 C), -5.96(60 C)
_____
                      -----
Tl+ sol oth/un 25°C var U
                                      1958MIa (8402) 162
                          B4 = -0.92
Medium:ZnI2
______
      sol oth/un 25°C var U I
                           K1=1.52 B2=1.94 1957KMa (8403) 163
                           B3=1.72
                           B4=1.24
                           Kso(TlL)=-7.24
Medium: LiI. In NaI K1=1.50, B2-1.96, B3=1.71, B4=1.32. In NH4I K1=1.45,
B2=1.92, B3=1.86, B4=1.44. In RbI: K1=1.52, B2=2.00, B3=1.87, B4=1.51
______
Tl+ sol NaClO4 25°C 4.0M U
                           K1=0.72 B2=0.90 1957NIa (8404) 164
                           K3=0.18
                           K4 = -0.38
                           Kso(T1L) = -6.72
                           K(T1L(s)=T1L)=-6.00
K(T1L(s)+L=T1L2)=-5.82, K(T1L(s)+2L=T1L3)=-5.64, K(T1L(s)+3L=T1L4)=-6.00.
By Tl/Hg electrode Kso=-6.73
-----
   con none 25°C 0.0 U
                                     1937DRa (8405) 165
                           Kso(TlL)=-7.19
______
      con none 26°C 0.0 U T
                                      1923B0a (8406) 166
                           Kso(T1L) = -7.20
I=0 corr. Kso=-7.93(9.9 C), -7.55(18 C)
______
T1+
      EMF oth/un 25°C dil U
                                      1912SPa (8407) 167
                           Kso(TlL)=-7.51
```

T1+	con oth/un	20°C	dil U		1903BOb (8408) 168
****	****	***	****	Kso(T1L)=-7.44	·*************************************
IO3- Iodate;		HL		CAS 7782-6	
Metal	Mtd Medium	Temp	Conc Cal Flag	gs Lg K values	Reference ExptNo
Tl+	sol oth/un	20°C	2.40M U	Kso(TlL(s))=-4.	1974FEa (8562) 169 66
Medium: Na	2504				
T1+	sol NaClO4	20°C	3.0M U	Kso(TlL(s))=-4.	1974FGe (8563) 170 31
T1+	vlt NaClO4	25°C	1.0M U	K1=0.15	1972BHb (8564) 171
Tl+	sol none	25°C	0.0 UTH	Kso(TlL)=-5.51	1953BGb (8565) 172
I=0 corr.	DH(so)=55.6	kJ mo	ol-1, DS=81 J	• •	40(0 C), -5.09(40 C)
T1+	sol none	25°C	0.0 U	Kso(TlL)=-5.51	1929MGa (8566) 173
Tl+	EMF oth/un	25°C	dil U	Kso(TlL)=-5.34	1912SPa (8567) 174
Tl+	con oth/un	20°C	dil U	Kso(T1L)=-5.66	1903BOb (8568) 175
	******	*****			*******
MoO4 Molybdate;		H2L	Molybdate	(443)	
Metal	Mtd Medium	Temp	Conc Cal Flag	gs Lg K values	Reference ExptNo
Tl+	sol NaClO4	20°C	3.00M U	Kso(T12L(s))=-6	1974FGe (8760) 176
******	******	*****	******		********
NH3 Ammonia		L		CAS 7664-4	11-7 (414)
Metal		Temp	Conc Cal Flag		Reference ExptNo
T1+	vlt oth/un	25°C			1975APd (9215) 177
Tl+ Medium: NH	-	23°C	2.0M U	K1=-0.9	1941BJa (9216) 178

```
sol oth/un 16°C var U K1=-0.92 1928J0a (9217) 179
T1+
K1=-0.87 by spec. (Job's method)
*************************
           HL Nitrite CAS 7782-77-6 (635)
NO2-
Nitrite;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ con oth/un 25°C 0.0 U K1=0.80 1957NBa (9407) 180
By Tl electrode K1=0.85
HL Nitrate CAS 7697-37-2 (288)
Nitrate;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ oth oth/un 20°C 3.00M U K1=-0.57 1979FEa (9947) 181
Method: densimetry
_____
   oth NaCl04 20°C 3.0M U K1=-0.55 1979FEb (9948) 182
Method: densitometry
______
Tl+ vlt oth/un 25°C 0.20M C K1=1.18 B2= 2.30 1975APd (9949) 183
                      B3=3.0
Method: polarography. Medium: 0.20 M NaNO3.
______
   cal oth/un 25°C 0.5M U K1=-0.3 1975FRa (9950) 184
Background salt: LiClO4; For: I=1.0 M, K1=-0.32
                Tl+ sol NaClO4 20°C 0.40M U M 1975GFa (9951) 185
                      B(T1(S203)(N03))=-0.08
                      B(T1(S203)(N03)2)=-0.37
-----
     sol NaCl04 20°C 3.0M U K1=-0.6 B2=-1.5 1974FGe (9952) 186
______
Tl+ con oth/un 25°C 0.0 U K1=0.29 1974MWc (9953) 187
Tl+ oth oth/un 25°C var U 1971JCa (9954) 188
                     K(T1(H20)2+L=T1(H20)L)=-0.4
Method: dilatometry, densometry
______
     vlt oth/un 30°C 1.0M U I K1=-0.19 1970B0d (9955) 189
Medium: KF. K1=-0.43(I=4)
_____
     sol NaNO3 30°C 0.10M U I K1=0.41
                               1969KMd (9956) 190
In LiNO3: K1=0.30; KNO3: 0.53; CsNO3: 0.65; Mg(NO3)2: 0.75
Tl+ con diox/w 25°C 16% U I K1=0.69 1968DFa (9957) 191
Medium: 16\% dioxan. K1=0.51(0\%), 0.60(7.8\%), 1.04(36.2\%), 1.54(52.5\%),
1.81(59.2\%), 2.61(70.0\%), 3.27(76.4\%)
```

```
Tl+ oth oth/un 25?°C 0.0 U K1=0.5 1966MBb (9958) 192
     ISE NaClO4 25°C 3.0M U H K1=-0.48
                               1965KMb (9959) 193
Method: amalgam electrode. Medium: LiClO4. DH(K1)=-25.9 kJ mol-1, DS=-96
______
     vlt oth/un 25°C 0.0 U K1=0.38
                            1961NRa (9960) 194
T1+
_____
      sol oth/un 25°C 0.0 U T H K1=0.33 1957NNa (9961) 195
K1=0.38(0 C), 0.31(40 C). DH(K1)=-2.7 kJ mol-1, DS=-4 J K-1 mol-1 (25 C)
_____
     con oth/un 25°C 0.0 U K1=0.38
_____
     con oth/un 18°C 0.0 U K1=0.26 19270Na (9963) 197
***********************************
            HL Azide
N3-
                        CAS 7782-79-8 (441)
Azide:
_____
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sol non-aq 25°C 100% U
                               1973BNa (10260) 198
Medium: MeCONMe2. Kso=-6.9
______
Tl+ vlt NaCl04 25°C 2.0M U K1=0.40 1962BSc (10261) 199
     sol oth/un 25°C 0.0 U T H K1=0.39 1957NNa (10262) 200
K1=0.45(10 C), 0.35(40 C). DH(K1)=-5.6 kJ mol-1, DS=-11 J K-1 mol-1
______
Tl+ cal oth/un 25°C 0.0 U H
                               1956GWc (10263) 201
DH(Kso(TlL(s))=46.6 \text{ kJ mol}-1
______
     ISE oth/un 25°C 0.0 U
                               1952SUa (10264) 202
                      Kso(TlL(s))=-3.66
**********************************
OH-
            HL
                Hydroxide (57)
Hydroxide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      nmr NaClO4 25°C 2.0M C I
                               1997SCa (12305) 203
                      *K1=-0.03
                      *B2=-1.2
Additional method: spectrophotometry. Data for I=2.0-8.0 M NaClO4.
-----
      cal NaCl04 25°C 3.00M U H K1=0.09 B2=-0.86 1973KKg (12306) 204
Medium: LiClO4. DH(K1)=7.5 kJ M-1, DS=26.8 J K-1 M-1; DH(K2)=20.9, DS=52.3
            Tl+ sp oth/un 25°C 1.00M U
                               1971CHa (12307) 205
                      K(Tl+ + TlOH++ = Tl2O+H)=1.7
______
```

```
sp NaClO4 25°C 0.50M U I K1=0.30
                                 1970KYa (12308) 206
K1=0.25(I=1), K1=0.09, B2=-0.8(I=3); K1=0.30(I=5)
In LiClO4: K1=0.09, B2=-0.82(I=3); K1=-0.08(I=5)). At I=0, K1=0.69
_____
   oth none 25°C 0.0 U K1=0.48 1962LIc (12309) 207
______
Tl+ kin none 25°C 0.0 U K1=0.85 1956BPa (12310) 208
      sol none 25°C 0.0 U T H K1=0.82 1953BGb (12311) 209
DH(K1)=1.5 kJ mol-1,DS=21 J K-1 mol-1. K1=0.81(0 C), 0.85(40 C)
_____
      kin oth/un 25°C 0.08M U I K1=0.22 1949BPb (12312) 210
Medium: 0.08 to 0.25 M. At I=0: K1=0.42. By conductivity, I=0, K1=0.49
********************************
                Phosphate CAS 7664-38-2 (176)
P04---
            H3L
Phosphate;
       Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp NaCl04 25°C 0.15M U K1=2.41 1971MMg (13348) 211 K(Tl+HL)=0.73
H4L
P207----
                Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from (HO)2PO.O.PO(OH)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp NaClO4 25°C 1.30M U K1=2.20 B2=3.40 1984FEa (13663) 212
                       K1=3.05 1971MMg (13664) 213
Tl+ sp NaClO4 25°C 0.15M U
                       K(T1+HL)=2.34
Tl+ vlt KNO3 35°C 2.00M U K1=1.69 B2=1.9 1952SDa (13665) 214
H5L
                          CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sol oth/un 20°C 2.40M U K1=1.3 B2=2.3 1974FEa (13914) 215
Medium:Na2SO4
*********************************
Re04-
                Perrhenate
                           (2581)
Rhenate(VII), Perrhenate;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
             ------
Tl+ sol none 25°C 0.0 C
                                 1988HHb (14111) 216
                        Kso(T1ReO4) = -4.92
Method: perrhenate ion selective electrode.
```

```
******************************
S--
             H2L Sulfide
                           CAS 7783-06-4 (705)
Sulfide:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
T1+
      oth none ? 0 U
                                     1990DKa (14485) 217
                           *Ks(T12S+H=2T1+HS)=-7.22
From recalculation of literature data.
Tl+ oth none 25°C 0 U
                                      1988LIa (14486) 218
                           Kso(T12S) = -24.5
                           *Kso(T12S) = -7.2
Derived from thermodynamic data and K(H+S=HS)=17.3.
_____
  ISE NaClO4 25°C 1.0M U
T1+
                                     1972GRa (14487) 219
                           K(T1+HL)=2.27
                           K(2T1+HL)=8.04
                           K(2T1+0H+3HL)=14.96
                           K(2T1+20H+2HL)=16.7
Kso=-21.15
      sol NaClO4 25°C 3.0M U
                                      1966GKc (14488) 220
                           *Kso(.5T12S(s))=1.36
-----
      vlt none 25°C 0.0 U
                                      1959KKa (14489) 221
                           Kso(T12L) = -20.0
I=0 corr. K(0.5T12L(s)+H=T1+0.5H2S(g))=0.46
Tl+ oth none 25°C 0.0 U
                                     1952GGc (14490) 222
                          Kso(T12L) = -19.15
From thermodynamic data
______
Tl+ oth none 25°C 0.0 U
                                     1952LAb (14491) 223
                           Kso(Tl2L)=-21
From thermodynamic data
Tl+ sol none 20°C 0.0 U
                                     1936RAa (14492) 224
                           Kso(T12L) = -22.19
I=0 corr. K(0.5T12L(s)+H=T1+0.5H2S(g))=0.37
_____
                    sol oth/un 18°C var U
                                      1931KOa (14493) 225
                           Kso(T12L) = -22.16
At 20 C: Kso=-23.92, K(0.5T12L+H=T1+0.5H2S(g))=-0.48
______
Tl+ sol oth/un 25°C var U T
                                     1909BZa (14494) 226
                           K = 0.41
                           Kso(T12L) = -22.15
K: K(0.5T12L(s)+H=T1+0.5H2L). K=-0.17(0 C), 0.69(40 C). Kso=-22.35(18 C)
```

```
SCN-
                   Thiocyanate CAS 463-56-9 (106)
               HL
Thiocyanate;
_____
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
------
Tl+ sp non-aq 25°C 100% C K1=0.15 1998AEa (15281) 227
Medium: N,N-Dimethylthioformamide. Methods: IR and FT Raman spectroscopy.
Ligand is S-bonded (thiocyanate). For N-bonding (isothiocyanate), K1=-0.52
______
      oth NaClO4 25°C 3.0M U I R K1=0.10
                                   1997BPa (15282) 228
IUPAC evaluation
______
T1+
      sol none 25°C 0.0 C
                                     1975PTe (15283) 229
                          Kso(T1SCN) = -3.74
Method: SCN ion selective electrode. Data for 10-40 C.
______
      ISE NaCl04 25°C 3.0M U I K1=0.08 B2=-0.08 1972FIb (15284) 230
T1+
                           B3 = -0.96
                           B4 = -1.22
Medium: LiClO4; K1=0.15, B2=-0.06, B3=-0.39(I=1); K1=0.12, B2=-0.11, B3=-0.47(I=2);
K1=0.13, B2=-0.03, B3=-0.50, B4=-1.4(I=4). Method: Tl amalgam electrode
_____
      oth none 25°C 0.0 U
                          K1=0.56 B2=0.37 1972FIb (15285) 231
T1+
                      B3=-0.30
-----
Tl+ ix oth/un 25°C var U
                          K1=0.46 B2=0.92 1971BSj (15286) 232
                          B3=0.30
                          B4 = 0.40
Tl+ ISE NaCl04 25°C 4.0M U I T K1=0.15 B2=-0.06 1971FRb (15287) 233
                          B3 = -0.42
                           B4 = -1.4
Medium: LiClO4; K1=0.17, B2=-0.05, B3=-0.36(I=1); 0.12, -0.12, -0.47(I=2);
K1=0.10, B2=-0.10, B3=-0.55, B4=-1.40(I=3). Method: Tl amalgam electrode
______
                         T K1=0.58 B2=0.36 1971FRb (15288) 234
       ISE none 25°C 0.0 U
T1+
                          B3 = -0.5
Medium: LiClO4, extrapolated to zero conc. Method: Tl amalgam electrode
     -----
                                1971FRb (15289) 235
Tl+ ISE NaClO4 25°C 1.0M U I M
                           B(T1C1L) = -0.05
                           B(T1C1L2) = -0.09
Medium: LiCl04. B(TlClL)=-0.15(I=0.5), -0.22(I=2), -0.15(I=3), -0.17(I=4); I=0
corr: 0.26. B(TICIL2) = -0.27(I=2), -0.35(I=3), -0.68(I=4); I=0 corr: 0.18. TI/Hg
______
      vlt KNO3 25°C 2.50M U K1=0.19
B3=-0.44
                           K1=0.19 B2=-0.04 19660La (15290) 236
T1+
_____
Tl+ sol NaCl04 25°C 4.0M U T K1=0.20 B2=-0.05 1965KMa (15291) 237
                          B3 = -0.58
```

B4=-0.80 K(K+T1L4)=-0.1

M = -1	K(K+11L4)=-0.1	
Medium: L	L1C104	
T1+	ISE NaClO4 25°C 3.0M U I T K1=0.11 B2=-0.06 1962K B3=-0.43 B4=-1.35	Cb (15292) 238
	LiClO4. By solubility K1=0.19, B2=-0.03, B3=-0.43, B4=-1.44 , 3 M LiClO4: K1=0.31, B2=0.35, B3=0.13. Also in 20, 40, 60	
T1+	sol NaClO4 20°C 4.60M U T B2=0.49 1961GSb (1 B3=0.17 Kso(TlL)=-3.27 K(TlL(s)=TlL)=-2.61 K(TlL(s)+L=TlL2)=-2.72	5293) 239
Kso(TlL)=	Tl/Hg electrode. K(TlL(s)+2L=TlL3)=-3.10, At 40 C: B2=0.41, =-2.72, K(TlL(s)+L=TlL2)=-2.32, K(TlL(s)+2L=TlL3)=-2.77	
Tl+ K1=0.53(I	oth KNO3 25°C 2.0M U I K1=0.49 1961PRa (1 I=3)	5294) 240
Medium: L	sol oth/un 25°C var U I K1=0.62 B2=0.57 1960K B3=0.13 B4=-0.34 LiL. In NaL K1=0.66, B2=0.60, B3=0.18, B4=-0.39. In KL: 0.7	Mb (15295) 241 4, 0.58,
0.25,-0.3	32. In CsL: 0.68,0.68,0.22,-0.48. In 8 M NaClO4:0.42,0.36,0	.01,-0.53
Tl+	sol none 25°C 0.0 U K1=0.85 1958BCa (1 K(TlL(s)=Tl+L)=-3.80 K(TlL(s)=TlL)=-2.94 B3=0.62	·
T1+ B6=0.46	vlt NaClO4 25°C 3.0M U T K1=0.64 B2=0.88 1958H K3=-0.18 K4=0.03 K5=-0.21 K6=-0.06	
T1+	sol oth/un 25°C dil U 1958MIb (1 Kso=-3.77	5298) 244
T1+	vlt KNO3 25°C 3.0M U T K1=0.19 B2=-0.20 1958P K3=-0.42 K4=-0.34 B4=-0.96	Da (15299) 245
T1+	ISE NaClO4 25°C 4.0M U K1=0.15 B2=0.00 1957N K3=-0.46 K4=-0.46	Ia (15300) 246

