

# Metals – I

## (Ferrous alloys)

**Instructor :** Prof. Bishakh Bhattacharya  
Dept. of Mechanical Engineering  
IIT Kanpur  
India

**E-mail** : *bishakh@iitk.ac.in*



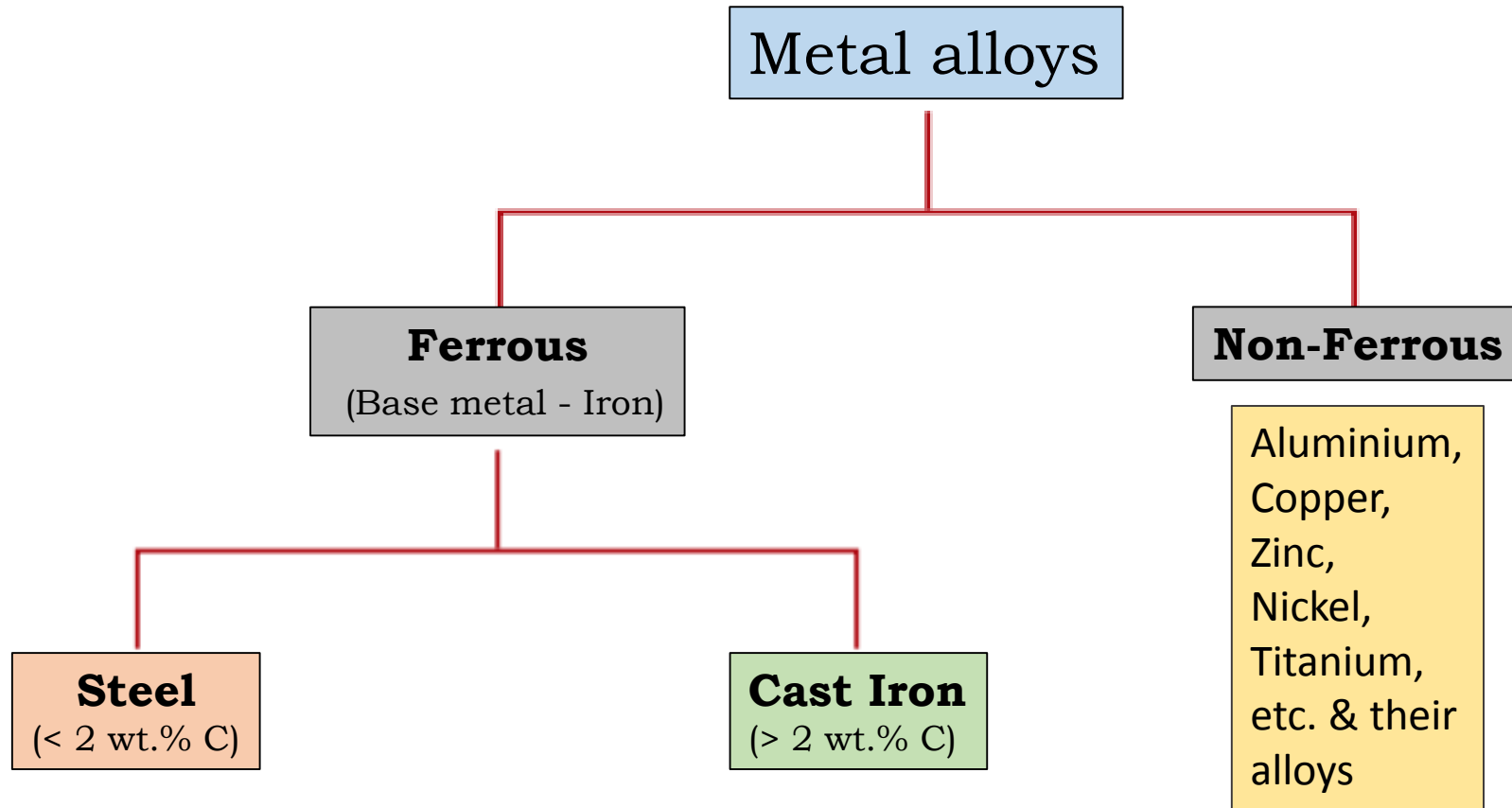
Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur

# Content

- ✓ Classification of Ferrous alloys
- ✓ Types of steel
- ✓ Effect of impurities
- ✓ Cast Iron

