



Virtual International Seminar on Theoretical Advancements

# Traveling in a space of approximations for modeling of photoinduced dynamic processes

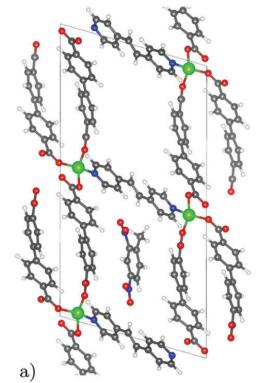
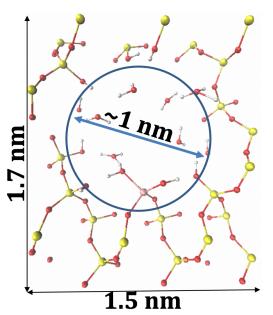
Dmitri Kilin

*Department of Chemistry and Biochemistry, North Dakota State University, Fargo ND 58108;*

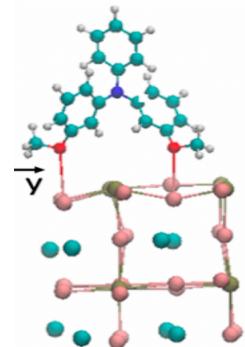
*Email: [Dmitri.Kilin@ndsu.edu](mailto:Dmitri.Kilin@ndsu.edu)*



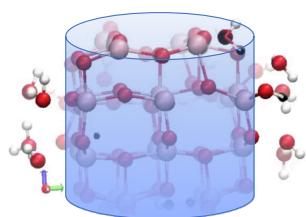
**3D**  
**Porous materials**



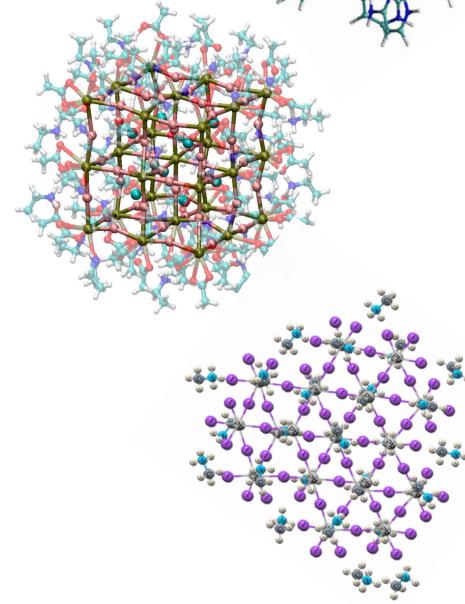
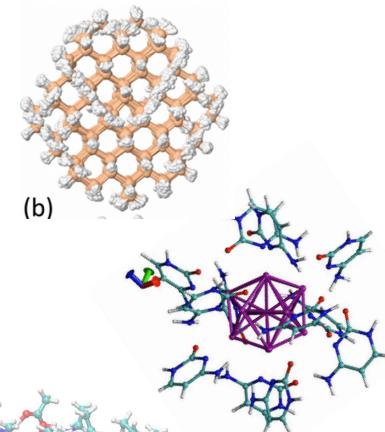
**2D**  
**Planar interfaces**



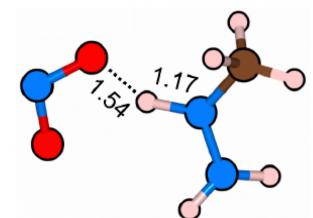
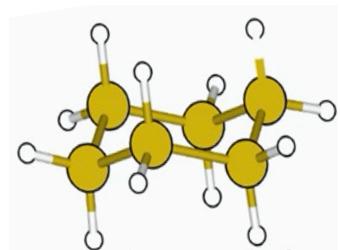
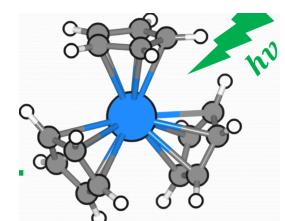
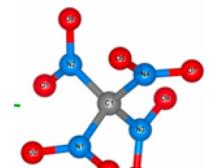
**1D**  
**nanowires**



**0D**  
**Quantum dots**



**0D**  
**molecules**



# Principles of Research

synthesis

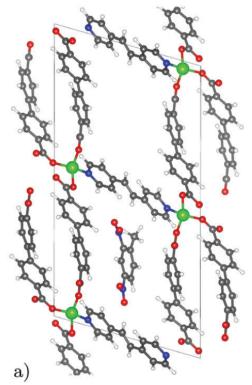
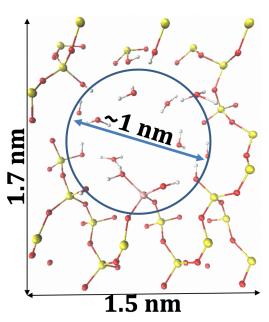


characterization

Study of mechanisms

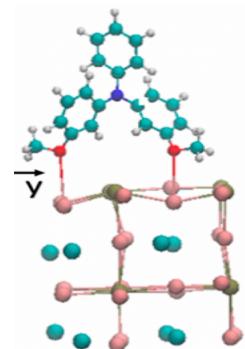


3D  
Porous materials

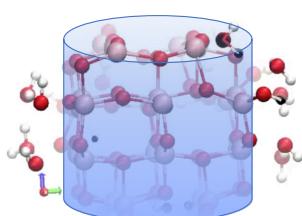


a)

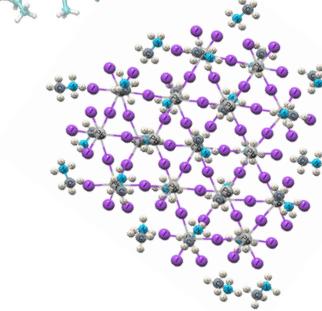
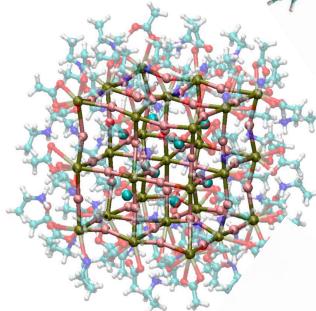
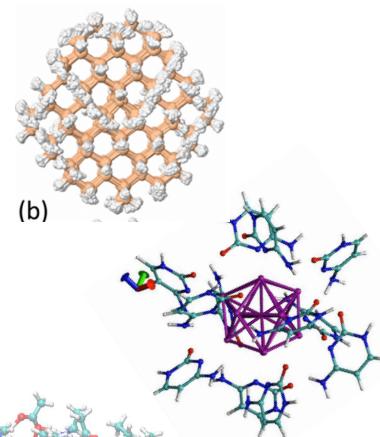
2D  
Planar interfaces



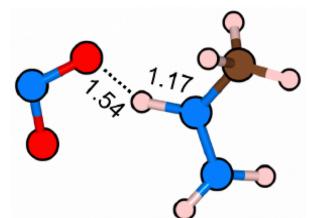
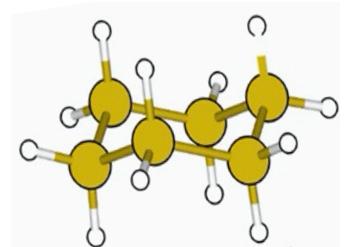
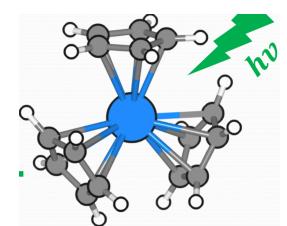
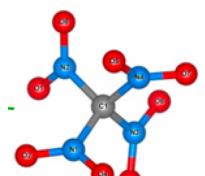
1D  
nanowires



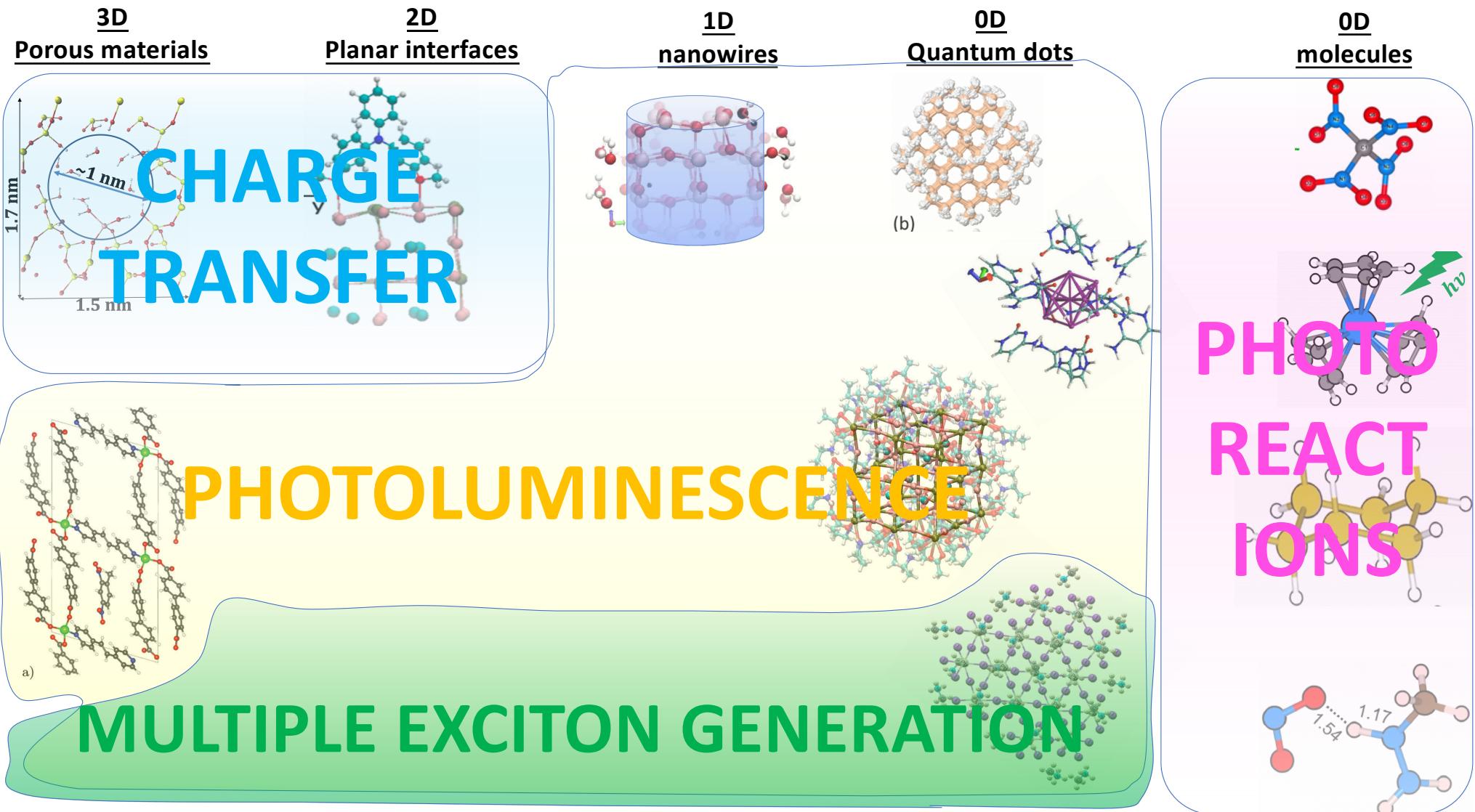
0D  
Quantum dots



0D  
molecules



**WHAT WOULD  
HAPPEN  
TO THESE COMPOUNDS  
UPON  
IRRADIATION  
BY LIGHT?**



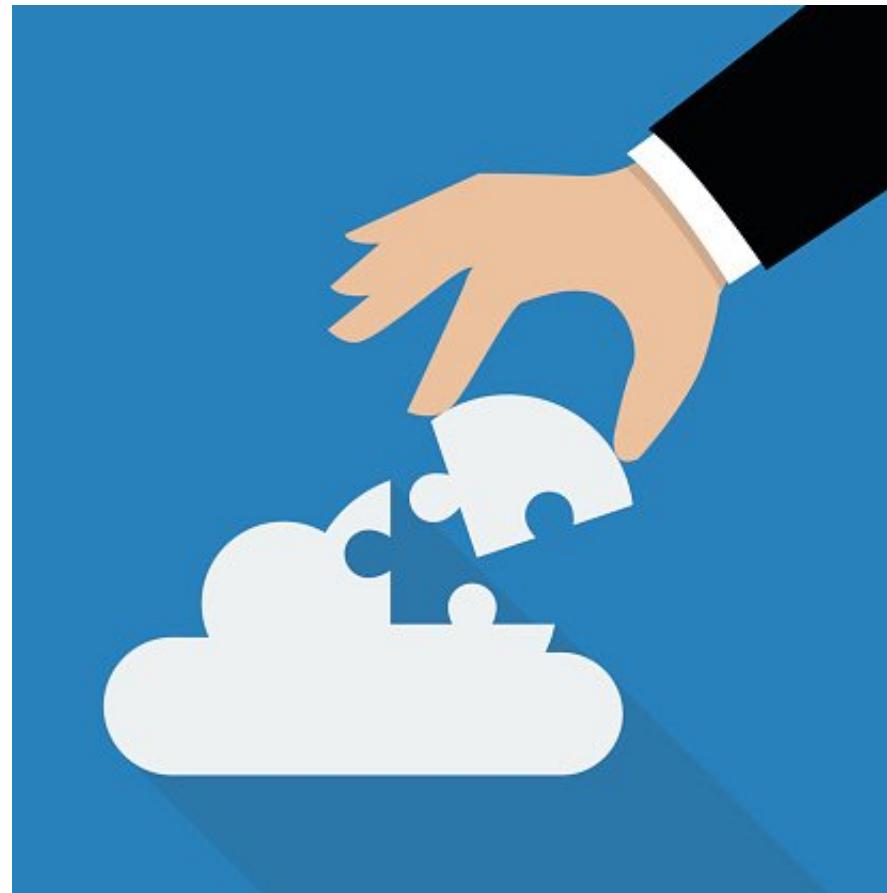


# WHAT'S MISSING?

**Is there a “ready-to-use” public software  
to model photoinduced dynamics?**



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to model photoinduced dynamics?**

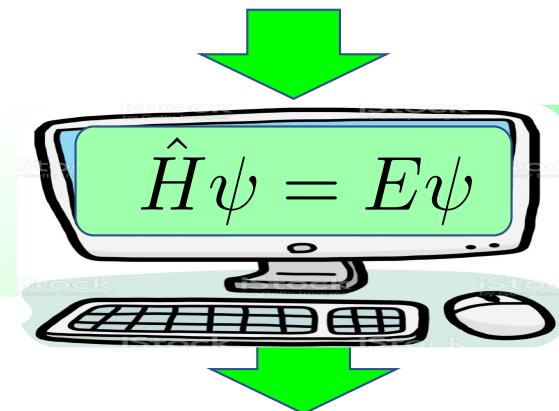


Timing: 1:41

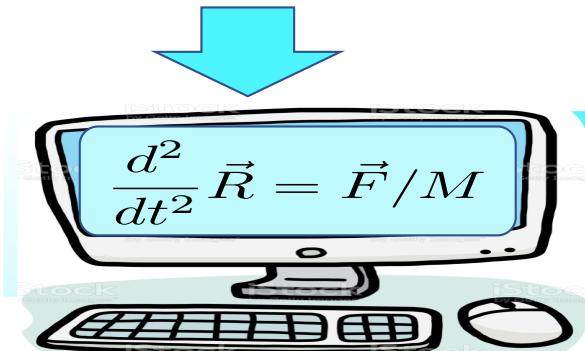
Electronic  
Degrees of freedom

Molecular  
Schrodinger Equation

Nuclear  
degrees of freedom



Nonadiabatic  
coupling



Dissipative  
Density Operator  
equation of motion

Time-dependent  
Perturbation theory

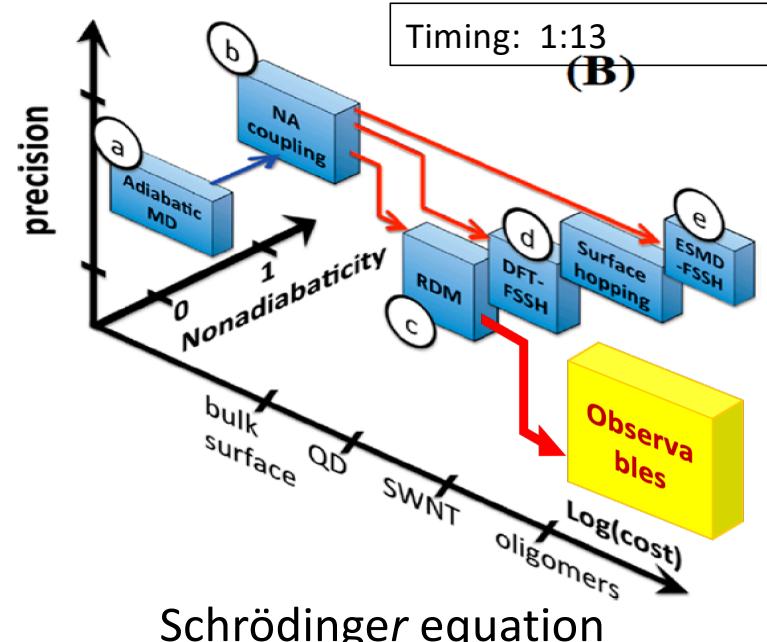
Observables

- Basic computation from standard software
- Excited state dynamics needs  
-homegrown software

## Adjustments of methodology

- Independent Orbitals Approximation
- Extract probabilities of non-radiative transitions from molecular dynamics, by “on-the-fly” protocol
- A very quick original code for dynamics of quantum state
- Time-independent coefficients in equation of motion enable time from 1fs to 1ns

Reference: Kilina, Kilin, Tretiak, *Chemical Reviews* 2015 115 (12), 5929-5978. cited 135



