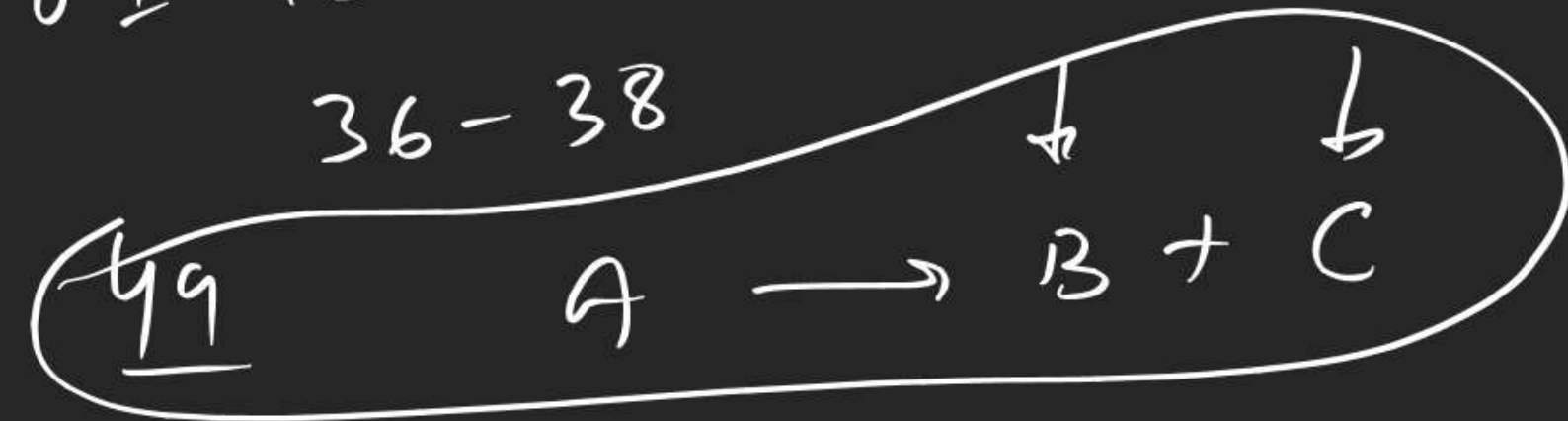


Q-1 48-51

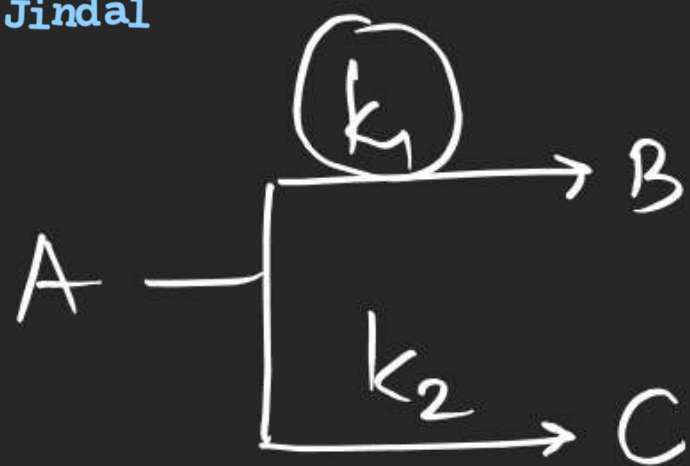
36-38



(37)

$$k = \frac{1}{60} \ln \frac{-3.8 - 13.1}{-3.8 - 11.6} = \frac{1}{t} \ln \frac{-3.8 - 13.1}{-3.8 - 0}$$

