

भारतीय प्रौद्योगिकी संस्थान पटना

Indian Institute of Technology Patna पाटलिपुत्र कॉलोनी, पटना - 800 013, भारत

Patliputra Colony, Patna - 800 013, India

DURATION: 2 HOURS MID-SEMESTER EXAMINATION M.M. 30 COURSE: INTRODUCTION TO NANOMATERIALS (PH 401) DATE: 17 – 09 – 2014

	COOL	SE: INTRODUCTION TO NANOMATERIALS (FIT 401) Attempt all Questions
•		Full Marks: 30
	hours	[0.5x5 = 2.5]
1.	Fill in t	the blanks: Electrical conductivity in nanosize ceramics compare to its bulk material.
	(i) (ii)	Electrical conductivity in nanosize ceramics upto a critical grain size (nanmeter Magnetic coercivity of ferromagnetic material upto a critical grain size (nanmeter
	(iii)	range). Quantum efficiency of semiconductor crystals with decrease in size in nanosize
		regime. Due to higher surface to volume ratio the catalytic efficiency of nanomaterial in parameterials compare to its bulk form.
	(iv)	Due to higher surface to volume ratio the catalytic emeters, sompare to its bulk form.
	(v)	Due to higher surface to volume ratio the catalytic embers, Electrical conductivity for metals in nanomaterials compare to its bulk form. [1x8 = 8]
2.	Answ	er all the questions:
	(i)	er all the questions: Write the summery of Richard Feynaman talk" there is plenty of room at the bottom". What is the difference between conventional engineering and nanoengineering?
	(ii)	What is the difference between conventional engineering
	(iii) (iv) (v)	Define nanoscience and nanotechnology. What characteristic of the butterfly morpho peleids limpida makes it beautiful colour? Explain. Describe the hydrophobicity of leaves of trees and wings of bird and, explain how the concept has been used to develop technology? Define the 0, 1, 2 & 3 degree/s of confinement of material. Give one example from each
	(vi)	
		what is the length scale order of strong confinement of electron inside the metal?
3	(vii)	the material increases with the decrease of grant size.
	(viii)	diale with refractive index of 1.57 die floating on trate.
	. 300	nm thick (3.05 mm radius) disks with refractive masks and the look it from the top? If somebody of water= 1.33). What will be the colour of sections, if we look it from the top? If somebody of water= 1.33).
	look	c at an angle of 60° to the surface of water, the colour of section this end [3
	Just	ify your answer.
4	4. Dra	cify your answer. w a graph between critical dimension (nm) and surface-to-volume ratio (nm ⁻¹) for sphere, cylinder [2]
	5. Def	I cube. Discuss the significance of the curve in hand meter responses for cubo-octahedral FC fine magic numbers in nanoscience. Write the structural magic numbers for cubo-octahedral FC [4]
	Fin [ve	noparticle. ective masses of electron and hole for InSb semiconductor are 0.014m _o and 0.4m _h respectively detective masses of electron and weak confinement region for the semiconductor. Sketch your result the strong, intermediate and weak confinement region for the semiconductor. Sketch your result the semiconductor, Planck's constant= 6.62x10 ⁻³⁴ .
		$_{\rm s}$ =9.31x10 ⁻³¹ kg]. and out the melting point of spherical Gold nanoparticle of radius 80nm and 2nm. Plot the melting out the melting point of Gold = 1336 $^{\circ}$ C. Boiling point
	po Go	wint versus radius of spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1335 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticle. Melting point of Gold = 1355 s, and a single spherical gold nanoparticl
	8. W	that is the resolution of projection/ lithography system with a 100kV electron search.
	9. Ex	efractive index) is 1.4 and projection angle 6-70 . Ose the equipment of the resist in lithography? It is positive and negative resist in lithography? It is positive and negative resist in lithography? It is positive and negative resist in lithography?

-----BEST OF LUCK-----

