

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 899 experiments for

(no ligands specified)

Metal : Li+

(no references specified)

(no experimental details specified)

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	EMF	mixed	25°C	10%	U	I		1974DKb K(Li+e=Li(s))=-51.40(-3041mV)	(627)	1
Medium: 10% w/w DMSO/H2O; K=-51.30(-3.035V,20%), -50.97(-3.015V,40%), -50.75(-3.002V,60%)										
Li+	oth	mixed	25°C	0.0	U	I		1972C0a K(Li+e+Li(s))=-50.73(3001mV)	(628)	2
Method: Estimated.K(Li+e=Li(s)).MeOH: -53.86(-3.186V).EtOH: -53.13(-3.143V) BuOH: -54.21(-3.207V).PentylOH: -54.58(3.229V).Me2CO: -54.21(-3.207V)										
Li+	oth	none	25°C	0.0	U	I		1972C0a K(Li+e=Li(s))=-50.73(3001mV)	(629)	3
Method: Estimated. MeCN: -56.05(-3.316V).HCOOH: -59.72(-3.533V) Also NH3 and N2H4										
Li+	con	non-aq	-71°C	100%	U			1972DBa K(Li + e(solv))=2.72 K(2Li=Li2)=1.95	(630)	4
Medium: NH3(liquid). Method:conductivity and magnetic susceptibility										
Li+	EMF	mixed	25°C	30%	U	I		1972KRb K(Li+e=Li(s))=-51.32(-3036mV)	(631)	5
Med.:30% w/w ethylene glycol/H2O; K=-51.18(-3.028V,50%), -50.95(-3.014V,70%) -51.15(-3.026V,90%), -52.35(-3.097V,100%)										
Li+	EMF	non-aq	25°C	100%	U	I		1972KRc K(Li+e=Li(s))=-52.08(-3081mV)	(632)	6
Medium: 30% w/w propylene glycol/MeOH. 0% Pr Glycol: K=-52.30(-3.094V) 50%: -52.00(-3.076V). 70%: -51.96(-3.074V). 100%: -52.00(-3.076V)										
Li+	EMF	none	25°C	0.0	M			1968HBb K(Li+e=Li(s))=-51.39,-3040.1mV	(633)	7
Li+	EMF	none	25°C	0.0	U			1967BHc K(Li+e=LiHg)=-36.99, -2188 mV	(634)	8

Li+ EMF none 25°C 0.0 M 1967BHc (635) 9
K(Li+e=LiHg)=-37.13, -2196.3 mV

Li+ EMF non-aq 25°C 100% U 1966LCa (636) 10
K'=-52.806, -3123.7 mV
Medium: CH3NHCHO. K': Li + Cl + Ag(s) = Li(s) + AgCl(s)

Li+ oth none 25°C 0.0 U 1952LAB (637) 11
K(Li+e)=-51.47(-3045 mV)

Li+ EMF none 25°C 0.0 U 1923LRa (638) 12
K(Li+e=Li(s))=-50.02(-2957.8mV

AsF6- L (8856)
Tetrafluoroarsenate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	C		K1=1.716	2002DDa (1040)	13
-----	-----	--------	------	------	---	--	----------	----------------	----

Medium: N,N-dimethylacetamide, 0.005-0.015 M LiAsF6.

BF4- HL (2497)
Tetrafluoroborate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	C		K1=1.477	2002DDa (1195)	14
-----	-----	--------	------	------	---	--	----------	----------------	----

Medium: N,N-dimethylacetamide, 0.005-0.015 M LiBF4.

Li+	con	non-aq	25°C	100%	C T		K1=2.50	2000VMa (1196)	15
-----	-----	--------	------	------	-----	--	---------	----------------	----

Medium: 2-Methoxyethanol. Data for 15-35 C.

Li+	con	non-aq	25°C	100%	C		K1=8.30 B(Li2BF4)=10.10	1997CHb (1197)	16
-----	-----	--------	------	------	---	--	----------------------------	----------------	----

Medium: THF. By conductivity, species M2L and L2M are equivalent.

Li+	con	non-aq	25°C	100%	U		K1=1.00	1991MHa (1198)	17
-----	-----	--------	------	------	---	--	---------	----------------	----

Medium: propylene carbonate

B04H4- HL Borate CAS 10043-35-3 (991)
Borate; B(OH)4-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	EMF	oth/un	25°C	0.0	C T H		K1=1.09	2000ZSb (1313)	18
-----	-----	--------	------	-----	-------	--	---------	----------------	----

Medium: 0.007-0.23 M LiCl. Method: Pt/H2 electrode. DH(K1)=0.64 kJ mol⁻¹, DS(K1)=22.9 J K⁻¹ mol⁻¹.

Li+	sp	oth/un	25°C	1.00M	U I		K1=0.73	1990RAa (1314)	19
-----	----	--------	------	-------	-----	--	---------	----------------	----

Medium: LiCl. Data at I=0 M and at pressures to 2041 atmos.

Li+	gl	NaCl	25°C	0.70M	U		K1=-0.05	1988RBa	(1315)	20

Br-		HL		Bromide			CAS 10035-10-6	(19)		
Bromide;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo	
Li+	con	non-aq	25°C	100%	C		K1=1.520	2002DDa	(2083)	21
Medium: N,N-dimethylacetamide, 0.005-0.015 M LiBr.										
Li+	con	non-aq	25°C	100%	C		K1=10.53 B(Li2Br)=13.33 K(2LiBr=Li2Br2)=1.70	1997CHb	(2084)	22
Medium: THF. By conductivity, species M2L and L2M are equivalent.										
Li+	con	non-aq	25°C	100%	U T		K1=2.23	1993TAa	(2085)	23
Medium: 2-methoxyethanol, -10 to 80 C										
Li+	con	non-aq	25°C	100%	U		K1=4.71	1982GRb	(2086)	24
Medium: octanol										
Li+	con	non-aq	25°C	100%	U		K1=0.76	1974HPb	(2087)	25
Medium: hexamethylphosphotriamide. K1 by Pitts eqn. K1=1.13 (Fuoss-Hsia eqn)										
Li+	con	mixed	25°C	0.10M	U I		K1=3.53	1973BHa	(2088)	26
In 99.9% w/w acetone/H2O. K1=3.62(100%),3.44(99.7%),3.38(99.4%),3.32(99%),3.21(98.5%),3.12(98%),2.96(97%),2.67(95%),2.15(90%),1.44(80%)										
Li+	con	mixed	25°C	0.1%	U I		K1=3.54	1973NIa	(2089)	27
Medium: 0.1% w/w MeOH/acetone. K1=3.42(0.3%), 3.19(1%), 3.03(2%), 2.78(5%), 2.50(=10%), 2.02(20%), 0.79(50%)										
Li+	con	non-aq	25°C	100%	U		K1=4.98	1973TKb	(2090)	28
Medium: liquid SO2										
Li+	kin	mixed	25°C	0.00	U I		K1=3.56	1972HBa	(2091)	29
In 99.995% w/w acetone/H2O.K1=3.54(99.894%),3.40(99.695%),3.32(99.395%),3.36(98.995%),3.28(98.495%). Data also by conductivity										
Li+	con	non-aq	25°C	100%	U		K1=1.26	1971BCa	(2092)	30
Medium: tetramethylurea										
Li+	kin	non-aq	25°C	100%	U		K1=3.63	1970BIa	(2093)	31
Medium: acetone. By conductivity :K1=3.67										
Li+	EMF	non-aq	25°C	100%	U		K1=0.40	1970SAb	(2094)	32
Medium: propene carbonate										
Li+	con	non-aq	25°C	100%	U		K1=1.28	1969MBf	(2095)	33

Medium: propene carbonate(0 corr)

Li+ con diox/w 25°C 50% U TI K1=1.11 1969SMc (2096) 34
In 50% w/w dioxan/H2O. K1=1.78(70%),3.08(77.5%). At 50 C: K1=1.63(50%),
3.34(70%),4.76(77.5%). Also 30, 35 and 40 C

Li+ con non-aq 40°C 100% U T K1=2.90 1967SMb (2097) 35
Medium:Me2CO. K1=2.80(25 C),2.77(30 C),2.85(35 C); also Me2CO-H2O mixtures

Li+ con non-aq 25°C 100% U K1=3.66 1966SAa (2098) 36
Medium: acetone

Li+ con alc/w 25°C 100% U K1=1.19 1966SMc (2099) 37
Medium:MeOH, also K1 values for MeOH-H2O mixtures

Li+ con non-aq 25°C 100% U K1=3.18 1965BFb (2100) 38
Medium: diaminoethane

Li+ oth non-aq 35°C 100% U K(2Li2Br2=Li4Br4)=1.3 1964TRb (2101) 39
Method:boiling point. Medium:Et2O

Li+ kin non-aq 0°C 100% U K1=0.41 1964WHa (2102) 40
Medium:DMF

Li+ con non-aq 0°C 100% U K1=4.58 1960LRb (2103) 41
Medium: liquid SO2, I=0 corr., 0.22 C

Li+ oth non-aq 16°C 100% U K(2LiBr=Li2Br2)=0.66 1959KEb (2104) 42
Method: freezing point; medium: CH3CO2H; m units.

Li+ con non-aq 30°C 100% U K1=6.14 1954JGa (2105) 43
Medium: CH3CO2H

BrO3- HL Bromate (6017)
Bromate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sol NaClO4 25°C 0.15M U I K1=-0.77 1963RSe (2422) 44
Medium: LiClO4. K1=-0.82 (I=0.20)

CO3-- H2L Carbonate CAS 465-79-6 (268)
Carbonate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sol oth/un 40°C var U T K(Li2L+CO2(g)=2Li+2HL)=-0.23 1958MLa (3261) 45

K=-2.63(200 C), -3.68(250 C), -4.91(290 C), m units

Li+ sol oth/un 20°C var U 1958VGa (3262) 46
Kso(Li2CO3(s))=-1.6

C6N6Fe---- H4L (2191)

Hexacyanoferrate (II); Fe(II)(CN)6----

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl oth/un 25°C 0.10M C TIH K1=1.37 1986CDc (3584) 47

B(Li2Fe(CN)6)=1.69

B(LiHFe(CN)6)=4.38

Data for 10-35 C and 0.05-1.0 M LiCl. DH(K1)=23.4 kJ mol⁻¹, DS(K1)=117

J K-1 mol⁻¹; DH(Li2Fe(CN)6)=9.62, DS=88; DH(LiHFe(CN)6)=20.1, DS=176.

Li+ EMF oth/un 25°C U K1=1.95 1969NSa (3585) 48

Assuming K(Li+Fe(CN)6)=1.3

Li+ oth oth/un 25°C 0.0 U K1=1.78 1966NSa (3586) 49

Medium: 0 corr. Method: electrical migration or transference number

Cl- HL Chloride CAS 7647-01-0 (50)

Chloride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C K1=1.660 2002DDa (5163) 50

Medium: N,N-dimethylacetamide, 0.005-0.015 M LiCl.

Li+ con non-aq 25°C 100% U I K1=1.5 1982GCb (5164) 51

Medium: DMF

Li+ con diox/w 25°C ? U K1=-0.092 1975MFa (5165) 52

Data for dioxan/H2O solution with a dielectric constant of 78.35

Further data available for solutions with varying dielectric constants

Li+ ISE non-aq 25°C 100% U K1=5.3 1974BMF (5166) 53

Medium: tributylphosphate

Li+ oth non-aq 25°C 100% U 1974GRa (5167) 54

Kd(2LiCl=Li2Cl2)=2.26

Medium: octanoic acid. Method: permittivity

Li+ con non-aq 25°C 100% U K1=1.27 1974HPb (5168) 55

Medium: hexamethylphosphotriamide, using Pitts equation. Using Fuoss-Hsia equation, K1=1.24

Li+ con non-aq 25°C 100% U K1=0.61 1972SKb (5169) 56

Medium: isopentylalcohol

Li+	con	non-aq	25°C	100%	U	K1=5.55	1971BHa	(5170)	57
Medium: acetone									
Li+	con	non-aq	25°C	100%	U	K1=2.2	1971ENa	(5171)	58
Medium: trifluoroethanol, K1=2.11 to 2.28									
Li+	con	non-aq	25°C	100%	U	K1=2.32	1971ETa	(5172)	59
Medium: propanol, also acetone-propanol mixtures. In 100% acetone: K1=5.32									
Li+	oth	non-aq	37°C	100%	U	Kd(2LiCl=Li2Cl2)=0.4	1971HMb	(5173)	60
Medium: ethanoic acid. Method: vapor phase osmometry									
Li+	con	non-aq	25°C	100%	U	K1=0.18	1971PGa	(5174)	61
Medium: N-methylformamide									
Li+	EMF	non-aq	25°C	100%	U	K1=1.70	1970SAb	(5175)	62
Medium: propene carbonate									
Li+	kin	non-aq	25°C	100%	U	K1=5.55	1969BEa	(5176)	63
Medium: acetone. By conductivity: K1=5.47									
Li+	con	alc/w	25°C	43%	U I	K1=0.40	1969DPa	(5177)	64
Medium: 43.5% w/w EtOH/H2O. K1=0.49(57.1%), 0.68(68.2%), 0.88(77.2%), 1.02(86.6%), 1.20(92.3%), 1.43(100%)									
Li+	con	non-aq	25°C	100%	U	K1=2.75	1969MBf	(5178)	65
Medium: propene carbonate									
Li+	oth	alc/w	25°C	100%	U	K1=0.5	1967MIc	(5179)	66
Method: from literature data. Medium: MeOH. K1=1.43(EtOH), 2.2(i-PrOH), 2.3(i-BuOH), 3(acetone)									
Li+	con	non-aq	25°C	100%	U	K1=5.48	1966SAa	(5180)	67
Medium: acetone									
Li+	oth	non-aq	18°C	100%	U	K1=0.7	1965DGa	(5181)	68
Method:freezing point. Medium: DMSO									
Li+	kin	non-aq	0°C	100%	U	K1=0.74	1964WHa	(5182)	69
Medium:DMF									
Li+	con	non-aq	20°C	100%	U	K1=5.3	1963MSd	(5183)	70
Medium:TBP, (BuO)3PO									
Li+	gl	diox/w	25°C	70%	U	K1=2.54	1963PGb	(5184)	71
Li+	con	non-aq	25°C	100%	U	K1=4.27	1962SHd	(5185)	72
Medium: CH3COOH. By EMF K1=3.98									

 Li+ con mixed 25°C 90% U K1=2.3 1961AMc (5186) 73
 Medium: 90% acetone/H2O

Li+ oth non-aq 16°C 100% U 1959KEb (5187) 74
 K(2LiCl=Li2Cl2)=0.44
 Method: freezing point in CH3COOH, m units.

Li+ con alc/w 25°C 100% U K1=1.75 1957GKa (5188) 75
 Medium: EtOH

Li+ EMF non-aq 25°C 100% U K1=7.08 1956BKa (5189) 76
 Medium: CH3COOH

Li+ con non-aq 25°C 100% U K1=7.13 1953SEa (5190) 77
 Medium: CH3COOH

Li+ con non-aq 30°C 100% U T K1=4.14 1952CSa (5191) 78
 Medium: cyclohexanol. K1=4.21(35 C)

 ClO3- HL Chlorate CAS 7790-93-4 (971)
 Chlorate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	diox/w	25°C	90%	U	I	K1=6.42	1966CKa (6046)	79

K1=1.42(64.5% dioxan)

 ClO4- HL Perchlorate CAS 7001-90-3 (287)
 Perchlorate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C		K1=1.349	2002DDa (6290)	80

Medium: N,N-dimethylacetamide, 0.005-0.015 M LiClO4.

Li+	con	non-aq	25°C	100%	M		K1=1.26	1999DSd (6291)	81
-----	-----	--------	------	------	---	--	---------	----------------	----

Medium: acetonitrile.

Li+	con	non-aq	25°C	100%	C	I	K1=7.34 B(Li2ClO4)=8.99	1997CHb (6292)	82
-----	-----	--------	------	------	---	---	----------------------------	----------------	----

Medium: THF. By conductivity, species M2L and L2M are equivalent.
 Also data for dimethoxyethane, ethyl acetate and THF/2-ethyl-1-hexanol.

Li+	gl	non-aq	25°C	100%	U	H	K1=5.11	1981TMb (6293)	83
-----	----	--------	------	------	---	---	---------	----------------	----

Medium: Glacial acetic acid. Alternative method: Spectrophotometry.
 DH(K1)=-52 kJ mol⁻¹

Li+	con	non-aq	25°C	100%	U		K1=7.68 B2=9.85	1979CCa (6294)	84
-----	-----	--------	------	------	---	--	-----------------	----------------	----

Medium: THF

Li+	con non-aq	25°C	100%	U	K1=1.20	1978CAa	(6295)	85
Medium: Acetonitrile								
Li+	con non-aq	25°C	100%	U	K1=5.09	1977BIb	(6296)	86
Medium: - tributylphosphate								
Li+	con non-aq	25°C	100%	U	K1=0.20	1974HPb	(6297)	87
Medium: hexamethylphosphotriamide. K1 by Pitt eqn. By Fuoss-Hsia: K1=1.13								
Li+	con non-aq	25°C	100%	U T	K1=7.68	1974JPa	(6298)	88
KT=K(LiL+Li)=2.18								
Medium: THF. At -15 C: K1=7.18, KT=2.14; -30 C: K1=6.97, KT=2.12								
Li+	oth non-aq	25°C	100%	U T H	K1=-0.51	1974PKc	(6299)	89
Medium: acetone. DH(K1)=3.68 kJ mol ⁻¹ . K1=-0.92(-90 C), -0.70(-45 C), -0.64(-25 C), -0.57(0 C), -0.39(45 C). Method: infrared spectra								
Li+	con mixed	25°C	15%	U I	K1=2.55	1974SPc	(6300)	90
Medium 15% w/w THF/H2O. K1=2.59(30%), 2.56(40%), 2.24(50%), 2.39(60%), 2.59(70%), 2.57(80%), 3.16(90%), 3.97(95%), 4.36(97%), 4.39(98%), 4.84(100%)								
Li+	con alc/w	25°C	100%	U	K1=1.14	1972DAa	(6301)	91
Medium: MeOH								
Li+	con non-aq	25°C	100%	U	K1=0.51	1972SKb	(6302)	92
Medium: isopentylalcohol								
Li+	con non-aq	25°C	100%	U	K1=3.23	1971BHa	(6303)	93
Medium: acetone								
Li+	con non-aq	25°C	100%	U	K1=0.18	1971PGa	(6304)	94
Medium: N-methylformamide								
Li+	con mixed	25°C	80%	U I	K1=1.75	1970ALa	(6305)	95
Medium: 80% w/w t-butanol/H2O. K1=2.25(85%), 3.08(90%), 3.87(95%), 4.12(97%)								
Li+	EMF mixed	25°C	0.10M	U I	K1=4.41	1970DCa	(6306)	96
Medium: dimethoxy-1,2-ethane, 0.1 M H2O. K1=4.2(H2O conc.:0.01 M)								
Li+	EMF non-aq	rt	100%	U	K1=4.5	1969BEb	(6307)	97
Medium: CF3COOH								
Li+	EMF non-aq	25°C	100%	U	K1(Li+)/K1(H+)=1.12	1968MKa	(6308)	98
Method: H electrode. Medium: pyridine								
Li+	EMF non-aq	20°C	100%	U	K1=0	1967PBa	(6309)	99
Method: H electrode. Medium: C4H8O, 0.1 M Bu4NC104								

Li+ con non-aq 25°C 100% U T K1=1.54 1966MWb (6310) 100
Medium: MeCN, also at 20 C, 30 C

Li+ con non-aq 25°C 100% U K1=1.83 1962Mwa (6311) 101
Medium:MeCN

Li+ oth non-aq 16°C 100% U K(2LiL=Li2L2)=0.89 1959KEb (6312) 102

Method: freezing point. Medium: CH3CO2H, m units

CrO4-- H2L Chromate CAS 7738-94-5 (2382)
Chromate;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth oth/un 25°C 0.0 U K1=0.7 1966MBb (6496) 103

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp oth/un 25°C 1.0M U I K1=0.34 1993MAa (6997) 104
K1 values over a range of pressures and ionic strengths

Li+ ISE NaClO4 25°C 1.0M U TI K1=-0.12 1984CTd (6998) 105

Li+ ISE NaClO4 25°C 1.00M C I K1=-0.12 1984Hca (6999) 106
Also in 1.0 M KNO3 (K1=-0.03) and 1.0 M NaNO3 (K1=-0.24).

Li+ oth oth/un 25°C ? U K1=0.25 1981ASa (7000) 107

Li+ ISE NaNO3 25°C 1.0M U K1=2.90 B2=3.67 1968SRd (7001) 108
Method: F membrane electrode

Li+ dis oth/un 25°C 0.0 U Kd=1.81 1964KYa (7002) 109

Kd: Li+F=Li(in BuOH)+F(in BuOH). Kd=3.84(Na+),3.86(K+),3.53(Cs+),3.69(NH4+)

Li+ cal oth/un 25°C 0.0 U H Kso=-2.77 1964SHb (7003) 110

DH(so)=4.5 kJ mol-1

H2O L Water CAS 7732-18-5 (6115)
Water

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth non-aq 25°C 100% U I K1=0.81 B2=1.47 1974BLa (7596) 111
Method: partial pressure. Medium: propene carbonate. In sulfolane: K1=0.65,

K2=0.5. In DMSO, K1=-0.7 (by N.M.R.)

Li+ nmr non-aq 36°C 100% U K1=0.81 B2=1.24 1971CBc (7597) 112
K3=0.28

Method:N.M.R.,Medium:Propene carbonate

Li+ sol non-aq 25°C 100% U K1=0.5 B2=0.7 1967CKa (7598) 113
Medium: MeCN

Li+ sp alc/w 25°C 100% U I K1=0.18 1953BJa (7599) 114
Medium: MeOH. Maximum value of n is 3 or 4. In EtOH K1=-0.3

I- HL Iodide CAS 10034-85-2 (20)
Iodide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C K1=1.478 2002DDa (8219) 115
Medium: N,N-dimethylacetamide, 0.005-0.015 M LiI.

Li+ oth diox/w 25°C 80% U I K1=2.23 1981ASa (8220) 116
K1=-0.21 in water

Li+ oth diox/w 25°C ? U K1=-0.21 1975MFa (8221) 117
Data for dioxan/H2O solution with a dielectric constant of 78.35
Further data available for solutions with varying dielectric constants

Li+ con non-aq 25°C 100% U K1=0.48 1974HPb (8222) 118
Medium: hexamethylphosphotriamide. Calculated using Pitts eqn. By Fuoss-Hsia
K1=1.11

Li+ con non-aq 25°C 100% U K1=4.22 1973TKb (8223) 119
Medium: liquid SO2

Li+ dis oth/un 25°C 0.0 U Kd(Li+I=Li(TBP)+I(TBP))=-0.54 1967RMe (8224) 120
In (i-amylO)2MePO: Kd=-1.03

Li+ dis oth/un 25°C 0.0 U Kd(Na+I=Na(TBP)+I(TBP))=-1.32 1967RMe (8225) 121
In (i-amylO)2MePO: Kd=-1.52

Li+ con diox/w 25°C 40% U I K1=-0.82 1966AMb (8226) 122
K1=0.40(60% dioxan), 1.38(70%), 2.28(80%), 2.82(87%), 3.85(91%), 4.50(95%)

Li+ con non-aq 25°C 100% U K1=2.16 1966SAa (8227) 123
Medium: acetone

Li+ con non-aq 25°C 100% U K1=2.97 1965BFb (8228) 124
Medium: diaminoethane

Li+ con oth/un 25°C 0.0 U K1=-0.04 1964PSh (9386) 133

NO3- HL Nitrate CAS 7697-37-2 (288)
 Nitrate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	C	I	K1=10.88 B(Li2NO3)=14.17	1997CHb (9747)	134
-----	-----	--------	------	------	---	---	-----------------------------	----------------	-----

Medium: THF. Also data for dimethoxyethane and THF/2-ethyl-1-hexanol mixtures. By conductivity, species M2L and L2M are equivalent.

Li+	sp	non-aq	25°C	100%	U		K1=-0.187	1979ITa (9748)	135
-----	----	--------	------	------	---	--	-----------	----------------	-----

Medium: N,N-Dimethylacetamide. Method: Raman spectroscopy

Li+	con	non-aq	25°C	100%	U		K1=0.72	1974HPb (9749)	136
-----	-----	--------	------	------	---	--	---------	----------------	-----

Medium: Hexamethylphosphotriamide. Using Pitts equation. Using Fuoss-Hsia eq K=1.18

Li+	con	non-aq	25°C	100%	U		K1=3.35	1974PHb (9750)	137
-----	-----	--------	------	------	---	--	---------	----------------	-----

Medium: MeCN

Li+	con	non-aq	25°C	100%	U		K1=9.77 K(LiL+Li)=2.3	1973WHa (9751)	138
-----	-----	--------	------	------	---	--	--------------------------	----------------	-----

Medium: THF

Li+	con	non-aq	25°C	100%	U		K1=0.58	1972SKb (9752)	139
-----	-----	--------	------	------	---	--	---------	----------------	-----

Medium: Isopentyl alcohol

Li+	con	non-aq	25°C	100%	U		K1=2.22	1971BCa (9753)	140
-----	-----	--------	------	------	---	--	---------	----------------	-----

Medium: Tetramethylurea

Li+	sp	oth/un		var	U		K1=-0.6	1971INb (9754)	141
-----	----	--------	--	-----	---	--	---------	----------------	-----

Method: Raman spectra

Li+	con	diox/w	25°C	72%	U	I	K1=1.59	1969SBe (9755)	142
-----	-----	--------	------	-----	---	---	---------	----------------	-----

In 74.6% dioxan: K1=2.12, 75.9%: 2.43, 77.8%: 2.76

Li+	EMF	non-aq	25°C	100%	U		K1(Li+)/K1(H+)=0.91	1968MKa (9756)	143
-----	-----	--------	------	------	---	--	---------------------	----------------	-----

Method: H electrode. Medium: C5H5N

Li+	ix	mixed	23°C	90%	U		K1=-0.54	1966WFa (9757)	144
-----	----	-------	------	-----	---	--	----------	----------------	-----

Medium: 90% i-PrOH, 0.5 M HL

Li+	con	alc/w	25°C	100%	U		K1=1.28	1963PSa (9758)	145
-----	-----	-------	------	------	---	--	---------	----------------	-----

Medium: EtOH, I=0 corr.

Li+	con	non-aq	25°C	100%	U		K1=3.39	1961KBa (9759)	146
-----	-----	--------	------	------	---	--	---------	----------------	-----

Medium: MeCN

Method: Freezing point. Medium: CH₃CO₂H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

method: NMR Li-7

Medium: THF, 0.1 M Bu₄NClO₄. H electrode

At $I=0$ corr.: $K1=-0.1$

*K1=-14.4

K1=1.13(93 C), 1.51(138 C), 1.42(182 C), 1.59(227 C), 1.76(271 C)

K1=0.26(5 C), 0.20(15 C), 0.19(35 C), 0.19(45 C). Method: H electrode

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

$$K(Li+HL)=0.79$$
$$K(Li+H_2L)=0.2$$
$$K(Li+HL)=0.72$$

Hypophosphate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.50M	U		K1=0.82	1967CMc (13415)	158

Ligand: O3POPHO2---, Medium: Me4NC1

P207---- H4L Pyrophosphate CAS 2466-09-3 (198)
Diphosphate; from (HO)2PO.O.PO(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	R4N.X	5°C	1.00M	U	H		1973VAa (13615)	159

Medium: Me4NN03, DH(K1)=1.3 kJ mol⁻¹. 35 C, I=0, DH(K1)=4.2. DH(Li+HL)=-0.8

Li+	gl	none	25°C	0.0	U	T	K1=3.1	1959W0a (13616)	160
-----	----	------	------	-----	---	---	--------	-----------------	-----

K1=3.3(40 C)

Li+	gl	R4N.X	25°C	1.00M	U		K1=2.39 K(Li+HL)=1.03	1957LWa (13617)	161
-----	----	-------	------	-------	---	--	--------------------------	-----------------	-----

Medium: Me4NC1

P208---- H4L CAS 13825-81-5 (2402)
Peroxodiphosphate, also cyclic metaposphates, thiophosphates etc.;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	kin	NaNO3	65°C	1.0M	C			1985GGb (13692)	162

K(Li+HP208)=0.92

Ligand is peroxydisulfate, S208----

Li+	gl	R4N.X	25°C	1.00M	U		K1=1.34 K(Li+HL)=0.70	1960CEa (13693)	163
-----	----	-------	------	-------	---	--	--------------------------	-----------------	-----

Medium: Me4NC1

P2W17061----- Polytungstate (2102)
alpha-Heterodiphospho-polytungstate (usually alpha1 isomer)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	1.0M	U		K1=3.61 K(Li+HL)=3.77 K(Li+H2L)=2.0	1982CCb (13724)	164

alpha2 isomer. For alpha1 isomer, K1>3.7, K(Li+HL)=1.6, K(Li+H2L)=0.6

P3010----- H5L CAS 10380-08-2 (1001)
Tripolyphosphate; from (HO)2PO.O.PO(OH).O.PO(OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	none	25°C	0.0	U	T	K1=3.9	1959W0a (13878)	165

K1=3.8(40 C)

 Li+ gl R4N.X 25°C 1.00M U K1=2.87 1957WLa (13879) 166
 K(Li+HL)=0.88

Medium: Me4NCl

P4013----- H6L Tetraphosphate (1102)

Tetraphosphate;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 1.0M U K1=2.64 1967WMa (14049) 167
 K(Li+HL)=1.59

Medium: Me4NCl

P6012----- H6L CAS 25268-83-1 (6590)

Dodecaoxohexaphosphate(III); anion of (PO.OH)₆

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 1.0M U K1=1.34 1960CEa (14061) 168
 K(Li+HL)=0.70

Medium: Me4NCl

SCN- HL Thiocyanate CAS 463-56-9 (106)

Thiocyanate;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% U K1=0.93 1994GGa (15130) 169

Medium: DMF

 Li+ sp non-aq 20°C 100% U I K1=3.87 1989GGa (15131) 170
 K1out=3.05

Medium: MeCN, by IR spectroscopy. Also data for MeCN containing R4NX salts

 Li+ cal NaClO₄ 25°C 0.50M U H K1=1.00 B2=1.07 1988ISb (15132) 171
 B4=3.12

Medium: LiClO₄ + 10%w/w Triton X-100. DH(K1)=-11.0 kJ mol⁻¹, DH(B2)=-31,
 DH(B4)=-32.6. DS(K1)=-18 J K⁻¹ mol⁻¹, DS(B2)=-84, DS(B4)=-50.

 Li+ sp non-aq 25°C 100% U K1=-0.125 B2=0.26 1979ITa (15133) 172

Medium: N,N-Dimethylacetamide. Method: Raman spectroscopy

 Li+ con non-aq 25°C 100% U K1=0.15 1971PGa (15134) 173

Medium: MeHNCHO

 Li+ sp non-aq 20°C 100% U K1=2.9 1970SSa (15135) 174

Medium: MeCN

S04-- H2L Sulfate CAS 7664-93-9 (15)

Sulfate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	NaCl	37°C	0.10M	C	I	K1=0.77	1982DRb (16304)	175
Data for I=0.03-0.50 M NaCl. At I=0.0 M, K1=1.12									
Li+	oth	oth/un	25°C	0.50M	U	TI	K1=0.77	1980GAb (16305)	176
Method: Ultrasonic absorption. Medium: Na2SO4									
Li+	con	none	25°C	0.0	U			1978FFa (16306)	177
							K(Li+LiSO4)=0.096		
Li+	oth	oth/un	25°C	.244M	U		K1=0.77	1975REa (16307)	178
Li+	oth	none	25°C	0.0	M		K1=0.7	1966MBb (16308)	179
Li+	con	oth/un	18°C	0.0	U		K1=0.64	1930RDa (16309)	180

SeCN- HL Selenocyanate CAS 73102-11-2 (440)
Selenocyanate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	20°C	100%	U		K1=2.4	1970SSa (16991)	181
Medium: acetonitrile									

SiW11039-----		H8L		(2464)					
alpha-Heterosilicon-polytungstate;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	1.0M	U		K1=4.1	1982CCb (17237)	182

VO4---		H3L		CAS 15457-75-7 (1586)					
Vanadate; VO2(OH)3-- or polymers									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	NaClO4	25°C	1.00M	U			1975KIc (17381)	183
							K(Li+H7PV12036)=1.92		
Li+	gl	R4N.X	20°C	0.10M	U			1963SGd (17382)	184
							K(Li+H15L10)=0.6		
							K(Li+H14L10)=1.6		
							K(Li+LiH14L10)=0.6		

CH2O2 HL Formic acid CAS 64-18-6 (37)
Methanoic acid; H.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	30°C	100%	U		K1=7.06	1954JGa (17622)	185
Medium: ethanoic acid									

CH3NO		L		Formamide			CAS 75-12-7	(3536)	
Methanoic acid amide; HCO.NH2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	C		K1=1.3 B2=2.0	1975NAa (17678)	186
Medium: CH3CN, 0.01 M Et4NClO4									

CH4O		L		Methyl alcohol			CAS 67-56-1	(597)	
Methanol; CH3.OH									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	C		K1=0.65 B2=0.83	1975NAa (17884)	187
Medium: CH3CN, 0.01 M Et4NClO4									

CH5O4P		H2L					CAS 2617-47-2	(1977)	
Hydroxymethylphosphonic acid; HO.CH2.PO3H2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	U		K1=0.72	1972WFa (18149)	188
Medium: (CH3)4NCl									

CH6O6P2		H4L		Medronic acid			CAS 1984-15-2	(2384)	
Methanediphosphonic acid; CH2(PO3H2)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.50M	U		K1=2.48 K(Li+HL)=0.82	1967CIa (18285)	189
Medium: Me4NCl									

C2H2O4		H2L		Oxalic acid			CAS 144-62-7	(24)	
Ethanedioic acid; (COOH)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	oth/un	37°C	0.15M	C	I	K1=0.79	1983DRb (18946)	190
Medium: 0.15 M LiNO3. Method: determination of protonation constant in LiNO3 and [Et4N]NO3 media. Data for I=0.0-1.0 M LiNO3. At I=0.0, K1=1.17.									

