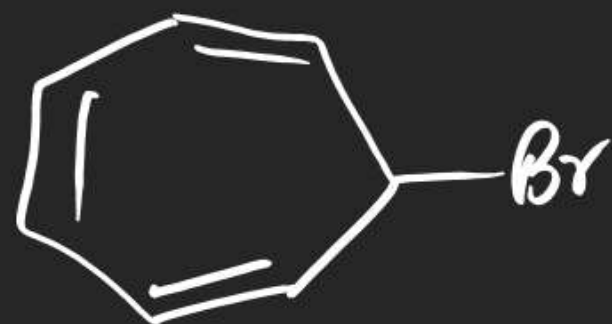
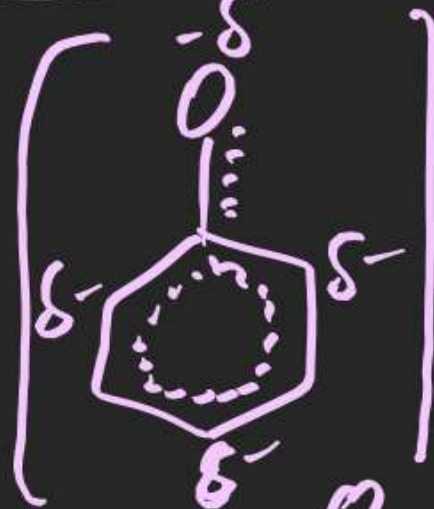


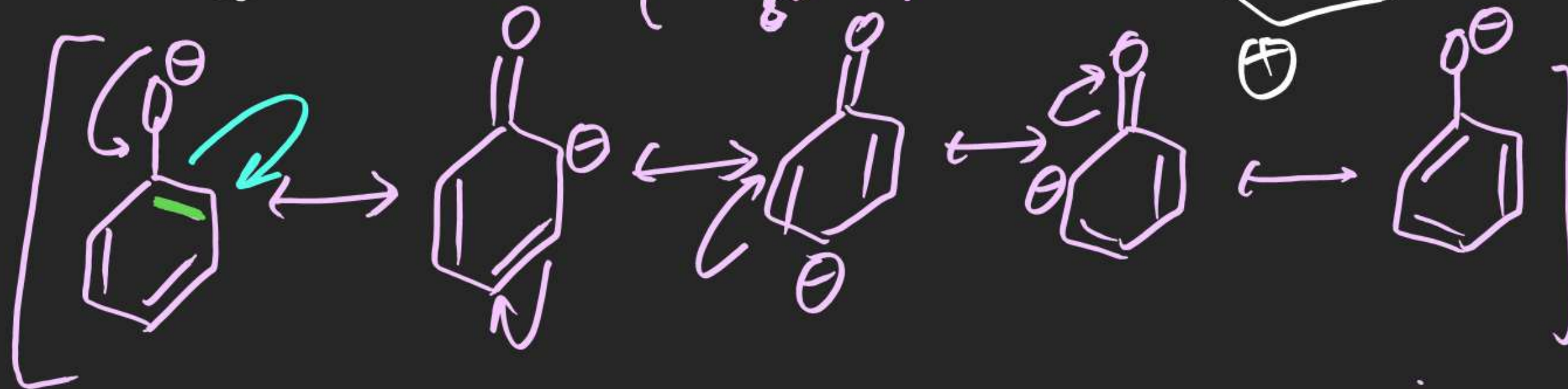
Ex-19

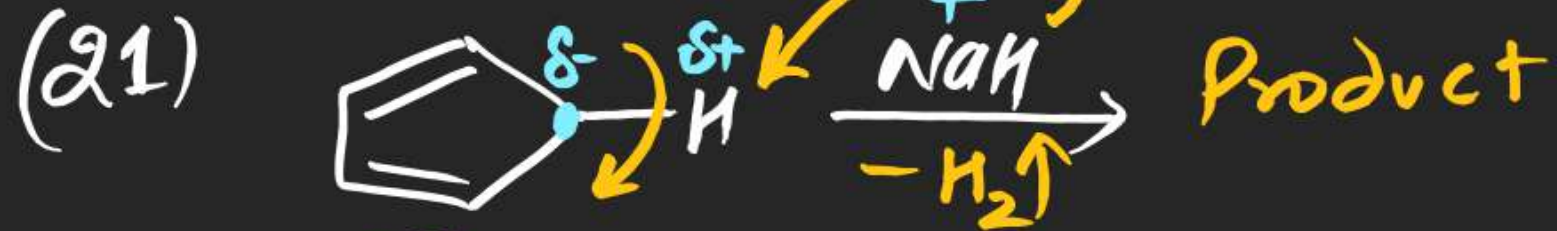


Ex-20:

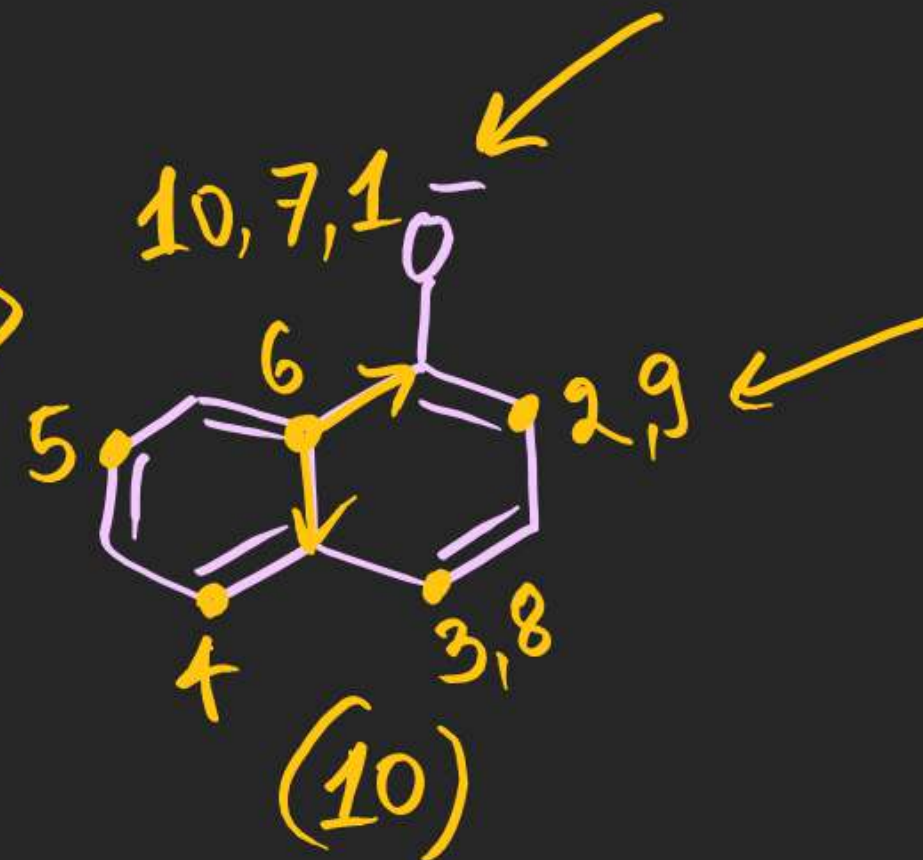
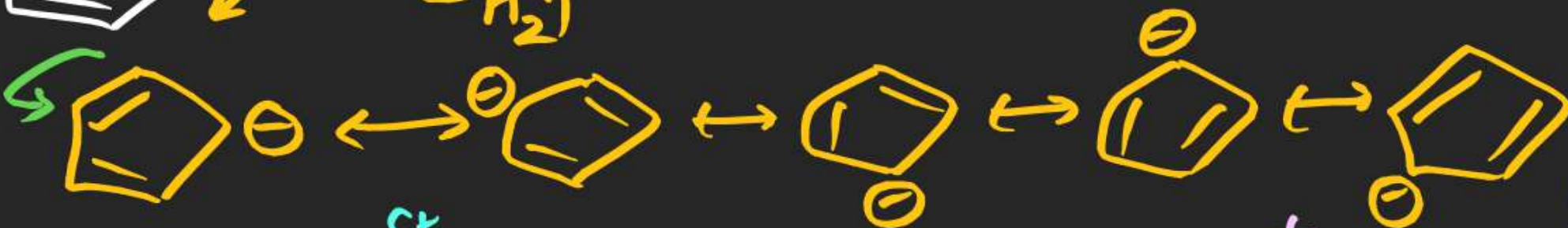


