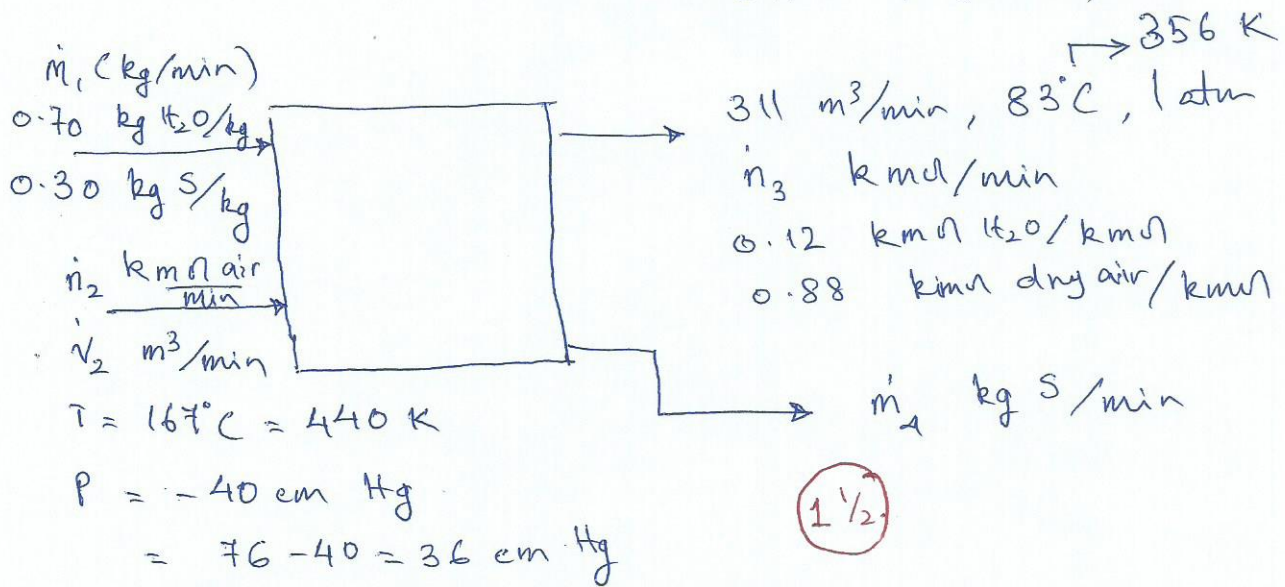


Quiz II (6 marks total)

S = solid (milk)



(or 1033 - 40 cm H₂O)
Physical laws (Ideal gas assumption)

$$\dot{n}_3 = 311 \frac{\text{m}^3}{\text{min}} \times \frac{1 \text{ atm}}{356 \text{ K}} \times \frac{\text{kmol} \cdot \text{K}}{0.08206 \text{ m}^3 \cdot \text{atm}} = 10.65 \frac{\text{kmol}}{\text{min}}$$

$$\dot{n}_3 = 10.65 \frac{\text{kmol}}{\text{min}} \quad (1)$$

H₂O balance: $0.70 \times \dot{m}_1 \times \frac{\text{kg H}_2\text{O}}{\text{kg}} \times \frac{\text{kg}}{\text{min}} = \dot{n}_3 \times 0.12 \times 18 \times \frac{\text{kmol}}{\text{min}} \times \frac{\text{kg H}_2\text{O}}{\text{kmol}} + \frac{\text{kg}}{\text{kmol}}$

$$\dot{m}_1 = 32.86 \frac{\text{kg}}{\text{min}} \quad (1)$$

Solid balance: $0.30 \times \dot{m}_1 = \dot{m}_4$

$$\dot{m}_4 = 9.85 \frac{\text{kg S}}{\text{min}} \quad (1/2)$$

Dry air balance: $\dot{n}_2 = 0.88 \times \dot{n}_3 \Rightarrow \dot{n}_2 = 9.37 \frac{\text{kmol}}{\text{min}} \quad (1/2)$

$$\dot{V}_2 = 9.37 \frac{\text{kmol}}{\text{min}} \times 0.08206 \frac{\text{m}^3 \cdot \text{atm}}{\text{kmol} \cdot \text{K}} \times \frac{440 \text{ K}}{36 \text{ cm Hg}} \times \frac{76 \text{ cm Hg}}{1 \text{ atm}}$$

$$\dot{V}_2 = 714.2 \text{ m}^3/\text{min} \quad (1)$$

Velocity Diameter not given. Assume it to be Δm

$$u_{\text{air}} = \frac{\dot{V}_2}{(\pi/4) D^2} = \frac{909 \text{ m}}{D^2 \text{ min}} = \frac{15.1 \text{ m}}{D^2 \text{ s}} \quad (1/2)$$

If velocity is too high then the powdered milk will fluidize and might be blown out of the reactor dryer.