

Ionic Formulae Worksheet (Colour Levels)

Use a method that suits you. Write the name of the compound along with your workings shown in your book.

<u>Red</u>	<u>Pink</u>	<u>Orange</u>	<u>Yellow</u>	<u>Green</u>	<u>Blue</u>	<u>Purple</u>
1. Hydrogen Fluoride 2. Hydrogen Chloride 3. Lithium Iodide 4. Potassium Bromide 5. Silver Chloride 6. Sodium Chloride	1. Barium Chloride 2. Calcium Iodide 3. Copper (II) Bromide 4. Nickel Bromide 5. Calcium Carbonate	1. Barium Oxide 2. Potassium Nitrate 3. Hydrogen Sulphate 4. Lithium Carbonate	1. Water 2. Calcium Fluoride 3. Magnesium Oxide 4. Silver Nitrate	1. Aluminium Bromide 2. Ammonium Chloride 3. Calcium Sulphate 4. Sodium Carbonate	1. Iron (ii) Oxide 2. Iron (iii) Oxide 3. Aluminium Sulphate	1. Copper (II) Carbonate 2. Aluminium Carbonate 3. Ammonium Sulphate
Attempt any 3, Get 3 right, progress to the next level!	Attempt any 3, Get 3 right, progress to the next level!	Attempt any 3, Get 3 right, progress to the next level!	Attempt any 2, Get 2 right, progress to the next level!	Attempt any 3, Get 3 right, progress to the next level!	Attempt all Get all correct, progress to the next level!	Attempt all Get all correct You have mastered Ionic Formulae!
						