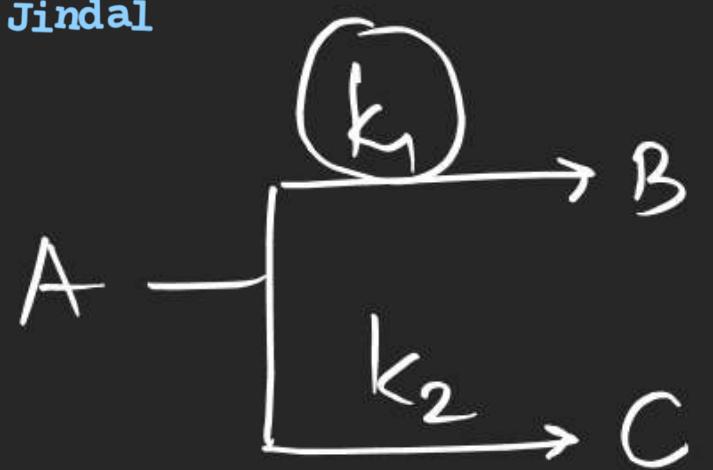


Q-1 48-51



(37) $k = \frac{1}{60} \ln \frac{-3.8 - 13.1}{-3.8 - 11.6} = \frac{1}{t} \ln \frac{-3.8 - 13.1}{-3.8 - 0}$

