

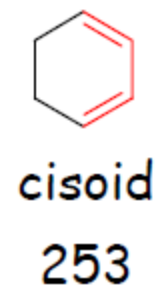
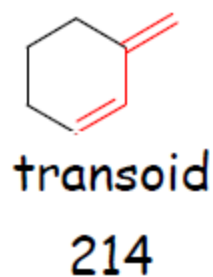
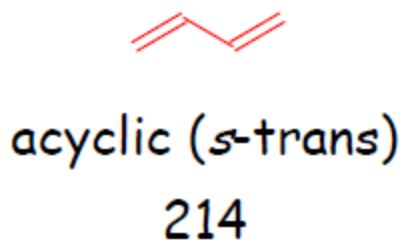
# **UV-Visible Spectroscopy**

# Prediction of $\lambda_{\max}$ : For Alkene

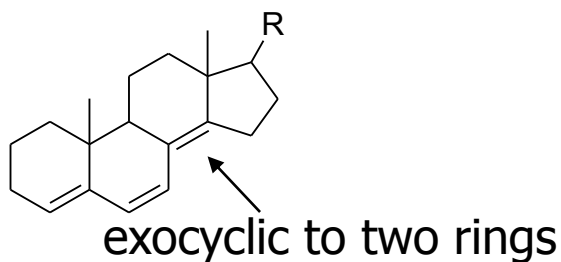
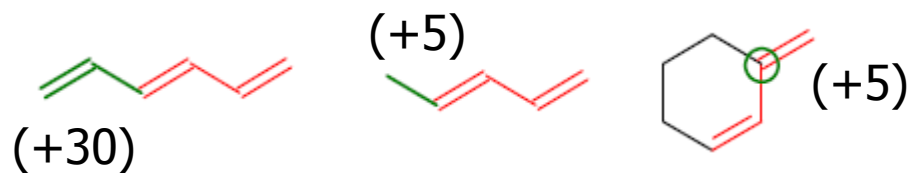
## Woodward-Fieser Rules:

For an empirical prediction of the wavelength for the lowest energy  $\pi \rightarrow \pi^*$  transition

A base value for  $\lambda_{\max}$  of the chromophore:

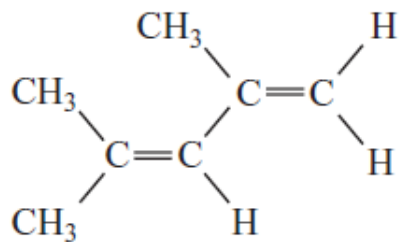


The incremental contribution of substituents is added to this base value from the group tables:



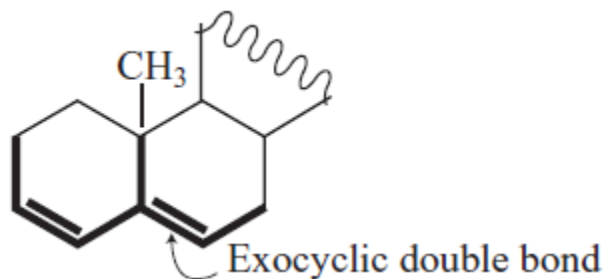
Group	Increment
Extended conjugation	+30
Each exo-cyclic C=C	+5
Alkyl	+5
-OCOCH <sub>3</sub>	+0
-OR	+6
-SR	+30
-Cl, -Br	+5
-NR <sub>2</sub>	+60

# Prediction of $\lambda_{\max}$ : Examples



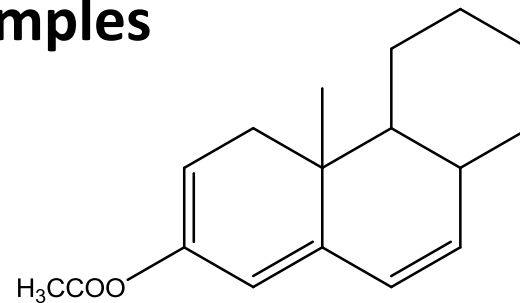
Transoid:	214 nm
Alkyl groups: $3 \times 5 =$	<u>15</u>
	229 nm

Observed: 228 nm

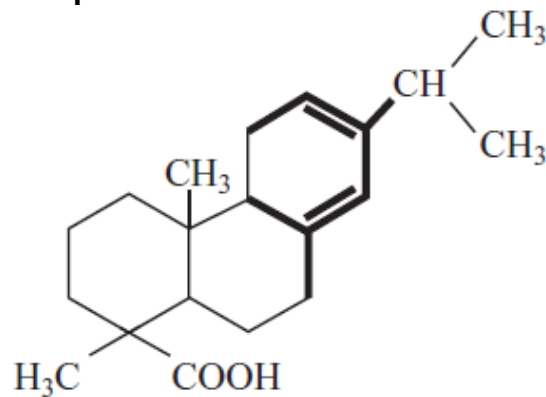


Transoid:	214 nm
Ring residues: $3 \times 5 =$	15
Exocyclic double bond:	<u>5</u>
	234 nm

