

SC-Database

Software version = 5.81 Data version = 4.62

Experiment list contains 153 experiments for

(no ligands specified)

Metal : Ge(IV)

(no references specified)

(no experimental details specified)

e- HL Electron (442)
Electron;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	kin	oth/un	25?°C	1.00M	U				1965REa	(513) 1
									$K(\text{Ge(IV)} + 2e = \text{Ge}^{++})=0$	
									$K(\text{Ge(IV)}+4e=\text{Ge(s)})=8.38, 124\text{mV}$	

Medium: H2SO4

Ge(IV)	EMF	none	25°C	0.0	U				1959Lba	(514) 2
									$K=-4.0(\text{brown GeO}, -118 \text{ mV})$	
									$K=-9.2(\text{yellow GeO}, -273 \text{ mV})$	
									$K(\text{Ge(II)}+2e=\text{Ge(s)})=7.81(231\text{mV})$	

K: $\text{GeO}_2(\text{s, hex})+2\text{H}+2e=\text{GeO}(\text{s})+\text{H}_2\text{O}$. $K(\text{H}_2\text{GeO}_3+4\text{H}+4e=\text{Ge(s)}+3\text{H}_2\text{O})=0.7(11 \text{ mV})$

Br- HL Bromide CAS 10035-10-6 (19)
Bromide;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	dis	oth/un	25°C	?	C				1991SOa	(1943) 3
									$K(\text{GeMe(OH)}+\text{H}+\text{L}=\text{GeMeL})=-2.57$	
									$K(\text{GeMe(OH)}_2+2\text{H}+2\text{L}=\text{GeMeL}_2)=-4.3$	
									$K(\text{GeMe(OH)}_3+3\text{H}+3\text{L}=\text{GeMeL}_3)=-4.7$	
									$K(\text{GeMe}_2(\text{OH})+\text{H}+\text{L}=\text{GeMe}_2\text{L})=-1.59$	

 $K(\text{GeMe}_2(\text{OH})_2+2\text{H}+2\text{L}=\text{GeMe}_2\text{L}_2)=-3.49$

Cl- HL Chloride CAS 7647-01-0 (50)
Chloride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	dis	oth/un	25°C	?	C				1991SOa	(4931) 4
									$K(\text{Ge(OH)}+\text{H}+\text{L}=\text{GeL(H}_2\text{O)})=-3.02$	
									$K(\text{Ge(OH)}_2+2\text{H}+2\text{L}=\text{GeL}_2)=-3.84$	
									$K(\text{Ge(OH)}_3+3\text{H}+3\text{L}=\text{GeL}_3)=-4.82$	
									$K(\text{Ge(OH)}_4+4\text{H}+4\text{L}=\text{GeL}_4)=-5.09$	

 $K(\text{MeGe(OH)}+\text{H}+\text{L}=\text{MeGeL})=-2.31$; $K(\text{MeGe(OH)}_2+2\text{H}+2\text{L}=\text{MeGe(OH)}_2)=-2.95$; $K(\text{MeGe(OH)}_3+3\text{H}+3\text{L}=\text{MeGeL}_3)=-3.81$; $K(\text{Me}_2\text{Ge(OH)}+\text{H}+\text{L}=\text{Me}_2\text{GeL})=-0.71$; $K(\text{Me}_2\text{Ge(OH)}_2+2\text{H}+2\text{L})=-2.2$

Ge(IV) sp oth/un ? var U 1961ADb (4932) 5
K5K6=-5.06

F- HL Fluoride CAS 7644-39-3 (201)
Fluoride;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	ISE	NaClO4	25°C	3.00M	U	I			1990CIc (6930)	6
								*B(2,-1)=2.9		
								*B(4,0)=7.18		
								*B(4,1)=6.65		
								*B(6,1)=9.94		

*B(6,2)=9.59. *B(p,q): $\text{Ge(OH)}_4 + \text{pHF} = \text{Ge(OH)}_x\text{F}_p + \text{qH} + 4-x \text{H}_2\text{O}$. Data also in 3.0 M LiClO4

Ge(IV)	dis	oth/un	20°C	?	U		K1=1.68	B2=3.03	1979NVa (6931)	7
							B3=4.18			
							B4=5.17			
							B5=6.07			
							B6=7.24			

Ge(IV)	ix	oth/un	?	?	U				1972PAb (6932)	8
							K6=3.21			

Ge(IV)	ix	KCl	?	0.50M	U				1968PMf (6933)	9
							K6=3.86			

Ge(IV)	ISE	oth/un	25°C	var	U	T			1965RKa (6934)	10
							K=-30.9			
							K(GeF6+2H2O=4H+6F+GeO2)=-25.8			

K: $\text{K}_2\text{GeF}_6(\text{s}) + 2\text{H}_2\text{O} = 2\text{K} + 4\text{H} + 6\text{F} + \text{GeO}_2(\text{s, hex})$

Ge(IV)	oth	NaCl	50°C	0.40M	U	T	H		1964RKb (6935)	11
								K(GeF5H2O+HF=GeF6+H3O)=0.34		

Method: chemical analysis. K=0.66(0 C), 0.62(10 C), 0.58(20 C), 0.50(30 C), 0.42(40 C). At 25 C: DH(K)=10.8 kJ mol⁻¹, DS=-26.3 J K⁻¹ mol⁻¹

Ge(IV)	oth	oth/un	25°C	dil	U	T			1964RKc (6936)	12
								K(GeF4H2O+HF+F=GeF6)=5.3		

