

START Experiments recorded for
 from SC-Database on Saturday, 01 January, 2000 at 00:10:19
 Software version = 5.81 Data version = 4.62
 Experiment list contains 101 experiments for
 (no ligands specified)
 2 metals : Nb(V), Nb++++
 (no references specified)
 (no experimental details specified)

e- HL Electron (442)
 Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	vlt	oth/un	25°C	1.00M	U				1965MHb	(709) 1
K'=14.40, 426 mV										

K': Nb6Cl12++++ + 2e=Nb6Cl12++.

Nb(V)	vlt	oth/un	25°C	12MM	U				1954CVa	(710) 2
K(Nb+e)=-3.60, -213 mV										

Medium 12 M HCl

Nb(V)	oth	none	25°C	0.0	U				1952LAb	(711) 3
K=-54.4(-650 mV)										
K(Nb(III)+3e)=-56 (-1100 mV)										

K: 0.5Nb2O4(s)+5H+5e=Nb(s)+2.5H2O. From thermodynamic data

Nb(V)	EMF	oth/un	18°C	6.25M	U	I			1938GGa	(712) 4
K=-10.84(-313 mV)										

Medium:H2SO4. K: Nb+2e=Nb(III). At I=4.95 M: K=-10.74(-310 mV), 3 M: -12.02 (-347 mV). Also at 18 C. At I=0: K(NbO+2H+2e=Nb(III)+H2O)=-11.88(-343 mV)

Nb(V)	EMF	oth/un	25°C	9.87M	U	I			1928KHa	(713) 5
K=-14.41(-426.1 mV)										

Medium: H2SO4. K(Nb+2e=Nb(III). At I=5.9: K=-13.02(-384.9 mV), I=3: -12.62 (-373.0 mV)

Br- HL Bromide CAS 10035-10-6 (19)
 Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	oth/un	25°C	var	M				1973LJb	(2150) 6
K(NbOOHL4+H+L=NbOL5+H2O)=-5.2										

Medium: HBr

Cl- HL Chloride CAS 7647-01-0 (50)
 Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	cal	non-aq	25°C	100%	U	HM		1993DSb (5278)	7	
Metal:Nb(IV). Medium:iso-Propyl ether. DH(Nb(H-1A)2B2(s)+2HL=NbL2(H-1A)2+2HB)=-216.4 kJ mol-1. A:Cyclopentadiene. B:CH3. Also for B=PhS and p-ClC6H4S										
Nb(V)	dis	KCl		var	U			1968SSf (5279)	8	
Kd(Nb+4Cl+3TBP(benzene))=-1.8										

F-		HL		Fluoride				CAS 7644-39-3 (201)		
Fluoride;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	ISE	non-aq	?	100%	C			1978GRa (7039)	9	
K6=6.88										
K(NbF5+NbF6=Nb2F11)=1.32										
Medium: liquid anhydrous HF										
Nb(V)	sp	mixed	?	20%	U			1973LCa (7040)	10	
K(NbH2O2+F)=2.62										
K(NbH2O2+2F)=1.72										
K(NbH2O2+3F)=1.58										
K(NbH2O2+4F)=1.52										
Medium: 20% H2SO4										
Nb(V)	ISE	NaClO4	25°C	0.50M	U			1972LOa (7041)	11	
K(NbOF2+F)=3.80										
K(NbOF3+F)=4.30										
K(NbOF4+F)=4.51										
K(NbOF5+F)=4.67										
Medium: (Na,H)ClO4. Nb(V)=NbO+++ . K(NbOF6+2H+F=NbF7+H2O)=11.4;										
K(NbF7+F)=3.08, K(NbF8+F)=4.0										
Nb(V)	EMF	KCl	25°C	3.0M	U			1970NEb (7042)	12	
K(Nb(OH)2F4+F)=2.51										
Nb(V)	dis	NaClO4	25°C	5.0M	U	T		1969ESa (7043)	13	
K6K7=10.66										
Nb(V)	sp	oth/un	?	17.0M	U		K1=7.12	1969PEc (7044)	14	
Medium: H2SO4										
Nb(V)	sol	KNO3	18°C	0.50M	U			1967Bnd (7045)	15	
Ks(Nb(OH)2F(s))=-5.22										
K(Nb(OH)4F+HF=Nb(OH)4F2+H)=3.6										
K(Nb(OH)4F+F)=6.8										
In 3 M HNO3: Ks(Nb(OH)4F2(s)=Nb(OH)4F2)=-4.82, K(Nb(OH)3F2+HF=Nb(OH)2F3)=4.2										

I-		HL		Iodide				CAS 10034-85-2 (20)		

Iodide;

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Nb(V)      cal non-aq 25°C 100% U   HM                      1993DSb (8273) 16
Metal:Nb(IV). Medium: Toluene or iso-Propyl ether. DH(Nb(H-1A)2B2(s)+2I2=
NbI2(H-1A)+2BI)=-289.9 kJ mol-1. A:Cyclopentadiene. B:CH3.
*****
OH-                HL      Hydroxide                      (57)
Hydroxide;
-----
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Nb(V)      gl  KCl      25°C 3.00M C   I                      1994EFa (11777) 17
                                K(Nb6O19+H=HNb6O19)=13.63
                                K(Nb6O19+2H=H2Nb6O19)=23.55
                                K(Nb6O19+3H=H3Nb6O19)=32.90
Values at I=0 corr: K=16.11, K(Nb6O19+2H)=27.97, K(Nb6O19+3H)=39.91.
K(Nb2O5(s)+5H2O=2Nb(OH)5)=-9.71. K(6Nb(OH)5=H3Nb6O19+5H)=-14.46.
-----
```