Schrödinger equation

$$i\hbar \frac{\partial}{\partial t} |\psi\rangle = H |\psi\rangle$$



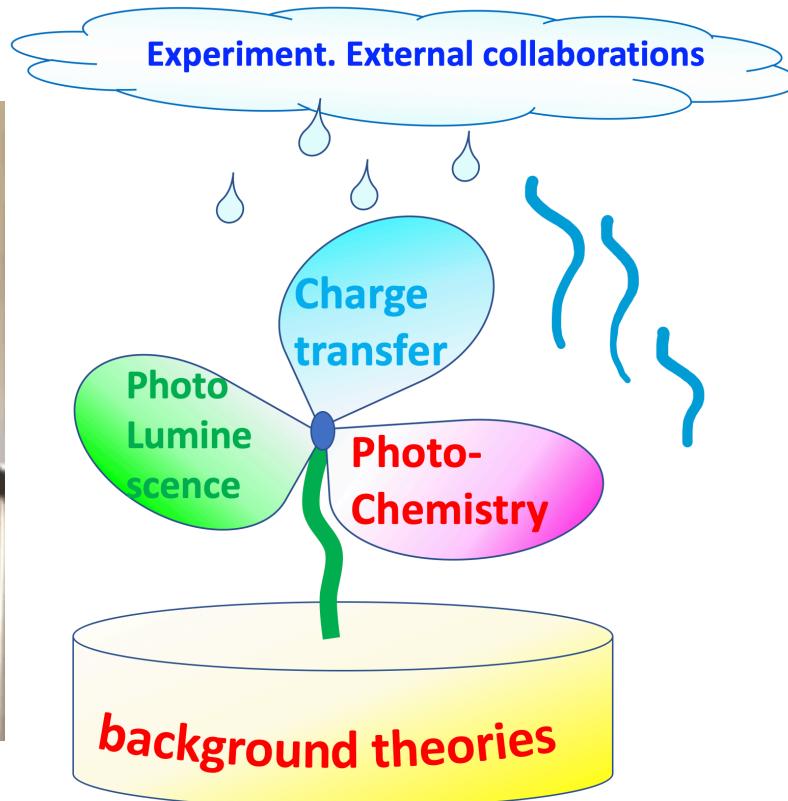
density operator

$$\rho = |\psi\rangle \langle \psi|$$

von Neumann equation

$$i\hbar \frac{\partial}{\partial t} \rho = H |\psi\rangle \langle \psi| - |\psi\rangle \langle \psi| H$$

# Collaboration between Theory and Experiment



Connection Between  
Computational Modeling and  
Experimental Observations

$$\langle \hat{A} \rangle = \int_{-\infty}^{+\infty} \Psi^* \hat{A} \Psi dx$$

$\langle \hat{A} \rangle$  Expectation value of  
 $\Psi(x,t)$  Experimental observable:  
 $\hat{A}$  An average result for a series  
of identical experiments  
 $\Psi$  Wavefunction of the model  
 $\hat{A}$  Quantum operator

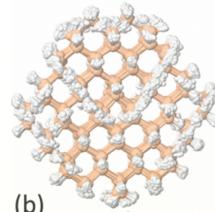
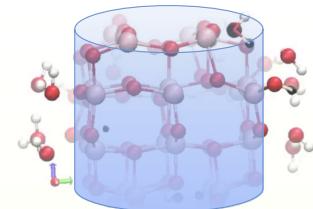
12

3D

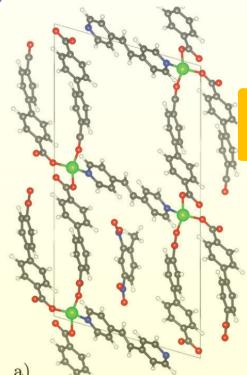
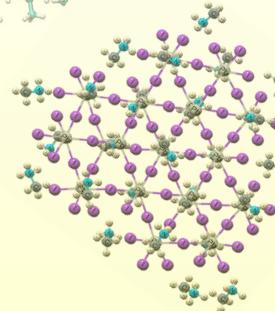
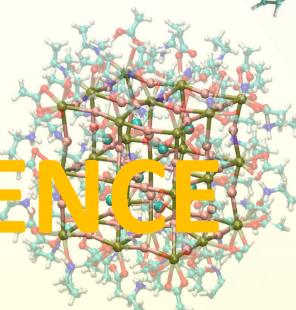
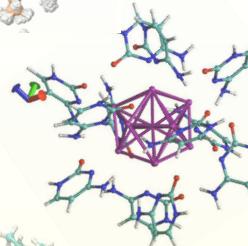
Porous materials

1D  
nanowires

0D  
Quantum dots



(b)



# PHOTOLUMINESCENCE

# PHOTOLUMINESCENCE

-At which wavelength the material emits light?



-Does emission occur at the same wavelength  
or in a range of frequencies?

-Does temperature enhance or suppresses emission?

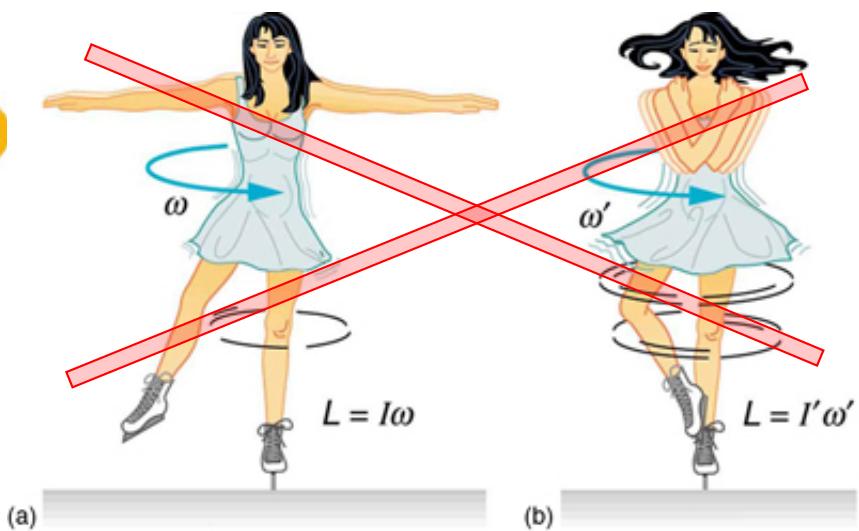
-What is more probable:  
to emit light or to convert energy into heat?

-Do surface adsorbates  
suppress or enhance photoluminescence?

# PHOTOLUMINESCENCE

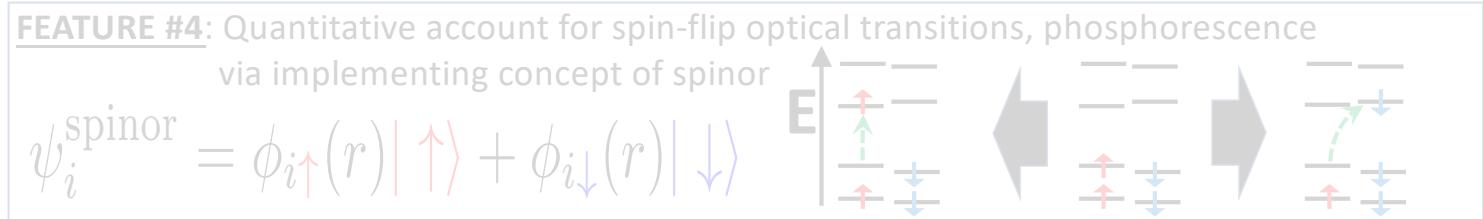
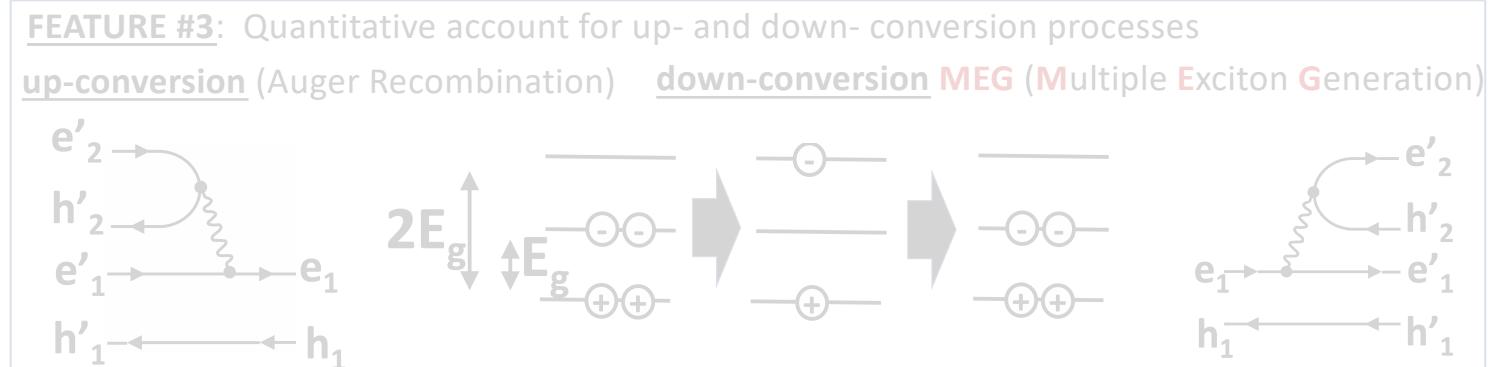
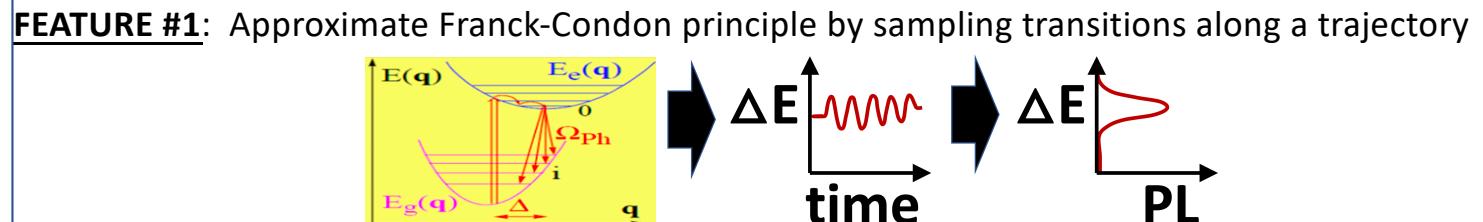


-materials composed of heavy transition metal ions and lanthanides may experience **violation of spin-conservation**



## Specific methods adjustments for modeling of photoluminescence

- sampling
- time integration
- up-/down- conversion
- spinors



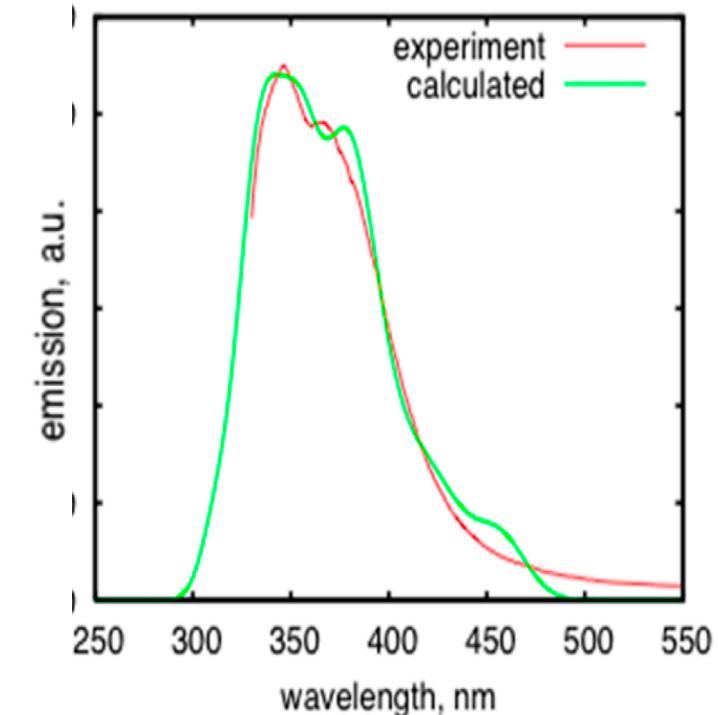
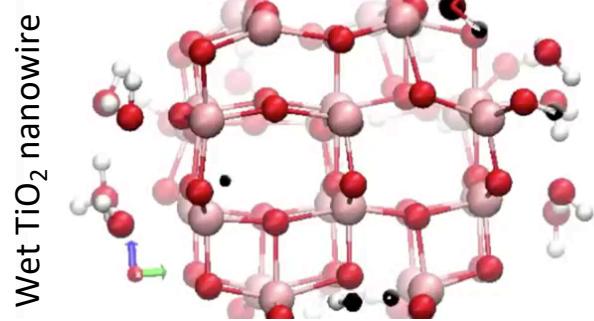
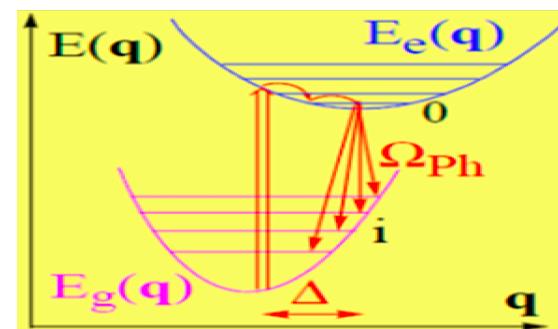
Research Direction: (ii) Photoluminescence

Timing: 2:29



Linewidth thermal broadening of PL signal via MD sampling  
Suggested abbreviation:

Molecular Dynamics Photo Luminescence (**MDPL**)



Justification: quantitatively match experimental data

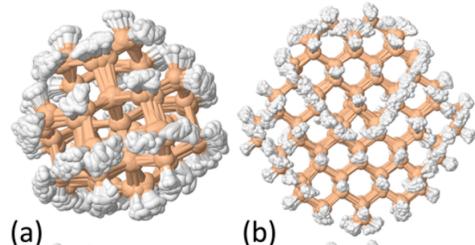
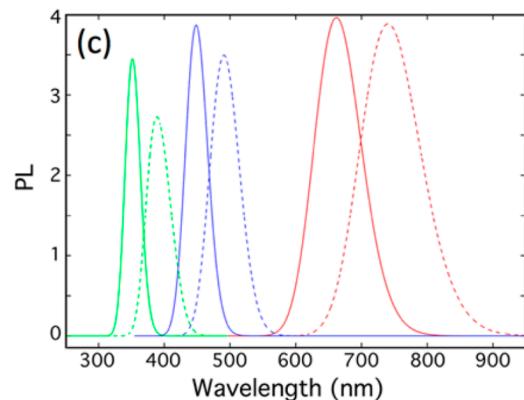
Reference: Vogel, Kilin, *J. Phys. Chem. C* 2015, 119, 27954–27964

Research Direction: (ii) Photoluminescence

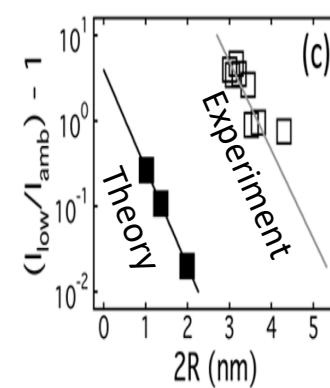
### Enhancing photoluminescence in silicon quantum dots by cooling



#### Theory

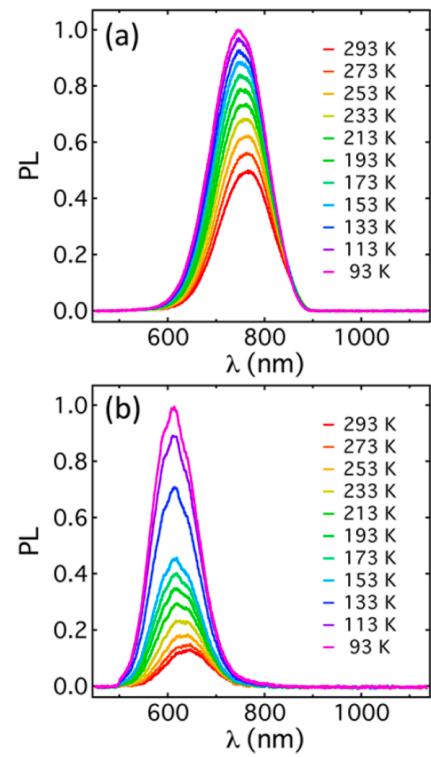


#### Comparison



Relative increase in PL upon cooling.  
experiments (open)  
computation (closed)  
fit to the same trend.

#### Experiment



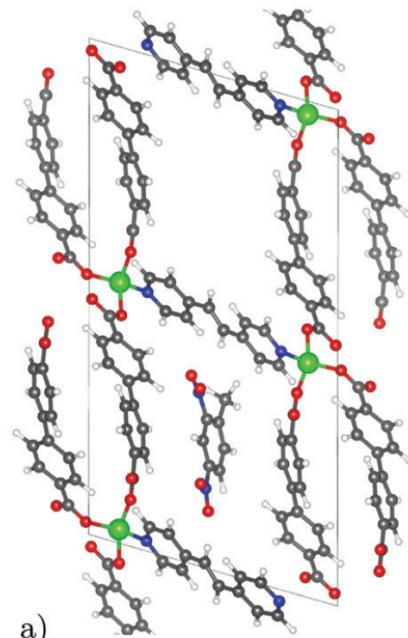
Justification: matches experimental trends

Reference: Brown, Vogel, Miller, Inerbaev, Anthony, Kortshagen, Kilin, Hobbie, *J. Phys. Chem. C* **2016**, *120*, 18909

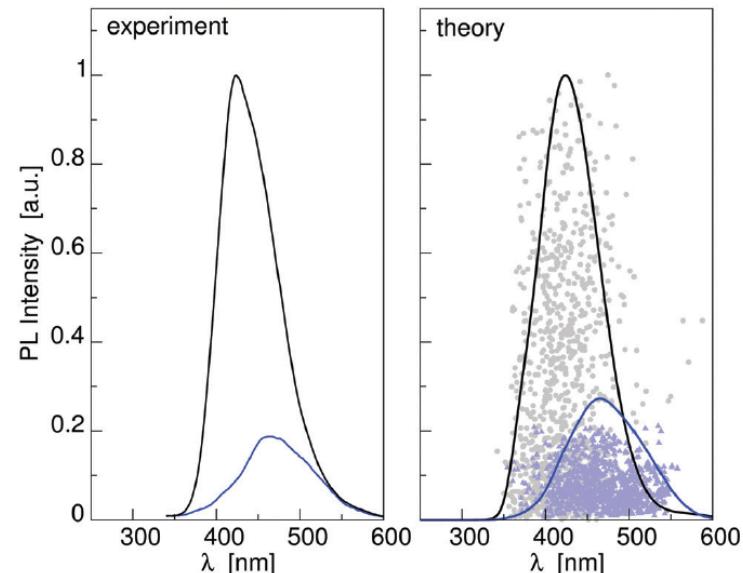
Research Direction: (ii) Photoluminescence



### Quenching of photoluminescence in a Zn-MOF sensor by nitroaromatic molecules



a)  
RPM3-Zn i.e.  $Zn_2(badc)_2(bpee)$   
 $bpee = 1,2\text{-bipyridylethene}$   
 $badc = 4,4'\text{-biphenyldicarboxylate}$   
 $DNT = 2,4\text{-dinitrotoluene}$



