

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 254 experiments for

(no ligands specified)

2 metals : Ti⁺⁺⁺, Ti⁺⁺⁺⁺

(no references specified)

(no experimental details specified)

e- HL Electron (442)

Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti ⁺⁺⁺	oth	oth/un	25°C	0.00	U				19630Ra	(962) 1
									K(Ti+e=Ti ⁺⁺)=-34, -2000 mV	

Ti ⁺⁺⁺	EMF	oth/un	0°C	var	U				1924FHa	(963) 2
									K(Ti+e=Ti(II))=-6.8(-370 mV)	

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti ⁺⁺⁺	sp	oth/un	25°C	1.61M	U	I		K1=1.61	1975FBa	(5788) 3
Ti ⁺⁺⁺	dis	NaClO4	25°C	4.0M	U				1975HKa	(5789) 4
									K3=0.51 K4=0.33	

Ti ⁺⁺⁺	EMF	non-aq	25°C	100%	U				1971DTb	(5790) 5
									K3=4.92 K4=2.92	

Medium: SeOCl₂, 0.5 M Et₄NClO₄

Ti ⁺⁺⁺	sp	oth/un	rt	var	U			B2=0.37	1971KGa	(5791) 6
									K(TiCl ₂ +3H+4Cl=H ₃ TiCl ₆)=-7.8	

Medium: HCl

Ti ⁺⁺⁺	sp	KCl	25°C	var	U			K1=0.56	1971PLa	(5792) 7
Ti ⁺⁺⁺	sp	oth/un	25°C	0.0	U			K1=-1	1967GAa	(5793) 8

Ti ⁺⁺⁺	ix	NaClO4		3.0M	U			K1=0.55 B2=0.15	1967NKe	(5794) 9
Metal:TiO ⁺⁺ . In LiCl var: K3=-1.03, K4=-1.1										

Ti ⁺⁺⁺	sp	oth/un	?	12.0M	U				1957JOb	(5795) 10
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Medium: HCl. K(Ti(III)Cl₂+Ti(IV)Cl₆=Ti₂Cl₇(?))=1.08

Ti+++ kin NaClO4 40°C 0.50M U K1=0.34 1954DQa (5796) 11

F- HL Fluoride CAS 7644-39-3 (201)
 Fluoride;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti+++ ix KCl ? 0.20M U 1966NAb (7255) 12
 K(TiO+L)=6.65
 K(TiOL+L)=5.09
 K(TiOL2+L)=4.58
 K(TiOL3+L)=4.06

Medium: HCl

OH- HL Hydroxide (57)
 Hydroxide;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti+++ gl KCl 25°C 1.00M C 1988PFa (12272) 13
 *K1=-2.59
 *B(2,2)=-3.03

Ti+++ vlt KCl 25°C 1.0M U 1983TMb (12273) 14
 K[Ti(OH)+H]=2.14

Ti+++ sol oth/un 18°C 1.00M U K1=12.30 B2=22.57 1981NMb (12274) 15
 B3=32.32

Ti+++ nmr oth/un var 1.0M U TI 1978SSd (12275) 16
 *K1=-3.9
 *K1=-3.7 (I=0.3)
 *K1=-3.6 (I=0.1)

Medium: KBr in D2O. Method: esr. 0-60 C. K(Ti(H2O)6=Ti(H2O)5OH+H)=-3.85

Ti+++ kin NaCl 25°C 1.0M C 1977Bmi (12276) 17
 *K1=-2.46

Ti+++ kin KCl 25°C 0.50M U I 1973BLc (12277) 18
 *K1=-1.4

In 0.5 M LiClO4, *K1=-1.4

Ti+++ kin oth/un 25°C 0.50M U 1973LBa (12278) 19
 *K1=-1.9

Medium: LiCl

Ti+++ vlt mixed 25°C U K1=14.0 B2=25.2 1972LIa (12279) 20
 B3=32.4

Medium: ethylene glycol and HCl varied

Ti+++ sp none 25°C 0.00 U 1971PLa (12280) 21
 *K(Ti+H2O=TiO+2H)=-4.5

 Ti+++ gl KCl 25°C 3.00M U 1970KBc (12281) 22
 *K1=-2.77
 *B(2,2)=-3.9

 Ti+++ gl none 25°C 0.0 U T K1=12.71 1962PFa (12282) 23
 K1=12.94(15 C),12.26(35 C)

 Ti+++ gl none ? 0.0 U 1957MOa (12283) 24
 Kso=-53.10

SCN- HL Thiocyanate CAS 463-56-9 (106)
 Thiocyanate;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti+++ vlt NaCl04 25°C 1.00M U K1=1.38 B2=1.78 1980TMb (15272) 25

 Ti+++ sp oth/un 25°C 1.70M U 1969DIb (15273) 26
 K1=0.18 to 0.7

 Ti+++ vlt oth/un 21°C 0.50M U 1958TDa (15274) 27
 K=-0.5(-0.03 mV)

Medium: HCl. K: K(TiOHL+H+e=TiL+H2O)

S04-- H2L Sulfate CAS 7664-93-9 (15)
 Sulfate;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti+++ ix oth/un 18°C 3.00M U K1=1.4 B2=2.3 1978NMa (16589) 28

 Ti+++ sp oth/un 25°C 1.12M U I K1=1.35 1975FBb (16590) 29

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
 Ethanedioic acid; (COOH)2

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti+++ sp oth/un 25°C 1.00M U K1=5.93 B2=11.33 1988LOa (19092) 30
 Medium: LiCl

 Ti+++ vlt NaNO3 25°C 0.20M U B2=12.11 1980MTb (19093) 31
 K(Ti+HL)=6.69

 Ti+++ ix NaCl04 17°C 1.08M U K1=7.15 B2=12.94 1979NMb (19094) 32
 B3=16.54

Medium: LiCl04

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Ti+++      sp  NaCl   10°C  1.0M U          K1=6.45   B2=12.53  1977CDa (19095)  33
-----
Ti+++      sp  KCl    25°C  0.10M U I      B2=8.58           1971PLa (19096)  34
0.2 M, B2=8.47; 0.3 M, B2=8.40; I=0.4, B2=8.33
*****
C2H5NO2          HL   Glycine          CAS 56-40-6  (85)
2-Aminoethanoic acid; H2N.CH2.COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ti+++      gl  KCl    25°C  1.00M C          K1=9.30           1988PFa (21734)  35
B(TiHL)=11.44
B(TiH-1L)=4.48
*****
C3H4O4          H2L   Malonic acid      CAS 141-82-2  (79)
Propanedioic acid; CH2(COOH)2
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ti+++      gl  KCl    25°C  1.00M C          K1=6.83   B2=11.82  1988PFa (24569)  36
K3=2.84
-----
Ti+++      sp  oth/un 15°C  0.50M U          K1=5.46           1986CDa (24570)  37
Medium: LiCl
*****
C3H7NO2          HL   Alanine          CAS 56-41-7  (86)
2-Aminopropanoic acid; H2N.CH(CH3).COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ti+++      EMF oth/un ?    ?  U          K1=8.50           1970FMb (26279)  38
*****
C3H7NO2          HL   B-Alanine        CAS 107-95-9  (575)
3-Aminopropanoic acid; H2N.CH2.CH2.COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ti+++      EMF oth/un ?    ?  U          K1=9.70           1970FMb (26483)  39
*****
C3H7NO3          HL   Serine          CAS 56-45-1  (49)
2-Amino-3-hydroxypropanoic acid; H2N.CH(CH2.OH)COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ti+++      EMF oth/un ?    ?  U          K1=7.60           1970FMb (27186)  40
*****
C3H9O3P          L          CAS 121-45-9  (1786)
Trimethylphosphite; (CH3O)3.P
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Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti+++	nmr	non-aq	20°C	100%	U T HM			1987SEc (28003)	41
K(Ti(2,4-C7H11)2 + L)=2.03									
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T. DH=47.7 kJ mol ⁻¹ , DS=126 J K ⁻¹ mol ⁻¹ . Medium: THF									

C3H9P		L						CAS 594-09-2	(1732)
Trimethyl phosphine; (CH3)3P									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti+++	nmr	non-aq	40°C	100%	U T HM			1987SEc (28057)	42
K(Ti(2,4-C7H11)2 + L)=2.69									
data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T. DH=60.7 kJ mol ⁻¹ ; DS=143 J K ⁻¹ mol ⁻¹ . Medium: THF									

C4H6O4		H2L			Me-Malonic Acid			CAS 516-15-2	(816)
Methylpropanedioic acid; HOOC.CH(CH3).COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti+++	sp	oth/un	15°C	0.50M	U		K1=5.4	1986CDa (30139)	43
Medium: LiCl									

C4H8N2O3		HL			Asparagine			CAS 70-47-3	(17)
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti+++	EMF	oth/un	?	?	U		K1=7.20	1970FMb (32734)	44

C5H5N		L			Pyridine			CAS 110-86-1	(31)
Pyridine, Azine;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti+++	vlt	non-aq	127°C	100%	U		K1=2.75	1967LTa (36684)	45
Medium: Fused ethylpyridinium bromide.									