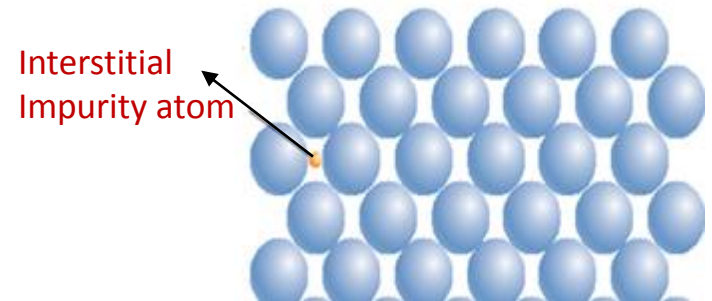


# Classification of Metal Alloys



# Iron

- The Iron Age began about 3000 years ago and **continues till today**.
- Carbon forms an **interstitial solid solution** when added to **iron** to form **Steel** as the **atomic radius** of the carbon (0.071 nm) atom is **much less** than that for iron (0.124 nm).
- Use of iron and steel has changed drastically the human development.
- Iron possesses **allotropy** - exist in two or more **different forms** in the **same** physical **state**.
  - ✓  $T < 770^{\circ}\text{C}$  : Ferrite ( $\alpha$ -iron), Ferromagnetic, BCC crystal structure.
  - ✓  $T = 770 - 912^{\circ}\text{C}$  :  $\beta$ -iron, paramagnetic, BCC crystal structure.
  - ✓  $T = 912 - 1394^{\circ}\text{C}$  :  $\gamma$ -iron (austenite), FCC crystal structure.
  - ✓  $T = 1394 - 1538^{\circ}\text{C}$  :  $\delta$ -iron, BCC crystal structure.



Another allotropic form called  
Hexaferrum at 10GPa pressure



Iron becomes Anti-  
ferromagnetic!



Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur

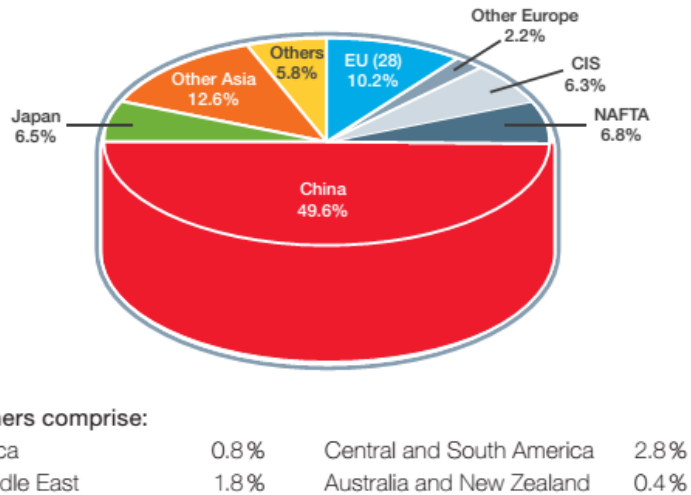
# World Crude Steel Production (1950 - 2015)

Reference: World Steel Association Report - 2016

- World crude steel production reached **1,621 million tonnes** for the year **2015**, in which **China** accounted for about **50 %** of the **global market for steel** (by volume).

