

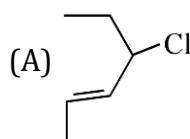


SOLUTION DPP-02

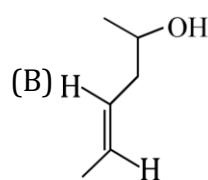
Link to View Video Solution: [Click Here](#)

1. Optical isomers are those which differ in their optical activity

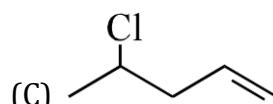
Geometrical isomerism are those having restricted rotation and different group on terminal carbon.



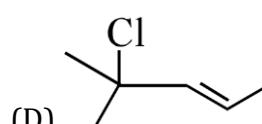
- Show geometrical isomerism
- Show optical isomerism



- Show geometrical isomerism
- Show optical isomerism

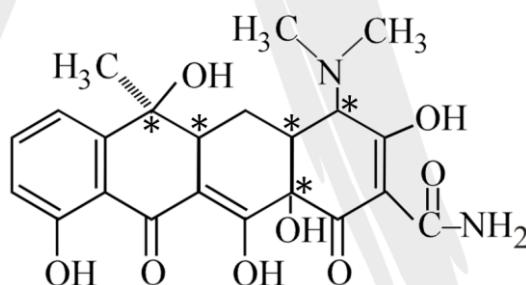


- Not Show geometrical isomerism.
- Show Optical isomerism.

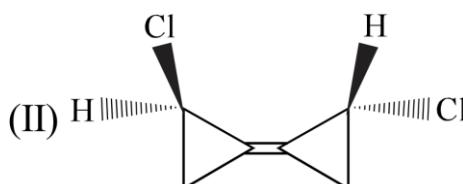
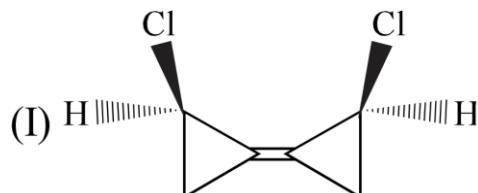


- Show geometrical isomerism.
- Not Show Optical isomerism.

2. Tetracycline: - Total chirality Centre = 5



3. For the given compounds, select the correct statements :

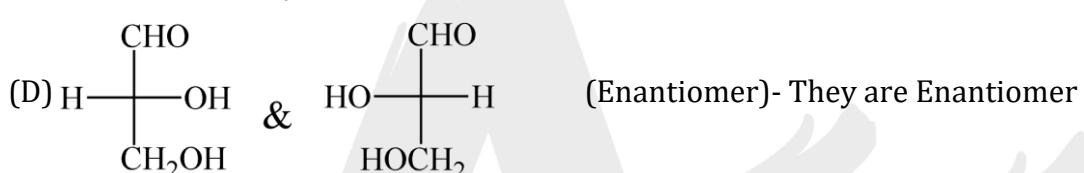
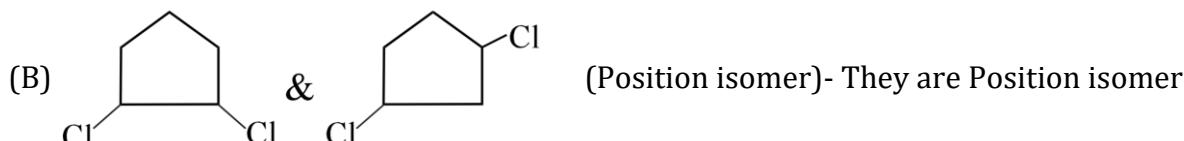
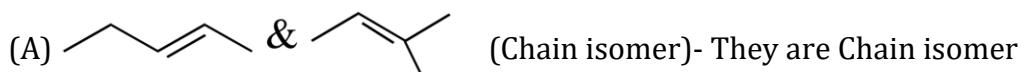


- (A) They are not optically inactive
- (B) Not enantiomeric
- (C) I is polar and II is non polar
- (D) Compound is chiral



Link to View Video Solution: [Click Here](#)

4. Which of the following statements are correct:



5. 1-heptanol

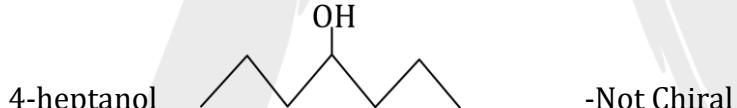
-Not Chiral



-It is Chiral



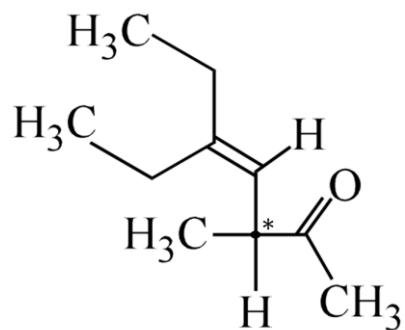
-It is Chiral



-Not Chiral

Hence option B is correct.