Method: chemical analysis, quinhydrone electrode. At 0 C: K=5.9

Ge(IV)	EMF	NaClO4	25°C	0.50M	U				1963BPb (6937)	13
								K(Ge(OH)4+4HF)=7.30		
								K(Ge(OH)4+5HF)=8.94		

MoO4-- H2L Molybdate (443)
Molybdate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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 Ge(IV) sp oth/un ? ? U 1960KRb (8735) 14
 $K(H_4GeO_4 + 4H_2Mo_3O_{10} = H_4GeMo_{12}O_{40} + 4H_2O) = 12.86$ (pH 2.40)

OH- HL Hydroxide (57)
 Hydroxide;

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl NaCl04 25°C 0.10M C 2000KAa (11543) 15
 $K(Ge(OH)_4 = GeO(OH)_3 + H) = -9.16$

 Ge(IV) gl oth/un 25°C 0.0 C T 1998PSb (11544) 16
 $K(GeO_2(s) + 2H_2O = Ge(OH)_4) = -5.02$
 $K(GeO(OH)_3 + H = Ge(OH)_4) = 9.32$

Method: solubility of GeO_2 (tetr) in dil KOH, 21-90 C. Also solubility data for GeO_2 at pH 1.5-10 at 25-350 C.

 Ge(IV) sol NaCl 25°C 0.10M C 1998PSc (11545) 17
 $K_s(GeO_2 + 2H_2O = Ge(OH)_4) = -1.38$

Method: solubility of GeO_2 (hex) in NaCl.

 Ge(IV) sol none RT 0.0 C 1990DEa (11546) 18
 $K_s(Ge(OH)_4 + 2H) = -19.26$
 $K(4Ge(OH)_4(s) + GeO_2(OH)_2) = 13.15$

K: $4Ge(OH)_4(s) + GeO_2(OH)_2 = Ge_5O_{11} + 9H_2O$

 Ge(IV) sp KNO3 25°C 0.10M U I K1=14.18 B2=27.98 1968NFa (11547) 19
 B3=41.52
 B4=54.81

K1=13.73, B2=29.28, B3=43.47, B4=56.98(I=1). Also when I=0.3, 0.5

 Ge(IV) dis oth/un 25°C U K1=14.78 B2=29.18 1966ANa (11548) 20
 B3=43.32
 B4=56.85

Medium: LiCl

 Ge(IV) sol oth/un 25°C var U 1964GZa (11549) 21
 $*K_s(Ge(OH)_2(s) + H = GeOH) = -1.26$
 $K(GeOH + H = Ge(II)) = -1.7 ?$

Ge as Ge^{++} ?

O2-- H2L Peroxide CAS 7772-84-1 (2813)
 Peroxide; -0.0-

 Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1960ARb (12665) 22
 $K(GeO(OH)_3 + 2H_2L) = 1.68$

S04-- H2L Sulfate CAS 7664-93-9 (15)
Sulfate;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) sp oth/un 70°C ? U 1974NOa (16226) 23
K(GeO2+HL)=-0.15

Medium: H_2SO_4

CH40	L	Methyl alcohol	CAS 67-56-1	(597)
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Methanol; CH₃.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	EMF	alc/w	20°C	100%	U	1964GUa	(17882)	24
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$$K(\text{Ge}(\text{H}-1\text{L})_3 + \text{H}-1\text{L}) = 13.65$$
$$K(\text{Ge}(\text{H}-1\text{L})_4 + \text{H} = \text{Ge}(\text{H}-1\text{L})_3) = 2.95$$

Method: H electrode. Medium: MeOH, 1.0 M Me₄NCI

C2H2O4 H2L Oxalic acid CAS 144-62-7 (24)

Ethanedioic acid; (COOH)₂

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	ix	oth/un	25°C	?	U	1964KSd (18914)	25
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 $K_3 = 3.5$

C2H4O3 HL Glycolic acid CAS 79-14-1 (33)

2-Hydroxyethanoic acid; $\text{HO}\cdot\text{CH}_2\cdot\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) gl KNO3 25°C 0.10M U 1975BPF (20552) 26

$$K(\text{Ge}(\text{OH})_2 + 2\text{L}) = 0.25$$

C2H6OS HL CAS 60-24-2 (841)

2-Mercaptoethanol; HS.CH₂.CH₂.OH

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) gl KCl 25°C 0.10M U 1963ATa (22065) 27

$$K(H_2GeO_3 + 2HL = GeOH(H_1L)_2 + H) = -4.22$$

C2H6O2	L	Ethyleneglycol	CAS 107-21-1	(924)
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1,2-Dihydroxyethane (Ethane-1,2-diol); $\text{HO.CH}_2.\text{CH}_2.\text{OH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) EMF KCl 25°C 0.10M U 1959ANa (22147) 28

$$K(\text{HGeO}_3 + \text{L}) = 0.17$$

$$K(\text{HGeO}_3+2\text{L})=-0.37$$

Method: quinhydrone electrode

C2H8O7P2 H4L HEDPA CAS 2809-21-4 (436)

1-Hydroxyethane-1,1-diphosphonic acid; $\text{CH}_3.\text{C}(\text{OH})(\text{PO}_3\text{H}_2)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	sp	NaNO3	20°C	0.10M	U				1983SBb (23376)	29
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$$K(\text{Ge}(\text{OH})_2+\text{H}_3\text{L})=2.52$$

C3H6O5S2 HL Xanthic acid CAS 151-01-9 (590)

