Round 12

1. CHEMISTRY

Toss Up: Multiple Choice

Which of the following factors will contribute to a decrease in oxygen in a pond?

- W) decreasing salinity
- X) increasing acidity
- Y) increasing temperature
- Z) increasing surface tension of the water

Toss Up Answer: Y

Bonus: Multiple Choice

When KCl dissolves in water, the solution cools noticeably to the touch. It may be concluded that

- W) the entropy increase overcomes the unfavorable heat of dissolution
- X) KCl is relatively insoluble in water
- Y) the entropy decreases when KCI dissolves
- Z) the boiling point of the solution will be less than 100 degrees Celsius

Bonus Answer: W

2. BIOLOGY

Toss Up: Multiple Choice

Which mollusk does not possess a radula?

W) Squid

- X) Scallop
- Y) Snail
- Z) Chiton

Toss Up Answer: X

Bonus: Short Answer

What is the scientific name for the job or role an organism plays in its habitat?

Bonus Answer: Niche

3. MATHEMATICS

Toss Up: Multiple Choice

If θ is an angle such that $\sin(\theta) < 0$ and $\cos(\theta) = 0$, where in the coordinate plane is it located?

- W) Between the 2nd and 3rd quadrants
- X) Between the 3rd and 4th quadrants
- Y) Between the 1st and 4th quadrants
- Z) Between the 1st and 2nd quadrants

Toss Up Answer: Y

Bonus: Short Answer

What is the remainder of x^10+x+1 divided by $(x-1)^2$?

Bonus Answer: 11x-8

4. ENERGY

Toss Up: Short Answer

Which compound contributes the most to global warming?

Bonus Answer: water

Bonus: Short Answer

What is China's largest renewable energy source?

Bonus Answer: Hydro

5. PHYSICS

Toss Up: Multiple Choice

An acceptor replacement atom in silicon might have electrons in its outer shell

W) 3

X) 4

Y) 5

Z) 6

Toss Up Answer: W

Bonus: Multiple Choice

When a forward bias is applied to a p-n junction the concentration of electrons on the p side:

W) increases slightly

- X) increases dramatically
- Y) decreases slightly
- Z) decreases dramatically

Bonus Answer: X

6. EARTH and SPACE

Toss Up: Short Answer

What mountain has the highest point in the Western Hemisphere?

Bonus Answer: Mt. Aconcagua

Bonus: Multiple Choice

Which of the following describes how sea ice is different from continental ice?

- W) Continental ice is denser than sea ice
- X) Sea ice is saline while continental ice is fresh
- Y) Sea ice floats on the water while continental ice sits on land
- Z) Sea ice is constantly changing while continental ice remains the same

Bonus Answer: Y

7. EARTH and SPACE

Toss Up: Short Answer

What is the second longest Peninsula in the world?

Bonus Answer: Baja California

Bonus: Short Answer

What is a system or group of parallel mountain ranges together with the intervening plateaus and other features?

Bonus Answer: Cordillera

8. BIOLOGY

Toss Up: Multiple Choice

What branching nerve fibers receive messages?

- W) Axons
- X) Axes
- Y) Dendrons
- Z) Dendrites

Toss Up Answer: Z

Bonus: Multiple Choice

What protein was first isolated in red blood cells?

W) Myoglobin

- X) Hemocyanin
- Y) Spectrin
- Z) Collagen

Bonus Answer: Y

9. EARTH and SPACE

Toss Up: Short Answer

Thousands of rivers drain the continental United States. These rivers primarily drain into which three bodies of water?

Bonus Answer: GULF OF MEXICO, ATLANTIC OCEAN (ACCEPT: GULF OF ST. LAWRENCE), PACIFIC OCEAN

(ACCEPT: SEA OF CORTEZ, GULF OF CALIFORNIA)

Bonus: Short Answer

What is the longest Peninsula in the world?

Bonus Answer: Malay Peninsula

10. BIOLOGY

Toss Up: Short Answer

Give the genus and species name of the organism that Beadle and Tatum worked with.

Bonus Answer: Neurospora crassa

Bonus: Short Answer

Archibald Garrod is credited with the discovery of which disease, characterized by blackened urine?

Bonus Answer: Alkaptonuria

11. BIOLOGY

Toss Up: Short Answer

Name the two types of cell death.

Bonus Answer: Apoptosis, Necrosis

Bonus: Short Answer

In C4 and CAM photosynthesis, which specific enzyme is used to fix CO2 in lieu of RuBisCO?

Bonus Answer: PEP Carboxylase

12. CHEMISTRY

Toss Up: Multiple Choice

Diamond is classified as

W) a covalent crystal

X) an ionic crystal

Y) a molecular crystal

Toss Up Answer: W

Bonus: Multiple Choice

A liquid substance that exhibits low intermolecular attractions is expected to have

- W) low viscosity, low boiling point, and low heat of vaporization
- X) high viscosity, low boiling point, and low heat of vaporization
- Y) low viscosity, high boiling point, and low heat of vaporization
- Z) low viscosity, low boiling point, and high heat of vaporization

Bonus Answer: W

13. PHYSICS

Toss Up: Multiple Choice

Monochromatic light is normally incident on a diffraction grating that is 1cm wide and has 10,000 slits. The first order line is deviated at a 30° angle. What is the wavelength, in nm, of the incident light?

