Fundamental Physical Constants

This booklet gives the latest values of the basic constants and conversion factors of physics and chemistry recommended by the Committee on Data for Science and Technology (CODATA) for international use. This, the 1998 set, replaces its immediate predecessor recommended by CODATA in 1986 and takes into account all of the data available through 31 December 1998.

The values given in these tables are a self-consistent set from a least squares evaluation produced by P J Mohr and B N Taylor (J. Phys. Chem. Ref. Data, 28(6), 1713–1852 (1999)). Energy conversion factors (these are apparent from their units) have been included in the table immediately below the appropriate quantities. The figures in parentheses () in the 'value' column represent the best estimates of the standard deviation uncertainties in the last two digits quoted, based on internal consistency. The International System of Units (SI) have been employed throughout this booklet.

CODATA was established in 1966 as an interdisciplinary committee of the International Council of Scientific Unions (ICSU), now the International Council for Science. It seeks to improve the quality, reliability, processing, management, and accessibility of data of importance to science and technology. In 1969 the Task Group on Fundamental Constants was set up to periodically review all the relevant data available at a given time, and to produce a self-consistent set of basic constants and energy conversion factors for international use.

The National Physical Laboratory (NPL) has the primary responsibility in the UK for the determination of the key fundamental constants. For further information contact the NPL Helpline.

Universal

speed of light in vacuum	c, c_0	299792458	$\mathrm{m}\cdot\mathrm{s}^{-1}$	(exact)
magnetic constant	μ_0	$4\pi \times 10^{-7}$	${ m N\cdot A^{-2}}$	
		$= 12.566370614\ldots\times10^{-7}$	${ m N\cdot A^{-2}}$	(exact)
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854187817\ldots \times 10^{-12}$	${\rm F\cdot m^{-1}}$	(exact)
characteristic impedance of vacuum $\sqrt{\mu_0/\epsilon_0} = \mu_0 c$	Z_0	376.730 313 461	Ω	(exact)
Newtonian constant of gravitation	G	$6.673(10) \times 10^{-11}$	$m^3 \cdot kg^{-1} \cdot s^{-2}$	1.5×10^{-3}
	$G/\hbar c$	$6.707(10) \times 10^{-39}$	$(\mathrm{GeV}/c^2)^{-2}$	1.5×10^{-3}
Planck constant	h	$6.62606876(52)\times10^{-34}$	$J \cdot s$	7.8×10^{-8}
in eV \cdot s		$4.13566727(16)\times10^{-15}$	$eV \cdot s$	3.9×10^{-8}
$h/2\pi$	\hbar	$1.054571596(82)\times 10^{-34}$	$J \cdot s$	7.8×10^{-8}
in eV \cdot s		$6.58211889(26)\times10^{-16}$	$eV \cdot s$	3.9×10^{-8}
Planck mass $(\hbar c/G)^{1/2}$	$m_{ m p}$	$2.1767(16) \times 10^{-8}$	kg	7.5×10^{-4}
Planck length $\hbar/m_{ m p}c=(\hbar G/c^3)^{1/2}$	$l_{ m p}$	$1.6160(12) \times 10^{-35}$	m	7.5×10^{-4}
Planck time $l_{\rm p}/c = (\hbar G/c^5)^{1/2}$	$t_{ m p}$	$5.3906(40) \times 10^{-44}$	S	7.5×10^{-4}

Electromagnetic

elementary charge	e	$1.602176462(63)\times 10^{-19}$	C	3.9×10^{-8}
	e/h	$2.417989491(95)\times 10^{14}$	${ m A\cdot J^{-1}}$	3.9×10^{-8}
magnetic flux quantum $h/2e$	Φ_0	$2.067833636(81)\times 10^{-15}$	Wb	3.9×10^{-8}
conductance quantum $2e^2/h$	G_0	$7.748091696(28) \times 10^{-5}$	S	3.7×10^{-9}
inverse of conductance quantum	G_0^{-1}	12906.403786(47)	Ω	3.7×10^{-9}
Josephson constant a $2e/h$	$K_{ m J}$	$483597.898(19) \times 10^9$	$\mathrm{Hz}\cdot\mathrm{V}^{-1}$	3.9×10^{-8}
von Klitzing constant ^b $h/e^2 = \mu_0 c/2\alpha$	$R_{ m K}$	25 812.807 572(95)	Ω	3.7×10^{-9}
Bohr magneton $e\hbar/2m_{\rm e}$	$\mu_{ m B}$	$927.400899(37) \times 10^{-26}$	${ m J\cdot T^{-1}}$	4.0×10^{-8}
in eV \cdot T ⁻¹		$5.788381749(43)\times 10^{-5}$	${ m eV\cdot T^{-1}}$	7.3×10^{-9}
	$\mu_{ m B}/h$	$13.99624624(56)\times 10^9$	$\mathrm{Hz}\cdot\mathrm{T}^{-1}$	4.0×10^{-8}
	$\mu_{ m B}/hc$	46.6864521(19)	$\mathrm{m}^{-1}\cdot\mathrm{T}^{-1}$	4.0×10^{-8}

^aSee the "Adopted values" table for the conventional value adopted internationally for realizing representations of the volt using the Josephson effect.

bSee the "Adopted values" table for the conventional value adopted internationally for realizing representations of the ohm using

the quantum Hall effect.

