```
Software version = 5.81 Data version = 4.62
Experiment list contains 914 experiments for
(no ligands specified)
Metal : Al+++
(no references specified)
(no experimental details specified)
************************************
               HL
                   Electron
                                (442)
e-
Electron:
          Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ vlt oth/un hi 100% U
                                      1971BSe (301) 1
                           K(Al + 2Al(1)=3Al+)=8.1
Medium: Na3AlF6(l); units of B?; 1015 C
                                      1952LAb (302) 2
     oth none 25°C 0.0 U
                           K(Al+3e=Al(s))=-84.3(-1660 \text{ mV})
**********************************
             H3L Arsenate
As04---
                              CAS 7778-39-4 (1557)
Arsenate:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      oth oth/un 25°C 0.0 U
                                      1990SAa (1128) 3
                           *K(AlAsO4(s)+H=Al+HAsO4)=-4.70
Calculated from thermodynamic data.
-----
      sol oth/un 22°C var U
Al+++
                                      1956CHc (1129) 4
                          Kso(All)=-15.80
***********************************
               HL Borate CAS 10043-35-3 (991)
B04H4-
Borate; B(OH)4-
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ nmr oth/un 25°C var C
                                      2004TSa (1301) 5
At 25 C by 27Al nmr. Medium: 0.02-0.2 m B(OH)3, pH 8.95.
K(A1(OH)4+B(OH)3=A1(OH)3BO(OH)2+H2O)=1.62.
Al+++
    sol oth/un 50°C var C T H
                                      2004TSa (1302) 6
Solubility of gibbsite or boehmite in 0.02-0.2 m B(OH)3, pH 8.95, 50-200 C
K(Al(OH)4+B(OH)3=Al(OH)3BO(OH)2+H2O)=1.58 (50 C), 1.46 (78), 1.25 (200).
                 sol none
              22°C 0.0 U
                                      1961SBc (1303) 7
                           Kso(Al(OH)3L3)?=-22.92
By spectrophotometry K1=7.62?, B2=14.64?, B3=20.0? B6=38.54?
**********************************
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SC-Database

```
Br-
              HL Bromide CAS 10035-10-6 (19)
Bromide:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ con non-aq -78°C 100% U T
                                    1977GPa (1717) 8
                          K(2AlBr3=AlBr4+AlBr2)=-6.6
Medium: CH3Br
______
Al+++ EMF non-aq 210°C 100% U T H
                                    1973TDa (1718) 9
                          K(2A14=A12L7+L)=-4.40
Medium: (Na,Al)Br. DH(K)=20.1 kJ mol-1; K=-4.30(225 C), -4.15(240 C) m units
______
Al+++ nmr oth/un 26°C var U HM 1972JOb (1719) 10
                          K(2A1C13L=A1C14+A1C12L2)=-0.5
                          K' = -0.1
                          K''(2A1C1L3=A1C12L2+A1L4)=-0.4
K': 2AlCl2L2=AlCl3L+AlClL3. DH(K)=0.2 kJ mol-1; DH(K')=-1.5; DH(K'')=-0.3
Al+++ con non-ag 21°C 100% U I
                                    1972SVa (1720) 11
                          K(A12L6=A1L2+A1L4)=-22.14
Medium: n-heptane. In benzene: K=-16.12
______
Al+++ con non-ag 25°C 100% U
                                    1964WEa (1721) 12
                          K3=4.9
                          K4=3.3
                          K(2AlBr3=A12Br6)=0.0
Medium:PhNO2
**********************************
CO3--
             H2L Carbonate CAS 465-79-6 (268)
Carbonate:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ gl NaCl 25°C 3.00M C I
                                    1987HSa (3141) 13
                          B(-4,2,1)=-16.61
                          B(-5,3,1)=-18.39
B(p,q,r); pH+qAl+rCO2(g)=HpAlq(CO2(g))r
Al+++ gl NaCl 25°C 0.60M C
                                    19810Fa (3142) 14
                          B(-4,2,1)=-20.41
                          B(-5,3,1)=-22.74
B(p,q,r): pH + qAl + rCO2(g) = HpAlq(CO2)r
**************************
             H3L Cyanocobaltate (5470)
Hexacyanocobaltate; [Co(CN)6]---
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      con none 25°C 0.00 U K1=4.30
                                   1971KKf (3485) 15
```

```
By kinetics, (K1out/K1)=-1.3
*************************************
             H3L
                  Ferricyanide
                              (2491)
Hexacyanoferrate (III); Fe(III)(CN)6---
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                       K1=4.30
      kin KNO3 25°C 0.01M C
                                   1983KLa (3629) 16
Method: stopped flow, by conductivity measurement.
********************************
                  Chloride
C1-
              HL
                            CAS 7647-01-0 (50)
Chloride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp non-aq 25°C 100% C IH K1=1.73 B2= 2.20 1998UKb (4443) 17
Medium: DMF, 0.2 M Et4NClO4. Also data for DMA, 0.2 M Bu4NClO4.
By calorimetry, DH(K1)=5.1 kJ mol-1, DH(B2)=24.4, DH(B3)=34.
Al+++ EMF non-aq 40°C 100% U
                                    1985KOa (4444) 18
                         K(A12C17+C1=2A1C14)=-17.0
Medium: N-1-butylpyridinium Cl. In 1-Me-3-ethylimidazolium Cl, K=-17.1
------
Al+++ con non-aq -78°C 100% U T H
                                    1977GPa (4445) 19
                          K(2A1C13=A1C14+A1C12)=-2.48
Medium: CH2Cl2, at -78 and 0 C. DH=-15 kJ mol-1,DS=-125 J K-1 mol-1
______
Al+++
     oth oth/un 550°C 100% U
                                    1974CHb (4446) 20
                         B4=5.60(x units)
Medium: (K,Al)Cl. Method: Raman
______
Al+++ EMF non-ag 300°C 100% U
                                    1974IKa (4447) 21
                         K(A12L6+L)=7.5
                         K(2A1L4=A12L7+L)=-5.5
Medium: (K,Al)Cl; x units. 300-450 C
______
Al+++ EMF non-aq 300°C 100% U
                                    1974IKa (4448) 22
                          K(A12L6+L)=9.5
                         K(2A1L4=A12L7+L)=-5.5
Medium: (Cs,Al)Cl; x units. 300-450 C
______
Al+++ EMF non-aq 300°C 100% U
                                    1973BBc (4449) 23
                         K(2A1L4=A12L7+L)=-7.83
Medium: (K,Al)Cl(51.7% KCl); m units
______
                                    1973BJc (4450) 24
Al+++ EMF non-aq 175°C 100% U T H
                         K(A1L3+A1L4)=4.38
Medium: (Na,Al)Cl;m units. DH=-54.4 kJ mol-1,DS=-38 J K-1 m-1(200 C), K=4.11
(200 C),3.53(250 C),3.00(300 C),2.60(355 C). K4=10.8(200 C)
______
```

```
1973SSi (4451) 25
Al+++ EMF non-aq 450°C 100% U
                            B4 = 25
                            K(A12L6+L)=5.70
                            K(2A1L4=A12L7+L)=-6.24
Medium: (Na,Al)Cl
Al+++ EMF non-aq 175°C 100% U T H
                                  1973TDa (4452) 26
                           K(2A1L4=A12L7+L)=-5.50
Medium: (Na,Al)Cl. DH(B)=28.0 kJ mol-1. K=-5.32(190 C), -5.00(210 C),
-4.87(225 C), -4.75(240 C) m units
-----
                       1972FKb (4453) 27
Al+++ EMF non-aq 175°C 100% U
                           K4=11.5
                           K(A1L3+A1L4)=4.4
Medium: (Na,Al)Cl
-----
                                1972KOa (4454) 28
Al+++ oth non-aq 300°C 100% U I
                           K(Co(A12C17)2+4C1)=6.5
Medium: molten (LiAl)Cl. K: (Co(Al2Cl7)2+4Cl=CoCl4+2Al2Cl7.
K=1.2(in (NaAl)Cl); 19.1((in KAl)Cl); 18.7(in (RbAl)Cl); 19.7(in (CsAl)Cl)
______
Al+++ EMF non-aq 300°C 100% U T 1972KOa (4455) 29
                           K(Al2L6+2AlL4=2Al2L7)=1
Medium: (xK,(1-x)Al)Cl; 0.35< x< 0.45
______
Al+++ oth non-aq 170°C 100% U
                                      19710Ra (4456) 30
                           K(A12L6+2A1L4=2A12L7)=2.1
Medium: (K,Al)Cl; m units. 170-240 C. Method: Raman. Error in abstract ?
                    Al+++ oth oth/un ? var U K2=0.5 1971SCc (4457) 31
                           K3 = -2.7
Method: ionophoresis
______
Al+++ EMF non-ag 175°C 100% U T
                                      1971TMa (4458) 32
                            k(2A1L4=A12L7+L)=-7.1
Medium: (Na,A1)Cl;K=-6.3(250 C),-5.7(300 C),-5.3(350 C),-5.0(400 C)
Al+++ con non-aq 25°C 100% U
                                     1970MLa (4459) 33
                           K = 5.0
Medium: CH3COCl. K: AlL3+CH3COL=CH3CO++AlL4
______
Al+++ oth non-aq 289°C 100% U
                                      1969JSb (4460) 34
                            K(2A1L4=A12L7+L)=-3.52
Medium: KAlCl4; m units. Method: gas chromatography
______
Al+++ con non-aq 25°C 100% U
                                      1966WIa (4461) 35
                           K3=4.45
                            K4 = 3.04
                            K(2A1L3+A1L2)=3.04
                            K(A12L5+L)=2.95
```

```
Medium: PhNO2
************************
                    Fluoride
                               CAS 7644-39-3 (201)
Fluoride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ nmr R4N.X 5°C 0.60M C
                            K1=6.42 B2=11.83 2000BTa (6696) 36
                            K3=3.99
                            K4 = 2.50
                            K5=0.84
Method: 19F nmr and potentiometry. Medium: NMe4Cl
In 3M KCl at 25 C: K1=6.35, K2=5.25, K3=4.11.
Al+++ sp oth/un 23°C 0.10M U
                                       1994KGa (6697) 37
                            Keff=10.7
Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=5.34 with HCl
______
Al+++ nmr oth/un 21°C var U
                                       1991NMa (6698) 38
                            K(Al(NDP)+F)=4.1
                            K(Al(NDP)F+F)=3.1
                            K(Al(NDP)F2+F)=1.8
K(Al+NDP)=7.8, K(AlF+NDP)=5.8, K(AlF2+NDP)=4.0, K(AlF3+NDP)=2.2. NDP=
nucleoside diphosphates: guanosine diphosphate or adenosine diphosphate.
______
                             K1=6.8 B2=13.04 1988JHa (6699) 39
Al+++ ISE non-ag 185°C 100% M
                            B3=19.7
                            B4=24.92
                            B5=28.79
Medium: molten KSCN. K1=mol-1 kg, B2=mol-2 kg2 etc.
______
       ISE KNO3 25°C 0.10M C K1=4.92 B2= 8.46 1988YYa (6700) 40
Method: fluoride ion selective electrode
______
Al+++
      sol oth/un 50°C var M M
                                       1987SMe (6701) 41
                            K(A1F3(s)=A1+3F)=-21.2
Also mixed hydroxo-fluoro complexes. Solubility using pH and pF electrodes
------
      ISE KNO3 25°C 0.10M C M K1=6.40 B2=11.64 1987YHa (6702) 42
                            K3=3.86, K4=2.7
K(A1A+F) = 5.41(H3A=NTA), 5.53(H3A=HEDTA), 4.95(H4A=EDTA), 3.14(H4A=CDTA)
                 Al+++ gl NaNO3 25°C 0.10M M TI
                                       1986COa (6703) 43
                            B(Al(OH)F)=15.58
                            B(A1(OH)F2)=20.0
                            B(A1(OH)3F)=30.01
                            K(AlOH+F)=6.13
K(AlOH+F)=5.97(35 C), 5.88(50 C); K(AlOH+2F)=10.36, 10.07(35 C), 9.88(50 C);
K(Al(OH)4+F)=-3.73, -3.34(35 C); I=1.0 M, K(Al(OH)3F+OH)=-3.43
```

```
ISE KNO3 25°C 0.10M C M K1=6.40
                                          B2=11.64 1986YUa (6704) 44
Al+++
                                 K3=3.86
                                 K4=2.75
                                 K(Al(edta)+F)=4.95
                                 K(Al(cdta)+F)=3.14
Method: F ion-selective electrode. K(AlHA+F)=5.7, K(AlA+F)=4.5,
K(AlFA+F)=3.9. H3A is citric acid.
______
        gl KNO3 25°C 0.50M M
                                              1985TZb (6705) 45
                                 K(A1(OH)4+L=A1(OH)3L+OH)=-2.20
Al(OH)2L2 may form, but no Ga complexes, detected.
Al+++ oth none 25°C 0 U
                                             1984DYa (6706) 46
                                 K(A1+F=A1(OH)F+H)=1.45
                                 K(A1+2F=A1(OH)F2+H)=6.21
                                 K(A1+F=A1(OH)2F+2H)=-4.76
                                 K(A1+3F=A1(OH)F3+H)=-0.77
Recalc. of lit. data. K(Al+F=Al(OH)3F+3H)=-10.97.
------
Al+++ gl R4N.X 25°C 0.20M U I
                                  K1=6.46 B2=11.44 1982KMa (6707) 47
                                 B3=15.16
                                 B4=17.83
                                 B5=19.29
                                 B6=20.46
      cal NaClO4 25°C 0.5M C
                                              1975VKb (6708) 48
                                 DH(Al+L)=2.68 \text{ kJ/mol}
                                 DH(Al+3L)=5.31 \text{ kJ/mol}
                                 DH(Al+2L)=4.9 \text{ kJ/mol}
Also data for 35 C
Al+++ EMF non-ag 210°C 100% U
                                             1973TDa (6709) 49
                                 K(2A1C14+L=A12C16L+2C1)=-3.5
                                 K(2AlBr4+L=Al2Br6L+2Br)=-4.0
Medium: (Na,Al)Cl and (Na,Al)Br, m units
                                 K1=6.45 B2=11.66 1971AMb (6710) 50
Al+++ ISE KNO3 25°C 0.10M U I
                                 K3=3.79
                                 K4=3.18
K1=6.51, K2=5.29, K3=3.76, K4=3.05(I=0.05); 6.32, 5.16, 3.85, 3.30(I=0.2);
6.14, 5.09, 3.93, 3.68(I=0.5); At I=0(corr): K1=6.69, K2=5.35, K3=5.68, K4=2.75
______
Al+++ ISE KNO3 37°C 0.10M U TI
                                          B2=11.73 1971AMb (6711) 51
                                  K1=6.49
                                 K3 = 3.86
                                 K4=3.38
K1=6.71, K2=5.26, K3=3.92, K4=3.29(I=0.05); 6.39, 5.17, 3.86, 3.38(I=0.2);
6.29,5.09,3.84,3.43. At I=0(corr): K1=6.68,K2=5.34,K3=3.94,K4=3.29
______
Al+++ EMF NaClO4 25°C 1.0M U H
                                              1971WTa (6712) 52
```

```
*K(Al+HF=AlF+H)=3.15
```

*K(AlF+HF=AlF2+H)=1.99

```
Method: quinhydrone electrode. By calorimetry: DH(*K1)=3.10 kJ mol-1,
DS=128 J K-1 mol-1
______
Al+++ nmr oth/un -15°C var U
                                       1970MWa (6713) 53
                            K(2A1F=A1+A1F2)=-0.8
                            K(2A1F2=A1F+A1F3)=-0.9
                            K(2A1F3=A1F2+A1F4)=-1.4
Medium: Al(NO3,F). Method: nmr
      ISE R4N.X 25°C 0.01M U I
                            K1=6.65 B2=12.09 1969BAa (6714) 54
                            K3=3.92
                            K4=2.38
Medium: NH4NO3. K1=6.40, K2=5.19, K3=3.91, K4=2.42(I=0.1); K1=6.29, K2=4.97,
K3=3.73, K4=2.50(I=0.3); I=0(corr): K1=6.98, K2=5.62, K3=4.05, K4=2.38
_____
      oth oth/un 782°C 100% U
                                       1969RLa (6715) 55
                            K5.K6=1.22
Medium: molten (Li,Al)F. Method: combination of thermodynamic data
______
      oth oth/un 25?°C var U K1=6.08 B2=11.10 1964BSc (6716)
                                                       56
Method:refractometry.
______
Al+++ oth non-aq 999°C 100% U
                                       1962RBa (6717) 57
                            K5K6=1.7
Method freezing point, ca.1000 C. Medium: Na3AlF6(1), ionic fraction units
______
Al+++ oth non-aq 930°C 100% U
                                       1960BGe (6718) 58
                            K5.K6=1.45
Method: freezing point. Medium: Na3AlF6(liquid). 1008-930 C
oth non-aq 999°C 100% U T H
                                       1960FFb (6719) 59
                            K5K6=1.05
In liquid Na3AlF6, 1000-1090 C. DH(K5K6)=-92.0KJ mol-1. K5K6=0.96(1030 C),
0.82(1075 C), 0.80(1090 C). Method: density
      oth non-aq 999°C 100% U
Δ1+++
                                       1960ROa (6720) 60
                            K5K6=0.75
Method: freezing point. Medium: Na3AlF3(1), mole fraction units
______
Al+++ oth non-ag 930°C 100% U
                                      1959BGh (6721) 61
                            K5K6=1.22
Method: freezing point. Medium: liquid Na3AlF6,930-1008 C. Ion fraction units
K5K6=1.4 to 1.5 in x units
Al+++ sp oth/un ? var U K1=6.4 1959BSg (6722) 62
______
Al+++ EMF KNO3 25°C 0.53M U I K1=6.16 B2=11.21 1959KGa (6723) 63
                            K3=3.91
```

In NH4NO3 K3=3.57, K4=2.64, K5=1.46, K6=0.04?

```
______
Al+++ cal oth/un 25°C 0.07M U H
                                 1959KGa (6724) 64
DH(K1)=4.4 kJ mol-1, DS=141 J K-1 mol-1. DH(K2)=3.9, DS=115; DH(K3)=0.8,
DS=80.3; DH(K4)=0.17, DS=51.9; DH(K5)=-1.5, DS=21
______
Al+++ cal none 25°C 0.0 U H
                                 1959SCe (6725) 65
DH(K1)=4.9 \text{ kJ mol-1}, DH(B2)=8.2, DH(B3)=9.1, DH(B4)=9.0, DH(B5)=9.5, DH(B6)=-5.2
______
Al+++ sol oth/un 25°C var U B2=9.06 1957TVa (6726) 66
______
Al+++ EMF none 25°C 0.0 U K1=7.00 1955PAa (6727) 67
-----
Al+++ cal none 25°C 0.0 U H
                                  1955PAa (6728) 68
At I=0 corr: DS(K1)=160J K-1 mol-1, DS(K2)=130, DS(K3)=84, DS(K4)=54, DS(K5)
=-8.4, DS(K6)=-25. Values also at I=0.07 M
Al+++ cal oth/un 25°C var U H
                                  1953LJa (6729) 69
DH(K1)=4.8 kJ mol-1, DS=130 J K-1 mol-1; DH(K2)=3.3, DS=110; DH(K3)=0.8, DS=
75; DH(K4)=1.2, DS=54; DH(K5)=-3.1, DS=21; DH(K6)=-6.5, DS=-13
______
Al+++ sp KNO3 ? 0.10M U K1=6.32 1950KLb (6730) 70
-----
Al+++ gl oth/un 18°C var U
                                 1949LAa (6731) 71
                        B6=ca.27
                       Ks(Al2F6(s)=Al+AlF6)=-9.4
Al+++ EMF KNO3 25°C 0.53M U K1=6.13 B2=11.15 1943B0a (6732) 72
                        K3=3.85
                        K4=2.74
                        K5=1.63
                        K6=0.47
B6=19.84. Method: quinhydrone electrode and redox
*********************************
HP03--
           H2L Phosphite CAS 13598-36-2 (6305)
Phosphite;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ nmr NaClO4 25°C 1.0M U K1=2.01 B2= 3.24 1999MHa (7502) 73
                        B(A12L2)=2.21
Method: Al nmr.
------
Al+++ nmr NaClO4 25°C 1.0M U
                                  1999MHa (7503) 74
                        K(Al+HL)=2.01
                        K(A1HL+HL=A1H2L2)=1.23
                        K(2A1HL=A12H2L2)=2.21
Method: nmr. L is H2PO3-.
```

```
nmr NaNO3 20°C 0.10M C K1=6.11
                                1991FWa (7504) 75
Method: 31P nmr.
************************************
                 Hypophosphite CAS 6303-21-5 (6304)
H2P02-
Hypophosphite;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ nmr NaClO4 25°C 1.0M U H K1=2.01 B2= 3.29 1999MHa (7634) 76
                        K3=1.26
                        B(A12L2)=1.11
Method: Al nmr. DH(K1)=9.3 kJ mol-1, DS=69 J K-1 mol-1. DH(K2)=7.1,
DS(K2)=48. DH(K3)=-0.3, DS(K3)=26. DH(A12L2)=46, DS(A12L2)=175.
Al+++ nmr NaClO4 25°C 1.0M U H K1=2.01 B2= 3.29 1999MHa (7635) 77
                        K3=1.26
                        K(2AlL=Al2L2)=1.11
Method: nmr. DH(K1)=9.3 kJ mol-1, DS=69 J K-1 mol-1.
DH(K2)=7.1, DS=48; DH(K3)=-0.3, DS=26; DH(Al2L2)=46, DS=175.
______
     nmr NaNO3 20°C 0.10M C K1=2.38
                                 1991FWa (7636) 78
Method: 31P nmr.
**********************************
I-
                          CAS 10034-85-2 (20)
                 Iodide
Iodide:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     con non-aq -78°C 100% U I
                                 1977GPa (7883) 79
                        K(2AII3=AII2+AII4)=-8.15
Medium: CH3I
************************************
MoO4 - -
            H2L
                 Molybdate
                            (443)
Molybdate;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaCl 25°C 0.60M C
                                  19890Ha (8710) 80
                       B(A1H6L6)=50.95
-----
     EMF oth/un 25°C U
                                 1971GLa (8711) 81
                       B6=ca.19
**********************************
            HL Hydroxide
                           (57)
Hydroxide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sol oth/un 50°C var C T H
                                 2004TSa (10887) 82
```

```
Ks(Al(OH)3+OH=Al(OH)4)=-0.83
Solubility of gibbsite or boehmite in NH3/NH4Cl at 50-200 C. At 78 C:
K's(AlOOH+OH+H2O=Al(OH)4)=-0.88; at 150 C, K's=-0.45, at 200 C,-0.11
______
Al+++ gl KCl 25°C 0.10M C I
                                             2001DJa (10888) 83
                                *K1=-4.81
                                *B(3,4)=-13.82
                                *B3=-14.17
                                *Ks(Al(OH)3)=10.38
Medium: 0.10 M LiCl, 0.005 M CTAB. Kw=-13.12. In 0.10 M LiCl, 0.001 M
Tiron, Kw=-13.40. In 0.010 M LiCl, Kw=-13.50
______
Al+++ sol NaCl 100°C 0.10M C
                                             2001PBa (10889) 84
                                *Kso(AlOOH)=4.20
                                *B4=-21.16
Solubility of boehmite in 0.1-5.0 m NaCl at 100-290 C, using a Pt/H2 conc
cell. At I=0, *Kso=3.46. Ks(AlOOH+2H2O=Al(OH)4+H)=-12.76 (-13.02 at I=0)
______
                             2001PBb (10890) 85
Al+++ sol NaCl 25°C 0.03M C
                                *B2 = -10.73
                                *B3=-15.46
                                *B4=-22.78
Calc from solubility of synth boehmite in 0.03 m NaCl using a Pt/H2 conc'n
cell. Data for at 100-290 C. At 100 C, *B2=-6.76, *B3=-10.95, *B4=-16.48.
.....
Al+++ gl NaCl 37°C 0.15M C
                                             2000GKb (10891) 86
                               *B4=-21.031
Al+++ gl KCl 30°C 3.0M C
                                             2000STc (10892) 87
                                *B(2,2)=-6.68
                                *B(3,6)=-20.90
                                *B(13,32)=-104.45
                                *B(13,35)=-117.78
                                _____
Al+++ gl NaClO4 25°C 1.0M C I
                                            1999CIb (10893) 88
                                B(1,1)=-5.48
                                B(1,2)=-10.3
                                B(2,2)=-8.0
                                B(3,4)=-13.47
B(13,32)=-104.8, B(pq):pAl+qH20=ALp(OH)(q-r)+qH
Al+++ sp NaClO4 25°C 8.0M U
                                            1998SCa (10894) 89
                                *B(4,3)=-11.7
                                *B(4,2)=-25.9
                                *B(5,4)=-35.4
                                *B(5,3)=-17.0
Medium: 8 M NaCl04. *B(6,5)=-45.1, *B(6,4)=-31.1, *B(6,3)=-16.9,
*B(7,5)=-40.7.
Al+++ gl NaCl 25°C 0.01M C
                                             19970Wa (10895) 90
```

```
*Kso(Al(OH)3)=10.35 (fresh)
                               *Kso(Al(OH)3)=9.0 (48 h)
Fresh indicates precipitate immediately after flashing reactants (10mM Al
and 30 mM NaOH) together (I=0.015 M).
Al+++ oth none 400°C 0.00 C T
                                            1995ANa (10896) 91
                               Kso(AlO1.5) = -5.666
                               K(A1(OH)3=A1(OH)4+H)=-3.359
From literature data on solubility of corundum (AlO1.5) in H2O and KOH.
Data for 400-700 C at 2000 bar.
______
       gl oth/un 25°C 0.10M C
                                           1995DJa (10897) 92
                               *K1=-5.62
                               *B2=-9.74
                               K(3A1+4H20=A13(0H)4+4H)=-13.7
Medium: LiCl
       sol NaCl 25°C 0.01M U T
                                            1994SHa (10898) 93
                               K(A1(OH)3(s)+3H=A1+3H20)=8.02
Gibbsite (Al(OH)3(s)) solubility measurements. Constant at I=0
______
Al+++ sol NaCl 25°C 0.10M C
                                            1994WPa (10899) 94
                               *Kso=8.31
                               *Kso=7.74 (I=0)
Gibbsite solubility study using H electrode. Data for 50-100 C, 0-5 M NaCl
extrapolated to 0 and 25 C. At I=0.1 M *K1=-5.31,*K2=-5.8,*K3=-6.8,*K4=-5.4
______
Al+++ sol oth/un 150°C var M TI
                                            1993BKa (10900) 95
                               *K1=-2.1
                               *B2=-4.8
                               *B3=-7.8
                               *B4=-13.6
Boehmite solubility study at 150,200,250 C and P=10 bars.
K(AlO(OH)(s)+3H=Al+3H2O)=1.53
______
Al+++
       sol none 170°C 0.0 C T
                                            1993CDc (10901) 96
                               K(A100H(s)+3H=A1+2H20)=0.81
                               K(A100H(s)+2H=A10H+H20)=-0.95
                               K(A100H+H=A1(OH)2)=-3.78
                               K(A100H+H20=A1(OH)3)=-6.76
AlooH is Boehmite. Data for 90-350 C. K(AlooH+2H2O=Al(OH)4+H)=-11.75.
*K1(Al)=-1.765, *K2=-2.82, *K3=-2.98, *K4=-4.99.
_____
       EMF NaCl 25°C 0.10M C TIH
Al+++
                                            1993PWa (10902) 97
                               *K1=-5.31
Data at I=0.1,0.3,1.0 and 5.0 M. 25-125 C. DH(*K1)=54.4 kJ mol-1.
H electrode. Using Pitzer evaluation. *K1(I=0)=-4.95
______
Al+++ sol none 80°C 0.0 C
                                            1992NLa (10903) 98
                               K(A1(OH)3(s)+3H=A1+3H2O)=5.00
```

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Al(OH)3(s) is gibbsite.
------
Al+++ sol NaCl 50°C 0.10M C TIH
                                       1992PWa (10904) 99
                            *Ks(Al(OH)3+3H)=6.919
Gibbsite solubility study using H electrode. I=0.1-5.0 M( NaCl), 30,50,70 C.
Pitzer ion interaction treatment of data. DH(*Kso)=-101.8 kJ mol-1
______
Al+++ EMF oth/un 25°C var C TIH
                                       1992WEa (10905) 100
                            K(Al(OH)4=Al(OH)3(s)+OH)=1.143
I=0.01-5.0 M, 6.4-80 C. Gibbsite solubility studies. DH(K)=-22.5 kJ mol-1
______
Al+++ gl NaCl 25°C 3.0M U
                                       1991MBa (10906) 101
                            *B(2,2)=-7.53
                            *B(2,4)=-16.50
                            *B(3,4)=-13.44
-----
                                  1991MBe (10907) 102
Al+++ gl NaNO3 25°C 3.0M C
                            *B(2,2)=-7.55
                            *B(2,4)=-16.41
                            *B(3,4)=-13.24
-----
                                1990MOa (10908) 103
Al+++ gl NaCl 25°C 0.60M C
                            *B(13,32)=-105.5
                            -----
Al+++ gl NaNO3 25°C 0.50M C
                                       1989DJa (10909) 104
                            *K1=-5.65
                            *B(2,2)=-7.03
                            *B(2,4)=-15.65
                            *B(3,4)=-12.60
Al+++ gl NaCl 25°C 0.60M C
                                       19890Ha (10910) 105
                            K(A1+3H20=A1(OH)3(s)+3H)=-10.5
                                       19890Ha (10911) 106
Al+++ gl NaCl 25°C 0.60M C
                           *Kso(Al(OH)3)=10.49 (4 h)
Precipitate aged for 4 hours.
Al+++ cal none 25°C 0.0 C
                                       1988HHc (10912) 107
                            B4=32.76
Heat capacity measurements on NaAl(OH)4 solutions and Al+++ solutions
at 10-55 C. At 0 C: B4=33.71; at 50 C: B4=32.19.
______
Al+++ gl NaCl 25°C 3.00M C
                                       1987HSa (10913) 108
                            *K1=-5.52
                            *B(3,4)=-13.96
                            *B(13,32)=-113.35
                           M 1987SMe (10914) 109
Al+++ sol oth/un 505°C var M
                            B(1,3,3)=30.69
                            B(1,2,2)=20.80
```

```
B(1,1,2)=16.24
B(1,y,x): Al(OH)4+yF+xH=Al(OH)(4-x)Fy+xH2O. Solubility using pH and pF elec.
-----
Al+++ gl NaCl 37°C 0.15M C
                               K1=8.577 1987VBa (10915) 110
                              B(A1L3)=26.138
                              B(A1L4)=29.044
                              B(Al3L11)=91.025
                              B(A16L15)=149.278
Constants also reported for data restricted to pH<8
Al+++ nmr none 20°C 0.0 U
                                          1985AEa (10916) 111
                              *K1=-4.93
Al+++ gl NaNO3 25°C 0.10M C
                                         1985BSa (10917) 112
                              *B1=-5.33
                              *B2=-10.91
                              *B(3,4)=-13.13
                              *B(13,32)=-107.47
                               19850Sb (10918) 113
Al+++ EMF NaCl 25°C 0.60M C
                             *Kso(Al(OH)3)=11.2 (20 h)
Precipitate aged for 20 h.
Al+++ dis oth/un 25°C var U T H
                                          1984CMc (10919) 114
                              *K1=-5.00
                              *B4=-22.20
20-70 C. DH(*K1)=49.4 kJ mol-1; DH(*B4)=177.6 kJ mol-1
                     -----
Al+++ oth none 25°C 0 U
                                         1984DYa (10920) 115
                              *K1=-5.0
                              *K2=-5.3
                              *K3=-5.9
                              *K4=-6.0
Recalc. of lit. data. B4=-22.24; *Kso(Gibbsite)=8.11, Ks(Al(OH)3=Al(OH)3)=
-8.13, Ks(Al(OH)3+2H)=3.12, Ks(Al(OH)3+H)=-2.22, Ks(Al(OH)3=Al(OH)4+H)=-14.13
______
Al+++ gl NaCl 25°C 0.60M C
                                         19830Sb (10921) 116
                              *B4=-23.46
------
Al+++ oth none 80°C 0.0 C T H
                                          1981CHa (10922) 117
                              Ks(AlOOH(s)+OH=Al(OH)4)=-0.70
Calculations based on literature solubility data for boehmite, AlOOH.
Ks(AlOOH(s)+2H2O=Al(OH)4+H)=-13.30; DH=51.9 kJ mol-1. Data for 80-300 C.
______
Al+++ gl NaCl 25°C 0.60M C
                                          19810Fa (10923) 118
                              *K1=-5.52
                              *B(3,4)=-13.57
                              *B(13,32)=-109.2
Al+++ gl oth/un 25°C 6.0M U
                                         1980BCa (10924) 119
```

```
*K1=-5.5
*B3=-10.1
*B(13,36)=-105
```

Additional method: 27Al nmr. Medium: 0.5 M AlCl3. Polymer is Al1304(OH)28

Al+++ sol none 25°C 0.0 U

1979MHa (10925) 120

*Kso=8.11

*Ks(Al(OH)3+2H)=3.12*Ks(A1(OH)3+H)=-2.02K(Al(OH)4+H)=14.0

Al+++ oth oth/un 20°C ? U

1979STa (10926) 121

Kso(Al(OH)3) = -32.60

K(Al(OH)3(s)=Al(OH)2+OH)=-12.6

*B2=-8.30

Medium: seawater. Method: Tyndallometry

Al+++ oth none 25°C 0.0 U

1977VLa (10927) 122

B3=33.96

-----1975TUa (10928) 123

Al+++ kin none 25°C 0.0 U I

*K1=-5.17*B(2,2)=-6.95

*B(13,32)=-100.7

Further data available for NaCl concentrations of 0.003 to 0.06M

B3=25.75

Al+++ sp NaCl04 25°C 0.10M U I K1=9.10 B2=17.65 1974NBe (10929) 124

K1=9.40, B2=18.27, B3=26.75(I=0.3). K1=9.70, B2=18.88, B3=27.62(I=0.5).

