## DEOXYURIDINE METABOLISM: URACIL PRODUCTION SPEED

## INTRODUCTION

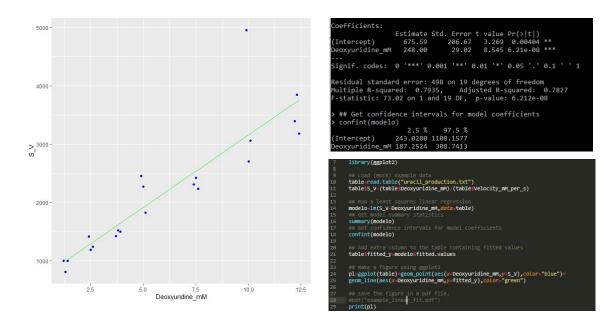
In an experiment, we constructed samples with differing concentrations of deoxyuridine in addition to a constant amount of enzyme ( $E_0=0.1~\mu\text{M}$ ) and the appropriate buffers. After a set period, we stopped the reaction (with sodium hydroxide in this case) and measured the amount of uracil produced via spectrophotometry.

## RESULTS AND CONCLUSIONS

Applying the Michaelis-Menten model, and, using linear models in R, inference of the Michaelis-Menten constant  $K_m$  for the system, as well as the maximum uracil production  $V_{\rm max}$ .

The starting point of the procedure is the Michaelis-Menten equation:

$$\frac{S}{V(uracil)} = \frac{K_m}{Vmax} + \frac{S}{Vmax}$$



$$Slope = 248.0 \, s^{-1} \rightarrow Vmax = Slope^{-1} = 0,00403225 \, s$$
 
$$indep. \, term = 675.6 \, s^{-1} \rightarrow K_M = i. \, term * Vmax = 0.00403225 * 675.6 = 2.72 \, mM$$

Value of the catalysis constant kcat:

$$Vmax = k_{cat} * E_0 \rightarrow k_{cat} = \frac{0,00403225 \, s}{0,0001 \, M} = 40,32 \, s/mM$$

For the case in which each molecule of complex has exactly the same probability of disintegrating originating a molecule of product or doing so giving back one molecule of substrate, values of k<sub>f</sub> and k<sub>r</sub>.

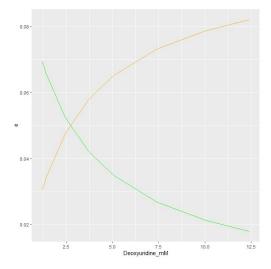
We assume  $k_r=k_f$ , then if:

$$K_{M} = \frac{(k_{r} + k_{cat})}{k_{f}} \rightarrow K_{M} = \frac{(k_{f} + k_{cat})}{k_{f}} \rightarrow k_{f} = \frac{k_{cat}}{K_{M} + 1} = \frac{40,32 \frac{s}{M}}{2.72M + 1} = 10.84 \text{ mM}$$

$$= k_{r}$$

Graphical representation of the predicted concentrations of complex and enzyme as a function of the remaining substrate: C(t) and E(t) in the range of deoxyuridine concentrations present in the input data.

$$E(t) = \frac{K_M E_0}{(K_M + S)}; C(t) = (E_0 * S)/(K_M + S)$$



```
## toad (mock) example data

table=read.table("uracil_production.txt")

tables_v=(table=Deoxyuridine_mM)/(table=Velocity_mM_per_s)

## Nun a least squares linear regression

modelo=lm(S_V-Deoxyuridine_mM,data=table)

## Steet model summary statistics

summary(modelo)

## Add extra column to the table containing fitted values

confint(modelo)

## Add extra column to the table containing fitted values

## amake a figure using geplot2

pl=geplot(table) egeom_point(asc(x=Deoxyuridine_mM,y=S_V),color="blue")+

geom_line(asc(x=Deoxyuridine_mM,y=fitted_y),color="green")

## save the figure in a pdf file.

## pdf("example_linear_fit.pdf")

print(pl)

## data=rbint(d,e,c)

print(table)

## pl=geplot(table)-

## peom_line(asc(x=Deoxyuridine_mM+0.1)/(table=Deoxyuridine_mM+2.720)

## data=rbint(d,e,c)

print(table)

## print(pl2)
```

Algebraic representation of each of them:

$$N1 = \begin{pmatrix} -1 \\ -1 \\ 1 \\ 2 \end{pmatrix}$$

$$N2 = \begin{pmatrix} 1 & -1 & 0 & 0 & 0 \\ 0 & 1 & -1 & 0 & 0 \\ 0 & 0 & 1 & -1 & 0 \\ 0 & 0 & 0 & 1 & -1 \end{pmatrix}$$

$$N3 = \begin{pmatrix} 1 & -1 & 0 & -1 \\ 0 & 2 & -1 & 0 \\ 0 & 0 & 0 & 1 \end{pmatrix}$$

$$N4 = \begin{pmatrix} 1 & -1 & 0 & -1 \\ 0 & 2 & -1 & 0 \\ 0 & 0 & 0 & 1 \end{pmatrix}$$

$$N5 = \begin{pmatrix} 1 & 0 & 0 \\ 0 & -1 & 1 \\ 0 & 1 & -1 \end{pmatrix}$$

System 1:

$$\frac{dS1}{dt} = -V1; \frac{dS2}{dt} = -V1; \frac{dS3}{dt} = V1; \frac{dS4}{dt} = 2V1$$

System 2:

$$\frac{dS1}{dt} = V1 - V2; \frac{dS2}{dt} = V3 - V2; \frac{dS3}{dt} = V3 - V4; \frac{dS4}{dt} = V4 - V5$$

System 3:

$$\frac{dS1}{dt} = V1 - V2 - V3$$

System 4:

$$\frac{dS1}{dt} = V1 - V2 - V4; \frac{dS2}{dt} = 2V3 - V2; \frac{dS3}{dt} = V4$$

System 5:

$$\frac{dS1}{dt} = V1; \frac{dS2}{dt} = V3 - V2; \frac{dS3}{dt} = V2 - V3$$

Kernel solutions

System 1:

$$0 = -V1$$
;  $0 = -V1$ ;  $0 = V1$ ;  $0 = 2V1$ 

System 2:

$$0 = V1 - V2$$
;  $0 = V3 - V2$ ;  $0 = V3 - V4$ ;  $0 = V4 - V5$   
 $0 = V1 = V2 = V3 = V4 = V5$ 

System 3:

$$0 = V1 - V2 - V3$$

System 4:

$$0 = V1 - V2 - V4$$
;  $0 = 2V3 - V2$ ;  $0 = V4$   
 $V1 = V2$ ;  $V3 = V2/2$ ;  $0 = V4$ 

System 5:

$$0 = V1; \ 0 = V3 - V2; \ 0 = V2 - V3$$
  
 $0 = V1; V3 = V2$ 

The configuration, for all the velocities, associated to the optimization of the objective function.

```
> optimum$objval
[1] 8
> optimum$solution
[1] 32 40 16 8 32 8 8 32 8 8
```

The output indicates the respective optimal velocities for maximum V10 which is  $8\,$  mM/s.