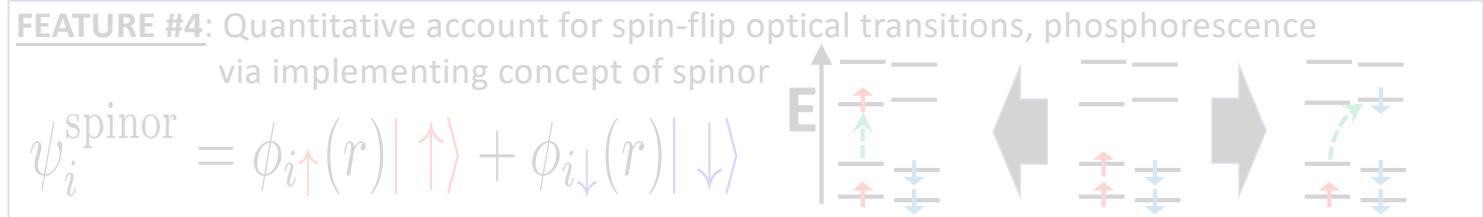
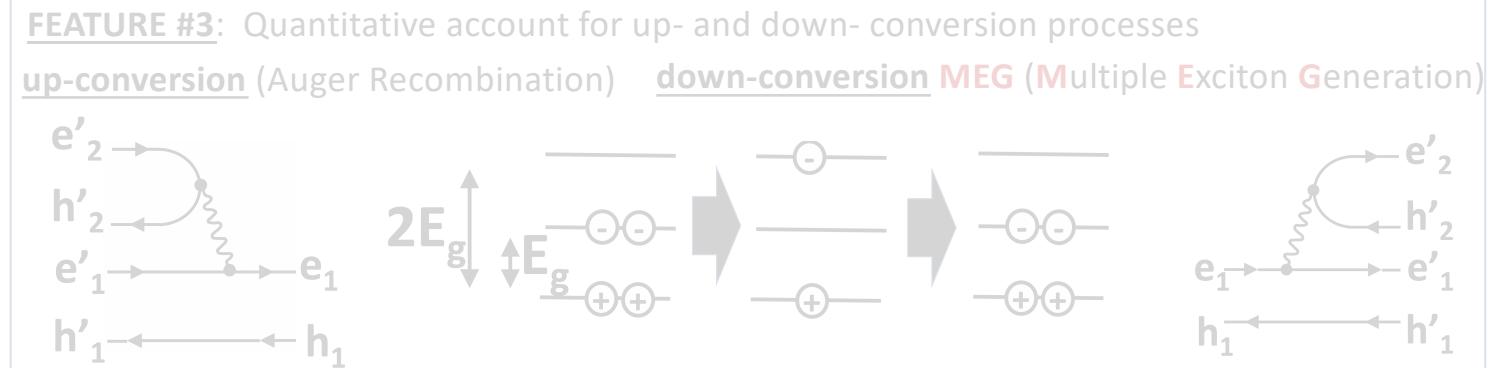
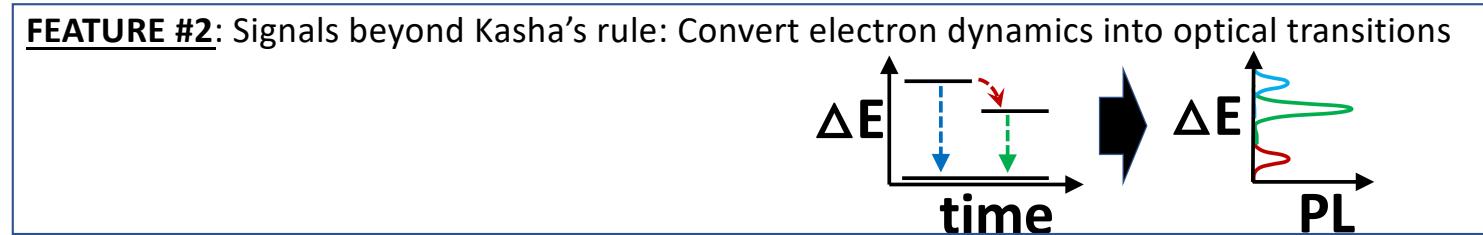
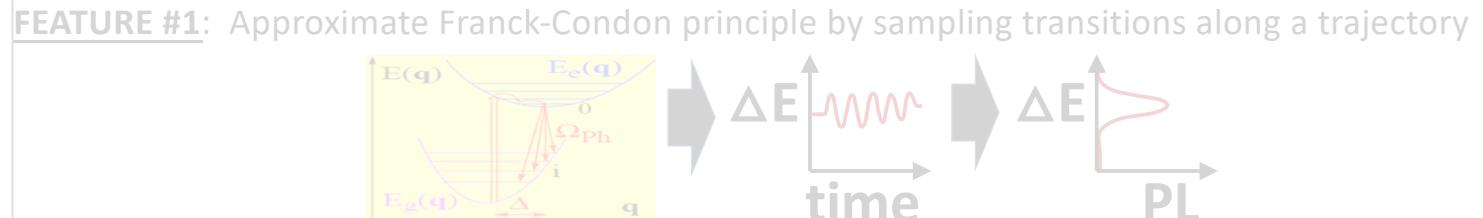
Intensity of PL by RPM3-Zn-MOF decreases with adsorption of explosive dinitro-toluene  
Upon photoexcitation  $(pbdc)^+(bpee)^-$   
electron migrates from  $(bpee)$  to  $(DNT)$   
Resulting excitations  $(pbdc)^+(DNT)^-$   
has CT character and low oscillator strength

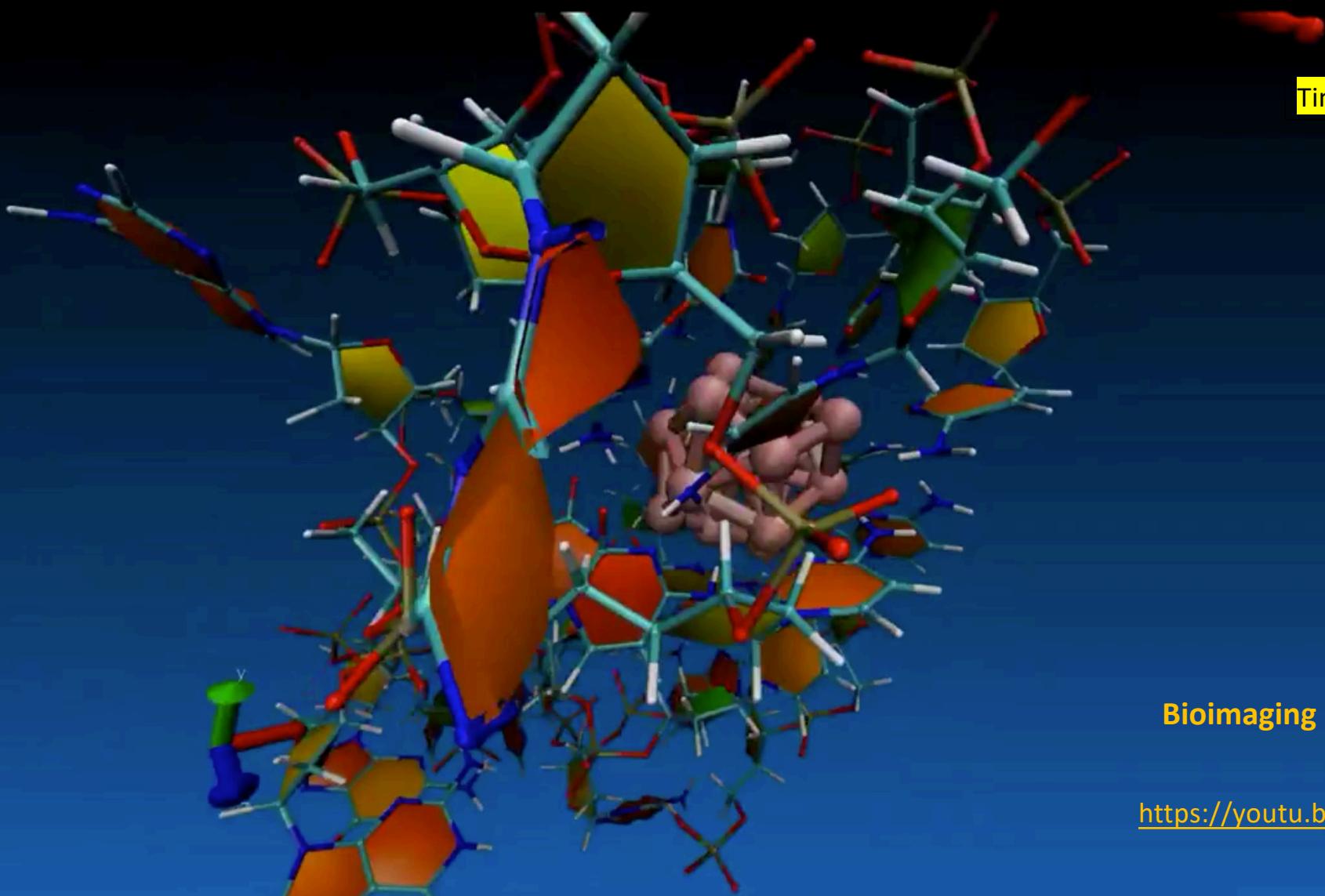
Justification: quantitatively matches experimental observables!

Reference: Jensen, Tan, Lustig, Kilin, Li, Chabal, Thonhouser *J. Mater. Chem. C*, **2019**, 7, 2625-2632

## Specific methods adjustments for modeling of photoluminescence

- sampling
- time integration
- up-/down- conversion
- spinors





Timing: 0:33

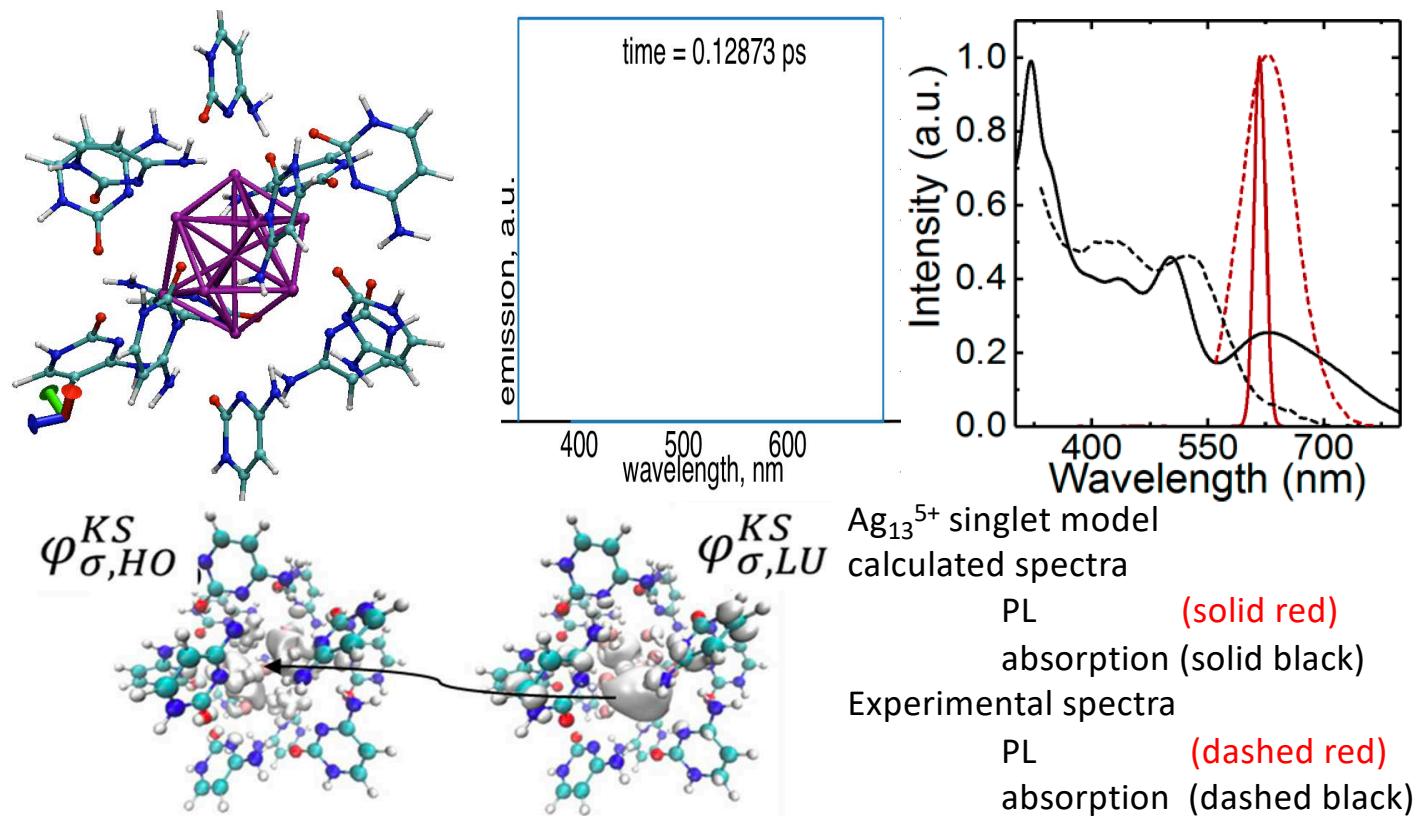
Bioimaging labels

<https://youtu.be/fhKn5HAcq8>

Research Direction: (ii) Photoluminescence

## Fluorescence in DNA-protected Silver Nanoclusters

Timing: 4:00

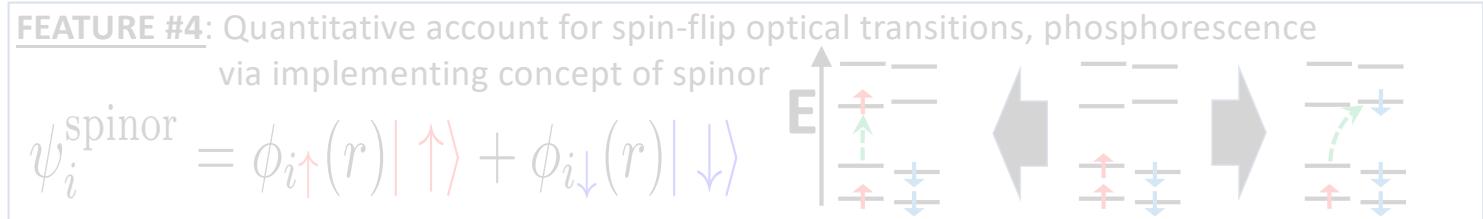
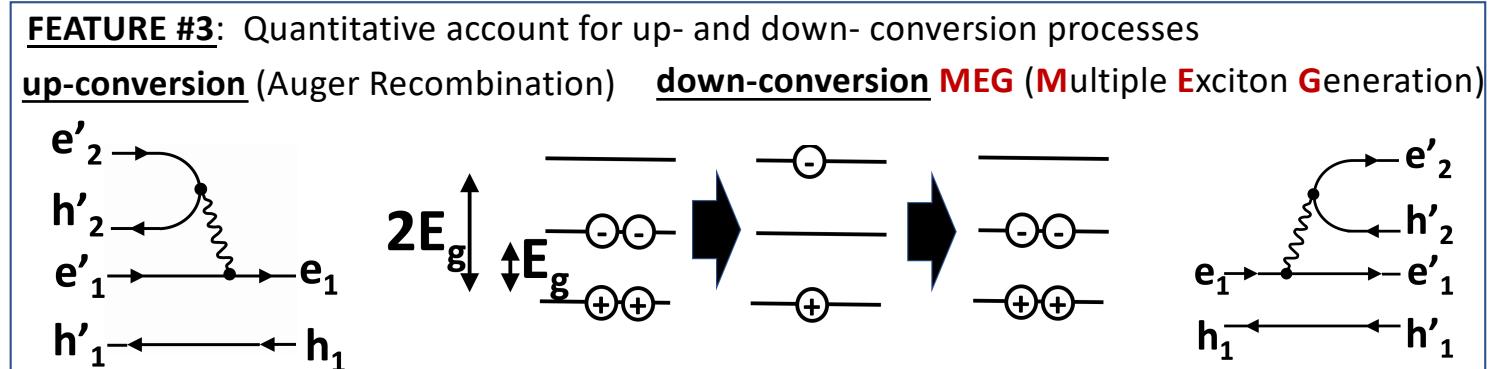
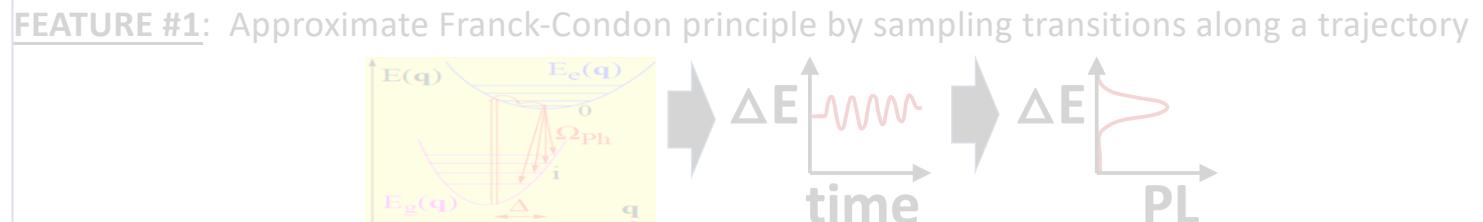


Justification: matches experimental trends, identified bright charge transfer excitations

Reference: Brown, Hobbie, Tretiak, Kilin, *J. Phys. Chem. C* **2017**, 121, 23875-23885