C2H4O2		HL		Acetic acid			CAS 64-19-7	(36)	
Ethanoic acid; CH3.COOH									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X I=0.02-1 M Et4NI	25°C	0.25M	C	TI	K1=0.13	1985DRa (20031)	191
Li+	gl	R4N.X	25°C	0.16M	U	I	K1=-0.13 K1=-0.10 (I=0.04); -0.13 (0.25); -0.09 (0.49); -0.02 (1.0)	1985RSa (20032)	192
Li+	con	alc/w K1=1.18 in water. In 50% dioxan/H2O:	25°C	100%	U	I	K1=1.73 K1=1.65; 60%: 2.04; 70%: 2.50	1981ASa (20033)	193
Li+	gl	non-aq Medium: Glacial acetic acid. Alternative method: Spectrophotometry. DH(K1)=-29.0 kJ mol ⁻¹	25°C	100%	U	H	K1=6.87	1981TMb (20034)	194
Li+	con	none Also data for MeOH (K1=1.834), 50% dioxan/H2O (K1=1.788) and 70% dioxane/H2O (K1=2.874).	35°C	0.0	C	I	K1=1.295	1979ASc (20035)	195
Li+	gl	oth/un	25°C	0.0	U		K1=0.26	1964AMa (20036)	196
Li+	gl	non-aq Medium: ethanoic acid	25°C	100%	U		K1=6.78	1964KLa (20037)	197
Li+	con	oth/un Medium: Li ethanoate. K1=-0.54(I=0.2), -0.52(I=0.5), -0.49(I=0.7)	18°C	0.10M	U	I	K1=-0.53	1964SUb (20038)	198
Li+	sp	non-aq Medium: ethanoic acid	25°C	100%	U		K1=6.20	1961PSa (20039)	199
Li+	EMF	non-aq Method: chloranil electrode. Medium: ethanoic acid	25°C	100%	U		K1=6.79	1956BKa (20040)	200
Li+	con	non-aq Medium: ethanoic acid	30°C	100%	U		K1=6.82	1954JGa (20041)	201

C2H5NO L Methylformamide CAS 123-39-7 (6268)
N-Methylformamide; HCO.NH.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq Medium: CH3CN, 0.01 M Et4NClO4	25°C	100%	C		K1=1.4 B2=2.4	1975NAa (20675)	202
C2H5NO2	HL	Glycine 2-Aminoethanoic acid; H2N.CH2.COOH					CAS 56-40-6 (85)		

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	oth/un	25°C	1.0M	U		K1=1.2	1987HAa (21603)	203

C2H6OS L DMSO CAS 67-68-5 (329)
Dimethylsulfoxide; (CH₃)₂S

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	M		K1=1.69 B2=2.99	1988NHa (22105)	204
Medium: MeCN, 0.01 M Et ₄ NClO ₄									
Li+	ISE	non-aq	25°C	100%	U T H		K1=1.70	1982NYa (22106)	205
Medium: MeCN									
Li+	ISE	non-aq	25°C	100%	C		K1=1.7 B2=3.0 B3=3.5	1975NAa (22107)	206

Medium: CH₃CN, 0.01 M Et₄NClO₄

C2H8N₂ L Ethylenediamine CAS 107-15-7 (23)
1,2-Diaminoethane; H₂N.CH₂.CH₂.NH₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	95%	U		K1=0.87	1993GSa (23186)	207
Medium: 95% w/w EtOH/H ₂ O, 0.05 M Et ₄ NClO ₄ , by competitive spectrophotometry									
Li+	gl	oth/un	25°C	0.10M	C I		K1=-0.20 K(Li+HL)=-0.65	1990CDb (23187)	208

Medium: 0.10 M LiCl. Data for I=0.25-1.0 M.

C2H8O₆P₂ H4L CAS 6145-33-1 (3543)
Ethane-1,1-diphosphonic acid; CH₃.CH(P₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.50M	U		K1=3.1 K(Li+HL)=0.99	1967CIa (23270)	209

Medium: Me₄NCl

C2H8O₇P₂ H4L HEDPA CAS 2809-21-4 (436)
1-Hydroxyethane-1,1-diphosphonic acid; CH₃.C(OH)(P₃H₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	U		K(Li+HL)=1.36 B(2Li+L)=4.78	1972WFa (23381)	210

Medium: (CH₃)₄NCl

Li+	gl	R4N.X	25°C	0.50M	U		K1=3.35 K(Li+HL)=1.08	1967CIa (23382)	211
-----	----	-------	------	-------	---	--	--------------------------	-----------------	-----

Medium: Me₄NCl

C3H4O4 H2L Malonic acid CAS 141-82-2 (79)
Propanedioic acid; CH₂(COOH)₂

Li+ gl R4N.X 37°C 0.25M C TI K1=0.95 1985DRa (24488) 212
B(LiHL)=5.63

Medium: 0.15 M LiNO₃. Method: determination of protonation constant in LiNO₃ and [Et₄N]NO₃ media. Data for I=0.0-1.0 M LiNO₃. At I=0.0, K₁=1.05.

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)
L-2-Hydroxypropanoic acid; CH₃.CH(OH).COOH

Li+ EMF oth/un 25°C ->0 U K1=0.20 1954DMb (25475) 214
Method: H electrode

C3H7NO L DMF CAS 68-12-2 (598)
N,N-Dimethylformamide; HCO.N(CH3)2

Li+ ISE non-aq 25°C 100% M K1=1.14 B2= 2.20 1999NMa (25659) 215
B3=1.66
B4=1.75

Medium: acetonitrile, 0.01 M Et₄NClO₄.

Medium: MeCN, 0.01 M Et₄NClO₄

Medium: MeCN

B3=1.8

Propane-2,2-diphosphonic acid; $\text{CH}_3.\text{C}(\text{P}(\text{O})(\text{OH})_2)_2.\text{CH}_3$

K(Li+HL)=1.38

Medium: Me4NCl

C4H4O4 H2L Maleic acid CAS 110-16-7 (111)
cis-Butenedioic acid; HOOC.CH:CH.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	37°C	0.25M	C	TI	K1=0.91 B(LiHL)=5.9	1985DRa (29099)	220

I=0.02-1 M Et4NI

Li+ gl oth/un 37°C 0.15M C I K1=0.72 1983DRb (29100) 221

Medium: 0.15 M LiNO3. Method: determination of protonation constant in LiNO3 and [Et4N]NO3 media. Data for I=0.0-1.0 M LiNO3. At I=0.0, K1=1.08.

C4H6O3 L CAS 108-32-7 (6267)

Propylene carbonate, 1,2-Propanediol cyclic carbonate, 4-Methyl-1,3-dioxolan-2-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	C		K1=0.5	1975NAa (29752)	222

Medium: CH3CN, 0.01 M Et4NClO4

C4H6O4 H2L Succinic acid CAS 110-15-6 (112)

1,4-Butanedioic acid; HOOC.CH2.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	37°C	0.25M	C	TI	K1=0.70 B(LiHL)=5.38	1985DRa (29991)	223

I=0.02-1 M Et4NI

Li+ gl oth/un 37°C 0.15M C I K1=0.42 B2= 0.60 1983DRb (29992) 224

Medium: 0.15 M LiNO3. Method: determination of protonation constant in LiNO3 and [Et4N]NO3 media. Data for I=0.0-1.0 M LiNO3. At I=0.0, K1=0.84.

C4H6O5 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	oth/un	25°C	0.10M	U		K1=0.38	1964RZa (30669)	225

Li+	gl	R4N.X	?	0.28M	U		K1=0.45	1963EDa (30670)	226
-----	----	-------	---	-------	---	--	---------	-----------------	-----

Medium: Me4NBr

C4H6O5 H2L Diglycolic acid CAS 110-99-6 (243)

Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOC.CH2.O.CH2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

 Li+ gl R4N.X 25°C 0.25M C TIH K1=0.70 1985DRa (30892) 227
 B(LiHL)=3.91
 0.02-1 M NEt4I. 12.5-48 C. DH(K1)=4 kJ mol⁻¹, DS=35; DH(LiHL)=10, DS=119

Li+ oth oth/un 30°C 1.00M U K1=-0.62 19730Ea (30893) 228
 Method: Raman spectroscopy. medium: LiCl

 C4H8O L THF CAS 109-99-9 (2537)
 Tetrahydrofuran; cyclo(-CH2.CH2.O.CH2.CH2-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ sp diox/w 25°C 100% U M 1990TPa (33188) 229
 K(Li(picrate)+L)=-0.02
 With 2-methyltetrahydrofuran K=-0.47; 2,5-dimethyl- K=-0.09; tetrahydropyran
 K=-0.54; dioxalane K=-0.47; hexamethyleneoxide -0.55

C4H9NO L CAS 127-19-5 (477)
 N,N-Dimethylacetamide; CH3.CO.N(CH3)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ ISE non-aq 25°C 100% U T H K1=1.76 1982NYa (33762) 230
 Medium: MeCN

Li+ ISE non-aq 25°C 100% U K1=1.39 B2=2.27 1976CWc (33763) 231
 B3=2.34
 B4=2.3

Medium: propylene carbonate

Li+ ISE non-aq 25°C 100% C K1=1.8 B2=2.9 1975NAa (33764) 232
 B3=3.3

Medium: CH3CN, 0.01 M Et4NClO4

C4H10O HL t-Butanol CAS 75-65-0 (1740)
 tert-Butanol, (CH3)3C.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C 100% U K1=8.0 1974ESa (34659) 233
 Medium: DMSO

C4H10O3 L CAS 111-46-6 (3579)
 2,2'-Oxydiethanol; (HO.CH2.CH2)2.O (Diethylene glycol)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C 100% C K1=2.7 1992MSe (34702) 234
 Medium: 100% MeOH. Anion: picrate. Also data for nitrophenolate anions.

C₄H₁₁N L (6678)
Dimethylethylamine; (CH₃)₂NCH₂CH₃

Li+ kin non-aq 20°C 100% U M 1993BCd (34823) 235
K=0.740

A:Di(iso-propyl)amine. B:N,N,N',N'-Tetramethylethylenediamine.

C4H11NO3 L Tris buffer CAS 77-86-1 (550)
2-Amino-2-(hydroxymethyl)-propan-1,3-diol; (HO.CH2)3C.NH2

Li+ gl R4N.X 25°C 1.00M C I K1=-0.23 1982SSf (35058) 236
In 90 % (v/v) DMSO/water mixture: K1=0.37

C5H5N	L	Pyridine	CAS 110-86-1	(31)
Pyridine, Azine;				

Li+ ISE non-aq 25°C 100% C K1=0.72 B2=0.43 1975NAa (36648) 237
Medium: CH3CN, 0.01 M Et4NClO4

C5H6N2 L 2-Aminopyridine CAS 504-29-0 (1478)
2-Aminoazine, 2-Pyridylamine; C5H4N.NH2

Li+ sp alc/w 25°C 95% U K1=0.76 1993GSa (37128) 238
Medium: 95% w/w EtOH/H2O, 0.05 M Et4NC104, by competitive spectrophotometry

C5H8O2	HL	Acetylacetone	CAS 123-54-6	(164)
Pentane-2,4-dione; CH ₃ .CO.CH ₂ .CO.CH ₃				

Li+ sp non-aq 25°C 100% U K1=4.76 1984AMa (38012) 239
In Dimethyl Sulfoxide (DMSO);

Li+ gl diox/w 30°C 75% U K1=4.75 B2=8.72 1975MMa (38013) 240

Li+ gl alc/w 25°C 100% U K1=2.8 1965LIa (38014) 241
Medium: MeOH, 0.1 M LiClO4. In EtOH: K1=4.6

C6H3N3O7	HL	Picric acid	CAS 88-89-1	(593)
----------	----	-------------	-------------	-------

2,4,6-Trinitrophenol; HO.C6H2(NO2)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	25°C	100%	C		K1=3.04	1999KKb (42122)	242
Medium: MIBK. Method: distribution of metal picrates into MIBK containing HO(CH2.CH2.O)n.C12H25, n=4, 6 or 8.									
Li+	dis	oth/un	25°C	dil	C		K(LiA+L)=2.53	1998TKa (42123)	243
Self medium, I<0.03 M. Method: Extraction of LiAl into dichloromethane. A is 18-crown-6.									
Li+	con	non-aq	25°C	100%	C I		K1=8.07 B(Li2L)=10.03	1997CHb (42124)	244
Medium: THF. By conductivity, species M2L and L2M are equivalent. Also data for dimethoxyethane and ethyl acetate.									
Li+	con	non-aq	25°C	100%	C I		K1=2.99	1996HHc (42125)	245
Medium: acetonitrile. Also data for benzonitrile and DMF.									
Li+	sp	non-aq	25°C	100%	U		K1=3.97	1980GRa (42126)	246
Medium: 2-butanol									
Li+	con	alc/w	30°C	100%	U I M		K1=2.72	1979PSa (42127)	247
Medium: isoPrOH. K(LiL+diethyleneglycol)=2.51; K(LiL+trien-glycol)=2.41. In H2O: K1=1.11									
Li+	dis	none	25°C	0.00	U		K1=1.13	1972IWc (42128)	248
Li+	dis	oth/un	25°C	var	U		K1=2.2	1970SSb (42129)	249
Method: paper chromatography									

C6H4N2O5 HL CAS 50-28-5 (505)									
2,4-Dinitrophenol; HO.C6H3(NO2)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C I		K1=4.29 B(Li2L)=6.71	1996HHc (42232)	250
Medium: acetonitrile. By conductivity, species M2L and L2M are equivalent. Also data for benzonitrile and DMF.									
Li+	con	non-aq	25°C	100%	U		K1=3.54	1973FGa (42233)	251
Medium: tetrahydrofuran									

C6H4N2O5 HL CAS 329-71-5 (507)									
2,5-Dinitrophenol; HO.C6H3(NO2)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C 100% C I K1=5.33 1996HHc (42244) 252
B(Li2L)=8.48
K(2LiL=Li2L2)=1.72

Medium: acetonitrile. By conductivity, species M2L and L2M are equivalent.
Also data for DMF.

C6H5NO3 HL 2-Nitrophenol CAS 88-75-5 (510)
2-Nitrohydroxybenzene; HO.C6H4.NO2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C K1=6.51 1996HHc (42737) 253
B(Li2L)=10.60
K(2LiL=Li2L2)=3.01

Medium: acetonitrile. By conductivity, species M2L and L2M are equivalent.

C6H5NO3 HL 4-Nitrophenol CAS 100-02-7 (454)
4-Nitrohydroxybenzene; HO.C6H4.NO2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C K1=4.96 1996HHc (42810) 254
B(Li2L)=8.30
K(2LiL=Li2L2)=3.18

Medium: acetonitrile. By conductivity, species M2L and L2M are equivalent.

Li+ con non-aq 25°C 100% U K1=6.99 1991AMa (42811) 255

Medium: THF

C6H5OCl HL 4-Chlorophenol CAS 106-48-9 (1631)
4-Chlorophenol; HO.C6H4.Cl

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=9.13 1991AMa (43054) 256

Medium: THF

C6H6O HL Phenol CAS 108-95-2 (457)
Hydroxybenzene, phenol; C6H5.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=10.16 1991AMa (43542) 257

Medium: THF

C6H6O2 H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

```

-----
Li+      sp  alc/w  25°C  95%  U      K1=1.31      1993GSa (43783) 258
Medium: 95% w/w EtOH/H2O, 0.05 M Et4NClO4, by competitive spectrophotometry
*****
C6H8N2          L          CAS 95-54-5 (2899)
1,2-Diaminobenzene, 1,2-Phenylenediamine; C6H4(NH2)2
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Li+      sp  alc/w  25°C  95%  U      K1=1.12      1993GSa (45271) 259
Medium: 95% w/w EtOH/H2O, 0.05 M Et4NClO4, by competitive spectrophotometry
*****
C6H8O6          H3L      Tricarballic CAS 99-14-9 (1620)
1,2,3-Propanetricarboxylic acid; HOOC.CH2.CH(COOH).CH2.COOH
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Li+      gl  oth/un 25°C  0.0  C  I      K1=1.473      1994DFc (45567) 260
                                B(LiHL)=7.322
                                B(LiH2L)=11.504
                                B(Li2L)=2.083
                                B(Li2HL)=6.923

```

```

Values at I=0 calculated from data for 0.04-1.0 M LiCl.
*****
C6H8O7          H3L      Citric acid  CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Li+      gl  oth/un 25°C  0.50M U  H      K1=1.10      1990DRa (46158) 261
                                B(LiHL)=6.49
                                B(Li2L)=1.66
DH(K1)=-3.0, DH(LiHL)=-3.6 and DH(Li2L)=-5.0 kJ mol-1.
-----

```

```

Li+      gl  KCl      37°C  0.15M C          K1=0.88  B2=1.13  1981CDb (46159) 262
-----

```

```

Li+      ISE oth/un 25°C  0.10M U      K1=0.83      1964RZa (46160) 263
*****
C6H9NO6          H3L      NTA          CAS 139-13-9 (191)
Nitrilotriethanoic acid; N(CH2.COOH)3
-----

```

```

Metal      Mtd Medium Temp Conc Cal Flags Lg K values      Reference ExptNo
-----
Li+      gl  oth/un 25°C  0.10M C TIH    K1=2.56      1985DRb (46894) 264
                                B(LiHL)=9.62
Data at 10-45 C and I=0.02-1.0 M in LiNO3. DH(K1)=8 kJ mol-1; DS=(K1)=74.
DH(LiHL)=15; DS(LiHL)=232 (by T coeff.)
-----

```

```

Li+      sp  R4N.X  25°C  0.10M C          K1=2.35      1985HAd (46895) 265
-----

```

Li+ gl R4N.X 20°C 0.10M U T K1=2.51 1963IFb (46896) 266
Medium: Me4HNO3

Li+ EMF oth/un 20°C 0.0 U K1=3.28 1945SKb (46897) 267
Method: H electrode

C6H10O2 HL CAS 815-57-6 (2261)
3-Methyl-pent-2,4-dione; CH3.CO.CH(CH3).CO.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr alc/w -60°C 100% U K1=1.1 1979RHa (47948) 268
Medium: CD3OD

C6H14O3 L Diglyme CAS 111-96-6 (6769)
bis-2-Methoxyethyl ether, 2,5,8-Trioxanonane; CH3.O.CH2CH2.O.CH2CH2.O.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C K1=3.2 1992MSe (51051) 269
Medium: 100% MeOH. Anion: picrate. Also data for nitrophenolate anions.

C6H15NO3 Triethanolamine CAS 102-71-6 (447)
Tris-(2-hydroxyethyl)amine; L

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 1.00M C I K1=-0.48 1982SSf (51297) 270
In 90 % (v/v) DMSO/water mixture: K1=0.49 (I=0.25 M)

Li+ con non-aq 25°C 100% U M K1=4.70 1976FGb (51298) 271
K(LiA+L)=2.34

A=Tetra-n-butylammonium-2,4-dinitrophenolate. Medium: Tetrahydrofuran

C6H15O15P3 H6L Ins(1,2,6)P3 CAS 28841-62-5 (6479)
D-myo-Inositol 1,2,6-trisphosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M U K1=2.38 1991BSa (51537) 272
B(LiHL)=11.24
B(LiH2L)=17.83
B(Li2L)=3.60

C6H16N2 L Tetrameen CAS 110-18-9 (124)
N,N,N',N'-Tetramethyl-1,2-diaminoethane; (CH3)2N.CH2.CH2.N(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ kin non-aq 0°C 100% U M 1993BCd (51648) 273

$$K(\text{Li}_2\text{A}_2\text{B}_2+2\text{L}=\text{Li}_2\text{B}_2\text{L}_2+2\text{A})=0.204$$

Metal:Li(0). Medium:Hexane. A:di(iso-propyl)amine. B:Tetrahydrofuran.

C6H16O3P2 L (2075)
Di(dimethylphosphinylmethyl) ether; Me2P(O)CH2.O.CH2.P(O)Me2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=4.04	1989KSa (51773)	274
Medium: tetrahydrofuran/CHCl3 4:1 (vol)									

Li+	sp	non-aq	25°C	100%	U		K1=1.83	1983YSb (51774)	275
Medium: tetrahydrofurane + CHCl3 (4:1); Li as 2,4-dinitrophenolate. In (CH3CN+CHCl3 1:1) K1=2.69. Data also for other phosphine oxides									

Li+	con	non-aq	25°C	100%	U		K1=4.04	1982YSa (51775)	276
Medium: tetrahydrofuran+CHCl3 4:1(vol); M is 2,4-dinitrophenolate									

C6H18N3OP L HMPA CAS 680-31-9 (603)
Hexamethylphosphoramide, Tris-(dimethylamino)phosphine oxide;((CH3)2N)3PO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	M		K1=3.32 B2=5.51	1988NHa (51981)	277
Medium: MeCN, 0.01 M Et4NClO4									

Li+	con	non-aq	25°C	100%	U	M	Kout(LiL+A)=5.8	1982GJb (51982)	278
Medium: 1,2-dichloroethane. A=tetraphenylborate									

Li+	ISE	non-aq	25°C	100%	U T H		K1=3.31	1982NYa (51983)	279
Medium: MeCN									

Li+	ISE	non-aq	25°C	100%	C		K1=3.3 B2=5.5	1975NAa (51984)	280
Medium: CH3CN, 0.01 M Et4NClO4									

C7H5NO HL CAS 767-00-0 (1632)
4-Cyanophenol; HO.C6H4.CN

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=8.04	1991AMa (52582)	281
Medium: THF									

C7H6O4 H3L CAS 303-38-8 (1398)
2,3-Dihydroxybenzoic acid; C6H3(OH)2.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	NaClO4	25°C	0	C I		K1=1.67	1992CRa (54468)	282

K(Li+LiL)=0.77

K(Li+HL)=0.75

K(Li+H2L)=-0.7

Extrapolated to I=0 from I=0.04 to I=0.81

C7H8O2 HL CAS 150-76-5 (6738)

4-Methoxyphenol; CH3O.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=11.07 1991AMa (56095) 283

Medium: THF. With 4-t-butylphenol K=10.87, 2-t-butylphenol K=10.13

C7H8O8P2 H4L (6892)

1,2-((Phenylenedioxy)methylene)diphosphonic acid); C6H4O2C(P03H2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.50M U K1=2.42 1985GMb (56169) 284

Medium: 0.5 M Me4NCl

C8H5N5O6 H3L Murexide (453)

Purpuric acid (Murexide is ammonium salt);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% U TIH K1=5.34 1995KSa (58518) 285

Medium: 10% w/w DMF/MeCN. DH(K1)=-5.0 kJ mol⁻¹, DS=86 J K⁻¹ mol⁻¹

Data also for 20 30, 40 w/w% DMF

Li+ sp alc/w 25°C 95% U K1=2.44 1993GSa (58519) 286

Medium: 95% w/w EtOH/H2O, 0.05 M Et4NClO4

Li+ sp non-aq 20°C 100% U K1=2.45 1992PSa (58520) 287

Medium: DMF, 0.01 M Me4NI

C8H5O2F3S HL TTA CAS 326-91-0 (165)

4,4,4-Trifluoro-1-(2-thienyl)butane-1,3-dione; F3C.CO.CH2.CO.C4H3S

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% C M 2002IIa (58639) 288

K(LiL+phen)=6.74

Medium: chlorobenzene. For extraction from 0.10 M KCl:

K(Li+HL(o)=LiL(o)+H)=-10.34; K(Li+HL(o)+phen(o)=LiL(phen)(o)+H)=-3.60.

Li+ gl alc/w 25°C 0.10M U I K1=3.2 1965LIa (58640) 289

Medium: MeOH, 0.1 M LiClO4. In EtOH: K1=5.3

C8H6O4 H2L Phthalic acid CAS 88-99-3 (113)

Benzene-1,2-dicarboxylic acid; C₆H₄(COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	37°C	0.25M	C	TI	K1=0.85 B(LiHL)=5.03	1985DRa (58985)	290

Medium: 0.02-1 M NEt₄I

Li+	gl	oth/un	37°C	0.15M	C	I	K1=0.65 B2= 0.65	1983DRb (58986)	291
-----	----	--------	------	-------	---	---	---------------------	-----------------	-----

Medium: 0.15 M LiNO₃. Method: determination of protonation constant in LiNO₃ and [Et₄N]NO₃ media. Data for I=0.0-1.0 M LiNO₃. At I=0.0, K₁=1.00.

C₈H₈O₃ HL Phenoxyacetic CAS 122-59-8 (1153)
Phenoxyethanoic acid; C₆H₅.O.CH₂.COOH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	none	25°C	0.0	C	TIH	K1=0.07	1985CDb (60038)	292

Calculated from protonation data for I=0.04-0.9 M LiCl. Data for 10-45 C.
DH(K₁)=7.7 kJ mol⁻¹, DS(K₁)=28 J K⁻¹ mol⁻¹.

C₈H₉N₃O₇ H₂L Uramildiacetic CAS 13055-06-5 (185)
5-Amino-2,4,6-trioxo-1,3-perhydrodiazimino-N,N-diethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	R4N.X	20°C	0.1M	C			1976ANb (60637)	293

DH₁= -7.61 kJ/mol

in Me₄NCI

Li+	gl	R4N.X	39°C	0.10M	U	TIH	K1=4.60	1963IFb (60638)	294
-----	----	-------	------	-------	---	-----	---------	-----------------	-----

Medium: Me₄NNO₃. K₁=4.90(20 C), 4.70(27 C), 4.57(34 C); DH(K₁)=-29.3 kJ mol⁻¹
DS=-4 J K⁻¹ mol⁻¹. At I=0 corr:K₁=5.61(20 C)

Li+	ISE	oth/un	20°C	0.0	U		K1=5.40	1948SBa (60639)	295
-----	-----	--------	------	-----	---	--	---------	-----------------	-----

C₈H₁₀O HL CAS 576-26-1 (1498)
2,6-Dimethylphenol; HO.C₆H₃(CH₃)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	22°C	100%	U	T H		1991JRa (60826)	296

Medium: dioxolane. 90-22 C. DH(Li₂L+2LiClO₄=2Li₂LCI₄)=0 kJ mol⁻¹, DS=58
In THF: DH=0, DS=28. In dioxalane DH(LiL₂+2LiBPh₄=2Li₂LBPh₄)=-5, DS=-5

C₈H₁₁O₂F₃ HL CAS 22767-90-4 (1249)
1,1,1-Trifluoro-5,5-dimethyl-2,4-hexanedione; F₃C.CO.CH₂.CO.CH(CH₃)₃

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ oth diox/w 25°C 75% U K1=4.61 B2=8.05 1979MMa (61304) 297

C8H12O2 HL Dimedone CAS 126-81-8 (1137)
 5,5-Dimethyl-1,3-cyclohexanedione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ gl alc/w 25°C 100% U I K1=1.5 1965LIa (61687) 298
 Medium: MeOH, 0.1 M LiClO4. In EtOH: K1=2.1

C8H16O4 L 12-Crown-4 CAS 294-93-9 (174)
 1,4,7,10-Tetraoxacyclododecane; cyclo(-O.(CH2.CH2.O)3.CH2.CH2-)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ nmr non-aq 25°C 100% C I K1=1.89 2001KZb (62683) 299
 Method: 7Li nmr. Medium: acetonitrile.
 Data for 20-80% w/w nitrobenzene/acetonitrile.

Li+ nmr non-aq 27°C 100% C K1=3.91 2000SMg (62684) 300
 Medium: acetonitrile. Method: 7Li nmr.

Li+ cal non-aq 25°C 100% C IH K1=3.52 1996DNa (62685) 301
 Medium: CH3CN. Data for LiX where X=AsF6-, BF4-, CF3SO3-, ClO4-. DH(K1)=
 -21.35 kJ mol⁻¹, DS=-4.6. In PC, K1=2.84, DH(K1)=-17.05, DS(K1)=-2.8.

Li+ nmr non-aq 27°C 1.0M C I K1=3.12 1996KAb (62686) 302
 Method: 7Li nmr. Medium: acetonitrile. Also data for nitromethane and
 20-80% w/w acetonitrile/nitromethane.

Li+ con alc/w 25°C 100% U I K1=1.320 1995DSb (62687) 303
 Medium : MeOH. In MeCN K1=3.140

Li+ con non-aq 25°C 100% U K1=3.2 1993EVa (62688) 304
 Medium: THF+CHCl3 (4:1 vol)

Li+ nmr non-aq 25°C 100% U K1=1.0 1989MGc (62689) 305
 Medium: tetrahydrofuran

Li+ con alc/w 25°C 100% U H T B2=2.73 1987BUa (62690) 306
 Medium: MeOH. DH(B2)=0 kJ mol⁻¹; DS=52.0 J K⁻¹ mol⁻¹

Li+ con non-aq 25°C 100% C K1=<0.0 1987ZBb (62691) 307
 Medium: MeOH.

Li+ con non-aq 25°C 100% U K1=3.40 1980HNa (62692) 308
 Medium: MeCN

Li+ vlt non-aq 25°C 100% U K1=2.93 1980MDa (62693) 309
 Medium: propylene carbonate

 Li+ nmr non-aq 27°C 100% C IH K1=4.25 1980SPb (62694) 310
 Method 7Li nmr. Medium: CH3CN. Also data for CH3NO2, PC, MeOH, acetone, PY
 DMSO, TMG, H2O. By calorimetry, DH(K1)=-16 kJ mol⁻¹, DS(K1)=27 J K⁻¹ mol⁻¹.

C8H17NO3 L CAS 41775-76-2 (6751)
 10-Aza-1,4,7-trioxacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	vlt	non-aq	25°C	100%	C		K1=4.5	2000HHa (62764)	311
-----	-----	--------	------	------	---	--	--------	-----------------	-----

Medium: acetonitrile, 0.1 M Et4NClO4. Method: dc polarography.

Li+ cal non-aq 25°C 100% C IH K1=4.24 1994DTa (62765) 312
 Medium: CH3CN. Data are for LiBF4. Data for LiAsF6 and Li(CF3SO3). DH(K1)=
 -19.91 kJ mol⁻¹, DS=14.4. In propylene carbonate, K1=3.69, DH=-14.63, DS=22

C8H18N2O2 L CAS 294-92-8 (654)
 1,7-Dioxo-4,10-diazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	vlt	non-aq	25°C	100%	C		K1=5.3	2000HHa (62844)	313
-----	-----	--------	------	------	---	--	--------	-----------------	-----

Medium: acetonitrile, 0.1 M Et4NClO4. Method: dc polarography.

Li+ sol non-aq 20°C 100% C K1=4.03 1983SLa (62845) 314
 Medium: CHCl3

C8H18O4 L Triglyme CAS 112-49-2 (2358)
 1,2-Bis(methoxyethoxy)ethane; CH3O.C2H4O.CH2.CH2.OC2H4.OCH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	U		K1=2.9	1993EVa (62988)	315
-----	-----	--------	------	------	---	--	--------	-----------------	-----

Medium: THF+CHCl3 (4:1 vol)

Li+	con	non-aq	25°C	100%	U	M		1982GJb (62989)	316
-----	-----	--------	------	------	---	---	--	-----------------	-----

Kout(LiL+A)=7.0
 Medium: 1,2-dichloroethane. A=picrate

C8H18O5 L Tetra-Et-Glycol CAS 112-60-7 (5664)
 2,2'-(Oxybis(2,2-ethanedioxy))-bis-ethanol; O(CH2.CH2.O.CH2.CH2.OH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	C		K1=3.3	1992MSe (63004)	317
-----	-----	--------	------	------	---	--	--------	-----------------	-----

Medium: 100% MeOH. Anion: picrate. Also data for nitrophenolate anions.

C8H19NO5 L Bis-tris CAS 6976-37-0 (2827)
 Bis-(2-hydroxyethyl)imino-tris(hydroxymethyl)methane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	1.00M	C	I	K1=-0.28	1982SSf (63062)	318
In 90 % (v/v) DMSO/water mixture: K1=0.61 (I=0.25 M)									

C8H20N4		L	Cyclen				CAS 294-90-6	(10)	
1,4,7,10-Tetraazacyclododecane; cyclo(-(NH.CH2.CH2.)4-)									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	EMF	non-aq	25°C	100%	U	I	K1=6.90	1996WPa (63294)	319
Medium: acetonitrile, 0.05 M NEt4ClO4. In propylene carbonate K1=5.6; in dimethylformamide K1=2.1									

C8H20O4P2		L					CAS 86536-56-3	(2076)	
1,2-Bis(2-dimethylphosphinylmethoxy)ethane; Me2P(O)CH2.O.CH2.CH2.O.CH2.P(O)Me2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=4.38	1989KSa (63311)	320
Medium: tetrahydrofuran/CHCl3 4:1 (vol)									

C9H5NOBr2		HL					CAS 521-74-4	(3279)	
5,7-Dibromo-8-hydroxyquinoline;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	?	100%	U		K1=5.54	1970PMc (63520)	321

C9H7NO		HL	Oxine				CAS 148-24-3	(504)	
8-Hydroxyquinoline (8-quinolinol);									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	27°C	100%	U	I	K1=1.96 B2= 2.36	1996MAb (64302)	322
Method: 7Li nmr. Medium: acetonitrile, 0.05 M LiClO4. Data for acetone (K1<0.5) and nitromethane (K1=1.87, K2=1.22).									

Li+	sp	alc/w	25°C	95%	U		K1=1.80	1993GSa (64303)	323
Medium: 95% w/w EtOH/H2O, 0.05 M Et4NClO4, by competitive spectrophotometry									

Li+	sp	non-aq	25°C	100%	U	I	K1=2.82 B2=4.54	1992GSa (64304)	324
Medium: MeCN. In acetone:K1=1.98; in MeOH:K1=0.91. By fluorimetry									

C9H11N3O7		H3L					(3877)		
N-(1-Methyl-2,4,6-trioxo-perhydropyrimidinyl)iminodiethanoic acid;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ gl R4N.X 20°C 0.10M U K1=4.86 1963IFb (66526) 325
Medium: Me4NNO3

C9H11O2F5 HL CAS 2145-68-8 (1251)

1,1,1,2,2-Pentafluoro-6,6-dimethyl-3,5-heptanedione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth diox/w 25°C 75% U K1=4.84 B2=8.23 1979MMA (66536) 326

C9H16O2 HL CAS 18362-64-6 (1134)

2,6-Dimethyl-3,5-heptanedione; (CH3)2.CH.CO.CH2.CO.CH(CH3)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl diox/w 30°C 75% U K1=9.23 B2=15.76 1975MMA (67746) 327

C9H20O6Cl2P2 L CAS 19928-93-7 (2633)

Dichloromethylenedi(phosphonic acid diethyl ester); Cl2C(PO.(OC2H5)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 22°C 100% U K1=1.11 1981SKd (68123) 328

Medium: CH3CN

C9H22O6P2 L CAS 1660-94-2 (2632)

Methylenedi(phosphonic acid diethyl ester) CH2(PO.(OC2H5)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 22°C 100% U K1=1.62 1981SKd (68260) 329

Medium: CH3CN

C10H6N2O5 HL (9002)

2,4-Dinitronaphthol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C K1=10.07 1997CHb (68444) 330

B(Li2L)=14.00

Medium: THF. By conductivity, species M2L and L2M are equivalent.

