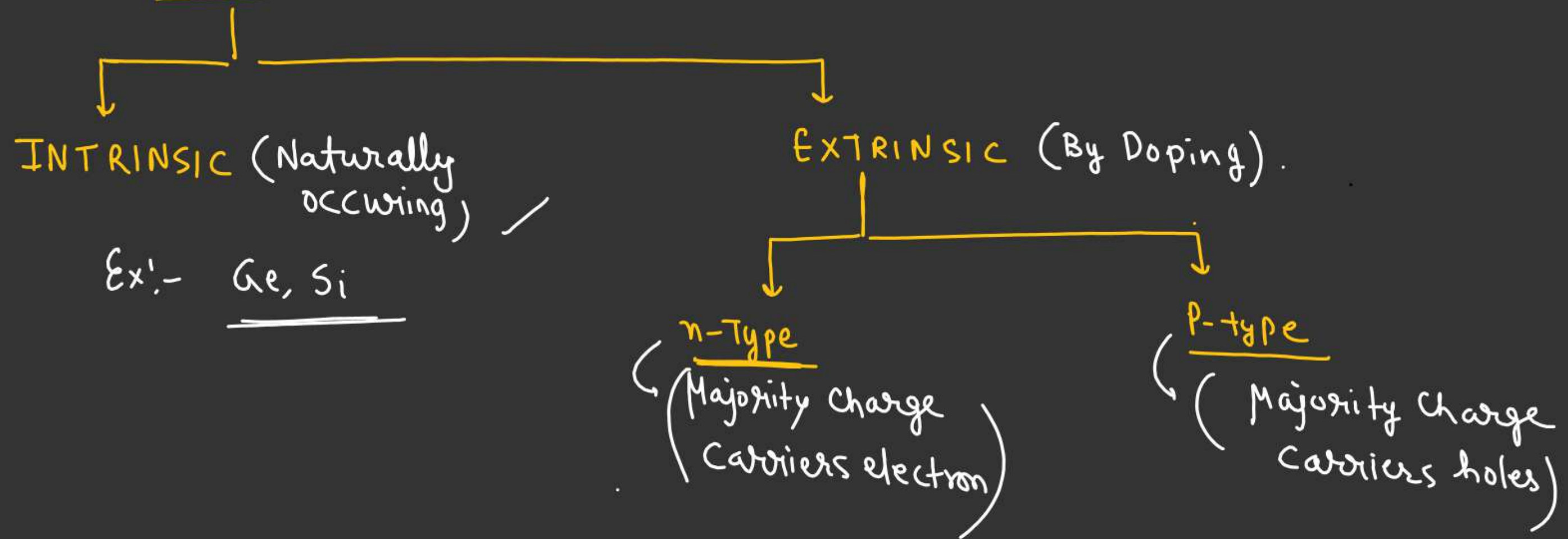


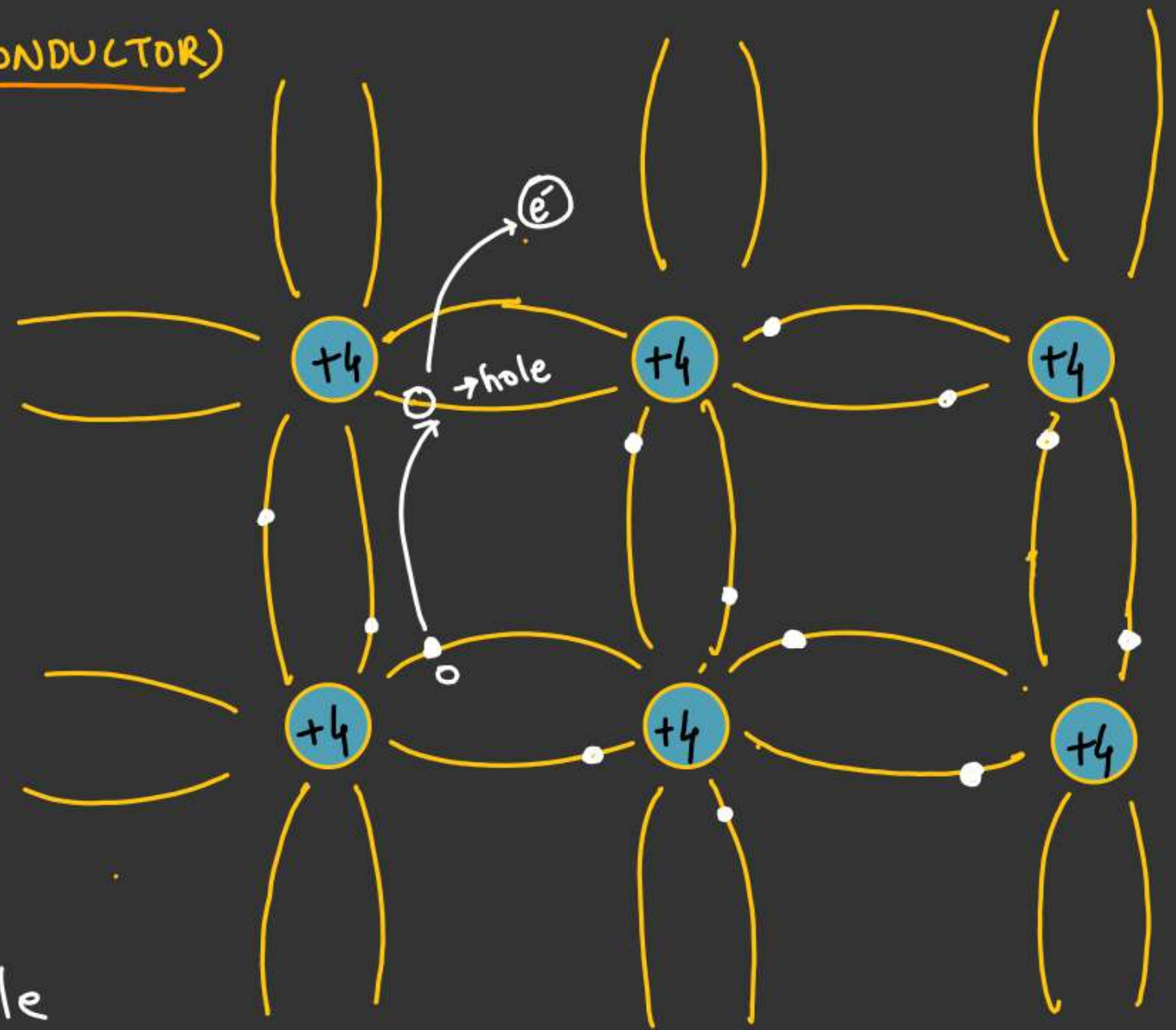
SEMI CONDUCTOR

- Much higher resistivity than metal.
- Temp Coeff of resistivity -ve & high
- Charge Carriers lower than metal.

TYPE

# INTRINSIC SEMI CONDUCTOR (PURE SEMI CONDUCTOR)

- Free from impurities ✓
- At OK behave as insulators  
↳ Temp.
- At high temp, the thermal energy of valance electron increases.  
An electron may break away from the Co-valent bond & become free to conduct



- To free an electron from Co-valent bond and simultaneously creation of hole required a kind of ionisation energy  $E_g$ .

