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Titration Wikipedia

Titration, also known as titrimetry, is a common laboratory method of quantitative chemical analysis that is used to determine the concentration of an identified analyte. Since volume measurements play a key role in titration, it is also known as volumetric analysis. A reagent, called the titrant or titrator is prepared as a standard solution. A known concentration and volume of titrant reacts

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Calibration curve Wikipedia

In analytical chemistry, a calibration curve, also known as a standard curve, is a general method for determining the concentration of a substance in an unknown sample by comparing the unknown to a set of standard samples of known concentration. A calibration curve is one approach to the problem of instrument calibration; other standard approaches may mix the standard into the unknown, giving

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[Federal Register Volume 79, Number 197 (Friday, October 10, 2014)][Proposed Rules][Pages 61383-61438]

From the Federal Register Online via the Government Printing Office [www.gpo.gov] [FR Doc No: 2014-24009] Vol. 79 Friday, No. 197 October 10, 2014 Part II Department of Labor ----- Occupational Safety and Health Administration ----- 29 CFR Parts 1910, 1915, 1917, et al. Chemical Management

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Chemical Substances Introduction ACGIH

Approved by the ACGIH Board of Directors on March 1, 1988. Special Note to User. These values are intended for use in the practice of industrial hygiene as guidelines or recommendations to assist in the control of potential workplace health hazards and for no other use.

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