

0-1 4-22
 5-1 5-11
 5-11 11

$$\textcircled{20} \quad P_A = \frac{y_A P_T}{2} = 2 \quad \textcircled{A}$$

$$P_B = \frac{y_B P_T}{3} = 3$$

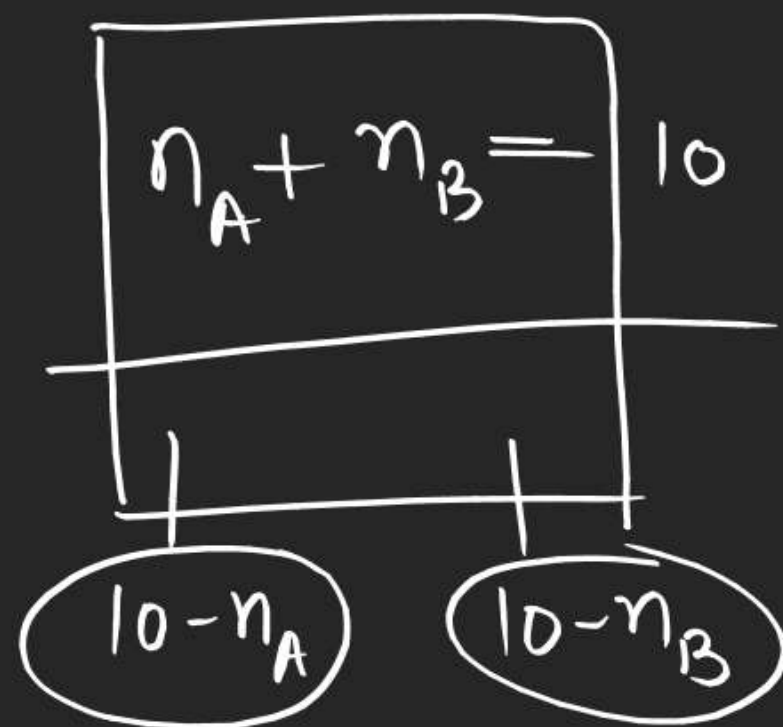
$\textcircled{21}$ Boiling point

$$760 \text{ torr} = 1 \text{ atm} = P_{\text{ext}} = \text{vap. pr.} = P_T$$

$$760 = P_T = x_A \times 900 + x_B \times 360$$

