

## Round 37

### 1. MATHEMATICS

Writer: Jessica Titensky

Toss Up: Short Answer

What is the amplitude of  $y=4\sin(5x+3)$

Bonus Answer: 4

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Bonus: Short Answer

What is the period of  $y=4\sin(5x+3)$  rounded to the nearest 100th

Bonus Answer: 1.26

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### 2. EARTH and SPACE

Writer: Shamaul Dilmohamed

Toss Up: Short Answer

What spectral type does the star Betelgeuse fall into?

Bonus Answer: M

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Bonus: Multiple Choice

When was the last major supernova in the Milky Way detected?

W) 1604

X) 1863

Y) 1973

Z) 1989

Bonus Answer: W

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### 3. PHYSICS

Writer: Yevgeniy Gorbachev

Toss Up: Short Answer

If a rocket has a specific impulse of 800 seconds, what is its exhaust velocity, in m/s? Assume standard gravity to be  $10 \text{ m/s}^2$ .

Bonus Answer: Answer: 8000 m/s (also acceptable: 8 km/s). Exhaust velocity is the product of specific impulse as a unit of time and standard gravity.

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Bonus: Short Answer

If an engine has a specific impulse of 500 seconds and a thrust of 10 kN, how long will it take the engine to burn 50 kg of fuel? Assume standard gravity to be  $10 \text{ m/s}^2$ .

Bonus Answer: Answer: 25 seconds.

Since  $F_{\text{thrust}} = G_{\text{standard}} \times I_{\text{sp}} \times R$ , where

$F_{\text{thrust}}$  = instantaneous thrust of the engine (in newtons)

$G_{\text{standard}}$  = standard gravity (usually  $9.81 \text{ m/s}^2$ , but for simplicity we round to 10)

$I_{\text{sp}}$  = specific impulse of the engine in seconds

$R$  = mass flow rate in kg/s

$10000 = 10 \times 500 \times R$ , so  $R = 2 \text{ kg/s}$

$50 \text{ kg} / 2 \text{ kg/s} = 25 \text{ s}$

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### 4. EARTH and SPACE

Writer: Shamaul Dilmohamed

Toss Up: Short Answer

If we could observe all forms of radiation, what would the brightest star be?

**Bonus Answer: Betelgeuse**

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**Bonus: Multiple Choice**

In 12000 years, which star will take the place of our North Star?

W) Aldebaran

X) Vega

Y) Sirius

Z) Rigel

**Bonus Answer: X**

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## 5. CHEMISTRY

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

The relation between pressure and temperature of two ideal gases is stated in

W) Gay-Lussac's Law

X) Boyle's Law

Y) Charles's Law

Z) Avogadro's Law

**Toss Up Answer: W**

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**Bonus: Short Answer**

What are the dimensions of the gas constant R in SI base units?

**Bonus Answer:  $\text{kg m}^2 \text{mol}^{-1} \text{K}^{-1} \text{s}^{-2}$  (accept as a fraction)**

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## 6. EARTH and SPACE

**Writer: Shantanu Jha**

**Toss Up: Short Answer**

What phenomena best explains why the sky is blue?

**Bonus Answer: Rayleigh Scattering**

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**Bonus: Short Answer**

What are the high intensity spots of light at the horizontal points of the 22 degree halo that may form around the sun due to ice crystals in the air called?

**Bonus Answer: Parhelia (ALSO ACCEPT: Sun Dogs, Parhelion)**

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## 7. MATHEMATICS

**Writer: Jessica Titensky**

**Toss Up: Short Answer**

What is  $\tan(\arcsin(9/41))$ ?

**Bonus Answer: 9/40**

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**Bonus: Short Answer**

What is  $\sin(\operatorname{arccot}(\tan(\operatorname{arccos}(3/5))))$ ?

