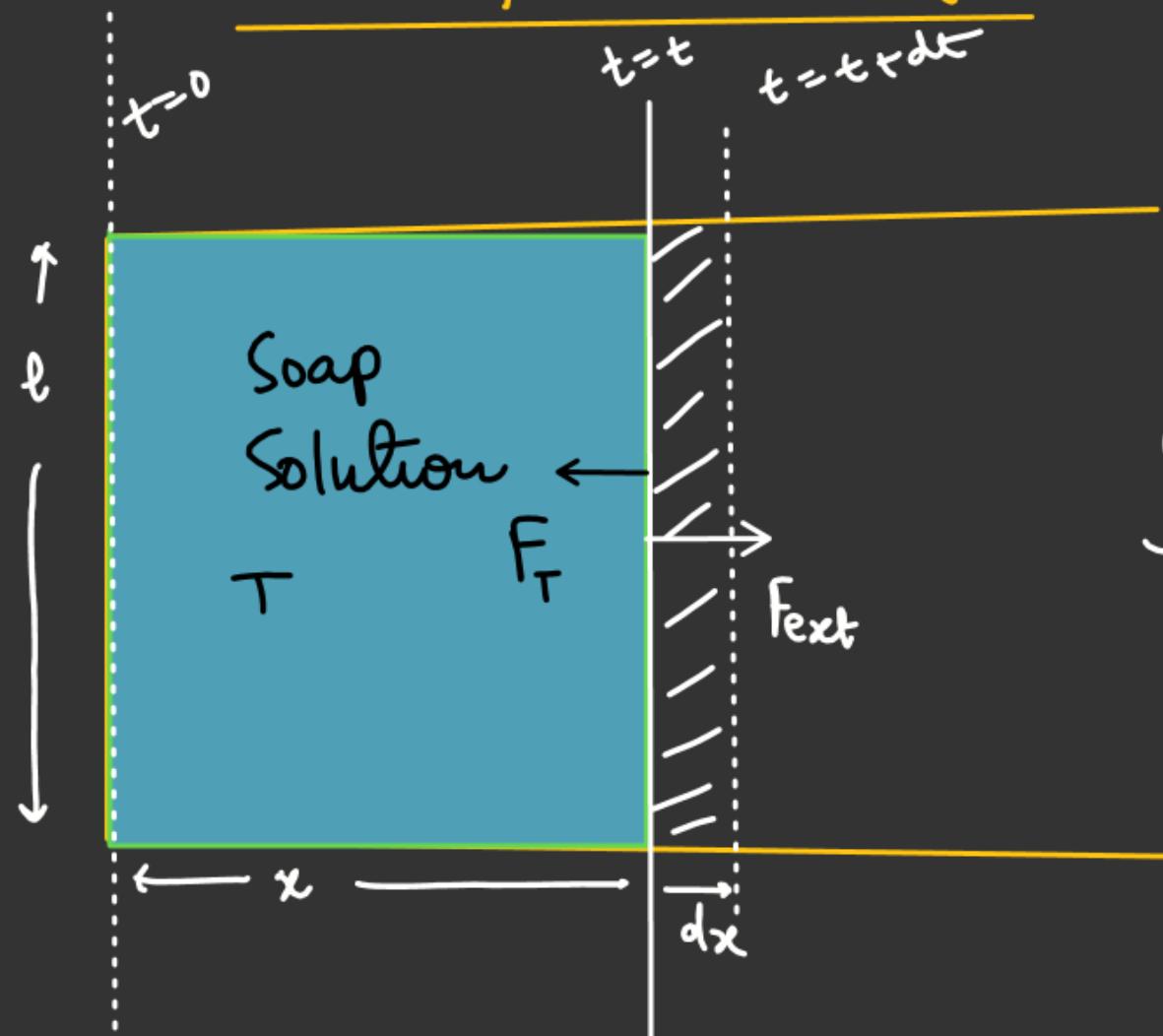




Surface Energy



$$\boxed{\frac{U}{A} = T}$$

$$dW_{\text{ext}} = F_{\text{ext}} \cdot dx$$

$$F_{\text{ext}} = f_T = T(2l)$$

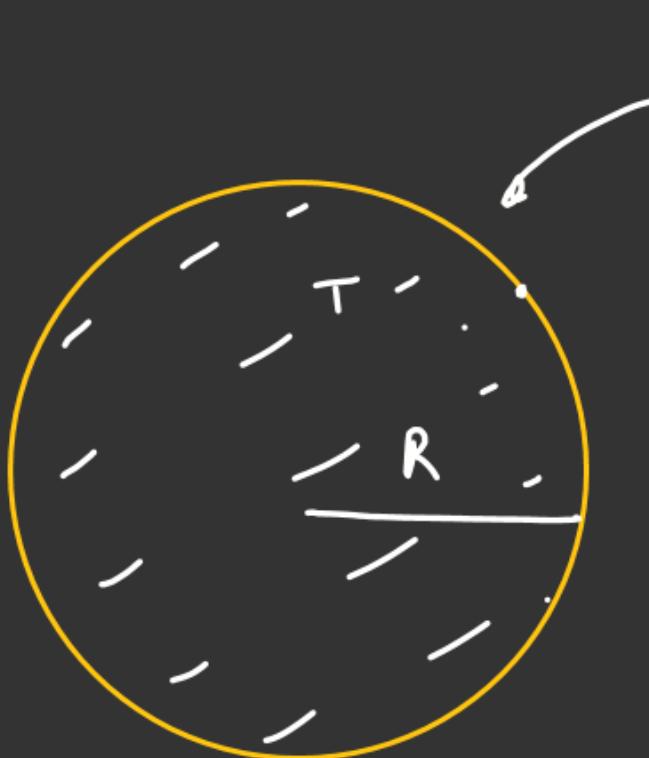
$$\int dW_{\text{ext}} = \int T \cdot 2(l \, dx)$$