C5H8O2		HL			Acetylacetone			CAS 123-54-6	(164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti+++	sp	oth/un	25°C	1.0M	U		K1=10.43 B2=18.82 B3=24.9	1967LBa (38103)	46

C5H9NO2		HL			Proline			CAS 147-85-3	(44)
Pyrrolidine-2-carboxylic acid; C4H8N.COOH									

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      EMF oth/un  ?      ?  U      K1=10.05      1970FMb (38648)  47
*****
C5H11NO2      HL      Valine      CAS 72-18-4  (43)
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      EMF oth/un  ?      ?  U      K1=8.20      1970FMb (40764)  48
*****
C6H5NO2      HL      Picolinic acid  CAS 98-98-6  (391)
2-Pyridine-carboxylic acid; C5H4N.COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      oth oth/un 25°C 3.00M U      K1=5.62  B2=11.10  1968PGa (42606)  49
B3=16.58
Medium: KBr. Method: coulometric titration
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Ti+++      gl  oth/un 25°C 3.0M U      K1=5.62  B2=11.10  1968PGc (42607)  50
K3=5.48
Medium: KBr
*****
C6H8O7      H3L      Citric acid      CAS 77-92-9  (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCCH2.CH(OH)(COOH).CH2COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      sp  oth/un 25°C 1.0M U      K(Ti+2H2L)=4.05      1979SFa (46279)  51
Medium: NaBr
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*****
C6H12N2O4      H2L      EDDA      CAS 5657-17-0  (119)
1,2-Diaminoethane-N,N'-diethanoic acid; HOOCCCH2.NH.CH2.CH2.NH.CH2.COOH
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-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      gl  KNO3  25°C 0.10M U      K1=8.75  B2=16.39  1962THb (49275)  52
*****
C6H13NO2      HL      Leucine      CAS 61-90-5  (47)
2-Amino-4-methylpentanoic acid; H2N.CH(CH2.CH(CH3)2)COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      EMF oth/un 25°C 0.10M U      K1=8.50      1970FMb (50113)  53
*****
C6H15O3P      L      CAS 122-52-1  (1723)
Triethylphosphite; (C2H5O)3P
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      nmr non-aq 17°C 100%  U T HM                        1987SEc (51514)  54
                                           K(Ti(2,4-C7H11)2 + L)=2.01
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T.
DH=44.4 kJ mol-1, DS=115 J K-1 mol-1. Medium: THF
*****
C6H15P      L                        CAS 554-70-1 (166)
Triethylphosphine; (C2H5)3P
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      nmr non-aq -4°C 100%  U T HM                        1987SEc (51549)  55
                                           K(Ti(2,4-C7H11)2 + L)=1.09
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T.
DH=41.8 kJ mol-1, DS=135 J K-1 mol-1. Medium: THF
*****
C7H6O3      H2L      Salicylic acid  CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      sp  KCl      20°C 1.00M U      M                        1973VGa (54309)  56
                                           K(TiA2+HL=TiA2L+H)=4.68
                                           K(TiH2A+H2L=(TiH2A)H2L)=11.7
H2A= 4-(2-pyridylazo)resorcinol
*****
C8H11P      L                        CAS 672-66-2 (2290)
Dimethyl-phenyl-phosphine; (CH3)2.P.C6H5
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      nmr non-aq 20°C 100%  U T HM                        1987SEc (61325)  57
                                           K(Ti(2,4-C7H11)2 + L)=2.38
Data for the reaction of open titanocene [Ti(2,4-C7H11)2] with L at var. T.
DH=54.0 kJ mol-1; DS=139 J K-1 mol-1. Medium: THF
*****
C10H8N2      L      2,2'-Bipyridyl  CAS 366-18-7 (25)
2,2'-Bipyridine; (C5H4N)2
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Ti+++      sp  KCl      50°C 0.50M U                        K1=2.86  B2=4.75  1970TNb (69653)  58
                                           B3=6.36
-----
Ti+++      con none      ?      0.0  U                        1959KMa (69654)  59
                                           K3=25.28
*****
C10H16N2O8      H4L      EDTA                        CAS 60-00-4 (120)
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1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	KCl	20°C	1.20M	U			K(Ti+H2L)=7.50	1973YPa (74225)	60

Ti+++	vlt	NaNO3	25°C	0.10M	U	T		K1=21.3 K(TiO+L)=17.3	1954PMb (74226)	61
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Ti+++	vlt	oth/un	?	0.10M	U			K1=17.7	1952BKa (74227)	62
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C12H8N2 L Phenanthroline CAS 66-71-7 (144)
1,10-Phenanthroline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	oth/un	20°C	?	U			K1=3.78 B2=8.29 B3=12.50	1969TNa (80522)	63

C20H14N2O5S H3L Solochrome 6B CAS 3564-14-5 (3507)
1-(1-Hydroxy-2-naphthylazo)-2-naphthol-4-sulfonic acid, Mordant Black3, Eriochrome blue-black B;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	oth/un	?	0.10M	U			K1=3.84 B2=7.82	1972TNa (99665)	64

C20H16N4O5S H2L EriochromeRed B CAS 14954-75-7 (3510)
4-(4,5-Dihydro-3-Me-5-oxo-1-Phe-1H-pyrazol-4-ylazo)-3-naphthol-1-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	oth/un	20°C	0.20M	U			K(Ti+HL)=6.14, pH=3.5-4.5 K(Ti+2HL)=14.49	1972TPb (99798)	65

C23H18O9S H4L Eriochrome cyan CAS 3564-18-9 (433)
4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchsone-5,5'-dicarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti+++	sp	oth/un	?	?	U			K1=5.92 B2=11.60	1973TPb (102637)	66

By polarography: K1=6.89, B2=10.39

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ vlt none 25°C 0.0 U 1961SKc (964) 67
 $K = -3.5 (-209 \text{ mV})$
 $K: \text{Ti}(\text{OHCl})_n + n\text{H} + e = \text{Ti}(\text{III}) + n\text{Cl} + n\text{H}_2\text{O}, n=1 \text{ or } 2$

Ti++++ EMF oth/un 20°C 5.0M U I 1958BGc (965) 68
 $K(\text{Ti}+e) = 2.58 (150 \text{ mV})$
 Medium: H_2SO_4 . $K = 2.24 (I=4; 130 \text{ mV})$, $1.86 (I=3; 108 \text{ mV})$, $1.58 (I=2, 92 \text{ mV})$.
 Data also in HCl : $K = 4.90 (I=10)$, $2.15 (I=4)$ and HBr : $K = 4.07 (I=7)$, $2.96 (I=3)$

Ti++++ vlt oth/un 21°C 0.50M U M 1958TDa (966) 69
 $K = -0.9 (-55 \text{ mV})$
 Medium: HCl . $K: \text{TiOH} + \text{H} + e = \text{Ti}(\text{III}) + \text{H}_2\text{O}$. $K(\text{TiOHSCN} + \text{H} + e = \text{TiSCN} + \text{H}_2\text{O}) = -0.5 (30 \text{ mV})$,

Ti++++ oth none 25°C 0.0 U 1952LAb (967) 70
 $K = -60.8 (-860 \text{ mV})$
 $K(\text{Ti} + 4e = \text{Ti}(\text{s})) = -60 (-880 \text{ mV})$
 $K: \text{TiO}_2(\text{s,hydr}) + 4\text{H} + 4e = \text{Ti}(\text{s}) + 2\text{H}_2\text{O}$. From thermodynamic data