Quantity	Symbol	Value	Unit	Relative std. uncert. $\mu_{\rm r}$
	$\mu_{ m B}/k$	0.671 7131(12)	${ m K\cdot T^{-1}}$	1.7×10^{-6}
nuclear magneton $e\hbar/2m_{\rm p}$	$\mu_{ m N}$	$5.05078317(20)\times10^{-27}$	$J \cdot T^{-1}$	4.0×10^{-8}
in $eV \cdot T^{-1}$		$3.152451238(24) \times 10^{-8}$	${ m eV\cdot T^{-1}}$	7.6×10^{-9}
	$\mu_{ m N}/h$	7.62259396(31)	$ m MHz \cdot T^{-1}$	4.0×10^{-8}
	$\mu_{ m N}/hc$	$2.54262366(10)\times 10^{-2}$	$\mathrm{m}^{-1}\cdot\mathrm{T}^{-1}$	4.0×10^{-8}
	$\mu_{ m N}/k$	$3.6582638(64)\times 10^{-4}$	$\mathbf{K}\cdot\mathbf{T}^{-1}$	1.7×10^{-6}
Atomic and Nuclear				
General				
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297352533(27) \times 10^{-3}$		3.7×10^{-9}
inverse fine-structure constant	α^{-1}	137.035 999 76(50)		3.7×10^{-9}
Rydberg constant $\alpha^2 m_{\rm e} c/2h$	R_{∞}	10973731.568549(83)	m^{-1}	7.6×10^{-12}
	$R_{\infty}c$	$3.289841960368(25)\times 10^{15}$	$_{\mathrm{Hz}}$	7.6×10^{-12}
	$R_{\infty}hc$	$2.17987190(17)\times 10^{-18}$	J	7.8×10^{-8}
$R_{\infty}hc$ in eV		13.60569172(53)	eV	3.9×10^{-8}
Bohr radius $\alpha/4\pi R_{\infty} = 4\pi\epsilon_0 \hbar^2/m_{\rm e}e^2$	a_0	$0.5291772083(19) \times 10^{-10}$	m	3.7×10^{-9}
Hartree energy $e^2/4\pi\epsilon_0 a_0 = 2R_{\infty}hc = \alpha^2 m_{\rm e}c^2$	$E_{ m h}$	$4.35974381(34) \times 10^{-18}$	J	7.8×10^{-8}
in eV		27.211 3834(11)	eV	3.9×10^{-8}
quantum of circulation	$h/2m_{ m e}$	$3.636947516(27)\times 10^{-4}$	$\mathrm{m}^2\cdot\mathrm{s}^{-1}$	7.3×10^{-9}
	$h/m_{ m e}$	$7.273895032(53)\times 10^{-4}$	$\mathrm{m}^2\cdot\mathrm{s}^{-1}$	7.3×10^{-9}
Electroweak				
Fermi coupling constant ^c	$G_{ m F}/(\hbar c)^3$	$1.16639(1)\times 10^{-5}$	${ m GeV}^{-2}$	8.6×10^{-6}
weak mixing angle ^d $\Theta_{\rm W}$ (on-shell scheme) $\sin^2 \Theta_{\rm W} = s_{\rm W}^2 \equiv 1 - (m_{\rm W}/m_{\rm Z})^2$	$\sin^2 \Theta_{ m W}$	0.2224(19)		8.7×10^{-3}

^cValue recommended by the Particle Data Group, Caso et al., Eur. Phys. J. C **3**(1–4), 1–794(1998)

dBased on the ratio of the masses of the W and M bosons $m_{\rm W}/m_{\rm Z}$ recommended by the Particle Data Group (Caso et al., 1998). The value for $\sin^2 \theta_{\rm W}$ they recommend, which is based on a particular variant of the modified minimal subtraction ($\overline{\rm MS}$) scheme, is $\sin^2 \hat{\theta}_{\rm W}(M_{\rm Z}) = 0.231\ 24(24)$.

