

START Experiments recorded for
 from SC-Database on Saturday, 01 January, 2000 at 01:03:30
 Software version = 5.81 Data version = 4.62
 Experiment list contains 158 experiments for
 (no ligands specified)
 4 metals : Rh+, Rh++, Rh+++, Rh++++
 (no references specified)
 (no experimental details specified)

I- HL Iodide CAS 10034-85-2 (20)
 Iodide;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | | | |
|-----|----|--------|------|------|---|---|--|----------------|---|
| Rh+ | sp | non-aq | 22°C | 100% | U | M | | 1975F0a (8353) | 1 |
|-----|----|--------|------|------|---|---|--|----------------|---|

K'(Rh(P(Ph)3)2COI+I) < -4.5

Medium: dichloromethane. K': Rh(P(Ph)3)2COI+I=Rh(P(Ph)3)COI2+P(Ph)3

| | | | | | | | | | |
|-----|----|--------|------|------|---|---|--|----------------|---|
| Rh+ | sp | non-aq | 22°C | 100% | U | M | | 1975F0a (8354) | 2 |
|-----|----|--------|------|------|---|---|--|----------------|---|

K(RhA2COI+I)=-1.30

Medium: dichloromethane. K: RhA2COI+I=RhACOI2+A, A=Triphenyl arsine

| | | | | | | | | | |
|-----|----|--------|------|------|---|---|--|----------------|---|
| Rh+ | sp | non-aq | 22°C | 100% | U | M | | 1975F0a (8355) | 3 |
|-----|----|--------|------|------|---|---|--|----------------|---|

K(RhA2COI+I)=-2.7

Medium: dichloromethane. K: RhA2COI+I=RhACOI2+A, A=Triphenyl stibine (Ph3Sb)

| | | | | | | | | | |
|-----|----|--------|---|------|---|---|--|----------------|---|
| Rh+ | sp | non-aq | ? | 100% | U | I | | 1972F0a (8356) | 4 |
|-----|----|--------|---|------|---|---|--|----------------|---|

K=2.4

Medium: DMF. K: Rh(Ph3P)2COC1+L=Rh(Ph3P)2COL+Cl. K(Rh((CO)2Cl2+2L=Rh(CO)2L2+2Cl)=1.9 (in 1,2-dichloromethane); 2.3 (MeCN); 1.5 (90% MeCN/H2O)

OH- HL Hydroxide (57)
 Hydroxide;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | | | |
|-----|-----|------|------|-----|---|--|--|-----------------|---|
| Rh+ | oth | none | 25°C | 0.0 | U | | | 1958MPa (12044) | 5 |
|-----|-----|------|------|-----|---|--|--|-----------------|---|

*Kso=3.31?

*Kso: K(1/2Rh2O(s)+H=1/2H2O+Rh); method:combination of thermodynamic data

C4H6O2 L Me methacrylate CAS 96-33-3 (815)

Methyl propenoate; CH2:CH.CO2.CH3

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | | | |
|-----|-----|-------|------|------|---|---|--|-----------------|---|
| Rh+ | nmr | alc/w | 20°C | 100% | U | M | | 1977HRa (29732) | 6 |
|-----|-----|-------|------|------|---|---|--|-----------------|---|

K(RhA+L)=0.5

Medium: MeOH. A=Ph2P.CH2.CH2.PPh2

| | | | |
|---------------------------|---|---------|--------------------|
| C6H6 | L | Benzene | CAS 71-43-2 (2143) |
| Benzene, cyclohexatriene; | | | |

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | |
|-----|-----------|------|------|---|---|-----------------|---|
| Rh+ | nmr alc/w | 20°C | 100% | U | M | 1977HRa (43170) | 7 |
|-----|-----------|------|------|---|---|-----------------|---|

$$K(RhA+L)=1.3$$

Medium: MeOH. A=Ph₂P.CH₂.CH₂.PPh₂

C6H14 L CAS 110-54-3 (2146)

n-Hexane; $\text{CH}_3.\text{CH}_2.\text{CH}_2.\text{CH}_2.\text{CH}_2.\text{CH}_3$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | |
|-----|-----------|------|------|---|---|-----------------|---|
| Rh+ | nmr alc/w | 20°C | 100% | U | M | 1977HRa (50626) | 8 |
|-----|-----------|------|------|---|---|-----------------|---|

$$K(RhA+L)=0.3$$

Medium: MeOH. A=Ph₂P.CH₂.CH₂.PPh₂

C7H8 L CAS 108-88-3 (2144)

Toluene; $C_6H_5.CH_3$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | |
|-----|-----------|------|------|---|---|-----------------|---|
| Rh+ | nmr alc/w | 20°C | 100% | U | M | 1977HRa (55786) | 9 |
|-----|-----------|------|------|---|---|-----------------|---|

$$K(RhA+L)=2.0$$

Medium: MeOH. A=Ph₂P.CH₂.CH₂.PPh₂

| | | | |
|------|---|--------------|--------------------|
| C8H8 | L | Vinylbenzene | CAS 100-42-5 (811) |
|------|---|--------------|--------------------|

Styrene; $C_6H_5.CH:CH_2$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | |
|-----|-----------|------|------|---|---|-----------------|----|
| Rh+ | nmr alc/w | 20°C | 100% | U | M | 1977HRa (59255) | 10 |
|-----|-----------|------|------|---|---|-----------------|----|

$$K(RhA+L)=1.3$$

Medium: MeOH. A=Ph₂P.CH₂.CH₂.PPh₂

| | | | |
|-------|---|----------|---------------------|
| C8H10 | L | p-Xylene | CAS 106-42-3 (2145) |
|-------|---|----------|---------------------|

1,4-Dimethylbenzene, 4-Xylene; CH3.C6H4.CH3

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

| | | | | | | | |
|-----|-----------|------|------|---|---|-----------------|----|
| Rh+ | nmr alc/w | 20°C | 100% | U | M | 1977HRa (60683) | 11 |
|-----|-----------|------|------|---|---|-----------------|----|

$$K(RhA+L)=2.7$$

Medium: MeOH. A=Ph₂P.CH₂.CH₂.PPh₂

C18H15Sb L CAS 603-36-1 (2654)

Triphenylantimony; $(\text{C}_6\text{H}_5)_3\text{Sb}$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

Rh+ sp non-aq 25°C 100% C I 20020Ra (97159) 12
K(RhCl(CO)L2+L)=2.21

Medium: CH2Cl2. In benzene, K=2.56; diethyl ether, K=2.87;
acetone, K=3.02; ethyl acetate, K=3.10.

C21H21P L CAS 6163-58-2 (600)
Tri(2-methylphenyl)phosphine (or 4-methyl where indicated); (CH3.C6H4)3P

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+ sp non-aq 25°C 100% U 1974TMa (101194) 13

K(H2(g)+RhClL3)=1.26
K(H2(g)+(RhClL2)2)=1.04

Medium: Toluene. Ligand: tri(4-methylphenyl)phosphine

BF4- HL (2497)
Tetrafluoroborate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++ dis oth/un 25°C 1.0M U M 1974TAb (1203) 14

K(Rh(phen)3+L)=1.69
K(Rh(phen)3L+L)=0.95

Medium: Na2SO4

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++ sp oth/un 35°C 1.50M U 1966BPb (2288) 15

K(Ru(NH3)5H2O+L)=-0.7

C2H3N L Cyanomethane CAS 75-05-8 (1399)
Acetonitrile; CH3.CN

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++ sp non-aq 25°C 100% U HM 1979DTa (19194) 16

K(Rh2(butanoate)4+L)=3.2
K(Rh2(butanoate)3L+L)=1.4

Medium: benzene. DH(K1)=-21, DH(K2)=-35 kJ mol-1 by calorimetry

C4H6N2 L N-Me-Imidazole CAS 616-47-7 (354)
N-Methyl-1,3-diazole; C3H3N2.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++ sp non-aq 25°C 100% U HM 1979DTa (29606) 17

