Subiectul B. ELEMENTE DE TERMODINAMICĂ

II.a.	MENTE DE TERMODINAMIOA
	$p_2V_2 = \vartheta_2RT$
	$T = T_0 + \theta$
	Rezultat final: ϑ_2 = 2,5 mol
b.	
	$\mu = \frac{m_{\text{amestec}}}{\vartheta_{\text{amestec}}}$
	$m_{\text{amestec}} = m_1 + m_2$
	$m_1 = \vartheta_1 \mu_1, \ m_2 = \vartheta_2 \mu_2$
	$\vartheta_{\text{amestec}} = \vartheta_1 + \vartheta_2$
	Rezultat final: μ ₂ ≈16 g/mol
C.	
	$ ho_{ m amestec} V_{ m amestec} = artheta_{ m amestec} RT$
	$V_{\text{amestec}} = V_1 + V_2$ şi $\vartheta_{\text{amestec}} = \vartheta_1 + \vartheta_2$
	Rezultat final: $p_{\text{amestec}} = 5,05 \cdot 10^5 \text{ Pa}$
d.	
	$\Delta m = \vartheta_1 \mu_1 - \vartheta_1 \mu_1$
	$\Delta m = \vartheta_1 \mu_1 - \vartheta_1 \mu_1$ $\Delta m = \frac{\vartheta_{am}}{2} \mu - \vartheta_1 \mu_1$
	Rezultat final: $\Delta m = 19,95 \text{ g}$