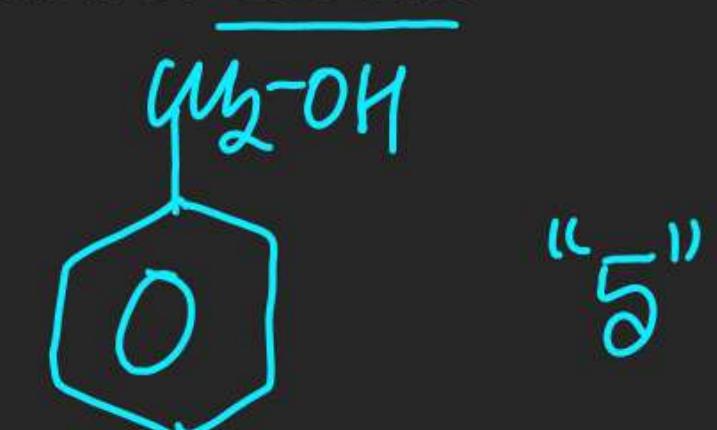
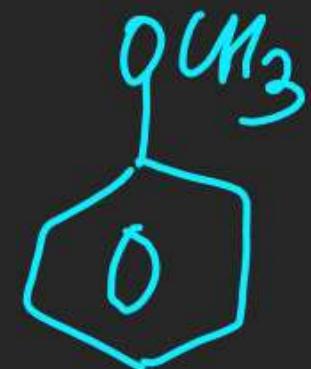
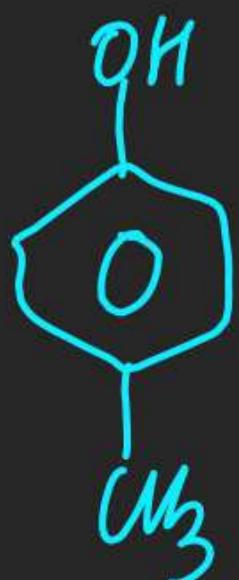
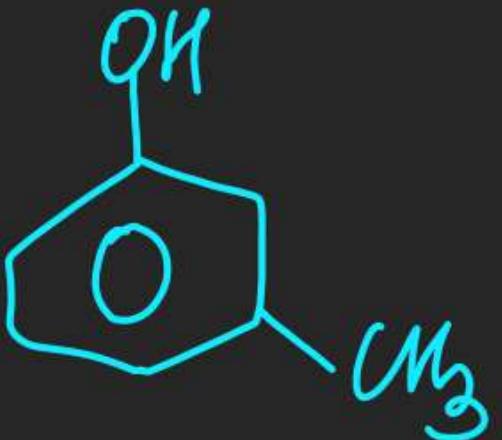
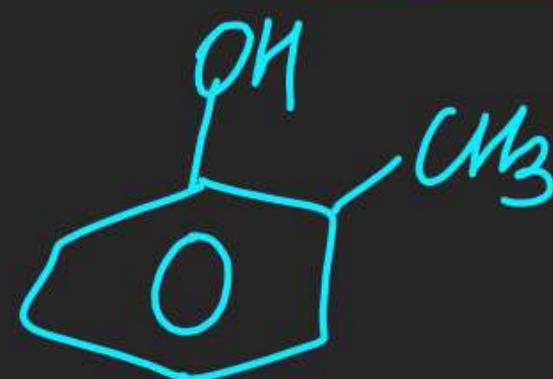


# Structural Isomerism

1. How many **benzenoid isomer** are possible for molecular formula of cresol?

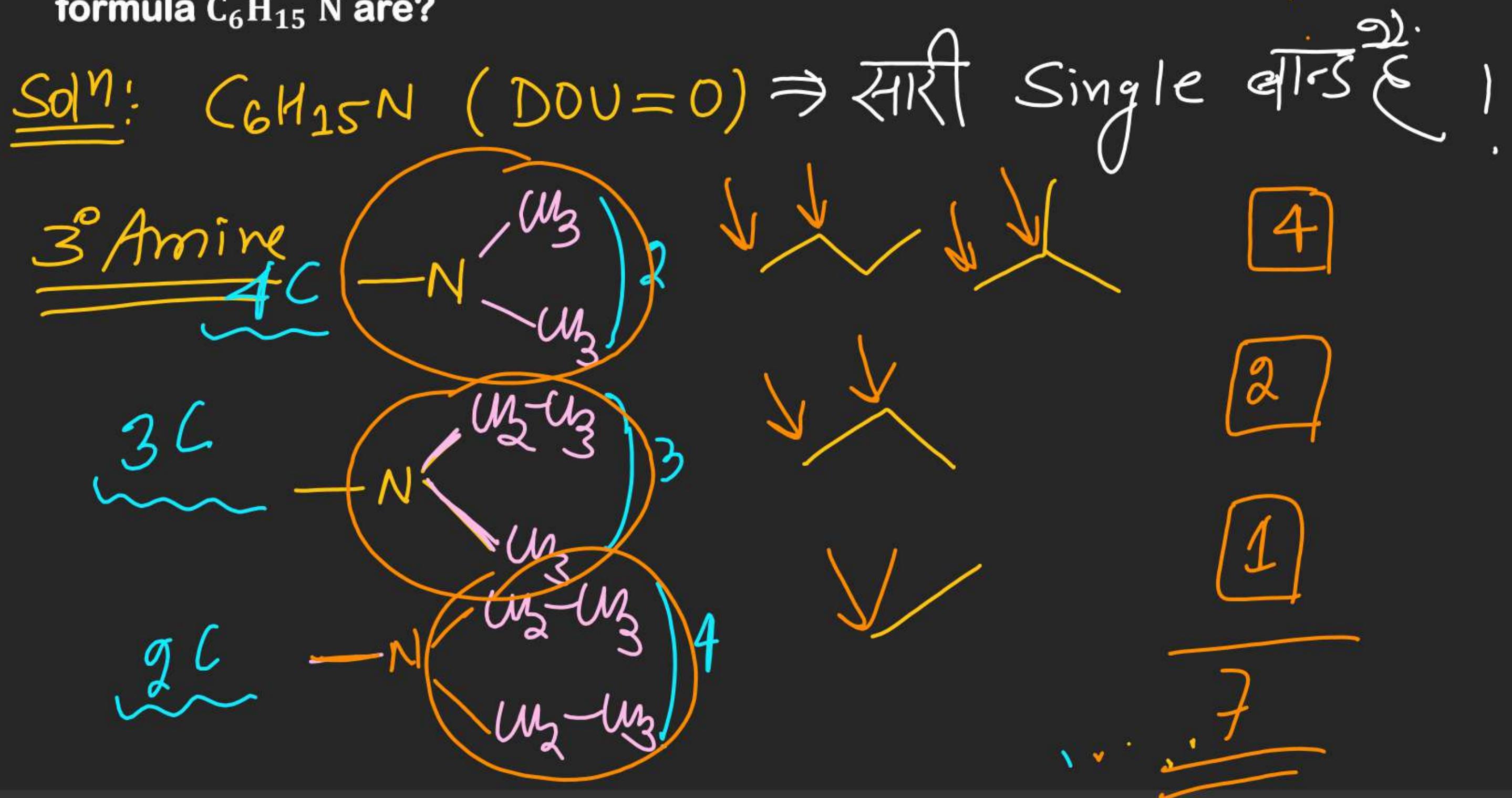
Soln!



"5"

# Structural Isomerism

4. Calculate the total number of structural isomers of  $3^{\circ}$ -amines for the molecular formula  $C_6H_{15}N$  are?