K1=10.4, B2=20.42, B3=30.16(I=1) Error in abstract?

sol R4N.X 25°C U I

1973CHc (10930) 125

Kso(Al(OH)3(s)=Al+3OH)=-30.55

Medium: NH4Cl. In LiCl, Kso=-33.15. In NaCl, Kso=-30.75. In KCl, Kso=-30.36.

In CaCl2, Kso=-31.00

Al+++ sol oth/un 23°C 0.10M U

Al+++ gl KCl 20°C 0.10M U

1972IYa (10931) 126

Kso(Al(OH)3(s)=Al+3OH)=-32.94

*Ks(Al(OH)3(s)+H2O=Al(OH)4+H)=-12.23 at 20 C

K1=12.3 1972SKa (10932) 127 B3=32.0

B4 = 36.6

Al+++ sol oth/un 30°C U 1971DBa (10933) 128

*Ks(A1L3(s)+H)=2.59

K(A1(OH)3(s)=A1(OH)3)=-3.92Ks(Al(OH)3+H2O=Al(OH)4)=-12.62

*Kso=11.40

```
Al+++ EMF KCl 62°C 1.00M U H
                                       1971MBc (10934) 129
                            *B(2,2)=-5.90
                            *B(3,4)=-10.74
DH(*B(2,2))=76.6 kJ mol-1 ,DS=113 J K-1 mol-1. DH(*B(3,4))=129.3, DS=176;
DH(*B(14,34))=1100, DS=1577
-----
Al+++ EMF KCl 99°C 1.00M U T
                                      1971MBc (10935) 130
                            *B(2,2)=-4.81
                            *B(3,4)=-8.20
                            *B(14,34)=-67.9
At 150 C: *B(2,2)=-3.95, *B(3,4)=-7.01, *B(14,34)=-55.7
Al+++ gl oth/un 25°C 0.07M U IH
                                      1971VPa (10936) 131
                            *K1(A1+H20=A10H+H)=-5.20
DH(*K1)=11.13 \text{ kJ mol}-1. *K1=-5.11(I=0.0025), -5.23(I=0.01), -5.20(I=0.02),
-5.30(I=0.03), -5.30(I=0.05)
_____
Al+++ gl none 25°C 0.00 U H
                                       1971VPa (10937) 132
                            *K1=-4.99
*DH(*K1)=11.42 kJ mol-1. Data also 10 C, 15 C, 20 C, 30 C, 35 C, 40 C
______
Al+++ sol none 25°C 0.00 U T H
                                       1970BSd (10938) 133
                           Ks(Al(OH)3(s)+OH=Al(OH)4)=-1.3
DH(Ks)=34.7 kJ mol-1. Ks=-1.05(35 C), -0.78(50 C), -0.63(60 C)(gibbsite)
______
Al+++ gl diox/w 25°C 55% U
                                       19700Ha (10939) 134
                            *B(2,2)=-6.95
                            *B(3,2)=-10.02
Medium: 55% w/w dioxan/H2O, 3 M LiClO4
______
Al+++ gl NaNO3 30°C 2.00M U T
                                       1969CBc (10940) 135
                            *B(8,4)=-27.0
                            *B(16,7)=-52.7
At 50 C, *B(4,8)=-25.5, *B(7,16)=-48.5
______
Al+++ gl none 30°C 0.00 U
                                       1969GFa (10941) 136
                            *K1=-4.61
                            *B(2,2)=-7.44
Al+++ kin none 15°C 0.00 U
                                       1969GFa (10942) 137
                            *K1 = -5.11
                            *B(2,2)=-8.03
______
                            K1=9.4 1969HEa (10943) 138
      oth none 50°C 0.0 U T
                            B4=32.3
Method:Literature.K1=9.5(60C),10.0(100C),10.8(150C),11.9(200C),13.1(250C)
14.7(300C),B4=32.2(60C+100C),33.0(150C),34.6(200C),36.7(250C),39.5(300C)
______
Al+++ oth none 60°C 0.0 U T
                                       1969HEa (10944) 139
```

*Kso=6.17

	timated data. *Kso=4.62(100 C) C), 0.76(300 C) (gibbsite)	
Al+++	sp NaClO4 25°C 0.10M U	K1=9.02 B2=17.59 1969NNc (10945) 140 B3=25.73
Al+++	sol oth/un 25°C 0.00 U	1969RPa (10946) 141 K(Al(OH)3(s)=Al(OH)2+OH)=-15.0 *Ks(Al(OH)3(s)+OH)=-14.40
	sol oth/un 25°C U T (20 C), -33.4(30 C)	1969SBf (10947) 142 Kso(Al(OH)3(s)=Al+3OH)=-33.7
Medium: A	oth oth/un 25°C dil U lCl3 dil. Method: Dissociation	
	kin NaClO4 25°C 1.00M U	1968SRc (10949) 144 *K1=-4.31
Al+++ Al(OH)3:	gl oth/un ? U gibbsite	1967FSb (10950) 145 Kso(Al(OH)3(s)=Al+3OH)=-31.8
Al+++	gl NaClO4 25°C 2.00M U	1965AVa (10951) 146 *B(2,2)=-7.07 *B(13,32)=-104.5
	con none 25°C 0.0 U	1965NTa (10952) 147 *K1=-4.5
	gl NaClO4 25°C 3.00M U	1964BIb (10953) 148 *B(7,17)=-48.8 *B(13,34)=-97.6
Al+++ Kso(Al(OH	sol none 25°C 0.0 U)3(s)=M+30H)=-32.43, *K1(Al+H2	1964PCa (10954) 149 K(Al(OH)3(s)+OH=Al(OH)4)=-0.68 *Kso=9.57 *Ks(Al(OH)3+2H=AlOH+2H2O)=5.27 *K(Al(OH)3+H=Al(OH)2+H2O)=1.01 O=AlOH+H)=-4.3
Al+++	gl NaClO4 50°C 3.0M U	1964PCa (10955) 150 *B(17,7)=-48.8 *B(34,13)=-97.6
Al+++	con none 25°C 0.0 U	1963FPc (10956) 151 *K1(Al+H2O=AlOH+H)=-5.02

```
Al+++ sol none 25°C 0.00 U
                                          1963RAa (10957) 152
                              *Ks=-12.45 for boehmite
                              *Ks=-13.84 for bayerite
                              *Ks=-14.57 for gibbsite
*Ks: A1(OH)3(s) + H2O=A1(OH)4 + H
-----
Al+++ sol none 25°C 0.00 U
                                          1962FPa (10958) 153
                              Kso(Al(OH)3(s)=Al+3OH)=-33.5
Al(OH)3:gibbsite
______
Al+++ con none 25°C 0.0 U
                                         1960FRa (10959) 154
                              *K1(Al+H2O=AlOH+H)=-5.02
                              Kso(Al(OH)3=Al+3OH)=-33.51
Gibbsite. Also pH and solubility
-----
Al+++ gl oth/un 25°C dil U
                                         1960G0b (10960) 155
                              *Ks(Al(OH)3+OH)=-12.6
                              *Ks(Al(OH)3+2H=AlOH+2H2O)=5.73
                              Ks(A1(OH)3=A1OH+2OH)=-22.28
                             ______
Al+++ sol none 25°C 0.0 U
                                         1958GTa (10961) 156
                              Ks(Al(OH)3(s)+OH=Al(OH)4)=-0.5
                              *Ks(A1(OH)3+H=A1(OH)2+H2O)=0.4
                              *Kso(Al(OH)3+3H=Al+3H2O)=9.04
                              Kso(Al(OH)3=Al+3OH)=-32.96
______
Al+++ gl none ? 0.0 U
                                          1957MOa (10962) 157
                              Kso = -33.45
______
Al+++ oth none 25°C 0.0 U
                                          1956DPa (10963) 158
                              *Kso=9.66(Al(OH)3 amorphous)
                              *Kso=8.55(alpha-Al203)
                              *Kso=7.98(AlOOH, boehmite)
                              *Kso=6.48(Al(OH)3,bayerite)
*Kso(Al(OH)3(s)+3H=Al+3H2O)=5.70(hydrargillite); *K(Al+4H2O=Al(OH)4+4H)=20.3
                               1955KEa (10964) 159
Al+++ oth none 25°C 0.0 U
                              *K1(A1+H20=A10H+H)=-4.96
                              *B(2,2)=-7.55
*B(2,2): 2Al+2H2O=Al2(OH)2+2H. Method: freezing point etc.
                              1955SCc (10965) 160
Al+++ EMF oth/un 20°C var U T
                              Kso(Al(OH)3) = -32.90
                              *Ks(Al(OH)3+OH)=-12.74
Kso=-31.72, *Ks=-12.87(30 C). Method: Sb electrode
Al+++ gl oth/un 20°C 0.60M U I
                                          1954FAa (10966) 161
                              *K1(Al+H2O=AlOH+H)=-5.97
                              *B(2,2)=-8.24
```

```
Medium: Ba(NO3)2. In 0.12 M Ba(NO3)2 *K1=-5.74, *B(2,2)=-8.06
______
Al+++ gl none 25°C 0.0 U T
                                       1954STa (10967) 162
                            *K1(A1+H20=A10H+H)=-4.98
*K1=-5.28(15 C),-5.15(20 C)
Al+++ gl none 25°C 0.0 U
                                      1953IYa (10968) 163
                            *K1(Al+H2O=AlOH+H)=-5.10
______
Al+++ gl oth/un 22°C var U
                                      1953KFa (10969) 164
                          Kso(Al(OH)3) = -29.92
_____
Al+++ EMF none 18°C 0.0 C I
                                       1950AFa (10970) 165
                            Kso(Al(OH)3) = -31.7
Method: H electrode. By solubility, dil. soln., B3=26.96
______
                            1949LAa (10971) 166
Al+++ gl oth/un 18°C 0.01M U I
                            *K1(Al+H20=AlOH+H)=-4.60
                            Kso(Al(OH)3(s)=Al+3OH)=-33.8
At I=0 corr: *Ks(Al(OH)3+OH)=-13, *Kso(Al(OH)3)=-34.0
                                 1943CFa (10972) 167
Al+++ oth oth/un 20°C var U
                            Ks(Al(OH)3(s)+OH=Al(OH)4)=1.82
                            *K=-12.19
Al+++ oth oth/un ? var U
                                      1943KTa (10973) 168
                           Kso=ca.-32
Al+++ cal oth/un 20°C 30% U
                                       1942RWa (10974) 169
Medium: 30% w/w NaOH. DH(Ks(Al(OH)3(s)+OH))=15.9 kJ mol-1
At 77.3 C: DH=22.8. Al(OH)3 as hydrargillite
Al+++ gl oth/un 25°C dil U
                                      19380Ka (10975) 170
                           Kso(M(OH)3(s)=M+3OH)=-31.7
                            *Ks(Al(OH)3+OH)=-11.92
______
Al+++ con oth/un 25°C var U
                                      1934MAa (10976) 171
                            K4=2.78
                            *K4=-11.22
                        -----
Al+++ oth oth/un 18°C var U
                                       1933FMa (10977) 172
                            Kso=-12.2(fresh)
                            Kso=-13.8(aged)
______
      EMF oth/un rt var C
                                      1930TRa (10978) 173
                            *Ks(Al(OH)3(s)+OH)=-12.98
                             1929FRa (10979) 174
Al+++ sol oth/un 18°C 0.62M U
                            *Ks(Al(OH)3+OH)=-12.44(fresh)
                            *Ks=-13.89(after 24 h)
```

```
Al+++ con oth/un 25°C ? U T
                                    1920HEa (10980) 175
                          Ks(Al(OH)3+OH=Al(OH)4)=1.54
Al(OH)3 fresh. At 25 C: K=1.85, or K=0.18(25 C,crystlline)
                         1920KOb (10981) 176
Al+++ oth oth/un 15°C var U
                          Ks(Al(OH)3(s)+OH=Al(OH)4)=1.60
                          *Ks = -12.40
______
                         K1=9.49 1913KUa (10982) 177
     kin oth/un 100°C 0.01M U
                        *K1(Al+H2O=AlOH+H)=-2.88
______
Al+++ sol oth/un 19°C var U
                                    1911SLa (10983) 178
                         Ks(Al(OH)3+OH=Al(OH)4)=-0.74
By solubility K=-0.48
      oth oth/un 25°C var U I *K1(Al+H2O=AlOH+H)=-4.29
-----
                                    1908DEa (10984) 179
                          1899LEa (10985) 180
Al+++ kin oth/un 77°C var U T
                        *K1(Al+H2O=AlOH+H)=-4.12
*K1=-3.4(99.7 C). At 25 C, I=0 corr: *K1=4.85
********************************
           H3L Phosphate CAS 7664-38-2 (176)
Phosphate;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.20M C M K1=13.50
                                    2001LEa (13081) 181
                          B(AlHL)=17.60
                          B(A1H2L)=19.65
                          B(A1H-1L)=8.37
                          B(A12L)=17.42
B(A12H-2L)=11.05, B(A12H-3L)=6.9. For citrate: B(A1H2(cit)L)=28.46,
B(AlH(cit)L)=25.02, B(Al(cit)L)=19.68, B(AlH-1(cit)L)=12.03. Also 31P nmr.
______
Al+++ gl KCl 25°C 0.20M C
                                    1996AKa (13082) 182
                          B(A1HL)=17.60
                          B(A12L)=16.65
                          B(A12H-1L)=14.21
                          B(A12H-3L)=7.42
-----
Al+++ gl NaClO4 25°C 3.00M C I
                                    1996CIa (13083) 183
                          B(1,1,1)=0.13
                          B(1,2,2)=1.04
                          B(3,5,2)=-0.81
                          B(3,6,3)=0.23
B(p,q,r): pAl+rH3PO4=AlpH-q(H3PO4)r+qH. B(3,8,3)=-6.11, B(3,6,4)=2.62,
B(3,8,5)=0.98, B(3,6,1)=-9.60. Values at I=0.0 M calculated.
______
```

```
1996CIa (13084) 184
Al+++ gl NaClO4 25°C 3.0M M I
                            K(Al+H3L=AlH2L+L)=0.13
                            K(A1+2H3L=A1HL+2H)=1.04
                            K(3A1+2H3L=A13HL2+5H)=-0.92
                            K(3A1+3H3L=A13H3L3+6H)=0.15
K(3A1+3H3L=A13HL3+8H)=-6.17, K(3A1+4H3L=A13H6L4+6H)=2.69. At I=0: K(A1+H3L=A13H6L4+6H)=2.69.
AlH2L+H)=0.6, K(3Al+2H3L=Al3HL2+5H)=0.3, K(3Al+3H3L=Al3H3L3+6H)=2.5
______
                            K1=15.32 1991DEa (13085) 185
Al+++ gl NaCl 37°C 0.15M U
                            B(A1HL)=17.79
                            B(A1H2L)=20.93
                            B(A12L)=18.72
                            B(A12H-2L)=12.58
______
Al+++ nmr NaNO3 20°C 0.10M C K1=17.26 1991FWa (13086) 186
Method: 31P nmr.
______
Al+++ gl NaCl 37°C 0.15M C
                            K1=15.660 B2=20.88 1990DFa (13087) 187
                            B(A1HL)=19.072
                            B(A1H2L)=22.247
                            B(A1H-2L2)=15.796
                            B(A1H-3L2)=6.667
______
Al+++ gl NaCl 25°C 0.15M C
                                      1988JVa (13088) 188
                            B(A1HL)=23.25
                            B(A1H2L)=26.18
                            B(A1HL2)=37.95
                             1978RPa (13089) 189
Al+++ con oth/un 25°C 0.06M U
                           K(A1+H2PO4)=3.06
Al+++ gl NaClO4 25°C 0.10M U M
                                  1975RMa (13090) 190
                            K(A1+HPO4)=9.17
                            K(Al+citrate+HPO4)=19.29
                            K(A1+NTA+HPO4)=23.89
                            K(A1+Cys+HPO4)=15.66
                            1974FGc (13091) 191
Al+++ ix R4N.X ? 0.20M U
                            K(2A1+H3L=A12HL+2H)=-1.96
-----
Al+++ sol none 25°C 0.0 U
                                       1961TGa (13092) 192
Ks(K3Al5H-10H2L8(H2O)18)(taranakite)=-22.5 ? Ks((NH4)3Al5H-10H2L8(H2O)18)=
19.3 ?. L-Al complex neglected
-----
      sol none 25°C 0.0 U
                                       1959LPb (13093) 193
                            Ks(Al(H2L)(OH)2)=-30.5
Ks(Al(H2L)H-2(H2O)2)(variscite)=-2.48
______
Al+++ sol oth/un 25°C var U
                                       1957TVa (13094) 194
                            Kso(All) = -10.41
```

Al+++	sol none	? 0.0 U	1955KJa (13095) 195 Ks(Al(H2L)(OH)2)=-28.0
Al+++	sol oth/u	n 19°C var U	1951ZHa (13096) 196 Kso(AlL)=-18.24
Al+++	sol NaCl	? 0.05M U	1950CJa (13097) 197 Kso(Al(H2L)(OH)2)=-29.55
Also quin	nydrone ele		1931BDb (13098) 198 K(Al+H2L)=3 ? K(AlH2L+H2L)=2.3 K(Al(H2L)2+H2L)=2.3 K(AlHL+H)=2.1 0+H)=2.1 and others
P207 Diphospha	te; from (H	H4L Pyrophosph 0)2P0.0.P0(OH)2	nate CAS 2466-09-3 (198)
Metal	Mtd Mediu	m Temp Conc Cal Fla	ags Lg K values Reference ExptNo
Al+++	gl KCl		K1=13.74 B2=19.77 1996AKa (13560) 199 B(AlHL)=17.03 B(AlH2L)=18.69 B(AlH-1L)=7.41 B(AlHL2)=25.64
		25°C 0.15M C	K1=14.30 1988JVa (13561) 200 B(AlHL)=19.20 B(AlH2L)=22.79
P3010	-	**************************************	CAS 10380-08-2 (1001) 0.0.PO(OH)2
Metal	Mtd Mediu	m Temp Conc Cal Fla	ags Lg K values Reference ExptNo
Al+++	gl KCl	25°C 0.20M C	K1=13.15 B2=19.14 1996AKa (13836) 201 B(AlHL)=16.65 B(AlH2L)=18.07 B(AlH-1L)=6.53 B(AlHL2)=24.43
Al+++	gl NaCl	25°C 0.15M C	K1=17.31 1988JVa (13837) 202 B(AlHL)=20.98 B(AlH-1L)=11.72
Al+++	gl KNO3	35°C 0.10M U	1980KHc (13838) 203 K(AlL+thr)=7.55 K(AlL+ala)=7.51

```
K(AlL+pro)=8.63
K(AlL+val)=7.90
```

```
K(All+gly)=7.97. For tyrosine: K(All+HA)=7.95, *K(All(HA))=-7.15.
K(AlL+Hgly-gly)=4.18, *K(AlL(Hgly-gly))=-5.53. Data for other aminoacids.
*************************
             H3L
                           CAS 13566-25-1 (235)
Cyclotrimetaphosphate;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ cal oth/un 25°C 0.10M C H K1=3.05 1983GGb (13944) 204
Medium: 0.10 M HCl. DH(K1)=19.6 kJ mol-1, DS(K1)=124 J K-1 mol-1.
***********************************
P4012----
                           CAS 13598-74-8 (234)
Cyclotetrametaphosphate;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      cal oth/un 25°C 0.10M C H K1=3.29 1983GGb (13996) 205
Medium: 0.10 M HCl. DH(K1)=37.8 kJ mol-1, DS(K1)=190 J K-1 mol-1.
***********************************
            H2L Sulfide
S--
                           CAS 7783-06-4 (705)
Sulfide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     vlt oth/un 25°C 0.72M C I
                                   1999AVb (14306) 206
                         K(Al+HL)=13.0
Method: determination of free S-- by cathodic stripping voltammetry.
Medium: seawater, pH 8.0, S=35. Also data for S=21 and 10.5.
********************************
             HL
                 Thiocyanate CAS 463-56-9 (106)
Thiocyanate;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 25°C 7.5M C K1=0.18 2001SZa (14800) 207
Method: Raman spectroscopy. Medium: NaSCN.
------
Al+++ cal non-aq 25°C 100% C IH K1=2.1 B2= 3.60 1996TSa (14801) 208
                         K3=1.4
                         K4=1.0
Medium: N,N-Dimethylformamide,0.20 M Et4NClO4. Also data at 0.4 M Et4NClO4
DH(K1)=9.0 \text{ kJ mol-1}, DH(K2)=13, DH(K3)=0, DH(K4)=12.
______
      sp non-aq 25°C 100% U IH K1=2.66 1985PWa (14802) 209
Al+++
Medium: dimethylsulphoxide. K1 extrapolated to I = 0.0
______
Al+++ sp none 22°C 0.0 U T K1=0.42 1963VMa (14803) 210
**********************************
```

```
S04--
           H2L Sulfate CAS 7664-93-9 (15)
Sulfate;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
                                  Reference ExptNo
Al+++ sol none 25°C 0.0 C T H 1999PPb (15966) 211
Ks(Ca6[A1(OH)6]2(SO4)3=6Ca+2A1(OH)4+3SO4+4OH)=-44.9. Data for 5-75 C.
DH(Ks)=204.6 kJ mol-1, DS(Ks)=170 J K-1 mol-1. Solid phase is ettringite.
Al+++ EMF NaCl 50°C 0.10M C T H K1=2.3 B2= 3.90 1999RWa (15967) 212
Method: Pt/H2 electrode. Data for 0.3-1.0 M NaCl, 50-125 C. DH(K1)=
-10 kJ mol-1, DS=10 J K-1 mol-1; DH(B2)=10, DS=100. At I=0, K1=3.7, B2=5.6
______
Al+++ EMF none 25°C 0.0 C T H K1=1.72 1989JJb (15968) 213
Method: Hg/Hg2SO4 electrode. Data for 5-35 C.
DH(K1)=56.1 \text{ kJ mol-1}, DS(K1)=221 \text{ J K-1 mol-1}.
Al+++ EMF KCl 25°C 1.00M C T H K1=3.35 1988MMa (15969) 214
K1/mol-1 kg; DH(K1)=6.6 kJ mol-1, DS=86.2 J K-1 mol-1(25 C); 50 C: K1=3.59;
75 C: 4.08; 100 C: 4.66; 125 C: 5.34
______
Al+++ kin KNO3 25°C 0.01M C K1=3.19
                                   1983KLa (15970) 215
Method: stopped flow, by conductivity measurement.
______
Al+++ cal NaClO4 25°C 1.0M C T H K1=0.45 B2= 0.76 1982LMb (15971) 216
Data for 25-70 C. DH(K1)=29.0 kJ mol-1, DH(K2)=10.3. At 70 C, K1=1.11,
DH(K1)=28.4; K2=0.91, DH(K2)=47.3
______
Al+++ sol none 25°C 0.0 C
                                    1982NOa (15972) 217
                          Kso(alunogen)=-7.0
                          Kso(jurbanite)=-17.8
                          Kso(alunite)=-85.4
                          Kso(basaluminite)=-117.7
Method: derived from literature data. Alunogen:Al2(SO4)3.17H2O. Jurbanite:
Al(SO4)OH.5H2O. Alunite:KAl3(SO4)2(OH)6. Basaluminite:Al4(SO4)(OH)10.5H2O
______
Al+++ nmr oth/un ? var U
                                   1972AGa (15973) 218
                         K1in=-0.6
------
Al+++ sol none 25°C 0.0 U
                                    1972IYa (15974) 219
                          Kso(Al(OH)2.56(L)0.22)=-29.6
In paper, formula expressed as: Al(OH)2.66(L)0.22, printing error?
______
Al+++ kin oth/un 25°C dil U K1=3.3 1971KKa (15975) 220
                         K1out=3.12
                         K1in=-0.7
------
Al+++ sp NaCl04 25°C 5.0M U IH K1=1.20 1971KVa (15976) 221
                          K1out=1.07
I=1 K1=1.48; I=3 K1=1.16; I=0(corr) K1=3.89
```

```
Al+++ EMF none 15°C 0.0 U T H K1=1.75 B2=2.25 1970SPd (15977) 222
At 5 C: K1=1.60; 25 C: K1=1.90, K2=0.80; 35 C: K1=2.08, K2=1.05.
DH(K1)=25 \text{ kJ mol-1, } DH(K2)=46
                 Al+++ nmr oth/un 25°C var U
                                 1969AGa (15978) 223
                        B(A1+HL)=0.5
Method: N.M.R.
______
Al+++ cal none 25°C 0.0 U H K1=3.01 B2=4.90 1969IEa (15979) 224
DH(K1)=9.6 kJ mol-1, DS=89.5 J K-1 mol-1; DH(K2)=3.3, DS=47.2
______
Al+++ sol oth/un 20°C var U T
                                 1969SBf (15980) 225
                        Kso(Al(OH)2.5(L)0.25)=-29.5
Kso(Al(OH)2.5(L)0.25)=-29.3(25 C), -29.1(30 C)
-----
Al+++ sol none 0.0 U
                                 1969SIa (15981) 226
                        Kso=ca.33 (fresh)
                        Ks(A1(OH)2.5(L)0.25)=-28.6
_____
Al+++ oth oth/un var U K1=2.57 1969SMi (15982) 227
Method: coagulation
______
Al+++ con oth/un 25°C 0.0 U K1=3.73
                              1965NTa (15983) 228
-----
Al+++ kin oth/un 25°C 0.10M U
                                 1963BLa (15984) 229
                        K(Al(aq)+L(aq)=Al(H20)L)=1.28
                        K(Al(H20)L=AlL)=-1 to -2
Method: pressure jump
-----
Al+++ EMF NaClO4 25°C 0.60M U I K1=1.30 B2=2.30 1962BWa (15985) 230
Method: Pb electrode. At I=0 corr. K1=3.2, K2=1.0
______
Al+++ sp oth/un 30°C 0.0 U K1=2.04 1962NAc (15986) 231
**********************************
                Silicate
Si03--
            H2L
                         CAS 7699-41-4 (747)
Silicate; SiO2(OH)2--
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ gl KCl 25°C 0.10M C T H
                                 1996PSb (17194) 232
                       K(A1+H4SiO4=A1H3SiO4+H)=-2.38
Data for 90 and 150 C. DH(Al+H4SiO4)=66.66 kJ mol-1, DS(Al+H4SiO4)=
177 J K-1 mol-1.
-----
      nmr oth/un RT 4.0M C
                                 1995LMd (17195) 233
Ks(Al(OH)2SiO2(OH)(s)+2OH=HSiO3+Al(OH)4)=-3.0
Method: 27Al nmr. Medium: 4.0 M NaOH.
______
Al+++ gl NaClO4 25°C 0.0 M
                                 1994FLa (17196) 234
```

```
K(A1+H2L=A1HL+H)=-2.50
```

```
Al+++ sol NaCl 25°C 0.01M U T
                                          1994SHa (17197) 235
                              *Ks(imogolite)=13.04
*Ks(imogolite):K(Al2SiO3(OH)4(s)+6H=2Al+Si(OH)4+3H2O). Constant at I=0
                     -----
Al+++
       sol NaNO3 25°C 0.10M U
                                          1986MKa (17198) 236
                              *Ks(kaolinite)=7.42
*Ks(kaolinite):Al2Si2O3(OH)4(s)+6H=2Al+2Si(OH)4+3H2O. Constants at I=0 corr
                          oth none 60°C 0.0 U T
                                          1969HEa (17199) 237
                              *Ks(Al2Si2O5(OH)4+6H)=4.75
Method: estimated data.(kaolinite,Al2Si2O5(OH)4).
*Ks=2.27(100 C); -0.12(150 C); -1.72(200 C); -2.98(250 C); -4.02(300 C)
-----
       oth none 60°C 0.0 U T
                                          1969HEa (17200) 238
                              *Ks(Al2Si2O5(OH)4+6H)=5.63
Method: estimated data.(dickite, Al2Si2O5(OH)4).
*Ks=3.10(100 C); 0.69(150 C); -0.90(200 C); -2.13(250 C); -3.09(300 C)
_____
       oth none 60°C 0.0 U T
Al+++
                                          1969HEa (17201) 239
                              *Ks(Al2Si2O5(OH)4+6H)=8.03
Method: estimated data.(halloysite,Al2Si2O5(OH)4).
*Ks=5.21(100 C); 2.50(150 C); 0.67(200 C); -0.75(250 C); -1.89(300 C)
_____
        oth none 150°C 0.0 U T
                                          1969HEa (17202) 240
Method:estimated data.*Kso(K0.6Mg0.25Al2.3Si3.5010(OH)2+8H)=0.81,6.85(60 C),
3.82(100 C),-1.23(200 C),-2.87(250 C),-4.29(300 C),(illite)
______
    oth none 150°C 0.0 U T 1969HEa (17203) 241
Δ1+++
Method:est.data. *Ks(KFe3AlSi3010(OH)2+10H)=10.8. *Ks=18.7(60 C),14.7(100 C)
8.0(200 C), 5.8(250 C), 3.9(300 C), (annite).
    oth none 150°C 0.0 U T
                                          1969HEa (17204) 242
                              *Kso=-2.37
Method:estimated data. *Kso(montmorillonite):(K0.33Al2.33Si3.67010(OH)2(s)+
12.7H).Also at 60-300 C: 3.00(60 C); 0.31(100 C); -4.17(200 C); -6.9(300 C).
______
        gl oth/un 400°C dil U I
Al+++
                                          1961HMa (17205) 243
                              K = 3.0
K: Na-feldspar(s)+H=Na-mica,paragonite(s)+3SiO2(s)+Na. Also other equilibria
______
Al+++ oth oth/un 200°C var U
                                       1959HEa (17206) 244
                              K = 4.9
By chemical analysis. P=1000 atm. Data also for mica to kaolinite etc.
K(1.5K-feldspar+H=0.5K-mica+3SiO2(s)+K)=3.55(300C), 2.7(400C), 2.1(500C).
********************************
                 HL Formic acid CAS 64-18-6 (37)
Methanoic acid; H.COOH
```

Metal	Mtd	Medium	Temp	Conc Cal	Flags	Lg K va	ues.	Refe	rence	ExptNo	
				0.01M C onductivit				 L983KLa	(1758	88) 245	-
Al+++	gl	NaNO3	25°C	1.0M U		B2=2.02	1	 L976KIb	(1758	39) 246	-
Al+++	gl	NaNO3	25°C	1.00M U		K1=1.3	1	L975KIb	(1759	90) 247	-
	oth	oth/un	25°C	1.00M U		K1=0.56		76 1 9	73TRc	(17591	-) 248
Al+++	ix ****	oth/un *****	25°C ****	1.0M U ******** Methyl	*****	K1=1.78 ******	1 ******	L962TSa *****	(1759	2) 249	
Metal	Mtd	Medium	Temp	Conc Cal	Flags	Lg K va	ues.	Refe	 rence	ExptNo	-
	∣ elec	trode. I	Mediur	100% U n: MeOH, 1	L.0 M		1L=A12(H .L)3+H-1L .)2+H-1L= .)3+H-1L=	H-1L)3) _)=11.1 =10.5 =5.5	=42.0	· 74) 250 ·*****	
CH5O3P Methylpho	sphon	ic acid	H2L ; CH3	.P03H2		CAS	13590-71	L-1 (1	752)		
Methylpho			; CH3	.PO3H2 Conc Cal	 Flags		13590-71			ExptNo	_
Methylpho	Mtd	Medium	; CH3	Conc Cal			13590-71 .ues .ues .=2.33 =-3.91	Refe	 rence		-
Methylpho Metal Al+++ *********************************	Mtd gl	Medium KCl	; CH3 Temp 25°C	Conc Cal 0.20M C	***** ic aci	Lg K val K1=6.48 B(AlH-1L) B(AlH-2L)	13590-71esese2:12e2:33e3:91e3:81e3:81	Refe 3 3 19	 rence 96AKa *****	(18122	- -) 251
Methylpho Metal Al+++ ******** CH606P2 Methanedi	Mtd gl *****	Medium KCl ******	; CH3 Temp 25°C ****** H4L cid; (Conc Cal 0.20M C	***** ic aci	Lg K val K1=6.48 B(A1H-1L) B(A1H-2L) B(A1H-1L2 ************************************	13590-71 .ues .ues .=2.33 .=-3.91 ?)=5.8 .******* 1984-15-	Refe 3 19 *******	 rence 96AKa *****	(18122	- -) 251 *
Methylpho Metal Al+++ ******** CH606P2 Methanedi Metal Al+++	Mtd gl ***** phospl Mtd 	Medium KCl ***** honic ac Medium KCl	; CH3 Temp 25°C ***** H4L cid; (Temp 25°C	Conc Cal 0.20M C ******* Medroni CH2(PO3H2) Conc Cal 0.10M U	***** ic aci)2 Flags	Lg K val K1=6.48 B(A1H-1L) B(A1H-1L2 ************************************	13590-71esese2.33e-3.91 2)=5.8******* 1984-15eseseseses	Refe 3 19 ****** -2 (23 Refe 01 19	****** 84) rence 	(18122 ****** ExptNo (18274	- -) 251 * - -) 252
Methylpho Metal Al+++ ******** CH606P2 Methanedi Metal Al+++	****** phospl Mtd g1 *****	Medium KCl ****** honic ac Medium KCl KCl	; CH3 Temp 25°C ****** H4L cid; (Temp 25°C	Conc Cal 0.20M C ******* Medroni CH2(PO3H2) Conc Cal 0.10M U	***** ic aci)2 Flags 	Lg K val K1=6.48 B(A1H-1L) B(A1H-1L2 ************************************	13590-71esese2.33e3.91e3.8******** 1984-15eseseseseseseseseseseseseseseseseses	Refe 3 19 ****** -2 (23 Refe 01 19	 rence 96AKa ****** 84) rence 67KLa	(18122 ****** ExptNo (18274	- -) 251 * - -) 252
Methylpho Metal Al+++ ******** CH606P2 Methanedi Metal Al+++ ******** C2H2O3 Glyoxylic	Mtd gl ***** phospl Mtd gl *****	Medium KCl ****** honic ac Medium KCl ******	; CH3 Temp 25°C ***** H4L cid; (Temp 25°C	Conc Cal 0.20M C ******* Medroni CH2(PO3H2) Conc Cal 0.10M U	***** ic aci)2 Flags *****	Lg K val K1=6.48 B(A1H-1L) B(A1H-1L2 ************************************	13590-71esese2:33e3:91e3:81e3:81e3:82e3:92e3:82e3:83	Refe 3 19 ****** -2 (23 Refe 01 19 *******	 rence 96AKa ****** 84) 67KLa *****	(18122 ****** ExptNo (18274	- -) 251 * -) 252 *