```
-----
Nb(V)      dis NaClO4 25°C 0.10M U                      1970GFb (11778) 18
                                *B(NbO2+H2O=NbO2OH+H)=-3.2
Medium: LiClO4
*****
O2--                H2L      Peroxide                      CAS 7772-84-1 (2813)
Peroxide; -O.O-
-----
```

```
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Nb(V)      sp  oth/un  25°C 95%   U T H                      1971VZa (12683) 19
                                K(NbOSO4+H2L)=3.67
Medium:95% H2SO4. K=3.83(15 C), 3.53(35 C), 3.41(45 C). DH=-25 kJ mol-1
63% H2SO4. K=2.73(25 C), 2.56(35 C), 2.42(45 C). DH=-29
-----
```

```
-----
Nb(V)      sp  oth/un  0°C 10%   U                      1969CKa (12684) 20
                                K(NbOSO4+H2L)=5
Medium: 10% H2SO4
-----
```

```
-----
Nb(V)      EMF KCl      0°C 1.0M U                      1969SPc (12685) 21
                                K(3NbO2L+H=HNb3O6L3)=13.08
K(3NbL3+3H2O+H=HNb3O3L6+3H2L)=4.50. In 3 M KCl: K(NbL4+OH=NbOL3+HL)=1.5,
K(HNbOL3+H2L+OH=NbL4+2H2O)=3.3
-----
```

```
-----
Nb(V)      sp  oth/un  ?   var   U                      1966BNa (12686) 22
                                K(Nb(OH)4HL+H)=2.7
-----
```