Ti++++ EMF oth/un 18°C 2.0M U 1908DFa (968) 71
 $K(\text{Ti}+e) = 0.95 (56 \text{ mV})$

 Br- HL Bromide CAS 10035-10-6 (19)
 Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	vlt	none	?	0.0	U		1962KSb (2332)	72	

 $K(\text{TiOH} + 2\text{Br}) = 1.56$

 Cl- HL Chloride CAS 7647-01-0 (50)
 Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	ISE	KNO_3	37°C	0.32M	U	M	1985TMb (5797)	73	

 $K(\text{TiA}_2\text{Cl} + \text{Cl} = \text{TiA}_2\text{Cl}_2) = 1.38$

A=Cyclopentadiene

Ti++++ sol KCl 20°C 3.50M U $K_1 = -0.5$ 1970GTc (5798) 74
 $K = -0.44$
 $K: \text{Ti}(\text{OH})_4(\text{s}) + 3\text{H} + 3\text{Cl} = \text{TiOHCl}_3 + 3\text{H}_2$. $*K_{\text{so}} = -2.3 (3.5 \text{ M NaClO}_4)$

Ti++++ sp NaClO_4 25°C 5.0M U I $K_1 = 0.60$ 1969VWa (5799) 75
 Medium: HClO_4 . $K_1 = 0.42 (I=6)$, $0 (I=7)$, $-0.05 (I=7.5)$, $-0.23 (I=8)$

Ti++++ dis oth/un var U 1968SSe (5800) 76
 $K_d(\text{Ti} + 4\text{Cl} + 2\text{TBP}(\text{benzene})) = -1.0$

 ClO4- HL Perchlorate CAS 7001-90-3 (287)
 Perchlorate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	con	oth/un	?	dil	U		K(TiOL+L)=1.61	1961KPd (6381)	77

Ti++++	dis	oth/un	?	?	U	M	Kd(TiOH+H+4L=TiL4(CH3COCH2CH(CH3)2))=-3.63	1955DEa (6382)	78
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F- HL Fluoride CAS 7644-39-3 (201)

Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	ISE	NaClO4	25°C	3.00M	U		K(Ti(OH)2+HF=Ti(OH)2F)=2.28 K(Ti(OH)2+4HF=TiF4)=8.34	1983CPa (7256)	79

Ti++++	kin	NaClO4	25°C	0.50M	U		K(TiO+HL=TiOL+H)>2.30	1977TTa (7257)	80
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Ti++++	ix	oth/un	?	?	U		K6=3.79	1972PAb (7258)	81
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Ti++++	kin	oth/un	25°C	dil	U	T	K(TiF4(OH)+HF+F=TiF6)=5.85	1972RTa (7259)	82
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At 0 C: K=6.48

Ti++++	EMF	oth/un	?	0.50M	U		K6=3.81	1967PMa (7260)	83
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Medium: HCl

Ti++++	EMF	NaClO4	25°C	3.0M	U		K1=>5.38 K2=4.35 K3=3.96 K4=3.72	1960CCa (7261)	84
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Metal:TiO ?. Method: quinhydrone electrode in HClO4

Ti++++	sp	KNO3	?	0.10M	U		K(TiO(?)+L)=6.44	1952KLa (7262)	85
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OH- HL Hydroxide (57)

Hydroxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	nmr	NaClO4	25°C	4.0M	C		K(3TiO=(TiO)3OH+H)=0.21 K(3TiO=Ti3O4+2H)=-0.42 K(4TiO=(TiO)4)=0.36	1987CMd (12284)	86

Method: 170 nmr and light scattering data.

Ti++++	gl	KNO3	37°C	0.32M	U		1985TMb (12285)	87
						*K(TiA2(H2O)2)=-3.5		
						*K(TiA2(H2O)OH)=-4.35		
A=cyclopentadiene								
Ti++++	EMF	NaCl	25°C	2.0M	U	H	1981EKb (12286)	88
Spectroscopy also used. K(TiO+2H2O=TiO(OH)2(sat.)+2H)=-4.7								
K(TiO(OH)2=TiO(OH)2(solid))=-27.3								
Ti++++	sp	oth/un	25°C	0.60M	U	I M	1981TMa (12287)	89
						K(TiL+S04)=1.28		
Ti++++	gl	NaCl	25°C	2.0M	U		1979EIa (12288)	90
						K(8TiO=(TiO)8(OH)12+12H)=-1.68		
Ti++++	sp	KNO3	25°C	0.10M	U	I	K1=14.15 B2=27.88 1971NAe (12289)	91
						K3=13.39		
						K4=13.06		
K1=14.29, K2=13.89, K3=13.58, K4=13.33(I=0.3); K1=14.40, K2=14.02, K3=13.75, K4=13.45(I=0.5); K1=14.70, K2=14.32, K3=14.05, K4=13.74(I=1)								
Ti++++	sol	NaClO4	20°C	3.50M	U		1970GTc (12290)	92
						Kso(Ti(OH)4)=-58.3		
Ti++++	sol	KCl	20°C	3.50M	U		1970GTc (12291)	93
						K(Ti(OH)4(s)=Ti(OH)2)=-30.4		
Ti++++	gl	KCl	?	0.10M	U		1970MMk (12292)	94
						*K1=-2.53		
						*K2=-4.58		
						*K3=-8.7		
						*K4=-14.3		
*K5=-21.5, *K6=-30.0. *Kn: Ti2O5 + nH2O=Ti2O5(OH)n + nH								
Ti++++	dis	NaClO4	25°C	1.00M	U		1969LSd (12293)	95
						B4=53.3		
Ti++++	gl	oth/un	25°C	3.00M	U		1968PGc (12294)	96
						*B(2,2)=-3.3		
						*K1=-2.25		
Medium: KBr								
Ti++++	sol	NaClO4	18°C	0.0	U	I	K1=18.0 B2=35.20 1964NLa (12295)	97
						K3=12.5		
						K4=11.0		
I=0 corr. In 0.1 M NaClO4: K(TiO2(s)+2H2O=Ti(OH)4)=-5.5								
Ti++++	dis	NaClO4	25°C	0.10M	U		1963LCb (12296)	98
						*K2=-1.8		

*K3=-2.4

*K4=-2.1

Ti++++ sol oth/un 25°C 0.1?M U 1963LCb (12297) 99
K(Ti(OH)4(s)=TiOH+3OH)=-39.4

Ti++++ sol none 25°C 0.0 U 1962BGa (12298) 100
K(TiO(OH)2(s)=TiO+2OH)=-29.0

Ti++++ sol none 25°C 0.0 U 1962BGa (12299) 101
K(TiO+OH=TiO(OH))=12.5
K(TiO(OH)2(s)=TiOOH+H)=-16.5

Ti++++ ix oth/un 18°C var U 1962NAd (12300) 102
K(TiO+H2O=TiO(OH)+H)=-1.3
K(TiO+OH=TiO(OH))=12.8

Ti++++ ix NaCl04 25°C var U 1960BHc (12301) 103
*K3=-0.3

Ti++++ dis NaCl04 25°C 2.0M U 1960GAa (12302) 104
*B2=-1.1

*B2: K(TiO+3H2O=Ti(OH)4+2H)

Ti++++ oth oth/un 0°C 12.0M U 1957MTa (12303) 105
*B4=-1.57(?)

