CHEMISTRY



CHAPTER – 2

vEy] {kijd , oayo. k **ACID, BASE & SALT**

iżu 1- vEy fdl s dgrsg&

mÜlj&vEy og i nkFlZgSft 1 dk Lokn [kVVk gkrk g\$ t ksuhysfyVel ds?kky dkyky dj nskg\$A tyh, foy; u englbMktu vk, u(H+) epr djrk gSrFlk elkrgij vEy dh vfHtO; k l sglbMtt u x\$ etpr gkrsg&A t sanci, HNO3, H2SO4 bR kin A

iżu 2- {kijd ; k HLe fdl sdgrsg\$|

milij&{lkjd og inkfkigSft i dk Lokn dMek gkrk g\$ yky fyVel dksuhyk cukrk gs bl dk t yk, foy; u (OH-) gkbMkt u vk, u elpr djrk gs rFlk vEy 1 s vfHlØ; k dj yo. k cukrk g\$ t JanaOH, CuO, CaO rFM Ca(OH)2 bR MnA

iżu 3- yo.k dh i fj HKkk mnkgj. k } kjk n**a**

mlkj&osinkFlZyo.kdgykrsg&tksfyVeli=kadsi£r mnklhugkrsg& èllegrilk vig kadschp villiø; k ds Qylo: i yo.k curs gs $Zn + 2HCl \longrightarrow ZnCl_2 + H_2 \uparrow$ singh

iżu 4- vEy dsiko xakkadksfy/ka

mlkj& vEy ds i kp xqk fuEufyf | kr g&%

- (i) VEy Lokn en [kVVsgkrsgn t S & ulufv l arjk]
- (ii) day vey to "kys glars gas to sk dkchetyd vey haqulay 1/2
- (iii) da vey lakijd, oagkiudkjd gkrsga tsal Yg; kjd veyA
- (iv) day veykal s vusd izdkj ds [kkn , oafolQkVd cuk, s t krsg& t \$ s &ulbfV& vEyA
- (v) day vey Loke; ij cijk izklo Mkyrsgs

iżu 5- {kg dsikp xqkkadksfy[ka

mÙkj& {kkj dsxqk fuEufyf[kr g&%

- (i) budk Lokn dMok gkrk g\$
- (ii) ; sl kcq t \$ sipdusgkrsg&rFkk Popk dksgkfu iggpkrsg&
- (iii) ; syky fyVel dksukyk dj nssg\$\frac{1}{2}
- (iv) ; sgYnh ds jax dksHyjk yky dj ngsgA
- (v) ; s v Eykads l kFk f0; k djds yo. k r Fkk t y cukrs g\$\mathbf{S}\mathbf{E}

iżu 6- {kijkadsmi; ks. crłost

mikj & {kgkadsmi; ks fuEufyf[kr g\$%%

- (i) bl dk mi; kx l kcq cukus esfd; k t krk g\$
- (ii) blga{kijh, cVij; kaeaiz Opr fd; k t krk gA
- (iii) buck mi; kx i \$V ky fj Qkbfuax vkf clkxt m/kx en i z pr gkrk g\$I
- (iv) dBkj ty dksengcukuseabl dk mi; kx fd; k t krk g\$\mathbb{S}\$

iżu 7 vEy rFlk {kj egvarj Li"V dj**a**

mlkj & vEy rFkk {kg esfuEufyf[kr vsrj g\$%%

	- 0	<i>v</i> - <i>v</i>				
A	v <i>E</i> y			{Hij	J	
(i) <i>b1</i>	dk Lokn kVVk	gkrk gå	(i)	bl dk lokn dM ; g yky fyVel dj nsk g\$	ok gk	rk g\$
(ii) ; g	g uhysfyVel i=	= dksyky dj	(ii)	;gykyfyVel	i=c	dks uhy k
119	rk gyl	, , , , , , , , , , , , , , , , , , ,		dj nirkgs	-0/,	/ 1 437 // 1 4
(iii) ; g	t y estoys gi	KOJ GHONIKE U	(iii)	;gtyesfoys g vk; u(OH-) nsr	kstj gkl	DIVINI KOM
(i-)	KU (H') IIKK GA	l hi di nakab		VK, U (OH) IIST	gal	i nakal
$(1V)$, ξ	Zanu ola 71	u Uj IIIK ga	(1V) ()	; g vEy dismnki bl dkpH eku 7 l	uu Cij	TIST K. GAL Ladenka (1
(V) DI	Chyn eu / 1.	oue gm n ga	(V)	VI CINPIT EN / I	SVICICI	gm Kga

iżu 8- vkjgkul }kjk nh xbZvEy rFkk HLe dh i fjHkkk n#

mlkj &vEy &vEy og inkfklgStksty en?kydj gkbMktu vk;u(H+) inku djrkg\$l

t I s & HCI, H₂SO₄, HNO₃, CH₃COOH bR, KnA

$$HCI \longrightarrow \frac{H_2O}{} \rightarrow H^+ + CI^-$$

HILE &HILE OG INKTIGST IST Y EN PROMINI I I INTO VIÇ U NISK GA t S S NaOH, KOH, NH4OH, Ca(OH)2 bR IInA

$$\mathbf{NaOH} \ \, \stackrel{\mathrm{H}_{2}\mathrm{O}}{-\!-\!-} \rightarrow \ \, \mathbf{Na}^{+} + \mathbf{OH}^{-}$$

