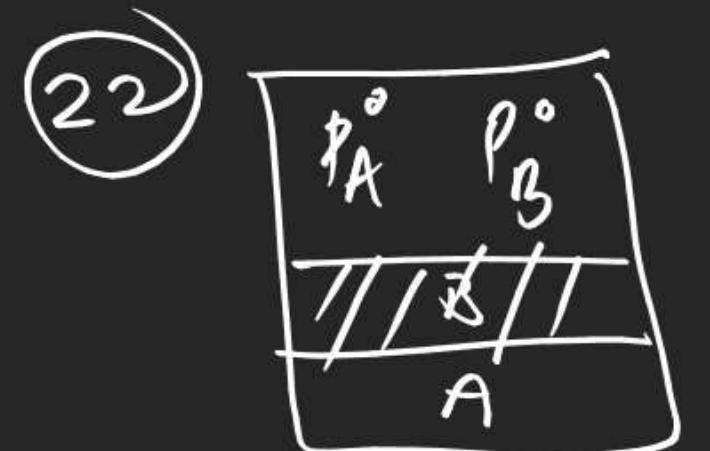


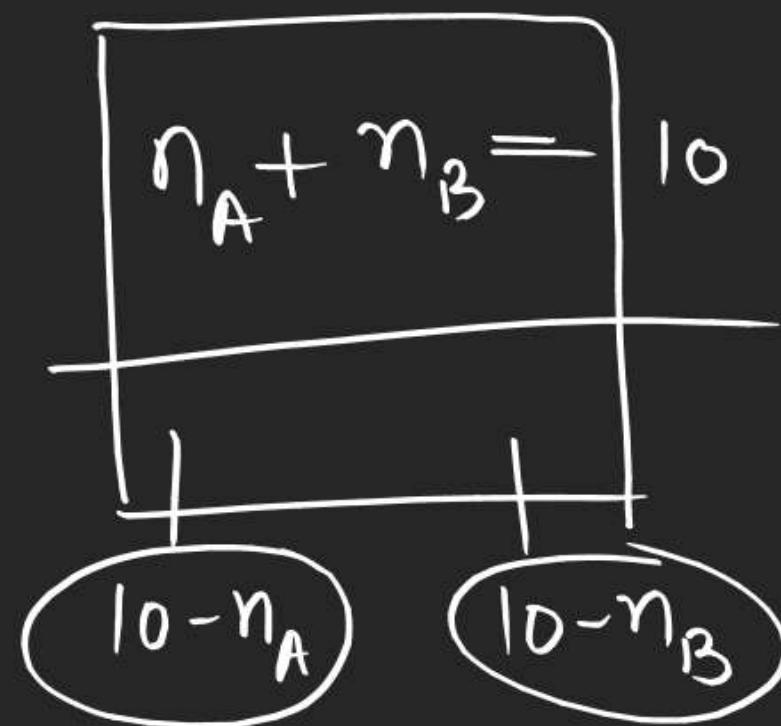
$$\textcircled{20} \quad P_A = \frac{y_A P_T}{y_B P_T} = \frac{2}{3}$$

A

$$\textcircled{21} \quad \text{Boiling point} \\ 760 \text{ torr} = \underline{1 \text{ atm}} = P_{\text{ext}} = \underline{\text{vap. pr.}} = P_T$$

