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SC-Database
Software version = 5.81 Data version = 4.62
Experiment list contains 101 experiments for
(no ligands specified)
2 metals : Nb(V), Nb++++
(no references specified)
(no experimental details specified)
***********************************
               HL
                   Electron
                                (442)
Electron:
          Mtd Medium Temp Conc Cal Flags Lg K values
                                       Reference ExptNo
______
Nb(V)
      vlt oth/un 25°C 1.00M U
                                     1965MHb (709) 1
                          K'=14.40, 426 mV
K': Nb6Cl12++++ + 2e=Nb6Cl12++.
Nb(V) vlt oth/un 25°C 12MM U
                                     1954CVa
                           K(Nb+e)=-3.60, -213 \text{ mV}
Medium 12 M HCl
-----
Nb(V) oth none 25°C 0.0 U
                                      1952LAb (711) 3
                           K=-54.4(-650 \text{ mV})
                           K(Nb(III)+3e)=-56 (-1100 \text{ mV})
K: 0.5Nb204(s)+5H+5e=Nb(s)+2.5H20. From thermodynamic data
______
Nb(V) EMF oth/un 18°C 6.25M U I
                                     1938GGa (712) 4
                           K=-10.84(-313 \text{ mV})
Medium: H2SO4. K: Nb+2e=Nb(III). At I=4.95 M: K=-10.74(-310 mV), 3 M: -12.02
(-347 mV). Also at 18 C. At I=0: K(NbO+2H+2e=Nb(III)+H2O)=-11.88(-343 mV)
______
     EMF oth/un 25°C 9.87M U I
                                      1928KHa (713) 5
                           K=-14.41(-426.1 \text{ mV})
Medium: H2SO4. K(Nb+2e=Nb(III). At I=5.9: K=-13.02(-384.9 mV), I=3: -12.62
(-373.0 \text{ mV})
HL
                   Bromide
                             CAS 10035-10-6 (19)
Br-
Bromide;
           ______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp oth/un 25°C var M
                                      1973LJb (2150) 6
                           K(NbOOHL4+H+L=NbOL5+H2O)=-5.2
Medium: HBr
**********************************
              HL Chloride CAS 7647-01-0 (50)
Chloride;
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
Nb(V) cal non-aq 25°C 100% U HM
                                 1993DSb (5278)
Metal:Nb(IV). Medium:iso-Propyl ether. DH(Nb(H-1A)2B2(s)+2HL=NbL2(H-1A)2+
2HB)=-216.4 kJ mol-1. A:Cyclopentadiene. B:CH3. Also for B=PhS and p-ClC6H4S
Nb(V) dis KCl var U
                          1968SSf (5279) 8
                       Kd(Nb+4Cl+3TBP(benzene))=-1.8
**********************************
            HL Fluoride CAS 7644-39-3 (201)
Fluoride;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
______
Nb(V) ISE non-aq ? 100% C
                                1978GRa (7039) 9
                        K6=6.88
                        K(NbF5+NbF6=Nb2F11)=1.32
Medium: liquid anhydrous HF
______
Nb(V) sp mixed ? 20% U
                                  1973LCa (7040) 10
                        K(NbH2O2+F)=2.62
                        K(NbH2O2+2F)=1.72
                        K(NbH2O2+3F)=1.58
                        K(NbH2O2+4F)=1.52
Medium: 20% H2SO4
-----
Nb(V) ISE NaClO4 25°C 0.50M U
                                  1972L0a (7041) 11
                        K(Nb0F2+F)=3.80
                        K(NbOF3+F)=4.30
                        K(NbOF4+F)=4.51
                        K(NbOF5+F)=4.67
Medium: (Na,H)ClO4. Nb(V)=NbO+++. K(NbOF6+2H+F=NbF7+H2O)=11.4;
K(NbF7+F)=3.08, K(NbF8+F)=4.0
______
Nb(V) EMF KCl 25°C 3.0M U
                                  1970NEb (7042) 12
                       K(Nb(OH)2F4+F)=2.51
______
     dis NaClO4 25°C 5.0M U T
                                 1969ESa (7043) 13
                       K6K7=10.66
------
Nb(V) sp oth/un ? 17.0M U K1=7.12 1969PEc (7044) 14
Medium: H2SO4
_____
Nb(V) sol KNO3 18?°C 0.50M U
                                  1967BNd (7045) 15
                        Ks(Nb(OH)2F(s))=-5.22
                        K(Nb(OH)4F+HF=Nb(OH)4F2+H)=3.6
                        K(Nb(OH)4F+F)=6.8
In 3 M HNO3: Ks(Nb(OH)4F2(s)=Nb(OH)4F2)=-4.82, K(Nb(OH)3F2+HF=Nb(OH)2F3)=4.2
**********************************
I-
            HL Iodide
                          CAS 10034-85-2 (20)
Iodide;
```

```
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      cal non-aq 25°C 100% U HM
                                    1993DSb (8273) 16
Metal:Nb(IV). Medium: Toluene or iso-Propyl ether. DH(Nb(H-1A)2B2(s)+2I2=
NbI2(H-1A)+2BI)=-289.9 kJ mol-1. A:Cyclopentadiene. B:CH3.