```
-----
Nb(V)      sp  mixed  23°C 97%   U                      1957AHb (12687) 23
                                K(2Nb(V)+3H2L)=12.70
Medium: 97.2% H2SO4.
-----
```

Nb(V) sp oth/un ? 96% U 1956SSc (12688) 24
 $K(\text{Nb(V)}+\text{H}_2\text{L})=3.6$

 SCN- HL Thiocyanate CAS 463-56-9 (106)
 Thiocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	alc/w	?	100%	U	I		K1=3.58 B2=6.74 B3=9.23	1964GSc (15186)	25

Medium: MeOH. In BuOH: K1=4.37, B2=8.58, B4=16.92. In Me2NCHO: K1=3.08, B2=6.11, B3=8.92, B4=11.55, B5=14.45, B6=16.72, B7=19.28. Nb added as NbCl5

 SO4-- H2L Sulfate CAS 7664-93-9 (15)
 Sulfate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	ix	oth/un	20°C	var	U			$K(\text{NbO}(\text{OH})_2+\text{L})=1.7$	1969MNC (16395)	26

Medium: H2SO4. By distribution: $K(\text{NbO}(\text{OH})_2+2\text{L}+2\text{H}=\text{NbOL}_2+2\text{H}_2\text{O})=3.12$
 In NH42SO4: $K(\text{NbO}(\text{OH})_3\text{L}+\text{L}+\text{H}=\text{NbO}(\text{OH})_2\text{L}+\text{H}_2\text{O})=1.09$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	oth/un	22°C	10.0M	U			$B(\text{Nb(III)}_4\text{Nb}_2)=4.3?$	1966GAc (16396)	27

Medium: H2SO4

 CH4O L Methyl alcohol CAS 67-56-1 (597)
 Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	EMF	alc/w	20°C	100%	U			$K(\text{Nb}+3\text{L}=\text{Nb}(\text{H}-1\text{L})_3+3\text{H}) > 1$ $K(\text{Nb}(\text{H}-1\text{L})_3+\text{H}-1\text{L})=14.3$	1971GSa (17891)	28

Medium: MeOH, 1 M Me4NCl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	EMF	alc/w	20°C	100%	U	M		$K(\text{NbA}(\text{L}')_4+2\text{L}'=\text{Nb}(\text{L}')_6+\text{A})=5.1$ $K(\text{Nb}(\text{H}-1\text{L})_4+\text{A})=10.84$ $K(\text{NbA}(\text{H}-1\text{L})_3+\text{L}')=12.4$ $K(\text{Nb}(\text{L}')_5+\text{HA}=\text{NbA}(\text{L}')_4+\text{L}')=5.18$	1965GBa (17892)	29

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. L'=H-1L; HA=acetylacetone

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	EMF	alc/w	20°C	100%	U	M		$K(\text{NbAL}'_3+\text{H}_2\text{A}+\text{L}'=\text{NbA}_2\text{L}'_2)=13.9$ $K(\text{NbA}_2\text{L}'_2+\text{H}_2\text{A}+\text{L}'=\text{NbA}_3\text{L}')=7.0$ $K(\text{NbA}(\text{L}')_3+\text{L}'=\text{NbA}(\text{L}')_4)=7.89$ $K(\text{NbAL}'_4+\text{NbAL}'_3=\text{Nb}_2\text{A}_2\text{L}'_7)=2.5$	1965GBa (17893)	30

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. L'=H-1L; H2B=catechol

Nb(V) EMF alc/w 20°C 100% U 1964GUa (17894) 31
K(Nb(H-1L)4+H-1L)=10.45
K(Nb(H-1L)5+H-1L)=5.45
K(Nb(H-1L)6+H=Nb(H-1L)5)=6.15
K(Nb(H-1L)7+H=Nb(H-1L)6)=11.15

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl

Nb(V) EMF alc/w 20°C 100% U 1964GUa (17895) 32
K(NbO(H-1L)2+H-1L)=10.51
K(NbO(H-1L)4+H)=6.03

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl

C2H2NBr L CAS 590-17-0 (4217)
Cyanomethyl bromide; Br.CH2.CN

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) nmr non-aq -60°C 100% U M 1971MBa (18364) 33
K(NbCl5A+L=NbCl5L+A)=0.99
K(NbCl5B+L=NbCl5L+B)=0.43

Medium: CHCl3. A=cyanomethyl fluoride, B=cyanomethyl chloride

Nb(V) nmr non-aq -60°C 100% U M 1971MBa (18365) 34
K(NbCl5A+L=NbCl5L+A)=0.55
K(NbCl5A+L=NbCl5L+A)=0.37

Medium: CHCl3. A=cyanomethyl fluoride, B=POCl3

C2H2NI L CAS 624-75-9 (4219)
Cyanomethyl iodide; I.CH2.CN

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) nmr non-aq -60°C 100% U M 1972MBb (18366) 35
K(NcCl5A+L=NbCl5L+A)=0.75
K(NbCl5B+L=NbCl5L+B)=1.18
K(NbCl5C+L=NbCl5L+C)=0.64

Medium: CHCl3. A=cyanomethyl bromide, B=cyanomethyl chloride, C=Et2O

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sol NaClO4 ? ? U 1970ZPa (18980) 36
K3=6.17

Metal ion is NbO+++. Medium : HClO4

Nb(V) sol oth/un 18°C 0.50M U B2=35.9 1968BMb (18981) 37
Metal ion is NbO+++

Nb(V) sol oth/un 18°C 0.50M U 1968Bmb (18982) 38
 $K(NbO(OH)_2 + L) = 9.08$

Nb(V) dis NaCl 20°C 4.50M U 1967K0d (18983) 39
 Medium: 4.5(NaCl or NaNO₃+2.5 M H). $K(Nb(OH)_4 + H_2L = Nb(OH)_4HL + H) = 3.55$
 $K(Nb(OH)_4 + 2H_2L = Nb(OH)_2L_2 + 2H) = 5.13$

Nb(V) EMF oth/un 25°C 0.50M U 1967NSb (18984) 40
