Chemistry 1141 Syllabus General Chemistry I Fall 2020

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Office hours: MTWR 2:00 – 2:50 P.M.

Lecture: TR 9:30 – 10:50 A.M. (MAS 020)

Lab: T 11:00 – 1:50 P.M. Section 04
R 11:00 – 1:50 P.M. Section 05

Attendance policy: Attendance at lectures is strongly recommended. Attendance at

laboratories is required. Two or more unexcused lab absences will result in a

grade of F for CHEM1141.

Excused absence policy: In case of illness, accident, family emergency, or university-sponsored

activity, you may be excused from labs, quizzes, and/or homework. In case

of a missed exam, a make-up exam will be provided.

For university-sponsored activities, an official excused absence slip must be obtained. This must be obtained in advance of the activity and given to

the instructor one-week before your absence.

For other absences, suitable documentation (such as a doctor's note, police accident report, etc.) must be provided within one-week of the excused absence. For absences longer than one week, an academic dean or the dean of students may issue you an excused absence which you can present to your instructor.

Unexcused absences will result in a grade of zero for the assignment.

Required materials: Chemistry: A Molecular Approach, 5/e

—Nivaldo Tro

Mastering Chemistry

—Bundled with the textbook or a separate access card

Chemistry 1141 Lab Manual, Fall 2020

—Andy Napper

A non-programmable scientific calculator (TI-30XIIS)
Safety goggles or visorgogs (ANSI Z-87 approved)

Final exam: Thursday December 10th, 8 A.M. (Massie 020)

Blackboard course-site:

Notes, handouts, and other useful pieces of information will be available at the following URLs:

http://blackboard.shawnee.edu http://chem1141.ssuchemistry.com

MasteringChemistry:

You should log on and create your account as soon as possible! Online homework will be assigned on a weekly basis. The homework set may consist of tutorials, homework problems, and review problems. Each homework set will be made available on Thursday by 5 P.M., and will be due the following Monday by 12 PM (NOON).

- You will be able to access online assignments on Blackboard. They are located in the "Online Assignments" tab on the left side of the screen. The first time you take an assignment you will be required to log onto your MasteringChemistry account. After doing so, your account will be linked to SSU's BlackBoard site, and you will not have to log in separately.
- The inclusive-access course-fee provides you with access to the textbook and an account with MasteringChemistry at a significant cost savings. The access-code for MasteringChemistry (for students who do not opt out of the inclusive-access plan) is:

PSMTCW-AUXIL-TRONE-WATAP-ABASH-MILES

Lecture material: We will be covering the following chapters in your textbook:

Chapter 1 Matter, Measurement, and Problem Solving

Chapter 2 Atoms and ElementsChapter 3 Molecules & Compounds

EXAM 1 (Thursday, September 24)

Chapter 4 Chemical Reactions & Chemical QuantitiesChapter 5 Introduction to Solutions & Aqueous Reactions

EXAM 2 (Thursday, October 22)

Chapter 6 Gases

Chapter 7 Thermochemistry

Chapter 8 The Quantum-Mechanical Model of the Atom

EXAM 3 (Thursday, November 12)

Chapter 9 Periodic Properties of the ElementsChapter 10 Chemical Bonding I: The Lewis Model

Chapter 11 Chemical Bonding II: Molecular Shapes, Valence Bond Theory, and

Molecular Orbital Theory EXAM 4 (<u>Tuesday</u>, December 1)

Exams 1 – 4:

Exams I - 4 will be held in Library 204 (Flohr) from 5 P.M. - 7 P.M. These will be joint exams that all sections of General Chemistry I will take together. A make-up exam will be available for students who have a time conflict or an official University excused absence.