```
K(TlL(s)=Tl+L)=-3.15
______
Tl+ sol NaClO4 25°C 4.0M U
                                1957NIa (15301) 247
                       K(TlL(s)=Tl+L)=-3.16
                       K(TlL(s)=TlL)=-3.00
                       K(TlL(s)+L=TlL2)=-3.15
                       K(T1L(s)+2L=T1L3)=-3.60
K(TlL(s)+3L=TlL4)=-3.97, K1=0.15, K2=-0.15, K3=-0.48, K4=-0.37
______
    vlt NaNO3 25°C 2.0M U K1=0.42 1956LSa (15302) 248
_____
     sol none 30°C 0.0 U
Tl+
                                1956SSb (15303) 249
                       K(T1L(s)=T1+L)=-3.64
Additional method: polarography
      sol none 25°C 0.0 U T H K1=0.80 1953BGb (15304) 250
DH(K1)=-12.38 \text{ kJ mol-1}, DS(K1)=-26.8(25 \text{ C}); K1=0.94(0 \text{ C}), 0.64(40 \text{ C})
             _____
Tl+ ISE none 25°C 0.0 U
                                1952SUa (15305) 251
                       K(T1L(s)=T1+L)=-3.77
______
     oth oth/un 20°C dil U
                               1903B0b (15306) 252
                       K(T1L(s)=T1+L)=-3.92
*****************************
           H2L Sulfite CAS 7782-99-2 (801)
S03--
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+
      oth oth/un 20°C 3.00M U K1=0.02
                                1979FEa (15478) 253
Method: densimetry
______
     oth NaClO4 20°C 3.0M U K1=-0.02 1979FEb (15479) 254
Method: densitometry
S04--
            H2L
                Sulfate
                         CAS 7664-93-9 (15)
Sulfate:
       Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     vlt oth/un 25°C 0.20M C B2=1.90
                             1975APd (16596) 255
Method: polarography. Medium: 0.20 M Na2SO4.
______
     cal oth/un 25°C 0.5M U K1=0.65 1975FRa (16597) 256
Background salt: LiClO4; For: I=1.0 M, K1=0.34;
     ISE oth/un 25°C 3.0M U K1=-0.48 B2=-0.89 1965KYd (16598) 257
Method:amalgam electrode. Medium: LiClO3. K1=-0.52 in summary
______
     vlt NaClO4 25°C 2.0M U K1=0.34
T1+
                               1962BSc (16599) 258
```

T1+	EMF oth/un	25°C 0.0			1960CRa (16600) 259
T1+		25°C 3.68			1957BMa (16601) 260
)=-1 kJ mc	ol-1, DS=23 J K-	
			U	K1=1.44	1930BDa (16603) 262
Tl+	con oth/un	18°C 0.0	U	K1=1.33	1930RDa (16604) 263
Tl+	con oth/un *******	18°C var ******* H2L Th	U ******* iosulfate	K1=1.85	1920DRa (16605) 264 *********** 28-7 (177)
Metal	Mtd Medium				Reference ExptNo
Method: de	ensimetry				.11 1979FEa (16902)
Tl+ Method: de	oth NaClO4 ensitometry	20°C 2.7	4 U	K1=0.79 B2=	1.13 1979FEb (16903)
			4 U		.94 1977PGa (16904)
			1 U		.72 1958NIa (16905)
				Kso(T12L)=-6.70	
T1+	vlt none			K1=1.91	1954NRb (16907) 270
Tl+ ******** Se Selenide;	ISE oth/un ********		******		1904EUa (16908) 271 *********
Metal	 Mtd Medium	Temp Conc	Cal Flags	lg K values	Reference ExptNo
T1+		25°C 0.0		-0	1964BUe (16950) 272
				Kso=-33.1	**************************************
SeCN- Selenocyar				e CAS 73102-	

	Mtd Medium	Temp C	onc Cal	Flags	Lg K val	ues	Refer	rence	ExptNo	
	vlt NaNO3			В:	3=1.12					
Se04 Selenate;	******		Selenat						*****	
Metal	Mtd Medium	Temp C	onc Cal	Flags	Lg K val	ues	Refer	ence	ExptNo	
I=0 corr. DH(so)=43.1	sol none Kso=-4.40(10 1 kJ mol-1 ********	0 C), -	4.13(20	K: C), -3	so(Tl2L) .88(30 (=-4.00 (2), -3.6			·	
V04	VO2(OH)3 (H3L					'5-7 (15			
Metal	Mtd Medium	Temp C	onc Cal	_	Lg K val		Refer			
	sol oth/un			K: K:	s(T1(V03 s(T14(V2	3))=-8.2 207))=-1	1964SMb 6 .8.59	(1739	3) 275	
CH4N2S	********** ide, Thiour	L	Thioure					****	*****	
Metal	Mtd Medium	Temp C	onc Cal	Flags	Lg K val	ues	Refer	rence	ExptNo	
	Mtd Medium									
T1+ T1+	con NaClO4 vlt KNO3	25°C 1	.00M U	 	K1=0.3		1998GZa	 (1786	1) 276	277
T1+ T1+ Method: po: ************************************	con NaClO4 vlt KNO3 larography.	25°C 1 25°C ******	.00M U 1.5M C		K1=0.3 K1=0.60 3=1.68	B2= 1	1998GZa 15 197	 (1786 '8DKb	1) 276 (17862)	
T1+ T1+ Method: po. ************************************	con NaClO4 vlt KNO3 larography. ********	25°C 1 25°C ****** H2L DH)2	.00M U 1.5M C ********	B: ******	K1=0.3 K1=0.60 3=1.68 *******	B2= 1 ******* 144-62-	1998GZa 15 197 ******** 7 (24)		1) 276 (17862) *****	
T1+ T1+ Method: po. ************************************	con NaClO4 vlt KNO3 larography. ******** c acid; (COC Mtd Medium vlt KNO3 captobenzoa	25°C 1 25°C ****** H2L DH)2 Temp C 30°C 1 te)L) =	.00M U 1.5M C ****** Oxalic onc Cal00M U 5.22	******* acid Flags	K1=0.3 K1=0.60 K1=0.60 3=1.68 ************************************	B2= 1 ****** 144-62	1998GZa 15 197 ******** 7 (24) Refer 1982GSa	(1786 78DKb ****** Pence (1910	1) 276 (17862) ****** ExptNo 7) 278	
T1+ T1+ Method: poi ************************************	con NaClO4 vlt KNO3 larography. ******** c acid; (COC Mtd Medium vlt KNO3 captobenzoa	25°C 1 25°C ****** H2L OH)2 Temp C 30°C 1 te)L) = 25°C 0	.00M U 1.5M C ******* Oxalic onc Cal00M U 5.22	#***** acid Flags	K1=0.3 K1=0.60 K1=0.60 3=1.68 ************************************	B2= 1 ****** 144-62- ues	1998GZa15 197 ******* 7 (24) Refer 1982GSa	(1786 (1786 (****** (1910	1) 276 (17862) ****** ExptNo 7) 278	
T1+ Method: po: ************************************	con NaClO4 vlt KNO3 larography. ******** c acid; (COC Mtd Medium vlt KNO3 captobenzoa gl NaClO4 vlt NaClO4	25°C 1 25°C ****** H2L DH)2 Temp C 30°C 1 te)L) = 25°C 0	.00M U 1.5M C ******* Oxalic onc Cal00M U 5.2215M U	#****** acid Flags	K1=0.3 K1=0.60 K1=0.60 3=1.68 ************************************	B2= 1 ****** 144-62- ues	1998GZa15 197 ******* 7 (24) Refer 1982GSa 1971MMg	(1786 (1786 (****** ****** (1910 (1910	1) 276 (17862) ****** ExptNo 7) 278 8) 279	

C2H4O2 Ethanoic a	cid; CH3.		Acetic	acid	CAS 64-1	.9-7 (36)
Metal	Mtd Medi	ım Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
	*****	****** HL	******* Glycin	*****		1937RDa (20204) 282 ***********************************
Metal	Mtd Medi	ım Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
Medium: Li	.C104					1974KUc (21735) 283
C2H8N2 1,2-Diamin	oethane; I	L H2N.CH2	•	nediami	ine CAS 107-	15-7 (23)
Metal	Mtd Medi	ım Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
By spectro	photometry	/ K1=0.	3			1928JOa (23237) 284
C3H4O4 Propanedio		H2L	Maloni		CAS 141-	
Metal	Mtd Medi	ım Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
T1+	sp NaCl(04 25°C ******	0.15M U	*****	K1=0.54 ************************************	
T1+ ******** C3H6O2S 3-Mercapto Metal	sp NaClo	04 25°C ******* H2L acid; I	0.15M U ******** HS.CH2.CH	****** 2.COOH Flags	K1=0.54 ************************************	1971MMg (24571) 285 ************ 96-0 (437) Reference ExptNo
T1+ ******** C3H6O2S 3-Mercapto Metal T1+ K1(30 C)=2	sp NaClo	H2L acid; I acid; I um Temp 20°C	0.15M U ******* HS.CH2.CH Conc Cal 0.10M U	****** 2.COOH Flags 	K1=0.54 ********* CAS 107- Lg K values K1=2.78	1971MMg (24571) 285 ************ *96-0 (437) Reference ExptNo 1968SGd (25230) 286
T1+ ******** C3H6O2S 3-Mercapto Metal T1+ K1(30 C)=2	sp NaClo ******** propanoic Mtd Medio EMF KNO3 *********	D4 25°C ******* H2L acid; I um Temp 20°C O C)=2.9 *******	0.15M U ******* HS.CH2.CH Conc Cal O.10M U 94 ********	******* 2.COOH Flags T ******	K1=0.54 ********* CAS 107- Lg K values K1=2.78	1971MMg (24571) 285 ************ 96-0 (437) Reference ExptNo 1968SGd (25230) 286
T1+ ******** C3H6O2S 3-Mercapto Metal T1+ K1(30 C)=2 ********* C3H7NO2 2-Aminopro	sp NaClo ******** propanoic Mtd Medio EMF KNO3 .85, K1(40 ********	H2L acid; I acid; I 20°C 0 C)=2.9 ********	0.15M U ******* HS.CH2.CH Conc Cal 0.10M U 94 ****** Alanin CH(CH3).	****** 2.COOH Flags T ******	K1=0.54 ********** CAS 107- Lg K values K1=2.78 ************* CAS 56-4	1971MMg (24571) 285 ************ 96-0 (437) Reference ExptNo 1968SGd (25230) 286
T1+ ******** C3H6O2S 3-Mercapto Metal T1+ K1(30 C)=2 ******** C3H7NO2 2-Aminopro Metal T1+ Medium: Li	sp NaClo ******** propanoic Mtd Medio EMF KNO3 .85, K1(40 ******* panoic ac: Mtd Medio gl NaClo Clo4	04 25°C ****** H2L acid; I um Temp 20°C 0 C)=2.9 ******* HL id; H2N um Temp	0.15M U ******* HS.CH2.CH Conc Cal 0.10M U 94 ****** Alanin CH(CH3).	******* 2.COOH Flags ****** e COOH Flags T	K1=0.54 ********* CAS 107- Lg K values K1=2.78 ********* CAS 56-4 Lg K values	1971MMg (24571) 285 *********** 96-0 (437) Reference ExptNo 1968SGd (25230) 286 ***********************************
Tl+ ******** C3H6O2S 3-Mercapto Metal Tl+ K1(30 C)=2 ********* C3H7NO2 2-Aminopro Metal Tl+ Medium: Li ********** C3H7NO2S	sp NaClo ******** propanoic Mtd Medio EMF KNO3 ******** panoic ac: Mtd Medio gl NaClo Clo4 ********	H2L acid; I ac	0.15M U ******* HS.CH2.CH Conc Cal 0.10M U 94 ******* Alanin CH(CH3).0 Conc Cal 0.10M U ******** Conc Cal	******* 2.COOH Flags ******* COOH Flags T	K1=0.54 ********* CAS 107- Lg K values K1=2.78 ********* CAS 56-4 Lg K values	1971MMg (24571) 285 ***********************************

```
gl NaCl 37°C 0.15M C
                        K1=3.26
T1+
                                  1989BCa (26844) 288
                        B(T1HL)=11.28
**********************************
              HL
                 Serine
                           CAS 56-45-1 (49)
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH
Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl NaClO4 25°C 0.10M U K1=1.53
                                  1974KUc (27187) 289
Medium: LiClO4
************************************
                           CAS 128-04-1 (2125)
Dimethyldithiocarbamic acid; (CH3)2N.CSSH
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      EMF non-aq 25°C 100% U B2=6.4
                                  1987USa (27279) 290
Medium: DMF, 0.1 M LiClO4
**********************************
                 Thiomalic acid CAS 70-49-5 (109)
             H3L
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
            30°C 0.10M U T H K1=3.58 1968SGa (30368) 291
      gl KNO3
K1(35C)=3.71, K1(40C)=3.78. DH=-24.0 kJ mol-1, DS=-7.1 J K-1 mol-1
*********************************
                 L-Tartaric acid CAS 87-69-4 (92)
             H2L
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      con oth/un 28°C ? U K1=1.39 1965SBa (31376) 292
*********************************
C4H7N04
                 Aspartic acid
                           CAS 56-84-8 (21)
             H2L
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl oth/un 25°C 0.1M U K1=2.5 B2= 4.00 1975KUb (31955) 293
In 0.1 M LiClO4
*********************************
             H2L
                 IDA
                           CAS 142-73-4 (118)
Iminodiethanoic acid; HN(CH2.COOH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ vlt oth/un 25°C 0.30M U K1=1.32 1970FUb (32379) 294
*********************************
```

C4H8N2O4 H2L CAS 39156-77-9 (3008) Hydrazine-N,N-diethanoic acid; H2N.N(CH2.COOH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ sp oth/un 20°C ? U K1=11.58 1972KVa (33115) 295 K(Tl+HL)=5.54

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ vlt alc/w 20°C 40% U T K1=1.74 B2=2.00 1972SCf (33368) 296 B3=3.25
Medium: 40% EtOH, 0.5 M NaClO4. 30 C: K1=1.70, B2=1.95, B3=3.20 ************************************
C4H9NO3 HL Threonine CAS 72-19-5 (48) 2-Amino-3-hydroxybutanoic acid; H2N.CH(CH(OH).CH3)COOH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ vlt NaClO4 25°C 0.10M C K1=0.90 B2= 1.95 1986SPb (34330) 297 Method: polarography. ************************************
C4H10N2O2 HL EDMA (2784) Diaminoethane-N-ethanoic acid; H2N.CH2.CH2.NH.CH2.COOH
Diaminoethane-N-ethanoic acid; H2N.CH2.CH2.NH.CH2.COOH
Diaminoethane-N-ethanoic acid; H2N.CH2.CH2.NH.CH2.COOH Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo Tl+ vlt NaClO4 25°C 0.30M U K1=1.30 1970FUb (34594) 298 ***********************************
Diaminoethane-N-ethanoic acid; H2N.CH2.CH2.NH.CH2.COOH
Diaminoethane-N-ethanoic acid; H2N.CH2.CH2.NH.CH2.COOH
Diaminoethane-N-ethanoic acid; H2N.CH2.CH2.NH.CH2.COOH

```
N-Acetylcysteine; CH3.CO.NH.CH(CH2.SH)COOH
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl NaCl 37°C 0.15M C K1=2.27 1989BCa (38818) 302
********************************
       H2L Glutamic acid CAS 56-86-0 (22)
C5H9N04
2-Aminopentanedioic acid; H2N.CH(CH2.CH2.COOH)COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ gl oth/un 25°C 0.1M U K1=2.35 B2= 4.05 1975KUb (39132) 303
In 0.1 M LiClO4
***********************************
C5H11N02S
            H2L Penicillamine CAS 52-66-4 (350)
DL-2-Amino-3-mercapto-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl NaCl 37°C 0.15M C K1=3.58 1989BCa (41284) 304 B(TlHL)=12.00
C5H11NS2
                         CAS 147-84-2 (2126)
Diethyldithiocarbamic acid; (CH3.CH2)2N.CSSH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     EMF non-aq 25°C 100% U B2=6.5
                              1987USa (41372) 305
Medium: DMF, 0.1 M LiClO4
______
     sp non-aq ? 100% U M
T1+
                               1968SRg (41373) 306
                    K(T1HA+HL=T1L+H2A)=3.53
Medium: CCl4. H2A=dithizone
*********************************
      H2L
               Ribose-5-phosph CAS 4300-28-1 (2756)
Ribose-5-phosphoric acid, Ribofuranoside 5 Phosphoric acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp NaCl04 25°C 0.15M U K1=0.87 1971MMg (41426) 307
*******************************
           H2L Catechol
                        CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH
-----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                Reference ExptNo
______
Tl+ gl KCl ? 0.10M U K1=7.05 1966TBa (43852) 308
*************************
               Citric acid CAS 77-92-9 (95)
           H3L
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH
```

Metal	Mtd Mediu	m Temp Conc Cal Fla	gs Lg K values	Reference ExptNo
Tl+	sp NaClO	4 25°C 0.15M U	K1=1.36	. ,
T1+	con oth/u	n 28°C ? U	K1=2.82	1965SBa (46284) 310
K1=0.97(I=	=0.3 M), 0.0	65(I=0.5)		1956SAb (46285) 311
C6H9N06		H3L partic acid;		5-84-1 (4367)
Metal	Mtd Mediu	m Temp Conc Cal Fla	gs Lg K values	Reference ExptNo
			K(T1+HL)=1.42	= 6.29 2005SNa (46381) 312
C6H9N06			CAS 139-:	**************************************
Metal	Mtd Mediu	m Temp Conc Cal Fla	gs Lg K values	Reference ExptNo
Tl+ IUPAC eval	•	20°C 0.10M C TIH	R K1=4.74	1982ANa (47054) 313
T1+	sp NaClO	4 25°C 0.15M U	K1=4.42	1971MMg (47055) 314
T1+	ix oth/u	n ? ? U	K1=5.00	1969KKf (47056) 315
T1+	gl KNO3	20°C 0.10M U		1967ABc (47057) 316
Tl+ Medium: Me	•	20°C 0.10M U	T K1=4.74	1963IFb (47058) 317
**************************************	vlt oth/u ********* niopropanoi	**************************************	**************************************	
Metal	Mtd Mediu	•	gs Lg K values	Reference ExptNo
Tl+ By polarog	gl oth/u graphy, B2=!	5.40	K1=2.75 B2	=5.42 1984SGa (48271) 319 ********
C6H12O2S		HL acid; CH3.(CH2)3.S.	CAS 2060	∂-61-7 (4375)
Metal	Mtd Mediu	m Temp Conc Cal Fla	gs Lg K values	Reference ExptNo

```
T1+
      vlt alc/w 30°C 50% U T H K1=1.25
                                B2=1.62
                                      1976SSg (49446) 320
                         B3=2.41
Medium: 50% EtOH, 0.1 M. At 40 C: K1=1.23, B2=1.60, B3=2.40
**********************************
                            CAS 56-87-1 (41)
C6H14N2O2
                  Lysine
2,6-Diaminohexanoic acid; H2N.(CH2)4.CH(NH2)COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      vlt NaClO4 30°C 0.10M C T H
                                   1983SDb (50838) 321
                         K(T1+HL)=1.64
                         K(T1+2HL)=3.07
Method: polarography. Medium pH 8.0. At 40 C, K(Tl+HL)=1.55,
K(Tl+2HL)=3.01. DH(Tl+HL)=-14.6 kJ mol-1, DH(Tl+2HL)=-10.7.
***********************************
C6H1502PS2
              HL
                              (2059)
0,0'-Dipropyl dithiophosphoric acid; (C3H7O)2P(S)SH
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp alc/w 25°C 100% U I K1=4.63
                                   1979SJd (51491) 322
Medium: = methanol; log K1 in other solvents: acetonitrile 4.35,
dioxan 5.16, tetrahydrofuran 6.38
C7H5N04
             H2L Dipicolinic aci CAS 449-83-2 (418)
2,6-Pyridinedicarboxylic acid; C5H3N.(C0OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      vlt NaCl04 25°C 0.5M C TI K1=2.33 1983PBa (52813) 323
T1+
Method: polarography. Also data for 15 C and 10% MeOH/H2O.
*****************************
C7H5O3Br
                            CAS 85-55-4 (1194)
5-Bromosalicylic acid; Br.C6H3(OH).COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl alc/w 30°C 50% M K1=8.28 B2=13.48 1978KDb (53311) 324
Medium: 50% v/v EtOH/H2O, 0.10 M NaClO4.
**********************************
                            CAS 321-14-2 (1113)
C7H5O3C1
5-Chlorosalicylic acid; Cl.C6H3(OH).COOH
  Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl alc/w 30°C 50% M
                          K1=8.55 B2=13.70 1978KDb (53347) 325
T1+
Medium: 50% v/v EtOH/H2O, 0.10 M NaClO4.
**********************************
                            CAS 119-30-2 (1114)
2-Hydroxy-5-iodobenzoic acid, 5-Iodosalicylic acid; I.C6H3(OH).COOH
```

Metal	Mtd Medium Temp Conc Cal Flags Lg K valu	es Reference ExptNo
Medium: 50	gl alc/w 30°C 50% M K1=8.00 0% v/v EtOH/H2O, 0.10 M NaClO4. ************************************	· · ·
C7H602S	H2L Thiosalicylic CAS 1-obenzoic acid; HS.C6H4.COOH	
Metal	Mtd Medium Temp Conc Cal Flags Lg K valu	es Reference ExptNo
Tl+	gl alc/w 25°C 50% U K1=4.68	1971RFa (53919) 327
Medium: 50	gl alc/w 17°C 50% U K1=3.66 0% EtOH, 0.05 M NaClO4 ************************************	·
C7H6O3	H2L Salicylic acid CAS 6 benzoic acid, Salicylic acid; H0.C6H4.COOH	9-72-7 (14)
Metal	Mtd Medium Temp Conc Cal Flags Lg K valu	es Reference ExptNo
	gl alc/w 17°C 50% U K1=3.90 0% EtOH, 0.05 M NaClO4	1970RBc (54313) 329
******** C7H6O4	con oth/un 28°C	********
Metal	Mtd Medium Temp Conc Cal Flags Lg K valu	es Reference ExptNo
	con oth/un 28°C ? U K(T1+HL)=2	• •
C7H605		49-91-7 (446)
Metal	Mtd Medium Temp Conc Cal Flags Lg K valu	es Reference ExptNo
	con oth/un 28°C ? U K(T1+H2L)= ************************************	
C7H606S		965-83-3 (399)
Metal	Mtd Medium Temp Conc Cal Flags Lg K valu	es Reference ExptNo
	con oth/un 25°C 0.01M U K(T1+HL=T1 ************************************	1962SSb (55058) 333 L+H)=2.38(?) ********