*******************************
OH-
              HL
                  Hydroxide
                              (57)
Hydroxide;
        Metal Mtd Medium Temp Conc Cal Flags Lg K values
_____
Nb(V) gl KCl 25°C 3.00M C I
                                    1994EFa (11777) 17
                          K(Nb6019+H=HNb6019)=13.63
                          K(Nb6019+2H=H2Nb6019)=23.55
                          K(Nb6019+3H=H3Nb6019)=32.90
Values at I=0 corr: K=16.11, K(Nb6019+2H)=27.97, K(Nb6019+3H)=39.91.
K(Nb205(s)+5H20=2Nb(OH)5)=-9.71. K(6Nb(OH)5=H3Nb6019+5H)=-14.46.
      dis NaClO4 25°C 0.10M U
Nb(V)
                                    1970GFb (11778) 18
                          *B(NbO2+H2O=NbO2OH+H)=-3.2
Medium: LiClO4
**********************************
             H2L Peroxide CAS 7772-84-1 (2813)
Peroxide; -0.0-
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) sp oth/un 25°C 95% U T H
                                    1971VZa (12683) 19
                          K(NbOSO4+H2L)=3.67
Medium:95% H2SO4. K=3.83(15 C), 3.53(35 C), 3.41(45 C). DH=-25 kJ mol-1
63% H2SO4. K=2.73(25 C), 2.56(35 C), 2.42(45 C). DH=-29
______
Nb(V) sp oth/un 0°C 10% U
                                    1969CKa (12684) 20
                          K(NbOSO4+H2L)=5
Medium: 10% H2SO4
Nb(V) EMF KCl 0°C 1.0M U
                                    1969SPc (12685) 21
                          K(3NbO2L+H=HNb3O6L3)=13.08
K(3NbL3+3H2O+H=HNb3O3L6+3H2L)=4.50. In 3 M KCl: K(NbL4+OH=NbOL3+HL)=1.5,
K(HNbOL3+H2L+OH=NbL4+2H2O)=3.3
______
Nb(V) sp oth/un ? var U
                                    1966BNa (12686) 22
                         K(Nb(OH)4HL+H)=2.7
-----
Nb(V) sp mixed 23°C 97% U
                                    1957AHb (12687) 23
                         K(2Nb(V)+3H2L)=12.70
Medium: 97.2% H2SO4.
-----
Nb(V) sp oth/un ? 96% U
                                    1956SSc (12688) 24
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K(Nb(V)+H2L)=3.6
*******************************
                  Thiocyanate CAS 463-56-9 (106)
Thiocyanate;
         Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp alc/w ? 100% U I
                         K1=3.58 B2=6.74 1964GSc (15186) 25
Nb(V)
                          B3=9.23
Medium: MeOH. In BuOH: K1=4.37, B2=8.58, B4=16.92. In Me2NCHO: K1=3.08,
B2=6.11, B3=8.92, B4=11.55, B5=14.45, B6=16.72, B7=19.28. Nb added as NbCl5
**********************************
S04--
                  Sulfate
                             CAS 7664-93-9 (15)
             H2L
Sulfate:
          Mtd Medium Temp Conc Cal Flags Lg K values
                                     Reference ExptNo
______
Nb(V) ix oth/un 20°C var U
                                    1969MNc (16395) 26
                          K(NbO(OH)2+L)=1.7
Medium: H2SO4. By distribution: K(NbO(OH)2+2L+2H=NbOL2+2H2O)=3.12
In NH42SO4: K(NbO(OH)3L+L+H=NbO(OH)2L+H2O)=1.09
______
Nb(V) sp oth/un 22°C 10.0M U
                                    1966GAc (16396) 27
                          B(Nb(III)4Nb2)=4.3?