(Ethoxy)dithiomethanoic acid; $\text{CH}_3.\text{CH}_2\text{O}.\text{CSSH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	sp	KNO3	20°C	0.10M	U	I			1982SGc (24873)	30
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$$K(\text{Ge}(\text{OH})_2+2\text{L})=8.81$$

C3H6O3 HL L-Lactic acid CAS 79-33-4 (82)

L-2-Hydroxypropanoic acid; $\text{CH}_3.\text{CH}(\text{OH}).\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	gl	KNO3	25°C	0.10M	U				1975BPf (25454)	31
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$$K(\text{Ge}(\text{OH})_2+2\text{L})=0.46$$

Ge(IV)	con	NaCl	18°C	1.0M	U				1957VAa (25455)	32
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$$K(\text{H}_2\text{GeO}_3+2\text{HL})=0.6(?)$$

Ge(IV)	gl	oth/un	18°C	0.0	U				1957VAa (25456)	33
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$$K(\text{H}_2\text{GeO}_3+2\text{HL})=1.9(?)$$

C3H8O2 L Propyleneglycol CAS 57-55-6 (2025)

Propan-1,2-diol; $\text{CH}_3.\text{CH}(\text{OH}).\text{CH}_2(\text{OH})$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	EMF	KCl	25°C	0.10M	U				1959ANa (27677)	34
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$$K(\text{HGeO}_3+\text{L})=0.28$$

$$K(\text{HGeO}_3+2\text{L})=0.06$$

Method: quinhydrone electrode.

C3H8O3 L Glycerol CAS 56-81-5 (2707)

Propane-1,2,3-triol; $\text{HO}.\text{CH}_2.\text{CH}(\text{OH}).\text{CH}_2.\text{OH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV)	gl	KCl	25°C	0.10M	U	I			1957ANa (27735)	35
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$$K(\text{HGeO}_3+\text{L})=1.21$$

$$K(\text{HGeO}_3+2\text{L})=1.94$$

$$K(\text{HGeO}_3+2\text{L})=1.105-1.700\sqrt{I}$$

C3H12N09P3 H6L NTPA CAS 6419-19-8 (2920)

Nitrilotris(methylenephosphonic acid); $\text{N}(\text{CH}_2\text{PO}_3\text{H}_2)_3$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp KNO3 20°C 0.10M U 1984SBa (28570) 36

$$K(\text{Ge}+\text{H}_2\text{L})=13.64$$

C4H605 H2L Malic acid CAS 617-48-1 (393)

2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; $\text{HOOC}.\text{CH}_2.\text{CH}(\text{OH}).\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) con NaCl 18°C 1.0M U 1957VAa (30639) 37

$$K(\text{H}_2\text{GeO}_3+2\text{H}_2\text{L})=0.68$$

Ge(IV) gl oth/un 18°C 0.0 U 1957VAa (30640) 38

$$K(\text{H}_2\text{GeO}_3+2\text{H}_2\text{L})=2.92$$

C4H606 H2L L-Tartaric acid CAS 87-69-4 (92)

L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; $\text{HOOC}.\text{CH}(\text{OH}).\text{CH}(\text{OH}).\text{COOH}$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaNO3 25°C 0.10M U 1973BPa (31266) 39

$$K(\text{Ge}(\text{OH})_2+\text{H}_2\text{L})=4.13$$

Ge(IV) gl NaCl 18°C 1.0M U 1957VAa (31267) 40

$$K(\text{H}_2\text{GeO}_3+\text{HL})=5.2$$

C4H1002 L Butanediol CAS 26171-83-5 (3574)

Butanediol (1,2-/1,3-/1,4- etc not stated)

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) EMF KCl 25°C 0.10M U 1959ANa (34667) 41

$$K(\text{HGeO}_3+\text{L})=0.64$$

$$K(\text{HGeO}_3+2\text{L})=0.04$$

Method: quinhydrone electrode.

C4H1003 L CAS 623-39-2 (3577)

3-Methoxypropan-1,2-diol; $\text{CH}_2(\text{OH}).\text{CH}(\text{OH}).\text{CH}_2.\text{OCH}_3$

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) oth KCl 25°C 0.10M U 1959ANa (34707) 42

$$K(\text{HGeO}_3+\text{L}=\text{GeO}_2\text{H}-2\text{L})=0.84$$

$$K(\text{HGeO}_3 + 2\text{L} = \text{HGeO}(\text{H} - 2\text{L})_2) = 0.58$$

Method: quinhydrone electrode

C4H11NO8P2 H5L CAS 2439-99-8 (2129)
N-Carboxymethyl-N,N-bis(methylenephosphonic acid); $\text{HOOCH}_2\text{N}(\text{CH}_2\text{PO}_3\text{H}_2)_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U				1988SBb (35109)	43
									$K(\text{Ge} + \text{HL}) = 17.1$	

Phosphate buffer pH=6

Ge(IV)	sp	KNO3	20°C	0.10M	U				1986SBb (35110)	44
									$K(\text{Ge}(\text{OH})_2 + \text{H}_2\text{L}) = 4.18$	

C5H4O3 HL Pyromeconic aci CAS 496-63-9 (3600)
3-Hydroxy-4H-pyran-4-one;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U				1967CBb (36272)	45
									$K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 2.86$	

