

# **UCC501 Sustainable Energy**

## **Homework (2)**

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**15<sup>th</sup> February 2014**

### **Question 1 (Energy Scales, CO2 Emissions and Renewables)**

- a) *What was the total Natural gas consumption in the UAE in 2011? Express in MTOE/year and GWh/year.*

Solution:

The total Natural Gas consumption in the UAE in 2011 was 2221 billion cubic feet.

$$1 \text{ billion cubic feet} = 0.026 \text{ MTOE}$$

Therefore, the total natural gas consumption in the UAE in 2011 was ([www.eia.gov](http://www.eia.gov))

$$0.026 * 2221 = 57.746 \text{ MTOE/year}$$

And we know

$$1 \text{ MTOE} = 11,630 \text{ GWh}$$

Therefore the total natural gas consumption in the UAE in 2011 was

$$\mathbf{11,630 * 57.746 = 671,585.98 \text{ GWh/year}}$$

- b) *What was the total annual CO2 emissions (in kg) from natural gas consumption in 2011? Base your answer on 1a. findings. Hint: Assume complete combustion.*

Solution:

From the [www.eia.gov](http://www.eia.gov) website, the total annual CO2 emissions from Natural gas consumption in the year 2011 for the UAE were 149.710 million metric tonnes.

$$\mathbf{CO_2 \text{ emissions}_{2011} = 149.71 * 10^9 \text{ kg}}$$

- c) *How much CO2 do we save in total by installing a solar system that will generate (throughout its lifetime) energy equal to 15% of natural gas secondary (useful) consumption in 2011? What is the size (in mW) of such a plant?*

*Assumptions:*

*Solar power obtains 1700 full load hours in the UAE*

*The fossil fuel conversion efficiency is 40%*

*Grid Transmission efficiency is 92%*

*Lifetime is 25 years*

*It is possible to convert any and all fossil fuel used to electricity instantaneously*

Solution:

We can use the following equation to calculate the size of the solar power plant:

$$\text{Solar Power} * 0.92 * \text{Lifetime} * 1700 = 15\% \text{ of Natural gas consumption} * 0.4$$

15% of the natural gas consumption in the year 2011=

$$0.15 * 671585 * 0.4 = 40295.1 \text{ GWh}$$

$$\text{Solar Power} = \frac{40295.1}{0.92 * 25 * 1700} = 1,030,565 \text{ kW} = 1030.565 \text{ MW}$$

$$\text{EROEI (Energy returned on energy invested)} = \frac{\text{Net Energy} + 1}{\text{Energy Expended}} \quad (\text{www.wikipedia.org})$$

Net Energy=15% of Natural gas consumption=100,737.75 GWh

Therefore, energy expended=5301.987 GWh

Primary Energy=energy expended/conversion efficiency

$$=13254.97 \text{ GWh} = 13254.97 * 10^6 \text{ kWh}$$

CO2 emissions from natural gas per kWh=0.553kg

(<http://www.eia.gov/tools/faqs/faq.cfm?id=74&t=11>)

Therefore the amount of CO2 emitted =  $7,329.998 * 10^6 \text{ kg}$

CO2 from 15% of the natural gas consumption= $55707.976 * 10^6 \text{ kg}$

$$\text{CO}_2 \text{ Emissions saved} = 48377.978 * 10^6 \text{ kg}$$

## **Question 2 (Economic Analysis)**

Solution:

In Attached Excel Workbook

## **Question 3 (Vehicle Kinetics, CO2 and Renewable Energy)**

a) Based on the principles of vehicle kinetics, estimate the propulsion power required (in Watt) for an average SEDAN (Toyota Camry VX50 type, gasoline). Use the following assumptions:

- Average car speed:  $100 \text{ km/h} = 27.778 \text{ m/s}$ .
- 80km trip.
- Effective aerodynamic cross sectional area =  $2.45 \text{ m}^2$ .

- Air density =  $1.2 \text{ kg/m}^3$ .
- Drag coefficient = 0.28.
- Rolling resistance coefficient = 0.01.
- Total vehicle mass: 1440 (vehicle) + 70 (passenger) + 40 (fuel) kg.