${\bf Electron,\ e^-}$				
electron mass	$m_{ m e}$	$9.10938188(72)\times10^{-31}$	kg	7.9×10^{-8}
in u, $m_e = A_r(e)$ u (electron relative mass times u)		$5.485799110(12) \times 10^{-4}$	u	2.1×10^{-9}
energy equivalent	$m_{ m e}c^2$	$8.18710414(64)\times 10^{-14}$	J	7.9×10^{-8}
in MeV		0.510998902(21)	MeV	4.0×10^{-8}
electron-muon mass ratio	$m_{ m e}/m_{ m \mu}$	$4.83633210(15)\times 10^{-3}$		3.0×10^{-8}
electron-tau mass ratio	$m_{ m e}/m_{ m au}$	$2.87555(47) \times 10^{-4}$		1.6×10^{-4}
electron-proton mass ratio	$m_{ m e}/m_{ m p}$	$5.446170232(12)\times 10^{-4}$		2.1×10^{-9}
electron-neutron mass ratio	$m_{ m e}/m_{ m n}$	$5.438673462(12)\times 10^{-4}$		2.2×10^{-9}
electron-deuteron mass ratio	$m_{ m e}/m_{ m d}$	$2.7244371170(58)\times 10^{-4}$		2.1×10^{-9}
electron to alpha particle mass ratio	$m_{ m e}/m_{ m lpha}$	$1.3709335611(29)\times10^{-4}$		2.1×10^{-9}
electron charge to mass quotient	$-e/m_{ m e}$	$-1.758820174(71)\times 10^{11}$	$C \cdot kg^{-1}$	4.0×10^{-8}
electron molar mass $N_{\rm A}m_{\rm e}$	$M(e), M_e$	$5.485799110(12) \times 10^{-7}$	$kg \cdot mol^{-1}$	2.1×10^{-9}
Compton wavelength $h/m_{\rm e}c$	$\lambda_{ m C}$	$2.426310215(18)\times 10^{-12}$	m	7.3×10^{-9}
$\lambda_{\rm C}/2\pi = \alpha a_0 = \alpha^2/4\pi R_{\infty}$	$\lambda_{ m C}$	$386.1592642(28)\times 10^{-15}$	m	7.3×10^{-9}
classical electron radius $\alpha^2 a_0$	$r_{ m e}$	$2.817940285(31)\times 10^{-15}$	m	1.1×10^{-8}
Thomson cross section $(8\pi/3)r_{\rm e}^2$	$\sigma_{ m e}$	$0.665245854(15)\times 10^{-28}$	m^2	2.2×10^{-8}
electron magnetic moment	$\mu_{ m e}$	$-928.476362(37)\times10^{-26}$	$J \cdot T^{-1}$	4.0×10^{-8}
to Bohr magneton ratio	$\mu_{ m e}/\mu_{ m B}$	-1.0011596521869(41)		4.1×10^{-12}
to nuclear magneton ratio	$\mu_{ m e}/\mu_{ m N}$	-1838.2819660(39)		2.1×10^{-9}
electron magnetic moment anomaly $ \mu_{\rm e} /\mu_{\rm B}-1$	$a_{ m e}$	$1.1596521869(41) \times 10^{-3}$		3.5×10^{-9}
electron g-factor $-2(1+a_e)$	$g_{ m e}$	-2.0023193043737(82)		4.1×10^{-12}
electron-muon magnetic moment ratio	$\mu_{ m e}/\mu_{ m \mu}$	206.766 9720(63)		3.0×10^{-8}
electron-proton magnetic moment ratio	$\mu_{ m e}/\mu_{ m p}$	-658.2106875(66)		1.0×10^{-8}
electron to shielded proton magnetic moment ratio ($\rm H_2O$, sphere, 25 °C)	$\mu_{ m e}/\mu_{ m p}'$	-658.2275954(71)		1.1×10^{-8}
electron-neutron magnetic moment ratio	$\mu_{ m e}/\mu_{ m n}$	960.920 50(23)		2.4×10^{-7}
electron-deuteron magnetic moment ratio	$\mu_{ m e}/\mu_{ m d}$	-2143.923498(23)		1.1×10^{-8}
electron to shielded helion magnetic moment ratio (gas, sphere, 25 °C)	$\mu_{ m e}/\mu_{ m h}'$	864.058 255(10)		1.2×10^{-8}

