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Zinc

Reaction	Baes and Mesmer, 1976	Powell and Brown, 2013	Brown and Ekberg, 2016
$Zn^{2+} + H_2O \rightleftharpoons ZnOH^+ + H^+$	-8.96 ± 0.05	-8.96 ± 0.05	-8.94 ± 0.06
$Zn^{2+} + 2 H_2O \rightleftharpoons Zn(OH)_2 + 2 H^+$	-16.9	-17.82 ± 0.08	-17.89 ± 0.15
$Zn^{2+} + 3 H_2O \rightleftharpoons Zn(OH)_3^- + 3 H^+$	-28.4 ± 0.2	-28.05 ± 0.05	-27.98 ± 0.10
$Zn^{2+} + 4 H_2O \rightleftharpoons Zn(OH)_4^{2-} + 4 H^+$	-41.2 ± 0.1	-40.41 ± 0.12	-40.35 ± 0.22
$2 \operatorname{Zn}^{2+} + \operatorname{H}_2\operatorname{O} \rightleftharpoons \operatorname{Zn}_2\operatorname{OH}^{3+} + \operatorname{H}^+$	-9.0 ± 0.1	-7.9 ± 0.2	-7.89 ± 0.31
$2 \operatorname{Zn^{2+}} + 6 \operatorname{H}_2 O \rightleftharpoons \operatorname{Zn_2}(OH)_6^{2-} + 6 \operatorname{H}^+$	-57.8 ± 0.1		
$ZnO(s) + 2 H^+ \rightleftharpoons Zn^{2+} + H_2O$	11.14 ± 0.03	11.12 ± 0.05	11.11 ± 0.10
ε -Zn(OH) ₂ (s) + 2 H ⁺ \rightleftharpoons Zn ²⁺ + 2 H ₂ O	11.50	11.38 ± 0.20	11.38± 0.20
β_1 -Zn(OH) ₂ (s) + 2 H ⁺ \rightleftharpoons Zn ²⁺ + 2 H ₂ O	11.73	11.72 ± 0.04	

11.77	11.76 ± 0.04	
11.71	11.70 ± 0.04	
11.82	11.81 ± 0.04	
1 :	1.71	1.71 11.70 ± 0.04

C.F. Baes and R.E. Mesmer, The Hydrolysis of Cations. Wiley, New York, 1976, pp. 292–293

P.L. Brown and C. Ekberg, Hydrolysis of Metal Ions. Wiley, 2016, pp. 676–700.

K. J. Powell, P. L. Brown, R. H. Byrne, T. Gajda, G. Hefter, A.-K. Leuz, S. Sjöberg, and H. Wanner, Pure and Applied Chemistry, 85, 2249–2311 (2013).