## Crude steel production (2015)

World total: 1,621 million tonnes

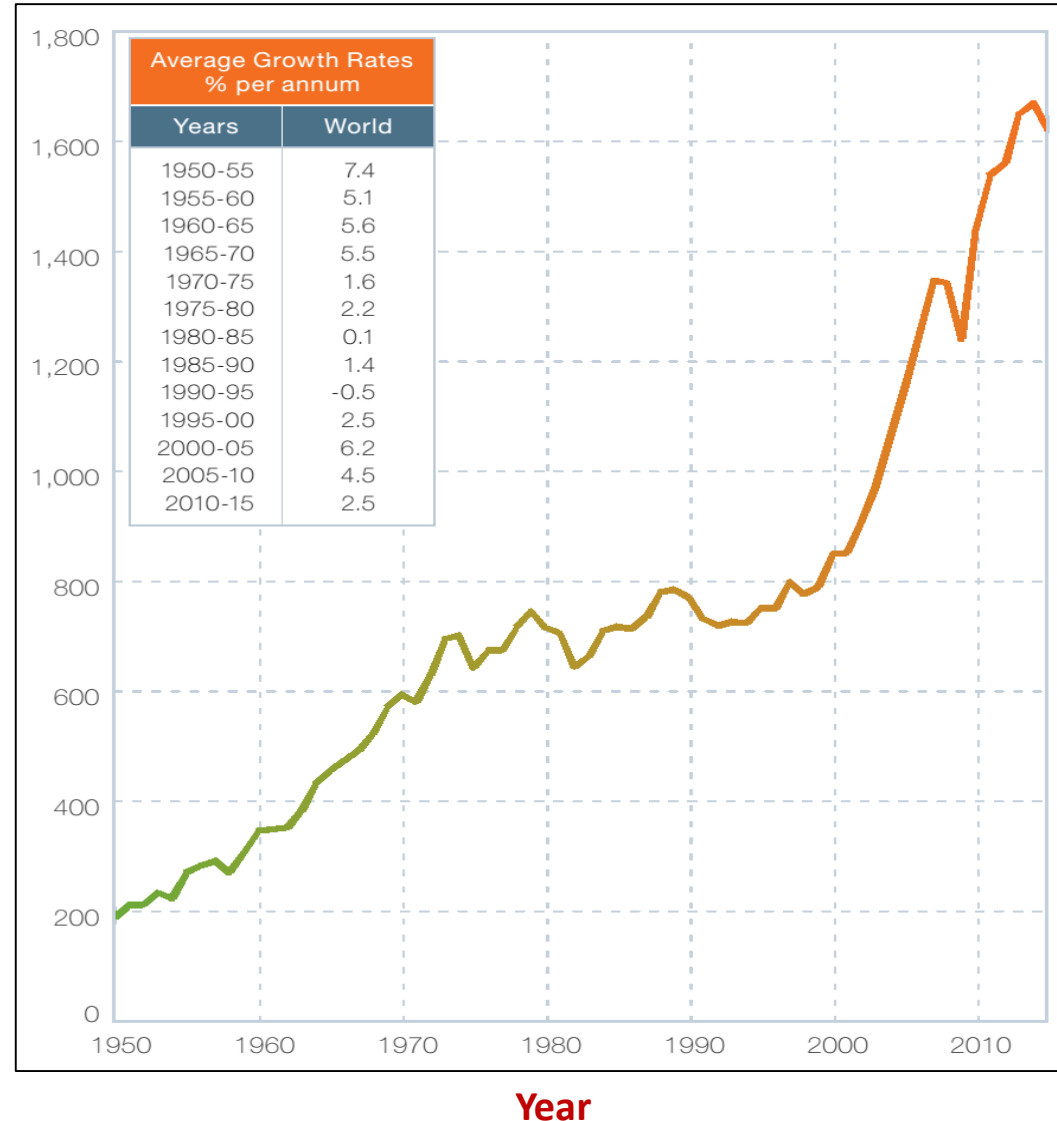


CIS: Commonwealth of Independent States

NAFTA: North American Free Trade Agreement(USA, Canada, Mexico)

EU: European Union

Million tonnes/annum



Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur

# Top Steel Producing Countries & Companies (2015)

Rank	Country	Million tonnes per annum
1	China	803.8
2	Japan	105.2
<b>3</b>	<b>India</b>	<b>89.4</b>
4	United States	78.8
5	Russia	70.9
6	South Korea	69.7
7	Germany	42.7
8	Brazil	33.3
9	Turkey	31.5
10	Ukraine	23.0

