

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 10 experiments for

(no ligands specified)

2 metals : Es++, Es+++

(no references specified)

(no experimental details specified)

C12H24O6 L 18-Crown-6 CAS 17455-13-9 (577)
1,4,7,10,13,16-Hexaoxacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Es++	dis alc/w	U	K1=4.70	1993MKa (83350)	1
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Medium: 10 mol/l H₂O in EtOH; for 100% H₂O K₁=2.64

ionic strength ~ 0.003 M, temp. is not indicated

C18H15B L CAS 960-71-4 (2107)

Triphenylboron; $B(C_6H_5)_3$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Es++	sol alc/w	25°C	80%	U	K1=0.79	B2=2.03	1988MKc (96974)	2
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e-	HL	Electron	(442)
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Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Es+++	oth none	25°C	0.0	U	1972MRd	(457)	3
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$$K(Es+e=Es(II)) = -26.2(-1.55V)$$

Method: Estimated data

Es+++	oth none	25°C	0.0	U	1969NBa	(458)	4
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$$K(Es+e=Es(II)) = -27.0(-1.6V)$$

Method: Estimated data

Cl-	HL	Chloride	CAS 7647-01-0	(50)
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Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Es+++ dis NaClO4 25°C 1.0M U K1=-0.02 1972HPa (4804) 5

OH- HL Hydroxide (57)

Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Medium: LiClO4

- 1993MKa N Mikheev,S Kulyukin,I Veleshko; Radiokhim.,35,43 (1993)
1988MKc N Mikheev,S Kulyukhin et al; Radiokhim.,30,218 (1988)
1974HHa S Hubert,M Hussonois,L Brillard et al; J.Inorg.Nucl.Chem.,36,2361 (1974)
1973HHd M Hussonois,S Hubert,L Brillard et al; Radiochem.Radioanal.Lett.,15,47
(1973)
1972HPa H Harmon,J Peterson,W McDowell; Inorg.Nucl.Chem.Lett.,8,57 (1972)
1972HPb H Harmon,J Peterson et al; J.Inorg.Nucl.Chem.,34,1381 (1972)
1972MCc W McDowell,C Coleman; J.Inorg.Nucl.Chem.,34,2837 (1972)
1972MRd N Mikheev,A Rumer; Radiokhim.,14,492(E:502) (1972)
1969NBa L Nugent,R Baybarz,J Burnett; J.Phys.Chem.,73,1177 (1969)
1965Bac R Baybarz; J.Inorg.Nucl.Chem.,27,1831 (1965)

EXPLANATORY NOTES

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END