

## Round 25

### 1. PHYSICS

Writer: Charles Zhang

Toss Up: Short Answer

A heated 8 kg ring with a radius of 4m cools as it rotates, causing it to shrink to a radius of 2m. If it was initially rotating at 6 rad/s, what is it's final angular velocity?

Bonus Answer: 24 rad/s

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Bonus: Short Answer

A 120-N child sits on a light swing and is pulled back and held with a horizontal force of 90 N. What is the magnitude of the tension force in each of the two supporting ropes?

Bonus Answer: 75 N

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### 2. MATHEMATICS

Writer: Benjamin Avrahami

Toss Up: Multiple Choice

What is the smallest positive 'taxicab' number?

W) 87

X) 91

Y) 95

Z) 100

Toss Up Answer: X

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Bonus: Short Answer

To what power do you have to raise any number for it to be 0, 1, or -1 in mod 7?

Bonus Answer: 3

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### 3. CHEMISTRY

Writer: Andrew Chen (Senior)

Toss Up: Multiple Choice

Certain atoms exhibit paramagnetic or diamagnetic bonding. Of the following below which choice is classified as paramagnetic?

W) Zinc (Zn)

X) Krypton (Kr)

Y) Helium (He)

Z) Oxygen (O)

Toss Up Answer: Z

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Bonus: Short Answer

Melting point is effected by bond strength. By name or number, order the following compounds in increasing order of melting point:

1. graphite

2. methane

3. lithium

4. sodium chloride

Bonus Answer: Methane, lithium, sodium chloride, graphite (2, 3 , 4, 1)

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### 4. MATHEMATICS

Writer: Steven Litvack-Winkler

Toss Up: Multiple Choice

For real numbers  $x$  and  $y$  on the closed interval from 0 to 1, let  $a=xy$  and  $b=(1-x)(1-y)$ . Find the maximum value over all choices of  $x$  and  $y$  for the minimum of  $a$  and  $b$ .

- W) 0
- X)  $1/4$
- Y)  $1/2$
- Z) 1

**Toss Up Answer: X**

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**Bonus: Short Answer**

One definition of the Cantor set is the set of all numbers  $x$  between 0 and 1 such that  $x$  has only the digits 0 and 2 in it's base 3 representation. Which of the following numbers is in the Cantor set.

- I.  $1/4$
- II.  $1/2$
- III.  $5/26$
- IV.  $7/10$

**Bonus Answer: 4 only**

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## 5. BIOLOGY

**Writer: Calvin Vuong**

**Toss Up: Short Answer**

Amino acids are attached to their respective tRNA molecules by which enzyme?

**Bonus Answer: aminoacyl-tRNA synthetase**

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**Bonus: Short Answer**

Which high energy molecule is responsible for forming the peptide bonds during translation elongation?

**Bonus Answer: GTP (ACCEPT: guanosine triphosphate)**

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## 6. CHEMISTRY

**Writer: Prangon Ghose**

**Toss Up: Multiple Choice**

Which of the following is not commonly produced by eletrolysis?

- W) NaOCl
- X) Al
- Y) Fe
- Z) NaOH

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

Which of the following pairs of constants are not mathematically related to eachother?

- W) Equilibrium constant and Gibbs Free Energy
- X) Rate Constant and Activation Energy
- Y) Standard Cell Voltage and Equilibrium Constant
- Z) Standard Cell Voltage and Rate Constant

**Bonus Answer: Z**

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## 7. BIOLOGY

**Writer: Mohammed Jamil**

**Toss Up: Multiple Choice**

Which buffer system prevent large changes in pH in the blood during gas exchange

- W) Hemoglobin
- X) Phosphate
- Y) Protein
- Z) Bicarbonate

**Toss Up Answer: Z**

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**Bonus: Multiple Choice**

How many ml of a 0.4 M HCl solution are required to bring the pH of 10 ml of a 0.4 M NaOH solution to 7.0 (neutral pH)?

- W) 4
- X) 40
- Y) 10
- Z) 20

**Bonus Answer: Y**

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**8. CHEMISTRY**

**Writer: Hanna Yang**

**Toss Up: Multiple Choice**

Which of the following is the lightest element with no stable isotopes?

- W) Tellurium
- X) Technetium
- Y) Promethium
- Z) Radon

**Toss Up Answer: X**

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**Bonus: Multiple Choice**

Which of the following is the strongest intermolecular force?

- W) Hydrogen Bonding
- X) Ionic Bonding
- Y) London Dispersion Force
- Z) Covalent Bonding

**Bonus Answer: W**

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**9. BIOLOGY**

**Writer: Josh Tish**

**Toss Up: Multiple Choice**

Acetone has a distinct smell, which many people associate with the smell of nail polish remover. What might the smell of acetone in the urine or on the breath of a patient indicate?

- W) The patient might be degrading too many amino acids from muscle proteolysis.
- X) The patient's body might be oxidizing too many fatty acids.
- Y) The patient might have enteritis and is absorbing endproducts of fermentation.
- Z) The patient might have fructose toxicity.

