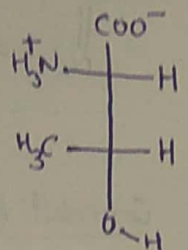


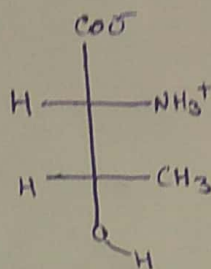
I

Threonine $\rightarrow (C_4H_9NO_3)$

(a) Stereoisomers

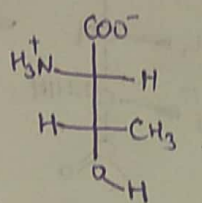


L-threonine

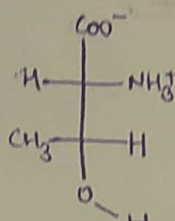


D-threonine

R \rightarrow clockwise
S \rightarrow anticlockwise



L-allo-threonine



D-allo-threonine

(b) (i) L-threonine $\rightarrow 2S, 3R$

(ii) D-threonine $\rightarrow 2R, 3S$

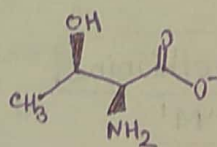
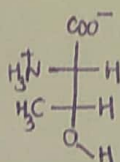
(iii) L-allo-threonine $\rightarrow 2S, 3S$

(iv) D-allo-threonine $\rightarrow 2R, 3R$

Enantiomeric pairs $\Rightarrow \{(i, ii); (iii, iv)\}$

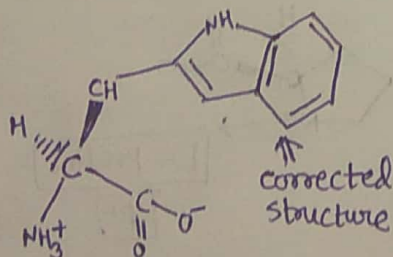
Diastereomeric pairs $\Rightarrow \{(i, iii); (i, iv); (ii, iii); (ii, iv)\}$

(c) $(2S, 3R) \rightarrow$ L-threonine



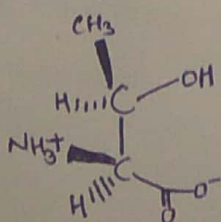
IUPAC name $\Rightarrow (2S, 3R)$ -2-amino-3-hydroxybutanoic acid.

2 (i)



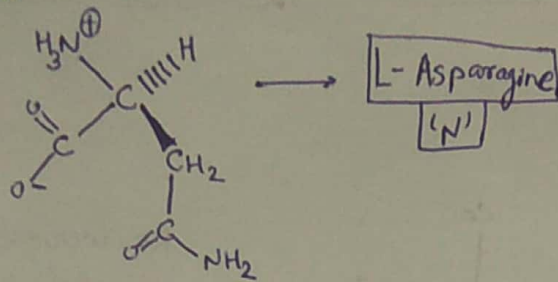
\rightarrow L-Tryptophan
[W]

(ii) Already correct structure \Rightarrow

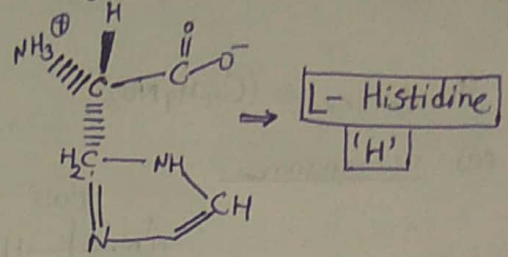


L-threonine
[T]

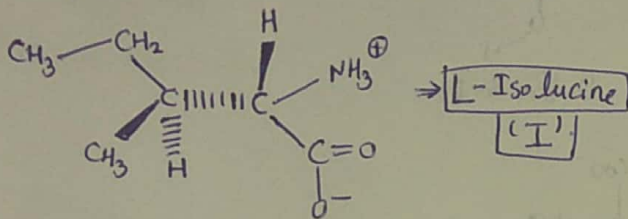
iii) Corrected structure.



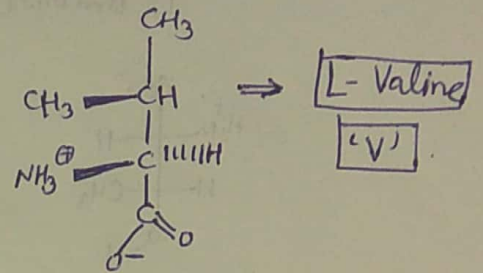
iv) Already correct structure.



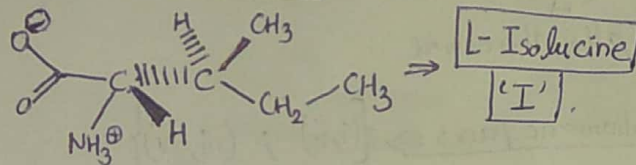
v) Corrected structure.