## Specific methods adjustments for modeling of photoluminescence

- sampling
- time integration
- up-/down- conversion
- spinors

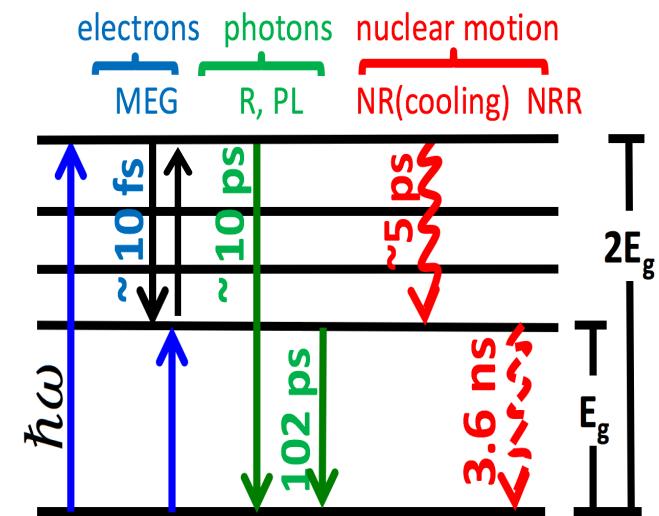
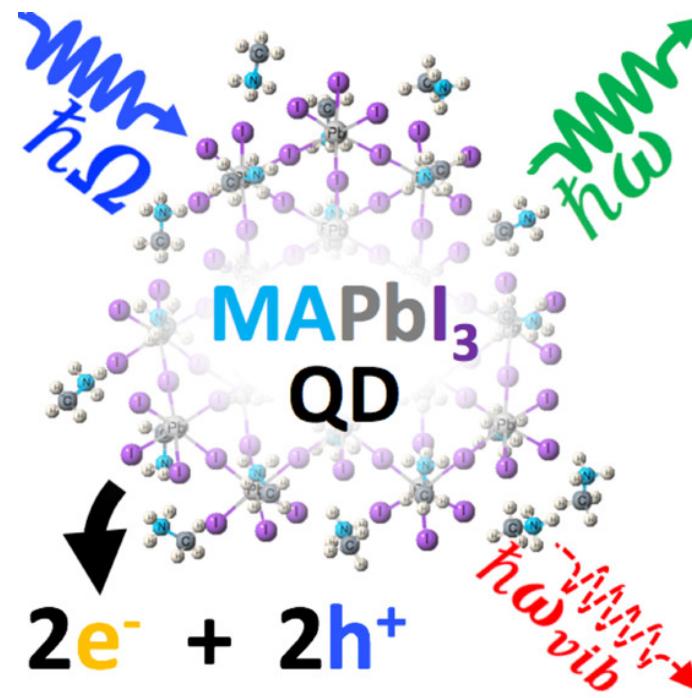


Research Direction: (ii) Photoluminescence

Timing: 2:43



### Quantum Confinement in Lead Halide Perovskite Quantum Dots



Justification: predict outcome of measurement prior to experimental observations

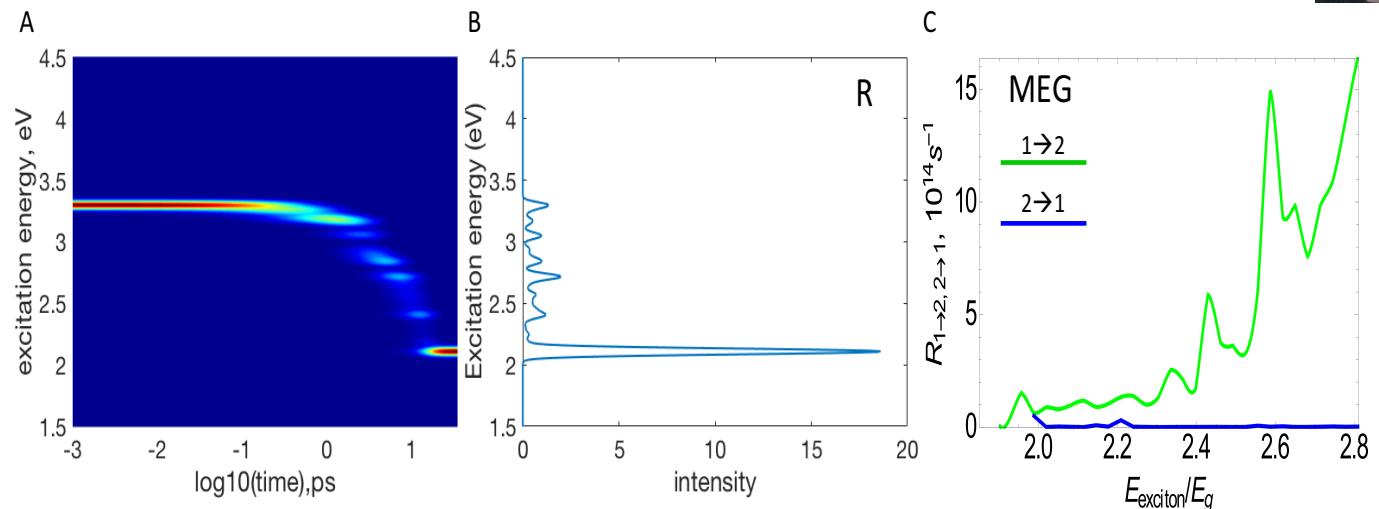
Reference: Vogel, Kryjevski, Inerbaev, Kilin *J. Phys. Chem. Lett.* **2017** 8, 3032

Research Direction: (ii) Photoluminescence

Timing: 2:35



### Quantum Confinement in Lead Halide Perovskite Quantum Dots



Signatures of size confinement were seen

- increased spacing of energy states
- localized electron density

The increased intra-band energy separation

- slows down the nonradiative relaxation
- by requiring multiple phonon modes to dissipate the hot electron energy.

**Multiple Exciton Generation (MEG)**

- is facilitated by confinement
- is beneficial for photovoltaics
- is adverse for light emission

**Ultra-small Quantum Dots**

- will be less efficient light emitters
- than mid-size ones

Justification: predict outcome of measurement prior to experimental observations

Reference: Vogel, Kryjevski, Inerbaev, Kilin *J. Phys. Chem. Lett.* **2017** 8, 3032

Research Direction: (ii) Photoluminescence

Timing: 0:59



## Quantum Confinement in Lead Halide Perovskite Quantum Dots



Letter



ARTICLE

DOI: 10.1038/s41467-018-06596-1 OPEN

### Low threshold and efficient multiple exciton generation in halide perovskite nanocrystals

Mingjie Li<sup>1</sup>, Raihana Begum<sup>2</sup>, Jianhui Fu<sup>1</sup>, Qiang Xu<sup>1</sup>, Teck Ming Koh<sup>2</sup>, Sjoerd A. Veldhuis<sup>2</sup>, Michael Grätzel<sup>3</sup>, Nripan Mathews<sup>2,4</sup>, Subodh Mhaisalkar<sup>1,2,4</sup> & Tze Chien Sum<sup>1</sup>

MEG<sup>36,37</sup>. Furthermore, another recent computational work<sup>38</sup> also demonstrated low-threshold MEG (close to  $2E_g$ ) in  $\text{MAPbI}_3$  NCs. The authors attributed that the stronger Coulomb coupling between the initial single-exciton and final-biexciton states, and the longer hot-carrier cooling of highly excited states were more conducive for MEG. Their calculated MEG process was on a time scale of tens of fs, which agrees well with our measured value of smaller than 90 fs. Although,  $\text{PbSe}$  or  $\text{PbS}$  also have similar small

38. Vogel, D. J., Kryjevski, A., Inerbaev, T. & Kilin, D. S. Photoinduced single- and multiple-electron dynamics processes enhanced by quantum confinement in lead halide perovskite quantum dots. *J. Phys. Chem. Lett.* **8**, 3032–3039 (2017).

Justification: predict outcome of measurement prior to experimental observations

Reference: Vogel, Kryjevski, Inerbaev, Kilin *J. Phys. Chem. Lett.* **2017** *8*, 3032-3039. cited 39 times

## Specific methods adjustments for modeling of photoluminescence

- sampling
- time integration
- up-/down- conversion
- spinors

**FEATURE #1:** Approximate Franck-Condon principle by sampling transitions along a trajectory

**FEATURE #2:** Signals beyond Kasha's rule: Convert electron dynamics into optical transitions

**FEATURE #3:** Quantitative account for up- and down- conversion processes

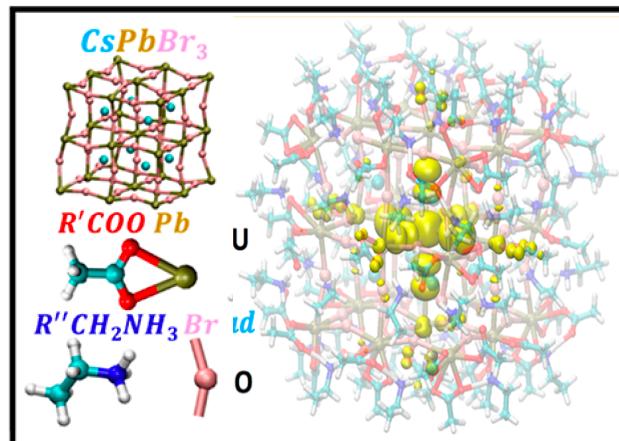
up-conversion (Auger Recombination)    down-conversion MEG (Multiple Exciton Generation)

**FEATURE #4:** Quantitative account for spin-flip optical transitions, phosphorescence via implementing concept of spinor

$$\psi_i^{\text{spinor}} = \phi_{i\uparrow}(r)|\uparrow\rangle + \phi_{i\downarrow}(r)|\downarrow\rangle$$

Research Direction: (ii) Photoluminescence

Timing: 4:11



Hot electron cooling

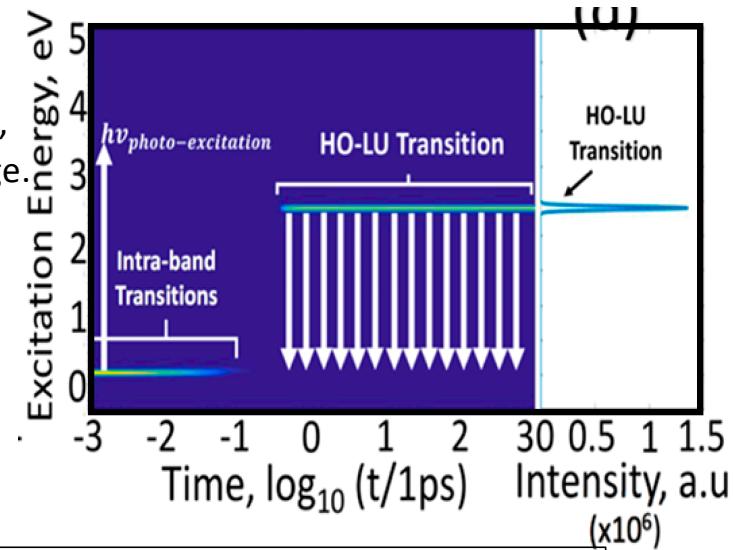
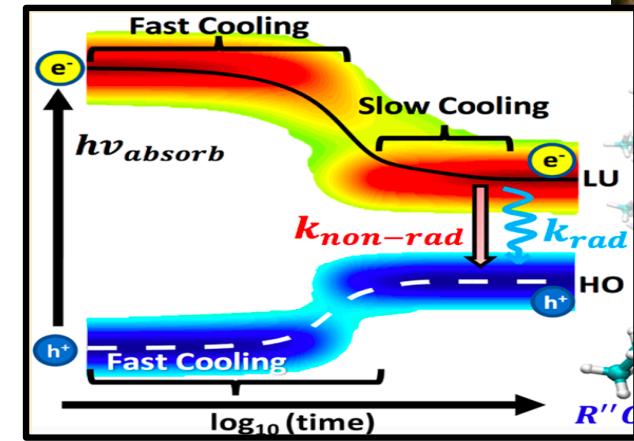
--slows down at the band edge,  
due to large SOC and strong confinement,  
which gives sub-gaps above the band edge.

Strong confined nanocrystals

--are better for photovoltaics

Weak confined nanocrystals

--are better for light-emitting

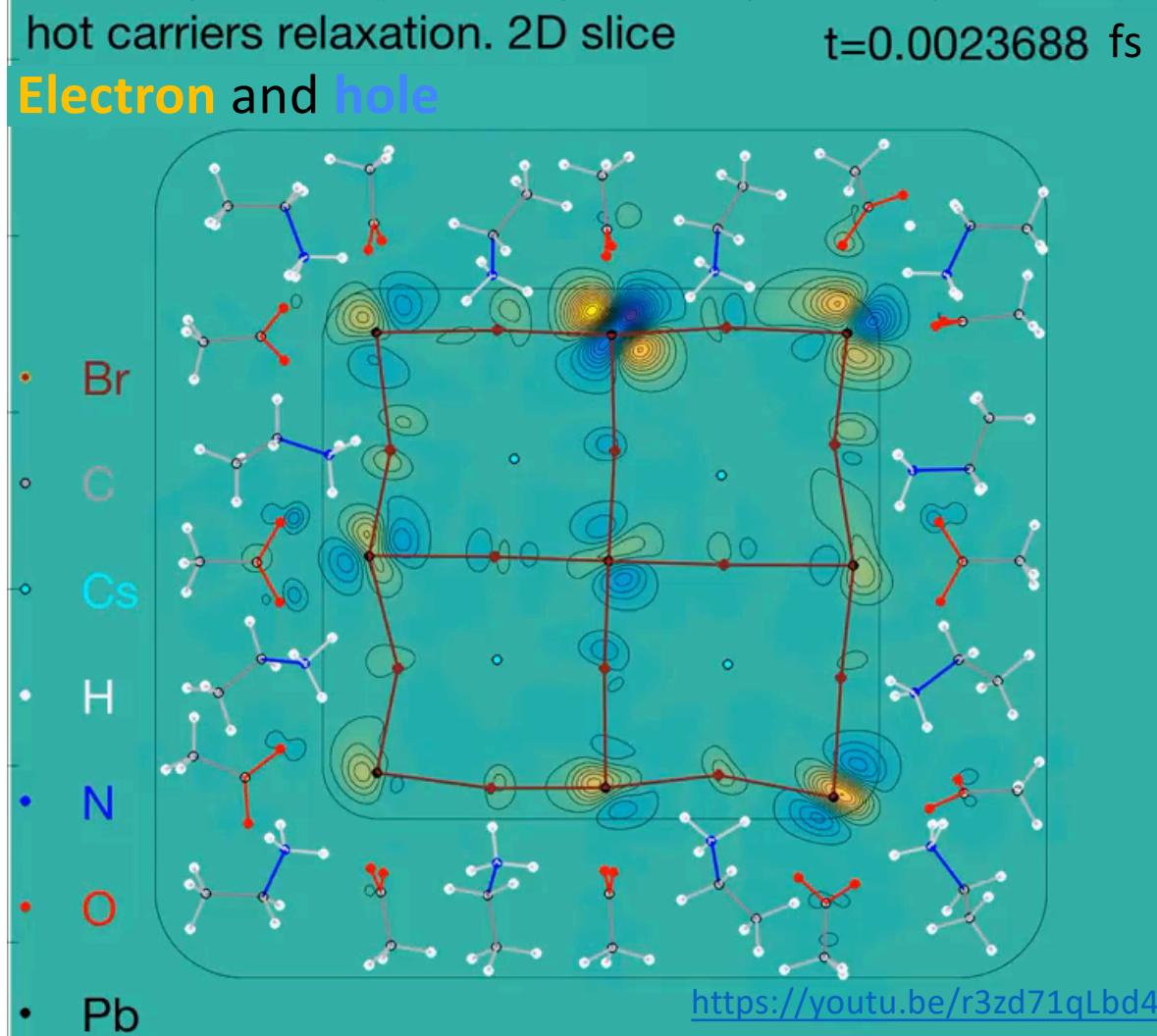


Justification: PLQY=53% matches experiment

28

Reference: Forde, Inerbaev, Hobbie, Kilin, *J. Am. Chem. Soc.* **2019**, 141, 10, 4388–4397

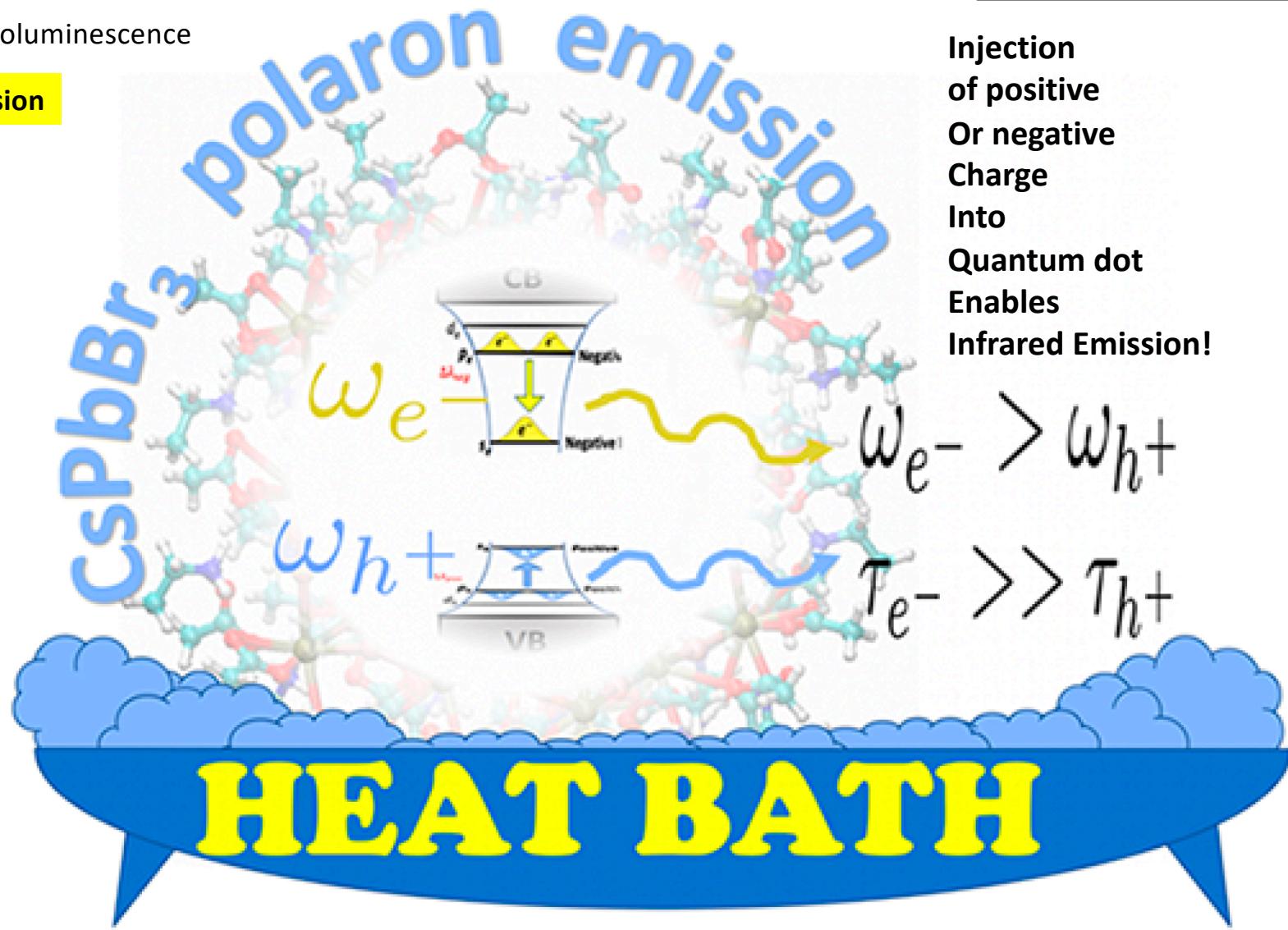
Research Direction: (ii) Photoluminescence