Soln

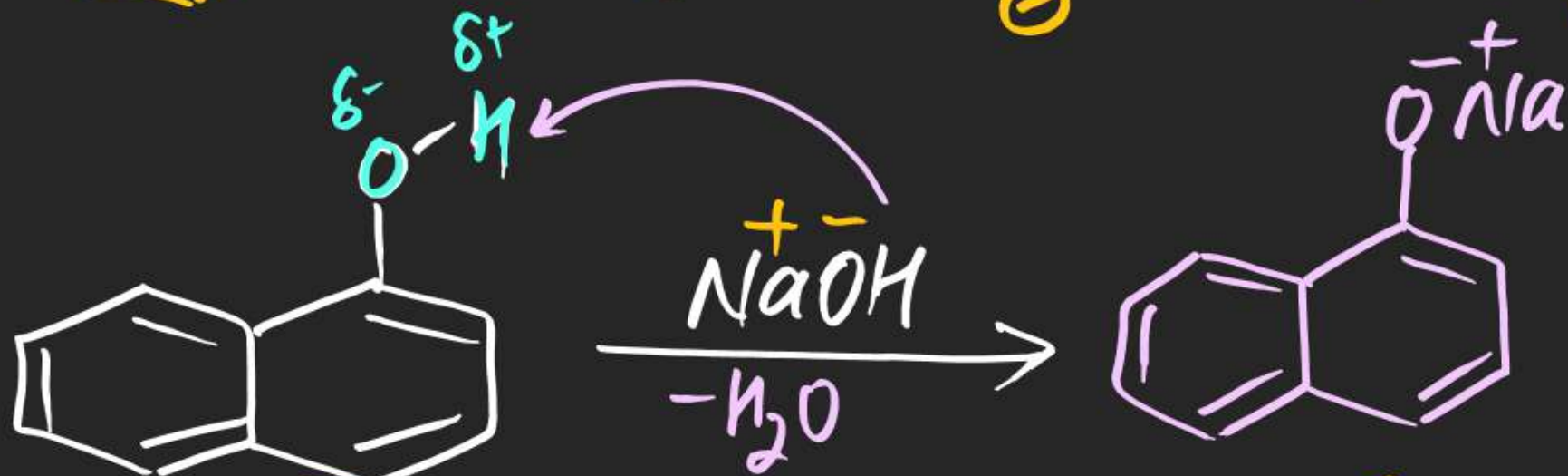




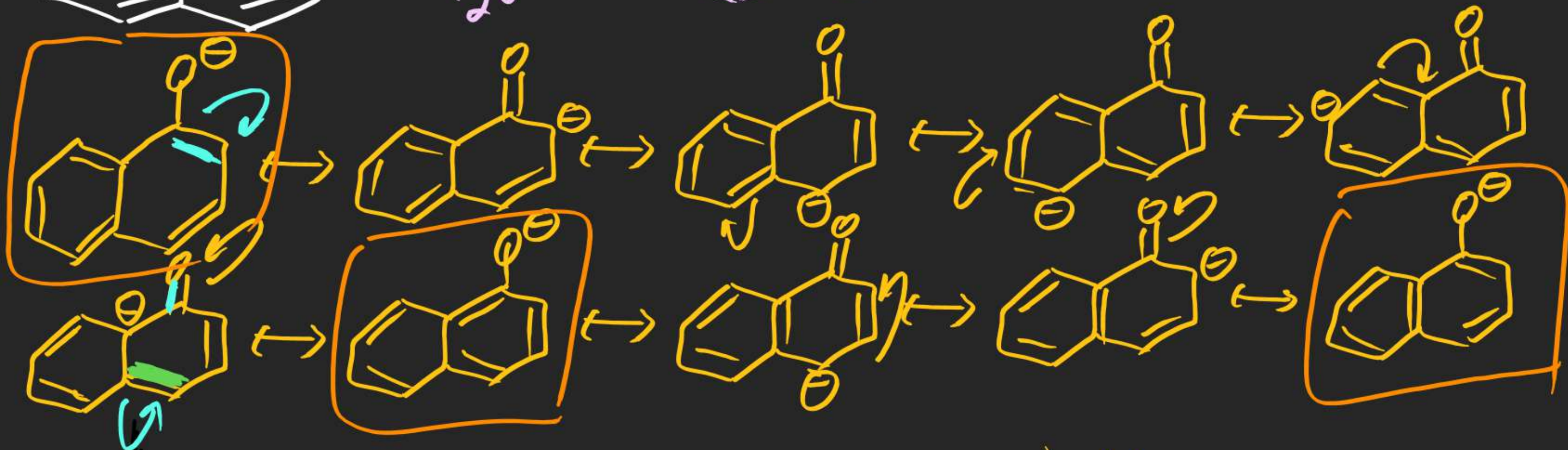
Soln!