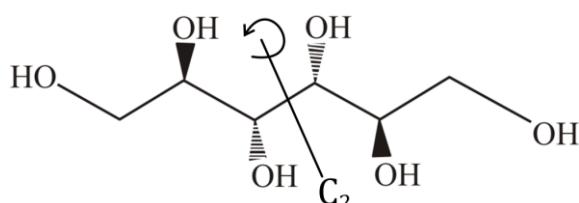
6. It shows optical isomerism



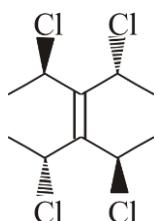


Link to View Video Solution: [Click Here](#)

7. C_2 axis of symmetry

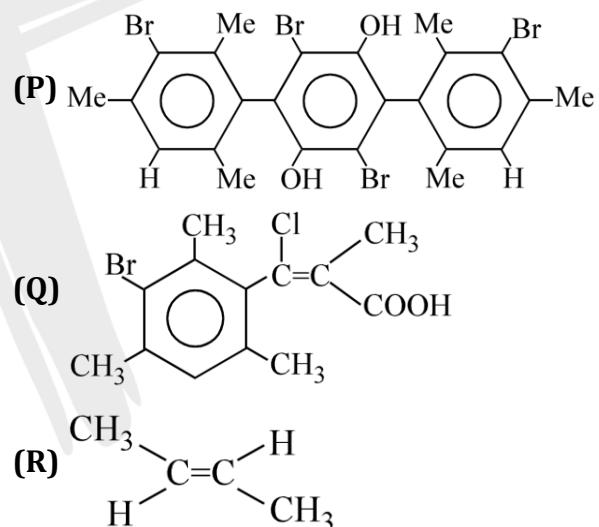


- 8.



- No Centre of Symmetry
- No Plane of Symmetry
- Optically active
- Cannot show E/Z as Group attach to Double Bond are same

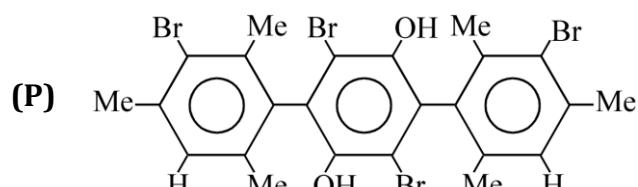
9. (A) Compound show geometrical isomerism



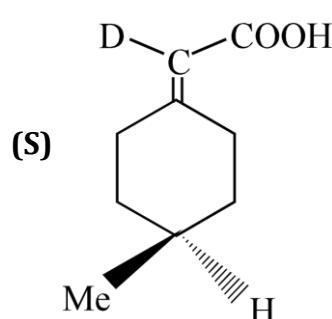


Link to View Video Solution: [Click Here](#)

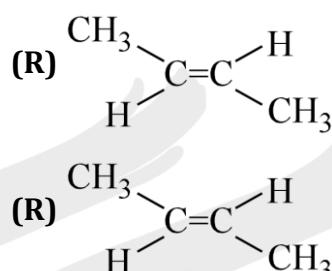
(B) Compound shows optical isomerism



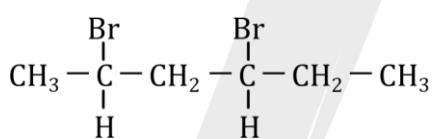
(C) Compound having plane of symmetry



(D) Compound having Centre of symmetry

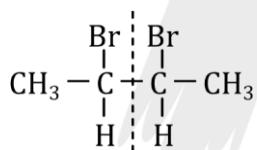


10.

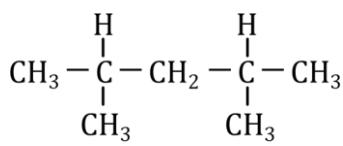


-No Plane of symmetry

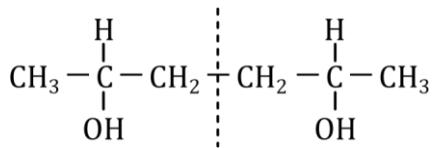
Not an Meso compound



-Plane of symmetry present
so a Meso compound.



-No Plane of symmetry
Not an Meso compound



-Plane of symmetry
so these are Meso compound.