

Quantum Mechanics

- 1) Rev. of de-Broglie's hypothesis.
- 2) Rev. of Prop. of matter waves.

Wave function ψ :

(1) A quantity ψ represents de-Broglie's waves as electric field vector represents light waves.

(2) The quantity ψ is called wavefn $\psi(x, y, z, t)$ is a fⁿ of space and time.

(3) ψ represents position of particle, However it is not possible to locate position of particle exactly. There is only a prob. of finding particle

(4) ψ is a complex quantity

$$\psi = A + iB \quad \text{and} \quad \psi^* = A - iB$$

$$|\psi|^2 = \psi \psi^* = A^2 + B^2 \Rightarrow \text{prob. of finding}$$

particle.

(5) The prob. of finding particle in vol. dv $= dx dy dz$ is given by $|\psi|^2 dx dy dz$

Requirements of acceptable wave fn

- (1) ψ must be finite
- (2) ψ must be single valued
- (3) ψ and its first derivative must be continuous.

①

① Limitations of QM CM

② Scope of QM

③ De-broglie's hypo.

④ Davisson Germer expt.

⑤ Uncertainty principle

⑥ v_g / v_p ; Matter waves, wave function, probability, Amplitude Normalization

⑦ Uncer

⑦ Schrodinger's eqⁿ TDSE \rightarrow

TISE.

Particle in a box problem (1D infinite potential well - bc's

⑧ Energy. momentum quantization
3D Eqⁿ + Degeneracy

Time dependent Schrodinger equation:

The 1D wave eqⁿ is

$$\frac{\partial^2 y}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 y}{\partial t^2}$$

and solⁿ of this eqⁿ is

$$y(x,t) = A e^{i(kx - \omega t)}$$

• 11/4 For matter waves

$$\frac{\partial^2 \psi}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 \psi}{\partial t^2} \rightarrow \textcircled{1}$$

and solⁿ $\psi(x,t) = A e^{i(kx - \omega t)}$

$$k = \frac{2\pi}{\lambda} = \frac{2\pi p}{h} = \frac{p}{\hbar} \rightarrow \textcircled{2}$$

$$\omega = 2\pi\nu = \frac{2\pi E}{h} = \frac{E}{\hbar}$$

$$\psi(x,t) = A e^{i\left[\frac{p}{\hbar}x - \frac{E}{\hbar}t\right]}$$

$$\psi(x,t) = A e^{-\frac{i}{\hbar}[Et - px]} \rightarrow \textcircled{3}$$

$$TE = KE + PE$$

$$E = \frac{p^2}{2m} + V$$

$$E\psi = \frac{p^2}{2m}\psi + V\psi \rightarrow \textcircled{4}$$

$$\psi = A e^{-i/h [Et - px]} \rightarrow \textcircled{2}$$

Diffⁿ eqⁿ $\textcircled{3}$ w.r.t x , we get

$$\frac{\partial \psi}{\partial x} = \left[A e^{-i/h [Et - px]} \right] \left[(-i/h) (-p) \right]$$

$$\frac{\partial^2 \psi}{\partial x^2} = \left[A e^{-i/h [Et - px]} \right] \left[(-i/h) (-p) (-i/h) (-p) \right]$$

$\underbrace{\hspace{10em}}_{\psi} \qquad \qquad \qquad \frac{i^2 p^2}{h^2} = -\frac{p^2}{h^2}$

$$\frac{\partial^2 \psi}{\partial x^2} = -\frac{p^2}{h^2} \psi$$

$$p^2 \psi = -h^2 \frac{\partial^2 \psi}{\partial x^2} \rightarrow \textcircled{5}$$

Diffⁿ eqⁿ $\textcircled{3}$ w.r.t t , we get

$$\frac{\partial \psi}{\partial t} = \left[A e^{-i/h (Et - px)} \right] \left[(-i/h) (E) \right] \frac{i}{2}$$

$\underbrace{\hspace{10em}}_{\psi}$

$$\frac{\partial \psi}{\partial t} = \frac{E \psi}{i h}$$

$$E \psi = i h \frac{\partial \psi}{\partial t} \rightarrow \textcircled{6}$$

From eqⁿ $\textcircled{4}$ $\textcircled{5}$ & $\textcircled{6}$

$$i h \frac{\partial \psi}{\partial t} = -\frac{h^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V \psi$$

$$\text{or, } \left[-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi = i\hbar \frac{\partial \psi}{\partial t} \right]$$