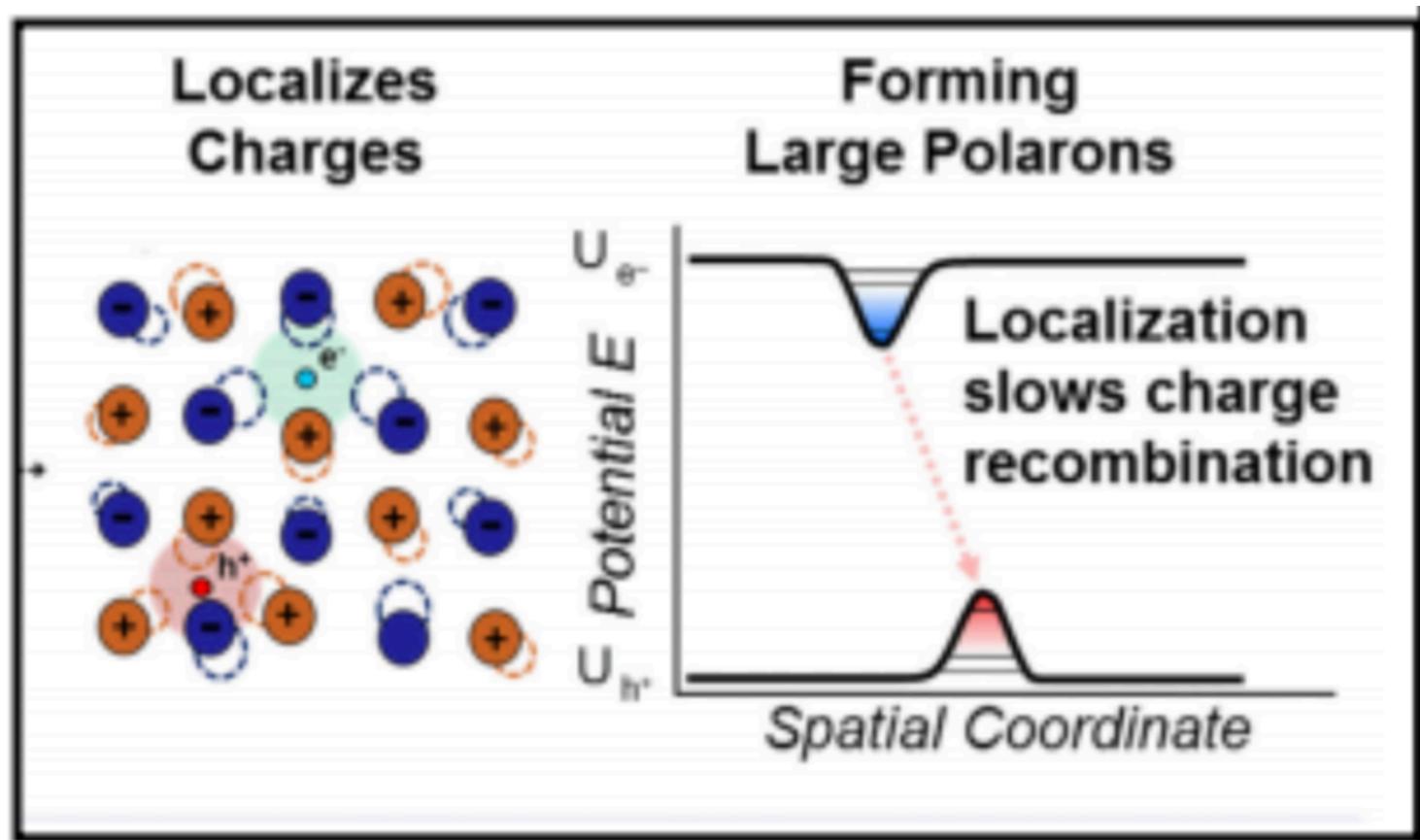
Reference: Forde, Inerbaev, Hobbie, Kilin, *J. Am. Chem. Soc.* **2019**, 141, 10, 4388–4397

Research Direction: (ii) Photoluminescence

Hot polaron Infrared emission

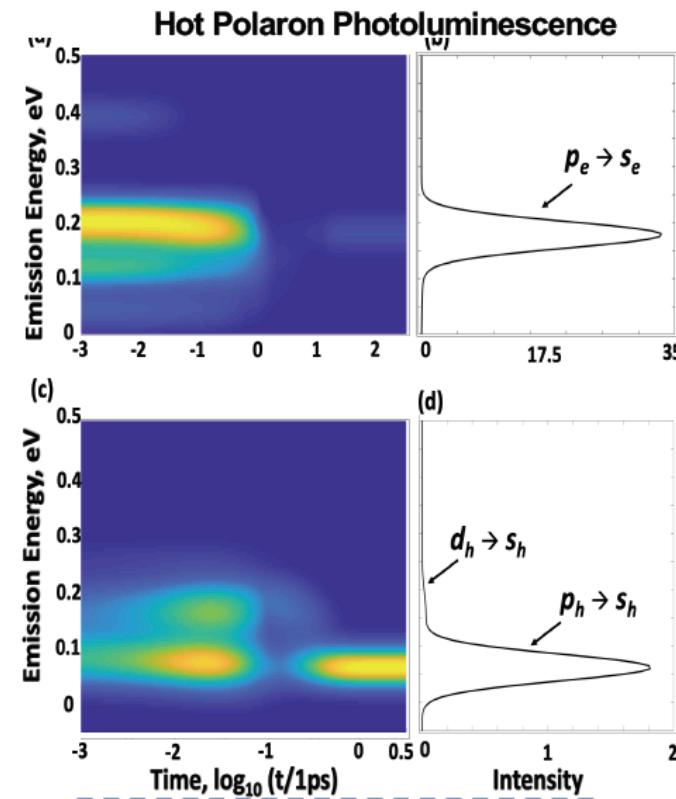
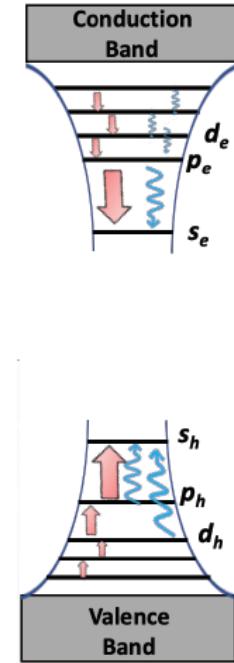


Hot polaron Infrared emission



Research Direction: (ii) Photoluminescence

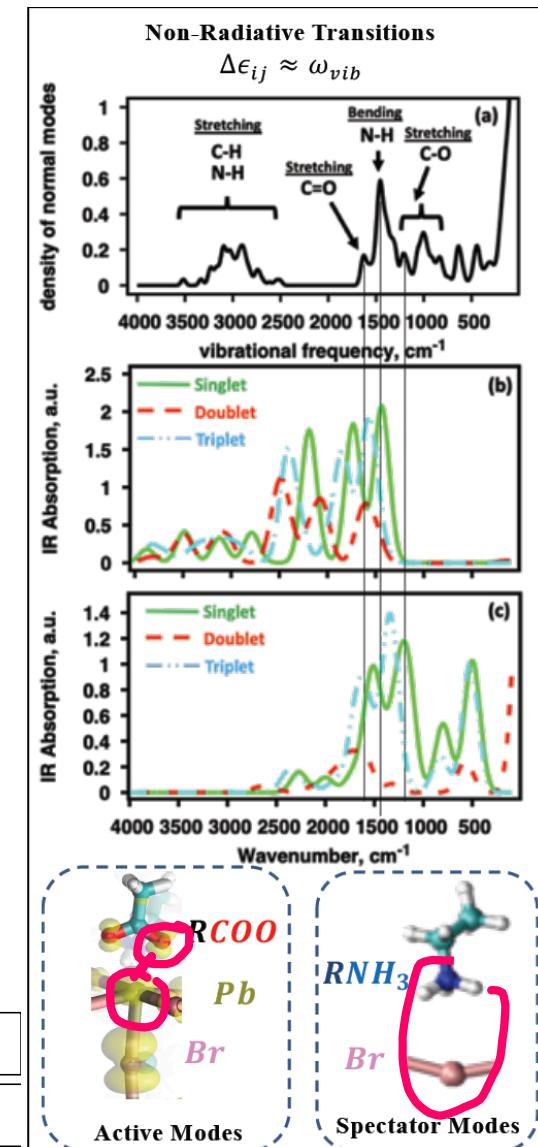
### Hot polaron Infrared emission



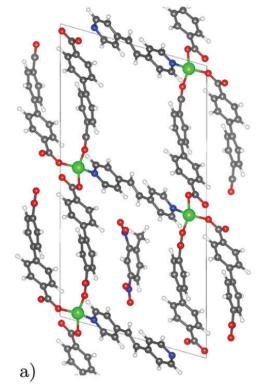
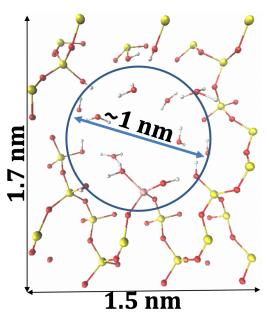
Model	$f_{PGS-RES}$	$k_r [1/\text{ns}]$	$k_{nr} [1/\text{ns}]$	PLQY	$\log(\text{PLQY})$
Positive-Singlet	0.14	$2.81 \times 10^{-4}$	$2.08 \times 10^3$	$1.35 \times 10^{-7}$	-7
Negative-Singlet	0.29	$5.10 \times 10^{-3}$	$1.64 \times 10^1$	$3.11 \times 10^{-4}$	-4

Justification: Prediction of a property ahead of experimental measurement

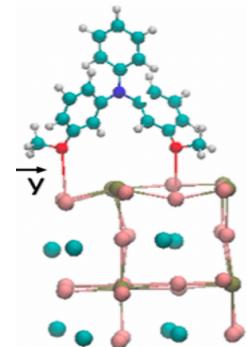
Reference: Forde, Inerbaev, Kilin, *J. Phys. Chem. C* **2020**, 124, 1, 1027–1041



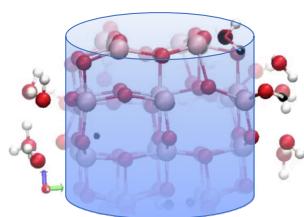
**3D**  
**Porous materials**



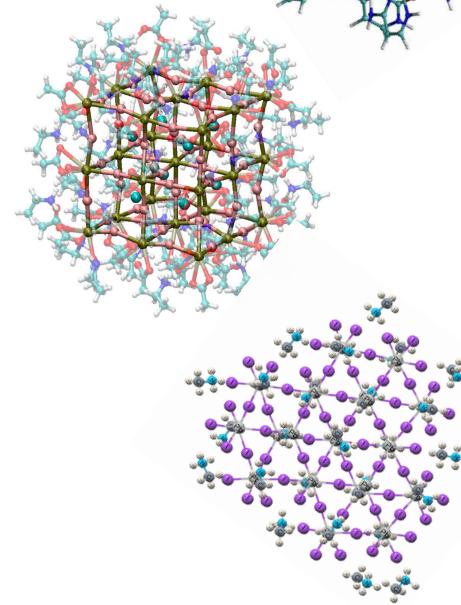
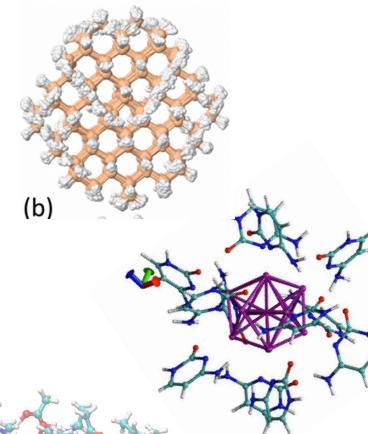
**2D**  
**Planar interfaces**



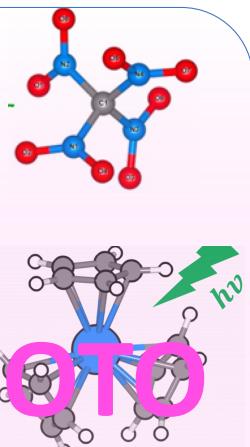
**1D**  
**nanowires**



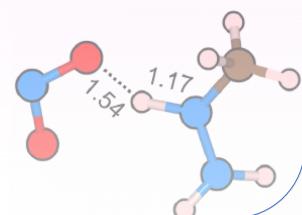
**0D**  
**Quantum dots**



**0D**  
**molecules**



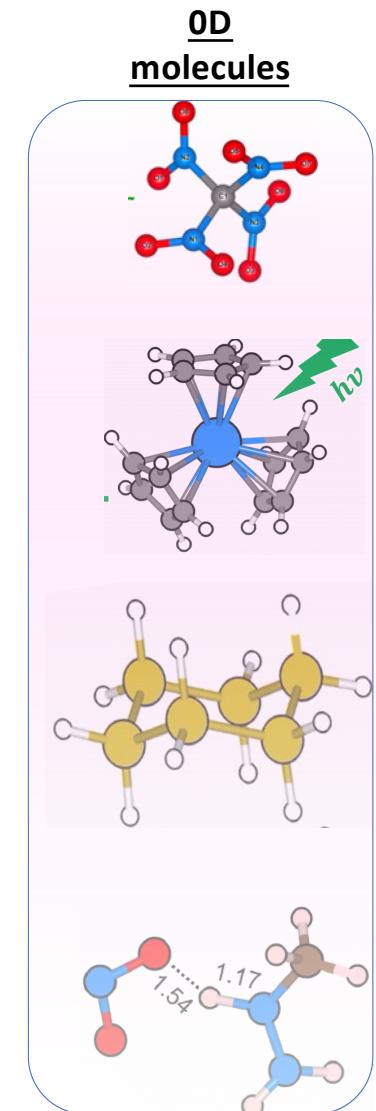
**PHOTO  
REACT  
IONS**



# PHOTOREACTIONS



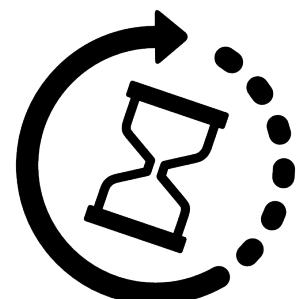
- does light change a molecule?
- which frequency / intensity of light facilitate most vivid reaction?
- which molecules are produced by a photoreaction?
- what is relative proportion of different product molecules?
- what would be a mass-spectrum for distribution of photoreaction' products?

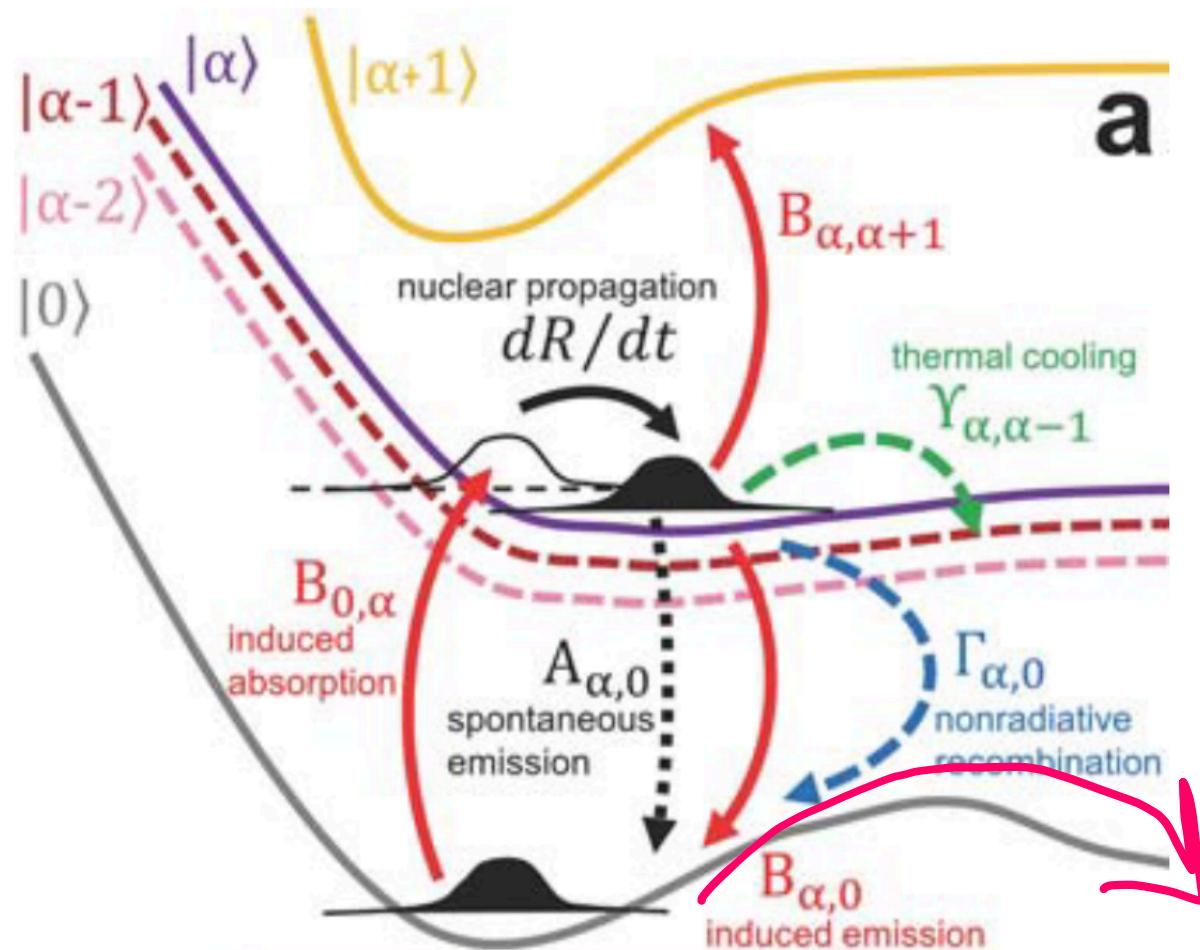


# PHOTOREACTIONS



- “fork road”  
**ambiguity of reaction pathways  
originating from the same precursor**
- Many isomers!
- A need to choose pathway/target isomer  
by tuning laser light parameters
- Very long time-scales, hard to simulate