W) 300

X) 500

Y) 877

Z) 1000

Toss Up Answer: X

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Bonus: Multiple Choice

A spectral line of a certain star is observed to be "red shifted" from a wavelength of 500nm to a wavelength of 1500nm. Interpreting this as a Doppler effect, the speed of recession of this star is:

W) .33c

X) .5c

Y) .71c

Z) .8c

Bonus Answer: Z

14. BIOLOGY

Toss Up: Short Answer

Which model is currently accepted as representative of the dynamic nature of the Golgi apparatus?

Bonus Answer: Cisternal Maturation model

Bonus: Short Answer

Which cytoskeletal element functions in cleavage furrow formation?

Bonus Answer: Microfilament

15. PHYSICS

Toss Up: Multiple Choice

The equation of continuity for fluid flow can be derived from the conservation of:

W) energy

X) mass

Y) angular momentum

Toss Up Answer: X

Bonus: Multiple Choice

A coil has a resistance of 60Ω and an impedance of 100Ω . Its reactance, in ohms, is:

W) 40

X) 60

Y) 80

Z) 117

Bonus Answer: Y

16. CHEMISTRY

Toss Up: Multiple Choice

Which element is expected to have the greatest polarizability?

W) Fe

X) Ca

Y) Ne

Z)S

Toss Up Answer: W

Bonus: Multiple Choice

A student observed that a small amount of acetone sprayed on the back of the hand felt very cool compared to a similar amount of water. Your explanation of this phenomena should be that

W) all organic compounds do this

- X) acetone has a lower viscosity and transfers heat quanta better
- Y) water has a higher heat capacity than acetone, therefore retaining more heat
- Z) the higher vapor pressure of acetone results in more rapid evaporation and heat loss

Bonus Answer: Z

17. MATHEMATICS

Toss Up: Short Answer

What is the 5th non-triangular number?

Bonus Answer: 8

Bonus: Short Answer

Name all the following that are true:

- 1. An icosahedron has 18 faces.
- 2. A regular hexahedron has 16 edges.
- 3. There are only nine regular polyhedra.
- 4. A regular octahedron has 4 times the volume of a regular tetrahedron with the same side length.

Bonus Answer: 2,4 (an icosahedron has 20 faces; a cube has 12 edges)

18. CHEMISTRY

Toss Up: Multiple Choice

Compared to ideal gases, real gases tend to have

W) larger volumes

- X) lower average kinetic energies
- Y) lower pressures
- Z) Both (W) and (Y)

Toss Up Answer: Z

Bonus: Multiple Choice

A gas has a density, at STP, of 3.48 g L^-1. The most reasonable formula for this compound is

W) C2H6 X) C6H6

Y) CCI4

Z) CaF2

Bonus Answer: X

19. MATHEMATICS

Toss Up: Short Answer

What is the inverse of the 2x2 matrix (row 1: 6 10), (row 2: 3 5)? Bonus Answer: The matrix has no inverse (b.c. determinant = 0).

Bonus: Short Answer

The legs of an isosceles triangle have a length of 10, and the altitudes to the legs have a length of 6. In simplified radical form, what is the length of the altitude to the base of the triangle?

Bonus Answer: 3 * sqrt(10) (Do not accept sqrt(90))

20. MATHEMATICS

Toss Up: Short Answer

If n is a positive integer, what is the smallest value of n such that n! + 1 is a perfect square?

Bonus Answer: 4

Bonus: Short Answer

If z1 = 3 - 4i and z2 = 7 + i, find the absolute value of z1z2 in simplest terms.

Bonus Answer: 25*sqrt(2)

21. EARTH and SPACE

Toss Up: Multiple Choice

The reason there aren't any active volcanoes in the Himalayan mountain range is because:

- W) The Indian plate is composed of basaltic igneous rock, which only melts at higher temperatures.
- X) It was formed by localized faults in the Earth's upper crust.
- Y) It was formed by the collision of two continental plates with no significant density differences.
- Z) It is composed of extinct volcanoes.

Toss Up Answer: Y

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Bonus: Short Answer

The South Georgia and the South Sandwich Islands were formed by the collision of which two tectonic plates?

Bonus Answer: South Sandwich Plate and South American Plate

22. CHEMISTRY

Toss Up: Multiple Choice

Sulfur forms the following compounds: SO2, SCl2, and SO32-. Which form of hybridization is NOT represented by these molecules?

W) sp

X) sp2

Y) sp3

Z) None of these

Toss Up Answer: W

Bonus: Multiple Choice

Which of the following is least related to the strength of a covalent bond?

W) vibrational frequency

X) bond order

Y) bond length

Z) bond direction

Bonus Answer: Z

23. PHYSICS

Toss Up: Short Answer

The Curie temperature is the temperature at which ferromagnets become paramagnets. What is the name for the point at which antiferromagnets become paramagnets?

Bonus Answer: The Neel temperature

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Bonus: Multiple Choice

A hall probe measures which of the following

W) Capcitance

X) Viscosity

Y) Magnetic Field

Z) Electric Field

Bonus Answer: Y

24. EARTH and SPACE

Toss Up: Multiple Choice

In the Bowen Reaction Series, which of the following minerals would solidify last in a mass of cooling magma?

W) Biotite Mica

X) Pyroxene

Y) Olivine

Z) Amphibole

Toss Up Answer: W

Bonus: Short Answer

What is the name given to any fragmental material produced by a volcanic eruption regardless of composition and fragment size?

Bonus Answer: Tephra

25. CHEMISTRY

Toss Up: Multiple Choice

Which of these molecules has the most pi bonds?

W) HCN

X) PF5

Y) NH3

Z) SO3

Toss Up Answer: W

Bonus: Short Answer

What name should be given to a molecule with the formula N2O5?

Bonus Answer: dinitrogen pentoxide