```
CAS 59-53-0 (1269)
C7H13N03S
            H2L
N-Acetyl-penicillamine; CH3.CO.NH.CH(COOH)C(CH3)2SH
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ gl NaCl 37°C 0.15M C K1=2.45 1989BCa (57493) 334
*********************************
                TTA
                         CAS 326-91-0 (165)
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ dis oth/un ? ? U K1=0.65 1969KKf (58687) 335
*********************************
C8H9N02S
             HL
                         CAS 6310-11-8 (4576)
3-Mercaptoacetamidophenol; HS.CH2.CO.NH.C6H4.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
   gl oth/un 17°C ? U K1=3.46 1973KPd (60384) 336
______
Tl+ oth alc/w 20°C 50% U K1=3.45 1972KPe (60385) 337
Medium: 50% v/v EtOH, 0.1 M NaClO4
*********************************
         H2L Uramildiacetic CAS 13055-06-5 (185)
C8H9N307
5-Amino-2,4,6-trioxo-1,3-perhydrodiazimino-N,N-diethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     cal R4N.X 20°C 0.1M C
                                1976ANb (60659) 338
                       DH1= -24.2 kJ/mol
in Me4NCl
______
      gl KNO3 39°C 0.10M U TIH K1=5.33 1963IFb (60660) 339
K1=5.99(20 C),5.76(27 C),5.41(34 C). DH(K1)=-64.4 kJ mol-1, DS=104 J K-1 m-1
At I=0 corr:K1=6.70(20 C)
***********************************
C8H11N08
                          CAS 24868-49-3 (2572)
2-Amino(N,N-diethanoic)-1,4-butanedioic acid;HOOCCH(N(CH2COOH)2)CH2COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KNO3 25°C 0.1M U K1=4.38 B2= 6.11 2005SNa (61187) 340 K(Tl+HL)=3.01
T1+
CAS 35039-85-1 (4537)
C8H12N2O8
1,2-Diaminoethane-N,N'-dimalonic acid; (HOOC)2.CH.NH.CH2.CH2.NH.CH(COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

Tl+	vlt KNO3	25°C 0.10M U	K1=3.75 K(T1+HL)=2.48	1973GSd (61527) 341		
			K(Tl+HL)=2.22 B(Tl2L)=1.0	1972KGc (61528) 342		
C8H12O4			CAS 1687-3	**************************************		
Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo		
Tl+ con oth/un 28°C ? U K1=1.74 1966SBa (61703) 343 ***********************************						
Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo		
At 30 C: H DS=57. DH	(1=1.77, B2= (B2)=29.3; [3.342. At 40 C: K1 S=159.	=1.87, B2=3.398.	3.176 1987GRb (62062) 344 DH(K1)=6.8 kJ mol-1		

Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo		
	vlt R4N.X .2 M Bu4NPF6		K1=9.1	1999BBc (62730) 345		
		25°C 100% C I for acetonitrile	K1=3.12	1993JHa (62731) 346		
	opylene car		K1=3.71	1982MDa (62732) 347		

Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo		
				1979SJd (63160) 348 *******		
C9H7NO HL Oxine CAS 148-24-3 (504) 8-Hydroxyquinoline (8-quinolinol);						
Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo		
T1+	dis oth/ur	, , , ,	K1=2.34	1969KKf (64360) 349		

```
******************************
C9H7N03S2
                        CAS 58447-10-2 (4675)
8-Mercaptoquinoline-5-sulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     EMF oth/un ? ? U K1=4.6
                          1968ABa (64431) 350
TAR
C9H7N3O2S
           H2L
                        CAS 2246-46-0 (707)
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values
______
Tl+ gl alc/w 25°C 50% U
                              1967NPb (64730) 351
                      K(T1+HL) < 3
Medium: 50% MeOH, 0.1 M NaClO4
*********************************
C9H11N05
           H2L
                        CAS 57362-11-5 (3876)
N-(2'-Furfuryl)iminodiethanoic acid; C4H3O.CH2.N(CH2.COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl KNO3 20°C 0.10M U K1=3.11 1963IFa (66452) 352
*********************************
C9H11N3O7
                          (3877)
N-(1-Methyl-2,4,6-trioxo-perhydropyrimidinyl)iminodiethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl R4N.X 20°C 0.10M U K1=5.79
                              1963IFb (66529) 353
Medium: Me4NNO3
**********************************
C9H14N2O9
                        CAS 56360-11-3 (2576)
2-Hydroxy-1,3-diaminopropane-N,N'-di(1,3-propanedioic acid)
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     EMF KNO3 25°C 0.10M U K1=3.29
T1+
                              1976DGc (67139) 354
                      K(T1+HL)=2.02
                      K(T1+T1L)=1.48
*********************************
C9H19NS2
                        CAS 150-11-8 (1154)
N,N-Di(n-butyl)dithiocarbamate; (C4H9)2N.CSSH
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     EMF non-aq 25°C 100% U B2=7.3 1987USa (67992) 355
Medium: DMF, 0.1 M LiClO4
**********************************
               2,2'-Bipyridyl CAS 366-18-7 (25)
C10H8N2
             L
```

2,2'-Bipyridine; (C5H4N)2							
Metal	Mtd Medium	n Temp Conc Cal Fla	gs Lg K values	Reference ExptNo			
(K1=2.63), ******	larography. EtOH/H2O (Medium pH 8.0. Al 2.70), dioxane/H20 *******	so values for 509 (2.92), ethyl ad *******	cetate/H2O (2.75). *********			
C10H11NO5 H3L CAS 100844-86-8 (2108) N-(2-Hydroxyphenyl)iminodiethanoic acid; H0.C6H4.N(CH2.COOH)2							
Metal	Mtd Medium	n Temp Conc Cal Fla	gs Lg K values	Reference ExptNo			
Tl+	gl KNO3		K1=4.79 K(T1+HL)=2.34	, ,			

Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo			
Tl+ gl KNO3 20°C 0.10M U K1=3.84 1963IFc (71277) 358 ************************************							
Metal	Mtd Medium	າ Temp Conc Cal Fla	gs Lg K values	Reference ExptNo			
Tl+ Medium: Me	4NN03		K1=5.73				

Metal	Mtd Medium	Temp Conc Cal Fla	gs Lg K values	Reference ExptNo			
Tl+ IUPAC eval	uation	25°C 0.10M C		1991SMa (73019) 360			
C10H16N2O8 1,2-Diamin	sp NaClO4 ******** oethane-N,N	25°C 0.15M U ************************************	K1=1.32 ************************************	1971MMg (73020) 361 ************************************			
Metal	Mtd Medium	n Temp Conc Cal Fla	gs Lg K values	Reference ExptNo			
Tl+	vlt KNO3		K1=3.30 K(T1+HL)=2.33	1973GKc (73188) 362			
		25°C 0.10M U		1972KGc (73189) 363			

K(Tl+HL)=2.14

C10H16N2O8			H4L	EDTA	******	**************************************	**************************************
Metal	Mtd	Medium	Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
T1+	nmr	NaClO4	25°C	1.00M C		(T1L+CN)=8.72 (T1L+SCN)=2.7	
Tl+	cal	KNO3	20°C	0.1M C	D	H1= -36.6 kJ/	1976ANb (74233) 365 mol
Tl+	sp	NaClO4	25°C	0.15M U		1>5	1971MMg (74234) 366
T1+	ix	oth/un	?	? U		K1=6.11	1969KKf (74235) 367
Tl+	gl	KNO3	20°C	0.10M U	T	K1=6.53	1967ABc (74236) 368
T1+	gl	R4N.X	20°C	0.10M U		K1=6.55 (T1+HL)=2.06	1963IFb (74237) 369
Medium: Me ******			****	******	*****	*****	*******
C10H16N5O1 Adenosine-		riphosp		ATP acid;		CAS 56-65	-5 (403)
Metal	Mtd	Medium	Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
Tl+ IUPAC eval	_		25°C	0.10M C	Т	K1=2.5	1991SMa (74832) 370
C10H17NO4	****	******	***** H2L	******	*****		1971MMg (74833) 371 ************************************
Metal	Mtd	Medium	Temp	Conc Cal	Flags	Lg K values	Reference ExptNo
Tl+ gl KNO3 20°C 0.10M U K1=3.40 1963IFb (74978) 372 ************************************							
							Reference ExptNo
******	_		****			*******	1963IFb (74992) 373 *********
C10H17N05			H2L			(3917)	

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N-(Tetrahydropyran-2-ylmethyl)iminodiethanoic acid;
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl KNO3 20°C 0.10M U K1=4.06 1963IFa (75007) 374
**********************************
C10H2005 L 15-Crown-5 CAS 33100-27-5 (576)
1,4,7,10,13-Pentaoxacyclopentadecane; cyclo(-(0.CH2.CH2)5-)
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con mixed 25°C 90% C TIH K1=6.14 1998MTa (76138) 375
Medium: 90% CH3CN/H20. Data for 20-35 C. DH(K1)=23.2 kJ mol-1, DS(K1)=39.8
J K-1 mol-1. In 50% CH3CN/H2O, K1=4.99, DH(K1)=31.1, DS(K1)=8.8.
-----
Tl+ sp non-aq 20°C 100% C K1=3.91 1993PSc (76139) 376
Method: spectrofluorescence. Medium: MeOH.
______
Tl+ cal non-aq 25°C 100% C H K1=3.31 1986ICa (76140) 377
Medium: MeOH. DH(K1)=-36.4 kJ mol-1, DS(K1)=-60 J K-1 mol-1.
______
Tl+ vlt KNO3 25°C 0.10M C K1=2.63 1985KTb (76141) 378
Method: d.c. polarography. Medium: 0.10 M HNO3.
By a.c. polarography, K1=2.72
______
Tl+ ISE non-aq 25°C 100% U K1=5.29 B2=6.74 1982MDa (76142) 379
Medium: propylene carbonate
______
Tl+ oth oth/un 25°C ? U K1=1.23 1977RLa (76143) 380
Method: ultrasound absorption
______
     cal oth/un 25°C 0.10M U H T K1=1.23
                              1976ITb (76144) 381
DH=-16.8 kJ mol-1.
************************************
       L Cryptand 2,1 CAS 31249-95-3 (835)
C10H22N2O3
4,7,13-Trioxa-1,10-diazacyclopentadecane (Trioxa(2,1)cryptand);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl R4N.X 25°C 0.05M C K1=2.2
                               1997BCc (76342) 382
Medium: 0.05 M Me4NClO4
______
Tl+ sp non-aq 20°C 100% C K1=3.56 1993PSc (76343) 383
Method: spectrofluorescence. Medium: MeOH.
**********************************
                Tetraglyme CAS 143-24-8 (121)
C10H22O5
             L
2,5,8,11,14-Pentaoxapentadecane; (CH3.0.CH2.CH2.0.CH2.CH2.)20
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

Medium: Me ******	ОН	alc/w ******				*****					(76476 *****	•	
C10H23O2PS 0,0'-Dipen		dithiop	HL nospho	oric a	ncid	; (C5H:		AS 2253-54 P(S)SH	4-5 ((206:	1)		
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K	values	Re	efer	ence l	Expt	No
Tl+ ******** C10H24N4 1,4,8,11-T	****	*****	***** L	***** Cyc	**** :lam	*****	*****)	********* CAS 295-37	***** -4 (8	**** 3)	*****		
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K	values	Re	efer	ence I	Expt	No
Tl+ Medium: 0. ******	2 M I	Bu4NPF6						2.3				•	
C11H11NO6 N-(2'-Carb			H3L				C	AS 1147-6!	5-5 ((425	_		
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K	values	Re	efer	ence I	Expt	No
Medium: Me	4NNO:							93					
C11H13NO5 N-(2'-Meth			H2L	iethar	noic	acid;		AS 4596-54 C6H4.N(CH	,	•	5)		
C11H13N05	oxypl	henyl)ir	H2L minodi				CH30.	C6H4.N(CH	2.COOF	1)2	, 	 Expt∣	 No
C11H13NO5 N-(2'-Meth Metal T1+ Medium: Me	oxypl Mtd gl 4NNO:	henyl)ir Medium R4N.X 3	H2L minod: Temp 20°C	Conc 	Cal U	Flags	CH30. Lg K K1=2.	C6H4.N(CHi values 46	2.COOH Re 19631	i)2 efer EFb	 ence I (78603	3) 3	 88
C11H13NO5 N-(2'-Meth Metal Tl+	oxypl Mtd gl 4NNO: ****	henyl)ir Medium R4N.X 3 *****	H2L minod: Temp 20°C *****	Conc 0.10N	Cal 1 U	Flags	CH30. Lg K K1=2. *****	C6H4.N(CH2 values 46 ********	2.COOH Re 1963] *****	1)2 efero [Fb ****	, ence I (78603 *****	3) 38 ****	 88
C11H13N05 N-(2'-Meth Metal T1+ Medium: Me ************************************	oxypl Mtd gl 4NNO: ****	henyl)ir Medium R4N.X 3 ******	H2L minod: Temp 20°C ***** H4L N,N',N	Conc 0.10N *****	Cal (Cal (1 U (****	Flags ******	CH30. Lg K K1=2. ******	values 46 ********** AS 4408-83	2.COOH R6 19631 ***** (CH2)2N	H)2 efero [Fb **** (923		3) 38 ****	 88 ***
C11H13NO5 N-(2'-Meth Metal T1+ Medium: Me ******** C11H18N2O8 1,3-Diamin	oxypl Mtd gl 4NNO: **** oprop Mtd	henyl)ir Medium R4N.X 3 ******* pane-N,N	H2L minod: Temp 20°C ***** H4L N,N',N	Conc 0.10M ***** N'-tet	Cal '**** Trae Cal	Flags ***** thanoic Flags	CH30. Lg K K1=2. ****** C acid Lg K Lg K K1=3.	values 46 AS 4408-83 1; ((H00C.0) values	2.COOH R6 1963] ***** 1-5 (CH2)2N R6	1)2 efero (923 1.CH)	******) 2.)2.0	 3) 38 ***** CH2 Exptl	 88 *** No
C11H13N05 N-(2'-Meth Metal T1+ Medium: Me ********* C11H18N208 1,3-Diamin Metal	oxypl Mtd gl 4NNO: **** oprop Mtd gl ****	henyl)ir Medium R4N.X 3 ******* pane-N,N Medium KNO3	H2L minod: Temp 20°C ***** H4L N,N',N Temp 20°C *****	Conc 0.10M ****** N'-tet Conc 0.10M *****	Cal '*** Crae Cal Output Cal Cal Cal Cal	Flags ***** thanoic Flags Flags *****	CH30. Lg K ****** C acid Lg K Lg K (T1++ *****	C6H4.N(CH2 values 46 AS 4408-82 (H00C.0 values 90 (HL)=2.7 ************************************	2.COOH R6 1963] ****** 1-5 (CH2)2N R6 1967# ******	1)2 efero **** (923 N.CH: efero ABc	****** (78603 ******) 2.)2.0 ence I		 88 *** No 89
C11H13N05 N-(2'-Meth Metal T1+ Medium: Me ********* C11H18N208 1,3-Diamin Metal T1+ ************ C11H18N209	oxypl Mtd gl 4NNO: **** opro Mtd gl ****	henyl)ir Medium R4N.X ******* pane-N,N Medium KNO3 *******	H2L minod: Temp 20°C ***** H4L N,N',N Temp 20°C ***** H4L propar	Conc 0.10M ***** N'-tet Conc 0.10M ******	Cal '*** Crae Cal '' '' '' '' '' '' Cal	Flags ****** thanoic Flags ******	CH30. Lg K ****** C acid C acid ((T1++ ****** Coutane	C6H4.N(CH2 values 46 ********* CAS 4408-8: I; ((H00C.0) values 90 HL)=2.7 ********* CAS 668-21 edioic) ac:	2.COOH R6 1963] ****** 1-5 (CH2)2N 1967A ****** -1 (2id	i)2 efero (923 i.CH ABc *****	/ ence I (78603 ******) 2.)2.((79472 *****	3 ***** CH2 Exptl 2) 3	 88 *** No 89 ***