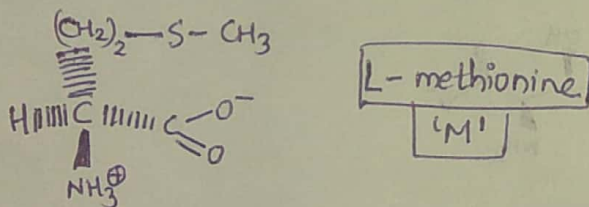
vi) Corrected structure



vii) Already correct structure.

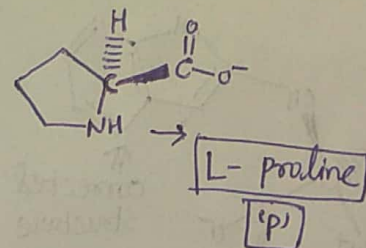
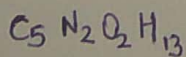
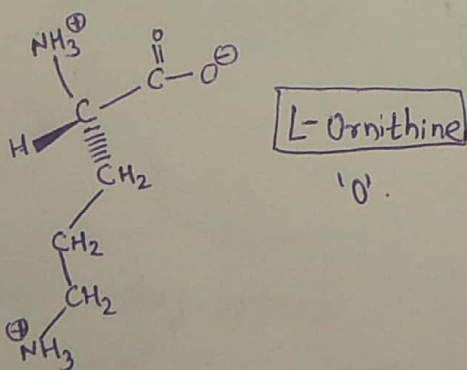


viii) Corrected structure.

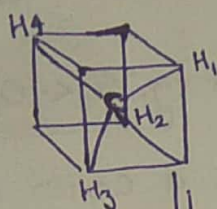


ix) Already correct structure.

(ix) Corrected structure

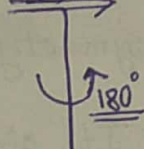


1

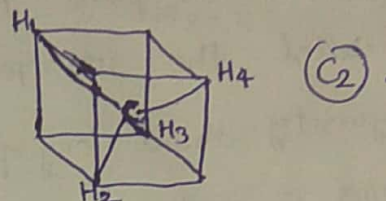


Methane

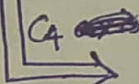
(a) C_2 axis of symmetry.



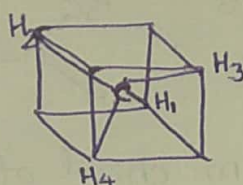
Rotation of CH_4 about the axis passing through the center by 180° gives same molecule back.



(b) $S_4 = C_4 + \sigma$.

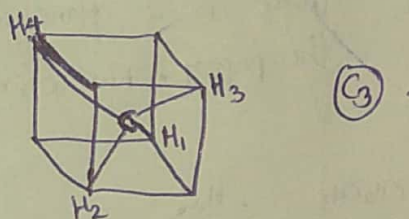
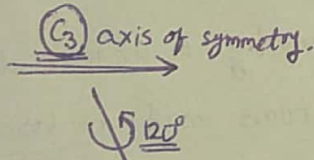
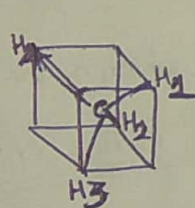


$S_4 \leftarrow C_4 + \sigma$.

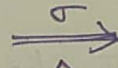
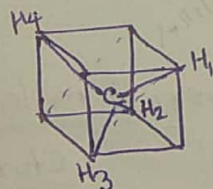


same as initial CH_4 structure $\Rightarrow S_4$ axis of symmetry.

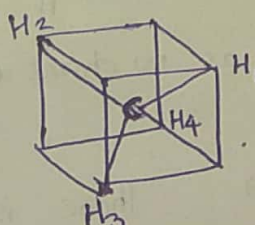
(c) Rotating CH_4 by 120° about axis through C & H_4 again gives the same result.



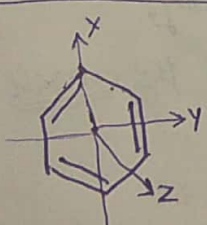
(d) $\sigma \Rightarrow$ By taking mirror plane along the diagonal plane, we get same element back again



mirror plane.



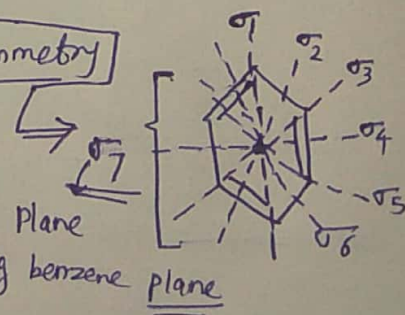
2



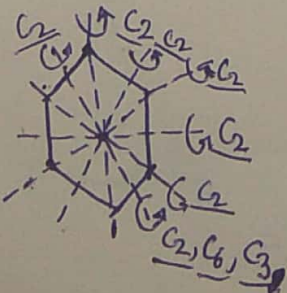
Reflection

planes.