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***********************************
C2H2O4
              H2L
                   Oxalic acid
                             CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2
                -----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++
      EMF NaCl 25°C 0.60M C
                         Μ
                                      2003BTa (18777) 254
                           B(AllF)=11.53
                           B(AllF2)=15.67
                           B(All2F)=15.74
                           B(AlL2F2)=19.09
Method: quinhydrone electrode and fluoride ISE.
Al+++ gl NaCl04 25°C 1.0M C I K1=6.03 B2=11.05 1999CIb (18778) 255
                           B3=15.02
                           B(2,2,2)=5.14; B(336)=-21.41
                           B((7,17,3)=-43.2; B(353)=-0.87
                           B(2,2,3)=-9.47
At I=0 (by SIT): K1=8.3, B2=14.1, B3=17.2
B(pqr):pAl+qH2O+rL=ALp(OH)(q-r)Lr+qH
                       -----
Al+++
      sol oth/un 80°C var U M K1=8.3
                                    1994FHa (18779) 256
                           B3=18.1
Alternatively data may be explained with formation of Al(OH)2L and Al(OH)L
-----
      sol NaCl 25°C 0.60M C
Al+++
                                      1986BHc (18780) 257
                           B(-9,3,1,0)=-21.87
                           B(-4,1,1,1)=-5.61
B(-9,3,1,0):3A1+H2L=A13(OH)7L(s)+9H; B(-4,1,1,1):Na+A1+H2L=NaA1(OH)2L(s)
-----
Al+++ gl NaCl 25°C 0.60M C
                                      1985S0a (18781) 258
                         Μ
                           B(-1,1,1)=1.40
                           B(-2,1,1)=1.43
                           B(-4,1,2)=1.85
                           B(-6,1,3)=1.26
B(-9,3,3) = -4.28; B(-10,2,4) = -4.62. B(p,q,r) for pH+qAl+r(H2L)=HpAlq(H2L)r
Al+++ gl NaCl 37°C 0.15M C
                           K1=5.02 B2=9.33 1982JCa (18782) 259
                           B3=12.41
                           B(A1HL)=6.63
                           B(AlH-1L2)=7.07
                           B(AlH-1L3)=9.52
                       -----
      nmr none 25°C 0.0 U
                           K1=7.18 B2=13.49 1977JBb (18783) 260
                          B3=17.53
______
      vlt NaClO4 25°C 1.00M U K1=4.90
                                     1970GMi (18784) 261
______
Al+++ gl NaClO4 25°C 1.00M U K1=4.85 1970GMi (18785) 262
______
```

Al+++		04 25°C 1.00M U	K1=6.06 B2=11.09 1968BCa (18786) 263 B3=15.12
Al+++		04 20°C 0.10M U	1963STc (18787) 264 B3=15.60
Al+++			K1=7.26 B2=12.11 1957BDc (18788) 265 K3=1.31
Al+++		32°C 1.0M U	
			B2=13 1949LAb (18790) 267 B3=16.3 ************
C2H3NO4 Nitroacet:		HL 2N.CH2.COOH	CAS 625-75-2 (2968)
Metal	Mtd Medi	um Temp Conc Cal Fla	ags Lg K values Reference ExptNo
Al+++ Medium: Ba	kin oth/u a(NO3)2	ın 18°C 0.20M U	K1=0.48 1949PEa (19206) 268 ************************************
C2H4O2 Ethanoic a	acid; CH3.		id CAS 64-19-7 (36)
Metal	Mtd Medi	um Temp Conc Cal Fl	ags Lg K values Reference ExptNo
	sol oth/unts at 60-2		K1=2.33 B2=4.10 1994BCa (19881) 269
Measuremer	nts at 25-1		K1=2.02 B2=3.5 1994PBa (19882) 270 M. Pitzer formalism and equations.
		un 80°C var U measurements.Const	K1=2.9 B2=4.8 1991FEa (19883) 271 ants at I=0
		25°C 0.60M C	1989MOa (19884) 272 K(2Al+HL=Al2(OH)2L+3H)=-7.98
Al+++	gl NaNO3	3 25°C 1.00M U	K1=1.4 1975KIb (19885) 273 ************************************
C2H5N02			CAS 56-40-6 (85)
Metal	Mtd Medi	um Temp Conc Cal Fl	ags Lg K values Reference ExptNo
Al+++	gl NaCl	37°C 0.15M C	2002DCa (21487) 274 B(Al2H-2L)=0.361

```
K1=5.91
Al+++ gl KCl 25°C 0.20M C
                                  1997KSa (21488) 275
                        B(AlH-1L)=1.08
                        B(A12H-1L)=4.35
Al+++ gl NaNO3 25°C 0.50M C
                                  1989DJa (21489) 276
                        B(A12H-2L2)=6.56
                        B(A1H-3L)=-7.53
_____
     oth NaClO4 35°C 0.01M U
                                  1984YSa (21490) 277
                       B3=19.40
Method: paper electrophoresis.
Al+++ gl KNO3 35°C 0.10M U M
                                  1980KHb (21491) 278
                        B(AlL(tripolyphosphate))=7.97
***********************
             HL Acetohydroxamic CAS 546-88-3 (2766)
Acetohydroxamic acid, N-Hydroxyacetamide; CH3.CO.NHOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl
            25°C 0.20M C
                        K1=8.15 B2=15.77 1995FKa (21804) 279
                        B3=21.5
                        B(AlH-1L2)=10.40
                        B(A1H-2L2)=1.04
*********************************
C2H7NS
                           CAS 60-23-1 (588)
2-Aminoethanethiol; H2N.CH2.CH2.SH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
                        K1=11.74 B2=22.94 1995LMa (22487) 280
     gl KCl 25°C 0.10M C
                        B(ALHL)=17.60
********************************
C2H7O3P
             H2L
                          CAS 71778-99-9 (1978)
Ethylphosphonic acid; CH3.CH2.PO3H2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ gl KCl
            25°C 0.20M C
                         K1=6.63 B2=12.1 1996AKa (22566) 281
                        B(AlH-1L)=2.34
                        B(A1H-2L)=-3.93
                        B(A1H-1L2)=6.1
*******************************
                 Ethylenediamine CAS 107-15-7 (23)
              L
1,2-Diaminoethane; H2N.CH2.CH2.NH2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     oth non-aq ? 100% U M
                                  1962BBa (23128) 282
```

K(4A12A6L=A18A24L4)=6.2

```
Method: freezing point. Medium: benzene. HA=isopropylalcohol
*****************************
                  HEDPA
                            CAS 2809-21-4 (436)
1-Hydroxyethane-1,1-diphosphonic acid; CH3.C(OH)(PO3H2)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.10M C
                          K1=19.12
                                   2002GKc (23356) 283
                         B(AlH-1L)=13.93
                         B(AlH-2L)=2.71
                         B(A1H3L2)=44.28
Al+++ gl KNO3 25°C 0.10M C
                         K1=22.7 B2=31.10 1998LDa (23357) 284
                         B(A1HL)=27.2
                         B(A1H2L)=29.1
                         B(A12HL)=38.7
                         B(A1H2L2)=43.1
B(AlH-1L)=17.6
              -----
Al+++ gl KCl 25°C 0.10M U
                          K1=15.29 B2=22.26 1967KLa (23358) 285
                         K(Al+H-1L)=21.37
                         K(A1+2H-1L))=25.87
                         K(2A1+H-1L))=27.25
                         K(2A1+L)=19.33
**********************************
                  IDPA
C2H9N06P2
             H4L
                            CAS 32545-63-4 (1335)
Imino-N,N-bis(methylenephosphonic acid); HN(CH2PO3H2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl KNO3 25°C 0.1M C
                         B2=20.60
                                   1985MMa (23451) 286
                         K(All(OH)+H)=6.92
*********************************
C2H16N5O4Co
                              (231)
Pentaammineoxalatocobalt(III); Co(NH3)5(HC2O4)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     sp NaClO4 28°C 0.30M U K1=1.74
                                   1974NDa (23472) 287
**********************************
                 Malonic acid CAS 141-82-2 (79)
Propanedioic acid; CH2(COOH)2
_____
      Mtd Medium Temp Conc Cal Flags Lg K values
                                    Reference ExptNo
______
    EMF NaCl 25°C 0.0 C TIH K1=7.49 B2=12.62 1998RPb (24383) 288
Method: Pt/H2 electrode. Calculated from data for 0.10-1.0 m NaCl, 5-75 C.
DH(K1)=19 \text{ kJ mol-1}, DS(K1)=208 \text{ J K-1 mol-1}; DH(B2)=29, DS(B2)=340.
______
```

Al+++ Method: a At 80 C,	atomic	absorp	35°C 0.00 C T tion and ion chr	B2=11.3 1995FYa (24384) 289 romatography.
Al+++	gl	KCl	25°C 0.10M C	K1=6.711 B2=11.53 1993PTa (24385) 290 K3=2.58 *K(AlL2)=-6.68
Al+++	gl	NaC1		K1=6.264 B2=11.111 1982JCa (24386) 291 B3=13.3
	J		25°C 1.00M U T	1974TGa (24387) 292 K(Al+HL)=3.58 K(Al+2HL)=5.99 K(Al+3HL)=8.58 82, K(Al+3HL)=8.98
Al+++	gl	NaC104	30°C 0.20M U	K1=5.24 B2=9.40 1967AMa (24388) 293
Al+++	gl	oth/un	35°C ? U	1958DBb (24389) 294 K3=4.06 B3=15.84
**************************************	L		**************************************	(8169)
	Mtd	Medium	•	lags Lg K values Reference ExptNo
Al+++ Also data ******** C3H6N2O3	gl a for «****	KNO3 the sch	25°C 0.10M C iff based formed ******** H2L	K1=4.05 B2= 6.05 1981TMe (24759) 295 d with pyridoxal. ex************ (7445) dd; CH3C(:NOH)CONHOH
Metal	Mtd	Medium	Temp Conc Cal F	lags Lg K values Reference ExptNo
Al+++	gl	NaC1	37°C 0.15M C TI	2000GKb (24824) 296 B(AlHL)=17.370 B(AlH2L2)=34.461 B(AlH3L3)=50.130 B(AlH2L3)=43.968
B(AlHL)=1	7.885	, B(AlH	2L2)=34.996, B3=	2)=27.322. At 25C, 0.1 M KNO3: =26.471, B(AlH3L3)=51.036.
C3H6O2 Propanoic	acid	; CH3.C	•	c acid CAS 79-09-4 (35)
Metal	Mtd	Medium	•	Flags Lg K values Reference ExptNo
Al+++	gl	NaCl	25°C 0.60M C	1989MOa (24980) 297

Κí	(2A1+HL=A12)	(OH))2L+3H)=-8.	038
\sim	ZAITHLEAIZ	UII	<i> </i> 	, 0 .	טכט

						K(2Al+HL=Al2(OH)2L+3H)=-8.038	
Al+++	gl	NaNO3	25°C	1.00M	U	K1=1.7 1975KIb (24981) 298	
Values als	o at	35 C,	45 C			K1=1.78 B2=3.4 1975TRa (24982) 29	19
C3H6O3 3-Hydroxyp			HL			CAS 81598-26-7 (2521)	
Metal	Mtd	Medium	Temp	Conc	Cal Flag	s Lg K values Reference ExptNo	
Al+++ *******	•					1972PKa (25259) 300 K(A1+3HL)=7.38 ************************************	
C3H6O3 L-2-Hydrox			HL	L-L	actic ac	id CAS 79-33-4 (82)	
Metal	Mtd	Medium	Temp	Conc	 Cal Flag 	s Lg K values Reference ExptNo	
Al+++	gl	NaCl	25°C	0.60M		1990MOa (25390) 301 K(2Al+HL=Al2L(OH)+3H)=-6.86 K(2Al+2HL=Al2H-4L2+6H)=-16.79	
K(13Al+4HL	=Al1	3L4(OH)	32+36l	H)=-10	6.9 		
Al+++	gl	NaCl	25°C	0.60M	C	K1=2.36 B2=4.42 1986MSb (25391) 30 K3=1.37 K(AlL2=AlH-1L2+H)=-3.45	2
Al+++ Values als				1.00M	UT	K1=4.26 B2=4.8 1975TRa (25392) 30	13
Al+++	sp	oth/un	?	?	U	1972PKa (25393) 304 K(Al+HL)=4.46 K(Al+2HL)=6.51	
Al+++	EMF	NaNO3	20°C	0.20M	U	K1=2.38 B2=4.56 1971HUb (25394) 30 B3=6.66	15
Al+++	sp	oth/un	?	?	U	1970PKd (25395) 306 B3=3.79 K(Al+L+20H=AlOH(H-1L))=8.49 K(Al+L+30H=Al(OH)2(H-1L))=24.6	
Al+++	•					K1=0.85 B2=2.92 1969PKc (25396) 30 K(Al+2H-1L)=23.05	17
********* C3H7NO2 2-Aminopro			HL	Ala	nine	**************************************	

Metal	Mtd	Medium	Temp	Conc Ca	l Flag	s Lg K value	Reference ExptNo
Al+++	gl	NaNO3	25°C	0.50M C		B(Al2H-2L2): B(AlH-3L)=-	
Al+++	J					B(AlL(tripo	1980KHb (26138) 309 lyphosphate))=7.51
C3H7NO2 N-Methylac			HL			(692	***************************************
Metal	Mtd	Medium	Temp	Conc Ca	l Flag	s Lg K value	s Reference ExptNo
Al+++	gl	KC1	25°C	0.20M C		K1=8.69 B3=22.41	32=16.21 2000FEc (26620) 310
**************************************			HL			************** (750)	*****************************
Metal	Mtd	Medium	Temp	Conc Ca	l Flag	s Lg K value	s Reference ExptNo
Al+++						B(AlH-1L2)=: B(AlH-2L2)=:	
******	*****	*****	*****	******	*****	*****	K*******
C3H7NO2S 2-Amino-3-	-merca		H2L	Cyste	ine	CAS 52 (CH2.SH)COOH	
2-Amino-3-		ptoprop	H2L panoio	Cyste acid;	ine H2N.CH 	CAS 52 (CH2.SH)COOH	
2-Amino-3- Metal Al+++	Mtd Mtd gl	ptoprop Medium NaClO4	H2L canoic Temp 25°C	Cyste acid; Conc Ca	ine H2N.CH 1 Flag M	CAS 52 (CH2.SH)COOH s Lg K value: 	-90-4 (96) s Reference ExptNo 1975RMa (26751) 312 te))=14.90 =18.89)=15.66
2-Amino-3 Metal Al+++ ********************************	Mtd gl *****	ptoprop Medium NaClO4	H2L panoic Temp 25°C	Cyste acid; Conc Ca 0.10M U	ine H2N.CH 1 Flag M *****	CAS 52 (CH2.SH)COOHs Lg K value: K1=6.43 B(AlL(citra: B(AlL(NTA)): K(Al+L+HPO4 ************************************	-90-4 (96) s Reference ExptNo 1975RMa (26751) 312 te))=14.90 =18.89
2-Amino-3 Metal Al+++ ************ C3H7NO3 2-Amino-3-	Mtd gl *****	ptoprop Medium NaClO4 *****	H2L Danoic Temp 25°C ***** HL anoic	Cyste acid; Conc Ca 0.10M U ******* Serin acid; H	ine H2N.CH 1 Flag M ***** e 2N.CH(CAS 52 (CH2.SH)COOH s Lg K value: K1=6.43 B(AlL(citra: B(AlL(NTA)): K(Al+L+HPO4 ************************************	-90-4 (96) s Reference ExptNo 1975RMa (26751) 312 te))=14.90 =18.89)=15.66 ***********************************
2-Amino-3 Metal Al+++ ********* C3H7NO3 2-Amino-3 Metal Al+++	Mtd gl ***** -hydro Mtd gl	ptoproperson Medium NaClO4 ***** xypropa Medium NaCl	H2L panoic Temp 25°C ***** HL anoic Temp 37°C	Cyste acid; Conc Ca 0.10M U ****** Serin acid; H Conc Ca	ine H2N.CH l Flag M ****** e 2N.CH(l Flag	CAS 52 (CH2.SH)COOH S Lg K value: K1=6.43 B(A1L(citra: B(A1L(NTA)): K(A1+L+HPO4 *********** CAS 56 CH2.OH)COOH S Lg K value: B(A12H-2L)=	-90-4 (96) Reference ExptNo 1975RMa (26751) 312 te))=14.90 =18.89)=15.66 **********************************
2-Amino-3	Mtd gl ***** -hydro Mtd gl	ptoproperson MediumNaClO4 ***** xypropa Medium NaCl	H2L panoic Temp ***** HL anoic Temp Temp 37°C	Cyste acid; Conc Ca 0.10M U ****** Serin acid; H Conc Ca 0.15M C	ine H2N.CH 1 Flag M ****** e 2N.CH(1 Flag 	CAS 52 (CH2.SH)COOH s Lg K value: K1=6.43 B(AlL(citra: B(AlL(NTA)): K(Al+L+HPO4 *********** CAS 56 CH2.OH)COOH s Lg K value: B(Al2H-2L)=6	-90-4 (96) S Reference ExptNo 1975RMa (26751) 312 te))=14.90 =18.89)=15.66 ***********************************

```
Method: electrophoresis
*******************************
                  3-Phosphono-Ala CAS 20263-06-3 (1509)
2-Amino-3-phosphonatopropanoic acid; (H2O3P)CH2.CH(NH2).COOH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KCl 25°C 1.00M C M K1=13.12
                                  1985SMd (27350) 316
Al+++
                          K(All+H)=3.22
ternary complexes with pyridoxal-5'-phosporic acid
**********************************
C3H8NO5P
                            CAS 23052-80-4 (1508)
3-Amino-3-phosphonatopropanoic acid; (H2O3P)(NH2)CH.CH2.COOH
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
             25°C 1.00M C
                         K1=10.93
      gl KCl
                                   1989MSb (27361) 317
                         K(All+H)=4.19
*****************************
                  Glyphosate CAS 1071-83-6 (1617)
C3H8N05P
             H3L
N-(Phosphonomethyl)glycine; H2O3P.CH2.NH.CH2.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KNO3 25°C 0.1M C K1=13.70 B2=22.05 1985MMa (27401) 318
                         B(A1HL)=16.18
                          B(A1HL2)=27.76
********************************
                  Phosphoserine CAS 17885-08-4 (1865)
             H3L
Serine dihydrogenphosphate, O-Phosphoserine; NH2.CH(CH2.OPO3H2).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KCl 25°C 0.20M C
                       K1=11.48 1998KLb (27463) 319
                          B(A1HL)=14.88
                          B(AlH-1L)=5.44
                         B(A1H-2L)=-2.63
Al+++ gl KCl 25°C 0.10M U
                          K1=8.50 B2=12.76 1997ZTa (27464) 320
______
Al+++ gl KNO3 25°C 0.10M C
                                B2= 8.81 1981TMe (27465) 321
                          K1=4.79
                          B3=10.68
Also data for the schiff based formed with pyridoxal.
**************************
                 Ala-hydroxamic CAS 16707-85-0 (1582)
              HL
2-Amino-N-hydroxypropanamide, Alanine hydroxamic acid; CH3.CH(NH2).CO.NH.OH
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl
             25°C 0.20M C B2=16.7
                                   1989FSa (27576) 322
```

```
B(A1HL)=14.35
                         B(A12L2)=22.21
                         B(A12H-1L2)=17.59
                         B(A12H-2L2)=12.63
Also B(A12H-3L2)=5.85; B(A12H-4L2)=-2.44; B(A1H-1L2)=9.62; B(A1H-2L2)=-0.16.
******************
C3H8N2O2
              HL
                              (6666)
beta-Alaninehydroxamic acid; NH2.CH2.CH2.CO.NHOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KCl
            25°C 0.20M C
                         B2=19.87 1995FKa (27606) 323
                         B(A1HL)=16.72
                         B(A1H2L2)=32.07
                         B(A1HL2)=27.04
                         B(AlH-1L2)=10.74
B(A1H-2L2)=0.04.
**********************************
2,3-Diphospho-D-glyceric acid; H2O3PO.CH2.CH(COOH)OPO3H2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                         B2=12.46 1990SKc (27802) 324
Al+++ gl KCl 25°C 0.20M C
                         B(AlHL)=13.12
                         B(A1H2L2)=24.42
                         B(A1HL2)=18.78
*********************************
                          CAS 471915-95-4 (8549)
2,3-Diamino-N-hydroxypropanamide;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.20M C K1=11.7 B2=17.60 2002ECa (27982) 325
                        B(A1HL2)=24.2
**********************************
                            CAS 17181-54-3 (7537)
1,3-Dihydroxypropyl-2-phosphoric acid; HOCH2CH(OPO3H2)CH2OH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaClO4 25°C 0.10M C
                         K1=5.30 B2=10.38 2003CCa (28029) 326
                         B(A12H-1L2)=11.72
                         B(A13H-5L2)=-3.42
                         B(A13H-6L2)=-10.03
                         B(A13H-7L2)=-18.05
Additional method: 31P nmr. B(Al3H-1L3)=20.53. Fixed values:
B(A1HL)=8.5, B(A1H2L)=9.4, B(A1HL2)=14.0.
*******************************
                            CAS 57-03-4 (2984)
C3H906P
             H2L
```

```
2,3-Dihydroxypropylphosphoric acid, Glycerol 1-phosphate; HO.CH2.CH(OH).CH2.OPO3H2
  -----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                  Reference ExptNo
_____
Al+++ gl NaClO4 25°C 0.10M C
                       K1=5.87
                              B2=10.31 2003CCa (28046) 327
                        B(A12H-1L2)=11.80
                        B(A13H-5L2)=-2.98
                        B(A13H-6L2)=-9.51
                        B(A13H-7L2)=-17.30
Additional method: 31P nmr. B(Al3H-1L3)=20.63. Fixed values:
B(A1HL)=8.5, B(A1H2L)=9.4, B(A1HL2)=14.0.
**********************************
C3H10N03P
            H2L
                            (1986)
1,1-Dimethyl-1-aminomethylphosphonic acid; H2N.C(CH3)2.PO3H2
------
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl KCl
            25°C 0.10M U
                        K1=11.42 B2=19.53 1969DMd (28073) 328
                        K(A1+HL)=5.59
                        K(A1+2HL)=9.61
                        K(A1+3HL)=13.07
*******************************
C4H2O4
            H2L
                Squaric acid
                         CAS 2892-51-5 (439)
3,4-Dihydroxy-3-cyclobutene-1,2-dione;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++
      ix R4N.X 25°C 1.00M U K1=1.85 B2=2.74 1972CSb (28635) 329
Medium: NH4ClO4
______
     ix NaClO4 25°C 0.30M U K1=2.83
                                1969TWa (28636) 330
*********************************
            H2L
                Maleic acid CAS 110-16-7 (111)
cis-Butenedioic acid; HOOC.CH:CH.COOH
_____
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl NaCl04 25°C 0.04M M K1=5.48 B2= 8.78 1993MYa (29046) 331
Δ1+++
Medium: 0.01 M HClO4, 0.005 M Al(ClO4)3.
***********************************
                6-Aminouricil CAS 873-83-6 (6213)
C4H5N302
4-Amino-2,6-dihydroxypyrimidine;
   -----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaClO4 30°C 0.10M U K1=13.32 B2=25.27 1986JDa (29421) 332
**********************
C4H603
                          CAS 600-18-0 (5474)
2-Ketobutanoic acid; CH3.CH2.CO.COOH
______
```

Metal	Mtd Medium Temp Conc Cal Flag	gs Lg K values Reference ExptNo	
		K1=2.09 B2=3.65 1982KMc (29746)	333
C4H6O3 3-Ketobut	HL Acetoacetion	c aci CAS 541-50-4 (5475)	
Metal	Mtd Medium Temp Conc Cal Fla	gs Lg K values Reference ExptNo	
Al+++	gl KCl 25°C 0.10M C	K1=1.21 B2=1.84 1982KMc (29750) ************************************	334
C4H6O4 1,4-Butan	H2L Succinic acedioic acid; H00C.CH2.CH2.COOH	cid CAS 110-15-6 (112)	
Metal 		gs Lg K values Reference ExptNo	
Al+++		K1=3.753 1998VBa (29928) 335 B(AlHL)=7.154 B(Al2H-2L)=-1.103 B(Al3H-1L2=9.114 B(Al3H-3L2)=1.385	
		K1=3.63 1997KSa (29929) 336 B(AlHL)=7.03 B(AlH-1L)=-0.53 B(AlH-2L)=-5.55	
Al+++		K1=4.17 1990FDa (29930) 337 B(AlHL)=7.18 B(AlH-2L)=-4.65 B(Al2H-3L)=-4.94	
 Al+++		K1=3.91 1987VBe (29931) 338 B(Al2H-3L)=-5.23 B(Al3H-2L2)=5.34	
 Al+++	gl NaClO4 25°C 0.50M U	K1=3.2 1984CDa (29932) 339 B(AlHL)=6.60 B(AlH-1L)=4.2	
Al+++	gl NaCl 37°C 0.15M C	1982JCa (29933) 340 B(AlH-1L)=-0.25 B(Al2H-1L)=2.88 B(AlH-2L)=-5.19	
Al+++	gl NaClO4 25°C 1.00M U	1974TGa (29934) 341 K(Al+HL)=3.84 K(Al+2HL)=5.93 K(Al+3HL)=8.98	

```
At 35 C: K(Al+HL)=3.78, K(Al+2HL)=6.40, K(Al+3HL)=9.52
**************************
                   Me-Malonic Acid CAS 516-15-2 (816)
              H2L
Methylpropanedioic acid; HOOC.CH(CH3).COOH
     Mtd Medium Temp Conc Cal Flags Lg K values
______
      gl NaCl 25°C 0.60M C
                                      1990MOb (30114) 342
Al+++
                           K(A1+H2L=A1L+2H)=-2.213
                           K(A1+2H2L=A1L2+4H)=-5.73
                           K(Al+3H2L=AlL3+6H)=-11.19
**********************************
              H2L
                   Thiodiacetic
                              CAS 123-93-3 (140)
2,2'-Thiodiglycolic acid, Thiodiethanoic acid; HOOC.CH2.S.CH2.COOH
-----
       Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
gl NaClO4 25°C 0.50M U
                           K1=1.93
                                      1972NAb (30210) 343
                          B(All(OH))=12.34
**********************************
                   Thiomalic acid CAS 70-49-5 (109)
              H3L
2-Mercaptosuccinic acid, 2-Sulfanyl-1,4-butanedioic acid; HOOC.CH(SH).CH2.COOH
______
                                     Reference ExptNo
Metal Mtd Medium Temp Conc Cal Flags Lg K values
-----
                            K1=8.63
Al+++ gl KCl 25°C 0.20M C
                                      1997KSa (30318) 344
                           B(A1HL)=12.99
                           B(AlH-1L)=4.05
***********************************
                   Malic acid CAS 617-48-1 (393)
C4H605
              H2L
2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                           K1=4.519
Al+++ gl NaCl 37°C 0.15M C
                                      2001VBa (30578) 345
                           B(A1HL)=7.032
                           B(A1HL2)=10.981
                           B(AlH-1L)=1.268
                           B(A12H-2L)=0.564
B(A12H-3L)=-3.054, B(A12H-3L2)=1.779, B(A12H-4L2)=-4.462,
B(A12H-1L3)=12.789, B(A13H-4L4)=10.132, B(A14H-5L4)=10.537.
_____
Al+++
      sp oth/un 23°C 0.10M U
                                      1994KGa (30579) 346
                           Keff=11.6
Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=5.34 with HCl
             37°C 0.15M C
                           K1=4.37 B2=8.17 1990FDa (30580) 347
Al+++
      gl NaCl
                           B(AlH-1L2)=4.11
```

```
gl NaNO3 25°C 0.50M M
Al+++
                       Μ
                                   1989MAa (30581) 348
                         B(-3,1,1)=-10.5
                         K(2A1H-2L=A12H-4L2)=-22.2
B(p,q,r): pH+qM+rH2L. K(UO2+A1+2H2L=UO2A1H-4L2+8H)=-8.06
                          K1=4.60 B2= 7.62 1987VBe (30582) 349
Al+++
      gl NaCl 37°C 0.15M U
                         B(A1HL)=6.87
                         B(AlHL2)=11.31
                         B(AlH-1L2)=4.31
                         B(A12H-2L2)=5.59
B(A12H-3L2)=2.50
-----
Al+++ gl NaClO4 25°C 0.01M U
                                   1976MPb (30583) 350
                         K(A1+H2L=A1H-1L+3H)=-5.39
                         K(A1H-2L+H)=4.72
                         K(A1H-3L+H)=7.60
 -----
     EMF KNO3 20°C 0.20M U
                         K1=5.34 B2=9.32 1969PVc (30584) 351
Al+++ sp NaClO4 29°C 1.0M U K1=3.32 1965MNa (30585) 352
*******************************
C4H605
                 Diglycolic acid CAS 110-99-6 (243)
             H2L
Di(carboxy)methyl ether, 2,2'-Oxydiethanoic acid; HOOC.CH2.O.CH2.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl NaCl04 25°C 0.10M U TIH K1=3.43 B2=6.42 1979SDc (30851) 353
-----
Al+++ gl NaClO4 25°C 0.50M U K1=3.16 B2=5.25 1972NAd (30852) 354
***********************************
                  DL-Tartaric acd CAS 133-37-9 (94)
             H2L
DL-Tartaric acid, DL-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaNO3 25°C 0.50M M
                       Μ
                                   1989MAa (31009) 355
                         B(-4,1,1)=-7.8
                         K(2A1H-2L=A12H-4L2)=-18.5
B(p,q,r): pH+qM+rH2L. K(UO2+Al+2H2L=UO2AlH-4L2+8H)=>-12
********************************
                  L-Tartaric acid CAS 87-69-4 (92)
C4H606
             H2L
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
  .-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaCl 37°C 0.15M C
                          K1=3.788
                                   2000DDb (31185) 356
                         B(AlH-1L)=1.165
                         B(A12H-1L2)=7.976
                         B(A12H-2L2)=5.347
                         B(A12H-3L2)=1.009
```

