Pericyclic Reaction

Pericyclic Reactions: π-Molecular Orbital

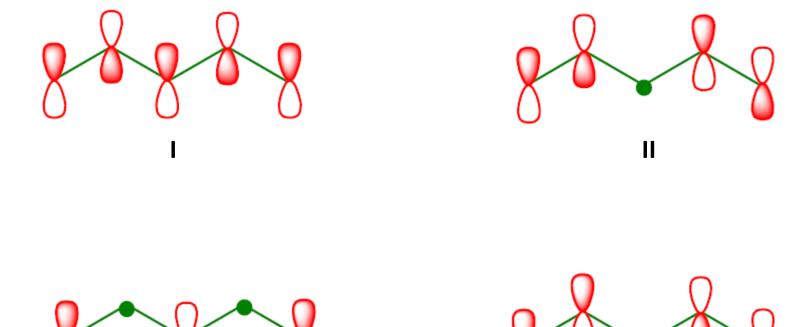


No of conjugated p orbitals $(n_p) = 5$ No of π -molecular orbitals = 5

$$n_{p}$$
is odd, $n_{BMO} = (n_{p}-1)/2 = 2$
 $n_{AMO} = (n_{p}-1)/2 = 2$
 $n_{NMO} = 1$

Pericyclic Reactions: π-Molecular Orbital

Indicate the following as BMO, AMO or NBO

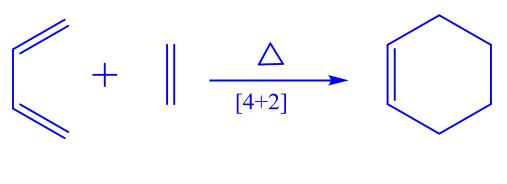


IV

Order of energy: I>II>III>IV

Ш

Pericyclic Reactions: Reactivity

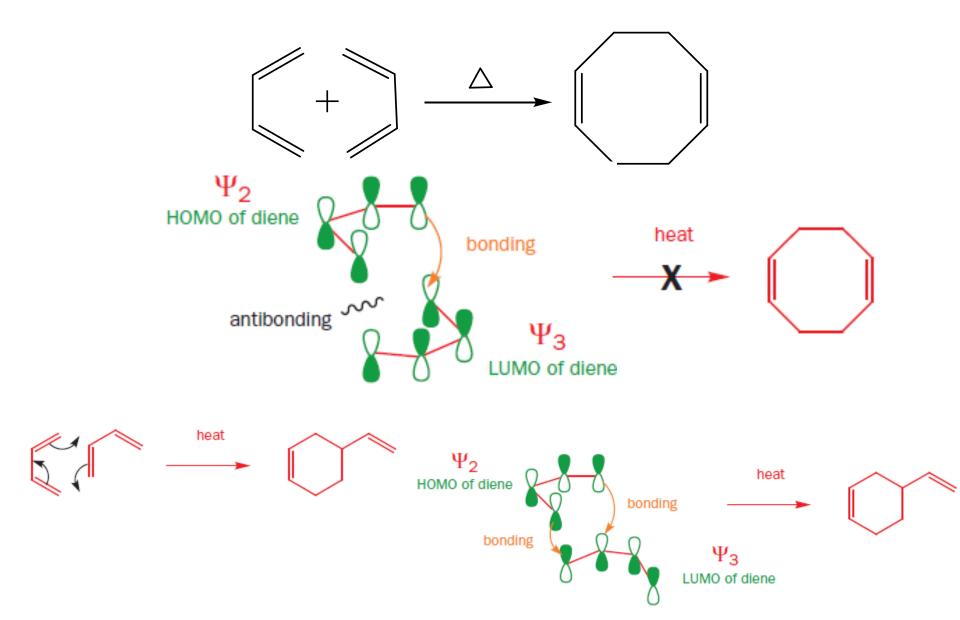


$$\| + \| \frac{\triangle}{[2+2]}$$

$$\left\| \begin{array}{c} + \\ \hline \end{array} \right\| \xrightarrow{\text{hv}} \left[\begin{array}{c} -1 \\ \hline \end{array} \right]$$