Medium: H2SO4
*********************************
              L Methyl alcohol CAS 67-56-1 (597)
Methanol; CH3.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
                                   Reference ExptNo
-----
      EMF alc/w 20°C 100% U
Nb(V)
                                    1971GSa (17891) 28
                          K(Nb+3L=Nb(H-1L)3+3H) > 1
                         K(Nb(H-1L)3+H-1L)=14.3
Medium: MeOH, 1 M Me4NCl
______
Nb(V)
     EMF alc/w 20°C 100% U M
                                    1965GBa (17892) 29
                          K(NbA(L')4+2L'=Nb(L')6+A)=5.1
                          K(Nb(H-1L)4+A)=10.84
                          K(NbA(H-1L)3+L)=12.4
                          K(Nb(L')5+HA=NbA(L')4+L)=5.18
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. L'=H-1L; HA=acetylacetone
______
Nb(V)
      EMF alc/w 20°C 100% U M
                                    1965GBa (17893) 30
                          K(NbAL'3+H2A+L'=NbA2L'2)=13.9
                          K(NbA2L'2+H2A+L'=NbA3L')=7.0
                          K(NbA(L')3+L'=NbA(L')4)=7.89
                          K(NbAL'4+NbAL'3=Nb2A2L'7)=2.5
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl. L'=H-1L; H2B=catechol
______
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1964GUa (17894) 31
       EMF alc/w 20°C 100% U
Nb(V)
                              K(Nb(H-1L)4+H-1L)=10.45
                              K(Nb(H-1L)5+H-1L)=5.45
                              K(Nb(H-1L)6+H=Nb(H-1L)5)=6.15
                              K(Nb(H-1L)7+H=Nb(H-1L)6)=11.15
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl
       EMF alc/w 20°C 100% U
Nb(V)
                                          1964GUa (17895) 32
                              K(NbO(H-1L)2+H-1L)=10.51
                              K(NbO(H-1L)4+H)=6.03
Method: H electrode. Medium: MeOH, 1.0 M Me4NCl
**********************************
                                  CAS 590-17-0 (4217)
Cyanomethyl bromide; Br.CH2.CN
        Mtd Medium Temp Conc Cal Flags Lg K values
                                           Reference ExptNo
______
Nb(V)
       nmr non-aq -60°C 100% U
                                          1971MBa (18364) 33
                              K(NbC15A+L=NbC15L+A)=0.99
                              K(NbC15B+L=NbC15L+B)=0.43
Medium: CHCl3. A=cyanomethyl fluoride, B=cyanomethyl chloride
______
        nmr non-aq -60°C 100% U
                                          1971MBa (18365) 34
Nb(V)
                              K(NbC15A+L=NbC15L+A)=0.55
                              K(NbC15A+L=NbC15L+A)=0.37
Medium: CHCl3. A=cyanomethyl fluoride, B=POCl3
*********************************
                                 CAS 624-75-9 (4219)
C2H2NI
Cyanomethyl iodide; I.CH2.CN
______
       Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
       nmr non-ag -60°C 100% U
                                          1972MBb (18366) 35
                              K(NcC15A+L=NbC15L+A)=0.75
                              K(NbC15B+L=NbC15L+B)=1.18
                              K(NbC15C+L=NbC15L+C)=0.64
Medium: CHCl3. A=cyanomethyl bromide, B=cyanomethyl chloride, C=Et20
*********************************
                H2L
                     Oxalic acid
                                CAS 144-62-7 (24)
Ethanedioic acid; (COOH)2
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
                        ? U
Nb(V)
       sol NaClO4 ?