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***********************************
                 16-Crown-5 CAS 55477-28-8 (1592)
C11H22O5
              L
1,4,7,10,13-Pentaoxacyclohexadecane; cyclo(-(0.CH2.CH2)5.CH2.CH2-)
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
             25°C 0.0 C K1=0.73
      con none
                                  1991TKa (79876) 391
Self medium (ca. 0.008M).
_____
      dis none 25°C 0.0 C
                                   1989TKc (79877) 392
                         K(T1L+A=T1AL(org))=2.22
Method: extraction of metal picrate/L from H2O into benzene.
K(Tl+HA(org)+L(org)=TlAL(org)+H)=0.96. HA is picric acid.
***********************
             H4L
C12H20N208
                           CAS 40623-42-5 (1101)
1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH2NHCH(COOH)CH2CH2COOH)2
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
T1+
      vlt KNO3 25°C 0.10M U K1=2.40 1973GKc (82102) 393
                        K(T1+HL)=1.80
_____
      gl KNO3 25°C 0.10M U
                         K1=2.20
                                   1972KGc (82103) 394
                       K(Tl+HL)=1.66
*******************************
             H4L
                 TEDTA
                            CAS 923-74-0 (3394)
2,2'-Thiobis(ethyliminodiethanoic acid); S(CH2.CH2.N(CH2.COOH)2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                         K1=4.47
                                  1967ABc (82477) 395
      gl KNO3 20°C 0.10M U
                        K(T1+HL)=3.85
************************************
                           CAS 923-73-9 (2112)
             H4L
                 EEDTA
Oxa-bis(ethyleneimino)diethanoic acid; ((HOOC.CH2)2N.CH2.CH2)20
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                         K1=4.47
                                  1967ABc (82569) 396
      gl KNO3 20°C 0.10M U
                        K(T1+HL)=4.0
*********************************
                            CAS 57721-99-0 (2508)
1,14-Diacetamido-3,6,9,12-tetraoxatetradecane; (CH2.0.CH2.CH2.0.CH2.CH2.CO.NH2)2
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con alc/w 25°C 100% U K1=1.35 1975CJa (83054) 397
Medium: MeOH
**********************************
C12H24O4S2
                            CAS 296-39-9 (4938)
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1,4,10,13-Tetraoxa-7,16-dithiacyclooctadecane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ nmr non-aq 25°C 100% U M
                                   1981RPa (83144) 398
                    K(T1C104+L) > 5
Medium: MeNO2. K(TlClO4+L)=0 in DMSO; 1.24 in DMF; 2.98 in acetone;
>5 in MeCN: 0.93 in H2O
L 18-Crown-6 CAS 17455-13-9 (577)
1,4,7,10,13,16-Hexaoxacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ ISE alc/w 25°C 100% C IH R K1=5.27 2003ADa (83651) 399
IUPAC Recommended. Medium: 0-0.1 M various. DH(K1)=-44 kJ mol-1
In H20: K1=2.2, DH(K1)=-20. In PC: K1=7.13
______
Tl+ con non-aq 25°C 25% C TIH K1=3.98 2003RZa (83652) 400
Medium: 25 mol % MeOH/benzonitrile. Data for 15-55 C. DH(K1)=23 kJ mol-1
DS(K1)=155 J K mol-1. Data for 40, 50 and 75 mol %
______
Tl+ con non-aq 25°C 100% C T H K1=3.82 2001SKc (83653) 401
Medium: DMF. Data for 15-45 C. DH(K1)=-27.2 kJ mol-1,
DS(K1)=-18 J K-1 mol-1. Also data for 40-80% w/w DMF/acetonitrile.
_____
Tl+
      vlt mixed 20°C 0.02M U I K1=4.75 2000RCb (83654) 402
                         K1=1.71 in 100%H20
Medium: 0.025 M Et4NCl in 75.78 %mass CH3CN in H2O
For 0.025 M Et4NCl in 79.17% mass DMFA/H2O K1=3.06
______
     vlt mixed 20°C 78% U K1=1.31
                                   2000RCb (83655) 403
                         K1=1.71 in 100% H20
Medium: 0.025 M Et4NCl in 34.78% (mass) propanol in H2O.
for 0.025 M Et4NCl in 34.21% CH3CN in H2O K1=2.52; for 38.8% DMFA K1=1.78
______
      vlt R4N.X 20°C 0.02M C I K1=1.71
                                   2000RCc (83656) 404
Method: SW polarography. Medium: 0.025 M Et4NCl. By DPP, K1<1.
Data for 0-76% w/w PrOH/H2O, 0-76% w/w AN/H2O and 0-79% w/w DMF/H2O.
______
      cal none 50°C 0.00 C T H K1=2.01 1995WIa (83657) 405
T1+
Method: isothermal flow calorimetry. Measurements at 1.52 MPa. Data for
55-125 C. DH(K1)=-19.4 kJ mol-1, DS(K1)=-22 J K-1 mol-1.
______
   con non-aq 25°C 100% C I K1=4.99
                                  1993JHa (83658) 406
Medium: acetone. Data for acetonitrile and DMF media.
______
Tl+ sp non-aq 20°C 100% C K1=4.95 1993PSc (83659) 407
Method: spectrofluorescence. Medium: MeOH.
______
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vlt non-aq 23°C 100% U K1=5.00 1992LLa (83660) 408
Several mixtures of MeCN/H2O, acetone/H2O, THF/H2O and DMSO/H2O
______
Tl+ ix none 25°C 0.0 U K1=2.0 1991BMb (83661) 409
Tl+ vlt non-aq 23°C 100% U I K1=4.90 1991LKa (83662) 410
Medium: acetone; 0.05 M Bu4NClO4. Also in other solvents
______
     vlt R4N.X 22°C 0.03M C I K1=<2 1991PSa (83663) 411
Medium: 0.025 M Et4NCl04. Method: differential pulse polarography. Data
for 15-75% w/w CH3CN/H2O, 0.025 M Et4NClO4.
______
      vlt non-aq 25°C 100% C I K1=7.73 1991SSb (83664) 412
Method: polarography. Medium: acetonitrile, 0.05 M Et4NClO4.
In DMF, K1=3.65.
_____
Tl+ vlt alc/w 25°C 100% U K1=5.55 1988LFa (83665) 413
Medium: MeOH. In Me2NCHO, K1=3.42
                    _____
Tl+ cal non-aq 25°C 100% C H K1=5.34 1986ICa (83666) 414
Medium: MeOH. DH(K1)=-45.65 kJ mol-1, DS(K1)=-50.7 J K-1 mol-1.
_____
Tl+ vlt KNO3 25°C 0.10M C K1=2.98 1985KTb (83667) 415
Method: d.c. polarography. Medium: 0.10 M HNO3.
By a.c. polarography, K1=3.06
______
Tl+ oth alc/w 25°C 100% U K1=5.04 1980WJa (83668) 416
Method: fluorimetry in CH30H
______
Tl+ cal oth/un 25°C 0.10M U H T K1=2.27 1976ITb (83669) 417
DH=-18.6 kJ mol-1.
______
Tl+ vlt R4N.X 25°C 0.10M C H T K1=2.2 1976KKf (83670) 418
DH(K1)=-22.6 kJ mol-1, DS=-34 J K-1 mol-1
**********************************
                           CAS 33941-15-0 (4939)
C12H25N05
1,4,7,10,13-Pentaoxa-16-azacyclooctadecane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     con mixed 25°C 40% C TIH K1=5.97 2003KSc (83713) 419
Medium: 40% w/w dimethylformamide/AN. Data for 15-45 C.
DH(K1)=-29.7 kJ mol-1, DS=14.5 J K-1 mol-1. Also data for 60-100% DMF/AN
______
Tl+ vlt non-aq 22°C 100% C I K1=3.3
                                  2001MRa (83714) 420
Medium: DMF, 0.025 M Et4NClO4. Method: differential pulse polarography.
Data for binary mixtures of DMF with MeOH, nitromethane, PrOH, AN.
*******************************
                            CAS 41775-36-4 (2470)
1,4,7,13-Tetraoxa-10,16-diazacyclooctadecane;
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     con non-aq 25°C 100% C I K1=>6.5
                                1993JHa (83730) 421
Medium: acetone. Data for acetonitrile and DMF media.
**********************************
        L Cryptand 2,2 CAS 23978-55-4 (925)
C12H26N2O4
4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
------
Tl+ con non-ag 25°C 100% C T H K1=4.00 2001SKc (83905) 422
Medium: DMF. Data for 15-45 C. DH(K1)=-14 kJ mol-1,
DS(K1)=29 J K-1 mol-1. Also data for 40-80% w/w DMF/acetonitrile.
______
Tl+ gl R4N.X 25°C 0.05M C K1=2.3 1997BCc (83906) 423
Medium: 0.05 M Me4NClO4
-----
Tl+ sp non-aq 20°C 100% C K1=3.69 1993PSc (83907) 424
Method: spectrofluorescence. Medium: MeOH.
______
Tl+ vlt R4N.X 22°C 0.03M C I K1=2.19 1991PSa (83908) 425
Medium: 0.025 M Et4NCl04. Method: differential pulse polarography. Data
for 15-75% w/w CH3CN/H2O, 0.025 M Et4NClO4.
______
      ISE non-ag 25°C 100% U I K1=7.54 1983CFa (83909) 426
Medium: CH3NO2. K1=6.81 in acetone; 3.41 in DMF; K1=2.38 in DMSO; 7.94 in
MeCN; 3.54 in MeOH; 7.05 in propylene carbonate
______
Tl+ gl R4N.X 24°C 0.10M C K1=1.1 1975ANa (83910) 427
*******************************
C12H26O6 L Pentaglyme
                       CAS 1191-87-3 (2498)
2,5,8,11,14,17-Hexaoxaoctadecane; (CH3.0.CH2.CH2.0.CH2.CH2.0.CH2.)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ con alc/w 25°C 100% U K1=1.90 1975CJa (84025) 428
Medium: MeOH
*************************
                         CAS 78-64-8 (2062)
0.0'-Dihexyl dithiophosphoric acid; (C6H13O)2P(S)SH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ sp alc/w 25°C 100% U K1=4.64 1979SJd (84114) 429
************************
                 CAS 296-35-5 (143)
1,4,7,10,13,16-Hexaazacyclooctadecane; cyclo(-(NH.CH2.CH2)6-)
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
Tl+ con non-aq 25°C 100% C T H K1=5.48
                             2001SKc (84358) 430
Medium: DMF. Data for 15-45 C. DH(K1)=-20 kJ mol-1,
DS(K1)=39 J K-1 mol-1. Also data for 40-80% w/w DMF/acetonitrile.
**********************************
                          CAS 1798-14-7 (921)
(Pentamethylenedinitrilo)tetraethanoic acid; ((HOOC.CH2)2N.CH2.CH2)2CH2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
   gl KNO3 20°C 0.10M U K1=3.73 1967ABc (86208) 431
                      K(T1+HL)=2.88
C13H2605
                           (6410)
15,15-Dimethyl-1,4,7,10,13-pentaoxacyclohexadecane;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ con none 25°C 0.0 C K1=0.55 2001KMb (86489) 432
*********************************
                19-Crown-6 CAS 55471-27-7 (8943)
1,4,7,10,13,16-Hexaoxacyclononadecane;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     con oth/un 25°C dil C K1=1.08 1999TMa (86506) 433
Self medium (T1NO3).
********************************
        L
                Benzo15-crown-5 CAS 14098-44-3 (608)
2,3-Benzo-1,4,7,10,13-pentaoxacyclopentadeca-2-ene;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ vlt non-aq 20°C 100% C K1=1.4 19990Ba (88380) 434
Medium: DMF, 0.10 M Bu4N[BPh4].
-----
Tl+ con non-aq 25°C 100% C I K1=2.90
                              1993JHa (88381) 435
Medium: acetone. Data for acetonitrile and DMF media.
______
Tl+ sp non-aq 20°C 100% C K1=3.65 1993PSc (88382) 436
Method: spectrofluorescence. Medium: MeOH.
______
Tl+ vlt non-aq 25°C 100% C I K1=5.41 1991SSb (88383) 437
Method: polarography. Medium: acetonitrile, 0.05 M Et4NClO4.
In DMF, K1 = \langle 2.5.
     vlt alc/w 25°C 100% U I K1=3.05
Medium: 0.05 M (C4H9)4NO4 in methanol. Data also in ethanol, propanol
butanol, 2-methylpropanol, 4-hydroxy-4-methyl-2-pentanone and others
______
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By a.c. po ******** C14H22N2O	.c. po olaro; *****	olarogragragraghy, ******	aphy. K1=2 ***** H4L	**************************************		· /
Metal	Mtd	Medium	Temp	Conc Cal Fla	ags Lg K values	Reference ExptNo
Tl+	gl	KNO3	20°C	0.10M U	K1=6.7 K(T1+HL)=1.7 K(T1L+H)=7.3	1979ABa (88800) 440
T1+	vlt	NaClO4	25°C	0.30M U	K1=3.85 K(TlL+H)=11.29	1969KTc (88801) 441
T1+	gl	KNO3	20°C	0.10M U	K1=6.7	1967ABc (88802) 442
T1+	vlt	KNO3	30°C	0.10M U	K1=5.84	1967SSe (88803) 443
T1+	vlt	KNO3	25°C	0.50M U	K1=5.33	1966PAc (88804) 444
Tl+ Medium: 10 *****	0% Me(-	M KNO		K1=5.58	1966PAc (88805) 445 *********
C14H23N3O			H5L		CAS 67-43-	` '
Diethylen	etria		ntaetl	nanoic acid; 	HOOC.CH2.N(CH2.C	-6 (238) H2.N(CH2.COOH)2)2
	etria		ntaetl	nanoic acid; 		` '
Diethylen	etria		ntaetl Temp	nanoic acid; 	HOOC.CH2.N(CH2.C	H2.N(CH2.COOH)2)2
Diethylend 	etria Mtd gl vlt	Medium KNO3	Temp 20°C	nanoic acid; Conc Cal Fla O.10M U	HOOC.CH2.N(CH2.CH egs Lg K values K1=5.97 K(T1+HL)=4.2 K(T1L+H)=8.8 K1=5.45 K(T1L+H)=8.81	H2.N(CH2.COOH)2)2 Reference ExptNo
Diethylend Metal Tl+ By d.c. po	etria Mtd gl vlt olaro	Medium KNO3 NaClO4	ntaetl Temp 20°C 25?°(By a	nanoic acid; Conc Cal Fla 0.10M U 0.40M U	HOOC.CH2.N(CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.	H2.N(CH2.COOH)2)2 Reference ExptNo 1979ABa (89414) 446 1968KNa (89415) 447
Diethylend Metal Tl+ By d.c. po Tl+	etriam	Medium KNO3 NaClO4 graphy. KNO3	Temp 20°C 25?°C By a 20°C	nanoic acid; Conc Cal Fla 0.10M U .c.: K1=5.53,	HOOC.CH2.N(CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.	Reference ExptNo 1979ABa (89414) 446 1968KNa (89415) 447
Diethylend	etria Mtd gl vlt olaro gl *****	Medium KNO3 NaClO4 graphy. KNO3	Temp 20°C 25?°(By a 20°C ******	nanoic acid; Conc Cal Fla 0.10M U Conc Cal Fla 0.10M U Conc Cal Fla 0.10M U	HOOC.CH2.N(CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.	Reference ExptNo
Diethylend	etriam	Medium KNO3 NaClO4 graphy. KNO3 ******	Temp 20°C 25?°(By a 20°C *****	nanoic acid; Conc Cal Fla 0.10M U .c.: K1=5.53, 0.10M U ***********************************	HOOC.CH2.N(CH2.CH2.CH2.CH2.CH2.CH2.CH2.NCH2.CH2.CH2.NCH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.	H2.N(CH2.COOH)2)2 Reference ExptNo 1979ABa (89414) 446 1968KNa (89415) 447 1967ABc (89416) 448 **********************************
Diethylend	etriam	Medium KNO3 NaClO4 graphy. KNO3 ****** O,O'-b: Medium NaClO4	Temp 20°C By a 20°C ***** is(2 Temp 25°C	Conc Cal Fla	HOOC.CH2.N(CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.CH2.	Reference ExptNo 1979ABa (89414) 446 1968KNa (89415) 447 1967ABc (89416) 448 **********************************

```
By polarography: K1=5.37
-----
   gl KNO3 20°C 0.10M U
                      K1=4.38 1963FCa (89952) 451
T1+
                     K(T1+HL)=3.85
***********************************
                     (6658)
C14H26N2O8 H2L
1,4,10,13-Tetraoxa-7,16-diaza-2,3-dicarboxycyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ gl R4N.X 25°C 0.10M U K1=2.7 1990AFa (90226) 452
*******************************
C14H28N2O4 L Cryptand 2,1,1 CAS 31250-06-3 (836)
1,10-Diaza-4,7,13,18-tetraoxabicyclo[8,5,5]eicosane (2,1,1);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl R4N.X 25°C 0.05M C K1=5.1 1997BCc (90448) 453
Medium: 0.05 M Me4NClO4
-----
Tl+ EMF non-aq 25°C 100% U I K1=2.97 1993LRa (90449) 454
Medium: triethylphosphate, 0.05 M Et4NClO4
Data also for tri-n-butylphosphate: K1=3.36
-----
Tl+ gl R4N.X 25°C 0.05M U K1=3.95 1991LRc (90450) 455
Tl+ ISE alc/w 25°C 100% C I K1=5.65 1989CSa (90451) 456
Medium: MeOH. Also in water (3.19), and EtOH (5.12).
______
Tl+ ISE non-aq 25°C 100% U K1=7.0 1988CSc (90452) 457
In acetonitrile
______
   ISE non-aq 25°C 100% C I K1=1.44 1985CKa (90453) 458
Medium: DMSO. In propylenecarbonate K1=6.58; in DMF K1=3.15
********************************
C14H28N2O4 L Cryptand 2,2,0 CAS 95334-31-9 (6544)
4,7,13,16-Tetraoxa-1,10-diazabicyclo[8.8.2]eicosane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      ISE non-ag 25°C 100% U I K1=10.4 1991LSb (90464) 459
Medium: MeCN, 0.05 M Et4NClO4. In MeOH: K1=7.8; in DMF: K1=6.7
*********************************
C14H28N2O7
                         (2509)
1,17-Diacetamido-3,6,9,12,15-pentaoxaheptadecane; 0((CH2.CH2.0)2.CH2.CH2.CO.NH2)2
------
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     con alc/w 25°C 100% U K1=1.82
                             1975CJa (90494) 460
```

