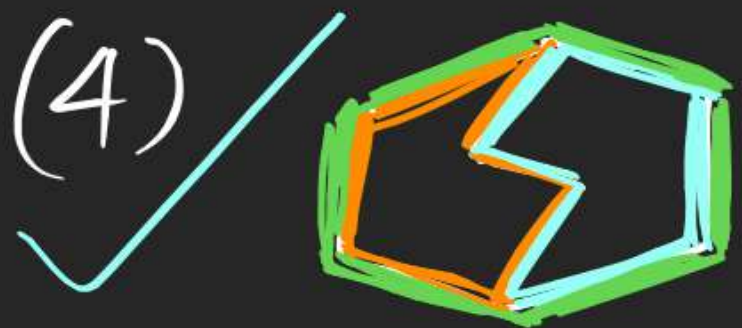
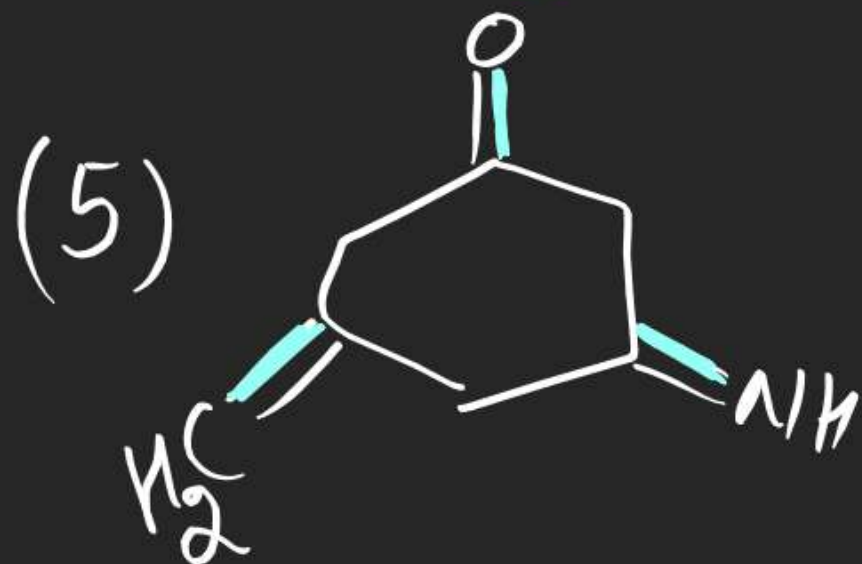


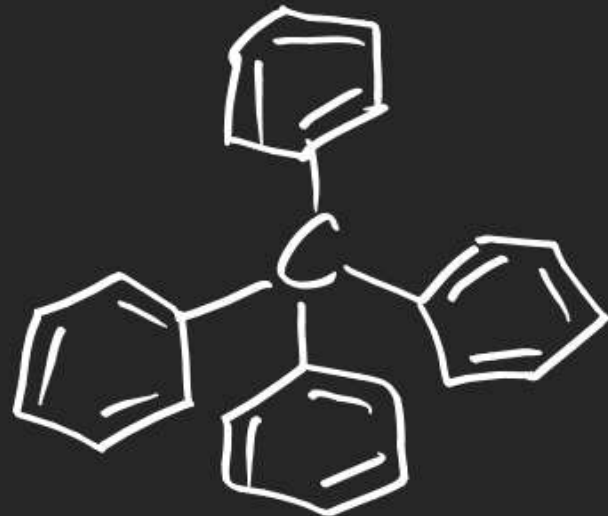
Theory Copy Discussion:

$$\begin{aligned} \text{DBE} &= T_R + T_\pi \\ &= \textcircled{2} + 0 = 2 \end{aligned}$$