C10H6O8 H4L Pyromellitic Ac CAS 89-05-4 (519)

Benzene-1,2,4,5-tetracarboxylic acid; C6H2.(COOH)4

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 1.0M C K1=1.70 1991DDb (68520) 331

B(LiHL)=6.85

B(LiH2L)=10.76

B(LiH3L)=13.00

B(Li2L)=2.50

Medium: 1.0 M Et4NI. B(Li2HL)=6.87.

Li+ gl R4N.X 25°C 0.25M C I K1=1.44 1990DDb (68521) 332

B(LiHL)=6.33

B(LiH2L)=10.04

B(LiH3L)=12.12

B(Li2HL)=6.15

Medium: 0.25 M Et4NI. Data for 0.08-0.99 M. B(Li2L)=1.97

C10H8N2 L 2,2'-Bipyridyl CAS 366-18-7 (25)

2,2'-Bipyridine; (C5H4N)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 27°C 100% U I K1=2.11 B2= 2.97 1996MAb (69596) 333

Method: 7Li nmr. Medium: acetonitrile, 0.05 M LiClO4. Data for acetone
(K1<0.5) and nitromethane (K1=2.44, K2=2.29).

Li+ sp alc/w 25°C 95% U K1=1.61 1993GSa (69597) 334

Medium: 95% w/w EtOH/H2O, 0.05 M Et4NClO4, by competitive spectrophotometry

Li+ sp non-aq 25°C 100% U I K1=1.87 1992GSa (69598) 335

Medium: MeCN. In acetone:K1=1.85; in MeOH:K1=0.45. By fluorimetry

C10H10O2 HL Benzoylacetone CAS 93-91-4 (197)

1-Phenylbutane-1,3-dione; C6H5.CO.CH2.CO.CH3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=0.54 1988YSb (70745) 336

Medium: acetonitrile

Li+ gl alc/w 25°C 100% U I K1=3.1 1965LIa (70746) 337

Medium: MeOH, 0.1 M LiClO4. In EtOH: K1=3.2

C10H11NO5 H3L CAS 100844-86-8 (2108)

N-(2-Hydroxyphenyl)iminodiethanoic acid; HO.C6H4.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl KNO3 20°C 0.10M U K1=2.20 1963IFb (71042) 338

C10H11NO7S H3L (3335)

N-(2-Sulfophenyl)iminodiethanoic acid; HO3S.C6H4.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF KCl 20°C 0.10M C K1=2.26 1947SWa (71067) 339

C10H1102F7 HL CAS 17587-22-3 (1252)

1,1,1,2,2,3,3-Heptafluoro-7,7-dimethyl-4,6-octanedione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth diox/w 25°C 75% U K1=4.75 B2=8.32 1979MMA (71111) 340

C10H12N2O4 H2L CAS 16598-05-3 (967)

2-Pyridylmethyliminodiethanoic acid; C5H4N.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl KNO3 20°C 0.10M U K1=1.71 1963IFc (71264) 341

C10H13N3O7 H3L (3912)

1,3-Dimethyluramil-N,N-diethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 20°C 0.10M U K1=4.91 1963IFb (71806) 342

Medium: Me4NNO3

C10H13N5O4 L Adenosine CAS 58-61-7 (2154)

Adenosine, Adenine-9-beta-D-ribofuranoside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 25°C 100% U M 1976PSc (71945) 343

K(LiCl+L)=0.8

Medium: DMSO

C10H14N5O7P H2L AMP-5 CAS 18422-05-4 (842)

Adenosine-5'-monophosphoric acid, 5-Adenylic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M C T K1=1.22 1991SMa (72459) 344

IUPAC evaluation

C10H14O HL CAS 98-54-4 (458)

4-(t-Butyl)-1-hydroxybenzene; C4H9.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=10.87 1991AMa (72610) 345

Medium: THF. With 2-t-butylphenol K=10.13

C10H15N5O10P2 H3L ADP CAS 20398-34-9 (2181)

Adenosine-5'-diphosphoric acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	C	T	K1=1.32	1991SMa (72988)	346
IUPAC evaluation									

C10H16N2O8		H4L		EDTA			CAS 60-00-4	(120)	
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestic acid;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	oth/un	25°C	0.10M	C	TIH	K1=2.90 B(LiHL)=10.85 B(Li2L)=3.05	1985DRb (73927)	347
Data at 10-45 C and I=0.02-1.0 M in LiNO3. DH(K1)=2 kJ mol-1; DS=60. DH(LiHL)=-2; DS=198; DH(Li2L)=3; DS=65.									
Li+	cal	R4N.X	20°C	1.0M	C		K1=2.66 DH1= 3.26 kJ/mol	1976ANb (73928)	348
in Me4NCl; for 0.1 M Me4NClK1=2.97; DH1=0.84 kJ/mol;									
Li+	sp	R4N.X	25°C	0.50M	U		K1=2.43	1973CSa (73929)	349
Medium: (CH3)4NCl									
Li+	vlt	R4N.X	20°C	0.10M	U		K1=3.15	1972BZc (73930)	350
Medium: (CH3)4NOH									
Li+	gl	oth/un	25°C	0.32M	U		K1=2.85 B2=3.68 K(Li+HL)=0.86	1965BCa (73931)	351
Medium: CsCl									
Li+	cal	oth/un	25°C	0.05M	U	H		1954CHa (73932)	352
Medium: LiCl. DH(K1)=-0.42 kJ mol-1, DS=54.3 J K-1 mol-1									
Li+	EMF	KCl	20°C	0.10M	U	T	K1=2.79	1947SAa (73933)	353
Method: H electrode									

C10H16N5O13P3		H4L		ATP			CAS 56-65-5	(403)	
Adenosine-5'-triphosphoric acid;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	C	IH R	K1=1.78	1991SMa (74755)	354
IUPAC evaluation. DH(K1)=-4 kJ mol-1 (tentative)									
Li+	gl	oth/un	25°C	0.25M	U	H	K(Li+LiHL)=1.35 B(LiHL)=6.79	1986RSa (74756)	355
Li+	gl	oth/un	25°C	0.32M	U		K1=1.7 B2=2.23	1965BCa (74757)	356

$$K(\text{Li}+\text{HL})=0.8$$

Medium: CsCl

C10H17N04 H2L CAS 2848-06-8 (3916)

N-(Cyclohexyl)iminodiethanoic acid; C6H11.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl KNO3 20°C 0.10M U K1=1.74 1963IFb (74975) 357

C10H17N05 H2L CAS 6243-06-7 (3326)

N-(2-Hydroxycyclohexyl)iminodiethanoic acid; HO.C6H10.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl KNO3 20°C 0.10M U K1=2.19 1963IFb (74989) 358

C10H17N05 H2L (3917)

N-(Tetrahydropyran-2-ylmethyl)iminodiethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl KNO3 20°C 0.10M U K1=1.7 1963IFa (75002) 359

C10H17N5O16P4 H5L AQP CAS 1062-98-2 (3341)

Adenosine-5'-tetraphosphoric acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M C T K1=2.22 1991SMa (75159) 360

IUPAC evaluation

C10H18N2O5 H2L (5608)

1-Oxa-4,7-diazacyclononane-N,N'-diethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl KNO3 25°C 0.10M U K1=1.42 1990CCa (75235) 361

C10H20O5 L 15-Crown-5 CAS 33100-27-5 (576)

1,4,7,10,13-Pentaoxacyclopentadecane; cyclo(-(O.CH2.CH2)5-)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ ISE alc/w 25°C 100% C I T K1=1.24 2003ADa (76036) 362

IUPAC Tentative. Medium: 0-0.1 M various.

Li+ nmr non-aq 25°C 100% C I K1=3.58 2001KZb (76037) 363

Method: 7Li nmr. Medium: acetonitrile.

Data for 20-80% w/w nitrobenzene/acetonitrile.

Li+ nmr non-aq 27°C 100% C K1=4.76 2000SMg (76038) 364
Medium: acetonitrile. Method: 7Li nmr.

Li+ vlt non-aq 25°C 100% C I K1=4.2 1999WKb (76039) 365
Medium: acetonitrile, 0.10 M Et4NClO4. Also data for TMS, propylene
carbonate, acetone, formamide, DMF, DMA, DMSO, MeOH, EtOH.

Li+ nmr non-aq 27°C 1.0M C I K1=4.8 1996KAb (76040) 366
Method: 7Li nmr. Medium: acetonitrile. Also data for nitromethane and
20-80% w/w acetonitrile/nitromethane.

Li+ con alc/w 25°C 100% U I K1=1.314 1995DSb (76041) 367
Medium : MeOH. In MeCN K1=3.580

Li+ vlt non-aq 25°C 100% C K1=7.2 1995KTb (76042) 368
Method: ion transfer polarography. Medium: nitrobenzene, 0.05 M
tetrabutylammonium tetraphenylborate.

Li+ cal non-aq 25°C 100% M H K1=3.42 1994BCd (76043) 369
Medium: acetone. DH(K1)=-12.9 kJ mol⁻¹, TDS=6.5

Li+ cal non-aq 25°C 100% C IH K1=4.44 1994DTa (76044) 370
Medium: CH3CN. Data are for LiBF4. Data for LiAsF6 and Li(CF3SO3). DH(K1)=
-25.34 kJ mol⁻¹, DS=0.0. In propylene carbonate, K1=4.21, DH=-20.44, DS=12

Li+ con non-aq 25°C 100% C T K1=5.3 1988TKa (76045) 371
Medium: MeCN

Li+ con non-aq 25°C 100% C I K1=1.21 1987ZBb (76046) 372
Medium: MeOH. In 70% w/w MeOH/H2O, K1=1.02.

Li+ con non-aq 25°C 100% U K1=3.60 1980HNa (76047) 373
Medium: MeCN

Li+ nmr non-aq 27°C 100% C IH K1=>4 1980SPb (76048) 374
Method 7Li nmr. Medium: CH3CN. Also data for CH3NO2, PC, MeOH, acetone, PY
DMSO, TMG, H2O. By calorimetry, DH(K1)=-21 kJ mol⁻¹, DS(K1)=>6 J K⁻¹ mol⁻¹.

Li+ dis non-aq 25°C 100% U K1=4.2 1980TYa (76049) 375
Medium: propylene carbonate

C10H21NO4 L CAS 66943-05-3 (5818)

1-Aza-4,7,10,13-tetraoxacyclopentadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	vlt	non-aq	25°C	100%	C		K1=5.2	2000HHa (76185)	376
-----	-----	--------	------	------	---	--	--------	-----------------	-----

Medium: acetonitrile, 0.1 M Et4NClO4. Method: dc polarography.

C10H22N2O3 L Cryptand 2,1 CAS 31249-95-3 (835)
 4,7,13-Trioxa-1,10-diazacyclopentadecane (Trioxa(2,1)cryptand);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	vlt	non-aq	25°C	100%	C		K1=5.4	2000HHa (76323)	377

Medium: acetonitrile, 0.1 M Et4NClO4. Method: dc polarography.

Li+	cal	non-aq	25°C	100%	M	H	K1=3.13	1994BCd (76324)	378
-----	-----	--------	------	------	---	---	---------	-----------------	-----

Medium: acetone. DH(K1)=-9.1 kJ mol⁻¹, TDS=8.7

Li+	sp	non-aq	20°C	100%	U		K1=2.3	1992PSa (76325)	379
-----	----	--------	------	------	---	--	--------	-----------------	-----

Medium: DMF, 0.01 M Me4NI

C10H22O5 L Tetraglyme CAS 143-24-8 (121)
 2,5,8,11,14-Pentaoxapentadecane; (CH3.O.CH2.CH2.O.CH2.CH2.)20

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	25°C	100%	C		K1=4.69	1998KSc (76456)	380

Medium: 1,2-dichloroethane.

Li+	con	non-aq	25°C	100%	U		K1=3.2	1993EVa (76457)	381
-----	-----	--------	------	------	---	--	--------	-----------------	-----

Medium: THF+CHCl3 4:1(vol)

Li+	con	non-aq	25°C	100%	U	M		1982GJb (76458)	382
-----	-----	--------	------	------	---	---	--	-----------------	-----

Kout(LiL+A)=5.7
 Medium: 1,2-dichloroethane. A=tetraphenylborate

C11H11NO6 H3L CAS 1147-65-5 (425)
 N-(2'-Carboxyphenyl)iminodiethanoic acid; HOOC.C6H4.N(CH2.COOH)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	20°C	0.10M	U		K1=2.05	1963IFb (77829)	383

Medium: Me4NN03

Li+	EMF	KCl	20°C	0.10M	U		K1=2.18	1950WIa (77830)	384
-----	-----	-----	------	-------	---	--	---------	-----------------	-----

Method: H electrode

C11H18N2O8 H4L PDTA CAS 4408-81-5 (1655)
 1,2-Diaminopropane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	oth	R4N.X	25°C	0.50M	U		K1=4.01	1971CSb (79306)	385

Method: polarimetry. Medium: Me4NOH

C11H20O2 HL Dipivaloylmeth. CAS 1118-71-4 (363)
 2,2,6,6-Tetramethyl-3,5-heptanedione; (CH3)3C.CO.CH2.CO.C(CH3)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	gl	diox/w	30°C	75%	U			K1=5.76 B2=10.37	1975Mma (79750)	386

C11H22O5		L		16-Crown-5				CAS 55477-28-8 (1592)		
1,4,7,10,13-Pentaoxacyclohexadecane; cyclo(-(O.CH2.CH2)5.CH2.CH2-)										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	none	25°C	0.0	C			K1=1.05	2001Kmb (79859)	387

Li+	dis	none	25°C	0.0	C	M			1989TKc (79860)	388
Method: extraction of metal picrate/L from H2O into benzene.										
K(Li+HA(org)+L(org)=LiAL(org)+H)=-1.74. HA is picric acid.										

Li+	con	non-aq	25°C	100%	C	I		K1=4.5	1988TKa (79861)	389
Medium: MeCN. In propylene carbonate K1=3.3										

C12H5N7O12		L		Dipicrylamine				CAS 131-73-7 (1942)		
Di(2,4,6-trinitrophenyl)amine; HN(C6H2(NO2)3)2										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	dis	non-aq	25°C	100%	C			K1=3.9	1998KSc (80078)	390
Medium: 1,2-dichloroethane.										

Li+	dis	oth/un	25°C	var	U			K1=1.7	1970SSb (80079)	391
Method: paper chromatography										

C12H6O12		H6L		Mellitic acid				(7400)		
Benzenehexacarboxylic acid; (C(COOH)6										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	ISE	R4N.X	25°C	0	C	I		K1=2.95	1996RSb (80113)	392
B(NaHL)=10.13										
B(NaH2L)=15.57										
B(NaH3L)=20.06										
B(NaH4L)=22.80										
B(Li2L)=4.80, B(Li2HL)=11.33 B(Li2H2L)=17.13, BLi2H3L)=20.84										
B(Li3L)=6.43,B(Li3HL)=12.95, B(Li4L)=7.83. I=0-3 M Et4NI etc.										

C12H8N2		L		Phenanthroline				CAS 66-71-7 (144)		
1,10-Phenanthroline;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	nmr	non-aq	27°C	100%	U	I		K1=2.28 B2= 3.98	1996MAb (80474)	393
Method: 7Li nmr. Medium: acetonitrile, 0.05 M LiClO4.										

Also data for acetone: K1=2.20, K2=1.93.

Li+ sp alc/w 25°C 95% U K1=2.24 1993GSa (80475) 394
Medium: 95% w/w EtOH/H2O, 0.05 M Et4NClO4, by competitive spectrophotometry

Li+ sp non-aq 25°C 100% U I K1=3.01 B2=4.88 1992GSa (80476) 395
Medium: MeCN. In acetone:K1=3.11, K2=2.00; in MeOH:K1=0.95. By fluorimetry

C12H16O4 L CAS 25887-95-6 (686)
2,3-Benzo-1,4,7,10-tetraoxacyclododeca-2-ene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% U K1=1.16 2000EGa (81675) 396
Method: fluorescence emission spectroscopy. Medium: acetonitrile.

Li+ cal non-aq 25°C 100% U H K1=1.05 1989SSd (81676) 397
Medium: CH3CN

Li+ cal non-aq 25°C 100% U H K1=1.05 B2=2.80 1988SSc (81677) 398
Medium: MeCN

Li+ cal alc/w 25°C 100% U H K1=1.34 1985LWa (81678) 399

C12H20N2O8 H4L BDTA CAS 868-43-9 (1742)
DL-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOC.CH2)2N.CH(CH3).CH(CH3).N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth R4N.X 25°C 0.50M U K1=5.26(D)
K1=2.60(meso)
K(Li+HL)=1.68
Method: polarimetry. Medium: Me4NCl

C12H20O4P2 L CAS 82154-47-0 (2915)
1,2-Di((2-dimethylphosphinyl)methoxy)benzene; C6H4(OCH2PO(CH3)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.31 1982YSa (82642) 401
Medium: tetrahydrofuran+CHCl3 4:1(vol); M is 2,4-dinitrophenolate

C12H22O2 HL CAS 93269-15-9 (1250)
2,2,4,6,6-Pentamethyl-3,5-heptanedione; (CH3)3C.CO.CH(CH3).CO.C(CH3)3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth diox/w 25°C 75% U K1=6.85 B2=10.54 1979MMA (82859) 402

C12H23NO5 L (6793)
10-Methoxycarbonylethyl-1,4,7-trioxa-10-azacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal alc/w 25°C 100% U H K1=2.71 1990KMb (82946) 403
Medium: MeOH. DH=-3.0 kJ mol⁻¹

C12H24N2O3 L Cryptand 1,1,1 CAS 37095-49-1 (6636)
4,10,15-Trioxa-1,7-diazabicyclo[5.5.5]heptadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 20°C 100% U K1=1.7 1992PSa (83018) 404
Medium: DMF, 0.01 M Me4NI

C12H24O2 HL Lauric acid CAS 143-07-7 (2540)
Dodecanoic acid, CH₃.(CH₂)₁₀.COOH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl oth/un 26°C 0.00 U K1=4.12 1976HYa (83113) 405
B(LiHL2)=9.06

C12H24O4 L CAS 26996-94-3 (2541)
Tetramethyl-12-crown-4

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=3.46 1980HNa (83124) 406
Medium: MeCN

C12H24O6 L 18-Crown-6 CAS 17455-13-9 (577)
1,4,7,10,13,16-Hexaoxacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 27°C 100% C I K1=4.96 2001KZa (83454) 407
Method: 7Li nmr. Medium: nitromethane. In acetonitrile, K1=2.25

Li+ nmr non-aq 25°C 100% C I K1=1.92 2001KZb (83455) 408
Method: 7Li nmr. Medium: acetonitrile.
Data for 20-80% w/w nitrobenzene/acetonitrile.

Li+ dis non-aq 25°C 100% U K1=8.20 2000KSa (83456) 409
Medium: 1,2-dichloroethane

Li+ nmr non-aq 27°C 100% U I K1=2.52 2000SMd (83457) 410
Method: 7Li nmr. Medium: acetonitrile (AN). Also data for 50% w/w AN/

nitrobenzene (K1=2.80) and 50% w/w AN/nitromethane (K1=2.98).

Li+ cal non-aq 25°C 100% C H K1=2.50 1999WBa (83458) 411
Medium: N,N-dimethylformamide. DH(K1)=-0.7 kJ mol-1.

Li+ dis non-aq 25°C 100% C I 1998TKa (83459) 412
K(Li+A+L(org))=LiAL(org))=2.440
Method: Extraction from aqueous phase (I<0.03, pH 10.6-11.8) into
dichloromethane. Data for many non-aqueous phases. HA is picric acid.

Li+ cal non-aq 25°C 100% C K1=4.74 1997DZa (83460) 413
Medium: benzonitrile. DH(K1)=-38.48 kJ mol-1, DS(K1)=-38.3 J K-1 mol-1.

Li+ nmr non-aq 27°C 1.0M C I K1=2.30 1996KAb (83461) 414
Method: 7Li nmr. Medium: acetonitrile. Also data for nitromethane and
20-80% w/w acetonitrile/nitromethane.

Li+ vlt non-aq 25°C 100% C K1=7.5 1995KTb (83462) 415
Method: ion transfer polarography. Medium: nitrobenzene, 0.05 M
tetrabutylammonium tetraphenylborate.

Li+ cal non-aq 25°C 100% M H K1=2.41 1994BCd (83463) 416
Medium: acetone. DH(K1)=-19.8 kJ mol-1, TDS=-6.1

Li+ dis non-aq 25°C 100% U 1993INa (83464) 417
B(Li2P2L)=7.43
K is the equilibrium constant for extraction of the metal picrate (P) into
CH2Cl2. For extraction from D2O, B=7.53

Li+ con non-aq 25°C 100% C K1=2.782 1990SAb (83465) 418
Medium: propylene carbonate.

Li+ con non-aq 25°C 100% U K1=3.73 1980HNa (83466) 419
Medium: MeCN

Li+ nmr non-aq 27°C 100% C IH K1=2.34 1980SPb (83467) 420
Method 7Li nmr. Medium: CH3CN. Also data for CH3NO2, PC, MeOH, acetone, PY
DMSO, TMG, H2O. By calorimetry, DH(K1)=ca.0 kJ mol-1, DS(K1)=45 J K-1 mol-1.

C12H25NO5 L CAS 33941-15-0 (4939)
1,4,7,10,13-Pentaoxa-16-azacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	vlt	non-aq	25°C	100%	C		K1=3.2	2000HHa (83707)	421
Medium: acetonitrile, 0.1 M Et4NClO4. Method: dc polarography. ***** C12H26NO2P L (7849) N,N-Diethylcarbamoylmethyl-(dipropylphosphineoxide; -----									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=4.0	1999ESa (83720)	422
In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate									

C12H26N2O4		L			Cryptand 2,2		CAS 23978-55-4	(925)	
4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	C		K1=3.50	2002NFa (83860)	423
Medium: 100% MeOH. Method: electrospray ionization mass spectrometry.									
Li+	cal	non-aq	25°C	100%	M	H	K1=1.52	1994BCd (83861)	424
Medium: acetone. DH(K1) < 0 kJ mol ⁻¹									
Li+	sp	non-aq	20°C	100%	U		K1=1.2	1992PSa (83862)	425
Medium: DMF, 0.01 M Me4NI									
Li+	ISE	non-aq	25°C	100%	U	I	K1=6.98	1983CFa (83863)	426
Medium: MeNO2									
Li+	sol	non-aq	20°C	100%	C		K1=4.08	1983SLa (83864)	427
Medium: CHCl3									

C12H26O6		L			Pentaglyme		CAS 1191-87-3	(2498)	
2,5,8,11,14,17-Hexaoxaoctadecane; (CH3.O.CH2.CH2.O.CH2.CH2.O.CH2.)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.2	1993EVa (84007)	428
Medium: THF+CHCl3 (4:1 vol). Also data for other solvents									

C12H27N3O3		L			THETAC		(7199)		
1,4,7-Tris(hydroxyethyl)-1,4,7-triazacyclononane									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	EMF	non-aq	25°C	100%	C		K1=3.13	1997WWa (84088)	429
Medium: MeOH, 0.05M Et4NClO4.									
Method: Ag/Ag+ electrode; by competition with Ag+.									

C12H32N4O12P4		H8L			DOTPH		CAS 91987-74-5	(229)	
1,4,7,10-Tetraazacyclododecane-N,N',N'',N'''-tetramethylenephosphonic acid;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	none	25°C	0	U	M		1996RSa (84414)	430
							B(LiTmDOTP)=1.58		
							B(Li2TmDOTP)=3.03		

B(Li3TmDOTP)=4.31
 B(LiTmDOTPH)=9.16
 B(LiTmDOTPH2)=16.0, B(Li2TmDOTPH)=10.57, B(Li3TmDOTPH)=11.79
 mixed-metal complexes in the Li(I)-Tm(III)-DOTP ternary system

C13H1006 HL CAS 156426-82-3 (8800)
 3-Acetoacetyl-7-methyl-2H,5H-pyrano(4,3-b)pyran-2,5-dione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	20°C	100%	C			1998FLb (85005)	431

K(Li+HL=LiL+H)=-3.01

Method: absorption and fluorescence spectroscopy. Medium: acetonitrile.

 C13H1804 L Bz-13-crown-4 CAS 62150-58-7 (552)
 2,3,6,7,9,10-Hexahydro-5H-1,4,8,11-Benzotetraoxacyclotridecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	alc/w	25°C	100%	U	H	K1=1.26	1985LWa (86047)	432
Li+	sol	non-aq	25°C	100%	U	I	K1=3? K2=1.26	19810Ja (86048)	433

Medium: CH2Cl2: K1=5(?), K2=1.70. In CH3CN: K1=2.40

 C13H2605 L (6410)
 15,15-Dimethyl-1,4,7,10,13-pentaoxacyclohexadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	none	25°C	0.0	C		K1=1.15	2001KMb (86477)	434
Li+	con	non-aq	25°C	100%	C	I	K1=4.1	1992TFa (86478)	435

Medium: acetonitrile. In propylene carbonate, K1=3.13.

 C13H2606 L 19-Crown-6 CAS 55471-27-7 (8943)
 1,4,7,10,13,16-Hexaoxacyclononadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C	I	K1=3.73	2000TMb (86498)	436

Medium: CH3CN. In other media, K1=2.29 (propylene carbonate), 1.72 (DMSO).

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	oth/un	25°C	dil	C		K1=0.79	1999TMa (86499)	437

Self medium (LiCl). For LiNO3, K1=0.72; for LiClO4, K1=0.77.

 C14H802 L Anthraquinone CAS 84-65-1 (2781)
 Anthraquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ vlt R4N.X 25°C 0.20M U K1=1.06 1975PTc (86623) 438

 C14H8O3 HL CAS 129-43-1 (2778)
 1-Hydroxyanthraquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ vlt R4N.X 25°C 0.20M U K1=2.94 B2=5.49 1975PTc (86629) 439

 C14H16N2O8 H4L CAS 40774-59-2 (1901)
 1,2-Diaminobenzene-N,N,N',N'-tetraethanoic acid; C6H4(N(CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ gl R4N.X 25°C 0.10M C H K1=2.02 1990NNA (87956) 440
 K(LiL+H)=5.63
 Medium: Et4NClO4. DH(K1)=10.4 kJ mol⁻¹. DS(K1)=73 J mol⁻¹ K⁻¹.

Li+ gl R4N.X 25°C 0.10M U K1=2.39 1985MHb (87957) 441
 K(LiL+H)=6.34
 K(Li+HL)=1.81
 K(LiHL+H)=4.44

Medium: 0.10 M Me4NCl.

 C14H20O5 L Benzo15-crown-5 CAS 14098-44-3 (608)
 2,3-Benzo-1,4,7,10,13-pentaoxacyclopentadeca-2-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ dis none 25°C dil C I M 2002THb (88294) 442
 K(LiL+A)=-0.32
 K(Li+A+L(org)=LiAL(org))=1.720

HA is picric acid. Data for several aryl and alkyl solvents.

Method: extraction of metal picrate into dichloromethane/L.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con none 25°C 0.0 C K1=0.76 2002TTa (88295) 443

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ nmr non-aq 25°C 100% C I K1=3.22 2001KZb (88296) 444

Method: 7Li nmr. Medium: acetonitrile.

Data for 20-80% w/w nitrobenzene/acetonitrile.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C 100% C K1=5.60 2000ICa (88297) 445

Medium: nitromethane.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ nmr non-aq 27°C 100% C K1=4.56 2000SMg (88298) 446

Medium: acetonitrile. Method: 7Li nmr.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ vlt non-aq 25°C 100% C I K1=3.8 1999WKb (88299) 447

Medium: acetonitrile, 0.10 M Et4NClO4. Also data for TMS, propylene

carbonate, acetone, formamide, DMF, DMA, DMSO, MeOH, EtOH.

Li+ nmr non-aq 27°C 1.0M C I K1=4.51 1996KAb (88300) 448
Method: 7Li nmr. Medium: acetonitrile. Also data for nitromethane and
20-80% w/w acetonitrile/nitromethane.

Li+ cal non-aq 25°C 100% U H K1=3.20 1989SSd (88301) 449
Medium: CH3CN

Li+ con non-aq 25°C 100% C I K1=4.46 1988TKb (88302) 450
Medium: MeCN. In propylene carbonate K1=3.77; in MeOH 2.31

Li+ sp non-aq 22°C 100% U K1=6.09 1987CCc (88303) 451
In deuteriochloroform

Li+ con non-aq 25°C 100% U K1=3.77 1982TAa (88304) 452
Medium: propylene carbonate

C14H22N2O8 H4L CDTA CAS 482-54-2 (200)
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	oth	R4N.X	25°C	0.50M	U		K1=6.11 K(Li+HL)=1.15	1971CSa (88710)	453
-----	-----	-------	------	-------	---	--	--------------------------	-----------------	-----

Method: polarimetry. Medium: Me4NOH

Li+	vlt	KNO3	30°C	0.10M	U		K1=4.13	1967SSe (88711)	454
-----	-----	------	------	-------	---	--	---------	-----------------	-----

C14H23N3O10 H5L DTPA CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	gl	KNO3	25°C	0.10M	C		K1=3.1	1960WAa (89308)	455
-----	----	------	------	-------	---	--	--------	-----------------	-----

C14H24N2O10 EGTA CAS 67-42-5 (349)
Ethyleneglycol-0,0'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	kin	KCl	25°C	1.50M	U		K1=1.17	1968TFb (89891)	456
-----	-----	-----	------	-------	---	--	---------	-----------------	-----

C14H26N2O7 H2L (1567)
1,4,10-Trioxa-7,13-diazacyclopentadecane-N,N'-diethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	gl	R4N.X	25°C	0.10M	C		K1=2.139	1987DDb (90195)	457
-----	----	-------	------	-------	---	--	----------	-----------------	-----

C14H26O5 L CAS 17454-48-7 (5039)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	ISE	oth/un	25°C	dil	A			K1=<1	1971FRa (90272)	458

C14H28N2O4 L Cryptand 2,1,1 CAS 31250-06-3 (836)										
1,10-Diaza-4,7,13,18-tetraoxabicyclo[8,5,5]eicosane (2,1,1);										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	C	H		K1=6.66	1999WBa (90387)	459
Medium: N,N-dimethylformamide. Method: competitive titration against Ag+, using Ag+ ISE. By calorimetry: DH(K1)=-38.0 kJ mol-1.										

Li+	gl	R4N.X	25°C	0.05M	C	H		K1=6.6	1996BCh (90388)	460
Medium: 0.05 M Et4NClO4. By calorimetry: DH(K1)=-20.2 kJ mol-1.										

Li+	cal	non-aq	25°C	100%	M	H		K1=11.80	1994BCd (90389)	461
Medium: acetone. DH(K1)=-63.0 kJ mol-1, TDS=4.0										

Li+	EMF	non-aq	25°C	100%	U			K1=6.44	1993LRa (90390)	462
Medium: triethylphosphate, 0.05 M Et4NClO4										

Li+	sp	non-aq	20°C	100%	U			K1=6.6	1992PSa (90391)	463
Medium: DMF, 0.01 M Me4NI										

Li+	gl	R4N.X	25°C	0.05M	U			K1=6.98	1991LRc (90392)	464

Li+	cal	alc/w	25°C	100%	U	H		K1=7.90	1986BUd (90393)	465
In MeOH. DH=-33.9 kJ mol-1										

Li+	gl	alc/w	25°C	95%	C			K1=7.93	1981ANa (90394)	466
Medium: 95% MeOH, 0.1 M Me4NCl										

Li+	ISE	non-aq	25°C	100%	U	I		K1=6.99	1981CRa (90395)	467
Medium: DMF. In DMSO: K1=5.84; in EtOH: 8.47; in MeCN: >10; in NMP: 6.43										

Li+	cal	oth/un	25°C	0.10M	C				1981LIc (90396)	468
Medium: piperidine/HCl buffer, pH 11.4. DH(K1)=-21.8 kJ mol-1.										

Li+	ix	non-aq	25°C	100%	U			K1=12.87	1981SAa (90397)	469
Medium: propylene carbonate										

Li+	ISE	non-aq	25°C	100%	U			K1=12.4	1980CRa (90398)	470
Medium: Propylene carbonate										

Li+	EMF	non-aq	25°C	100%	C			K1=5.3	1979BLb (90399)	471
Method: Ag electrode; competition with Ag+. Medium: MeOH, 0.05 M Me4NClO4.										

Li+ ISE alc/w 25°C 100% U K1=8.04 1978CSb (90400) 472
Medium: MeOH

Li+ cal R4N.X 25°C 0.06M C H 1976KLc (90401) 473
Medium: 0.057 M Me4NBr. Method: flow microcalorimetry.
DH(K1)=-21.3 kJ mol⁻¹, DS(K1)=34 J K⁻¹ mol⁻¹.

Li+ gl R4N.X 25°C 0.05M C I K1=5.5 1975LSc (90402) 474
In 95% MeOH: K1=7.58; 100%: > 6

C14H28N2O4 L Cryptand 2,2,0 CAS 95334-31-9 (6544)
4,7,13,16-Tetraoxa-1,10-diazabicyclo[8.8.2]eicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	U	I	K1=7.8	1991ALa (90462)	475
Medium: MeCN, 0.05 M Et4NClO4. In acetone K1=8.9, MeOH K1=4.0, DMF K1=3.5, in pyridine K1=4.0.									

C14H30N2O2P L (2094)
P-(N,N-Diethylamidocarbonyl)methyl-P,P-dibutylphosphine oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=4.0	1999ESa (90554)	476
In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate									

Li+ con non-aq 25°C 100% U K1=3.11 1988YKa (90555) 477
Medium: tetrahydrofuran

C14H30N2O4P L (2096)
P-(N,N-Diethylamidocarbonyl)methyl-P,P-dibutoxyphosphine oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=2.89	1988YKa (90558)	478
Medium: tetrahydrofuran									

C14H30N2O4 L CAS 85726-93-8 (644)
4,10-Dimethyloxyethylidene-1,7-dioxy-4,10-diazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sol	non-aq	20°C	100%	C		K1=4.09	1983SLa (90562)	479
Medium: CHCl3									

C14H30N2O4 L CAS 31255-13-7 (2448)
N,N'-Dimethyl-cyclo-1,10-diaza-4,7,13,16-tetraoxa-octadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=<2	2004KVa (90581)	480
Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.									

C14H30N2O5 L (6722)									
7,13-Bis(2-hydroxyethyl)-1,4,10-trioxa-7,13-diazacyclopentadecane									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	U I		K1=2.36	1993RPa (90630)	481
Medium: dimethylformamide, 0.05 M Et4NClO4. By competition with Ag+.									
In methanol, K=2.85.									

C14H30O7 L CAS 1072-40-8 (2499)									
2,5,8,11,14,17,20-Heptaioxaheneicosane; CH3.O.(CH2.CH2.O)6.CH3									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	25°C	100%	C		K1=6.75	1998KSc (90698)	482
Medium: 1,2-dichloroethane.									