Observed: 235 nm

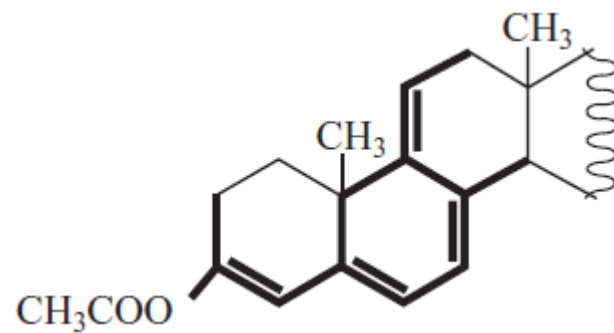


Homoannular diene	253 nm
Extra conjugation	30 nm
one exocyclic C=C	+ 5 nm
3 alkyl subs.	<u>+15 nm</u>
	303 nm
Experimental value	306 nm



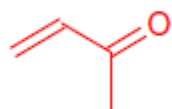
Cisoid:	253 nm
Alkyl substituent:	5
Ring residues: $3 \times 5 =$	15
Exocyclic double bond:	<u>5</u>
	278 nm
Observed:	275 nm

# Prediction of $\lambda_{\text{max}}$ : Examples



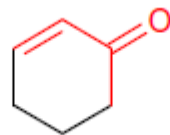
Observed: 355 nm

# Prediction of $\lambda_{\max}$ : For Unstaturated Ketone



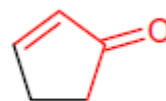
acyclic (*s*-trans)

215



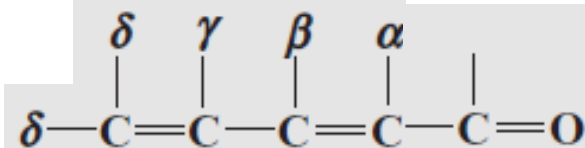
6-membered ring

215



5-membered ring

202



Double-bond-extending conjugation

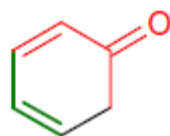
30

Alkyl group or ring residue

$\alpha$  10

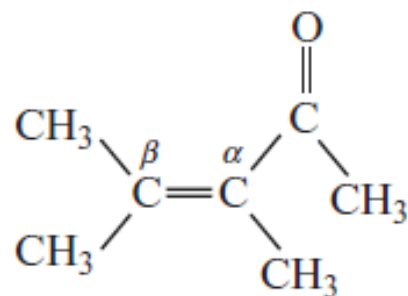
$\beta$  12

$\gamma$  and higher 18



homocyclic  
(cisoid) diene

39



Acyclic enone: 215 nm

$\alpha$ -CH<sub>3</sub>: 10

$\beta$ -CH<sub>3</sub>:  $2 \times 12 = \frac{24}{249 \text{ nm}}$

Observed: 249 nm

Polar groupings:

—OH  $\alpha$  35

$\beta$  30

$\delta$  50

—OCOCH<sub>3</sub>  $\alpha, \beta, \delta$  6

—OCH<sub>3</sub>  $\alpha$  35

$\beta$  30

$\gamma$  17

$\delta$  31

—Cl  $\alpha$  15

$\beta$  12

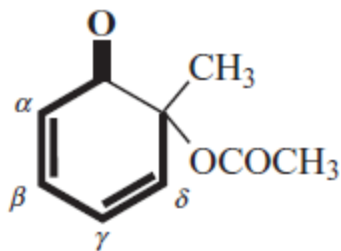
—Br  $\alpha$  25

$\beta$  30

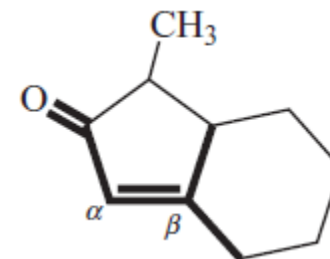
—NR<sub>2</sub>  $\beta$  95

Exocyclic double bond 5

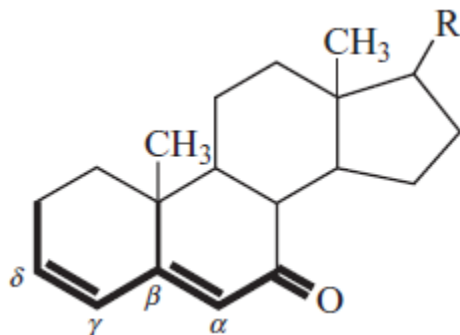
# Prediction of $\lambda_{\max}$ : For Unstaturated Ketone



