Table S1: Estimated sub-cooled liquid saturation vapour pressures at 298 K. N_Vp (Nannoolal vapour pressure), MY_Vp (Myrdal and Yalkowsky vapour pressure), EVAPORATION (EVAPORATION vapour pressure), SIMPOL (SIMPOL vapour pressure), N_Tb (Nannoolal boiling point), SB_Tb (Stein and Brown boiling point), Literature (2-nitrophenol, 3-methyl-2-nitrophenol, 4-methyl-2-nitrophenol, 5-fluoro-2-nitrophenol, 4-nitrophenol from (Schwarzenbach et al., 1988), 3-nitrophenol from (Ribeiro da Silva et al., 1992) 2-nitrobenzaldehyde, 3-nitrobenzaldehyde from (Perry et al., 1984), 2-nitrobenzoic acid, 3-nitrobenzoic acid, 4-nitrobenzoic acid from (Ribeiro Da Silva et al., 1999), 4-methyl-3-nitrophenol acid, 3-methyl-4-nitrophenol from (Bannan et al., 2017), 4-methyl-2-nitrophenol, 4-methyl-3-nitrophenol, 3-methyl-4-nitrophenol from (Dang et al., 2019)). This table contains the corresponding information used in Fig. 8.

Compound	N_Vp &	N_Vp &	MY_Vp	MY_Vp	SIMPOL	EVAPORATION	Literature	Literature	Experimental
	N_Tb	SB_Tb	& N_Tb	& SB_Tb	(Pa)	(Pa)	(Pa)	(previous	(Pa)
	(Pa)	(Pa)	(Pa)	(Pa)				KEMS	
								studies)	
								(Pa)	
2-nitrophenol	1.37E+03	1.80E+02	1.13E+03	1.22E+02	3.58E+01	1.37E+04	2.69E+01	nan	1.38E-03
3-methyl-2-									
nitrophenol	3.58E+02	5.53E+01	3.25E+02	4.49E+01	1.35E+01	4.47E+03	1.12E+01	nan	1.22E-02
4-methyl-2-									
nitrophenol	3.58E+02	5.53E+01	3.25E+02	4.49E+01	1.35E+01	4.47E+03	9.63E+00	5.97E-04	3.29E-03
5-fluoro-2-									
nitrophenol	1.03E+03	1.98E+02	9.67E+02	1.68E+02	3.58E+01	1.37E+04	3.05E+00	nan	5.01E-03
4-amino-2-									
nitrophenol	6.70E+00	2.36E+00	6.79E+00	2.36E+00	9.35E-01	1.37E+04	nan	nan	9.29E-02
3-nitrophenol	1.11E+03	1.70E+02	9.37E+02	1.22E+02	3.58E+01	1.37E+04	9.40E-02	1.14E-02	nan
4-methyl-3-									
nitrophenol	3.58E+02	5.53E+01	3.25E+02	4.49E+01	1.35E+01	4.47E+03	nan	4.85E-03	6.85E-02
4-chloro-3-									
nitrophenol	2.09E+02	3.22E+01	1.87E+02	2.58E+01	3.58E+01	1.37E+04	nan	nan	5.80E-02
4-nitrophenol	9.60E+02	1.69E+02	8.04E+02	1.22E+02	3.58E+01	1.37E+04	1.95E-01	5.14E-03	nan
3-methyl-4-									
nitrophenol	3.58E+02	5.53E+01	3.25E+02	4.49E+01	1.35E+01	4.47E+03	nan	3.78E-03	5.86E-02
2-fluoro-4-	1.03E+03	1.98E+02	9.67E+02	1.68E+02	3.58E+01	1.37E+04	nan	nan	6.42E-02