## Structural Isomerism

6. Mention the specific type of isomerism exhibited by each of the following pairs:

(a) 1,2-dichloro ethane and 1,1-dichloro ethane

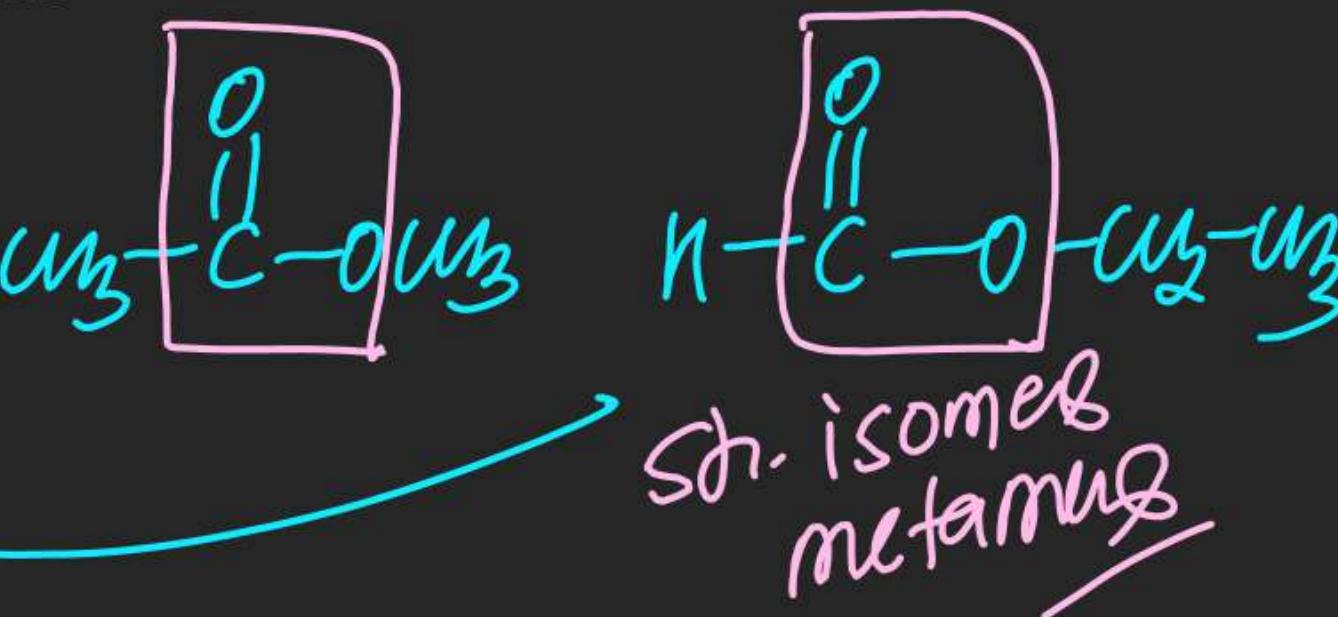
(b) Propanoic acid and methyl acetate

(c) Methyl acetate and ethyl formate

(d) o-Nitrophenol and P-nitrophenol

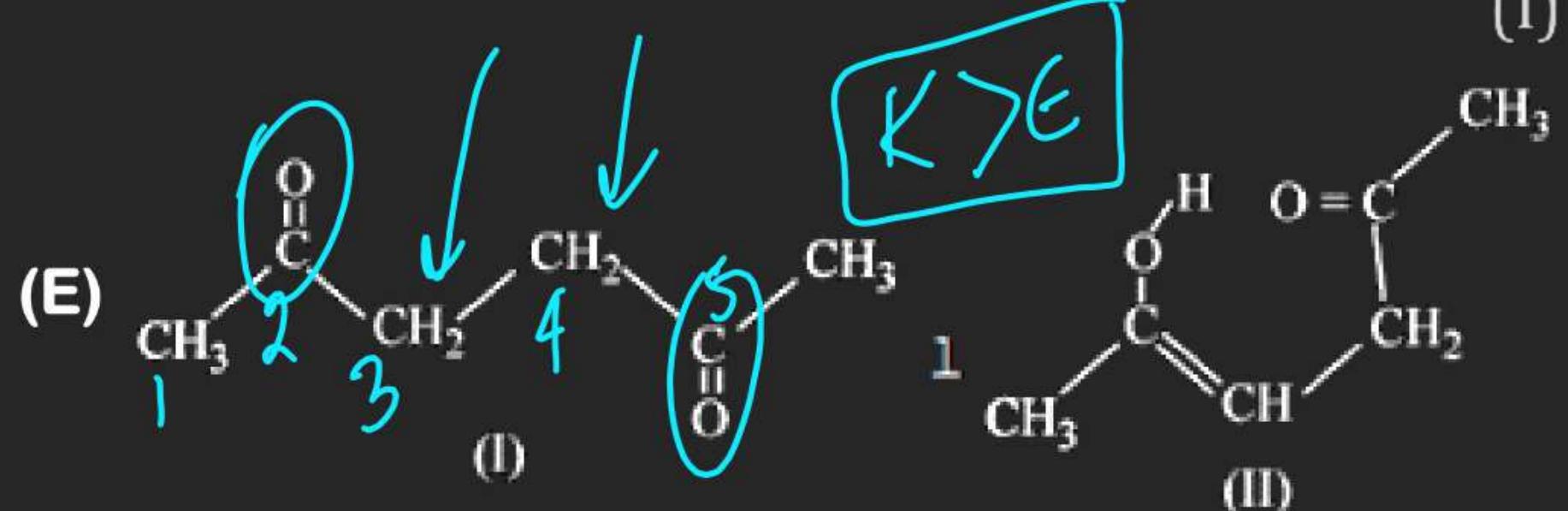
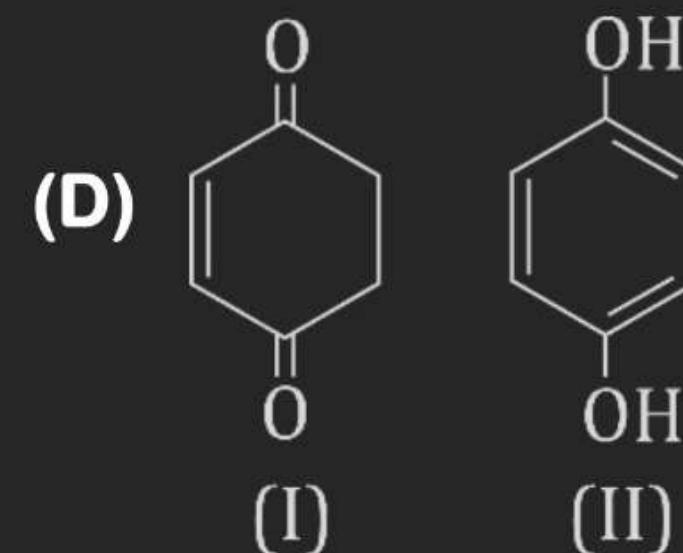
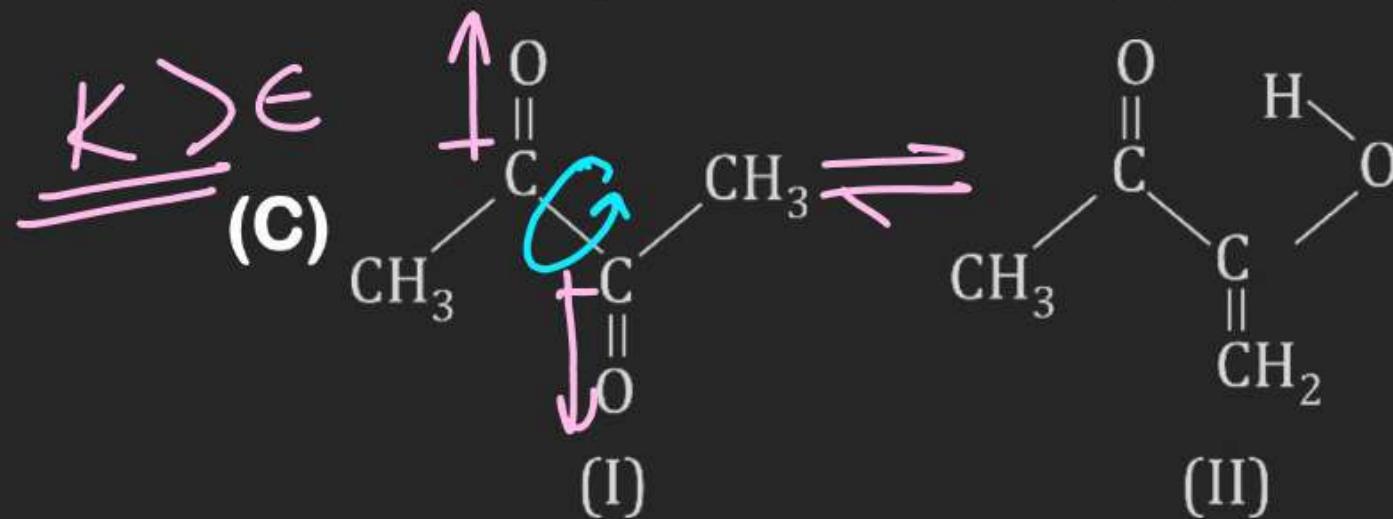
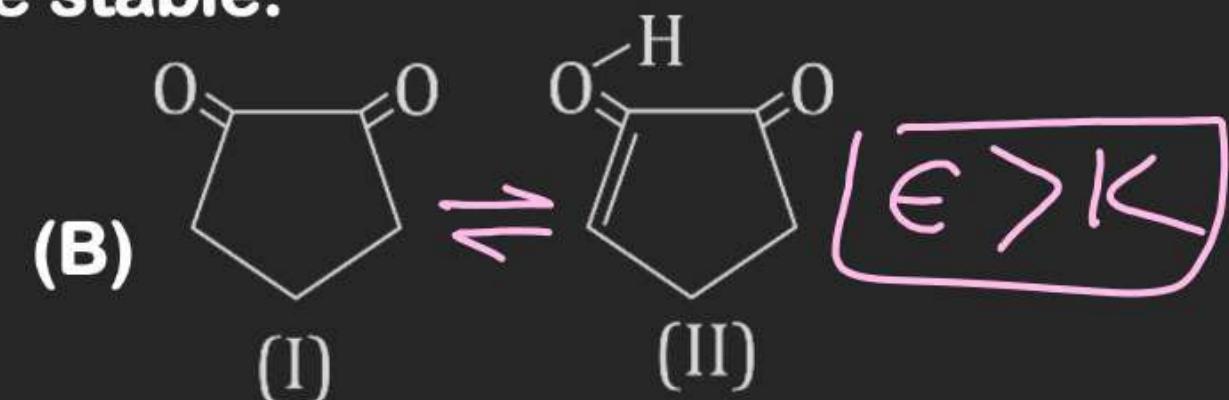
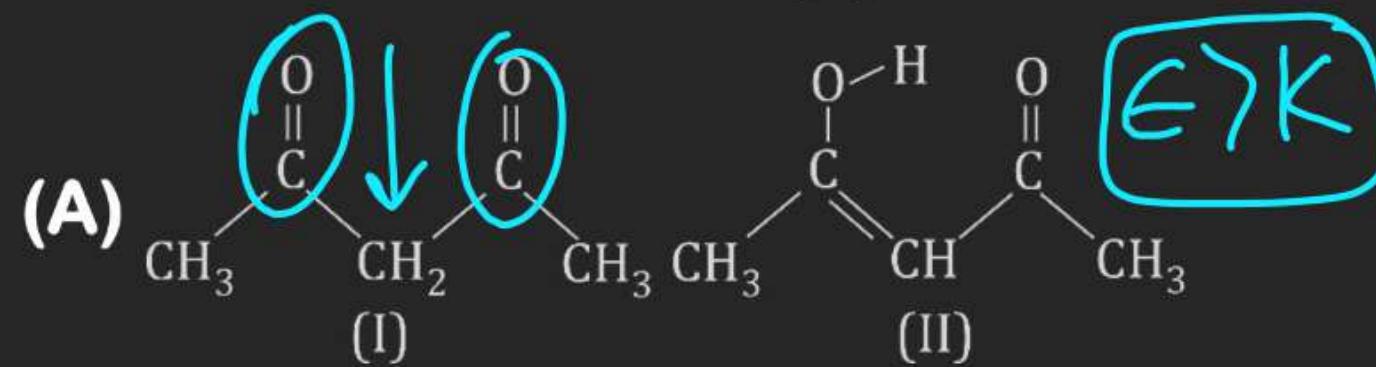
(e) Anisole and o-cresol