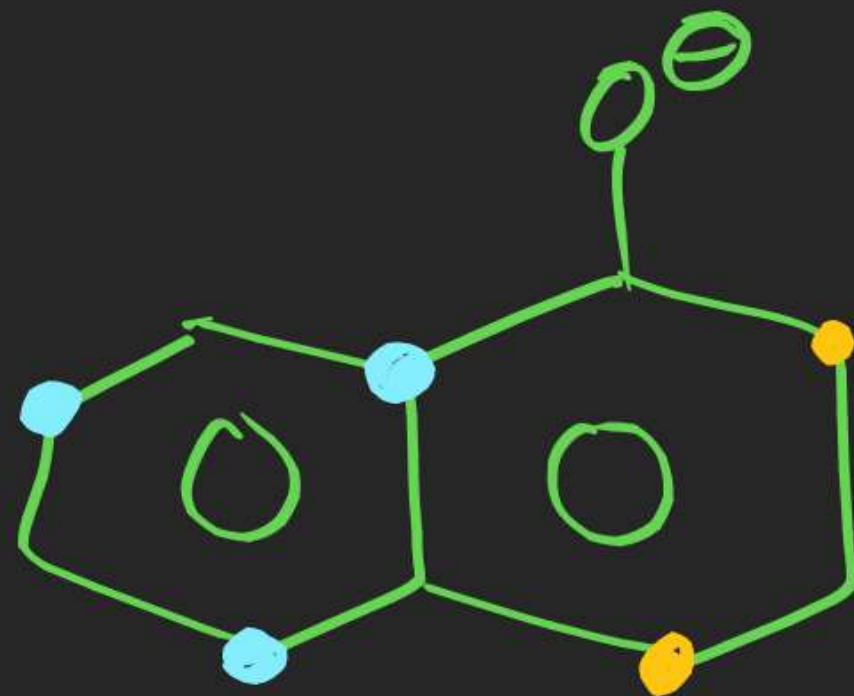
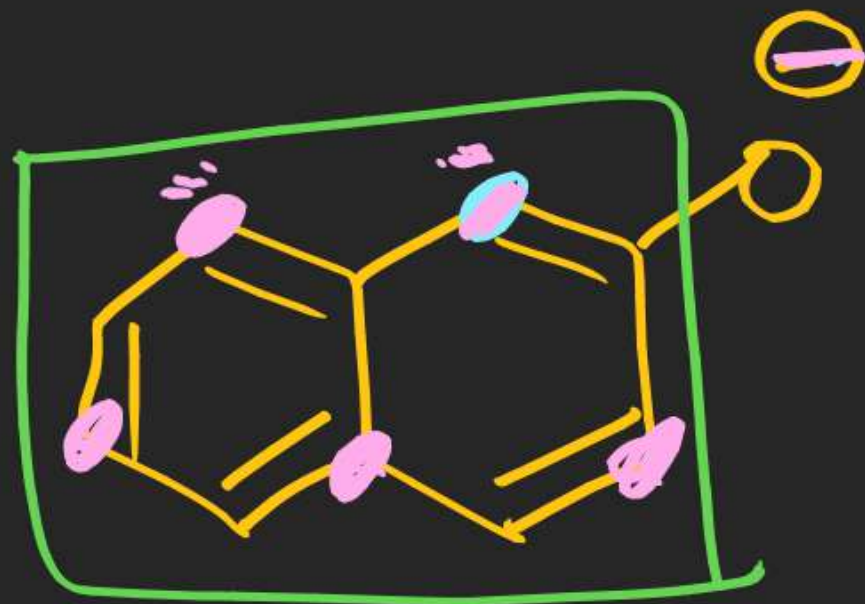


(22)



Soln!



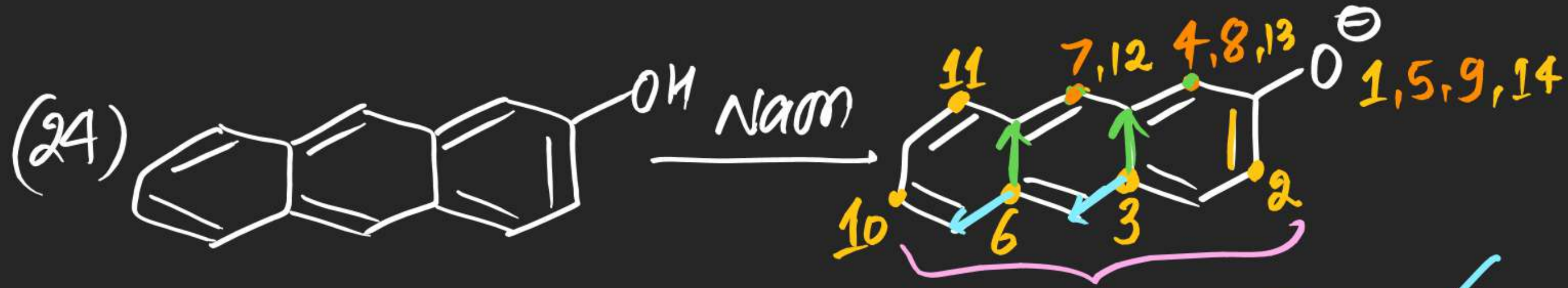


$$\underbrace{3}_{\text{wavy}} + \underbrace{2}_{\text{wavy}} + 4 = 9$$

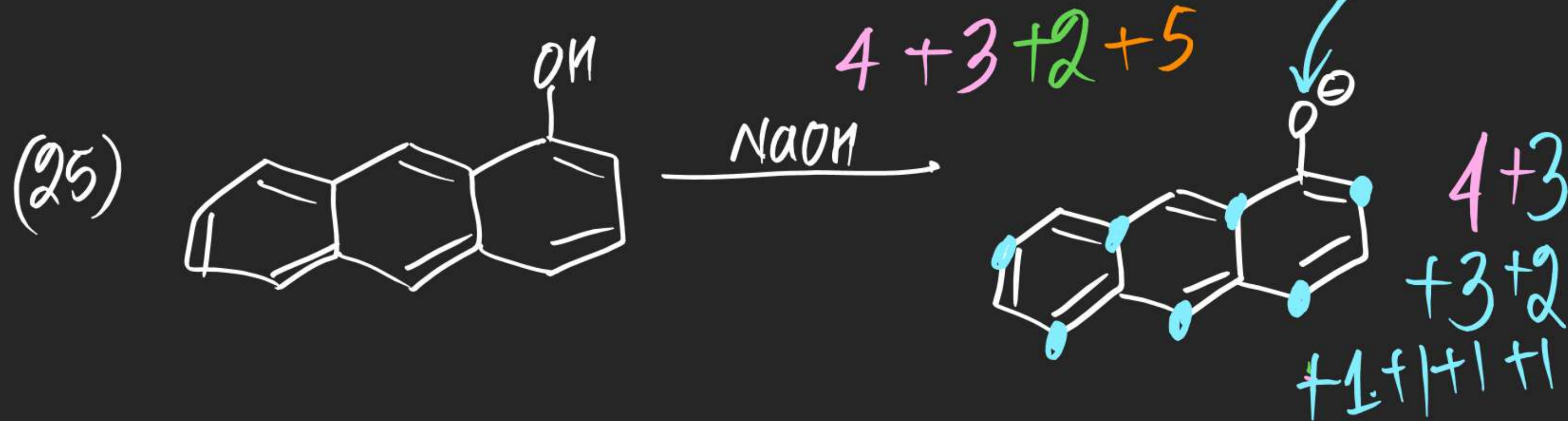
$$\begin{array}{r} 3 + 2 + 2 \\ + 3 \\ \hline 10 \end{array}$$