$$n_e = C e^{-E_g/2KT}$$

✓  
No of electrons,  
set free

For given  $E_g$ ,  $n$  increases as the temp increases.

→ Each free electron creates one hole, so in an intrinsic semiconductor no of holes ( $n_h$ ) and no of electrons  $n_e$  are equal.

$$\underline{n_e = n_h = n_i}$$

→ Total current in intrinsic semiconductor is  $I = \underline{(I_e + I_h)}$  ✓



## DOPING:-

↙  
Addition of desirable impurity to a pure semiconductor so as to increase its conductivity. ✓



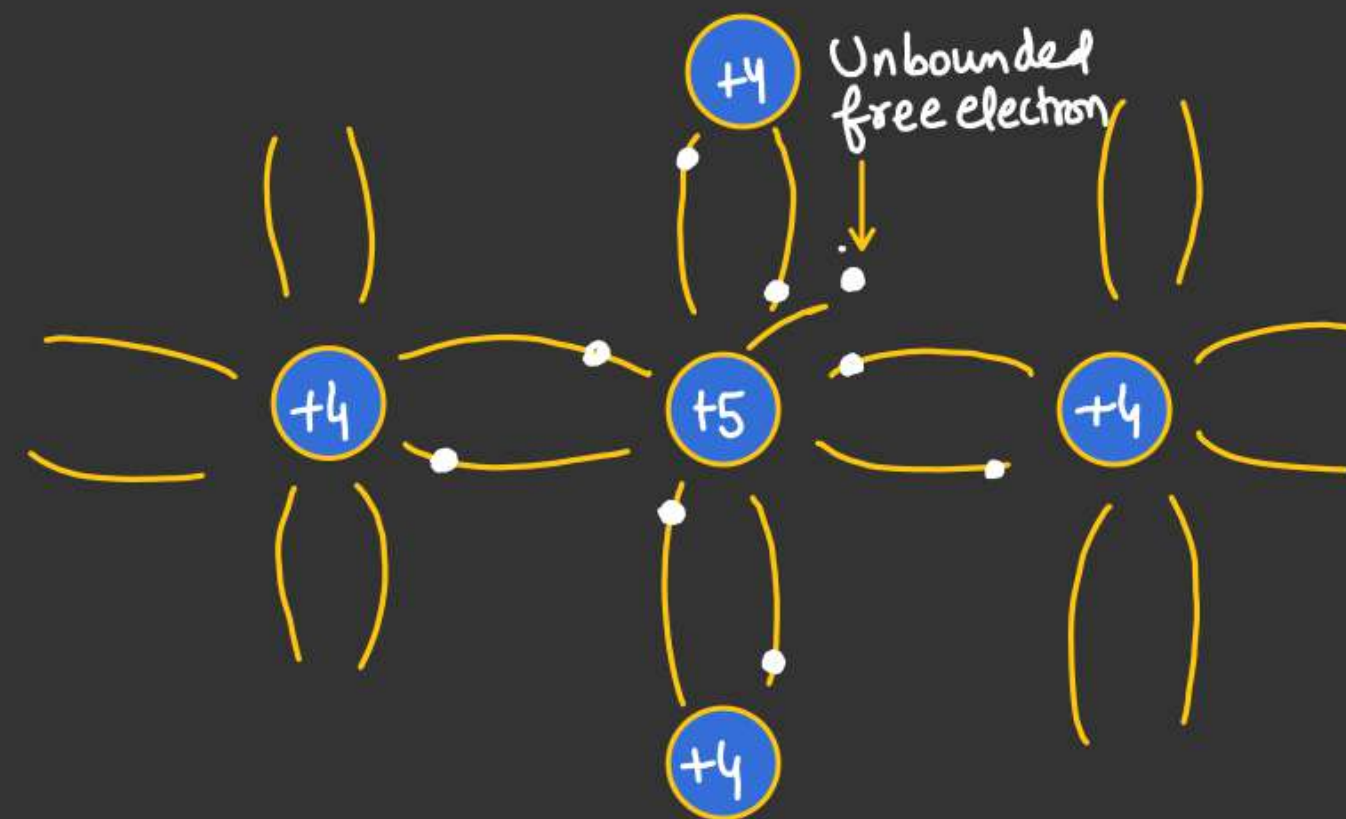
## TYPE OF DOPENTS (IMPURITY)