University ADA statement:

Any student who believes s/he may need an academic accommodation based on the impact of a documented disability should first contact a Coordinator in the Office of Accessibility Services, Hatcher Hall, 740-351-3106 to schedule a meeting to identify potential reasonable academic accommodation(s). Students are strongly encouraged to initiate the academic accommodation process in the early part of the semester or as soon as the need is recognized. After meeting with the Coordinator, students are encouraged to meet with their instructors during the instructor's office hours to discuss their specific needs related to their disability. The academic accommodation letter will be sent to the instructor and student via secure e-mail prior to the semester start date. Any questions regarding the academic accommodations on the letter should be addressed to the Coordinator of Accessibility Services. If a student does not make a timely request for academic accommodations and/or fails to meet with the Coordinator of Accessibility Services, a reasonable academic accommodation might not be able to be provided.

General education program:

Chemistry 1141 counts towards the Natural Science component of the General Education Program (GEP) and addresses *Scientific Reasoning*.

Laboratory information:

Safety goggles or visorgogs are required to be worn for all laboratories. They must meet ANSI Z87 requirements (normally this information is permanently stamped on the goggles). Laboratory coats are recommended, but not required. Full length pants or full-length skirts are required to be worn in lab. Shoes that cover all parts of your feet are also required. If you are improperly dressed for lab, you will be asked to leave and awarded a zero for the lab assignment.

Lab reports must be turned in *at the end* of each week's lab. Reports must have your full name clearly written on the front page to receive a grade.

Order of labs:

Week	Tue (section 04)		Thu (section 05)	
Beginning				
	Blue	Gray	Blue	Gray
August 24	I	2	I	2
August 31	2	I	2	I
September 7	3	4	3	4
September 14	4	3	4	3
September 21	5	6	5	6
September 28	6	5	6	5
October 5	7	8	7	8
October 12	8	7	8	7
October 19	9	10	9	10
October 26	10	9	10	9
November 2	ΙΙ	I 2	ΙΙ	I 2
November 9	I 2	11	I 2	ΙΙ
November 16	13	13	13	13
November 23	No lab		Thanksgiving break	
November 30	No lab		Final exam week	

Laboratories:

- 1. Check-in, safety, and conversions
- 2. LABSTER: Matter and phase changes
- 3. Accuracy and precision
- 4. LABSTER: Atomic structure (principles)
- 5. Determining the formula of a hydrate
- 6. LABSTER: Stoichiometric calculations
- 7. Double-replacement reactions
- 8. LABSTER: Titration: neutralize an acid lake contamination
- 9. Titrations (handout)
- 10. LABSTER: Periodic table of elements: Get table organized in time!
- 11. Determining the enthalpy of neutralization
- 12. **LABSTER**: Ionic and covalent bonds
- 13. VIRTUAL LAB: TBD

Labster: These virtual labs are available in BlackBoard and should be completed by the 11:59 pm on the Sunday after your assigned lab date. For example: If your regular lab meets on Thursday, November 12 and on this day you are assigned the Labster virtual lab, "Ionic and covalent bonds," you will need to complete the virtual lab exercise on BlackBoard by 11:59 pm on Sunday, November 15.

- Note: These virtual labs may be taken any number of times, however the grade will be based upon your first attempt!
- You will need to access these labs on a desktop or laptop computer using the most recent version of either the *Chrome* or *Firefox* web browser.

Grading scale:

% Grade	% Grade	% Grade	% Grade	% Grade
>93 A	87 – 90 B+	77 - 80 C+	67 - 70 D+	<60 F
90 – 93 A–	83 – 87 B	73 - 77 C	63 - 67 D	
	80 – 83 B–	70 – 73 C–	60 – 63 D–	

Note: To receive a grade of C – or higher, you must *either*

- pass (>59.5 %) at least three of the four semester exams, or
- score greater than a 30th percentile on the ACS standardized final If you fail to meet both requirements, the maximum issued grade will be a D+.

Grading errors:

If you notice a grade error on BlackBoard for quizzes, exams, etc.—you need to bring it to the instructor's attention in writing within one week of the due date (for an online assignment) or one week from the assignment being handed back (lab/exam assignments).

Who should take this course?

The typical audience for this course is: science, engineering, pre-pharmacy, pre-medicine, and science education majors. You may also be taking this course if you are interested in chemistry (yay!), are seeking to satisfy the natural sciences general education category, or curious about how things work.