$$\begin{aligned} \text{DBE} &= T_R + T_\pi \\ &= 1 + 3 \\ &= 4 \end{aligned}$$

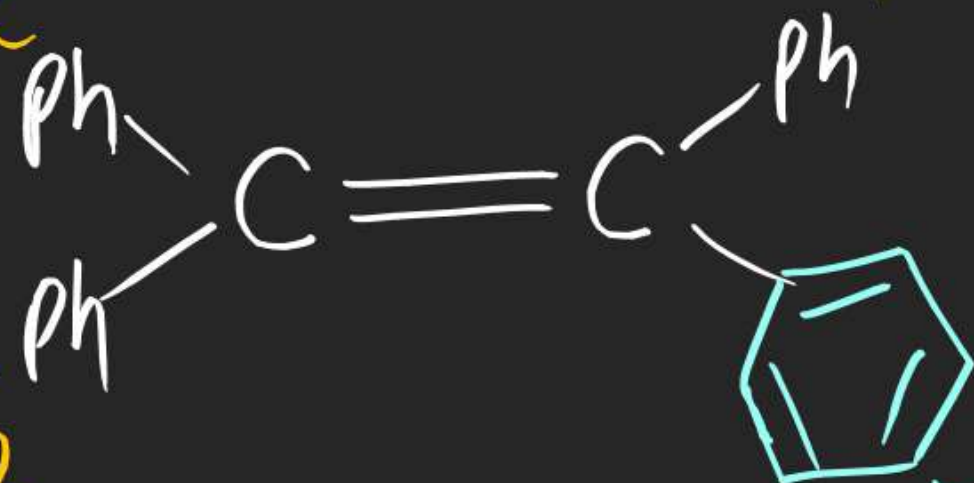
(6)  $\text{Ph}_4\text{C}$



$$\text{DBE} = 16$$

$$= 4 + 12$$

(7)  $\text{Ph}_4\text{C}_2$



$$\text{DBE} = 4 + 12 + 1$$

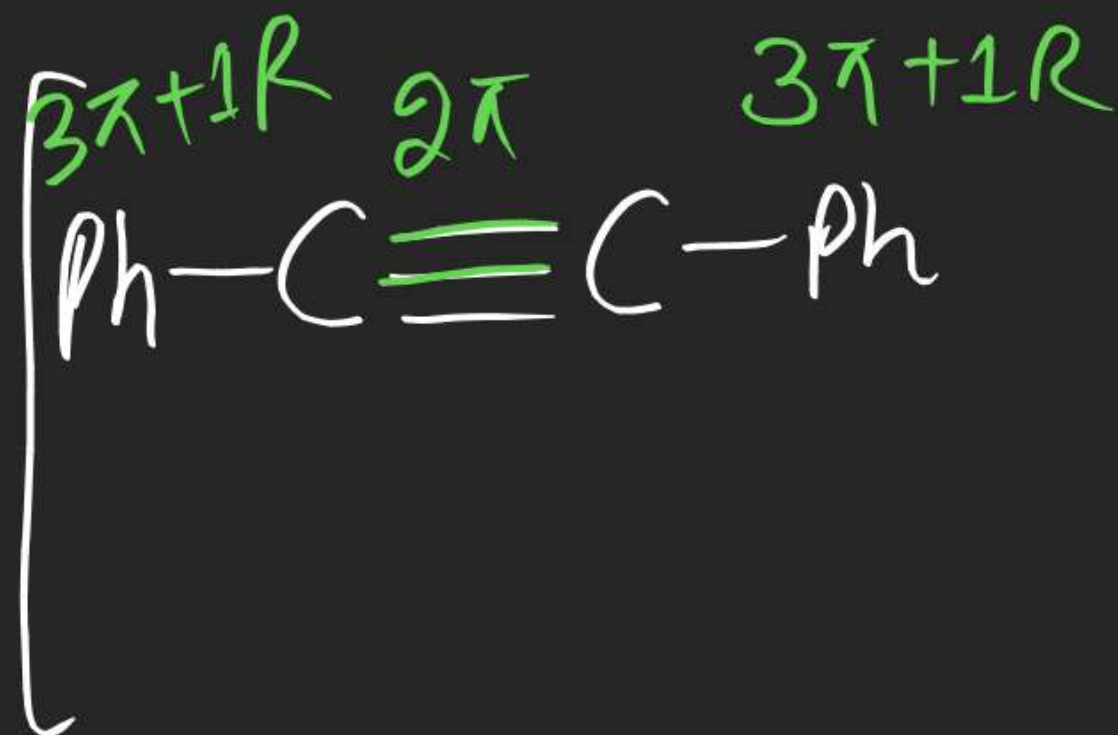
$$= 4 + 13$$

$$= 17$$