Solution:

The mechanical propulsion power required can be obtained from the following equation:

$$P_v = \text{mechanical power} + \text{aerodynamic drag} + \text{rolling resistance}$$

$$P_v = \frac{m_v \cdot u^3}{2 \cdot d} + \frac{1}{2} \cdot \rho_{air} \cdot A \cdot c_d \cdot u^3 + m_v \cdot g \cdot c_{rr} \cdot u$$

$$P_v = \frac{1550 \cdot 27.778^3}{2 \cdot 80000} + \frac{1}{2} \cdot 1.2 \cdot 2.45 \cdot 0.28 \cdot 27.778^3 + 1550 \cdot 9.79 \cdot 0.01 \cdot 27.778$$

$$P_v = 207.64 + 8822.02 + 4215.14$$

$$\mathbf{P_v = 13244.8W}$$

b) For the CV in a), calculate the amount of energy (in MWh) required for travelling an average total distance of 36000 km/year (considering all efficiencies).

Solution:

If the internal combustion engine efficiency is assumed to be 25%:

Average travelling time = distance/average speed = 36000/100 = 360 hours

The total annual amount of energy required for a single CV operation = power/motor efficiency\*average travelling time = 13.2448/0.25\*360

$$\mathbf{E_T = 19.073 MWh}$$

c) Calculate annual energy required (in MWh) for 500,000 vehicles registered in Abu Dhabi and estimate the corresponding CO<sub>2</sub> emissions (in kg) caused by the vehicle activity.

Solution:

The total annual amount of thermal energy needed for all CV's operation =  
 $19.073 \times 500,000 = 9,536,500$  MWh.

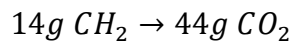
Using the gross heat of combustion of one cubic meter of gasoline to be 12.26kWh/kg  
([http://wogone.com/science/the\\_energy\\_and\\_fuel\\_data\\_sheet.pdf](http://wogone.com/science/the_energy_and_fuel_data_sheet.pdf))

1kg of gasoline = 12.26kwh

777,854,812.4kg of gasoline = 9,536,500 MWh

The chemical equation of gasoline combustion is  $CH_2 + \frac{3}{2}O_2 \rightarrow CO_2 + H_2O$  (Tester, Drake, Driscoll, Golay, Peters, Sustainable Energy: Choosing Among Options Chapter 7:pg.316).

(The molar mass of  $CH_2 = 12 + 2 = 14$ g/mol, the molar mass of  $CO_2 = 12 + 16 + 16 = 44$  g/mol)



$$777,854,812.4 \text{ kg } CH_2 \rightarrow 2,444,686,553 \text{ kg } CO_2$$

d) What is the rate of solar power installations that the UAE would have needed to install per year for the next 5 years so that the CO<sub>2</sub> emissions from the CVs are kept constant – at the level of 50% less than now? Assume that the number of CVs grows at a rate of 3% per year. Use the assumptions in 1.c.

Solution:

The target CO<sub>2</sub> emissions, 50% of now, is  $\frac{1}{2} \cdot 2,444,686,553 = 1,222,343,277$  kg/year

In 5 years' time the CO<sub>2</sub> emissions of cars in Abu Dhabi would have increased to

$$2,444,686,553 \times 1.03^5 = 2,834,061,741 \frac{kg}{year}$$

Making the amount of CO<sub>2</sub> that needs to be reduced

$$2,834,061,741 - 1,222,343,277 = 1,611,718,464$$

This means that  $1,611,718,464 \times \frac{14}{44} = 512,819,511.3$  kg of gasoline needs to be reduced.

This amount of gasoline equates to

$512,819,511.3 \text{ kg gasoline} * \frac{12.26 \text{ kWh}}{\text{kg}} = 6,287,167.21 \text{ MWh}$  that needs to be installed over 5 years.

The average number of hours of full load sunlight in a year in the UAE is 1700.

That means that  $6,287,167.21 \div 1700 = 3,698.34 \text{ MW}$  need to be installed over the next 5 years.

The yearly capacity that needs to be installed is  **$3,698.67 \div 5 = 739.67 \text{ MW}$**