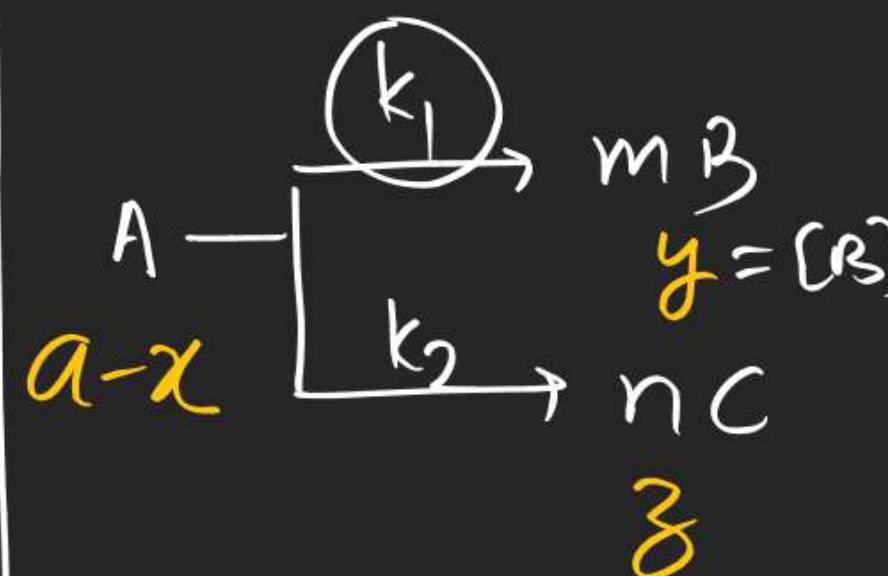
 γ_t γ_∞



$$[A]_t = [A]_0 e^{-(k_1 + k_2)t}$$

$$[B]_t = \frac{k_1}{k_1 + k_2} x$$

$$[C]_t = \frac{k_2}{k_1 + k_2} x$$



~~$$x = my + nz$$~~

$$x = \left(\frac{y}{m}\right) + \left(\frac{z}{n}\right)$$

$$\frac{dx}{dt} = \frac{1}{m} \frac{dy}{dt} + \frac{1}{n} \frac{dz}{dt}$$

$$-\frac{d[A]}{dt} = \frac{1}{m} \frac{d[B]}{dt} + \frac{1}{n} \frac{d[C]}{dt}$$

$$\frac{1}{m} \frac{d[B]}{dt} = k_1 [A]$$

$$\frac{1}{n} \frac{d[C]}{dt} = k_2 [A]$$

$$-\frac{d[A]}{dt} = (k_1 + k_2) [A]$$

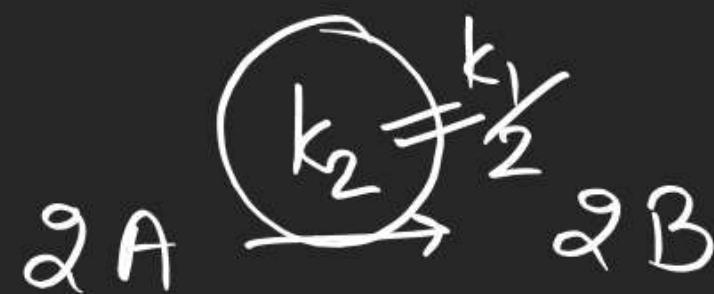
$$[A]_t = [A]_0 e^{-(k_1 + k_2)t}$$

$$[B] = mx \frac{k_1}{k_1 + k_2} x$$

$$[C] = n \frac{k_2}{k_1 + k_2} x$$



$$-\frac{d[A]}{dt} = k_1 [A]$$



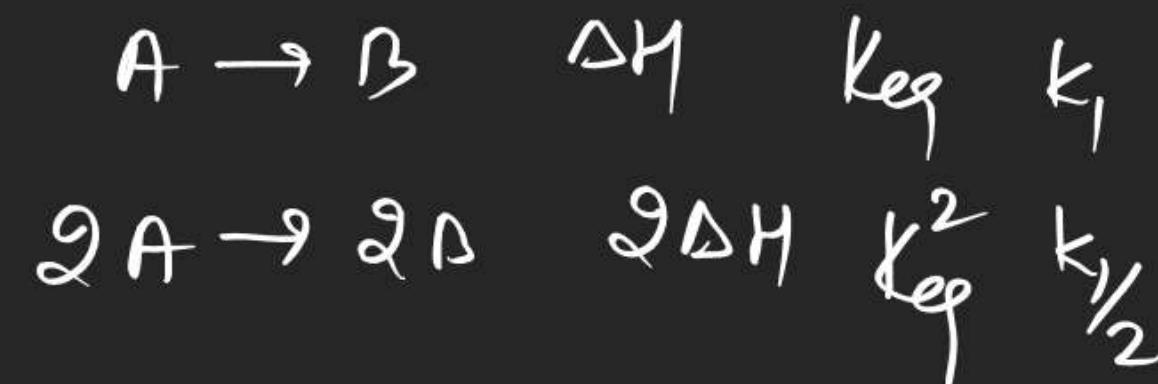
$$-\frac{1}{2} \frac{d[A]}{dt} = k_2 [A]$$

