
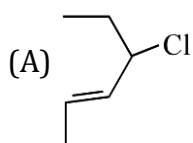


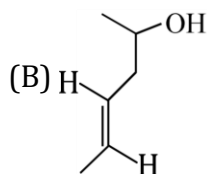
SOLUTION DPP-02

Link to View Video Solution:  [Click Here](#)

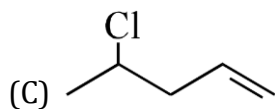
1. Optical isomers are those which differ in their optical activity
Geometrical isomerism are those having restricted rotation and different group on terminal carbon.



- Show geometrical isomerism
- Show optical isomerism

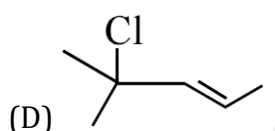


- Show geometrical isomerism
- Show optical isomerism



Not Show geometrical isomerism.

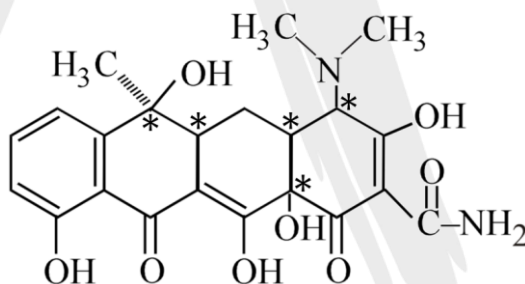
Show Optical isomerism.



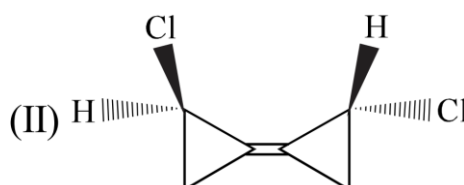
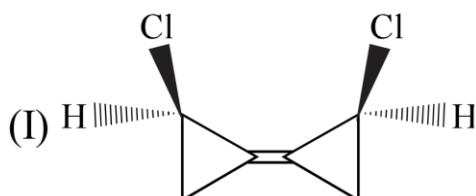
Show geometrical isomerism.

Not Show Optical isomerism.


2. Tetracycline: - Total chirality Centre = 5



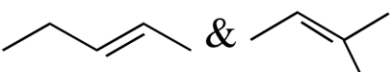
3. For the given compounds, select the correct statements :

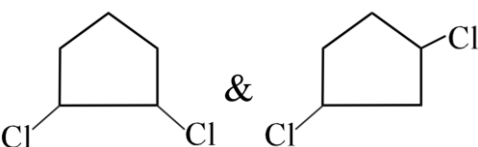


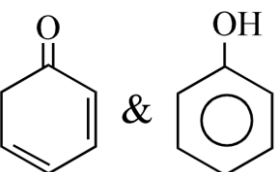
- (A) They are not optically inactive
- (B) Not enantiomeric
- (C) I is polar and II is non polar
- (D) Compound is chiral

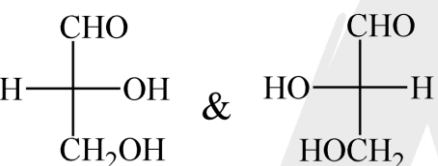
Link to View Video Solution:  [Click Here](#)