- iżu 9- vk; uhdj. k ds vhèhij ij v Eyhadk foHht u fdu oxkteafd; k t krk g\$\\
 o. ku dja
- mlhj & vk; uhdj. k ds vhèhlj ij vEyhadk foHht u nhsoxheeafd; k t hrk g&%
 - (i) izy vēy (Strong Acid) the vēy ty en ?haydj yxhkx i whī % vk; fur ghalj ghb/Mht u vk; u(H+) izhu djrsgh ts sæ ghb/Mhðyhijd vēy (HCl), uhbf/d vēy (HNO3), l YÝ; hjd vēy (H2SO4) bR; hnA
 - (ii) now vey (Weak Acid) os vey that y earloydj fl Qzvhikd: i Is vk, fur ghragh ml anow vey dgragh that sadhchhud vey (H2CO3), I hivd vey (CH3COOH), now vey gh chid vey (H3BO3) Hh, d now vey gh ft l dk mi; ha, whi hivd da: i en ghrk gh
- iżu 10- foy; u esmifLFkr vEy dh ek=k ds vu¶ kj vEykadk foHkt u fdu oxkresfd; k t krk gS\ o. ku djst
- mÙij & fo; yu eami fLFkr vEy dh ek=k ds vuq kj vEykadks nks oxkæeackVk x; k g&%
 - (i) 1 kæ vEy (Concentrated Acid) t c foy; u envEy dh vfèkd ek=k
 mifLFkr jgrh gSrksml s l kæ vEy dgrsgM
 l kæ vEy ent y dh ek=k de jgrh gM
 - (ii) ruqvEy (Dilute Acid) t c foy; u envEy dh ek=k de jgrh gS rksml sruqvEy dgrsgM ruqvEy ent y dh ek=k vfèkd jgrh gM
- iżu 11- vEy rFkk {kkj ds vkjgsu; l fl) ktr dh l hekvkodk mYys[k djst mÙkj & vkjgsu; l fl) ktr ds nksk fuEufyf[kr gsi&%
 - (i) bl fl) kur ds vun kj v Ey H; Opr; koxd gs v kg {kkj OH; Opr; koxd gs v kg {kkj OH; Opr; koxd gs v kg {kkj OH; Opr; koxd ks of oh v Eykor F kk {kkj ko dh Q k f; k bl fl) kur ds v kèkkj i j ughadh t k l drh
 - (ii) æo vekku; kekNH4NO3 dkvkpj.kvEyh; gkrkg& bl dhQk[;kbl fl) kUr dsvkekkj ij ughadh tkl drh
 - (iii) tyh, foy; u esHCI rksvEy ekuk t krkgSfclarqx\$ h, voLFkk es; k vl; foyk, d \talkscst hu\2esvEy ughrekuk t krk1

- iżu 12- vk, ukdj. k dsvieklj i j HilekadksfdrusHkxkaeackVk x; k g\$\ o. ki djal milij&vk, ukdj. k ds vieklj i j Hilekadks nks oxkreackVk x; k g\$%
 - (i) izy Hle (Strong base)- os Hle t kst yk; foy; u eqi whith vk; fur gkclj dkQh ek=k eqgkb MkW kbM vk; u (OH-) izhu djrsga ml s izy Hle; k izy {kkj dgrsga t \$ &\text{NaOH}, KOH izy Hle ga
 - (ii) ngg/ Hile (Weak base) & os Hile the tylt foy; u enfl QZváhr% vk; fur gholj de ek=k enghbNhhl hbM(OH-) inhu djrsgh mls ngg/ Hile; k ngg/ {hhj dgyhrsgh t\$ &vekhu; e ghbNhhl hbM (NH₄OH), d\$Y k; e ghbNhhl hbMCa(OH)₂ |
- iżu 13-1 pod (Indicator) foll solgrsgs ; storusially olsglirsgs i fj. Hiller oljad milij&l pod , i s i nkfligkrs gst ks vi us jax i fjorit ols } kj k i nkflilos v Ey h ; k {lkj h; ; k mnkl hu gkus olh l pouk ners gst rhu l kekt; l pod fy Vel i=| feFklby v kj/kt rFkk QhukhQFksyu gst l pod olks v Ey {kkj d l pod kkh olgrs gst ; s nks i zlkj ols gkrs gsk/k
 - (i) i Mird I pod & i Mird I pod dsvarxT fyVel i = rFlkgYnh vkrs gA pqllhj| yky xkHh i ÙkA
- (ii) I ays kr 1 pod & bl dsvarxir feFkby vkjiht rFkkfQukNQFkfyu gM

iżu 14 xż fufez 1 pod vki ds scuk, sks

milij &gYnh dsNkV&NkVsVqlMrdksty dslkik xje dj Nku ysrsg& gYnh l sikir foy; u dksnk vyx&vyx ij[kufy; knenysrsg&, d ij[kufh enlkcq dkfoy; u Mkyrsg& ij[kuyh dsfoy; u dkihyk jax yky&Hyjs jax encny tkrkg\$ftllslkcq ds{kljh; gknsdhigpku gkrhg& nkjs ij[kuyhenfljdkMkyrsg& fljdkij[kuyhenj[ksfoy; u dsjax endkbZ ifjor% ughdjrk vr%fljdk vEyh; g&

- iżu 15- vEykadh 'kDr dsckjseavki D; k t kursg&| fdu&fdu rjhdkal s vEykadh 'kDr dh ryuk dh t k l drh g\$|
- mÙhj & vEy dst yh; foy; u envEy }hjk ihir ghbMht u vh; uhndh eh=k l s ml dh vEyh; 'hDr dk fuèhhj.k ghrkgN

निम्न तरीकों से अम्लों की तुलना की जा सकती है-

(i) VEV ds 1 Ar ds vhèhij & t S & dhoLud; h&xd & i h&k 1 s i Air v EV &1 hbf V et v Ev | v kV t S y d v Ev A

- (ii) vh. od l jpuk ds vhehij i j vEy & ghbMk vEy & HCI, HBr, HI
- (iii) *izy vey dsvieldj ij & vey t ylt, foy; u es*ai*vlit%vk, fur gla*rk gSft l dsvk, ulidj. k dk vák yxHx 100% glark gSl

iżu 16- yo. k fdl s dgrsg&|; s fdrus i zdkj ds gkrsg&| i fj Hkk kr dj.kl mÙkj & yo. k os; k&xd g&ft udk fuekZk fdl h v Ey dk fdl h {kkj d ds vfHkØ; k ds Qy Lo: i gkrk gS, oaft l eav Ey v. kyds; k&xd eami fLFkr gkb/kkt u

mnkl huhdj. k vfHkØ; k eadkbZvEy fdl h {kkjd dsl kFk vfHkØ; k dj yo. k , oat y cukrk gSA

t \$ s&1 kM; e gkbMkM kbM , oagkbMkDyksjd vEy dsmnkl huhdj. k vfHkO; k en1 kèkj. k ued yo. k , oat y cukrsg\$