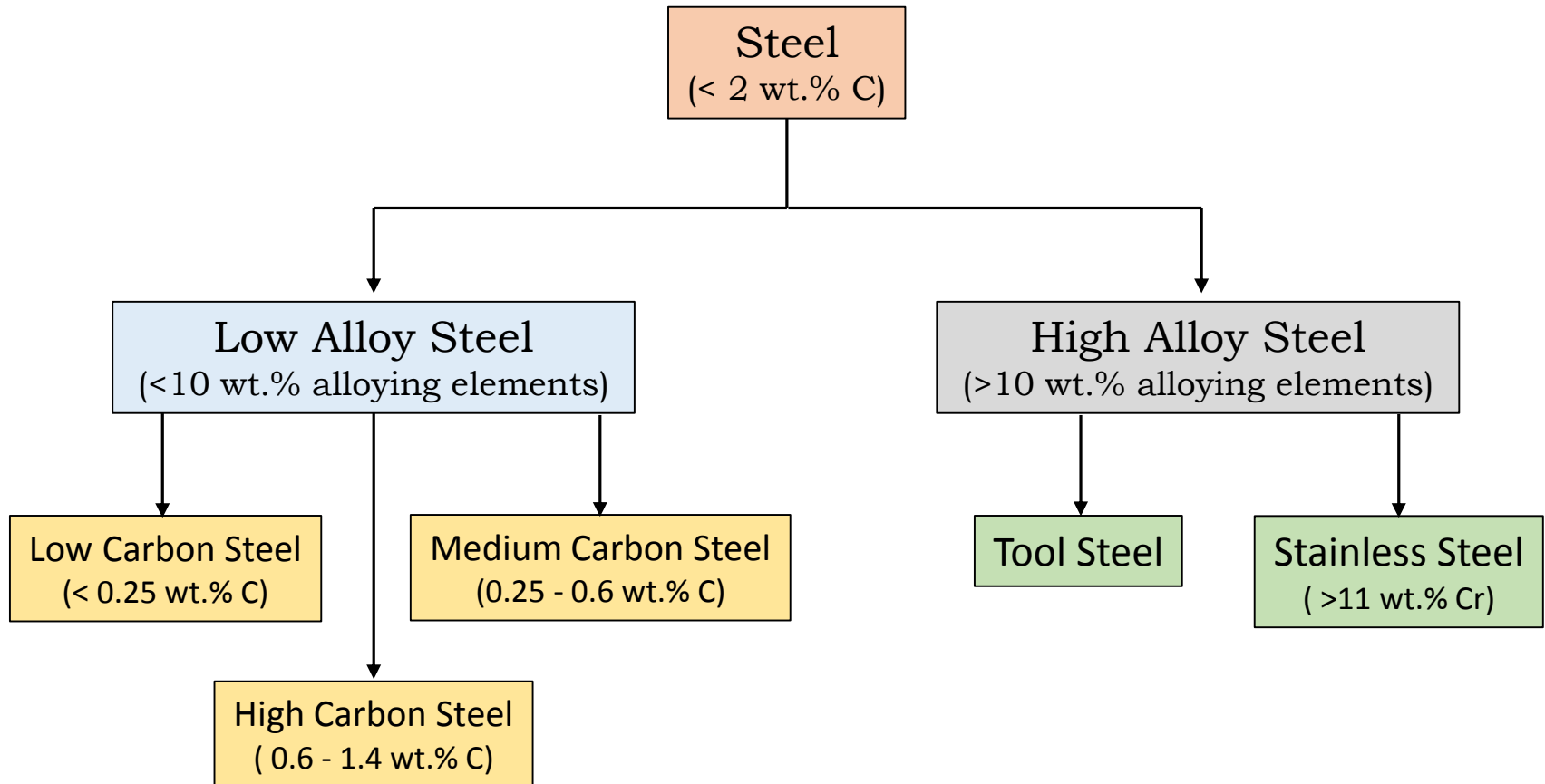
Rank	Company	Million tonnes per annum (2015)
1	Arcelor Mittal	97.14
2	Hesteel Group	47.75
3	NSSMC	46.37
4	POSCO	41.97
5	Baosteel Group	34.94
6	Shagang Group	34.21
7	Ansteel Group	32.50
8	JFE Steel Corporation	29.83
9	Shougang Group	28.55
<b>10</b>	<b>Tata Steel Group</b>	<b>26.31</b>
<b>26</b>	<b>SAIL</b>	<b>14.34</b>
<b>30</b>	<b>Jindal Steel Limited</b>	<b>12.42</b>

Reference: World Steel Association Report - 2016

Rank 10, 26 & 30 are held by Indian group of companies



Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur



**Low alloy steel** is further divided into **Plain carbon steel** and **Alloy steel** of respective category.





As per **American Iron and Steel Institute (AISI)** definition for **Plain carbon steel**:

- ✓ When no minimum content is specified for alloying element (Cr, Co, Mo, Ni, Ti, W, V, Zr, etc.) to be added to obtain a desired effect.