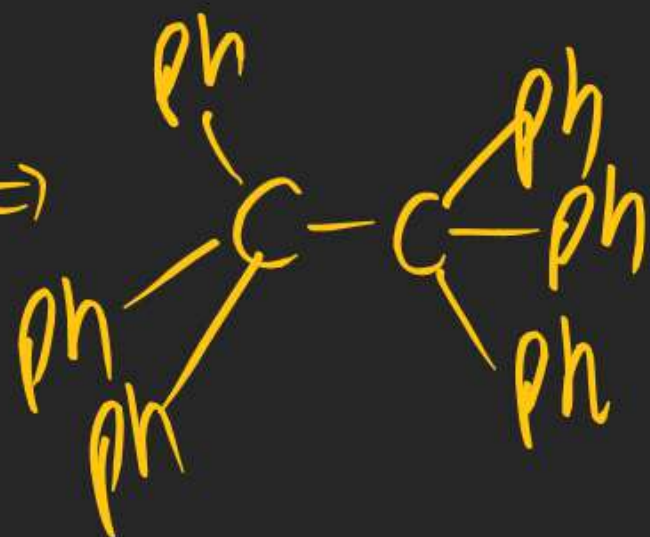
(8)  $\text{Ph}_2\text{C}_2$

$$\text{DBE} = \textcircled{10}$$

$$= 2 + 8$$



(9)  $\text{Ph}_6\text{C}_2 \Rightarrow$



$$\text{DBE} = 24$$

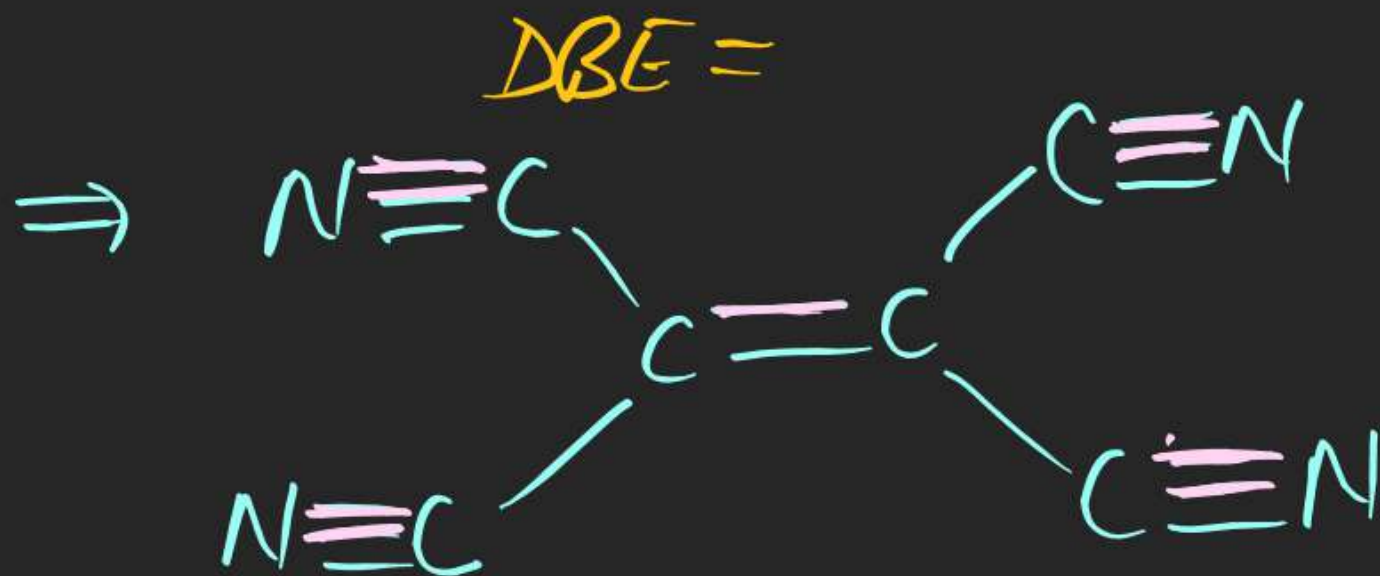
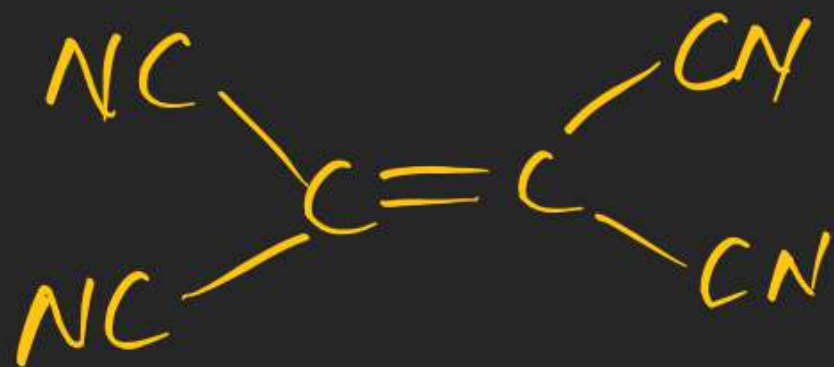
$$= 6 + 18$$

(10)



$$DBE = 2 + 5 = 7$$

(11)



$$DBE = 0 + 9 = 9$$

(12)