$\textcircled{22}$

P_A°	P_B°
///	///
A	



$$y_A P_T = x_A P_A^0$$

$$\frac{n_A}{10} (P_T) = \frac{10 - n_A}{10} \times 200$$

$$\frac{n_B}{10} P_T = \frac{10 - n_B}{10} \times 100$$

$$\frac{n_A}{n_B} = \frac{10 - n_A}{10 - n_B} \times \frac{2}{1}$$

$$n_A^2 = (10 - n_A)^2 \times 2$$

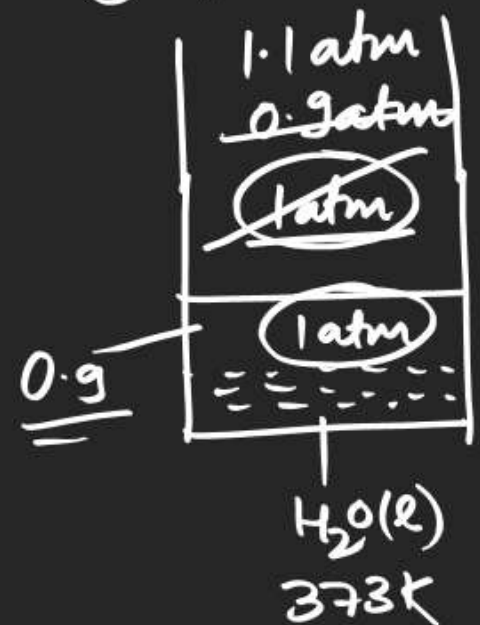
$$n_A = (10 - n_A) \times \sqrt{2}$$

