# Round 12

### 1. PHYSICS

Writer: Aaron Gee

Toss Up: Multiple Choice

The collision between a photon and a free electron was first explained by which of the following scientists?

W) Compton

X) Hertz

Y) Einstein

Z) Newton

Toss Up Answer: W

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# **Bonus: Multiple Choice**

The Tesla and the Gauss are units of measure of

W) magnetic field strength

X) conductance

Y) light

Z) electrical current

**Bonus Answer: W** 

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#### 2. PHYSICS

Writer: Aaron Gee

Toss Up: Multiple Choice

Whose principle or law states that each point on a wavefront may be considered a new wave source?

W) Snell's Law

X) Huygen's principle

Y) Young's Law

Z) Hertz's Law

Toss Up Answer: X

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# Bonus: Multiple Choice

The wave nature of light is demonstrated by which of the following?

W) Diffraction

X) Color

Y) Length

Z) Speed of light

Bonus Answer: W

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### 3. PHYSICS

Writer: Mohammed Jamil Toss Up: Short Answer

Given that the specific heat capacity of water is 11 times that of copper, calculate the mass of copper at a temperature of 100 °C required to raise the temperature of 200 g of water from 20.0 °C to 24.0 °C, assuming no energy is lost to the surroundings.

Bonus Answer: 0.116 kg

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#### **Bonus: Short Answer**

1 kg of water at a temperature of 45 °C is mixed with 1.5 kg of alcohol at 20 °C. Find the final temperature of the mixture.

Take the specific heat capacity of water to be 4200 J kg -1 K -1 and the specific heat capacity of alcohol to be 2400 J kg -1 K -1 . Assume no other exchange of heat occurs.

Bonus Answer: 33°C

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### 4. PHYSICS

Writer: Aaron Gee Toss Up: Short Answer

What is the resulting electrical potential, in volts, when a charge of 12

coulombs is applied to a 1 farad capacitor?

Bonus Answer: 12 volts

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**Bonus: Short Answer** 

If 1000 pounds is applied to a spring with spring constant of 100

pounds per inch on top of a hydraulic piston, how many pounds of force is transferred to the piston?

Bonus Answer: 1,000

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#### 5. PHYSICS

Writer: Mohammed Jamil Toss Up: Multiple Choice

Which statement describes a situation when polarization could not occur?

W) Light waves are reflected.

- X) Light waves are scattered.
- Y) Microwaves pass through a metal grid.
- Z) Sound waves pass through a metal grid.

Toss Up Answer: Z

# **Bonus: Multiple Choice**

An electromagnetic wave has a wavelength that is numerically of the same order of magnitude as the diameter of a nucleus.

In which region of the electromagnetic spectrum does the wave occur?

W) Gamma ray

X) X-ray

Y) Visible light

Z) Infra-red

Bonus Answer: W

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### 6. MATHEMATICS

Writer: Aaron Gee

Toss Up: Multiple Choice

Solve the following equation for x:

 $x^2 - 20x + 19 = 0$ 

W) 9, 10

X) 10 and 9

Y) 21 and 19

Z) 19 and 1

Toss Up Answer: Z

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**Bonus: Short Answer** 

Convert log base 4 of 53 into a base 10 expression:

Bonus Answer: log 53/ log4

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#### 7. MATHEMATICS

Writer: William Xiang
Toss Up: Short Answer

Find the integral of  $2x^3 + 2$  with respect to x in simplest terms.

Bonus Answer:  $(x^4)/2 + 2x$ 

**Bonus: Multiple Choice** 

Which of the following is a conic section?

W) CylinderX) Triangle

Y) Hyperbola

Z) Asymptote

**Bonus Answer: Y** 

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### 8. MATHEMATICS

Writer: William Xiang
Toss Up: Short Answer

Find the first derivative of  $10x^3 + 3x^2 + 3$  in the simplest form.

Bonus Answer: 30x^2 + 6x

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**Bonus: Short Answer** 

How many points of inflection does the graph  $x^3 + 5$  have?

