

Phonons and Electron-Phonon Couplings

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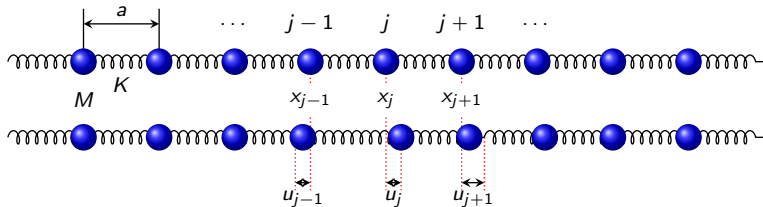


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1 Introduction

1D Chain of Atoms — 1 Atom per Unit

A 1D chain of N equally spaced atoms at $R_j(t) = x_j + u_j(t)$



The Newton's Equation

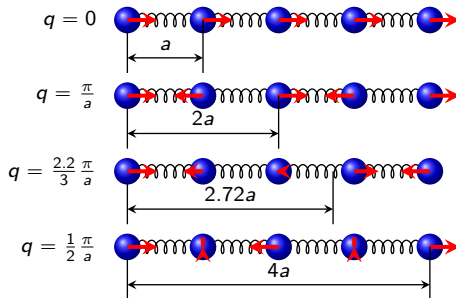
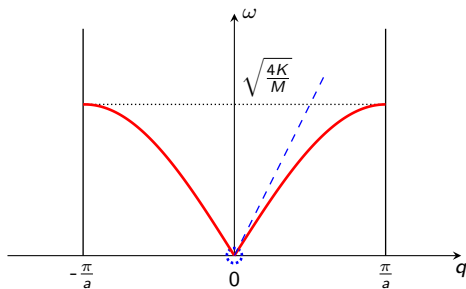
$$M \frac{d^2 u_j}{dt^2} = K(u_{j+1} + u_{j-1} - 2u_j) \quad j = 1, \dots, N$$

Assume the solution has the form $u_j(t) = \frac{A_q}{\sqrt{M}} e^{i(qx_j - \omega t)}$, then ¹

$$\begin{aligned} \omega^2 &= \frac{K}{M} (2 - e^{iqa} - e^{-iqa}) \\ &= \frac{2K}{M} (1 - \cos qa) \\ \Rightarrow \quad \omega &= \sqrt{\frac{4K}{M}} \left| \sin \frac{qa}{2} \right| \end{aligned}$$

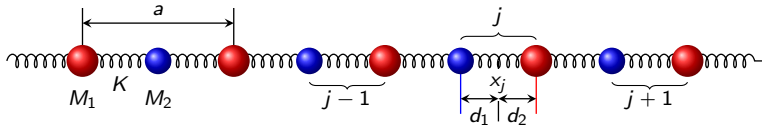
¹ $u_j(t)$ here is complex. In practice, take the real part, i.e. $\text{Re}[u_j(t)]$.

1D Chain of Atoms



1D Chain of Atoms — 2 Atoms per Unit

A 1D chain with 2 atoms in each unit: $R_j^s(t) = x_j + d_s + u_j^s(t)$; $s = 1, 2$



The Newton's Equation

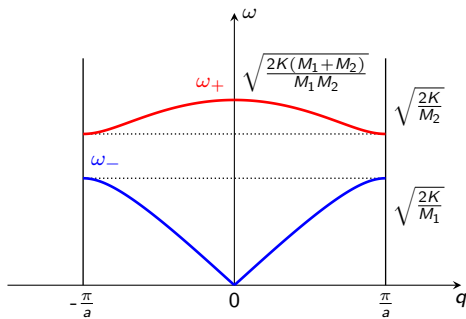
$$\begin{aligned} M_1 \frac{d^2 u_j^1}{dt^2} &= K(u_j^2 + u_{j-1}^2 - 2u_j^1) \\ M_2 \frac{d^2 u_j^2}{dt^2} &= K(u_j^1 + u_{j+1}^1 - 2u_j^2) \end{aligned} \quad \Rightarrow \quad \begin{cases} u_j^1(t) = \frac{A_q}{\sqrt{M_1}} e^{i(qx_j - \omega t)} \\ u_j^2(t) = \frac{B_q}{\sqrt{M_2}} e^{i(qx_j - \omega t)} \end{cases}$$