(f) Phenol and Cyclohexa-2,4-dien-1-one



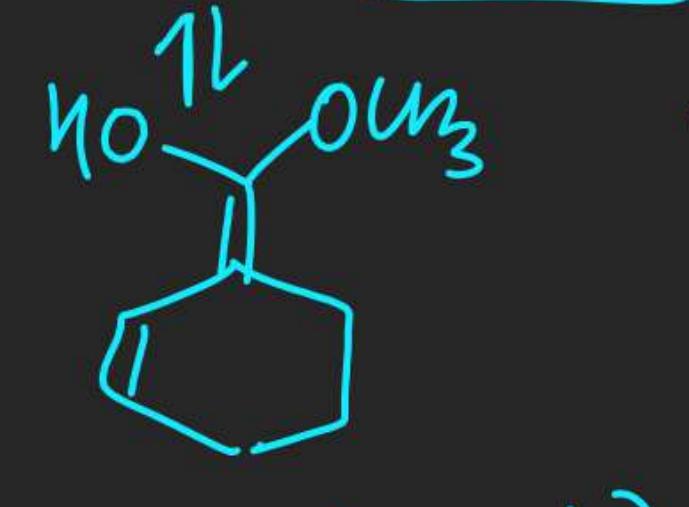
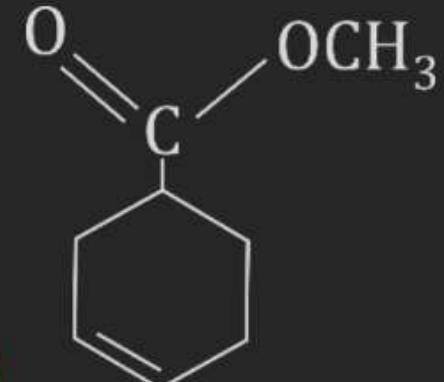
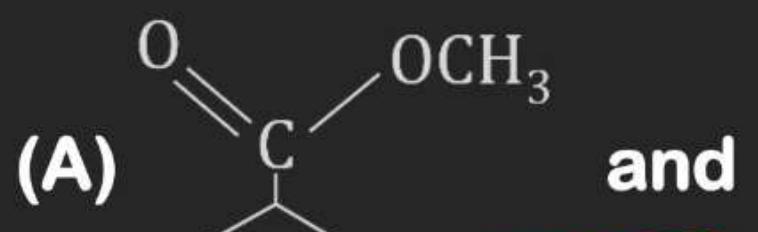
# Structural Isomerism

8. In each of the following pairs which is more stable:

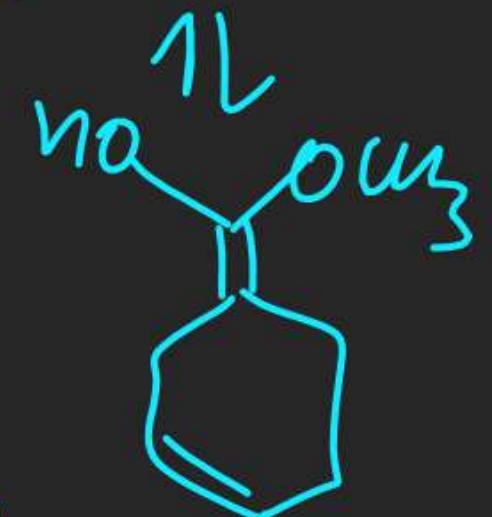


# Structural Isomerism

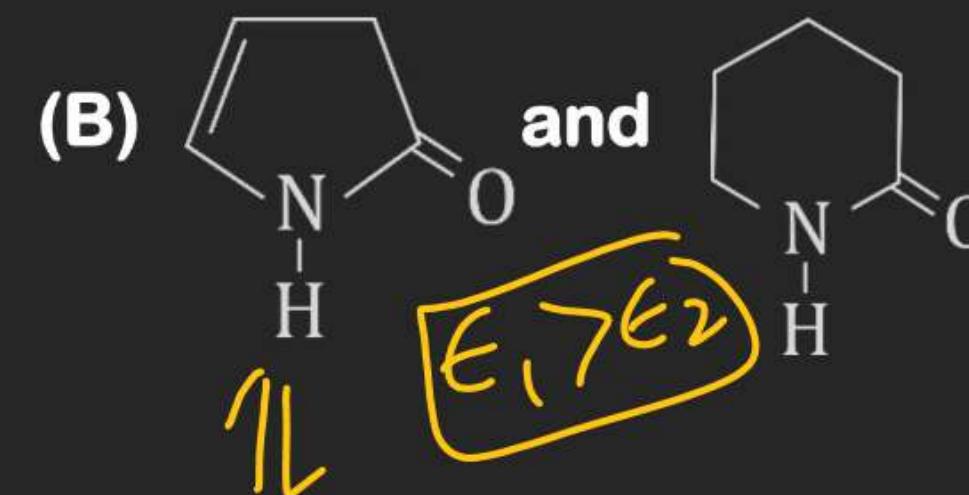
10. In each of the following pairs which will have less enol content:



Conjugate



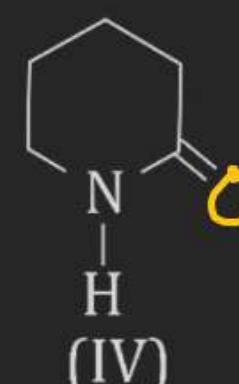
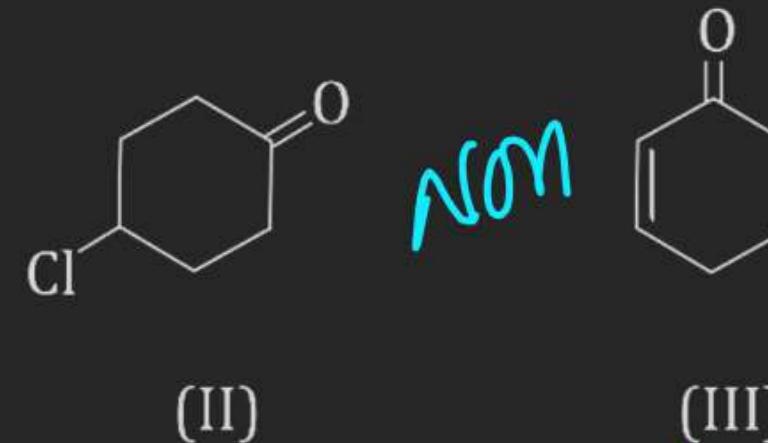
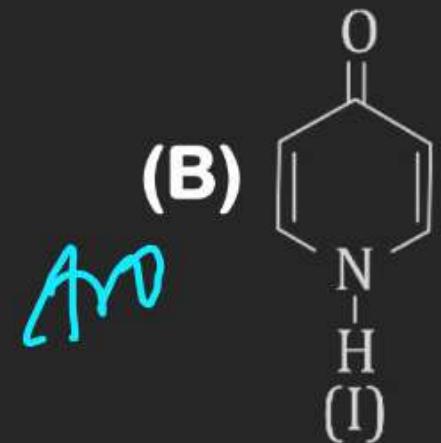
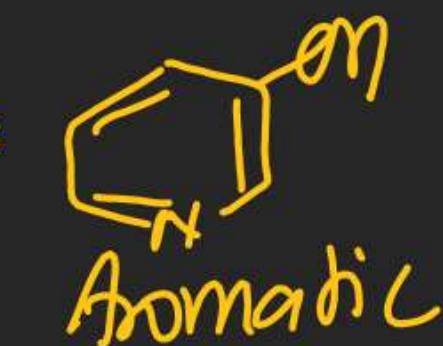
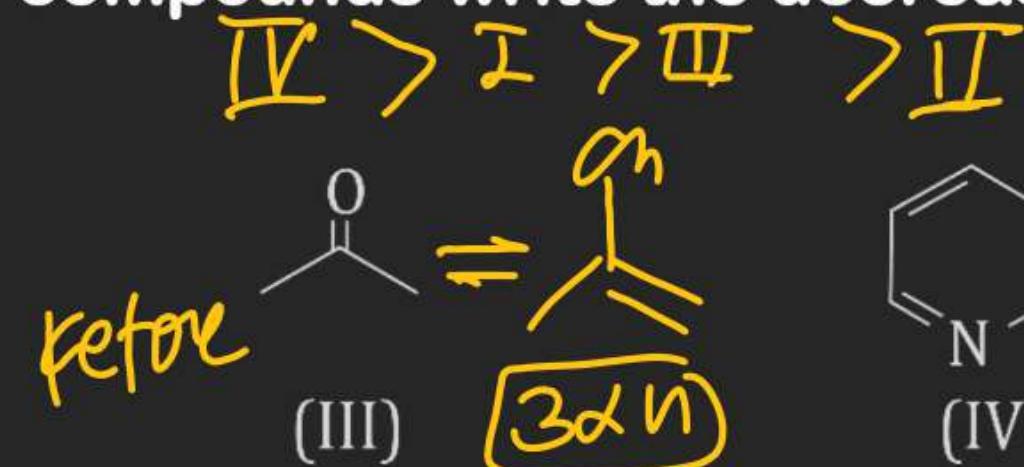
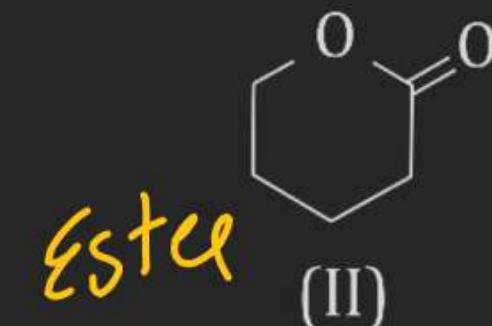
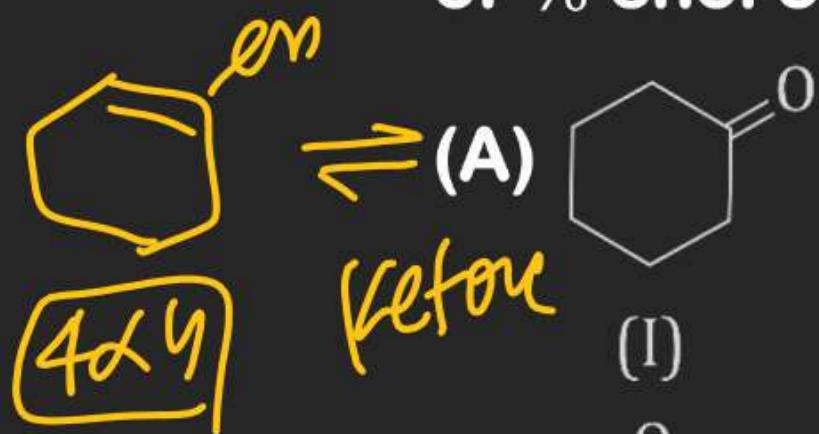
Not Conjugate