K(Rh2(butanoate)4+L)=9.0

Medium: benzene. DH(K1)=-52, DH(K2)=-44 kJ mol⁻¹ by calorimetry

| | | | | |
|------------------|---|----------|--------------|------|
| C5H5N | L | Pyridine | CAS 110-86-1 | (31) |
| Pyridine, Azine; | | | | |

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K | values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|---|--------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|---|--------|-----------|--------|

Rh++ sp non-aq 25°C 100% U HM 1979DTa (36673) 18
K(Rh2(butanoate)4+L)=8.2
K(Rh2(butanoate)3L+L)=4.4

Medium: benzene. DH(K1)=-47, DH(K2)=-47 kJ mol⁻¹ by calorimetry

| | | | | | |
|---------------------------------------------------|---|------------|--------|----------|-------|
| C5H11N | L | Piperidine | CAS | 110-89-4 | (105) |
| Perhydropyridine; cyclo(-CH2.CH2.CH2.NH.CH2.CH2-) | | | | | |
| | | | C5H11N | | |

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

Rh++ sp non-aq 25°C 100% U HM 1979DTa (40457) 19
K(Rh2(butanoate)4+L)=9.0
K(Rh2(butanoate)3L+L)=5.0

Medium: benzene. DH(K1)=-55, DH(K2)=-52 kJ mol⁻¹ by calorimetry

C6H11O3P L CAS 824-11-3 (7548)
4-Ethyl-2,6,7-trioxa-1-phosphabicyclo[2,2,2]octane; CH3CH2C(CH2O)3P

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K | values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|---|--------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|---|--------|-----------|--------|

Rh++ nmr non-aq -60°C 100% U 1998KTa (49006) 20
K(Rh₂(OAc)₄+L)=2.48
K(Rh₂(OAc)₄+2L)=4.98

Method: 31P nmr. Medium: CD2Cl2

C6H12 L CAS 592-41-6 (2771)
1-Hexene; CH₂:CH(CH₂)₃.CH₃

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

Rh++ sp non-aq 25°C 100% U M K(RhA+L)=1.62 1987DMb (49013) 21

A=trifluoroacetate. Also data for A=perfluorobutyrate; and for L=styrene, cyclohexene, 2,5-dimethyl-2,4-hexadiene, 2-methoxypropene, and more

C7H16N02P L CAS 38432-39-2 (7549)
N,N-Diethylamine-1,3,2-dioxaphosphorinan; CH2(CH2O)2PN(CH2CH3)2

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

Rh++ nmr non-aq -60°C 100% U M 1998KTa (58028) 22
K(Rh2(OAc)4+L)=3.38

K(Rh2(OAc)4+2L)=5.48
K(Rh2(OAc)4+L+P)=5.56

Method: 31P nmr. Medium: CD2Cl2.

P: 4-Ethyl-2,6,7-trioxa-1-phosphabicyclo[2,2,2]octane.

C9H15O6P H3L CAS 2848-01-3 (5882)
Tris(2-carboxyethyl)phosphine; P(CH2.CH2.COOH)3

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|-------|-----|-------|----|------------------------------|-----------------|--------|
| Rh++ | sp | NaClO4 | 25°C | 0.10M | U | HM | | | 1988AMa (67600) | 23 |
| | | | | | | | | Keff(Rh2AB2+L=Rh2ABL+B)=7.04 | | |
| | | | | | | | | Keff(Rh2ABL+L=Rh2AL2+B)=4.60 | | |

Medium: LiClO4. A=(O2CCH3)4, B=H2O

C12H21N2P L CAS 115305-74-3 (5884)
Bis-(3-aminopropyl)phenylphosphine; C6H5P(CH2.CH2.CH2.NH2)2

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|-------|-----|-------|----|---------------------------|-----------------|--------|
| Rh++ | sp | NaClO4 | 25°C | 0.10M | U | HM | | | 1988AMa (82710) | 24 |
| | | | | | | | | K(Rh2AB2+L=Rh2ABL+B)=6.72 | | |
| | | | | | | | | K(Rh2ABL+L=Rh2AL2+B)=5.06 | | |

Medium: LiClO4. A=(O2CCH3)4, B=H2O

C15H15O2P HL CAS 85209-41-2 (4067)
3-(Diphenylphosphino)propanoic acid; (C6H5)2P.CH2.CH2.COOH

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|-------|-----|-------|----|---------------------------|-----------------|--------|
| Rh++ | sp | NaClO4 | 25°C | 0.10M | U | HM | | | 1988AMa (91918) | 25 |
| | | | | | | | | K(Rh2AB2+L=Rh2ABL+B)=6.60 | | |
| | | | | | | | | K(Rh2ABL+L=Rh2AL2+B)=5.26 | | |

Medium: LiClO4. A=(O2CCH3)4, B=H2O

C16H20NP L CAS 115290-71-6 (5883)
Diphenyl-(2-N,N-dimethylaminoethyl)phosphine; (C6H5)2P.CH2.CH2.N(CH3)2

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|-------|-----|-------|----|---------------------------|-----------------|--------|
| Rh++ | sp | NaClO4 | 25°C | 0.10M | U | HM | | | 1988AMa (93949) | 26 |
| | | | | | | | | K(Rh2AB2+L=Rh2ABL+B)=6.40 | | |
| | | | | | | | | K(Rh2ABL+L=Rh2AL2+B)=4.46 | | |

Medium: LiClO4. A=(O2CCH3)4, B=H2O

C18H15O3PS HL CAS 54262-24-7 (327)
4-(Diphenylphosphino)benzenesulfonic acid; (C6H5)2P.C6H4.SO3H

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

Rh++ sp NaClO4 25°C 0.10M U HM 1988AMa (97114) 27
 Keff(Rh2AB2+L=Rh2ABL+B)=7.20
 Keff(Rh2ABL+L=Rh2AL2+B)=5.38

Medium: LiClO4. A=(O2CCH3)4, B=H2O

C18H15P L CAS 603-35-0 (621)

Triphenylphosphine; (C6H5)3P

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++ sp non-aq 25°C 100% U HM 1992LPa (97147) 28

In CHCl3. K(Rh2(H-1A)3A2L+L=Rh2(H-1A)3AL2+A)=3.6, K(Rh2(H-1A)2A2L2+2L=Rh2-(H-1A)2L4+2A=2.9, K(Rh2(H-1A)2AL3+L=Rh2(H-1A)2L4+A)=1.0. HA=CH3COOH

C32H48N2O2 H2L CAS 103595-81-6 (7708)

N,N'-Ethylenebis(3,5-di-tert-butylsalicylalimine);

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++ nmr non-aq 23°C 100% U H 2000BNa (105792) 29

Method: 1H nmr. Medium: C6D6. For K(2RhL=Rh2L2), DH=ca. -55.9 kJ mol-1, DS=ca. -113 J K-1 mol-1.

e- HL Electron (442)

Electron;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ EMF none 25°C 0.00 U 1971ARa (884) 30

K(Rh + 3e=Rh(s))=38.44(758mV)

Rh+++ oth none 25°C 0.0 U 1952LAb (885) 31

K=44.0(870 mV)

K: 0.5Rh2O3(s)+3H+3e=Rh(s)+1.5H2O. K(RhCl6+3e=Rh(s)+6Cl)=21.8(440 mV).
 From thermodynamic data

 Rh+++ gl NaClO4 18°C 1.0M U I 1938GAa (886) 32

K(Rh(VI)+3e=Rh)=77(1480 mV)

Medium: HClO4. In 0.15 M HNO3: K=76(1460 mV)