NaOH + HCl → NaCl + H₂O

yo.k dsixlkj fuEufyf[kr g\$8%

i jek kgfdl h èkkrg}kjk folfkki r gkrsg\$

- (i) Leht, yo. h(Normal Salt) & og yo. kft l l svk, uhvr H i jek hq; k gkb/MkfDl y l eg ughnjgrk g\$ ml sl hekt, yo. k dgrsg& ; g vEy , oaHte dsi wk/mnkl huhdj. k dsQyLo: i curk g& t \$ & AnaCl, HCl, NaNO3, Na2SO4 bR, kmA
- (ii) VEyk, yo.k(Acidic Salt) &fdl h vEy ds v. kqenmi fl.Fkr fol.Fkhi u ; kK; gkb/kkt u i jek kqdksekkrq} kjk vákr%fol.Fkhi r djusdsQylo: i cus yo. k dks vEyk, yo. k dgrsg&

v Flok

os yo.k t ks fdl h Hkle } kjk fdl h v Ey ds vi wkl mnkl huhdj.k ds Qy Lo: i curs g\$ ml s v Ey h; yo.k dgrs g\$ bl en fo L Fkki u ; k\$; gkb Mkt u gkrs g\$ t \$ &NaHSO4, KHSO4, bR; km A

(iii) Helt yo. h(Basic Salt) & os Hele ft uds v. lqea, d 1 s vfêkd OH 1 eg gkrsga v Eyka} kjk v kixel ; i 1 smnkt hu gkelj Hkiled yo. k cukrk ga t \$ s & Pb(OH)NO3. b1 eafol Ekki u ; ki; gkb Mith kbM evyel gkrsga

iżu 17- pH ds vielkij ij yo. kwadk oxlicij. k dj. k

mlkj &vEy vkf HLe dhiNfr; kpH vkekkjr yo.kkadsfoy; u rhu rjg ds gkrsg&%

(i) mnkl hu yo. k foy; u (Natural Salt Solution) & izy vEy, oa izy
HLe ds yo. k foy; u mnkl hu gkrs g&l budkpH eku 7 gkrk g&l; s

fyVel dk jax i fjor Zi ughidjr A t S &NaCl, KCl, NaNO3, Na2SO4 bR km A

- (ii) VEyh, yo.kfo; yu(Acidic Salt Solution) izy VEy, oanqZy HLe dsyo.kfoy; u vEyh; ghrsgM bl dkpH ehu 7 lsde ghrkgM; s fyVel dkyky jax uhykesi fjofræ djrsgs tøsh4CI, (NH4)2SO4 A
- (iii) Hely yo.kfoy; u(Basic Salt Solution) izy He, oangly vEy ds yo. k foy; u {kkj k; gkrsg& budk pH eku 7 l s v fêkd gkrk g& t \$ s- Na₂CO₃, K₃PO₄ bR kinA

iżu 18- yo.k dsl lelu, xglkadksfy[kd mlkj & yo.k dsxqk fuEufyf[kr g&%

- (i) izy vEy rFlk izy HLe I scus yo. Wordk t yh; foy; u mnkl hu glark gSrFlk foy; u dkpH ehu 7 glark gA t I s KCI, NaCI, KNO3 bR kinA
- (III) izy vEy rFlknqZy HLe 1 scusyo. Wadkt yh, foy; u vEyh, gkrk gs ts-NH4CI, FeCI3, FeSO4 bR kinA
- (iii) ngZy vEy rFlkizy Hile I scusyo. Wadkt yh, foy; u {kijh, gkrk gsrFlk foy; u dk pH eku 7 ls vfekd gkrk gs tss- Na2CO3, NaHCO₃, CH₃COONa bR, kinA

izu 19-pH *Ldsy D; k gS* (SPL)

mlkj &1909 bZ eal kj.Ll u usH+ vk; u dh l kazk dksQ Dr djusdsfy, pH fpå dkmi; kx fd; k bl es0 l s14 rd dh l 4; k; gkrsg\$ bl spH Ldsy dgk t krk g\$

iżu 20- pH eku D; k gS by R mnki hr

mÜlij &xhe v. hqihr yhVj enQDr ghbMht u vk, uhnds 1 kmzk ds_ . hhled y?haxqhd dhspH ehu dgk t hrk gA

pH =
$$-\log[H^+] = \log\left[\frac{1}{H^+}\right]$$

'hg ty dkpH eku 7 gkrk g\$

iżu 21-pH eku dk D; k egko gS

mÜhj & gehjsnSud thu espH vR, Ur eghbi whZLFhhu j[krk gSl bl dseghb

fuEufyf[kr g&%

- (i) ty dkpH eku Kkr djdsirk yxk; k t krk gSfd ty fdl dk; Zds fy, mi; Ør gSl
- (ii) jDr dkpH eku Kkr djdsirk yxk; k t krk gSfd jDr 'k) gS; k V'k) A
- (iii) feVVh dkpH eku Kkr djdsirk yxk, k t krk gSfd bl eødkø&l h Ql y mxk, h t k l drh g\$l
- (iv) vusd jlk, fud vfHsØ; k, spH }kjk fu; s=r dh t krh g\$t t\$ s&t y vi?kVu vfHsØ; kJ fd. ou bR, kinA
- (v) ikpu rædspH dkirk yxkdj jkxkødht kudkjhiktr dht krhg& (1.0)
- (vi) nkrkadspH eku esifjorZi gkusij nkr u"V gkus yxrsgA (5.5)
- (vii) ty dkpH , d fuf'pr l hek ds vænj jgusij jgusokyst yh; t ho t hfor jgrsgæ
- iżu 22-mnki hukdj. k vfHkO; k l svki D; k l e>rsg& mnkgj.k}kjkl e>koA mlkj &vEy rFkk {kkjd dsvfHkO; k dsQyLo: i yo.k rFkk t y curk g\$ ml s mnki hukdj.k vfHkO; k dgrsg\$

NaOH + HCI → NaCI + H₂O

iżu 23- usty iksk (cht) dh D; k fo kkrk gs

milij &usly, d'hkfkir i kökk gSt kst avyknenmi t rk gsl bl dsi fik, knenMaduqk cky gkrsgsl vxj xyrh l s Nwfy; k t k, rks Mad t s k nn Zgkrk gsl bu ckyknen esfkukbel v Ey elk l ho gkrus els elkj. k nn Zgkrk gsl Mad e kjus els LFkhu i j Mkhl i köks elh i ikh jx Musi j bykt gkst krk gsl; si köks v fêkelr j usly els i kl i k; st krsgsl Mkhl i kökkal selip Hile; k {klj fuelyrsgsl t ks v Ey els i kko elks mnkl hu elj nersgsl

iżu 24- vPNsQl y dsfy, feVVh dkpH eku 5.5 – 7.0 gkuk pkg, A fdl ku feVVh en puk D; kn feykrk gS|