$$760 = P_T = x_A \times 900 + x_B \times 360$$





$$y_A P_T = x_A P_A^0$$

$$\frac{n_A}{10} P_T = \frac{10 - n_A}{10} \times 200$$

$$\frac{n_B}{10} P_T = \frac{10 - n_B}{10} \times 100$$

$$\frac{n_A}{n_B} = \frac{10 - n_A}{10 - n_B} \times \frac{2}{1}$$

$$\gamma_A^2 = (10 - n_A)^2 \times 2$$

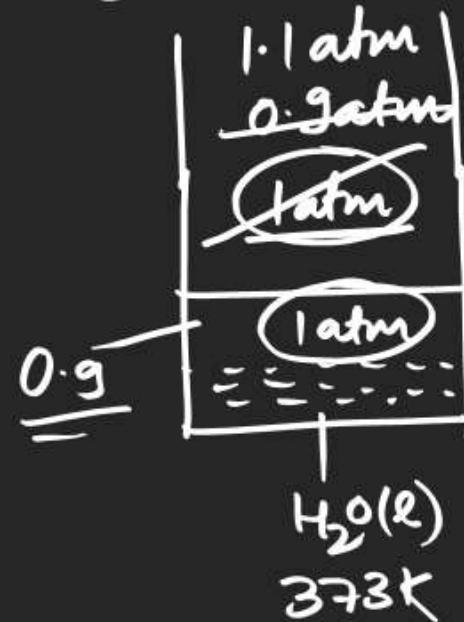
$$n_A = (10 - n_A) \times \sqrt{2}$$

$$n_A = \frac{10\sqrt{2}}{1 + \sqrt{2}}$$

$$n_B = \frac{10}{1 + \sqrt{2}}$$

① By changing external pressure at constant temp

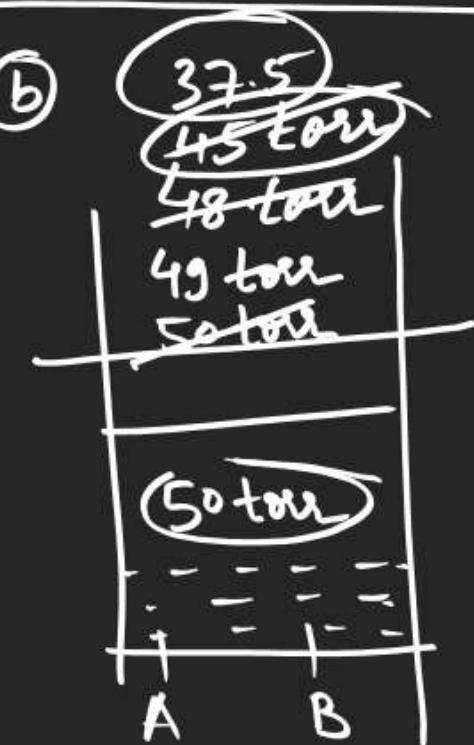
a) for pure liquid



$P_{ext} > \text{vap. pr}$ or
only liq

$P_{ext} < \text{vap. pr}$
only gas

b)