 Rh+++ EMF oth/un 18°C 0.10M U I 1937GGa (887) 33

K(Rh(IV)+e=Rh)=24.2(1400 mV)

Medium: H2SO4. In 0.5 M H2SO4: K=24.8(1430 mV)

Br- HL Bromide CAS 10035-10-6 (19)

Bromide;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sol oth/un 25°C 0.10M U I 1985PSc (2289) 34

$K_{out}(Rh(phen)_3+Br)=0.65$
 $K_{out}(Rh(phen)_3+2Br)=0.56$
 Also $K_{out}(1:1 \text{ complex})=0.55$ ($I=0.25 \text{ M}$), 0.50 ($I=0.5 \text{ M}$), 0.44 ($I=0.75 \text{ M}$)
 and $K_{out}(1:2 \text{ complex})=0.49$ ($I=0.25 \text{ M}$), 0.37 ($I=0.5 \text{ M}$), 0.27 ($I=0.75 \text{ M}$)

Rh+++ sp non-aq 25°C 100% U M 1976B0a (2290) 35
 $K(RhA_4+RhA_4Br_2=Rh_2A_8Br_2)=4.3$
 Medium: MeCN. A=cyclohexylisocyanide

Rh+++ kin NaClO4 35°C 1.50M U T 1970BPb (2291) 36
 $K(Rh(NH_3)_5H_2O+L)=-1.23$
 By spec. $K=-0.77$. At 70 C: $K(trans-Rh(en)_2(H_2O)_2+L)=-0.28$ (by kinetics)

Rh+++ kin NaClO4 35°C 4.0M U 1969RSa (2292) 37
 $K(RhCl_5+L)=0.10$

Rh+++ kin NaClO4 65°C 4.0M U T 1968M0b (2293) 38
 $K(Rh(NH_3)_5+L)=-1.1$
 $K=-1.0(25 \text{ C})$

Rh+++ vlt oth/un 25°C 0.0 U K1=14.3 B2=16.3 1961CPb (2294) 39
 B3=17.6
 B4=18.4
 B5=17.2
 Additional method: spectrophotometry.

Rh+++ oth oth/un 84°C 0.0 U 1939LAa (2295) 40
 $K(Rh(NH_3)_5+L)=3.20$
 Method:chemical analysis

 CN- HL Cyanide CAS 74-90-8 (230)
 Cyanide;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|-----------|
| Rh+++ | nmr | oth/un | 25°C | ? | U | | | | 1994RGa | (2759) 41 |
| | | | | | | | | B6=47 | | |

Method: correlation with nmr parameters.

 CO3-- H2L Carbonate CAS 465-79-6 (268)
 Carbonate;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|------------------------------|-----------|-----------|
| Rh+++ | kin | NaClO4 | 25°C | 2.0M | C | | | | 2000KYb | (3366) 42 |
| | | | | | | | | * $K(Rh(NH_3)_5HCO_3)=-6.32$ | | |

*K is for loss of proton from HCO_3^- .

 Cl- HL Chloride CAS 7647-01-0 (50)
 Chloride;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|----------------------------------------------------|-----|--------|------|-------|-----|-------|----|---------------------------------------------------------------------------------------------------------|----------------|--------|
| Rh+++ | sol | oth/un | 25°C | 0.10M | U | I | | 1985PSc Kout(Rh(phen)3+Cl)=0.43 Also Kout=0.32 (I=0.25 M), 0.17 (I=0.5 M), 0.03 (I=0.75 M). | (5616) | 43 |
| Rh+++ | kin | NaCl | 50°C | 3.0M | C | | | 1975PHa K(fac-RhCl3(H2O)3+L)=6.3 K(RhCl3(H2O)2L+L)=5.3 K(RhCl3(H2O)L2+L)=0.039 | (5617) | 44 |
| Rh+++ | vlt | NaClO4 | 25°C | 1.0M | U | | | K1=2.62 B3=5.94 B4=7.42 B5=8.79 | 1974MMd (5618) | 45 |
| Medium: HClO4 | | | | | | | | | | |
| Rh+++ | con | non-aq | 25°C | 100% | U | T | | 1971PWb K(cis-Ru(en)2L2+L)=2.41 K(trans-Ru(en)2L2+L)=1 | (5619) | 46 |
| Medium: DMSO | | | | | | | | | | |
| Rh+++ | kin | NaClO4 | 35°C | 1.50M | U | T | M | 1970BPb K(trans-Rh(NH3)5H2O+L)=-1.3 | (5620) | 47 |
| 65 C: K=-0.74 | | | | | | | | | | |
| Rh+++ | kin | NaClO4 | 45°C | 1.50M | U | | M | 1970BPb K(trans-Rh(en)2BrH2O+L)=-0.19 | (5621) | 48 |
| Rh+++ | kin | NaClO4 | 35°C | 4.0M | U | | | 1969RSa K6=-1.1 | (5622) | 49 |
| Rh+++ | sp | NaClO4 | 90°C | 4.0M | U | | | K1=2.49 B3=6.15 B4=7.6 B5=8.1 B6=7.8 | 1969SEa (5623) | 50 |
| Medium: HClO4 | | | | | | | | | | |
| Rh+++ | kin | none | 87°C | 0.0 | U | H | | 1968LBb 77-97 C, DH(Rh(NH3)5+L)=14.6 kJ mol-1, DS=112 J K-1 mol-1 | (5624) | 51 |
| Rh+++ | kin | NaClO4 | 65°C | 4.0M | U | T | | 1968MOb K(Rh(NH3)5+L)=-0.74 K1=-0.80(25 C). In 4 M LiClO4: K=-0.60(65 C); 5 M NaClO4: -0.66(65 C) | (5625) | 52 |
| Rh+++ | kin | NaClO4 | 45°C | 4.0M | U | T | | 1967RSa K5=0.77 | (5626) | 53 |
| Medium: HClO4. K5=0.90(30 C),0.83(35 C),0.80(40 C) | | | | | | | | | | |

Rh+++ ISE KNO3 55°C 0.10M U T M 1966BPc (5627) 54
K(Ag+RhCl6)=4.68

2nd Metal:Ag+. K=5.69(25 C),5.18(35 C)

Rh+++ EMF NaClO4 55°C 0.10M U M 1966BPg (5628) 55
K(RhCl3(H2O)2OH+H)=4.8
K(RhCl4(H2O)OH+H)=6.0
K(RhCl5OH+H)=7.3

Also solubility data with AgCl

Rh+++ sp NaClO4 85°C 0.10M U T H 1966BPh (5629) 56
K2K3=3.65

Also chemical analysis. Medium: HClO4. K2K3=4.25(25 C), 4.11(40 C),
4.06(55 C), 3.83(70 C), DH(K2K3)=-20.1 kJ mol⁻¹

Rh+++ kin NaClO4 85°C 2.50M U T K1=-0.15 1966SHb (5630) 57
K(RhOH+L)=-0.40
K1=-0.05(75 C),-0.10(80 C); K(RhOH+L)=-0.22(75 C),-0.30(80 C)

Rh+++ oth NaCl 40°C var U 1965BPe (5631) 58
K3=1.69
K4=0.47
K5=-0.51

Method:electrophoresis

Rh+++ EMF NaClO4 25°C 0.10M U 1965BPg (5632) 59
K4=1.39
K5=0.55
K6=-0.23

Medium:HClO4

Rh+++ kin NaClO4 25°C 0.10M U 1965BPg (5633) 60
K(Hg+RhCl6)=7.3

2nd Metal:Hg++. Medium:HClO4

Rh+++ kin NaClO4 35°C 4.0M U T 1965RHa (5634) 61
K6=-1.14
Medium: HClO4. K6=-0.72(15 C),-0.85(20 C),-0.93(25 C)

Rh+++ sp NaClO4 120°C 6.0M U K1=>3 K2=>3 1963WRa (5635) 62
K3=3
K4=2.4
K5=1.4
K6=-0.25