Symbol

Value

 $\underline{\mathrm{Quanti}}\mathrm{ty}$

 ${\it Relative std.}$

uncert. $\mu_{\rm r}$

Unit

Quantity	Symbol	Value	Unit	Relative std. uncert. $\mu_{\rm r}$
electron gyromagnetic ratio $2 \mu_{\rm e} /\hbar$	$\gamma_{ m e}$	$1.760859794(71) \times 10^{11}$	$\mathrm{s}^{-1}\cdot\mathrm{T}^{-1}$	4.0×10^{-8}
	$\gamma_{ m e}/2\pi$	28 024.9540(11)	${ m MHz}\cdot{ m T}^{-1}$	4.0×10^{-8}
$Muon,\mu^-$				
muon mass	$m_{ m \mu}$	$1.88353109(16) \times 10^{-28}$	kg	8.4×10^{-8}
in u, $m_{\mu} = A_{\rm r}(\mu)$ u (muon relative atomic mass times u)		0.1134289168(34)	u	3.0×10^{-8}
energy equivalent	$m_{\mu}c^2$	$1.69283332(14)\times 10^{-11}$	J	8.4×10^{-8}
$\mathrm{in}\ \mathrm{MeV}$		105.6583568(52)	${ m MeV}$	4.9×10^{-8}
muon-electron mass ratio	$m_{ m \mu}/m_{ m e}$	206.7682657(63)		3.0×10^{-8}
muon-tau mass ratio	$m_{ m \mu}/m_{ m au}$	$5.94572(97)\times 10^{-2}$		1.6×10^{-4}
muon-proton mass ratio	$m_{ m \mu}/m_{ m p}$	0.1126095173(34)		3.0×10^{-8}
muon-neutron mass ratio	$m_{ m \mu}/m_{ m n}$	0.1124545079(34)		3.0×10^{-8}
muon molar mass $N_{\rm A} m_{\rm \mu}$	$M(\mu), M_{\mu}$	$0.1134289168(34)\times 10^{-3}$	$\mathrm{kg}\cdot\mathrm{mol}^{-1}$	3.0×10^{-8}
muon Compton wavelength $h/m_{\mu}c$	$\lambda_{\mathrm{C},\mu}$	$11.73444197(35) \times 10^{-15}$	m	2.9×10^{-8}
$\lambda_{\mathrm{C},\mu}/2\pi$	$\lambda_{\mathrm{C},\mu}$	$1.867594444(55)\times 10^{-15}$	m	2.9×10^{-8}
muon magnetic moment	μ_{μ}	$-4.49044813(22) \times 10^{-26}$	$J \cdot T^{-1}$	4.9×10^{-8}
to Bohr magneton ratio	$\mu_{ m \mu}/\mu_{ m B}$	$-4.84197085(15)\times10^{-3}$		3.0×10^{-8}
to nuclear magneton ratio	$\mu_{ m \mu}/\mu_{ m N}$	-8.89059770(27)		3.0×10^{-8}
muon magnetic moment anomaly $ \mu_{\mu} /(e\hbar/2m_{\mu})-1$	$a_{oldsymbol{\mu}}$	$1.16591602(64)\times 10^{-3}$		5.5×10^{-7}
muon g-factor $-2(1+a_{\mu})$	$g_{oldsymbol{\mu}}$	-2.0023318320(13)		6.4×10^{-10}
muon-proton magnetic moment ratio	$\mu_{ m \mu}/\mu_{ m p}$	-3.18334539(10)		3.2×10^{-8}
Tau, τ^-				
tau mass ^e	$m_{ au}$	$3.16788(52)\times10^{-27}$	kg	1.6×10^{-4}
in u, $m_{\tau} = A_{\rm r}(\tau)$ u (tau relative atomic mass time u)		1.907 74(31)	u	1.6×10^{-4}
energy equivalent	$m_{\tau}c^2$	$2.84715(46) \times 10^{-10}$	J	1.6×10^{-4}
$\mathrm{in}\ \mathrm{MeV}$		1777.05(29)	${ m MeV}$	1.6×10^{-4}
tau-electron mass ratio	$m_{ au}/m_{ m e}$	3477.60(57)		1.6×10^{-4}
tau-muon mass ratio	$m_{ au}/m_{ extsf{\mu}}$	16.8188(27)		1.6×10^{-4}
tau-proton mass ratio	$m_{ au}/m_{ m p}$	1.89396(31)		1.6×10^{-4}

^eThis and all other values involving m_{τ} are based on the values of $m_{\tau}c^2$ in MeV recommended by the Particle Data Group (Caso et al., 1998), but with a standard uncertainty of 0.29 MeV rather than the quoted uncertainty of -0.26 MeV, +0.29 MeV.