$$W_{\text{ext}} = 2T(l \cdot x)$$

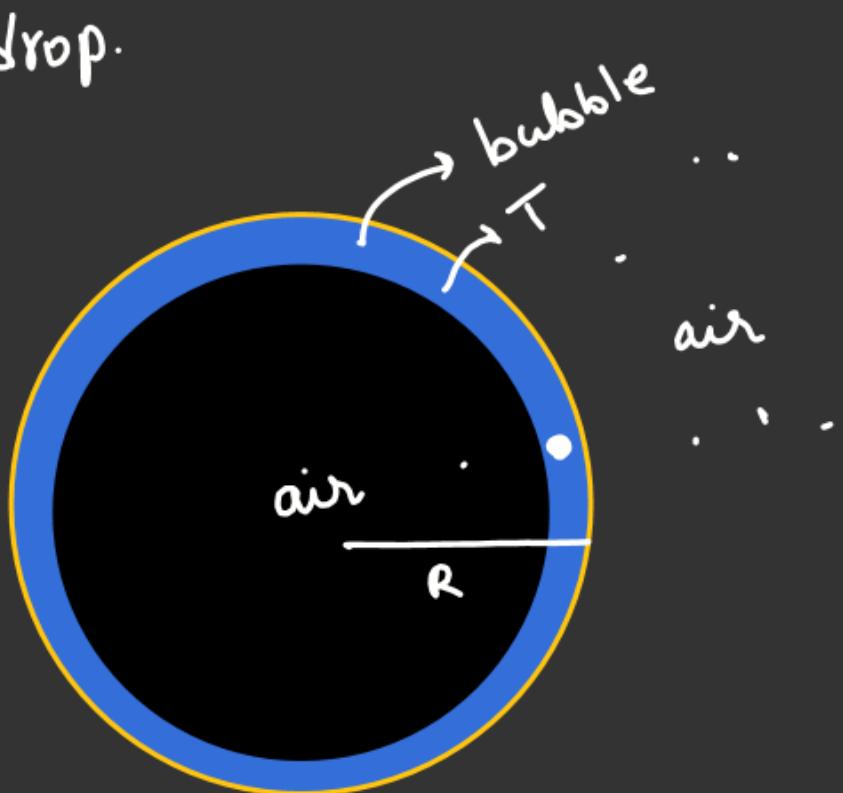
||

$$\text{Surface energy} \quad U = T A_{\text{eff}}$$

$$A_{\text{eff}} = \left(A \times \frac{\text{No of contact}}{\text{Contact}} \right)$$