Rh+++ vlt NaClO4 25°C 1.0M U K1=2.45 B2=4.54 1958CPb (5636) 63
K3=1.38
K4=1.16
K5=1.67

medium: HC104. B6=8.43

$$K(\text{Rh}(\text{NH}_3)_5 + \text{L}) = 4.95$$

| | | | | |
|-------------------|----|-------------|---------------|-------|
| C104-Perchlorate; | HL | Perchlorate | CAS 7001-90-3 | (287) |
|-------------------|----|-------------|---------------|-------|

$$K(Rh(en)_3L)=0.93$$

| | | | | |
|-----------------|----|----------|---------------|-------|
| F- Fluoride; | HL | Fluoride | CAS 7644-39-3 | (201) |
|-----------------|----|----------|---------------|-------|

$$K(\text{trans-Rh}(\text{NH}_3)_5(\text{H}_2\text{O}+\text{F}))=0.41$$

Halides, comparative (for book data under ligand 80)

$$K_{out}(\text{Rh}(\text{NH}_3)_5\text{H}_2\text{O}+\text{Cl}) = -0.81$$

Kout=-1.02(35C), -1.27(Br,35C). Ai I=0.2: Kout=0.42(F,45 C), 1.06(F,75 C)
Plus data at other I values

$$K(RhAI+I)=3.7$$
$$\text{RhA} = \text{trans-Rh(en)}_2$$
$$K(\text{Rh}(\text{NH}_3)_5\text{I}) = 2.68$$

DH(Cl)=-5.4 kJ mol⁻¹, DS=-60 J K⁻¹ mol⁻¹; DH(Br)=-2.9, DS=-49; DH(I)=7.5, DS=-26.8

$$K(\text{RhACl}_2 + \text{I} = \text{RhAClI} + \text{Cl}) = 0.80?$$

RhA=trans-Rh(en)₂. K=0.78(85 °C), 0.85(90 °C). Also other halogen complexes

```

-----
Rh+++      sp  KCl      85°C 1.50M U    M                      1965BPd (7434) 71
                                     K(RhCl2+Br=RhABrCl+Cl)=0.29
                                     K(RhABrCl+Br=RhABr2+Cl)=-0.10
RhA=trans-Rh(en)2
*****
I-          HL      Iodide          CAS 10034-85-2 (20)
Iodide;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Rh+++      sol oth/un 25°C 0.10M U    I                      1985PSc (8357) 72
                                     Kout(Rh(phen)3+I)=0.79
                                     Kout(Rh(phen)3+2I)=1.05
Also Kout (1:1 complex)=0.71 (I=0.25 M), 0.67 (I=0.5 M), 0.60 (I=0.75 M)
and Kout (1:2 complex)=0.88 (I=0.25 M), 0.68 (I=0.5 M), 0.56 (I=0.75 M)
-----
Rh+++      sp  non-aq 25°C 100% U    I M                      1976B0a (8358) 73
K:RhA4+RhA4I2=Rh2A8I2, A=cyclohexylisocyanide. In MeCN: K=3.7; in DMSO: 3.8;
In nitromethane: 3.4; in acetone: 2.5
-----
Rh+++      kin NaClO4 35°C 4.0M U                      1969RSa (8359) 74
                                     K(Rh(Cl)5+L)=0.48
*****
NH3          L      Ammonia          CAS 7664-41-7 (414)
Ammonia
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Rh+++      sol R4N.X 25°C 1.00M U                      1995MPa (9205) 75
                                     Kout(Rh(NH3)6+L)=0.93
Medium: NH4ClO4
-----
Rh+++      sp  NaClO4 25°C 1.00M C T H                      1992SPa (9206) 76
K((Rh2L8(OH)2(H2O)=2(cis-RhL4(OH)(H2O)))=-1.84
Data also for other equilibria between mononuclear and binuclear species.
-----
Rh+++      gl  NaClO4 25°C 1.00M C    H                      1986SKa (9207) 77
                                     *K1(cis-RhL4)=-6.39
                                     *K2(cis-RhL4)=-8.36
                                     *K1(trans-RhL4)=-4.86
                                     *K2(trans-RhL4)=-8.29
cis-RhL4: DH(*K1)=44.9 kJ mol-1; DH(*K2)=42.6;
trans-RhL4: DH(*K1)=34.0 kJ mol-1; DH(*K2)=36.7
-----
Rh+++      sp  NaClO4 125°C 0.10M U T                      1981BJa (9208) 78
                                     K(RhL5OH+HL=RhL6+H2O)=1.92
For temperatures 111.5, 135, 142 C, K=2.36, 1.80, 1.63 respectively
-----
Rh+++      kin NaClO4 125°C 0.10M U T                      1981BJa (9209) 79

```

K(RhL5OH+HL=RhL6+H2O)=2.04
 For temperatures 111.5, 135, 142 C, K=2.39, 1.67, 1.67 respectively

Rh+++ gl oth/un 25°C var U 1968TOb (9210) 80

K(RhHL4+L)=3.7
 K(RhEtL4+L)=9.4
 K(Rh(C2F4H)L4+L)=9.7

 NO2- HL Nitrite CAS 7782-77-6 (635)
 Nitrite;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg K values | Reference | ExptNo |
|-------|-----|--------|------|------|-------|-------|---------------------------------------------------------|----------------|--------|
| Rh+++ | sol | oth/un | 26°C | 3.0M | U T H | | K1=6.16 B3=11.53 B4=12.20 B5=12.85 B6=13.99 | 1983ZYa (9404) | 81 |

pH=9-10, NaNO2 aqueous solution

Rh+++ kin NaClO4 35°C 4.0M U 1969RSa (9405) 82

K(RhCl5+L)=0.04

 N3- HL Azide CAS 7782-79-8 (441)
 Azide;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|--------------------|-----------------|--------|
| Rh+++ | kin | oth/un | 77°C | var | U | | K(Rh(NH3)5L+H)=2.2 | 1970DLa (10255) | 83 |

Medium: HClO4

Rh+++ kin oth/un 60°C var U M 1968STb (10256) 84

K(Rh(NH3)5L+H)=1.95

 OH- HL Hydroxide (57)
 Hydroxide;

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg K values | Reference | ExptNo |
|-------|-----|--------|------|------|-------|-------|-----------------------------------------------------------------|-----------------|--------|
| Rh+++ | gl | NaClO4 | 25°C | 1.0M | C T H | | *K(H2OA2Rh(OH)RhA2H2O)=-2.372 *K((HO)A2Rh(OH)RhA2H2O)=-9.128 | 1982HNb (12045) | 85 |

A is 1,2-diaminoethane. K(A2Rh(OH)2RhA2+H2O=H2OA2Rh(OH)RhA2(OH))=1.05
 DH(*K((H2O)A2Rh(OH)RhA2(H2O)))=28 kJ mol-1, DS=49 J K-1 mol-1.