- ① Pentavalent → 5 Valance electrons.  
Arsenic (As), Antimony (Sb), & Phosphorus (P)
- ② Trivalent → 3 Valance bond  
Indium (In), Boron (B) & Aluminium (Al)

# EXTRINSIC SEMICONDUCTOR

## N-TYPE

- Doped with pentavalent impurity with Si or Ge.
- When pentavalent impurity atom substitutes the tetravalent Si atom it uses four of its 5 valence electrons in forming 4-covalent bonds with neighbouring Si atom while the 5<sup>th</sup> electron is loosely bound to the impurity atom.



A very small amount of ionisation energy  $\approx (0.01 \text{ eV})$  is required to free this  $e^-$ .

- N-Type semiconductors electrons are majority charge carriers.  
 $(n_e \gg n_h)$



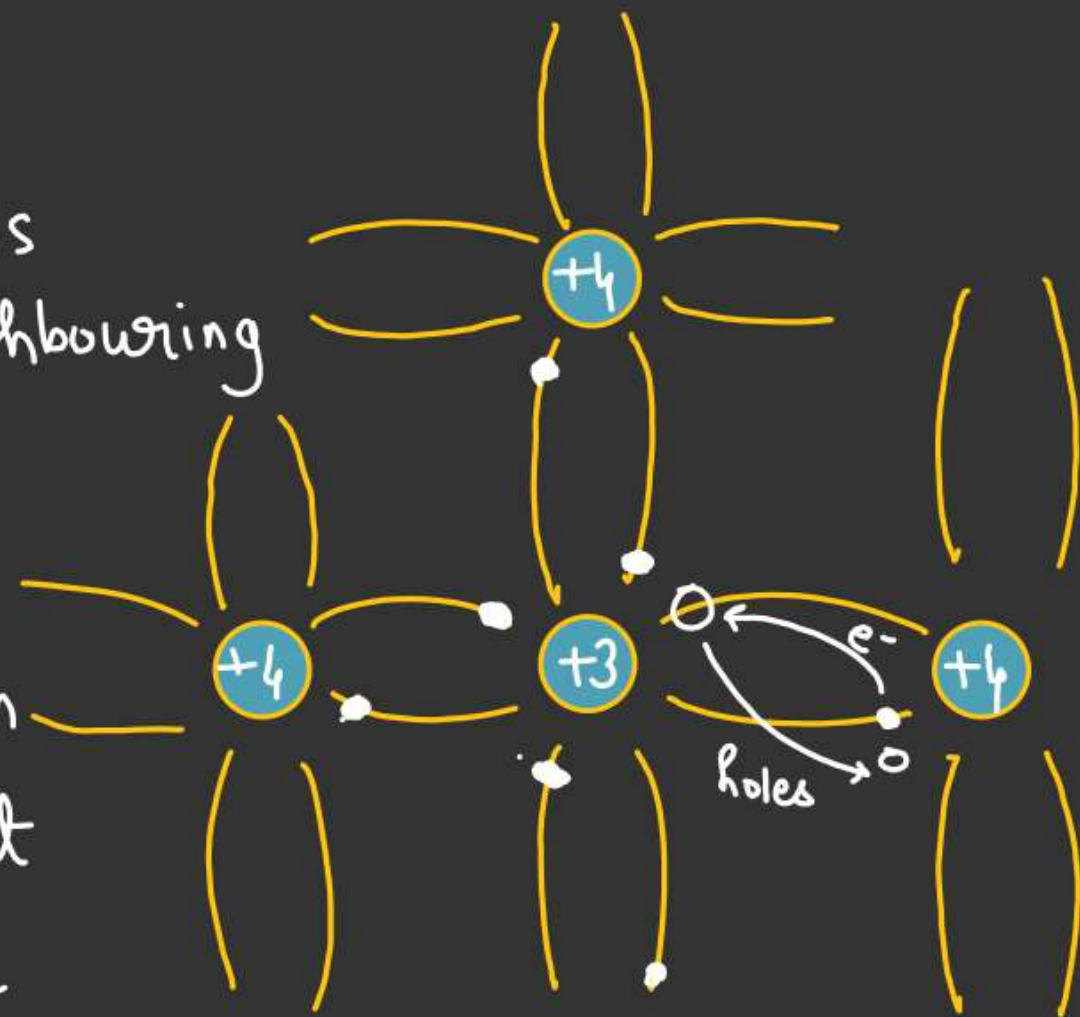
P-TYPE

→ Obtained by doping the trivalent impurity with Si or Ge.

