ESO 201A: Thermodynamics 2016-2017-I semester

Gas Power Cycle: part 4

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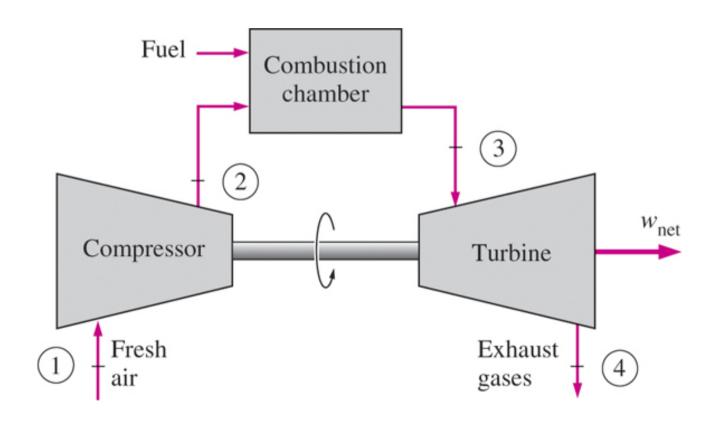
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Learning Objectives

- Evaluate the performance of gas power cycles for which the working fluid remains a gas throughout the entire cycle.
- Develop simplifying assumptions applicable to gas power cycles.
- Review the operation of reciprocating engines.
- Analyze both closed and open gas power cycles.
- Solve problems based on the Otto, Diesel, and Brayton cycles.

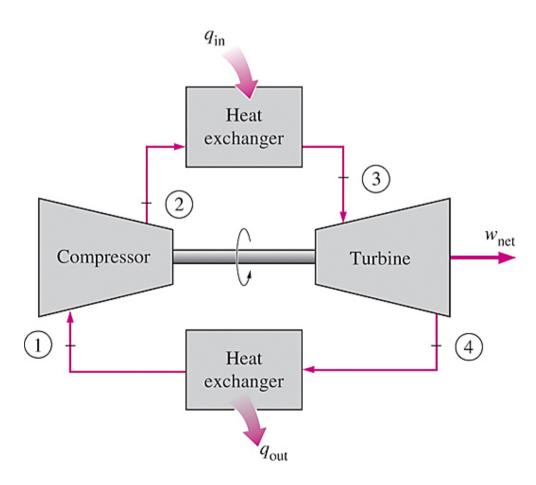
Ideal Brayton Cycle

- In reality, gas turbines operate on an open cycle
- Fresh air is continuously drawn into the compressor and exhaust gases are thrown out



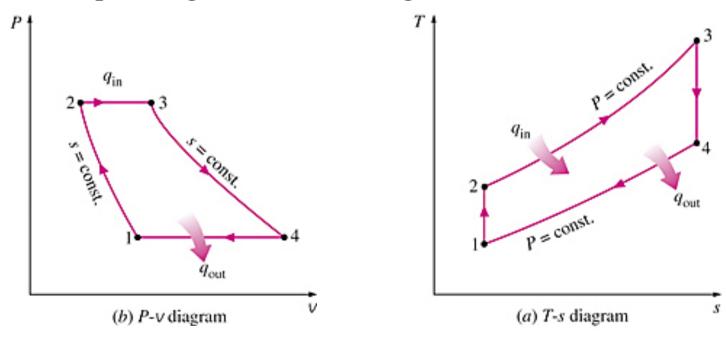
Ideal Brayton Cycle (cont.)

- The open gas-turbine cycle can be modeled as a closed cycle
- The combustion process is replaced by a constant-pressure heataddition process and the exhaust process is replaced by a constantpressure heat-rejection process



Ideal Brayton Cycle (cont.)

- The idealized closed loop cycle is the Brayton cycle, which consists of the following four internally reversible processes
 - $1\rightarrow 2$ Isentropic compression
 - $2\rightarrow 3$ Constant-pressure heat addition
 - 3→4 Isentropic expansion
 - 4→1 Constant-pressure heat rejection
- The corresponding P-v and T-s diagrams are shown below



Thermodynamic Analysis

- The four processes of the Brayton cycle are executed in steadyflow devices
- When changes in kinetic and potential energies are neglected, the energy balance for one of the processes can be expressed as

$$(q_{\text{in}} - q_{\text{out}}) + (w_{\text{in}} - w_{\text{out}}) = h_{\text{exit}} - h_{\text{inlet}}$$

• Therefore, heat transfers to and from the working fluid are

$$q_{\rm in} = h_3 - h_2 = c_p (T_3 - T_2)$$

$$q_{\text{out}} = h_4 - h_1 = c_p (T_4 - T_1)$$

Thermal Efficiency

• The thermal efficiency of the ideal Brayton cycle under the coldair-standard assumptions becomes

$$\eta_{\text{th, Brayton}} = \frac{w_{\text{net}}}{q_{\text{in}}} = 1 - \frac{q_{\text{out}}}{q_{\text{in}}}$$

$$= 1 - \frac{c_p (T_4 - T_1)}{c_p (T_3 - T_2)}$$

$$= 1 - \frac{T_1 (T_4 / T_1 - 1)}{T_2 (T_3 / T_2 - 1)}$$

• Processes $1 \rightarrow 2$ and $3 \rightarrow 4$ are isentropic, and $P_2 = P_3$ and $P_4 = P_1$

$$\frac{T_2}{T_1} = \left(\frac{P_2}{P_1}\right)^{(k-1)/k} = \left(\frac{P_3}{P_4}\right)^{(k-1)/k} = \frac{T_3}{T_4}$$

Thermal Efficiency (cont.)

