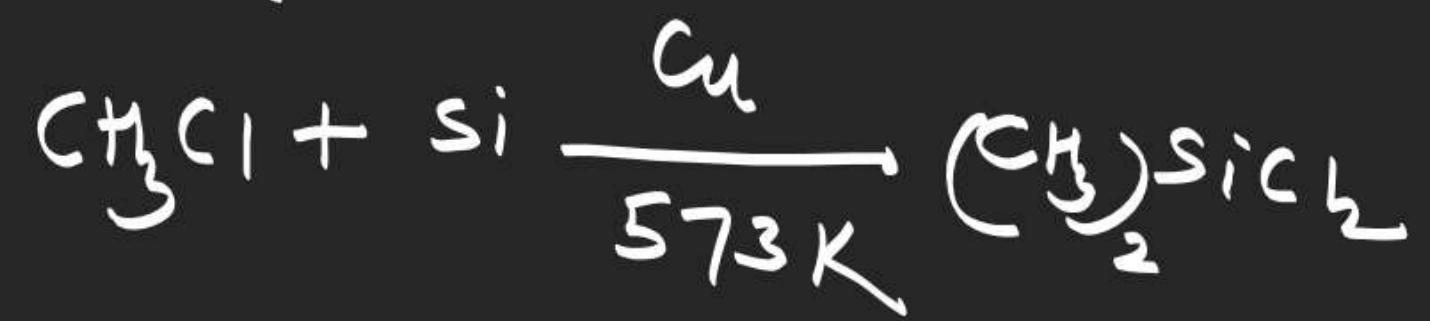
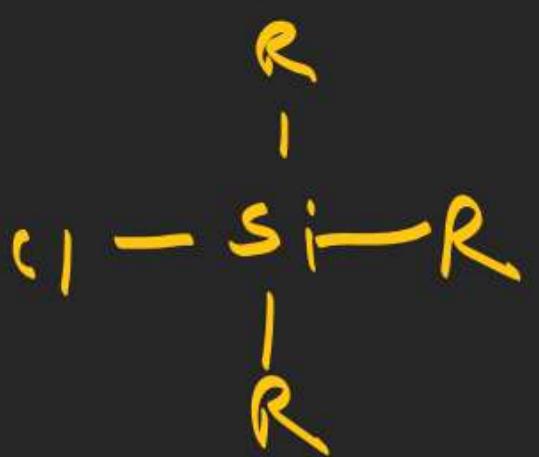
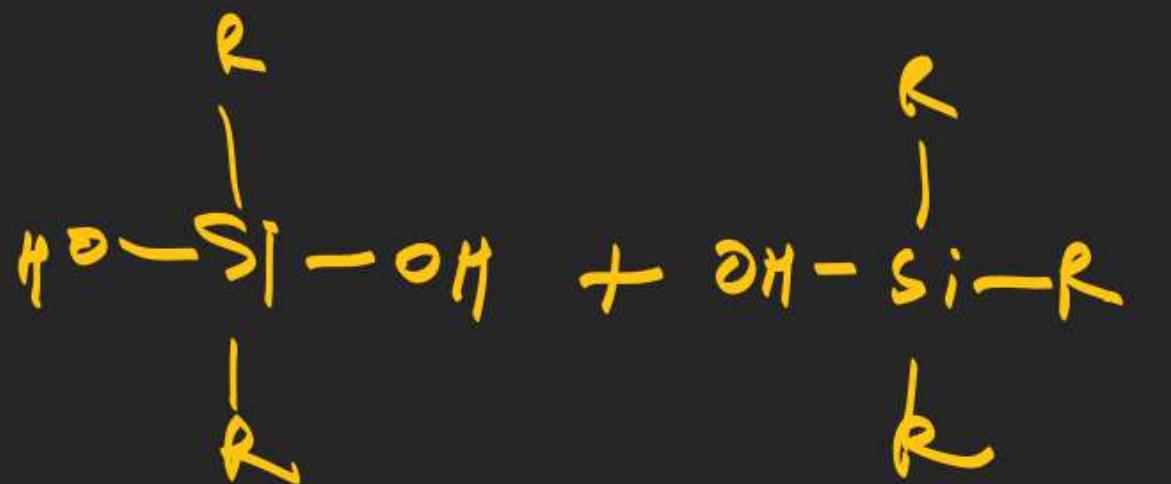


Prep of silicone



Ques Which of the following molecule can act as
Chain stopping unit for silicones.

