

All  $\text{Al}_2\text{O}_3 \cdot 2\text{H}_2\text{O}$  [bauxite]

① crushing

② conc. → leaching

bauxite  $\xrightarrow{\quad}$  Red bauxite having impurities of  $\left. \begin{matrix} \text{Fe}_2\text{O}_3 \text{(maj)} \\ \text{SiO}_2 \\ \text{TiO}_2 \end{matrix} \right\} \text{min.}$

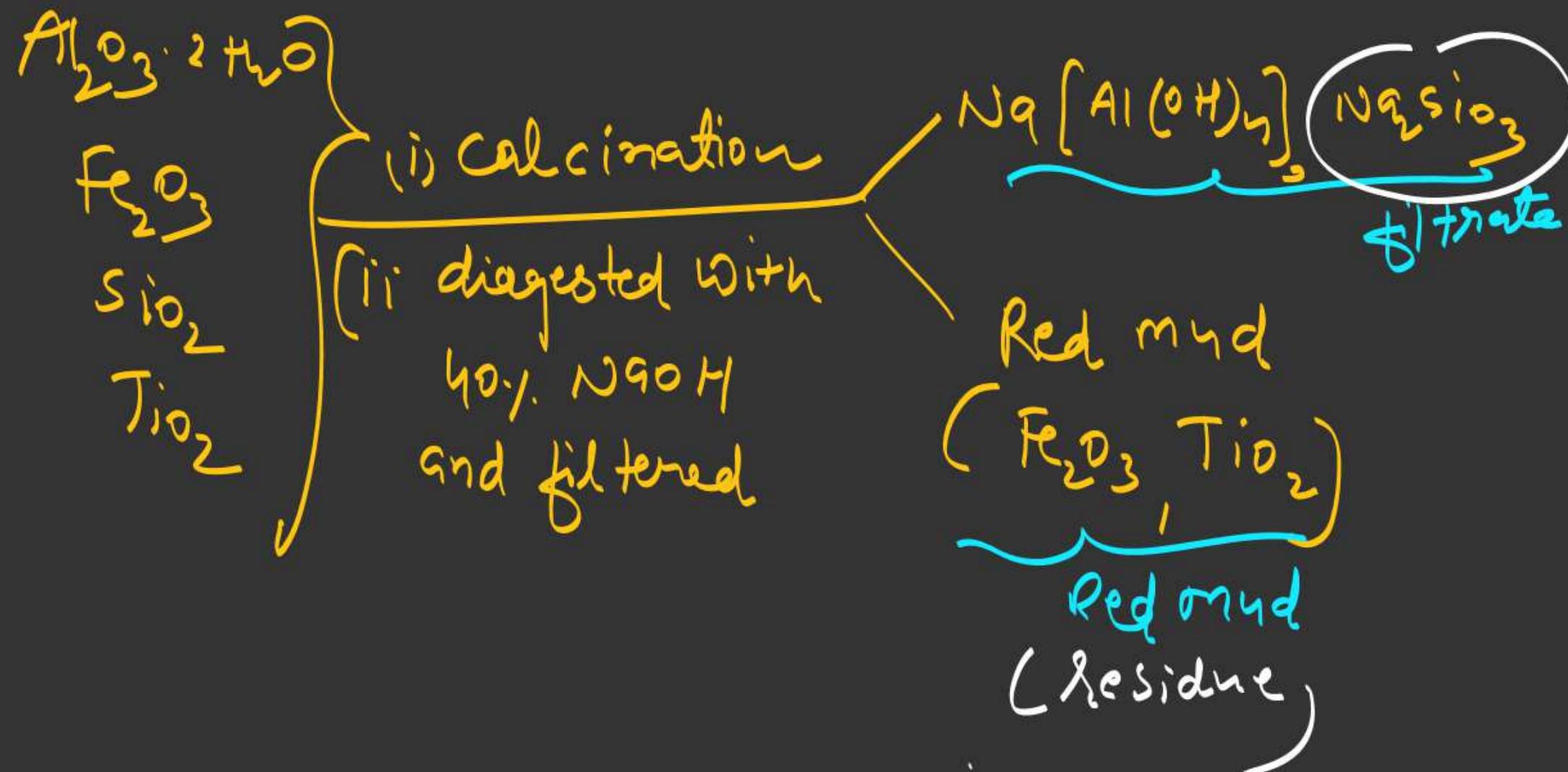
White bauxite having impurities of  $\left. \begin{matrix} \text{SiO}_2 \text{(maj.)} \\ \text{TiO}_2 \\ \text{Fe}_2\text{O}_3 \end{matrix} \right\} \text{min.}$