```
B(A12H-4L2)=-5.101
-----
Al+++ sp oth/un 23°C 0.10M U
                                     1994KGa (31186) 357
                          Keff=11.7
Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=6.00 with HCl
______
Al+++ gl NaCl 25°C 0.60M C
                                     1990MOd (31187) 358
                          B(1,-2,1)=-3.44
                          B(2,-5,2)=-6.30
                          B(2,-6,2)=-8.91
                          B(2,-7,2)=-13.12
B(2,-8,2)=-18.95. B(p,q,r): pAl+qH+rH2L=AlpHq(H2L)r
                          B2=7.65 1984MMb (31188) 359
Al+++ gl KNO3 25°C 0.10M C
                          B(AlH-1L)=1.18
                          *K(AlH-1L)=-5.15
                          *K(A1L2)=-3.72
                          K(A1H-3L2+2H)=12.67
______
Al+++ gl NaClO4 25°C 0.10M U K1=5.62 B2=9.95 1972MRc (31189) 360
Meso Tartaric acid: K1=5.32, K2=4.45.
Al+++ oth oth/un 20°C 0.0 U
                           K1=6.35
                                    1967FRa (31190) 361
                          K(A1+HL)=3.43
                          K(AlOH+L=AlH-1L)=9.05
                          K(Al(OH)2+L=AlH-2L)=10.92
                          K(Al(OH)3+L=AlH-2LOH)=8.37
Method: optical rotation. K(Al(OH)4+L=Al(H-2L)(OH)2)=8.89, B(Al2L)=2,
K(Al+AlH-1L)=3, K(Al+H-2L)=4, K(AlL+H2L)=0.66, K(AlH-1L+H2L)=1.12 plus others
_____
Al+++ oth oth/un 20°C ? U
                                     1967PTa (31191) 362
                          K(AlOH+L)=1.8
Method: refraction
-----
Al+++ dis NaCl04 20°C 0.10M U B2=9.56 1963STc (31192) 363
*******************************
              H2L Aspartic acid CAS 56-84-8 (21)
C4H7N04
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.20M C K1=7.87 1997KSa (31811) 364
                          B(A1HL)=11.76
                          B(AlH-1L)=3.30
                          B(A1H-2L)=-2.32
          Al+++ gl NaClO4 25°C 0.50M U
                                     1984CDa (31812) 365
                          K(Al+HL)=2.16
```

K(A1(OH)+HL)=3.03

```
Al+++ gl KNO3 35°C 0.10M U
                                1980KHb (31813) 366
                     B(AlHL(tripolyphosphate))=8.29
    gl NaClO4 25°C 0.10M U K1=16.25
K3=11.50
                       K1=16.29 B2=30.69 1972SSe (31814) 367
*******************************
            H2L
                TDA
                         CAS 142-73-4 (118)
Iminodiethanoic acid; HN(CH2.COOH)2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 25°C 0.10M U K1=8.10 1997YSa (32195) 368
______
Al+++ gl NaClO4 25°C 0.10M U T K1=8.62 B2=16.12 1981DSa (32196) 369
At 35 C: K1=8.42, B2=16.62; 45 C: 8.28, 15.49
-----
Al+++ gl NaClO4 25°C 0.50M U K1=8.10 B2=15.07 1971LNb (32197) 370
C4H7N05
            H2L
                           (1234)
N-Hydroxyiminodiethanoic acid; HO.N(CH2.COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KNO3 25°C 0.10M C K1=5.1
                                1987AKa (32426) 371
*************************
C4H8N02C1
                           (8170)
3-Chloro-2-aminobutanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KNO3
            25°C 0.10M C K1=4.13 B2= 6.33 1981TMe (32468) 372
Also data for the schiff based formed with pyridoxal.
******************************
               Asparagine CAS 70-47-3 (17)
C4H8N2O3
            HL
2-Aminobutanedioic acid 4-amide; H2N.CH(CH2.CO.NH2).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
           25°C 0.20M C K1=5.50 1997KSa (32681) 373
     gl KCl
                      B(AlH-1L)=1.31
******************************
                Gly-Gly CAS 556-50-3 (54)
C4H8N2O3
Glycyl-glycine; H2N.CH2.CO.NH.CH2.COOH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      gl KNO3 35°C 0.10M U
                                1980KHb (33017) 374
                     B(AlHL(tripolyphosphate))=4.18
```

C4H8N2O4 N(1)-Hydro	xyası					(6369) droxamic acid;				DΗ
Metal	Mtd					s Lg K values				
						B2=19.26 B(AlHL)=16.27 B(AlH2L2)=31.7 B(AlHL2)=26.80 B(AlH-1L2)=10.	6 36	·	·	
C4H8O2 n-Butanoic			HL	2.C00H		CAS 107-9	2-6	(1118)		
Metal		Medium		Conc Cal	Flag	Lg K values				
	****	******	****** HL	1.00M U ******	****	K1=1.6 ************************************	****	*******		
Metal	Mtd	Medium	Temp (Conc Cal	Flag	s Lg K values		Reference	ExptNo	
Al+++ gl diox/w 30°C 50% U K1=3.45 1956IFa (33403) 377 ***********************************										
Metal	Mtd	Medium	Temp (Conc Cal	Flag	s Lg K values		Reference	ExptNo	
Al+++ dis NaClO4 35°C 0.10M U M K1=7.68 1995TKa (34094) 378 Method: Paper electrophoresis; Ternary complexes with NTA. ***********************************										
Method: Pa ******** C4H9NO3	per (****	electrop	ohores: ****** HL	is; Terna ****** Threon:	ary co **** ine	omplexes with N ************************************	TA. ****	· *******	·	
Method: Pa ********* C4H9NO3 2-Amino-3- Metal	per (**** hydro Mtd	electrop ******* oxybutan Medium	ohores: ****** HL noic ad Temp (is; Tern ******* Threon: cid; H2N Conc Cal	ary co ***** ine .CH(CH Flags	omplexes with N ************************************	TA. ***** -5 (**************************************	****** ExptNo	
Method: Pa ************************************	per (**** hydro Mtd gl	electrop ******* oxybutan Medium NaCl	ohores: ****** HL noic ad Temp (is; Tern ******* Threon: cid; H2N Conc Cal 0.15M C	ary co ***** ine .CH(CH Flags	omplexes with None with No	TA. **** -5 (200 177	**************************************	****** ExptNo	
Method: Pa ******** C4H9NO3 2-Amino-3 Metal Al+++	hydro Mtd gl	electroper ******* oxybutaner Mediumer NaCl KCl	ohores: ***** HL noic ac Temp (37°C (is; Tern: ******* Threon: cid; H2N Conc Cal 0.15M C	ary co ***** ine .CH(CI Flags	omplexes with N ********* CAS 72-19 H(OH).CH3)COOH S Lg K values B(Al2H-2L)=-0. K1=5.51 B(AlH-1L)=0.94	TA. **** -5 (200 177 199	**************************************	******* ExptNo (7) 379	
Method: Pa ******** C4H9NO3 2-Amino-3 Metal Al+++	hydro Mtd gl gl	electroper ******* oxybutaner ** Mediumer ** NaCl** KCl** NaCl** NaCl04	ohores: ***** HL noic ac Temp (37°C (25°C (35°C (is; Tern: ****** Threon: cid; H2N Conc Cal 0.15M C 0.20M C	ary co	omplexes with N ********** CAS 72-19 H(OH).CH3)COOH COURT CH3 COURT CH3 COURT CH3 COURT CH3 COURT CH3 COURT CH3 CAS 72-19 H(OH).CH3 CH3 CH3 CH3 CH3 CH3 CH3 CH3	TA. **** -5 (200 177 199 13.04	**************************************	ExptNo (7) 379 (8) 380 (34289)	381

```
***********************************
C4H100
               L
                  Ether
                            CAS 60-29-7 (3573)
Diethyl ether (ethyl ether, ethoxyethane); C2H5.O.C2H5
        Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      oth oth/un 20°C 0.07M U
                          K1=1.34
                                   1965PBc (34651) 383
Method: mass spectrograph. Medium: AlI3
*********************
                            CAS 471915-94-3 (8550)
C4H11N302
2,4-Diamino-N-hydroxybutanamide;
-----
      Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
______
Al+++
      gl KCl 25°C 0.20M C
                          B2=19.81
                                    2002ECa (35175) 384
                         B(AlHL)=17.3
                         B(A1HL2)=27.2
                         B(A1H2L2)=34.0
*******************************
C5H2O2F6
                  HFA
                            CAS 1522-22-1 (195)
1,1,1,5,5,5-Hexafluoropentane-2,4-dione; F3C.CO.CH2.CO.CF3
   ______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      nmr non-aq 25°C 100% U H
                                    1964PCa (35920) 385
Method:NMR, medium:CHCl3. DG(trans-AlL3=cis-AlL3)=3.8 kJ mol-1,DH=1.0,DS=-8
*******************************
                            CAS 13161-30-3 (5582)
C5H5N02
              HL
1-Hydroxypyridin-2(1H)-one, 2-Hydroxypyridine 1-oxide;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.20M C B2=12.5
                                  2000FEc (36751) 386
-----
      gl KCl
             25°C 0.10M U
                                B2=15.54 1993LMc (36752) 387
Al+++
                         K1=8.16
                         K3=6.05
**********************************
C5H5N02
                            CAS 16867-04-2 (2316)
2,3-Dihydroxypyridine, 3-Hydroxypyridin-2(1H)-one; C5H3N(OH)2
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++
      gl NaCl 25°C 0.60M U T
                                B2=16.34 1999DBa (36779) 388
                          K1=8.59
                         B3=23.11
                         B(AlH-1L3)=13.85
At 37 C, K1=8.19, B2=16.03, B3=21.77, B(AlH-1L3)=13.0
Al+++
      kin NaClO4 34°C 0.10M C
                                    1979BMb (36780) 389
                         K(Al+H2L=AlL+2H)=-1.85
Method: stopped-flow.
```

```
***********************************
C5H5N03
             H2L
                           CAS 99110-85-7 (2195)
1,4-Dihydroxy-2-pyridinone;
  -----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++
     gl KCl
            25°C 0.10M C
                         B2=21.20
                                  1992CMc (36841) 390
                         B3=25.16
                         B(AlHL)=16.5
                         B(A1HL2)=27.22
                         B(A1H2L2)=32.64
**********************************
                            CAS 642-93-3 (5476)
C5H605
             H2L
3-Methyl-2-oxobutanedioic acid HOOC.CO.CH(CH3).COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values
                                    Reference ExptNo
-----
     gl KCl 25°C 0.10M C K1=6.16 B2=10.80 1982KMc (37479) 391
******************************
                 Acetylacetone CAS 123-54-6 (164)
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      dis NaClO4 25°C 0.10M C K1=8.7
                                   1986SNa (37910) 392
Method: rate of distribution of volatile ligand between aqueous phase and
inert gas phase. K(H+L)=9.17 assumed.
     oth NaCl04 25°C 0.10M C I T K1=8.2 B2=15.7 1982SLc (37911) 393
Al+++
                         B3=21.4
IUPAC evaluation. I=0 corr.: K1=8.6, B2=16.5, B3=22.3
      gl oth/un 30°C 0.0 U
                         K1=8.6
                                B2=16.5 1955IFa (37912) 394
                         K3=5.8
*********************************
C5H9N02
              HL
                 Proline
                           CAS 147-85-3 (44)
Pyrrolidine-2-carboxylic acid; C4H8N.COOH
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KNO3 35°C 0.10M U
                       М
                                   1980KHb (38601) 395
                      B(AlL(tripolyphosphate))=8.63
***************
              HL
                 Hydroxyproline CAS 51-35-4 (416)
4-Hydroxy-2-pyrrolidinecarboxylic acid; C4H7N(OH)(COOH)
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KNO3 35°C 0.10M U
                                   1980KHb (38719) 396
                         B(AlL(tripolyphosphate))=7.83
```

C5H9N04		H2L Glutamic acacid; H2N.CH(CH2.CH	•
Metal	Mtd Medi		gs Lg K values Reference ExptNo
		37°C 0.15M C	2003DBa (39059) 397 B(AlH2L)=15.015 B(Al2H-1L)=6.680 B(AlHL2)=18.850 B(AlH2L2)=23.425
B(A13H-4L2	2)=3.810,	B(Al3H-4L3)=11.220.	
Al+++	gl KCl		K1=7.29 1997KSa (39060) 398 B(AlHL)=10.88 B(AlH-1L)=2.55 B(Al2L)=9.46
		04 25°C 0.50M U	1984CDa (39061) 399 K(Al+HL)=2.30 K(Al(OH)+HL)=3.50
		35°C 0.10M U M	
			K1=15.12 B2=29.40 1972SSe (39063) 401 K3=9.20 ************
C5H9N04			CAS 4408-64-4 (190)
		um Temp Conc Cal Fla	gs Lg K values Reference ExptNo
Al+++ **********************************	gl NaCl ******	04 25°C 0.50M C ******** H4L PMIDA	K1=7.55 1984NAa (39238) 402 **************** CAS 5994-61-6 (2433) ; H203P.CH2.N(CH2.COOH)2
Metal	Mtd Medi	um Temp Conc Cal Fla	gs Lg K values Reference ExptNo
			K1=14.7 1980VRa (39666) 403 K(Al+HL)=7.3 ************************************
	ntanedioic		H(CH2.CH2.CO.NH2)COOH
Metal	Mtd Medi	um Temp Conc Cal Fla	gs Lg K values Reference ExptNo
Al+++	gl KCl	25°C 0.20M C	K1=5.61 1997KSa (39811) 404 B(AlH-1L)=1.33

```
***********************************
C5H10N2O4
             HL
                          CAS 1955-67-5 (6736)
2-Aminopentanoic-5-hydroxamic acid; HOOC.CH(NH2).CH2.CH2.CO.NOH
 -----
   Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++
     gl KCl
            25°C 0.20M C
                        B2=19.89
                                 1995FKa (40077) 405
                        B(A1HL)=16.65
                        B(A1H2L2)=32.79
                        B(A1HL2)=27.62
                        B(AlH-1L2)=9.9
B(AlH-2L2)=1.10
***********************************
                            (7178)
2,5-Dihydroxypentanoic acid; HOCH2CH2CH2CH0HCOOH
-----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                  Reference ExptNo
______
                        K1=2.04
Al+++ gl NaCl 25°C 0.10M C
                                 1994BHa (40324) 406
                        *K(AlL) = -3.14
                        *K(AlH-1L)=-2.25
C5H11N02
             HL
                Valine
                          CAS 72-18-4 (43)
2-Amino-3-methylbutanoic acid; H2N.CH(CH(CH3)2)COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values
______
     gl KNO3
           35°C 0.10M U
                                 1980KHb (40687) 407
                     B(AlL(tripolyphosphate))=7.97
********************************
             HL
                Methionine
                          CAS 63-68-3 (42)
2-Amino-4-(methylthio)butanoic acid; H2N.CH(CH2.CH2.S.CH3)COOH
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      oth oth/un 25°C 0.10M C
                       K1=7.00 B2=11.50 1998TEb (41077) 408
Method: electrophoresis. Medium: 0.1 M HClO4.
***********************************
            H2L
                Penicillamine CAS 52-66-4 (350)
DL-2-Amino-3-mercapto-3-methylbutanoic acid; (CH3)2C(SH)CH(NH2)COOH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      oth NaCl04 35°C 0.10M C K1=11.50 B2=15.05 1996TKb (41252) 409
Method: paper electrophoresis.
********************************
                Ribose-5-phosph CAS 4300-28-1 (2756)
            H2L
Ribose-5-phosphoric acid, Ribofuranoside 5 Phosphoric acid;
-----
      Mtd Medium Temp Conc Cal Flags Lg K values
                                Reference ExptNo
Metal
```

```
Al+++ gl KCl 25°C 0.20M C
                       K1=5.63
                                1996AKa (41420) 410
                       B(AlH-1L)=1.69
                       B(A1H-2L)=-4.83
*********************************
                Picric acid CAS 88-89-1 (593)
             HL
2,4,6-Trinitrophenol; HO.C6H2(NO2)3
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                     K1=1.05 1955BKa (42091) 411
Al+++ sp oth/un 21°C 0.40M U
                       B3=3.12
**********************************
C6H4N2O5
                          CAS 50-28-5 (505)
2,4-Dinitrophenol; HO.C6H3(NO2)2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp oth/un 21°C 0.40M U K1=0.89 1955BKa (42223) 412
Medium: 0.2-0.7 (some EtOH)
**********************************
C6H4N2O6
                          CAS 7659-29-2 (2694)
            H2L
1,2-Dihydroxy-3,5-dinitrobenzene; (HO)2.C6H2(NO2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl KCl
           25°C 0.10M C B2=21.80
                              2004GAa (42263) 413
                       B3=31.68
**********************************
             L
C6H5NC12
                Dichloroaniline CAS 554-00-7 (761)
2,4-Dichloroaniline; H2N.C6H3(C1)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     sp diox/w 25°C 100% U
                                1976BSa (42347) 414
                      K(AlCl3+L)=2.14
**********************************
C6H5NC12
                Dichloroaniline CAS 95-76-1 (759)
3,4-Dichloroaniline; H2N.C6H3(Cl)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp diox/w 25°C 100% U
                                1976BSa (42352) 415
                      K(AlC13+L)=2.83
**********************************
                Picolinic acid CAS 98-98-6 (391)
C6H5N02
             HL
2-Pyridine-carboxylic acid; C5H4N.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

Al+++ Method: 27	nmr NaCl 7Al nmr.	25°C 1.5M C	K1=4.11 B2= 7.90 2000LKb (424	94) 416
Al+++	gl NaCl	37°C 0.15M C	K1=4.35 B2=8.48 1990FDa (4249 B(AlH-1L2)=4.15 B(AlH-2L)=-3.19	95) 417
Al+++	gl KNO3	25°C 0.15M U	K1=4.487 B2=8.419 1988JJa (4249 B(Al(OH)L2)=17.589 B(Al(OH)2L)=21.73	96) 418
Al+++	gl NaClO	4 25°C 0.50M C	K1=4.62 B2=8.60 1986MNb (4249 B3=12.3	97) 419
			K1=4.497 B2= 8.27 1985JJa (4249 B(AlH-1L2)=17.668 B(Al2H-3L2)=39.27	·
C6H5NO3		H2L carboxylic acid;	CAS 609-71-2 (5910)	* * *
Metal	Mtd Mediu	m Temp Conc Cal Flag	gs Lg K values Reference Expt	No
		25°C 0.60M C By spectrophotometry	K1=12.48 1999MTb (42722) 43 B(AlHL)=18.02 B(AlH2L2)=34.73 B(AlH3L3)=49.92 B(AlH2L3)=43.0	 21
**************************************	********		CAS 874-24-8 (4356)	***
Metal			gs Lg K values Reference Expt	 No
Al+++	gl NaCl		K1=11.89 B2=21.13 1999MTb (4274 B(AlHL)=16.91 B(AlH2L2)=32.62 B(AlHL2)=27.2 B(AlH3L3)=46.91	 47) 422
By spectro	ophotometry	HL3)=34.7, B3=27.04 : B(AlHL)=16.89	· ·	***
C6H5N03		H2L carboxylic acid;	CAS 10128-71-9 (8910)	
Metal	Mtd Mediu	m Temp Conc Cal Flag	gs Lg K values Reference Expt	No
Al+++	gl NaCl	25°C 0.60M C	K1=10.84 B2=19.50 2002DYa (427) B3=26.09	58) 423

B(AlHL)=15.97 B(AlH2L2)=30.78 B(AlHL2)=25.47

```
B(A1H3L3)=44.05, B(A1H2L3)=38.69, B(A1HL3)=32.67.
**********************************
                          CAS 609-70-1 (8911)
4-Hydroxypyridine-3-carboxylic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaCl 25°C 0.60M C
                      B2=20.5
                                 2002DYa (42775) 424
                        B3=25.76
                        B(AlHL)=18.19
                        B(A1H2L2)=35.15
                        B(A1HL2)=28.7
B(A1H3L3)=50.76, B(A1H2L3)=42.92, B(A1HL3)=34.77.
*******************************
             H2L
                 3-Nitrocatechol CAS 6665-98-1 (2685)
1,2-Dihydroxy-3-nitrobenzene; O2N.C6H3(OH)2
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl KCl 25°C 0.10M M K1=14.17 B2=26.30 1986HAb (42855) 425
Al+++
                   B3=35.81
***********************************
             H2L 4-Nitrocatechol CAS 3316-09-4 (890)
1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl KCl 25°C 0.10M C
                        K1=13.89 B2=26.33 2004GAa (42911) 426
Al+++
                       B3=37.08
______
Al+++ gl KCl 25°C 0.10M C
                        K1=13.75 B2=25.44 1997DSa (42912) 427
                        B3=34.38
                       B(AlH-1L2)=17.93
______
Al+++
      gl KCl 25°C 0.10M M
                       K1=13.74 B2=25.39 1984HAd (42913) 428
                        B3=34.31
********************************
                           CAS 635-22-3 (763)
C6H5N2O2C1
3-Nitro-4-chloroaniline; H2N.C6H3(C1)(NO2)
 Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp diox/w 25°C 100% U
                                 1976BSa (42977) 429
                     K(AlCl3+L)=1.67
***********************************
              L o-Chloroaniline CAS 95-51-2 (3088)
2-Chloroaniline (1-amino-2-chlorobenzene); Cl.C6H4.NH2
```

```
Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
 sol NaNO3 55°C 0.10M C T K1=<-2.37
                               1998YFa (43199) 430
At 80 C, K1<-2.50.
************************************
C6H6N06P
                         CAS 330-13-2 (5865)
            H2L
4-Nitrophenylphosphoric acid; NO2.C6H4.O.PO.(OH)2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
                       K1=4.80
            25°C 0.20M C
     gl KCl
                               1996AKa (43246) 431
                      B(AlH-1L)=1.19
                      B(A1H-2L)=-5.5
************************************
                          (8281)
3-Hydroxy-2-amidocarboxypyridine, Hydroxypicolinamide;
-----
    Mtd Medium Temp Conc Cal Flags Lg K values
-----
           25°C 0.10M C K1=7.54 B2=13.79 1990ARa (43373) 432
     gl KNO3
****************************
C6H6N2O2
                m-Nitroaniline CAS 99-09-2 (464)
3-Nitroaminobenzene; H2N.C6H4.NO2
    Mtd Medium Temp Conc Cal Flags Lg K values
                                Reference ExptNo
______
     sp diox/w 25°C 100% U
                               1976BSa (43387) 433
                      K(AlC13+L)=2.35
***********************************
                p-Nitroaniline CAS 100-01-6 (465)
C6H6N2O2
             L
4-Nitroaminobenzene; H2N.C6H4.NO2
______
      Mtd Medium Temp Conc Cal Flags Lg K values
                                Reference ExptNo
-----
Al+++
      sp diox/w 25°C 100% U
                               1976BSa (43404) 434
                      K(AlCl3+L)=1.52
*********************************
                         CAS 2504-83-8 (1141)
Imidazolylpyruvic acid; C3H3N2.CH2.CO.COOH
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
            25°C 0.10M U K1=9.0
     gl KCl
                            B2=16.10 1975SDa (43451) 435
*******************************
C6H602
            H2L
                Catechol
                         CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
Al+++ gl KCl 25°C 0.20M C M K1=16.20 B2=29.26 1993KAa (43715) 436
                         B3=37.95
                         B(A12H-2L2)=24.05
B(AlLA)=26.92, B(AlLA2)=35.2, B(AlL2A)=37.12. H2A=salicylic acid
Al+++ sp KCl 25°C 0.10M C K1=16.22 1989SMa (43716) 437
______
Al+++ gl KCl 25°C 0.10M C
                         K1=16.89 B2=30.55 1985KPa (43717) 438
                         K3 = 8.98
                         *K(AlL) = -6.07
                         *K(A1L2)=-8.10
______
Al+++ gl KNO3 25°C 0.10M C
                                   1984MMb (43718) 439
                         K(A1+H2L=A1L+2H)=-6.08
                         K(All+H2l=All2+2H)=-9.18
                         K(All2+H2L=All3+2H)=-13.52
                         K(All2+H)=6.03
.....
Al+++ gl NaCl 25°C 0.60M U
                                   19830Sa (43719) 440
                         B(-2,1,1)=-6.337
                         B(-4,1,2)=-15.44
                         B(-6,1,3)=-28.62
                         B(-5,1,2)=-23.45
B(-9,3,3)=-29.91. B(p,q,r):pH+qAl+r(H2L)
            ______
Al+++ gl KNO3 25°C 0.20M U K1=15.31 B2=27.67 1982H0b (43720) 441
                        K3=7.74
Al+++ gl KCl 25°C 0.20M U
                         K1=16.27 B2=29.75 1970G0a (43721) 442
                         K3 = 9.00
______
Al+++ gl KNO3 20°C 0.10M U
                         K1=16.9 B2=30.50 1969HBa (43722) 443
                        K3 = 8.9
-----
Al+++ gl KNO3 ? 0.20M U K1=16.56 B2=32.20 1964DMa (43723) 444
                        K3=13.65
*******************************
C6H6O3
                  Pyrogallol CAS 87-66-1 (696)
             H3L
1,2,3-Trihydroxybenzene; C6H3(OH)3
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 22°C 0.20M U K1=24.50 B2=44.55 1970G0b (43946) 445
                         K3=13.40
Al+++ gl KNO3 ? 0.20M U
                                   1967DMa (43947) 446
                         K(Al+HL)=14.3
                         K(AlHL+HL)=13.5
                         K(Al(HL)2+HL)=11.9
**********************************
```

```
C6H6O3
                Isomaltol CAS 3420-59-5 (5885)
             HL
1-(3-Hydroxy-2-furanyl)ethanone;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl 25°C 0.15M C K1=5.66 B2=10.42 1989LCa (44031) 447
                       K3 = 4.03
************************
                Maltol
                         CAS 118-71-8 (2442)
3-Hydroxy-2-methyl-4H-pyran-4-one;
_____
     Mtd Medium Temp Conc Cal Flags Lg K values
______
Al+++ nmr NaCl 25°C 1.50M C T H
                               2002YPa (44073) 448
Method: 27Al and 170 nmr measurements at 0-65 C.
DH(Al+HL=AlL+H)=22 kJ mol-1; DH(AlL+HL=AlL2+H)=28 kJ mol-1.
______
     gl NaCl 25°C 0.15M C
                       K1=8.44
                             B2=15.54 1991JSb (44074) 449
                      B3=22.16
-----
Al+++ gl NaCl 25°C 0.60M C
                                1988HOa (44075) 450
                       B(-1,1,1)=-0.130
                       B(-2,1,2)=-0.956
                       B(-3,1,3)=-2.669
                       B(-4,2,2)=-7.203
B(p,q,r): pH+qAl+r(HL)=HpAlq(HL)r
-----
      gl KNO3 25°C 0.10M U
                       K1=7.7
                             B2=15.25 1969CBb (44076) 451
Al+++
                      K3 = 6.65
*******************************
                Kojic acid
                         CAS 501-30-4 (1800)
             HL
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++
     gl NaCl 25°C 0.60M C
                                 1988HOa (44194) 452
                       B(-1,1,1)=-0.371
                       B(-2,1,2)=-1.499
                       B(-3,1,3)=-3.564
                       B(-4,2,2)=-7.656
B(p,q,r): pH+qAl+r(HL)=HpAlq(HL)r
______
Al+++ sp KCl 25°C 0.10M C K1=7.66 1987PEa (44195) 453
-----
     EMF KCl 21°C 0.10M U
                        K1=7.7 B2=14.2 19590Kb (44196) 454
                       B3=19.5
Method: H electrode
**********************************
                          CAS 7134-09-0 (3687)
3,4-Dihydroxybenzenesulfonic acid; (HO)2.C6H3.SO3H
```

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K valu	ues	Reference	ExptNo	
Al+++							K3=9.3				455
**************************************			H4L	Tir	on		CAS 1	149-45-1	******* (104)	*****	
Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K valu	ues	Reference	ExptNo	
Al+++				0.20M	C		K1=15.75 B3=38.97 B(AlAL)=22 B(AlA2L)=3	2.48	2002FCa	(44397)	456
A is aceto	hydro	oxamic a	acid.								
Al+++	gl	KNO3	25°C	0.10M	C		K(Al+H2L=# K(AlL+H2L= K(AlL2+H2I	198 -=(AlL+2H -AlL2+2H	88YYa (443 [,] 3.11 =-6.26	98) 457	
Al+++	gl	KCl	30°C	0.10M	U		K1=15.48 K3=12.72	B2=30.09	9 1980BDe	(44399)	458
Data for I At I=0, K1						40 C.		S values.			
Al+++ Method: st			34°C	0.10M	С		K(Al+H2L=		79BMb (444 2.62	00) 459	
Al+++	gl	KNO3	20°C	0.10M	U		K3=9.7				460
Al+++	gl	NaNO3	25°C	0.20M	U				5 1968ASa		461
Al+++ By glass e								196	65DMa (444	03) 462	
Al+++	gl	oth/un	25°C	0.0	U		K1=19.02 K3=2.4	B2=31.10	0 1957NAd	(44404)	463
	****	*****	*****						******	*****	
C6H7N Aminobenze	ene, a	aniline	L ; C6H5		lin	e	CAS 6	62-53-3	(583)		
Metal	Mtd	Medium	Temp	Conc	 Cal	Flags	Lg K valu	ues 	Reference	ExptNo	
Al+++	sol	NaNO3	80°C	0.10M	C		K1=<-0.93	3 199	98YFa (448	67) 464	
**************************************	****	******	***** HL	*****	***	*****			********* 5	******	
COLLYNOS			IIL				CAS .	19365-01-6	(0//1)		

```
1-Methyl-3-hydroxy-2-pyridinone;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.10M C K1=9.41 B2=17.79 1992CMc (45023) 465
                         B3=25.10
**********************************
                            CAS 17184-19-9 (5888)
3-Hydroxy-2-methylpyridin-4(1H)-one;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaCl 25°C 0.15M C I
                         K1=11.87 B2=22.54 1989CNa (45047) 466
                          B3=32.05
Data also at I=0.6 M(NaCl): K1=11.43, B2=21.73, B3=30.41
*******************************
                            CAS 701-64-4 (5866)
C6H7O4P
Phenyl phosphoric acid; C6H5O.PO(OH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.20M C
                         K1=5.29 1996AKa (45230) 467
                          B(AlH-1L)=1.51
                          B(A1H-2L)=-4.5
**********************************
                  Tricarballylic CAS 99-14-9 (1620)
             H3L
1,2,3-Propanetricarboxylic acid; HOOC.CH2.CH(COOH).CH2.COOH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaCl 37°C 0.15M C
                          K1=5.44
                                 1982JAc (45561) 468
                         B(A1HL) = 8.85
                         B(AlH-1L)=1.88
*****************
C6H806
             H2L Ascorbic acid CAS 50-81-7 (285)
Ascorbic acid (Vitamin C);
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl 25°C 0.60M C
                                    19920Na (45624) 469
                          B(-1,1,1)=-2.59
                          B(-6,3,1)=-18.38
                          B(-9,3,4)=-24.19
B(p,q,r); pH+qAl+rHL=Hp(Al)q(HL)r
Note: L-ascorbic acid is here defined as HL
**********************************
             H3L
                 Isocitric acid CAS 1637-73-6 (2527)
2-Hydroxy-3-carboxypentanedioic acid; HOOC.CH(OH).CH(COOH).CH2.COOH
-----
Metal
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
K1=6.905
Al+++ gl KCl 25°C 0.10M C
                                     1995PTa (45731) 470
                            B(A1HL)=9.55
                            B(AlH-1L)=3.06
                            B(A12H-3L2)=4.08
An alternative model gave K1=6.96, B(AlHL)=9.37, B(AlH-1L)=2.92,
B(A13H-4L3)=10.36.
H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.20M C K1=7.85 B2=12.73 2001LBa (46012) 471
                           B(A1HL)=10.18
                            B(AlH-1L)=4.27
                            B(A1H-2L)=-1.77
                            B(AlH-1L2)=7.81
B(A1H-2L2)=0.4, B(A13H-4L3)=16.34.
______
Al+++ cal oth/un 25°C 0.04M C H
                                      1997WDa (46013) 472
DH(Al+L=Al(OH)L+H)=41.6 \text{ kJ mol}-1; DH(Al(OH)L+3OH=Al(OH)4+L)=-29.2 \text{ kJ mol}-1
DH(Al+4OH=Al(OH)4)=-43.4 \text{ kJ mol}-1.
______
Al+++ sp oth/un 23°C 0.10M U
                                      1994KGa (46014) 473
                           Keff=9.4
Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=5.34 with HCl
Al+++ gl NaCl 37°C 0.15M C
                           K1=8.34 B2=13.56 1990FDa (46015) 474
                            B(A1HL)=10.78
                            B(AlH-1L)=5.30
                            B(AlH-1L2)=7.99
                            B(A1H-2L)=-0.59
______
Al+++ gl NaNO3 25°C 0.50M M M
                                      1989MAa (46016) 475
                           K(Al+H3L=AlH-1L+4H)=-6.6
                            K(2A1H-1L=A12H-2L2)=-14.8
K(UO2+A1+2H4L=A1UO2H-2L2+8H)=-8.21
______
Al+++ gl NaCl 37°C 0.15M C
                            K1=8.246 B2=13.068 1989VBa (46017) 476
                            B(A1HL)=10.502
                            B(A1H-2L)=6.777
                            B(AlH-2L2)-0.209
                            B(A12H-2L2)=12.694
 ______
Al+++ gl NaCl 25°C 0.60M C
                                      19880Hb (46018) 477
                           B(-4,1,1)=-8.48
                            B(-5,1,1)=-14.71
                            B(-16,3,3)=-47.11
```

```
B(p,q,r)=pH+qAl+rH3L=HqAlq(H3L)r
______
Al+++ gl NaCl 37°C 0.15M U
                          K1=8.34 B2=13.69 1987FDb (46019) 478
                           B(A1HL)=10.79
                           B(AlH-1L)=5.29
                           B(A1H-2L)=-0.53
                           B(AlH-1L2)=8.21
_____
                         K1=8.25 B2=13.07 1987VBe (46020) 479
Al+++ gl NaCl 37°C 0.15M U
                          B(A1HL)=10.50
                          B(A1H-2L2)=-0.21
                           B(A12H-2L2)=12.69
                           B(A13H-4L3)=15.08
B(A1H-1L2)=6.78
             Al+++ gl KCl 25°C 0.10M C
                          K1=8.10 B2=12.90 1986GPc (46021) 480
                          B(AlHL)=10.81
                          K(All2=AlH-1l2+H)=-6.10
                          K(Al+L+HL)=11.14
                           B(A1HL2)=16.84
K(AlH-1L2=AlH-2L2+H)=-7.17
-----
                         K1=8.0 B2=13.00 1986MAa (46022) 481
Al+++ NaCl 25°C 0.15M U
                           K(A1+HL)=4.7
                           K(AlHL=AlL+H)=-2.5
                           K(All=ALH-1L+H)=-3.4
25-37 C. From a survey of literature data
______
Al+++ kin NaNO3 205°C var U K1=10.72 1984LKa (46023) 482
                          K(A1+HL)=6.56
                          K(A1+H2L)=2.91
------
                          K1=7.98 1984MMb (46024) 483
Al+++ gl KNO3 25°C 0.10M C
                          K(All+H)=2.94
                          K(All=AlH-1L+H)=-3.31
______
Al+++ gl NaCl 25°C 0.60M C
                                     19830Sd (46025) 484
                          B(-2,1,1)=-2.68
                          B(-3,1,1)=-4.925
                           B(-6,1,2)=-12.53
                           B(-13,3,3)=-21.77
B(p,q,r): pH + qAl + rH3L = AlqHp(H3L)r
Al+++ gl NaCl 25°C 0.60M C
                                     19830Se (46026) 485
                          B(-2,1,1)=-2.68
                           B(-3,1,1)=-4.925
                           B(-6,1,2)=-12.53
                          B(-13,3,3)=-21.77
B(p,q,r): pH+qAl+r(H3L)=HpAlq(H3L)r
```