**Toss Up Answer: X**

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**Bonus: Multiple Choice**

Which description best describes what would happen to the carbon cycle if all detritivores suddenly went on "strike" and stopped working?

W) Carbon would increase in inorganic mass, while the atmospheric reservoir of carbon would continue to increase and plants would not be jeopardized.

X) Carbon would accumulate in organic mass, the atmospheric reservoir of carbon would decline, and plants would eventually be starved for CO<sub>2</sub>.

Y) Carbon would increase in organic mass, while the atmospheric reservoir of carbon would increase and plant-life would be starved for CO<sub>2</sub>.

Z) Carbon would decrease in organic mass, while the atmospheric reservoir of carbon would increase with the result that plant-life would be starved for CO<sub>2</sub>.

**Bonus Answer: X**

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## 10. MATHEMATICS

**Writer: Benjamin Avrahami**

**Toss Up: Short Answer**

In simplest form, express the surface area to volume ratio of a cube with side length  $s$ .

**Bonus Answer:  $6:s$**

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**Bonus: Short Answer**

Calculate the harmonic mean of the first two perfect numbers and 10.

**Bonus Answer:  $4 * \sqrt{105}$**

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## 11. BIOLOGY

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

Which of the following statements concerning photosynthetic light reactions is false?

W) Photosystem I contains the P700 reaction center.

X) The electron transport chain following photosystem I drives chemiosmosis.

Y) NADP<sup>+</sup> is the final electron acceptor.

Z) Electrons are replenished in photosystem II by the splitting of water.

**Toss Up Answer: X**

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**Bonus: Multiple Choice**

Which of the following molecules is not present in the electron transport chain immediately following photosystem II?

W) cytochrome complex

X) plastoquinone

Y) ferredoxin

Z) plastocyanin

**Bonus Answer: Y**

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## 12. ENERGY

**Writer: Aaron Gee**

**Toss Up: Short Answer**

Cadmium and boron are used in a nuclear reactor to absorb which subatomic particle?

**Bonus Answer: Neutron**

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**Bonus: Short Answer**

The electrical resistance of a semiconducting wire is least likely to be reduced by a decrease in which of the following four wire characteristics?  
Radius, Melting Point, Length, or Temperature?

**Bonus Answer: Melting Point**

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## 13. BIOLOGY

**Writer: Josh Tish**

**Toss Up: Multiple Choice**

Polarity in the developing *Drosophila* embryo is determined by:

- W) a protein gradient of the segmentation protein engrailed
- X) a protein gradient of the bicoid protein expressed from maternal mRNA
- Y) a protein gradient of the gap protein hunchback
- Z) expression of the segmentation protein engrailed throughout the embryo

**Toss Up Answer: X**

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**Bonus: Multiple Choice**

When a four-cell *Xenopus* embryo is divided into ventral and dorsal halves, the half containing the Nieuwkoop center will develop:

- W) ventralized features
- X) dorsalized features
- Y) as a normal embryo
- Z) only to the 8-cell stage

**Bonus Answer: X**

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## 14. PHYSICS

**Writer: Charles Zhang**

**Toss Up: Multiple Choice**

A cube with 2-m sides and a bulk modulus of  $4 \times 10^5 \text{ N/m}^2$  (READ 4 times 10 to the sixth newton meters squared). When it's subjected to a pressure of  $2 \times 10^5 \text{ Pa}$ , what is the resulting volume of the cube?

- W)  $0.5\text{m}^3$
- X)  $2\text{m}^3$
- Y)  $3\text{m}^3$
- Z)  $4\text{m}^3$

**Toss Up Answer: Z**

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**Bonus: Multiple Choice**

A 4.0-m long steel beam with a cross-sectional area of  $1.0 \times 10^{-2} \text{ m}^2$  and a Young's modulus of  $2.0 \times 10^{11} \text{ N/m}^2$  is wedged horizontally between two vertical walls. In order to wedge the beam, it is compressed by 0.020mm. If the coefficient of static friction between the beam and the walls is 0.70, what is the maximum mass (including its own) it can bear without slipping?

- W) 700kg
- X) 710kg
- Y) 900kg
- Z) 910kg

**Bonus Answer: X**

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## **15. BIOLOGY**

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

Where do the microtubules of the spindle originate during mitosis in both plant and animal cells?

- W) centromere
- X) centrosome
- Y) centriole
- Z) kinetochore

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

If cells in the process of dividing are subjected to colchicine, at which stage will mitosis be arrested?

- W) anaphase
- X) prophase
- Y) metaphase
- Z) interphase

**Bonus Answer: Y**

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## **16. ENERGY**

**Writer: Elias Milborn**

**Toss Up: Multiple Choice**

Which of the following is measured by total harmonic distortion, a standard for gauging the quality of power provided to customers?

- W) How distorted the waveform is from a pure sinusoidal waveform
- X) How distorted the waveform is from a triangular waveform
- Y) The difference between a sinusoidal waveform and a triangular waveform
- Z) the maximum value of a sinusoidal wave form with respect to the reference point of the waveform

**Toss Up Answer: W**

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**Bonus: Multiple Choice**

Which of the following is NOT an accurate representation of a limitation on the production and use of bioplastics?