 Rh+++ gl NaClO4 25°C 1.00M C M 1980SFb (12046) 86

*K1(RhA5(H2O))=-6.93
 *K1(cis-RhA4)=-6.40
 *K1(trans-RhA4)=-4.92

A=NH3. *K2=-8.32

Rh+++ sp NaClO4 20°C 1.0M U 1960C0c (12058) 98
*K1=-2.92

```

-----
Rh+++      sp   oth/un 25°C   dil  U                      1959FAa (12059) 99
                                     *K1=-3.43?
                                     K(Rh(OH)3(s)=RhOH+2OH)=-22.32
-----
Rh+++      gl   oth/un ?25    dil  U                      1959GVa (12060) 100
                                     *K1(Rh(en)3) < -12
-----
Rh+++      oth none 25°C   0.0  U                      1958MPa (12061) 101
                                     *Kso=2.56?
*Kso: K(0.5 Rh2O3(s)+3H=1.5 H2O+Rh); method:combination of thermodynamic
data
-----
Rh+++      sol oth/un 20°C   dil  U                      1956JOa (12062) 102
                                     K(Rh(OH)3(s)+H=Rh(OH)2)=-5
-----
Rh+++      sp   oth/un 20°C   dil  U                      1930GFa (12063) 103
                                     *K1(Rh(NH3)6)=-11.14
-----
Rh+++      kin none 15°C   0.0  U                      1928BVa (12064) 104
                                     *K1(Rh(NH3)5(H2O))=-5.86
*****
P3010-----      H5L                      CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Rh+++      sp   oth/un 25°C 1.00M U                      1978FPa (13899) 105
                                     K(Rh+2HP3010)=16.37
                                     K(Rh+2P3010)=26.25
K(Rh(H2O)2Cl4+2H2L=RhH2L2+2H+4Cl)=-2.43
*****
SCN-      HL      Thiocyanate      CAS 463-56-9 (106)
Thiocyanate;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Rh+++      nmr oth/un 25°C   ?  U                      1994RGa (15243) 106
                                     B6=35
Method: correlation with nmr parameters.
-----
Rh+++      ISE oth/un 25°C 0.10M U                      1975LMa (15244) 107
                                     K(Rh(NH3)5NCS+Ag)=3.38
-----
Rh+++      kin NaClO4 35°C 4.0M U                      1969RSa (15245) 108
                                     K(RhCl5+L)=0.96
*****
S04--      H2L      Sulfate      CAS 7664-93-9 (15)
Sulfate;
-----

```

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-----------------------------------|-----|--------|------|-------------|-----|-------|----|---------------------------------------------------|-----------|--------|
| Rh+++ | con | oth/un | 25°C | 0.01M | U | M | | 1977SPa (16520) | 109 | |
| | | | | | | | | K(Rh(NH3)5Cl+S04)=2.90 K(Rh(NH3)5NO2+S04)=2.52 | | |
| Rh+++ | kin | NaClO4 | 65°C | 1.0M | U | I M | | 1973M0a (16521) | 110 | |
| | | | | | | | | Kout(Rh(NH3)5+L)=0.5 Kin(Rh(NH3)5+L)=1.0 | | |
| When I=4 M: Kout=-0.20, Kin=1.4 | | | | | | | | | | |
| Rh+++ | sol | NaClO4 | 25°C | 3.0M | U | HM | | 1972MRe (16522) | 111 | |
| | | | | | | | | K(Rh(en)3+L)=0.15 K(Rh(en)3L+L)=0.11 | | |
| Rh+++ | sp | NaClO4 | 65°C | 4.0M | U | T | | 1968M0b (16523) | 112 | |
| | | | | | | | | K(Rh(NH3)5+L)=-0.15 | | |
| K=0.0(25 C) | | | | | | | | | | |
| ***** | | | | | | | | | | |
| S2O3-- | | H2L | | Thiosulfate | | | | CAS 73686-28-7 | (177) | |
| Thiosulfate; | | | | | | | | | | |
| ***** | | | | | | | | | | |
| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
| Rh+++ | con | oth/un | 25°C | 0.01M | U | M | | 1977SPa (16898) | 113 | |
| | | | | | | | | K(Rh(NH3)5Cl+S2O3)=2.50 | | |
| ***** | | | | | | | | | | |
| CH4N2S | | L | | Thiourea | | | | CAS 62-56-6 | (51) | |
| Thiocarbamide, Thiourea; (H2N)2CS | | | | | | | | | | |
| ***** | | | | | | | | | | |
| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
| Rh+++ | gl | NaClO4 | 25°C | 1.00M | U | | | 1997SAa (17853) | 114 | |
| | | | | | | | | *K(Rh3(OH)4(H2O)11)=-3.10 | | |
| ***** | | | | | | | | | | |
| C2H3N3S | | L | | | | | | CAS 3179-31-5 | (4221) | |
| 1,2,4-Triazoline-3-thione; | | | | | | | | | | |
| ***** | | | | | | | | | | |
| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
| Rh+++ | sp | KCl | ? | 1.20M | U | | | 1972RPb (19246) | 115 | |
| | | | | | | | | B3=28.5 | | |
| Medium: HCl | | | | | | | | | | |
| ***** | | | | | | | | | | |
| C2H6OS | | L | | DMSO | | | | CAS 67-68-5 | (329) | |
| Dimethylsulfoxide; (CH3)2.SO | | | | | | | | | | |
| ***** | | | | | | | | | | |
| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
| Rh+++ | cal | non-aq | 24°C | 100% | U | HM | | 1976Lda (22121) | 116 | |

Medium: benzene. $K((\text{RhACl})_2 + 2\text{L} = 2\text{RhALCl}) = 1.9$, A=1,5-Cyclooctadiene.

DH=-16 kJ mol⁻¹

C2H8N2 L Ethylenediamine CAS 107-15-7 (23)

1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp NaClO4 25°C 1.00M C 1983HSc (23227) 117

*K(trans-RhL2)=-4.47

*K(trans-Rh(OH)L2)=-7.91

C3H4N2 L Imidazole CAS 288-32-4 (90)

1,3-Diazole, imidazole; C3H4N2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal oth/un 25°C 0.10M U HM 1977DSa (23920) 118

K(Rh2(O2CCH2OCH3)4+L)=3.94

K(Rh2(O2CCH2OCH3)4L+L)=2.40

Medium: phosphate buffer, pH 7.4

Rh+++ cal oth/un 25°C 0.10M U 1976DSa (23921) 119

K(Rh2(O2CCH2OCH3)4+L)=3.94

K(Rh2(O2CCH2OCH3)4L+L)=2.40

C3H6O2S H2L Thiolactic acid CAS 79-42-5 (366)

2-Mercaptopropionic acid; CH3.CH(SH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ gl NaClO4 25°C 0.0 C TIH K1=9.12 B2=16.42 1985SNc (25167) 120

K3=5.35

Data for I=0.10-1.0 M NaClO4, extrapolated to I=0.0. Data for 35 and 45 C

DH(K1)=-48.3 kJ mol⁻¹, DS=13; DH(K2)=-31.6, DS=33; DH(K3)=-21.9, DS=29

C3H7NO2 HL Alanine CAS 56-41-7 (86)

2-Aminopropanoic acid; H2N.CH(CH3).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ gl NaClO4 25°C 0.0 C TIH K1=15.95 B2=23.95 1985SNc (26255) 121

K3=4.35

Data for I=0.10-1.0 M NaClO4, extrapolated to I=0.0. Data for 35 and 45 C

DH(K1)=-167 kJ mol⁻¹, DS=-255; DH(K2)=-52.7, DS=-25; DH(K3)=-17.5, DS=25

C3H7NO2S H2L Cysteine CAS 52-90-4 (96)

2-Amino-3-mercaptopropionic acid; H2N.CH(CH2.SH)COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ gl NaClO4 25°C 0.0 C TIH K1=8.60 B2=11.95 1985Snc (26832) 122
K3=2.25

Data for I=0.10-1.0 M NaClO4, extrapolated to I=0.0. Data for 35 and 45 C
DH(K1)=-61.5 kJ mol⁻¹, DS=-42; DH(K2)=-18.0, DS=-25; DH(K3)=-17.6, DS=-17

C3H7NS L CAS 758-16-7 (476)
N,N-Dimethylthioformamide; HCS.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal non-aq 24°C 100% U HM 1976Lda (27258) 123
Medium: benzene. K((RhACl)2+2L=2RhALCl)=6.0, A=1,5-cyclooctadiene.
DH=-37 kJ mol⁻¹