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Medium: MeOH
**********************************
                21-Crown-7
                         CAS 33089-36-0 (2264)
1,4,7,10,13,16,19-Heptaoxacycloheneicosane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      cal non-aq 25°C 100% C H K1=4.55 1986ICa (90542) 461
Medium: MeOH. DH(K1)=-40.1 kJ mol-1, DS(K1)=-47.3 J K-1 mol-1.
********************************
                          CAS 1072-40-8 (2499)
2,5,8,11,14,17,20-Heptaoxaheneicosane; CH3.0.(CH2.CH2.0)6.CH3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con alc/w 25°C 100% U K1=2.30
T1+
                               1975CJa (90713) 462
Medium: MeOH
**********************************
                          CAS 16195-35-0 (27)
5-(4-Chlorophenylazo)-8-hydroxyquinoline; Cl.C6H4.N:N.C9H5N.OH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp oth/un 25°C 0.10M U B2=7.81 1978KIa (90949) 463
*********************************
                          CAS 4312-09-8 (989)
5-Phenylazo-8-hydroxyguinoline; C6H5.N:N.C9H5N.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ sp oth/un 25°C 0.10M U B2=8.03 1978KIa (91271) 464
********************************
C15H30N2O3
                         CAS 72640-82-5 (6040)
4,7,13-Trioxa-1,10-diazabicyclo[8.5.5]eicosane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     EMF non-aq 25°C 100% U K1=2.18 1993LRa (92528) 465
Medium: triethylphosphate, 0.05 M Et4NClO4
______
      gl R4N.X 25°C 0.05M U H K1=2.42
                              1991LRc (92529) 466
DH(K1)=-61.2 kJ mol-1, DS=13.6 J K-1 mol-1
********************************
C16H2406
                Benzo18-crown-6 CAS 14098-24-9 (513)
2,3-Benzo-1,4,7,10,13,16-hexaoxacyclooctadeca-2-ene;
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con non-aq 25°C 100% C T H K1=2.76
                               2001SKc (94467) 467
Medium: DMF. Data for 15-35 C. DH(K1)=-25 kJ mol-1,
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DS(K1)=-30 J K-1 mol-1. Also data for 40-80% w/w DMF/acetonitrile.
_____
      con none 20°C 0.0 C T H K1=1.71
                                 1990TAa (94468) 468
Data for 15-32 C. At 15 C, K1=1.75; at 30 C, K1=1.66
At 25 C, DH(K1)=-9.7 kJ mol-1, DS(K1)=-0.13 J K-1 mol-1.
-----
Tl+ con none 25°C 0.0 U K1=1.68 1989TKa (94469) 469
T1+
      cal non-ag 25°C 100% C H K1=4.37 1986ICa (94470) 470
Medium: MeOH. DH(K1)=-39.1 kJ mol-1, DS(K1)=-47.7 J K-1 mol-1.
*****************************
                      CAS 61696-54-6 (6104)
C16H24014
1,4,7,10,13,16-Hexaoxacyclooctadeca-2,3,11,12-tetracarboxylic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl R4N.X 25°C 0.10M M K1=3.9
                                 1991FGb (94503) 471
                        B(T1HL)=8.3
Medium: 0.10 M Et4NNO3.
***********************************
                            (7444)
1-Aza-4,7,10,13-tetraoxa-1-phenyl-cyclopentadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ con mixed 25°C 90% C TIH K1=4.05 1998MTa (94521) 472
Medium: 90% CH3CN/H20. Data for 20-35 C. DH(K1)=18.6 kJ mol-1, DS(K1)=14.8
J K-1 mol-1. In 50% CH3CN/H20, K1=3.58, DH(K1)=7.9, DS(K1)=41.9.
***************
                 (6659)
            H4L
C16H26N2O12
1,4,10,13-Tetraoxa-7,16-diaza-2,3,11,12-tetracarboxycyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Tl+ gl R4N.X 25°C 0.10M U K1=3.7 1990AFa (94593) 473
                       B(T1HL)=13.3
******************************
                  CAS 130190-52-2 (6660)
C16H26N2O12
1,4,10,13-Tetraoxa-7,16-diaza-2,3,7,16-tetracarboxycyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
- - - '
Tl+ gl R4N.X 25°C 0.10M U K1=5.4 1990AFa (94607) 474 B(TlHL)=14.2
*******************************
C16H2606 L
                   CAS 57721-93-4 (2502)
2,5,8,11,14,17-Hexaoxa-9,10-benzo-octadeca-9-ene; C6H4(0.(CH2.CH2.0)2.CH3)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
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con alc/w 25°C 100% U K1=1.73 1975CJa (94634) 475
T1+
Medium: MeOH
**********************************
C16H32N2O5 L
               Cryptand 2,2,1 CAS 31364-42-8 (837)
1,10-Diaza-4,7,13,16,21-pentaoxabicyclo[8,8,5]tricosane (2,2,1);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ gl R4N.X 25°C 0.05M C K1=7.0
                               1997BCc (95298) 476
Medium: 0.05 M Me4NClO4
-----
Tl+ sp non-aq 20°C 100% C K1=>6
                               1993PSc (95299) 477
Method: spectrofluorescence. Medium: MeOH.
Tl+ ISE alc/w 25°C 100% C I K1=10.76 1989CSa (95300) 478
Medium: MeOH. Also in EtOH (11.01).
______
      sp non-aq 25°C 100% U K1=11.9
                              1988CSc (95301) 479
In acetonitrile
-----
      ISE non-aq 25°C 100% C I K1=6.80
                                1985CKa (95302) 480
Medium: DMSO. In DMF K1=8.61; in propylenecarbonate K1=12.13
-----
  kin R4N.X 25°C 0.10M U K1=6.8 1980GBa (95303) 481
**********************************
                         CAS 60598-04-1 (1530)
4,7-Dimethyl-1,4,7,10-tetraaza-13,18-dioxabicyclo[8,5,5]eicosane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ gl R4N.X 25°C 0.10M U K1=3.9 1978LMa (95473) 482
*******************************
C16H3406
                          CAS 57721-92-3 (2501)
2,5,8,15,18,21-Hexaoxadocosane; CH3.0.(CH2.CH2.0)2.(CH2)6.0.(CH2.CH2.0)2.CH3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ con alc/w 25°C 100% U
                                1975CJa (95487) 483
Medium: MeOH
************************************
                          CAS 1191-91-9 (2500)
2,5,8,11,14,17,20,23-Octaoxatetracosane; CH3.0.(CH2.CH2.O)7.CH3
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con alc/w 25°C 100% U K1=2.55 1975CJa (95496) 484
Medium: MeOH
************************************
                          CAS 83296-49-5 (2063)
0,0'-Dioctyl dithiophosphoric acid; (C8H170)2P(S)SH
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp alc/w 25°C 100% U K1=4.63 1979SJd (95504) 485
**************************
                          CAS 142565-14-8 (6562)
4,7,13,16-Tetraoxa-1,10-diazabicyclo[8.8.5]tricosane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      gl R4N.X 25°C 0.05M C I K1=4.10 1992CGb (96752) 486
Medium: Et4NClO4. In MeOH: K1=6.48; in DMF K1=5.05
*******************************
                           CAS 15196-73-3 (2359)
2,3-(4'-Dimethylethylbenzo)-1,4,7,10,13-pentaoxacyclopentadeca-2-ene;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+ gl non-aq 25°C 100% U K1=4.13 B2=6.35 1982MDa (97815) 487
Medium: propylene carbonate
**********************************
            H6L TTHA
                          CAS 869-52-3 (694)
C18H30N4O12
Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     vlt NaClO4 25°C 0.50M U K1=4.91 1977CNa (98097) 488
B(TlHL)=14.64
********************************
                  CAS 57721-61-7 (2510)
C18H3409 L
3,6,9,12,15-Pentaoxaheptadecane-1,17-dioic acid diethyl ester
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      con alc/w 25°C 100% U K1=1.44 1975CJa (98399) 489
Medium: MeOH
***********************************
                 Cryptand 2,2,2 CAS 23978-09-8 (514)
1,10-Diaza-4,7,13,16,21,24-hexaoxabicyclo[8.8.8]hexacosane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     vlt non-aq 25°C 100% C I K1=12.3 1999FKb (98751) 490
Medium: acetonitrile, 0.10 M Et4NClO4. Method: cyclic voltammetry.
Also in: DMF (K1=7.8), DMSO (6.2), MeOH (10.0), acetone (10.3) etc.
______
Tl+ gl R4N.X 25°C 0.05M C K1=6.2
                              1997BCc (98752) 491
Medium: 0.05 M Me4NClO4
-----
   sp non-aq 20°C 100% C K1=>7
                                 1993PSc (98753) 492
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Method: spectrofluorescence. Medium: MeOH.
-----
     vlt R4N.X 22°C 0.03M C I K1=6.55 1991PSa (98754) 493
Medium: 0.025 M Et4NClO4. Method: differential pulse polarography. Data
for 15-75% w/w CH3CN/H2O, 0.025 M Et4NClO4.
_____
Tl+ ISE non-aq 25°C 100% U IH K1=12.4 1988CSc (98755) 494
In CH3CN. In CH3CN/water mixtures:mole fraction 0.8, K=10.9; 0.5, K=9.2;
0.3, K=8.4; 0.0, K=6.6
______
Tl+ ISE alc/w 25°C 100% C K1=8.06 1985CKa (98756) 495
_____
   kin R4N.X 25°C 0.10M U K1=6.4 1980GBa (98757) 496
______
Tl+ EMF non-aq 25°C 100% C I K1=6.2 1979BLb (98758) 497
Method: Tl electrode. Medium: MeOH, 0.05 M Me4NClO4.
Also K1=6.3 (H2O), 6.2 (DMSO), 13.4 (CH3CN).
______
Tl+ EMF oth/un 25°C 0.05M C I K1=6.4 1978YTa (98759) 498
Method: Tl amalgam electrode. Electrolyte not stated.
In MeOH, 0.05 M: K1=10.1. In DMSO, 0.10 M: K1=6.1
______
Tl+ gl R4N.X 25°C 0.10M C H K1=5.5 1975ANa (98760) 499
Medium: Me4NNO3. DH(K1)=-55.2 kJ mol-1, DS=-61.9
______
Tl+ gl R4N.X 25°C 0.05M C K1=6.8 1975LSc (98761) 500
*******************************
C19H39N3O5 L CAS 60598-00-7 (1537)
4-Methyl-1,4,10-triaza-7,13,16,21,24-pentaoxa-bicyclo[8,8,8]hexacosane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl R4N.X 25°C 0.10M U K1=6.3 1978LMa (99498) 501
*******************************
C20H2406 L DiBz-18-Crown-6 CAS 14187-32-7 (604)
2,3:11,12-Dibenzo-1,4,7,10,13,16-hexaoxacyclooctadeca-2,11-diene
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ con non-aq 25°C 100% C TIH K1=2.45
Medium: DMF. Data for 15-55 C. Also data for 25-75% mol% DMF/AN.
DH(K1)=-29 \text{ kJ mol}-1, DS(K1)=-146 J K-1 mol}-1.
_____
Tl+ con non-aq 25°C 100% C I K1=4.73 1993JHa (100247) 503
Medium: acetone. Data for acetonitrile and DMF media.
______
Tl+ sp non-aq 20°C 100% C K1=4.42 1993PSc (100248) 504
Method: spectrofluorescence. Medium: MeOH.
______
   vlt non-aq 23°C 100% U K1=4.60 1991LKa (100249) 505
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medium: acetone; 0.05 M Bu4NClO4. Also in other solvents
-----
      vlt non-ag 25°C 100% C I K1=4.78 1991SSb (100250) 506
Method: polarography. Medium: acetonitrile, 0.05 M Et4NCl04.
In DMF, K1=\langle 2.5.
-----
Tl+ vlt alc/w 25°C 100% U K1=3.38 1988LFa (100251) 507
Medium: MeOH, In Me2NCHO, K1=1.96
______
Tl+ vlt non-aq 25°C 100% U I K1=4.90 1978HKc (100252) 508
Medium: CH3CN, 0.05M Bu4NCl04
-----
Tl+ nmr non-aq 29°C 100% U K1=2.48
                                 1977SZa (100253) 509
Medium: DMF
Tl+ sol none 25°C 0.0 U I K1=1.50
                                  1975SNa (100254) 510
********************************
                             (2504)
2,5,8,11,14,17,20,23-Octaoxa-12,13-benzotetracosa-12-ene; C6H4(0.(CH2.CH2.0)3.CH3)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ con alc/w 25°C 100% U K1=2.45 1975CJa (100527) 511
Medium: MeOH
***********************************
         L DiCy-18-crown-6 CAS 16069-36-6 (1653)
2,3:11,12-Dicyclohexyl-1,4,7,10,13,16-hexaoxacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ vlt non-aq 22°C 100% C I K1=3.2 2002RYa (100712) 512
Method: DPP in DMF, 0.025 M Et4NClO4. By conductivity, K1=3.30.
Data for 0-100 mol% DMF/H2O, and MeOH/H2O, AN/H2O and PrOH/H2O mixtures.
______
Tl+ con non-aq 25°C 100% C TIH K1=3.3 2001RKa (100713) 513
Medium: DMF. Data for 15-55 C. Also data for 25-75% mol% DMF/AN.
DH(K1)=121 \text{ kJ mol-1}, DS(K1)=339 \text{ J K-1 mol-1}.
______
      con non-aq 25°C 100% C T H K1=3.53
                               2001SKc (100714) 514
Medium: DMF. Data for 15-45 C. DH(K1)=-28 kJ mol-1,
DS(K1)=-28 J K-1 mol-1. Also data for 40-80% w/w DMF/acetonitrile.
-----
Tl+ con non-aq 25°C 100% C I K1=6.23 1993JHa (100715) 515
Medium: acetone. Data for acetonitrile and DMF media.
______
Tl+ sp non-aq 20°C 100% C K1=4.67 1993PSc (100716) 516
Method: spectrofluorescence. Medium: MeOH.
______
      vlt non-aq 25°C 100% C I K1=7.54 1991SSb (100717) 517
Method: polarography. Medium: acetonitrile, 0.05 M Et4NClO4.
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In DMF, K1=3.55.
______
      vlt alc/w 25°C 100% U K1=4.95
                                1988LFa (100718) 518
Medium: MeOH, In Me2NCHO, K1=3.30
-----
Tl+ vlt KNO3 25°C 0.10M C K1=3.20
                               1985KTb (100719) 519
Method: d.c. polarography. Medium: 0.10 M HNO3.
By a.c. polarography, K1=3.18
T1+
     cal oth/un 25°C 0.10M U
                                1976ITb (100720) 520
                       K1=2.44 (cis-syn-cis isomer)
                       K1=1.83 (cis-anti-cis isomer)
DH(Syn)=-15.1 and DH(Anti)=-17.9 kJ mol-1.
C20H40N2O4
                           (6625)
1,10-Diaza-4,7,13,16-tetraoxabicyclo[8.8.8]hexacosane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ gl non-aq 25°C 100% C I K1=6.19 1992LSc (100779) 521
Medium: MeCN, 0.05 M Et4NClO4. In DMF K1=3.1
**********************************
C20H42N404
                         CAS 39678-14-3 (1543)
4,7-Dimethyl-1,4,7,10-tetraaza-13,16,21,24-tetraoxa-bicyclohexacosane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl R4N.X 25°C 0.10M U K1=5.5
K(Tl+HL)=1.9
                                1978LMa (100894) 522
C20H4302PS2
                          CAS 2253-89-0 (2064)
0,0'-Didecyl dithiophosphoric acid; (C10H210)2P(S)SH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp alc/w 25°C 100% U K1=4.64 1979SJd (100904) 523
**********************************
C22H2807 L Dibenzo-21-Cr-7 CAS 14098-41-0 (2876)
2,3:11,12-Dibenzo-1,4,7,10,13,16,19-heptaoxacycloheneicosane-2,11-diene;
_____
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con mixed 25°C 40% C TIH K1=3.29
                                 2003KSc (102060) 524
Medium: 40% w/w dimethylformamide/AN. Data for 15-45 C.
DH(K1)=-48 kJ mol-1, DS(K1)=-101 J K-1 mol-1. Also data for 60-100% DMF/AN
-----
Tl+ con non-aq 25°C 100% C I K1=5.07 1993JHa (102061) 525
Medium: acetone. Data for acetonitrile media.
-----
     cal non-aq 25°C 100% C H K1=4.03
                               1986ICa (102062) 526
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Medium: MeOH. DH(K1)=-36.9 kJ mol-1, DS(K1)=-46.3 J K-1 mol-1.
*******************************
                             (2506)
2,5,8,13,16,19-Hexaoxa-9,10:11,12-dibenzoeicosa-9,11-diene;
(-C6H4.0.(CH2.CH2.0)2.CH3)2
__________
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con alc/w 25°C 100% U K1=1.13 1975CJa (102136) 527
Medium: MeOH
**********************************
              L Bz-Cryptand 222 CAS 31250-18-7 (2269)
5,6-Benzo-4,7,13,16,21,24-hexaoxa-1,10-diazabicylo[8:8:8]hexacosa-5-ene;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
    ISE alc/w 25°C 100% C I K1=8.71 1989CSa (102285) 528
Medium: MeOH. Also in water (5.84), EtOH (8.58), propylene carbonate(10.73)
and dimethylformamide (6.79).
______
      ISE non-ag 25°C 100% U K1=10.3 1988CSc (102286) 529
In acetonitrile
***********************************
                            CAS 39678-22-3 (1542)
4,7,13,16-Tetramethyl-1,4,7,10,13,16-hexaaza-21,24-dioxabicyclohexacosane;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                       K1=4.1 1978LMa (102493) 530
      gl R4N.X 25°C 0.10M U
                        K(Tl+HL)=1.9
*********************************
C23H23N05
                           CAS 218619-58-0 (7808)
Dibenzo-pyridino-18-crown-6;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      vlt non-aq 22°C 100% C I K1=<1
                                   2001MRa (102667) 531
Medium: DMF, 0.025 M Et4NCl04. Method: differential pulse polarography.
Data for binary mixtures of DMF with MeOH, nitromethane, PrOH, AN.
______
      EMF alc/w 25°C 100% C T H K1=3.74 2001SZb (102668) 532
T1+
Medium: methanol, 0.5 M Bu4NClO4. Method: Ag electrode, using competitive
complexation with Ag+. Data for 5-35 C. DH=-52.0 kJ mol-1, DS=-97
(7368)
9-(2'-Hydroxy-5'-methylbenzyl)-3,6,12,15-Tetraoxa-9,21-diazabicyclo[15.3.1]heneicos
      Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
cal alc/w 25°C 100% U T H K1=4.34 1997ZBa (102782) 533
T1+
Medium: MeOH. Data also for several similar 5'-substituted ligands
***********************************
                              (7369)
9-(2'-Pyridylmethyl)-3,6,12,15-tetraoxa-19-methyl-21-hydroxy-9-azabicyclo[15.3.1]he
neicosatriene;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      cal alc/w 25°C 100% U H K1=4.45 1997ZBa (102787) 534
Medium: MeOH
************************************
              L DiBz-24-Crown-8 CAS 14174-09-5 (580)
2,3:14,15-Dibenzo-1,4,7,10,13,16,19,22-octaoxacyclotetracosa-2,14-diene;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con non-aq 25°C 100% C I K1=4.90
                                  1993JHa (103180) 535
Medium: acetone. Data for acetonitrile media.
______
       cal non-aq 25°C 100% C H K1=3.40
                                   1986ICa (103181) 536
Medium: MeOH. DH(K1)=-30.0 kJ mol-1, DS(K1)=-35.6 J K-1 mol-1.
*******************
                             CAS 71735-94-9 (7414)
1,4,7,10,13,16,19,22,25-Nonaoxacycloheptacosane-2,3,11,12,20,21-hexacarboxylic
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ gl R4N.X 25°C 0.10M M K1=3.4
                                   1991FGb (103311) 537
                         B(T1HL)=8.0
Medium: 0.10 M Et4NNO3.
************************************
2,5,8,11,14,17,20,23,26,29-Decaoxa-15,16-benzo-triconta-15-ene;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      con alc/w 25°C 100% U K1=2.80
                                   1975CJa (103401) 538
Medium: MeOH
******************************
                            CAS 239470-22-5 (8948)
4'-Carboxybenzo-30-crown-10;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      con non-aq 25°C 100% C T H K1=5.52 1999RGa (103778) 539
Medium: acetonitrile. Data for 5-35 C. DH(K1)=-70.6 kJ mol-1, DS(K1)=
-132 J K-1 mol-1.
*******************************
```

```
C26H36N2O6
                 DiBzCryptand222 (746)
             L
5,6,14,15-Dibenzo-4,7,13,16,21,24-hexaoxa-1,10-diazabicyclo[8.8.8]hexacosan-5,14-di
         Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ ISE non-aq 25°C 100% U K1=10.2 1988CSc (104148) 540
In acetonitrile
______
     ISE alc/w 25°C 100% C I K1=7.9 1985CKa (104149) 541
Medium: MeOH. In propylenecarbonate K1=9.81; in DMF K1=6.14; in DMSO K1=4.58
**********************************
C26H38N2O4
                            CAS 80757-23-9 (2450)
N,N'-Bis(benzyl)-1,10-diaza-4,7,13,16-tetraoxacyclooctadecane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ con non-aq 25°C 100% C T H K1=3.51 2001SKc (104192) 542
Medium: DMF. Data for 15-35 C. DH(K1)=-44 kJ mol-1,
DS(K1)=-81 J K-1 mol-1. Also data for 40-80% w/w DMF/acetonitrile.
**********************************
C26H3808
2,5,8,11,16,19,22,25-Octaoxa-12,13:14,15-dibenzohexacosa-12,14-diene;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con alc/w 25°C 100% U K1=1.81 1975CJa (104221) 543
Medium: MeOH
***********************************
C28H40010 L DiBz-30-crown10 CAS 104946-67-0 (1776)
2,3:17,18-Dibenzo-1,4,7,10,13,16,19,22,25,28-decaoxacyclotriaconta-2,17-diene;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      con mixed 25°C 40% C TIH K1=2.89 2003KSc (104914) 544
Medium: 40% w/w dimethylformamide/AN. Data for 15-45 C.
DH(K1)=-47 kJ mol-1, DS(K1)=-101 J K-1 mol-1. Also data for 60-100% DMF/AN
______
      con mixed 25°C 40% C TIH K1=3.81
                                   2003KSc (104915) 545
Medium: 40% w/w dimethylformamide/AN. Data for 15-45 C.
DH(K1)=-77 kJ mol-1, DS(K1)=-189 J K-1 mol-1. Also data for 60-100% DMF/AN
_______
Tl+ con non-aq 25°C 100% C I K1=5.39 1993JHa (104916) 546
Medium: acetone. Data for acetonitrile media.
______
      sp non-aq 20°C 100% C K1=4.53 1993PSc (104917) 547
T1+
Method: spectrofluorescence. Medium: MeOH.
_____
  con non-ag 25°C 100% U I K1=6.30 1991ASb (104918) 548
Medium: 1,2-dichlorethane. In nitromethane: K1=5.48; in MeCN: K=5.15;
```