$$DBE = 4 + 8 = 12$$

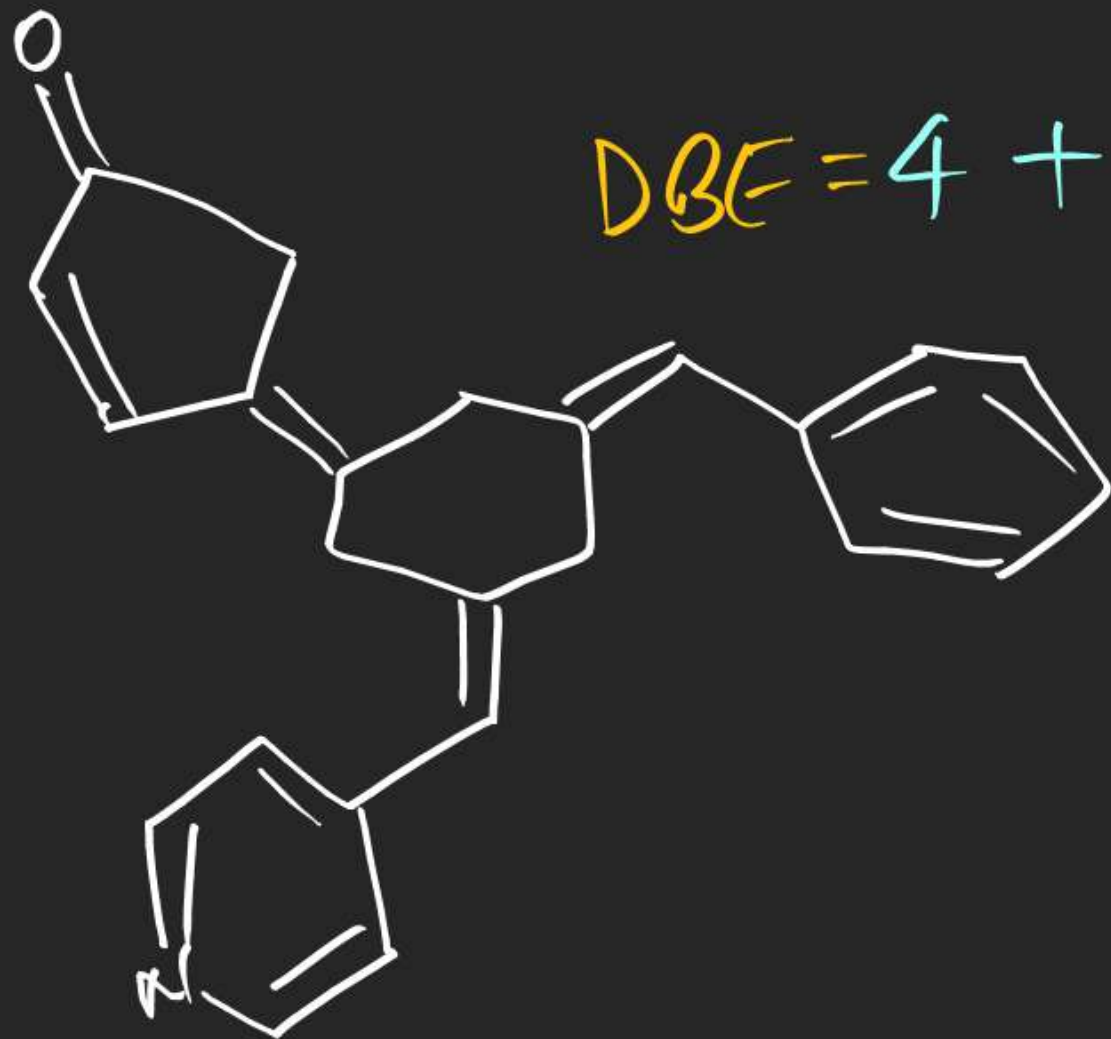


(13)



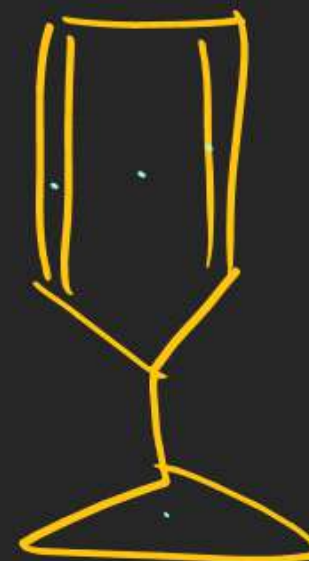
$$DBE = 4$$

(14)



$$DBE = 4 + 11 = 15$$

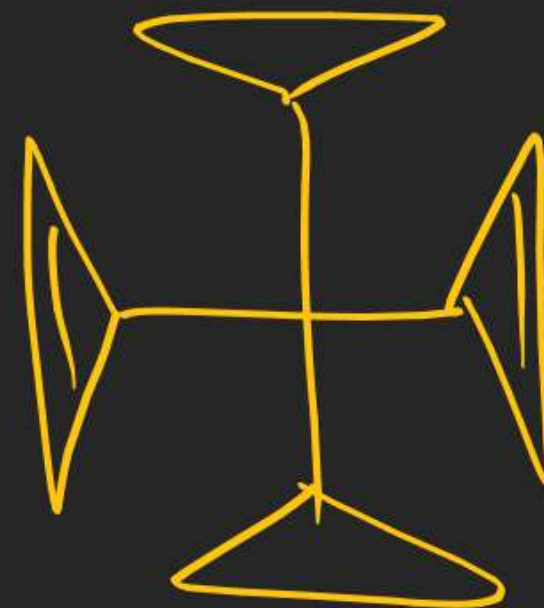
(15)