4. Which of the following statements are correct:

(A)  (Chain isomer)- They are Chain isomer

(B)  (Position isomer)- They are Position isomer

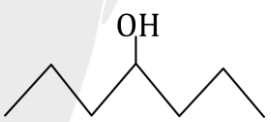
(C)  (Tautomer's)- They are Tautomer's

(D)  (Enantiomer)- They are Enantiomer

5. 1-heptanol  -Not Chiral

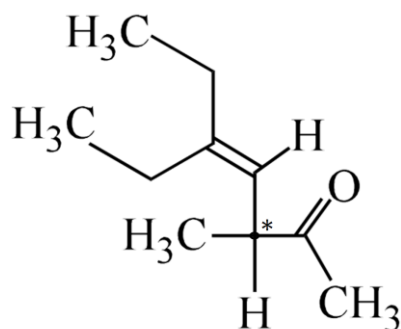
2-heptanol  -It is Chiral

3-heptanol  -It is Chiral

4-heptanol  -Not Chiral

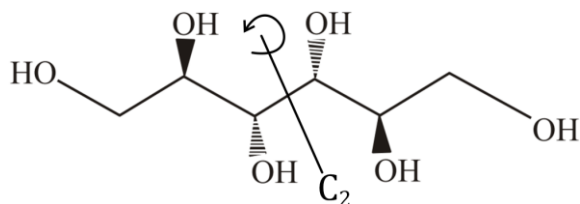
Hence option B is correct.

6. It shows optical isomerism

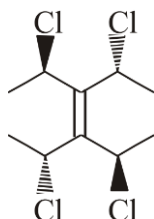


Link to View Video Solution: [Click Here](#)

7. C_2 axis of symmetry



8.



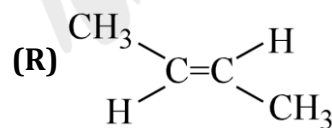
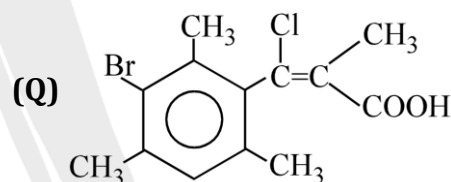
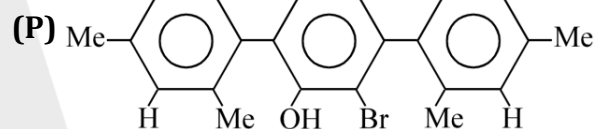
-No Centre of Symmetry


-No Plane of Symmetry

-Optically active

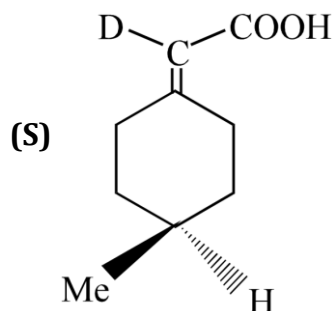
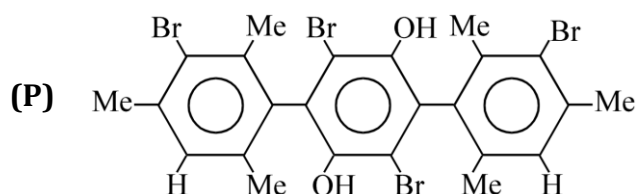
-Cannot show E/Z as Group attach to Double Bond are same

9. (A) Compound show geometrical isomerism

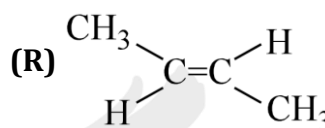


Link to View Video Solution:  [Click Here](#)

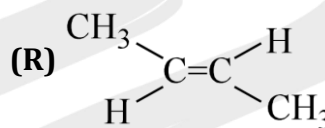
(B) Compound shows optical isomerism



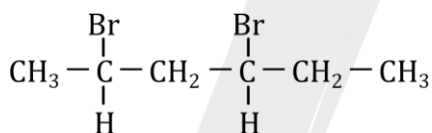
(C) Compound having plane of symmetry



(D) Compound having Centre of symmetry



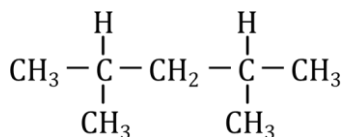
10.



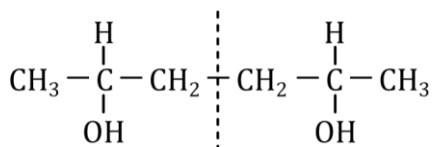
-No Plane of symmetry
Not an Meso compound



-Plane of symmetry present
so a Meso compound.



-No Plane of symmetry
Not an Meso compound



-Plane of symmetry
so these are Meso compound.