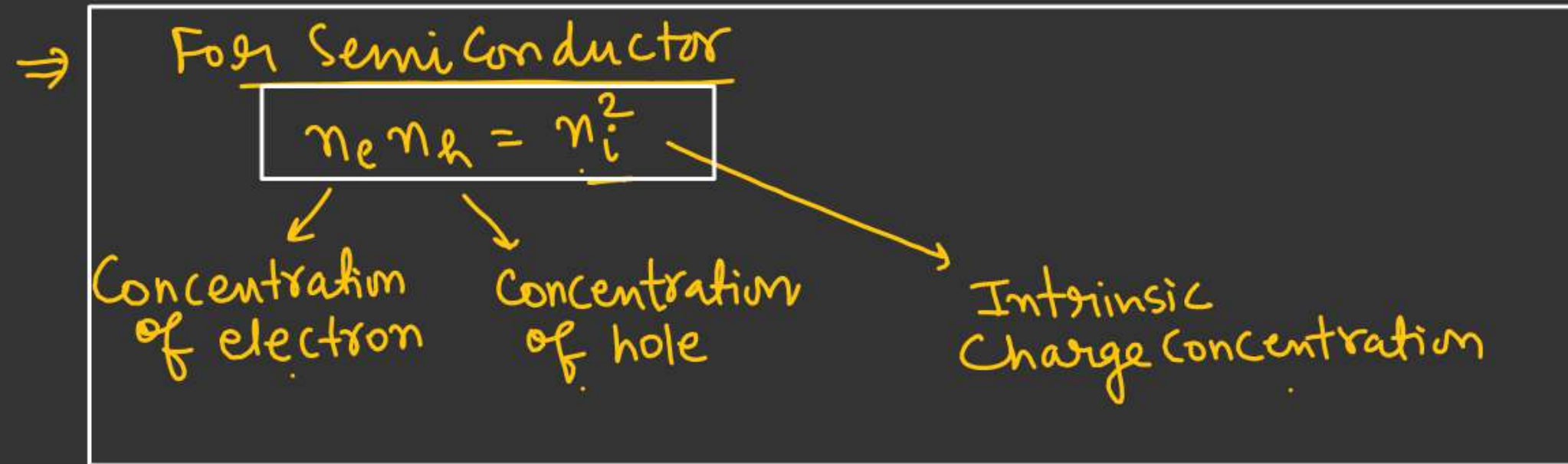
→ Impurity atoms uses its three valance electrons in forming Co-valent bonds with three neighbouring Si atoms.

one Co-valent bond with a neighbouring Si atom is left incomplete due to deficiency of atom.

An electron from neighbouring Si-Si Co-valent bond can slide into this vacant bond creating a vacancy or hole in that bond.



→ Hole is now available for conduction  
Current in P-type due to conduction of hole.  
 $(n_h \gg n_e)$