mlkj & feVVh dkpH eku 5.5-7.0 dschp jgusij Ql y vPNsgkrsg& feVVh ds vR; fêkd vEyh; ; k {kkjh; gkusij i kkkkadh of) chêkr gkst krh g\$\mathbb{S}\text{left} feVVh ds vfêkd vEyh; gkusij ml eadyh pwh Hafkjk pwk; k d\$Y'k; e dkck&\mathbb{S}\text{Vh Mkydj ml dkpH fu; k=r fd; k t krk g\$\mathbb{S}\text{ bu jkl k; fud i nkFkkeds Hafled gkusdsdkj.k; sfeVVh ds vfrfjDr vEyh; rk dksde dj nsrs

ga vr%fdl ku puk feykrk ga

iżu 25- {kijkadsegRoiwkZjkl k; fud xgkkadksfy[ka

mlkj & {kkjkndsegRoivkZjklk; fud xqk fuEufyf[kr g&%

- (i) èllary hal si0; k &{llij dip èllary hal si0; k dj H₂ x\$ mbi lu djrsg\$l Zn + 2NaOH → Na₂ZnO₂ + H₂ ⅓l kbN; e ft ad\$V½
- (ii) Ok, ql sfO; k & dqN {kdj ok, qesimifLFkr CO_2 l sfO; k djrsgSh $2NaOH + <math>CO_2 \longrightarrow Na_2CO_3$
- (iii) VEykal sf0; k&{kij vEykal sf0; k djds yo. k r\$kj djrsg& NaOH + HCl → NaCl + H₂O
- (iv) yo. kkal sf0; k& rkck ykgk ft ad vkin dsyo. k {kkjkal sf0; k djrs g\$ vk\$ v?kyu'khy èkkfod gkb/kkM kbM r\$kj djrsg\$ ZnSO4 + 2NaOH ---> Na2SO4 + Zn(OH)2

iżu 26- gekjsnkud t hou en v Ey kn ds pkj mi; kx crkon mlkj &gekjsnkud t hou en v Ey ds mi; kx fu Eufyf[kr g&%

- (i) fljdk gelijs Hikt u dis i dius vlij ml dh ljj{ik rFik vlipkj cukus esi die vkrk gsl
- (ii) gelgsist esHCI ghudhjd thok hy hadhsu"V dj nsrk gsI thsHhstu ds 1 hFk i gsp thrsgsI
- (iii) VkjVSjd vEy csdx i kmMj cukuses dhe vkrk gSl
- (iv) dkolud vey is inkikeenizer gkrk gs

iżu 27- vEykadh gekjst kou eaD; k gku; k g&

milij &vEykal sgknokyh gkfu; kj fuEnfyf[kr g&%

- (i) ; slthodk kdkvkadksu"V djrsg&
- (ii) I knz v Ey Popk v Ky dkey v a kn dk x Mily {Kr i gopkrs g M
- (iii) de [kk/ inkfk&dks [kjkc dj nssg&
- iżu 28- 1 kfM; e DykjkbM ¼ kèklj. k ued½d\$ scuk; kt krk g\$\ b1 dsnkseq; j1 k; fud xqk rFkk mi; kx crkost
- mÙkj&xeZl kkM; e ij Dykjhu xS i økSr djusij l kkM; e DykjkbM curk gSl 2Na + Cl₂ \longrightarrow 2NaCl

jkt k, fuct xqk&%

(i) ; g , d vk; fud ; k&xd g\$ t k vfr ?kyu'khy g\$

- (ii) ; g , d 'or joknkj inkFkZg\$\ mi; kx &%
- (i) Hhat u cuhus en
- (ii) gkb/Mkt u Dykjkb/M(HCI), csdax i km/Nj] l ksM; e ckb/zkckasV] l ksM; e gkb/MkN kbM vkin dsfuekzk est

iżu 29- 1 kèkij. k ued dh i kilr dgk&dgk; gkrh g\$\ Li "V dj. k mÙij & 1 kèkij. k ued fuEufyf[kr 1 krkn] s i kir gkrk g\$%

- (i) leghty&lequids [kijsty dkscM&cM&xM<knen, d= dj l wZds izlk kenok ir gknsnrsgn ok'i u dsckn Bkl ued dsjosikir gkrsgn
- (ii) [Hut ued Akuko; k pVVkuko 1 skvklVsy; k es ued [kuko 1 s fuckyk t krkgs bl dsfy, t ehu ds vanj, d i Ei ?ki krsgs ft 1 es rhu 1 solshih ukfy; k gkrhgs ckgj okyh uyh 1 sxeZt y vanj i sos k djk; k t krkgs ft 1 1 sued dk foy; u r\$ kj gkrkgs 1 cl svanj okyh uyh 1 sgkclj mPp nkc i j gok dk >kalk vanj Hkt k t krkgs ued ds foy; u dkschp okyh uyh 1 sgkclj ckgj fucky nrkgs foy; u dks Nkudj ok ir djusij ued i Hr gkrkgs
- (iii) > hykal s& jkt LFkku dh l kkkj > hyl vestjdk dh xxV l kVV ysdl : l dh ysd , YVu l s Kkh ued r\$kj gkrk g\$l bl s t y dsok'i hdj.k l s i kUr fd; k t krk g\$l

iżu 30- 1 kèkij. k ued gok eaD; ka i 1 kt us yxrk g\$\

mÙig & l hèlkg. k ued env'hø) ds: i eneskuhf'k, e DykghoMjgrk gs MgCl₂ , d i konh i nkflæst ksueh l kfkrk gs bl h dkj. k l hèlkg. k ued [kyh gok en j [kus i j i l ht us yxrk gs

iźu 31-1 kM; e głóMMM kbM (NaOH) dß scuk k t krk g\$\ b1 dsmi; kx crłod
mlkj&l kM; e głóMMM kbM dks Dykj , Ydyh fofèk }kjk cuk; k t krk g\$\ b1 s
dkIVd 1 kWk Hh dgrsg\$\ b1 sfo/qr vi?kVu fofèk }kjk cuk; k t krk g\$\
I kM; e DykjkbM dst yh; foy; u enfo/qr èkkjk i pkgr djusij; g
vi?kVr gkdj 1 kM; e głóMMM kbM, Dykjhu rFkk głóMkt u cukrk g\$\
2NaCl + H₂O → 2NaOH + Cl₂ + H₂↑

mi: kx&%

- (i) èlary la ds xht gVlus est
- (ii) I kcqu/ viektZl rFkk dkxt dsfuekZk est
- (iii) Ñf=e Qlbcj/ Ñf=e oL=/ jsˈks vkfn dsfuekZk est

iżu 32-1 ksM; e ckbZkckZsV ; k [kkusdk] ksMk dS scuk; k t krk gS\ b] dsnks ef; jkl k; fud xgk rFkk mi; kx crhod

mÙkj &1 ksM; e ckbZlkckUsV clks veksu; k 1 ksVk fofêk ; k 1 kYos fofêk } kjk cuk; k t krk g\$