$$n_A = \frac{10\sqrt{2}}{1 + \sqrt{2}}$$

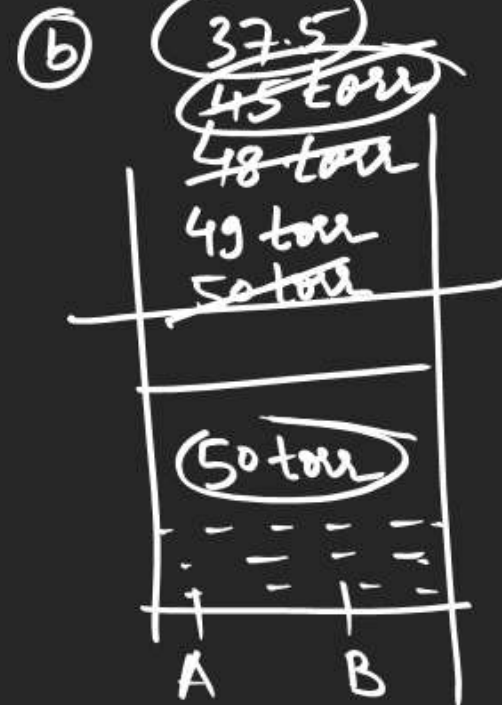
$$n_B = \frac{10}{1 + \sqrt{2}}$$

① By changing external pressure at constant temp

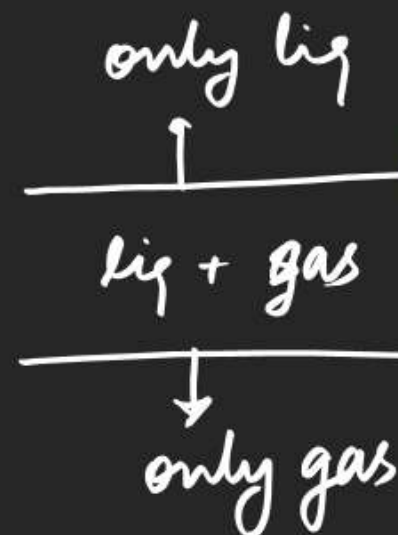
② for pure liquid



$P_{ext} > P_{vap}$
only liq
 $P_{ext} < P_{vap}$
only gas

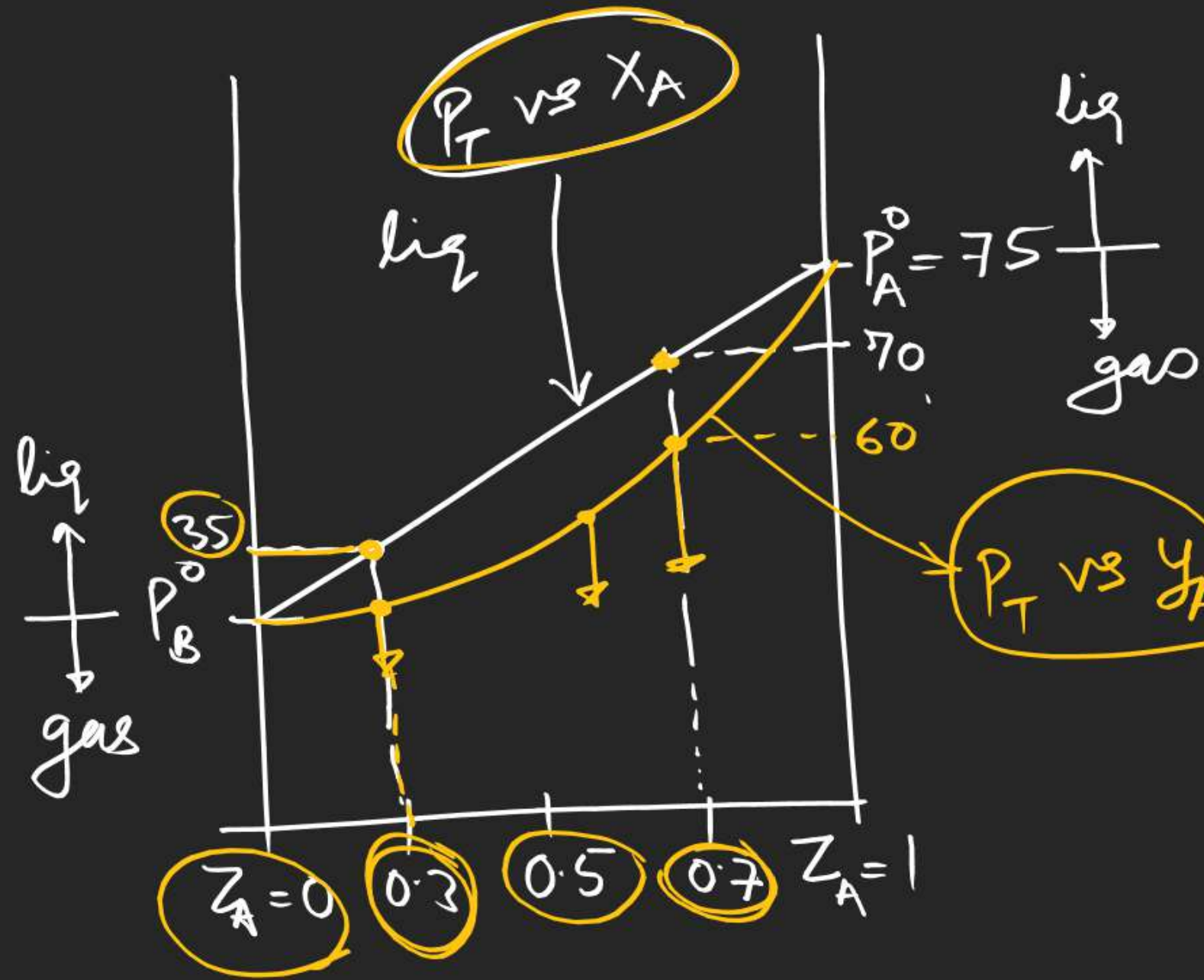


