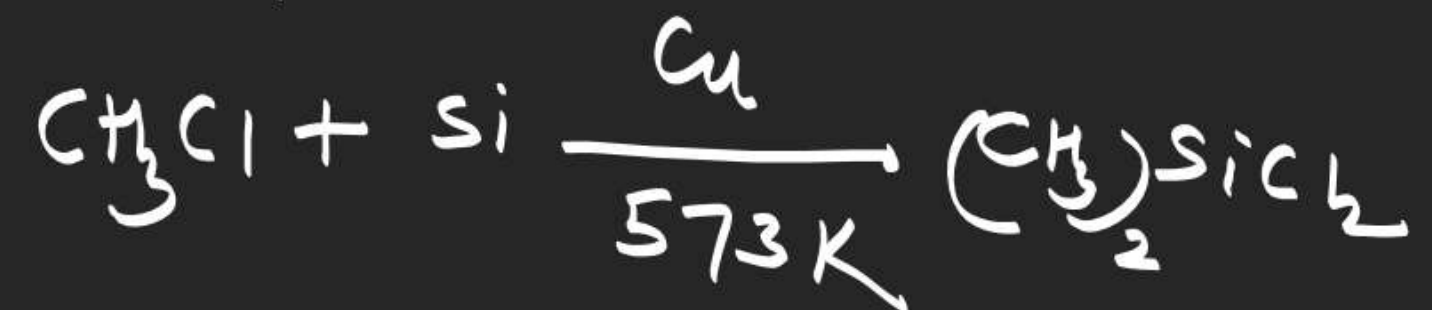
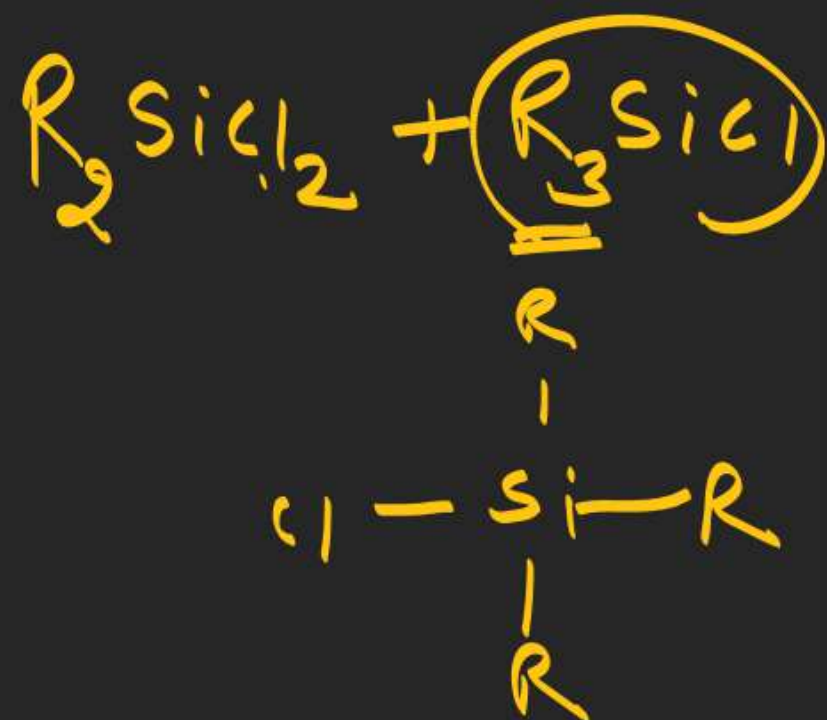
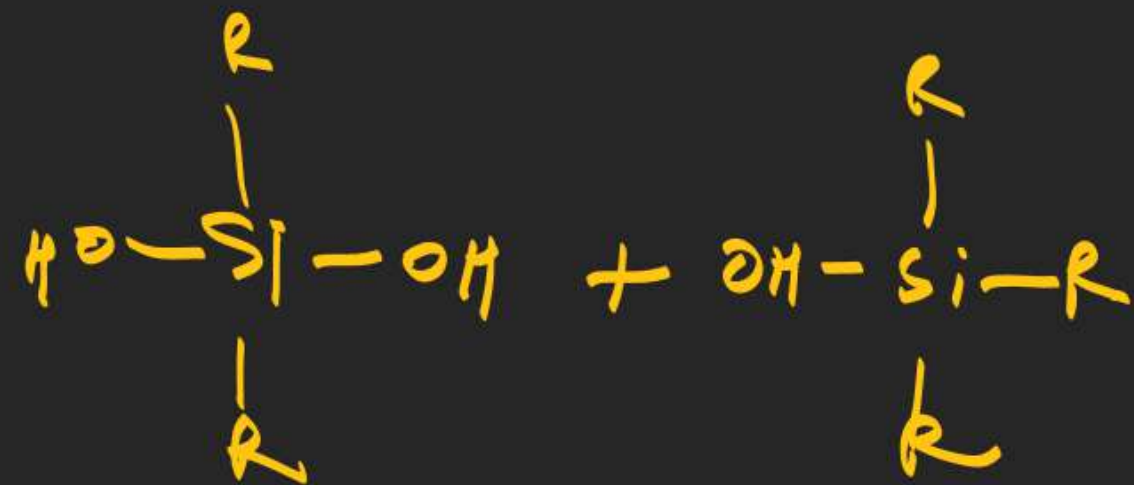


Prep of silicone



Ques Which of the following molecule can act as chain stopping unit for silicones.