I kfM; e dkckffV dstyk; ?kky esco₂ xS i zkfgr djusij I kfM; e cholehekas dk volki istr gkrk gs

 $Na_2CO_3 + H_2O + CO_2 \longrightarrow 2NaHCO_3$

jkl k; fud xgk &%

- (i) ; g , d joknkj 1 Qsn Bkd i nkFkZgSl
- (ii) [kkuk i dkrsle; tc; g xeZgkrk gSrks; g lkM; e dkckus] ty rFlk dlca MbykM lbM x\$ nsckg\$ vr% [kkuk dls 'ki/krk 1 si plus $\frac{dsfy, \ bl \ dk \ mi; \ kx \ fd; \ k \ t \ krk \ gS}{A''ek}$ $2NaHCO_3 \xrightarrow{A''ek} Na_2CO_3 + H_2O + CO_2$

mi; ks &%

- (i) bldk mi; kx vfXu'kled ds: i engkrk gSl
- (ii) bl dk mi; kx csdax i kmMj ds fuekZk ea fd; k t krk gsl

iżu 33-1 k/M; e dkck/W lékkrus dk 1 kMk/2dS scuk, k t krk gS\ b1 dsnksef; jkl k, fud xgk rFkk mi; kx crhod

milij& bl dk jkl k; fud uke l kfM; e dkckus Mclk gkbMs gs ft l dk l # Na₂CO₃. 10H₂O gkrk gs

> csdax 1 kWk dksxje djusij 1 ksM; e dkckusV curk gs $2NaHCO_3 \longrightarrow Na_2CO_3 + H_2O + CO_2$

iMr 1 kM; e dkckus dksty 1 s fOLVyhÑr djus 1 s èkkou 1 kWk iter gterk gs

 $Na_2CO_3 + 10H_2O \longrightarrow NaCO_3 \cdot 10H_2O$

jkl k; fud xqk &%

- (i) ; g 1 Qsn i kjn'kki joknkj i nkiki gkrk g\$\igst\}
- (ii) bl dksxje djusij ; g 10 v.kgjokty ds[kksnskg\$ vk\$ fut Z/k Na₂CO₃ cukrk gs

 Na_2CO_3 . $10H_2O \longrightarrow Na_2CO_3 + 10H_2O$ fut Z/h, Na2CO3 dks l kWk {kkj ; k l kWk jk/k dgrsg& mi; ks & %

- (i) bl dk mi; kx [kkjk t y dksengcukusesgkrk gs
- (ii) oh 'kax I kNk ds: i est
- (iii) dkxt m/kx est
- (iv) dkp m/kx est
- iżu 34- mRQgvu fdl s dgrs g\$\square\text{mRQgvu inf'\text{\text{iz}} djus okys, d;\text{\text{\text{bx}}} dk\\ u\text{\text{w} fy[\text{\text{\text{bh}}, d vfH\text{\text{\text{i}}}\text{\text{i}};\text{\text{k} n\text{\text{n\text{c}}}j l e>\text{\text{\text{o}}\text{\text{h}}}
- mÜlj&ok, qesı[loyk NlsN+nsısij 1 ksM; e dlıckisV jok ty [lluclj 1 Qsn ivliZesı cny tkrkgsI bl ds10 v.lqesıl s9 v.lqfudky dj ok; qesMy esipystkrs gsI 1 Qsn vikjn'lizl ilmMj ep tkrkg\$ft1s1 ksM; e dlıckisV eksukglbMV dgrsgsI b1 f0; k dksmRQoyu dgrsgsI

 Na_2CO_3 . $10H_2O \longrightarrow Na_2CO_3H_2O + 9H_2O$ $mRQyu\ inf'kT\ djusokys; kSxd\ lkM; e\ dkckUN\ gkrk\ gS$

- iżu 35- viXu 'Wed ; # } kjk vkx cq-kusdh fØ; k dks j kl k; fùd vfHkØ; k } kjk l e>kod
- milij&l ksM; e chbzikckus dk mi; ks vfXu'kked; æknen Hh fd; k t krk g\$\ vfXu'kked; æ en\ ha\co_3; k\co_2\ so_4 jgrsg\$\ vkx yxusij bl; æ dh ?ky\hij nkc Mkyk t krk g\$\ ft l l s\\ \text{NaHCO}_3 rFk\co_4\ ijLij l Eidzen vkdj \co_2 x\$\ cukrsg\$\ ; g x\$\ rt h l sckgj fudydj vkx dkscq\ k nsh g\$\

 $2NaHCO_3 + H_2SO_4 \longrightarrow Na_2SO_4 + 2H_2O + 2CO_2 \uparrow$

- iżu 36-fojat d pwł d s s cuk k t krk g s b l d s nks e f; j kl k, fud x q k r F kk mi; kx crkoa
- mlkj&bl dkjkl k; fud uke d\$Y'; e vkM hDykjkbM gkrkg\$I bl dkl # CaOCI₂ gkrk g\$I bl s Gylfpax i kmMj ; k purs dk DykjkbM Hth dgrsg\$I Bkl 'kfd cq-s purs i j 313 K ; k40°C i j Dykjhu x\$I i pkgr djus i j fojat d pukIcurk g\$I

$$Ca(OH)_2 + Cl_2 \xrightarrow{313K} CaOCl_2 + H_2O$$

jkl k; fucl xqk &%

- (i) ; g 1 Qsn pwl2gSft 1 esDykjhu t\$ h xik gkrh g\$
- (ii) ; g ruqH₂SO₄ dh vYi ek=k dsl kFk vfHkØ; k dj d\$Y'k; e l YQ\$/] gkbMkDyk§jd vEy rFkk vkM ht u n\$rk g\$\(\)