Li+	con	non-aq	25°C	100%	U		K1=3.3	1993EVa (90699)	483
Medium: THF+CHCl3 (4:1 vol). Also data for other solvents									

C15H11N3 L CAS 1148-79-4 (488)									
2,2':6'2"-Terpyridine; C5H4N.C5H3N.C5H4N									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	27°C	100%	U		K1=3.24	1996MAb (91159)	484
Method: 7Li nmr. Medium: nitromethane, 0.05 M LiClO4.									

C15H12O2 HL Diphenylacac CAS 120-46-7 (362)									
1,3-Diphenylpropane-1,3-dione, Dibenzoylmethane; C6H5.CO.CH2.CO.C6H5									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	100%	U		K1=4.1	1965LIa (91552)	485
Medium: MeOH, 0.1 M LiClO4									

Li+	gl	diox/w	30°C	75%	U		K1=5.95	1954FUa (91553)	486

C15H15O2P L CAS 76229-99-7 (2091)									
(Methylcarbonyl)methyldiphenylphosphine oxide; Ph2P(O)CH2C(O)Me									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=2.24	1988YSb (91913)	487
Medium: acetonitrile									

 C15H17O3P L CAS 40410-38-6 (5736)
 Methyl-(diphenoxymethyl)phosphine oxide; MePO(CH2.O.Ph)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=2.05	1989TKb (91987)	488

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

 C15H18N2O8 H4L CAS 101455-18-9 (1902)
 1-Methyl-3,4-diaminobenzene-N,N,N',N'-tetraethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	U		K1=2.31 K(LiL+H)=6.16 K(Li+HL)=1.43	1985MHb (92084)	489

Medium: 0.10 M Me4NCl.

 C15H23NO3 L CAS 84227-47-4 (5814)
 N-Benzyl-1-aza-4,7,10-Trioxacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	C	IH	K1=4.31	1996DNa (92257)	490

Medium: CH3CN. Data for LiX where X=AsF6-,BF4-,CF3SO3-,ClO4-. DH(K1)=-27.44 kJ mol-1, DS=-9.3. In PC, K1=4.59, DH(K1)=-24.70, DS(K1)=5.0.

 C15H24N2O2P L (7846)
 N,N-Diethylcarbamoymethyl-(P-phenyl-P-propylphosphineoxide);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=4.0	1999ESa (92329)	491

In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate

 C15H30N2O3 L CAS 72640-82-5 (6040)
 4,7,13-Trioxa-1,10-diazabicyclo[8.5.5]eicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	EMF	non-aq	25°C	100%	U	IH	K1=1.99	1993LRa (92517)	492

Medium: triethylphosphate, 0.05 M Et4NClO4. DH(K1)=-34.8 kJ mol-1, DS=91.9 J K-1 mol-1; Data also for tri-n-butylphosphate: K1=2.36

Li+	gl	R4N.X	25°C	0.05M	U		K1=2.40	1991LRc (92518)	493
-----	----	-------	------	-------	---	--	---------	-----------------	-----

Li+	ISE	non-aq	25°C	100%	U	I	K1=4.15	1990LAa (92519)	494
-----	-----	--------	------	------	---	---	---------	-----------------	-----

Medium: MeCN, 0.05 M Et4NClO4. In MeOH: K1=3.00; in DMF: K1=1.80; in DEF K1=1.72, in dimethylacetamide K1=1.85

Li+ kin non-aq 25°C 100% C K1=2.80 1987ABe (92520) 495
Medium: dimethylformamide.

C15H33N3O3 L CAS 220811-82-5 (7916)
1,4,7-Tris((S)-2-hydroxypropyl)-1,4,7-triazacyclononane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF non-aq 25°C 100% C I K1=3.39 2001WBa (92575) 496
Medium: methanol, 0.05 M Et4NClO4. In DMF, K1=3.29. Competition with Ag+.
Also data for the 1,4,7-tris((S)-2-hydroxy-2-phenylethyl- derivative.

C15H36N09P3 L CAS 37909-50-5 (2634)
(N,N-Dimethylamine)methylenetris(phosphonic acid diethyl ester);
(CH3)2N.C(CH2.PO(OC2H5)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 22°C 100% U K1=2.10 1981SKd (92604) 497
Medium: CH3CN

C16H18N02P L CAS 32159-22-1 (2098)
P-(N-Ethylamidocarbonyl)methyl-P,P-diphenylphosphine oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=3.31 1988YKa (93768) 498
Medium: tetrahydrofuran

C16H20O3P2 L CAS 82154-46-9 (2914)
Dimethylphosphinomethyl-diphenylphosphinomethyl-ether; Me2PO.CH2.O.CH2.PO(C6H5)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=3.92 1982YSa (94099) 499
Medium: tetrahydrofuran+CHCl3 4:1(vol); M is 2,4-dinitrophenolate

C16H24O5 L (2245)
1,3-Benzo-18-crown-5, 1,3-Benzo-5,8,11,14,17-pentaoxacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H 1979KLa (94344) 500
K(Li(picrate)+L)=2.0

Medium: CHCl3

Li+ dis non-aq 24°C 100% C 1977MTc (94345) 501

K(LiA+L)=2.0

Method: extraction of metal picrate (A) from H2O into CDCl3 containing L.

Data for the 5'-bromo, 5'-t-butyl, 5'-methoxy and 5'-cyanobenzo-derivs

C16H24O5 L AN(MOEO)2E CAS 60232-72-6 (2246)
18-Methoxy-16-methyl-3,6,9,12-tetraoxabicyclo[12.3.1]octadeca-1(18),14,16-triene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	25°C	100%	U	H		1979KLa (94354)	502
							K(Li(picrate)+L)=3.6		

Medium: CHCl3

C16H24O5 L CAS 75507-20-9 (605)
Benzyloxymethyl-1,4,7,10-tetraoxacyclododecane, Benzyloxymethyl-12-crown-4;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	22°C	100%	C			1984CBa (94360)	503
							K(Li+A+L(org)=LiAL(org))=<0		

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.

For extraction into 1,2-dichloroethane, K=<0.

C16H24O6 L Benzo18-crown-6 CAS 14098-24-9 (513)
2,3-Benzo-1,4,7,10,13,16-hexaoxacyclooctadeca-2-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C		K1=4.24	2000ICa (94422)	504

Medium: nitromethane.

Li+	cal	non-aq	25°C	100%	C	H	K1=2.44	1999WBa (94423)	505
-----	-----	--------	------	------	---	---	---------	-----------------	-----

Medium: N,N-dimethylformamide. DH(K1)=-0.5 kJ mol⁻¹.

Li+	sp	non-aq	22°C	100%	U		K1=5.77	1987CCc (94424)	506
-----	----	--------	------	------	---	--	---------	-----------------	-----

In deuterochloroform

C16H24O14 H4L CAS 61696-54-6 (6104)
1,4,7,10,13,16-Hexaoxacyclooctadeca-2,3,11,12-tetracarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	M		K1=3.8 B(LiHL)=8.5	1991FGb (94496)	507

Medium: 0.10 M Et4NNO3.

C16H26NO2P L (2093)
P-(N,N-Diethylamidocarbonyl)methyl(P-phenyl)(P-butyl)phosphine oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=3.9	1999ESa (94543)	508

In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate

Li+ con non-aq 25°C 100% U K1=3.38 1988YKa (94544) 509
Medium: tetrahydrofuran

C16H28N4O8 H4L DOTA CAS 60239-18-1 (1017)
1,4,7,10-Tetraazacyclododecane-1,4,7,10-tetraethanoic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M C K1=4.32 1982DSa (94912) 510

C16H30O6 L CAS 17454-53-4 (5148)
Cyclohexyl-18-crown-6;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF oth/un 25°C dil A 1971FRa (95101) 511

K1<0.7

C16H32N2O4 L Cryptand 1,2,1H CAS 119017-36-6 (6587)
4,7,14,20-Tetraoxa-1,10-diazabicyclo[8.7.5]docosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl alc/w 25°C 95% M K1=4.21 1990LNa (95118) 512

Medium: 95% MeOH, 0.05 M Bu4NBr. For the 9,13-dihydroxy- analogue: K1 < 2

C16H32N2O5 L Cryptand 2,2,1 CAS 31364-42-8 (837)
1,10-Diaza-4,7,13,16,21-pentaoxabicyclo[8.8.5]tricosane (2,2,1);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% M M K1=6.12 1999DSd (95229) 513

K(LiL+ClO4)=0.94

Medium: acetonitrile.

Li+ cal non-aq 25°C 100% C H K1=3.48 1999WBa (95230) 514

Medium: N,N-dimethylformamide. DH(K1)=-15.4 kJ mol⁻¹.

Li+ gl R4N.X 25°C 0.05M C K1=3.4 1996BCh (95231) 515

Medium: 0.05 M Et4NClO4.

Li+ cal non-aq 25°C 100% M H K1=8.11 1994BCd (95232) 516

Medium: acetone. DH(K1)=-38.1 kJ mol⁻¹, TDS=7.3

Li+ sp non-aq 20°C 100% U K1=3.96 1992PSa (95233) 517

Medium: DMF, 0.01 M Me4NI

Li+ cal alc/w 25°C 100% U H K1=4.69 1986BUd (95234) 518

In MeOH. DH=-10.3 kJ mol⁻¹

Li+ nmr non-aq 25°C 100% U K1=7.33 1986CHc (95235) 519
In CDCl₃ saturated with D₂O

Li+ ISE non-aq 25°C 100% C I K1=2.63 1985CKa (95236) 520
Medium: DMSO. In propylenecarbonate K1=9.67

Li+ gl alc/w 25°C 95% C K1=4.46 1981ANa (95237) 521
Medium: 95% MeOH, 0.1 M Me₄NCl

Li+ ISE non-aq 25°C 100% U I K1=10.33 1981CRa (95238) 522
Medium: MeCN. In DMF: K1=3.58; in EtOH: 5.34; in DMSO: 2.77; in NMP: 3.48

Li+ ISE non-aq 25°C 100% U K1=9.6 1980CRa (95239) 523
Medium: Propylene carbonate

Li+ ISE alc/w 25°C 100% U K1=5.38 1978CSb (95240) 524
Medium: MeOH

Li+ cal R4N.X 25°C 0.06M C H 1976KLc (95241) 525
Medium: 0.057 M Me₄NBr. Method: flow microcalorimetry.
DH(K1)=0 kJ mol⁻¹, DS(K1)=48 J K⁻¹ mol⁻¹.

Li+ gl R4N.X 25°C 0.05M C I K1=2.50 1975LSc (95242) 526
In 95% MeOH: K1=4.18; 100%: > 5

C16H32N4O4 L (6794)
4,10-Bis(N,N-dimethylethanamido)-1,7-dioxo-4,10-diazacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal alc/w 25°C 100% U H K1=5.38 1990KMb (95320) 527
Medium: MeOH. DH=-12.7 kJ mol⁻¹

C16H32O7 L (6411)
15-(2,5-Dioxahexyl)-15-methyl-1,4,7,10,13-pentaoxacyclohexadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% C I K1=4.45 1992TFa (95387) 528
Medium: acetonitrile. In propylene carbonate, K1=3.06.

C16H34N2O5 L (6953)
7,13-Bis(2-methoxyethyl)-1,4,10-trioxa-7,13-diazacyclopentadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF alc/w 25°C 100% U I K1=3.01 1994LLa (95416) 529
Medium: MeOH, 0.05M Et₄NClO₄. Also data for acetonitrile: K=9.13, PC: K=7.0

DMF: K=2.23, H2O: K<2 and pyridine: K=5.08. Method: by competition with Ag+.

C16H34N2O6 L CAS 69930-74-1 (1321)
 N,N'-Bis(2-hydroxyethyl)-1,7,10,16-tetraoxa-4,13-diazacyclooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	U		K1=2.29	1993RPa (95451)	530

Medium: dimethylformamide, 0.05 M Et4NClO4. By competition with Ag+.

 C16H34N4O2 L CAS 60598-04-1 (1530)
 4,7-Dimethyl-1,4,7,10-tetraaza-13,18-dioxabicyclo[8,5,5]eicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	U		K1=3.8	1978LMa (95471)	531

In CH3OH, K1>4.0. In 95 vol% CH3OH, K1>3.8.

 C16H34O8 L CAS 1191-91-9 (2500)
 2,5,8,11,14,17,20,23-Octaoxatetracosane; CH3.0.(CH2.CH2.0)7.CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.6	1993EVa (95492)	532

Medium: THF+CHCl3 (4:1 vol). Also data for other solvents

 C16H36N4 L CAS 54622-44-5 (147)
 5,5,7,12,12,14-Hexamethyl-1,4,8,11-tetraazacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	non-aq	25°C	100%	U		K1=3.8	1986STb (95540)	533

Medium: THF:CHCl3 4:1 v/v. Metal ions as 2,4-dinitrophenolates

 C16H36N4O4 L (6703)
 1,4,7,10-Tetrakis(2-hydroxyethyl)-1,4,7,10-tetraazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	EMF	non-aq	25°C	100%	U I		K1=8.07	1996WPa (95574)	534

Medium: acetonitrile, 0.05 M NEt4ClO4. In propylene carbonate K1=8.90

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	100%	C I		K1=3.09	1993TCa (95575)	535

Medium: MeOH, 0.05 M Et4NClO4. In DMF, K1=2.99

 C17H13N5O5 HL CAS 90163-26-1 (5212)
 1-(4'-(5'-Hydroxy-3'-methyl-1'-phenyl)pyrazolylazo)4-nitrobenzoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ sp mixed ? 77% U K1=4.24 1968DZa (95776) 536
Medium: 77% acetone

C17H20N4O6 HL Riboflavin CAS 83-88-5 (1438)
7,8-Dimethyl-10(D-1'-ribityl)isoalloxazine, Vitamin B2, Vitamin H

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sol oth/un 22°C U K1=-0.2 1980Lda (96339) 537

Medium: variable LiClO4 content 0.1-2.5 M

The same constant measured spectrophotometrically: K1=-1.2

C17H21O5P L (5732)
Methyldi(2-methoxyphenoxy)methylphosphine oxide; Me.PO(CH2.O.C6H4.OMe)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=2.65 1989TKb (96392) 538

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C17H24N2O10 HL CAS 217972-81-1 (8163)
9-(2-Hydroxy-3,5-dinitrophenoxy)methyl-1,4,8,11-tetraoxacyclotetradecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% C 1990SSe (96433) 539

K(Li+HL(org))=LiL(org)+H=-6.1

Method: extraction from aqueous phase (0.10 M MOPS, pH 7.3) into

1,2-dichloroethane. Data for 1,2-dialkyl- derivatives.

C17H24N4O11 L CAS 94616-60-1 (1039)
2,4,6-Trinitrophenylaminomethyl-15-crown-5

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp mixed 25°C 16% U K1=2.28 1984BP a (96464) 540

K(Li+HL)=1.09

C17H26O5 L CAS 92818-18-3 (8987)
12-[(Phenylmethoxy)methyl]-1,4,7,10-tetraoxacyclotridecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 22°C 100% C 1984CB a (96509) 541

K(Li+A+L(org))=LiAL(org)=0.9

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.

For extraction into 1,2-dichloroethane, K=1.84. In H2O, K(LiA+L)=3.80.

C17H26O5 L CAS 92818-15-0 (8986)
5-[(Phenylmethoxy)methyl]-1,4,7,10-tetraoxacyclotridecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	22°C	100%	C		K(Li+A+L(org)=LiAL(org))=1.59	1984CBa (96511)	542
Extraction of metal picrate from H2O into CDCl3. HA is picric acid. For extraction into 1,2-dichloroethane, K=2.94. In H2O, K(LiA+L)=4.43.									

C17H26O6		L					CAS 99159-90-7	(688)	
2,3-Benzo-1,4,7,10,13,16-hexaoxacyclononadeca-2-ene;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	22°C	100%	U		K1=5.33	1987CCc (96522)	543
In deuterochloroform									

C17H27NO5		L					CAS 98269-22-8	(8844)	
13-(2-Methoxyphenyl)-1,4,7,10-tetraoxa-13-azacyclopentadecane;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	RT	50%	C	I	K1=1.5	2002GNe (96544)	544
Medium: 50% v/v MeOH/H2O, pH 7.4 (0.01 M Tris buffer), 0.1 M Me4NCl. In 10% MeOH/H2O, K1=1.1.									

C17H34N2O4		L					CAS 142565-14-8	(6562)	
4,7,13,16-Tetraoxa-1,10-diazabicyclo[8.8.5]tricosane;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	EMF	non-aq	25°C	100%	C		K1=5.36	1993DLb (96745)	545
Medium: propylene carbonate, 0.05 M Et4NClO4.									

Li+	gl	R4N.X	25°C	0.05M	C	I	K1=2.08	1992CGb (96746)	546
Medium: Et4NClO4. In MeOH: K1=2.30; in DMF K1=2.21; in MeCN: K1=6.07									

C17H34N4O4S		L					CAS 503465-04-1	(9247)	
4,7,13,16-Tetraoxa-1,10,21,23-tetraazabicyclo[8.8.7]pentacosane-22-thione;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=1.23	2004KVa (96759)	547
Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.									

C17H36N4		L					(6788)		
12,17-Dimethyl-1,9,12,17-tetraazabicyclo[7.5.5]nonadecane;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	NaCl	25°C	0.15M	C		K1=2.6	1996BFc (96773)	548

 C17H37N5 L CAS 122874-65-1 (5903)
 5,12,17-Trimethyl-1,5,9,12,17-pentaazabicyclo[7.5.5]nonadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	mixed	25°C	80%	C		K1=5.0	1996BFc (96790)	549

Medium: 80% v/v DMSO/H2O, 0.15 M NaCl.

Li+	gl	NaCl	25°C	0.15M	C	H		1989BBe (96791)	550
-----	----	------	------	-------	---	---	--	-----------------	-----

DH(K1)=-2.1 kJ mol⁻¹, DS(K1)=54.3 J K⁻¹ mol⁻¹

 C17H38O2P2 L CAS 21245-67-8 (2100)
 Methylenebis(dibutylphosphine oxide); Bu2P(O)CH2P(O)Bu2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=5.0	1999ESa (96813)	551

In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate

Li+	con	non-aq	25°C	100%	U		K1=3.92	1988YKa (96814)	552
-----	-----	--------	------	------	---	--	---------	-----------------	-----

Medium: tetrahydrofuran

 C18H150P L CAS 791-28-6 (32)
 Triphenylphosphine oxide; (C6H5)3PO

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C	M	K1=4.873 K(LiClO4+L)=4.53 K(LiI+ClO4)=-0.265	1990SAb (97095)	553

Medium: propylene carbonate.

Li+	con	non-aq	25°C	100%	U		K1=1.95	1988YSb (97096)	554
-----	-----	--------	------	------	---	--	---------	-----------------	-----

Medium: acetonitrile

Li+	con	non-aq	25°C	100%	U		Kout(LiI+A)=3.9	1982GJb (97097)	555
-----	-----	--------	------	------	---	--	-----------------	-----------------	-----

Medium: 1,2-dichloroethane. A=tetraphenylborate

Li+	con	non-aq	25°C	100%	U		K(LiI+L)=2.6	1969SSi (97098)	556
-----	-----	--------	------	------	---	--	--------------	-----------------	-----

Medium: CH3CN

 C18H20O5 L CAS 14262-60-3 (5616)
 2,3:11,12-Dibenzo-1,4,7,10,13-pentaoxacyclopentadeca-2,11-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=3.003	2002YEa (97478)	557

Method: fluorescence spectroscopy. Medium: acetonitrile.

C18H22NO2P L (2092)
(N,N-Diethylamidocarbonyl)methyldiphenylphosphine oxide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C C K1=4.1 1999ESa (97507) 558
In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate

Li+ con non-aq 25°C 100% U K1=3.69 1988YKa (97508) 559
Medium: tetrahydrofuran

C18H22O5 L (5737)
1,7-Di(2-methoxyphenyl)-1,4,7-trioxaheptane; MeO.C6H4.O.C2H4.O.C2H4.O.C6H4.OMe

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=1.65 1989TKb (97566) 560
Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C18H27N2O3F L CAS 173417-90-8 (6571)
23-Fluoro-4,7,20-trioxa-1,10-diazatricyclo[8.7.5.1,12,16]tricos-12,14,16(23)triene
;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF non-aq 25°C 100% C H K1=4.34 1999BHa (97748) 561
Medium: MeOH, 0.05 M Et4NClO4. By calorimetry DH(K1)=-5.5 kJ mol⁻¹.
Method: by competition with Ag⁺, using Ag/Ag⁺ electrode.

C18H28N2O3 L CAS 154148-31-9 (6510)
4,7,20-Trioxa-1,10-diazatricyclo[8.7.5.1,12,16]tricos-12,14,16(23)-triene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF non-aq 25°C 100% C H K1=0.81 1999BHa (97771) 562
Medium: MeOH, 0.05 M Et4NClO4. By calorimetry DH(K1)=-2.1 kJ mol⁻¹.
Method: by competition with Ag⁺, using Ag/Ag⁺ electrode.

C18H28O5 L CAS 92818-19-4 (8988)
2-[(Phenylmethoxy)methyl]-1,4,8,11-tetraoxacyclotetradecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 22°C 100% C 1984CBa (97819) 563
K(Li+A+L(org))=LiAL(org))=2.08

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.

For extraction into 1,2-dichloroethane, K=3.29. In H2O, K(LiA+L)=4.91.

C18H28O5 L CAS 92818-28-05 (8989)
6-[(Phenylmethoxy)methyl]-1,4,8,11-tetraoxacyclotetradecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	dis	non-aq	22°C	100%	C				1984CBa (97821)	564
-----	-----	--------	------	------	---	--	--	--	-----------------	-----

K(Li+A+L(org))=LiAL(org))=2.30

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.

For extraction into 1,2-dichloroethane, K=2.69. In H2O, K(LiA+L)=5.15.

C18H28O6 L Benzo20-crown-6 (6354)

2,3-Benzo-1,5,8,11,14,18-Hexaoxacos-2-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	22°C	100%	U			K1=5.48	1987CCc (97836)	565
-----	----	--------	------	------	---	--	--	---------	-----------------	-----

In deuterochloroform

C18H28O6 L AN(MOEOE)20 CAS 60232-73-7 (2247)

21-Methoxy-19-methyl-3,6,9,12,15-pentaoxabicyclo[15.3.1]heneicos-1(21),17,19-triene
;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	dis	non-aq	25°C	100%	U	H			1979KLa (97847)	566
-----	-----	--------	------	------	---	---	--	--	-----------------	-----

K(Li(picrate)+L)=4.00

Medium: CHCl3

C18H28O6 L CAS 100433-53-6 (607)

Benzyloxymethyl-1,4,7,10,13-pentaoxacyclopentadecane, Benzyloxymethyl-15-crown-5;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	dis	non-aq	22°C	100%	C				1984CBa (97852)	567
-----	-----	--------	------	------	---	--	--	--	-----------------	-----

K(Li+A+L(org))=LiAL(org))=2.09

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.

In H2O, K(LiA+L)=4.94

C18H28O7 L Benzo21-crown-7 (6355)

2,3-Benzo-1,4,7,10,13,16,19-Heptaoxaheneicos-2-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	22°C	100%	U			K1=5.60	1987CCc (97857)	568
-----	----	--------	------	------	---	--	--	---------	-----------------	-----

In deuterochloroform

C18H32N2O8 L CAS 24951-52-8 (2560)

Cryptand-2,2,2-dilactam

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

 Li+ nmr non-aq 33°C 100% U I K1=2.64 1977HPa (98133) 569
 Medium: pyridine. In THF: K1=3.12; in MeCN: 3.13

C18H36N2O6 L Cryptand 3,2,1 (7303)
 1,10-Diaza-4,7,13,16,19,24-hexaoxabicyclo[8,11,5]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	cal	alc/w	25°C	95%	U	H	K1=3.14	1997Zia (98421)	570
-----	-----	-------	------	-----	---	---	---------	-----------------	-----

Medium: 95% v/v MeOH/H2O, 0.1 M. DH(K1)=-11.7 kJ mol⁻¹, DS=20.8 J K⁻¹ mol⁻¹

C18H36N2O6 L Cryptand 2,2,2 CAS 23978-09-8 (514)
 1,10-Diaza-4,7,13,16,21,24-hexaoxabicyclo[8.8.8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	M	M	K1=6.21	1999DSd (98629)	571
-----	-----	--------	------	------	---	---	---------	-----------------	-----

K(LiI+ClO4)=0.72

Medium: acetonitrile.

Li+	cal	R4N.X	25°C	0.10M	C	H		1996BCh (98630)	572
-----	-----	-------	------	-------	---	---	--	-----------------	-----

Medium: 0.10 M Et4NClO4. DH(K1)=-6.4 kJ mol⁻¹.

Li+	EMF	non-aq	25°C	100%	C	I	K1=6.98	1995DGa (98631)	573
-----	-----	--------	------	------	---	---	---------	-----------------	-----

Medium: acetonitrile, 0.05 M Et4NClO4. In benzonitrile, K1=8.18.
 Competitive method with Ag/Ag+ electrode.

Li+	cal	non-aq	25°C	100%	M	H	K1=4.62	1994BCd (98632)	574
-----	-----	--------	------	------	---	---	---------	-----------------	-----

Medium: acetone. DH(K1)=-23.9 kJ mol⁻¹, TDS=2.4

Li+	sp	non-aq	20°C	100%	U		K1=2.3	1992PSa (98633)	575
-----	----	--------	------	------	---	--	--------	-----------------	-----

Medium: DMF, 0.01 M Me4NI

Li+	cal	alc/w	25°C	100%	U	H	K1=2.46	1986BUd (98634)	576
-----	-----	-------	------	------	---	---	---------	-----------------	-----

In MeOH. DH=-3.7 kJ mol⁻¹

Li+	cal	non-aq	25°C	100%	U	H		1986DGa (98635)	577
-----	-----	--------	------	------	---	---	--	-----------------	-----

DH1 = -59.1 kJ mol⁻¹. Medium: nitromethane

Li+	cal	non-aq	25°C	100%	U	H		1985DGa (98636)	578
-----	-----	--------	------	------	---	---	--	-----------------	-----

Medium: propylene carbonate. DH1 = -36.4 kJ mol⁻¹

Li+	cal	non-aq	25°C	100%	U	H		1985DGa (98637)	579
-----	-----	--------	------	------	---	---	--	-----------------	-----

Medium: acetonitrile. DH1 = -29.8 kJ mol⁻¹

Li+	ISE	non-aq	25°C	100%	M		K1=11.49	1985DGb (98638)	580
-----	-----	--------	------	------	---	--	----------	-----------------	-----

Medium: nitromethane

Li+	gl	alc/w	25°C	95%	C		K1=1	1981ANa (98639)	581
-----	----	-------	------	-----	---	--	------	-----------------	-----

Medium: 95% MeOH, 0.1 M Me4NC1

Li+ ISE non-aq 25°C 100% U I K1=6.97 1981CRa (98640) 582
Medium: MeCN. In DMSO: < 1.0; in EtOH: < 2.3; in N-methylpropionamide: 2.97

Li+ ISE non-aq 25°C 100% U K1=6.9 1980CRa (98641) 583
Medium: Propylene carbonate

Li+ EMF non-aq 25°C 100% C I K1=4.3 1979BLb (98642) 584
Method: Ag electrode; competition with Ag+. Medium: MeOH, 0.05 M
Me4NC104. Also K1=2 (H2O), <2.0 (DMSO), 6.7 (CH3CN).

Li+ EMF oth/un 25°C 0.05M C I K1=<1.4 1978YTa (98643) 585
Method: competition with Tl+, using Tl amalgam electrode.
Electrolyte not stated. In DMSO, 0.10 M: K1<1

Li+ nmr non-aq 30°C 100% U I K1=2.94 1975CDa (98644) 586
Medium: pyridine. In aqueous soln: K1=0.99

Li+ gl R4N.X 25°C 0.05M C I K1=<2.0 1975LSc (98645) 587
In 95% MeOH: K1=1.8; 100%: 2.6

C18H36N4O4 L (6795)
4,10-Bis(N,N-dimethylpropanamido)-1,7-dioxa-4,10-diazacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Li+ cal alc/w 25°C 100% U H K1=2.99 1990KMb (98782) 588
Medium: MeOH. DH=-23.8 kJ mol-1

C18H36N6O3 L (6790)
1,4,7-Tris(N,N-dimethylethanamido)-1,4,7-triazacyclononane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Li+ gl R4N.X 25°C 0.10M M K1=3.91 1990KMb (98799) 589
Medium: 0.10 M Me4NNO3

C18H38N2O6 L CAS 85726-94-9 (645)
4,10-Dimethoxyethoxyethylidene-1,7-dioxo-4,10-diazacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Li+ sol non-aq 20°C 100% C K1=3.98 1983SLa (98822) 590
Medium: CHCl3

C18H38N2O6 L CAS 72911-99-0 (649)
4,13-Bis(2-methoxyethyl)-1,7,10,16-tetraoxo-4,13-diazacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sol non-aq 20°C 100% C K1=3.89 1983SLa (98840) 591
Medium: CHCl3

C18H3809 L Glyme-9 CAS 25990-94-7 (7806)
2,5,8,11,14,17,20,23,26-Nonaoxaheptacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	dis	non-aq	25°C	100%	C		K1=7.53	1998KSc (98875)	592
-----	-----	--------	------	------	---	--	---------	-----------------	-----

Medium: 1,2-dichloroethane.

C19H2306P L (5731)
1,2:8,9-Dibenzo-5-methylphosphinyl-3,7,10,13,16-pentaoxacyclohexadeca-1,8-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	U		K1=3.26	1989TKb (99346)	593
-----	-----	--------	------	------	---	--	---------	-----------------	-----

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C19H24N02P L (2095)
P-(N,N-Diethylamidocarbonyl)methyl,P,P-diphenylphosphine oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	U		K1=4.12	1988YKa (99348)	594
-----	-----	--------	------	------	---	--	---------	-----------------	-----

Medium: tetrahydrofuran

C19H3005 L CAS 92818-26-3 (8991)
10-[(Phenylmethoxy)methyl]-1,4,8,12-tetraoxacyclopentadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	dis	non-aq	22°C	100%	C			1984CBa (99431)	595
-----	-----	--------	------	------	---	--	--	-----------------	-----

K(Li+A+L(org))=LiAL(org))=1.04

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.
For extraction into 1,2-dichloroethane, K=1.91. In H2O, K(LiA+L)=3.89.

C19H3005 L CAS 92818-23-0 (8990)
2-[(Phenylmethoxy)methyl]-1,4,8,12-tetraoxacyclopentadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	dis	non-aq	22°C	100%	C			1984CBa (99433)	596
-----	-----	--------	------	------	---	--	--	-----------------	-----

K(Li+A+L(org))=LiAL(org))=0.78

Extraction of metal picrate from H2O into CDCl3. HA is picric acid.
For extraction into 1,2-dichloroethane, K=1.36. In H2O, K(LiA+L)=3.65.

C19H39N305 L CAS 60598-00-7 (1537)
4-Methyl-1,4,10-triaza-7,13,16,21,24-pentaoxa-bicyclo[8,8,8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.10M	U		K1=1.5	1978LMa (99493)	597
In 95 vol% MeOH, K1=4.0.									

C20H22O4		L					CAS 82645-28-1	(8945)	
o,o'-(Triethyleneglycoldiyl)-(Z)-stilbene;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C		K1=5.47	2000ICa (99928)	598
Medium: nitromethane.									

C20H22O6		L					(6834)		
1,8-Bis(2-Formyphenoxy)-3,6-dioxaoctane; (CH2.0.CH2.CH2.0.C6H4.CHO)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.3	1993EVa (99932)	599
Medium: THF+CHCl3 (4:1 vol)									

C20H24N2O5		L					CAS 165815-06-5	(8936)	
N-(2-Pyridylmethylene)-4-aminobenzo-15-crown-5;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C I M		K(ZnA2L+Li)=4.13	2002YPc (99952)	600
Medium: MeCN, 0.10 M n-Bu4NPF6. By 1H nmr in CDCl3, K(ZnA2L+Li)=4.06.									
A is p-thiocresol.									

C20H24O6		L					DiBz-18-Crown-6 CAS 14187-32-7	(604)	
2,3:11,12-Dibenzo-1,4,7,10,13,16-hexaoxacyclooctadeca-2,11-diene									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	25°C	100%	C I		K1=<0.3	2001KZb (100160)	601
Method: 7Li nmr. Medium: acetonitrile.									
Data for 20-80% w/w nitrobenzene/acetonitrile.									
Li+	con	non-aq	25°C	100%	C		K1=4.79	2000ICa (100161)	602
Medium: nitromethane.									
Li+	oth	oth/un	25°C	0.04M	C		K1=-0.3	1998TIa (100162)	603
Method: capillary electrophoresis.									
Medium: 0.005 M phosphate buffer, pH 7.1, 0.04 M MCl.									
Li+	nmr	non-aq	27°C	1.0M	C I		K1=0.86	1996KAb (100163)	604
Method: 7Li nmr. Medium: acetonitrile. Also data for nitromethane and									

20-80% w/w acetonitrile/nitromethane.

Li+ vlt non-aq 25°C 100% U K1=11.3 1990SPa (100164) 605
Medium: 1,2-dichloroethane

Li+ con non-aq 25°C 100% U K1=4.06 1986STb (100165) 606
Medium: THF:CHCl3 4:1 v/v. M as 2,4-dinitrophenolate

Li+ con non-aq 25°C 100% U K1=3.48 1985YKa (100166) 607
Medium: EtOH+CHCl3 1:1; M is used in nitrophenolate form

Li+ con non-aq 25°C 100% U M Kout(LiL+A)=3.2 1982GJb (100167) 608
Medium: 1,2-dichloroethane. A=tetraphenylborate

C20H26O6 L CAS 84884-14-0 (2236)
2,3-Naphtho-18-crown-6, 2,3-Naphtho-1,4,7,10,13,16-hexaoxacyclooctadeca-2-ene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H K(M(picrate)+L)=4.35 1979KLa (100347) 609
Medium: CHCl3

C20H32O7 L AN(MOEEO)2E (2248)
24-Methoxy-22-methyl-3,6,9,12,15,18-hexaoxabicyclo[18.3.1]-tetracos-1(24),20,22-triene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H K(Li(picrate)+L)=3.5 1979KLa (100492) 610
Medium: CHCl3

C20H32O8 L Benzo24-crown-8 (6356)
2,3-Benzo-1,4,7,10,13,16,19,22-Octaoxatetracos-2-ene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 22°C 100% U K1=5.52 1987CCc (100497) 611
In deuterochloroform

C20H34N4O HL (7763)
14,19-Dimethyl-1,11,14,19-tetraazatricyclo[9.5.5.14,8]docosa-4,6,8(22)-trien-22-ol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.15M C K(Li+L=LiH-1L+H)=-9.1 2000Mfa (100513) 612
Medium: 0.10 M NMe4Cl.

