## Neutralisation Equations

ACID + BASE SALT	+ WATER
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Complete the following equations in both words and formulae:

1.	Hydrochloric Acid	+	Sodium Hydroxide	<b>→</b>	Sodium Chloride	+	Water
	H CI	+	Na OH	<b></b> ▶		+	
2.	Nitric Acid	+	Sodium Hydroxide	>	Sodium Nitrate	+	Water
	H NO <sub>3</sub>	+	Na OH	▶		+	
3.	Nitric Acid	+	Zinc Hydroxide	<del></del>		+	
	H NO <sub>3</sub>	+	Zn (OH) <sub>2</sub>	—▶	<del></del>	+	
4.	Sulphuric Acid	<del>- -</del>	Calcium Hydroxide	—>	-	+	***************************************
	H <sub>2</sub> SO <sub>4</sub>	+	Ca (OH) <sub>2</sub>	>		+	
5.	Hydrochloric Acid	+	Calcium Hydroxide	—-▶		+	
	H CI	+	Ca (OH) <sub>2</sub>	>		+	
6.	Sulphuric Acid	+	Ammonium Hydroxide	<b></b>		+	
	H <sub>2</sub> SO <sub>4</sub>	+	NH₄ OH	>		+	