Al+++	gl NaCl	37°C 0.15M C	K1=7.87 1982JAc (46027) 486 B(AlHL)=10.12 B(AlH-1L)=4.64 B(AlH-1L2)=8.8
Al+++	gl NaCl	25°C 0.12M C	1981RMb (46028) 487 K(Al+H-1L)=18.0 K(AlL(OH)2+2H)=18.4
Al+++	gl NaClO	4 25°C 0.10M U M	K1=8.65 1975RMa (46029) 488 B(AlL(Cys))=14.90 K(Al+L+HPO4)=19.29
		4 33°C 0.25M U	1961PPa (46030) 489 K(Al+H3L=AlL+3H)=-4.7 K(AlH-1L+H)=3.5 K(Al(OH)H-1L+H=AlH-1L)=6.8 ***********************************
C6H9NO6 N-Carboxyn	nethyl-L-as	H3L partic acid;	CAS 41035-84-1 (4367)
Metal	Mtd Mediu	m Temp Conc Cal Fla	gs Lg K values Reference ExptNo
Al+++	gl KCl	25°C 0.20M C	K1=2.24 1997KSa (46374) 490 B(AlHL)=6.15 B(AlH-1L)=-1.91
C6H9N06		**************************************	**************************************
Metal	Mtd Mediu	m Temp Conc Cal Fla	gs Lg K values Reference ExptNo
Al+++	gl NaCl	25°C 0.60M C	K1=11.097 19900Hb (46686) 491 B(AlH-1L)=5.67 B(AlH-2L)=-2.52 B(Al2H-2L2)=13.16
			K1=10.80 1984NAa (46687) 492
Al+++	gl NaClO	4 25°C 0.10M U T 2=18.57; 45 C: 10.1	K1=10.53 B2=19.08 1981DSa (46688) 49 8, 18.32
Al+++	gl KNO3	35°C 0.10M U	K1=11.61 1980KHb (46689) 494
			K1=12.72 1975RMa (46690) 495 B(AlL(Cys))=18.89 K(Al+L+HPO4)=23.89
Al+++	•	4 25°C 0.20M U	K1=11.37 1967BDb (46691) 496 LOH+H)=5.09, K(AlL(OH)2+H)=8.28

Al+++ dis NaCl04 20°C 0.10M U T K1=9.5 1963STC (46692) 497 Al+++ gl KCl 20°C 0.10M U K1=>10 1948SBa (46693) 498 K(AlLOH+H)=5.8 K(AlLOH+H)=5.8 K(AlLOH+H)=5.8 K(AlLOH+H)=5.8 K(AlLOH+H)=5.8 K(AlLOH+H)=6.6 **********************************										
K(AlLOH+H)=5.8	Al+++	dis	NaC104	20°C	0.10M U	T K1=9.5	5 19	63STc	(46692)	497
C6H9N3O2						K(AlLOH K(AlL(H+H)=5.8 DH)2+H)=8.6		•	
Al+++ gl NaCl 37°C 0.15M C 2002DCa (47529) 499 B(Al2H-2L)=1.163 Al+++ gl NaCl04 25°C 0.50M U 1984CDa (47530) 500 K(Al(OH)+L)=8.45 ***********************************	C6H9N3O2			HL	Histid:	ine CA	AS 71-00-1	(1)	*****	****
### B(A12H-2L)=1.163 Al+++ g1 NaCl04 25°C 0.50M U	Metal	Mtd	Medium	Temp	Conc Cal	Flags Lg K v	values	Refer	ence Exp	otNo
K(Al(OH)+HL)=3.62 K(Al(OH)+L)=8.45 ***********************************	Al+++	gl	NaCl	37°C	0.15M C	B(Al2H		02DCa	(47529)	499
C6H100452 H2L CAS 7244-02-2 (438) 1,2-Bis(carboxymethylthio)ethane; H0OC.CH2.S.CH2.CH2.S.CH2.COOH Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo Al+++ gl NaCl04 25°C 0.50M C K1=2.05 1985NAb (48234) 501 B(AlHL)=12.46 ***********************************						K(Al(O	H)+HL)=3.62 H)+L)=8.45		, ,	
Al+++ gl NaCl04 25°C 0.50M C K1=2.05 1985NAb (48234) 501 B(AlHL)=12.46 ***********************************	C6H10O4S2			H2L		CA	AS 7244-02-2	(438		*****
B(AlHL)=12.46 ***********************************	Metal	Mtd	Medium	Temp	Conc Cal	Flags Lg K v	values	Refer	rence Exp	otNo
C6H1008		•				B(AlHL))=12.46		•	
Al+++ gl KNO3 25°C 0.10M C B(AlH-1L)=1.57 *K(AlH-1L)=-3.76 ***********************************	C6H1008			H2L	Sacchai	ric acid CA	AS 87-73-0	(1191))	
B(AlH-1L)=1.57 *K(AlH-1L)=-3.76 ***********************************	Metal	Mtd	Medium	Temp	Conc Cal	Flags Lg K v	values	Refer	ence Exp	otNo
**************************************	Al+++	gl	KNO3	25°C	0.10M C	•	1L)=1.57	 84MMb	(48467)	502
Al+++ gl NaClO4 25°C 0.50M C K1=7.49 1984NAa (48687) 503 Al+++ gl KNO3 35°C 0.10M U K1=9.33 1980KHb (48688) 504 ***********************************	C6H11N05			H2L	HIMDA	************ CA cid; HO.CH2.0	*********** AS 93-62-9 CH2.N(CH2.CO	(192)	******	****
Al+++ gl NaClO4 25°C 0.50M C K1=7.49 1984NAa (48687) 503 Al+++ gl KNO3 35°C 0.10M U K1=9.33 1980KHb (48688) 504 ***********************************				-					-	
Al+++ gl KNO3 35°C 0.10M U K1=9.33 1980KHb (48688) 504 ************************************					0.50M C	K1=7.4	1 9 1 9	84NAa	(48687)	503
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	********* C6H11N05	****	******	***** H2L	0.10M U ******	K1=9.3	33 19 ******* (1233)	80KHb	(48688)	504
	Metal	Mtd	Medium	Temp	Conc Cal	Flags Lg K v	values	Refer	rence Exp	tNo

```
Al+++ gl KNO3 25°C 0.10M C K1=5.9 1987AKa (48839) 505
*************************
                          CAS 4726-83-4 (5911)
N,N-Dihydroxyhexanediamide; HN(OH).CO.(CH2)4.CO.NH(OH)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ gl NaNO3 25°C 0.10M C K1=14.20 1989EHa (49332) 506
*************************
                        CAS 498-43-1 (5803)
C6H12O6
3-Deoxy-D-ribohexanoic acid:
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                       K1=1.97
Al+++ gl NaCl 25°C 0.10M C
                                1994BHa (49529) 507
                        *K(AlL)=-2.83
                       K(AlH-1L=AlH-3L+2H)=-9.17
********************************
                Gluconic acid CAS 526-95-4 (904)
            HL
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaNO3 25°C 0.10M C
                                 1995E0a (49695) 508
                       B(AlH-1L)=-0.84
                       B(A1H-3L)=-10.70
-----
Al+++ gl NaCl 25°C 0.10M C
                        K1=2.01
                                 1994BHa (49696) 509
                       *K(AlL)=-2.89
                       K(AlH-1L=AlH-3L+2H)=-9.30
______
Al+++ gl KNO3 25°C 0.10M C
                        K1=1.98
                                 1984MMb (49697) 510
                        *K(AlL) = -2.87
                        K(AlH-1L=AlH-3L+2H)=-9.29
*********************************
C6H13N02
             HL
                Leucine
                          CAS 61-90-5 (47)
2-Amino-4-methylpentanoic acid; H2N.CH(CH2.CH(CH3)2)COOH
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++
     oth NaClO4 35°C 0.10M C
                              B2=14.01 1986SGd (50058) 511
                       K1=7.92
                       B3=18.90
Method: electrophoresis
***********************************
                          CAS 4312-93-0 (4386)
Hexanohydroxamic acid; CH3.CH2.CH2.CH2.CH2.CO.NH.OH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl
            25°C 0.20M C K1=8.32 B2=16.17 2000FEc (50226) 512
```

```
B(AlH-1L2)=11.26
*******************************
                    Bicine
                                CAS 150-25-4 (2124)
N,N-Bis(2-hydroxyethyl)glycine; (HO.CH2.CH2)2N.CH2.COOH
       Mtd Medium Temp Conc Cal Flags Lg K values
                       _____
       gl KNO3
               25°C 0.10M C
Al+++
                                        1984MMb (50340) 513
                            K(Al+HL)=3.38
                            K(AlHL=AlH-2L+3H)=-13.72
********************************
C6H1309P
                                CAS 59-56-3 (3049)
alpha-D-Glucose-1-phosphoric acid; Glucopyranose-1-phosphoric acid;
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
       gl KCl
               25°C 0.20M C
Al+++
                             K1=5.60 B2=10.41 2001CRa (50620) 514
                            B(A1H2L)=ca.9.40
                            K(A1HL)=8.50
                            B(A1HL2)=14.04
                            B(A12H-1L2)=11.08
B(A12H-2L2)=5.73, B(A12H-3L2)=0.84, B(A12H-4L2)=-5.09, B(A1H-3L)=-11.80.
Additional methods: 13C, 27Al and 31P nmr.
*********************************
                    Arginine
C6H14N4O2
                HL
                                CAS 74-79-3 (40)
2-Amino-5-guanidopentanoic acid; H2N.CH((CH2)3.NH.C(:NH)(NH2)COOH
______
       Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                    ? U T K1=6.63
       gl oth/un 25°C
                                    B2=12.86 1960PEd (51001) 515
7 C: K1=6.67, K2=6.38
******************************
C6H15N3O2
                                CAS 52760-35-7 (6670)
Lysine hydroxamic acid; H2N.(CH2)4.CH(NH2)CO.NHOH
______
      Mtd Medium Temp Conc Cal Flags Lg K values
                                          Reference ExptNo
-----
Al+++
      gl KCl 25°C 0.20M C
                             B2=21.28
                                        2002ECa (51424) 516
                            B(A1H2L)=24.21
                            B(A1H2L2)=36.94
                            B(A1HL2)=29.84
*********************************
C6H15N3O3
                                 (6613)
1,3,5-Triamino-1,3,5-trideoxy-cis-inositol,5-Amino-5-deoxy-streptamine;
-----
       Mtd Medium Temp Conc Cal Flags Lg K values
                                          Reference ExptNo
Al+++
       gl KCl
              25°C 0.10M C
                             K1=11.8 B2=18.8
                                           1993HGa (51447) 517
                            B(A1HL2)=25.3
                            *K(A1L2) = -8.1
```

```
*K(AlH-1L2)=-8.9
*K(AlH-2L2)=-9.1
```

```
*K(A1H-3L2)=-9.7
**********************************
                 Ins(1,2,6)P3 CAS 28841-62-5 (6479)
C6H15O15P3
             H6L
D-myo-Inositol 1,2,6-trisphosphoric acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl R4N.X 25°C 0.10M C
                        K1=13.72
                                  1995MBb (51533) 518
                        B(A1HL)=19.18
                        B(A12L)=19.72
                        B(AlH-1L)=6.10
Medium: 0.10 M (n-Bu)4NBr.
***********************************
             H6L
                          CAS 88269-39-0 (8168)
C6H15O15P3
D-myo-Inositol-1,4,5-triphosphoric acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                        K1=13.37 1995MBb (51543) 519
Al+++ gl R4N.X 25°C 0.10M C
                        B(A1HL)=18.98
                        B(AlH-1L)=5.84
                        B(AlH-2L)=-1.82
Medium: 0.10 M (n-Bu)4NBr.
***********************************
             H4L
                           CAS 4721-22-6 (3708)
C6H16O6P2
Hexane-1,6-diphosphonic acid; H2O3P(CH2)6PO3H2
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.10M U K1=14.66 1967KLa (51791) 520
*******************************
                           CAS 609-99-4 (400)
3,5-Dinitrosalicylic acid; (O2N)2.C6H2(OH).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                       K1=7.53 B2=13.24 1996MMa (52461) 521
Al+++ gl KNO3 30°C 0.10M C
                       K3 = 4.24
-----
     gl NaClO4 25°C 0.10M U
                        K1=8.81 B2=15.39 1979LTc (52462) 522
                       K3=4.34
*********************************
             H2L Quinolinic acid CAS 89-00-9 (567)
2,3-Pyridinedicarboxylic acid; C5H3N.(COOH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     nmr NaCl 25°C 1.33M U K1=3.72 B2= 7.10 2001LKc (52620) 523
```

```
Method: 27Al NMR spectroscopy.
_____
     gl NaClO4 25°C 0.50M C K1=4.50 B2=7.00 1986MNb (52621) 524
CAS 499-80-9 (566)
C7H5N04
           H2L
2,4-Pyridinedicarboxylic acid; C5H3N.(COOH)2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaClO4 25°C 0.50M C K1=4.35 B2=7.30 1986MNb (52649) 525
*************************
C7H5N04
                        CAS 100-26-5 (2528)
2,5-Pyridinedicarboxylic acid, Isocinchomeronic acid; C5H3N.(COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl NaClO4 25°C 0.50M U K1=3.95 B2=7.24 1970NAb (52664) 526
*******************************
               Dipicolinic aci CAS 449-83-2 (418)
2,6-Pyridinedicarboxylic acid; C5H3N.(COOH)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp NaCl 25°C 1.00M U
                    K1=11.37 1993NRa (52751) 527
                      K(A1+HL)=7.44
_____
     gl NaCl04 25°C 0.50M U K1=4.87 B2=8.32 1968NAc (52752) 528
By spectrophotometry: K1=4.85
*************************************
           H2L Nitrosalicylic CAS 85-38-1 (1416)
C7H5N05
2-Hydroxy-3-nitrobenzoic acid; HO.C6H3(NO2).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KNO3 30°C 0.10M C K1=10.22 B2=18.20 1996MMa (52972) 529
                      K3=6.87
                      K(A1L3+H)=6.56
Nitrosalicylic CAS 96-97-9 (148)
           H2L
2-Hydroxy-5-nitrobenzoic acid; HO.C6H3(NO2).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.10M C
                      K1=10.65 B2=19.81 2004GAa (53035) 530
                     B3=25.74
Al+++ gl KCl 30°C 0.10M C
                      K1=10.91 B2=19.67 1996MMa (53036) 531
                      K3=5.36
                     K(A1L3+H)=6.24
```

```
gl NaClO4 25°C 0.10M U
                       K1=11.11 B2=19.73 1979LTc (53037) 532
Al+++
                       K3=6.13
-----
     sp oth/un 25°C 0.10M C
                                1979PTb (53038) 533
                       K(Al+HL=AlL+H)=1.11
     sp NaClO4 29°C 1.00M U
                                1976DDa (53039) 534
                       K(Co(en)2(NH3)L+A1)=-0.45
**********************************
C7H5O3Br
                          CAS 85-55-4 (1194)
5-Bromosalicylic acid; Br.C6H3(OH).COOH
  Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp NaClO4 29°C 1.00M U
                                 1976DDa (53307) 535
                       K(Co(en)2(NH3)L+A1)=-0.72
******************************
                Tropolone
                         CAS 533-75-5 (3129)
2-Hydroxycyclohepta-2,4,6-trien-1-one;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaCl 25°C 0.60M C
                                 19900Ha (53665) 536
                       K(Al+3HL=AlL3(s)+3H)=11.21
                       Kso = -31.05
*********************************
                Benzoic Acid CAS 65-85-0 (462)
Benzenecarboxylic acid; C6H5.COOH
______
                                Reference ExptNo
Metal Mtd Medium Temp Conc Cal Flags Lg K values
-----
Al+++ gl NaCl 25°C 0.60M C
                                 19910Ha (53821) 537
                       B(-1,1,1)=-2.67
                       B(-3,2,1)=-7.446
B(p,q,r); pH+qAl+rHL=HpAlq(HL)r
______
     gl NaClO4 25°C 0.50M U
                                 1970NLa (53822) 538
                       B(All(OH))=12.1
******************
                              *********
                Salicylic acid CAS 69-72-7 (14)
            H2L
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp oth/un 23°C 0.10M U
                                 1994KGa (54139) 539
                       Keff=4.8
Method: spectrophotometric using pyrocatechol violet. Tris buffer adjusted
to a pH=5.34 with HCl
     Al+++ gl NaCl 25°C 0.60M C T
                                 19830Sc (54140) 540
```

```
B(-2,1,1)=-3.052
                          B(-4,1,2)=-8.39
                          B(-5,1,2)=-15.99
                          B(-6,1,2)=-25.31
B(p,q,r); pH+qAl+r(H2L)=HpAlq(H2L)r
Al+++ gl NaCl 25°C 0.12M C
                         T K1=13.7 B2=26.70 1981RMb (54141) 541
                          K3=10.73
                          K(A1H-2L+2H)=10.40
                          K(A1H-3L+H)=9.37
      sp NaClO4 29°C 1.00M U
                                    1976DDa (54142) 542
                          K(Co(en)2(NH3)L+A1)=-0.46
Al+++ kin NaClO4 25°C 0.10M U
                                    1975SVa (54143) 543
                          K(Al+HL=AlL+H)=0.067
_____
     gl KNO3 20°C 0.10M U
                                 B2=23.20 1969HBb (54144) 544
                          K1=12.9
                         K3 = 6.6
Al+++ EMF NaNO3 18°C 1.0M U
                                    1961COb (54145) 545
                         K(A1+3HL=A1HL2+H2L)=4.5
Method: quinhydrone electrode
______
Al+++ sp NaClO4 27°C 0.20M U I
                                    1959DAa (54146) 546
                          K(Al+HL=AlL+H)=-0.18
K=0.26(I=0), 0.06(I=0.02), 0.01(I=0.05), -0.13(I=0.10). Recalculated values
______
Al+++ sp oth/un 27°C ->0 U K1=14.11 1959DAa (54147) 547
************************************
                             CAS 99-06-9 (1370)
3-Hydroxybenzoic acid; HO.C6H4.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl 25°C 0.60M C
                                    19910Ha (54375) 548
                          B(-1,1,1)=-2.59
                          B(-3,2,1)=-7.453
B(p,q,r); pH+qAl+rHL=HpAlq(HL)r
***********************************
                             CAS 303-38-8 (1398)
C7H604
2,3-Dihydroxybenzoic acid; C6H3(OH)2.COOH
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.10M C
                          K1=10.62 B2=19.20 2004GAa (54461) 549
                          B(A12H-3L2)=10.30
                         B(AlH-1L2)=13.50
_____
Al+++ gl KCl 25°C 0.20M C T K1=10.32 B2=18.26 1993KAa (54462) 550
```

B(AlH-1L2)=11.56 B(AlH-2L2)=1.74 B(Al2H-2L2)=13.62 B(Al2H-3L2)=8.87

```
*********************************
                 Resorcylic acid CAS 89-86-1 (876)
            H3L
2,4-Dihydroxybenzoic acid, b-Resorcylic acid; C6H3(OH)2.COOH
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.20M C
                       T K1=8.71 B2=15.03 1993KAa (54514) 551
                        B(A12H-2L2)=9.1
                        B(AlH-1L2)=7.21
C7H604
            H3L
                          CAS 409-79-9 (1115)
2,5-Dihydroxybenzoic acid; C6H3(OH)2.COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KCl 25°C 0.20M C T K1=9.74 B2=17.17 1993KAa (54580) 552
                       B(A12H-2L2)=11.5
                        B(AlH-1L2)=9.97
______
                       Т
Al+++ gl NaClO4 25°C 0.50M C
                                 1985CDa (54581) 553
                        K(Al+HL)=10.4
                       K(A1+2HL)=18.15
.....
Al+++ sp NaClO4 20°C 0.09M U TIH
                                 1971SGa (54582) 554
                        K(A1+H2L=A1HL+H)=-3.96
K(Al+H2L=AlHL+H)(I=0.01)=-3.49, (I=0.20)=-4.47. DH=22.7 kJ mol-1, DS=173
********************************
            H3L
                 g-Resorcylic ac CAS 303-07-1 (1624)
2,6-Dihydroxybenzoic acid; C6H3(OH)2.COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                       T K1=12.79 B2=23.67 1993KAa (54603) 555
Al+++
      gl KCl 25°C 0.20M C
                        B(A12H-2L2)=17.2
                        B(A1H-1L2)=16.46
***********************************
                 Protocatechuic CAS 99-50-3 (875)
C7H604
            H3L
3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH
  Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl KCl 25°C 0.20M C
                       K1=16.47 B2=29.38 1993KAa (54657) 556
Al+++
                        B3=38.35
                        B(A1HL)=21.09
                        B(A12H-2L2)=25.03
```

```
sp none 25°C 0 U K1=15.33 1990SMd (54658) 557
Al+++
______
                          K1=16.1
      gl NaClO4 25°C 0.50M C
Al+++
                                     1985CDa (54659) 558
                          K(A1+HL)=7.8
Al+++ gl KCl 25°C 0.10M C
                          K1=16.87 B2=29.88 1985KPa (54660) 559
                           K3 = 8.76
                           *K(AlL) = -5.77
                           *K(A1L2)=-8.39
                           K(A1+H2L=A1H2L)=2.85
K(All+H)=4.66
           ______
                           K1=15.03 B2=27.64 1982H0b (54661) 560
Al+++
      gl KNO3 25°C 0.20M U
                          K3=9.91
***********************************
                              CAS 610-02-6 (3725)
2,3,4-Trihydroxybenzoic acid; (HO)3.C6H2.C0OH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl
             25°C 0.20M C
                                  B2=15.94 1993KAa (54719) 561
                           K1=9.10
                           B(A1H-1L2)=10.26
                           B(A1H-2L2)=2.26
                           B(A1H-2L3)=6.79
                           B(A12H-2L2)=11.09
B(Al2H-3L2)=6.49, B(Al2H-4L2)=-0.69. Ligand as H2L
******************************
                   Gallic acid
                             CAS 149-91-7 (446)
C7H605
              H4L
3,4,5-Trihydroxybenzoic acid; C6H2(OH)3.COOH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl 25°C 0.60M C
                                     19820Sa (54748) 562
                          B(-5,3,1)=-12.52
                           B(-9,4,3)=-20.25
B(p,q,r); pH+qAl+r(H3L)=HpAlq(H3L)r. K(Al+H3L+4H20=AlL.H20(s)+3H)=-6.2
Al+++
       gl NaCl 25°C 0.60M C
                                     19810Sa (54749) 563
                           B(-2,1,1)=-4.933
                           B(-3,1,1)=-9.43
                           B(-6,1,2)=-21.98
                           B(-9,1,3)=-37.69
                          B(-8,2,3)=-22.65, B(-9,2,3)=-27.81,
B(p,q,r): pH + qAl + rH3L = HpAlqH3Lr.
B(-10,2,3)=-32.87, B(-11,2,3)=-39.56.
*******************************
                              CAS 5965-83-3 (399)
C7H606S
              H3L
5-Sulfosalicylic acid, 2-Hydroxy-5-sulfobenzoic; HO3S.C6H3(OH).COOH
 -----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
```

```
Al+++ gl NaCl04 25°C 0.50M C K1=11.8 B2=21.2 1985CDa (54930) 564
Al+++ sp NaClO4 25°C 0.10M U
                                    1977PTa (54931) 565
                         K(Al+HL=AlL+H)=0.88
                      M 1976DDa (54932) 566
Al+++ sp NaClO4 29°C 1.00M U
                         K(Co(en)2(NH3)L+A1)=-0.30
-----
Al+++ gl KNO3 20°C 0.10M U
                         K1=12.3 B2=22.00 1969HBb (54933) 567
                         K3=5.8
Al+++ gl NaClO4 30°C 0.20M U K1=12.20 B2=22.21 1967AMa (54934) 568
Al+++ sp NaClO4 31°C 0.20M U TI
                                   1963NAa (54935) 569
                          K(A1+HL=A1L+H)=0.23
K=0.79(I=0.02), 0.55(I=0.05), 0.37(I=0.10). Recalculated values
At I=0.1 M: K=0.22(15 C),0.41(28 C),0.53(40 C); DH=20.5 kJ mol-1, DS=58
______
Al+++ sp NaClO4 25°C 0.10M U K1=12.91 B2=22.92 1960BSb (54936) 570
By glass electrode K1=13.20, K2=9.63, K4=6.06
______
Al+++ sp oth/un ? 0.20M U TI
                                   1957NAa (54937) 571
                         K(Al+HL=AlL+H)=0.19
K=0.36(I=0.02),0.31(I=0.05). Recalculated values
*********************************
                             CAS 56507-30-3 (2659)
3,5-Disulfosalicylic acid; (HO3S)2.C6H2(OH).COOH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaClO4 25°C 0.50M C T K1=11.51 B2=20.19 1978LAa (55091) 572
****************************
C7H7N02
                            CAS 495-18-1 (184)
Benzohydroxamic acid; C6H5.CO.NH.OH
 ·
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.20M C K1=7.57 B2=14.60 2000FEc (55494) 573
*******************************
          L Thiobenzamide CAS 2227-79-4 (1660)
Thiobenzamide; C6H5.CS.NH2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sp non-aq 25°C 100% U
                                    1977SWa (55702) 574
                         K(A1C13+L)=2.20
Medium: Et20
***********************************
                             CAS 99-52-5 (470)
2-Methyl-4-nitro-aminobenzene; CH3.C6H3(NO2).NH2
```

Metal	Mtd M	1edium	Temp	Conc	Cal	Flags	s Lg K values	Reference ExptNo
Al+++	·	diox/w					K(AlCl3+L)=2.2	
	*****	*****	****	*****	k***	*****		********
C7H8N2O2 2-Nitro-4-	methyl	Laminob	L enzer	ne; CH	13.C	6H3(N0	CAS 89-62 D2).NH2	-3 (466)
Metal	Mtd M	1edium	Temp	Conc	Cal	Flags	s Lg K values	Reference ExptNo
Al+++	sp n	non-aq	25°C	100%	U -	ТН	K(AlBr3+L)=1.3	1988DNa (55887) 576
By fibre-c to 25 C.	ptic p	hotome	etry i	in die	ethy.	lethe	` '	l-1 from data at -15
********	*****	*****	****	****	k***	*****		*************
C7H8N2O2			L		12 6	CU3 /***		2-4 (467)
3-Nitro-4-	methyl	Laminob	enzer	ne; Ch	13.C	6H3(N0	02).NH2	
Metal	Mtd M	1edium	Temp	Conc	Cal	Flags	s Lg K values	Reference ExptNo
Al+++	sp d	diox/w	25°C	100%	U			1976BSa (55903) 577
	·						K(AlCl3+L)=2.8	8
	*****	******	****	****	k***	*****		**************
C7H8N2O2			L	12 661	12 (N/	00 \ NI		5-2 (764)
4-Nitro-3-	metnyı	.aniiin	ie; Ci	13 . Cbi	13 (NC	J2).NF	12 	
Metal	Mtd M	Medium	Temp	Conc	Cal	Flags	s Lg K values	Reference ExptNo
Al+++	sp d	diox/w	25°C	100%	U		K(AlCl3+L)=1.8	1976BSa (55917) 578
*******	*****	*****	****	****	***	****	•	*********
C7H9NO 2-Ethoxypy	ridine	e; C5H4	L IN.OC2	2H5			CAS 14529	-53-4 (2473)
Metal	Mtd M	1edium	Temp	Conc	Cal	Flags	s Lg K values	Reference ExptNo
Al+++	sp n	non-aq	25°C	100%	U	М	K(AlCl3+L)=4.2	1981SKe (56378) 579
Medium: DN		ና ችችጥጥጥጥ	• ች ች ት ጥ •	የችጥ ተ	የ ች ጥ ጥ •	ችችጥጥጥ	,	******
C7H9NO	****	* * * * * * *	L	*****		****		7-79-6 (2635)
N-Ethylpyr	idine-	-2-one:	_)C5H4N	N-C2H	H5	CAS 13337	75 0 (2033)
Metal	Mtd M	Medium	Temp	Conc	Cal	Flags	s Lg K values	Reference ExptNo
Al+++	sp n	non-aq	25°C	100%	U	М		1981SKe (56400) 580
							K(AlC13+L)=4.6	1
Medium: DM *******		*** ***	:****	k****	k***	*****	******	******

```
C7H9N02
                            (8381)
             H2L
                 DHB
3,4-Dihydroxybenzylamine;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.10M C
                                  2002AOa (56411) 581
                        B(-2,1,1)=-5.58
                        B(-4,1,2)=-13.74
                        B(-6,1,3)=-25.51
                        B(-7,1,3)=-35.55
B(-8,1,3)=-46.39, B(-9,1,3)=-56.79, B(-6,2,2)=-19.58
B(p,q,r) defined for the protonated ligand, H3L+: pH+qAl+rH3L=HpAlq(H3L)r
**************************
C7H9N02
                          CAS 30652-11-0 (2458)
3-Hydroxy-1,2-dimethylpyridin-4(1H)-one; (OH)(CH3)(O:)C5H2N.CH3
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.10M C K1=12.20 B2=23.25 2004SGc (56423) 582
                       B3=32.62
-----
Al+++ gl KCl 25°C 0.10M C K1=12.20 B2=23.25 1994MRa (56424) 583
                       K3 = 9.37
______
            25°C 0.10M C K1=12.20 B2=23.25 1992CMb (56425) 584
Al+++ gl KCl
                       K3 = 9.37
-----
Al+++ gl NaCl 25°C 0.15M C I K1=11.91 B2=22.83 1989CNa (56426) 585
                        B3=32.35
Data also at I=0.6 M(NaCl): K1=11.57, B2=22.01, B3=30.90
********************************
                           CAS 157070-43-4 (7154)
3-Hydroxy-5-methyl-2-(N-methylformamido)furan;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaCl 25°C 0.60M C
                        K1=7.28 B2=13.42 1994LSc (56445) 586
                        K3=4.83
                        B(-1,1,1)=0.19
                        B(-2,2,1)=-0.76
                        B(-3,3,1)=-3.02
B(-7,3,3)=-10.43, B(-4,2,1)=-12.08. B(p,q,r): pH+qHL+rM=Hp(HL)qMr.
**************
C7H1007P2
             H4L
                           CAS 2809-26-9 (8731)
1-Phenyl-1-hydroxymethylene-1,1-diphosphonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
                        K1=18.50
Al+++ gl KNO3 25°C 0.10M C
                                  2002GKc (56762) 587
                        B(AlH-1L)=13.75
```

