



## DPP 05

1. A nucleus moving with velocity  $\vec{v}$  emits an  $\alpha$ -particle. Let the velocities of the  $\alpha$ -particle and the remaining nucleus be  $\vec{v}_1$  and  $\vec{v}_2$  and their masses be  $m_1$  and  $m_2$ , then
  - (A)  $\vec{v}, \vec{v}_1$ , and  $\vec{v}_2$  must be parallel to each other
  - (B) none of the two of  $\vec{v}, \vec{v}_1$ , and  $\vec{v}_2$  should be parallel to each other
  - (C)  $\vec{v}_1 + \vec{v}_2$  must be parallel to  $\vec{v}$
  - (D)  $m_1\vec{v}_1 + m_2\vec{v}_2$  must be parallel to  $\vec{v}$
  
2. In an  $\alpha$ -decay, the kinetic energy of  $\alpha$ -particle is 48 MeV and Q value of the reaction is 50 MeV. The mass number of the mother nucleus is (assume that daughter nucleus is in ground state)
  - (A) 96
  - (B) 100
  - (C) 104
  - (D) none of these
  
3. In which of the following processes, the number of protons in the nucleus increase?
  - (A)  $\alpha$ -decay
  - (B)  $\beta^-$ -decay
  - (C)  $\beta^+$ -decay
  - (D) k-capture
  
4. Atomic masses of two isobars  $^{64}_{29}\text{Cu}$  and  $^{64}_{30}\text{Zn}$  are 63.9298 u and 63.9292 u, respectively. It can be concluded from this data that
  - (A) both the isobars are stable
  - (B)  $^{64}\text{Zn}$  is radioactive, decaying to  $^{64}\text{Cu}$  through  $\beta$ -decay
  - (C)  $^{64}\text{Cu}$  is radioactive, decaying to  $^{64}\text{Zn}$  through  $\beta$ -decay
  - (D)  $^{64}\text{Cu}$  is radioactive, decaying to  $^{64}\text{Zn}$  through  $\gamma$ -decay
  
5. If a nucleus such as  $^{226}\text{Ra}$  that is initially at rest undergoes  $\alpha$ -decay, then which of the following statements is true?
  - (A) The alpha particle has more kinetic energy than the daughter nucleus.
  - (B) The alpha particle has less kinetic energy than the daughter nucleus.
  - (C) The alpha particle and daughter nucleus both have same kinetic energy
  - (D) We cannot say anything about kinetic energy of alpha particle and daughter nucleus.
  
6. Consider one of fission reactions of  $^{235}\text{U}$  by thermal neutrons  $^{235}_{92}\text{U} + n \rightarrow ^{94}_{38}\text{Sr} + ^{140}_{54}\text{Xe} + 2n$ . The fission fragments are however unstable and they undergo successive  $\beta$ -decay until  $^{94}_{38}\text{Sr}$  becomes  $^{94}_{40}\text{Zr}$  and  $^{140}_{54}\text{Xe}$  becomes  $^{140}_{58}\text{Ce}$ . The energy released in this process is  
 [Given:  $m(^{235}\text{U}) = 235.439$  u,  $m(n) = 1.00866$  u,  $m(^{94}\text{Zr}) = 93.9064$  u,  $m(^{140}\text{Ce}) = 139.9055$  u,  $1\text{u} = 931 \text{ MeV}$ ]
  - (A) 156 MeV
  - (B) 208 MeV
  - (C) 456 MeV
  - (D) cannot be computed

7. The binding energies of nuclei X and Y are  $E_1$  and  $E_2$ , respectively. Two atoms of X fuse to give one atom of Y and an energy Q is released. Then,

(A)  $Q = 2E_1 - E_2$       (B)  $Q = E_2 - 2E_1$   
(C)  $Q < 2E_1 - E_2$       (D)  $Q > E_2 - 2E_1$

8. In the disintegration series

$${}_{92}^{238}\text{U} \xrightarrow{\alpha} {}_X^A\text{X} \xrightarrow{\beta^-} {}_Z^AY$$

the values of Z and A, respectively, will be

(A) 92,326      (B) 88,230  
(C) 90,234      (D) 91,234

9. In the case of thorium ( $A = 232$  and  $Z = 90$ ), we obtain an isotope of lead ( $A = 208$  and  $Z = 82$ ) after some radioactive disintegrations. The number of  $\alpha$ - and  $\beta$ -particles emitted are, respectively,

(A) 6,3      (B) 6,4      (C) 5,5      (D) 4,6



**ANSWER KEY**

1. (D)
2. (B)
3. (B)
4. (C)
5. (A)
6. (B)
7. (B)
8. (D)
9. (B)