$\pi$  in aromatic U

# Structural Isomerism

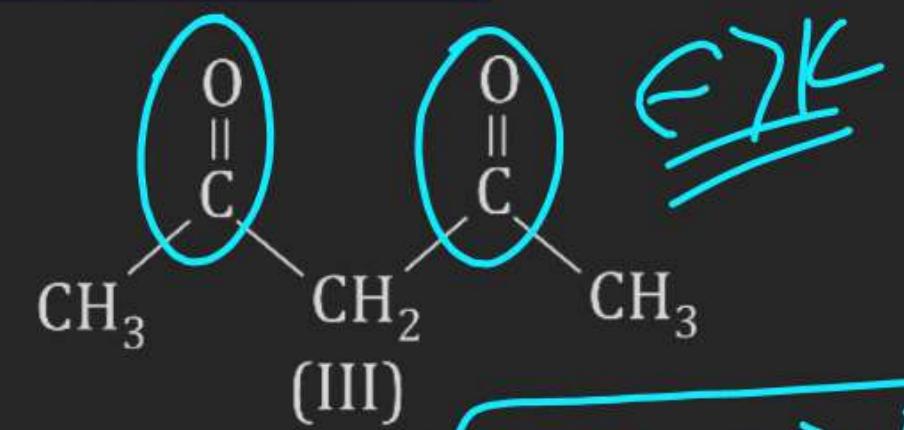
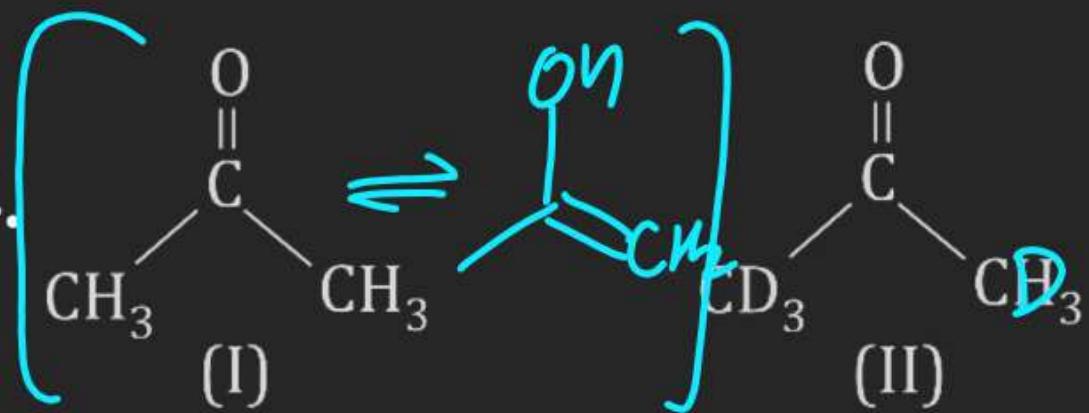
12. In each of the following sets of compounds write the decreasing order of % enol content.



$\text{III} > \text{I} > \text{IV} > \text{II}$

## Structural Isomerism

14.



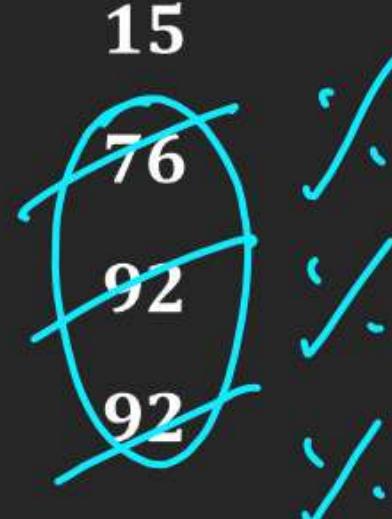
IV > II > I

Among these give ease of enolization.

# Structural Isomerism

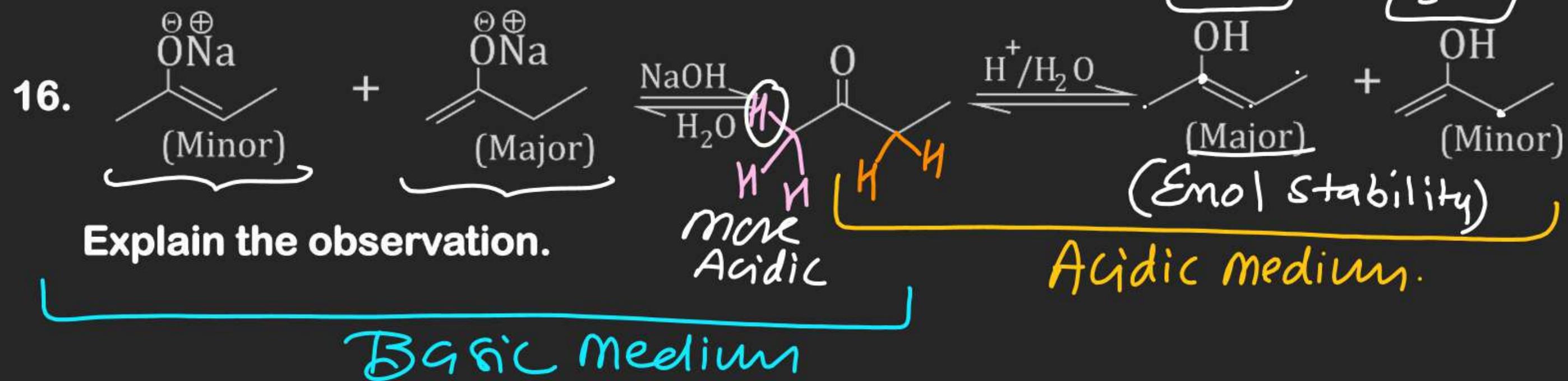
15. % enol content of acetylacetone in following solvents is found as:

Solvent	% enol content
H <sub>2</sub> O	15
Liquid state	76
hexane	92
gas phase	92



Explain the observation.

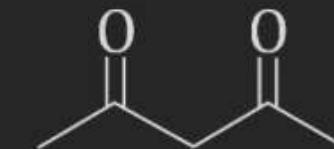
# Structural Isomerism



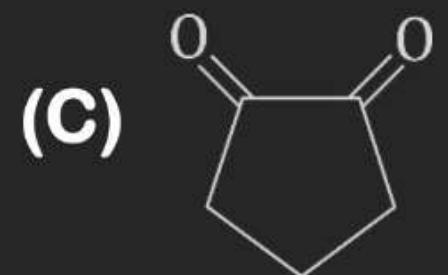
# Structural Isomerism

17. Decreasing order of enol content of the following. (along with proper explanation).

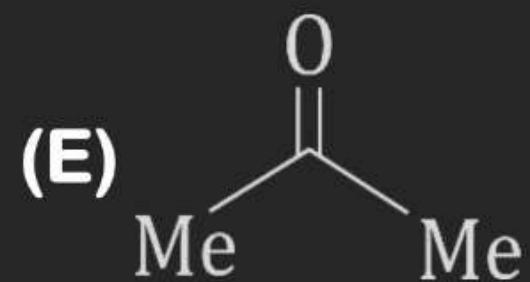
(A)



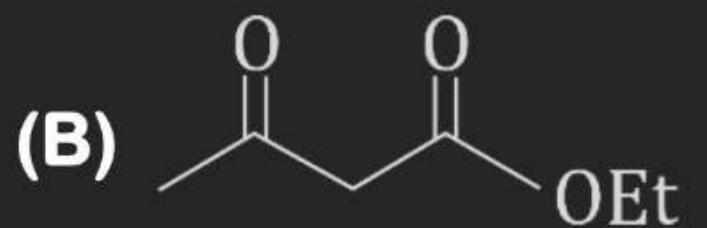
(C)



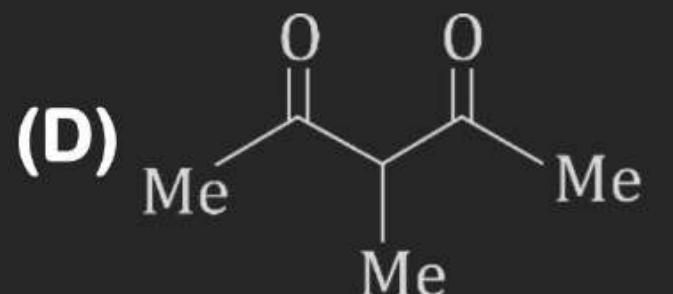
(E)