2Ca(OCI)CI + $H_2SO_4 \longrightarrow CaCI_2 + CaSO_4 + 2HCIO$ HCIO \longrightarrow HCI + O

bl izlkj iktr vkth htu uot kr vkth htu gkrk gSA vr%bl dk mi;kx fojatd ds: i eafd;k t krk gSI

mi; ks &%

- (i) bl dkmi; kk ty dk'k) djuseadhVk kqk'kd ds: i eafd; k t krkg\$\mathbb{B}
- (ii) dkxt rFlk diMk m/kx esfojt d ds: i est

iżu 37- dyh&pwk d\$ scuk, k t krk g\$\ bl dsnkseq; jkl k; fud xqk rFkk mi Hkx crhost

mÙkj&bl dk jkl k; fud uke d\$Y'k; e vkW kbM(CaO) g\$

t c puki RFkj dks 1000°C I sde rki ij , d HkVBhesexje fd; k t krk gSrks pukki RFkj VWdj dykkpuk rFkk CO2 cukrk gSl

 $CaCO_3 \xrightarrow{797^{\circ}C-997^{\circ}C} CaO + CO_2 \uparrow$

jkt k, fuet xqk %

- (i) ; g 1 Qsn cjoknkj i nkFkZgSl bl dk np. kkel 2597°C gksck gSl
- (ii) ; g' t y l si ErfO; k dj dSY' k; e gkbMMM kbM rFkk Å''ekepr djrkgSCaO + H₂O \longrightarrow Ca(OH)₂ \downarrow + $\mathring{A}''ek$

bl s cq-k pwk Hh dgrs gM bl i £0; k dks Hjduk; k i Mr pws dks Hjdh pwk dgk t krk gM

jkl k, fud xgk &%

- (i) I head m/ksx ead
- (ii) Cyhpax i kmNj cukus est
- (iii) dhp dsmRi hnu en

iżu 38- IykIVj vkt i sj. i ds scuk, k t krkgs bl dsnkej; jki k, fud xqk rFkk mi; kx crkos

mÙhj&b] dk jkl k, fud uhe d\$Y'k, e] YQ\$V vèhghb\M\$V g\$I ft] dk] # Ca\$O4 . $\frac{1}{2}$ H $_2$ O ghrk g\$I

blsvěkty; ktr dsyk, e l YQN dgrsgst blsl áki esp.o.p dgk t krk gsl

ft II e dks bLikr ds cjru e st 100°C rki ij xje djus ij ; g ty ds $1\frac{1}{2}$ v. ky ks dk R; kx dj IykLVj v kQ i sjl curk gSl

 $CaSO_4$. $2H_2O \xrightarrow{100^{\circ}C} CaSO_4$. $\frac{1}{2} H_2O + \frac{3}{2} H_2O$

jkl k; fud xqk &%

- (i) ; g , d 1 Qsn pullgs
- (ii) ty ds l kFk feydj ; g dMk rFkk fNnz fpr cu t krk g\$\mi; k\text{k}\%
- (i) 'kY; fpfdRl k ea Hh gfNAI; ka dks t kNus e A
- (ii) ehrZhadk 1 hpk cuhus est
- (iii) vilujkiko inkikoukusest
- iżu 39- fuEu dsdkj.k crkos&
 - (i) ihry rFlk rkcs ds cjru en ngh rFlk [kVVs i nkFlZD; kn ughn j [kuk pkfg, |
- mÜhj&nghrFlk [kVVsinkFlktenvEy gkrkgN vEy èkkryknlsvfklØ; kdj yo.k rFlkH2 xS cukrsgN ft] sinkFlZ[kkus; kK; ughnjgrkgN] kFkghngh , oa[kVVsinkFlktdksrkcsdscjruknenj[kk t k, xk rksvEy dh fØ; k ds dkj.k cjru] 4kkfjr gkstk, xkA
- (ii) *vey dk t yh, foy; u fo/q dk phyu D; hadjrk g\$*| mÙhj&vey t y e*n?hydj èhu , oa_ . k dk fuehZk djrk g\$*| HCl → H+ + Cl-
- (iii) 'hyd ghbMhDyhsid x\$ fyVel i= dk jx D; haughacnyrh g\$\ m\ddot\hyd ghbMhDyhsid x\$ eaghbMht u vk; u(H+) ughajgrk g\$\ bl fy, ; g
 vEyh; vfHky{k k inf'hT ughadjrk g\$\ ft l dsdhj.k fyVel i= dsjx
 dhsughacnyrh g\$\
- (iv) vki for ty enfo/q dkpkyu D; knuglngkrk cfYd o'lkit y engkrk g\$\)
 milig&vki for ty en dkbIvk; fud ; kixd foys uglngkrh ft I ds dkj. k; s
 vk; uknenfo?kVr uglngkrsgh o'lkit y ok; qhIy I sgkrsgq Hhe ij fxjrs
 I e; ok; q ds vEyl; x\$\mathbb{s} \text{CO}_2, \text{SO}_2, \text{NO}_2 bR; kin dks ?kyk nrk g\$\mathbb{l}
 ft I I sfofHili izlkj ds vEy Øe'lkodkchud vEy (H2CO3), I YD; jil vEy
 (H2SO3), ukbVi vEy (HNO2); kukbfVtl vEy (HNO3) cukrsgh ; svEy
 vk; uknenfo?kVr gkrsgh bl fy, o'lkit y fo/qr dk pkyu djrsgh