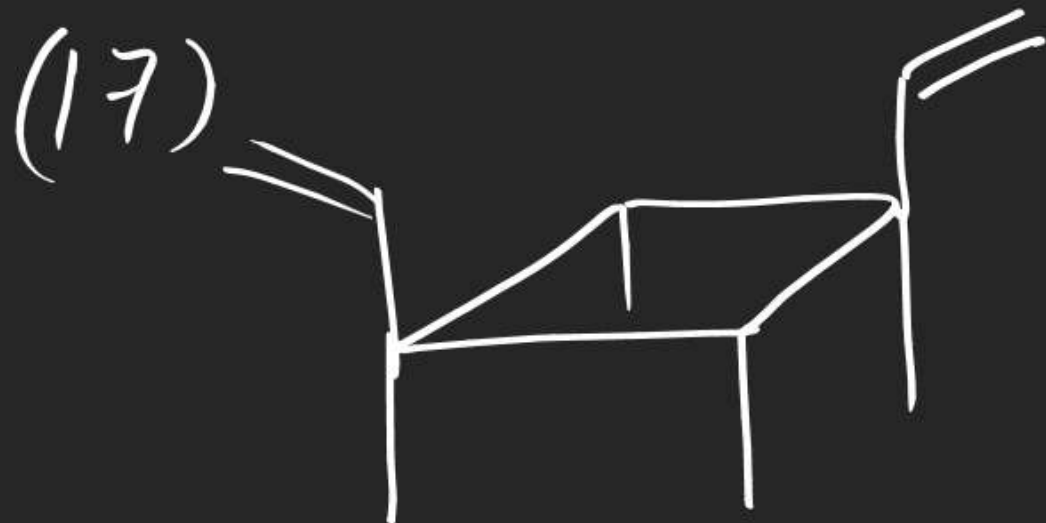


$$DBE = 2 + 2 = 4$$

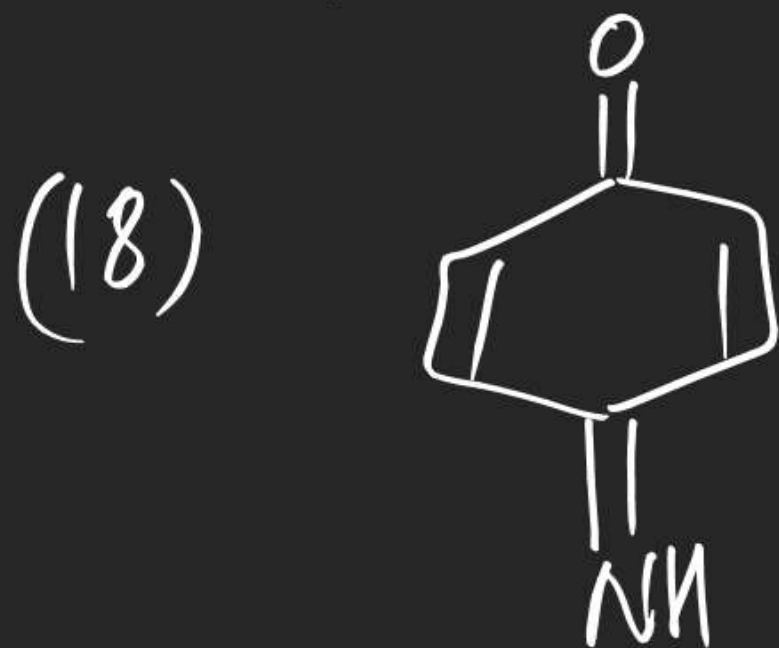
(16)