                                          1970ZPa (18980) 36
                              K3=6.17
Metal ion is NbO+++. Medium : HClO4
Nb(V) sol oth/un 18°C 0.50M U
                              B2 = 35.9
                                          1968BMb (18981) 37
Metal ion is NbO+++
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Nb(V) sol oth/un 18°C 0.50M U
                                  1968BMb (18982) 38
                       K(NbO(OH)2+L)=9.08
K(NbO(OH)2+L)=9.08
      dis NaCl 20°C 4.50M U
                                  1967KOd (18983) 39
Medium: 4.5(NaCl or NaNO3+2.5 M H). K(Nb(OH)4+H2L=Nb(OH)4HL+H)=3.55
K(Nb(OH)4+2H2L=Nb(OH)2L2+2H)=5.13
______
Nb(V) EMF oth/un 25°C 0.50M U
                                  1967NSb (18984) 40
                        K(Nb(OH)4+2HL)=12.11
                       K(Nb(OH)4+2HL+L)=17.15
**************
             L Cyanomethane CAS 75-05-8 (1399)
Acetonitrile; CH3.CN
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V) nmr non-ag -60°C 100% U M 1974GMa (19189) 41
                      K(NbBr5A+L=NbBr5L+A)=0.15
Medium: CH2Cl2. A=t-butylcyanide
-----
Nb(V) nmr non-aq -60°C 100% U M
                                  1972MBb (19190) 42
                     K(NbC15A+L=NbC15L+A)=0.46
Medium: CHCl3. A=1,4-dioxan
______
Nb(V) nmr non-aq -60°C 100% U M
                                  1971MBa (19191) 43
                     K(NbC15A+L=NbC15L+A)=0.76
Medium: CHCl3. A=1-chloro-4-cyanobenzene. K=0.34, A=cyanobenzene;
K=0.68, A=cyanoethane; K=0.38, A=dimethylether.
*************************
C2H6NOC12P L CAS 667-43-0 (910)
Dichloro(dimethylamine)phosphine oxide; (CH3)2N.P(0)Cl2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
_____
Nb(V) nmr non-aq -60°C 100% U M 1974GMa (21900) 44
                       K(NbBr5A+L=NbBr5L+A)=0.72
Medium: CH2Cl2, A=acetonitrile
*******************************
                         CAS 115-10-6 (4214)
Dimethyl ether; CH3.O.CH3
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) nmr non-aq -40°C 100% U M
                                  1972MBb (22019) 45
                      K(NbC15A+L=NbC15L+A)=0.08
Medium: CHCl3. A=dioxan. K=0.38, A=1-chloro-4-cyanobenzene (-60 C)
                Nb(V) nmr non-aq -60°C 100% U M 1971MBa (22020) 46
                       K(NbC15A+L=NbC15L+A)=0.38
Medium: CHCl3. A=1-chloro-4-cyanobenzene
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***********************************
C2H6S
                           CAS 75-18-3 (151)
              L
Dimethyl sulfide; CH3.S.CH3
______
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
                       М
      nmr non-aq -60°C 100% U
                                  1974GMa (22190) 47
Nb(V)
                        K(NbBr5A+L=NbBr5L+A)=1.52
Medium: CH2Cl2, A=t-butylcyanide
    nmr non-aq -60°C 100% U
                                   1972MBb (22191) 48
                         K(NbC15A+L=NbC15L+A)=1.32
Medium: CHCl3. A=t-butylnitrile
************************
             H2L
C3H404
                 Malonic acid
                           CAS 141-82-2 (79)
Propanedioic acid; CH2(COOH)2
______
      Mtd Medium Temp Conc Cal Flags Lg K values
                                  Reference ExptNo
·
      dis NaCl 20°C 4.50M U T
                                   1967KOd (24509) 49
Medium: 4.5(NaCl or NaNO3+2.5 HCl). K(Nb(OH)4+H2L=Nb(OH)4HL+H)=1.72
*****************************
C3H904P
                            CAS 512-56-1 (2431)
Trimethyl phosphate; (CH30)3.P:0
  -----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) nmr non-ag -60°C 100% U
                                   1972BMb (28024) 50
                         K(NbC15L+A=NbC15A+L) > 2.0
K(NbCl50PCl3+L=NbCl5L+0PCl3) > 7.0,. A=tris(dimethylamino)phosphine oxide
Medium: CHCl3
**********************************
             H2L
                 Succinic acid CAS 110-15-6 (112)
1,4-Butanedioic acid; HOOC.CH2.CH2.COOH
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      dis NaCl 20°C 4.50M U
                                   1967KOd (30004) 51
Nb(V)
K(Nb(OH)4+H2L=Nb(OH)4HL+H)=1.53, Medium: (NaCl or NaNO3+2.5 HCl)
*********************************
                 Malic acid
             H2L
                            CAS 617-48-1 (393)
2-Hydroxybutane-1,4-dioic acid, Hydroxy-succinic acid; HOOC.CH2.CH(OH).COOH
    Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      dis NaCl
             20°C 4.50M U
                                   1967KOd (30682) 52
K(Nb(OH)4+H2L=Nb(OH)4HL+H)=2.01. Medium: NaCl or NaBO3 + 2.5 M HCl
*******************************
                  L-Tartaric acid CAS 87-69-4 (92)
