

1. The lifetime of a given atom in an excited state is  $10^{-8} \text{ s}$ . It comes to the ground state by emitting a photon of wavelength  $5800 \text{ \AA}$ . Find the energy uncertainty and wavelength uncertainty of the photon. Use the minimum time-Energy uncertainty principle  $\Delta E \Delta t = \hbar/2$ .

**Solution:**

Here we will use the time-Energy uncertainty principle  $\Delta E \Delta t = \hbar/2$ .

For the given problem the photon can be emitted at any instant during the time interval  $\Delta t = 10^{-8} \text{ s}$ .

$\therefore$  The energy uncertainty of the photon is

$$\Delta E = \frac{\hbar}{2\Delta t} = \frac{0.527 \times 10^{-34} \text{ J.s}}{10^{-8} \text{ s}} = 0.527 \times 10^{-26} \text{ J}$$

If  $\lambda$  is the wavelength of the photon then,

$$E = \frac{hc}{\lambda} \Rightarrow \lambda = \frac{hc}{E} \Rightarrow \Delta \lambda = hc \left( \frac{-\Delta E}{E^2} \right)$$

So the uncertainty in wavelength is

$$\begin{aligned} \Delta \lambda &= \frac{hc}{E^2} \Delta E = \frac{hc\lambda^2}{h^2c^2} \Delta E = \frac{\lambda^2}{hc} \Delta E \\ &= \frac{(5.8 \times 10^{-7} \text{ m})^2 \times (0.527 \times 10^{-26} \text{ J})}{(6.62 \times 10^{-34} \text{ J.s}) \times (3 \times 10^8 \text{ m/s})} = 0.89 \times 10^{-14} \text{ m} \sim 9 \times 10^{-8} \text{ \AA} \end{aligned}$$

2. Find the uncertainty in the velocity of a particle if the uncertainty in its position is equal to its (a) de Broglie wavelength (b) Compton wavelength. Use the minimum position and momentum uncertainty relation.

**Solution:**

From the uncertainty principle we know that

$$\Delta x \Delta p = \hbar/2$$

$$\Delta v = \frac{\hbar}{2m\Delta x}$$

(a) If the uncertainty in position is equal to the de Broglie wavelength

$$\Delta x = \lambda = h/mv$$

So the uncertainty in velocity is

$$\Delta v = \frac{\hbar}{2m\Delta x} = v/4\pi$$

(b) If the uncertainty in position is equal to the Compton wavelength

$$\Delta x = \lambda_C = \frac{h}{mc}$$

So,

$$\Delta v = \frac{\hbar}{2m\Delta x} = c/4\pi$$

Therefore, the uncertainty in velocity is of the order to the speed of light in vacuum.

3. Check if  $\Psi = Ae^{i(kx-\omega t)}$  and  $\Psi = A\sin(kx - \omega t)$  are acceptable solutions of the time-dependent Schroedinger's equation. The time-dependent Schroedinger's equation is given by

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + U\Psi$$

**Solution:**

The time-dependent Schroedinger's equation is given by

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + U\Psi$$

**For  $\Psi = Ae^{i(kx-\omega t)}$ :**

$$\frac{\partial \Psi}{\partial t} = -i\omega Ae^{i(kx-\omega t)} = -i\omega \Psi$$

$$\frac{\partial \Psi}{\partial x} = ik\Psi$$

$$\frac{\partial^2 \Psi}{\partial x^2} = i^2 k^2 \Psi = -k^2 \Psi$$

Inserting these in the Schroedinger's equation yields

$$\begin{aligned} i\hbar(-i\omega\Psi) &= -\frac{\hbar^2}{2m}(-k^2\Psi) + U\Psi \\ \Rightarrow \left(\hbar\omega - \frac{\hbar^2 k^2}{2m} - U\right)\Psi &= 0 \end{aligned}$$

Using  $E = h\nu = \hbar\omega$  and  $p = \hbar k$ , we obtain

$$\left(E - \frac{p^2}{2m} - U\right)\Psi = 0$$

Note that the quantity on the left is equal to zero for the non-relativistic case as  $E = K.E + U = \frac{p^2}{2m} + U$ .