A) $k_1 = k_2$

B) $2k_1 = k_2$

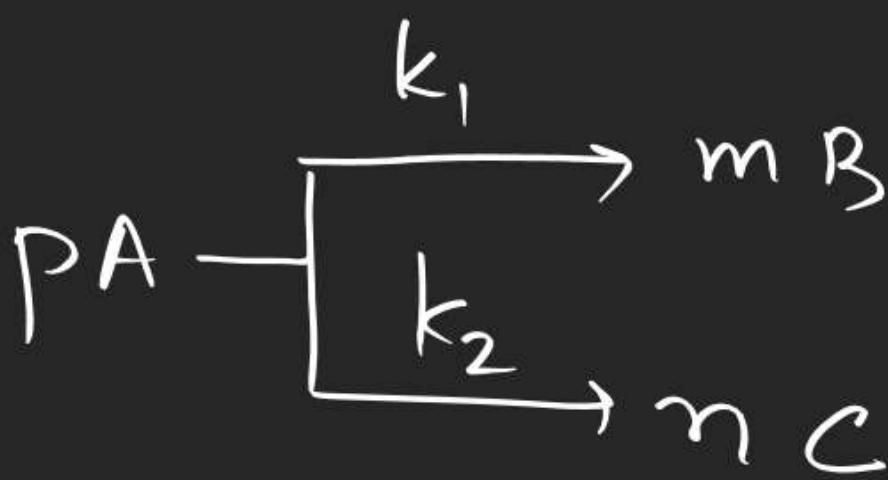
~~C) $k_1 = 2k_2$~~

D) $k_2 = k_1^2$

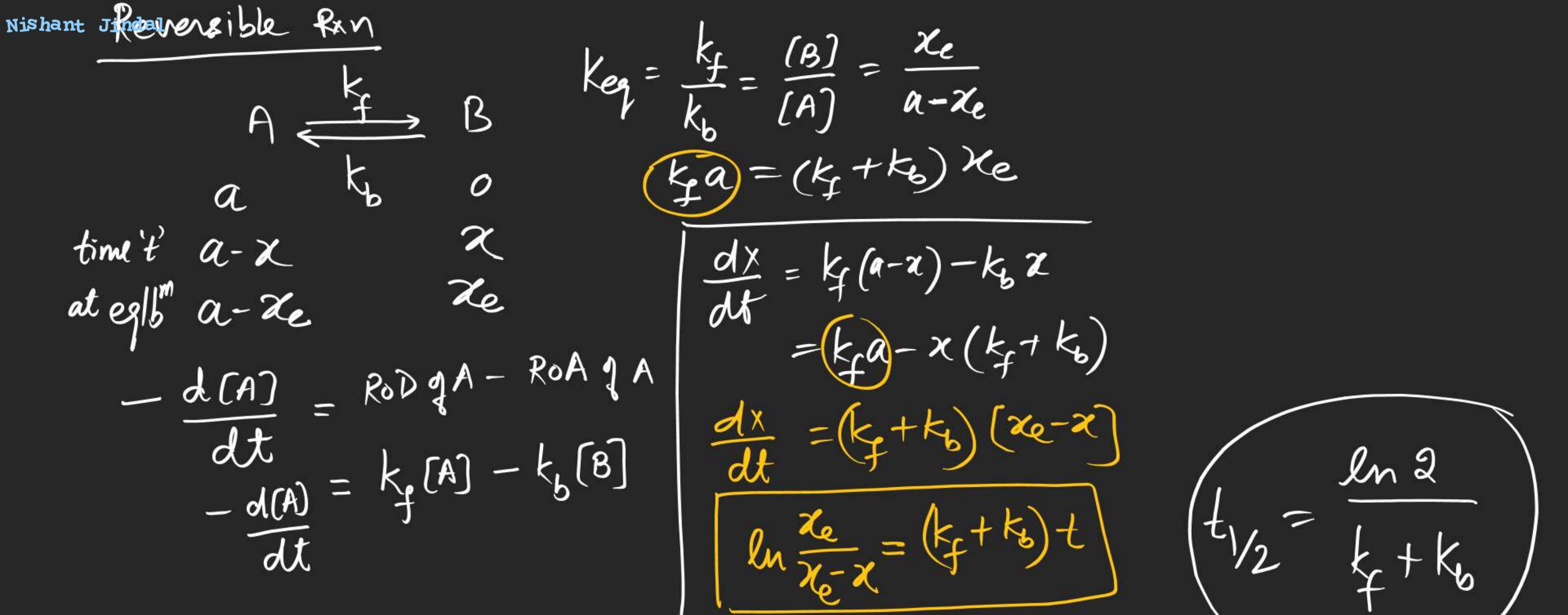


$$k_1 = 2k_2$$

$$\frac{k_1}{2} = k_2$$



$$[B] = \frac{k_1 \times x \times \frac{m}{p}}{k_1 + k_2}$$



$t_{1/2}$ when $x = x_e/2$

$$t_{1/2} = \frac{\ln 2}{k_f + k_b}$$

Sequential reaction :-



for 'A'

$$-\frac{d[A]}{dt} = k_1 [A]$$

$$\Rightarrow [A]_t = [A]_0 e^{-k_1 t}$$

for 'I'

$$\begin{aligned}\frac{d[I]}{dt} &= R_o A - R_o D \\ &= k_1 [A] - k_2 [I]\end{aligned}$$