B(AlH-2L)=3.57 B(AlH3L2)=43.41

K(Al+HL)=12.50 B(Al2L)=23.09 K(Al+HL+L)=19.68 ***********************************					B(AlH3L2)=				
Al+++ gl KCl 25°C 0.10M U K1=18.97 B2=26.31 1969DMd (56938) 588 K(Al+HL)=12.50 B(Al2L)=23.09 K(Al+HL+L)=19.68 ***********************************	C7H11N06P2	2		H4L	CAS 4			******	
K(Al+HL)=12.50 B(Al2L)=23.09 K(Al+HL+L)=19.68 K(Al+HL)=19.68 K(Al+GL)=16.89 K(Al+G	Metal	Mtd	Medium	Temp Conc Cal	Flags Lg K valu	ies	Reference	ExptNo	
C7H1109P		J			K(Al+HL)=1 B(Al2L)=23 K(Al+HL+L)	12.50 3.09)=19.68		,	588
Al+++ gl NaNO3 25°C 0.50M C B(AlH-6L2)=18.42 B(AlH-6L)=16.06 **********************************	C7H1109P			H5L	(50	941)			
B(AlH-6L2)=18.42 B(AlH-6L)=16.06 **********************************	Metal	Mtd	Medium	Temp Conc Cal	Flags Lg K valu	ies	Reference	ExptNo	
C7H1202	Al+++	gl	NaNO3	25°C 0.50M C	,	=18.42	99SEa (5702	23) 589	
Al+++ gl NaCl 25°C 0.60M C 19910Ha (57227) 590 B(-1,1,1)=-3.48 B(-3,2,1)=-8.04 B(p,q,r); pH+qAl+rHL=HpAlq(HL)r ***********************************	C7H12O2			HL	CAS 9	98-89-5 ((2793)	******	
B(-1,1,1)=-3.48 B(-3,2,1)=-8.04 B(p,q,r); pH+qAl+rHL=HpAlq(HL)r ************************************	Metal	Mtd	Medium	Temp Conc Cal	Flags Lg K valu	ies	Reference	ExptNo	
<pre>%************************************</pre>	Al+++	gl	NaCl	25°C 0.60M C		-3.48	910Ha (5722	27) 590	
C8H6O4					******	*******	·*******	******	
Al+++ gl NaCl 25°C 0.60M C K1=2.94 B2=4.97 1988HBa (58943) 591	C8H604			H2L Phthali	c acid CAS 8				
Al+++ gl NaCl 25°C 0.60M C K1=2.94 B2=4.97 1988HBa (58943) 591				-				=	
B(p,q,r): pH+qAl+rL=HpAlqLr					K1=2.94 B(-2,2,1)= B(-4,3,1)=	B2=4.97 =-2.50 =-8.47			591
**************************************	B(p,q,r):	pH+q	Al+rL=H	pAlqLr	2(-)-)-/	0,07			
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo	**************************************	****	*****	************* H4L	**************************************	*******		•	
	Metal	Mtd	Medium	Temp Conc Cal	Flags Lg K valu		Reference	ExptNo	

```
Al+++ gl KCl 25°C 0.20M C
                          K1=3.90
                                    1993KAa (59077) 593
                         B(AlH-1L)=0.96
                         B(A1H-2L2)=-0.36
                         B(A1H-3L2)=-8.11
                         B(A12H-3L2)=1.36
B(Al3H-5L3)=2.27, B(Al3H-6L3)=-2.11. Ligand as H2L ?
*********************************
                            CAS 2153-11-9 (4570)
C8H8N02C1
N-Chloroacetyl-N-phenylhydroxylamine; Cl.CH2.CO.N(C6H5).OH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl mixed 30°C 50% U
                          K1=6.94
                                B2=14.52 1971GSc (59284) 594
                         K3 = 6.62
Medium: 50% acetone/H2O, 0.5 M NaClO4
************************
C8H8N2O2
                              (3821)
1-(2'-Hydroxyphenyl)-4-oxo-2,3-diazabut-1-ene;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sp alc/w 19°C 28% U
                                    1963HOc (59324) 595
                         K(?)=4.7
                         K(?)=9
Medium: 28% EtOH, 0.025 M, acetate buffer
*********************************
                  2-Acetylphenol CAS 118-93-4 (1888)
C8H802
              HL
2-Hydroxyacetophenone; HO.C6H4.CO.CH3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl 25°C 0.60M C K1=7.34
                                    1999MTa (59456) 596
                         B(AlH-1L)=1.97
By spectrophotometry: K1=7.30
*************************************
                            CAS 614-75-5 (4475)
C8H803
             H<sub>2</sub>L
2-Hydroxyphenylethanoic acid; HO.C6H4.CH2.COOH
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++
     gl NaCl 25°C 0.60M C
                         K1=8.06
                                    1999MTa (59714) 597
                         B(AlH-1L)=3.11
Confirmed by H-nmr
*************************
                  Mandelic Acid CAS 611-72-3 (80)
              HL
2-Phenyl-2-hydroxyethanoic acid; C6H5.CH(OH).COOH
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      EMF oth/un ? ? U K1=13.91 B2=26.90 19680Sb (59810) 598
```

K3=11.98

	K3=11.98
**************************************	CAS 520-45-6 (4478)
3-Acetyl-2-hydroxy-6-methylpyran-4-one, D	· · · ·
Metal Mtd Medium Temp Conc Cal Flags	Lg K values Reference ExptNo
Al+++ gl diox/w 35°C 50% U Medium: 50% dioxan, 0.1 M NaClO4 ************************************	K1=5.39 B2=10.32 1971MAa (60082) 599 ***********************************
C8H8O5 H2L 7-Oxy-bicyclo[2.2.1]-hept-5-ene-2,3-dicar	CAS 5629-08-3 (679) boxylic acid;
Metal Mtd Medium Temp Conc Cal Flags	Lg K values Reference ExptNo
	1988HYa (60124) 600 B(AlHL)=9.279 B(Al+L=Al(OH)L+H)=1.565 B(AlHL2)=13.585
C8H9NOS HL N-(Mercaptoacetyl)aniline (thioglycolanil	CAS 4822-44-0 (3240)
Metal Mtd Medium Temp Conc Cal Flags	Lg K values Reference ExptNo
Al+++ gl diox/w 30°C 50% U Medium: 0.1 M NaClO4 ************************************	·
C8H9NO2 L 5'-Deoxypyridoxal	CAS 1849-49-6 (5907)
Metal Mtd Medium Temp Conc Cal Flags	Lg K values Reference ExptNo
Al+++ gl KCl 25°C 1.00M C **************** C8H9NO2 HL Glycolanilide;	
Metal Mtd Medium Temp Conc Cal Flags	Lg K values Reference ExptNo
Al+++ gl diox/w 30°C 50% U Medium: 50% dioxan, 0.1 M NaClO4 ************************************	K1=8.48 B2=16.17 1973ABb (60250) 603 ***********************************
Metal Mtd Medium Temp Conc Cal Flags	Lg K values Reference ExptNo
Al+++ gl KCl 25°C 0.20M C	K1=7.84 B2=15.10 2000FEc (60281) 604 B(AlH-1L2)=10.77

```
gl mixed 30°C 50% U
                         K1=9.20 B2=17.30 1971GSc (60282) 605
                         K3=7.53
Medium: 50% acetone/H2O, 0.5 M NaClO4
**********************************
                            CAS 6310-11-8 (4576)
3-Mercaptoacetamidophenol; HS.CH2.CO.NH.C6H4.OH
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl oth/un 17°C ? U K1=6.72 B2=13.01 1973KPd (60382) 606
**********************************
                 Pyridoxal
                           CAS 65-22-5 (110)
3-Hydroxy-5-(hydroxymethyl)-2-methyl-4-pyridinecarboxaldehyde;
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KNO3 25°C 0.10M C K1=2.45 1981TMe (60424) 607
*******************************
C8H9N04
                             (4520)
Dehydroethanoic acid oxime;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl diox/w 35°C 50% U
                                   1971MAa (60488) 608
                         K(Al+HL)=5.11
                         K(A1+2HL)=9.57
Medium: 50% dioxan, 0.01 M NaClO4
**********************************
C8H10N06P
             H3L
                 Codecarboxylase CAS 41468-25-1 (2555)
Pyridoxal-5-phosphoric acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl KCl
            25°C 1.00M C M K1=10.48
                                  1985SMd (60701) 609
                         K(All+H)=4.23
ternary complexes with 2-amino-3-phosphonopropionic acid
*******************************
             H4L
                           CAS 137172-86-2 (6612)
SS-Oxydisuccinic acid; O(CH(COOH)CH2.COOH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
                         K1=8.43
Al+++ gl KCl 25°C 0.10M C
                                  1992MMa (60902) 610
                         K(All+H)=3.42
                         *K(AlL) = -5.31
                         *K(AlH-1L)=-6.43
                         K(A1+HL)=5.87
*********************************
                            CAS 84852-72-2 (6611)
C8H1009
             H4L
```

```
meso-Oxydisuccinic acid; O(CH(COOH)CH2.COOH)2
_____
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
                          K1=9.17
Al+++ gl KCl 25°C 0.10M C
                                     1992MMa (60914) 611
                           K(All+H)=3.37
                           *K(All) = -5.30
                           *K(AlH-1L)=-6.96
                           K(A1+HL)=6.57
************************************
                                (5894)
1-Hydroxy-3-oxapentane-1,2,4,5-tetracarboxylic acid;
HO.CH(COOH).CH(COOH).O.CH(COOH).CH2(COOH)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
             25°C 0.10M C
                           K1=7.63 1989MMd (60926) 612
      gl KCl
                           K(A1L+H)=2.98
                           K(AlH-1L+H)=5.05
********************************
                   Dopamine CAS 579-59-9 (251)
              H2L
2-(3',4'-Dihydroxyphenyl)ethylamine; (HO)2.C6H3.CH2.CH2.NH2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KCl
             25°C 0.20M C
                                     1989KSd (61075) 613
                           K(Al+HL)=15.63
                           K(AlHL+HL)=12.98
                           K(A1H2L2+HL)=8.95
At pH 7: K1eff=8.01, K2eff=5.36, K3eff=1.33
***********************************
                             CAS 30652-12-1 (5889)
3-Hydroxy-2-methyl-1-ethylpyridin-4-one;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl NaCl 25°C 0.15M C
                          K1=11.75 B2=22.52 1989CNa (61090) 614
                           B3=32.17
**********************************
              H2L Noradrenaline CAS 138-65-8 (253)
Norepinephrine, 3,4-Dihydroxyphenylethanolamine; (HO)2C6H3.CH(CH2.NH2).OH
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++
      gl KCl
             25°C 0.20M C
                                     1989KSd (61160) 615
                           K(Al+HL)=15.60
                           K(A1HL+HL)=12.99
                           K(A1H2L2+HL)=9.27
                           B(A1HL)=25.33
B(AlH2L2)=48.08, B(AlH3L3)=67.05. At pH 7:K1eff=8.31, K2eff=5.70, K3eff=1.98
```

```
************************************
C8H12N2O7
                            (9050)
Aspartyl-aspartic acid;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                        K1=7.54
Al+++
     gl KCl
            25°C 0.20M C
                                2003KFa (61471) 616
                        B(AlHL)=11.09
                        B(A1H2L)=14.42
                        B(AlH-1L)=3.01
**********************************
                               ********
C8H12O7P2
                            (7244)
1-Hydroxy-2-phenylethane-1,1-diphosphonic acid; HO.C(PO(OH)2)2.CH2C6H5
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl KNO3
                        K1=18.74 2002GKc (61738) 617
            25°C 0.10M C
                        B(AlH-1L)=13.26
                        B(A1H-2L)=2.57
                        B(A1H3L2)=44.74
********************************
                 Gly-Ala-Ala CAS 6491-25-4 (6783)
C8H15N3O4
             HL
Glycyl-alanyl-alanine;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.10M C K1=21.85 1983IMb (62249) 618
*******************************
                          CAS 38937-66-5 (5912)
N,N-Dihydroxyoctanediamide; HN(OH).CO.(CH2)6.CO.NH(OH)
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     Mtd Medium Temp Conc Cal Flags Lg K values
Al+++ gl NaNO3 25°C 0.10M C K1=14.59 1989EHa (62538) 619
Bis-tris
C8H19N05
                          CAS 6976-37-0 (2827)
Bis-(2-hydroxyethyl)imino-tris(hydroxymethyl)methane;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++
     sol NaCl 25°C 0.10M U T
                                 1990WPa (63054) 620
                        K = 2.63
K=2.6 at 50 C. K: Al(OH)4+H2L=Al(OH)2L+2H2O
*******************************
C9H6NOC1
                          CAS 130-16-5 (1268)
5-Chloro-8-hydroxyquinoline;
Metal Mtd Medium Temp Conc Cal Flags Lg K values
-----
Al+++ gl diox/w 25°C 60% U
                                 1973SCd (63658) 621
```

```
Medium: 60% dioxan, 0.1 M NaClO4
********************************
            H2L
                Ferron
                          CAS 547-91-1 (275)
7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI
  Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp NaCl 25°C 1.0M C K1=8.0
                                1988NMa (63776) 622
Medium pH 1.8-3.5
         sp KCl
           25°C 0.14M U I
                       K1=7.9 B2=15.10 1982GTa (63777) 623
                       K3=7.2
In Al(III)-ferron-CTMAC solution: K1=8.5, K2 < 5.8, K3 > 10.8
______
      sp oth/un ? dil U B2=12.5
                                1971BRf (63778) 624
______
Al+++ gl KNO3 28°C 0.10M U K1=6.76 B2=13.76 1971LSb (63779) 625
                       K1=7.6 B2=14.7 1961LSa (63780) 626
Al+++ gl KCl 25°C 0.10M U
                       K3 = 5.6
                       K(A1(OH)L2+H)=5.0
*******************************
                       CAS 5437-99-0 (3865)
5-Nitro-8-hydroxyquinoline;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl diox/w 25°C 60% U
                                1973SCd (63860) 627
                      B3=21.42
Medium: 60% dioxan, 0.1 M NaClO4
********************************
C9H607
                          CAS 609-98-3 (4591)
2-Hydroxybenzene-1,3,5-tricarboxylic acid; HO.C6H2(COOH)3
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ EMF NaCl 25°C 0.10M U
                                1971PPe (64004) 628
                       K(A1+H2L)=4.97
*********************************
C9H607
                          CAS 54176-76-0 (4592)
5-Hydroxybenzene-1,2,4-tricarboxylic acid; HO.C6H2(COOH)3
  ----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     EMF NaCl 25°C 0.10M U
                                1971PPe (64008) 629
                       K(A1+H2L)=4.40
********************************
                         CAS 148-24-3 (504)
                0xine
8-Hydroxyquinoline (8-quinolinol);
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     oth oth/un 25°C 0.05M U
                                1989WHa (64233) 630
                       K(Al+HL=AlL+H)=0.08
By fluorescence on silica immobilized ligand
______
Al+++ gl diox/w 25°C 50% U K1=11.17 B2=22.59 1978THc (64234) 631
                       B3=32.74
                       B(A1HL)=14.56
                       B(A1HL2)=25.73
-----
Al+++ gl diox/w 25°C 60% U
                                1973SCd (64235) 632
                      B3=33.75
Medium: 60% dioxan, 0.1 M NaClO4
-----
     sp alc/w ? 20% U
                                1971BRf (64236) 633
                      B3=33.42
********************************
                       CAS 1127-45-3 (4614)
C9H7N02
8-Hydroxyquinoline-N-oxide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl diox/w 30°C 50% U K1=9.30 B2=18.55 1970GMb (64399) 634
Medium: 50% dioxan, 0.3 M NaClO4
***********************************
C9H7NO4S
                Sulfoxine
                        CAS 84-88-8 (448)
            H2L
8-Hydroxyguinoline-5-sulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un ? ? U K1=9.69 B2=18.70 1973BIb (64524) 635
********************************
C9H7N3O2S
            H2L
                         CAS 22525-35-3 (4673)
4-(2'-Thiazolylazo)-1,2-dihydroxybenzene; C3H2NS.N:N.C6H3(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp alc/w 22°C 2% U
                                1973PPa (64665) 636
                      K(A10H+2HL)=14.62
*****************************
       HL
C9H8BrNOS
                         CAS 292149-06-5 (8797)
4-Bromo-2-(4,5-dihydro-2-thiazolyl)phenol;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp alc/w 25°C 73% C
                                2000LTb (64748) 637
                       K(A1+HL=A1L+H)=3.746
Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H2O, 0.10 M NaClO4.
```

```
By fluorescence, K(Al+HL=AlL+H)=3.590.
**************************************
                              CAS 292149-08-7 (8799)
2-(4,5-Dihydro-2-thiazolyl)-4-nitrophenol;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ sp alc/w 25°C 73% C
                                     2000LTb (64810) 638
                           K(Al+HL=AlL+H)=4.65
Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H2O, 0.10 M NaClO4.
********************************
                             CAS 292149-08-7 (8798)
C9H8N2O3S
2-(4,5-Dihydro-2-thiazolyl)-5-nitrophenol;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sp alc/w 25°C 73% C
                                     2000LTb (64812) 639
                           K(Al+HL=AlL+H)=4.08
Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H2O, 0.10 M NaClO4.
***********************
              H2L
                              CAS 148292-08-4 (7219)
Nordesferriferrithiocin; (HO)C5NH3.C3NSH3(COOH)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                           B2=22.0 1996LHa (64814) 640
Al+++ sp KCl 25°C 0.10M C
                           K(A1L2+H)=6.8
                           K(A1HL2+H)=3.7
*****************
                           CAS 15851-62-4 (3886)
C9H8N2O4S
              H2L
7-Amino-8-hydroxyquinoline-5-sulfonic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp NaNO3 25°C 0.10M C K1=8.52 1995SOa (64823) 641
*************************
              H3L Caffeic acid CAS 331-39-5 (6037)
3-(3,4-Dihydroxyphenyl)propenoic acid; (HO)2C6H3.CH:CH.COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
.....
Al+++ gl KCl 25°C 0.10M C
                                     2002AOa (64917) 642
                           B(-2,1,1)=-4.88
                           B(-3,1,1)=-9.45
                           B(-4,1,1)=-15.53
                           B(-6,1,2)=-22.24
B(-7,1,2)=-30.73, B(-9,1,3)=-39.23.
B(p,q,r): pH+qAl+rH3L=HpAlq(H3L)r
Al+++ sp none 25°C 0 U K1=15.06 1990SMd (64918) 643
```

```
************************************
C9H9NOS
                           CAS 101821-30-1 (8796)
2-(4,5-Dihydro-2-thiazolyl)phenol;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
Al+++
      sp alc/w 25°C 73% C
                                   2000LTb (65028) 644
                         K(Al+HL=AlL+H)=3.75
Medium: 73.2% v/v MeOH, 2.4% DMF, 24.4% H2O, 0.10 M NaClO4.
By fluorescence, K(Al+HL=AlL+H)=3.50.
HL Sulfathiazole
C9H9N302S2
                           CAS 72-14-0 (8357)
4-Amino-N-2-thiazolyl-benzenesulfonamide;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl alc/w 25°C 50% C K2=6.70
                                  1999GAa (65132) 645
Medium: 50% EtOH/H2O, 0.10 M NaNO3.
**********************
C9H11N02
                 Phenylalanine
                           CAS 63-91-2 (2)
2-Amino-3-phenylpropanoic acid; H2N.CH(CH2.C6H5)COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      gl KNO3
             35°C 0.10M U
                                   1980KHb (65923) 646
                        B(AlL(tripolyphosphate))=7.55
*******************************
C9H11NO3
             H2L
                 Tyrosine
                           CAS 60-18-4 (4)
2-Amino-3-(4-hydroxyphenyl)propanoic acid; HO.C6H4.CH2.CH(NH2).COOH
     -----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      gl KNO3 35°C 0.10M U
                                   1980KHb (66210) 647
                         B(AlHL(tripolyphosphate))=7.95
********************************
C9H11N04
                            CAS 95215-59-1 (8724)
1-(2'-Carboxyethyl)-2-methyl-3-hydroxy-4-pyridinone;
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
             25°C 0.10M C K1=13.04 B2=24.06 2002SGb (66302) 648
      gl KNO3
******************************
C9H11N04
                 DOPA
                            CAS 59-92-7 (5)
             H3L
2-Amino-3-(3,4-dihydroxyphenyl)propanoic acid;H2NCH(CH2C6H3(OH)2)COOH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                Al+++ gl KCl 25°C 0.20M C
                                   1989KSd (66392) 649
                         K(Al+HL)=16.03
                         K(AlHL+HL)=13.21
```

```
K(AlH2L2+HL)=9.12
```

```
At pH 7: K1eff=8.08, K2eff=5.36, K3eff=1.17
-----
           25°C 0.12M U K1=19.60 1978RMc (66393) 650
Al+++ gl NaCl
********************************
            H3L
                 PhosphoTyrosine CAS 41863-47-2 (5813)
Phosphotyrosine; 4-((OH)2P(0).0)C6H4.CH2.CH(NH2)COOH
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                      K1=10.37 1998KLb (66552) 651
Al+++ gl KCl 25°C 0.20M C
                        B(A1HL)=14.42
                        B(AlH-1L)=4.08
C9H13N02
                            (7151)
1,2-Diethyl-3-hydroxy-4-pyridinone
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KCl 25°C 0.10M C K1=13.42 B2=25.06 1994MRa (66793) 652
                       K3 = 8.48
**********************************
C9H13N03
            H2L
                 (-)Adrenaline
                          CAS 51-43-4 (252)
4-(1-Hydroxy-2-(methylamino)ethyl)-1,2-dihydroxybenzene,
Epinephrine; CH3NHCH(OH)C6H3(OH)2
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.20M C
                                 1989KSd (66857) 653
                        K(Al+HL)=15.57
                        K(AlHL+HL)=13.03
                        K(A1H2L2+HL)=9.25
                        B(A1HL) = 25.81
B(AlH2L2)=49.08, B(AlH3L3)=68.57. At pH 7:K1eff=8.22, K2eff=5.68, K3eff=1.90
*******************************
C9H13N2O9P
             H3L
                 UMP-5
                          CAS 58-97-9 (2948)
Uridine-5'-monophosphoric acid;
 ______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KCl
            25°C 0.20M C
                        K1=11.18 B2=17.49 1996AKa (66969) 654
Al+++
                     B(AlHL)=14.83
*********************************
C9H14N3O8P
            H2L
                 CMP-5
                           CAS 63-37-6 (1243)
Cytidine-5'-monophosphoric acid, Cytidilic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++
     gl KCl 25°C 0.20M C
                        K1=6.08 B2=9.7 1996AKa (67248) 655
                        B(A1HL)=9.0
```

```
B(AlH-1L)=0.39
```

```
*********************************
                            CAS 445253-97-4 (8732)
[(Dimethoxyphosphinyl)hydroxyphenylmethyl]phosphoric acid;
   Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KNO3 25°C 0.10M C
                                   2002GKc (67375) 656
                         B2=12.21
                         B(A1H-3L)=-11.91
                         B(AlHL2)=16.30
                         B(AlH-1L2)=6.06
                         B(A1H-2L2)=-0.78
*********************************
C9H18N2O4
             H2L
                            CAS 18992-11-5 (5913)
N,N-Dihydroxynonanediamide; HN(OH).CO.(CH2)7.CO.NH(OH)
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaNO3 25°C 0.10M C K1=15.55 1989EHa (67937) 657
C9H24N3O9P3
             H6L
                 NOTPH
                           CAS 83843-39-3 (224)
1,4,7-Triazacyclononane-N,N',N"-tris(methylenephosphonic acid);
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                         K1=18.6
Al+++ gl KNO3 25°C 1.00M U
                                  1990BSd (68310) 658
                         K(Al+HL)=12.6
                         K(Al+H2L)=10.3
                         K(A1+H3L)=7.6
**********************************
             H4L
                 Pyromellitic Ac CAS 89-05-4 (519)
Benzene-1,2,4,5-tetracarboxylic acid; C6H2.(COOH)4
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++
     gl NaCl 25°C 0.10M C
                         K1=4.44
                                  1998NPa (68507) 659
                         B(A1HL)=8.44
                         B(A1H-1L)=-0.25
********************************
C10H805S
                           CAS 16223-97-7 (2392)
1,2-Dihydroxynaphthalene-4-sulfonic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl NaCl 25°C 0.60M C
                                   19830Sb (69806) 660
                         B(-2,1,1)=-5.343
                         B(-3,1,1)=-11.24
                         B(-4,1,2)=-13.115
                         B(-5,1,2)=-21.15
B(p,q,r); pH+qAl+rH2L=HpAlqH2Lr. B(-6,1,3)=-24.47
```

Metal	Mtd Mediu	n Temp Conc Cal Flag	s Lg K values Reference ExptNo									
Al+++	gl NaClO	4 25°C 0.50M C	K1=15.7 B2=29.1 1985CDa (69834) 661									
Al+++	gl NaNO3	25°C 0.10M U	K1=16.48 B2=29.82 1984NHa (69835) 662 B3=39.12									
	J		K1=15.10 B2=27.88 1976LAe (69836) 663 B3=37.47									

Metal	Mtd Mediu	n Temp Conc Cal Flag	s Lg K values Reference ExptNo									
Al+++	gl NaNO3	25°C 0.10M U	K1=17.18 B2=30.10 1990HWa (69920) 664									
Al+++	gl NaClO		K1=12.9 B2=22.5 1985CDa (69921) 665									
Al+++	kin NaClO	4 25°C 0.50M C	1981BMg (69922) 666									
Method: sto	pped-flow		K(Al+H2L=AlL+2H)=-2.82									
Al+++	gl KNO3	20°C 0.10M U	K1=17.1 B2=29.80 1969HBb (69923) 667									
Al+++	gl NaClO	4 30°C 0.20M U	K1=17.16 B2=30.41 1967AMa (69924) 668									
Al+++ By glass el		? 0.10M U <1=17.40, K2=16.86	K1=17.22 1965DMa (69925) 669									
		n 25°C 0.01M U	1957JAc (69926) 670 K(Al+H2L=AlL+2H)=-3.87									

Metal	Mtd Mediu	m Temp Conc Cal Flag	s Lg K values Reference ExptNo									
Medium: EtO)H	? 100% U	K1=8? 19620Ba (70043) 671									
C10H9N0 5-Methyl-8-		HL	CAS 5541-67-3 (999)									
Metal	Mtd Mediu	m Temp Conc Cal Flag	s Lg K values Reference ExptNo									

```
1978THc (70063) 672
Al+++ gl diox/w 25°C 50% U
                        B2 = 20.52
                        B3=30.53
                        B(A1HL)=15.08
                        B(A1HL2)=25.24
**********************************
                           CAS 49608-51-7 (8280)
4,5-Dihydro-2-(2-hydroxyphenyl)-4-thiazolecarboxylic acid,
Deazademethyldesferrithiocin;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
------
     sp alc/w 25°C 75% C
                                  2000LTb (70167) 673
                        K(A1+H2L=A1L+2H)=3.38
Medium: 75% v/v MeOH/H2O, 0.10 M NaClO4.
By fluorescence, K(Al+H2L=AlL+2H)=3.26
-----
Al+++ gl KNO3 25°C 0.10M C K1=12.22 B2=22.82 1990ARa (70168) 674
**********************
C10H10N2O3S
                           CAS 76045-30-2 (7218)
Desferriferrithiocin,
2-(3-Hydroxypyridin-2-yl)-4-methyl-4,5-dihydrothiazole-4-carboxylic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
            25°C 0.10M C
                        B2=23.6
Al+++ sp KCl
                                 1996LHa (70558) 675
                        K(A1L2+H)=6.6
                        K(A1HL2+H)=3.3
______
Al+++ gl KNO3 25°C 0.10M C B2=22.2 1990ARa (70559) 676
********************
                           CAS 137528-47-3 (8725)
1-(3'-Carboxypropyl)-2-methyl-3-hydroxy-4-pyridinone;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++
     gl KNO3 25°C 0.10M C
                        K1=13.03 B2=22.97 2002SGb (71755) 677
                        B3=31.27
                        B(A1HL2)=27.34
                        B(A1H2L2)=30.26
********************************
                           CAS 18422-05-4 (842)
C10H14N507P
             H2L
                 AMP-5
Adenosine-5'-monophosphoric acid, 5-Adenylic acid;
 -----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                        K1=6.14
Al+++ gl KCl 25°C 0.20M C
                                 1996AKa (72439) 678
                     B(AlH-1L)=1.90
______
Al+++ gl KCl 25°C 0.20M U
                        K1=6.17 B2=10.35 1991KSb (72440) 679
                        B(A1H-1L)=2.02
```

```
********************************
C10H14N508P
            H3L
               GMP - 5
                        CAS 85-32-5 (2947)
Guanosine-5'-monophosphoric acid;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      gl KCl 25°C 0.20M C K1=11.66 1996AKa (72585) 680
                      B(AlHL)=14.91
TMP-5
C10H15N2O8P
                         CAS 365-07-1 (2949)
Thymidine-5'-monophosphoric acid, Thymidylic acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.20M C K1=11.92 B2=18.66 1996AKa (72698) 681
                       B(AlHL)=15.41
                       B(AlH-1L)=5.6
*********************************
                       CAS 20398-34-9 (2181)
C10H15N5O10P2
            H3L ADP
Adenosine-5'-diphosphoric acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
___________
                     K1=7.82 B2=12.16 1991KSb (72976) 682
Al+++ gl KCl 25°C 0.20M U
                       B(A1HL)=10.98
                       B(AlH-1L)=2.94
                       B(AlH-1L2)=5.01
______
                      K1=10.03 1987JVa (72977) 683
Al+++ gl NaCl 25°C 0.15M U
                       B(A1H-1L)=4.18
********************************
                EDDS
            H4L
                         CAS 52759-67-8 (1100)
1,2-Diaminoethane-N,N'-di-1,4-butanedioic acid; (CH2.NH.CH(COOH)CH2.COOH)2
_____
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
           20°C 0.10M U K1=14.8
      dis KNO3
                              1968MJa (73111) 684
Method: paper electrophoresis
**********************************
C10H16N2O8
           H4L EDTA
                         CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     cal none 25°C 0.0 C H
                                19900Ba (73567) 685
Medium: pH 8.7. DH(K1)=48.96 kJ mol-1.
Al+++ gl KNO3 25°C 0.50M C
                                1989TBa (73568) 686
                       K(All+H)=2.41
                       *K(AlL) = -5.81
```