L-Tartaric acid, L-2,3-Dihydroxybutanedioic acid; HOOC.CH(OH).CH(OH).COOH
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Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
             20°C 1.00M U B2=33.5
      dis KCl
                                   1971LFc (31314) 53
Metal ion: NbO+++
Nb(V) oth oth/un ? ? U K1=25.15 B2=33.00 1969EMa (31315)
                                                 54
Metal ion: NbO+++
______
      dis NaCl 20°C 4.50M U
                                   1967KOd (31316) 55
K(Nb(OH)4+H2L=Nb(OH)4HL+H)=2.34. Medium: NaCl or NaNO3 + 2.5 M HCl
**********************************
             H2L
                  Aspartic acid
                            CAS 56-84-8 (21)
Aminobutanedioic acid; H2N.CH(CH2.COOH).COOH
-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V)
     dis KNO3 20°C 2.00M U
                                   1969KOb (31898) 56
                         K(NbO2+H2L)=4.82
Medium: HNO3
**********************************
                 1,4-Thioxane CAS 15980-15-1 (4266)
              L
1,4-Oxathiane; cyclo(-0.CH2.CH2.S.CH2.CH2-)
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) nmr non-aq -60°C 100% U M 1972MBb (33190) 57
                         K(NbC15A+L=NbC15L+A)=0.08
Medium: CHCl3. A=t-butyl nitrile
************************
                            CAS 110-01-0 (150)
Tetrahydrothiophene; cyclo(-CH2.CH2.S.CH2.CH2-)
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V)
     nmr non-ag -60°C 100% U
                                   1972MBb (33739) 58
                         K(NbC15A+L=NbC15L+A)=1.80
A=t-butyl mercaptan. Medium: CHCl3
**********************************
                 1,4-Dithiane
                           CAS 505-29-3 (4255)
1,4-Dithiane; cyclo-(S.CH2.CH2.S.CH2.CH2-)
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      nmr non-aq -60°C 100% U
                                   1972MBb (33743) 59
                         K(NbC15A+L=NbC15L+A)=0.16
A=t-butvl nitrile. Medium: CHCl3
**********************************
                  Ether
                           CAS 60-29-7 (3573)
Diethyl ether (ethyl ether, ethoxyethane); C2H5.O.C2H5
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Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
     nmr non-aq -60°C 100% U M
                                   1972MBb (34652) 60
                         K(NbClA+L=NbClL+A)=0.11
A=cyanomethyl bromide. Medium: CHCl3. When A=cyanomethyl chloride, K=0.54
**********************************
                           CAS 352-93-2 (4259)
Diethyl sulfide; C2H5.S.C2H5
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values
______
      nmr non-aq -60°C 100% U M
Nb(V)
                                   1974GMa (34719) 61
                        K(NbBr5A+L=NbBr5L+A)=0.59
A=t-butyl nitrile. Medium: CH2Cl2
______
Nb(V) nmr non-aq -40°C 100% U M
                                   1972MBb (34720) 62
                         K(NbC15A+L=NbC15L+A)=0.66
A=dimethyl ether. Medium: CHCl3. When A=acetone, K=0.88
*********************************
                  Acetylacetone CAS 123-54-6 (164)
              HL
Pentane-2,4-dione; CH3.CO.CH2.CO.CH3
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     EMF non-aq 20°C 100% U M
                                   1971GSa (38035) 63
                         K(NbA5+HL=NbA4L+HA)=5.18
                         K(NbA3L+A)=12.40
                         K(NbA4L+2A=NbA6+L)=5.06
Medium: CH3OH, 1 M Me4NCl. A=CH3OH
**********************************
               L t-Butylnitrile CAS 7188-38-7 (913)
t-Butylcyanide; (CH3)3C.CN
______
      Mtd Medium Temp Conc Cal Flags Lg K values
                                    Reference ExptNo
______
Nb(V) nmr non-aq -60°C 100% U M
                                   1974GMa (38453) 64
                         K(NbBr5A+L=NbBr5L+A)=1.68
Medium: CH2Cl2, A=dimethylether
______
Nb(V) nmr non-aq -60°C 100% U M
                                   1972MBb (38454) 65
                        K(NbC15A+L=NbC15L+A)=0.42
Medium: CHCl3. A=dimethyl ether. When A=1,4-dioxan, K=0.50
______
      nmr non-aq -60°C 100% U M
                                   1971MBa (38455) 66
                         K(NbC15A+L=NbC15L+A)=0.04
Medium: CHCl3. A=cyanomethane. When A=cyanoethane, K=0.72
**********************************
                  Cupferron
                           CAS 135-20-6 (637)
N-Nitrosophenylhydroxylamine; C6H5.N(OH).NO
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Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V) sp alc/w 25°C 50% U
                                    1967LSb (43419) 67
                         K(NbOL2+L)=4.83
Medium: 50% EtOH, 0.1 M (NH4)2SO4