(B)

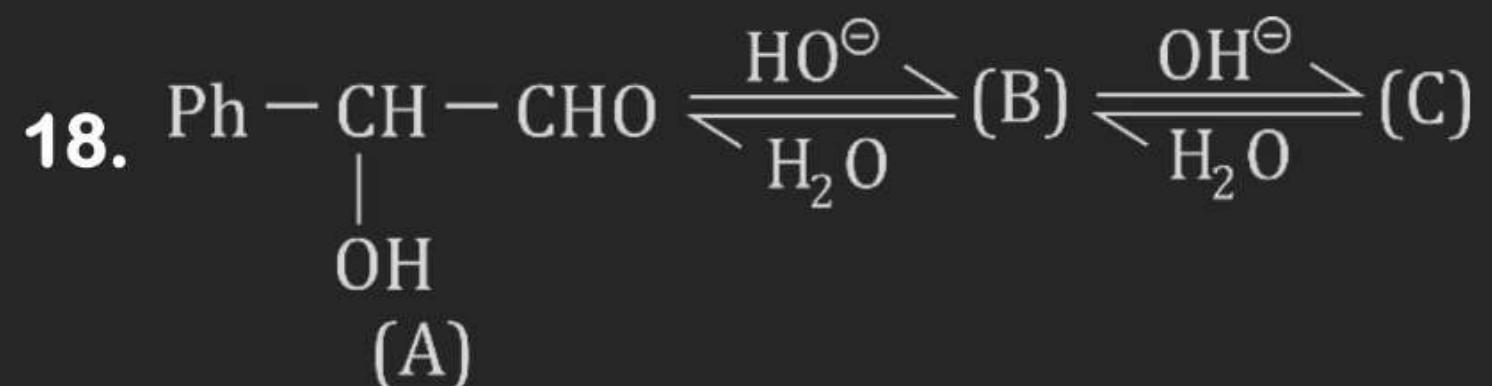


(D)

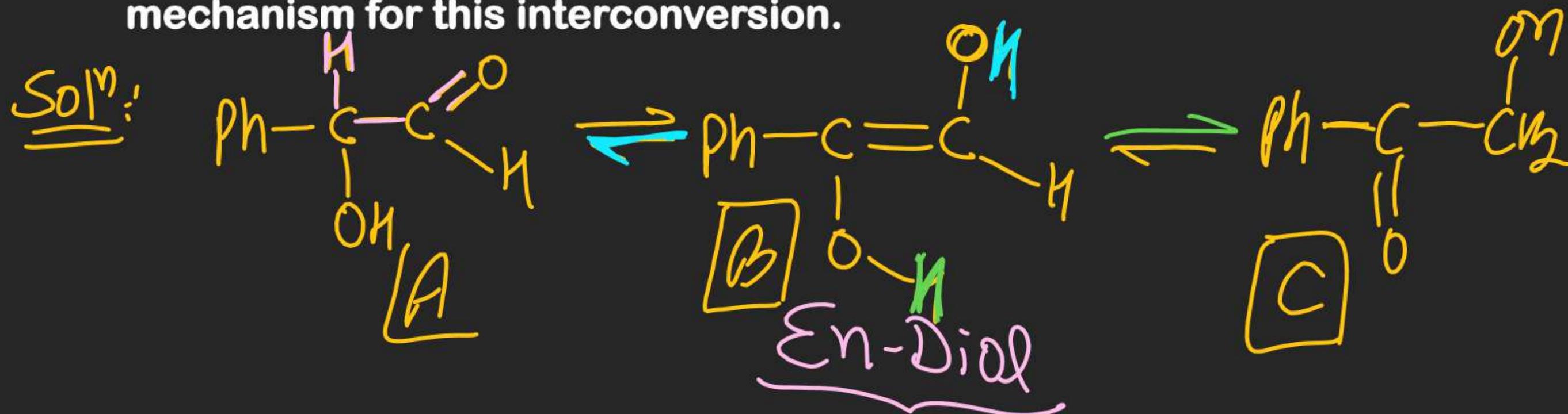


C > A > D > B > E

# Structural Isomerism



(A), (B) and (C) are structural isomers and isomerization is effectively carried out by trace of base. Give structure of (B) and (C) and also write base catalysed mechanism for this interconversion.



# Stereo Isomerism

**Q.2** If 'X' is total number of plane of symmetry, 'Y' is total number of two fold axis of symmetry and 'Z' is total number of four fold alternate axis of symmetry present in  $\text{CH}_4$ . Then find the value of  $(X + Y - Z)$ .

(A) 3

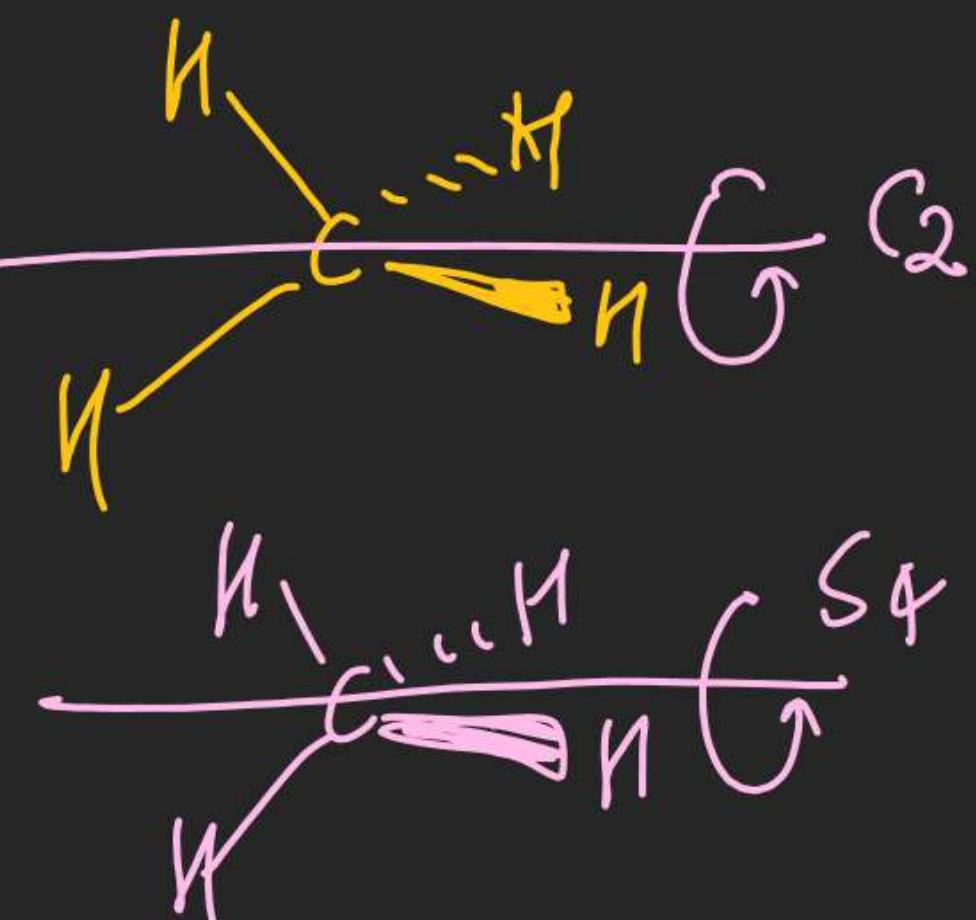
(B) 4

(C) 5

(D) 6

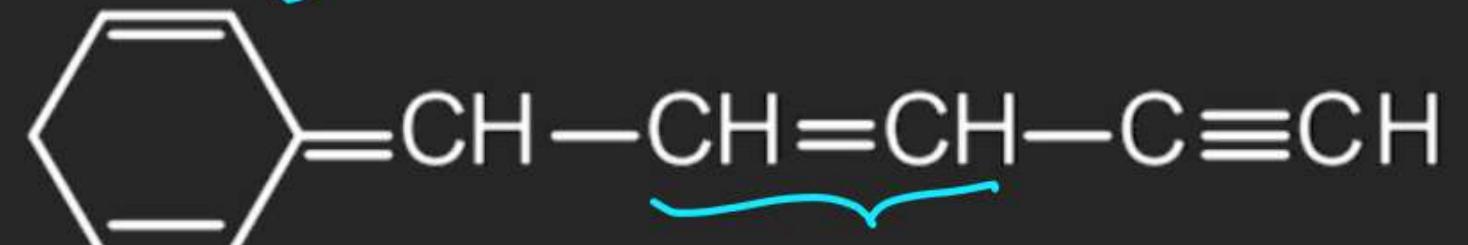


$$\begin{aligned}
 X &= \text{POS} = 6 \\
 Y &= C_2 = 3 \\
 Z &= S_4 = 3 \\
 \hline
 X+Y-Z &= 6+3-3
 \end{aligned}$$



## Stereo Isomerism

Q.4 The number of cis-trans isomer possible for the following compound.