Dew point

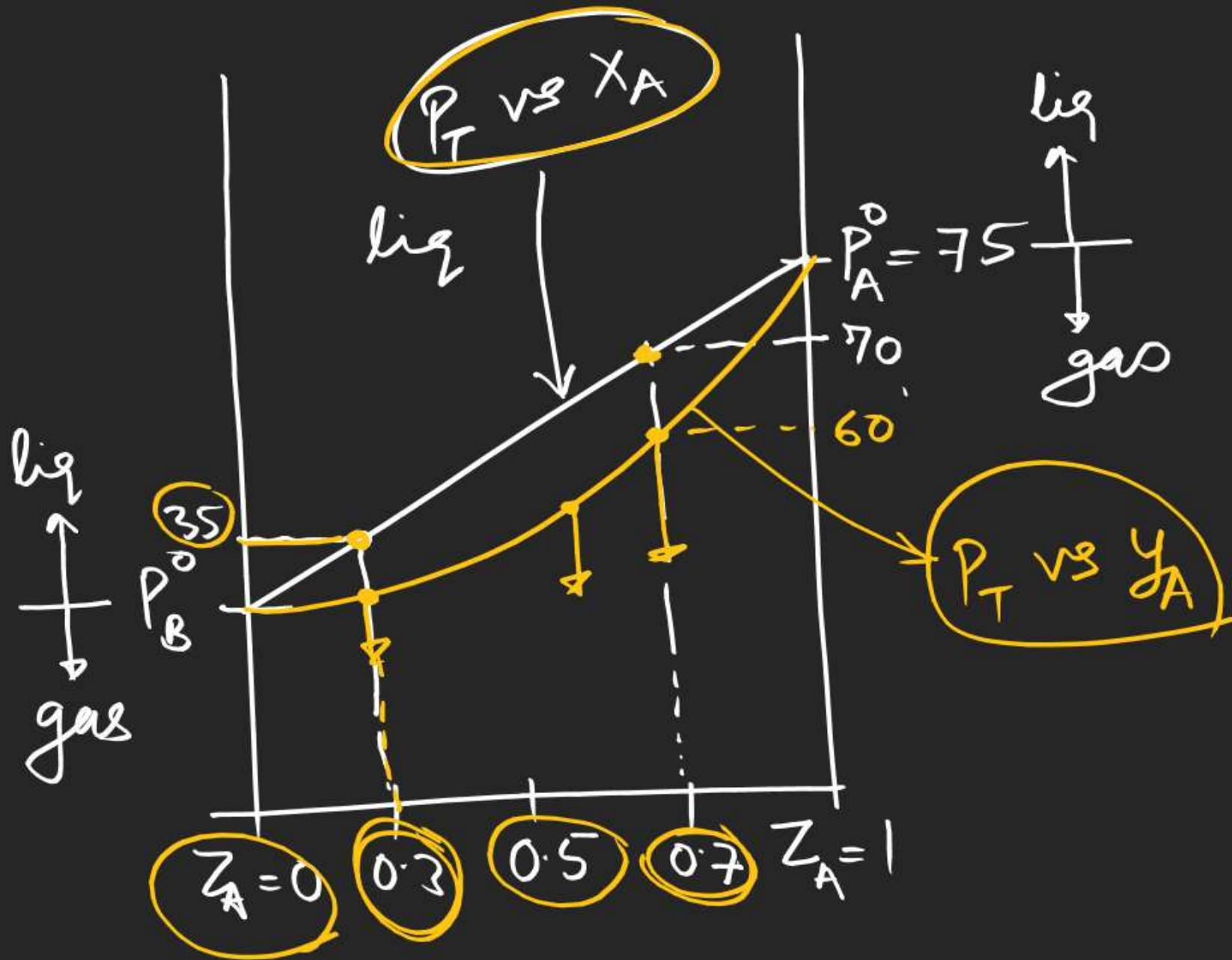
only liq

liq + gas

only gas

$$P_T = x_A P_A^0 + x_B P_B^0$$

$$\frac{1}{P_T} = \frac{y_A}{P_A^0} + \frac{y_B}{P_B^0}$$

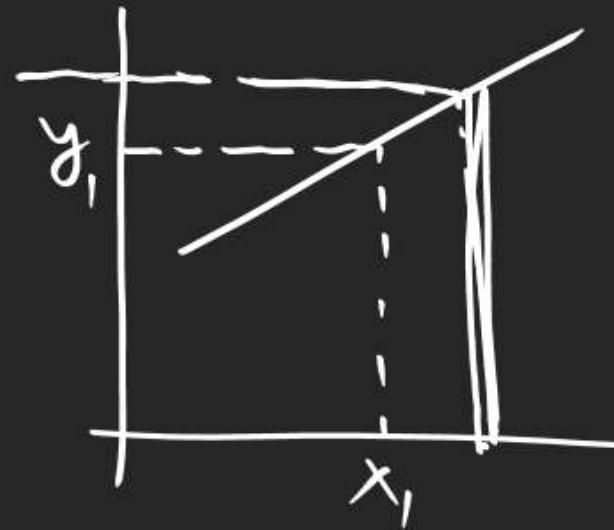


$$P_T = X_A P_A^0 + X_B P_B^0$$

