



Sofia Gama 21 November 2021

Neodymium(III)

Reaction	Baes and Mesmer, 1976	NIST46	Neck et al., 2009	Brown and Ekberg, 2016
$Nd^{3+} + H_2O \rightleftharpoons Nd(OH)^{2+} + H^+$	-8.0	-8.0	-7.4 ± 0.4	-8.13 ± 0.05
$Nd^{3+} + 2 H_2O \rightleftharpoons Nd(OH)_2^+ + 2 H^+$	(-16.9)		-15.7 ± 0.7	
$Nd^{3+} + 3 H_2O \rightleftharpoons Nd(OH)_{3(aq)} + 3 H^+$	(-26.5)		-26.2 ± 0.5	
$Nd^{3+} + 4 H_2O \rightleftharpoons Nd(OH)_4^- + 4 H^+$	(-37.1)	-37.4	-40.7 ± 0.7	
$2 \text{ Nd}^{3+} + 2 \text{ H}_2\text{O} \rightleftharpoons \text{Nd}_2(\text{OH})_2^{4+} + 2 \text{ H}^+$	-13.86	-13.9		-15.56 ± 0.20
$3 \text{ Nd}^{3+} + 5 \text{ H}_2\text{O} \rightleftharpoons \text{Nd}_3(\text{OH})_5^{4+} + 5 \text{ H}^+$	<-28.5			-34.2 ± 0.3

$Nd(OH)_3(s) + 3 H^+ \rightleftharpoons Nd^{3+} + 3 H_2O$	18.6		17.2 ± 0.4	17.89 ± 0.09
$Nd(OH)_3(s) \rightleftharpoons Nd^{3+} + 3 OH^{-}$		-23.2 ± 0.9	-21.5 (act)	
			-23.1(inact)	

C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976.

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 135-145.

V. Neck, M. Altmaier, T. Rabung, J. Lützenkirchen and T. Fanghänel. Pure Appl. Chem., 81, 1555–1568 (2009).

NIST46, NIST Critically Selected Stability Constants of Metal Complexes: Version 8.0. Available at: www.nist.gov/srd/nist46