C20H36O6 L DiCy-18-crown-6 CAS 16069-36-6 (1653)
2,3:11,12-Dicyclohexyl-1,4,7,10,13,16-hexaoxacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 27°C 100% C I K1=5.60 2001KZa (100663) 613
Method: 7Li nmr. Medium: nitromethane. In acetonitrile, K1=3.41

Li+ nmr non-aq 25°C 100% C I K1=2.51 2001KZb (100664) 614
Method: 7Li nmr. Medium: acetonitrile.
Data for 20-80% w/w nitrobenzene/acetonitrile.

Li+ dis non-aq 25°C 100% U K1=9.26 2000KSa (100665) 615
Medium: 1,2-dichloroethane

Li+ nmr non-aq 27°C 1.0M C I K1=3.14 1996KAb (100666) 616
Method: 7Li nmr. Medium: acetonitrile. Also data for nitromethane and
20-80% w/w acetonitrile/nitromethane.

Li+ dis non-aq 25°C 100% U H 1979KLa (100667) 617
K(Li(picrate)+L)=2.28
Medium: CHCl3

Li+ ISE oth/un 25°C dil A K1=0.6 1971FRa (100668) 618
Data for isomer A

C20H40N2O4 L (6625)
1,10-Diaza-4,7,13,16-tetraoxabicyclo[8.8.8]hexacosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl non-aq 25°C 100% C I K1=3.7 1992LSc (100776) 619
Medium: MeCN, 0.05 M Et4NClO4. In MeOH K1=2.2; in DMF K1=1.9; in H2O K1<2

C20H40N2O7 L Cryptand 3,2,2 CAS 31255-22-8 (1763)
Cryptand 3,2,2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.05M C I K1=<2.0 1975LSc (100815) 620
In 95% MeOH: K1 < 2; 100%: 2.3

C20H42N4O4 L CAS 39678-14-3 (1543)
4,7-Dimethyl-1,4,7,10-tetraaza-13,16,21,24-tetraoxa-bicyclohexacosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M U I K1=2.4 1978LMa (100889) 621
In CH3OH, K1>4.0, in 95 vol% CH3OH, K1=3.8.

C20H42O5 L CAS 9002-92-0 (8207)
1-Hydroxy-11-oxydodecane-3,6,9-trioxaundecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+		dis non-aq	25°C	100%	C			K1=1.61	1999KKb (100902)	622

Medium: MIBK. Method: distribution of metal picrates in H2O/MIBK(ligand) system. Also data for L= HO(CH2.CH2.O)n.(CH2)11.CH3, n=6 and 8.

C20H44N4O4 L CAS 102202-74-4 (6041)
1,4,7,10-Tetra-(2-hydroxypropyl)-1,4,7,10-tetraazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+		EMF non-aq	25°C	100%	C I			K1=7.65	1997DMd (100928)	623

Method: Ag electrode; competitive titration. Medium: acetonitrile, 0.05 M Et4NClO4. Also data for PC (K1=6.7), MeOH (4.0), DMF (3.24), H2O (<2).

C20H44N4O4 L (6730)
1,4,7,10-Tetra-(2-methoxyethyl)-1,4,7,10-tetrazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+		gl non-aq	25°C	100%	U I			K1=9.34	1996SDa (100942)	624

Medium: MeCN, 0.05 M Et4NClO4. In MeOH: K1=4.1, DMF: 3.61, DMSO: 2.82, propylene carbonate: 8.0

Li+		gl R4N.X	25°C	0.10M	C			K1=<2.0	1993SFb (100943)	625
-----	--	----------	------	-------	---	--	--	---------	------------------	-----

Medium: 0.1 M Et4NClO4.

C21H19OP L CAS 29942-64-1 (2087)
C-Methylcarbonylmethylenetriphenylphosphorane; Ph3P:CHC(O)CH3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+		con non-aq	25°C	100%	U			K1=2.75	1988YSb (101145)	626

Medium: acetonitrile

C21H24O8 L CAS 78708-41-5 (799)
2,3:9,10-Dibenzo-1,4,8,11,14-pentaoxacyclohexadeca-2,9-diene-6-oxyethanoic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+		gl alc/w	25°C	80%	M H			K1=3.27	1985AEb (101267)	627

Medium: 80% w/w MeOH/H2O, pH=11. By calorimetry: DH(K1)=-24.8 kJ mol⁻¹, DS(K1)=-20.4 J K⁻¹ mol⁻¹.

C21H27O8P L CAS 71817-08-8 (6905)
1,2:10,11-Dibenzo-16-methylphosphonyl)-3,6,9,12,15,17,20-heptaoxacycloeicosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	nmr	non-aq	20°C	100%	U		K1=2.3	1982BGe (101300)	628
-----	-----	--------	------	------	---	--	--------	------------------	-----

Medium: Acetone-D6; Method - 1H NMR

C21H30N7O17P3		H4L		NADPH			CAS 2646-71-1	(7185)	
---------------	--	-----	--	-------	--	--	---------------	--------	--

Nicotinamide adenine dinucleotide phosphate reduced;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	nmr	none	RT	0	U			1995MMf (101374)	629
-----	-----	------	----	---	---	--	--	------------------	-----

K1eff=1.70

Keff(2Li+L)=0.58

Medium: D2O, pH 8.5-9.5. Coordination site is the adenine or nicotinamide phosphate

C21H30O2P2		L					(7851)		
------------	--	---	--	--	--	--	--------	--	--

P'P'-Diphenyl-P,P-dibutylmethylenediphosphinedioxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C		C		K1=5.1	1999ESa (101385)	630
-----	-----	--------	------	--	---	--	--------	------------------	-----

In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate

C21H31O7P3		L					CAS 82154-48-1	(2916)	
------------	--	---	--	--	--	--	----------------	--------	--

Methyl di((2-dimethylphosphinylmethoxy)phenoxy)methyl)phosphineoxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	con	non-aq	25°C	100%	U		K1=4.54	1982YSa (101420)	631
-----	-----	--------	------	------	---	--	---------	------------------	-----

Medium: tetrahydrofuran+CHCl3 4:1(vol); M is 2,4-dinitrophenolate

L=CH3P(O)[CH2OC6H4OCH2P(O)(CH3)2]2

C21H42N4O6S		L					CAS 503465-05-2	(9248)	
-------------	--	---	--	--	--	--	-----------------	--------	--

4,12,18,21,26,29-Hexaoxa-1,7,9,15-tetraazabicyclo[13.8.8]hentriacontane-8-thione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	gl	alc/w	25°C	95%	C		K1=1.30	2004KVa (101464)	632
-----	----	-------	------	-----	---	--	---------	------------------	-----

Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.

C21H42N6O3		L					(6791)		
------------	--	---	--	--	--	--	--------	--	--

1,5,9-Tris(N,N-dimethylethanamido)-1,5,9-triazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	gl	R4N.X	25°C	0.10M	M		K1=4.21	1990KMb (101475)	633
-----	----	-------	------	-------	---	--	---------	------------------	-----

Medium: 0.10 M Me4NNO3

C22H20N2O4 L CAS 207461-96-9 (8955)
(5Z)-12,13,20,21-Tetrahydrotribenzo[b,f,l][1,8,11,14,4,5]tetraoxadiazacyclohexadecine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	RT	100%	C		K1=3.42	2000GDa (101696)	634

Medium: acetonitrile.

C22H25O3P L CAS 97745-35-2 (2069)
Adamantyl(diphenoxy)phosphonyl

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sol	non-aq	25°C	100%	U		K1=3.01	1987TCa (101924)	635

Medium: CH2Cl2, 2% MeCN. Metal as picrate

C22H26O5 L CAS 160978-39-2 (8944)
o,o'-(Tetraethyleneglycoldiyl)-(Z)-stilbene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C		K1=6.0	2000ICa (101998)	636

Medium: nitromethane.

C22H28N2O6 L CAS 449740-17-4 (8937)
N-(2-Pyridylmethylene)-4-aminobenzo-18-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C	M	K(ZnA2L+Li)=1.86	2002YPc (102017)	637

Medium: MeCN, 0.10 M n-Bu4NPF6. A is p-thiocresol.

C22H28O7 L Dibenzo-21-Cr-7 CAS 14098-41-0 (2876)
2,3:11,12-Dibenzo-1,4,7,10,13,16,19-heptaoxacycloheicosane-2,11-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=4.3	1993EVa (102050)	638

Medium: THF+CHCl3 (4:1 vol)

C22H30N2O2P L CAS 97937-88-7 (2097)
P-(N,N-Dibutylamidocarbonyl)methyl,P,P-diphenylphosphine oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.58	1988YKa (102099)	639

Medium: tetrahydrofuran

C22H32O7P2 L (2078)
1,5-Bis(2-(dimethylphosphinylmethoxy)phenoxy)-3-oxapentane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+		con non-aq	25°C	100%	U			K1=4.02	1989KSa (102207)	640

Medium: tetrahydrofuran/CHCl3 4:1 (vol)

C22H36N2O6 L Bz-Cryptand 222 CAS 31250-18-7 (2269)
5,6-Benzo-4,7,13,16,21,24-hexaoxa-1,10-diazabicyclo[8:8:8]hexacosa-5-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.05M	U	H		K1=1.7	1998DBa (102276)	641

Medium: 0.05 M Et4NClO4. By calorimetry: DH(K1)=-5.7 kJ mol-1,

Li+	gl	oth/un	25°C	0.02M	U	H		K1=2.19	1980CKa (102277)	642
-----	----	--------	------	-------	---	---	--	---------	------------------	-----

DH=-12.5 kJ mol-1. Alternative method: calorimetry

C22H36O9 L Benzo-27-Crown9 CAS 63144-76-3 (2842)
2,3-Benzo-1,4,7,10,13,16,19,22,25-nonanoxacycloheptacosa-2-ene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	non-aq	22°C	100%	U			K1=5.47	1987CCc (102300)	643

In deuteriochloroform

C22H40O6 L CAS 76993-47-0 (2340)
2,5,8,11,14,17-Hexaoxatricyclo[22.4.0.0(18,23)]octacosane (trans-cis-trans isomer)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	nmr	non-aq	24°C	100%	U	M			1981BEb (102371)	644

K(Li(picrate)+L)=5.5

Medium: CDCl3

C22H44N2O8 L Cryptand 3,3,2 CAS 132162-57-3 (1762)
Cryptand 3,3,2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	gl	R4N.X	25°C	0.05M	C			K1=<2	1975LSc (102429)	645

C22H44N6O5S2 L CAS 503465-08-5 (9241)
9,20,23,28,31-Pentaoxa-1,4,6,12,14,17-hexaazabicyclo[15.8.8]tritriacontane-5,13-dithione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C			K1=<2	2004KVa (102439)	646

Medium: 95% MeOH/H₂O, 0.01 M Et₄NClO₄.

C22H₄₆N₂O₈ L CAS 85726-96-1 (647)
4,10-Dimethyloxyethoxyethoxyethylidene-1,7-dioxo-4,10-diazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sol	non-aq	20°C	100%	C		K1=4.19	1983SLa (102455)	647

Medium: CHCl₃

C22H₄₆N₂O₈ L CAS 85726-97-2 (650)
4,13-Dimethyloxyethoxyethylidene-1,7,10,16-tetraoxo-4,13-diazaoctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sol	non-aq	20°C	100%	C		K1=3.88	1983SLa (102458)	648

Medium: CHCl₃

C22H₄₈N₆O₂ L CAS 39678-22-3 (1542)
4,7,13,16-Tetramethyl-1,4,7,10,13,16-hexaaza-21,24-dioxabicyclohexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	U		K1=3.5	1978LMa (102488)	649

C23H₂₁O₂P L CAS 1474-32-4 (2089)
C,C-Di(methylcarbonyl)methylenetriphenylphosphorane; Ph₃P:C(C(O)Me)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=2.13	1988YSb (102643)	650

Medium: acetonitrile

C23H₂₂N₄O₄ HL CAS 207800-89-3 (8966)
19,20,22,23-Tetrahydro-9-methyl-11,7-metheno-7H-dibenzotrioxatetraazacycloeicosin-25-ol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	diox/w	25°C	50%	C	I	K1=1.70	2001INa (102645)	651

Medium: 50% v/v dioxane/H₂O, 3% v/v triethylamine; pH 12. In 50% v/v dioxane/H₂O with Et₄NOH, K1=2.94.

C23H₃₀N₂O₄ L CAS 361454-16-2 (8960)
N-(Phenylmethylene)-4-(1,4,7,10-tetraoxa-13-azacyclopentadec-13-yl)benzamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	RT	100%	C		K1=2.77	2001AVa (102751)	652

Method: spectrophotometric titration. Medium: acetonitrile.

C23H30N4O7 L CAS 356535-57-4 (8845)
13-[2-Methoxy-4-[(4-nitrophenyl)azo]phenyl]-1,4,7,10-tetraoxa-13-azacyclopentadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp alc/w RT 50% C K1=ca.0.5 2002GNe (102768) 653
Medium: 50% v/v MeOH/H2O, pH 7.4 (0.1M Tris buffer), 0.1 M Me4NCl.

C24H24N2O4 L (5741)
1,10-Di(8-quinolyl)-1,4,7,10-tetraoxadecane; C9H6N.O.C2H4.O.C2H4.O.C2H4.O.C9H6N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.9 1989BEa (102938) 654
Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C24H25O7P L (2067)
Phenylphosphonyldibenzo-17-crown-6

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sol non-aq 25°C 100% U K1=2.38 1987TCa (102965) 655
Medium: CH2Cl2, 2% MeCN

C24H32O6 L ANAN(MOE0)2E (2242)
2,3:4,5-Di(1,3-(2-methoxy-5-methylbenzo))-9,12,15,18-tetraoxacyclooctadeca-2,4-diene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H 1979KLa (103071) 656
K(Li(picrate)+L)=4.76
Medium: CHCl3

C24H32O6 L AN(MOE0M)2AN (2244)
23,24-Dimethoxy-10,21-dimethyl-3,6,14,17-tetraoxatricyclo-tetracos-1(23),8(24),9,11,19,21hexaene

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H 1979KLa (103077) 657
K(Li(picrate)+L)=2.95
Medium: CHCl3

C24H32O6 L DP(OE0E0)2E CAS 60985-77-5 (2237)
3,4:5,6-Bis(2-methylbenzo)-2,7,10,13,16,19-hexaoxacyclodocosa-3,5-diene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

C24H36O10P2 L (5726)
1,4-Bis(2-(diethoxyphosphinylmethoxy)phenyl)-1,4-dioxabutane;
2(EtO)2PO.CH2O.C6H4.O.CH2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=3.7 1989EVa (103296) 666
Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C24H42O10 L (2505)
2,5,8,11,14,17,20,23,26,29-Decaoxa-15,16-benzo-triconta-15-ene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 22°C 100% U K1=5.46 1987CCc (103397) 667
In deuteriochloroform

C24H44O5 L (2341)
16,18,23,25-Tetramethyl-2,5,8,11,14-pentaoxatricyclo(22.4.0.0(15,20))pentacosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 24°C 100% U M K(Li(picrate)+L)=4.0 1981BEb (103410) 668
Medium: CDCl3

C24H48N2O9 L Cryptand 3,3,3 CAS 132162-61-9 (1761)
Cryptand 3,3,3

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.05M C K1=<2 1975LSc (103465) 669

C24H48N6O6S2 L CAS 503465-10-9 (9242)
9,12,23,26,31,34-Hexaoxa-1,4,6,15,17,20-hexaazabicyclo[18.8.8]hexatricontane-5,16-dithione;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl alc/w 25°C 95% C K1=<2 2004KV a (103506) 670
Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.

C24H48N8O4 L (6789)
1,4,7,10-Tetrakis(N,N-dimethylethanamido)-1,4,7,10-tetraazacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M M K1=5.23 1990KM b (103516) 671
Medium: 0.10 M Me4NNO3

C24H50N2O6 L CAS 85726-95-0 (646)
4,10-Dibutoxyethoxyethylidene-1,7-dioxo-4,10-diazacyclododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sol	non-aq	20°C	100%	C		K1=4.05	1983SLa (103529)	672

Medium: CHCl3

C25H22O2P2 L CAS 207-21-8 (2099)
Methylenebis(diphenylphosphine oxide); Ph2P(O)CH2P(O)Ph2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=5.6	1999ESa (103631)	673

In tetrahydrofuran; alkali metal is used as 2,4-dinitrophenolate

Li+	con	non-aq	25°C	100%	U		K1=4.73	1988YKa (103632)	674
-----	-----	--------	------	------	---	--	---------	------------------	-----

Medium: tetrahydrofuran

Li+	con	non-aq	25°C	100%	U		K1=4.9	1984YKa (103633)	675
-----	-----	--------	------	------	---	--	--------	------------------	-----

Medium: tetrahydrofuran + CHCl3 4:1, Li as 2,4-dinitrophenolate

Li+	oth	non-aq	22°C	100%	U		K1=2.5	1978YSa (103634)	676
-----	-----	--------	------	------	---	--	--------	------------------	-----

Medium: 1:1 v/v CH3CN:CHCl3 1:1 v/v. Li as LiCl; for LiI K1=2.3

Li+	con	non-aq	25°C	100%	U			1969SSi (103635)	677
-----	-----	--------	------	------	---	--	--	------------------	-----

K(LiI+L)=3.3

Medium: CH3CN

C25H26N4O5 HL CAS 207800-93-9 (8967)
19,20,22,23,25,26-Hexahydro-9-methyl-11,7-metheno-7H-dibenzotetraoxatetraazacyclotricosin-28-ol

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	diox/w	25°C	50%	C		K1=2.36	2001INa (103652)	678

Medium: 50% v/v dioxane/H2O, 3% v/v triethylamine, pH 12.

C25H28N4O4S+ L CAS 423763-92-2 (8996)
3-Ethyl-2-[4-(2,3,5,6,8,9-hexahydro-1,4,7,10-benzotetraoxacyclododecin-12-yl)buta-1,3-dienyl]benz

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=1.50	2002GVc (103660)	679

Medium: acetonitrile, 0.1 M Et4NClO4.

C25H29NO7 L FQC CAS 215095-38-8 (8804)
4'-(Dimethylamino)-2,7-(3,6,9-trioxaundecane-1,11-dioxy)flavone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	ns	100%	C		K1=1.51	2000LXa (103680)	680
Medium: acetonitrile. By fluorescence, K1=1.68.									

C25H37N2O7P		L					CAS 202407-79-2	(8035)	
26,27-Dimethoxy-3,7,24-triMe-11,14,17,20-tetraoxa-2,4-diaza-phosphatricycloheptacosahexaeneoxide;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	20°C	100%	C			1998DDc (103758)	681
							K(LiP+L)=3.48		
Medium: CHCl3. P is picrate.									

C25H50N2O4		L					(2317)		
N,N'-Diheptyl-N,N',5,5'-tetramethyl-3,7-dioxanonanediarnide;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	25°C	100%	U	I M		19800Ea (103827)	682
							K(Li(ClO4)+L)=6.0		
							K(Li(ClO4)L+L)=2.3		
Medium CH2Cl2. In CH3CN: K(Li(ClO4)+L)=3.0, K(Li(ClO4)L+L)=1.0. In pyridine: K(Li(ClO4)+L)=0.04. In MeCN: K(Li(ClO4)L+L)=1.23									

C25H50N4O8S		L					CAS 503465-06-3	(9249)	
4,7,15,18,24,27,32,35-Octaoxa-1,10,12,21-tetraazabicyclo[19.8.8]heptatriacontane-11-thione;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=2.84	2004KV a (103845)	683
Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.									

C26H20		L					CAS 2039-68-1	(1741)	
Tetraphenylethylene; (C6H5)2C:C(C6H5)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	20°C	100%	U	T	K1=4.27	1975LLa (103853)	684
Medium: THF. K1=4.24 (10 C); 4.20 (0 C); 4.18(-10 C); 4.10 (-30 C); 4.05 (-40 C); 4.02 (-50 C); 4.00 (-70 C)									

C26H21OP		L					CAS 33078-07-8	(2088)	
C-Phenylcarbonylmethylenetriphenylphosphorane; Ph3P:CHC(O)Ph									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo

Li+ con non-aq 25°C 100% U K1=2.42 1988YSb (103856) 685
Medium: acetonitrile

C26H24N4O5 L CAS 188838-26-8 (7359)

Dipyrido[3,2-a:2',3'-c]-phenazo-(1,4,7,10,13-pentaoxacyclopentadecane);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C I 2002YPb (103900) 686

K(CuLA2+Li)=3.37

Medium: MeCN, 0.10 M Bu4NPF6. By nmr, K=3.07. Also data for acetone/
0.01 M Bu4NPF6: K=1.97 (1.75 by nmr). A is triphenylphosphine.

Li+ sp non-aq 25°C 100% C I 2002YPb (103901) 687

K(ZnLA2+Li)=3.76

Medium: MeCN, 0.10 M Bu4NPF6. A is CH3.C6H4.SH

Li+ sp non-aq 25°C 100% U I M 1997YLa (103902) 688

K(Ru(II)(bpy)2L+Li)=3.45

Medium: CH3CN;0.1M NBu4PF6. In (CH3)2CO: K=1.64. Data also for
bis(4,4'-di-tert-butylbipyridyl) and bis(phenanthroline) RuL complexes.

C26H24O2P2 L (6648)

Bis(diphenylphosphinyl)ethane; (C6H5)2PO.CH2CH2.PO(C6H5)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.3 1990EAb (103912) 689

Medium: THF+CHCl3 4:1(vol). Metal as 2,4-dinitrophenolate

C26H24O3P2 L (7158)

1,3-Bis(diphenylphosphinyl)-2-oxopropane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C C K1=4.2 1999TEa (103919) 690

In: tetrahydrofuran/CHCl3 4:1 v/v

Li+ oth non-aq 25°C 100% U K1=4.2 1995TEa (103920) 691

Medium: tetrahydrofuran:CHCl3 4:1 (v/v).

Metal ion is used as 2,4-dinitrophenolate.

C26H28N2O5 L (2155)

1,13-Di-(8-quinolyl)-1,4,7,10,13-tetraoxatridecane; C9H6N.O.(CH2.CH2.O)4.C9H6N

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 27°C 100% C IH K1=4.3 1996TJa (103979) 692

K(LiL+Li)=2.45

Method: 7Li nmr. Medium: acetonitrile. Data for 27-67 C. DH(K1)=-16 kJ

mol-1, DS=30 J K-1 mol-1; DH(LiL+Li)=10, DS=81. Also data in nitromethane.

C26H34N4 L CAS 677034-80-9 (9063)

1-(2-{10-[2-Piperazinoethyl]-9-anthryl}ethyl)piperazine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=4.72 K(LiL+Li)=2.74	2003GHa (104075)	693

Method: fluorescence spectroscopy. Medium: acetonitrile, 0.05 M Et4NClO4.

C26H36N2O6 L DiBzCryptand222 (746)

5,6,14,15-Dibenzo-4,7,13,16,21,24-hexaoxa-1,10-diazabicyclo[8.8.8]hexacosan-5,14-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	U	IH		1988DSa (104137)	694

Medium: MeCN. DH(K1)=-33.0 k J mol-1. Also data in propylene carbonate.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	ISE	non-aq	25°C	100%	U	M	K1=6.06	1987DSa (104138)	695

Medium: acetonitrile

C26H40O11P2 L (5727)

1,7-Bis(2-(diethoxyphosphinylmethoxy)phenyl)-1,4,7-trioxaheptane;2(EtO)2PO.CH2OC6H4C2H4OC2H4)2O

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.7	1989EVa (104244)	696

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C26H45N3O6 L CAS 111928-04-2 (8968)

7-Phenyl-4,10,16,19,24,27-hexaoxa-1,7,13-triazabicyclo[11.8.8]nonacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	none	25°C	dil	C		K1=5.96 K(Li+A+L(org))=LiAL(org))=3.10	1987BBf (104280)	697

Method: extraction of metal picrate from H2O into CHCl3.

C26H48O6 L (2342)

19,21,26,28-Tetramethyl-2,5,8,11,14,17-hexaoxatricyclo[22.4.0.0(18,23)]octacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	24°C	100%	U	M	K(Li(picrate)+L)=5.1	1981BEb (104310)	698

Medium: CDCl3

C26H52N2O5 L Cryptand 221D CAS 62002-40-8 (8956)
5-Decyl-4,7,13,16,21-pentaoxa-1,10-diazabicyclo[8.8.5]tricosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	M	M	K1=5.91 K(LiL+ClO4)=1.14	1999DSd (104322)	699

Medium: acetonitrile.

C26H52N6O7S2 L CAS 503465-16-5 (9245)
4,12,20,26,29,34,37-Heptaoxa-1,7,9,15,17,23-hexaazabicyclo[21.8.8]nonatriacontane-8,16-dithione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=<2	2004KVa (104340)	700

Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.

C26H52N6O7S2 L CAS 503465-12-1 (9243)
9,12,15,26,29,34,37-Heptaoxa-1,4,6,18,20,23-hexaazabicyclo[21.8.8]nonatricontane-5,19-dithione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=<2	2004KVa (104350)	701

Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.

C26H54N2O10 L CAS 85726-99-4 (652)
4,13-Dimethyloxyethoxyethoxyethylidene-1,7,10,16-tetraoxy-4,13-diazaoctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sol	non-aq	20°C	100%	C		K1=4.03	1983SLa (104361)	702

Medium: CHCl3

C27H26O2P2 L (6811)
1,2-Bis(2-Diphenylphosphinyl)-1-methylethane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=4.0	1990EAb (104397)	703

Medium: THF+CHCl3 4:1(vol). Metal as 2,4-dinitrophenolate. Data also for 1,1-dimethyl, 1-hexyl, 1-heptyl, 1-octyl and 1-decyl analogues

C27H26O3P2 L (6812)
1,2-Bis(2-Diphenylphosphinyl)-1-hydroxymethylethane;
(C6H5)2PO.CH(CH2OH)CH2.PO(C6H5)2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C 100% U K1=4.2 1990EAb (104402) 704
 Medium: THF+CHCl3 4:1(vol). Metal as 2,4-dinitrophenolate. Data also for
 3-hydroxypropyl analogue

C27H26O3P2 L (7159)
 1,4-Bis(diphenylphosphinyl)-2-oxobutane;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth non-aq 25°C 100% U K1=4.9 1995TEa (104407) 705
 Medium: tetrahydrofuran:CHCl3 4:1 (v/v).
 Metal ion is used as 2,4-dinitrophenolate.

C27H32N05S+ L CAS 423763-94-4 (8997)
 3-Ethyl-2-[4-(2,3,5,6,8,9,11,12-octahydro-1,4,7,10,13-benzopentaoxacyclopentadecin-
 15-yl)butadien

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=4.16 2002GVc (104517) 706
 Medium: acetonitrile, 0.1 M Et4NClO4.

C27H33N07 L FLC CAS 223390-37-2 (8805)
 2-[4-Dimethylaminophenyl]-6-methyl-3-(1,4,7,10-tetraoxacyclododec-2-ylmethoxy)-4H-1
 -Benzopyran-4;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq ns 100% C K1=3.02 2000LXa (104525) 707
 Medium: acetonitrile. Method: fluorescence spectroscopy.

C27H47N3O6 L (8029)
 Tripodal ionophore 3;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C 2001LFa (104625) 708
 K(LiP+L=LiPL)=5.11

Method: Analyses by spectrophotometry. Medium: chloroform. P is picrate.

C28H24N2O4 L (5742)
 5,6-Benzo-1,10-di(8-quinolyl)-1,4,7,10-tetraoxadecane;
 C9H6N.O.C2H4.O.C6H4.O.C2H4.O.C9H6N

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.4 1989BEa (104676) 709
 Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C28H28O3P2 L (6815)

1,5-Bis(diphenylphosphinyl)-3-oxapentane; O(CH₂.CH₂.PO(C₆H₅)₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=5.15	1993EBa (104712)	710

Medium: CH₃CN

Li+	con	non-aq	25°C	100%	U			K1=5.6	1993Eva (104713)	711
-----	-----	--------	------	------	---	--	--	--------	------------------	-----

Medium: THF+CHCl₃ (4:1 vol). Also data for other solvents

Li+	con	non-aq	25°C	100%	U			K1=5.3	1992BEa (104714)	712
-----	-----	--------	------	------	---	--	--	--------	------------------	-----

Medium: THF+CHCl₃ (4:1 vol)

C28H28O4P2 L (7891)

1,6-Bis(diphenylphosphinyl)-2,5-dioxohexane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C		C			K1=5.7	1999TEa (104722)	713

In: tetrahydrofuran/CHCl₃ 4:1 v/v

C28H30N2O2P2 L CAS 68745-29-9 (5707)

N,N'-Bis(diphenylphosphinylmethyl)-1,2-diaminoethane; ((C₆H₅)₂PO.CH₂.NH.CH₂-)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=4.4	1984YKa (104727)	714

Medium: tetrahydrofuran + CHCl₃ 4:1, Li as 2,4-dinitrophenolate

C28H32N2O6 L (5743)

1,16-Di(8-quinolyl)-1,4,7,10,13,16-hexaoxahexadecane; C₉H₆N.O.(C₂H₄O)₅.C₉H₆N

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=4.4	1989BEa (104750)	715

Medium: tetrahydrofuran/CHCl₃ 4:1 (volume)

C28H35O7P L CAS 90275-27-7 (2068)

Adamantylphosphonyldibenzo-17-crown-6

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sol	non-aq	25°C	100%	U			K1=3.86	1987TCa (104768)	716

Medium: CH₂Cl₂, 2% MeCN. Metal as picrate

C28H36N2O7S2 HL CAS 150196-54-6 (7735)

3-(3-Sulfopropyl)-2-[4-[N-(1,4,7,10,13-pentaoxa-16-azacyclooctadeca)]]styryl-benzot hiazolium;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+ sp non-aq 18°C 100% C K1=2.2 1997LHa (104784) 717
Medium: acetonitrile.

C28H4008 L AN(MOE0EOM)2AN (2243)
29,30-Dimethoxy-13,27-dimethyl-3,6,9,17,20,23-hexaoxatricyclo-tricon-1,11,13,15,2
5,27-hexaene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H 1979KLa (104857) 718
K(Li(picrate)+L)=2.75

Medium: CHCl3

C28H40010 L DiBz-30-crown10 CAS 104946-67-0 (1776)
2,3:17,18-Dibenzo-1,4,7,10,13,16,19,22,25,28-decaoxacyclotriacont-2,17-diene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U I K1=4.68 1991ASb (104892) 719
Medium: 1,2-dichloroethane. In nitromethane: K1=4.49

Li+ vlt non-aq 25°C 100% U K1=14.2 1990SPa (104893) 720
Medium: 1,2-dichloroethane

C28H44012P2 L (5728)
1,10-Bis(2-(diethoxyphosphinylmethoxy)phenyl)-1,4,7,10-tetraoxadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.0 1989EVa (104946) 721
Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C28H5205 L (2339)
16,16,18,18,23,23,25,25-Octamethyl-2,5,8,11,14-pentaoxatricyclo(22.4.0.0(15,20))pen
tacosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 24°C 100% U M 1981BEb (105010) 722
K(Li(picrate)+L)=3.9

Medium: CDCl3

C28H5206 L (5352)
Di(t-butylcyclohexyl)-18-crown-6

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth oth/un 25°C dil U K1=<0.9 1970MSa (105016) 723

C28H56N2O6 L Cryptand 222D CAS 69878-46-2 (8957)
5-Decyl-4,7,13,16,21,24-hexaoxa-1,10-diazabicyclo[8.8.8]hexacosane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	M	M	K1=5.28 K(LiL+ClO4)=1.20	1999DSd (105030)	724

Medium: acetonitrile.

C28H56N6O8S2 L CAS 503465-18-7 (9246)
4,12,15,23,29,32,37,40-Octaoxa-1,7,9,18,20,26-hexaazabicyclo[24.8.8]dotetracontane-8,19-dithione;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=<2	2004KV a (105041)	725

Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.

C28H56N6O8S2 L CAS 503465-14-3 (9244)
9,12,15,18,29,32,37,40-Octaoxa-1,4,6,21,23,26-hexaazabicyclo[24.8.8]dotetratricontane-5,22-dithio

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	95%	C		K1=<2 B2= 6.03	2004KV a (105051)	726

Medium: 95% MeOH/H2O, 0.01 M Et4NClO4.

C29H30O3P2 L CAS 176849-77-7 (7160)
1,6-Bis(diphenylphosphinyl)-2-oxohexane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	oth	non-aq	25°C	100%	U		K1=4.8	1995TEa (105080)	727

Medium: tetrahydrofurane:CHCl3 4:1 (v/v).
Metal ion is used as 2,4-dinitrophenolate.

C29H30O3P2 L CAS 176849-78-8 (7161)
1,6-Bis(diphenylphosphinyl)-3-oxohexane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	oth	non-aq	25°C	100%	U		K1=4.8	1995TEa (105085)	728

Medium: tetrahydrofurane:CHCl3 4:1 (v/v).
Metal ion is used as 2,4-dinitrophenolate.

C29H30O4P2 L (7897)
1,7-Bis(diphenylphosphinyl)-2,6-dioxoheptane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C C K1=5.8 1999TEa (105090) 729
In: tetrahydrofuran/CHCl3 4:1 v/v

C29H35N05 L CAS 201154-06-5 (7825)

N-(1-Pyrenylmethyl)-1,4,7,10,13-pentaoxa-16-azacyclooctadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp mixed 25°C 90% C 1997KKa (105101) 730

K(LiSCN+L)=1.68

Method: fluorescence emission. Medium: MeOH/CHCl3 (9:1 v/v).

C29H36N06S+ L CAS 423763-96-6 (8998)

2-[4-(2,3,5,6,8,9,11,12,14,15-Decahydro-1,4,7,10,13,16-benzohexaoxacyclooctadecin-1
8-yl)butadien

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=1.98 2002GVc (105106) 731

Medium: acetonitrile, 0.1 M Et4NClO4.