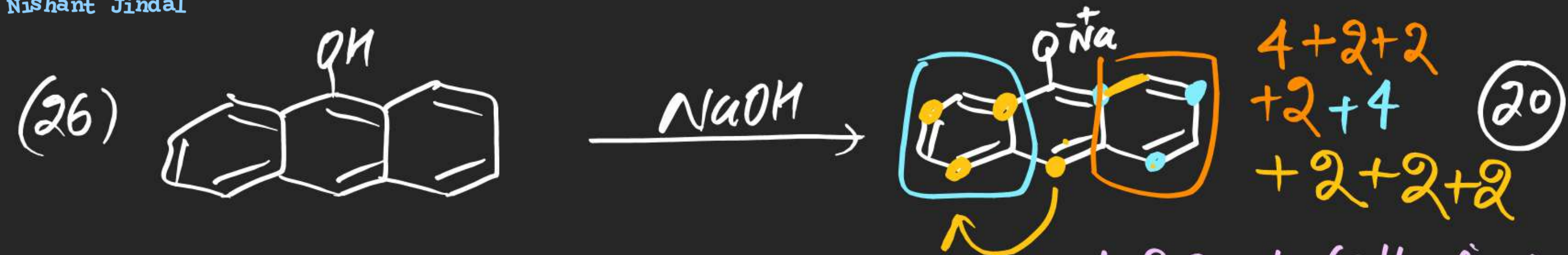
(9)



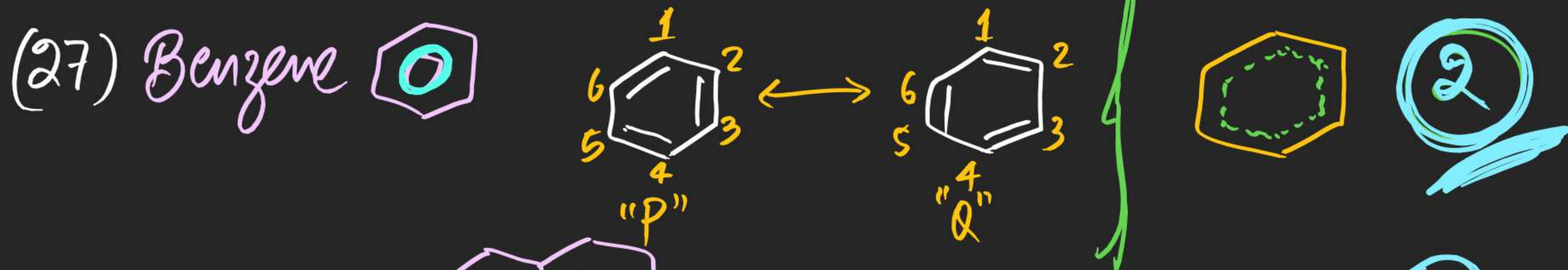
(14)



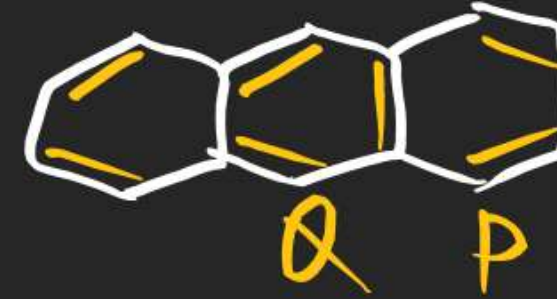
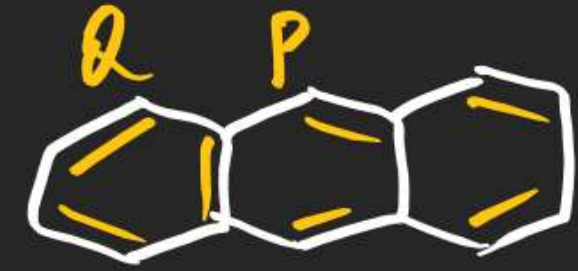
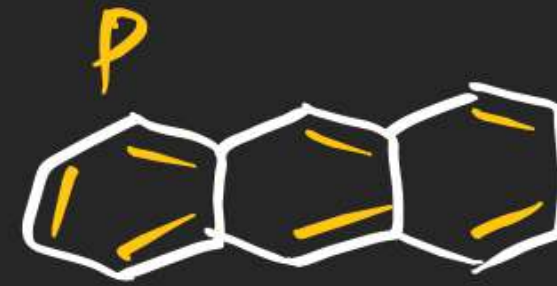
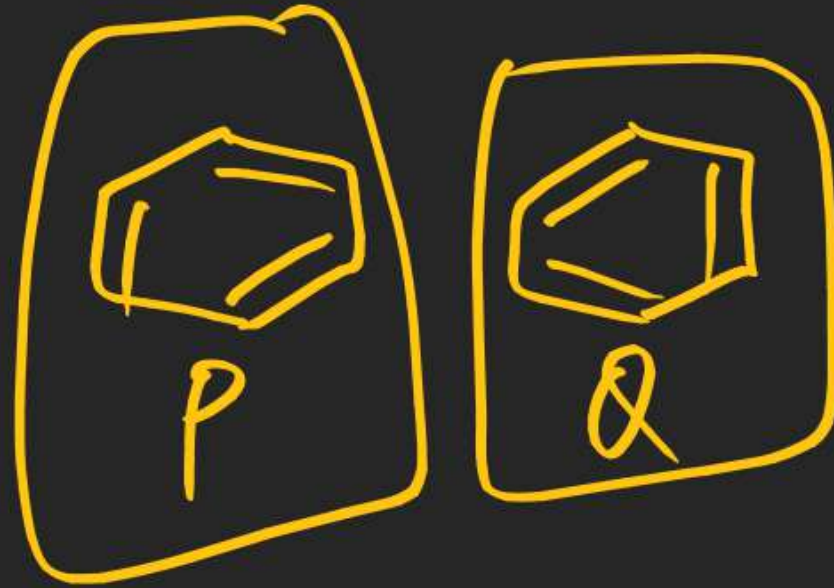
(16)



(#) Find total no. of Neutral Benzenoid R.S of following.