```
in acetone: K=5.03
-----
       vlt non-aq 25°C 100% C I K1=5.47 1991SSb (104919) 549
Method: polarography. Medium: acetonitrile, 0.05 M Et4NCl04.
In DMF, K1=\langle 2.5.
In DMF, K1=<2.5.
Tl+ ISE non-aq 25°C 100% U K1=5.35 1982MDa (104920) 550
Medium: propylene carbonate
********************************
             L Bis(15-crown-5)
                             (6879)
Methylene-bis(4'-(2,3-benzo-1,4,7,10,13-pentaoxacyclopentadeca-2-ene));
CH2(C14H1905)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+ cal alc/w 25°C 80% U H K1=2.04 1990LTa (105140) 551
Medium: 80% v/v MeOH/H2O. DH(K1)=-102.0 kJ mol-1. Also data for a large
range of benzo-15-crown-5 dimers with 4'-bridging groups up to 10 carbons.
************************************
              L Anthracene-22 (3329)
C30H40N2O4
6,9,17,20-Tetraoxa-3,12-diaza[14:8](9,10)anthracenophane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      sp alc/w 25°C 100% U K1=8.52 1989FDa (105282) 552
Medium: MeOH, 0.1 M Bu4NClO4
************************************
C32H44N2O4 L
                              (6164)
7,10,17,20-Tetraoxa-4,13-diaza[16:8](9,10)anthracenophane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ sp alc/w 25°C 100% U K1=7.07 1989FDa (105763) 553
Medium: MeOH, 0.1 M Bu4NClO4
*************************
                            CAS 210485-26-0 (3146)
15,31-Diethylhexadecahydroanthra[2,3-b:6,7-b']bis[1,4,7,10,13]pentaoxacyclopentadec
in:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      vlt non-ag 20°C 100% C K1=2.8 19990Ba (106082) 554
Medium: DMF 0.10 M Bu4N[BPh4]. Data for other 15,31-dialkyl derivatives.
*************************************
            H2L Lasalocid CAS 25999-20-6 (2335)
C34H5408
Lasalocid acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
   nmr non-ag 20°C 100% C
                                   1998MLa (106162) 555
```

K(T1+HL)=1.8

```
Medium: CD3OD. Method: 13C nmr.
*********************************
             HL Monensin CAS 17090-79-8 (737)
Monensin, 1,6-dioxaspiro[4,5]decane derivative;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     con non-aq 25°C 100% C K1=4.30 1997PBb (106539) 556
Medium: acetonitrile. Additional method: potentiometry with ISE.
By calorimetry, DH(K1)=-24 kJ mol-1, DS(K1)=0 J K-1 mol-1
______
      vlt non-aq 23°C 100% U I K1=10.6 1994FRa (106540) 557
Medium: MeCN. In PrCN: K1=9.9; acetone: 9.9; DMF: 7.2; N-Me-pyrrolidinone:
6.0; NN-DMA: 6.0; DMSO: 4.5; Diethylformamide: 4.1; Di-Et-acetamide: 4.1
-----
Tl+ vlt non-aq 23°C 100% U I K1=4.5 1994RCa (106541) 558
In DMSO/MeCN mixt: mol. fract. DMSO=1. At mf: K1=10.6, 0.2: 6.0; 0.5: 5.1.
In DMSO/acetone: mf DMSO=0: K1=9.9; 0.5: 5.6. DMSO/HMPT: mf 0:1.9, 0.5: 2.2
______
     vlt alc/w 25°C 100% U K1=3.31 1978HPa (106542) 559
Method: Cyclic voltametry
CAS 210485-29-3 (3260)
Hexadecahydro-15,31-bis(2-methylpropyl)anthra[2,3:6,7]bis[1,4,7,10,13]pentaoxacyclo
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+ vlt non-aq 20°C 100% C K1=1.6 19990Ba (106702) 560
Medium: DMF 0.10 M Bu4N[BPh4]. Data for other 15,31-dialkyl derivatives.
**********************
                Chloride CAS 7647-01-0 (50)
C1-
Chloride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl++ oth NaCl04 23°C 1.00M U K1=4.79 B2=8.07 1974DSa (5874) 561 K3=1.11
Method: Pulse radiolysis.
-----
Tl++ oth NaClO4 ? 1.0M U K1=4.8 B2=8.10 1974DSa (5875) 562
                       K3=1.1
Method: pulse radiolysis
*********************************
                        CAS 294-95-1 (8604)
C8H1602S2
1,7-Dioxa-4,10-dithiacyclododecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
cal non-aq 25°C 100% C H K1=3.87 B2= 3.87 1986BUe (62627) 563
DH(K1)=-2.9 \text{ kJ mol}-1, DS(K1)=64.1 \text{ J K}-1 \text{ mol}-1; DH(K2)=-7.7.
Medium: MeOH.
***********************************
                  12-Crown-4 CAS 294-93-9 (174)
1,4,7,10-Tetraoxacyclododecane; cyclo(-0.(CH2.CH2.0)3.CH2.CH2-)
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl++ cal non-aq 25°C 100% C H K1=3.22 B2= 3.22 1986BUe (62733) 564
DH(K1)=-9.4 \text{ kJ mol}-1, DS(K1)=30 \text{ J K}-1 \text{ mol}-1; DH(K2)=-10.6.
Medium: MeOH.
**********************************
                            CAS 294-92-8 (654)
C8H18N2O2
1,7-Dioxo-4,10-diazacyclododecane;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     cal non-aq 25°C 100% C H K1=2.48 B2= 2.48 1986BUe (62851) 565
DH(K1)=-28.5 \text{ kJ mol-1}, DS(K1)=-48.3 \text{ J K-1 mol-1}; DH(K2)=8.
Medium: MeOH.
***********************************
C12H24O4S2
                             CAS 296-39-9 (4938)
1,4,10,13-Tetraoxa-7,16-dithiacyclooctadecane;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                          K1=3.93
      cal non-aq 25°C 100% C H
                                    1986BUe (83145) 566
Medium: MeOH. DH(K1)=-11.2 kJ mol-1, DS(K1)=37.3 J K-1 mol-1.
******************************
                            CAS 17455-13-9 (577)
                  18-Crown-6
1,4,7,10,13,16-Hexaoxacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      cal non-aq 25°C 100% C H
                          K1=5.22
                                    1986BUe (83671) 567
Medium: MeOH. DH(K1)=-50.9 kJ mol-1, DS(K1)=-71.1 J K-1 mol-1.
*****************************
                  Cryptand 2,2
                            CAS 23978-55-4 (925)
4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       cal non-ag 25°C 100% C H K1=3.06 1986BUe (83911) 568
Medium: MeOH. DH(K1)=-21.2 kJ mol-1, DS(K1)=-43.0 J K-1 mol-1.
********************************
             HL Electron
                          (442)
Electron;
______
      Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
```

T1+++	kin NaClO4 25°C 0.30M U I	1974FFb (980) 569 K(Tl+e)=5.6 (330mV) K(Tl(II)+e)=37.5 (2.22V)
	EMF oth/un 135°C 100% U	1969APa (981) 570 K(Tl + 2Tl(s)=3Tl+) > 51.2
Medium: (Na,K,Al)Cl	
	EMF NaCl04 25°C 3.00M U	1966JOa (982) 571 K(Tl+2I=Tl+ + I2)=25.34
Tl+++	EMF oth/un 25°C 0.50M U I	1962BBc (983) 572 K(Tl+2e=Tl(I))=26.0(770 mV)
	25 mV). In 0.5 to 5 M HClO4 K=4	04 K=41.1(1215 mV). In 0.5 to 5M HNO3
T1+++	EMF NaClO4 25°C 3.0M U	K(Tl+2e=Tl(I))=43.28(1280 mV)
	EMF none 25°C 0.0 U	1952KJa (985) 574 K=0.7(20 mV)
Method: ar	nperometry. K: 0.5Tl2O3(s)+1.5H	l20+2e=T1(I)+30H
	EMF none 25°C 0.0 U	1943STa (986) 575 K(Tl+2e=Tl(I))=43.3(1280 mV)
Tl+++ Medium: HM	EMF oth/un 0°C 1.0M U TI	1936NGa (987) 576 K(Tl+2e=Tl(I))=44.04(1193 mV) 3. At 25 C K=41.61(1230.3 mV)
********* Br- Bromide;		**************************************
Metal	Mtd Medium Temp Conc Cal Flag	s Lg K values Reference ExptNo
T1+++	nmr NaClO4 25°C 3.0M C	1990BGc (2362) 577 B3=21.9 B4=25.7
Medium: 3	.0 M HClO4. Method: 206Tl nmr.	
Tl+++		K1=9.28 B2=16.70 1967YKa (2363) K3=5.4 K4=3.6 K5=1.5 K(T1L+H2O=T1OHL+H)=-1.84
T1+++		K1=9.51 B2=16.88 1964LRa (2364) B3=22.30 B4=26.43

```
Tl+++ EMF NaClO4 20°C 7.0M U H K1=9.62 B2=17.06 1963AGa (2365) 580
                            K3=5.53
                            K4=4.14
                            B4=26.73
                            K5 < -0.4
Medium: 4 M NaClO4, 3 HClO4. By calorimetry: DH(K1)=-37.5 kJ mol-1, DS=56.0
J K-1 mol-1. DH(K2)=-25.5, DS=55.6; DH(K3)=-19.1, DS=40.5; DH(K4)=-8.9, DS=48.8
______
Tl+++ EMF oth/un ? var U
                                       1961EVa (2366) 581
                         B4=19.7
------
                            K1=8.3 B2=14.6 1960BTc (2367) 582
Tl+++ EMF NaClO4 20°C 0.40M U
                            K3=4.6
                            K4=3.1
                            K5 = 2.5
                            K6=1.7
Tl+++ EMF NaCl04 25°C 2.20M U
                            K1=8.9 B2=16.4 1956PVa (2368) 583
                            K3=5.7
                            K4 = 4.0
                            K5 = 3.1
                            K6=2.4
Tl+++ EMF oth/un 25°C var U
                                       1950BJa (2369) 584
                           B4 = 20.2
______
Tl+++ EMF none 18°C 0.0 U K1=9.7 B2=16.6 1949BEa (2370) 585
                            K3 = 4.6
                            K4=2.7
********************************
                           CAS 74-90-8 (230)
              HL Cyanide
Cyanide;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ nmr NaClO4 25°C 1.0M C
                                       1998MGb (2765) 586
                            K(Pt(CN)4+Tl+CN)=19.9
                            K(Pt(CN)4+T1+2CN)=30.7
                            K(Pt(CN)4+T1+3CN)=38.6
                            K(Pt(CN)4+T1+4CN)=44.8
Method: 195Pt and 205Tl nmr. K(2Pt(CN)4+Tl+2CN)=32.1.
Tl+++ nmr oth/un 25°C 3.00M U M K1=12.7 B2=25.5 1996BBc (2766) 587
                            B3=34.0
                            B4=41.3
                            B(TlLCl)=19.1
                            B(T1LC12)=22.3
Medium: LiCl04/HCl04. B(TlLCl3)=24.6, B(Tl(L)2Cl)=28.6, B(Tl(L)2Cl2)=30.9,
B(T1(L)3C1)=36.4.
```

```
Tl+++ nmr NaCl04 25°C 4.0M C K1=13.21 B2=26.50 1989BGb (2767) 588
                        B3=35.17
                        B4=42.61
Method: 205Tl and 13C nmr. Medium: 3 M LiClO4/1 M NaClO4.
                 Tl+++ kin NaClO4 30°C 0.50M U
                            1955PDa (2768) 589
                       K2/K1 = -0.82
Medium: 0.5 M(H,NaClO4)
                 Tl+++ con oth/un 25°C var U
                                  1950BJa (2769) 590
                        B4=35
**********************************
            HL Chloride CAS 7647-01-0 (50)
Chloride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ nmr oth/un 25°C 3.0M C
                                  1989BGc (5876) 591
                        K5=ca. -0.30
Method: 205Tl nmr. Medium: 3.0 M HClO4/HCl.
-----
Tl+++ EMF NaCl04 25°C 3.0M U K1=7.04 B2=12.32 1971BSd (5877) 592
                       B3=15.30
                        B4=17.36
                        K(T1L+H20=T1L(OH)+H)=-1.87
-----
Tl+++ sp oth/un 23°C ? U
                                 1970SCb (5878) 593
                     K5=-0.07
-----
Tl+++ EMF oth/un 25°C 1.0M U
                        K1=5.88 B2=10.40 1969CPd (5879) 594
                        B3=12.94
                        B4=14.58
Medium: H2SO4
-----
Tl+++ oth non-aq 26°C 100% U
                                  1968WSb (5880) 595
                       K5=0.66
Method:Raman spectra. Medium:MeNO2
_____
     ISE NaCl04 25°C 4.0M U H K1=7.10 B2=12.50 1965KMd (5881) 596
                        B3=16.00
                        B4=18.50
Medium: 3 M LiClO4,1 HClO4. By calorimetry: DH(K1)=-32.6 kJ mol-1, DH(K2)=
-15.5, DH(K3)=DH(K4)=0; DS(K1)=29.3 J K-1 mol-1, DS(K2)=50.2, DS3=66.9, DS4=25.1
Tl+++ oth oth/un var U
B6/B4=-0.7
-----
                                  1965SPb (5882) 597
Method:Raman spectra
-----
Tl+++ dis oth/un 25°C 0.0 U TIH
                                 1964NUa (5883) 598
```

```
K3=3.03
```

```
K4=1.47
K4=1.51(0 C), 1.32(35 C); DH(K4)=-21.3 kJ mol-1, DS=-41.8 J K-1 mol-1
In 0.5 M NaCl04:K4=1.38(25 C). In 96% D20, I=0: K4=1.31(25 C)
                      1964PFa (5884) 599
Tl+++ dis NaCl 25°C var U
                           K(T1C14+H)=1
-----
                            1964PMa (5885) 600
Tl+++ ix NaClO4 20°C 1.50M U
                 K4=0.6
______
Tl+++ cal NaClO4 25°C 3.0M U IH K1=7.16 B2=12.60 1964WGa (5886) 601
                            K3=3.55
                            K4=2.17
                            B4=18.33
Also solubility, redox. DH(K1) = -22.8 \text{ kJ mol} - 1, DH(K2) = -18.4, DH(K3) = -4.6,
DH(K4)=-1.3; DS(K1)=60.6 J K-1 mol-1, DS(K2)=42.2, DS(K3)=52.7, DS(K4)=37.2
______
T1+++ EMF NaCl04 20°C 7.0M U T K1=7.54 B2=13.38 1963AGa (5887) 602
                            K3=3.41
                            K4=2.79
                            B4=19.58
                            K5 < -1.2
Medium: 4 M NaClO4), 3 M HClO4. At 25 C:K1=7.48, B2=13.26, B3=16.65, B4=19.45
______
    cal NaClO4 25°C 7.0M U H
                                       1963AGa (5888) 603
Medium: 4 M NaCl04, 3 M HCl04, DH(K1)=-25.2 kJ mol-1, DH(K2)=-16.9, DH(K3)=-4.5,
DH(K4)=-0.7. DS(K1)=58.1 J K-1 mol-1, DS(K2)=53.9, DS(K3)=49.7, DS(K4)=51.0
Tl+++ EMF oth/un 25°C 3.0M U I K1=7.78 B2=12.87 1963KIa (5889) 604
                            K3=3.29
                            K4=2.16
At I=0.5:K1=7.05, K2=4.97, K3=2.41, K4=1.89, B4=16.32
______
      EMF NaCl04 25°C 3.0M U IH K1=7.18 B2=12.94 1961WGa (5890) 605
T1+++
                            K3=3.15
                            K4=2.22
Medium: HClO4. K1 by solubility. In 0.5 M HClO4 K1=6.78, K2=5.26, K3=2.52,
K4=1.72. In 3 M : DS(K1)=62 J K-1 mol-1, DS(K2)=46, DS(K3)=45, DS(K4)=42
______
Tl+++ ISE NaCl04 30°C 3.0M U K1=7.30 B2=12.48 1960BAb (5891) 606
                           K3 = 3.08
                           K4=2.36
______
Tl+++ ISE NaClO4 20°C 0.40M U
                            K1=7.50 B2=12.00 1960BTc (5892) 607
                            K3=2.75
                            K4=2.25
                            K5=1.95
```

K6=1.75

			H 1960GAc (5893) 608 , DH(B2)=-41.8, DH(B3)=DH(B4)=-46.4
T1+++	ix none 2		K2=3.04 1958HOa (5894) 609 K3=2.08 K4=0.52
T1+++	dis none 3	0°C 0.0 U	1957HVa (5895) 610 B5=17.47
T1+++	ISE NaClO4 2		K1=6.25 B2=11.40 1956PVa (5896) 611 K3=3.10 K4=2.5 K5=2.15 K6=1.80
	dis none 3	0°C 0.0 U	1955HVa (5897) 612 B5=17.56
	con oth/un 2	5°C var U	1950BJa (5898) 613 B4=15.4
T1+++	ISE none 1		K1=8.1 B2=13.60 1949BEa (5899) 614 K3=2.2
*******	******	******	K4=ca.2.2 ************
F- Fluoride;		HL Fluorid	e CAS 7644-39-3 (201)
Metal	Mtd Medium T	emp Conc Cal	Flags Lg K values Reference ExptNo
T1+++	sol non-aq	0°C 100% U	1961CZa (7269) 615 Ks(TlF3(s)=TlF3)=-3.68 Ks(TlF3(s)+F=TlF4)=-1.89
	quid HF, I=0 ******		*********
I- Iodide;			CAS 10034-85-2 (20)
Metal	Mtd Medium T	emp Conc Cal	Flags Lg K values Reference ExptNo
	EMF NaClO4 2 HClO4,4 M Na		1966JOa (8409) 616 B4=35.66
	sol none 2		1957KMa (8410) 617 B4=32.15 K(TlL3(s)+L=TlL4)=0.0

```
EMF oth/un 25°C var U
                                  1906MAa (8411) 618
                        B4=30.29
************************************
                           CAS 7664-41-7 (414)
NH3
                 Ammonia
Ammonia
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ gl R4N.X 25°C 10.0M U
                                  1968LVa (9218) 619
                         B(Ti(OH)2L)=35.3
                         B(Ti(OH)2L2)=40.0
                         B(Ti(OH)2L3)=42.3
                         B(Ti(OH)2L4)=43.8
Medium: 10 M NH4NO3
______
                         K1=4.6 B2=9.30 1967LKb (9219) 620
Tl+++ gl R4N.X 25°C 10.0M U
                         K3 = 2.3
                         K4=1.5
                         B4=13.0
Metal: T1(OH)2+. Medium: NH4NO3
*****************************
NO2-
              HL
                 Nitrite
                           CAS 7782-77-6 (635)
Nitrite:
          Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     kin NaClO4 25°C 0.30M U
                                  1977TGb (9408) 621
                         K(T1L+HL=T1L2+H)=2.81
                         K(T1L2+HL=T1L3+H)=0.84
*******************************
                 Nitrate
NO3 -
             HL
                           CAS 7697-37-2 (288)
Nitrate;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      cal NaClO4 25°C 3.0M U H K1=0.90
                                 1967MKb (9964) 622
(H/Li)ClO4. DH(K1)=0 kJ mol-1, DS=17 J K-1 mol-1
         Tl+++ gl NaClO4 25°C 3.0M U
                         K1=0.90 B2=0.12 1965KYc (9965) 623
                         B3=1.10
                         B(T1H-2L)=-2.10
                         B(T1H-1L2)=-0.32
                         B(T1H-1L3)=-0.40
Medium: LiClO4
Tl+++ sp NaCl04 22°C 2.0M U I K1=0.45
                                 1957BWa (9966) 624
K1=0.30(I=3), 0.67(I=1.15), 0.92(I=0.57)
______
Tl+++ sp oth/un 18°C var U T K1=0.18
                                 1954PEb (9967) 625
```