$$DBE = 4 + 2 = 6$$





$$\text{DBE} = 1 + 2$$
$$= 3$$

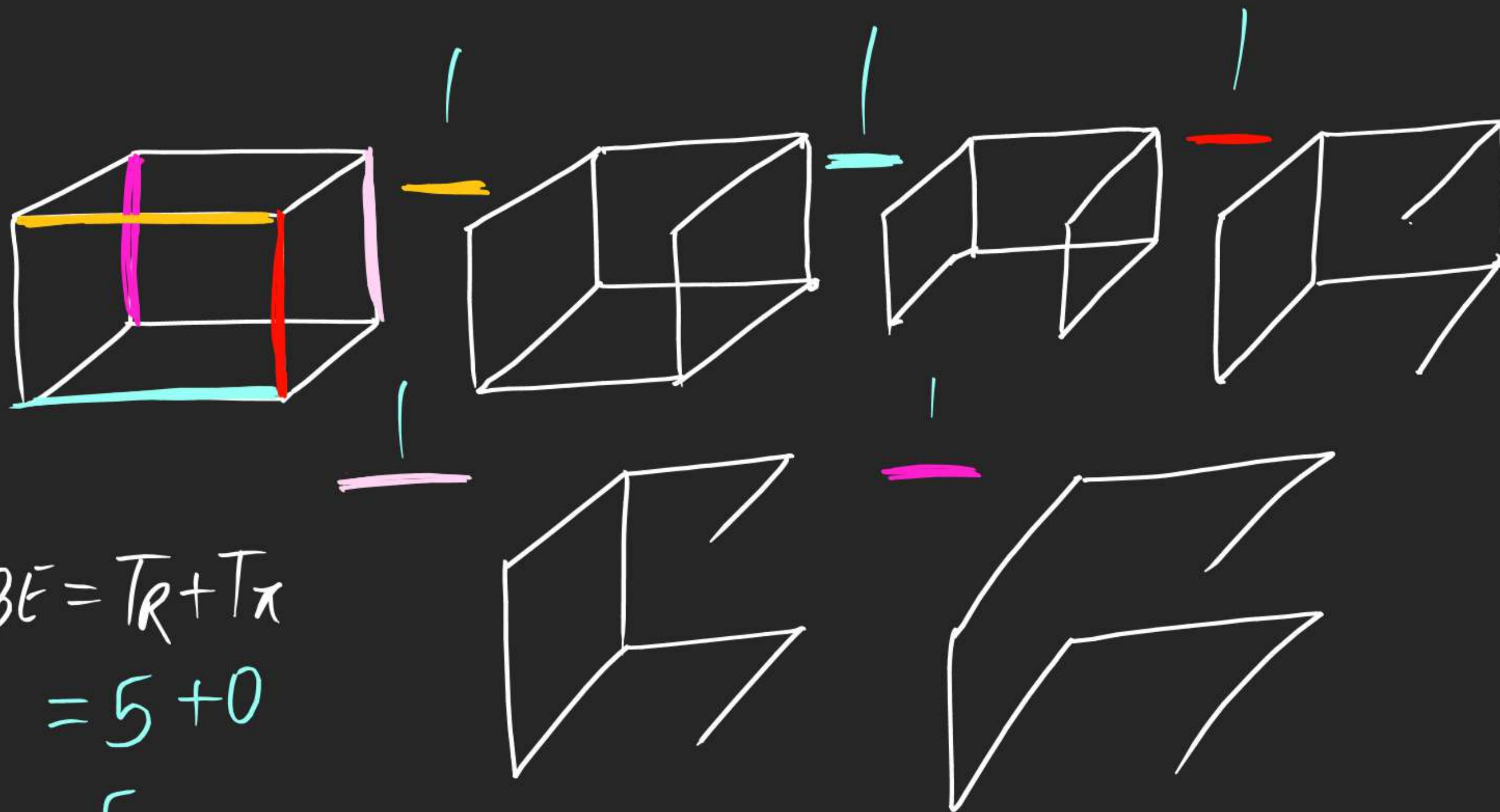


$$\text{DBE} = 5$$





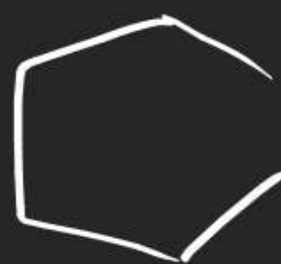
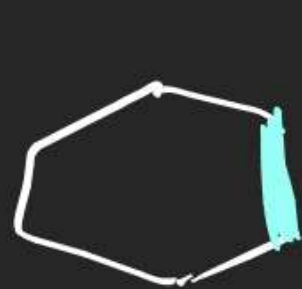
(19)



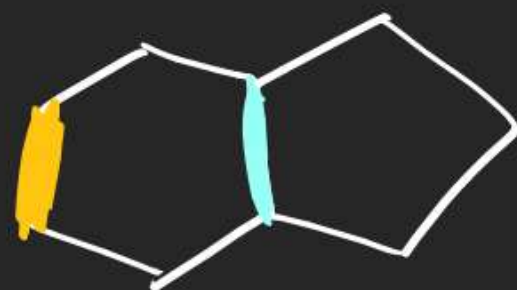
$$\begin{aligned}
 DBE &= T_R + T_\pi \\
 &= 5 + 0 \\
 &= 5
 \end{aligned}$$

## Basic Organic Chemistry

(20) (a)

 $(T_R=1)$ Total No. of Ring= minimum No. of Bond  
By Breaking of that

(b)

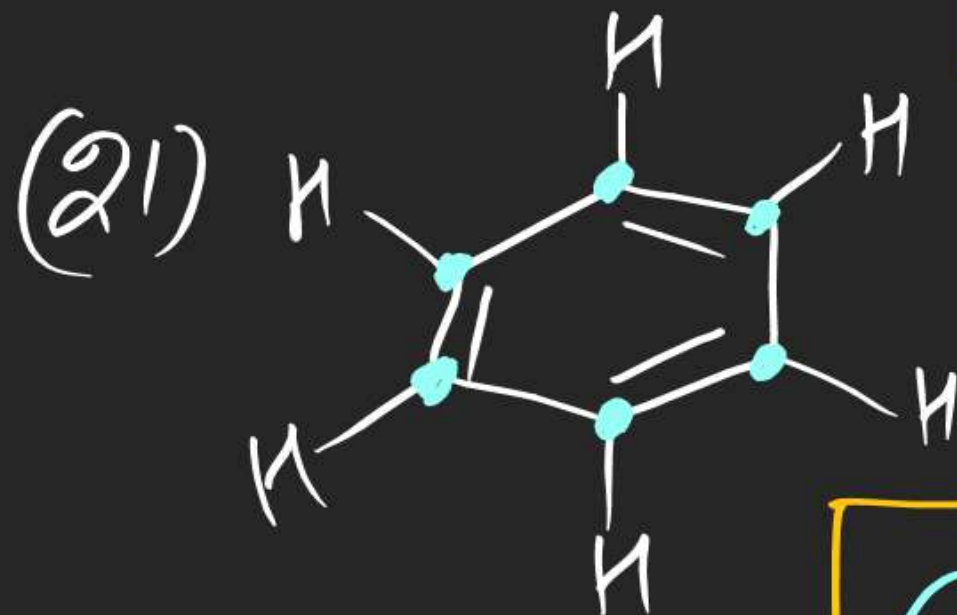
 $(T_R=2)$ Acyclic compound  
is obtained.

