

Model parameters

Diffusivities

- * Radial and axial diffusivity of oxygen in pear tissue

$$D_{u_r} = 2.8 \times 10^{-10} \text{ m}^2/\text{s} \quad , \quad D_{u_z} = 1.10 \times 10^{-9} \text{ m}^2/\text{s}$$

- * Radial and axial diffusivity of carbon dioxide in pear tissue

$$D_{v_r} = 2.32 \times 10^{-9} \text{ m}^2/\text{s} \quad , \quad D_{v_z} = 6.97 \times 10^{-9} \text{ m}^2/\text{s}$$

Respiration kinetic parameters

- * Maximum oxygen consumption rate

$$V_{mu} = V_{mu,ref} \exp \left[\frac{E_{a,vmu,ref}}{R_g} \left(\frac{1}{T_{ref}} - \frac{1}{T} \right) \right]$$

with T_{ref} a reference temperature (in degree Kelvin), T the actual temperature (in degree Kelvin) and $R_g = 8.314 \text{ J}/(\text{mol K})$ the universal gas constant.

- Maximum oxygen consumption rate at $T_{ref} = 293.15 \text{ K}$ ($= 20^\circ\text{C}$):

$$V_{mu,ref}: 2.39 \times 10^{-4} \text{ mol}/(\text{m}^3 \text{ s}).$$

- Activation energy for oxygen consumption:

$$E_{a,vmu,ref}: 80200 \text{ J/mol}.$$

- * Maximum fermentative carbon dioxide production rate:

$$V_{mf v} = V_{mf v,ref} \exp \left[\frac{E_{a,vmf v,ref}}{R_g} \left(\frac{1}{T_{ref}} - \frac{1}{T} \right) \right].$$

- Maximum fermentative carbon dioxide production rate at $T_{ref} = 293.15\text{K}$:

$$V_{mf v,ref}: 1.61 \times 10^{-4} \text{ mol}/(\text{m}^3 \text{ s}).$$

- Activation energy for fermentative carbon dioxide production:

$$E_{a,vmf v,ref}: 56700 \text{ J/mol}.$$

- * Michaelis-Menten constants and respiration quotient.

- Michaelis-Menten constant for oxygen consumption:

$$K_{mu}: 0.4103 \text{ mol}/\text{m}^3.$$

- Michaelis-Menten constant for non-competitive carbon dioxide inhibition:

$$K_{mv}: 27.2438 \text{ mol}/\text{m}^3.$$

- Michaelis-Menten constant of oxygen inhibition on fermentative carbon dioxide production:

$$K_{mf u}: 0.1149 \text{ mol}/\text{m}^3.$$

- The respiration quotient:

$$r_q: 0.97.$$

Convective mass transfer coefficients

$$h_u = 7 \times 10^{-7} \text{ m/s} \quad , \quad h_v = 7.5 \times 10^{-7} \text{ m/s}.$$

Ambient conditions

- * Atmospheric pressure

$$p_{atm} = 101300 \text{ Pa}.$$

- * Temperature

$$T = T_{cel} + 273.15$$

with T the temperature in degree Kelvin (K) and T_{cel} the temperature in degree Celsius ($^{\circ}$ C).

- * Ambient oxygen and carbon dioxide concentrations

$$C_{uamb} = \frac{p_{atm} \eta_u}{R_g T}$$

$$C_{vamb} = \frac{p_{atm} \eta_v}{R_g T}$$

with p_{atm} (Pa) the atmospheric pressure, η_u and η_v the fraction (percentage) oxygen and carbon dioxide in the (controlled) atmosphere, respectively; R_g the universal gas constant (J/(mol K)); and T the ambient temperature in degree Kelvin (K).

Note: $0 \leq \eta_u \leq 1$, $0 \leq \eta_v \leq 1$.

Simulations

Compute steady-state oxygen and carbon dioxide concentration profiles for the following storage conditions.

Storage description	$T_{cel}(^{\circ}\text{C})$	η_u (%)	η_v (%)
Orchard	25	20.8	0.04
Shelf life	20	20.8	0
Refrigerator	7	20.8	0
Precooling	-1	20.8	0
Disorder innducing	-1	2	5
Optimal CA	-1	2	0.7

Note: To compute C_{uamb} and C_{vamb} the fractions η_u and η_v should be numbers between 0 and 1.