C30H32O4P2 L (6816)

1,8-Bis(diphenylphosphinyl)-3,6-dioxaoctane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=5.50 1993EBa (105229) 732

Medium: CH3CN. Data also for 3,5,8-trioxa, 3,5,8,11-tetraoxa and 3,5,8,11-pe
ntaoxa analogues

Li+ con non-aq 25°C 100% U K1=5.1 1992BEa (105230) 733

Medium: THF+CHCl3 (4:1 vol)

C30H32O5P2 L (7892)

1,9-Bis(diphenylphosphinyl)-2,5,8-trioxononane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C C K1=5.1 1999TEa (105236) 734

In: tetrahydrofuran/CHCl3 4:1 v/v

C30H34N2O2P2 L CAS 68743-31-3 (2066)

Diaminoethane-N,N'-di-2-ethyldiphenylphosphine oxide; (CH2.NH.C2H4.P(O)(C6H5)2)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.77 1986STb (105241) 735

Medium: THF:CHCl3 4:1 v/v. M as 2,4-dinitrophenolate

C30H36N8O3 Furan-cryptand CAS 121954-37-8 (7451)

39,40,41-Trioxa-1,4,11,14,17,24,29,36-octaazapentacyclo[12.12.12.1.1.1]henLetetracododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	sp	non-aq	25°C	100%	U		K1=3.91	1996AAb (105255)	736
Medium: MeCN. L = 39,40,41-Trioxa-1,,4,11,14,17,24,29,36-octaazapentacyclo[12.12.12.1(6,9).1(19,22).1(31,34]hentetetraconta-4,6,8....dodecaene									

C30H36O6 L ANANAN(MOE)20 (2239)
2,3,4,5,6,7,8,9,10-Tri(1,3-(2-methoxy-5-methylbenzo))-12,15,18-trioxacyclooctadeca-2,5,8-triene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	dis	non-aq	25°C	100%	U	H		1979KLa (105261)	737
K(Li(picrate)+L)=5.25									

Medium: CHCl3

C30H37N5O7	HL						CAS 552856-74-3	(8846)	
7-[2-Methoxy-4-[(4-nitrophenyl)azo]phenyl]-13-(2-methoxyphenyl)-1,4,10-trioxa-7,13-diazacyclopent;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	sp	alc/w	RT	50%	C		K1=1.8	2002GLb (105267)	738
Medium: 50% MeOH/H2O, pH 7.4 (0.1 M Tris buffer), 0.1 M Me4NCl.									

C30H38N2O4	L						(5828)		
Trimethoxyphenylcryptand 3,1.									
25,26,27-Trimethoxy-5,10,15-trimethyl-22-oxa-1,19-diazatetra-									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	nmr	non-aq	25°C	100%	U		K1=<4.94	1986CHc (105273)	739
In CDCl3. L=25,26,27-Trimethoxy-5,10,15-trimethyl-22-oxa-1,19-diazatetracyclo[24.1(3,7).1(8,12).1(13,17)]heptacosa-3,5,7,8,10,12,13,15,17-nonaene									

C30H38N2O8	L						CAS 137571-97-2	(6821)	
Anthraquinone[2.2]cryptand;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+	nmr	non-aq	21°C	100%	U		K1=4.61 B2=6.83	1992CSc (105278)	740
Method:NMR. Medium:CD3CN									

C30H42O10P4	L						CAS 97910-31-1	(2083)	
Tris-((2-(dimethylphosphinylmethoxy)phenoxy)methyl)phosphine oxide;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	-------------	-----------	--------

Li+ con non-aq 25°C 100% U K1=4.49 1989KSa (105302) 741
Medium: tetrahydrofuran/CHCl3 4:1 (vol)

C30H48O13P2 L CAS 112120-14-6 (5729)
1,13-Bis(2-(diethoxyphosphinylmethoxy)phenyl)-1,4,7,10,13-pentaoxatridecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.0 1989EVa (105344) 742
Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C31H34O4P2 L (7157)
1,9-Bis(diphenylphosphinyl)-3,7-dioxononane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth non-aq 25°C 100% U K1=4.7 1995TEa (105526) 743
Medium: THF:CHCl3 4:1 v/v. Li as 2,4-dinitrophenolate. Also other si
milar ligands

C32H28O4P2 L CAS 88928-04-5 (2072)
1,2-Dihydroxybenzene bis(diphenylphosphinylmethyl) ether

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C C K1=4.7 1999TEa (105576) 744
In: tetrahydrofurane/CHCl3 4:1 v/v

Li+ con non-aq 25°C 100% U K1=4.40 1989KSa (105577) 745
Medium: tetrahydrofuran/CHCl3 4:1 (vol)

C32H29O3P3 L CAS 21851-89-8 (2640)
P,P,P',P'',P'''-Pentaphenyldimethylenetri(phosphineoxide); (Ph2P(O)CH2)2P(O)Ph

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% U M 1981SPb (105583) 746
K(LiI+L)=2.69

Medium: CH3CN

C32H36O5P2 L CAS 137728-07-5 (6837)
1,11-Bis(diphenylphosphinyl)-3,6,9-trioxaundecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=5.4 1992BEa (105646) 747
Medium: THF+CHCl3 (4:1 vol)

C32H36O6P2 L (7893)

1,12-Bis(diphenylphosphinyl)-2,5,8,11-tetraoxododecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C		C			K1=4.8	1999TEa (105651)	748
-----	-----	--------	------	--	---	--	--	--------	------------------	-----

In: tetrahydrofuran/CHCl3 4:1 v/v

C32H43N2O7S HL CAS 189057-31-6 (7756)
3-(4-Carboxybutyl)-2-[4-[N-(1,4,7,10,13-pentaoxa-16-azacyclooctadeca)]]styryl-benzo
thiazolium;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	18°C	100%	C			K1=1.9	1997LHa (105757)	749
-----	----	--------	------	------	---	--	--	--------	------------------	-----

Medium: acetonitrile.

C32H44O12P2 L CAS 112120-16-8 (5738)
3,4:9,10:15,16-Tribenzo-1,18-di(diethoxyphosphinyl)-2,5,8,11,14,17-hexaoxaoctadeca-
3.9.15-triene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C	100%	U			K1=3.7	1989BEa (105777)	750
-----	-----	--------	------	------	---	--	--	--------	------------------	-----

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C32H48N2O3 L CAS 170801-55-5 (8952)
1,5-Bis[2,2'-azo-4,4'-(1,1,3,3-tetramethylbutyl)phenoxy]-3-oxapentane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	RT	100%	C			K1=3.28	2000GDa (105795)	751
-----	----	--------	----	------	---	--	--	---------	------------------	-----

Medium: acetonitrile.

C32H52O14P2 L CAS 112120-15-7 (5730)
1,13-Bis(2-(diethoxyphosphinylmethoxy)phenyl)-1,4,7,10,13,16-hexaoxahexadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C	100%	U			K1=3.7	1989EVa (105824)	752
-----	-----	--------	------	------	---	--	--	--------	------------------	-----

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C33H28O2P2 L CAS 118448-50-3 (2085)
C-Methylcarbonyl,C-diphenylphosphinylmethylenetriphenylphosphorane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C	100%	U			K1=2.98	1988YSb (105871)	753
-----	-----	--------	------	------	---	--	--	---------	------------------	-----

Medium: acetonitrile

C33H39N11 L Pyr-cryptand CAS 141258-00-6 (7452)

1,4,12,15,18,26,31,39,42,43,44-Undecaazapentacyclo[13.13.13.1.1.1]tetratetraconta pentadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	U			K1=2.36	1996AAb (105918)	754
Medium: CH3CN. L = 11,4,12,15,18,26,31,39,42,43,44-undecazapentacyclo[13.13.13.1(6,10).1(20,24).1(33,37)]tetratetraconta-4-6-8-10(44),11...pentadecaene										

C33H41N3O6		L						(8027)		
Tripodal ionophore ;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C				2001LFa (105924)	755
K(LiP+L=LiPL)=5.37										
Method: Analyses by spectrophotometry. Medium: chloroform. P is picrate.										

C34H36N4O10		H4L		CCE				(7373)		
N,N'-Bis(2-hydroxy-5-nitrobenzyl)4,13-diazadibenzo-18-crown-6;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	alc/w	25°C	70%	U			K1=9.75 B2=17.90	1995VZa (106008)	756
K3=6.20										
K4=3.60										

Medium: 70% MeOH

C34H38O12P2		L						(6906)		
1,2:10,11:15,16:24,25-Tetrabenzo-13,27-di(methylphospha)-3,6,9,12,14,17,20,23,27,28-10-crown-28										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	oth	non-aq	22°C	100%	U			K1=1.9	1978YSa (106040)	757
Medium: 1:1 v/v EtOH+CHCl3. Li as acetate salt										

C34H40O6P2		L						CAS 137728-08-6 (6838)		
1,14-Bis(diphenylphosphinyl)-3,5,8,11-tetraoxatetradecane;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=5.6	1992BEa (106044)	758
Medium: THF+CHCl3 (4:1 vol)										

C34H40O7P2		L						(7894)		
1,15-Bis(diphenylphosphinyl)-2,5,8,11,14-pentaoxopentadecane;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+ con non-aq 25°C C K1=4.7 1999TEa (106051) 759
In: tetrahydrofuran/CHCl3 4:1 v/v

C34H44N2O5 L CAS 101671-92-5 (5825)
Trimethoxyphenylcryptand 3,1,1.
30,31,32-Trimethoxy-5,10,15-trimethyl-22,27-dioxo-1,9-diaza....

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 25°C 100% U K1=13.79 1986CHc (106069) 760
CDCl3. L=30,31,32-Trimethoxy-5,10,15-trimethyl-22,27-dioxa-1,19-diazapentacyclo[17.5.5.1(3,7).1(8,12).1(13,17)]dotriaconta-3,5,7,8,10,12,13,15,17nonaene

C34H53O8Br H2L CAS 38784-08-6 (2336)
5-Bromolasalacid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl alc/w 25°C 100% M H K(Li+HL)=1.8 1988PJ a (106099) 761
Medium: MeOH. DH = 8.2 kJ mol⁻¹; DS = 62

C34H54O8 H2L Lasalacid CAS 25999-20-6 (2335)
Lasalacid acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ nmr non-aq 20°C 100% C K(Li+HL)=0.0 1998MLa (106142) 762
Medium: CD3OD. Method: 13C nmr.

Li+ dis oth/un 25°C 0.0 U K1=2.2 1992LPb (106143) 763

Li+ gl alc/w 25°C 100% M H K(Li+HL)=1.9 1988PJ a (106144) 764
Medium: MeOH. DH = 4.9 kJ mol⁻¹; DS = 53

Li+ gl alc/w 25°C 100% U K(Li+2HL)=1.44 1982BDc (106145) 765

Medium: MeOH

C35H45N9 L CAS 312304-65-7 (7962)
29,32,35-TriMe-1,14,29,32,35,38,39,40,41-Nonaazahexacyclohentetraconta-3,5,7,8,10,12,16,18,20,21,

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl R4N.X 25°C 0.10M U K1=3.4 2001BBa (106203) 766
K(LiL+H)=9.5
K(LiHL+H)=9.3

Medium: 0.10 M NMe4NO3.

C36H32N2O6 L (5744)
5,6:11,12-Dibenzo-1,16-di(8-quinolyl)-1,4,7,10,13,16-hexaoxahehexadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.1 1989BEa (106220) 767
Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C36H36N24O12 L Cucurbituril CAS 283175-97-3 (6744)
Cucurbit[6]uril;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal mixed 25°C 50% C H K1=2.38 1998BJb (106263) 768
Medium: 50% (v/v) HCOOH/H2O. DH(K1)=-3.4 kJ mol⁻¹

Li+ sp none 25°C 0 U K1=2.23 B2=2.73 1994HKA (106264) 769

C36H36O4P2 L (2073)
3-t-Butyl-1,2-dihydroxybenzene bis(diphenylphosphinylmethyl) ether

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.25 1989KSA (106281) 770
Medium: tetrahydrofuran/CHCl3 4:1 (vol)

C36H36O6P2 L CAS 103990-64-3 (2077)
1,2-Bis(2-(diphenylphosphinylmethoxy)ethoxy)benzol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=4.20 1989KSA (106285) 771
Medium: tetrahydrofuran/CHCl3 4:1 (vol)

C36H40O4S2 L ANAN(MSM)2ANAN CAS 1129-04-9 (2240)
Tetra(1,3-(2-methoxy-5-methylbenzo))-9,18-dithiacyclooctadeca-2,5,12,14-tetraene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H 1979KLa (106295) 772
K(Li(picrate)+L)=2.96

Medium: CHCl3

C36H40O6 L ANANAN(MOM)2AN CAS 1129-07-2 (2238)
Tetra(1,3-(2-methoxy-5-methylbenzo))-12,18-dioxacyclooctadeca-2,5,8,14-tetraene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U H 1979KLa (106301) 773
K(Li(picrate)+L)=2.91

Medium: CHCl3

C36H4407P2 L (5725)
1,17-Di(diphenylphosphinyl))-3,6,9,12,15-pentaoxaseptadecane;
Ph2PO.C2H4(0.C2H4)4OC2H4POPh2

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.5	1992BEa (106336)	774

Medium: THF+CHCl3 (4:1 vol)

Li+	cal	non-aq	25°C	100%	U		K1=3.5 K(Li+LiL)=2.70	1991SGa (106337)	775
-----	-----	--------	------	------	---	--	--------------------------	------------------	-----

C36H4408P2 L (7895)
1,18-Bis(diphenylphosphinyl)-hexaoxooctadecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=4.6	1999TEa (106345)	776

In: tetrahydrofurane/CHCl3 4:1 v/v

C36H47N3O6 L (8028)
Tripodal ionophore 2;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K(LiP+L=LiPL)=5.21	2001LFa (106375)	777

Method: Analyses by spectrophotometry. Medium: chloroform. P is picrate.

C36H48N2O6 L CAS 101695-36-7 (5826)
Trimethoxyphenylcryptand 3,2,1.
33,34,35-Trimethoxy-5,10,15-trimethyl-22,25,30-trioxa-1,19-diaza-

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	25°C	100%	U		K1=9.79	1986CHc (106379)	778

In CDCl3. L=33,34,35-trimethoxy-5,10,15-trimethyl-22,25,30-trioxa-1,19-diaza pentacyclo[17.8.5.1(3,7).1(8,12).1(13,17)]pentatriaconta-3,5,7,8,...nonaene

C36H52O14P2 L (5739)
3,4:12,13:21,22-Tribenzo-1,24-di(diethoxyphosphinyl)-2,5,8,11,14,17,20,23-octaoxate tracosatriene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.9	1989BEa (106397)	779

Medium: tetrahydrofuran/CHCl₃ 4:1 (volume)

C36H58N10O10S4 H5L CAS 136685-24-0 (6875)

(1-Cys-,1'-Cys,4-Cys-,4'-Cys)-dithiobis(Ac-1-Cys-Pro-D-Val-4-Cys-NH₂);

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	gl	non-aq	20°C	100%	U			K1=2.60	1993EAa (106442)	780

Method: circular dichroism. Medium: MeCN, ClO₄-

C36H62O11 HL Monensin CAS 17090-79-8 (737)

Monensin, 1,6-dioxaspiro[4,5]decane derivative;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C			K1=4.4	1997PBb (106511)	781

Medium: acetonitrile. Additional method: potentiometry with ISE.

By calorimetry, DH(K1)=-31 kJ mol⁻¹, DS(K1)=-21 J K⁻¹ mol⁻¹.

Li+	vlt	non-aq	25°C	100%	C	I		K1=12.1	1997WRa (106512)	782
Method: cyclic voltammetry. Medium: acetonitrile, 0.05 M Et ₄ NClO ₄ . In DMSO K1=3.8; in acetone, K1=11.0; in hexamethylphosphoric triamide, K1<1.										

Li+	vlt	non-aq	23°C	100%	U	I		K1=12.1	1994FRa (106513)	783
Medium: MeCN. In PrCN: K1=12.2; acetone: 11.0; DMF: 6.0; Me-pyrrol.: 4.7; NN-DMA: 4.3; DMSO: 3.8; Di-Et-formamide: 3.5; Di-Et-acetamide: 2.8; PC: 11.5										

Li+	ISE	alc/w	25°C	100%	M			K1=3.60	1984CTa (106514)	784
Medium: MeOH										

Li+	ISE	non-aq	25°C	100%	M			K1=5.90	1984CTa (106515)	785
Medium: N,N-dimethylformamide. In DMSO K1=3.71										

Li+	ISE	alc/w	25°C	100%	U			K1=5.35	1984CTb (106516)	786
Medium: EtOH										

Li+	vlt	alc/w	25°C	100%	U			K1=3.3	1978HPa (106517)	787
Method: Cyclic voltametry										

C38H30O2P2 L CAS 118448-51-4 (2086)

C-Phenylcarbonyl,C-diphenylphosphinylmethylenetriphenylphosphorane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=2.56	1988YSb (106640)	788

Medium: acetonitrile

C38H32O3P2 L (6804)

1,3-Bis(2-Diphenylphosphinylphenyl)-2-oxapropane; O(CH₂.C₆H₄(PO.(C₆H₅)₂)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.5	1993BEb (106643)	789
Medium: THF+CHCl3 4:1(vol)									

C38H32O4P2		L					(1320)		
1,4-Di(2-diphenylphosphinylphenyl)-1,4-dioxabutane; Ph2PO.C6H4.O.CH2.CH2.O.C6H4.P(O)Ph2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.8	1991EBa (106649)	790
Medium: THF+CHCl3 4:1(vol)									

C38H40O6P2		L					(6833)		
1,2-Bis(2-(2-(diphenylphosphinyl)ethoxy)ethoxy)benzene; C6H4(OCH2CH2OCH2CH2PO(C6H5)2)2									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.5	1993EVa (106660)	791
Medium: THF+CHCl3 (4:1 vol). Also data for other solvents									

C38H48O8P2		L					CAS 145864-37-5 (6839)		
1,20-Bis(diphenylphosphinyl)-3,5,8,11,14,17-hexaoxaicosane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.0	1992BEa (106681)	792
Medium: THF+CHCl3 (4:1 vol)									

C38H48O9P2		L					(7896)		
1,21-Bis(diphenylphosphinyl)-2,5,8,11,14,17,20-heptaioxoheneicicosane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C		C		K1=4.6	1999TEa (106686)	793
In: tetrahydrofuran/CHCl3 4:1 v/v									

C38H52N2O7		L					CAS 101671-93-6 (5827)		
Trimethoxyphenylcryptand 3,2,2. 36,37,38-Trimethoxy-5,10,15-trimethyl-22,25,30,33-tetraoxa-1,19-									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	nmr	non-aq	25°C	100%	U		K1=7.26	1986CHc (106691)	794
CDCl3. L=36,37,38-trimethoxy-5,10,15-trimethyl-22,25,30,33-tetraoxa-1,19-dia zapentacyclo[17.8.8.1(3,7).1(8,12).1(13,17)]octatriaconta-3,5,7,8...nonaene									

C40H36O4P2		L					(6805)		

1,6-Bis(2-Diphenylphosphinylphenyl)-2,5-dioxahexane; (CH₂.O.CH₂.C₆H₄(PO(6H₅)₂)₂)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C	100%	U			K1=5.2	1993BEb (106734)	795
-----	-----	--------	------	------	---	--	--	--------	------------------	-----

Medium: THF+CHCl₃ 4:1(vol)

C₄₀H₃₆O₅P₂ L CAS 86341-96-0 (5724)
1,7-Di(2-diphenylphosphinyl)phenyl-1,4,7-trioxaheptane; Ph₂PO.C₆H₄.O.C₂H₄.O.C₂H₄.O.C₆H₄.POPh₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C	100%	U			K1=4.6	1991EBa (106746)	796
-----	-----	--------	------	------	---	--	--	--------	------------------	-----

Medium: THF+CHCl₃ 4:1(vol). Data also for 1,4,7,10-tetraoxa,1,4,7,10,13-pentaoxa and 1,4,7,10,13,16-hexaoxa and 4-tributyl analogues

C₄₀H₄₄O₄P₂ L (2074)
3,5-Di(t-butyl)-1,2-dihydroxybenzene bis(diphenylphosphinylmethyl)ether

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	con	non-aq	25°C	100%	U			K1=4.72	1989KSa (106765)	797
-----	-----	--------	------	------	---	--	--	---------	------------------	-----

Medium: tetrahydrofuran/CHCl₃ 4:1 (vol)

C₄₀H₄₆O₇ L CAS 177723-37-4 (8912)
25,27-Diethoxycalix[4]arenecrown-5, 1,3-alternate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	dis	non-aq	22°C	100%	C	M			1996CPa (106772)	798
-----	-----	--------	------	------	---	---	--	--	------------------	-----

K(LiA+L(org))=LiAL(org))=4.93

Medium: CHCl₃ saturated with H₂O. Method: extraction of LiA into CHCl₃/L solution. HA is picric acid. For the cone conformation, K=4.74.

C₄₀H₄₆O₈ L CAS 161282-95-7 (8680)
25,27-Dimethoxycalix[4]arene-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	25°C	100%	C			K1=<=1	1995CUa (106777)	799
-----	----	--------	------	------	---	--	--	--------	------------------	-----

Medium: methanol, 0.01 M Et₄NCl.

C₄₀H₄₈O₈ L AN2DP(OEOEO)₂E (2235)
3,4,5,6-Bis(3-methyl-5-(2-methoxy-5-methylbenzo))-2,7,10,13,16,19-hexaoxacyclodocosa-3,5-diene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	dis	non-aq	25°C	100%	U	H			1979KLa (106795)	800
-----	-----	--------	------	------	---	---	--	--	------------------	-----

K(Li(picrate)+L)=5.01

Medium: CHCl₃

C40H50N20010 L CAS 143902-45-8 (8935)

Decamethylcucurbit[5]uril;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal mixed 25°C 50% C H K1=1.99 2000ZKb (106807) 801

Medium: 50% v/v formic acid/H₂O. DH(K1)=-14.4 kJ mol⁻¹, DS(K1)=-10 J K⁻¹ mol⁻¹.

C40H52N404 L CAS 205066-94-0 (8760)

Tetraphenyl-1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraethanol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ ISE non-aq 25°C 100% C K1=3.13 1998WLc (106823) 802

Medium: DMF, 0.05 M Et₄NClO₄.

Ligand is (all-R)-(all-α)-Tetraphenyl-

C40H52O14P2 L CAS 127832-94-4 (5740)

2,3:9,10:15,16:21-Tetrabenzo-1,24-di(diethoxyphosphinyl)-2,5,8,11,14,17,20,23-octaoxatetracosane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=3.6 1989BEa (106828) 803

Medium: tetrahydrofuran/CHCl₃ 4:1 (volume)

C41H42O6 L CAS 151832-07-4 (6874)

9-(Dimethylethyl)-29,30,31,32,33-pentamethoxy-23-oxahexacyclotritriacontapentadecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ dis non-aq 25°C 100% U 1993HSa (106871) 804

K(Li(picrate)+L)=9.38

Medium: CDCl₃ saturated with D₂O. With 23-thia- analogue K=7.96

C42H40O4P2 L (7153)

1,2-Bis(2-(2-(diphenylphosphinyl)ethyl)phenoxy)ethane

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ oth non-aq 25°C 100% U K1=5.2 1995TEa (106912) 805

Medium: THF:CHCl₃ 4:1 v/v. Li as 2,4-dinitrophenolate

C42H40O4P2 L (6809)

1,6-Bis(2-Diphenylphosphinylphenyl)-3,4-dimethyl-2,5-dioxahexane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=4.9	1993BEb (106917)	806
Medium: THF+CHCl3 4:1(vol)									

C42H40O5P2		L					CAS 163172-12-6	(2080)	
Bis((2-diphenylphosphinylmethyl)phenyl)diethyleneglycol ether;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.3	1993BEb (106927)	807
Medium: THF+CHCl3 4:1(vol)									

Li+	con	non-aq	25°C	100%	U		K1=4.31	1989KSa (106928)	808
Medium: tetrahydrofuran/CHCl3 4:1 (vol)									

C42H40O7P2		L					CAS 95651-36-8	(2079)	
1,7-Di(2-(diphenylphosphinylmethoxy)phenyl)-1,4,7-trioxaheptane;									
(Ph2PO.CH2.O.C6H4.O.C2H4)2O									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.65	1989KSa (106937)	809
Medium: tetrahydrofuran/CHCl3 4:1 (vol)									

Li+	con	non-aq	25°C	100%	U		K1=3.65	1989TKb (106938)	810
Medium: tetrahydrofuran/CHCl3 4:1 (volume)									

C42H50O7		L					CAS 177723-38-5	(8793)	
1,3-Diisopropoxycalix[4]arene-crown-5, 1,3-alternate;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	22°C	100%	C	M		1996CPa (106952)	811
K(LiA+L(org))=LiAl(org))=4.78									
Medium: CHCl3 saturated with H2O. Method: extraction of LiA into CHCl3/L									
solution. HA is picric acid. For the cone conformation, K=4.70.									

C42H68N2O4		L					CAS 188593-77-3	(8954)	
2,17-Didodecyl-6,7,9,10,12,13-hexahydro-dibenzo[b,f][1,8,11,14,4,5]tetraoxadiazacyclohexadecine									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	RT	100%	C		K1=4.0	2000GDa (106974)	812
Medium: acetonitrile.									

C43H42O4P2		L					(7156)		
1,3-Bis((2-diphenylphosphinyl)phenoxy)propane;									

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+		oth non-aq	25°C	100%	U		K1=4.4	1995TEa (107000)	813
Medium: THF:CHCl3 4:1 v/v. Li as 2,4-dinitrophenolate. Also other similar ligands									

C43H42O6P2		L					(5734)		
1,7-Di((2-diphenylphosphynylmethoxy)phenyl)-1,7-dioxahепtane; (Ph2PO.CH2O.C6H4.O.C2H4)2CH2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+		con non-aq	25°C	100%	U		K1=3.46	1989TKb (107005)	814
Medium: tetrahydrofuran/CHCl3 4:1 (volume)									

C43H43NO4P2		HL					(8538)		
Methyl[bis-(2-diphenylphosphorylmethyl)phenoxyethyl]amine;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+		cal non-aq	25°C	100%	U	H	K1=4.56	1998SBb (107007)	815
Medium: MeCN Calorimetric titration of LiNCS. DH(K1)=-17.5 kJ mol-1									

C44H22N4O12Br8S4		H6L					CAS 176173-80-1 (6959)		
2,3,7,8,12,13,17,18-Octabromo-5,10,15,20-tetrakis(4-sulfonatophenyl)porphyrin;									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	NaNO3	25°C	0.1M	C			1996TNa (107039)	816
							K(Li+H2L=LiL+2H)=-18.81		

C44H30N8Br8		L					(7212)		
2,3,7,8,12,13,17,18-Octabromo-5,10,15,20-tetrakis(N-methylpyridinium-4-yl)porphin(+++);									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	NaNO3	25°C	0.1M	C		K1=4.21	1998TNa (107086)	817
							K(Li+HL=LiL+H)=-8.80		
Li+	sp	oth/un	25°C	0.10M	C			1996RHb (107087)	818
							K1eff=2.98		

C44H36O4P2		L					(6810)		
1,2-Bis(2-Diphenylphosphinylphenylmethoxy)benzene; C6H4(OCH2.C6H4(PO(C6H5)2)2									
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+		con non-aq	25°C	100%	U		K1=4.3	1993BEb (107091)	819

Medium: THF+CHCl3 4:1(vol)

C44H38N8 H2L CAS 48242-70-2 (6629)

5,10,15,20-Tetrakis(1-methylpyridinium-4-yl)porphine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	NaNO3	25°C	0.50M	C		K1=2.58	1998IHb (107106)	820

For the 2-pyridyl analogue, K1=3.28

C44H42O6P2 L (6806)

1,12-Bis(2-Diphenylphosphinylphenyl)-2,5,8,11-tetraoxadodecane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=5.2	1993BEb (107110)	821

Medium: THF+CHCl3 4:1(vol)

C44H44O5P2 L (5735)

1,7-Di((2-diphenylphosphinylmethoxy)phenyl)-4-oxaheptane; (Ph2PO.CH2O.C6H4.C3H6)20

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.18	1989TKb (107114)	822

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C44H44O5P2 L (5733)

1,7-Di(2-(diphenylphosphinylethyl)phenyl)-1,4,7-trioxaheptane;
(Ph2PO.C2H2.C6H4.OC2H4)20

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	oth	non-aq	25°C	100%	U		K1=4.6	1995TEa (107119)	823

Medium: THF:CHCl3 4:1 v/v. Li as 2,4-dinitrophenolate

Li+	con	non-aq	25°C	100%	U		K1=4.05	1989TKb (107120)	824
-----	-----	--------	------	------	---	--	---------	------------------	-----

Medium: tetrahydrofuran/CHCl3 4:1 (volume)

C44H44O6P2 L CAS 126763-09-5 (7790)

1,8-Bis[2-(diphenylphosphinylmethyl)phenoxy]-3,6-dioxaoctane;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	U	H	K1=4.90	1998SBb (107128)	825

Medium: MeCN Calorimetric titration of LiNCS. DH(K1)=-25.2 kJ mol⁻¹

C44H50N2O10 H2L CAS 329183-28-0 (8807)

25,27-Bis(carboxymethoxy)-26,28-bis[(N,N-diethylaminocarbonyl)methoxy]calix[4]arene
;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	non-aq	25°C	100%	C		K1=4.28 B(Li2L)=7.11	2000ABb (107144)	826

Medium: MeOH, 0.05 M Et4NClO4.

C44H52N4O8 L CAS 246035-33-6 (2925)
25,27-Bis(N,N-diethylaminocarbonylmethoxy)-26,28-bis(aminocarbonylmethoxy)calix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=<1	1999USa (107159)	827

Medium: MeOH, 0.10 M Et4NCl

C44H52O10 L CAS 163317-54-2 (9089)
1,3-Calix[4]-bis-crown-5;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C	I	K1=1.80	1996AAe (107165)	828

Medium: acetonitrile. In 100% MeOH, K1<=1.

C44H54O8 L CAS 161282-98-0 (8679)
25,27-Bis(1-propyloxy)calix[4]arene-crown-6, 1,3-alternate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=<=1	1995CUa (107176)	829

Medium: methanol, 0.01 M Et4NCl.

C44H54O8 L CAS 161282-96-8 (8678)
25,27-Bis(2-propyloxy)calix[4]arene-crown-6, 1,3-alternate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=<=1	1995CUa (107182)	830

Medium: methanol, 0.01 M Et4NCl.

C44H56O4 H4L (7294)
4-Tert-butyl-calix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	U		K1=2.9 B(Li2L)=4.04	1996ABa (107187)	831

Medium: MeCN

C45H39O3P3 L CAS 73218-92-5 (5679)
1,3,5-Tris(diphenylphosphinylmethyl)-benzene; C6H3(CH2.PO(C6H5)2)3

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=5.1	1984YKa (107213)	832
Medium: tetrahydrofuran + CHCl3 4:1, Li as 2,4-dinitrophenolate										

C45H48NO6P3		L						(7953)		
Tris[2-(diphenylphosphorylmethoxy)ethyl]amine;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	U	H		K1=4.08	1998SBb (107219)	833
Medium: MeCN Calorimetric titration of LiNCS. DH(K1)=-28.1 kJ mol-1										

C45H48N3O3P3		L						CAS 90179-28-5 (5682)		
N,N',N''-tris(Diphenylphosphinylmethyl)-1,4,7-triazacyclononane;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=5.6	1984YKa (107226)	834
Medium: tetrahydrofuran + CHCl3 4:1, Li as 2,4-dinitrophenolate										

C46H40O6P2		L						(6814)		
1,2-Bis((2-(2-diphenylphosphinyl)phenoxy)ethoxy)benzene;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=7.0	1991EBa (107241)	835
Solvent : Tetrahydrofurane + CHCl3 4:1(vol);										

C46H46N2O4		L						CAS 185118-12-1 (7824)		
N,N'-Bis(1-pyrenylmethyl)-1,4,10,13-tetraoxa-7,16-diazacyclooctadecane;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	mixed	25°C	90%	C			K(LiSCN+L)=0.48	1997KKa (107250)	836
Method: fluorescence emission. Medium: MeOH/CHCl3 (9:1 v/v).										

C46H46O7P2		L						(6807)		
1,15-Bis(2-Diphenylphosphinylphenyl)-2,5,8,11,14-pentaoxapentadecane;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=4.9	1993BEb (107260)	837
Medium: THF+CHCl3 4:1(vol)										

C46H48O6P2		L						(7155)		
1,8-Bis(2-(2-(diphenylphosphinyl)ethyl)phenoxy)-3,6-dioxaoctane										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	oth	non-aq	25°C	100%	U			K1=4.6	1995TEa (107271)	838
Medium: THF:CHCl3 4:1 v/v. Li as 2,4-dinitrophenolate. Also other similar ligands										

C46H48O9P2		L						CAS 95651-38-0	(2082)	
1,5-Bis(2-(2-(diphenylphosphinylmethoxy)ethoxy)phenoxy)-3-oxapentane;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=4.18	1989KSa (107280)	839
Medium: tetrahydrofuran/CHCl3 4:1 (vol)										

C48H44O8P2		L						CAS 95651-37-9	(2081)	
1,2-Bis(2-(2-(diphenylphosphinylmethoxy)phenoxy)ethoxy)benzol;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=3.75	1989KSa (107361)	840
Medium: tetrahydrofuran/CHCl3 4:1 (vol)										

C48H50O8P2		L						(6808)		
1,18-Bis(2-Diphenylphosphinylphenyl)-2,5,8,11,14,17-hexaoxananodecane;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U			K1=4.6	1993BEb (107365)	841
Medium: THF+CHCl3 4:1(vol)										

C48H54N06P3		L						(7975)		
Tris(3-oxa-5-(diphenylphosphoryl)pentyl]amine;										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	U	H		K1=4.16 B(Li2L)=6.04 B(Li3L)=9.99	1998SBb (107376)	842
Medium: MeCN Calorimetric titration of LiNCS. DH(K1)=-32.8 kJ mol ⁻¹ DH(Li2L)=-68.4, DH(Li3L)=-46.7										

C48H54N6O8		L						CAS 449738-94-7	(8791)	
1,7-Dioxa-4,10-diazacyclododecane-4,10-bis[methylene-8-(1,3,3-trimethyl-6-nitro-spirobenzopyran)]										
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	C			K1=6.62	2002NFa (107384)	843
Medium: 100% MeOH. Method: electrospray ionization mass spectrometry.										

C48H54O10P4 L CAS 97910-30-0 (2084)
Tris((2-(diphenylphosphinylmethoxy)ethoxy)methyl)phosphine oxide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+		con non-aq	25°C	100%	U		K1=4.45	1989KSa (107388)	844

Medium: tetrahydrofuran/CHCl3 4:1 (vol)

C48H60O8 H2L R-Bu-Calixarene CAS 147513-53-9 (6705)
4-tert-Butylcalix[4]areneedicarboxylic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	100%	C		K1=4.5 B(Li2L)=7.6 B(LiHL)=12.4	1993ABb (107403)	845

Medium: MeOH, 0.01 M Et4NClO4. Data also for di-tert-butyl ester

C48H60O12 L CAS 157769-14-7 (9090)
1,3-Calix[4]-bis-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C	I	K1=2.3	1996AAe (107411)	846

Medium: acetonitrile. In 100% MeOH, K1<=1.

