## FIRST SEMESTER

ROH No. UN 23

B. Tech. (Group-A)

## END SEMESTER EXAMINATION

NOVEMBER-2010

AC-104 APPLIED CHEMISTRY

Time: 3 Hours

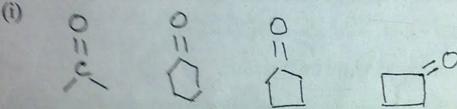
Max. Marks

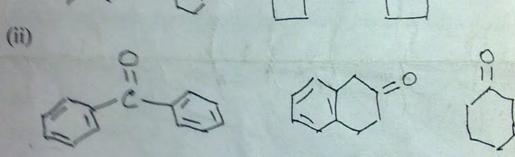
70

Note: Answer any FIVE questions.

Assume suitable missing data, if any.

1[a] Arrange the following molecules in their increasing order of stretching frequency of > C = 0. Discuss the reason.





- [b] Classify the polymers on the basis of stereochemistry. Explain with the help of suitable examples.
- [c] Write down the chemical structure of following indicators used in volumetric analysis. Are these water soluble? Explain in detail.
  - (i) Phenolphthalein
  - (ii) Methyl orange
  - (iii) EBT

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2[a] 25 ml of a solution containing Na<sub>2</sub>CO<sub>3</sub> and NaHCO<sub>3</sub> requires 5ml of 0.1 N HCl for titration using phenolphthalein as an indicator. The titration was repeated with the same volume of this solution but using methyl orange as an indicator requires only 12.5 ml of 0.1 N HCl for titration. Calculate the amount of both Na<sub>2</sub>CO<sub>3</sub> and NaHCO<sub>3</sub> in the solution.

6[2

D

[c

of green chemistry. Explain atleast one with examples.