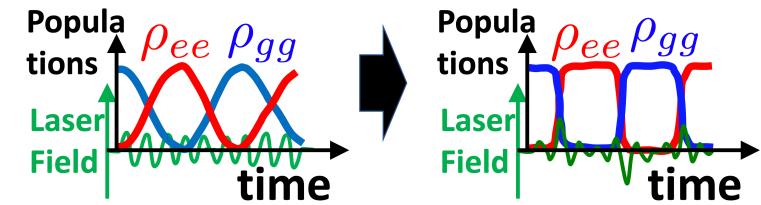
## Specific methods adjustments for modeling of photoreactions

Timing: 2:09

**FEATURE #1:** Substantially shorten computational resources/simulation time by  
\*avoiding time-dependence of total energy via **RWA** (Rotating Wave Approximation)

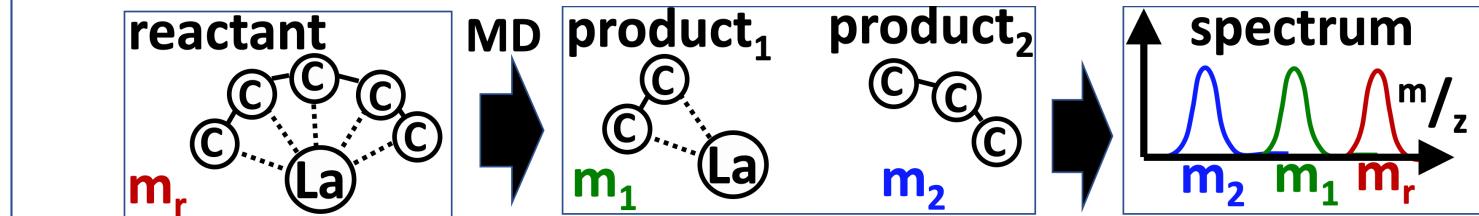
\*replacing continuous change of state' population by instantaneous hops

\*use of "pulse-area theorem"

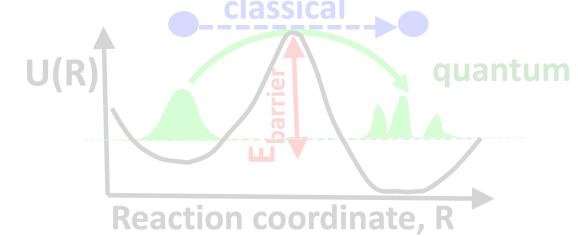


**FEATURE #2:** developed method which models irradiation of the model with  
\*more than two photons

**FEATURE #3:** mass-spectra became available via unique computational procedure



**FEATURE #4:** activation energies of reactions are computed more accurately  
\* via quantum tunneling of quantized nuclei

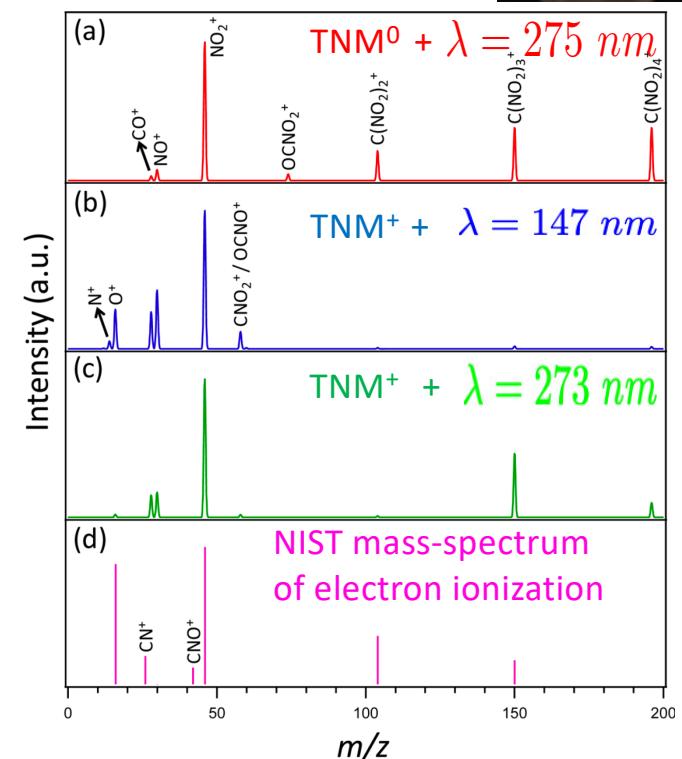
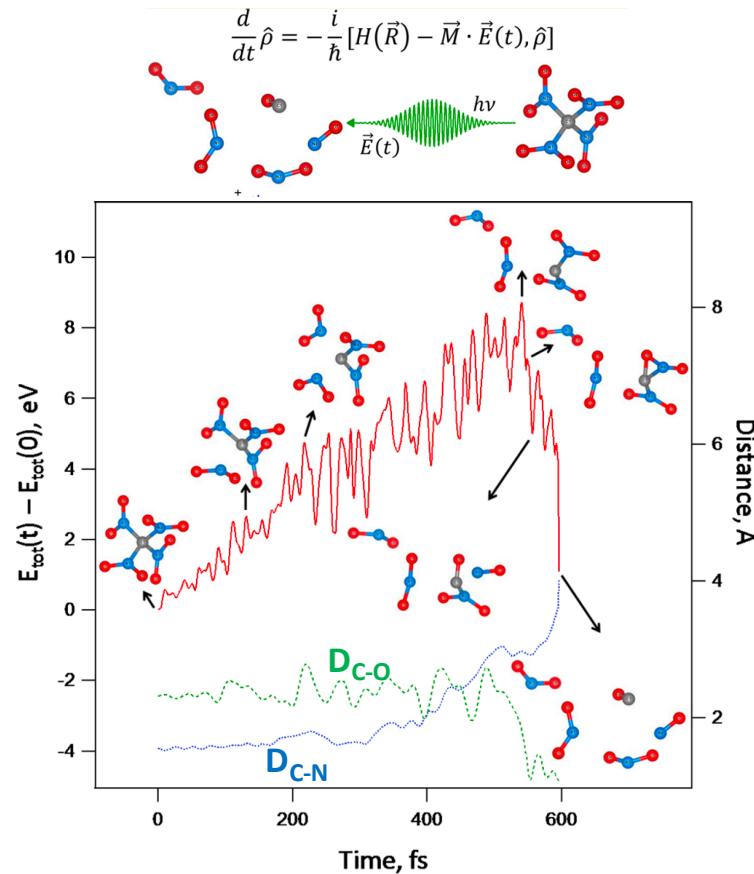
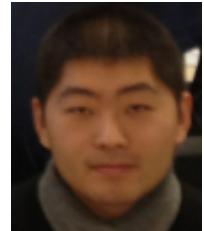


Timing: 0:25



Research Direction: (iii) Light-induced Reactions

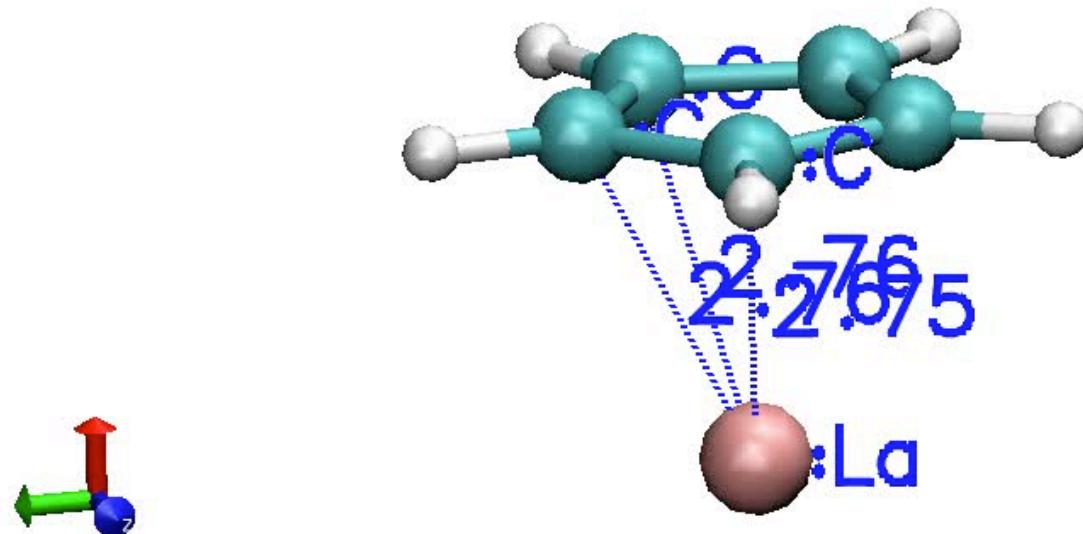
### Photofragmentation of Tetranitromethane



Justification: matches experimental trends, control of reaction outcome by light wavelength and charge

Reference: Han, Rasulev, Kilin, *J. Phys. Chem. Lett.* **2017**, 8, 3185–3192

ALL STAGES OF  
PHOTOREACTION



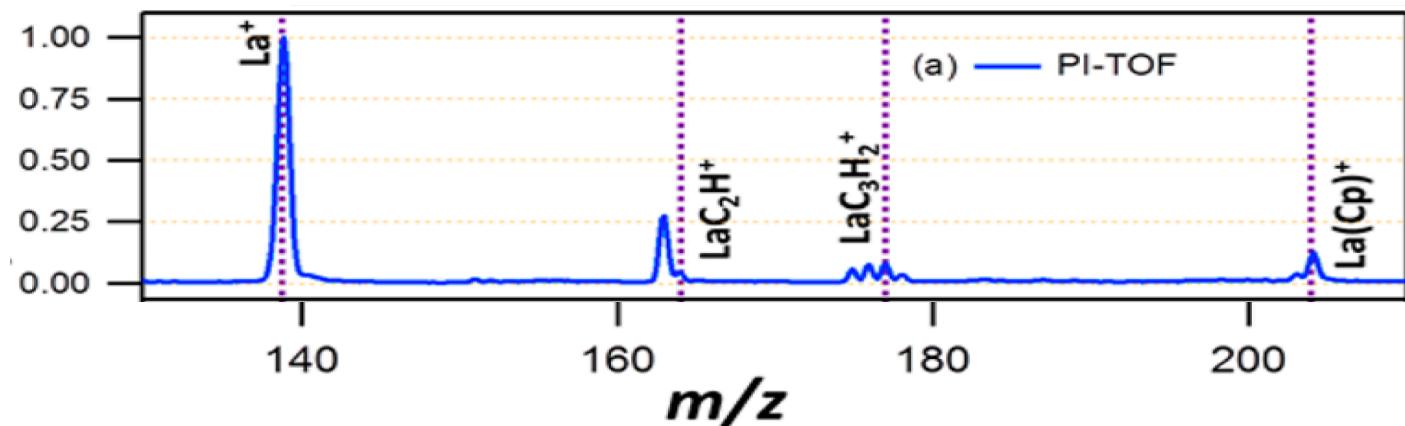
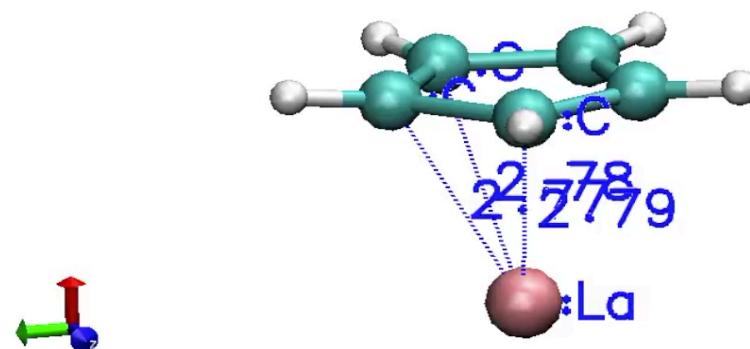
Justification: Quantitatively matches experiment: mass and intensity in the mass-spectra

Reference: Han, Meng, Rasulev, May, Berry, Kilin, *J. Chem. Theory Comput.* **2017**, 13, 4281–4296

Research Direction: (iii) Light-induced Reactions

**STAGE 1:  $\text{LaCp} \rightarrow \text{LaC}_3\text{H}_2^+$**

Timing: 0:29

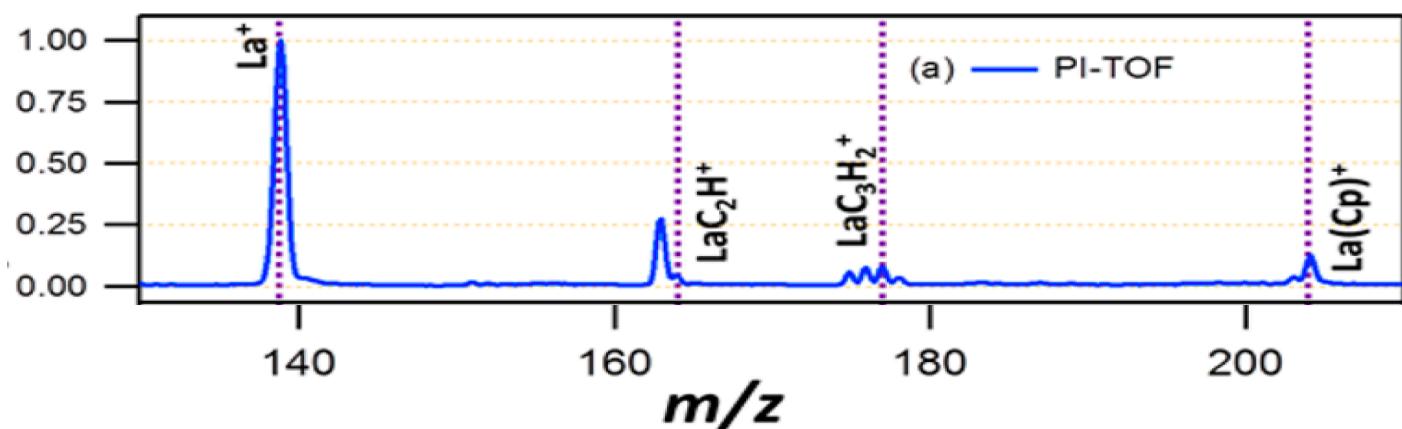
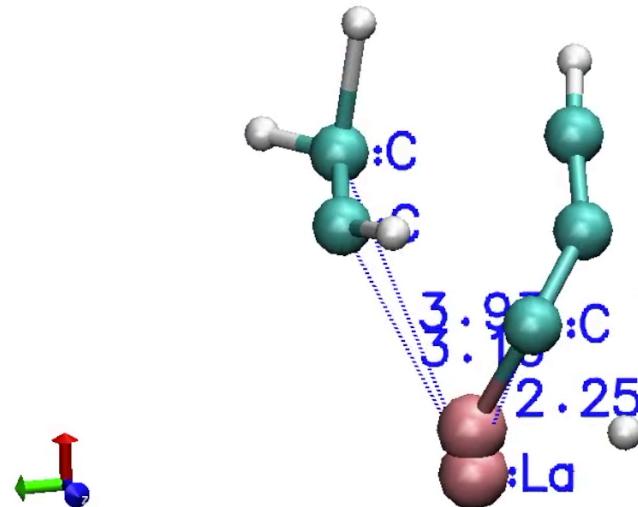


Reference: Han, Meng, Rasulev, May, Berry, Kilin, *J. Chem. Theory Comput.* **2017**, 13, 4281–4296

Research Direction: (iii) Light-induced Reactions

**STAGE 2:  $\text{LaCp} \rightarrow \text{LaC}_2\text{H}_3^+$**

Timing: 0:12

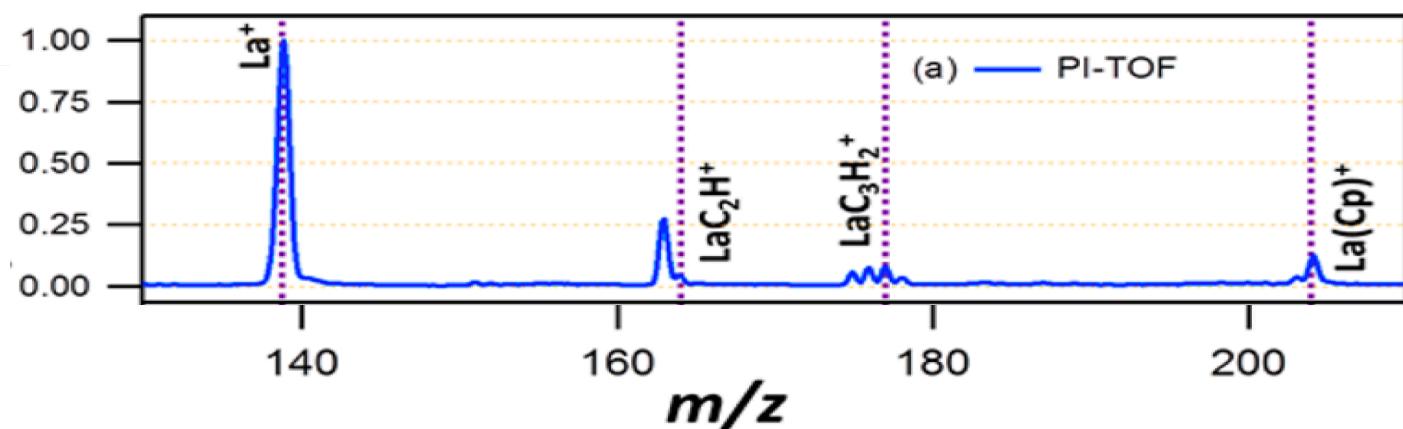
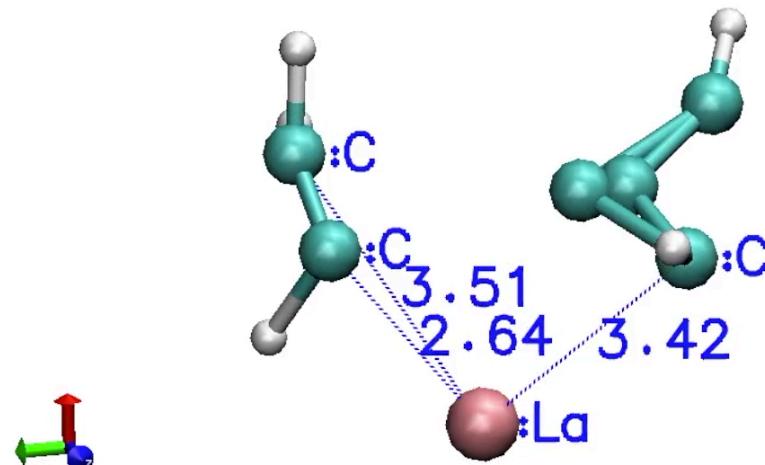