 $K(Nb(OH)_4 + 2HL) = 12.11$
 $K(Nb(OH)_4 + 2HL + L) = 17.15$

C2H3N L Cyanomethane CAS 75-05-8 (1399)
 Acetonitrile; CH₃CN

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1974Gma (19189)	41
-------	-----	--------	-------	------	---	---	--	-----------------	----

$K(NbBr_5A + L = NbBr_5L + A) = 0.15$

Medium: CH₂Cl₂. A=t-butylcyanide

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1972MBb (19190)	42
-------	-----	--------	-------	------	---	---	--	-----------------	----

$K(NbCl_5A + L = NbCl_5L + A) = 0.46$

Medium: CHCl₃. A=1,4-dioxan

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1971MBa (19191)	43
-------	-----	--------	-------	------	---	---	--	-----------------	----

$K(NbCl_5A + L = NbCl_5L + A) = 0.76$

Medium: CHCl₃. A=1-chloro-4-cyanobenzene. $K = 0.34$, A=cyanobenzene;
 $K = 0.68$, A=cyanoethane; $K = 0.38$, A=dimethylether.

C2H6NOC12P L CAS 667-43-0 (910)
 Dichloro(dimethylamine)phosphine oxide; (CH₃)₂N.P(O)Cl₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1974Gma (21900)	44
-------	-----	--------	-------	------	---	---	--	-----------------	----

$K(NbBr_5A + L = NbBr_5L + A) = 0.72$

Medium: CH₂Cl₂, A=acetonitrile

C2H6O L CAS 115-10-6 (4214)
 Dimethyl ether; CH₃O.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Nb(V)	nmr	non-aq	-40°C	100%	U	M		1972MBb (22019)	45
-------	-----	--------	-------	------	---	---	--	-----------------	----

$K(NbCl_5A + L = NbCl_5L + A) = 0.08$

Medium: CHCl₃. A=dioxan. $K = 0.38$, A=1-chloro-4-cyanobenzene (-60 C)

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1971MBa (22020)	46
-------	-----	--------	-------	------	---	---	--	-----------------	----

$K(NbCl_5A + L = NbCl_5L + A) = 0.38$

Medium: CHCl₃. A=1-chloro-4-cyanobenzene

C2H6S L CAS 75-18-3 (151)

Dimethyl sulfide; CH₃.S.CH₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U	M	1974G	Ma	(22190)	47
-------	-----	--------	-------	------	---	---	-------	----	---------	----

$$K(\text{NbBr}_5\text{A}+\text{L}=\text{NbBr}_5\text{L}+\text{A})=1.52$$

Medium: CH₂Cl₂, A=t-butylcyanide

Nb(V) nmr non-aq -60°C 100% U M 1972MBb (22191) 48

$$K(NbCl_5A+L=NbCl_5L+A)=1.32$$

Medium: CHCl₃. A=t-butyl nitrile

C3H4O4	H2L	Malonic acid	CAS 141-82-2 (79)
--------	-----	--------------	-------------------

Propanedioic acid; $\text{CH}_2(\text{COOH})_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	dis	NaCl	20°C	4.50M	U	T	1967K	Od	(24509)	49
-------	-----	------	------	-------	---	---	-------	----	---------	----

Medium: 4.5(NaCl or NaNO₃+2.5 HCl). $K(Nb(OH)_4+H_2L=Nb(OH)_4HL+H)=1.72$

C3H9O4P L CAS 512-56-1 (2431)

Trimethyl phosphate; (CH₃)₃P:O

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U	M	1972Bm	b (28024)	50
-------	-----	--------	-------	------	---	---	--------	-----------	----

$$K(NbCl_5L+A=NbCl_5A+L) > 2.0$$

$K(NbCl_5OPCl_3 + L = NbCl_5L + OPCl_3) > 7.0$,. A=tris(dimethylamino)phosphine oxide

Medium: CHCl₃

C4H6O4	H2L	Succinic acid	CAS 110-15-6	(112)
--------	-----	---------------	--------------	-------

1,4-Butanedioic acid; $\text{HOOC} \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	---	--------	-----------	--------

Nb(V) dis NaCl 20°C 4.50M U 1967K0d (30004) 51

$K(Nb(OH)_4 + H_2L = Nb(OH)_4HL + H) = 1.53$, Medium: (NaCl or $NaNO_3 + 2.5 HCl$)

C4H6O5	H2L	Malic acid	CAS 617-48-1	(393)
--------	-----	------------	--------------	-------

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; $\text{HOOC} \cdot \text{CH}_2 \cdot \text{CH}(\text{OH}) \cdot \text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K	values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	---	--------	-----------	--------

Nb(V) dis NaCl 20°C 4.50M U 1967K0d (30682) 52

$K(Nb(OH)_4 + H_2L = Nb(OH)_4HL + H) = 2.01$. Medium: NaCl or NaBO₃ + 2.5 M HCl

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)

L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; $\text{HOOC} \cdot \text{CH}(\text{OH}) \cdot \text{CH}(\text{OH}) \cdot \text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	dis	KCl	20°C	1.00M	U			B2=33.5	1971LFc (31314)	53