$$U = (T \cdot 4\pi R^2)$$



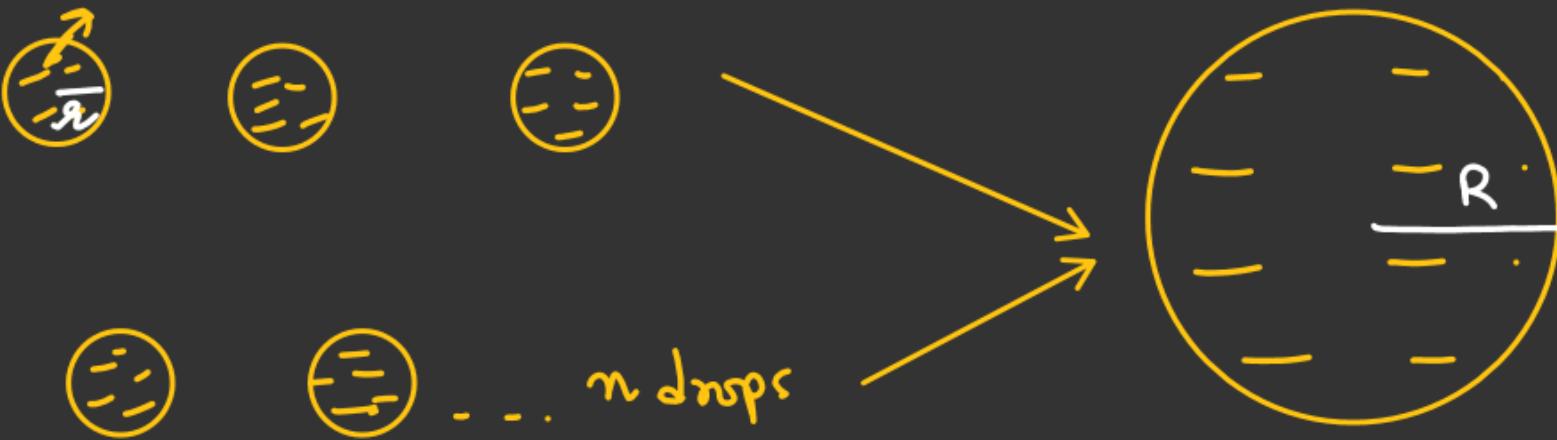
$$U = T \times 4\pi R^2 \times 2$$

$$U = (T \cdot 8\pi R^2)$$

~~∴~~

n -liquid drops coalesce to form a single bigger liquid drop.

(γ = Surface tension)



$$n \cdot \frac{4}{3} \pi r^3 = \frac{4}{3} \pi R^3$$

$$\Delta U = U_f - U_i$$

$$R = (n^{1/3}, r)$$

$$= T \cdot 4\pi R^2 - n \cdot 4\pi r^2 \cdot T$$

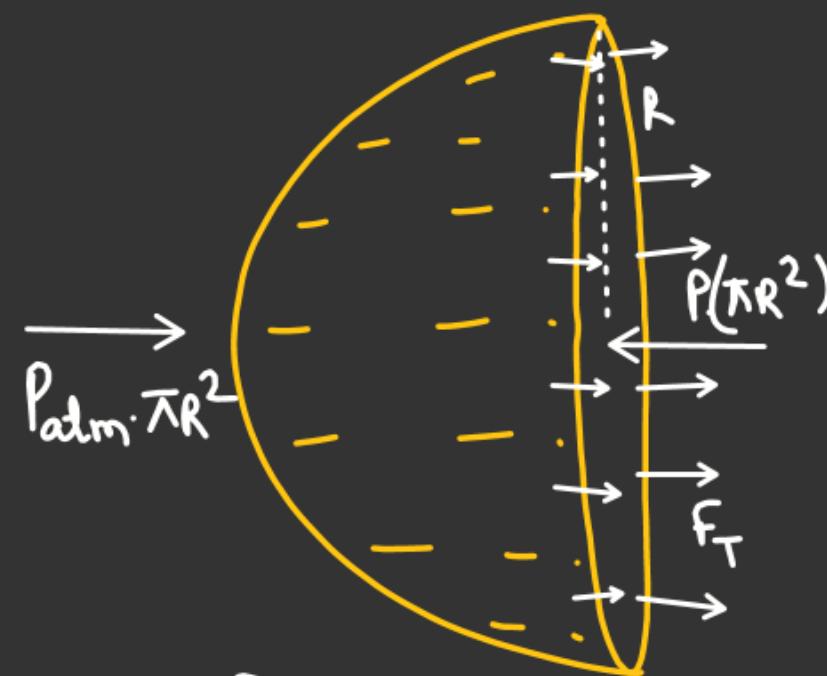
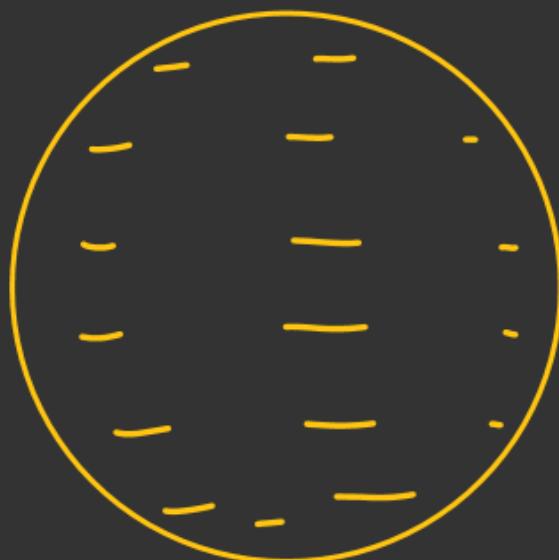
$$U_i = n(T \cdot 4\pi r^2)$$