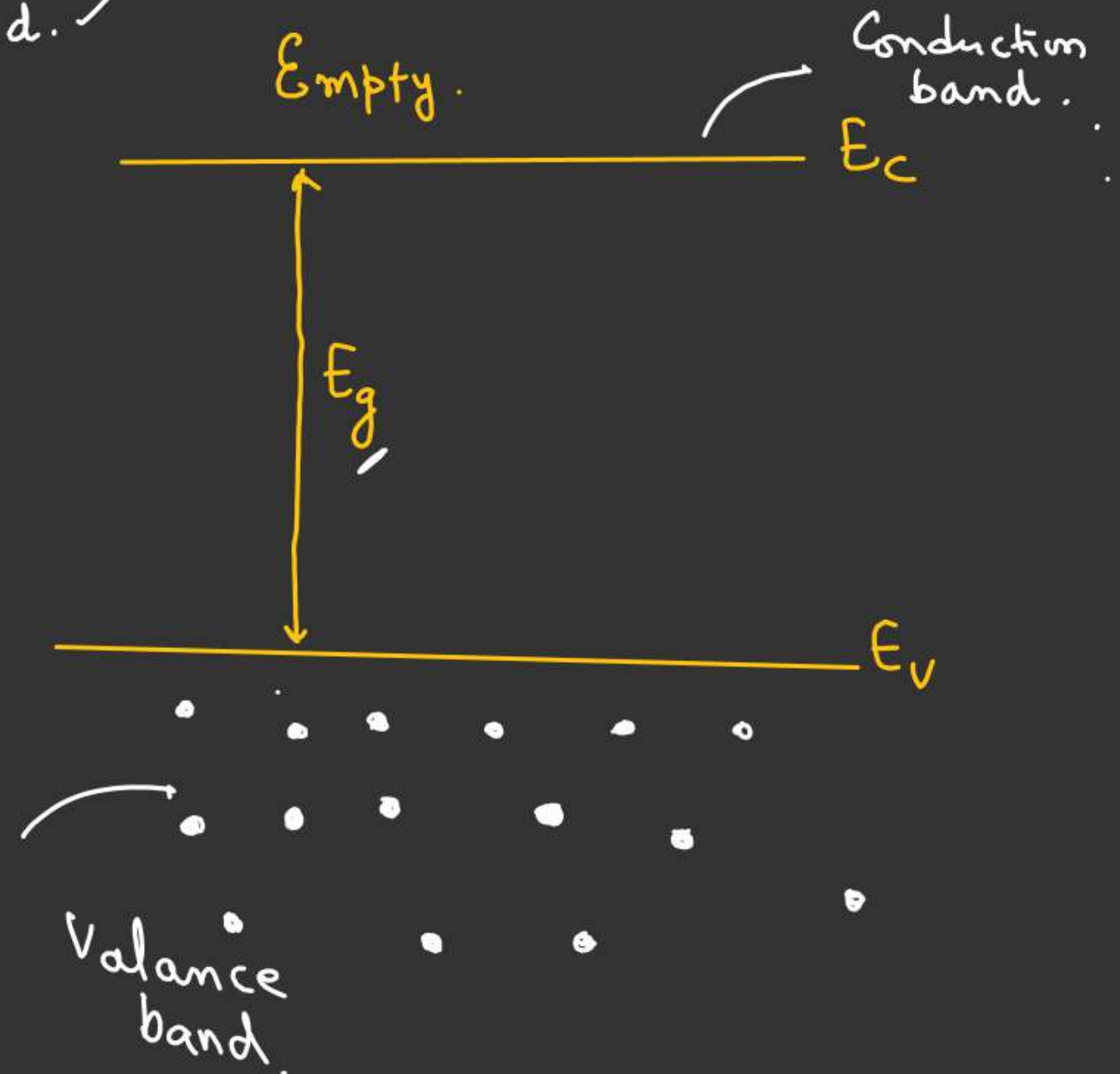
# ENERGY BAND THEORY IN SOLIDS

→ Two Energy Bands  $\begin{cases} \text{Valance Band} \\ \text{Conduction Band} \end{cases}$  ✓

→  $E_v$  = Highest Valance band energy

$E_c$  = Lowest conduction band energy

$$\underline{E_g} = E_c - E_v$$





