

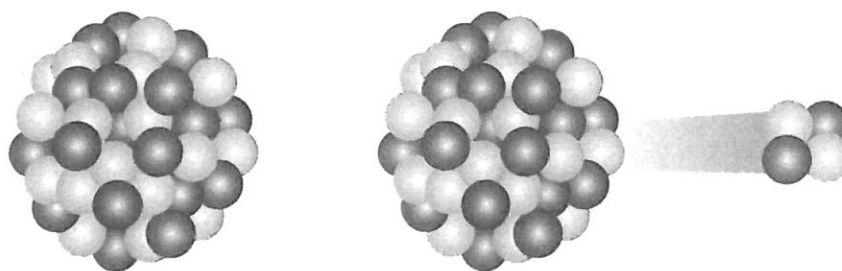
# 1.6

## Half-life decay

### Science inquiry

 Logical/mathematical  Visual/spatial

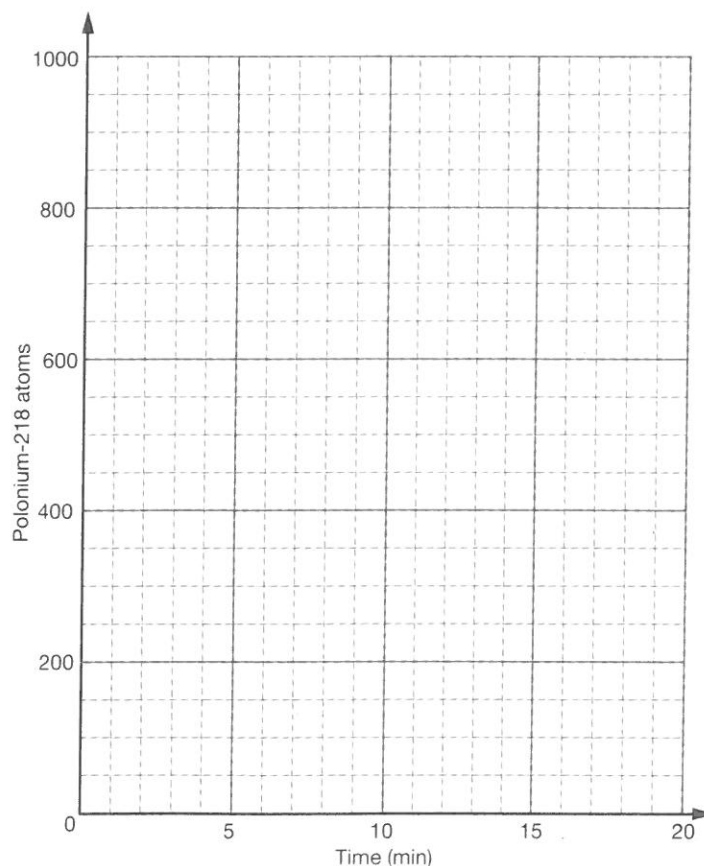
Jonathon has exactly 1000 polonium-218 atoms. He knows that polonium-218 decays rapidly into lead-214 through alpha decay.



To measure the half-life of polonium-218, Jonathon counts the number of polonium-218 atoms he has left after every 2 minutes. His results are written in the table below.

Time (min)	0	2	4	6	8	10	12	14	16	18	20
Number of polonium-218 atoms	1000	635	430	280	155	85	65	35	25	15	10

- Plot** the data in the table on the axes provided. Plot only the data points, with no connecting lines.



2 Draw a line of best fit through the data points, to show the general trend.

3 State how many polonium-218 atoms Jonathon has after:

(a) 4 min \_\_\_\_\_

(b) 14 min \_\_\_\_\_

(c) 9 min \_\_\_\_\_

(d) 19 min \_\_\_\_\_

4 Determine the time (min) when Jonathon has:

(a) one-half of the original number of polonium-218 atoms  
\_\_\_\_\_

(b) one-quarter of the original number of polonium-218 atoms  
\_\_\_\_\_

(c) one-eighth of the original number of polonium-218 atoms  
\_\_\_\_\_

(d) one-sixteenth of the original number of polonium-218 atoms  
\_\_\_\_\_

5 Use your graph to calculate the half-life of polonium-218.  
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