Quantity	Symbol	Value	Unit	Relative std. uncert. $\mu_{\rm r}$
tau-neutron mass ratio	$m_{ au}/m_{ m n}$	1.891 35(31)		1.6×10^{-4}
tau molar mass $N_{\rm A} m_{ m au}$	$M(au), M_{ au}$	$1.90774(31)\times 10^{-3}$	$\mathrm{kg}\cdot\mathrm{mol}^{-1}$	1.6×10^{-4}
tau Compton wavelength $h/m_{\tau}c$	$\lambda_{\mathrm{C}, au}$	$0.69770(11) \times 10^{-15}$	m	1.6×10^{-4}
$\lambda_{\mathrm{C}, au}/2\pi$	$\lambda_{\mathrm{C},\tau}$	$0.111042(18)\times 10^{-15}$	m	1.6×10^{-4}
Proton, p				
proton mass	$m_{ m p}$	$1.67262158(13) \times 10^{-27}$	kg	7.9×10^{-8}
in u, $m_p = A_r(p)u$ (proton relative atomic mass times u)		1.00727646688(13)	u	1.3×10^{-10}
energy equivalent	$m_{ m p}c^2$	$1.50327731(12)\times 10^{-10}$	J	7.9×10^{-8}
$\mathrm{in}\ \mathrm{MeV}$		938.271 998(38)	${ m MeV}$	4.0×10^{-8}
proton-electron mass ratio	$m_{ m p}/m_{ m e}$	1836.1526675(39)		2.1×10^{-9}
proton-muon mass ratio	$m_{ m p}/m_{ m \mu}$	8.880 244 08(27)		3.0×10^{-8}
proton-tau mass ratio	$m_{ m p}/m_{ m au}$	0.527994(86)		1.6×10^{-4}
proton-neutron mass ratio	$m_{ m p}/m_{ m n}$	0.99862347855(58)		5.8×10^{-10}
proton charge to mass quotient	$e/m_{ m p}$	$9.57883408(38)\times10^7$	$C \cdot kg^{-1}$	4.0×10^{-8}
proton molar mass $N_{\rm A}m_{\rm p}$	$M(\mathbf{p}), M_{\mathbf{p}}$	$1.00727646688(13) \times 10^{-3}$	$\mathrm{kg}\cdot\mathrm{mol}^{-1}$	1.3×10^{-10}
proton Compton wavelength $h/m_{\rm p}c$	$\lambda_{ ext{C,p}}$	$1.321409847(10) \times 10^{-15}$	m	7.6×10^{-9}
$\lambda_{ m C,p}/2\pi$	$\lambda_{\mathrm{C,p}}$	$0.2103089089(16) \times 10^{-15}$	m	7.6×10^{-9}
proton magnetic moment	$\mu_{ m p}$	$1.410606633(58)\times 10^{-26}$	${ m J\cdot T^{-1}}$	4.1×10^{-8}
to Bohr magneton ratio	$\mu_{ m p}/\mu_{ m B}$	$1.521032203(15)\times 10^{-3}$		1.0×10^{-8}
to nuclear magneton ratio	$\mu_{ m p}/\mu_{ m N}$	2.792847337(29)		1.0×10^{-8}
proton g-factor $2\mu_{\rm p}/\mu_{\rm N}$	$g_{ m p}$	5.585 694 675(57)		1.0×10^{-8}
proton-neutron magnetic moment ratio	$\mu_{ m p}/\mu_{ m n}$	-1.45989805(34)		2.4×10^{-7}
shielded proton magnetic moment(H ₂ O, sphere, 25 °C)	$\mu_{ m p}'$	$1.410570399(59)\times 10^{-26}$	$J \cdot T^{-1}$	4.2×10^{-8}
to Bohr magneton ratio	$\mu_{ m p}'/\mu_{ m B}$	$1.520993132(16)\times 10^{-3}$		1.1×10^{-8}
to nuclear magneton ratio	$\mu_{ m p}'/\mu_{ m N}$	2.792 775 597(31)		1.1×10^{-8}
proton magnetic shielding correction $1 - \mu_{\rm p}'/\mu_{\rm p}$ (H ₂ O, sphere, 25 °C)	$\sigma_{ m p}'$	$25.687(15) \times 10^{-6}$		5.7×10^{-4}
proton gyromagnetic ratio $2\mu_{\rm p}/\hbar$	$\gamma_{ m p}$	$2.675\ 222\ 12(11) \times 10^8$	$\mathrm{s}^{-1}\cdot\mathrm{T}^{-1}$	4.1×10^{-8}
	$\gamma_{\mathrm{p}}/2\pi$	42.577 4825(18)	$ m MHz \cdot T^{-1}$	4.1×10^{-8}
shielded proton gyromagnetic ratio $2\mu'_p/\hbar$ (H ₂ O, sphere, 25 °C)	$\gamma_{ m p}'$	$2.67515341(11)\times10^8$	$s^{-1} \cdot T^{-1}$	4.2×10^{-8}
	$\gamma_{ m p}'/2\pi$	42.576 3888(18)	$ m MHz \cdot T^{-1}$	4.2×10^{-8}