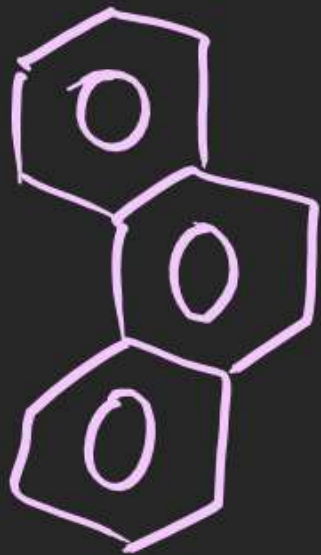


(29) Anthracene



④

(30) Phenanthrene



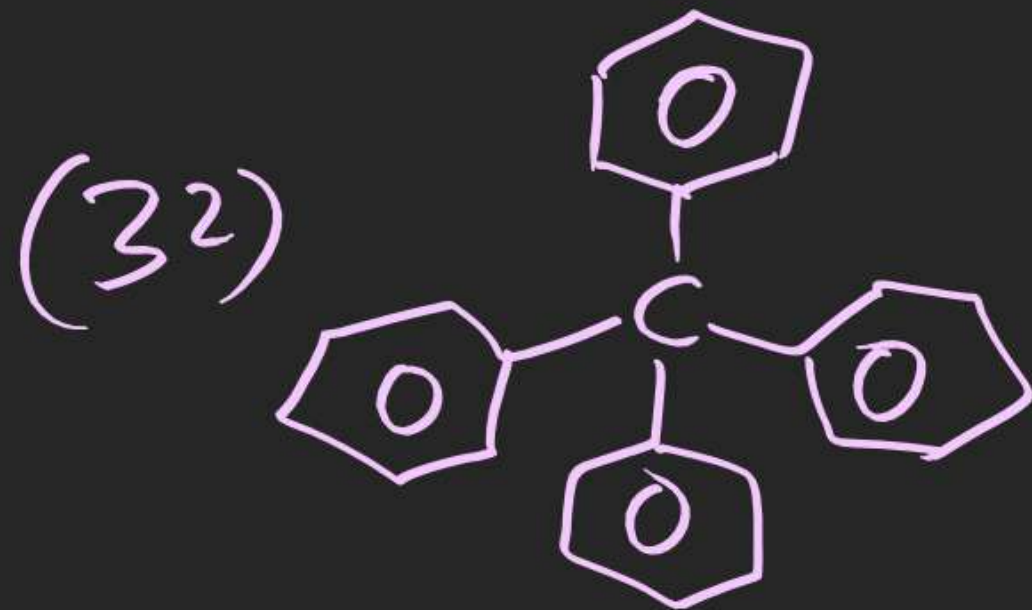
P
Q
P



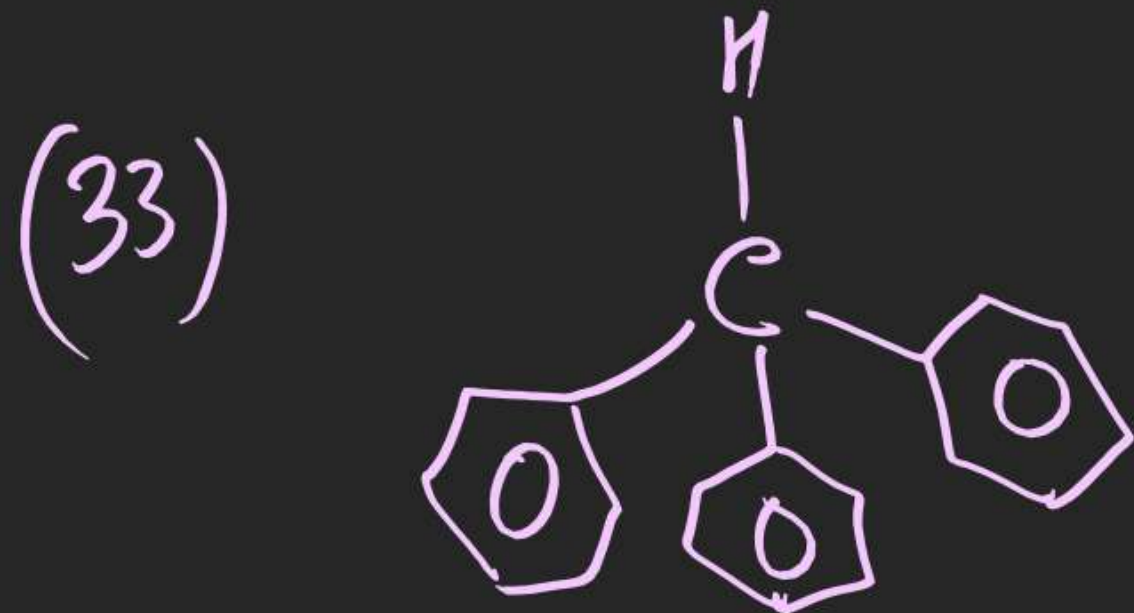
⑤



$$\begin{matrix} P & P \\ P & Q \\ Q & P \\ Q & Q \end{matrix} \quad 2^2 = 4$$



$$2^4 = 16$$



$$2^3 = 8$$

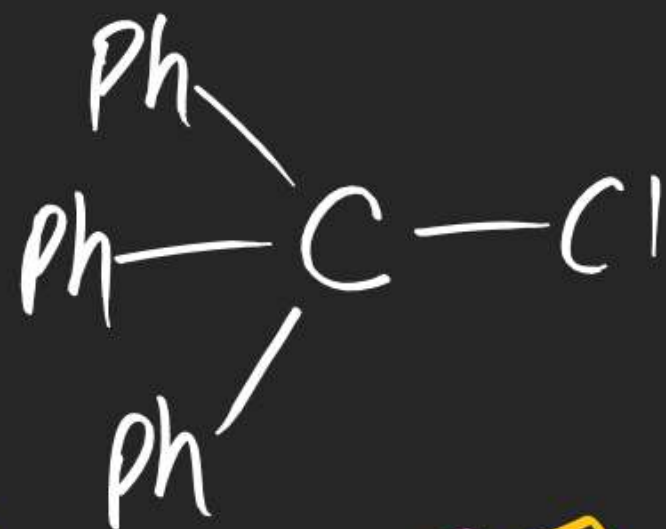


$$2^2 = 4$$

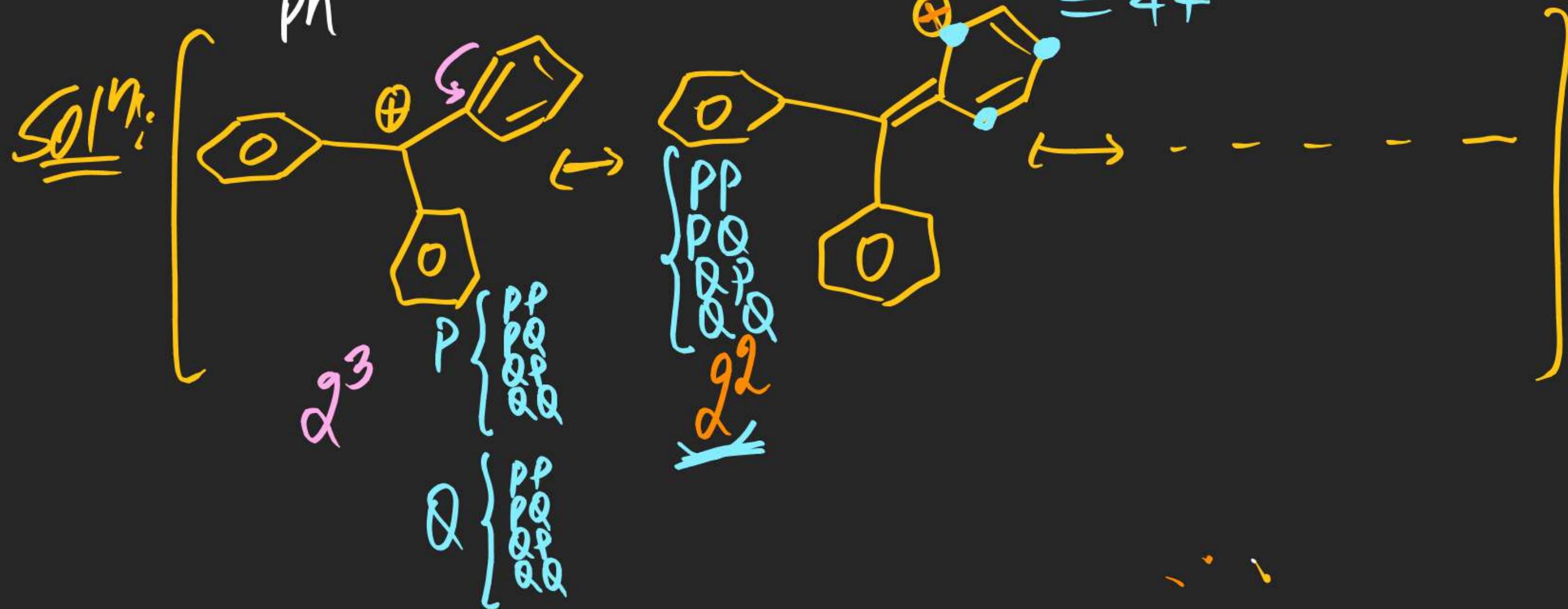


$$2^1 = 2$$

(36) Total no. of RS of Product of following Reaction.

