H. atom Problem. H=(KE) + (KE) + V potential energy  $V = -\frac{e^{\tau}}{4\pi\epsilon_{\delta}\tau} \leftarrow \tau_{\alpha}\lambda_{i\omega}$   $\Psi(\tau, \theta, \phi) = R(\tau) \cdot Y(\theta, \phi)$ 

Internal PM,  $\frac{1}{\mu} = \frac{1}{m_1}$ WI - et 47 E, Y

$$\frac{1}{2^{n}} \nabla^{2} \Psi(r, \theta, \phi) - \frac{e^{2}}{4\pi \epsilon_{0}r} \Psi(r, \theta, \phi) = E \Psi(r, \theta, \phi)$$

$$\nabla^{2} = \frac{1}{r^{2}} \frac{\partial}{\partial r} \left(r^{2} \frac{\partial}{\partial r}\right) + \frac{1}{r^{2} \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta}\right)$$

$$\Psi(r, \theta, \phi) = N. R(r) \Psi(\theta, \phi) + \frac{1}{r^{2} \sin \theta} \frac{\partial}{\partial \phi}$$
Normally  $A_{r, m_{e}}$ 

R(r): Rentum Number (n)

$$R = 0$$
 $R = 0$ 
 $R =$ 

Rydberg const RH = 109679 cm 4TE. 1 = 0.53 A  $Q^{o} =$ 

Rng (r) = Nnhm, (1r / nao) Lng (e-r/nao)  $\langle r \rangle_n \sim n^2 a_0$ Bohr Radino Orbital (-7, 0, 0) = EY(0,0)  $Y_n E_n$ Degeneracy

Orbital 
$$\rightarrow \frac{e^{-}}{\pi} \frac{\text{Wave } f^{n}}{\sqrt{2\pi}} = \frac{vishalize}{\sqrt{2\pi}}$$

$$\int_{0}^{R} \int_{0}^{R} d\theta \int_{0}^{\sqrt{2\pi}} d\phi \left(r^{2}\sin\theta\right) = \frac{4}{3}\pi^{2}$$

$$|\psi|^{2} \qquad \qquad \forall \varphi \in \text{Ap} \left(-\frac{r}{a_{0}}\right)$$

$$|\psi|^{2} \qquad \qquad \psi_{100} \propto \left(2 - \frac{r}{a_{0}}\right)^{2\pi} \left(-\frac{r}{a_{0}}\right)$$

$$|\psi|^{2} \qquad \qquad \psi_{200} \propto \left(2 - \frac{r}{a_{0}}\right)^{2\pi} \left(-\frac{r}{a_{0}}\right)$$

$$n = 1$$
,  $l = 0$   $\longrightarrow$   $] S - orbital$ 
 $n = 2$ ,  $l = 0$   $\longrightarrow$   $2S - orbital$ 
 $l = 1$   $\longrightarrow$   $m_{k} = -1$   $\longrightarrow$   $p - orbital$ 
 $m_{k} = +1$   $\longrightarrow$   $p - orbital$ 
 $l = 0, 1, 2, \dots$   $(n-1)$   $\stackrel{n-1}{\sum}$   $(2k+1)$ 
 $m_{k} = 0, \pm 1, \pm 2, \dots$   $\pm 2$   $= n$