Medium:12-17 M HCl. Method:partial pressure of HCl

Ti++++ oth none 25°C 0.0 U 1952LAb (12304) 106
K(TiO(OH)2(s)=TiO+2OH)=ca.-29

method:combination of thermodynamic data

O2-- H2L Peroxide CAS 7772-84-1 (2813)

Peroxide; -0.0-

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp NaCl04 25°C 1.00M U TI 1984THa (12706) 107
K(TiO+H2O2=TiO2+H2O)=3.94

Ti++++ sp oth/un 25°C 2.0M U M K1=4.13 1975MTb (12707) 108

Ti++++ sp NaCl04 25°C 5.46M U TI 1968VWa (12708) 109
K(TiO+H2L)=4.17

K=4.01(I=3.23), 3.94(I=2.11), 3.86(I=1.00). In 1 M HCl04: K=4.42(5 C);
4.15(15 C); 3.87(25 C); 3.55(40 C). Also at I=0 corr

Ti++++ sp none 25°C 0.0 U H 1967VWa (12709) 110
K(TiO+H2L)=3.71

DH=-43.9 kJ mol-1, DS=-75.7 J K-1 mol-1

Ti++++	sp	oth/un	?	var	U		1966BVc (12710) 111
						$K(\text{TiOHL}+\text{H})=-2$ $K(\text{TiO}+\text{HL})=13.85$	
Ti++++	kin	oth/un	18°C	0.30M	U		1966LIa (12711) 112
						$K(\text{TiO}+\text{HL})=12.29$	
Ti++++	sp	oth/un	?	var	U		1964JPa (12712) 113
						$K(\text{TiOC2O4}+\text{H2L})=6.15$	
Ti++++	sp	oth/un	?	1.50M	U	M	1963PJa (12713) 114
							$K(\text{TiOSO4}+\text{H2L}=\text{TiLSO4}+\text{H2O})=4.2$
Ti++++	sp	oth/un	?	?	U		1961VIb (12714) 115
						$K(\text{TiO}+\text{H2L})=4.31$	
Ti++++	sp	oth/un	?	?	U	M	1961VIb (12715) 116
						$K(\text{TiOH2L}+2\text{H}+\text{A}=\text{TiH2LA})=24.82$	
H4A=EDTA							
Ti++++	sp	NaCl04	25°C	3.0M	U	M	1960CLa (12716) 117
						$K(\text{TiO}+\text{H2L})=3.51$ $K(\text{TiOF}+\text{H2L})=4.22$	
Ti++++	dis	NaCl04	25°C	1.0M	U		1960GAa (12717) 118
						$K(\text{TiO}+\text{H2L})=3.9$ $K(\text{TiO}+2\text{H2L})=6.3$	
Ti++++	sp	oth/un	rt	var	U	M	1957MOB (12718) 119
						$K(\text{TiOA}+\text{H2L}=\text{TiAL}+\text{H2O})=6.37$ $K(\text{TiOH2L}+\text{A}=\text{TiAL}+\text{H2O})=20.43$	
H4A=EDTA							
Ti++++	sp	oth/un	rt	var	U		1954GAa (12719) 120
						$K(\text{TiO}+\text{H2L})=4.27$	
Ti++++	sp	oth/un	20°C	var	U		1951BVA (12720) 121
						$K(\text{TiO}+\text{H2L})=4.05$	
Medium: HCl							
Ti++++	sp	oth/un	20°C	30%	U		1948STa (12721) 122
						$K(\text{TiO}+\text{H2L})=5.05$	
Medium:H2SO4.							
Ti++++	sp	oth/un	20°C	var	U		1937RUa (12722) 123
						$K(\text{Ti}(\text{OH})6(?) + \text{H2L})=3.95$	

P04--- H3L Phosphate CAS 7664-38-2 (176)
Phosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	vlt	oth/un	25°C	?	U		1982KNb (13345) 124 pKa(Ti(H2PO4)6)=18.29 (?) pKa(Ti(H2PO4)3)=10.37 (?)		
Ti++++	EMF	NaCl	25°C	2.0M	U	H	1981EKb (13346) 125 K(TiO+HL)=4.48 Ks(TiOHL)=-114.8		
Ti++++	sol	oth/un	20°C	0.50M	U	I	1970GSh (13347) 126 K(TiO+HL)=10.15 K(TiOHL(H2O)x(s)+2H=TiO+H3L)=-5.19 or -6.3 depending on form At I=0 corr: Ks=-14.2 or -15.3		

SCN-		HL					Thiocyanate CAS 463-56-9 (106)		
Thiocyanate;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	sp	NaClO4	25°C	0.10M	U	I	T K1=0.45	1980MTa (15275) 127	
Ti++++	sp	NaClO4	25°C	0.50M	U		T K(TiO+L)=0.52	1977TTa (15276) 128	
In LiClO4. By kinetic methods, K(TiO+L)=0.49									
Ti++++	vlt	oth/un	25°C	0.40M	C		1977VPb (15277) 129 K(TiOH+SCN)=1.65		
Method: cyclic chronopotentiometry. Medium: 0.40 M SCN-.									
Ti++++	sp	non-aq	?	100%	U	I	K1=2.31 B2=4.33 B3=6.25 B4=8.15 B5=10.05 B6=11.94	1973SMd (15278) 130	
Medium:acetone. In DMF: K1=3.20,B2=6.25,B3=9.15,B4=11.91,B5=14.61,B6=17.26									
Ti++++	dis	non-aq	?	100%	U		Kd=1.92	1955DEa (15279) 131	
Kd: K(TiOH+H+4L=TiL4(Me-i-Bu-ketone)+H2O)									
Ti++++	sp	NaClO4	?	1.0M	U		K(TiOH+L)=1.7	1953DEa (15280) 132	

S04--		H2L					Sulfate CAS 7664-93-9 (15)		
Sulfate;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

Ti++++ dis oth/un 20°C 8.0M U 1969BMg (16591) 133
 $K(\text{TiO}+\text{L})=2.23$
 $K(\text{TiO}+2\text{L})=4.11$
 $\text{B3}=11.42$

Medium: H2SO4

Ti++++ sp NaClO4 25°C 4.0M U I 1969VWa (16592) 134
 $K(\text{TiO}+\text{L})=2.26$
 $K(\text{TiO}+2\text{L})=3.80$

Medium: HClO4; $K_1=2.15(I=3)$, $2.47(I=5)$ $2.52(I=0)$

Ti++++ sp oth/un 26°C 1.30M U T 1966GSg (16593) 135
 $\text{B}(\text{TiCr complex})=1.5$
 $\text{B}(\text{TiFe complex})=1.4$ complexes not defined

Ti++++ ix oth/un 18°C var U $K_1=2.40$ 1962NAe (16594) 136

Ti++++ ix oth/un ? var U 1960BHc (16595) 137
 $K(\text{Ti}(\text{OH})_2+\text{HL})=-0.19$
 $K(\text{Ti}(\text{OH})_3+\text{HL})=1.05$

Medium: H2SO4

CH4O L Methyl alcohol CAS 67-56-1 (597)
Methanol; CH3.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ EMF alc/w 20°C 100% U 1971GSa (17908) 138
 $K(\text{Ti}+2\text{L}=\text{Ti}(\text{H}-1\text{L})_2+2\text{H}) > 1$
 $K(\text{Te}(\text{H}-1\text{L})_2+\text{H}-1\text{L})=12.82$
 $K(3\text{TiL}'_3+2\text{L}'=\text{Ti3L}'_{11})=27.47$
 $K(\text{Ti3L}'_{11}+\text{L}'=\text{Ti3L}'_{12})=9.72$

Medium: MeOH, 1 M Me4NCl. $K(2\text{Ti3L}'_{12}+3\text{L}'=3\text{Ti2L}'_9)=13.84$. $\text{L}'=\text{H}-1\text{L}$ (i.e.CH3O)

Ti++++ EMF alc/w 20°C 100% U 1964GUa (17909) 139
 $K(\text{Ti}_2(\text{H}-1\text{L})_8+\text{H}-1\text{L})=11.3$

Method: H electrode. Medium: MeOH, 1.0 M Me4NCl

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ vlt NaNO3 25°C 0.20M U $K_1=6.26$ $\text{B2}=10.78$ 1980MTb (19097) 140
 $K(\text{Ti}+2\text{HL})=5.69$

Ti++++ vlt NaClO4 25°C 1.00M U $K_1=5.18$ 1979TGa (19098) 141

Ti++++ gl NaClO4 25°C 0.50M U 1977VWa (19099) 142
 $K(\text{Ti}(\text{OH})_2+\text{L})=7.90$

$$K(\text{Ti}(\text{OH})_2+2\text{L})=13.2$$

 Ti++++ ix oth/un ? ? U K1=6.51 B2=11.97 1967MNa (19100) 143
 Metal ion: TiO++

Ti++++ dis NaClO4 20°C 0.10M U 1963STc (19101) 144
 $K(\text{TiO}+2\text{L})=10.7$

Ti++++ gl oth/un 25°C 0.03M U 1960GSa (19102) 145
 $K(\text{TiO}+\text{L})=9.7$
 $K(\text{TiOL}+\text{L})=5.11$

