Round 2

1. BIOLOGY

Toss Up: Multiple Choice

Which of the following proteins is involved in forming the cleavage furrow of a cell during cytokinesis?

W) G-coupled proteins

X) Keratin

Y) Myosin

Z) Tubulin

Toss Up Answer: Y

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Bonus: Short Answer

Name the method in which a new species develops due to a population separated from one another by geographical isolation.

Bonus Answer: Allopatric speciation (OR geographic speciation)

2. MATHEMATICS

Toss Up: Short Answer

What is the 5th non-triangular number?

Bonus Answer: 8

Bonus: Short Answer

Name all the following that are true:

- 1. An icosahedron has 18 faces.
- 2. A regular hexahedron has 16 edges.
- 3. There are only nine regular polyhedra.
- 4. A regular octahedron has 4 times the volume of a regular tetrahedron with the same side length.

Bonus Answer: 2,4 (an icosahedron has 20 faces; a cube has 12 edges)

3. PHYSICS

Toss Up: Multiple Choice

The rate of heat flow by conduction through a slab does NOT depend upon the

- W) temperature difference between opposite faces of the slab
- X) thermal conductivity of the slab
- Y) slab thickness
- Z) specific heat of the slab

Toss Up Answer: Z

Bonus: Multiple Choice

Inside a room at a uniform comfortable temperature, metallic objects generally feel cooler to the touch than wooden objects do. This is because:

- W) a given mass of wood contains more heat than the same mass of metal
- X) metal conducts heat better than wood
- Y) the equilibrium temperature of metal in the room is lower than that of wood
- Z) the human body, being organic, resembles wood more closely than it resembles metal

Bonus Answer: X

4. EARTH and SPACE

Toss Up: Short Answer

After reaching what mass can a white dwarf no longer remain stable?

Bonus Answer: The Chandrasekhar Limit

Bonus: Short Answer

What is the pressure that prevents the gravitational collapse of a stable white dwarf?

Bonus Answer: Electron degeneracy pressure

5. MATHEMATICS

Toss Up: Short Answer

Which of the following statements concerning prime numbers are true?

- 1. If a and n are relatively prime than there are infinitely many primes that leave a remainder of a when divided by n.
- 2. All integers less than p are quadratic residues modulo p
- 3. There are infinitely many prime numbers

Bonus Answer: 1 and 3

Bonus: Short Answer

If a1+a2+a3+a4+a5=17 for positive integers a1,a2,a3,a4,a5, then maximize the product a1a2a3a4a5?

Bonus Answer: 432

6. MATHEMATICS

Toss Up: Multiple Choice

A point with coordinates (3,5) is rotated 90 degrees clockwise about the point (2,1) and is then reflected across the y-axis. What are the coordinates of the resulting image?

W) (-5,-3)

X)(-4,0)

Y) (-6,0)

Z) (-5,-1)

Toss Up Answer: Y

Bonus: Multiple Choice

AC=BD=(AB)(sqrt(2)) [AC equals BD equals AB times the square root of 2]

- W) The circumcircle of ABC is centered at the midpoint of AC
- X) The midlines of ABCD are perpendicular
- Y) The perpendicular bisectors of AB and CD are the same line
- Z) None of the above.

Bonus Answer: Z

7. EARTH and SPACE

Toss Up: Short Answer

What hypothetical structure is believed to mark the end of the Sun's Hill Sphere?

Bonus Answer: Oort Cloud

Bonus: Short Answer

The time between two successive meridian transits of the Sun as observed from a stationary spot on the Earth's surface is called

Bonus Answer: A day (a solar day)

8. BIOLOGY

Toss Up: Multiple Choice

Which of the following is a neurotransmitter important for the functioning of myofilaments?

- W) Dopamine
- X) Acetylcholine
- Y) Serotonin
- Z) Acetaminophen

Toss Up Answer: X

Bonus: Multiple Choice

Which of the following is not a layer of the epidermis in thin skin?

- W) Stratum corneum
- X) Stratum lucidum
- Y) Stratum spinosum
- Z) Stratum basale

Bonus Answer: X

9. EARTH and SPACE

Toss Up: Short Answer

What are the three types of asteroids, classified based on composition?