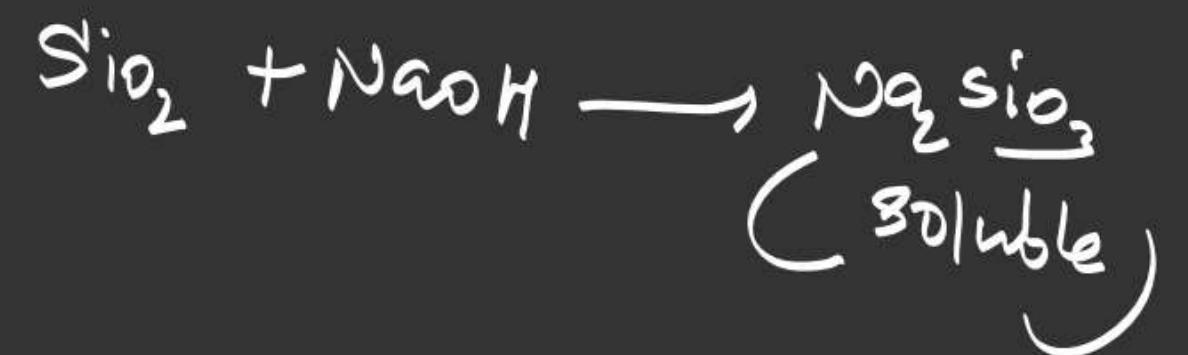
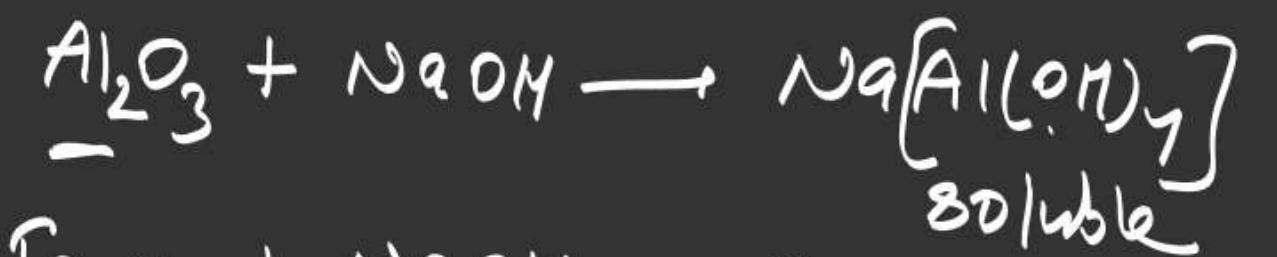
Sorbe K's process