C5H5N L Pyridine CAS 110-86-1 (31)
Pyridine, Azine;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	cal	non-aq	25°C	100%	U	H			1967MOb (36638)	46
Medium: n-hexane. $\text{DH}(\text{GeF}_4(\text{l}) + 2\text{L}(\text{l}) = \text{GeF}_4\text{L}_2(\text{c})) = -202.3 \text{ kJ mol}^{-1}$, $\text{DH}(\text{GeF}_4(\text{g}) + 2\text{L}(\text{l}) = \text{GeF}_4\text{L}_2(\text{c})) = -224$; $\text{DH}(\text{GeCl}_4(\text{g}) + 2\text{L}(\text{l}) = \text{GeCl}_4\text{L}_2(\text{c})) = -207$. Plus others										

C5H10NO7P H4L PMIDA CAS 5994-61-6 (2433)
N-(Phosphonomethyl)iminodiethanoic acid; $\text{H}_2\text{O}_3\text{P}.\text{CH}_2.\text{N}(\text{CH}_2.\text{COOH})_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U				1988SBb (39676)	47
									$K(\text{Ge}(\text{OH}) + \text{HL}) = 10.4$	

Phosphate buffer pH 6

Ge(IV)	sp	KNO3	20°C	0.10M	U				1986SBb (39677)	48
									$K(\text{Ge}(\text{OH})_2 + \text{HL}) = 6.48$	

C5H10O5S2 HL CAS 110-50-9 (591)
(Butoxy)dithiomethanoic acid; $\text{CH}_3.\text{CH}_2.\text{CH}_2.\text{CH}_2\text{O}.\text{CSSH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U	I			1982SGc (40158)	49

$$K(\text{Ge}(\text{OH})_2 + 2\text{L}) = 8.82$$

C5H10O4 L Deoxy-Ribose CAS 533-67-5 (7470)
2-Deoxy-D-ribose, 2-Deoxy-D-erythro-pentose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	U				1979HUa (40327)	50
									$K(\text{H}_2\text{GeO}_3 + \text{L}) = 3.44$	

C5H10O5 L D-Arabinose CAS 10323-20-3 (3606)
D-Arabinose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	U				1959ATa (40335)	51
									$K(\text{HGeO}_3 + 2\text{L} = \text{HGeO}(\text{H} - 2\text{L})_2) = 3.52$	

C5H10O5 L D-Xylose CAS 58-86-6 (3607)
D-Xylose;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	U				1959ATa (40362)	52
									$K(\text{HGeO}_3 + 2\text{L} = \text{HGeO}(\text{H} - 2\text{L})_2) = 3.38$	

C5H10O5 L L-Arabinose CAS 5328-37-0 (1616)
L-Arabinose

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	U				1959ATa (40370)	53
									$K(\text{HGeO}_3 + 2\text{L} = \text{HGeO}(\text{H} - 2\text{L})_2) = 3.63$	

C6H03Cl3 HL CAS 69173-78-0 (3668)
Trichlorohydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U				1966BBb (42031)	54
									$K(\text{Ge}(\text{OH})_4 + 2\text{HL}) < 1.4$	

C6H2N2O8 H2L Nitroanilic aci CAS 479-22-1 (3669)
3,6-Dinitro-2,5-dihydroxy-1,4-benzoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl04	25°C	3.0M	U				1967BBa (42034)	55
									$K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 4.9$	

Medium: LiCl04

C6H2O4Cl2 H2L Chloranilic acid CAS 87-88-7 (1281)
3,6-Dichloro-2,5-dihydroxy-1,4-benzoquinone;

Ge(IV) sp KCl 25°C 0.50M U 1964BBb (42052) 57
K(Ge(OH)2L2+H)=0.8

Ge(IV) sp oth/un 25°C 2.50M U 1962NFa (42053) 58
 K(Ge(OH)₄+HL)=7.64
 K(Ge(OH)₃L+HL)=6.30
 K(Ge(OH)₂L₂+HL)=5.65

C6H4N2O6 H2L CAS 7659-29-2 (2694)
1,2-Dihydroxy-3,5-dinitrobenzene; (HO)2.C6H2(NO2)2

C6H4O4 H2L CAS 615-94-1 (1280)
2,5-Dihydroxy-1,4-benzoquinone;

Ge(IV) sp NaCl 25°C 0.50M U 1966BBb (42306) 61
K(Ge(OH)4+2HL=Ge(OH)2L2)=9.1

C6H4O5 H2L Comenic acid CAS 499-78-5 (2544)
3-Hydroxypyran-4-one-6-carboxylic acid;

C6H4O10S2	H4L	Euthiochronic a	(3670)
3,6-Disulfo-2,5-dihydroxy-1,4-benzoquinone;			

Ge(IV) sp NaCl 25°C 0.50M U 1967BBa (42333) 63
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.35

C6H5NO4 H2L 3-Nitrocatechol CAS 6665-98-1 (2685)
1,2-Dihydroxy-3-nitrobenzene; O2N.C6H3(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl04 25°C 0.10M U 1970NLc (42861) 64
B3=59.59

C6H5NO4 H2L 4-Nitrocatechol CAS 3316-09-4 (890)
1,2-Dihydroxy-4-nitrobenzene; O2N.C6H3(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp KCl 25°C 0.10M U 1967PBd (42928) 65
K(Ge(OH)4+3H2L=GeL3+2H)=3.90

C6H5O2Cl H2L 4-Cl-Catechol CAS 2138-22-9 (1656)
1,2-Dihydroxy-4-chlorobenzene; Cl.C6H3(OH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.10M U 1967PBd (43082) 66
K(Ge(OH)4+3H2L=GeL3+2H)=0.65