**OR**

- ✓ When the specified minimum amount for copper (Cu) does not exceed 0.40 percent.

**OR**

- ✓ When the maximum content for any of the following elements does not exceed the percentages: Manganese (1.65), Silicon (0.60), Copper (0.60).

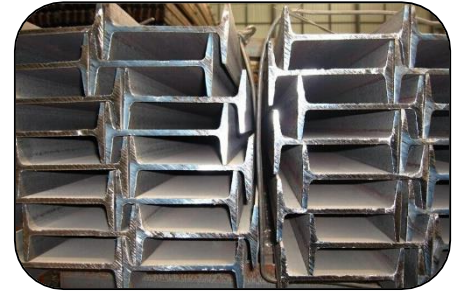
• Effects of **increasing carbon** content in steel are:

- ✓ Increase in hardness & strength.
- ✓ Decrease in weldability.
- ✓ Decrease in ductility.
- ✓ Decreased machinability (about 0.2 to 0.25 wt.% C provides the best machinability).



# Low Carbon Steel

- Contain less than about 0.25 wt.% C (Mild steel).
- Relatively soft and weak.
- Outstanding ductility (25% EL) & toughness.
- Also, high machinability and weldability.
- Least expensive to produce.
- Tensile strength (415-550 MPa).



## Low alloy steel:

- Contains alloys such as Cu, V, Ni & Mo up to 10 wt.%
- High strength & corrosion resistance than plain low carbon steel.
- Tensile strength up to 700 MPa.



## Applications:

Beams, Channels, nuts, bolts, wires, tin cans, etc.



# Medium Carbon Steel

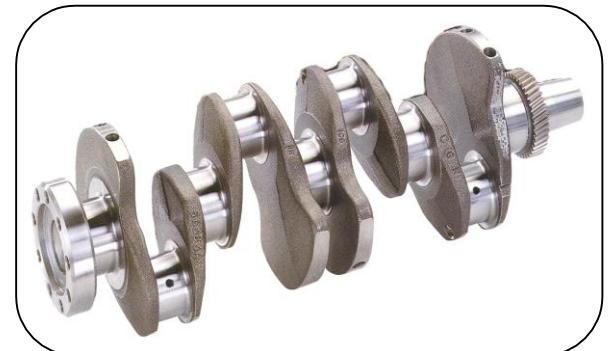
- Contain 0.25 - 0.6 wt.% C.
- Stronger than low-C steels but of low ductility and toughness.
- Good wear resistance.
- Plain carbon steel (Tensile strength up to 850 MPa) & alloy steel (Tensile strength up to 1900 MPa)
- **Applications:** Railway wheels & tracks, gears, crankshafts, etc.



Rail wheels



Gears



Crankshaft



# High Carbon Steel

- 0.6 - 1.4 wt. % C .
- Hardest, strongest and least ductile carbon steel.
- Can be alloyed with other metals to form very hard and wear resistance material (e.g. Cr, Ni, W, Mo and V).
- **Applications:** Cutting tools, embossing dies, saws, concrete drills, etc.



Die



Circular saw



Concrete drill



# High Alloy Steel(>10 wt.% alloys) - Tool Steel

- ✓ Commonly used in drill bits & other rotating cutting tools.
- ✓ It can withstand higher temperatures without losing its hardness & toughness.
- ✓ **Example**
  - ❖ 18-4-1 HSS: 18% tungsten, 4% chromium, 1% vanadium with a carbon content of 0.6 - 0.7%.
  - ❖ Cobalt high speed steel – increased heat resistance
  - ❖ Molybdenum high speed steel – Mo increases hardness and wear resistance.

Also cost effective replacement for tungsten in tool steels.



Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur

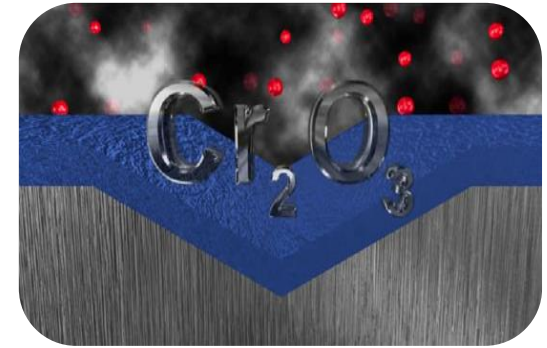
# High Alloy Steel - Stainless Steel

- Highly resistant to corrosion in a variety of environment.
- Predominant alloy: **Chromium (at least 11 wt.%)**.
- **Example:** 18/8 stainless steel - 18% chromium and 8% nickel.
- **Applications:**
  - ✓ Cryogenic vessels.
  - ✓ Food processing equipment's.
  - ✓ Gas turbines parts.
  - ✓ High-temperature steam boilers.
  - ✓ Heat-treating furnaces.
  - ✓ Nuclear power generating units.