-------	-----	-----	------	-------	---	--	--	---------	-----------------	----

Metal ion: NbO^{+++}

Nb(V)	oth	oth/un	?	?	U			K1=25.15 B2=33.00	1969EMa (31315)	54
-------	-----	--------	---	---	---	--	--	-------------------	-----------------	----

Metal ion: NbO^{+++}

Nb(V)	dis	NaCl	20°C	4.50M	U				1967K0d (31316)	55
-------	-----	------	------	-------	---	--	--	--	-----------------	----

$\text{K}(\text{Nb}(\text{OH})_4 + \text{H}_2\text{L} = \text{Nb}(\text{OH})_4\text{HL} + \text{H}) = 2.34$. Medium: NaCl or NaNO_3 + 2.5 M HCl

C4H7NO4 H2L Aspartic acid CAS 56-84-8 (21)
Aminobutanedioic acid; $\text{H}_2\text{N} \cdot \text{CH}(\text{CH}_2 \cdot \text{COOH}) \cdot \text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	dis	KNO_3	20°C	2.00M	U				1969K0b (31898)	56
-------	-----	----------------	------	-------	---	--	--	--	-----------------	----

$\text{K}(\text{NbO}_2 + \text{H}_2\text{L}) = 4.82$

Medium: HNO_3

C4H8OS L 1,4-Thioxane CAS 15980-15-1 (4266)
1,4-Oxathiane; $\text{cyclo}(-\text{O} \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{S} \cdot \text{CH}_2 \cdot \text{CH}_2 -)$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U	M			1972MBb (33190)	57
-------	-----	--------	-------	------	---	---	--	--	-----------------	----

$\text{K}(\text{NbCl}_5\text{A} + \text{L} = \text{NbCl}_5\text{L} + \text{A}) = 0.08$

Medium: CHCl_3 . A=t-butyl nitrile

C4H8S L CAS 110-01-0 (150)
Tetrahydrothiophene; $\text{cyclo}(-\text{CH}_2 \cdot \text{CH}_2 \cdot \text{S} \cdot \text{CH}_2 \cdot \text{CH}_2 -)$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U				1972MBb (33739)	58
-------	-----	--------	-------	------	---	--	--	--	-----------------	----

$\text{K}(\text{NbCl}_5\text{A} + \text{L} = \text{NbCl}_5\text{L} + \text{A}) = 1.80$

A=t-butyl mercaptan. Medium: CHCl_3

C4H8S2 L 1,4-Dithiane CAS 505-29-3 (4255)
1,4-Dithiane; $\text{cyclo}(\text{S} \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{S} \cdot \text{CH}_2 \cdot \text{CH}_2 -)$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	nmr	non-aq	-60°C	100%	U				1972MBb (33743)	59
-------	-----	--------	-------	------	---	--	--	--	-----------------	----

$\text{K}(\text{NbCl}_5\text{A} + \text{L} = \text{NbCl}_5\text{L} + \text{A}) = 0.16$

A=t-butyl nitrile. Medium: CHCl_3

C4H10O L Ether CAS 60-29-7 (3573)

Diethyl ether (ethyl ether, ethoxyethane); C₂H₅.O.C₂H₅

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	nmr	non-aq	-60°C	100%	U	M		1972MBb (34652)	60	
K(NbCl ₅ A+L=NbCl ₅ L+A)=0.11										
A=cyanomethyl bromide. Medium: CHCl ₃ . When A=cyanomethyl chloride, K=0.54										

C ₄ H ₁₀ S		L						CAS 352-93-2 (4259)		
Diethyl sulfide; C ₂ H ₅ .S.C ₂ H ₅										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	nmr	non-aq	-60°C	100%	U	M		1974GMa (34719)	61	
K(NbBr ₅ A+L=NbBr ₅ L+A)=0.59										
A=t-butyl nitrile. Medium: CH ₂ Cl ₂										

Nb(V)	nmr	non-aq	-40°C	100%	U	M		1972MBb (34720)	62	
K(NbCl ₅ A+L=NbCl ₅ L+A)=0.66										
A=dimethyl ether. Medium: CHCl ₃ . When A=acetone, K=0.88										

C ₅ H ₈ O ₂		HL		Acetylacetone				CAS 123-54-6 (164)		
Pentane-2,4-dione; CH ₃ .CO.CH ₂ .CO.CH ₃										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	EMF	non-aq	20°C	100%	U	M		1971GSa (38035)	63	
K(NbA ₅ +HL=NbA ₄ L+HA)=5.18										
K(NbA ₃ L+A)=12.40										
K(NbA ₄ L+2A=NbA ₆ +L)=5.06										
Medium: CH ₃ OH, 1 M Me ₄ NC ₁ . A=CH ₃ OH										

C ₅ H ₉ N		L		t-Butylnitrile				CAS 7188-38-7 (913)		
t-Butylcyanide;(CH ₃) ₃ C.CN										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	nmr	non-aq	-60°C	100%	U	M		1974GMa (38453)	64	
K(NbBr ₅ A+L=NbBr ₅ L+A)=1.68										
Medium: CH ₂ Cl ₂ , A=dimethylether										

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1972MBb (38454)	65	
K(NbCl ₅ A+L=NbCl ₅ L+A)=0.42										
Medium: CHCl ₃ . A=dimethyl ether. When A=1,4-dioxan, K=0.50										