Noutron n				
Neutron, n				
neutron mass	$m_{ m n}$	$1.67492716(13)\times 10^{-27}$	kg	7.9×10^{-8}
in u, $m_n = A_r(n)$ u (neutron relative atomic mass times u)		1.00866491578(55)	u	5.4×10^{-10}
energy equivalent	$m_{\rm n}c^2$	$1.50534946(12)\times 10^{-10}$	J	7.9×10^{-8}
in MeV		939.565330(38)	${ m MeV}$	4.0×10^{-8}
neutron-electron mass ratio	$m_{ m n}/m_{ m e}$	1838.6836550(40)		2.2×10^{-9}
neutron-muon mass ratio	$m_{ m n}/m_{ m \mu}$	8.892 484 78(27)		3.0×10^{-8}
neutron-tau mass ratio	$m_{ m n}/m_{ m au}$	0.528722(86)		1.6×10^{-4}
neutron-proton mass ratio	$m_{ m n}/m_{ m p}$	1.00137841887(58)		5.8×10^{-10}
neutron molar mass $N_{\rm A} m_{\rm n}$	$M(\mathbf{n}), M_{\mathbf{n}}$	$1.00866491578(55)\times 10^{-3}$	$\mathrm{kg}\cdot\mathrm{mol}^{-1}$	5.4×10^{-10}
neutron Compton wavelength $h/m_{ m n}c$	$\lambda_{\mathrm{C,n}}$	$1.319590898(10) \times 10^{-15}$	m	7.6×10^{-9}
$\lambda_{ m C,n}/2\pi$	$\lambda_{\mathrm{C,n}}$	$0.2100194142(16)\times 10^{-15}$	m	7.6×10^{-9}
neutron magnetic moment	$\mu_{ m n}$	$-0.96623640(23) \times 10^{-26}$	$J \cdot T^{-1}$	2.4×10^{-7}
to Bohr magneton ratio	$\mu_{ m n}/\mu_{ m B}$	$-1.04187563(25)\times 10^{-3}$		2.4×10^{-7}
to nuclear magneton ratio	$\mu_{ m n}/\mu_{ m N}$	-1.91304272(45)		2.4×10^{-7}
neutron g-factor $2\mu_{\rm n}/\mu_{\rm N}$	$g_{ m n}$	-3.82608545(90)		2.4×10^{-7}
neutron-electron magnetic moment ratio	$\mu_{ m n}/\mu_{ m e}$	$1.04066882(25)\times 10^{-3}$		2.4×10^{-7}
neutron-proton magnetic moment ratio	$\mu_{ m n}/\mu_{ m p}$	-0.68497934(16)		2.4×10^{-7}
neutron to shielded proton magnetic moment ratio (H_2O , sphere, 25 °C)	$\mu_{ m n}/\mu_{ m p}'$	-0.68499694(16)		2.4×10^{-7}
neutron gyromagnetic ratio $2 \mu_{\rm n} /\hbar$	$\gamma_{ m n}$	$1.83247188(44) \times 10^8$	$s^{-1} \cdot T^{-1}$	2.4×10^{-7}
	$\gamma_{\rm n}/2\pi$	29.1646958(70)	${ m MHz}\cdot{ m T}^{-1}$	2.4×10^{-7}
Deutron, d				
1		0.040,700,00/00)	,	T 0 10 9
deuteron mass in u, $m_{\rm d} = A_{\rm r}({\rm d})$ u (deuteron	$m_{ m d}$	$3.34358309(26) \times 10^{-27}$	kg	7.9×10^{-8}
relative atomic mass times u)		2.01355321271(35)	u	1.7×10^{-10}
energy equivalent	$m_{\rm d}c^2$	$3.00506262(24)\times10^{-10}$	J	7.9×10^{-8}
in MeV		1875.612762(75)	MeV	4.0×10^{-8}
deuteron-electron mass ratio	$m_{ m d}/m_{ m e}$	3670.4829550(78)		2.1×10^{-9}
deuteron-proton mass ratio	$m_{ m d}/m_{ m p}$	1.99900750083(41)		2.0×10^{-10}
deuteron molar mass $N_{\rm A}m_{\rm d}$	$M(d), M_d$	$2.01355321271(35)\times10^{-3}$	$\mathrm{kg}\cdot\mathrm{mol}^{-1}$	1.7×10^{-10}

Symbol

Quantity

Value

 ${\it Relative std.}$

uncert. $\mu_{\rm r}$

Unit

ty S	Symbol	Value	Unit	Relative std uncert. $\mu_{\rm r}$
on magnetic moment μ	$\mu_{ m d}$	$0.433073457(18) \times 10^{-26}$	$J \cdot T^{-1}$	4.2×10^{-8}
Bohr magneton ratio μ	$\mu_{ m d}/\mu_{ m B}$	$0.4669754556(50) \times 10^{-3}$		1.1×10^{-8}
uclear magneton ratio	$\mu_{ m d}/\mu_{ m N}$	0.8574382284(94)		1.1×10^{-8}
on-electron magnetic μ at ratio	$\mu_{ m d}/\mu_{ m e}$	$-4.664345537(50) \times 10^{-4}$		1.1×10^{-8}
on-proton magnetic μ at ratio	$\mu_{ m d}/\mu_{ m p}$	0.3070122083(45)		1.5×10^{-8}
on-neutron magnetic μ at ratio	$\mu_{ m d}/\mu_{ m n}$	-0.44820652(11)		2.4×10^{-7}
n, h				
$\mathrm{mass}^{\mathrm{f}}$	$m_{ m h}$	$5.00641174(39) \times 10^{-27}$	kg	7.9×10^{-8}
$m_{\rm h} = A_{\rm r}({\rm h}){\rm u}$ (helion elative atomic mass times u)		3.014 932 234 69(86)	u	2.8×10^{-10}
rgy equivalent	$m_{ m h}c^2$	$4.49953848(35)\times 10^{-10}$	J	7.9×10^{-8}
MeV		2808.39132(11)	${ m MeV}$	4.0×10^{-8}
electron mass ratio	$m_{ m h}/m_{ m e}$	5495.885238(12)		2.1×10^{-9}
proton mass ratio	$m_{ m h}/m_{ m p}$	2.99315265850(93)		3.1×10^{-10}
molar mass $N_{\rm A} m_{ m h}$	$M(\mathrm{h}), M_{\mathrm{h}}$	$3.01493223469(86)\times 10^{-3}$	$kg \cdot mol^{-1}$	2.8×10^{-10}
d helion magnetic moment $~\mu$ phere, 25 °C)	$\mu_{ m h}'$	$-1.074552967(45) \times 10^{-26}$	$J \cdot T^{-1}$	4.2×10^{-8}
Bohr magneton ratio	$\mu_{ m h}'/\mu_{ m B}$	$-1.158671474(14) \times 10^{-3}$		1.2×10^{-8}
uclear magneton ratio	$\mu_{ m h}'/\mu_{ m N}$	-2.127497718(25)		1.2×10^{-8}
d helion to proton tic moment ratio (gas, , 25 °C)	$\mu_{ m h}'/\mu_{ m p}$	-0.761766563(12)		1.5×10^{-8}
d helion to shielded proton tic moment ratio (2O, spheres, 25 °C)	$\mu_{ m h}'/\mu_{ m p}'$	-0.7617861313(33)		4.3×10^{-9}
d helion gyromagnetic $ \mu_{\rm h}' /\hbar$	$\gamma_{ m h}'$	$2.037894764(85)\times 10^8$	$\mathrm{s}^{-1}\cdot\mathrm{T}^{-1}$	4.2×10^{-8}
	$\gamma_{ m h}'/2\pi$	32.434 1025(14)	${ m MHz}\cdot{ m T}^{-1}$	4.2×10^{-8}
particle, α				
	m_{lpha}	$6.64465598(52)\times10^{-27}$	kg	7.9×10^{-8}
$m_{\alpha} = A_{r}(\alpha)u$ (alpha article relative atomic mass mes u)		4.001 506 1747(10)	u	2.5×10^{-10}
particle, α particle mass $u, m_{\alpha} = A_{r}(\alpha)u$ (alpha article relative atomic mass		$6.64465598(52) \times 10^{-27}$	kg	7.9 ×

