Lanthanide Ground Shell Electron Configuration per Ionization State

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June 24, 2020

| | Charge States | | | | | | | | | |
|-------------------|-------------------------|--------------------------|-------------------|----------------|--------------------|------------------------------|----------------------------|------------------------------|----------------------------|---------------------------------------|
| Element Name (Z) | I | II | III | IV | V | VI | VII | VIII | IX | X |
| Lanthanum (57) | [Xe] $5d^16s^2$ | $[Xe]5d^2$ | $[Xe]5d^1$ | $[Cd]5p^6$ | $[Cd]5p^5$ | $[\mathrm{Cd}]5p^4$ | $[\mathrm{Cd}]5p^3$ | $[\mathrm{Cd}]5p^2$ | $[\mathrm{Cd}]5p^1$ | $\left[\text{Kr} \right] 4d^1 05s^2$ |
| Cerium (58) | $[Xe]4f^15d^16s^2$ | [Xe] $4f^15d^2$ | $[Xe]4f^2$ | $[Xe]4f^1$ | $[Cd]5p^6$ | $[\mathrm{Cd}]5p^5$ | $[\mathrm{Cd}]5p^4$ | $[\mathrm{Cd}]5p^3$ | $[\mathrm{Cd}]5p^2$ | $[\mathrm{Cd}]5p^1$ |
| Praseodymium (59) | $[Xe]4f^36s^2$ | $[Xe]4f^36s^1$ | [Xe] $4f^{3}$ | $[Xe]4f^2$ | $[Xe]4f^1$ | $[\mathrm{Cd}]5p^6$ | $[\mathrm{Cd}]4f^15p^4$ | $[\mathrm{Cd}]4f^15p^3$ | $[\mathrm{Cd}]4f^15p^2$ | $[\mathrm{Cd}]4f^2$ |
| Neodymium (60) | [Xe] $4f^46s^2$ | [Xe] $4f^46s^1$ | $[Xe]4f^4$ | [Xe] $4f^{3}$ | $[Xe]4f^2$ | $[\mathrm{Cd}]4f^25p^5$ | $[\mathrm{Cd}]4f^25p^4$ | $[\mathrm{Cd}]4f^25p^3$ | $[\mathrm{Cd}]4f^25p^2$ | $\boxed{ [\mathrm{Cd}]4f^25p^1}$ |
| Promethium (61) | $[{ m Xe}]4f^{5}6s^{2}$ | $[Xe]4f^56s^1$ | [Xe] $4f^{5}$ | $[Xe]4f^4$ | $[Xe]4f^3$ | $[\mathrm{Cd}]4f^35p^5$ | $[\mathrm{Cd}]4f^35p^4$ | $[\mathrm{Cd}]4f^35p^3$ | $[\mathrm{Cd}]4f^35p^2$ | $\boxed{ [\mathrm{Cd}]4f^35p^1}$ |
| Samarium (62) | $[Xe]4f^66s^2$ | $[Xe]4f^66s^1$ | [Xe] $4f^{6}$ | [Xe] $4f^{5}$ | [Xe] $4f^4$ | [Xe] $4f^{3}$ | $[\mathrm{Cd}]4f^45p^4$ | $[\mathrm{Cd}]4f^45p^3$ | $[\mathrm{Cd}]4f^45p^2$ | $\boxed{ [\mathrm{Cd}]4f^45p^1}$ |
| Europium (63) | $[Xe]4f^{7}6s^{2}$ | $[Xe]4f^{7}6s^{1}$ | [Xe] $4f^{7}$ | [Xe] $4f^{6}$ | [Xe] $4f^{5}$ | $[{ m Cd}]4f^{5}5p^{5}$ | $[\mathrm{Cd}]4f^55p^4$ | $[\mathrm{Cd}]4f^55p^3$ | $[\mathrm{Cd}]4f^55p^2$ | $\left[\text{Cd} \right] 4f^5 5p^1$ |