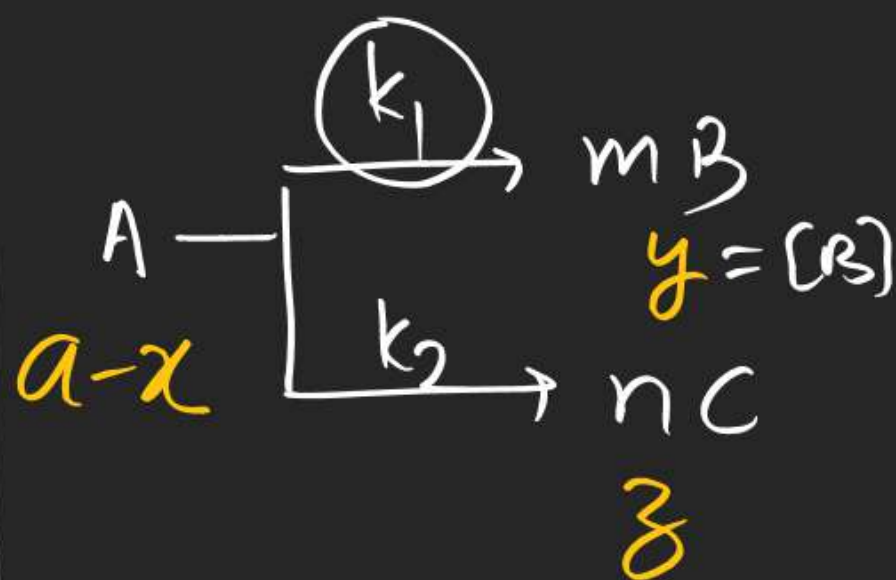
 $r_t$  $r_\infty$



$$[A]_t = [A]_0 e^{-(k_1 + k_2)t}$$

$$[B]_t = \frac{k_1}{k_1 + k_2} x$$

$$[C]_t = \frac{k_2}{k_1 + k_2} x$$



~~$$x = my + nz$$~~

$$x = \left(\frac{y}{m}\right) + \left(\frac{z}{n}\right)$$

$$\frac{dx}{dt} = \frac{1}{m} \frac{dy}{dt} + \frac{1}{n} \frac{dz}{dt}$$

$$-\frac{d[A]}{dt} = \frac{1}{m} \frac{d[B]}{dt} + \frac{1}{n} \frac{d[C]}{dt}$$

$$\frac{1}{m} \frac{d[B]}{dt} = k_1 [A]$$

$$\frac{1}{n} \frac{d[C]}{dt} = k_2 [A]$$