**Bonus Answer: 3/5**

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## 8. PHYSICS

**Writer: Aaron Gee**

**Toss Up: Short Answer**

A military cannon fires a boy into the air at an angle of  $45^\circ$  above the horizontal, reaching a max height  $y$  above his original launch height. The cannon is now aimed so that it fires straight up into the air at an angle of  $90^\circ$  to the horizontal. What is the maximum height reached by the same boy now? (let  $y$  represent height)

**Bonus Answer:  $2y$**

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**Bonus: Short Answer**

a

**Bonus Answer: a**

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## 9. CHEMISTRY

**Writer: Jason Weng**

**Toss Up: Short Answer**

How many hydrogen atoms does a molecule of acetone have?

**Bonus Answer: 6**

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**Bonus: Short Answer**

What is the electron configuration of  $\text{Cr}^{3+}$  using noble gas notation?

**Bonus Answer:  $[\text{Ar}]3d^3$**

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## 10. PHYSICS

**Writer: Shantanu Jha**

**Toss Up: Multiple Choice**

What is the S.I. unit for luminous intensity?

- W) Lumen
- X) Candela
- Y) Newton
- Z) Watts

**Toss Up Answer: X**

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**Bonus: Multiple Choice**

For an isotropic source how many candelas equals 3 lumens?

- W)  $12\pi$
- X)  $10\pi$
- Y)  $4\pi$
- Z)  $2\pi$

**Bonus Answer: W**

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## 11. CHEMISTRY

**Writer: Jason Weng**

**Toss Up: Multiple Choice**

What is the most reasonable pH of the solution when the salt formed by reacting hydrochloric acid and aluminum hydroxide is dissolved in water?

- W) 7
- X) 4
- Y) 10
- Z) 0

**Toss Up Answer: X**

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**Bonus: Short Answer**

125 mL from a 4M perchloric acid stock solution is reacted with excess sodium hydroxide. How many moles of the salt are formed if the system is 50% efficient?

**Bonus Answer: 0.25 mol; 1/4 mol**

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**12. PHYSICS**

**Writer: Shantanu Jha**

**Toss Up: Short Answer**

If the sound intensity is 10,000 times the threshold of hearing then what is the intensity in decibels?

**Bonus Answer: 40dB**

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**Bonus: Short Answer**

At what standard frequency would 60 decibels have a loudness of 60 phons?

**Bonus Answer: 1000Hz**

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**13. BIOLOGY**

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

Neurotransmitters are received by which kinds of receptors?

W) G protein coupled receptors

X) receptor tyrosine kinases

Y) ion gated channels

Z) intracellular receptors

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

Which of the following is the immediate effect of a signal molecule binding to a receptor tyrosine kinase molecule?

W) It forms a dimer with another receptor tyrosine kinase.

X) It initiates a phosphorylation cascade.

Y) It becomes activated with phosphate groups.

Z) It attaches to a scaffolding protein.

**Bonus Answer: W**

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**14. CHEMISTRY**

**Writer: Jason Weng**

**Toss Up: Short Answer**

What is the oxidation number of chromium in the dichromate ion?

**Bonus Answer: +6**

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**Bonus: Short Answer**

Name the following:  $\text{CuCr}_2\text{O}_7$ .

**Bonus Answer: Copper (II) dichromate; or Cupric dichromate**

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**15. PHYSICS**

**Writer: Shantanu Jha**

**Toss Up: Multiple Choice**

What is the relativistic mass of a particle with a rest mass of 8g traveling at a speed of  $.6c$ ?

W) 4g

- X) 8g
- Y) 10g
- Z) 12g

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

What is the length of a 1 meter rod traveling on a spaceship going at .8c, as measured by an astronaut on the ship?

- W) .8m
- X) .6m
- Y) 1m
- Z) 1.25m

**Bonus Answer: Y**

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## 16. BIOLOGY

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

Which of the following intermolecular interactions is not exclusive to a single polypeptide's tertiary structure?

- W) hydrophobic interactions
- X) polar interactions
- Y) hydrogen bonding
- Z) disulfide bridges

**Toss Up Answer: Y**

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**Bonus: Short Answer**

Polypeptide folding is aided by which type of cavity-structured proteins?

