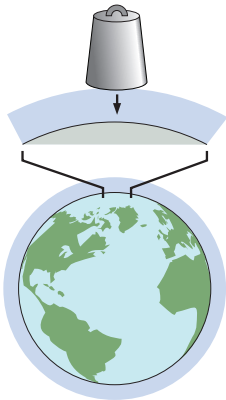


# WHAT IS THE MASS OF CONTEMPORARY CO<sub>2</sub>?



$$\begin{aligned}
 & \left\{ \begin{aligned} p_{\text{atmosphere}} &\approx 10^5 \text{ Pascal} \approx 10^5 \frac{\text{N}}{\text{m}^2} \\ A_{\text{earth}} &\approx f \times 10^{14} \text{ m}^2 \\ g &\approx 10 \frac{\text{m}}{\text{s}^2} \end{aligned} \right. \\
 & \downarrow \\
 & m_{\text{atmosphere}} \approx \frac{p_{\text{atmosphere}} \times A_{\text{earth}}}{g} \approx \frac{10^5 \text{ N}}{\text{m}^2} \times \frac{1 \text{ s}^2}{10 \text{ m}} \times f \times 10^{14} \text{ m}^2 \\
 & \quad \approx f \times 10^{18} \text{ kg} \\
 & \left\{ \begin{aligned} mw_{\text{air}} &\approx f \times 10^{-2} \frac{\text{kg}}{\text{mol}} \approx f \times 10^{25} \frac{\text{molecules}}{\text{kg}} \\ c_{\text{CO}_2} &\approx \frac{420 \text{ CO}_2 \text{ molecules}}{10^6 \text{ air molecules}} \end{aligned} \right. \\
 & \downarrow \\
 & N_{\text{CO}_2} \approx m_{\text{atmosphere}} \times mw_{\text{air}} \times c_{\text{CO}_2} \\
 & \quad \approx f \times 10^{18} \text{ kg} \times \frac{f \times 10^{25} \text{ molecules}}{\text{kg}} \times \frac{420 \text{ CO}_2}{10^6 \text{ molecules}} \\
 & \quad \approx f \times 10^{40} \text{ CO}_2 \text{ molecules} \\
 & \left\{ \begin{aligned} mw_{\text{CO}_2} &\approx 0.044 \frac{\text{kg}}{\text{mol}} \approx 10^{25} \frac{\text{molecules}}{\text{kg}} \end{aligned} \right. \\
 & \downarrow \\
 & m_{\text{CO}_2} \approx \frac{N_{\text{CO}_2}}{mw_{\text{CO}_2}} \approx \frac{f \times 10^{40} \text{ CO}_2 \text{ molecules}}{10^{25} \text{ molecules / kg}} \approx f \times 10^{15} \text{ kg CO}_2
 \end{aligned}$$