fyVel dk jax i fjor Zi ughidjr A t S &NaCl, KCl, NaNO3, Na2SO4 bR km A

- (ii) VEyh, yo.kfo; yu(Acidic Salt Solution) izy VEy, oanqZy HLe dsyo.kfoy; u vEyh; ghrsga bl dkpH ehu 7 lsde ghrkga; s fyVel dkyky jax uhykesi fjofræ djrsgs tøsh4CI, (NH4)2SO4 A
- (iii) Hely yo.kfoy; u(Basic Salt Solution) izy He, oangly vEy ds yo. k foy; u {kkj k; gkrsg& budk pH eku 7 l s v fêkd gkrk g& t \$ s- Na₂CO₃, K₃PO₄ bR kinA

iżu 18- yo.k dsl lelu, xglkadksfy[kd mlkj & yo.k dsxqk fuEufyf[kr g&%

- (i) izy vEy rFlk izy HLe I scus yo. Wordk t yh; foy; u mnkl hu glark gSrFlk foy; u dkpH ehu 7 glark gA t I s KCI, NaCI, KNO3 bR kinA
- (III) izy vEy rFlknqZy HLe 1 scusyo. Wadkt yh, foy; u vEyh, gkrk gs ts-NH4CI, FeCI3, FeSO4 bR kinA
- (iii) ngZy vEy rFlkizy Hile I scusyo. Wadkt yh, foy; u {kijh, gkrk gsrFlk foy; u dk pH eku 7 ls vfekd gkrk gs tss- Na2CO3, NaHCO₃, CH₃COONa bR, kinA

izu 19-pH *Ldsy D; k gS* (SPL)

mlkj &1909 bZ eal kj.Ll u usH+ vk; u dh l kazk dksQ Dr djusdsfy, pH fpå dkmi; kx fd; k bl es0 l s14 rd dh l 4; k; gkrsg\$ bl spH Ldsy dgk t krk g\$

iżu 20- pH eku D; k gS by R mnki hr

mÜlg &xhe v. hgi fir yhVj en Q Dr ghbMht u vk, uhn ds 1 kmzk ds_ . hhled y?haxqhd dhspH ehu dgk t hrk gA

pH =
$$-\log[H^+] = \log\left[\frac{1}{H^+}\right]$$

'hg ty dkpH eku 7 gkrk g\$

iżu 21-pH eku dk D; k egko gS

mÜhj & gehjsnSud thu espH vR, Ur eghbi whZLFhhu j[krk gSl bl dseghb

fuEufyf[kr g&%

- (i) ty dkpH eku Kkr djdsirk yxk; k t krk gSfd ty fdl dk; Zds fy, mi; Ør gSl
- (ii) jDr dkpH eku Kkr djdsirk yxk; k t krk gSfd jDr 'kg gS; k v'kg A
- (iii) feVVh dkpH eku Kkr djdsirk yxk, k t krk gSfd bl eødkø&l h Ql y mxk, h t k l drh g\$l
- (iv) vusd jl k, fud vfHsØ; k, spH }kjk fu; s=r dh t krh g\$t t\$ s&t y vi?kVu vfHsØ; kJ fd. ou bR, kfnA
- (v) ikpu rædspH dkirk yxkdj jkxkødh t kudkjh ikkr dh t krhg& (1.0)
- (vi) nkrkadspH eku esifjorZi gkusij nkr u"V gkus yxrsgA (5.5)
- (vii) ty dkpH , d fuf'pr l hek ds vænj jgusij jgusokyst yh; t ho t hfor jgrsgæ
- iżu 22-mnki hukdj. k vfHkO; k l svki D; k l e>rsg& mnkgj.k}kjkl e>koA mlkj &vEy rFkk {kkjd dsvfHkO; k dsQyLo: i yo.k rFkk t y curk g\$ ml s mnki hukdj.k vfHkO; k dgrsg\$

NaOH + HCI → NaCI + H₂O

iżu 23- usty iksk (cht) dh D; k fo kkrk gs

milij &usly, d'kk[kh; ikskkgSt kst axyksesmit rkgsl bl dsifik; ksesaMaduqk cky gkrsgsl vxj xyrh l s Nvhy; k t k, rks Mad t s k nn Zgkrk gsl bu ckyksesesfkukbd v Ey dk l ko gkrus ds dkj. k nn Zgkrk gsl Mad ekjus ds LFkku i j Mkkl i ksksdh i ikh jx Musi j bykt gkst krk gsl; si ksks v fêkdr j usly dsikl i k; st krsgsl Mkkl i ksksl s dpl Hile; k {kj fudyrsgsl t ks v Ey ds i Hko dks mnkl hu dj nsrsgsl