C3H10N2 L Propanediamine CAS 109-76-2 (123)
1,3-Diaminopropane; H2N.CH2.CH2.CH2.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp NaClO4 25°C 1.0M C 19840Sa (28321) 124
*K1(cis-RhL2)=-6.15
*K2(cis-RhL2(OH))=-8.20
*K1(trans-RhL2)=-4.39
*K2(trans-RhL2(OH))=-8.20

C4H6O4S H3L Thiomalic acid CAS 70-49-5 (109)
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp oth/un 30°C ? U K1=21.6 1966SNb (30360) 125
By glass electrode: K2=8.4

C4H8S L CAS 110-01-0 (150)
Tetrahydrothiophene; cyclo(-CH2.CH2.S.CH2.CH2-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal non-aq 24°C 100% U HM 1976Lda (33741) 126
Medium: benzene. K((RhACl)2+2L=2RhALCl)=1.4, A=1,5-cyclooctadiene.
DH=-22 kJ mol⁻¹

C4H9NO L Morpholine CAS 110-91-8 (318)
Perhydro-1,4-oxazine, Tetrahydro-1,4-oxazine; C4H8NO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal non-aq 24°C 100% U HM 1976Lda (33794) 127
Medium: benzene. K((RhACl)2+2L=2RhALCl)=3.9, A=1,5-cyclooctadiene.

DH=-41 kJ mol⁻¹

C5H5N L Pyridine CAS 110-86-1 (31)

Pyridine, Azine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal oth/un 25°C 0.10M U HM 1977DSa (36674) 128

K(Rh₂(O₂CCH₂OCH₃)₄+L)=4.52

K(Rh₂(O₂CCH₂OCH₃)₄L+L)=2.81

Medium: phosphate buffer, pH 7.4

Rh+++ sp alc/w 25°C 100% U T M 1977PVa (36675) 129

K(RhA₂Cl₂+L)=1.98

A=o-dimethylaminophenyldimethylarsine. Also with many substituted pyridines

Medium: MeOH

Rh+++ cal non-aq 24°C 100% U IHM 1976LDa (36676) 130

Medium: benzene. K((RhACl)₂+2L=2RhALCl)=3.04, A=1,5-Cyclooctadiene.

DH=-28 kJ mol⁻¹. In triethylphosphate, K=3.61, DH=-30 kJ mol⁻¹

C5H6 HL Cyclopentadiene CAS 542-92-7 (4288)

Cyclopentadiene; cyclo(-CH:CH.CH₂.CH:CH-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp NaClO₄ 25°C 0.20M U M 1999CEa (37081) 131

*K(RhL(H₂O)₃)=-6.47

K(2RhL(OH)=(RhL)₂(u-OH)₃)=-8.9

K(RhL+Cl)=2.1

K(RhL+Br)=2.8

K(RhL+CN-py)=3.2, K(RhL+py-nia)=3.6, K(RhL+py)=4.6, K(RhL+dms)=3.4,

K(RhL+tu)>6, K(RhL+SCN)=5.1. dms: dimethylsulfide; py-nia: nicotinamide.

C5H₁₀N₄O₃ L CAS 54376-69-1 (8335)

N,N'-Carbonylbis(2-aminoacetamide);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ gl NaClO₄ 25°C 0.10M U TIH K₁=9.75 B₂=14.90 1980SAC (40139) 132

Data for 0.075-0.15 M. At I=0, K₁=10.20, K₂=5.75. Also data for 30 C.

DH and DS values.

C5H₁₁N L Piperidine CAS 110-89-4 (105)

Perhydropyridine; cyclo(-CH₂.CH₂.CH₂.NH.CH₂.CH₂-) C5H₁₁N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal non-aq 24°C 100% U HM 1976LDa (40458) 133

Medium: benzene. K((RhACl)₂+2L=2RhALCl)=5.4, A=1,5-Cyclooctadiene.

Medium: benzene. $K((\text{RhACl})_2 + 2\text{L} = 2\text{RhALCl}) = 3.67$, $A = 1,5\text{-Cyclooctadiene}$.

2-Amino-3-(4'-imidazolyl)propanoic acid; $\text{H}_2\text{N}.\text{CH}(\text{CH}_2.\text{C}_3\text{H}_3\text{N}_2)\text{COOH}$

$$K(\text{Rh}_2(\text{O}_2\text{CCH}_2\text{OCH}_3)_4 + \text{L}) = 4.38$$

Medium: phosphate buffer, pH 7.4

S-Ethyl-2-mercaptobutane; $\text{CH}_3.\text{CH}(\text{SCH}_2.\text{CH}_3)\text{CH}_2.\text{CH}_3$

B3=6.25

2-Mercaptobenzoic acid; HS.C6H4.COOH

Benzamide; $\text{C}_6\text{H}_5.\text{CO}.\text{NH}_2$
$$K(\text{Rh}(\text{NH}_3)_5 + \text{H}^+ - 1\text{L}) = 2.2$$

1-Azabicyclo[2.2.2]octane;

Rh+++ cal non-aq 24°C 100% U HM 1976Lda (57423) 139
 Medium: benzene. $K((\text{RhACl})_2+2\text{L}=2\text{RhALCl})=1.3$, A=1,5-Cyclooctadiene.
 DH=-29 kJ mol⁻¹

C7H16S L (6899)
 S-Ethyl-2-methyl-2-mercaptobutane; $\text{H}_3\text{C}.\text{CH}_2.\text{S}.\text{C}(\text{CH}_3)_2.\text{CH}_2.\text{CH}_3$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|-------------------------|-----------------|--------|
| Rh+++ | EMF | non-aq | 25°C | 100% | U | I | | K1=2.32 B2=4.12 B3=5.81 | 1990MRc (58097) | 140 |

Medium: Dimethylformamide, 0.1 M NaClO₄; Rh/Pt-electrode
 In acetone: K1=3.71, B2=6.99, B3=9.27

C8H13NO6S H3L (5675)
 2-Mercapto-1-aminoethane-N,N,S-triethanoic acid; $\text{HOOC}.\text{CH}_2.\text{S}.\text{CH}_2.\text{CH}_2.\text{N}(\text{CH}_2\text{COOH})_2$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|-------|-----|-------|----|-----------------------------|-----------------|--------|
| Rh+++ | gl | oth/un | 25°C | 0.10M | U | | | K(RhL+H)=3.65 K(RhHL+H)=3.2 | 1983ESa (61830) | 141 |

C8H18S L CAS 544-40-1 (2346)
 Bis(n-butyl)sulfide; $\text{C}_4\text{H}_9.\text{S}.\text{C}_4\text{H}_9$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|--------------------------|-----------------|--------|
| Rh+++ | EMF | non-aq | 25°C | 100% | U | I | | K1=4.05 B2=7.75 B3=10.80 | 1990MRc (63008) | 142 |

Medium: Acetone, 0.1 M NaClO₄; Rh/Pt-electrode. In DMF K1=2.61, B2=4.70, B3=6.56

C9H7N3O2S H2L TAR CAS 2246-46-0 (707)
 4-(2'-Thiazolylazo)-resorcinol; $\text{C}_3\text{H}_2\text{NS}.\text{N}:\text{N}.\text{C}_6\text{H}_3(\text{OH})_2$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|---------------|-----------------|--------|
| Rh+++ | sp | oth/un | ? | ? | U | | | K(Rh+HL)=8.45 | 1968BIb (64723) | 143 |

| | | | | | | | | | | |
|-------|----|-------|------|-----|---|--|--|---------|-----------------|-----|
| Rh+++ | sp | alc/w | 25°C | 50% | U | | | K(?)=12 | 1967NPb (64724) | 144 |
|-------|----|-------|------|-----|---|--|--|---------|-----------------|-----|

