The ozone layer surrounding the earth's atmosphere acts as a shield by absorbing the harmful ultraviolet radiations of the sunlight to protect the living organisms from the harmful effects of UV rays.

The depletion or reduction in the ozone layer is caused due to the use of various ozonedepleting substances by human beings.

Advantage

The industries are the major sources of ozone-depleting substances such as chemical Chlorofluorocarbons (CFCs), carbon tetrachloride, methyl chloroform, methyl bromide, and nitrogen oxides, halogens, etc.

These chemicals continuously find a way to reach the ozone and react with it which has led to the depletion of ozone and an ozone hole is formed.

Ozone

Buy airconditioning and
refrigeration
equipment that do
not use HCFCs as
refrigerant.

Solutions

Depletion of the ozone layer has no effect on a region or a country. In reality, the entire planet is vulnerable to its consequences.

Buy aerosol products that do not use HCFCs or CFCs as propellants.

Increased levels of UV radiation result in a higher rate of skin cancer and eye issues.

Let's have a look at some of the ozone layer depletion solutions.

Ozone is formed when heat and sunlight cause chemical reactions between oxides of nitrogen (NOX) and Volatile Organic Compounds (VOC),

Causes

The nitrogenous compounds such as NO2, NO, N2O are highly responsible for the depletion of the ozone layer.

The ozone layer has been found to be depleted by certain natural processes such as Sun-spots and stratospheric winds.

It can worsen bronchitis,

emphysema and asthma.

reduce lung function and

inflame the linings of the

may permanently scar

lung tissue.

lungs. Repeated exposure

"Bad" ozone also can

Chlorofluorocarbons or

CFCs are the main

cause of ozone layer

depletion. These are

released by solvents,

spray aerosols,

refrigerators, air-

conditioners, etc.

Breathing ozone can trigger a variety of health problems including chest pain, coughing, throat irritation and congestion.

The ozone layer acts as a natural filter, absorbing most of the sun's burning ultraviolet (UV) rays.

Stratospheric ozone depletion leads to an increase in UV -B that reach the earth's surface, where it can disrupt biological processes and damage a number of materials.

Disadvantage