Dew point



$$P_T = X_A P_A^0 + X_B P_B^0$$

$$\frac{1}{P_T} = \frac{y_A}{P_A^0} + \frac{y_B}{P_B^0}$$

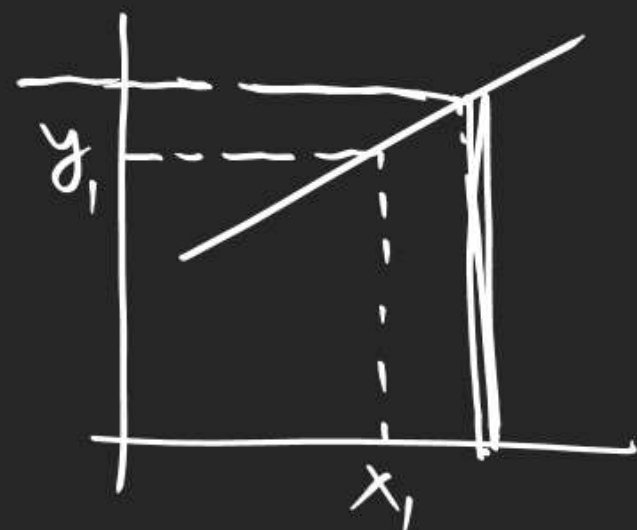


$$P_T = X_A P_A^0 + X_B P_B^0$$

$$\underline{P_T} = \underline{P_B^0} + \underline{X_A} (P_A^0 - P_B^0)$$

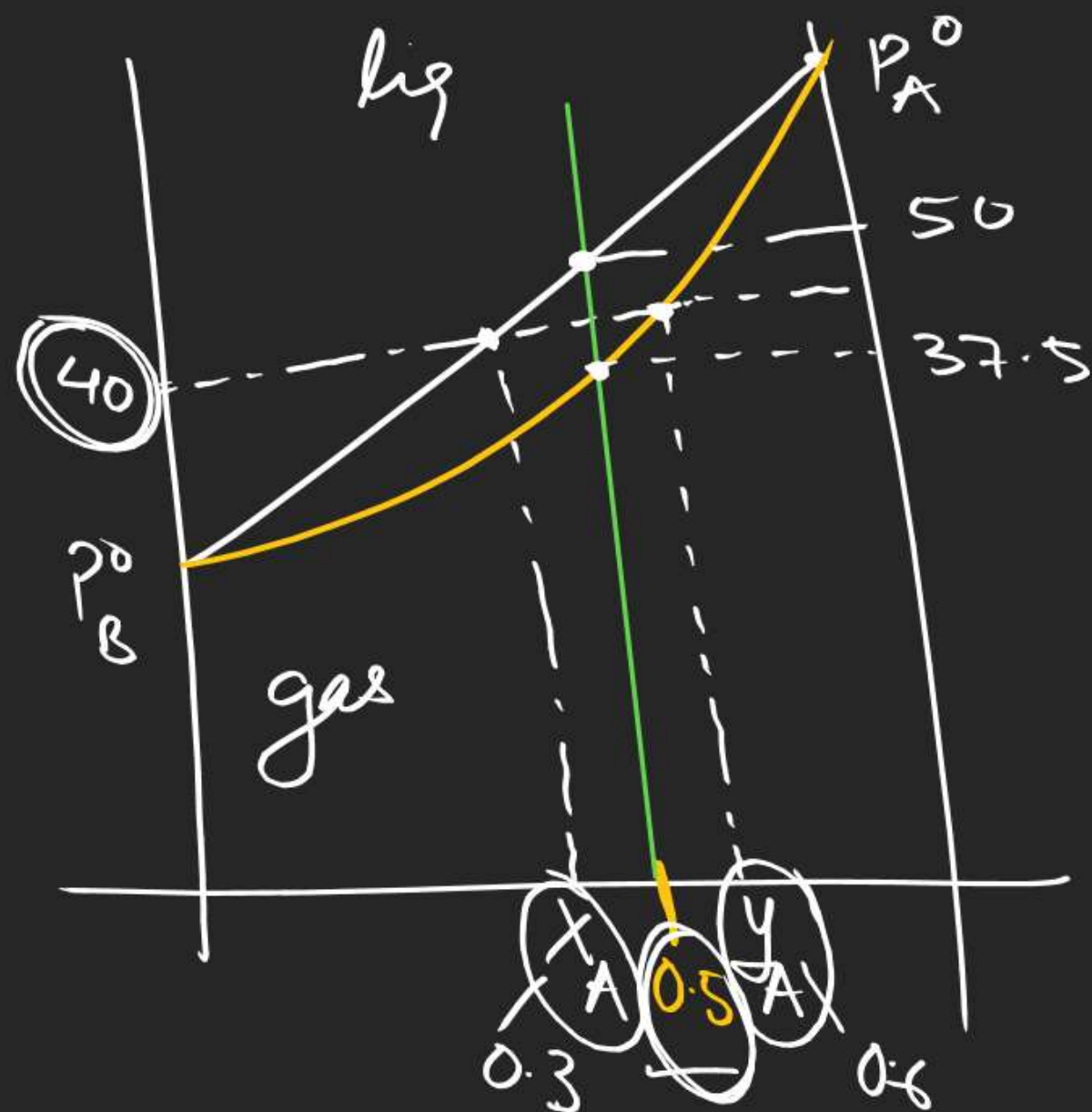
$$\frac{1}{P_T} = \frac{y_A}{P_A^0} + \frac{y_B}{P_B^0}$$

