

0-1 1-4

5-1 1-8

$$\text{Power} = 40 \times 0.8$$

①

$$\frac{\frac{4}{3}\pi r_p^3}{\frac{4}{3}\pi r_{\text{atom}}^3} = \frac{(1.5 \times 10^{-15})^3}{(0.05 \times 10^{-9})^3}$$

⑧

5000 Å

$$\left(\frac{hc}{\lambda} \times N_A \right)$$

$$= \frac{6.62 \times 10^{-34} \times 3 \times 10^8}{5000 \times 10^{-10}} \times N_A$$

(7)

$$\underline{330 \text{ Joule}} = h\nu \times \textcircled{N}$$

↑
no. of photon

$$mvr = \frac{nh}{2\pi} \quad \text{--- (1)}$$

$$r = \frac{n^2 h^2}{4\pi^2 K Z e^2 m}$$

$$= 0.529 \frac{n^2}{Z} \text{ \AA}$$

$$\frac{mv^2}{r} = \frac{K Z e^2}{r^2} \quad \text{--- (2)}$$

$$v = \frac{2\pi K Z e^2}{nh}$$

$$= 2.188 \times 10^6 \frac{Z}{n} \text{ m/sec}$$

for H, He^+, Li^{2+} ← single e^- system

$$TE = -13.6 \frac{Z^2}{n^2} \text{ eV}$$

$$KE = -TE$$

$$PE = 2TE$$

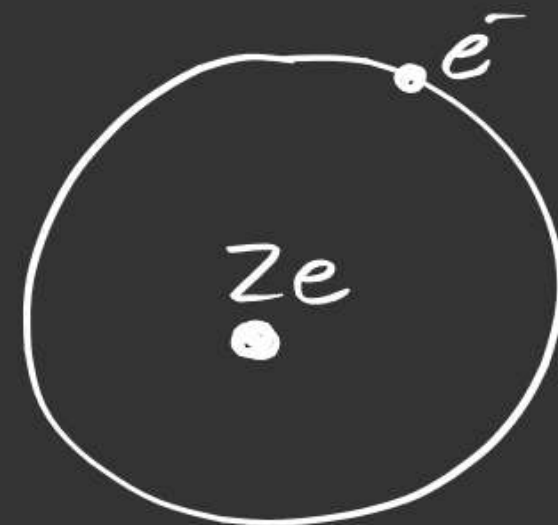
$$\begin{aligned}\text{Total Energy} &= \text{KE} + \text{PE} \\ &= \frac{1}{2}mv^2 - \frac{KZe^2}{r}\end{aligned}$$

$$\text{T.E} = \frac{1}{2} \left(\frac{KZe^2}{r} \right) - \frac{KZe^2}{r}$$

$$\text{TE} = -\frac{1}{2} \frac{KZe^2}{r}$$

$$\text{T.E} = - \frac{2\pi^2 k^2 Z^2 e^4 m}{n^2 h^2}$$

$$\text{TE} = -13.6 \frac{Z^2}{n^2} \text{eV}$$



$$U_r = \frac{Kq_1q_2}{r} + U_\infty$$

↑
PE at r distance

frequency (f) :- no. of round by e^- in 1 sec

$$f = \frac{v}{2\pi r}$$

$$f \propto \frac{Z/n}{n^2/Z}$$

$$\propto \frac{Z^2}{n^3}$$

$$\text{Time period (T)} = \frac{2\pi r}{v} = \frac{1}{f}$$

$$T \propto n^3/Z^2$$

$$\text{Electrostatic force} = \frac{KZe^2}{r^2} \propto \frac{Z \times Z^2}{n^4} = \frac{Z^3}{n^4}$$

Q. find the expression of r , v & T.E if

(i) mass of e^- is increased to 2 times

(2) if charge of e^- & proton are increased to 2 times.