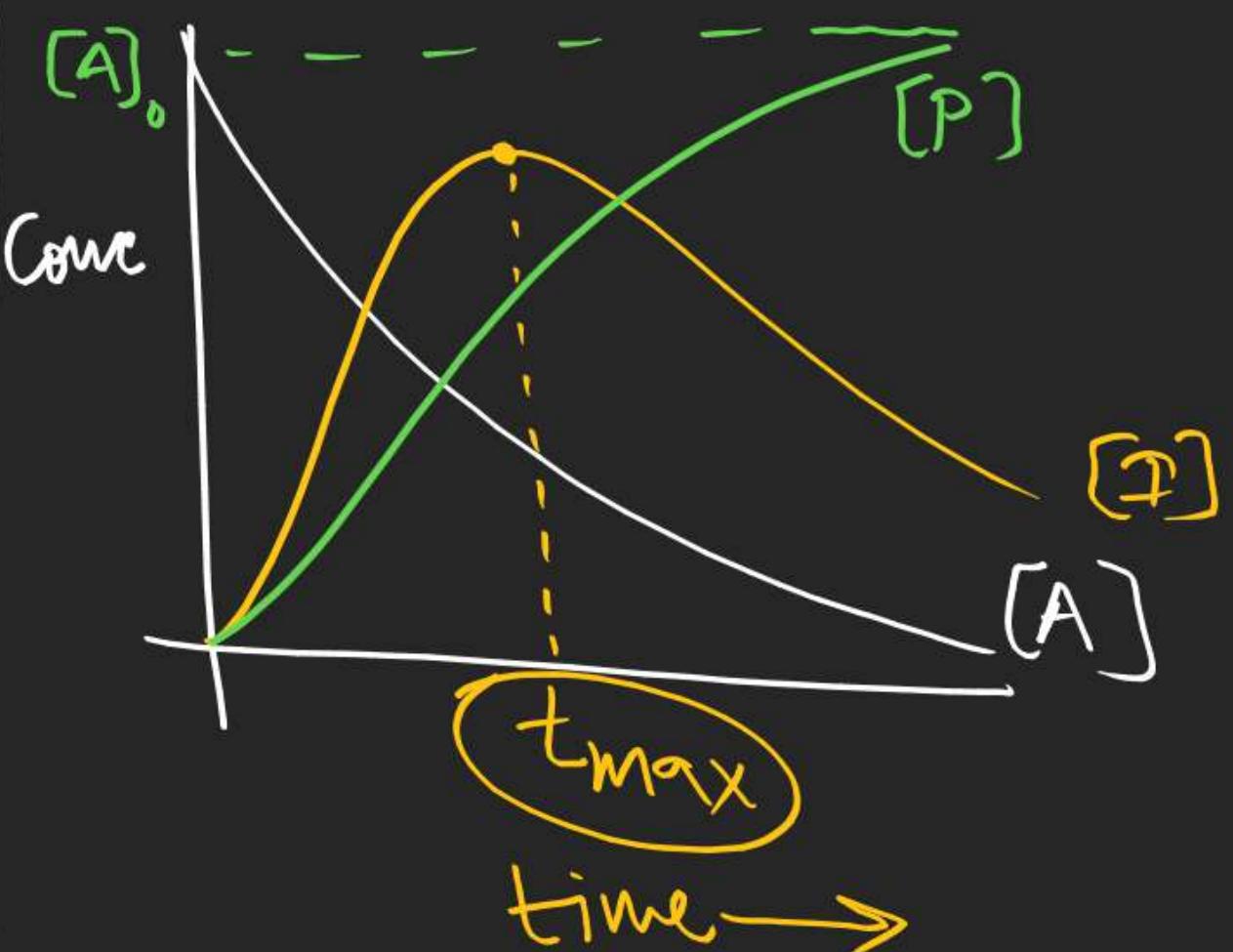
$$[I] = \frac{k_1}{k_2 - k_1} [A]_0 \left\{ e^{-k_1 t} - e^{-k_2 t} \right\}$$