- W) using crops for plastic diverts plants from the food supply
- X) bioplastics are unlikely to disintegrate in a landfill
- Y) bioplastics produce a variety of pollutants when burned
- Z) Individuals at home would find it more difficult to compost bioplastics

**Bonus Answer: Y**

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## **17. CHEMISTRY**

**Writer: Olivia Gallager**

**Toss Up: Short Answer**

How many sigma and pi bonds does 1,3 butadiene have, respectively?

**Bonus Answer: 9 sigma, 2 pi**

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**Bonus: Short Answer**

S orbital overlap has what shape?

**Bonus Answer: spherical, sphere**

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**18. PHYSICS**

**Writer: Jessica Titensky**

**Toss Up: Multiple Choice**

Which has the same units as joules?

W) Newton / meter

X) Pascal \* meter<sup>2</sup>

Y) Coulomb \* volt

Z) Kilogram \* meter / second<sup>2</sup>

**Toss Up Answer: Y**

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**Bonus: Short Answer**

What is the derived unit for Newton / meter<sup>2</sup>?

**Bonus Answer: Pascal**

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**19. CHEMISTRY**

**Writer: Olivia Gallager**

**Toss Up: Multiple Choice**

Which of the following is a strong nucleophile?

W) ethanol

X) butanol

Y) Bromide

Z) t-butoxide

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

What causes Coordination compounds to be different colors

W)  $\Delta E$

X) s orbital overlap

Y) VESPR

Z) p orbital overlap

**Bonus Answer: W**

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**20. PHYSICS**

**Writer: Shantanu Jha**

**Toss Up: Short Answer**

Which of the 6 quarks is the lightest?

**Bonus Answer: Up**

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**Bonus: Multiple Choice**

What is the "formal" SI unit of elementary particle spin?

W) joule-second

X) spin-h (planck constant)

Y) joule-plancktime

Z) spinor-second

**Bonus Answer: W**

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## **21. BIOLOGY**

**Writer: Josh Tish**

**Toss Up: Multiple Choice**

In plant cells grown in the presence of a metabolic poison that specifically inhibits mitochondrial F1 ATP synthase, one would expect:

- W) the pH difference across the cristae to be greater than normal.
- X) the pH difference across the cristae to be less than normal.
- Y) the electron transport chain to become inoperative.
- Z) oxygen consumption to cease.

**Toss Up Answer: W**

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**Bonus: Short Answer**

With proofreading, what is approximately the error rate of DNA polymerase?

**Bonus Answer: One error in every  $10^9$  bases.**

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## **22. PHYSICS**

**Writer: Shantanu Jha**

**Toss Up: Multiple Choice**

What quantity of a magnet determines the torque it will experience in an external magnetic field?

- W) Multipole Expansion
- X) Dipole Inversion
- Y) Magnetic Moment
- Z) Hysteresis Loop

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

What Dutch physicist, living from 1626-1695, pioneered the use of the pendulum in clocks?

- W) Willems Gravesande
- X) Hans Christian Oersted
- Y) Nicolas Hartsoeker
- Z) Christiaan Huygens

**Bonus Answer: Z**

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## **23. CHEMISTRY**

**Writer: Siam Muquit**

**Toss Up: Short Answer**

By name or number, which of the following is associated with inversion of stereochemistry?

- 1. Sn1
- 2. Sn2
- 3. E1
- 4. E2

**Bonus Answer: Sn2 only (2 only)**

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**Bonus: Short Answer**



Which rule states that in E1 and E2 reactions, the more substituted double bond is more likely to occur?

**Bonus Answer: Saytzeff rule**

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## 24. PHYSICS

**Writer: Charles Zhang**

**Toss Up: Multiple Choice**

A wave's equation is given as  $y = 0.1 \sin(3x + 10t)$  (READ AS: y equals 0.1 times sine of open parentheses 3x plus 10t close parentheses). What is the angular wave number?

W) 0.3

X) 1

Y) 10

Z) 3

**Toss Up Answer: Z**

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**Bonus: Short Answer**

An EM wave has a magnetic field with an amplitude of 200 Teslas. What is the amplitude of the wave's electric field in N/c (READ AS: newtons per coulomb)?

**Bonus Answer:  $6 \times 10^{10}$  N/c**

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## 25. PHYSICS

**Writer: Charles Zhang**

**Toss Up: Short Answer**

An object is dropped from an altitude of one Earth radius above Earth's surface. In terms of M, the mass of Earth, and R, Earth's radius, what is the speed of the object just before it hits Earth?

**Bonus Answer:  $\sqrt{GM/R}$  (READ AS: square root of GM over R)**

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**Bonus: Short Answer**

Two particles, each of mass m, are a distance d apart. If an external force brings a third particle, with mass 2m, from far away to a resting point midway between the two particles, what is the work done by this force?

**Bonus Answer:  $-8Gm^2/d^2$**

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