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Tetle of the experement:

magnetic susceptibility of a Paramagnetic material

objective:

To determine the magnetic susceptibility (x) of the given paramagne -tic solution using anincke's method

Apparatus:

- -> Electromagnet and power supply
- -> Gauss meter and Hall probe
- >> paramagnetic salt (Mnsoy. #20)
  - >> Digital balance
- specific gravity bottle and distilled The water died roll the Bet
  - > auencke's tube
    - -> Travelling microscope

Formula:

Ysa = 2 40 19h ; B = Mott

Psot = Powater (mass of solution) = m3-m1
mass of water) = m2-m1

Table:	current (D) amp	rectoomagnet magneticflux, B causs
	0	0
2	0.5	-13.8
3	10 to 120 4105 175	28
4	1.5 Long stom	24242.6 MASO
5	2	-55.3
6	ampana. Sati anim	-68.2
80170and	1 1930 SH 10	-78.9
1008 NJ 18	m 2193.5	-86.)
9	4	-94.5

Table: 2 Height of the solution for different magnetic flux.

S.No	current (I) amp	Magneli flux, B Gauss	BA	mentscus posttom for B-0,a cm	posetton	change on height h=b-a cm	change in height h=b=a(cm)
I.E.		-28	784	1.465	1.56	0.095	0.095
2	2	-55.3	3058.09	1.465	1.772	0.307	0.307
3	13	-78.9	6225.21	1.465	1.876	0.411	0.411
4	4	-94.5	8930.25	1.465	1.979	0.514	0.514

Calculations: 
$$900 - 17.0668$$
 $900 - 17.0668$ 
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 $900 - 17.0668$ 

= 27.2132 1 Megar - 1 Millions to 2201 23.7822

11-11 (84) 10 22 01 = 1.1442 Kg/m3 1009

Density value = 1.14 m1=17.06689m m2 = 40.849gm m3 = 44.2800gm Mass of empty specific gravity bottle, m1 = 12.0668 gm Mass of specific gravity bottle with water, m2 =40.849 gm Mass of specific gravity bottle with soult solution, m3 = 44.2800 gm Density of water, Pwater = 1000 kg/m3 Density of solution Psol = 1.144 kg/m3 The paramagnetic succeptibility of salt solution from the relation Xsol = 2 Morsolgh Graph of B us I Graph of h vs B2.

$$P_{SOI} = 1.14 \text{ kg/m}^3$$
 $X_{SOI} = 2 M_0 P_{SOI} 9 h$ 
 $B^2$ 

$$\chi^{201} = \overline{5 \times 10 \times 1.11 \times 4.8 \times 1}$$

Slope 
$$(\frac{h}{BL})$$
 from graph =  $18 \times 10^3$   
 $9 = 9.8 \, \text{m/s}^{\frac{1}{2}}$ ;  $u_0 = u_{11} \times 10^{-\frac{3}{2}} + 1/m$   
 $x_{501} = 2 \times u_{11} \times 10^{-\frac{3}{2}} \times 1.14 \times 9.8 \times 18 \times 10^{3}$   
 $x_{501} = 0.505$ 

## Result:

Susceptibility of Mnsoy. H20 = 0.505.



