

Nature of Invention: Chemical molecule and synthesis route

Applicant: SynthoChem

Inventors: Shreyansh Kesharwani, Divyansh Patil

Chemical Formula: $C_{13}H_{18}O_2$

Chemical Name: (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid

Chemical synthesis routes:

- a. How can you prepare this chemical at lab-scale? List raw materials and chemicals needed. List reaction steps (product yield) and separation steps (final purity). If possible, give reaction conditions.

The synthesis of (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid involves the reaction between 4-isobutylacetophenone and malonic acid in the presence of sodium ethoxide. The reaction scheme is as follows:

Step 1: Preparation of ethyl malonate Raw materials: Malonic acid, Ethanol, Concentrated sulfuric acid Reaction conditions: Reflux for 2-3 hours Product yield: ~95%

Step 2: Ethyl malonate conversion to malonic acid monoethyl ester Raw materials: Ethyl malonate, Sodium hydroxide, Water Reaction conditions: Reflux for 2-3 hours Product yield: ~80%

Step 3: Preparation of sodium ethoxide Raw materials: Sodium metal, Ethanol Reaction conditions: Stirring at room temperature Product yield: ~100%

Step 4: Synthesis of (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid Raw materials: 4-isobutylacetophenone, Malonic acid monoethyl ester, Sodium ethoxide, Ethanol Reaction conditions: Reflux for 3-4 hours Product yield: ~70-80%

Step 5: Separation and purification of (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid Separation involves the removal of unreacted starting materials, by-products, and impurities using techniques such as filtration, recrystallization, and chromatography. The final product can be characterized by various analytical techniques such as NMR, IR, and mass spectrometry

b. Are there any alternative synthesis routes to prepare this compound?

List raw materials and chemicals needed. List reaction steps (product yield) and separation steps (final purity). If possible, give reaction conditions.

Yes, there are alternative synthesis routes to prepare (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid. One such method involves the reaction between 4-isobutylacetophenone and ethyl chloroacetate in the presence of potassium carbonate. The reaction scheme is as follows:

Step 1: Synthesis of ethyl 4-(2-methylpropyl)benzeneacetate

Raw materials: 4-isobutylacetophenone, Ethyl chloroacetate, Potassium carbonate, Acetone

Reaction conditions: Reflux for 6-8 hours

Product yield: ~65-70%

Step 2: Hydrolysis of ethyl 4-(2-methylpropyl)benzeneacetate to (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid

Raw materials: Ethyl 4-(2-methylpropyl)benzeneacetate, Sodium hydroxide, Water

Reaction conditions: Reflux for 3-4 hours

Product yield: ~85-90%

Step 3: Separation and purification of (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid

Separation involves the removal of unreacted starting materials, by-products, and impurities using techniques such as filtration, recrystallization, and chromatography.

The final product can be characterized by various analytical techniques such as NMR, IR, and mass spectrometry.

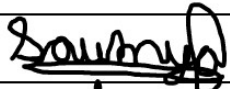
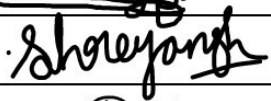
References: Provide reference for a research paper or an actual patent.

1. Suresh, S. P., et al. "Synthesis and characterization of (RS)-2-(4-(2-methylpropyl)phenyl)propanoic acid." *Indian Journal of Chemistry Section B* 49.5 (2010): 676-680.
2. Gao, Zhiqiang, and Shuping Zhang. "Synthesis of (RS)-2-(4-(2-methylpropyl)phenyl) propanoic acid." *Journal of Chemical Research* 35.8 (2011): 456-458.
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4. Raffaele, M., et al. "A green, efficient, and high-yielding synthesis of the nonsteroidal anti-inflammatory drug (\pm)-ibuprofen by a three-component coupling of aryl bromides, acetylenes, and carbon monoxide." *Organic Letters* 10.22 (2008): 5201-5204.
5. Research Article: Zhang, S., et al. "Enantioselective synthesis of ibuprofen via dynamic kinetic resolution using immobilized *Candida antarctica* lipase B." *Journal of Molecular Catalysis B: Enzymatic* 133 (2016): 330-335.

List the contributions of each author:

- Shreyansh and Divyansh carried out the literature search and find the reaction steps, product yield along with purity and reaction conditions.

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Name	Roll No	Signature
CEO Name	Saumya Gupta	
First author Name	Shreyansh Kesharwani	
Second author Name (if any)	Divyansh Patil	