Bonus Answer: C-type (accept carbonaceous), M-type (accept metal-rich), S-type (accept silicate)

(Accept in any order)

Bonus: Short Answer

Order these three types of asteroid from most common to least common in the asteroid belt.

Bonus Answer: C-type (accept carbonaceous), S-type (accept silicate), M-type (accept metal-rich)

About 75% of visible asteroids in the asteroid belt are carbonaceous, about 17% are silicate asteroids, and about 10% are metal-rich asteroids

10. BIOLOGY

Toss Up: Multiple Choice

Which of the following is a ground tissue with thick secondary cells walls reinforced with lignin?

- W) Collenchyma
- X) Parenchyma
- Y) Cleochyma
- Z) Sclerenchyma

Toss Up Answer: Z

Bonus: Short Answer

What anatomical structure is a projection of the cell membrane used for phagocytosis?

Bonus Answer: Pseudopods/pseudopodia

11. ENERGY

Toss Up: Multiple Choice

Which of the following is closest to the percentage of US electricity generated from natural gas in 2014?

W) 5%

X) 15%

Z) 40%

Toss Up Answer: Y

Bonus: Multiple Choice

Which of these states contain the largest natural gas field in the US?

W) Pennsylvania

X) Texas

Y) Louisiana

Z) Colorado

Bonus Answer: W

12. EARTH and SPACE

Toss Up: Short Answer

What paradox asks why we haven't found extraterrestrial life yet?

Bonus Answer: Fermi Paradox

Bonus: Short Answer

If a star is said to be 1.4 times more massive than the Sun, respectively, what threshold is it said to have passed, and what will it become in the future?

Bonus Answer: 1) Chandrasekhar limit, 2) supernova

13. PHYSICS

Toss Up: Multiple Choice

The zeroth law of thermodynamics allows us to define:

W) work

X) pressure

Y) temperature

Z) thermal equilibrium

Toss Up Answer: Y

Bonus: Short Answer

Which physicist contributed to the understanding of electrical circuits and coined the term "black body" radiation?

Bonus Answer: Kirchoff

14. BIOLOGY

Toss Up: Multiple Choice

What structure is present in cilia and flagella?

W) 9+0

X) 9+1

Y) 9+2

Z) 9+3

Toss Up Answer: Y

Bonus: Short Answer

Dicots feature vascular bundles arranged in what formation within the stem?

Bonus Answer: Arranged in a ring.

15. MATHEMATICS

Toss Up: Multiple Choice

MULTIPLE CHOICE: tossup

For an infinite sequence of numbers $a_0,a_1,...$, which of the following best describes the polynomial P(x) = sum(i=0) to infinity) $a_i * x^i$.

- W) Reimann-Zeta function
- X) Generating function
- Y) Newtonian series
- Z) Tchebychev polynomial

Toss Up Answer: X

Bonus: Short Answer

In the Pell equation $x^2-3y^2=1$ the smallest solution is x=2 and y=1. The next smallest solution is x=7 and y=4. Compute the third smallest solution.

Bonus Answer: x=26, y=15

Also accept (26,15)

16. CHEMISTRY

Toss Up: Multiple Choice

Which of the following is not a reason for the use of nitrogen in many explosives?

- W) these explosives form gases, which expand
- X) the bond between two nitrogen atoms is very strong
- Y) these explosives form toxic compounds such as cyanide
- Z) nitrogen is widely available

Toss Up Answer: Y

Bonus: Multiple Choice

Which of the following is closest to the bond angle in ammonia?

- W) 120 degrees
- X) 109.5 degrees
- Y) 107 degrees
- Z) 100 degrees

Bonus Answer: Y

17. PHYSICS

Toss Up: Multiple Choice

In order for two sound waves to produce audible beats, it is essential that the two waves have:

- W) the same amplitude
- X) slightly different amplitudes
- Y) the same number of harmonics
- Z) slightly different frequencies

Toss Up Answer: Z

Bonus: Multiple Choice

A 200-cm organ pipe with one end open is in resonance with a sound wave of wavelength 270cm. The pipe is

operating in its

- W) fundamental frequency
- X) second harmonic
- Y) third harmonic
- Z) fourth harmonic

Bonus Answer: X

18. CHEMISTRY

Toss Up: Multiple Choice

When the reaction of Silver Nitrate (aqueous) and Sodium Chloride (aqueous) goes to completion, what is formed?