K(AlL+F)=4.8 K(AlL+S)=10.3

							K(AlL+S)=10.3			
Al+++	gl	KNO3	25°C	0.50M	С	м М	K(AlL+H)=2.41 *K(AlL)=-5.81 K(AlL+F)=4.8 K(Al(OH)L+HS=Al		(73569)	687
Al+++	gl	NaCl	25°C	0.12M	С		K1=15.3 *K(AlL)=-5.73 *K(AlH-1L)=-9.8		(73570)	688
Al+++	gl	KNO3	35°C	0.10M	U		K1=16.95	1980KHb	(73571)	689
Al+++ Medium: K2				1.08M	U		K1=15.9	1969SVd	(73572)	690
		KNO3		0.10M	U		K1=16.5 K(Al+HL)=3.4 K(AlL+OH)=8.0	1967ABb	(73573)	691
Al+++	sp	NaCl04	25°C				K1=16.01 .OH+H)=5.87, K(Al		(73574))=10.31	692
Al+++ K1=16.84(3							T K1=16.7 K(AlL+H)=2.77 kJ mol-1, DS=489		(73575) ol-1	693
Al+++ Medium: KC		NaClO4	20°C	0.10M	U		B(A1L(OH))=25.0	4	(73576)	694
Al+++ DH(K1)=52.						Н			(73577)	695
Al+++	gl	KNO3	15°C	0.10M	U		K1=16.11 K(AlL+H)=2.0 K(AlLOH+H)=5.16		(73578)	696
Al+++							T K1=16.13 K(Al+HL)=8.4 K(AlLOH+H)=5.89 K(AlL(OH)2+H)=9	.97	(73579)	
C10H16N501 Adenosine-	3P3		H4L	ATP	***	****	CAS 56-65-		· * * * * * * * * * * * * * * * * * * *	* * * * *
Metal	Mtd	Medium	Temp	Conc (Cal	Flag	gs Lg K values	Refer	rence Exp	otNo

```
Al+++ nmr oth/un 25°C ? U
                                1994DBa (74697) 698
                       K1eff=2.4
                       K2eff=2.1
At pH 7.4
------
Al+++ gl KCl 25°C 0.20M U
                       K1=7.92 B2=12.47 1991KSb (74698) 699
                       B(AlHL)=11.30
                       B(AlH-1L)=2.46
                       B(A1H-1L2)=4.84
Al+++ gl NaCl 25°C 0.15M U
                                1987JVa (74699) 700
                      B(A1HL)=12.47
***********************************
               HEDTA
C10H18N2O7
           H3L
                         CAS 150-39-0 (392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KNO3 35°C 0.10M U K1=15.68 1980KHb (75329) 701
_____
                       K1=14.4 1967ABb (75330) 702
Al+++ ISE KNO3 25°C 0.10M U
                       K(All+H)=2.4
                       K(All+OH)=9.3
______
     sp NaCl04 25°C 0.20M U K1=13.96 1967BDb (75331) 703
By glass electrode: K(AlL+H)=2.14, K(AlLOH+H)=4.89, K(AlL(OH)2+H)=9.19
Al+++ ISE KNO3 20°C 0.10M U T H
                       K1=12.43
                                1966MCa (75332) 704
                       K(All+H)=5.08
K1=12.6(30 C),12.9(40 C). DH(K1)=37.6 kJ mol-1, DS=372 J K-1 mol-1
*******************************
4-tert-Butylcyclohexanone;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ nmr non-aq 25°C 100% U T
                                1993HMb (75586) 705
                       K(A1(OR)3+L)=2.08
Medium: C6D6. At 15 C: K=2.39; 37 C: K=1.81
**********************************
                          CAS 5578-84-7 (5914)
C10H20N2O4
N,N-Dihydroxydecanediamide; HN(OH).CO.(CH2)8.CO.NH(OH)
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaNO3 25°C 0.10M C K1=15.29 1989EHa (75798) 706
********************
                Cryptand 2,1 CAS 31249-95-3 (835)
C10H22N2O3
4,7,13-Trioxa-1,10-diazacyclopentadecane (Trioxa(2,1)cryptand);
______
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl R4N.X 25°C 0.05M U K1=9.3 1999BDb (76306) 707
Medium: Et4NClO4
*********************************
                           CAS 74385-48-1 (897)
2-(1H-Tetrazol-5-ylazo)chromotropic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp NaClO4 25°C 0.10M U
                                   1982PRa (76949) 708
                       K(Al+H2L=AlL+2H)=-4.16
*********************************
                            CAS 92-70-6 (1130)
2-Hydroxy-3-naphthoic acid (3-Hydroxy-2-naphthoic acid);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 60°C 0.02M U TIH
                                   1967GSf (77112) 709
                         K(A1+HL=A1L+H)=5.42
y=4.55(20 C),4.77(30 C),4.90(40 C),5.18(50 C). DH=-39.7 kJ mol-1, DS=217?
At 20 C: K=4.38(I=0.06), 4.26(I=0.09), 4.17(I=0.11), 4.07(I=0.16), 3.97(I=0.21).
_____
Al+++ sp oth/un 25°C 0.0 U I K1=13.38 1966MAh (77113) 710
In KCl: K(Al+HL=AlL+H)=-0.54+4.072sqrtI/(1+2.22sqrtI)-0.03I (?)
********************************
C11H804
                            CAS 7555-37-5 (4812)
3-Acetyl-4-hydroxycoumarin
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl diox/w 35°C 50% U K1=4.52 B2=8.22 1971MAa (77170) 711
Medium: 50% dioxan, 0.01 M NaClO4
*************************
        HL
C11H804
                           CAS 6724-42-1 (6183)
8-Formyl-7-hydroxy-4-methyl-2H-1-benzopyran-2-one; CHO.C9H3O(:0)(CH3)(OH)
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
Al+++ gl alc/w 35°C 70% U K1=6.83 B2=12.22 1988KRc (77196) 712
******************************
                           CAS 66695-90-7 (1996)
1-Hydroxy-4-sulfo-2-naphthoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaClO4 25°C 0.50M C K1=12.64 B2=21.89 1988LKa (77219) 713
                        B(A1H-1L2)=13.98
                        B(A1H-2L2)=5.32
K1 measured by spectrophotometry
```

```
**********************************
                             CAS 6407-91-6 (1994)
C11H806S
              H3L
1-Hydroxy-7-sulfo-2-naphthoic acid:
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaClO4 25°C 0.50M C K1=13.88 B2=23.52 1988LKa (77238) 714
                          B(AlH-1L2)=15.22
                          B(A1H-2L2)=6.54
K1 measuerd by spectrophotometry
***********************
C11H806S
              H3L
                             CAS 15509-36-1 (2658)
3-Hydroxy-7-sulfo-2-naphthoic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
.....
Al+++ sp NaClO4 25°C 0.10M U I K1=11.934
                                   1980LPf (77247) 715
In 0.1 M KCl: K1=11.316
**********************************
                             CAS 6407-90-5 (2683)
C11H807S
              H4L
1,7-Dihydroxy-4-sulfo-2-naphthoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaClO4 25°C 0.50M C K1=14.95 B2=22.14 1982LAa (77262) 716
                          B3=28.21
                          B(A1HL2)=30.80
                          B(A1HL)=20.00
Al+++ gl NaClO4 25°C 0.50M C
                          K1=14.92 B2=22.14 1982LKc (77263) 717
                          B3=28.21
                          B(A1HL) = 20.00
                          B(A1HL2)=30.80
********************************
C11H807S
              H4L
                            CAS 6470-93-5 (8345)
3,5-Dihydroxy-7-sulfo-2-naphthoic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ gl NaClO4 25°C 0.50M C K1=16.00 B2=24.09 1982LAa (77268) 718
                          B3=30.85
                          B(A1HL) = 20.65
                          B(A1HL2)=32.44
**********************************
C11H809S2
              H4L
                             CAS 67097-84-1 (1995)
1-Hydroxy-4,7-disulfo-2-naphthoic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaClO4 25°C 0.50M C K1=12.38 B2=21.55 1988LKa (77274) 719
```

B(A1H-1L2)=13.66

```
K1 measured by spectrophotometry
***********************
C11H809S2
                            CAS 67097-83-0 (1618)
3-Hydroxy-5,7-disulfo-2-naphthoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaClO4 25°C 0.50M C K1=10.81 B2=19.26 1978LAa (77293) 720
***********************
                         CAS 92609-55-3 (4827)
5-Acetyl-8-hydroxyquinoline;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl diox/w 25°C 60% U
                                    1973SCd (77327) 721
                         B3 = 26.64
Medium: 60% dioxan, 0.1 M NaClO4
*******************************
                            CAS 4321-82-7 (4829)
C11H9N04
             H2L
3-Acetyl-4-hydroxycoumarin oxime;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl diox/w 35°C 50% U
                                    1971MAa (77412) 722
                          K(A1+HL)=4.32
                          K(A1+2HL)=7.69
Medium: 50% dioxan, 0.01 M NaClO4
***********************************
       H2L PAR
C11H9N3O2
                            CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ vlt NaClO4 25°C 0.10M U
                                    1975TBc (77526) 723
                         K(A1+2HL)=23.52
Al+++ sp NaCl04 20°C 0.10M U K1=11.5 1967SNb (77527) 724
***********************************
C11H10N2O6S H3L
                             (7533)
7-(2-Carboxymethyl)amino-8-hydroxyquinoline-5-sulfonic acid; HOOCCH2NHC9H4N(OH)HSO3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl NaNO3 25°C 0.10M M
                          K1=11.23 B2=22.17 1996S0a (77682) 725
                          K(A1+HL)=10.52
                          K(A1+2HL)=21.17
***********************************
                             CAS 1147-65-5 (425)
N-(2'-Carboxyphenyl)iminodiethanoic acid; HOOC.C6H4.N(CH2.COOH)2
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     sp oth/un 25°C dil U
                                  1970DPb (77821) 726
                       K(A1+HL)=4.60 at pH 3
****************
                 Gly-Phe CAS 3321-03-7 (829)
C11H14N2O3
             HL
Glycyl-phenylalanine; H2N.CH2.CO.NH.CH(CH2.C6H5).COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     gl KNO3 35°C 0.10M U
                                  1980KHb (78811) 727
                       B(AlL(tripolyphosphate))=4.33
***********************************
C11H15N04
                           CAS 480436-59-7 (8726)
1-(4'-Carboxybutyl)-2-methyl-3-hydroxy-4-pyridinone;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.10M C K1=13.23 B2=23.41 2002SGb (79035) 728
                        B3=31.60
                        B(A1HL2)=27.76
                        B(A1H2L2)=30.69
**********************************
C11H17N2O6P
                             (5908)
3(((3-Hydroxy-2,5-dimethyl-4-pyridinyl)methylene)amino)-3-phosphonopropanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl
            25°C 1.00M C
                         K1=18.68 B2=25.10 1989MSb (79186) 729
                        K(All+H)=4.86
                        K(A1HL+H)=3.23
***********************************
                           CAS 4408-81-5 (923)
1,3-Diaminopropane-N,N,N',N'-tetraethanoic acid; ((HOOC.CH2)2N.CH2.)2.CH2
-----
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      gl KNO3 20°C 0.10M U K1=16.33 1964LAa (79420) 730
By polarography: K1=16.31
**************************************
             H4L
                 HDPTA
                           CAS 3148-72-9 (431)
1,3-Diamino-2-hydroxypropane-N,N,N',N'-tetraethanoic acid;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
___________
Al+++ gl KNO3 25°C 0.10M U
                         K1=15.2
                                  1971KRa (79540) 731
                        K(A1(OH)L+H)=6.70
                        K(A1(OH)2L+H)=8.70
                        B(A12L)=16.6
```

```
K(A12(OH)L+H)=1.0
```

```
K(A12(OH)2L+H)=3.55, K(A12(OH)3L+H)=6.20
*****************************
C12H10N2O2
                           CAS 2050-14-8 (3378)
2,2'-Dihydroxyazobenzene; HO.C6H4.N:N.C6H4.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
.....
Al+++ sp KCl 25°C 0.10M U
                                   1962KMa (80699) 732
                         K(Al+H2L=AlL+2H)=3.1
                         K(All+H2l=All2+2H)=7.4(?)
***********************************
C12H10N2O3
             H3L
                            CAS 69323-27-9 (3971)
2,2',4'-Trihydroxyazobenzene; HO.C6H4.N:N.C6H3(OH)2
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp oth/un 25°C 0.10M U
                                   1967NNa (80720) 733
                         K(Al+HL)=12.28
                         K(A10H+2HL)=29.55
********************************
C12H1006S
                           CAS 41279-65-6 (1397)
             H4L
3,3',4,4'-Tetrahydroxydiphenylsulfone; (HO)2.C6H3.S(:0)2.C6H3.(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KNO3 25°C 0.20M U K1=14.93 B2=27.58 1982HOb (80802) 734
                         K3=9.24
**********************************
C12H11N02
              L
                           CAS 49744-73-2 (1602)
3-Hydroxy-2-methyl-1-phenyl-4-pyridone; (0)(CH3)(OH).C5H2N-C6H5
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl
             25°C 0.15M C T
                         K1=11.36 B2=21.78 1991ZRa (80821) 735
                         B3=30.74
                         B3(eff)=24.74
At 37C: K1=11.86, B2=23.13, B3=32.44, B3(eff)=25.81
B3(eff) in 0.15M NaCl, pH 7.4
**********************************
C12H12N03Cl
                              (1055)
2-Chloro-4-dimethylamino-benzylidenepyruvic acid; (CH3)2N.C6H3C1.CH:CH.CO.COOH
   .....
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp NaClO4 25°C 0.50M C K1=1.979 1984MTa (80961) 736
********************
C12H12N2O6S
             H3L
                              (7532)
7-(2-Carboxyethyl)amino-8-hydroxyquinoline-5-sulfonic acid;
HOOC(CH2)2.NH.C9H4N(OH)(SO3H)
```

```
Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
 .....
      gl NaNO3 25°C 0.10M M
                         K1=11.88 B2=22.34 1996S0a (81106) 737
                         K(Al+HL)=10.91
                         K(A1+2HL)=21.20
********************************
C12H13N03
                             (1054)
4-Dimethylamino-benzylidenepyruvic acid; (CH3)2N.C6H4.CH:CH.CO.COOH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ sp NaClO4 25°C 0.50M C K1=2.563 1984MTa (81191) 738
********************************
              L Sulfadimidine CAS 57-68-1 (6167)
C12H14N4O2S
2-(4-Aminobenzolsulfamido)-4,6-dimethylpyrimidine;
_______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl alc/w 25°C 50% C K1=8.55 B2=15.00 1999GAa (81366) 739
Medium: 50% EtOH/H2O, 0.10 M NaNO3.
********************
C12H14O14
             H6L
                            CAS 111451-17-3 (5895)
3,6-Dioxaoctane-1,2,4,5,7,8-hexacarboxylic acid; (CH2(COOH).CH(COOH).0.CH(COOH)-)2
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.10M C
                                   1989MMd (81415) 740
                         K1=8.56
                         K(All+H)=5.73
                         K(A1HL+H)=3.23
                         K(AlL+Al)=1.2
                         K(AlH-1L+H)=4.95
B(A12H-1L)=10.96
*********************************
C12H17N3010
                 Asp-Asp-Asp
             H4L
                             (6445)
Aspartyl-aspartyl-aspartic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                         K1=8.45
      gl KCl
             25°C 0.20M C
                                   2003KFa (81735) 741
                         B(A1HL)=12.33
                         B(A1H2L)=15.51
                         B(AlH-1L)=3.00
*********************************
C12H18N2O5S
                            CAS 80459-15-0 (1595)
             H2L
2-Nitroso-5-(N-propyl-3-sulfopropylamino)phenol;
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     gl KNO3 25°C 0.10M C K1=7.47 B2=14.41 1988YSc (81805) 742
```

```
************************************
C12H19N02
                             (5890)
              HL
3-Hydroxy-2-methyl-1-hexylpyridin-4-one;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl NaCl 25°C 0.15M C K1=11.51 B2=22.49 1989CNa (81976) 743
Al+++
                        B3=31.71
*********************************
C12H20N2O8
                  BDTA
                            CAS 868-43-9 (1742)
DL-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOC.CH2)2N.CH(CH3).CH(CH3).N(CH2.COOH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
            20°C 0.10M U K1=18.5 1965JMb (82282) 744
      oth KNO3
Method: electrophoresis
CAS 22968-57-6 (3992)
C12H20N208
meso-2,3-Diaminobutane-N,N,N',N'-tetraethanoic acid;
(HOOC.CH2)2N.CH(CH3).CH(CH3).N(CH2.COOH)2
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
            20°C 0.10M U K1=16.5
      oth KNO3
                                   1965JMb (82383) 745
Method: electrophoresis
*************************************
              HL
                 Lactobionic acd CAS 96-82-2 (2487)
C12H22O12
4-O-Beta-D-Galactopyranosyl-D-gluconic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl NaNO3 25°C 0.10M C
                                   1995E0a (82927) 746
                        B(AlH-3L)=-11.98
********************************
C12H26N2O4
               L
                 Cryptand 2,2
                            CAS 23978-55-4 (925)
4,7,13,16-Tetraoxa-1,10-diazacyclooctadecane;
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
      gl R4N.X 25°C 0.05M U K1=9.4
                                   1999BDb (83812) 747
Medium: Et4NClO4
*********************************
                 SDS
C12H2604S
              HL
                           CAS 151-21-3 (2522)
Dodecyl sulfate; CH3(CH2)11.0S03H
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ dis NaNO3 25°C 0.10M C
                                   1994BSb (83979) 748
                         Kout(AlA+L=AlAL)=1.95
```

```
At pH 6.88. A=Ferrioxamine B.
*********************************
1,3,5-Trideoxy-1,3,5-tris(dimethylamino)-cis-inositol;
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
 -----
      gl KCl
              25°C 0.10M C
                          K1=14.23 B2=26.25 1995HKb (84067) 749
Al+++
                          B(A1HL2)=30.50
Other models also considered., e.g. K1=14.21, B2=26.23, B(AlH-1L)=8.71
-----
                          K1=16.74 B2=30.25 1995HKb (84068) 750
Al+++
       gl KCl
              25°C 0.10M C
                          B(A1HL2)=34.37
Other models also considered., e.g. K1=16.62, B2=30.22, B(AlH-1L)12.53
Al+++
      gl KCl
             25°C 0.10M U
                          K1=14.3 B2=26.4 1992KHa (84069) 751
                          B(AlH-1L)=8.9
                          B(A1HL2)=30.6
****************************
                             CAS 719-41-5 (3397)
C13H8O3
1-Hydroxyxanthone (1-Hydroxy-9-xanthenone)
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       sp alc/w 25°C 50% U K1=10.37
                                    1968GDb (84494) 752
Medium: 50% EtOH, 0.1 M NaClO4
***********************************
              H2L
                              (6878)
C13H9N03
2-Hydroxy-1-carbazole carboxylic acid;
                     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 25°C ? U
                                    1993TPa (84596) 753
                          K(Al+HL=AlL+H)=0.97
                          K(AlLOH+H)=6.2
                          K(All(OH)2+2H)=11.9
                          K(Al2LOH+3H=AlL+Al)=11.8
K(Al3L(OH)4+4H=AlL+2Al)=12.7. Method: fluorescence spectroscopy
C13H9N3OS
                  TAN
                             CAS 1147-56-4 (4030)
1-(1',3'-Thiazol-2'-ylazo)-2-hydroxynaphthalene;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      vlt NaClO4 25°C 0.10M U
                         B2=14.28
                                    1975TBc (84614) 754
********************************
                            CAS 28467-51-8 (898)
C13H9N308S3
2-(2-Thiazolylazo)chromotropic acid;
-----
      Mtd Medium Temp Conc Cal Flags Lg K values
Metal
                                    Reference ExptNo
```

```
sp NaClO4 25°C 0.10M U
                                  1982PRa (84663) 755
                        K(A1+H2L=A1L+2H)=-2.70
***********************************
C13H10N2O4
                           CAS 13245-57-3 (4983)
N-4-Nitrobenzoyl-N-phenylhydroxylamine; O2N.C6H4.CO.N(C6H5)OH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl mixed 30°C 50% U K1=8.27 B2=15.89 1971GSc (84882) 756
                        K3=6.73
Medium: 50% v/v acetone/H2O, 0.5 M NaClO4
**********************************
C13H10N2O4
                           CAS 2029-61-0 (178)
N-Phenyl-2-nitrobenzohydroxamic acid; O2N.C6H4.CO.N(C6H5).OH
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl mixed 30°C 50% U K1=6.60 B2=14.07 1971GSc (84895) 757
                        K3=6.73
Medium: 50% v/v acetone/H2O, 0.5 M NaClO4
******************************
C13H10N2O4
              HL
                            CAS 17120-18-2 (220)
N-Phenyl-3-nitrobenzohydroxamic acid; O2N.C6H4.CO.N(C6H5).OH
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl mixed 30°C 50% U K1=8.32 B2=16.07 1971GSc (84907) 758
                        K3=6.77
Medium: 50% v/v acetone/H2O, 0.5 M NaClO4
********************************
              L
                 3-Stilbazole (6869)
(3-Pyridyl)styrene; C5H4N.CH:CH.C6H5
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp non-aq 25°C 100% U M
                                  1993IWa (85006) 759
                         K(AlACl+L)=-2.54 (L is trans)
                         K(AlACl+L)=-2.62 (L is cis)
Medium: Dichloroethane. H2A: Tetraphenylporphyrin
*********************************
                 4-Stilbazole
(4-Pyridyl)styrene; C5H4N.CH:CH.C6H5
-----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                   Reference ExptNo
______
Al+++ sp non-aq 25°C 100% U M
                                  1993IWa (85007) 760
                         K(AlACl+L)-1.98 (L is trans)
                         K(AlACl+L)=-1.98 (L is cis)
Medium: Dichloroethane. H2A: Tetraphenylporphyrin
```

```
************************************
C13H11N02
              HL
                            CAS 304-88-1 (181)
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl diox/w 25°C 50% U K1=9.65 B2=19.31 1972GDb (85136) 761
                         B3=27.81
Medium: 50% dioxan, 0.25 M NaClO4
Al+++ gl mixed 30°C 50% U
                         K1=8.98 B2=17.01 1971GSc (85137) 762
                         K3=7.52
Medium: 50% v/v acetone, 0.5 M NaClO4
**********************
C13H11N05
              HL
                  Oxolinic acid CAS 14698-29-4 (2755)
1-Ethyl-6,7-dioxymethylene-quinoline-4-one-3-carboxylic acid;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp oth/un 25°C 0.05M C
                                   2000MPa (85217) 763
                         K1eff=6.39
Medium: 0.05 M chloroethanoate buffer, pH=3.0. Method: spectrofluorimetry.
*******************************
                            CAS 23117-22-8 (6287)
N-Benzoyl-4-hydroxylaminobenzene sulfonic acid; C6H5.CO.N(OH).C6H4HSO3
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaNO3 30°C 0.50M U
                         K1=6.98 B2=13.50 1976GMc (85220) 764
                         B3=18.88
************************
                  Tenoxicam
                            CAS 59804-37-4 (8393)
4-Hydroxy-2-methyl-N-2'-pyridinyl-2H-thien[2,2-e]-1,2-thiazine-3-carboxamide-1,1-di
     Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl mixed 25°C 50% C K2=5.5 2002MWa (85286) 765
Medium: 50% v/v CH3CN/H2O, 0.05 M NaNO3.
*******************************
                            CAS 83-61-4 (950)
C14H807S
             H3L
                  DASA
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl 25°C 0.10M C
                                   2002AOa (86709) 766
                         B(-2,1,1)=-2.550
                         B(-4,1,2)=-5.236
                         B(-10,2,4)=-22.77
B(p,q,r) defined for the deprotonated ligand H2L-: pH+qAl+rH2L=HpAlq(H2L)r
```

```
B(-2,1,1) determined by spectrophotometry.
-----
Al+++ gl NaCl04 25°C 0.10M M T K1=14.11 B2=26.69 1989C0a (86710) 767
______
Al+++ gl oth/un 25°C 0.00 U
                               1988RCa (86711) 768
                    B(A1H2L2)=12.88
_____
     sp NaClO4 20°C 0.10M M
                      K1=14.11 B2=26.92 1987C0a (86712) 769
                     B(Al(OH)L2)=34.1
-----
     EMF oth/un ? 0.10M U
                     K1=11.31 B2=17.37 1972GBc (86713) 770
______
     sp NaClO4 rt 0.10M U
                               1971NOc (86714) 771
                      K(A1+2H2L)=11.5
-----
Al+++ sp oth/un 25°C 0.10M U
                               1968BNa (86715) 772
                      K(Al+2HL)=12.06
*********************************
1,4-Dihydroxyanthraquinone-2-sulfonic acid, quinizarin-2-sulfonic acid;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
___________
Al+++ gl KNO3 25°C 0.05M C
                    K1=9.04
                               1993PFa (86777) 773
                       B(AlH-1L)=4.33
                       B(AlH-2L)=-1.47
                       B(A12H-1L)=8.89
                      B(A12H-3L)=0.80
********************************
                 CAS 530-28-9 (2574)
C14H10N02F3
            HL
N-(3-Trifluoromethylphenyl)-2-aminobenzoic acid; HOOC(C6H4)NH(C6H4)CF3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl mixed 22°C 90% U K1=5.95
                               1982GKb (86896) 774
Medium: 90% DMF/H20
**********************************
                         CAS 30782-99-1 (5045)
C14H1007S
            H5L
1,2,5,10-Tetrahydroxyanthracene-3-sulfonic acid (Leucoalizarin red S)
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp NaClO4 ? 0.10M U
                               1971NPb (86934) 775
                      K(A1+H3L)=7.9
                      K(A1+H4L)=6.3
**********************************
                         CAS 1105-53-9 (5084)
1,5-Bis(2-hydroxy-5-sulfophenyl)-3-cyanoformazan;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
Al+++ gl NaNO3 20°C 0.10M U K1=16.40 1971SEa (87017) 776
*********************
C14H1102NF2S
                           CAS 51679-49-3 (2928)
N-((3-Difluoromethylthio)phenyl)anthranilic acid;HOOC(C6H4).NH.(C6H4).S.CHF2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl mixed 22°C 90% U K1=6.32
                                1982GKb (87026) 777
Medium: 90% DMF/H20
*********************************
                           CAS 51679-50-4 (2929)
C14H1102NF2S
N-((4-Difluoromethylthio)phenyl)anthranilic acid;HOOC(C6H4).NH.(C6H4).S.CHF2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl mixed 22°C 90% U K1=6.12
                                  1982GKb (87031) 778
Medium: 90% DMF/H20
**********************************
C14H16N2O5S
                           CAS 390426-77-4 (8803)
1-n-Butyl-5-sulfo-8-hydroxyquinoline-7-carboxamide;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp NaCl04 25°C 0.01M C K1=8.4
                               B2=16.20 2001LAa (87905) 779
Medium: 0.01 M HClO4. Method: spectrophotometric titration.
**********************************
                 Pipemidic acid CAS 51940-44-4 (2535)
C14H17N503
8-Ethyl-5,8-dihydro-5-oxo-2-(1-piperazinyl)pyrido[2,3-d]pyrimidine-6-carboxylic
acid:
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp oth/un 25°C 0.05M U
                                  2000MPa (88055) 780
                        K1eff=6.07
Medium: 0.05 M chloroethanoate, pH=5.5. Method: spectrofluorimetry.
For Cinoxacin, K1eff=5.601-Ethyl-3-carboxy-6,7-methylenedioxy-4-cinnolone
******************************
             H4L
                 CDTA
                           CAS 482-54-2 (200)
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 35°C 0.10M U
                         K1=19.18 1980KHb (88572) 781
_____
                               1967ABb (88573) 782
Al+++ ISE KNO3 25°C 0.10M U
                         K1=18.9
                        K(A1+HL)=3.4
                        K(All+OH)=6.3
-----
Al+++ sp NaCl04 25°C 0.20M U K1=18.50 1967BAc (88574) 783
```

```
By glass electrode: K(AlL+H)=2.29, K(AlL(OH)+H)=7.82
     -----
                           K1=18.63 1966MCa (88575) 784
       ISE KNO3 20°C 0.10M U T H
                           K(All+H)=2.59
K1=18.8(30 C),19.15(40 C); DH(K1)=-46.0 kJ mol-1, DS=510 J K-1 mol-1
Al+++
       dis NaClO4 20°C 0.10M U
                                     1963STc (88576) 785
                           B(All(OH)) = 26.61
Medium: KClO4
Al+++ vlt KNO3 20°C 0.10M U
                           K1=17.63 1954SGa (88577) 786
                           K(AllOH+H)=7.58
                           K(All+H)=3.93
C14H23N308
              H4L
                   NODASA
                             CAS 210217-93-9 (8716)
1,4,7-Triazacyclononane-1-succinic acid-4,7-diethanoic acid;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ nmr oth/un 20°C 0.12M C K1=18.50
                                    2002AMb (89008) 787
Medium: 0.02 M Al(NO3)3, pH 2.0. Method: 27Al nmr.
**********************************
C14H23N3010
              H5L
                   DTPA
                              CAS 67-43-6 (238)
Diethylenetriamine-pentaethanoic acid; HOOC.CH2.N(CH2.CH2.N(CH2.COOH)2)2
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      ISE KNO3 25°C 0.10M C
                                     1996YHa (89144) 788
                           K(All+H)=5.18
                           K(A1HL+F)=5.3
                           K(All+F)=2.9
Method: Fluoride ISE.
______
Al+++ gl KNO3 35°C 0.10M U
                           K1=20.66 1980KHb (89145) 789
<del>-</del>
Al+++ ISE KNO3 25°C 0.10M U
                           K1=18.7
                                     1967ABb (89146) 790
                           K(A1+HL)=4.3
                           K(All+OH)=6.6
                        _____
                           K1=18.4 1966MCa (89147) 791
      ISE KNO3 20°C 0.10M U T H
                           K(All+H)=4.63
K1=18.51(25 C),18.62(30 C),18.80(40 C). At 25 C: DH(K1)=33 kJ mol-1, DS=472
C14H24N2O10
                   EGTA
                             CAS 67-42-5 (349)
Ethyleneglycol-0,0'-bis(2-aminoethyl ether)-N,N,N',N'-tetraethanoic acid; H4L
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
                           K1=13.90
Al+++ gl NaClO4 25°C 0.20M U
                                     1967BDb (89837) 792
                           K(All+H)=3.97
```