By HgC204: $K(\text{TiOL}+\text{L})=4.4$

Ti++++ sp NaClO4 ? 0.02M U 1959BDa (19103) 146
 $K(\text{TiO}+\text{L})=6.6$
 $K(\text{TiOL}+\text{L})=3.3$
 Medium: HClO4

Ti++++ sp oth/un ? ? U K1=6.60 B2=9.90 1959BDd (19104) 147

Ti++++ con oth/un ? ? U K1=2.67 1959BSb (19105) 148

Ti++++ oth oth/un ? ? U K1=1.35 1956KPa (19106) 149

 C2H8N2 L Ethylenediamine CAS 107-15-7 (23)
 1,2-Diaminoethane; H2N.CH2.CH2.NH2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	oth	non-aq	?	100%	U	I M			1962BBa (23236)	150

$$K(\text{TiA4}+\text{L})=2.8$$

Method: freezing point. Medium: benzene. HA=isopropyl alcohol
 In cyclohexane: $K(\text{TiA4}+\text{L})=3.7$, $K(\text{TiA4L}+\text{TiA4})=-5.3$

 C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
 L-2-Hydroxypropanoic acid; CH3.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	NaCl	?	0.30M	U				1971ZPa (25555)	151

$$K(\text{Ti}(\text{OH})_3+2\text{L})=8.61 \text{ at pH } 4$$

 C4H6O5 H2L Malic acid CAS 617-48-1 (393)
 2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	20°C	0.30M	U	M			1970ZHa (30740)	152

$$K(\text{Ti}(\text{OH})_3+\text{L})=6.74$$

Ti++++ dis oth/un 25°C 0.10M U 1968GPc (30741) 153

K(TiO+L)=6.81

C4H6O6 H2L L-Tartaric acid CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; H00C.CH(OH).CH(OH).COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ ix NaClO4 ? 2.00M U 1973ZGc (31374) 154

K(TiO+HL)=2.50

K(TiOHL+HL)=2.12

Ti++++ dis NaClO4 20°C 0.10M U 1963STc (31375) 155

K(TiO+2L)=9.7

C5H5N L Pyridine CAS 110-86-1 (31)
Pyridine, Azine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ oth none 25°C 0.0 U HM 1958ERb (36685) 156

DG(TiF4(s)+2L(g)=TiF4L2(s))=-24.2 kJ mol⁻¹, DH=-46, DS=-75. Data also for
TiCl4, TiBr4 and TiI4

C5H8O2 HL Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ EMF alc/w 20°C 100% U M 1971GSa (38104) 157

K(TiA2L+A)=11.25

K(2TiA3L+3A=Ti2A9+2L)=8.80

K(TiAL2+A)=11.90

K(TiAL2+TiA2L2=Ti2A3L4)=1.95

Medium: MeOH. HA=CH3OH

K(TiA2L2+A=TiA3L+L)=1.35, K(2TiA2L2+5A=Ti2A9+4L)=11.50

C6H6N2O2 HL Cupferron CAS 135-20-6 (637)
N-Nitrosophenylhydroxylamine; C6H5.N(OH).NO

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ dis NaClO4 10°C 3.00M U K1=10.39 B2=20.46 1969Aic (43426) 158

K3=9.75

K4=9.43

C6H6O HL Phenol CAS 108-95-2 (457)
Hydroxybenzene, phenol; C6H5.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

 Ti++++ sp oth/un 25°C 0.02M U M 1981VMa (43546) 159
 $K(\text{TiO}A_2+2\text{HL}=\text{TiO}A\text{L}_2+\text{HA}+\text{H})=-1.66$

H2A=oxalic acid

C6H6O2 H2L Catechol CAS 120-80-9 (534)
 1,2-Dihydroxybenzene, pyrocatechol; H0.C6H4.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ sp oth/un ? 70% U I 1973SPd (43845) 160

$K(\text{TiO}+\text{HL})=1.85$

Medium: H2SO4. In 95% H2SO4, $K(\text{TiO}+\text{HL})=2.56$, $K(\text{TiO}+2\text{HL})=4.75$

 Ti++++ EMF R4N.X 20°C 1.00M U M 1971GSa (43846) 161

$K(\text{Ti}A_2+\text{H}_2\text{L}+2A=\text{Ti}A_2\text{L}+2\text{HA})=27.35$

$K(2\text{Ti}A_2\text{L}+A)=13.80$

Medium: MeOH, 1.0 M Me4NCl. HA=MeOH

$K(\text{Ti}_2A_5\text{L}_2+4\text{H}_2\text{L}+3A=2\text{TiL}_3+8\text{HA})=34.11$

 Ti++++ sp oth/un ? 1.20M U I M B2=50.1 1970PLd (43847) 162

Medium: 1.2 M H2L; Medium: 0.5 M HCl; $B(\text{Ti}A\text{L}_2)=59.4$;

Medium: unknown; $K(\text{Ti}(\text{OH})_2+A+\text{L})=29.3$, H2A=oxalic acid

 Ti++++ sp alc/w 20°C 100% U I 1966SCe (43848) 163

$K(\text{TiO}+\text{HL})=6.1$

Medium: MeOH. $K=4.2(0\%), 4.7(25\%), 5.1(50\%), 5.5(75\%)$. In EtOH/H2O:

$K=4.56(25\%), 4.98(50\%), 5.17(75\%), 6.15(100\%)$

 Ti++++ sp mixed 20°C 100% U I 1966SCe (43849) 164

$K(\text{TiO}+\text{HL})=6.39$

Medium: propanol. $K=4.36(25\%), 4.64(50\%), 5.07(75\%)$. In 2-propanol:

$K=4.28(25\%), 4.73(50\%), 4.9(75\%), 6.75(100\%)$

 Ti++++ sp NaCl04 20°C 0.10M U 1963SOa (43850) 165

$K(\text{TiO}+2\text{H}_2\text{L}=\text{TiO}(\text{HL})_2+2\text{H})=-1.9$

$K(\text{TiO}(\text{HL})_2+\text{H}_2\text{L}=\text{TiL}_3+2\text{H})=-4.7$

$K(\text{TiO}+2\text{H}+3\text{L})=61.6$

 Ti++++ sp oth/un 22°C 0.50M U 1961SKa (43851) 166

$K(\text{TiO}+\text{L})=22.5$

$K(\text{TiOL}+\text{L})=15.9$

Medium: acetate buffer. At I=0.05 M: $K(\text{TiO}+\text{L})=18.8$, $K(\text{TiOL}+\text{L})=17.7$

C6H6O3 H3L Pyrogallol CAS 87-66-1 (696)
 1,2,3-Trihydroxybenzene; C6H3(OH)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++ sp oth/un 23°C 96% U 1981BMe (43985) 167

K(Ti+H3L)=2.13

Medium: 96% H2SO4

Ti++++ sp oth/un ? 95% U 1973SPd (43986) 168

K(TiO+H3L)=1.49

Medium: H2SO4

Ti++++ sp alc/w 20°C 100% U I 1966SCe (43987) 169

K(TiO+H2L)=6.2

Medium: 100% MeOH. K=4.5(0%), 5.1(25%), 5.4(50%), 5.8(75%). 18-20 C

In EtOH: K=4.8(25%), 5.7(50%), 6.0(75%), 6.4(100%)

Ti++++ sp mixed 20°C 100% U I 1966SCe (43988) 170

K(TiO+H2L)=6.6

Medium: 100% propanol. K=4.6(25%), 4.8(50%), 5.6(75%). 18-22 C

In 2-propanol: K=4.7(25%), 4.7(50%), 5.6(75%), 6.8(100%)

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
4,5-Dihydroxybenzene-1,3-disulfonic acid; (HO)2.C6H2(SO3H)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp NaClO4 20°C 0.10M U 1963SOa (44502) 171

K(TiO+2H2L=TiO(HL)2+2H)=-0.3

K(TiO(HL)2+H2L=TiL3+2H)=-2.9

K(TiO+2H+3L=TiL3)=57.6

C6H7N L Aniline CAS 62-53-3 (583)
Aminobenzene, aniline; C6H5.NH2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ dis oth/un ? ? U M 1972BAc (44882) 172