7 planes of symmetry

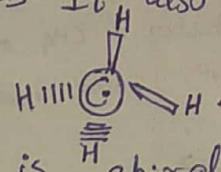


9 rotation axes of symmetry.

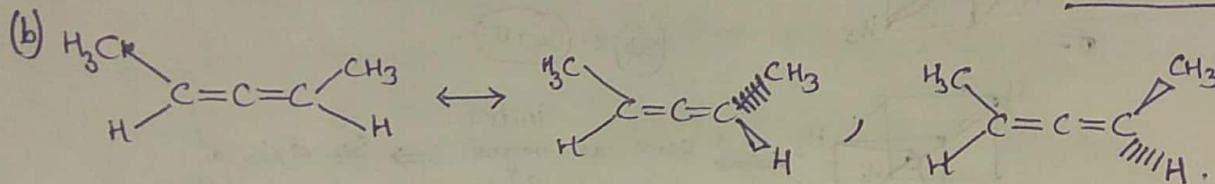


3

→ An improper rotation is a combination of a rotation about an axis followed by a reflection in the plane perpendicular to the rotation axis and the rotation axis is called the improper axis of symmetry. for e.g. $S_4 \leftarrow C_4 + \sigma_H$. CH_4 has axis of symmetry about any of 4 C-H bonds. It also has a center of symmetry by keeping C as the center.



(a) No symmetry exist in this element. N here is chiral. Also all 4 Carbons attached to N are chiral. \therefore Overall, molecule is chiral.



There is no plane of symmetry and no chiral atoms but overall molecule is chiral.

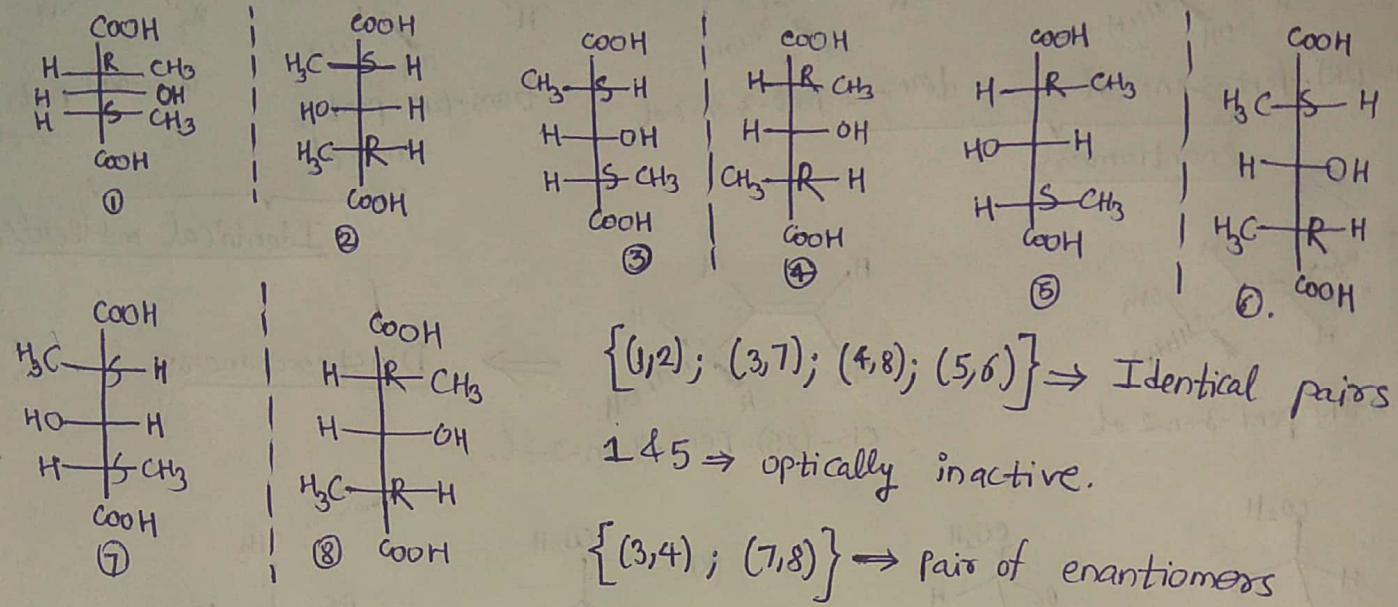
(c) There is a plane of symmetry in the molecule which is the plane of the paper. No chiral atoms and molecule is not chiral.

(d) No symmetry exist as well as no chiral atoms, but the overall molecule is chiral.

(e) There is 1 plane of symmetry. No chiral atoms are there and the overall molecule is not chiral.

4

diastereomers of 2,4-dimethyl-3-hydroxy pentane-1,5-dicarboxylic acid.



5