(c)

 $(T_R=2)$



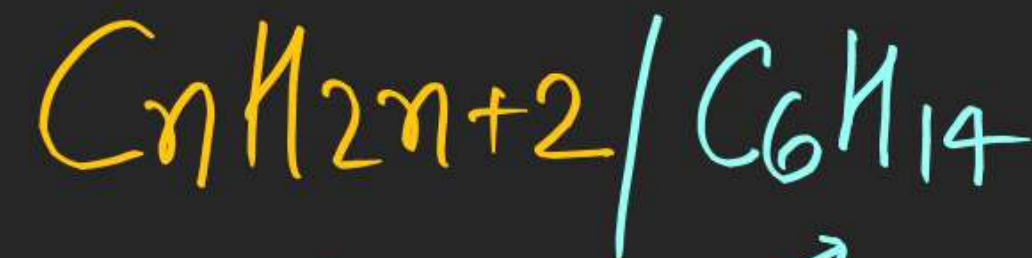
## Basic Organic Chemistry



$$\text{DBE} = 4$$



Alkane G.F



$$\Delta n_H = 8$$

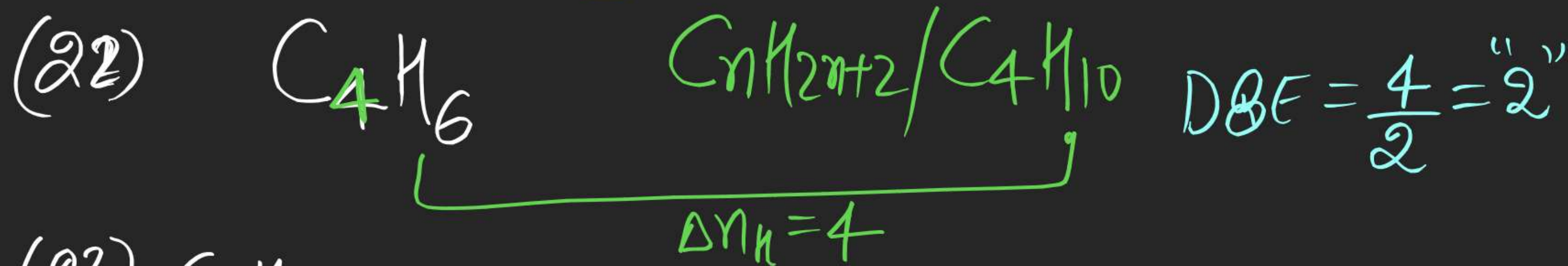
loss of 2H ..... gemmate = 1 DBE

loss 1H ..... = 1 DBE

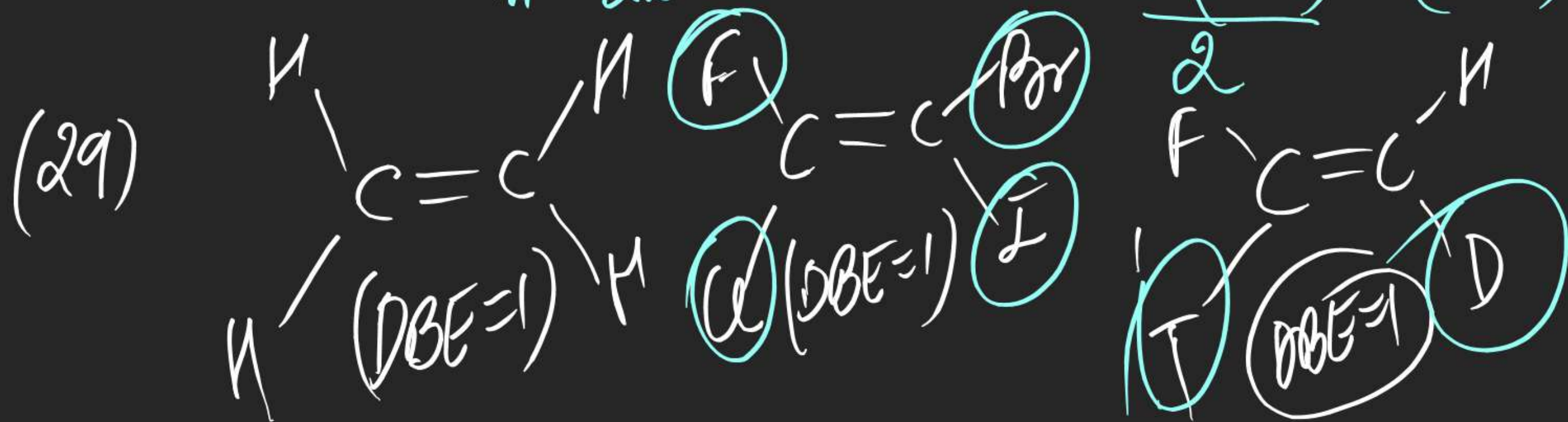
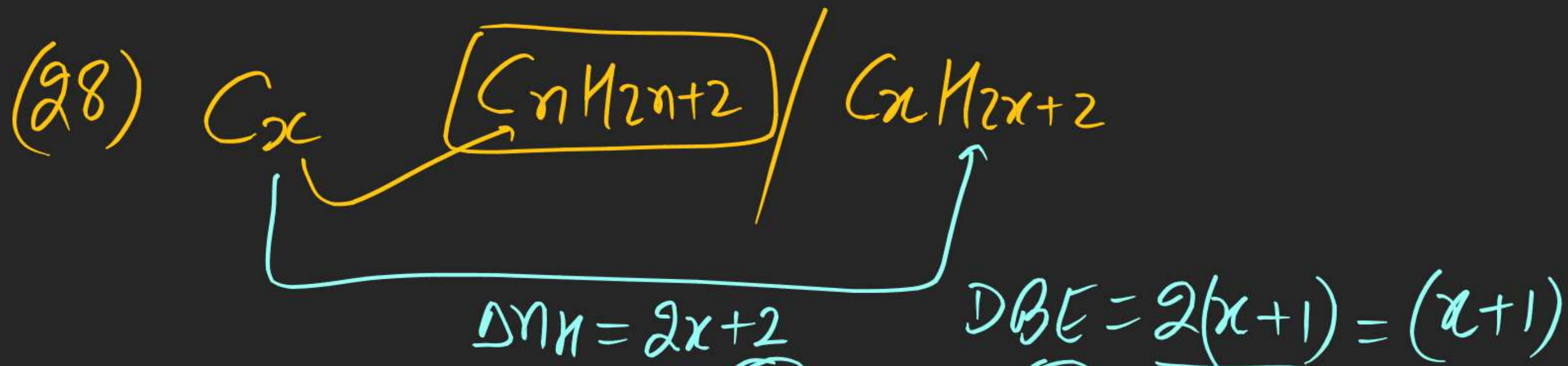
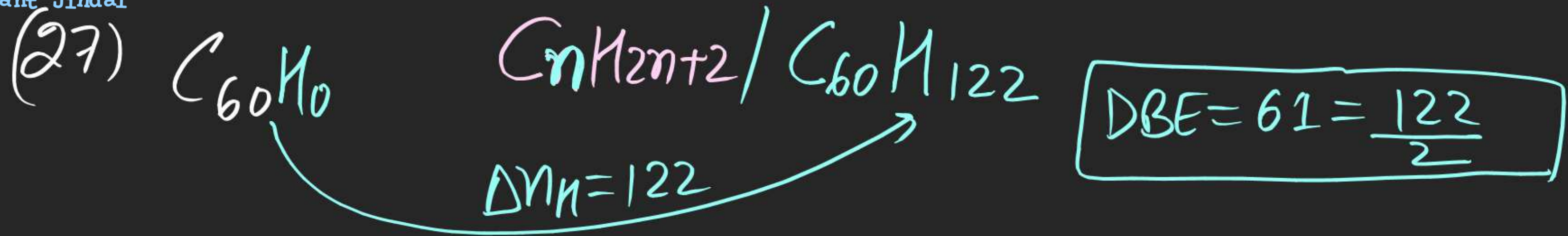
.....  $\Delta n_H$

$$\boxed{\frac{\Delta n_H}{2} = \text{DBE}}$$

$$DBE = \frac{Q}{2} = 4$$









## Basic Organic Chemistry

Case (i) If monovalent atoms are present  $\rightarrow (-D, -T, -F, -Cl, -Br, -I)$   
 then consider these atoms like "H"





## Basic Organic Chemistry

Sol<sup>n</sup> (26):mono  
valent