Reduction to TISE

$$\text{TDSE is } -\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi = i\hbar \frac{\partial \psi}{\partial t}$$

$$\therefore \psi(x, t) = \psi(x) \phi(t)$$

$$\therefore -\frac{\hbar^2}{2m} \phi(t) \frac{d^2 \psi(x)}{dx^2} + V \psi(x) \phi(t) = i\hbar \psi(x) \frac{d\phi(t)}{dt}$$

dividing by $\psi(x) \phi(t)$, we get

$$\underbrace{-\frac{\hbar^2}{2m} \frac{1}{\psi(x)} \frac{d^2 \psi(x)}{dx^2}}_{\text{fn of } x} + V = \underbrace{i\hbar \frac{1}{\phi(t)} \frac{d\phi(t)}{dt}}_{\text{fn of } t}$$

correct only if LHS and RHS are equal to constant. let $RHS = E (\text{Constant})$

$$\therefore -\frac{\hbar^2}{2m} \frac{1}{\psi(x)} \frac{d^2 \psi(x)}{dx^2} + V = E$$

$$\left[-\frac{\hbar^2}{2m} \frac{d^2 \psi(x)}{dx^2} + V\psi(x) = E\psi(x) \right] \Rightarrow -\frac{\hbar^2}{2m} \nabla^2 \psi + V\psi = E\psi$$

$$\therefore H = -\frac{\hbar^2}{2m} \nabla^2 + V$$

$$\boxed{H\psi = E\psi}$$

free particle:

For free particle, $V=0$

Consider an electron freely propagating along x-axis and not acted by any force.

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \psi = E\psi$$

$\therefore V=0$

$$-\left(\frac{\hbar^2}{2m}\right) \frac{d^2\psi}{dx^2} = E\psi$$

$$\frac{d^2\psi}{dx^2} + \left(\frac{2mE}{\hbar^2}\right) \psi = 0$$

$$k^2 = \frac{2mE}{\hbar^2} \rightarrow (1)$$

$$\frac{d^2\psi}{dx^2} + k^2 \psi = 0 \rightarrow (2)$$

The general solⁿ is

$$\psi = A e^{ikx} + B e^{-ikx} \rightarrow (3)$$

$A, B \Rightarrow$ constant

\therefore wave is propagating only along x-axis

we can write

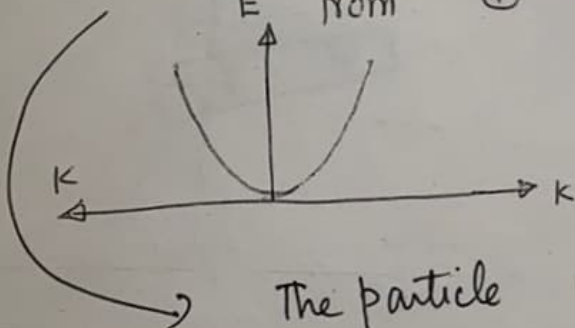
$$\psi = A e^{i(kx - \omega t)} \rightarrow (4)$$

As particle is free, no boundary condition

can be applied i.e. no restriction on k

E from (1)

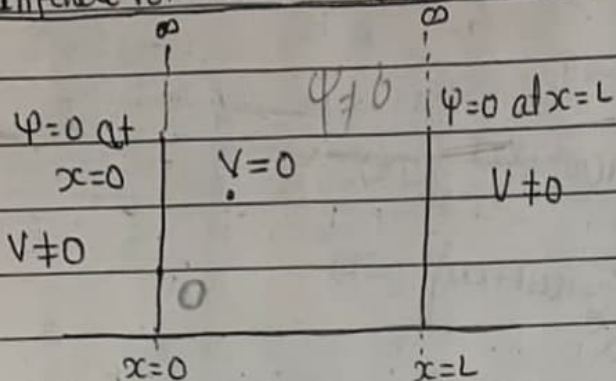
$$E = \frac{\hbar^2 k^2}{2m} = \frac{\hbar^2 k^2}{8m\pi^2}$$
$$E \propto k^2$$



The particle is permitted to have any values of E

Energy is not quantized i.e. a freely moving particle possesses a continuous energy spectrum.