$$y = mx + c$$



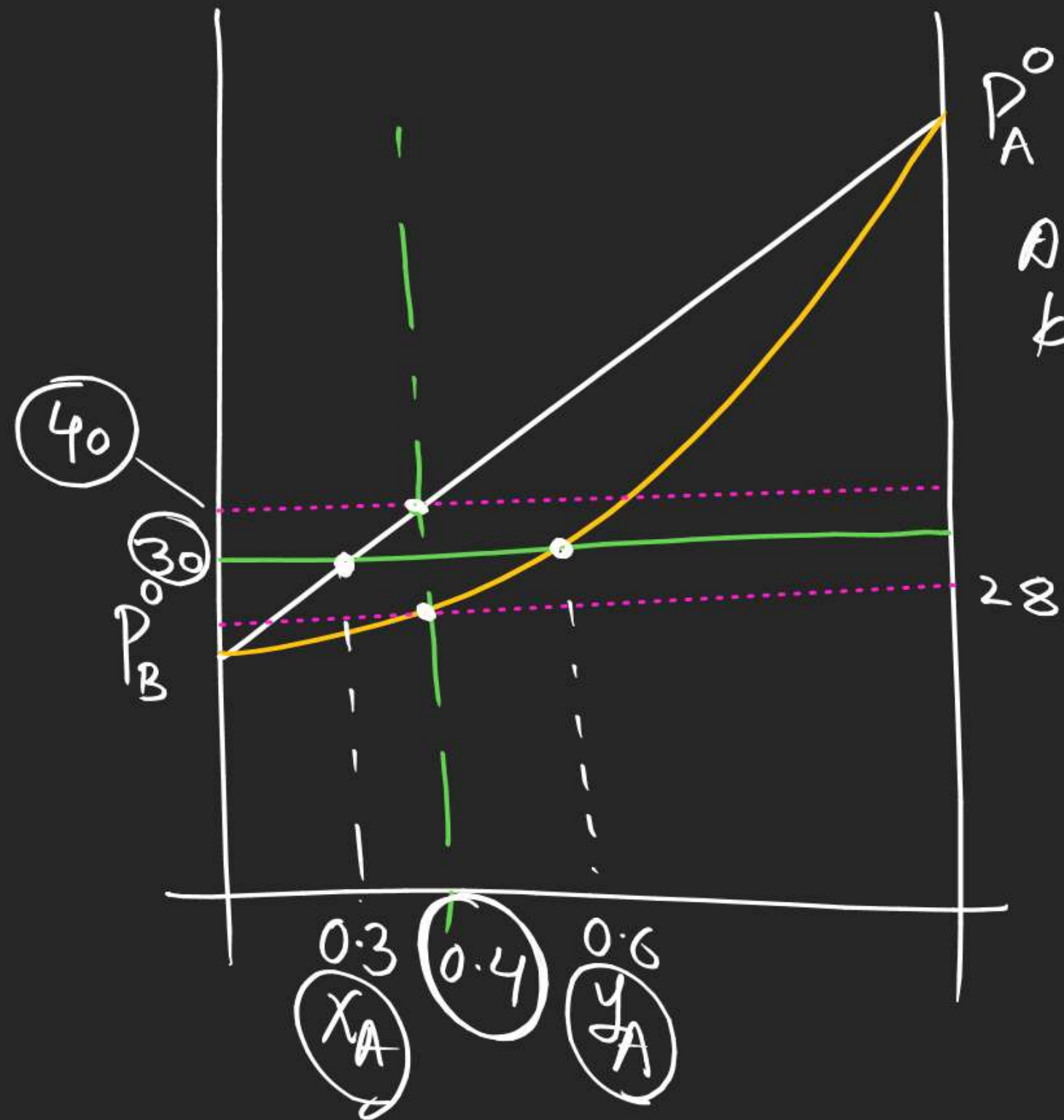
$$\underline{40} = P_T = P_B^0 + \underline{X_A}(P_A^0 - P_B^0)$$

$$\frac{1}{40} = \frac{1}{P_T} = \frac{y_A}{P_A^0} + \frac{y_B}{P_B^0}$$



Initially
 $X_A = 0.5$ $X_B = 0.5$
 at 40 torr

$X_A = 0.3$ $y_A = 0.6$
 $X_B = 0.7$ $y_B = 0.4$



A solution having $X_A = 0.4$ is kept at 30 torr. find

$$X_A = 0.3$$

$$X_B = 0.7$$

$$y_A = 0.6$$

$$y_B = 0.4$$

(b) By changing temp at constant external pressure.

@ for pure liq

Electro
JEE-Adv