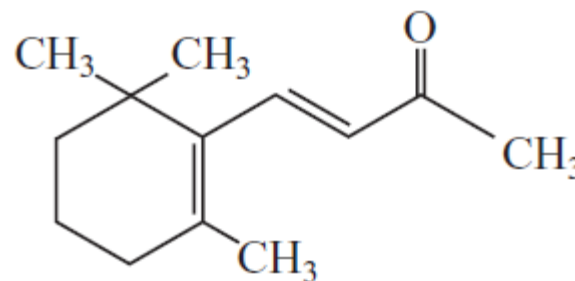
Six-membered enone:	215 nm
Double-bond-extending conjugation:	30
Homocyclic diene:	39
$\delta$ -Ring residue:	18
	<hr/>
	302 nm
Observed:	300 nm



Five-membered enone:	202 nm
$\beta$ -Ring residue: $2 \times 12 =$	24
Exocyclic double bond:	5
	<hr/>
	231 nm
Observed:	226 nm



Six-membered enone:	215 nm
Double-bond-extending conjugation:	30
$\beta$ -Ring residue:	12
$\delta$ -Ring residue:	18
Exocyclic double bond:	5
	<hr/>
	280 nm
Observed:	280 nm



Woodward-Fieser Rules is not valid for conjugation with more than four double bonds

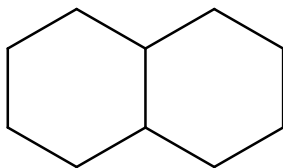
## Summary and Problem

*Bands of high intensity that appear above 210 nm generally represent either an unsaturated ketone, a diene, or a polyene.*

*Simple ketones, acids, esters, amides, and other compounds containing both  $p$  systems and unshared electron pairs show two absorptions: an  $n$  to  $\pi^*$  transition at longer wavelengths (>300 nm, low intensity) and a  $\pi$  to  $\pi^*$  transition at shorter wavelengths (<250 nm, high intensity).*

With conjugation (enones), the  $\lambda_{\text{max}}$  of the  $\pi$  to  $\pi^*$  band moves to longer wavelengths and can be predicted by Woodward's Rules.

A compound with the molecular formula of  $\text{C}_{10}\text{H}_{14}$  has the basic skeleton as shown below. Provide the exact molecular structure of the compound, if the  $\lambda_{\text{max}}$  for the compound is 234 nm.