(A) 2

(B) 4

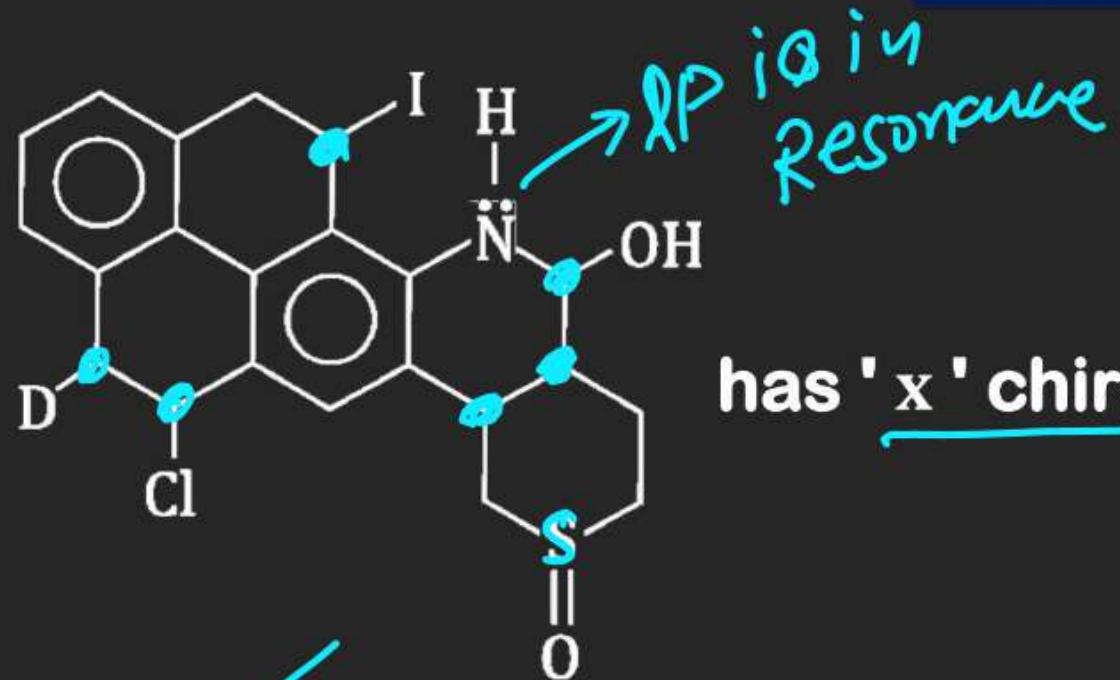
(C) 6

(D) 8

Cis & Trans

## Stereo Isomerism

Q.5



has 'x' chiral centre then find the value of x :

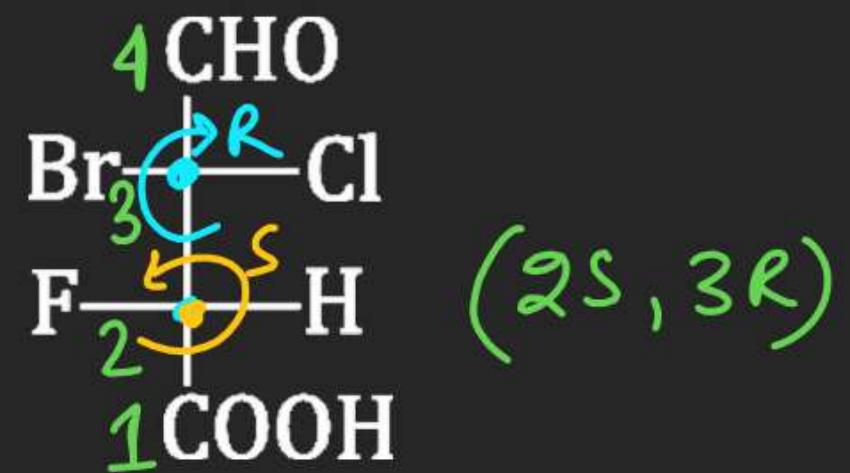
(A) 7

(B) 8

(C) 6

(D) 5

Q.8



Configuration of compound is :

(A) 2S, 3 S

(B) 2R, 3 S

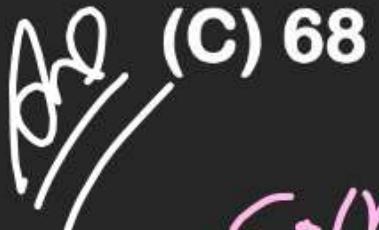
(C) 2R, 3R

(D) 2 S, 3R

# Stereo Isomerism

Q.10 Minimum molecular weight of a hydrocarbon containing minimum number of C-atom to show optical isomerism.

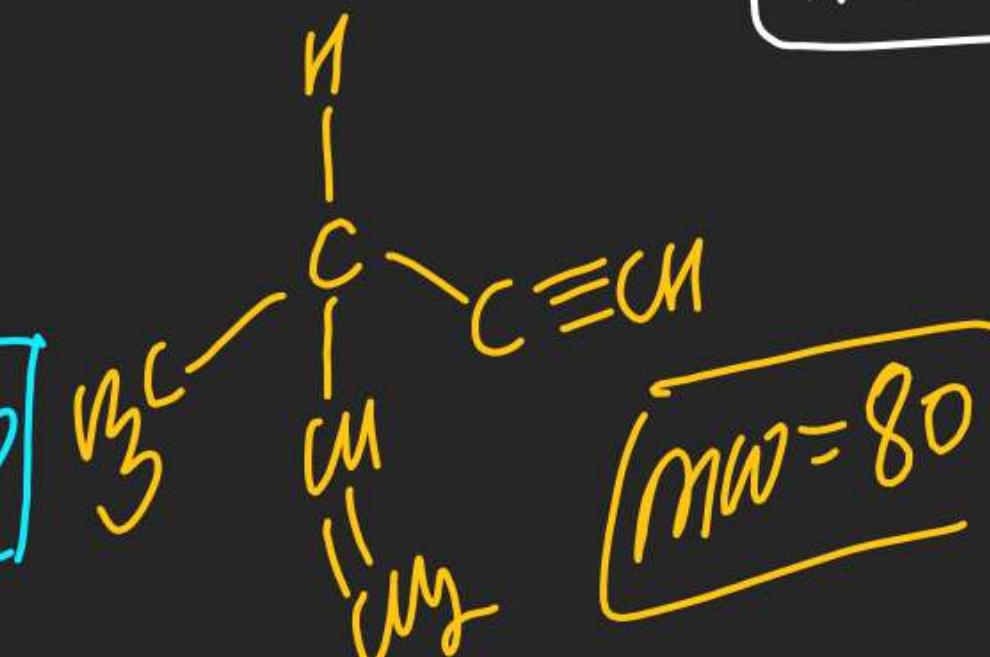
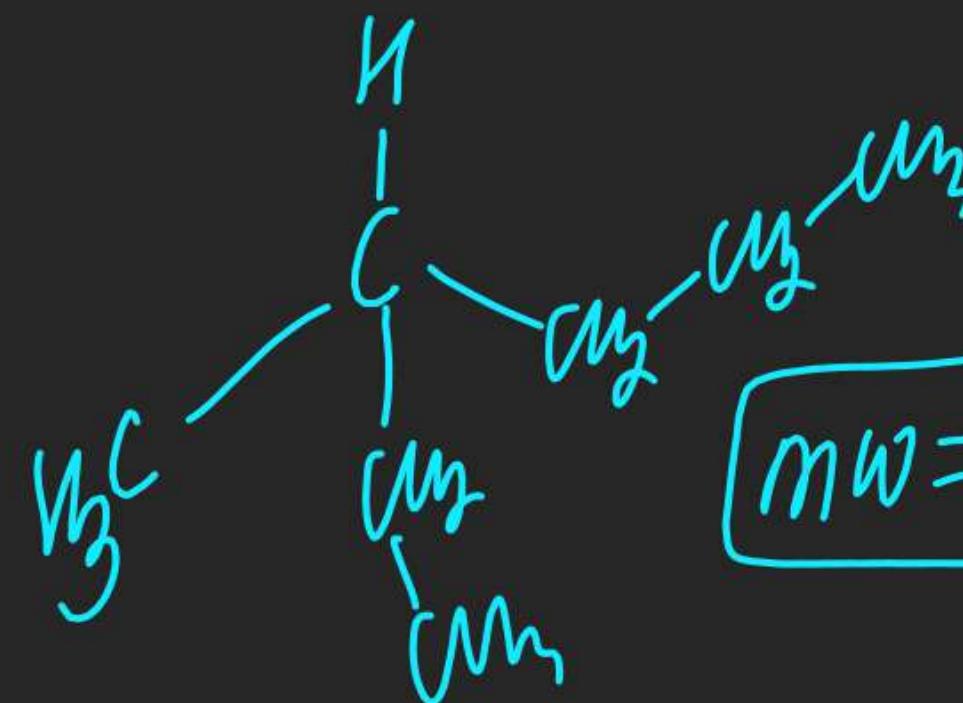
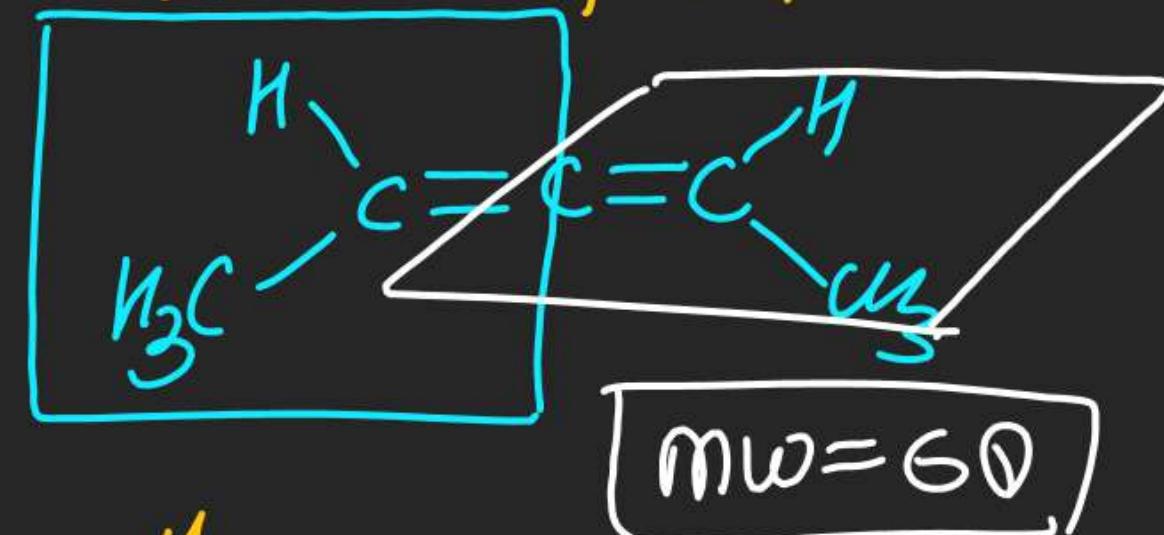
- (A) 100
- (B) 80
- (C) 68
- (D) 70