**Bonus Answer: Chaperonins**

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## 17. MATHEMATICS

**Writer: Jessica Titensky**

**Toss Up: Short Answer**

How many seconds are in a day

**Bonus Answer: 86400**

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**Bonus: Multiple Choice**

How many days are in a second

- W)  $1.16 \times 10^{-3}$
- X)  $1.16 \times 10^{-4}$
- Y)  $1.16 \times 10^{-5}$
- Z)  $1.16 \times 10^{-6}$

**Bonus Answer: Y**

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## 18. BIOLOGY

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

Cellulose and glycogen differ in

- W) their alpha/beta glucose configurations
- X) branching

Y) their ability to be metabolized

Z) all of the above

**Toss Up Answer: Z**

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**Bonus: Short Answer**

N-Acetylglucosamine is the monomer of which common polysaccharide?

**Bonus Answer: Chitin**

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## 19. MATHEMATICS

**Writer: Jessica Titensky**

**Toss Up: Short Answer**

How many distinct roots does  $x^3 - 6x^2 + 32$  have

**Bonus Answer: 2**

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**Bonus: Short Answer**

What is the remainder when  $x^3 - 6x^2 + 32$  is divided by  $x - 1$

**Bonus Answer: 27**

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## 20. BIOLOGY

**Writer: Calvin Vuong**

**Toss Up: Short Answer**

Name all of the following that are amphipathic: integral proteins, cholesterol, phospholipids, triacylglycerol, mannose

**Bonus Answer: integral proteins and phospholipids**

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**Bonus: Multiple Choice**

Which of the following is not true about a part of the phospholipid?

W) They contain two fatty acid chains.

X) Their heads contain a net charge.

Y) They have a phosphate group in their heads.

Z) They have a choline group in their heads.

**Bonus Answer: X**

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## 21. CHEMISTRY

**Writer: Ashneel Das**

**Toss Up: Multiple Choice**

Which of the following substances cannot be decomposed further chemically?

W) Carbon Dioxide

X) Water

Y) Silicon

Z) Ammonia

**Toss Up Answer: Y**

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**Bonus: Short Answer**

If 0.5 moles of NaCl are dissolved in 2 kg of water, what is the molality of the resulting solution?

**Bonus Answer: 0.25 or 1/4**

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## 22. BIOLOGY

**Writer: Calvin Vuong**

**Toss Up: Multiple Choice**

Exocytotic vesicles are most frequently exported via the

- W) endoplasmic reticulum
- X) the nuclear envelope
- Y) the trans Golgi
- Z) the cis Golgi

**Toss Up Answer: Y**

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**Bonus: Multiple Choice**

Endosomes formed as a result of receptor-mediated endocytosis are most frequently coated with which type of proteins?

- W) G proteins
- X) clathrins
- Y) pseudopodium
- Z) microtubules

**Bonus Answer: X**

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## **23. BIOLOGY**

**Writer: Hanna Yang**

**Toss Up: Multiple Choice**

Which of the following is a retrovirus?

- W) HIV
- X) Hepatitis B virus
- Y) Poliovirus
- Z) Influenza A virus

**Toss Up Answer: W**

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**Bonus: Short Answer**

Where do DNA viruses usually replicate?

**Bonus Answer: In the nucleus.**

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## **24. BIOLOGY**

**Writer: Hanna Yang**

**Toss Up: Multiple Choice**

Which of the following is present in plant cells but not animal cells?

- W) Nuclei
- X) Plasmodesmata
- Y) Mitochondria
- Z) Vacuoles

**Toss Up Answer: X**

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**Bonus: Short Answer**

What is the name of the theory that explains the existence of mitochondria (and why they have their own DNA)?

**Bonus Answer: Endosymbiotic Theory**

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## **25. BIOLOGY**

**Writer: Hanna Yang**

**Toss Up: Multiple Choice**

What transmissible agent causes Mad Cow Disease (BSE)?

- W) Viron
- X) Virus
- Y) Bacteria
- Z) Prion

**Toss Up Answer: Z**

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**Bonus: Multiple Choice**

Which of the following are prions made up of?

- W) Proteins, only
- X) Nucleic Acids, only
- Y) Proteins and Nucleic Acids, only
- Z) Proteins and Carbohydrates, only

**Bonus Answer: W**

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