$$-\frac{d[A]}{dt} = (k_1 + k_2) [A]$$

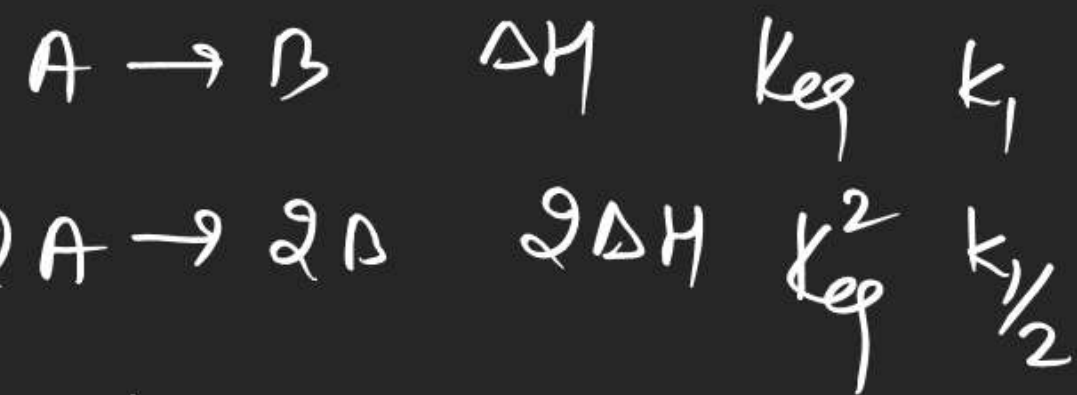
$$[A]_t = [A]_0 e^{-(k_1 + k_2)t}$$

$$[B] = m \times \frac{k_1}{k_1 + k_2} x$$

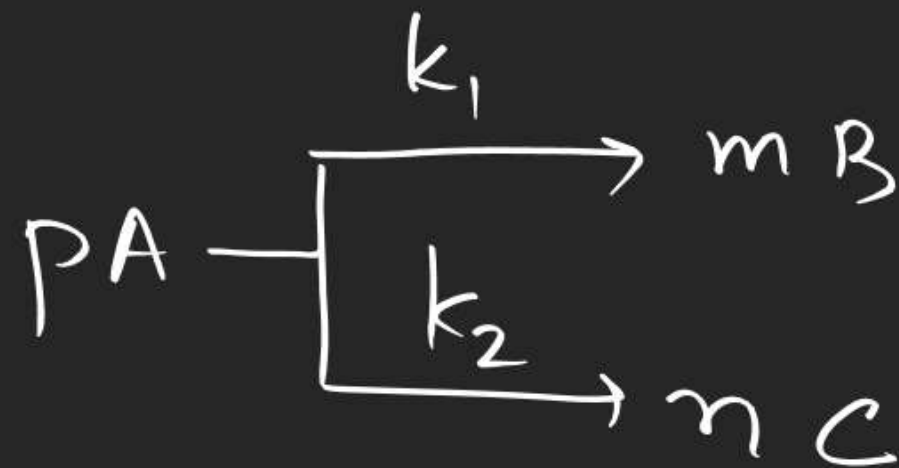
$$[C] = n \frac{k_2}{k_1 + k_2} x$$



$$-\frac{d[A]}{dt} = k_1 [A]$$



$$-\frac{1}{2} \frac{d[A]}{dt} = k_2 [A]$$

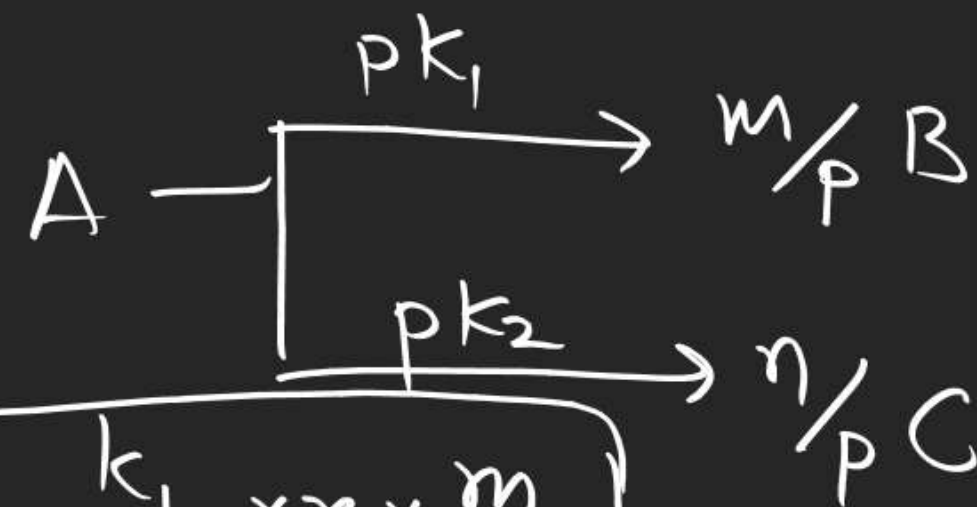


(A)  $k_1 = k_2$

$$k_1 = 2k_2$$

(B)  $2k_1 = k_2$

$$k_1/2 = k_2$$



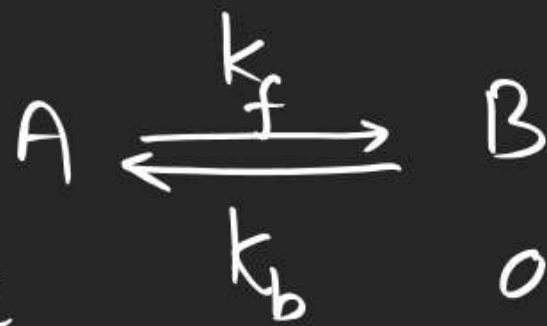
$$[B] = \frac{k_1}{k_1 + k_2} \times \frac{m}{p}$$