Medium: 50% MeOH, 0.1 M NaClO₄

C9H18O2S L (6900)
 S-Butyl-O-(2-butyl)thiocarbonate; $\text{CH}_3.\text{CH}_2.\text{CH}(\text{CH}_3).\text{O}.\text{CO}.\text{S}.\text{CH}_2.\text{CH}_2.\text{CH}_2.\text{CH}_3$

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|
|-------|-----|--------|------|------|-----|-------|----|----------|-----------|--------|

Rh+++ EMF non-aq 25°C 100% U K1=1.96 B2=3.45 1990MRc (67961) 145
B3=4.81

Medium: Dimethylformamide, 0.1 M NaClO₄; Rh/Pt-electrode

C10H14N5O7P H2L AMP-5 CAS 18422-05-4 (842)

Adenosine-5'-monophosphoric acid, 5-Adenylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ cal oth/un 25°C 0.10M U HM 1977DSa (72484) 146

K(Rh₂(O₂CCH₂OCH₃)₄+L)=3.45

K(Rh₂(O₂CCH₂OCH₃)₄L+L)=2.70

Medium: phosphate buffer, pH 7.4

Rh+++ sp oth/un 22°C 0.10M U M 1975RHa (72485) 147

K(RhA+L)=3.18

K(RhB+L)=3.28

K(RhY+L)=3.63

K(RhAL+L)=2.20

At pH 7. K(RhBL+L)=2.31; K(RhCL+L)=2.54. HA=CH₃.CO.COOH; HB=CH₃.COOH;

HC=C₂H₅.COOH

C10H15N5O10P2 H3L ADP CAS 20398-34-9 (2181)

Adenosine-5'-diphosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp oth/un 22°C 0.10M U K1=3.06 B2=5.10 1975RHa (73012) 148

Cation is RhCH₃COO⁺ and ionic medium is a K-phosphate buffer

C10H16N5O13P3 H4L ATP CAS 56-65-5 (403)

Adenosine-5'-triphosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp oth/un 22°C 0.10M U 1975RHa (74815) 149

K(Rh(CH₃COO)+L)=3.27

K(Rh(CH₃COO)L+L)=2.11

K(Rh(C₂H₅COO)+L)=3.65

K(Rh(C₂H₅COO)L+L)=2.52

Medium: 0.1 M phosphate buffer, pH 7.5

C12H24O2S L (6901)

S-Pentyl-O-(hexyl)thiocarbonate; C₆H₁₃.O.CO.S.C₅H₁₁

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ EMF non-aq 25°C 100% U K1=1.89 B2=3.30 1990MRc (83114) 150

B3=4.51

Medium: Dimethylformamide, 0.1 M NaClO₄; Rh/Pt-electrode

C12H26S L CAS 6294-31-3 (5697)
S,S-Dihexylsulfide; C6H13.S.C6H13

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ EMF non-aq 25°C 100% U K1=3.98 B2=7.54 1990MRc (84034) 151
B3=10.48

Medium: Acetone, 0.1 M NaClO4; Rh/Pt-electrode

C13H9N3OS HL TAN CAS 1147-56-4 (4030)
1-(1',3'-Thiazol-2'-ylazo)-2-hydroxynaphthalene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh+++ sp mixed ? 40% U K1=9.84 1971IBa (84617) 152
Medium: 40% dimethylformamide

e- HL Electron (442)
Electron;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++++ EMF oth/un 25°C 0.25M U I 1948DSa (888) 153
K(Rh(IV)+e=Rh)=23.3(1380 mV)

Medium: H2SO4. In 3 M H2SO4: K=24.4(1440 mV), 0.5 M: 24.3(1435 mV)

Rh++++ EMF oth/un 22°C 0.10M U 1937GGa (889) 154
K(Rh(VI)+2e)=50(1460 mV)

Medium: H2SO4

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++++ sp oth/un 25°C ? U 1974TMa (5638) 155
K4=3.48

OH- HL Hydroxide (57)
Hydroxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Rh++++ kin NaClO4 25°C 1.00M U 1977KLa (12065) 156
K(RhO(OH)+OH=RhO(OH)2)=4.48
K(RhO(OH)2+OH=RhO(OH)3)=3.54

Rh++++ kin oth/un 25°C .007M U 1977KYa (12066) 157
K(RhO2+OH)=7.3

C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)
2,2'-Bipyridine; (C5H4N)2

| Metal | Mtd | Medium | Temp | Conc | Cal | Flags | Lg | K values | Reference | ExptNo |
|--------|-----|--------|------|------|-----|-------|----|-------------------------------|-----------------|--------|
| Rh++++ | sp | none | 25°C | 0.0 | U | | | | 1982CCc (69637) | 158 |
| | | | | | | | | K(2Rh(I)L2=[RhL2]2)=4.0 | | |
| | | | | | | | | K(Rh(I)L2+H3O=RhL2(H)H2O)=7.3 | | |
| | | | | | | | | K([Rh(I)L2]2+H=[RhL2]2H)=9.3 | | |

REFERENCES

- 2002ORa S Otto,A Roodt; Inorg.Chim.Acta,331,199 (2002)
2000BNa A Bunn,Y Ni,B Wayland; Inorg.Chem.,39,5576 (2000)
2000KYb Y Kitamura,L Yano,K Fujimori; Bull.Chem.Soc.Jpn.,73,2025 (2000)
1999CEa S Cayemittes,H Elias,A Merbach; Inorg.Chem.,38,4309 (1999)
1998Kta A Kudryavtsev,A Teleshev,W Linert; Inorg.Chim.Acta,267,293 (1998)
1997SAa L Spiccia,J Aramini,S Crimp,A Drljaca; J.Chem.Soc.,Dalton Trans.,4603 (1997)
1995MPa V Mironov,G Pashkov et al; Zh.Neorg.Khim.,40,1670 (1995)
1994RGa M Read,J Glaser,I Persson,M Sandstrom; J.Chem.Soc.,Dalton Trans.,3243 (1994)
1992LPa P Lahuerta,J Paya,A Bianchi; Inorg.Chem.,31,5336 (1992)
1992SPa J Springborg; Acta Chem.Scand.,46,956,1047 (1992)
1990MRC V Maistrenko,I Rusakov et al; Zh.Obshch.Khim.,60,851 (1990)
1988AMa M Aquino,D Macartney; Inorg.Chem.,27,2868 (1988)
1987DMb M Doyle,S Mahapatra et al; Inorg.Chem.,26,3070 (1987)
1986SKa L Skibsted; Acta Chem.Scand.,A40,364 (1986)
1985PSc A Pyartman,M Sof'in; Zh.Neorg.Khim.,30,1230 (1985)
1985Snc C Sharma,S Narvi; Thermochim.Acta,90,1 (1985)
19840Sa B Oby,L Skibsted; Acta Chem.Scand.,A38,399 (1984)
1983ESa N Ezerskaya,L Schubochkin; Zh.Neorg.Khim.,28,1515(854) (1983)
1983HSc K Howland,L Skibsted; Acta Chem.Scand.,A37,647 (1983)
1983ZYa Zhang Jianmin,Yang Bingyu,H Z,M G; Acta Chimica Sinica,284 (1983)
1982CCc M Chou,C Creutz,D Mahajan et al; Inorg.Chem.,21,3989 (1982)
1982HNb M Hancock,B Nielsen,J Springborg; Inorg.Chim.Acta,41,25 (1982)
1981BJa S Balt,A Jelsma; J.Chem.Soc.,Dalton Trans.,1289 (1981)
1980SAC P Srivastava,S Adhya,B Banerjee; J.Indian Chem.Soc.,57,985 (1980)
1980SFb L Skibsted,P Ford; Acta Chem.Scand.,A34,109 (1980)
1979DTa R Drago,S Tanner,R Richman et al; J.Am.Chem.Soc.,101,2897 (1979)
1978FPa N Falendish,E Parkhomenko et al; Zh.Neorg.Khim.,23,2130(1170) (1978)
1977DSa K Das,E Simmons,J Bear; Inorg.Chem.,16,1268 (1977)
1977HRa J Halpern,D Riley,S Chan et al; J.Am.Chem.Soc.,99,8055 (1977)
1977KLa V Kalinina,V Lyakushina; Zh.Neorg.Khim.,22,3325(1814) (1977)
1977KYa V Kalinina,K Yatsimirskii et al; Zh.Neorg.Khim.,22,2488(1344) (1977)
1977PVa A Peloso,L Volponi; J.Chem.Soc.,Dalton Trans.,2356 (1977)
1977SPa M Sofin,A Pyartman et al; Zh.Fiz.Khim.,51,1281 (1977)
1976BOa A Balch,M Olmstead; J.Am.Chem.Soc.,98,2354 (1976)
1976DSa K Das,J Bear; Inorg.Chem.,15,2093 (1976)