$$= 4\pi T(R^2 - n \cdot r^2)$$

$$U_f = T \cdot 4\pi R^2$$

$$\boxed{\Delta U = \underbrace{4\pi r^2 \cdot T \times n}_{\Downarrow} (n^{1/3} - 1)}$$

$$\Delta U = U_i (n^{1/3} - 1)$$

~~Q&A~~EXCESS PRESSURE INSIDE A LIQUID DROP

$$F_T = T \times 2\pi R$$

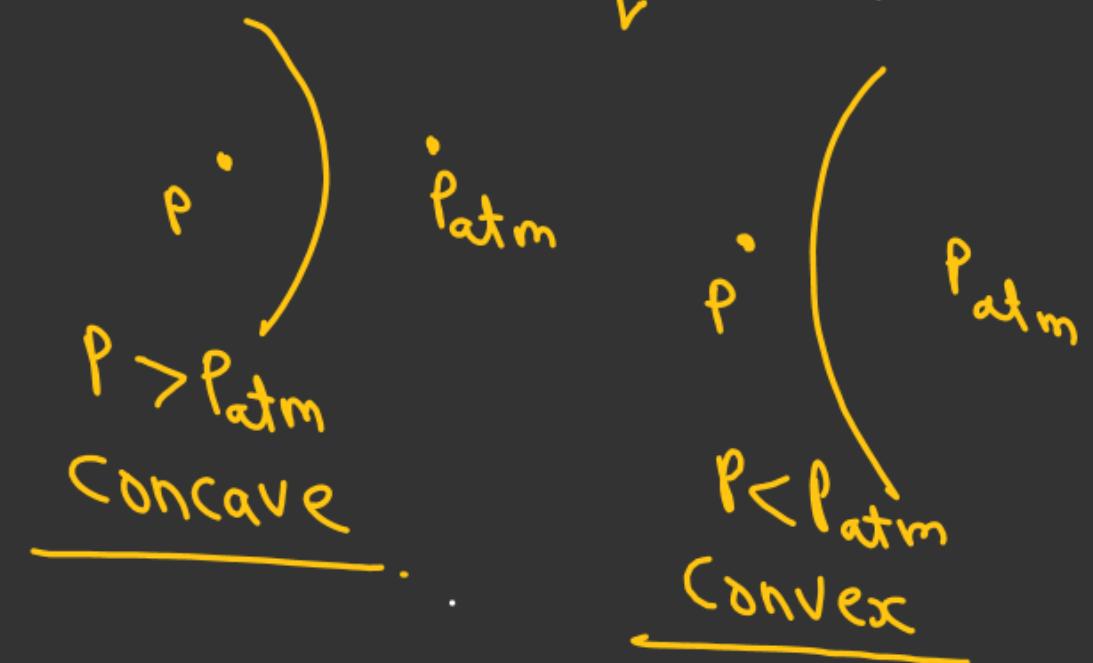
A + Equilibrium

$$P_{atm} \cdot \pi R^2 + F_T = P \cdot \pi R^2$$

$$P_{atm} \cdot \pi R^2 + T \cdot 2\pi R = P \cdot \pi R^2$$

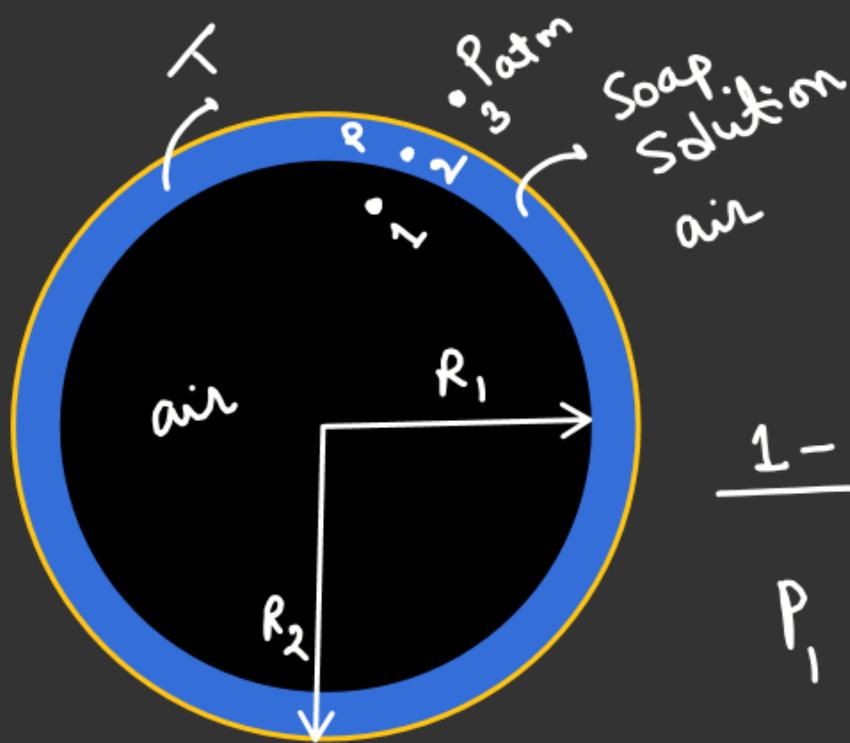
~~$$P - P_{atm} = \frac{2T}{R}$$~~

Excess pressure
inside a liquid drop.



~~*X~~

Excess pressure inside a Soap bubble

1-2

$$P_1 - P_2 = \frac{2T}{R_1}$$

2-3

$$P_2 - P_{atm} = \frac{2T}{R_2}$$

if $R_1 \approx R_2 \approx R$

$$P_1 - P_{atm} = \frac{4T}{R}$$

~~*X~~

$$\Delta P = \frac{4T}{R}$$

Excess pressure
inside a Soap bubble.

$$P_1 - P_{atm} = 2T \left(\frac{1}{R_1} + \frac{1}{R_2} \right)$$

Excess pressure for any liquid Surface

$$\Delta P = 2T \left[\frac{1}{R_1} \pm \frac{1}{R_2} \right]$$

R_1 & R_2 be the Radius of Curvature of the two Curve Surfaces

(+) → When radius of Curvature on Same Side.