***********************************
                         CAS 120-80-9 (534)
             H2L
                  Catechol
1,2-Dihydroxybenzene, pyrocatechol; HO.C6H4.OH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) EMF alc/w 20°C 100% U
                       Μ
                                    1971GSa (43797) 68
                         K(NbA3L+A)=7.89
                         K(NbA3L+NbA4L=Nb2A7L2)=2.50
                         K(NbA3L+H2L+A=NbA2L2+2HA)=13.9
                         K(NbA2L2+H2L+A=NbAL3+2HA)=6.98
Medium: MeOH, 1.0 M Me4NCl. HA=CH3OH
*******************************
                  Pyrogallol CAS 87-66-1 (696)
             H3L
1,2,3-Trihydroxybenzene; C6H3(OH)3
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Nb(V) sp oth/un 23°C 96% U
                                    1981BVa (43971) 69
                         K(Nb(V)+H3L)=5.62
Medium: 96% H2SO4. In 85%: K(Nb(V)+H3L)=2.30
*******************************
                            CAS 599-71-3 (4398)
Benzenesulfohydroxamic acid; C6H5.S02.NH.OH
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
      sp oth/un 20°C 1.0M U M
                                   1971GVa (45071) 70
Medium: 1-10 M HCl. K(NbOCl5+2H2L=(H2L)2NbOCl5)=4.91
*********************************
                 Ascorbic acid CAS 50-81-7 (285)
             H2L
Ascorbic acid (Vitamin C);
______
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V) sp oth/un ? ? U K1=9.4 1966SAb (45649) 71
**********************************
             H3L Citric acid CAS 77-92-9 (95)
2-Hydroxypropane-1,2,3-tricarboxylic acid; HOOCCH2.CH(OH)(COOH).CH2COOH
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Nb(V) dis oth/un 20°C 4.50M U
                                    1967KOd (46195) 72
Medium: 4.5(NaCl or NaNO3+2.5 HCl). K(Nb(OH)4+H3L=Nb(OH)4H2L+H)=2.94 ?
```

```
***********************************
C6H12O7
             HL Gluconic acid CAS 526-95-4 (904)
D-Gluconic acid, 2,3,4,5,6-Pentahydroxyhexanoic acid; HO.CH2(CHOH)4.COOH
  -----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     gl oth/un 25°C 0.10M U
                                  1968L0b (49738) 73
Nb(V)
                        K(Nb(OH)n+L)=2.78
                         K(Nb(OH)nH-1L+H)=7.82
*******************************
                         CAS 1885-81-0 (4433)
1-Chloro-4-cyanobenzene; Cl.C6H4.CN
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Nb(V) nmr non-aq -60°C 100% U M
                                  1971MBa (52380) 74
                        K(NbC15A+L=NbC15L+A)=1.91
A=cyanomethyl bromide. When A=cyanomethyl iodide, K=1.16
*********************************
                 Cyanobenzene CAS 100-47-0 (4406)
Cyanobenzene, benzonitrile; C6H5.CN
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Nb(V) nmr non-aq -60°C 100% U M
                                  1971MBa (52569) 75
                       K(NbC15A+L=NbC15L+A)=0.04
Medium: CHCl3. A=dimethyl ether
*********************
             H2L Salicylic acid CAS 69-72-7 (14)
2-Hydroxybenzoic acid, Salicylic acid; HO.C6H4.COOH
-----
                                  Reference ExptNo
      Mtd Medium Temp Conc Cal Flags Lg K values
Nb(V) sp oth/un 25°C 0.04M U
                                  1962BVa (54266) 76
                        K(NbO+2L)=22.60
******************************
             H3L Protocatechuic CAS 99-50-3 (875)
3,4-Dihydroxybenzoic acid; C6H3(OH)2.COOH
-----
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V) sp oth/un 20°C 0.10M U
                                  1964SHa (54685) 77
                         K(NbO2+H3L=NbO2LH2+H)=2.3
                         K(NbO2LH2+H3L=NbOL2H2+H)=1.3
                        K(NbO2+3HL+4H)=63.1
*******************************
                           CAS 89-73-6 (204)
2-Hydroxybenzohydroxamic acid (salicylhydroxamic acid); HO.C6H4.CO.NHOH
______
Metal
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
```

```
Nb(V) sp oth/un 20°C dil U M
                                   1972LVa (55604) 78
                       K(NbOC15+2H3L)=4.58
*******************************
            H2L
                            CAS 5263-74-1 (2738)
C9H6N2O5S
7-Nitroso-8-hydroxyquinoline-5-sulfonic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) gl alc/w 27°C 50% C H K1=6.15 B2=11.53 1986EAa (63877)
*****************************
             H2L TAR
C9H7N3O2S
                            CAS 2246-46-0 (707)
4-(2'-Thiazolylazo)-resorcinol; C3H2NS.N:N.C6H3(OH)2
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V) sp alc/w 25°C 50% U
                                   1967NPb (64716) 80
                         K(NbO3+H2L)=9.5(?)
