Simulations of the motion of atoms and molecules. Trajectories of molecules (position versus time) are calculated by numerically solving equations of motion originating from the Newton laws:

Force=mass* acceleration, F=m*a

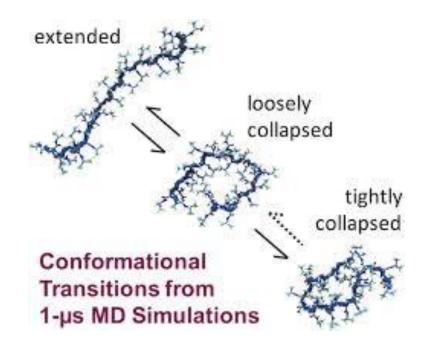
Forces between interacting particles and potential energy are defined by molecular mechanics force fields.

Main area of applications: materials science and modeling of biomolecules.

Examples of potential: gravitation field, electrostatic field

Potential energy – characterises forces that depend only on the position of the particle in space. Consequently, the potential energy also depends only on the position

Force field – the set of forces



http://ringo.ams.stonybrook.edu/
index.php/MD_Simulation:
 _Protein_in_Water

https://pubs.acs.org/doi/pdf/10.1021/acs.jpcb.6b09165?rand=fu7wimf5

https://www.ks.uiuc.edu/Research/STMV/images/virus_webpage.full.png

Theoretical and Computational Biophysics Group Beckman Institute University of Illinois at Urbana-Champaign

- Each atom is treated as a single particle
- Each particle has an radius and polarizability
- Interactions are treated as "springs" with an equilibrium distance (harmonic oscillators)
- Atoms are treated as a hard spheres (balls)
- Collisions are perfectly elastic (the total kinetic energy and momentum is conserved, no changes in the particle shape)
- Atomic charges remain unchanged during the motion
- Bonds cannot be created (or destroyed)

Limitations

bigger systems for longer timescales - longer computation time

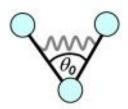
- protein aggregation on atomistic level, adsorption and diffusion mechanisms
- enzymes activity and insight into detailed catalytic activity mechanisms

- Build the molecule find the (x,y,z) position of all atoms
- Create (or apply existing) the force field
- Find the optimal structure (energy minimization)
- Solve the Newton's equations of motions of all atoms in the force field for each time point

$$U(\vec{R}) = \frac{1}{2} \sum_{bonds} K_b (b - b_0)^2 +$$

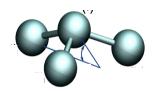
$$b_0$$

$$\frac{1}{2} \sum_{angles} K_{\theta} (\theta - \theta_0)^2 +$$



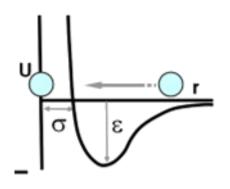
$$\sum_{dihedrals} K_{\phi} \left[1 + \cos(n\phi - \delta) \right] +$$

$$\sum_{impropers} K_{\omega} \left(\omega - \omega_0 \right)^2 +$$

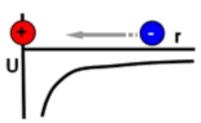


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$$\sum_{i>j} \left(\frac{A_{ij}}{r_{ij}^{12}} - \frac{B_{ij}}{r_{ij}^{6}} \right) +$$



$$\sum_{i>j} \frac{q_i q_j}{4\pi \varepsilon_0 r_{ij}}$$



Energy minimization methods - to compute the equilibrium configuration of molecules

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The Verlet algorithm

Taylor expansion

$$\vec{r}_{i}(t + \Delta t) = \vec{r}_{i}(t) + \frac{d\vec{r}_{i}(t)}{dt} \Delta t + \frac{1}{2} \frac{d^{2}\vec{r}_{i}(t)}{dt^{2}} \Delta t^{2} + \frac{1}{6} \frac{d^{3}\vec{r}_{i}(t)}{dt^{3}} \Delta t^{3} + O(\Delta t^{4})$$

$$\vec{r}_{i}(t - \Delta t) = \vec{r}_{i}(t) - \frac{d\vec{r}_{i}(t)}{dt} \Delta t + \frac{1}{2} \frac{d^{2}\vec{r}(t)}{dt^{2}} \Delta t^{2} - \frac{1}{6} \frac{d^{3}\vec{r}_{i}(t)}{dt^{3}} \Delta t^{3} + O(\Delta t^{4})$$

$$\vec{r}_i(t+\Delta t) = 2\vec{r}_i(t) - \vec{r}_i(t-\Delta t) + \frac{d^2\vec{r}_i(t)}{dt^2} \Delta t^2 + O(\Delta t^4)$$

The Verlet algorithm

Velocity:

$$\vec{v}_i(t) = \frac{d\vec{r}_i(t)}{dt} = \frac{\vec{r}_i(t + \Delta t) - \vec{r}_i(t - \Delta t)}{2\Delta t}$$

Kinetic Energy:

$$E(t) = \sum_{i=1}^{N} \frac{m\vec{v}_i(t)^2}{2}$$

The Verlet algorithm

The position of the atom is updated with Δt step

$$\vec{r}_i(t + \Delta t) = \vec{r}_i(t) + \vec{v}_i(t)\Delta t + \frac{1}{2}\vec{a}_i(t)\Delta t^2$$

The velocity is updated with $(\Delta t + \frac{\Delta t}{2})$ step:

$$\vec{v}_i \left(t + \frac{\Delta t}{2} \right) = \vec{v}_i (t) + \frac{1}{2} \vec{a}_i (t) \Delta t$$

Acceleration at $(t + \Delta t)$:

$$\vec{a}_i(t + \Delta t) = -\left(\frac{1}{m_i}\right) \nabla U(\vec{r}_i(t + \Delta t))$$

Velocity at:

$$(t + \Delta t) \qquad \vec{v}_i (t + \Delta t) = \vec{v}_i \left(t + \frac{\Delta t}{2} \right) + \frac{1}{2} \vec{a}_i (t + \Delta t) \Delta t$$

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