Infinite Potential well (Particle in box)



TISE,
$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V\psi = E\psi \quad \text{--- (1)}$$

$V = 0$ when $0 < x < L$

$V \neq 0$ when $x \leq 0$ and $x \geq L$

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi$$

$$\frac{d^2\psi}{dx^2} + \frac{2mE\psi}{\hbar^2} = 0$$

$$\hbar = \left(\frac{h}{2\pi}\right)$$

$$\frac{d^2\psi}{dx^2} + \frac{2m}{\hbar^2} E\psi = 0$$

$x=0; \psi(x)=0$
 $x=L; \psi(x)=0$

(9)

$$\frac{d^2\psi}{dx^2} + \left(\frac{8m\pi^2 E}{h^2}\right) \psi = 0 \rightarrow (2)$$

$$\text{let } k^2 = \frac{8\pi^2 m E}{h^2}$$

$$\frac{d^2\psi}{dx^2} + k^2 \psi = 0 \rightarrow (3)$$

The solⁿ is

$$\psi = Ae^{ikx} + Be^{-ikx} \rightarrow (4)$$

When $x=0$, $\psi=0$

$$0 = A + B$$

$$\therefore B = -A$$

From (4)

$$\psi = A \left(e^{ikx} - e^{-ikx} \right) \frac{2i}{2i}$$

$$\therefore \text{At } 0 = \left(\frac{e^{i0} - e^{-i0}}{2i} \right)$$

$$\psi = (2iA) \sin kx \rightarrow \textcircled{5}$$

when $x = L$, $\psi = 0$

$$0 = 2iA \sin kL, \quad kL = 0; kL = n\pi$$

$$\Rightarrow kL = n\pi$$

$$k = \left(\frac{n\pi}{L}\right) \rightarrow \textcircled{6}$$

From $\textcircled{5}$ & $\textcircled{6}$,

$$\psi = (2iA) \sin \left(\frac{n\pi x}{L}\right)$$

$$\text{or, } \psi_n = (2iA) \sin \left(\frac{n\pi x}{L}\right) \rightarrow \textcircled{7}$$

From normalized condⁿ

$$\int_0^L |\psi_n|^2 dx = 1$$

$$\int_0^L (2iA)^2 \sin^2 \left(\frac{n\pi x}{L}\right) dx = 1$$

$$\sin 0 = \frac{1 - \cos 20}{2}$$

(11)

$$\therefore \frac{(2iA)^2}{2} \int_0^L \left[1 - \frac{\cos 2n\pi x}{L} \right] dx = 1$$

$$\frac{(2iA)^2}{2} \left[L - \left(\frac{\sin 2n\pi x / L}{2n\pi / L} \right) \right]_0^L = 1$$

$$\frac{(2iA)^2}{2} \left[L - \left(\frac{\sin 2n\pi x}{2n\pi / L} \right) - 0 \right] = 1$$

$$\downarrow$$

0

$$(2iA)^2 \left(\frac{L}{2} \right) = 1$$

$$2iA = \sqrt{\left(\frac{2}{L} \right)} \rightarrow (8)$$

From eqn (7) & (8)

$$\psi_n = \sqrt{\left(\frac{2}{L} \right)} \sin \left(\frac{n\pi x}{L} \right) \rightarrow (9)$$

$$|\Psi_n|^2 = \left(\frac{2}{L}\right) \sin^2\left(\frac{n\pi x}{L}\right) \rightarrow (10)$$

also $\therefore k = \frac{n\pi}{L}$

$$k^2 = \frac{n^2 \pi^2}{L^2}$$

$$\therefore k^2 = \frac{8\pi^2 m E}{h^2}$$

$$\frac{8\pi^2 m E}{h^2} = \frac{n^2 \pi^2}{L^2}$$

$$E = \left(\frac{n^2 h^2}{8mL^2} \right)$$

$$2mE = k^2 \hbar^2$$

$$E = \frac{\hbar^2 k^2}{2m}$$

$$= \frac{\hbar^2}{2m} \cdot \frac{n^2 \pi^2}{L^2}$$

$$E = \frac{n^2 h^2}{8mL^2}$$

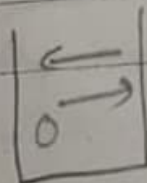
$$E_n = \left(\frac{n^2 h^2}{8mL^2} \right) \rightarrow (11)$$

\therefore Energy is quantized and depend on (n) & (L) .