```
K1=0.41(10 C)
*******************************
                 Azide
                            CAS 7782-79-8 (441)
Azide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ EMF NaClO4 50°C 3.0M U T K1=2.81 B2=4.98 1965VOa (10265) 626
                         B3=5.70
K1=3.08(12.7 \text{ C}), 3.0(20 \text{ C}), 2.90(35 \text{ C}); B2=5.60(12.7 \text{ C}), 5.38(20 \text{ C}),
5.18(35 C); B3=7.04(12.7 C), 6.90(20 C), 6.40(35 C)
**********************************
OH-
                 Hydroxide
              HL
                             (57)
Hydroxide;
        Metal Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
                   - -
_____
      oth none 25°C 0.0 M
                                   1986BGe (12313) 627
                         *K1 = -0.18
Application of specific ion interaction theory (SIT) to literature data.
Ks(0.5T1203(s)+3H=T1+1.5H20)=-2.95
_____
      EMF oth/un 25°C 1.00M U K1=14.41 B2=27.08 1979YRa (12314) 628
Medium: 0.56 mol.parts CH3CN in H2O K1=15.40, B2=28.85
-----
Tl+++ sp none 20°C 0.0 U K2=7.7
                                  1976BAa (12315) 629
_____
Tl+++ kin none 25°C 0.00 U
                                   1974LPb (12316) 630
                         K(Me2T1 + OH)=1.05
                         K(2Me2T1OH=(Me2T1OH)2)=10.18
______
Tl+++ cal NaClO4 25°C 3.00M U H
                                   1973KKg (12317) 631
                         *K1=-1.41
                         *K2=-1.15
Medium: LiClO4. DH(*K1)=100 kJ M-1, DS=310 J K-1 M-1; DH(*K2)=109, DS=343
______
     sp diox/w 25°C 70% U I
                                   1972KKh (12318) 632
                         *K1 = -0.5
Medium: 70% w/w dioxan/H2O, 3 M LiClO4> In 0.5 to 1 M DMSO, 3 M LiClO4,
*K1=-0.6
-----
Tl+++ sp oth/un 20°C U
                                   1971KYb (12319) 633
                         *K1 = -1.51
-----
      sol oth/un 25°C
T1+++
                   U
                                   1970IEb (12320) 634
                         K(T1L3(s)+L)=-5.4
                         K(T1L3(s)+2L)=-5.6
                         K(T1L3(s)+3L)=-6.4
     sol oth/un 22°C 0.02M U I
                                   1970VTa (12321) 635
```

```
Kso(T1(OH)3)=-36.7
```

```
Medium: Tl(NO3)3 at I=0.025(hydrolysis neglected); Kso=-38.3(I=0)
______
Tl+++ sp NaCl04 25°C 0.10M U I K1=12.82 B2=25.27 1969BNc (12322) 636
                           B3=37.46
K1 12.96, B2=25.45, B3=37.70(I=0.3); K1=13.10, B2=25.65, B3=37.98(I=0.5);
K1=13.52, B2=26.33, B3=38.80(I=1)
-----
Tl+++ EMF R4N.X 25°C U K1=15.7 B2=30.7 1968LVa (12323) 637
Medium: NH4NO3. Also data in presence of py, en
-----
T1+++ gl R4N.X 25°C 10.0M U I K1=15.7 B2=30.7 1967LKc (12324) 638
Medium: 10 M NH4NO3. In 2 M Mg(NO3)2: K1=15.40, B2=28.66. In 2 M en(HNO3)2:
K1=15.45, B2-28.64
______
Tl+++ EMF NaClO4 25°C 3.00M U
                                      1964KYb (12325) 639
                           *K1=-1.18
                           *K2=-1.42
Medium: 3 M LiClO4
T1+++ EMF NaCl04 25°C 3.00M U
                                      1963KOb (12326) 640
                           *K1=-1.14
                           *K2=-1.43
Tl+++ sp NaClO4 25°C 3.0M U TI
                                     1961RWa (12327) 641
                           *K1=-1.16
*K1=-1.10(40 C). Same in H2O & D2O. 1.5 M NaClO4 *K1=-1.07(25 C),-1.01(40 C)
______
Tl+++ kin none 25°C 0.0 M
                                      1959LPa (12328) 642
                           K(Me2T1+L)=1.04
                           K(2Me2T1OH=(Me2T1OH)2)=0.3
Tl+++ oth oth/un 32°C satd U
                                     1958VRa (12329) 643
                           K=-6.90(?)
Medium:saturated Na2SO4. K:TlCl3+3H2O=Tl(OH)3+3H+3Cl). Method:freezing point
______
Tl+++ oth none 25°C 0.0 U
                                      1958VSa (12330) 644
                           *Kso(Tl(OH)3)=-2.15
                           *Kso(T1203) = -2.60
Method:combination of thermodynamic data
______
Tl+++ EMF NaCl04 25°C 3.0M U
                                      1957SCd (12331) 645
                           Kso = -45.0
                           *Kso=-2.34
Tl+++ oth oth/un ? var U
                                      1957SKa (12332) 646
                          Kso=-38
Medium: H2SO4. Method: tyndallometry
-----
      EMF NaClO4 25°C 3.0M U
                                      1953BIa (12333) 647
```

*K1=-1.14 *K2=-1.49

				KZ1.47		
T1+++	gl oth/un 2	25°C ?		Kso=-37		(12334) 648
T1+++	kin NaClO4 2	25°C 3.68M		*K1=0.81		(12335) 649
*K1=0.97(3	5 C),1.10(45	C)				
T1+++	kin NaClO4 2	25°C 6.0M	1 U T	*K1=0.51		(12336) 650
*K1=0.72(3	2.2 C),0.84(4	41.8 C)				
T1+++	EMF none 2	25°C 0.0	U	Kso=-45.20	1951SUa	(12337) 651
T1+++	gl none 1	18°C 0.0		Kso=-43.6		(12338) 652
				*K1=-0.2		
T1+++	gl oth/un 2			Kso=-34.1		(12339) 653
T1+++	sol NaClO4 2	25°C var	U	Kso=-43.81 *Kso=-1.81	1936SHa	(12340) 654
	EMF oth/un 2			Kso=-42.90 *Kso=-1.13		(12341) 655
**********				**************************************		
Thiocyanat		1112	.ocyana cc	CA3 403 30	J (100)	'
Metal	Mtd Medium 1	Temp Conc	Cal Flag	s Lg K values	Refer	rence ExptNo
T1+++	kin NaClO4 2	20°C 2.0M	1 U	K(TlL+Tl)=1.65 K(TlL+L+H)=1.38	1990GKb	(15307) 656
T1+++	vlt NaClO4 2	25°C 2.0M	1 U	V/T121 11 _T1212\		(15308) 657
******	******	******	******	K(T12L+L=T12L2) *********		*******
SO3 Sulfite;	ŀ	H2L Sul	fite.	CAS 7782-9	9-2 (801	1)
Metal	Mtd Medium 1	Temp Conc	Cal Flag	s Lg K values	Refer	rence ExptNo
T1+++	EMF oth/un 2	 25°C var	U		 1957ВЈа	(15480) 658

B4=ca.34

SO4 Sulfate;		H2L S	ulfate	CAS 7664-9	3-9 (15)
Metal	Mtd Medium	Temp Con	c Cal Flags	Lg K values	Reference ExptNo
T1+++	cal NaClO4 redox. Med:	25°C 3.0	0M U H LiClO4, 0.5	K1=2.27 5 M HClO4. DH(K1	1967MKb (16606) 659 .)=-11 kJ mol-1,
Tl+++ Medium: Li		25°C 3.0	ЭМ U	K1=1.95 B2=3 K(Tl+HL)=1.23 K(Tl+2HL)=2.12 K(Tl+HL+L)=3.00	
 Tl+++	sol oth/ur	1 25°C va	r U	K1=1	1960HEa (16608) 661
T1+++	kin NaClO4	25°C 3.6	8M U	K1=0.3	1957BMa (16609) 662
			hiosulfate	CAS 73686-	28-7 (177)
Metal	Mtd Medium	Temp Con	c Cal Flags		Reference ExptNo
T1+++	EMF oth/ur	ı 25°C va	r U		1950BJa (16909) 663
SeO3 Selenite;		H2L S	elenite	CAS 7783-0	, ,
			c Cal Flags	Lg K values	Reference ExptNo
	sol oth/ur		r U	Kso(T12L3)=-38.	1959MIa (17076) 664 7
******* C2H2O2 Ethanedial		L G	lyoxal	CAS 107-22	-2 (2017)
		Temp Con	c Cal Flags	Lg K values	Reference ExptNo
T1+++ ******** C2H2O2C12		H 18°C 0.2 ************************************	2M U TI *******	K1=0.56 B2=1	.53 1980IAa (18372) ************************************
 Metal	Mtd Medium	Temp Con	c Cal Flags	Lg K values	Reference ExptNo
 Tl+++	sp oth/ur	15°C 0.5	 ∂M C	K1=0.72	1984CDb (18401) 666

```
Medium: 0.50 M LiCl.
***********************************
                Glyoxylic acid CAS 298-12-4 (1142)
Glyoxylic acid; OHC.COOH
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp none 30°C 0.0 C K1=2.93
                               1984GSf (18431) 667
**********************************
                Oxalic acid CAS 144-62-7 (24)
            H2L
Ethanedioic acid; (COOH)2
------
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     dis NaClO4 20°C 0.10M U
                                1963STc (19111) 668
                       B3=16.9
**********************************
C2H3O2C1
                Chloroacetic
                         CAS 79-11-8 (34)
Chloroethanoic acid; ClCH2.COOH
  Mtd Medium Temp Conc Cal Flags Lg K values
                                 Reference ExptNo
_____
      sp oth/un 15°C 0.50M C K1=1.38
                               1984CDb (19387) 669
Medium: 0.50 M LiCl.
***********************************
                         CAS 141-46-8 (2016)
2-Hydroxyethanal; HO.CH2.CH0
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     kin NaClO4 50°C 0.68M U TI K1=0.26 1980IAa (19514) 670
********************************
C2H402
                Acetic acid CAS 64-19-7 (36)
Ethanoic acid; CH3.COOH
_____
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp oth/un 15°C 0.50M C K1=2.59
                               1984CDb (20205) 671
Medium: 0.50 M LiCl.
------
    EMF NaClO4 25°C 3.0M U
T1+++
                             B2=11.28 1965KYe (20206) 672
                       K1=6.17
                       B3=15.10
                       B4=18.3
                       B(T1(OH)L)=18.41
                       B(T1(OH)L2)=22.9
Medium: LiClO4. K(Tl+HL+L)=7.97, B(Tl(OH)2L)=30.1
********************************
                Glycolic acid CAS 79-14-1 (33)
2-Hydroxyethanoic acid; HO.CH2.COOH
```

Metal	Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	
T1+++	sp none 30°C 0.0 C K1=3.40 B2= 5.00 1984GSf (20638) 67	3
C2H5N02	kin NaClO4 75°C 0.20M U TI K1=0.77 1980IAa (20639) 674 ************************** HL Glycine CAS 56-40-6 (85) anoic acid; H2N.CH2.COOH	
Metal	Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	
******	gl NaCl 37°C 0.15M C M 1988BGa (21736) 675 K(Tl(CH3)2+L)=1.189 ***********************************	
C2H6OS Dimethylsu	L DMSO CAS 67-68-5 (329) lfoxide; (CH3)2.SO	
	Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	
Tl+++	vlt non-aq 25°C 100% U K1=1.3 1976GBa (22126) 676 2Cl2, 0.1M Bu4NClO4; Metal ion (C6F5)2Tl+	
Medium: CH	vlt non-aq 25°C 100% U K1=0.9 1976GBa (22127) 677 2Cl2, 0.1M Bu4NClO4; Metal ion (m-CH3C6H4)2Tl+ ************************************	
C2H6O2 1,2-Dihydr	L Ethyleneglycol CAS 107-21-1 (924) oxyethane (Ethane-1,2-diol); HO.CH2.CH2.OH	
Metal	Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	
******** C2H8N2	kin NaClO4 75°C 0.56M U TI K1=-0.49 1980IAa (22158) 678 ************************ L Ethylenediamine CAS 107-15-7 (23) oethane; H2N.CH2.CH2.NH2	
Metal	Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	
Tl+++ Method: 20	nmr non-aq 25°C 100% U K1=7.6 B2=12.80 2001MIa (23238) 679 K3=2.64 5Tl and 1H NMR spectroscopy. Medium: pyridine, 0.5 M NaClO4.	9
Tl+++ Medium: CH	vlt non-aq 25°C 100% U K1=4.6 1976GBa (23239) 680 2Cl2, 0.1M Bu4NClO4; Metal ion (m-CH3C6H4)2Tl+	
T1+++	vlt non-aq 25°C 100% U K1=6.3 1976GBa (23240) 681 2Cl2, 0.1M Bu4NClO4; Metal ion (C6F5)2Tl+	
T1+++	gl oth/un 25°C 2.0M U 1967LKa (23241) 682 K(Tl(OH)2+L)=13.0 K(Tl+20H+L)=41.64	

```
Medium: 2M L(HNO3)2
**********************************
                         CAS 68-12-2 (598)
N,N-Dimethylformamide; HCO.N(CH3)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     vlt non-ag 25°C 100% U K1=0.7 1976GBa (25667) 683
Medium: CH2Cl2, 0.1M Bu4NClO4; Metal ion (m-CH3C6H4)2Tl+
-----
                           1976GBa (25668) 684
    vlt non-aq 25°C 100% U K1=1.0
Medium: CH2Cl2, 0.1M Bu4NClO4; Metal ion (C6F5)2Tl+
*********************************
                       CAS 107-95-9 (575)
            HL
               B-Alanine
3-Aminopropanoic acid; H2N.CH2.CH2.COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ EMF NaCl04 25°C 1.00M U K1=13.28 B2=24.90 1977YKc (26484) 685
                       B3=37.98
Medium: LiClO4
**********************************
C3H7N02S
            H2L
                Cysteine
                         CAS 52-90-4 (96)
2-Amino-3-mercaptopropanoic acid; H2N.CH(CH2.SH)COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ gl NaCl 37°C 0.15M C
                                1988BGa (26845) 686
                       K(T1(CH3)2+L)=3.621
                       K(T1(CH3)2+H+L)=11.850
                       K(T1(CH3)2+2L)=5.349
C4H606
                L-Tartaric acid CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ EMF NaCl04 20°C 1.0M U K1=11.57 B2=12.81 1962BTb (31377) 687
                      B3=13.34
*********************************
                        CAS 110-86-1 (31)
C5H5N
                Pyridine
Pyridine, Azine;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      vlt non-aq 25°C 100% U K1=1.3
                                1976GBa (36686) 688
Medium: CH2Cl2, 0.1M Bu4NClO4; Metal ion (m-CH3C6H4)2Tl+
_____
      vlt non-aq 25°C 100% U K1=2.0
                            1976GBa (36687) 689
Medium: CH2Cl2, 0.1M Bu4NClO4; Metal ion (C6F5)2Tl+
```

```
gl oth/un 25°C 4.0M U
                                     1966LKb (36688) 690
                          K(T1(OH)2+L)=0.7
                          B(T1(OH)2L)=29.1
                          K(T1(OH)2+2L)=2.4
                          B(T1(OH)2L2)=30.8
Medium: C5H5NHNO3. K(T1(OH)2+4L)=2.5, B(T1(OH)2L4)=31.0
********************************
                  Acetylacetone CAS 123-54-6 (164)
              HL
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ dis NaCl04 20°C 0.10M U K1=8.88 B2=16.88 1969BFb (38105) 691
                          K3=7.82
************************
                  N-Acetyl-Cys CAS 616-91-1 (1187)
C5H9N03S
              H2L
N-Acetylcysteine; CH3.CO.NH.CH(CH2.SH)COOH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl NaCl 37°C 0.15M C M
                                    1988BGa (38819) 692
                          K(T1(CH3)2+L)=2.622
************************************
C5H11N02S
              H2L Penicillamine CAS 52-66-4 (350)
DL-2-Amino-3-mercapto-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
              Tl+++ gl NaCl 37°C 0.15M C
                        М
                                     1988BGa (41285) 693
                          K(T1(CH3)2+L)=3.814
                          K(T1(CH3)2+H+L)=11.853
                          K(T1(CH3)2+2L)=5.217
*******************************
C6H9N06
                  NTA
                             CAS 139-13-9 (191)
              H3L
Nitrilotriethanoic acid; N(CH2.COOH)3
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      EMF NaCl04 20°C 1.0M U I T K1=20.9
                                     1967ABc (47060) 694
Medium: HClO4. In 1 M NaClO4: B2=32.5
______
Tl+++ sp oth/un 20°C ? U
                                     1966KAc (47061) 695
                          K(T1+H2L)=4.38
                          K(T1L+H3L)=5.37
                          K1=18 1965KMc (47062) 696
Tl+++ gl oth/un 25°C 1.0M U I
                          K(T1+3H2L)=17.64
Medium: HNO3. In 1 M NaCl: K(TlCln+H)=2.5
*********************************
```

