



Sofia Gama 21 November 2021

Cerium(III)

Reaction	Baes and Mesmer, 1976	NIST46	Brown and Ekberg, 2016
$Ce^{3+} + H_2O \rightleftharpoons Ce(OH)^{2+} + H^+$	-8.3	-8.3	-8.31 ± 0.03
$2 \text{ Ce}^{3+} + 2 \text{ H}_2\text{O} \rightleftharpoons \text{Ce}_2(\text{OH})_2^{4+} + 2 \text{ H}^+$	<-15.5		-16.0 ± 0.2
$3 \text{ Ce}^{3+} + 5 \text{ H}_2\text{O} \rightleftharpoons \text{Ce}_3(\text{OH})_5^{4+} + 5 \text{ H}^+$	-33.5		-34.6 ± 0.3
$Ce(OH)_3(s) + 3 H^+ \rightleftharpoons Ce^{3+} + 3 H_2O$	19.9		18.5 ± 0.5
$Ce(OH)_3(s) \rightleftharpoons Ce^{3+} + 3 OH^-$		-22.1 ± 0.9	

C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976.

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 135–145.

NIST46, NIST Critically Selected Stability Constants of Metal Complexes: Version 8.0. Available at: www.nist.gov/srd/nist46