iżu 24- vPNsQl y dsfy, feVVh dkpH eku 5.5 – 7.0 gkuk pkg, A fdl ku feVVh en puk D; kn feykrk gS|

mlkj & feVVh dkpH eku 5.5-7.0 dschp jgusij Ql y vPNsgkrsg& feVVh ds vR; fêkd vEyh; ; k {kkjh; gkusij i kkkkadh of) chêkr gkst krh g\$\mathbb{S}\text{left} feVVh ds vfêkd vEyh; gkusij ml eadyh pwh Hafkjk pwk; k d\$Y'k; e dkck&\mathbb{S}\text{Vh Mkydj ml dkpH fu; k=r fd; k t krk g\$\mathbb{S}\text{ bu jkl k; fud i nkFkkeds Hafled gkusdsdkj.k; sfeVVh ds vfrfjDr vEyh; rk dksde dj nsrs

ga vr%fdl ku puk feykrk ga

iżu 25- {kijkadsegRoiwkZjkl k; fud xgkkadksfy[ka

mÙig & {kig ka cis eg Roi wk? j ki k; fuci xqk fuEufyf[kr g&%

- (i) èlkry kal sf0; k &{kij dip èlkry kal sf0; k dj H₂ x\$ mbi lii djrsg\$\frac{\pi}{2}\$ Zn + 2NaOH → Na₂ZnO₂ + H₂
- (ii) Ok, ql sfO; k& dqN {kdj ok, qesimifLFkr CO_2 l sfO; k djrsgSh $2NaOH + <math>CO_2 \longrightarrow Na_2CO_3$
- (iii) V E V ka l s f O; $k \& \{ k j \mid V E V ka l s f O ; k d j d s y o . k r s k j d j r s g s a NaOH + HCI <math>\longrightarrow$ NaCI + H₂O
- (iv) yo. kkal sf0; k& rkck ykgk ft ad vkin dsyo. k {kijkal sf0; k djrs g\$ vk\$ v?kyu'khy èkkhod gkbMkM kbM r\$kj djrsg\$ ZnSO4 + 2NaOH ---> Na2SO4 + Zn(OH)2

iżu 26- gekisnSud t hou envEykndspki mi; kx crkon mÙkj &gekisnSud t hou envEy dsmi; kx fuEufyf[kr g&%

- (i) fljdk gelijs Hkt u dlø i dlus vlij ml dh ljj{lk rFkk vlipkj culus est dlæ vkrk gsl
- (ii) gelgsisV esHCI glaudligd t lok lay ladlsu"V dj nsrk gsI t ls Hkt u ds 1 liFk i gsp t krs gsI
- (iii) VkjVSjd vEy csdx i kmMj cukuses dhe vkrk gSl
- (iv) dkolud vEy is inkflueenizipr glark gs

iżu 27- vEykadh gekjst kou eaD; k gku; k g&

mÙhj &vEykal sgkuskyh gkfu; kj fuEufyf[kr g&%

- (i) ; slth dk kdk kadk u"V djrsg&
- (ii) I knz v Ey Popk v Ky dkey v a kn dk x Mily {Kr i gopkrs g M
- (iii) de [kk/ inkfk&dks [kjkc dj nerses]
- iżu 28- 1 kfM; e DykjkbM ¼ kèkij. k ued½d\$ scuk; kt krk g\$\ b1 dsnkseq; j1 k; fud xqk rFkk mi; kx crkost
- mÙkj&xeZl kkM; e ij Dykjhu xS i økSr djusij l kkM; e DykjkbM curk gSl 2Na + Cl₂ \longrightarrow 2NaCl

jkt k, fuct xqk&%

(i) ; g , d vk; fud ; k&xd g\$ t k vfr ?kyu'khy g\$

- (ii) ; g , d 'or joknkj inkFkZg\$\ mi; kx &%
- (i) Hhat u cuhus en
- (ii) gloMkt u DylgloM(HCI), csdax i kmNj] I ksM; e cloZlkckuN] I ksM; e gloMkM loM vkin dsfuelZk esl

iżu 29- 1 kèkij. k ued dh i kilr dgk&dgk; gkrh g\$\ Li "V dj. k mlkj & 1 kèkij. k ued fuEufyf[kr 1 krkn] s i kir gkrk g\$%