K(TiA3+2HL)=0.27

H2A=pyrocatechol. pH 3-4

C6H8O6 H2L Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp NaClO4 20°C 0.10M U 1963SOB (45662) 173

K(TiO+2HL)=24.8

K(TiO+H2L)=3.1

K(TiO+2H2L)=6.25

K(TiO+2H+3HL)=39.3

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCCH2.CH(OH)(COOH).CH2COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	vlt	NaClO4	25°C	0.50M	U		K(Ti+H3L)=2.98	1978TGa (46280)	174
Ti++++	ix	NaClO4	?	2.00M	U		K(TiO+H2L)=2.91 K(TiOH2L+H2L)=2.49	1973ZGc (46281)	175
Ti++++	sp	NaCl	?	0.30M	U		K(Ti(OH)2+L+HL)=16.28	1971ZHa (46282)	176

C6H9NO6		H3L		NTA			CAS 139-13-9	(191)	
Nitrilotriethanoic acid; N(CH2.COOH)3									

Ti++++	dis	NaClO4	20°C	0.10M	U		K(TiO+L)=12.3	1963STc (47053)	177

C6H12O7		HL		Gluconic acid			CAS 526-95-4	(904)	
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH									

Ti++++	sp	oth/un	25°C	dil	U	I	Keff(TiO+L)=4.66 pH 3 Keff(TiO+L)=4.43 pH 6	1969Mca (49766)	178

C7H6O3		H2L		Salicylic acid			CAS 69-72-7	(14)	
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH									

Ti++++	cal	NaClO4	25°C	0.50M	U		DH(TiO+L)=-51.4 kJ mol-1	1981VVb (54310)	179
for 0.1 M NaClO4 DH1=-46.4 kJ/mol (25 C); -47.7 kJ/mol (15 C); -46.9 kJ/mol (30 C)									
Ti++++	sp	oth/un	20°C	0.10M	U	I	K(TiO+L)=15.66 K(TiO+2L)=24.36	1962BVb (54311)	180
In 2-3 M KNO3: K(TiO+2L)=24.63									
Ti++++	sp	oth/un	35°C	?	U		K1=6.09	1959DGd (54312)	181

C7H6O4		H3L		Resorcylic acid			CAS 89-86-1	(876)	
2,4-Dihydroxybenzoic acid, b-Resorcylic acid; C6H3(OH)2.COOH									

Ti++++	sp	oth/un	25°C	0.10M	U	I	K(TiO+L)=15.66 K(TiO+2L)=24.36	1962BVb (54311)	180

Ti++++ sp oth/un ? 0.10M U 1972AKc (54543) 182
K(Ti(OH)3+2H2L)=6.48

Ti++++ sp oth/un ? ? U 1970AKb (54544) 183
K(TiO+2H2L)=6.01

C7H6O4 H3L Protocatechuic CAS 99-50-3 (875)
3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp NaClO4 20°C 0.10M U 1963SOa (54703) 184
K(TiO+2H3L=TiO(H2L)2+2H)=-1.35
K(TiO(H2L)2+H3L=TiL3+5H)=-3.9
K(TiO+2H+3L)=58.6

C7H6O5 H4L CAS 610-02-6 (3725)
2,3,4-Trihydroxybenzoic acid; (HO)3.C6H2.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp mixed ? 70% U I 1972SPb (54722) 185
K(TiO+H4L)=2.82
Medium: 70% H2SO4. In 95% H2SO4, K=2.44

Ti++++ sp oth/un 18°C 0.10M U 1971AKe (54723) 186
K(Ti(OH)3+H2L)=3.54
K(Ti(OH)3+2HL)=7.61
K(Ti(OH)3+H2L): dil HCl; K(Ti(OH)3+2HL): pH=2.8-7.5

C7H6O5 H4L Gallic acid CAS 149-91-7 (446)
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp mixed ? 95% U 1972SPb (54766) 187
K(TiO+H4L)=2.35
Medium: 95% H2SO4

C7H6O6S H3L CAS 585-42-2 (6136)
2-Hydroxy-4-sulphobenzoic acid, 4-sulfosalicylic acid; HO.C6H3(COOH)(HSO3)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ vlt NaClO4 21°C 0.10M U I 1977UBa (54805) 188
K(TiO+HL=Ti(OH)L)=4.1
In 0.6 M NaClO4: K(Ti(OH)L2+HL=TiL3+H2O)=1.7

C7H6O6S H3L CAS 5965-83-3 (399)

5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; H03S.C6H3(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	sp	KCl	20°C	1.0M	U	M		1973VGa (55056) 189		
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K(TiA2+HL=TiA2L+H)=3.44

K(TiH2A+H2L=(TiH2A)H2L)=11.40

H2A=4-(2-pyridylazo)-resorcinol

Ti++++	sp	oth/un	20°C	0.10M	U			1963S0a (55057) 190		
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K(TiO+2H+3L)=42.2

K(TiO+2HL)=5.4

K(TiO+HL)=3.1

C7H7NO2 H2L Salicylaldehyde oxime; CAS 94-67-7 (1486)

2-Hydroxybenzaldehyde oxime; H0.C6H4.CH:N.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	gl	KCl	25°C	0.10M	U	I	K1=16.30 B2=31.15	1968MDe (55314) 191		
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K1=18.5(I=0),18.29(I=0.01),17.74(I=0.025),17.35(I=0.05),16.86(I=0.075);

K2=17.2(I=0),16.88(I=0.01),16.62(I=0.025),16.07(I=0.05),15.66(I=0.075)

C7H8O2 HL CAS 150-19-6 (4412)

5-Methoxyphenol; H0.C6H4.OCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	sp	mixed	?	90%	U			1973SPd (56097) 192		
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K(TiO+HL)=1.92

Medium: 90% H2SO4

C8H5O2F3S HL TTA CAS 326-91-0 (165)

4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	dis	NaClO4	25°C	1.0M	U		K1=7.87 B2=15.52	1969LSe (58686) 193		
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K3=7.45

K4=7.23

C9H6O4 H2L Esculetin CAS 305-01-1 (3853)

6,7-Dihydroxycoumarin;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ti++++	sp	alc/w	20°C	20%	U			1964JSb (63954) 194		
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K(?)=8.8

Medium: 20% EtOH, 0.4 M NaClO4

C9H7NO HL Oxine CAS 148-24-3 (504)
8-Hydroxyquinoline (8-quinolinol);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis	NaClO4	20°C	1.0M	U			K1=13.22 B2=25.94 K3=12.26 K4=11.0	1967SLa (64359)	195

C9H7N3O2S H2L TAR CAS 2246-46-0 (707)
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	alc/w	25°C	50%	U				1967NPb (64729)	196
									K(TiO+HL)=13	

Medium: 50% MeOH, 0.1 M NaClO4

C9H10O2 L 4-Tolyl-acetate CAS 140-39-6 (3857)
Ethanoic acid 4-methylphenyl ester; CH3.CO.O.C6H4.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	non-aq	60°C	100%	U T H				1966GSd (65382)	197
									K(TiCl4+L)=1.93	

Medium: 1,2-dichloroethane. K=2.60(25 C). DH=-37.6 kJ mol⁻¹, DS=-71 J K⁻¹m⁻¹

C10H6O8Br2S2 H4L CAS 58425-38-0 (2003)
2,7-Dibromo-1,8-dihydroxy-naphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	NaClO4	20°C	0.10M	U				1975MDa (68535)	198
									B(Ti(OH)2(HL)2)=10.98	

C10H6O8Cl2S2 H4L CAS 6155-33-5 (4761)
2,7-Dichlorochromotropic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	NaClO4	20°C	0.50M	U				19700Mb (68537)	199
									K(TiO+2HL)=7.38	

C10H8O2 H2L CAS 569-42-6 (4699)
1,8-Dihydroxynaphthalene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	25°C	0.10M	U				1968BNc (69754)	200
									K(Ti(OH)2+2H2L=Ti(OH)2(HL)2+2H)=15.34	

$K(\text{Ti}(\text{OH})_3 + 3\text{H}_2\text{L} = \text{Ti}(\text{OH})(\text{HL})_3 + \text{H}) = 28.0$