Sol'n:

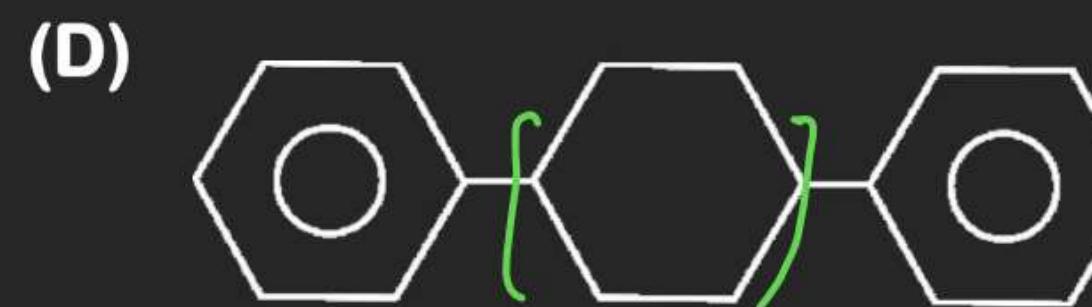
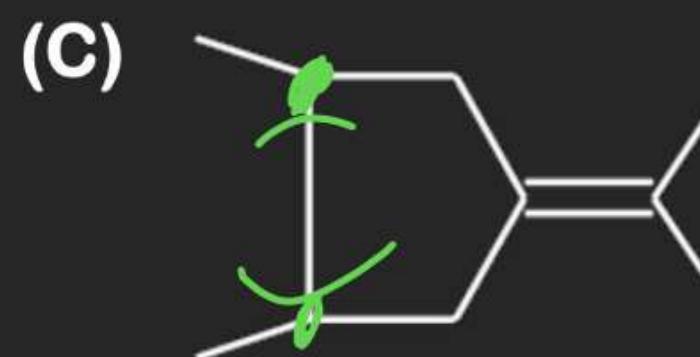
Compound must have  
Chiral centre

Compound may have  
absence of Sn



# Stereo Isomerism

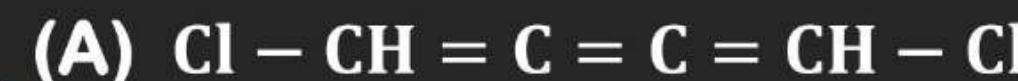
Q.11 Compounds which can show both optical as well as geometrical isomerism:



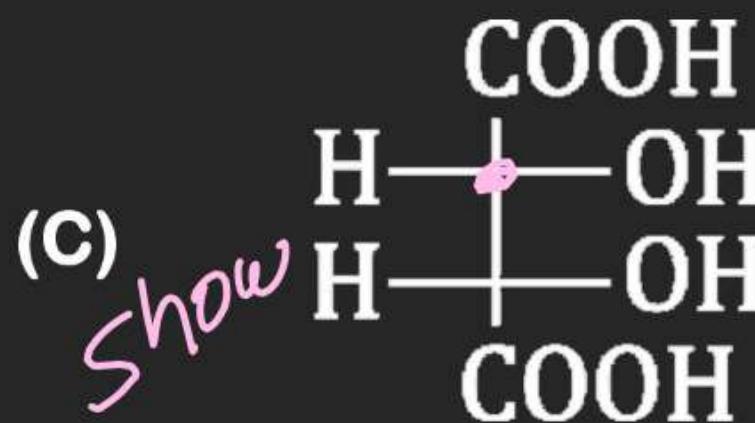
# Stereo Isomerism

**Q.14** Which of the following will not show optical isomerism.

~~(A)~~

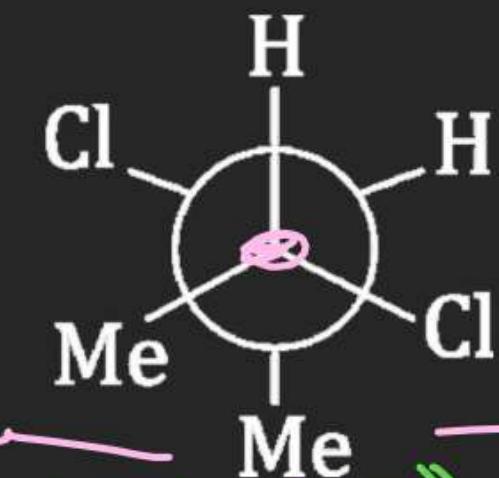


(C)



Show

(D)



(i)

Chiral centre present

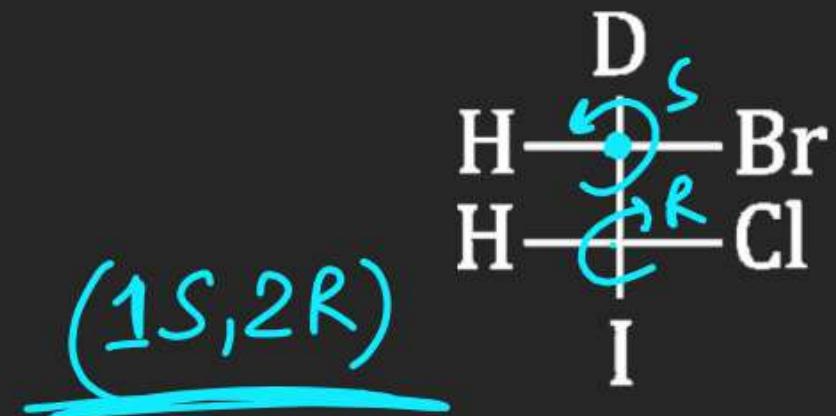
(ii)

Sn absent

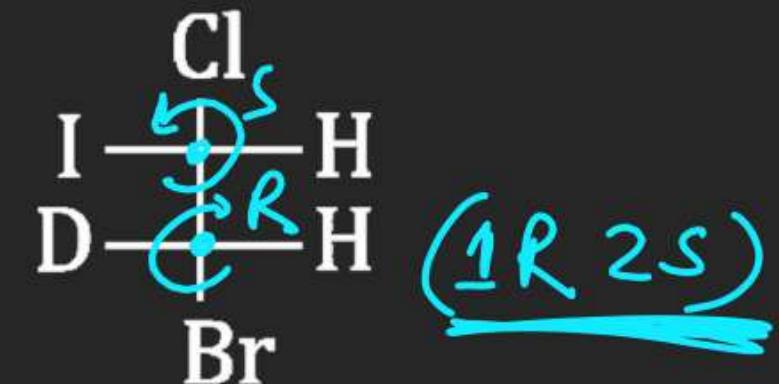
$\Rightarrow$  wd show optical

# Stereo Isomerism

Q.16 The two compounds given below are :