$n=1$ \Rightarrow Zero point energy

$n=2$: First Exc. state

$n=3$ Second "



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$N =$

$n=3$

E_3

$n=2$

E_2

$n=1$

E_1

ψ_n

$n=2$

$n=1$

$x=0$

$x=L$

$\psi(x)$

$|\psi_n|^2$

ψ_2

ψ_1

$x=0$

$x=L$

$|\psi(x)|^2$

$n=2$

$n=1$

① for $n=1$ $\therefore \psi_n = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$

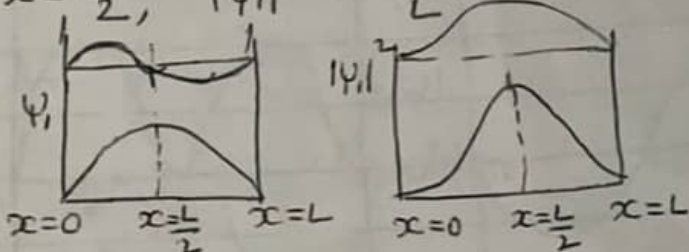
$$\psi_1 = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi x}{L}\right)$$

at $x=0$ and $x=L$, $\psi=0$

$$x = \frac{L}{2} \quad \psi_1 = \sqrt{\frac{2}{L}} \Rightarrow \text{max}$$

$|\psi_1|^2$ is zero at $x=0$ and $x=L$

$$\text{at } x = \frac{L}{2}, |\psi_1|^2 = \frac{2}{L} \Rightarrow \text{max}$$

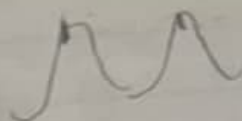


② for $n=2$ $\psi_2 = \left(\sqrt{\frac{2}{L}}\right) \sin\left(\frac{2\pi x}{L}\right)$

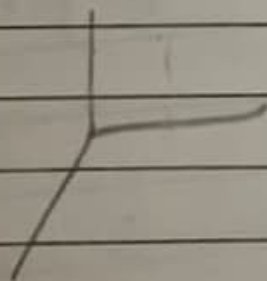
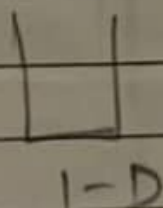
$$\text{for } x = \frac{L}{4} \text{ and } x = \frac{3L}{4}$$

ψ_2 is max

$|\psi_2|^2$ is also max



- ① Energy is quantized and depend on principal quantum no. n and width (L) .
- ② Energy cannot be zero
 if $E=0$
 $\Rightarrow p=0$
 $\lambda=\infty$
 \therefore particle cannot trapped in box.
- ③ ψ_n can be -ve or +ve but $|\psi|^2$ is only +ve
- ④ for $n=1$ Max probability at centre.



Thus first excited state is three-fold degenerate

Eigen values and Eigen function
for particle in box.

$$E_n = \frac{n^2 h^2}{8mL^2}$$

$$\text{and } \psi_n = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$$

The possible values of E
are called eigen values and the
corresponding ψ are
called eigen function.

Degeneracy:

For different combination of quantum numbers we may have same energy value but wavefn are different.

Such quantum states having same energy are called degenerate.

for example: $a=b=c \Rightarrow (a)(b)(c) = a^3$

$$\psi_{112} = \sqrt{\frac{8}{abc}} \sin \frac{\pi x}{a} \sin \frac{\pi y}{b} \sin \frac{2\pi z}{c}$$

$$\psi_{121} = \sqrt{\frac{8}{abc}} \sin \frac{\pi x}{a} \sin \frac{2\pi y}{b} \sin \frac{\pi z}{c}$$

$$\psi_{211} = \sqrt{\frac{8}{abc}} \sin \frac{2\pi x}{a} \sin \frac{\pi y}{b} \sin \frac{\pi z}{c}$$

are different, but corresponding energies are same.

for first ψ_{112} , $n_x=1$, $n_y=1$, $n_z=2$

$$n^2 = n_x^2 + n_y^2 + n_z^2 = 6$$

$$E_{112} = \frac{6h^2}{8ma^2}$$

$$\text{for other two } E_{121} = E_{211} = \frac{6h^2}{8ma^2}$$

The number of different states with a certain value of energy is known as degree of degeneracy.

Thus first excited state is three-fold degenerate