Nb(V)	nmr	non-aq	-60°C	100%	U	M		1971MBa (38455)	66	
K(NbCl ₅ A+L=NbCl ₅ L+A)=0.04										
Medium: CHCl ₃ . A=cyanomethane. When A=cyanoethane, K=0.72										

C ₆ H ₆ N ₂ O ₂		HL		Cupferron				CAS 135-20-6 (637)		

N-Nitrosophenylhydroxylamine; C₆H₅.N(OH).NO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	sp	alc/w	25°C	50%	U				1967LSb (43419)	67
									K(NbOL2+L)=4.83	

Medium: 50% EtOH, 0.1 M (NH₄)₂SO₄

C₆H₆O₂ H₂L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C₆H₄.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	EMF	alc/w	20°C	100%	U	M			1971GSa (43797)	68
									K(NbA3L+A)=7.89	
									K(NbA3L+NbA4L=Nb2A7L2)=2.50	
									K(NbA3L+H2L+A=NbA2L2+2HA)=13.9	
									K(NbA2L2+H2L+A=NbAL3+2HA)=6.98	

Medium: MeOH, 1.0 M Me₄NC₄l. HA=CH₃OH

C₆H₆O₃ H₃L Pyrogallol CAS 87-66-1 (696)
1,2,3-Trihydroxybenzene; C₆H₃(OH)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	sp	oth/un	23°C	96%	U				1981BVa (43971)	69
									K(Nb(V)+H3L)=5.62	

Medium: 96% H₂SO₄. In 85%: K(Nb(V)+H₃L)=2.30

C₆H₇NO₃S HL CAS 599-71-3 (4398)
Benzenesulfohydroxamic acid; C₆H₅.SO₂.NH.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	sp	oth/un	20°C	1.0M	U	M			1971GVa (45071)	70
									Medium: 1-10 M HCl. K(NbOC15+2H2L=(H2L)2NbOC15)=4.91	

C₆H₈O₆ H₂L Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	sp	oth/un	?	?	U		K1=9.4		1966SAb (45649)	71
-------	----	--------	---	---	---	--	--------	--	-----------------	----

C₆H₈O₇ H₃L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH₂.CH(OH)(COOH).CH₂COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Nb(V)	dis	oth/un	20°C	4.50M	U				1967K0d (46195)	72
-------	-----	--------	------	-------	---	--	--	--	-----------------	----

Medium: 4.5(NaCl or NaNO₃+2.5 HCl). K(Nb(OH)₄+H₃L=Nb(OH)₄H₂L+H)=2.94 ?

C6H12O7 HL Gluconic acid CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH₂(CHOH)₄.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) gl oth/un 25°C 0.10M U 1968L0b (49738) 73

K(Nb(OH)_n+L)=2.78

K(Nb(OH)_nH-1L+H)=7.82

C7H4NCl L CAS 1885-81-0 (4433)

1-Chloro-4-cyanobenzene; Cl.C₆H₄.CN

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) nmr non-aq -60°C 100% U M 1971MBa (52380) 74

K(NbCl₅A+L=NbCl₅L+A)=1.91

A=cyanomethyl bromide. When A=cyanomethyl iodide, K=1.16

C7H5N L Cyanobenzene CAS 100-47-0 (4406)

Cyanobenzene, benzonitrile; C₆H₅.CN

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) nmr non-aq -60°C 100% U M 1971MBa (52569) 75

K(NbCl₅A+L=NbCl₅L+A)=0.04

Medium: CHCl₃. A=dimethyl ether

C7H6O3 H2L Salicylic acid CAS 69-72-7 (14)

2-Hydroxybenzoic acid, Salicylic acid; HO.C₆H₄.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un 25°C 0.04M U 1962BVa (54266) 76

K(NbO₂+2L)=22.60

C7H6O4 H3L Protocatechuic CAS 99-50-3 (875)

3,4-Dihydroxybenzoic acid; C₆H₃(OH)₂.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un 20°C 0.10M U 1964SHa (54685) 77

K(NbO₂+H₃L=NbO₂LH₂+H)=2.3

K(NbO₂LH₂+H₃L=NbO₂LH₂+H)=1.3

K(NbO₂+3HL+4H)=63.1

C7H7NO3 H2L CAS 89-73-6 (204)

2-Hydroxybenzohydroxamic acid (salicylhydroxamic acid); HO.C₆H₄.CO.NHOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	oth/un	20°C	dil	U	M		$K(NbOCl_5 + 2H_3L) = 4.58$	1972LVa (55604)	78

C9H6N2O5S		H2L						CAS 5263-74-1	(2738)	
7-Nitroso-8-hydroxyquinoline-5-sulfonic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	gl	alc/w	27°C	50%	C	H		$K_1 = 6.15$ $B_2 = 11.53$	1986EAa (63877)	79

C9H7N3O2S		H2L		TAR				CAS 2246-46-0	(707)	
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	alc/w	25°C	50%	U			$K(NbO_3 + H_2L) = 9.5(?)$	1967NPb (64716)	80
Medium: 50% MeOH, 0.1 M NaClO4										

C10H8O8S2		H4L		Chromotropic ac				CAS 148-25-4	(1875)	
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	NaCl	20°C	0.10M	U	I		$K(NbO_2 + 3L + 4H) = 64.7$	1964SHa (69962)	81
In 3 M NaClO4: $K(NbO_2 + 2H + 2L) = 42.5$										

C10H12O5		H3L						CAS 121-79-9	(3895)	
3,4,5-Trihydroxybenzoic acid propyl ester; (HO)3.C6H2.CO.O.CH2.CH2.CH3										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	mixed	22°C	5%	U			$K(?) = 3.48$	1968LSc (71685)	82