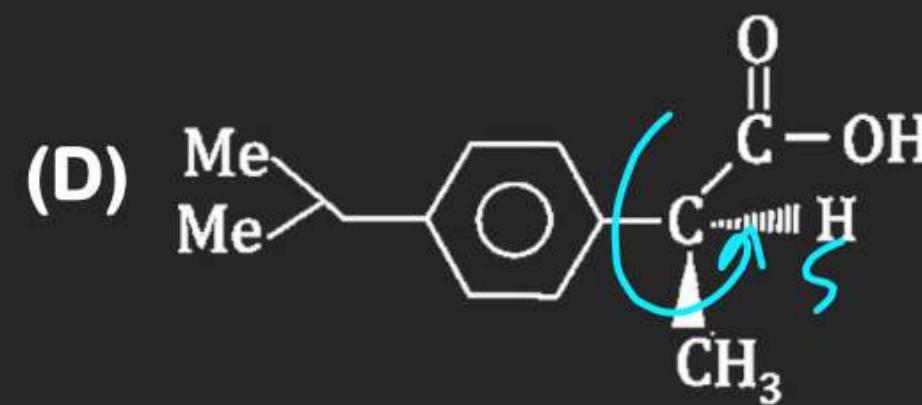
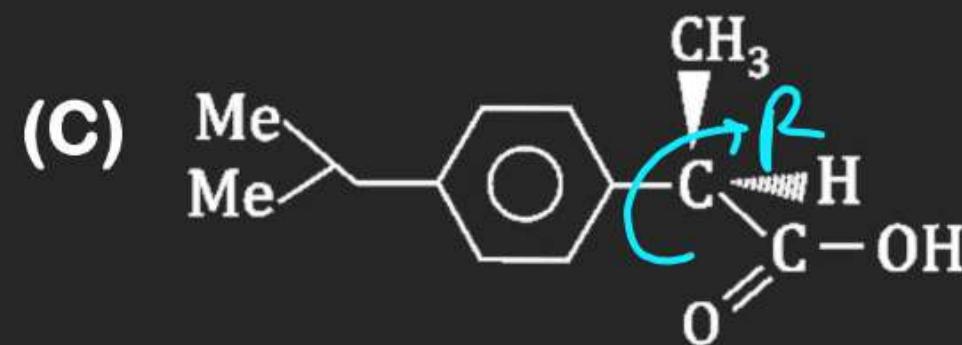
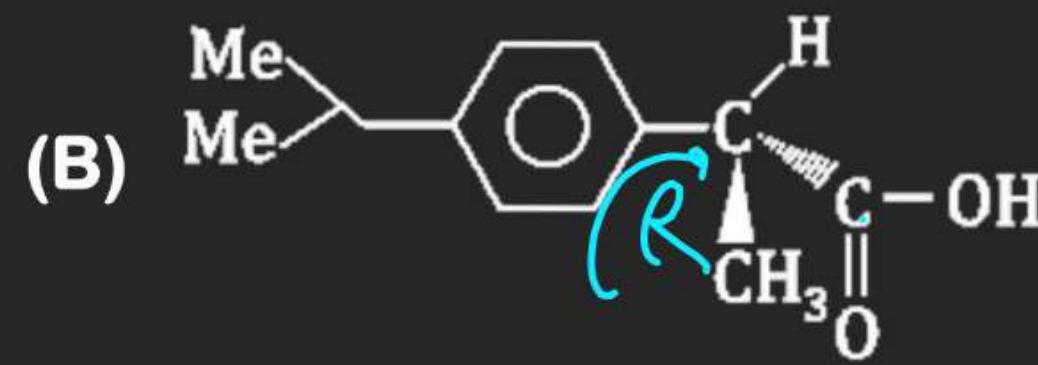
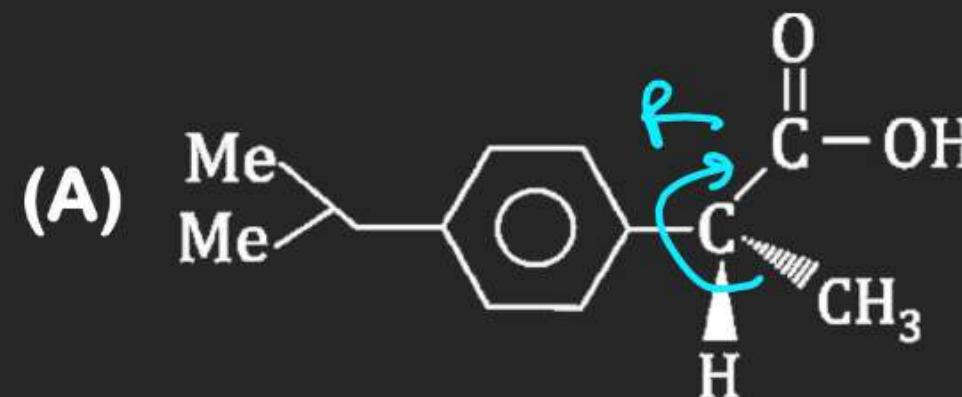
- (A) Enantiomers  
(C) Optically inactive



- (B) Diastereomers  
(D) Identical

# Stereo Isomerism

Q.18 The S-ibuprofen is responsible for its pain reliving property. Which one of the structures shown is S-ibuprofen :



# Stereo Isomerism

Q.20 Number of possible stereoisomers of glucose are:

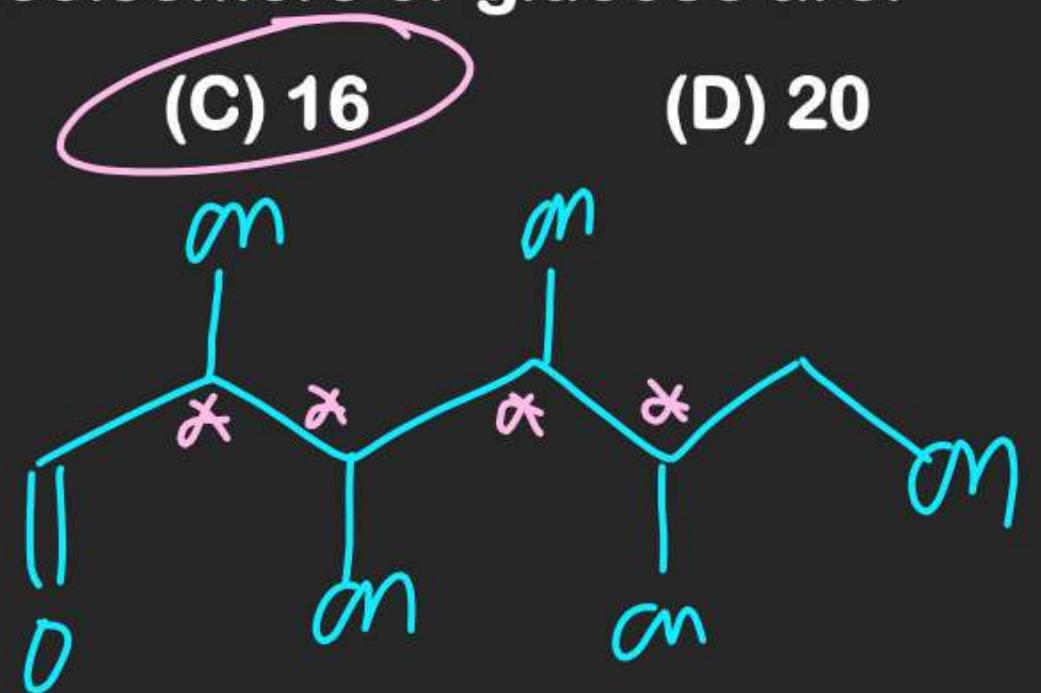
(A) 10

(B) 8

(C) 16

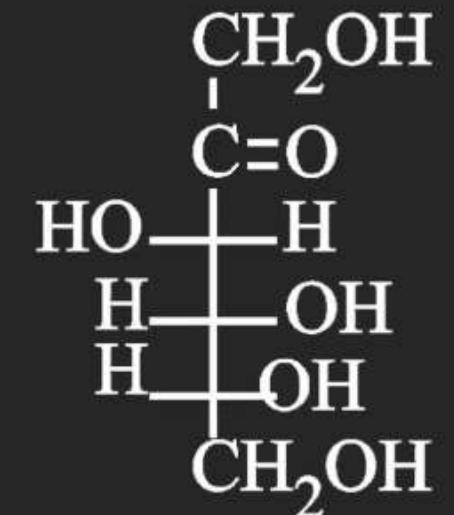
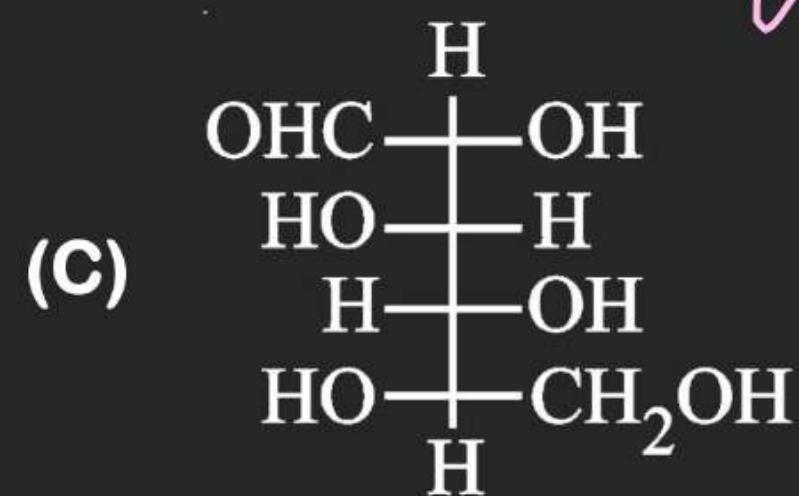
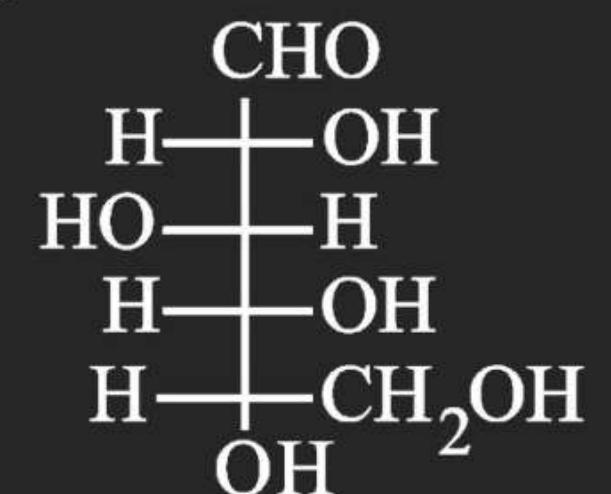
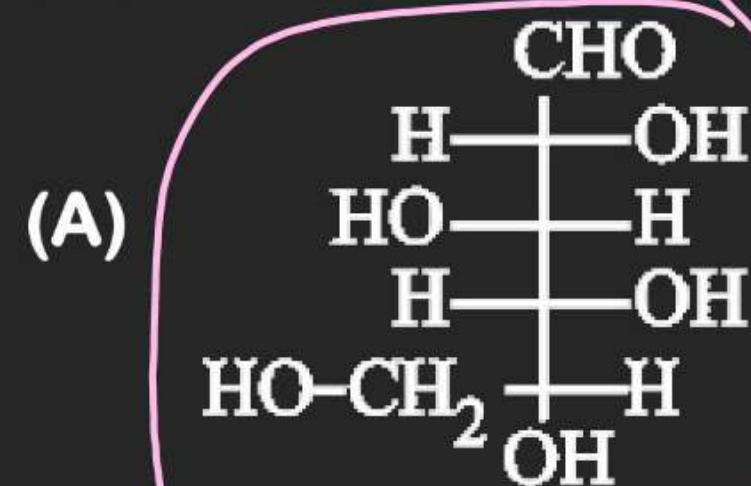
(D) 20

$$\begin{aligned} &= 2^4 \\ &\equiv 16 \end{aligned}$$



# Stereo Isomerism

Q.21 Which of the following is not D sugar :



HW

Stereo sheet

$\alpha$ -1 Complete

$\alpha$ -2 Complete

$\alpha$ -3 An