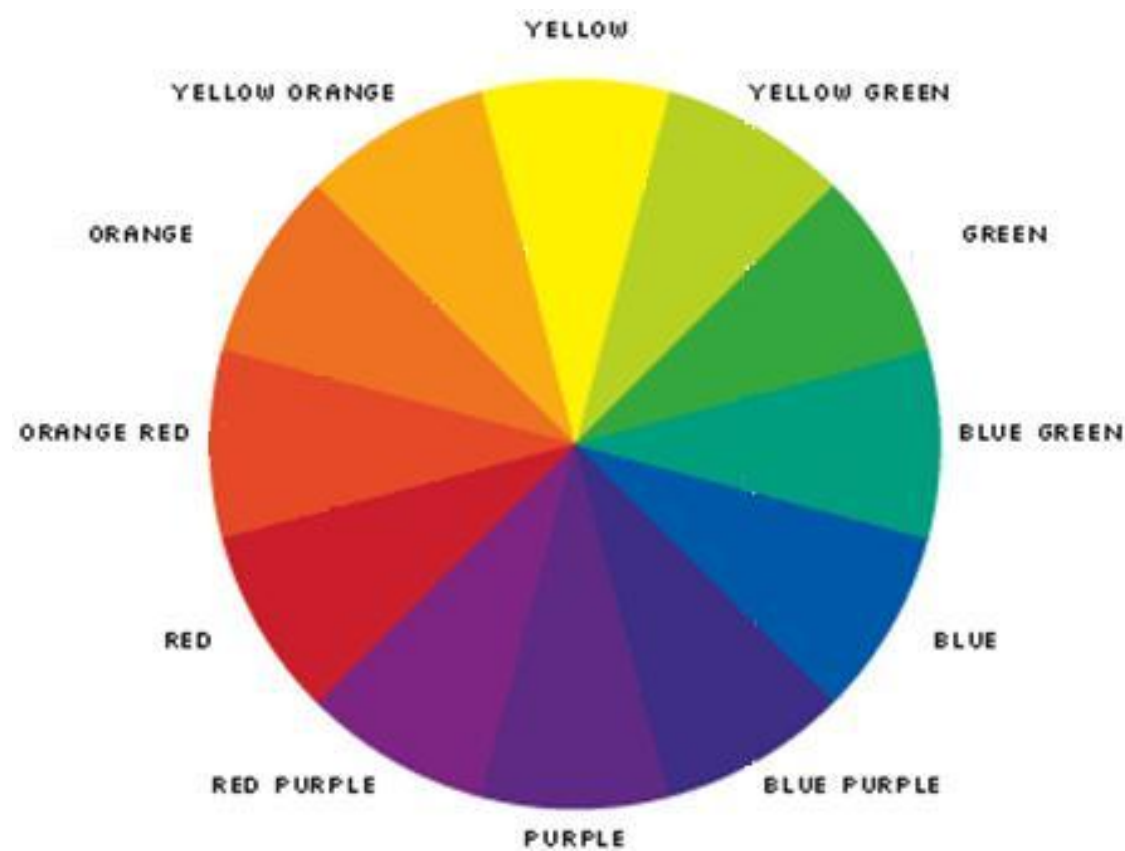
# VISIBLE SPECTRA: COLOR OF COMPOUNDS

electromagnetic spectrum lying between about 400 and 750 nm is the **visible** region  
Light waves with wavelengths between these limits appear colored to the human eye

If a substance absorbs visible light, it appears to have a color; if not, it appears white

compounds that absorb light in the visible region of the spectrum do not possess the color corresponding to the wavelength of the absorbed light

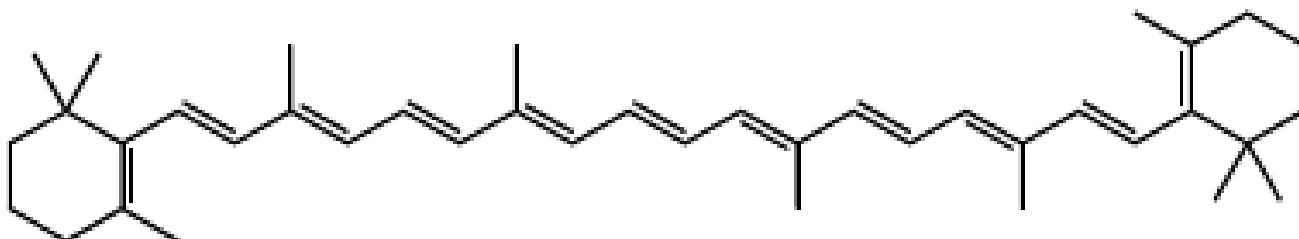
Compounds show  
**complementary** color



Absorbed color	$\lambda_{\text{abs}}$	Observed color
Violet	400	Yellow
Blue	450	Orange
Blue-green	500	Red
Yellow-green	530	Red-violet
Yellow	550	Violet
Orange-red	600	Blue-green
Red	700	Green



# VISIBLE SPECTRA: COLOR OF COMPOUNDS



**$\beta$ -carotene,  $\lambda_{\text{max}} = 455 \text{ nm}$**  (Blue region)

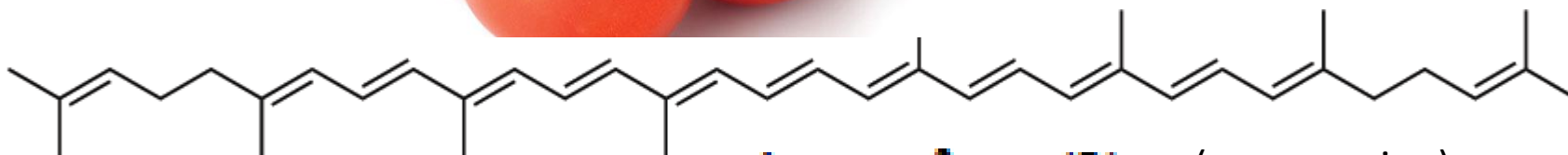


$$\lambda_{\text{max}} = 114 + 5(\# \text{ alkyl substituents}) + n(48 - 1.7n) - 16.5(\# \text{ endo}) - 10(\# \text{ exo})$$