Medium: 50% MeOH, 0.1 M NaClO4
********************************
             H4L Chromotropic ac CAS 148-25-4 (1875)
1,8-Dihydroxynaphthalene-3,6-disulfonic acid;
-----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Nb(V) sp NaCl 20°C 0.10M U I
                                   1964SHa (69962) 81
                         K(NbO2+3L+4H)=64.7
In 3 M NaClO4: K(NbO2+2H+2L)=42.5
**********************************
        H3L
C10H12O5
                            CAS 121-79-9 (3895)
3,4,5-Trihydroxybenzoic acid propyl ester; (HO)3.C6H2.CO.O.CH2.CH2.CH3
 ______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V) sp mixed 22°C 5% U
                                   1968LSc (71685) 82
                         K(?)=3.48
Medium: 5% 1-PrOH, carbonate buffer
********************************
C10H16N2O8
             H4L
                 EDTA
                            CAS 60-00-4 (120)
1,2-Diaminoethane-N,N,N',N'-tetraethanoic acid, Sequestric acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb(V) dis KNO3 20°C 2.0M U
                                   1969KKf (73986) 83
                        K(NbO2+H3L)=10.54
Nb(V) vlt KCl ? 0.40M U K1=39.4 1969SVd (73987) 84
Nb(V) vlt oth/un 20°C 1.0M U
                                   1967VSd (73988) 85
                         K(Nb(OH)2+L)=40.78
```

```
***********************************
C11H9N302
             H2L
                  PAR
                            CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2
    -----
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp oth/un 25°C ? U
                                   1967ADa (77564) 86
Nb(V)
                         K(?)=4.3
C13H11N02
                            CAS 304-88-1 (181)
N-Phenylbenzohydroxamic acid; C6H5.CO.N(C6H5).OH
 .-----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      dis KCl 20°C 1.0M U
Nb(V)
                                    1971LFc (85167) 87
                         B((Nb0)L3)=53.1
************************
             H3L
                  DASA
                            CAS 83-61-4 (950)
1,2-Dihydroxyanthraquinone-3-sulfonic acid, Alizarin Red S;
  -----
      Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
Nb(V)
      sp oth/un ? ? U
                                    1968ADa (86745) 88
                         B((NbO)L2)=8.33
***********************************
C14H15N4OBr
                            CAS 14337-50-9 (5095)
5-(5-Bromo-2-pyridylazo)-2-ethylamino-4-hydroxy-1-methylbenzene;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
     sp oth/un ? ? U
                                    1969EMa (87766) 89
Nb(V)
                       Μ
                         K(NbOA+L)=20.63
H2A=tartaric acid.
*******************************
C14H16N40
              HL
                  PAAC
                            CAS 13059-69-3 (5067)
5-Ethylamino-4-methyl-2-(2'-pyridylazo)phenol;
   Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
-----
      sp oth/un ?