Reference: Han, Meng, Rasulev, May, Berry, Kilin, *J. Chem. Theory Comput.* **2017**, 13, 4281–4296

Research Direction: (iii) Light-induced Reactions

**STAGE 3:  $\text{LaCp} \rightarrow \text{La}^+$**

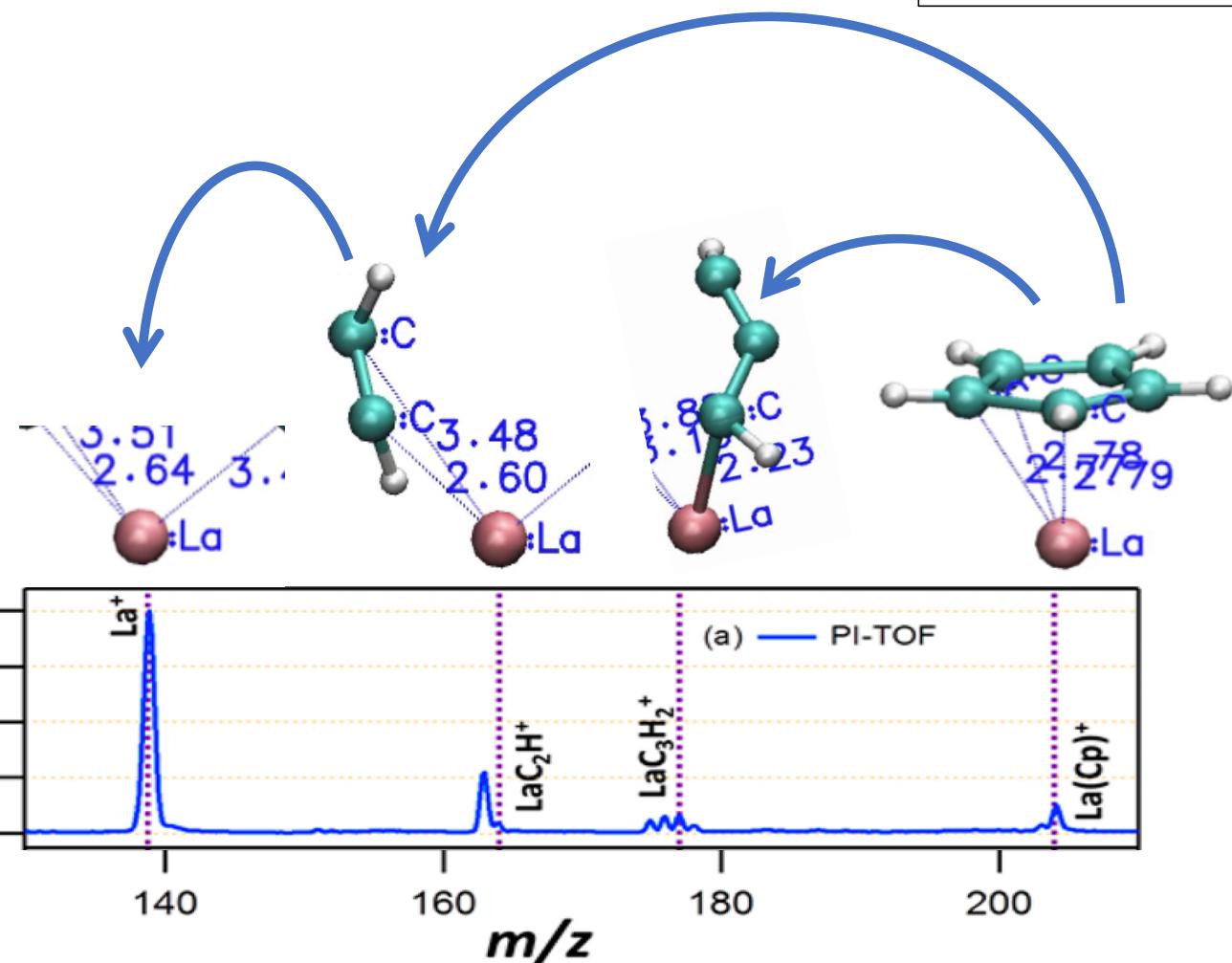
Timing: 0:11



**Reference:** Han, Meng, Rasulev, May, Berry, Kilin, *J. Chem. Theory Comput.* **2017**, 13, 4281–4296

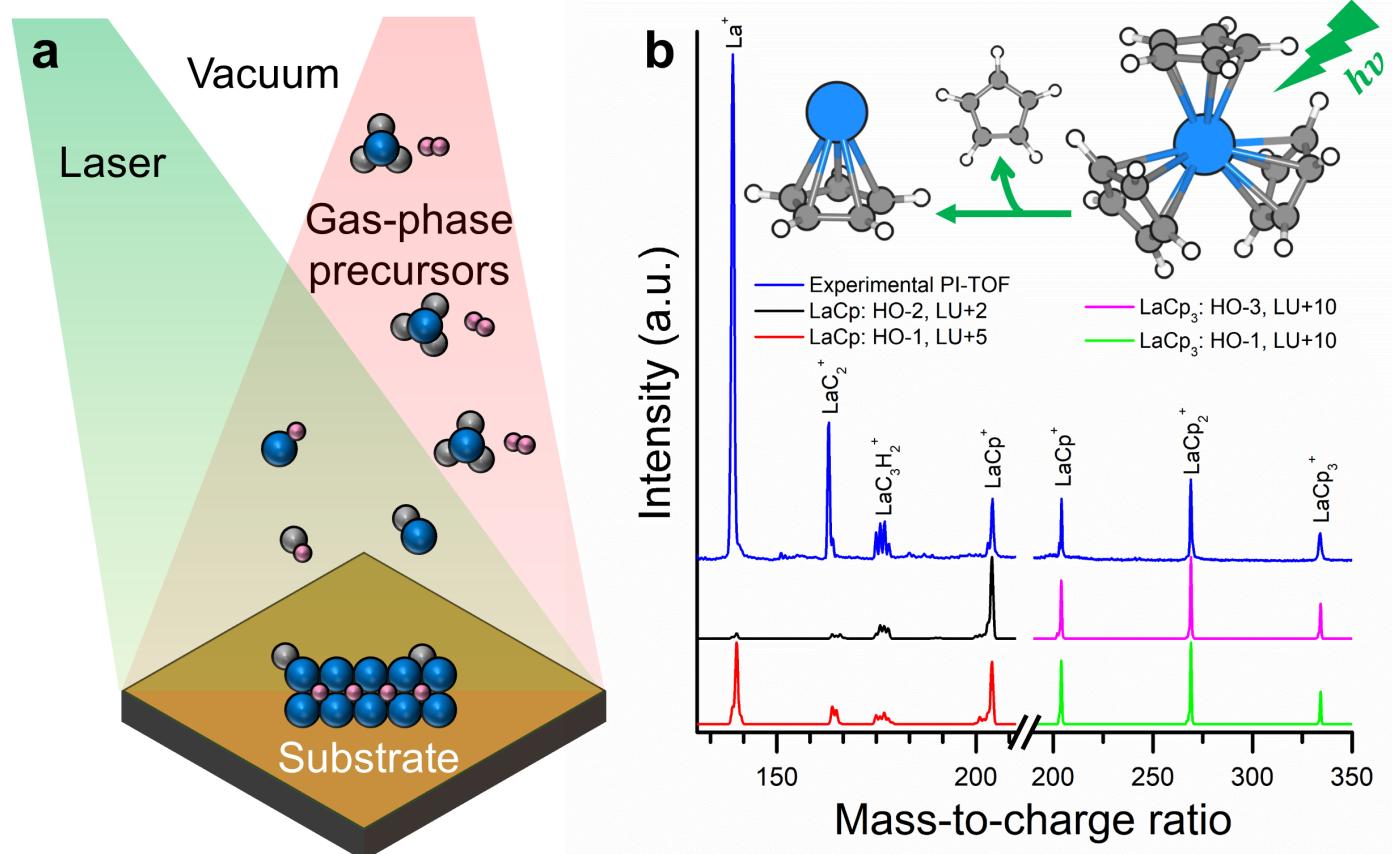
Research Direction: (iii) Light-induced Reactions

Timing: 0:17



Reference: Han, Meng, Rasulev, May, Berry, Kilin, *J. Chem. Theory Comput.* **2017**, 13, 4281–4296

Research Direction: (iii) Light-induced Reactions



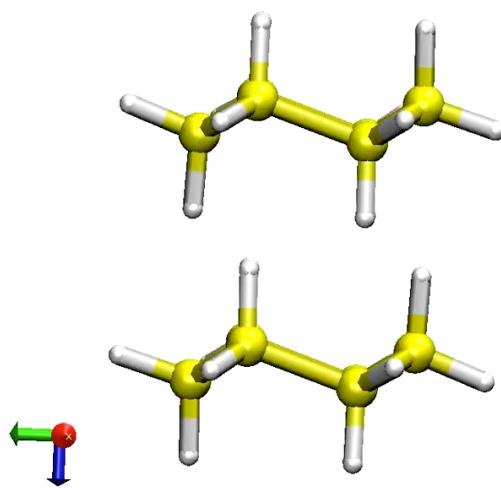
**Justification:** Quantitatively matches experiment: mass and intensity in the mass-spectra

**Reference:** Han, Meng, Rasulev, May, Berry, Kilin, *J. Chem. Theory Comput.* **2017**, 13, 4281–4296

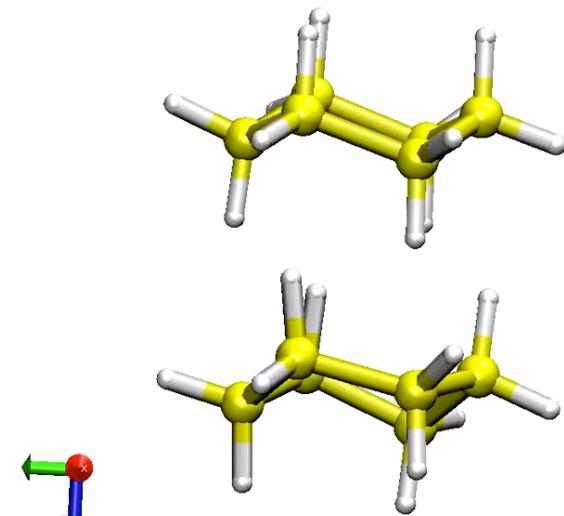
<https://youtu.be/5JmGb5vx-dE>

<https://youtu.be/V-whfiCBRg8>

## Molecular Dynamics

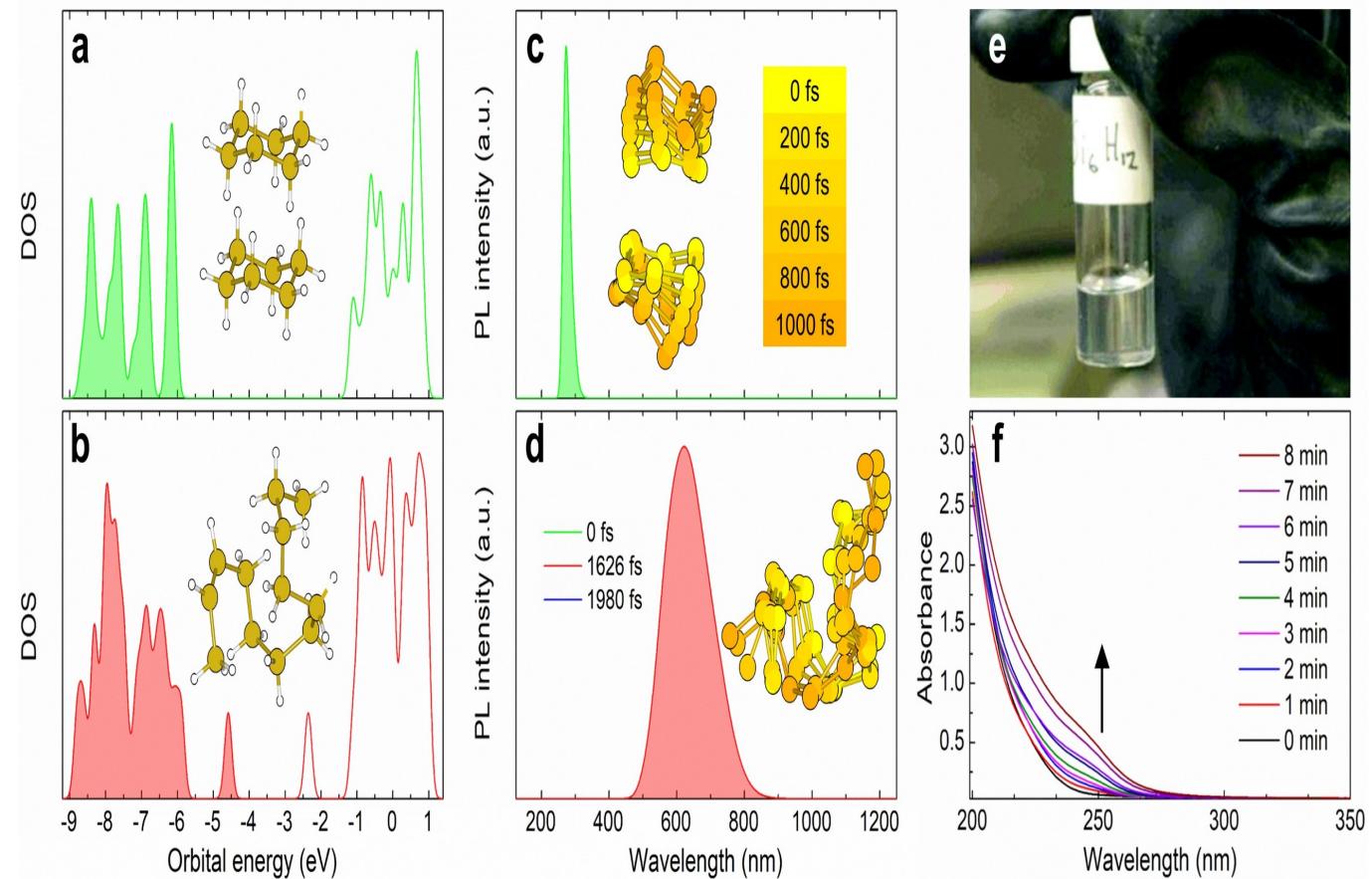


## Optimized Intermediates



**Justification:** matches experimental trends

**Reference:** Han, Anderson, Hobbie, Boudjouk, Kilin, *J. Phys. Chem. Lett.* **2018**, 9, 4349–4354



**Justification:** matches experimental trends

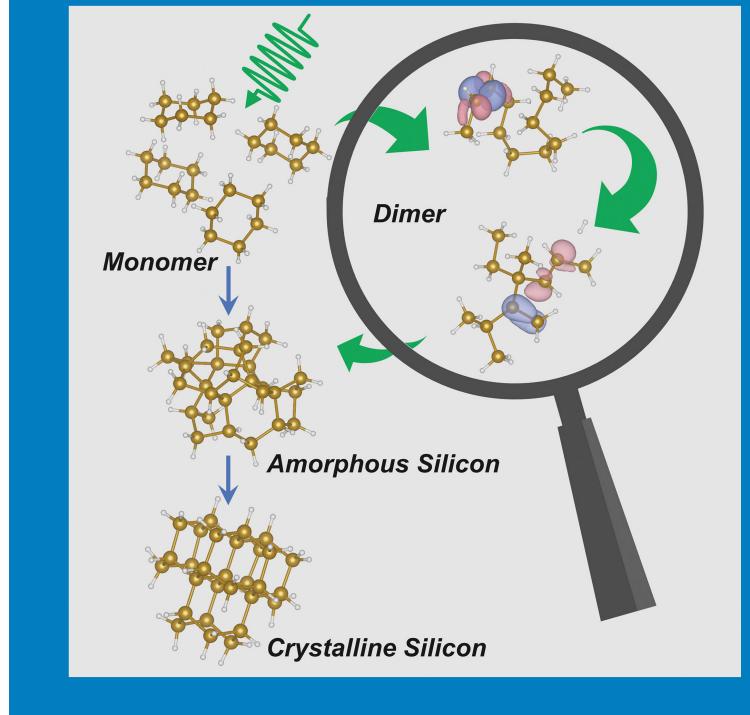
**Reference:** Han, Anderson, Hobbie, Boudjouk, Kilin, *J. Phys. Chem. Lett.* **2018**, 9, 4349–4354

Research Direction: (iii) Light-induced Reactions

Timing: 0:06

THE JOURNAL OF  
PHYSICAL CHEMISTRY  
*L**e**t**t**e**r*s

August 2, 2018 | Volume 9, Number 15



Justification: matches experimental trends

ACS Publications  
Most Trusted. Most Cited. Most Read.