**Bonus Answer: 1** 

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### 9. MATHEMATICS

Writer: Larry Wong
Toss Up: Short Answer

What is the value of 125 to the ¾ power?

Bonus Answer: 25

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**Bonus: Short Answer** 

What is the argument of 1 + sqrt 3 i?

Bonus Answer: 60

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# 10. BIOLOGY

Writer: Henry Zheng Toss Up: Short Answer

What is another name for the voice box?

**Bonus Answer: Larynx** 

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**Bonus: Short Answer** 

Which part of the larynx is responsible for sound production?

Bonus Answer: Intrinsic laryngeal muscles (accept intrinsic, intrinsic muscles)

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## 11. BIOLOGY

Writer: Henry Zheng Toss Up: Short Answer

What state was the first to use DNA to exonerate an accused serial rapist/murderer?

Bonus Answer: Virginia

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**Bonus: Short Answer** 

Who was the first person to be exonerated based on contradictory DNA evidence?

**Bonus Answer: David Vasquez** 

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#### 12. BIOLOGY

Writer: Henry Zheng Toss Up: Short Answer

What state was the first to use DNA to capture a serial rapist/murderer?

Bonus Answer: Virginia

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**Bonus: Short Answer** 

Who was the first murderer to be convicted on the basis of DNA evidence?

Bonus Answer: Timothy Wilson Spencer (accept Spencer, Timothy Spencer, Southside Strangler"

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### 13. BIOLOGY

Writer: Henry Zheng Toss Up: Multiple Choice

What light-sensitive cells in the eye detect colors?

W) RodsX) ConesY) Retina

Z) Lens

Toss Up Answer: X

**Bonus: Short Answer** 

What are the three types of cones?

Bonus Answer: blue, green, red (accept in any order)

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#### 14. BIOLOGY

Writer: Jason Weng Toss Up: Short Answer

What is it called when a protein loses its 3-dimensional shape?

**Bonus Answer: Denatured** 

**Bonus: Short Answer** 

Of the 4, name all the following that are not common amino acids: histidine, isoleucine, glutamine, isovaline

Bonus Answer: Isovaline

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#### 15. BIOLOGY

Writer: Jason Weng Toss Up: Short Answer

Which human organ, comprised mostly of lymphoid, is essential for the growth of T cells?

**Bonus Answer: Thymus** 

**Bonus: Short Answer** 

What cells are responsible for forming blood platelets?

**Bonus Answer: Megakaryocytes** 

### 16. BIOLOGY

Writer: Jason Weng Toss Up: Multiple Choice

Which of the following blood types has blood cells that possess antibodies for type A antigens?

W) A X) AB

Y) B

Z) O

Toss Up Answer: Y

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**Bonus: Short Answer** 

Which molecule produced by alveolar cells cause the decrease in surface tension within alveoli?

Bonus Answer: Pulmonary surfactant

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### 17. BIOLOGY

Writer: Mohammed Jamil Toss Up: Multiple Choice

Which type of cell has a large number of glycoproteins on the cell surface membrane?

W) Ciliated cell

X) Goblet cell

Y) Lymphocyte

Z) Red blood cell

Toss Up Answer: Y

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**Bonus: Short Answer** 

Name the bond that forms between glucose molecules in polysaccharides, such as amylose.

Bonus Answer: Glycosidic bond

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### 18. CHEMISTRY

Writer: Jason Weng

Toss Up: Multiple Choice

How much heat, in kJ, is needed to heat 1 kg of water 1 Kelvin if the specific heat of water is 4.18?

W) 4.18 kJ

X) 41.8 kJ

Y) 418 kJ

Z) 4180 kJ

Toss Up Answer: Z

**Bonus: Multiple Choice** 

What is the molecular geometry of phosphorous triflouride according to the VSEPR theory?

W) Bent

X) Seesaw

Y) Trigonal pyramid

Bonus Answer: Y

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#### 19. CHEMISTRY

Writer: Jason Weng

Toss Up: Multiple Choice

According to the VSEPR theory, what shape does methane assume?