C6H5O4Cl HL Chlorokojic aci (3086)
3-Chloro-5-hydroxy-2-hydroxymethyl-4-pyrone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967CBb (43133) 67
K(Ge(OH)4+2HL=Ge(OH)2L2)=2.33

C6H5O4I HL Iodokojic acid CAS 40838-33-3 (3681)
3-Iodo-5-hydroxy-2-hydroxymethyl-4-pyrone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967CBb (43143) 68
K(Ge(OH)4+2HL=Ge(OH)2L2)=2.49

C6H6O2 H2L Catechol CAS 120-80-9 (534)
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl NaCl 25°C 0.10M M 1998PSc (43768) 69

$K(\text{Ge}(\text{OH})_4 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H} + 4\text{H}_2\text{O}) = -1.39$

Method: solubility of $\text{GeO}_2(\text{hex})$ in 0.1 m NaCl/0.01-0.05 m H_2L .

Ge(IV)	gl	oth/un	25°C	0.0	U		1963ANc (43769)	70
						$K(\text{HGeO}_3 + 3\text{H}_2\text{L} = \text{HGeL}_3) = 8.67$		

Ge(IV)	gl	KCl	25°C	0.10M	U		1959AMa (43770)	71
						$K(\text{H}_2\text{GeO}_3 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H}) = -0.77$		

C6H6O3	H3L	Pyrogallol	CAS 87-66-1	(696)
1,2,3-Trihydroxybenzene; $\text{C}_6\text{H}_3(\text{OH})_3$				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
Ge(IV)	gl	oth/un	25°C	0.0	U		1963ANc (43961)	72	
						$K(\text{HGeO}_3 + 3\text{H}_3\text{L} = \text{HGe}(\text{HL})_2) = 9.05$			

Ge(IV)	gl	KCl	25°C	0.10M	U		1959AMa (43962)	73
						$K(\text{H}_2\text{GeO}_3 + 3\text{H}_3\text{L} = \text{Ge}(\text{HL})_3 + 2\text{H}) = -0.22$		

C6H6O3	HL	Maltol	CAS 118-71-8	(2442)
3-Hydroxy-2-methyl-4H-pyran-4-one;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
Ge(IV)	gl	KN03	20°C	0.10M	C		1979MBf (44088)	74	
						$K(\text{GeO}_2 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 4.2$			
						$K(\text{GeO}_2 + 3\text{HL} + \text{H} = \text{GeL}_3 + 2\text{H}_2\text{O}) = 8.3$			

Ge(IV)	sp	NaCl	25°C	0.50M	U		1966BBb (44089)	75
						$K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 3.90$		
						$K(\text{Ge}(\text{OH})_4 + 3\text{HL} + \text{H} = \text{GeL}_3) = 8.05$		

C6H6O3	HL	Allomaltol	CAS 644-46-2	(2688)
5-Hydroxy-2-methyl-4H-pyran-4-one;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
Ge(IV)	sp	NaCl	25°C	0.50M	U		1967CBb (44127)	76	
						$K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 3.43$			

C6H6O4	HL	Kojic acid	CAS 501-30-4	(1800)
5-Hydroxy-2-(hydroxymethyl)-4H-pyran-4-one;				

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
<hr/>									
Ge(IV)	gl	KN03	20°C	0.10M	C		1979MBf (44219)	77	
						$K(\text{GeO}_2 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 3.2$			
						$K(\text{GeO}_2 + 3\text{HL} + \text{H} = \text{GeL}_3 + 2\text{H}_2\text{O}) = 6.0$			

Ge(IV) sp NaCl 25°C 0.50M U 1967CBb (44220) 78
 $K(\text{Ge}(\text{OH})_4 + 2\text{HL} = \text{Ge}(\text{OH})_2\text{L}_2) = 2.81$

C6H6O8S2 H4L Tiron CAS 149-45-1 (104)
 4,5-Dihydroxybenzene-1,3-disulfonic acid; $(\text{HO})_2\text{C}_6\text{H}_2(\text{SO}_3\text{H})_2$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) sp NaCl 25°C 1.00M U I 1967PBd (44453) 79
 $K(\text{Ge}(\text{OH})_4 + 2\text{H}_2\text{L} = \text{GeL}_2) = 3.89$
 $K'(\text{Ge}(\text{OH})_4 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H}) = 3.70$
 $K = 2.30(I=0.11), 3.10(I=0.26), 3.50(I=0.50)$

Ge(IV) gl KCl 25°C var U 1966ATc (44454) 80
 $K(\text{Ge}(\text{OH})_4 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H}) = -2.307 + 27.49\text{SQRTI} / (1 + 2.851\text{SQRTI}) - 0.370\text{I}$

Ge(IV) gl KCl 25°C 0.10M U 1959AMa (44455) 81
 $K(\text{Ge}(\text{OH})_4 + 3\text{H}_2\text{L} = \text{GeL}_3 + 2\text{H}) = -2.74$

C6H8O7 H3L Citric acid CAS 77-92-9 (95)
 2-Hydroxypropane-1,2,3-tricarboxylic acid; $\text{HOOCCH}_2\text{CH}(\text{OH})(\text{COOH})\text{CH}_2\text{COOH}$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) gl NaCl 25°C 0.10M M 1998PSc (46126) 82
 $K(\text{Ge}(\text{OH})_4 + 2\text{H}_3\text{L} = \text{Ge}(\text{OH})_2(\text{H}_2\text{L})_2 + 2\text{H}_2\text{O}) = 6.2$, $K(\text{Ge}(\text{OH})_4 + \text{H}_2\text{L} = \text{Ge}(\text{OH})_3\text{HL} + \text{H}_2\text{O}) = 2.4$. Method: solubility of $\text{GeO}_2(\text{hex})$ in 0.1 m NaCl/0.02 m H3L.