$$\underline{P_T} = P_B^0 + \underline{X_A (P_A^0 - P_B^0)}$$

$$\frac{1}{P_T} = \frac{Y_A}{P_A^0} + \frac{Y_B}{P_B^0}$$

$$y = mx + c \quad \underline{40} = P_T = P_B^0 + X_A(P_A^0 - P_B^0)$$

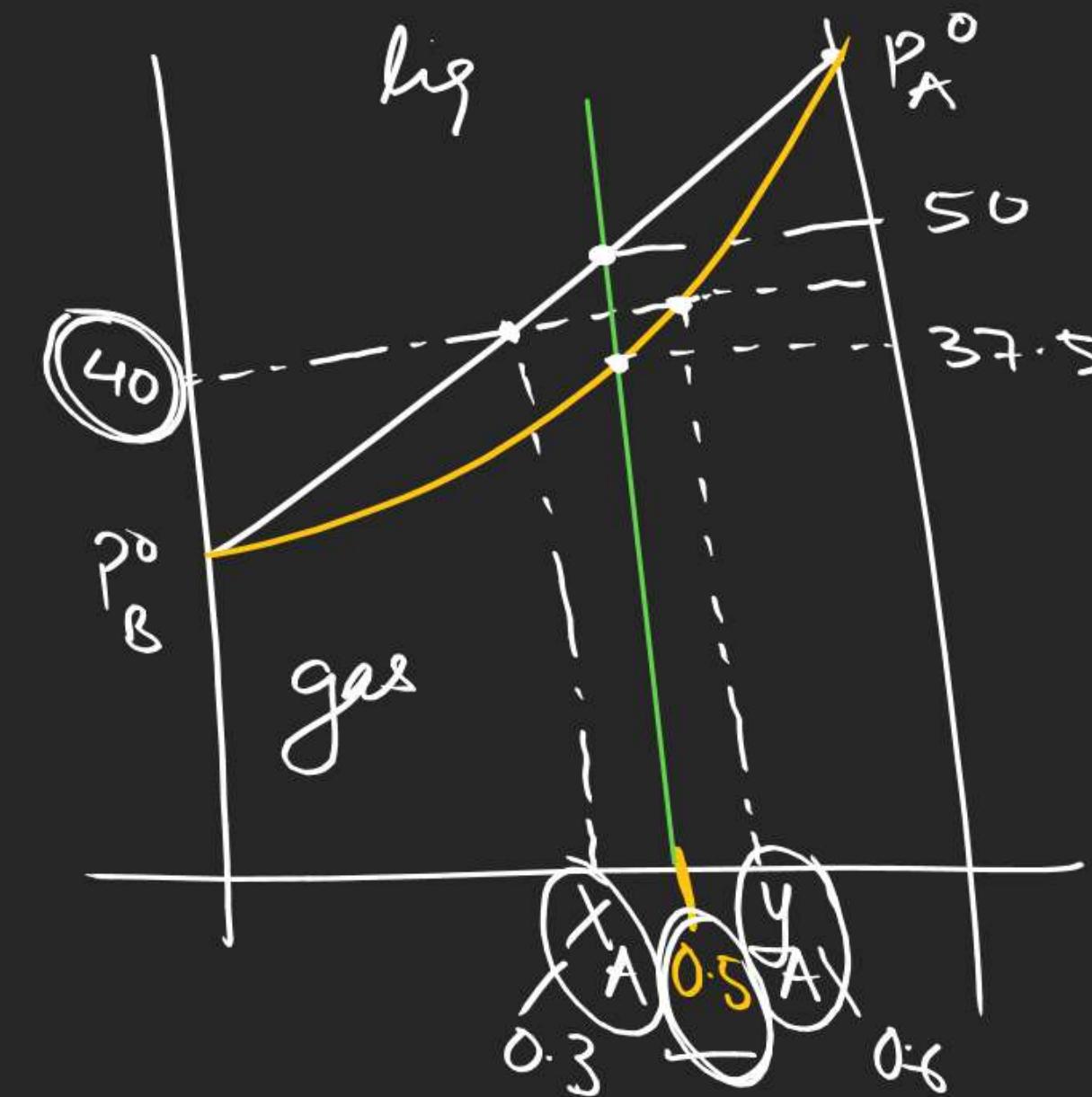