(1) Ans $r = \frac{0.529}{2} \frac{n^2}{Z} \text{ \AA}$

$$v = 2.188 \times 10^6 \frac{Z}{n} \text{ m/sec}$$

$$T.E = -2 \times 13.6 \frac{Z^2}{n^2}$$

$$r = \frac{n^2 h^2}{4\pi^2 K Z e^2 m}$$

$$r = \frac{0.529}{4} \frac{n^2}{Z}$$

$$v = \frac{2\pi K Z e^2}{nh}$$

$$= 2.188 \times 10^6 \frac{Z}{n} \times 4$$

$$T.E = \frac{-2\pi^2 K^2 Z^2 e^4 m}{n^2 h^2}$$

$$= -13.6 \frac{Z^2}{n^2} \times 16$$

$$r_{n,z}$$

Q

find ratio of

$$(1) \frac{r_{2,2}}{r_{1,1}} \quad \left(\frac{2}{1} \right)$$

$$(2) \frac{v_{2,1}}{v_{1,2}} \quad \frac{2/n}{2/1} \quad \left(\frac{1}{4} \right)$$

$$(3) \frac{TE_{2,2}}{TE_{1,1}} \quad \frac{Z^2/n^2}{Z^2/n^2}$$

$$(4) \frac{KE_{2,1}}{KE_{1,1}} \quad \frac{1/4}{1/1}$$

$$(5) \frac{PE_{1,3}}{PE_{3,1}} \quad \frac{9/1}{1/9}$$

$$(6) \frac{f_{2,1}}{f_{1,1}} \quad \frac{1/8}{1/1} \quad \frac{Z^2/n^3}{Z^2/n^3}$$

$$(7) \frac{T_{2,2}}{T_{1,1}} = \frac{8/4}{1/1} \quad \frac{n^3/z^2}{n^3/z^2}$$

$$(8) \frac{(\text{Electrostatic force})_{2,2}}{(\text{Electrostatic force})_{1,1}} \quad \frac{2^3/n^4}{1^3/n^4} \quad \left(\frac{1}{2} \right)$$

$$(9) \frac{(\text{angular momentum})_{2,1}}{(\text{angular momentum})_{1,1}} \quad \frac{2}{1}$$

$$mvr = \frac{nh}{2\pi}$$

$\propto \frac{r}{n} \cdot \frac{n^2}{r}$

$$TE = -13.6 \frac{Z^2}{n^2}$$

TE of H

TE of He⁺

$$n=1$$

$$-13.6$$

$$\updownarrow 10.2$$

$$-54.4$$

$$n=2$$

$$-3.4$$

$$\updownarrow 1.89$$

$$-13.6$$

$$n=3$$

$$-1.51$$

$$\updownarrow 0.66$$

$$-6.04$$

$$n=4$$

$$-0.85$$

$$\updownarrow 0.31$$

$$-3.4$$

$$n=5$$

$$-0.54$$

$$-2.16$$

$$-1.51$$

$$n=\infty \quad TE=0$$

$$T.E = -13.6 \frac{Z^2}{n^2}$$

'H'