```
C6H11N05
                           CAS 93-62-9 (192)
             H2L
                 HIMDA
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ gl KCl 20°C 0.10M U
                                   1978VMa (48799) 697
                         K(T1H-1L2+H)=7.53
                         K(T1H-2L2+H)=10.11
______
      sp oth/un 20°C ? U
                         B2=19.24
                                   1971KOc (48800) 698
                         K(T1+2HL)=5.66
**********************************
C6H20N2O12P4
             H8L
                  EDTPA
                            CAS 1429-50-1 (434)
Ethane-1,2-bis(iminobis(methylenephosphonic acid)); ((H2O3PCH2)2NCH2.)2
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
sp NaClO4 20°C 1.00M U
                                   1974KPc (52366) 699
                        K(T1+H5L)=5.74
********************************
                 Tropolone
             HL
                           CAS 533-75-5 (3129)
2-Hydroxycyclohepta-2,4,6-trien-1-one;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     dis non-aq 25°C 100% C
                                   2001NCa (53696) 700
                         K(T1L3+T0P0)=1.35
TOPO is trioctylphosphane oxide. Medium: CCl4.
*********************************
                 Thiosalicylic CAS 147-93-3 (236)
C7H602S
             H2L
2-Mercaptobenzoic acid; HS.C6H4.COOH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl alc/w 25°C 50% U K1=8.96 B2=17.61 1971RFa (53921) 701
                        K3=5.42
**********************************
C7H603
             H2L
                  Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
    gl alc/w 25°C 50% U K1=12.73 B2=24.74 1971RFa (54315) 702
-----
      gl KNO3 25°C 0.10M U K1=12.96
                                  1967ASa (54316) 703
*******************************
                            CAS 5965-83-3 (399)
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; HO3S.C6H3(OH).COOH
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
Tl+++ gl KNO3 25°C 0.10M U K1=12.41 1967ASa (55059) 704
*************************
C7H13N03S
                            CAS 59-53-0 (1269)
N-Acetyl-penicillamine; CH3.CO.NH.CH(COOH)C(CH3)2SH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ gl NaCl 37°C 0.15M C M
                                   1988BGa (57494) 705
                        K(T1(CH3)2+L)=2.628
******************
C8H502F3S
                 TTA
                           CAS 326-91-0 (165)
4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S
                    Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      dis NaClO4 20°C 0.10M U K1=9.11 B2=17.39 1969BFb (58688) 706 K3=7.63
*******************************
C8H12N2O8
                            CAS 35039-85-1 (4537)
1,2-Diaminoethane-N,N'-dimalonic acid; (HOOC)2.CH.NH.CH2.CH2.NH.CH(COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ EMF KNO3 25°C 0.10M U K1=35.78
                                   1973GSd (61529) 707
                        K(T1+HL)=27.80
Using glass/Pt electrodes, values are 35.78, 27.80
*******************************
                       CAS 547-91-1 (275)
             H2L
                 Ferron
7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ sp oth/un 25°C 0.10M U
                                   1968BNb (63831) 708
                        K(T10H+L)=30.1
******************************
              HL Oxine
                           CAS 148-24-3 (504)
8-Hydroxyquinoline (8-quinolinol);
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ sp alc/w 25°C 20% U
                                   1968BNb (64361) 709
                         K(T1(OH)2+L)=10.34
Medium: EtOH
Tl+++ oth oth/un ? ? U
                                   1957PKa (64362) 710
                       Kso = -32.4
***********************************
                 TAR
                           CAS 2246-46-0 (707)
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ sp NaClO4 ? 0.10M U
                                1969HSd (64731) 711
                       K(T1+HL)=13.41
                       K(T1HL+HL)=12.35 in 30%ethanol
-----
Tl+++ sp alc/w 25°C 50% U
                                1967NPb (64732) 712
                       K(Tl+HL)=12.0
Medium: 50% MeOH, 0.1 M NaClO4
*********************************
C9H11NOS
4-Phenylthiourethane; C6H5.NH.CO.S.C2H5
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     kin diox/w 25°C 1% U K1=2.35
                                1992SSd (65671) 713
Constants also for related R.C6H4.NHCOS.Et and R.C6H5.NHCOS.C6H4.R'
********************************
C9H14N2O9
            H4L
                          CAS 56360-11-3 (2576)
2-Hydroxy-1,3-diaminopropane-N,N'-di(1,3-propanedioic acid)
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     EMF KNO3 25°C 0.10M U K1=34.95
K(Tl+HL)=27.12
                                1976DGf (67140) 714
***********************************
C10H8N2
                2,2'-Bipyridyl CAS 366-18-7 (25)
2,2'-Bipyridine; (C5H4N)2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ dis NaNO3 25°C 1.0M U K2=5.5
                               1962KMb (69656) 715
                       K3 = 4.89
                       B3=20.05
 ______
Tl+++ EMF oth/un 25°C 1.0M U K1=9.40 B2=16.10 1961KMa (69657) 716
Benzoylacetone CAS 93-91-4 (197)
             HL
1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     dis NaClO4 20°C 0.10M U
                       K1=11.92 B2=22.84 1969BFb (70778) 717
                      K3=9.76
***********************************
            H4L EDDS
                         CAS 52759-67-8 (1100)
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH2.NH.CH(COOH)CH2.COOH)2
-----
Metal
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
EMF KNO3 25°C 0.10M U
                        K1=35.12
                                 1973GKe (73190) 718
                       K(T1+HL)=28.10
*********************************
C10H16N2O8
            H4L EDTA
                           CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ gl NaCl04 25°C 1.00M C
                                 1995MAa (74238) 719
                        K(T1L+OH)=7.90
                        K(T1L+C1)=2.52
                        K(TlL+Br)=3.70
                        K(T1L+I)=5.47
K(T1L+SCN)=2.94, K(T1L+N3)=3.56, K(T1L+py)=2.58, K(T1L+en)=8.65,
K(TlL+phen)=4.75, K(TlL+bpy)=3.21, K(TlL+Gly)=5.9, K(TlL+IDA)=4.72
______
Tl+++ gl NaClO4 25°C 1.00M C
                                 1989TBa (74239) 720
                        K(T1L(OH)+H)=6.0
                        K(T1L+C1)=2.3
                        K(TlBr)=3.5
                        K(TlL+I)=5.9
______
Tl+++ gl KNO3 25°C 0.10M U
                        K1=35.30 1973GKe (74240) 721
                        K(T1+HL)=27.54
-----
     EMF NaClO4 25°C 1.00M U
T1+++
                                 1971KMe (74241) 722
                        K(T1+CoL)=5.10
                        K(T1+2CoL)=9.97
By spectrophotometry: K(Tl+CoL)=5.02.
     oth NaClO4 25°C 1.0M U
T1+++
                                 1971KMe (74242) 723
                        K(T1+CrL)=5.45
                        K(T1+2CrL)=10.0
Method: platinum electrode. By spectrophotometry, K(Tl+CrL)=5.31
______
    EMF NaClO4 20°C 1.0M U M T K1=37.8 1967ABc (74243) 724
-----
Tl+++ sp oth/un 19^{\circ}C dil U M K1=24.0
                                 1966KAb (74244) 725
                        K(FeL+Tl=TlL+Fe)=0.086
T:18-20
______
Tl+++ gl oth/un 20°C 0.40M U
                                 1960BTa (74245) 726
                       K(T1LOH+H)=7.8
______
Tl+++ gl oth/un 20°C 0.10M U
                        K1=22.5 1960BTd (74246) 727
                      K(T1L+H)=2.30
______
Tl+++ vlt oth/un 20°C 1.0M U K1=5.81 1957BVb (74247) 728
-----
```

```
gl none 15°C 0.0 U
T1+++
                       K1 = 24.95
                                1956STa (74248) 729
                       K(T1L+H)=1.7
**********************************
                HEDTA
           H3L
                          CAS 150-39-0 (392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp oth/un 20°C ? U M K1=19.72 1967KAe (75519) 730
                      K(T1+FeL=T1L+Fe)=0.66
*******************************
                     CAS 4097-90-9 (3315)
C10H28N6
                PENTEN
N,N,N',N'-Tetra-(2-aminoethyl)diaminoethane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl NaClO4 25°C 1.0M C K1=37.1 2001GLb (76881) 731
B(T1HL)=39.7
*****************************
                     CAS 1141-59-9 (636)
C11H9N3O2
            H2L
                PAR
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     sp oth/un ? ? U
                                1971BRd (77590) 732
                      K(T1(OH)2+HL)=24.17
-----
Tl+++ sp NaClO4 ? 0.10M U
                                1969HSd (77591) 733
                     K(T1+HL)=17.93
-----
     sp oth/un 25°C ? U
                                1966DMf (77592) 734
                      K(?)=9.9
***********************************
            H4L
                          CAS 38539-29-0 (2573)
1,3-Diaminopropane-N,N'-di(1,4-butanedioic acid)
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
. - - -
                       K1=26.32 1976DGf (79374) 735
Tl+++ EMF KNO3 25°C 0.10M U
                      K(M+T1L)=20.45
*********************************
C11H18N2O8
                          CAS 4408-81-5 (923)
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.)2.CH2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                 Reference ExptNo
___________
Tl+++ EMF NaCl04 20°C 1.0M U K1=30.9 1967ABc (79473) 736
**********************************
                          CAS 668-21-1 (2562)
2-Hydroxy-1,3-diaminopropane-N,N'-di(1,4-butanedioic) acid
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     EMF KNO3 25°C 0.10M U
T1+++
                        K1 = 29.90
                                 1976DGf (79607) 737
                        K(M+T1L)=22.04
**************
C12H8N2
             L Phenanthroline CAS 66-71-7 (144)
1,10-Phenanthroline;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     dis NaNO3 25°C 1.0M U
                       K2=7.4
                                 1962KMb (80523) 738
                        K3=5.82
                        B3=24.3
Tl+++ EMF oth/un 25°C 1.0M U K1=11.57 B2=18.30 1961KMa (80524) 739
**************************
C12H20N2O8
                           CAS 40623-42-5 (1101)
1,2-Diaminoethane-N,N'-di(2-pentane-1,5-dioic acid); (CH2NHCH(COOH)CH2CH2COOH)2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ gl KNO3 25°C 0.10M U K1=35.25
                                 1973GKe (82104) 740
                        K(T1+HL)=27.85
2nd method: platinum electrode.
********************************
C12H20N208S
            H4L
                 TEDTA
                          CAS 923-74-0 (3394)
2,2'-Thiobis(ethyliminodiethanoic acid); S(CH2.CH2.N(CH2.COOH)2)2
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      EMF NaClO4 20°C 1.0M U I K1=31.8
                                 1967ABc (82478) 741
In 1 M HClO4: K1=32.3
***********************************
C12H20N2O9
            H4L EEDTA
                          CAS 923-73-9 (2112)
Oxa-bis(ethyleneimino)diethanoic acid; ((HOOC.CH2)2N.CH2.CH2)20
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      EMF NaClO4 20°C 1.0M U I K1=32.8
                                 1967ABc (82570) 742
In 1 M HClO4: K1=33.4
______
     sp oth/un 20°C ? U
                       K1=23.08 1967KAc (82571) 743
T1+++
                        K(FeL+Tl=TlL+Fe)=0.51
**********************************
                          CAS 52559-82-7 (8963)
C12H24O3S3
1,4,7-Trioxa-10,13,16-trithiacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
nmr non-aq 25°C 100% C
T1+++
                                     1984KMf (83120) 744
                       Н
                           K((CH3)2T1C104+L)=2.09
Method: 1H nmr. Medium: CD3CN. DH(K)=-12 kJ mol-1, DS(K)=-22 J K-1 mol-1.
_____
Tl+++ nmr non-ag 25°C 100% C H
                                     1984KMf (83121) 745
                           K((C2H5)2T1C104+L)=1.56
Method: 1H nmr. Medium: CD3CN. DH(K)=-21 kJ mol-1, DS(K)=-40 J K-1 mol-1.
*******************************
C12H24O3S3
                             CAS 63919-49-3 (8964)
1,7,13-Trioxa-4,10,16-trithiacyclooctadecane;
______
      Mtd Medium Temp Conc Cal Flags Lg K values
______
Tl+++ nmr non-aq 25°C 100% C H
                                     1984KMf (83122) 746
                           K((CH3)2T1C104+L)=1.68
Method: 1H nmr. Medium: CD3CN. DH(K)=-9.6 kJ mol-1, DS(K)=-62 J K-1 mol-1.
_____
Tl+++ nmr non-ag 25°C 100% C H
                                     1984KMf (83123) 747
                           K((C2H5)2T1C104+L)=1.49
Method: 1H nmr. Medium: CD3CN. DH(K)=-41 kJ mol-1, DS(K)=-110 J K-1 mol-1
***********************************
C12H24O4S2
                             CAS 296-39-9 (4938)
1,4,10,13-Tetraoxa-7,16-dithiacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      nmr alc/w 25°C 100% C H
                                     1984KMf (83146) 748
                           K((CH3)2T1C104+L)=0.93
Method: 1H nmr. Medium: CD3OH. DH(K)=-18 kJ mol-1, DS(K)=-41 J K-1 mol-1.
-----
Tl+++ nmr non-aq 25°C 100% C
                                     1984KMf (83147) 749
                           K((C2H5)2T1C104+L)=>3.0
Method: 1H nmr. Medium: CD3CN.
*******************************
C12H24O4S2
                             CAS 52559-81-6 (8965)
1,4,7,13-Tetraoxa-10,16-dithiacyclooctadecane;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      nmr alc/w 25°C 100% C H
                                     1984KMf (83148) 750
                           K((CH3)2T1C104+L)=1.90
Method: 1H nmr. Medium: CD3OH. DH(K)=-26 kJ mol-1, DS(K)=-50 J K-1 mol-1.
*****************
C12H2405S
               L
                  Thia-18-crown-6 CAS 52559-79-2 (2263)
1-Thia-4,7,10,13,16-pentaoxacyclooctadecane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ nmr non-aq 25°C 100% C
                                     1984KMf (83157) 751
                           K((C2H5)2T1C104+L)=>3.0
```

```
Method: 1H nmr. Medium: CD3CN.
_____
      nmr non-aq 25°C 100% C
                                  1984KMg (83158) 752
                        K((C2H5)2T1C104+L)=>3.0
Method: 1H nmr. Medium: CD3CN.
***********************************
C13H9N307S3
            H3L
                           CAS 2172-27-2 (5007)
1-(2-Thiazolylazo)-2-naphthol-3,6-disulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sp NaClO4 ? 0.10M U B2=21.43
***********************************
                           CAS 1798-14-7 (921)
C13H22N2O8
             H4L
(Pentamethylenedinitrilo)tetraethanoic acid; ((HOOC.CH2)2N.CH2.CH2)2CH2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
    EMF NaClO4 20°C 1.0M U K1=31.3
                              1967ABc (86209) 754
********************************
C14H9N3O3S
             H2L
                           CAS 22026-06-8 (5081)
1-(2'-Thiazolylazo)-2-naphthol-3-carboxylic acid;
                   Cal Flags Lg K values Reference ExptNo
Metal Mtd Medium Temp Conc Cal Flags Lg K values
              ? 40% U
Tl+++ sp mixed
                                  1972BZb (86844) 755
                        K(T1+2HL)=26.65
Medium: 40% v/v HCON(CH3)2, 0.1 M NaClO4
*******************************
C14H15N4OBr
                          CAS 14337-50-9 (5095)
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     sp oth/un ? ? U
                                  1967GUa (87770) 756
                        K(?)=5.59
**********************************
C14H16N4O
                 PAAC
                          CAS 13059-69-3 (5067)
5-Ethylamino-4-methyl-2-(2'-pyridylazo)phenol;
______
                                 Reference ExptNo
     Mtd Medium Temp Conc Cal Flags Lg K values
·
     sp oth/un ?
                 ? U
                                  1967GKb (88021) 757
                       K(?)=6.68
H4L
                          CAS 482-54-2 (200)
C14H22N2O8
                 CDTA
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;
  -----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
Tl+++ gl NaNO3 25°C 1.00M C
                       М
                                   1995MAa (88806) 758
                         K(T1L+OH)=7.20
                         K(TlL+Cl)=1.86
                         K(TlL+Br)=2.80
                         K(T1L+I)=4.79
K(T1L+SCN)=2.21, K(T1L+N3)=3.28, K(T1L+en)=7.68, K(T1L+Hen)=5.87,
K(T1L+phen)=3.64, K(T1L+Hphen)=2.77, K(T1L+bpy)=2.20, K(T1L+oxalate)=2.10
-----
Tl+++ EMF NaCl04 20°C 1.0M U K1=38.3 1967ABc (88807) 759
********************************
                  DTPA
                            CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ EMF NaClO4 20°C 1.0M U I K1=46.0
                                  1967ABc (89417) 760
In 1 M HClO4: K1=48.0
**********************************
             H4L HMDTA
                            CAS 1633-00-7 (920)
1,6-Diaminohexane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.CH2)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ sp oth/un 19^{\circ}C .001M U
                                   1967KAc (89609) 761
                         K(T1+HL)=9.72
                         K(T1+H2L)=2.52
                         K(T1+H3L)=2.28
********************************
C15H10N3O5C1S
             H3L
7-[(2-Hydroxy-5-chlorophenyl)azo]-8-hydroxyquinoline-5-sulfonic
acid;C6H3C1(OH)N=NC9H4N(OH)(SO3H)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
    sp KNO3 25°C 0.10M M K1=25.46 1997PKb (90956) 762
********************************
C15H1006S
                           CAS 17356-57-5 (4058)
Flavonol-2'-sulfonic acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ sp NaCl04 25°C 0.10M U K1=9.2 B2=16.4 1967YTb (90999) 763
*********************************
C15H11N30
              HL
                  PAN
                            CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ sp oth/un ? ? U
                                   1971BRe (91244) 764
                         K(T1(OH)2+L)=16.70
```

```
C15H11N3O4S
            H2L 1-PAN-4S
                             (7010)
2-(2-Pyridylazo)-1-naphthol-4-sulfonic acid;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Tl+++ sp KNO3 25°C 0.10M U K1=14.23 B2=26.62 1980VHa (91327) 765
***********************************
C15H12N2O5
                            CAS 1562-85-2 (5111)
Gallocyanine;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
Tl+++ sp oth/un ? ? U K1=5.50 B2=11.21 1973TPb (91442) 766
By polarography: K1=6.79, B2=11.24
***********************************
C17H14N2O2
                           CAS 4551-69-3 (698)
4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     dis NaCl04 20°C 0.10M U K1=8.2 B2=15.2 1969BFc (95903) 767
B3=21.5
******************************
C17H16N8
                            (5235)
1,5-Di-(1'-methylbenzimidazolyl-2')formazan;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                 Tl+++ dis oth/un ? ? U M 1966LGa (96118) 768
                        K(T1A2+HL=T1A2L+H)=4.05
HA=ethanoic acid
************************************
C18H30N4O12 H6L
                TTHA
                           CAS 869-52-3 (694)
Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Tl+++ gl NaNO3 25°C 1.00M C I
                                  2000CLa (98098) 769
                         K(T1L+H)=4.95
                         K(T1HL+H)=2.61
                         K(T1H2L+H)=1.4
In 1.0 M NaClO4, K(TlL+H)=5.05, K(TlHL+H)=2.55, K(TlH2L+H)=1.75.
______
Tl+++ gl NaNO3 25°C 1.00M C
                                  2000CLa (98099) 770
                         K(T1L+Co)=4.45
                         K(Tll+Ni)=5.68
                         K(T1L+Cu)=6.65
                         K(T1L+Zn)=4.44
K(T1L+Cd)=4.26, K(T1L+Pb)=4.20
```

```
************************************
C22H20N2
                DiMe-naphidine CAS 13138-48-2 (1809)
3,3'-Dimethylnaphthidine, 4,4'-Diamino-3,3'-dimethyl-1,1'-binaphthyl
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp oth/un 25°C dil U K1=3.95 1971CBb (101692) 771
**************************
C25H20N8
                           (5341)
1,5-Di(1'-methylnaphth[1,2-d]imidazolyl-2)formazan;
______
     Mtd Medium Temp Conc Cal Flags Lg K values
______
Tl+++ dis oth/un ? ? U M
                                 1966LGa (103600) 772
                        K(T1A2+HL=T1A2L+H)=2.74
HA=ethanoic acid
*********************************
                            (5342)
1,5-Di(3'-methylnaphth[1,2-d]imidazolyl-2)formazan;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
______
Tl+++ dis oth/un ? ? U M
                                 1966LGa (103601) 773
                       K(T1A2+HL=T1A2L+H)=2.74
HA=ethanoic acid
*********************************
                          CAS 16858-02-9 (933)
N,N,N',N'-Tetrakis-(2-pyridylmethyl)-diaminoethane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Tl+++ gl NaNO3 25°C 1.00M C
                                 1995MAa (104012) 774
                        K(T1L+OH)=8.40
                        K(T1L+C1)=3.56
                        K(TlL+Br)=4.17
                        K(T1L+I)=5.56
K(T1L+SCN)=2.76, K(T1L+N3)=4.33, K(T1L+phen)=2.34, K(T1L+oxalate)=2.9
******************************
                Xylenol orange CAS 63721-85-5 (432)
C31H32N2O13S
            H6L
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulf
onic acid:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp oth/un 25°C 0.10M U
                                 1969BRb (105502) 775
                        K(2T1+2H2L)=8.03
     sp oth/un 25°C ? U
                                1966DMd (105503) 776
                       K(?)=4.8
```

```
C37H44N2O13S
                     H6L
                            MeThymol Blue
                                               (428)
3,3'-Bis(N,N-di(carboxymethyl)aminomethyl)thymolsulfonephthalein;
______
         Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
------
Tl+++ sp oth/un 25°C 0.10M C
                                                        1997ASa (106622) 777
                                        K1eff=3.69
Medium: 0.10 M acetate buffer, pH 5.0.
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EXPLANATORY NOTES
  DATA Flags are :-
       T Data at other TEMPERATURES
        I Data with various BACKGROUNDS
       H Data for THERMOCHEMICAL quantities
       M Data for TERNARY Complexes
  EVALUATION Flags are :-
        T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC
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