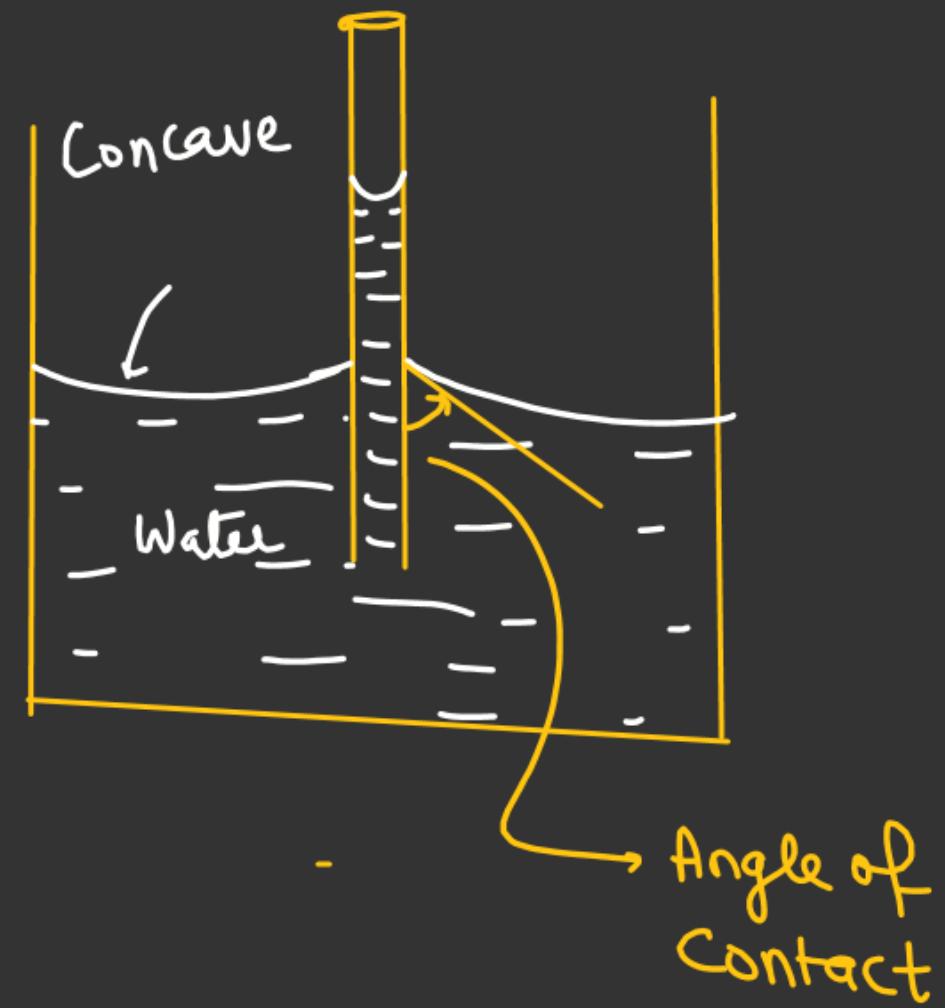
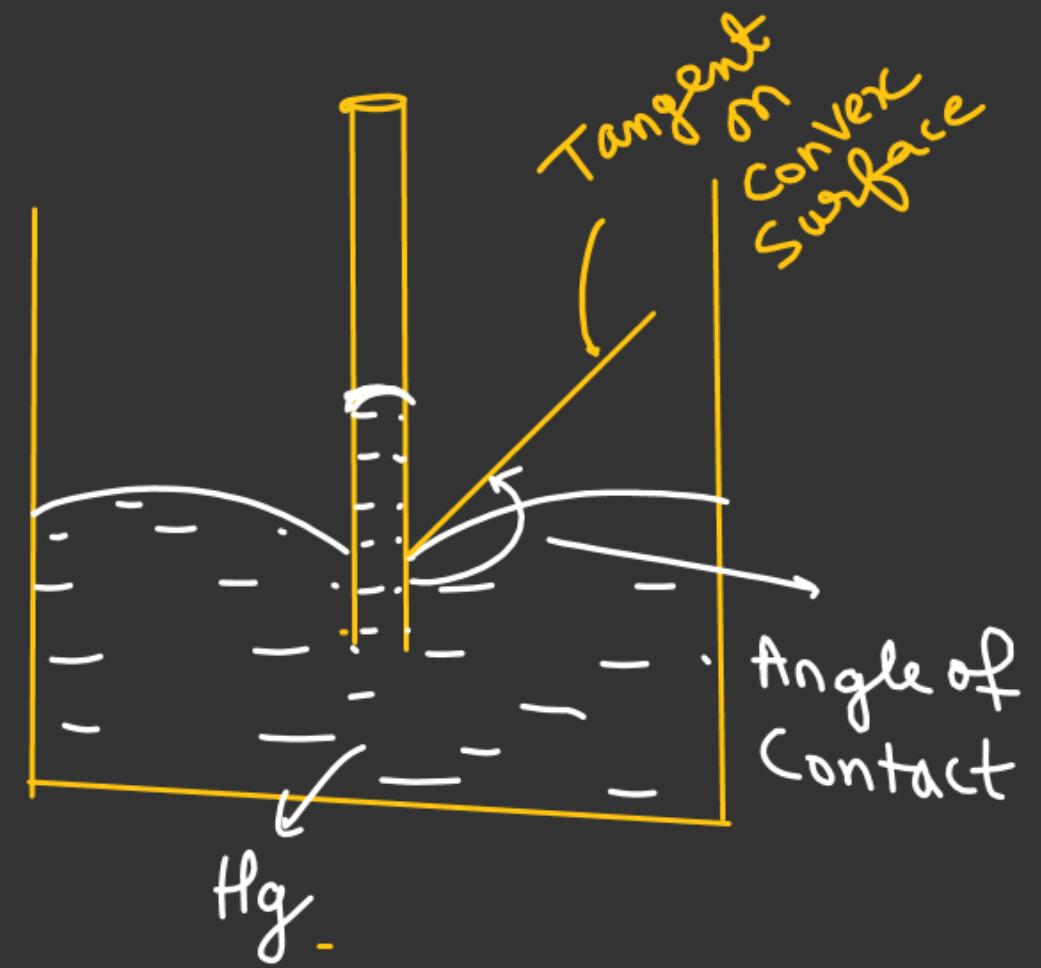
(-) → When radius of Curvature on opposite side.