Is chemistry hard?

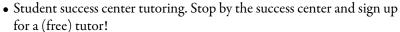
Yes. But not impossible. Consider setting aside several hours a week to practice end-of-chapter homework problems, forming a study group, rereading your Mastering Chemistry assignments, reading the textbook, and quizzing yourself. Reviewing old material every few weeks has been shown to dramatically improve retention of material in college!

Do I have to attend every lecture?

Not attending lecture tends to correlate with doing poorly in chemistry. Do not get into the habit of missing class if you can possibly avoid it. At mid-term I will be reporting your expected grade and attendance to the Registrar. This information will be provided to your academic advisor(s), as well as student support services.

What should I do if I need help?

If you need help—don't wait too long before you seek it out! The following is a partial list of options that are available to you:



- Browse my course website for chapter objectives, old exams, lecture notes, quizzes, etc.
- YouTube. Amazing selection of videos on any topic you can think about. The *Khan Academy* videos are an excellent place to start.
- Office hours. I hold four office hours a week over three days. Stop by if you have any questions about the course!



An ion with a paws-itive charge.
 The cutest ion ever.

low to study for this class	Buy a composition notebook to work problems in.				
	☐ <i>Skim</i> through the textbook section before you come to each class				
	☐ After each class, but before the next class, go through the Example				
	problems in the chapter. Do the "For Practice" problems after each				
	example. The answers are in Appendix IV.				
	☐ In a separate notebook, answer the problems at the end of each chapter				
	that go over the relevant sections. The answers to the odd-numbered				
	problems are in Appendix III.				
	At the end of each chapter:				
	☐ Take the self-assessment quiz. The answers are printed at the bottom of the assignment (upside down).				
	☐ In your composition notebook, work the Cumulative problems as well as any Challenge problems that are assigned.				
	☐ One week before each exam, thoroughly read your notes, being sure to work out any problems yourself that we went over. Try covering up my				
	worked answers with a blank piece of paper and then working them yourself.				
	□ Re-work the end-of-chapter and in-chapter problems				
	☐ Print off a practice exam and take it in a timed fashion. Print off the answers and then grade yourself.				
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Hint: 90 % of your studying in general chemistry should consist of working problems!

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End-of-chapter problems
                                (Bolded and underlined problems are cumulative/challenge)
                 Chapter 1
                                37, 43, 45, 59, 69, 73, 77, 87, 101, 123, 129, 133, 141
                  Chapter 2
                                47, 51, 53, 61, 67, 71, 81, 87, 97, 105, 117, 125
                  Chapter 3
                                29, 35, 37, 45, 47, 49, 61, 65, 73, 83, 89, 109, 115, 121, 125, 141
                 Chapter 4
                                19, 23, 33, 43, 49, <u>67, 73</u>
                 Chapter 5
                                21, 25, 31, 33, 37, 43, 47, 53, 57, 59, 63, 65, 73, 81, 89
                                31, 33, 37, 45, 55, 59, 65, 75, 81, 89,
                 Chapter 6
                                94 (p_{\text{ideal}} = 4.480 \text{ atm}, p_{\text{vdW}} = 4.272 \text{ atm}), 95, 103, 119, 127, 137
                 Chapter 7
                                49, 59, 63, 67, 75, 79, 81, 85, 91, 99, 105, 113, 121
                 Chapter 8
                                37, 39, 43, 45, 49, 59, 61, 67, 69, 73, 81, 95
                 Chapter 9
                                39, 47, 49, 55, 57, 59, 63, 65, 69, 71, 75, 85, 87, 93, 101, 111, 113
                Chapter 10
                                39, 53, 55, 63, 65, 73, 77, 81, 83, 91, 95, 97, 113, 119
                Chapter 11
                                31, 35, 39, 47, 51, 53, 55, 57, 59, 61, 65, 83, 87, 97, 103, 107
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Disclaimer: All dates and policies are subject to change as announced in class.