Ge(IV) sp NaNO3 25°C 0.10M U 1973BPa (46127) 83
 $K(\text{Ge}(\text{OH})_2 + \text{H}_3\text{L}) = 2.01$ pH 1-2

C6H9NO6 H3L NTA CAS 139-13-9 (191)
 Nitritotriethanoic acid; $\text{N}(\text{CH}_2\text{COOH})_3$

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) sp KNO3 20°C 0.10M U 1988SBb (46843) 84
 $K(\text{Ge}(\text{OH})_2 + \text{HL}) = 4.06$

Phosphate buffer pH=6

Ge(IV) sp KNO3 20°C 0.10M U 1986SBb (46844) 85
 $K(\text{Ge}(\text{OH})_2 + \text{HL}) = 3.82$

Ge(IV) gl KNO3 20°C 0.10M U 1981MMe (46845) 86
 $K(\text{GeO}_2 + \text{H}_2\text{L} = \text{Ge}(\text{OH})_2\text{L}) = 4.42$

C6H10O7 HL Glucuronic acid CAS 6556-12-3 (599)
 D-Glucuronic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
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Ge(IV) gl KCl 25°C 0.10M M K1=1.70 1986HPb (48420) 87

C6H11NO5 H2L HIMDA CAS 93-62-9 (192)
N-(2-Hydroxyethyl)iminodiethanoic acid; HO.CH2.CH2.N(CH2.COOH)2

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp KNO3 20°C 0.10M U 1988SBb (48739) 88
K(Ge(OH)2+L)=8.42

Phosphate buffer pH=6

C6H12O5S2 HL CAS 123-97-7 (6144)
Pentoxidythiomethanoic acid; C5H11.0.C(S)SH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp KNO3 20°C 0.10M U I 1982SGc (49411) 89
K(Ge(OH)2+2L)=8.72

C6H12O5 L L-Rhamnose CAS 634-74-2 (3659)
6-Deoxy-L-mannose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M U 1959ATa (49507) 90
K(HGeO3+2L=HGeO(H-2L)2)=3.24

C6H12O6 L D-Fructose CAS 57-48-7 (1561)
D-Fructose

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl NaCl 25°C 0.10M M 1998PSc (49548) 91
K(GeO(OH)3+2L=Ge(OH)(H-2L)2+3H2O)=5.48.
2.4. Method: solubility of GeO2(hex) in 0.1 m NaCl/0.02 m L.

Ge(IV) gl KCl 25°C var U I 1963NFa (49549) 92
K(HGeO3+2L)=4.273+1.155SQRTI

Ge(IV) gl KCl 25°C 0.10M U 1958ANa (49550) 93
K(HGeO3+2L=HGeO(H-2L)2)=5.48

C6H12O6 L D-Galactose CAS 59-23-4 (1559)
D-Galactose

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C var U I 1963NFa (49565) 94
K(HGeO3+2L)=2.117+1.297SQRTI

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Ge(IV)      gl  KCl    25°C 0.10M U                1958ANa (49566) 95
                                         K(HGeO3+2L=HGeO(H-2L)2)=3.29
*****
C6H12O6          L    D-Glucose          CAS 492-62-6 (1560)
D-Glucose
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ge(IV)      gl  KNO3   20°C 0.10M M                1980MBc (49589) 96
                                         K(GeO2+2H2L=Ge(OH)L2+H)=-6.33
                                         K'(Ge(OH)L2+H2L=GeL3+H)=-10.6
For L=D-dulcitol, K=-3.88, K'=-10.0; L=D-adonitol, K=-5.43, K'=-10.6.
-----
Ge(IV)      gl  KCl    25°C var U I                1963NFa (49590) 97
                                         K(HGeO3+2L)=1.451+1.178SQRTI
-----
Ge(IV)      gl  KCl    25°C 0.10M U                1958ANa (49591) 98
                                         K(HGeO3+2L=HGeO(H-2L)2)=3.46
*****
C6H12O6          L    D-Mannose          CAS 3458-28-4 (1562)
D-Mannose
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ge(IV)      gl  KCl    25°C 0.10M U                1958ANa (49606) 99
                                         K(HGeO3+2L=HGeO(H-2L)2)=4.13
*****
C6H12O6          L    Sorbose          CAS 87-79-6 (930)
L(-)-Sorbose;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ge(IV)      gl  KCl    25°C 0.10M U                1959ATa (49614) 100
                                         K(HGeO3+2L=HGeO(H-2L)2)=5.35
*****
C6H12O6          L    Inositol          CAS 87-89-8 (2285)
myo-Inositol, meso-Inositol;
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ge(IV)      gl  KCl    25°C 0.10M U                1967FAa (49638) 101
                                         K(HGeO3+2L=HGeO(H-2L)2)=2.140
*****
C6H12O7          HL   Gluconic acid    CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH
-----
Metal      Mtd Medium Temp Conc Cal Flags Lg K values          Reference ExptNo
-----
Ge(IV)      gl  KCl    25°C 0.10M M                K1=2.06          1986HPb (49721) 102

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C6H13NO6 HL CAS 84518-56-9 (4387)
2-Amino-2-deoxy-D-gluconic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	gl	NaCl04	25°C	0.10M	U	M		2000KAa (50532)	103
							B(GeO(OH)H-1L)=3.01		
							B(Ge(OH)2(H-1L)2)=6.63		
							B(GeO(OH)H-2L)=-6.23		
							B(Ge(OH)(H-1L)2)=14.35		

Metal is Ge(OH)4. Also data for ternary species Ge(OH)4ML, M = Zn, Cd, Pb.

