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### **Maxwell Boltzmann distribution Wikipedia**

In summary, the typical speeds are related as follows:  $v_{rms} < v_{avg} < v_{mp}$ . The root mean square speed is directly related to the speed of sound  $c$  in the gas, by  $v_{rms} = \sqrt{\gamma} c$ , where  $\gamma$  is the adiabatic index,  $f$  is the number of degrees of freedom of the individual gas molecule. For the example above, diatomic nitrogen (approximating air) at 300 K,  $\gamma = 1.4$  and  $f = 5$

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