Reference: [www.surfox.com](http://www.surfox.com)



Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur



Passivation



Christian Friedrich  
Schönbein



# Effect of alloying elements on Steel

S.No.	Element	Effects
1.	Boron (B)	✓ Improves hardenability without the loss of machinability.
2.	Chromium (Cr)	✓ Improves oxidation (at high temperature) and corrosion resistance. ✓ Corrosion resistance may also be enhanced by Ni and Mo additions.
3.	Cobalt (Co) & Tungsten (W)	✓ Improves strength and hardness at elevated temperatures.
4.	Sulphur (S)	✓ Improves machinability when combined with manganese. ✓ Alone it increases brittleness & lowers impact strength and ductility.
5.	Manganese (Mn)	✓ Improves hardenability & wear resistance. ✓ Counteracts the brittleness caused by Sulphur.
6.	Molybdenum (Mo)	✓ Improves hardenability, toughness. ✓ Improves elevated-temperature strength, creep resistance.
7.	Nickel (Ni)	✓ Increases strength and hardness without sacrificing ductility and toughness.
8.	Vanadium	✓ Increases strength, hardness, wear resistance and resistance to shock impact at high temperature.
9.	Titanium	✓ Improves strength. ✓ Deoxidizes steels.



## Relative effect on Steel

	Cr	Mn	Mo	Ni	Ti	W	V
Hardenability	++	++	++	+	++	++	+++
High temperature Strength	+		++	++	+	++	++
Ductility & Toughness		+		++			
Wear resistance	+		+		+	++	+
Promote fine grain size			+		++	+	+++
Corrosion resistance	++		+	+			

**Hardness** is a material property & is a resistance to penetration, scratching, etc.

**Hardenability** is a way to indicate a **material's potential** to be hardened by heat treatment.





# Cast Iron Types

(> 2 wt.% C)



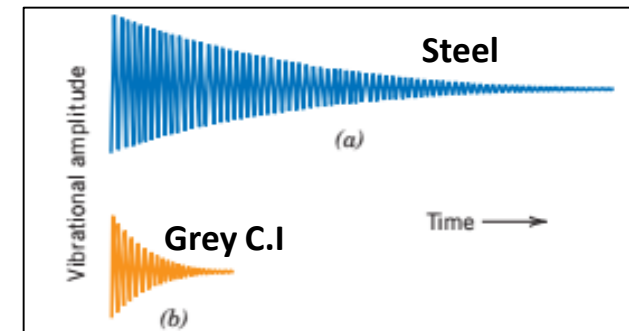
Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur

# 1. Grey Cast Iron

- ✓ Carbon content varies from 2.5 – 4.0 wt.%.
- ✓ Graphite exists in the form of flakes.
- ✓ Graphite flakes gives **self-lubricating** property and **vibration damping** capability.
- ✓ Strength and ductility are much higher under compressive loads.
- ✓ Tensile strength = 120 – 280 MPa.
- ✓ **Application:** Base structures for machines and heavy equipment that are exposed to vibration.



Grey Cast Iron microstructure



**Damping capacity**



## 2. White/Chilled Cast Iron

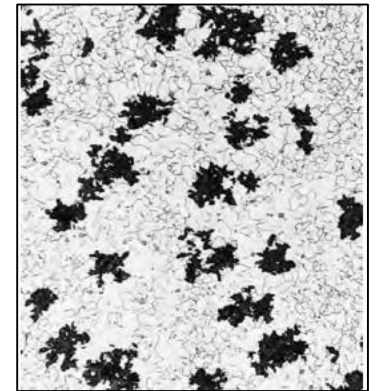
- ✓ No graphite, carbon in the form of carbide (cementite – hardest constituent of iron)
- ✓ Formed by rapidly cooling molten iron.
- ✓ Very hard, wear and corrosion resistant.
- ✓ Almost non - machinable.
- ✓ Application: Rollers in rolling mills.