Thus  $\Psi = Ae^{i(kx - \omega t)}$  is a solution of the Schroedinger's equation.

**For  $\Psi = A\sin(kx - \omega t)$ :**

Following the similar process as above we will reach at

$$-i\hbar\omega\cos(kx - \omega t) = \left(\frac{\hbar^2 k^2}{2m} + U\right)\sin(kx - \omega t)$$

This equation is generally not satisfied for all  $x$  and  $t$ . Hence  $\Psi = A\sin(kx - \omega t)$  is not an acceptable solution of the time-dependent Schroedinger equation. This function however, is a solution of the classical wave equation.

4. The normalized wave function of the ground state of the Quantum harmonic oscillator is given by  $\psi(x) = C_0 e^{-\alpha x^2}$ , where  $C_0 = \left(\frac{m\omega}{\pi\hbar}\right)^{1/4}$  and  $\alpha = \frac{m\omega}{2\hbar}$ .  $m$  is the mass and  $\omega$  is the angular frequency of the oscillator.

Compute the  $\Delta x \Delta p$  for this state, where  $\Delta x$  and  $\Delta p$  are the uncertainties in the position  $x$  and momentum  $p$ , respectively. Please comment over the result whether it is consistent with the uncertainty principle. **Use the Gaussian integral**  $\int_{-\infty}^{\infty} e^{\alpha x^2 + \beta x} dx = \sqrt{\frac{\pi}{-\alpha}} e^{\beta^2 / 4\alpha}$ .

**Solution:** We have  $\langle x \rangle = \int_{-\infty}^{\infty} x |\psi(x)|^2 dx = 0$  as integrand is odd function.

$\langle x^2 \rangle = \int_{-\infty}^{\infty} x^2 |\psi(x)|^2 dx$ . Solving the integral by integration by parts and using the Gaussian integral formula we have  $\langle x^2 \rangle = \frac{\hbar}{2m\omega}$ .

Now  $\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2} = \sqrt{\frac{\hbar}{2m\omega}}$ .

$\langle p \rangle = -i\hbar \int_{-\infty}^{\infty} \psi^*(x) \frac{d\psi(x)}{dx} dx = 0$ . Again integrand here is odd function.

$\langle p^2 \rangle = -\hbar^2 \int_{-\infty}^{\infty} \psi^*(x) \frac{d^2\psi(x)}{dx^2} dx$ . Again performing the integral by parts and using the Gaussian integral we have,

$$\langle p^2 \rangle = \frac{m\hbar\omega}{2}.$$

Therefore,  $\Delta p = \sqrt{(\langle p^2 \rangle - \langle p \rangle^2)} = \sqrt{\frac{m\hbar\omega}{2}}$ .

Now  $\Delta x \Delta p = \hbar/2$ . This is, for the optimum state (Gaussian state) the product of the uncertainties of position and momentum is the smallest value allowed by Heisenberg's Uncertainty relation:

$$\Delta p \Delta x \geq \hbar/2.$$

5. An electron is described by the wave function

$$\psi(x) = \begin{cases} 0, & \text{for } x \leq 0 \\ Ce^{-x}(1 - e^{-x}), & \text{for } x > 0, \end{cases}$$

where  $x$  is in  $nm$  and  $C$  is a constant.

(a) Determine the value of  $C$  that normalizes  $\psi(x)$ .

(b) Where is the electron most likely to be found?

(c) Calculate the average position or expectation value of the position  $\langle x \rangle$  for the electron. Compare this with the most likely position, and comment on the difference.