• Substituting these expressions into the thermal efficiency relation yields

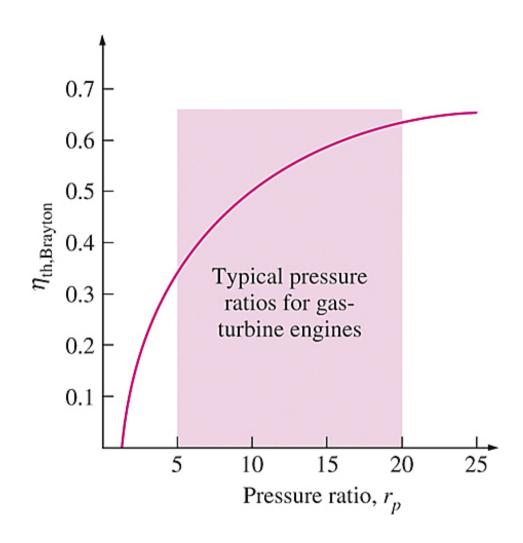
$$\eta_{\text{th, Brayton}} = 1 - \frac{1}{r_P^{(k-1)/k}}$$

• Where r_P is the pressure ratio

$$r_P = \frac{P_2}{P_1}$$

Thermal Efficiency (cont.)

- The thermal efficiency increases with both the pressure ratio (r_P) and the specific heat ratio (k)
- The plot to the right shows the thermal efficiency as a function of the compression ratio
- The two major application areas of gas-turbine engines are aircraft propulsion and electric power generation

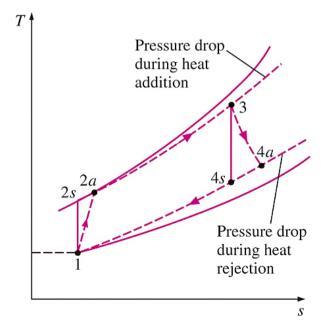


Deviation of Actual Gas-Turbine Cycles from Idealized Ones

• The deviation of actual compressor and turbine behavior, due to irreversibility, can be accurately accounted for by utilizing the isentropic efficiencies of the turbine and compressor

$$\eta_C = \frac{w_s}{w_a} \cong \frac{h_{2s} - h_1}{h_{2a} - h_1}$$
 and $\eta_T = \frac{w_a}{w_s} \cong \frac{h_3 - h_{4a}}{h_3 - h_{4s}}$

• The actual and isentropic states of a gas-turbine cycle are illustrated below



Problem

Air is used as the working fluid in a simple ideal Brayton cycle that has a pressure ratio of 12, a compressor inlet temperature of 300 K, and a turbine inlet temperature of 1000 K. Determine the required mass flow rate of air for a net power output of 70 MW, assuming both the compressor and the turbine have an isentropic efficiency of (a) 100 percent and (b) 85 percent. Assume constant specific heats at room temperature.

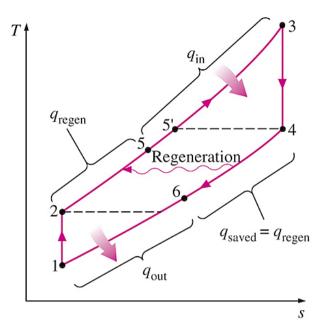
Properties The properties of air at room temperature are $c_p = 1.005 \text{ kJ/kg} \cdot \text{K}$ and k = 1.4 (Table A-2).

THE BRAYTON CYCLE WITH REGENERATION

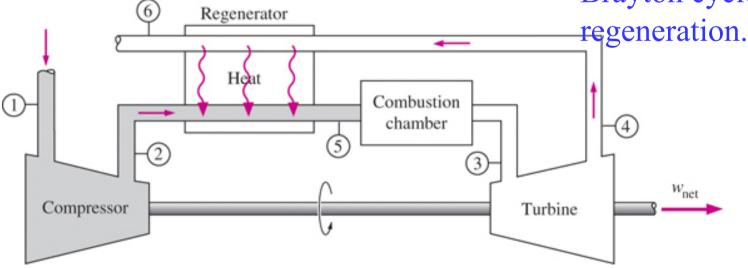
In gas-turbine engines, the temperature of the exhaust gas leaving the turbine is often considerably higher than the temperature of the air leaving the compressor.

Therefore, the high-pressure air leaving the compressor can be heated by the hot exhaust gases in a counter-flow heat exchanger (a *regenerator* or a *recuperator*).

The thermal efficiency of the Brayton cycle increases as a result of regeneration since less fuel is used for the same work output.



T-s diagram of a Brayton cycle with regeneration.



A gas-turbine engine with regenerator.

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Summary

- Basic considerations in the analysis of power cycles
- The Carnot cycle and its value in engineering
- Air-standard asssumptions
- An overview of reciprocating engines
- Otto cycle: The ideal cycle for spark-ignition engines
- Diesel cycle: The ideal cycle for compression-ignition engines
- Brayton cycle: The ideal cycle for gas-turbine engines
- The Brayton cycle with regeneration
- The Brayton cycle with intercooling, reheating, and regeneration