~~(C)~~  $k_1 = 2k_2$

(D)  $k_2 = k_1^2$



# Reversible rxn



a

0

time 't'

a-x

x

at eq<sup>m</sup>

a-x<sub>e</sub>

x<sub>e</sub>

$$-\frac{d[A]}{dt} = R_{\text{of}} \text{ of } A - R_{\text{or}} \text{ of } A$$

$$-\frac{d[A]}{dt} = k_f[A] - k_b[B]$$

$$K_{eq} = \frac{k_f}{k_b} = \frac{[B]}{[A]} = \frac{x_e}{a-x_e}$$

$$k_f a = (k_f + k_b) x_e$$

$$\frac{dx}{dt} = k_f(a-x) - k_b x$$

$$= k_f a - x(k_f + k_b)$$

$$\frac{dx}{dt} = (k_f + k_b)(x_e - x)$$

$$\ln \frac{x_e}{x_e - x} = (k_f + k_b)t$$

$$t_{1/2} \text{ when } x = x_e/2$$

$$t_{1/2} = \frac{\ln 2}{k_f + k_b}$$

Sequential reaction :  $\rightarrow$



for 'A'

$$-\frac{d[A]}{dt} = k_1 [A]$$

$$\rightarrow [A]_t = [A]_0 e^{-k_1 t}$$

for 'I'

$$\frac{d[I]}{dt} = R_0 A - R_0 D$$

$$= k_1 [A] - k_2 [I]$$

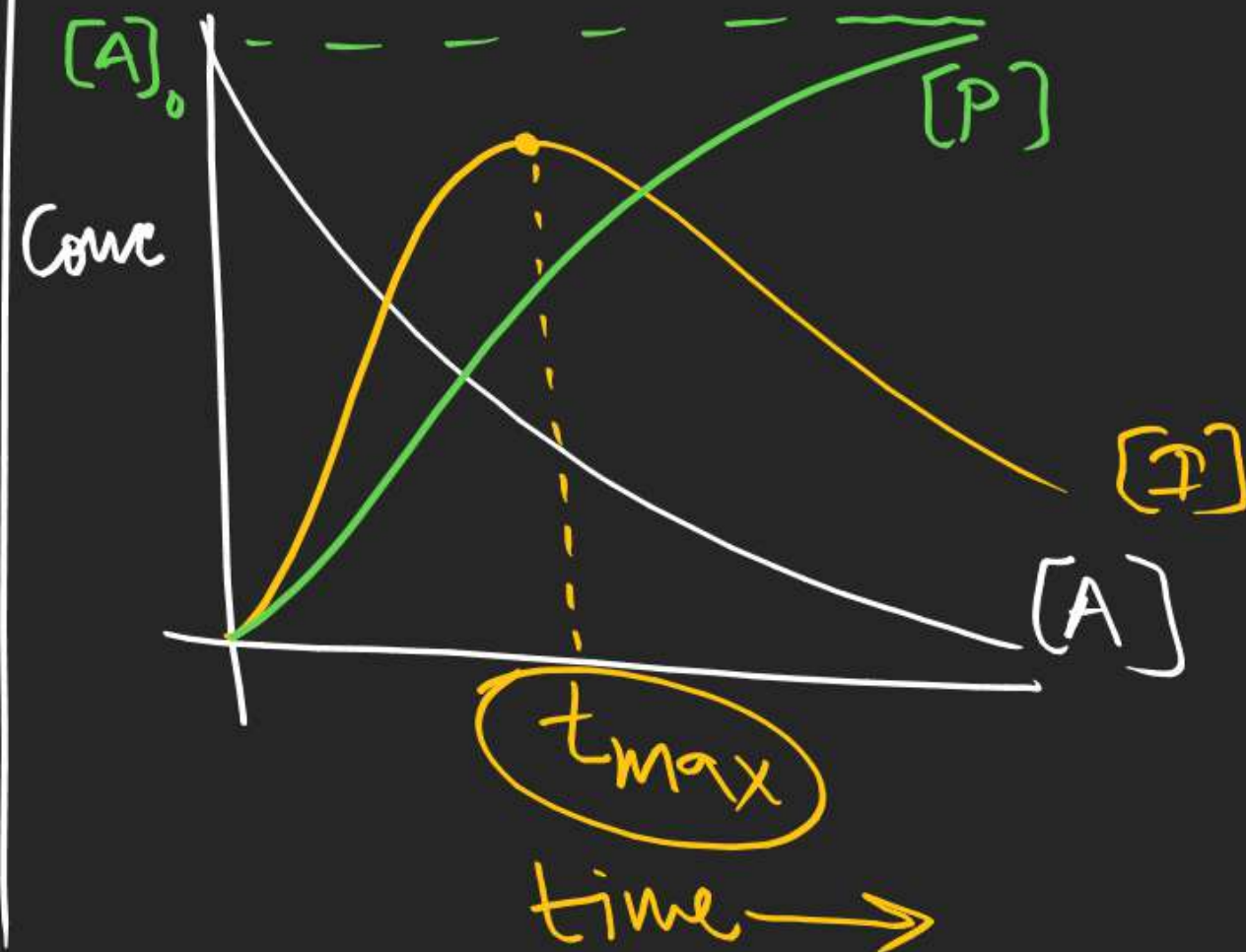
$$[I] = \frac{k_1}{k_2 - k_1} [A]_0 \left\{ e^{-k_1 t} - e^{-k_2 t} \right\}$$