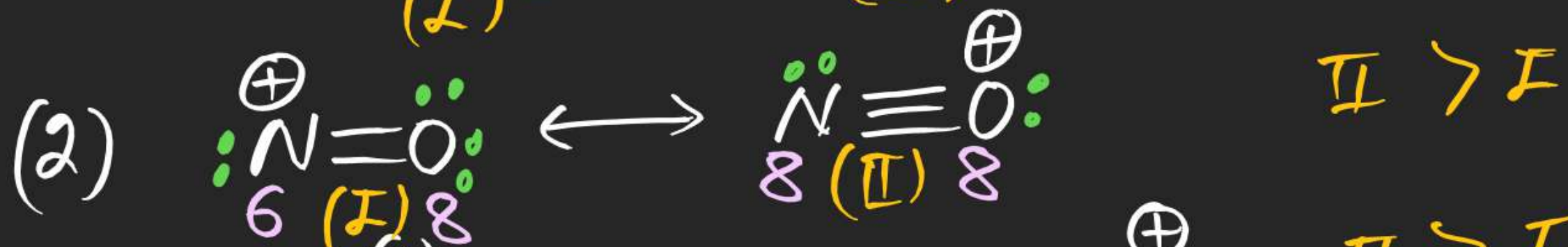
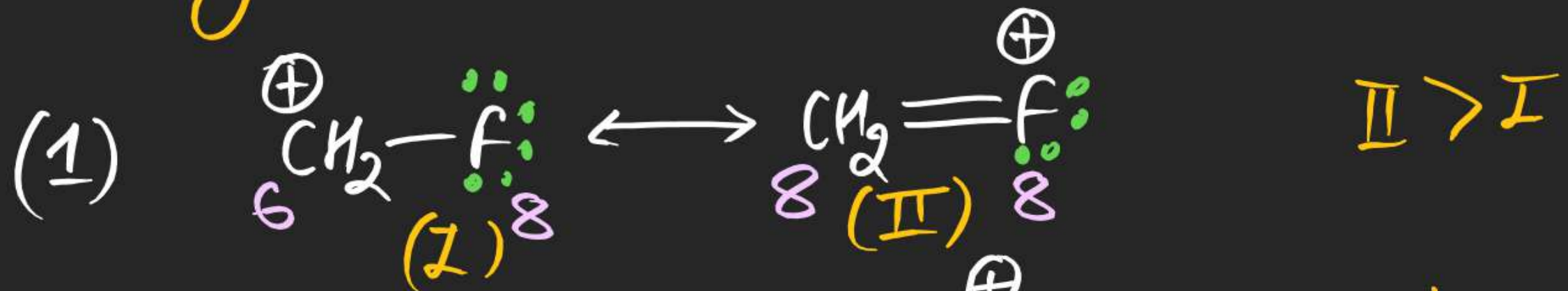


$$\begin{aligned}
 & \text{+ in ph} \quad \text{+ in ph} \quad \text{+ in ph} \\
 & [2^3 + 3 \times 2^2 + 3 \times 2^2 + 3 \times 2^2] \\
 & = 8 + 12 + 12 + 12 \\
 & = 44
 \end{aligned}$$

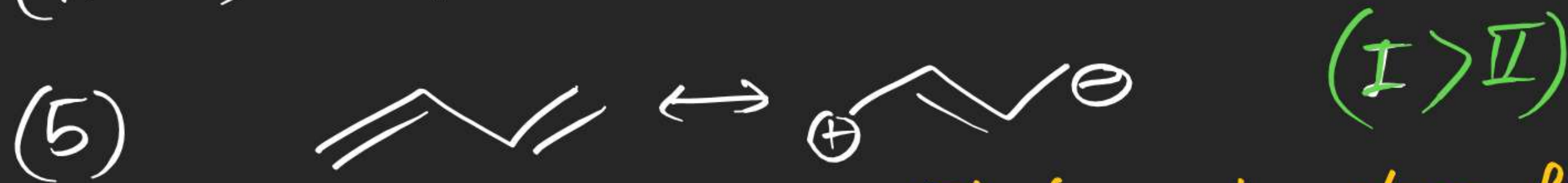


(#) Rules for Stability of RS :

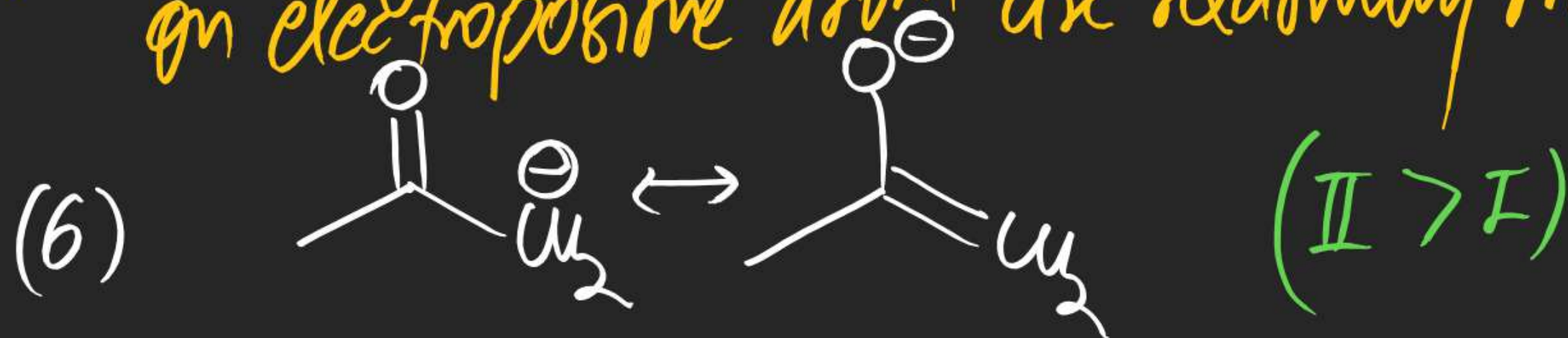
(1) RS having complete octet is more stable than having incomplete octet.