Excess pressure for thin film

$$\Delta P = 4T \left[\frac{1}{R_1} \pm \frac{1}{R_2} \right]$$

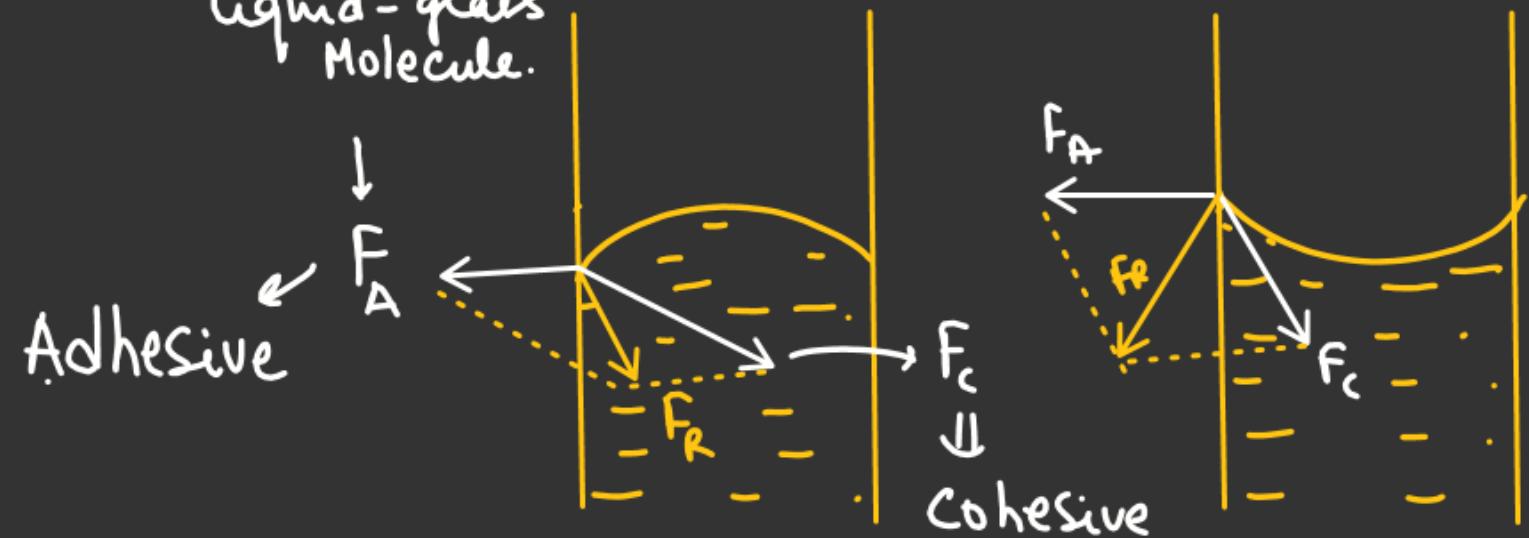


Angle of Contact

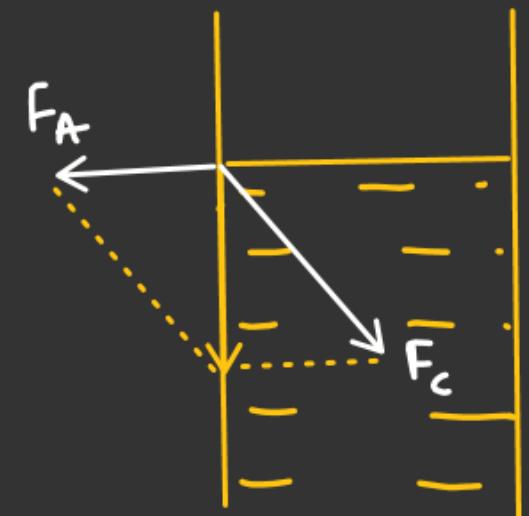


Shape of liquid Meniscus

Liquid-glass
Molecule.