W) Trigonal pyramid

X) Linear

Y) Tetrahedron

Z) Octahedron

Toss Up Answer: Y

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## **Bonus: Multiple Choice**

How many pi and sigma bond are present in hexene, respectively?

W) 17 pi, 1 sigmaX) 1 pi, 17 sigma

Y) 2 pi, 16 sigma

Z) 16 pi, 2 sigma

Bonus Answer: X

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#### 20. CHEMISTRY

Writer: Mohammed Jamil Toss Up: Short Answer

What is the chemical equation(s) for the reversible reaction(s) in the contact process

Bonus Answer:  $2SO2(g) + O2(g) \Rightarrow 2SO3(g)$ 

### **Bonus: Multiple Choice**

Which set of conditions would promote the fastest rate for the converter reaction in the contact process?

- W) No catalyst, high pressure, low temperature. Catalyst, high pressure, high temperature. Catalyst, low pressure, low temperature. Catalyst, low pressure, high temperature.
- X) Catalyst, high pressure, high temperature.
- Y) Catalyst, low pressure, low temperature
- Z) No catalyst, high pressure, low temperature. Catalyst, high pressure, high temperature. Catalyst, low pressure, low temperature. Catalyst, low pressure, high temperature.

**Bonus Answer: X** 

### 21. CHEMISTRY

Writer: Mohammed Jamil Toss Up: Multiple Choice

What suggests that metal, M, is not in Group I of the Periodic Table?

- W) M has a bright, silvery appearance and is a good conductor of electricity.
- X) M is hard and difficult to cut
- Y) M produces an alkaline solution when it reacts with water.
- Z) M produces hydrogen gas when it reacts with water.

### Toss Up Answer: X

### **Bonus: Multiple Choice**

Lactic acid, CH3CH(OH)CO2H, causes pain when it builds up in muscles.

Which reagent reacts with both of the -OH groups in lactic acid?

W) acidified potassium dichromate(VI)

X) ethanol

Y) sodium

Z) sodium hydroxide

**Bonus Answer: Y** 

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### 22. CHEMISTRY

Writer: Andrew Chen Toss Up: Short Answer

What is the common name of the simplest molecule with a carbonyl group?

Bonus Answer: formaldehyde

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### **Bonus: Multiple Choice**

What would be the best indicator for the titration of a weak base with a strong acid?

W) phenolphthalein

X) bromothymol blue

Y) methyl yellow

Z) methyl violet

Bonus Answer: Y

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### 23. EARTH and SPACE

Writer: Mohammed Haque Toss Up: Multiple Choice

What metal is Mercury mainly comprised of?

W) Lithium

X) Magnesium

Y) Iron

Z) Lead

Toss Up Answer: Y

# **Bonus: Multiple Choice**

How long is Mercury's orbital period (Earth days)?

W) 88 days

X) 165 days

Y) 49 days

Z) 57 days

Bonus Answer: W

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# 24. EARTH and SPACE

Writer: Mohammed Haque Toss Up: Multiple Choice

Which of the 3 following components needed for the formation of a star?

- W) Helium, lithium, and pressure.
- X) Helium, pressure, and time.
- Y) Hydrogen, pressure, and time.
- Z) Magnesium, pressure, and time.

Toss Up Answer: Y

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# **Bonus: Multiple Choice**

What gas is Venus's atmosphere mainly composed of?

- W) Carbon Monoxide
- X) Hydrogen Gas
- Y) Carbon Dioxide
- Z) Carbon is a gaseous form

Bonus Answer: Y

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# 25. EARTH and SPACE

Writer: Henry Zheng Toss Up: Short Answer

What is the most abundant gas in the atmosphere?

**Bonus Answer: Nitrogen** 

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**Bonus: Multiple Choice** 

An example of a density-dependent factor is

W) weather

X) climate

Y) air

Z) food

**Bonus Answer: Z**