 $^{^{\}mathrm{f}}$ The helion, symbol h, is the nucleus of the 3 He atom.

Quantity	Symbol	Value	Unit	Relative std. uncert. $\mu_{\rm r}$
energy equivalent	$m_{\alpha}c^2$	$5.97191897(47) \times 10^{-10}$	J	7.9×10^{-8}
in MeV		3727.37904(15)	${ m MeV}$	4.0×10^{-8}
alpha particle-electron mass ratio	$m_{lpha}/m_{ m e}$	7294.299508(16)		2.1×10^{-9}
alpha particle-proton mass ratio	$m_{lpha}/m_{ m p}$	3.9725996846(11)		2.8×10^{-10}
alpha particle molar mass $N_{\rm A} m_{\alpha}$	$M(\alpha), M_{\alpha}$	$4.0015061747(10) \times 10^{-3}$	$kg \cdot mol^{-1}$	2.5×10^{-10}
Physico-Chemical				
Avogadro constant	$N_{ m A}, L$	$6.02214199(47)\times10^{23}$	mol^{-1}	7.9×10^{-8}
atomic mass constant		$1.66053873(13)\times10^{-27}$	kg	7.9×10^{-8} 7.9×10^{-8}
atomic mass constant $m_{\rm u} = \frac{1}{12} m(^{12}{\rm C}) = 1{\rm u} = 10^{-3} \ {\rm kg \cdot mol}^{-1}/N_{\rm A}$	$m_{ m u}$	1.000 330 73(13) × 10	ку	7.9 × 10
energy equivalent	$m_{ m u}c^2$	$1.49241778(12) \times 10^{-10}$	J	7.9×10^{-8}
in MeV		931.494 013(37)	${ m MeV}$	4.0×10^{-8}
Faraday constant g $N_{\rm A}e$	F	$96\ 485.3415(39)$	$C \cdot \text{mol}^{-1}$	4.0×10^{-8}
molar Planck constant	$N_{ m A}h$	$3.990312689(30)\times 10^{-10}$	$J \cdot s \cdot mol^{-1}$	7.6×10^{-9}
	$N_{ m A}hc$	0.11962656492(91)	$J\cdot m\cdot mol^{-1}$	7.6×10^{-9}
molar gas constant	R	8.314472(15)	$J \cdot \mathrm{mol}^{-1} \cdot \mathrm{K}^{-1}$	1.7×10^{-6}
Boltzmann constant $R/N_{\rm A}$	k	$1.3806503(24)\times 10^{-23}$	${ m J\cdot K^{-1}}$	1.7×10^{-6}
in $eV \cdot K^{-1}$		$8.617342(15)\times 10^{-5}$	${ m eV\cdot K^{-1}}$	1.7×10^{-6}
	k/h	$2.0836644(36)\times 10^{10}$	$\mathrm{Hz}\cdot\mathrm{K}^{-1}$	1.7×10^{-6}
	k/hc	69.50356(12)	$\mathrm{m}^{-1}\cdot\mathrm{K}^{-1}$	1.7×10^{-6}
molar volume of ideal gas RT/p				
$T = 273.15 \; \mathrm{K}, p = 101.325 \; \mathrm{kPa}$	$V_{ m m}$	$22.413996(39) \times 10^{-3}$	$\mathrm{m}^3\cdot\mathrm{mol}^{-1}$	1.7×10^{-6}
Loschmidt constant $N_{\rm A}/V_{\rm m}$	n_0	$2.6867775(47)\times 10^{25}$	m^{-3}	1.7×10^{-6}
$T = 273.15 \; \mathrm{K}, p = 100 \; \mathrm{kPa}$	$V_{ m m}$	$22.710981(40) \times 10^{-3}$	$\mathrm{m}^3\cdot\mathrm{mol}^{-1}$	1.7×10^{-6}
Sackur-Tetrode constant (absolute entropy constant) ^h $\frac{5}{2} + \ln[(2\pi m_{\rm u}kT_1/h^2)^{3/2}kT_1/p_0]$				
$T_1 = 1 \text{ K}, p_0 = 100 \text{ kPa}$	S_0/R	-1.1517048(44)		3.8×10^{-6}
$T_1 = 1 \text{ K}, p_0 = 101.325 \text{ kPa}$		-1.1648678(44)		3.7×10^{-6}
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3c^2$	σ	$5.670400(40)\times10^{-8}$	$W \cdot m^{-2} \cdot K^{-4}$	7.0×10^{-6}
first radiation constant $2\pi hc^2$	c_1	$3.74177107(29) \times 10^{-16}$	$\mathrm{W}\cdot\mathrm{m}^2$	7.8×10^{-8}