$\Delta n_H = 2$

$$\text{DBE} = \frac{\Delta n_H}{2} = \frac{2}{2} = 1$$

Sol<sup>n</sup>

$$\text{DBE} = T_R + T_\pi$$

$$\Rightarrow 3 = T_R + T_\pi$$

(37) DBE of a compound is "3"  
Then correct statement is/are

(A) Compound must have 3  $\pi$  Bonds

(B) Compound must have 3 Rings

(C) Compound must have 2 Rings & 1  $\pi$  Bond

Ans (D) All statement 'A' 'B' & 'C' is incorrect.



## Basic Organic Chemistry

(38) Compound  $\text{CH}_3-\text{CH}=\text{C}(\text{CH}_3)_n$  containing "21" DBE. Find 'n'.

(A) 21

(B) 22

(C) 23

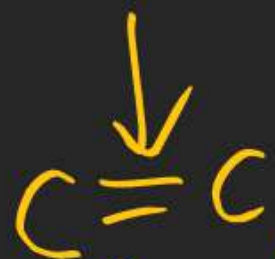
(D) 24

(E) NOT

19 DBE

DBE

1



2C

2



3C

⋮

20C

19

$n=20$

$$x+y+xy=7$$

$$y+z+yz=3$$

$$z+x+zx=1$$