C6H14O6 L D-Dulcitol CAS 608-66-2 (3663)
D-Galactitol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	gl	KCl	25°C	0.10M	U			1959ARa (51061)	104
							K(HGeO3+2L=HGeO(H-2L)2)=4.71		

C6H14O6 L D-Mannitol CAS 69-65-8 (3664)
D-Mannitol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	gl	KNO3	20°C	0.10M	C			1979MBf (51081)	105
							K(GeO2+3H2L=GeL3+2H+2H2O)=-13.7; K(GeO2+2H2L=Ge(OH)L2+H+H2O)=-4.0;		
							K(Ge(OH)L2+H2O=GeL3+H+H2O)=-9.7		
Ge(IV)	gl	NaCl	25°C	0.50M	U			1973PAb (51082)	106
							K(Ge(OH)4+L+H2O=GeH-1(OH)4L+H)=-6.43, K(Ge(OH)4+2L+H2O=GeH-1(OH)4L2+H)=-3.95		
							K(2Ge(OH)4+2L+2H2O=(Ge(OH)4)2H-2L2+2H)=-10.68		

Ge(IV)	gl	KCl	25°C	var	U			1963NFA (51083)	107
							K(HGeO3+2L)=3.394+1.055SQRTI		

Ge(IV)	gl	KCl	25°C	0.10M	U			1959ARa (51084)	108
							K(HGeO3+2L=HGeO(H-2L)2)=4.53		

C6H14O6 L Glucitol CAS 50-70-4 (2878)
D-Sorbitol;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg K values	Reference	ExptNo
Ge(IV)	gl	KNO3	20°C	0.10M	C			1979MBf (51103)	109
							K(GeO2+3H2L=GeL3+2H+2H2O)=-12.3; K(GeO2+2H2L=Ge(OH)L2+H+H2O)=-3.7;		
							K(Ge(OH)L2+H2O=GeL3+H+H2O)=-8.6		

Ge(IV)	gl	KCl	25°C	0.10M	U			1959ARa (51104)	110
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K(GeHL3+H)=2.53

Ge(IV) con NaCl 18°C 1.0M U 1957VAa (59837) 117
K(H2GeO3+2HL)=2.0

Ge(IV) gl oth/un 18°C 0.0 U 1957VAa (59838) 118
K(H2GeO3+2HL)=2.92

C9H5NOC12 HL CAS 773-76-2 (3278)
5,7-Dichloro-8-hydroxyquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967TMd (63543) 119
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.7

C9H6NO4IS H2L Ferron CAS 547-91-1 (275)
7-Iodo-8-hydroxyquinoline-5-sulfonic acid; (HO)(HO3S)C9H4NI

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967TMd (63805) 120
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.78

C9H7N L CAS 119-65-3 (487)
Isoquinoline;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) cal non-aq 25°C 100% U H 1967MOb (64026) 121
Medium: n-hexane. Many data; DH(GeF4(1)+2L(1)=GeF4L2(c))=-149.2 kJ mol-1
DH(GeF4(g)+2L(1)=GeF4L2(c))=-170.9, DH(GeCl4(1)+2L(1)=GeCl4L2(c))=-93.2

C9H7NO HL Oxine CAS 148-24-3 (504)
8-Hydroxyquinoline (8-quinolinol);

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967TMd (64279) 122
K(Ge(OH)4+2HL=Ge(OH)2L)=6.61

C9H7NO4S H2L Sulfoxine CAS 84-88-8 (448)
8-Hydroxyquinoline-5-sulfonic acid;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967TMd (64549) 123
K(Ge(OH)4+2HL=Ge(OH)2L2)=6.55

C9H28N3O15P5 10L DTPPH CAS 15827-60-8 (2921)

Diethylenetriamine-N,N,N',N'',N''-penta(methylphosphonic acid);
H2O3PCH2.N(CH2CH2.N(CH2PO3H2)2)2 H

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KNO3	20°C	0.10M	U			K(Ge+H6L)=9.45	1984SBa (68410)	124

C10H6O3 HL CAS 83-72-7 (3294)
2-Hydroxy-1,4-naphthoquinone;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			K(Ge(OH)4+2HL=Ge(OH)2L2) < 3.0	1966BBb (68460)	125

C10H8O5S H3L DHNSA (877)
2,3-Dihydroxynaphthalene-6-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	KCl	25°C	0.10M	U			K(H2GeO3+3H2L=GeL3+2H)=2.0	1967PBd (69849)	126

C10H8O8S2 H4L Chromotropic ac CAS 148-25-4 (1875)
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	gl	KNO3	20°C	0.10M	C			K(GeO2+3H2L=GeL3+2H+2H2O)=-4.8	1979MBf (69953)	127

Ge(IV)	sp	KCl	25°C	0.10M	U			K(H3GeO4+3H2L=HGeL3+2H)=2.30	1967PBd (69954)	128
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C10H9NO HL 8-OH-Quinaldine CAS 826-81-3 (998)
2-Methyl-8-hydroxyquinoline;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			K(Ge(OH)4+2HL=Ge(OH)2L2)=3.4	1967Tmd (70047)	129