for t_{max} $\frac{d[I]}{dt} = 0$

$$t_{max} = \frac{\ln(k_2/k_1)}{k_2 - k_1}$$

$$\frac{d[P]}{dt} = k_2 [I]$$

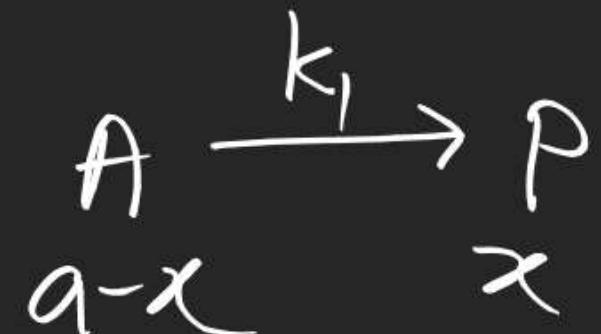
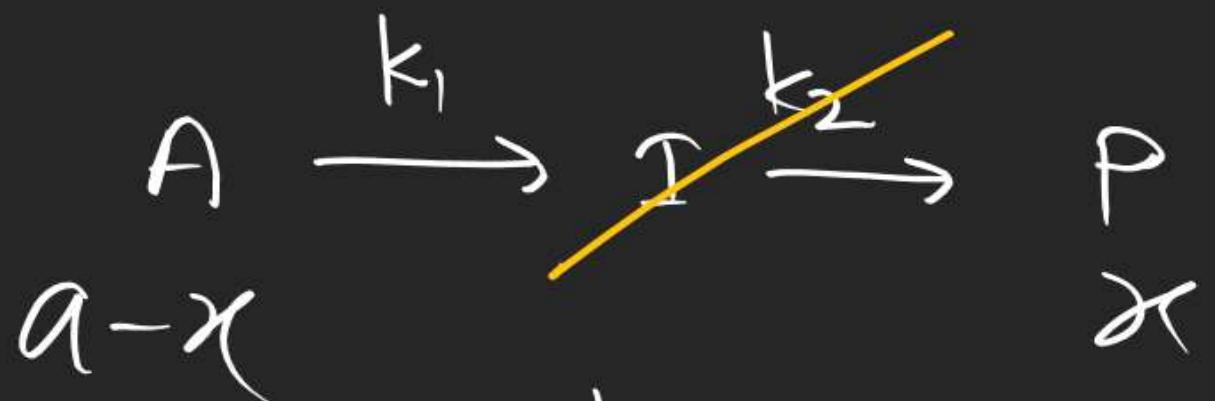
$$[P] = [A]_0 \left\{ 1 - \frac{k_2 e^{-k_1 t} - k_1 e^{-k_2 t}}{k_2 - k_1} \right\}$$



Case-I $k_2 \gg k_1$

$$[P] = [A]_0 \left\{ 1 - \frac{k_2 e^{-k_1 t}}{k_2 - k_1} + \frac{k_1 e^{-k_2 t}}{k_2 - k_1} \right\}$$

$$\underline{\underline{[P] = [A]_0 \left\{ 1 - e^{-k_1 t} \right\}}} = x = \text{Amt of A reacted}$$



$$\left. \begin{aligned} [I] &= \frac{k_1}{k_2 - k_1} [A] \left\{ e^{-k_1 t} - e^{-k_2 t} \right\} \\ &= \frac{k_1}{k_2} [A]_0 e^{-k_1 t} x \end{aligned} \right\}$$

Kinetics

$$\begin{array}{l} v = \text{L} \\ s = \text{L} \end{array}$$

(52)
39 - 41

→ TD 1 & 2 JEE Adv