- (v) ty dh vuajfleher en vey dh Q oghj veyh, D; ha ughaghark |
- mllij&vEy døy ty dhmiflFlfr engløMnt u vk, u mrilu djrsgn gløMnt u vk, u dhmiflFlfr dsdlj.k vEykndk Q oglij vEylt, glork gn vr%t y dhvuqflFlfr engløMnt u vk, u ughcurn bl dlj.k vEy vi uk vEylt Q oglij ughdjrld
- iżu 40- rkt snik dspH eku 6 gkrk gA ngh cu t kusi j bl dspH eku eAD; k ifjorZu gksk |
- mÙkj&ngh en y\$DVd vEy gkrk g\$I vFkkT~tc nèk l sngh cu t krk g\$rksog vfèkd vEyk, gkst krk g\$I bl fy, nøk pH dk eku 6 l sde gkst k, xkA
- iżu 41-, d Xokyk rkt snik es FlkMk csdax 1 kMk feykrk g\$
 - (a) rkt k nijk dspH eku dks6 l scny dj FkkNk {kij k, D; kacuk nsck g\$\]
 - (b) bl nik dksngh cuus en vièkd le; D; kn yxrk g\$\square\$
- mÙhj&(a) nùk esacsclax I kNk feykclj {khjh; cuk fn; k t krk gSt rhict nùk v fèkcl I e; rd jg I dst v FkkZ~nùk QVs ught nùk QVus dk rhPi; ZgSnùk dk [kVVk gks t kukt
 - (b), I snikenngh cuusenvièkd le; bl fy, yxrkgSid {kijh, niki gys mnkl hu gkrk gSrc ngh curk gI vr%nik ds yIDVd vEy dks i gys ml enmi fLFkr {kij dksmnkl hu djuk gkrk gSrc ngh curk gI
- iżu 42- vki dkrhu ij [kufy; k nh xbZgM bl enls, d envkl for ty, oa'kk nksenls, d envEyh, foy; u rFkk nh/jsen{kkjh, foy; u gM; fn vki dk doy yky fyVel i= fn; k t krk gSrk vki iR, kd ij [kuyh enj [ksx; s inkFkZdh igpku dN s djaks|
- miligity ky fy Vel i= clks ckjht ckjh ls rhuka i j [kufy; ka ea Mkyrs ga t ks foy; u yky fy Vel i= clks uhyk clj nsrk ga og {kkjh; foy; u ga vc uhyk gq fy Vel i= clks ckjht ckjh ls 'kk nksi j [kufy; kaeaMkyrsga t ksfoy; u uhyk fy Vel i= clks yky clj nsrk ga og v Eyh; foy; u ga 'kk cpk foy; u vkl for t y ga bl ea yky, oa uhys fy Vel i= i j clkb Zi Elko ugha i Merk ga

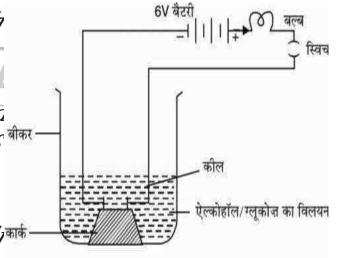
iżu 43-, YdkgkW, oa Xyrdkt tSs; kSxdknen HhgkbMktu gkrsg\$ ysdu budkoxkHj.kvEy dhrjg ughgkrkgS, dfØ; kdyki}kjkblslkcr dja

mÙhj&, Ydhgy, oa Xyndht ty en ?hyus i j ghbMhtu vh; uhn(H+) ds: i en vFhht~; s vh; uhn en fo?hVr ughnghrs ghl bl fy, buds ?hhy fo/qr dh

pkyu ughidjrsg**a**

blslker djusdsfy, fp=kufkj midj.kkadksltk, ktkrkg&chdjes vYdkgy dk?kky ysrsg&blesfo/g èkljk i pkgr dhtkrhg&geikrsg&fo cYc ughat yrkg&bllsfl) gkrkg& fd, Ydkgy dk?kky fo/gr dkpkyu ughadjrkg&

ge chdj esiXyvdkt dk?kky ysdj**** izkx dksngjkrsg\$1 cYc fQj Hhugh tyrkg\$1 bl l sfl) gkrkg\$fd Xyvdkt



Page No. 20, Fig. 2.2

dk?Msy Hh fo/gr dk pkyu ughadjrk g\$ vr%, Ydkgy rFkk Xyvdkt t\$ s; k\$xdknengkbNkt u gkrsgg Hh budk oxhdj.k vEy dh rjg ughagkrk g\$ i zu 44- D; k gkrk g\$t c&%

- (i) rugl Y¶; hjd vEy dh vfHhO; k t Lrk l sghrh g\$
- (ii) ruggloMoykijd vEy dh vfHo; k eStulf'k, e/l sgloth gSl
- (iii) rugl Y¶; kjd vEy dh vfHkO; k , Y; kjefu; e/l sgkrh g\$\forall 1
- (iv) ruggloMkDyksjd vEy dh vfHkO; k ykgk I sgkrh g\$
- (v) rugl Yy; fijd vEy eaBh 1 hM; e dhchas feyhrsgs
- (vi) rugl Y¶; hjd vEy] nhunkj ft ad ds 1 kFk vfHhO; k djrk g\$
- mlkj&(i) ruq1 Y¶; kjd vEy dh vfkkØ; k t Lrk 1 s djk, h t krh gSrksft æd 1 YQ\$V rFkk gkbNkt u x\$\square\$ curk g\$

$$Zn + H_2SO_4 \longrightarrow ZnSO_4 + H_2\uparrow$$

(ii) ruqgkbMkDyksjd vEy dh vfHkO; k t c eSkulf'k, e 1 sdjk, h t krh gSrkseSkulf'k, e DykjkbM rFkk gkbMkt u xS curk gSl

$$Mg + 2HCI \longrightarrow MgCl_2 + H_2\uparrow$$

(iii) ruql Y¶; hjd vEy dh vfHhO; k , Y; hefu; e 1 s djk; h t krh gSrks , Y; Gefu; e 1 YQV rFlk glbMkt u x\$ curk g\$

$$2AI + 3H_2SO_4 \longrightarrow AI_2(SO_4)_3 + 3H_2\uparrow$$

(iv) rugghbMhDyksjd vEy dh vfHsO; k ykgk l sdjk, h t krhgSrksQsjl DykjkoM curk gSrFkk gkbMkt u x\$ epr gkrh g\$

$$3Fe + 6HCI \longrightarrow 3FeCl_2 + 3H_2\uparrow$$

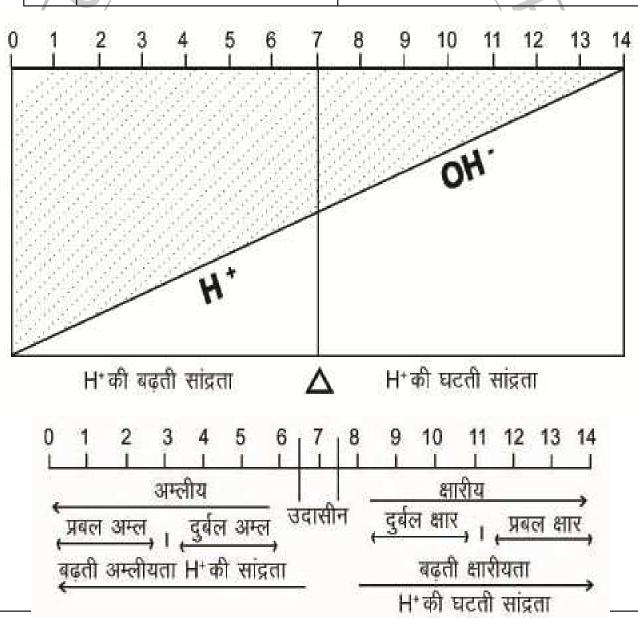
(v) rugl Y¶; kjd vEy eal kM; e dkckWy feykrsg&rks l kM; e l YQV curk gSrFWH2O, oaCO2 curs gS