| Gadolinium (64) | $[Xe]4f^75d^16s^2$ | $[Xe]4f^{7}5d^{1}6s^{1}$ | [Xe] $4f^75d^1$ | $[Xe]4f^7$ | [Xe] $4f^6$ | $[{\rm Cd}]4f^65p^5$ | $[\mathrm{Cd}]4f^65p^4$ | $[\mathrm{Cd}]4f^65p^3$ | $[\mathrm{Cd}]4f^65p^2$ | $\boxed{ [\mathrm{Cd}]4f^55p^2}$ |
| Terbium (65) | $[Xe]4f^{9}6s^{2}$ | $[Xe]4f^96s^1$ | $[Xe]4f^9$ | [Xe] $4f^{8}$ | $[Xe]4f^7$ | $[\mathrm{Cd}]4f^75p^5$ | $[\mathrm{Cd}]4f^85p^3$ | $[\mathrm{Cd}]4f^75p^3$ | $[\mathrm{Cd}]4f^75p^2$ | $\boxed{ [\mathrm{Cd}]4f^65p^2}$ |
| Dysprosium (66) | $[Xe]4f^{10}6s^2$ | [Xe] $4f^{10}6s^1$ | [Xe] $4f^{10}$ | [Xe] $4f^9$ | [Xe] $4f^{8}$ | $[{\rm Cd}]4f^85p^5$ | $[{\rm Cd}]4f^85p^4$ | $[\mathrm{Cd}]4f^95p^2$ | $[\mathrm{Cd}]4f^85p^2$ | [Cd] $4f^85p^1$ |
| Holmium (67) | $[Xe]4f^{11}6s^2$ | [Xe] $4f^{11}6s^1$ | [Xe] $4f^{11}$ | [Xe] $4f^{10}$ | $[{ m Xe}]4f^9$ | $[{\rm Cd}]4f^95p^5$ | $[\mathrm{Cd}]4f^95p^4$ | $[\mathrm{Cd}]4f^95p^3$ | $[Cd]4f^{10}5p^1$ | $\boxed{ [\mathrm{Cd}]4f^95p^1}$ |
| Erbium (68) | $[Xe]4f^{12}6s^2$ | [Xe] $4f^{12}6s^1$ | [Xe] $4f^{12}$ | [Xe] $4f^{11}$ | $[{ m Xe}]4f^{10}$ | $[\mathrm{Cd}]4f^{10}5p^5$ | $[\mathrm{Cd}]4f^{10}5p^4$ | $[\mathrm{Cd}]4f^{10}5p^{3}$ | $[{\rm Cd}]4f^{10}5p^2$ | $[Cd]4f^{10}5p^1$ |
| Thulium (69) | $[Xe]4f^{13}6s^2$ | [Xe] $4f^{13}6s^1$ | [Xe] $4f^{13}$ | [Xe] $4f^{12}$ | [Xe] $4f^{11}$ | $[\mathrm{Cd}]4f^{11}5p^{5}$ | $[\mathrm{Cd}]4f^{11}5p^4$ | $[Cd]4f^{11}5p^3$ | [Cd] $4f^{11}5p^2$ | $\boxed{[\mathrm{Cd}]4f^{11}5p^1}$ |
| Ytterbium (70) | $[Xe]4f^{14}6s^2$ | [Xe] $4f^{14}6s^1$ | [Xe] $4f^{14}$ | [Xe] $4f^{13}$ | [Xe] $4f^{12}$ | $[\mathrm{Cd}]4f^{12}5p^5$ | $[\mathrm{Cd}]4f^{12}5p^4$ | $[Cd]4f^{12}5p^3$ | $[Cd]4f^{12}5p^2$ | $[Cd]4f^{12}5p^1$ |
| Lutetium (71) | $[Xe]4f^{14}5d^16s^2$ | [Xe] $4f^{14}6s^2$ | $[Xe]4f^{14}6s^1$ | [Xe] $4f^{14}$ | [Xe] $4f^{13}$ | $[{ m Xe}]4f^{12}$ | $[\mathrm{Cd}]4f^{13}5p^4$ | $[\mathrm{Cd}]4f^{13}5p^3$ | $[\mathrm{Cd}]4f^{13}5p^2$ | $\boxed{ [\mathrm{Cd}]4f^{12}5p^2}$ |