White Cast Iron microstructure

## 3. Malleable Cast Iron

- ✓ Formed by **heating white C.I** between 800-900°C for a **prolonged time** in a neutral atmosphere (to prevent oxidation) leads to the decomposition of the cementite, **forming graphite** in the form of **clusters**.
- ✓ Highly shock resistant or tough.
- ✓ Tensile strength = 350 – 450 MPa.
- ✓ Can be hammered to small thickness.
- ✓ **Applications:** Connecting rods, transmission gears, and differential cases for the automotive industry and flanges, pipe fittings, and valve parts.

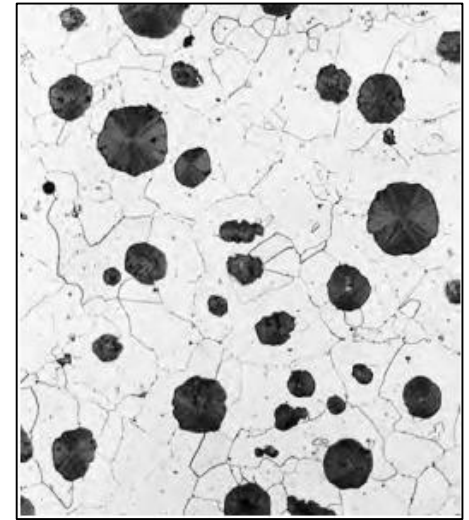


Malleable Cast Iron microstructure



## 4. Ductile/Nodular/Spheroidal Cast Iron

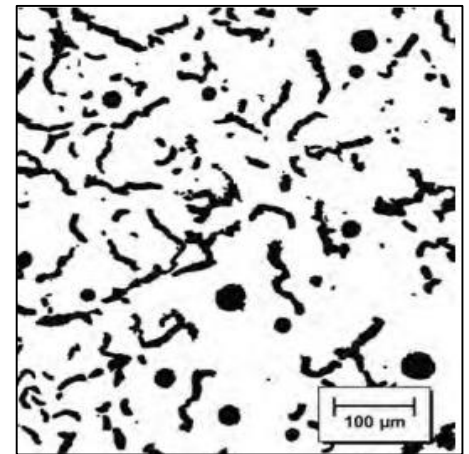
- ✓ Obtained by adding small amount of Magnesium (0.1-0.8%) to molten Grey C.I leading to the formation of graphite in the forms of spheres.
- ✓ High fluidity.
- ✓ High Tensile strength (400 – 900 MPa).
- ✓ Tough, wear resistant.
- ✓ Good machinability and weldability.
- ✓ Designated as SG 900/2 representing tensile strength and % elongation.



Ductile Cast Iron microstructure

## 5. Mottled/Compacted Cast Iron

- ✓ Product in between Grey and ductile C.I
- ✓ Carbon partly free and combined form.
- ✓ Graphite has worm-like appearance.
- ✓ Higher thermal conductivity.
- ✓ Better resistance to thermal shock
- ✓ Lower oxidation at elevated temperatures
- ✓ Application: diesel engine blocks, exhaust manifolds, gearbox housings, flywheels, etc.



Compact Cast Iron microstructure



# Effect of Impurities on Cast Iron

## 1. Silicon (Si)

- ✓ Provides formation of free graphite, makes iron soft and easily machinable.
- ✓ Produces sound casting free from blow-holes as having high affinity for oxygen.

## 2. Sulphur (S)

- ✓ Makes C.I hard and brittle.
- ✓ Above 0.1% makes gives unsound casting.

## 3. Manganese (Mn)

- ✓ Makes C.I hard by formation of carbide.
- ✓ Keeps control over harmful effects of sulphur.

## 4. Phosphorous (P)

- ✓ Imparts fusibility & fluidity but induces brittleness.



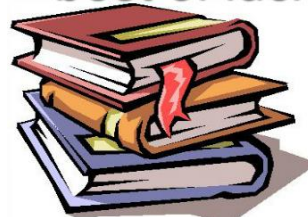
In the **next lecture**, we will learn:

- **Metals (Non-Ferrous alloys)**

- ✓ Classification

- ✓ Properties

best of luck



Smart Materials Structures and Systems  
Laboratory  
IIT Kanpur