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Absolute zero Wikipedia

Absolute zero is the lowest limit of the thermodynamic temperature scale, a state at which the enthalpy and entropy of a cooled ideal gas reach their minimum value, taken as 0. The fundamental particles of nature have minimal vibrational motion, retaining only quantum mechanical, zero-point energy-induced particle motion. The theoretical temperature is determined by extrapolating the ideal gas

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Absolute Zero and the Conquest of Cold Tom Shachtman

I'm really interesting in the pursuit of absolute zero. And this seemed like the book. I confess, I've struggled with it, even trying to skip sections, but I just can't make it through it.

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Absolute value Wikipedia

The real absolute value function is continuous everywhere. It is differentiable everywhere except for $x = 0$. It is monotonically decreasing on the interval $(-, 0]$ and monotonically increasing on the interval $[0, +)$. Since a real number and its opposite have the same absolute value, it is an even function, and is hence not invertible. The real absolute value function is a piecewise linear

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Zero Point Crystalinks

Zero Point Energy. In a quantum mechanical system such as the particle in a box or the quantum harmonic oscillator, the lowest possible energy is called zero-point energy.. Zero-point energy is the lowest possible energy that a quantum mechanical physical system may have; it is the energy of its ground state.

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