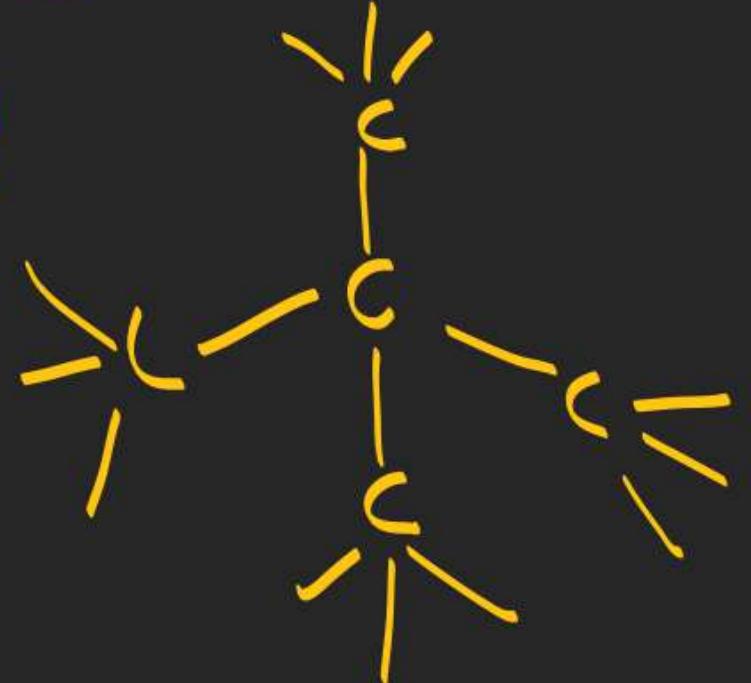
- ① R_2SiCl_2
- ② RSiCl_3
- ③ R_3SiCl
- ④ SiCl_4



allotropes of C

Diamond
graphite
fullerene

(i) Diamond



sp^3
tet
nonplanar
nonconductor

graphite

on

(T/F)

non metallic covalent +
bond is present within
layer of graphite

(T)

sp^2

planar

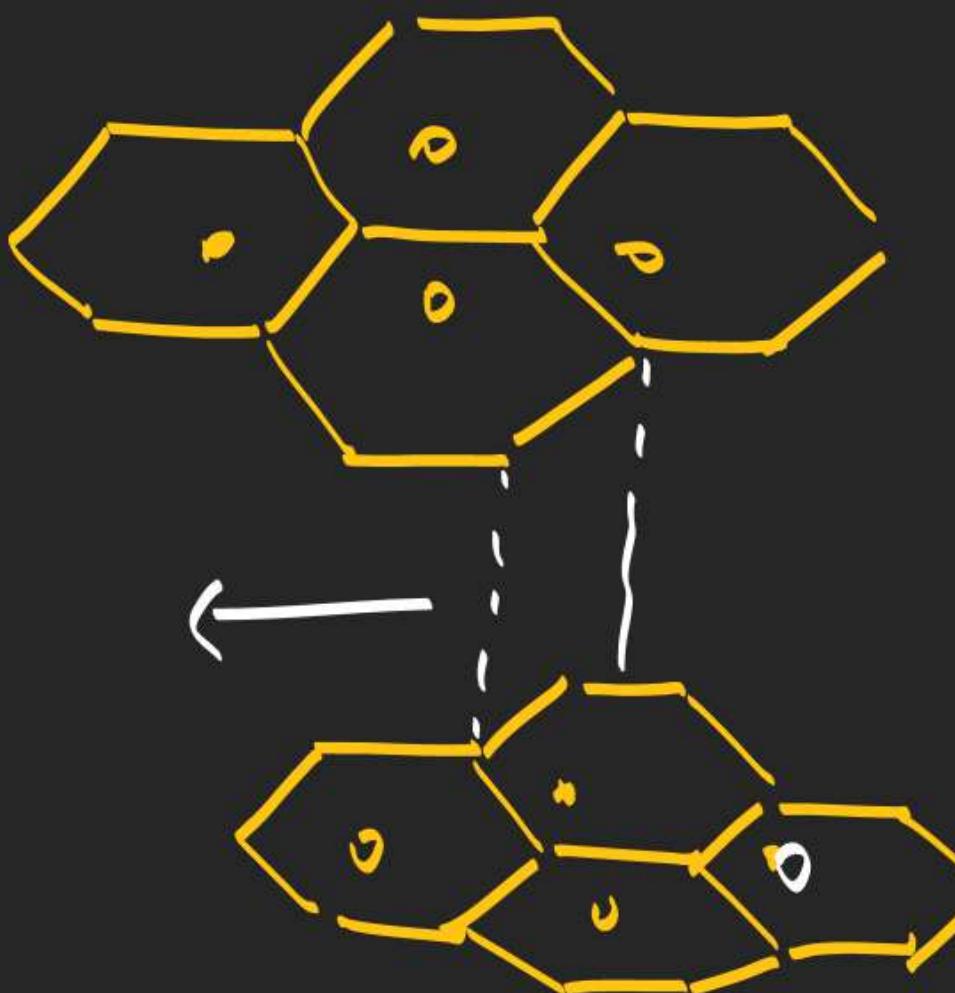
aromatic
Conductor

Hexagonal planar 2D sheet

like structure.

(Vander
Waal's
force)

V.W.F



60 ⇒

sp^2

aromatic

fullerene has

no dangling bond.

20 Hexagonal Rings

12 - five memb. rings [fix]
(Pentagonal Rings)

Each Hexagonal Rings fused with Pentagonal
and Hexagonal.
and each Pentagonal Rings fused with only
Hexagonal.

find the no of Hexagonal Rings
in C_{80}

$$\frac{g}{2} = n + 10$$

g = total no of carbon

n = Hexagonal Rings

$$\frac{80}{2} = n + 10$$

$$\underline{n = 30}$$