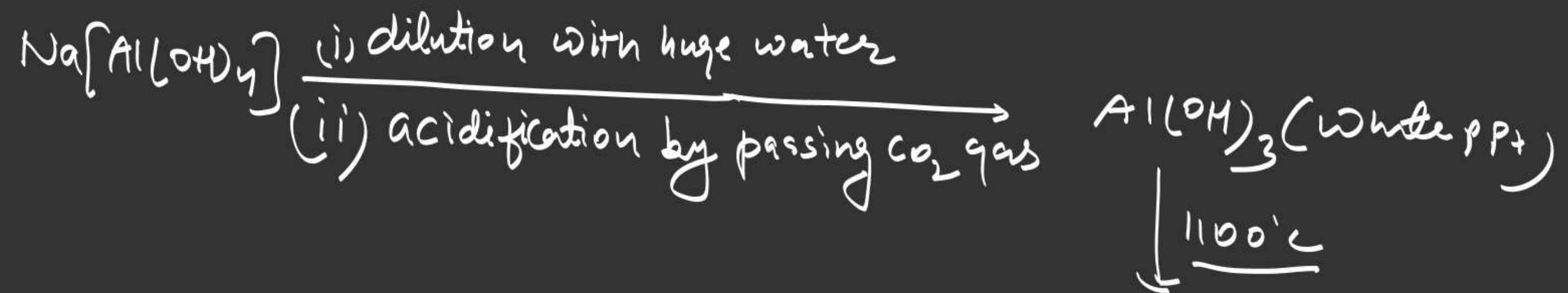
## Bayer's process

① Leaching agent  $\rightarrow \text{NaOH}$

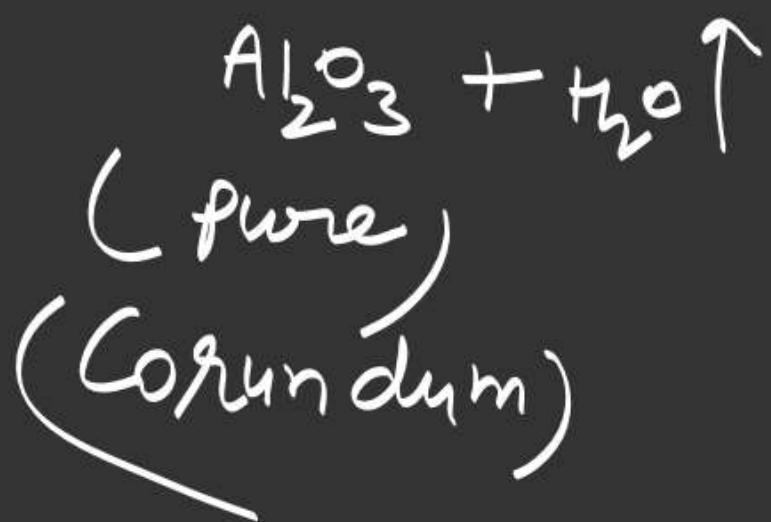
based on Amphoteric nature of  $\text{Al}_2\text{O}_3$