gThe numerical value of F to be used in coulometric chemical measurements is 96 485.3432(76) $[7.9 \times 10^{-8}]$ when the relevant current is measured in terms of representations of the volt and ohm based on the Josephson and quantum Hall effects and the internationally adopted conventional values of the Josephson and von Klitzing constants K_{J-90} and R_{K-90} given in the "Adopted Values" table.

hThe entropy of an ideal monoatomic gas of relative atomic mass $A_{\rm r}$ is given by $S = S_0 + \frac{3}{2}R\ln A_{\rm r} - R\ln(p/p_0) + \frac{5}{2}R\ln(T/K)$.

Quantity	Symbol	Value	Unit	Relative std. uncert. $\mu_{\rm r}$
first radiation constant for spectral radiance $2hc^2$	$c_{ m 1L}$	$1.191042722(93) \times 10^{-16}$	$\mathrm{W}\cdot\mathrm{m}^2\cdot\mathrm{sr}^{-1}$	7.8×10^{-8}
second radiation constant hc/k	c_2	$1.4387752(25)\times 10^{-2}$	$m \cdot K$	1.7×10^{-6}
Wien displacement law constant $b = \lambda_{\text{max}} T = c_2/4.965 114 231 \dots$	b	$2.8977686(51) \times 10^{-3}$	$m \cdot K$	1.7×10^{-6}
Adopted Values				
molar mass of $^{12}\mathrm{C}$	$M(^{12}\mathrm{C})$	12×10^{-3}	$\mathrm{kg}\cdot\mathrm{mol}^{-1}$	(exact)
molar mass constant $M(^{12}C)/12$	$M_{ m u}$	1×10^{-3}	$kg \cdot mol^{-1}$	(exact)
conventional value of Josephson constant $^{\rm j}$	$K_{ ext{J-90}}$	483 597.9	$\mathrm{GHz}\cdot\mathrm{V}^{-1}$	(exact)
conventional value of von Klitzing constant k	$R_{ ext{K-90}}$	25 812.807	Ω	(exact)
standard atmosphere	p_0	101 325	Pa	(exact)
standard acceleration of gravity	$g_{ m n}$	9.80665	$\mathrm{m}\cdot\mathrm{s}^{-2}$	(exact)
X-ray Values				
Cu x unit: $\lambda(\text{CuK}\alpha_1)/1$ 537.400	$xu(CuK\alpha_1)$	$1.00207703(28) \times 10^{-13}$	m	2.8×10^{-7}
Mo x unit: $\lambda(\text{MoK}\alpha_1)/707.831$	$xu(MoK\alpha_1)$	$1.00209959(53)\times10^{-13}$	m	5.3×10^{-7}
ångstrom star: $\lambda({\rm WK}\alpha_1)/0.2090100$	Å*	$1.00001501(90)\times10^{-10}$	m	9.0×10^{-7}
lattice parameter $^{\rm l}$ of silicon (in vacuum, 22.5 °C)	a	$543.102088(16) \times 10^{-12}$	m	2.9×10^{-8}
$\{220\}$ lattice spacing of silicon $a/\sqrt{8}$ (in vacuum, 22.5 °C)	d_{220}	$192.0155845(56)\times10^{-12}$	m	2.9×10^{-8}
molar volume of silicon $M(\mathrm{Si})/\rho(\mathrm{Si}) = N_{\mathrm{A}}a^3/8$ (in	$V_{ m m}({ m Si})$	$12.0588369(14) \times 10^{-6}$	$\mathrm{m}^3\mathrm{mol}^{-1}$	1.2×10^{-7}

vacuum, 22.5 °C)

ⁱThe relative atomic mass $A_r(X)$ of particle X with mass m(X) is defined by $A_r(X) = m(X)/m_u$, where $m_u = m(^{12}C)/12 = M_u/N_A = 1$ u is the atomic mass constant, N_A is the Avogadro constant, and u is the atomic mass unit. Thus the mass of particle X in u is $m(X) = A_r(X)u$ and the molar mass of X is $M(X) = A_r(X)M_u$.

^jThis is the value adopted internationally for realizing representations of the volt using the Josephson effect.

^kThis is the value adopted internationally for realizing representations of the ohm using the quantum Hall effect.

¹This is the lattice parameter (unit cell edge length) of an ideal single crystal of naturally occurring Si free of impurities and imperfections, and is deduced from lattice spacing measurements on extremely pure and nearly perfect single crystals of Si by correcting for the effects of impurities.