METALS  $\Rightarrow$  (Valance band  $\rightarrow$  Completely filled  $\checkmark$   
Conduction band  $\rightarrow$  Partially filled)

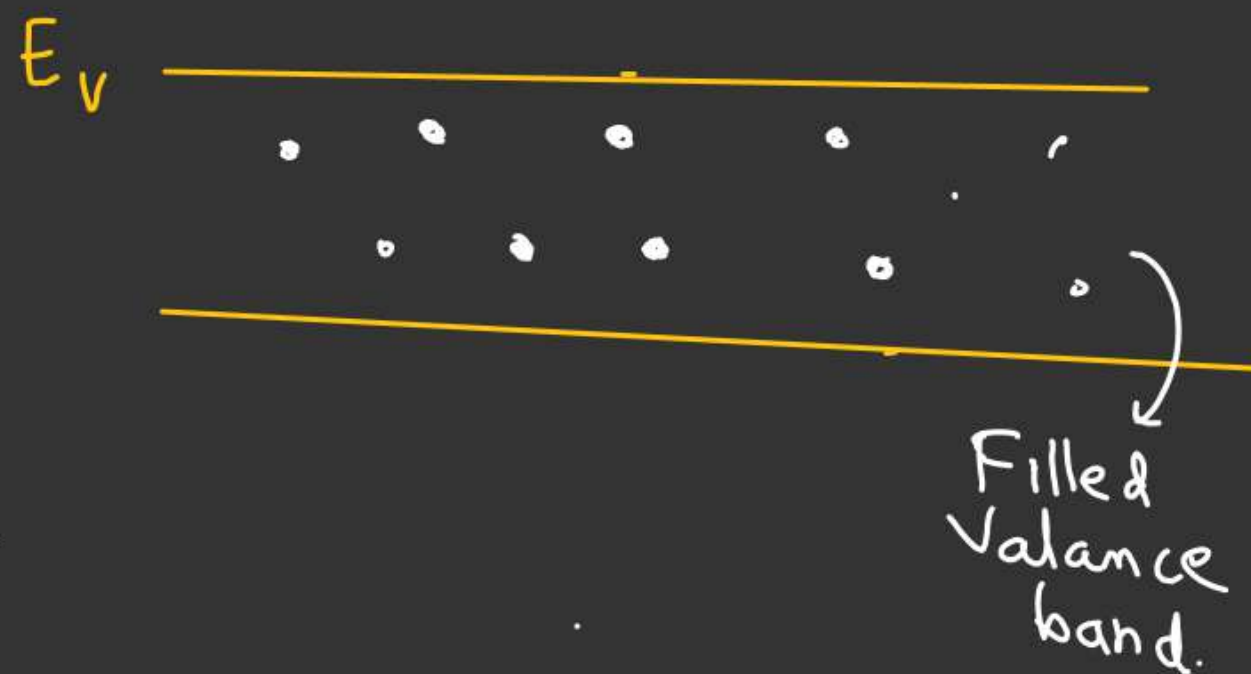
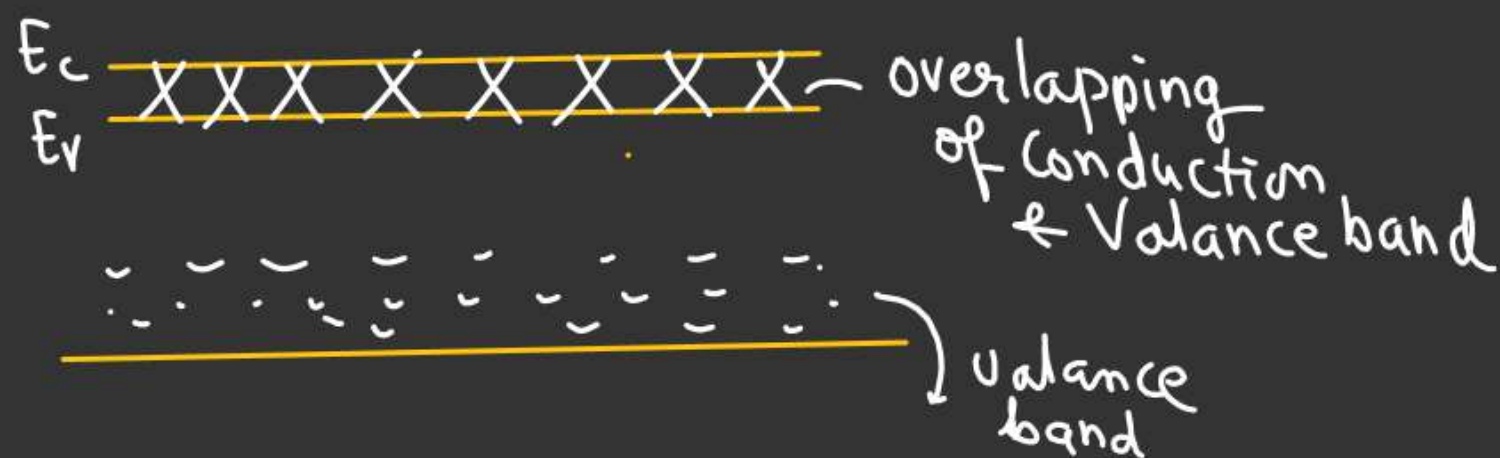
Two type of Energy band structure.