1976LDA M Li,R Drago; J.Am.Chem.Soc.,98,5129 (1976)
 1976MNa N Masleyi,B Nabivanets et al; Ukr.Khim.Zh.,42,247 (1976)
 1975FOa D Forster; J.Am.Chem.Soc.,97,951 (1975)
 1975LMA G Lalor,H Miller; J.Inorg.Nucl.Chem.,37,1832 (1975)
 1975PAa M Pavelich; Inorg.Chem.,14,982 (1975)
 1975PHA D Palmer,G Harris; Inorg.Chem.,14,1316 (1975)
 1975RHa L Rainen,R Howard,A Kimball et al; Inorg.Chem.,14,2752 (1975)
 1975ZFa A Zanella,P Ford; Inorg.Chem.,14,700 (1975)
 1974MMd M Mihailov,V Mihailova,V Khalkin; J.Inorg.Nucl.Chem.,36,115 (1974)
 1974PKa A Pyartman,N Kolobov,L Merkuleva et al; Zh.Neorg.Khim.,19,1691(E:920)
 (1974)
 1974TAb T Takamatsu; Bull.Chem.Soc.Jpn.,47,118 (1974)
 1974TMA C Tolman,P Meakin,D Lindner,J Lesson; J.Am.Chem.Soc.,96,2762 (1974)
 1973MOa F Monacelli; Inorg.Chim.Acta,7,65 (1973)
 1973NNA C Nair,H Nigam; Curr.Sci.,42,495 (1973)
 1972FOa D Forster; Inorg.Chem.,11,1686 (1972)
 1972MRe V Mironov,G Ragulin,I Umova et al; Zh.Fiz.Khim.,46,257(E:155) (1972)
 1972RPb A Radushev,E Prokhorenko; Zh.Anal.Khim.,27,11,2209 (1972)
 1971ARA J Amosse,M Rubaud et al; Compt.Rend.,273C,1708 (1971)
 1971IBA V Ivanov,A Busev,V Gresl,A Zagruzina; Zh.Anal.Khim.,26,8,1553 (1971)
 1971IBb B Ivanov-Emin,L Borzova et al; Zh.Neorg.Khim.,16,2766(E:1474) (1971)
 1971PWb D Palmer,D Watts; Inorg.Chem.,10,281 (1971)
 1970BPb H Bott,A Poe,K Shaw; J.Chem.Soc.(A),1745 (1970)
 1970CHb A Cunningham,D House,H Powell; Australian J.Chem.,23,2375 (1970)
 1970DLA C Davis,G Lalor; J.Chem.Soc.(A),445 (1970)
 1969RSa D Robb,M Steyn,H Kruger; Inorg.Chim.Acta,3,383 (1969)
 1969SEa V Shlenskaya,O Efremenko et al; Izv.Akad.Nauk(USSR),18,1643 (1969)
 1968BIb A Busev,V Ivanov,V Gresl; Zh.Anal.Khim.,23,10,1570 (1968)
 1968BPb H Bott,A Poe,K Shaw; J.Chem.Soc.,Chem.Comm.,793 (1968)
 1968LBb G Lalor,G Bushnell; J.Chem.Soc.(A),2520 (1968)
 1968MOb F Monacelli; Inorg.Chim.Acta,2,263 (1968)
 1968STb P Staples; J.Chem.Soc.(A),2731 (1968)
 1968TOb K Thomas,J Osborn,A Powell,G Wilkinson; J.Chem.Soc.(A),1801 (1968)
 1967BPb H Bott,A Poe; J.Chem.Soc.(A),205 (1967)
 1967NPb G Nickless,F Pollard,T Samuelson; Anal.Chim.Acta,39,37 (1967)
 1967PSb A Poe,K Shaw,M Wendt; Inorg.Chim.Acta,1,371 (1967)
 1967RSa W Robb,M de,V Steyn; Inorg.Chem.,6,616 (1967)
 1966BPb H Bott,A Poe,K Shaw; J.Chem.Soc.,Chem.Comm.,793 (1966)
 1966BPc A Belyaev,B Ptitsyn; Izv.Sib.Otd.Akad.Nauk SSR,136 (1966)
 1966BPf E Bounsall,A Poe; J.Chem.Soc.(A),286 (1966)
 1966BPg A Belyaev,B Ptitsyn; Zh.Neorg.Khim.,11,1345 (1966)
 1966BPh A Belyaev,B Ptitsyn; Zh.Neorg.Khim.,11,1565 (1966)
 1966SHb K Swaminathan,G Harris; J.Am.Chem.Soc.,88,4411 (1966)
 1966SNb S Sinha,H Nigam,S Sangal; Chim.Anal.,48,515 (1966)
 1965BPd H Bott,A Poe; J.Chem.Soc.,5931 (1965)
 1965BPe E Blasius,W Preetz; Z.Anorg.Chem.,335,1 (1965)
 1965BPg A Belyaev,B Ptitsyn; Zh.Obshch.Khim.,35,1887 (1965)
 1965RHa W Robb,G Harris; J.Am.Chem.Soc.,87,4472 (1965)
 1964PHb W Plumb,G Harris; Inorg.Chem.,3,542 (1964)
 1963WRa W Wolsey,C Reynolds,J Kleinberg; Inorg.Chem.,2,463 (1963)

1961CPb D Cozzi,F Pantani; J.Electroanal.Chem.,2,72;230 (1961)
 1960COc M Cola; Gazz.Chim.Ital.,90,1037 (1960)
 1959FAa J Forrester,G Ayres; J.Phys.Chem.,63,1979 (1959)
 1959GVa A Grinberg,L Vrublevskaya et al; Zh.Neorg.Khim.,4,1018 (1959)
 1958CPb D Cozzi,F Pantani; J.Inorg.Nucl.Chem.,8,385 (1958)
 1958MPa J van Muylder,M Pourbaix; Cebelcor Rapp.Tech.,59;62 (1958)
 1956JOa C Jorgensen; Acta Chem.Scand.,10,500;518 (1956)
 1952LAb W Latimer; "Oxidation Potentials",Prentice Hall,NY (1952)
 1948DSa F Dwyer,H Schafer; J.Proc.Roy.Soc.,NSW,82,294 (1948)
 1939LAa A Lamb; J.Am.Chem.Soc.,61,699 (1939)
 1938GAa G Grube,H Autenrieth; Z.Elektrochem.,44,296 (1938)
 1937GGa G Grube,B Gu; Z.Elektrochem.,43,397 (1937)
 1930GFa A Grunberg,G Faermann; Z.anorg.Chem., 193,193 (1930)
 1928BVA J Brondted,K Volqvartz; Z.Phys.Chem., 134,97 (1928)

EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
 I Data with various BACKGROUNDS
 H Data for THERMOCHEMICAL quantities
 M Data for TERNARY Complexes

EVALUATION Flags are :-

END Experiments recorded for
 from SC-Database on Saturday, 01 January, 2000 at 01:03:30