                  ? U
                                    1969EMa (88019) 90
Nb(V)
                         K(NbOA+L)=21.0
H2A=tartaric acid
********************************
             H4L
C14H22N2O8
                  CDTA
                            CAS 482-54-2 (200)
trans-1,2-Diaminocyclohexane-N,N,N',N'-tetraethanoic acid;
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
Metal Mtd Medium Temp Conc Cal Flags Lg K values
      vlt oth/un 25°C 2.0M U
                                    1970PLa (88728) 91
Nb(V)
                         K(?)=15.60 pH 5
```

```
***********************************
C15H13N02
             HL
                          CAS 7369-44-0 (4066)
N-3-Diphenylpropenohydroxamic acid;
  -----
   Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      dis KCl
            20°C 1.0M U
                                 1971LFc (91639) 92
Nb(V)
                       K(NbO+3L=59.7
******************************
C15H18N40
                          CAS 14337-52-1 (5124)
5-Diethylamino-2-(2-pyridylazo)phenol;
-----
     Mtd Medium Temp Conc Cal Flags Lg K values
                                   Reference ExptNo
Nb(V)
     sp oth/un ? ? U
                      М
                                 1969EMa (92097) 93
                        K(NbOA+L)=20.63
H2A=tartaric acid
********************************
                          CAS 3567-23-5 (5202)
5-Chloro-2-hydroxy-3-(2-hydroxy-1-naphthylazo)-benzenesulfonic acid;
______
     Mtd Medium Temp Conc Cal Flags Lg K values
                                  Reference ExptNo
___________
Nb(V) EMF oth/un ? ? U
                     Μ
                                 1971ENc (92770) 94
                        K(NbOA+L)=27.5
H2A=tartaric acid
*******************************
            H4L
                            (5174)
C16H11N3O10S2
2-Hydroxy-1-(2'-hydroxy-4'-nitro)phenylazo-3,6-disulfonaphthalene;
______
     Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
     sp oth/un 25°C ? U
                                 1971RCd (92882) 95
                       K(?)=5.51
********************************
C17H17NO3
                          CAS 58434-59-6 (1213)
2'-Hydroxy-4-methoxy-5'-methylbenzylidene acetophenone oxime
-----
                                Reference ExptNo
      Mtd Medium Temp Conc Cal Flags Lg K values
______
     sp oth/un 30°C 8.00M U
                                 1980GKa (96190) 96
Nb(V)
                        K(NbO(SCN)+L)=2.67
                        K(NbO(SCN)L+L)=2.27
****************************
C17H21N50
                           (5223)
3-Amino-1-hydroxy-6-(2-N-methylanabasinyl-alpha-azo)benzene;
      Mtd Medium Temp Conc Cal Flags Lg K values
-----
      sp oth/un ? ? U
                                 1967TTa (96389) 97
Nb(V)
```

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K(?)=11.36
```

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*********************************
                    Xylenol orange CAS 63721-85-5 (432)
               H6L
5,5'-Bis-N,N-bis(carboxymethyl)aminomethyl-4'-hydroxy-3,3'-dimethylfuchsone-2"-sulf
onic acid;
            ______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
      sp oth/un 25°C ? U
                                       1967ADc (105482) 98
                           K(?)=4.7
********************************
                L Methyl alcohol CAS 67-56-1 (597)
CH40
Methanol; CH3.OH
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb++++ EMF alc/w 20°C 100% U
                                        1971GSa (17896) 99
                            K(Nb(L')2+L')=12.6
                            K(2Nb(L')3+3L'=Nb2(L')9)=23.9
Medium: MeOH, 1 M LiCl
**********************************
              H2L PAR
C11H9N302
                                CAS 1141-59-9 (636)
4-(2'-Pyridylazo)-1,3-dihydroxybenzene; C5H4N.N:N.C6H3(OH)2
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb++++ sp oth/un ? ? U M
                                        1969EMa (77565) 100
                            K(NbOA+L)=21.22
H2A=tartaric acid
***********************************
C22H14N4O16Cl2S4 H8L ClSulfophenol S CAS 2103-73-3 (4156)
2,7-Bis(5'-chloro-2'-hydroxy-3'-sulfo-phenylazo)chromotropic acid;
______
Metal Mtd Medium Temp Conc Cal Flags Lg K values Reference ExptNo
______
Nb++++ sp KNO3 20°C 0.20M U
                                        1965BSe (101484) 101
                           B(NbO2+6H+L)=53.0
Metal: Nb(III)
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EXPLANATORY NOTES

DATA Flags are :-

- T Data at other TEMPERATURES
- I Data with various BACKGROUNDS
- H Data for THERMOCHEMICAL quantities
- M Data for TERNARY Complexes

END