## Quantum Mechanics (PH101) Course Instructors: Pankaj Mishra and Tapan Mishra Tutorial-8 due on Wednesday, 6th of November, 2019 (8:00Hrs IST)

- 1. The lifetime of a given atom in an excited state is  $10^{-8}$  s. It comes to the ground state by emitting a photon of wavelength 5800 Å . Find the energy uncertainty and wavelength uncertainty of the photon. Use the minimum time-Energy uncertainty principle  $\Delta E \Delta t = \hbar/2$ .
- 2. Find the uncertainty in the velocity of a particle if the uncertainty in its position is equal to its (a) de Broglie wavelength (b) Compton wavelength. Use the minimum position and momentum uncertainty relation.
- 3. Check if  $\Psi = Ae^{i(kx-\omega t)}$  and  $\Psi = Asin(kx-\omega t)$  are acceptable solutions of the time-dependent Schroedinger's equation. The time-dependent Schroedinger'e equation is given by

$$i\hbar\frac{\partial\Psi}{\partial t} = -\frac{\hbar^2}{2m}\frac{\partial^2\Psi}{\partial^2x} + U\Psi$$

4. The normalized wave function of the ground state of the Quantum harmonic oscillator is given by  $\psi(x) = C_0 e^{-\alpha x^2}$ , where  $C_0 = \left(\frac{m\omega}{\pi\hbar}\right)^{1/4}$  and  $\alpha = \frac{m\omega}{2\hbar}$ . m is the mass and  $\omega$  is the angular frequency of the oscillator.

Compute the  $\Delta x \Delta p$  for this state, where  $\Delta x$  and  $\Delta p$  are the uncertainties in the position x and momentum p, respectively. Please comment over the result whether it is consistent with the uncertainty principle. Use the Gaussian integral  $\int_{-\infty}^{\infty} e^{\alpha x^2 + \beta x} dx = \sqrt{\frac{\pi}{\alpha}} e^{\beta^2/4\alpha}$ .

5. An electron is described by the wave function

$$\psi(x) = \begin{cases} 0, & \text{for } x \le 0\\ Ce^{-x}(1 - e^{-x}), & \text{for } x > 0, \end{cases}$$

where x is in nm and C is a constant.

- (a) Determine the value of C that normalizes  $\psi(x)$ .
- (b) Where is the electron most likely to be found?
- (c) Calculate the average position or expectation value of the position  $\langle x \rangle$  for the electron. Compare this with the most likely position, and comment on the difference.

6. A particle is represented by the wavefunction at time t=0 by

 $\Psi(x) = A(a^2 - x^2)$  if  $-a \le x \le a$  and zero at all other places. Here A and a are constant.

- (a) Determine the normalization constant A.
- (b) What is the expectation value of x at t = 0?
- (c) What is the expectation value of p at t = 0?
- (d) Evaluate  $\langle x^2 \rangle$  and  $\langle p^2 \rangle$  at t = 0.
- (e) Obtain the uncertainty relation  $(\Delta x \Delta p)$  and comment over your result whether you are getting minimum uncertainty relation or not.