C10H9NO4S H2L CAS 29021-67-8 (3926)
2-Methyl-8-hydroxyquinoline-5-sulfonic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	NaCl	25°C	0.50M	U			K(Ge(OH)4+2HL=Ge(OH)2L2)=2.2	1967Tmd (70198)	130

C10H16N2O8	H4L	EDTA	CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;			

Ge(IV) EMF NaCl 25°C 0.50M U 1969NVa (73827) 132
K(Ge(OH)₄+H₄L)=5.3

Ge(IV) vlt NaClO4 25°C 0.10M U 1967K0c (73829) 134
 $K(\text{Ge}(\text{OH})_4 + \text{H}_4\text{L} = \text{Ge}(\text{OH})_2\text{H}_2\text{L}) = 4.80$
 $K(\text{Ge}(\text{OH})_4 + \text{H}_3\text{L} = \text{Ge}(\text{OH})_2\text{HL}) = 4.58?$

C10H18N2O7	H3L	HEDTA	CAS 150-39-0	(392)
N-(Hydroxyethyl)diaminoethane-N,N',N'-triethanoic acid;				

C11H18N2O8	H4L	PDTA	CAS 4408-81-5	(1655)
1,2-Diaminopropane-N,N,N',N'-tetraethanoic acid;				

C12H9N2O2C1 H2L CAS 29600-20-2 (2638)
4-Chlorobenzene-(1-azo-1')-3',4'-dihydroxybenzene; ClC6H5.N:N.C6H3(OH)2

C12H22O11 L Turanose CAS 547-25-1 (2701)
3-O-D-Glucopyranosyl-D-fructose;

Ge(IV) gl KCl 25°C 0.10M M K1=2.32 B2=3.25 1986HPb (82866) 138

C12H22O11 L alpha-Lactose CAS 5989-81-1 (2486)
4-D-Beta-D-Galactopyranosyl-alpha-D-glucose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=1.97 1986HPb (82875) 139

C12H22O11 L Maltose CAS 6363-53-7 (2705)
4-O-alpha-D-Glucopyranosyl-D-glucose, Maltobiose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=1.48 1986HPb (82880) 140

C12H22O11 L Cellobiose CAS 528-50-7 (2697)
4-O-beta-D-Glucopyranosyl-D-glucose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=1.58 1986HPb (82886) 141

C12H22O11 L Melibiose CAS 66009-10-7 (2699)
6-O-D-Galactopyranose-D-glucose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=2.32 B2=3.24 1986HPb (82890) 142

C12H22O11 L Gentiobiose CAS 554-91-6 (2698)
6-O-D-Glucopyranosyl-D-glucose, Amygdalose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=1.27 1986HPb (82893) 143

C12H22O11 L Trehalose CAS 6138-23-4 (2700)
D-Glucopyranosyl-D-glucopyranoside;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=1.12 1986HPb (82900) 144

C12H22O11 L Sucrose CAS 57-50-1 (2523)
beta-D-Fructofuranosyl-alpha-D-glucopyranoside; Saccharose;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=1.00 1986HPb (82910) 145

C12H24O11 L Maltitol CAS 585-88-6 (2709)
4-O-alpha-D-Glucopyranosyl-D-glucitol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=3.22 1988HLA (83683) 146

C12H24O11 L Lactitol CAS 535-94-4 (2710)
4-O-beta-D-Galactopyranosyl-D-glucitol;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) gl KCl 25°C 0.10M M K1=3.08 1988HLA (83686) 147

C14H8O7S H3L DASA CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp oth/un 25°C 0.10M U 1972NFb (86732) 148
B3=52.80

Medium: acetate

C15H11N3O HL PAN CAS 85-85-8 (572)
1-(2-Pyridylazo)-2-naphthol; C5H4N.N:N.C10H6.OH

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp oth/un 27°C ? U M 1974ZSa (91219) 149
Keff(GeCl4+L)=3.3

C18H12O6 H2L (4124)
2,5-Dihydroxy-3,6-diphenoxy-1,4-benzoquinone;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaCl 25°C 0.50M U 1967BBa (96886) 150
K(Ge(OH)4+2HL=Ge(OH)2L2)=8.8

C19H12O9Br2S H6L Bromo Pyrog.Red CAS 16574-43-9 (706)
5',5''-Dibromopyrogallolsulfonephthalein;

Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo

Ge(IV) sp NaNO3 ? 0.10M U 1969NMa (99011) 151
K(Ge(OH)3+3H2L)=12.9

C22H20O13 H5L Carminic acid CAS 1260-17-9 (714)

Carminic acid;

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	oth/un	25°C	?	U				1970BRa (101702)	152
								K(Ge(OH) ₄ +H ₅ L)=4.58		

Medium: conc H₂SO₄

C₂₈H₁₅N₄O₄ L CAS 82-22-4 (3522)
1,1'-Iminodianthraquinone; (1,1'-dianthrimide)

Metal	Mtd	Medium	Temp	Conc	Cal	Flags	Lg	K values	Reference	ExptNo
Ge(IV)	sp	mixed	?	93%	U				1968LNa (104653)	153
								K(HGeO ₂ +HL)=2.35(?)		

Medium: 93.2% H₂SO₄

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EXPLANATORY NOTES

DATA Flags are :-

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS
- H Data for THERMOCHEMICAL quantities

M Data for TERNARY Complexes

END