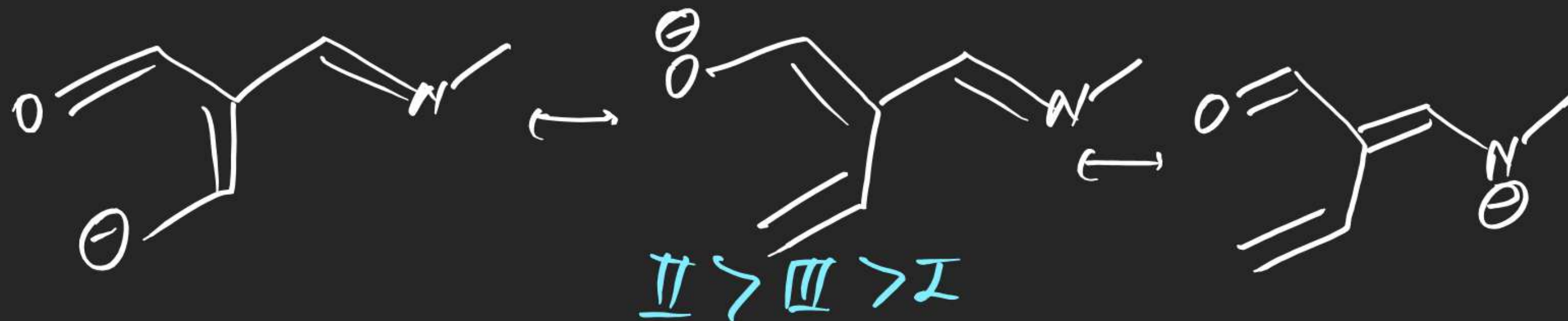
(2) RS having higher no. of Covalent Bond or less charge is more stable.



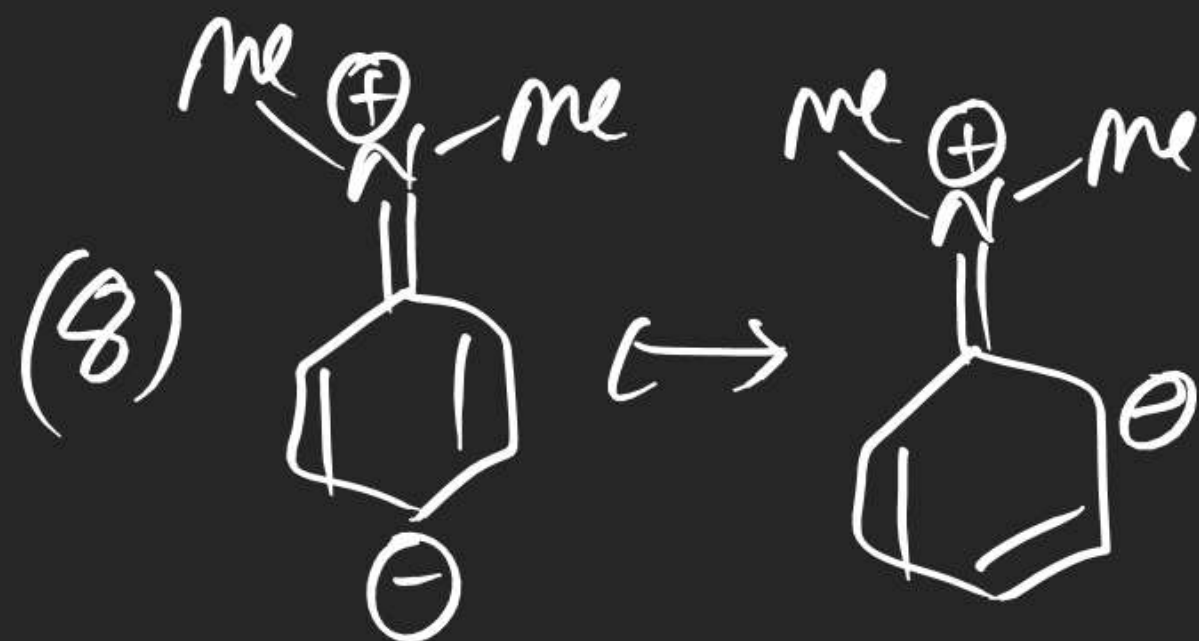
(3) RS having (-)ve charge on Electronegative atom & (+)ve charge on electropositive atom are relatively more stable.



(7)

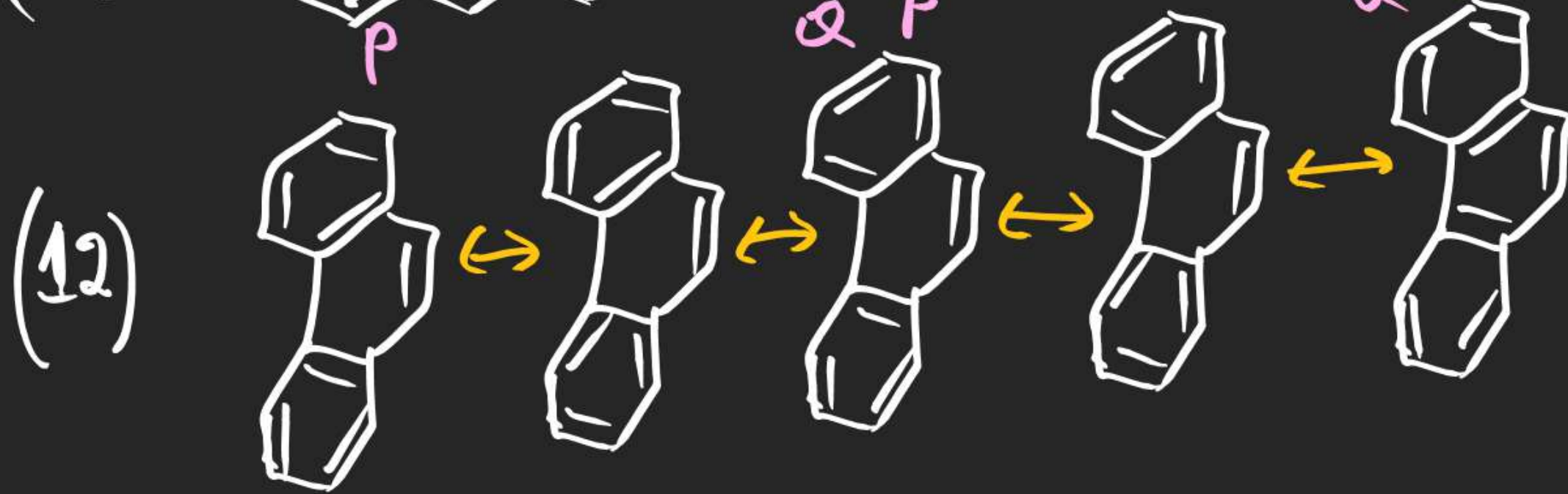
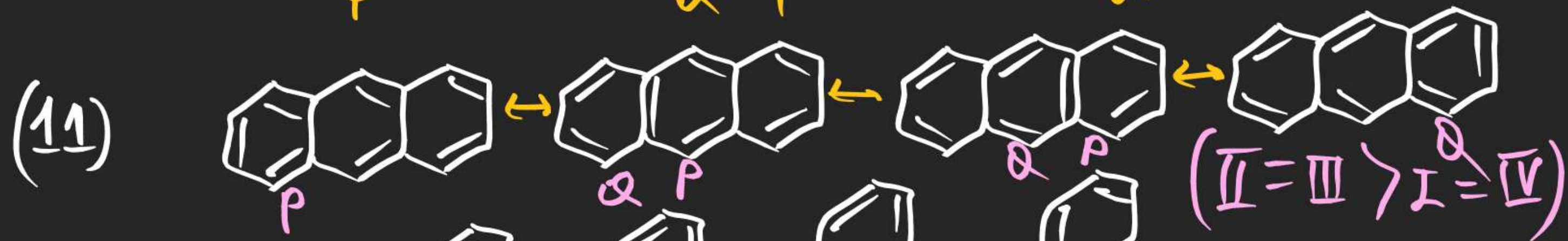


(4) RS having opp. charges closer & like charges away are more stable.