Either there is energy gap b/w the Completely filled Valance band & the partially filled Conduction band.

Ex!- (Alkali Metals, Noble Metals  
third group Metals)

OR  $\checkmark$

Conduction and Valance band overlap  $\checkmark$

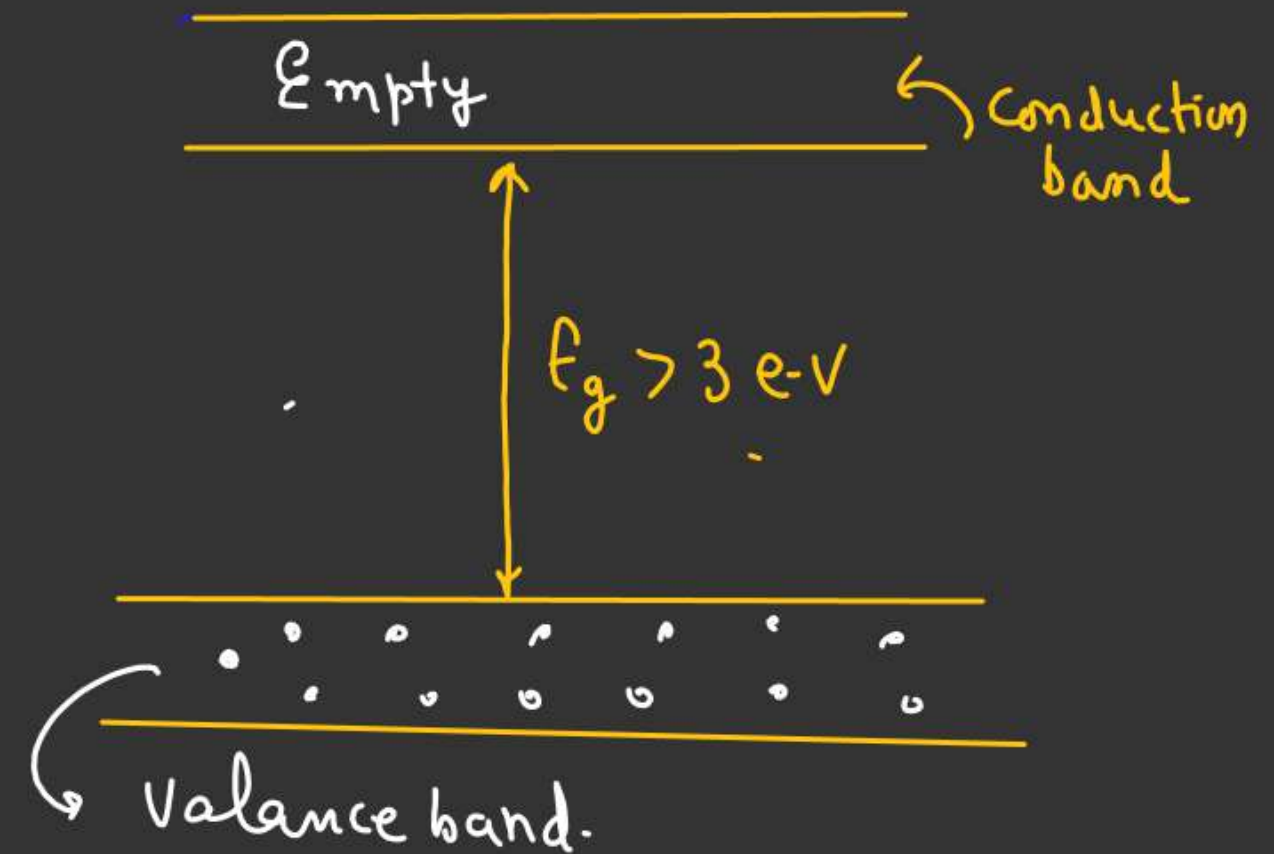


## FERMI LEVEL

Highest Energy level in the conduction band at absolute zero is called FERMIL LEVEL and energy corresponding to this level is called FERMIL Energy level. ✓

## INSULATORS ✓

- ↳ In insulators, valance band is completely filled while the conduction band is empty.
- ↳ Large energy gap b/w the valance band & conduction band ( $E_g > 3 \text{ eV}$ )





# ENERGY BAND IN SEMI CONDUCTOR

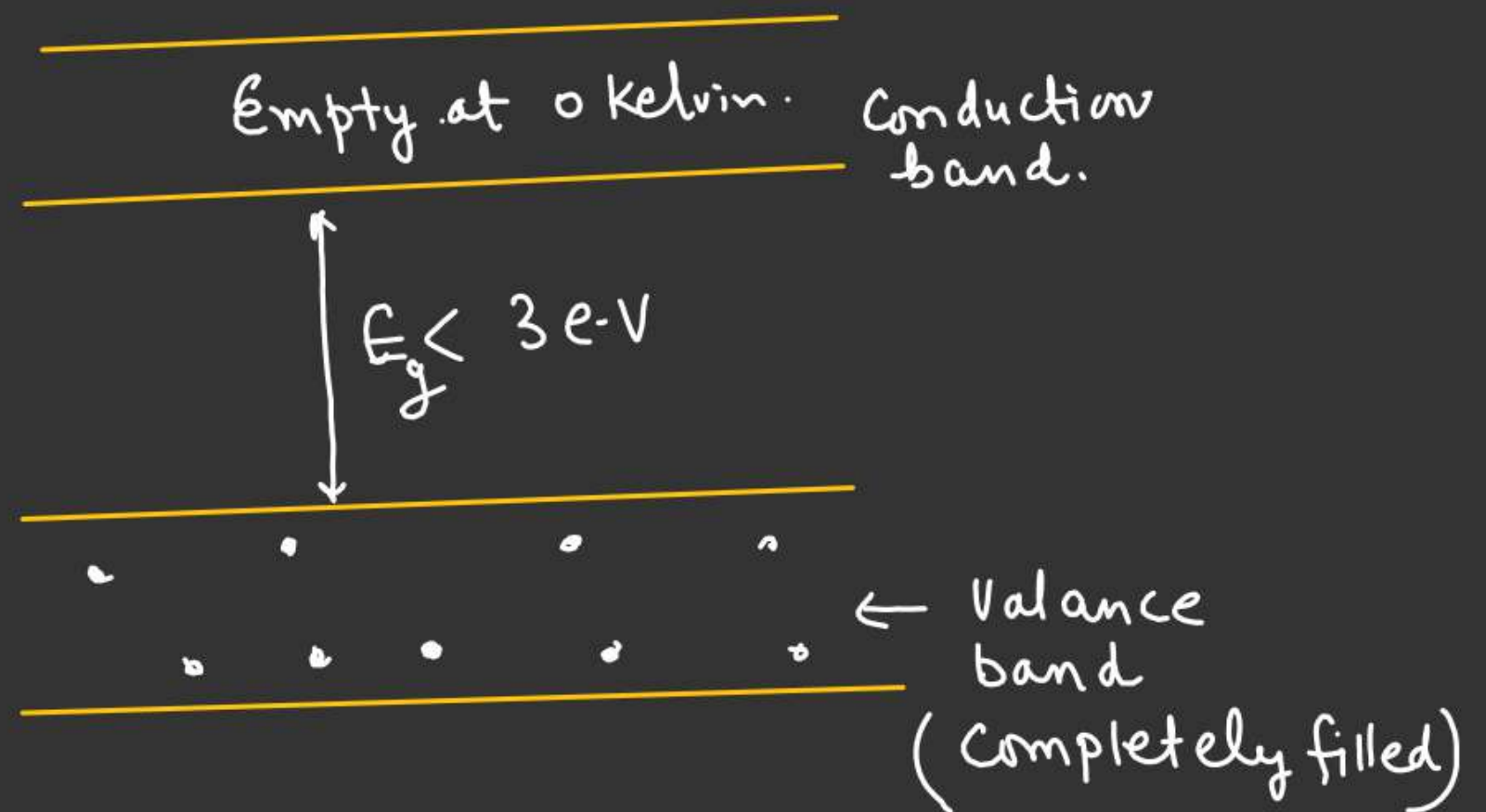
## INTRINSIC SEMI CONDUCTOR

At 0K, Conduction band empty.  
(Temp)  $\nwarrow$  Valance band completely filled

$$\left[ \begin{array}{l} E_g = 1.17 \text{ e-v for Si} \\ E_g = \underline{0.74 \text{ e-v}} \text{ for Ge} \end{array} \right.$$

At higher temp ( $> 0K$ ) Some Valance electron jump to Conduction band where they are free to conduct.