K(AlLOH+H)=5.20 K(AlL(OH)2+H)=8.42

```
K1 by spectrophotometry
***********************************
C14H25N307
              H3L
                               (5397)
1-0xa-4,7,10-triazacyclododecane-4,7,10-triethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
       gl KCl 25°C 0.10M C
                          K1=12.5
                                     1993DSa (90079) 793
                          K(All+H)=3.28
**********************************
               L
                   Cryptand 2,1,1 CAS 31250-06-3 (836)
C14H28N2O4
1,10-Diaza-4,7,13,18-tetraoxabicyclo[8,5,5]eicosane (2,1,1);
       Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       gl R4N.X 25°C 0.05M U
                          K1=12.9
                                     1999BDb (90345) 794
Medium: Et4NClO4
***********************************
                              CAS 16195-35-0 (27)
5-(4-Chlorophenylazo)-8-hydroxyquinoline; Cl.C6H4.N:N.C9H5N.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sp oth/un 25°C 0.10M U
                                     1978KIa (90946) 795
                          B3=11.91
************************************
C15H10N3O5C1S
7-[(2-Hydroxy-5-chlorophenyl)azo]-8-hydroxyquinoline-5-sulfonic
acid;C6H3C1(OH)N=NC9H4N(OH)(SO3H)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
       kin KNO3 25°C 0.10M C K1=23.3
                                     1996PKa (90952) 796
********************************
              H5L Quercetin
                             CAS 117-39-5 (5101)
3,5,7-Trihydroxy-2-(3',4'-dihydroxyphenyl)-1-benzopyran-4-one;
-----
      Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
-----
     sp non-aq 25°C 100M C
                                     2001ADb (91020) 797
                           K1eff=-2.34
Medium: MeOH, 0.2 M acetate buffer, pH 5.0. K1eff: Al+HnL=AlL
******************************
C15H10010S
              H5L
                              CAS 141896-20-0 (8182)
2-(3,4-Dihydroxyphenyl)-3.5-dihydroxy-7-(sulphooxy)-4H-1-benzopyran-4-one,
Ouercetin-7-sulfonic;
Metal
       Mtd Medium Temp Conc Cal Flags Lg K values
                                      Reference ExptNo
```

```
sp non-aq 25°C 100M C
                                 2001ADb (91028) 798
                       K1eff=-1.73
Medium: MeOH, 0.2 M acetate buffer, pH 5.0. K1eff: Al+HnL=AlL
*******************************
                Quercetin S F CAS 25001-18-7 (1520)
            H5L
3,5,7,3',4'-Pentahydroxy-5'-sulfoflavone; (H0)3(0)C9H2O.C6H2(SO3H)(OH)2
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp NaClO4 20°C 0.10M U K1=5.11
                                 1989KOa (91031) 799
_____
Al+++ sp NaClO4 20°C 0.10M U
                                 1976KTb (91032) 800
                       B(A1H4L)=7.56
                       B(A12H2L)=20.9
**********************************
C15H11N02
                           (430)
2-(2'-Hydroxyphenyl)-8-hydroxyquinoline; HO.C6H4.C9H5N.OH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
 Al+++ gl diox/w 25°C 50% U K1=19.8 B2=34.70 1974CCb (91056) 801
*********************
                PAN
                          CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH
-----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                  Reference ExptNo
______
    vlt NaClO4 25°C 0.10M U B2=21.90 1975TBc (91206) 802
-----
                        K1=12.86 1973TBa (91207) 803
      vlt alc/w 25°C 50% U
Medium: 50% EtOH, 0.06 M (H/Na)ClO4
************************************
                          CAS 4312-09-8 (989)
5-Phenylazo-8-hydroxyguinoline; C6H5.N:N.C9H5N.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 25°C 0.10M U K1=4.37
                                 1978KIa (91266) 804
***********************************
                            (5130)
C15H11N3O4S
7-Phenylazo-8-hydroxyquinoline-5-sulfonic acid;
  -----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp NaNO3 25°C 0.10M C K1=7.53 1995SOa (91334) 805
********************
C15H11N3O5S
                         CAS 111248-75-0 (8411)
5-(2'-Hydroxy-5'-phenylazo)-8-quinolinol;
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ sp oth/un RT dil C
                                   1985IBa (91340) 806
                         K1eff=6.73
                         B2eff=11.20
                         B3eff=15.64
Medium: Britton and Robinson buffer, pH 6.6
*******************************
                    CAS 654637-45-3 (9237)
C15H1103Cl
7,8-Dihydroxyflavylium chloride;
______
     Mtd Medium Temp Conc Cal Flags Lg K values
______
Al+++ sp none 25°C 0.0 C K1=9.11 2003MMa (91401) 807
***********************************
             H4L Luteolinidin CAS 1154-78-5 (9239)
2-(3,4-Dihydroxyphenyl)-5,7-dihydroxy-1-benzopyrilium chloride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp none 25°C 0.0 C K1=6.81
                                2003MMa (91403) 808
***********************************
C15H12O2
              HL
                            CAS 1214-47-7 (951)
3-Phenyl-1-(2'-hydroxyphenyl)-2-propen-1-one, 2'-hydroxychalkone;
C6H5.CH:CH.CO.C6H4.OH
______
     Mtd Medium Temp Conc Cal Flags Lg K values
                                    Reference ExptNo
______
Al+++ gl diox/w 30°C 60% U K1=11.35 B2=21.05 1975KKc (91576) 809
*************************
                 Catechin CAS 154-23-4 (2737)
             H4L
3,3',4',5,7-Pentahydroxyflavone; (H0)3(0)C9H60.C6H3(OH)2
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl
            25°C 0.10M C
                                   2002IIb (91815) 810
                         K(A1+H4L=A1H2L+2H)=-5.57
                         K(A1H2L=A1H2L(OH)+H)=-5.76
                         K(A1H2L+H4L=A1(H2L)2+2H)=-8.3
For (-)-epicatechin, K(A1+H4L=A1H2L+2H)=-5.75, K(A1H2L=A1H2L(OH)+H)=-6.0,
K(A1H2L+H4L=A1(H2L)2+2H)=-8.2.
Al+++ gl NaClO4 25°C 0.50M C
                         K1=18.3
                                   1985CDa (91816) 811
                         K(Al+HL)=12.4
Al+++ gl KCl 25°C 0.10M C
                                   1985KPa (91817) 812
                         K(A1+H2L)=17.10
                         K(A1H2L+H2L)=13.89
                         K(A1(H2L)2+H2L)=9.93
                         K(A1(OH)H2L+H)=5.98
```

```
K(A1(H2L)2=A1(OH)(H2L)2+H)=-8.22. For the epicatechin dimer H8L constants
for AlH6L, Al(H6L)2 and Al(OH)H6L given in Austral. J Chem., (1985) 38, 879
**************************
C15H1407
                           CAS 970-73-0 (1796)
Epigallocatechin;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.10M C
                                 2002IIb (91819) 813
                        K(A1+H5L=A1H3L+2H)=-5.75
                        K(A1H3L=A1H3L(OH)+H)=-5.23
                        K(A1H3L+H5L=A1(H3L)2+2H)=-7.16
****************************
C15H15NO2 HL
                          CAS 61-68-7 (2927)
N-(2,3-Dimethylphenyl)anthranilic acid; HOOC(C6H4).NH.(C6H3)(CH3)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl mixed 22°C 90% U K1=6.56 1982GKb (91829) 814
Medium: 90% DMF/H20
**********************************
C16H11N2O5ClS HL
                            (7535)
2-[(2-Hydroxy-5-chlorophenyl)azo]-1-hydroxynaphthalene-4-sulfonic
acid;HO.C6H3ClN=NC10H5N(OH)HSO3
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ kin KNO3 25°C 0.10M C K1=15.26 1996PKa (92769) 815
***********************************
C16H11N3O6S H3L
                            (1047)
7-(4-Carboxyphenylazo)-8-hydroxyquinoline-5-sulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp NaCl 25°C 0.10M U K1=9.86 B2=16.84 19840Fa (92850) 816
******************************
C16H11N3O7S
            H3L
                          CAS 116946-37-3 (1598)
7-Hydroxy-((4-carboxyphenyl)azo)-8-hydroxy-5-quinolinesulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaCl 22°C 0.10M C K1=7.49 1988BEa (92852) 817
C16H11N3O10S2
            H4L Chromotrope 2B CAS 548-80-1 (896)
2-((4-Nitrophenyl)azo)chromotropic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp NaClO4 25°C 0.10M U
                                 1982PRa (92861) 818
                        K(A1+H2L=A1L+2H)=-4.70
```

```
************************************
C16H12N2O5S
           H3L SolochromeVio R CAS 94205-83-1 (4093)
1-(2'-Hydroxy-5'-sulfophenylazo)-2-naphthol;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 25°C 0.0 U K1=18.4 B2=31.6 1962CRa (93021) 819
*********************************
C16H12N2O8S2
           H4L Chromotrope 2R CAS 4197-07-3 (2604)
2-(Benzeneazo)-chromotropic acid, Acid Red 29
______
     Mtd Medium Temp Conc Cal Flags Lg K values
                                Reference ExptNo
Al+++ gl NaClO4 25°C 0.10M U K1=16.70 B2=25.70 1975MPa (93059) 820
Al+++ gl KCl 20°C 0.10M U K1=18.41 1964PCa (93060) 821
*************************
C16H12N2O11S3
                        CAS 548-81-2 (5180)
2-(4'-Sulfophenylazo)chromotropic acid,
2-(4-sulfophenylazo)-1,8-dihydroxyaphthalene-3,6-diHSO3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaClO4 25°C 0.10M U K1=10.00 B2=13.80 1975MPa (93091) 822
Thorin I CAS 3688-92-4 (2609)
C16H13N2O10AsS2
           H5L
1-((2-Arsonophenyl)azo)-2-hydroxy-3,6-naphthalyldisulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
     sp oth/un 25°C ? U
                               1968GSe (93184) 823
                     K(?)=10.5
************************************
8-Hydroxy-7-((2-hydroxy-5-carboxyphenyl)azo)-5-quinoline sulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp NaCl 22°C 0.10M C K1=14.15 1988BEa (93287) 824
**********************************
                         CAS 125653-94-3 (9238)
C16H13O3Cl
7,8-Dihydroxy-4-methylflavylium chloride;
_____
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ sp none 25°C 0.0 C K1=8.30 2003MMa (93393) 825
C16H13O5Cl
3',4',7-Trihydroxy-3-methoxyflavylium chloride; (HO)2C6H3.C9H4O(OH)(OCH3)Cl
______
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                      sp none 25°C 0.00 U
                                       1997EFa (93396) 826
                           K(A1+H2L=A1L+2H)=-4.04
Al+++ sp NaCl 25°C 0.50M U
                                       1994DEa (93397) 827
                            K(A1+HL=A1L+H)=0.36
                            K(A1+H2L=A1L+2H)=-4.04
For the hemiacetal, K(Al+H2L=AlL+2H)=-6.39.
*********************************
                               CAS 38214-71-0 (8453)
3-(2-Hydroxy-5-methylphenyl)-5-phenylpyrazole;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl diox/w 27°C 70% C K1=10.75 B2=21.05 1994SNa (93418) 828
                           K3 = 9.50
Medium: 70% v/v dioxane/H2O, 0.10 M NaClO4.
**********************************
C16H16N2O2
                                (5159)
4-Hydroxyphenacylidene-4-dimethylaminoaniline; HOC6H4COCH:NC6H4N(CH3)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp alc/w 28°C 100% U
                                       1970GGa (93660) 829
                            K(A1+2HL)=8.59
Medium: MeOH
*******************************
                   Penicillin V CAS 87-08-1 (943)
Phenoxymethylpenicillinic acid, 4-Thia-1-azabicyclo[3.2.0]heptane-2-carboxylic
acid;
        ._____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.10M M T H K1=8.20 B2=13.20 1983SBc (93814) 830
Also data for 35 C. DH(B2)=1.1 kJ mol-1, DS(B2)=253 J K-1 mol-1.
**********************************
C16H18N3O3F HL Norfloxacin CAS 70458-96-7 (7141)
1-Ethyl-6-fluoro-1,4-dihydro-4-oxo-7[1-piperazinyl]-3-quinoline carboxylic acid;
C16H18N3O3F
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl oth/un 25°C 0.10M C
                           K1=8.83
                                      1995DJa (93827) 831
                            B(A1HL)=14.60
                            K(A1+L=A1L(OH)3+3H)=-14.85
Medium: LiCl
*********************************
                   Chlorogenic acd CAS 327-97-9 (2844)
3-(3',4'-Dihydroxycinnamoyl)-1,3,4,5-tetrahydroxycyclohexane carboxylic acid;
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl KCl
            25°C 0.10M C
                                 2002AOa (93898) 832
                        B(-2,1,1)=-3.91
                        B(-3,1,1)=-8.17
                        B(-4,1,1)=-13.79
                        B(-6,1,2)=-19.28
B(-7,1,2)=-27.65, B(-9,1,3)=-34.01.
B(p,q,r): pH+qAl+rH3L=HpAlq(H3L)r
****************************
                          CAS 61696-54-6 (6104)
1,4,7,10,13,16-Hexaoxacyclooctadeca-2,3,11,12-tetracarboxylic acid;
 .....
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl R4N.X 25°C 0.05M C M K1=9.0 1998TSb (94490) 833
                        B(A1HL)=13.1
                        B(A1H2L)=17.3
Medium: 0.05 M Et4NCl04. Also ternary complexes, MAlH-nL, where M=Na, K,
Cs, Ca, Sr, Ba.
DOTA
C16H28N4O8
            H4L
                          CAS 60239-18-1 (1017)
1,4,7,10-Tetraazacyclododecane-1,4,7,10-tetraethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaNO3 25°C 0.20M C K1=17.0 1995KKa (94876) 834
*******************************
                Cryptand 2,2,1 CAS 31364-42-8 (837)
1,10-Diaza-4,7,13,16,21-pentaoxabicyclo[8,8,5]tricosane (2,2,1);
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      gl R4N.X 25°C 0.05M U K1=11.3 1999BDb (95175) 835
Medium: Et4NClO4
*********************************
        H4L
                            (1597)
4-Hydroxy-((2-hydroxy-5-carboxyphenyl)azo)-naphthalenesulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ gl NaCl 22°C 0.10M C K1=15.27 1988BEa (95936) 836
*******************************
C17H16N2O2
                          CAS 65840-98-4 (8454)
3-(2-Hydroxy-5-methyphenyl)-5-(4-methoxyphenyl)pyrazole;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
Al+++ gl diox/w 27°C 70% C
                       K1=10.55 B2=20.80 1994SNa (96027) 837
                        K3=7.45
```

```
Medium: 70% v/v dioxane/H2O, 0.10 M NaClO4.
************************
                  Ciprofloxacin CAS 189257-90-7 (7142)
1-Cyclopropyl-6-fluoro-1,4-dihydro-4-oxo-7[1-piperazinyl]-3-quinoline carboxylic
        ._____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.20M C
                                    1996TBc (96223) 838
                          B(A1H2L2)=29.33
                          B(A1H2L3)=35.8
                          B(A1H3L3)=43.26
*******************************
C18H20N2O6
                            CAS 10328-28-6 (3501)
Ethylenedinitrilo-N,N'-bis(2'-hydroxyphenyl)-N,N'-diethanoic acid;
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KCl 25°C 0.10M C K1=25.78
                                  1993MMa (97393) 839
                         K(A1L+H)=3.73
*********************************
             H4L EHPG
C18H20N2O6
                            CAS 10328-28-6 (429)
N,N'-Ethylene-bis-(2-(2'-hydroxyphenyl))glycine; (HOOCCH(C6H4OH)NHCH2.)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp oth/un 25°C 0.10M C
                                    2003YFc (97420) 840
                         K1eff=9.38
Method: UV difference spectrophotometry. Medium: 0.10 M HEPES, pH 7.4.
-----
Al+++ gl NaCl 25°C 0.12M C K1=24.48 1981RMb (97421) 841
*******************************
              HL Ofloxacin CAS 82419-36-1 (7789)
a-Fluoro-3-methyl-10-(4-methyl-1-piperazinyl)-7-oxo-2,3-dihydro-7H-pyrido-1,4-benzo
xazine-6-COOH:
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                          K1=10.37
Al+++ gl KCl 25°C 0.10M C I
                                    2001DJa (97453) 842
                          B(A1HL)=16.40
Medium: 0.10 M LiCl, 0.001 M Triton. In 0.10 M LiCl, 0.005 M CTAB,
K1=11.56, B(A12H-2L)=3.6.
*******************************
C18H22N4O4
                            CAS 2444-14-6 (3502)
             H2L
N,N'-Bis(2-pyridylmethyl)diaminoethane-N,N'-diethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaCl 25°C 0.16M C
                                   1997CRa (97539) 843
                          K(Al+L=AlL(OH)+H)=6.37
```

```
K(All(OH)+H=All)=4.48
***************************
                            CAS 53431-86-0 (5266)
Ethylenebis(imino(2-hydroxyphenyl)methylene(methyl)phosphinic acid);
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                                1970DMc (97673) 844
     EMF oth/un ? ? U
                         K1 = 20.0
                        K(A1+HL)=15.36
**********************************
             H6L
                           CAS 869-52-3 (694)
                 TTHA
Triethylenetetraaminehexaethanoic acid;((HOOC.CH2)2N.CH2.CH2.N(CH2.COOH).CH2)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl R4N.X 25°C 0.10M C
                         K1=20.23 1998ACc (98007) 845
                        K(All+H)=5.97
                        K(All+Al)=9.55
                        K(A12L(OH)+H)=4.68
                        K(A12L(OH)2+2H)=9.87
Medium: N(CH3)4NO3.
-----
Al+++ ISE KNO3 25°C 0.10M C M
                                  1996YHa (98008) 846
                        K(All+H)=5.94
Method: Fluoride ISE.
______
Al+++ gl KNO3 35°C 0.10M U
                        K1=18.74 1980KHb (98009) 847
Al+++ gl KNO3 25°C 0.10M U
                        K1=19.7 1970HAa (98010) 848
                        K(All+H)=5.85
                        K(A12L+20H)=15.9
By ion-selective electrode (Hg): B(Al2L)=28.6. By redox: B(Al2L)=28.9
**************************
                 TETA
                           CAS 60239-22-7 (1019)
             H4L
1,4,8,11-Tetraazacyclotetradecane-1,4,8,11-tetraethanoic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaNO3 25°C 0.20M C K1=16.3 1995KKa (98188) 849
*******************************
                 Cryptand 2,2,2 CAS 23978-09-8 (514)
C18H36N2O6
1,10-Diaza-4,7,13,16,21,24-hexaoxabicyclo[8.8.8]hexacosane;
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl R4N.X 25°C 0.05M U K1=10.6 1999BDb (98513) 850
Al+++
Medium: Et4NClO4
**********************************
                 Pyrogallol red CAS 85531-30-2 (638)
Pyrogallolsulfonephthalein;
```

```
Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                  ? U
      sp oth/un 25°C
                                    1968GSa (98997) 851
                          K(?)=5.0
C19H12O9Br2S
              H6L
                  Bromo Pyrog.Red CAS 16574-43-9 (706)
5',5"-Dibromopyrogallolsulfonephthalein;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
·
------
      sp oth/un 20°C 0.05M U
                                    1970BLb (99010) 852
                          K(A1+H4L)=5.03
C19H13N3O7S2
              H3L
                  Naphthylazoxine CAS 56932-43-5 (276)
8-Hydroxy-7-(6'-sulfo-2'-naphthylazo)-quinoline-5-sulfonic acid;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp NaCl 25°C 0.10M U K1=7.88
                                    1982H0a (99056) 853
**********************************
C19H13N3O11S3
                             CAS 37469-13-9 (1883)
8-Hydroxy-7((8'-hydroxy-3',6'-disulfo-1'-naphthyl)azo)quinole-5-sulfonic acid;
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
              25°C 0.10M U K1=16.03 1983IOa (99058) 854
Al+++ sp NaCl
*********************************
                  Pyrocatechol Vi CAS 369596-29-2 (709)
              H4L
C19H1407S
Pyrocatechol Violet,
3-[3,4-Dihydroxyphenyl-3-hydroxy-4-oxo-2,5-cyclohexadien-1-ylidenemethyl-b.;
------
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      gl KCl
              25°C 0.10M C
                                    1995SSa (99101) 855
                          B(-1,1,1)=-0.23
                          B(-2,1,2)=-1.02
                          B(-3,1,3)=-2.57
                          B(-4,1,2)=-10.21
B(p,q,r): pH+qAl+r(H3L-). B(-5,1,3)=-13.03, B(-6,1,3)=-21.10, B(-7,1,3)=
-30.46, B(-8,1,3)=-40.75, B(-9,1,3)=-52.0, B(-6,3,3)=-5.07, B(-16,6,6)=-23.2
Al+++
              25°C 0.50M U
       sp KCl
                                    1974CMc (99102) 856
                          K(Al+H3L=AlH2L+H)=0.12
                          K(A1H2L+H3L=A1H3L2+2H)=-6.10
                          K(AlH3L2+H2L=AlH4L3+H)=-1.87
                          K(2A1+H3L=A12HL+2H)=-0.23
K(A1+H2L)=7.9, K(A1+H2L+HL)=19.4, K(A1+H2L+2HL)=27.3, K(2A1+HL)=17.3
Ligand: Pyrocatechol sulfophthalein
```

```
gl KCl 25°C 0.20M U K1=25.12 B2=47.39 1970G0a (99103) 857
Al+++
                         K3 = 20.74
**********************************
(4-Phenyl-3-pyridyl)styrene;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp non-aq 25°C 100% U M
                                   1993IWa (99120) 858
                         K(AlACl+L)=-2.02 (L is trans)
                         K(AlACl+L)=-2.52 (L is cis)
Medium:Dichloroethane. H2A:Tetraphenylporphyrin
*****************************
            H4L Alizarin Comp. CAS 3952-78-1 (671)
C19H15N08
(3,4-Dihydroxy-2-anthraquinonyl-methyl)iminodiethanoic acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp diox/w 20% U
                                   1973INa (99129) 859
                         K(A1+HL)=14.3
                         B(A12L)=25.3
Medium: 20% dioxan, 0.1 M
**********************************
            H3L Folic acid CAS 75708-92-8 (194)
Pteroylglutamic acid;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Al+++ gl KNO3 30°C 0.10M U I K1=4.65 B2=8.85 1970NDa (99283) 860
                         K3=4.10
I=0: K1=5.80, K2=4.70, K3=4.65. I=0.01: K1=5.25, K2=4.55, K3=4.50.
I=0.05: K1=4.80, K2=4.28, K3=4.15
************************************
                 TETMAHA
                             (7468)
1,4,8,11-Tetraazacyclotetradecane-N,N',N"-tris(N-methylacetohydroxamic acid);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.10M C
                         K1 = 21.02
                                  1999GGa (99501) 861
                         B(A1HL)=30.72
                         B(A1H2L)=35.60
                         B(A1H3L)=40.71
                         B(A1H4L)=44.17
*********************************
C20H11N09S2
                           CAS 65501-73-7 (8982)
             H3L
6-Hydroxy-5-dibenzo[a,j]phenoxazone-8,11-disulfonic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp KCl RT 1.0M C
                                   1980NLa (99533) 862
```

```
K(Al+HL=AlL+H)=-0.39
```

cetylpyri	5, K1eff=4.77. Data for solutions with Septone: inium surfactants ************************************	
C20H13N3O7	H3L Eriochrome Bl T CAS 1787-6 xy-2-naphthylazo)-6-nitro-2-naphthol-4-sulfoni	1-7 (997)
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
Al+++ Medium: Na	sp oth/un 20°C 0.10M U K(Al+2HL)=12.51 2SO4	1980PKa (99559) 863
**************************************	gl NaClO4 25°C 0.10M U K1=9.56 B2=13 ************************************	**************************************
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
**************************************	gl NaClO4 25°C 0.10M U K1=11.58 B2=20 ***********************************	******* 5-4 (3508)
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	
	sp oth/un 25°C 0.10M U K(Al+2HL)=41.97 K(Al0H+2HL)=40.0	1967NNc (99686) 866 62
C20H24N2O6		9-6 (2208)
Metal	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
	sp oth/un 25°C 0.10M C K1eff=8.88 difference spectrophotometry. Medium: 0.10 M N	2003YFc (99987) 867
Al+++ ******** C20H35N503	gl NaCl 25°C 0.12M C K1=24.78 ***********************************	1981RMb (99988) 868 ******
	Mtd Medium Temp Conc Cal Flags Lg K values	Reference ExptNo
	gl NaNO3 25°C 0.20M C K1=16.1 ***********************************	

```
C21H21012
            H5L
                         CAS 50986-17-9 (7770)
3-O-beta-D-Glucopyranosyldelphinidin ion;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp none 25°C 0.00 U
                               1997EFa (101191) 870
                      K(A1+H2L=A1L+2H)=-3.53
*********************************
            L G-Rubrofusarin CAS 63174-98-1 (7067)
2-Methyl-5,6-dihydroxy-6-O-B-D-galactosyl-8-methoxy-naphtho-pyrone;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp NaCl04 25°C 1.00M C K1=8.91 1995PDa (101213) 871
***********************************
C22H1409
                        CAS 4431-00-9 (3513)
Aurintricarboxylic acid;
_____
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp oth/un ? ? U K1=16.6
                            1972PKa (101491) 872
********************************
C22H18N4O14As2S2
            H8L
               Arsenazo III
                        CAS 1668-00-4 (1148)
2,7-Bis(2'-arsonophenylazo)chromotropic acid;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 25°C 0.10M U
                               1975MIa (101609) 873
                      K(A1+H4L=A1H2L+2H)=-1.17
                      K(A10H+H5L=A1(OH)H3L+2H)=0.47
*********************************
C22H18011
                         CAS 989-51-5 (2270)
Epigallocatechin gallate;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl KCl 25°C 0.10M C
                               2002IIb (101680) 874
                      K(A1+H8L=A1H6L+2H)=-4.47
                      K(AlH6L=AlH5L(OH)+H)=-4.74
********************************
C22H23N2O8C1
            H2L
               Aureomycin
                        CAS 56235-18-8 (3515)
Chlorotetracycline;
  Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl oth/un 20°C 0.01M U K1=7.2 1956ARd (101758) 875
*******************
C22H24N2O8
           H2L
               Tetracycline CAS 60-54-8 (2201)
Tetracycline;
______
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl NaNO3 25°C 0.10M C K1=12.5 1992GAa (101807) 876
_____
Al+++ gl oth/un 20°C 0.01M U
                        K1=7.4 B2=13.80 1956ARd (101808) 877
                        K3 = 5.4
********************************
C22H24N2O9
             H2L
                 Oxotetracycline CAS 79-57-2 (2202)
Oxytetracycline, 5-Hydroxy-tetracycline;
______
      Mtd Medium Temp Conc Cal Flags Lg K values
_____
Al+++ gl oth/un 20°C 0.01M U K1=7.0 1956ARd (101881) 878
********************************
C22H31N506
                            CAS 813432-03-0 (9200)
             H2L
Imino-bis(acetyl(1-(3'-aminopropyl)-3-hydroxy-2-methyl-4-pyridinone);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ gl KNO3 25°C 0.10M C
                       K1=20.35
                                   2004SGc (102190) 879
                         B(A1HL) = 25.37
                         B(A1H2L)=27.71
                         B(A1H3L)=30.44
                         B(A12L3)=60.18
B(Al2HL3)=66.21, B(Al2H2L3)=71.72, B(Al2H3L3)=76.46, B(Al2H-2L2)=32.05
********************************
             H4L Chrome azurol S CAS 1667-99-8 (711)
C23H1609Cl2S
Chromazurol S;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Al+++ sp KCl 25°C 0.10M C I
                                   1995HPa (102539) 880
                         K(AlL+HL=AlL+H)=2.01
                         K(3A1+2HL=A13L2+2H)=12.29
                         K(2A1+2HL=A12HL2+H)=12.92
At I=0.60M (KCl): K(Al+HL=AlL+H)=1.80; K(3Al+2HL=Al3L2+2H)=11.99
K(4A1+2HL=A14H-2L2+4H)=7.45
                                1969TKb (102540) 881
Al+++ sp oth/un ? ? U
                         B2=12.85
                        K(A1+2HL)=6.82
 Al+++ sp KCl 30°C 0.20M U K1=4.32 1963SDh (102541) 882
******************************
                        (7200)
             H2L
C23H1809S
                 ECR
3''-Sulfo-3,3'-dimethyl-4-hydroxyfuchson-5,5'-dicarboxylic acid
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp KCl 25°C 0.10M C
                                   1996HKa (102619) 883
                         B(Al-HL)=1.75
```

```
B(Al3-H2L2)=13.44
B(Al4-H3L5)=29.07
B(Al4-H4L5)=25.30
```

B(Al1-HL) determined by spectrophotometry, the other values by potentiometry B(Al4-H5L5)=20.67 species formed at higher alkali addition rate. C23H1809S H4L Eriochrome cyan CAS 3564-18-9 (433) 4'-Hydroxy-3,3'-dimethyl-2''-sulfofuchsone-5,5'-dicarboxylic acid; ______ Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo sp oth/un 25°C 0.10M U K1=13.66 1975EPa (102625) 884 B(A1HL)=18.25B(A1H2L)=22.29Al+++ sp R4N.X 25°C 0.10M U 1973NNb (102626) 885 K(Al+H2L=AlHL+H)=1.9K(AlHL+2HL)=8.1Medium: (Na,NH4)Cl *********************************** C24H29N3O12S3 H6L 1,2,3-Tris((2-hydroxy-5-sulfobenzyl)amino)propane; -----Metal Mtd Medium Temp Conc Cal Flags Lg K values ______ Al+++ gl NaCl 25°C 0.16M C K1=22.8 1997C0a (103014) 886 ******************************* H6L C24H42N6O12 (6546) 1,4,7,10,13,16-Hexaazacyclooctadecane-N,N',N",N"",N"",N""'-hexaethanoic acid; Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo ______ Al+++ gl NaNO3 25°C 0.20M C K1=22.09 1995KKa (103370) 887 ****************************** C25H2009 CAS 2947-64-0 (4166) 4',3''-Dihydroxy-3,3',4''-trimethylfuchsone-5,5',5''-tricarboxylic acid, Chromoxane violet R Reference ExptNo Mtd Medium Temp Conc Cal Flags Lg K values _____ Al+++ sp oth/un ? 0.10M U K1=10.42 1967LMf (103603) 888 ******************************* H3L Desferrioxamine CAS 70-51-9 (2488) C25H48N608 Desferrioxamine B; NH2.((CH2)5.NOH.CO.C2H4.CO.NH)2.(CH2)5.NOH.CO.CH3 ______ Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo K1=23.9 2000FEc (103800) 889 Al+++ gl KCl 25°C 0.20M C B(A1HL) = 33.8B(A1H2L)=36.6

```
K(Al+H2L)=16.3
                         K(A1+HL)=23.0
*K(AlH2L)=-2.8, *K(AlHL)=-9.9.
    gl KCl 25°C 0.10M C
                         K1=24.50
Al+++
                                   1989EHa (103801) 890
                         K(A1+HL)=24.14
                         K(AlHL+H)=1.18
                         K(All+H)=9.43
**********************************
                 Semi-Xylenol O
3-(N,N-Di(carboxymethyl)aminomethyl)-2-cresolsulfonephthalein;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
                        K1=16.7
Al+++ gl oth/un 25°C 0.10M U
                                  1979MUa (103942) 891
                         K(Al+HL)=9.1
                         K(All+H)=3.2
                         K(Al(OH)L+H)=7.2
********************************
C26H33N3O12S3
                             (7354)
1,1,1-Tris(((2-hydroxy-5-sulfobenzyl)amino)methyl)ethane;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaCl 25°C 0.16M C K1=22.5
                                   1997C0a (104061) 892
****************************
C26H48N6010
                            CAS 207388-25-8 (7648)
Triethylenetetramine-N,N,N',N"',N"'-hexaethanoic acid NN-bis(butanamide);
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
                                    Reference ExptNo
______
Al+++ gl R4N.X 25°C 0.10M C
                         K1=14.59
                                   1998ACc (104304) 893
                         K(All+H)=4.40
                         K(All(OH)+H)=7.53
                         K(All+Al)=5.58
                         K(A12L(OH)+H)=3.16
Medium: N(CH3)4NO3. K(Al2L(OH)2+2H)=6.92.
Adriamycin CAS 25316-40-9 (2407)
C27H29N011
Doxorubicin;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl KNO3 25°C 0.05M C
                                   1993PFa (104452) 894
                         B(AlHL)=16.70
                         B(A1H-1L)=7.45
                         B(A1H-2L)=-0.46
********************************
C27H30N4O18S3
                 TRENCAMS
             H9L
                           CAS 252906-93-7 (7599)
```

3,3',3"-[Nitrilotris(2,1-ethanediyliminocarbonyl)]tris(4,5-dihydroxybenzenesulfonic

```
acid);
        Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ gl NaClO4 25°C 0.10M C K1=39.97
                                    2002BBd (104479) 895
                          B(A1HL)=46.27
                          B(A1H3L)=54.79
K1 by spectrophotometry using competitive reaction with edta, pH 7.2.
***********************
             H4L
                  Rutin
                             CAS 153-18-4 (4169)
3,3',4',5,7-Pentahydroxyflavone-3-beta-rutinoside;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp non-aq 25°C 100M C
                                    2001ADb (104505) 896
                         K1eff=-1.92
Medium: MeOH, 0.2 M acetate buffer, pH 5.95. K1eff: Al+HnL=AlL
*******************************
                  Cyanin CAS 2611-67-8 (9240)
2-(3,4-Dihydroxyphenyl)-3,5-bis(beta-D-glucopyranosyloxy)-7-hydroxy-1-benzopyrilium
chloride;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Al+++ sp none 25°C 0.0 C K1=6.74 2003MMa (104512) 897
**********************************
              L Vitamin D3 CAS 67-97-0 (6103)
C27H440
7-Dehydrocholesterol, Cholecalciferol
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      gl alc/w 25°C 70% C K1=12.4 B2=24.40 2003MYc (104613) 898
Medium: 70% v/v EtOH/H2O, 0.10 M KNO3.
************************************
             H6L Enterobactin
C30H27N3O15
                            CAS 28384-96-5 (2259)
Enterobactin; cyclo-((OH)C6H3(OH).CO.NH.CH.CO.CH2)3
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp KCl
             25°C 0.10M C
                                   1991LRa (105189) 899
                          K(All+H)=5.15
                          K(A1HL+H)=3.4
                          K(A1H3L+H)=2.6
**********************************
C30H45N4O6P3
                             CAS 182250-11-9 (8686)
             H3L
Tris(4-(phenylphosphinato)-3-methyl-3-azabutyl)amine;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      nmr NaCl 25°C 0.16M C
                                    1996LRc (105321) 900
```

K(A1+H3L)=0.93K(A1+2H3L)=3.45

Method: 27Al nmr. Medium pH 1.5. *********************************** Xylenol orange CAS 63721-85-5 (432) 5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulf onic acid; ______ Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo ______ Al+++ gl KNO3 25°C 0.10M C M K1=18.18 B2=23.40 1998GBa (105450) 901 K(All+H)=7.45K(AlHL+H)=5.14K(All2+2Mg)=4.03K(All2+2Ba)=1.12K(All2+Ba+Mg)=2.79______ gl NaClO4 25°C 0.10M U B2=27.01981MYa (105451) 902 K(Al+HL)=16.3K(Al(OH)HL+H=AlHL)=5.8K*(A12L)=13.4K*(A12L): K((A1(OH))2L+2H=A12L+2H2OAl+++ sp NaClO4 20°C 0.10M U 1969NNb (105452) 903 K(2A1+H3L)=13.64K(A1+H2L+H3L)=11.79********************************* C32H37N09S H4L SemiMeThymolBlu (427)3-(N,N-Di(carboxymethyl)-aminomethyl)thymolsulfonephthalein; ______ Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo ______ K1=17.9Al+++ gl oth/un 25°C 0.10M U 1979MUa (105663) 904 K(A1+HL)=9.3K(All+H)=3.5K(A1(OH)L+H)=7.6********************************** O-TRENSOX CAS 169209-69-2 (7370) C36H33N7O15S3 H6L Tris-N-(2-aminoethyl-(8-hydroxyquinoline-5-sulphonato-7-carboxamido))amine; ______ Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo _____ sp NaCl04 25°C 0.10M C K1=21.40 2002BBd (106239) 905 Method: spectrophotometry using competitive reaction with edta, pH 7.2. ______ Al+++ sp NaClO4 25°C 0.01M C 2001LAa (106240) 906 B(A1HL)=24.8B(A12HL)=30.4B(A13HL)=34.3Medium: 0.01 M HClO4. Method: spectrophotometric titration.

```
*******************************
C37H44N2O13S
              H6L
                   MeThymol Blue
                               (428)
3,3'-Bis(N,N-di(carboxymethyl)aminomethyl)thymolsulfonephthalein;
_____
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp oth/un 25°C 0.10M C
                                      1997ASa (106583) 907
                           K1eff=6.84
                           K2eff=6.61
Medium: 0.10 M acetate buffer, pH 5.0.
Al+++ gl NaClO4 25°C 0.10M U
                                      1982MYa (106584) 908
                           B(A1HL)=29.14
                           B(A12L)=26.8
Al+++ sp NaClO4 22°C 1.0M U
                                     1967LMg (106585) 909
K(2A1+2H4L=A1H3L+A1H4L+H)=3.88(?),
K(2A1+4H4L=A1(H3L)2+A1(H4L)2+2H)=6.02(?)
***********************************
C40H47N3010
                              CAS 86728-01-0 (5503)
Bis(3-(((2-hydroxy-5-methylbenzyl)amino)methyl)-2-hydroxy-5-methylbenzyl)amine-trie
thanoic acid
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                            K1=15.29 1983YMa (106785) 910
Al+++ gl oth/un 25°C 0.10M U
                           K(AlH-1L+H)=6.19
                           K(A1H-2L+H)=8.23
                           K(A1H-3L+H)=10.02
                           K(All+H)=3.05
*********************************
Polymer
                   Fulvic acid
                               (1523)
Fulvic acid;
            .....
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Al+++ sp oth/un 22°C ? U H
                                      1993LMd (108176) 911
                           K1eff=4.6 (type I site)
                           K2eff=3.5 (type I site)
pH=3.5; method: Synchronous scan fluorescence spectroscopy. Fulvic acid from
northern coniferous forest. K=5.0 and 4.2 (type II site); 5.3 (type III)
****************************
Polymer
               L
                               (3532)
Human transferrin;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
               Al+++ sp KNO3 25°C 0.10M C
                                      1994HCa (108204) 912
                           Keff(Al+HCO3L)=13.72
                           Keff(Al+AlHCO3L)=12.72
```

```
Keff(Al+L)=7.6
```

At pH 7.4 in 0.1M N-(2-hydroxyethyl)piperazine-N'-2-ethanesulfonic acid, (HEPES) and 5mM HCO3

Al+++ sp oth/un 25°C 0.10M U

1990HSb (108205) 913

Keff(Al+L)=13.5
Keff(AlL+L)=12.5

Medium: 0.1 M N-(2-Hydroxyethyl)piperazine-N'-ethanesulfonic acid and 5 mM NaHCO3, pH 7.4.

Al+++ sp oth/un 25°C 0.10M U

1987MSc (108206) 914

K1eff=12.9 K1eff=12.3

Medium: 0.1 M Tris buffer, pH 7.4; 0.027 M HCO3-. By competition with the Al-NTA complex.

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EXPLANATORY NOTES
  DATA Flags are :-
       T Data at other TEMPERATURES
       I Data with various BACKGROUNDS
       H Data for THERMOCHEMICAL quantities
       M Data for TERNARY Complexes
  EVALUATION Flags are :-
       T or IUP=T signifies EVALUATION RATING = Tentative by IUPAC
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