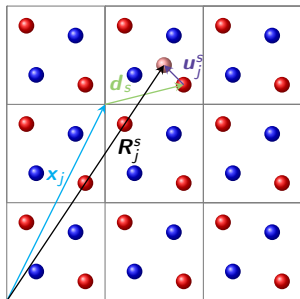
We then have

$$\begin{pmatrix} \frac{2K}{M_1} & \frac{-K}{\sqrt{M_1 M_2}} (1 + e^{-iqa}) \\ \frac{-K}{\sqrt{M_1 M_2}} (1 + e^{iqa}) & \frac{2K}{M_2} \end{pmatrix} \begin{pmatrix} A_q \\ B_q \end{pmatrix} = \omega^2 \begin{pmatrix} A_q \\ B_q \end{pmatrix}$$

$$\Rightarrow \quad \omega_{\pm}^2 = \frac{K}{M_1 M_2} \left((M_1 + M_2) \pm \sqrt{M_1^2 + M_2^2 + 2M_1 M_2 \cos qa} \right)$$

1D Chain of Atoms





- \mathbf{x}_j : the position of unit cell j
- \mathbf{d}_s : the equilibrium position of the atom s in the cell
- \mathbf{u}_{js}^s : displacement from the equilibrium position for the atom s in the cell j
- \mathbf{R}_{js}^s : the position of the atom s in the cell j

$$\begin{aligned}\mathbf{R}_{js}^s(t) &= \mathbf{x}_j + \mathbf{d}_s + \mathbf{u}_{js}^s(t) \\ &= \mathbf{r}_{js}^s + \mathbf{u}_{js}^s(t)\end{aligned}$$

$$R_{j\alpha}^s(t) = r_{j\alpha}^s + u_{j\alpha}^s(t) \quad (\alpha = x, y, z)$$

The total energy can be written as

$$E_{\text{tot}}(\{\mathbf{R}_{js}^s(t)\}) = E_{\text{tot}}^0(\{\mathbf{r}_{js}^s\}) + \sum_{js\alpha} \frac{\partial E_{\text{tot}}^0}{\partial u_{j\alpha}^s} u_{j\alpha}^s + \frac{1}{2} \sum_{js\alpha} \frac{\partial^2 E_{\text{tot}}^0}{\partial u_{j\alpha}^s \partial u_{kt\beta}^s} u_{j\alpha}^s u_{kt\beta}^s + \dots$$

- The expression is exact if we take all the orders in the expansion.
- All the derivatives are taken at the equilibrium positions $\{\mathbf{r}_{js}^s\}$, i.e. $\frac{\partial E_{\text{tot}}^0}{\partial u_{js}^s} = 0$.
- Harmonic approximation: truncated at *second* order.

Within the harmonic approximation, the Newton's equation for the atom s in cell j

$$M_s \frac{d^2 u_{j\alpha}^s(t)}{dt^2} = - \frac{\partial E_{\text{tot}}}{\partial u_{j\alpha}^s} = - \sum_{kt\beta} \frac{\partial^2 E_{\text{tot}}^0}{\partial u_{j\alpha}^s \partial u_{kt\beta}^t} u_{kt\beta}^t = - \sum_{kt\beta} C_{s\alpha, t\beta}(k-j) u_{kt\beta}^t \quad (1)$$

The ansatz of the solution

$$u_{j\alpha}^s(t) = \frac{\eta_{\alpha}^s(\mathbf{q})}{\sqrt{M_s}} e^{i\mathbf{q} \cdot \mathbf{x}_j} e^{-i\omega t} \quad (2)$$

Substitute Eq. 2 into Eq. 1

$$\omega^2(\mathbf{q}) \eta_{\alpha}^s = \sum_{t\beta} \left[\sum_k \frac{1}{\sqrt{M_s M_t}} \frac{\partial^2 E_{\text{tot}}^0}{\partial u_{j\alpha}^s \partial u_{kt\beta}^t} e^{i\mathbf{q} \cdot (\mathbf{x}_k - \mathbf{x}_j)} \right] \eta_{\beta}^t = \sum_{t\beta} D_{s\alpha, t\beta}(\mathbf{q}) \eta_{\beta}^t$$

In matrix form

$$\underbrace{\begin{pmatrix} \ddots & & \\ & D_{s\alpha, t\beta}(\mathbf{q}) & \\ & & \ddots \end{pmatrix}}_{3N_a \times 3N_a} \underbrace{\begin{pmatrix} \vdots \\ \eta_{\beta}^t(\mathbf{q}) \\ \vdots \end{pmatrix}}_{3N_a} = \omega^2(\mathbf{q}) \underbrace{\begin{pmatrix} \vdots \\ \eta_{\beta}^t(\mathbf{q}) \\ \vdots \end{pmatrix}}_{3N_a}$$

where N_a is the number of atoms in a cell.

Thank you!