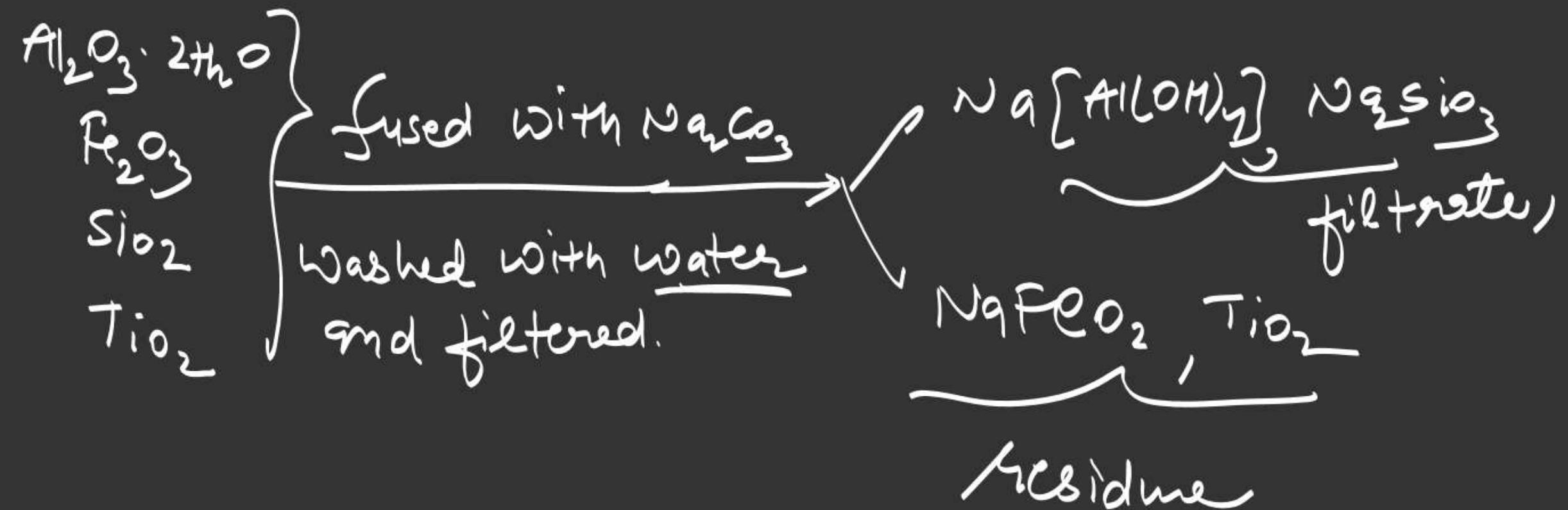


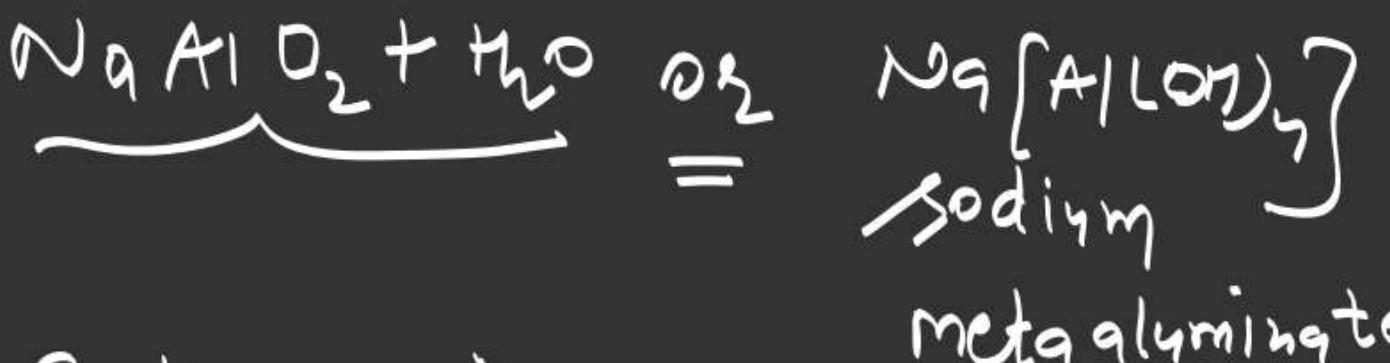
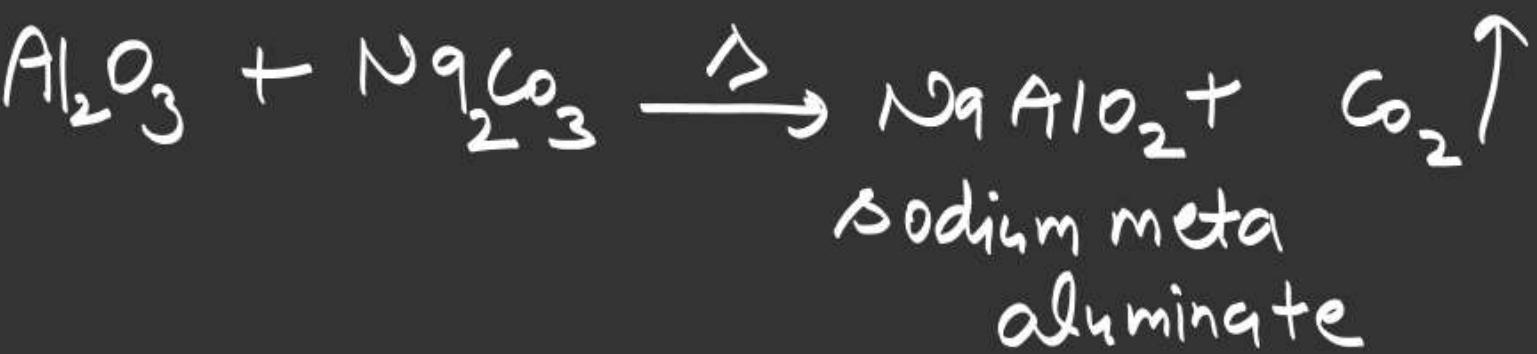
110°C