C48H64O4 L CAS 105880-81-7 (8677)
tert-Butylcalix-4-arene tetramethyl ether;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=5.10	2004BCb (107421)	847

Medium: acetonitrile, 0.01 M Et4NClO4.

C52H62N6O10 ; L CAS 190781-91-0 (8792)
1,4,10,13-Tetraoxa-7,16-diazacyclododecane-7,16-bis[methylene-8-(trimethyl-6-nitro-spirobenzopyra

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	C		K1=6.85	2002NFa (107480)	848

Medium: 100% MeOH. Method: electrospray ionization mass spectrometry.

C52H64O12 H4L R-Bu-Calixarene CAS 113215-72-8 (6704)
5,11,17,23-Tetra-(t-butyl)-25,26,27,28-tetrakis[(hydroxycarbonyl)methoxy]calix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	gl	alc/w	25°C	100%	C		K1=7.89	1993ABb (107491)	849

B(LiHL)=18.93
 B(LiH2L)=27.98
 B(LiH3L)=35.68

In methanol; 0.01 M (CH3CH2)4NClO4

C52H68N4O8 CAS 150588-24-2 (3074)
 25,26,27,28-Tetrakis-(N,N-diethylaminocarbonylmethoxy)calix[4]arene; L

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C	H	K1=3.0	1999USa (107499)	850

Medium: MeOH, 0.10 M Et4NCl. By calorimetry: DH(K1)=-1 kJ mol-1

C52H68N4O8 L (4823)
 25,27-Bis(N,N-diethylaminocarbonylmethoxy)-26,28-bis(N-butylaminocarbonylmethoxy)calix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=<1	1999USa (107508)	851

Medium: MeOH, 0.10 M Et4NCl

C52H72O6 L (9263)
 5,11,17,23-Tetra(t-butyl)-25,27-dimethoxy-26,28-dimethoxyethoxycalix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	mixed	25°C	100%	C		K1=5.41	2004BCb (107526)	852

Medium: acetonitrile, 0.01 M Et4NClO4.

C54H74O7 L (7302)
 25,27-Dimethoxy-4-tert-butylcalix[4]arene-crown-5;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	22°C	100%	U		K1=4.53	1996SCa (107542)	853

Medium: CHCl3 saturated with H2O

Data also for other substituted t-butylcalix[4]arene-crown-5 analogues

C54H90N6O18 L Valinomycin CAS 2001-95-8 (2142)
 Valinomycin, Potassium Ionophore

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	dis	non-aq	22°C	100%	C	M	K(LiA+L(org))=LiAL(org))=5.83	1996CPa (107554)	854

Medium: CHCl3 saturated with H2O. Method: extraction of LiA into CHCl3/L solution. HA is picric acid.

C56H60O12 L CAS 157769-17-0 (9091)

1,3-Calix[4]-bis-benzo-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	25°C	100%	C			K1=1.5	1996AAe (107578)	855
-----	----	--------	------	------	---	--	--	--------	------------------	-----

Medium: acetonitrile.

C56H64O10 L CAS 405108-40-9 (8249)
1,2-Di-O-[2-(2-benzyloxyethoxy)ethyl]-3,4,5,6-tetra-O-benzyl-myo-inositol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	dis	non-aq	25°C	100%	C			K(Li.pic+L(org))=LiL.pic)=2.19	2001SSb (107587)	856
-----	-----	--------	------	------	---	--	--	--------------------------------	------------------	-----

Distribution of picrate salt into CHCl₃/HL.

K: Li.pic(aq)+L(org)=LiL.pic(org). Data for series of myo-inositol ligands

C56H72O8 L CAS 123311-74-0 (6160)
Tetramethyl-t-butylcalix[4]arenetetraetone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	alc/w	25°C	100%	U I			K1=2.7	1989ACb (107598)	857
-----	----	-------	------	------	-----	--	--	--------	------------------	-----

Medium: MeOH. In CH₃CN, K1=5.8

C56H72O12 L (8751)
Tetramethyl-4-t-Butylcalix[4]arenetetraethanoate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	EMF	non-aq	25°C	100%	C IH			K1=5.61	1995DGa (107602)	858
-----	-----	--------	------	------	------	--	--	---------	------------------	-----

Medium: acetonitrile, 0.05 M Et₄NClO₄. In benzonitrile, K1=5.63.
Competitive method: Ag/Ag⁺ electrode. DH(K1)=-37.80, DS=-19.4.

C56H78O8 L CAS 122356-76-7 (8681)
Tetra-tert-butyl-1,3-dimethoxycalix[4]arene-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	25°C	100%	C			K1=<=1	1995CUa (107607)	859
-----	----	--------	------	------	---	--	--	--------	------------------	-----

Medium: methanol, 0.01 M Et₄NCl.

C56H80O8 L (9259)
5,11,17,23-Tetra(t-butyl)-25,26,27,28-tetramethoxyethoxycalix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

Li+	sp	non-aq	25°C	100%	C H			B2=9.23	2004BCb (107614)	860
-----	----	--------	------	------	-----	--	--	---------	------------------	-----

Medium: acetonitrile, 0.01 M Et₄NClO₄. By calorimetry: DH(B2)=-28.3
kJ mol⁻¹, DS(B2)=81.7 J K⁻¹ mol⁻¹.

C58H78O11 HL CAS 465527-74-6 (9287)
7,13,19,25-Tetra-t-butyl-28-methoxy-27,29,30-triethylacetate-2,3-dihomo-3-oxacalix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	C		K1=2.6	2001MAa (107623)	861

Medium: MeOH, 0.01 M Et4NCl.

C58H80O10 L (9264)
5,11,17,23-Tetra-t-butyl-25,27-di(2-methoxyethoxy)-26,28-di(ethylacetate)calix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C	H	K1=5.99 B2=10.72	2004BCb (107632)	862

Medium: acetonitrile, 0.01 M Et4NClO4. DH(K1)=-33.8 kJ mol⁻¹, DS(K1)=1.0 J K⁻¹ mol⁻¹; DH(B2)=-19.0, DS(B2)=141.1.

C60H54N06P3 L (8067)
Tris[2-diphenylphosphoryl]phenoxyethylamine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	U	H	K1=4.30 B(Li2L)=6.35 B(Li3L)=8.22	1998SBb (107639)	863

Medium: MeCN Calorimetric titration of LiNCS. DH(K1)=-10.7 kJ mol⁻¹
DH(Li2L)=-16.6, DH(Li3L)=-19.3

C60H72O4 L (9260)
5,11,17,23-Tetra(t-butyl)-25,27-dimethoxy-26,28-diphenylmethoxycalix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C		K1=3.54	2004BCb (107643)	864

Medium: acetonitrile, 0.01 M Et4NClO4.

C60H80O12 L CAS 97600-39-0 (6158)
Tetraethyl-4-t-butylcalix[4]arenetetraethanoate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	C	H	K1=6.25	2002ASc (107652)	865

Medium: acetonitrile. DH(K1)=-45.83 kJ mol⁻¹, DS(K1)=-42.36 J K⁻¹ mol⁻¹.

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	EMF	non-aq	25°C	100%	C	I	K1=6.20	1995DGa (107653)	866

Medium: acetonitrile, 0.05 M Et4NClO4. Competitive method: Ag/Ag+ electrode. Also data for solvent benzonitrile and for tetrabutyl deriv.

Li+ sp alc/w 25°C 100% U I K1=2.6 1989ACb (107654) 867
Medium: MeOH. In CH3CN, K1=6.4

C60H82N2O10 L CAS 155377-20-1 (8806)
5,11,17,23-Tetra-butyl-25,27-bis(carboxymethoxy)-bis[(N,N-diethylaminocarbonyl)methoxy]calix[4]ar

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl non-aq 25°C 100% C K1=4.32 2000ABb (107667) 868
Medium: MeOH, 0.05 M Et4NClO4.

C60H84N4O8 L CAS 246035-32-5 (2735)
25,27-Bis(N,N-diethylaminocarbonylmethoxy)-26,28-bis(aminocarbonylmethoxy)-t-butylcalix[4]arene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=<1 1999USa (107680) 869
Medium: MeOH, 0.10 M Et4NCl

C62H78N2O4S2 L (8158)
5,11,17,23-Tetrakis(1,1-dimethylethyl)-25,27-bis(2-methylthioethoxy)...calix(4)arene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal non-aq 25°C 100% U H K1=5.42 2002NRa (107686) 870
Method: microcalorimetry. Medium: MeCN.. DH(K1)=-26.2 kJ mol-1
In benzonitrile K1=5.88, DH=-37.6

C62H84O14 L CAS 135581-11-2 (8630)
9,23-Dioxpentacyclo[23.3.1.13,7.111.15.117.21]dotriacontane, ethanoic acid derivative;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=3.2 1991ACc (107695) 871
Medium: acetonitrile, 0.01 M Et4NClO4.

C63H60N06P3 L (8437)
Tris[2-(diphenylphosphorylmethyl)phenoxyethyl]amine;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal non-aq 25°C 100% U H K1=3.48 1998SBb (107720) 872
B(Li2L)=4.56
B(Li3L)=7.60

Medium: MeCN Calorimetric titration of LiNCS. DH(K1)=-20.2 kJ mol-1

DH(Li2L)=-40.7, DH(Li3L)=-27.1

C64H60012 L CAS 211870-40-5 (4258)

Calix[4]arene-bis(dibenzo)crown-6;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=2.18 1999Lda (107735) 873
Medium: acetonitrile, 0.01 M Et4NClO4.

C64H6206P4 L (6813)

1,2-Bis(4,5-di(diphenylphosphinyl)-pent-1-oxy)benzene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=6.0 1990EAb (107740) 874
Medium: THF+CHCl3 4:1(vol). Metal as 2,4-dinitrophenolate

C64H64012 L CAS 162898-44-4 (9092)

1,3-Calix[4]-bis-naphtho-crown-6;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=1.2 1996AAe (107745) 875
Medium: acetonitrile.

C64H72N404P4 L CAS 104786-07-4 (2065)

1,4,7,10-Tetra(diphenylphosphinylethyl)-1,4,7,10-tetraazacyclododecane;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ con non-aq 25°C 100% U K1=5.91 1986STb (107753) 876
Medium: THF:CHCl3 4:1 v/v. M as 2,4-dinitrophenolate

C64H8006 L (9262)

5,11,17,23-Tetra-t-butyl-25,27-di(phenylmethoxy)-26,28-di(2-methoxyethoxy)-calix[4]arene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C H K1=3.76 B2= 7.88 2004BCb (107762) 877
Medium: acetonitrile, 0.01 M Et4NClO4. DH(K1)=-30.5 kJ mol⁻¹
DS(K1)=-30.6 J K⁻¹ mol⁻¹; DH(B2)=-20.0, DS(B2)=83.6.

C64H8607 L CAS 182684-17-9 (7455)

4-tert-Butylcalix[5]crown-4 trimethylester;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp alc/w 25°C 100% U K1=1.5 1996AAc (107769) 878

Medium MeOH, 0.1M Et4NCl. Data also for the crown-5 and crown-6 analogues

C66H80O8 L (9261)
5,11,17,23-Tetra(t-butyl)-25,27-diethoxycarbonylmethoxy-26,28-diphenylmethoxycalix[4]arene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=4.30 2004BCb (107777) 879

Medium: acetonitrile, 0.01 M Et4NClO4.

C68H76N4O4 L CAS 123207-92-1 (7812)
5,11,17,23-Tetra-t-butyl-[25,26,27,28-tetrakis(2-pyridylmethyl)oxy]calix(4)arene;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ EMF non-aq 25°C 100% C IH K1=5.95 1999DCa (107785) 880

Medium: acetonitrile, 0.05 M Bu4NClO4. Method: by competition with Ag+.

By calorimetry: K1=5.95, DH(K1)=-23.91 kJ mol⁻¹, DS(K1)=33.7 J K⁻¹ mol⁻¹.

C68H92N4O8 L CAS 133801-01-1 (7184)
4-tert-Butylcalix[4]arene tetrapyrrolidinylamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal alc/w 25°C 100% U H 1995ABc (107791) 881

Medium: 100% Methanol. DH(K1)=6 kJ mol⁻¹, DS(K1)=77 J K⁻¹ mol⁻¹.

C68H96O8 L (6161)
Tetra-t-butyl-4-t-butylcalix[4]arenetetraketone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp alc/w 25°C 100% U K1=1.8 1989ACb (107795) 882

Medium: MeOH, 0.1 M Et4NCl

C68H100N4O8 L CAS 246035-35-8 (3034)
25,27-Bis(N,N-diethylaminocarbonylmethoxy)-26,28-bis(N-butylaminocarbonylmethoxy)-t-butylcalix[4]

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ sp non-aq 25°C 100% C K1=<1 1999USa (107805) 883

Medium: MeOH, 0.10 M Et4NCl

C68H100N4O8 L CAS 114155-16-7 (7183)
4-tert-Butylcalix[4]arene tetradiethylacetamide;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ cal alc/w 25°C 100% U IH 1995ABc (107815) 884
 Medium: 100% Methanol. DH(K1)=-7 kJ mol⁻¹, DS(K1)=50 J K⁻¹ mol⁻¹.
 In acetonitrile, K1>8.5, DH(K1)=-55 kJ mol⁻¹, DS(K1)=-22 J K⁻¹ mol⁻¹.

Li+ dis non-aq 20°C 100% C M 1988AGa (107816) 885
 K(Li+A+L(org))=LiAL(org))=7.11

Method: extraction of metal picrate into CHCl₃/L solution. HA is picric acid.

 C69H102N4O9 L CAS 116352-85-3 (9286)
 para-t-Butyldihomooxacalix[4]arene tetra(diethyl)amide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	C		K1=3.81	2004MFa (107835)	886

Medium: MeOH, 0.01 M Et₄NCl.

 C72H68O10P4 L CAS 88928-02-3 (5680)
 Tetrakis-4',5',4",5"- (diphenylphosphinylmethyl)-2,3:11,12-dibenzo-18-crown-6;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	con	non-aq	25°C	100%	U		K1=3.09	1985YKa (107847)	887

Medium: EtOH+CHCl₃ 1:1; M is used in nitrophenolate form

 C75H100O15 L CAS 152495-34-6 (7033)
 Penta-tert-butylpentakis(ethoxycarbonylmethyloxy)calix[5]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	U		K1=1.0	1993BMa (107860)	888

Medium: MeOH, 0.1 M Et₄NCl.

 C76H80O8 L (6162)
 5,11,17,23-Tetra-t-butyl-25,26,27,28-tetra(benzoyl)methoxycalix[4]arene;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	U		K1=6.3	1989ACb (107870)	889

Medium: CH₃CN

 C77H82O9 L CAS 253317-20-3 (9288)
 p-Tert-butyldihomooxacalix[4]arene tetraphenylketone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	C I		K1=2.3	1999MAb (107894)	890

Medium: MeOH, 0.01 M Et₄NCl. In acetonitrile, K1=3.6.

 C78H90O10P2 L CAS 160638-26-6 (9130)

5,11,17,23-Tetra-*t*-butyl-bis(diethylcarbamoylmethoxy)-bis(diphenylphosphinoylmethoxy)calix[4]aren

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	alc/w	20°C	100%	C			K1=2.88	2003YVa (107900)	891
Medium: 100% EtOH, 0.01 M Et4NBr. Ligand is cone isomer. For paco isomer, K=2.64. Also data for bis(diethyl ester) analogues.										

C85H120O15		L						CAS 152495-35-7	(7034)	
Penta- <i>tert</i> -butylpentakis(<i>tert</i> -butoxycarbonylmethoxy)calix[5]arene;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	alc/w	25°C	100%	U			K1=1.5	1993BMa (107917)	892
Medium: MeOH, 0.1 M Et4NCl.										

C90H120O18		L						CAS 92003-62-8	(6159)	
Hexaethyl-4- <i>t</i> -butylcalix[6]arenehexaethanoate;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	cal	non-aq	25°C	100%	C			K1=4.37	1997DZa (107942)	893
Medium: benzonitrile. DH(K1)=-21.04 kJ mol ⁻¹ , DS(K1)=13.1 J K ⁻¹ mol ⁻¹ .										

Li+	sp	non-aq	25°C	100%	U	I		K1=3.7	1989ACb (107943)	894
Medium: CH3CN										

C90H130O15		L						CAS 269057-78-5	(3334)	
5,11,17,23,29-Penta- <i>tert</i> -octylcalix[5]arene-31,32,33,34,35-pentaethanoate pentamethyl ester;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	sp	non-aq	25°C	100%	C	I		K1=2.28	2000AAa (107951)	895
Medium: methanol, 0.01 M Et4NCl. Also data for acetonitrile, 0.01 M Et4NCl and for the pentaethyl ester.										

Polymer		H2L			X-14885A			(4547)		
Antibiotic X14885A, calcium ionophore										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Li+	gl	alc/w	25°C	100%	U			K1=4.1	1989ABb (108076)	896
Medium: MeOH										

Polymer								(4181)		
Phosphatidic acid;										

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
-------	-----	--------	------	------	-----	-------	----	----------	-----------	--------

 Li+ gl oth/un 24°C 0.10M U K1=1.3 1966AKa (108271) 897

Polymer (4192)

Polyacrylic acid and 7.5% divinylbenzene copolymer

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl oth/un 25°C 0.2M U K1=0.29 1957GFa (108304) 898

Polymer HL (3531)

Polyacrylic acid;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Li+ gl oth/un 25°C 0.20M U K'=0.28 1957GFa (108323) 899

Medium: LiCl. See reference for definitions

REFERENCES

- 2004BCb L Baklouti,J Cherif,R Abidi,F Arnaud-Neu; Org.Biomol.Chem.,2,2786 (2004)
 2004KVa T Kirichenko,V Vetrogon,N Lukyanenko; Anal.Chim.Acta,505,277 (2004)
 2004MFa P Marcos,S Felix,J Ascenso,M Segurado; New J.Chem.,28,748 (2004)
 2003ADa F Arnaud-Neu,R Delgado,S Chaves; Pure & Appl.Chem.,75,71 (2003)
 2003GHa J Geue,N Head,A Ward,S Lincoln; Aust.J.Chem.,56,917 (2003)
 2003YVa M Yaftian,M Vahedpour,H Abdollahi; J.Inclusion Phenom.,47,129 (2003)
 2002ASc M Ashram; J.Inclusion Phenom.,42,25 (2002)
 2002DDa D Das,B Das,D Hazra; J.Solution Chem.,31,425 (2002)
 2002GLb T Gunnlaugsson,J Leonard; J.Chem.Soc.,Perkin Trans.,II,1980 (2002)
 2002GNe T Gunnlaugsson,M Nieuwenhuyzen; J.Chem.Soc.,Perkin Trans.,II,141 (2002)
 2002GVC S Gromov,A Vedernikov,E Ushakov,U Edlund; Helv.Chim.Acta,85,60 (2002)
 2002IIa K Ishimori,H Imura,K Ohashi; Anal.Chim.Acta,454,241 (2002)
 2002NFa M Nakamura,T Fujioka,H Sakamoto,K Kimura; New J.Chem.,26,554 (2002)
 2002NRA A D de Namor,N.Rami,O Piro et al.; J.Phys.Chem.B,106,779 (2002)
 2002PLa K Popov,L Lajunen,A Popov et al; Inorg.Chem.Comm.,5,223 (2002)
 2002THb Y Takeda,K Hashimoto,D Yoshiyama; J.Inclusion Phenom.,42,313 (2002)
 2002TTa Y Takeda,M Tanaka,H Yamada,S Katsuta; J.Coord.Chem.,55,459 (2002)
 2002YEa G Yapar,C Erk; J.Inclusion Phenom.,42,145 (2002)
 2002YPb V Yam,L-L Pui,K-K Cheung; Inorg.Chim.Acta,335,77 (2002)
 2002YPc V Yam,Y Pui,K Cheung,N Zhu; New J.Chem.,26,536 (2002)
 2001AVa L Antonov,M Vladimirova,M Mitewa; J.Inclusion Phenom.,40,23 (2001)
 2001BBa C Bazzicalupi,A Bencini,A Bianchi,F Pina; Inorg.Chem.,40,6172 (2001)
 2001INa H Inerowicz; J.Inclusion Phenom.,39,211 (2001)
 2001KMb S Katsuta,T Motoyama,Y Takeda,M Ouchi; Bull.Chem.Soc.Jpn.,74,311 (2001)
 2001KZa E Kharkhaneei,M Zebrajadian,M Shamsipur; J.Solution Chem.,30,323 (2001)
 2001KZb E Karkhaneei,M Zebrajadian,M Shamsipur; J.Inclusion Phenom.,40,309 (2001)
 2001LFa H-J Lu,Y-T Fan,Y-J Wu; Polyhedron,20,3281 (2001)
 2001LWa L-D Li,Y Wei,A-J Tong; Anal.Chim.Acta,427,29 (2001)

- 2001MAa P Marcos, J Ascenso, M Segurado, J Pereira; *Tetrahedron*, 57, 6977 (2001)
- 2001SSb K Sureshan, M Shashidhar, A Varma; *J.Chem.Soc., Perkin Trans.II*, 2298 (2001)
- 2001WBa J Weeks, M Buntine, S Lincoln; *J.Chem.Soc., Dalton Trans.*, 1939 (2001)
- 2000AAA F Arnaud-Neu, Z Asfari, B Souley; *J.Chem.Soc., Perkin Trans.II*, 495 (2000)
- 2000ABb F Arnaud-Neu, S Barbosa, A Casnati; *New J.Chem.*, 24, 967 (2000)
- 2000EGa C Erk, A Gocmen; *Talanta*, 53, 137 (2000)
- 2000GDa L Goldenberg, N Denisov, J Biernat; *J.Inclusion Phenom.*, 38, 171 (2000)
- 2000HHa M Hojo, I Hisatsune, H Tsurui, S Minami; *Anal.Sci.Jpn.*, 16, 1277 (2000)
- 2000ICa H Inerowicz, J Chojnacki, A Merz; *J.Inclusion Phenom.*, 38, 123 (2000)
- 2000KSa Y Kikuchi, Y Sakamoto; *Anal.Chim.Acta*, 403, 325 (2000)
- 2000LXa H Li, H Xie, P Wang, S Wu; *New J.Chem.*, 24, 105 (2000)
- 2000MFa M Micheloni, M Formica, V Fusi; *Eur.J.Inorg.Chem.*, 51 (2000)
- 2000SMD M Shamsipur, T Madrakian; *Polyhedron*, 19, 1681 (2000)
- 2000SMg M Shamsipur, T Madrakian; *J.Coord.Chem.*, 52, 139 (2000)
- 2000TMB Y Takeda, Y Mochizuki, Y Matsuzaki; *J.Inclusion Phenom.*, 37, 179 (2000)
- 2000VMA P Victor, P Muhuri, B Das, D Hazra; *J.Phys.Chem.B*, 104, 5350 (2000)
- 2000ZKb X Zhang, K Krakowiak, J Bradshaw, R Izatt; *Ind.Eng.Chem.Res.*, 39, 3516 (2000)
- 2000ZSb J-Z Yang, B Sun, P-S Song; *Thermochim.Acta*, 352/3, 69 (2000)
- 1999BHa H Buschmann, J Hermann, H Plenio; *Chem.Eur.J.*, 5, 2566 (1999)
- 1999DCa A Danil de Namor, E Castellano, L Salazar; *Phys.Chem.Chem.Phys.*, 1, 285 (1999)
- 1999DSd A D'Aprano, B Sesta, V Mauro, M Salomon; *J.Inclusion Phenom.*, 35, 451 (1999)
- 1999ESa V Evreinov, Z Safronova, A Yarkevich et al; *Zh.Obshch.Khim.*, 69, 1088 (1999)
- 1999KKb Y Kikuchi, M Kubota, K Sawada; *Bull.Chem.Soc.Jpn.*, 72, 2437 (1999)
- 1999LDA V Lamare, J-F Dozol, S Fuangswasdi; *J.Chem.Soc., Perkin Trans.II*, 271 (1999)
- 1999MAb P Marcos, J Ascenso, M Segurado, J Pereria; *J.Phys.Org.Chem.*, 12, 695 (1999)
- 1999NMA T Nakamura, K Makino, M Yanagisawa; *Bull.Chem.Soc.Jpn.*, 72, 2459 (1999)
- 1999TEa V Tsvetkov, V Evreinov et al; *Zh.Obshch.Khim.*, 69, 1080 (1999)
- 1999TMA Y Takeda, Y Mochizuki, M Tanaka, Y Kudo; *J.Inclusion Phenom.*, 33, 217 (1999)
- 1999USa R Ungaro, M Schwing-Weill, G Wipff; *J.Chem.Soc., Perkin Trans.II*, 1727 (1999)
- 1999WBa G Wenz, H-J Buschmann, E Schollmeyer; *J.Coord.Chem.*, 48, 465 (1999)
- 1999WKb E Wagner-Czauderna, M Kalinowski; *J.Coord.Chem.*, 46, 265 (1999)
- 1998BJb H-J Buschmann, K Jansen, C Meschke; *J.Solution Chem.*, 27, 135 (1998)
- 1998DBa D Dantz, H Buschmann, E Schollmeyer; *Polyhedron*, 17, 1891 (1998)
- 1998DDc P Delangle, J-P Dutasta, J-P Declercq; *Chem.Eur.J.*, 4, 100 (1998)
- 1998FLb S Fery-Forgues, D Lavabre, A Rochal; *New J.Chem.*, 1531 (1998)
- 1998IHb M Islam, P Hambright; *Transition Met.Chem.*, 23, 727 (1998)
- 1998KSc Y Kikuchi, Y Sakamoto, K Sawada; *J.Chem.Soc., Faraday Trans.*, 94, 105 (1998)
- 1998KZa E Karkhaneei, J Zolgharnein; *J.Coord.Chem.*, 46, 1 (1998)
- 1998MLa M Mimouni, R Lyazghi, J Juillard; *New J.Chem.*, 367 (1998)
- 1998SBb V Solov'ev, V Baulin, A Varnek et al.; *J.Chem.Soc., Perkin Trans.II*, 1489 (1998)
- 1998TIA T Takayanagi, T Iwashido, S Motomizu; *Bull.Chem.Soc.Jpn.*, 71, 1373 (1998)
- 1998TKa Y Takeda, A Kawarabayashi, K Endo; *Anal.Sci.Jpn.*, 14, 215 (1998)
- 1998TNa M Tabata, J Nishimoto, T Kusano; *Talanta*, 46, 703 (1998)
- 1998WLC S Whitbread, S Lincoln, K Wainwright; *J.Am.Chem.Soc.*, 120, 2862 (1998)
- 1997CHb Z Chen, M Hojo; *J.Phys.Chem.B*, 101, 10896 (1997)
- 1997DMd R Dhillon, S Madbak, F Ciccone, S Lincoln; *J.Am.Chem.Soc.*, 119, 6126 (1997)
- 1997DZa A Danil de Namor, M Zapata-Ormachea; *J.Phys.Chem.B*, 101, 6772 (1997)

- 1997KKa K Kubo, N Kato, T Sakurai; *Bull.Chem.Soc.Jpn.*, 70, 3041 (1997)
- 1997LHa I Lednev, R Hester, J Moore; *J.Chem.Soc., Faraday Trans.*, 93, 1551 (1997)
- 1997PBb Y Pointud, C Bernard, J Juillard; *J.Solution Chem.*, 26, 479 (1997)
- 1997STa A Saito, H Tomari, G Choppin; *Inorg.Chim.Acta*, 258, 145 (1997)
- 1997WRa E Wagner-Czauderna, J Rzeszotarska; *Ber.Buns.Phys.Chem.*, 101, 1154 (1997)
- 1997WWa S Whitbread, J Weeks, S Lincoln; *Australian J.Chem.*, 50, 853 (1997)
- 1997YLa V W-H Yam, V W-M Lee, F Ke, K-W Siu; *Inorg.Chem.*, 36, 2124 (1997)
- 1997ZTa X Zhang, R Izatt, K Krakowiak; *Inorg.Chim.Acta*, 254, 43 (1997)
- 1996AAb R Abidi, F Arnaud-Neu, M Drew, J Nelson; *J.Chem.Soc., Perkin Trans.II*, 2747 (1996)
- 1996AAc F Arnaud-Neu, R Arnecke, J Gordon; *J.Chem.Soc., Perkin Trans.II*, 1855 (1996)
- 1996AAe F Arnaud-Neu, Z Asfari, B Souley, J Vicens; *New J.Chem.*, 20, 453 (1996)
- 1996ABa R Abidi, M Baker, J Harrowfield; *Inorg.Chim.Acta*, 246, 275 (1996)
- 1996BCh H-J Buschmann, E Cleve, E Schollmeyer; *J.Coord.Chem.*, 39, 293 (1996)
- 1996BFc A Bencini, V Fusi, C Giorgi, M Micheloni; *J.Chem.Soc., Perkin Trans.II*, 2297 (1996)
- 1996CPa A Casnati, A Pochini, R Ungaro, D Reinhoudt; *Chem.Eur.J.*, 2, 436 (1996)
- 1996DNa A de Namor, J Ng, M Tanco, M Salomon; *J.Phys.Chem.*, 100, 14485 (1996)
- 1996HHc M Hojo, H Hasegawa, N Hiura; *J.Phys.Chem.*, 100, 891 (1996)
- 1996KAb E Karkhaneei, A Afkhami, M Shamsipur; *J.Coord.Chem.*, 39, 33 (1996)
- 1996MAb T Madrakian, A Afkhami, M Shamsipur; *Polyhedron*, 15, 3647 (1996)
- 1996Rhb R Richards, K Hammons, M Joe et al; *Inorg.Chem.*, 35, 1940 (1996)
- 1996RSa J Ren, A Sherry; *Inorg.Chim.Acta*, 246, 331 (1996)
- 1996RSb A de Robertis, C de Stefano, C Foti; *Ann.Chim.(Rome)*, 86, 155 (1996)
- 1996SCa N Sabbatini, A Casnati, C Fischer; *Inorg.Chim.Acta*, 252, 19 (1996)
- 1996SDa A Stephens, R Dhillon et al; *Inorg.Chem.*, 35, 2019 (1996)
- 1996TJa M Tayyebbeh, Z Javad, S Mojtaba; *J.Coord.Chem.*, 40, 121 (1996)
- 1996TNa M Tabata, J Nishimoto, A Ogata; *Bull.Chem.Soc.Jpn.*, 69, 673 (1996)
- 1996WPa S Whitbread, S Politis, S Lincoln; *J.Chem.Soc., Dalton Trans.*, 1379 (1996)
- 1995ABC F Arnaud-Neu, G Barrett, S Fanni, D Marrs; *J.Chem.Soc., Perkin Trans.II*, 453 (1995)
- 1995CUa A Casnati, R Ungaro, M Schwing, D Reinhoudt; *J.Am.Chem.Soc.*, 117, 2767 (1995)
- 1995DGa A Danil de Namor, E Gil, M Llosa Tanco; *J.Phys.Chem.*, 99, 16776 (1995)
- 1995DSb A D'Aprano, M Salomon, V Mauro; *J.Solution Chem.*, 24, 685 (1995)
- 1995Ksa H Khajesharifi, M Shamsipur; *J.Coord.Chem.*, 35, 289 (1995)
- 1995KTb Y Kudo, Y Takeda, H Matsuda; *J.Electroanal.Chem.*, 396, 333 (1995)
- 1995MMf S Mazzini, R Mondelli, E Ragg, L Scaglioni; *J.Chem.Soc., Perkin Trans.II*, 285 (1995)
- 1995TEa E Tsvetkov, V Evreinov, V Baulin et al; *Zh.Obshch.Khim.*, 65, 1421(1300) (1995)
- 1995VZa B Vaidya, J Zak et al; *Anal.Chem.(USA)*, 67, 4101 (1995)
- 1994BCd H Buschmann, E Cleve, E Schollmeyer; *J.Solution Chem.*, 23, 569 (1994)
- 1994DFc C de Stafano, C Foti, A Gianguzza; *Talanta*, 41, 1715 (1994)
- 1994DTa A de Namor, M Tanco, M Salomon, J Ng; *J.Phys.Chem.*, 98, 11796 (1994)
- 1994FRa S Filipek, J Rzeszotarska, M Kalinowski; *Monatsh.Chem.*, 125, 801 (1994)
- 1994GGa P Gans, J Gill, P Langdon; *J.Chem.Soc., Faraday Trans.*, 90, 315 (1994)
- 1994Hka R Hoffman, W Knoche, C Fenn, H-J Buschmann; *J.Chem.Soc., Faraday Trans.*, 90, 1507 (1994)
- 1994LLa P Lye, G Lawrance, M Maeder et al; *J.Chem.Soc., Dalton Trans.*, 793 (1994)
- 1993ABb F Arnaud-Neu, G Barrett et al; *Inorg.Chem.*, 32, 2644 (1993)