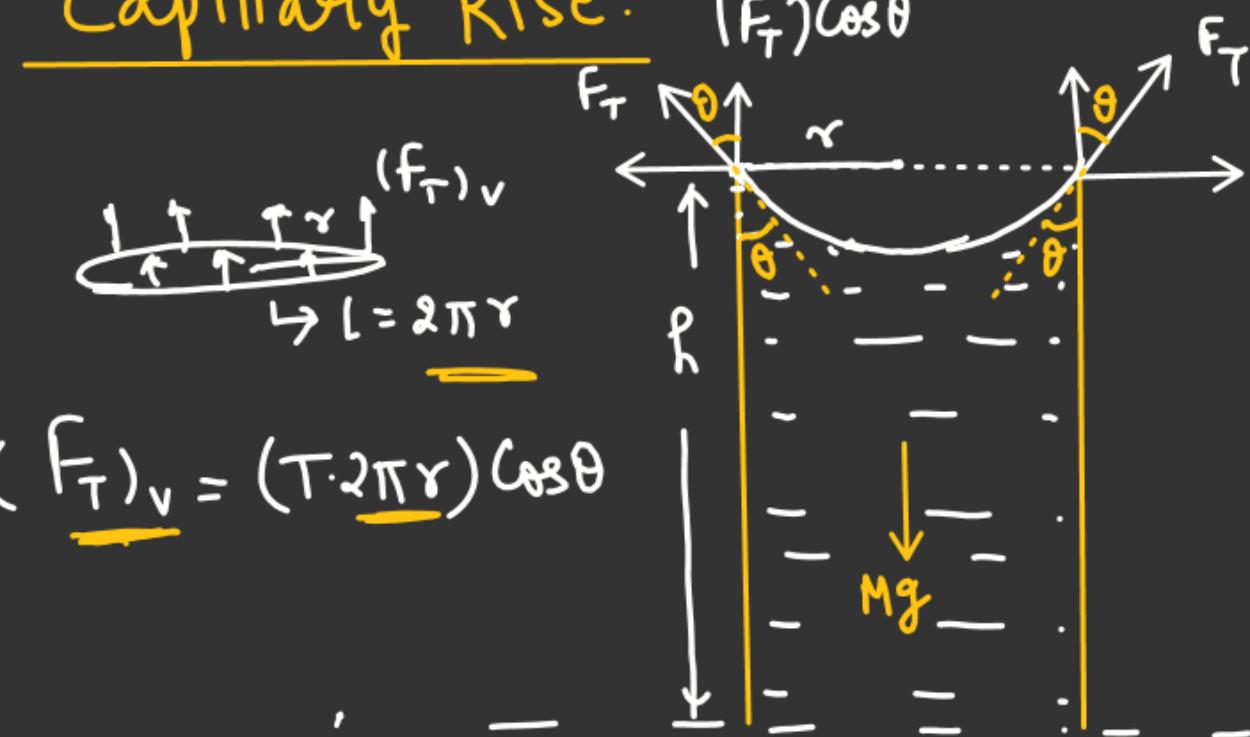
$$\underline{F_c > F_A} \Rightarrow \text{Convex Meniscus.}$$



$$\underline{F_A > F_c} \quad \text{Concave Meniscus.}$$



Capillary Rise.



$$(F_T)_v = (T \cdot 2\pi r) \cos \theta$$

$$T \cdot 2\pi r \cos \theta = Mg$$

$$M = (\pi r^2 h) \rho$$

$$T \cdot 2\pi r \cos \theta = (\pi r^2 h) \rho g$$

$$\boxed{h = \frac{2T \cos \theta}{\rho g r}}$$

θ = Angle of Contact

r = radius of capillary

ρ = density of liquid.

T = Surface tension of liquid



If height of Capillary is insufficient

$$h = \frac{2T \cos \theta}{\rho g r}$$

$$\cos \theta = \left(\frac{r}{R} \right)$$

$$h = \frac{2T}{\rho g R}$$

θ = Angle of Contact

r = radius of Capillary

ρ = density of liquid.

T = Surface tension of liquid

R = Radius of Curvature
of liquid surface

$$Rh = \frac{2T}{\rho g}$$

$$Rh = \text{Constant}$$

$$\frac{R_1}{R_2} = \frac{h_2}{h_1} \Leftarrow R_1 h_1 = R_2 h_2 = C$$

$$h_1 > h_2$$

$$R_1 < R_2$$