1

nitrophenol									
3-fluoro-4-									
nitrophenol	1.03E+03	1.98E+02	9.67E+02	1.68E+02	3.58E+01	1.37E+04	nan	nan	3.32E-02
2-									
nitrobenzaldehyde	8.33E+02	1.56E+02	7.94E+02	1.33E+02	6.08E-01	2.86E+02	2.39E+00	nan	2.15E+00
3-									
nitrobenzaldehyde	6.80E+02	1.46E+02	6.73E+02	1.33E+02	6.08E-01	2.86E+02	9.70E-01	nan	2.75E-01
2-chloro-5-									
nitrobenzaldehyde	1.46E+02	3.04E+01	1.45E+02	2.83E+01	6.08E-01	2.86E+02	nan	nan	8.41E-02
4-									
nitrobenzaldehyde	5.95E+02	1.45E+02	5.87E+02	1.33E+02	6.08E-01	2.86E+02	nan	nan	1.93E-01
2-nitrobenzoic									
acid	1.12E+01	9.32E-01	2.28E+01	2.18E+00	3.91E-03	1.09E+00	4.44E-03	nan	nan
5-chloro-2-									
nitrobenzoic acid	1.42E+00	1.50E-01	4.48E+00	5.85E-01	3.91E-03	1.09E+00	nan	nan	1.40E-02
3-nitrobenzoic									
acid	8.59E+00	8.48E-01	1.90E+01	2.18E+00	3.91E-03	1.09E+00	5.05E-03	nan	1.90E-03
4-methyl-3-									
nitrobenzoic acid	2.56E+00	2.52E-01	7.63E+00	9.26E-01	1.47E-03	3.57E-01	3.07E-03	nan	5.76E-02
2-chloro-3-									
nitrobenzoic acid	1.42E+00	1.50E-01	4.48E+00	5.85E-01	3.91E-03	1.09E+00	nan	nan	6.29E-03
2-hydroxy-5-									
nitrobenzoic acid	1.15E+00	1.35E-02	5.50E+00	1.18E-01	4.16E-03	1.09E+00	nan	nan	1.87E-02
4-nitrobenzoic									
acid	7.28E+00	8.42E-01	1.64E+01	2.18E+00	3.91E-03	1.09E+00	1.94E-02	nan	nan
3-methyl-4-									
nitrobenzoic acid	2.56E+00	2.52E-01	7.63E+00	9.26E-01	1.47E-03	3.57E-01	7.71E-03	nan	3.04E-01

References

20

1992.

Bannan, T. J., Booth, A. M., Jones, B. T., O'meara, S., Barley, M. H., Riipinen, I., Percival, C. J. and Topping, D.: Measured Saturation Vapor Pressures of Phenolic and Nitro-aromatic Compounds, Environ. Sci. Technol, 51(7), 3922–3928, doi:10.1021/acs.est.6b06364, 2017.

Dang, C., Bannan, T., Shelley, P., Priestley, M., Worrall, S. D., Waters, J., Coe, H., Percival, C. J. and Topping, D.: The effect of structure and isomerism on the vapour pressures of organic molecules and its potential atmospheric relevance, Aerosol Sci. Technol., 1–32, doi:10.1080/02786826.2019.1628177, 2019.

Monte, M. J. S., Hillesheim, D. M., Monte, M. J. S. and Hillesheim, D. M.: Thermodynamic study on the sublimation of six methylnitrobenzoic

- acids, J. Chem. Thermodyn., 33, 103–112, doi:10.1006/jcht.2000.0729, 2001.

 Perry, R. H., Green, D. W. and Maloney, J. O.: Perry's Chemical engineers' handbook., 6th ed., McGraw-Hill., 1984.

 Ribeiro da Silva, M. A. V., Reis, A. M. M. V., Monte, M. J. S., Bártolo, M. M. S. S. F. and Rodrigues, J. A. R. G. O.: Enthalpy of combustion, vapour pressures, and enthalpy of sublimation of 3-nitrophenol, J. Chem. Thermodyn., 24(6), 653–659, doi:10.1016/S0021-9614(05)80037-0,
 - Ribeiro Da Silva, M. A. V, Agostinha, M., Matos, R., Monte, M. J. S., Hillesheim, D. M., Marques, M. C. P. O. and Vieira, N. F. T. G.: Enthalpies for combustion, vapour pressures, and enthalpies of sublimation of three methoxy-nitrobenzoic acids. Vapour pressures and enthalpies of sublimation of the three nitrobenzoic acids, J. Chem. Thermodyn., 31, 1429–1441, 1999.
- Schwarzenbach, R. P., Stierli, R., Folsom, B. R. and Zeyer, J.: Compound Properties Relevant for Assessing the Environmental Partitioning of Nitrophenols, Environ. Sci. Technol, 22, 83–92 [online] Available from: https://pubs.acs.org/sharingguidelines (Accessed 9 August 2018), 1988.