- (i) leghty&lequids [kijsty dkscM&cM&xM<knen, d= dj l wZds izlk kenok ir gknsnrsgn ok'i u dsckn Bkl ued dsjosikir gkrsgn
- (ii) [Nut ued Alkula; k pVVkula I skvlu Vsy; k es ued [Nula I s fuckyk t krkgs bl dsfy, t ehu ds vanj, d i li ?klj krsgs ft l es rhu l adshih ukfy; k gkrh gst ckgj okyh uyh l sxeZt y vanj i askk djk; k t krkgs ft l l sued dk foy; u r\$ kj gkrkgs l cl svanj okyh uyh l sgkclj mPp nkc ij gok dk >kalk vanj Hkt k t krkgs ued ds foy; u dkschp okyh uyh l sgkclj ckgj fudky nrkgs foy; u dks Nkudj ok ir djusij ued i ktr gkrk gs
- (iii) > hykal s& jkt LFkku dh l kkkj > hyl vestjdk dh xxV l kVV ysdl : l dh ysd , YVu l s Kkh ued r\$kj gkrk g\$l bl s t y dsok'i hdj.k l s i kUr fd; k t krk g\$l

iżu 30- 1 kèkij. k ued gok eaD; ka i 1 kt us yxrk g\$\

mÙig & l hèlkg. k ued env'hø) ds: i eneskuhf'k, e DykghoMjgrk gs MgCl₂ , d i konh i nkflæst ksueh l kfkrk gs bl h dkj. k l hèlkg. k ued [kyh gok en j [kus i j i l ht us yxrk gs

iźu 31-1 kM; e głóMMM kbM (NaOH) dß scuk k t krk g\$\ b1 dsmi; kx crłod
mlkj&l kM; e głóMMM kbM dks Dykj , Ydyh fofèk }kjk cuk; k t krk g\$\ b1 s
dkIVd 1 kWk Hh dgrsg\$\ b1 sfo/qr vi?kVu fofèk }kjk cuk; k t krk g\$\
I kM; e DykjkbM dst yh; foy; u enfo/qr èkkjk i pkgr djusij; g
vi?kVr gkdj 1 kM; e gkbMMM kbM, Dykjhu rFkk gkbMkt u cukrk g\$\
2NaCl + H₂O → 2NaOH + Cl₂ + H₂↑

mi: kx&%

- (i) èlary la ds xht gVlus est
- (ii) I kcyı/ viekt ZI rFkk dkxt dsfuekZk est
- (iii) Ñf=e Qlbcj/ Ñf=e oL=/ jsˈks vkfn dsfuekZk est

iżu 32-1 ksM; e ckbZkckZsV ; k [kkusdk] ksMk dS scuk; k t krk gS\ b] dsnks ef; jkl k; fud xgk rFkk mi; kx crhod

mÙkj &1 ksM; e ckbZlkckUsV clks veksu; k 1 ksVk fofêk ; k 1 kYos fofêk } kjk cuk; k t krk g\$

I kfM; e dkckffV dstyk; ?kky esco₂ xS i pkgr djusij I kfM; e cholehekas dk volki istr gkrk gs

 $Na_2CO_3 + H_2O + CO_2 \longrightarrow 2NaHCO_3$

jkl k; fud xgk &%

- (i) ; g , d joknkj 1 Qsn Bkd i nkFkZgSl
- (ii) [kkuk i dkrsle; tc; g xeZgkrk gSrks; g lkM; e dkckus] ty rFlk dlca MbykM lbM x\$ nsckg\$ vr% [kkuk dls 'ki/krk 1 si plus $\frac{dsfy, \ bl \ dk \ mi; \ kx \ fd; \ k \ t \ krk \ gS}{A''ek}$ $2NaHCO_3 \xrightarrow{A''ek} Na_2CO_3 + H_2O + CO_2$

mi; ks &%

- (i) bldk mi; kx vfXu'kled ds: i engkrk gSl
- (ii) bl dk mi; kx csdax i kmMj ds fuekZk ea fd; k t krk gsl

iżu 33-1 k/M; e dkck/W lékkrus dk 1 kMk/2dS scuk, k t krk gS\ b1 dsnksef; jkl k, fud xgk rFkk mi; kx crhod

milij& bl dk jkl k; fud uke l kfM; e dkckijV Mclk gkbMV g\$ ft l dk l # Na₂CO₃. 10H₂O gkrk gs

csdax I kWk dks xje djus ij I ksM; e dkckusV curk gs

 $2NaHCO_3 \longrightarrow Na_2CO_3 + H_2O + CO_2$

iMr 1 kM; e dkckus dksty 1 s fOLVyhÑr djus 1 s èkkou 1 kWk iter gterk gs

 $Na_2CO_3 + 10H_2O \longrightarrow NaCO_3 \cdot 10H_2O$

jkl k; fud xqk &%

- (i) ; g 1 Qsn i kjn'kki joknkj i nkiki gkrk g\$\igst\}
- (ii) bl dksxje djusij ; g 10 v. kyjok t y ds [kksnsk g\$ vk\$ fut Z/k; Na₂CO₃ cukrk gs

 Na_2CO_3 . $10H_2O \longrightarrow Na_2CO_3 + 10H_2O$ fut Z/h, Na2CO3 dks l kWk {kkj ; k l kWk jk/k dgrsg&