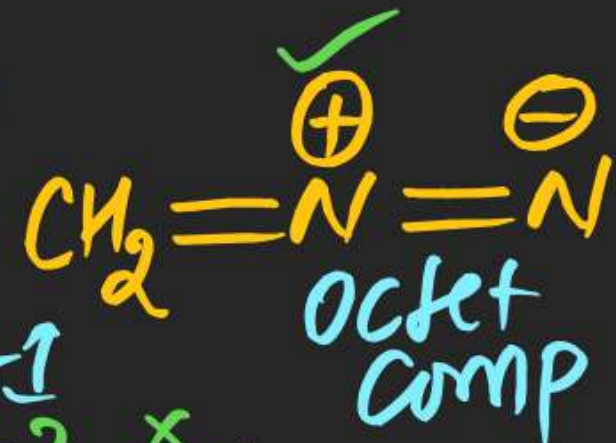


$\text{II} > \text{I}$

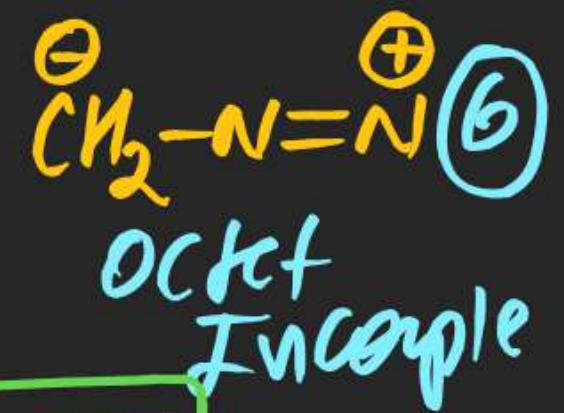
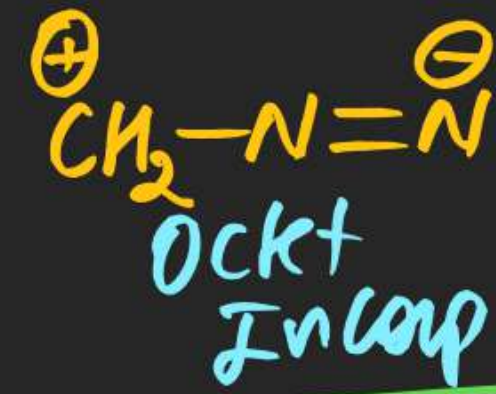
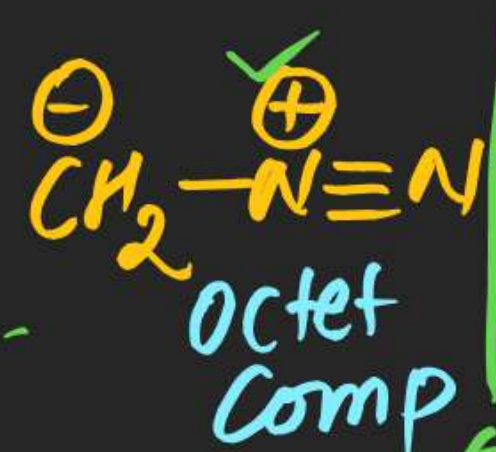
⑤ RS having higher no. of Benzenoid segment is more stable.



IIT Ad
(13)



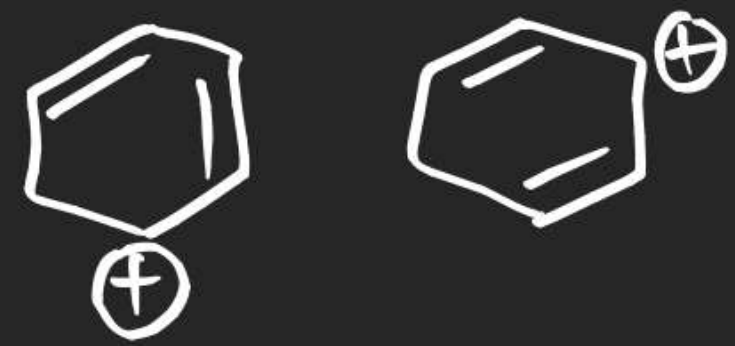
Rule-1
 Rule-2 x
 Rule-3 ✓



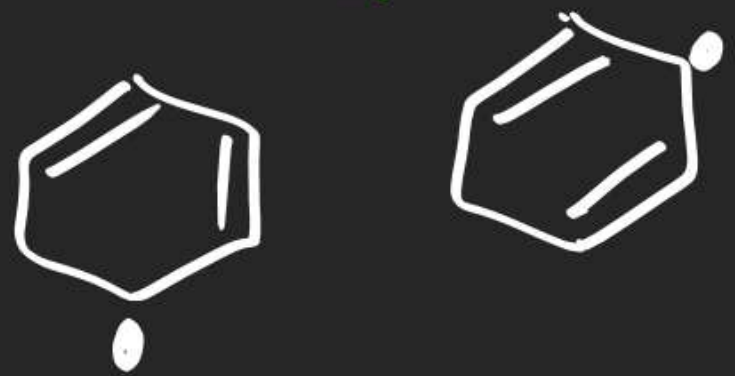
Rule-2 x
 Rule-3 ✓

$\text{I} > \text{II} > \text{III} > \text{IV}$

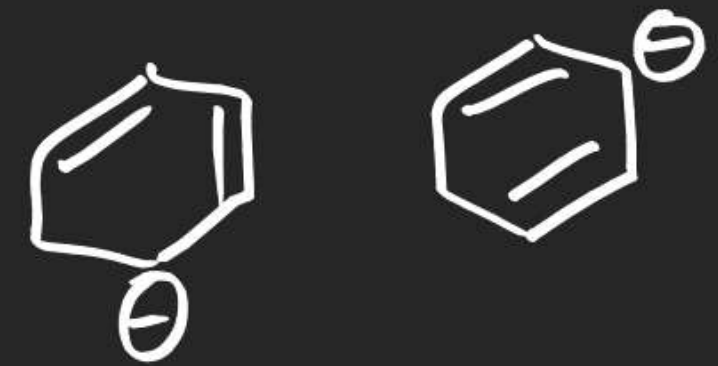
(14)



(15)



(16)



(17)



! Resonance / Mesomeric Effect:-