for  $t_{\max}$   $\frac{d[I]}{dt} = 0$

$$t_{\max} = \frac{\ln(k_2/k_1)}{k_2 - k_1}$$

$$\frac{d[P]}{dt} = k_2 [I]$$

$$[P] = [A]_0 \left[ 1 - \frac{k_2 e^{-k_1 t} - k_1 e^{-k_2 t}}{k_2 - k_1} \right]$$

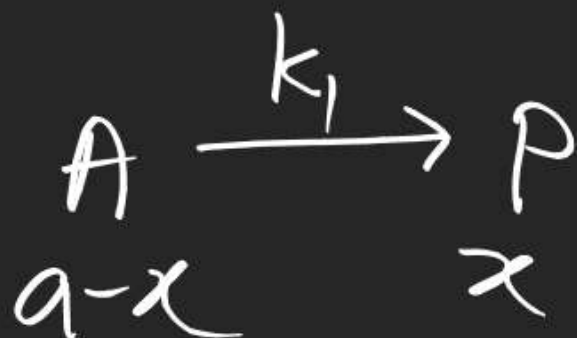
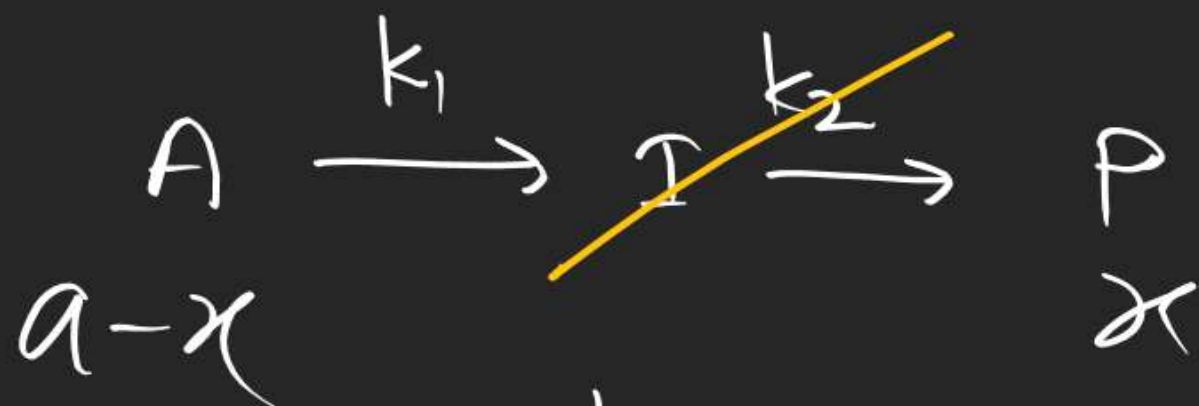




Case-I  $k_2 \gg k_1$ 

$$[P] = [A]_0 \left\{ 1 - \frac{k_2 e^{-k_1 t}}{k_2 - k_1} + \frac{k_1 e^{-k_2 t}}{k_2 - k_1} \right\}$$

$$\underline{\underline{[P] = [A]_0 \{ 1 - e^{-k_1 t} \}}} = x = \text{Amt of A reacted}$$



$$[I] = \frac{k_1}{k_2 - k_1} [A] \left\{ e^{-k_1 t} - e^{-k_2 t} \right\}$$

$$= \frac{k_1}{k_2} [A]_0 e^{-k_1 t}$$

x

## Kinetics

U-I	52
S-I	39-41

→ T.D 1 & 2 JEE Adv