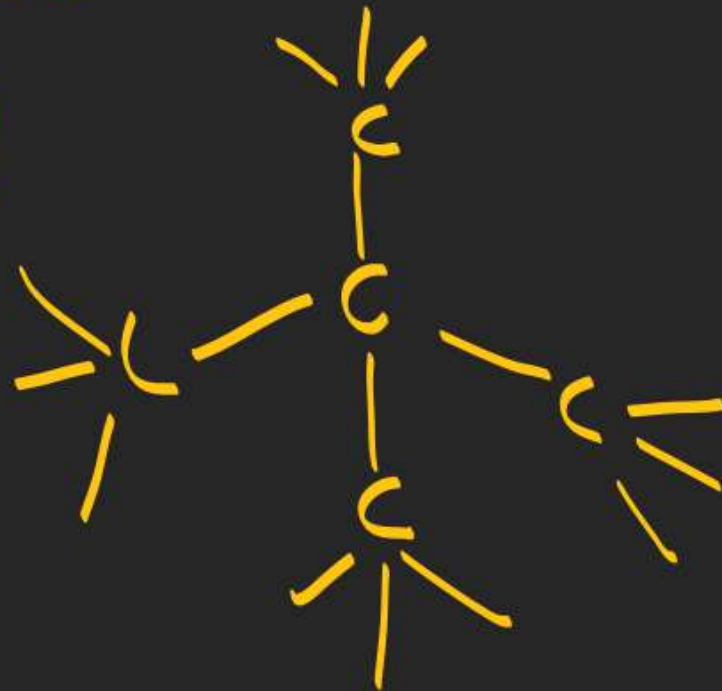
- (1) R_2SiCl_2 (2) RSiCl_3
~~(3)~~ R_3SiCl (4) SiCl_4



allotropes of C

← Diamond
graphite
fullerene

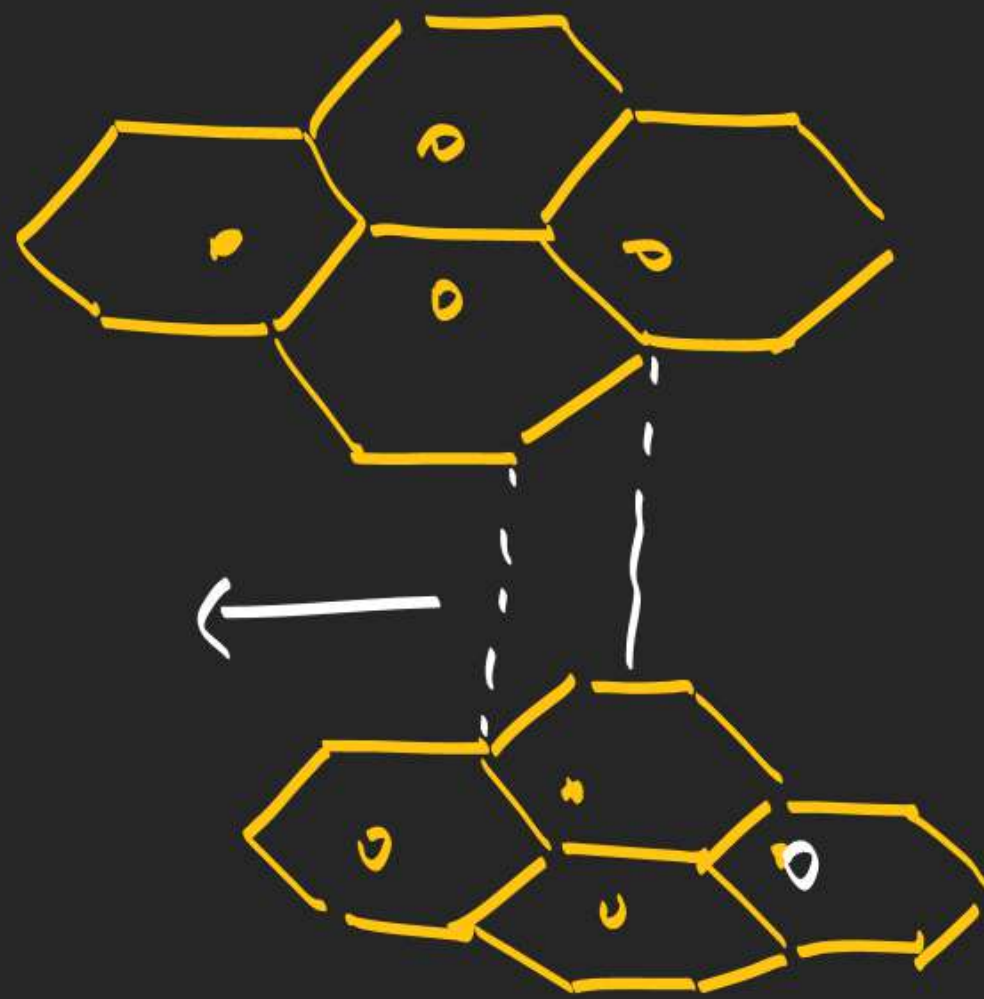
(i) Diamond



sp^3
tet
non planar
non conductor

graphite

guy



V.W.F

(Vander
Waal's
force)

(TIF)
non metallic covalent
bond is present within
layer of graphite

sp^2 (T)

planar

aromatic

Conductor

Hexagonal planar 2D sheet

like structure.

60 \Rightarrow

sp^2
aromatic

20 Hexagonal Rings

12 - five memb. Rings [fix]
(Pentagonal Rings)

each Hexagonal Rings fused with Pentagonal
and Hexagonal.
and each Pentagonal Rings fused with only
Hexagonal.

Fullerene has
no dangling bond.

find the no of Hexagonal Rings
in C_{80}

$$\frac{q}{2} = n + 10$$

q = total no of Carbon

n = Hexagonal Rings

$$\frac{80}{2} = n + 10$$

$$\underline{n = 30}$$