Pericyclic Reactions: Reactivity

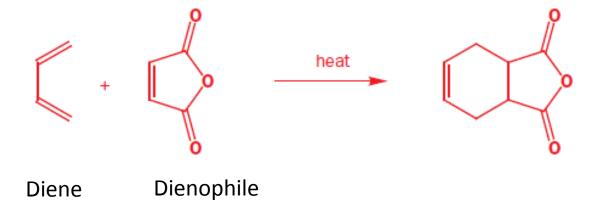
Based on FMO theory, predict whether the following reaction is feasible or not



Pericyclic Reactions: Reactivity

Based on FMO theory, predict whether the following reaction is feasible or not

Pericyclic Reactions: Selectivity of Diels-Alder Reaction





worked at the University of Kiel and discovered this reaction in 1928. They won the **Nobel Prize** in 1950.



Otto Diels (1876–1954)

Kurt Alder (1902–58)

Pericyclic Reactions: Selectivity in Diels-Alder Reaction

Pericyclic Reactions: Selectivity in Diels-Alder Reaction

Geometry of dienophile is retained: cis-alkene give cis-product and trans-alkene gives trans-product

Pericyclic Reactions: Selectivity in Diels-Alder Reaction

OAC
$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

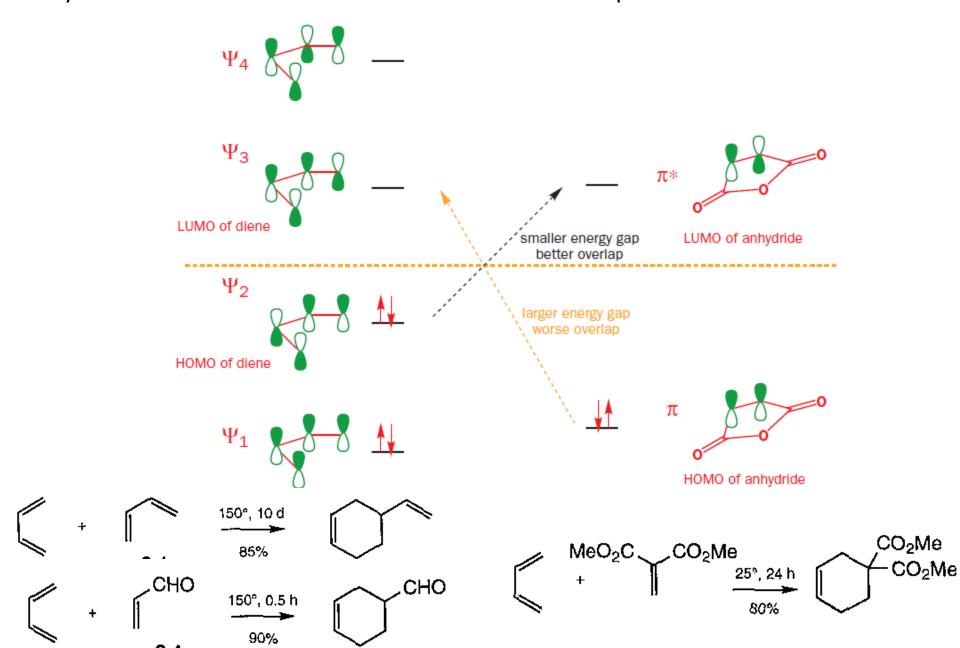
$$CO_{2}CH_{3}$$

$$CO_{2}CH_{3}$$

$$R_1$$
 R_2
 R_3

Pericyclic Reactions: Selectivity of Diels-Alder Reaction

Why HOMO of diene interact with the LUMO of Dienophile?



Looking forward

Pericyclic Reaction:

Reactivity and Selectivity of electrocyclic reaction Pericyclic Reactions in Biological Systems

Course material will be uploaded after 17:00 h on every Friday @

http://www.iitg.ac.in/ckjana/ckjana/Teaching.html