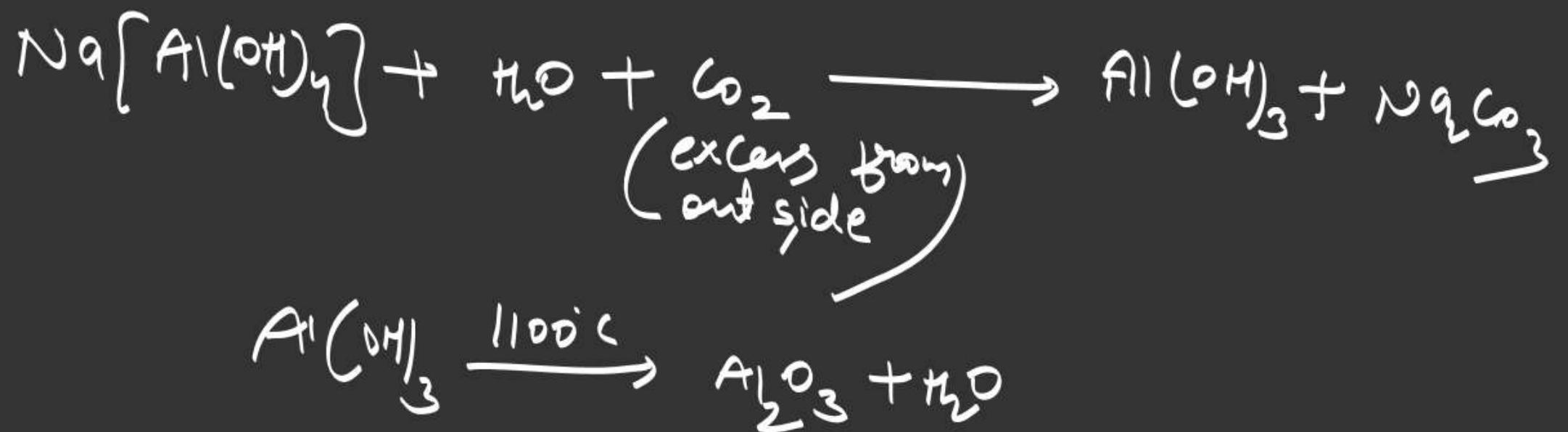




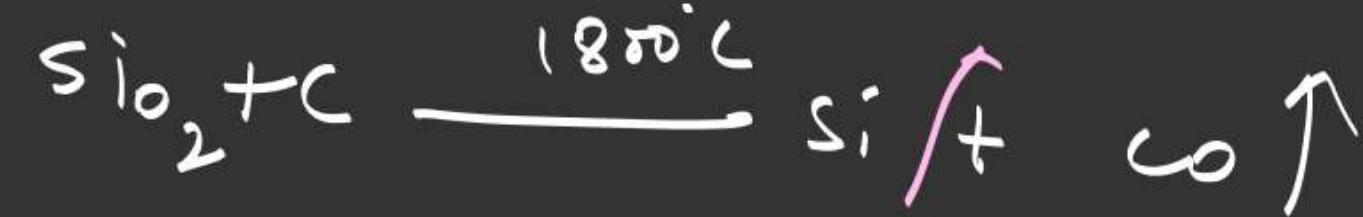
## Hall process







Serpelk's Process

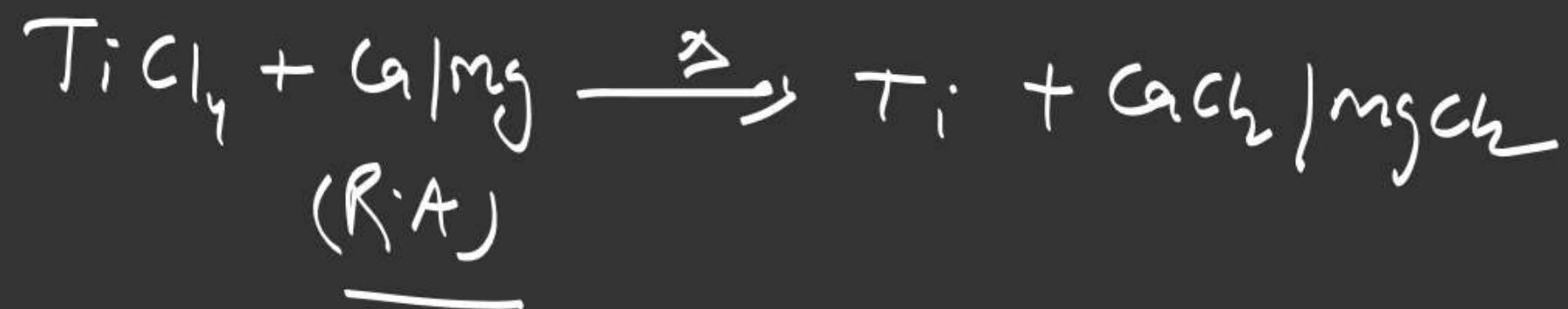


Pidgeon Process → extraction of Mg from dolomite



Reduction



Kroll process

# Electrolytic reduction [Hall-Heroult] process

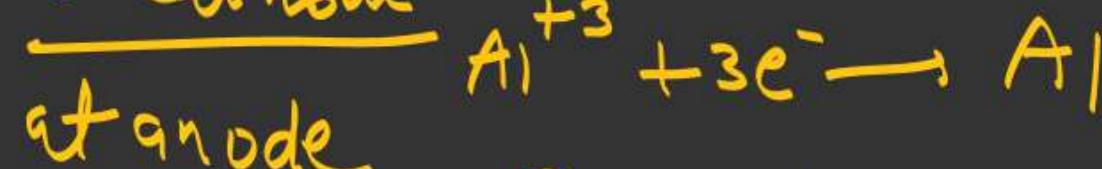
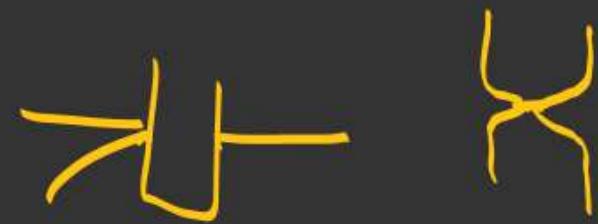


Problems in electrolytic reduction of Al<sub>2</sub>O<sub>3</sub>

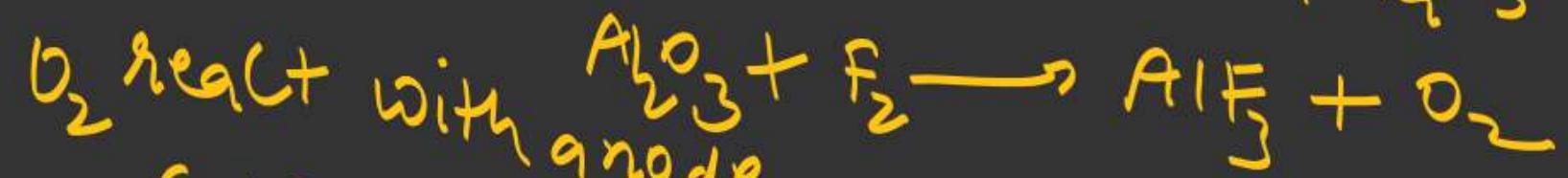
- ① high m.p. of  $\text{Al}_2\text{O}_3$
  - ②  $\text{Al}_2\text{O}_3$  is bad conductor
- for removal these problem  $\text{NaAlF}_6$  and  $\text{CaF}_2$

# Function of $\text{Na}_3\text{AlF}_6$ and $\text{CaF}_2$

- ① ↑ conductivity of  $\text{Al}_2\text{O}_3$   
 ② ↓ m.p. of  $\text{Al}_2\text{O}_3$
- 



$\text{F}_2$  further react with  $\text{Al}_2\text{O}_3$



## function of layer of coke powder

- ① it prevents corrosion of graphite rod.
- ② it does not allow to escape to Heat

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Purification

Hoops