- W) A Gas
- X) Water
- Y) An Acid
- Z) A Precipitate

Toss Up Answer: Z

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Bonus: Multiple Choice

When does the value of pH equal the value of pKa?

- W) when [HA]=[A-] (Read as: the molarity of the conjugate base equals the molarity of the conjugate acid)
- X) at the end point of a titration
- Y) everywhere in the buffer region
- Z) in the Henderson-Hasselbach equation

Bonus Answer: W

19. PHYSICS

Toss Up: Multiple Choice

A transverse traveling sinusoidal wave on a string has a frequency of 100Hz, a wavelength of 0.040m, and an amplitude of 2.0mm. The maximum velocity in m/s of any point on the string is

W) 0.2

X) 1.3

Y) 4

Z) 15

Toss Up Answer: X

Bonus: Multiple Choice

The time required for a small pulse to travel from A to B on a stretched cord shown is NOT altered by changing

- W) the linear mass density of the cord
- X) the length between A and B
- Y) the shape of the pulse
- Z) the tension in the cord

Bonus Answer: Y

20. CHEMISTRY

Toss Up: Multiple Choice

Which one of these metals doesn't produce a red flame in a flame test?

W) Rubidium

X) Strontium

Y) Cesium

Z) Lithium

Toss Up Answer: Y

Bonus: Short Answer

A solution contains some BaSO4 at 25* C. It is found to contain 4x10^-5 mole/liter of Ba^2+ ions. What is the Ksp of the BaSO4 in scientific notation?

Bonus Answer: 1.6*10^-9

21. PHYSICS

Toss Up: Multiple Choice

Find the angular frequency of oscillations in a LC circuit if the total inductance is 2 henry and the total capacitance is 8 fored

W) 1/4

X) 1/2

Y) 1/3

Z) 1/5

Toss Up Answer: W

Bonus: Short Answer

An LC circuit has an inductance of 20mH and a capacitance of 5.0 µF. At time

t = 0 the charge on the capacitor is 3.0 μ C and the current is 7.0mA. The total energy in joules is:

Bonus Answer: 9.0 * 10^-7 (accept equivalent forms)

22. CHEMISTRY

Toss Up: Multiple Choice

Which of the following salts is responsible for the browning of pretzels?

W) Sodium Chloride

X) Sodium Hydroxide

Y) Potassium Chloride

Z) Potassium Carbonate

Toss Up Answer: X

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Bonus: Short Answer

Consider the reaction A + 2B yields 4C. Assume all species in the reaction are gaseous. If the reaction is at equilibrium, and the concentration of every species is 2 molar, calculate the equilibrium constant of the reaction.

Bonus Answer: 2

23. CHEMISTRY

Toss Up: Short Answer

What is the oxidation state of chlorine in perchloric acid?

Bonus Answer: +7

Bonus: Short Answer

The formula for the combustion of methane is CH4 + 2O2 yields CO2 + 2H2O. Given that the heat of formation of methane gas is -75 kilojoules per mole, the heat of formation of carbon dioxide is -400 kilojoules per mole, and the heat of formation of water is -300 kilojoules per mole, find the heat of combustion for methane.

Bonus Answer: 925kj/mol

24. CHEMISTRY

Toss Up: Multiple Choice

Which noble gas is the most reactive?

W) Xenon

X) Radon

Y) Helium

Z) Argon

Toss Up Answer: W

Bonus: Short Answer

How many carbon atoms are in a single molecule of trinitrotolulene, or TNT?

Bonus Answer: 7

25. CHEMISTRY

Toss Up: Short Answer

Which nitrogen hydride is the most stable?

Bonus Answer: ammonia

Bonus: Multiple Choice

Why isn't silicon a suitable candidate for life like carbon is?

W) Silicon is too heavy

X) Silicon atoms are too big to form pi bonds

Y) Silicon is more reactive

Z) Silicon is rarer in the earth's crust

Bonus Answer: X