**Solution:**

(a) We have the normalization condition:

$$\int_{-\infty}^{\infty} |\psi|^2 dx = 1. \text{ We get } |C| = 2\sqrt{3} \text{ nm}^{-1/2}.$$

(b) The most likely place  $x_m$  for the electron to be is where  $|\psi(x)|^2$  is maximum, or, in this case where  $\psi(x)$  is maximum. We have

$$\frac{d\psi(x)}{dx} = 0 \Rightarrow C[e^{-2x} - e^{-x}(1 - e^{-x})] = 0 \Rightarrow x = \ln 2 \quad nm = 0.693 \quad nm.$$

(c) Since electron state is in the stationary state. Its average position is given by

$$\langle x \rangle = \int_{-\infty}^{\infty} \psi(x)x\psi^*(x) = C^2 \int_0^{\infty} xe^{-2x}[1 - e^{-2x}]^2 = 12 \int_0^{\infty} xe^{-2x}[1 - 2e^{-x} + e^{-2x}]$$

Using integration by parts we have,

$$\langle x \rangle = C^2 \left[ \frac{1}{4} - \frac{2}{9} + \frac{1}{16} \right] = \frac{13}{12} \quad nm \simeq 1.083 \quad nm$$

6. A particle is represented by the wavefunction at time  $t = 0$  by

$\Psi(x) = A(a^2 - x^2)$  if  $-a \leq x \leq a$  and zero at all other places. Here  $A$  and  $a$  are constant.

(a) Determine the normalization constant  $A$ .

(b) What is the expectation value of  $x$  at  $t = 0$ ?

(c) What is the expectation value of  $p$  at  $t = 0$ ?

(d) Evaluate  $\langle x^2 \rangle$  and  $\langle p^2 \rangle$  at  $t = 0$ .

(e) Obtain the uncertainty relation  $(\Delta x \Delta p)$  and comment over your result whether you are getting minimum uncertainty relation or not.

**Solution:**

(a) The normalization condition is

$$\begin{aligned} 1 &= \int_{-\infty}^{\infty} |\Psi(x)|^2 dx \\ &= \int_{-a}^a |\Psi(x)|^2 dx \\ &= \int_{-a}^a A^2(a^2 - x^2)^2 dx = 2A^2 \left[ a^5 - \frac{2}{3}a^5 + \frac{a^5}{5} \right] = \frac{16}{15}A^2a^5 \\ \Rightarrow A &= \sqrt{\frac{15}{16a^5}} \end{aligned}$$

(b) The expectation value of  $x$  at  $t = 0$  is

$$\langle x \rangle = \int_{-\infty}^{\infty} x |\Psi(x)|^2 dx$$

$$\langle x \rangle = \int_{-a}^a x A^2 (a^2 - x^2)^2 dx$$

This integral is zero since the integrand is an odd function of  $x$ .

(c)  $\langle p \rangle$  will also be zero due to the above reason given in (b).

(d) Expectation value of  $x^2$  at  $t = 0$  is

$$\begin{aligned} \langle x^2 \rangle &= \int_{-\infty}^{\infty} \Psi^*(x) x^2 \Psi(x) dx = \int_{-\infty}^{\infty} x^2 |\Psi(x)|^2 dx \\ &= \int_{-a}^a x^2 A^2 (a^2 - x^2)^2 dx = \frac{a^2}{7} \end{aligned}$$

(e) Expectation value of  $p^2$  at  $t = 0$  is

$$\begin{aligned} \langle p^2 \rangle &= \int_{-\infty}^{\infty} \Psi^*(x) p^2 \Psi(x) dx = \int_{-a}^a \Psi^*(x) \left( -\hbar^2 \frac{d^2}{dx^2} \right) \Psi(x) dx \\ &= \frac{5\hbar^2}{2a^2} \end{aligned}$$

$$\begin{aligned} \text{(f) } (\Delta x)^2 &= \langle x^2 \rangle - \langle x \rangle^2 = \langle x^2 \rangle = \frac{a^2}{7} \\ (\Delta p)^2 &= \langle p^2 \rangle - \langle p \rangle^2 = \langle p^2 \rangle = \frac{5\hbar^2}{2a^2} \\ (\Delta x)^2 (\Delta p)^2 &= \frac{5}{14} \hbar^2 \\ \Delta x \Delta p &= \sqrt{\frac{5}{14}} \hbar \end{aligned}$$

**Note that here we get  $\Delta x \Delta p > \hbar/2$  because the given wave function is non-Gaussian.**