He⁺

Li²⁺

$$n=1$$

$$2$$

$$3$$

$$n=2$$

$$4$$

$$6$$

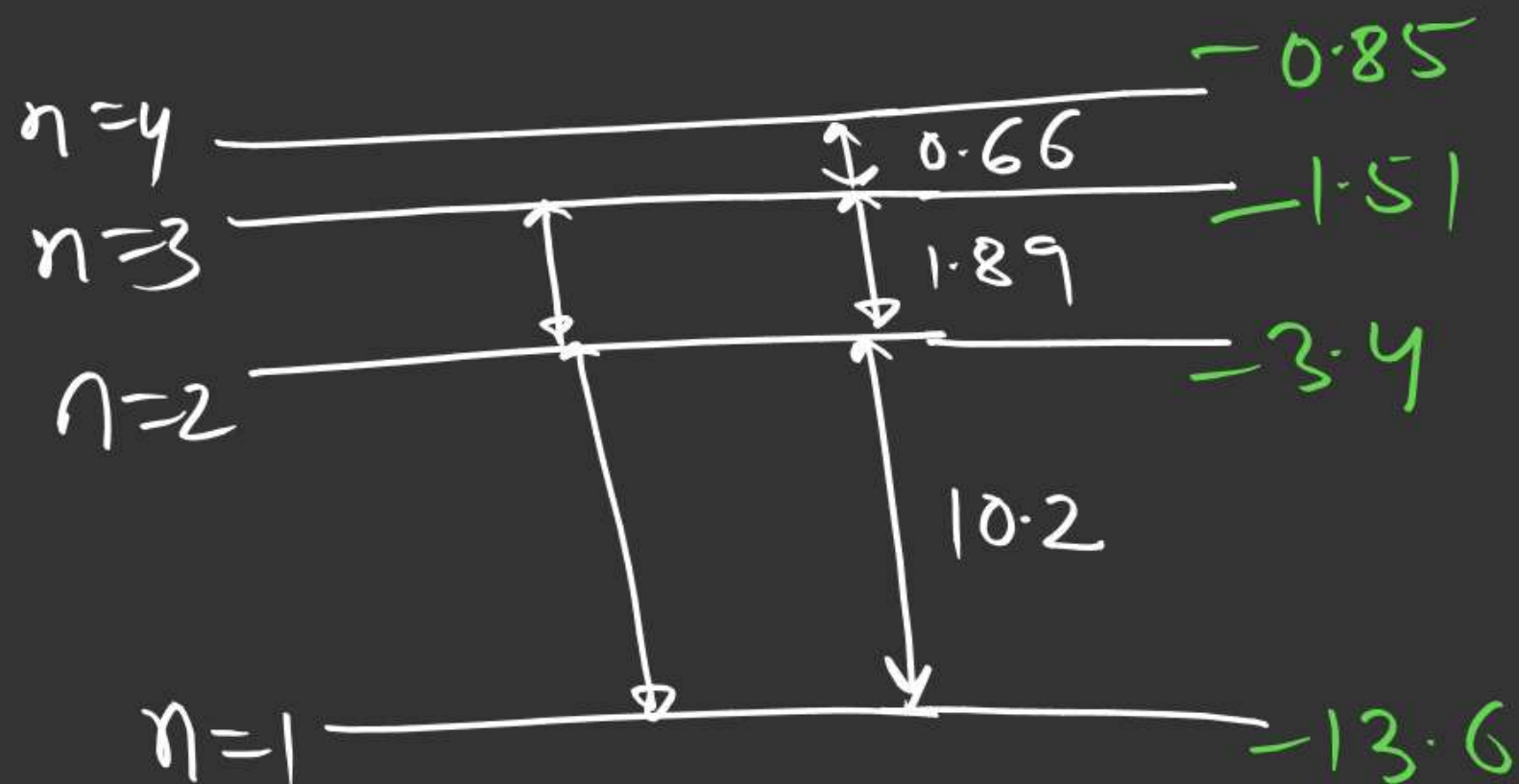
$$n=3$$

$$6$$

$$9$$

$$E_2 - E_1 > E_3 - E_2 > E_4 - E_3 > E_5 - E_4$$

$$n=\infty \text{ ————— } 0 \quad \bigcirc$$



$$E_2 - E_1 > E_{\infty} - E_2$$

$$= 10.2 \quad \underline{\underline{= 3.4}}$$

find Energy of photon emitted when an e^- jumps from

① $2 \rightarrow 1$ in 'H' atom

② $3 \rightarrow 2$ in "

③ $3 \rightarrow 1$ in "

④ $2 \rightarrow 1$ in He^+

⑤ $3 \rightarrow 1$ in He^+

$$0 - 1$$

$$5 - 11$$

$$5 - 1$$

$$9 - 16$$