So Intrinsic Semi Conductor acquires Small Conductivity above 0 Kelvin



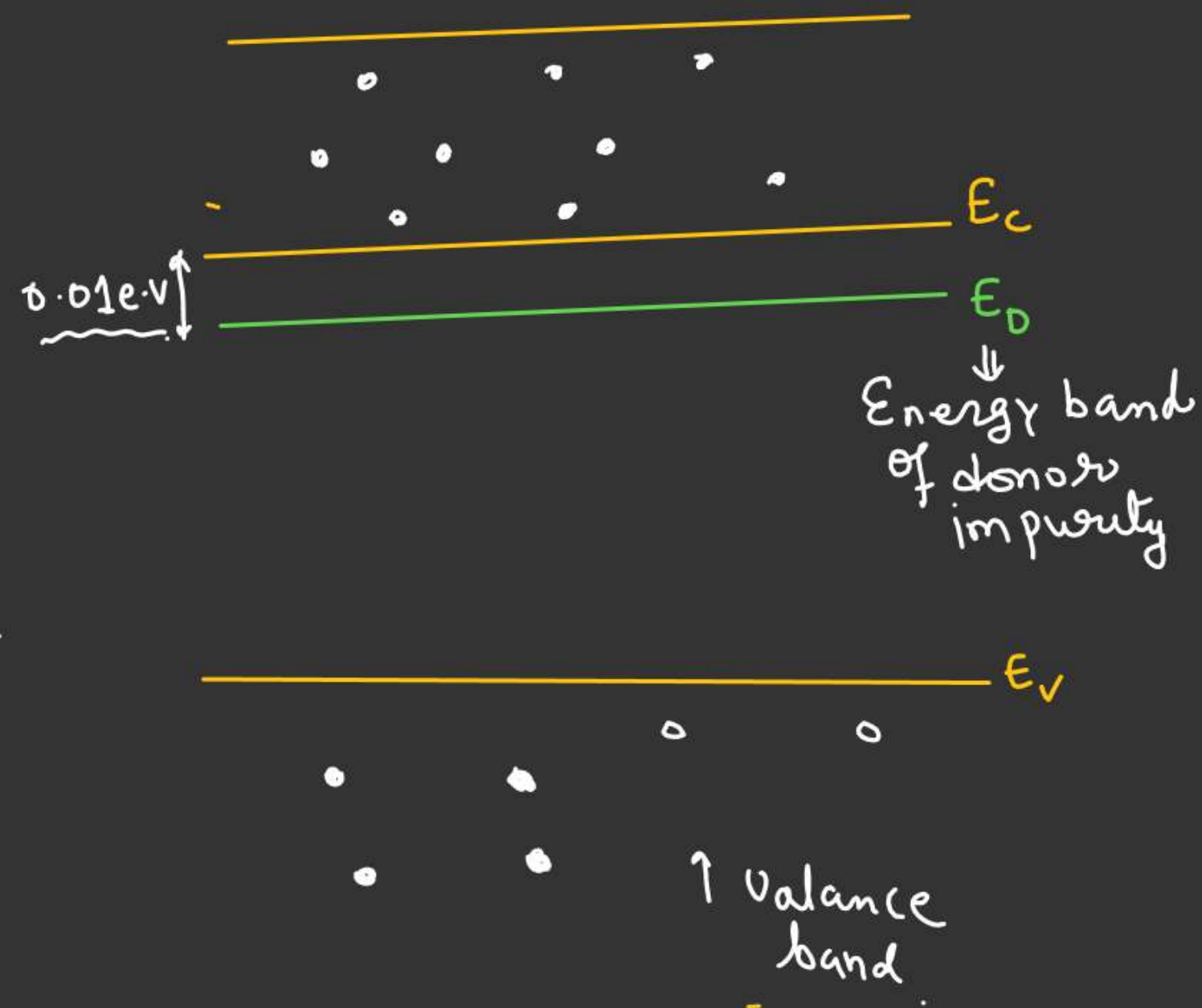


# EXTRINSIC SEMICONDUCTOR

## N-Type

Extra fifth electron is very weakly attracted by the donor impurity. A very small energy is required to free electrons from the donor impurity.

Imp ⇒ Conduction band has more electrons as they have been contributed both in thermal excitation & donor impurity



P-Type

Each acceptor impurity creates a hole which can be easily filled by an electron of Si-Si covalent band.

A very small energy require by an electron of valance band to move into this hole.

Acceptor energy level slightly above the valance band.