Ti++++ sp non-aq ? 100% U I 1966SCa (69755) 201

$K(\text{TiOH} + \text{HL} = \text{TiOHL} + \text{H}) = 7.0$: acetone

K=6.38: in dimethylformamide

K=6.08: in dioxan

K=6.33: in ethanol

K=6.10 in methanol, K=6.52 in propanol, K=6.85 in 2-propanol

C10H8O4 H2L 4-Me-Esculetin CAS 529-84-0 (3890)

4-Methyl-6,7-dihydroxycoumarin

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp alc/w 20°C 20% U 1964JSb (69791) 202

B3=10.7

Medium: 20% EtOH, 0.4 M NaClO4

C10H8O4 H2L 4-Me-daphnetin CAS 2107-77-9 (6317)

7,8-Dihydroxy-4-methylcoumarin;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp alc/w ? 50% U 1976SSe (69792) 203

$K(\text{TiO} + \text{HL}) = 8.37$

$K(\text{TiOHL} + \text{HL}) = 7.33$

$K(\text{TiO}(\text{HL})_2 + \text{HL}) = 5.70$

C10H8O5S H3L DHNSA (877)

2,3-Dihydroxynaphthalene-6-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl oth/un 20°C 0.10M U 1963SOb (69865) 204

$K(\text{TiO} + 2\text{L}) = 38.1$

$K(\text{TiO} + 3\text{L}) = 54.7$

$K(\text{TiO} + 2\text{H} + 3\text{L}) = 56.5$

$K(\text{TiO} + 2\text{H}_2\text{L} = \text{TiO}(\text{HL})_2 + 2\text{H}) = -0.69$

C10H8O8S2 H4L Chromotropic ac CAS 148-25-4 (1875)

1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp NaClO4 25°C 0.10M U 1975BUb (69972) 205

K1eff=7.72 at pH 5.32

B2eff=11.85 at pH 5.32

B(2,2)eff=16.11 at pH 5.32

Ti++++ sp oth/un 25°C 0.10M U 1968BNc (69973) 206

$K(\text{Ti}(\text{OH})_2 + 2\text{H}_2\text{L} = \text{Ti}(\text{OH})_2(\text{HL})_2 + 2\text{H}) = 14.36$

$K(\text{Ti}(\text{OH})_3 + 3\text{H}_2\text{L} = \text{Ti}(\text{OH})(\text{HL})_3 + \text{H}) = 25.92$

Ti++++ sp mixed 20°C 100% U I 1966CSb (69974) 207

$K(\text{TiO} + \text{HL}) = 7.34$

$K(\text{TiO} + 2\text{HL}) = 12.17$

Medium: DMF/H₂O, TiO 0.005 M: K₁=4.80(0% DMF), 5.60(25%), 6.26(50%-74%);

With 0.0025 TiO:K₁=5.05(0%), 5.57(25%), 6.12(50%), 6.57(74%), 7.38(100%)

Ti++++ sp NaClO₄ 20°C 0.10M U 1963SOa (69975) 208

$K(\text{TiO} + 2\text{L}) = 40.5$

$K(\text{TiO} + 3\text{L}) = 56.4$

$K(\text{TiO} + 2\text{H} + 3\text{L} = \text{TiL}_3) = 60.5$

$K(\text{TiOL}_2 + 2\text{H} = \text{TiO}(\text{HL})_2) = 4.4$

Ti++++ sp oth/un 20°C 0.10M U B₂=6.18 1959SOc (69976) 209
B₃=10.59

Ti++++ sp oth/un 20°C 0.10M U 1957BPc (69977) 210

$K(\text{Ti} + \text{H}_2\text{L}) = 3.99$

C₁₀H₁₂O₂ HL CAS 1946-74-3 (202)

3-Isopropyltropolone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp alc/w 25°C 50% U B₂=21.17 1961HSa (71609) 211
B₃=28.95

Medium: 50% EtOH, 0.01 M

C₁₀H₁₆N₂O₈ H₄L EDTA CAS 60-00-4 (120)

1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ vlt KNO₃ 25°C 0.20M U K₁=22.61 1986ZFa (74228) 212

Ti++++ gl KNO₃ 25°C 0.10M C T H K₁=18.47 1985HWc (74229) 213

Data for 5-35 C. Metal is TiO⁺⁺. Method: Hg and glass electrodes,
competition with Hg⁺⁺. DH(K₁)=-31.3 kJ mol⁻¹, DS(K₁)=250 J K⁻¹ mol⁻¹.

Ti++++ sp NaClO₄ 20°C 1.0M U 1971KNa (74230) 214

$K(\text{TiO} + \text{L}) = 18.15$

$K(\text{TiO} + \text{HL}) = 12.08$

Ti++++ dis NaClO₄ 20°C 0.10M U T 1963STc (74231) 215

$K(\text{TiO} + \text{L}) = 17.5$

C₁₁H₉N₃O₂ H₂L PAR CAS 1141-59-9 (636)

4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C₅H₄N.N:N.C₆H₃(OH)₂

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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Ti++++    sp  oth/un 20°C 1.00M U                        1974LKd (77587) 216
K(Ti(OH)2+HL+A=Ti(OH)2HLA)=2.72(HA-ethanoic acid); 21.12(HA=chloroethanoic
acid);20.08(HA=dichloroethanoic acid); 19.74(HA=trichloroethanoic acid)
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Ti++++    sp  oth/un 20°C 1.00M U                        1974LKd (77588) 217
B(TiL2A2)=47.43(HA-ethanoic acid); 46.68(HA=chloroethanoic acid);
45.81(HA=dichloroethanoic acid); 45.45(HA=trichloroethanoic acid)
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Ti++++    sp  oth/un  ?      ?  U      M                  1967SHa (77589) 218
                                           K(TiOA+HL)=13.25
HA=ethanoic acid

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C13H11NO2          HL                      CAS 304-88-1 (181)
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Ti++++    dis oth/un 20°C 1.0M U                    K1=11.77 B2=23.40 1970LSd (85181) 219
                                           K3=11.49
                                           K4=11.35

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C13H14NO3P        H2L                      CAS 19316-85-7 (1466)
2-Hydroxyphenyl-N-phenylaminomethylphosphinic acid;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Ti++++    gl  NaClO4 20°C 0.10M U                    K1=7.20      1985SIb (85566) 220
*****
C13H14N3O5P        H2L                      CAS 80767-75-5 (1467)
2-Hydroxy-4-nitrophenyl-N-(2-pyridylmethyl)aminomethylphosphinic acid;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Ti++++    gl  NaClO4 20°C 0.10M U                    K1=7.20      1985SIb (85644) 221
*****
C13H14N3O5P        H2L                      CAS 80767-76-6 (1468)
2-Hydroxy-4-nitrophenyl-N-(3-pyridylmethyl)aminomethylphosphinic acid;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Ti++++    gl  NaClO4 20°C 0.10M U                    K1=8.10      1985SIb (85657) 222
*****
C13H15N2O3P        H2L                      CAS 80767-72-2 (1460)
2-Hydroxyphenyl-(N-2-pyridylmethylamino)methylphosphinic acid;
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Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
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Ti++++ gl NaClO4 20°C 0.10M U K1=11.80 1985SIa (85784) 223

C13H15N2O3P H2L CAS 80767-73-3 (1461)
2-Hydroxyphenyl-(N-3-pyridylmethylamino)methylphosphinic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl NaClO4 20°C 0.10M U K1=11.70 1985SIa (85797) 224

C13H15N2O3P H2L CAS 80767-74-4 (1462)
2-Hydroxyphenyl-(N-4-pyridylmethylamino)methylphosphinic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl NaClO4 20°C 0.10M U K1=11.75 1985SIa (85810) 225

C13H15N2O4P H3L CAS 80767-78-8 (1463)
2-Hydroxyphenyl-(N-2-pyridylmethylamino)methylphosphonic acid;
C6H4(OH)CH(PO3H2).NH.CH2.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl NaClO4 20°C 0.10M U K1=16.80 1985SIa (85823) 226

C13H15N2O4P H3L CAS 85946-85-6 (1464)
2-Hydroxyphenyl-(N-3-pyridylmethylamino)methylphosphonic acid;
C6H4(OH)CH(PO3H2).NH.CH2.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl NaClO4 20°C 0.10M U K1=16.90 1985SIa (85836) 227