Medium: 5% 1-PrOH, carbonate buffer										

C10H16N2O8		H4L		EDTA				CAS 60-00-4	(120)	
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequesteric acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	dis	KNO3	20°C	2.0M	U			$K(NbO_2 + H_3L) = 10.54$	1969KKf (73986)	83

Nb(V)	vlt	KCl	?	0.40M	U			$K_1 = 39.4$	1969SVd (73987)	84

Nb(V)	vlt	oth/un	20°C	1.0M	U				1967VSd (73988)	85

$$K(\text{Nb}(\text{OH})_2 + \text{L}) = 40.78$$

C11H9N3O2 H2L PAR CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un 25°C ? U 1967ADa (77564) 86

$$K(?) = 4.3$$

C13H11NO2 HL CAS 304-88-1 (181)
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) dis KCl 20°C 1.0M U 1971LFc (85167) 87

$$B(\text{NbO})\text{L}_3 = 53.1$$

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un ? ? U 1968ADa (86745) 88

$$B(\text{NbO})\text{L}_2 = 8.33$$

C14H15N4OBr HL CAS 14337-50-9 (5095)
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un ? ? U M 1969EMa (87766) 89

$$K(\text{NbOA} + \text{L}) = 20.63$$

H2A=tartaric acid.

C14H16N4O HL PAAC CAS 13059-69-3 (5067)
5-Ethylamino-4-methyl-2-(2'-pyridylazo)phenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un ? ? U M 1969EMa (88019) 90

$$K(\text{NbOA} + \text{L}) = 21.0$$

H2A=tartaric acid

C14H22N2O8 H4L CDTA CAS 482-54-2 (200)
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) vlt oth/un 25°C 2.0M U 1970PLa (88728) 91

K(?)=15.60 pH 5

C15H13NO2 HL CAS 7369-44-0 (4066)

N-3-Diphenylpropenohydroxamic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) dis KCl 20°C 1.0M U 1971LFC (91639) 92

K(NbO+3L)=59.7

C15H18N4O HL CAS 14337-52-1 (5124)

5-Diethylamino-2-(2-pyridylazo)phenol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un ? ? U M 1969EMa (92097) 93

K(NbOA+L)=20.63

H2A=tartaric acid

C16H11N2O5ClS H3L CAS 3567-23-5 (5202)

5-Chloro-2-hydroxy-3-(2-hydroxy-1-naphthylazo)-benzenesulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) EMF oth/un ? ? U M 1971ENC (92770) 94

K(NbOA+L)=27.5

H2A=tartaric acid

C16H11N3O10S2 H4L (5174)

2-Hydroxy-1-(2'-hydroxy-4'-nitro)phenylazo-3,6-disulfonaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un 25°C ? U 1971RCd (92882) 95

K(?)=5.51

C17H17NO3 HL CAS 58434-59-6 (1213)

2'-Hydroxy-4-methoxy-5'-methylbenzylidene acetophenone oxime

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un 30°C 8.00M U M 1980GKa (96190) 96

K(NbO(SCN)+L)=2.67

K(NbO(SCN)L+L)=2.27

C17H21N5O HL (5223)

3-Amino-1-hydroxy-6-(2-N-methylanabasiny1-alpha-azo)benzene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Nb(V) sp oth/un ? ? U 1967TTa (96389) 97
 $K(?)=11.36$

 C31H32N2O13S H6L Xylenol orange CAS 63721-85-5 (432)
 5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sul
 fonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb(V)	sp	oth/un	25°C	?	U			$K(?)=4.7$	1967ADc (105482)	98

 CH4O L Methyl alcohol CAS 67-56-1 (597)
 Methanol; CH3.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb+++	EMF	alc/w	20°C	100%	U			$K(Nb(L')_2+L')=12.6$ $K(2Nb(L')_3+3L'=Nb_2(L')_9)=23.9$	1971GSa (17896)	99

Medium: MeOH, 1 M LiCl

 C11H9N3O2 H2L PAR CAS 1141-59-9 (636)
 4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb+++	sp	oth/un	?	?	U	M		$K(NbOA+L)=21.22$	1969EMa (77565)	100

H2A=tartaric acid

 C22H14N4O16Cl2S4 H8L ClSulfofenol S CAS 2103-73-3 (4156)
 2,7-Bis(5'-chloro-2'-hydroxy-3'-sulfo-phenylazo)chromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Nb+++	sp	KNO3	20°C	0.20M	U			$B(NbO_2+6H+L)=53.0$	1965BSe (101484)	101

Metal: Nb(III)

REFERENCES

- 1994EFa N Etxebarria, L Fernandez, J Madariaga; J.Chem.Soc., Dalton Trans., 3055 (1994)
 1993DSb H Diogo, J Simoni et al; J.Am.Chem.Soc., 115, 2764 (1993)
 1986EAa M El-Haty, F Adam; Bull.Soc.Chim.Fr., I, 351 (1986)
 1981BVa F Bajraktari, D Veselinovic et al; Bull.Soc.Chim.Beograd, 46, 291 (1981)
 1980GKa S Gholse, R Kharat; Indian J.Chem., 19A, 823 (1980)
 1978GRa R Gut, J Rueede; J.Coord.Chem., 8, 47 (1978)