Fieser-Kuhn Rules

$$= 114 + 5(10) + 11(48 - 1.7 \cdot 11) - 16.5(2) - 0$$

$$= 453.3 \text{ nm}$$

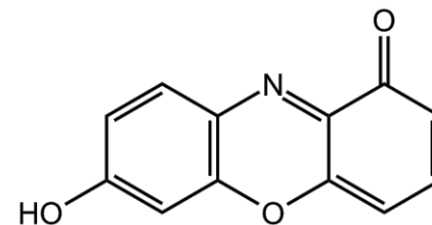
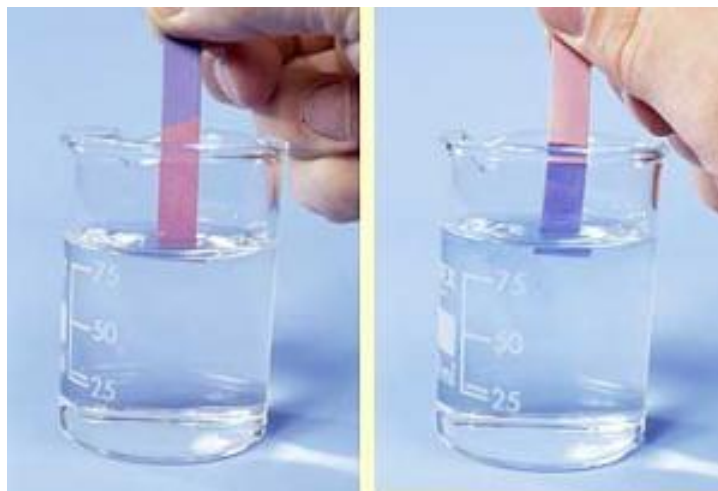


**lycopene,  $\lambda_{\text{max}} = 474 \text{ nm}$**  (green region)

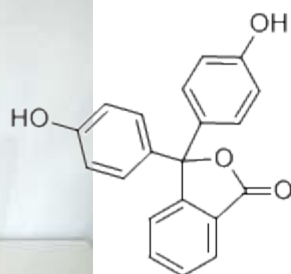
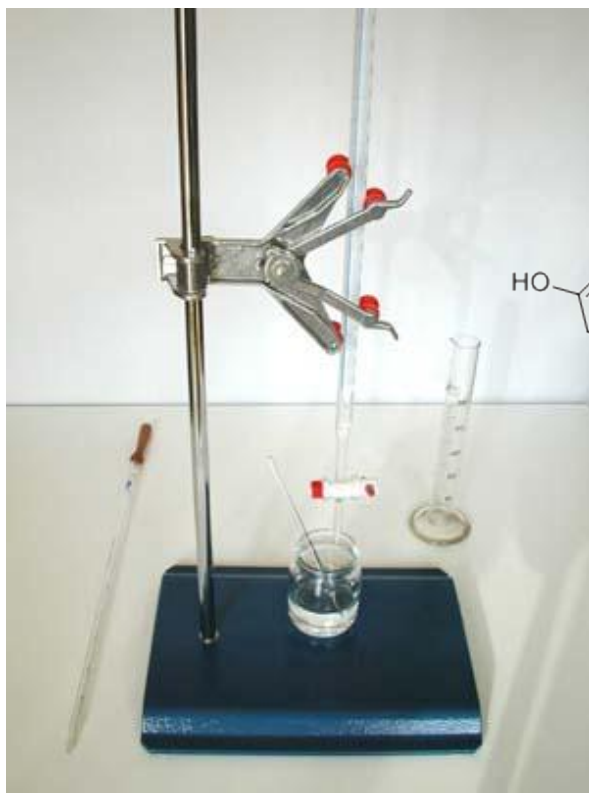
$$\lambda_{\text{max}} = 114 + 5(8) + 11(48 - 1.7 \cdot 11) - 0 - 0$$

$$= 476 \text{ nm}$$

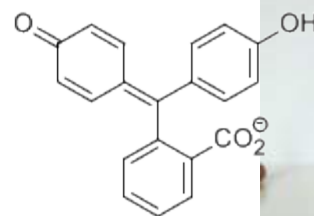
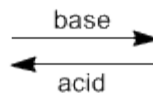
# VISIBLE SPECTRA: COLOR OF COMPOUNDS



7-hydroxyphenoxazone



(colorless)



(pink)



## Problem

A cyclic carbonyl compound **A** with molecular formula of  $C_5H_6O$ . It shows a strong absorption at 214 nm. If a hydrogen is replaced by a  $CH_3$  group, the  $\lambda_{max}$  shifted to 226 nm. Provide the exact molecular structure of the compound **A**, and the compound with  $\lambda_{max}=226$ .

# Looking forward

## IR and NMR spectroscopy

**Course material** will be uploaded **after 17:00 h** on **every Friday@**

<http://www.iitg.ac.in/ckjana/ckjana/Teaching.html>