- 1993BCd M Bernstein,D Collum; J.Am.Chem.Soc.,115,8008 (1993)
- 1993BEb A Bovin,V Evreinov et al.; Izv.Akad.Nauk USSR,(5)952 (1993)
- 1993BMa G Barrett,A McKervey,J Malone et al; J.Chem.Soc.,Perkin Trans.II,1475 (1993)
- 1993DLb R Dhillon,S Lincoln; Australian J.Chem.,47,123 (1993)
- 1993EAa C Garcia-Echeverria,F Albericio et al; J.Am.Chem.Soc.,115,11663 (1993)
- 1993EBa V Evreinov,V Baulin et al.; Izv.Akad.Nauk USSR,(3)518 (1993)
- 1993Eva V Evreinov,Z Vostroknutova et al; Zh.Neorg.Khim.,38(9),1519 (1993)
- 1993GSa J Ghasemi,M Shamsipur; J.Coord.Chem.,28,231 (1993)
- 1993Hsa R Helgeson,B Selle et al; J.Am.Chem.Soc.,115,11506 (1993)
- 1993INa Y Inoue,K Nakagawa,T Hakushi; J.Chem.Soc.,Dalton Trans.,1333,2279 (1993)
- 1993LRa S Lincoln,T Rodopoulos; Inorg.Chim.Acta,205,23 (1993)
- 1993MAa S Manohar,G Atkinson; J.Solution Chem.,22,859 (1993)
- 1993RPa T Rodopoulos,P Pittet,S Lincoln; J.Chem.Soc.,Dalton Trans.,1055 (1993)
- 1993SFb A Stephens,S Lincoln; J.Chem.Soc.,Dalton Trans.,2123 (1993)
- 1993TAa L Tassi; J.Chem.Soc.,Faraday Trans.,89,733 (1993)
- 1993TCa M Turonek,P Clarke et al; Inorg.Chem.,32,2195 (1993)
- 1992BEa V Baulin,V Evreinov et al.; Izv.Akad.Nauk USSR,(5)1161 (1992)
- 1992CGB P Clarke,J Gulbis,S Lincoln et al; Inorg.Chem.,31,3398 (1992)
- 1992CRA A Casale,A de Robertis,F Licastro; Ann.Chim.(Rome),82,13 (1992)
- 1992CSc Z Chen,O Schall et al; J.Am.Chem.Soc.,114,444 (1992)
- 1992GSa J Ghasemi,M Shamsipur; J.Coord.Chem.,26,337 (1992)
- 1992LPb R Lyazghi,Y Pointud,J Juillard; J.Chem.Soc.,Faraday Trans.,88,1017 (1992)
- 1992LSc S Lincoln,A Stephens; Inorg.Chem.,31,5067 (1992)
- 1992MGa J Medina,T Goodnow et al; J.Am.Chem.Soc.,114,10583 (1992)
- 1992MPa V Mironov,G Pashkov,T Stupko et al.; Zh.Neorg.Khim.,37,2545 (1992)
- 1992MSe D Mishra,U Sharma,V Bhagwat; J.Indian Chem.Soc.,69,70 (1992)
- 1992PSa H Parham,M Shamsipur; Polyhedron,11,987 (1992)
- 1992TFa Y Takeda,I Fujimaki,S Ochiai,K Aoki; J.Inclusion Phenom.,13,129 (1992)
- 1991ACc F Arnaud-Neu,S Cremin,D Cunningham; J.Inclusion Phenom.,10,329 (1991)
- 1991ALa A A-Hamdan,S Lincoln; Inorg.Chem.,30,462 (1991)
- 1991AMa E Arnett,K Moe; J.Am.Chem.Soc.,113,7288 (1991)
- 1991ASb M Amini,M Shamsipur; Inorg.Chim.Acta,183,65 (1991)
- 1991BSa H Bieth,G Schlewer,B Spiess; J.Inorg.Biochem.,41,37 (1991)
- 1991DDb A de Robertis,C de Stefano; Talanta,38,439 (1991)
- 1991EBa V Evreinov,V Baulin et al.; Izv.Akad.Nauk USSR,(9)1993 (1991)
- 1991FGB F Fronczek,R Gandour,T Fyles; Can.J.Chem.,69,12 (1991)
- 1991JRa L Jackmanm,E Rakiewicz; J.Am.Chem.Soc.,113,1202 (1991)
- 1991LRc S Lincoln,T Rodopoulos; Inorg.Chim.Acta,190,223 (1991)
- 1991MHa P Muhuri,D Hazra; J.Chem.Soc.,Faraday Trans.,87,3511 (1991)
- 1991MPa V Mironov,G Pashkov,T Stupko et al.; Zh.Neorg.Khim.,36,2434 (1991)
- 1991SGa V Solovev,L Govorkova et al.; Izv.Akad.Nauk USSR,(3)575 (1991)
- 1991SMa R Smith,A Martell,Y Chen; Pure & Appl.Chem.,63,1015 (1991)
- 1990CCa M Cabral,J Costa,R Delgado et al; Polyhedron,9,2847 (1990)
- 1990CDB A Casale,A De Robertis,F Licastro; J.Chem.Research(S),204 (1990)
- 1990DDb P Daniele,A de Robertis,C de Stafano; Ann.Chim.(Rome),80,177 (1990)
- 1990DRa P Daniele,A de Robertis,C de Stefano; J.Chem.Res.(S),300 (1990)
- 1990EAb V Evreinov,A Antoshin et al.; Izv.Akad.Nauk USSR,(4)873 (1990)
- 1990KMB R Katakya,K Matthes et al; J.Chem.Soc.,Perkin Trans.II,1425 (1990)

- 1990LAa S Lincoln, A A-Hamdan; *Inorg.Chem.*, 29, 3584 (1990)
- 1990LNa N Lukyanenko, N Nazarova, V Vetrogon et al; *Polyhedron*, 9, 1369 (1990)
- 1990NNA N Nakasuka, Y Natsume, M Tanaka; *Inorg.Chem.*, 29, 147 (1990)
- 1990RAa L Rowe, G Atkinson; *J.Solution Chem.*, 19, 149 (1990)
- 1990SAb M Salomon; *J.Solution Chem.*, 19, 1225 (1990)
- 1990SPa Z Samec, P Papoff; *Anal.Chem.(USA)*, 62, 1010 (1990)
- 1990SSe Y Shibutani, H Sakamoto, K Hayano; *Anal.Chim.Acta*, 375, 81 (1990)
- 1990TPa Tsvetanov, Petrova, Dimov, Panayotov +; *J.Solution Chem.*, 19, 425 (1990)
- 1989ABb A Albrecht, S Blanc, D Boyd, G Jeminet; *J.Am.Chem.Soc.*, 111, 8598 (1989)
- 1989ACb F Arnaud-Neu, E Collins, M Deasy et al; *J.Am.Chem.Soc.*, 111, 8681 (1989)
- 1989BBb A Bencini, A Bianchi, A Borselli et al; *Inorg.Chem.*, 28, 4279 (1989)
- 1989BEa A Bovin, V Evreinov et al; *Izv.Akad.Nauk(USSR)*, 11, 2611 (1989)
- 1989EVA I Evreinov, Z Vostroknutova et al; *Izv.Akad.Nauk(USSR)*, 1, 60 (1989)
- 1989GGa P Gans, J Gill, P Longdon; *J.Chem.Soc., Faraday Trans.I*, 85, 1835 (1989)
- 1989KSA T Kron, E Sinyavskaya, E Tsvetkov; *Izv.Akad.Nauk(USSR)*, 11, 2451 (1989)
- 1989MGc A Mishoustin, N Generalova, A Tsivadze; *Zh.Neorg.Khim.*, 34, 1421(801) (1989)
- 1989SSd N Strakhova, V Solovev, O Raevskii; *Koord.Khim.*, 15(4) 483 (1989)
- 1989TKb E Tsvetkov, T Kron, E Sinyavskaya; *Izv.Akad.Nauk(USSR)*, 11, 2456 (1989)
- 1989TKc Y Takeda, T Kimura, Y Kudo, H Matsuda; *Bull.Chem.Soc.Jpn.*, 62, 2885 (1989)
- 1988AGA A Arduini, E Ghidini, R Ungaro, F Ugozzoli; *J.Inclusion Phenom.*, 6, 119 (1988)
- 1988DSa A Danil de Namor, F Salazar; *J.Chem.Soc., Faraday Trans.I*, 84, 3539 (1988)
- 1988ISb S-I Ishiguro, T Sotobayashi et al; *Inorg.Chem.*, 27, 1152 (1988)
- 1988NHa T Nakamura, H Higuchi, K Izutsu; *Bull.Chem.Soc.Jpn.*, 61, 1020 (1988)
- 1988PJa Y Pointud, J Juillard; *J.Chem.Soc., Faraday Trans.I*, 84, 959 (1988)
- 1988RBA H Rogers, C van den Berg; *Talanta*, 35, 271 (1988)
- 1988SSc V Solovev, N Strakhova, O Raevskii; *Izv.Akad.Nauk(USSR)*, 10, 2400 (1988)
- 1988TKa Y Takeda, K Katsuta, Y Inoue et al; *Bull.Chem.Soc.Jpn.*, 61, 627 (1988)
- 1988TKb Y Takeda, T Kumazawa; *Bull.Chem.Soc.Jpn.*, 61, 655 (1988)
- 1988YKa K Yatsimirskii, M Kabachnik et al; *Izv.Akad.Nauk(USSR)*, 1, 53 (1988)
- 1988YSb K Yatsimirskii, E Sinyavskaya et al; *Izv.Akad.Nauk(USSR)*, 9, 2094 (1988)
- 1987ABe A Abou-Hamdan, I Brereton, A Hounslow; *J.Inclusion Phenom.*, 5, 137 (1987)
- 1987BBf R Bartsch, D Babb, B Knudsen; *J.Inclusion Phenom.*, 5, 515 (1987)
- 1987BUa H-J Buschmann; *J.Solution Chem.*, 16, 181 (1987)
- 1987CCc B Czech, A Czech, B Knudsen et al; *Gazz.Chim.Ital.*, 117, 717 (1987)
- 1987DDb R Delgado, J da Silva et al; *Polyhedron*, 6, 29 (1987)
- 1987DSa A Danil de Namor, S Salazar et al; *J.Chem.Soc., Faraday Trans.I*, 83, 2663 (1987)
- 1987HAa L Harju; *Talanta*, 34, 817 (1987)
- 1987TCa V Tkachev, A Chaikovskaya et al; *Izv.Akad.Nauk(USSR)*, 12, 2745 (1987)
- 1987ZBb D Zollinger, E Bulten, A Christenhusz; *Anal.Chim.Acta*, 198, 207 (1987)
- 1986BUD H-J Buschmann; *Inorg.Chim.Acta*, 125, 31 (1986)
- 1986CDc S Capone, A De Robertis, S Sammartano; *Thermochim.Acta*, 102, 1 (1986)
- 1986CHc D Cram, S Ho; *J.Am.Chem.Soc.*, 108, 2998 (1986)
- 1986DGA A Danil de Namor, L Ghouseini, T Hill; *J.Chem.Soc., Faraday Trans.I*, 82, 349 (1986)
- 1986RSa A de Robertis, C de Stefano et al; *J.Chem.Res.(S)*, 164 (1986)
- 1986STb E Sinyavskaya, L Tsymbal et al; *Izv.Akad.Nauk(USSR)*, 1, 176 (1986)
- 1985AEb R Adamic, E Eyring, S Petrucci, R Bartsch; *J.Phys.Chem.*, 89, 3752 (1985)
- 1985CDB A Casale, A De Robertis, S Sammartano; *Thermochim.Acta*, 95, 15 (1985)

- 1985CKa M Chantooni, I Kolthoff; J. Solution Chem., 14, 1 (1985)
- 1985DGa A Danil de Namor, L Ghouseini; J. Chem. Soc., Faraday Trans. I, 81, 781 (1985)
- 1985DGB A Danil de Namor, L Ghouseini et al; J. Chem. Soc., Faraday Trans. I, 81, 2459 (1985)
- 1985DRa P Daniele, A de Robertis et al; J. Chem. Soc., Dalton Trans., 2353 (1985)
- 1985DRb P Daniele, C Rigano, S Sammartano; Anal. Chem. (USA), 57, 2956 (1985)
- 1985GGB B Gupta, A Gupta, S Gupta, Y Gupta; Indian J. Chem., 24A, 927 (1985)
- 1985GMB G Gross, T Medved, S Novak et al; Zh. Obshch. Khim., 55, 734 (1985)
- 1985HAD L Harju; Finn. Chem. Lett. 235 (1985)
- 1985LWa Y Liu, Y Wang, Z Guo, S Yang etc; Huaxue Tongbao (Chem. China), 5 (1985)
- 1985MHb A Mederos, J Herrera, J Felipe; An. Quim., 81, 152 (1985)
- 1985RSa A de Robertis, C de Stefano, C Rigano +; J. Chem. Res. (S), 42 (1985)
- 1985YKa K Yatsimirskii, M Kabachnik et al; Zh. Neorg. Khim., 30, 976(549) (1985)
- 1984AMa E Arnett, S Maroldo et al; J. Am. Chem. Soc., 106, 6759 (1984)
- 1984BPa B Bubnis, G Pacey; Talanta, 31, 1149 (1984)
- 1984CBa B Czech, D Babb, B Son, R Bartsch; J. Org. Chem., 49, 4805 (1984)
- 1984CTa B Cox, N Truong, J Rzeszotarska et al; J. Chem. Soc., Faraday Trans. I, 80, 3275 (1984)
- 1984CTb B Cox, Ng van Truong et al; J. Am. Chem. Soc., 106, 5965 (1984)
- 1984CTd C Chan, N Tioh, G Hefter; Polyhedron, 3, 845 (1984)
- 1984HCa G Hefter, C Chan, N Tioh; Anal. Chem. (USA), 56, 749 (1984)
- 1984YKa K Yatsimirskii, M Kabachnik et al; Zh. Neorg. Khim., 29, 884(510) (1984)
- 1983CFa B Cox, P Firman, H Hurst et al; Polyhedron, 2, 343 (1983)
- 1983DGA P Daniele, M Grasso, C Rigano et al; Ann. Chim. (Rome), 73, 495 (1983)
- 1983DRb P Daniele, C Rigano, S Sammartano; Thermochim. Acta, 62, 101 (1983)
- 1983SLa Sheng Huaiyu, S Li, H Lu, D Cheng; Acta Chimica Sinica, 1127 (1983)
- 1983YSb K Yatsimirskii, E Sinyavskaya et al; Zh. Neorg. Khim., 28, 1410(795) (1983)
- 1982BDc J Bolte, C Demuynck, G Jeminet; Can. J. Chem., 60, 981 (1982)
- 1982BGe V Bidzilya, L Golovkova et al; Teoret. Eksper. Khim., 18, 65 (1982)
- 1982CCb R Contant, J Ciabrini; J. Chem. Res. (S), 50 (1982)
- 1982DRb P Daniele, C Rigano, S Sammartano; Inorg. Chim. Acta, 63, 267 (1982)
- 1982DSa R Delgado, J da Silva; Talanta, 29, 815 (1982)
- 1982GCB D Gill, J Cheema; Electrochim. Acta, 27, 1267 (1982)
- 1982GJB W Gilkerson, M Jackson; J. Am. Chem. Soc., 104, 1218 (1982)
- 1982GRb M Grigo; Thermochim. Acta, 11, 529 (1982)
- 1982NYa T Nakamura, Y Yumoto, K Izutsu; Bull. Chem. Soc. Jpn., 55, 1850 (1982)
- 1982SSf H Sigel, K Scheller, B Prijs; Inorg. Chim. Acta, 66, 147 (1982)
- 1982TAa Y Takeda; Bull. Chem. Soc. Jpn., 55, 2040 (1982)
- 1982YSa K Yatsimirskii, E Sinyavskaya et al; Zh. Neorg. Khim., 27, 1148(644) (1982)
- 1981ANA G Anderegg; Helv. Chim. Acta, 64, 1790 (1981)
- 1981ASa J Ananthaswamy, B Setturah, T N-Rao; Electrochim. Acta, 26, 925 (1981)
- 1981BEb T Bell; J. Am. Chem. Soc., 103, 1163 (1981)
- 1981CDB V Cucinotta, P Daniele, C Rigano et al; Inorg. Chim. Acta, 56, L45 (1981)
- 1981CRa B Cox, J G-Rosas, H Schneider; J. Am. Chem. Soc., 103, 1384 (1981)
- 1981LIC G Liesegang; J. Am. Chem. Soc., 103, 953 (1981)
- 1981OJa U Olscher, J Jagur-Grodzinski; J. Chem. Soc., Dalton Trans., 501 (1981)
- 1981SAa A Speiss, F Arnaud-Neu, M Schwing-Weill; Inorg. Nucl. Chem. Lett., 17, 253 (1981)
- 1981SKd E Sinyavskaya, M Konstantinovskaya et al; Zh. Neorg. Khim., 26, 1800(971) (1981)

1981SPb E Sinyavskaya, S Pisareva et al; Zh. Neorg. Khim., 26, 1274 (686) (1981)
 1981TMb C Tang, J McLean jnr; Inorg. Chem., 20, 2652 (1981)
 1980CKa B Cox, D Knop, H Schneider; J. Phys. Chem., 84, 320 (1980)
 1980CRa R Cox, J G-Rosas, H Schneider; J. Phys. Chem., 84, 3178 (1980)
 1980GAb T Gilligan, G Atkinson; J. Phys. Chem., 84, 208 (1980)
 1980GRa W Gilkerson, B Roberts; J. Am. Chem. Soc., 102, 5181 (1980)
 1980HNa H Hopkins, A Norman; J. Phys. Chem., 84, 309 (1980)
 1980Lda L Lugina, N Davidenko; Zh. Neorg. Khim., 25, 1454 (1980)
 1980MDa J Massaux, J Desreux, G Duyckaerts; J. Chem. Soc., Dalton Trans., 865 (1980)
 1980OEa U Olsher, G Elgavish et al; J. Am. Chem. Soc., 102, 3338 (1980)
 1980SPb A Smetana, A Popov; J. Solution Chem., 9, 183 (1980)
 1980TYa Y Takeda, H Yano, M Ishibashi et al; Bull. Chem. Soc. Jpn., 53, 72 (1980)
 1979ASc J Ananthaswamy, B Setherum, T Rao; Indian J. Chem., 18A, 123 (1979)
 1979BLb J Bessiere, M Lejaille; Anal. Lett., 12, 753 (1979)
 1979CCa H Cachet, A Cyrot, M Fekir et al; J. Phys. Chem., 83, 2419 (1979)
 1979ITa D Irish, S-Y Tang, H Talts et al; J. Phys. Chem., 83, 3268 (1979)
 1979KLa K Koenig, G Lein, P Stucker et al; J. Am. Chem. Soc., 101, 3553 (1979)
 1979MMA B Martin, D Martin; J. Inorg. Nucl. Chem., 41, 1503 (1979)
 1979PSa N Poonia, S Sarad, A Jayakumar et al; J. Inorg. Nucl. Chem., 41, 1759 (1979)
 1979RHa M Raban, D Haritos; J. Am. Chem. Soc., 101, 5178 (1979)
 1978CAa P Carman; J. Solution Chem., 7, 845 (1978)
 1978CSb B Cox, H Schneider, J Stroka; J. Am. Chem. Soc., 100, 4746 (1978)
 1978FFa F Fisher, A Fox; J. Solution Chem., 7, 561 (1978)
 1978HPa J Hooderheide, A Popov; J. Solution Chem., 7, 357 (1978)
 1978LMA J Lehn, F Montavon; Helv. Chim. Acta, 61, 67 (1978)
 1978YSa K Yatsimirskii, E Sinyavskaya, T Kudrya; Dokl. Akad. Nauk SSSR 240, 100 (1978)
 1978YTa E Yee, J Tabib, M Weaver; J. Electroanal. Chem., 96, 241 (1978)
 1977BIb E Belousov, V Ivanov; Zh. Fiz. Khim., 51, 939 (1977)
 1977HPa A Hourdakakis, A Popov; J. Solution Chem., 6, 299 (1977)
 1977MTc S Moore, T Tarnowski, M Newcomb, D Cram; J. Am. Chem. Soc., 99, 6398 (1977)
 1976ANb G Anderegg; Z. Naturforsch. 31B, 786 (1976)
 1976CWc G Clune, W Waghorne, B Cox; J. Chem. Soc., Faraday Trans. I, 72, 1295 (1976)
 1976FGb H Flora, W Gilkerson; J. Phys. Chem., 80, 679 (1976)
 1976HYa S Harada, T Yasunaga K Tamura et al; J. Phys. Chem., 80, 313 (1976)
 1976KLC E Kauffmann, J Lehn, J Sauvage; Helv. Chim. Acta, 59, 1099 (1976)
 1976PSc A Plaush, R Sharp; J. Am. Chem. Soc., 98, 7973 (1976)
 1975CDa Y Cahen, J Dye, A Popov; J. Phys. Chem., 79, 1289 (1975)
 1975KIc L Kourbatova, A Ivakin, E Voronova; Koord. Khim., 1, 1481 (1975)
 1975LLa B Lundgren, G Levin, S Claesson et al; J. Am. Chem. Soc., 97, 262 (1975)
 1975LSc J Lehn, J Sauvage; J. Am. Chem. Soc., 97, 6700 (1975)
 1975MFa C Mattina, R Fuoss; J. Phys. Chem., 79, 1604 (1975)
 1975MMA B Martin, D Martin; J. Inorg. Nucl. Chem., 37, 1079 (1975)
 1975NAa T Nakamura; Bull. Chem. Soc. Jpn., 48, 1447 (1975)
 1975PTc I Piljac, M Tkalec, B Grabaric; Anal. Chem. (USA), 47, 1369 (1975)
 1975REa E Reardon; J. Phys. Chem., 79, 422 (1975)
 1974BLa R Benoit, S Lam; J. Am. Chem. Soc., 96, 7385 (1974)
 1974BMF E Belousov, V Mironov et al; Zh. Fiz. Khim., 48, 1521 (E:892) (1974)
 1974DKb A Das, K Kundu; J. Chem. Soc., Faraday Trans. I, 70, 1452 (1974)
 1974ESa J Exner, E Steiner; J. Am. Chem. Soc., 96, 1782 (1974)

1974GRa T Grunwald; J.Am.Chem.Soc.,96,2387 (1974)
 1974HPb E Hanna,A Pethybridge,J Prue,D Spiers; J.Solution Chem.,3,563 (1974)
 1974JPa P Jagodzinski and S Petrucci; J.Phys.Chem.,78,917 (1974)
 1974PHb D Pirson,P Huyskens; J.Solution Chem.,3,503 (1974)
 1974PKc I Pereygin,M Klimchuk; Zh.Fiz.Khim.,48,2481(E:1466) (1974)
 1974SPc A Shkodin,V Podolyanko,L Gritsan; Elektrokhim.,10,31(E:24) (1974)
 1973BHa P Beronius,A Holmgren,A Nilsson; Acta Chem.Scand.,27,670 (1973)
 1973CSa J Carr,D Swartzfager; J.Am.Chem.Soc.,95,3569 (1973)
 1973FGa H Flora,W Gilkerson; J.Phys.Chem.,77,1421 (1973)
 1973NIa A Nilsson; Acta Chem.Scand.,27,2722 (1973)
 19730Ea R Oertel; Inorg.Chem.,12,3000 (1973)
 1973TKb S Takezawa,Y Kondo,N Tokura; J.Phys.Chem.,77,2133 (1973)
 1973VAa V Vasilev,S Aleksandrova; Zh.Neorg.Khim.,18,2055(E:1089) (1973)
 1973WHa H Wang,P Hemmes; J.Am.Chem.Soc.,95,5119 (1973)
 1972BZc A Bobrovsky,Y Zaremsky; Zh.Anal.Khim.,27,8,1472 (1972)
 1972COa E Constantinescu; Rev.Roumaine Chim.,17,1819 (1972)
 1972DAa A D'Aprano; J.Phys.Chem.,76,2920 (1972)
 1972DBa A Demortier,M de Becker,G Lepoutre; J.Chim.Phys.,69,380 (1972)
 1972HBa A Holmgren,P Beronius; Acta Chem.Scand.,26,3881 (1972)
 1972IWc T Iwachido; Bull.Chem.Soc.Jpn.,45,432 (1972)
 1972KRb K Kundu,A Rakshit,M Das; Electrochim.Acta,17,1921 (1972)
 1972KRc K Kundu,A Rakshit,M Das; J.Chem.Soc.,Dalton Trans.,381 (1972)
 1972SKb P Skabichevskii; Zh.Fiz.Khim.,46,532(E:310) (1972)
 1972WFa H Wada,Q Fernando; Anal.Chem.,44,1640 (1972)
 1971BCa B Barker,J Caruso; J.Am.Chem.Soc.,93,1341 (1971)
 1971BHa H Brookes,M Hotz,A Spong; J.Chem.Soc.(A),2410 (1971)
 1971CBc D Cogley,J Butler,E Grunwald; J.Phys.Chem.,75,1477 (1971)
 1971CSa J Carr,D Swartzfager; Anal.Chem.,43,1520 (1971)
 1971CSb J Carr,D Swartzfager; Anal.Chem.,43,583 (1971)
 1971ENa D Evans,J Nadas,M Matesisch; J.Phys.Chem.,75,1708 (1971)
 1971ETa D Evans,J Thomas,J Nadas,M Matesisch; J.Phys.Chem.,75,1714 (1971)
 1971FRa H Frensdorff; J.Am.Chem.Soc.,93,600 (1971)
 1971HMb P Henry,O Marks; Inorg.Chem.,10,373 (1971)
 1971INb D Irish,D Nelson,M Brooker; J.Chem.Phys.,54,654 (1971)
 1971PGa R Paul,D Gill,J Singla,S Narula; Indian J.Chem.,9,63 (1971)
 1970ALA F Accasina,R de Lisi,M Goffredi; Electrochim.Acta,15,1209 (1970)
 1970BIa P Beronius,U Isacsson,A Nilsson; Acta Chem.Scand.,24,189 (1970)
 1970DCa G Demange-Guerin,A Caillet; Compt.Rend.,271C,189 (1970)
 1970MSa S McLaughlin,G Szabo,G Eisenman et al; 14th.Bio.Soc.Baltimore,p.96a (1970)
 1970PMc N Poluektov,S Meshkova,E Melenteva; Zh.Anal.Khim.,25,7,1314 (1970)
 1970SAb M Salomon; J.Electroanal.Chem.,25,1 (1970)
 1970SSa J Songstad,L Stangeland; Acta Chem.Scand.,24,804 (1970)
 1970SSb K Sano,M Sakuma,S Motomizu et al; Bull.Chem.Soc.Jpn.,43,2457 (1970)
 1969BEa P Beronius; Acta Chem.Scand.,23,1175 (1969)
 1969BEb J Bessiere; Bull.Soc.Chim.Fr.,3353 (1969)
 1969DPa A Dill,O Popovych; J.Chem.Eng.Data,14,156 (1969)
 1969MBf L Mukherjee,D Boden; J.Phys.Chem.,73,3965 (1969)
 1969NSa G Nichugovskii,V Shvedov; Zh.Neorg.Khim.,14,299(E:156) (1969)
 1969SBe B Sesta,M Berardelli; Ricerca Sci.,39,795;803 (1969)

1969SMe D Singh,A Mishra; Indian J.Chem.,7,86 (1969)
 1969SSi E Sinyavskaya,Z Sheka,K Yatsimirskii; Zh.Neorg.Khim.,14,940;2083;3081
 (1969)
 1968DZa V Dziomko,S Zelichenok,I Markovich; Zh.Anal.Khim.,23,2,170 (1968)
 1968HBb R Huston,J Butler; J.Phys.Chem.,72,4263 (1968)
 1968MKa L Mukherjee,J Kelly,W Baranetsky,J Sica; J.Phys.Chem.,72,3410 (1968)
 1968SRd K Srinivasan,G Rechnitz; Anal.Chem.,40,509 (1968)
 1968TFb M Tanaka,S Funahashi,K Shirai; Inorg.Chem.,7,573 (1968)
 1967BHC J Butler,R Huston,P Hsu; J.Phys.Chem.,71,3294 (1967)
 1967CIa R Carroll,R Irani; Inorg.Chem.,6,1994 (1967)
 1967CKa M Chantooni,un,I Kolthoff; J.Am.Chem.Soc.,89,1582 (1967)
 1967CMc R Carroll,R Mesmer; Inorg.Chem.,6,1137 (1967)
 1967MIc S Manahan,R Iwamoto; J.Electroanal.Chem.,13,411 (1967)
 1967PBa J Perichon,R Buvet; Bull.Soc.Chim.Fr.,3697 (1967)
 1967RMe A Rozen,A Mikhailichenko; Zh.Neorg.Khim.,12,741 (1967)
 1967SMb D Singh,A Mishra; Bull.Chem.Soc.Jpn.,40,2801 (1967)
 1967SSe R Sundaresan,S Saraiya,A Sundaram; Curr.Sci.,36,255 (1967)
 1967WMa J Watters,S Matsumoto; J.Inorg.Nucl.Chem.,29,2955 (1967)
 1966AKa M Abramson,R Katzman,H Gregor,R Curci; Biochemistry,5,2207 (1966)
 1966AMb G Atkinson,Y Mori; J.Chim.Phys.,45,4716 (1966)
 1966CKa A Campbell,E Kartzmark,B Oliver; Can.J.Chem.,44,925 (1966)
 1966LCa E Luksha,C Criss; J.Phys.Chem.,70,1496 (1966)
 1966MBb W Masterton,L Berka; J.Phys.Chem.,70,1924 (1966)
 1966MWb S Minc,L Werblan; Roczn.Chem.,40,1537;1753 (1966)
 1966NSa G Nichugovskii,V Shvedov; Radiokhim.,8,118 (1966)
 1966SAa L Savedoff; J.Am.Chem.Soc.,88,664 (1966)
 1966SMc D Singh,A Mishra; Indian J.Chem.,4,308 (1966)
 1966WFa H Waki,J Fritz; J.Inorg.Nucl.Chem.,28,577 (1966)
 1965BCa J Botts,A Chashin,H Young; Biochemistry,4,1788 (1965)
 1965BFb I Bellobono,G Favini; Ann.Chim.(Italy),55,32 (1965)
 1965DGA J Dunnett,R Gasser; Trans.Faraday Soc.,61,922 (1965)
 1965LIa D Luehrs,R Iwamoto,J Kleinberg; Inorg.Chem.,4,1739 (1965)
 1964AMa D Archer,C Monk; J.Chem.Soc.,3117 (1964)
 1964KLa O Kolling,J Lambert; Inorg.Chem.,3,202 (1964)
 1964KYa Y Kurokawa,N Yui; J.Chem.Soc.Jpn.,85,397 (1964)
 1964OHa H Ohtaki; Acta Chem.Scand.,18,521 (1964)
 1964PCa Personal Communication etc; Chem.Soc.Spec.Publ.,no.17 (1964)
 1964PSH P Protsenko,O Shokina,N Chekhunova; Zh.Fiz.Khim.,38,1857 (1964)
 1964RZa G Rechnitz,S Zamochnick; Talanta,11,1061 (1964)
 1964SHb C Stephenson,H Hopkins,C Wulff; J.Phys.Chem.,68,1427 (1964)
 1964SUB M Suryanarayana; Curr.Sci.33,520 (1964)
 1964TRb T Talalaeva,A Rodionov,K Kocheshkov; Dokl.Akad.Nauk SSSR,154,174 (1964)
 1964WHa W Weaver,J Hutchison; J.Am.Chem.Soc.,86,261 (1964)
 1963EDa L Erikson,J Dembo; J.Phys.Chem.,67,707 (1963)
 1963IFa H Irving,J Frausto da Silva; J.Chem.Soc.,1144 (1963)
 1963IFb H Irving,J Frausto da Silva; J.Chem.Soc.,448;458;3308 (1963)
 1963IFc H Irving,J Frausto da Silva; J.Chem.Soc.,945 (1963)
 1963MSd D Morris,E Short; J.Inorg.Nucl.Chem.,25,291 (1963)
 1963PGb E Purlee,E Grunwald; J.Phys.Chem.,67,1364 (1963)
 1963PSa G Parfitt,A Smith; Trans.Faraday Society,59,257 (1963)

1963RSe R Ramette, J Spencer; J.Phys.Chem., 67, 944 (1963)
 1963SGd G Schwarzenbach, G Geier; Helv.Chim.Acta, 46, 906 (1963)
 1962Mwa S Minc, L Werblan; Electrochim.Acta, 7, 257 (1962)
 1962SHd A Shkodin; Zh.Fiz.Khim., 36, 595 (1962)
 1961AMc V Atkins, C Monk; J.Chem.Soc., 1817 (1961)
 1961KBa I Kolthoff, S Bruckenstein, M Chantooni; J.Am.Chem.Soc., 83, 3927 (1961)
 1961PSa P Proll, L Sutcliffe; Trans.Faraday Society, 57, 1078 (1961)
 1961WLa J Wright, W Lindsay, T Druga; US AEC - Report TID-4500 (WAPD-TM-204)
 (1961)
 1960CEa M Crutchfield, J Edwards; J.Am.Chem.Soc., 82, 3533 (1960)
 1960LRb N Lichtin, K Rao; J.Phys.Chem., 64, 945 (1960)
 1960Waa E Wanninen; Acta Acad.Aboensias, XXI, 17 (1960)
 1959KEb J Kenttamaa; Suomen Kem., B32, 9; 55; 68; 220 (1959)
 1959WOa J Wolhoff, J Overbeek; Rec.Trav.Chim., 78, 759 (1959)
 1958MLa W Marshall, F Loprest, C Secoy; J.Am.Chem.Soc., 80, 5646 (1958)
 1958VGa G Volkov, V Grinevich; Zh.Neorg.Khim., 3, 1968 (1958)
 1957GFa H Gregor, M Frederick; J.Polymer Sci., 23, 451 (1957)
 1957GKa J Graham, G Kell, A Gordon; J.Am.Chem.Soc., 79, 2356 (1957)
 1957HUa S Hughes; J.Chem.Soc., 634 (1957)
 1957LWa S Lambert, J Watters; J.Am.Chem.Soc., 79, 4262; 5606 (1957)
 1957WLa J Watters, S Lambert, E Loughran; J.Am.Chem.Soc., 79, 3651 (1957)
 1956BKa S Bruckenstein, I Kolthoff; J.Am.Chem.Soc., 78, 2974 (1956)
 1956SAC R Smith, R Alberty; J.Phys.Chem., 60, 180 (1956)
 1954Cha R Charles; J.Am.Chem.Soc., 76, 5854 (1954)
 1954DMb C Davies, C Monk; Trans.Faraday Society, 50, 132 (1954)
 1954FUa W Fernelius, L van Uitert; Acta Chem.Scand., 8, 1726 (1954)
 1954GMb F Gimblett, C Monk; Trans.Faraday Society, 50, 965 (1954)
 1954JGa M Jones, E Griswold; J.Am.Chem.Soc., 76, 3247 (1954)
 1953BJa J Bjerrum, C Jorgensen; Acta Chem.Scand., 7, 951 (1953)
 1953SEa T Smith, J Elliott; J.Am.Chem.Soc., 75, 3566 (1953)
 1952CSa K Cruse, H Stohr; Z.Elektrochem., 56, 561 (1952)
 1952FYa W Fyfe; J.Chem.Soc., 2018; 2023 (1952)
 1952LAB W Latimer; "Oxidation Potentials", Prentice Hall, NY (1952)
 1950Wia A Willi; Diss.Univ.Zurich (1950)
 1948SBa G Schwarzenbach, W Bierdermann; Helv.Chim.Acta, 31, 331; 456; 678 (1948)
 1947SAa G Schwarzenbach, H Ackermann; Helv.Chim.Acta, 30, 1798 (1947)
 1947SWa G Schwarzenbach, A Willi, R Bach; Helv.Chim.Acta, 30, 1303 (1947)
 1945SKb G Schwarzenbach, E Kampitsch, W Beidermann; Helv.Chim.Acta, 28, 828 (1945)
 1942DMA L Darken, H Meier; J.Am.Chem.Soc., 64, 621 (1942)
 1941BJa J Bjerrum; Thesis, repr.1957, P.Haase & Son, Copenhagen (1941)
 1930RDa E Righellato, C Davies; Trans.Faraday Society, 26, 592 (1930)
 1928HEa H von Halban, J Eisenbrand; Z.Phys.Chem., 132, 401; 433 (1928)
 1923Lra G Lewis, M Randall; Thermodynamics, McGraw-Hill., p.417 (1923)

EXPLANATORY NOTES

DATA Flags are :-

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS

H Data for THERMOCHEMICAL quantities
M Data for TERNARY Complexes

EVALUATION Flags are :-

T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC
R or IUP=R signifies EVALUATION RATING = Recommended by IUPAC

END