 1974GMA R Good, A Merbach; Helv.Chim.Acta, 57, 1192 (1974)

- 1973LCa G Latysh,A Chernyak,T Serebrennikova; Zh.Neorg.Khim.,18,1014(E:533)
(1973)
- 1973LJb J Land,M Jackson; J.Less Common Metals,33,283 (1973)
- 1972BMb J Bunzli,A Merbach; Helv.Chim.Acta,55,2867 (1972)
- 1972LOa J Land,C Osborne; J.Less Common Metals,92,147 (1972)
- 1972LVA F Lobanov,V Voskov,I Gibalo et al; Zh.Neorg.Khim.,17,10,2683 (1972)
- 1972MBb A Merbach,J Bunzli; Helv.Chim.Acta,55,580 (1972)
- 1971ENc S Elinson,T Nezhnova; Zh.Anal.Khim.,26,8,1535 (1971)
- 1971GSa R Gut,E Schmid,J Serrallach; Helv.Chim.Acta,54,593;609 (1971)
- 1971GVA I Gibalo,V Voskov,L Kulikova et al; Zh.Anal.Khim.,26,11,2145 (1971)
- 1971LFC F Lobanov,V Feskova,I Gibalo; Zh.Neorg.Khim.,16,3,776 (1971)
- 1971MBA A Merbach,J Bunzli; Helv.Chim.Acta,54,2536;2543 (1971)
- 1971RCd L Rygalov,A Chibisov; Zh.Anal.Khim.,26,5,850 (1971)
- 1971VZa V Vasilev,G Zaitseva,V Malakhova;
Zh.Neorg.Khim.,16,675;2142(E:360;1144) (1971)
- 1970GFb R Guillaumont,J Franck et al; Radiochem.Radioanal.Lett.,4,73 (1970)
- 1970NEb G Neumann; Ark.Kemi.,32,229 (1970)
- 1970PLa G Popa,R Lereh,I Cazacu; An.Univ.Bucuresti,Chim.,19,9 (1970)
- 1970ZPa E Zhurennikov,D Pobezhimoskaya; Radiokhim.,12,1,105 (1970)
- 1969CKa A Chernyak,V Khomutnikov,A Batsuev et al; Zh.Neorg.Khim.,14,1251(E:655)
(1969)
- 1969EMA S Elison,L Maltseva; Zh.Anal.Khim.,24,10,1524 (1969)
- 1969ESa J Erskine,M Sink,L Varga; Anal.Chem.,41,70 (1969)
- 1969KKf H Koch,H Kupsch; Z.Naturforsch.,24B,398 (1969)
- 1969KOb C Konecny; Z.Phys.Chem.,240,196;225 (1969)
- 1969MNC E Mazurenko,B Nabivanets; Zh.Neorg.Khim.,14,3286(E:1732) (1969)
- 1969PEc A Pilipenko,O Eremenko; Ukr.Khim.Zh.,35,681 (1969)
- 1969SPc B Spinner; Rev.Chim.Minerales,6,319 (1969)
- 1969SVd V Sochevanov,G Volkova; Zh.Neorg.Khim.,14,118 (1969)
- 1968ADa B Agarwala,A Dey; Chim.Anal.(Paris),50,233 (1968)
- 1968BMb A Babko,E Mazurenko,B Nabivanets; Zh.Neorg.Khim.,13,3,718 (1968)
- 1968LOb J Land,C Osborne; J.Less Common Metals,14,349 (1968)
- 1968LSc J Land,C Stillwell; J.Less Common Metals,14,231 (1968)
- 1968SSf V Startsev,Y Sannikov,S Stroganov et al; Zh.Neorg.Khim.,13,1608 (1968)
- 1967ADa B Agarwala,A Dey; Curr.Sci.,36,544 (1967)
- 1967ADc B Agarwala,A Dey; J.Indian Chem.Soc.,44,691 (1967)
- 1967BNd A Babko,B Nabivanets,V Lukachina; Zh.Neorg.Khim.,12,2965 (1967)
- 1967KOD C Kovecny; Z.Phys.Chem.,235,39 (1967)
- 1967LSb J Land,J Sanchez-Caldas; J.Less Common Metals,12,41 (1967)
- 1967NPb G Nickless,F Pollard,T Samuelson; Anal.Chim.Acta,39,37 (1967)
- 1967NSb A Nevzorov,O Songina; Zh.Neorg.Khim.,12,1259 (2388) (1967)
- 1967TTa N Turakhonova,S Talipov; Uzbeksk.Khim.Zh.,3,23 (1967)
- 1967VSd G Volkova,V Sochevanov; Zh.Neorg.Khim.,12,222 (433) (1967)
- 1966BNA A Babko,B Nabivanets,I Lukianets; Zh.Neorg.Khim.,11,1257 (1966)
- 1966GAc Y Goroshchenko,M Andreeva; Zh.Neorg.Khim.,11,2233 (1966)
- 1966SAb K Stolyarov,I Amantova; Vestnik Leningr.Univ.,4,141;155;10,133 (1966)
- 1965BSe B Budesinsky,B Savvin; Z.Anal.Chem.,214,189 (1965)
- 1965GBa R Gut,H Buser,E Schmid; Helv.Chim.Acta,48,878 (1965)
- 1965MHb R McCarley,B Hughes,F Cotton,R Zimmerman; Inorg.Chem.,4,1491 (1965)
- 1964GSc A Golub,A Sych; Zh.Neorg.Khim.,9,1085 (1964)

1964GUa R Gut; Helv.Chim.Acta,47,2262 (1964)
1964SHa L Sommer,J Havel; Collec.Czech.Chem.Comm.,29,690 (1964)
1962BVa A Babko,A Volkova; Zh.Neorg.Khim.,7,1216 (2345) (1962)
1957AHb N Adler,C Hiskey; J.Am.Chem.Soc.,79,1827,1831,1834 (1957)
1956SSc H Schafer,F Schulte; Zh.Anal.Khim.,149,73 (1956)
1954CVa D Cozzi,S Vivarelli; Z.Elektrochem.,58,177 (1954)
1952LAB W Latimer; "Oxidation Potentials",Prentice Hall,NY (1952)
1938GGa G Grube,H Grube; Z.Elektrochem.,44,771 (1938)
1928KHa S Kiehl,D Hart; J.Am.Chem.Soc.,50,2337 (1928)

EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
I Data with various BACKGROUNDS
H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

END Experiments recorded for
from SC-Database on Saturday, 01 January, 2000 at 00:10:19