

Homework 2 - Due 2/20/2013

John Kitchen

2013-02-13 Mon

1 Calculate the number of atoms/cm² in the following surfaces:

- a) FCC(100) Cu, Pt
- b) FCC(111) Cu, Pt
- c) BCC(100) Mo, W
- d) BCC(110) Mo, W

1.1 Solution :solution:

1.1.1 a)

Cu: 3.61 Å Pt: 3.92 Å

There are 2 atoms on the fcc(100) face, and the area of the face is a^2 . 1 Å = 1e-10 m

```
1 m = 1
2 cm = 0.01 * m
3 Ang = 1e-10 * m
4
5 print 'Cu(100): {0:1.2E} atoms/cm^2'.format(2.0 / (3.61 * Ang / cm)**2)
6 print 'Pt(100): {0:1.2E} atoms/cm^2'.format(2.0 / (3.92 * Ang / cm)**2)
```

Cu: 1.53E+15 atoms/cm²

Pt: 1.30E+15 atoms/cm²

1.1.2 b)

We need the area of an fcc(111) unit cell. We can compute this from the primitive lattice vectors, which form a primitive fcc(111) cell containing one atom.

```
1 import numpy as np
2 m = 1
3 cm = 0.01 * m
4 Ang = 1e-10 * m
5
```

```

6  a = 3.61
7  b = a / 2.0
8
9  v1 = np.array([0.0, b, b])
10 v2 = np.array([b, 0.0, b])
11 A = np.linalg.norm(np.cross(v1, v2))* Ang**2 /(cm**2)
12
13 print 'Cu(111): {0:1.2E} atoms/cm^2'.format(1.0 / A)
14
15 a = 3.92
16 b = a / 2.0
17
18 v1 = np.array([0.0, b, b])
19 v2 = np.array([b, 0.0, b])
20 A = np.linalg.norm(np.cross(v1, v2))* Ang**2 /(cm**2)
21 print 'Pt(111): {0:1.2E} atoms/cm^2'.format(1.0 / A)

```

```

Cu(111): 1.77E+15 atoms/cm^2
Pt(111): 1.50E+15 atoms/cm^2

```

The 111 surface is more densely packed than the (100) surface.

1.1.3 c)

Mo = 3.15 W = 3.16

The bcc(100) surface only has one atom in it, and the area is a^2 .

```

1  m = 1
2  cm = 0.01 * m
3  Ang = 1e-10 * m
4  print 'Mo(100): {0:2e} atoms/cm^2'.format(1.0 / (3.15**2 * Ang**2 / cm**2))
5  print 'W(100): {0:2e} atoms/cm^2'.format(1.0 / (3.16**2 * Ang**2 / cm**2))

```

```

Mo(100): 1.007811e+15 atoms/cm^2
W(100): 1.001442e+15 atoms/cm^2

```

1.1.4 d)

The bcc(110) surface has two atoms in it. The area of the unit cell is $a*\sqrt{(2)}*a$.

```

1  import numpy as np
2
3  m = 1
4  cm = 0.01 * m
5  Ang = 1e-10 * m
6  print 'Mo(110): {0:2e} atoms/cm^2'.format(2.0 / (3.15**2 * np.sqrt(2) * Ang**2 / cm**2))
7  print 'W(110): {0:2e} atoms/cm^2'.format(2.0 / (3.16**2 * np.sqrt(2) * Ang**2 / cm**2))

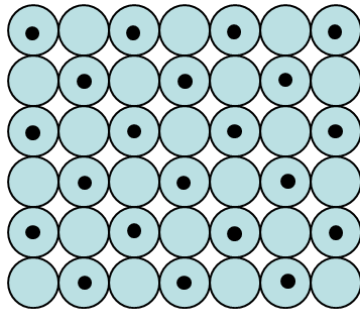
```

```

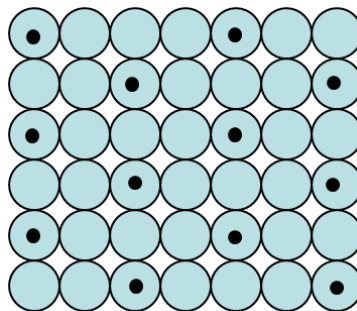
Mo(110): 1.425259e+15 atoms/cm^2
W(110): 1.416253e+15 atoms/cm^2

```

2 In the following image the large blue circles are the surface metal atoms and the small black circles represent adsorbates.



A.



B.

- Give an example of what surface the large circles could represent.
- Describe the overlayer in each case using matrix notation. If possible, also express the overlayer using Wood's notation.
- Calculate the reciprocal lattice vectors for each case.
- Sketch the LEED pattern you expect for these two surfaces.

2.1 solution :solution:

2.1.1 a. This could be an fcc or bcc (100) surface.

2.1.2 b)

on A:

$$\begin{bmatrix} 1 & 1 \\ 1 & -1 \end{bmatrix}$$

In Wood's notation, $p(\sqrt{2} \times \sqrt{2})R45^\circ$

on B:

$$\begin{bmatrix} 2 & 1 \\ 0 & 2 \end{bmatrix}$$

This can be described as $c(4 \times 2)$

2.1.3 c)

```
1 import numpy as np
2
3 uc1 = np.array([[1, 1], [1, -1]])
```

```

4
5 print np.linalg.inv(uc1.T)

```

```

[[ 0.5  0.5]
 [ 0.5 -0.5]]

```

```

1 import numpy as np
2
3 uc1 = np.array([[2, 1], [0, 2]])
4
5 print np.linalg.inv(uc1.T)

```

```

[[ 0.5  0. ]
 [-0.25 0.5 ]]

```

2.1.4 d)