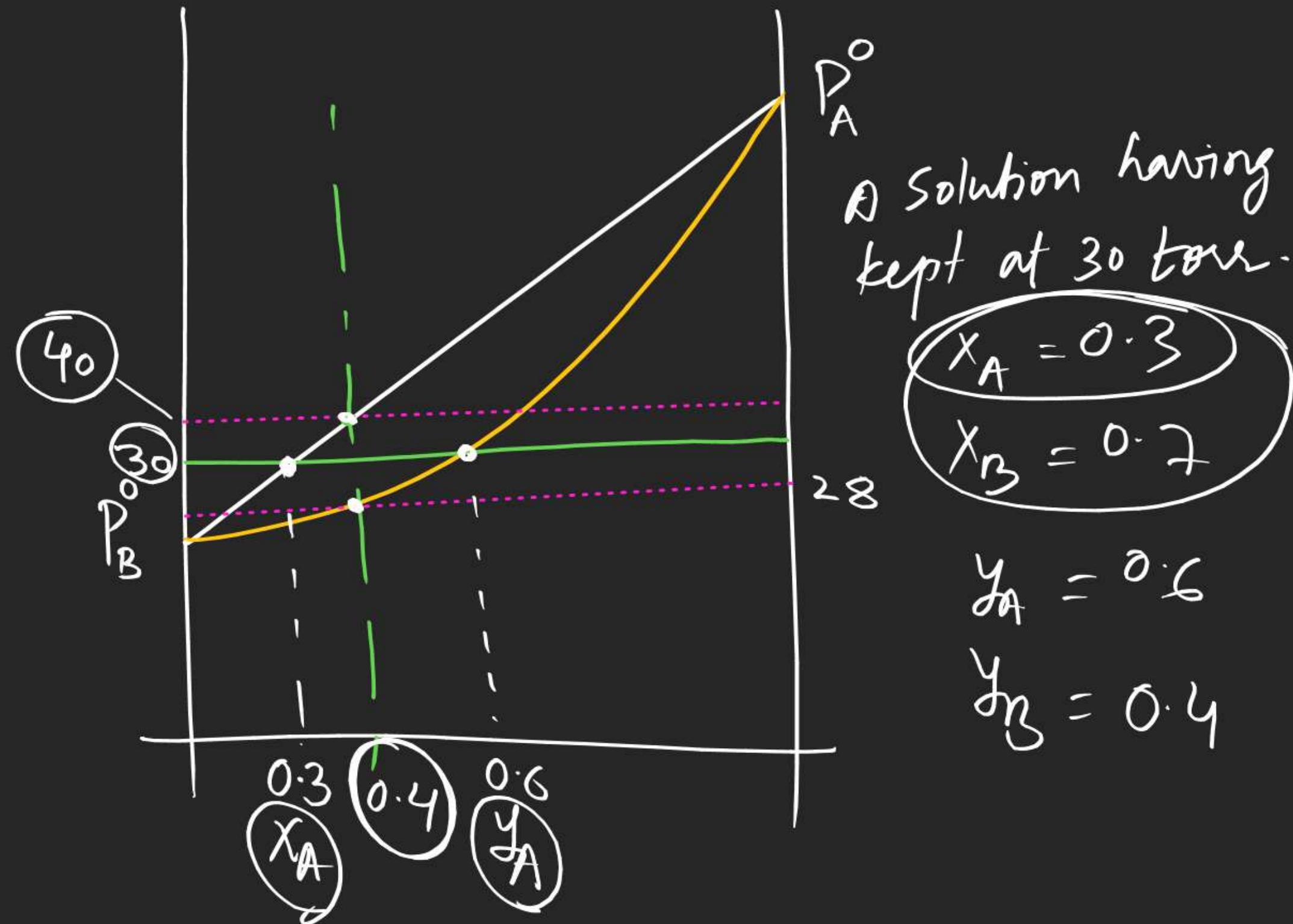
$$\frac{1}{40} = \frac{1}{P_T} = \frac{Y_A}{P_A^0} + \frac{Y_B}{P_B^0}$$

Initially
 $X_A = 0.5 \quad X_B = 0.5$

at 40 torr

$$Y_A = 0.6 \\ Y_B = 0.4$$





⑥ By changing temp at constant external pressure.

⑦ for pure lig