C13H15N2O4P H3L CAS 85946-86-7 (1465)
2-Hydroxyphenyl-(N-4-pyridylmethylamino)methylphosphonic acid;
C6H4(OH)CH(PO3H2).NH.CH2.C5H4N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl NaClO4 20°C 0.10M U K1=17.15 1985SIa (85849) 228

C13H20N04P H3L (1471)
2-Hydroxyphenyl-N-(cyclohexylamino)methylphosphonic acid;
C6H4(OH)CH(PO3H2).NH.C6H11

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl NaClO4 20°C 0.10M U K1=15.05 1985SIb (86094) 229

C14H12O3 HL Benzilic acid CAS 76-93-7 (710)
Diphenylglycolic acid, (benzilic acid); (C6H5)2C(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	vlt	KNO3	25°C	0.20M	U		K1=9.40	1986ZFa (87352)	230

C14H16NO3P		H2L					CAS 25881-35-0	(1469)	
Phenyl-N-(benzylamino)methylphosphonic acid; C6H5.CH(PO3H2).NH.CH2.C6H5									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	gl	NaClO4	20°C	0.10M	U		K1=12.60	1985SIb (87813)	231

C14H16NO4P		H3L					CAS 61146-25-6	(1470)	
2-Hydroxyphenyl-N-(benzylamino)methylphosphonic acid; C6H4(OH)CH(PO3H2).NH.CH2.C6H5									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	gl	NaClO4	20°C	0.10M	U		K1=15.60	1985SIb (87826)	232

C14H17N2O4P		H3L					(1472)		
2-Hydroxyphenyl-N-(2-(2'-pyridyl)ethylamino)methylphosphonic acid;C6H4(OH)CH(PO3H2)NHCH2CH2C5H4N									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	gl	NaClO4	20°C	0.10M	U		K1=15.60	1985SIb (88046)	233

C14H22N2O8		H4L		CDTA			CAS 482-54-2	(200)	
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	gl	KNO3	25°C	0.10M	C T H		K1=20.91	1985HWc (88797)	234
Data for 5-35 C. Metal is TiO++. Method: Hg and glass electrodes, competition with Hg++. DH(K1)=-37.8 kJ mol-1, DS(K1)=274 J K-1 mol-1.									

Ti++++	sp	NaClO4	20°C	0.10M	U		K1=18.23 K(TiO+HL)=11.14 K(TiO+H2L)=8.33	1972NKc (88798)	235

Ti++++	dis	NaClO4	20°C	0.10M	U		K(TiO+L)=19.9	1963STc (88799)	236

C14H23N3O10		H5L		DTPA			CAS 67-43-6	(238)	
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	gl	KNO3	25°C	0.10M	C T H		K1=23.05	1985HWc (89412)	237
Data for 5-35 C. Metal is TiO++. Method: Hg and glass electrodes,									

competition with Hg⁺⁺. DH(K1)=-104.8 kJ mol⁻¹, DS(K1)=89.7 J K⁻¹ mol⁻¹.

Ti++++ sp oth/un 20°C dil U K1=23.38 1970KAf (89413) 238
K(Ti+HL)=14.51

C14H24N2O10 EGTA CAS 67-42-5 (349)
Ethyleneglycol-0,0'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ gl KNO3 25°C 0.10M C T H K1=20.08 1985HWc (89949) 239

Data for 5-35 C. Metal is TiO⁺⁺. Method: Hg and glass electrodes,
competition with Hg⁺⁺. DH(K1)=-80.3 kJ mol⁻¹, DS(K1)=115 J K⁻¹ mol⁻¹.

C15H10O7 H5L Quercetin CAS 117-39-5 (5101)
3,5,7-Trihydroxy-2-(3',4'-dihydroxyphenyl)-1-benzopyran-4-one;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp non-aq 25°C 100% U 1969DSc (91025) 240

K(?)=4.30

Medium: BuOH

C15H13NO2 HL CAS 7369-44-0 (4066)
N-3-Diphenylpropenohydroxamic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ dis oth/un 20°C 1.0M U K1=13.3 B2=26.40 1970LSd (91645) 241

K3=12.9

K4=12.7

C16H11NO3 HL HPBI CAS 41836-94-6 (7740)
3-Phenyl-4-benzoyl-5-isoxazolone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ dis non-aq 30°C 100% U 1999SPa (92687) 242

Kd=1.91

Kd: TiO+2HL(org)=TiOL2(org)+2H.

Method: Solvent extraction, H2O/xylene.

C16H14O5 H3L CAS 966-64-3 (5143)
2,3,7-Trihydroxy-9-propylfluorone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ti++++ sp alc/w ? 4% U 1967NBa (93591) 243

K(TiOH+2H2L)=32.72

K(Ti(OH)2+2H2L)=25.70

Medium: 4% EtOH, 0.1 M

C17H14N2O2 L CAS 4551-69-3 (698)

4-Benzoyl-3-methyl-1-phenyl-2-pyrazolin-5-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	dis	oth/un	20°C	1.0M	U			K1=8.11 K3=7.76 K4=7.58	1969LSb (95902)	244

C17H17NO3 HL CAS 58434-59-6 (1213)

2'-Hydroxy-4-methoxy-5'-methylbenzylidene acetophenone oxime

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	30°C	8.00M	U	M		K(TiO(SCN)+L)=2.29 K(TiO(SCN)L+L)=1.80	1980GKa (96192)	245

C18H26N2O6P2 H4L CAS 53431-86-0 (5266)

Ethylenebis(imino(2-hydroxyphenyl)methylene(methyl)phosphinic acid);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	EMF	oth/un	?	?	U			K(TiO+2H2L)=8.46 B(TiO+2L)>15	1970DMc (97675)	246

C19H12O6 H4L Salicylfluorone (5269)

9-(2-Hydroxyphenyl)-2,3,7-trihydroxy-6-fluorone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	alc/w	?	4%	U			K(TiOH+2H2L)=26.19 K(Ti(OH)2+2H2L)=26.19	1967NBa (98996)	247

Medium: 4% EtOH, 0.1 M

C19H15NO8 H4L Alizarin Comp. CAS 3952-78-1 (671)

(3,4-Dihydroxy-2-anthraquinonyl-methyl)iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ti++++	sp	oth/un	RT	dil	C			K1eff=4.2 B2eff=8.6	1982EDa (99141)	248

Medium: borax buffer, pH 10.

C21H19N3O8S H4L MeNaptholOrange (4151)

N-(1'-Hydroxy-4'-(4''-sulfophenylazo)-2'-naphthylmethyl)-iminodiethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	sp	NaNO3	20°C	0.20M	U		B2=22.96	1963BUb (101142)	249

C23H16O9Cl2S H4L Chrome azurol S CAS 1667-99-8 (711)									
Chromazurol S;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	sp	oth/un	?	?	U			1969TKc (102575)	250
							K(Ti(OH)2+HL)=5.64		

C23H24N4O2 L Trichachnine CAS 1251-85-0 (2606)									
4,4'-Diantipyrilmethane,									
4,4'-phenylmethylene-bis-(1,2-dihydro-1,5-dimethyl-2-phenylpyrazol-3-one									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	sp	oth/un	20°C	1.0M	U			1962BTc (102679)	251
							B3=7.89		

C27H30O16 H4L Rutin CAS 153-18-4 (4169)									
3,3',4',5,7-Pentahydroxyflavone-3-beta-rutinoside;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	sp	alc/w	RT	50%	C			2000KMa (104509)	252
Medium: 50% EtOH/H2O. K(TiO(ox)2+2HL=TiO(0x)2L2+2H)=10.80 at pH 6.50.									

C31H32N2O13S H6L Xylenol orange CAS 63721-85-5 (432)									
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulfonic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ti++++	sp	NaNO3	20°C	0.20M	U			1963BGa (105500)	253
							B(Ti2L2)=57.8		

Ti++++	sp	NaClO4	25°C	0.50M	U	I		19630Ta (105501)	254
							K(TiO+H6L=TiOH5L+H)=3.46		
In 0.05 M HClO4: K(TiO+H6L+H2O2=TiH6LH2O2)=37.68									

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EXPLANATORY NOTES

DATA Flags are :-

T Data at other TEMPERATURES
 I Data with various BACKGROUNDS
 H Data for THERMOCHEMICAL quantities
 M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC

END