[www.acs.org](http://www.acs.org)

Reference: Han, Anderson, Hobbie, Boudjouk, Kilin, *J. Phys. Chem. Lett.* **2018**, 9, 4349–4354

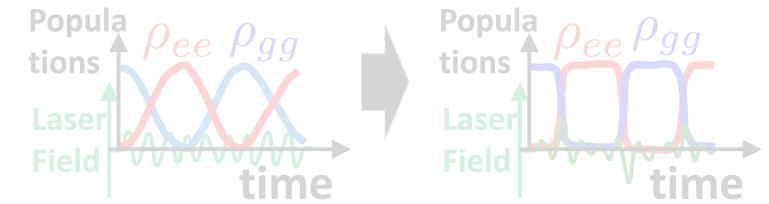
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Timing: 2:09

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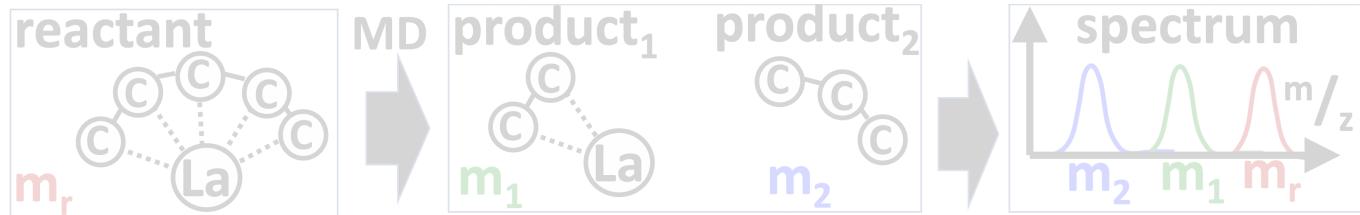
\* replacing continuous change of state' population by instantaneous hops

\* use of "pulse-area theorem"

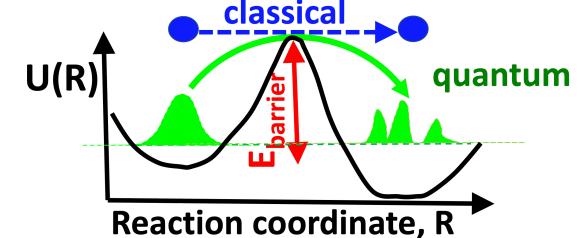


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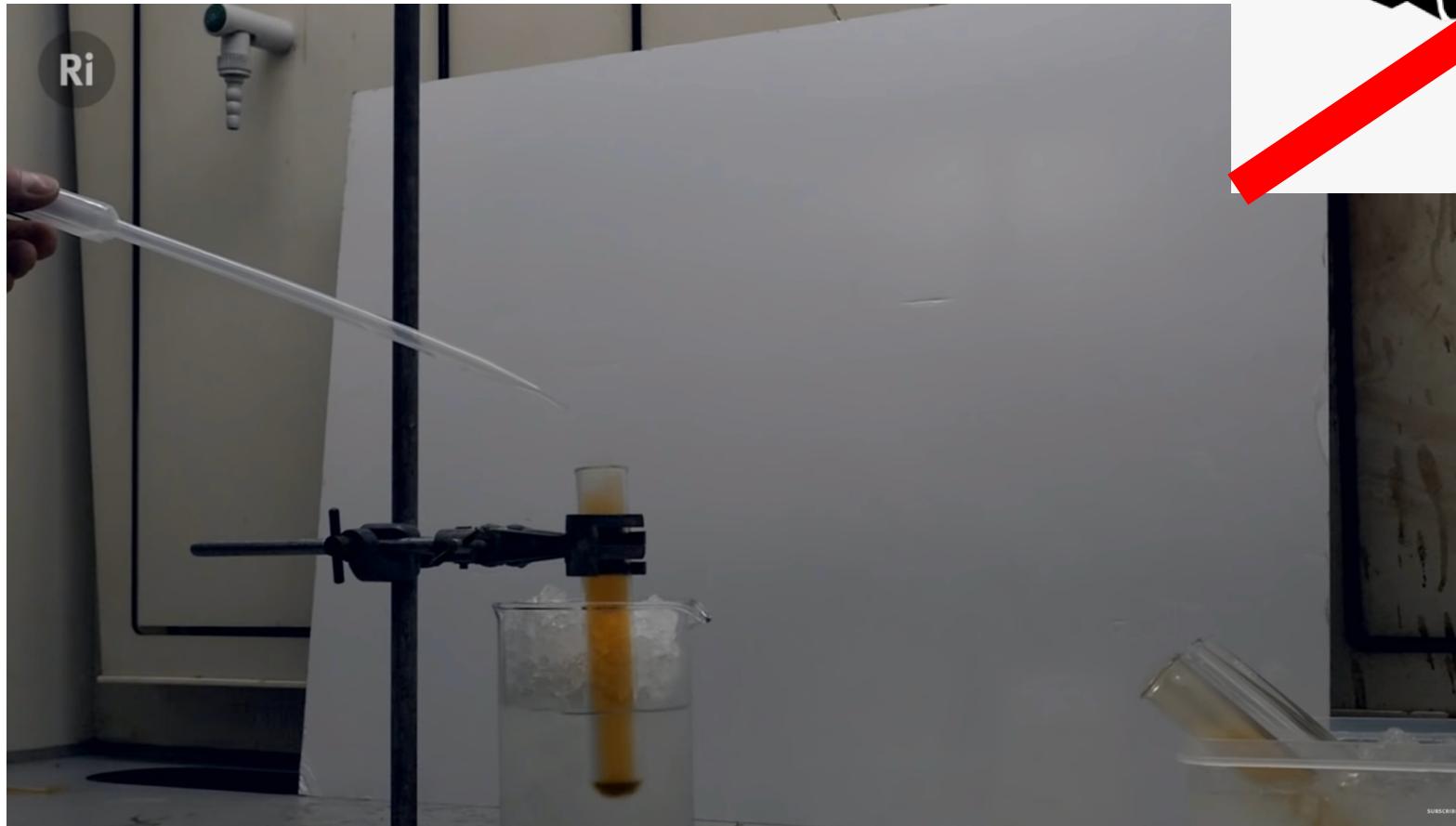


00:00:07;23

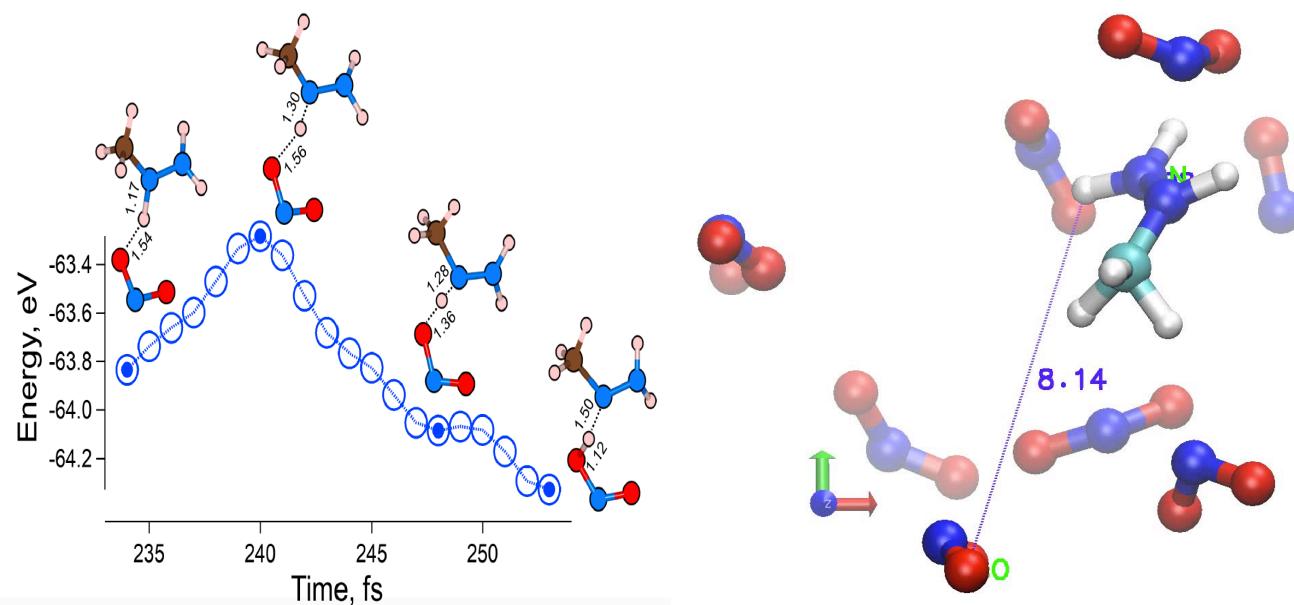


**Multiple ignitions, maneuvering, start and stop with confidence:**

- avoid sparkplugs**
- use igniting-while-mixing fuel and oxidizer**



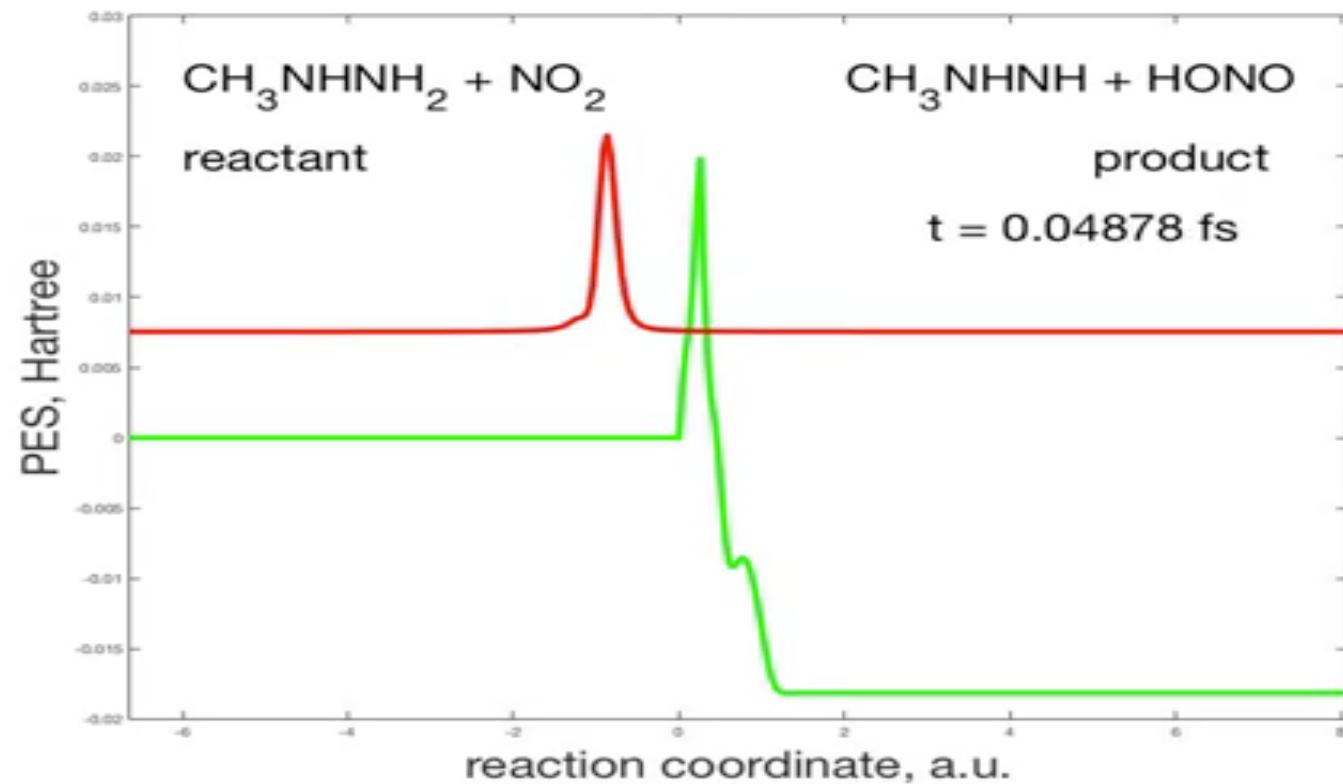
### Role of Quantum Tunneling for Proton Transfer: Case Study of Monomethylhydrazine and Nitrogen Dioxide Initiation



Justification: matches experimental trends

Reference: Han, Hobbie, Kilin *J. Phys. Chem. Lett.* **2019**, 10, 2394–2399

**Role of Quantum Tunneling for Proton Transfer:  
Case Study of Monomethylhydrazine and Nitrogen Dioxide Ignition**



Justification: matches experimental trends

Reference: Han, Hobbie, Kilin *J. Phys. Chem. Lett.* **2019**, 10, 2394–2399

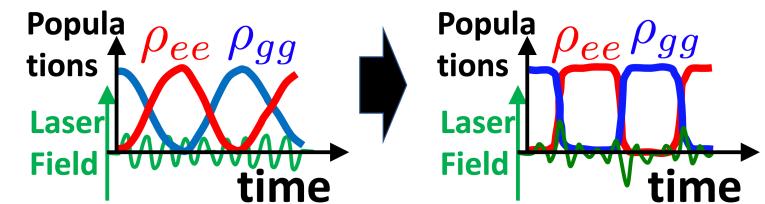
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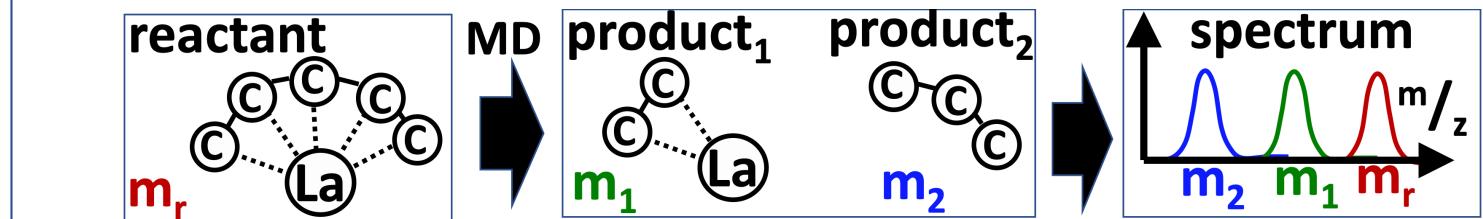
\* replacing continuous change of state' population by instantaneous hops

\* use of "pulse-area theorem"



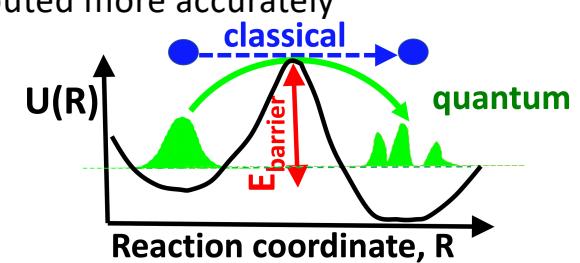
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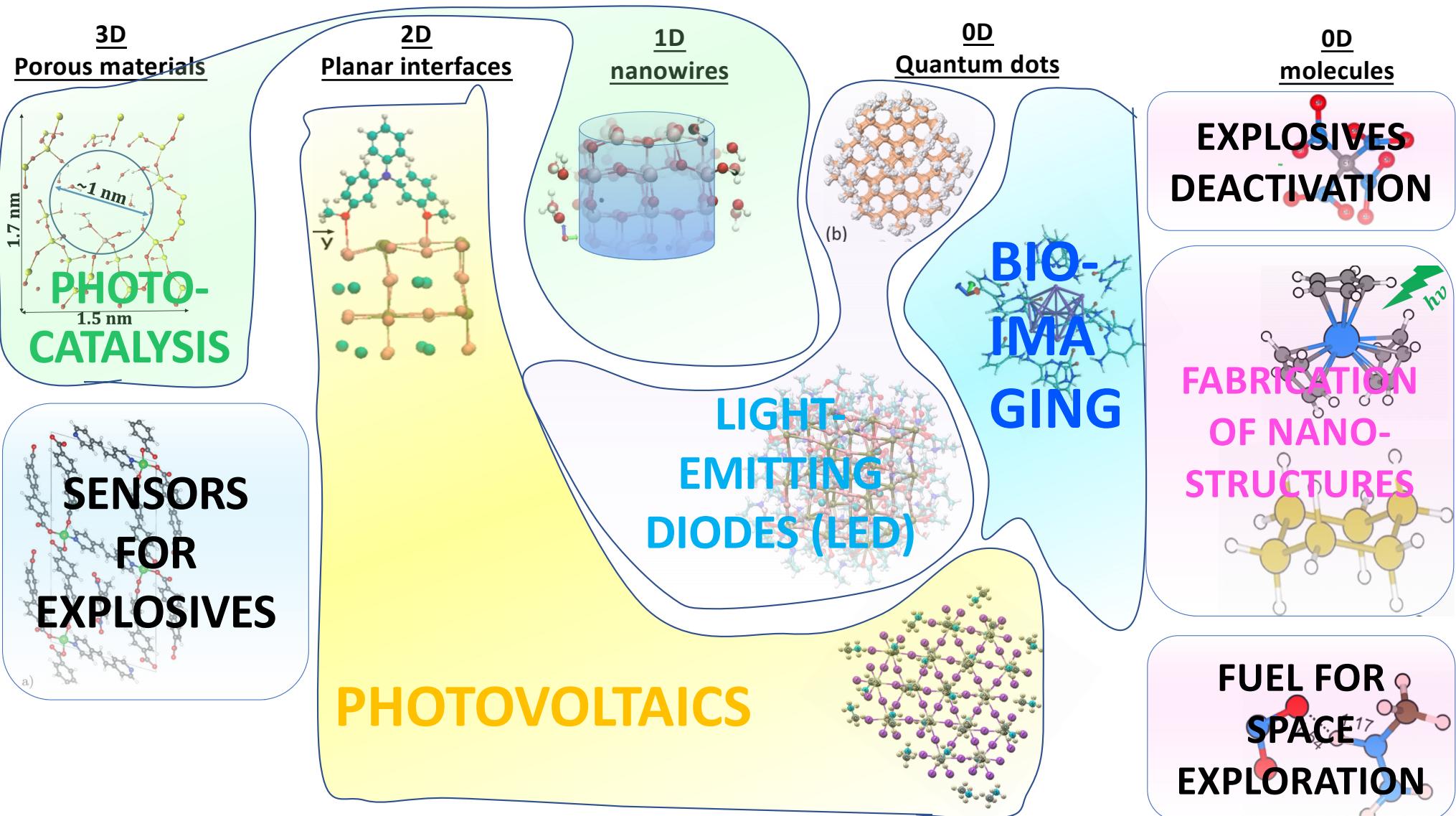
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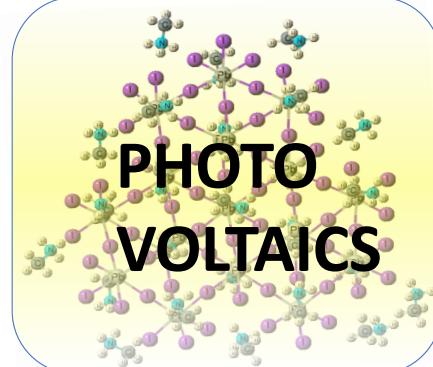
quantum dots



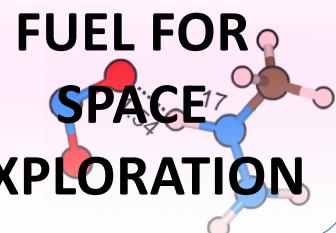
molecules



LIGHT  
EMITTING  
DIODES



FABRICATION  
OF NANO-  
STRUCTURES



## Acknowledgement

(i) Charge transfer

NSF CHE-2004197

*"Modeling of Charge Transfer Processes  
in Heterostructured Nanocomposites"*

ND EPSCOR seed

DOE NERSC facility DE-AC02-05CH11231

*"Computational Modeling of Photo-catalysis and  
Photo-induced Charge Transfer Dynamics on Surfaces"*

(ii) Photoluminescence

NSF CHE-1800476

*"D3SC: Integrated Studies on Designing Organometallic Complexes  
with Nonlinear Absorption and Near-Infrared Emission"*

LANL CINT facility visit

(iii) Photoreactions

NSF CHE-1944921

*"CAREER: Investigation of laser-driven  
chemical reactions by molecular dynamics"*

local DOE CCAST seed

All projects:

Department and College startup



**Special  
Thank you  
to all attendees of this seminar:  
face-to-face  
and online!**