 $Na_2CO_3 + H_2SO_4 \longrightarrow Na_2SO_4 + H_2O + CO_2 \uparrow$

fofHili foy; i	ika dk p	H eku	fofHeli i nliflæenmi fLFkr vEy
foy; u	-	pH <i>eku</i>	iNVfrd I Mr — VEy
vek'k, jl	_	1	fljdk — ,1 HVd vEy
uhrwj1/ (5)/	_	2.5	
fljdk	_	3.0	1 arjk
VekVj jl	_	4.1	beyh — VKVIjd vEy
il huk	_	4.5	
vEy o'WZ	_	5.6	VekVj — vktt Syd vEy
ishc	_	6	[kVVk nvk 4agh4 y\$DVd vFy
nvk	_	6.5	
'A ty	_	7	uhew — 1 kbfVet vEy
Vkl W	_	7.3	phVh 4usVy12dk Mad — eshubod 4Qhbe2l12vEy
	_	7.4	
fi Ük	_ '	7.5–7.8	1.c — elsyd vEy
puk t y		11.0	ve: n — vlut fyd vEy
ykj 4kusdsig) ykj 4kusdscka uhewjl	1/5/2-	8	eD[ku — C, Vhjd vEy
ykj 4 kkus as chi	12-	6	GIIV9
	7.0	10	pk — Vsud vEy
jax jigr iş	_		olk — LVI; fjd vEy
Xkt j dk jl	_	6	
		5	e/kgD[kh dk Mkd— eFkukbd vEy
VekVj dk jl	_	4	I; kt — , Idkktzl vEy
uy dkty		8	, 2012/2017/29
1 M NaOH		14	
1 M HCI		0	
		.0 /37.8	
feld vill exus	K, K	10.5	

iżu 45- dkolud vly vkj vdkolud vly envarj Li "V dja mlij&dkolud vly vkj vdkolud vly enfulufyf[kr varj gak%

S.No.	dichid viy	vdkokud vky
1.	fl fVet vEy&bl vEy dkmi; kx	xakd dk vEy ; k1 Y¶; kjd vEy&b1
	[kk/ inkfkkidsifjj{k k vkg	vEy dk mi; kx cVjlf jl k; fud
	Lokin"Brk dsfy, gkrk g\$	[kkn] isV] fMVjtsV] gkbMkDykGja
		vEy vkin dsiuekik esimi ; kixh gsi
2.	, fl fVd vEy&fl jdk ds: i	glombykjd vey&clik e I kQ
	es v pkj dks [kVVk cukus ds	djus PVC dsmriknu esiç; Pr
	dhe envhrk ga S	gkrk g\$
3.	VNZjd vEy&csdx ikmNj	
	cukus esiç; Dr gkrk gsl	TNT/Mg ukekbV vkin foLQkVd
		ds mRi knu en gkrk gN
	9/	



jax i fjolkti

S.No.		v <i>Eyl</i> ţ	v <i>Eyh</i> ;
I.	fyVeI	yhy	uhyk
II.	estky vlijte	yky	ihyk
III.	fQukWi FkS/hu	jazghu	xykch
IV.	gYnh	ihyk	yky&H ij k
V.	pqllhj	yky&c&uh	ihyk
VI.	yky xkHh dk i rk		gjk

vkW QSVjh 1 pod D; k g\$ mudsuke fy[ka

mÙlj&dlp , I si nlFllglærsgå ft udhxák v Eyh; r Flk {lkj h; e lè; e esfHlu&fHlu gkrhgA mlgavkW QDVjhl pod dgrsgA tSk:kt/yox dkrsy/oSuykb=A

		jlk, fud 1#
(I) <i>I zvejej</i>	d	CaCO ₃
(II) I kMk ok k	d	Na ₂ CO ₃
(III) okt ax 1 kMk	d	Na ₂ CO ₃ .10 H ₂ O
(IV) fojat d pudZ	d	CaOCI ₂
(V) uhyk Fllefik ½rýr; ki	½ &	CuSO ₄ CO ₃ .5H ₂ O
(VI) csdax 1 kMx	d	NaHCO ₃
(VII) <i>IykLVj vkQ isjl</i>	d	$(CaSO_4)_2.H_2O$; // $CaSO_4.\frac{1}{2}H_2O$
(VIII) dkIVd ikVkI	d	KOH
(VIII) lyklvj vklet isjl (VIII) dklvd ikvkl (IX) lkjk	&	CaNO ₃
(X) yhQax x\$	de	N ₂ O
(XI) ukl knj	d	Na₄CI
(XII) yky fl lhjv	d	Pb ₃ O ₄
(XIII) y\$DVd vEy	d	$C_3H_6O_3$
(XIV) VWZjd vEy	d	$C_4H_6O_6$
(XV) QHEZI VEY YEFKUR	bod v	Ey1/2 & CH ₄₂ O ₂

(XVI)	vk u t Syd vEy	E	$C_2H_2O_4$
(XVII)	QHIQHjd vEy	E	H ₃ PO ₄
(XVIII)	dkchud vEVy	d	H ₂ CO ₃
(XIX)	, IdhcZl vEy	d	$C_6H_8O_6$
(XX)	; hjd vEy	Æ	$C_5H_4N_4O_3$

<i>ja</i> x PH	eku	
(I) Xkk yky (Dark Red)	1A	0
(II) yky (Red)	_	1
(III) Xgjk yky (Dark Red)	-	2
(IV) ukjakh yky (Orange Red)	_	4 VEylt
(V) <i>ukjaxh i hyk</i> (Orange Yellow)	_	5
(VII) gfjr ilyk (Greenish Yellow)	_	6
(VIII) gjk (Green)	_	7 } mnkl hu
(IX) gfjr uhyk (Greenish Blue)	_	8
(X) <i>ul</i> y/k(Blue)	-	9
(XI) gfjr ihyk (Navy Blue)	_	10 Skih
(XII) theath (Purple)	-	11
(XIII) Xkk theyth (Dark Purple)	-	12
(XIV) C&uh(Violet)	_	13-14
(XIII) xkkt keyh (Dark Purple) (XIV) ckuh (Violet)	R.	B. Silva